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## Aerosol hygroscopicity in the marine atmosphere: a closure study using high-resolution, size-resolved AMS and multiple-RH DASH-SP data

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#### Abstract

We have conducted the first closure study to couple high-resolution aerosol mass spectrometer (AMS) composition data with size-resolved, multiple-RH, high-time-resolution hygroscopic growth factor (*GF*) measurements from the differential aerosol sizing and hygroscopicity spectrometer probe (DASH-SP). These data were collected off the coast of Central California during seven of the 16 flights carried out during the MASE-II field campaign in July 2007. Two of the seven flights were conducted in airmasses that originated over the continental United States. These flights exhibited elevated organic volume fractions (*VF*<sub>organic</sub>=0.46±0.22, as opposed to 0.24±0.18 for all other flights),
corresponding to significantly suppressed *GFs* at high RH (1.61±0.14 at 92% RH, as compared with 1.91±0.07 for all other flights), more moderate *GF* suppression at intermediate RH (1.53±0.10 at 85%, compared with 1.58±0.08 for all other flights, and no measurable *GF* suppression at low RH (1.31±0.06 at 74%, compared with 1.31±0.07 for all other flights). Organic loadings were slightly elevated in above-cloud aerosols,

- as compared with below-cloud aerosols, and corresponded to a similar trend of significantly suppressed *GF* at high RH, but more moderate impacts at lower values of RH. A hygroscopic closure based on a volume-weighted mixing rule provided excellent agreement with DASH-SP measurements ( $R^2$ =0.79). Minimization of root mean square error between observations and predictions indicated mission-averaged organic
- <sup>20</sup> *GFs* of 1.20, 1.43, and 1.46 at 74, 85, and 92% RH, respectively. These values agree with previously reported values for water-soluble organics such as dicarboxylic and multifunctional acids, and correspond to a highly oxidized, presumably water-soluble, organic fraction (O:C=0.92±0.33). Finally, a backward stepwise linear regression revealed that, other than RH, the most important predictor for *GF* is *VF*<sub>organic</sub>, indicating
- <sup>25</sup> that a simple emperical model relating *GF*, RH, and the relative abundance of organic material can provide accurate predictions of hygroscopic growth in the marine atmosphere.

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#### 1 Introduction

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Atmospheric aerosols change size with fluctuations in relative humidity, with a magnitude dictated by chemical composition. Because this hygroscopic response determines particle size, it influences direct climate forcing attributed to aerosols. Further, subsaturated hygroscopic growth factor ( $D_{p,wet}/D_{p,dry}$ ) is strongly correlated with CCN activity (Prenni et al., 2001). Given the importance of aerosol water uptake on both the direct

- and indirect light scattering properties of aerosols, incomplete understanding of aerosol hygroscopicity has been identified as a major limitation in estimations of climate forcing (IPCC, 2007).
- <sup>10</sup> With a firm foundational understanding of hygroscopic properties of inorganic aerosol constituents, there has been a significant shift in focus toward organic hygroscopicity in the last decade, as a number of theoretical (e.g., Clegg and Seinfeld, 2006; Topping et al., 2005a,b), laboratory (Peng et al., 2001; Choi and Chan, 2002a; Peng and Chan, 2001; Choi and Chan, 2002b; Sjogren et al., 2007; Virkkula et al., 1999; Cruz and Pan-
- <sup>15</sup> dis, 2000; Prenni et al., 2001; Cocker et al., 2001a,b; Hameri et al., 2002; Saathoff et al., 2003; Prenni et al., 2003; Petters et al., 2006; Varutbangkul et al., 2006; Sjogren et al., 2007; Prenni et al., 2007; Moore and Raymond, 2008; Rood et al., 1985; Carrico et al., 1998, 2000; Dougle et al., 1998; Magi and Hobbs, 2003; Kim et al., 2006; Kreisberg et al., 2001; Hegg et al., 2006; Massling et al., 2007), and chamber (e.g., Ansari
- and Pandis, 2000; Cocker et al., 2001a,b; Duplissy et al., 2008) studies have sought to address how the presence of organics affects the water uptake characteristics of atmospheric aerosol. Despite advances in understanding hygroscopic characteristics of organic-containing aerosols, measurements of aerosol hygroscopicity in field campaigns have remained relatively sparse.
- <sup>25</sup> Closure studies, which attempt to reconcile simultaneously measured hygroscopic and chemical data, link laboratory studies of hygroscopicity, theoretical models for water uptake, and field measurements of aerosol-water interactions. The standard method for predicting hygroscopic growth from composition data is based on volume-

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weighted water uptake by the individual chemical constituents. While it is usually possible to predict water uptake for the inorganic fraction of atmospheric aerosols, the wealth of organic species in the atmosphere, combined with limited understanding of organic aerosol hygroscopicity, has led investigators to assign the water uptake of organics as the particle growth not explained by inorganic constituents (e.g., Malm et al., 2005).

An important approach in hygroscopicity closure is to represent organics as a combination of water-soluble and insoluble fractions, corresponding to oxygenated (OOA) and hydrocarbon-like (HOA) carbon, respectively. One notable study in an urban atmosphere (Gysel et al., 2007) used high-resolution chemical data from the aerosol mass

- <sup>10</sup> spectrometer (AMS, Jayne et al., 2000) to separate the organic component into OOA and HOA by using m/z 44 and 57 signals, respectively, with a deconvolution method developed by Zhang et al. (2005). In the study, division of organics between hygroscopically active and inactive fractions provided good agreement between predicted and measured hygroscopicity values. This study represents an improvement in the hygroscopic treatment of organics over prior work, but applies only to urban aerosol and is limited by both low time resolution and single-RH conditions inherent in HTDMA
- and is limited by both low time resolution and single-RH conditions inherent in HTDMA systems. The current study presents data obtained during seven flights in the marine

atmosphere off the coast of Central California during seven hights in the marine atmosphere off the coast of Central California during the second Marine Stratus/Stratocumulus Experiment (MASE-II). The dataset is the first to combine hightime-resolution, size-resolved AMS chemistry with high-time-resolution, size-resolved hygroscopic data at multiple RH values from the Differential Aerosol Sizing and Hygroscopicity Spectrometer Probe (DASH-SP). Hygroscopic studies have previously been carried out in the marine atmosphere (see Table 1), but none with chemical and hygroscopic data as highly size. time. and PH resolved as these prevented here.

 $_{\rm 25}$   $\,$  scopic data as highly size-, time-, and RH-resolved as those presented here.

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#### 2 Experimental

#### 2.1 MASE-II experiment

The data presented here were obtained during a series of seven cloud probing flights carried out as part of the second Marine Stratus/Stratocumulus Experiment (MASE-II) <sup>5</sup> field campaign during July 2007. The MASE II experiment was the second of two airborne field campaigns directed toward measurement of aerosol-cloud relationships in marine stratocumulus in the eastern Pacific Ocean. The Marine Stratus/Stratocumulus Experiment (MASE) was carried out in 2005 off the coast of Monterey, California (Lu et al., 2007), and MASE II was undertaken in 2007 in the same region. Both experiments were carried out in the month of July, when marine stratocumulus are prevalent over the region, and utilized the CIRPAS Twin Otter aircraft. In each campaign, comprehensive airborne measurements were made of aerosol and cloud properties

in areas both perturbed and unperturbed by local anthropogenic emissions. Tables 2 and 3 list the flights carried out during MASE II and the instrument payload onboard
 the Twin Otter, respectively. The present study addresses measurements of the hygroscopic properties of marine aerosols during MASE II. Other flights probed emissions from a large bovine source and a large container ship, and these data are presented elsewhere (Sorooshian et al., in preparation; Murphy et al., in preparation).

#### 2.2 Aerosol composition measurements

<sup>20</sup> Non-refractory aerosol chemical species were characterized by the Aerodyne compact Time of Flight Aerosol Mass Spectrometer (cToF-AMS, Drewnick et al., 2004a,b). In the AMS, particles with vacuum aerodynamic diameters  $(D_{p,va})$  50 nm $\leq D_{p,va} \leq$ 800 nm are focused by an aerodynamic lens, pass through a 3.5% chopper, and are vaporized at 500°C. The chopper has three modes of operation, which detect background mass spectra, ensemble average mass spectra over all particle sizes, or size-resolved mass spectra. After vaporization, all molecules are ionized via electron impact, and undergo

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time-of-flight mass analysis. In addition to sulfate, nitrate, and ammonium, the AMS detects total organic mass as a well-established sum of mass spectra. AMS data allows partitioning of organic species into oxygenated (OOA) and hydrocarbon-like (HOA) carbon with the use of m/z 44 and 57 markers, respectively, and a deconvolution method described by Zhang et al. (2005). Kondo et al. (2007) found that in aerosols collected in Tokyo, OOA and HOA correlated strongly with water-soluble organic carbon (WSOC) and water-insoluble organic carbon (WIOC), respectively, with nearly all WSOC being OOA and nearly all HOA being WIOC. These methods allow estimation of the fraction of aerosol comprised of WSOC and WIOC. Finally, O:C ratios are determined by using mass concentration at m/z 44 and a parameterization presented in Aiken et al. (2007).

AMS  $D_{p,va}$  is converted to volume-equivalent diameter  $(D_{p,ve})$  by assuming constant particle density of 1.65 g/cm<sup>3</sup>. Data represent size-resolved AMS chemistry, averaged over 10 nm size bins. Since hygroscopicity measurements were carried out for particles of dry electrical mobility diameter  $(D_{p,em,dry})$  equal to 150, 175, and 200 nm, the corresponding AMS size bins were  $145 \le D_{p,ve} \le 155$  nm,  $170 \le D_{p,ve} \le 180$  nm, and  $195 \le D_{p,ve} \le 205$  nm, respectively.

2.3 Hygroscopicity measurements

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Hygroscopicity measurements were carried out with the Differential Aerosol Sizing and Hygroscopicity Spectrometer Probe (DASH-SP, Brechtel Mfg; Sorooshian et al., 2008). <sup>20</sup> Ambient particles pass through a nation dryer before receiving a uniform charge distribution in a <sup>210</sup>Po neutralizer. A cylindrical, single-classification differential mobility analyzer (DMA) then size selects particles into narrow ranges of mobility-equivalent diameters ( $D_{p,em}$ ) between 0.1 and 1.0 µm. The resulting monodisperse aerosol is split into five separate flows. One channel provides a redundant measurement of to-

tal particle concentration at the DMA-selected size with a water condensation particle counter (TSI Model 3831). The remaining four channels consist of parallel nafion humidification chambers (Perma Pure, LLP, Model MD-070-24FS-4), followed by cor-

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respondingly humidified custom optical particle counters (OPCs). In the OPC sample volume, particles pass through a focused laser beam ( $\lambda$ =532 nm, World Star Technologies, Model TECGL-30) and scatter light in proportion to size  $(D_p)$  and refractive index (RI). Forward-scattered light is collected and focused on a photomultiplier tube, and 5 the resulting electrical pulse is recorded by a high-speed data acquisition computer. An iterative data processing algorithm, based on laboratory calibrations with salts of known refractive indices, is used to determine the best fit on a solution surface relating electrical pulse height, size, and refractive index. The hygroscopic growth factor  $(GF = D_{p,wet}/D_{p,drv})$  is corrected for the RI change caused by particulate liquid water at elevated RH. In the current study, hygroscopicity was measured at dry sizes corresponding to D<sub>n em</sub> of 150, 175, and 200 nm. Multiple RH sensors in the nation tubes and OPCs controlled RHs to dry (<8%), 74%, 85%, and 92%, with RH uncertainty of 1.5%. Low particle loadings inherent in the marine atmosphere required increased online collection times at each DMA size step, but usually < one minute was sufficient to overcome counting statistic limitations. Overall uncertainty in GF calculations is 4.5%. 15 Assuming particles to be uniform, non-light-absorbing spheres allows the assumption that the intensity of scattered light is a function of only RI and  $D_p$ . This assumption also allows calculation of dry, 'effective' RI from the known DMA-selected  $D_p$  and measured scattered light intensity.

20 2.4 Hygroscopic closure

A volume-weighted mixing rule was used to perform a hygroscopic closure using AMS and DASH-SP data, under the assumption of independent and additive water uptake by individual constituents in each particle:

$$GF_{\text{mixed}}(a_w) = \left(\sum_i \epsilon_i GF_i(a_w)^3\right)^{1/3}$$

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where  $GF_{mixed}$  is the overall particle GF,  $a_w$  is the water activity,  $GF_i$  is the hygroscopic growth factor for pure species *i*, and  $\varepsilon_i$  is the volume fraction of species *i*. At equilibrium,  $a_w$ =RH (Seinfeld and Pandis, 2006). Values for  $\varepsilon_i$  were calculated for the following species, using AMS masses of ammonium  $(NH_4^+)$ , nitrate  $(NO_3^-)$ , sulfate  $(SO_4^{2-})$ , and total organic: ammonium nitrate  $(NH_4NO_3)$ , ammonium sulfate  $((NH_4)_2SO_4)$ , ammonium bisulfate  $(NH_4HSO_4)$ , sulfuric acid  $(H_2SO_4)$ , and organic. Partitioning between sulfate species was determined on the basis of the ammonium to sulfate molar ratio.  $(NH_4)_2SO_4$  is assumed to have a *GF* of unity at 74% since particles are exposed to RH well below its effloresence point before subsequent rehumidification. The organic fraction was assumed to be hydrophilic WSOC, based on evidence of a high degree of organic oxygenation from AMS mass spectra (see Sect. 3.3). Values of *GF<sub>i</sub>* for the organic fraction were calculated as those necessary to minimize the root mean square error in comparing predictions with measured hygroscopicity.

#### 3 Results

- <sup>15</sup> *GF* values for seven flights (RF 7, 10, 11, 12, 13, 14, and 16) are presented in Table 2. Table 4 presents measured *GFs* at each RH and each  $D_{p,dry}$ . Typically, multiple measurements were made on each leg, at each size, for each flight. The error reported is  $\pm$  one standard deviation in these multiple measurements. When only one measurement was made at a given size in a given leg, it is reported without error. Ship plumes were encountered on flights 10 and 16, as evidenced by brief, significantly elevated particle number concentration. Analysis of *GF* measurements in these presumptive plumes is not presented in the present work.
  - 3.1 Airmass origin

Relative to the other flights, RF 12 and 16 exhibited significantly suppressed water uptake at high RH for all dry sizes. During these two flights, 92% *GF* values for 200 8, 16789–16817, 2008

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nm  $D_{p,dry}$  particles averaged 1.61±0.14, as compared with 1.91±0.07 for all other flights. These low-*GF* flights corresponded to significantly elevated total organic, as measured by the AMS. Mass concentration averaged 1.97±1.71µg/m<sup>3</sup> organic (as opposed to 0.58±0.63 µg/m<sup>3</sup> for all other flights), corresponding to volume fraction  $^{5}$  organic (*VF*<sub>organic</sub>) of 0.46±0.22 (as opposed to 0.24±0.18 for all other flights).

Back-trajectory analysis suggests that the MASE-II flights can be categorized by airmass origin as either clean/marine (flights 7, 10, 11, 13, 14) or polluted/continental (flights 12, 16). Figure 1 shows 92% *GF* measurements for 200 nm  $D_{p,dry}$  particles, with corresponding 48-hr HYSPLIT (available at http://www.arl.noaa.gov/ready/ hysplit4.html) back-trajectories identifying airmass origin. Note that the low *GFs* and high *VF*<sub>organic</sub> measured on flights 12 and 16 correspond with airmass origins over the continental United States, while higher *GFs* and lower *VF*<sub>organic</sub> measured on other flights correspond to airmass origins over the clean marine environment. It is interesting to note that trajectories at sea level have marine origins for all flights, including

- <sup>15</sup> low-*GF* flights 12 and 16, suggesting that aloft airmass origin is a more significant factor in determining aerosol characteristics. Further, airmasses corresponding to clean, high-*GF* flights often pass either directly over or very near the San Francisco Bay Area, but these trajectories do not appear to significantly affect aerosol *GF* or *VF*<sub>organic</sub>. It is also noteworthy that *GF* values at low RH were not significantly suppressed, with val-
- <sup>20</sup> ues of  $1.31\pm0.06$  at 74% (compared with  $1.31\pm0.07$  for all other flights). *GF* values at intermediate *RH* were moderately suppressed in the continental airmass, measuring  $1.53\pm0.10$  at 85% (compared with  $1.58\pm0.08$  for all other flights). In other words, the effect of the high-*VF*<sub>organic</sub>, polluted/continental airmass is to significantly suppress *GF* at high RH, while having no measurable effect on aerosol water uptake at low RH and a moderate impact at intermediate RH.

#### 3.2 Hygroscopicity trends

No size-dependent hygroscopicity was observed over the range of measured  $D_{p,dry}$ , despite slightly elevated  $VF_{\text{organic}}$  at smaller sizes. This suggests that minor size-

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dependent changes in composition had too small an impact on particle GF to be detected by the DASH-SP.

- Figure 2 shows below- and corresponding above-cloud 92% GF values for all flights. There exists a ubiquitous trend of higher below-cloud aerosol GF values at high RH when compared with top of cloud hygroscopicity values (1.88±0.14 below cloud, ver-5 sus 1.78±0.18 above cloud). Marker sizes, proportional to  $VF_{\text{organic}}$  suggest a trend of higher organic loading above cloud in several flights (0.27±0.15 below cloud, versus  $0.31\pm0.21$  above cloud). While variability is rather large in these measurements, the trend of elevated VF<sub>organic</sub> above-cloud does correspond to suppressed GF at high RH. There is a trend of more moderate GF suppression at intermediate RH 10 in the higher  $VF_{\text{organic}}$  above-cloud layer (1.53±0.06 above cloud, as compared with 1.58±0.07 below cloud). Unlike the continental-influenced flights, there is evidence of more significant suppression of low-RH GFs in above-cloud legs, with 74% GF values of  $1.31\pm0.05$  above cloud, compared with  $1.36\pm0.05$  below. It appears, then, that the elevated organic loadings typical of above-cloud legs are correlated with GF suppression at high RH, and more moderate GF suppression at lower RHs.
  - 3.3 Hygroscopic closure

A hygroscopic closure was performed, using volume-weighted hygroscopic contributions from each chemical constituent identified by the AMS. O:C ratios for the flights <sup>20</sup> presented were 0.92±0.33, with very similar ratios for below, above, and free troposphere legs on all flights. This result indicates a consistently uniform, highly oxidized organic component (Aiken et al., 2007), with little'fresh,' or HOA-type organic carbon. With this evidence, the organic fraction was treated as a bulk, water-soluble constituent (Kondo et al., 2007), as opposed to being speciated into a soluble and insoluble OOA <sup>25</sup> and HOA according to Zhang et al. (2005). As described in Sect. 2.4, *GF* values

were calculated for the organic component of the aerosol by minimizing root mean square error when comparing measured GFs with volume-weighted closure predictions. Mission-averaged organic GFs were determined to be 1.20, 1.43, and 1.46 at



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Interactive Discussion

74, 85, and 92% RH, respectively. These values agree well with those of hydrophilic multifunctional and dicarboxylic acids studied by Peng et al. (2001). Closure results are presented in Fig. 3. Markers in Fig. 3 are color-coded according to relative humidity, and marker size is proportional to  $VF_{\text{organic}}$ . The volume-weighted hygroscopic closure utilizing size-resolved AMS chemistry achieves good agreement with the 675 DASH-SP *GF* measurements, with an  $R^2$  of 0.79. Agreement is better at lower values of RH, owning to smaller *GF* magnitudes and less overall *GF* variability.

Aside from the obvious RH dependence of *GF* values, the clearest trend in Fig. 3 is that of larger markers (higher  $VF_{\text{organic}}$ ) at low *GF* transitioning to smaller markers (lower  $VF_{\text{organic}}$ ) at high values of *GF* for the same RH. The clarity and regularity of this trend reveal the importance of the organic fraction in determining *GF* values, and suggest that, of all the chemical species considered in the closure, the organic fraction is a particularly crucial parameter in predicting *GF*.

- 3.4 Simplified parameterization
- To further investigate the relative importance of each parameter in quantifying aerosol water uptake, and to detemine the simplest statistical model still capable of accurately predicting *GF*, a backward stepwise linear regression was performed. The process, which eliminates predictors one-by-one to generate increasingly simplified linear representations of data, started with over 60 predictors, ranging from PILS and AMS chemi cal parameters to atmospheric data. The result is a two-parameter model that predicts *GF* as a function of RH and *VF*<sub>organic</sub>:

 $GF = -0.314 + 2.33(RH) - 0.30(VF_{\text{organic}})$ 

Figure 4 demonstrates the accuracy with which this model predicts DASH-SP *GF* values over measured the range of RH and  $VF_{\text{organic}}$ . It is noteworthy that the  $R^2$  for this model is 0.77, indicating that the simple, two-parameter model explains only 2% less variability than the full volume-weighted chemical closure, which contains significantly more information (i.e. multiple inorganic chemical species and their individual *GFs*, in

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addition to organic fraction with its associated *GF*). So even with information on the specific nature of organics in the atmosphere, it appears as though accurate predictions of particle water uptake in the marine atmosphere can be made simply on the basis of relative humidity and the relative abundance of organics in the aerosol. While this model accurately predicts *GF* over the range of RH and *VF*<sub>organic</sub> measured during MASE-II, it should not be used when RH is outside the range 74%  $\leq$  RH  $\leq$  92%, or where *VF*<sub>organic</sub> is less than 0.1.

#### 4 Discussion

Comparing mixed organic-inorganic particles with those comprised entirely of inorganic
 salts, there is a strong RH-dependence in the effect of organics on hygroscopicity (Peng et al., 2001). During the MASE-II field campaign, *GF* values at 74% RH averaged ~1.3. Organic *GFs* were inferred to be 1.20 at 74% RH, suggesting that they contributed significantly to overall aerosol water uptake at low RH. *GF* values at 85% RH averaged ~1.6 during the campaign, and so the organic *GF* of 1.43 calculated for 85% RH suggests that organics played a less significant role, but still influenced water uptake at intermediate RH. An organic *GF* value of 1.46 was calculated for 92% RH, while 92% *GF* measurements averaged ~1.8, suggesting that organics contributed little to overall *GF* at high RH.

Inorganic salts exhibit deliquescent behavior as RH is increased. Many organics do not deliquesce or crystallize when RH is increased or decreased, respectively, but instead retain water at RH values well below the RH of deliquescence (RHD) of the inorganic salts with which they often co-exist in ambient particles. As a result, at RH values below the salt RHD, the presence of organics enhances water uptake. Thus, the effect of organics is to contribute significantly to overall water uptake at low values of RH in mixed organic-inorganic particles. At high RH, on the other hand, organics tend

to take up significantly less water than the inorganic constituents with which they coexist in ambient particles. Therefore, at RH values above the inorganic RHD, organics





appear to suppress water uptake relative to that which a pure inorganic particle would exhibit.

Figure 5 shows the behavior of a pure ammonium sulfate (AS) particle, pure organic acid (OA) particle, and a mixed organic acid and ammonium sulfate (OA/AS) particle
over the range 50%≤RH≤94%. Note that the mixed OA/AS particle shows smooth growth with RH, as opposed to the deliquescent behavior exhibited by the pure AS particle. The tendency of organics to retain water at low RH causes water uptake behavior for the OA/AS particle to follow that of the descending (i.e. efflorescence) branch of the pure AS growth curve. Since pure OA takes up less water at high RH that of pure AS, the growth curve for the OA/AS particle is suppressed, compared with that of pure AS. The overall result, as predicted by thermodynamic theory, is that the presence of OA leads to enhanced water uptake at low RH and suppressed *GF* at high RH.

Virkkula et al. (1999) concluded that the most important factor contributing to *GF*<sup>15</sup> suppression at high RH was the volume fraction of organic present in an aerosol. Others have suggested that the exact chemical identity of the organic constituents is not especially important, and that for an organic component classified as either oxidized or hydrocarbon-like, its relative abundance determines its effect on *GF* values (McFiggans et al., 2005; Moore and Raymond, 2008). Results presented here from a stepwise
<sup>20</sup> linear regression on *GF* data from the marine atmosphere suggest that the single most important factor in predicting *GF* (aside from RH) is, indeed, *VF*<sub>organic</sub>.

In most instruments that measure aerosol hygroscopicity, residence times for humidification are on the order of seconds, much longer than the equilibration time for most inorganics with water vapor. Sjogren et al. (2007) noted, however, that parti-

cles with high volume fraction organic material may require as long as 40 s to achieve equilibrium with water vapor. If such long times are necessary to achieve equilibrium, hygroscopic measurement methods suitable for the field will tend to overpredict *GF* at low RH (water vapor does not evaporate completely from the particle during the drying process), while underpredicting *GF* at high RH (insufficient humidification time is

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provided for the organic fraction to achieve thermodynamic equilibrium with water vapor). In electrodynamic balance (EDB, (Cohen et al., 1987a,b,c)) studies, suspended particles are subjected to extended exposure to water vapor (minutes to hours), establishing equilibrium. Some organics exhibit extremely high deliquescence relative

- <sup>5</sup> humidities (DRH) (e.g., oxalic acid), while others exhibit gradual hygroscopic growth at low RH and substantial growth at high RH (e.g., malonic acid) (Peng et al., 2001). It is possible, given the wide range of organic species in the atmosphere and correspondingly wide range of hygroscopic properties associated with those species, and the relatively short humidification times in the DASH-SP and other similar instruments,
- that the effects attributed to organics may reflect some kinetic instrumental limitations.

#### 5 Conclusions

We report a hygroscopic closure study for the marine aerosol, using high-resolution AMS chemical data coupled with highly time-resolved, multiple-RH hygroscopicity measurements from the DASH-SP.

- <sup>15</sup> Airmasses originating from continental locations showed elevated organic loading, and corresponded to significant *GF* suppression at high RH. More moderate *GF* suppression was measured at intermediate RH and no impact was observed at low RH. A comparison of above-cloud with below-cloud aerosol indicated that a slightly organicenriched layer above cloud corresponded with suppressed *GFs* at high RH. A volume-
- <sup>20</sup> weighted hygroscopic closure provided excellent agreement with measured *GFs*, and mission-averaged organic *GFs* at 74, 85, and 92% were calculated to be 1.20, 1.43, and 1.46, respectively. These *GF* values are relatively high when compared with many previous estimations of organic *GF*, but agree well with values reported for dicarboxylic and multifunctional acids. These high organic *GFs* are indicative of the highly oxidized
- state of the aged organic fraction. A simplified parameterization for predicting *GF* was developed using a stepwise linear regression method. This parameterization is a function of only RH and  $VF_{\text{organic}}$ , and explains only 2% less variability than does

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the volume-weighted chemical closure. The GF predictions and parameterizations presented here accurately predict GFs from high-resolution chemical inputs, and may be broadly applicable to the marine environment, where some oxidized organics are present in an otherwise clean, atmosphere.

The richness of this hygroscopic/chemical data set underlies the significance of coupling the DASH-SP instrument with the AMS. The improved time resolution available with the DASH-SP eliminates much of the time-averaging of AMS chemical data that is otherwise necessary with less highly time-resolved HTDMA hygroscopicity measurements. The importance of simultaneous *GF* measurements at multiple RH values is
 demonstrated by a simplified parameterization for predicting *GF* as a function of RH and *VF*<sub>organic</sub>; a result potentially important for efficiently representing aerosol-water interactions in global models.

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Investigators	Year	Study Area	Reported GFs
Hegg	1996	Eastern Pacific	1.4–2.0
Berg	1998	Pacific and Southern Oceans	1.56–1.78
Kotchenruther	1999	Mid-Atlantic	1.81–2.3 <sup>a</sup>
Gasso	2000	Eastern Atlantic	$\gamma = 0.51 - 0.75^{b}$
Swietlicki	2000	Northeastern Atlantic	1.6–1.8
Zhou	2001	Arctic Ocean	1.4–1.9
Hegg	2002	Eastern Pacific	$\gamma = 0.20.7^{b}$
Vakeva	2002	Northeastern Atlantic	1.3–1.4 <sup>c</sup>
Massling	2003	Atlantic and Indian Oceans	1.6–2.0
Hegg	2006	Eastern Pacific	1.3–1.5
Kaku	2006	Eastern Pacific	1.2–1.7
Tomlinson	2007	Southeastern Pacific	1.3–1.7 <sup>d</sup>

 Table 1. Previous marine aerosol hygroscopicity studies.

<sup>a</sup> Ratio of light scattering coefficient at 80% RH to 30% RH.

<sup>b</sup>  $\gamma$  from the expression  $b_{sp}(\text{RH})/b_{sp}(\text{RH}_0) = ((1 - \text{RH}/100)/(1 - \text{RH}_0/100))^{-\gamma}$  (Kasten, 1969).

<sup>c</sup> Aitken mode particles.

<sup>d</sup> 85% RH.

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#### Table 2. MASE-II flights.

Flight Number (RF)	Date	Type of Flight
1	10-Jul-07	Cloud/aerosol interactions
2	11-Jul-07	Cloud/aerosol interactions
3	12-Jul-07	Bovine source probe
4	14-Jul-07	Cloud/aerosol interactions
5	15-Jul-07	Cloud/aerosol interactions
6	16-Jul-07	Coordinated ship plume probe
7	21-Jul-07	Cloud/aerosol interactions
8	22-Jul-07	Cloud/aerosol interactions
9	23-Jul-07	Cloud/aerosol interactions
10	24-Jul-07	Cloud/aerosol interactions
11	25-Jul-07	Cloud/aerosol interactions
12	26-Jul-07	Cloud/aerosol interactions
13	28-Jul-07	Cloud/aerosol interactions
14	29-Jul-07	Cloud/aerosol interactions
15	30-Jul-07	Bovine source probe
16	31-Jul-07	Cloud/aerosol interactions

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#### Table 3. MASE-II instrument payload.

Parameter	Instrument	Time Resolution	Detection Limit	Size Range
Particle Number Concentration	Condensation Particle Counter (TSI CPC 3010)	1 s	0–10 000 cm <sup>-3</sup>	$D_p \ge 10 \text{ nm}$
Particle Number Concentration (including ultrafine)	Condensation Particle Counter (TSI CPC 3025)	1 s	$0-100000~{\rm cm}^{-3}$	$D_{ ho}$ ≥3 nm
Aerosol Size Distribution	Scanning differential mobility analyzer (Dual Automated Classified Aerosol Detector, DACAD)	73 s	N/A	10–700 nm
Aerosol Size Distribution	Passive Cavity Aerosol Spectrometer Probe (PCASP)	1 s	N/A	0.1–2.6 μm
Separation of Cloud Droplets From Interstitial Aerosol	Counterflow Virtual Impactor	N/A	N/A	N/A
Cloud Droplet Size Distribution	Phase Doppler Interferometer (PDI)	1 s	N/A	4–200 μm
Cloud and Drizzle Drop Size Distribution	Forward Scattering Spectrometer Probe (FSSP)	1 s	N/A	1–46 µm
Cloud Droplet Liquid Water Content	Light Diffraction (Gerber PVM-100 probe)	1 s	N/A	5–50 nm
Aerosol Bulk Ionic Composition and Soluble Organic Composition	Particle-Into-Liquid Sampler (PILS)	5 min	0.02–0.28 μg/m <sup>3</sup>	1 µm
Aerosol Bulk Composition	Aerodyne Time-of-Flight Aerosol Mass Spectrometer (TOF-AMS)	1 s or 15 s	0.25 μg/m <sup>3</sup>	40 nm≤ <i>D<sub>p,va</sub></i> ≤1 μm
Aerosol Hygroscopicity	Differential Aerosol Sizing and Hygroscopicity Spectrometer Probe (DASH-SP)	15 - 100 s	N/A	150 nm≤ <i>D</i> <sub>ρ</sub> ≤1 μm
Soot Absorption	Photoacoustic Absoprtion Spectrometer (PAS)	1 s	1 Mm <sup>-1</sup>	10 nm–5 µm
Soot Absorption	Particle Soot Absorption Photometer (PSAP)	≥1 s	N/A	N/A

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Interactive Discussion



#### 150 nm Flight Below Above FT Ship Plume 74% 85% 92% 74% 85% 92% 74% 85% 92% 74% 85% 92% 7 $1.40 \pm 0.04$ 1.57±0.03 1.82±0.05 1.26±0.06 1.55±0.04 1.78±0.01 1.40±0.03 1.55±0.03 1.84±0.04 10 1.42±0.04 1.60±0.06 $1.80 \pm 0.04$ 1.37±0.10 1.57±0.04 1.71±0.11 1.42±0.03 1.56±0.04 1.70±0.07 $1.42 \pm 0.04$ 1.62±0.03 1.78±0.04 11 1.44±0.05 1.61±0.06 1.80±0.07 1.40±0.07 1.51±0.02 1.64±0.05 1.38±0.07 1.52±0.06 1.65±0.10 $1.51 \pm 0.03$ 12 $1.38 \pm 0.03$ $1.58 \pm 0.03$ $1.63 \pm 0.04$ $1.34 \pm 0.04$ $1.55 \pm 0.01$ $1.38 \pm 0.02$ $1.60 \pm 0.01$ $1.63 \pm 0.01$ $1.39 \pm 0.01$ $1.61 \pm 0.05$ $1.82 \pm 0.02$ $1.35 \pm 0.01$ $1.61 \pm 0.03$ $1.75 \pm 0.05$ 1.27±0.05 $1.49 \pm 0.07$ $1.61 \pm 0.08$ 13 14 $1.36 \pm 0.05$ $1.65 \pm 0.21$ $1.82 \pm 0.14$ $1.39 \pm 0.02$ $1.62 \pm 0.01$ 1.87±0.09 $1.37 \pm 0.03$ $1.60 \pm 0.03$ 1.77±0.06 1.41±0.02 1.58±0.05 $1.64 \pm 0.08$ 1.33±0.03 1.52±0.03 $1.54 \pm 0.03$ $1.29 \pm 0.04$ 1.49±0.07 1.52±0.06 1.38±0.02 1.59±0.02 1.61±0.04 16

Table 4. GF results for below, above, free troposphere (FT), and ship plume measurements.

175 nm												
Flight	Below	Above				FT			Ship Plume			
	74%	85%	92%	74%	85%	92%	74%	85%	92%	74%	85%	92%
7	1.36±0.06	1.54±0.02	1.95±0.03	1.28±0.04	1.53±0.03	1.91±0.06	1.24±0.08	1.52±0.03	1.87±0.10			
10	1.41±0.06	$1.59 \pm 0.05$	1.93±0.02	1.34±0.05	1.51±0.01	1.83±0.06	1.33±0.02	1.60±0.03	1.89±0.05	1.34±0.02	1.55±0.05	1.88±0.06
11	1.31±0.08	1.48±0.07	1.82±0.09	1.31±0.03	1.44±0.09	1.77±0.03	1.33±0.04	1.53±0.12	1.77±0.23			
12	1.33±0.02	1.55±0.04	1.63±0.08	1.26±0.06	1.48±0.08	1.51±0.07	1.35±0.02	1.57±0.01	1.60±0.01			
13	1.34±0.03	1.58±0.04	2.00±0.05	1.3	1.55	1.78	1.24±0.05	1.44±0.06	1.63±0.10			
14	1.34±0.01	1.63±0.02	2.03±0.01	1.32±0.02	1.58±0.02	1.91±0.05	1.33±0.02	$1.50 \pm 0.08$	1.79±0.22			
16	1.35±0.04	$1.54 \pm 0.03$	1.62±0.06	1.28±0.02	1.43±0.04	1.45±0.04	1.24±0.04	1.38±0.06	1.40±0.10	1.33±0.03	1.52±0.04	1.55±0.03

	200 nm											
Flight	Below	Above				FT			Ship Plume			
	74%	85%	92%	74%	85%	92%	74%	85%	92%	74%	85%	92%
7	1.37±0.04	1.57±0.03	1.98±0.03	1.22±0.06	1.56±0.05	1.92±0.0.01	1.21±0.01	1.54±0.08	1.91±0.04			
10	1.41±0.09	1.72±0.09	2.01±0.05	1.33±0.02	1.57±0.07	1.87±0.04	1.35±0.01	1.63±0.06	1.93±0.06	1.33±0.04	1.66±0.06	1.95±0.05
11	1.25±0.11	1.45±0.17	1.86±0.07	1.26	1.47	1.83	1.36±0.16	1.56±0.15	1.89±0.20			
12	1.33±0.03	1.64±0.03	1.73±0.05	1.29±0.03	1.48±0.10	1.55±0.12	1.37±0.05	1.64±0.05	1.69±0.04			
13	1.36±0.05	1.68±0.05	2.01±0.04	1.3	1.61	1.85	1.21±0.08	1.45±0.07	1.78±0.12			
14	1.42±0.11	1.65±0.05	1.98±0.10	1.31±0.05	1.60±0.03	1.96±0.01	1.35±0.09	1.59±0.10	1.89±0.11			
16	1.35±0.02	$1.59 \pm 0.05$	1.77±0.08	1.26±0.02	1.45±0.08	1.53±0.10	1.23±0.06	$1.40 \pm 0.10$	1.40±0.11	1.33±0.03	1.62±0.04	1.66±0.06

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Interactive Discussion





**Fig. 1.** 92% *GF* values for 200 nm particles are shown for below and above cloud flight legs. 48-hr HYSPLIT back-trajectories show airmass origin for each flight. Flights 12 and 16 show both suppressed *GF* values and continental, polluted airmass origin.

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**Fig. 2.** Below- and above-cloud 92% *GF* values for 200 nm particles as a function of altitude, with lines connecting measurements made on the same "trip" from bottom to top of cloud. Marker size is proportional to  $VF_{\text{organic}}$ . "a" and "b" designate separate "trips" during the same flight.

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**Fig. 3.** Measured *GF* values versus volume-weighted predictions. Markers are color-coded with RH, and marker size is proportional to  $VF_{\text{organic}}$ , as determined by the AMS. The blue line represents 1:1.

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# **Fig. 4.** Predictions of *GF* from the simplified parameterization. Markers are color-coded with RH, and marker size is proportional to $VF_{\text{organic}}$ , as determined by the AMS. The blue line represents 1:1.

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**Fig. 5.** Comparison of growth curves for pure ammonium sulfate, pure hydrophilic organic, and mixed organic/inorganic particles. Ammonium sulfate curves calculated from AIM (Clegg and Seinfeld, 2006).