

**Climate change and
tropospheric ozone**

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Impact of climate change on tropospheric ozone and its global budgets

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Abstract

We present the chemistry-climate model UM_CAM in which a relatively detailed tropospheric chemical module has been incorporated into the UK Met Office's Unified Model version 4.5. We obtain good agreements between the modelled ozone/nitrogen species and a range of observations including surface ozone measurements, ozone sonde data, and some aircraft campaigns.

Four 2100 calculations assess model responses to projected changes of anthropogenic emissions (SRES A2), climate change (due to doubling CO₂), and idealised climate change associated changes in biogenic emissions (i.e. 50% increase of isoprene emission and doubling emissions of soil-NO_x). The global tropospheric ozone burden increases significantly for all the 2100 A2 simulations, with the largest response caused by the increase of anthropogenic emissions. Climate change has diverse impacts on O₃ and its budgets through changes in circulation and meteorological variables. Increased water vapour causes a substantial ozone reduction especially in the tropical lower troposphere (>10 ppbv reduction over the tropical ocean). On the other hand, an enhanced stratosphere-troposphere exchange of ozone, which increases by 80% due to doubling CO₂, contributes to ozone increases in the extratropical free troposphere which subsequently propagate to the surface. Projected higher temperatures favour ozone chemical production and PAN decomposition which lead to high surface ozone levels in certain regions. Enhanced convection transports ozone precursors more rapidly out of the boundary layer resulting in an increase of ozone production in the free troposphere. Lightning-produced NO_x increases by about 22% in the doubled CO₂ climate and contributes to ozone production.

The response to the increase of isoprene emissions shows that the change of ozone is largely determined by background NO_x levels: high NO_x environment increases ozone production; isoprene emitting regions with low NO_x levels see local ozone decreases, and increase of ozone levels in the remote region due to the influence of PAN chemistry. The calculated ozone changes in response to a 50% increase of isoprene

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emissions are in the range of between -8 ppbv to 6 ppbv. Doubling soil- NO_x emissions will increase tropospheric ozone considerably, with up to 5 ppbv in source regions.

1 Introduction

Tropospheric ozone (O_3) has important chemical and radiative roles and has been a focus of many modelling studies. It can be a regional pollutant; high levels of ozone are harmful to human health and vegetation. O_3 is the primary source of the hydroxyl radical (OH), which plays a key role in the oxidizing capacity of the atmosphere. It is also important because of its radiative impact; ozone is currently the third most important greenhouse gas after carbon dioxide (CO_2) and methane (CH_4).

During the industrial era, human activities have changed the chemical composition of the atmosphere considerably. Increasing surface emissions of methane, carbon monoxide (CO), volatile organic compounds (VOCs) and nitrogen oxides ($\text{NO}_x = \text{NO} + \text{NO}_2$), produced by biomass burning and fossil-fuel combustion, have caused tropospheric O_3 concentrations to increase significantly (Volz and Kley, 1988; Thompson, 1992; Marengo et al., 1994). The total amount of tropospheric O_3 is estimated to have increased by 30% globally since 1750, which corresponds to an average positive radiative forcing of 0.35 W m^{-2} (Houghton et al., 2001). Further increases of tropospheric O_3 are anticipated in response to continuing increases in surface emissions.

Tropospheric ozone is formed as a secondary photochemical product of the oxidation of CO and hydrocarbons in the presence of NO_x . Its short chemical lifetime results in an inhomogeneous distribution and a stronger dependence on changes in source gas emissions than for longer-lived greenhouse gases. The complex O_3 chemistry in the troposphere requires a comprehensive chemical mechanism describing NO_x/VOC chemistry to be incorporated in a 3-dimensional chemistry/climate model to simulate the global O_3 distribution and to assess climate feedbacks.

Numerous studies of the evolution of tropospheric O_3 changes since preindustrial times and the associated radiative forcings have been carried out using various chem-

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ical transport or climate models (e.g., [Hauglustaine et al., 1994](#); [Forster et al., 1996](#); [Roelofs et al., 1997](#); [Berntsen et al., 2000](#); [Brasseur et al., 1998](#); [Stevenson et al., 1998a](#); [Mickley et al., 1999](#); [Hauglustaine and Brasseur, 2001](#); [Grenfell et al., 2001](#)). However there is often significant difference between models in their predictions of ozone change, (see, e.g. [Houghton et al., 2001](#)), even though these models normally reproduce present-day observations “satisfactory”. Future changes of tropospheric O₃ will depend on how the emissions of ozone precursors change in the future and also on how the climate will change. Continuing emissions of NO_x and VOCs are predicted to increase tropospheric O₃, but the anticipated rise in temperature and humidity will likewise have an impact. A number of studies have suggested that an anticipated warmer and wetter climate would slow down the increase in O₃ abundance compared to an unchanged climate ([Brasseur et al., 1998](#); [Johnson et al., 1999](#); [Stevenson et al., 2000](#)).

We have reported earlier studies with a tropospheric chemical module (identical to the off-line CTM TOMCAT, see [Law et al., 1998](#)) incorporated into the UK Met Office (UKMO) Unified Model (UM) version 4.4. The chemistry comprised NO_x/CO/CH₄/NMVOCs(C2-C3 alkanes, HCHO, CH₃CHO and Acetone). The model was used to assess tropospheric ozone changes between 2000 and 2100 using the SRES A2 scenario ([Zeng and Pyle, 2003](#)). We assessed the feedback on chemical ozone production following increased water vapour, but, in contrast to some of the earlier studies, found that large-scale dynamical changes in a future climate led to an increase in tropospheric ozone through enhanced stratosphere-troposphere exchange ([Zeng and Pyle, 2003](#)). An increased STE in a future climate was also reported by [Collins et al. \(2003\)](#) and [Sudo et al. \(2003\)](#). However detailed studies on the feedbacks between climate change and tropospheric composition are still limited. [Stevenson et al. \(2005\)](#) discuss the impact of changes in physical climate on tropospheric chemical composition; the climate feedbacks are dominantly negative (e.g. reduced tropospheric ozone burden and lifetime, and shortened methane lifetime), but more modelling studies are needed in order to reach a common consensus on climate feedbacks. Most

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recently, [Stevenson et al. \(2006\)](#) reported an ensemble modeling study from 25 models for the 4th IPCC assessment. Simulations for the assessment contrasted a 2000 atmosphere with the year 2030, including runs with changed precursor emissions only, and a run considering both the changes of emissions and climate. The model-model differences are largest when the climate change scenario is considered.

It is likely that climate change will also affect emissions of trace gases from the biosphere. Isoprene is a reactive biogenic compound, emitted by several plant species and with a global source comparable to methane ([Guenther et al., 1995](#)). Isoprene emission is sensitive to temperature (e.g., [Monson and Fall, 1989](#); [Sharkey et al., 1996](#)), CO₂ concentration ([Rosenstiel et al., 2003](#)) and water availability ([Pegoraro et al., 2005](#)), amongst other factors. These driving variables are expected to change in the future, hence there has been some interest in projecting future isoprene emissions and quantifying their effect on atmospheric composition ([Sanderson et al., 2003](#); [Hauglustaine et al., 2005](#); [Wiedinmyer et al., 2006](#)). Estimates of the emission at the end of the 21st century range from 640 to 890 TgC/yr ([Sanderson et al., 2003](#); [Lathiere et al., 2005](#); [Wiedinmyer et al., 2006](#)), with the biggest driver being the projected increase in surface temperature. Changes in the emission of NO_x by microorganisms in soils is another important climate-biosphere feedback, due to the importance of NO_x in tropospheric photochemistry. These emissions are expected to increase in a warmer, wetter atmosphere ([Yienger and Levy, 1995](#)).

In this paper, we present the results from a version of our chemistry-climate model (UM_CAM) updated to include an isoprene oxidation scheme. We calculate ozone changes for a 2100 climate with associated chemical and dynamical changes (e.g. ozone production/destruction, stratosphere-troposphere exchange). We chose the SRES A2 scenario for the year 2100, which predicts relatively large emission increases, in order to explore a large range of influences on future air quality. Currently, there are large uncertainties in the estimation of biogenic emissions. Here, we carry out an initial assessment of the impact of idealised increases in isoprene emissions and altered

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soil-NO_x emissions on future tropospheric ozone. A paper by Young et al. (2007)¹ will discuss in detail the role of isoprene on ozone formation.

The base climate model is the UM version 4.5. We describe the model in Sect. 2. In Sect. 3 the experimental setup is given. In Sect. 4, we present the present-day simulation and compare modelled O₃, NO_x and PAN to observations. Section 5 presents future simulations: impacts on tropospheric O₃ from the anthropogenic emissions and from changes in meteorology are discussed; idealised changes in the biogenic emission, related to climate change, are also assessed. The tropospheric ozone budgets for the various cases in Sects. 4 and 5 are analysed. Conclusions are gathered in Sect. 6.

2 Model description

2.1 Climate model

The UM is developed and used at the UKMO for weather prediction and climate research (Cullen, 1993; Senior and Mitchell, 2000; Johns et al., 2003). Here we use the 19 level UM version 4.5. It uses a hybrid sigma-pressure vertical coordinate, and the model domain extends from the surface up to 4.6 hPa. The horizontal resolution is 3.75° by 2.5°. The model's meteorology is forced using prescribed sea surface temperatures (SSTs).

We adopt an improved tracer advection scheme (A. R. Gregory, private communication, 2001) to replace the existing scheme in the UM which is based on Roe's flux redistribution method (Reo, 1985). The Roe "scheme" has the advantage of being a monotonic method but suffers from low accuracy at the climate resolution. The new tracer transport scheme is based on the 1-D NIRVANA scheme of Leonard et al.

¹Young, P. J., Zeng, G., and Pyle, J. A.: Isoprene chemistry in a future atmosphere, in preparation, 2007.

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(1995) using the same extension to 3-D/sphere as the Roe scheme. The new scheme is conservative, monotonic and more accurate than the Roe scheme at a lesser computational cost. It is considerably less diffusive in the vertical than the Roe scheme.

Convection is parameterized by a penetrative mass flux scheme (Gregory and Rowntree, 1990) in which buoyant parcels are modified by entrainment and detrainment to represent an ensemble of convective clouds. The convection scheme has been tested using ^{222}Rn experiments; the agreement with observations is reasonable (Stevenson et al., 1998b).

The Edwards and Slingo (1996) radiation code is used in the UM. Absorption by water vapour, carbon dioxide, and O_3 are included in both longwave and shortwave calculations. Absorption by methane, nitrous oxide, CFC-11 and CFC-12 are also included in the longwave scheme. Water vapour is a basic model variable. Prescribed monthly zonal mean O_3 climatology fields are used in the radiation scheme unless otherwise stated. Mixing ratios of other gases are assumed to be global constants.

2.2 Chemical module

The tropospheric chemical mechanism includes CO, methane and NMVOC oxidation as previously used in the off-line transport model TOMCAT (Law et al., 1998) and earlier versions of the UM+chemistry model (Zeng and Pyle, 2003, 2005). An isoprene oxidation scheme adopted from Poschl et al. (2000) was recently added into the model (see Young, 2007 for more details). Reaction rates are taken from the recent IUPAC (Atkinson et al., 1999) and JPL (DeMore et al., 1997) evaluations. Chemical integrations are performed using an implicit time integration scheme, IMPACT (Carver and Stott, 2000), with a 15 min time step. The model includes 60 species and 174 chemical reactions. Two tracers, O_x and NO_x , are treated as chemical families. OH, HO_2 and other short-lived peroxy radicals are assumed to be in steady state. The model uses the diurnal varying photolysis rates calculated off-line in a 2-D model (Law and Pyle, 1993) and interpolated to 3-D fields. Loss of trace species by dry deposition is included using deposition velocities which are calculated using prescribed deposition

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5 velocities at 1 m height (largely taken from [Valentin, 1990](#) and [Zhang et al., 2003](#)), depending on season, time of the day and on the type of surface (grass, forest, dessert, water, snow/ice) and are extrapolated to the middle of the lowest model layer using a formula described by [Sorteberg and Hov \(1996\)](#). Wet deposition of soluble species is represented as a first order loss using model-calculated large-scale and convective rainfall rates. A detailed description of the dry and wet deposition schemes is given by [Giannakopoulos et al. \(1999\)](#). Instead of explicit stratospheric chemistry in the model, daily concentrations of O_3 , NO_y and CH_4 are prescribed at the top three model layers (29.6, 14.8 and 4.6 hPa) using output from the 2-D model, to produce a realistic annual cycle of these species in the stratosphere. Note that the scheme includes the important stratospheric NO_x/HO_x chemistry and is applied in the lowermost stratosphere (i.e. below 30 hPa) but that no halogen chemistry is included.

3 Experimental setup

15 We have performed 5 simulations (see Table 1). The baseline run A covers the years 1996–2000 using emissions for year 2000 and is used to verify the model performance against observations. Run B uses 2100 emissions to assess changes of tropospheric composition only due to changes in these anthropogenic emissions. Run C calculates future changes due to changes in both anthropogenic emissions and the climate using 2100 emissions (same as run B) and a double CO_2 climate forcing with appropriate SSTs. In runs B and C biogenic emissions are held constant. Run D is based on run C but uses elevated isoprene emissions to assess the sensitivity of the model to increased biogenic emissions which may be associated with climate change. Similar to run C, run E also accounts for increased soil NO_x emissions in addition to the anthropogenic emissions. All runs were for several years (3–5 years) with averaged data analysed for each scenario.

25 Emissions are seasonally varying but have no inter-annual variability except for NO_x produced from lightning which is climate-dependent. In detail, the anthropogenic emis-

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sions of NO_x, CO and NMVOCs for the present-day are from the recently published emission scenarios by the International Institute of Applied System Analysis (IIASA) which was described in detail by Dentener et al. (2005) and references therein. Anthropogenic emissions appropriate to 2100 are based on the IPCC Special Report on Emission Scenarios (SRES) for the year 2100 (Nakićenović et al., 2000). We chose the A2 scenario to demonstrate the sensitivity to assumed large emission changes.

We include 512 TgC/yr total annual emissions of isoprene (Guenther et al., 1995) in the base run. For run D, isoprene emissions are increased by 50% relative to the base run (768 TgC/yr). This increased isoprene emission sits roughly in the middle of the 2100 estimates of Sanderson et al. (2003), Lathiere et al. (2005), and Wiedinmyer et al. (2006) (640–890 TgC/yr) although, unlike their experiments, we do not account for any change in the distribution of the vegetation and simply scale up the present day emissions. Run D is used to assess the sensitivity of the modelled 2100 atmosphere to an increase in isoprene emission, rather than be a prediction of future isoprene emissions.

Large uncertainties exist in estimating the global soil-biogenic NO_x emissions. In our base run, we take the data from Yienger and Levy (1995) and scale to 7 TgN/yr which is close to the upper end of their estimation for the 1990s. Yienger and Levy (1995) related soil-NO_x emissions to biome, soil temperature, precipitation and fertilizer application. They estimated a 25% increase from 1990 to 2025 in response to a warmer, wetter climate. In our soil-NO_x perturbation run E, we simply double the present-day value to develop a scaled emission field appropriate for a 2100 atmosphere sensitivity experiment.

For all the runs, other natural emissions are kept the same as in 2000 and are taken from the EDGAR3.2 global emission inventory (Oliver and Berdowski, 2001). Biomass burning is based on the Global Fire Emission Data averaged for 1997–2002 (van der Werf et al., 2003). Lightning-produced NO_x is calculated as a function of the cloud top height using the parameterization of Price and Rind (1992, 1994) and scaled close to 4 Tg(N)yr⁻¹ for the present day simulation. For the simulation with the future climate

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forcing, a 22% increase of NO_x was found as a result of the increase in convection in a warmer and wetter climate. Methane concentrations are constrained throughout the model domain to reduce the spin-up time and eliminate possible trends. A summary of the emissions is given in Table 2. All runs were for several years (3-5 years) with averaged data analysed for each scenario.

We use observed monthly mean sea surface temperature and sea ice climatology compiled at the Hadley centre (GISST 2.0) to drive the present-day climate. The future climate is driven by SSTs produced by the Hadley Centre coupled ocean-atmosphere GCM (HadCM3) run with IS92a emissions for the year 2090–2100 (Johns et al., 2003; Cox et al., 2004). In the radiation scheme, the same present-day O_3 climatology from Li and Shine (1995) is used for all model runs; other trace gases mixing ratios are fixed at present-day levels with only CO_2 doubled in the future climate runs. Prescribed stratospheric O_3 and NO_x , and photolysis rates are kept the same for all model runs.

4 Present-day simulation

4.1 Ozone

Figure 1 shows modelled monthly mean distributions of surface O_3 for January, April, July and October. Surface O_3 is generally higher in the NH than in the SH due to the higher emissions of ozone precursors there. Seasonally, the highest surface O_3 level occurs in July due to intensified photochemical production of O_3 , while in winter the O_3 level is generally low with higher values over the ocean. The model calculation shows that surface O_3 is low all year round in the equatorial western Pacific region; this area (the “warm pool”) is characterized by high sea surface temperatures and strong convection and precipitation which lead to efficient chemical destruction of O_3 and the strong lifting of O_3 precursors. It is evident that the long range transport of O_3 and its precursors results in elevated O_3 away from its sources. Note, for example, the high ozone values over the Atlantic, downstream of the eastern US, and in the Pacific, to

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the east of the Asian Pacific rim.

Figure 2 shows zonally averaged O_3 concentrations for January, April, July and October. At the equator the air at all heights is characterized by relatively low O_3 , as a result of convective transport of O_3 -poor air from the surface up to the tropical tropopause. At higher latitudes, downward transport of O_3 -rich air dominates. The northern-hemispheric stratosphere/troposphere exchange (STE) is strong in April while the O_3 concentration is low in the tropical region reflecting stronger upwards transport. In the southern hemisphere, STE maximizes in austral spring. Photochemical production of O_3 is strongest in summer in the NH and convection leads to efficient transport of O_3 precursors out of the boundary layer and mixing into the free troposphere.

We have compared the baseline simulation with a wide range of long term observations. Figure 3 shows the simulated and observed monthly mean O_3 concentrations near the surface. The observational data are from the World Data Centre for Surface Ozone (WDSO) (<http://gaw.kishou.go.jp/wdcgg.html>), with major contributions from CMDL. We have selected stations with data covering the years 1996–2004 where possible. The ozone concentrations from the simulation are averaged over 1996–2005. Most of the observations are well reproduced by the model. For the northern-hemisphere extratropical remote sites, the observations are characterized by a spring maximum and a summer minimum associated with relatively strong downward flux of O_3 from the stratosphere in the spring and efficient photochemical destruction in summer. A summer maximum over polluted continental areas (e.g. at Hohenpeissenberg) associated with intensified photochemical production is well reproduced by the model. There are some discrepancies between the model and the measurements. At Barrow, the observed spring minimum in surface O_3 may well be associated with bromine chemistry (Barrie et al., 1988) which is not represented in the model. The observed summer minima at Ryori and Bermuda are weak in the model simulation. At the southern tropical sites, the observations indicate an austral spring maximum which is well captured by the model. The year-round low O_3 value in Samoa is well simulated. The austral spring maxima at Cuiaba, produced by biomass burning, is also accurately

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reproduced. The observed surface O₃ concentrations at southern middle and high latitudes shows a winter maximum and a summer minimum in the lower troposphere. This feature is not well simulated by the model. The seasonal cycle produced by the model is relatively weak and does not reflect that of the measurements. The model over-estimates summer O₃ concentrations in Baring Head and also underestimates O₃ values by up to 20 ppbv in the austral winter-spring over Antarctica. The cause of these discrepancies needs to be investigated further.

We have also compared the modelled vertical ozone concentrations to O₃ sonde measurements (Logan, 1999) made between 1985–1995 (Fig. 4). Note that the ozone concentrations from the baseline simulation are averaged over 1996–2005. The model simulation agrees reasonably well with the observations. The model captures very well the strong vertical O₃ concentration gradient shown in the measurements in middle to high latitudes in both hemispheres. However, the model overestimates spring-summer mid-upper tropospheric O₃ concentrations to some degree (e.g. at Edmonton). The mid tropospheric maxima in the northern subtropical sites at Kagoshima and Hilo are well simulated by the model. In the southern tropical sites at Natal and Samoa, the model simulates the steep decrease of O₃ concentrations in the lower troposphere but cannot reproduce well the elevated O₃ concentrations in the middle and upper troposphere, especially in Natal, indicating possibly that there is not enough convective lifting of O₃ and its precursors to the middle and upper troposphere during the biomass burning season.

In general the model does a good job in simulating observed O₃. However, to understand factors affecting O₃ it is necessary also to look at the processes involved in ozone production, destruction and transport. The ozone budget depends critically on the concentration of ozone precursors and comprises the chemical production and destruction of O₃, stratosphere/troposphere exchange (STE) and dry deposition at the surface. Although models can generally reproduce the observations of O₃ concentrations, there are large differences in tropospheric O₃ budgets between different models (see, e.g., Table 4.12 of IPCC, Houghton et al., 2001 and Table 5 of Shindell et al., 2001): the

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net chemical production varies from -810 to 550 Tg/year; the flux from the stratosphere to the troposphere from 390 to 1440 Tg/year; and the dry deposition from 533 to 1237 Tg/year. Most recent intermodel comparison shows that differences in O_3 budget calculations are reduced among models for the present-day scenario (see Stevenson et al., 2006). In our calculations (see Table 3 – scenario A), the net influx from the stratosphere is 452 Tg/year which is within the range reported by Houghton (2001) and Stevenson et al. (2006). The main chemical reactions contributing to O_3 production are reactions between NO and hydroxyl peroxide/other peroxy radicals (RO_2). The chemical destruction channels are mainly through the reactions $H_2O+O(^1D)$ and O_3+HO_x . The net chemical production (NCP) of 512 Tg/year calculated from these main terms is within the reported range. Note that it is a small residual of two large production and destruction terms which are 3620 to 3108 Tg/year, respectively, in our calculation. Gross ozone production vary greatly across models on present-day simulations (2300 to 5300 Tg/year) (Stevenson et al., 2006) and are likely due to differences in complexities of chemical mechanisms included (see also discussions by Wu et al., 2007). Our dry deposition of 1035 Tg/year is at the high end of the model range. The total tropospheric burden of 314 Tg is within the range seen in other models. We use a 150 ppbv O_3 threshold to define tropospheric air. All the budget calculations are the global sum below this threshold.

4.2 Nitrogen species

The model calculated NO_2 column averaged for year 2000 is in good agreement with the GOME measurement (not shown). Here we emphasize speciated comparisons; we compare some measured and modelled NO_x and PAN vertical profiles. The observation data are from short-term aircraft campaigns compiled by Emmons et al. (2000) and should not necessarily compare in detail with model results from a climate simulation. Nevertheless, the modelled NO_x concentrations are generally in reasonable agreement with observations especially in the mid troposphere (see Fig. 5). The low NO_x concentrations observed over the remote Pacific regions are well represented by the model.

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The model does a good job in reproducing higher NO_x mixing ratios in the lower troposphere during February-March (PEM-West-B) in East Asia, arising from the strong influence of local anthropogenic emissions. Note, in particular, that the “C” shaped profile found in observation along the Japanese coast is well simulated by the model.

5 Biomass burning in Africa and South America during September-November (TRACE-A) leads to a large near-surface enhancement of NO_x mixing ratios in the surrounding regions. The NO_x profile in East Brazil is well reproduced but the model underestimates NO_x mixing ratios in the lower troposphere in South Africa during the biomass burning season. The higher mixing ratios of NO_x seen in the middle to upper tropospheric
10 over the South Atlantic during September-November are from biomass burning emissions that have been transported from the continents (PEM-Tropics-A and Tracer-A) (see Emmons et al., 2000 and references therein); this is not reproduced by the model. Note that the modelled data are the average over a decade; hence they do not capture interannual variability of emissions (which are the same for every year of the model
15 run) and meteorological conditions. Savage et al. (2007)² have recently pointed to the importance of interannual variations in meteorology for explaining observed NO_x .

A comparison of modelled and observed PAN is shown in Fig. 6. Modelled PAN depends strongly on the magnitude of the VOC and NO_x emissions sources and the regional meteorology, as well as on the precise hydrocarbon degradation scheme included. With isoprene chemistry in this version of the model, modelled PAN has been
20 improved considerably compared to the previous version without isoprene chemistry which systematically underestimated PAN (not shown). The model simulates well the increase of PAN with altitude over the oceans. The peak observed in the Pacific and Atlantic oceans in the 4–8 km region during PEM-Tropics-A and Tracer-A, associated with the transport of PAN from South America, Australia and Africa, is reproduced by the model. PAN profiles over the China Coast and Japan observed during PEM West
25 B reflect strong outflow of pollutants from Asia to the North Pacific with high values

²Savage, N. S., Pyle, J. A., Braesicke, P., et al.: The sensitivity of NO_2 columns to interannual variability of meteorology and emissions: a model-GOME study, submitted, 2007.

seen near the surface; the model simulates this feature well. PAN over the Philippine Sea peaks in the middle troposphere reflecting the transport of PAN from the Asian continent (PEM West B). However, the seasonal change of PAN over the Philippine sea is not well captured by the model. High levels of PAN observed during September–November (TRACE A) are the result of biomass burning in Africa and South America and the transport to the South Atlantic ocean. The model well simulates the vertical profiles of PAN in these regions. Addition of isoprene to the model has led to a much improved NO_y distribution compared with our earlier model simulations.

5 Tropospheric composition changes between 2000 and 2100

5.1 Response to anthropogenic emission changes

Figure 7 shows calculated changes in surface O_3 for January and July between 2000 and 2100 assuming only changes in emissions (i.e. Run B–Run A). In the Northern Hemisphere, increases of O_3 peaking above 40 ppbv are calculated over the polluted continents, with the largest increase of O_3 in the Far East in summer. The areas of larger ozone increase are regions where rapid economic growth and population increase are predicted. In January, there are significant increases of O_3 over the oceans. For the Pacific region this corresponds to an outflow of pollutants from Asia, highlighting the potential importance of the Asian plume and its impact on global O_3 levels in the future. In the Southern Hemisphere, O_3 increases of 30 ppbv are calculated in South Africa and South America. The long range transport of O_3 from these regions is evident: There is a background O_3 increase of up to 5–15 ppbv in the southern hemisphere in remote oceanic areas. Increased surface emissions of O_3 precursors not only contribute to O_3 formation in the source region (air quality) but also increase the O_3 level in remote regions through long-range transport.

Figure 8a shows the calculated zonally averaged O_3 changes between 2000 and 2100 (B-A). The largest increase of O_3 occurs in the northern hemisphere subtropical

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free troposphere; it is a consequence of increases in emissions in northern latitudes and weak destruction of O_3 above the boundary layer. The O_3 increase in the stratosphere results from transport of O_3 precursors from the troposphere, although the increase is relatively small compared to the background. Figure 8b shows NO_x changes for B-A; largest increases are in the boundary layer where the direct emissions occur, and in the upper tropical troposphere where its lifetime is long. There, large increases of NO_x result in substantial O_3 production.

The global tropospheric O_3 budgets for scenario B are shown in Table 3. The net stratospheric influx is about 5% smaller than in scenario A. This small net decrease is the result of a larger tropical troposphere-to-stratosphere O_3 flux in B that more than offsets the middle latitude O_3 influx from the stratosphere. Chemical production increases substantially as a result of increasing emissions of O_3 precursors. Chemical destruction also increases in response to the increased O_3 . The overall net chemical production of O_3 is nearly three times that of scenario A. With higher O_3 , the dry deposition increases by a factor of 1.7. The average tropospheric O_3 burden increases from 314 to 549 Tg.

OH controls the oxidizing capacity in the troposphere and its distribution depends critically on NO_x and hydrocarbons. Increases of NO_x and O_3 tend to increase OH and increases of CO and CH_4 depress OH. Figure 8c shows changes in OH in response to changes in NO_x /VOCs emissions; there are increases throughout the tropical troposphere with the largest increase in the upper troposphere, corresponding to the increase of NO_x there. OH decreases in a large area of the NH and some of the SH as a result of increases of VOCs over the continents. Although the tropospheric OH burden increases by 17% from run A to run B, the methane lifetime decreases only slightly from 11.3 to 11.2 years (see Table 3); the methane lifetime is mainly influenced by the changes in lower tropospheric OH.

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5.2 Response to climate change

The impact of climate change on the O₃ distribution and its budget is assessed by considering the differences between run C and run B, where only differences arise from the doubling CO₂ (see Table 1). Figure 9 shows the differences in temperatures caused by a doubled CO₂ forcing. The average temperature increase is 2–3 K at the surface (higher at high latitudes) and reaches 9 K in the upper tropical troposphere. Cooling in the lower stratosphere occurs in the double-CO₂ climate. Specific humidity increases throughout the troposphere with substantial increase in the tropical boundary layer by 20% (not shown).

Figure 10 shows zonal mean changes of O₃ due to climate change for January, April, July and October. It shows that for all seasons enhanced chemical destruction, due to increased water vapour in the projected future climate, dominate O₃ changes in the tropical troposphere. A pronounced feedback is the substantial increase of O₃ (over 200 ppbv) in the extratropical low stratosphere which is a response to changes in circulation; the enhanced Brewer-Dobson circulation more rapidly lifts O₃-poor air upwards in the tropics and transports O₃-rich air into high latitudes. This leads to an O₃ reduction in the upper tropical troposphere and an O₃ buildup at high latitudes in the lower stratosphere, (in part also due to reduced ozone destruction in the cooler lower stratosphere, consistent with our earlier finding based on an older model version (Zeng and Pyle, 2003)). In a recent multimodel comparison (Butchart et al., 2006) most participating models also produce an increase in the stratosphere-troposphere mass exchange rate in response to growing greenhouse gas concentrations. Consequently, the enhanced STE transports stratospheric O₃ more rapidly to the troposphere leading to significant increases of O₃ in the free troposphere; For the NH this feature is most pronounced in April, shown in Fig. 10b when STE normally maximizes. The influence of STE peaks in Austral winter/spring for the SH, leading to increased O₃ in the free troposphere which also propagates to the lower troposphere. The elevated O₃ levels over the southern midlatitudes and the Antarctic in July and October shown

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in Figure 10c,d seems linked due to increased stratosphere to troposphere transport of O_3 and low surface O_3 deposition rates at these locations (namely water and snow surfaces). Note that the decreases of O_3 along the tropopause are due to tropopause lifting in a changed climate.

Responses of ground-level O_3 to climate change are highly complex. Figure 11 shows monthly mean surface O_3 changes for January, April, July and October, in response to a doubling of CO_2 . O_3 changes are predominately negative through increased water vapour for all the seasons, with the largest decrease over tropical oceans. However, we note some prominent, seasonally-varying O_3 increases, e.g., over some of the polar regions, over the Southern Ocean in Austral winter, and over some of the continents, with largest increases over Amazonia, Africa, North America, southern and central Europe in summer. The significant increase of O_3 over the Arctic in April shown in Fig. 11b (most pronounced in February/March, not shown) may be due to an intensified poleward transport of O_3/O_3 precursors from North America and Europe, possibly associated with increased STE, but needs further investigation. There is also some increase of O_3 over the Arctic in October following summer increases of surface O_3 over Europe and North America. We also note that, to some extent, increases of surface O_3 are linked to increased surface temperatures. A number of mechanisms can lead to increased O_3 production following increased temperature: 1) favoured production of HO_x due to mostly positive temperature-dependencies of CH_4/VOC oxidation reaction rate constant which fuel ozone formation; our calculations show that increases of HO_x correlate closely to the increase in O_3 , and 2) the faster decomposition of PAN which releases NO_2 leading to regionally increase of ozone production, especially in the NH polluted regions. Over Amazonia, southern Africa and southern Europe the model predicts a reduced humidity following the increased temperature associated with reduced soil moisture (Cox et al., 2004), reducing the O_3 destruction in those regions.

The climate change also comprises changes of convection, which play an important role in redistributing O_3 and its precursors in the troposphere (see Lawrence et al. 2003; Doherty et al., 2005). Figure 12a displays ΔNO_x for runs C-B; increases of

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NO_x occur mainly in the tropical upper and middle troposphere, which are likely due to increased deep convection and the increased lightning activity respectively. Enhanced convection can transport NO_x and other O₃ precursors to the upper troposphere more efficiently, while intensified lightning produces NO_x directly in the free troposphere, leading to increased O₃ chemical production in the free troposphere. On the other hand, increased convection could bring O₃-poor air (e.g. from the Pacific “warm pool”) to the upper troposphere which contributes to O₃ decreases there. The O₃ increases at 5–10 km over the tropics in July and into October shown in Fig. 10 are most likely associated with increased convection/lightning. Note that the large negative changes of NO_x above 10 km are partly due to strengthened circulation associated with the double-CO₂ climate forcing, and in part due to increased formation of HNO₃ from NO_x in extratropical latitudes, favoured by the lower temperatures.

In a future climate the OH concentration will be modified following the increase of water vapour, which can subsequently modify the oxidizing capacity of the atmosphere. Responding to a double-CO₂ climate, increases of OH occur throughout the troposphere (Fig. 12b) with an important feedback on the methane lifetime. Methane is an important greenhouse gas and is also a key trace gas controlling background O₃ concentrations. In these calculations the methane lifetime has shortened considerably (by 1.8 years) in response to the double-CO₂ forcing, due not only to increased OH concentrations but also to the increased reaction rate coefficient of OH+CH₄ which has a strongly positive temperature-dependence.

Impacts of the climate change on the chemical and dynamical processes that affect tropospheric O₃ are reflected in the O₃ budget. Budget calculations (Table 3) show that with climate change included, the tropospheric O₃ burden reduces slightly, as a result of several competing processes. The most significant positive feedback is a 80% increase of net flux of O₃ from the stratosphere to the troposphere. Both chemical production and destruction show increases under the climate change. The largest negative chemical change is through reaction O(¹D)+H₂O following photolysis of O₃. Positive chemical changes are mainly through increased reaction fluxes of NO+HO₂

(by 4.5%) and $\text{NO}+\text{CH}_3\text{O}_2$ (by 13%), which lead to O_3 production. We note the larger relative increase in $\text{NO}+\text{CH}_3\text{O}_2$; the driving factor is the strong positive temperature-dependence of the methane oxidation by OH which favours CH_3O_2 production at higher temperatures (Recall that CH_4 concentrations are fixed at the same value for runs B and C). Higher CH_4 levels could have a significant impact on tropospheric O_3 in a future warmer climate. Finally, note that enhanced convection is reflected in a 26% increase of lightning- NO_x emission.

We have shown here that climate change has diverse impacts on O_3 production, loss and transport, and that the oxidizing capacity of the troposphere is modified by climate change. The feedbacks of climate change on tropospheric ozone are complex. In particular, changes of surface O_3 in response to climate change vary regionally and seasonally. More studies are needed to address in detail, for example, how changes in transport patterns can affect surface O_3 changes.

5.3 Response to climate change induced natural emission changes

5.3.1 Increased isoprene emissions

Relatively to scenario C, we increase isoprene emissions by 50% globally to assess the possible impact on O_3 . Note that the major emission regions are in the tropics in the maritime continent, in South America and Africa. The Southeast USA is a regionally important extra tropical source. Our calculation shows that increasing isoprene emissions has little impact on the global tropospheric ozone burden, which decreases by less than 1% (see Table 3). However, the spatial distribution of ozone is modified; Fig. 13a shows that ozone generally increases in the northern hemisphere throughout the model domain and decreases in the equatorial and southern subtropical regions. The largest negative change of O_3 occurs between 5–10 km in the southern tropics where NO_x concentrations are low. Budget calculations (see Table 3) show that with extra isoprene emissions, the gross chemical production is reduced slightly due to a reduced NO_x level which is consumed by elevated RO_2 radicals from isoprene oxida-

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tion to form PAN. O₃ loss increases due to increased destruction by reactions with HO₂ and isoprene, respectively. Following the reduced OH level, the global methane lifetime increases by 0.6 years.

The largest impact of the increased isoprene emissions occurs in summer. Fig. 14a shows O₃ changes at the surface for July; O₃ generally decreases over the isoprene source regions where NO_x levels are also low, as a result of ozone destruction (less O₃ production) in the NO_x-limited regime. In high NO_x regions (Europe and Asia), O₃ increases by up to 4–6 ppbv due to increased peroxy radicals from the degradation of isoprene which contribute positively to ozone production in the NO_x-rich (VOC-limited) environment. We also find elevated ozone concentrations away from the main emitting sources (e.g. over the North Atlantic and western Africa). This suggests that PAN plays an important role in ozone formation; PAN can transport NO_x away from its source and contribute to ozone production in remote regions. In our simulation background O₃ concentrations has increased by around 1 ppbv except over the southern oceans. We will consider a range of future isoprene scenarios in more detail (Young et al., 2007¹; Young, 2007).

Of course, the link between climate and isoprene emission is more complicated than the simple scaling up of the emissions in this sensitivity experiment. Besides the effect of temperature, the magnitude and spatial distribution of isoprene emission is highly dependent on the plant species. Thus, any future natural or anthropogenic land use change, such as the drying of the Amazon rain forest (Cox et al., 2004) or increase in crop growth, would have a large impact on the isoprene emission field. Furthermore, increases in atmospheric CO₂ may well decrease isoprene emission (Rosensteil et al., 2003; Arneth et al., 2007). Other climate-related factors such as water availability, changes in the flux of photosynthetically active radiation (PAR), nutrient delivery and air pollution, will also effect isoprene and other biogenic emissions, either directly or through their impact on primary productivity.

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5.3.2 Increased soil-NO_x emissions

We double the soil-NO_x emission globally in this experiment and compare run E to run C to assess the impact associated with increasing soil-NO_x emissions. The major emissions regions are the tropics and subtropics with the strongest sources from agriculture, grassland, and tropical rain forests. Results show that increased soil NO_x has a substantial positive feedback on tropospheric ozone (3% increase of the tropospheric burden, see Table 3). Both gross chemical production and net chemical production increase compared to run C, which is driven by increased NO_x levels. Figure 13b shows that changes in zonal mean O₃ are positive globally with a peak in the southern subtropics. Increases of O₃ at the surface are largely in the source region but are also due to transport to remote regions through long-range transport (Fig. 14b). This indicates that emission changes in the tropics have a significant impact on O₃ formation. O₃ precursors are subjected to faster transport and are more chemically active in that region.

These calculations of the effects of natural emissions on O₃ are very simple and the results are merely indicative of possible impacts. However the impacts are potentially significant, and more detailed studies are needed to project future changes of natural emissions which can be included in models.

6 Conclusions

We have evaluated an updated tropospheric chemistry model which is incorporated into a version of the UK Met Office climate model. The model is satisfactory in modelling present-day observed tropospheric ozone and nitrogen species. The ozone budget falls within reported ranges. We calculate a net stratospheric to tropospheric ozone flux of 452 Tg/year, a gross O₃ chemical production of 3620 Tg(O₃)/year, and a gross O₃ chemical destruction of 3108 Tg/year. However, the gross chemical production of O₃ is relatively low compared to a recent multimodel study.

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Calculations for a series of 2100 scenarios suggest that projected significant increase of anthropogenic emissions of ozone precursors could contribute to large ozone increases throughout the troposphere. A pessimistic (large emissions) scenario (SEES A2) leads to an unacceptable increases of surface ozone. Such that these would be significant exceedences of suggested health-related thresholds. An assessment of the impact of climate change on global tropospheric ozone reveals a number of important feedbacks. Increased water vapour leads to increased O₃ destruction in the tropics, whereas enhanced stratosphere-troposphere exchange increases the net O₃ flux to the troposphere. The O₃ changes at the surface in a future climate are complex and regionally varying, and are strongly influenced by changes in temperature, humidity, STE, and hemispheric transport patterns. We pay attention to some positive changes: in particular, we find elevated O₃ over polluted continents especially during summer months; increase of background O₃ over the Southern Ocean and the Antarctic during austral winter/spring; and intensified poleward transport of pollutants from Europe and North America leading to elevated O₃ in the Arctic, in particular during winter/spring. Recent studies of the response of tropospheric O₃ to climate change reveal diverse model responses (Shindell et al., 2006; Brasseur et al., 2006). Multi-model studies are important to achieve a consensus on the impact of future climate change on tropospheric O₃, in particular at ground-level.

Changes in convection in a double CO₂ climate can modify the NO_x distribution; Enhanced convection lifts NO_x and other ozone precursors more efficiently in the tropical region which contribute positively to the O₃ chemical production through elevated NO_x and HO_x in that region. The associated change in lightning-produced NO_x is approximately a 22% increase in our calculation and contributes positively to tropospheric ozone formation. However, the effect of the convection on ozone budgets are still uncertain (Doherty et al., 2005) and further studies are needed to quantify to what extent the tropospheric ozone budget is influenced.

Climate change modifies the tropospheric oxidizing capacity considerably. The methane lifetime is shortened by 1.8 years when climate change is included in the

calculation, due to increased OH concentrations in a more humid climate. In contrast, the methane lifetime was not changed significantly in run B which considered just the increases of anthropogenic emissions. In a warmer and wetter climate, methane can play a significant role in ozone formation due to the strong positive temperature dependence of its oxidation rate coefficient. Further studies are needed to assess the role of methane on ozone formation, particularly in a changed climate.

In addition to considering changing anthropogenic emissions in the 2100 climate change experiment, we have also examined idealised changes of some natural emissions and their impact on tropospheric ozone. With a 50% increase of the isoprene emission, changes in surface ozone range from -8 ppbv to 6 ppbv. The impacts are regionally varying and have a relatively strong seasonal cycle. The largest decreases of surface ozone occur over isoprene source regions (Amazonia, US, Africa and South East Asia) and the largest increases are over China and Europe in summer where NO_x levels are high. Generally the response to changed isoprene depends on whether the chemical regime is NO_x - or VOC-limited: so we predict zonal mean ozone decrease in the southern hemisphere and tropics, with ozone increases in the north hemisphere. The increased isoprene emission increases methane lifetime by 0.6 years, which is large compared with changing anthropogenic emissions. By doubling soil- NO_x emissions, we obtain substantial increases of ozone over emitting regions and a 0.4 years reduction on methane lifetime. Changes in biogenic emissions, which are mainly from tropics and subtropics, can significantly affect the oxidizing capacity of the troposphere. More study is needed to reduce the uncertainty in present estimates of natural emissions. Accurate projections of the future biogenic emissions are crucial to assessing chemistry-climate-biosphere feedbacks.

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Table 1. Simulations.

Simulations	Emissions	Meteorology	Description
Run A	IIASA-2000	2000s	Present-day
Run B	A2-2100	2000s	Future (A+ Δ anthropogenic emissions)
Run C	A2-2100	2100s	Future (B+climate change)
Run D	A2-2100+ Δ isoprene	2100s	Future (C+ Δ isoprene emissions)
Run E	A2-2100+ Δ soil-NO _x	2100s	Future (C+ Δ soil-NO _x emissions)

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Table 2. Summary of annual total emissions.

	IIASA-2000	A2-2100(B)	A2-2100(C)	A2-2100(D)	A2-2100(E)
CO (Tg CO)					
Fossil fuel	470	1720	1720	1720	1720
Biomass burning	507	507	507	507	507
Ocean/vegetation	100	100	100	100	100
NO _x (Tg NO ₂)					
Fossil fuel	91.4	333	333	333	333
Biomass burning	33.4	33.4	33.4	33.4	33.4
Soil	23	23	23	23	46
Aircraft	2.58	5.67	5.67	5.67	5.67
Lightning	10.3	10.3	12.6	12.6	12.6
NMVOCs (Tg VOC)					
Industrial source	116	306	306	306	306
Biomass burning	31.2	31.2	31.2	31.2	31.2
Natural (isoprene)	580	580	580	870	580
CH ₄ (ppbv)	1760	3731	3731	3731	3731

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Table 3. Tropospheric budgets. Units are Tg/yr unless stated differently.

	2000 (A)	2100 (B)	2100 (C)	2100 (D)	2100 (E)
NO+HO ₂	2464	5095	5322	5311	5556
NO+CH ₃ O ₂	811	1833	2074	2014	2179
NO+RO ₂	346	625	627	695	643
Total chemical production	3620	7550	8022	8018	8376
O ¹ D+H ₂ O	1737	3167	3794	3756	3943
OH+O ₃	343	667	749	707	806
HO ₂ +O ₃	965	2239	2384	2441	2477
Isoprene+O ₃	64	57	54	106	52
Total chemical loss	3108	6128	6980	7029	7276
Net chemical production	512	1422	1042	989	1100
STE	452	430	773	774	772
Dry deposition	1035	1767	1695	1684	1746
O ₃ burden (Tg)	314	549	530	527	546
OH burden (Mg)	172	201	223	207	234
Global mean CH ₄ lifetime (yr)	11.3	11.2	9.4	10.0	9.0
Lightning NO _x emission (Tg[NO ₂]/yr)	10.3	10.3	12.6	12.6	12.6

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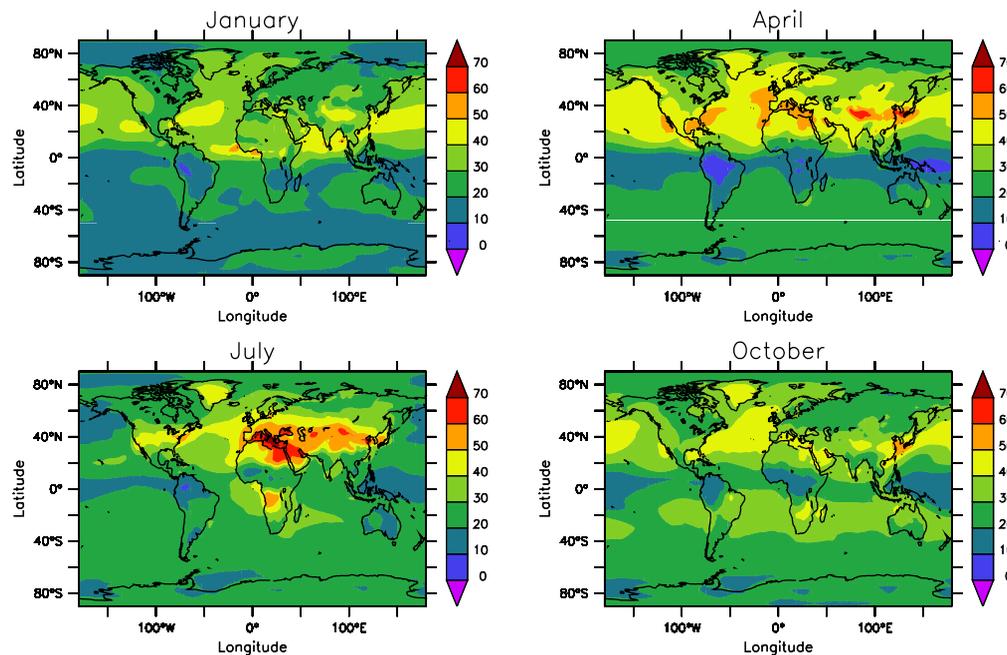


Fig. 1. Modelled surface O_3 (ppbv) in January, April, July and October from the present-day simulation.

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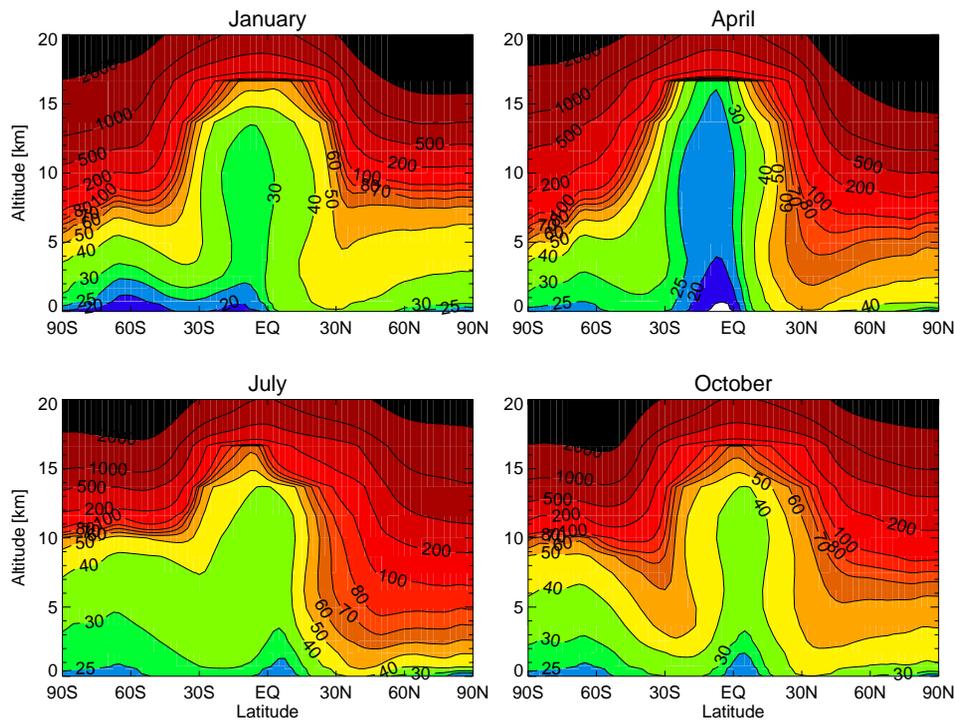
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Fig. 2. Zonal mean O_3 (ppbv) profile in January, April, July and October from the present-day simulation.

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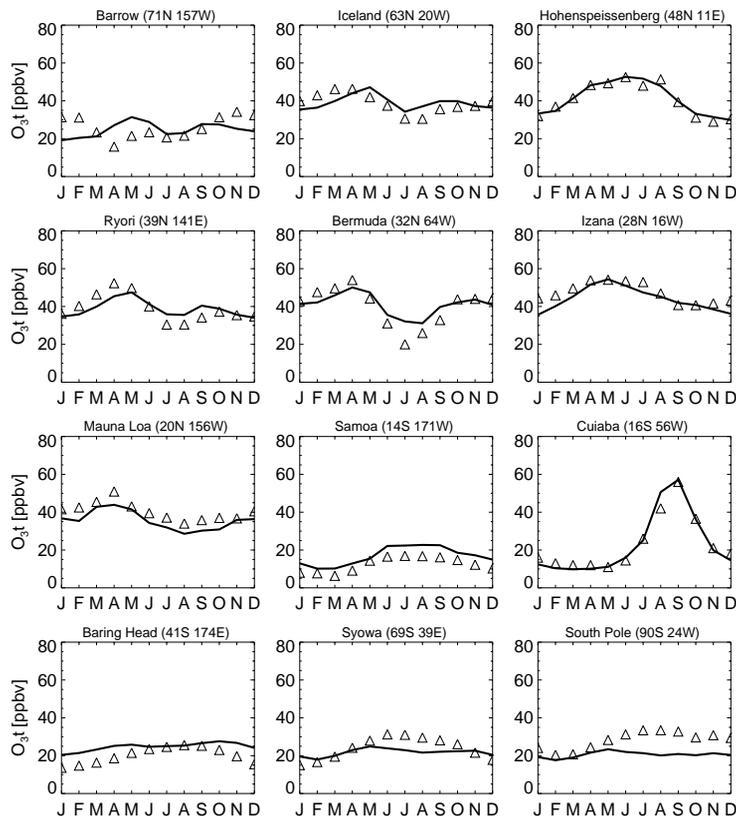


Fig. 3. Observed (symbols) and simulated (lines) surface ozone (ppbv). Data from the World Data Centre for Surface Ozone (see text).

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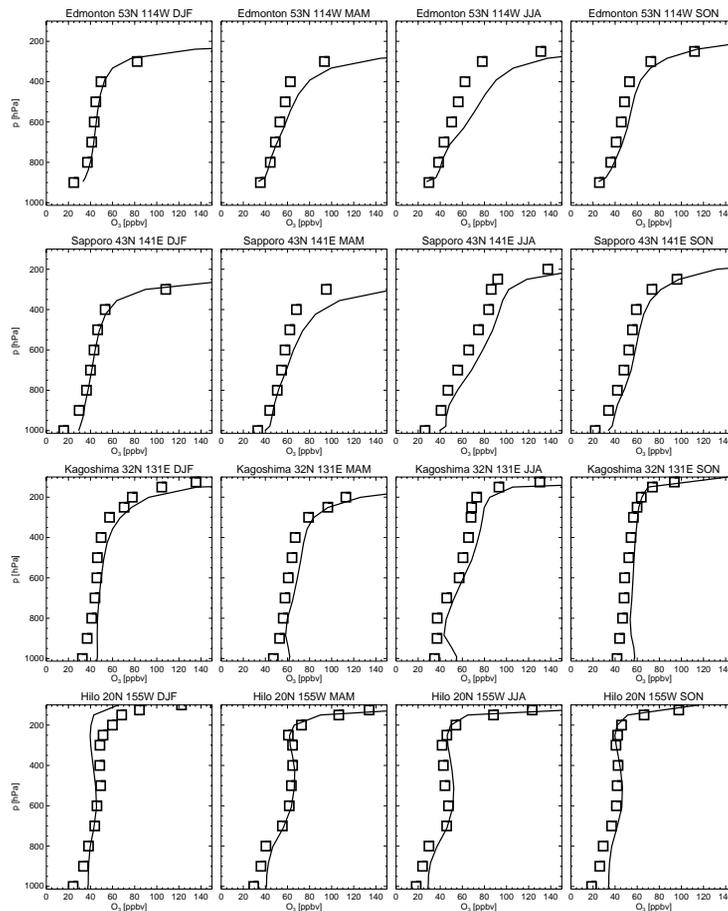


Fig. 4. Seasonal averaged O_3 profile (ppbv) by measurements (symbols) and by calculations (lines). Data from Logan (1999).

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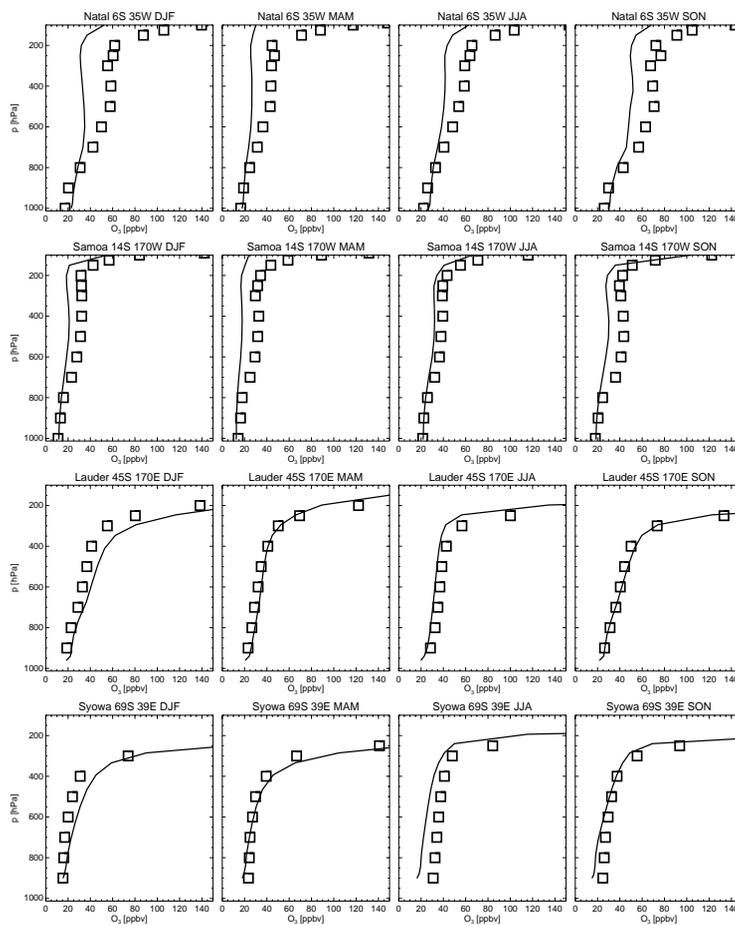


Fig. 4. Continued.

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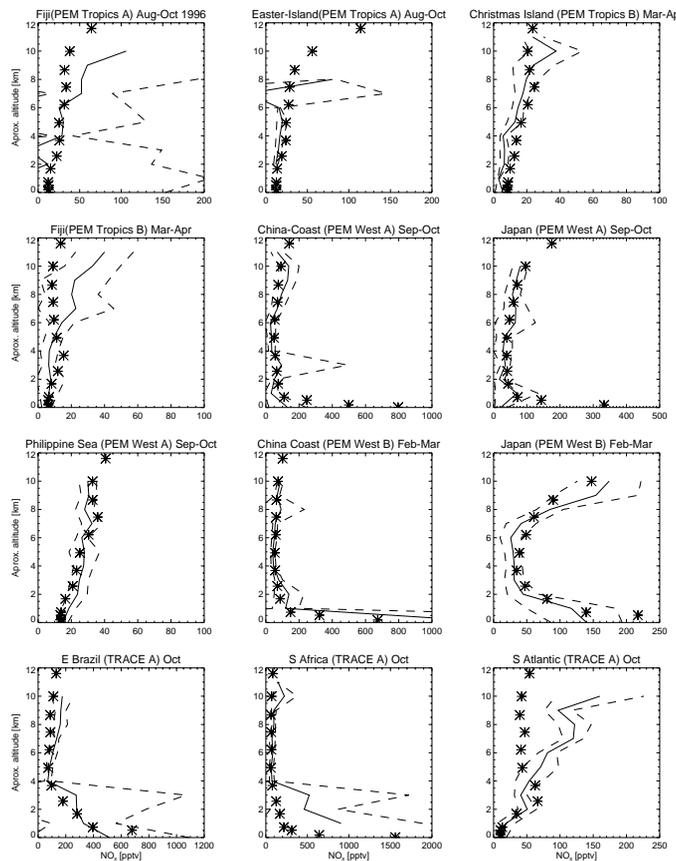


Fig. 5. Observed and simulated profiles of NO_x (pptv) for various locations and seasons. Solid and dashed lines indicate measured mean values and standard deviations respectively. Model calculations are indicated by symbols. Observation data are taken from Emmons et al. (2000).

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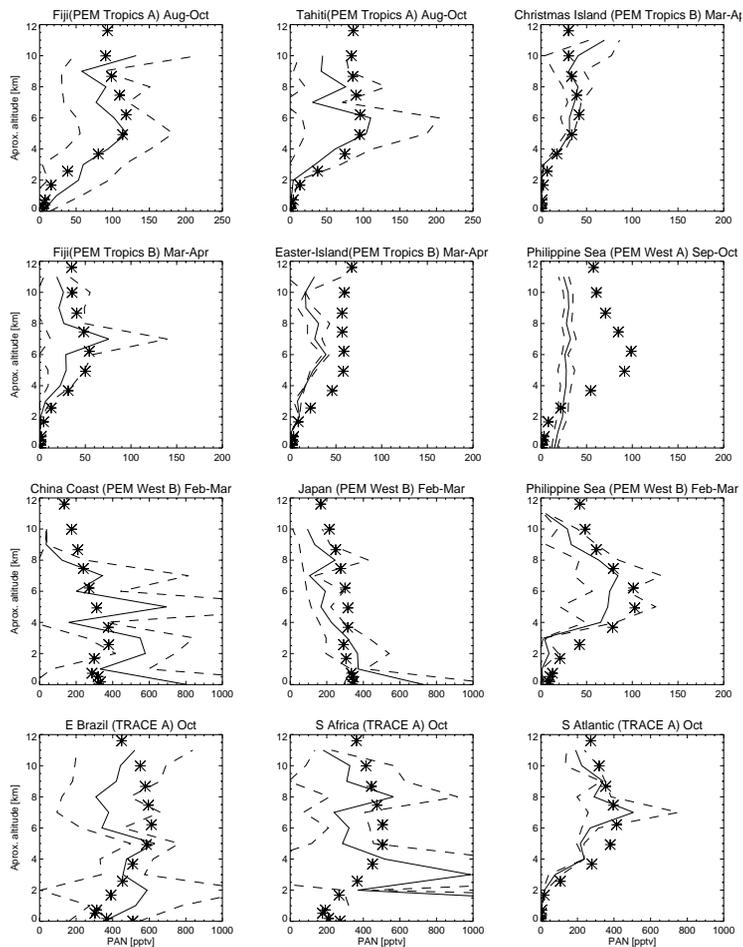


Fig. 6. Observed and simulated PAN (pptv) as indicated in Fig. 5.

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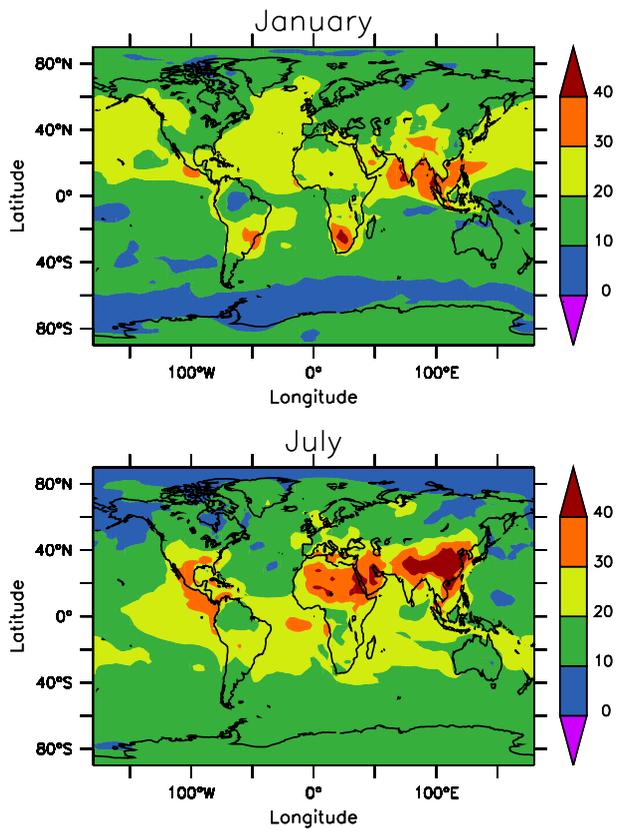


Fig. 7. Changes in surface O₃ (ppbv) between 2000 and 2100 due to anthropogenic emission changes, for January and July.

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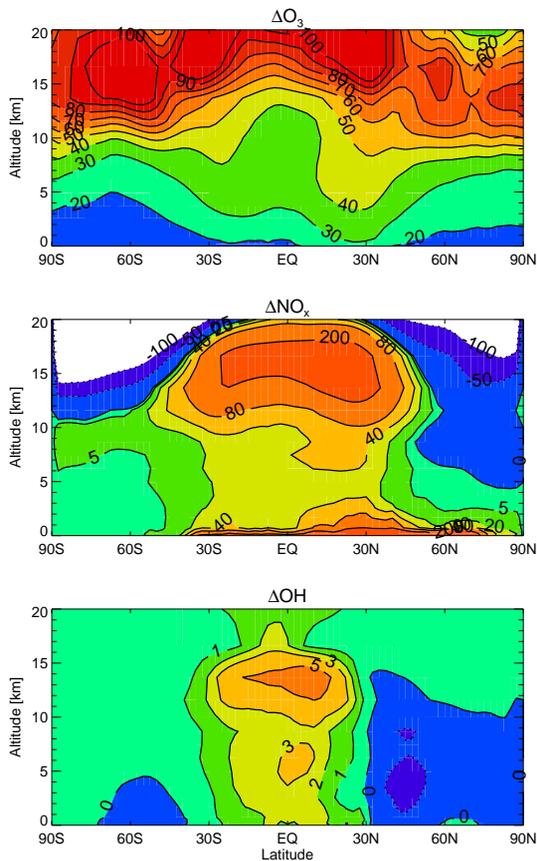


Fig. 8. Changes in zonal and annual mean O_3 (ppbv), NO_x (pptv) and OH (10^5 molecules/cm³) between 2000 and 2100 due to anthropogenic emission changes.

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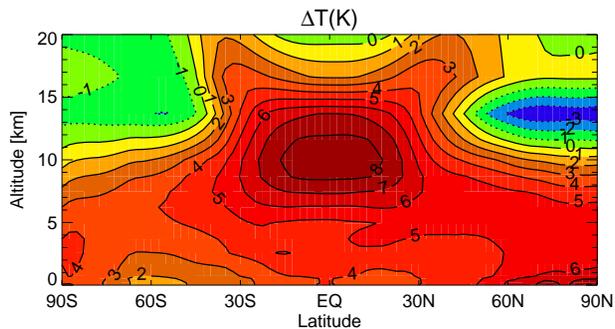


Fig. 9. Changes in zonal and annual mean temperature (K) between 2000 and 2100 (double CO_2).

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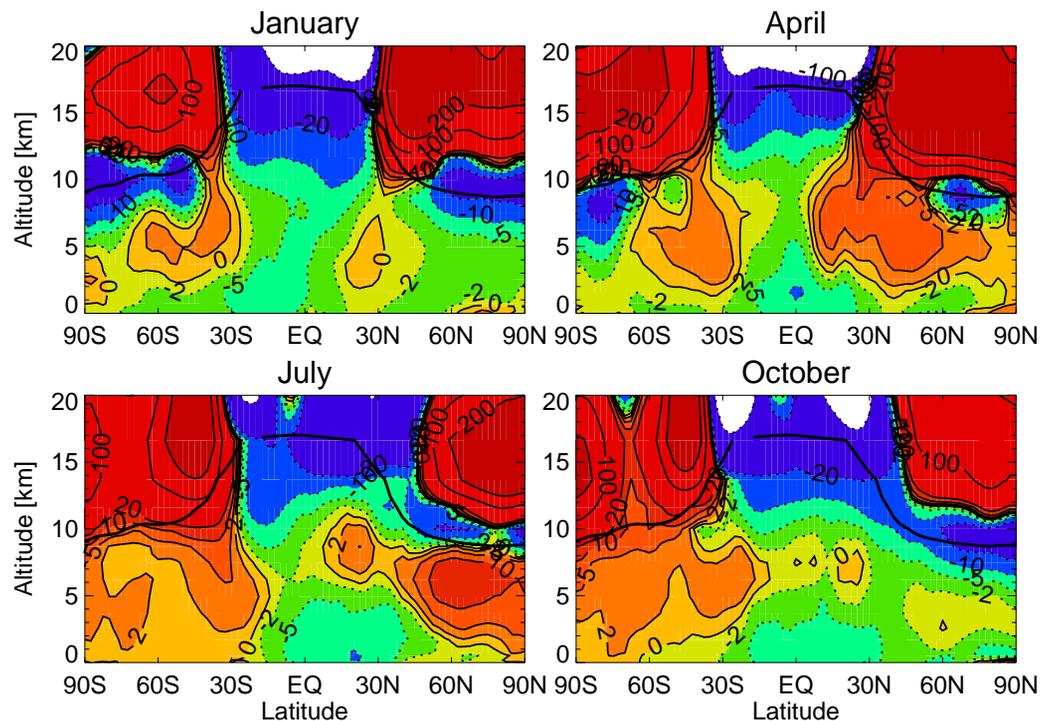


Fig. 10. Changes in zonal mean O₃ (ppbv) between 2000 and 2100 due to climate change (C-B), for January, April, July and October.

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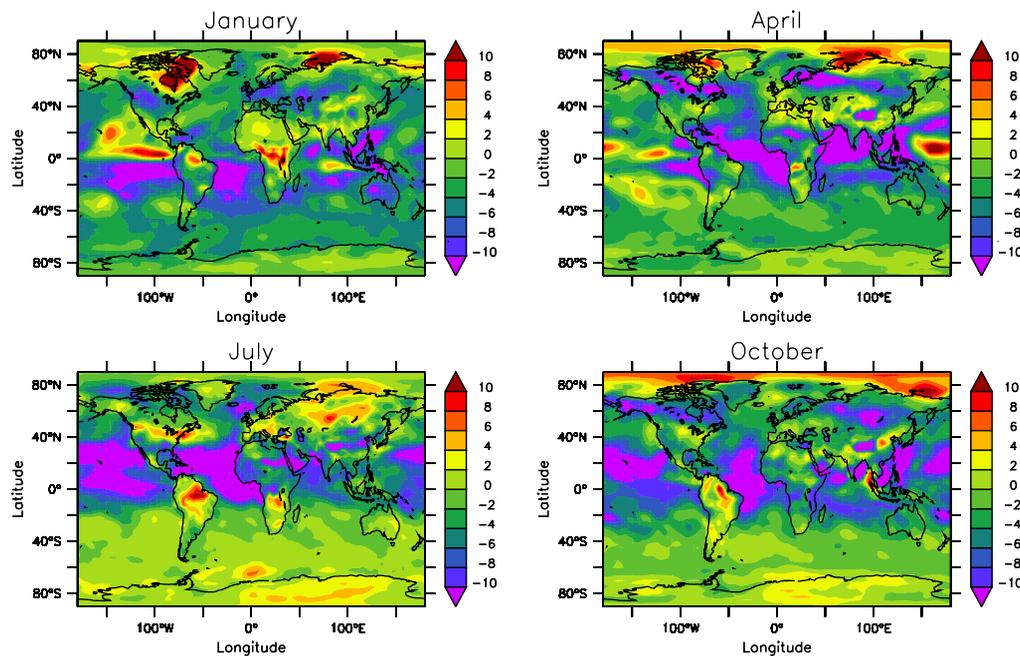


Fig. 11. Changes in surface O₃ (ppbv) between 2000 and 2100 due to climate change, for January, April, July and October.

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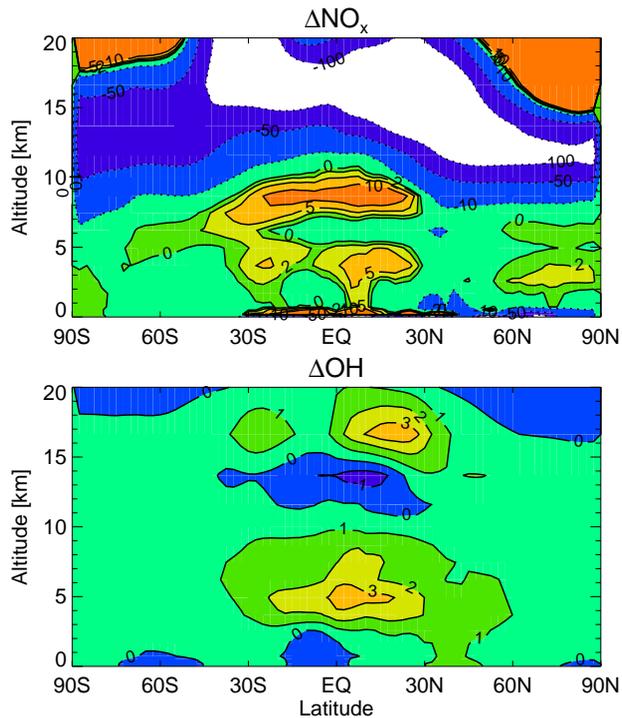


Fig. 12. Changes in annual and zonal mean NO_x (pptv) and OH (10^5 molecules/cm³) due to climate change.

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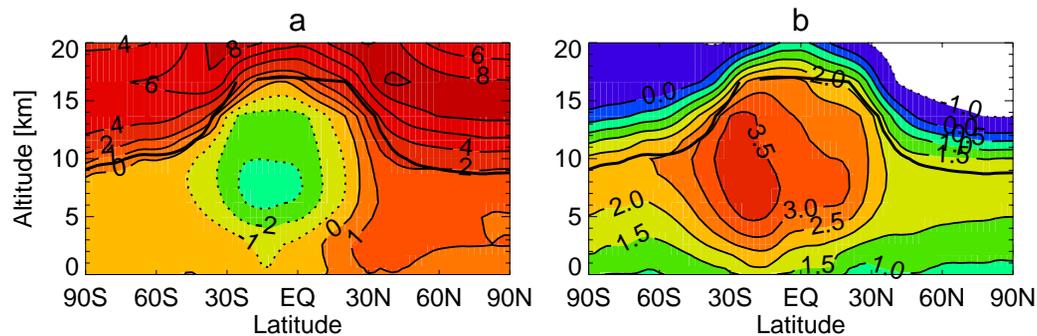


Fig. 13. Changes in annual and zonal mean O₃ (ppbv) due to **(a)** increase of isoprene emissions (D-C) and **(b)** increase of soil-NO_x emissions (E-C).

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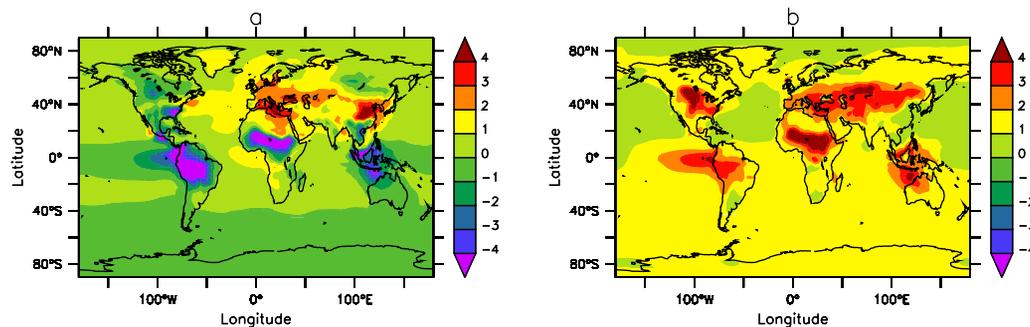


Fig. 14. Changes in surface O_3 (ppbv) due to **(a)** increase of isoprene emissions (D-C) and **(b)** increase of soil- NO_x emissions (E-C). Both changes are for July.

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