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The continental source of glyoxal estimated by the synergistic use of spaceborne measurements and inverse modelling

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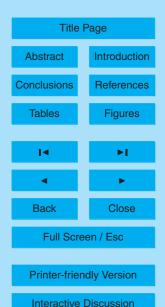
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Abstract

Tropospheric glyoxal and formaldehyde columns retrieved from the SCIAMACHY satellite instrument in 2005 are used with the IMAGESv2 global chemistry-transport model and its adjoint in a two-compound inversion scheme designed to estimate the continental source of glyoxal. The formaldehyde observations provide an important constraint on the production of glyoxal from isoprene in the model, since the degradation of isoprene constitutes an important source of both glyoxal and formaldehyde. Current modelling studies underestimate largely the observed glyoxal satellite columns, pointing to the existence of an additional land glyoxal source of biogenic origin. We include an extra glyoxal source in the model and we explore its possible distribution and magnitude through two inversion experiments. In the first case, the additional source is represented as a direct glyoxal emission, and in the second, as a secondary formation through the oxidation of an unspecified glyoxal precursor. Besides this extra source, the inversion scheme optimizes the primary glyoxal and formaldehyde emissions, as well as their secondary production from other identified non-methane volatile organic precursors of anthropogenic, pyrogenic and biogenic origin.

In the first inversion experiment, the additional direct source, estimated at 36 Tg/yr, represents 38% of the global continental source, whereas the contribution of isoprene is equally important (30%), the remainder being accounted for by anthropogenic (20%) and pyrogenic fluxes. The inversion succeeds in reducing the underestimation of the glyoxal columns by the model, but it leads to a severe overestimation of glyoxal surface concentrations in comparison with in situ measurements. In the second scenario, the inferred total global continental glyoxal source is estimated at 108 Tg/yr, almost two times higher than the global a priori source. The extra secondary source is the largest contribution to the global glyoxal budget (50%), followed by the production from isoprene (26%) and from anthropogenic NMVOC precursors (14%). A better performance is achieved in this case, as the updated emissions allow for a satisfactory agreement of the model with both satellite and in situ glyoxal observations.

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1 Introduction

Glyoxal (CHOCHO) and formaldehyde (HCHO) are short-lived intermediate products in the oxidation of non-methane volatile organic compounds (NMVOCs) emitted by vegetation, fires and anthropogenic activities. They are also directly emitted during fossil fuel and biofuel combustion and biomass burning. Both compounds absorb in the UV-visible spectral region and have been measured by the SCIAMACHY satellite sensor since 2003, and more recently, by the OMI and GOME-2 instruments.

Although current modelling studies succeed in reproducing reasonably well the variability and the magnitude of the observed HCHO columns (Fu et al., 2007; Barkley et al., 2008; Stavrakou et al., 2009a), a large underestimation of the glyoxal columns has been reported (Myriokefalitakis et al., 2008; Fu et al., 2008). These discrepancies point to the existence of additional land and marine sources, and/or to an underestimation of the known sources. This study aims at determining the global distribution and strength of the glyoxal continental sources. To this purpose, we utilize synergistically HCHO and CHOCHO satellite columns in a two-compound inverse modelling framework based on the adjoint of the IMAGES model (Müller and Stavrakou, 2005). In this approach, information gained from HCHO can help to constrain the sources of CHOCHO, since both gases have common precursors and are interrelated through chemistry (e.g. through the OH fields).

The grid-based inverse modelling approach used in this study is designed to estimate the fluxes emitted from every model grid cell for each emission category (Stavrakou and Müller, 2006). The adjoint model technique is used to minimize the bias between the model and the observations, while accounting for the errors on the observations and on the emission variables to be determined. This method has been recently used to constrain the pyrogenic and biogenic NMVOC sources of HCHO based on a 4-year satellite dataset (Stavrakou et al., 2009b). A similar setup will be used in this study in order to derive a "top-down" inventory of the glyoxal lanfd fluxes.

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In the current inversion framework, primary formaldehyde and glyoxal emission and secondary production from NMVOC precursors (anthropogenic, biogenic and pyrogenic) are optimized for the year 2005, as well as an additional missing source of glyoxal of biogenic origin. Two inversion experiments are investigated. In the first, the extra source is assumed to be directly emitted glyoxal, and in the second, glyoxal is formed through oxidation of an unknown glyoxal precursor (UVOC) with a lifetime of five days. The impact of the choice for the UVOC lifetime is assessed through sensitivity simulations. The magnitudes and distributions of the glyoxal sources inferred by the inversion case studies are discussed, and the optimized glyoxal concentrations are evaluated against in situ measurements.

The UVOC precursor of the second inversion should not be considered as a single molecular species, but as a generic species possibly representing a collection of compounds leading to glyoxal formation after a number of non-specified intermediate steps. These compounds could be directly emitted or produced in the oxidation of short-lived biogenic precursors like terpenoids. Detailed knowledge on the oxidation mechanisms and glyoxal yield of such compounds is still lacking. Moreover, the release of glyoxal from organic aerosols is another potentially significant source of glyoxal.

The next section provides a description of the IMAGESv2 model. Particular focus is given on the description of the chemistry of the glyoxal precursors (Sect. 2.3) and to the formation of secondary organic aerosols (Sect. 2.4) due to its importance as a glyoxal loss process. A short description of the glyoxal and formaldehyde observed columns is given in Sect. 2.5. The global glyoxal budget calculated with the model is discussed in Sect. 3. The setup for the inversion and the optimization results are presented in Sect. 4. Conclusions are drawn in the last section.

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2 The model and the measurements

2.1 The IMAGESv2 CTM

The global IMAGESv2 chemistry-transport model (Müller and Stavrakou, 2005; Stavrakou et al., 2009a) is run at the horizontal resolution of 4° × 5° and is vertically discretized at 40 sigma-pressure levels extending from the surface to 44 hPa. The model calculates the concentrations of 80 trace gases using a one-day time step. Advection is driven by monthly mean ECMWF reanalysed wind fields, whereas daily ECMWF fields are used for temperature, water vapor, cloud optical depths, and planetary boundary layer mixing. The model accounts for the impact of diurnal variations of the chemical compounds through correction factors computed via a diurnal cycle simulation with a 20-min time step. The diurnal profiles are also used to estimate the CHOCHO and HCHO concentration at the satellite overpass time (10:00 LT) from the daily averaged values calculated with a time step of one day. They depend on the location and the month. The comparison with ground-based observations uses the diurnally averaged glyoxal concentrations. The model is confronted with the data following a 4-month initialization. The calculated monthly averaged HCHO columns account for the sampling times of observations at each location.

2.2 NMVOC emissions

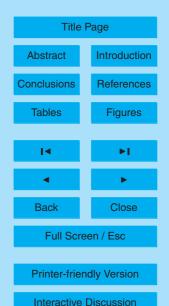
The model comprises emissions from 20 explicit NMVOCs (ethene, acetylene, isoprene, α -pinene, glycolaldehyde, ethane, propane, propene,formaldehyde, glyoxal, acetaldehyde, methylglyoxal, acetone, methyl vinyl ketone, biacetyl, acetic acid, methanol, benzene, toluene and xylene). All NMVOCs are formaldehyde precursors, and among them, ethene, acetylene, isoprene, α -pinene, glycolaldehyde, benzene, toluene, and xylene are precursors of glyoxal.

Biogenic emissions of isoprene are obtained from the MEGAN-ECMWF inventory (Müller et al., 2008). Their global estimate for 2005 is equal to 449 Tg. Emissions of

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monoterpenes (represented by the model compound α -pinene) are taken from Guenther et al. (1995) and amount to 144 Tg/yr. The biogenic source of methanol is calculated as in Jacob et al. (2005) and is estimated at 152 Tg/yr. Other biogenic sources are as in Müller and Stavrakou (2005).

The vegetation fire emissions are provided by the Global Fire Emission Database (GFEDv2) (van der Werf et al., 2006) and the speciation from Andreae (2007). Their global estimate amounts to 95 Tg for 2005, including 6.5 Tg ethene, 4.7 Tg glycolaldehyde, 5.9 Tg glycoxal, 1.6 Tg benzene, 0.94 Tg toluene and 0.28 Tg xylene.

Biofuel emissions are estimated using emission ratios to CO from Andreae (2007) and biofuel CO emissions from EDGARv3.3 inventory for 1997 (Olivier et al., 2001; Olivier, 2002). In addition, emissions associated with fossil fuel use and industrial activities are provided by the POET database (Olivier et al., 2003). For acetylene, biofuel emission (3.3 Tg/yr) and technological emissions (3 Tg/yr) have been adjusted in order to reproduce aircraft campaign measurements. The global anthropogenic emissions of ethene, glycolaldehyde and glyoxal are 5.4, 1.9, and 3.2 Tg, respectively. Aromatics emissions are obtained from the REanalysis of the TROpospheric chemical composition over the past 40 years (RETRO) database (Schultz et al., 2007). Their annual estimates are 3.5, 6.1 and 5.2 Tg for benzene, toluene and xylene, respectively.

2.3 NMVOC chemistry

The NMVOC degradation mechanism has been thoroughly presented in Stavrakou et al. (2009a). It has been designed so as to reproduce the HCHO yields of the quasi-explicit Master Chemical Mechanism (MCM) (Saunders et al., 2003) for pyrogenic and biogenic NMVOCs, and it has been further adjusted to account for the glyoxal production from the NMVOCs. Hereafter, updates of that mechanism regarding the glyoxal chemistry are briefly presented.

The degradation of isoprene is adapted from the Mainz Isoprene Mechanism (Pöschl et al., 2000) and has been further modified to include a production of two OH radicals in the reaction of first generation peroxy radicals from isoprene with HO_2 (Lelieveld

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et al., 2008). Further, the rate of the reaction of isoprene with OH is reduced by 30% in order to account for the isoprene/OH segregation effect, as recommended by Butler et al. (2008). Finally, the isoprene nighttime oxidation by NO₃ has been aligned with the MCM scheme. One-day box model simulations starting at 06:00 a.m. under high (low) NOx levels yield a molar yield of glyoxal of 9.9% (8.4%), in excellent agreement with the corresponding MCM yield of 10.1% (8.8%). For a simulation starting at 19:30 p.m., the molar glyoxal yield is estimated at 35%, very close to the MCM vield of 34%, under high or low NOx conditions. According to our chemical mechanism, reaction with OH represents 85% of the global isoprene sink, followed by ozonolysis (10%) and oxidation by NO₃ (5%). The calculated globally averaged molar glyoxal yield from isoprene is found to be 7.2%, by 16% higher than in Fu et al. (2008). The mean molar glyoxal yield from monoterpenes is 2.8%, 63.6% from acetylene, 5.7% from ethene, and 9.2% from glycolaldehyde. The formation yield of glyoxal from aromatics is taken to be the mean of the literature values reported in Arey et al. (2009): 33.3% for benzene, 22.3% for toluene and, taking into account for the relative contribution of o-, m-, and p-xylene in the total xylene emission (Middleton et al., 1990), the glyoxal yield from the lumped xylene compound is estimated at 17.3%. Note that laboratory studies conducted under close to atmospheric conditions suggest that the glyoxal yield from toluene is higher (30.6%, Volkamer et al., 2001) than assumed in this study.

Other identified precursors with significant glyoxal yields, like trimethylbenzenes, ethylbenzene, ethyltoluenes, butenes, acrolein, etc. (Volkamer et al., 2005b) are not included in the model, either due to the lack of available emission data for these compounds on the global scale, or to their estimated minor contribution to the global annual glyoxal production. For instance, the global anthropogenic emission of trimethylbenzenes, estimated at 1 Tg/yr in the RETRO database (Schultz et al., 2007), would yield only about 0.04 Tg/yr of glyoxal, whereas the glyoxal produced from the degradation of methylbutenol is estimated at 0.35 Tg/yr (Fu et al., 2008).

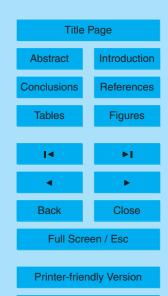
The glyoxal photodissociation reactions combine absorption cross sections by Volkamer et al. (2005a) with literature quantum yields from Sander et al. (2006). The

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OH-reaction rate is taken from Feierabend et al. (2008).

2.4 Aerosol emissions and SOA formation

Secondary organic aerosols (SOA) are assumed to be formed rapidly and irreversibly as soon as glyoxal is processed in liquid clouds. The distribution of cloud volume fraction and liquid water content are obtained from ECMWF analyses for 2005.

The irreversible uptake of glyoxal by aqueous aerosols is assumed to proceed via a uniform uptake probability, $\gamma = 2.9 \cdot 10^{-3}$ (Liggio et al., 2005). This value has been derived from laboratory experiments conducted under atmospheric conditions using ammonium sulphate seed aerosols at 55% relative humidity. Note however, that the rapidity and irreversibility of the glyoxal uptake has been questioned by Kroll et al. (2005) through experiments conducted in the absence of light and oxidants. Nevertheless, recent laboratory studies demonstrate that the uptake is greatly enhanced by the presence of oxidants and light (Volkamer et al., 2009). Another evidence for rapid glyoxal uptake is provided by direct measurements of gas-phase glyoxal (Volkamer et al., 2007), where the underestimation of the observed glyoxal concentrations can be fully accounted either through a parameterization by irreversible uptake into the aerosols with γ equal to $3.7 \cdot 10^{-3}$, or through reversible partitioning to aerosol liquid water with an effective Henry's constant K_H estimated at 4.10^9 M atm⁻¹, which, although higher than the K_H in pure water by four orders of magnitude, is found to be consistent with recent experimental values (Ip et al., 2009). Due to the high uncertainty associated to the processes leading to the SOA formation from glyoxal, two sensitivity experiments are conducted assuming either no glyoxal uptake on aerosols, or that the uptake proceeds with the coefficient $\gamma=3.7\cdot10^{-3}$ (Volkamer et al., 2007), which is by 27% higher than in Liggio et al. (2005). The results of these sensitivity studies are discussed in Sects. 3 and 4.2.

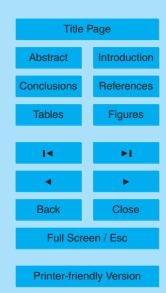
Aqueous aerosols in IMAGESv2 include liquid inorganic aerosols and hydrophilic carbonaceous aerosols. Dust and sea-salt aerosols are neglected. Inorganic aerosols (the sulfate-ammonium-nitrate-water system) are calculated using EQSAM (EQuilib-

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rium Simplified Aerosol Model) (Metzger et al., 2002), with deliquescence relative humidities (DRH) of Nenes et al. (1998). In order to account for the hysteresis of particle phase transitions (Wang et al., 2008), inorganic aerosols are assumed to be solid only when relative humidity never exceeded the DRH in the past 24 h. The description of ₅ sulfate formation follows Pham et al. (1995) with several improvements. Anthropogenic emissions (72.6 TqS/yr globally) are provided by EDGAR-FT2000 (Olivier et al., 2005). Non-eruptive volcanic emissions (7.2 TqS/yr, including 2.4 TqS/yr of non-SO₂ sulfur compounds emitted as SO₂ in the model) are provided by Andres and Kasgnoc (1998). The GFEDv2 database and the emission ratios of Andreae (2007) result in global emissions of 1.2 TqS/yr due to vegetation fires in 2005. Natural sulfur emissions include the oceanic DMS source obtained from Dentener et al. (2006) (18.3 TqS/yr) and a small biospheric emission of H₂S (0.5 TgS/yr) (Pham et al., 1995). The model includes the gas-phase oxidation of DMS and H₂S to form SO₂ (and methanesulfonic acid in the case of DMS) and the SO₂ oxidation by OH to form sulfate, with rates from Sander et al. (2006). The pH-dependent aqueous-phase oxidation of SO₂ by O₃ and H₂O₂ to form sulfate is included using kinetic data and Henry constants reported in Tie et al. (2001), with cloud volume fractions and liquid water content from ECMWF analyses.

Ammonia emissions due to domestic animals and synthetic fertilizers ($19\,TgN/yr$ globally) are obtained from Bouwman et al. (2002). Vegetation fires emitted $6.1\,TgN\,H_3-N$ in 2005, based on GFEDv2 and emission ratios of Andreae (2007). The other NH $_3$ emissions (totalling $19.4\,Tg/yr$) are provided by the GEIA inventory (Bouwman et al., 1997). The temporal variability of emissions due to soils and manure is driven by soil temperature (Park et al., 2004), whereas the seasonal variation of emissions due to crops and fertilizers follows the number of daylight hours (Adams et al., 1999).

The emissions of organic carbon (OC) and black carbon (BC) aerosols from anthropogenic activities (Bond et al., 2004) and biomass burning (van der Werf et al., 2006) are included in the model. Both OC and BC have hydrophilic and hydrophobic components. The conversion of hydrophobic to hydrophilic OC and BC due to aging is represented by an e-folding time of 6 h in tropical regions (Bellouin et al., 2007) and

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one day elsewhere (Martin et al., 2003). The mass ratio of organic matter to OC is taken to 2.1, a value valid for rural areas (Pang et al., 2006). The effective diameter of dry organic aerosols is taken to 145 nm, based on field campaigns over Amazonia (Rissler et al., 2006) and Africa (Bellouin et al., 2007). The hygroscopic growth factor is calculated by setting the activity coefficient of water to 1.1, a value typical for α -pinene SOA (Compernolle et al., 2009) leading to growth factors in good agreement with field measurements over Amazonia during the dry season (Rissler et al., 2006; Mircea et al., 2005).

2.5 Observed CHOCHO and HCHO columns

We use glyoxal and formaldehyde column abundancies retrieved from the SCIA-MACHY (Scanning imaging absorption spectrometer for atmospheric cartography) UV-visible sounder onboard ENVISAT satellite for 2005. The differential optical absorption spectroscopy (DOAS) technique (Platt et al., 1994) has been applied for both compounds. The spectral window found to be optimal for the fitting is 436–457 nm for CHOCHO (Wittrock et al., 2006) and 328.5–346 nm for HCHO (De Smedt et al., 2008). For CHOCHO, the Volkamer et al. (2005a) absorption cross sections are used, and for the HCHO retrieval the Cantrell et al. (1990) values shifted by 0.08 mm (Stavrakou et al., 2009b).

The columns are binned onto the model resolution and monthly averaged. The error on the CHOCHO columns is assumed to comprise an absolute error of 2×10^{14} molecules cm⁻² and a 30% relative error, whereas the error on the HCHO columns includes the reported retrieval error (De Smedt et al., 2008) and a model error amounting to 20% of the column, based on sensitivity calculations with the IMAGESv2 model (Stavrakou et al., 2009a). Seasonally averaged observed glyoxal and formaldehyde columns are displayed in Fig. 1. A thorough discussion on the seasonal variability of the measured columns can be found in De Smedt et al. (2008), Wittrock et al. (2006), and Vrekoussis et al. (2009). As this study focuses on land sources, oceanic data of both compounds are not used to constrain the inversion. Moreover, data in the vicin-

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ity of the south Atlantic anomaly (26.1° S, width=16.7°, 47.4° W, width=28.4°, Nichitiu et al., 2004) are also excluded.

3 The glyoxal global budget

The global a priori budget of glyoxal (in absence if any additional biogenic source) is summarized in Table 1. About 55%, 27% and 18% of the global glyoxal source are due to biogenic, anthropogenic, and pyrogenic emissions, respectively. The largest sink of glyoxal is photolysis (51%), followed by SOA formation (22%), OH oxidation (18%), and dry and wet deposition (8%). The global glyoxal source is estimated to 56 Tg/yr and the global lifetime to 2.5 h, in good agreement with literature values (45 Tg/yr, 2.9 h in Fu et al., 2008, 56 Tg/yr, 3 h in Myriokefalitakis et al., 2008). The global annual burden, estimated at 15.8 Gg/yr, is consistent with the Fu et al. (2008) estimate of 15 Gg/yr, and is found to be by 20% lower than in Myriokefalitakis et al. (2008). This discrepancy is mostly due to the SOA formation sinks which are not considered in Myriokefalitakis et al. (2008).

Directly emitted glyoxal from fires constitutes the major contribution to the total pyrogenic glyoxal production (60%), whereas the remainder is secondary production, mostly from acetylene oxidation (24%), aromatics (5%), glycolaldehyde (4%), and ethylene (7%). About 60% of the total anthropogenic glyoxal is generated through acetylene oxidation, followed by primary glyoxal (20%), and aromatics (15%). Biogenically emitted glyoxal is only of secondary origin, the isoprene oxidation being responsible for more than 90% of the total source, the remainder being formed from monoterpenes and ethene.

The estimated glyoxal sink associated with SOA formation in clouds (4.7 Tg/yr) is found to be in good agreement with the previous estimate of Fu et al. (2008), whereas the irreversible uptake on aerosols, amounting to 5.9 Tg/yr on OC/BC and to 1.8 Tg/yr on sulfates, is significantly higher than in Fu et al. (2008). The resulting mean lifetime against SOA conversion (ca. 11 h), is about two times shorter than in Fu et al. (2008).

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The lifetimes against photolysis and OH oxidation are calculated to be 4.8 h and 14 h, respectively, in good agreement with the study of Myriokefalitakis et al. (2008). Note that the glyoxal aerosol sink is even higher when using the uptake coefficient taken from Volkamer et al. (2007) (3.7·10⁻³) and is estimated at 9.2 Tg/yr.

Annually averaged observed and simulated CHOCHO columns are compared in the top panels of Fig. 2. The model is found to underestimate the observations over all continents. The model/data biases and spatiotemporal correlations averaged over large regions are summarized in Table 2. On the global scale, the a priori simulation underestimates the data by 57%, and yields an overall correlation coefficient of 0.5. The correlation is stronger over Northern America and Asia (>0.6), but falls to zero over Europe. The model/data discrepancies over the oceans are much more important (Fig. 2), but their investigation lies out of scope of this study.

The global HCHO budget is estimated at 1660 Tg/yr. According to the model calculations, the global HCHO production is due to the oxidation of methane (60%), isoprene (30%), anthropogenic (7%), and pyrogenic NMVOCs (3%), whereas photolysis is the major sink (73%), followed by the OH oxidation (23%), and deposition (4%) (Stavrakou et al., 2009a). Systematic comparisons between the monthly mean observed and simulated HCHO columns over a large number of regions, presented in Stavrakou et al. (2009a), show that the model captures the spatial pattern of the observed HCHO columns and succeeds guite well in reproducing the intensity of the observed maxima.

Monthly mean HCHO columns from SCIAMACHY are compared to modelled columns (Fig. 3, right) averaged over several tropical regions. The seasonality and intensity of the observed HCHO columns are well reproduced by the model, except for an overestimation over Indonesia and an underestimation over southern Africa. The comparisons for glyoxal on the same figure confirm the model underestimation noted above. Furthermore, in many cases, the seasonality of the observed columns is severely misrepresented.

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4 Inverse modelling

4.1 Setup for the inversion

The inversion is performed on a grid cell basis, i.e. the fluxes are updated at every grid cell and month (Stavrakou and Müller, 2006; Stavrakou et al., 2009b), except when the a priori emissions are very small (see below). Four emission categories are optimized: i) anthropogenic NMVOC emissions of glyoxal and its precursors (except for acetylene) ii) all pyrogenic NMVOCs, as in Stavrakou et al., 2009b), iii) identified biogenic NMVOC precursors (BVOC) (isoprene, monoterpenes, methanol, and ethene), and iv) unidentified biogenic VOCs (UVOC). The global a priori UVOC source is distributed as the leaf area index obtained from the Moderate Resolution Imaging Radiospectrometer (MODIS) satellite (http://modis.gsfc.nasa.gov/data/atbd/) and is scaled globally to 20 Tg/yr (Fig. 4, upper panel).

Emissions are not optimized in grid cells with very low emissions. An emission parameter is optimized in a grid cell only when the a priori maximum flux in this cell is higher than 10^{10} molecules cm⁻² s⁻¹ for the first three categories and 5×10^8 molecules cm⁻² s⁻¹ for the latter. The thresholds has been adjusted so as to avoid the existence of large disparities in the number of unknowns to be optimized from the different categories. The total number of control parameters is equal to ca. 30 000. The errors on the emission parameters are assumed to be factor of two. Correlations between the a priori errors on the control parameters are introduced through the off-diagonal elements of the error covariance matrix, in the same way as in Stavrakou et al. (2009b) for the inversion using HCHO data.

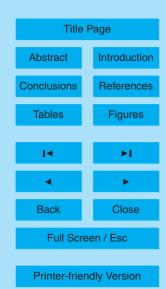
Among the above-mentioned species, isoprene, monoterpenes, glycolaldehyde, and ethene are precursors of both glyoxal and formaldehyde. Although the chemical feedbacks between glyoxal and formaldehyde are weak, performing a simultaneous inversion of both species constitutes the best way to take advantage from the constraints provided from both satellite datasets, while accounting for all chemical interdependencies.

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Two inversion experiments are conducted. In inversion A, UVOC is directly emitted glyoxal, whereas in inversion B, UVOC forms glyoxal upon oxidation by OH with a molar yield of unity and a reaction rate of 2.86×10^{-12} molecules $^{-1}$ cm 3 s $^{-1}$, which is adjusted so that the global lifetime of UVOC is equal to five days. Sensitivity calculations with shorter and longer lifetimes for UVOC will also be addressed.

4.2 Results

Inferred annual mean glyoxal columns from both inversions are illustrated in Fig. 2. In both cases the inversion reproduces reasonably well the observations, although an underestimation persists due to the a priori constraints on the emissions. Inversion B, however, provides a better match with the observed patterns and magnitudes over the continents, as well as near the coasts over the tropical oceans, due to the secondary glyoxal production from UVOC inside the transported air masses. The columns inferred from inversion A (Fig. 3, green) remain too low over southeast Asia and miss the observed seasonality, whereas the deduced columns from inversion B allow for an excellent model/data agreement over all displayed regions. Overall, the globally averaged relative bias over land is decreased from -57% in the a priori, to -27% and -19% in scenarios A and B, respectively, whereas the spatiotemporal correlation coefficient increases from 0.5 in the a priori to 0.73 and 0.77 in cases A and B. The lowest biases (<-10%) and highest correlations (>0.8) are obtained over Tropical America in both inversions, and over Asia in inversion B.

The a priori and optimized annual global distribution of the UVOC source is displayed in Fig. 4. The a priori flux and the ratios of the optimized by the a priori fluxes for the different emission categories are illustrated in Figs. 5 and 6. The column enhancement of inversion A (Fig. 2) is achieved by strengthening the primary UVOC source to 36 Tg/yr, and by moderately increasing anthropogenic emissions as well as biogenic isoprene fluxes (Table 1). The optimized isoprene source amounts to 476 Tg/yr, in excellent agreement with a previous inversion constrained by HCHO columns (Stavrakou et al., 2009b). A similar increase in the isoprene source is deduced from the experiment B,

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whereas the secondary UVOC source is estimated at 56 Tg/yr. The optimized tropical UVOC source (40 Tg/yr) represents two thirds of the total UVOC source and is strongly enhanced with respect to its a priori value.

A decrease by about 20% is found for the pyrogenically emitted NMVOCs in both inversions. The inferred global terrestrial source amounts to 94 Tg/yr in inversion A and to 108 Tg/yr in inversion B. At the continental scale, the derived mean annual isoprene flux decrease by 25% over Northern America and the strong increase over southern Africa are similar to the inversion results using only HCHO columns (Stavrakou et al., 2009b). The a posteriori anthropogenic emissions are enhanced compared to the a priori, especially over the Far East, where the increase exceeds a factor of two in inversion A. Further, the pyrogenic emissions inferred over northern Africa are decreased by 50%, in qualitative agreement with previous studies (Myriokefalitakis et al., 2008).

The uncertainties associated to the aforementioned glyoxal source estimates are large, but difficult to quantify. They are mostly due to possible biases in the satellite retrievals, to uncertainties in the chemical mechanisms of the NMVOCs, and to the representation of the glyoxal sinks in the model.

Comparisons with surface glyoxal concentration measurements at continental midlatitude sites of the Northern Hemisphere are provided in Table 3 and Fig. 7. Whereas a close-to-zero average bias is calculated in the a priori simulation (without UVOC source), inversion A leads to a strong overestimation (+34 ppt), which would have been even larger if the settings of the inversion (e.g. a priori errors on the emission parameters) were chosen so as to allow for a better match with the satellite data. The computed bias after optimization B is very close to the a priori, whereas its slightly negative value is explained by the inferred decrease of the isoprene emissions over Northern America, as illustrated on Fig. 6. Further, the root mean square deviation in inversion B is by 16% lower than in the a priori. These comparisons suggest that the space-based glyoxal columns are compatible with the available in situ measurements when the extra glyoxal source is assumed to be secondary. However, in situ glyoxal measurements in tropical continental regions, certainly critical in order to evaluate the

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model findings, are currently missing from the literature.

The sensitivity tests C and D use the setup of inversion B. Aerosol uptake of glyoxal is ignored in inversion C and a higher uptake probability is assumed in inversion D (see Sect. 2.4). The results show that our conclusions regarding the magnitude of the UVOC 5 source and the agreement with the independent measurements of Table 3 do not depend significantly on the assumptions made for the uptake of glyoxal by aerosols. More precisely, when ignoring the SOA formation sink on aerosols, the inferred global glyoxal and UVOC flux amount to 95 Tg/yr and 45 Tg/yr, respectively, and the global glyoxal lifetime is by 16% longer (2.9 h) compared to the lifetime of inversion B. The agreement with the ground-based measurements is similar to that of inversion B, except for the mean average bias (Table 3) which becomes positive in this inversion (1.1 pptv) due to the longer residence time of glyoxal near the surface. When the higher value $\gamma = 3.7 \cdot 10^{-3}$ for the uptake coefficient is used, the estimated global annual UVOC and glyoxal source, and the glyoxal lifetime are guite similar to those of inversion B, namely, 57 Tq, 110 Tq and 2.6 h, respectively, whereas the predicted SOA sink on aerosols amounts to 13.2 Tg/yr. In this case, the average bias with the observations of Table 3 is calculated to be -2 pptv.

However, the results show a stronger sensitivity to the choice of the UVOC lifetime. Sensitivity tests where the UVOC lifetime ranges between one (inversion E) and 20 days (inversion F) give results qualitatively similar to inversion B, with inferred global terrsestrial glyoxal source estimated between 96–138 Tg/yr, and UVOC flux between 42 and 84 Tg/yr. These inversions underestimate the satellite data by 13–21%, whereas the agreement with in situ measurements worsens compared to the case B results. Inversions with UVOC lifetimes shorter than one day give results more similar to the inversion A, and lead to a significant overestimation of the surface glyoxal mixing ratio. The main results of the performed tests are summarized in Fig. 8.

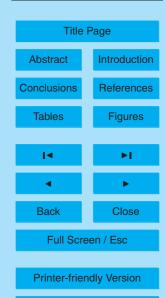
Both optimizations improve significantly the modelled HCHO column, bringing it much closer to the observations (Fig. 3). This is achieved mostly by decreasing the Indonesian biogenic fluxes by up to 40% and strongly increasing the isoprene emis-

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sions over southern Africa, whereas less significant changes are inferred over other regions (Stavrakou et al., 2009b).

5 Conclusions

The continental source of glyoxal has been constrained by glyoxal and formaldehyde columns retrieved from the SCIAMACHY sensor for 2005, based on a two-compound grid-based inversion scheme using the IMAGESv2 CTM and its adjoint. Formaldehyde provides a useful constraint on the emissions of isoprene, an important precursor of both glyoxal and formaldehyde. Fluxes of anthropogenic, pyrogenic, identified biogenic NMVOCs, and uncharacterized organic compounds (UVOC), initially distributed as the MODIS leaf area index, are determined by the inversion. Two possibilities have been explored. The UVOC is either direct glyoxal flux (case A) or an unknown glyoxal precursor with a lifetime of five days (case B).

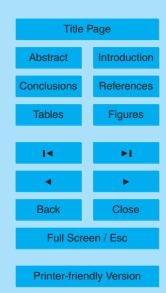
The continental glyoxal source (56 Tg/yr in the a priori), is estimated at 94 Tg/yr in inversion A, including a primary UVOC source of 36 Tg/yr, and at 108 Tg/yr in inversion B, including a secondary UVOC source of 54 Tg/yr. The isoprene source, major contributor to the global glyoxal source, exhibits only a slight increase with respect to its priori value on the global scale, but large deviations at the continental scale, in agreement with an inversion study constrained by formaldehyde columns only (Stavrakou et al., 2009b). Global anthropogenic fluxes are enhanced over the Far East, especially in inversion A, but stay close to their a priori values over other regions. Pyrogenic emissions are reduced by 25% on the global scale, the decrease being more important over the African continent. The emission updates enable significant reduction of the model/data discrepancies for the glyoxal and formaldehyde, demonstrating the capability to reconcile the two datasets in a single scheme. The global mean model/data bias, –57% in the a priori simulation, is reduced to –27% and to –19% in inversions A and B, respectively, whereas the globally calculated spatiotemporal correlation coeficient increases from 0.5 to 0.77 in inversion B.

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Validation using in situ glyoxal measurements in rural settings suggests that the missing glyoxal source is secondary (inversion B), since a large additional direct emission of glyoxal (inversion A) leads to a large overestimation of surface glyoxal concentrations. However, the lack of in situ measurements in the continental tropics makes difficult to evaluate the model results over these strongly emitting regions. Note also that errors in the satellite retrievals, the chemical degradation mechanisms, and the a priori spatiotemporal flux distribution may introduce significant uncertainties in the estimated sources. Further, the inferred glyoxal source depends on the choice of the UVOC lifetime and could range between 96 and 138 Tg/yr, according to sensitivity calculations, whereas it is found to be weakly sensitive to the choice of the uptake coefficient in the parameterization of SOA formation sink from aerosols. This sink, estimated between 10 Tg/yr and 13 Tg/yr, is found to be much higher than in a previous model study (Fu et al., 2008). The large relevance of SOA formed from aqueous aerosols has been also pointed out in Volkamer et al. (2009). More work is definitely needed in order to improve the representation of these processes in large scale models.

More glyoxal measurements, both space-based and in situ, would be a valuable contribution to our understanding of the emission mechanisms, the precursors of this compound, as well as the chemical mechanism controlling the local glyoxal concentrations.

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9, 13593-13628, 2009

Estimation of the glyoxal land source

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References

- Adams, P. J., Seinfeld, J. H., and Koch, D. M.: Global concentrations of tropospheric sulfate, nitrate, and ammonium aerosol simulated in a general circulation model, J. Geophys. Res., 104(D11), 13791–13823, 1999. 13601
- Andreae, M. O. and Merlet, P.: Emission of trace gases and aerosols from biomass burning, Global Biogeochem. Cy., 15, 955–966, 2001, and updates from Andreae, M. O., personal communication, 2007. 13598, 13601
 - Andres, R. J. and Kasgnoc, A. D.: A time-averaged inventory of subaerial volcanic sulfur emissions, J. Geophys. Res., 103, 25251–25261, 1998. 13601
 - Arey, J., Obermeyer, G., Aschmann, S. M., Chattopadhyay, S., Cusick, R. D., and Atkinson, R.: Dicarbonyl Products of the OH Radical-Initiated Reaction of a Series of Aromatic Hydrocarbons, Environ. Sci. Technol., 43(3), 683–689, 2009. 13599
 - Barkley, M. P., Palmer, P. I., Kuhn, U., Kesselmeier, J., Chance, K., Kurosu, T. P., Martin, R. V., Helmig, D., and Guenther, A.: Net ecosystem fluxes of isoprene over tropical South America inferred from Global Ozone Monitoring Experiment (GOME) observations of HCHO columns, J. Geophys. Res., 113, D20304, doi:10.1029/2008JD009863, 2008. 13595
 - Bellouin, N., Boucher, O., Haywood, J., et al.: Improved representation of aerosols for HadGEM2, Meteorological Office Hadley Centre, Technical Note 73, Met Office, Exeter, United Kingdom, 2007. 13601, 13602
- Bond, T. C., Streets, D. G., Yarber, K. F., Nelson, S. M., Woo, J.-H., and Klimont, Z.: A technology-based global inventory of black and organic carbon emissions from combustion, J. Geophys. Res., 109, D14203, doi:10.1029/2003JD003697, 2004. 13601
 - Bouwman, A. F., Lee, D. S., Asman, W. A. H., Dentener, F. J., van der Hoek, K. W., and Olivier, J. G. J.: A global high-resolution emission inventory for ammonia, Global Biogeochem. Cy., 11, 561–578, 1997. 13601
 - Bouwman, A. F., Boumans, L. J. M., and Batjes, N. H.: Estimation of global NH3 volatilization loss from synthetic fertilizers and animal manure applied to arable lands and grasslands, Global Biogeochem. Cy., 16(2), 1024, doi:10.1029/2000GB001389, 2002. 13601
 - Butler, T. M., Taraborrelli, D., Brühl, C., Fischer, H., Harder, H., Martinez, M., Williams, J., Lawrence, M. G., and Lelieveld, J.: Improved simulation of isoprene oxidation chemistry with the ECHAM5/MESSy chemistry-climate model: lessons from the GABRIEL airborne field campaign, Atmos. Chem. Phys., 8, 4529–4546, 2008,

ACPD

9, 13593–13628, 2009

Estimation of the glyoxal land source



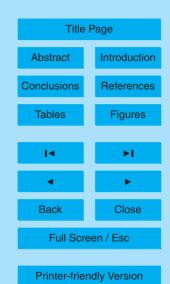


- http://www.atmos-chem-phys.net/8/4529/2008/. 13599
- Cantrell, C. A., Davidson, J. A., McDaniel, A. H., Shetter, R. E., and Calvert, J. G.: Temperature-dependent formaldehyde cross sections in the near-ultraviolet spectral region, J. Phys. Chem, 94, 3902–3908, 1990. 13602
- Compernolle, S., Ceulemans, K., and Müller, J.-F.: Influence of non-ideality on condensation to aerosol, Atmos. Chem. Phys., 9, 1325–1337, 2009, http://www.atmos-chem-phys.net/9/1325/2009/. 13602
 - Dentener, F., Kinne, S., Bond, T., Boucher, O., Cofala, J., Generoso, S., Ginoux, P., Gong, S., Hoelzemann, J. J., Ito, A., Marelli, L., Penner, J. E., Putaud, J.-P., Textor, C., Schulz, M., van der Werf, G. R., and Wilson, J.: Emissions of primary aerosol and precursor gases in the years 2000 and 1750 prescribed data-sets for AeroCom, Atmos. Chem. Phys., 6, 4321–4344, 2006, http://www.atmos-chem-phys.net/6/4321/2006/. 13601
 - De Smedt, I., Müller, J.-F., Stavrakou, T., van der A, R., Eskes, H., and Van Roozendael, M.: Twelve years of global observations of formaldehyde in the troposphere using GOME and SCIAMACHY sensors, Atmos. Chem. Phys., 8, 4947–4963, 2008, http://www.atmos-chem-phys.net/8/4947/2008/. 13602
 - Feierabend, K. J., L., Zhu, R. K., Talukdar, and J. B. Burkholder,: Rate Coefficients for the OH + HC(O)C(O)H (Glyoxal) Reaction between 210 and 390 K, J. Phys. Chem. A, 112(1), 73–82, 2008. 13600
- Fu, T.-M., D. J., Jacob, P. I., Palmer, K., Chance, Y. X., Wang, B., Barletta, D. R., Blake, J. C., Stanton, and M. J. Pilling,: Space-based formaldehyde measurements as constraints on volatile organic compound emissions in east and south Asia and implications for ozone, J. Geophys. Res., 112, D06312, doi:10.1029/2006JD007853, 2007. 13595
 - Fu, T.-M., D. J., Jacob, F., Wittrock, J. P., Burrows, M., Vrekoussis, and D. K. Henze,: Global budgets of atmospheric glyoxal and methylglyoxal, and implications for formation of secondary organic aerosols, J. Geophys. Res., 113, D15303, doi:10.1029/2007JD009505, 2008. 13595, 13599, 13603, 13610
 - Giglio, L.: (2007), Characterization of the tropical diurnal fire cycle using VIRS and MODIS observations, Remote Sens. Environ., 108, 407–421, 2007.
- Guenther, A., Hewitt, C. N., Erickson, D., Fall, R., Geron, C., Graedel, T., Harley, P., Klinger, L., Lerdau, M., McKay, W. A., Pierce, T., Scholes, B., Steinbrecher, R., Tallamraju, R., Taylor, J., and Zimmerman, J.: A global model of natural volatile organic compound emissions, J. Geophys. Res., 100, 8873–8892, 1995. 13598

9, 13593–13628, 2009

Estimation of the glyoxal land source

T. Stavrakou et al.



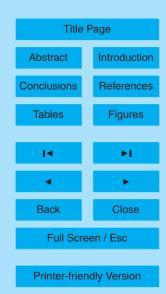


- Ieda, T., Kitamori, Y., Mochida, M., Hirata, R., Hirano, T., Inukai, K., Fujinuma, Y., and Kawamura, K.: Diurnal variations and vertical gradients of biogenic volatile and semi-volatile organic compounds at the Tomakamai larch forest station in Japan, Tellus B, 59, 177–186, 2006. 13620
- Ip, H. S. S., Huang, X. H. H., and Yu, J. Z.: Effective Henry's law constants of glyoxal, glyoxylic acid, and glycolic acid, Geophys. Res. Lett., 36, L01802, doi:10.1029/2008GL036212, 2009. 13600
 - Jacob, D. J., Field, B. D., Li, Q., Blake, D. R., de Gouw, J., Warneke, C., Hansel, A., Wisthaler, A., Singh, H. B., and Guenther, A.: Global budget of methanol: constraints from atmospheric observations, J. Geophys. Res., 110, D08303, doi:10.1029/2004JD005172, 2005, 13598
 - Kroll, J., Ng, N. L., Murphy, S. M., Varutbangkul, V., Flagan, R. C., and Seinfeld, J. H.: Chamber studies of secondary organic aerosol growth by reactive uptake of simple carbonyl compounds, J. Geophys. Res., 110, D23207, doi:10.1029/2005JD006004, 2005. 13600
 - Lee, Y.-N., Zhou, X., and Hallock, K.: Atmospheric carbonyl compounds at a rural southeastern United States site, J. Geophys. Res., 100(D12), 25 933–25 944, 1995. 13620
 - Lee, Y.-N., Zhou, X., Kleinman, L. I., et al.: Atmospheric chemistry and distribution of formaldehyde and several multioxygenated carbonyl compounds during the 1995 Nashville/Middle Tennessee Ozone Study, J. Geophys. Res., 103(D17), 22449–22462, 1998. 13620
 - Lelieveld, J., Butler, T. M., Crowley, J. N., Dillon, T. J., Fischer, H., Ganzeveld, L., Harder, H., Lawrence, M. G., Martinez, M., Taraborrelli, D., and Williams, J.: Atmospheric oxidation capacity sustained by a tropical forest, Nature, 452, 737–740, 2008. 13598
 - Liggio, J., Li, S.-M., and McLaren, R.: Reactive uptake of glyoxal by particulate matter, J. Geophys. Res., 110(D10304), doi:10.1029/2004JD005113, 2005. 13600
 - Martin, R. V., Jacob, D. J., Yantosca, R. M., Chin, M., and Ginoux, P.: Global and regional decreases in tropospheric oxidants from photochemical effects of aerosols, J. Geophys. Res., 108(D3), 4097, doi:10.1029/2002JD002622, 2003. 13602
 - Matsunaga, S., Mochida, M., and Kawamura, K.: Variation on the atmospheric concentrations of biogenic carbonyl compounds and their removal processes in the northern forest at Moshiri, Hokkaido Island in Japan, J. Geophys. Res., 109, D04302, doi:10.1029/2003JD004100, 2004. 13620
 - Metzger, S., Dentener, F., Pandis, S., and Lelieveld, J.: Gas/aerosol partitioning, 1, A computationally efficient model, J. Geophys. Res., 107(D16), 4312, doi:10.1029/2001JD001102, 2002. 13601

9, 13593-13628, 2009

Estimation of the glyoxal land source

T. Stavrakou et al.





- Middleton, P., Stockwell, W. R., and Carter, W. P.: Aggregation and analysis of volatile organic compound emissions for regional modeling, Atmos. Environ., 24A(5), 1107–1133, 1990. 13599
- Mircea, M., Facchini, M. C., Decesari, S., Cavalli, F., Emblico, L., Fuzzi, S., Vestin, A., Rissler, J., Swietlicki, E., Frank, G., Andreae, M. O., Maenhaut, W., Rudich, Y., and Artaxo, P.: Importance of the organic aerosol fraction for modeling aerosol hygroscopic growth and activation: a case study in the Amazon Basin, Atmos. Chem. Phys., 5, 3111–3126, 2005, http://www.atmos-chem-phys.net/5/3111/2005/. 13602
- Moortgat, G. K., Grossmann, D., Boddenberg, A., et al.: Hydrogen peroxide, organic peroxides and higher carbonyl compounds determined during BERLIOZ campaign, J. Atmos. Chem., 42, 443–463, 2002. 13620
- Munger, J., Jacob, D., Daube, B., Horowitz, L., Keene, W., and Heikes, B.: Formaldehyde, glyoxal, and methylglyoxal in air and cloudwater at a rural mountain site in central Virginia, J. Geophys. Res., 100(D5), 9325–9333, 1995. 13620
- Müller, J.-F. and Stavrakou, T.: Inversion of CO and NO_x emissions using the adjoint of the IMAGES model, Atmos. Chem. Phys., 5, 1157–1186, 2005, http://www.atmos-chem-phys.net/5/1157/2005/. 13595, 13597, 13598
 - Müller, J.-F., Stavrakou, T., Wallens, S., De Smedt, I., Van Roozendael, M., Potosnak, M. J., Rinne, J., Munger, B., Goldstein, A., and Guenther, A. B.: Global isoprene emissions estimated using MEGAN, ECMWF analyses and a detailed canopy environment model, Atmos. Chem. Phys., 8, 1329–1341, 2008, http://www.atmos-chem-phys.net/8/1329/2008/. 13597

20

- Myriokefalitakis, S., Vrekoussis, M., Tsigaridis, K., Wittrock, F., Richter, A., Brühl, C., Volkamer, R., Burrows, J. P., and Kanakidou, M.: The influence of natural and anthropogenic secondary sources on the glyoxal global distribution, Atmos. Chem. Phys., 8, 4965-4981, 2008, http://www.atmos-chem-phys.net/8/4965/2008/. 13595, 13603, 13604, 13607
- Nenes, A., Pilinis, C., and Pandis, S. N.: Isorropia: A new thermodynamic model for multiphase multicomponent inorganic aerosols, Aquat. Geochem., 4, 123–152, 1998. 13601
- Nichitiu, F., Drummond, J. R., Zou, J., and Deschambault, R.: Solar particle events seen by the MOPITT instrument, J. Atmos. Sol.-Terr. Phy., 66, 1797–1803, 2004. 13603
- Olivier, J. G. J., Berdowski, J. J. M., Peters, J. A. H. W., Bakker, J., Visschedijk, A. J. H., and Bloos, J.-P. J.: Applications of EDGAR. Including a description of EDGAR 3.0: reference database with trend data for 1970–1995, RIVM report no. 773301 001/NOP report

ACPD

9, 13593–13628, 2009

Estimation of the glyoxal land source





- no. 410200 051, RIVM, Bilthoven, 2001. 13598
- Olivier, J.G. J.: Part III: Greenhouse gas emissions. 1. Shares and trends in greenhouse gas emissions; 2. Sources and methods: greenhouse gas emissions for 1990 and 1995 in "CO₂ emissions from fuel combustion 1971–2000", 1–31, International Energy Agency, Paris, ISBN 92-64-09794-5, 2002. 13598
- Olivier, J.G. J., Peters, J., Granier, C., Pétron, G., Müller, J.-F., and Wallens, S.: Present and future surface emissions of atmospheric compounds, POET Report 2, EU project EVK2-1999-00011, 2003. 13598
- Olivier, J. G. J., Van Aardenne, J. A., Dentener, F., Ganzeveld, L., and Peters, J. A. H. W.: Recent trends in global greenhouse gas emissions: regional trends and spatial distribution of key sources, in: "Non-CO₂ Greenhouse Gases (NCGG-4)", coord.: van Amstel, A., 325–330, Millpress, Rotterdam, ISBN-90-5966-043-9, 2005. 13601
- Pang, Y., Thrpin, B. J., and Gundel, L. A.: On the importance of organic oxygen for understanding organic aerosol particles, Aerosol Sci. Technol., 40(2), 128–133, 2006. 13602
- Park, R. J., Jacob, D. J., Field, B. D., Yantosca, R. M., and Chin, M.: Natural and transboundary pollution influences on sulfate-nitrate-ammonium aerosols in the United States: Implications for policy, J. Geophys. Res., 109, D15204, doi:10.1029/2003JD004473, 2004 13601
 - Pham, M., Müller, J.-F., Brasseur, G. P., Granier, C., and Megie, G.: A three-dimensional study of the tropospheric sulfur cycle, J. Geophys. Res., 100(D12), 26 061–26 092, 1995. 13601
- Platt, U.: Differential Optical Absorption Spectroscopy (DOAS), in: Air Monitoring by Spectroscopic Techniques, edited by: Sigrist, M. W., John Wiley and Sons, Inc., New York, 1994. 13602
 - Pöschl, U., von Kuhlmann, R., Poisson, N., and Crutzen, P. J.: Development and intercomparison of condensed isoprene oxidation mechanisms for global atmospheric modeling, J. Atmos. Chem., 37, 29–52, 2000. 13598
 - Rissler, J., Vestin, A., Swietlicki, E., Fisch, G., Zhou, J., Artaxo, P., and Andreae, M. O.: Size distribution and hygroscopic properties of aerosol particles from dry-season biomass burning in Amazonia, Atmos. Chem. Phys., 6, 471–491, 2006, http://www.atmos-chem-phys.net/6/471/2006/. 13602
- Sander, S. P., Finlayson-Pitts, B. J., Friedl, R. R., Golden, D. M., Huie, R. E., Keller-Rudek, H., Kolb, C. E., Kurylo, M. J., Molina, M. J., Moortgat, G. K., Orkin, L. V., Ravishankara, A. R., and Wine, P. H.: Chemical Kinetics and Photochemical data for use in atmospheric studies, Evaluation number 15, NASA Panel for data evaluation, JPL Publication 06-2, Jet Propulsion

9, 13593–13628, 2009

Estimation of the glyoxal land source

T. Stavrakou et al.





Laboratory, Pasadena, 2006. 13599, 13601

20

- Saunders, S. M., Jenkin, M. E., Derwent, R. G., and Pilling, M. J.: Protocol for the development of the Master Chemical Mechanism, MCM v3 (Part A): tropospheric degradation of non-aromatic volatile organic compounds, Atmos. Chem. Phys., 3, 161–180, 2003, http://www.atmos-chem-phys.net/3/161/2003/. 13598
- Schultz, M. G., Backman, L., Balkanski, Y., et al.: REanalysis of the TROpospheric chemical composition over the past 40 years (RETRO): A long-term global modeling study of tropospheric chemistry, Jlich/Hamburg, Germany, 48/2007 report on Earth System Science of the Max Planck Institute for Meteorology, Hamburg, ISSN 1614-1199, available at: http://retro.enes.org, 2007 13598, 13599
- Spaulding, R. S., Schade, G. W., Goldstein, A. H., and Charles, M. J.: Characterization of secondary atmospheric photooxidation products: Evidence for biogenic and anthropogenic sources, J. Geophys. Res., 108(D8), 4247, doi:10.1029/2002JD002478, 2003. 13620
- Stavrakou, T. and Müller, J.-F.: Grid-based versus big region approach for inverting CO emissions using Measurement of Pollution in the Troposphere (MOPITT) data, J. Geophys. Res., 111, D15304, doi:10.1029/2005JD006896, 2006. 13595, 13605
- Stavrakou, T., Müller, J.-F., De Smedt, I., Van Roozendael, M., van der Werf, G. R., Giglio, L., and Guenther, A.: Evaluating the performance of pyrogenic and biogenic emission inventories against one decade of space-based formaldehyde columns, Atmos. Chem. Phys., 9, 1037–1060, 2009a, http://www.atmos-chem-phys.net/9/1037/2009/. 13595, 13597, 13598, 13602, 13604
- Stavrakou, T., Müller, J.-F., De Smedt, I., Van Roozendael, M., van der Werf, G. R., Giglio, L., and Guenther, A.: Global emissions of non-methane hydrocarbons deduced from SCIA-MACHY formaldehyde columns through 2003–2006, Atmos. Chem. Phys., 9, 3663–3679, 2009, http://www.atmos-chem-phys.net/9/3663/2009/. 13595, 13602, 13605, 13606, 13607, 13609
- Tie, X., Brasseur, G., Emmons, L., Horowitz, L., and Kinnison, D.: Effects of aerosols on tropospheric oxidants: A global model study, J. Geophys. Res., 106, 2931–2964, 2001. 13601
- van der Werf, G. R., Randerson, J. T., Giglio, L., Collatz, G. J., Kasibhatla, P. S., and Arellano Jr., A. F.: Interannual variability in global biomass burning emissions from 1997 to 2004, Atmos. Chem. Phys., 6, 3423–3441, 2006, http://www.atmos-chem-phys.net/6/3423/2006/. 13598, 13601

ACPD

9, 13593-13628, 2009

Estimation of the glyoxal land source





- Volkamer, R., Platt, U., and Wirtz, K.: Primary and Secondary Glyoxal Formation from Aromatics: Experimental Evidence for the Bicycloalkyl-Radical Pathway from Benzene, Toluene, and p-Xylene, J. Phys. Chem. A, 105, 7865–7874, 2001. 13599
- Volkamer, R., Spietz, P., Burrows, J. P., and Platt, U.: High-resolution absorption cross section of Glyoxal in the UV/vis and IR spectral ranges, J. Photochem. Photobio. A, 172, 35–46, 2005a. 13599, 13602
- Volkamer, R., Barnes, I., Platt, U., Molina, L. T., and Molina, M. J.: Remote Sensing of Gly-oxal by Differential Optical Absorption Spectroscopy (DOAS): Advancements in Simulation Chamber and Field Experiments, NATO Sciences Series, IV. Earth and Environmental Sciences, Vol. 62 "Environmental Simulation Chambers: Application to Atmospheric Chemical Processes", edited by: Barnes, I. and Rudzinski, K. J., Kluwer Academic Publishers, Dordrecht, The Netherlands. ISBN: 1-4020-4231-0, 2005b. 13599
- Volkamer, R., San Martini, F., Salcedo, D., Molina, L. T., Jimenez, J. L., and Molina, M. J.: A Missing Sink for Gas-Phase Glyoxal in Mexico City: Formation of Secondary Organic Aerosol, Geophys. Res. Lett., 34, L19807, doi:10.1029/2007GL030752, 2007. 13600, 13604
- Volkamer, R., Ziemann, P. J., and Molina, M. J.: Secondary Organic Aerosol Formation from Acetylene (C₂H₂): seed effect on SOA yields due to organic photochemistry in the aerosol aqueous phase, Atmos. Chem. Phys., 9, 1907-1928, 2009, http://www.stmas.chem.phys.net/0/1007/2000/, 13600, 13610
 - http://www.atmos-chem-phys.net/9/1907/2009/. 13600, 13610
- Vrekoussis, M., Wittrock, F., Richter, A., and Burrows, P. J.: Temporal and spatial variability of glyoxal as observed from space, Atmos. Chem. Phys. Discuss., 9, 8993-9042, 2009, http://www.atmos-chem-phys-discuss.net/9/8993/2009/. 13602
 - Wang, J., Hoffmann, A. A., Park, R. J., Jacob, D. J., and Martin, S. T.: Global distribution of solid and aqueous sulfate aerosols: Effect of the hysteresis of particle phase transitions, J. Geophys. Res., 113, D11206, doi:10.1029/2007JD009367, 2008. 13601
 - Wittrock F., Richter, A., Oetjen, H., Burrows, J. P., Kanakidou, M., Myriokefalitakis, S., Volkamer, R., Beirle, S., Platt, U., and Wagner, T.: Simultaneous global observations of glyoxal and formaldehyde from space, Geophys. Res. Lett., 33, L16804, doi:10.1029/2006GL026310, 2006. 13602

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Table 1. Global annual glyoxal source calculated with IMAGESv2 for 2005 in the a priori simulation and in inversions A and B. A priori and optimized flux estimates are expressed in Tg/yr.

Sources		A priori	Inversion A	Inversion B
Anthopogenic	primary	3.2	5	3.4
	secondary	11.9	13.3	12.1
Pyrogenic	primary	5.9	4.7	4.3
	secondary	3.9	2.9	2.6
Biogenic		31	32.5	32
UVOC	primary		35.6	
	secondary			54
Total		56	94	108
Sinks				
Photolysis		28.7	41.9	62.4
OH oxidation		10	14.1	18
SOA in clouds		4.7	9.3	10
SOA in aerosols		7.7	11.6	10.4
Dry deposition		2.7	12.1	3.9
Wet deposition		1.8	4.7	3.1
Lifetime (hr)		2.5	2.4	2.5
Burden (Gg)		15.8	25	32.3

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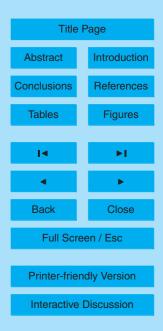




Table 2. Mean model/data biases and spatiotemporal correlation coefficients per region calculated for the a priori simulation, and after the inversions A and B.

Regions	A priori		Inversion A		Inversion B	
	Bias	r	Bias	r	Bias	r
N. America (30-70 N, 60-170 W)	-48.6	0.66	-21.4	0.65	-19.5	0.67
Tropical America (30 S-20 N, 30-80 W)	-63.7	0.43	-3.2	0.86	-2.5	0.84
Europe (35–65 N, 10 W–55 E)	-62.8	0.00	-32.2	0.05	-7.6	0.12
Africa (40 S-30 N, 20 W-45 E)	-54.2	0.47	-31.0	0.66	-26.2	0.70
Asia (10 S-55 N, 60-160 E)	-54.8	0.62	-23.8	0.74	-9.7	0.85
Oceania (10-40 S, 110-170 E)	-17.3	0.57	-6.3	0.75	-5.6	0.72
Globe	-56.9	0.50	-26.7	0.73	-18.7	0.77

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Table 3. Comparison between in situ measurements of glyoxal concentrations, a priori concentrations, and concentrations inferred by the inversions. Numbers are in ppt. RMSD is the root mean square deviation. References: [1] Lee et al. (1995), [2] Lee et al. (1998), [3] Munger et al. (1995), [4] Spaulding et al. (2003), [5] leda et al. (2006), [6] Matsunaga et al. (2004), [7] Moortgat et al. (2002).

Location	Time	Observ.	A priori	Α	В
Georgia (32.5N, 82W) [1]	Jul-Aug 1991	18.0	65.1	132.6	55.4
Georgia (32.5N, 82W) [1]	Jun 1992	83.0	48.8	132.6	45.9
Nashville (37N, 87W) [2]	Jul 1995	67.0	89.0	64.7	58.5
Nashville (37N, 87W) [2]	Jul 1995	26.0	31.9	13.8	17.9
Pinnacles (39N, 78W) [3]	Sept 1990	44.0	48.9	138.1	51.0
Blodgett Forest (39N, 120W) [4]	Aug-Sep 2000	27.0	25.4	59.4	31.1
Tomakomai (42.7N, 141.6E) [5]	Sep 2003	26.0	7.4	45.3	19.3
Uryu, Japan (44.3N, 142.2E) [6]	Aug 2002	18.0	13.2	70.8	27.2
Pabsthum (52.5N, 12.5E) [7]	Aug 1998	38.0	18.8	67.6	26.3
Average bias (pptv)			0.14	34.3	-1.3
RMSD			6.2	15.7	5.2

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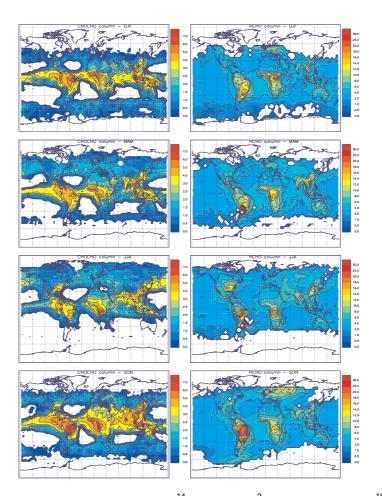
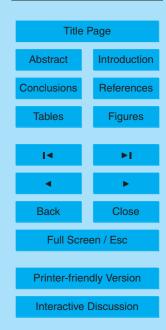


Fig. 1. Averages of the observed CHOCHO (left, in 10¹⁴ molecules cm⁻²) and HCHO (right, in 10¹⁵ molecules cm⁻²) vertical columns for winter (DJF), spring (MAM), summer (JJA) and autumn (SON) 2005. Blank regions correspond to an absence of valid data due to large zenith angles, south Atlantic Anomaly, or missing cloud information.

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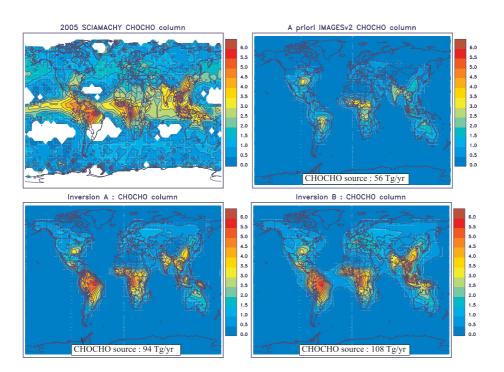
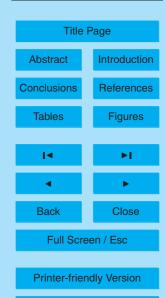


Fig. 2. Annually averaged CHOCHO vertical columns observed (top left), simulated with IM-AGESv2 (top right), and deduced from inversion (bottom panels). The global annual CHOCHO source for each case is given inset. Units are 10¹⁴ molecules cm⁻².

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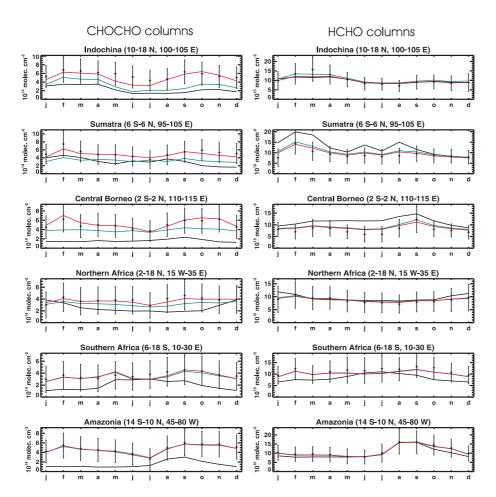


Fig. 3. Monthly mean CHOCHO (left) and HCHO (right) column abundancies simulated with IMAGESv2 and observed by SCIAMACHY (black dots) in 2005. Black solid lines correspond to the a priori simulation, green and red to inversions A and B, respectively. CHOCHO columns are expressed in 10¹⁴ molecules cm⁻² and HCHO columns in 10¹⁵ molecules cm⁻².

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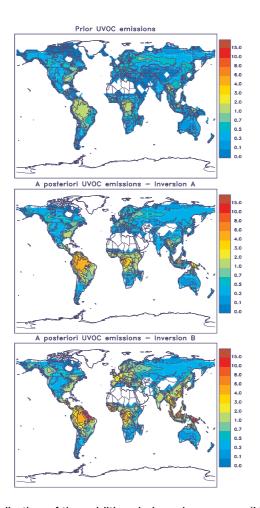
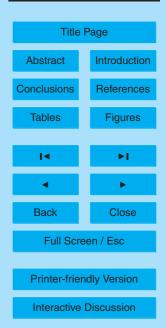


Fig. 4. Global annual distribution of the additional glyoxal precursor (UVOC) source inferred by the inversions. Units are 10¹⁰ molecules cm⁻² s⁻¹.

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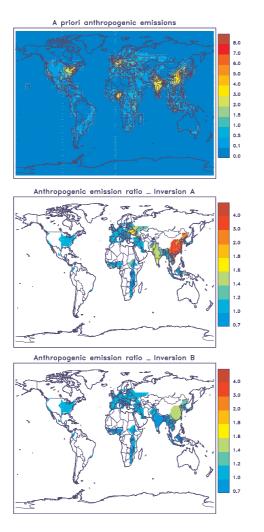


Fig. 5. Global annual a priori anthropogenic flux of glyoxal precursors, and the ratio of the optimized by the a priori estimate inferred by the inversion studies.

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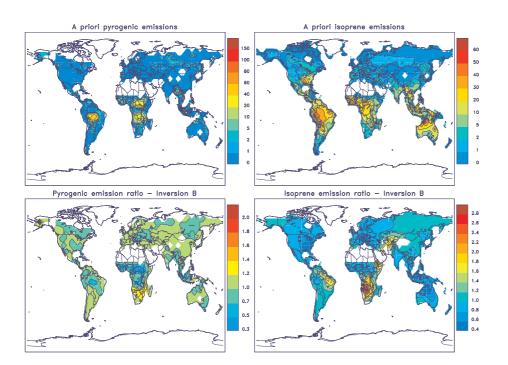
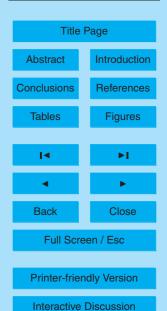


Fig. 6. Global annual a priori pyrogenic NMVOC and biogenic isoprene emissions (upper panel) and the ratios of the optimized by the a priori fluxes deduced by the inversion B (lower panel).

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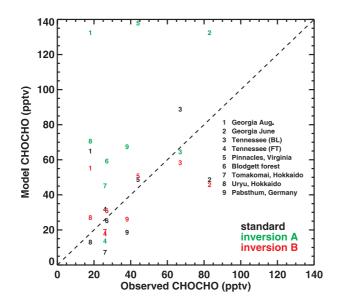


Fig. 7. In situ measurements of glyoxal vs. a priori (in black) and optimized (in green for inversion A and red for inversion B) glyoxal concentrations (see Table 3).

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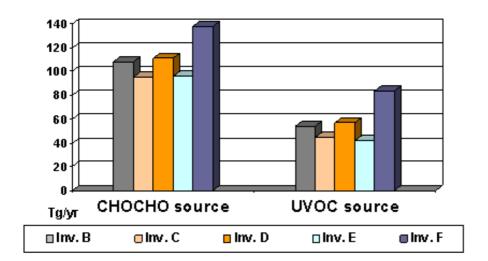


Fig. 8. Global glyoxal source and UVOC source inferred from the performed sensitivity inversions: B (standard with UVOC lifetime equal to five days and uptake coefficent equal to 2.9·10⁻³), C (as B, but assuming no aerosol uptake of glyoxal), D (as B, but with uptake coefficient equal to 3.7·10⁻³), E (as B, with UVOC lifetime of one day), F (as B, with UVOC lifetime of 20 days).

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