

1 Anonymous Referee #3

2 Received and published: 1 September 2015

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4 *[A0]* For clarity and visual distinction, the referee comments or questions are listed
5 here in black and are preceded by bracketed, italicized numbers (e.g. *[1]*). Authors'
6 responses are offset in blue below each referee statement with matching numbers
7 (e.g. *[A1]*). Page and line numbers refer to online ACPD version.

8
9 The manuscript reports immersion-mode INP number concentrations as a function of
10 aerodynamic size at six ground sites in North America and one in Europe. Size- resolved
11 particle samples were collected using a model 110R or 120R Moudi. The ice-nucleating
12 ability of particles was then determined by a microscope-based immersion freezing
13 apparatus (MOUDI-DFT technique). The authors found that both supermicron and coarse
14 mode aerosol particles were a significant component of the INP population. The paper is
15 well written and of interest. I suggest publication after the following few comments have
16 been addressed.

17
18 We thank the referee for his/her helpful comments!

19
20 General remarks:

21 *[1]* It is known that not all particles colliding with a plate adhere to it. As offline ice
22 nucleation analysis cannot coat (e.g. with vacuum grease) the substrate located on the
23 impaction plate, particle rebound in the Moudi impactor should be discussed. Generally
24 speaking, the rebound increases with aerosol diameter and decreasing air relative
25 humidity. The average relative humidity values (r.h.) in the seven locations range from
26 48% to 97%. Therefore, the rebound should differ in the sampling sites. At Amphitrite
27 Point, where r.h. during sampling was 97%, the rebound should be much lower with
28 respect to Colby, where the r.h. was 48%. In marine air the bouncing should be lower due
29 to hygroscopic particles.

30
31 *[A1]* Thank you for bringing up this point. In the revised manuscript the issue of
32 particle bounce will be discussed.

33
34 *[2]* The interest of the paper would be enhanced if particle number concentration and size
35 distribution were considered in each site during sampling. This information could allow
36 calculation of the ratio between INP number concentration and the corresponding particle
37 concentration in each size bin, and the correlation between INP number concentration and
38 aerosol.

39
40 *[A2]* We agree that this information would be very useful to report. To carry out
41 these types of calculations we would need the size distribution of the particle
42 population at each site, measured at the same times as the INP measurements.
43 Unfortunately, this data was not measured at all sites (e.g. UBC), and for the cases
44 where this information was measured, accessing this data and then carryout out these
45 calculations would require a large amount of work. We hope the Editor agrees that
46 these calculations are not required for the current publication. However, we will

47 pursue these types of calculations in future publications where we intend to report
48 INP values from these sites in more detail.

49
50 **[3]** Hygroscopic particles sampled in marine sites form droplets on the examined area
51 sooner than insoluble particles (which are considered efficient ice nuclei). Could this
52 feature influence the INP concentrations measured with the MOUDI-DFT technique?

53
54 **[A3]** Differences in particle hygroscopicity between sampling locations should not
55 influence the droplet freezing technique (DFT), and therefore the measured INP
56 concentrations. The relative humidity of the gas flow in the DFT during droplet
57 growth was held at approximately 120 %. At this relative humidity water will
58 readily condense on both ambient samples containing particles and directly on the
59 hydrophobic glass cover slip (as observed in the blank experiments).

60
61 To address the referee's comment we will add the following to the revised
62 manuscript:

63
64 "Droplet growth by water condensation in the DFT occurs in the same manner for
65 samples containing particles and clean hydrophobic glass cover slips (no particles
66 deposited and rinsed with ultrapure water). Therefore, water condenses uniformly on
67 the cover slip, and droplets combine as they grow to a final size of $97 \pm 42 \mu\text{m}$ (mean
68 diameter and 1 standard deviation). On average, more than 99 % of particles become
69 incorporated into the droplets as this occurs."

70
71 **[4]** Paragraph 2.2: Size-resolved INP number concentrations

72 This paragraph should be broadened by summarizing the most important points of the
73 technique used, reported in the paper of Huffman et al. (2013) and Mason et al. (2015).
74 For instance, the total area of each stage and of the analyzed area should be indicated, and
75 the problem of the non-uniformity of aerosol deposit in each stage of the MOUDI should
76 be addressed. An additional point should be clarified. Huffman et al. (2013) found that
77 the maximum concentration of IN detected for any given slide with the microscope
78 freezing technique is roughly 0.6 -0.9 L⁻¹ (depending on the number of droplets
79 condensed and the total volume of air sampled) and the maximum concentration of IN
80 determined by the microscope technique is small compared to the maximum
81 concentration determined with the CFDC method. The submitted paper reports
82 concentrations up to 10 INPs L⁻¹ (T = -25°C, size interval 5.6 - 10 μm , at Colby, KS).
83 Which is the maximum number of droplets that can be formed on the area (1.2 mm²)
84 analyzed by the DFT?

85
86 **[A4]** This paragraph will be expanded to address the referee's comments, as well as
87 the other referee's comments. For example, Values of A_{deposit} , f_{nu} , and f_{ne} used in
88 Equation (1) will be given in the Supplement to provide more information on
89 calculations involving the MOUDI-DFT.

90
91 Regarding the difference in the upper limit of the INP concentrations reported here
92 and in Huffman et al. (2013): in the current study we have used shorter sampling

93 times than those of Huffman et al. (2013), allowing us to detect greater INP number
94 concentrations. This information will be added to the revised manuscript. The
95 largest number of droplets condensed in an experiment for this study was 75.
96

97 Minor remarks:

98 **[5]** Page 20531 - Line 12 and following: “Freezing events were rare at temperatures
99 warmer than -15°C and are therefore not reported”. This statement appears contradict
100 what it is said afterwards, i.e.: - Page 20532, Line 9 and following: “. . .the major source
101 of INPs at Amphitrite Point during the study period was likely biological particles from
102 local vegetation . . .” - Page 20532, Line 20 and following: “. . .the highest concentrations
103 of INPs at a freezing temperature of -25°C were found at the Colby, KS sites. . . aerosol
104 sampling was conducted adjacent to soya and sorghum fields . . .This high concentration
105 of INPs is consistent with previous work of Garcia et al. (2012) . . . and Bowers et al.
106 (2011). . .”
107

108 For instance, Garcia et al. (2012) measured an INP concentration of about 1 L⁻¹ at T = -
109 10°C. Bowers et al. (2011) found greater INPs downwind of corn fields than in air
110 samples collected from the suburban and forest land-use types, at T > -10°C. Generally
111 speaking, biological aerosols (bacteria, spores, fungi, pollen) are activated as ice nuclei
112 prevalently at temperatures warmer than -10°C (Möhler et al., 2007). Therefore, at sites
113 like Amphitrite Point, Colby and Saclay, a fraction of aerosol particles should be
114 activated even at T > -15°C.
115

116 **[A5]** A small fraction of the droplets did freeze at temperatures above 15 °C (see
117 response **[A7]** for referee 1). This will be noted in the revised manuscript.
118

119 **[6]** At Labrador Sea only one sample was available. What does the uncertainty reported
120 in Fig. 2 and Fig. 4 mean?
121

122 **[A6]** As only one sample is available for the Labrador Sea location, the uncertainties
123 reported in Fig. 2 is the uncertainty in the INP concentration from Equation (1). We
124 now realize this has led to confusion. To reduce confusion, we will remove the error
125 bars for the Labrador Sea data in Fig 2 and indicate in the figure caption, that no
126 error bars for the Labrador Sea data are reported since only one sample was available
127 for this location. All other error bars in the figure represent the standard error of the
128 mean.
129

130 In Fig. 4, the uncertainty at each location is given as the 25th and 75th percentile INP
131 size as indicated in the figure caption.
132

133 **[7]** We note that the lower concentration of INP was obtained at Alert, where higher air
134 volume was sampled (about 32 m³) and the highest concentration at Colby, where the
135 sampled air volume was the lowest (about 8 m³). Is this a fortuitous event?
136

137 **[A7]** We do not have any reason to expect the INP concentration is related to the
138 sampling volume. Nevertheless we carried out an additional analysis of the Alert

139 data to further investigate this question. Sampling times at Alert varied from
140 approximately 2.3 to 46.1 hours (with an average of 17.6 h as reported in Table 1).
141 The linear correlation coefficients (R) of INP concentration vs. sample volume were
142 0.45, 0.04, and -0.39 at freezing temperatures of -15, -20, and -25 °C, respectively.
143 Furthermore, at temperatures of -15 and -20 °C the sample with the largest INP
144 concentration had the second largest sample volume. Hence, the Alert data suggests
145 the INP concentration is not related to the sampling volume.

146

147 **[8]** As all graphs report in order: Alert NU, Whistler Mountain, Amphitrite Point,
148 Labrador Sea, Saclay France, UBC Campus, Colby, KS, please follow the same order in
149 the Paragraph: 2.1: Samplings Sites and in: Table 1

150

151 **[A8]** The change in order will be made in the revised manuscript.