

1 **Distribution and sources of bioaccumulative air**  
2 **pollutants at Mezquital Valley, Mexico, as reflected by**  
3 **the atmospheric plant *Tillandsia recurvata* L.**

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12  
13 **Abstract**

14 Mezquital Valley (MV), a Mexican wastewater-based agricultural and industrial region,  
15 is a “hot spot” of regulated air pollutants emissions, but the concurrent unregulated  
16 ones, like hazardous metals and polycyclic aromatic hydrocarbons (PAH), remain  
17 undocumented. A biomonitoring survey with the epiphytic *Tillandsia recurvata* was  
18 conducted there to detect spatial patterns and potential sources of 20 airborne elements  
19 and 15 PAH. The natural  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  ratios of this plant helped in source  
20 identification. The regional mean concentrations of most elements was two (Cr) to over  
21 40 times (Ni, Pb, V) higher than reported for *Tillandsia* in other countries. Eleven  
22 elements, pyrene and chrysene had 18-214% higher mean concentration at the industrial  
23 south than at the agricultural north of MV. The total quantified PAH (mean, 572 ng g<sup>-1</sup>;  
24 range, 142.6 - 2,568) were composed by medium (65%, phenanthrene to chrysene), low  
25 (28%, naphthalene to fluorene) and high molecular weight compounds (7%,  
26 Benzo(b)fluoranthene to indeno(1,2,3-cd)pyrene). The  $\delta^{13}\text{C}$  (mean, -14.6‰; range, -  
27 15.7‰ to -13.7‰) was lower (<-15‰) near the major petroleum combustion sources.  
28 The  $\delta^{15}\text{N}$  (mean, -3.0‰; range, -9.9‰ to 3.3‰) varied from positive at  
29 agriculture/industrial areas to negative at rural sites. Factor analysis provided a five-  
30 factor solution for 74% of the data variance: 1) crustal rocks, 39.5% (Al, Ba, Cu, Fe, Sr,  
31 Ti); 2) soils, 11.3%, contrasting contributions from natural (Mg, Mn, Zn) and saline  
32 agriculture soils (Na); 3) cement production and fossil fuel combustion, 9.8% (Ca, Ni,

1 V, chrysene, pyrene); 4) probable agricultural biomass burning, 8.1% (K and  
2 benzo(*g,h,i*)perylene), and 5) agriculture with wastewater, 5.2% ( $\delta^{15}\text{N}$  and P). These  
3 results indicated high deposition of bioaccumulative air pollutants at MV, especially at  
4 the industrial area. Since *T. recurvata* reflected the regional differences in exposition, it  
5 is recommended as a biomonitor for comparisons within and among countries where it  
6 is distributed: southern USA to Argentina.

7

## 8 **1 Introduction**

9 Monitoring airborne metals, polycyclic aromatic hydrocarbons (PAH) and other  
10 bioaccumulative compounds with living organisms (biomonitoring) is a technique in  
11 use and refinement since at least the 1960's. It is a multipurpose environmental tool for  
12 exploring pollutant occurrence and dispersion trends at different geographical/time  
13 scales, to identify emission sources, estimate atmospheric deposition and relate  
14 biological/ecological changes to air pollution (Aboal et al., 2006; Conti and Cecchetti,  
15 2001; Pirintsos and Lopi, 2008; Segala et al., 2008; Wolterbeek, 2002). Since  
16 bioaccumulative air pollutants usually travel in particles, biomonitoring is done  
17 preferentially with organisms that rely on the atmosphere as primary source of moisture  
18 and nutrients, such as lichens, mosses and some vascular plants. This dependence is  
19 indicated by positive correlations between the concentration of pollutants in the  
20 biomonitor tissues and the amount of atmospheric deposition (Sloff, 1995; St. Clair et  
21 al., 2002; Wolterbeek, 2002).

22 We report results from a biomonitoring survey of airborne metals and PAH at Mezquital  
23 Valley (MV) conducted in the context of the MILAGRO 2006 field campaign in  
24 Mexico. This region is environmentally better known by severe soil and water pollution  
25 problems caused by over 100 years of agricultural irrigation with untreated sewage  
26 water from Mexico City (Cifuentes et al., 1994; Down et al., 1999; Friedel et al., 2000;  
27 Siebe, 1994; Vázquez-Alarcón et al., 2001). Mezquital Valley is also a major Mexican  
28 "hot spot" in emissions of particulate matter,  $\text{SO}_2$  and other regulated pollutants by the  
29 Tula-Tepeji-Vito industrial corridor located there (SEMARNAT-INE, 2006).  
30 Nevertheless, there is very little information on the expected concurrence of unregulated  
31 air toxics at MV, such as hazardous metals, PAH and other persistent organic pollutants.

1 This survey explored MV for spatial deposition trends and potential sources of 20 trace  
2 and major elements and 15 hazardous PAH using the “ball moss” (*Tillandsia recurvata*  
3 L.) as a natural receptor. This epiphytic Bromeliaceae is very common at the study  
4 region, where it mainly grows on mesquite trees (*Prosopis laevigata* (Willd) M.C.  
5 Johnst.). Adult individuals are spherical, ca. 10-12 cm in diameter, easy to recognize in  
6 the field and collect by hand. They have a reduced stem and non-functional roots. Most  
7 of their biomass is formed by linear leaves profusely covered with absorptive trichomes.  
8 This latter morphological feature increases the ability of this plant to capture moisture  
9 and particles directly from the surrounding air, which justifies naming it as an “air” or  
10 “atmospheric” plant (Schmitt et al. 1989). *Tillandsia recurvata* has some physiological  
11 resistance to high levels of O<sub>3</sub> and SO<sub>2</sub> (Benzig et al., 1992), which may partially  
12 explain why is it so abundant at the polluted MV. This and other atmospheric *Tillandsia*  
13 species have been used as *in situ* air pollution biomonitors in some southern US States  
14 and Latin-American countries, where this genus is exclusively distributed (e.g., De  
15 Sousa et al., 2007; Husk et al., 2004; Pignata et al., 2002; Pyatt et al., 1999; Schrimpf,  
16 1984; Smodiš et al., 2004; Wannaz et al., 2006a,b). Successful transplanting of  
17 *Tillandsia* from rural into urban areas for similar purposes has been reported by  
18 Brighina et al. (2002); Figuereido et al. (2007) and Malm et al. (1998).

19 The natural  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  stable isotope ratios of *T. recurvata* were determined as  
20 additional information to identify regional sources of air pollution. These ratios express  
21 the  $^{13}\text{C}$  to  $^{12}\text{C}$  and  $^{15}\text{N}$  to  $^{14}\text{N}$  proportions in the samples relative to their proportions in  
22 standard materials. They are sensitive to spatial and time variations of the isotopic  
23 composition of the C and N sources used by plants (Bukata and Kyser, 2007; Liu et al.,  
24 2008; Norra et al., 2005; Skinner et al., 2006; Solga et al., 2006). Since most  
25 constitutive plant carbon derives from  $^{12}\text{CO}_2$  fixed directly from the atmosphere by  
26 photosynthesis, plant  $\delta^{13}\text{C}$  ratios are particularly useful in detecting air pollution by the  
27  $^{12}\text{C}$ -rich emissions from fossil fuel combustion. These emissions have significantly  
28 lower  $^{13}\text{CO}_2$ : $^{12}\text{CO}_2$  ratios than unpolluted air, as indicated by  $\delta^{13}\text{C}$  values around -25‰,  
29 -30‰, -40‰ and -8‰ for carbon, petroleum, natural gas and clean air, respectively  
30 (Pichlmayer et al., 1998). The mixing of fossil fuel emissions with clean air is thus  
31 related to the commonly low air  $\delta^{13}\text{C}$  values at urban/industrial areas compared to  
32 rural/remote locations (e.g., -12‰ vs. -8‰, respectively, Boutton, 1991). Because of the  
33 increase in the air concentrations of  $^{12}\text{CO}_2$  caused by fossil fuel emissions and the

1 preferential photosynthetic fixation of this form of carbon dioxide over  $^{13}\text{C}\text{O}_2$ , plants  
2 exposed to such pollution develop lower  $\delta^{13}\text{C}$  values than those growing under clean air.  
3 In contrast, the plant  $\delta^{15}\text{N}$  changes related to air pollution sources cannot be interpreted  
4 straightforward because of the variety of N compounds involved, their emission and  
5 deposition forms, and the complex isotopic fractionation processes occurring within and  
6 between the sources and the plant receptors. The natural  $\delta^{15}\text{N}$  values of land plants  
7 range roughly from -7‰ to +9‰ (Kelly et al., 2005). They can change because of the  
8 isotopic composition of the N sources and the ability of each plant species to  
9 discriminate against  $^{15}\text{N}$ . In general, for plants obtaining nutrients exclusively from  
10 atmospheric sources, this ratio decreases if the deposited N compounds are dominated  
11 by  $\delta^{15}\text{N}$  values lower than theirs. Such compounds are usually reduced N species  
12 abundant at, but not restricted to, agriculture/farming areas: e.g., ammonia ( $\text{NH}_3$ ,  $\delta^{15}\text{N}$ : -  
13 10‰ to -40‰, several sources). Contrarily, this ratio increases in plants predominantly  
14 exposed to compounds with  $\delta^{15}\text{N}$  values higher than theirs, mainly oxidized N species  
15 ( $\text{NO}_x$ ), which are more common to urban/industrial areas because of the heavier use of  
16 fossil fuels (Jung et al., 1997; Skinner et al., 2006; Solga et al., 2005). Since agriculture  
17 and industry coexist at our study area, we expected the *T. recurvata*  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  to  
18 reflect these land uses.

19 A data matrix with 50 MV sites and 38 chemical variables was reduced by standard  
20 univariate statistical techniques. Exploratory factor analysis (FA) of selected variables  
21 and mapping allowed us to detect the major regional emission sources. As far as we  
22 know, this is the first report on the bioaccumulative air pollution of MV using  
23 biomonitoring techniques.

24

## 25 **2 Materials and methods**

### 26 **2.1 Study area**

27 Mezquital Valley is located in Hidalgo State, Mexico, ca. 60 km NW of Mexico City  
28 (Figure 1). It is a semiarid region (ca. 2,429 km<sup>2</sup>) with about 500 thousand inhabitants  
29 distributed in medium-size towns (e.g., Tula and Tepeji del Río) and many small  
30 villages. The mean annual temperature and precipitation range from 16.5°C and 432  
31 mm at the north portion of the valley, which is mainly dedicated to agriculture, to  
32 17.2°C and 647 mm at the industrialized south. The mean elevation at the agriculture

1 flatlands is 2,000 m above mean sea level, and up to 2,400 m in the surrounding  
2 mountains. The original valley's vegetation was a xerophytic shrubland with "mesquite"  
3 tree as dominant element. This vegetation disappeared at the agricultural areas and is  
4 now restricted to the foothills and highlands. Over 90,000 ha of MV are currently  
5 dedicated to agriculture with raw sewage water from Mexico City. This is the major  
6 cause of chemical and biological contamination of the regional soils (Siebe and  
7 Cifuentes 1995). The Tepeji-Tula-Atitalaquia industrial corridor has about 140  
8 industries, including a petroleum refinery (325 thousand barrels/day); a 1500 MW  
9 electricity power plant fueled with residual fuel oil and, secondarily, with natural gas,  
10 and over 20 cement plants. That power plant is the largest emitter of air toxics among  
11 similar Mexican facilities (CEPAL-SEMARNAT, 2004). The regional cement  
12 production relies on raw limestone material obtained nearby by dusty quarries. Cement  
13 is processed using residual fuel oil, petroleum coke and other materials, like used tires  
14 and industrial wastes. Other MV industries include non-ferrous metal manufacturing,  
15 textile, chemical, processed food, and disposal, recycling or incineration of waste  
16 materials. Cabrera et al., (2003) estimated the MV emissions of total suspended  
17 particles (TSP) from industrial and mobile sources in 21,538 ton year<sup>-1</sup>. This is a rather  
18 large amount, representing ca. 80% of the same emissions at the much larger and higher  
19 populated Mexico City (27,308 ton/year, SMA 2006). The contribution from natural  
20 and agricultural sources to the MV air pollution is still unknown. Two major regional  
21 natural/crustal sources of particulate material are the Cretaceous limestone (El Doctor  
22 Formation) that covers about 60% of MV, mainly exposed at the E-SE sector (Silva-  
23 Mora, 1997), and Lower Tertiary igneous rocks, roughly located at the valley's west  
24 half.

25

## 26 **2.2 Sampling and sample preparation**

27 *Tillandsia recurvata* was sampled at 50 sites throughout MV in the late spring and early  
28 summer, 2006 (Figure 1). Most sites were at mountain foothills where mesquite trees,  
29 from which the biomonitor was obtained, are more common. In a few sites lacking this  
30 tree, sampling was done on alternative trees and cacti. The mean elevation of the  
31 sampling sites was 2200 m (min, 1879 m; max, 2435 m). Three composite samples per  
32 site were collected from tree branches at  $\geq 1$  m from the ground, each composed by six  
33 to eight "ball moss" individuals from different trees. Cross contamination between

1 samples was avoided by wearing new dust-free latex gloves per sample. Samples were  
2 taken to the laboratory in air-dried condition in brown paper bags, and stored at ambient  
3 temperature until processing (late September-October, 2006). They were manually  
4 cleaned prior chemical analyses by removing plant dead parts as well as insects, feathers  
5 and spider webs, among other materials extraneous to the samples, but were not  
6 subjected to further cleaning procedures, as recommended for air pollution  
7 biomonitoring with plants (Rossini Oliva and Raitio, 2003). The samples were then  
8 subdivided into three portions for metal, PAH and isotopic analyses, and stored in  
9 polyethylene bags at -40°C until analyses. Since *T. recurvata* is perennial, potential age-  
10 related variability in pollutant bioaccumulation was minimized by analyzing only the  
11 newest 3-4 pairs of leaves per shoot. Thus, the time period of environmental exposition  
12 was assumed to be similar for all sites, probably representing one or two years.

13

#### 14 **2.2.1 Metals**

15 The glass and Teflon material for sample digestion was cleaned by immersion in 10%  
16 HNO<sub>3</sub> (24 h) followed by profuse rinsing with deionized water. The plant samples were  
17 oven-dried to constant weight (70°C, three days) and ground to fine powder with agate  
18 mortar and pestle. A fraction of powder (0.15-0.2 g per sample) was digested with 6.0  
19 ml HNO<sub>3</sub> (65%, Merck), 4.0 ml HCl (38%, JT Baker, Ultrex II), and 0.2 ml HF (48%,  
20 JT Baker, Ultrex II) in a microwave oven (Anton Paar, Multiwave 3000) using the “pine  
21 needles” program: phase I (power, 1400 W; ramp, 10 min; hold, 10 min; fan 1), phase 2  
22 (power, 0 W; hold, 30 min; fan, 2). Digestion temperature: 180°C. The digests were  
23 filtered with Whatman 4 paper and brought to 25 ml with deionized water.

24 The elements were determined by inductively coupled plasma/optical emission  
25 spectrometry (ICP-OES, Perkin–Elmer, Optima 3200 DV) following the USEPA 6010C  
26 method (EPA, 2000). The sample injection flow was 1.0 ml min<sup>-1</sup>. Calibration curves  
27 were prepared with diluted element standard solutions (High Purity). QC repeatability  
28 was checked up by injecting element mixtures (Ultra Scientific) every 10-15 samples.  
29 The percent recovery of seven certified elements in seven equally processed samples of  
30 the NIST 1575a standard reference material (trace elements in pine needles) was (mean  
31 ± coefficient of variation): Al (102.2 ± 2.0), Ca (113.5 ± 10.1), Fe (117.3 ± 7.5), Mg  
32 (84.8 ± 13.9), P (96.5 ± 2.0), Zn (83.0 ± 5.1). The recovery of Ba was low (35.8 ±

1 6.4%), but was very good for In ( $103.7 \pm 2.9\%$ ), which was added as surrogate from a  
2  $5.0 \mu\text{g ml}^{-1}$  standard solution.

3

#### 4 **2.2.2 PAH**

5 The samples were organically extracted in microwave oven and analyzed by high  
6 performance liquid chromatography (HPLC) for the 15 PAH listed in Table 1. The  
7 glassware and Teflon materials for sample preparation were cleaned with liquid  
8 detergent (Liqui Nox), running water and consecutive rinsing with bidistilled water,  
9 acetone and dichloromethane (DCM). Samples (5 to 8 g, air-dry weight) were ground  
10 with ceramic mortar and pestle under liquid nitrogen. A fraction of powder (0.25-0.5 g)  
11 was oven-dried ( $90^{\circ}\text{C}$ , 24 h) for dry weight determination and about 2.0 g were  
12 extracted in 10-sample batches with 30 ml DCM in microwave oven (CEM, model  
13 MarsX). The oven was programmed as USEPA 3546 method: potency, 1200 W;  
14 pressure, 100 psi; temperature,  $115^{\circ}\text{C}$ ; time, 15 min; total extraction time, 30 min;  
15 cooling, 2 h. Each extraction batch included one pure DCM blank and one positive  
16 laboratory control sample (LCS). As a surrogate for the vegetal matrix, the LCS were  
17 added 0.5 g of dehydrated and deactivated Cromosorb (60/80 mesh, Supelco, Inc;  
18  $400^{\circ}\text{C}$  for 4h) and 200  $\mu\text{l}$  of standard mixture containing  $5 \mu\text{g ml}^{-1}$  of each target PAH  
19 (Chemservice). The extracts were filtered through Millipore membrane and  
20 concentrated to 1.0 ml under a gentle stream of ultrapure nitrogen supplied with a  
21 nitroevaporator (8158, N EVAP111, Organomation Associates, Inc). The concentrates  
22 were cleaned up with glass chromatographic columns (40 cm length x 1.5 cm ID) filled  
23 with deactivated ( $400^{\circ}\text{C}$  for 4 h) alumina-silica-gel (USEPA 3610B and 3630C). The  
24 columns were packed from bottom to top with 1.0 cm silanized glass fiber (Alltech)  
25 humidified with DCM, 10 g alumina (Baker, Inc.), 3.0 g silica gel (60/200 mesh,  
26 Mallinckroft) and 2.0 g anhydrous sodium sulfate (JT Baker) dissolved in DCM. For  
27 final PAH separation, extra DCM (30 ml) was added to the columns. Eluates were  
28 captured in glass vials, concentrated to 1.0 ml under ultrapure nitrogen, changed into  
29 acetonitrile (4.0 ml, Burdick & Jackson), filtered with  $0.2 \mu\text{m}$  acrodiscs (Pall Gelman  
30 Laboratory), reconcentrated to 500  $\mu\text{l}$  with ultrapure nitrogen and stored at  $-40^{\circ}\text{C}$  in  
31 amber glass vials until analyses.

32 The extracts were analyzed with a liquid chromatograph (Agilent HP, 1100 series)  
33 equipped with Nucleosil column (Macherey-Nagel, 265 mm, 100-5 C18 PAH), an

1 automatic sample injector, and DAD and fluorescence detectors. The analyses were  
2 done in 10-sample batches plus one blank and one positive control. Each batch had its  
3 own calibration curve with seven concentrations (0.0625 to 5.0  $\mu\text{g ml}^{-1}$ ) of stock  
4 acetonitrile solution containing 5  $\mu\text{g ml}^{-1}$  of each target PAH. Samples were added 150  
5  $\mu\text{l}$  of 4,4'-difluorobifenil (Chemservice) solution in acetonitrile (4  $\mu\text{g ml}^{-1}$ ) as internal  
6 standard. Solvents A (methanol, 50%; acetonitrile, 25%; HPLC water, 25%) was  
7 injected from 0 to 3 min gradually exchanged by solvent B (acetonitrile) from 3 to 21  
8 min; only solvent B was injected from 21 to 35 min. Injection volume, 5  $\mu\text{l}$ ; flux, 0.3  
9 ml/min; column temperature, 25°C; wavelength (FLD: 275-495 nm, Table 2; DAD: 230  
10 nm); total analysis time per sample extract, 30 min. Method validation parameters:  
11 linearity ( $R^2 < .98$ ); accuracy and precision (RSD < 3%); detection limit, 1.0  $\mu\text{g ml}^{-1}$ ;  
12 quantification limits, 0.01 to 0.03  $\mu\text{g ml}^{-1}$ , depending on the compound (Table 1).

13

### 14 **2.2.3 $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$**

15 Air-dried samples (200-500  $\mu\text{g}$ ) were ground with cleaned ceramic mortar and pestle  
16 under liquid nitrogen, and dried at 80°C for 24 h. The  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  ratios were  
17 determined with a dual carbon and nitrogen analyzer coupled to a continuous flow  
18 isotope ratio mass spectrometer (Europa Scientific). Equipment precision was 0.1 ‰ for  
19 C and 0.2 ‰ for N. The isotopic composition is defined by:

20

$$21 \quad \delta^{13}\text{C} \text{ or } \delta^{15}\text{N} (\text{‰}) = (R_{\text{sample}} / R_{\text{standard}} - 1) \times 1000$$

22

23 where R is the sample or standard ratio of the heavy to the light isotope:  $^{13}\text{C}/^{12}\text{C}$  and  
24  $^{15}\text{N}/^{14}\text{N}$ . The V-PDB (Viena Pee Dee Belemnite) and atmospheric nitrogen were used as  
25 standards to determine  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$ , respectively (Mariotti, 1974; Coplen, 1995).

26

## 27 **2.3 Statistics**

28 The raw site-pollutant data base was regionally summarized by simple central tendency  
29 and dispersion statistics. The geographical distribution of pollutant was explored by  
30 mapping with a geostatistical gridding method for irregularly spaced data (Surfer, Ver.  
31 7.05, Kriging method) and comparing concentrations between and among MV areas  
32 with *t*-tests or ANOVA, as appropriate. Potential emission sources were explored by  
33 factor analyses (FA) with principal component extraction and normalized varimax



1 rotation (Statistica, Ver. 6.0). Since 50 sampling sites was a small number of cases, FA  
2 was used conservatively. Only 20 pollutants were included in FA to keep a 2.5 site to  
3 variable ratio. The standardized median site values were used for FA. Pollutants  
4 excluded from this analysis had one or more of the following problems: they were not  
5 detected or below quantification levels at  $\geq 8$  sampling sites (i.e., 16% of sites: Pb, Sb,  
6 BaA, BbF, BkF, BaP and IcdP); high skewness ( $>2$ ) and/or kurtosis ( $>7$ ); strong  
7 deviation from normality even after log transformation (Shapiro-Wilk test); no  
8 significant correlation with any other pollutant, because they tended to form factors with  
9 single pollutants (e.g., Cd and Cr), or contrarily, very high correlation ( $r > .9$ ), which  
10 may lead to multicollinearity. All low and some medium molecular weight PAH, which  
11 predominantly disperse in the gas phase, were also excluded from FA (NAP, ACY,  
12 ACE, FLN, PHE, ANT and FLT). Finally, the pollutant selection for FA was aided by  
13 considering element signal to noise ratios (SNR), which relate the regional to the site  
14 variability, as defined by Wolterbeek et al. (1996), and the element enrichment factors  
15 (EF) calculated as

$$16 \quad \quad \quad EF = (X_s/Al_s)/(X_r/Al_r)$$

17 where  $X_s$  and  $Al_s$  are the sample element and aluminum concentrations (ppm),  
18 respectively;  $X_r$  and  $Al_r$  are the element and aluminum concentrations (ppm) in the main  
19 crustal rocks of MV: limestone, according to Lozano and Bernal (2005), and acidic  
20 igneous rocks, using averaged data from an internet data base by Surendra (2001).

21

## 22 **3 Results and discussion**

23 Mezquital Valley was biomonitoring with *T. recurvata* to detect the regional dispersion  
24 trends and potential emission sources of airborne metals and PAH.

25

### 26 **3.1 Metals**

27 Table 2 summarizes the regional element concentrations in the biomonitor. Calcium  
28 was the most abundant element (regional mean, 1.3% on dry weight basis; min, 0.5%;  
29 max, 4.5%). This indicated high regional exposure to limestone dust at MV, which is  
30 exacerbated by the cement industry. Predominance of Ca in lichen biomonitors growing  
31 near a cement plant was recently reported by Branquinho et al. (2007). They considered

1 this element as the best indicator for cement-dust. Other major geological elements  
2 followed Ca in abundance (%): K (0.76) > Al (0.42) > Na (0.34) > Mg (0.29) > Fe  
3 (0.17) > P (0.05) > Ti (0.02) > Mn (0.008).

4 The trace element constituted individually less than 0.005% of the *T. recurvata* dry  
5 biomass: V (0.0044) > Zn (0.0042) > Ba (0.0038) > Pb (0.0034) > Sr (0.0031) > Li  
6 (0.0018) > Ni (0.0016) > Cu (0.0007) > Cr (0.0006) > Sb (0.0005) > Mo (0.0004) > Cd  
7 (0.0002). The prevalence of V, an element typically abundant in crude oils (Bairwise  
8 1990), can be attributable to emissions from the major industrial users of petroleum  
9 fuels at MV. This will be detailed later.

10 At the gross level of data reduction in Table 2, the element concentrations in *T.*  
11 *recurvata* reflected the composition of the MV acidic crustal rocks and agriculture soils  
12 (Figure 2). According to this figure, there was a closer chemical similarity between the  
13 biomonitor and the igneous rocks ( $R^2 = 0.86$ ) than for soils ( $R^2 = 0.78$ ) and limestone  
14 ( $R^2 = 0.64$ ). Thus, a high proportion of the spatial element variability in the biomonitor  
15 could be expected to derive from these sources.

16 Most measured elements had higher mean concentrations than reported for other  
17 *Tillandsia* biomonitors in other US and Latin American areas. This is shown in Table 3  
18 by the element concentration ratios between MV and other countries (MV:OT). This  
19 ratio was  $\geq 2$  in 25 out of 44 cases in Table 3, which implies  $\geq 100\%$  higher  
20 bioaccumulation by the Mexican *Tillandsia*. The difference was larger for  
21 anthropogenic elements such as Ni, Pb and V, whose respective concentrations at MV  
22 were 79, 66 and up to 43 times higher than measured in *Tillandsia* biomonitors from  
23 Argentina (Wannaz et al., 2006a, b). Assuming similar interspecies ability to capture  
24 and retain airborne metals, such differences indicate that regional atmospheric  
25 deposition levels at MV are at least as high as indicated by those ratios. The MV:OT  
26 ratio was  $< 1.0$  only in 11 cases; i.e., lower element concentration in at MV (Table 3).  
27 Most of them involved Cu and Zn biomonitoring within urban/industrial areas at São  
28 Paulo, Brazil (Figuereido et al., 2007), Cali and Medellín, Colombia (Schrimpff, 1984).  
29 This is attributable to the fewer rural sites included in these studies, which are usually  
30 less exposed to anthropogenic emissions and tend to lower the regional mean estimates.

### 31 **3.2 PAH**

1 Table 4 summarizes the regional concentration and variability of PAH in the  
2 biomonitor. The sum of quantified PAH per site ranged from 143 to 2,568 ng g<sup>-1</sup>, with  
3 regional mean and median equal to 572 and 439 ng g<sup>-1</sup>, respectively, indicating a data  
4 distribution skewed to the right. Apart from PHE and CHR, which were lognormally  
5 distributed, the rest of PAH had neither normal nor lognormal distributions. The  
6 regional variability of PAH depended on the compound, ranging from 58% (ACY) to  
7 165% (FLN). The medium molecular weight PAH (MMW: PHE, ANT, FLT, PYR,  
8 BaA and CHR) was the most abundant group, representing 64.5% of the total measured  
9 PAH. The low molecular weight PAH (LMW: NAP, ACY, ACE and FLN) constituted  
10 28.1%, and the high molecular weight (HMW: BbF, BkF, BaP, BghiP and IcdP) only  
11 7.3%. The relatively low amount of the most hazardous PAH group (HMW) deserves  
12 further research to be explained. Hypothetically, it may be attributed to  
13 photodegradation enhanced by the high solar irradiation rates received in this tropical  
14 semiarid region and the scarce protection from it provided by the very open local  
15 vegetation.

16 The most frequent PAH were FLT, FLN, PHE and PYR, which were present at all  
17 sampled sites, as well as CHR, NAP and BghiP, which were missing only at one or two  
18 sites. Most HMW PAH had low to very low frequencies. For instance, BaP had  
19 quantifiable levels at only nine sites, and BaA, BkF and IcdP at the two sites closest to  
20 the petroleum refinery and the electrical power plant (Fig. 1). The most abundant  
21 compounds were FLT (24.2% of total quantified PAH, regional average), PHE (17.1%)  
22 and NAP (10.4%).

23 Table 5 compares the mean total PAH in *T. recurvata* at MV with other biomonitoring  
24 studies using *Tillandsia*. Such comparisons are, however, limited by the several factors  
25 affecting the total sum of PAH; e.g., the number and type of measured compounds, the  
26 biomonitor's ability to capture and retain them, and design aspects of the biomonitoring  
27 studies; e.g., number and type of sampling sites, length of exposure and sampling  
28 strategy (*in situ* vs. transplanted samples). No difference was clear between the total  
29 PAH at MV and one heavily polluted site at Mexico City measured by Wang et al.,  
30 (2007) with pine needles, a morphologically and physiologically different receptor. The  
31 mean total PAH reported for *T. usneoides* from polluted sites in Río de Janeiro (De  
32 Sousa et al., 2007) doubled the mean value at MV. Nevertheless, the most polluted MV  
33 sites had similar or slightly different total PAH (1,284-1,748 ng g<sup>-1</sup>) than the polluted

1 sites at Rio de Janeiro, and no MV site had lower total PAH than unpolluted Brazilian  
2 sites (93 to 175 ng g<sup>-1</sup>). Levels of total PAH at MV were 70% to 140% higher than  
3 recorded for two different *Tillandsia* species transplanted from Costa Rica into  
4 downtown Florence, Italy, for eight months (Brighina et al., 2002).

### 6 **3.3 $\delta^{13}\text{C}$ and $\delta^{15}\text{N}$**

7 The spatial patterns of the  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  in *T. recurvata* reflected quite well the main  
8 land uses at MV. Interestingly, these ratios showed that this epiphyte is naturally less  
9 discriminatory against <sup>13</sup>C, but more discriminatory against <sup>15</sup>N than most terrestrial  
10 plants.

11 The regional  $\delta^{13}\text{C}$  ratio (mean, -14.6‰; min, -15.7‰; max, -13.7‰) was similar to  
12 known values for this species (-15.3‰ to -13.2‰, Martin, 1994) and within range for  
13 *Tillandsia* with crassulacean acid metabolism (CAM;  $\delta^{13}\text{C}$  >-20‰, Crayn et al., 2004);  
14 i.e., plants which fix CO<sub>2</sub> predominantly at night as a water-saving adaption to arid  
15 environs (Pierce et al., 2002). This ratio was the most homogeneous of the measured  
16 variables. It varied only 2‰ throughout MV (CV, 9.0%). Its less negative values  
17 occurred at the predominantly rural NW sector of the valley ( $\delta^{13}\text{C}$  = -14.3‰ ± 0.13,  
18 mean ± standard error), being on average 0.6‰ higher than at the industrialized SE  
19 sector (-14.9‰ ± 0.12; *p* <.006 from one-way ANOVA by sectors). This suggests that  
20 *T. recurvata* has a “background”  $\delta^{13}\text{C}$  signature similar to values recorded at the NW  
21 sector of MV (-14.3‰, on average).

22 The  $\delta^{13}\text{C}$  values of *T. recurvata* decreased from the rural periphery of MV to the core  
23 industrial area, from values ≥-14.5‰ to ≤-15‰ (Figure 3a). This negative shift reflected  
24 the increased emissions of <sup>13</sup>C-depleted compounds from industrial fossil fuel  
25 combustion. The crude oils processed there have an average  $\delta^{13}\text{C}$  = -27‰, as inferred  
26 from available data for Mexican petroleum (e.g., Macko and Parker, 1983; Prinzhofer et  
27 al., 2000); i.e., 12.4‰ more negative than the mean value for *T. recurvata*. The effect  
28 from petroleum sources was further recognized by the regional correlation of  $\delta^{13}\text{C}$  with  
29 the Ni/V ratio (*r* = 0.39, *p* = .01), which was even closer (*r* = 0.70, *p* = .001) when  
30 considering only the sampling sites along a belt transect in the predominant wind  
31 direction (NE-SW). Since Ni and V are the predominant trace metals in crude oils and  
32 keep relatively constant mutual proportions, their ratios (Ni/V or V/Ni) are used for

1 geochemical classification of crude oils (Bairwise, 1990) and as environmental  
2 indicators for petroleum sources (Ganor et al., 1988). Figure 4 illustrates the parallel  
3 pattern followed by  $\delta^{13}\text{C}$  and Ni/V along a NE-SW transect, including a sharp decline at  
4 the closest downwind sites from the main petroleum sources (km 0). The regional mean  
5 Ni/V for *T. recurvata* was  $0.49 \pm 0.29$  ( $n = 50$ ). The lower ( $< 0.20$ ) and higher values  
6 ( $> 0.8$ ) of this ratio occurred, respectively, near and far from the petroleum facilities. The  
7 Mexican crude oils used at MV have Ni/V values ranging from 0.07 to 0.24, and 0.17  
8 on average, calculated from data by Castro and Vazquez (2009) and Pena et al., (1993).  
9 The biomonitor tended to acquire Ni/V values similar to crude oil at sites exposed to  
10 petroleum combustion emissions (Fig. 4).

11 The spatial pattern of  $\delta^{15}\text{N}$  reflected clearly the agriculture, industrial and rural land  
12 uses at MV. The regional mean was  $-3.0\text{‰}$  (min,  $-9.9\text{‰}$ ; max,  $3.3\text{‰}$ ), which  
13 represented a  $6\text{‰}$  to  $9.6\text{‰}$   $^{15}\text{N}$  enrichment respect to other Mexican atmospheric  
14 *Tillandsia* from less polluted habitats ( $\delta^{15}\text{N} = -9.0\text{‰}$  to  $-12.6\text{‰}$ , Hietz and Wanek,  
15 2003; Hietz et al., 1999). The lowest, probably “background” signature for *T. recurvata*,  
16 occurred at the rural SW corner of MV ( $\delta^{15}\text{N} \approx -7.0\text{‰}$ ). Since the  $\delta^{15}\text{N}$  of land plants  
17 range roughly from  $-7\text{‰}$  to  $+9\text{‰}$  (Kelly et al., 2005), this *Tillandsia* belongs to the most  
18 discriminative plants against  $^{15}\text{N}$ . This suggests that it could rely on biologically usable  
19 atmospheric compounds with low  $\delta^{15}\text{N}$ , such as gaseous  $\text{NH}_3$  and its derivative aerosol  
20  $\text{NH}_4$  (Krupa, 2003). Another contributing factor may be a symbiotic association with  
21 bacteria able to convert atmospheric  $^{14}\text{N}_2$  into usable salts for *T. recurvata*. Puente and  
22 Bashan (1994) documented the occurrence of such bacteria in Mexican *T. recurvata*, but  
23 their function needs to be proven.

24 The *T. recurvata*  $\delta^{15}\text{N}$  pattern at MV changed from positive ( $> 3.0\text{‰}$ ) to negative ( $< -$   
25  $6.0\text{‰}$ ) along the predominant wind direction (NE-SW, Fig. 3b). The most positive to  
26 slightly negative values occurred at the NNE sector of MV, near agriculture and  
27 industrial/urban areas, whereas the most negative were recorded at the farther WSW  
28 rural sites. Similarly high to low plant  $\delta^{15}\text{N}$  spatial contrasts have been documented for  
29 other industrial/urban or agriculture areas respect to rural/remote locations (Gerdol et  
30 al., 2002; Jung et al., 1997; Pearson et al., 2000; Solga et al., 2005). Such contrasts are  
31 attributed in general to the higher N emission/deposition by human activities in the  
32 former areas, an explanation that finds support from the positive correlations often  
33 observed between the plant  $\delta^{15}\text{N}$  and the total N deposition (e.g., Pardo et al., 2007;

1 Skinner et al., 2006). Two major, probably supplementary, hypotheses have been  
2 proposed about what causes the higher plant  $\delta^{15}\text{N}$  at the urban/industrial areas: a)  
3 increased exposure to some oxidized N species ( $\text{NO}_x$ ) from fossil fuel combustion, and  
4 b) fractionating losses of the lighter  $^{14}\text{N}$  compounds from the plants by leaching or  
5 nitrification processes with subsequent concentration of  $^{15}\text{N}$  in the remaining N pools  
6 (Jung, 1997). Solga et al. (2005) reported a  $\delta^{15}\text{N}$  spatial difference between mosses near  
7 agriculture areas (lower values) and rural environs (higher values) negatively correlated  
8 with the  $\text{NH}_4:\text{NO}_3$  ratio of the bulk N deposition. This implies that increasing the  
9 proportion of  $\text{NO}_3$  and compounds with high  $\delta^{15}\text{N}$ , like  $\text{N}_2\text{O}$  (+6.7 to +7.0‰,  
10 Brenninkmeijer et al., 2003), in the bulk N deposition would shift positively this plant  
11 ratio. The same effect would result by lowering the proportion of  $\text{NH}_4$  and other  $^{15}\text{N}$ -  
12 depleted compounds.

13 There may be a variety of primary and secondary N atmospheric compounds  
14 contributing to the  $\delta^{15}\text{N}$  spatial pattern shown by *T. recurvata*, but specifying which of  
15 them accounted for the observed distribution of this ratio was beyond the reach of this  
16 survey. However, it can be hypothesized that the higher  $\delta^{15}\text{N}$  values near the agriculture  
17 areas may be related to the heavy use of untreated wastewater for irrigation. There are  
18 no  $\delta^{15}\text{N}$  data for this water, but it is reasonable to assume high positive values, as  
19 reported elsewhere for wastewater ( $\delta^{15}\text{N} = +10\text{‰}$  to  $+22\text{‰}$ ). Soils and plants exposed to  
20 such  $^{15}\text{N}$ -enriched water tend to acquire its high  $\delta^{15}\text{N}$  values (Cole et al., 2004; Wigand  
21 et al., 2007). This is most likely the case at MV, where agriculture soils have a net  
22 positive N balance despite the substantial loss that may occur by volatilization from the  
23 open wastewater channels and during irrigation (Siebe, 1998). However, it can be  
24 hypothesized that gaseous and particulate emissions of compounds with high  $\delta^{15}\text{N}$  (e.g.,  
25  $\text{N}_2\text{O}$  and  $\text{NO}_3$ , respectively) from those channels and soils are large enough to convert  
26 the naturally negative  $\delta^{15}\text{N}$  of *T. recurvata* into positive values. Recent, spatially limited  
27 measurements (5 ha) on the emissions of  $\text{N}_2\text{O}$  from local agriculture soils showed rates  
28 varying from 0.01-0.04  $\text{mg m}^{-2} \text{h}^{-1}$  prior irrigation with wastewater to 3.4  $\text{mg m}^{-2} \text{h}^{-1}$   
29 after irrigation, which are similar to fluxes from artificially fertilized soils (González-  
30 Méndez, et al. 2008). Other regional activities, like burning agriculture debris, may also  
31 contribute with high  $^{15}\text{N}$  emissions.

1 The  $\delta^{15}\text{N}$  values at the core industrial area were similar to those at agriculture sites (0%  
2 to +3%). This could be due both to *in situ*  $\text{NO}_x$  emissions from fossil fuel combustion  
3 sources (signaled with C, E and R in Fig. 3b) and to downwind transport of  $^{15}\text{N}$ -  
4 enriched compounds from the main agriculture areas located immediately to the north of  
5 the industrial park.

### 6 7 **3.4 Distribution of pollutants and identification of major emission** 8 **sources**

9 Some insights on the element spatial dispersion at MV can be derived from their  
10 regional variability in the biomonitor (Table 2). The average coefficient of variation was  
11 58% and ranged from 25% (Li) to 124% (Sb). The elements with lower variability, such  
12 as the same Li and K (33%) were in general more homogeneously distributed  
13 throughout the region than those more variable (e.g., V, 87%, Cr, 96%). The former  
14 elements are more likely to be emitted by widely distributed sources; e.g., soil or crustal  
15 rocks, whereas the latter ones may derive from localized emission sources.

16 In looking for general spatial patterns at MV, the predominantly agricultural north was  
17 compared to the industrial south using the median site concentrations and straight *t*  
18 tests. Eleven elements (Al, Ba, Ca, Fe, Mg, Mo, Ni, Sr, Ti, V and Zn) plus CHR and  
19 PYR had 18 to 214% higher concentrations at the industrial area ( $p < .05$ , Table 6).  
20 None was significantly more concentrated at the agricultural north. Such spatial  
21 difference is partially illustrated in Fig. 5. This figure plots the site normalized median  
22 concentration of Ca, CHR, PYR and V within a belt transect oriented along the  
23 predominant wind direction (NNE to SSW), including the main types of particle  
24 emitters in the region. At the industrial area ( $\geq 0$  km in Fig. 5), those pollutants had site  
25 concentrations up to 5.2 (Ca), 4.2 (CHR), 4.8 (PYR) and 29 (V) times higher than at the  
26 farthest north, rural location (Mixquiahuala), located 37 km from the industrial area  
27 (Fig. 5). At sites near or within the main north agriculture area (-2 to -30 km in Fig. 5),  
28 most elements and PAH had intermediate concentrations.

29 Figure 6 illustrates four selected distribution patterns of pollutants as mapped with the  
30 median site data. Vanadium had a well-defined concentration peak ( $>175 \text{ mg kg}^{-1}$ )  
31 located 3.7 km to the south of the petroleum refinery and the power plant (Fig. 6a). Its  
32 concentration dropped centrifugally from there, but at faster rate toward the north than

1 to the south, indicating air transport in this latter direction. The V concentration  
2 increased again southward (ca., 12 km from the main peak) probably due to further  
3 fossil fuel emissions from cement industries, which are also heavy users of residual  
4 petroleum and other materials (petcoke, natural gas, vehicle tires and a variety of  
5 residual materials).

6 Chromium had also a single concentration peak ( $>20 \text{ mg kg}^{-1}$ , Fig. 6b). In contrast with  
7 V, this was located at the north of the valley; i.e., upwind from the industrial park. At  
8 the rest of the sampling areas, the biomonitor accumulated rather similar Cr levels ( $\approx 5$   
9  $\text{mg kg}^{-1}$ ). Two potential sources causing that Cr peak are nearby polluted agriculture  
10 soils and limestone dust from quarrying.

11 Calcium (Fig. 6c) presented two high concentration spots, one at the SE ( $>36,000 \text{ mg}$   
12  $\text{kg}^{-1}$ ) and another at the NE ( $>25,000 \text{ mg kg}^{-1}$ ) of the study region, coinciding with two  
13 major limestone areas where mining for cement materials occurs. In general, this  
14 element was more concentrated in the biomonitor at the East half of MV where limestone  
15 rocks predominate.

16 Figure 6d shows the distribution of total PAH. The highest concentrations were  
17 recorded at the south of MV, mainly to the WSW of the core industrial area. The lowest  
18 levels occurred at rural NW sector, upwind respect to the urban/industrial areas. The  
19 tree concentration peaks in Fig. 6d were determined by the most abundant MMW (FLT  
20 and PHE) and LMW (NAP) PAH, probably reflecting vehicular emissions from the  
21 closest urban settlements.

22 Prior to factor analysis, the element signal-to-noise ratio (SNR) and enrichment factors  
23 (EF) were considered as criteria for variable selection (see Statistic section). The  
24 average SNR was 4.1 (min, 2.1, Cd; max, 7.3, Ca, Table 2), a good ratio in signal  
25 information for a biomonitoring survey. Comparatively, Wolterbeek et al., (1996)  
26 reported average SNR values from 1.3 to 3.4 for lichens, mosses and tree bark  
27 biomonitors in Europe.

28 Figure 7 shows the mean element EFs relative to the Al concentrations in the regional  
29 limestone and igneous rocks. It includes only elements quantified in both rock types.  
30 Enrichment values close to 1.0 are indicative of crustal origin. Higher values,  
31 subjectively  $\geq 5$ , indicate enrichment by anthropogenic emissions. Since K, Ni, P, V and  
32 Zn were more enriched than expected from both rock types, they are likely to derive



1 from industrial (Ni, V and Zn) and, tentatively, agricultural sources (K and P). In  
2 contrast, Na was enriched only respect to limestone; whereas Ca and Cu were enriched  
3 respect to igneous rocks. The rest of elements in Fig. 5, Ba, Cr, Mg, Mn, Sr and Ti,  
4 probably originated from crustal/soil sources. The natural emission of some elements  
5 can be enhanced by activities like mining. This could be the case for Ca and Cr, whose  
6 highest site concentrations occurred near limestone mining areas.

7 Table 7 shows a factor analysis solution for the 50 MV sites x 20 pollutants  
8 standardized data. Five factors (F1 to F5) or potential sources with eigenvalues >1.0  
9 accounted for 74% of the data variance. Most of it (39.5%) was associated to F1, which  
10 was loaded by Ti >Ba >Fe >Al >Sr >Cu (loadings >0.6), and to some extent by Ca  
11 (0.49). Such composition indicates crustal sources, including both igneous rocks and  
12 limestone (Ca). According to the site scores from FA (not shown), this factor was  
13 geographically centered at the MV agriculture/industrial frontier, 1.5 km from the  
14 refinery in NE direction, where the first four elements (Ti to Al) peaked in  
15 concentration. This suggests soil dust traveling N to S from the agriculture areas as the  
16 element vehicle. Since most F1 elements are considerably more concentrated in the  
17 local igneous rocks than in limestone (igneous to limestone concentration ratios: Al,  
18 106:1; Ba, 48:1; Cu, 3:1; Fe, 1,201:1; Ti 340:1, with data by Lozano and Bernal, 2005;  
19 Verma-Surendra, 2001), this factor appears more related to soil sources composed  
20 predominantly by igneous parental material.

21 Factor 2 (11.3% of variance) was loaded (>0.60) by Mn >Mg >Zn, and slightly by Cu  
22 (0.44). Interestingly, Na had a negative loading (-0.71) on this factor. This reflected a  
23 major spatial contrast between Na, with higher concentrations (4,000 to 5,000 mg kg<sup>-1</sup>)  
24 within or near agriculture areas (Fig. 8); whereas the other elements were more  
25 concentrated at the farther rural areas to the south (Mn >100 mg kg<sup>-1</sup>) and SE of MV  
26 (Mg >4,000 mg kg<sup>-1</sup>). Although Zn had high levels at some industrial sites (>60 mg kg<sup>-1</sup>),  
27 it kept similar concentrations at the SE rural area, resembling the Mg pattern. This  
28 factor may be also related to soil sources, but distinguishing saline agriculture soils at  
29 the MV lowlands (Na) from natural soils (Mg, Mn and Zn). Sodium is the main cation  
30 in irrigation wastewater at MV (Cervantes-Medel and Armienta, 2004; Jimenez and  
31 Chávez, 2004), which is attributed transport by wastewater from the saline areas near  
32 Mexico City Siebe (1998). However, potential inputs from other sources for the  
33 elements loading on this factor cannot be excluded. For instance, industrial particulate

1 matter traveling N to S (Cu and Zn) and gasoline vehicle emissions (Mn) from a major  
2 line source located at the S limit of MV, the Mexico City-Querétaro highway.

3 Factor 3 (F3) accounted for 9.8% of the variance and was significantly loaded (>0.60)  
4 by PYR >V >CRY >Ni >Ca and Zn (0.57). It is clearly an anthropogenic factor with  
5 pollutants from fossil fuel combustion and cement production (Ca). The presence of Ca  
6 on this factor is explained by the spatial concurrence of the main petroleum combustion  
7 sources and the major limestone quarries and cement plants, which are also heavy users  
8 of fossil fuels, at the SSE sector of MV. The negative loading of  $\delta^{13}\text{C}$  on this factor  
9 added further evidence to associate it with petroleum combustion. Figure 9 illustrates  
10 how these pollutants separate in the factor space from the predominantly crustal/soil  
11 elements (Al, Ba, Cu, Fe, Sr and Ti).

12 Factor 4 (F4) had 8.1% of the variance associated with it. It was loaded by K and  
13 benzo(*g,h,i*)perylene, deriving tentatively from biomass burning, which may include  
14 agriculture debris in the flatlands and forest fires in the highlands at the W side of MV.

15 Factor 5 (F5) accounted only by 5.2% of the variance. It was loaded by the  $\delta^{15}\text{N}$  isotope  
16 ratio and P. This factor may reflect agriculture activities with wastewater, characterized  
17 by large emissions of N compounds, mainly  $\text{NH}_3$ ,  $\text{NH}_4$  and  $\text{NO}_3$  (Cervantes-Medel and  
18 Armienta, 2004; Jimenez and Chávez, 2004) and P (Ortega-Larrocea et al., 2001).  
19 Phosphorus may also derive from pesticides used in agriculture, which often contain  
20 this element. The negative loading of  $\delta^{13}\text{C}$  on this factor added evidence on the regional  
21 difference between agriculture and industrial (fossil fuel combustion) land uses.

22 Although FA did not extracted pure factors/sources, it acceptably distinguished  
23 contributions from the most relevant sources of particulate matter into the MV air. Most  
24 element variability in the biomonitor, 64% out of 74% total explained variance,  
25 appeared associated with crustal/soil sources and agriculture activities (F1 + F2 + F4 +  
26 F5), and only 10% (F3) was attributable to emissions from industrial fossil fuel  
27 combustion, including power generation, petroleum refining and cement production.  
28 Increasing the number of sampling sites, especially around some major fixed sources,  
29 such as the power plant and cement factories, may allows us to differentiate their  
30 particular contributions to the load of bioaccumulative atmospheric pollutants in the  
31 study region.

## 32 **4 Conclusions**

1 This survey produced previously unknown information on the regional levels, sources  
2 and dispersion of airborne metals and PAH for Mezquital Valley, Mexico. *Tillandsia*  
3 *recurvata* was confirmed as a good natural receptor for those pollutants. The site to site  
4 concentration variability, as indicated by high signal-to-noise ratios for most metals  
5 (>3.0), reflected the differential long-term exposition to most measured pollutants,  
6 especially to the anthropogenic ones. Some anthropogenic metals (e.g., Ni, Pb and V)  
7 had particularly high concentrations respect to values for similar biomonitors in other  
8 countries, indicating that atmospheric deposition should be several times higher at this  
9 Mexican region. The levels of PAH were more similar to values for *Tillandsia*  
10 biomonitors, with a few exceptions (e.g., ACY and IcdP). The southern industrial  
11 portion of MV was more polluted than the agricultural north, as implied by the higher  
12 concentrations of 11 elements and two PAH. From the chemical correlation between the  
13 biomonitor and the regional rocks, results from factor analysis and the geographic  
14 patterns of pollutants, it is concluded that crustal/soil sources had a higher relative  
15 impact than agriculture and industry on the spatial variability of pollutants. The high  
16 “background”  $\delta^{13}\text{C}$  and low  $\delta^{15}\text{N}$  of *T. recurvata* make this epiphyte a sensitive  
17 biomonitor for  $^{13}\text{C}$ -depleted and  $^{15}\text{N}$ -enriched sources. It may be thus recommendable to  
18 use it as a common natural receptor for comparative studies within and among Latin  
19 American countries and the southern US states, where it is naturally distributed. This  
20 would minimize potential variability from using different species.

21

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28

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1 Table 1. Target PAH and HPLC analytical parameters.

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PAH	FLD, $\lambda_{ex}$ - $\lambda_{em}$ <sup>1</sup> (nm)	RT <sup>2</sup> (min)	QL <sup>3</sup> ( $\mu\text{g ml}^{-1}$ )
Naphthalene, NAP	275-350	4.29	0.03
Acenaphthylene, ACY	275-350	4.95	0.03
Acenaphthene, ACE	275-350	6.75	0.01
Fluorene, FLN	275-350	7.30	0.01
Phenanthrene, PHE	275-350	8.59	0.01
Anthracene, ANT	275-450	10.45	0.01
Fluoranthene, FLT	275-450	11.94	0.01
Pyrene, PYR	275-410	13.14	0.03
Benzo(a)anthracene, BaA	275-410	17.29	0.03
Chrysene, CHR	275-410	18.47	0.01
Benzo(b)fluoranthene, BbF	275-410	21.32	0.01
Benzo(k)fluoranthene, BbF	275-410	23.02	0.03
Benzo(a)pyrene, BaP	275-410	24.28	0.03
Benzo(g,h,i)perylene, BgP	275-410	27.56	0.01
Indeno(1,2,3-cd)pyrene, IcdP	275-495	29.01	0.03

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<sup>1</sup>Fluorescence detector excitation (ex) and emission (em) wave length; <sup>2</sup>retention time; <sup>3</sup>Quantification limit.

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Table 2. Regional element concentrations in *T. recurvata* (mg kg<sup>-1</sup>).

Element	N	n	Mean	Median	Range	CV (%)	S/N	<i>p</i> ( <i>r</i> )	<i>p</i> ( <i>log</i> )
Al	150	50	4,155	3,969	290 - 11,900	41.1	4.5		>.05
Ba	150	50	38.1	33.2	8.9 - 131	51.0	4.1		
Ca	149	50	12,556	10,190	2,883 - 46,780	55.2	7.3		>.05
Cd	150	50	2.2	1.9	0.9 - 6.5	42.7	2.1		>.05
Cr	139	49	6.1	4.5	0.8 - 40.4	96.3	6.1		
Cu	150	50	7.0	5.9	2.0 - 48.4	71.6	3.5		>.05
Fe	149	50	1,579	1,444	401 - 4,368	41.1	3.5		
K	150	50	7,568	7,559	1,398 - 16,380	33.3	2.5	>.05	
Li	150	50	18.1	17.0	6.0 - 34.0	24.8	3.2		>.05
Mg	150	50	2,906	2,815	868 - 6,285	33.8	2.7		>.05
Mn	149	50	80.3	75.2	14.2 - 327.4	39.4	4.3		>.05
Mo	145	47	4.0	4.0	0.3 - 28.6	76.6	5.5		
Na	149	50	3,236	3,046	424 - 10,280	57.6	6.1	>.05	
Ni	123	50	15.9	14.0	1.6 - 44.2	60.4	4.5		>.05
P	150	50	501	486	163 - 1,642	45.1	3.8		>.05
Pb	86	31	33.1	28.6	1.4 - 226	92.0	3.6		
Sb	25	9	4.0	2.9	0.2 - 25.4	124.3	3.6		
Sr	148	50	30.5	28.4	1.9 - 83.2	44.8	2.8		>.05
Ti	150	50	196	177	58.0 - 532	42.8	2.8		
V	150	50	43.4	29.7	6.5 - 220	86.8	5.5		>.05
Zn	150	50	41.9	36.4	3.7 - 192	57.0	4.0		

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N, number of samples; n, number of sampling sites with quantifiable amount of elements; CV, coefficient of variation; S/N, mean signal to noise ratio; *p*(*r*) and *p*(*log*), probability associated to Shapiro-Wilks normality test for raw and log<sub>10</sub> transformed data, respectively; empty cells, non-normal distribution.

1 Table 3. Elements in *Tillandsia* biomonitors from other countries (OC,) respect to *T.*  
 2 *recurvata* at Mezquital Valley, Mexico (MV). Units: mg kg<sup>-1</sup>; values in parentheses,  
 3 MV:OC ratio.  
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Element	MV	Argentina <sup>1</sup>	Argentina <sup>2</sup>	Brazil <sup>3</sup>	Colombia <sup>4</sup>	Colombia <sup>5</sup>	US <sup>6</sup>
Al	4,155			1,112 (3.7)			
Ba	38.1			36.8 (1)			
Ca	12,556			3,312 (3.8)			4,160 (3.0)
Cd	2.2				0.4 (5.3)	0.82 (2.6)	
Cr	6.1			2.2 (2.8)			
Cu	7.0	0.9 (7.8)	0.9 (7.0)	19.3 (0.4)	52.7 (0.1)	30.5 (0.2)	7.1 (1.0)
Fe	1,579	223 (7.1)	257 (6.1)	1,144 (1.4)			575 (2.7)
K	7,568			5,890 (1.3)			2,550 (3.0)
Mg	2,906			2,025 (1.4)			414 (7.0)
Mn	80.3	22.6 (3.6)	22.3 (3.6)	174 (0.5)			52.8 (1.5)
Na	3,236			1,923 (1.7)			
Ni	15.9	0.2 (79.4)	0.2 (79.4)		11.9(1.3)	52.8 (0.3)	
Pb	33.1	0.5 (66.2)	0.5 (66.2)		40.6 (0.8)	26.3 (1.3)	
Sb	4.0			0.5 (8.0)			
V	43.4	1.0 (43.4)	1.1 (39.5)	2.7 (16.1)			
Zn	41.9	4.0 (10.5)	3.8 (11.0)	111 (0.4)	119 (0.4)	379 (0.1)	73.5 (0.6)

5 <sup>1</sup>Wannaz et al. (2006a), *T. capillaris*; <sup>2</sup>Wannaz et al. (2006b), *T. permutata*; <sup>3</sup>Figueredo et al. (2007), *T. usneoides*, average value  
 6 for the two-months transplanting period presenting the maximum concentration, excluding the control site, <sup>4,5</sup>Schrimppf (1984), Cali  
 7 and Medellín cities, respectively ; <sup>6</sup>Husk et al. (2004), *T. usneoides* in Florida, samples recollected in 1998. Empty cells: data not  
 8 reported.  
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Table 4. Regional PAH concentrations (ng g<sup>-1</sup>) in *T. recurvata*.

HAP	N	n	Mean	Median	Range	CV(%)
NAP	141	49	82.3	49.6	6.0 - 607	122.8
ACY	85	40	88.6	78.8	10.2 - 292	61.9
ACE	51	32	19.6	13.4	1.1 - 87.0	95.6
FLN	129	50	37.6	21.4	7.2 - 386	134.6
PHE	150	50	136	114	9.9 - 624	60.3
ANT	92	42	9.0	7.1	1.7 - 27.0	63.0
FLT	150	50	192	72.2	19.3 - 1,674	164.9
PYR	148	50	25.9	22.1	0.03 - 158	88.0
BaA	2	2	12.8	12.8	0.2 - 25.4	139.7
CHR	127	49	22.5	18.2	1.9 - 92.9	72.6
BbF	75	40	18.3	13.2	0.1 - 131	118.0
BkF	2	2	13.5	13.5	1.7 - 25.3	123.6
BaP	12	9	41.7	39.9	2.4 - 87.6	58.2
BghiP	112	48	28.1	22.5	4.0 - 134	73.7
IcdP	2	2	67.0	67.0	22.2 - 112	94.4
LMW	150	50	164	128	12.4 - 738	81.2
MMW	150	50	376	261	74.1 - 2,131	96.8
HMW	122	50	42.5	32.8	0.1 - 250	85.0
ΣHAPs	150	50	572	439	142.6 - 2,568	76.7

N, number of samples; n, number of sampling sites with quantifiable amount of PAH; CV, coefficient of variation.

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1 Table 5. PAH in *Tillandsia* biomonitors from other countries (OC) respect to *T.*  
 2 *recurvata* at Mezquital Valley, Mexico (MV). Units: ng g<sup>-1</sup>; values in parentheses,  
 3 MV:OC ratio.  
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HAPs	MV	Mexico City <sup>1</sup>	Sao Paulo, Brazil <sup>2</sup>		Florence, Italy <sup>3</sup>	
			Polluted sites	Unpolluted sites	<i>T. caput-medusae</i>	<i>T. bulbosa</i>
NAP	82.3		70.7 (1.2)	11.4 (7.2)		
ACY	88.6	8.3 (10.7)	10.2 (8.7)	1.4 (63.3)	23.4 (3.8)	55.1 (1.6)
ACE	19.6	1.7(11.5)	13.7 (1.4)	8.6 (2.3)	53.6 (0.4)	53.9 (0.4)
FLN	37.6		24.6 (1.5)	9.5 (4.0)	4.0 (9.4)	112.1 (0.3)
PHE	136	15.8 (8.6)	198.2 (0.7)	18.6 (7.3)		
ANT	9.0	2.39 (3.8)	23.3 (0.4)	1.5 (6.0)	6.9 (1.3)	14.0 (0.6)
FLT	192	66.9 (2.9)	211.5 (0.9)	20.1 (9.6)		
PYR	25.9	118 (0.2)	145.1 (0.2)	14.2 (1.8)		
BaA	12.8	11.6 (1.1)	41.9 (0.3)	3.9 (3.3)	24.5 (0.5)	10.8 (1.2)
CHR	22.5	49.8 (0.5)	119.4 (0.2)	18.3 (1.2)	34.9 (0.6)	18.5 (1.2)
Bb+kF	31.8	47.2 (0.7)	111.6 (0.3)	15.9 (2.0)		
BaP	41.7	9.2 (4.5)	72.7 (0.6)	4.0 (10.4)	8.26 (5.0)	35.0 (1.2)
BghiP	28.1	39.7 (0.7)	29.3 (1.0)	5.0 (5.6)	28.2 (1.0)	
IcdP	67.0	28.2 (2.4)	31.1 (2.2)	3.9 (17.2)	33.2 (2.0)	12.4 (5.4)
ΣPAH	572	563 (1.0)	1,067 (0.5)	138 (4.1)	238 (2.4)	330 (1.7)

<sup>1</sup>Hwang et al. (2007): measurements in *Pinus maximartinezii* needles at one site with vehicle/industrial emissions. <sup>2</sup>After de Souza Pereira et al. (2007): average summer and winter measurements in *T. usneoides*. <sup>3</sup>Brighigna et al. (2002): eight-month transplanting study at an urban site. Empty cells: data not reported.

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Table 6. Elements (mg kg<sup>-1</sup>) and PAH (ng g<sup>-1</sup>) in *T. recurvata* at Mezquital Valley. Mean ± SD, n = 25.

Pollutant	North	South	D <sub>S-N</sub> <sup>1</sup>
Al	3,724 ± 918.7	4,625 ± 2003.8	24.2
Ba	31.0 ± 6.62	43.5 ± 18.20	40.3
Ca	9,453 ± 4073.5	15,577 ± 7078.9	64.8
Cd	2.1 ± 0.83	2.1 ± 0.55	
Cr	5.9 ± 6.06	4.7 ± 2.15	
Cu	6.1 ± 1.95	6.7 ± 2.87	
Fe	1,283 ± 278.0	1,848 ± 701.9	44.0
K	7,078 ± 2058.2	7,786 ± 1477.9	
Li	17.0 ± 3.23	18.7 ± 3.83	
Mg	2,647 ± 692.7	3,111 ± 970.6	17.5
Mn	77.3 ± 17.99	79.4 ± 23.66	
Mo	3.2 ± 1.71	4.8 ± 3.25	50.0
Na	3,215 ± 1414.4	2,815 ± 1421.4	
Ni	11.2 ± 4.95	20.8 ± 10.25	85.7
P	478 ± 130.5	505 ± 234.6	
Sr	26.8 ± 7.23	34.8 ± 14.30	29.9
Ti	164 ± 34.5	230 ± 92.5	40.2
V	21.0 ± 7.01	65.9 ± 38.89	213.8
Zn	31.4 ± 6.90	52.0 ± 27.82	65.6
NAP	55.5 ± 39.22	107 ± 144.0	
FLN	26.9 ± 25.10	46.9 ± 73.27	
PHE	113 ± 54.6	144 ± 68.6	
FTN	136 ± 265.6	159 ± 171.0	
PYR	16.1 ± 8.63	28.8 ± 11.89	78.9
CRY	14.8 ± 7.17	27.3 ± 16.76	84.5
BGP	25.5 ± 12.00	28.0 ± 15.64	

<sup>1</sup>D<sub>S-N</sub>, significant South – North percent difference (Student t, *p* < 0.05).

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Table 7. Factor analysis of pollutants in *T. recurvata* (n = 50).

	F1	F2	F3	F4	F5
Ti	<b>0.89</b>	0.07	0.25	0.16	0.17
Ba	<b>0.87</b>	0.26	0.14	-0.07	0.17
Fe	<b>0.86</b>	0.08	0.32	0.14	0.19
Al	<b>0.78</b>	-0.30	0.30	0.08	0.22
Sr	<b>0.72</b>	0.26	0.25	-0.15	0.21
Cu	<b>0.61</b>	0.44	0.09	0.27	0.05
Mn	0.25	<b>0.70</b>	-0.14	0.18	-0.18
Mg	0.35	<b>0.68</b>	0.13	0.24	0.21
Zn	0.24	<b>0.66</b>	0.57	0.08	0.19
Na	0.22	<b>-0.71</b>	-0.08	0.37	0.06
PYR	0.17	0.05	<b>0.81</b>	0.12	0.03
V	0.33	0.19	<b>0.79</b>	-0.06	0.29
CRY	0.23	0.24	<b>0.78</b>	0.05	0.34
Ni	0.12	-0.28	<b>0.73</b>	-0.04	-0.26
Ca	0.49	0.04	<b>0.62</b>	0.05	0.33
B <sub>g</sub> P	-0.20	0.13	0.14	<b>0.72</b>	-0.15
K	0.31	-0.02	-0.02	<b>0.71</b>	0.04
δ <sup>15</sup> N	0.26	-0.19	-0.05	-0.2	<b>0.69</b>
P	0.42	0.27	0.17	0.08	<b>0.67</b>
δ <sup>13</sup> C	-0.08	-0.01	-0.19	-0.01	<b>-0.73</b>
% variance	39.5	11.3	9.8	8.1	5.2

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1 Figure Captions

2 Figure 1. Mezquital Valley study area. Symbols: dots, sampling sites; stars, main urban  
3 settlements; E, electricity power plant; R, petroleum refinery; C, cement plants. The  
4 isoline numbers indicate topographic elevation (m).

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6 Figure 2. Elements in *T. recurvata* and crustal/soil sources at Mezquital Valley ( $\log_{10}$   
7 transformed mean values, ppm = mg kg<sup>-1</sup>). Geochemical data: acidic igneous rocks,  
8 calculated from Verma-Surendra (2001); agricultural soils, median values among  
9 Flores-Delgado et al., (1992); Hernández-Silva et al., (1994); Huerta et al., (2002);  
10 Lucho-Constantino et al., (2005) and Siebe (1994); limestone (Lozano and Bernal  
11 2005).

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13 Figure 3. Spatial distribution of the natural *T. recurvata*  $\delta^{13}\text{C}$  (a) and  $\delta^{15}\text{N}$  (b) isotopic  
14 ratios (per mil, ‰) at Mezquital Valley.

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17 Figure 4. Median  $\delta^{13}\text{C}$  (triangles) and Ni:V ratios (squares) in *T. recurvata* at sites along  
18 a transect oriented with the predominant wind direction at Mezquital Valley (arrow).  
19 The main petroleum combustion sources are located at km 0. The  $\delta^{13}\text{C}$  values were  
20 added 16 to allow the logarithmic transformation of the y-axis. The fitting lines are two-  
21 period running means.

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23 Figure 5. Normalized median site concentration of two metals and two PAH in *T.*  
24 *recurvata* along a NNE-SSW transect at Mezquital Valley. The 100 value represents the  
25 maximum site concentration: Ca, 32,600 mg kg<sup>-1</sup>; CRY, 82.1 ng g<sup>-1</sup>; PYR, 65.9 ng g<sup>-1</sup>;  
26 V, 197.2 mg kg<sup>-1</sup>. The border between the main agricultural and industrial areas is  
27 roughly located at km 0.

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29 Figure 6. Spatial distribution of V (a), Cr (b), Ca (c) and total PAH (d) in *T. recurvata* at  
30 Mezquital Valley. Units: mg kg<sup>-1</sup>.

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32 Figure 7. Mean element enrichment factors in *T. recurvata* at Mezquital Valley.

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34 Figure 8. Spatial distribution of Na in *T. recurvata* at Mezquital Valley. Units: mg kg<sup>-1</sup>.

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38 Figure 9. Factor 3 vs. factor 1 plot. F1 includes elements from mainly crustal origin; F3,  
39 elements and PAH from petroleum fuel combustion and cement production (Ca). The  
40 third circle includes elements related to agriculture soil sources (F2, F4 and F5, not  
41 shown).

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