

1 Authors' response to Referees

2

3 Dear ACP Referees,

4 We thank you for your thorough work and valuable comments. We have made most of the
5 suggested changes; where this was not possible, a more thorough explanation is given below.

6

7 *Referee 1*

8 *Comments from Referee*

9 General comments: One of the motivations for this study is non-stomatal ozone deposition.
10 Significant non-stomatal ozone fluxes have been frequently observed, but a general
11 explanation has not been given yet. While there are no obvious reaction sites for ozone
12 reaction in leaf surface waxes, dissolved or attached terpenes could react effectively with
13 ozone. This idea has been around for a while, but although it was not entirely supported by
14 first experiments, a thorough characterization of the system is still missing and might enable
15 more successful experiments in the future. From this background, the present contribution
16 adds importantly to the knowledge of the fate of terpenes shortly after synthesis and their
17 possible role in ozone deposition.

18 The authors compare the composition of mono- and sesquiterpenes emitted from pine shoots
19 with their abundance in needle waxes. There are common compounds but also some
20 compounds which appear only in one of the compartments. Alternative ways of transport are
21 discussed. Although the general message of the manuscript is clear, the presentation of the
22 results doesn't seem appropriate to me. Fig. 1 C shows results for compounds where no
23 detection limits are given (α -humulene, aromadendrene). Especially the part with missing
24 standards for some of the compounds measured remains weak and the high amount of
25 sesquiterpenes claimed (up to 50%) doesn't seem to be sufficiently corroborated. Which of
26 the three mentioned compounds (cadinene, cubebene, murolene) would be most abundant?

27 There were also very high differences between repetitions of the same tree (e.g., Tree 2,
28 myrcene: 9, 6, and 372g m⁻²; Tree 4, limonene: 15, 355, and 60g m⁻²). While notable
29 variations between the emissions are mentioned in the discussion, these differences are not

1 discussed. Have similarly large differences been reported before or how could they be
2 explained? Could this be an indication that the solvent was not equally effective?

3 Specific comments: Due to the indicated artifact and some other unexplained structures,
4 Figure 1C is not very fortunate. It should be possible to find a better series of photographs, or
5 sketch to illustrate the relevant features.

6 P, 5, L. 21: Were the 'handheld pumps' operated by persons and how could they do this
7 evenly for 30 minutes? If they were machine controlled, why were they handheld?

8 Technical corrections: P. 2, l. 30: 'or', not 'on'

9

10 ***Response to comments***

11 *The authors compare the composition of mono- and sesquiterpenes emitted from pine shoots*
12 *with their abundance in needle waxes. There are common compounds but also some*
13 *compounds which appear only in one of the compartments. Alternative ways of transport are*
14 *discussed. Although the general message of the manuscript is clear, the presentation of the*
15 *results doesn't seem appropriate to me. Fig. 1 C shows results for compounds where no*
16 *detection limits are given (a-humulene, aromadendrene). Especially the part with missing*
17 *standards for some of the compounds measured remains weak and the high amount of*
18 *sesquiterpenes claimed (up to 50%) doesn't seem to be sufficiently corroborated. Which of the*
19 *three mentioned compounds (cadinene, cubebene, murolene) would be most abundant?*

20 This comment points to Fig 1 C, but judged by the content it is meant to be 2 C, and our
21 response is based on this assumption. There are essentially two points in the comment:

22 1) We have now added the missing detection limits for a-humulene and aromadendrene.
23 These compounds do not exist in blank samples, but we calculated the blank levels by
24 integrating background noise of the chromatogram.

25 2) In addition to quantifying compounds known to be emitted from pine shoots, we wanted to
26 search for any indication of possible additional compounds in the waxes. For this reason, we
27 searched the library for candidate compounds for all unidentified large peaks we observed.
28 However, since we did not have the standards, we do not know the actual responses, and an
29 analysis of the possible relative abundances of these compounds is not possible.

30

1 *There were also very high differences between repetitions of the same tree (e.g., Tree 2,*
2 *myrcene: 9, 6, and 372g m⁻²; Tree 4,limonene: 15, 355, and 60g m⁻²). While notable*
3 *variations between the emissions are mentioned in the discussion, these differences are not*
4 *discussed. Have similarly large differences been reported before or how could they be*
5 *explained? Could this be an indication that the solvent was not equally effective?*

6 This is a very valuable comment. The variation in the terpene content of the epicuticular
7 waxes cannot be explained by variation in wax yield (i.e. solvent effectiveness). Even though
8 there is variation in wax yield (per needle area), this variation does not correspond to the
9 variation observed in the terpenes. We do not know of previous studies with similar
10 methodology, so there is nothing to compare to. It is possible that some of the variation was
11 caused by the sampling procedure. Despite the short sampling time, it is possible that the
12 emissions caused by plucking needles had sufficient time to adsorb onto other needles that
13 were subsequently picked into a sample. Other possible causes of variation include small
14 cracks, insect bites or pathogens, in the bark near some of the needles. E.g. insect bites are
15 known to induce both local and systemic terpene emissions (Heijari et al., 2011). Some of
16 these may well have escaped visual inspection. One very likely source is true natural variation
17 between needles grown in different parts of the branch/canopy. Very little is known on this
18 topic, but since terpene synthesis is light-dependent, it is very likely that there are differences
19 (Juho Aalto, personal communication). The wax yields have been added to Appendix B and
20 the possible causes of variation have been discussed more thoroughly.

21

22 *Specific comments: Due to the indicated artifact and some other unexplained structures,*
23 *Figure 1C is not very fortunate. It should be possible to find a better series of photographs, or*
24 *sketch to illustrate the relevant features.*

25 The photographs collaged to produce 1C are quite old, and we know their quality could be
26 better. However, we feel that the image illustrates an important feature not often discussed in
27 literature (the fact that the epicuticular waxes are actually present not just on the surface but
28 all around the epicuticular cell). Unfortunately it is not possible for us to acquire a better
29 photograph, and a drawing would not be sufficiently credible for this purpose.

30

1 *P, 5, L. 21: Were the 'handheld pumps' operated by persons and how could they do this*
2 *evenly for 30 minutes? If they were machine controlled, why were they handheld?*

3 “Handheld” referred actually only to the small size of the pumps; they were battery-operated.
4 The wording has been changes to “small pumps” to avoid unnecessary confusion.

5

6 *Technical corrections: P. 2, l. 30: 'or', not 'on'*

7 These mistakes have been corrected, thank you for noticing them!

8

9 ***Changes made in the manuscript based on these comments***

10 P2 L 30: “on” corrected to “or”

11 P5 L 21: “Small pumps were used to pull the sample through the tube (70 ml/min).” (instead
12 of “handheld pumps”

13 P6 L7-8: Added: “The detection limits were ... 0.05 ng/sample for α -humulene and
14 aromadendrene, ...”

15 P7 L22: Added: “The limits of detection ... were 0.15-0.30 ng/sample for ..., α -humulene,
16 aromadendrene...”

17 P9 L26 New paragraph: “The is remarkable variation observed in the terpene content of the
18 epicuticular waxes, and this variation cannot be explained by variation in the amount of
19 extracted wax. Possible natural causes of variation include small cracks, insect bites or
20 pathogens in the bark near some of the needles. E.g. insect bites are known to induce both
21 local and systemic terpene emissions (Heijari et al., 2011). Some of these may well have
22 escaped visual inspection. One feasible source is true natural variation between needles grown
23 in different parts of the branch or canopy, due to the light-delendent nature of terpene
24 synthesis. Very little is known on this topic, but it is very likely that there are notable
25 differences (Juho Aalto, personal communication). Some of the variation, however, may have
26 been caused by the sampling procedure itself. Despite the short sampling time, it is possible
27 that the emissions caused by plucking needles had sufficient time to adsorb onto other needles
28 that were subsequently picked into a sample. “

29 APPENDIX B: Added: Wax yields

1 References: Added: Heijari, J., Blande, J.D. and Holopainen, J.K. Feeding of large pine
2 weevil on Scots pine stem triggers localised bark and systemic shoot emission of volatile
3 organic compounds, *Environ Exp Bot*, 71, 390-398, 2011.

4

5

6 ***Referee 2***

7 ***Comments from Referee***

8 General comments: Authors have measured whole shoot-level mono- and sesquiterpene
9 emissions of Scots pine seedlings and analysed the needle surface waxes for the same
10 compounds. The aim of the work was to determine if the same terpenes can be found on the
11 epicuticles as in shoot emissions. This approach is needed to better understand the
12 mechanisms how plant release BVOCs in the atmosphere and if there is a temporal storage of
13 BVOCs on plant surfaces.

14 Main observations were that shoot emissions and wax extracts were dominated by
15 monoterpenes and the proportion of some sesquiterpenes was higher in the wax extracts than
16 in whole shoot emissions. Authors have discussed about the pathways of mono- and
17 sesquiterpenes to needle cuticle also considering external sources. Their conclusion was the
18 “any BVOCs found in the extract were most likely not a result of stomatal emissions but
19 rather compounds that had been associated to the epicuticle”. However, whole discussion is
20 based on the assumption that needle emissions are the only source of needle epicuticular
21 mono- and sesquiterpenes. External redeposition is mentioned, but other possible external
22 sources are not discussed. These might include e.g. emissions from the bark of studied
23 branches or other branches and stem, but also emission from neighboring plants. Authors
24 should mention these other pathways of needle deposition of BVOCs.

25 Specific comments:

26 P 4, L24. The analysis is based only on four seedlings, so crafted shoots representing the same
27 genotype was a good choice.

28 P5, L12. Air flow in the shoot chambers was rather high. How much this may stimulate
29 monoterpene and sesquiterpene emission from bark?

1 P 6, L 22-23. Three replicate samples were reported. How they were collected? Was each of
2 those composed of 20 needles or were these 20 needles divided to 3 subsamples?

3 P10, L9. Redeposition plant's own BVOCs on epicuticular waxes might not be the only
4 pathway. Adsorption of sesquiterpenes on epicuticular wax layer from external plant sources
5 and their emission back to atmosphere is reported (Li & Blande 2015). As Scots pine bark is
6 important monoterpene and sesquiterpene emitter (e.g. Ghirardo et al. 2012, Heijari et al.
7 2011). There could be a possibility that part of detected sesquiterpenes on epicuticular wax
8 may originate from earlier sesquiterpene emission from bark of the focal plant and
9 neighboring plants and adsorbed on needles?

10 P 14. Appendix B. Authors should discuss about potential reason for high variation in
11 monoterpene content in replicate samples within each tree. As the same sample has high
12 emission of all common resin monoterpenes (e.g. tree 2 s3 and tree 4 s2), it may suggest e.g.
13 high bark emission from micro cracks near these needles. Together with high sesquiterpene
14 content in some of the needle samples localized biotic stress by e.g. fungal pathogen or mites
15 might also explain these.

16 References

17 Ghirardo et al. (2010) Plant, Cell and Environ. 33, 781–792

18 Li, T & Blande JD, (2015) Global Change Biology 21, 1993–2004

19 Heijari et al. (2011) Environ. Exp. Bot. 71, 390–398.

20

21 ***Response to comments***

22 *Main observations were that shoot emissions and wax extracts were dominated by*
23 *monoterpenes and the proportion of some sesquiterpenes was higher in the wax extracts than*
24 *in whole shoot emissions. Authors have discussed about the pathways of mono- and*
25 *sesquiterpenes to needle cuticle also considering external sources. Their conclusion was the*
26 *“any BVOCs found in the extract were most likely not a result of stomatal emissions but*
27 *rather compounds that had been associated to the epicuticle”. However, whole discussion is*
28 *based on the assumption that needle emissions are the only source of needle epicuticular*
29 *mono- and sesquiterpenes. External redeposition is mentioned, but other possible external*
30 *sources are not discussed. These might include e.g. emissions from the bark of studied*

1 *branches or other branches and stem, but also emission from neighboring plants. Authors*
2 *should mention these other pathways of needle deposition of BVOCs.*

3 It is true that the possible sources of the redeposited terpenes are not discussed; this is indeed
4 a valuable remark. We have added a mention of the possible pathways in the discussion.

5

6 *P5, L12. Air flow in the shoot chambers was rather high. How much this may stimulate*
7 *monoterpene and sesquiterpene emission from bark?*

8 Since the chamber encloses the whole shoot, some of the emissions measured come from the
9 bark/stem of the shoot, not only the needles. This is true of any shoot measurement done with
10 a similar chamber setup. The biomass inside such a chamber is typically 10-25 % wood
11 material (including bark). The needles are more active terpene emitters than the wood/bark,
12 but there is some evidence of compound-specific variation (Anni Vanhatalo, personal
13 communication). It is also likely that the needles are more susceptible to any air current
14 induced disturbance than the bark.

15

16 *P 6, L 22-23. Three replicate samples were reported. How they were collected? Was each of*
17 *those composed of 20 needles or were these 20 needles divided to 3 subsamples?*

18 We took three separate samples of 20 needle pairs each. This information has been added to
19 the text.

20

21 *P10, L9. Redeposition plant's own BVOCs on epicuticular waxes might not be the only*
22 *pathway. Adsorption of sesquiterpenes on epicuticular wax layer from external plant sources*
23 *and their emission back to atmosphere is reported (Li & Blande 2015). As Scots pine bark is*
24 *important monoterpene and sesquiterpene emitter (e.g. Ghirardoet al. 2012, Heijari et al.*
25 *2011). There could be a possibility that part of detected sesquiterpenes on epicuticular wax*
26 *may originate from earlier sesquiterpene emission from bark of the focal plant and*
27 *neighboring plants and adsorbed on needles?*

28 It is true that the possible sources of the redeposited terpenes are not discussed; this is indeed
29 a valuable remark. We have added a mention of the possible pathways in the discussion.

30

1 *P 14. Appendix B. Authors should discuss about potential reason for high variation in*
2 *monoterpene content in replicate samples within each tree. As the same sample has high*
3 *emission of all common resin monoterpenes (e.g. tree 2 s3 and tree 4 s2), it may suggest e.g.*
4 *high bark emission from micro cracks near these needles. Together with high sesquiterpene*
5 *content in some of the needle samples localized biotic stress by e.g. fungal pathogen or mites*
6 *might also explain these.*

7 This is a very valuable comment. The variation in the terpene content of the epicuticular
8 waxes cannot be explained by variation in wax yield (i.e. solvent effectiveness). Even though
9 there is variation in wax yield (per needle area), this variation does not correspond to the
10 variation observed in the terpenes. We do not know of previous studies with similar
11 methodology, so there is nothing to compare to. It is possible that some of the variation was
12 caused by the sampling procedure. Despite the short sampling time, it is possible that the
13 emissions caused by plucking needles had sufficient time to adsorb onto other needles that
14 were subsequently picked into a sample. Other possible causes of variation do indeed include
15 small cracks, insect bites or pathogens, in the bark near some of the needles. Some of these
16 may well have escaped visual inspection. One very likely source is true natural variation
17 between needles grown in different parts of the branch/canopy. Very little is known on this
18 topic, but since terpene synthesis is light-dependent, it is very likely that there are differences
19 (Juho Aalto, personal communication). The wax yields have been added to Appendix B and
20 the possible causes of variation have been discussed more thoroughly.

21

22 *References*

23 *Ghirardo et al. (2010) Plant, Cell and Environ. 33, 781–792*

24 *Li, T & Blande JD, (2015) Global Change Biology 21, 1993–2004*

25 *Heijari et al. (2011) Environ. Exp. Bot. 71, 390–398.*

26

27 These references are an excellent addition to the manuscript. They have been used to improve
28 the text and added in the references list.

29

30

1 *Changes made in the manuscript based on these comments*

2 P6 L 23: Changed to "...and then took needle samples (three separate samples of 20 needle
3 pairs each) in darkness for the wax analysis."

4 P9 L26 New paragraph: "The is remarkable variation observed in the terpene content of the
5 epicuticular waxes, and this variation cannot be explained by variation in the amount of
6 extracted wax. Possible natural causes of variation include small cracks, insect bites or
7 pathogens in the bark near some of the needles. E.g. insect bites are known to induce both
8 local and systemic terpene emissions (Heijari et al., 2011). Some of these may well have
9 escaped visual inspection. One feasible source is true natural variation between needles grown
10 in different parts of the branch or canopy, due to the light-dependent nature of terpene
11 synthesis. Very little is known on this topic, but it is very likely that there are notable
12 differences (Juho Aalto, personal communication). Some of the variation, however, may have
13 been caused by the sampling procedure itself. Despite the short sampling time, it is possible
14 that the emissions caused by plucking needles had sufficient time to adsorb onto other needles
15 that were subsequently picked into a sample. "

16 P11 L8: Rewritten start to first chapter: "In theory, there are three mechanisms for the
17 terpenes produced by a plant to end up on the needle surface. The first one is (dry)
18 redeposition after emission from either the tree itself (needles, bark or other parts) or
19 neighbouring trees. Terpene emission from one plant individual and redeposition onto another
20 has been reported, more markedly for sesqui- than monoterpenes (Himanen et al., 2010, Li
21 and Blande, 2015)."

22 P11 L26: Changed to: "The second option is transport in the aqueous layer ... This route is
23 naturally only available to terpenes produced by the needle itself, and the effectiveness of the
24 route depends on the existence of such a continuous water film..."

25 APPENDIX B: Added: Wax yields

26 References: Added:

27 Ghirardo, A., Koch, K., Taipale, R., Zimmer, I., Schnitzler, J.-P. And Rinne, J. Determination
28 of de novo and pool emissions of terpenes from four common boreal/alpine trees by ¹³C₂O₂
29 labelling and PTR-MS analysis, Plant Cell Environ, 33, 781-792, 2010.

- 1 Heijari, J., Blande, J.D. and Holopainen, J.K. Feeding of large pine weevil on Scots pine stem
2 triggers localised bark and systemic shoot emission of volatile organic compounds, *Environ*
3 *Exp Bot*, 71, 390-398, 2011.
- 4 Himanen, S.J., Blande, J.D., Klemola, T., Pulkkinen, J., Heijari, J., and Holopainen, J.K.
5 Birch (*Betula* spp.) leaves adsorb and re-release volatiles specific to neighbouring plants – a
6 mechanism for associational herbivore resistance? *New Phytol*, 186, 722-732, 2010.
- 7 Li, T. and Blande, J. Associational susceptibility in broccoli: mediated by plant volatiles,
8 impeded by ozone, *Global Change Biol*, 21, 1993-2004, 2015.
- 9

1 Role of needle surface waxes in dynamic exchange of 2 mono- and sesquiterpenes

3 J. Joensuu¹, N. Altimir¹, H. Hakola², M. Rostás³, M. Raivonen⁴, M. Vestenius², H.
4 Aaltonen², M. Riederer⁵ and J. Bäck¹,

5 [1]{Department of Forest Sciences, University of Helsinki, Finland}

6 [2]{Finnish Meteorological Institute, Helsinki, Finland}

7 [3]{Bio-Protection Research Centre, Lincoln University, Christchurch, New Zealand}

8 [4]{Division of Atmospheric Sciences, University of Helsinki, Finland}

9 [5]{Julius-von-Sachs-Institut für Biowissenschaften, University of Würzburg, Germany}

10 Correspondence to: J. Joensuu (johanna.joensuu@helsinki.fi)

11

12 Abstract

13 Biogenic volatile organic compounds (BVOCs) produced by plants have a major role in
14 atmospheric chemistry. The different physicochemical properties of BVOCs affect their
15 transport within and out of the plant as well as their reactions along the way. Some of these
16 compounds may accumulate in or on the waxy surface layer of conifer needles and participate
17 in chemical reactions on or near the foliage surface. The aim of this work was to determine
18 whether terpenes, a key category of BVOCs produced by trees, can be found on the
19 epicuticles of Scots pine (*Pinus sylvestris* L.) and, if so, how they compare with the terpenes
20 found in shoot emissions of the same tree. We measured shoot-level emissions of pine
21 seedlings at a remote outdoor location in Central Finland and subsequently analysed the
22 needle surface waxes for the same compounds. Both emissions and wax extracts were clearly
23 dominated by monoterpenes, but the proportion of sesquiterpenes was higher in the wax
24 extracts. There were also differences in the terpene spectra of the emissions and the wax
25 extracts. The results, therefore, support the existence of BVOC associated to the epicuticular
26 waxes. We briefly discuss the different pathways for terpenes to reach the needle surfaces and
27 the implications for air chemistry.

28

1 1 Introduction

2 At the border of the atmosphere and Earth's ecosystems, the living layer of vegetation is an
3 active player interacting with its surroundings in multiple ways. Plants absorb, transmit and
4 produce compounds like water, oxygen and carbon, as well as a myriad of more complex
5 molecules such as volatile organic compounds (VOCs). In addition to this biological activity,
6 plant surfaces provide area for adsorption, desorption and chemical reactions. These
7 phenomena are affected by both environmental conditions and the structure (species, canopy
8 layers etc.) of the vegetation – in turn shaping itself in response to the environment it grows
9 in. The result of these interactions is an extremely complex and dynamic network of
10 simultaneous processes.

11 Biogenic VOCs (BVOCs) produced by plants have a major role in atmospheric chemistry.
12 They affect the formation and destruction of ozone in the troposphere and participate in
13 aerosol formation processes (e.g. Kulmala et al., 2004, Tunved et al., 2006). Despite
14 considerable progress in recent years, aerosol-related processes are a major source of
15 uncertainty in climate estimates (IPCC 2014). Biogenic VOC emissions dominate over those
16 of anthropogenic origin both globally (Guenther et al., 1995) and in the sparsely populated
17 regions of Northern Europe, especially in the summertime (Simpson et al., 1999, Lindfors et
18 al., 2000).

19 Terpenes (monoterpenes ($C_{10}H_{16}$) and sesquiterpenes ($C_{15}H_{24}$)) represent a reactive subgroup
20 of BVOCs that are produced in different plant tissues and during various physiological
21 processes (e.g. Loreto and Schnitzler, 2010). Plants are known to use these compounds in
22 their interactions with insects and other plants, and they may help the plant to adapt to abiotic
23 stress (see Holopainen and Gershenzon, 2010 for a review). BVOC emissions in the Eurasian
24 taiga are dominated by monoterpenes (Guenther et al., 1995, Tarvainen et al., 2007, Rinne et
25 al., 2009), but boreal forest trees also produce significant amounts of e.g. sesquiterpenes
26 (Hakola et al., 2006, Holzke et al., 2006, Ruuskanen et al., 2007), which are generally more
27 reactive than monoterpenes (Atkinson and Arey, 2003, Appendix A). Many terpenes are
28 produced constitutively, but synthesis can also be induced by biotic and abiotic stresses such
29 as herbivory or heat (Holopainen and Gershenzon, 2010, Loreto and Schnitzler, 2010). Plants
30 store terpenes either in specialised storage structures like the resin canals of conifers ~~or~~
31 nonspecifically in the mesophyll tissue (Niinemets et al., 2004).

1 On their way from the plant interior to the atmosphere, the terpenes, mostly rather lipophilic
2 in nature (Niinemets and Reichstein, 2003, Appendix A), must first cross the lipophilic cell
3 membranes and then the hydrophilic apoplast before evaporating into the air spaces inside the
4 leaf. It was long assumed that this transfer happens purely by diffusion, but new evidence
5 suggests active transport out of the cells (Widhalm et al., 2015). Finally, emission into the
6 atmosphere occurs first by gas-phase diffusion through the stomata and the leaf boundary
7 layer, where the conditions are significantly affected by the leaf (Schuepp, 1993), and then by
8 turbulent transport. The driving force of diffusion is the concentration gradient between the
9 leaf interior and the atmosphere. The leaf cuticle is generally considered an effective barrier
10 for plant-produced volatiles, preventing direct emission (Niinemets and Reichstein, 2003).

11 The different physicochemical properties of terpenes affect their transport within and out of
12 the needle as well as their reactions along the way (Atkinson and Arey, 2003, Niinemets and
13 Reichstein, 2003, Appendix A). For example volatility (described by Henry's law constant H ;
14 $\text{Pa m}^3 \text{mol}^{-1}$) and partitioning between the lipid and aqueous phases (octanol-water partition
15 coefficient K_{OW}) vary between compounds, as do reaction rates with oxidants such as O_3 .

16 Terpenes participate in many chemical reactions at and near the needle surfaces. For example,
17 terpenes can protect the plant from oxidative stressors such as ozone (O_3) by reacting with it
18 before it reaches the sensitive tissues inside the leaves (Loreto and Schnitzler, 2010). BVOC
19 reactions are known to be a major factor in non-stomatal O_3 deposition in forests (Goldstein et
20 al., 2004, Bouvier-Brown et al., 2009). The terpene- O_3 reactions can occur in the atmosphere
21 after terpene emission, but they can also take place in the leaf boundary layer, in the air spaces
22 or aqueous phase inside the leaf – or on the leaf surface (Altimir et al., 2006). In addition to
23 gas-phase reactions, heterogeneous reactions are known to play a key role in BVOC
24 chemistry (Shen et al., 2013). It has been suggested that some of the BVOCs produced by
25 foliage could be attached to the epicuticular waxes (Sabljić et al., 1990, Welke et al., 1998),
26 providing additional protection against oxidants, but scientific knowledge on this issue is
27 currently very limited. At least in theory BVOCs also affect the formation of water films on
28 leaf surfaces (Rudich et al., 2000, Sumner et al., 2004), thereby enhancing O_3 deposition
29 mediated by surface wetness.

30 The surfaces of conifer needles are both complex and dynamic in nature. As they grow,
31 needles are covered with a waxy layer secreted by the epicuticular cells (Fig. 1). This layer is
32 lipophilic and hydrocarbons are known to be taken up in it (Binnie et al., 2002, Brown et al.,

1 1998, Welke et al., 1998). With time and weathering, the surfaces undergo chemical and
2 structural changes (Barnes and Brown, 1990, Huttunen and Laine, 1983). Irregularities in the
3 surface provide sites for water adsorption (Rudich et al., 2000). As a result, the originally
4 water-repellent surface becomes more wettable as it wears down. Compounds accumulating
5 on the surface change the characteristics of both the surface and the water film that forms on it
6 (Neinhuis and Barthlott, 1997, Burkhardt and Eiden, 1994). Such water films are ubiquitous
7 when the ambient relative humidity is above 70 % – a common condition in boreal areas –
8 and can even extend through the stomata, creating a pathway for water-soluble compounds
9 between the leaf inside and the surface (Burkhardt et al., 2012).

10 Thus it is plausible that plant-derived terpenes with varying chemical properties could
11 accumulate on foliage surfaces in amounts and proportions difficult to predict and participate
12 in reactions with other compounds. Because of their importance for both atmospheric
13 chemistry and the plant's adaptation to stress, it is necessary to analyze how the surface
14 processes might change the composition of terpenes reaching the free atmosphere.

15 The aim of this work was to determine whether terpenes can be found on the epicuticles of
16 Scots pine (*Pinus sylvestris* L.) and, if so, to compare the spectra of the terpenes with those
17 found in shoot emissions. To our knowledge this is the first time shoot terpene emissions are
18 compared with terpenes on needle surfaces of the same tree.

19

20 **2 Materials and methods**

21 We measured shoot-level emissions of pine seedlings at a remote outdoor location in Central
22 Finland (Hyytiälä, 61°51'N, 24°17'E). The subsequent needle surface wax analysis was
23 performed in the laboratory of the Finnish Meteorological Institute in Helsinki.

24 The plant material consisted of four grafted Scots pine seedlings, grown for five years in an
25 outdoor plant nursery field. Grafted material was selected to reduce variation in the emissions,
26 since it is well known that the spectrum of terpene emissions depends, among other factors,
27 on the genetic background (Bäck et al., 2012). The height of the seedlings was 1.5–2 m. The
28 trees were transplanted in 15 l plastic pots in May 2013. The plants were kept outdoors in
29 light shade and were well watered. Emission measurements were done during the first days of
30 August. Scots pine terpene emissions have an annual and a diurnal pattern (Hakola et al.,
31 2006, Holzke et al., 2006, Ruuskanen et al., 2007, Aalto et al., 2015); the measurement period

1 was selected to capture sesquiterpene emissions that peak in the summer (Hakola et al., 2006,
2 Tarvainen et al., 2005).

3 We aimed to measure the terpene emissions of each seedling once in similar environmental
4 conditions close to noon and to take three needle samples from each seedling for subsequent
5 wax analysis.

6

7 **2.1 Terpene emissions at shoot level**

8 We measured terpene emissions from the seedlings with a dynamic chamber. The chamber
9 consisted of a steel frame, coated with PTFE tubing, and a FEP bag supported by the frame
10 (volume 4.5 l). The chamber was fitted with an inlet and outlet tube made of PTFE. An
11 external pump, with an active carbon filter and an ozone scrubber, pushed air through the
12 chamber (2.5 l/min). The chamber system is described in more detail in Hakola et al. (2006).

13 A healthy mid-crown branch was selected for the emission measurement. Before
14 measurement, the tip of the branch (approximately 30 cm) was gently fitted in the frame. The
15 measured section included needles grown in 2013 and 2012. The growth of the new needles
16 was not quite complete at the time of measurement. The FEP bag was then pulled over the
17 frame, the pump was started and the system was left to stabilize for 30 minutes to minimize
18 the effect of emissions induced by handling.

19 A sample flow was then directed through adsorbent tubes (Tenax-TA and Carbopack-B)
20 attached to the inlet and outlet tubes with a stainless steel T piece. The resin filling of the tube
21 adsorbs terpenes, which can later be desorbed and analyzed. ~~Handheld-Small~~ pumps were
22 used to pull the sample through the tube (70 ml/min). The sampling time was 30 minutes,
23 after which the chamber was removed. The air temperature inside and the PAR above the
24 chamber were measured during chamber closure with thermistors (Philips KTY 80/110) and
25 quantum sensor (LI-190SZ), respectively. During the 60-minute closure, the temperature
26 inside the chamber increased by 1.5–3 degrees Centigrade. The same chamber was used to
27 measure all the seedlings. To minimize the effect of changing light conditions, the
28 measurements were done between 10 AM and 1 PM, which allowed us to measure one tree
29 per day. Each tree was measured once. After emission measurement and needle sampling (as
30 described below), the measured shoot was cut and weighed for fresh and dry mass. A 10 %
31 subsample was taken and weighed separately. For this subsample, we measured needle

1 dimensions (length, width and thickness) and calculated needle area according to Tirén
2 (1927). This needle area was then used to estimate the needle area for the shoot using the
3 respective dry weights of the subsample and main sample.

4 The contents of the adsorbent tubes were analyzed at the Finnish Meteorological Institute
5 with a thermal desorber (Perkin-Elmer TurboMatrix 650 ATD) connected to a gas
6 chromatograph – mass spectrometer (Perkin-Elmer Clarus 600) with HP-1 column (60 m, i.d.
7 0.25 mm). The detection limits were 0.04 ng/sample for camphene, 0.05 ng/sample for α -
8 humulene and aromadendrene, 0.10–0.15 ng/sample for α -pinene, β -pinene and carene, 0.20–
9 0.42 ng/sample for sabinene, limonene, 1,8-cineol, bornylacetate and β -caryophyllene and
10 0.55–0.64 ng/sample for other sesquiterpenes. The measured compounds were identified
11 using authentic standards and NIST library.

12 The observed emission rate (E , $\mu\text{g}/\text{m}^2/\text{h}$) was calculated based on the two concentrations of
13 each compound as

$$14 \quad E = \frac{(C_2 - C_1)}{A} F \quad (1)$$

15 Where C_2 is the concentration in the outlet air ($\mu\text{g}/\text{m}^3$), C_1 is the concentration in the inlet air
16 ($\mu\text{g}/\text{m}^3$), F is the flow rate into the enclosure (m^3/h) and A is the needle area of the measured
17 shoot (m^2). From E , we obtained the spectra of emitted compounds (% of total emissions).

18

19 **2.2 Terpenes in the epicuticular waxes**

20 To detect the presence of terpenes associated to the epicuticular surfaces, we collected the
21 waxy material from the needle surfaces for subsequent terpene analysis.

22 After each emission measurement, we darkened the measured tree for 30 minutes to close the
23 stomata and minimize stomatal terpene emission and then took needle samples (three separate
24 samples of 20 needle pairs each) in darkness for the wax analysis. The needles were
25 immediately stored in a liquid nitrogen dry shipper until analysis (two weeks later).

26 We collected the epicuticular wax layer by dipping each needle pair in 5 ml dichloromethane
27 for 15 seconds. The dipping time was optimized in a preliminary experiment to remove most
28 of the wax layer but to keep the solvent from reaching the inside of the needle through
29 stomata (visual inspection under a stereo microscope). We took special care to use only intact

1 needles and to not immerse the cut base of the needle in the solvent. This was done to prevent
2 compounds originating inside the needle from getting into the extract. Dipping the needles
3 while they were frozen should also minimize the extraction of compounds from inside the
4 needle. After wax extraction, the needles were weighed for fresh and dry mass and measured
5 for their dimensions (width, length and thickness). From these dimensions, needle surface
6 area was approximated according to Tirén (1927).

7 The obtained extract was evaporated to 1 ml volume with pure nitrogen gas. The reduced
8 extract was then analyzed with a gas chromatograph (Agilent 6890N) with a mass
9 spectrometric detector (Agilent 5973) to identify terpenes. A JandW DB-5MS column (30 m,
10 i.d. 0.25 mm) and a 5 m pre-column (Agilent FS) were used for the chromatography. The
11 limits of detection were estimated from the standard deviations of blank samples and were
12 | 0.15-0.30 ng/sample for p-cymene, bornyl acetate, α -humulene, aromadendrene and iso-
13 longifolene, 0.48–0.72 ng/sample for α -pinene, camphene, myrcene, 1,8-cineol and
14 longicyclene and 1.55–2.29 ng/sample for β -pinene, 3-carene and β -caryophyllene. The
15 analysis method is described in more detail in Vestenius et al. (2011). The compounds to be
16 identified were not predetermined, and hence we did not have calibration standards for all of
17 them. Some of the compounds were therefore identified and quantified only tentatively, using
18 the reference from another compound. After the analysis the extract was left to evaporate, and
19 the solid wax residue left in the vial was weighed (Mettler AT2000).

20 For an estimation of the terpenes lost during the evaporation, we performed a separate
21 evaporation test, letting known concentrations of selected terpenes evaporate as described
22 above. The test gave no indication of any significant loss of terpenes associated with the
23 method.

24

25 **3 Results**

26 The weather conditions during the experiment were slightly variable. The first two days
27 (measuring emissions from trees 1 and 2) were relatively warm (+19–21 °C during the
28 measurements) but partly cloudy. The last two days were sunny and warm, especially the last
29 day (+21–24 °C). This deserves notice, since the amount of terpenes emitted by a plant is
30 affected by temperature, irradiation and humidity that on one hand regulate the biosynthetic
31 processes that produce BVOCs and on the other hand affect volatilization and diffusion rates
32 (Lerdau and Gray, 2003, Niinemets et al., 2004, Tarvainen et al., 2005).

1

2 **3.1 Terpenes in shoot emissions**

3 The shoot emissions were clearly dominated by monoterpenes (96–98 % of total terpene
4 emissions, Fig. 2). Sesquiterpenes amounted to 0–2 % of total emissions. The compounds
5 found in each group and the variation in their emissions are presented in detail in Appendix B
6 and Fig. 2.

7 The most abundant monoterpenes were α -pinene (36–58 % of total emissions), myrcene (13–
8 36 %) and carene (12–18 %). The emitted sesquiterpenes included α -humulene (0–1 % of
9 total emissions), aromadendrene (0–0.5 %) and longicyclene (0–0.8 %). None of the
10 identified sesquiterpenes was detected in the emissions of all four pine seedlings, and one
11 seedling showed no sesquiterpene emission. In addition, 1,8-cineol was observed in the
12 emissions, as was a small percentage of bornyl acetate.

13

14 **3.2 Terpenes in epicuticular waxes**

15 The wax yield from the pine needles was 0.0066–0.0114 g/ g DW (average 0.0075 g/g) or
16 0.43–1.23 g/m² of needles (average 0.76 g/m²) (Appendix B). As for the shoot emissions, the
17 epicuticular wax extracts were dominated by monoterpenes (76–93 % of total terpene
18 amount). The proportion of sesquiterpenes, however, was notably higher than in emissions:
19 5–21 %. Taking into account the six unidentified sesquiterpenes for which we did not have
20 standards for (described below), the proportion of total sesquiterpenes in the waxes rises to 7–
21 50 % (average 34 %).

22 The results for different compounds were highly variable also in the wax analysis (Appendix
23 B). The variation in the terpene content of the epicuticular waxes cannot be explained by
24 variation in wax yield. Even though there is variation in wax yield (per needle area), this
25 variation does not correspond to the variation observed in the terpenes. The most abundant
26 monoterpenes in the waxes were α -pinene (10–57 % of total), carene (11–26 %) and limonene
27 (2–40 %) (Fig. 2). For sesquiterpenes, the highest amounts were measured for β -
28 caryophyllene (4–16 % of total), iso-longifolene (0–9 %) and humulene (0.5–3 %). Of the
29 sesquiterpenes seen in shoot emissions, only α -humulene was found in the surface waxes. Iso-
30 longifolene was found in the waxes but not in emissions. In addition to the pre-selected

1 compounds (with standards available), we detected six unidentified sesquiterpenes, some in
2 relatively high proportions. This group is likely to include cadinene, cubebene and murolene.
3 Also 1,8-cineol was found in the waxes, but in much smaller proportion than in emissions.

4

5 **4 Discussion**

6 **4.1 The terpene spectra in emissions and pine epicuticular waxes**

7 The composition of the emitted pine shoot terpenes measured in this study is generally in the
8 range observed by others (Bäck et al., 2012, Hakola et al., 2006, Holzke et al., 2006,
9 Tarvainen et al., 2005), allowing for the natural variation in BVOC emission and the
10 differences in methodology. The pine seedlings in our study emitted more than twice as much
11 α -pinene than carene, thus representing the pinene or intermediate chemotype described in
12 Bäck et al. (2012). The fact that the pine seedlings were grafted (genetically identical
13 canopies) is likely to have reduced the variation in the results. Grafted seedlings have the
14 advantage of providing, at least in theory, identical replicates that should only show variation
15 caused by differences either in the environmental conditions or life histories (mechanical
16 injuries, insect attacks and similar). Nevertheless, notable variation in the emissions was
17 observed, underlining the importance of the effects of varying conditions and life history
18 experienced by individual trees on their terpene emissions.

19 The amount of terpenes found in the epicuticular waxes is the equivalent to 4–84 hours of the
20 measured emissions for the same compound (per m² of needle surface), depending on the
21 compound. For example, it would take the shoot on average 14 h to emit the amount of α -
22 pinene that was present on the needle surfaces. For myrcene the time would be 9 hours, for
23 carene 24 hours and for limonene 84 hours. For most sesquiterpenes this comparison cannot
24 be done, because they were found in either only emissions or only epicuticular waxes, but for
25 α -humulene the equivalent time would be 34 hours.

26 The is remarkable variation observed in the terpene content of the epicuticular waxes, and this
27 variation cannot be explained by variation in the amount of extracted wax. Possible natural
28 causes of variation include small cracks, insect bites or pathogens in the bark near some of the
29 needles. E.g. insect bites are known to induce both local and systemic terpene emissions
30 (Heijari et al., 2011). Some of these may well have escaped visual inspection. One feasible
31 source is true natural variation between needles grown in different parts of the branch or

1 canopy, due to the light-dependent nature of terpene synthesis. Very little is known on this
2 topic, but it is very likely that there are notable differences (Juho Aalto, personal
3 communication). Some of the variation, however, may have been caused by the sampling
4 procedure itself. Despite the short sampling time, it is possible that the emissions caused by
5 plucking needles had sufficient time to adsorb onto other needles that were subsequently
6 picked into a sample.

7 The short exposure to the solvent and the fact that the stomata were virtually closed means
8 that any BVOCs found in the extract were most likely not a result of stomatal emissions but
9 rather compounds that had been associated to the epicuticle. In studies with extracts from
10 crushed needles, the proportion of mono- and sesquiterpenes has been found to be in the same
11 range as observed here for both emissions and epicuticular waxes. For example Manninen et
12 al., (2002) reported a mean total monoterpene ratio of 67 % for a Scots pine provenance from
13 central Finland and listed α -pinene and carene as the major monoterpenes in the needles. In
14 our study, these two were among the main compounds in both emissions and waxes.
15 Achotegui-Castells et al. (2013) reported camphene, α -pinene, β -pinene, β -caryophyllene and
16 germacrene D as the most abundant terpenes in Scots pine needles. Limonene, in our study
17 the third most abundant compound in waxes, was notably less abundant in whole needles
18 (Achetegui-Castells et al., 2013, Manninen et al., 2002). On the other hand camphene was
19 relatively more abundant both in the whole-needle extracts (Achetegui-Castells et al., 2013,
20 Manninen et al., 2002) and in the emissions in our study than in the needle waxes. This is a
21 strong indication that the solvent used in our study did not reach the needle interior during the
22 procedure.

23 In the epicuticular waxes, we observed six unidentified sesquiterpenes, some in relatively
24 high proportions. Although this group is likely to include cadinene, cubebene and murolene,
25 the exact identification and quantification of these compounds would require a more detailed
26 study. Naturally, the possible role of these compounds in the emissions remains unknown, but
27 their existence in the waxes suggests that the production of sesquiterpenes in Scots pine
28 deserves more attention.

29 It is interesting to note that despite the large variation there is some indication that the most
30 water-soluble compound in our study, 1,8-cineol, (Appendix A) was relatively more abundant
31 in the emissions, while the compounds with a large K_{OW} (more likely to partition into the lipid
32 than the water phase), like α -humulene, β -caryophyllene and iso-longifolene, were relatively

1 more abundant in the surface waxes. This finding is in line with the results of Welke et al.
2 (1998), who found the cuticular matrix to be a much stronger sink for limonene than for
3 isoprene from air. The compounds with the highest reactivities towards ozone (α -humulene
4 and β -caryophyllene; Appendix A) were more abundant in the epicuticular waxes than
5 emissions. Since the inlet air used in our experiment was scrubbed of ozone, the result is not
6 due to O₃-VOC reactions inside the chamber.

8 4.2 The fate of terpenes on leaf surfaces

9 In theory, there are three mechanisms for the terpenes produced by a needle-plant to end up on
10 the needle surface. The first one is (dry) redeposition after emission from either the tree itself
11 (needles, bark or other parts) or neighbouring trees. Terpene emission from one plant
12 individual and redeposition onto another has been reported, more markedly for sesqui- than
13 monoterpenes (Himanen et al, 2010, Li and Blande, 2015).—This route is more likely for the
14 less volatile terpenes like longicyclene and p-cymene (Appendix A). The most lipophilic
15 terpenes, such as β -caryophyllene and α -humulene, are also the most reactive ones. Although
16 they are more likely to bind into or onto the lipophilic wax layer, they are also most unlikely
17 to survive in the air phase long enough for redeposition to happen (Atkinson and Arey, 2003).
18 The observed spectra, with β -caryophyllene observed in the pine epicuticular waxes but not in
19 the shoot emissions and with α -humulene being relatively more abundant in the waxes than
20 the emissions, are an indication that this route can be considered of minor importance. This
21 conclusion is supported by Cape et al. (2009), who observed that α -pinene did not dissolve or
22 adsorb into a wax layer to enhance O₃ removal. Another way for the emitted compounds to
23 bind onto the epicuticular waxes after emission into the air is absorption or adsorption into or
24 onto the layer of water on the surface.

25 The second option is transport in the aqueous layer extending from the outer needle surface
26 through the stoma all the way into the substomatal cavity, as suggested by Burkhardt et al.
27 (2012). This route is naturally only available to terpenes produced by the needle itself, and the
28 The-effectiveness of this-the route depends on the existence of such a continuous water film,
29 and also on the water-solubility and diffusion capabilities in water of the compound in
30 question. Because of their low water solubility, it has often been assumed that the reactions of
31 terpenes in the aqueous phase do not contribute significantly to the total reactions. Wang et al.

1 (2012) however propose that the reactions of biogenic unsaturated hydrocarbons happening
2 on wet surfaces, like those of plants growing in nature, can have a significant effect on ozone
3 deposition. In this work, we cannot differentiate between compounds that were in or on the
4 epicuticular waxes from those that may have been bound in the surface water. The most
5 water-soluble of the detected compounds was 1,8-cineol, which was present in greater
6 proportion in shoot emissions than epicuticular waxes. It is then possible that some of the 1,8-
7 cineol emitted from the shoot is redeposited onto the surface.

8 The third alternative is direct transport from the production sites inside the cells through the
9 plant cuticle. In xerophytic plants, such as conifers, the cuticle has a strongly layered
10 structure. The insoluble lipid cutin is partly embedded as intracuticular wax under the cuticle
11 proper, not as an even layer but forming legs towards the epidermal cell wall (Evert, 2007,
12 Fig. 1). The production of surface waxes takes place in epidermal cells during the first few
13 weeks and months of needle growth (Kinnunen et al, 1998), and they are transported via
14 microchannels or diffusion to the surface (Evert, 2007). Despite some reports of terpene
15 emissions through the cuticle (e.g. Guenther et al, 1991), this route is usually considered
16 negligible for terpene emissions (Niinemets and Reichstein, 2003) because of the
17 considerably slower diffusion rate of terpenes within the cuticle than in air or water. It does
18 not, however, contradict the notion that terpenes might be transported into the epicuticulum
19 and accumulate there. Theoretically, this mode of transport would be more effective for the
20 most lipophilic compounds like α -humulene and β -caryophyllene (Kirsch et al., 1997,
21 Appendix A). Indeed, these compounds were found in greater proportion in the waxes than in
22 the emissions, suggesting that this may be an important pathway for lipophilic terpenoids.

23

24 **4.3 Implications for gas-phase chemistry**

25 Once in the gas phase, plant-emitted terpenes can react in various ways. They can undergo
26 photolysis or react with hydroxyl or nitrate radicals or ozone (Atkinson and Arey, 2003). The
27 relative importance of the different reaction pathways depends on atmospheric conditions,
28 time of day and the compound in question. Ozone reactions target double bonds in the terpene
29 molecule (Atkinson and Arey, 2003). The most O₃-reactive compounds have two or three of
30 these double bonds in their structure (Atkinson and Arey, 2003, Appendix A).

1 The available reaction rate coefficients for O₃-BVOC reactions are almost exclusively for the
2 gas phase (Appendix A). This makes rate calculations of reactions happening on wet plant
3 surfaces challenging. There is evidence that the reaction rates of terpenes on solid and liquid
4 surfaces can be faster than in the gas phase (Shen et al., 2013, Enami et al., 2010), but because
5 of the almost unlimited variation in surface properties and aqueous solutions found in nature,
6 a single coefficient will never cover all circumstances. For a rough estimate of the O₃
7 scavenging potential of the terpenes we found on the needle surfaces, we calculated their
8 “reaction time” or how many hours worth of non-stomatal deposition of O₃ each compound
9 could react with, assuming there were no other sinks, as

10

$$11 \quad Time = \frac{n_{terp}}{dep_{O_3}} \quad (2)$$

12 Where Time is the reaction time (h), n_{terp} is the amount of the terpene in question ($\mu\text{g}/\text{m}^2$) and
13 dep_{O_3} ($\mu\text{g}/\text{m}^2/\text{h}$) is non-stomatal deposition towards the shoot.

14

15 Similarly to Fares et al. (2012), we assumed that each molecule of any terpene can react with
16 one molecule of O₃, even though some terpenes have more than one double bond available
17 while others have none. Assuming a total O₃ deposition of 30 $\text{ng}/\text{m}^2/\text{s}$ towards the shoot with
18 40 % non-stomatal deposition (realistic values for Scots pine in the area in the summer as
19 reported by Altimir et al., 2006), the terpenes present on the surfaces could in theory react
20 with 5 hours of nonstomatal O₃ deposition.

21 Although simple, our calculation shows that the terpenes found in needle surface waxes could
22 act as a significant O₃ sink. The extent to which this actually happens depends on two factors:
23 how much of the atmospheric ozone reaches the terpenes within a given time, and how fast
24 the terpene supply is replenished. The fact that reactive terpenes were present on the needle
25 surfaces indicates that under the conditions of this study, the terpene supply is renewed at
26 least at a rate comparable to the amount of O₃ reaching the storage site. Assessing these
27 factors would present an interesting question for future research.

28

1 Appendix A: Physicochemical properties of BVOCs (at 25 °C)

2 The values for molecular mass (M), water solubility, Henry's law constant (H), saturated
 3 vapor pressure (VP) and octanol-water partition coefficient (KOW) from Copolovici and
 4 Niinemets (2005) unless otherwise marked. Reaction rate constants from Shu and Atkinson
 5 (1995) unless otherwise marked.

	M g/mol	Water sol. mol/m ³	H Pa m ³ /mol	VP Pa	log[KOW] mol/mol	Rate constants for gas phase reactions, cm ³ /molec s		
						OH	O ₃	NO ₃
p-cymene	134.2	0.21	935	197**	4.1	8.5x10 ⁻¹² ***		
α-pinene	136.2	0.0411	13590	558**	4.66	5.4x10 ⁻¹¹ *	8.7x10 ⁻¹⁷ *	6.1x10 ⁻¹² *
β-pinene	136.2	0.0592	6826	404**	4.42	5.7x10 ⁻¹¹ ***	1.2x10 ⁻¹⁷ ***	
camphene	136.2	0.0419	3238	136*	4.56	5.7x10 ⁻¹¹ ***	1.1x10 ⁻¹⁷ ***	
Δ ³ -carene	136.2	0.0214	13640 *	292*	4.61	8.8x10 ⁻¹¹ *	3.7x10 ⁻¹⁷ *	9.1x10 ⁻¹² *
limonene	136.2	0.0886	2850	253*	4.49	1.5x10 ⁻¹⁰ ***	4.4x10 ⁻¹⁶ ***	
myrcene	136.2	0.0421	6300	265*	4.34	1.9x10 ⁻¹⁰ ***	4.4x10 ⁻¹⁶ ***	
1,8-cineole	154.2	19.1	13.27	253*	2.61	2.3x10 ⁻¹¹ ***		
bornyl acetate	196.3	0.118 **	44.3 ***	30.4***	3.86 **	7.7x10 ⁻¹² ***		
longicyclene	204.4	0.966 ***	2422 ***	11.5***	5.60 ***	9.4x10 ⁻¹² ***		
iso-longifolene	204.4	0.375 ***	25939 ***	6.4***	6.12 ***	9.6x10 ⁻¹¹ ***	1.1x10 ⁻¹⁷ ***	
β-caryophyllene	204.4	0.245 ***	69914 ***	4.2***	6.30 ***	2.0x10 ⁻¹⁰	1.2x10 ⁻¹⁴	1.9x10 ⁻¹¹
aromadendrene	204.4	0.345 ***	29688 ***	5.3***	6.13 ***	6.2x10 ⁻¹¹ ***	1.2x10 ⁻¹⁷ ***	
α-humulene	204.4	0.0683 ***	165160 ***	2.0***	6.95 ***	2.9x10 ⁻¹⁰	1.2x10 ⁻¹⁴	3.5x10 ⁻¹¹

6
 7 Water sol., H, VP, log[KOW]: *)Niinemets and Reichstein (2002) **)Niinemets and Reichstein (2003)
 8 ***)ChemSpider. Reaction rate constants: *)Rinne et al., (2007) ***)ChemSpider.

9
 10
 11
 12

1 Appendix B: BVOCs in shoot emissions and surface waxes

		a-pinene	myrcene	carene	limonene	b-pinene	camphene	p-cymene	monoterpenes total	a-humulene	aromadendrene	longicyclene	iso-longifolene	b-caryophyllene	unknown1	unknown2	unknown3	unknown4	unknown5	unknown6	sesquiterpenes total	1,8-cineol	bornyacetate	others total	TOTAL	
Emissions, $\mu\text{g}/\text{m}^2/\text{h}$	Tree 1	6.8	6.7	2.6	0.0	1.6	0.8	0.0	18.4	0.0	0.0	0.0	0.0	0.0							0.0	0.4	0.0	0.4	18.8	
	Tree 2	12.4	4.9	3.4	0.7	2.0	3.1	0.0	26.5	0.2	0.0	0.2	0.0	0.0							0.5	0.4	0.1	0.5	27.5	
	Tree 3	13.2	3.6	4.0	0.0	1.0	0.7	0.0	22.5	0.3	0.0	0.0	0.0	0.0							0.3	0.1	0.0	0.1	22.9	
	Tree 4	20.0	6.0	7.9	5.3	3.3	1.2	0.4	44.1	0.2	0.2	0.0	0.0	0.0							0.4	0.4	0.1	0.5	45.0	
	Min	6.8	3.6	2.6	0.0	1.0	0.7	0.0	18.4	0.0	0.0	0.0	0.0	0.0							0.0	0.1	0.0	0.1	18.8	
	Max	20.0	6.7	7.9	5.3	3.3	3.1	0.4	44.1	0.3	0.2	0.2	0.0	0.0							0.5	0.4	0.1	0.5	45.0	
	Mean	13.1	5.3	4.4	1.5	2.0	1.4	0.1	27.9	0.2	0.1	0.1	0.0	0.0							0.3	0.4	0.0	0.4	28.6	
	SD	5.4	1.4	2.4	2.5	1.0	1.2	0.2	11.3	0.1	0.1	0.1	0.0	0.0							0.2	0.2	0.0	0.2	11.5	
In waxes, $\mu\text{g}/\text{m}^2$	Tree 1 s 1	62.9	1.0	29.7	2.3	0.0	2.9	0.0	98.8	1.1	0.0	0.0	0.2	9.2	0.2	3.5	21.4	8.5	0.0	2.7	46.8	0.2	3.5	3.7	149	
	Tree 1 s 2	408	3.6	147	44.2	21.1	10.0	9.1	642	11.8	0.0	0.0	39.8	83.2	13.0	24.2	158	104	2.6	26.8	464	3.3	10.1	13.3	1120	
	Tree 1 s 3	20.1	2.5	9.9	6.2	0.0	0.0	0.1	38.8	0.4	0.0	0.0	0.3	3.0	0.1	2.3	16.8	8.5	0.0	0.8	32.1	0.0	1.1	1.1	72.1	
	Tree 2 s 1	120	9.3	39.8	20.7	0.0	3.9	0.5	194	5.2	0.0	0.0	7.7	39.8	5.3	12.8	62.5	43.4	1.3	17.3	195	1.1	3.8	4.9	394	
	Tree 2 s 2	59.0	5.8	32.2	18.2	11.9	4.4	0.5	132	4.8	0.0	0.0	1.3	25.0	2.7	5.0	29.4	10.8	1.3	14.7	94.9	1.1	10.2	11.2	238	
	Tree 2 s 3	213	372	463	856	83.9	0.0	0.0	1988	14.5	0.0	0.0	0.0	112	0.0	3.1	18.7	5.7	0.0	1.9	156	18.7	3.9	22.6	2166	
	Tree 3 s 1	152	21.6	71.9	61.2	8.4	6.9	1.1	324	4.4	0.0	0.0	3.4	36.0	2.2	8.1	48.1	17.7	0.6	11.4	132	2.0	5.6	7.6	463	
	Tree 3 s 2	76.3	11.6	25.8	38.7	9.8	7.0	1.9	171	2.2	0.0	0.0	4.5	14.3	0.6	6.8	49.9	15.0	1.1	1.9	96.4	1.3	5.4	6.7	274	
	Tree 3 s 3	305	22.4	132	62.4	12.5	11.0	1.5	547	11.2	0.0	0.0	25.0	83.2	12.0	21.8	108	61.8	4.6	48.9	376	2.6	12.4	15.0	938	
	Tree 4 s 1	421	7.0	81.6	14.5	20.2	18.1	3.1	565	8.6	0.0	0.0	66.0	69.7	10.9	19.4	159	64.3	3.8	12.3	414	3.3	21.4	24.7	1004	
	Tree 4 s 2	207	101	152	355	39.0	10.0	2.6	867	4.8	0.0	0.0	7.6	37.0	1.6	8.6	70.3	10.7	1.5	0.0	142	8.4	5.9	14.3	1023	
	Tree 4 s 3	82.5	21.6	69.3	60.2	11.4	2.8	0.0	248	4.7	0.0	0.0	1.8	34.4	2.5	8.0	30.9	12.7	1.3	18.9	115	1.7	1.5	3.1	366	
	Min	20.1	1.0	9.9	2.3	0.0	0.0	0.0	38.8	0.4	0.0	0.0	0.0	3.0	0.0	2.3	16.8	5.7	0.0	0.0	32.1	0.0	1.1	1.1	72.1	
	Max	421	372	463	856	83.9	18.1	9.1	1988	14.5	0.0	0.0	66.0	112	13.0	24.2	159	104	4.6	48.9	464	18.7	21.4	24.7	2166	
Mean	177	48.3	105	128	18.2	6.4	1.7	485	6.1	0.0	0.0	13.1	45.5	4.3	10.3	64.4	30.3	1.5	13.2	189	3.6	7.1	10.7	684		
SD	137	106	123	248	23.4	5.2	2.6	537	4.4	0.0	0.0	20.6	33.9	4.9	7.6	51.1	31.4	1.5	14.2	146	5.2	5.7	7.6	598		
Emissions, $\mu\text{g}/\text{m}^2/\text{h}$	Tree 1	6.8	6.7	2.6	0.0	1.6	0.8	0.0	18.4	0.0	0.0	0.0	0.0	0.0							0.0	0.4	0.0	0.4	18.8	
	Tree 2	12.4	4.9	3.4	0.7	2.0	3.1	0.0	26.5	0.2	0.0	0.2	0.0	0.0							0.5	0.4	0.1	0.5	27.5	
	Tree 3	13.2	3.6	4.0	0.0	1.0	0.7	0.0	22.5	0.3	0.0	0.0	0.0	0.0							0.3	0.1	0.0	0.1	22.9	
	Tree 4	20.0	6.0	7.9	5.3	3.3	1.2	0.4	44.1	0.2	0.2	0.0	0.0	0.0							0.4	0.4	0.1	0.5	45.0	
	Min	6.8	3.6	2.6	0.0	1.0	0.7	0.0	18.4	0.0	0.0	0.0	0.0	0.0							0.0	0.1	0.0	0.1	18.8	
	Max	20.0	6.7	7.9	5.3	3.3	3.1	0.4	44.1	0.3	0.2	0.2	0.0	0.0							0.5	0.4	0.1	0.5	45.0	
	Mean	13.1	5.3	4.4	1.5	2.0	1.4	0.1	27.9	0.2	0.1	0.1	0.0	0.0							0.3	0.4	0.0	0.4	28.6	
	SD	5.4	1.4	2.4	2.5	1.0	1.2	0.2	11.3	0.1	0.1	0.1	0.0	0.0							0.2	0.2	0.0	0.2	11.5	
	In waxes, $\mu\text{g}/\text{m}^2$	Tree 1 s 1	62.9	1.0	29.7	2.3	0.0	2.9	0.0	98.8	1.1	0.0	0.0	0.2	9.2	0.2	3.5	21.4	8.5	0.0	2.7	46.8	0.2	3.5	3.7	149
		Tree 1 s 2	408	3.6	147	44.2	21.1	10.0	9.1	642	11.8	0.0	0.0	39.8	83.2	13.0	24.2	158	104	2.6	26.8	464	3.3	10.1	13.3	1120
		Tree 1 s 3	20.1	2.5	9.9	6.2	0.0	0.0	0.1	38.8	0.4	0.0	0.0	0.3	3.0	0.1	2.3	16.8	8.5	0.0	0.8	32.1	0.0	1.1	1.1	72.1
		Tree 2 s 1	120	9.3	39.8	20.7	0.0	3.9	0.5	194	5.2	0.0	0.0	7.7	39.8	5.3	12.8	62.5	43.4	1.3	17.3	195	1.1	3.8	4.9	394
		Tree 2 s 2	59.0	5.8	32.2	18.2	11.9	4.4	0.5	132	4.8	0.0	0.0	1.3	25.0	2.7	5.0	29.4	10.8	1.3	14.7	94.9	1.1	10.2	11.2	238
		Tree 2 s 3	213	372	463	856	83.9	0.0	0.0	1988	14.5	0.0	0.0	0.0	112	0.0	3.1	18.7	5.7	0.0	1.9	156	18.7	3.9	22.6	2166
Tree 3 s 1		152	21.6	71.9	61.2	8.4	6.9	1.1	324	4.4	0.0	0.0	3.4	36.0	2.2	8.1	48.1	17.7	0.6	11.4	132	2.0	5.6	7.6	463	
Tree 3 s 2		76.3	11.6	25.8	38.7	9.8	7.0	1.9	171	2.2	0.0	0.0	4.5	14.3	0.6	6.8	49.9	15.0	1.1	1.9	96.4	1.3	5.4	6.7	274	
Tree 3 s 3		305	22.4	132	62.4	12.5	11.0	1.5	547	11.2	0.0	0.0	25.0	83.2	12.0	21.8	108	61.8	4.6	48.9	376	2.6	12.4	15.0	938	
Tree 4 s 1		421	7.0	81.6	14.5	20.2	18.1	3.1	565	8.6	0.0	0.0	66.0	69.7	10.9	19.4	159	64.3	3.8	12.3	414	3.3	21.4	24.7	1004	
Tree 4 s 2		207	101	152	355	39.0	10.0	2.6	867	4.8	0.0	0.0	7.6	37.0	1.6	8.6	70.3	10.7	1.5	0.0	142	8.4	5.9	14.3	1023	
Tree 4 s 3		82.5	21.6	69.3	60.2	11.4	2.8	0.0	248	4.7	0.0	0.0	1.8	34.4	2.5	8.0	30.9	12.7	1.3	18.9	115	1.7	1.5	3.1	366	
Min		20.1	1.0	9.9	2.3	0.0	0.0	0.0	38.8	0.4	0.0	0.0	0.0	3.0	0.0	2.3	16.8	5.7	0.0	0.0	32.1	0.0	1.1	1.1	72.1	
Max		421	372	463	856	83.9	18.1	9.1	1988	14.5	0.0	0.0	66.0	112	13.0	24.2	159	104	4.6	48.9	464	18.7	21.4	24.7	2166	
Mean	177	48.3	105	128	18.2	6.4	1.7	485	6.1	0.0	0.0	13.1	45.5	4.3	10.3	64.4	30.3	1.5	13.2	189	3.6	7.1	10.7	684		
SD	137	106	123	248	23.4	5.2	2.6	537	4.4	0.0	0.0	20.6	33.9	4.9	7.6	51.1	31.4	1.5	14.2	146	5.2	5.7	7.6	598		

2

</

		a-pinene	myrcene	carene	limonene	b-pinene	camphene	p-cymene	monoterpenes total	a-humulene	aromadendrene	longicyclene	iso-longifolene	b-caryophyllene	sesquiterpenes total	1,8-cineol	bomyacetate	others total	TOTAL
Emissions,	Tree 1	35.9	35.7	13.6	0.0	8.4	4.0	0.0	97.7	0.0	0.0	0.0	0.0	0.0	0.0	2.3	0.0	2.3	100.0
% of total	Tree 2	44.9	17.8	12.3	2.7	7.2	11.4	0.2	96.4	0.8	0.0	0.8	0.0	0.0	1.7	1.6	0.3	1.9	100.0
	Tree 3	57.5	15.8	17.4	0.1	4.5	3.0	0.0	98.3	1.2	0.0	0.0	0.0	0.0	1.2	0.5	0.0	0.5	100.0
	Tree 4	44.4	13.4	17.5	11.8	7.3	2.7	0.8	97.9	0.5	0.5	0.0	0.0	0.0	0.9	1.0	0.2	1.2	100.0
	Min	35.9	13.4	12.3	0.0	4.5	2.7	0.0	96.4	0.0	0.0	0.0	0.0	0.0	0.0	0.5	0.0	0.5	100.0
	Max	57.5	35.7	17.5	11.8	8.4	11.4	0.8	98.3	1.2	0.5	0.8	0.0	0.0	1.7	2.3	0.3	2.3	100.0
	Mean	45.7	20.7	15.2	3.7	6.8	5.3	0.2	97.6	0.6	0.1	0.2	0.0	0.0	1.0	1.4	0.1	1.5	100.0
	SD	8.9	10.2	2.6	5.6	1.7	4.1	0.4	0.8	0.5	0.2	0.4	0.0	0.0	0.7	0.8	0.1	0.8	0.0
In waxes,	Tree 1 s 1	55.7	0.8	26.3	2.0	0.0	2.6	0.0	87.4	0.9	0.0	0.0	0.2	8.1	9.2	0.2	3.1	3.3	100.0
% of total	Tree 1 s 2	51.6	0.5	18.5	5.6	2.7	1.3	1.2	81.3	1.5	0.0	0.0	5.0	10.5	17.1	0.4	1.3	1.7	100.0
	Tree 1 s 3	46.1	5.8	22.6	14.2	0.0	0.0	0.3	89.0	0.8	0.0	0.0	0.8	6.9	8.5	0.0	2.6	2.6	100.0
	Tree 2 s 1	47.7	3.7	15.8	8.2	0.0	1.5	0.2	77.1	2.1	0.0	0.0	3.1	15.8	20.9	0.4	1.5	1.9	100.0
	Tree 2 s 2	33.9	3.3	18.4	10.5	6.8	2.5	0.3	75.7	2.7	0.0	0.0	0.7	14.4	17.8	0.6	5.8	6.4	100.0
	Tree 2 s 3	10.0	17.4	21.7	40.0	3.9	0.0	0.0	93.0	0.7	0.0	0.0	0.0	5.2	5.9	0.9	0.2	1.1	100.0
	Tree 3 s 1	40.7	5.8	19.2	16.3	2.3	1.8	0.3	86.3	1.2	0.0	0.0	0.9	9.6	11.7	0.5	1.5	2.0	100.0
	Tree 3 s 2	38.4	5.8	13.0	19.5	4.9	3.5	0.9	86.1	1.1	0.0	0.0	2.2	7.2	10.6	0.6	2.7	3.4	100.0
	Tree 3 s 3	44.8	3.3	19.4	9.2	1.8	1.6	0.2	80.3	1.6	0.0	0.0	3.7	12.2	17.5	0.4	1.8	2.2	100.0
	Tree 4 s 1	57.3	1.0	11.1	2.0	2.7	2.5	0.4	77.0	1.2	0.0	0.0	9.0	9.5	19.6	0.4	2.9	3.4	100.0
	Tree 4 s 2	22.2	10.9	16.4	38.1	4.2	1.1	0.3	93.2	0.5	0.0	0.0	0.8	4.0	5.3	0.9	0.6	1.5	100.0
	Tree 4 s 3	28.3	7.4	23.7	20.6	3.9	1.0	0.0	84.9	1.6	0.0	0.0	0.6	11.8	14.0	0.6	0.5	1.1	100.0
	Min	10.0	0.5	11.1	2.0	0.0	0.0	0.0	75.7	0.5	0.0	0.0	0.0	4.0	5.3	0.0	0.2	1.1	100.0
	Max	57.3	17.4	26.3	40.0	6.8	3.5	1.2	93.2	2.7	0.0	0.0	9.0	15.8	20.9	0.9	5.8	6.4	100.0
	Mean	39.7	5.5	18.8	15.5	2.8	1.6	0.3	84.3	1.3	0.0	0.0	2.2	9.6	13.2	0.5	2.0	2.5	100.0
	SD	14.1	4.8	4.4	12.6	2.1	1.0	0.4	6.0	0.6	0.0	0.0	2.6	3.6	5.4	0.3	1.5	1.5	0.0

1

2

3 Acknowledgements

4 Anni Vanhatalo, Ditte Mogensen, Theo Kurtén and Pontus Roldin are acknowledged for their
5 valuable help before, during and after the experiment. We thank the Natural Resources
6 Institute Haapastensyrjä unit for the grafted plant material. The research was supported by the
7 Academy of Finland Center of Excellence (grant no. 272041), Maj and Tor Nessling
8 foundation, the Finnish Society of Forest Science and the Doctoral Programme in Sustainable
9 use of renewable natural resources (AGFOREE). N. Altimir thanks VOCBAS for supporting
10 the exchange visit where the initial idea for this study was generated.

11

1 **References**

- 2 Aalto, J, Porcar-Castell, A, Atherton, J, Kolari, P, Pohja, T, Hari, P, Nikinmaa, E, Petäjä, T
3 and Bäck, J. Onset of photosynthesis in spring speeds up monoterpene synthesis and leads to
4 emission bursts, *Plant Cell Environ*, doi:10.1111/pce.12550, 2015.
- 5 Achotegui-Castells, A, Llusà, J, Hódar, J and Peñuelas, J. Needle terpene concentrations and
6 emissions of two coexisting subspecies of Scots pine attacked by the pine processionary moth
7 (*Thaumetopoea pityocampa*), *Acta Physiol Plant*, 35, 3047-3058, 2013.
- 8 Altimir, N., Kolari, P., Tuovinen, J.-P., Vesala, T., Bäck, J., Suni, T., Kulmala, M. and Hari,
9 P. Foliage surface ozone deposition: a role for surface moisture? *Biogeosciences*, 3, 209-228,
10 2006.
- 11 Atkinson, R. and Arey, J. Gas-phase tropospheric chemistry of biogenic volatile organic
12 compounds: a review, *Atmos Environ*, 37, Supplement No 2, S197-S219, 2003.
- 13 Barnes, J.D. and Brown, K.A. The influence of ozone and acid mist on the amount and
14 wettability of the surface waxes in Norway spruce [*Picea abies* (L.) Karst], *New Phytol*, 114,
15 531-535, 1990.
- 16 Binnie, J., Cape, J.N., Mackie, N. and Leith, I.D. Exchange of organic solvents between the
17 atmosphere and grass – the use of open top chambers, *Sci Total Environ*, 285, 53, 2002.
- 18 Bouvier-Brown, N.C., Holzinger, R., Palitzsch, K. and Goldstein, A.H. Large emissions of
19 sesquiterpenes and methyl chavicol quantified from branch enclosure measurements, *Atmos*
20 *Environ*, 43, 389-401, 2009.
- 21 Brown, R.H.A., Cape, J.N. and Farmer, J.G. Partitioning of chlorinated solvents between pine
22 needles and air, *Chemosphere*, 36, 1799-1680, 1998.
- 23 Burkhardt, J. and Eiden, R. Thin water films on coniferous needles, *Atmos Environ* 28, 2001-
24 2011, 1994.
- 25 Burkhardt, J., Basi, S., Pariyar, S. and Hunshe, M. Stomatal penetration by aqueous solutions
26 – an update involving leaf surface particles, *New Phytol*, 196, 774-787, 2012.
- 27 Bäck, J., Aalto, J., Henriksson, M., Hakola, H., He, Q. and Boy, M. Chemodiversity of a
28 Scots pine stand and implications for terpene air concentrations, *Biogeosciences*, 9, 689-702,
29 2012.

1 Cape, J.N., Hamilton, R. and Heal, M.R. Reactive uptake of ozone at simulated leaf surfaces:
2 Implications for “non-stomatal” ozone flux, *Atmos Environ*, 43, 1116-1123, 2009.

3 ChemSpider (www.chemspider.com). Accessed November 15. 2015.

4 Copolovici, L.O. and Niinemets, Ü. Temperature dependencies of Henry’s law constants and
5 ocanol/water partition coefficients for key plant volatile monoterpenoids, *Chemosphere*, 61,
6 1390-1400, 2005.

7 Enami, S., Mishra, H., Hoffmann, M.R. and Colussi, A.J. Protonation and oligomerization of
8 gaseous isoprene on mildly acidic surfaces: implications for atmospheric chemistry, *J of Phys*
9 *Chem*, 116, 6027-6032, 2012.

10 Evert, R.F. *Esau’s Plant Anatomy. Meristems, Cells and Tissues of the Plant Body: Their*
11 *structure, function and development.* Wiley, New Jersey. 2007.

12 Fares, S., Weber, R., Park, J.-H., Gentner, D., Karlik, J. and Goldstein, A.H. Ozone deposition
13 to an orange orchard: Partitioning between stomatal and non-stomatal sinks, *Environ Pollut*,
14 169, 258-266, 2012.

15 [Ghirardo, A., Koch, K., Taipale, R., Zimmer, I., Schnitzler, J.-P. And Rinne, J. Determination](#)
16 [of de novo and pool emissions of terpenes from four common boreal/alpine trees by 13CO2](#)
17 [labelling and PTR-MS analysis, *Plant Cell Environ*, 33, 781-792, 2010.](#)

18 Goldstein, A.H., McKay, M., Kurpius, M.R., Schade, G.W., Lee, A., Holzinger, R. and
19 Rasmussen, R.A. Forest thinning experiment confirms ozone deposition to forest canopy is
20 dominated by reaction with biogenic VOCs, *Geophys Res Lett*, 31, L22106, 2004.

21 Guenther, A. B., Monson, R. K. and Fall, R. Isoprene and monoterpene emission rate
22 variability: Observations with Eucalyptus and emission rate algorithm development, *J*
23 *Geophys.Res.* 96, 10799–10808, 1991.

24 Guenther, A., Hewitt, C.N., Erickson, D., Fall, R., Geron, C., Graedel, T., Harley, P., Klinger,
25 L., Lerdau, M., McKay, W.A., Pierce, T., Scholes, B., Steinbrecher, R., Tallamraju, R.,
26 Taylor, J. and Zimmermann, P. A global model of natural volatile organic compound
27 emissions, *J Geophys Res*, 100, 8873-8892, 1995.

28 Hakola, H., Laurila, T., Lindfors, V., Hellén, H., Gaman, A. and Rinne, J. Variation of the
29 VOC emission rates of birch species during the growing season. *Boreal Environ Res*, 6 237-
30 249, 2001.

- 1 Hakola, H., Tarvainen, V., Bäck, J., Ranta, H., Bonn, B., Rinne, J. and Kulmala, M. Seasonal
2 variation of mono- and sesquiterpene emission rates of Scots pine, *Biogeosciences*, 3, 93-101,
3 2006.
- 4 [Heijari, J., Blande, J.D. and Holopainen, J.K. Feeding of large pine weevil on Scots pine stem](#)
5 [triggers localised bark and systemic shoot emission of volatile organic compounds, *Environ*](#)
6 [*Exp Bot*, 71, 390-398, 2011.](#)
- 7 [Himananen, S.J., Blande, J.D., Klemola, T., Pulkkinen, J., Heijari, J., and Holopainen, J.K.](#)
8 [*Birch \(Betula spp.\) leaves adsorb and re-release volatiles specific to neighbouring plants – a*](#)
9 [*mechanism for associational herbivore resistance? *New Phytol*, 186, 722-732, 2010.*](#)
- 10 Holopainen, J.K. and Gershenson, J. Multiple stress factors and the emission of plant VOCs,
11 *Trends Plant Sci*, 15, 176-184, 2010.
- 12 Holzke, C., Hoffmann, T., Jaeger, L., Koppmann, R. and Zimmer, W. Diurnal and seasonal
13 variation of monoterpene and sesquiterpene emissions from Scots pine (*Pinus sylvestris* L.).
14 *Atmos Environ*, 40, 3174-3185, 2006.
- 15 Huttunen, S. and Laine, K. Effects of air-borne pollutants on the surface wax structure of
16 *Pinus sylvestris* L. needles, *Ann Bot Fenn*, 20, 79-86, 1983.
- 17 IPCC. Climate Change 2014: Synthesis Report. Contribution of Working Groups I, II and III
18 to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change [Core
19 Writing Team, R.K. Pachauri and L.A. Meyer (eds.)]. IPCC, Geneva, Switzerland, 2014.
- 20 Kajos, M.K., Hakola, H., Holst, T., Nieminen, T., Tarvainen, V., Maximov, T., Petäjä, T.,
21 Arneeth, A. and Rinne, J. Terpenoid emissions from fully grown east Siberian *Larix cajanderi*
22 trees, *Biogeosciences*, 10, 4705-4719, 2013.
- 23 Kinnunen, H., Manninen, S., Peura, R., Laakso, K. and Huttunen, S. SEM–EDS image
24 analysis as a tool for scoring the epicuticular wax tube distribution on *Pinus sylvestris*
25 needles—evaluation using a UV-B field experiment. *Chemosphere*, 36, 847-852, 1998.
- 26 Kirsch, T., Kaffarnik, F., Riederer, M. and Schreiber, L. Cuticular permeability of the three
27 tree species *Prunus laurocerasus* L., *Ginkgo biloba* L. and *Juglans Regia* L.: comparative
28 investigation of the transport properties of intact leaves, isolated cuticles and reconstituted
29 cuticular waxes, *J Exp Bot*, 48, 1035-1045, 1997.

1 Kulmala, M., Suni, T., Lehtinen, K.E.J., Dal Maso, M., Boy, M., Reissell, A., Rannik, Ü.,
2 Aalto, P., Keronen, P., Hakola, H., Bäck, J., Hoffmann, T., Vesala, T. and Hari, P. A new
3 feedback mechanism linking forests, aerosols and climate, *Atmos Chem Phys*, 4, 557-562,
4 2004.

5 Lerda, M. and Gray, D. Ecology and evolution of light dependent and light-independent
6 phytogetic volatile organic carbon, *New Phytol*, 157, 199-211, 2003.

7 [Li, T. and Blande, J. Associational susceptibility in broccoli: mediated by plant volatiles,](#)
8 [impeded by ozone, *Global Change Biol*, 21, 1993-2004, 2015.](#)

9 Lindfors, V., Laurila, T., Hakola, H., Steinbrecher, R. and Rinne, J. Modeling speciated
10 terpenoid emissions from the European boreal forest, *Atmos Environ*, 34, 4983-4996, 2000.

11 Loreto, F. and Schnitzler, J.-P. Abiotic stresses and induced BVOCs, *Trends Plant Sci*, 15,
12 154-166, 2010.

13 Manninen, A.-M., Tarhanen, S., Vuorinen, M. and Kainulainen, P. Comparing the variation of
14 needle and wood terpenoids in Scots pine provenances, *J Chem Ecol*, 28, 211-228, 2002.

15 Neinhuis, C. and Barthlott, W. Characterization and distribution of water-repellent, self-
16 cleaning plant surfaces, *Ann Bot*, 79, 667,677, 1997.

17 Niinemets, Ü. and Reichstein, M. A model analysis of the effects of nonspecific
18 monoterpene storage in leaf tissues on emission kinetics and composition in Mediterranean
19 sclerophyllous *Quercus* species, *Global Biogeochem Cy*, VOL. 16, NO. 4, 2002.

20 Niinemets, Ü. and Reichstein, M. Controls on the emission of plant volatiles through stomata:
21 differential sensitivity of emission rates to stomatal closure explained, *J Geophys Res*, 108,
22 issue D7, 2003.

23 Niinemets, Ü., Loreto, F. and Reichstein, M. Physiological and physicochemical controls on
24 foliar volatile organic compound emissions, *Trends Plant Sci*, 4, 180-186, 2004.

25 Rinne, J., Taipale, R., Markkanen, T., Ruuskanen, T.M., Hellén, H., Kajos, M.K., Vesala, T.
26 and Kulmala, M. Hydrocarbon fluxes above a Scots pine canopy: measurements and
27 modelling, *Atmos Chem Phys*, 7, 3361-3372, 2007.

28 Rinne, J. Bäck, J. and Hakola, H. Biogenic volatile organic compounds emissions from the
29 Eurasian taiga: current knowledge and future directions. *Boreal Environ Res*, 14, 807-826,
30 2009.

1 Rudich, Y., Benjamin, I., Naaman, R., Thomas, E., Trakhtenberg, S. and Ussyshkin, R.
2 Wetting of hydrophobic organic surfaces and its implications to organic aerosols in the
3 atmosphere, *J of Phys Chem A*, 104, 5238-5245, 2000.

4 Ruuskanen, T.M., Hakola, H., Kajos, M.K., Hellén, H., Tarvainen, V. and Rinne, J. Volatile
5 organic compound emissions from Siberian larch, *Atmos Environ*, 41, 5807-5812, 2007.

6 Sabljic, A., Güsten, H., Schönherr, J. and Riederer, M. Modeling Plant Uptake of Airborne
7 Organic Chemicals. 1. Plant Cuticle/Water Partitioning and Molecular Connectivity. *Environ*
8 *Sci Technol*, 24, 1321-1326, 1990.

9 Schuepp, P.H. Leaf boundary layers, *New Phytol*, 125, 477-507, 1993.

10 Shen, X., Zhao, Y. and Chen, Z. Heterogeneous reactions of volatile organic compounds in
11 the atmosphere, *Atmos Environ*, 68, 297-314, 2013.

12 Shu, Y. and Atkinson, R. Atmospheric lifetimes and fates of a series of sesquiterpenes, *J*
13 *Geophys Res*, 100, 7275-7281, 1995

14 Simpson, D., Winiwarter, W., Börjesson, G., Cinderby, S., Ferreira, A., Guenther, A., Hewitt,
15 N., Janson, R., Khalil, M.A.K., Owen, S., Pierce, T., Puxbaum, H., Shearer, M., Skiba, U.,
16 Steinbrecher, R., Tarrason, L. and Öquist, M.G. Inventorying emissions from nature in
17 Europe, *J Geophys Res*, 104, 8113-8152, 1999.

18 Sumner, A., Menke, E.J., Dubowski, Y., Newberg, J.T., Penner, R.M., Hemminger, J.C.,
19 Wingen, L.M., Brauers, T. and Finlayson-Pitts, B.J. The nature of water on surfaces of
20 laboratory systems and implications for heterogeneous chemistry in the troposphere. *Phys*
21 *Chem Chem Phys*, 6, 604-613, 2004.

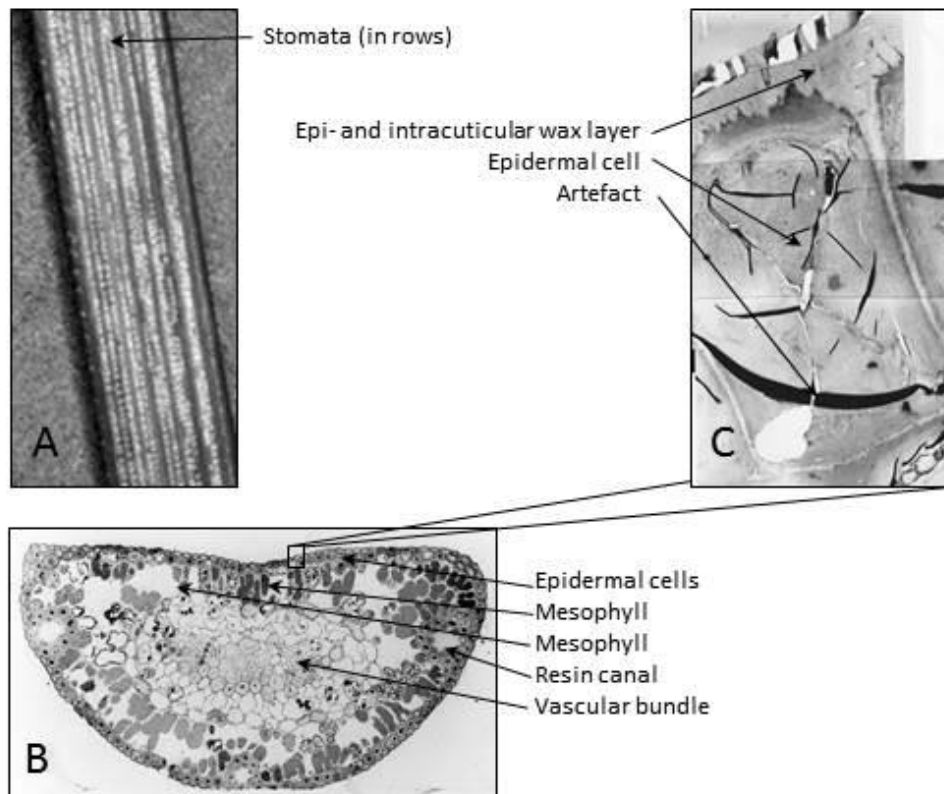
22 Tarvainen, V., Hakola, H., Hellén, H., Bäck, J., Hari, P. and Kulmala, M. Temperature and
23 light dependence of the VOC emissions of Scots pine, *Atmos Chem Phys*, 5, 989-998, 2005.

24 Tarvainen, V., Hakola, H., Rinne, J., Hellén, H. and Haapanala, S. Towards a comprehensive
25 emission inventory of terpenoids from boreal ecosystems, *Tellus B*, 59, 526-534, 2007.

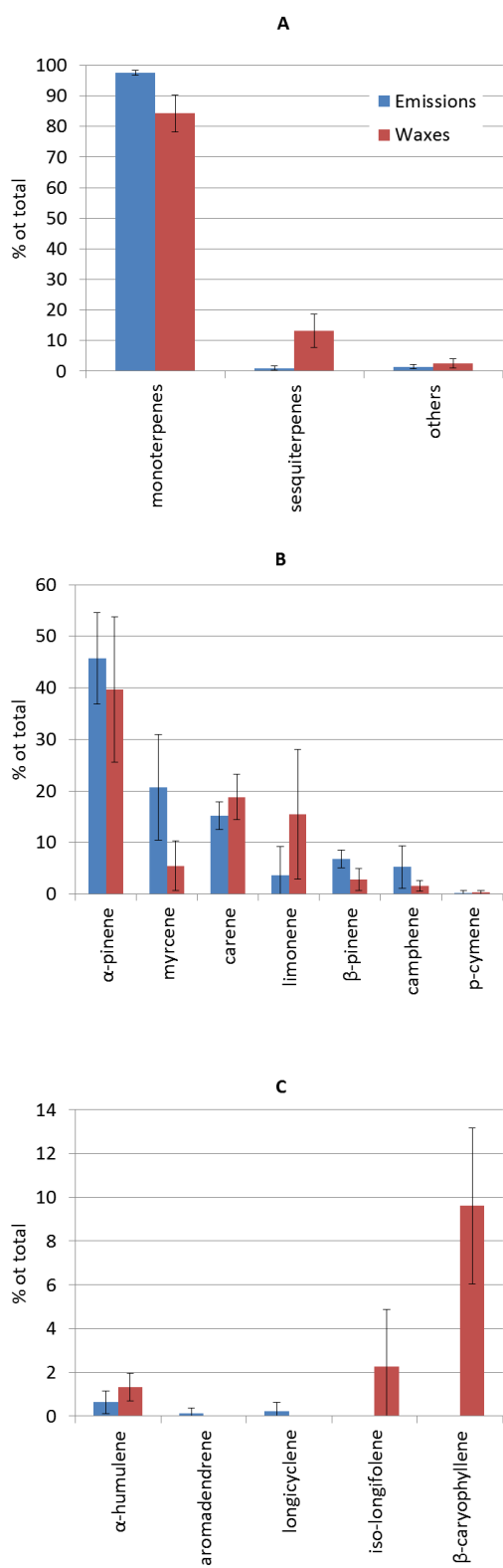
26 Tirén, L. Om barrytans storlek hos tallbestind. *Meddelanden från statens skogsförsöksanstalt*
27 23, 295-336, 1927.

28 Tunved, P., Hansson, H.-C., Kerminen, V.-M., Ström, J., Dal Maso, M., Lihavainen, H.,
29 Viisanen, Y., Aalto, P., Komppula, M. and Kulmala, M. High natural aerosol loading over
30 boreal forests. *Science*, 312, 261-263, 2006.

- 1 Vestenius, M., Leppänen, S., Anttila, P., Kyllönen, K., Htakkal, J., Hellén, H., Hyvärinen, A.-
2 P. and Hakola, H. Background concentrations and source apportionment of polycyclic
3 aromatic hydrocarbons in south-eastern Finland, *Atmos Environ*, 45, 3391-3399, 2011.
- 4 Welke, B., Ettliger, K. and Riederer, M. Sorption of volatile organic chemicals in plant
5 surfaces. *Environ Sci Technol*, 32, 1099-1104, 1998.
- 6 Widhalm, J.R., Jaini, R., Morgan, J.A. and Dudareva, N. Rethinking how volatiles are
7 released from plant cells, *Trends Plant Sci*, doi:10.1016/j.tplants.2015.06.009, 2015.
- 8 Wang, H.L., Huang, D., Zhang, X., Zhao, Y. and Chen, Z.M. Understanding the aqueous
9 phase ozonolysis of isoprene: distinct product distribution and mechanism from the gas phase
10 reaction, *Atmos Chem Phys*, 12, 7187-7198, 2012.
- 11



1
 2 Fig. 1. Pine needle structure. A. The abaxial side of the needle with rows of stomata covered
 3 with epicuticular waxes. B. Cross-section of a needle. C. An epidermal cell with epicuticular
 4 layer.
 5



1

2 Fig. 2. Relative amounts of terpenes in the pine shoot emissions and needle surface waxes,
 3 average % of total, with standard deviation. A: relative abundancies of each compound group,
 4 B: monoterpenes, C: sesquiterpenes. The unknown sesquiterpenes found in the waxes are not
 5 included.