

***Interactive comment on “Secondary organic aerosol formation from toluene photooxidation under various NO<sub>x</sub> conditions and particle acidity” by G. Cao and M. Jang***

**G. Cao and M. Jang**

Received and published: 30 October 2008

We would like to thank the reviewer for the valuable comments.

Response to reviewer 2:

Comment 1. The reviewer questions the assumption that SOA density remains 1.4 g/ml under all experimental conditions in this study, because the compositions of SOA in different conditions are different.

**Response:** In this study, we didn't measure the SOA density experimentally. We agree that SOA densities depend on SOA compositions and may vary under different experimental conditions. However, the current limited knowledge of SOA compositions

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

Discussion Paper

hinders the accurate estimation of SOA densities. The findings of toluene SOA densities at different NO<sub>x</sub> concentrations are not very consistent. For example, Sato et al. (2007) reported that the density of SOA remained 1.4 g/ml at different initial NO<sub>x</sub> concentrations (0.2 ppm and 1 ppm). But Ng et al. (2007) found that the toluene SOA density was 1.45 g/ml at low NO<sub>x</sub> level and 1.24 g/ml at high NO<sub>x</sub> level. If 1.24 g/ml is used as the SOA density for the high NO<sub>x</sub> conditions, our assumption in this study (SOA density = 1.4 g/ml) probably overestimates the SOA yields at high NO<sub>x</sub> levels by 0.01%-0.02%, which can not significantly change our interpretation and conclusions. Therefore, we would like to remain our assumption for the toluene SOA density.

Comment 2. The reviewer comments on the Figure 5C that besides mechanisms addressed in this manuscript, another possibility proposed by Hurley et al. (2001) may explain the curve shape of aerosol growth profile.

**Response:** We thank the reviewer for providing the valuable information regarding the Fig. 5C. The additional sentences have been included at the end of section 3.2 (line 3 on Page 14481) of the revised manuscript.

"Hurley et al. (2001) who simulated the toluene SOA formation at high NO<sub>x</sub> levels using a chemical kinetic model have seen the similar SOA growth curves as observed in this study, which indicates another possible mechanism for toluene SOA formation: not only first-generation but also second-generation oxygenated products can contribute the toluene SOA formation. However, this mechanism may not very significant in our system with high initial toluene concentrations where first-generation products are dominant."

---

Interactive comment on Atmos. Chem. Phys. Discuss., 8, 14467, 2008.

[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)[Discussion Paper](#)