- 1 Referee #4, Gabor Vali
- 2 Received and published: 11 September 2015
- 3 4

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[A0] For clarity and visual distinction, the referee comments or questions are listed here in black and are preceded by bracketed, italicized numbers (e.g. [1]). Authors' responses are offset in blue below each referee statement with matching numbers (e.g. [A1]). Page and line numbers refer to online ACPD version.

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9 The problem addressed in this paper - the sizes of atmospheric ice nucleating particles - is

10 important and is far from having been adequately resolved in previous studies.

11 The paper does leave the impression of a somewhat brief, preliminary study for exploring

12 the potential of the relatively recent MOUDI-DFT technique. That is a legitimate goal

13 and it has been successful. But there are some shortcomings that deserve to be mentioned.

14 No ancillary data are presented, such as aerosol size distributions.

15 The aerosol were not brought to a standardized humidity so the sizing would be more

16 consistently comparable. Possible aging of the samples was not cotrolled for. Other

17 questions and suggestions are listed in the following sections. Some of the points are

18 perhps marginal, as this paper focuses on the sizes of the INPs not on their concentration;

19 nonetheless, the authors should discuss these problems and reflect on possible

implications for the results. Some caveats would be in order and some changes should beincluded when the paper is revised.

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We thank the referee for his helpful comments!

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## 25 Technical issues:

Since the MOUDI-DFT method employed in this paper has been described in at least two
previous publications, and is used here with no changes that would require special
attention, the method is taken as established and few details are given. Nonetheless, there

are some questions that can be esked regarding some aspects of the method.

- 30
- 31 *[1]* Background

32 Unless I missed it, it seems that the background level, or detection limit, has not been 33 established. There is mention of sudden freezing at -25C (pg. 20531 line 14) and this may 34 be an indication of the detection limit. But the background levels at other temperatures 35 are still open to question. Tests with the intake air passed through an efficient fiter 36 upstream of the sampler would have vielded blanks for establishing the background. This 37 may have made the condensation of drops difficult but that could have been circumvented 38 by some steps like higher supersaturation, producing drops meachanically, etc. None of 39 this is simple, of course.

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*[A1]* To address the referee's comments the average freezing temperature and
standard deviation of the blanks as well as the warmest freezing temperatures of the
blanks will be reported.

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- 45 [2] Sample volume
- 46 The volume of the air sample that is evaluated for INP content by Eq. 1 is taken as the

ratio of the DFT view area to the area of the glass slip while the total volume is for theentire MOUDI sample. Isn't the ratio of glass slip to MOUDI disk missing here?

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50 [A2] In Equation (1), the number of INPs found using the droplet freezing technique 51 is first multiplied by the fraction of the total sample area to the analysed area ( $A_{deposit}$ / 52  $A_{DFT}$ ) and is then divided by the total volume of air sampled.  $A_{deposit}$  is the area of the 53 MOUDI disk that the referee is referring to above. To clarify this point, in the revised 54 manuscript  $A_{deposit}$  will be referred to as the total area of the sample deposit on the 55 MOUDI impaction plate.

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57 *[3]* INP content of drops

58 The assumption of equal drop sizes from Vali (1971) has been maintained in Eq. 1 in 59 spite of the large range of droplet sizes (a factor 20 or more in volume) that form via 60 condensation in these experiments. This can't be simply ignored.

61 The issue raised in the previous paragraph is linked to the question what is the relevant

62 measure of INP content per drop in these experiments. The Vali (1971) proceedure is

63 formulated for the case when all drops are derived from the same bulk sample of water

64 with a random partitioning of INPs in the drops. In that case, all drops can be assumed to

have the same probability of containing any type of INP. With the condensation method

66 employed in these experiments the validity of this assumption needs to be examined

carefully, considering what determines where the drops form on the plates and what
 particles they contain. If condensation is initiated on the INPs themselves, the question

reduces to whether there is any dependence of the supersaturation required for

70 condensation and the INP activity of that particle. This in turn depends on what kind of

71 internal mixtures consitute the INPs. If the INPs become included into the drops as those

drops grow from a condensation nucleus that is not an INP, than the assumption of random probability of INP content per drop seems to be fulfilled. The main point is that

the authors should justify the use of the Vali (1971) formula for these experiments. It is

not obvious how the original assumptions apply here. Additional possible problems relate
to the adhesion of collected particles on the supporting surface. Perhaps the authors have
looked into whether the drops move particles along the surface as they expand during
condensation, or cover them up. The same question needs to be asked for the period of

- evaporation: are particles left behind as the drops shrink? There is a hint that perhaps this
- 80 is the case in Fig 2(c) of Mason et al. (2015a) with particles(?) seen in circular patterns.
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82 [A3] Droplet growth by water condensation in the DFT occurs in the same manner 83 for samples containing particles and clean hydrophobic glass cover slips (no particles 84 deposited and rinsed with ultrapure water). Therefore, water condenses uniformly on 85 the cover slip, and these droplets then combine as they grow. Particles become 86 incorporated into the droplets as this occurs. Analysis of the Ucluelet samples, with 87 similar observations over the entire dataset, shows that only a small fraction (on 88 average < 1%) of particles visible with the optical microscope remain outside of the 89 droplets once they reach their final size.

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We have also investigated binning the measurements based on droplet size and
residual inclusion area, using 2–4 bins, to see how this would influence our results.

- 93 The samples used in the binning analysis are those from the CSU intercomparison
- study noted in the Mason et al (2015a) instrument paper. We found in 98% of the
- samples the INP number concentration found using the binned data was within theexperimental uncertainty of the unbinned data. This information will be added to the
- 97 revised manuscript to address the referee's comments.
- 98
- 99 [4] Cooling rate

100 The reported cooling rate in the freezing experiments is -10C min<sup>-1</sup>. Thermal lags 101 throughout the system, specially within the drops but also within the stage, could be a 102 problem with such rapid cooling. Has that been considered? How uniform was the rate of

- 103 cooling throughout an experiment?
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[A4] The possibility of a thermal lag using a cooling rate of -10 °C min<sup>-1</sup> was 105 106 investigated in detail previously using a similar droplet freezing technique (Koop et al., 1998), and the thermal lag in that study was shown to be small at a cooling rate of 107 108 10 °C min<sup>-1</sup> (the correction was 0.2 °C at -33 °C). In addition we tested the possibility 109 of a thermal lag by investigating the freezing of pure water droplets using a cooling 110 rate of -5 and -10 °Cmin<sup>-1</sup> (unpublished results). The median freezing temperature 111 determined with the two cooling rates of -5 and -10 °C min<sup>-1</sup> were  $-36.3 \pm 1.5$  and -112  $36.7 \pm 2.0$  °C, respectively, with uncertainty given as the standard deviation.

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114 To determine the stability of the cooling rate we cooled the freezing cell from -5 °C 115 to -37 °C three times using the set cooling rate of -10 °Cmin<sup>-1</sup>. The average cooling 116 rate was found to be  $-9.8 \pm 0.4$  °C based on the three experiments. In Figure 1 of this 117 document (see below) we have plotted the cooling rate from three experiments over 118 the temperature range of -5 to -37 °C measured at a frequency of 1 Hz.

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Fig 1. The cooling rate of three experiments over the temperature range of -5 to -37
°C measured at a frequency of 1 Hz.

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125 **[5]** Correction factors

126 Two corrections factors are included in Eq. (1). The first one,  $f_{nu}$  is explained in some 127 detail in the referenced paper (Mason et al. 2015a). The second one,  $f_{ne}$  is less clear as the

referenced paper (Koop et al. 1997) doesn't show this quantity directly and here it is

applied to a formula (Vali 1971) that is differently derived. This requires some

- 130 explanation. Numerical values for the correction factors should be given.
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*[A5]* Values of these correction factors will be added to the Supplement. In addition,
Equation 1 will be separated into 2 equations; similar to our earlier paper that
describes the DFT-MOUDI technique (Mason et al. 2015a). The correction factors
should be clearer when 2 equations are included. In addition, the Koop correction
will be discussed in more detail in the revised manuscript.

- 136 137
- 138 Results:

139 *[6]* Were the overall average and median concentrations reported calculated with

140 weighting by the number of samples for each location? In light of the sample numbers

141 varying from 1 to 34, giving equal weight to each results in some distortion of the data.

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*[A6]* In our analysis the overall average and median concentrations were calculated
by giving each location (i.e. sampling site) equal weight. We felt that weighting by
the sample number would skew the average toward the sites of Amphitrite Point,
Saclay, and Alert while largely ignoring the remaining locations. This information
will be included in the revised manuscript for clarity.

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[7] On page 20534 long-range transport is stated to preferentially reduce the numbers of
 large particles. That is not an obvious result for INPs, specially since the mixture of
 components that make up the INPs is unknown.

- 152 153
- [A7] To address the referee's comments, the discussion on long-range transport will be deleted. Specifically, lines 18-27 on page 20534 will be deleted.
- 154 155

*[8]* There are many facets of the results obtained in this work and the authors have
selected meaningful diagrams to present those results. Yet, it would have been useful to
see the results also in the form of temperature versus concentration. That would also be a

158 see the results also in the form of temperature versus concentration 159 good way to compare present results with other published works.

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[A8] To address the referee's comments in the revised manuscript a spectrum of INP concentration vs. temperature for each location will be added to the Supplement.

164 **[9]** Results from the Colby site lie at the upper ends of the data both in concentration and 165 in size. It is stated that sampling was during and following combine operations. The range 166 of variability in concentrations and in sizes for these 3 samples is no different from other 167 sites. Is this relative uniformity related to having combine operations in the area, if not 168 directly near the site, throughout the sampling periods? On page 20258 it is indicated that

169 sampling was 3-10 m from the fields. Isn't that an error and kilometers were meant?

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- *[A9]* It is possible that additional combine activities in the region also contributed to
  the measured INP population. However, we don't have information that can confirm
  this possibility.
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The stated proximity of 3–10 m is correct. Equipment was set up in a mobile laboratory parked next to the fields.

- 178 [10] There is interesting aspect of the data that the authors might want to explore. The 179 question arises: are the INPs of different sizes of the same composition and have the same 180 surface properties? If so, the probability of a nucleating site being found on any size 181 particle would be the same. A rough estimate of the relative proportions indicated in Fig. 182 3 of the paper for two sizes indicates a negative answer to the question. Using a ratio of 183 about 1.3 for INPs greater than 1µm to 2.5µm and using those as nominal sizes whose 184 surface areas would differ by a factor of 6.25, the indication is that the larger particles 185 have a much lower probability per unit surface area to contain an ice nucleus. That would 186 be something to ponder. This calculation could be made more precise and extended to the 187 full range of sizes covered by the sampling. There are broad implications of this type of 188 analyses so presentation of the results in this form would make a valuable addition to the 189 paper.
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191 [A10] We assume here that the referee is suggesting that we calculate ice-nucleation 192 active surface site density  $(n_s)$  values as a function of size. We certainly agree these 193 calculations would be very interesting. To carry out these types of calculations we 194 would need the size distribution of the particle population at each site, measured at 195 the same times as the INP measurements. Unfortunately, this data was not measured 196 at all sites (e.g. UBC), and for the cases were this information was measured, 197 accessing this data and then carryout out these calculations would require a large 198 amount of work. We hope the Editor agrees that these calculations are not required 199 for the current publication. However, we will pursue these types of calculations in 200 future publications where we intend to report INP values from these sites in more 201 detail.

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203 *[11]* It is pointed out in the paper that the size range of INPs examined is restricted to > 204 0.1 $\mu$ m aerodynamic size and that further studies are needed to examine the contributions 205 of smaller size particles to INP populations. An additional aspect of INP sizing would be 206 worth mentioning. The size distributions of atmospheric particles undergo many changes 207 due to capturing, scavenging and other processes. Few particles are of single

- 208 composition. The measurements here reported refer to sizes of internally mixed particles 209 and also of aggregates. Thus, small INPs may be detected at larger sizes if they adhere to 210 other particles. This may take place already at the origin (soil especially) but also during 211 atmospheric transport. A difficult issue, no doubt, but one that need to be recognized.
- 212
- 213 *[A11]* This is a good point raised by the referee. The last paragraph of the Summary 214 and Conclusions section will be rewritten to address the referee's comment.
- and Conclusions section will be rewritten to address the referee's comment.
- 215 Specifically the last paragraph will be revised to the following:

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217	"One caveat to this study is that our measurements were confined to aerosol particle
218	sizes greater than either 0.10 or 0.18 µm. If there were a significant contribution
219	from INPs of smaller sizes (Vali, 1966; Schnell and Vali, 1973; Pummer et al., 2012;
220	Augustin et al., 2013; Fröhlich-Nowoisky et al., 2015; O'Sullivan et al., 2015; Tong
221	et al., 2015; Wilson et al., 2015), and these smaller sizes did not coagulate or get
222	scavenged by larger particles, the values presented here would represent upper limits
223	to the contribution of supermicron and coarse mode particles to the total INP
224	population. Additional studies exploring the relative atmospheric abundance of INPs
225	$< 0.10 \ \mu m$ are necessary (Hader et al., 2014). Future studies of the size distribution
226	of INPs should also include measurements of particle mixing state to determine if
227	particles are internally or externally mixed at the locations where the size distribution
228	of INPs are being measured."

229 230 Supp

Supplement:
[12] It is good to have the authors' analyses of the works cited for comparison of size

determinations. It helps to illustrate the sparsity of data on this important point.

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All but one of the works cited employed aerosol samplers. The exception is the Vali

235 (1966) report which is based on size discrimination by filtering of liquid samples. It

would be helpful to readers to point out this difference and its possible implications,

namely that soluble components of the INP get removed and particle might break up

when immersed in water. Both of these factors introduce a bias in the comparison of sizes, with corresponding aerosol sizes likely to have been considerably larger than the

sizes, with corresponding acrossing acrossing intervention in the opposite size found in water. Coagulation of particles in the liquid would add a bias in the opposite directions but is less likely to have been a factor.

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*[A12]* These points will be noted in section S1.1 of the Supplement where the Vali
(1966) measurements are discussed. Thank you for the suggestion.