

Role of OH variability in the stalling of the global atmospheric CH₄ growth rate from 1999 to 2006

J. McNorton^{1,2}, M. P. Chipperfield^{1,2}, M. Gloor³, C. Wilson^{1,2}, W. Feng^{1,4},
G. D. Hayman⁵, M. Rigby⁶, P. B. Krummel⁷, S. O'Doherty⁶, R. G. Prinn⁸, R. F. Weiss⁹,
D. Young⁶, E. Dlugokencky¹⁰, and S. A. Montzka¹⁰

1. School of Earth and Environment, University of Leeds, Leeds, LS2 9JT, UK.

2. National Centre for Earth Observation, University of Leeds, LS2 9JT, UK.

3. School of Geography, University of Leeds, Leeds, LS2 9JT, UK.

4. National Centre for Atmospheric Science, University of Leeds, LS2 9JT, UK.

5. Centre for Ecology and Hydrology, Wallingford, UK.

6. School of Chemistry, University of Bristol, Bristol, BS8 1TS, UK.

7. CSIRO Oceans and Atmosphere Flagship, Aspendale, Victoria, Australia.

8. Center for Global Change Science, Massachusetts Institute of Technology, Cambridge, MA 02139, USA.

9. Scripps Institution of Oceanography, University of California, San Diego, CA 92093, USA.

10. National Oceanic and Atmospheric Administration, Boulder, USA.

Abstract

The growth in atmospheric methane (CH₄) concentrations over the past two decades has shown large variability on a timescale of several years. Prior to 1999 the globally averaged CH₄ concentration was increasing at a rate of 6.0 ppb/yr, but during a stagnation period from 1999 to 2006 this growth rate slowed to 0.6 ppb/yr. From 2007 to 2009 the growth rate again increased to 4.9 ppb/yr. These changes in growth rate are usually ascribed to variations in CH₄ emissions. We have used a 3-D global chemical transport model, driven by meteorological reanalyses and variations in global mean hydroxyl (OH) concentrations derived from CH₃CCl₃ observations from two independent networks, to investigate these CH₄ growth variations. The model shows that between 1999 and 2006, changes in the CH₄ atmospheric loss contributed significantly to the suppression in global CH₄ concentrations relative to the pre-1999 trend. The largest factor in this is relatively small variations in global mean OH on a timescale of a few years, with minor contributions of atmospheric transport of CH₄ to its sink region and of atmospheric temperature. Although changes in emissions may be important during the stagnation period, these results imply a smaller variation is required to explain the observed CH₄ trends. The contribution of OH variations to the renewed CH₄ growth after 2007 cannot be determined with data currently available.

1. Introduction

The global mean atmospheric methane (CH_4) concentration has increased by a factor of 2.5 since the pre-industrial era, from approximately 722 ppb in 1750 to 1803.2 ± 0.7 ppb in 2011 (Etheridge et al., 1998; Dlugokencky et al., 2005). Over this time period methane has accounted for approximately 20% of the total direct anthropogenic perturbation of radiative forcing by long-lived greenhouse gases ($0.48 \pm 0.05 \text{ W/m}^2$), the second largest contribution after CO_2 (Cicerone et al., 1988; Myhre et al., 2013). This long-term methane increase has been attributed to a rise in anthropogenic emissions from fossil fuel exploitation, agriculture, waste management and biomass burning (Dlugokencky et al., 2011). Predictions of future CH_4 levels require a complete understanding of processes governing emissions and atmospheric removal.

Since the mid-1980s measurements of CH_4 in discrete atmospheric air samples collected at surface sites have been used to observe changes in the interannual growth rate of CH_4 (Rigby et al., 2008; Dlugokencky et al., 2011, Kirschke et al., 2013). Nisbet et al. (2014) showed that between 1984 and 1992 atmospheric CH_4 increased at ~ 12 ppb/yr, after which the growth rate slowed to ~ 3 ppb/yr. In 1999 a period of near-zero growth began which continued until 2007. In 2007 this stagnation period ended and up until 2009 average growth increased again to ~ 4.9 ppb/yr (Rigby et al., 2008; Dlugokencky et al., 2011).

The reasons for the pause in CH_4 growth are not well understood. Bousquet et al. (2006) performed an atmospheric transport inversion study to infer an increase in anthropogenic emissions since 1999. Similarly, the EDGAR v3.2, bottom-up anthropogenic emission inventory, an updated inventory to that used as an a priori by Bousquet et al. (2006), shows a year-on-year increase in anthropogenic CH_4 emissions between 1999 and 2006 (Olivier et al., 2005). This would suggest that a decrease in anthropogenic emissions is not the likely cause of the pause in growth during this period. A second potential explanation is a reduction in wetland emissions between 1999 and 2006, which is in part compensated by an increase in anthropogenic emissions (Bousquet et al., 2006). However, more recently, Pison et al. (2013) used two atmospheric inversions alongside a process-based model and found much more uncertainty in the role wetlands played in the pause in growth over this period. Their study found a negative trend in Amazon basin emissions between 2000 and 2006 from the process-based model and a positive trend from the inversion estimates.

Dlugokencky et al. (2003) argued that the behaviour of global mean CH_4 up to around 2002 was characteristic of the system approaching steady state, accelerated by decreasing emissions at high northern latitudes in the early 1990s and fairly constant emissions elsewhere. However, since then there have been notable perturbations to the balance of sources and sinks (Rigby et al., 2008). The observed growth since 2007 has been, at least partly, attributed to increases in wetland (Bousquet et al., 2011) and anthropogenic emissions (Bousquet et al., 2011). Recent changes in emissions are not well constrained and the reasons for the renewed growth are also not fully understood (Nisbet et al., 2014).

Atmospheric chemistry has also been hypothesised to play a role in past variations in CH_4 growth rates. The major (90%) sink of atmospheric CH_4 is via reaction with the hydroxyl radical, OH. Variations in the global mean concentration of OH ($[\text{OH}]$), or changes to the

reaction rate through changes in temperature, therefore have the potential to affect CH₄ growth. Previous studies have suggested that an increase in atmospheric OH concentration may have been at least partly responsible for a decrease in the CH₄ growth rate (Karlsdottir and Isaksen et al., 2000; Lelieveld et al., 2004; Wang et al., 2004; Fiore et al., 2006). This rise in OH has been attributed to an increase in lightning NO_x (Fiore et al., 2006), a decrease in column O₃ (Wang et al., 2004) and changes in atmospheric pollutants (Karlsdottir and Isaksen et al., 2000). The abundance of other species such as H₂O and CH₄ also determine the concentration of OH (Lelieveld et al., 2004). Prinn et al. (2005) and Voulgarakis et al. (2015) suggested that major global wildfires and El Nino Southern Oscillation (ENSO) events could influence [OH] variability.

Warwick et al. (2002) investigated the impact of meteorology on atmospheric CH₄ growth rates from 1980 to 1998, i.e. well before the observed recent pause. They concluded that atmospheric conditions could be an important driver in the interannual variability (IAV) of atmospheric CH₄. In similar studies a combination of atmospheric dynamics and changes in emissions were shown to explain some of the earlier past trends in atmospheric CH₄ (Fiore et al., 2006; Patra et al., 2009). This paper builds on these studies to investigate the chemical and non-chemical atmospheric contribution to the recent variations in CH₄ growth. By ‘non-chemical’ we mean transport-related influences, although the loss of CH₄ is ultimately due to chemistry as well. We use a 3-D global chemical transport model to simulate the period from 1993 to 2011 and to quantify the impact of variations in [OH] and meteorology on atmospheric CH₄ growth.

2. Data and Models

2.1 NOAA and AGAGE CH₄ Data and Derived OH

We have used surface CH₄ observations from 19 National Oceanographic and Atmospheric Administration/Earth System Research Laboratory (NOAA/ESRL) cooperative global air sampling sites (Dlugokencky et al., 2014) over 1993-2009 (see Table 1). To calculate the global average concentration, measurements were interpolated across 180 latitude bins, which were then weighted by surface area. We have also used the same method to derive global mean CH₄ based on 5 sites from the Advanced Global Atmospheric Gases Experiment (AGAGE) network (Prinn et al., 2000; Cunnold et al., 2002; Prinn et al., 2015).

Montzka et al. (2011) used measurements of methyl chloroform (CH₃CCl₃) from an independent set of flasks sampled approximately weekly at a subset of NOAA air sampling sites to derive global [OH] anomalies from 1997 to 2007 and found only a small interannual variability ($2.3 \pm 1.5\%$). They argued that uncertainties in emissions are likely to limit the accuracy of the inferred inter-annual variability in global [OH], particularly before 1997. At that time the emissions were large but decreasing rapidly due to the phaseout of CH₃CCl₃ production and consumption, and the large atmospheric gradients were also more difficult to capture accurately with only few measurement sites. Instrument issues caused an interruption to their CH₃CCl₃ time series in 2008/9. We have averaged these (based on the red curve in Figure 3 of Montzka et al.) into yearly anomalies to produce relative interannual variations in the mean [OH]. Similarly, Rigby et al. (2013) used CH₃CCl₃ measurements from on-site instrumentation operated continuously within the 5-station AGAGE network in a 12-box model

to produce yearly global [OH] anomalies from 1995 (the date from which data from all 5 stations is available) to 2010. These two timeseries, which convert anomalies in the CH_3CCl_3 decay rate into anomalies in [OH] using constant temperature, correspond to the best estimate of [OH] variability from the two measurement networks by the groups who operate them. We then applied these two series of yearly global anomalies uniformly to the global latitude-height [OH] field used in the recent TransCom CH_4 model intercomparison (see Patra et al., 2011), which itself was derived from a combination of semi-empirically calculated tropospheric OH distributions (Spivakovsky et al. 2000; Huijnen et al., 2010) and 2-D model simulated stratospheric loss rates (Velders, 1995). For consistency between the model experiments, both sets of yearly anomalies were scaled so that the mean [OH] between 1997 and 2007 (the overlap period where NOAA and AGAGE anomalies are both available) equalled the TransCom [OH] value. In the rest of this paper we refer to these two OH datasets as ‘NOAA-derived’ and ‘AGAGE-derived’.

These two calculations of yearly [OH] anomalies use slightly different assumptions for CH_3CCl_3 emissions after 2002. Before that year they use values from Prinn et al. (2005). The NOAA data then assumed a 20% decay in emission for each subsequent year (Montzka et al., 2011), while AGAGE used United Nations Environment Programme (UNEP) consumption values (UNEP, 2015). Holmes et al. (2013) suggested that inconsistencies in CH_3CCl_3 observations between the AGAGE and NOAA networks also limit understanding of OH anomalies for specific years due to an unexplained phasing difference of up to around 3 months. As we are interested in the impact of [OH] changes over longer time periods (e.g. 2000 – 2006), this phase difference will be less important. We have investigated the impact of the different CH_3CCl_3 observations and assumed emissions on the derived [OH] anomalies (see Section 3.1).

2.2 TOMCAT 3-D Chemical Transport Model

We have used the TOMCAT global atmospheric 3-D off-line CTM (Chipperfield, 2006) to model atmospheric CH_4 and CH_3CCl_3 concentrations. The TOMCAT simulations were forced by winds and temperatures from the 6-hourly European Centre for Medium-Range Weather Forecasts (ECMWF) ERA-Interim reanalyses (Dee et al., 2011). They covered the period 1993 to 2011 with a horizontal resolution of $2.8^\circ \times 2.8^\circ$ and 60 levels from the surface to ~60 km.

The TOMCAT simulations use annually repeating CH_4 emissions, which have been scaled to previous estimates of 553 Tg/yr (Ciais et al., 2013), taken from various studies (Fiore et al., 2006; Curry et al., 2007; Bergamaschi et al., 2009; Pison et al., 2009; Spahni et al., 2011; Ito et al., 2012). Annually-repeating anthropogenic emissions (except biomass burning) were calculated from averaging the EDGAR v3.2 (2009) inventory from 1993 to 2009 (Olivier and Berowski, 2001). Biomass burning emissions were calculated using the GFED v3.1 inventory and averaged from 1997 to 2009 (van der Werf et al., 2010). The Joint UK Land Environment Simulator (JULES) (Best et al., 2011; Clark et al., 2011; Hayman et al., 2014) was used to calculate a wetland emission inventory between 1993 and 2009, which was then used to produce a mean annual cycle. Annually-repeating rice (Yan et al., 2009), hydrate, mud volcano, termite, wild animal and ocean (Matthews et al., 1987) emissions were taken from the

TransCom CH₄ study (Patra et al., 2011). The methane loss fields comprised an annually repeating soil sink (Patra et al., 2011), an annually repeating stratospheric loss field (Velders, 1995) and a specified zonal mean [OH] field. This does not account for longitudinal variations in [OH], which are considered to be negligible compared to latitudinal variations. To create a reasonable spatial distribution the model was spun up for 15 years prior to initialising the simulations, using emission data from 1977 to 1992 where available and annual averages otherwise. Before reinitialising the model in 1993, concentrations were scaled using the model and observed global concentrations to remove any imbalance.

Fifteen TOMCAT simulations were performed each with a CH₄ tracer and a CH₃CCl₃ tracer. The runs had differing treatments of meteorology (winds and temperature) and [OH] (see Table 2). Simulations with repeating [OH] fields (RE_xxxx) used the TransCom dataset. The other runs with varying [OH] used the NOAA-derived or AGAGE-derived [OH] fields based on the original published work or our estimates (see Section 3.1). For these runs, the mean [OH] field is used where the respective NOAA or AGAGE-derived [OH] is unavailable or uncertain (before 1997 / after 2007 for NOAA and before 1997 / after 2009 for AGAGE). The five simulations with fixed wind and temperature fields (with labels ending in FTFW) used the ERA-Interim analyses from 1996 repeated for all years. The five simulations with varying winds and fixed temperature (with labels ending in FTVW) used zonal mean temperature fields averaged from 1993-2009, any influence from the relatively small longitudinal temperature variations is unlikely to have a noticeable impact. We also derive our own [OH] anomalies from the anomaly in the CH₃CCl₃ loss rate, which combines variations in atmospheric OH concentration with variations in temperature which affect the rate constant of the CH₃CCl₃ + OH reaction. To quantify the importance of this temperature effect we also performed 5 model runs which allow both winds and temperature to vary interannually according to ERA-Interim data (labels ending VTVW). Fixed temperature simulations are used for general analysis because the derived OH anomalies already implicitly contain temperature variations.

3. Results

3.1 Correlation of CH₄ variations with OH and temperature

We first investigate the extent to which variations in the observed CH₄ growth rate correlate with variations in derived [OH]. Figure 1a shows the published NOAA-derived and AGAGE-derived global [OH] anomalies along with the annual CH₄ growth rate estimated from the NOAA and AGAGE measurements. The two [OH] series show the similar behaviour of negative anomalies around 1997 and 2006/7, and an extended period of more positive anomalies in between. For the time periods covered by the NOAA (1997-2007) and AGAGE (1997-2009) CH₃CCl₃ observations, the two derived [OH] timeseries show negative correlations with the CH₄ growth from NOAA (regression coefficient, $R = -0.32$) and AGAGE ($R = -0.64$). Only the AGAGE [OH] correlation, from the longer timeseries, is statistically significant at the 90% level. We assume that this correlation arises from variability in [OH] driving variability in CH₄ growth, although the correlation could be the result of a bidirectional effect, whereby decreased also CH₄ acts to increase [OH]. We note that Spivakovsky et al., (2000) showed a 25% (~450 ppb) change in model CH₄ equates to a 5-6% change in [OH]. This far exceeds the annual growth observed, therefore this effect is likely to be small.

However, the concentration of others species which affect OH such as CO and volatile organic compounds (VOCs) can co-vary with the methane concentration, for example during years with high biomass burning emissions, so the effect may be larger than suggested by the Spivakovsky et al., (2000) study.

We can use a simple ‘global box model’ (see Supplement S1) to estimate the [OH] variations required to fit the observed CH₄ growth rate variations assuming constant CH₄ emissions and temperature (black line in Figure 1b). This provides a crude guide to the magnitude of OH variations which could be important for changes in the CH₄ budget. Our results are consistent with those of Montzka et al. (2011) who performed a similar analysis on the NOAA CH₄ data. The required [OH] rarely exceeds their CH₃CCl₃-derived interannual variability (IAV) range of [OH] ($\pm 2.3\%$, shown as shading in the figure). Also shown in Figure 1b are the published estimates of the global mean OH anomalies from Figure 1a, converted to concentration units (see Section 2.1). The relative interannual variations in [OH] required to fit the CH₄ observations match the CH₃CCl₃-derived [OH] variations in many years, for example from 1998-2002 (see Montzka et al., 2011). Some of the derived variations in [OH] exceed that required to match the CH₄ growth rate, with larger negative anomalies in the early and later years and some slightly larger positive values in middle of the period.

Figures 1c and 1d show our estimates of [OH] using NOAA and AGAGE observations and two assumptions of post-2000 CH₃CCl₃ emissions (see Section 2.1) in a global box model. The figures also compare our OH estimates with the NOAA-derived and AGAGE-derived [OH] anomalies based on the work of the observation groups (Figure 1a). Our results demonstrate the small impact of using different observations and post-2000 emission assumptions (compare filled and open red circles for the two panels). For these box model results there is also only a very small effect of using annually varying temperature (compare red and blue lines). In later years the choice of observations has a bigger impact than the choice of emissions on the derived [OH]. For AGAGE-derived values (Figure 1d) our estimates agree well with the published values of Rigby et al., (2013), despite the fact we use a global box model while they used a more sophisticated 12-box model. In contrast, there are larger differences between our values and the NOAA-derived OH variability published by Montzka et al. (2011) (Figure 1c), despite both studies using box models. In particular, around 2002-2003 we overestimate the positive anomaly in [OH]. We also estimate a much more negative OH anomaly in 1997 than Montzka et al., though we slightly underestimate the published AGAGE-derived anomaly in that year (Figure 1d). Tests show that differences between our results and the NOAA box model are due to the treatment of emissions. This suggests a larger uncertainty in the inferred low 1997 [OH] value, when emissions of CH₃CCl₃ were decreasing rapidly, although reasons why atmospheric [OH] might have been anomalously low were discussed by Prinn et al. (2005). In the subsequent analysis we use the OH variability from the published NOAA and AGAGE studies as input to the 3-D model.

3.2 TOMCAT Simulations

Overall, Figure 1 shows the potential importance of small, observationally derived variations in OH concentrations to impact methane growth. We now investigate this quantitatively in the framework of a 3-D CTM.

3.2.1 Methyl Chloroform

The TOMCAT simulations include a CH_3CCl_3 tracer. This allows us to verify that our approach of using a global OH field, scaled by derived anomalies, allows the model to reproduce the observed magnitude and variability of CH_3CCl_3 decay accurately. Figure 2a shows that the model, with the imposed [OH] field, does indeed simulate the global decay of CH_3CCl_3 very well. This justifies our use of the ‘offline’ [OH] field, as models with interactive tropospheric chemistry can produce a large range in absolute global mean [OH] and therefore in lifetimes of gases such as CH_3CCl_3 . For example, Voulgarakis et al., (2013) analysed the global mean [OH] from various 3D models and found a range of 0.65×10^6 to 1.34×10^6 molecules cm^{-3} . Furthermore, Montzka et al., (2011) discussed how photochemical models typically show smaller interannual variability than CH_3CCl_3 -derived OH, again suggesting that the models are not accurately capturing all relevant processes. Figure 2a also shows that the global mean CH_3CCl_3 from the NOAA and AGAGE networks differ by ~2.5ppt around 1993-1996, but since then this difference has become smaller.

The observed and modelled CH_3CCl_3 decay rate anomalies (calculated using the method of Holmes et al., (2013) with a 12-month smoothing) are shown in Figures 2b and 2c (different panels are used for AGAGE and NOAA comparisons for clarity). The model and observation-derived results both tend to show a faster CH_3CCl_3 decay (more positive anomaly) in the middle of the period, with slower decay at the start and end. The anomalies for the NOAA and AGAGE-derived OH show periodic variations on a timescale of 2-3 yrs but with a phase shift between the two datasets of 3 months, as noted by Holmes et al., (2013). The model runs with OH variability prescribed from the observations and varying winds also show these periodic variations with correlation coefficients ranging from 0.71 – 0.90. The correlation values for these runs using varying OH are all larger than the run using repeating OH (for RE_FTVW $R=0.62$ compared to AGAGE data and 0.67 compared to NOAA data). Note that for CH_3CCl_3 decay there are only small differences between the 3-D simulations which use varying temperatures and the corresponding runs which use fixed temperature (e.g. simulation RE_VTVW versus RE_FTVW). This agrees with the results of Montzka et al (2011) based on their box model. This shows that the largest contribution from the CH_3CCl_3 decay rate anomaly comes from variations in atmospheric OH concentration, rather than atmospheric temperature. The simulations with repeating winds show less variability in the CH_3CCl_3 decay rate, particularly in the period 1999-2004, but the small difference suggests that the interannual variability in the observed CH_3CCl_3 decay rate is driven primarily by the variations in the OH concentration. The remaining interannual variability in run RE_FTFW is due to variations in emissions.

Figure 3 shows the CH_3CCl_3 decay and decay rate anomalies at four selected stations, two from the NOAA network and two from the AGAGE network. The good agreement in the global CH_3CCl_3 decay in Figure 2 is also seen at these individual stations. At the AGAGE stations of Mace Head and Gape Grim, the model runs with varying OH perform better in capturing the decay rate anomalies than the runs with repeating OH. However, the impact of variability in the winds (solid lines versus dotted lines) is more apparent at these individual stations compared to the global means. At the NOAA station of Mauna Loa the model run with varying

OH and varying winds also appears to perform better in capturing the observed variability in CH_3CCl_3 decay. At the South Pole the observed variability is small, except in 2000-2002. This feature is not captured by the model.

In summary, Figures 2 and 3 show that the global OH fields that we have constructed from different datasets can perform well in capturing the decay of CH_3CCl_3 and its anomalies both globally and at individual stations. Although, the interannual variability in global mean OH has been derived from these CH_3CCl_3 observations, the figures do show that the reconstructed model OH fields (which also depend on the methodology discussed in Section 2) perform well in simulating CH_3CCl_3 within the 3D model. Therefore, we would argue that these fields are suitable for testing the impact of OH variability on the methane growth rate. Even so, it is important to bear in mind that these fields may not represent the true changes in atmospheric OH, particularly if the interannual variability in CH_3CCl_3 emissions was a lot different to that assumed here. However, we would again note that we are focussing on the impact of multi-year (≥ 2 years) variability which appears more robustly determined by the networks under differing assumptions of temperature and emissions than year-to-year variability.

3.2.2 Methane

Figure 4 shows deseasonalised modelled surface CH_4 from the 3-D CTM simulations compared with in-situ observations from a northern high-latitude station (Alert), two tropical stations (Mauna Loa and Tutuila), a southern high-latitude station (South Pole) and the global average of the NOAA and AGAGE stations. The global comparisons are shown for simulations both with varying and repeating meteorology. Figure 5 shows the global annual CH_4 growth rates with a 12-month smoothing (panel a) and differences between the model and NOAA and AGAGE observations (panels b and c). The changes in the modelled global mean CH_4 over different time periods are given in Table 3.

Figure 4 shows that in 1993, at the end of the model spin-up, the simulations capture the global mean CH_4 level well, along with the observed values at a range of latitudes. The exception is at high northern latitudes. However, these differences are not important when investigating the change in the global growth rate. The global change in atmospheric CH_4 in all simulations from 1993 to the end of 2009 is between 75 and 104 ppb, compared to 56 and 66 ppb in the observations.

Model run RE_FTFW does not include interannual variations in atmospheric transport or CH_4 loss. Therefore, and also given the lack of change in emissions the modelled CH_4 gradually approaches a steady state value of ~ 1830 ppb (Figure 4f). The rate of CH_4 growth decreases from 7.9 ppb/yr (1993-1998) to 1.4 ppb/yr (2007-2009). Compared to run RE_FTFW, the other simulations introduce variability on this CH_4 evolution.

Run RE_FTVW includes interannual variability in wind fields which may alter the transport of CH_4 from the source (emission) to the sink regions. The largest difference between runs RE_FTFW and RE_FTVW occurs after 2000 (Figure 4f). During the stagnation period (1999-2006) run RE_FTVW has a smaller growth rate of 3.5 ppb/yr compared to 4.1 ppb/yr in run

RE_FTFW, showing that variations in atmospheric transport made a small contribution to the slowdown in global mean CH₄ growth.

Compared to run RE_FTVW, runs AP_FTVW, AL_FTVW, NP_FTVW and NL_FTVW include CH₃CCl₃-derived interannual variations in [OH] which introduce large changes in modelled CH₄, which are more in line with the observations (Figure 4e and 5). These runs produce turnarounds in the CH₄ growth in 2001/2 (becomes negative) and 2005/6 (returns to being positive). For AGAGE-derived [OH] (runs AP_FTVW, AL_FTVW) the large negative anomaly in OH in 1997 produces a significant increase in CH₄ prior to the turnaround in 2001.

Table 3 summarises the change in global mean CH₄ over different time periods. These periods are defined by the key dates in the observed record, i.e. 1999 and 2006 as the start and end dates of the stagnation period. Comparison of Figure 4e and Table 3 shows, however, that the timing of the largest modelled change in growth rate do not necessarily coincide with those dates. That is understandable if other factors not considered here, e.g. emission changes, are contributing to the change in global CH₄ concentration. It does mean that the summary model values in Table 3 do not capture the full impact of the changes in [OH] and winds within the stagnation period. Figure 4e shows that model runs with varying OH perform better in simulating the relative CH₄ trend from 1999 to around 2004.

Table 3 shows that runs NP_FTVW and NL_FTVW (NOAA-derived [OH]) produce a small modelled CH₄ growth of 2.5-3.1 ppb/yr during the stagnation period 1999-2006, compared to 1.0 ppb/yr for run AP_FTVW (AGAGE-derived [OH]). The AGAGE results are slightly larger than the observed growth rate of 0.6-0.7 ppb/yr. Runs AL_FTVW, AP_FTVW, NL_FTVW and NP_FTVW capture the observed strong decrease in the CH₄ growth rate. With the exception of AP_FTVW between 1999 and 2006 (p-value = 0.37) all trends, over all three time periods, are statistically significant at the 90% level. Clearly, these runs demonstrate the significant potential for relatively small variations in mean [OH] to affect CH₄ growth. Excluding the stagnation period the mean modelled CH₄ lifetime in run NP_FTVW is 9.4 years, but this decreases slightly by 0.01 years during the stagnation period. For run AP_FTVW there is a decrease of 0.18 years from 9.6 years between the same intervals. The results from all the CTM simulations during 1999-2006 indicate that the accuracy of modelled CH₄ growth is improved by accounting for interannual variability in [OH] as derived from CH₃CCl₃ observations, and interannual variability in meteorology.

The variation of [OH] after 2007 cannot be determined from the available NOAA data so run NP_FTVW used the mean [OH] field for all subsequent years. The modelled CH₄ increase of 3.5 ppb/yr underestimates the observations (4.9 ppb/yr). Should the lower [OH] of 2007 have persisted then the model would have produced a larger increase in CH₄, in better agreement with the observations. The AGAGE-derived [OH] for 2007-2009 (run AP_FTVW) produces a larger CH₄ growth relative to the previous years (8.8 ppb/yr). Runs RE_FTFW (1.4 ppb/yr) and RE_FTVW (1.8 ppb/yr) both show a decreased rate of growth during the final 5 years, consistent with a system approaching steady state.

Figure 5a shows the global CH₄ growth rate derived from the AGAGE and NOAA networks together with selected model simulations. Figures 5b and c show the differences between the

model simulations and the NOAA and AGAGE observations, respectively. The runs which include variations in [OH] agree better with the observed changes, i.e. larger R values in panel (a) and the model lines are closer to the y=0 line in panels (b) and (c), especially in the first 5 years of the stagnation period. It is interesting to note that the relative impacts of wind and temperature variations are larger for CH₄ than for CH₃CCl₃ (compare simulations RE_FTFW, RE_FTVW and RE_VTVW in Figures 2 and 5a). The temperature dependences of the OH loss reactions are similar for the two species (see Supplement S1) but the impact of variability in transport is likely to be greater for CH₄ due to its stronger spatial gradients than for CH₃CCl₃. Figure S2 in the Supplementary Material shows the very weak horizontal gradients in CH₃CCl₃ in its period of atmospheric decay due to small emissions. In contrast, variations in emissions lead to large spatial gradients in CH₄ which can then couple with variability in transport. This lack of spatial variability in CH₃CCl₃ is an advantage when using this species to derive OH variability as it reduces the possible complication from transport variability. The impact of variability in temperature will remain, however. In principle, it would be possible to use a 3-D inverse model with realistic temperature fields to derive a time-dependent 3-D OH field which is consistent with the CH₃CCl₃ decay. However, there are not enough observations to constrain such a model. Using the TOMCAT model, in Supplement S2 we test whether differences in the distribution of the CH₃CCl₃ and CH₄ observation networks will affect the anomaly signal derived by the application of the same OH field. The results there show that the differences in the distribution of the observing stations is not likely to be important.

4. Discussion and Conclusions

Our model results suggest that variability in atmospheric [OH] played a key role in the observed recent variations in CH₄ growth, particularly during the CH₄ stagnation period between 1999 and 2006. The 3-D CTM calculations show that during the stagnation period, variations in atmospheric conditions in the tropical lower to mid-troposphere could potentially account for an important component of the observed decrease in global CH₄ growth. Within this, small increases in [OH] were the largest factor, while variations in transport from source to sink regions made a smaller contribution. Note again, however, that the ultimate loss of CH₄ is still due to chemistry. The role of atmospheric temperature variations is factored into the observationally derived OH, but model experiments show that changes in the OH concentration itself is most important. The remainder of the variation can be ascribed to other processes not considered in our runs such as emission changes. There are also measurement uncertainties to consider and the possible underrepresentation of the global mean CH₃CCl₃ which will affect the derived OH concentration. Our results are consistent with an earlier budget study which analysed 1991 to 2004 and found that variations in [OH] were the main control of variations in atmospheric CH₄ lifetime (65%), with temperature accounting for a smaller fraction (35%) (Fiore et al., 2006). However, they were not able to study the full period of the pause in CH₄ growth and did not impose observation-based [OH] variations. As we have noted here the CH₄ lifetime can also be affected by emissions distributions which affects transport to the main loss regions.

Prior to the stagnation period the simulation using AGAGE-derived [OH] (9.7-10.4 ppb/yr) overestimates CH₄ growth when compared to observations (6.0-7.1 ppb/yr), which degrades

the agreement with the observed CH₄ variations. A likely cause of this is inaccuracies in derived [OH] in 1997 when emissions still played a large role in the observed CH₃CCl₃ and the e-fold decay had not yet stabilised (Montzka et al., 2011).

We have not accounted for expected variations in CH₄ emissions in this study. We can conclude that although global CH₄ emissions do vary year-to-year, the observed trend in CH₄ growth between 1999 and 2006 was impacted by changing atmospheric processes that affected CH₄ loss. Changes in emissions are still important over this time period and likely still dominate CH₄ variations over other time periods. The observed changes in growth rates during ENSO events in e.g. 1998 are poorly captured by the meteorological changes considered here and can be attributed to changes in emissions through changing precipitation and enhanced biomass burning (Hodson et al., 2011). The renewed growth of CH₄ in 2007 is also poorly captured by all model simulations without varying [OH]. The observed decrease in AGAGE and NOAA-derived [OH] coincides with the increase in CH₄ growth in 2007, although the currently available data do not allow for a more detailed investigation of the possible contribution of [OH] changes in this recent increase.

Despite the differences in year-to-year variability in [OH] derived from CH₃CCl₃ observations (Holmes et al., 2013), we find that [OH] variability derived from two different networks of surface CH₃CCl₃ observations over multi-year periods provide insights into atmospheric CH₄ variations. Improved quantification of the role of OH variability will require efforts to reduce uncertainties associated with estimating [OH]. Estimates of global mean [OH] in recent years from CH₃CCl₃ observations are becoming increasingly difficult because CH₃CCl₃ levels are currently <5 ppt; hence this may limit the accuracy of derived [OH] and its variability in future years (Lelieveld et al., 2006). Wennberg et al. (2004) also noted that there can be time variations in the net flux of CH₃CCl₃ by the oceans, which could potentially affect the derived [OH] concentrations and which were not considered in our analysis. However, the impact of interannual variability in this flux are not likely to be important. For the period considered in this study, Figure 2 of Wennberg et al., (2004) shows that the CH₃CCl₃ flux into the ocean decreased from the largest value in 1997 to almost zero in recent years, which mimics CH₃CCl₃ emissions. Including the estimated 1997 ocean flux in our box model decreased the OH anomaly for that year by 0.8%. This change would decrease in magnitude in the subsequent years. Overall, accurate estimates of [OH] beyond 2009 will require more sophisticated analysis of CH₃CCl₃ observations, derivation from other species or improved representation of [OH] in photochemical models.

Overall our study suggests that future atmospheric trends in CH₄ are likely to be strongly influenced by not only emissions but also by changes in processes that affect atmospheric loss. Therefore, to be realistic, predictions of these future trends need to explicitly account for likely variations in [OH], the major sink, and possibly other processes related to tropospheric and stratospheric chemistry.

Acknowledgements: JRM thanks NERC National Centre for Earth Observation (NCEO) for a studentship. CW, MPC and MG acknowledge support from NERC grants GAUGE (NE/K002244/1) and AMAZONICA (NE/F005806/1). GDH acknowledges support from the

447 European Space Agency through its Support to Science Element initiative (ALANIS Methane),
448 NCEO and the NERC MAMM grant (NE/I028327/1). SAM acknowledges support in part from
449 NOAA Climate Program Office's AC4 program. AGAGE is supported by NASA grants
450 NNX11AF17G to MIT and NNX11AF15G and NNX11AF16G to SIO, by NOAA, UK
451 Department of Food and Rural Affairs (DEFRA) and UK Department for Energy and Climate
452 Change (DECC) grants to Bristol University, and by CSIRO and Australian Bureau of
453 Meteorology. MR is supported by a NERC Advanced Fellowship (NE/I021365/1). Model
454 calculations were performed on the Arc1 and Archer supercomputers.

References

- Bergamaschi, P., Frankenberg, C., Meirink, J. F., Krol, M., Villani, M. G., Houweling, S., Dentener, F., Dlugokencky, E. J., Miller, J. B., Gatti, L. V., Engel, A., and Levin, I.: Inverse modeling of global and regional CH₄ emissions using SCIAMACHY satellite retrievals, *J. Geophys. Res.*, 114, 10.1029/2009jd012287, 2009.
- Best, M. J., Pryor, M., Clark, D. B., Rooney, G. G., Essery, R. L. H., Ménard, C. B., Edwards, J. M., Hendry, M. A., Porson, A., Gedney, N., Mercado, L. M., Sitch, S., Blyth, E., Boucher, O., Cox, P. M., Grimmond, C. S. B., and Harding, R. J.: The Joint UK Land Environment Simulator (JULES), model description – Part 1: Energy and water fluxes, *Geosci. Model Dev.*, 4, 677-699, 10.5194/gmd-4-677-2011, 2011.
- Bousquet, P., Ciais, P., Miller, J. B., Dlugokencky, E. J., Hauglustaine, D. A., Prigent, C., Van der Werf, G. R., Peylin, P., Brunke, E. G., Carouge, C., Langenfelds, R. L., Lathiere, J., Papa, F., Ramonet, M., Schmidt, M., Steele, L. P., Tyler, S. C., and White, J.: Contribution of anthropogenic and natural sources to atmospheric methane variability, *Nature*, 443, 439-443, 10.1038/nature05132, 2006.
- Bousquet, P., Ringeval, B., Pison, I., Dlugokencky, E. J., Brunke, E. G., Carouge, C., Chevallier, F., Fortems-Cheiney, A., Frankenberg, C., Hauglustaine, D. A., Krummel, P. B., Langenfelds, R. L., Ramonet, M., Schmidt, M., Steele, L. P., Szopa, S., Yver, C., Viovy, N., and Ciais, P.: Source attribution of the changes in atmospheric methane for 2006–2008, *Atmos. Chem. Phys.*, 11, 3689-3700, 10.5194/acp-11-3689-2011, 2011.
- Chipperfield, M. P.: New version of the TOMCAT/SLIMCAT off-line chemical transport model: Intercomparison of stratospheric tracer experiments. *Q. J. R. Meteorol Soc.*, 132, 1179–1203, 2006.
- Ciais, P., Sabine, C., Bala, G., Bopp, L., Brovkin, V., Canadell, J., Chhabra, A., DeFries, R., Galloway, J., Heimann, M., Jones, C., Le Quere, C., Myneni, R. B., Piao, S., and Thornton, P.: Carbon and other biogeochemical cycles, in: *Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change*, Cambridge University Press, 2013.
- Cicerone, R. J., and Oremland, R. S.: Biogeochemical aspects of atmospheric methane, *Global Biogeochem. Cycles*, 2, 299-327, 1988.
- Clark, D. B., Mercado, L. M., Sitch, S., Jones, C. D., Gedney, N., Best, M. J., Pryor, M., Rooney, G. G., Essery, R. L. H., Blyth, E., Boucher, O., Harding, R. J., Huntingford, C., and Cox, P. M.: The Joint UK Land Environment Simulator (JULES), Model description – Part 2: Carbon fluxes and vegetation, *Geosci. Model Dev.*, 4, 701-722, 10.5194/gmd-4-701-2011, 2011.
- Cunnold, D., Steele, L., Fraser, P., Simmonds, P., Prinn, R., Weiss, R., Porter, L., O'Doherty, S., Langenfelds, R., and Krummel, P.: In situ measurements of atmospheric methane at GAGE/AGAGE sites during 1985–2000 and resulting source inferences, *J. Geophys. Res.*, 107, ACH 20-21-ACH 20-18, 2002.
- Curry, C. L.: Modeling the soil consumption of atmospheric methane at the global scale, *Global Biogeochem. Cycles*, 21, GB4012, doi:10.1029/2006GB002818, 2007.

497 Dee, D., Uppala, S., Simmons, A., Berrisford, P., Poli, P., Kobayashi, S., Andrae, U.,
 498 Balmaseda, M., Balsamo, G., Bauer, P., et al.: The ERA - Interim reanalysis:
 499 Configuration and performance of the data assimilation system, *Q. J. R. Meteorol. Soc.*,
 500 137, 553-597, 2011.

501 Dlugokencky, E. J., Houweling, S., Bruhwiler, L., Masarie, K., Lang, P., Miller, J., and Tans,
 502 P.: Atmospheric methane levels off: Temporary pause or a new steady - state?,
 503 *Geophys. Res. Lett.*, 30, doi:10.1029/2003GL018126, 2003.

504 Dlugokencky, E. J., Myers, R., Lang, P., Masarie, K., Crotwell, A., Thoning, K., Hall, B.,
 505 Elkins, J., and Steele, L.: Conversion of NOAA atmospheric dry air CH₄ mole fractions
 506 to a gravimetrically prepared standard scale, *J. Geophys. Res.*, 110, D18306, 2005.

507 Dlugokencky, E. J., Nisbet, E. G., Fisher, R., and Lowry, D.: Global atmospheric methane:
 508 budget, changes and dangers, *Philos. Trans. R. Soc. A*, 369, 2058-2072,
 509 10.1098/rsta.2010.0341, 2011.

510 Dlugokencky, E. J., P.M. Lang, A.M. Crotwell, K.A. Masarie, M.J. Crotwell, Atmospheric
 511 Methane Dry Air Mole Fractions from the NOAA ESRL Carbon Cycle Cooperative
 512 Global Air Sampling Network, 1983-2013, Version: 2014-06-24. Available at
 513 ftp://aftp.cmdl.noaa.gov/data/trace_gases/ch4/flask/surface/. Accessed July 6, 2014.

514 Etheridge, D. M., Steele, L. P., Francey, R. J., and Langenfelds, R. L.: Atmospheric methane
 515 between 1000 A.D. and present: Evidence of anthropogenic emissions and climatic
 516 variability, *J. Geophys. Res.*, 103, 15,979-15,993, 10.1029/98jd00923, 1998.

517 Fiore, A. M., Horowitz, L. W., Dlugokencky, E. J., and West, J. J.: Impact of meteorology
 518 and emissions on methane trends, 1990–2004, *Geophys. Res. Lett.*, 33, L12809,
 519 10.1029/2006gl026199, 2006.

520 Hayman, G.D., et al., Comparison of the HadGEM2 climate-chemistry model against in-situ
 521 and SCIAMACHY atmospheric methane data, *Atmos. Chem. Phys.*, 14, 13,257-13,280,
 522 2014.

523 Hodson, E. L., Poulter, B., Zimmermann, N. E., Prigent, C., and Kaplan, J. O.: The El Niño-
 524 Southern Oscillation and wetland methane interannual variability, *Geophys. Res. Lett.*,
 525 38, L08810, 10.1029/2011gl046861, 2011.

526 Holmes, C. D., Prather, M. J., Søvde, O., and Myhre, G.: Future methane, hydroxyl, and their
 527 uncertainties: key climate and emission parameters for future predictions, *Atmos.*
 528 *Chem. Phys.*, 13, 285-302, 2013.

529 Huijnen, V., Williams, J., Weele, M. v., Noije, T. v., Krol, M., Dentener, F., Segers, A.,
 530 Houweling, S., Peters, W., and de Laat, J.: The global chemistry transport model TM5:
 531 description and evaluation of the tropospheric chemistry version 3.0, *Geosci. Model*
 532 *Dev.*, 3, 445-473, 2010.

533 Ito, A., and Inatomi, M.: Use of a process-based model for assessing the methane budgets of
 534 global terrestrial ecosystems and evaluation of uncertainty, *Biogeosciences*, 9, 759-773,
 535 10.519/bg-9-759-2012, 2012.

536 Karlsdottir, S. and Isaksen, I.S.A.: Changing methane lifetime: Possible cause for reduced
 537 growth. *Geophys. Res. Lett.*, 27(1), 93-96, 2000.

538 Kirschke, S., Bousquet, P., Ciais, P., Saunois, M., Canadell, J. G., Dlugokencky, E. J.,
 539 Bergamaschi, P., Bergmann, D., Blake, D. R., Bruhwiler, L., et al.: Three decades of
 540 global methane sources and sinks, *Nature Geosci.*, 6, 813-823, 2013.

541 Lelieveld, J., Dentener, F., Peters, W., and Krol, M.: On the role of hydroxyl radicals in the
 542 self-cleansing capacity of the troposphere, *Atmos. Chem. Phys.*, 4, 2337-2344, 2004.

543 Lelieveld, J., Brenninkmeijer, C. A. M., Joeckel, P., Isaksen, I. S. A., Krol, M. C., Mak, J. E.,
 544 Dlugokencky, E., Montzka, S. A., Novelli, P. C., Peters, W. and Tans, P. P.: New
 545 Directions: Watching over tropospheric hydroxyl (OH), *Atmospheric Environment*, 40,
 546 5741-5743, 2006.

547 Matthews, E., and Fung I.: Methane emissions from natural wetlands: Global distribution,
 548 area, and ecology of sources. *Global Biogeochem. Cycles*, 1, 61–86, 1987.

549 Montzka, S. A., Krol, M., Dlugokencky, E., Hall, B., Jöckel, P., and Lelieveld, J.: Small
 550 interannual variability of global atmospheric hydroxyl, *Science*, 331, 67-69, 2011.

551 Myhre, G., Shindell, D., Bréon, F., Collins, W., Fuglestad, J., Huang, J., Koch, D.,
 552 Lamarque, J., Lee, D., Mendoza, B., Nakajima, T., Robock, A., Stephens, G.,
 553 Takemura, T., and Zhang, H.: Anthropogenic and natural radiative forcing, in: *Climate*
 554 *Change 2013: The Physical Science Basis. Contribution of Working Group I to the*
 555 *Fifth Assessment Report of the Intergovernmental Panel on Climate Change*,
 556 Cambridge University Press, 2013.

557 Naik, V., Voulgarakis, A., Fiore, A. M., Horowitz, L., Lamarque, J.-F., Lin, M., Prather, M.
 558 J., Young, P., Bergmann, D., and Cameron-Smith, P.: Preindustrial to present-day
 559 changes in tropospheric hydroxyl radical and methane lifetime from the Atmospheric
 560 Chemistry and Climate Model Intercomparison Project (ACCMIP), *Atmos. Chem.*
 561 *Phys.*, 13, 5277-5298, 2013.

562 Nisbet, E. G., Dlugokencky, E. J., and Bousquet, P.: Atmospheric science. Methane on the
 563 rise - again, *Science*, 343, 493-495, 10.1126/science.1247828, 2014.

564 Olivier, J.G.J., Berdowski J.J.M.: Global emissions sources and sinks, in: *The Climate*
 565 *System*, edited by: Berdowski J, Guicherit R, and Heij BJ., IISBN 9058092550, A. A.
 566 Balkema Publishers/Swets & Zeitlinger Pub., Lisse, The Netherlands, 33–78, 2001.

567 Olivier, J. G., Van Aardenne, J. A., Dentener, F. J., Pagliari, V., Ganzeveld, L. N., and Peters,
 568 J. A.: Recent trends in global greenhouse gas emissions: regional trends 1970–2000 and
 569 spatial distribution of key sources in 2000, *Environmental Sciences*, 2, 81-99, 2005.

570 Patra, P. K. , Takigawa, M., Ishijima, K., Choi, B.-C., Cunnold, D., J. Dlugokencky, E.,
 571 Fraser, P., J. Gomez-Pelaez, A., Goo, T.-Y., Kim, J.-S., Krummel, P., Langenfelds, R.,
 572 Meinhardt, F., Mukai, H., O'Doherty, S., G. Prinn, R., Simmonds, P., Steele, P.,
 573 Tohjima, Y., Tsuboi, K., Uhse, K., Weiss, R., Worthy, D., and Nakazawa, T.: Growth
 574 rate, seasonal, synoptic, diurnal variations and budget of methane in the lower
 575 atmosphere, *J. Meteorol. Soc. Japan*, 87, 635-663, 10.2151/jmsj.87.635, 2009.

576 Patra, P. K., Houweling, S., Krol, M., Bousquet, P., Belikov, D., Bergmann, D., Bian, H.,
 577 Cameron-Smith, P., Chipperfield, M. P., and Corbin, K.: TransCom model simulations
 578 of CH₄ and related species: linking transport, surface flux and chemical loss with CH₄

579 variability in the troposphere and lower stratosphere, *Atmos. Chem. Phys.*, 11, 12,813-
580 12,837, 2011.

581 Patra, P., Krol, M., Montzka, S., Arnold, T., Atlas, E., Lintner, B., Stephens, B., Xiang, B.,
582 Elkins, J., and Fraser, P.: Observational evidence for interhemispheric hydroxyl-radical
583 parity, *Nature*, 513, 219-223, 2014.

584 Pison, I., Bousquet, P., Chevallier, F., Szopa, S., and Hauglustaine, D.: Multi-species
585 inversion of CH₄, CO and H₂ emissions from surface measurements, *Atmos. Chem.*
586 *Phys.*, 9, 5281-5297, 2009.

587 Pison, I., Ringeval, B., Bousquet, P., Prigent, C., and Papa, F.: Stable atmospheric methane in
588 the 2000s: key-role of emissions from natural wetlands, *Atmos. Chem. Phys.*, 13,
589 11,609-11,623, 10.5194/acp-13-11609-2013, 2013.

590 Prinn, R., Weiss, R., Fraser, P., Simmonds, P., Cunnold, D., Alyea, F., O'Doherty, S.,
591 Salameh, P., Miller, B., and Huang, J.: A history of chemically and radiatively
592 important gases in air deduced from ALE/GAGE/AGAGE, *J. Geophys. Res.*, 105,
593 17,751-17,792, 2000.

594 Prinn, R. G.: Evidence for variability of atmospheric hydroxyl radicals over the past quarter
595 century, *Geophys. Res. Lett.*, 32, L07809, 10.1029/2004gl022228, 2005.

596 Prinn, R.G., R.F. Weiss, P.J. Fraser, P.G. Simmonds, S. O'Doherty, P. Salameh, L. Porter, P.
597 Krummel, R.H.J. Wang, B. Miller, C. Harth, B. Grealley, F.A. Van Woy, L.P. Steele, J.
598 Mühle, G. Sturrock, F.N. Alyea, J. Huang, and D.E. Hartley, The ALE / GAGE
599 AGAGE Network (DB1001), Carbon Dioxide Information Analysis Center (CDIAC),
600 U.S. Department of Energy (DOE), <http://cdiac.esd.ornl.gov/ndps/alegage.html>, 2015.

601 Rigby, M., Prinn, R. G., Fraser, P. J., Simmonds, P. G., Langenfelds, R., Huang, J., Cunnold,
602 D. M., Steele, L. P., Krummel, P. B., and Weiss, R. F.: Renewed growth of atmospheric
603 methane, *Geophys. Res. Lett.*, 35, L22805, 2008.

604 Rigby, M., Prinn, R. G., O'Doherty, S., Montzka, S. A., McCulloch, A., Harth, C. M., Mühle,
605 J., Salameh, P., Weiss, R. F., and Young, D.: Re-evaluation of the lifetimes of the
606 major CFCs and CH₃CCl₃ using atmospheric trends, *Atmos. Chem. Phys.*, 13, 2691-
607 2702, 2013.

608 Sander, S.P., et al., Chemical Kinetics and Photochemical Data for Use in Atmospheric
609 Studies Evaluation Number 17. *JPL Publication* 10-6, Jet Propulsion Laboratory,
610 Pasadena, USA, 2011.

611 Spahni, R., Wania, R., Neef, L., van Weele, M., Pison, I., Bousquet, P., Frankenberg, C.,
612 Foster, P. N., Joos, F., Prentice, I. C., and van Velthoven, P.: Constraining global
613 methane emissions and uptake by ecosystems, *Biogeosciences*, 8, 1643-1665,
614 10.5194/bg-8-1643-2011, 2011.

615 Spivakovsky, C., Logan, J., Montzka, S., Balkanski, Y., Foreman-Fowler, M., Jones, D.,
616 Horowitz, L., Fusco, A., Brenninkmeijer, C., and Prather, M.: Three-dimensional
617 climatological distribution of tropospheric OH: Update and evaluation, *J. Geophys.*
618 *Res.*, 105, 8931-8980, 2000.

- UNEP, The UNEP Environmental Data Explorer, as compiled from United Nations Environment Programme . United Nations Environment Programme.
<http://ede.grid.unep.ch>, 2015.
- van der Werf, G. R., Randerson, J. T., Giglio, L., Collatz, G., Mu, M., Kasibhatla, P. S., Morton, D. C., DeFries, R., Jin, Y. v., and van Leeuwen, T. T.: Global fire emissions and the contribution of deforestation, savanna, forest, agricultural, and peat fires (1997–2009), *Atmos. Chem. Phys.*, 10, 11707-11735, 2010.
- Velders, G. J. M.: Description of the RIVM 2-dimensional stratosphere model, RIVM Rapport 722201002, 1995.
- Voulgarakis, A., Naik, V., Lamarque, J.-F., Shindell, D. T., Young, P., Prather, M. J., Wild, O., Field, R., Bergmann, D., and Cameron-Smith, P.: Analysis of present day and future OH and methane lifetime in the ACCMIP simulations, *Atmos. Chem. Phys.*, 13, 2563-2587, 2013.
- Voulgarakis, A., Marlier, M.E., Faluvegi, G., Shindell, D.T., Tsigaridis, K. and Mangeon, S.: Interannual variability of tropospheric trace gases and aerosols: The role of biomass burning emissions, *J. Geophys. Res.: Atmos.*, 120(14), 7157-7173, 2015.
- Wang, J.S., Logan, J.A., McElroy, M.B., Duncan, B.N., Megretskaia, I.A. and Yantosca, R.M.: A 3 - D model analysis of the slowdown and interannual variability in the methane growth rate from 1988 to 1997, *Global Biogeochem. Cycles*, 18(3), 2004.
- Warwick, N. J., Bekki S., Law K. S., Nisbet E. G., and Pyle, J. A.: The impact of meteorology on the interannual growth rate of atmospheric methane, *Geophys. Res. Lett.*, 29, doi:10.1029/2002GL015282, 2002.
- Wennberg, P. O., Peacock, S., Randerson, J. T., and Bleck, R.: Recent changes in the air - sea gas exchange of methyl chloroform, *Geophys. Res. Lett.*, 31, L16112, 2004.
- Yan, X., Akiyama, H., Yagi, K., and Akimoto, H.: Global estimations of the inventory and mitigation potential of methane emissions from rice cultivation conducted using the 2006 Intergovernmental Panel on Climate Change Guidelines, *Global Biogeochem. Cycles*, 23, GB2002, 10.1029/2008gb003299, 2009.

650 **Table 1.** List of NOAA and AGAGE stations which provided CH₄ and CH₃CCl₃ observations.

Site Code	Site Name	Lat. (°N)	Lon. (°N)	Altitude (km)	CH ₄	CH ₃ CCl ₃	Start Date ⁺⁺	End Date
ABP	Arembepe, Brazil	-12.77	-38.17	0	NOAA		27/10/2006	12/01/2010
ALT	Alert, Canada	82.45	-62.51	0.2	NOAA	NOAA	10/06/1985	Ongoing
ASC	Ascension Island, UK	7.97	-14.4	0.09	NOAA		11/05/1983	Ongoing
BRW	Barrow, USA	71.32	-156.61	0.01	NOAA	NOAA	06/04/1983	Ongoing
CGO	Cape Grim, Australia	-40.68	144.69	0.09	NOAA/AGAGE	NOAA/AGAGE	19/04/1984	Ongoing
HBA	Halley Station, UK	-75.61	-26.21	0.03	NOAA		17/01/1983	Ongoing
ICE	Storhofdi, Iceland	63.4	-20.29	0.12	NOAA		02/10/1992	Ongoing
KUM	Cape Kumukahi, USA	19.5	-154.8	0.02	-	NOAA	-	-
LEF	Park Falls, USA	45.9	-90.3	0.47	-	NOAA	-	-
MHD	Mace Head, Ireland	53.33	-9.9	0.01	NOAA/AGAGE	AGAGE**	03/06/1991	Ongoing
MLO	Mauna Loa, USA	19.54	-155.58	3.4	NOAA	NOAA	06/05/1983	Ongoing
NWR	Niwot Ridge, USA	40.05	-105.59	3.52	NOAA	NOAA	21/06/1983	Ongoing
PAL	Pallas-Sammaltunturi, Finland	67.97	24.12	0.56	NOAA		21/12/2001	Ongoing
PSA	Palmer Station, USA	-64.92	-64	0.01	NOAA	**	01/01/1983	Ongoing
RPB	Ragged Point, Barbados	13.17	-59.43	0.02	NOAA/AGAGE	AGAGE	14/11/1987	Ongoing
SEY	Mahe Island, Seychelles	-4.68	55.53	0	NOAA		12/05/1983	Ongoing
SMO	Tutuila, American Samoa	-14.25	-170.56	0.04	NOAA	NOAA/AGAGE	23/04/1983	Ongoing
SPO	South Pole, USA	-89.98	-24.8	2.81	NOAA	NOAA	20/02/1983	Ongoing
STM	Ocean Station M, Norway	66	2	0	NOAA		29/04/1983	27/11/2009
SUM	Summit, Greenland	72.6	-38.42	3.21	NOAA	**	23/06/1997	Ongoing
THD	Trinidad Head, USA	41.1	-124.1	0.1	AGAGE	AGAGE**	09/1995	Ongoing
ZEP	Ny-Alesund, Norway/Sweden	78.91	11.89	0.47	NOAA		11/02/1994	Ongoing

652 ++For NOAA CH₃CCl₃ data the record starts in 1992 at 7 of the 9 stations used here. It started
653 in 1995 for KUM and 1996 for LEF.

654 **NOAA flask data from these sites was not used in the present study or in Montzka et al.,
655 (2011).

656 **Table 2.** Summary of the fifteen TOMCAT 3-D CTM simulations.

Run	OH time variation	Meteorology ^b	
		Winds ^c	Temperature ^d
RE_FTFW	Repeating ^a	Fixed	Fixed
RE_FTVW	Repeating ^a	Varying	Fixed
RE_VTVW	Repeating ^a	Varying	Varying
AP_FTFW	AGAGE (Rigby et al., 2013)	Fixed	Fixed
AP_FTVW	AGAGE (Rigby et al., 2013)	Varying	Fixed
AP_VTVW	AGAGE (Rigby et al., 2013)	Varying	Varying
AL_FTVT	AGAGE (this work)	Fixed	Fixed
AL_FTVW	AGAGE (this work)	Varying	Fixed
AL_VTVW	AGAGE (this work)	Varying	Varying
NP_FTFW	NOAA (Montzka et al., 2011)	Fixed	Fixed
NP_FTVW	NOAA (Montzka et al., 2011)	Varying	Fixed
NP_VTVW	NOAA (Montzka et al., 2011)	Varying	Varying
NL_FTFW	NOAA (this work)	Fixed	Fixed
NL_FTVW	NOAA (this work)	Varying	Fixed
NL_VTVW	NOAA (this work)	Varying	Varying

657 a. Annually repeating [OH] taken from Patra et al. (2011).

658 b. Varying winds and temperatures are from ERA-Interim.

659 c. Fixed winds using repeating ERA-Interim winds from 1996.

660 d. Fixed temperatures use zonal mean ERA-Interim temperatures averaged over 1993-2009.

Table 3. Calculated methane changes over different time periods from selected TOMCAT experiments and the NOAA and AGAGE observation networks. Standard errors shown are calculated from statistically independent unsmoothed monthly global CH₄ growth data.

Model run or observation network	Global mean ΔCH_4 in ppb (ppb/yr)			
	2009-1993	1998-1993	2006-1999	2009-2007
RE_FTFW	85.0 (5.0 \pm 0.2)	47.2 (7.9 \pm 0.1)	32.9 (4.1 \pm 0.1)	4.3 (1.4 \pm 0.1)
RE_FTVW	82.2 (4.8 \pm 0.2)	48.2 (8.0 \pm 0.3)	27.8 (3.5 \pm 0.3)	5.4 (1.8 \pm 0.3)
RE_VTVW	74.6 (4.4 \pm 0.2)	45.6 (7.6 \pm 0.2)	23.1 (2.9 \pm 0.2)	5.3 (1.8 \pm 0.2)
AP_FTVW ^a	97.7 ^e (5.7 \pm 0.4)	62.3 ^e (10.4 \pm 0.5)	8.2 ^g (1.0 \pm 0.4)	26.4 (8.8 \pm 0.6)
AL_FTVW ^b	104.2 ^e (6.1 \pm 0.4)	58.4 ^e (9.7 \pm 0.4)	17.3 (2.2 \pm 0.5)	27.5 (9.2 \pm 0.5)
NP_FTVW ^c	86.2 ^f (5.1 \pm 0.3)	49.7 ^f (8.3 \pm 0.3)	24.8 (3.1 \pm 0.4)	10.6 ^f (3.8 \pm 0.7)
NL_FTVW ^d	91.4 ^f (5.4 \pm 0.5)	58.8 ^f (9.8 \pm 0.5)	20.1 (2.5 \pm 0.6)	11.3 ^f (3.8 \pm 1.0)
NOAA obs.	56.1 (3.3 \pm 0.3)	36.0 (6.0 \pm 0.4)	4.8 (0.6 \pm 0.3)	14.7 (4.9 \pm 0.4)
AGAGE obs.	66.3 (3.9 \pm 0.4)	42.6 (7.1 \pm 0.9)	5.6 (0.7 \pm 0.7)	17.4 (5.8 \pm 0.7)

a. Taken from Rigby et al. (2013) and Patra et al. (2011).

b. Using 1997-2009 relative annual changes in mean [OH] derived from AGAGE data (Cunnold et al., 2002).

c. Taken from Montzka et al. (2011) and Patra et al. (2011).

d. Using 1997-2007 relative annual changes in mean [OH] derived from NOAA data (Prinn et al., 2015).

e. Value using mean [OH] from 1993-1996.

f. Value using mean [OH] from 1993-1996 and 2008-2011.

g. Trend value not statistically significant at the 90% level.

Figures

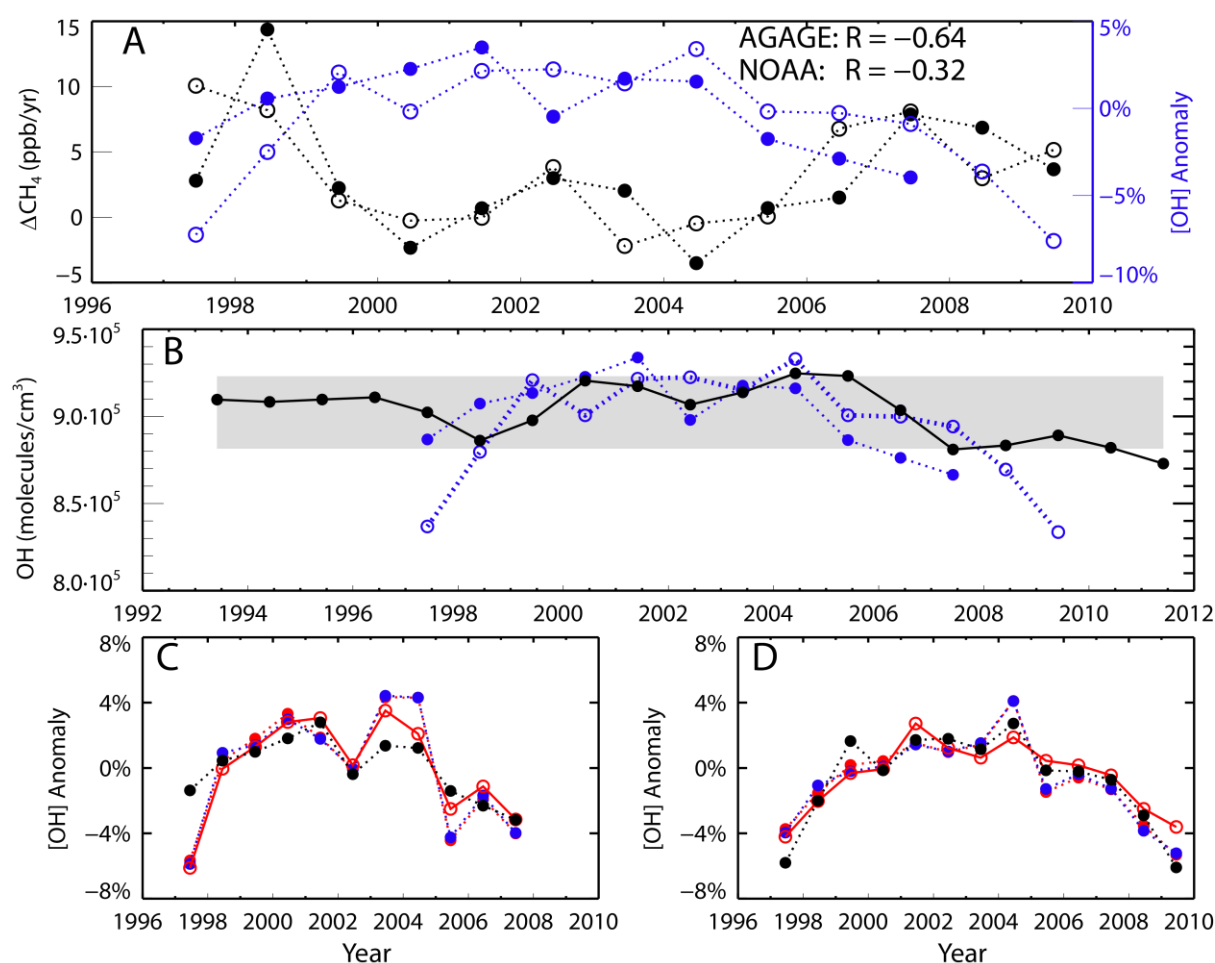
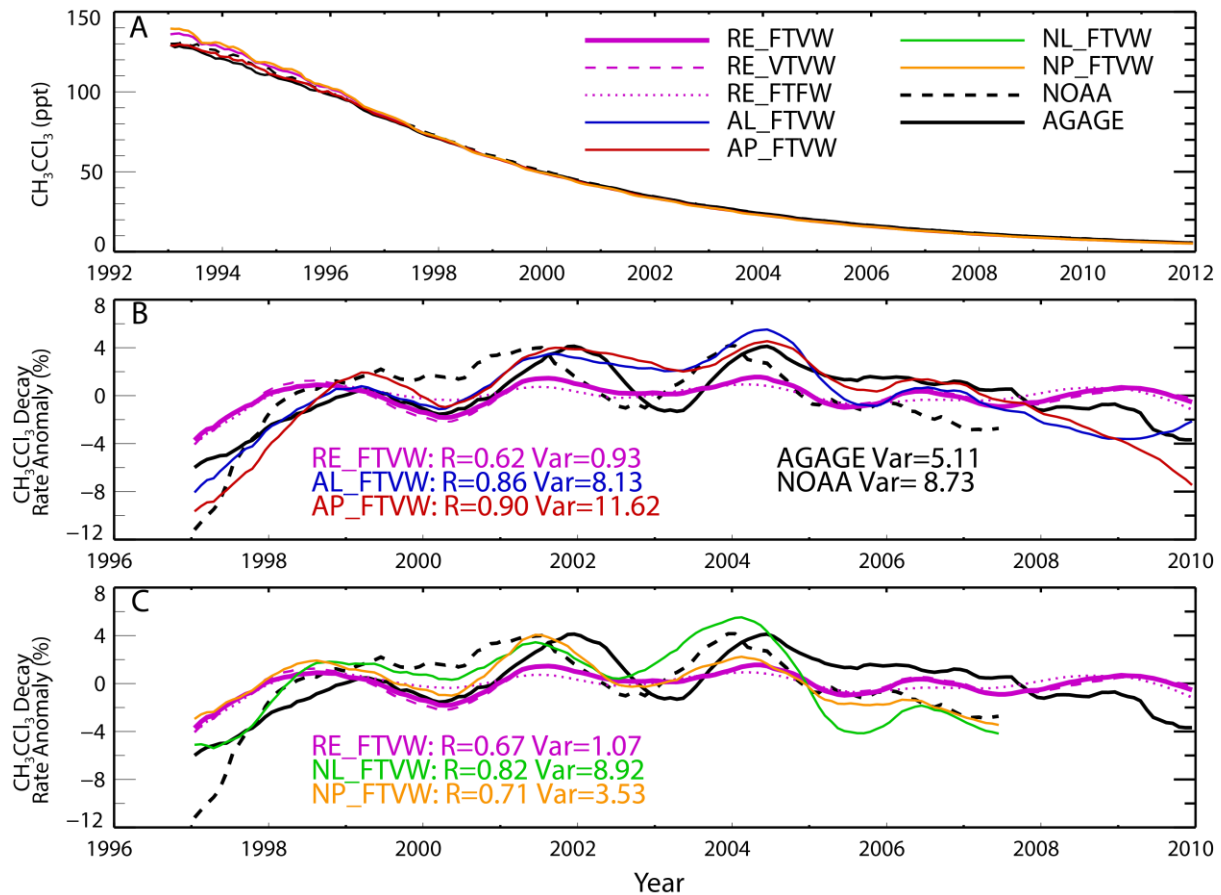


Figure 1. (a) Annual global CH_4 growth rate (ppb/yr) derived from NOAA (filled black circles) and AGAGE (open black circles) data (left hand y-axis), and published annual global $[\text{OH}]$ anomalies derived from NOAA (filled blue circles, 1997-2007) and AGAGE (open blue circles, 1997-2009) CH_3CCl_3 measurements (right hand y-axis) (see text). (b) Annual mean $[\text{OH}]$ (molecules/ cm^3) required for global box model (see Supplement S1) to fit yearly variations in NOAA CH_4 observations assuming constant emissions and temperature ($E=553$ Tg/yr, $T=272.9$ K), based on Montzka et al. (2011) (solid black line). The shaded region denotes $[\text{OH}]$ deviation of $\pm 2.3\%$ from the 1993-2011 mean. Also shown are the NOAA- and AGAGE-derived anomalies from panel (a) for an assumed mean OH (see Section 2.1). (c) Our estimates of $[\text{OH}]$ derived from NOAA CH_3CCl_3 calculated using a global box model (Supplement S1) using repeating (blue) and varying (red) annual mean temperature and the CH_3CCl_3 emission scenario from UNEP (2015) (filled circles and dashed lines). Also shown for varying temperatures are results using the emissions of Montzka et al (2011) (red open circles and solid line) based on (Prinn et al. 2005) and the NOAA-derived values from panel (a) (black dashed line and circles). (d) As panel (c) but for OH derived from AGAGE CH_3CCl_3 observations.

693



694

695 **Figure 2.** (a) Global mean surface CH_3CCl_3 (ppt) from NOAA (black dashed) and AGAGE
696 (black solid) observations from 1993 to 2012. Also shown are results from five TOMCAT
697 simulations with fixed temperatures and varying winds (see Table 1). (b) Global surface
698 CH_3CCl_3 decay rate anomalies from NOAA and AGAGE along with model runs RE_FTVW,
699 AL_FTVW and AP_FTVW (solid lines). Results from runs RE_FTFW and RE_VTVW are
700 shown as a purple dotted line and dashed line, respectively. Observation and model anomalies
701 are smoothed with a 12-month running average. Values given represent correlation coefficient
702 when compared to AGAGE observations and variance. The decay rate anomaly is calculated
703 from global mean CH_3CCl_3 values using equation (1) from Holmes et al., (2013), expressed as
704 a percentage of the typical decay with a 12-month smoothing. (c) As panel (b) but for model
705 runs NL_FTVW and NP_FTVW, along with RE_FTVW, RE_VTVW and RE_FTFW, and
706 correlation coefficients for comparison with NOAA observations. The model results are split
707 across panels (b) and (c) for clarity.

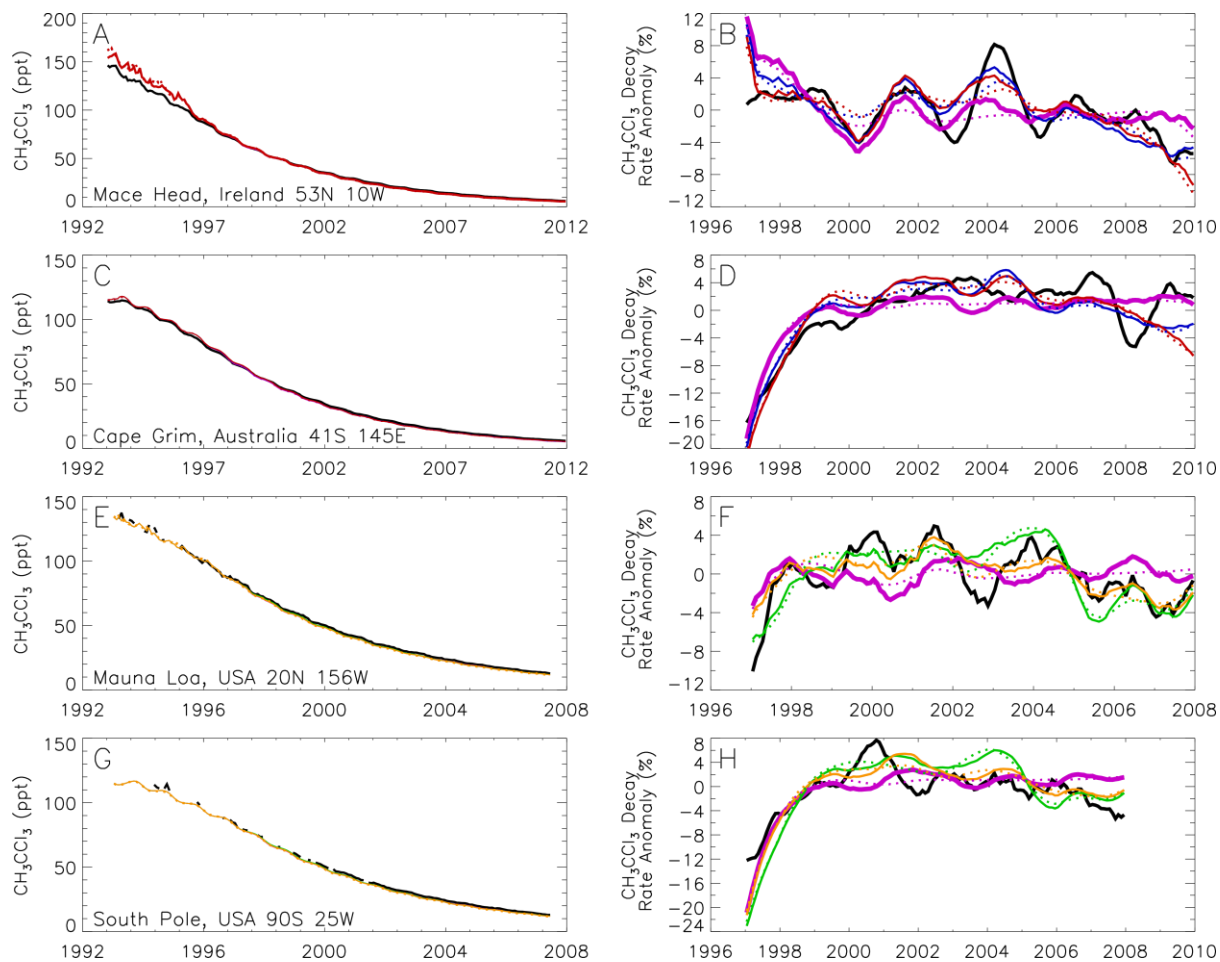


Figure 3. (Left) Observed mean surface CH_3CCl_3 (ppt) (black line) from (a) Mace Head (AGAGE), (c) Cape Grim (AGAGE), (e) Mauna Loa (NOAA) and (g) South Pole (NOAA). Also shown are results from five TOMCAT simulations with fixed temperatures and varying winds (FTVW, for legend see Figure 2a). (Right): Surface CH_3CCl_3 decay rate anomalies at the same station as the corresponding left column plot for observations (black), TOMCAT simulations with varying winds (FTVW, solid coloured lines) and TOMCAT simulations with fixed winds (FTFW, dotted lines). Comparisons at NOAA (AGAGE) stations show only comparisons with runs using NOAA (AGAGE)-derived OH, along with runs RE_FTVW and RE_FTFW in all panels.

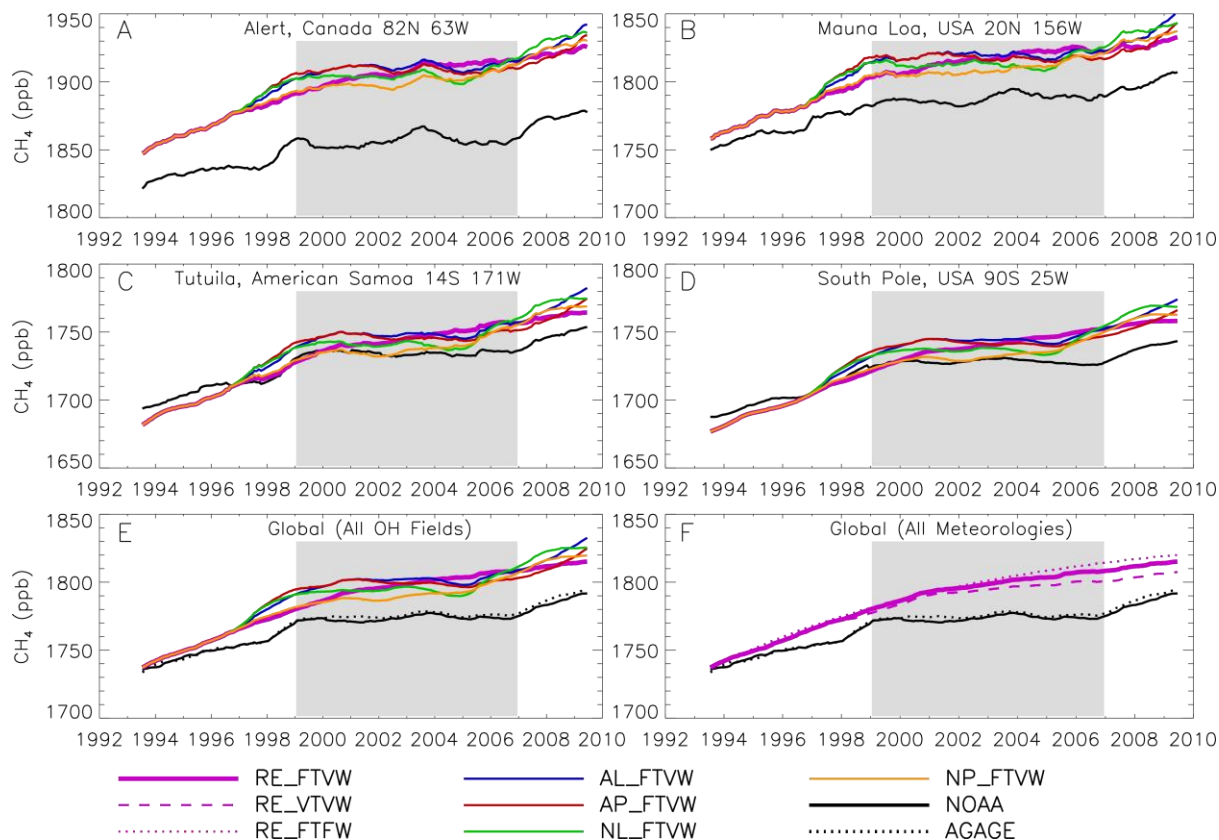


Figure 4. (a, b, c and d) Deseasonalised surface CH₄ (ppb) from 4 NOAA sites (black solid line) from 1993 to 2009. Also shown are results from five TOMCAT 3-D CTM simulations with fixed temperatures and varying winds (FTVW, see **Table 2**). (e) Deseasonalised global mean surface CH₄ from NOAA (black solid) and AGAGE (black dashed) observations along with five TOMCAT simulations with different treatments of OH. (f) Same as (e) but for TOMCAT simulations using repeating OH (RE) and different treatments of winds and temperature. All panels use observation and model values which are smoothed with a 12-month running average. The shaded region marks the stagnation period in the observed CH₄ growth rate.

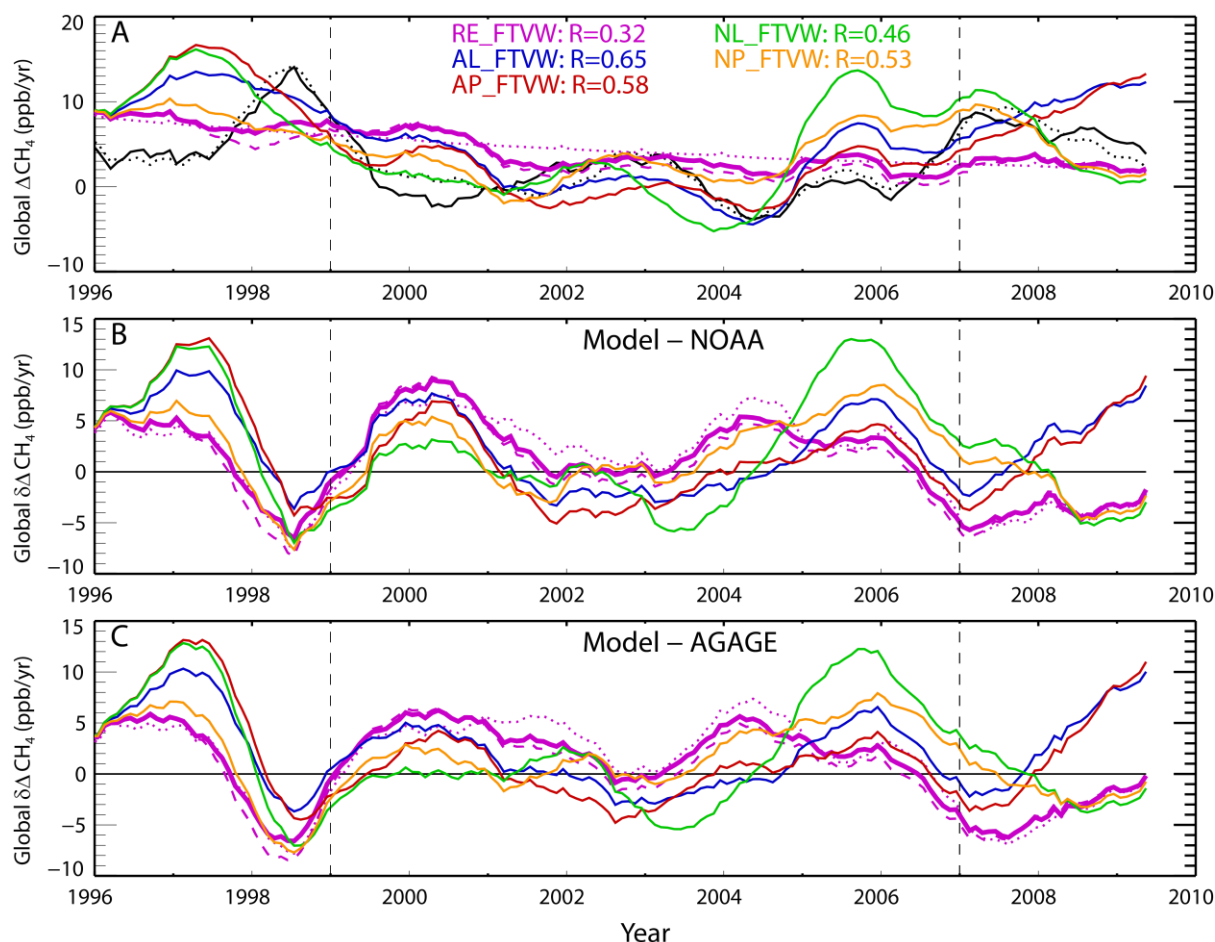


Figure 5. (a) The smoothed variation in the global annual CH_4 growth rate (ppb/yr) derived from NOAA (black solid) and AGAGE (black dashed) observations. Also shown are the smoothed growth rates from five TOMCAT 3-D CTM simulations with fixed temperatures and varying winds (FTVW, see Table 1). Values in legend give correlation coefficient between model run and NOAA observations. Also shown are results from runs RE_FTVW and RE_VTVW as a purple dotted line and dashed line, respectively. (b) The difference in smoothed growth rate between TOMCAT simulations and NOAA observations shown in panel (a). (c) Same as (b) except using differences compared to AGAGE observations. The vertical dashed lines mark the start and end of the stagnation period in the observed CH_4 growth rate (1999 – 2006).