

***Interactive comment on* “Measurement-based modeling of bromine chemistry at the Dead Seaboundary layer – Part 2: The influence of NO<sub>2</sub> on bromine chemistry at mid-latitude areas” by E. Tas et al.**

**E. Tas et al.**

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We are very grateful to the reviewer for the level of attention given to our paper, and for the comments that are very helpful in improving the paper

**Scientific comments:**

We have edited the new version of the manuscript according to the reviewers comments. We refer here to all cases in which further clarification or details are needed.

- According the reviewers comment we will add the following as the first sentence in the experimental section (Sec. 2.1):

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The model used for the present study is essentially the same as used for a companion paper (Tas et al., 2006).

In addition to that, according to the reviewers suggestion, the assumptions underlying the treatment of the heterogeneous reactions H1 and H2 including the used surface area of aerosols and the uptake coefficients will be added to the revised version.

- We agree with the referee that the discussion about the significance of reactions H3 and H4, under the condition of the present study, should be based on their rates compared to other reactions that were included in the mechanism. Therefore, the following will be included (replacing lines 2-8, page 7741):

In order to investigate the effect of these reactions on the analysis described in the present paper, their potential influence on BrOX levels should be evaluated, based on the NO<sub>2</sub> fluxes used in the present study,. For this purpose the rate of reactions H3 and H4 should be compared with these of the rate limiting step reactions of cycles 3a, 3b, 4 and 5. The rate of reaction H2 is the slowest of all limiting rate reactions of these cycles. The rate of reaction H2 was compared to the calculated rates of reactions H3 and H4, assuming uptake coefficients ( $\gamma$  for reactions H3 and H4 of 0.16 (Seisel et al., 1997) and 0.023, respectively. The calculations show that the daily average rates (06:00 and 20:00 LT) of reactions H3 and H4 are lower than the rates of reaction H2, by at least 1.5 and 3.5 orders of magnitude, respectively, for all of the simulations described in the present study. Thus, it appears that the influence of reactions H3 and H4 on BrOx production can be neglected, under the conditions used for the present study. However, it should be remarked that reactions H3 and H4 are expected to have higher influence on BrOx production and NO<sub>2</sub> lost during nighttime, and also in cases that the available salt surface area is large enough for significant surface reactions.

- It may appear that a comparison between the periodic two stage cyclic process described in the present paper and the cyclic behavior of ozone-RBS-NO<sub>x</sub> as discussed by Evans et al., 2003, is beyond the scope of the present study. However, we think

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that such a discussion may be both interesting and helpful for a better understanding of RHS activity in other areas, both in mid-latitude and Polar Regions. Thus, we have decided to include the following:

It should be remarked that the relationship between  $\text{BrO}_x$ , ozone and  $\text{NO}_x$  during ozone depletion events, as described by the two stages periodic process, are not limited only to the Dead Sea area. A similar relationship between these species, during ozone depletion events at the Arctic boundary layer, was discovered by Evans et al. 2003 and was defined as An Oscillatory System in Bromine-Catalyzed  $\text{O}_3$  depletion. Because of the relatively small area of the evaporation ponds, and the high intensity of RBS activity, the time period and the magnitude of these oscillations at this area are expected to be small, in comparison with other areas. Thus, Evans et al. 2003 reported a much larger time period (3 days) for the oscillations at the Arctic, compared to the 1 to 3 hours time period of the oscillations described in the present paper. The time period in the first case was long enough, so that a photochemical recovery of ozone occurred. Advection of ozone is expected to be the main cause for ozone recovery in cases that the time period is short enough, as was evident at the Dead Sea evaporation ponds. It is also could be that the recovery of ozone by advection may be limited only to the more outer parts of the area at which RHS activity occur. This may be the case, when the area at which RHS activity occurs is big enough, that the traveling time of fresh ozone from outside of this area is high enough.

#### Literature cited:

-A reference to the paper by Fan and Jacob (1992) was added.

-The original references for the work carried out in Polar Regions, is replaced by reference of Simpson et al. (2007).

**Definition:** RHS and RBS will be defined in the new manuscript (at the beginning of sect. 1).

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An information about the location, relevant features and the potential influence of the Dead Sea Works on RBS activity, will be included in the section describes the model simulations (currently Sect. 2.1). . Page 7731, lines 21 to 25: We have rephrased the original sentence by dividing it into two sentences as follows:

All of the results and discussion presented in Sect. 3.1 are directly based on the FULL simulation. The results and discussion presented in Sect. 3.2 refer to a set of simulations based on the FULL simulation and on the NOB simulation, changing only the NO<sub>2</sub> fluxes, as described in more detail in Sect. 3.2.

Page 7736, line 13:

The average values of ozone and BrO<sub>x</sub>, as well as all other species that were discussed in this section, are taken only over the time of RBS activity (i.e., between 6:00 and 20:00 LT). Since this was not clear, the following will replace the original sentence:

In this section the investigation is limited only to the influence of NO<sub>2</sub> mixing ratios on the average mixing ratios of BrO, Br, BrO<sub>x</sub> and O<sub>3</sub> obtained during daytime RBS activity (6:00-20:00 LT). Its influence on the diurnal profile of these species is ignored .

Page 7736, line 23: We have changed the original two sentences (page 7736,lines 20-24) in a way that is less redundant, but still emphasizes that fact that for each simulation only a single multiplication factor, n, was used:

For each of these 11 simulations, the value of the NO<sub>2</sub> flux obtained in the FULL scenario, was multiplied by a single multiplication factor, n, at each update time (every 15 min). For each simulation, a different value for n was used, varying from 0.1 to 100. .

- Page 7739, line 24: The sentence was rephrased according the reviewers comment:

For example, several studies (e.g., Beine et al., 1997; Sander et al., 1999; Ridley and Orlando., 2003) have showed that an increase in the concentrations of NO<sub>x</sub> impedes the destruction of ozone due to an increase in the concentrations of bromine-containing nitrogen reservoirs.

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Page 7741, line 12:

According to the referee comment, the sentence will be rephrased:

The influence of NO<sub>2</sub> on both the diurnal profiles and daily average production of Br and BrO was investigated in this study, based on measurements performed at the Dead Sea evaporation ponds.

### Figures and Tables:

Table 1: The title will be changed according to the referee suggestion: Average mixing ratios of various species obtained for different NO<sub>2</sub> flux values over the time period 8230;.

We apologize for the mistake in the unit of the average flux of NO<sub>2</sub> during FULL simulation. The value was checked again and the formats of the number and units were changed as follows: The average NO<sub>2</sub> flux used for FULL simulation is 3.61 × 10<sup>13</sup> molec·cm<sup>-2</sup>·s<sup>-1</sup> during RBS activity (06:008211;20:00 LT). .

Figure 3 :caption: A description of the link between reactions G2, G7 and G9 will be given in the legend of Fig. 3. In addition to that, the caption of Fig. 3 was edited to include the connection between reactions G2, G7 and G9.

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Fig.4: caption: last sentence:

The referee advice is accepted.

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