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Interactive Comment

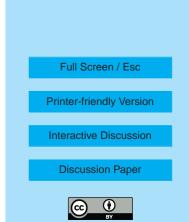
Interactive comment on "Relating CCN activity, volatility, and droplet growth kinetics of β -caryophyllene secondary organic aerosol" by A. Asa-Awuku et al.

Anonymous Referee #1

Received and published: 8 July 2008

The authors present the results from a short series of experiments examining the CCN activity of secondary organic aerosol derived from the oxidation of b-caryophyllene. Their observations provide a detailed and important characterization of the potential for sesquiterpene SOA to contribute to CCN activity.

The manuscript is generally well-written, and the introduction does a good job of discussing the key concepts in the field and extensively cover the existing literature. That said, the paper could be significantly improved if the authors would carefully lay out in the first two sections a more detailed account of the five experiments included in the study, and the motivations for choosing which instruments to include for each exper-



iment. The use of a summary table (i.e., Table 1) is a practical one, but in this case many key details are missing and the logic behind the progression of experiments seems somewhat difficult to follow.

One significant concern is the lack of repetition for the experiments used to support the key findings for the study. Only two experiments are included for which data from both CCN counters are available, and only one experiment included the use of the thermal denuder. The authors use these limited results as the sole basis for the first of their "major findings". Their argument is supported by the data, but the limited amount of information leaves available alternative explanations, including something as unfortunate as an undetected instrument malfunction. It seems odd that such a large impact of semivolatile SOA on CCN activity would exist for b-caryophyllene where (as the authors indicate) there is no observed difference between the instruments for monoterpene SOA. The differences between the CCN counters when measuring the effect of the oxidation pathway is not sufficiently explained, again suggesting the possibility of instrumental variability. One possible check on the consistency of the SD CCN counter would be to compare the results of experiment 1 & 4, which were carried out at very similar conditions; the authors presumably have this data and should include it in the manuscript. Ideally, more replicates of the experiment would be added to the study, but the reviewer acknowledges that this is probably not feasible.

Despite some skepticism that the presence of semivolatile hygroscopic material fully explains the observed discrepancy between the two instruments, the results of experiment 2 with the thermal denuder are quite exciting. Clearly the hygroscopic component is more likely to evaporate under dry conditions. This raises an interesting question-would this same material be continue to dominate the overall volatility under the humid conditions of the CCN counters and the natural environment? Within the CFSTGC CCN counter, the aerosol sample would be humidified and probably deliquesced before significant heating occurs; this is substantially different from the conditions within the thermal denuder.

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The remaining analyses are well developed, including the major finding that the less volatile material remaining after thermal denudation is connected to slower growth kinetics. In light of the limited nature of the available data, it would be preferable if the concluding statements were more nuanced, but the authors do not radically overreach in this regard.

In addition to the above major comments, there are the following minor comments:

Page 10111, Line 25: What fraction of the SOA volatilizes in the thermodenuder?

Page 10112, Lines 4-26: What are the measurement uncertainties for the SD CCN counter, and for the CFSTGC CCN counter operating in various modes? The SMCA technique would seem to have particularly high measurement uncertainties under atmospherically relevant aerosol concentrations. Note that we are not told the actual particle concentrations for the experiments described in this study.

Page 10115, Lines 8-10: The density assumption is critical to the subsequent soluble fraction calculation. How sensitive are these subsequent calculations to the choice of density?

Page 10119, Lines 1-3: A minor point, but to my eye the differences between Figs. 2 & 3 seem to be in slope rather than shape (i.e., Fig. 3 looks sigmoidal only because the transition occurs more rapidly than the Fig. 2 data).

Page 10127, Line 5: I would hesitate to extend the observed results to sesquiterpenes generally.

Interactive comment on Atmos. Chem. Phys. Discuss., 8, 10105, 2008.

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