Reply to RC1

3 General Comments: Hygroscopicity and volatility are the key properties of atmospheric 4 aerosols that can significantly impact the climate directly by interacting with solar radiation 5 and indirectly by affecting cloud microphysics. Hygroscopic aerosol components can uptake 6 substantial water under sub-saturated conditions and thus can significantly change aerosol 7 size distribution and promote both heterogeneous and aqueous-phase reactions. Severe haze 8 event caused by high mass loading of aerosols is one of the most critical environmental issues 9 faced by China. Although the underlying cause is still an area of active research, it is well 10 recognized that ever-increasing economical expansion and demands for energy have led to 11 tremendous emissions of primary air pollutants (e.g., VOCs, NOx, SO2) in China. Many 12 mitigation measures have been proposed to reduce air pollutant emissions but their effectiveness still needs to be verified in practice. The "Victory Day Parade" event has 13 provided the atmospheric scientists such an opportunity to examine the aerosol properties 14 15 with/without those anthropogenic emissions. Therefore, this dataset by itself is a valuable 16 contribution. The manuscript is well written and is certainly within the scope of ACP. The 17 measurements techniques are well established and the methodology is generally sound. 18 Overall, the research was well planned and carefully executed. I would recommend the 19 manuscript for publication in ACP after the authors address the following comments. 20 21 22 Specific Comments: 23 1. L231: As the authors pointed out that "Air quality has a strong correlation with 24 local wind direction in Beijing", how did the authors manage to separate the effects of 25 meteorological conditions from that of control measures? 26 27 Re: You are right. Separating the effects of meteorological conditions is important to 28 analyze the impact of emission control. When selecting the comparing periods, we 29 tried to find the periods of similar meteorological conditions, especially for Clean1 30 and Clean2 periods. You can refer to Figure S2 and Table S1 in the supplementary material, we have compared the wind and other meteorological variables during 31 32 different periods, the meteorological parameters were similar, especially for Clean1 33 and Clean2 periods. 34 35 2. L259 and L294: It is interesting to note that PM mass concentration decreased by 36 a factor of 8 between polluted and clean periods while precursors only decreased by less than 50%. Please comment on this. 37 38 Re: This is a good point. In practice, the relationship between PM₁ and precursors 39 40 always is not linear. The PM1 mass concentration is mainly determined by the larger 41 particles (Accumulation mode particles), most of them are from condensation of fine

42 particles and direct emission. Many control measures (such as stopping construction

43 activities, shutting down or limiting factory production) would directly decrease the

1

1 2 44 PM₁ emission. However, the particles converted from gas precursors always are the 45 fine particles (Aitken nuclei mode), which have a smaller contribution to PM₁. 46 47 3. L311-314: The authors may also consider the possibility that during daytime strong 48 vertical mixing can bring down air masses transported from long distance that were 49 more aged. However, during nighttime thermal inversion would cap the ground level, 50 where fresh primary emissions would dominate. 51 52 Re: Yes, during nighttime the increase of hydrophobic species (like BC) from primary 53 emissions and thermal inversion would make BC accumulation, these factors would decrease the aerosol hygroscopicity. This has been addressed as "In the evening, thermal inversion 54 55 would cap the ground level and a number of low hygroscopic primary particles (like black 56 carbon, BC) emitted from local diesel trucks and heavy-duty vehicles results in the decrease 57 in kmean during nighttime (see L303-306). 58 59 4. L319: How did "H2O2" form during nighttime? This is more likely due to N2O5 60 chemistry. 61 62 Re: The heterogeneous reactions in the atmosphere are complex, the reactive material 63 includes H₂O₂, HO₂, N₂O₅, O₃ and so on. At night the formation of NO₃ and N₂O₅ 64 becomes important, NO_x can be oxidized to NO_3 through heterogeneous reactions. NO_3 and 65 NO_2 can be further combined to N_2O_5 . Then nitrate can be formed rapidly through NO_3 and 66 N₂O₅. This rapid conversion enhances the available nitrate and may add significantly to the 67 nitrate available for condensation, enhancing night-time concentrations compared to those observed in the day (Dall'Osto et al., 2009). This has been addressed as "NOx could be 68 69 transformed into hydrophilic nitrate rapidly through NO3 and N2O5 (Dall'Osto et al., 2009)" in 70 the revised version of the paper (see L310-312). 71 72 5. L324-330: "During NPF events, ... condensation of VOCs". Did the authors mean 73 that these NPF events (Clean1 case) were caused by organic precursors? If these 74 organics were less hygroscopic, why did kappa-mean increase? Was there any possibility 75 that these 40 nm particles were due to primary emissions, such as automobile 76 exhaust or cooking? 77 78 Re: Yes, we think organic cursors may take some role during NPF in Clean1 case. Note that 79 the hygroscopicity of 40-nm particles during Clean1 period had a "slight" increase, this was 80 likely caused by the increase of OA oxidative level. The 40-nm particles can be from both 81 growth of new formed particles and primary emissions. This has been addressed as "...40-nm 82 particles from local sources were always hydrophobic and a very small amount of 83 hygroscopic particles were produced through the nucleation and growth from gaseous 84 precursors" in the revised version of the paper (see L317-319). 85 6. L377: "With the processes of ... (i.e., SFmean increases)." Why did the aging, coagulation 86 87 and growth processes cause decrease in particle volatility if the precursor gases 88 were the same? This is more likely due to the formation of more oxidized oxygenated 89 organic aerosols. 90

91 Re: Yes. The aging, cogulation and growth processes can enhance the oxidative level of

- 92 organics, and there are more refractory organics (like polymer-type organics) produced. This
- 93 has been addressed "With the processes of particle aging, collision, and growth, they then
- 94 decrease in volatility (i.e., SFmean increases). This is because these processes can enhance
- 95 the oxidative level of organics and there are more refractory organics (like polymer-type 96 organics) produced." in the revised version of the paper (see L367-370).
- 97
- 98 7. Figure 8: How was the linear fit calculated for each size? Did the authors try

99 orthogonal distance regression (ORD), which I believe will generate more reasonable

- 100 results especially for the 40 nm case?
- 101

102 Re: Thanks for your advice. In this paper, I used the least square fitting for each size. Here I

103 tried to use ORD method to the linear fit (Figure 1.). The result by ORD method is similar to

104 our linear fit results.



105

- 106 Figure 1. Comparisons between the number fractions of the nearly-hydrophobic group (NF_{NH})
- 107 and the non-volatile group (NF_{NV}) with orthogonal distance regression (ORD) fitting method.
- 108
- 109

110 Technical comments:

- 111 1. L138-147: remove period before semicolon;
- 112 Re: Revised, thanks! This has been addressed in the revised version of the paper (see 113 L138-144).
- 114 2. L248: Particle mass concentration is denoted by "" throughout the manuscript,
- 115 which, however, could be misleading, since "" is commonly used to represent density.
- Please replace ""with "m" or "mass" or something less misleading. 116
- 117 Re: Revised, thanks! This has been addressed as "PM10-400 nm" in the revised version of the 118 paper (see L247)
- 119 2. L376: dominated by sulfate and "organics" not "VOCs".
- 120 Re: Revised, thanks! This has been addressed as "Wehner et al. (2009) also showed that
- 121 ~97% of newly-formed particles are volatile because they are dominated by sulfate and organics." in the revised version of the paper (see L367)
- 122

123	3. L459: change "corrected" to "correlated".
124	Re: Revised, thanks! This has been addressed in the revised version of the paper (see
125	L449)
126	
127	References:
128	Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization
129	of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009.
130	
131	
101	
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133	
134	Reply to RC2
135	
136	Wang et al. presents their results of hygroscopicity and volatility measurements during and
137	after the Victory Day parade period in China. Aerosol hygroscopicity describes the interaction
138	of aerosols and water vapor, and Aerosol volatility reflect the mixing state of nonvolatile
139	aerosol particles. In addition to their importance of climate effects, the variation of them
140	indicates the change in primary emissions, particle aging process, and regional transport. This
141	dataset provide us new insights into the haze formation mechanisms in Beijing. The paper is
142	well written and organized. Overall, the manuscript is within the scope of ACP. The reviewer
143	would recommend the manuscript for publication in ACP after some minor revisions.
144	
145	Specific Comments:
146	Line 61-64: The authors highlight the uncertainty of aerosol on climate. However, the author
147	discuss the aerosol chemistry indicated from hygroscopicity and volatility throughout the
148	paper. In fact, the pollution formation mechanism remain unclear in Beijing.
149	
150	Re: Revised, Thanks! This has been addressed as "the pollution formation mechanism and
151	the climate effects of aerosols remain highly uncertain due to the highly variable physical and
152	chemical properties of aerosols, as well as complex mechanisms that govern aerosol-climate
153	interactions in the revised version of the paper (see L60-63).
154	Line 250, 251. The outputs define the 40 nm and 150 nm particles as freeh and are existing
155	narticles, respectively. The authors should address the validity of the definition. In fact
157	freshly emitted refractory particles (like BC) are primarily within the 150 nm to 240 nm
158	diameter range (I evy et al. 2013). In contrast, it takes several hours for the growth of 40 pm
150	even during NPF events. The reviewer suggests that the 40 nm particles are from local
160	sources whereas the 150 nm particles are influenced by long-range transport and vertical
161	mixing
162	inizing.
162	Re: Revised. Thanks! This has been addressed as "Particles with Dn equal to 40 nm
100	received, mainto, mis nuo ocon addressed as i i articles with Dp equal to to fill

4

164 represent local-impacted particles and particles with Dp equal to 150 nm represent regionaltransport particles." in the revised version of the paper (see L244-246). 165 166 167 Line 314-320: The reviewer highly suspects that the nitrate is responsible for the appearance 168 of more hygroscopic mode in the 40 nm particles in the early morning. How can form H2O2 169 during the night? The presence of nitrate in Aitken mode particles is very difficult and the 170 addition of nitrate was initially promoted by sulfate condensation (Ye et al., 2010). 171 172 Re: The heterogeneous reactions in the atmosphere are complex, the reactive material 173 includes H_2O_2 , HO_2 , N_2O_5 , O_3 and so on. At night the formation of NO₃ and N_2O_5 174 becomes important, NO_x can be oxidized to NO₃ by through heterogeneous reactions. NO₃ 175 and NO₂ can be further combined to N₂O₅. Then nitrate can be formed rapidly through NO₃ 176 and N_2O_5 . This rapid conversion enhances the available nitrate and may add significantly to 177 the nitrate available for condensation, enhancing night-time concentrations compared to those 178 observed in the day (Dall'Osto et al., 2009). In addition, nitrate in locally produced aerosol 179 was present mainly in particles smaller than 300 nm. In our study, ACSM data also suggested 180 nitrate mass concentration increased sharply at night, especially during the non-control period (Zhao et al., 2016, Figure 8.). This has been addressed as "During the Clean2 period, there is 181 182 another obvious peak at about 0300 LT in the early morning, likely related to the increase in 183 nitrate. Because there was a large amount of NOx emitted from traffic sources in the evening 184 during the non-control period, with the PBL height reduction and ambient temperature 185 decrease, NOx could be transformed into hydrophilic nitrate rapidly through NO3 and N2O5 (Dall'Osto et al., 2009)." in the revised version of the paper (see L306-311). 186 187 188 Line 372: The vehicles emit both gas and particle pollutants, regardless of emission control 189 conditions. However, the NPF is favored when the concentration of pre-existing particles is 190 lower due to emission control. 191 192 Re: Revised, thanks! This has been addressed as "As stated previously, during the Clean1 193 period VOCs has a weaker decrease than SO2 and NOx, this would lead to more VOC-194 formed organic particles formed, which are normally highly volatile." in the revised version 195 of the paper (see L362-365). 196 197 Line 400-402: Do the authors think the emission of refractory particles at night is larger than 198 at daytime? The increase of number fraction may be caused by the slower particle aging and 199 weaker vertical mixing. 200 201 Re: Yes, the sampling site is located in the downtown of the city. A number of diesel vehicles 202 and heavy trucks only can be allowed to enter the downtown at night, this made the increase 203 of refractory particles at night. It can be reflected from the diurnal cycle of BC (Zhao et al., 204 2017, Section 3.5). Then the slower particle aging and weaker vertical mixing made the 205 external-mixed BC accumulated. This has been addressed as "...and then the slower particle 206 aging and weaker vertical mixing made the external-mixing BC accumulated" in the revised 207 version of the paper (see L395-396).

- 208
- Line 426-430: The authors should pay attention to the contradictory statements of $\sigma\kappa$ -PDF "always" exceeds 0.08, ..., the mean $\sigma\kappa$ -PDF of 40-nm particles during the Clean1 period "is
 - 5

211	equal to" 0.08.
212	
213	Re: Revised, thanks! This has been addressed as "ok-PDF always exceeds 0.08 but the 40-
214	nm particles during the Clean1 period" in the revised version of the paper (see L418-419).
215	
216	Technical comments:
217	Line 325: The term of hydrophilic should be replaced by "hygroscopic", because hygroscopic
218	growth do not take place at a high RH for many hydrophilic substance.
219	Re: Revised, thanks! This has been addressed in the revised version of the paper (see L318).
220	Line 336: The condensation species are sulfate, nitrate, OA other than SO2, NOx, and VOCs.
221	Re: Revised, Thanks! This has been addressed in the revised version of the paper (see L329-
222	330).
223	
224	Ye, X.N., Ma, Z., Hu, D.W., Yang, X., Chen, J.M., 2010. Size-resolved hygroscopicity of
225	submicrometer urban aerosols in Shanghai during wintertime. Atmospheric Research
226	99, 353-364.
227	
228	References:
228 229	References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization
228 229 230	References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009.
228 229 230 231	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Xu. Justicky into average characterization the 2015. Cline Victors Departments for a series of the ser
228 229 230 231 232	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Y.: Insights into aerosol chemistry during the 2015 China Victory Day parade: results from simulaneous measurements at ground hund and 260 m in Deliing. Atmos. Chem. Phys. 17, 2215.
228 229 230 231 232 233 234	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Y.: Insights into aerosol chemistry during the 2015 China Victory Day parade: results from simultaneous measurements at ground level and 260 m in Beijing, Atmos. Chem. Phys., 17, 3215-2222, 10.5104/com 17, 2017, 2017.
228 229 230 231 232 233 234 225	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Y.: Insights into aerosol chemistry during the 2015 China Victory Day parade: results from simultaneous measurements at ground level and 260 m in Beijing, Atmos. Chem. Phys., 17, 3215-3232, 10.5194/acp-17-3215-2017, 2017.
228 229 230 231 232 233 234 235	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Y.: Insights into aerosol chemistry during the 2015 China Victory Day parade: results from simultaneous measurements at ground level and 260 m in Beijing, Atmos. Chem. Phys., 17, 3215-3232, 10.5194/acp-17-3215-2017, 2017.
228 229 230 231 232 233 234 235 236	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Y.: Insights into aerosol chemistry during the 2015 China Victory Day parade: results from simultaneous measurements at ground level and 260 m in Beijing, Atmos. Chem. Phys., 17, 3215-3232, 10.5194/acp-17-3215-2017, 2017.
228 229 230 231 232 233 234 235 236 237	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Y.: Insights into aerosol chemistry during the 2015 China Victory Day parade: results from simultaneous measurements at ground level and 260 m in Beijing, Atmos. Chem. Phys., 17, 3215-3232, 10.5194/acp-17-3215-2017, 2017.
228 229 230 231 232 233 234 235 236 237 238	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Y.: Insights into aerosol chemistry during the 2015 China Victory Day parade: results from simultaneous measurements at ground level and 260 m in Beijing, Atmos. Chem. Phys., 17, 3215-3232, 10.5194/acp-17-3215-2017, 2017.
228 229 230 231 232 233 234 235 236 237 238 239	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Y.: Insights into aerosol chemistry during the 2015 China Victory Day parade: results from simultaneous measurements at ground level and 260 m in Beijing, Atmos. Chem. Phys., 17, 3215-3232, 10.5194/acp-17-3215-2017, 2017.
228 229 230 231 232 233 234 235 236 237 238 239 240	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Y.: Insights into aerosol chemistry during the 2015 China Victory Day parade: results from simultaneous measurements at ground level and 260 m in Beijing, Atmos. Chem. Phys., 17, 3215-3232, 10.5194/acp-17-3215-2017, 2017.
228 229 230 231 232 233 234 235 236 237 238 239 240 241	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Y.: Insights into aerosol chemistry during the 2015 China Victory Day parade: results from simultaneous measurements at ground level and 260 m in Beijing, Atmos. Chem. Phys., 17, 3215-3232, 10.5194/acp-17-3215-2017, 2017.
228 229 230 231 232 233 234 235 236 237 238 239 240 241	 References: Dall'Osto M., Harrison R.M., Coe H., Williams P.I. and Allan J.D.: Real time chemical characterization of local and regional nitrate aerosols, Atmos. Chem. Phys., 9, 3709-3720, 2009. Zhao J., Du W., Zhang Y., Wang Q., Chen C., Xu W., Han T., Wang Y., Fu P., Wang Z., Li Z. and Sun Y.: Insights into aerosol chemistry during the 2015 China Victory Day parade: results from simultaneous measurements at ground level and 260 m in Beijing, Atmos. Chem. Phys., 17, 3215-3232, 10.5194/acp-17-3215-2017, 2017.

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244	Enhanced hydrophobicity and volatility of submicron aerosols under
245	severe emission control conditions in Beijing
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271 Abstract. A series of strict emission control measures were implemented in Beijing and 272 the surrounding seven provinces to ensure good air quality during the 2015 China 273 Victory Day parade, rendering a unique opportunity to investigate anthropogenic 274 impact of aerosol properties. Submicron aerosol hygroscopicity and volatility were 275 measured during and after the control period using a hygroscopic and volatile tandem differential mobility analyzer (H/V-TDMA) system. Three periods, namely, the control 276 277 clean period (Clean1), the non-control clean period (Clean2), and the non-control 278 pollution period (Pollution), were selected to study the effect of the emission control 279 measures on aerosol hygroscopicity and volatility. Aerosol particles became more 280 hydrophobic and volatile due to the emission control measures. The hygroscopicity 281 parameter (κ) of 40–200 nm particles decreased by 32.0%–8.5% during the Clean 1 period relative to the Clean2 period, while the volatile shrink factor (SF) of 40-300 nm 282 283 particles decreased by 7.5%-10.5%. The emission controls also changed the diurnal variation patterns of both the probability density function of κ (κ -PDF) and the 284 285 probability density function of SF (SF-PDF). During Clean1 the κ-PDF showed one 286 nearly-hydrophobic (NH) mode for particles in the nucleation mode, which was likely 287 due to the dramatic reduction in industrial emissions of inorganic trace gases. Compared 288 to the Pollution period, particles observed during the Clean1 and Clean2 periods exhibited a more significant non-volatile (NV) mode throughout the day, suggesting a 289 more externally-mixed state particularly for the 150 nm particles. Aerosol 290 291 hygroscopicities increased as particle sizes increased, with the greatest increases seen during the Pollution period. Accordingly, the aerosol volatility became weaker (i.e., SF 292

increased) as particle sizes increased during the Clean1 and Clean2 periods, but no
apparent trend was observed during the Pollution period. Based on a correlation
analysis of the number fractions of NH and NV particles, we found that a higher number
fraction of hydrophobic and volatile particles during the emission control period.

297 1. Introduction

298 China, as the world's second largest economy, is facing severe air pollution problems due to its rapid economic growth. This has led to highly elevated aerosol 299 300 concentrations, especially in urban regions such as Beijing, Shanghai, and Guangzhou 301 (Hsu et al., 2012; Huang et al., 2014). Every year, high levels of fine particulate matter 302 (PM) have caused many severe haze days in these regions, that may pose a great health 303 hazard and changes in the regional climate because of aerosol direct and indirect climate 804 effects (Z. Li et al., 2016; G. X. Wu et al., 2016). However, the pollution formation 805 mechanism and the climate effects of aerosols still remain highly uncertain due to the 306 highly variable physical and chemical properties of aerosols, as well as complex B07 mechanisms that govern aerosol-climate interactions (Tao et al., 2012; Wang et al., 808 2014)(Tao et al., 2012).

Aerosol hygroscopicity and volatility are two important physical properties describing the process of haze formation and its effects on climate. Aerosol hygroscopicity describes the interaction of aerosols and water vapor under sub- and supersaturation conditions, and is a vital parameter to the aerosol life cycle, aerosol activation ability, and aerosol direct and indirect climate effects (Swietlicki et al., 2008; 域代码已更改

314	Tao et al., 2012; Bian et al., 2014). Aerosol volatility is a physical parameter correlated
315	with carbonaceous aerosols, commonly used to study the aerosol mixing state and aging
316	level (Wehner et al., 2009; S. L. Zhang et al., 2016). To date, there are many ways to
317	measure aerosol hygroscopicity and volatility, but the most popular one is the
318	Hygroscopic and Volatile Tandem Differential Mobility Analyzer (H/V-TDMA)
319	system because it can measure these properties in great detail (Swietlicki et al., 2008).
320	The Chinese government took many drastic measures to reduce the emissions of
321	air pollutants from industry, road traffic, and construction sites, especially during some
322	great events such as the 2008 Summer Olympic Games, the 2014 Asia-Pacific
323	Economic Cooperation. Swift and drastic improvement in air quality (Huang et al.,
324	2015; Shi et al., 2016) provide unique opportunities to investigate the effects of
325	emissions on air quality. To our knowledge, previous studies have usually focused on
326	aerosol chemistry, sources, and transport (Wang et al., 2010; Gao et al., 2011; Sun et
327	al., 2016b), but not on the effects of emission controls on aerosol hygroscopicity and
328	volatility. Due to the importance of the two factors on describing the process of haze
329	formation as well as the effect on climate, it is necessary to investigate the changes in
330	aerosol hygroscopicity and volatility when emission control measures are in place.
331	To guarantee good air quality in Beijing during the 2015 China Victory Day parade,

the Chinese government implemented much stricter emission control measures than
normally done in Beijing and the surrounding seven provinces from 20 August to 3
September. The control measures consisted of a ban on driving vehicles every other day,
shutting down or limiting factory production, stopping construction activities, and so

336 on. These emission control measures successfully ensured a continuous stretch of 15 days of blue sky, vividly named "Parade Blue" (H. Li et al., 2016). During and after the 337 parade emission control period, we conducted in situ measurements of submicron 338 339 aerosol chemical and physical properties in Beijing. Size-resolved chemical 840 compositions were also obtained (Zhao et al., 2017) (Zhao et al., 2016, published). The 341 average PM less than 1 µm in diameter (PM1) concentration was 19.3 µg m⁻³ during the parade emission control period, 57% lower than that after the control period. All 342 343 chemical species decreased during the control period, but their decreasing percentages 344 were different.

345 This study period is unique for investigating aerosol properties during low PM level 346 periods. This paper will further evaluate the impact of emission controls on the hygroscopicity and volatility of submicron aerosols, which may bring some insight into 347 348 how to reduce pollution in the future. Furthermore, investigating aerosol hygroscopicity and volatility with and without emission controls will help in understanding 349 350 environmental and climate changes in general. This paper is structured as follows. 351 Section 2 describes the instrumentation and data used, and section 3 introduces the methods to data analysis. Aerosol hygroscopicity and volatility during different periods 352 353 were compared and discussed in section 4. Conclusions and summary are given in 354 section 5.

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355 2. Experimental methods

356 2.1. Sampling site and meteorology

357 The submicron aerosol hygroscopicity and volatility were measured in situ from 358 26 August to 7 October 2015 using the H/VTDMA system located at the Institute of 359 Atmospheric Physics (IAP), Chinese Academy of Sciences (39.97°N, 116.37°E), which is located between the north 3rd and 4th ring road in northern Beijing. The sampling 360 361 instruments were put into a white container at ground level and an air conditioner was used to maintain the temperature at 20-25°C inside the container. Meteorological 362 363 variables, including temperature (T), relative humidity (RH), wind speed (WS), and 364 wind direction (WD), were measured at different heights of a 325 m meteorological 365 tower, the tower located ~20 m west of the container. To eliminate the influence of 366 buildings on wind, we selected the 280-m wind direction and 8-m wind speed as references in this study. Simultaneously, particle number concentrations (10-600 nm) 367 368 were also measured by a scanning mobility particle sizer (SMPS) located at the 260-m level of the tower. The SMPS is equipped with a long differential mobility analyzer 369 (DMA, Model 3081A, TSI Inc.) and a condensation particle counter (CPC, Model 3775, 370 371 TSI Inc.). In addition, the measurement of aerosol chemical composition using a High-372 Resolution Aerosol Mass Spectrometer (HR-AMS) and an Aerosol Chemical Speciation Monitor (ACSM) were deployed at ground level and at the 260-m level of 373 374 the tower, respectively. The HR-AMS was situated in a sampling room located on the 375 rooftop of a two-story building (~8 m), ~25 m north from the container. An analysis of 876 the aerosol chemical composition has been done (Zhao et al., 2017) (Zhao et al., 2016,

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377 published).

378 2.2. Instrumentation and operation

379 The H/V-TDMA system developed by the Guangzhou Institute of Tropical and Marine Meteorology (ITMM) was used to measure the submicron aerosol 380 381 hygroscopicity and volatility. The H-TDMA system (H-mode) shown in Figure 1 consists of four main parts: (1) a nafion dryer (Model PD-70T-24ss, Perma Pure Inc., 382 USA) and a bipolar neutralizer (Kr85, TSI Inc.). The nation dryer ensured that the RH 383 384 of the sample flow was below 20% over the entire measurement period, and the bipolar 885 neutralizer was used to equilibrate the charge of particles (Wiedensohler, 1988)-; (2) 386 the first differential mobility analyzer (DMA1, Model 3081L, TSI Inc.): The DMA1 was 387 used to select quasi-monodisperse particles of a certain diameter through a fixed electric 888 voltage. The diameters selected were 40, 80, 110, 150, and 200 nm-; (3) a nafion humidifier (Model PD-70T-24ss, Perma Pure Inc., USA). The nafion humidifier was 389 390 used to humidify the aerosol flow from the DMA1 to a defined RH. In the study, we set 891 RH to 90%-; (4) the second DMA (DMA2, same model as the DMA1) and a 392 condensation particle counter (CPC, Model 3772, TSI Inc.). The DMA2 and the CPC 393 were used together to measure the number size distribution of the humidified particles. 394 An automated valve located between the DMA₁ and the nation humidifier directly 395 connects the DMA1 with the CPC. This can be used to measure the 10-400 nm particle 396 number size distribution (PNSD) by varying the electric voltage of the DMA1. Details 397 about the design of the HTDMA system and its applications are given by Tan et al.

(2013a). 398

399 The design of the V-TDMA system (V-mode) is similar to that of the H-TDMA system, except that the nation humidifier in the V-TDMA system was replaced by a 400 401 heating tube that induces the evaporation of volatile materials. The heating tube was an 80-cm long stainless steel tube with an inner diameter of 8 mm. With a sample flow 402 rate of 1.0 L min⁻¹, its residence time (~ 2.4 s) in the heated section is sufficient for the 403 volatile materials to be effectively vaporized (Cheung et al., 2016). In this study, the 404 405 heating temperature was set to 300°C. The residual particles of volatile compounds at 406 this temperature, such as sulfates, nitrates, and most organics, are mainly refractory non-volatile organic carbon (such as polymer-type organics), and sea salts (Philippin et 407 408 al., 2004; Wehner et al., 2009; Cheung et al., 2016; Ma et al., 2016). Particle were measured at the diameters of 40, 80, 110, 150, 200 and 300 nm. The H/V-TDMA system 409 410 has been successfully used in previous studies (Tan et al., 2013b; Cheung et al., 2016; 411 Tan et al., 2016).

The hygroscopic growth factor (GF) at a given RH and the volatile shrink factor 412 413 (SF) at a certain temperature are defined as the ratio of the conditional diameter to the dry diameter, with respect to RH and T, respectively: 414

- 415 $GF = D_p(RH)/D_{0dry}$,
- 416

(1)

(2)

 $SF = D_p(T)/D_{0dry}$.

Here, $D_p(RH)$ refers to the particle diameter measured at RH = 90%, $D_p(T)$ refers to the 417 particle diameter measured at $T = 300^{\circ}$ C, and D_{0dry} refers to the dry diameter set by the 418 DMA1. The measured distribution function (MDF) versus GF or SF can be calculated 419

420 with the number concentration from CPC data downstream from the DMA1 and the 421 DMA₂. However, the MDF is a skewed and smoothed integral transformation of the 422 particles' actual growth/shrink factor probability density function (GF-PDF or SF-PDF) 423 due to the effect of the DMA diffusion transfer function (Swietlicki et al., 2008; Gysel et al., 2009). In this study, the TDMA fit algorithm (Stolzenburg et al., 1988, 2008) was 424 used to retrieve the GF-PDF and the SF-PDF. The TDMAfit algorithm assumes that 425 426 groups in the PDF following one or more lognormal distribution functions (Gaussian 427 shape), thus allowing for the possibility that particles of a given type are not all identical.

428 **3. Data analysis**

429 3.1. Hygroscopicity parameter

According to the Köhler theory (Petters et al., 2007), the hygroscopicity parameter
κ can be used to depict the hygroscopicity of particles at different RHs. Using H-TDMA
data, κ is calculated as:

433
$$\kappa(GF, D_d) = (GF^3 - 1) \cdot \left[\frac{1}{RH} \exp\left(\frac{4\sigma_{s/a}M_w}{RT\rho_w D_d GF}\right) - 1\right] \quad , \tag{3}$$

where RH is the default value of the H-TDMA, $\sigma_{s/a}$ is the surface tension of the solution/air interface, M_w is the molecular weight of water, R is the universal gas constant, T is the temperature, ρ_w is the density of water, D_d is the diameter of the dry particles (equivalent to D_{0dry} as mentioned above), and GF from equation (1). In this study, T used in the κ calculation is 23°C (the average temperature of the inner container) and $\sigma_{s/a}$ is assumed to be the same as the surface tension of the pure water/air interface (about 0.0723 N m⁻¹ at 23°C).

441 3.2. Statistics of κ-PDF and SF-PDF

442 The probability distribution function of κ (κ -PDF, $c(\kappa, D_d)$) derived from the GF-443 PDF was normalized as $\int c(\kappa, D_d) d\kappa = 1$. The ensemble mean hygroscopicity 444 parameter is then defined as the number-weighted mean GF of κ -PDF over the whole κ 445 range:

446
$$\kappa_{\text{mean}} = \int_0^\infty \kappa c(\kappa, D_d) d\kappa$$
 (4)

447 The standard deviation of κ -PDF is:

448
$$\sigma_{\kappa-PDF} = \left(\int_0^\infty (\kappa - \kappa_{mean})^2 c(\kappa, D_d) d\kappa\right)^{\frac{1}{2}} \quad . \tag{5}$$

449 The calculated statistical parameters of SF-PDF ($c(SF, D_d)$) are similar to those of κ -450 PDF, so SF can be used instead of κ and $c(SF, D_d)$ instead of $c(\kappa, D_d)$ in these 451 equations.

452 3.3. Classification of different hygroscopic and volatile groups

The mixing state of ambient aerosol particles is complex due to different sources, different aging processes, and so on. Different hygroscopic and volatile groups had been used around the Beijing region using H-TDMAs and V-TDMAs (Massling et al., 2009; Liu et al., 2011; S. L. Zhang et al., 2016). Based on previous studies and our measurements (Figure S1), ambient aerosol particles were classified into three hygroscopic groups and three volatile groups, where κ and SF are used here to define the boundaries for each group:

460 Nearly-Hydrophobic, NH: $\kappa < 0.1$;

461 Less-Hygroscopic, LH: $0.1 \le \kappa \le 0.2$;

- More-Hygroscopic, MH: $0.2 \le \kappa$; 462
- Non-Volatile, NV: $SF \ge 0.88$; 463
- Slight-Volatile, SV: $0.88 > SF \ge 0.55$; 464
- Very-Volatile, VV: SF < 0.55. 465
- 466 The number fraction (NF) for each hygroscopic group with the boundary of [a, b]
- is defined as: 467

468
$$NF = \int_{a}^{D} c(\kappa, Dp) d\kappa \quad . \tag{6}$$

469 The number fraction of each volatile group also can be calculated using a similar 470 equation.

471 4. Results and discussion

4.1. Overview of measurements 472

4.1.1. Meteorological conditions during the sampling period 473

474 Air quality has a strong correlation with local wind direction in Beijing. Previous 475 studies have shown that high PM concentrations usually correspond to southerly winds, 476 while low PM concentration are generally related to northerly winds, because there 477 were more high concentration air pollutants from source locations south of the Beijing area (Wehner et al., 2008; Wang et al., 2010; Gao et al., 2011). Figure 2 displays time 478 479 series of WD at 280 m, WS at 8 m, ambient T, and RH. During the emission control 480 period, the prevailing winds were northerly, except for the period from 29 August to 30 481 August due to the influence of accumulated precipitation. During the non-control period, 482 the prevailing winds changed due to the influence of weather systems. Two cold fronts 17

passed through on two different days, i.e., on the night of 9 September and in the early morning of 30 September. During these frontal passages, the prevailing winds were northerly, but on other days, the prevailing winds were southerly and the meteorological parameters showed obvious diurnal cycle patterns. Over the measurement period, the average ambient T and RH were 21.9°C and 62.4%, respectively.

488 4.1.2. Time series of κ-PDF and SF-PDF, and the division of clean and pollution 489 periods

490 Figure 3 shows the time series of 10-400 nm particle mass concentrations (P10-400 mmPM_{10-400 nm}) derived from PNSD measurements and the time series of κ-PDF and SF-491 492 PDF with 40 nm and 150 nm particles as examples. Particles with D_p equal to 40 nm 493 represent local-impacted particles fresh particles and particles with Dp equal to 150 nm 494 represent regional-transport particlespre-existing particles. Several haze events during the non-control period can be seen from the time series of p10-400 nm PM10-400 nm, which 495 shows the rapid accumulation of particle mass concentration. Based on mass 496 concentrations and weather conditions, we selected several clean and pollution periods 497 to study the differences in aerosol hygroscopicity and volatility for two different cases 498 499 (Figure 3). To further study the effect of emission controls, we divided the clean period 500 into two periods: Clean1 (control clean period) and Clean2 (non-control clean period). During the Clean1, Clean2, and Pollution periods, the average P10 400 mm PM10-400 nm was 501 6.9±2.8, 6.0±4.2, and 51.0±25.6 µg m⁻³, respectively. There was no significant 502 precipitation during the three selected periods. The time series of ĸ-PDF and SF-PDF 503 504 (Figure 3b-e) showed evident changes and fluctuations in the measurements. The 18

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prominent differences in κ-PDF and SF-PDF during the three periods will be discussed
in the following sections.

507 A wind rose diagram (Figure S2) was used to compare winds during the different 508 periods. During the Clean1 and Clean2 periods, wind directions were similar, mainly 509 from the north and northwest. During the Pollution period, the wind direction had the characteristics of mountain-valley breezes where the wind direction changed routinely 510 at midnight and changed the wind direction from southerly to northerly (Figure 2). The 511 512 change in wind direction at night would reduce pollution in the short term (Sun et al., 513 2016b). Even so, the prevailing wind direction was southerly during the Pollution period (Figure S2), which was favorable for the transport of pollutants from the more 514 515 populated and more industrialized south and southeast to Beijing. The mean WS and RH were similar during all periods, but the mean temperature during the Clean2 period 516 517 was lower than during the other periods due to the influence of cold fronts (Table S1). In summary, the meteorological parameters of the Clean1 and Clean2 cases were similar 518 519 expect for the ambient T. This provided the opportunity to compare the differences in 520 aerosol properties between control and non-control periods. The high level of PM 521 during the Pollution case can also provide a good opportunity to compare differences 522 between clean and polluted environments.

523 4.2. Diurnal variation

524 4.2.1. Diurnal variation in the aerosol size distribution

525 Figure 4a shows the diurnal variation in total number concentration of 10–400 nm

526	particles (N_{10-400 \ \text{nm}}). In general, N_{10-400 \ \text{nm}} is higher at night and lower during the day
527	due to the influence of changes in the planetary boundary layer (PBL). However, a
528	significant peak in $N_{\rm 10-400\ nm}$ is also seen at noontime because of new particle formation
529	(NPF) events (Figure S3). NPF started at about 0900 local time (LT) during the Clean1
530	and Clean2 periods. During the Clean1 period, the $N_{\rm 10-400\ nm}$ peak was lower than that
531	observed during the Clean2 period, and the peak in the Clean1 case appeared two hours
532	earlier than that in the Clean2 case (1200 LT during Clean1 and 1400 LT during Clean2).
533	This illustrates that the strength of the NPF was weaker during the Clean1 period than
534	during the Clean2 period, and that it was likely related to the decrease in precursors
535	during the Clean1 period. H. Li et al. (2016) have reported that during the parade control
536	period, the precursors SO_2 , NO_x , and volatile organic compounds (VOCs) decreased by
537	36.5%, 49.9%, and 32.4%, respectively. The relatively higher ambient temperature
538	during the Clean1 period was also unfavorable for NPF (Kulmala et al., 2004).
539	Figure 4b compares diurnal variations in total mass concentration of 10-400 nm
540	particles (<u>PM_{10-400 nm}P_{10-400 nm}</u>) during the three periods. No clear increase in P_{10-400}
541	$\frac{PM_{10-400 \text{ nm}}}{M_{10-400 \text{ nm}}}$ is seen while N _{10-400 nm} sharply increases during the Clean1 and Clean2
542	daytime periods. This is because the D _p for most particles was less than 100 nm, which
543	contributed little to <u>P10 400 nm</u> PM10-400 nm. During the Clean1 and Clean2 periods, PM10-
544	400 mm P+10-400 mm had an obvious diurnal variation, which could be attributed to the
545	evolution of the PBL. As is known, the lower PBL at night aids in the accumulation of
546	pollutants (Achtert et al., 2009). However, this effect was weak in the pollution case
547	because of the change in wind direction from southerly to northerly at midnight, which

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548 could partly offset the influence of the PBL.

549

4.2.2. Diurnal variation in aerosol hygroscopicity

550 Figure 5a shows the diurnal variation in size-resolved κ_{mean} during the three periods. κ_{mean} shows a peak during daytime, and is always higher than that observed during 551 552 nighttime. This is because more highly aged particles due to photochemical reactions 553 cause the increase in κ_{mean} during daytime. In the evening, thermal inversion would cap the ground level and a number of low hygroscopic primary particles (like black carbon, 554 555 BC) emitted from local diesel trucks and heavy-duty vehicles results in the decrease in 556 κ_{mean} during nighttime (Liu et al., 2011; S. L. Zhang et al., 2016). During the Clean2 period, there is another obvious peak at about 0300 LT in the early morning, likely 557 related to the increase in nitrate. Because there was a large amount of NOx emitted from 558 559 traffic sources in the evening during the non-control period, with the PBL height 560 reduction and ambient temperature decrease, NOx could be transformed into 561 hydrophilic nitrate rapidly through NO3 and N2O5 (Dall'Osto et al., 2009). through 562 heterogeneous reactions with dissolved H2O2 during nighttime (Seinfeld et al., 2016). 563 This was also verified from comparisons of the nitrate diurnal cycle with and without 564 emission controls (Zhao et al., 2017) (Zhao et al., 2016, published). 565 Figure 5b shows the diurnal variation in κ -PDF for particles with D_p equal to 40

566 nm (i.e., <u>local-impacted</u> newly-formed particles) during the three periods. During the 567 Clean1 period, the κ -PDF has a quasi-unimodal shape (only in the hydrophobic mode). 568 During NPF events, κ_{mean} increases slightly, indicating that <u>40-nm particles from local</u>

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569	sources were always hydrophobic and a very small amount of hygroscopic hydrophilie	
570	particles were produced through the nucleation and growth from gaseous precursors.	
571	This is likely because the secondary formation of hydrophilic sulfate and nitrate was	
572	suppressed due to low concentrations of SO_2 and NO_X during the parade control period	
573	(H. Li et al., 2016). Most of the new particles should consist of less hygroscopic	
574	organics that are formed by oxidation and condensation of VOCs. By contrast, during	
575	the Clean2 period, the $\kappa\text{-PDF}$ with D_p of 40 nm shows either a bimodal or quasi-	
576	trimodal distribution and exhibits a large diumal variation during the day. Interestingly,	
577	when the NPF event occurred at about 0900 LT, the number fraction of the hydrophobic	
578	mode quickly decreased and the hydrophilic mode increased (Figure 5b), suggesting	
579	the conversion of externally mixed particles to internally mixed particles due to the	
580	species condensation (sulfate, nitrate and organics) from the photochemical reaction of	
581	SO2, NOx and VOCs. the condensation of gas precursors (including SO2, NOx, as well	_
582	as VOCs). A similar phenomenon was also observed by Z.J. Wu et al. (2016). For the	
583	Clean1 case, much less of these gases were in the atmosphere due to the emission	
584	control. Around 1700 LT, the fraction of hydrophobic mode particles increased again,	
585	mainly due to substantial traffic emissions at rush hour. However, during the Pollution	
586	period, the κ -PDF shows a bimodal shape during the day. The hydrophilic mode	
587	becomes stronger in the early morning and in the afternoon, which is attributed to the	
588	NO_x heterogeneous reactions at night, and the aging and growth of pre-existing particles	
589	during the day.	

590 In summary, the diurnal variations in κ -PDF for 40-nm particles were significantly

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(帶格式的:下标 (帶格式的:下标) 591 different during the three periods and the emission control appeared to change the 592 diurnal pattern of κ -PDF, mainly due to the decrease in gas precursors, like SO₂ and 593 NO_x, the reduction of which will suppress the formation of hydrophilic matter.

594 However, the K-PDF for 150-nm D_p particles (i.e., regional-transport particlespre-595 existing particles) had a similar diurnal variation pattern during the three periods (Figure 5c) and showed NH and MH modes. The number fraction of the MH mode 596 increased significantly during daytime. There are different reasons for the diurnal 597 598 variations. One reason is that during daytime, strong photochemistry can produce a 599 large number of condensable vapors, such as sulfuric acid and secondary organic 600 species, which can condense onto pre-existing particles and enhance their water 601 absorbing capacity (Z. J. Wu et al., 2016). Another reason is that when the sun rises, 602 the PBL height increases and older particles are well-mixed, making them more 603 hydrophilic (-S. L. Zhang et al., 2016). However, the MH mode is much more evident 604 for the Pollution case and may be related to the higher mass fractions of inorganic salts 605 and more internal-mix ed particles during the Pollution period.

606 4.2.3. Diurnal variation in aerosol volatility

Figure 6a shows the diurnal variation in SF_{mean} , which shows similar trends for all three cases, with the lowest SF_{mean} in the afternoon (1200–1500 LT) and the highest SF_{mean} in the morning (0700–0900 LT) for all particles with D_p ranging from 40-300 nm. The diurnal variations illustrate that particles had a higher volatility during the day than at night. This feature is more obvious for those small particles observed during the

Clean1 period. During the Clean1 period, particle volatility increased dramatically (i.e.,
$\mathrm{SF}_{\mathrm{mean}}$ decreased) along with the occurrence of NPF events, suggesting that the earliest
newly-formed matter (before \sim 1200 LT) were always volatile at 300°C. As stated
previously, during the Clean1 period VOCs has a weaker decrease than SO2 and NOx,
this would lead to more VOC-formed organic particles formed, which are normally
highly volatile. and likely due to the dramatic reduction in soot particles, the primary
emissions were VOCs from motor vehicles in urban Beijing. This would lead to more
VOC-formed organic particles, which are normally highly volatile. Therefore, the
highest volatility was observed during the Clean1 period. Wehner et al. (2009) also
showed that $\sim 97\%$ of newly-formed particles are volatile because they are dominated
by sulfate and organics VOCs. With the processes of particle aging, collision, and
growth, they then decrease in volatility (i.e., SF_{mean} increases). This is because these
processes can enhance the oxidative level of organics and there are more refractory
organics (like polymer-type organics) produced.
For 40-nm particles, the lowest SF_{mean} appeared two hours later during the Clean2
period (~1500 LT) than during the Clean1 period (~1300 LT). This is probably because
NPF lasted longer during the Clean2 period. For larger particles, the coating effect of
condensable vapors onto pre-existing particles was the major reason behind the
intensification of their volatility during NPF events (Wehner et al., 2009; Cheung et al.,
2016). The SF_{mean} decreased little compared to that for 40-nm particles. By comparison,
the diurnal variation in SF_{mean} for 40-nm particles during the Pollution period changed

more smoothly, likely because under a polluted environment, the mass fractions of all 633

chemical species were relatively stable (Sun et al., 2016a) and the particles were well-mixed with highly aging levels.

636 Figure 6b and 6c show the diurnal variation in SF-PDF for 40 nm and 150 nm 637 particles. The SF-PDF normally has an NV mode and a SV or VV mode. The NV mode consists of non-volatile particles, like BC particles, which do not shrink when aerosols 638 are heated. SV and VV modes suggest a mixture of volatile (e.g., organics) and non-639 640 volatile matter that shrink when aerosols are heated (Kuhn et al., 2005). The two SF-641 PDF modes suggest that the particles during the observed periods were mostly 642 externally-mixed. The 40-nm and 150-nm SF-PDF show similar diurnal patterns. 643 During daytime, active aging processes facilitated the mixing of primary particles with 644 secondary species, leading to the transformation of externally-mixed particles to internally-mixed particles, and weakening the NV mode. In particular, this effect was 645 646 stronger during the Clean1 period than during the other periods. This may be due to the reduction in emissions of soot particles during the control period. In the evening and 647 648 the early morning, the number fraction of NV-mode particles increased again because 649 a large amount of refractory particles (like BC) were emitted from traffic sources or cooking, and then the slower particle aging and weaker vertical mixing made the 650 651 external-mixing BC accumulated in the nocturnal boundary layer (S. L. Zhang et al., 2016). The number fraction of NV-mode 150-nm particles in the Clean1 case had a 652 stronger increase than that of 40-nm particles in the evening and early morning. This is 653 because freshly emitted refractory particles (like BC) are primarily within the 150 nm 654 655 to 240 nm diameter range (Levy et al., 2013). Furthermore, compared with the Pollution case, the number fraction of NV-mode 150-nm particles are much higher during the Clean1 and Clean2 cases. This may reflect the fact that soot particles in a polluted environment can be coated and aged quickly through the heterogeneous reactions of VOCs and other precursor gases (like SO_2 , NO_x), which are usually present in extremely high concentrations during polluted days in urban Beijing (Guo et al., 2014; Sun et al., 2016a).

662 Overall, the diurnal variation in aerosol volatility is different between clean and 663 polluted periods. NPF can enhance volatility through the formation of volatile matter 664 and the coating effect of condensation vapors. Particles observed during the control 665 period showed two significant NV and VV modes during the day, suggesting a more 666 externally-mixed state, particularly for the larger particles.

667 4.3. Size-resolved particle hygroscopic and volatile properties

Table 1 summarizes the size-resolved mean κ , the growth spread factor ($\sigma_{\kappa-PDF}$) of 668 669 κ -PDF, size-resolved SF during different periods, and the change in percentages of κ 670 and SF due to the emission control policy. The $\sigma_{\kappa\text{-PDF}}$, defined as the standard deviation 671 of κ -PDF, is an indication of the mixing state of aerosol particles. A higher σ_{κ -PDF generally suggests a higher degree of external mixing (Sjogren et al., 2008; Liu et al., 672 673 2011; Jiang et al., 2016). Liu et al. (2011) chose $\sigma_{\kappa-PDF} = 0.08$ as the cut-off point for 674 high external mixing and quasi-internal mixing. In this study, $\sigma_{\kappa-PDF}$ always exceeds 675 0.08 but the 40-nm particles during the Clean1 period, indicating that the particle 676 population was more externally mixed in urban Beijing. The mean $\sigma_{\kappa-PDF}$ of 40-nm particles during the Clean1 period is equal to 0.08, suggesting that during the
control period, 40-nm particles had a low degree of external mixing. This is also seen
in the quasi-unimodal distribution of 40-nm κ-PDF (Figure 5b).

680 During the selected three periods, aerosol particles were more hygroscopic (i.e., ĸ increased) with increase in particle size (Figure 7a). The most significant trend is seen 681 in the Pollution case where κ increases from 0.16 to 0.42 when D_p changes from 40 nm 682 to 200 nm, but only increases from 0.10 to 0.25 for the Clean1 case and from 0.14 to 683 684 0.28 for the Clean2 case. This is because particles with a larger size are usually 685 composed of more inorganic salts or oxidized organics, especially in a polluted environment (Swietlicki et al., 2008; Achtert et al., 2009; Fors et al., 2011; Sun et al., 686 687 2016a). Meanwhile, the increase in $\sigma_{\kappa-PDF}$ with the increase in particle size illustrates that there were more external mixing particles with larger sizes (Table 1). Accordingly, 688 689 aerosol volatility became weaker (SF increased) as particle size increased during the Clean1 and Clean2 periods, but no apparent trend was observed for the Pollution period 690 (Figure 7b). This finding is consistent with that reported by Wehner et al. (2009). 691

Figure 6 also shows that all particles were less hygroscopic and more volatile during the control Clean1 period than during the non-control Clean2 period, which can also be seen from the variation in chemical composition. Based on HR-AMS measurements, secondary inorganic aerosols (SIA) had larger decreases than organic aerosols (OA) during the parade control period. The positive matrix factorization of OA further illustrates that primary OA (POA) had similar decreases as secondary OA (SOA). However, more-oxidized SOA had larger decreases than less-oxidized SOA

699 (Zhao et al., 2017)(Zhao et al., 2016, published). SIA is always more hydrophilic than 700 OA and more-oxidized SOA is also more hydrophilic than less-oxidized SOA (Jimenez 701 et al., 2009; Chang et al., 2010; Rickards et al., 2013; Zhang et al., 2014; F. Zhang et 702 al., 2016). Therefore, the increased fraction of POA emissions, but weakened age 703 processing due to a sharp reduction in SO2 and NOx, lead to particles being less hygroscopic during the control period. Meanwhile, particles become relatively more 704 volatile due to the high number of POA particles because OA volatility is generally 705 706 inversely correlated corrected with the O:C ratio (an indicator of oxidation state) 707 (Jimenez et al., 2009).

708 To quantify the effects of emission control on aerosol hygroscopicity and volatility, 709 Table 1 also gives the change in percentages of κ and SF during the control Clean1 period compared with that during the non-control Clean2 period. Results show that ĸ 710 711 decreased by 32.0-8.5% from 40 nm to 200 nm during the control period, with a more significant reduction for small particles, while SF reduced by 7.5-10.5% from 40 nm 712 713 to 300 nm. The significant decrease in aerosol hygroscopicity is favorable for 714 decreasing the aerosol water content, thus suppressing the evolution of regional air 715 pollution (like liquid-phase chemical reaction processes in the atmosphere) (Arellanes 716 et al., 2006; Ye et al., 2011; Bian et al., 2014)(Arellanes et al., 2006; Bian et al., 2014),

717 and eventually improving atmospheric visibility.

In addition, because of the reduced hygroscopicity, fewer particles would be activated as cloud condensation nuclei, which is a critical parameter in evaluating the aerosol indirect effect. Thus, our study is important for investigating environmental and 域代码已更改

721 climate changes, and should inspire both scientists and policy makers to think more

722 deeply about the issue of heavy air pollution in China.

723 4.4.Relationship between nearly-hydrophobic and non-volatile particles

For submicron particles, non-volatile (NV) part particles at 300°C were normally the major nearly-hydrophobic (NH) part particles because both their main components are soot particles (Massling et al., 2009; Wehner et al., 2009). S. L. Zhang et al. (2016) compared the relationship between the number fraction of measured non-volatile particles (NF_{NV}) and nearly hydrophobic (NF_{NH}) particles and found those two groups are very likely to be dominated by the same component. In this study, we also analyze the relationship of NF_{NH} and NF_{NV} particles as shown Figure 8.

731 The results show that Aitken mode particles (40 nm and 80 nm) have a very weak linear relationship between NF_{NH} and NF_{NV}, likely because Aitken mode particles are 732 not as aged. There are a large number of hydrophobic, but volatile, particles such as 733 734 POA and less-oxidized SOA. Accumulation mode particles (> 100 nm) show a 735 relatively better linear correlation between NF_{NH} and NF_{NV}, i.e., correlation coefficients (R²) are 0.26, 0.55, and 0.62 for 110, 150, and 200 nm particles, 736 737 respectively. The higher R² for the larger particles may arise because larger particles 738 are highly aged particles from cloud processes. Also, freshly emitted refractory and 739 hydrophobic matter is mostly in the accumulation mode (Levy et al., 2013). The best-740 fit regression line for the accumulation mode particles is always lower than the 1:1 line. This can be attributed not only to externally mixed SIA and volatile organics 741

(completely volatile), which are not taken into account for calculation, but also to thatsome medium/high volatile organics are nearly-hydrophobic.

744 There were obvious differences in NF_{NH} and NF_{NV} during the three selected 745 periods. For the Clean1 and Clean2 cases, NF_{NH} and NF_{NV} were larger than those obtained for the Pollution case, but more scatter was seen. This is likely related to the 746 influence of NPF events, during which secondary aerosol material had more complex 747 chemical compositions due to the different sources of precursors. A higher NF_{NH}/NF_{NV} 748 749 ratio was seen during the Clean1 period than during the other two periods, illustrating 750 that a higher number fraction of hydrophobic and volatile particles during the control 751 period.

752

753 5. Conclusions and Summary

In this study, a H/V-TDMA system was used to measure submicron aerosol 754 755 hygroscopic and volatile properties in Beijing during and after the parade emission control period. Three periods, namely, the control clean period (Clean1), the non-756 757 control clean period (Clean2), and the non-control pollution period (Pollution), were 758 selected to study the effect of emission control on aerosol hygroscopicity and volatility. 759 When emission control measures were in place, particles became more 760 hydrophobic and volatile compared to particles in the non-control period. The k of 40-200 nm particles decreased by 32.0-8.5% during the Clean1 period relative to the 761 762 Clean2 period, while SF of 40-300 nm particles decreased by 7.5-10.5%. The diurnal

variations of κ -PDF were significantly different during the three selected periods, 763 764 especially for small particles. During the Clean 1 period, the κ-PDF of 40-nm particles 765 always showed a quasi-unimodal distribution and had a weaker diurnal variation than 766 that observed during the Clean2 period. This demonstrates that emission control measures can change the diurnal variation pattern of K-PDF due to the reduction in gas 767 precursors like SO₂ and NO_x, which suppresses the formation of hydrophilic matter. 768 The diurnal variation in aerosol volatility was different between clean and polluted 769 770 periods. NPF appears to enhance aerosol volatility through the formation of volatile 771 matter and the coating effect of condensable vapors. The particles observed during the 772 control period showed two significant modes during the day, i.e., NV and VV modes, 773 and a more externally-mixed state particularly for larger particles.

774 Aerosol particles became more hygroscopic (i.e., κ increases) as the particle size 775 increased during the three periods. The trend was greatest for the Pollution case where κ increased from 0.16 to 0.42 when D_p changed from 40 nm to 200 nm, but only 776 777 increased from 0.10 to 0.25 for the Clean1 case and from 0.14 to 0.28 for the Clean2 778 case. Meanwhile, the increase in $\sigma_{\kappa-PDF}$ (i.e., the standard deviation of $\kappa-PDF$) with the increase in particle size also illustrates that there were more external mixing particles 779 780 with larger sizes. Accordingly, aerosol volatility became weaker (SF increased) as 781 particle size increased during the Clean1 and Clean2 periods, but no apparent trend was observed for the Pollution period. 782

783 Our results suggest that emission control measures weaken submicron aerosol784 hygroscopicity, and that aerosol particles are more hygroscopic in a polluted

785	environment. The significant decrease in aerosol hygroscopicity is favorable for	
786	suppressing the evolution of regional air pollution. In addition, because of the reduced	
787	hygroscopicity, fewer particles would be activated as could condensation nuclei, which	
788	is a critical parameter in evaluating the aerosol indirect effect. Thus, our study is	
789	important for investigating environmental and climate changes, and should inspire both	
790	scientists and policy makers to think more deeply about the issue of heavy air pollution	
791	in China from a broader perspective.	
792		
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1085	Table 1. Summary of size-resolved mean κ , the κ -PDF growth spread factor ($\sigma_{\kappa-PDF}$),
1086	size-resolved mean SF during the selected three periods, and the change in percentage
1087	of κ and SF during the control Clean1 period compared with the non-control Clean2
1088	period.

		40 nm	80 nm	110 nm	150 nm	200 nm	300 nm
Clean l	κ	0.10±0.05	0.11±0.06	0.15±0.07	0.20±0.10	0.25±0.12	
	$\sigma_{\kappa\text{-PDF}}$	0.08±0.03	0.10±0.03	0.12±0.03	0.14±0.03	0.15±0.04	_
	SF	0.55±0.08	0.60±0.07	0.64±0.06	0.66±0.05	0.67±0.05	0.70±0.06
Clean2	κ	0.14±0.06	0.17±0.08	0.20±0.10	0.24±0.12	0.28±0.13	
	$\sigma_{\kappa\text{-PDF}}$	0.11±0.04	0.13±0.03	0.15±0.03	0.17±0.03	0.19±0.04	_
	SF	0.60±0.06	0.66±0.07	0.70±0.07	0.72±0.07	0.74±0.06	0.78±0.06
Pollution	к	0.16±0.08	0.24±0.08	0.30±0.09	0.36±0.10	0.42±0.12	_
	$\sigma_{\kappa\text{-PDF}}$	0.12±0.02	0.13±0.02	0.14±0.02	0.14±0.0.02	0.15±0.04	_
	SF	0.65±0.06	0.65±0.06	0.65±0.05	0.65±0.05	0.65±0.06	0.66±0.07
(Clean1 – Clean2) Clean2	к	-32.0%	-31.9%	-26.1%	-17.5%	-8.5%	
	SF	-7.5%	-9.4%	-9.2%	-8.7%	-10.1%	-10.5%



1099 red), and (b) ambient temperature at 8 m (in blue) and relative humidity at 8 m (in red)

1100 during the control and non-control periods.







1108 Figure 4. Diurnal variation in mobility diameter (D_p) 10-400 nm particles (a) number

1109 concentration $(N_{10-400 \text{ nm}})$ and (b) mass concentration $(p_{10-400 \text{ nm}} \underline{PM}_{10-400 \text{ nm}})$ for the

带格式的: 下标



1110 Clean1 (in red), Clean2 (in green), and Pollution (in blue) cases.



1113 $\kappa\text{-PDF}$ for particles with D_p equal to 40 nm, and (c) $\kappa\text{-PDF}$ for particles with D_p equal

to 150 nm during the Clean1, Clean2, and Pollution periods.



1116 Figure 6. Diurnal variation of (a) mean SF (SF_{mean}) for different mobility diameters,



1118 equal to 150 nm during the Clean1, Clean2, and Pollution periods.

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1120 Figure 7. Size-resolved (a) κ and (b) SF during Clean1 (C1), Clean2 (C2), and

1121 Pollution (P) periods. The figure shows the mean κ or SF (solid square markers) with

1122 boxes showing the 25th, 50th, and 75th percentiles. The extremities show the 5th and

1123 95th percentiles.

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1126 (NF_{NH}) and the non-volatile group (NF_{NV}) for the Clean1 (in green), Clean2 (in blue),

1127 and Pollution (in red) periods.

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