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# Slower CCN growth kinetics of anthropogenic aerosol compared to biogenic aerosol observed at a rural site

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#### Abstract

Growth rates of water droplets were measured with a static diffusion cloud condensation chamber in May–June 2007 at a rural field site in Southern Ontario, Canada, 70 km north of Toronto. Observations were made during periods when the winds were from the south and the site was impacted by anthropogenic air from the US and Southern Ontario as well as during a 5-day period of northerly wind flow when the aerosol was dominated by biogenic sources. The growth of droplets on anthropogenic size-selected particles centred at 0.1  $\mu$ m diameter and composed of approximately 40% organic and 60% ammonium sulphate (AS) by mass, was delayed on the order of 1 second compared to a pure AS aerosol. Simulations of the growth rate indicate that a lowering of the water mass accommodation coefficient from  $\alpha_c=1$  to an average of  $\alpha_c=0.044$  is needed (assuming an insoluble organic with hygroscopicity parameter,  $\kappa_{org}$ , of zero). In contrast, the growth rate of the aerosol of biogenic character, consisting of >80% organic, was similar to that of pure AS. Simulations of the predominantly biogenic aerosol

- <sup>15</sup> show agreement between the observations and simulations when  $\kappa_{\rm org}$ =0.05–0.2 and  $\alpha_c$ =1. Inhibition of water uptake by the anthropogenic organic applied to an adiabatic cloud parcel model in the form of a constant low  $\alpha_c$  increases the number of droplets in a cloud compared to pure AS. If the  $\alpha_c$  is assumed to increase with increasing liquid water on the droplets, then the number of droplets decreases which could diminish the <sup>20</sup> indirect effect. The slightly lower  $\kappa_{\rm org}$  in the biogenic case decreases the number of
- droplets in a cloud compared to pure AS.

#### 1 Introduction

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Aerosol particles affect the atmospheric radiation budget directly by scattering or absorbing radiation and indirectly by acting as cloud condensation nuclei (CCN). If the amount of cloud liquid water does not change with changes in CCN, then an increase in the number of droplets leads to a decrease in the size of the droplets with increas-

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ing CCN, thus influencing the radiative properties and lifetimes of clouds. The extent of the cloud indirect effect of aerosol particles is highly uncertain (IPCC, 2007), and part of this uncertainty is understanding how the organic components contribute to the formation of cloud droplets. Carbonaceous material represents a significant fraction

- of atmospheric fine particle composition (Zhang et al., 2007). Mixtures of organic and inorganic components are complex and water uptake is difficult to characterize. The literature contains many examples of laboratory studies of the CCN behaviour of organic aerosol particles (e.g. Abbatt et al., 2005; Cruz and Pandis, 1997; Shantz et al., 2003; Shulman et al., 1996) but currently, the focus is moving away from studying organics alone in the laboratory and moving towards studying the CCN character of ambient aerosol fill formation of the analysis.
- aerosols from different sources (i.e. mixtures that may contain organics) (e.g. Chang et al., 2007; Ruehl et al., 2008; Shantz et al., 2008; Stroud et al., 2007).

Two significant sources of organic aerosols are the oxidation products of biogenic emissions (e.g. monoterpenes) and anthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as well as oxidation products of enthropogenic emissions (primary as oxidation products of enthropogenic emissions

- <sup>15</sup> dation products of anthropogenic volatile organic compounds, or VOCs). Organic biogenic aerosols have been found to be reasonably CCN active in smog chamber studies (e.g. Duplissy et al., 2008; Hartz et al., 2005; Prenni et al., 2007; VanReken et al., 2005). Duplissy et al. (2008) found the aerosol formed from  $\alpha$ -pinene oxidation became more CCN active over time when exposed to sunlight. VanReken et al. (2005) found
- the CCN activity to decrease as more oxidation products condensed on the aerosol in a dark experiment. Field observations of biogenic aerosols suggest that this organic contributes to the CCN activity (Leaitch et al., 1999; Shantz et al., 2008). In contrast, several field studies focussed on anthropogenic emissions suggest that CCN closure is easily achieved if an insoluble organic is assumed, showing that the sulphate present
- in the aerosol dominated the CCN activity (Broekhuizen et al., 2006; Medina et al., 2007; Stroud et al., 2007). A number of studies suggest that the contribution of the organic components to the CCN activity increases when more of the organic component is assumed to be soluble (e.g. Chang et al., 2007; Ming and Russell, 2004).

The recent introduction of a single hygroscopicity parameter,  $\kappa$ , reduces the large

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number of physical and chemical parameters needed to describe the water uptake by growing droplets to one value (Petters and Kreidenweis, 2007). For aerosols consisting of multiple components, the value of  $\kappa_{tot}$  is given by a simple mixing rule:  $\kappa_{tot} = \sum \varepsilon_i \kappa_i$ , where  $\varepsilon_i$  is the volume fraction and *i* is the index representing the individual species (Petters and Kreidenweis, 2007). This parameter has been determined for organic particles resulting from the oxidation of biogenic VOCs in smog chamber studies as 0.1 (Duplissy et al., 2008; Prenni et al., 2007) and 0.15 (Engelhart et al., 2008) and 0.02–0.05 in a biogenic field study (Shantz et al., 2008).

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Chuang et al. (1997) showed that neglecting kinetic limitations on the water uptake of cloud droplets can lead to overestimations in cloud radiative forcing calculations and thus kinetics should be considered. The water vapour mass accommodation coefficient,  $\alpha_c$ , can have a strong effect on the condensation rate of water and remains an outstanding uncertainty in quantifying the indirect effect of aerosols on climate forcing. Literature values of  $\alpha_c$  span 2 orders of magnitude, from 0.01 to 1 (e.g. Davidovits et al. 2004; Lagraphic and Streub 2001; Magnutication and Streub 200

<sup>15</sup> al., 2004; Laaksonen et al., 2005; Marek and Straub, 2001; Mozurkewich, 1986, and the references therein) and have been estimated to be as low as  $10^{-5}$  for organic films on cloud droplet surfaces (Chuang, 2003). Ruehl et al. (2008) found that approximately 60% of the ambient CCN of urban, polluted regional and continental origins grew into cloud droplets at a similar rate as ammonium sulphate (AS). They observed a number of cases that had a lower  $\alpha_c$  compared to AS at various sites.

In this paper, observations from a field study at Egbert, Ontario, Canada during the late spring of 2007 are used to contrast the CCN cloud droplet water uptake by aerosol particles containing organics from anthropogenic vs. biogenic sources. The CCN water uptake growth kinetics were studied using a cloud droplet growth model that utilizes  $\kappa$ 

<sup>25</sup> to determine the hygroscopicity of the organic aerosol (Shantz et al., 2008). The rate of growth of aerosol particles containing anthropogenic organic components is found to be reduced.

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#### 2 Experiment

The Egbert 2007 field campaign took place May-June 2007 at a rural field site in Southern Ontario, Canada, 70 km north of Toronto (44.23 N, 79.78 W, see Fig. 1). Previous studies of the atmospheric aerosol at this site are discussed by Chang et al. (2007) and Rupakheti et al. (2005).

#### 2.1 Instrumentation

Much of the instrumentation used during the Egbert 2007 field campaign is described by Slowik et al. (2009), Vlasenko et al. (2009) and Chan et al. (2009). The focus of this work is on data from a static thermal diffusion chamber (University of Wyoming Model 100 CCN counter) used to study the initial growth rates of cloud droplets at a constant supersaturation (Shantz et al., 2003, 2008). The experimentally determined supersaturation used in this work is  $S=0.35\pm0.05\%$ , found using nearly monodisperse ammonium sulphate particles as discussed in Shantz et al. (2003, 2008). The polydisperse ambient aerosol was sampled directly into the CCN counter and a TSI 3022

- <sup>15</sup> Condensation Particle Counter (CPC) through a 3/8"OD stainless steel tube approximately 3 m in length with an intake point approximately 1 m above the roof of the building. In order to reduce the effect of dry particle size on the droplet growth rates, ambient aerosol particles were also size-selected with a TSI 3081 Long Differential Mobility Analyzer (DMA) during periods with measurable particle number concentrations.
- <sup>20</sup> After size selection, the nearly monodisperse particle flow was split for measurement using the CPC and the CCN counter. Measured ambient growth rates from the CCN counter were compared to those of pure ammonium sulphate (AS) calibration aerosol at similar size and number concentration as the measured ambient aerosol. The AS calibration aerosol was nebulized and then size-selected with the DMA.
- <sup>25</sup> Ambient aerosol particle size distributions between 0.01  $\mu$ m and 0.4  $\mu$ m geometric diameter were determined with a Scanning Mobility Particle Sizer (SMPS) comprised of a TSI 3071 DMA and a TSI 3010 CPC (TSI Inc., St. Paul, MN, USA).



An Aerodyne time-of-flight (C-ToF) Aerosol Mass Spectrometer (AMS) (Aerodyne Research, Inc., Billerica, MA, USA) provides size-resolved, non-refractory chemical composition of submicron aerosol particles measured in the range of approximately  $0.07-0.7 \,\mu$ m vacuum aerodynamic diameter ( $D_{va}$ ), where for spherical partifices  $D_{va} = \rho_p D_{geo}$ , where  $\rho_p$  is the particle density and  $D_{geo}$  is the geometric diameter. Detailed descriptions of the AMS design, operation, quantification methods and calibration procedures are given elsewhere (Canagaratna et al., 2007; Drewnick et al., 2005; Jayne et al., 2000; Jimenez et al., 2003; Liu et al., 2007). The mass concentrations have been corrected for an average collection efficiency of 0.6 determined for this study (Slowik et al., 2009).

#### 3 Results and discussion

#### 3.1 Time series overview

Figure 2 shows a time series of wind speed and direction (Fig. 2a, dates/times are in local time, LT, Eastern Daylight Savings Time, EDT) and mass concentrations from
the AMS (Fig. 2b) (Slowik et al., 2009). Air reaching the site from the south is typically influenced by anthropogenic sources from the US and the highly populated Southern Ontario region (e.g. 7 June in Fig. 1a). During times of southerly winds (Fig. 2a, red shades), high mass concentrations of organic and sulphate were measured (e.g. 31 May–3 June and 7–9 June inclusive).

When winds were from the north (Fig. 2a, blue shades), the aerosol mass concentrations were usually lower (Fig. 2b). An exception to this was near the end of the study, 9–13 June, when north winds persisted (with few anthropogenic sources) and air reaching the site has travelled over a large fetch of mixed forests (e.g. 13 June in Fig. 1b). Figure 2b shows the organic mass concentration increased steadily over this

time period. The gas-phase and aerosol signatures as well as regional modelling indicates that the origin of most of the organic aerosol during this period was from the





oxidation of biogenic precursors (Slowik et al., 2009).

#### 3.2 CCN growth kinetics of monodisperse aerosol

Examples of droplet growth curves of size-selected ambient particles compared to AS particles at *S*=0.35% are shown in Figs. 3a and 4a. The CCN counter detector voltage
<sup>5</sup> (Fig. 3a) is proportional to the cross section of the growing cloud droplets. The data of Fig. 3 were collected during a period when the winds were from the southwest (Fig. 1a) and the aerosol had crossed over populated Southern Ontario and the Ohio valley. The AS calibration aerosol begins to accumulate water at approximately 2 s and grows through 8 s. After 8 s, the water droplets begin to fall from the detection region and the signal decreases. The initial growth of the ambient anthropogenic aerosol is delayed relative to the AS particles.

A droplet growth model that mimics the chamber supersaturation (Shantz et al., 2003, 2008) was used to investigate possible explanations for the delayed growth of the ambient aerosol. The model is initiated with nearly monodisperse particles with approximately 87% of the particles centred at 0.1  $\mu$ m and 13% of the particles centred at 0.14  $\mu$ m (the doubly charged particle mode that exit the DMA). Figure 3b shows the model results limited to 5 s as well as the observed growth rates for the AS and ambient aerosols from Fig. 3a. Only the first 5 s are shown as the model does not include the droplet gravitational settling at larger droplet sizes. The left and right axes are scaled according to the AS calibration results (Shantz et al., 2008). We assume that the water mass accommodation coefficient,  $\alpha_c$ , for the AS case is unity. Figure 3c shows the AMS time-of-flight chemical composition measurements. All conversions from  $D_{va}$  to  $D_{geo}$  were performed assuming spherical particles and using the following material densities:  $\rho_{AS}$ =1.77 g cm<sup>-3</sup> and  $\rho_{org}$ =1.3 g cm<sup>-3</sup>. The mass fractions of the ambient

<sup>25</sup> particles were estimated from the data shown in Fig. 3c to be 62% for AS and 38% for the organic component at  $D_{va}$ =0.13–0.19  $\mu$ m (corresponding to  $D_{geo}$ =0.08–0.12  $\mu$ m with midpoint  $D_{geo}$ =0.1  $\mu$ m). Initially, we assumed the surface tension of water, a mass

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accommodation coefficient of  $\alpha_c$ =1 (Davidovits et al., 2004; Laaksonen et al., 2005; Mozurkewich, 1986), a  $\kappa_{AS}$  of 0.61 (Petters and Kreidenweis, 2007) and a  $\kappa_{org}$  of 0 (i.e. the organic was assumed to not contribute to the initial water uptake). Although there was a slight delay in the ambient simulations compared to the AS simulations, it is not nearly as significant as the observed delay, suggesting that the organic compounds

present in the ambient aerosol inhibited water uptake by these CCN.

There are 2 adjustable parameters for matching the simulations to the observed growth curves,  $\kappa_{\rm org}$  and  $\alpha_c$ . Since  $\kappa_{\rm org}$ =0 is already overestimating the observed delay, increasing  $\kappa_{\rm org}$  will not help achieve agreement between the observations and simulations. Instead, lower values of  $\alpha_c$  (0.04–0.06) produce delays closer to that observed.

- This result may be consistent with Stroud et al. (2007) who found that  $\alpha_c = 0.07$  was needed to achieve closure in a case study of a pollution episode. If  $\alpha_c$  increases with the accumulation of water on the particles, then it may explain the rise in the ambient growth curve after 4 s. The effect of an increasing  $\alpha_c$  with water uptake will be discussed further in Sect. 3.4.
  - This analysis was performed for the 6 anthropogenic monodisperse cases observed on 3 different days during the field campaign and the results are summarized in Table 1. The delays in the initial growth rates of the CCN droplets were estimated by comparing the ambient growth rate with the corresponding AS calibration growth rate at the same size and number concentration. The  $\alpha_c$  ranges from 0.02 to 0.06 and the mean  $\alpha_c$  is 0.044.

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The size-selected aerosol sampled during the biogenic period is shown in Fig. 4a. Figure 4b shows this period was dominated by an organic aerosol. Growth of the  $0.12 \,\mu$ m biogenic aerosol is not delayed relative to the AS aerosol even though the majority of the aerosol is organic (Fig. 4b). Although our previous field studies have indicated a CCN-active biogenic aerosol (Leaitch et al., 1999; Shantz et al., 2008), the comparable CCN activity between the organic-dominated aerosol and the pure AS

aerosol observed at Egbert was unexpected.

For simulations of the ambient aerosols, we applied an approximate chemical com-

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position of 12% AS ( $\kappa_{AS}$ =0.61) and 88% organic mass fractions (varying  $\kappa_{org}$ ) based on the AMS mass distribution (Fig. 4b) and assumed that  $\alpha_c$ =1. For the  $\kappa_{org}$ =0 case, there is more delay simulated (not shown) than was observed, suggesting that the organic component has an elevated value of  $\kappa_{org}$  and is contributing to the water uptake.

<sup>5</sup> The aerosol numbers are extremely low in this case and the estimation of  $\kappa_{\text{org}}$  from the comparison of the simulations and observations can only be constrained within a range of 0.05–0.2. This of course is based on an  $\alpha_c$  of 1, but it is possible that the value of  $\kappa_{\text{org}}$  is higher than 0.2, in which case  $\alpha_c$  must be significantly lower than 1.

The AMS data suggest that the anthropogenic aerosol was more oxygenated than the biogenic aerosol (Slowik et al., 2009); the latter aerosol perhaps consisted of fresh, small and soluble organic compounds that activate readily. The fact that the more oxygenated anthropogenic aerosol exhibits characteristics that suggest it slows down the rate of water uptake and that the less oxygenated biogenic aerosol takes up water readily is counter to the more common concepts of organic CCN activation (Kanakidou

- et al., 2005). We suggest that the more oxygenated anthropogenic organic components may have a higher average molecular weight and be less soluble, and that the water uptake suppression may be due to organic films arising from the HOA coating the aerosol; however, we are comparing organic compounds from different sources and the individual types of organics measured in both of these cases are unknown. Ruehl et
- <sup>20</sup> al. (2008) observed at least 1 day of moderate to strong condensational droplet growth inhibition relative to AS at each of the urban, polluted regional and continental sites, consistent with the present observations.

#### 3.3 CCN growth kinetics of polydisperse aerosol

We compare the anthropogenic and biogenic CCN abilities further using polydisperse data from different days at S=0.35%. Because we could not produce an AS calibration with the exact polydisperse ambient aerosol size distribution, we rely on the model described in Sect. 3.2 to study the effects of chemical composition on water uptake.

Figure 5a shows the droplet growth rates of the anthropogenic aerosol observed



from the CCN counter for 1 June. The error bars show the uncertainties that arise from errors in the calibrated supersaturation. The model size distribution assumption (based on the SMPS) was set up as shown in Fig. 5b. From the chemical mass distribution from the AMS (Fig. 5b), the growth rates were modelled assuming modes 1

- <sup>5</sup> and 2 consist of 50% AS and 50% organic and mode 3 consists of 60% AS and 40% organic (mass fractions) assuming  $\kappa_{\rm org}$ =0. The best match with the observations is for the assumption of  $\alpha_c$ =0.07 and  $\alpha_c$  can be constrained within the supersaturation uncertainty to  $\alpha_c$ =0.05–0.15, which are at the upper end of the result for the size-selected anthropogenic aerosol (Sect. 3.2).
- <sup>10</sup> Following the same procedure for a biogenic case (Fig. 6, with chemical composition assumptions that all 3 modes are 20% AS and 80% organic), we find that  $\kappa_{org}$ =0.2 (0.05 in the first few seconds of growth) and  $\alpha_c$ =1 gives the best agreement between the model and the observed growth curves. The range  $\kappa_{org}$ =0.05-0.36 demonstrates the values within the supersaturation uncertainty. The biogenic organic must be con-<sup>15</sup> tributing to the water uptake in this case as the levels of sulphate are low. These poly-
- <sup>15</sup> tributing to the water uptake in this case as the levels of sulphate are low. These polydisperse experiments verify the findings from Sect. 3.2, which had higher uncertainties in the monodisperse biogenic case due to the much lower number concentrations.

#### 3.4 Adiabatic cloud parcel model simulations

The results from the previous sections were used in an adiabatic cloud parcel model (Shantz et al., 2003) to consider the impact of the initial water uptake delay on the number concentrations of cloud droplets. Table 2 shows the number of activated droplets  $(N_d)$  above the maximum supersaturation in a cloud for updraft velocities of 20 cm s<sup>-1</sup> and 100 cm s<sup>-1</sup> and for aerosol number and chemical size distributions represented by the anthropogenic and biogenic observations. All size distributions are scaled to 1500 cm<sup>-3</sup> total number concentration for simpler comparison.

The biogenic simulation included  $\kappa_{\rm org}$ =0.2 and  $\alpha_c$ =1 (Sect. 3.3) as well as a base case of pure AS. These assumptions lead to a lowering of the  $N_d$  to 0.80–0.89 of the  $N_d$  obtained for pure AS due to a lower  $\kappa_{\rm org}$  (0.2) than  $\kappa_{AS}$  (0.61).

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The anthropogenic simulation included  $\kappa_{org}=0$  and  $\alpha_c=0.044$  (the average value found in Sect. 3.2) as well as a varying  $\alpha_c$  that increased with increasing liquid water content (LWC) in each particle size bin. The base cases were run with the same chemical size distributions but with  $\alpha_c=1$  and then for pure AS and  $\alpha_c=1$ . The delay observed for the anthropogenic cases modelled with  $\alpha_c$  of 0.044 leads to an  $N_d$  that is 1.13–1.53 times higher than the case with the same chemical compositions but with  $\alpha_c=1$  (Table 2). This lower  $\alpha_c$  also results in an  $N_d$  that is identical (for the 20 cm s<sup>-1</sup> simulation) and 1.22 times higher (for the 100 cm s<sup>-1</sup> simulation) compared to that for pure AS ( $\alpha_c=1$ ). If we assume  $\alpha_c$  increases with the accumulation of water on the

- <sup>10</sup> particles, the  $N_d$  decreases compared to the constant  $\alpha_c$  simulations in the 100 cm s<sup>-1</sup> simulation. In this case, the larger particles activate readily, increasing their  $\alpha_c$  and depleting the excess water vapour so that the smaller sizes, for which the  $\alpha_c$  remains lower, are unable to activate. Thus, depending on if and how  $\alpha_c$  changes during the condensation process, the anthropogenic organic components may either enhance or diminich the indirect effect.
- <sup>15</sup> diminish the indirect effect. Overall, these simulations suggest that the delay in the anthropogenic aerosol water uptake and the biogenic aerosol's lower  $\kappa_{\rm org}$  compared to  $\kappa_{AS}$  may lead to slightly lower  $N_d$  than pure AS. The exception to this is if the  $\alpha_c$ remains low throughout the droplet growth process; a possibility that seems unlikely.

A second effect that shows up in the modelling is of some significance to the in-<sup>20</sup> direct effect. Hypothetically, if the updraft velocity is the same but one case has a reduced growth rate compared to the other that leads to reduced  $N_d$ , then the adiabatic cloud LWC has to be reduced. Even if the  $N_d$  increase, persistently low growth rates (i.e.  $\alpha_c$ =0.044) reduce the LWC in the case of the higher updraft speed (Table 2). At the lower updraft speed, the reduction in LWC for the variable  $\alpha_c$  is about 0.03%, whereas the reduction for the higher updraft is about 0.4% compared to pure AS. This indicates that at least at cloud base and for higher updraft speeds, there can be insufficient time for the droplets to reach equilibrium when organic components from anthropogenic sources are present in the aerosol.

With a persistent smaller  $\alpha_c$ , the  $N_d$  are higher because the cloud base supersatu-



ration is increased. However, the effect of the higher  $N_d$  may be offset by a reduction in LWC. In the more likely situation where the  $\alpha_c$  starts relatively low and increases as the droplets increase in size, the effects of anthropogenic organics on the growth rates is to reduce both the cloud  $N_d$  and LWC. This result should not suggest that anthropogenic organics in general overall reduce the indirect effect of aerosol particles because the organic aerosol also influences the number concentrations of CCN and may enhance the indirect effect through that process (Leaitch et al., 2009). It is important to understand all of the ways that the organic aerosol can influence clouds.

#### 4 Conclusions

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- <sup>10</sup> For a study with both air of biogenic and anthropogenic influence, kinetics of the initial growth of cloud droplets in a CCN counter were measured and simulated. The organic components in the anthropogenic aerosol delayed the initial growth of these particles into cloud droplets when compared with a pure ammonium sulphate (AS) aerosol (*S*=0.35%, 0.1  $\mu$ m particles). The best agreement between the observations and simulations was found for  $\kappa_{org}$ =0 and an average mass accommodation coefficient of  $\alpha_c$ =0.044. A different effect was observed for fresher biogenic particles, where the organic-dominated particles exhibited similar initial growth rates as pure AS particles. The best agreement between the observations and simulations in these cases was found for  $\kappa_{org}$ =0.05–0.2 and  $\alpha_c$ =1. Higher values of  $\kappa_{org}$  are possible but that requires decreased  $\alpha_c$  in order to match the modelled and observed growth rates. The delay
- in the initial growth of the cloud droplets on anthropogenic aerosols suggest organic inhibition, possibly due to the formation of an organic film that coated the aerosol. Even though the biogenic aerosol appears to be fresher, it takes up water readily perhaps because the fresh secondary organic compounds from the oxidation of forest emissions
- are soluble. Ruehl et al. (2008) observed droplet growth inhibition (normalized to labgenerated AS) for aerosols in urban, polluted and continental air masses, consistent with the findings from our study.





Applying these findings in an adiabatic cloud parcel model, we find that the delay in the anthropogenic aerosol water uptake could increase (constant low  $\alpha_c$ ) or decrease (increasing  $\alpha_c$  with increasing liquid water on the droplets) the  $N_d$  compared to either an aerosol with the same chemical composition and  $\alpha_c=1$  or a pure AS aerosol also with  $\alpha_c=1$ . If  $\alpha_c$  starts at a lower value and increases as more water accumulates on

the droplet, the model results show a lower  $N_d$  and LWC, which slightly reduces the indirect effect. The biogenic aerosol's  $\kappa_{\text{org}}$  also appears to lead to slightly lower  $N_d$  compared with the pure AS simulation.

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#### References

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Abbatt, J. P. D., Broekhuizen, K., and Kumar, P. P.: Cloud condensation nucleus activity of internally mixed ammonium sulfate/organic acid aerosol particles, Atmos. Environ., 39, 4767– 4778, 2005.

Broekhuizen, K., Chang, R. Y. W., Leaitch, W. R., Li, S. M., and Abbatt, J. P. D.: Closure between measured and modeled cloud condensation nuclei (CCN) using size-resolved aerosol

- compositions in downtown Toronto, Atmos. Chem. Phys., 6, 2513–2524, 2006, http://www.atmos-chem-phys.net/6/2513/2006/.
  - Canagaratna, M. R., Jayne, J. T., Jimenez, J. L., Allan, J. D., Alfarra, M. R., Zhang, Q., Onasch, T. B., Drewnick, F., Coe, H., Middlebrook, A., Delia, A., Williams, L. R., Trimborn, A. M.,

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Northway, M. J., DeCarlo, P. F., Kolb, C. E., Davidovits, P., and Worsnop, D. R.: Chemical and microphysical characterization of ambient aerosols with the aerodyne aerosol mass spectrometer, Mass Spectrom. Rev., 26, 185–222, 2007.

Chan, T. W., Huang, L., Leaitch, W. R., Sharma, S., Brook, J. R., Slowik, J. G., Abbatt, J. P.

<sup>5</sup> D., Brickell, P. C., Liggio, J., Li, S.-M., and Moosmüller, H.: Determination of OM/OC ratios and specific attenuation coefficients in ambient fine PM at a rural site in southern Ontario: Implications for emission sources, particle aging, and radiative forcing, Atmos. Chem. Phys. Discuss., in review, 2009.

Chang, R. Y. W., Liu, P. S. K., Leaitch, W. R., and Abbatt, J. P. D.: Comparison between mea-

- <sup>10</sup> sured and predicted CCN concentrations at Egbert, Ontario: Focus on the organic aerosol fraction at a semi-rural site, Atmos. Environ., 41, 8172–8182, 2007.
  - Chuang, P. Y., Charlson, R. J., and Seinfeld, J. H.: Kinetic limitations on droplet formation in clouds, Nature, 390, 594–596, 1997.

Chuang, P. Y.: Measurement of the timescale of hygroscopic growth for atmospheric aerosols, J. Geophys. Res.-Atmos., 108, 4282, doi:10.1029/2002JD002757, 2003.

Cruz, C. N., and Pandis, S. N.: A study of the ability of pure secondary organic aerosol to act as cloud condensation nuclei, Atmos. Environ., 31, 2205–2214, 1997.

Davidovits, P., Worsnop, D. R., Jayne, J. T., Kolb, C. E., Winkler, P., Vrtala, A., Wagner, P. E., Kulmala, M., Lehtinen, K. E. J., Vesala, T., and Mozurkewich, M.: Mass accom-

- <sup>20</sup> modation coefficient of water vapor on liquid water, Geophys. Res. Lett., 31, L22111, doi:10.1029/2004GL020835, 2004.
  - Drewnick, F., Hings, S. S., DeCarlo, P., Jayne, J. T., Gonin, M., Fuhrer, K., Weimer, S., Jimenez, J. L., Demerjian, K. L., Borrmann, S., and Worsnop, D. R.: A new time-of-flight aerosol mass spectrometer (TOF-AMS) Instrument description and first field deployment, Aerosol Sci. Technol., 39, 637–658, 2005.
  - Duplissy, J., Gysel, M., Alfarra, M. R., Dommen, J., Metzger, A., Prevot, A. S. H., Weingartner, E., Laaksonen, A., Raatikainen, T., Good, N., Turner, S. F., McFiggans, G., and Baltensperger, U.: Cloud forming potential of secondary organic aerosol under near atmospheric conditions, Geophys. Res. Lett., 35, L03818, doi:10.1029/2007GL031075, 2008.
- <sup>30</sup> Engelhart, G. J., Asa-Awuku, A., Nenes, A., and Pandis, S. N.: CCN activity and droplet growth kinetics of fresh and aged monoterpene secondary organic aerosol, Atmos. Chem. Phys., 8, 3937–3949, 2008,

http://www.atmos-chem-phys.net/8/3937/2008/.

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#### Anthropogenic and biogenic CCN growth kinetics

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and Pandis, S. N.: Cloud condensation nuclei activation of monoterpene and sesquiterpene secondary organic aerosol, J. Geophys. Res.-Atmos., 110, 1–8, 2005. Jayne, J. T., Leard, D. C., Zhang, X. F., Davidovits, P., Smith, K. A., Kolb, C. E., and Worsnop, D. B.: Development of an aerosol mass spectrometer for size and composition analysis of

Hartz, K. E. H., Rosenorn, T., Ferchak, S. R., Raymond, T. M., Bilde, M., Donahue, N. M.,

- D. R.: Development of an aerosol mass spectrometer for size and composition analysis of submicron particles, Aerosol Sci. Technol., 33, 49–70, 2000.
  - Jimenez, J. L., Jayne, J. T., Shi, Q., Kolb, C. E., Worsnop, D. R., Yourshaw, I., Seinfeld, J. H., Flagan, R. C., Zhang, X. F., Smith, K. A., Morris, J. W., and Davidovits, P.: Ambient aerosol sampling using the Aerodyne Aerosol Mass Spectrometer, J. Geophys. Res.-Atmos., 108, 8425, doi:10.1029/2001JD001213, 2003.
- Kanakidou, M., Seinfeld, J. H., Pandis, S. N., Barnes, I., Dentener, F. J., Facchini, M. C., Van Dingenen, R., Ervens, B., Nenes, A., Nielsen, C. J., Swietlicki, E., Putaud, J. P., Balkanski, Y., Fuzzi, S., Horth, J., Moortgat, G. K., Winterhalter, R., Myhre, C. E. L., Tsigaridis, K., Vignati, E., Stephanou, E. G., and Wilson, J.: Organic aerosol and global climate modelling:
- <sup>15</sup> a review, Atmos. Chem. Phys., 5, 1053–1123, 2005, http://www.atmos-chem-phys.net/5/1053/2005/.
  - Laaksonen, A., Vesala, T., Kulmala, M., Winkler, P. M., and Wagner, P. E.: Commentary on cloud modelling and the mass accommodation coefficient of water, Atmos. Chem. Phys., 5, 461–464, 2005,
- <sup>20</sup> http://www.atmos-chem-phys.net/5/461/2005/.

10

- Leaitch, W. R., Bottenheim, J. W., Biesenthal, T. A., Li, S. M., Liu, P. S. K., Asalian, K., Dryfhout-Clark, H., Hopper, F., and Brechtel, F.: A case study of gas-to-particle conversion in an eastern Canadian forest, J. Geophys. Res.-Atmos., 104, 8095–8111, 1999.
- Leaitch, W. R., Russell, L. M., Lohmann, U., Shantz, N. C., Garrett, T., Toom-Sauntry, D., Strapp, J. W., Hayden, K. L., Marshall, J., Worsnop, D., and Jayne, J.: Do carbonaceous
  - particles produce brighter clouds than sulphate particles?, Nature Geosci., in review, 2009. Liu, P. S. K., Deng, R., Smith, K. A., Williams, L. R., Jayne, J. T., Canagaratna, M. R., Moore, K.,
  - Onasch, T. B., Worsnop, D. R., and Deshler, T.: Transmission efficiency of an aerodynamic focusing lens system: Comparison of model calculations and laboratory measurements for the Aerodyne Aerosol Mass Spectrometer, Aerosol Sci. Technol., 41, 721–733, 2007.
- the Aerodyne Aerosol Mass Spectrometer, Aerosol Sci. Technol., 41, 721–733, 2007.
   Marek, R. and Straub, J.: Analysis of the evaporation coefficient and the condensation coefficient of water, Int. J. Heat Mass Transf., 44, 39–53, 2001.

Medina, J., Nenes, A., Sotiropoulou, R. E. P., Cottrell, L. D., Ziemba, L. D., Beckman, P. J.,

9, 13775–13799, 2009

#### Anthropogenic and biogenic CCN growth kinetics





and Griffin, R. J.: Cloud condensation nuclei closure during the International Consortium for Atmospheric Research on Transport and Transformation 2004 campaign: Effects of size-resolved composition, J. Geophys. Res.-Atmos., 112, D10S31, doi:10.1029/2006JD007588, 2007.

- <sup>5</sup> Ming, Y. and Russell, L. M.: Organic aerosol effects on fog droplet spectra, J. Geophys. Res.-Atmos., 109, 13–14, 2004.
  - Mozurkewich, M.: Aerosol Growth and the Condensation Coefficient for Water a Review, Aerosol Sci. Technol., 5, 223–236, 1986.
  - Petters, M. D. and Kreidenweis, S. M.: A single parameter representation of hygroscopic growth
- and cloud condensation nucleus activity, Atmos. Chem. Phys., 7, 1961–1971, 2007, http://www.atmos-chem-phys.net/7/1961/2007/.
  - Prenni, A. J., Petters, M. D., Kreidenweis, S. M., DeMott, P. J., and Ziemann, P. J.: Cloud droplet activation of secondary organic aerosol, J. Geophys. Res., 112, D10223, doi:10.1029/2006JD007963, 2007.
- Ruehl, C. R., Chuang, P. Y., and Nenes, A.: How quickly do cloud droplets form on atmospheric particles?, Atmos. Chem. Phys., 8, 1043–1055, 2008, http://www.atmos-chem-phys.net/8/1043/2008/.
  - Rupakheti, M., Lohmann, U., Leaitch, W. R., Hayden, K., Brickell, P., Lu, G., Li, S.-M., Toom-Sauntry, D., Bottenheim, J. W., Brook, J. R., Vet, R., Jayne, J. T., and Worsnop, D. R.: An
- <sup>20</sup> intensive study of the size and composition of submicron atmospheric aerosols at a rural site in Ontario, Canada, Aerosol Sci. Technol., 39, 722–736, 2005.
  - Shantz, N. C., Leaitch, W. R., and Caffrey, P. F.: Effect of organics of low solubility on the growth rate of cloud droplets, J. Geophys. Res.-Atmos., 108, 4168–4176, 2003.

Shantz, N. C., Leaitch, W. R., Phinney, L., Mozurkewich, M., and Toom-Sauntry, D.: The effect

of organic compounds on the growth rate of cloud droplets in marine and forest settings, Atmos. Chem. Phys., 8, 5869–5887, 2008,

http://www.atmos-chem-phys.net/8/5869/2008/.

- Shulman, M. L., Jacobson, M. C., Charlson, R. J., Synovec, R. E., and Young, T. E.: Dissolution behavior and surface tension effects of organic compounds in nucleating cloud droplets,
- <sup>30</sup> Geophys. Res. Lett., 23, 277–280, 1996.
  - Slowik, J. G., Stroud, C., Bottenheim, J. W., Brickell, P. C., Chang, R. Y.-W., Liggio, J., Makar,
    P. A., Martin, R., Moran, M. D., Shantz, N. C., Sjostedt, S. J., van Donkelaar, A., Vlasenko,
    A., Wiebe, H. A., Xia, A. G., Zhang, J., Leaitch, W. R., and Abbatt, J. P. D.: A large biogenic

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#### Anthropogenic and biogenic CCN growth kinetics

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organic aerosol source from Eastern Canadian forests, Nature Geosci., submitted, 2009. Stroud, C. A., Nenes, A., Jimenez, J. L., DeCarlo, P. F., Huffman, J. A., Bruintjes, R., Nemitz, E., Delia, A. E., Toohey, D. W., Guenther, A. B., and Nandi, S.: Cloud activating properties of aerosol observed during CELTIC, J. Atmos. Sci., 64, 441–459, 2007.

- <sup>5</sup> VanReken, T. M., Ng, N. L., Flagan, R. C., and Seinfeld, J. H.: Cloud condensation nucleus activation properties of biogenic secondary organic aerosol, J. Geophys. Res.-Atmos., 110, D07206, doi:10.1029/2004JD005465, 2005.
  - Vlasenko A., Slowik, J. G., Bottenheim, J. W., Brickell, P. C., Chang, R. Y.-W., Macdonald, A. M., Shantz, N. C., Sjostedt, S. J., Wiebe, H. A., Leaitch, W. R., and Abbatt, J. P. D.: PTR-MS
- VOC measurements at a rural Ontario site: sources and correlations to aerosol composition, J. Geophys. Res., in review, 2009.
- Zhang, Q., Jimenez, J. L., Canagaratna, M. R., Allan, J. D., Coe, H., Ulbrich, I., Alfarra, M. R., Takami, A., Middlebrook, A. M., Sun, Y. L., Dzepina, K., Dunlea, E., Docherty, K., De-Carlo, P. F., Salcedo, D., Onasch, T., Jayne, J. T., Miyoshi, T., Shimono, A., Hatakeyama, S., Takegawa, N., Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S., Weimer, S., Demerjian, K., Williams, P., Bower, K., Bahreini, R., Cottrell, L., Griffin, R. J., Rautiainen, J., Sun, J. Y., Zhang, Y. M., and Worsnop, D. R.: Ubiquity and dominance of oxygenated species in organic aerosols in anthropogenically-influenced Northern Hemisphere midlatitudes, Geophys. Res. Lett., 34, L13801, doi:10.1029/2007GL029979, 2007.

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Table 1. Observations from the size-selected nearly monodisperse ambient aerosol of stronger	
anthropogenic influence. Simulations were used to determine $\alpha_c$ for the anthropogenic cases	
compared to AS calibration simulations (assuming $\alpha_c$ of unity). The organics in the simulations	
are assumed to have $\kappa_{\rm org}$ of zero.	

Date (dd mmm)	Time D <sub>geo</sub> Π mm) (μm) (μ (		Number conc. (cm <sup>-3</sup> )	Organic mass fraction at D <sub>geo</sub>	Inorganic mass fraction at D <sup>a</sup> <sub>geo</sub>	Approx. delay time (sec)	$\alpha_c$ estimated range	$\alpha_c$ midpoint of estimated range	
27 May	13:33–14:05	0.12	95–105	0.53	0.47	1	0.03-0.05	0.04	
7 Jun	16:37-16:42	0.10	285–295	0.40	0.60	1	0.04-0.07	0.055	
7 Jun	16:44–16:58	0.10	265-275	0.40	0.60	1.25	0.03-0.05	0.04	
7 Jun	17:01-18:00	0.10	215–225	0.38	0.62	1	0.04-0.06	0.05	
7 Jun	18:00-18:17	0.10	205–215	0.39	0.61	0.75	0.05-0.07	0.06	
8 Jun	11:26-11:46	0.10	65–75	0.48	0.52	1	0.01-0.03	0.02	
						Average $\alpha_c$		0.044	
					Standard de	eviation of $a_c$		0.014	

<sup>a</sup> Inorganic is assumed to be ammonium sulphate with  $\kappa_{AS}$  of 0.61.

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**Table 2.** Simulations from the adiabatic cloud parcel model for anthropogenic and biogenic case studies. The total mass fractions, hygroscopicity parameters ( $\kappa$ ) and mass accommodation coefficients ( $\alpha_c$ ) are unitless. The number of droplets ( $N_d$ ) found in these simulated clouds is determined well above the maximum supersaturation (60 m above cloud base) for 2 updraft velocities (u), 20 and 100 cm s<sup>-1</sup>. "Anth" is anthropogenic aerosol, "Biog" is biogenic aerosol and "AS" indicates a simulation that modelled the ambient size distribution from the date indicated but with the composition as pure ammonium sulphate. All simulations are scaled to the same total number concentration of 1500 cm<sup>-3</sup>. Also, only particles >0.05  $\mu$ m are simulated here.

Total Mass Fraction							u=20 cm s <sup>-1</sup>	u = 100 cm s <sup>-1</sup>	$u=20 \ cm s^{-1}$	u=10 cm s <sup>-</sup>
Date (dd mmm)	Case	Org.	Inorg. <sup>a</sup>	κ <sub>org</sub>	K <sub>AS</sub>	α <sub>c</sub>	LWC (g m <sup>-3</sup> )	LWC (g m <sup>-3</sup> )	N <sub>d</sub> (cm <sup>-3</sup> )	N <sub>d</sub> (cm <sup>-3</sup>
7 Jun	Anth	0.37	0.63	0	0.61	0.044	0.12816	0.13513	269	936
7 Jun	Anth	0.37	0.63	0	0.61	Vary 0.01 to 1 <sup>b</sup>	0.13068	0.13535	238	565
7 Jun	Anth	0.37	0.63	0	0.61	1	0.13082	0.13535	238	612
7 Jun	AS	0	1	N/A	0.61	1	0.13115	0.13539	269	768
11 Jun	Biog	0.78	0.22	0.2	0.61	1	0.13135	0.13544	292	885
11 Jun	AS	0	1	N/A	0.61	1	0.13148	0.13992	364	995

<sup>a</sup> Inorganic is assumed to be ammonium sulphate.

<sup>b</sup>  $\alpha_c$  is assumed to increase as the amount of liquid water on the droplets increase.

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**Fig. 1.** Back trajectory analyses from the NOAA Air Resources Laboratory HYSPLIT program for **(a)** an anthropogenically influenced period (7 June) and **(b)** a biogenically influenced period (13 June). Egbert, Ontario is the trajectory end point.



**Fig. 2.** Time Series from Egbert 2007 of **(a)** wind speed, colour-coded to wind direction (reds=south winds, blues=north winds) and **(b)** organic, sulphate and nitrate data from the C-ToF AMS. Indicated in black lines are the case study periods discussed in this paper. Dates/times are in EDT, local time (LT).

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**Fig. 3.** 7 June 2007, 17:00–18:00 LT anthropogenic case with south winds. (a) Observations of CCN counter droplet growth rates (shown as a voltage) size-selected to  $0.1 \,\mu$ m at S=0.35% and number concentrations of  $215-225 \,\mathrm{cm}^{-3}$ . The x-axis is the sample time as the particles activate and grow into cloud droplets. (b) The observed growth rates from the CCN counter from (a) correspond to the left y-axis. The right y-axis show the droplet cross-sectional area calculated in the simulations. The simulations assume 62% AS ( $\kappa_{AS}=0.61$ ) and 38% organic mass fractions ( $\kappa_{org}=0$ ) for a nearly monodisperse aerosol centred around  $0.1 \,\mu$ m. This figure is constrained to 5 s because the model does not simulate the droplet gravitational settling. (c) Mass distributions from the AMS particle time-of-flight measurements (right y-axis) and number size distributions (left y-axis) from the SMPS. The x-axis shows the geometric diameter (converted from  $D_{va}$  assuming a particle density of 1.63 g cm<sup>-3</sup>, calculated as described in Sect. 3.2).





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