

# Influence of modelled soil biogenic NO emissions on related trace gases and the atmospheric oxidizing efficiency

J. Steinkamp<sup>1</sup>, L. N. Ganzeveld<sup>2</sup>, W. Wilcke<sup>3</sup>, and M. G. Lawrence<sup>1</sup>

<sup>1</sup>Department of Atmospheric Chemistry, Max-Planck-Institute for Chemistry, Mainz, Germany

<sup>2</sup>Department of Environmental Sciences, Chairgroup Earth System Sciences, Wageningen University and Research Centre, Wageningen, Netherlands

<sup>3</sup>Geographic Institute, Johannes Gutenberg University, Mainz, Germany

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Correspondence to: J. Steinkamp (steinkam@mpch-mainz.mpg.de)

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## Abstract

The emission of nitric oxide (NO) by soils (SNOx) is an important source of oxides of nitrogen ( $\text{NO}_x = \text{NO} + \text{NO}_2$ ) in the troposphere, with estimates ranging from 4 to 21 Tg of nitrogen per year. Previous studies have examined the influence of SNOx on ozone ( $\text{O}_3$ ) chemistry. We employ the ECHAM5/MESSy earth system model to go further in the reaction chain and investigate the influence of SNOx on lower tropospheric  $\text{NO}_x$ ,  $\text{O}_3$ , peroxyacetyl nitrate (PAN), nitric acid ( $\text{HNO}_3$ ), the hydroxyl radical (OH) and the lifetime of methane ( $\tau_{\text{CH}_4}$ ). We show that SNOx is responsible for a significant contribution to the  $\text{NO}_x$  mixing ratio in many regions, especially in the tropics. On the other hand in some regions SNOx has a negative feedback on the lifetime of  $\text{NO}_x$  through  $\text{O}_3$  and OH, which results in regional increases in the mixing ratio of  $\text{NO}_x$  despite lower total emissions in a simulation without SNOx. Furthermore, the concentration of OH is substantially increased due to SNOx, resulting in an enhanced oxidizing efficiency of the global troposphere, reflected in a  $\sim 10\%$  decrease in  $\tau_{\text{CH}_4}$  due to soil NO emissions.

## 1 Introduction

Nitric oxide (NO) in the soil is produced by the microbial processes of nitrification and denitrification (Firestone and Davidson, 1989). The NO emission originates from a natural pool of nitrogen and a fraction from fertilizer application (Yienger and Levy II, 1995; Stehfest and Bouwman, 2006). The estimates of NO emitted yearly by soils (hereafter called SNOx) ranges from 4 to 21 Tg(N) (Yienger and Levy II, 1995; Davidson and Kingerlee, 1997, and references therein). NO reacts rapidly with other atmospheric compounds, establishing an equilibrium between NO and nitric dioxide ( $\text{NO}_2$ ). These two species are frequently referred to the oxides of nitrogen ( $\text{NO}_x$ ). SNOx is topped by the anthropogenic combustion of fossil fuels ( $20\text{--}24 \frac{\text{Tg(N)}}{\text{yr}}$ ) (Holland et al., 1999) and is comparable to the production of  $\text{NO}_x$  from lightning and biomass burning, but especially in remote continental regions of the mid- and low-latitudes SNOx is the dom-

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inant source of  $\text{NO}_x$ . Through reactions, deposition and stomatal uptake directly within the vegetation layer not all  $\text{NO}$  emitted by the soil escapes the canopy layer as  $\text{NO}_x$  (Yienger and Levy II, 1995; Ganzeveld et al., 2002b). In this work SNO<sub>x</sub> refers to the flux from the vegetation to the atmosphere. The fraction of  $\text{NO}_x$  that reaches the atmosphere reacts as a catalyst for production of ozone ( $\text{O}_3$ ), one of the greenhouse gases. This  $\text{O}_3$  production is driven by the oxidation of carbon monoxide (CO) and volatile organic compounds (VOC), if the concentration of  $\text{NO}$  is higher than about 5–30  $\frac{\text{pmol}}{\text{mol}}$  (Brasseur et al., 1999). Atmospheric  $\text{NO}_x$  is also involved in the production of the hydroxyl radical (OH), which is responsible for the oxidation and depletion of methane ( $\text{CH}_4$ ), another greenhouse gas. Beyond these climate related issues, high  $\text{NO}_x$  mixing ratios also have a direct impact on human health.  $\text{NO}_x$  is removed from the atmosphere by reaction with hydroxyl radicals (OH) or oxidation to dinitrogen pentoxide ( $\text{N}_2\text{O}_5$ ) and subsequent deposition as nitric acid ( $\text{HNO}_3$ ). It can also react with organic tracers to form peroxy nitrates, mainly peroxyacetyl nitrate (PAN), which, once it is lifted to higher altitudes, can be transported over large distances releasing  $\text{NO}_x$  when it is transported back downward again.

Previous model studies of the influence of SNO<sub>x</sub> on atmospheric chemistry mainly focused either on the  $\text{NO}_x$  source itself, or on  $\text{O}_3$ , mostly on a regional scale. Jaeglé et al. (2005) examined the global partitioning of  $\text{NO}_x$  sources using inverse modelling and the space-based  $\text{NO}_2$  column derived by GOME (Global Ozone Monitoring Experiment). Their a posteriori SNO<sub>x</sub> ( $8.9 \frac{\text{Tg(N)}}{\text{yr}}$ ) is 68% greater than their a priori SNO<sub>x</sub> ( $5.3 \frac{\text{Tg(N)}}{\text{yr}}$ ). Based on this, Jaeglé et al. (2005) suggest that the influence of SNO<sub>x</sub> on background  $\text{O}_3$  could be underestimated in current chemistry transport models (CTMs). Bertram et al. (2005) come to a similar conclusion by inverse modelling using another satellite sensor (SCIAMACHY) above the western United States, computing an underestimation of 60%. Delon et al. (2007) modelled higher  $\text{O}_3$  concentrations with higher SNO<sub>x</sub> above Western Africa. For Europe, Simpson (1995) found that SNO<sub>x</sub> hardly has any influence on controlling the  $\text{O}_3$  mixing ratio. Isaksen and Hov (1987) already investigated the influence of changes in the emission intensity of different relevant trace gases

on the oxidizing efficiency through an increase in OH concentration with increased  $\text{NO}_x$  emissions, but they neglected SNOx.

In this study, we take these analyses a step further and follow the reaction chain from SNOx through  $\text{O}_3$  and OH to its global influence on the oxidizing efficiency of the atmosphere. To do so, we compare two model runs with a state-of-the-art global 3-D global chemistry climate model. One is a simulation with all relevant emissions and reactions (BASE), and the second simulation is without SNOx (NOBIONO = “No biogenic NO”). We expect a considerable influence of SNOx on the mixing ratios and distribution of related global tropospheric trace gases ( $\text{NO}_x$ , PAN,  $\text{HNO}_3$ ,  $\text{O}_3$  and OH). Furthermore the global oxidizing efficiency, indicated by the lifetime of  $\text{CH}_4$  ( $\tau_{\text{CH}_4}$ ), is expected to decrease ( $\tau_{\text{CH}_4}$  increases) if we exclude  $\text{NO}_x$  emission from soils.

In the following section we briefly describe the model setup. We then compare the relevant tracer mixing ratios from the BASE simulation versus the NOBIONO simulation. In the final section we present our conclusions and outlook.

## 2 Model description and setup

For this study the Modular Earth Submodel System version 1.2 (MESSy) coupled to the general circulation model ECHAM5 is employed. MESSy connects, through a standardized interface, submodels for different processes with bidirectional feedbacks (Jöckel et al., 2005, 2006). The meteorology for these simulations is driven by sea surface temperature (SST) from the AMIP IIb dataset (Taylor et al., 2000). The calculation of SNOx in the BASE simulation is based on the algorithm of Yienger and Levy II (1995), which is the most widely used SNOx algorithm in CTMs (Ganzeveld et al., 2002a; Jaeglé et al., 2005; Delon et al., 2007). This calculation is performed in the submodel EMDEP (Ganzeveld et al., 2006).  $\text{NO}_x$  produced by lightning is calculated in the submodel LNOX ( $2.2 \frac{\text{Tg(N)}}{\text{yr}}$ ). The remaining sources of  $\text{NO}_x$  ( $49 \frac{\text{Tg(N)}}{\text{yr}}$ ) are read in from the offline EDGAR database (Olivier et al., 1994) by the submodel OFFLEM (Kerkweg et al., 2006). Other relevant emissions are calculated either by the EMDEP

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or OFFLEM submodel.

A model spinup time of eleven months (January–November 1994) was chosen and the data of the period December 1994–December 1995 is analysed here. To achieve an identical meteorology of both simulations feedback through trace gases and water vapor is switched off. Table 1 recapitulates the setup of the two simulations.

### 3 Results and discussion

The emission of NO from soils in the BASE simulation accounts for 15% of the total annual global NO emissions (Table 2). The interannual variability of SNO<sub>x</sub> is low in the model (Steinkamp, 2007). The largest emissions are calculated for tropical regions. During JJA there are some exceptions further north in northern America, Europe and north-eastern China. These are fertilizer induced emissions in agricultural regions (Fig. 1 and Table 2).

The data is analysed by season with a focus on the winter and summer season. There is a notable seasonal variation with larger SNO<sub>x</sub> in the summer period of each hemisphere and with a larger contribution of SNO<sub>x</sub> to the total NO emissions during the northern hemispheric spring and summer (Table 2). The first point can be explained by the temperature dependence and the second one by the greater landmasses in the Northern Hemisphere. In the northern mid-latitudes SNO<sub>x</sub> plays a less important role relative to other NO<sub>x</sub> emissions, except during the JJA period.

#### 3.1 Influence of SNO<sub>x</sub> on related trace gases

The column mean mixing ratios of NO<sub>x</sub>, PAN, HNO<sub>3</sub> and O<sub>3</sub> and the column mean concentration of OH in the gridcells (weighted by the air mass in the gridcells) in the lower troposphere (below 500 hPa; hereafter “LT”) from the BASE simulation are compared with the values from the NOBIONO simulation.

As expected, in the LT the difference between the NO<sub>x</sub> column mean mixing ratio in the NOBIONO simulation versus the BASE simulation is well-correlated with SNO<sub>x</sub> in

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all regions (Table 3). A low correlation is computed for the Northern Hemisphere during DJF, as could be expected due to the small SNO<sub>x</sub> compared to the anthropogenic emissions.

In the low-latitudes and in the northern mid-latitudes the correlation between SNO<sub>x</sub> and the difference in the PAN column mean mixing ratio of the two simulations is low (Table 3). This is likely due to the stronger dependence of PAN formation on VOC mixing ratios than on NO<sub>x</sub> mixing ratios (Singh et al., 1986). In contrast, there is a much better correlation in the southern mid-latitudes between the difference in the LT PAN column mixing ratio and SNO<sub>x</sub>. This suggests a dominating role of NO<sub>x</sub> in the formation of PAN in the mid-latitudes of the Southern Hemisphere. This hypothesis is partially confirmed by the calculation of the correlation coefficient of the difference in the PAN column mixing ratio with SNO<sub>x</sub> up to a lower altitude (750 hPa). R<sup>2</sup> increases slightly for the northern mid- and low-latitudes and stays nearly the same for the southern mid-latitudes (Table 3). In the low-latitudes, convective downdrafts and subsiding airmasses, combined with the strong temperature dependence of the decomposition of PAN decreases the correlation.

The correlation between SNO<sub>x</sub> and the difference in the LT O<sub>3</sub> column mean mixing ratio is lower than for NO<sub>x</sub>. This is partly due to the longer lifetime of O<sub>3</sub>, which is better mixed in the LT. Furthermore the production of O<sub>3</sub> is not only determined by the NO<sub>x</sub> mixing ratio, but also by the concentration of VOC. The correlation of the OH column mean concentration difference in the LT with SNO<sub>x</sub> is similar to O<sub>3</sub>. OH is a very short lived tracer, whose production depends on the one hand on the photolysis of O<sub>3</sub> and the water vapor concentration and on the other hand on the reaction of NO with HO<sub>2</sub> in the troposphere. This results, depending on the dominating reaction, in a higher or lower correlation of the OH column concentration difference versus SNO<sub>x</sub> than the O<sub>3</sub> column mixing ratio difference versus SNO<sub>x</sub>.

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### 3.1.1 NO<sub>x</sub>

The global mean mixing ratio of NO<sub>x</sub> in the LT during DJF decreases by 6% in the NOBIONO simulation compared to the BASE simulation. During JJA it decreases by 12%. In both cases the decrease in the mixing ratio is less than the contribution of SNO<sub>x</sub> (12% and 17% respectively). The maximum decrease is 78% in DJF and 68% in JJA, while the maximum absolute decreases in the DJF and JJA periods are 139 and 209  $\frac{\text{pmol}}{\text{mol}}$ , respectively (figures with absolute differences can be found in the supplement: <http://www.atmos-chem-phys-discuss.net/8/10227/2008/acpd-8-10227-2008-supplement.pdf>).

Interestingly, during DJF the mixing ratio above large parts of the Northern Hemisphere even increases by up to 3.5% (Fig. 2a), with the largest absolute increase of 15.2  $\frac{\text{pmol}}{\text{mol}}$  above Europe. In the JJA period the maximum relative increase of 4.8% is larger than in the DJF period, but the maximum absolute difference is only 3.0  $\frac{\text{pmol}}{\text{mol}}$  (Fig. 2b).

A similar result has been noted for model sensitivity simulations with and without NO<sub>x</sub> from lightning (Stockwell et al., 1999; Labrador et al., 2005), in which a decrease in near surface NO<sub>x</sub> mixing ratios was computed for similar regions with increasing production of NO<sub>x</sub> by lightning. Although NO<sub>x</sub> produced by lightning is formed in the free troposphere and SNO<sub>x</sub> originates from the surface, we achieve comparable results with SNO<sub>x</sub>. To explain why the NO<sub>x</sub> mixing ratio decreases less than the relative decrease in the emission of the NOBIONO simulation compared to the BASE simulation, and why it increases during the DJF period in large areas in the Northern Hemisphere, the feedback through O<sub>3</sub> and OH has to be taken into account. Stockwell et al. (1999) assumed that the general increase in O<sub>3</sub> with lightning NO<sub>x</sub> causes an increase in OH. This OH reduces the lifetime of NO<sub>x</sub> ( $\tau_{\text{NO}_x}$ ) through Eq. (1) above regions with high non-lightning NO<sub>x</sub> sources. Labrador et al. (2005) showed that the conversion to HNO<sub>3</sub> via N<sub>2</sub>O<sub>5</sub> also contributes to the shorter  $\tau_{\text{NO}_x}$  (Eq. 2) with higher NO<sub>x</sub> emissions.



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Similarly we find that without SNO<sub>x</sub>, O<sub>3</sub> and OH levels decrease, resulting in enhanced  $\tau_{\text{NO}_x}$ , and due to Eqs. (1) and (2) the NO<sub>x</sub> mixing ratio increases in regions with low SNO<sub>x</sub>. The changes in HNO<sub>3</sub>, O<sub>3</sub> and OH related to this are discussed in the following sections.

In the vertical direction the strongest effects of SNO<sub>x</sub> are simulated near the surface (DJF: 53%, JJA: 42%), and a decrease of up to 10 to 25% for higher altitudes in the zonal mean is calculated if SNO<sub>x</sub> was switched off (Fig. 3). The effect of convective transport to higher altitudes has a stronger influence on the difference in the total burden between 500 and 250 hPa during DJF (relative: 11.3%, absolute: 1.85 Gg) than during JJA (relative: 6.5%, absolute: 1.22 Gg). This is because the main regions, where the convective transport is most effective, are in the Southern Hemisphere especially the Amazon Basin and the southern tropics of Africa (not shown).

### 3.1.2 PAN

The LT PAN mixing ratio decreases globally by 4% during DJF and 8% during JJA without SNO<sub>x</sub>. In both periods the PAN mixing ratio decreases nearly everywhere above the continents (Fig. 4). Above the tropical oceans, especially during JJA, there is a high relative but a negligible absolute increase in the PAN mixing ratio associated with a decrease in SNO<sub>x</sub>. As mentioned above, the formation of PAN in the northern mid- and low latitudes relies more on VOC than on NO<sub>x</sub>, but more on NO<sub>x</sub> in the southern mid-latitudes. This explains the larger decrease during DJF than during JJA. There is

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also no increase of PAN in the Northern Hemisphere during DJF despite higher  $\text{NO}_x$  mixing ratios, which confirms a dominating role of VOC in PAN formation.

Interestingly, in the upper troposphere between 500 hPa and 250 hPa the largest decrease in the PAN mixing ratio is during DJF (7%), whereas it is 5% during JJA. In the zonal mean of the relative difference in PAN mixing ratio with and without SNOx (Fig. 5), the effect of convective transport in the lower latitudes is more effective during DJF than during JJA. At the higher altitudes PAN does not increase anymore, due to its longer lifetime resulting in better mixing.

The differences in the PAN mixing ratio should be interpreted with caution, because the model generally overestimates its levels compared to observations (Jöckel et al., 2006), though this may improve with a new isoprene oxidation scheme (Butler, personal communication).

### 3.1.3 $\text{HNO}_3$

The global LT mixing ratio of  $\text{HNO}_3$  decreases by 16% (DJF) and 15% (JJA) without SNOx. The greatest decrease occurs above continental regions of the low-latitudes and in the summer months in the Northern Hemisphere (Fig. 6). The larger decrease in the mixing ratio of  $\text{HNO}_3$  compared to the decrease of  $\text{NO}_x$  mixing ratio is because the formation of  $\text{HNO}_3$  is not only determined by the  $\text{NO}_x$  mixing ratio, but also relies on the mixing ratios of  $\text{O}_3$  and OH, which also decrease, as discussed in the following sections.

Nitric acid is mainly deposited on aerosol particles, taken up by cloud water or directly deposited on the earth's surface. The deposition of  $\text{HNO}_3$  is decreased by 14% throughout the year without SNOx. During DJF the decrease is 12% and during JJA it is 15%.

### 3.1.4 $\text{O}_3$

The mixing ratio of  $\text{O}_3$  in the NOBIONO simulation compared to the BASE simulation in the LT decreases by 6% in both seasons, with the greatest decline above the conti-

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nents (Fig. 7). The maximum relative decrease during DJF is 37% and during JJA it is 28%. The maximum absolute decrease ( $17.6 \frac{\text{nmol}}{\text{mol}}$ ) occurs during DJF above Australia (Fig. 7a). In contrast to what was found for  $\text{NO}_x$ , there is no region with increasing  $\text{O}_3$  mixing ratios. The decrease of the  $\text{NO}_x$  mixing ratio in the LT during JJA (12%) is less effective in reducing the  $\text{O}_3$  mixing ratio than the decrease in the  $\text{NO}_x$  mixing ratio during DJF (6%). This is because the formation of  $\text{O}_3$  through  $\text{SNO}_x$  competes with other strong sources of  $\text{NO}_x$  during JJA in the Northern Hemisphere, whereas  $\text{SNO}_x$  is relatively much more important for the formation of  $\text{O}_3$  during DJF in the Southern Hemisphere. Furthermore, as was noted above for the PAN formation in the Northern Hemisphere the simulated  $\text{O}_3$  production may be more dependent on VOC than  $\text{NO}_x$ .

In the zonal mean distribution (not shown) a similar pattern of the influence of convection can be seen as already discussed for  $\text{NO}_x$  and PAN. But due to the longer lifetime of  $\text{O}_3$  the relative change has only a maximum of 17% in DJF and 14% in JJA. These changes are smaller than for  $\text{NO}_x$  and are more evenly distributed above all latitudes and also in the vertical direction. In the vertical direction there is, as with the horizontal, no increase in the  $\text{O}_3$  mixing ratio throughout the troposphere.

### 3.1.5 OH

Excluding the contribution of  $\text{SNO}_x$ , the mean OH concentration declines by 10% during DJF and 8% during JJA in the LT. The largest relative decrease is 58% during DJF and 42% during JJA above the tropical land regions. During DJF the decrease is shifted to the southern tropics and to the northern tropics during JJA (Fig. 8). Note that during JJA an absolute increase above the Antarctic region is calculated, but the OH concentration here is less than  $1 \times 10^4 \frac{\text{molec}}{\text{cm}^3}$ .

The decrease is on the one hand induced directly by  $\text{NO}_x$  through Eq. (3), and on the other hand indirectly by the lower  $\text{O}_3$  mixing ratio, leading to less primary OH production, and therefore in a decrease of the OH concentration in the LT.



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The largest relative decrease in the zonal mean concentration of OH is 18% during DJF and 14% during JJA. This maximum of the relative decrease in the OH concentration without SNO<sub>x</sub> is nearly detached from the surface, despite the surface source of SNO<sub>x</sub> (Fig. 9). At the surface OH production is mainly related to the reaction of O(<sup>1</sup>D) with water, while at higher altitudes it depends more on the reaction of NO with HO<sub>2</sub> (Eq. 3). In the zonal mean the shift to the Southern Hemisphere during DJF is stronger than the shift during JJA to the Northern Hemisphere.

### 3.2 Influence of SNO<sub>x</sub> on the oxidizing efficiency

The oxidation of CO and VOC in the atmosphere is mainly driven by OH. As a measure for the oxidizing efficiency of the atmosphere,  $\tau_{\text{CH}_4}$  is calculated for both simulations according to Lawrence et al. (2001). The trend of monthly mean values is depicted in Fig. 10. The mean  $\tau_{\text{CH}_4}$  averaged for one year (December 1994 to November 1995) for the BASE simulation is 7.26 years and 7.93 years for the NOBIONO simulation, a 9.2% increase without SNO<sub>x</sub>. The maximum prolongation of 0.89 years occurs in December 1994 (10.3%) and January 1995 (10.4%).

The changes in  $\tau_{\text{CH}_4}$  are not equally distributed over the globe. In the southern and low-latitudes the relative influence is noticeably greater than in the northern latitudes (Fig. 11). This agrees with the smaller relative change in the OH concentration in the northern latitudes (Fig. 8). In the vertical direction, the relative changes are slightly larger above 500 hPa, even though the origin of SNO<sub>x</sub> is at the surface. Beginning from the surface source of SNO<sub>x</sub> and following the reaction chain from NO<sub>x</sub> over O<sub>3</sub> and OH in each step, the relative difference of our two simulations becomes smaller near the surface and larger at higher altitudes. This trend corroborates the larger relative change of the oxidizing efficiency at higher altitudes. However, only ~15% of the absolute amount of CH<sub>4</sub> in the troposphere is oxidized above 500 hPa (Lawrence et al., 2001).

Labrador et al. (2004) modelled a decrease of 15% in  $\tau_{\text{CH}_4}$  in a simulation with 5 Tg(N) NO<sub>x</sub> produced by lightning relative to one with no lightning NO<sub>x</sub>. Compared to this, SNO<sub>x</sub> is somewhat less effective in altering the oxidizing efficiency of the atmo-

sphere, which is interesting, given that CH<sub>4</sub> oxidation is more effective near the surface where SNO<sub>x</sub> is emitted, due to the strong temperature dependence of the reaction of OH with CH<sub>4</sub>. The change in the oxidizing efficiency due to lightning NO<sub>x</sub> is larger than due to SNO<sub>x</sub>, even though the total emission rate is lower. This is because at higher altitudes the NO:NO<sub>2</sub> ratio is greater, so that with more NO the NO<sub>x</sub> lifetime is not diminished as strongly as near the surface. Furthermore at higher altitudes more NO results in higher OH yields by reaction with HO<sub>2</sub>.

#### 4 Conclusions and outlook

The emission of NO from soils plays an important role for chemical reactions in the atmosphere. Lower global mean NO<sub>x</sub> mixing ratios without SNO<sub>x</sub> lead to lower global O<sub>3</sub> mixing ratios in the LT. The lower O<sub>3</sub> mixing ratios result in lower OH concentrations. This results in an enhanced lifetime of NO<sub>x</sub> in regions with other dominating sources of NO<sub>x</sub>. Hence the NO<sub>x</sub> mixing ratios increases in some regions, despite lower emissions when SNO<sub>x</sub> is neglected in our NOBIONO simulation. From this it follows, that although NO<sub>x</sub> is a short-lived tracer it indirectly influences chemical processes in regions with low SNO<sub>x</sub> through feedback with O<sub>3</sub> and OH. By following the reaction chain up to PAN and HNO<sub>3</sub>, we detected a dominating role of SNO<sub>x</sub> compared to VOC in the mid-latitudes of the Southern Hemisphere. Also by following the reaction chain (SNO<sub>x</sub>→NO<sub>x</sub>→O<sub>3</sub>→OH), the magnitude of relative effects are shifted step by step to higher altitudes in the troposphere.

Through reaction of NO with HO<sub>2</sub> SNO<sub>x</sub> is directly involved in the production of OH. SNO<sub>x</sub> also has, through O<sub>3</sub>, an indirect influence on OH production. With OH formed by SNO<sub>x</sub> through these pathways  $\tau_{\text{CH}_4}$  is decreased considerably, and the influence of SNO<sub>x</sub> on the tropospheric oxidizing efficiency is considerable, in the range of 10%.

The notable modelled influence of SNO<sub>x</sub> on directly and indirectly related trace gases shown in this work supports further efforts to improve the parameterization of SNO<sub>x</sub> in CTMs, as also proposed by Jaeglé et al. (2005).

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**Table 1.** Setup of the ECHAM5/MESSy model and used submodels.

Horizontal resolution	T42 ( $\sim 2.8^\circ \times 2.8^\circ$ )
Vertical resolution	L31 (up to 10 hPa)
Internal timestep	20 min
Timestep of output	5 h
Period of simulation	1994–1995

Used submodels	Calculation of	Literature ref.
CLOUD	Clouds and precipitation	Jöckel et al. (2006)
CONVECT	Convection	Tost et al. (2006b)
CVTRANS	Convective tracer transport	Tost (2006)
EMDEP <sup>a</sup>	Emission and deposition	Ganzeveld et al. (2006)
JVAL	Rates of photolysis	Jöckel et al. (2006)
LNOX	Lightning NO <sub>x</sub>	Tost et al. (2007)
MECCA	Chemical atmospheric reactions <sup>b</sup>	Sander et al. (2005)
OFFLEM	Offline emissions	Kerkweg et al. (2006)
RAD4ALL	Radiation	Jöckel et al. (2006)
SCAV	Wet deposition	Tost et al. (2006a)
TNUDGE	Tracer nudging	Kerkweg et al. (2006)
TROPOP	Calculation of the tropopause	Jöckel et al. (2006)

<sup>a</sup> Extended version; soil NO emissions switched off in NOBIONO simulation

<sup>b</sup> Tropospheric reaction with NMHC and without halogens

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**Table 2.** Simulated total NO<sub>x</sub> emissions, SNO<sub>x</sub> in Tg(N) of the BASE simulation and in brackets relative contribution of SNO<sub>x</sub> to the total NO emissions for different regions and periods.

Season <sup>a</sup>	Global		Low-latitudes (30° N–30° S)		Mid-latitudes (30° N–60° N) (30° S–60° S)			
	total	soil	total	soil	total	soil	total	soil
DJF	12.53	1.53 (12%)	7.09	1.34 (19%)	4.93	0.06 (1%)	0.46	0.12 (26%)
MAM	11.93	2.03 (17%)	6.07	1.60 (26%)	5.40	0.35 (6%)	0.40	0.08 (20%)
JJA	13.67	2.34 (17%)	7.04	1.43 (20%)	6.15	0.88 (14%)	0.33	0.04 (12%)
SON	13.03	1.81 (14%)	7.33	1.45 (20%)	5.27	0.29 (6%)	0.38	0.07 (18%)
All	51.17	7.72 (15%)	27.52	5.82 (21%)	21.76	1.58(7%)	1.57	0.31 (20%)

<sup>a</sup> DJF = December 1994, January, February 1995; MAM = March, April, May 1995; JJA = June, July, August 1995; SON = September, October, November 1995

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**Table 3.** Correlation coefficient ( $R^2$ ) between surface SNO<sub>x</sub> flux values and the difference (NOBIONO–BASE) of the tracer burden in the overlying lower troposphere (> 500 hPa for all gases and > 750 hPa for PAN values in brackets) by gridcell, both averaged over the corresponding period; only gridcells with a land surface fraction of at least 75% were included.

Season <sup>a</sup>	NO <sub>x</sub>	PAN	HNO <sub>3</sub>	O <sub>3</sub>	OH	NO <sub>x</sub>	PAN	HNO <sub>3</sub>	O <sub>3</sub>	OH
Global (N=2462)						Low-latitudes 30° N–30° S (N=646)				
DJF	0.86	0.53 (0.60)	0.43	0.62	0.58	0.68	0.17 (0.25)	0.18	0.29	0.22
MAM	0.90	0.47 (0.53)	0.60	0.58	0.64	0.79	0.14 (0.24)	0.42	0.29	0.33
JJA	0.82	0.27 (0.35)	0.37	0.37	0.43	0.69	0.07 (0.07)	0.15	0.21	0.22
SON	0.86	0.44 (0.52)	0.43	0.55	0.58	0.66	0.14 (0.24)	0.07	0.14	0.14
Year	0.88	0.46 (0.54)	0.51	0.56	0.60	0.72	0.11 (0.23)	0.16	0.19	0.21
Northern 30° N–60° N (N=637)			Mid-latitudes			Southern 30° S–60° S (N=46)				
DJF	0.55	0.02 (0.09)	0.40	0.07	0.31	0.80	0.45 (0.45)	0.62	0.67	0.39
MAM	0.82	0.19 (0.30)	0.43	0.20	0.27	0.76	0.71 (0.60)	0.45	0.70	0.59
JJA	0.75	0.10 (0.22)	0.26	0.13	0.18	0.83	0.46 (0.51)	0.54	0.46	0.69
AON	0.80	0.09 (0.19)	0.40	0.19	0.32	0.87	0.74 (0.77)	0.65	0.77	0.47
Year	0.77	0.12 (0.23)	0.33	0.16	0.16	0.79	0.66 (0.65)	0.64	0.78	0.56

<sup>a</sup> See Table 2 for abbreviations

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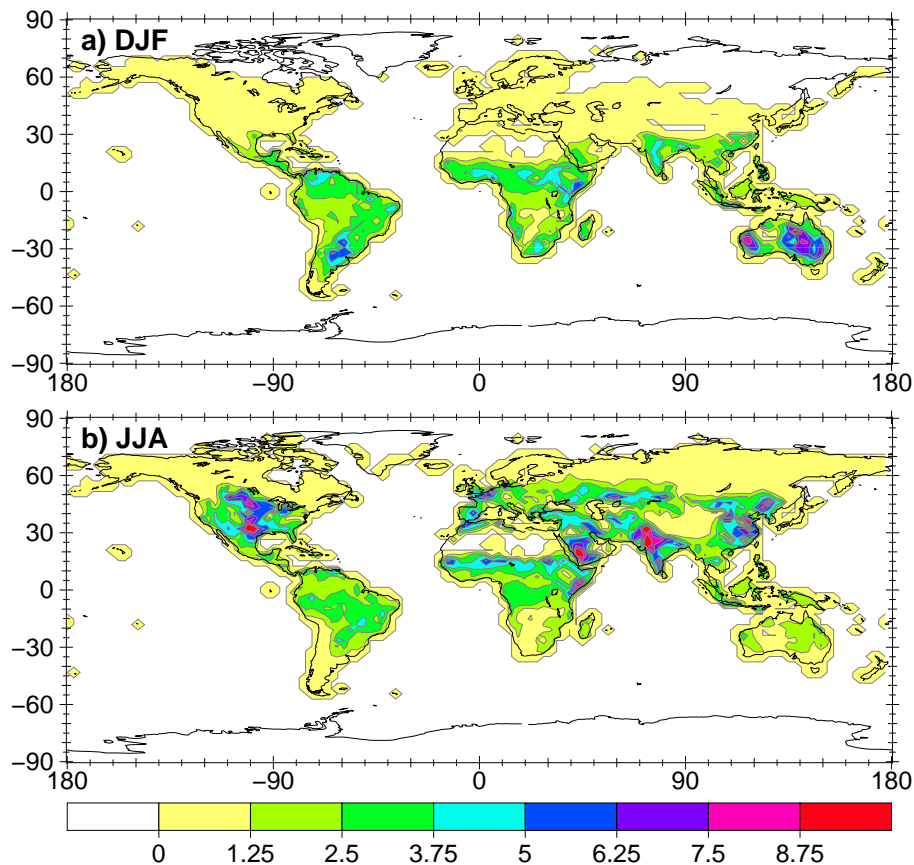
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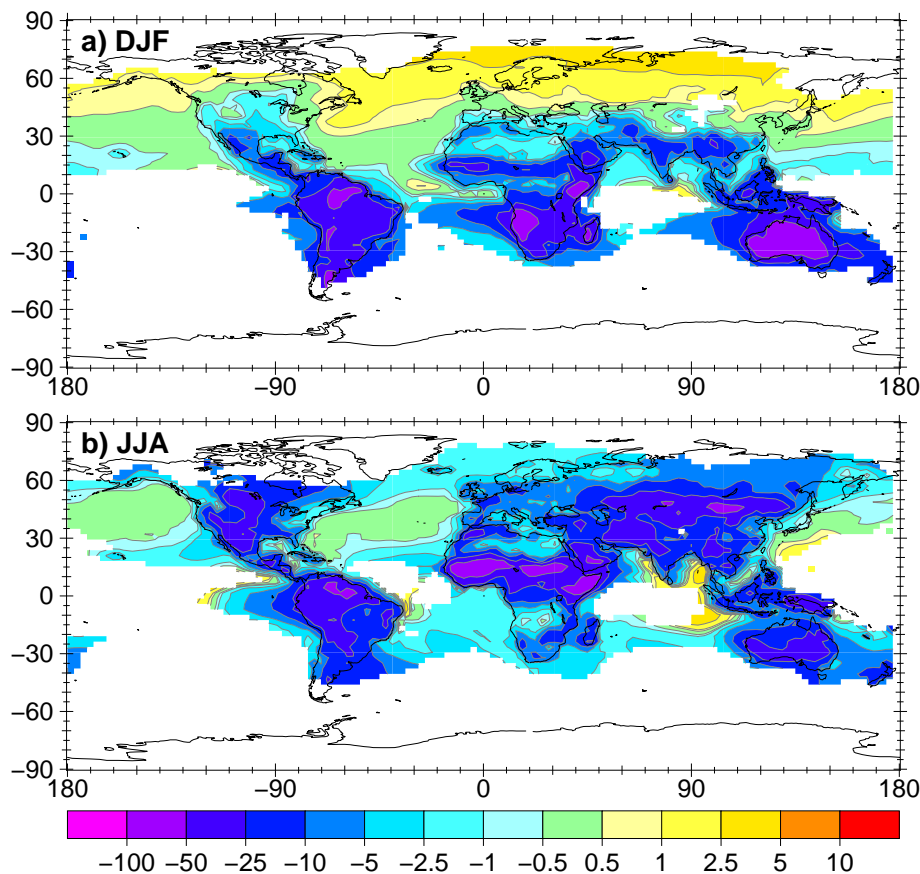


**Fig. 1.** Simulated SNOx flux for **(a)** December 1994 to February 1995 and **(b)** June to August 1995 in  $\frac{\text{ng}}{\text{m}^2 \text{s}}$ .

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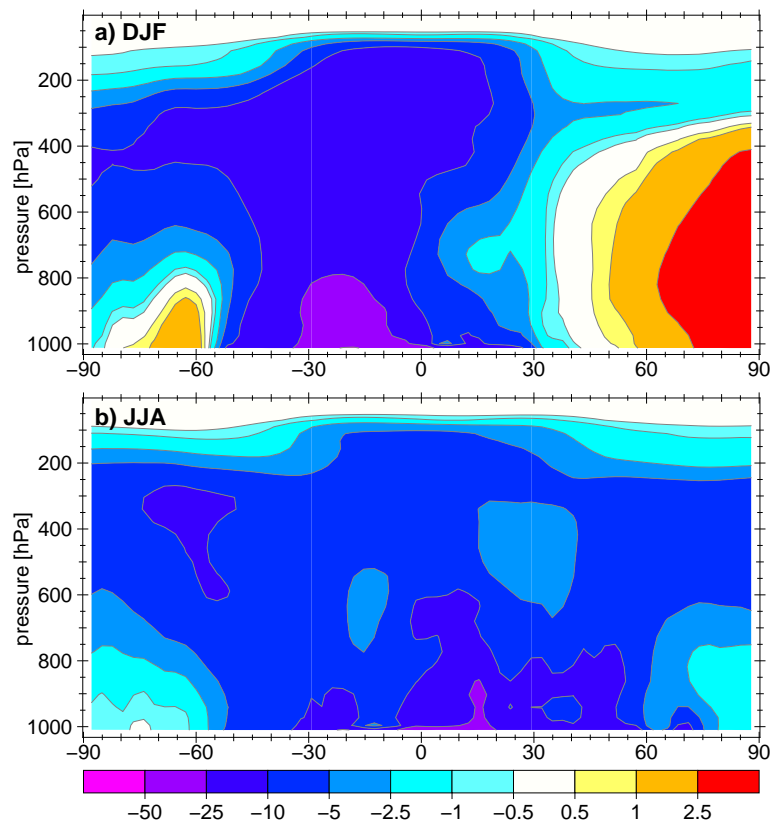
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**Fig. 2.** Relative difference ( $\frac{\text{NO}_{\text{BIONO}} - \text{BASE}}{\text{BASE}} \times 100\%$ ) of the lower tropospheric mixing ratio of  $\text{NO}_x$  in % – values below  $30 \frac{\text{pmol}}{\text{mol}}$  in the BASE simulation excluded from calculation – averaged for **(a)** December, January, February and **(b)** June, July and August.

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**Fig. 3.** Zonal mean relative difference ( $\frac{\text{NOBIONO}-\text{BASE}}{\text{BASE}} \times 100\%$ ) of  $\text{NO}_x$  mixing ratio in % averaged for **(a)** December, January, February and **(b)** June, July and August. Note that the y-axis is linearly scaled, since the focus of this work lies in the lower troposphere.

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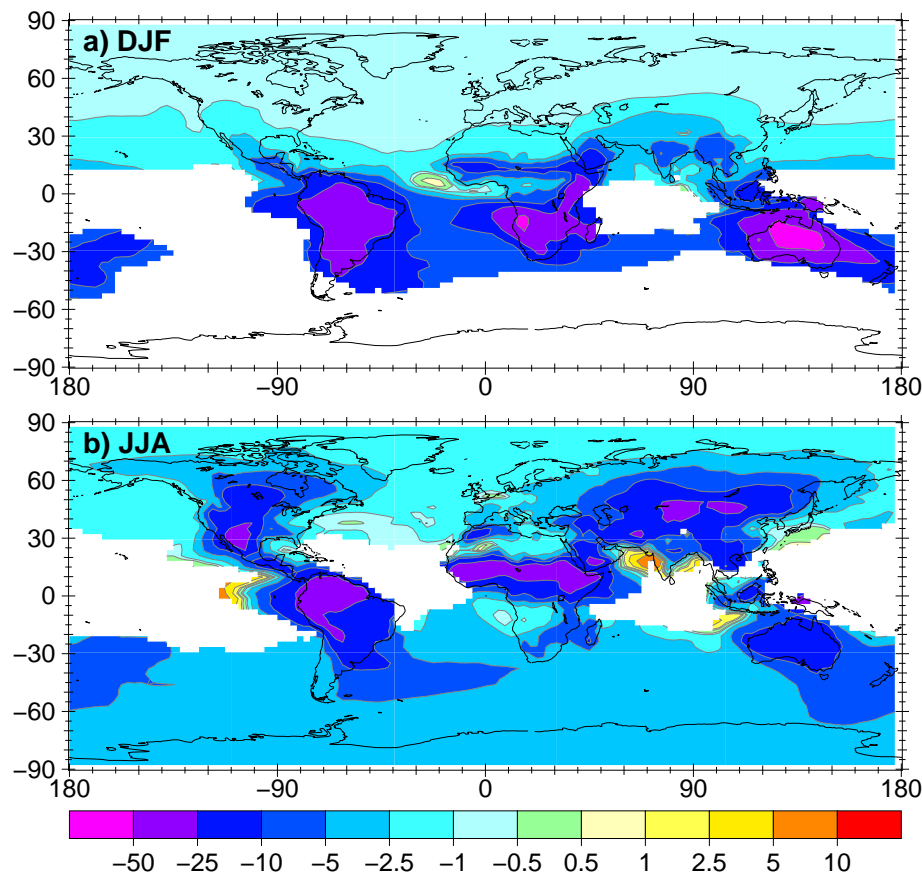
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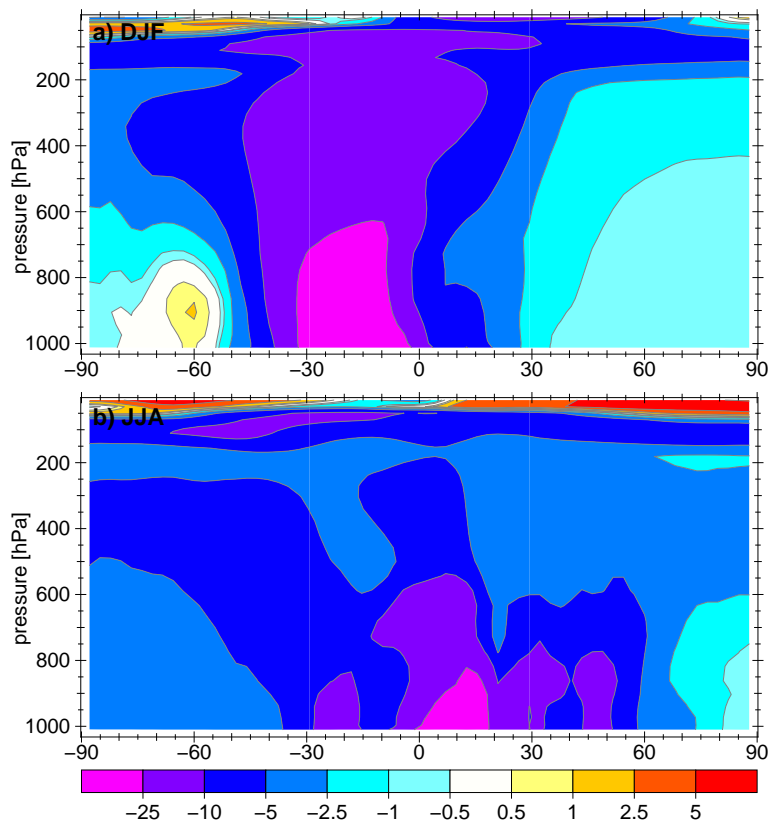
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**Fig. 4.** Relative difference ( $\frac{\text{NO}_{\text{BIONO}} - \text{BASE}}{\text{BASE}} \times 100\%$ ) of the lower tropospheric mixing ratio of PAN in % – values below  $50 \frac{\text{pmol}}{\text{mol}}$  in the BASE run excluded from calculation – averaged for (a) December, January, February and (b) June, July and August.

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**Fig. 5.** Zonal mean relative difference ( $\frac{\text{NOBIONO}-\text{BASE}}{\text{BASE}} \times 100\%$ ) of PAN mixing ratio in % averaged for **(a)** December, January, February and **(b)** June, July and August.

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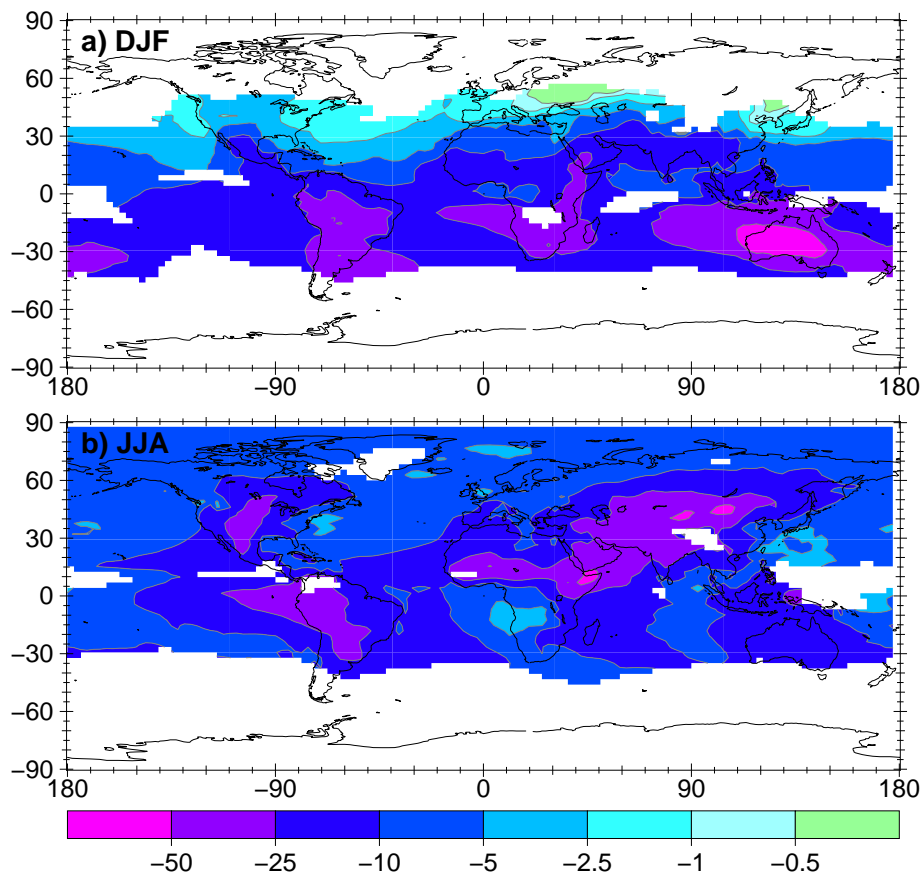
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**Fig. 6.** Relative difference ( $\frac{\text{NOBIONO}-\text{BASE}}{\text{BASE}} \times 100\%$ ) of the lower tropospheric mixing ratio of  $\text{HNO}_3$  in % – values below  $30 \frac{\text{pmol}}{\text{mol}}$  in the BASE simulation excluded from calculation – averaged for **(a)** December, January, February and **(b)** June, July and August.



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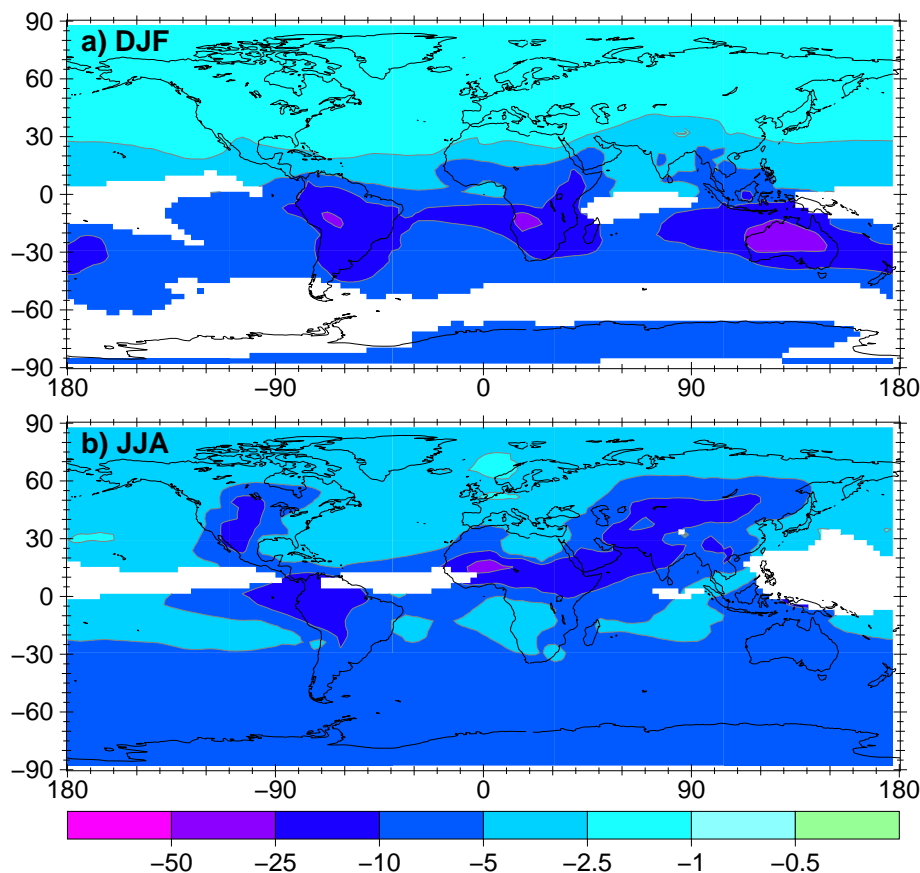
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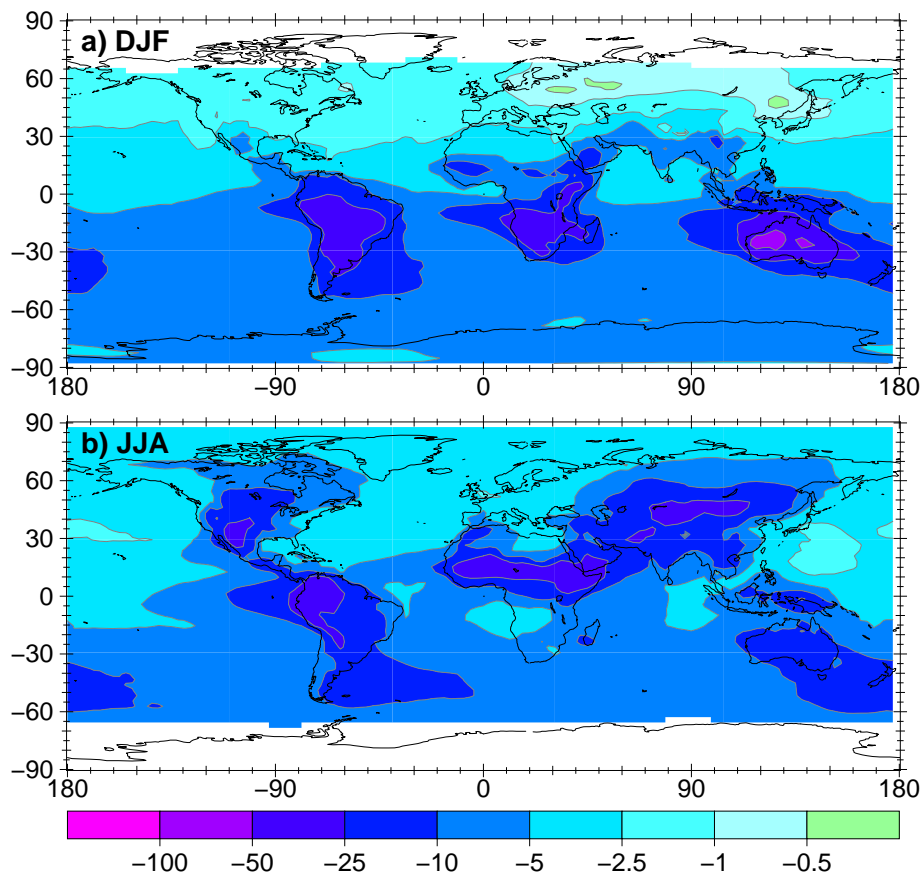
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**Fig. 7.** Relative difference ( $\frac{\text{NOBIONO}-\text{BASE}}{\text{BASE}} \times 100\%$ ) of the lower tropospheric mixing ratio of  $\text{O}_3$  in % – values below  $25 \frac{\text{nmol}}{\text{mol}}$  in the BASE simulation excluded from calculation – averaged for **(a)** December, January, February and **(b)** June, July and August.

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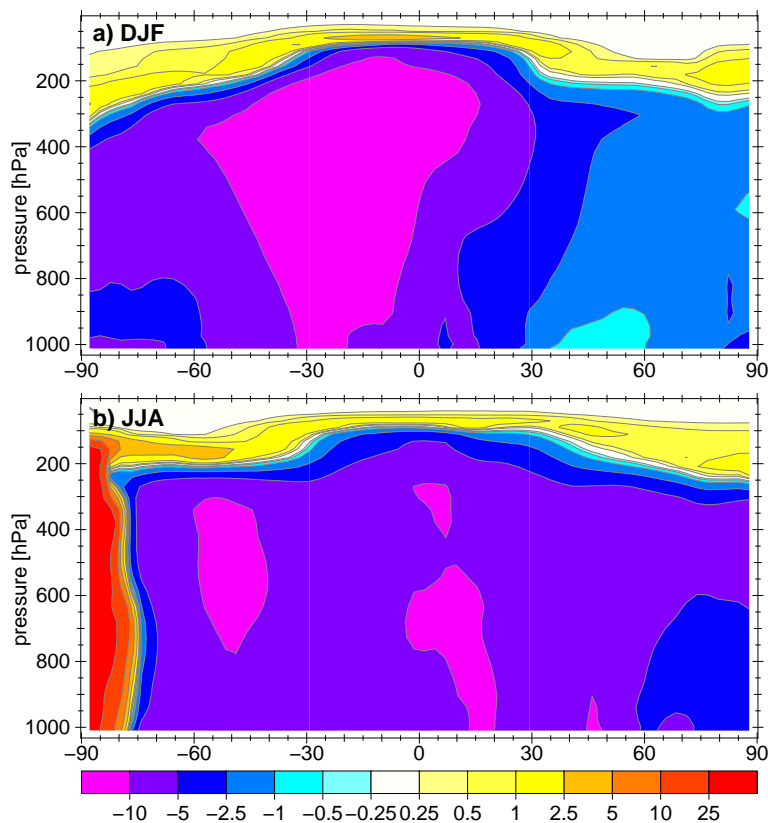


**Fig. 8.** Relative difference ( $\frac{\text{NOBIONO}-\text{BASE}}{\text{BASE}} \times 100\%$ ) of the lower tropospheric concentration of OH in % – values below  $10^4 \frac{\text{molec}}{\text{cm}^3}$  in the BASE simulation excluded from calculation – averaged for **(a)** December, January, February and **(b)** June, July and August.

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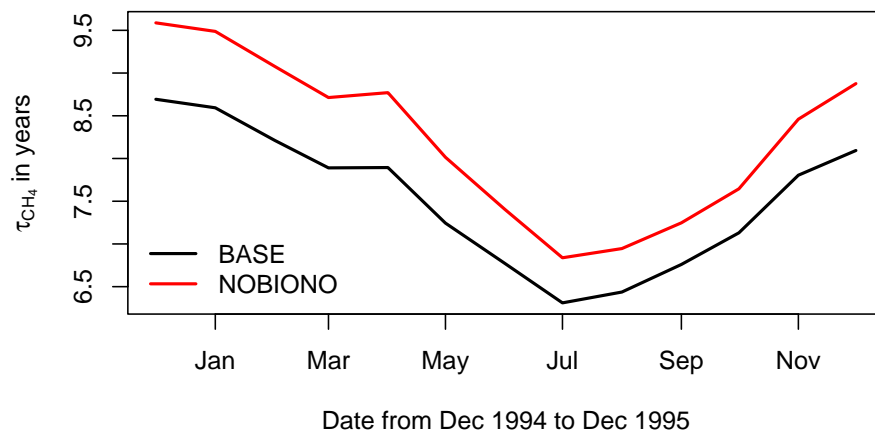


**Fig. 9.** Zonal mean relative difference ( $\frac{\text{NOBIONO-BASE}}{\text{BASE}} \times 100\%$ ) of OH mixing ratio in % averaged for (a) December, January, February and (b) June, July and August.

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**Fig. 10.** Seasonal cycle of monthly mean lifetime of CH<sub>4</sub> from December 1994 to December 1995 in years (calculated according to [Lawrence et al., 2001](#)).

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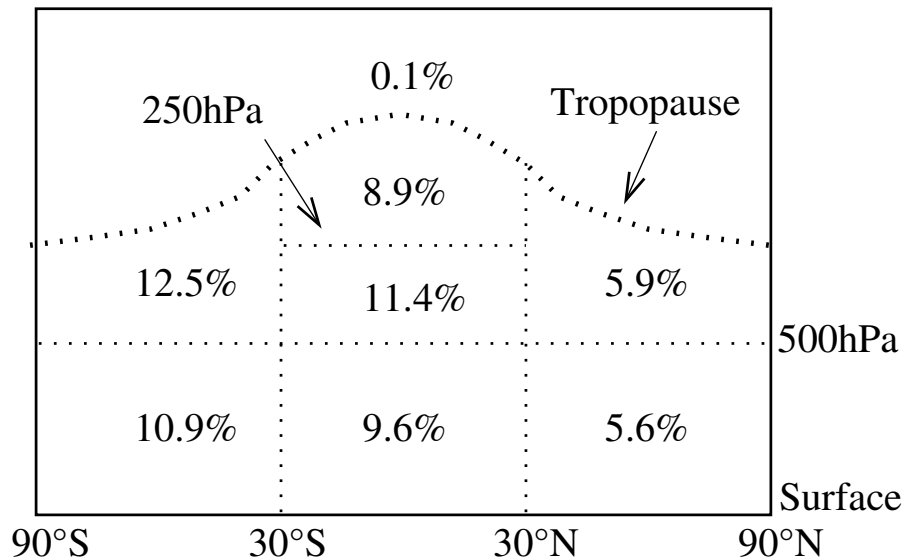
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**Fig. 11.** Relative increase of  $\tau_{\text{CH}_4}$  ( $\frac{\tau_{\text{CH}_4, \text{NOBIONO}} - \tau_{\text{CH}_4, \text{BASE}}}{\tau_{\text{CH}_4, \text{BASE}}} \times 100\%$ ) in various zonal subdomains of the atmosphere (calculated after Lawrence et al., 2001).

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