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**Airborne emission
measurements over a
megacity**

T. Karl et al.

Emissions of volatile organic compounds inferred from airborne flux measurements over a megacity

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Abstract

Toluene and benzene are used for assessing the ability to measure disjunct eddy covariance (DEC) fluxes of Volatile Organic Compounds (VOC) using Proton Transfer Reaction Mass Spectrometry (PTR-MS) on aircraft. Statistically significant correlation between vertical wind speed and mixing ratios suggests that airborne VOC eddy covariance (EC) flux measurements using PTR-MS are feasible. City-average midday toluene and benzene fluxes are calculated to be on the order of $15.5 \pm 4.0 \text{ mg/m}^2/\text{h}$ and $4.7 \pm 2.3 \text{ mg/m}^2/\text{h}$ respectively. These values argue for an underestimation of toluene and benzene emissions in current inventories used for the Mexico City Metropolitan Area (MCMA). Wavelet analysis of instantaneous toluene and benzene measurements during city overpasses is tested as a tool to assess surface emission heterogeneity. High toluene to benzene flux ratios above an industrial district (e.g. 10–15) including the International airport (e.g. 3–5) and a mean flux (concentration) ratio of 3.2 ± 0.5 (3.9 ± 0.3) across Mexico City indicate that evaporative fuel and industrial emissions play an important role for the prevalence of aromatic compounds. Based on a tracer model, which was constrained by BTEX (Benzene/Toluene/Ethylbenzene/m,p,o-Xylenes) compound concentration ratios, the fuel marker methyl-tertiary-butyl-ether (MTBE) and the biomass burning marker acetonitrile (CH_3CN), we show that a combination of industrial, evaporative fuel, and exhaust emissions account for >90% of all BTEX sources. Our observations suggest that biomass burning emissions play a minor role for the abundance of BTEX compounds (0–10%) in the MCMA.

1 Introduction

Volatile organic compounds (VOCs) represent a major component of photochemical smog by fueling tropospheric ozone production (Atkinson, 2003). More recently their importance for the organic aerosol budget in the atmosphere has been demonstrated in the field (e.g. de Gouw et al., 2005, 2008; Volkamer et al., 2006; Weber et al.,

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2007). Accurate predictions of VOCs in chemical transport (CT) model simulations rely heavily on their emission strengths. Recent field observations suggest significant uncertainties of anthropogenic VOC emission inventories in major metropolitan areas (e.g. Warneke et al., 2007; Jobson et al., 2004; Karl et al., 2002; Zhao et al., 2004).

5 Emission inventories for developing Megacities are considered particularly uncertain. Air pollution management relies on accurate predictions of VOC to NO_x ratios in order to determine effective ozone reduction strategies (e.g. Liu et al., 1987; Sillman, 1995; Kleinman et al., 2005). Conflicting results on VOC versus NO_x sensitivities for ozone production rates have been published for Mexico City (see Stephens et al., 2008, in
10 this special issue). Based on a regional CT model Tie et al. (2007) show that variable VOC emission estimates translate into large offsets of calculated daytime ozone mixing ratios in Mexico City. The sensitivity of modeled ozone concentrations to VOC emission inputs has also been demonstrated by adjusting biogenic emission maps in the US. For example, the difference between two biogenic emission inventories (BEIS 1
15 and BEIS 2) almost doubled the frequency of modeled ozone exceedances (e.g. mixing ratios >80 ppbv) in the Eastern US (Pierce et al., 1998). As a consequence the assessment of emission inventories has important implications for policy decisions.

It is common practice to use measured VOC concentrations as one important constraint for CT models. The atmospheric concentration of a reactive compound however
20 can be seen as balance between emission, deposition, transport and chemistry. With so many degrees of freedom, concentration measurements alone make it hard to diagnose uncertainties in CT models. Currently bottom-up emission inventories are typically tuned manually or by data assimilation techniques (Arellano et al., 2007) so that modeled concentration fields match those observed. This can result in uncertain as-
25 sumptions. For example, based on a modeling study West et al. (2004) suggested an overall increase of the VOC emission inventory (CAM, 2001) by a factor of 2–3 to match observed VOC concentrations in Mexico City. Their work was contrasted by Velasco et al. (2005) who argued that the urban emission inventory for Mexico City was generally consistent with their ambient air measurements. More recently, Lei et al. (2007) used

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a revised emission inventory (CAM, 2004) and adjusted the initial emission estimates until “satisfactory agreement” between modeled concentrations and observations was reached. Some of their adjusted emissions for aromatic compounds in their lumped chemical scheme (e.g. ARO1, ARO2) were a factor of 2.5 higher than estimates by Velasco et al. (2007). It was noted that species lumping in condensed chemical schemes can lead to additional uncertainty (Lei et al., 2007).

In order to disentangle surface exchange from other processes effecting the distribution of reactive trace gases, direct flux measurements can add one important additional constraint on the atmospheric cycle of VOCs and help lessen uncertainties of emission inventories. The most direct method for quantifying surface fluxes is the eddy covariance (EC) method based on atmospheric turbulence measurements (Kaimal, 1972). EC is widely used for measurements of the exchange of energy and air constituents in the atmosphere (Stull, 1988). The main challenge of this technique is the requirement of sampling rates on the order of 10 Hz; however, this can be relaxed by the introduction of disjunct sampling strategies (Lenschow et al., 1994). Ground based EC and disjunct eddy covariance (DEC) methods are increasingly deployed for gas and aerosol flux measurements over forested (Guenther and Hills, 1998; Karl, 1999; Rinne et al., 2001; Karl et al., 2002; Spirig et al., 2005; Graus et al., 2006; Lee et al., 2005; Pressley et al., 2006) and urban areas (e.g. Nemtizi et al., 2002; Velasco et al., 2005). In contrast these measurements have been limited to very few reactive gases on aircraft: For example Faloon et al. (2005) measured EC fluxes of dimethylsulfide (DMS) and ozone (O₃) over the ocean and inferred entrainment rates for these species. To date ozone is the only reactive gas that has been commonly measured by EC on aircraft (e.g. Lenschow et al., 1981; Mauder et al., 2007).

For the first time, we test the ability to measure airborne toluene and benzene fluxes over Mexico City, taking advantage of the measurement capabilities on the NCAR-C130 aircraft during the MIRAGE-MEX/MILAGRO project in 2006. We compare these fluxes to current emission inventories used in regional models. As precursor to secondary organic aerosol (e.g. Sato et al., 2007; Ng et al., 2007) and ozone formation

(e.g. Atkinson et al., 1980; Tie et al., 2007), aromatic (BTEX) compounds, such as toluene, are important ingredients of urban photochemical smog. Due to their toxicity BTEX compounds are of particular concern from a health related perspective. For example chronic exposure to benzene, a known carcinogen, can lead to bone marrow damage, leukemia and depression of the immune system (e.g. Rinsky et al., 1987). A good understanding of source distributions and atmospheric transformations of BTEX compounds is therefore needed to investigate their impact on urban and regional atmospheric chemistry and human health.

2 Experimental description

2.1 MIRAGE-Mex (Megacity impacts on regional and global environments – Mexico)

General information on the experimental design of MIRAGE-Mex which was part of the MILAGRO field project (Megacity Initiative: Local and Global Research Observations) is summarized by Doran et al. (2007) and Molina et al. (2008). A meteorological overview can be found in Fast et al. (2007). As part of MIRAGE-Mex twelve research flights were conducted with the NCAR C-130 research aircraft (<http://www.eol.ucar.edu/instrumentation/aircraft/C-130>). Figure 1a shows all flight tracks plotted on top of a topographical map showing Central Mexico. Here we focus on a subset of individual flights in the vicinity of Mexico City. Figure 1b shows a zoomed map on top of which typical flight patterns in and around Mexico City are plotted. Research flights (RF) 1 (4 March 2006), 6 (18 March 2006), 8 (22 March 2006) and 12 (29 March 2006) included flight legs in Mexico City and are color coded in Fig. 1b. The highlighted portion of the flight tracks (in blue) in the Mexico City Metropolitan Area (MCMA) is focus of the present investigation. For orientation Fig. 1b also depicts surface supersites (T0, T1 and T2) as well as two separate locations where flux measurements were conducted on the ground in 2003 (T-1₂₀₀₃) and 2006 (T-1₂₀₀₆). Due to tight air traffic regulations each MCMA approach was planned in an exactly iden-

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tical fashion. Before entering the MCMA a profile at 1350, 1650 and 2050 m above ground was flown south-east, which was followed by a city transect (~40 km) in the planetary boundary layer (PBL). City overpasses during research flights 1, 6, 8 and 12 (RF1, RF6, RF8 and RF12) were typically conducted 585±50 m above ground in the lower part of a well developed (e.g. 2500–3000 m deep) mixed layer. Disjunct eddy covariance (DEC) measurements were attempted on 3 of 4 flights (RF6, RF8 and RF12). Planetary boundary layer heights (PBL) were 3600±400 m, 3500±300 m and 3000±300 m respectively during research flights in the MCMA (Shaw et al., 2006), resulting in normalized flight altitudes (z/h) of 0.16, 0.17 and 0.2 above ground. The GPS corrected wind signal during RF6 was compromised by operational noise problems and DEC analysis for RF6 is therefore omitted. VOC data collected during RF1 and RF6 however are included in the analysis of concentration ratios.

2.2 VOC sampling

A Proton-transfer-reaction mass spectrometer (PTR-MS) was deployed on the NCAR C-130 aircraft for fast VOC measurements. The instrument was operated at 2.0 mbar (110 Td) (Lindinger et al., 1998). Measurements reported here were obtained at a sampling rate of 10 Hz and a repetition rate of 1 Hz. Periodic calibrations using two VOC standards (Matheson TriGas, USA; Apel Riemer Environmental Inc, USA) were conducted during pre-flight operations. The sensitivity of the PTRMS was on the order of 20–70 cps/ppbv, which was somewhat lower during the MIRAGE-MEX field project than typically encountered (e.g. 100–600 cps/ppbv). We attribute this to a different secondary electron multiplier (MasCom, MC-217, Germany) with a lower detection efficiency used during this field deployment. Intercomparison with two complementary VOC systems (TOGA – Trace Organic Gas Analyzer – and WAS – whole air canister sampling; Apel et al., 2007) however showed excellent agreement for most VOCs within the combined uncertainties. Table 1 lists VOCs measured by PTRMS along with sensitivities, detection limits and results from an intercomparison with TOGA and WAS.

While Jobson et al. (2004) have reported a 16% overestimation of benzene mixing ra-

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tios measured with their PTRMS instrument attributed to fragmentation of higher alkyl-benzenes (e.g. ethyl-benzene), Rogers et al. (2006) showed generally good agreement for benzene inferred from an intercomparison between PTRMS and GC-FID. Comparable results for benzene measurements were also obtained in different urban environments (Warneke et al., 2001). The overall agreement between 3 independent VOC sampling methods on the C-130 during this study suggests that, within the uncertainty, benzene measurements obtained from PTRMS showed minimal bias due to the fragmentation of higher alkyl-benzenes.

The PTRMS sampling inlet consisted of a 2 m long 1/4" Teflon (PFA) tube pumped by a diaphragm pump (KNF Neuberger, UNF726.3, USA), where a portion of this flow (~250 sccm) was diverted into a pressure controlled inlet of the PTRMS instrument, such that the overall delay time was less than 3 s. Zero air was periodically back-flushed through the whole inlet system to determine instrumental background.

Whole air sampling (WAS) of hydrocarbons was based on a similar setup used for earlier studies where stainless steel cans were shipped to University of California Irvine Analytical Laboratories for further GC-FID and GC-MS analysis (e.g. Blake and Rowland, 1995).

Instrument specific details of TOGA (Trace Organic Gas Analyzer) are described in great depth elsewhere (Apel et al., 2003, 2007). Briefly, the system used here is composed of an inlet, a cryogenic preconcentrator and a gas chromatograph (GC) coupled to a mass spectrometer (Agilent 5973). Three traps (a water, an enrichment and a cryofocus trap) are used without the use of adsorbents. The GC is fitted with a Restek MTX-624, 0.18 μm , 8 m column using Helium as carrier gas (flow rate of 1 ml/min). The initial GC oven temperature (30°C) is held for 10 s followed by heating to 140°C at a rate of 110°C/min. The system was calibrated with an in-house gravimetrically prepared mixture, which was dynamically diluted with scrubbed ambient (outside aircraft) air to mixing ratios near typically observed levels. Compound dependent detection limits ranged from sub-pptv to 40 pptv.

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2.3 WRF-CHEM

The Weather Research and Forecasting model coupled with chemistry (WRF-Chem) provides the capability to simulate chemistry and aerosols from cloud scales to regional scales. It is a fully coupled meso-scale model that treats emissions, transport, multi-phase chemistry, radiation and dry deposition of major gaseous and particulate pollutants simultaneously. The configuration of the model used here is based on CBMz chemistry and the MOSAIC aerosol module (Zaveri et al., 2008). A detailed model description can be found in Fast et al. (2006), who have evaluated ozone and aerosol predictions with observations obtained from a field campaign in Houston, Tx, USA. During MIRAGE-MEX two official emission inventories (CAM01, 2001, and CAM04, 2004) were implemented for anthropogenic pollutants. Biogenic emissions were based on MEGAN (Guenther et al., 2006). Pyrogenic emissions were driven by satellite observations of fire counts scaled to a biomass burning emission model (Wiedinmyer et al., 2006).

3 Results and discussion

3.1 Flux calculations

The scalar conservation equation of an atmospheric constituent can be written as:

$$\frac{\partial C}{\partial t} + U \frac{\partial C}{\partial x} + \frac{\partial F}{\partial z} = S, \quad (1)$$

where C is the mean concentration of a scalar, U the mean wind speed, S source and sink terms and F the turbulent flux. The turbulent flux F is defined as the average over the fluctuating terms of vertical wind speed (w') and concentration (C'):

$$F = \overline{w' C'}. \quad (2)$$

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In order to capture all turbulent terms sampling rates of 10 Hz averaged over ~30 min are commonly used on the ground. At 5 m/s wind speeds this would relate to spatial scales of 9 km. Disjunct sampling methods have been introduced to relax the time requirement between consecutive samples (Rinne et al., 2001; Karl et al., 2002; Spirig et al., 2005). The turbulent flux is then calculated as the discrete covariance between w' and C' :

$$F = \sum_i w'_i C'_i. \quad (3)$$

The advantage of disjunct eddy covariance (DEC) is that it allows time for sample processing, while maintaining a 10 Hz sampling frequency. The disadvantage is that DEC limits the possibilities for cospectral analysis (e.g. investigation of the inertial subrange is not possible). In case of PTR-MS, DEC allows scanning various ions of interest while sampling at 10 Hz.

For turbulent statistics measured on aircraft Lenschow et al. (1994) have investigated the impact of statistical errors on sampling intervals. For a systematic error in the lower part of the mixed layer they derive:

$$\text{systematic error} \leq \frac{2.2z_i(z/z_i)^{1/2}}{L}, \quad (4)$$

where z_i is the planetary boundary layer height, z is the flight altitude and L is the length of the flight leg. For typical conditions described in this paper we calculate systematic errors (se) (e.g. $z_i=3000$ m, $z/z_i=0.16$, $L\sim 40\,000$ m) to be $<7\%$.

The random error (re) can be estimated according to,

$$\text{random error} \leq 1.75 \left(\frac{z}{z_i}\right)^{1/4} \left(\frac{z_i}{L}\right)^{1/2}, \quad (5)$$

and, for conditions described above, would result in $re <30\%$.

Mann and Lenschow (1994) presented equations to estimate the systematic and random fluxes as a function of spatial scales and correlation coefficient. Applying their

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Eq. (27), we estimate a systematic error of 10% to the measured flux for a 10 km long flight segment across the city. Random errors (Eq. 17, Mann and Lenschow, 1994) are calculated between 20–40% depending on the entrainment to surface flux ratio. In order to account for delay times between instantaneous wind speed and VOC concentration measurements we performed a cross correlation analysis. As an example Fig. 2 shows the correlation coefficient (r) between vertical wind speed and toluene mixing ratios as a function of lag time for RF12. At its peak r exhibits a value of 0.3, which is in the range expected for a turbulent boundary layer (e.g. Mann and Lenschow, 1994). Random errors calculated from Eq. (5) are also plotted in Fig. 2 as horizontal lines and generally agree with the amount of variability obtained from the cross correlation analysis. The maximum correlation falls within the expected delay time window measured on the ground by spiking a VOC standard into the sampling inlet. The delay time between individual flights inferred from the correlation analysis was determined to be 1.5 ± 0.1 s.

We tested two independent methods for calculating turbulent toluene fluxes based on Eq. (3): The first method was conventional Fast Fourier Transformation (FFT), which computes an average flux over the entire flight leg. The second was wavelet transformation, which computes an instantaneous correlation over a chosen bandwidth. Wavelet analysis is increasingly used for turbulent flux calculations (e.g. Mauder et al., 2007 and references within). Here we implemented a wavelet transformation routine outlined by Torrence and Compo (1998) using the Morlet wavelet (Thomas and Foken, 2005). For more information on wavelet analysis used in atmospheric research we refer to Torrence and Compo (1998). Briefly, two advantages of wavelet transforms include that (1) it does not rely on the ergodic hypothesis and therefore does not require stationarity, and (2) it allows investigating time resolved spectral contributions to the measured flux. This makes wavelet analysis an attractive alternative for calculating covariances from airborne measurements, because it allows for investigating spatially heterogeneous surface emission patterns (e.g. Mauder et al., 2007).

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As an example Fig. 3 depicts the wavelet correlation analysis between toluene and GPS corrected vertical wind speed (vws) during RF12. The top panel depicts detrended toluene mixing ratios and vws over Mexico City. A two-dimensional wavelet spectrum between these variables is shown in the middle panel. Strong positive correlations – contributions to the measured toluene flux – are evident on time scales between 16–60 s, about 10 km after the airplane descended into the boundary layer over Mexico City. High correlation coefficients spatially coincide with the portion of the flight leg where the airplane flew North of Downtown across Mexico City.

The integral over all eddy contributions (y-dimension in Fig. 3, middle panel) is plotted in the lower panel and represents the spatially apportioned instantaneous toluene flux. The integral can be chosen for different timescales (bandwidths). As an example fluxes calculated for three different bandwidths (0.1–32 s, 32–64 s and 0.1–64 s) are shown. It can be seen that most of the contribution to the measured toluene flux is captured within the 0.1–32 s bandwidth. For further analysis we use the 0.1–64 s bandwidth, which should retain most of the turbulent flux contribution.

The integral in x-dimension (middle panel) results in a global wavelet spectrum which should be comparable to a conventional FFT spectrum. A comparison between the global wavelet and FFT spectrum is shown in Fig. 4. Also shown is the FFT spectrum for sensible heat ($w'T'$) sampled at 20 Hz. Due to the disjunct VOC sampling strategy, the spectral analysis for the toluene cross-covariance calculation is restricted to <0.2 Hz. The two scalars exhibit similar co-spectral peaks at $3e-2$ and $7e-2$ Hz, corresponding to spatial scales of 3–4 and 8–9 km. Toluene fluxes obtained by these two methods were 12.6 ± 3.8 mg/m²/h (wavelet) and 14.8 ± 4.4 mg/m²/h (FFT) for RF 12 and 18.8 ± 5.6 mg/m²/h (wavelet) and 15.5 ± 4.7 mg/m²/h (FFT) for RF8. Benzene fluxes were 3.7 ± 1.0 mg/m²/h (wavelet) and 4.0 ± 1.2 mg/m²/h (FFT) for RF12 and 6.3 ± 1.9 mg/m²/h (wavelet) and 5.0 ± 1.5 mg/m²/h (FFT) for RF8.

3.2 Uncertainty analysis

Several factors need to be considered when extrapolating measured aircraft fluxes (e.g. in the present case: $z/z_i=0.2$, $F_{0.2}$) to surface emissions. For slowly reacting species, such as toluene and benzene, entrainment fluxes (F_e) can be approximated by a linear flux relationship throughout the PBL (e.g. Faloon et al., 2005). The magnitude of entrainment and surface fluxes can be experimentally determined by measuring fluxes at several heights throughout the PBL and calculating the flux divergence (third term in Eq. 1). Since flux divergence measurements were not available, we estimate lower and upper limits of the bias that would arise for surface fluxes based on PBL growth measurements during this study (Shaw et al., 2006): We calculate dz_i/dt between 0.15–0.25 m/s. Taking typical measured concentration jumps dC across the PBL top of 3.5–7 ppbv for toluene and benzene, $F_{0.2}/F_s$ is calculated on the order of –19% to +5%. Using the best estimate F_s would be systematically 7% higher than $F_{0.2}$.

The influence of advection (second term in Eq. 1) can be estimated based on measured horizontal concentration gradients (dC/dx) and horizontal wind speeds (u), choosing a coordinate system where v equals zero. Typical values encountered for toluene during RF8 and 12 were $u=5$ – 10 m/s, and $dC=1$ ppbv; this leads to an advection flux of 1–2.5 mg/m²/h (e.g. 1 ppbv × 585 m × 5 m/s/30 000 m ~ 1.3 mg/m²/h) or 8–17% of $F_{0.2}$.

We discussed systematic and random errors associated with airborne flux measurements in Sect. 3.1. The systematic error for a subset of the total flight leg can be calculated according to Mann and Lenschow (1994):

$$se = b \frac{z_i z^{0.5}}{1 / \left(\frac{1}{L_{rm}} - \frac{1}{L} \right)}, \quad (6)$$

where b is 1.2 (for $z/z_i=0.2$), z_i is the height of the PBL, z is the flight altitude, L is the length of the flight leg and L_{rm} is a sub-length of the total flight leg L . For $L_{rm}=10$ – 15 km and $L=30$ km, the systematic error would then lie between 5–10%, depending

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over which spatial scales the flux was averaged.

In summary flux measurements performed at $z/z_i=0.2$ ($F_{0.2}$) likely yield a lower limit of actual surface fluxes F_s . The best estimate of this lower limit is 13% (range: -6%–30%) with an associated random error of 28%.

5 3.3 BTEX (Benzene/Toluene/Ethylbenzene/m, p, o-Xylenes) sources in Mexico City

3.3.1 Comparison with CAM04 and CAM01 emission inventory

Figure 5 (left panel) depicts the spatial distribution of the CAM (2004) emission inventory for toluene, which was adjusted according to Lei et al. (2007). As reference we plot the C-130 flight track along with horizontal wind vectors on top of the emission map.

10 The middle panel shows the same flight track color coded by the instantaneous toluene flux (average of RF8 and Rf12) on top of a digital elevation map. The size of the circles represent 90% of the flux footprint calculated according to Weil and Horst (1992) using instantaneous horizontal wind speed measurements. The figure also shows the 2006 supersite (T0) and the 2003 and 2006 flux sites (T-1). High toluene flux contributions
15 are evident above the International Airport and an industrial area in the northern part (indicated by circles). The toluene/benzene flux ratio (Fig. 5 right panel) indicates a particularly high toluene contribution over the industrial area (e.g. up to ~15; mean ratio of 10). This is comparable to a concentration ratio of 8.8, which was measured in the vicinity during a study in 2003 (Velasco et al., 2007). These observations point
20 towards distinct industrial pollution sources of aromatic compounds. It is noted that the flux ratio between two compounds can pinpoint pollution fingerprints much more precisely than concentration ratio measurements because in a well mixed boundary layer the spatial variation of mixing ratios is significantly smaller than the spatial variation of surface emissions. Concentration ratios can therefore be seen more like a city average
25 value. The flux ratio is confined within the flux footprint (e.g. 90% of the flux contribution), which, at flight levels flown during this study ($z/z_i=0.2$), was typically between 0.5–2 km.

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In order to compare toluene emissions on a more representative modeling scale we show average fluxes over the entire flight leg together with the CAM01 (2001) (West et al., 2001) and the adjusted CAM04 (2004) (Lei et al., 2007) emission inventory during times when the C-130 flew over the city in Fig. 6. Emissions inferred from the CAM01 and the adjusted CAM04 inventory were confined within 192 997–194 985° N and 992 667–988 893° W. Mean fluxes measured over this industrialized part of Mexico City are higher than predicted by both emission inventories (e.g. 50% for the adjusted CAM04 and almost a factor of 3 for the CAM01 inventory). This does not entirely come as a surprise as discrepancies between reported and actual Industrial emissions have been found previously in other North American cities. For example emissions from counties along the Houston ship channel, TX, USA, were significantly underestimated by inventories compiled for 2000 (Wert et al., 2003). We also find evidence of evaporative losses, which typically contain a higher toluene fraction and have to be considered as important urban pollution sources for aromatic compounds. For example high toluene fluxes and a high toluene/benzene flux ratio (3–5) were observed over the International airport. Aircraft engines burn fuel efficiently resulting in low toluene/benzene ratios (e.g. 0.6 g/g; Gerstle et al., 2002). Emissions from jet aircraft alone are therefore not likely to explain these toluene emission enhancements. Measured and modeled toluene mixing ratios for RF8 (22 March 2006: 19:00–20:00 UTC) are shown in Fig. 7. The average toluene mixing ratio measured in the PBL ($z/z_i=0.2$) along the flight track over the city was 5.6 ± 4.0 ppbv; it was 8.2 ± 1.5 ppbv when flying over T0. For comparison time-interpolated measurements during RF8 on the ground at T0 showed values of 15.8 ± 0.4 ppbv. Mixing ratios on the ground up to 70 ppbv were reported at this site. WRF-CHEM model runs based on the CAM01 emission inventory capture the extent of the MC plume well, but underestimate the average mixing ratios across the plume (3.1 ± 0.6 ppbv). Better agreement with measurements is achieved when using the adjusted CAM04 emission inventory (e.g. 4.1 ± 1.6 ppbv), but the extent and location of the concentration plume is shifted to the east. While detailed analysis of the exact spatial distribution of the concentration plume might go beyond what is reasonable for

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a large scale regional model, our measurement/model concentration comparison nevertheless points towards an underestimation of toluene emissions in current emission inventories. From direct flux measurements (Fig. 6) we find that this underestimation could be as large as 50% and up to a factor of 3 for the adjusted CAM04 and the CAM01 inventory respectively.

3.3.2 BTEX emission ratios

BTEX compounds, defined as benzene, toluene, ethylbenzene and the sum of o,p+m xylene, are important NMHC (non-methane hydrocarbons) for urban air chemistry. For the MCMA area depicted within the blue box in Fig. 1b their combined average mixing ratio was about 10% of the total observed NMHC mixing ratio, excluding oxygenated VOCs, which to a large extent are also produced photochemically. Due to their high reactivity, BTEX compounds accounted for ~30–36% of the total observed NMHC reactivity in this region in 2006.

On-road emissions are considered to be an important source for BTEX compounds in Mexico City. Zavala et al. (2006) derived yearly on-road emissions for toluene ($E_{\text{toluene}}=10\,100\pm 2200$ tons/year) and benzene ($E_{\text{benzene}}=4090\pm 850$ tons/year), which would result in an emission ratio of 2.5. From our airborne DEC flux measurements we obtain an average toluene/benzene flux ratio of 3.2 ± 0.5 . Localized enhanced toluene/benzene flux ratios were evident above the International airport and an industrial district in the northern part of the city (e.g. 10–15). We can test whether these flux ratios are consistent with observed concentration ratios: From an x/y-weighted regression between toluene and benzene we obtain an average concentration ratio of 3.9 ± 0.3 (RF1, RF6, RF8 and RF12). This is consistent with previous work published in Mexico City: Rogers et al. (2006) found an average concentration ratio of 4.0 based on ground based observations in different parts of Mexico City. Velasco et al. (2007) reported a toluene/benzene concentration ratio of 4.9 from urban sources at one particular site (T-1₂₀₀₃). The fact that emission ratios between VOCs should be consistent with observed concentration ratios can be used to gain additional information on

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VOC source distributions. For example typical emissions from fuel combustion yield a toluene/benzene ratio of 2.5. A high ratio (e.g. 3.9) inferred from our concentration measurements suggests additional toluene sources:

Biomass burning

5 It has been previously suggested that biomass burning activities, such as fires, residual waste burning, home cooking etc., could play a significant role for urban air quality in Mexico City (Yokelson et al., 2007; deCarlo et al., 2007). Typical toluene/benzene ratios inferred for biomass burning activities are 0.6 ± 0.3 (e.g. Yokelson et al., 2007; Merlet and Andreae, 2001), remarkably similar to those from urban burning activities:
10 0.6 ± 0.5 (Lemieux et al., 2004). Such low ratios from burning activities can therefore not account for the observed difference between ambient concentration measurements and typical fuel combustion emission profiles.

Other sources

Vega et al. (2000) investigated emission factors from various anthropogenic sources.
15 While their toluene/benzene ratio from exhaust emissions (2.5) was similar to that reported more recently by Zavala et al. (2006), they found high ratios (e.g. up to 7) from gasoline vapor depending on the refining grade of the fuel. Otherwise, direct industrial toluene emissions could also be responsible for high toluene to benzene concentration ratios. Velasco et al. (2007) reported a concentration ratio of 8.8 for an industrial district in the Northern part of Mexico City. From measurements obtained during this
20 study a toluene to benzene flux ratio of 10–15 was observed above the same industrial region (19°29'47.80"N 99°9'53.50"W). More recently Fortner et al. (2008) have observed distinct toluene plumes advected over T0, suggesting a significant influence from industrial sources. Together these observations show the potential importance
25 of evaporative fuel and industrial sources for aromatic compounds, in particular for toluene.

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In order to put these observations in formal context of a chemical tracer model, we relate observed concentration ratios (R^{observed}) to various source profile ratios according to

$$R_i^{\text{observed}} = \sum_{j=1}^N \alpha_{ij} S_j, \quad (7)$$

where α is the fraction of each ratio i from source j , N is the total number of sources and S the relative contribution of each source. Calculations based on Eq. (7) are restricted to C-130 measurements in the well mixed PBL in the MCMA (see Fig. 1b), where chemical VOC transformations exert a minor influence on the outcome of the regression model. Here we consider three lumped source categories: exhaust, evaporation+industrial and biomass burning emissions. We apportion benzene, toluene and C2-benzenes (the sum of xylenes and ethylbenzene) in Mexico City using a tracer method that includes methyl-tertiary-butyl-ether (MTB) and acetonitrile (CH_3CN). BTEX compounds can be emitted from all source categories. MTBE is a gasoline additive and is mainly released from fuel emissions. CH_3CN is thought to be a unique tracer for biomass burning, although there have been reports of direct cyanide (sum of HCN and CH_3CN) emissions from cars without a catalyst (VW AG, 1988). We account for this by including upper and lower limits of cyanide emission factors in our regression model. Table 2 lists a range of VOC/benzene, VOC/toluene and VOC/C2-benzene source ratios used to calculate relative source contributions (S_j). Exhaust and gasoline emissions were compiled from various reports (Vega et al., 2000; Zavala et al., 2006; Velasco et al., 2007) including measurements conducted during MIRAGE-Mex. Biomass burning emission ratios were based on Yokelson et al. (2007) and Lemieux et al. (2004) and direct observations of fire plumes near Mexico City. Relative source contributions for exhaust, evaporative+industrial and biomass burning emissions are summarized in Fig. 8. These were calculated for data collected in the MCMA area (blue box shown in Fig. 1b).

Based on this regression model we determine exhaust and evaporative emissions

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(gasoline and industry) as the largest contribution responsible for the abundance of aromatic compounds. The sum of these emissions account for 90–100% of the benzene, toluene and C2-benzenes loading; a fraction of 0–10% can be attributed to burning activities (e.g. forest fires, trash burning), mostly for benzene. Significant evaporative industrial and gasoline emissions inferred by the tracer model confirm direct observations of source signatures based on direct flux measurements (Sect. 3.3.1), which show distinct emission ratio enhancements over Mexico City. Among others, these emission hotspots corresponded to an industrial district in the North and the International airport. The large fraction of evaporative sources shows similarity with other VOC source categories. For example, leakage from liquefied petroleum gas (LPG) has previously been identified as a major contributor to the abundance of some alkanes (e.g. propane, i-butane, n-butane) (Blake and Rowland, 1995; Vega et al., 2000).

For benzene we find the highest contribution from exhaust emissions. A significant fraction of benzene and toluene can also come from evaporative fuel and industrial sources, which dominate emissions for C2-benzenes.

4 Conclusions

We show that VOC eddy covariance measurements on aircraft equipped with turbulent measurement capabilities are feasible using a Proton-transfer-reaction mass spectrometer. It is demonstrated that spatially variable distributions of VOC emissions, which could pose a problem for surface sites, can be assessed from aircraft measurements. Fluxes of toluene and benzene are compared with two emission inventories and indicate that midday surface emissions are most likely underestimated. Flux ratios between toluene and benzene show distinct industrial pollution sources over the city. Based on a tracer model, exhaust, industrial and fuel evaporative emissions are determined as major sources for BTEX compounds in the Mexico City Metropolitan Area (MCMA). We find that biomass burning contributes a comparably small amount to the MCMA BTEX mix (~10%, mostly for benzene). Future airborne flux measurements

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will help reducing uncertainties of anthropogenic and biogenic emission maps used in regional and global CT models. This will ultimately lead to improved simulations of tropospheric chemistry and a better understanding of air pollution management.

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Table 1. VOCs measured by PTRMS, TOGA and WAS.

VOC	Sensitivity [cps/ppbv]	Detection Limit ^a [pptv]	Slope PTRMS/TOGA	Slope PTRMS/WAS
Methanol	41	230	1.20	NA
Acetone+Propanal	70	40	1.04	NA
Acetonitrile	59	34	0.82	NA
Benzene	37	24	1.21	0.88
Toluene	21	55	1.10	0.80

^a For a 5 s integration time.[Title Page](#)[Abstract](#)[Introduction](#)[Conclusions](#)[References](#)[Tables](#)[Figures](#)[I◀](#)[▶I](#)[◀](#)[▶](#)[Back](#)[Close](#)[Full Screen / Esc](#)[Printer-friendly Version](#)[Interactive Discussion](#)

Table 2. Typical range of emission ratios for three different source categories.

	Emission Ratios (g/g)	Exhaust a,b,c,d	Evaporative+Industrial b,c,e,f	Biomass Burning g,h,i
Benzene	toluene/benzene	2.1–2.5	7.0–10.0	0.4–0.6
	MTBE/benzene	0.7	1.8–3.5	0.0
	CH ₃ CN/benzene	0.06–0.10	0.0	0.5–0.8
Toluene	benzene/toluene	0.4–0.5	0.10–0.14	1.6–2.5
	MTBE/toluene	0.2	0.3–0.6	0.0
	CH ₃ CN/toluene	0.03–0.06	0.0	0.7–0.9
C2-benzenes	benzene/C2-benzenes	0.2–0.4	0.3–0.6	2.6–6.0
	MTBE/C2-benzenes	0.1–0.2	0.8–1.2	0.0
	CH ₃ CN/C2-benzenes	0.03–0.06	0.0	0.8–1.5

^a Zavala et al. (2006).

^b Vega et al. (2000).

^c Canister measurements made by TOGA during this study.

^d VW AG (1988).

^e EC fluxes over industrial area made during this study.

^f Velasco et al. (2007).

^g Yokelson et al. (2007).

^h Lemieux et al. (2004).

ⁱ Andreae and Merlet (2001).

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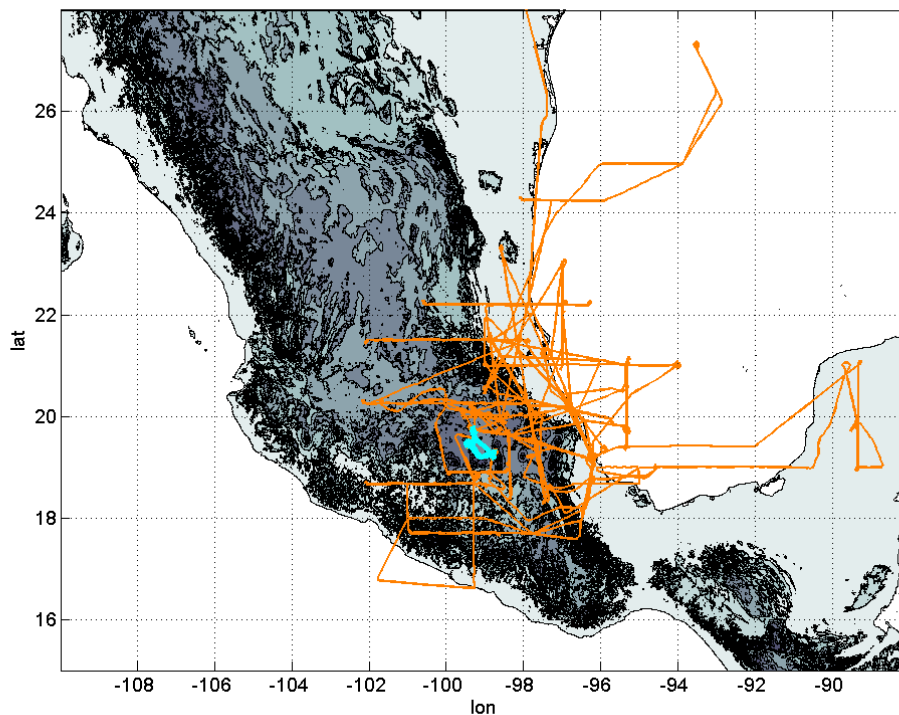


Fig. 1a. Map showing all C-130 flight tracks plotted on top of a topographical map of Mexico. The blue segment highlights flight tracks near and in the Mexico City Metropolitan Area (MCMA).

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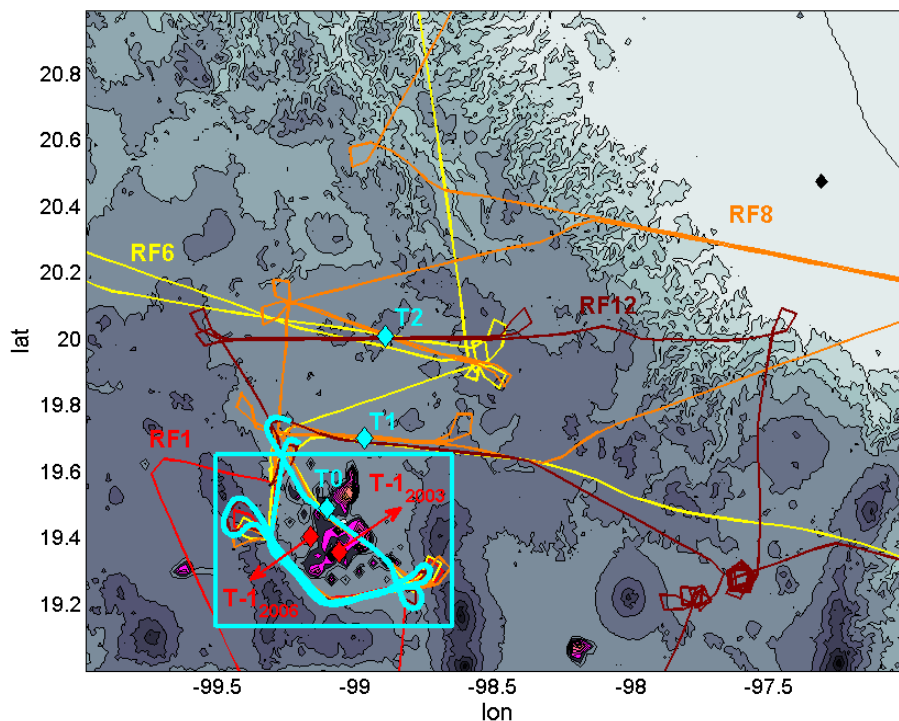


Fig. 1b. Flight tracks flown during research flights 1, 6, 8 and 12. Ground super sites (T0,T1 and T2) are depicted by blue symbols. Ground based urban flux measurements (T-1) for the 2003 and 2006 studies are shown by red symbols. The blue box highlights the area which is focus of the present study.

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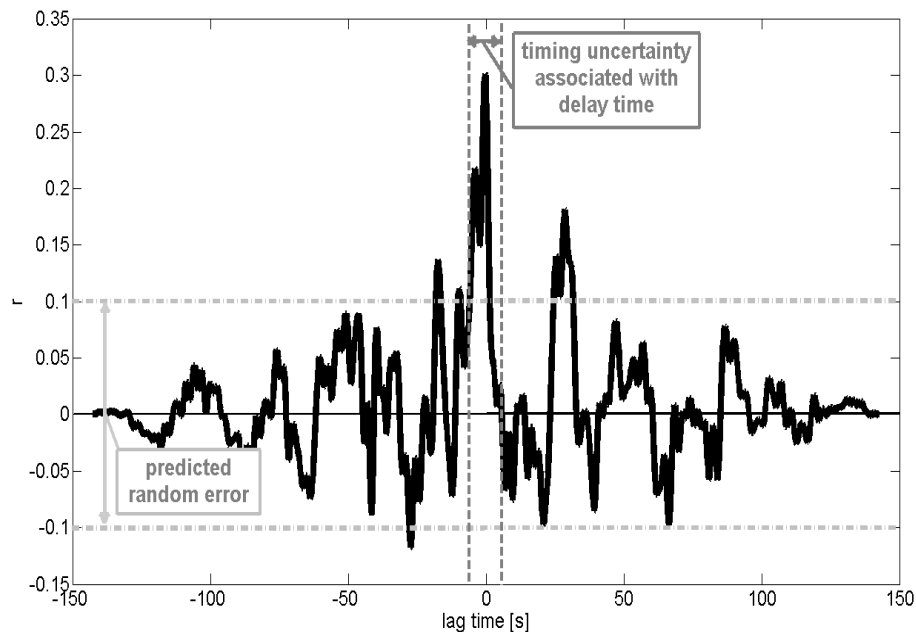


Fig. 2. Correlation coefficient (r) between toluene mixing ratios and vertical wind speed plotted as a function of lag time. Vertical lines indicate the lag time window measured on the ground. Horizontal lines depict the estimated random error associated with the flux measurement.

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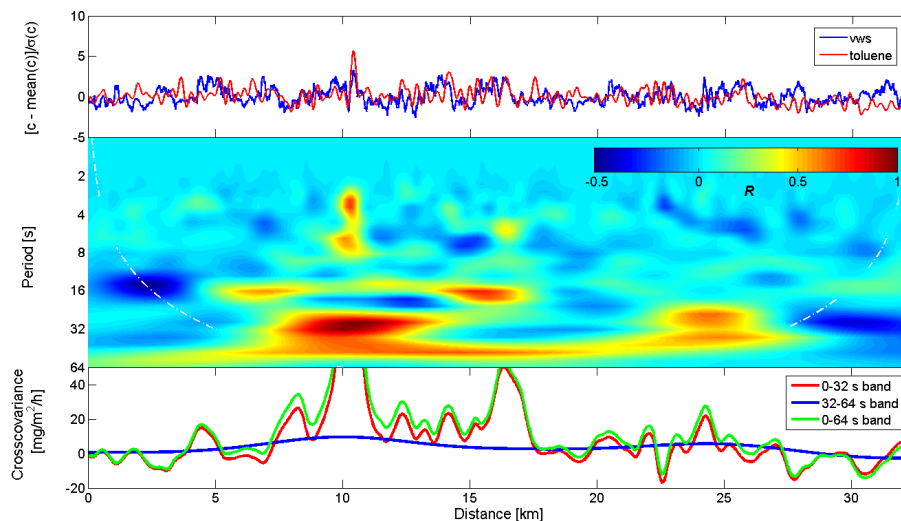


Fig. 3. Wavelet crosscorrelation along the PBL flight leg over MCMA: Top panel: De-trended vertical wind speed (blue) and toluene (red) for RF12. Middle panel: Correlation coefficient (R) obtained from the wavelet cross spectrum between toluene and vertical wind speed. Dashed curve indicates the cone of influence. Bottom panel: Instantaneous toluene flux for different bandwidths.

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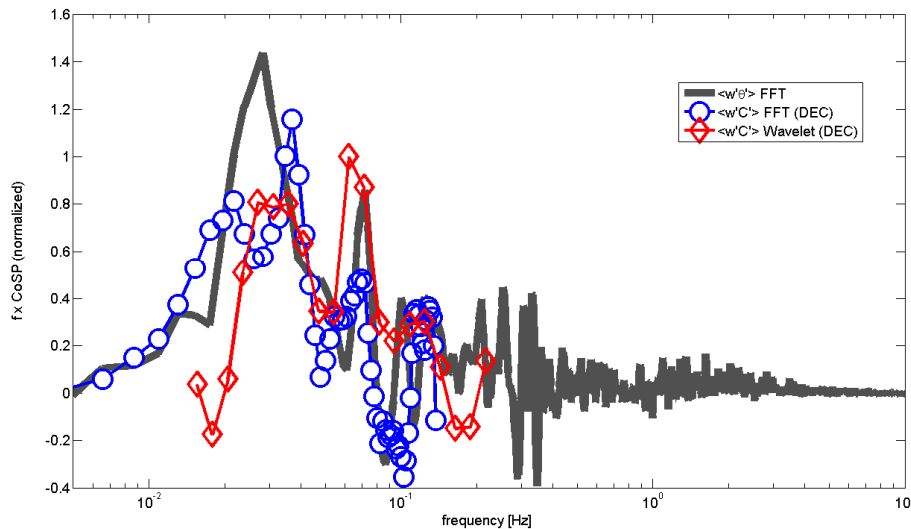


Fig. 4. FFT Cospectra between vertical wind speed and temperature ($w'T'$, grey) and toluene and vertical windspeed (blue circles) for RF12. The wavelet spectrum for toluene and vertical windspeed is depicted by red diamonds.

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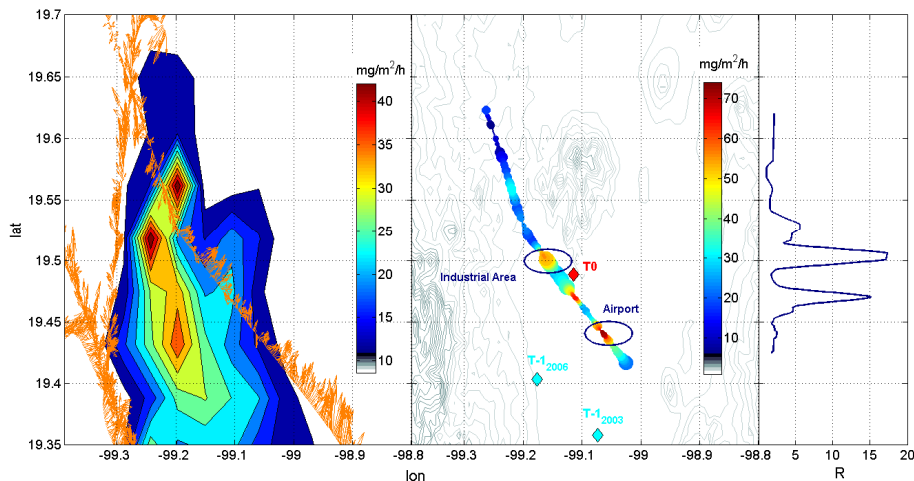


Fig. 5. Comparison between the adjusted CAM04 emission map (left panel) with instantaneous flux measurements across the city (middle panel). The feather plot on top of the emission map (left panel) depicts the C-130 flight track across the city and shows the instantaneous horizontal wind speed vector. The middle panel shows toluene fluxes across MC on top of an elevation map. The right panel shows the ratio between instantaneous toluene to benzene fluxes measured across the city for RF12.

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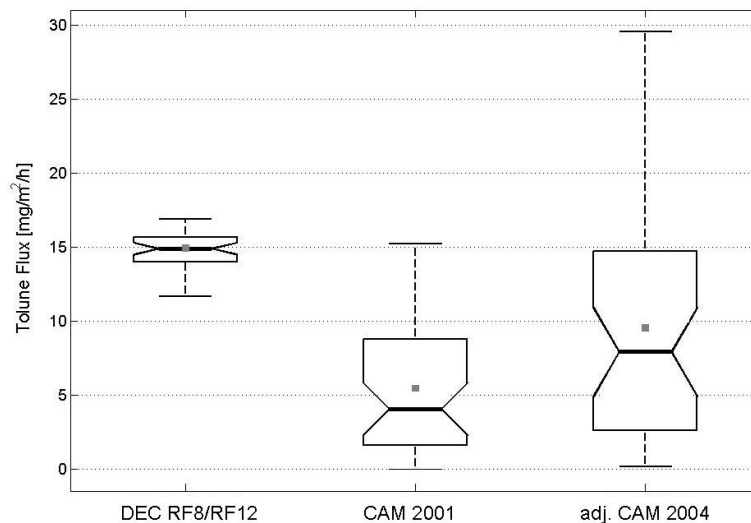


Fig. 6. Statistical boxplot showing toluene fluxes measured for RF8 and RF12 and toluene emissions inferred from the CAM 2001 and adjusted CAM 2004 emission inventories. The statistical box plot depicts lower quartile, median and upper quartile (vertical lines of notched box), mean (gray squares connected with dashed line) and whiskers (include 95% of data).

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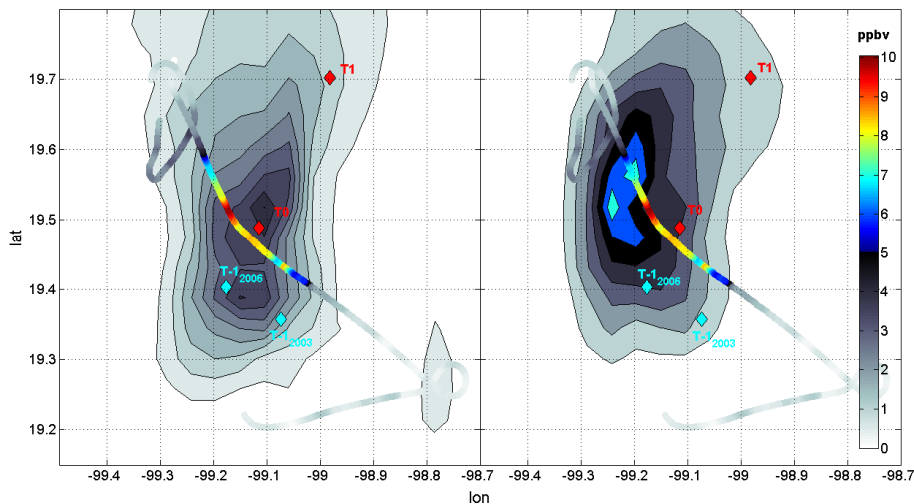


Fig. 7. Comparison between toluene mixing ratios measured during RF8 and modeled with WRF-CHEM using CAM 2001 (left panel) and the adjusted CAM 2004 (right panel) emission inventory.

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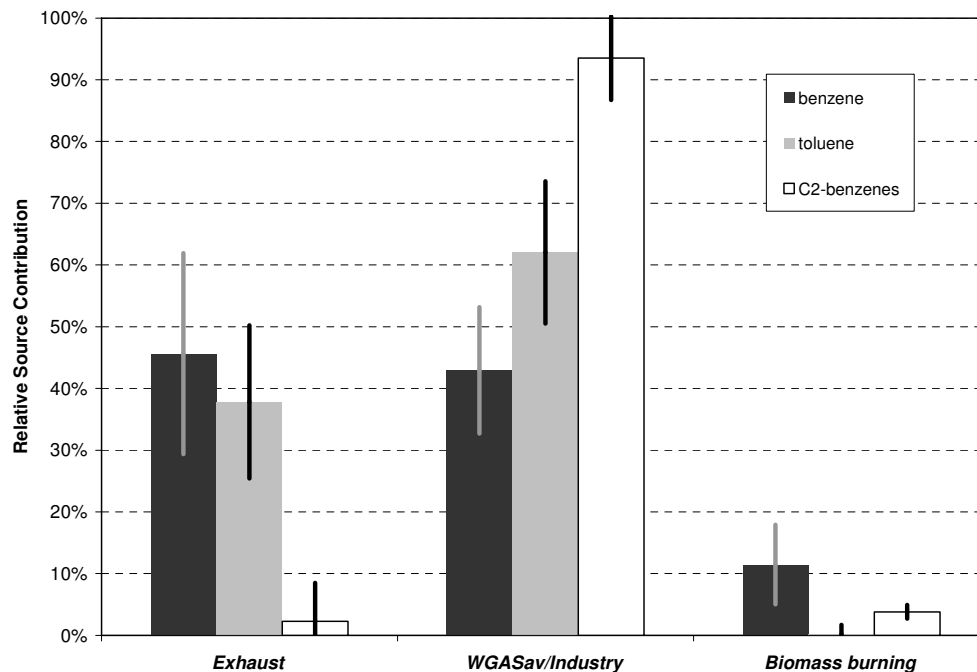


Fig. 8. Relative source contributions for benzene, toluene and C2-benzenes for 3 source categories: exhaust, evaporative fuel plus industrial (WGASav/Industry) and biomass burning.

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