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**Decoupling peroxyacetyl nitrate from ozone in Chinese outflows observed
at Gosan Climate Observatory**

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26 **Abstract**

27 We measured peroxyacetyl nitrate (PAN) and other reactive species such as O₃, NO₂, CO,
28 and SO₂ with aerosols including mass, organic carbon (OC), and elemental carbon (EC) in
29 PM_{2.5} and K⁺ in PM_{1.0} at Gosan Climate Observatory in Korea (33.17°N, 126.10°E) during
30 October 10 to November 6, 2010. PAN was determined through fast gas chromatography
31 with luminol chemiluminescence detection at 425 nm every 2 min. The PAN concentrations
32 ranged from 0.1 (detection limit) to 2.4 ppbv with a mean of 0.6 ppbv. For all measurements,
33 PAN was unusually better correlated with PM_{2.5} (Pearson correlation coefficient, $\gamma = 0.79$)
34 than with O₃ ($\gamma = 0.67$). In particular, the O₃ level was highly elevated with SO₂ at midnight,
35 along with a typical midday peak when air was transported rapidly from the Beijing areas.
36 The PAN enhancement was most noticeable during the occurrence of haze under stagnant
37 conditions. In Chinese outflows slowly transported over the Yellow Sea, PAN gradually
38 increased up to 2.4 ppbv at night, in excellent correlation with a concentration increase of
39 PM_{2.5} OC and EC, PM_{2.5} mass, and PM_{1.0} K⁺. The high K⁺ concentration and OC/EC ratio
40 indicated that the air mass was impacted by biomass combustion. This study highlights PAN
41 decoupling with O₃ in Chinese outflows and suggests PAN as a useful indicator for
42 diagnosing continental outflows and assessing their perturbation on regional air quality in
43 northeast Asia.

44

45 Key words: PAN, O₃, PM_{2.5}, Chinese outflow, Haze, Biomass combustion

46 1. Introduction

47 At the surface, ozone is primarily photochemically produced, and the contribution from the
48 stratosphere is generally small. Ozone is formed through reactions of various precursors such
49 as CO, CH₄, volatile organic compounds (VOCs), and NO_x (e.g., Brasseur et al., 1999; Jacob,
50 2000; Nielsen et al., 1981). Likewise, peroxyacetyl nitrate (PAN) is a secondary product of
51 urban air pollution and a significant oxidant in the atmosphere (e.g., Hansel and Wisthaler,
52 2000; La Franchi et al., 2009; Lee et al., 2012; Liu et al., 2010; Roberts et al., 2007). PAN is
53 solely produced by the photochemical reaction between the peroxyacetyl radical and nitrogen
54 dioxide, and the peroxyacetyl radical is derived from the OH oxidation or photolysis of
55 VOCs such as acetaldehyde, methylglyoxal, and acetone (e.g., Fischer et al., 2014; La
56 Franchi et al., 2009; Lee et al., 2012). For this reason, PAN is a very useful indicator of
57 photochemical air pollution. As thermal decomposition is a major PAN sink in the
58 troposphere (Beine et al., 1997; Jacob, 2000; Kenley and Hendry, 1982; Talukdar et al., 1995),
59 the lifetime of PAN depends on temperature. For example, the PAN lifetime is ~5 years at
60 -26°C and 1 h at 20°C (Fischer et al., 2010; Zhang et al., 2011). At high altitudes above ~7
61 km, photolysis becomes the most important loss process for PAN (Talukdar, et al., 1995).
62 Besides, PAN is less soluble compared to nitric acid and is more easily transported to the free
63 troposphere after it is released from scavenge in lower temperature (e.g., Zhu et al., 2017).
64 Thus, PAN can be an indicator of NO_y concentration in the free troposphere and a guide for
65 the long-range transport of NO_x in remote regions (Jacob, 1999).

66 In the past decades, PAN was measured not only in urban areas (Aneja et al., 1999;
67 Gaffney et al., 1999; Grosjean et al., 2002; Lee et al., 2008; Tanimoto et al., 1999; Zhang et
68 al., 2014) but also in background regions (Fischer et al., 2011; Kanaya et al., 2007; Lee et al.,
69 2012; Tanimoto et al., 2002), onboard aircraft (Tereszchuk et al., 2013), and ships (Roberts et

70 al., 2007). PAN concentrations were in the range of a few ppbv in urban areas close to VOCs
71 and NO_x sources (Lee et al., 2008; Zhang et al., 2011). In [the most](#) remote regions, PAN
72 [concentrations](#) were generally in the range of a few pptv (Gallagher et al., 1990; Mills et al.,
73 2007; Muller and Rudolph, 1992; Staudt et al., 2003).

74 [Although NO_x concentration has recently declined in China \(Gu et al., 2013; Liu et al.,](#)
75 [2016a; Krotkov et al., 2016\), NO_x and VOCs have gradually increased in East Asia,](#)
76 [particularly China during the last couple of decades \(Akimoto, 2003; Liu et al., 2010; Ohara](#)
77 [et al., 2007; Zhao et al., 2013\). It led to an increase in the concentrations of photochemical](#)
78 [byproducts such as PAN and O₃ not only in East Asia \(Liu et al., 2010; Wang et al., 2010;](#)
79 [Zhang et al., 2009; Zhang et al., 2011; Zhang et al., 2014\) but also in North America \(Fischer](#)
80 [et al., 2010; Fischer et al., 2011; Jaffe et al., 2007; Zhang et al., 2008\). These results were](#)
81 [also demonstrated by the GEOS-Chem model \(Zhang et al., 2008\). In addition to urban](#)
82 [plumes, PAN was reported to be enhanced by biomass combustion \(Alvarado et al., 2010;](#)
83 [Coheur et al., 2007; Zhu et al., 2015; Zhu et al., 2017\), such as open burning and use of](#)
84 [biofuel, which is used to take place often in China after crop harvesting \(Cao et al., 2006;](#)
85 [Duan et al., 2004\). Recent satellite studies have also observed the increased PAN in plumes](#)
86 [associated with anthropogenic emissions in eastern China and boreal fires in Siberia \(Zhu et](#)
87 [al., 2015; Zhu et al., 2017\). In this context, PAN is a useful indicator for diagnosing Chinese](#)
88 [outflows and assessing their perturbation on regional air quality in the northwestern Pacific](#)
89 [region.](#)

90 Gosan Climate Observatory (GCO) is an ideal place to monitor Asian outflows and their
91 transformation and to estimate their impact on air quality over the northern Pacific region
92 (Lee et al., 2007; Lim et al., 2012). In the present study, PAN was first measured
93 continuously at GCO to characterize its variation and source in relation to O₃ and to

94 understand the influence of Chinese outflows on the regional air quality.

95

96 **2. Experiments**

97 PAN measurements were conducted at GCO (33.17°N, E126.10°E) on Jeju Island from
98 October 19 to November 6, 2010. GCO is located on a cliff at the western edge of Jeju Island.

99 PAN was determined through fast gas chromatography (GC) with luminol
100 chemiluminescence detection, which is described in detail elsewhere (Gaffney et al., 1998;
101 Lee et al., 2008; Marley et al., 2004). Here, we briefly describe the measurement method.

102 Ambient air was pumped through a 1.6-m PFA tubing (1/4 inch outer diameter) from the
103 roof of the two-story container into a six-port two-position switching injection valve
104 (Cheminert C22, Valco Instruments (Houston, TX, USA)) at 100 ml/min controlled by Mass
105 flow controller (Lee et al., 2012; Lee et al., 2008). The residence time of the inlet was less
106 than 2 seconds. PAN and NO₂ (and peroxypropyl nitrate (PPN) if present) were separated
107 along a 10-m capillary GC column (DB-1, J&W Scientific, Folsom, CA, USA), whose end
108 was connected to a luminol cell where the column effluent reacted with luminol, giving off
109 luminescent light (Lee et al., 2008; Lee et al., 2012). The concentrations of PAN and other
110 species were determined from the chemiluminescence signals detected by a gated photon
111 counter (HC135-01, Hamamatsu, Bridgewater, NJ, USA) at 425 nm, which was set at 800 V
112 and operated at room temperature (Gaffney et al., 1998; Lee et al., 2012; Lee et al., 2008).

113 PAN was calibrated against standards synthesized by the nitration of peracetic acid in n-
114 tridecane (Gaffney et al., 1984; Gregory, 1990; Lee et al., 2008). A few microliter aliquots of
115 standard solution were injected through an injection valve and then mixed with zero air
116 (99.999 %) in a 5 L Tedlar bag. After being left for a few minutes for equilibrium, it was
117 injected into GC-luminol instrument and NO_x chemiluminescence instrument with a

118 molybdenum converter (42C, Thermo Electron Corporation, Franklin, MA, USA). The
119 calibration was completed within 5 minutes to prevent thermal decomposition of the PAN
120 (Kourtidis et al., 1993; Lee et al., 2008). These calibration procedures were carried out on the
121 assumption that the PAN was completely converted to NO in the molybdenum converter. The
122 detection limit of PAN defined by 3σ of the lowest standard was no greater than 100 pptv
123 (Lee et al., 2008). The overall measurement uncertainty and precision was estimated to be 16 %
124 and 5%, respectively (Lee et al., 2012). NO_x instrument was calibrated with NO standard gas.

125 Gaseous species including O₃, NO, NO₂, CO, and SO₂ were measured by UV absorption,
126 chemiluminescence with a molybdenum converter, non-dispersive infrared, and pulse UV
127 fluorescence method, respectively (NIER, 2016a). The measurements were made in
128 compliance with guidelines for installation and operation of an air pollution monitoring network
129 (NIER, 2016b). Calibration was conducted before and after the experiment, following the
130 regular checkup procedure. Detection limits of O₃, NO_x, CO, and SO₂ are 2 ppb, 0.1 ppb,
131 0.05 ppm, 0.1 ppb, respectively (NIER, 2016b).

132 Aerosol species, including PM_{2.5} mass and PM_{2.5} OC and EC were measured and recorded
133 along with meteorological parameters (relative humidity, temperature, and wind speed).
134 Water-soluble ions of PM_{1.0} were collected by a particle-into-liquid sampler and analyzed by
135 ion chromatography. The detailed results of the aerosol measurements can be found in Shang
136 et al. (2017).

137 For the air parcel at 850 m a.s.l., the three-day backward trajectories were calculated every
138 hour using NOAA Air Resources Laboratory (ARL) Hybrid Single-Particle Lagrangian
139 Integrated Trajectory (HYSPLIT) model (version 4) (Draxler and Rolph, 2012; Rolph, 2012,
140 <http://www.arl.noaa.gov/ready/hysplit4.html>). In addition, O₃ and PAN concentrations were
141 calculated using a global chemistry model, the Community Atmosphere Model with

142 Chemistry (CAM-Chem), a component of the Community Earth System Model (CESM)
143 (Lamarque et al., 2012; Tilmes et al., 2015). The CAM-chem results shown here follow the
144 configuration used for the HTAP2 (Hemispheric Transport of Air Pollution, Phase 2)
145 intercomparison (e.g., Stjern et al., 2016). CAM-chem is nudged to observed meteorology
146 (GEOS-5) to reproduce the actual period of the observations (Oct 2010). The emissions used
147 in the model are the HTAP2 inventory (Janssens-Maenhout, et al., 2015), which include the
148 "MIX" Asian emissions inventory. Biomass burning emissions are from the Global Fire
149 Emissions Database (GFED3) (Randerson et al., 2013).

150

151 **3. Results**

152 In the present experiments, PAN concentrations range from 0.1 to 2.4 ppbv, with an
153 average of 0.6 ppbv. This mean value is lower than those observed in other Asian megacities:
154 Beijing (1.41 ppb in the summer), Pearl River Delta region (1.32 ppb in the summer), and
155 Seoul (0.8 ppb in the early summer); similar to those of suburban areas in China, e.g.,
156 Lanzhou (0.76 ppb in the summer); and higher than those of urban and rural sites in Japan,
157 e.g., Tokyo (up to 0.6 ppb in the fall), Rishiri Island (~0.5 ppb in spring) or in the western
158 coast of the US, e.g., Sacramento (0.45 ppb in the summer), Mt. Bachelor (0.144 ppb in the
159 spring and early summer), off the western coast of the US (0.65 ppb in the spring), and over
160 the remote North Pacific (total PAN < 0.3 ppb in spring) (Bertram et al., 2013; Fischer et al.,
161 2011; La Franchi et al., 2009; Lee et al., 2008; Roberts et al., 2004; Tanimoto et al., 1999;
162 Tanimoto et al., 2002; Wang et al., 2010; Zhang et al., 2009; Zhang et al., 2011). Because the
163 PAN lifetime is greatly dependent on temperature, its concentration decreases with increasing
164 distance from the source regions. The PAN concentrations calculated in this study thus lie in-
165 between the levels for the East Asian megacities and the northern Pacific. The distributions of

166 all measured species, including PAN and O₃, are presented in Fig. 1. In particular, there are
167 several periods characterized by high concentrations of PAN, O₃, and PM_{2.5}. In terms of PAN,
168 four periods are particularly interesting (Fig. 1). High O₃ concentrations were observed
169 during October 31–November 2 [episode 1] but did not coincide with high PAN
170 concentrations. During October 28–29 [episode 2], NO₂ was noticeably increased. On the
171 other hand, PAN and O₃ concentrations were both high during October 20–21 [episode 3] and
172 November 4–5 [episode 4]. Episodes 3 and 4 are characterized by haze, while episodes 1 and
173 2 are characterized by urban influence in the Korean and Beijing outflows, respectively. Haze
174 is reported by Korea Meteorological Administration (KMA) as a meteorological phenomenon
175 when visibility is 1~ 10 km and relative humidity is less than 75 %.

176 In the present study, PAN correlates reasonably well with O₃ ($\gamma = 0.67$) and even better
177 with PM_{2.5} ($\gamma = 0.79$). In general, O₃ and PAN exhibit typical diurnal variation with a
178 maximum recorded in the afternoon, which results in a good correlation between the two
179 (Brasseur et al., 1999; Gaffney et al., 1999; Ridley et al., 1990; Schrimpf et al., 1995; Wang
180 et al., 2010). In this study, however, the O₃ peak was often found in the early morning and
181 late afternoon for several days (Fig. 1). Observing the diurnal variations in the entire PAN
182 concentration measurement set (Fig. 2), the maximum was clearly recorded in the morning
183 with the highest outliers, which is rather similar to that of PM_{2.5}. The diurnal pattern of NO₂
184 shows little variation, even though its concentrations were increased in the morning along
185 with PAN. This first measurement of PAN at GCO reveals that PAN is not always coupled
186 with O₃, which was not typically observed at remote sites in previous studies (e.g., Fischer et
187 al., 2010; Lee et al., 2012).

188

189 4. Discussion

190 **4.1. Decoupling of PAN from O₃**

191 To examine the detailed mechanism of the decoupling of PAN from O₃, the daily
192 maximum concentrations of PAN and O₃ were further explored. The recorded daily PAN
193 maxima were generally in good correlation with O₃, albeit the relationship did not seem to
194 hold at high concentrations of PAN and O₃ (Fig. 3). The daily maxima were then categorized
195 into four groups according to the time when each O₃ and PAN maximum was recorded: “O₃
196 day-PAN day,” “O₃ day-PAN night,” “O₃ night-PAN day,” and “O₃ night-PAN night.” The
197 day interval started from 08:00 and ended at 18:00 (local time), based on the times of sunrise
198 and sunset during the experiment period. While the high PAN concentrations were associated
199 with the “O₃ day-PAN day” group (cross symbols in Fig. 3), the enhanced O₃ concentration
200 was recorded in the “O₃ night-PAN night” group (star symbols in Fig. 3). The “O₃ night-PAN
201 night” group unexpectedly held more data points than the “O₃ day-PAN day” group, even
202 though the “O₃ night-PAN night” group concentrations were lower (Fig. 3). In addition, there
203 were several days classified in the “O₃ night-PAN day” (marked by diamond) and “O₃ day-
204 PAN night” groups, but with less frequency and lower concentrations. These results indicate
205 that the decoupling of PAN from O₃ was primarily due to the elevated concentrations of O₃
206 and PAN at night. While PAN reached the maximum during the day on Oct 20 and Nov 5,
207 their concentrations were increased from the previous day through the night. The four high
208 PAN and O₃ episodes identified in this study fall under the category of “O₃ night-PAN night”
209 or “O₃ day-PAN day”. These two cases will be further examined to identify the chemical and
210 physical processes responsible for PAN being decoupled from O₃, instead of being coupled
211 with PM_{2.5}. The overall characteristics of the four episodes are summarized in Table 1.

212

213 **4.2. Export of O₃ from Asian continents (episodes 1 & 2)**

214 High O₃ concentrations were encountered around midnight on three consecutive days from

215 October 31 to November 2 (episode 1), during which SO₂ reached its maximum
216 concentration (Fig. 1). The backward trajectories of air masses revealed that air passed
217 through the Beijing area during this period (Fig. 4). The wind was strong (13.5 m/s on
218 average) and the recorded O₃ maximum (80.6 ppbv) was concurrent with the PAN maximum
219 (0.9 ppbv) around midnight on November 1st (Fig. 1).

220 All these results indicate that the air was heavily influenced by outflow from the Beijing
221 area, as previously hypothesized (Lim et al., 2012), and that the nighttime enhancement of O₃
222 and PAN resulted from the fast transport of relatively less-aged urban plumes. Because the
223 overall correlation between O₃ and PAN was the best with the highest daily $\Delta O_3/\Delta PAN$
224 among all cases discussed in this study, episode 1 likely represents an event of rapid transport
225 from the Beijing area (Fig. 5a).

226 In previous studies, the nighttime enhancement of O₃ was observed at GCO (e.g., Lee et
227 al., 2007) in association with pollutant-laden air coming from Beijing. Similarly, Banta et al.
228 (1998) pointed out that the evening O₃ maximum was due to long-range transport of O₃ from
229 nearby urban areas. Wang et al. (2011) reported that the O₃ lifetime was about two days in
230 East China during the summer, which is sufficient time for air to travel to GCO. Therefore,
231 the nighttime maximum of O₃ can be attributed to the export of O₃ from megacities in China.
232 It causes PAN to be decoupled from O₃ because PAN levels remained low, even though there
233 was good correlation between the two species. Another night maximum of O₃ was recorded
234 on October 29. Note that NO_x was highly elevated during October 28–29 (episode 2) (Fig. 1).
235 However, O₃ level was relatively low, leading to the lowest daily $\Delta O_3/\Delta PAN$ among all
236 episodes. In this case, air masses passed through the Korean Peninsula, carrying low O₃ being
237 titrated by high NO_x (Brasseur et al., 1999; Jacobson, 2005). These two urban plumes are
238 well contrasted in terms of O₃ and NO_x levels (Table 1), depending on the degree of aging.

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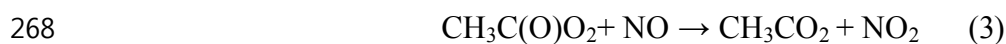
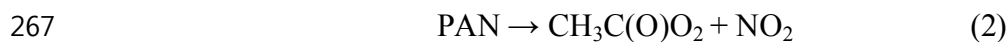
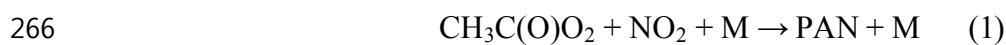
240 **4.3. PAN enhancement upon occurrence of haze (episodes 3 & 4)**

241 In this study, two haze events were observed in the very beginning (October 20–21;
242 episode 3) and the end of the study period (November 4–5; episode 4). As the nighttime O₃
243 peak was attributed to the transport from nearby urban areas to Jeju Island, the two haze
244 episodes were also observed in association with continental outflows. The first haze event
245 occurred on October 18th and lingered until October 21st, during which O₃ concentrations
246 were gradually elevated. A second peak was recorded around midnight of October 19th and
247 20th, and the maximum was reached in the afternoon of October 20th (Figs. 1 and 3). In this
248 episode, the maximum concentrations of O₃ and PAN were 78.9 ppbv and 2.0 ppbv,
249 respectively, on October 29th, when the highest NO₂ concentration (12.7 ppbv) was observed
250 under low wind speed (6.6 m/s daily average). The air mass trajectories suggest the influence
251 of the Korean Peninsula, particularly the Seoul metropolitan area, in addition to East China
252 (Fig. 4).

253 In the second haze event (episode 4), an air mass was slowly transported from East China,
254 including the Jiangsu province, under stagnant condition which was developed by an
255 anticyclone system (Fig. 4). We measured the highest concentrations of all aerosol species
256 including the PM_{2.5} mass as well as PAN and O₃, which were 156 μg/m³, 2.4 ppbv, and 87.5
257 ppbv, respectively. Other reactive gases such as CO, SO₂, and NO₂ were also highly elevated.
258 Note that PAN and O₃ gradually increased through the night, leading to a nighttime maximum
259 of both species on November 4th. It is likely that the pre-formed PAN and O₃ were
260 continuously transported into Gosan at night.

261 PAN is formed through the reaction of the peroxyacetyl radical and nitrogen dioxide (Eq. 1)
262 and decomposed at high temperature (Eq. 2), returning these radicals. Unless the NO

263 concentration is high (Eq. 3), the peroxyacetyl radical recombines with NO₂, producing PAN.
264 Thus, the total lifetime of PAN depends on the NO₂/NO ratio and temperature (Eq. 4)
265 (Brasseur et al., 1999).



$$269 \quad T_{eff} = T_d \left(1 + \frac{k_1[\text{NO}_2]}{k_2[\text{NO}]} \right) [\text{sec}^{-1}] \quad (4)$$

270 where T_d and T_{eff} indicate the lifetime against decomposition and the effective lifetime of
271 PAN (Brasseur et al., 1999). The effective lifetime of PAN was estimated through Eq. 4 using
272 the rate constants proposed by Brasseur et al. (1999), Jacobson (2005), and Maricq and
273 Szente (1996).

274 During the haze event, NO was close to the detection limit, while NO₂ was greatly
275 enhanced. Owing to the high NO₂/NO ratio, the effective lifetime of PAN increased by 57
276 times; this possibly contributed to the gradual increase in PAN through the night on
277 November 4th. Fischer et al. (2014) also reported that, at night, PAN can be produced from
278 the reaction of acetaldehyde with the nitrate radical.

279 Besides PM_{2.5}, PAN was also well correlated with PM_{2.5} OC and EC not only during this
280 haze episode but also during the entire measurement period (Fig. 6a and b). Furthermore, the
281 enhancement of PAN was concurrent with that of OC and K⁺, resulting in excellent
282 correlation between them (Fig. 6b and d). In fact, the ΔOC/ΔEC ratio of episode 4 was much
283 higher (7) than those of the other episodes (~2.5) (Fig. 6c). The fraction of PM_{2.5} against
284 PM_{2.5} was also the highest in this episode, indicating significant contribution of secondary
285 aerosols. These observations suggest that air masses were affected by biomass combustion
286 (e.g., Ram et al., 2008, 2012; Saarikoski et al., 2008).

287 According to previous studies, PAN can be produced in plumes through biomass

288 combustion (Alvarado et al., 2010; Coheur et al., 2007; Liu et al., 2016b; Tereszchuk et al.,
289 2013). In northeast China, open burnings related to agricultural activities frequently occur
290 during the spring and fall (Duan et al., 2004; Yang et al., 2005). Kudo et al. (2014) also
291 reported that, upon burning crop residue in Yangtze region, the levels of oxygenated VOCs
292 were elevated together with NO_x. In addition, biofuel is used for cooking and heating and as
293 an energy source in China's industry (Cao et al., 2006).

294 Therefore, PAN is likely to increase when haze occurs and fine aerosols are transformed as
295 air masses carrying combustion emissions are slowly transported from China over the Yellow
296 Sea. Additionally, the results of this study imply that PAN can be used as a robust tracer for
297 continental outflows in northeast Asia, to identify transport- and chemical transformation-
298 dominant regimes. In a transport-dominant regime, O₃ export was distinguished by the
299 highest levels of primary gaseous species such as SO₂ and relatively low levels of PAN. In
300 contrast, fine aerosol species are enhanced in a chemical transformation regime, leading to
301 haze events with relatively more enhanced PAN compared to O₃.

302 Finally, the measured O₃ and PAN concentrations were compared to results from a global
303 chemistry model CAM-Chem. In the model simulation, O₃ and PAN were highly
304 underestimated during the episodes observed in Chinese outflows, although the variation
305 around average level of O₃ and PAN was well captured (Fig. 7). The elevated PAN
306 concentration was underestimated in the model (Oct 20–21 and Nov 4–5), especially when air
307 was impacted by biomass combustion. The timing of the O₃ diurnal variability was captured
308 by the model, although the magnitude of the variation was underestimated. These results
309 reveal that the current understanding of Chinese outflow is still not sufficient, thereby causing
310 uncertainty in estimating its effect on air quality in the northwestern Pacific Rim.

311 5. Conclusions

312 The first measurements of PAN, reactive gases, and aerosol species were conducted at
313 GCO during October 19 to November 6, 2010. The average concentration of PAN was 0.6
314 ppbv with a maximum of 2.4 ppbv, which was lower than those in major cities in East Asia
315 but much higher than the background concentrations in other regions. In addition, PAN and
316 O₃ concentrations were well correlated ($\gamma = 0.67$). However, the comparison of the daily
317 maxima of PAN and O₃ highlighted that they were not proportionally enhanced. That is,
318 either PAN was relatively more elevated than O₃ or the highest O₃ was associated with low
319 levels of PAN. Unexpectedly, both PAN and O₃ often reached their maxima at night. As a
320 result, PAN was decoupled from O₃ and better correlated with the PM_{2.5} mass ($\gamma = 0.79$) than
321 with O₃. In this study, these high-concentration episodes were all encountered in association
322 with continental outflows, and thus, two high-O₃ and two high-PAN events were recorded
323 and investigated in detail.

324 During the O₃ episodes, both O₃ and PAN concentrations reached their maximum values at
325 night. In episode 1 (Oct. 31 to Nov. 2), the O₃ concentration was increased to 80.6 ppbv, with
326 a high SO₂ concentration under strong wind. It was a typical Beijing plume observed in the
327 study region. In comparison, NO₂ was greatly increased in episode 2 (Oct. 28–29) when the
328 air masses were affected by urban emissions from Korean Peninsula. Although the maximum
329 O₃ level was lower during episode 2, these two cases demonstrated well how O₃ was exported
330 from the East Asian continent.

331 The remaining two episodes were highlighted by enhanced PAN concentrations and
332 characterized by haze occurrence. During episode 3 (Oct. 20–21), PAN and O₃ concentrations
333 increased up to 2.0 ppbv and 78.9 ppbv, respectively, with high NO_x levels, probably
334 influenced by emissions from Korea. Episode 4 (Nov. 4–5) was characterized by the highest

335 concentrations of almost all measured species, including PAN, O₃, PM_{2.5} mass, and PM_{1.0}
336 species; the maximum recorded concentrations of PAN, O₃, and PM_{2.5} mass during this
337 interval were 2.4 ppbv, 87.5 ppbv, and 156 μg/m³, respectively. Note that, along with PM_{2.5}
338 and O₃, PAN was gradually increased through the night. In this episode, an air mass was
339 slowly transported from eastern China. With depleted NO, the effective lifetime of PAN was
340 greatly extended. In addition, PAN concentration showed good correlation with OC, EC, and
341 K⁺; in fact, the correlation of PAN with K⁺ was comparable to that of OC with K⁺. These
342 results, in conjunction with the high AOC/EC (7), imply that the observed haze was mainly
343 caused by the emissions produced by biomass combustion. These results suggest that PAN is
344 a useful tool for distinguishing continental outflows that were typically observed in northeast
345 Asia.

346 The comparison between the measured and calculated concentrations using the CAM-
347 Chem-HTAP2 model showed that the model underestimated the O₃ and PAN levels in
348 Chinese outflows, particularly for haze incidence. These results reveal that Chinese outflows
349 are still poorly understood and not well captured in the model.

350

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650 **Tables**

651

652 Table 1. Chemical and meteorological characteristics of the four episodes.

	Episode 1	Episode 2	Episode 3	Episode 4
Type	Transport dominant	Transport dominant	Chemical transformation	Chemical transformation
Event	O ₃ export	O ₃ export	Haze	Haze
O ₃ (ppbv)	60.2 (80.6)	45.6 (62.8)	59.7 (78.9)	61.8 (87.5)
PAN (ppbv)	0.5 (0.9)	0.5 (0.8)	1.2 (2.0)	1.3 (2.4)
PM _{2.5} (µg/m ³)	34 (62)	23 (36)	50 (76)	77 (156)
SO ₂ (ppbv)	4.3 (12.9)	2.0 (4.4)	2.6 (5.4)	4.4 (9.5)
NO ₂ (ppbv)	3.7 (7.3)	6.2 (12.1)	6.2 (12.7)	6.1 (9.9)
Wind Speed (m/s)	13.5 (16.0)	9.5 (16.1)	6.6 (10.2)	5.0 (7.7)

653 *Measurements are given for the average with the maximum in the parenthesis.

654 **Figure Captions**

655

656 Figure 1. Temporal variations (against local time) of measured species, including PAN, PM_{2.5},
657 O₃, NO₂, NO, SO₂, and, CO, and meteorological parameters, including relative
658 humidity, temperature, and wind speed in fall 2010. Episodes 1–4, described in the
659 main text, are shaded in blue and yellow.

660 Figure 2. Diurnal variations in the concentrations of O₃, NO₂, PAN, and PM_{2.5}, measured at
661 GCO in the fall of 2010 (5 min data of O₃, NO₂, 2 min data of PAN, and 1 h data of
662 PM_{2.5}).

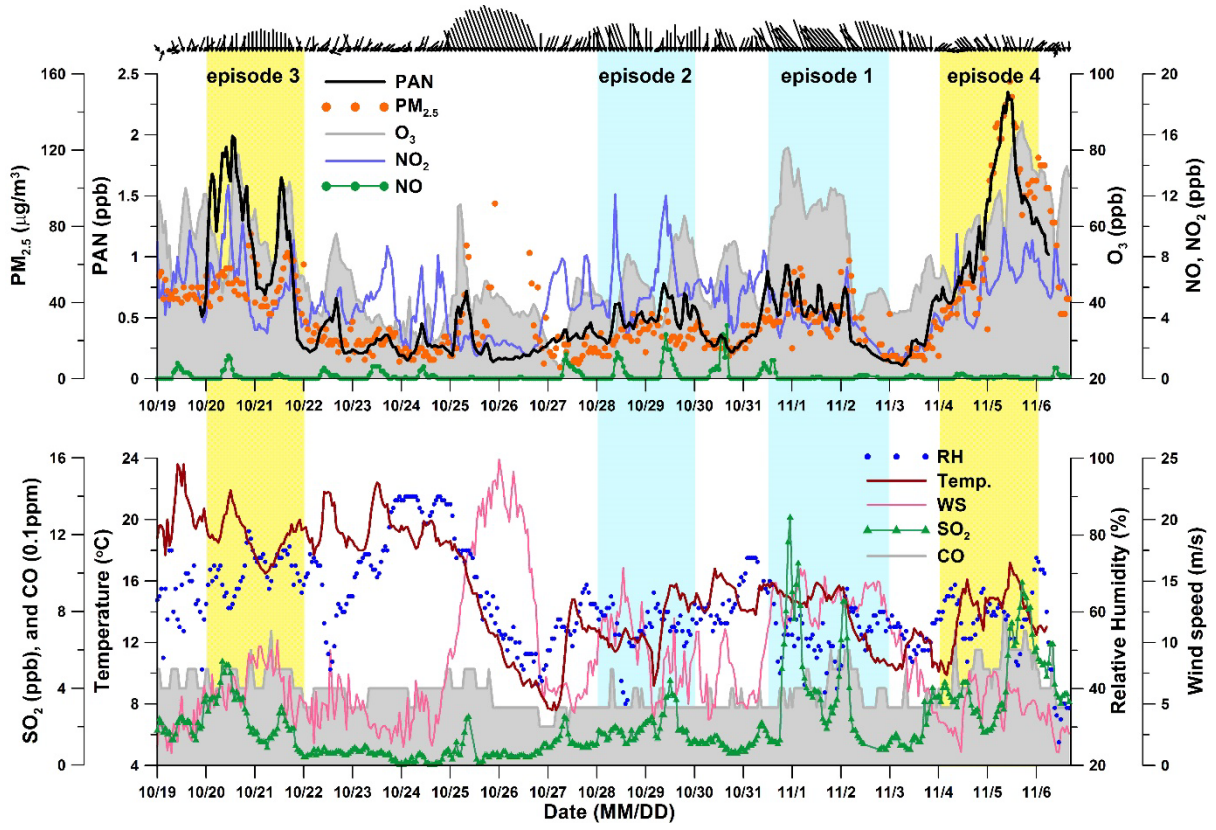
663 Figure 3. Comparison of O₃ with the PAN daily maxima. The time when the daily maximum
664 appears is classified as daytime (08–18 h) and nighttime (the rest) based on the
665 time of sunrise and sunset. Numerals indicate the days.

666 Figure 4. The three-day NOAA HYSPLIT backward trajectories of air masses for every one
667 hour observed at GCO during episode 1 (Oct. 31 to Nov. 2), episode 2 (Oct. 28–29),
668 episode 3 (Oct. 20–21), and episode 4 (Nov. 4–5). They are colored according to
669 the level of (a) PAN, (b) O₃, (c) NO₂, and (d) PM_{2.5} at GCO at the time of the
670 trajectory initialization. The trajectories north of 50°N are not shown. For these
671 horizontal trajectories, (e) vertical heights are given.

672 Figure 5. Correlations among PAN, PM_{2.5}, O₃, and NO₂: (a) O₃ and PAN, (b) NO₂ and PAN,
673 and (c) O₃ and PAN. The lines in (a) and (b) represent the linear regression for each
674 episode.

675 Figure 6. Correlations among PAN, K⁺ ion of PM_{1.0}, and carbon components of PM_{2.5} for
676 three cases: (a) PM_{2.5} mass and PAN, (b) PM_{2.5} OC and PAN, (d) PM_{2.5} OC and EC,
677 and (d) PM_{1.0} K⁺ and PAN. The lines represent the linear regression for each
678 episode.

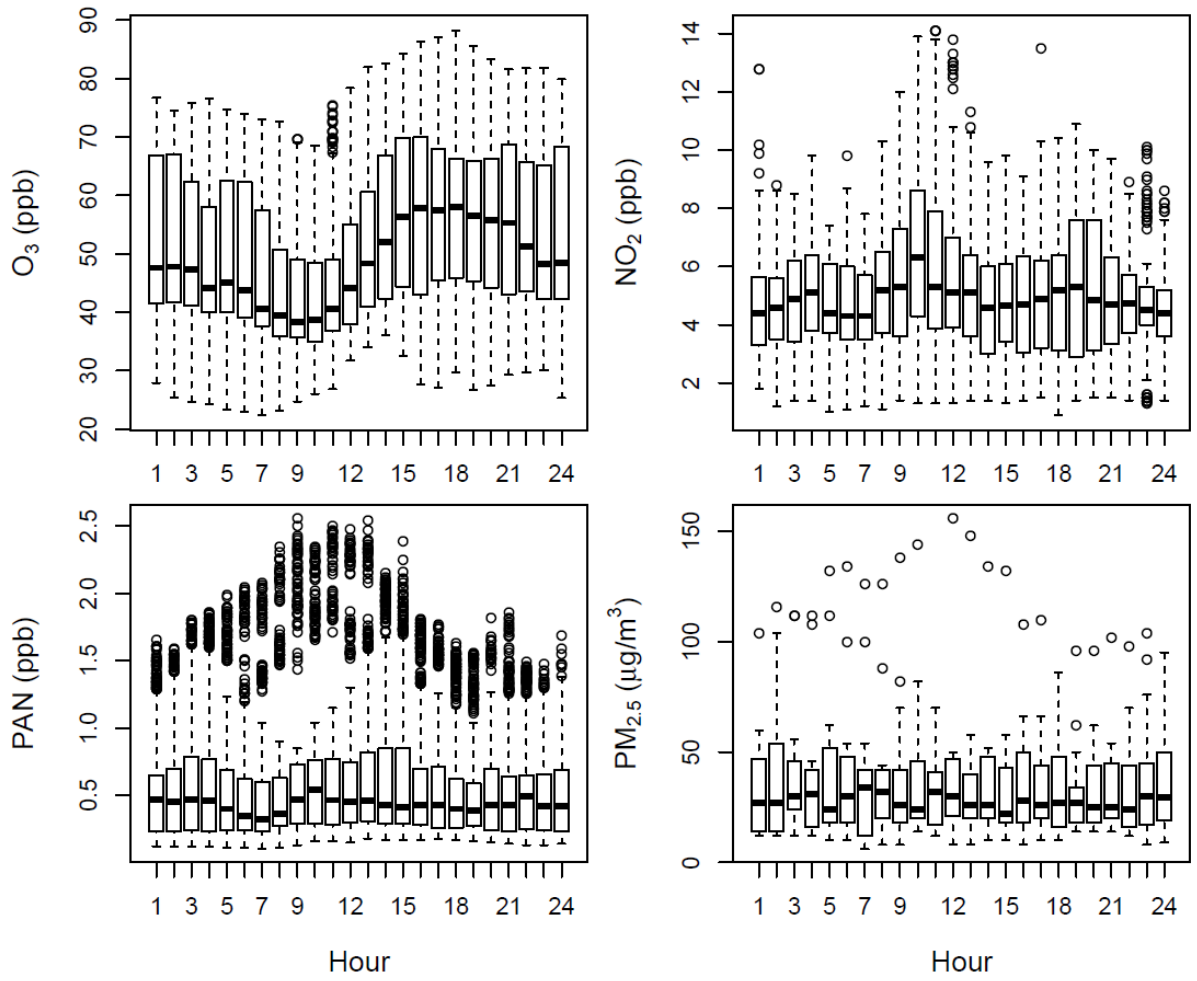
679 Figure 7. Comparison between the observed and calculated (a) PAN and (b) O₃
680 concentrations by CAM-chem model. Time is given in local time and four episodes
681 are shaded.



682

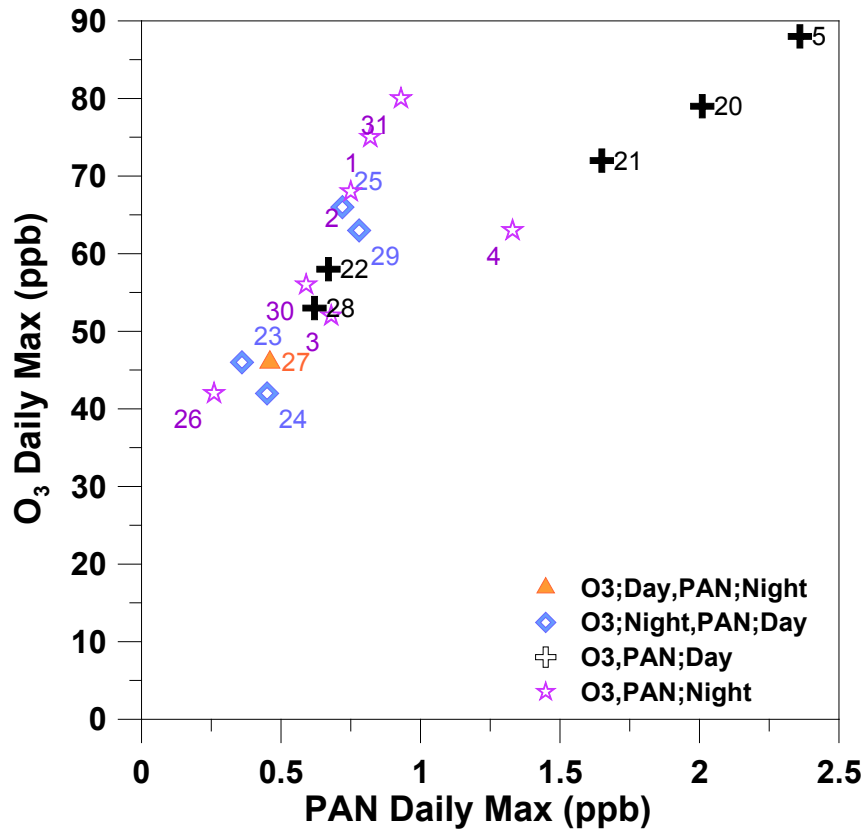
683 Figure 1.

684



685

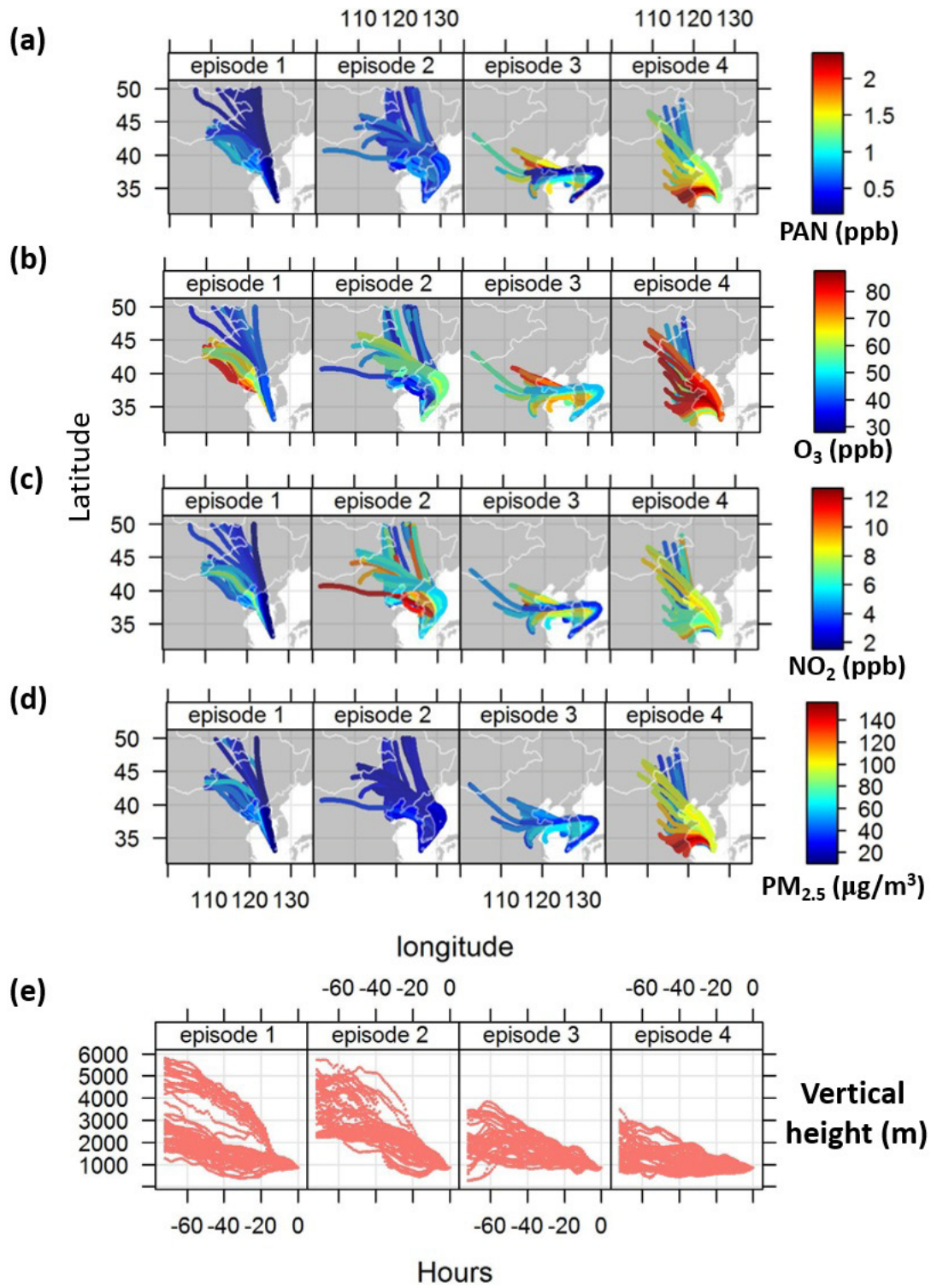
686 Figure 2.



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688 Figure 3.

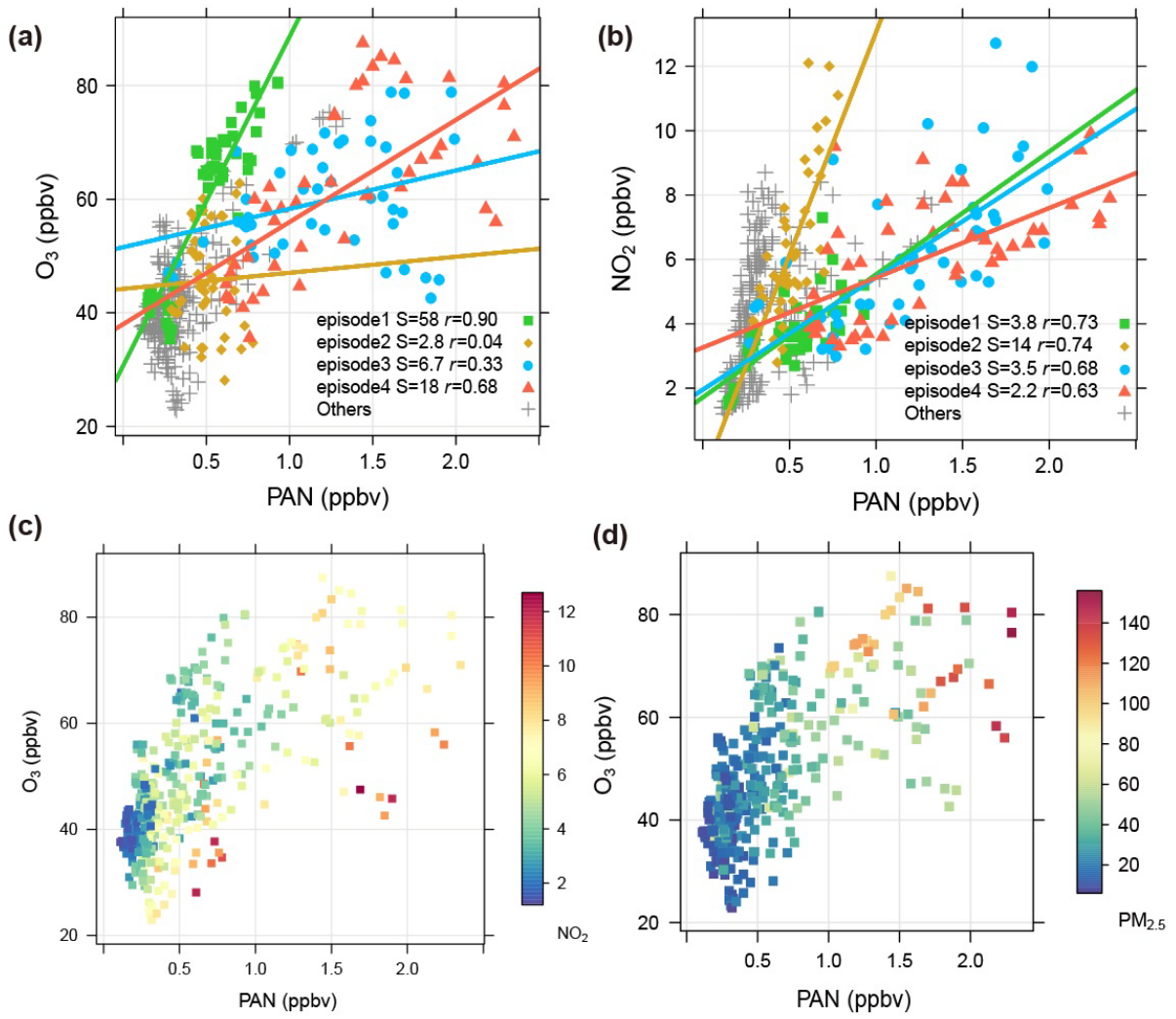
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690 Figure 4.

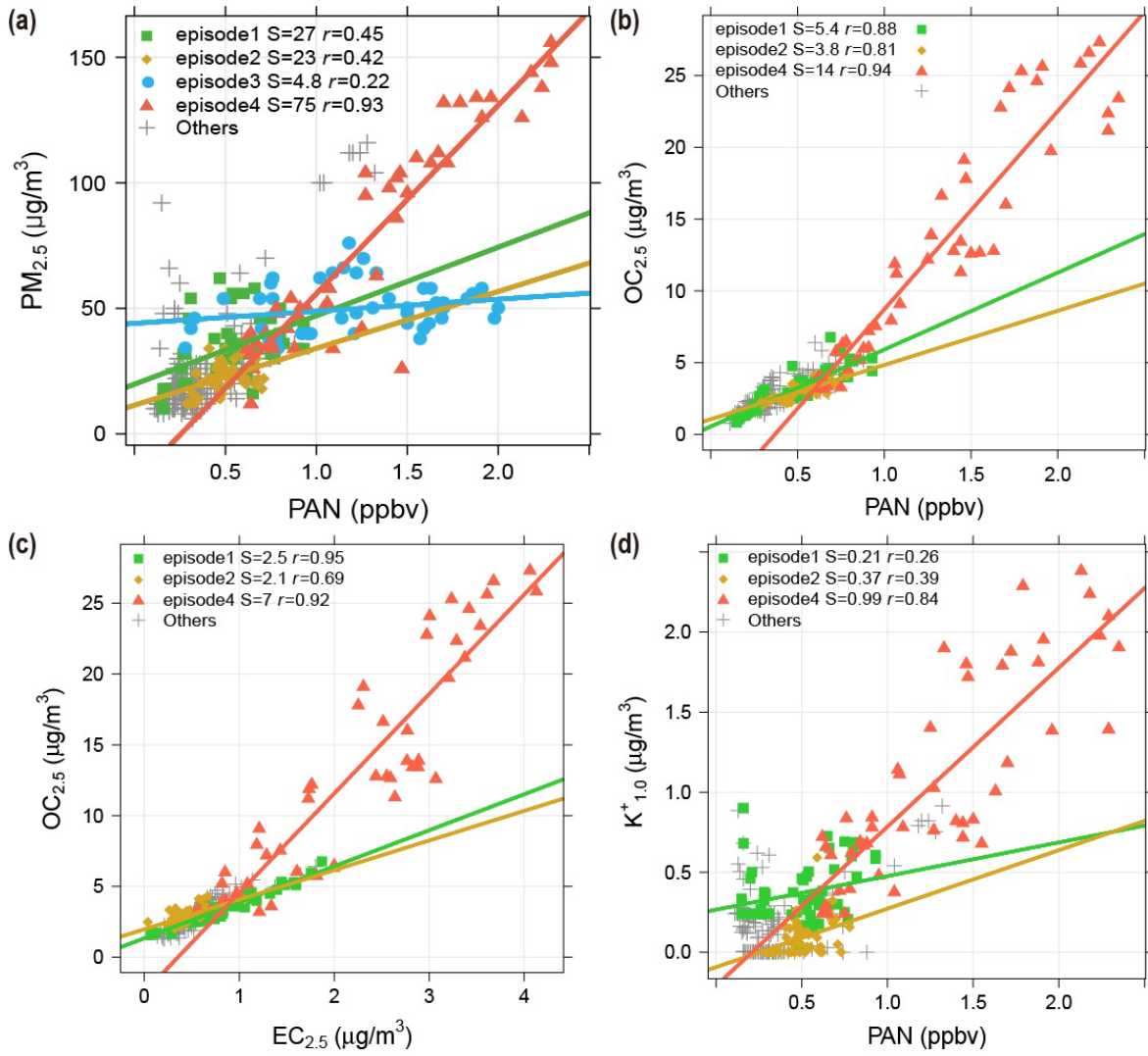
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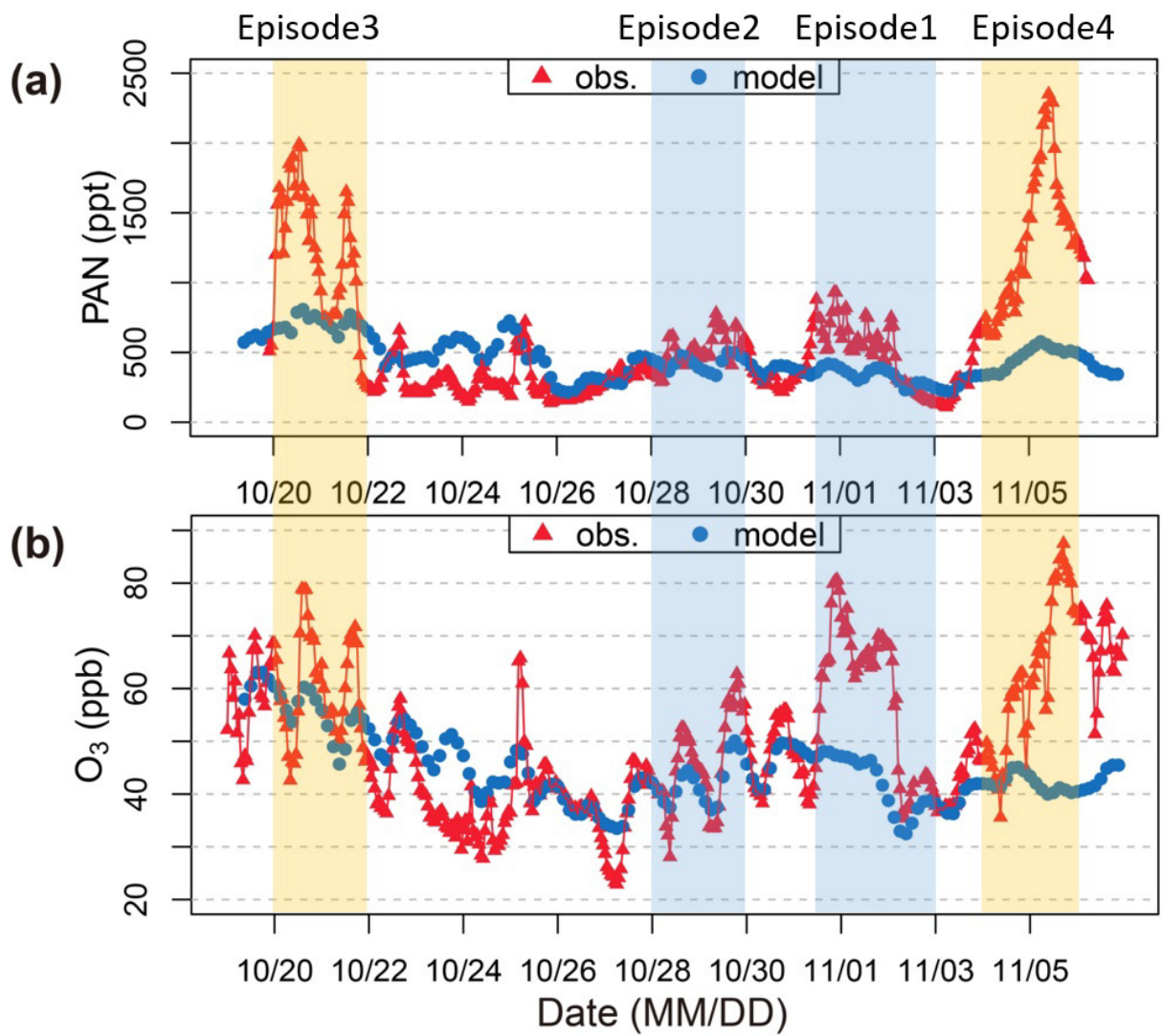
693 Figure 5.

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696 Figure 6.



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698 Figure 7.