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Interactive comment on “Multiphase modeling of nitrate photochemistry in the quasi-liquid layer (QLL): implications for NO_x release from the Arctic and coastal Antarctic snowpack” by C. S. Boxe and A. Saiz-Lopez

C. S. Boxe and A. Saiz-Lopez

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Author Response to Anonymous Referee #1 comments to “Multiphase Modeling of nitrate photochemistry in the quasi-liquid layer (QLL): implications for NO_x release from the Arctic and coastal Antarctic snowpack,” by C. S. Boxe and A. Saiz-Lopez.

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1. We appreciate Anonymous Referee #1's comments on our paper.
2. As shown on pages 6016–6017 of the manuscript, using recently obtained laboratory

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data {Grannas et al. (2007); Takenaka et al. (1996)}, we reformulated the model parameterizations, which do not affect the model output simulations since photochemistry at the top-most layers (top 14 m) governs the release of NO and NO₂. Therefore, calculation of kmix depends on the enhanced rate constants recently obtained from the lab, the boundary layer height, and an estimated first-order rate constant (equation 11, page 6017).

3. As suggested by Anonymous Referee #1, the minor comments will be corrected in the revised manuscript.

References Grannas, A. M., Bausch, A. R., and Mahanna, K. M.: Enhanced aqueous photochemical reaction rates after freezing, *J. Phys. Chem. A*, 111, 11043-111049, 2007. Takenaka, N., Ueda, A., Daimon, T., Bandow, H., Dohmaru, T., and Maeda, Y.: Acceleration mechanism of chemical reaction by freezing: the reaction of nitrous acid with dissolved oxygen, *J. Phys. Chem.*, 100, 13874-13884, 1996.

Interactive comment on *Atmos. Chem. Phys. Discuss.*, 8, 6009, 2008.

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