

**Bromoform in the  
TTL**

B.-M. Sinnhuber and  
I. Folkins

# Estimating the contribution of bromoform to stratospheric bromine and its relation to dehydration in the tropical tropopause layer

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## Abstract

The contribution of bromoform to the stratospheric bromine loading is estimated using the one-dimensional tropical mean model of [Folkins and Martin \(2005\)](#), which is constrained by observed mean profiles of temperature and humidity. In order to reach the stratosphere, bromoform needs to be lifted by deep convection into the tropical tropopause layer (TTL), above the level of zero radiative heating. The contribution of bromoform to stratospheric bromine depends then critically on the rate of removal of the degradation products of bromoform (collectively called  $\text{Br}_y$  here) from the TTL, which is believed to be due to scavenging by falling ice. This relates the transport of short-lived bromine species into the stratosphere to processes of dehydration in the TTL. In the extreme case of dehydration occurring only through overshooting deep convection, the loss of  $\text{Br}_y$  from the TTL may be negligible and consequently bromoform will fully contribute with its boundary layer mixing ratio to the stratospheric bromine loading, i.e. with 3 pptv for an assumed 1 pptv of bromoform in the boundary layer. For the other extreme that  $\text{Br}_y$  is removed from the TTL almost instantaneously, the model calculations predict a contribution of about 0.5 pptv for the assumed 1 pptv of boundary layer bromoform. While this gives some constraints on the contribution of bromoform to stratospheric bromine, it is argued that a more precise number cannot be given until the mechanisms of dehydration in the TTL are better understood.

## 1. Introduction

In recent years it has been realized that in addition to the long-lived ozone depleting substances such as the CFCs or halons, also short-lived halogen compounds can contribute significantly to the stratospheric halogen loading ([Ko et al., 2003](#)). In particular short-lived bromine compounds can have a relatively large impact, as already a few parts per trillion by volume (pptv) of extra bromine in the stratosphere from short-lived source gases can lead to significant increases in stratospheric ozone depletion ([Salaw-](#)

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itch et al., 2005).

Bromoform ( $\text{CHBr}_3$ ) is probably the most abundant short-lived bromine specie and may be an important source for stratospheric bromine (Sturges et al., 2000). Bromoform is mainly emitted from the oceans with large regional differences in the emission rates and near surface air concentrations (Quack and Wallace, 2003; Quack et al., 2004). Average bromoform mixing ratios in the marine boundary layer are probably in the range of about 1 to 2 pptv (Ko et al., 2003), i.e., carrying about 3 to 6 pptv of bromine, but locally much higher mixing ratios have been observed (Quack and Wallace, 2003).

Two previous model studies (Dvortsov et al., 1999; Nielsen and Douglass, 2001) have concluded that bromoform may contribute about 1 pptv to the stratospheric bromine loading. Both model results indicated that about half of the supply of bromine to the stratosphere comes from oxidation of bromoform in the troposphere which is then transported as inorganic bromine ( $\text{Br}_y$ ) to the stratosphere. In addition to uncertainties in the rate of convective transport of bromoform into the upper troposphere and lower stratosphere, the critical point in these studies is the lifetime of  $\text{Br}_y$  in the troposphere.

In steady state the rate of  $\text{Br}_y$  production due to the chemical loss of  $\text{CHBr}_3$  (ignoring other sources of  $\text{Br}_y$  for the moment) equals the rate of loss of  $\text{Br}_y$  due to washout:

$$\frac{[\text{Br}_y]}{\tau_w} = 3 \times \frac{[\text{CHBr}_3]}{\tau_{\text{chem}}}. \quad (1)$$

Here  $\tau_{\text{chem}}$  is the chemical lifetime of  $\text{CHBr}_3$  and  $\tau_w$  is the lifetime of  $\text{Br}_y$  due to washout. The factor of 3 comes from the fact that each molecule of  $\text{CHBr}_3$  contains 3 bromine atoms. For typical values of  $\tau_{\text{chem}}$  and  $\tau_w$ , which are both in the order of 10 days (see below), 1 pptv of  $\text{CHBr}_3$  could sustain about 3 pptv of  $\text{Br}_y$ . Under these assumptions, the total amount of bromine that could be transported into the stratosphere is then according to Eq. (1) given by  $3[\text{CHBr}_3] (1 + \tau_w/\tau_{\text{chem}})$ , i.e., it depends on the ratio between the lifetime of  $\text{Br}_y$  due to washout and the chemical lifetime of  $\text{CHBr}_3$  (Ko et al., 1997).

The most important pathway of air entering the stratosphere is in the tropics through

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the tropical tropopause layer (TTL). In order to reach the stratosphere, air masses need to be lifted by deep convection into the TTL, above the level of zero radiative heating (Fig. 1). The level of zero radiative heating marks the transition from large-scale subsidence to large scale ascent, as diagnosed from diabatic heating rates ( $\omega_r$ ).

5 The divergence in the large-scale vertical velocities is then balanced by detrainment from deep convection (with detrainment rate  $d_c$ ). Note that only air masses detraining from deep convection above the level of zero radiative heating (at an altitude of about 15.5 km, or about 365 K potential temperature) will reach the stratosphere. This makes the level of zero radiative heating a useful definition for the lower boundary of the TTL.

10 The other important level to be considered here is the cold point (at an altitude of about 17 km or about 380 K potential temperature): air masses reaching the stratosphere need to be dehydrated to mixing ratios corresponding roughly to the saturation mixing ratio at the cold point. The mechanism of dehydration in the TTL is not well understood and different hypothesis are currently being discussed. In one possible  
15 scenario, dehydration occurs gradually during large scale ascent. It has been realized (Holton and Gettelman, 2001) that during the relatively slow ascent through the TTL there is a high probability for airmasses being transported horizontally through the coldest regions of the tropical tropopause. In such a scenario, scavenging by falling ice could lead to some loss of  $\text{Br}_y$  from the TTL. In a different scenario (Sherwood and  
20 Dessler, 2000), dehydration may occur completely in overshooting convection so that air detraining from deep convection is already dry. In such a scenario, the loss of  $\text{Br}_y$  from the TTL may be negligible. In any case, the rate of removal of  $\text{Br}_y$  from the TTL is highly uncertain, depending not only on the mechanism of dehydration, but also on the chemical partitioning of  $\text{Br}_y$  in the TTL.

25 Here in this paper, the contribution of bromoform to the stratospheric bromine loading is estimated using the one dimensional tropical mean model of Folkins and Martin (2005), which is constrained by observed mean profiles of temperature and humidity. This removes some of the uncertainty of previous studies that relied on convective mass fluxes from parameterized convection schemes (e.g., Nielsen and Douglass,

2001). Because the rate of removal of Br<sub>y</sub> from the TTL is highly uncertain, depending on the mechanism of dehydration in the TTL, we perform calculations over a range of Br<sub>y</sub> lifetimes.

## 2. Model

### 2.1. One-dimensional model

The model calculations in this work use tropical mean convective mass fluxes from the diagnostic one-dimensional model of [Folkins and Martin \(2005\)](#), which is constrained by observed profiles of temperature and humidity. Briefly, downward mass fluxes in the clear sky part of the tropical atmosphere are calculated from radiative and evaporative cooling rates:

$$\omega_r = \frac{Q_r}{\sigma}$$

$$\omega_e = -\frac{L_v e}{c_p \sigma},$$

with  $Q_r$  the radiative cooling rate,  $e$  the evaporative moistening rate,  $L_v$  the heat of vaporization,  $c_p$  the specific heat, and  $\sigma$  the static stability:

$$\sigma = -\frac{T}{\theta} \frac{\partial \theta}{\partial p}.$$

It is then assumed that the clear sky downward mass flux is balanced by an upward convective mass flux ( $\omega_c$ ) so that

$$\omega_r + \omega_e + \omega_c = 0.$$

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The vertical divergence of the convective mass flux then gives the convective detrainment rate

$$d_c = \frac{\partial (\omega_r + \omega_e)}{\partial p}.$$

Profiles of  $\omega = \omega_r + \omega_e$  and  $d_c$  are shown in Fig. 2. Note that  $d_c$  is zero at and above the cold point. This is of some importance for the interpretation of the results of the present study, as it is assumed here that  $\text{Br}_y$  is removed from the atmosphere by falling rain or ice only below the cold point. The convective turnover time (inverse of detrainment rate) used in the present study agrees well with estimates from Dessler (2002) in the TTL, based on measurements of ozone and carbon monoxide. However, while the detrainment rate from Folkins and Martin (2005) is zero above 17 km, Dessler (2002) estimates a convective turnover time of 50 to 90 days at 17 km.

Using the vertical mass fluxes  $\omega_r + \omega_e$  and the detrainment rate  $d_c$  from Folkins and Martin (2005), tropical mean trace gas profiles can be calculated according to the following equation:

$$\frac{\partial [X]}{\partial t} + (\omega_r + \omega_e) \frac{\partial [X]}{\partial p} = P - L[X] + d_c ([X]_c - [X]). \quad (2)$$

Here  $[X]$  is the trace gas mixing ratio,  $P$  and  $L$  are the chemical production and loss rates, respectively, and  $[X]_c$  is the trace gas mixing ratio of air parcels detraining from convective clouds, which is essentially a free parameter of the model. Folkins and Martin (2005) show that ozone profiles calculated from Eq. (2) agree well with observed tropical mean ozone profiles.

Here we now use Eq. (2) to calculate tropical mean profiles of bromoform and  $\text{Br}_y$ :

$$\frac{\partial [\text{CHBr}_3]}{\partial t} + (\omega_r + \omega_e) \frac{\partial [\text{CHBr}_3]}{\partial p} = -\frac{[\text{CHBr}_3]}{\tau_{\text{chem}}} + d_c ([\text{CHBr}_3]_c - [\text{CHBr}_3]) \quad (3)$$

$$\frac{\partial [\text{Br}_y]}{\partial t} + (\omega_r + \omega_e) \frac{\partial [\text{Br}_y]}{\partial p} = 3 \frac{[\text{CHBr}_3]}{\tau_{\text{chem}}} - \frac{[\text{Br}_y]}{\tau_w} + d_c (0 - [\text{Br}_y]). \quad (4)$$

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For the calculation of bromoform, the surface mixing ratio is kept fixed (at 1 pptv unless otherwise noted) and it is assumed that air detraining from deep convection also carries 1 pptv of bromoform. There is no in situ production of bromoform and the loss rate is specified by the chemical lifetime  $\tau_{\text{chem}}$  of  $\text{CHBr}_3$  due to photolysis and reaction with OH, as discussed in the next section. It is then assumed that every molecule of bromoform lost produces 3 molecules of  $\text{Br}_y$ . The loss of  $\text{Br}_y$  due to washout is then further specified by a free parameter  $\tau_w$  that in the model calculations presented here is constant throughout the troposphere up to the cold point. Because the major  $\text{Br}_y$  species and in particular HBr are highly soluble, it is assumed here that air detraining from convection has zero  $\text{Br}_y$ .

The set of Eqs. (3) and (4) is solved numerically on a vertical grid with 200 m resolution using an upwind advection scheme with leap-frog time stepping over 500 days with a timestep of 0.01 days. The relatively long integration time is needed to reach a steady state solution in the lower stratosphere.

## 2.2. Bromoform chemistry

Loss of bromoform occurs mainly by photolysis and reaction with OH (Ko et al., 2003). Photolysis of bromoform is calculated here using the temperature dependent absorption cross sections from Sander et al. (2002) for a tropical atmosphere and averaged over the diurnal and annual cycles. Loss by reaction with OH is calculated using the temperature dependent reaction rate constant from Sander et al. (2002) and the idealized OH profile of Nielsen and Douglass (2001). The resulting loss rates and corresponding lifetime of bromoform are shown in Fig. 3. It is clear that uncertainties in the assumed OH profile will have only a minor impact on the calculated total bromoform loss rate because except for the lower troposphere the total loss is dominated by the photolysis rate.

### 3. Results and discussion

The calculated tropical mean bromoform profile using the mass fluxes and chemical loss rates of Sect. 2 is shown in Fig. 4. The model calculations are compared with bromoform observations from the NASA DC-8 aircraft during the PEM Tropics-A campaign and from the NASA WB57 high-altitude aircraft during the ACCENT campaign. The observations during PEM Tropics-A were performed in the Central and Eastern Pacific during August to October 1996. The PEM-Tropics data shown here were taken from the compilation of aircraft measurements by Emmons et al. (2000). Observations during ACCENT were performed during April and September 1999 between 5° N and 40° N at around 90° W (e.g., Tuck et al., 2004). Indicated in Fig. 4 are the mean values and the range between minimum and maximum values over all measurements during the two campaigns. The model calculations of bromoform agree well with the mean bromoform profile from the two campaigns, in particular if a mean boundary layer and detrainment mixing ratio of 0.75 pptv is assumed in the model calculations (dashed line). (Note that the model results are linear in the assumed detrainment mixing ratios.) However, one has to be careful not to overinterpret the comparison between model and observations. It is not clear how representative the bromoform observations during the PEM Tropics-A and ACCENT campaigns are for the tropics in general, given the relatively large variability of bromoform. Observations during the STRAT campaign in 1996 showed much smaller bromoform values in the upper troposphere and lower stratosphere (Nielsen and Douglass, 2001), maybe because the observations during STRAT were less influenced by deep convection than the observations during ACCENT.

The model calculations of total bromine emitted from bromoform (defined as  $Br_y + 3 \times CHBr_3$ ) are shown in Fig. 5. The results shown here assume a mean boundary layer and detrainment mixing ratio of 1 pptv. As the model results are linear in the bromoform detrainment mixing ratios they can be directly scaled to give results for different boundary layer bromoform mixing ratios. The calculations here are performed for a range of  $Br_y$  lifetimes between 0 days (i.e. instantaneous removal of  $Br_y$ ) and in-

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5 finity (i.e. no loss of  $\text{Br}_y$  from the TTL). Most model calculations assume or compute a washout lifetime of  $\text{Br}_y$  of about 10 to 30 days (Dvortsov et al., 1999; Nielsen and Douglass, 2001; von Glasow et al., 2004). This would result in about 1 pptv of total bromine in the stratosphere due to bromoform (assuming a mean boundary layer mixing ratio of 1 pptv), confirming the results of the previous studies by Dvortsov et al. (1999) and Nielsen and Douglass (2001).

10 For a  $\text{Br}_y$  lifetime of 0 days the contribution of bromoform to stratospheric bromine is about 0.5 pptv. This situation constitutes a lower limit for the bromine contribution due to bromoform and could – at least in principle – be derived from bromoform measurements in the TTL. At the opposite extreme is the situation, where the loss of  $\text{Br}_y$  from the TTL is negligible (infinite  $\text{Br}_y$  lifetime). For this case we expect that the mixing ratio of bromoform detraining from deep convection in the TTL will fully contribute to the stratospheric bromine loading, i.e., the assumed 1 pptv of bromoform in the boundary layer will add 3 pptv to the stratospheric bromine loading. However, this situation constitutes only an upper boundary for the contribution of bromoform to the stratospheric bromine loading if it is assumed that air detraining from deep convection does not contain any  $\text{Br}_y$ . In reality a certain fraction of the inorganic bromine ( $\text{Br}_y$ ) in the troposphere will consist of species with low solubility (most importantly bromine monoxide,  $\text{BrO}$ ) so that  $\text{Br}_y$  could be transported into the TTL by deep convection. 20 Unfortunately the tropospheric bromine chemistry is not well understood at present: Indications for a global mean background of about 1 pptv of  $\text{BrO}$  in the troposphere (e.g., Sinnhuber et al., 2005) would require – according to our present understanding of the tropospheric bromine chemistry – an additional unknown source of bromine (von Glasow et al., 2004).

25 Without a better understanding of the  $\text{Br}_y$  lifetime in the TTL – and thus a better understanding of the mechanisms of dehydration in the TTL – it will probably not be possible to get a precise estimate for the contribution of bromoform to stratospheric bromine. From the present study our best estimate is somewhere between 1 and 3 pptv. This number is basically consistent with estimates of the contribution from

short-lived bromine compounds based on measurements of BrO in the stratosphere (Pfeilsticker et al., 2000; Sinnhuber et al., 2002, 2005; Salawitch et al., 2005).

The results presented here for bromoform could in principle also be applied to other short-lived bromine gases such as CH<sub>2</sub>BrCl, CH<sub>2</sub>Br<sub>2</sub>, CHBrCl<sub>2</sub> and CHBr<sub>2</sub>Cl.

5 However, these substances have lifetimes of 70 to 150 days (Ko et al., 2003) as compared to about 25 days for CHBr<sub>3</sub>. Moreover, they contain only one or two bromine atoms, as compared to three for CHBr<sub>3</sub>. Consequently, for these substances the transport of degradation products (Br<sub>y</sub>) into the stratosphere will be less important than it is for bromoform. Thus alternatively their contribution to the stratospheric bromine  
10 loading may be estimated similar to that of long-lived species, e.g. through observed correlations with CFC-11 (Wamsley et al., 1998).

#### 4. Conclusions

We have presented an estimate for the contribution of bromoform to the stratospheric bromine loading. Using reasonable assumptions for the mean bromoform mixing ratio in the tropical boundary layer of about 1 pptv and for the tropospheric lifetime of Br<sub>y</sub> due to washout of 10 to 30 days the calculations show that bromoform contributes  
15 about 1 pptv to the stratospheric bromine loading. This confirms earlier results that used similar assumptions (Dvortsov et al., 1999; Nielsen and Douglass, 2001). One improvement compared to these earlier studies is that here we use tropical mean convective mass fluxes that are diagnosed from observations. This removes part of the  
20 uncertainty of previous studies that rely on convective mass fluxes from parameterized convections schemes. The fact that the calculated tropical mean bromoform profile agrees well with available observations gives some confidence. However, given the relatively large variability of bromoform in the atmosphere it is not clear how representative the available observations are.  
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The contribution of bromoform to the stratospheric bromine loading depends critically on the loss of Br<sub>y</sub> due to falling rain or ice. In fact, what is of importance here is

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the lifetime of  $\text{Br}_y$  in the TTL between the level of zero radiative heating and the cold point. Knowledge of the mechanisms and rates of dehydration in the TTL are very limited at present, which limits our ability to estimate the contribution of bromoform to stratospheric bromine.

5 In the extreme case that all dehydration in the TTL occurs already in overshooting deep convection (Sherwood and Dessler, 2000) the loss of  $\text{Br}_y$  from the TTL may be negligible. In this case bromoform will contribute fully with its boundary layer mixing ratio to the stratospheric bromine loading, i.e., with 3 pptv for an assumed mean boundary layer mixing ratio of 1 pptv. In principle this value could be even higher if also  
10 degradation products of bromoform are transported into the TTL by deep convection. (Our calculations here assumed that the degradation products – collectively called  $\text{Br}_y$  here – are soluble and will be removed in deep convecting clouds.)

Our ability to quantify the contribution of bromoform to stratospheric bromine is thus not only limited by uncertainties in the strength and distribution of bromoform sources  
15 and the transport of bromoform into the tropopause region, but also by the loss rate of  $\text{Br}_y$  in the TTL, which depends on the mechanism of dehydration in the TTL.

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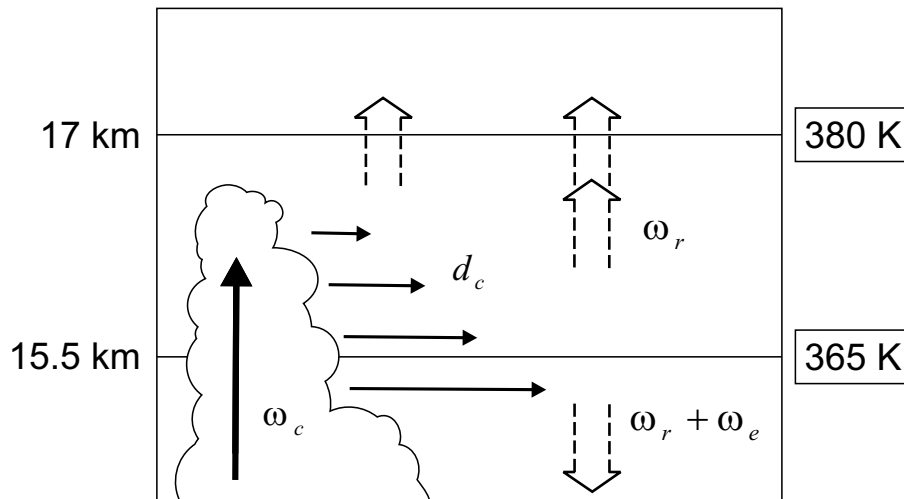
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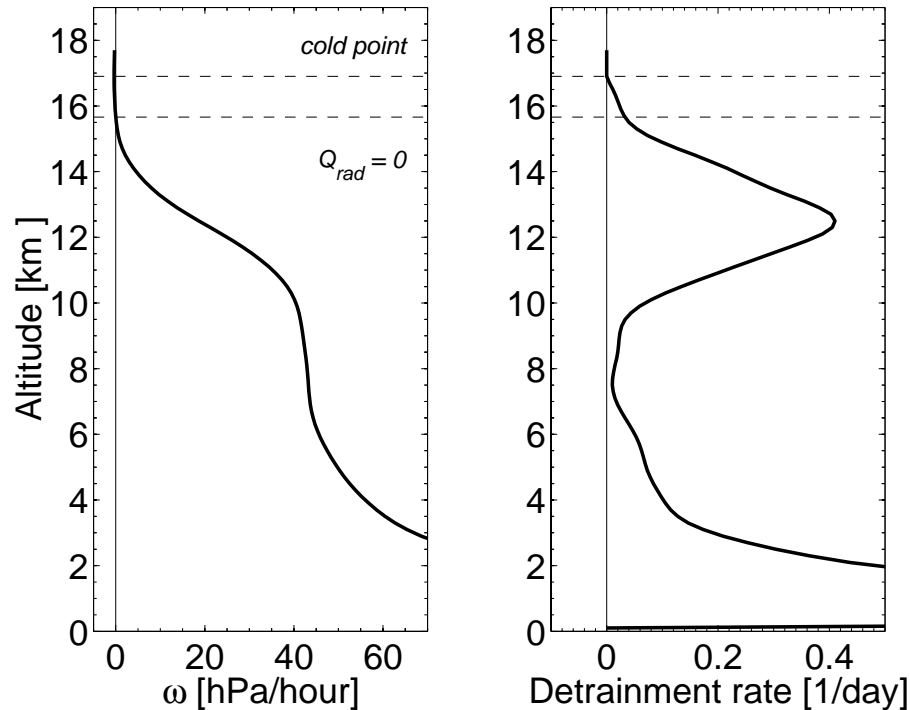
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**Fig. 1.** Schematic overview of transport in the tropical tropopause region. The level of zero radiative heating at an altitude of about 15.5 km (about 365 K potential temperature) marks the transition from large scale subsidence to large scale ascent, as diagnosed from diabatic heating rates ( $\omega_r$ ). Any material transported into the stratosphere through the tropical tropopause thus has to be transported upward by deep convection ( $\omega_c$ ) and detrain above the level of zero radiative heating (with detrainment rate  $d_c$ , schematically indicated by horizontal arrows). The line at 17 km (about 380 K potential temperature) indicates the approximate position of the cold point tropopause. See also the text in Sect. 2.1 for further discussion.

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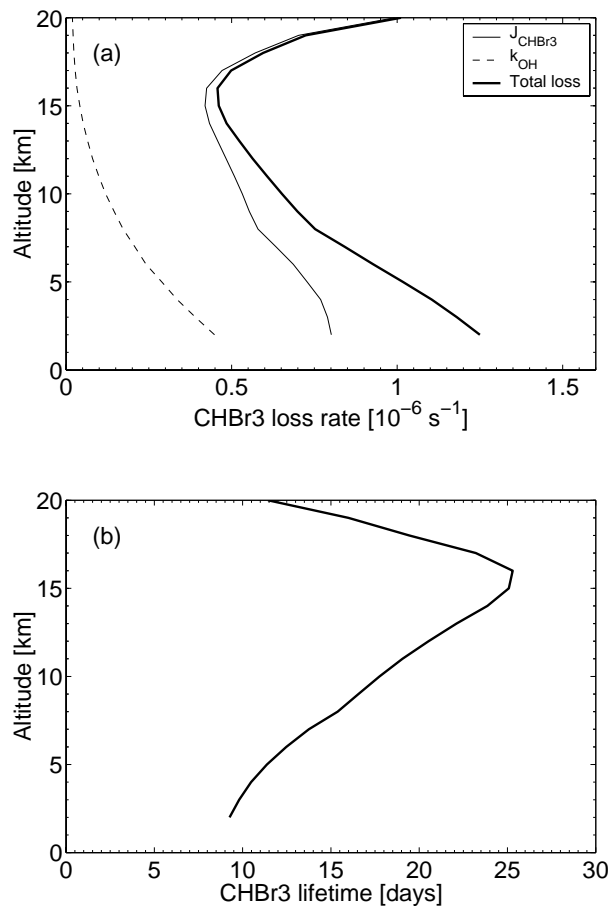
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**Fig. 2.** Clear sky downward mass flux ( $\omega = \omega_r + \omega_e$ , left panel) and detrainment rate ( $d_c$ , right panel) from Folkins and Martin (2005). The dashed lines indicate the level of zero radiative heating (which marks the transition from large scale subsidence to large scale ascent) and the cold point.

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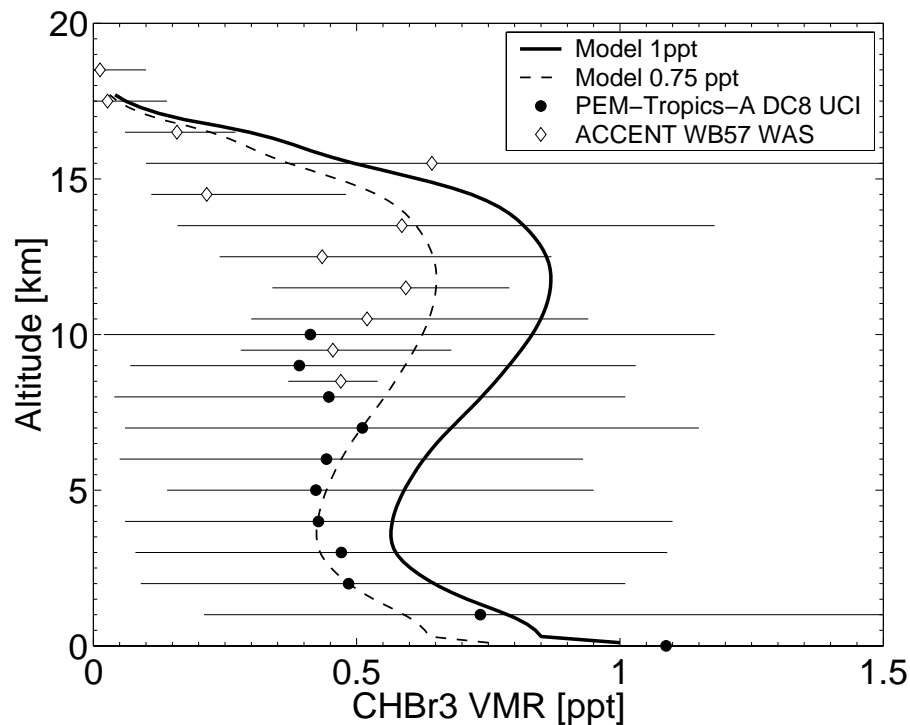
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**Fig. 3.** (a) Calculated bromoform loss rates from photolysis (thin solid line) and reaction with OH (dashed line) for a tropical mean profile, together with the total loss rate (thick solid line). (b) Bromoform lifetime  $\tau_{\text{chem}}$  in days (inverse of total loss rate).

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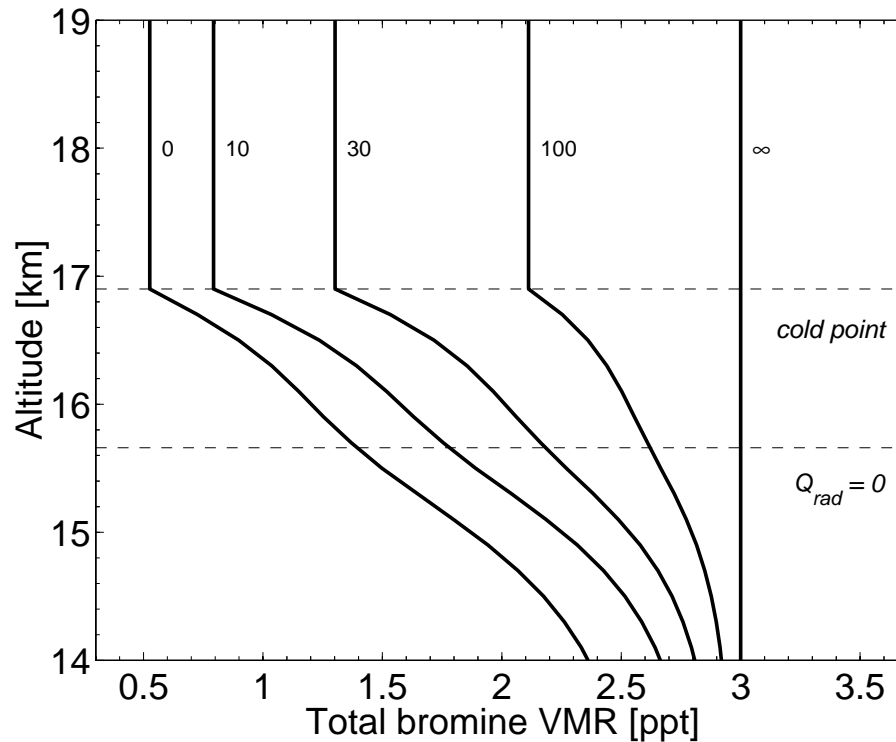


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**Fig. 4.** Modeled profile of bromoform assuming 1 pptv (solid line) and 0.75 pptv (dashed line) in the boundary layer, together with observations of bromoform from the PEM Tropics-A (August to October 1996, dots) and ACCENT (April and September 1999, diamonds) campaigns. For the campaign data mean values and the range between minimum and maximum values of the individual observations are shown.

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**Fig. 5.** Calculated total bromine released from bromoform (defined as  $\text{Br}_y + 3 \times \text{CHBr}_3$ ) for different loss rates of  $\text{Br}_y$  (numbers given in the graph with  $\text{Br}_y$  lifetime in days). The calculations assume 1 pptv of bromoform in the boundary layer and no detrainment of  $\text{Br}_y$  from convection (see text for discussion).

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