Atmos. Chem. Phys. Discuss., 8, S3180–S3181, 2008 www.atmos-chem-phys-discuss.net/8/S3180/2008/ © Author(s) 2008. This work is distributed under the Creative Commons Attribute 3.0 License.



## **ACPD**

8, S3180-S3181, 2008

Interactive Comment

## Interactive comment on "1-D air-snowpack modeling of atmospheric nitrous acid at South Pole during ANTCI 2003" by Wei Liao and D. Tan

## P. Anderson

philip.s.anderson@bas.ac.uk

Received and published: 29 May 2008

I draw the attention to the subject of "model selection", with reference to Equation (3) and the comment at the bottom of the page on 9734: levels at 30 cm > levels at 10 cm, depite hardly any actinic flux.

Using the simplest diffusion model will explain this, assuming 1. local near equilibrium of chemsitry and therefore C proportional upon actinic flux 2. steady state profiles 3. true diffusion term (term 1 on RHS eq 3)

d/dz(D.dC/dz) that is, the gadient in diffusivity, D, is incuded in the model. e.g. if actinic derived concentration is given as

Aexp(k.z)

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

**Discussion Paper** 



with k = 1/0.2 (typical from figure 3) z depth (negative) and diffusivity, D is given as

 $D = 5e-4 * exp(k_D.z)$ 

with  $k_D = 1/0.05$  (which also looks reasonable, and describes the enhanced diffusivity near the surface due to ventilation), then solving

dC/dt = Aexp(k.z) + d/dz(D.dC/dz)

with boundary conditions of C = 0 at z = 0 and -infinity produces a peak in C at a depth of 30 cm.

Parsimony would imply that the data presented cannot therefore discriminate between the chemistry and the diffusion.

Interactive comment on Atmos. Chem. Phys. Discuss., 8, 9731, 2008.

## **ACPD**

8, S3180-S3181, 2008

Interactive Comment

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

**Discussion Paper** 

