

1 **Global combustion sources of organic aerosols: Model comparison**
2 **with 84 AMS factor analysis data sets**

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12 **Abstract**

13 Emissions of organic compounds from biomass, biofuel and fossil fuel combustion
14 strongly influence the global atmospheric aerosol load. Some of the organics are
15 directly released as primary organic aerosol (POA). Most are emitted in the gas phase
16 and undergo chemical transformations (i.e., oxidation by hydroxyl radical) and form
17 secondary organic aerosol (SOA). In this work we use the global chemistry climate
18 model EMAC with a computationally efficient module for the description of organic
19 aerosol (OA) composition and evolution in the atmosphere (ORACLE). The
20 tropospheric burden of open biomass and anthropogenic (fossil and biofuel)
21 combustion particles is estimated to be 0.59 Tg and 0.63 Tg, respectively, accounting
22 for about 30% and 32% of the total tropospheric OA load. About 30% of the open
23 biomass burning and 10% of the anthropogenic combustion aerosols originate from
24 direct particle emissions while the rest is formed in the atmosphere. A comprehensive
25 dataset of aerosol mass spectrometer (AMS) measurements along with factor-analysis
26 results from 84 field campaigns across the Northern Hemisphere are used to evaluate
27 the model results. Both the AMS observations and the model results suggest that over
28 urban areas both POA (25-40%) and SOA (60-75%) contribute substantially to the
29 overall OA mass while further downwind and in rural areas the POA concentrations
30 decrease substantially and SOA dominates (80-85%). EMAC does a reasonable job in
31 reproducing POA and SOA levels during most of the year. However, it tends to
32 underpredict POA and SOA concentrations during winter indicating that the model
33 misses wintertime sources of OA (e.g., residential biofuel use) and SOA formation
34 pathways (e.g., multiphase oxidation).

35

36 **1. Introduction**

37

38 Organic aerosol (OA) is a major contributor to fine particulate matter mass with
39 potentially harmful effects on the environment and human health (Lelieveld et al.,
40 2013; Poschl, 2005), however, the sources are poorly understood (Kanakidou et al.,
41 2005; Goldstein and Galbally, 2007; Donahue et al., 2009; Tsigaridis et al., 2014). OA
42 comprises primary organic aerosol (POA), i.e., directly emitted in the particulate
43 phase, and secondary organic aerosol (SOA), formed within the atmosphere from the
44 oxidation of gas-phase precursors. POA constitutes the particulate OA fraction

45 emitted by anthropogenic combustion processes (i.e., fossil fuels, biofuels) and open
46 biomass burning (i.e., savanna and forest fires). Anthropogenic combustion emissions
47 of particulate organic carbon (OC) are estimated at 13.9 Tg C yr⁻¹ for the year 2005
48 (Clarke et al., 2007). OC emissions from open biomass burning range from 13.5 Tg C
49 yr⁻¹ to 21.4 Tg C yr⁻¹ during the decade since 2000 (Van der Werf et al., 2010). POA
50 emitted from combustion sources can evaporate rapidly during atmospheric dilution
51 depending on ambient concentrations (Robinson et al., 2010; Ranjan et al., 2012; May
52 et al., 2014). The phase partitioning of the emitted POA depends on the volatility
53 distribution of the emissions. This distribution includes low volatility (LVOC; $C^* <$
54 $0.32 \mu\text{g m}^{-3}$), semivolatile (SVOC; $0.32 \mu\text{g m}^{-3} < C^* < 320 \mu\text{g m}^{-3}$), and intermediate
55 volatility (IVOC; $3.2 \times 10^2 \mu\text{g m}^{-3} < C^* < 3.2 \times 10^6 \mu\text{g m}^{-3}$) organic compounds. The
56 corresponding emission factors can be measured using dilution samplers and are
57 estimated as a function of the saturation concentration of the emitted organic
58 compounds (Grieshop et al., 2009). Traditional emission inventories (e.g., Clarke et
59 al., 2007; Van Der Werf et al., 2010) account only for a small fraction of the emitted
60 IVOCs since they are based on filter samples collected at aerosol concentrations up to
61 $10^4 \mu\text{g m}^{-3}$ (Shrivastava et al., 2008; Robinson et al., 2010). The amount of IVOC
62 emissions missing in traditional inventories is estimated to be between 0.25 and 2.8
63 times POA emissions, depending on the type of the source (Shrivastava et al., 2008;
64 Robinson et al., 2010).

65 Organic emissions further downwind mix with background air, resulting in cooling
66 and dilution and altering the gas-particle partitioning. The organic compounds that
67 remain in the gas phase can undergo chemical transformations (i.e., oxidation by
68 hydroxyl radical), become less volatile and may be transferred into SOA (Donahue et
69 al., 2006). Therefore, in addition to direct emissions of POA, it is important to
70 understand the potential of combustion emissions to contribute to SOA formation.
71 Numerous studies have indicated that SOA usually exceeds POA even in urban
72 environments with substantial primary emissions (Jimenez et al., 2009; Stone et al.,
73 2009; Sun et al., 2011; Mohr et al., 2012; Hayes et al., 2013). However, the overall
74 contribution of combustion emissions to ambient SOA and OA remains uncertain
75 (Chirico et al., 2010; Miracolo et al., 2011; Samy and Zielinska, 2010; Gentner et al.,
76 2012; Bahreini et al., 2012; Gordon et al., 2014). Together with the OA mass
77 concentration, the hygroscopic, chemical and optical properties continue to change

78 because of chemical processing by gas-phase oxidants (Jimenez et al., 2009). These
79 changes affect the OA radiative forcing on climate by direct and indirect effects, the
80 latter through cloud formation (Poschl, 2005; McFiggans et al., 2006; IPCC; 2013).

81 Mass spectrometry has been widely used in aerosol analyses because of the
82 universal, sensitive and rapid detection of aerosol components (Suess and Prather,
83 1999). The Aerosol Mass Spectrometer (AMS) (Jimenez et al., 2003) has been the
84 most commonly used instrument in recent years. AMS is capable of quantitatively
85 measuring the OA mass concentrations with high time and particle size resolution
86 (Takegawa et al., 2005; Zhang et al., 2005b). Several factor analysis techniques have
87 been employed to extract information about processes and sources of OA. These
88 techniques include principal component analysis (CPCA; Zhang et al., 2005a),
89 multiple component analysis (MCA; Zhang et al., 2007), hierarchical cluster analysis
90 (Marcolli et al., 2006), the Multilinear Engine (ME-2; Lanz et al., 2008), and positive
91 matrix factorization (PMF; Paatero and Tapper, 1994; Paatero, 1997), with the latter
92 being the most commonly used (Lanz et al., 2007; Nemitz et al., 2008; Aiken et al.,
93 2009; Ulbrich et al., 2009; DeCarlo et al., 2010; Mohr et al., 2012; Hayes et al., 2013;
94 Crippa et al., 2014; Carbone et al., 2014; Chen et al., 2015).

95 PMF allows the classification of OA into several types based on different temporal
96 and mass spectral signatures. Two major components often resolved by the analysis of
97 the AMS measurements are hydrocarbon-like organic aerosol (HOA) and oxygenated
98 organic aerosol (OOA) (Zhang et al., 2007; Jimenez et al., 2009). Biomass burning
99 OA (BBOA), marine-related OA (MOA) and cooking OA (COA) are other OA
100 components that PMF may identify as important components of the observed OA
101 (Lanz et al., 2010; Mohr et al., 2012; Kostenidou et al., 2013; Crippa et al., 2013a).
102 HOA correlates with combustion tracers (e.g., CO, EC, and NO_x) and is considered as
103 a surrogate for fossil fuel combustion POA (Lanz et al., 2007; Ulbrich et al., 2009;
104 Crippa et al., 2014). BBOA correlates with tracers from biomass burning (e.g.,
105 acetonitrile, levoglucosan, and potassium) and is considered a surrogate of biomass
106 burning POA (Aiken et al., 2010; Lanz et al., 2010; Crippa et al., 2014). OOA often
107 correlates with secondary pollutants (e.g. ozone, sulfate, and nitrate) and is considered
108 a surrogate for SOA (Jimenez et al., 2009; Ng et al., 2011;). However, Crippa et al.
109 (2014) have reported that the OOA correlation with secondary inorganic species
110 might not be very high in at least some field campaigns. OOA can include SOA from

111 various precursors, such as anthropogenic and biogenic VOCs, as well as SVOCs and
112 IVOCs from fossil fuel, biofuel and open biomass burning. PMF often classifies OOA
113 into two subtypes that differ in the degree of oxidation: a more strongly oxygenated
114 low-volatility OOA (LV-OOA) often correlating with sulfate, and a less oxygenated
115 semivolatile OOA (SV-OOA) usually correlating with nitrate (Jimenez et al., 2009;
116 Crippa et al., 2014). Field campaigns in the Northern Hemisphere have shown that
117 HOA accounts for approximately one third of the OA in urban sites and OOA
118 accounts for the remaining two thirds, while OOA represents roughly 95% of the OA
119 in rural/remote regions (Zhang et al., 2007).

120 Global chemistry climate and chemical transport models systematically
121 underpredict OA levels, especially over and downwind of anthropogenic source
122 regions (Tsigaridis et al., 2014). At the same time, global models tend to predict a
123 dominance of POA at mid-latitudes in the Northern Hemisphere while measurements
124 indicate the opposite (Henze et al., 2008; Tsigaridis et al., 2014). The same models
125 indicate that the formation of SOA from biogenic sources greatly exceeds that from
126 anthropogenic sources. The shortcomings in many OA models are partially due to the
127 assumption that POA is non-volatile and nonreactive (Kanakidou et al., 2005; Jimenez
128 et al., 2009). To address these shortcomings, Donahue et al. (2006) developed the
129 volatility basis set (VBS) framework which assumes that POA emissions are
130 semivolatile and photochemically reactive and uses logarithmically spaced volatility
131 bins to distribute POA upon emission. Recently, several regional-scale modeling
132 studies have accounted for the semivolatile nature and chemical aging of organic
133 compounds demonstrating improvements in reproducing the OA concentrations and
134 chemical composition (Robinson et al., 2007; Shrivastava et al., 2008; Murphy and
135 Pandis, 2009; Tsimpidi et al., 2010; Tsimpidi et al., 2011; Hodzic et al., 2010;
136 Fountoukis et al., 2011; Bergstrom et al., 2012; Athanasopoulou et al., 2013; Zhang et
137 al., 2013; Fountoukis et al., 2014). However, only few global modeling studies have
138 yet adopted the VBS approach to simulate the SOA formation from the chemical
139 aging of SVOC and IVOC emissions (Jathar et al., 2011; Tsimpidi et al., 2014;
140 Shrivastava et al., 2015; Hodzic et al., 2015). According to these studies, the modeled
141 tropospheric burden of POA is 0.09-0.94 Tg and of SOA 1.8-2.8 Tg.

142 In this work we use ORACLE, a computationally efficient module for the
143 description of organic aerosol composition and evolution in the atmosphere (Tsimpidi

144 et al., 2014), to estimate the impact of open biomass burning and anthropogenic
145 combustion emissions and their chemical aging on global OA budgets and
146 distributions. An extensive global dataset of AMS measurements and factor-analysis
147 results from 84 field campaigns in the Northern Hemisphere are used in combination
148 with the model results during the period of 2001-2010. This integrated effort provides
149 further insights into the temporal and geographical variability of the OA particles,
150 emission strengths and the chemical processing of organics from combustion sources.

151

152 **2. Model description and application**

153

154 **2.1 EMAC Model**

155 The ECHAM/MESSy Atmospheric Chemistry (EMAC) model is a numerical
156 chemistry and climate simulation system that includes sub-models describing the
157 lower and middle atmosphere processes (Jöckel et al., 2006). EMAC includes
158 submodels that describe gas-phase chemistry (MECCA; Sander et al., 2011),
159 inorganic aerosol microphysics (GMXe; Pringle et al., 2010), cloud microphysics
160 (CLOUD; Jöckel et al., 2006), aerosol optical properties (AEROPT; Lauer et al.,
161 2007), dry deposition (DRYDEP; Kerkweg et al., 2006a), sedimentation (SEDI;
162 Kerkweg et al., 2006a), cloud scavenging (SCAV; Tost et al., 2006), emissions
163 (ONLEM and OFFLEM; Kerkweg et al., 2006b), and organic aerosol formation and
164 growth (ORACLE; Tsimpidi et al., 2014).

165 The removal of gas and aerosol organic compounds through dry deposition is
166 calculated with the DRYDEP submodel (Kerkweg et al., 2006a) based on the big-leaf
167 approach, and the dry deposition velocities depend on physical and chemical
168 properties of the surface cover (e.g., the roughness length, soil pH, leaf stomatal
169 exchange, etc.). The sedimentation of aerosols is calculated with the SEDI submodel
170 (Kerkweg et al., 2006a) using a first-order trapezoid scheme. In-cloud scavenging and
171 rainout of gas and aerosol species are treated by the SCAV submodel (Tost et al.,
172 2006). The effective Henry's law coefficient used for calculating the scavenging rates
173 of LVOCs, SVOCs, and IVOCs is 10^5 M atm^{-1} .

174 The spectral resolution used in this study is T63L31, corresponding to a horizontal
175 grid resolution of $1.875^\circ \times 1.875^\circ$ and 31 vertical layers extending to 10 hPa at about 25
176 km altitude. EMAC has been run for 11 years, covering the period 2000–2010, and

177 the first year is used as spin-up. EMAC has been extensively described and evaluated
178 against in situ observations and satellite measurements that include filter-based
179 particulate matter concentrations, aerosol optical depth, acid deposition, gas-phase
180 mixing ratios, and meteorological parameters (Jöckel et al., 2006; Pozzer et al.,
181 2012a; Pozzer et al., 2012b; Karydis et al., 2016). Tsimpidi et al. (2014) performed an
182 in depth evaluation of the EMAC calculated total OA over different continents by
183 using measurements from the EMEP network over Europe, the IMPROVE network
184 over North America, and several short-term field campaigns over East Asia,
185 subtropical West Africa, the Amazon rainforest, and the Canadian boreal forest. The
186 present work focuses on the model evaluation for the individual OA components (i.e.,
187 POA and SOA). In addition, the statistical evaluation of EMAC results for the
188 inorganic components summarized in Tables S1-S3 in the Supplementary Material.

189

190 **2.2 ORACLE Module**

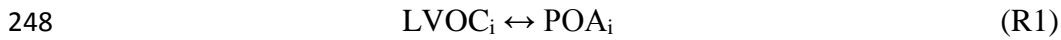
191 ORACLE is a computationally efficient submodel for the description of OA
192 composition and evolution in the atmosphere which has been implemented into the
193 EMAC model by Tsimpidi et al. (2014). ORACLE simulates a variety of semivolatile
194 organic species and reaction products and separates them into groups of compounds
195 with logarithmically spaced effective saturation concentrations.

196 In this study, primary organic emissions from open biomass burning and from
197 anthropogenic sources (i.e., fossil fuel and biofuel) are simulated using separate
198 surrogate species for each source category. They are subdivided into three groups of
199 organic compounds: low volatility, LVOCs ($10^{-2} \mu\text{g m}^{-3}$), semi-volatile, SVOCs (10^0
200 and $10^2 \mu\text{g m}^{-3}$) and intermediate volatility organic compounds, IVOCs (10^4 and $10^6 \mu\text{g}$
201 m^{-3}). These organic compounds are allowed to partition between the gas and aerosol
202 phases resulting in the formation of fPOA (anthropogenic POA from fossil fuel and
203 biofuel combustion) and bbPOA (natural POA from open biomass burning). VOCs are
204 distinguished into anthropogenic and biogenic and their oxidation products are
205 distributed in four volatility bins with effective saturation concentrations of 10^0 , 10^1 ,
206 10^2 , and $10^3 \mu\text{g m}^{-3}$ at 298 K by using the aerosol mass yields (Table S4) by Tsimpidi
207 et al., (2014). Gas-phase photochemical reactions that change the volatility of the
208 organics are taken into account and their oxidation products (SOA-sv, SOA-iv, and
209 SOA-v) are simulated separately in the module to keep track of their origin. The

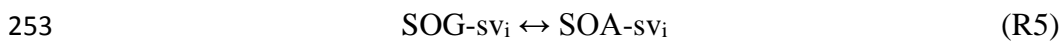
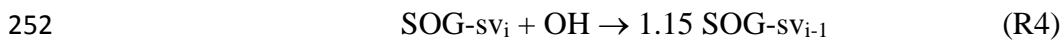
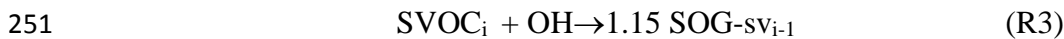
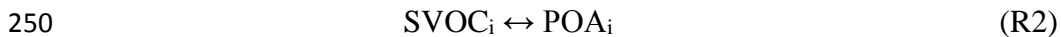
210 suffixes -sv, -iv and -v to the term SOA define category of precursors (SVOCs,
211 IVOCs, and VOCs, respectively). For the current application, SOA components are
212 divided into four groups based on their source: anthropogenic from fossil fuel and
213 biofuel combustion sources (fSOA), natural from open biomass burning (bbSOA),
214 SOA from anthropogenic (aSOA-v) and biogenic (bSOA-v) VOCs. This study focuses
215 on the OA produced from primary combustion sources and discusses in detail results
216 for the first two types of SOA (fSOA and bbSOA). The model set up for simulating the
217 formation of aSOA-v and bSOA-v and the corresponding results can be found in
218 Tsimpidi et al. (2014). In addition, in this work ORACLE has been modified to
219 distinguish the formation of fresh SOA and aged SOA by adding additional tracers to
220 the model. The first generation oxidation products of SVOCs, IVOCs, and VOCs are
221 characterized as fresh while SOA produced from any additional oxidation step is
222 grouped together and considered aged (Figure 1). LVOCs are not allowed to
223 participate in photochemical reactions since they are in the lowest volatility bin. This
224 assumption may introduce a small bias in our results only under extremely clean
225 conditions ($OA \leq 10^{-2} \mu\text{g m}^{-3}$) where part of LVOC is in the gas phase. Adding another
226 bin in the volatility distribution to accurately represent the extremely low volatility
227 organic compounds (e.g., ELVOCs with C^* lower than 10^{-3}) would be useful only for
228 studying new particle formation, which is outside the scope of the current work. The
229 volatilities of SVOCs and IVOCs are reduced by a factor of 10^2 as a result of the OH
230 reaction with a rate constant of $2 \times 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ and a 15% increase in mass
231 is assumed to account for two added oxygen atoms (Tsimpidi et al., 2014). This
232 formulation is comparable to a number of global and regional studies which assume
233 two orders of magnitude reduction in volatility and up to 50% increase in mass per
234 reaction (Grieshop et al. 2009; Hodzic et al., 2010; Pye and Seinfeld 2010, etc.).
235 Shrivastava et al. (2011) even used seven orders of magnitude reduction in volatility
236 per reaction. However, despite the fact that most of the studies assume that each
237 oxidation reaction of SVOC and IVOC reduces the volatility of the precursor by one
238 (e.g., Tsimpidi et al., 2010; Jathar et al., 2011; Bergstrom et al., 2012) or two orders of
239 magnitude, the oxidation products can be up to four orders of magnitude lower in
240 volatility than the precursor (Kroll and Seinfeld, 2008). Furthermore, ORACLE
241 calculates the fraction of the semivolatile organic compounds that condenses to (or
242 evaporates from) the particle phase by assuming bulk equilibrium and that all organic

243 compounds form a pseudo-ideal solution (Tsimpidi et al., 2014). Overall, the primary
244 aerosol formation from the phase partitioning of the freshly emitted LVOCs and
245 SVOCs, as well as the formation of SOA from the photo-oxidation of SVOCs and
246 IVOCs are described by the following reactions:

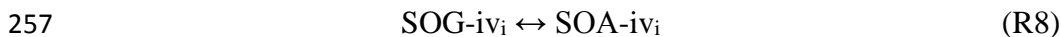
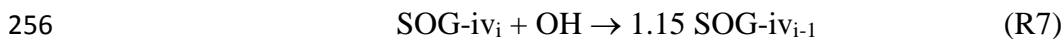
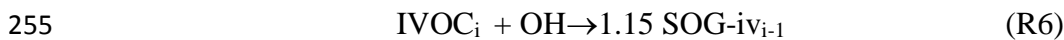
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259 where i is the original volatility bin and $i-1$ is the volatility bin with saturation
260 concentration reduced by a factor of 10^2 . The term SOG corresponds to secondary
261 organic gas that is produced by at least one chemical reaction in the atmosphere. The
262 symbol “ \leftrightarrow ” denotes the equilibrium between the gas and the aerosol phases. It is
263 worth mentioning that the production of RO_2 as an intermediate after the oxidation of
264 SVOC and IVOC has been omitted since it would be essential only in cases where
265 these reactions are a potentially significant sink of OH (i.e. in concentrated smoke
266 plumes) (Alvarado et al., 2015). The model set-up and the different aerosol types and
267 chemical processes that are simulated by ORACLE for this study are illustrated in
268 Figure 1. More details about ORACLE can be found in Tsimpidi et al. (2014).

269

270 **2.3 Emission inventory**

271 The CMIP5 RCP4.5 emission inventory (Clarke et al., 2007) is used for the
272 anthropogenic POA emissions from fossil fuel and biofuel combustion sources. The
273 open biomass burning emissions from savanna burning and forest fires are based on
274 the Global Fire Emissions Database (GFED v3.1; van der Werf et al., 2010). These
275 emission datasets report the mass of the OC emitted. Therefore, in order to determine

276 the total organic matter (OM) emitted (including any additional species associated
277 with the carbon) OM/OC values of 1.3 for anthropogenic POA and 1.6 for open
278 biomass burning POA are used. These values are based on the OM:OC ratios
279 estimated by Canagaratna et al. (2015) for HOA and BBOA, respectively.
280 Furthermore, the above emission datasets are monthly resolved and treat POA as non-
281 reactive and non-volatile. However, only a fraction of this organic material is directly
282 emitted in the aerosol phase as POA. Most of it is rapidly transferred to the gas phase
283 where it can undergo chemical transformations and form SOA. Therefore, key input
284 for the accurate description of these compounds and their chemical aging is the
285 volatility distribution at 298 K. Figure 2 depicts the volatility distributions assumed
286 for this study which cover a range of 10^{-2} to $10^4 \mu\text{g m}^{-3}$ for open biomass burning
287 (May et al., 2013) and 10^{-2} to $10^6 \mu\text{g m}^{-3}$ for fossil and biofuel combustion emissions
288 (Robinson et al., 2007). Emission inventories are based on samples collected at
289 aerosol concentrations up to $10^4 \mu\text{g m}^{-3}$ (Shrivastava et al., 2008; Robinson et al.,
290 2010). As a result IVOC emissions with $C^* > 10^4 \mu\text{g m}^{-3}$ are missing from the
291 traditional emission inventories and have to be accounted for by assigning additional
292 emissions in this volatility range. We assume that the missing IVOC emissions from
293 anthropogenic combustion are 1.5 times the traditional OA emissions included in the
294 inventory (Shrivastava et al., 2008; Tsimpidi et al., 2010), therefore the sum of the
295 emission factors is 2.5. No additional IVOC emissions are assumed in the $C^* > 10^4 \mu\text{g}$
296 m^{-3} bins for open biomass burning and therefore the sum for the biomass burning
297 emission factors is unity. As a result, 40% of the biomass burning OA emissions
298 represents IVOCs with $C^* = 10^4 \mu\text{g m}^{-3}$ (Table 2). The sensitivity of our results to
299 these assumptions will be discussed in a subsequent article in preparation. Overall, the
300 decadal average global emission flux of SVOCs and IVOCs is 44 Tg yr^{-1} from
301 anthropogenic combustion sources and 28 Tg yr^{-1} from open biomass burning sources.

302

303 **3. Model evaluation methodology**

304

305 **3.1 Factor analysis of AMS measurements**

306

307 During the period 2001-2010, 84 field campaigns were performed in the Northern
308 Hemisphere using the AMS for measuring ambient OA concentrations in urban,

309 urban-downwind, and rural environments (Figure 3). Information for each of these
310 campaigns is given in Tables S5-S7. The OA source apportionment for all sites was
311 taken from the literature (Tables S5-S7) and performed using factor-analysis
312 techniques classifying OA as HOA, corresponding to POA from fossil fuel
313 combustion, and OOA, corresponding to SOA. Therefore, AMS HOA is compared
314 with modeled fPOA, which is emitted and remains in the aerosol phase without
315 undergoing chemical reactions, and AMS-OOA is compared with modeled SOA (the
316 sum of SOA-sv, SOA-iv, and SOA-v), formed from the oxidation of gas phase
317 precursors (SVOCs, IVOCs, and VOCs). At many locations, PMF and other factor
318 analysis techniques identified two subtypes of OOA that differ in volatility and
319 oxidation state: semi-volatile OOA (SV-OOA) and low-volatility OOA (LV-OOA).
320 There are different potential interpretations of SV-OOA and LV-OOA. SV-OOA
321 often correlates with semi-volatile species such as ammonium nitrate and is less
322 oxygenated, consistent with relatively fresh SOA (Zhang et al., 2011; Ng et al., 2011).
323 LV-OOA usually correlates with nonvolatile secondary species such as sulfate, and is
324 highly oxygenated, consistent with regional aged OA (Zhang et al., 2011; Ng et al.,
325 2011). Recently, Ehn et al. (2014) found a direct pathway which leads to the
326 formation of fresh LV-OOA from the oxidation of several biogenic VOCs. Here we
327 test the hypothesis that SV-OOA corresponds to the first generation products and LV-
328 OOA to the later generation ones. Therefore, AMS SV-OOA is compared with the
329 fraction of SOA-sv, SOA-iv, and SOA-v from the first oxidation step of SVOCs,
330 IVOC, and VOCs as it is tracked separately (henceforth EMAC fresh SOA). Then
331 AMS LV-OOA is compared with the fraction of SOA-sv, SOA-iv, and SOA-v from
332 any additional oxidation step (henceforth EMAC aged SOA). Finally, in a few field
333 campaigns, e.g., in the Alps (Lanz et al., 2010), residential wood burning was found
334 to be a major source of OA. However, residential wood burning is included in EMAC
335 as fPOA. To account for this inconsistency for the AMS data sets that include BBOA
336 we compare the sum of the simulated fPOA and bbPOA (henceforth EMAC POA) to
337 the sum of the AMS HOA and BBOA (henceforth AMS POA). In data sets where OA
338 from cooking activities has been resolved by the PMF analysis, AMS COA has not
339 been taken into account for the model evaluation since these emissions are not
340 included in our emission inventory.

341

342 3.2 Evaluation metrics

343

344 The mean bias (MB), mean absolute gross error (MAGE), normalized mean bias
345 (NMB), normalized mean error (NME), and the root mean square error (RMSE) are
346 used to assess the model performance:

$$347 \quad MAGE = \frac{1}{N} \sum_{i=1}^N |P_i - O_i| \quad (1) \quad MB = \frac{1}{N} \sum_{i=1}^N (P_i - O_i) \quad (2)$$

$$348 \quad NME = \frac{\sum_{i=1}^N |P_i - O_i|}{\sum_{i=1}^N O_i} \quad (3) \quad NMB = \frac{\sum_{i=1}^N (P_i - O_i)}{\sum_{i=1}^N O_i} \quad (4)$$

$$349 \quad RMSE = \left[\frac{1}{N} \sum_{i=1}^N (P_i - O_i)^2 \right]^{\frac{1}{2}} \quad (5)$$

350 where O_i is the observed campaign average value of the i th OA component, P_i is the
351 corresponding modelled value during the same period, and N is the total number of
352 comparisons used for the evaluation. NME (in %) and MAGE (in $\mu\text{g m}^{-3}$) provide an
353 estimate of the overall discrepancy between predictions and observations, while NMB
354 (in %) and MB (in $\mu\text{g m}^{-3}$) are sensitive to systematic errors. RMSE (in $\mu\text{g m}^{-3}$) is the
355 root of the mean square error, which incorporates both the variance of the prediction
356 and its bias. Both NME and MAGE inherently include the corresponding bias, which
357 is the reason why their magnitude is equal or larger than NMB and MB respectively.
358 For an unbiased prediction, NME and MAGE express the variance. When NME and
359 NMB or MAGE and MB are close, the discrepancy is explained as a systematic bias
360 rather than scatter. When NME/MAGE exceeds NMB/MB, part of the discrepancy
361 between predictions and observations is explained as scatter. To determine the effects
362 of the site type, geographical location and the seasonal cycle on the model results, the
363 evaluation metrics are calculated separately for urban, urban-downwind, and rural
364 sites; European, North American, and Asian sites; and for four seasons (winter,
365 spring, summer, and autumn).

366

367 4. Model results

368

369 **4.1 OA from anthropogenic combustion**

370

371 **4.1.1 Geographical distribution**

372 Figure 4 depicts the simulated, decadal average global surface concentrations of
373 fPOA and fSOA from anthropogenic SVOC and IVOC sources (fossil and biofuel
374 combustion). The average surface concentration of fPOA is $0.1 \mu\text{g m}^{-3}$. Higher fPOA
375 concentrations (up to $14 \mu\text{g m}^{-3}$) are simulated over densely populated and highly
376 industrialized areas (e.g., Eastern China, Northern India, Central Europe, etc.) where
377 there are substantial anthropogenic combustion emissions. Downwind of the sources
378 fPOA concentrations decrease substantially since they are diluted and a large fraction
379 is predicted to evaporate during transport. This results in a highly inhomogeneous
380 spatial distribution of fPOA concentrations (Figure 4a). In contrast, fSOA is more
381 regionally distributed with high concentrations (up to $9.5 \mu\text{g m}^{-3}$) downwind of the
382 anthropogenic sources due to its continuous production and long-range transport from
383 SVOCs and IVOCs (Figure 4b). This results in a continental fSOA background of
384 $0.5\text{-}1 \mu\text{g m}^{-3}$ and in concentrations of around $1 \mu\text{g m}^{-3}$ over marine regions close to
385 anthropogenic sources (e.g., Arabian Sea, Yellow Sea). The average surface
386 concentration of fSOA is $0.26 \mu\text{g m}^{-3}$ with 73% of it originating from the oxidation of
387 IVOC emissions. This result supports the hypothesis of several recent studies that
388 IVOC emission and oxidation may be a significant SOA source that has been missing
389 from chemistry climate models (Jathar et al., 2011; Tsimpidi et al., 2014). The
390 relatively small contribution (25%) of SVOCs to total fSOA follows from its low
391 emissions compared to the IVOCs (two times lower) and by the fact that a significant
392 fraction of SVOCs stays in the aerosol phase (as POA) without undergoing chemical
393 reactions (Tsimpidi et al., 2014).

394 The fraction of fossil and biofuel combustion OA (fOA) that is formed through the
395 oxidation of gas phase species, fSOA/fOA, is consistently high with values ranging
396 from around 20% close to the sources to 100% away in remote regions with a global
397 average of 83% at the surface (Figure 5a). This suggests that over urban areas both
398 fPOA and fSOA contribute significantly to the overall OA mass while further
399 downwind and in rural areas SOA formation dominates since POA decreases
400 substantially due to dilution and evaporation. The OA due to anthropogenic
401 combustion sources contributes significantly to total OA over the continents in the

402 Northern Hemisphere (Figure 5b). The highest contribution is predicted over Eastern
403 China (83%) and the lowest over the Southeastern US (23%). Over mid-latitude
404 oceans, the contribution of fOA to total OA is also high (around 60%) due to the long-
405 range transport of SOA. On the other hand, fOA/OA is very low (0-10%) over the
406 tropical and boreal forest regions in contrast to the significant bbOA and bSOA-v
407 concentrations over these areas. The eastern part of the Eurasian boreal forest is an
408 exception since the lower emissions of bbOA together with the considerable amount
409 of fSOA transported from Europe results in fOA/OA fractions of about 40%. Overall,
410 the predicted global average fOA/OA is 38%. This result highlights the importance of
411 anthropogenic emissions for global OA levels, also suggested by other recent studies
412 (Carslaw et al., 2013; Lee et al., 2013; Spracklen et al., 2011).

413

414 **4.1.2 Temporal profile**

415 Table 1 shows the decadal average tropospheric burden of fPOA and fSOA. The
416 decadal average tropospheric burden of total fOA is 0.63 Tg (10% fPOA and 90%
417 fSOA). The tropospheric fSOA/fOA is higher than at the surface since SVOC and
418 IVOC continue forming fSOA at higher altitudes (Tsimpidi et al., 2014).

419 The wintertime burden of fPOA is 36% higher than its annual average value
420 (Figure 6a). This increase is partially driven by the seasonality of the emissions since
421 anthropogenic OA emissions are 12% higher during winter compared to the annual
422 average. Furthermore, the lower temperatures that occur during winter in the Northern
423 Hemisphere drive the gas-particle partitioning of freshly emitted SVOCs to the
424 aerosol phase resulting in higher fPOA concentrations. At the same time, less SVOCs
425 are available in the gas phase to react with the lower wintertime OH resulting in
426 reduced formation of fSOA. The wintertime tropospheric burden of fSOA is 16%
427 lower than the annual average value (Figure 6a) representing 87% of the fOA. During
428 summer, the photooxidation of SVOC and IVOC is significantly enhanced; however,
429 the increase in fSOA mass is compensated by evaporation due to the high
430 temperatures resulting in an overall increase of only 3% compared to the annual
431 average values. High temperatures during summer also result in a significant decrease
432 of fPOA due to evaporation, i.e., a 27% decrease compared to the annual average
433 tropospheric burden (Figure 6a). Overall, the tropospheric fSOA:fOA during summer
434 increases to 93%. The highest fSOA concentrations are predicted during spring (i.e.,

435 April) when photochemistry is active and the moderately low temperatures favor the
436 partitioning into the aerosol phase (Figure 6a).

437 Figure 6b depicts the annual tropospheric fOA variability over the simulated years
438 (2001 to 2010). The variability of the model predicted fOA is very low ($\pm 4\%$) since
439 anthropogenic emissions are assumed to have small differences between the simulated
440 years (Clarke et al., 2007). The anthropogenic OC emissions from fossil and biofuel
441 combustion increase by $1.23 \text{ Tg C yr}^{-1}$ (10%) during the simulated decade. Over Asia
442 and Africa, anthropogenic OC emissions have increased by 12% and 33%,
443 respectively, during the simulated decade. On the other hand, anthropogenic OC
444 emissions over North America have decreased by 15% during the same period. Over
445 South America, anthropogenic OC emissions have decreased up to the year 2005
446 (10%) and then remained about constant until the end of the decade. Over Europe,
447 anthropogenic OC emissions have increased up to the year 2005 (5%) and then started
448 to decrease reaching 4% lower emissions (compared to 2001) by the end of the
449 decade. However, the simulated fOA tropospheric burdens over the continents (Figure
450 6b) do not reflect this clear trend of emissions since other factors (i.e., meteorology)
451 play an important role. Overall, the lowest fOA global tropospheric burden is
452 calculated during the years 2001 and 2003 (0.61 Tg yr^{-1}) and the highest during the
453 year 2009 (0.66 Tg yr^{-1}).

454

455 **4.2 OA from open biomass burning**

456

457 **4.2.1 Geographical distribution**

458 Figure 7 depicts the simulated decadal average global surface concentrations of
459 bbPOA and bbSOA. The average surface concentration of bbPOA is $0.11 \mu\text{g m}^{-3}$. The
460 highest bbPOA concentrations (up to $7.7 \mu\text{g m}^{-3}$) are predicted over the tropical
461 rainforests (i.e., Amazon, Congo, and Southeast Asia) and the boreal forests (i.e.,
462 Alaska, Canada, and Russia) due to substantial emissions from forest and savannah
463 fires. Similar to fPOA, bbPOA levels rapidly decrease as the air masses travel from
464 the sources due to dilution and evaporation (Figure 7a). The average surface
465 concentration of bbSOA is $0.15 \mu\text{g m}^{-3}$. In contrast to anthropogenic combustion
466 emissions, IVOCs are assumed to account for only 40% of the total open biomass
467 burning emissions. Nevertheless, the model predicts that the bbSOA formed due to

468 the oxidation of IVOCs (46%) is similar to that from the oxidation of SVOCs (54%).
469 This result corroborates our finding that IVOCs are a significant source of SOA.
470 bbSOA concentrations are more spatially homogeneous compared to bbPOA reaching
471 high levels (up to $6.4 \mu\text{g m}^{-3}$) over a wide area covering most of South America,
472 Central and South Africa, Southeastern Asia, including Indonesia (Figure 7b). The
473 atmosphere over the South Atlantic Ocean is also strongly influenced by long-range
474 transport of bbSOA from the Congo Basin ($1\text{-}3 \mu\text{g m}^{-3}$). Over these areas, the
475 atmospheric conditions are favorable for the photochemical oxidation of SVOCs and
476 IVOCs. On the other hand, over the boreal forests, the low temperatures favor the
477 partitioning of SVOCs into the particulate phase forming bbPOA, and at the same
478 time the photo-oxidation of IVOCs is slow. This results in moderate average bbSOA
479 concentrations around $0.5 \mu\text{g m}^{-3}$.

480 Figure 8a depicts the predicted decadal average contribution of bbSOA to total
481 bbOA (bbSOA/bbOA) at the surface. bbSOA/bbOA is high with values ranging from
482 around 35% over the tropical and boreal forests to 85% in areas downwind and over
483 the oceans. The global average bbSOA/bbOA at the surface is predicted to be 72%.
484 This result indicates that even though the biomass burning emissions are distributed in
485 relatively low volatility bins ($C^* \leq 10^4 \mu\text{g m}^{-3}$), bbSOA still exceeds primary biomass
486 burning OA on a global scale. Figure 8b depicts the decadal average surface
487 contribution of bbOA to total OA (bbOA/OA). As expected, bbOA contributes
488 significantly to total OA over the tropical and boreal forests (around 60%) while it has
489 a smaller impact on OA levels over the mid-latitude continents of the Northern
490 Hemisphere. This result does not include other types of biomass combustion (e.g., for
491 residential heating) that often contribute significantly in urban areas (Chen et al.,
492 2007; Wang et al., 2007; Lanz et al., 2010). High bbOA contributions are also
493 predicted downwind of the boreal forests (up to 80%). Furthermore, the bbOA/OA
494 ratio is high (50-90%) off the west coasts of Africa, South America and Indonesia.
495 These high values are due to the chemical aging of biomass burning SVOCs and
496 IVOCs in contrast to the chemical products of biogenic VOCs which are not allowed
497 to participate in additional photochemical reactions (Tsimpidi et al., 2014). Overall,
498 the global average bbOA/OA is predicted to be 26%.

499

500 **4.2.2 Temporal evolution**

501 The decadal average tropospheric burden of total bbOA is 0.59 Tg yr⁻¹ (30%
502 bbPOA, 70% bbSOA) (Table 1). The fraction of bbOA that is secondary is less than
503 that of fOA (90%).

504 The interannual variability of bbPOA and bbSOA is high due to the seasonality of
505 fires (Figure 9a). During July to September (dry season) intense wildfires are reported
506 over the tropics related to the low precipitation and high temperatures. This results in
507 high biomass burning emissions which together with the intense photochemical
508 activity result in bbOA tropospheric burdens of up to 1.4 Tg yr⁻¹ during August
509 (130% higher than the annual average). The lowest bbOA tropospheric burdens are
510 estimated during the wet season (0.21 Tg yr⁻¹ during April, 64% lower than the annual
511 average). Furthermore, during the dry season OA consists mainly of bbOA over the
512 tropical rainforests due to the intense wildfires while during the wet season OA
513 consists mainly of biogenic SOA since biomass burning emissions are low. As a
514 result, the bbOA/OA has a significant seasonal variability as well; during the dry
515 season the global average bbOA/OA increases significantly (e.g., 41% during August;
516 not shown) while during the wet season it is significantly lower (e.g., 11% during
517 March; not shown).

518 The decadal variability of the model predictions is also important since open
519 biomass burning emissions can vary significantly from year to year (Figure 9b). The
520 years 2001 and 2009 had relatively low fire activity (13.5 Tg C yr⁻¹) and the bbOA
521 annual tropospheric burden was 0.47 Tg yr⁻¹ (21% lower than the decadal average).
522 During these two years tropospheric bbOA was lower over both the Amazon and the
523 Congo basin (Figure 9b). The year of 2010 on the other hand was characterized by
524 severe wildfires, especially in the Amazon region (OC emissions were twice the
525 decadal average) resulting in a global bbOA source of 0.72 Tg yr⁻¹ (21% higher than
526 the decadal average). Over the Congo Basin, the calculated tropospheric burden peaks
527 during the years 2005 and 2010 (Figure 9b) while over the Amazon Basin, the highest
528 burdens are calculated during the years 2007 and 2010. The above results are
529 consistent with Chen et al. (2013) who analyzed satellite data to detect the fire activity
530 over the Amazon rainforest and reported a twofold increase in fire activity during
531 2010 compared to 2009.

532

533 **5. Comparison with AMS data**

534

535 **5.1 Evaluation over urban areas**

536 The spatial resolution used in the current application as well as in most global
537 model applications (Tsigaridis et al., 2014) can introduce potentially significant errors
538 over urban areas. Other issues can also add to the model/measurement discrepancy
539 over cities. For example, global models, including EMAC, lack OA emissions from
540 residential and commercial cooking activities (Tsigaridis et al., 2014). However,
541 cooking can be an important source of OA that can contribute significantly to
542 measured POA (around 50%) and total OA (15%-20%) over urban areas (Sun et al.,
543 2011; Mohr et al., 2012; Ge et al., 2012; Hayes et al., 2013). Therefore our analysis
544 and use of the corresponding urban AMS datasets should be viewed as an effort to
545 quantify the magnitude of these errors. In addition, there have been a number of
546 recent studies using global atmospheric chemistry models to investigate the link
547 between premature mortality and atmospheric aerosols in urban and rural
548 environments (Lelieveld et al., 2015). Evaluating global models over urban locations
549 can provide useful information about their potential biases in these locations.

550 AMS observations indicate that over urban areas the POA (sum of HOA and
551 BBOA) concentration is relatively high while further downwind and in rural areas it
552 decreases substantially due to dilution and evaporation (Figure 10a). The model is
553 able to reproduce this trend, however, it significantly underpredicts (NMB=-65%,
554 Table 2) the high values of POA over urban areas and especially over densely
555 populated areas such as Beijing, Tokyo and Mexico City (Table S5; Figure 11a). This
556 underprediction appears to be typical for global models (Tsigaridis et al., 2014) and is
557 partly associated with the limited spatial resolution of the model (the size of a grid
558 cell used typically exceeds the size of most urban centers) and the lack of COA
559 emissions. The model underestimates SOA (NMB=-33%, Table 2) over densely
560 populated areas such as Beijing and Mexico City (Table S5; Figure 12a) partially due
561 to its limited spatial resolution. In addition, the lack of COA emissions can be
562 considered as a possible cause of OOA underestimation by the model over urban and
563 urban-downwind areas (see below) given that COA can be oxidized and form SOA
564 over the urban center and further downwind. Overall, the underestimation of OA over
565 urban locations indicates that global exposure studies (Lelieveld et al., 2015) provide

566 a lower limit of the actual contribution of OA to premature mortality over large urban
567 areas.

568 Given that the model cannot sufficiently reproduce the concentrations of POA and
569 SOA over urban locations, AMS data from these locations is not included for the
570 seasonal, continental and total (annual) evaluation of the model presented below.
571 Especially for the seasonal model evaluation, most of the urban field campaigns were
572 conducted either during winter or summer. Therefore, including these locations in our
573 analysis is expected to bias the model performance during winter and summer leading
574 to a potential misinterpretation of the corresponding seasonal results.

575

576 **5.2 Spatial evaluation**

577

578 **5.2.1 POA**

579 Over urban-downwind locations, the model does a better job than urban locations
580 in reproducing the measured POA values (Table 3). This can be verified by focusing
581 on specific field campaigns that provide data from both the urban center and urban-
582 downwind locations over the same period of time (i.e., MILAGRO over Mexico City
583 and MEGAPOLI over Paris). Over these areas the model captures the measured POA
584 concentrations downwind of the urban center (Table S6; Figure 11b) but it
585 significantly underpredicts the POA concentrations measured in the urban center
586 (Table S5; Figure 11a). Overall, in urban-downwind and rural areas the model
587 captures the lower POA levels (Figure 10a; Figure 11). Over urban-downwind areas,
588 the model slightly underpredicts POA (NMB=-15%) while over rural areas it
589 overpredicts by $0.04 \mu\text{g m}^{-3}$ (Table 3). However, over rural areas with high BBOA
590 concentrations (e.g, Massognex, Payerne, etc.) the model underpredicts POA (Table
591 S7; Figure 11c) indicating that biomass burning and/or biofuel use in residential areas
592 may be underestimated in the emission inventory.

593 In Europe, the model underestimates POA concentrations (NMB=-23%; Table 3).
594 However, the comparison of simulated fPOA with AMS HOA (i.e, excluding BBOA
595 from the comparison) suggests that the model overpredicts POA over Europe with a
596 NMB=20% (not shown). This result underscores the underestimated emissions from
597 residential biofuel use as a prominent cause of the model bias over Europe. The
598 possible underestimation of biomass/biofuel burning emissions in European

599 residential areas has also been reported by other studies (Bergstrom et al., 2012;
600 Kostenidou et al., 2013; Denier van der Gon et al., 2015). Over North America, the
601 model reproduces well the measured HOA (Table 3). Over Asia, the model
602 overestimates the low values of POA measured by AMS (Table 3) mainly due to the
603 high simulated bbPOA concentrations (45% of total POA) transported from the boreal
604 forests of Northeast Asia.

605

606 **5.2.2 SOA**

607 Both AMS and EMAC model results indicate that SOA (and OOA) is high over all
608 environments considered (Table 4). The highest concentrations are found over urban
609 locations (AMS-OOA= $4.33 \mu\text{g m}^{-3}$ and EMAC-SOA= $2.97 \mu\text{g m}^{-3}$) while further
610 downwind SOA concentrations decrease by 37% over rural locations according to
611 both AMS and EMAC results (Figure 10b). This indicates that the initial emissions of
612 VOCs, IVOCs and SVOCs are photo-oxidized rapidly in the urban environment
613 producing SOA, while their atmospheric aging and further production of SOA is
614 offset by dilution as the air masses travel from the urban centers. EMAC does a
615 reasonable job in reproducing SOA concentrations (Table 4), however, a systematic
616 underprediction is found in all types of environments. The best model performance is
617 achieved over urban downwind locations (NMB=-25%) followed by urban and rural
618 areas (NMB=-31% and -32%, respectively). The model significantly underpredicts
619 SOA over specific urban-downwind and rural areas. In most of these cases the field
620 campaign was short (up to 1 week) and the results were subject to specific pollution
621 episodes which cannot be captured by our model (e.g., Puy de Dome, Table S7;
622 Figure 12c).

623 Over the continents, the largest SOA underestimation is found over Europe
624 (NMB=-39%). Similar to POA, it is mostly driven by the model underperformance
625 over sites with high biomass burning sources and biofuel use (e.g, Harkingen and
626 Payern, CH; Figure 12). Over North America, the model simulates well the SOA
627 formation with NMB=-15%. IEPOX-SOA, a type of SOA likely formed via
628 processing of later generation isoprene products in aqueous acidic aerosols, has been
629 recently suggested as an important source of SOA close to isoprene emissions (Hu et
630 al., 2015). The model does not simulate SOA formation from aqueous-phase reactions
631 and therefore does not produce IEPOX-SOA, which may lead to an underestimation

632 of SOA over some sites in North America that are strongly influenced by isoprene
633 emissions (e.g. over the Pinnacle state park, NY; Table S7; Figure 12c). Over Asia,
634 the model slightly underestimates SOA with NMB=-22% (Table 4; Figure 12).

635 In most of the available datasets (41 out of 84), PMF provides information for the
636 two subtypes of OOA (LV-OOA and SV-OOA). Both PMF and EMAC results
637 indicate that aged SOA (or LV-OOA) is higher than fresh SOA (or SV-OOA)
638 regardless of the type of environment (Tables 5, 6). However, in North America,
639 AMS SV-OOA is slightly higher than LV-OOA while EMAC calculations indicate
640 the opposite (Tables 5, 6). Despite this discrepancy, the model reproduce well both
641 the fresh SOA (NMB=-29%) and aged SOA (NMB=-20%) over North America while
642 over Europe the underestimation is larger (Tables 5, 6) The EMAC performance is
643 better over urban locations where it reproduces the high levels of aged SOA with
644 NMB=-21% and NME=43% (Table 2). Over urban-downwind and rural locations
645 EMAC underpredicts aged SOA with NMB=-47% and -38%, respectively (Table 5).
646 The performance of the model for fresh SOA is better compared to aged SOA (Table
647 5), with the exception of North America, indicating that the modeled OA aging
648 parameterization may underestimate the SOA produced from chemical reactions
649 during transport and requires improvements. Similar to aged SOA, the best
650 performance of the model for fresh SOA is obtained over urban locations (NMB=-
651 12%).

652

653 **5.3 Seasonal evaluation**

654

655 **5.3.1 POA**

656 The model performs best during summer (RMSE=0.4, NMB=-3%), followed by
657 autumn (RMSE=0.37, NMB=-15%) and spring (RMSE=0.52, NMB=21%). During
658 winter EMAC underpredicts POA with NMB=-34% (Table 3; Figure 13a). This result
659 corroborates our hypothesis that residential biofuel emissions may be underestimated
660 in the inventory since residential heating is expected during winter. Furthermore,
661 since vehicle catalysts require a certain temperature to work to full efficiency,
662 emissions from gasoline and diesel engines are significantly higher during the warm-
663 up phase of the car (Westerholm et al., 1996). Typically, the additional emissions
664 during the warm-up phase (or cold-start emissions) are not accounted for in emission

665 inventories, which are based on measurements at an ambient temperature of 23 °C
666 (Weilenmann et al., 2009). However, cold-start emissions increase considerably at
667 lower ambient temperatures varying by more than one order of magnitude between 23
668 and -20 °C (Weilenmann et al., 2009), and thus significant underestimations of OA
669 emissions from the transport sector can be expected during wintertime. Kopacz et al.
670 (2010) provide a global estimate of CO sources by adjoint inversion of satellite
671 datasets and reported an underestimation of CO sources during the winter season due
672 to larger than expected CO emissions from vehicle cold starts and residential heating.
673 Errors in the POA volatility distributions can also explain parts of the
674 model/measurement discrepancy. An overestimation of the fresh POA volatility will
675 favor its evaporation resulting in an underestimation of POA levels by the model.
676 Another source of the POA underestimation by EMAC may be the treatment of wet
677 deposition. The sensitivity of the results to the emission and deposition
678 parameterizations (e.g., the Henry's law constants for the organic vapors) will be
679 tested in a subsequent article in preparation.

680 According to recent studies (Cappa and Wilson, 2012; Aumont et al., 2012; Zhang
681 et al., 2013), not all oxidation products of SVOCs and IVOCs can be assigned to the
682 OOA mass fraction since they are not sufficiently oxidized. Fountoukis et al. (2014)
683 assumed that 50% of the simulated SOA-sv and SOA-iv is still considered as HOA by
684 the AMS analysis and found significant improvements in the view of the modeled
685 bias for POA. In this study we tested this hypothesis and we considered POA to be the
686 sum of fPOA and bbPOA and 50% of the SOA-sv and SOA-iv produced from the
687 first oxidation step of SVOCs and IVOCs, respectively. We assumed that SOA-sv and
688 SOA-iv produced during subsequent oxidation steps together with all the SOA-v are
689 sufficiently oxidized to be considered 100% OOA. Following this hypothesis the
690 model performance improved during winter (NME=55% $\mu\text{g m}^{-3}$ and NMB=-28%) and
691 autumn (NME=50% $\mu\text{g m}^{-3}$ and NMB=1%) and deteriorated during spring
692 (NME=110% $\mu\text{g m}^{-3}$ and NMB=49%) and summer (NME=71% $\mu\text{g m}^{-3}$ and
693 NMB=16%) when the oxidation of SVOCs and IVOCs is enhanced significantly.

694

695 **5.3.2 SOA**

696 The best performance of the model is found for spring (NME=46%, NMB=-24%)
697 followed by the autumn (NME=52%, NMB=-25%) and summer (NME=44%,

698 NMB=-28%) (Table 4; Figure 13b). However, during winter the model strongly
699 underpredicts OOA concentrations (NME=80%, NMB=-80%). The overall
700 underprediction of OOA concentrations indicates that the model is missing an
701 important source or formation pathway of SOA. Possible underestimation of
702 residential biofuel emissions in our model, identified during the spatial and seasonal
703 evaluation of simulated POA, can lead to an underestimation of SOA formed from the
704 oxidation of these emissions during winter. Fountoukis et al. (2015) also reported low
705 modeled SOA values compared to AMS OOA over the Paris region and attributed this
706 discrepancy to the transformation of BBOA to OOA without the presence of sunlight
707 reported by some recent studies (Bougiatioti et al., 2014; Crippa et al., 2013b).
708 Underestimation of cold-start vehicle emissions during winter can also lead to a
709 significant underestimation of SOA, since SOA produced from organic compounds
710 emitted during the warm-up phase can be 3-7 times higher than SOA produced when
711 the catalyst is hot (Gordon et al., 2014). Furthermore, ORACLE assumes that the only
712 source of SOA is the homogeneous gas-phase photochemical oxidation of SOA
713 precursors. Therefore, the negative bias of the model during winter may also be
714 explained by its inability to simulate SOA formed from aqueous-phase and other
715 heterogeneous reactions, including processes like oligomerization. Such processes
716 should be taking place in all seasons. However, during the photochemically active
717 periods (e.g., summer) there are other chemical pathways (e.g., reactions with OH and
718 ozone) to convert the organic precursors to SOA. Adding to this the increased
719 presence of lower-level clouds during winter and early spring compared to summer in
720 North Hemisphere mid-latitudes (Stubenrauch et al., 2006), one would expect a higher
721 importance of heterogeneous oxidation in winter. Finally, the underprediction of SOA
722 by the model during winter may be also associated with an overestimation of
723 atmospheric removal.

724 PMF and EMAC results indicate that aged SOA levels exceed those of fresh SOA
725 during all seasons. The EMAC performance for aged SOA appears to be better during
726 spring (NMB=-33%), summer (NMB=-36%), and autumn (NMB=-32%), and much
727 worse during winter (NMB=-91%) (Table 5; Figure 13c). The overall performance of
728 the model for fresh SOA (NME=60%, NMB=-30%) (Table 6, Figure 13d) appears to
729 be better than aged SOA (NME=71%, NMB=-40%) which supports our conclusions
730 from the spatial model evaluation that the atmospheric aging of SOA may be

731 underestimated by EMAC. However, this apparent discrepancy may be partially due
732 to our assumption that LV-OOA corresponds only to multiple generational SOA. This
733 is not consistent with recent studies that reported formation of LV-OOA from the first
734 oxidation step of biogenic VOCs (Ehn et al., 2014). During winter, EMAC also
735 underestimates the fresh SOA levels (NMB=-79%). This underprediction of both
736 fresh and aged SOA during winter suggests that one or more important wintertime
737 SOA formation pathways are missing in our model.

738

739 **5.4 OA composition**

740 According to PMF results, the OOA/OA ratio increases downwind of the urban
741 centers and in rural areas (from 61% over urban environments to 86% over remote
742 areas; Figure 14a). This is generally consistent with the EMAC predictions. The
743 predicted SOA/OA fraction increases downwind of the urban centers (from 76% over
744 urban locations to 80% over rural areas). This change is lower than the PMF estimates
745 but could be explained by the uncertainty of the PMF analyses (Figure 14a).
746 Alternatively, this may indicate that EMAC tends to underpredict the aging rate of
747 OA. OOA/OA is consistently high during all seasons (around 80%) with the highest
748 ratio predicted in summer (90%) and the lowest in winter (74%) (Figure 14b). The
749 model predicts high SOA/OA during all seasons except winter (Figure 14b). The
750 highest SOA/OA ratio is predicted during summer (87%) when the photo-oxidation of
751 SOA is enhanced. The low SOA/OA during winter (47%) once again shows the
752 inability of EMAC model to reproduce the observed SOA levels during that season.

753 Both PMF and EMAC indicate that aged SOA is higher than fresh SOA in all types
754 of environment and seasons (Figure 15). PMF results suggest that LV-OOA/OOA is
755 higher over urban-downwind environments (69%), while EMAC aged SOA/SOA is
756 similar over all types of locations (59%) (Figure 15a). The high fresh SOA fraction
757 estimated over rural areas by both PMF and EMAC (around 40%) indicates that fresh
758 SOA production occurs even remote from the sources. The composition of OOA
759 exhibits a seasonal cycle as well since AMS results indicate that LV-OOA/OOA is
760 higher during winter (73%) and lower during summer (57%) (Figure 15b). EMAC
761 predicts the highest aged SOA/SOA fraction during spring (68%) and the lowest
762 during winter (53%) without any clear seasonal pattern (Figure 15b).

763

764 **6. Conclusions**

765 This study estimates the impact of open biomass burning and anthropogenic
766 combustion emissions (from fossil and biofuels) of SVOCs and IVOCs to global OA
767 budgets and distributions. The EMAC simulations indicate that the tropospheric
768 burden of OA consists of 32% fOA and 30% bbOA. Furthermore, 90% of fOA and
769 70% of bbOA is predicted to be secondary. These results support recent findings from
770 global studies that have also reported strong contributions of SOA from
771 anthropogenic sources to global OA concentrations (Spracklen et al., 2011; Carslaw et
772 al., 2013; Lee et al., 2013; Tsimpidi et al., 2014).

773 The tropospheric burdens of fOA and bbOA follow a clear seasonal pattern. fOA is
774 higher during the boreal summer (0.63 Tg) and lower during winter (0.57 Tg), while
775 bbOA is higher during the dry season in the tropics (1.15 Tg during August) and
776 lower during the wet season (0.17 Tg during April). The simulated spatial distribution
777 of fOA and bbOA is driven by the sources of their precursors and atmospheric
778 transport. Higher fPOA concentrations occur over densely populated and highly
779 industrialized areas of the Northern Hemisphere while further downwind fPOA
780 decreases substantially due to dilution and evaporation. On the other hand, fSOA
781 maintains similar levels downwind of the anthropogenic sources due to the continued
782 chemical transformations. bbPOA concentrations peak over the tropical and the boreal
783 forests while bbSOA has high concentrations over a wide area covering most of South
784 America, Central and South Africa, Southeastern Asia, including Indonesia and even
785 parts of the Southern Atlantic Ocean.

786 AMS results from 84 field campaigns performed at continental locations in the
787 Northern Hemisphere during the examined period (2001-2010) have been used to
788 provide further insights into the composition of OA in three different types of
789 environments: urban, urban-downwind and rural areas, during four seasons. The
790 spatial analysis of AMS and EMAC results indicate that over urban areas POA is
791 highest while further downwind and in rural areas decreases substantially due to
792 dilution and evaporation. On the other hand, SOA is found to be high over all types of
793 environments. This results in an increase of the SOA/OA ratio downwind of the urban
794 centers. The seasonal analysis of the results does not include the urban areas since the
795 model cannot reproduce the high OA concentrations over urban environments due to
796 its limited spatial resolution. The seasonal evaluation of the model results against the

797 AMS measurements showed a major weakness of the model associated with
798 calculated POA and SOA concentration levels during winter. This indicates that the
799 model is probably missing both an important source and a formation pathway of OA,
800 which becomes increasingly important during boreal winter. Possible causes include
801 the underestimation of residential biofuel emissions during winter, the
802 underestimation of vehicle cold-start emissions, the neglect of aqueous-phase and
803 heterogeneous oxidation reactions in the model, and the overestimation of the
804 atmospheric removal of POA and freshly formed SOA.

805 AMS results indicate that OA consists of 15% HOA and 85% OOA on average
806 during all seasons. EMAC is able to reproduce this dominance of OOA and its results
807 suggest that SOA accounts for 80% of total OA. At many locations, PMF analysis
808 identified two subtypes of OOA that differ in volatility and oxidation state (LV-OOA
809 and SV-OOA). PMF results indicate that LV-OOA is higher than SV-OOA regardless
810 of the season or the type of environment. The overall LV-OOA/OOA fraction during
811 the four seasons is 63% according to AMS measurement analysis. Assuming that SV-
812 OOA corresponds to fresh SOA (first generation oxidation products) and LV-OOA
813 corresponds to aged SOA (later generation oxidation products), EMAC is able to
814 reproduce the PMF results predicting a dominance of aged SOA during all seasons
815 (59% of the total SOA on average).

816

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824

825 **8. References**

826 Alvarado, M. J., Lonsdale, C. R., Yokelson, R. J., Akagi, S. K., Coe, H., Craven, J. S.,
827 Fischer, E. V., McMeeking, G. R., Seinfeld, J. H., Soni, T., Taylor, J. W.,
828 Weise, D. R., and Wold, C. E.: Investigating the links between ozone and
829 organic aerosol chemistry in a biomass burning plume from a prescribed fire in
830 California chaparral, *Atmos. Chem. Phys.*, 15, 6667-6688, 10.5194/acp-15-
831 6667-2015, 2015.

832 Aiken, A. C., Salcedo, D., Cubison, M. J., Huffman, J. A., DeCarlo, P. F., Ulbrich, I.
833 M., Docherty, K. S., Sueper, D., Kimmel, J. R., Worsnop, D. R., Trimborn, A.,
834 Northway, M., Stone, E. A., Schauer, J. J., Volkamer, R. M., Fortner, E., de
835 Foy, B., Wang, J., Laskin, A., Shutthanandan, V., Zheng, J., Zhang, R.,
836 Gaffney, J., Marley, N. A., Paredes-Miranda, G., Arnott, W. P., Molina, L. T.,
837 Sosa, G., and Jimenez, J. L.: Mexico City aerosol analysis during MILAGRO
838 using high resolution aerosol mass spectrometry at the urban supersite (T0) -
839 Part 1: Fine particle composition and organic source apportionment, *Atmo.*
840 *Chem. Phys.*, 9, 6633-6653, 2009.

841 Aiken, A. C., de Foy, B., Wiedinmyer, C., DeCarlo, P. F., Ulbrich, I. M., Wehrli, M.
842 N., Szidat, S., Prevot, A. S. H., Noda, J., Wacker, L., Volkamer, R., Fortner, E.,
843 Wang, J., Laskin, A., Shutthanandan, V., Zheng, J., Zhang, R., Paredes-
844 Miranda, G., Arnott, W. P., Molina, L. T., Sosa, G., Querol, X., and Jimenez, J.
845 L.: Mexico city aerosol analysis during MILAGRO using high resolution
846 aerosol mass spectrometry at the urban supersite (T0) - Part 2: Analysis of the
847 biomass burning contribution and the non-fossil carbon fraction, *Atmo. Chem.*
848 *Phys.*, 10, 2010.

849 Athanasopoulou, E., Vogel, H., Vogel, B., Tsimpidi, A. P., Pandis, S. N., Knote, C.,
850 and Fountoukis, C.: Modeling the meteorological and chemical effects of
851 secondary organic aerosols during an EUCAARI campaign, *Atmos. Chem.*
852 *Phys.*, 13, 625-645, 2013.

853 Aumont, B., Valorso, R., Mouchel-Vallon, C., Camredon, M., Lee-Taylor, J., and
854 Madronich, S.: Modeling SOA formation from the oxidation of intermediate
855 volatility n-alkanes, *Atmo. Chem. Phys.*, 12, 7577-7589, 2012.

856 Bahreini, R., Middlebrook, A. M., de Gouw, J. A., Warneke, C., Trainer, M., Brock,
857 C. A., Stark, H., Brown, S. S., Dube, W. P., Gilman, J. B., Hall, K., Holloway,
858 J. S., Kuster, W. C., Perring, A. E., Prevot, A. S. H., Schwarz, J. P., Spackman,
859 J. R., Szidat, S., Wagner, N. L., Weber, R. J., Zotter, P., and Parrish, D. D.:
860 Gasoline emissions dominate over diesel in formation of secondary organic
861 aerosol mass, *Geophys. Res. Lett.*, 39, doi: 10.1029/2011gl050718, 2012.

862 Bergstrom, R., van der Gon, H. A. C. D., Prevot, A. S. H., Yttri, K. E., and Simpson,
863 D.: Modelling of organic aerosols over Europe (2002-2007) using a volatility
864 basis set (VBS) framework: application of different assumptions regarding the
865 formation of secondary organic aerosol, *Atmos. Chem. Phys.*, 12, 8499-8527,
866 2012.

867 Bougiatioti, A., Stavroulas, I., Kostenidou, E., Zarnpas, P., Theodosi, C., Kouvarakis,
868 G., Canonaco, F., Prevot, A. S. H., Nenes, A., Pandis, S. N., and Mihalopoulos,
869 N.: Processing of biomass-burning aerosol in the eastern Mediterranean during
870 summertime, *Atmo. Chem. Phys.*, 14, 4793-4807, 2014.

871 Cappa, C. D., and Wilson, K. R.: Multi-generation gas-phase oxidation, equilibrium
872 partitioning, and the formation and evolution of secondary organic aerosol,
873 *Atmo. Chem. Phys.*, 12, 9505-9528, 2012.

874 Canagaratna, M. R., Jimenez, J. L., Kroll, J. H., Chen, Q., Kessler, S. H., Massoli, P.,
875 Ruiz, L. H., Fortner, E., Williams, L. R., Wilson, K. R., Surratt, J. D., Donahue,
876 N. M., Jayne, J. T., and Worsnop, D. R.: Elemental ratio measurements of
877 organic compounds using aerosol mass spectrometry: characterization,
878 improved calibration, and implications, *Atmo. Chem. Phys.*, 15, 253-272, 2015

879 Carbone, S., Aurela, M., Saarnio, K., Saarikoski, S., Timonen, H., Frey, A., Sueper,
880 D., Ulbrich, I. M., Jimenez, J. L., Kulmala, M., Worsnop, D. R., and Hillamo,

881 R. E.: Wintertime Aerosol Chemistry in Sub-Arctic Urban Air, *Aerosol Sci.*
882 *Tech.*, 48, 313-323, 2014.

883 Carslaw, K. S., Lee, L. A., Reddington, C. L., Mann, G. W., and Pringle, K. J.: The
884 magnitude and sources of uncertainty in global aerosol, *Faraday Discuss.*, 165,
885 495-512, 2013.

886 Chen, L. W. A., Watson, J. G., Chow, J. C., and Magliano, K. L.: Quantifying PM_{2.5}
887 source contributions for the San Joaquin Valley with multivariate receptor
888 models, *Environ. Sci. Tech.*, 41, 2818-2826, 2007.

889 Chen, Q., Farmer, D. K., Rizzo, L. V., Pauliquevis, T., Kuwata, M., Karl, T. G.,
890 Guenther, A., Allan, J. D., Coe, H., Andreae, M. O., Poschl, U., Jimenez, J. L.,
891 Artaxo, P., and Martin, S. T.: Submicron particle mass concentrations and
892 sources in the Amazonian wet season (AMAZE-08), *Atmo. Chem. Phys.*, 15,
893 3687-3701, 2015.

894 Chen, Y., Morton, D. C., Jin, Y., Gollatz, G. J., Kasibhatla, P. S., van der Werf, G. R.,
895 DeFries, R. S., and Randerson, J. T.: Long-term trends and interannual
896 variability of forest, savanna and agricultural fires in South America, *Carbon*
897 *Management*, 4, 617-638, 2013.

898 Chirico, R., DeCarlo, P. F., Heringa, M. F., Tritscher, T., Richter, R., Prevot, A. S. H.,
899 Dommen, J., Weingartner, E., Wehrle, G., Gysel, M., Laborde, M., and
900 Baltensperger, U.: Impact of aftertreatment devices on primary emissions and
901 secondary organic aerosol formation potential from in-use diesel vehicles:
902 results from smog chamber experiments, *Atmo. Chem. Phys.*, 10, 11545-11563,
903 2010.

904 Clarke, L., Edmonds, J., Jacoby, H., Pitcher, H., Reilly, J., and Richels, R.: Scenarios
905 of greenhouse gas emissions and atmospheric concentrations (Part A) and
906 review of integrated scenario development and application (Part B). A report by
907 the U.S. climate change science program and the subcommittee on global
908 change research, 2007.

909 Crippa, M., El Haddad, I., Slowik, J. G., DeCarlo, P. F., Mohr, C., Heringa, M. F.,
910 Chirico, R., Marchand, N., Sciare, J., Baltensperger, U., and Prevot, A. S. H.:
911 Identification of marine and continental aerosol sources in Paris using high
912 resolution aerosol mass spectrometry, *Journal of Geophysical Research-*
913 *Atmospheres*, 118, 1950-1963, 10.1002/jgrd.50151, 2013a.

914 Crippa, M., DeCarlo, P. F., Slowik, J. G., Mohr, C., Heringa, M. F., Chirico, R.,
915 Poulain, L., Freutel, F., Sciare, J., Cozic, J., Di Marco, C. F., Elsasser, M.,
916 Nicolas, J. B., Marchand, N., Abidi, E., Wiedensohler, A., Drewnick, F.,
917 Schneider, J., Borrmann, S., Nemitz, E., Zimmermann, R., Jaffrezo, J. L.,
918 Prevot, A. S. H., and Baltensperger, U.: Wintertime aerosol chemical
919 composition and source apportionment of the organic fraction in the
920 metropolitan area of Paris, *Atmo. Chem. Phys.*, 13, 961-981, 2013b.

921 Crippa, M., Canonaco, F., Lanz, V. A., Aijala, M., Allan, J. D., Carbone, S., Capes,
922 G., Ceburnis, D., Dall'Osto, M., Day, D. A., DeCarlo, P. F., Ehn, M., Eriksson,
923 A., Freney, E., Ruiz, L. H., Hillamo, R., Jimenez, J. L., Junninen, H., Kiendler-
924 Scharr, A., Kortelainen, A. M., Kulmala, M., Laaksonen, A., Mensah, A., Mohr,
925 C., Nemitz, E., O'Dowd, C., Ovadnevaite, J., Pandis, S. N., Petaja, T., Poulain,
926 L., Saarikoski, S., Sellegri, K., Swietlicki, E., Tiitta, P., Worsnop, D. R.,
927 Baltensperger, U., and Prevot, A. S. H.: Organic aerosol components derived
928 from 25 AMS data sets across Europe using a consistent ME-2 based source
929 apportionment approach, *Atmo. Chem. Phys.*, 14, 2014.

930 DeCarlo, P. F., Ulbrich, I. M., Crounse, J., de Foy, B., Dunlea, E. J., Aiken, A. C.,
931 Knapp, D., Weinheimer, A. J., Campos, T., Wennberg, P. O., and Jimenez, J.
932 L.: Investigation of the sources and processing of organic aerosol over the
933 Central Mexican Plateau from aircraft measurements during MILAGRO, *Atmo.*
934 *Chem. Phys.*, 10, 5257-5280, 2010.

935 Denier van der Gon, H. A. C., Bergström, R., Fountoukis, C., Johansson, C., Pandis,
936 S. N., Simpson, D., and Visschedijk, A.: Particulate emissions from residential
937 wood combustion in Europe – revised estimates and an evaluation, *Atmos.*
938 *Chem. Phys. Discuss.*, 14, 31719-31765, 2014.

939 Donahue, N. M., Robinson, A. L., Stanier, C. O., and Pandis, S. N.: Coupled
940 partitioning, dilution, and chemical aging of semivolatile organics, *Environ. Sci.*
941 *Technol.*, 40, 2635-2643, 2006.

942 Donahue, N. M., Robinson, A. L., and Pandis, S. N.: Atmospheric organic particulate
943 matter: From smoke to secondary organic aerosol, *Atmo. Environ.*, 43, 94-106,
944 2009.

945 Ehn, M., Thornton, J. A., Kleist, E., Sipila, M., Junninen, H., Pullinen, I., Springer,
946 M., Rubach, F., Tillmann, R., Lee, B., Lopez-Hilfiker, F., Andres, S., Acir, I.-
947 H., Rissanen, M., Jokinen, T., Schobesberger, S., Kangasluoma, J., Kontkanen,
948 J., Nieminen, T., Kurten, T., Nielsen, L. B., Jorgensen, S., Kjaergaard, H. G.,
949 Canagaratna, M., Maso, M. D., Berndt, T., Petaja, T., Wahner, A., Kerminen,
950 V.-M., Kulmala, M., Worsnop, D. R., Wildt, J., and Mentel, T. F.: A large
951 source of low-volatility secondary organic aerosol, *Nature*, 506, doi:
952 10.1038/nature13032, 2014.

953 Fountoukis, C., Racherla, P. N., van der Gon, H. A. C. D., Polymeneas, P.,
954 Charalampidis, P. E., Pilinis, C., Wiedensohler, A., Dall'Osto, M., O'Dowd, C.,
955 and Pandis, S. N.: Evaluation of a three-dimensional chemical transport model
956 (PMCAMx) in the European domain during the EUCAARI May 2008
957 campaign, *Atmos. Chem. and Phys.*, 11, 10331-10347, 2011.

958 Fountoukis, C., Megaritis, A. G., Skyllakou, K., Charalampidis, P. E., Pilinis, C., van
959 der Gon, H., Crippa, M., Canonaco, F., Mohr, C., Prevot, A. S. H., Allan, J. D.,
960 Poulain, L., Petaja, T., Tiitta, P., Carbone, S., Kiendler-Scharr, A., Nemitz, E.,
961 O'Dowd, C., Swietlicki, E., and Pandis, S. N.: Organic aerosol concentration
962 and composition over Europe: insights from comparison of regional model
963 predictions with aerosol mass spectrometer factor analysis, *Atmo. Chem. Phys.*,
964 14, 9061-9076, 2014.

965 Fountoukis, C., Megaritis, A. G., Skyllakou, K., Charalampidis, P. E., Denier van der
966 Gon, H. A. C., Crippa, M., Prévôt, A. S. H., Freutel, F., Wiedensohler, A.,
967 Pilinis, C., and Pandis, S. N.: Simulating the formation of carbonaceous aerosol
968 in a European Megacity (Paris) during the MEGAPOLI summer and winter
969 campaigns, *Atmos. Chem. Phys. Discuss.*, 15, 25547-25582, 2015.

970 Ge, X., Setyan, A., Sun, Y., and Zhang, Q.: Primary and secondary organic aerosols
971 in Fresno, California during wintertime: Results from high resolution aerosol
972 mass spectrometry, *Journal of Geophysical Research-Atmospheres*, 117,
973 10.1029/2012jd018026, 2012.

974 Gentner, D. R., Isaacman, G., Worton, D. R., Chan, A. W. H., Dallmann, T. R., Davis,
975 L., Liu, S., Day, D. A., Russell, L. M., Wilson, K. R., Weber, R., Guha, A.,
976 Harley, R. A., and Goldstein, A. H.: Elucidating secondary organic aerosol from
977 diesel and gasoline vehicles through detailed characterization of organic carbon

978 emissions, Proceedings of the National Academy of Sciences of the United
979 States of America, 109, 18318-18323, 2012.

980 Goldstein, A. H., and Galbally, I. E.: Known and unexplored organic constituents in
981 the earth's atmosphere, *Environ. Sci. Tech.*, 41, 1514-1521, 2007.

982 Gordon, T. D., Presto, A. A., May, A. A., Nguyen, N. T., Lipsky, E. M., Donahue, N.
983 M., Gutierrez, A., Zhang, M., Maddox, C., Rieger, P., Chattopadhyay, S.,
984 Maldonado, H., Maricq, M. M., and Robinson, A. L.: Secondary organic aerosol
985 formation exceeds primary particulate matter emissions for light-duty gasoline
986 vehicles, *Atmo. Chem. Phys.*, 14, 2014.

987 Grieshop, A. P., Logue, J. M., Donahue, N. M., and Robinson, A. L.: Laboratory
988 investigation of photochemical oxidation of organic aerosol from wood fires 1:
989 measurement and simulation of organic aerosol evolution, *Atmos. Chem. Phys.*,
990 9, 1263-1277, 2009.

991 Hayes, P. L., Ortega, A. M., Cubison, M. J., Froyd, K. D., Zhao, Y., Cliff, S. S., Hu,
992 W. W., Toohey, D. W., Flynn, J. H., Lefer, B. L., Grossberg, N., Alvarez, S.,
993 Rappenglueck, B., Taylor, J. W., Allan, J. D., Holloway, J. S., Gilman, J. B.,
994 Kuster, W. C., De Gouw, J. A., Massoli, P., Zhang, X., Liu, J., Weber, R. J.,
995 Corrigan, A. L., Russell, L. M., Isaacman, G., Worton, D. R., Kreisberg, N. M.,
996 Goldstein, A. H., Thalman, R., Waxman, E. M., Volkamer, R., Lin, Y. H.,
997 Surratt, J. D., Kleindienst, T. E., Offenberg, J. H., Dusanter, S., Griffith, S.,
998 Stevens, P. S., Brioude, J., Angevine, W. M., and Jimenez, J. L.: Organic
999 aerosol composition and sources in Pasadena, California, during the 2010
1000 CalNex campaign, *J. Geophys. Res. Atmo.*, 118, 9233-9257, doi:
1001 10.1002/jgrd.50530, 2013.

1002 Henze, D. K., Seinfeld, J. H., Ng, N. L., Kroll, J. H., Fu, T. M., Jacob, D. J., and
1003 Heald, C. L.: Global modeling of secondary organic aerosol formation from
1004 aromatic hydrocarbons: high- vs. low-yield pathways, *Atmo. Chem. Phys.*, 8,
1005 2405-2420, 2008.

1006 Hodzic, A., Jimenez, J. L., Madronich, S., Canagaratna, M. R., DeCarlo, P. F.,
1007 Kleinman, L., and Fast, J.: Modeling organic aerosols in a megacity: potential
1008 contribution of semi-volatile and intermediate volatility primary organic
1009 compounds to secondary organic aerosol formation, *Atmos. Chem. Phys.*, 10,
1010 5491-5514, 2010.

1011 Hodzic, A., Kasibhatla, P. S., Jo, D. S., Cappa, C., Jimenez, J. L., Madronich, S., and
1012 Park, R. J.: Rethinking the global secondary organic aerosol (SOA) budget:
1013 stronger production, faster removal, shorter lifetime, *Atmos. Chem. Phys.*
1014 *Discuss.*, 2015, 32413-32468, 10.5194/acpd-15-32413-2015, 2015.

1015 Hu, W. W., Campuzano-Jost, P., Palm, B. B., Day, D. A., Ortega, A. M., Hayes, P. L.,
1016 Krechmer, J. E., Chen, Q., Kuwata, M., Liu, Y. J., de Sa, S. S., McKinney, K.,
1017 Martin, S. T., Hu, M., Budisulistiorini, S. H., Riva, M., Surratt, J. D., St Clair, J.
1018 M., Isaacman-Van Wertz, G., Yee, L. D., Goldstein, A. H., Carbone, S., Brito,
1019 J., Artaxo, P., de Gouw, J. A., Koss, A., Wisthaler, A., Mikoviny, T., Karl, T.,
1020 Kaser, L., Jud, W., Hansel, A., Docherty, K. S., Alexander, M. L., Robinson, N.
1021 H., Coe, H., Allan, J. D., Canagaratna, M. R., Paulot, F., and Jimenez, J. L.:
1022 Characterization of a real-time tracer for isoprene epoxydiols-derived secondary
1023 organic aerosol (IEPOX-SOA) from aerosol mass spectrometer measurements,
1024 *Atmospheric Chemistry and Physics*, 15, 11807-11833, 10.5194/acp-15-11807-
1025 2015, 2015.

1026 IPCC: (Intergovernmental Panel on Climate Change): The physical science basis.
1027 Contribution of working group I to the fifth assessment report of the
1028 intergovernmental panel on climate change. T.F. Stocker, D. Qin, G.-K.
1029 Plattner, M. Tignor, S.K. Allen, J. Boschung, A. Nauels, Y. Xia, V. Bex, and
1030 P.M. Midgley (eds.). Cambridge University Press, Cambridge, United Kingdom
1031 and New York, NY, USA, 2013.

1032 Jathar, S. H., Farina, S. C., Robinson, A. L., and Adams, P. J.: The influence of semi-
1033 volatile and reactive primary emissions on the abundance and properties of
1034 global organic aerosol, *Atmos. Chem. Phys.*, 11, 7727-7746, 2011.

1035 Jimenez, J. L., Jayne, J. T., Shi, Q., Kolb, C. E., Worsnop, D. R., Yourshaw, I.,
1036 Seinfeld, J. H., Flagan, R. C., Zhang, X. F., Smith, K. A., Morris, J. W., and
1037 Davidovits, P.: Ambient aerosol sampling using the Aerodyne Aerosol Mass
1038 Spectrometer, *J. Geophys. Res. Atmo.*, 108, doi: 10.1029/2001jd001213, 2003.

1039 Jimenez, J. L., Canagaratna, M. R., Donahue, N. M., Prevot, A. S. H., Zhang, Q.,
1040 Kroll, J. H., DeCarlo, P. F., Allan, J. D., Coe, H., Ng, N. L., Aiken, A. C.,
1041 Docherty, K. S., Ulbrich, I. M., Grieshop, A. P., Robinson, A. L., Duplissy, J.,
1042 Smith, J. D., Wilson, K. R., Lanz, V. A., Hueglin, C., Sun, Y. L., Tian, J.,
1043 Laaksonen, A., Raatikainen, T., Rautiainen, J., Vaattovaara, P., Ehn, M.,
1044 Kulmala, M., Tomlinson, J. M., Collins, D. R., Cubison, M. J., Dunlea, E. J.,
1045 Huffman, J. A., Onasch, T. B., Alfarra, M. R., Williams, P. I., Bower, K.,
1046 Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S., Weimer, S., Demerjian,
1047 K., Salcedo, D., Cottrell, L., Griffin, R., Takami, A., Miyoshi, T., Hatakeyama,
1048 S., Shimojo, A., Sun, J. Y., Zhang, Y. M., Dzepina, K., Kimmel, J. R., Sueper,
1049 D., Jayne, J. T., Herndon, S. C., Trimborn, A. M., Williams, L. R., Wood, E. C.,
1050 Middlebrook, A. M., Kolb, C. E., Baltensperger, U., and Worsnop, D. R.:
1051 Evolution of organic aerosols in the atmosphere, *Science*, 326, 1525-1529,
1052 2009.

1053 Jöckel, P., Tost, H., Pozzer, A., Bruehl, C., Buchholz, J., Ganzeveld, L., Hoor, P.,
1054 Kerkweg, A., Lawrence, M. G., Sander, R., Steil, B., Stiller, G., Tanarhte, M.,
1055 Taraborrelli, D., Van Aardenne, J., and Lelieveld, J.: The atmospheric chemistry
1056 general circulation model ECHAM5/MESSy1: consistent simulation of ozone
1057 from the surface to the mesosphere, *Atmos. Chem. Phys.*, 6, 5067-5104, 2006.

1058 Kanakidou, M., Seinfeld, J. H., Pandis, S. N., Barnes, I., Dentener, F. J., Facchini, M.
1059 C., Van Dingenen, R., Ervens, B., Nenes, A., Nielsen, C. J., Swietlicki, E.,
1060 Putaud, J. P., Balkanski, Y., Fuzzi, S., Horth, J., Moortgat, G. K., Winterhalter,
1061 R., Myhre, C. E. L., Tsigaridis, K., Vignati, E., Stephanou, E. G., and Wilson,
1062 J.: Organic aerosol and global climate modelling: a review, *Atmos. Chem.
1063 Phys.*, 5, 1053-1123, 2005.

1064 Karydis, V. A., Tsimpidi, A. P., Pozzer, A., Astitha, M., and Lelieveld, J.: Effects of
1065 mineral dust on global atmospheric nitrate concentrations, *Atmos. Chem. Phys.*,
1066 16, 1491-1509, 10.5194/acp-16-1491-2016, 2016.

1067 Kerkweg, A., Buchholz, J., Ganzeveld, L., Pozzer, A., Tost, H., and Jöckel, P.:
1068 Technical Note: An implementation of the dry removal processes DRY
1069 DEPosition and SEDimentation in the Modular Earth Submodel System
1070 (MESSy), *Atmos. Chem. Phys.*, 6, 4617-4632, 2006a.

1071 Kerkweg, A., Sander, R., Tost, H., and Jöckel, P.: Technical note: Implementation of
1072 prescribed (OFFLEM), calculated (ONLEM), and pseudo-emissions
1073 (TNUDGE) of chemical species in the Modular Earth Submodel System
1074 (MESSy), *Atmos. Chem. Phys.*, 6, 3603-3609, 2006b.

1075 Kopacz, M., Jacob, D. J., Fisher, J. A., Logan, J. A., Zhang, L., Megretskaia, I. A.,
 1076 Yantosca, R. M., Singh, K., Henze, D. K., Burrows, J. P., Buchwitz, M.,
 1077 Khlystova, I., McMillan, W. W., Gille, J. C., Edwards, D. P., Eldering, A.,
 1078 Thouret, V., and Nedelec, P.: Global estimates of CO sources with high
 1079 resolution by adjoint inversion of multiple satellite datasets (MOPITT, AIRS,
 1080 SCIAMACHY, TES), *Atmos. Chem. Phys.*, 10, 855-876, 10.5194/acp-10-855-
 1081 2010, 2010.
 1082 Kostenidou, E., Kaltsonoudis, C., Tsiflikiotou, M., Louvaris, E., Russell, L. M., and
 1083 Pandis, S. N.: Burning of olive tree branches: a major organic aerosol source in
 1084 the Mediterranean, *Atmo. Chem. Phys.*, 13, 8797-8811, 2013.
 1085 Kroll, J. H., and Seinfeld, J. H.: Chemistry of secondary organic aerosol: Formation
 1086 and evolution of low-volatility organics in the atmosphere, *Atmos. Environ.*, 42,
 1087 3593-3624, 2008.
 1088 Lanz, V. A., Alfarra, M. R., Baltensperger, U., Buchmann, B., Hueglin, C., and
 1089 Prevot, A. S. H.: Source apportionment of submicron organic aerosols at an
 1090 urban site by factor analytical modelling of aerosol mass spectra, *Atmo. Chem.*
 1091 *Phys.*, 7, 1503-1522, 2007.
 1092 Lanz, V. A., Alfarra, M. R., Baltensperger, U., Buchmann, B., Hueglin, C., Szidat, S.,
 1093 Wehrli, M. N., Wacker, L., Weimer, S., Caseiro, A., Puxbaum, H., and Prevot,
 1094 A. S. H.: Source attribution of submicron organic aerosols during wintertime
 1095 inversions by advanced factor analysis of aerosol mass spectra, *Environ. Sci.*
 1096 *Tech.*, 42, 214-220, 2008.
 1097 Lanz, V. A., Prevot, A. S. H., Alfarra, M. R., Weimer, S., Mohr, C., DeCarlo, P. F.,
 1098 Gianini, M. F. D., Hueglin, C., Schneider, J., Favez, O., D'Anna, B., George, C.,
 1099 and Baltensperger, U.: Characterization of aerosol chemical composition with
 1100 aerosol mass spectrometry in Central Europe: an overview, *Atmo. Chem. Phys.*,
 1101 10, 10453-10471, 2010.
 1102 Lauer, A., Eyring, V., Hendricks, J., Joeckel, P., and Lohmann, U.: Global model
 1103 simulations of the impact of ocean-going ships on aerosols, clouds, and the
 1104 radiation budget, *Atmos. Chem. Phys.*, 7, 5061-5079, 2007.
 1105 Lee, L. A., Pringle, K. J., Reddington, C. L., Mann, G. W., Stier, P., Spracklen, D. V.,
 1106 Pierce, J. R., and Carslaw, K. S.: The magnitude and causes of uncertainty in
 1107 global model simulations of cloud condensation nuclei, *Atmos. Chem. Phys.*,
 1108 13, 8879-8914, 2013.
 1109 Lelieveld, J., Barlas, C., Giannadaki, D., and Pozzer, A.: Model calculated global,
 1110 regional and megacity premature mortality due to air pollution, *Atmos. Chem.*
 1111 *Phys.*, 13, 7023-7037, 2013.
 1112 Lelieveld, J., Evans, J. S., Fnais, M., Giannadaki, D., and Pozzer, A.: The contribution
 1113 of outdoor air pollution sources to premature mortality on a global scale,
 1114 *Nature*, 525, 367-+, 10.1038/nature15371, 2015.
 1115 Marcolli, C., Canagaratna, M. R., Worsnop, D. R., Bahreini, R., de Gouw, J. A.,
 1116 Warneke, C., Goldan, P. D., Kuster, W. C., Williams, E. J., Lerner, B. M.,
 1117 Roberts, J. M., Meagher, J. F., Fehsenfeld, F. C., Marchewka, M., Bertman, S.
 1118 B., and Middlebrook, A. M.: Cluster analysis of the organic peaks in bulk mass
 1119 spectra obtained during the 2002 New England air quality study with an
 1120 Aerodyne aerosol mass spectrometer, *Atmo. Chem. Phys.*, 6, 5649-5666, 2006.
 1121 May, A. A., Levin, E. J. T., Hennigan, C. J., Riipinen, I., Lee, T., Collett, J. L.,
 1122 Jimenez, J. L., Kreidenweis, S. M., and Robinson, A. L.: Gas-particle

1123 partitioning of primary organic aerosol emissions: 3. Biomass burning, *J.*
1124 *Geophys. Res. Atmos.*, 118, 11327-11338, doi: 10.1002/jgrd.50828, 2013.

1125 May, A. A., Nguyen, N. T., Presto, A. A., Gordon, T. D., Lipsky, E. M., Karve, M.,
1126 Gutierrez, A., Robertson, W. H., Zhang, M., Brandow, C., Chang, O., Chen, S.,
1127 Cicero-Fernandez, P., Dinkins, L., Fuentes, M., Huang, S.-M., Ling, R., Long,
1128 J., Maddox, C., Massetti, J., McCauley, E., Miguel, A., Na, K., Ong, R., Pang,
1129 Y., Rieger, P., Sax, T., Tin, T., Thu, V., Chattopadhyay, S., Maldonado, H.,
1130 Maricq, M. M., and Robinson, A. L.: Gas- and particle-phase primary emissions
1131 from in-use, on-road gasoline and diesel vehicles, *Atmospheric Environment*,
1132 88, 247-260, 10.1016/j.atmosenv.2014.01.046, 2014.

1133 McFiggans, G., Artaxo, P., Baltensperger, U., Coe, H., Facchini, M. C., Feingold, G.,
1134 Fuzzi, S., Gysel, M., Laaksonen, A., Lohmann, U., Mentel, T. F., Murphy, D.
1135 M., O'Dowd, C. D., Snider, J. R., and Weingartner, E.: The effect of physical
1136 and chemical aerosol properties on warm cloud droplet activation, *Atmo. Chem.*
1137 *Phys.*, 6, 2593-2649, 2006.

1138 Miracolo, M. A., Hennigan, C. J., Ranjan, M., Nguyen, N. T., Gordon, T. D., Lipsky,
1139 E. M., Presto, A. A., Donahue, N. M., and Robinson, A. L.: Secondary aerosol
1140 formation from photochemical aging of aircraft exhaust in a smog chamber,
1141 *Atmo. Chem. Phys.*, 11, 4135-4147, 2011.

1142 Mohr, C., DeCarlo, P. F., Heringa, M. F., Chirico, R., Slowik, J. G., Richter, R.,
1143 Reche, C., Alastuey, A., Querol, X., Seco, R., Penuelas, J., Jimenez, J. L.,
1144 Crippa, M., Zimmermann, R., Baltensperger, U., and Prevot, A. S. H.:
1145 Identification and quantification of organic aerosol from cooking and other
1146 sources in Barcelona using aerosol mass spectrometer data, *Atmo. Chem. Phys.*,
1147 12, 1649-1665, 2012.

1148 Murphy, B. N., and Pandis, S. N.: Simulating the formation of semivolatile primary
1149 and secondary organic aerosol in a regional chemical transport model, *Environ.*
1150 *Sci. Technol.*, 43, 4722-4728, 2009.

1151 Nemitz, E., Jimenez, J. L., Huffman, J. A., Ulbrich, I. M., Canagaratna, M. R.,
1152 Worsnop, D. R., and Guenther, A. B.: An eddy-covariance system for the
1153 measurement of surface/atmosphere exchange fluxes of submicron aerosol
1154 chemical species - First application above an urban area, *Aerosol Sci. Tech.*, 42,
1155 636-657, 2008.

1156 Ng, N. L., Canagaratna, M. R., Jimenez, J. L., Zhang, Q., Ulbrich, I. M., and
1157 Worsnop, D. R.: Real-Time Methods for Estimating Organic Component Mass
1158 Concentrations from Aerosol Mass Spectrometer Data, *Environ. Sci. Tech.*, 45,
1159 910-916, 2011.

1160 Paatero, P., and Tapper, U.: Positive matrix factorization-A nonnegative factor model
1161 with optimal utilization of error-estimates of data values, *Environmetrics*, 5,
1162 111-126, 1994.

1163 Paatero, P.: Least squares formulation of robust non-negative factor analysis,
1164 *Chemometrics and Intelligent Laboratory Systems*, 37, 23-35, 1997.

1165 Poschl, U.: Atmospheric aerosols: Composition, transformation, climate and health
1166 effects, *Angew. Chem.-Int. Edit.*, 44, 7520-7540, 2005.

1167 Pozzer, A., Zimmermann, P., Doering, U. M., van Aardenne, J., Tost, H., Dentener,
1168 F., Janssens-Maenhout, G., and Lelieveld, J.: Effects of business-as-usual
1169 anthropogenic emissions on air quality, *Atmos. Chem. Phys.*, 12, 6915-6937,
1170 10.5194/acp-12-6915-2012, 2012a.

1171 Pozzer, A., de Meij, A., Pringle, K. J., Tost, H., Doering, U. M., van Aardenne, J., and
1172 Lelieveld, J.: Distributions and regional budgets of aerosols and their precursors
1173 simulated with the EMAC chemistry-climate model, *Atmos. Chem. Phys.*, 12,
1174 961-987, 2012b.

1175 Pringle, K. J., Tost, H., Message, S., Steil, B., Giannadaki, D., Nenes, A., Fountoukis,
1176 C., Stier, P., Vignati, E., and Lelieveld, J.: Description and evaluation of GMXe:
1177 a new aerosol submodel for global simulations (v1), *Geoscientific Model
1178 Development*, 3, 391-412, 2010.

1179 Pye, H. O. T., and Seinfeld, J. H.: A global perspective on aerosol from low-volatility
1180 organic compounds, *Atmos. Chem. Phys.*, 10, 4377-4401, 2010.

1181 Ranjan, M., Presto, A. A., May, A. A., and Robinson, A. L.: Temperature
1182 Dependence of Gas-Particle Partitioning of Primary Organic Aerosol Emissions
1183 from a Small Diesel Engine, *Aerosol Science and Technology*, 46, 13-21,
1184 10.1080/02786826.2011.602761, 2012.

1185 Robinson, A. L., Donahue, N. M., Shrivastava, M. K., Weitkamp, E. A., Sage, A. M.,
1186 Grieshop, A. P., Lane, T. E., Pierce, J. R., and Pandis, S. N.: Rethinking organic
1187 aerosols: Semivolatile emissions and photochemical aging, *Science*, 315, 1259-
1188 1262, 2007.

1189 Robinson, A. L., Grieshop, A. P., Donahue, N. M., and Hunt, S. W.: Updating the
1190 conceptual model for fine particle mass emissions from combustion systems, *J.
1191 Air Waste Manage.*, 60, 1204-1222, 2010.

1192 Samy, S., and Zielinska, B.: Secondary organic aerosol production from modern
1193 diesel engine emissions, *Atmo. Chem. Phys.*, 10, 609-625, 2010.

1194 Sander, R., Baumgaertner, A., Gromov, S., Harder, H., Joeckel, P., Kerkweg, A.,
1195 Kubistin, D., Regelin, E., Riede, H., Sandu, A., Taraborrelli, D., Tost, H., and
1196 Xie, Z. Q.: The atmospheric chemistry box model CAABA/MECCA-3.0,
1197 *Geoscientific Model Development*, 4, 373-380, 2011.

1198 Shrivastava, M. K., Lane, T. E., Donahue, N. M., Pandis, S. N., and Robinson, A. L.:
1199 Effects of gas particle partitioning and aging of primary emissions on urban and
1200 regional organic aerosol concentrations, *J. Geophys. Res. Atmos.*, 113, doi:
1201 10.1029/2007jd009735, 2008.

1202 Shrivastava, M., Fast, J., Easter, R., Gustafson, W. I., Jr., Zaveri, R. A., Jimenez, J. L.,
1203 Saide, P., and Hodzic, A.: Modeling organic aerosols in a megacity: comparison
1204 of simple and complex representations of the volatility basis set approach,
1205 *Atmos. Chem. Phys.*, 11, 6639-6662, 2011.

1206 Shrivastava, M., Easter, R. C., Liu, X., Zelenyuk, A., Singh, B., Zhang, K., Ma, P.-L.,
1207 Chand, D., Ghan, S., Jimenez, J. L., Zhang, Q., Fast, J., Rasch, P. J., and Tiitta,
1208 P.: Global transformation and fate of SOA: Implications of low-volatility SOA
1209 and gas-phase fragmentation reactions, *Journal of Geophysical Research:
1210 Atmospheres*, 120, 4169-4195, 10.1002/2014JD022563, 2015.

1211 Spracklen, D. V., Jimenez, J. L., Carslaw, K. S., Worsnop, D. R., Evans, M. J., Mann,
1212 G. W., Zhang, Q., Canagaratna, M. R., Allan, J., Coe, H., McFiggans, G., Rap,
1213 A., and Forster, P.: Aerosol mass spectrometer constraint on the global
1214 secondary organic aerosol budget, *Atmo. Chem. Phys.*, 11, 12109-12136, 2011.

1215 Stone, E. A., Zhou, J., Snyder, D. C., Rutter, A. P., Mieritz, M., and Schauer, J. J.: A
1216 Comparison of Summertime Secondary Organic Aerosol Source Contributions
1217 at Contrasting Urban Locations, *Environ. Sci. Tech.*, 43, 3448-3454, 2009.

1218 Stubenrauch, C. J., Chedin, A., Radel, G., Scott, N. A., and Serrar, S.: Cloud
 1219 properties and their seasonal and diurnal variability from TOVS path-B, *Journal*
 1220 *of Climate*, 19, 5531-5553, 10.1175/jcli3929.1, 2006.
 1221 Suess, D. T., and Prather, K. A.: Mass spectrometry of aerosols, *Chemical Reviews*,
 1222 99, 3007-+, 1999.
 1223 Sun, Y. L., Zhang, Q., Schwab, J. J., Demerjian, K. L., Chen, W. N., Bae, M. S.,
 1224 Hung, H. M., Hogrefe, O., Frank, B., Rattigan, O. V., and Lin, Y. C.:
 1225 Characterization of the sources and processes of organic and inorganic aerosols
 1226 in New York city with a high-resolution time-of-flight aerosol mass
 1227 spectrometer, *Atmospheric Chemistry and Physics*, 11, 1581-1602,
 1228 10.5194/acp-11-1581-2011, 2011.
 1229 Takegawa, N., Miyazaki, Y., Kondo, Y., Komazaki, Y., Miyakawa, T., Jimenez, J. L.,
 1230 Jayne, J. T., Worsnop, D. R., Allan, J. D., and Weber, R. J.: Characterization of
 1231 an Aerodyne Aerosol Mass Spectrometer (AMS): Intercomparison with other
 1232 aerosol instruments, *Aerosol Sci. Tech.*, 39, 760-770, 2005.
 1233 Tost, H., Jockel, P. J., Kerkweg, A., Sander, R., and Lelieveld, J.: Technical note: A
 1234 new comprehensive SCAVenging submodel for global atmospheric chemistry
 1235 modelling, *Atmos. Chem. Phys.*, 6, 565-574, 2006.
 1236 Tsigaridis, K., Daskalakis, N., Kanakidou, M., Adams, P. J., Artaxo, P., Bahadur, R.,
 1237 Balkanski, Y., Bauer, S. E., Bellouin, N., Benedetti, A., Bergman, T., Bernsten,
 1238 T. K., Beukes, J. P., Bian, H., Carslaw, K. S., Chin, M., Curci, G., Diehl, T.,
 1239 Easter, R. C., Ghan, S. J., Gong, S. L., Hodzic, A., Hoyle, C. R., Iversen, T.,
 1240 Jathar, S., Jimenez, J. L., Kaiser, J. W., Kirkevåg, A., Koch, D., Kokkola, H.,
 1241 Lee, Y. H., Lin, G., Liu, X., Luo, G., Ma, X., Mann, G. W., Mihalopoulos, N.,
 1242 Morcrette, J. J., Mueller, J. F., Myhre, G., Myriokefalitakis, S., Ng, N. L.,
 1243 O'Donnell, D., Penner, J. E., Pozzoli, L., Pringle, K. J., Russell, L. M., Schulz,
 1244 M., Sciare, J., Seland, O., Shindell, D. T., Sillman, S., Skeie, R. B., Spracklen,
 1245 D., Stavroukou, T., Steenrod, S. D., Takemura, T., Tiitta, P., Tilmes, S., Tost, H.,
 1246 van Noije, T., van Zyl, P. G., von Salzen, K., Yu, F., Wang, Z., Wang, Z.,
 1247 Zaveri, R. A., Zhang, H., Zhang, K., Zhang, Q., and Zhang, X.: The AeroCom
 1248 evaluation and intercomparison of organic aerosol in global models, *Atmo.*
 1249 *Chem. Phys.*, 14, 10845-10895, 2014.
 1250 Tsimpidi, A. P., Karydis, V. A., Zavala, M., Lei, W., Molina, L., Ulbrich, I. M.,
 1251 Jimenez, J. L., and Pandis, S. N.: Evaluation of the volatility basis-set approach
 1252 for the simulation of organic aerosol formation in the Mexico City metropolitan
 1253 area, *Atmos. Chem. Phys.*, 10, 525-546, 2010.
 1254 Tsimpidi, A. P., Karydis, V. A., Zavala, M., Lei, W., Bei, N., Molina, L., and Pandis,
 1255 S. N.: Sources and production of organic aerosol in Mexico City: insights from
 1256 the combination of a chemical transport model (PMCAMx-2008) and
 1257 measurements during MILAGRO, *Atmos. Chem. Phys.*, 11, 5153-5168, 2011.
 1258 Tsimpidi, A. P., Karydis, V. A., Pozzer, A., Pandis, S. N., and Lelieveld, J.: ORACLE
 1259 (v1.0): module to simulate the organic aerosol composition and evolution in the
 1260 atmosphere, *Geoscientific Model Development*, 7, 3153-3172, 2014.
 1261 Ulbrich, I. M., Canagaratna, M. R., Zhang, Q., Worsnop, D. R., and Jimenez, J. L.:
 1262 Interpretation of organic components from Positive Matrix Factorization of
 1263 aerosol mass spectrometric data, *Atmo. Chem. Phys.*, 9, 2891-2918, 2009.
 1264 van der Werf, G. R., Randerson, J. T., Giglio, L., Collatz, G. J., Mu, M., Kasibhatla,
 1265 P. S., Morton, D. C., DeFries, R. S., Jin, Y., and van Leeuwen, T. T.: Global fire

1266 emissions and the contribution of deforestation, savanna, forest, agricultural,
1267 and peat fires (1997-2009), *Atmos. Chem. Phys.*, 10, 11707-11735, 2010.

1268 Wang, Q., Shao, M., Liu, Y., William, K., Paul, G., Li, X., Liu, Y., and Lu, S.: Impact
1269 of biomass burning on urban air quality estimated by organic tracers:
1270 Guangzhou and Beijing as cases, *Atmo. Environ.*, 41, 8380-8390, doi:
1271 10.1016/j.atmosenv.2007.06.048, 2007.

1272 Weilenmann, M., Favez, J.-Y., and Alvarez, R.: Cold-start emissions of modern
1273 passenger cars at different low ambient temperatures and their evolution over
1274 vehicle legislation categories, *Atmos. Environ.*, 43, 2419-2429, 2009.

1275 Westerholm, R., Christensen, A., and Rosen, A.: Regulated and unregulated exhaust
1276 emissions from two three-way catalyst equipped gasoline fuelled vehicles,
1277 *Atmos. Environ.*, 30, 3529-3536, 1996.

1278 Zhang, Q., Alfarra, M. R., Worsnop, D. R., Allan, J. D., Coe, H., Canagaratna, M. R.,
1279 and Jimenez, J. L.: Deconvolution and quantification of hydrocarbon-like and
1280 oxygenated organic aerosols based on aerosol mass spectrometry, *Environ. Sci.
1281 Tech.*, 39, 4938-4952, 2005a.

1282 Zhang, Q., Canagaratna, M. R., Jayne, J. T., Worsnop, D. R., and Jimenez, J. L.:
1283 Time- and size-resolved chemical composition of submicron particles in
1284 Pittsburgh: Implications for aerosol sources and processes, *J. Geophys. Res.-
1285 Atmos.*, 110, doi: 10.1029/2004jd004649, 2005b.

1286 Zhang, Q., Jimenez, J. L., Canagaratna, M. R., Allan, J. D., Coe, H., Ulbrich, I.,
1287 Alfarra, M. R., Takami, A., Middlebrook, A. M., Sun, Y. L., Dzepina, K.,
1288 Dunlea, E., Docherty, K., DeCarlo, P. F., Salcedo, D., Onasch, T., Jayne, J. T.,
1289 Miyoshi, T., Shimo, A., Hatakeyama, S., Takegawa, N., Kondo, Y.,
1290 Schneider, J., Drewnick, F., Borrmann, S., Weimer, S., Demerjian, K.,
1291 Williams, P., Bower, K., Bahreini, R., Cottrell, L., Griffin, R. J., Rautiainen, J.,
1292 Sun, J. Y., Zhang, Y. M., and Worsnop, D. R.: Ubiquity and dominance of
1293 oxygenated species in organic aerosols in anthropogenically-influenced
1294 Northern Hemisphere midlatitudes, *Geophys. Res. Lett.*, 34, doi:
1295 10.1029/2007gl029979, 2007.

1296 Zhang, Q., Jimenez, J. L., Canagaratna, M. R., Ulbrich, I. M., Ng, N. L., Worsnop, D.
1297 R., and Sun, Y. L.: Understanding atmospheric organic aerosols via factor
1298 analysis of aerosol mass spectrometry: a review, *Anal. Bioanal. Chem.*, 401,
1299 3045-3067, 2011.

1300 Zhang, Q. J., Beekmann, M., Drewnick, F., Freutel, F., Schneider, J., Crippa, M.,
1301 Prevot, A. S. H., Baltensperger, U., Poulain, L., Wiedensohler, A., Sciare, J.,
1302 Gros, V., Borbon, A., Colomb, A., Michoud, V., Doussin, J. F., van der Gon, H.
1303 A. C. D., Haefelin, M., Dupont, J. C., Siour, G., Petetin, H., Bessagnet, B.,
1304 Pandis, S. N., Hodzic, A., Sanchez, O., Honore, C., and Perrussel, O.:
1305 Formation of organic aerosol in the Paris region during the MEGAPOLI
1306 summer campaign: evaluation of the volatility-basis-set approach within the
1307 CHIMERE model, *Atmos. Chem. Phys.*, 13, 5767-5790, 2013.

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1312 **Table 1.** Predicted tropospheric burden in Tg of organic aerosol components during the
1313 decade 2001-2010.

OA component	Tropospheric burden (Tg)	Monthly Standard Deviation (σ)
fPOA	0.06	0.01
fSOA	0.57	0.06
bbPOA	0.18	0.13
bbSOA	0.42	0.27
aSOA	0.44	0.08
bSOA	0.31	0.10
OA	1.98	0.54

1314

1315 **Table 2.** Statistical evaluation of EMAC results against AMS measurements over
 1316 urban locations of the Northern Hemisphere during 2001-2010.
 1317

EMAC Element	AMS Element	Number of datasets	Mean Observed ($\mu\text{g m}^{-3}$)	Mean Predicted ($\mu\text{g m}^{-3}$)	MAGE ($\mu\text{g m}^{-3}$)	MB ($\mu\text{g m}^{-3}$)	NME (%)	NMB (%)	RMSE ($\mu\text{g m}^{-3}$)
POA*	HOA+BBOA	23	2.70	0.98	1.73	-1.72	64	-64	2.58
SOA	OOA	23	4.25	2.85	1.97	-1.40	46	-33	2.50
Aged SOA	LV-OOA	10	3.43	2.72	1.47	-0.72	43	-21	2.04
Fresh SOA	SV-OOA	10	2.14	1.88	0.69	-0.26	32	-12	0.81

1318

1319 * Sum of fPOA and bbPOA

1320

1321 **Table 3.** Statistical evaluation of EMAC POA (sum of fPOA and bbPOA) against
 1322 AMS POA (sum of HOA and BBOA) in the Northern Hemisphere during 2001-2010.

1323

Site Type^a	Number of datasets	Mean Observed ($\mu\text{g m}^{-3}$)	Mean Predicted ($\mu\text{g m}^{-3}$)	MAGE ($\mu\text{g m}^{-3}$)	MB ($\mu\text{g m}^{-3}$)	NME (%)	NMB (%)	RMSE ($\mu\text{g m}^{-3}$)
Urban Downwind	15	0.82	0.64	0.38	-0.18	47	-22	0.50
Rural/Remote	46	0.43	0.47	0.37	0.04	87	9	0.5
Continent^b								
Europe	42	0.61	0.47	0.36	-0.14	59	-23	0.47
N. America	10	0.51	0.50	0.29	-0.01	57	-3	0.37
Asia	9	0.15	0.69	0.54	0.54	363	363	0.72
Season^c								
Winter	6	1.18	0.74	0.60	-0.44	51	-37	0.76
Spring	30	0.42	0.53	0.41	0.11	97	26	0.52
Summer	14	0.50	0.44	0.30	-0.06	59	-13	0.39
Autumn	11	0.49	0.42	0.27	-0.07	54	-15	0.37
Total	61	0.53	0.51	0.38	-0.02	71	-3	0.50

1324 ^a Statistics are calculated for a specific site type during all seasons

1325 ^b Statistics are calculated for a specific continent excluding the values from urban areas

1326 ^c Statistics are calculated for a specific season excluding the values from urban areas

1327

1328 **Table 4.** Statistical evaluation of EMAC SOA against AMS OOA in the Northern
 1329 Hemisphere during 2001-2010.

1330

Site Type^a	Number of datasets	Mean Observed ($\mu\text{g m}^{-3}$)	Mean Predicted ($\mu\text{g m}^{-3}$)	MAGE ($\mu\text{g m}^{-3}$)	MB ($\mu\text{g m}^{-3}$)	NME (%)	NMB (%)	RMSE ($\mu\text{g m}^{-3}$)
Urban Downwind	15	2.98	2.07	1.20	-0.91	40	-30	1.77
Rural/Remote	46	2.72	1.86	1.45	-0.86	54	-32	2.09
Continent^b								
Europe	42	2.47	1.49	1.59	-0.98	64	-39	2.28
N. America	10	3.29	2.78	0.91	-0.51	28	-15	1.37
Asia	9	3.68	2.89	1.00	-0.79	27	-22	1.11
Season^c								
Winter	6	2.81	0.50	2.31	-2.31	82	-82	2.65
Spring	30	2.22	1.79	0.97	-0.43	44	-20	1.18
Summer	14	4.30	2.89	2.04	-1.41	47	-33	3.20
Autumn	11	2.35	1.78	1.22	-0.57	52	-25	1.39
Total	61	2.78	1.91	1.39	-0.87	50	-31	2.02

1331 ^a Statistics are calculated for a specific site type during all four seasons

1332 ^b Statistics are calculated for a specific continent excluding the values from urban areas

1333 ^c Statistics are calculated for a specific season excluding the values from urban areas

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1335 **Table 5.** Statistical evaluation of EMAC aged SOA against AMS LV-OOA in the
 1336 Northern Hemisphere during 2001-2010.

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Site Type^a	Number of datasets	Mean Observed ($\mu\text{g m}^{-3}$)	Mean Predicted ($\mu\text{g m}^{-3}$)	MAGE ($\mu\text{g m}^{-3}$)	MB ($\mu\text{g m}^{-3}$)	NME (%)	NMB (%)	RMSE ($\mu\text{g m}^{-3}$)
Urban Downwind	8	1.77	0.94	1.28	-0.83	72	-47	1.55
Rural/Remote	33	1.65	1.02	1.17	-0.63	71	-38	1.69
Continent^b								
Europe	35	1.71	0.98	1.24	-0.73	73	-43	0.47
N. America	6	1.45	1.17	0.87	-0.28	60	-20	1.00
Asia	-	-	-	-	-	-	-	-
Season^c								
Winter	3	2.36	0.20	2.16	-2.16	91	-91	2.36
Spring	18	1.06	0.81	0.82	-0.25	77	-24	1.03
Summer	11	2.64	1.55	1.79	-1.09	68	-41	2.47
Autumn	9	1.49	1.01	0.89	-0.48	59	-32	1.10
Total	41	1.68	1.01	1.19	-0.67	71	-40	1.67

1338 ^a Statistics are calculated for a specific site type during all four seasons

1339 ^b Statistics are calculated for a specific continent excluding the values from urban areas

1340 ^c Statistics are calculated for a specific season excluding the values from urban areas

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1342 **Table 6.** Statistical evaluation of EMAC fresh SOA against AMS SV-OOA in the
 1343 Northern Hemisphere during 2001-2010.

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Site Type^a	Number of datasets	Mean Observed ($\mu\text{g m}^{-3}$)	Mean Predicted ($\mu\text{g m}^{-3}$)	MAGE ($\mu\text{g m}^{-3}$)	MB ($\mu\text{g m}^{-3}$)	NME (%)	NMB (%)	RMSE ($\mu\text{g m}^{-3}$)
Urban Downwind	8	0.81	0.64	0.41	-0.17	51	-21	0.76
Rural/Remote	33	1.03	0.70	0.64	-0.33	62	-32	0.85
Continent^b								
Europe	35	0.90	0.63	0.56	-0.27	62	-30	0.81
N. America	6	1.51	1.07	0.80	-0.44	53	-29	0.96
Asia	-	-	-	-	-	-	-	-
Season^c								
Winter	3	0.87	0.18	0.69	-0.69	79	-79	0.76
Spring	18	0.54	0.37	0.46	-0.17	86	-31	0.60
Summer	11	1.89	1.22	0.96	-0.67	51	-36	1.27
Autumn	9	0.83	0.86	0.39	0.03	47	4	0.52
Total	41	0.99	0.69	0.60	-0.30	60	-30	0.83

1345 ^a Statistics are calculated for a specific site type during all four seasons

1346 ^b Statistics are calculated for a specific continent excluding the values from urban areas

1347 ^c Statistics are calculated for a specific season excluding the values from urban areas

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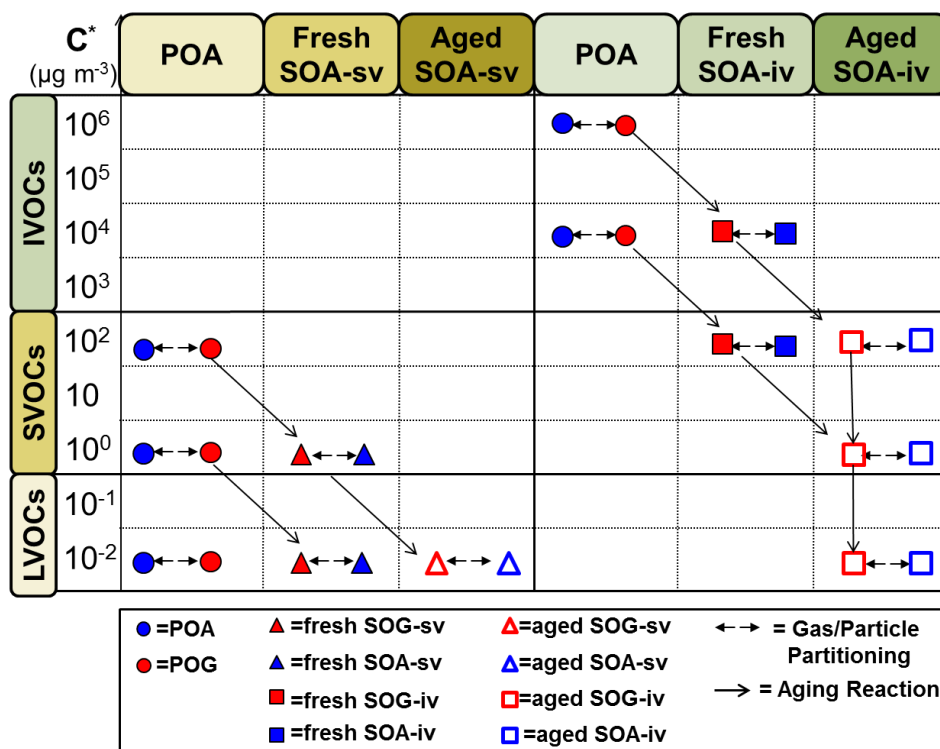
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1361 **Figure 1:** Schematic of the VBS resolution and the formation of SOA from SVOC

1362 and IVOC emissions. Red indicates the vapor phase and blue the particulate phase.

1363 The circles represent primary organic material that can be emitted either in the gas or

1364 in the aerosol phase. Filled triangles and squares indicate the formation of SOA from

1365 SVOCs and IVOCs, respectively, by fuel combustion and biomass burning sources

1366 from the first oxidation step (fresh SOA). Open triangles and squares represent SOA

1367 formed in additional oxidation steps (aged SOA) from SVOCs and IVOCs by the

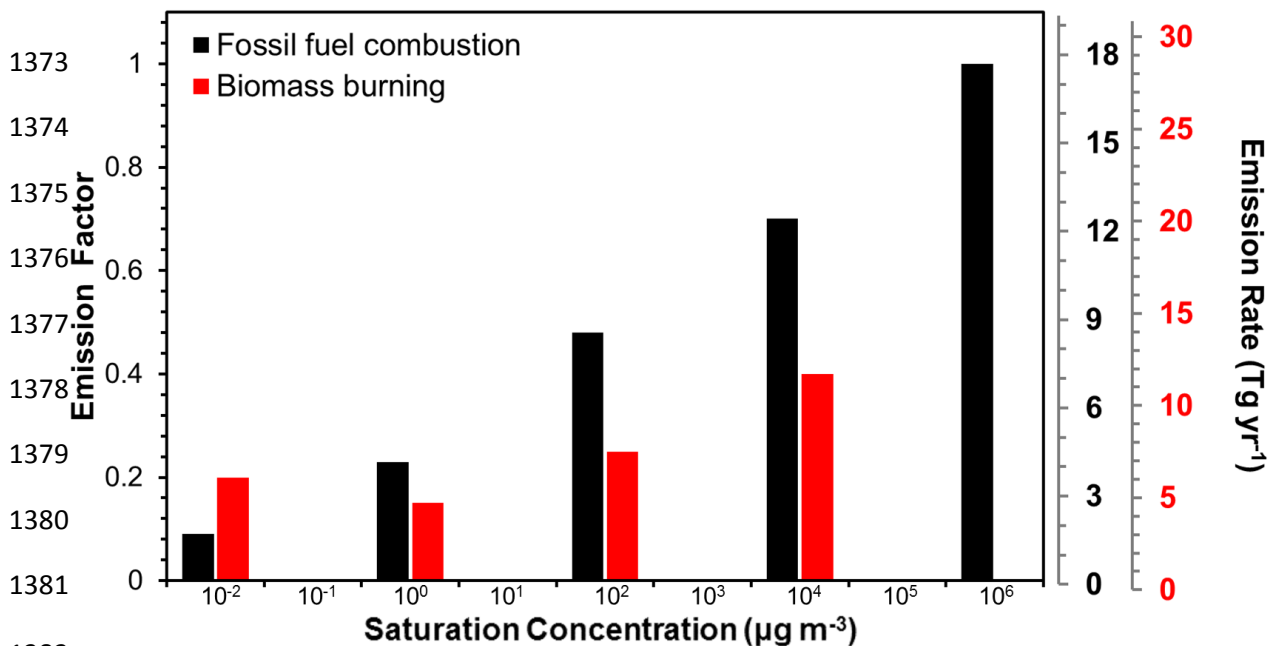
1368 same sources. The partitioning processes, the aging reactions of the organic

1369 compounds, and the names of the species used to track all compounds are also shown.

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1383 **Figure 2.** Volatility distribution for fuel combustion (in black) and
1384 biomass burning (in red) organic emissions. The emission factors for
1385 fuel combustion emissions are derived from Robinson et al. (2007)
1386 while for biomass burning POA emissions are from May et al. (2013)
1387 (shown in the primary y-axis). The corresponding emission rates are
1388 also shown in the secondary y-axis.

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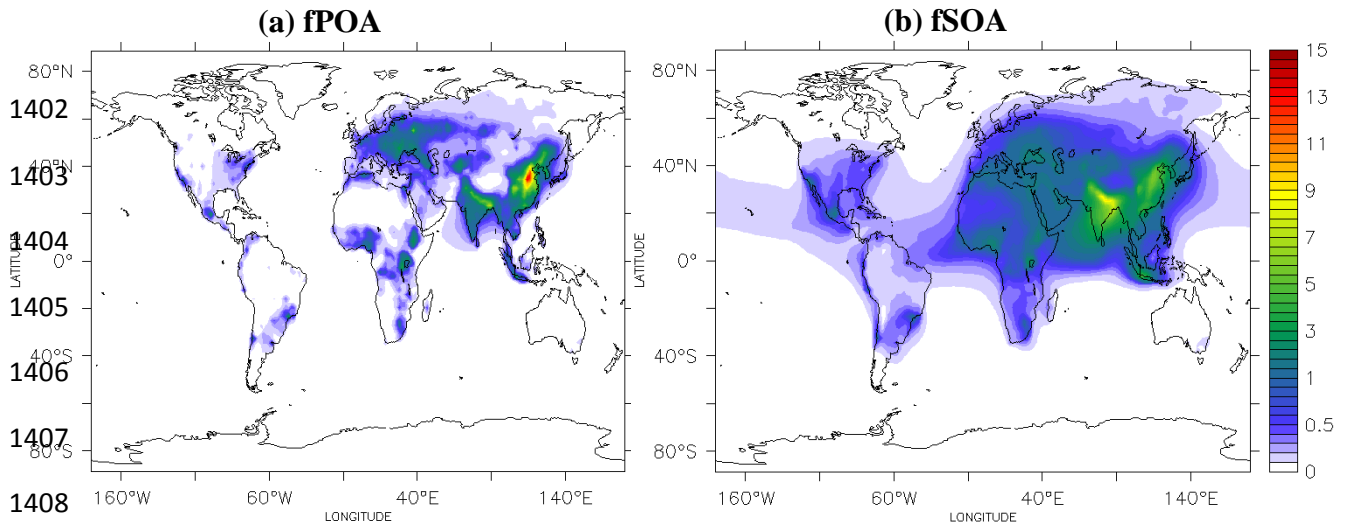
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1398 **Figure 3:** Location of the field measurement campaigns used for evaluating the model
1399 during 2001-2010. Urban, urban downwind and rural/remote areas are represented by
1400 red, blue, and green colors respectively.

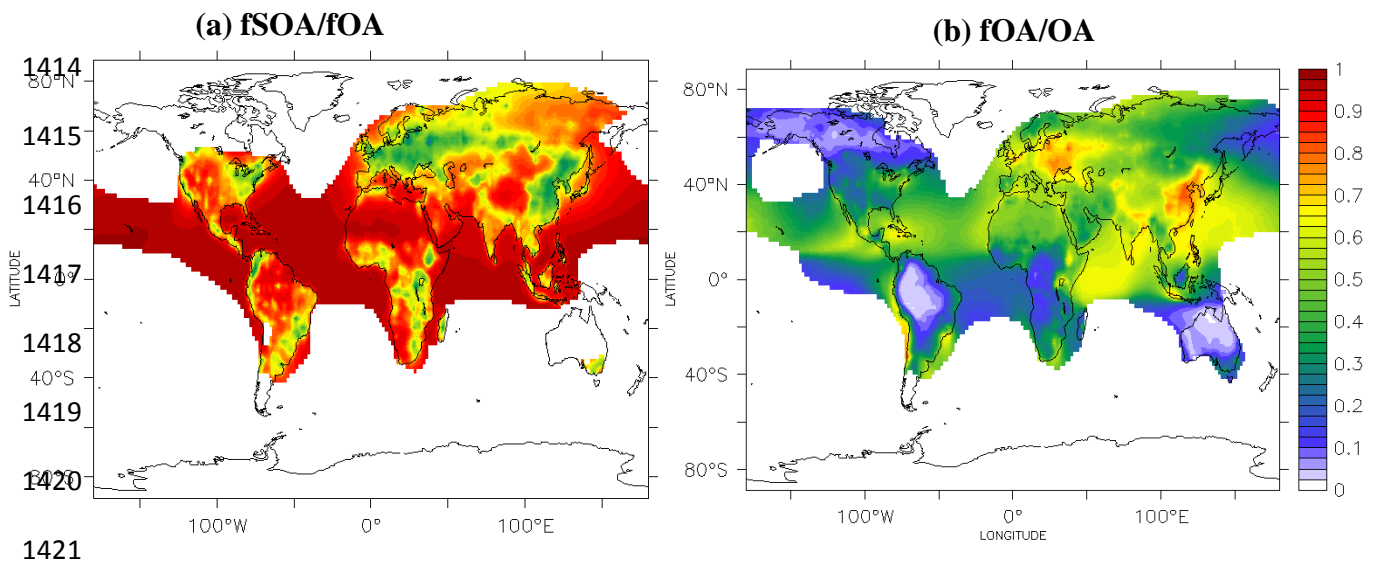
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1410 **Figure 4:** Predicted average surface concentrations (in $\mu\text{g m}^{-3}$) of: (a) POA from fuel
 1411 combustion sources (fPOA) and (b) SOA from the oxidation of SVOCs and IVOCs
 1412 from fuel combustion sources (fSOA) during the years 2001-2010.

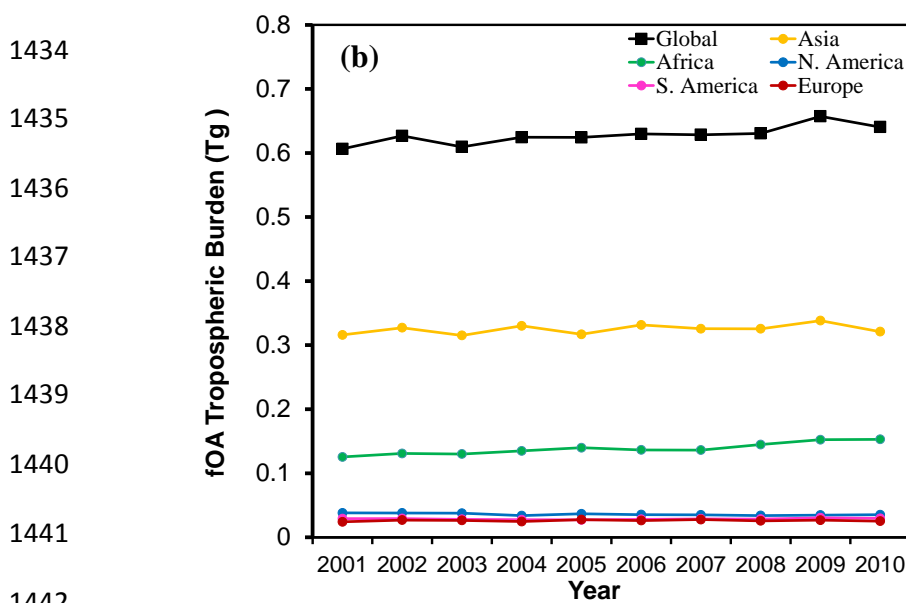
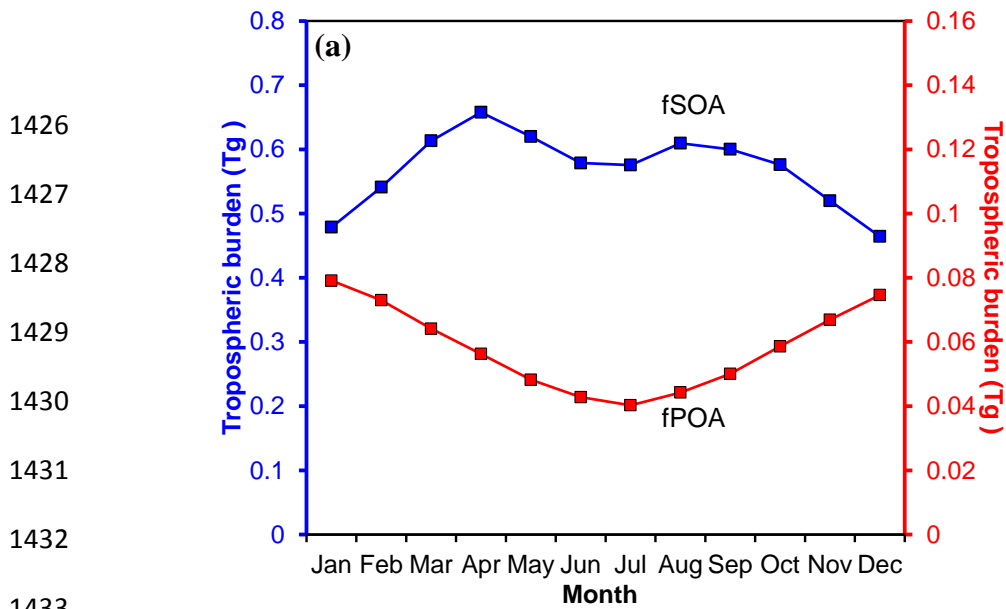
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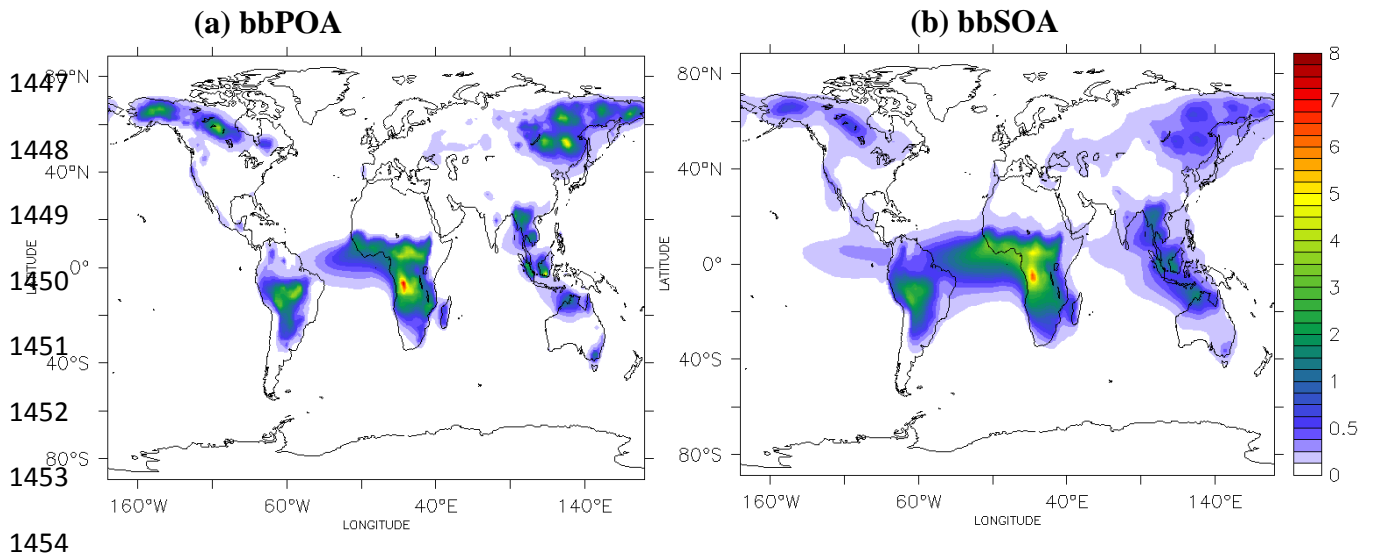
1422 **Figure 5:** Predicted ratio of (a) fuel combustion SOA (fSOA) to total fuel combustion
 1423 OA (sum of fPOA and fSOA) and (b) fuel combustion OA to total OA (sum of fOA,
 1424 bbOA, aSOA, and bSOA) during the years 2001-2010.

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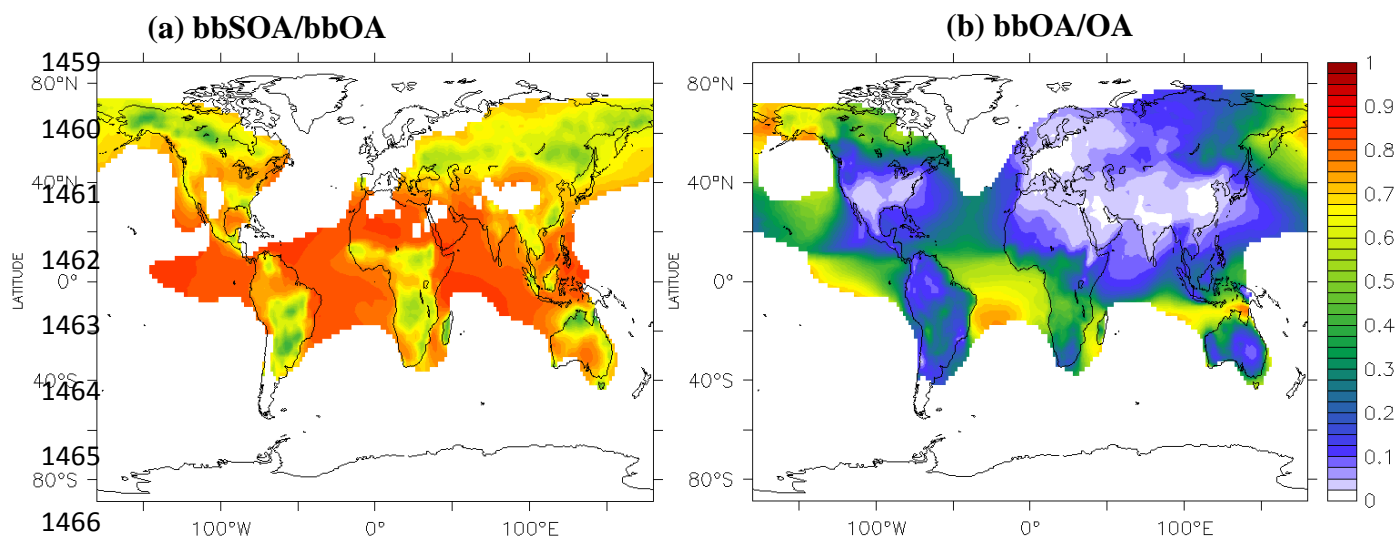
1443 **Figure 6:** (a) Average predicted tropospheric burden (Tg) of fSOA (in blue, primary y-
 1444 axis) and fPOA (in red, secondary y-axis) and (b) annually averaged tropospheric
 1445 burden of total fuel combustion OA (fOA) during 2001-2010.

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 1455 **Figure 7:** Predicted average surface concentrations (in $\mu\text{g m}^{-3}$) of: **(a)** POA from
 1456 biomass burning sources (bbPOA) and **(b)** SOA from the oxidation of SVOCs and
 1457 IVOCs from biomass burning sources (bbSOA) during the years 2001-2010.

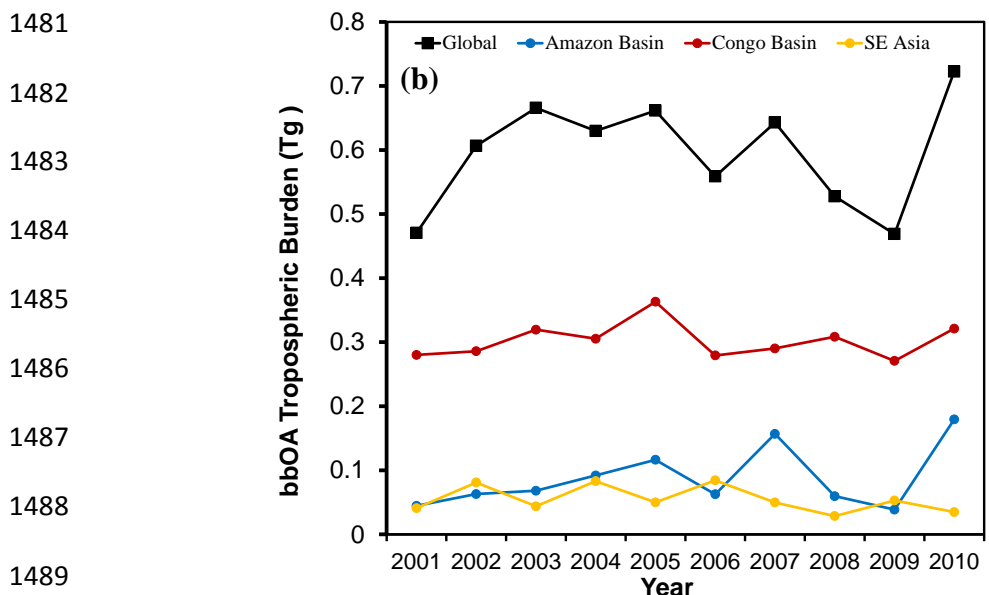
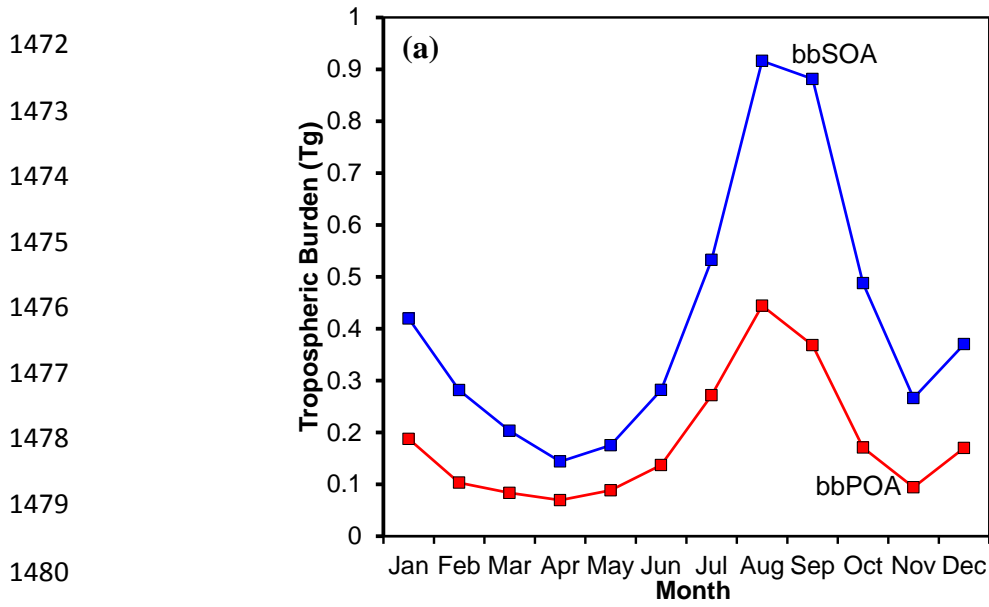
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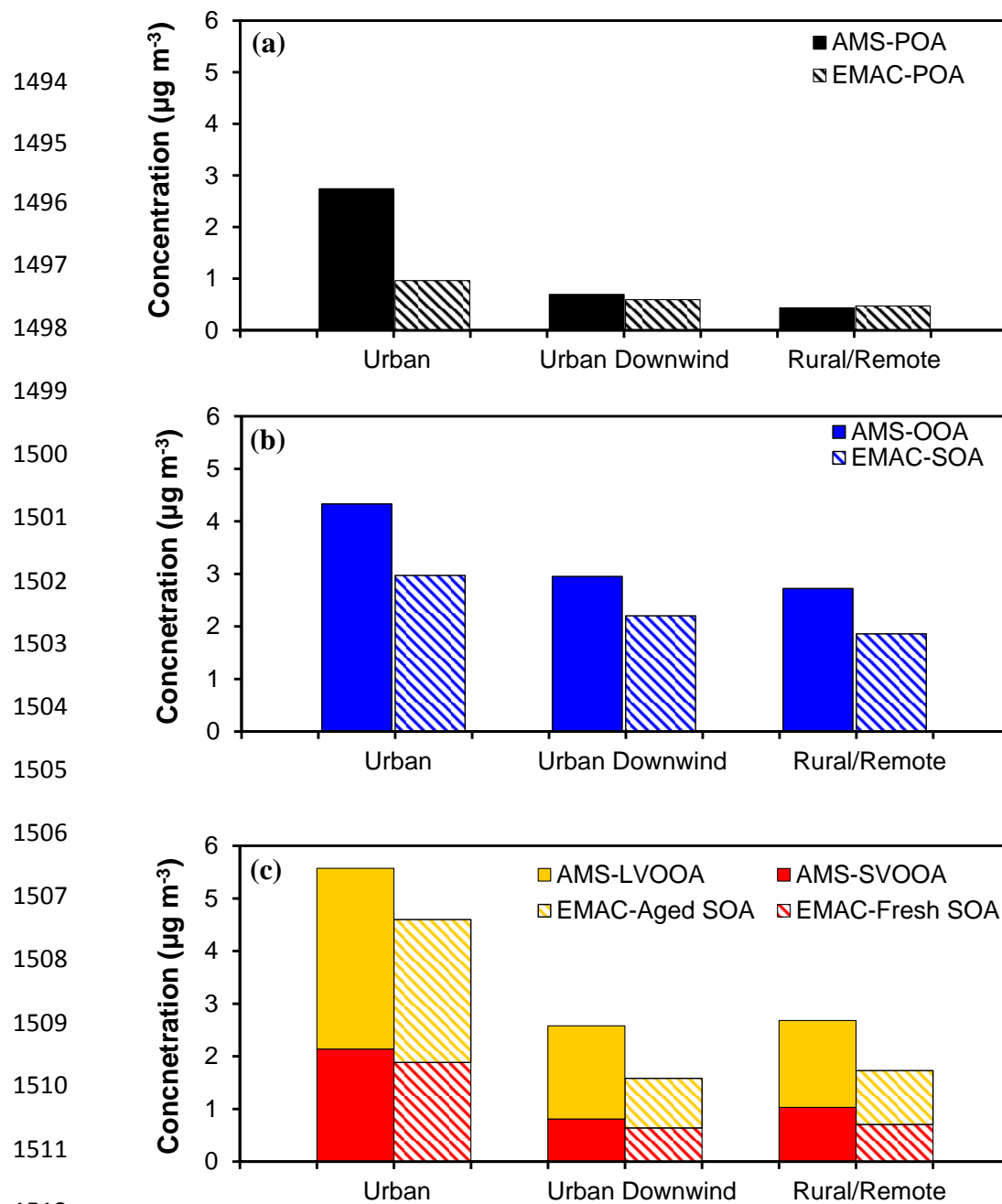
1468 **Figure 8:** Predicted ratio of (a) biomass burning SOA (bbSOA) to total biomass
 1469 burning OA (sum of bbPOA and bbSOA) and (b) biomass burning OA to total OA
 1470 (sum of fOA, bbOA, aSOA, and bSOA) during the years 2001-2010.

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1490 **Figure 9:** (a) Monthly averaged predicted tropospheric burden (Tg) of bbSOA (in
 1491 blue) and bbPOA (in red) and (b) Annual average tropospheric burden of total
 1492 biomass burning OA (bbOA) during 2001-2010.

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1513 **Figure 10:** Comparison of average (a) EMAC predicted POA to AMS-POA (sum of
 1514 AMS-HOA and AMS-BBOA) (b) EMAC predicted SOA to AMS-OOA, and, (c)
 1515 EMAC predicted fresh and aged SOA to AMS-SVOOA and AMS-LVOOA from 84
 1516 data sets over urban, urban downwind and rural/remote areas during 2001-2010.

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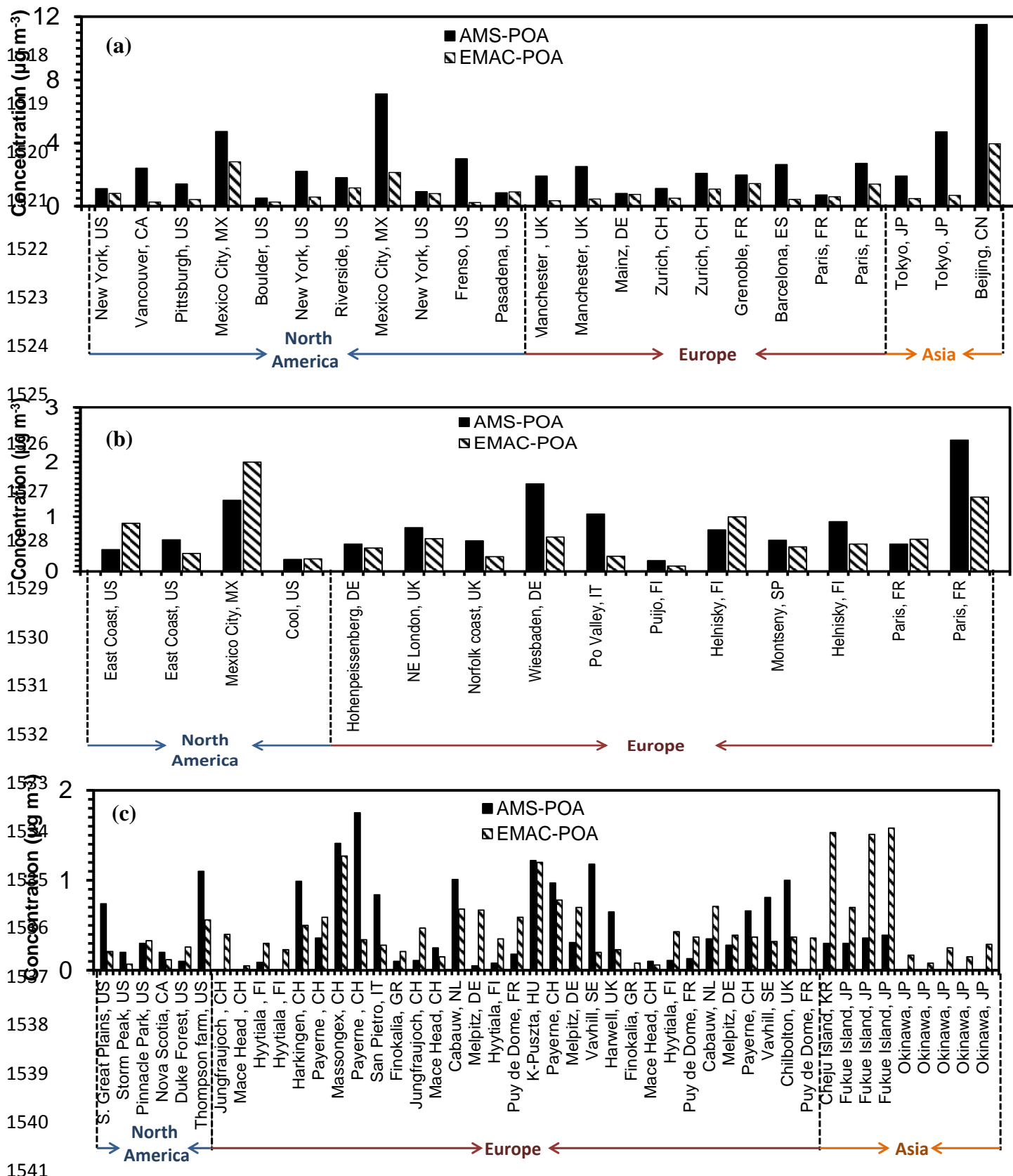


Figure 11: Comparison of EMAC POA (sum of fPOA and bbPOA) to AMS POA (sum of HOA and BBOA) from 84 data sets worldwide over (a) urban, (b) urban downwind and (c) rural/remote areas during 2001-2010.

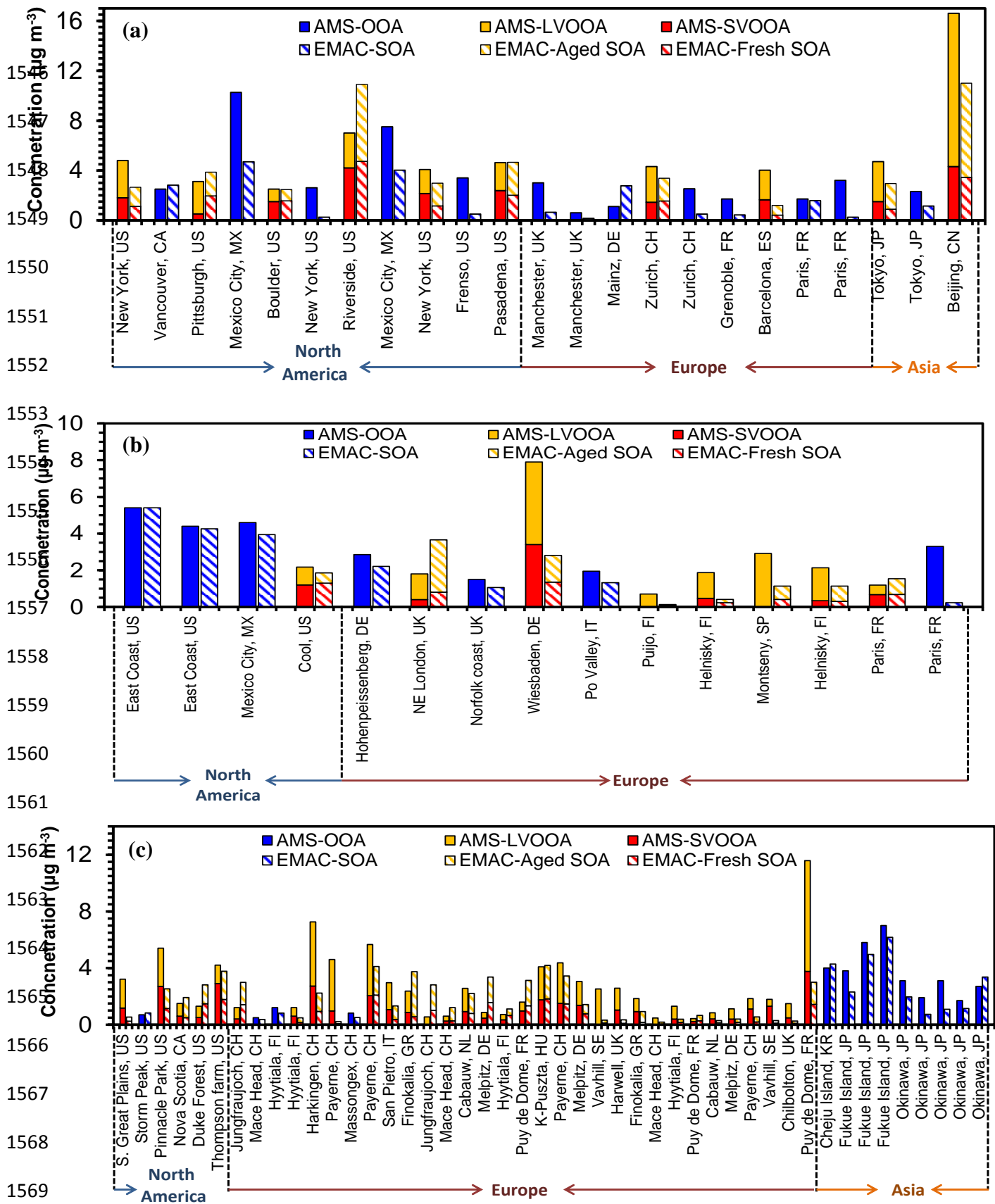
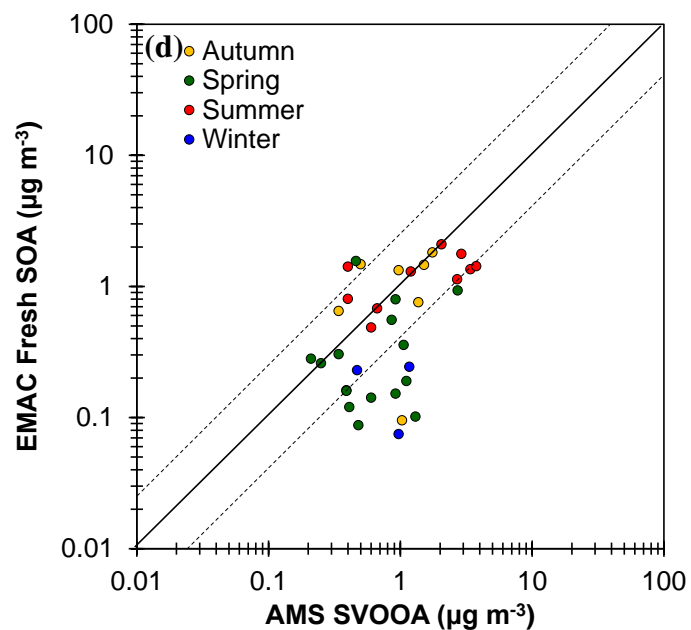
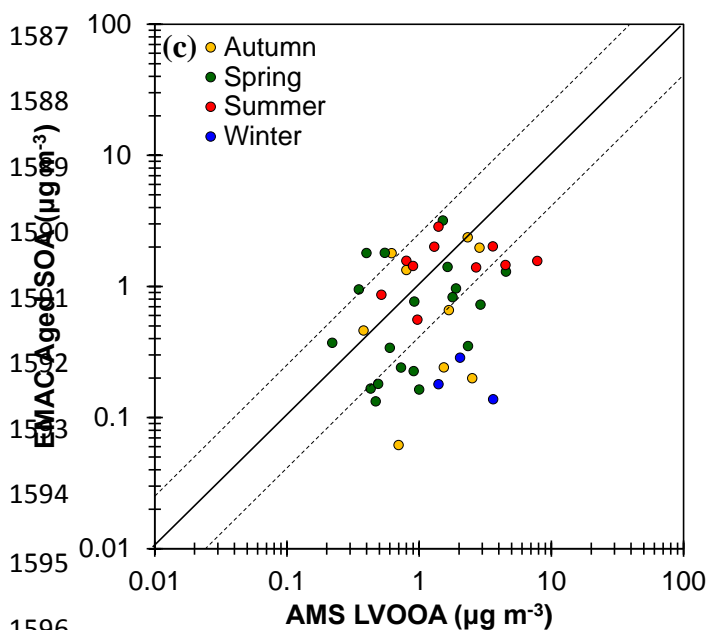
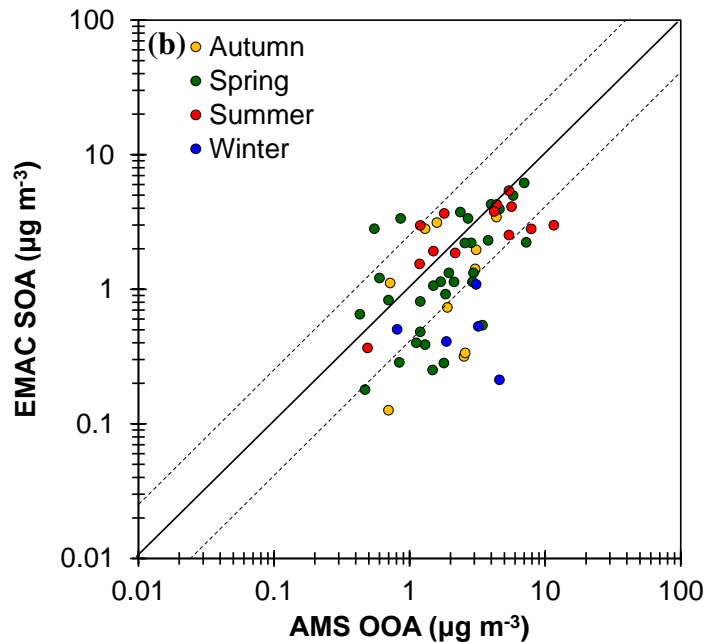
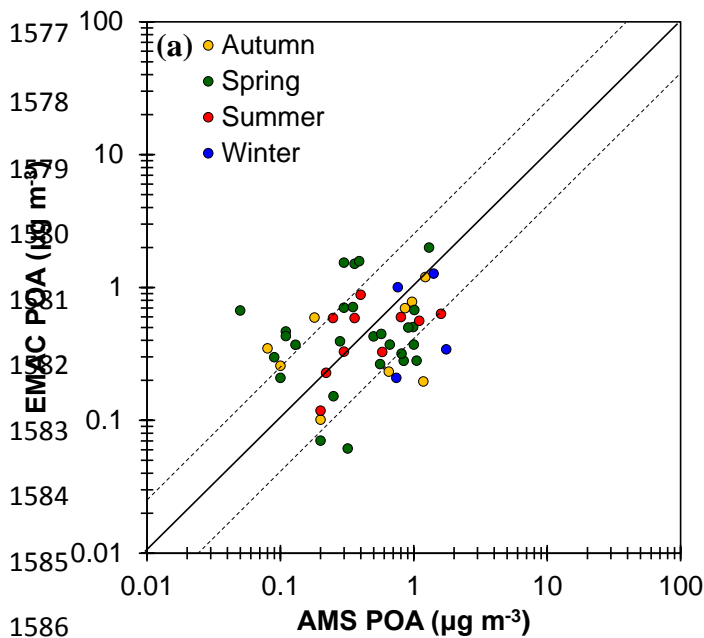


Figure 12: Comparison of EMAC SOA (fresh SOA and aged SOA) to AMS OOA (SV-OOA and LV-OOA) from 84 data sets worldwide over (a) urban, (b) urban downwind and (c) rural/remote areas during 2001-2010.

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1598 **Figure 13:** Scatterplots comparing model results to AMS for: (a) POA, (b) OOA, (c)
1599 LV-OOA, and (d) SV-OOA concentrations (in $\mu\text{g m}^{-3}$) in the Northern Hemisphere
1600 during 2001-2010. Each point represents the data set average value and is colored
1601 based on the season of the field campaign. Also shown are the 1:1, 2:1, and 1:2 lines.

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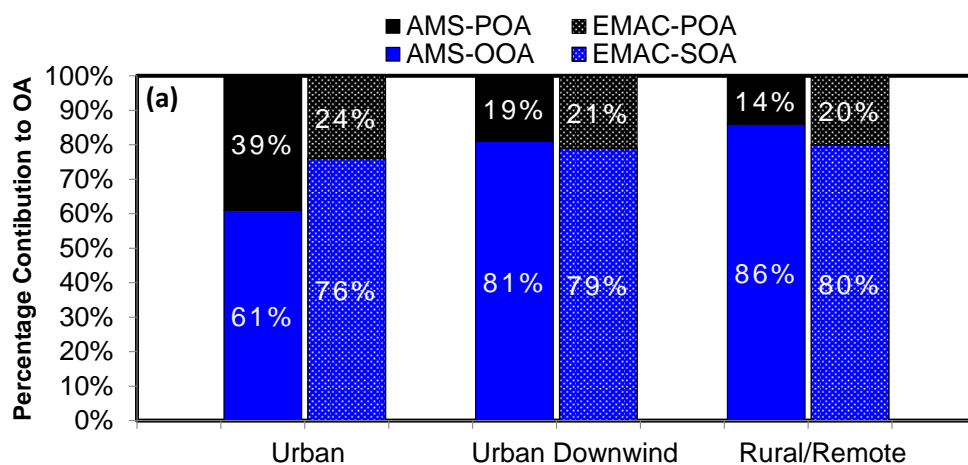
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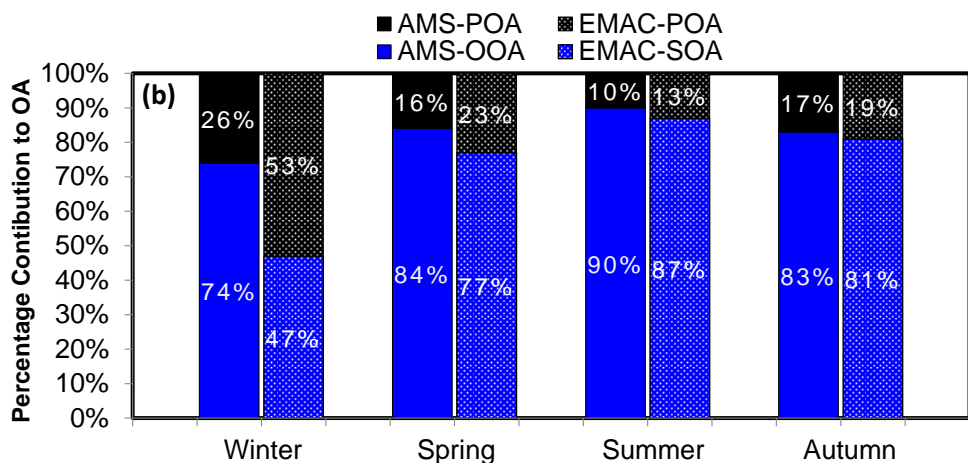
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Figure 14: (a) Spatial and (b) seasonal composition of total OA mass calculated from EMAC and AMS results in the Northern Hemisphere during 2001-2010.

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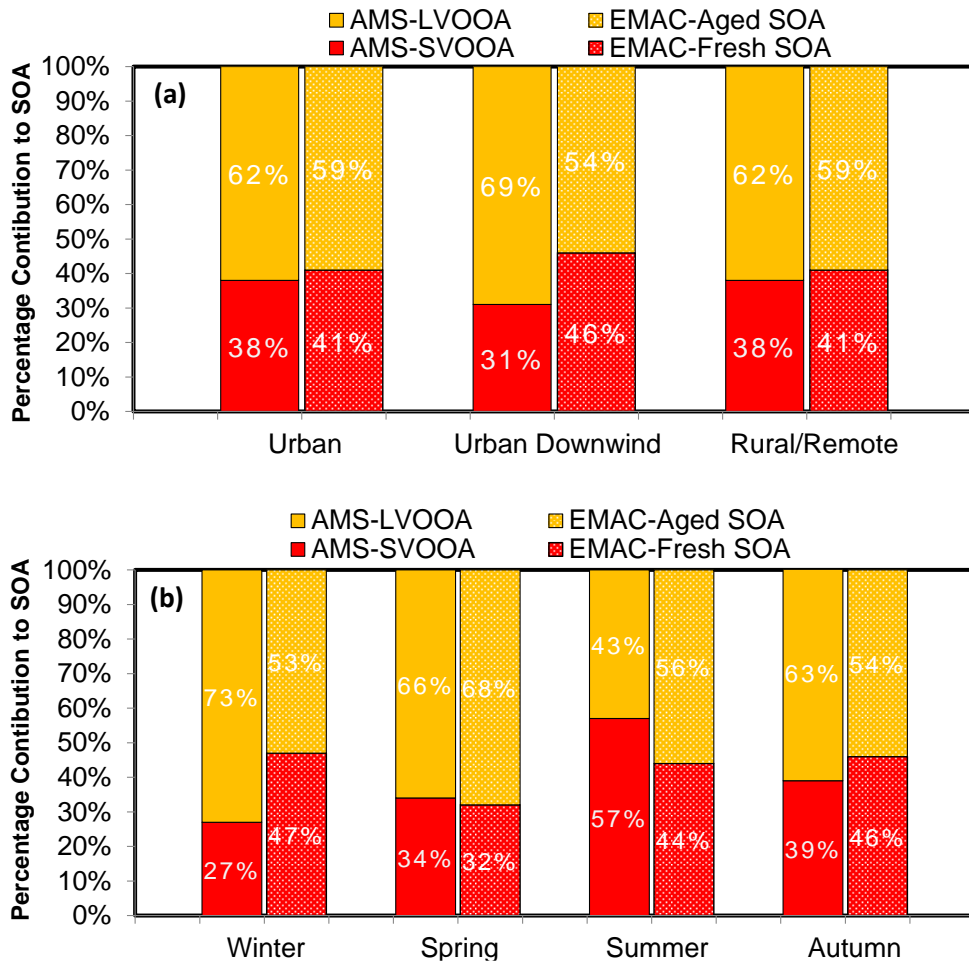


Figure 15: (a) Spatial and (b) seasonal composition of SOA and OOA mass calculated from EMAC and AMS results, respectively, in the Northern Hemisphere during 2001-2010.