

# Four years (2011-2015) of Total Gaseous Mercury Measurements from the Cape Verde Atmospheric Observatory

Katie A Read<sup>1</sup>, Luis M Neves<sup>2</sup>, Lucy J Carpenter<sup>1</sup>, Alastair C Lewis<sup>1</sup>, Zoe Fleming<sup>3</sup>, and John Kentisbeer<sup>4</sup>

<sup>1</sup>National Centre for Atmospheric Science (NCAS), Department of Chemistry, University of York, York, YO10 5DD, UK

<sup>2</sup>Instituto Nacional de Meteorologia Geofisica (INMG), Delegação de São Vicente, Monte, CP15, Mindelo, Rep of Cape Verde

<sup>3</sup>National Centre for Atmospheric Science (NCAS), University of Leicester, Leicester, LE1 7RH, UK

<sup>4</sup>Centre for Ecology and Hydrology (CEH), Bush Estate, Penicuik, Midlothian, EH26 0QB, UK

Correspondence to: Katie A. Read (katie.read@york.ac.uk)

**Abstract.** Mercury is a chemical with widespread anthropogenic emissions that is known to be highly toxic to humans, ecosystems and wildlife. Global anthropogenic emissions are around 20% higher than natural emissions and the amount of mercury released into the atmosphere has increased since the industrial revolution. In 2005 the European Union and United States adopted measures to reduce mercury use, in part to offset the impacts of increasing emissions in industrialising countries. The changing regional emissions of mercury have impacts on a range of spatial scales. Here we report four years (Dec 2011 – Dec 2015) of Total Gaseous Mercury (TGM) measurements at the Cape Verde Observatory (CVO), a global WMO-GAW station located in the sub-tropical remote marine boundary layer. Observed total gaseous mercury concentrations were between 1.03 and 1.33 ng m<sup>-3</sup> (10<sup>th</sup>, 90<sup>th</sup> percentiles), close to expectations based on previous interhemispheric gradient measurements. We observe a decreasing trend in TGM ( $-0.05 \pm 0.04$  ng m<sup>-3</sup> yr<sup>-1</sup>,  $-4.2\% \pm 3.3\%$  yr<sup>-1</sup>) over the four years consistent with the reported decrease of mercury concentrations in North Atlantic surface waters and reductions in anthropogenic emissions. The decrease was more visible in the summer (Jul-Sep) than in the winter (Dec-Feb), when measurements were impacted by air from the African continent and Sahara/Sahel regions. African air masses were also associated with the highest and most variable TGM concentrations. We suggest that the less pronounced downward trend inclination in African air may be attributed to poorly controlled anthropogenic sources such as artisanal and small-scale gold mining (ASGM) in West Africa.

## 1 Introduction

Mercury is present in the atmosphere in three main forms; gaseous elemental mercury Hg<sup>0</sup>, which is the most common form in the gas phase, oxidized mercury Hg<sup>II</sup> (GOM or RGM), and Hg-bound to particulate matter (PBM). Total Gaseous Mercury (TGM) is the combined measurement of Hg<sup>0</sup> (or Gaseous Elemental Mercury (GEM)) + RGM, with Hg<sup>0</sup> typically contributing around 90-99% of the total Hg or TGM.

Anthropogenic sources of mercury account for around 30% of the total amount and include emissions from coal burning, mining, cement production, oil refining and waste incineration. One third of the anthropogenic emissions are thought to come from deliberate biomass burning with Africa as the single largest continental source; therefore in this region there could be an influence from Sahel African biomass burning during the months of November through to February (Roberts et al, 2009, De Simone et al., 2015). Hg<sup>0</sup> reacts slowly with atmospheric oxidants with a global lifetime of around 6-8 months (Selin et al., 2007; Holmes et al., 2010), and so can be transported to remote regions. When oxidized to less volatile Hg<sup>II</sup>, it can be deposited either through wet deposition processes (precipitation-scavenging) or by surface uptake (Gustin et al., 2012; Schroeder and Munthe, 1998; Sather et al., 2013; Wright et al., 2014). Hg<sup>0</sup> also undergoes slow dry deposition through air-surface exchange with both terrestrial and aquatic surfaces (Zhang et al., 2009; Wang et al., 2016). Once deposited, transformation to highly toxic species such as the neurotoxic methylmercury allows bioaccumulation in food chains and poses a health risk to humans and a damaging effect to ecosystems (US EPA, 1997). Previously deposited mercury can also be reduced back to Hg<sup>0</sup> through the natural weathering of

1 mercury-containing rocks, geothermal activity, or from volcanic activity, and then re-emitted back to  
2 the atmosphere (Smith et al., 2008)(Qureshi., 2012).

3  
4 Reactions of  $\text{Hg}^0$  to  $\text{Hg}^{\text{II}}$  with the hydroxyl radical (OH) and ozone ( $\text{O}_3$ ) were historically accepted as  
5 the dominant photochemical oxidation mechanisms (Bergan and Rodhe, 2001;Lin et al., 2006;Seigneur  
6 et al., 2006;Selin et al., 2007;Pongprueksa et al., 2008). Recent work has suggested that there may be  
7 significant other oxidants such as atomic halogens (Holmes et al., 2010;Wang et al., 2014) and more  
8 complex two-step oxidation schemes, which include further reactions with  $\text{NO}_2$  and  $\text{HO}_2$ , however the  
9 kinetics are highly uncertain (Goodsite et al., 2004) (Goodsite et al., 2004;Holmes et al., 2010).  
10 Heterogeneous oxidation in clouds may also contribute but is not experimentally proven (Ariya et al.,  
11 2009;Calvert and Lindberg, 2005).

12  
13 Strode et al. (2007) estimated that 36% of all mercury emissions in the northern hemisphere come from  
14 the ocean both through primary emission (ocean upwelling and mercury-containing rocks) and from re-  
15 emission of previously deposited mercury (as  $\text{Hg}^{\text{II}}$ ), but this increases to 55% as you move into the  
16 southern hemisphere (Strode et al., 2007). The major anthropogenic source affecting the remote  
17 marine boundary layer is likely to be long-range transport of  $\text{Hg}^0$  from combustion (smelting, waste  
18 incineration, chemical plants) rather than from  $\text{Hg}^{\text{II}}$ , which is more likely to deposit regionally due to  
19 its relatively short lifetime of 4.8 hours (Zhang et al., 2012). Other industrial sources for Hg include  
20 artisanal and small-scale gold mining (ASGM), which are known to occur in West Africa (Telmer and  
21 Velga, 2009; UNEP, 2013) and will likely regionally influence the measurements described here. For  
22 the 2013 UNEP global assessment, ASGM emission data were compiled from field and industry  
23 reports but with an uncertainty of ca.  $\pm 43\%$  due to the multitude and varying nature of ASGM sites. In  
24 recent years, global emissions from ASGM, and in particular the proportion of global emissions  
25 attributed to South America and Sub-Saharan Africa, appear to be increasing; however this assumption  
26 may be due to improved reporting (Muntean et al., 2014). The majority of global anthropogenic  
27 emissions of Hg to the atmosphere in 2010 are associated with ASGM (37%), with one third thought to  
28 be from sub-Saharan Africa (UNEP, 2013).

29  
30 A community strategy developed by the EU was adopted in 2005 and listed 20 actions to reduce  
31 mercury emissions, cut mercury supply and demand, and to protect people against exposure. This  
32 strategy had a strong focus on the need to take a global approach and included actions relating to  
33 multilateral negotiations for the conclusion of a legally binding convention on mercury  
34 ([http://ec.europa.eu/environment/chemicals/mercury/strategy\\_en.htm](http://ec.europa.eu/environment/chemicals/mercury/strategy_en.htm)). The UNEP Global Mercury  
35 partnership led by the US Environmental Protection Agency took a similar approach  
36 (<http://www.unep.org/chemicalsandwaste/Metals/GlobalMercuryPartnership/tabid/1253/Default.aspx>)  
37 and these initiatives formed the basis of the Minamata Convention on Mercury, which was agreed in  
38 2013 and is a global treaty to protect human health and the environment from the adverse effects of  
39 mercury (<http://www.mercuryconvention.org/>).

40  
41 It has been a source of contradiction that in the northern hemisphere, while both measured atmospheric  
42 Hg concentrations and wet deposition fluxes have been decreasing since 1990 (Soerensen et al., 2012)  
43 and 1996-2013 (Slemr et al., 2011;Weigelt et al., 2015); global Hg emissions during this period were  
44 calculated to be increasing (Pacyna et al., 2010;Streets et al., 2011). Very recently, however, Zhang et  
45 al., (2016), using a revised inventory and the global model GEOS-CHEM, have shown that global Hg  
46 emissions may also be decreasing. They suggest that a large discrepancy in the emissions data was  
47 from locally deposited mercury close to coal-fired utilities. It is thought that this source has declined  
48 more rapidly than was previously predicted due to shifts in mercury speciation from air pollution  
49 control technology targeted at  $\text{SO}_2$  and  $\text{NO}_x$  (Zhang et al., 2016). Flue gas desulfurization (FGD) -  
50 which controls  $\text{SO}_2$  emissions - washes out  $\text{Hg}^{\text{II}}$ , whilst selective catalytic reduction (SCR) to control  
51  $\text{NO}_x$  emissions also oxidises  $\text{Hg}^0$  to  $\text{Hg}^{\text{II}}$ . These effects of FGD, in addition to the recent phase-out of  
52 Hg from commercial products (UNEP Minamata Convention on Mercury) and lower global estimates  
53 from small-scale gold mining, serve to explain the globally decreasing atmospheric concentrations in  
54 the model. Zhang et al. (2016) also found that the larger emission decreases observed in North America  
55 and Europe globally offset the increases from other major polluted regions e.g. from coal-fired utilities  
56 in East Asia (Pacyna et al., 2010;Pirrone et al., 2013).

57  
58 Using data from ship cruises, Soerensen et al. (2012) observed a significant decreasing trend of  
59 atmospheric mercury concentrations over the North Atlantic of  $-0.046 \text{ ng m}^{-3} \text{ yr}^{-1}$  ( $-2.5\% \text{ yr}^{-1}$ ), with  
60 smaller trends at more southern latitudes (Soerensen et al., 2012). They suggest that this decline is due

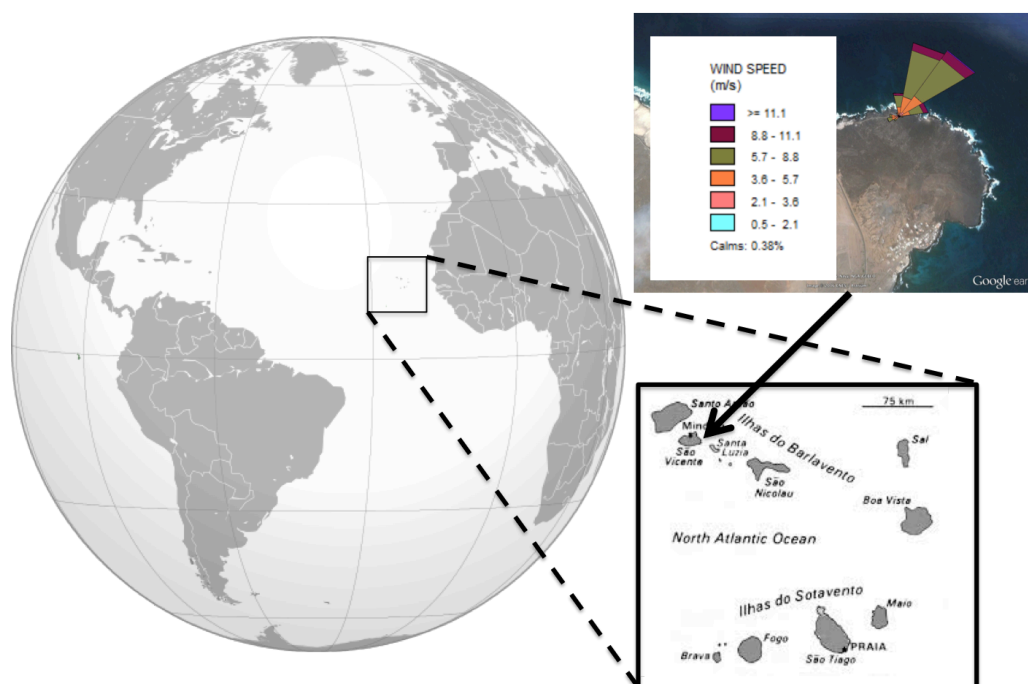
1 to decreasing oceanic evasion driven by declining subsurface water  $Hg^0$  concentrations ( $-5.7\% \text{ yr}^{-1}$   
2 since 1999, (Mason et al., 2012)).

3  
4 Here we report four years (Dec 2011 – Dec 2015) of TGM measurements at the Cape Verde  
5 Observatory (CVO), a clean marine background station located in the subtropical Atlantic. The  
6 measurements presented here are part of the EU Global Mercury Observation System (GMOS)  
7 network. The GMOS network of sites was established in 2011 with the aim of addressing known gaps  
8 in the spatial and temporal measurement of mercury, as well as improving knowledge of Hg speciation.  
9 The data is being used to validate regional and global scale atmospheric Hg models in order to improve  
10 understanding of global Hg transport, deposition and re-emission as well as providing a contribution to  
11 future international policy development and implementation ([www.gmos.eu](http://www.gmos.eu)).

## 12 2 Experimental

13  
14  
15 The CVO was established in 2006 as a multilateral project between the UK, Germany and Republic of  
16 Cape Verde. Long-term atmospheric measurements include reactive trace gases including ozone,  
17 carbon monoxide, nitrogen oxides and volatile organic compounds (National Centre for Atmospheric  
18 Sciences (NCAS), University of York, UK), long lived greenhouse gases (Max-Planck Institute (MPI),  
19 Jena, Germany), and physical and chemical characterisation of aerosol (Leibniz Institute for  
20 Tropospheric Research (TROPOS), Leipzig, Germany). Details of the measurements and  
21 characteristics of the station can be found in Carpenter et al., (2010).

22  
23 The CVO is positioned on the northeast side of Sao Vicente (16.85°N, 24.87°W), one of ten islands in  
24 the Cape Verde archipelago (Fig. 1). The island is of volcanic origin and the CVO is situated 50m  
25 from the coastline. The climate is warm (mean annual air temperature is  $24.0^\circ\text{C} \pm 2.0^\circ\text{C}$ ) and dry with  
26 extremely low annual rainfall ( $<200 \text{ mm}$ ), which occurs mostly during the rainy season of July-  
27 November. The site receives air masses from the northeasterly trade winds for 95% of the time, which  
28 have travelled typically for five days over the ocean. Research flights carried out over the CVO in  
29 summer 2007 established that the boundary layer is well mixed (Read et al., 2008). There is no coastal  
30 shelf to the island at this location point, and the CVO conditions are considered to be representative of  
31 the North Atlantic open ocean boundary layer. Radiosonde and ceilometer data show there is no  
32 diurnal pattern evident in boundary layer heights, which suggests no systematic difference between  
33 day-time and night-time entrainment rates (Carpenter et al., 2010).



35  
**Figure 1: Cape Verde site location. Top right, image from Google earth: V7.1.5.1557 (6<sup>th</sup> July 2016). São Vicente, Cape Verde, 16°51'59.60"N, 24° 52'03.60"W, Eye altitude 2.70 km. Wind rose for the measurement period is coloured by wind speed.**

1 Air is sampled from the main laboratory glass manifold (10 m height of inlet, 2" diameter, residence  
2 time 4 seconds) and then through a 2 m length of ¼" Teflon tubing and a particulate filter which is  
3 changed every two months. The entire inlet is heated. A TEKRAN 2537B analyser (Tekran Inc.,  
4 Toronto, Canada) was used for the TGM measurements and is described in detail elsewhere (Steffen et  
5 al., 2014) and so only a brief summary is presented here. The analytical principle collects the TGM  
6 onto gold traps with subsequent thermal desorption and detection by atomic fluorescence spectroscopy  
7 ( $\lambda = 253.7$  nm, (Bloom and Fitzgerald, 1988)). It is however likely that the measurement at this site is  
8 of GEM rather than TGM since RGM is lost very easily to any salt deposits in the inlet lines and filters.  
9 Samples of 5 L volume are obtained every 5 minutes ( $1 \text{ L min}^{-1}$  flow rate) with a detection limit of  
10 around  $0.1 \text{ ng m}^{-3}$ , using a dual trap set-up. Concentrations in  $\text{ng m}^{-3}$  are reported at a standard pressure  
11 of 1013 hPa and a standard temperature of 273.14K. Calibrations are performed every 72 hours using  
12 an internal mercury permeation source which injects a known amount of  $\text{Hg}^0$  into mercury-free zero air  
13 (using a TEKRAN Zero Air filter, part no: 90-25360-00). The calibration consists of a zero and a span  
14 on each channel. The effective span was  $19.08 \text{ ng m}^{-3}$  for a sample volume of 5 L. The permeation  
15 rate was externally validated using manual injections of saturated mercury vapour taken from a Tekran  
16 2505 mercury vapour calibration unit and after 5 years found to be within  $\sim 3.6\%$  of the instrumental  
17 set-point. The detection limit of the instrument was  $0.1 \text{ ng m}^{-3}$ .

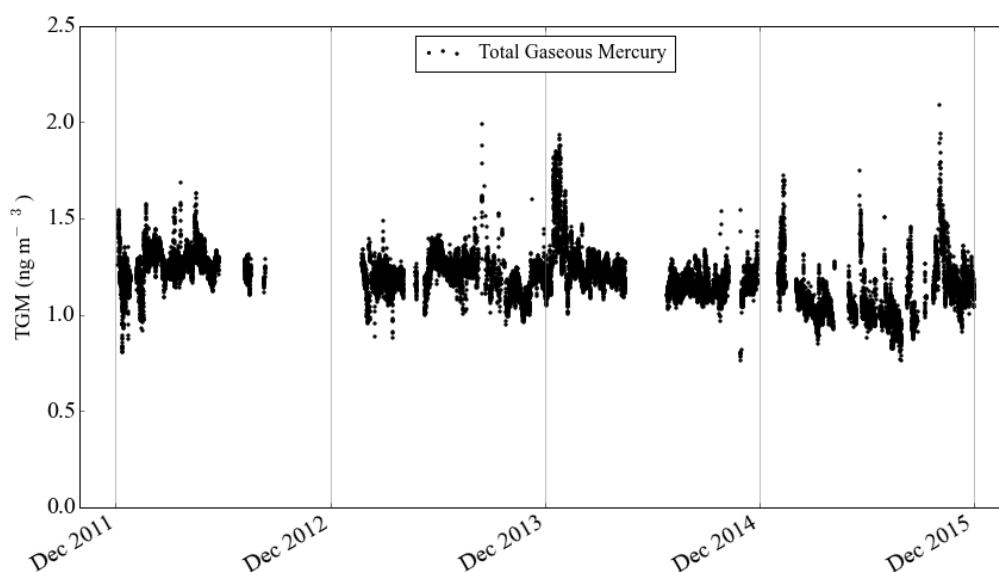
18 Instruments to make trace gas and meteorological measurements are provided by the Atmospheric  
19 Measurement Facility (AMF), which is part of the National Centre for Atmospheric Science (NCAS).  
20 Ozone measurements were made using a UV photometric analyser (Thermo Electron Corporation).  
21 The instrument had a detection limit of 0.05 ppb and a precision of  $<1$  ppbV. Carbon monoxide data  
22 presented here was measured using a Vacuum UV fluorescence technique (Aerolaser 5001). It was  
23 sensitive to 1 ppbV and linear up to 100 ppm. The accuracy of the measurements was  $<2$  ppbV. In  
24 2012 the Global Atmospheric Watch (GAW) audited these measurements and the report can be found  
25 at [http://www.wmo.int/pages/prog/arep/gaw/documents/CVO\\_2012.pdf](http://www.wmo.int/pages/prog/arep/gaw/documents/CVO_2012.pdf). Nitrogen oxide  
26 measurements were made using a low detection, high accuracy, (sensitivity 0.3 and 0.35 pptV and  
27 accuracy of 5.5% and 5.9% for NO and NO<sub>2</sub> respectively) chemiluminescence analyser (AQD Inc.).  
28 The meteorological measurements presented here were made using a Campbell Scientific Automatic  
29 Weather station. For more information about the present NCAS instrumentation at the CVAO refer to  
30 [https://www.ncas.ac.uk/index.php/en/the-facility-amf/291-amf-main-category/cvao/cvao-amf-](https://www.ncas.ac.uk/index.php/en/the-facility-amf/291-amf-main-category/cvao/cvao-amf-instrumnets/1557-cvao-amf-instrumnets)  
31 [instrumnets/1557-cvao-amf-instrumnets](https://www.ncas.ac.uk/index.php/en/the-facility-amf/291-amf-main-category/cvao/cvao-amf-instrumnets/1557-cvao-amf-instrumnets).

32  
33 Four years of data are presented here obtained between 5<sup>th</sup> December 2011 and 5<sup>th</sup> December 2015. In  
34 calculating annual statistics, we have used data from 1 Dec – 30 Nov. The data was quality controlled  
35 using the central GMOS-Data Quality Management (G-DQM) system (Cinnirella et al., 2014; D'Amore  
36 et al., 2015). The G-DQM allows harmonization of data across the network and is able to acquire and  
37 process data in near real time allowing immediate diagnosis of issues. It was developed using  
38 harmonized Standard Operating Procedures, which had been established over many years by European  
39 and Canadian monitoring networks, together with recent literature (Brown et al., 2010; Gay et al.,  
40 2013; Steffen et al., 2012). An additional filter has been applied to the data presented here to exclude  
41 periods when the relative humidity was higher than 90%, as the data was prone to increased  
42 uncertainties due to water condensing in the instrument. Instrument issues led to some significant data  
43 gaps; a lamp failure caused major data gaps between July-August 2012 and May-June 2014, whilst a  
44 pump failure caused downtime between October 2012-January 2013.

### 45 **3 Results and discussion**

#### 46 **3.1 Statistics and seasonal cycles**

47  
48  
49  
50 The mean TGM concentration over 2011-2015 was  $1.191 \pm 0.128 \text{ ng m}^{-3}$  and the four-year time-series  
51 is shown in Fig. 2. Sprovieri et al. (2016) showed that the CVO measurements (site referred to as CAL  
52 rather than CVO) fit well within the north-south gradient of TGM data. Other sites of reference, which  
53 receive background air similar in origin to CVO, include Mace Head, Ireland, Nieuw Nickerie,  
54 Suriname, and Cape Point, South Africa (Table 1). The remoteness of the CVO is reflected in the small  
55 variability of the TGM measurements, compared to other sites.  
56



1  
2  
3 **Figure 2: Time-series (December 2011-December 2015) of TGM data measured at the Cape Verde**  
4 **Observatory.**

5  
6 The data shown in Table 1 illustrate the dominating effect of emissions from the northern hemisphere  
7 compared to the southern hemisphere, with Mace Head (53°20'N, 9°54'W) TGM concentrations  
8 averaging 7-9% higher than those observed at Cape Point. The site at Nieuw Nickerie experiences 10%  
9 higher concentrations in the air arriving from the north compared to the south (Muller et al., 2012) and  
10 is additionally impacted by emissions from biomass burning and gold mining from South America  
11 (Sprovieri et al., 2010). Comparisons to ship-borne field campaigns in the Atlantic made between 1977  
12 and 2001 (Sprovieri et al., 2010) show that the data from CVO are more comparable with southern  
13 Atlantic conditions than the northern Atlantic.  
14

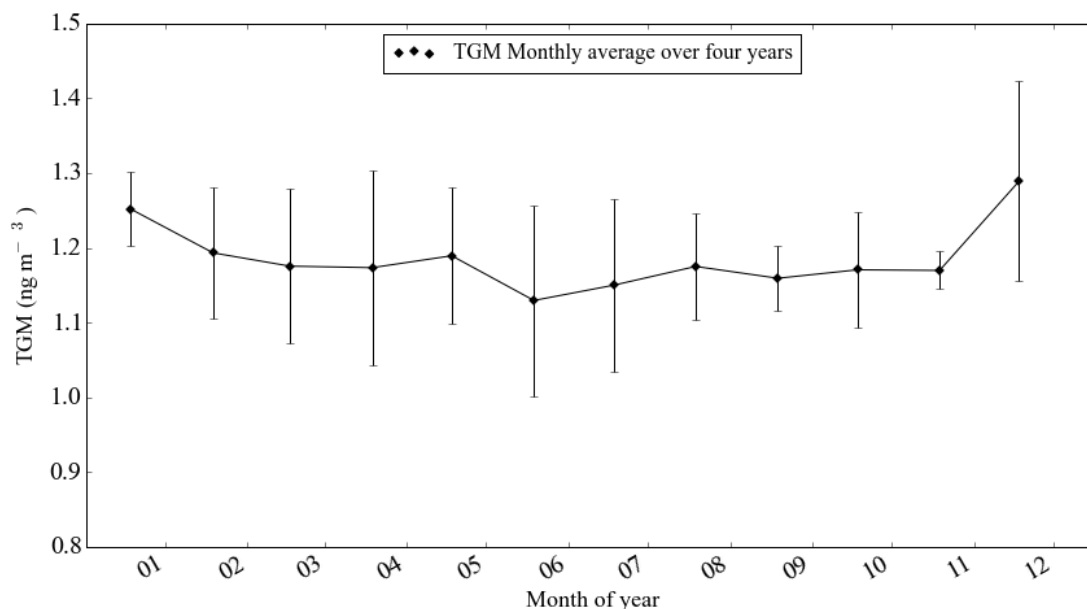
Site (Latitude, Longitude)	Mean $\pm$ standard deviation for 2013 (ng m <sup>-3</sup> )	Mean $\pm$ standard deviation for 2014 (ng m <sup>-3</sup> )
Mace Head, Ireland (53°20'N, 9°54'W)	1.46 $\pm$ 0.17	1.41 $\pm$ 0.14
Calhau, Rep of Cape Verde (16°51'N, 24°52'W)	1.22 $\pm$ 0.14	1.20 $\pm$ 0.09
Nieuw Nickerie, Suriname (5°56'N, 56° 59'W)	1.13 $\pm$ 0.42	1.28 $\pm$ 0.46
Cape Point, South Africa (33° 56'S, 18°28'E)	1.03 $\pm$ 0.11	1.09 $\pm$ 0.12

15  
16 **Table 1. Average TGM concentrations and standard deviation statistics from comparable sites in 2013 and**  
17 **2014. Data from Sprovieri et al., (2016).**

18  
19 The CVO TGM monthly mean data shows a weak seasonal cycle ( $1.289 \pm 0.134$  ng m<sup>-3</sup> December  
20 maximum,  $1.130 \pm 0.128$  ng m<sup>-3</sup> June minimum, Fig. 3) with generally higher concentrations in winter  
21 and lower in summer. This cycle is similar, both in shape and magnitude, to that observed within sub-  
22 tropical maritime air masses at Mace Head, which is shallower than for other air masses (Weigelt et al.,  
23 2015) and generally not so defined as that of other remote sites in the Northern Hemisphere (Temme et  
24 al., 2007;Holmes et al., 2010). Selin et al., (2007) show that the mean seasonal amplitude of 12  
25 northern mid-latitude sites between the maximum in January (winter) and minimum August (summer)  
26 is  $0.19$  ng m<sup>-3</sup>, compared to the CVO amplitude of  $0.14$  ng m<sup>-3</sup> (December-June).

27 A smaller seasonal cycle may mean that O<sub>3</sub> plays a more dominant role in the oxidation of Hg<sup>0</sup>  
28 compared to other oxidants such as OH (Temme et al., 2007;Selin et al., 2007;Holmes et al., 2010).  
29 The equatorial nature of the CVO site means that solar irradiance and water vapour are high year-round  
30 (Carpenter et al., 2010;Whalley et al., 2010). This may lead to a less pronounced change in oxidation  
31 capacity between summer and winter, when compared to sites at higher latitudes.

1 An influence of air masses from the southern hemisphere without any pronounced seasonal variation  
 2 (Slemr et al., 2015) may be another reason for the smaller amplitude in the seasonal cycle at the CVO.  
 3 However, air mass back trajectory analyses show that the CVO receives very little air representative of  
 4 the southern hemisphere (~1.3% of all data, Fig. 1, Supplementary Information). Further, the highest  
 5 frequency of southerly air masses arriving at the CVO occurs during August and September, which  
 6 would serve to increase the mercury seasonal cycle amplitude rather than reduce it.  
 7



8 **Figure 3: Seasonal cycle of TGM at CVO. The bars represent the standard deviation of the monthly**  
 9 **averages.**

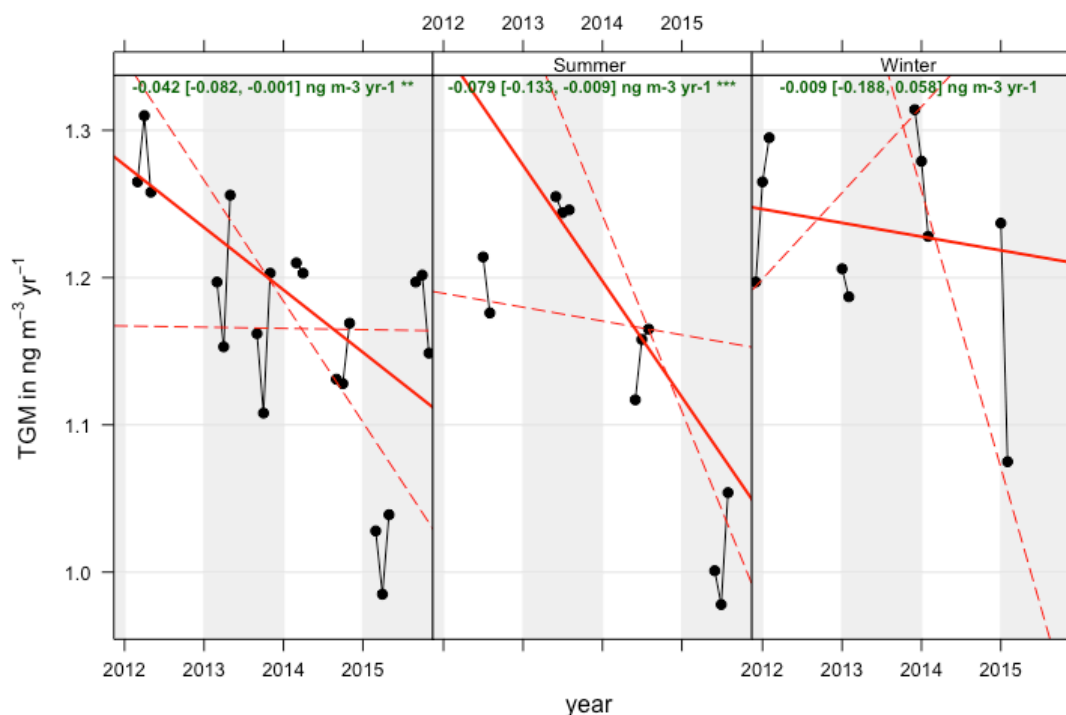
10 Anthropogenic emissions of mercury affecting the Atlantic region include emissions from coal  
 11 combustion, which tend to have maximum impact in February-March due to a dominance of air from  
 12 continental regions such as North America. This is also observed in the seasonal distribution of  
 13 anthropogenic combustion tracers such as carbon monoxide (Selin et al., 2007; Weigelt et al., 2015;  
 14 Read et al., 2009). Ocean emissions of Hg<sup>0</sup> from the reduction of Hg<sub>aq</sub><sup>II</sup> to Hg<sub>aq</sub><sup>0</sup>, driven by increased  
 15 biological production are at a maximum in June in the NH but December in the SH (Strode et al.,  
 16 2007). The seasonal trend may also be affected by meteorological differences in seasonal circulation  
 17 patterns and cycles in boundary layer heights, clouds, precipitation and dry deposition characteristics  
 18 (Dastoor and Larocque, 2004; Selin et al., 2007).  
 19

### 20 3.2 Four-year trends

21  
 22 The Theil-Sen function (Theil 1950; Sen 1968) was used to evaluate the 4-year dataset inclination  
 23 based on monthly TGM medians by season. The results are shown in Fig. 4. In this function the  
 24 slopes between all x, y pairs are calculated and the Theil-Sen estimate is the median of all these slopes.  
 25 This analysis was performed using the Openair package in R (Carlsaw et al., 2012). The advantage of  
 26 using this function is that it gives accurate confidence intervals and is resistant to outliers. Statistics  
 27 used for this plot can be found in Table 1 in the Supplementary information.  
 28

29 Over four years the data shows a weak downward inclination ( $0.042 \pm 0.04 \text{ ng m}^{-3} \text{ yr}^{-1}$ ,  $p < 0.01$   
 30 significance level). This decrease is more significant in the data collected during the Cape Verdean  
 31 summer ( $-0.079 \pm 0.054 \text{ ng m}^{-3} \text{ yr}^{-1}$ ,  $p < 0.001$  in June-August) than in the winter ( $-0.009 \pm 0.179 \text{ ng m}^{-3}$   
 32  $\text{yr}^{-1}$ ,  $p < 0.1$  in December-February). Previous studies have shown a stronger decreasing trend in air  
 33 that has been influenced by anthropogenic emissions, for which there is more ready detection of the  
 34 impact of regulation (Selin et al., 2007; Cole et al., 2014; Weigelt et al., 2015; Zhang et al., 2016).  
 35

1



2

3

4

**Figure 4: TGM trends for the full year and then separated by season at the Cape Verde Observatory. The green text shows the slope estimate, with 99% confidence intervals in brackets.**

5

6

7

8

9

10

11

12

13

14

15

16

17

18

The seasonal trends calculated here imply that there are differences in the sources of mercury that affect the winter months compared to the summer months, potentially with a smaller decline in emissions over this winter period. This may be because CVO measures Hg coming from the same source region throughout the year but that the emission from that source has not declined as much in winter as it has in summer, e.g. from residential burning. Alternatively the difference could be explained by a difference in air mass between seasons bringing air from different sources that have experienced different trends in emissions over the years. We consider this latter scenario to be the more likely explanation, since air masses originating from continental Africa, which may be influenced by ASGM or biomass burning, frequently reach the CVO in winter (Carpenter et al., 2010), but are more rare in summer. A further alternative explanation for the difference in trends between seasons would be a change in global oxidant concentrations (such as OH) and that this effect had a seasonal dependence, but there is no evidence to support this from studies that estimate OH fields (Hartmann et al., 2013).

19

20

21

22

23

24

25

26

27

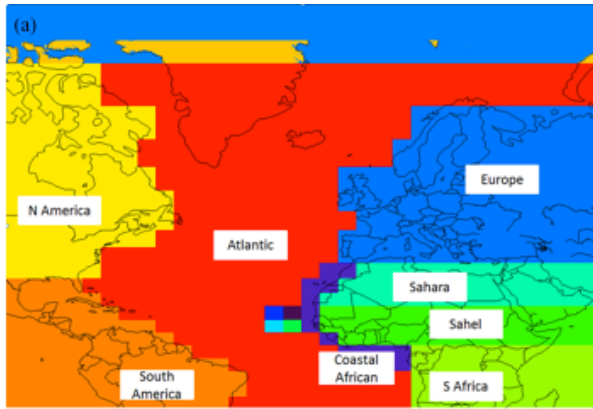
28

29

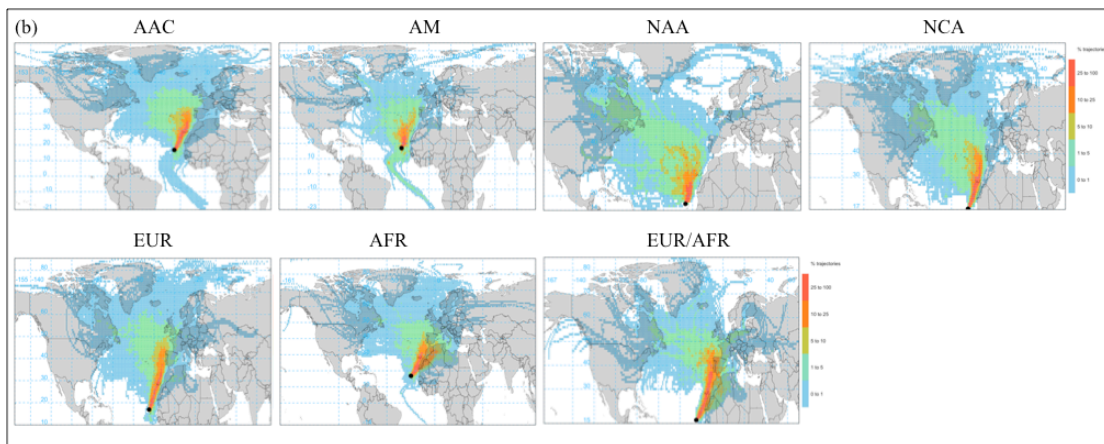
30

31

In order to understand better the drivers of the TGM behaviour, observations were classified according to the origin and pathways of air masses arriving at the CVO over a ten-day period using the UK Met Office NAME dispersion model in passive tracer mode (Ryall et al., 2001). The air mass classifications have been used previously for evaluating the source regions of reactive trace gases arriving at CVO (Carpenter et al., 2010). For this study eight geographical regions were defined (Coastal African, polluted Marine, Saharan Africa, Sahel Africa, North America, Atlantic marine South America, and, Tropical Africa) and from these, 7 air mass types are classified based on the percentage time spent over each of the 8 regions (Figures 5a and b). These are: Atlantic and African Coastal (AAC), Atlantic marine (AM), North American and Atlantic (NAA), North American and coastal African (NCA), European (with minimal African influence) (EUR), African (with minimal European influence) (AFR) and European and African (EUR/AFR). The eight regions are shown in Figure 5a and a trajectory frequency footprint of the trajectories (using all of the data from the measurement period), for each of the 7 classifications is shown in Figure 5b.



1



2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

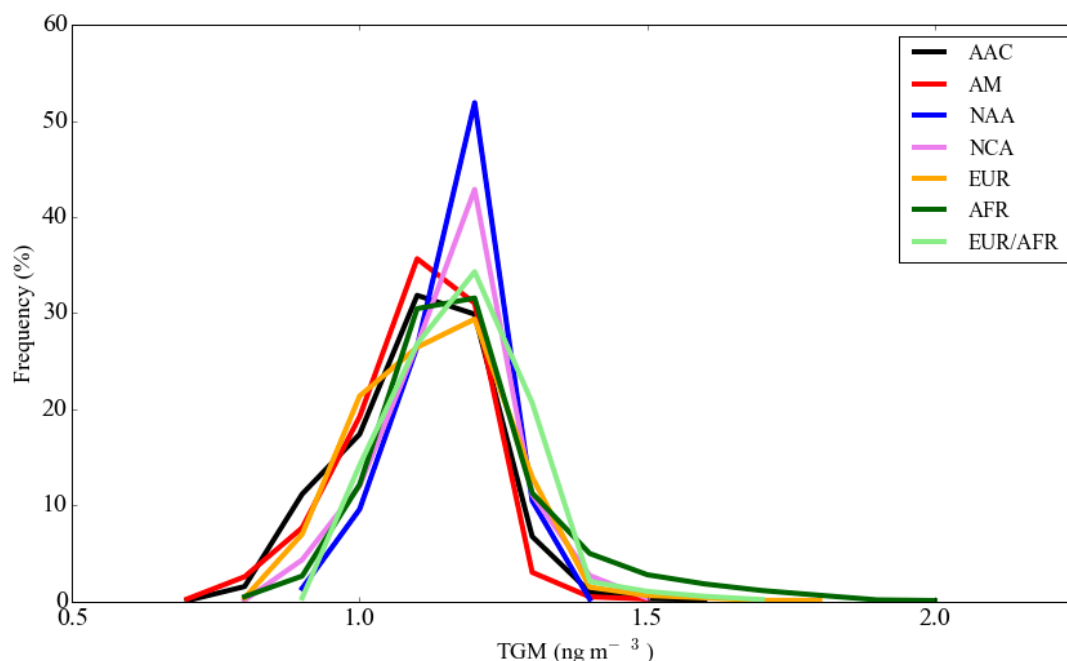
20

21

Figure 5: a) Boundary definition of the eight geographical regions, Coastal African, polluted Marine, Saharan Africa, Sahel Africa, North America, Atlantic marine South America, and, Tropical Africa. b) Trajectory frequency maps for each of the seven air mass types using HYSPLIT trajectories and Openair, AAC - Atlantic and African Coastal, AM - Atlantic Marine, NAA - North American and Atlantic, NCA - North American and Coastal African, EUR - European (with minimal African influence), AFR - African (with minimal European influence) and EUR/AFR - European and African.

Figure 6 shows histograms representing the data in each of the 7 classifications and Table 2 details the associated statistics. The lowest variability in TGM was observed in air that had travelled the longest period since contact with continental sources even though these would have been subjected to greatest potential for ocean emissions (AM, NCA, NAA). The lowest concentrations ( $1.144 \pm 0.109 \text{ ng m}^{-3}$ ) were observed in Atlantic and African coastal air (AAC). During late summer we occasionally ( $\sim 2\%$  of time) receive air that has been influenced by the Southern hemisphere (Figure 5b and Supplementary Information). At these times the ozone mixing ratios drop to  $\sim 15 \text{ ppbV}$  and there can be a rare occurrence of rain leading to spikes in the mostly low TGM concentrations (Figure 7). Otherwise, the highest and most variable concentrations of total gaseous mercury (mean of  $1.23 \pm 0.16 \text{ ng m}^{-3}$ ) are observed in air originating from continental Africa (AFR).





1  
2 **Figure 6: Histograms of observed TGM classified by airmass for the full 2011-2015 dataset.**

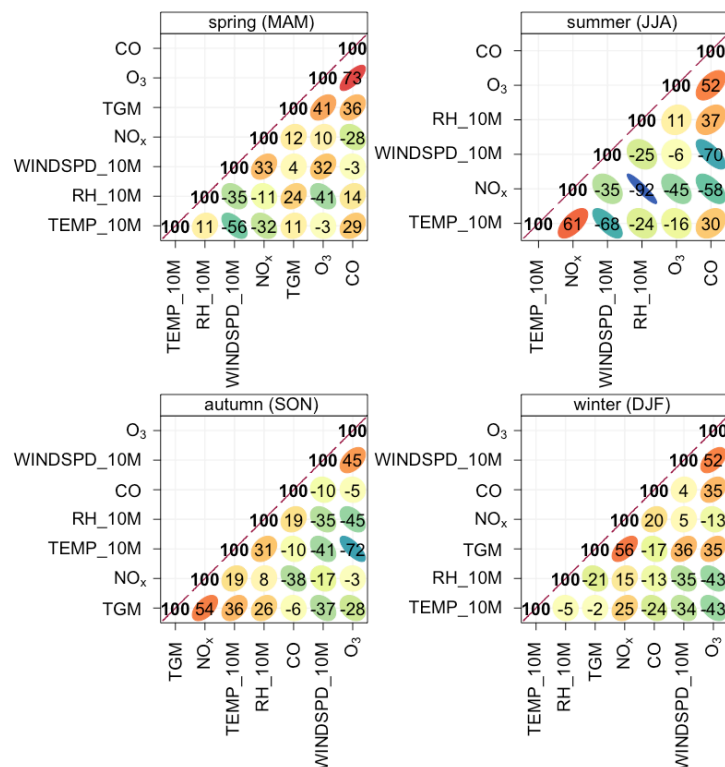
Air mass	Mean +/- 1 sigma standard deviation (25 <sup>th</sup> -75 <sup>th</sup> percentiles, number of points) ng m <sup>-3</sup>	% Time the site receives air mass
AM	1.14 +/- 0.11 (1.08-1.22, 432)	6%
AAC	1.15 +/- 0.12 (1.08-1.24, 1633)	24%
NAA	1.21 +/- 0.08 (1.17-1.27, 449)	7%
NCA	1.20 +/- 0.10 (1.15-1.27, 975)	15%
EUR	1.18 +/- 0.13 (1.08-1.26, 1161)	17%
AFR	1.23 +/- 0.16 (1.14-1.29, 1448)	22%
EUR/AFR	1.22 +/- 0.11 (1.16 -1.30, 586)	9%

3  
4 **Table 2. Statistics for the individual air mass classified data.**

5  
6 Paired t-tests were performed (Wilcoxon signed rank test, R) using the air mass datasets and the AFR  
7 dataset showed a significant difference (significance level <95%) in the mean concentration when  
8 compared to AAC, NAA, NCA, EUR, EUR/AFR and AM. This suggests that the air classified as AFR  
9 may be influenced by sources with different longer-term emissions trends to those experienced when  
10 other air masses are detected (t-test results can be found in Table 2 in the Supplementary Information).

11  
12 Biomass burning, of both anthropogenic and biogenic origins, is prevalent in Africa. In the Northern  
13 Hemisphere Africa burning occurs primarily in the Sahel, moving from the northern to the southern  
14 Sahel between November and February (Roberts et al., 2009). From Figure 5b it would appear that  
15 there are few trajectories which originate from this region, however an influence from biomass burning  
16 could be one explanation for the variable and sometimes higher, mercury concentrations within AFR  
17 air masses. Previous studies have found a relationship between TGM and carbon monoxide during  
18 such episodes (Slemr et al., 2006, Brunke et al, 2012). Figure 7 shows a correlation analysis using a  
19 matrix method between pairs of data at the CVO for AFR air, separated by season. O<sub>3</sub> and CO show a  
20 strong positive correlation particularly in spring, but also in summer and winter consistent with their  
21 shared pollution sources and of CO being a precursor of O<sub>3</sub> over long transport times. Higher wind  
22 speeds tend to be associated with air masses that have travelled further (from continental regions), and  
23 therefore have undergone greater photochemical production, which might explain the positive  
24 correlation of O<sub>3</sub> with wind speed. In spring a strong positive correlation of TGM with CO and with  
25 O<sub>3</sub> suggests a shared anthropogenic source when these concentrations are at their seasonal high in the  
26 northern hemisphere. The lack of correlation of O<sub>3</sub> with CO in autumn however suggests a more  
27 localised source for CO, perhaps from biomass burning, but this is not reflected in a positive

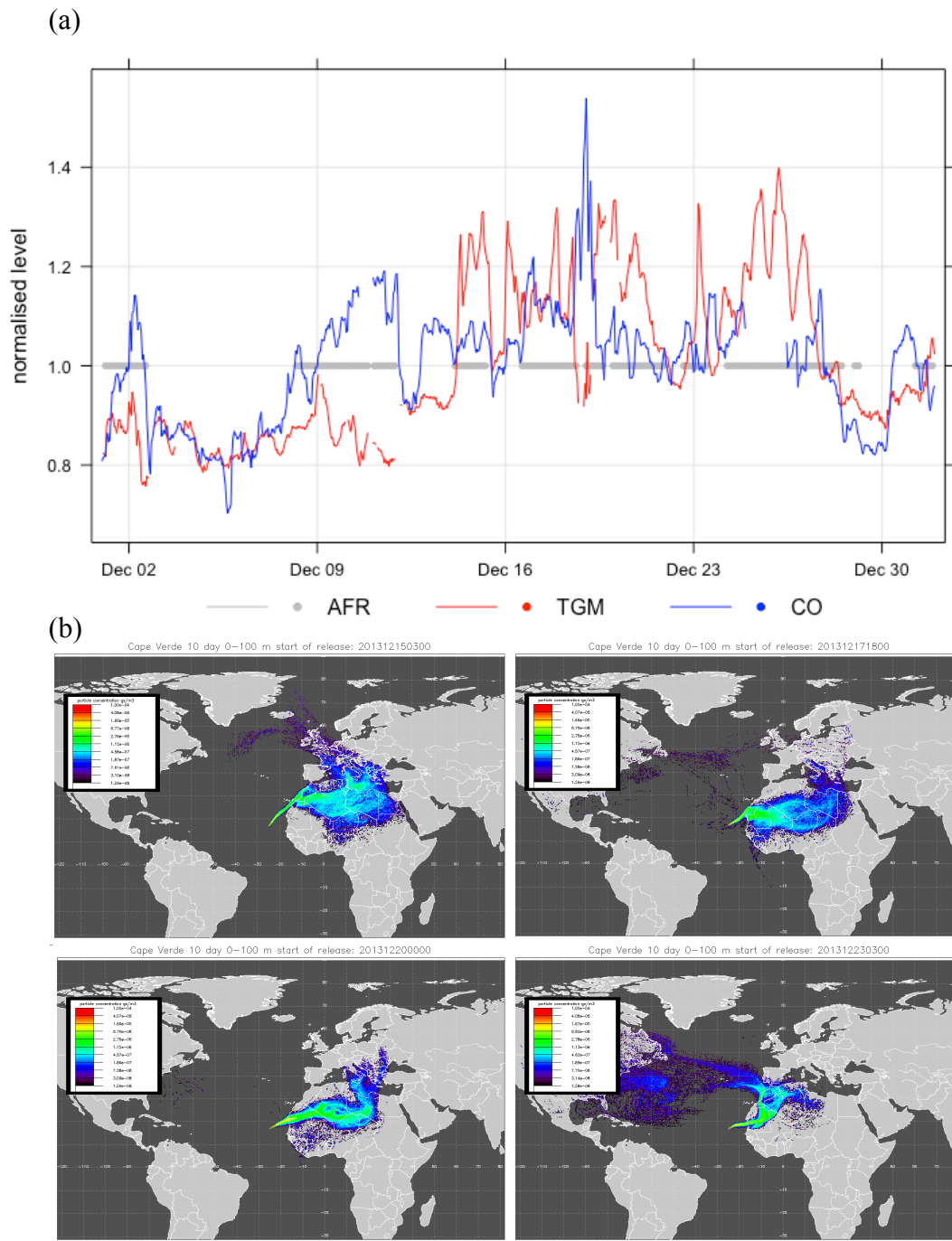
1 relationship with the TGM concentrations. Instead, TGM has a strong correlation with NO<sub>x</sub> (NO +  
 2 NO<sub>2</sub>), which suggests that the TGM concentrations are influenced by anthropogenic pollution sources  
 3 within closer proximity, for example from pollution emitted from cities on the coast of West Africa.  
 4 This may also explain the high variability in the TGM concentrations observed in the air classified as  
 5 AAC, which has also travelled over the African coast; although the concentrations in that air mass are  
 6 lower due to longer time spent over the ocean. Air classified as AFR and EUR/AFR shows the highest  
 7 levels of TGM and these may be better explained by an additional source further in-land. If sources of  
 8 mercury from small-scale artisanal gold mining (ASGM) from West Africa have any impact on these  
 9 measurements and trends then this may be reflected in a weaker correlation with CO in autumn and  
 10 winter. This is because we speculate that the activity might be more commonly carried out in the dry  
 11 season (November to April) when crops can't be grown; however there is little evidence to support  
 12 this. There is no data for TGM in AFR in summer.  
 13



14  
 15 **Figure 7: A correlation matrix separated by season to show the correlation between pairs of data, using the**  
 16 **corPlot function in Openair (Carslaw et al., 2012). The ellipses are visual representations of a scatter plot.**  
 17 **The colour scale highlights the strength of the correlation (red being the strongest and blue the weakest),**  
 18 **and the number is the r<sup>2</sup> of the data. The data was daily averaged before correlating to remove any bias**  
 19 **from diurnal variability. The order the variables appear is due to their similarity with one another, through**  
 20 **hierarchical cluster analysis**  
 21

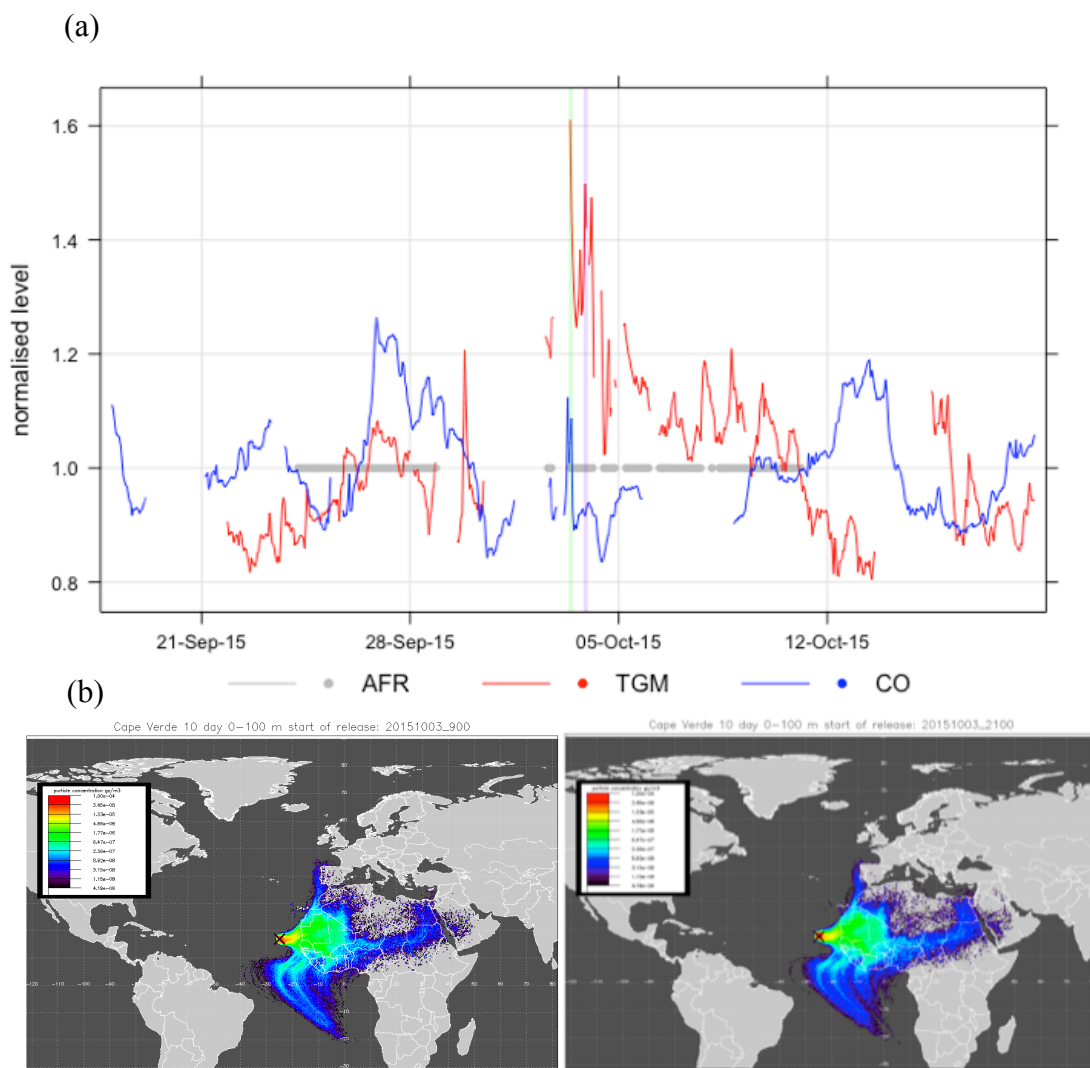
22 We next consider episodes when TGM concentrations were enhanced, to investigate the potential  
 23 influence of biomass burning on the measurements. On two occasions TGM exceeded 1.7 ng m<sup>-3</sup> (~0.5  
 24 ng m<sup>-3</sup> higher than the mean levels detailed in Table 1). The first period was during December 2013;  
 25 Figure 8a shows normalised CO and TGM concentrations during this month. The shaded periods  
 26 correspond to AFR trajectories, shown in detail in Figure 8b. The figure shows that for this period of  
 27 relatively elevated concentrations in AFR classified air, there is no significant correlation with CO, nor  
 28 do the trajectories originate over the biomass-burning region of the Sahel. Thus, biomass burning  
 29 appears not to be a contributor to the high TGM concentrations during this period.  
 30  
 31  
 32  
 33  
 34

1



2  
3  
4 **Figure 8: (a) Time-series of TGM and carbon monoxide. The plot is normalised by dividing by the**  
5 **compounds' mean value and the grey dots indicate when the air was characterized as AFR. The shaded**  
6 **periods correspond to the 10-day back trajectories in (b). Clockwise from top left: 15<sup>th</sup> December 2013**  
7 **03:00, 17<sup>th</sup> December 2013 18:00, 23<sup>rd</sup> December 2013 03:00, and 20<sup>th</sup> December 2013 00:00.**

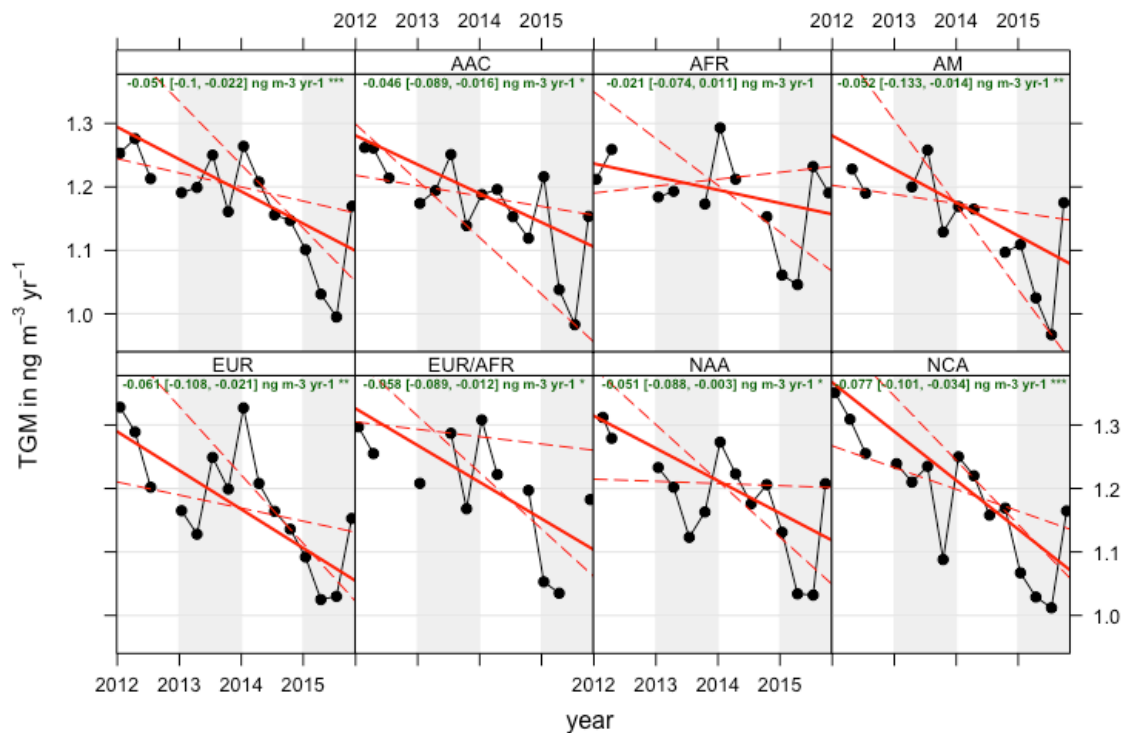
8  
9 A similar analysis was performed for the period 19<sup>th</sup> September 2015 until the 19<sup>th</sup> October 2015 and  
10 the corresponding plots and trajectories are shown in Figure 9. In this case the period of elevated  
11 concentrations is shorter with the episode lasting around a week. From the trajectories the air may  
12 have been influenced by air from the biomass region in Sahel Africa, which is at its most northern  
13 location in the month of October (Roberts et al., 2009). However, CO was not elevated during the  
14 period of peak [TGM] suggesting that biomass burning was not the source.  
15  
16



1  
2 **Figure 9: a) Time-series of TGM and CO. The plot is normalised by dividing by the compounds mean value**  
3 **and the grey dots indicate when the air was characterized as AFR. The shaded periods correspond to the**  
4 **10-day back trajectories in b), left, 3<sup>rd</sup> October 2015 9:00 and right, 3<sup>rd</sup> October 2015 21:00.**  
5

6 It has been previously established that West Africa is an important source region for ASGM activity  
7 (Telmer and Velga, 2009) but it is difficult to determine whether West Africa is a growing source of  
8 emissions since data has been limited and is subject to large uncertainty. It is likely however that the  
9 ASGM emissions are less regulated than anthropogenic emissions from coal combustion, ferrous/non-  
10 ferrous metal and cement production from Europe and the US (UNEP, 2013), and so is less likely to be  
11 decreasing in source strength.  
12

13 A further analysis using the Theil-Sen function was performed to evaluate trends in the 4-year dataset  
14 within individual air masses, based on seasonal TGM medians (Figure 10).  
15  
16



**Figure 10: 4-year TGM trends using a Theil-Sen function based on seasonal TGM medians separated by air mass. The data for all air masses is shown in the top left panel with the rest of the panels showing the data separated by the seven air mass classifications. The green text shows the slope estimate and the 95% confidence intervals are in brackets. In each case the solid red line shows the trend estimate and the dashed red lines show the 95% confidence intervals for the trend based on resampling methods.**

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30

The overall data (top left panel) show a decrease of  $-0.051 \text{ ng m}^{-3} \text{ yr}^{-1}$  (95% confidence interval of  $-0.1$  to  $-0.022 \text{ ng m}^{-3} \text{ yr}^{-1}$ ), as shown in the green text. Over the 4-year period the concentration changes were (in  $\text{ng m}^{-3} \text{ yr}^{-1}$ ):  $-0.046 \pm 0.037$ ,  $-0.021 \pm 0.043$ ,  $-0.052 \pm 0.060$ ,  $-0.061 \pm 0.044$ ,  $-0.058 \pm 0.039$ ,  $-0.051 \pm 0.043$ ,  $-0.077 \pm 0.034$ , for AAC, AFR, AM, EUR, EUR/AFR, NAA and NCA respectively (See Table 3 in the Supplementary information for the number of points used to derive monthly medians). The symbol “\*\*\*” in green indicates that the decrease is significant to the 0.001 level, “\*\*” to 0.01 and “\*” 0.05. Decreasing trends were observed in all the air masses with the largest trends observed in NCA and EUR suggesting controls on anthropogenic emissions are having an effect. Although there is some overlap in the 95% confidence intervals, the African (AFR) air clearly shows the smallest decreasing trend and the only positive upper confidence interval over the 4 years compared to all of the other air masses.

From an analysis of trajectories it was found that the AFR air was within the Sahel region outlined earlier for only around 1-2% of the time and mostly in October when the burning is at its most northern point (Figure 5b). In order to evaluate whether these air masses biased the decrease in AFR air, a sensitivity analysis was performed on the ThielSen analysis with the October data removed. The data flagged as southerly was also removed. The updated concentration change was  $-0.022 \pm 0.036 \text{ ng m}^{-3} \text{ yr}^{-1}$ , thus the filtering had essentially no effect on the AFR trend.

Weigelt et al., (2015) observed an annual decrease in TGM concentrations at Mace Head ( $53^{\circ}20'N$ ,  $9^{\circ}54'W$ , 10 m asl) of between  $-0.021$  and  $-0.023 \text{ ng m}^{-3} \text{ yr}^{-1}$  between 1996-2013, close to the upper confidence levels shown here (Weigelt et al., 2015). Within sub-tropical maritime classified air masses (air from south of  $28^{\circ}N$ , west of  $10^{\circ}W$ ) the decrease was  $-0.016 \pm 0.002 \text{ ng m}^{-3} \text{ yr}^{-1}$ . Further south at Cape Point, the trend for 17 years (1996-2013) was reported as  $-0.018 \text{ ng m}^{-3} \text{ yr}^{-1}$  (95% significance interval of  $-0.035$  to  $-0.013 \text{ ng m}^{-3} \text{ yr}^{-1}$ ). These concentration decreases are lower than the global calculation using the GEOS-CHEM model for the Northern Atlantic of  $-0.04 \text{ ng m}^{-3} \text{ yr}^{-1}$  (Soerensen et al., 2012), but may not take into account the very recent decreases in emissions (Weigelt et al., 2015; Soerensen et al., 2012). The overall concentration decrease (2011-2015) observed here of  $-0.05 \text{ ng m}^{-3} \text{ yr}^{-1}$  is more similar to the GEOS-CHEM study.

## 4 Conclusions

We report a four-year decreasing trend in total gaseous mercury (TGM) concentrations over the subtropical north Atlantic of  $-0.05 \pm 0.04 \text{ ng m}^{-3} \text{ yr}^{-1}$ , a rate of decline that is in agreement with a GEOS-CHEM analysis for the Northern Atlantic (Soerensen et al., 2012). A downward trend in concentration was observed in 6 out of 7 different air mass types, all associated broadly with long-range transport of air from the US and Europe over the north Atlantic ocean to the measurement location. The smallest and least significant downward trend ( $-0.02 \pm 0.03 \text{ ng m}^{-3} \text{ yr}^{-1}$ ) was observed in air that was influenced by West Africa, where emissions are less understood and may well be static or possibly still increasing. The UNEP Global Mercury Assessment report in 2013 suggested that more work is needed to improve emissions estimates for West African sources including field measurements around artisanal and small-scale gold mining (ASGM) sites.

## Data availability

The data presented here is freely available at the Centre for Environmental Data Analysis (CEDA) at <http://catalogue.ceda.ac.uk/uuid/0ae5eb7ce3ad4885a7223dd7b69f4db6>. Other levels of data are available within the GMOS central database upon request at [http://sdi.iaa.cnr.it/geoint/publicpage/GMOS/gmos\\_historical.zul](http://sdi.iaa.cnr.it/geoint/publicpage/GMOS/gmos_historical.zul) (GMOS Database, 2014).

## Author contribution

K. A. Read, L.J Carpenter, A. C. Lewis, J. Kentisbeer contributed to the preparation of the manuscript. K. A. Read and L.M. Neves made the measurements. Z. Fleming ran the NAME trajectories and assigned the air mass classifications.

## Acknowledgements

The authors acknowledge the Natural Environmental Research Council (NERC) and the Atmospheric Measurement Facility (AMF), National Centre for Atmospheric Science (NCAS) for their continued funding of the Cape Verde Observatory. The measurements of TGM were initiated due to financial support from the EU FP7-ENV-2010 project “Global Mercury observation System” (GMOS, Grant Agreement no 265113).

## References

- Ariya, P. A., Sun, J., Eltouny, N. A., Hudson, E. D., Hayes, C. T., and Kos, G.: Physical and chemical characterization of bioaerosols - Implications for nucleation processes, *International Reviews in Physical Chemistry*, 28, 1-32, 10.1080/01442350802597438, 2009.
- Bergan, T., and Rodhe, H.: Oxidation of elemental mercury in the atmosphere; Constraints imposed by global scale modelling, *Journal of Atmospheric Chemistry*, 40, 191-212, 10.1023/a:1011929927896, 2001.
- Bloom, N., and Fitzgerald, W. F.: Determination of Volatile Mercury species at the Picogram level by Low-Temperature Gas Chromatography with Cold-vapour Atomic Fluorescence Detection, *Analytica Chimica Acta*, 208, 151-161, 10.1016/s0003-2670(00)80743-6, 1988.
- Brown, R. J. C., Pirrone, N., van Hoek, C., Horvat, M., Kotnik, J., Wangberg, I., Corns, W. T., Bieber, E., and Sprovieri, F.: Standardisation of a European measurement method for the determination of mercury in deposition: results of the field trial campaign and determination of a measurement uncertainty and working range, *Accreditation and Quality Assurance*, 15, 359-366, 10.1007/s00769-010-0636-2, 2010.
- Brunke, E.-G, Ebinghaus, R., Kock, H.H., Labuschagne, C., and Slemr, F.: Emissions of mercury in southern Africa derived from long-term observations at Cape Point, South Africa, *Atmospheric Chemistry and Physics*, 12, 7465-7474, 10.5194/acp-12-7465-2012, 2012.
- Calvert, J. G., and Lindberg, S. E.: Mechanisms of mercury removal by O-3 and OH in the atmosphere, *Atmospheric Environment*, 39, 3355-3367, 10.1016/j.atmosenv.2005.01.055, 2005.

1  
2 Carpenter, L. J., Fleming, Z. L., Read, K. A., Lee, J. D., Moller, S. J., Hopkins, J. R., Purvis, R. M.,  
3 Lewis, A. C., Muller, K., Heinold, B., Herrmann, H., Fomba, K. W., van Pinxteren, D., Muller, C.,  
4 Tegen, I., Wiedensohler, A., Muller, T., Niedermeier, N., Achterberg, E. P., Patey, M. D., Kozlova, E.  
5 A., Heimann, M., Heard, D. E., Plane, J. M. C., Mahajan, A., Oetjen, H., Ingham, T., Stone, D.,  
6 Whalley, L. K., Evans, M. J., Pilling, M. J., Leigh, R. J., Monks, P. S., Karunaharan, A., Vaughan, S.,  
7 Arnold, S. R., Tschritter, J., Pohler, D., Friess, U., Holla, R., Mendes, L. M., Lopez, H., Faria, B.,  
8 Manning, A. J., and Wallace, D. W. R.: Seasonal characteristics of tropical marine boundary layer air  
9 measured at the Cape Verde Atmospheric Observatory, *Journal of Atmospheric Chemistry*, 67, 87-140,  
10 10.1007/s10874-011-9206-1, 2010.  
11  
12 Carslaw, D.C. and K. Ropkins, (2012). openair — an R package for air quality data  
13 analysis. *Environmental Modelling & Software*. Volume 27-28, 52-61.  
14  
15 Cinnirella, S., D'Amore, F., Bencardino, M., Sprovieri, F., and Pirrone, N.: The GMOS cyber(e)-  
16 infrastructure: advanced services for supporting science and policy, *Environmental Science and*  
17 *Pollution Research*, 21, 4193-4208, 10.1007/s11356-013-2308-3, 2014.  
18  
19 Cole, A. S., Steffen, A., Eckley, C. S., Narayan, J., Pilote, M., Tordon, R., Graydon, J. A., St Louis, V.  
20 L., Xu, X. H., and Branfireun, B. A.: A Survey of Mercury in Air and Precipitation across Canada:  
21 Patterns and Trends, *Atmosphere*, 5, 635-668, 10.3390/atmos5030635, 2014.  
22  
23 D'Amore, F., Bencardino, M., Cinnirella, S., Sprovieri, F., and Pirrone, N.: Data quality through a web-  
24 based QA/QC system: implementation for atmospheric mercury data from the global mercury  
25 observation system, *Environmental Science-Processes & Impacts*, 17, 1482-1491,  
26 10.1039/c5em00205b, 2015.  
27  
28 Dastoor, A. P., and Larocque, Y.: Global circulation of atmospheric mercury: a modelling study,  
29 *Atmospheric Environment*, 38, 147-161, 10.1016/j.atmosenv.2003.08.037, 2004.  
30  
31  
32 De Simone, F., S. Cinnirella, S. Gencarelli, C. N., Yang, X., Hedgecock, I. M., and Pirrone, N.: Model  
33 Study of Global Mercury Deposition from Biomass Burning, *Environmental Science and Technology*,  
34 49, 6712-6721, DOI 10.1021/acs.est.5b00969, 2015.  
35  
36 Duncan, B. N., Logan, J. A., Bey, I., Megretskaia, I. A., Yantosca, R. M., Novelli, P. C., Jones, N. B.,  
37 and Rinsland, C. P.: Global budget of CO, 1988-1997: Source estimates and validation with a global  
38 model, *Journal of Geophysical Research-Atmospheres*, 112, 10.1029/2007jd008459, 2007.  
39  
40 Gay, D. A., Schmeltz, D., Prestbo, E., Olson, M., Sharac, T., and Tordon, R.: The Atmospheric  
41 Mercury Network: measurement and initial examination of an ongoing atmospheric mercury record  
42 across North America, *Atmospheric Chemistry and Physics*, 13, 11339-11349, 10.5194/acp-13-11339-  
43 2013, 2013.  
44  
45 Goodsite, M. E., Plane, J. M. C., and Skov, H.: A theoretical study of the oxidation of Hg<sup>0</sup> to HgBr<sub>2</sub>  
46 in the troposphere, *Environmental Science & Technology*, 38, 1772-1776, 10.1021/es034680s, 2004.  
47  
48 Gustin, M. S., Weiss-Penzias, P. S., and Peterson, C.: Investigating sources of gaseous oxidized  
49 mercury in dry deposition at three sites across Florida, USA, *Atmospheric Chemistry and Physics*, 12,  
50 9201-9219, 10.5194/acp-12-9201-2012, 2012.  
51  
52 Hartmann, D.L., A.M.G. Klein Tank, M. Rusticucci, L.V. Alexander, S. Brönnimann, Y. Charabi, F.J.  
53 Dentener, E.J. Dlugokencky, D.R. Easterling, A. Kaplan, B.J. Soden, P.W. Thorne, M. Wild and P.M.  
54 Zhai, 2013: Observations: Atmosphere and Surface. In: *Climate Change 2013: The Physical Science*  
55 *Basis. Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel*  
56 *on Climate Change [Stocker, T.F., D. Qin, G.-K. Plattner, M. Tignor, S.K. Allen, J. Boschung, A.*  
57 *Nauels, Y. Xia, V. Bex and P.M. Midgley (eds.)]. Cambridge University Press, Cambridge, United*  
58 *Kingdom and New York, NY, USA.*  
59

1 Holmes, C. D., Jacob, D. J., Corbitt, E. S., Mao, J., Yang, X., Talbot, R., and Slemr, F.: Global  
2 atmospheric model for mercury including oxidation by bromine atoms, *Atmospheric Chemistry and*  
3 *Physics*, 10, 12037-12057, 10.5194/acp-10-12037-2010, 2010.  
4  
5 Lan, X., Talbot, R., Castro, M., Perry, K., and Luke, W.: Seasonal and diurnal variations of  
6 atmospheric mercury across the US determined from AMNet monitoring data, *Atmospheric Chemistry*  
7 *and Physics*, 12, 10569-10582, 10.5194/acp-12-10569-2012, 2012.  
8  
9 Lin, C. J., Pongprueksa, P., Lindberg, S. E., Pehkonen, S. O., Byun, D., and Jang, C.: Scientific  
10 uncertainties in atmospheric mercury models I: Model science evaluation, *Atmospheric Environment*,  
11 40, 2911-2928, 10.1016/j.atmosenv.2006.01.009, 2006.  
12  
13 Mason, R. P., Choi, A. L., Fitzgerald, W. F., Hammerschmidt, C. R., Lamborg, C. H., Soerensen, A.  
14 L., and Sunderland, E. M.: Mercury biogeochemical cycling in the ocean and policy implications,  
15 *Environmental Research*, 119, 101-117, 10.1016/j.envres.2012.03.013, 2012.  
16  
17 Muller, D., Wip, D., Warneke, T., Holmes, C. D., Dastoor, A., and Notholt, J.: Sources of atmospheric  
18 mercury in the tropics: continuous observations at a coastal site in Suriname, *Atmospheric Chemistry*  
19 *and Physics*, 12, 7391-7397, 10.5194/acp-12-7391-2012, 2012.  
20  
21 Muntean, M., Janssens-Maenhout, G., Song, S. J., Selin, N. E., Olivier, J. G. J., Guizzardi, D., Maas,  
22 R., and Dentener, F.: Trend analysis from 1970 to 2008 and model evaluation of EDGARv4 global  
23 gridded anthropogenic mercury emissions, *Science of the Total Environment*, 494, 337-350,  
24 10.1016/j.scitotenv.2014.06.014, 2014.  
25  
26 Nair, U. S., Wu, Y. L., Walters, J., Jansen, J., and Edgerton, E. S.: Diurnal and seasonal variation of  
27 mercury species at coastal-suburban, urban, and rural sites in the southeastern United States,  
28 *Atmospheric Environment*, 47, 499-508, 10.1016/j.atmosenv.2011.09.056, 2012.  
29  
30 Pacyna, E. G., Pacyna, J. M., Sundseth, K., Munthe, J., Kindbom, K., Wilson, S., Steenhuisen, F., and  
31 Maxson, P.: Global emission of mercury to the atmosphere from anthropogenic sources in 2005 and  
32 projections to 2020, *Atmospheric Environment*, 44, 2487-2499, 10.1016/j.atmosenv.2009.06.009,  
33 2010.  
34  
35 Pirrone, N., Aas, W., Cinnirella, S., Ebinghaus, R., Hedgecock, I. M., Pacyna, J., Sprovieri, F., and  
36 Sunderland, E. M.: Toward the next generation of air quality monitoring: Mercury, *Atmospheric*  
37 *Environment*, 80, 599-611, 10.1016/j.atmosenv.2013.06.053, 2013.  
38  
39 Pongprueksa, P., Lin, C. J., Lindberg, S. E., Jang, C., Braverman, T., Bullock, O. R., Ho, T. C., and  
40 Chu, H. W.: Scientific uncertainties in atmospheric mercury models III: Boundary and initial  
41 conditions, model grid resolution, and Hg(II) reduction mechanism, *Atmospheric Environment*, 42,  
42 1828-1845, 10.1016/j.atmosenv.2007.11.020, 2008.  
43  
44 Qureshi, A., MacLeod, M., Sunderland, E. and Hungerbühler, K. (2011) Exchange of Elemental  
45 Mercury between the Oceans and the Atmosphere, in *Environmental Chemistry and Toxicology of*  
46 *Mercury* (eds G. Liu, Y. Cai and N. O'Driscoll), John Wiley & Sons, Inc., Hoboken, NJ, USA.  
47 doi: 10.1002/9781118146644.ch12  
48  
49 Read, K. A., Mahajan, A. S., Carpenter, L. J., Evans, M. J., Faria, B. V. E., Heard, D. E., Hopkins, J.  
50 R., Lee, J. D., Moller, S. J., Lewis, A. C., Mendes, L., McQuaid, J. B., Oetjen, H., Saiz-Lopez, A.,  
51 Pilling, M. J., and Plane, J. M. C.: Extensive halogen-mediated ozone destruction over the tropical  
52 Atlantic Ocean, *Nature*, 453, 1232-1235, 10.1038/nature07035, 2008.  
53  
54 Read, K. A., Lee, J. D., Lewis, A. C., Moller, S. J., Mendes, L., and Carpenter, L. J.: Intra-annual  
55 cycles of NMVOC in the tropical marine boundary layer and their use for interpreting seasonal  
56 variability in CO, *Journal of Geophysical Research-Atmospheres*, 114, 10.1029/2009jd011879, 2009.  
57  
58 Roberts, G., Wooster, M.J. and Lagoudakis, E.: Annual and diurnal african biomass burning temporal  
59 dynamics, *Biogeosciences*, 6, 849-866, 2009.  
60



1 Ryall, D. B., Derwent, R. G., Manning, A. J., Simmonds, P. G., and O'Doherty, S.: Estimating source  
2 regions of European emissions of trace gases from observations at Mace Head, Atmospheric  
3 Environment, 35, 2507-2523, 10.1016/s1352-2310(00)00433-7, 2001.  
4  
5 Sather, M. E., Mukerjee, S., Smith, L., Mathew, J., Jackson, C., Callison, R., Scrapper, L., Hathcoat,  
6 A., Adam, J., Keese, D., Ketcher, P., Brunette, R., Karlstrom, J., and Van der Jagt, G.: Gaseous  
7 oxidized mercury dry deposition measurements in the Four Corners area and Eastern Oklahoma,  
8 U.S.A, Atmospheric Pollution Research, 4, 168-180, 10.5094/apr.2013.017, 2013.  
9  
10 Schroeder, W. H., and Munthe, J.: Atmospheric mercury - An overview, Atmospheric Environment,  
11 32, 809-822, 10.1016/s1352-2310(97)00293-8, 1998.  
12  
13 Seigneur, C., Vijayaraghavan, K., and Lohman, K.: Atmospheric mercury chemistry: Sensitivity of  
14 global model simulations to chemical reactions, Journal of Geophysical Research-Atmospheres, 111,  
15 10.1029/2005jd006780, 2006.  
16  
17 Selin, N. E., Jacob, D. J., Park, R. J., Yantosca, R. M., Strode, S., Jaegle, L., and Jaffe, D.: Chemical  
18 cycling and deposition of atmospheric mercury: Global constraints from observations, Journal of  
19 Geophysical Research-Atmospheres, 112, 10.1029/2006jd007450, 2007.  
20  
21 Sen, Pranab Kumar: Estimates of the regression coefficient based on Kendall's tau, Journal of the  
22 American Statistical Association **63**: 1379-1389,doi:10.2307/2285891, JSTOR 2285891, MR 0258201,  
23 1968.  
24  
25 Slemr, F., Ebinghaus, R., Simmonds, P.G., Jennings, S.G.: European emissions of mercury derived  
26 from long-term observations at mace head, on the western Irish coast, Atmospheric Environment, 40,  
27 6966-6974, doi:10.1016/j.atmosenv.2006.06.013, 2006.  
28  
29 Slemr, F., Brunke, E.-G., Ebinghaus, R., and Kuss, J.: Worldwide trend of atmospheric mercury since  
30 1995, Atmos. Chem. Phys., 11, 4779-4787, doi:10.5194/acp-11-4779-2011, 2011.  
31  
32 Smith, C.N., Kesler, S.E., Blum, J.D., and Rytuba, J.J.: isotope geochemistry of mercury in source  
33 rocks, mineral deposits and spring deposits of the California Coast Ranges, USA, Earth and Planetary  
34 Science Letters, 269, 3-4, 399-407, doi:10.1016/j.epsl.2008.02.029, 2008.  
35  
36 Soerensen, A. L., Jacob, D. J., Streets, D. G., Witt, M. L. I., Ebinghaus, R., Mason, R. P., Andersson,  
37 M., and Sunderland, E. M.: Multi-decadal decline of mercury in the North Atlantic atmosphere  
38 explained by changing subsurface seawater concentrations, Geophysical Research Letters, 39,  
39 10.1029/2012gl053736, 2012.  
40  
41 Sprovieri, F., Pirrone, N., Ebinghaus, R., Kock, H., and Dommergue, A.: A review of worldwide  
42 atmospheric mercury measurements, Atmospheric Chemistry and Physics, 10, 8245-8265,  
43 10.5194/acp-10-8245-2010, 2010.  
44  
45 Sprovieri, F., Pirrone, N., Bencardino, M., D'Amore, F., Carbone, F., Cinnirella, S., Mannarino, V.,  
46 Landis, M., Ebinghaus, R., Weigelt, A., Brunke, E. G., Labuschagne, C., Martin, L., Munthe, J.,  
47 Wangberg, I., Artaxo, P., Morais, F., Barbosa, H. D. J., Brito, J., Cairns, W., Barbante, C., Dieguez, M.  
48 D., Garcia, P. E., Dommergue, A., Angot, H., Magand, O., Skov, H., Horvat, M., Kotnik, J., Read, K.  
49 A., Neves, L. M., Gawlik, B. M., Sena, F., Mashyanov, N., Obolkin, V., Wip, D., Bin Feng, X., Zhang,  
50 H., Fu, X. W., Ramachandran, R., Cossa, D., Knoery, J., Maruszczak, N., Nerentorp, M., and Norstrom,  
51 C.: Atmospheric mercury concentrations observed at ground-based monitoring sites globally  
52 distributed in the framework of the GMOS network, Atmospheric Chemistry and Physics, 16, 11915-  
53 11935, 10.5194/acp-16-11915-2016, 2016.  
54  
55 Steffen, A., Scherz, T., Olson, M., Gay, D., and Blanchard, P.: A comparison of data quality control  
56 protocols for atmospheric mercury speciation measurements, Journal of Environmental Monitoring, 14,  
57 752-765, 10.1039/c2em10735j, 2012.  
58

1 Steffen, A., Bottenheim, J., Cole, A., Ebinghaus, R., Lawson, G., and Leitch, W. R.: Atmospheric  
2 mercury speciation and mercury in snow over time at Alert, Canada, *Atmospheric Chemistry and*  
3 *Physics*, 14, 2219-2231, 10.5194/acp-14-2219-2014, 2014.  
4  
5 Streets, D. G., Devane, M. K., Lu, Z. F., Bond, T. C., Sunderland, E. M., and Jacob, D. J.: All-Time  
6 Releases of Mercury to the Atmosphere from Human Activities, *Environmental Science &*  
7 *Technology*, 45, 10485-10491, 10.1021/es202765m, 2011.  
8  
9 Strode, S. A., Jaegle, L., Selin, N. E., Jacob, D. J., Park, R. J., Yantosca, R. M., Mason, R. P., and  
10 Slemr, F.: Air-sea exchange in the global mercury cycle, *Global Biogeochemical Cycles*, 21,  
11 10.1029/2006gb002766, 2007.  
12  
13 Temme, C., Blanchard, P., Steffen, A., Banic, C., Beauchamp, S., Poissant, L., Tordon, R., and Wiens,  
14 B.: Trend, seasonal and multivariate analysis study of total gaseous mercury data from the Canadian  
15 atmospheric mercury measurement network (CAMNet), *Atmospheric Environment*, 41, 5423-5441,  
16 10.1016/j.atmosenv.2007.02.021, 2007.  
17  
18 Theil, H. : A rank-invariant method of linear and polynomial regression analysis. I, II, III, *Nederl.*  
19 *Akad. Wetensch., Proc.* **53**: 386–392, 521–525, 1397–1412, MR 0036489, 1950.  
20  
21 UNEP: UNEP: Technical background report for the Global Mercury Assessment 2013. Arctic  
22 Monitoring and Assessment programme, Oslo, Norway/UNEP Chemicals Branch, Geneva,  
23 Switzerland, 2013.  
24  
25 US EPA: Mercury Study Report to Congress, Fate and transport of mercury in the Environment, vol  
26 111, EPA-452/R-97-005, US environmental Protection Agency, US Government Printing Office,  
27 Washington, DC, 1997  
28  
29 Wang, X., Lin, C. J., and Feng, X.: Sensitivity analysis of an updated bidirectional air-surface  
30 exchange model for elemental mercury vapor, *Atmospheric Chemistry and Physics*, 14, 6273-6287,  
31 10.5194/acp-14-6273-2014, 2014.  
32  
33 Wang, X., Lin, C. J., Lu, Z. Y., Zhang, H., Zhang, Y. P., and Feng, X. B.: Enhanced accumulation and  
34 storage of mercury on subtropical evergreen forest floor: Implications on mercury budget in global  
35 forest ecosystems, *Journal of Geophysical Research-Biogeosciences*, 121, 2096-2109,  
36 10.1002/2016jg003446, 2016.  
37  
38 Weigelt, A., Ebinghaus, R., Manning, A. J., Derwent, R. G., Simmonds, P. G., Spain, T. G., Jennings,  
39 S. G., and Slemr, F.: Analysis and interpretation of 18 years of mercury observations since 1996 at  
40 Mace Head, Ireland, *Atmospheric Environment*, 100, 85-93, 10.1016/j.atmosenv.2014.10.050, 2015.  
41  
42 Weinzierl, B., Sauer, D., Esselborn, M., Petzold, A., Veira, A., Rose, M., Mund, S., Wirth, M.,  
43 Ansmann, A., Tesche, M., Gross, S., and Freudenthaler, V.: Microphysical and optical properties of  
44 dust and tropical biomass burning aerosol layers in the Cape Verde region-an overview of the airborne  
45 in situ and lidar measurements during SAMUM-2, *Tellus*, 63B, 589-618, doi: 10.1111/j.1600-  
46 0889.2011.00566, 2011.  
47  
48 Whalley, L. K., Furneaux, K. L., Goddard, A., Lee, J. D., Mahajan, A., Oetjen, H., Read, K. A.,  
49 Kaaden, N., Carpenter, L. J., Lewis, A. C., Plane, J. M. C., Saltzman, E. S., Wiedensohler, A., and  
50 Heard, D. E.: The chemistry of OH and HO<sub>2</sub> radicals in the boundary layer over the tropical Atlantic  
51 Ocean, *Atmospheric Chemistry and Physics*, 10, 1555-1576, 10.5194/acp-10-1555-2010, 2010.  
52  
53 Wright, G., Gustin, M. S., Weiss-Penzias, P., and Miller, M. B.: Investigation of mercury deposition  
54 and potential sources at six sites from the Pacific Coast to the Great Basin, USA, *Science of the Total*  
55 *Environment*, 470, 1099-1113, 10.1016/j.scitotenv.2013.10.071, 2014.  
56  
57 Zhang, L. M., Wright, L. P., and Blanchard, P.: A review of current knowledge concerning dry  
58 deposition of atmospheric mercury, *Atmospheric Environment*, 43, 5853-5864,  
59 10.1016/j.atmosenv.2009.08.019, 2009.  
60

1 Zhang, Y., Jaegle, L., van Donkelaar, A., Martin, R. V., Holmes, C. D., Amos, H. M., Wang, Q.,  
2 Talbot, R., Artz, R., Brooks, S., Luke, W., Holsen, T. M., Felton, D., Miller, E. K., Perry, K. D.,  
3 Schmeltz, D., Steffen, A., Tordon, R., Weiss-Penzias, P., and Zsolway, R.: Nested-grid simulation of  
4 mercury over North America, *Atmospheric Chemistry and Physics*, 12, 6095-6111, 10.5194/acp-12-  
5 6095-2012, 2012.

6  
7 Zhang, Y. X., Jacob, D. J., Horowitz, H. M., Chen, L., Amos, H. M., Krabbenhoft, D. P., Slemr, F., St  
8 Louis, V. L., and Sunderland, E. M.: Observed decrease in atmospheric mercury explained by global  
9 decline in anthropogenic emissions, *Proceedings of the National Academy of Sciences of the United*  
10 *States of America*, 113, 526-531, 10.1073/pnas.1516312113, 2016.