# Theoretical study of mixing in liquid clouds. Part 1: classical concept

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# Alexei Korolev<sup>1</sup>, Alex Khain<sup>2</sup>, Mark Pinsky<sup>2</sup>, and Jeffrey French<sup>3</sup>

- [1] Environment Canada, Cloud Physics and Severe Weather Section, Toronto, Canada
- [2] Department of Atmospheric Sciences, the Hebrew University of Jerusalem, Israel
- 9 [3] University of Wyoming, Laramie, WY, USA
- 10 Correspondence to: A. Korolev (alexei.korolev@ec.gc.ca) Korolev (alexei.korolev@canada.ca)

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#### Abstract

Relationships between basic microphysical parameters are studied within. The present study considers final stages of in-cloud mixing in the framework of classical concept of homogeneous and extreme inhomogeneous mixing. Analytical expressions Simple analytical relationships between basic microphysical parameters were obtained for homogeneous and numerical simulation <del>of</del>extreme inhomogeneous mixing based on the adiabatic consideration. It was demonstrated that during homogeneous mixing the functional relationships between droplet concentration, extinction <del>coefficient, liquid water content, and mean volume droplet</del>the moments of the droplets size<del>, formed a</del> the final distribution hold only during primary stage of mixing are presented. The expressions are Subsequent random mixing between already mixed parcels and undiluted cloud parcels breaks these relationships. However, during extreme inhomogeneous mixing the functional relationships between the microphysical parameters hold both for primary and subsequent mixing. The obtained relationships can be used to identify the type of mixing for from in-situ observations obtained. Th <u>effectiveness of the developed method was demonstrated using in-situ data collected in convective</u> clouds. The analysis suggests It was found that for the specific set of observations investigated here, in situ measurements, the interaction between cloudy and entrained environments is was, dominated by inhomogeneous mixing. Lastly, an analysis of different response times of the cloud environmen

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undergoing mixing is presented. Comparisons of different characteristic times suggest that within the same mixing environment depending on mixing fraction some volumes may be dominated by homogeneous mixing whereas others by inhomogeneous mixing, extreme inhomogeneous mixing.

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## 1. 1 Introduction

Turbulent mixing is an important non-adiabatic process in the atmosphere that to a large extent determines spatial gradients of many thermodynamic (e.g. temperature, humidity) and cloud microphysical parameters (e.g. hydrometeor concentrations, extinction coefficient, condensed water content) and as such, needs to be properly described in numerical simulations of clouds and weather predictions. Entrainment and mixing occurs during the entire lifetime of a cloud and is active not only near cloud edges, but it is important throughout the whole cloud volume. Mixing of cloudy and entrained air results in changes to the shape of the droplet size distribution through partial droplet evaporation and can also lead to changes in droplet concentration through complete evaporation of some fraction of droplets, and dilution. The shape of the droplet size distribution plays key role in the initiation of precipitation and radiative properties of clouds.

The treatment of mixing in numerical simulations of clouds and precipitation formation remains a challenging problem. Besides the issues related to the way to describe mixing in numerical schemes, there is a fundamental problem of identifying a scenario or path, that mixing events should follow. Since Through the pioneering works of Latham and Reed (1977) and Baker et al. (1980) two explicitly alternative scenarios of mixing were identified. In the first scenario the turbulent mixing rapidly stirs the environment unifyinghomogenizing the fields of temperature and humidity. After Following that, all of the droplets undergo partial evaporation under the same conditions. The result of this mixing is the droplet population with reduced sizes, but their total amount number that remains unchanged. This type of mixing is referred to as homogeneous. In the second scenario, mixing occurs more slowly such that the population of droplets experiences different amount of sub-saturation. Some number of droplets completely evaporates, while others experience no evaporation until the entirety of the entrained air becomes saturated. After Following

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that turbulence mixes the rest of the droplets with the saturated, but droplet-free environment. During this type of mixing the size of droplets remains unchanged; however, their concentration total number is reduced. This type of mixing is called extreme inhomogeneous. The intermediate case when some fraction of droplets evaporates partially, the another other fraction evaporates completely, and thea third fraction remains without changes unchanged is in some works referred to as inhomogeneous (e.g. Baker and Latham, 1980).

The conditions for homogeneous and extreme inhomogeneous mixing and their effects on precipitation formation have been debated in cloud physics over forty years. There are a number of numerical simulations and theoretical efforts on studying different aspects of mixing and its effect on cloud microphysics (e.g. Baker and Latham, 1982; Jensen and Baker, 1989; Su et al., 1989; Lasher-Trapp et al., 2005; Jeffrey, 2007; Andrejczuk et al., 2009; Kumar et al., 2013; Jarecka et al., 2013, and many others). A comprehensive review of the works on the effect of turbulence and mixing on cloud dropletsdroplet formation can be found in Devenish et al. (2012).

A number of studies were dedicated to identifying type of mixing based on in-situ observations. Most of the previous in-situ observations provide evidence supporting inhomogeneous mixing (e.g. Hill and Choularton, 1985; Paluch, 1986; Bower and Choularton, 1988; Blyth and Latham, 1991; Gerber et al., 2008, Lu et al. 2011; Beals et al. 2016). However, works of Jensen and Baker (1989), Paluch and Baumgardner (1989), Burnet and Brenguier (2007), Lehmann et al. (2009) suggest), Lu et al. (2011) suggested occurrence of homogeneous mixing. So, at the moment it appears that both types of mixing may occur in liquid clouds. However, the environmental conditions resulting from governing one or the other type of mixing remain not well understood.

Early experimental work on identifying type of mixing from in-situ observations were based on the analysis of spatial variability of the shapes of individual droplet size distributions (e.g. Paluch and Knight, 1984; Paluch, 1986; Bower and Choularton, 1988). However, the The effectiveness of this method involving the analysis of overlya large number of individual size spectra turned out to be quite low. Another technique utilized expected functional relationships between droplet concentration (\*N\*)(N) and droplet radius (\*r\*), which is diameter (D) specific to each type of mixing. Thus, during extreme inhomogeneous mixing the droplet size is expected to remain unchanged, whereas the concentration will vary. However, during During homogeneous mixing the droplet size and concentration in cloud will be related to each other in a certain way, depending on

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the mixing fraction and the humidity of the entrained air. This fact was used in observational studies enfor identifying the type of mixing from "mixing diagrams" that related  $\frac{N}{N}$  and  $\frac{1}{N}$  for different regimes of mixing (e.g. Burnet and Brenguier, 2007; Gerber et al., 2008; Lehmann et al., 2009).

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The use of the mixing diagrams to some extent facilitated identification of type of mixing. However, in many cases scatter in the relationships between  $\frac{N}{N}$  versus  $\frac{1}{N}$   $\frac{N}{N}$  was too large, which hindered identifying hindering identification of the type of mixing (Burnet and Brenguier, 2007). To resolve this problem many researchers used other complementary measurements supporting identification of the type of mixing (e.g. Gerber et al., 2008; Lehmann et al., 2009).

Besides the effect on  $N_{\bullet}$  and  $r_{\bullet}D_{\nu_{\bullet}}$  the type of mixing is anticipated to manifest itself in relationships between other moments of the droplet size distribution,  $f(r) \cdot f(D)$ . Such relationships may provide insight into the mixing process and identify type of mixing. With the exception of the work by Hill and Choularton (1985), who correlated concentration and liquid water content, there have been few attempts to use any other microphysical parameters for identification of type of mixing.

In order to fill this gap, this study presents a theoretical analysis of relationships between different moments of f(r)f(D) within the framework of homogeneous and extreme inhomogeneous mixing. The analysis is focused on the first four moments of f(r)f(D) corresponding to the droplet concentration f(D) moment, integral radius f(D) diameter f(D) moment), extinction coefficient f(D) (f(D) moment) and mean volume radius f(D) (f(D) moment) and f(D) moment). It is shown that the set of newly obtained relationships obtained can be effectively used to identify between the moments provide a more robust identification of type of mixing from in-situ observations, measurements as compared to conventional f(D) relationships used in mixing diagrams. Relationships between moments may be useful for parameterization of mixing in numerical simulations of clouds and climate, interpretations of remote sensing measurements.

This paper constitutes the first in a series of three papers. The present paper It considers the final stage of mixing based on the formal definitions of homogeneous and extreme inhomogeneous mixing. These two types of mixing present two extreme regimes of mixing. The following two papers provide a detailed analysis of the time dependent processes during

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homogeneous (Pinsky et al. 2015a., 2016a) and inhomogeneous (Pinsky et al. 2015b., 2016b) mixing- where non-extreme regimes are considered as well.

The present This paper is arranged in the following way. Section 2 presents analysis of the analytical relationship between N,  $N\overline{r}$ ,  $\beta$ , q,  $\overline{r}_3N$ ,  $N\overline{D}$ ,  $\beta$ , q,  $D_v$  and mixing fraction  $\mu\mu$  for the cases of homogeneous and extreme inhomogeneous mixing. The In Sect. 3 the obtained analytical relationships are compared with the results of numerical simulation of N,  $\beta$ , q,  $\overline{r}_3N$ ,  $\beta$ , q,  $D_v$  formed after at the final stage of mixing are discussed in section 3. Section 4 presents results of simulation of progressive mixing and it effect of the relationships between moments. Examples of relationship between N,  $\beta$ , q,  $\overline{r}_3N$ ,  $\beta$ , q and  $D_v$  from in-situ observations are presented in section 4. Section Sect. 5 presents analysis of characteristic response times of a cloud environment during homogeneous and inhomogeneous mixing. The The discussion and concluding remarks are presented in section Sect. 6— and 7.

#### **2. 2** Effect of mixing on microphysical variables

#### 2.1. Phenomenological consideration

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150 151 The conceptual diagrams of homogeneous and extreme inhomogeneous mixing are shown on Fig. 1. During the first stage of extreme inhomogeneous mixing the <u>sub-saturated-subsaturated</u> parcel is engulfed into the cloudy environment (Fig. 1-a1\_la1). Then, the droplets at the interface of the sub-saturated parcel and the cloud environment undergo complete evaporation until the air within the engulfed volume reaches saturation (Fig. 1-a2\_la2). After that the saturated but droplet free parcel mixes with the rest of the cloud environment (Fig. 1-a3\_la3). The result of the inhomogeneous mixing is that the cloud parcel has reduced droplet concentration and the droplet sizes remain unchanged.

In the case of homogeneous mixing after entraining into a cloud (Fig.1-b1\_1b1), the subsaturated subsaturated parcel "instantly" mixes up with its cloud environment (Fig.1-b2), which leads 1b2) leading to undersaturation of the total volume. Then, all droplets throughout the mixed volume undergo simultaneous evaporation until the equilibrium state is reached. The result of homogeneous mixing is a cloud parcelvolume with reduced concentration of droplets and droplets with reduced sizes (Fig.1-b3).—1b3).

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The difference between these two-Based on mass and energy conservation considerations the final state of the bulk parameters (i.e. liquid water mixing fraction, humidity, temperature, etc.) is the same for both types of mixing is as follows: However, in the case of extreme inhomogeneous mixing the saturation is reached through complete evaporation of some fraction of droplets, whereasand their sizes remain constant. In Whereas in case of homogeneous mixing the saturation is reached through a uniform evaporation of droplets, whereasand the total number of droplets in the diluted parcel remains unchanged. It should be noted, that in both cases of mixing the droplet concentration decreases due to dilution by the entrained mixed droplet free sub-saturated parcel.

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The following discussion will be specifically focused aton the microphysical properties formed at the final stage of the homogeneous and extreme inhomogeneous mixing. The processes occurring during mixing state (i.e. transition  $1a\rightarrow 2a$  and  $1b\rightarrow 2b$  in Fig. 1) remain outside the frame of this work. Following the formalism of homogeneous and extreme inhomogeneous mixing, the process of mixing reaches the final stage when (1) the entrained and cloud environment are mixed up and the spatial gradients of the microphysical  $(N, \beta, q, N, \beta, q, etc.)$  and environmental (T, S, e, etc.) parameters approach to zero; (2) the diffusional process related to droplet evaporation comes into equilibrium. The second condition is completed when (a) the environment reaches saturation state, or (b) the entire population of droplets is completely evaporated, if the mixedentrained air is toosufficiently dry.

The above description of homogeneous and extreme inhomogeneous mixing is highly idealized. Actual in-cloud mixing does not occur as a sequence of discrete events (Fig.1) that individually come to equilibrium only to be followed by next discrete mixing events. But rather it is occurring continuously on a cascade of different spatial and time scales. Broadwell and Breidenthal (1982) summarized the experimental evidence and proposed the following description of mixing in turbulent shear layers. Mixing takes place in a series of events. Two shear layers exchange mass by engulfing parcels from an opposite layer into localized zones. The initially large-scale filaments of the two gases break down towards smaller scales due to the action of turbulence. The turbulence stretches the interface between the gases and enhances the molecular diffusion across the increasing surface. The actual mixing of the engulfed volume is a molecular diffusion process that is most effective after the break down volumes reduce to the Kolmogorov viscosity scale. It is anticipated that the reaction of the ensemble of droplets from entrainment is a combination of homogeneous and inhomogeneous mixing with domination of one type of mixing

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over the other depending on the characteristic spatial and time scales of the environment determined by turbulence, cloud microphysics, state parameters and stage of mixing.

## 2.2 Methodology

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volumes.

The foregoing discussion will be focused on mixing between saturated cloud parcels and outof-cloud sub-saturated air. The cloud parcel contains droplets with monodisperse radius-rade radius rade age <u>diameter  $\bar{D}_{1}$ , liquid mixing ratio  $q_0q_1$  and number concentration  $N_0$ .  $N_1$ . The initial temperature</u> in the cloud parcel is  $T_1, T_{12}$  relative humidity  $S_1 = 1, RH_1 = 1$ , where  $S = e/E_s(T)$  (The RH  $e/e_s(T)$  (the explanation of variables variable notations is provided in Appendix ATable 1). The second parcel is droplet free  $(N_2 = 0) \cdot (N_2 = 0)$  sub-saturated with initial relative humidity  $S_2 < 0$  $RH_2 < 1$  and temperature  $T_2$ - $T_2$ . The mixing occurs isobarically, i.e. during mixing p-const.p const. At the endfinal stage of the mixing the temperature and humidity formed in the resulting parcel are  $\frac{T_mT}{T_m}$  and  $\frac{S_m}{S_m}$  (Appendix B). RH (appendix A). The process of mixing is completed when the mixed parcel reaches equilibrium due to the air saturation (i.e.  $S_m = 1$ ). RH = 1, or due to the complete evaporation of droplets. In the latter case the final humidity is  $\frac{S_m < 1-RH}{S_m < 1-RH} \le 1$ . The effect of the vertical velocity and vertical travel of the mixing parcels on temperature  $T_m$ , humidity  $S_m$ , of <u>final T, RH</u>, and condensed water  $q_m q$  is not considered here, i.e. vertical velocity  $u_z = 0$ ,  $u_z = 0$ Without the loss of generality the masses of the cloudy and sub-saturated volumes prior to the mixing are assumed to have a unit masses, i.e.  $m_1 = 1$  and  $m_2 = 1$ ,  $m_2 = 1$ . The mixing process will be considered as mixing of the  $\mu\mu$  fraction of the cloud volume parcel with the  $(1-\mu)$  $(1 - \mu)$  fraction of the second (sub-saturated) volume parcel. The mixing cloud fraction changing may vary within the range of  $0 \le \mu \le 1$ , Therefore, the mass of the resulting mixed parcel is equal to  $m_1\mu + (1-\mu)m_2 = 1$ ,  $m_1\mu + (1-\mu)m_2 = 1$ . This approach simplifies

# 2.3 Effect of mixing on liquid water and temperature

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the consideration of mixing and allows considering all possible proportions of the mixing of two

The mixing ratio of liquid water q formed at the final stage of mixing is determined by the

211 mass of the mixing cloud water  $\mu q_1$  and amount of evaporated water required to saturate the newly

212 <u>formed mixed volume</u>  $\delta q_m$ . The mass balance of liquid water for the mixing volume yields

$$q = \mu q_1 - \delta q_m \tag{1}$$

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$$\delta q_{m} = \frac{c_{p}R_{v}T_{m0}^{2}}{L^{2}} \ln \left( \frac{1 + \frac{e_{s}(T_{m0})R_{a}L^{2}}{pc_{pa}R_{v}^{2}T_{m0}^{2}}}{1 + RH_{m0}\frac{e_{s}(T_{m0})R_{a}L^{2}}{pc_{pa}R_{v}^{2}T_{m0}^{2}}} \right) \cong -\frac{S_{m0}}{A_{2}}$$

$$(2)$$

is the mixing ratio of liquid water required to saturate 1kg of volume with temperature  $T_{m0}$  and

<u>humidity</u>  $RH_{m0}$  (appendix A);  $T_{m0}$ ,  $RH_{m0}$  and  $S_{m0}$ , are the temperature, relative humidity formed

and supersaturation formed in the volume after instantaneous air mixing, but before droplets start

evaporating (appendix A);  $e_S(T_{m0})$  is saturation vapor pressure at temperature  $T_{m0}$ .

Eq. (1) is a non-linear function of  $\mu$ , since  $T_{m0}$ ,  $e_{m0}$  and thus  $\delta q_m$  depend on  $\mu$ . Eq.(1) can be

simplified, if  $T_1 = T_2$ . In this case  $T_{m0} = T_1 = T_2$ , and  $e_S(T_{m0}) = e_S(T_1) = e_S(T_2)$ . Given that,

222 the expression under logarithm in Eq.(2) can be expanded in series resulting in (appendix B)

$$\delta q_m = (1 - \mu) \delta q^*, \tag{3}$$

224 where

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$$\delta q^* = \frac{c_p R_v T_2^2}{L^2} \ln \left( \frac{1 + \frac{e_S(T_2) R_a L^2}{p c_{pa} R_v^2 T_2^2}}{1 + R H_2 \frac{e_S(T_2) R_a L^2}{p c_{pa} R_v^2 T_2^2}} \right) \cong -\frac{S_2}{A_2}$$
(4)

226 is the mixing ratio of liquid water required to saturate 1 kg of the entrained dry air. Substituting

227 Eq.(3) in Eq.(1) gives

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$$q = \mu q_1 - (1 - \mu) \delta q^*,$$
 (5)

The value of  $\delta q^*$  does not depend on  $\mu$ , and Eq. (5) is a simple linear function of  $\mu$ . The

230 comparisons with numerical simulations showed, that Eq.(5) provides accuracy within few

percent, when the temperature difference  $|T_1 - T_2| < 2$ °C. Although, in many cases  $|T_1 - T_2|$  may

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vary a wide range reaching 10°C or higher, clouds with  $|T_1 - T_2| < 2$ °C are quite common.

Therefore, for the sake of simplicity, Eq.(5) and the assumption  $T_1 \approx T_2$  will be used in the

234 following consideration of mixing.

It should be noted that, Eqs (1) and (5) are valid for the cases, when  $\mu > \mu_{cr}$ . Here  $\mu_{cr}$  is

critical mixing fraction, which separates partial and complete evaporation of cloud water in the

mixing volume (section 2.4). Cases when  $\mu \le \mu_{cr}$  correspond to complete evaporation of droplets,

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The temperature at the final stage of mixing can be estimated as (appendix C)

$$T = T_{m0} - \frac{(1-\mu)\delta q^* L}{c_{pa}}, \quad \text{when } \mu > \mu_{cr}$$
 (6a)

$$T = T_{m0} - \frac{\mu q_1 L}{c_{pa}} \qquad \text{when } \mu \le \mu_{cr}$$
 (6b)

Eqs. (1), (5), (6) were obtained based on mass and energy conservation, and they do not

depend on how mixing proceeds. Therefore, Eqs. (1), (5), (6) are valid for both homogeneous and

244 inhomogeneous mixing.

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2.4 Complete evaporation

As mentioned in section 2.2 the process of mixing is complete only after reaching equilibrium

by saturating the mixed volume or by evaporating of all cloud droplets depending on the mixing

fraction  $\mu_c$ . The critical mixing fraction  $\mu_{cr}$ , corresponding to evaporation of all droplets, can be

250 found from Eq.(5) when q = 0, i.e.

$$\mu_{cr} = \frac{\delta q^*}{q_1 + \delta q^*} \tag{7}$$

252 <u>Critical mixing fraction separates  $\mu$  in two subranges: (a)  $1 \ge \mu > \mu_{cr}$  where q is described</u>

253 <u>by Eqs.(1) or (5) and  $RH_m = 1$ ; (b)  $\mu_{cr} \ge \mu \ge 0$  where q = 0 and  $RH_m \le 1$ .</u>

For the general case when  $T_1 \neq T_2$ ,  $\mu_{cr}$ , can be found by solving the non-linear equation

$$\mu_{cr}q_1 - \delta q_m(\mu_{cr}) = 0 \tag{8}$$

Figure 2 shows comparisons of dependences of  $\mu_{cr}$  vs.  $q_1$  calculated from Eq. (7) and those

deduced from a numerical model (Sect. 3). Critical mixing fraction  $\mu_{cr}$  is also shown by black

258 stars in Fig. 4. The locations of the stars in Fig.4 coincide well with the locations, where the

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modeled microphysical moments become zero. The obtained agreement between analytical and 259 modeled  $\mu_{cr}$  in Figs. 2 and 4 validates the developed approach. 260 261 2.5 Extreme inhomogeneous mixing 262 263 Within the framework of the extreme inhomogeneous mixing some fraction of droplets 264 undergo complete evaporation, whereas the rest of the droplets remain unchanged. Therefore, such <u>a</u> process results in scaling the droplet size distribution  $f(r)_{\tau} f(D)_{\underline{1}}$  i.e. 265  $f(r) = kf_0(r)$ (1) 266  $f(D) = kf_1(D)$ (9) 267 where  $\frac{k}{k}$  is some coefficient dependent on  $\frac{\mu\mu}{k}$  and the initial environmental parameters of the 268 mixing volumes,  $\frac{f_0(r)f_1(D)}{f_1(D)}$  is the droplet size distribution before mixing. 269 270 Eq.1 Equation (9), yields relationship relationships between pairs n-thnth and mk-th moments  $\frac{M_n}{M_{0n}} = \frac{M_m}{M_{0m}}$ 271  $\frac{M_n}{M_{n1}} = \frac{M_k}{M_{k1}}$ 272 (10)where  $M_n = \int_0^\infty f(r)r^n dr M_n = \int_0^\infty f(D)D^n dD / \int_0^\infty f(D)dD$  is the *n*-th<u>nth</u> moment of f(r) - f(D)273 274 Therefore, it is anticipated that for extreme inhomogeneous mixing droplet number concentration 275  $N \in N$  (0th moment), extinction coefficient  $\beta = \beta$  (2nd moment), liquid water content q-mixing ratio **a** (3rd moment), along with other moments, will correlate with each other, i.e. 276  $\frac{N}{N_0} - \frac{\beta}{\beta_0} - \frac{q}{q_0}$ 277  $\frac{N}{N_1} = \frac{\beta}{\beta_1} = \frac{q}{q_1}$ 278 (11)One of the consequences of Eqs. 1-3 (9)-(11) is conservation of that the characteristic droplet 279 sizes, i.e.  $\bar{r}$ ,  $\bar{r}_2$ ,  $\bar{r}_3$ ,  $r_{eff}$   $\bar{D}$ ,  $D_{2}$ ,  $D_{v}$ ,  $D_{eff}$  will remain constant during inhomogeneous mixing. 280 The liquid water mixing ratio q-resulting from extreme inhomogeneous mixing can be written base 281 282 on For the mass balance of vapor case  $T_1 = T_{24}$  and liquid water as

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$$283 \qquad \frac{q = \mu q_0 - \delta q}{q}, \tag{4}$$

where  $\delta q$  is  $\mu > \mu_{cr}$  Eqs. (5) and (11) yield the amount of liquid water per 1 kg of the dry air required

285 to saturate the volume formed as a result of mixing of the cloud and entrained parcels before dropldt

286 evaporation. The value of  $\delta q$  can be found as (Appendix B

$$S_{q} = \frac{L^{2}}{c_{p}R_{v}T_{2}^{2}} \ln \left( \frac{1 + \frac{E_{s}(T_{m0})R_{a}L^{2}}{pc_{p}R_{v}^{2}T_{m0}^{2}}}{1 + S_{m0}\frac{E_{s}(T_{m0})R_{a}L^{2}}{pc_{p}R_{v}^{2}T_{m0}^{2}}} \right)$$
(5)

288 Here  $T_{m0}$  and  $S_{m0}$  are the temperature and relative humidity formed in the parcel after mixing, bu

289 before droplets start evaporating (Appendix B),  $E_s(T_{m0})$  is saturation humidity at temperature  $T_{m0}$ . Since

290  $T_{m0}$ ,  $S_{m0}$  and  $E_s$  are functions of the ratio of  $\mu$ , then  $\delta q$  is function of  $\mu$  as well.

291 Combining Eq.4 and Eq.3 yields dependence of N-versus μ-N vs. μ

$$292 N = N_0 \left( \mu - \frac{\delta q}{q_0} \right) (6)$$

293 Eqs. 4 and 5 describe changes of q and  $\delta q$  for a general case, when the temperatures in the cloud (

294  $T_1$ ) and sub-saturated  $(T_2)$  parcels are different. When  $T_1 = T_2$ , the temperature after mixing, but before

droplet evaporation, remain the same, i.e.  $T_{m0} = T_1 = T_2$ . For this case the amount of liquid water

evaporated after mixing can be estimated as the amount of evaporated liquid required to saturate the

297 entrained parcel

$$298 \qquad \frac{\delta q - (1 - \mu)\delta q^*}{q^*}, \tag{7}$$

299 where

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295296

300

301

$$\mathcal{E}q^* = \frac{L^2}{c_p R_v T_2^2} \ln \left( 1 + \frac{E_S(T_2) R_a L^2}{p c_p R_v^2 T_2^2} \right) + S_2 \frac{E_S(T_2) R_a L^2}{p c_p R_v^2 T_2^2}$$
(8)

is the amount of liquid water required to saturate 1kg of the entrained sub-saturated volume. Thereford,

302 Eq.4 can be rewritten as

303 
$$q = \mu q_0 - (1 - \mu) \delta q^*$$
, (9)

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Eq.8 does not take into account the air temperature changes due to the latent heat of droplet

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$$\beta = \beta_{1} \left( \mu - \frac{(1-\mu)\delta q^{*}}{q_{1}} \right)$$

For a general case when  $T_1 \neq T_2$  the term  $(1 - \mu)\delta q^*$  in Eqs. (12) and (13) should be replaced

308 309

by  $\delta q_m(\mu)$  (Eq.(2)).

310 311

 $\underline{2.6}$ Eqs. 8 and 9 provide accuracy within few percent when the temperature difference  $|T_1 - T_2| < 2^{\circ}C$ .

312

It should be noted that the mass balance Eqs. 4 and 9 do not take into account changes of the size of

313

the volume-V-due to changes of temperature caused by evaporation of droplets. The temperature

314

depression  $\Delta T = |T_1 - T_2|$  due to evaporation depends on  $\delta q$  and it may reach a few degrees depending

315

on- $\mu$ - and  $q_0$  (sections 3a,d). For isobaric processes the relative changes of volume can be estimated as

316

 $\frac{\Delta V}{V} = \frac{\Delta T}{T}$ . So, for the lower troposphere  $\Delta T$ =1C, would result in a relative changes of volume less than

317

0.3%. Therefore, the evaporative cooling within few degrees will not produce any significant effect on-A

318

 $_{r}\beta$ - and  $_{q}$ -during mixing, and the corrections of the volume changes can be neglected.

319

Similar to Eq.6 the dependence of concentration versus  $\mu$  for the case  $T_1 \approx T_2$  can be found as

320

$$N = N_0 \left( \mu - \frac{(1-\mu)\delta q^*}{q_0} \right)$$

The dependences for other moments versus- u- for extreme inhomogeneous mixing can be found the

321 322

same way as in Eq.6 or Eq.10.

323 324

2.4. Homogeneous mixing

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Based on mass conservation one may conclude that the amount of evaporated water does not depend

326

on how the mixing occurred. Therefore, the mass balance equations Eqs. 4 and 9 are valid for bot

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homogeneous and inhomogeneous mixing.

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For homogeneous mixing, when  $\mu > \mu_{cr}$  the droplet number concentration changes only due to dilution of the cloud parcel by the entrained air, i.e.

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$$\frac{N}{N} = \mu \tag{11}$$

$$\frac{N}{N_1} = \mu$$
 (14)

Assuming  $T_1 \approx T_2$   $T_1 = T_{24}$  and substituting Eq. 11 into Eq. 9 yields dependence q-versus N-for

333 homogeneous mixing (5) in (14) yields:

335

334 
$$\frac{N}{N_0} = \frac{q + \delta q^*}{q_0 + \delta q^*}$$
 (12)

Eq.12 suggests linear relationship between 
$$N$$
 and  $q$   $\frac{N}{N_1} = \frac{q + \delta q^*}{q_1 + \delta q^*}$ 

336 (15)

As follows from Eq. (15) N and q are linearly related for homogeneous mixing. However, not linear relationships exist between other moments. Thus, substituting the definition of the liquid

water mixing ratio  $q = 4\pi \rho_w N \overline{r}_3^3 / 3q = \pi \rho_w N D_v^3 / 6\rho_a$  in Eq. 12 (15) yields the relationship

between changes of mean volume droplet size and concentration

341 
$$\frac{\overline{r}^3}{\overline{r}_{30}^3} = 1 + \frac{\delta q^*}{q_0} \left( 1 - \frac{N_0}{N} \right). \tag{13}$$

342 
$$\frac{D_{v}^{3}}{D_{v1}^{3}} = 1 + \left(1 - \frac{N_{0}}{N}\right) \frac{\delta q^{*}}{q_{1}}$$
 (16a)

343 
$$\frac{D_{v}^{3}}{D_{vl}^{3}} = \frac{q}{q_{l}} \left( \frac{q_{l} + \delta q^{*}}{q + \delta q^{*}} \right)$$
(16b)

In a similar way the relationship between the extinction coefficient  $\beta = Q\pi N \overline{r}_2^2$ , concentration

345  $\beta = Q\pi N D_2^2 / 4$ . N and liquid waterq can be obtained written as

346 
$$\frac{\beta}{\beta_0} = \frac{N}{N_0} \left( 1 + \frac{\delta q^*}{q_0} \left( 1 - \frac{N_0}{N} \right) \right)^{\frac{2}{3}}$$
 (14a)

$$\frac{\beta}{\beta_0} = \frac{q}{q_0} \left( \frac{q + \delta q^*}{q_0 + \delta q^*} \right)^{\frac{5}{2}}$$
 (14b)

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$$\frac{\beta}{\beta_{1}} = \frac{N}{N_{1}} \left( 1 + \left( 1 - \frac{N_{0}}{N} \right) \frac{\delta q^{*}}{q_{1}} \right)^{\frac{2}{3}}$$
 (17a)

$$\frac{\beta}{\beta_1} = \left(\frac{q}{q_1}\right)^{\frac{2}{3}} \left(\frac{q + \delta q^*}{q_1 + \delta q^*}\right)^{\frac{1}{3}}$$
(17b)

350 In Eqs. 14a,b (17a) and (17b) it is assumed that  $\frac{1}{r_2} \approx \frac{1}{r_3}$ .  $D_2 \approx D_{\nu}$ .

Substituting in Eq.<del>13</del> (16) the expression for the time of phase relaxation

352  $\tau_p = 1/bN\overline{D}$  (e.g. Squires 1953; Korolev and Mazin, 2003)

$$\frac{\tau_p = \frac{1}{bN\overline{r}}}{15}$$

and assuming  $\bar{r} \approx \bar{r}_3, \bar{D} \approx D_v$  yields

349

359

355 
$$\frac{\tau}{\tau_0} = \frac{N_0}{N} \left( 1 + \frac{\delta q^*}{q_0} \left( 1 - \frac{N}{N_0} \right) \right)^{-\frac{1}{3}}$$
 (16)

For the cases when the temperature difference  $\frac{|T_1 - T_2|}{|T_1 - T_2|} |T_1 - T_2|$  exceeds a few degrees, the

effect of  $\mu \mu$  on  $T_m$  and  $S_m T_m$  and  $S_m$  should be taken into consideration in the calculations of

evaporated water. For such cases  $\frac{\delta q}{\delta q_m}$  (Eq.5) (2) should be used instead of  $\frac{\delta q^*}{\delta q}$  (Eq.8).  $\delta q^*$ 

Using Eq. 11 
$$\delta q_m$$
 can be presented as a function of  $\frac{N}{N_0} \frac{N}{N_1}$  i.e.  $\delta q(\mu) = \delta q \left(\frac{N}{N_0}\right)$ 

361  $-\delta q_m(\mu) = \delta q_m \left(\frac{N}{N_1}\right)$ . Replacing Eq.9\_(5) by Eq.4(1) in the above consideration, the equations

362 Eqs. 12, 13, 14, 16 (15)-(18) can be rewritten as

$$\frac{N}{N_0} = \frac{q + \delta q \left(\frac{N}{N_0}\right)}{q_0} \tag{17}$$

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<sup>&</sup>lt;sup>4</sup> Equalities  $\bar{r}_2 \approx \bar{r}_3$  and  $\bar{r} \approx \bar{r}_3$  are valid for a relatively narrow droplet size distributions

364 
$$\frac{\overline{r_3}^3}{\overline{r_{30}}^3} = 1 - \frac{\delta q \left(\frac{N}{N_0}\right)}{q_0} \frac{N_0}{N}$$
 (18)

$$\frac{\beta}{\beta_0} = \frac{N}{N_0} \left( 1 - \frac{\delta q \left( \frac{N}{N_0} \right)}{q_0} \frac{N}{N_0} \right)^{\frac{2}{3}} \frac{N}{N_1} = \frac{q + \delta q_m \left( \frac{N}{N_1} \right)}{q_1}$$

366 (19

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367 
$$\frac{q^{\frac{1}{3}}\left(q + \delta q\left(\frac{\beta}{\beta_{0}}\right)\right)^{\frac{2}{3}}}{\frac{\beta}{\beta_{0}}} = \frac{D_{v}^{3}}{q_{0}} = 1 - \frac{\delta q_{m}\left(\frac{N}{N_{1}}\right)}{q_{1}} \frac{N_{1}}{N} = \frac{q}{q + \delta q_{m}\left(\frac{q}{q_{1}}\right)}$$

368 (20)

$$\frac{\tau_p}{\tau_{p0}} = \frac{N_0}{N} \left[ 1 - \frac{\delta q \left( \frac{N}{N_0} \right)}{q_0} \frac{N_0}{N} \right]^{-\frac{1}{3}}$$

370 
$$\frac{\beta}{\beta_{1}} = \frac{N}{N_{1}} \left[ 1 - \frac{\delta q_{m} \left( \frac{N}{N_{1}} \right)}{q_{1}} \frac{N}{N_{1}} \right]^{\frac{2}{3}} = \frac{q^{\frac{2}{3}} \left( q + \delta q_{m} \left( \frac{\beta}{\beta_{1}} \right) \right)^{\frac{1}{3}}}{q_{1}}$$
 (21)

371 Equations 17-21 are non-linear and 
$$\frac{\tau_p}{\tau_{p1}} = \frac{N_1}{N} \left( 1 - \frac{\delta q_m \left( \frac{N}{N_0} \right)}{q_1} \frac{N_1}{N} \right)^{-\frac{1}{3}}$$

372 (22)

Eqs. (19)–(22) can be solved numerically.

375 **2.7 Degenerate case** 

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As follows from Eq.(5. Complete evaporation), if

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377 The amount of evaporated water is determined by the initial and final values of the state parameters 378 (e.g. T, e, p) in the mixed parcel, and it does not depend on the type of mixing. Therefore, Eqs valid for both homogeneous and inhomogeneous mixing. 379 For the general case when  $T_1 \neq T_2$  the value of  $\delta q$  is a function of  $\mu$ , i.e.  $\delta q(\mu)$  (Eq.5). Then, in 380 order to find the critical value of  $\mu_{cr}$  , when all liquid water evaporates (i.e. q=0) a non-linear equation 381 382 should be solved  $\mu_{cr}q_0 \quad \delta q(\mu_{cr}) = 0$ 383 384 If the temperature of the entrained and cloud parcels are the same (i.e.,  $T_1 = T_2$ ), then the critical 385 mixing fraction for the case q=0 can be computed from Eq.9 as  $\mu_{cr} = \frac{\delta q^*}{q_0 + \delta q^*}$ 386 Eq.23 yields the condition for the mixing fraction of entrained and cloud parcels such that if  $\mu < \mu_{cr}$  theh 387 all liquid water in the parcel after completing the mixing evaporates (q=0). 388 Figure 2 shows comparisons of dependences of  $\mu_r$  versus  $q_0$  for different  $S_0$  calculated from Eq.28 389 390 and those deduced from a numerical model. As it is seen the agreement appeared to be reasonably good 391 Critical ratios of mixing are also shown by black stars in Figs. 4ab (Eq. 23). As seen from Figs. 4ab the locations of the black stars coincide well with the locations  $-\frac{(1-\mu)}{\mu}\frac{\delta q^*}{q_1}$  << 1\_\_\_\_\_ 392 393 then  $q_1 \ge q \gg \delta q^*$ . If the condition in Eq. (23) is valid, then the terms associated with  $\delta q^*$  in Eqs. 394 (15)-(18) can be neglected. This results in correlation of all moments, i.e.  $N/N_1 = \beta/\beta_1 = q/q$ 395 (compare with Eq.(11)). This corresponds to the degenerate case, when the difference between the 396

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homogeneous and inhomogeneous mixing vanishes. Thus, the dimensionless parameter  $\xi$ 

formed during extremely inhomogeneous mixing.

moments for homogeneous and extremely inhomogeneous mixing.

 $\frac{1-\mu}{\mu}\frac{\delta q^*}{q_1}$  can be used for characterization of proximity of the homogeneous mixing moments to those

changes of  $\xi$ , i.e.  $0 \le \xi \le 1$  for the mixing without complete evaporation of droplets. The

degenerate case corresponds to  $\xi \to 0$ , whereas  $\xi \to 1$  corresponds to maximum difference of the

The range of  $\mu$  in  $\xi$  is limited by  $\mu_{cr} < \mu \le 1$ , so that  $0 < \frac{1-\mu}{\mu} \le \frac{q_1}{\delta q^*}$ . This gives the range of

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As follows from Eqs. (4) and (23) approaching to the degenerate case  $(\xi \to 0)$  occurs, when one of the following conditions or their combination is satisfied: (a)  $RH_2 \to 1$ ; (b)  $E_s(T) \to 0$  at low temperatures; (c)  $q_1 \gg \delta q^*$ ; (d)  $\mu \to 1$ . The effect of RH, T,  $q_1$  and  $\mu$  on mixing will be demonstrated in Sect.3.

Figure 3 shows dependence of  $\xi$  vs.  $\mu$ . The grey area in Fig.3 indicates the region where identification of type of mixing from in-situ measurements (Sect.5) may be hindered due to proximity of the moments for homogeneous and inhomogeneous mixing. Thus for  $\delta q^*/q_1 = 0.01$  identification of type of mixing is ambiguous for nearly the entire range of  $\mu$ .

For the general case, when  $T_1 \neq T_2$ , it should be  $\xi = \frac{|\delta q_m(\mu)|}{|\mu q_1|}$ . An absolute value  $|\delta q_m(\mu)|$  should be used in  $\xi$  since  $\delta q_m(\mu)$  can be negative (Appendix A, Fig.A1) if mixing results in supersaturation Sect. 3.4).

The coefficient  $\xi$  may be useful for identification type of mixing from in-situ observations. It is worth nothing, that the ratio  $\frac{\delta q^*}{q_1} \cong \frac{s_2}{A_2 q_1}$  is equal to the parameter R (Pinsky et al. 2015ab), which plays an important role in determining scenarios of droplet evaporation in turbulent environment.

# 3 Comparisons with numerical simulations

where modeled -LWC and -N become zero. The obtained agreement between analytical and modeled  $\mu_{rr}$  in Figs. 3 and 4 validates the developed approach.

Numerical simulations of the final stages of homogeneous and inhomogeneous, mixing were

# 3. Modeling

performed in order to examine the accuracy and limitations of the analytical expressions obtained in the previous section. The simulations have been conducted with the help of a parcel model similar to that described in Korolev (1995). The ensemble of droplets in the simulation was assumed to be monodisperse. For the case of extreme inhomogeneous mixing the amount of evaporated water  $\Delta q$  required to saturate the mixed volume was calculated first. If  $\Delta q < \mu q_1$ , then the concentration of evaporated droplets was calculated as  $N_{ev} = \frac{\Delta q}{m_d} \rho_a$ , where  $m_d = \pi \rho_w D^3 / 6$ . Then, the concentration of the remaining droplets  $N = N_1 - N_{ev}$  was

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432 recalculated based of the calculation on the volume formed after mixing. If  $\Delta q \ge \mu q_1$ , then all droplets evaporate, and N = 0.

For the case of inhomogeneous mixing the amount of evaporated water  $\Delta q$  required to saturate the mixed volume was calculated first. If  $\Delta q < q_0$ , then the concentration of evaporated droplets was

436 calculated as  $N_{ev} = \frac{\Delta q}{\overline{m_d}} \rho_a$ , where  $-\overline{m_d} = 4\pi \rho_w \overline{r_3}^3/3$  is the average mass of a droplet. Then the

concentration of the remaining droplets  $N = N_0 - N_{ev}$  was recalculated based of the calculation on the

volume formed after mixing. If  $\Delta q \ge q_0$ , then all droplets evaporate and N=0

For the case of homogeneous mixing in the first step the engulfed parcel instantly mixes with the cloud parcel resulting in a new humidity  $S_{m0}$ ,  $RH_{m0}$ , temperature  $T_{m0}$  and volume  $V_{m0}$ . Then  $V_{m0}$ . After that the droplets start evaporating until either their complete evaporation or saturation over liquid is reached. The calculations stopped when, either T < D < 0.1 mm / D = 0.1 mm /

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# 3.1, Effect of mixing ratio fraction

Figure 34 shows the results of the simulation of the dependence of the droplet number concentration (N), droplet integral radius ( $N\bar{r}$ ), extinction coefficient ( $\beta$ ), liquid water mixing ratio (q) mean cube droplet radius ( $\bar{r}_3$ ), and time of phase relaxation ( $\tau_p$ ), relative humidity ( $S_{m0}$ ,  $S_m$ ) and find temperature ( $T_m$ ) versus mixing fraction  $\mu_{\tau}$  different moments and state parameters vs.  $\mu_{\tau}$ . The calculations were performed for different saturation ratios relative humidity of the entrained parcel  $S_{20} = 0.2$ ,  $RH_2 = 0.2$ , 0.5, 0.8 and 0.95. As seen from Fig.34 for the case of homogeneous mixing only N and q are linearly related with  $\mu_{\tau}$ ,  $\mu_{\tau}$ , the rest of the variables have non-linear dependences on  $\mu_{\tau}\mu_{\tau}$ . For the case of inhomogeneous mixing all f(r)f(D) moments and droplet sizesizes linearly depend on the mixing fraction when  $\mu > \mu_{cr}$ . When  $\mu \leq \mu_{cr}\mu_{\tau}$ . Note, for  $\mu \leq \mu_{cr}$  all liquid evaporates, moments are equal to zero.

Since the amount of the evaporated liquid water does not depend on the type of mixing, the dependences of  $q(\mu)q(\mu)$  are the same for both homogeneous and inhomogeneous mixing (Fig. 3-4a). The type of mixing has the most pronounced effect on the droplet concentration

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(Fig. 3b4b) and droplet sizes (Fig. 3e). The obtained results are in a good agreement with the analytical predictions discussed in section 2.4e).

Figure 3g4g shows the dependences  $S_{m0}(\mu)$  and  $S_m(\mu)$   $RH_{m0}$  and RH vs.  $\mu$ . Here  $S_{m0}(\mu)$   $RH_{m0}$  is the relative humidity at the initial stage of homogeneous mixing before droplets start evaporating (Fig. 1b2). Figure 3h presents dependences of  $T_m(\mu)$  for different  $S_2$ . It is worth noting that  $S_m(\mu)$  comparisons of modeled  $T(\mu)$  and those calculated from Eqs.(6a,b) and  $T_m(\mu)$  do not depend on the type of mixing and they are the same for homogeneous and inhomogeneous mixing. Basically the (C4). The independence of  $q(\mu)$ ,  $S_m(\mu)q(\mu)$ ,  $RH(\mu)$  and  $T_m(\mu)T(\mu)$  on type of mixing (Fig.4a,g,h) is the consequence of the mass and energy conservation laws, which are not contingent on type of mixing.

## 3.2 Effect of humidity of entrained air.

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486 487 The diagrams in Fig. 4 <u>5a-c</u> show the dependences of normalized  $\beta$ , q and  $\bar{r}_3$  versus  $N/N_0$   $\beta$ , q and  $D_v$  vs.  $N/N_0$  calculated from numerical simulations and analytical equations from section Sect. 2. The calculations were performed for different humidity of the entrained air  $S_2$ ;  $RH_2$ . As seen from Fig. 4 <u>5a-c</u>, the normalized dependences q(N),  $\beta(N)$  and  $\bar{r}_3(N)$  for homogeneous mixing q(N),  $\beta(N)$  and  $D_v(N)$ , tend to approach the line of extreme inhomogeneous mixing when relative humidity  $S_2RH_2$  approaches to 100%. As it was indicated in previous studies (e.g. Burnet and Brenguier 2007), mixing with the entrained saturated air  $(S_2 = 1)$  represents of 1. This is consistent with the degenerate case, when there is no difference between homogeneous or inhomogeneous mixing.  $\xi \to 0$  (Sect. 2.7). In this case droplets behave as a passive admixture, and they do not interact with the environment. This effect is clearly seen from the diagrams in Fig. 4.

# 3.3 Effect of liquid water contentmixing ratio

Figure 5 presents the normalized dependences q(N),  $\beta(N)$  and  $\overline{r}_3(N)$  calculated for different initial liquid water  $q_0$  in the cloud parcel, but the same humidity of the entrained parcel  $S_2$ =0.5. Figure 5 shows 5d-f demonstrate the sensitivity of q(N),  $\beta(N)$  and  $D_v(N)$  to liquid water mixing ratio  $q_1$ . It is seen, that the increase of  $q_0$  results in q(N),  $\beta(N)$  and  $\overline{r}_3(N)$  (which were  $q_1$  results in q(N)  $\beta(N)$  and  $D_v(N)$  (calculated for homogeneous mixing) approaching towards  $\overline{q(N)}$ ,  $\beta(N)$  q(N)

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 $\beta(N)$  and  $\frac{1}{r_3}(N)D_{\nu}(N)$  for the inhomogeneous mixing. In other words, the sensitivity of the microphysical parameters to the type of mixing increases with the decrease of  $q_0$ ,  $q_1$ . From a practical viewpoint it means, that from in-situ observations the difference between homogeneous and inhomogeneous mixing is anticipated to be more pronounced for the cases with a relatively low  $\frac{1}{1} \frac{1}{1} \frac{1}{1}$ 

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# 3.4 Effect of temperature $T_1 = T_2$

Figure 5g-j shows the effect of temperature on the normalized q(N),  $\beta(N)$  and  $D_{\nu}(N)$  for  $T_1 = T_2$ . Figure 6 shows the effect of temperature on the normalized dependences q(N),  $\beta(N)$  and  $\overline{r_3}(N)$  calculated for the case  $T_1 = T_2$ . The relative humidity of the entrained parcel was assumed to be the same for all cases ( $S_2$ =0.5). Figure 6 suggests that the difference between the f(r)5g-j indicate that the difference between the moments becomes most pronounced at warm temperatures, whereas at cold temperatures (e.g. T=30C) q(N),  $\beta(N)$  and  $\overline{r_3}(N)$  T=-30°C), q(N),  $\beta(N)$  and  $D_{\nu}(N)$  for homogeneous mixing are approaching those for the extreme inhomogeneous mixing limit.

Such behavior is explained by the fact that the amount of liquid water deficit  $\frac{\partial q}{\partial q_m}$  decreases with decreasing temperature. The effect of T on  $\frac{\partial q}{\partial q_m}$  is demonstrated in (appendix A, Fig. 81 (Appendix BA1)). At low temperatures ( $T < -30^{\circ}\text{C}(T = -30^{\circ}\text{C})$ ) the amount of evaporated water  $\frac{\partial q}{\partial q_m}$  is so small, that homogeneous mixing with undersaturated dry out-of-cloud air will have approximately the same effect as mixing with saturated air (i.e. degenerate case, Sect. 2.7).

It is well established that isobaric Overall, as follows from Fig. 5 the results the analytical predictions (Sect. 2) turned out to be in a good agreement with numerical simulations.

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# 3.5 Effect of temperature $T_1 \neq T_2$

Isobaric mixing of two nearly saturated parcels having different temperatures results volumes with  $T_1 \neq T_2$  may result in the formation of supersaturated airenvironment (e.g. Rogers, 1976; Bohren and Albrecht, 1998). However, as it was shown in Korolev and Isaac (2000), the isobaric Mixing resulting in supersaturation is different in principle from the mixing of saturated cloudy and

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undersaturated air may also result in supersaturated air. The formation with evaporating droplets. In 517 518 this case the meaning of homogeneous and inhomogeneous mixing becomes ambiguous 519 Formation of <del>supersaturated air supersaturation</del> leads to different dependences between NF, β, q 520  $\overline{r}$  and  $N N \overline{D}_1 \beta_1 q_1 \overline{D}_1$  and N as compared to those shown in Figs. 3-6-4, when  $T_1 = T_2$ . In the case 521 discussed above, mixing between cloud and sub-saturated entrained air with 522 evaporation of some fraction of cloud liquid water, i.e. it was always  $\delta q < 0$ . However, mixing that cau 523 supersaturation will result in condensation of liquid water (e.g.  $\delta q > 0$  ), which is different of the case considered above.  $T_1 = T_{2}$ 524 Figure 76 presents a set of diagrams similar to those in Fig. 34, but calculated for the eloud and 525 entrained sub-saturated parcels having different temperatures  $T_1$  and  $T_2$  cases when  $T_1 \le T_2$ . It turns 526 out that for the case of extreme inhomogeneous mixing the temperature difference between  $T_1$  and 527 528  $T_2T_1$  and  $T_2$  breaks down linear dependences between  $\mu$  and f(r) of the microphysical moments 529 (e.g.  $N\overline{r}$  , eta , g -Fig.7a,b,c). This happens because positive supersaturation ( $S_{m0}>1$ ), which may form 530 after mixing, will result in an increase of  $q_{\tau}$ ,  $\beta_{\tau}$ ,  $\bar{\tau}_{3}$ , etc. The droplet concentration still holds the linea 531 relationship with  $\mu$  (Fig. 7b), since no formation of new droplets were allowed when  $S_{m0} > 1$ .  $N\overline{D}$ ,  $\beta$ 532 q Fig. 6a,c,d) vs. μ. Figure 87 presents the effect of the temperature difference  $\Delta T \Delta T$  on the normalized 533 dependences q(N),  $\beta(N)$  and  $\bar{r}_1(N)$ . As seen from Fig. 8 for large temperature differences  $\Delta T = 10$ 534 535  $^{
m e}$ C, the values of q , eta ,  $ar{ au}$  for inhomogeneous mixing exceed those for inhomogeneous mixing for the 536 same concentration N -It is worth nothing that in real q(N),  $\beta(N)$  and  $D_{\nu}(N)$ . In clouds, high supersaturation resulting 537 from isobaric mixing may lead to activation of interstitial CCN, which may slow down the growt 538 539 of  $\bar{r}_3$  and increase  $\frac{4N}{2}$  and decrease  $D_v$  (Korolev and Isaac, 2000). However, no activation of new droplets during isobaric mixing was allowed in this study. For the cases when  $\frac{S_{m0} > 1RH_{m0}}{S_{m0}}$ 540 (Fig.8, section AB 7, AB on line 1) the condensed water was uniformly distributed between available 541 droplets, Therefore, q(N),  $\beta(N)$  and therefore, q(N),  $\beta(N)$  and  $\overline{r_3(N)}D_v(N)$  calculated for 542 homogeneous and extremely inhomogeneous mixing to coincide with each other. 543

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Numerical simulations also showed, that the effect of temperature on the result of mixing is more pronounced for the cases when the cloud temperature is warmer than that of the entrained air, i.e.  $T_1 > T_2$ ,  $T_1 > T_2$ , as compared to the cases with  $T_1 < T_2$ .  $T_1 < T_2$ .

## 3.5. Multiple 4. Progressive mixing events

#### 4.1 Effect on microphysical parameters

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In the previous sections the mixing between cloud and sub-saturated volumes, was considered as a single event, i.e.  $\mu$  fraction of the cloudy air mixed up with  $(1 - \mu)$  fraction of entrained dry air, Such mixing will be referred to as "primary" mixing. Primary mixing results in an ensemble of elementary volumes characterized by a set of microphysical and state parameters i.e.  $\overline{\tau}(\mu)$ ,  $N(\mu)$ ,  $N(\mu)$ ,  $N(\mu)$ ,  $N(\mu)$ ,  $N(\mu)$ ,  $N(\mu)$ , and what is important, these parameters have functional relationships between each other.

In reality mixing is a continuous process. It does not stop after the primary mixing. The elementary volumes formed after primary mixing continue to progressively mixing between mix with each other, cloud environment and newly entrained air.

The second stage of mixing will result in an ensemble of elementary volumes characterized by a set of parameters  $\overline{T^{(2)}}$ ,  $N^{(2)}$ ,  $T^{(2)}$ ,  $D^{(2)}$ ,  $N^{(2)}$ ,  $N^{(2)}$ ,  $N^{(2)}$ ,  $N^{(2)}$ , etc. Here the superscript indicates the stage of mixing. After the second stage the mixed volumes undergo subsequent stages of mixing. The conceptual diagram of the progressive mixing is shown in Fig. 9. This is a highly idealized scheme of mixing. However, it helps understand some features of homogeneous mixing.

The progressive The idealised conceptual diagram of the progressive mixing is shown in Fig. 8. As mentioned in Sect. 2.1, the actual process of mixing is indeed much more complex than the sequence of discrete events portrayed in Fig. 8. However, as it will be shown below, this simplified consideration of allows establishing main features of evolution of relationships between the microphysical moments affected by mixing. The obtained results facilitates identification of type of mixing from in-situ measurements.

Progressive mixing was simulated with the help of a numerical model, where parcels were randomly mixed with each other and with the cloud environment. The mixing fraction  $\mu\mu$  was also set to be random during each mixing event. Models of randomstochastic mixing have been

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used in a number of studies (e.g. Krueger et al., 1997, Su et al., 1998, Burnet and Brenguier, 2007). Burnet and Brenguier (2007) allowed mixing between cloud and entrained parcels at each stage of mixing. In the present approach work the mixing with analysis of progressive mixing is expanded to examine its effect on the entrained environment was allowed only at relationship between moments of the primary stage. After that the mixing continued with the cloud air and between the mixed volumes droplet size distribution.

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The results of the progressive mixing for the first four stages are presented in Fig. 10.9. As seen from Fig. 10.9 the functional relationship between the pairs of microphysical and state parameters exists only for the primary stage. However for For higher mixing stages these functional relationships breakdown break down. Thus, cloud volumes with the same  $\overline{N}^{(2)}$  will be associated with an ensemble of droplets with  $N^{(2)}$  may have different  $\overline{\tau}^{(2)}$ . Figure 109 also shows that the regions of scattering of  $\overline{q}(N)$ ,  $\beta(N)$  and  $\overline{r}_3(N)$ . D<sub>v</sub>(N) for stages 2, 3 and 4 are limited from above by the inhomogeneous mixing (red dashed lines) and from below by primary homogeneous mixing (red solid lines).

Figure 10 presents a conceptual N-q diagram explaining breaking the functional relationships during progressive homogeneous mixing. After the first stage of mixing the N-q points will be scattered along the line OB and point C. The line OB corresponds to the ensemble of points with RH = 1. Therefore, result of mixing between two saturated volumes randomly selected on AB, will remain on the same line. Point C corresponds to the ensemble of points with N = 0,  $RH_2 \le RH_C(\mu^{(1)}) \le 1$ , where  $0 \le \mu^{(1)} < \mu_{cr}$ . Therefore, mixing between point A (Fig.10) and point C, when RH = 1 will result in scattering along the line AC (degenerate case). Points resulted from mixing between A (RH = 1) and point C, when  $RH_2 \le RH_C < 1$ , will scattered over the ensemble of dashed lines shown in Fig.10. These lines will fill the sector CAB. Random mixing between points on the line OB and C, will eventually fill the entire sector COB. The same consideration can be applied to progressive mixing between other moments.

During the progressive mixing  $N^{(n)}$ ,  $q^{(n)}$  and  $q^{(n)}$  and  $q^{(n)}$  and  $q^{(n)}$  formed in the elementary parcels tend to approach those in the undiluted cloud, i.e.  $N_0$ ,  $p_0$ ,  $q_0$  and  $q_0$ ,  $q_0$ ,  $q_0$  and  $q_0$ ,  $q_0$ ,

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and  $\bar{\tau}_3^{(n)} D_{\nu}^{(n)}$  during the progressive mixing can be seen in Fig. 10.9, where the scattering of normalized  $q^{(n)}(N)$ ,  $\beta^{(n)}(N)$ ,  $\beta^{(n)}(N)$  and  $\bar{\tau}_3^{(n)}(N)$  becomes denser towards the top-right corner (1,1) with the increase of the stage of mixing.

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Another interesting feature of It is worth noting that progressive mixing with the dry air does not break the functional relationships between the moments. This case is equivalent to detrainment of cloudy environment into dry air. It can be shown that Eq.(14) remain valid at any stage of progressive homogeneous mixing is that it results in a population of points along the line corresponding to inhomogeneous mixing (dashed red line in Fig.10) In other words during homogeneous with dry air only, i.e.  $N_j/N_1 = \mu^{(1)} \cdots \mu^{(j-1)} \mu^{(j)}$  where  $\mu^{(j)}$  is the mixing some-fraction at the j-th stage of elementary volumes may develop properties of inhomogeneous mixing. Similar conclusion was derived from a more comprehensive analysis of Eqs. (15)-(24) also remain valid for the progressive mixing in Pinsky et al. (2015b), with the dry air only.

As follows from Eq.4 (9) for the case of extreme inhomogeneous mixing the progressive mixing does not affect the functional relations between  $N^{(n)} - \beta^{(n)} - \overline{r_3^{(n)}} N^{(n)} - \overline{r_3^{(n)}}$ 

## 3.6. Droplet 4.2 Effect on droplet size distribution distributions.

Figure 11 shows modeled droplet size distributions averaged over an ensemble the ensembles of elementary volumes corresponding to the first four stages of homogeneous mixing. As seen from Fig. 11a,b,c, d for the case with  $T_1 = T_2 T_1 = T_2$  the droplet size distributions are broadenbroadened towards small sizes. Depending on the stage of mixing and mixing fraction  $\mu$  the size distributions formed in each elementary volume may be unimodal or multimodal. However, due to the random nature of the modal sizes formed during mixing, the average size distributions become smooth and unimodal (Fig.11a-d).

Broadening of the droplet size distributiondistributions towards small sizes during homogeneous mixing is well known and it was demonstrated in a number of studies (e.g. Baker and Latham, 1982; Jensen and Baker, 1989; Jeffery, 2007; Kumar et al., 2013). However, if the

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difference  $|T_1 - T_2|$  exceeds few degrees and the humidity of the sub-saturated parcel is not too low (e.g.  $\geq$ 80%), then homogeneous mixing may result in supersaturation of the mixed parcel and subsequent droplet growth. In this case-mixing results in supersaturation (section 3.4), then the droplet size distribution may broaden towards large sizes (Fig.11e h)-larger sizes (Fig. 11e-h). For this to occur, both the temperature difference between the cloud and the environment  $|T_1 - T_2|$  and the relative humidity of the environment  $|T_1|$  must be sufficiently large. Such conditions are inherently unstable, however, this might occur in regions that have been moistened through prior cloud detrainment. Thus homogeneous mixing may result in broadening of droplet size distributions towards either smaller or larger sizes (Fig.11).

The examples in Fig.11 show that homogeneous mixing results in broadening of size distribution. Depending on the temperature difference  $|T_1-T_2|$  and the relative humidity of the entrained air the broadening may occur both towards smaller and towards larger sizes. Contrary to homogeneous mixing, extreme inhomogeneous mixing does not result in broadening of droplet size distributions.

These results were obtained in the frame of the formalism of homogeneous and inhomogeneous mixing. The following two works in this series (Pinsky et al. 2015a, ... 2016a, b) show that f(r) will discuss the broadening towards small droplets may occur of polydisperse and monodisperse f(D) during both homogeneous and inhomogeneous mixing in greater details.

3.7. Summary

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### ₩5 Identification of type of mixing from in-situ observations

The purpose of this section is to attempt identifying type of mixing based on examining relationships between basic microphysical parameters  $N_{\star}\beta_{\star}LWC_{\star}D_{v}$  measured from in-situ.

# 5.1 Expected relationships between the moments

Prior proceeding with the analysis of in-situ data we summarize the results of the previous consideration on how homogeneous and extreme inhomogeneous mixing is expected to manifest itself in relationships between basic microphysical parameters, such as N,  $\beta$ , q and  $\bar{r}_3$ . The objective of this analysis is to facilitate examination of in situ measurements in order to identify the type of mixing within the conventional framework of homogeneous and extreme inhomogeneous mixing. N,  $\beta$ , q and  $D_{v_2}$ 

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662	Tallania Tan antonna inhama ann minina dha mlatian shin hatanan dha naim af N. O.N.	Formathed Seek Times New Devices 12 et Seek selew
662	Following For extreme inhomogeneous mixing the relationship between the pairs of $\frac{N}{r}$ , $\frac{\beta}{\beta}$ .	Formatted: Font: Times New Roman, 12 pt, Font color: Black Formatted: Indent: First line: 0.75 cm, Don't adjust space
663	$\beta$ and $q\underline{q}$ are determined by a-linear relationship $M_n = \alpha_{nm} M_m$ dependences $M_n = \alpha_{nk} M_k$ (Eq.2).	between Latin and Asian text, Don't adjust space between Asian text and numbers, Tab stops: Not at 13.65 cm
664	10) at any stage of mixing. As follows from Eq.3 (11) the slopes $\alpha_{nm}\alpha_{nk}$ for $q(N)-\beta(N)q(N)$ ,	Formatted
665	$\beta(N)$ and $\frac{q(\beta)}{q(\beta)}q(\beta)$ are equal to the ratios $\frac{q_0/N_0}{\rho_0/N_0} - \frac{q_1/N_1}{\rho_0/N_0} + \frac{\beta_1/N_1}{\rho_0/N_0}$ and $\frac{q_0/\beta_0}{\rho_0/N_0} + \frac{\beta_1/N_0}{\rho_0/N_0} + \beta_1$	Formatted
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666	forming through vertical motions (e.g. Sc, Ac, Cu, Cb etc.) $N_0$ , $\beta_0 q_1/\beta_1$ , respectively, where $N_1$ , $\beta_1$	Formatted: Font: Times New Roman, 12 pt, Font color:
667	and $q_0q_1$ correspond to undiluted adiabatic values. The values of $N_0$ , $\beta_0N_1$ , $\beta_1$ and $q_0$ depends	Formatted: Font: (Default) Times New Roman, 12 pt, Font color: Black
668	of q <sub>1</sub> may vary depending on the location of measurements inside the cloud and the environmental	Formatted: Font: Times New Roman, 12 pt, Font color:
660	conditions at the cloud base. Thus, the ediabetic value of a guise function of clovetion above the	Formatted: Font: Times New Roman, 12 pt, Font color:
669	conditions at the cloud base. Thus, the adiabatic value of $q_0q_1$ is a function of elevation above the	Formatted: Font: Times New Roman, 12 pt, Font color:
670	cloud base $\Delta Z$ . The concentration $N_0 \Delta Z$ , whereas $N_1$ depends on the vertical velocity at the cloud	Formatted
674	have a send the consent load. The estimation of fixing the control of the best AZ and a	Formatted: Font: Times New Roman, 12 pt, Font color:
671	base $u_z u_{z_z}$ and the aerosol load. The extinction coefficient $\beta_0$ is determined by both $\Delta Z$ and $u_z$ .	Formatted: Font: Times New Roman, 12 pt, Font color:
672	Therefore, for the measurements collected in the same cloud, but at different altitudes, it is anticipated	Formatted
673	that the the scattering of $N_0$ , and $N_0$	
674	measurements $q(N) - q - N$ points will be aligned along an ensemble of different lines associated	Formatted: Font: Times New Roman, 12 pt, Font color: Black
675	withdetermined by $q_1/N_1$ , which are specific to different $q_0/N_0$ -cloud volumes. The conceptual	Formatted
676	diagram of the scattering of $\frac{q}{q} - \frac{N}{q} - \frac{N}{q}$ measurements in a cloud having experienced with extreme	Formatted
677	inhomogeneous mixing is shown in Fig. 12a. The scatter diagrams of q β and N β for other	
678	moments (e.g. $q - \beta N - \beta$ ) will have the same similar patterns as that in Fig. 12a.	Formatted
679	For the case of homogeneous mixing the functional relationship between the pairs of $\frac{N}{r}$ , $q_{\tau}$	
680	$\beta$ and $\overline{r_3}$ is $N$ , $\beta$ , $q$ and $D_v$ are disrupted by a progressive mixing. As shown in section 3eSect. 4.1	Formatted
681	the ensemble of points of $N - q$ , and $\beta - N$ , $\beta$ and $q$ will be scattered within a sector, which is limited	Formatted
682	by lines determined by Eq. <del>3 ((11) (extreme inhomogeneous mixing) and Eq.12 and Eq.15a</del> Eqs.	
683	(15)-(17) (primary homogeneous), respectively (Fig. 109). What is important, is that the top of the	
684	cones of the scatter plots $q(N)$ and $\beta(N)$ sectors for $q(N)$ and $\beta(N)$ correspond to points $\{N_0, q\}$	Formatted: Font: Times New Roman, 12 pt, Font color:
685	$[N_1, q_1]$ and $[N_0, \beta_0]$ , $[N_1, \beta_1]$ respectively. Since $N_0, \beta_0 N_1, \beta_1$ and $q_0 q_1$ may vary both within	Formatted: Font: Times New Roman, 12 pt, Font color:
686	the same cloud—as well as from cloud to cloud, it is anticipated that the $N-q_7N$ , $\beta$ and $\beta$	Formatted: Font: Times New Roman, 12 pt, Font color:  Formatted: Font: Times New Roman, 12 pt, Font color:
687	measurements will be scattered within an ensemble of conessectors as shown in Fig. 12b.	Formatted
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In the case of inhomogeneous mixing, which occurs within a cloud volume with the same  $N_0$  and  $q_0$ , the scattered points are aligned along the same line, e.g.  $q=\frac{q_0}{N_0}N$ . Such scattering patterns are easy to identify even when there are a limited number of measured points. However, in the case of homogeneous mixing the points will be scattered within a large area limited by the cone with the top point  $[N_0, q_0]$  (Fig.12b). Even in case of a relatively large number of measurements the identification of such scattering patterns may be hindered because of the low density of points scattered over a large area. Ultimately, when looking at the feasibility of identifying homogeneous and inhomogeneous mixing by means of microphysical measurements it can be concluded that, when using the analysis of scatter-diagrams q(N),  $\beta(N)$  and  $q(\beta)$ , inhomogeneous mixing is easier identify than homogeneous mixing

#### 4. In-situ observations

This section presents an attempt to identify type of mixing based on examining relationships between basic microphysical parameters N,  $\beta$ , LWC,  $\bar{r_3}$  measured from in situ. These parameters present different moments of droplet size distribution and depending on instrumentation set used airborne, they can be either measured directly or calculated from measurements of f(r).

**3.1.** It is important to note that that during homogeneous mixing prior reaching equilibrium, functional relationships between the microphysical moments do not exist either. After the instant mixing of cloud fraction  $\mu$  with entrained air (Fig. 1b(2)),  $q_{m0} = \mu q_0$  and  $N_{m0} = \mu N_0$ . This state corresponds to point D in Fig.10. After that droplets start evaporating until liquid mixing ratio reaches point A (Fig.10), which corresponds to the equilibrium state (RH = 1). Therefore, during evaporation time q - N points will be scattered along the line AD. Since, point D can be located anywhere on D0, the ensemble of D1 points corresponding to non-equilibrium state will fill the D2 area.

Thus, the absence of the functional relationships between the moments during homogeneous mixing may occur both during progressive mixing and during primary mixing prior reaching the equilibrium state. The evaporation time required to reach equilibrium during homogeneous mixing is discussed in details in Pinsky et al. (2016b), and it is usually limited by few tens of seconds. However, progressive mixing is not limited in time. Therefore, it is very likely that no functional relationship between microphysical parameters will be observed during in-situ measurements.

Fig.12 demonstrated a fundamental difference in scattering of q - N for homogeneous and extreme inhomogeneous mixing, which will be used to facilitate identification of type of mixing in the following section.

#### **5.2** Results of observations

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The measurements were obtained on the University of Wyoming King Air aircraft during the COPE-MED project in South-Western part of UK during July-August 2013 (Leon et al. 2015) 2016). The UW King Air was equipped with a suite of microphysical instruments, including DMT Cloud Droplet probe (CDP) and PMS Forward Scattering Spectrometer Probe (FSSP-100). These probes were), designed for measurements of droplet sizes and their concentrations in the nominal size ranges 1-\_50μm and 3-47μm, respectively μm.

The measurements were focused on characterizing microphysical evolution of convective clouds and <del>effect of entrainment on precipitation formation. </del>Figure 13 shows a time series of droplet concentration, extinction coefficient, liquid water content and mean volume droplet diameter measured by the CDP during transit through a convective cell on 18 July 2013. The CDP data were sampled at 10Hz, which corresponds to approximately 10m spatial averaging. Visual examination of the spatial changes of N, β and LWCN, β and LWC shows strong correlation. The amplitude of changes of these parameters reaches nearly one hundred percent with respect to their maximum. Contrary to that, the spatial variations of  $\overline{D}$  and  $\overline{D}_3\overline{D}$  and  $D_v$  are quite conservative and their values remain nearly constant. With the exception of two cloud holes between 13:50:42 and 13:50:44, the amplitude of fluctuations of  $\overline{\mathcal{D}}_3 D_{\nu}$  does not exceed 8% with standard deviation of 2.2%.

Figure 14 shows scatter diagrams of  $\frac{LWC(N), \beta(N), LWC(\beta)}{LWC(N), \beta(N), LWC(\beta)}$ and  $\overline{D}_3(N)D_v(N)$  measured by the CDP during seven consecutive penetrations of the same convective cell extended over a period of approximately 19 minutesmin. One of these penetrations is shown in Fig. 13. The measurements were conducted at <del>an altitude H= <u>H</u> = </del>5500m and <del>temperatur</del> 7=120. The relative humidity of the ambient air was approximately 20\_%. At the beginning of the sampling no precipitation size particles were observed in the cloud. However, by the end of the sampling period some raindrops and ice crystals were present in the cloud. Despite the presence of some precipitation size particles, the scatter diagrams in Fig. 14a, b, and

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demonstrate high correlation between pairs <del>N , β and LWC , N , β and LWC , The</del> mean volume Formatted 748 diameter in Fig.\_14c shows very little changes from 19 µm to 17 µm µm when concentration 749 changes from 1100 to 500 cm<sup>-3</sup> to 500cm<sup>-3</sup>. However, for N < 200cmN < 200 cm<sup>-3</sup>, the volume 750 751 diameter decreases to 12-15 µmµm. Red lines in Fig. 14 indicate q(N),  $\beta(N)$ ,  $LWC(\beta)$ , q(N),  $\beta(N)$ ,  $LWC(\beta)$  and  $\overline{D}_3(N)$ ,  $D_v(N)$ **Formatted** 752 Formatted: Font: Times New Roman, 12 pt, Font color: 753 calculated for the 1st stage of homogeneous mixing. The calculations were performed for **Formatted** 754 monodisperse  $\frac{f(r)f(D)}{D_0}$  with  $\frac{D_0 = 18.5 \text{ } \mu\text{m}D_1 = 18.5 \mu\text{m}}{D_1 = 18.5 \mu\text{m}}$ ,  $N_1 = 1100 \text{ } \text{cm}^3$ , and  $\frac{N_0 = 1100 \text{ } \text{cm}^3}{D_0 = 18.5 \mu\text{m}}$ . The Formatted: Font: Times New Roman, 12 pt, Font color: **Formatted** 755 <del>initial</del>-state parameters <del>were considered to be the same as during the measurements. Comparisons of</del> dependences  $\frac{q(N), -\beta(N), -LWC(\beta)}{q(N), \beta(N), LWC(\beta)}$  and  $\frac{\overline{D_3}(N)}{D_v(N)}$  based on in-situation Formatted: Font color: Black 756 **Formatted** 757 measurements with those obtained from numerical simulations of homogeneous mixing show minor difference for high concentrations  $\frac{500\text{cm}}{700}$  cm<sup>-3</sup>  $\frac{N}{100\text{cm}}$  N < 1100 cm<sup>-3</sup> (Fig. 14a,b, c 758 759 Simulation\_also shows that for this specific case the difference between homogeneous and inhomogeneous mixing does not exceed 10% when  $\frac{500 \text{cm}^{-3}}{N} < \frac{1100 \text{cm}^{-3}}{100 \text{cm}^{-3}} < \frac{N}{N} < \frac{1100 \text{cm}^{-3}}{N} < \frac{N}{N} < \frac{1100 \text{cm}^{-3}}{N} < \frac{N}{N} < \frac{1100 \text{cm}^{-3}}{N} < \frac{N}{N} < \frac{N}{N$ 760 **Formatted** Such difference remains within the errors of measurements. Therefore, in this specific cloud for the 761 762 regions with  $\frac{N > 500 \,\text{N}}{200} \,\text{cm}^3$  the type of mixing cannot be unambiguously identified from the Formatted (... Formatted: Font color: Black 763 analysis of the dependences q(N),  $\beta(N)$ ,  $\beta(N)$ ,  $\beta(N)$ ,  $\beta(N)$ ,  $\beta(N)$ ,  $\beta(N)$ ,  $\beta(N)$ However,  $D_n(N)$ . This is consistent with the assessment of feasibility of segregation of 764 homogeneous and inhomogeneous mixing in Fig.3 (dashed line). Since for homogeneous mixing 765 Formatted: Font: Times New Roman, 12 pt, Font color:  $N \propto \mu_{L}$  than Fig.3 suggests good separation of the moments for N > 700 cm<sup>-3</sup>. 766 For the regions with  $N < 500 \text{cm}^{-3}$  the deviation between homogeneous 767 Formatted: Indent: First line: 0.75 cm. Don't adjust space between Latin and Asian text, Don't adjust space between <del>simulation</del>mixing <u>simulations</u> and in-situ measurements <u>in Fig.14</u> becomes well pronounced and i Asian text and numbers 768 Formatted: Font: Times New Roman, 12 pt, Font color: extends beyond possible errors of measurements. This suggests that the mixing is in these regions 769 Formatted 770 is dominated by the extreme inhomogeneous type. Figure 15 shows the same type of diagrams as in Fig. 14, which were measured during 45 Formatted: Indent: First line: 0.75 cm, Don't adjust space 771 between Latin and Asian text, Don't adjust space between consecutive traverses through an ensemble of deep convective cells. The sampling altitude varied Asian text and numbers, Tab stops: Not at 13.65 cm 772 Formatted: Font: Times New Roman, 12 pt, Font color: 773 in the range 3000m $\prec H \leq H \leq 4500$ m, temperature  $-\frac{110}{100} \leq T \leq 00$ , relative humidity in **Formatted** the vicinity of clouds 15%< RH< 65,%. The cloud measurements were extended over a period of 774 775 2h 13m2 h 13 m, which is suggestive that the convective cells were sampled at different stages of

their lifetime. At the sampling level the concentration of raindrops varied from zero to few per

liter, and their diameter did not exceed 2 mm. 2mm.

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and  $\frac{LWC(\beta)}{LWC(\beta)}$  (Fig. 15a,  $b_7$  and d) is limited by the sector, which originates from the zero point, as in Fig. 12a. Analysis of the measurements showed that scatter diagrams LWC(N),  $\beta(N)$  $-LWC(\beta)$  and  $\overline{D}_3(N)$  calculated for the data points LWC(N),  $\beta(N)$ ,  $LWC(\beta)$  in each individual cloud <u>traverse</u> are <del>very similar to those in Fig.14. It appeared that the well aligned along the lines witl</del> different slopes of LWC(N),  $\beta(N)$ ,  $LWC(\beta)$  vary from cloud to cloud. So, after (e.g. Fig. 14). After averaging over the ensemble of clouds, the area of the scattered points willturned out to be located inside a sector limited by the lines with smallest and largest slopes. As follows from Eq. 3 the slopes of q(N) -  $\beta(N)$  and  $q(\beta)$  are governed by the ratios  $q_0/N_0$  -  $\beta_0/N_0$ and  $q_0/\beta_0$ , respectively, where  $N_0$ ,  $\beta_0$  and  $q_0$  correspond to undiluted adiabatic values. As discussed in section 3.7 changes in  $oldsymbol{eta}_0$  and  $oldsymbol{q}_0$  can be explained by (a) different altitude of measurements abo the cloud base, or (b) different cloud base temperatures in different clouds measured during this day. The changes in  $N_0$  can be caused by (a) different vertical velocities  $u_z$  at the cloud base in different cloud, or (b) spatial and temporal changes of the aerosol load during measurements. lead to different concentration of activated CCN. The combination of the above different slopes  $q_0/N_0$  -  $\beta_0/N_0$  - and  $q_0/\beta_0$  , which may be the  $\frac{LWC(N)}{B(N)}$  and  $\frac{LWC(B)}{B(N)}$  in Fig.15. A distinguishing characteristic of homogeneous mixing is that during evaporation the concentration tends to approach a non-zero value (Figs.4-6), whereas during inhomogeneous mixing it approaches zer The above analysis of the type of dependences LWC(N),  $\beta(N)$ ,  $LWC(\beta)$  and  $\overline{D}_3(N)$  in Fig.14 a 15 clearly shows  $N \rightarrow 0$ , when  $q \rightarrow 0$ . This is another argument in favor that the mixing was dominated

What is interesting that the scattering of the measurements  $\frac{LWC(N)}{\beta(N)}$ ,  $\frac{\beta(N)}{\beta(N)}$ 

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by inhomogeneous type.

3.2. Limitations of identifying type of mixing from in-situ measurements

Comparisons of the scatterdiagrams LWC(N),  $\beta(N)$  and  $LWC(\beta)$  in Figs.14 and 15 with the conceptual diagrams in Fig.12 unambiguously suggest that interaction between cloud and environment in the studied clouds was dominated by inhomogeneous mixing. It should be emphasized that analysis of a stand alone mixing diagram  $N - D_{\nu}$  would not allow unambiguously draw such conclusion.

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808 6. Discussion

One of the assumptions in most past studies is that for a sequence of the cloud samples collected along the flight path, the adiabatic values of  $N_1$ ,  $q_1$ ,  $\beta_1$ ,  $D_1$  and environmental parameters  $e_2$  and  $T_2$  remain the same. In fact these parameters may vary both within the same cloud or sequence of samples clouds, and the amplitude of their variations depends on microphysical and thermodynamical properties inside and outside the cloud environment. This variation will result in an ensemble of relationships  $M_n = F_{nk}(M_k)$ , and enhance scattering of the data points. In such cases identification of the type of mixing based on the  $N - D_v$  diagram may result in confusion between homogeneous and inhomogeneous mixing. As demonstrated in Sect. 5, consideration of N - q and  $N - \beta$  diagrams may provide a better identification type of mixing.

Strictly speaking the identification of type of mixing from particle probe measurements as it was performed in Sect. 5 is incomplete. It allows establishing correlation between microphysical moments and makes a formal conclusion about the mixing type, however it does not allow judgement about stage of mixing (i.e. whether mixing is complete by reaching equilibrium). In most previous studies, including this one, identification of type of mixing was based on the assumption that the sampled cloud volume is in equilibrium state (RH = 1), and that it reached the final stage of mixing (Fig.1 a2, a3, b3). It is possible that at the moment of measurement the process of mixing is not complete and the droplet free filaments remained undersaturated (Fig.1 a1, b1, b2). In this case the relationship between different moments may be well described as  $M_n = \alpha_{nk} M_{nk}$  and the mixing be confused with inhomogeneous mixing.

In order to identify stage of mixing, high frequency collocated measurements of temperature and humidity are required. Unfortunately current technology does not allow such measurements yet.

Identification of type of mixing from in-situ observations is based on examination of relationships between moments of the size distributions measured along the flight path. The basic assumption underlying this analysis is that the cloud droplets in each averaging interval originate at the same altitude. Otherwise, the effects of mixing on size distributions will be superimposed with environment is not affected by other non-adiabatic processes affecting size distributions, which may affect the relationships between the microphysical parameters. Precipitating particles along with the presence of ice may also.

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Thus, collision-coalescence, riming or Wegener-Bergeron-Findeisen processes may change the droplet number concentration and liquid water content, and therefore will, affect the relationship between the moments. Altogether it limits Activation of interstitial CCN will result in breaking correlation between the moments due to formation of large concentration of droplets. Broad size distributions may also hinder identification of type of clouds suitable for examining homogeneous and inhomogeneous mixing. The due to partial evaporation of small droplets (Pinsky et al. 2016a)

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It is anticipated that most suitable candidates for that matter to study mixing-entrainment process are non-precipitating convective clouds and stratocumulus clouds. However, as it was seen from above, high correlation between N,  $\beta$  and q-may exist even in presence of small amounts of precipitation size particles with relatively narrow droplet size distributions.

Another limitation is based on the assumption that the sampled cloud parcel is in equilibrium state and that it reached the final stage of mixing. It is possible that during mixing the cloud mixed with the entrained air, but at the moment of measurement it remained undersaturated and the droplets did not complete changing their sizes. In this case the relationship between different moments may be well described as  $M_n = \alpha_{nm} M_m$  and the mixing be confused with inhomogeneous mixing. In order to identify cases like this, accurate high frequency measurements of relative humidity are required. Unfortunately such measurements were not available during the in situ observations described above.

Another limiting factor is that the above consideration did not account for the effect of the vertical velocity. The vertical changes of the relative humidity in adiabatic parcels are described by the equation (Korolev and Field, 2008)

$$\frac{1}{S} \frac{dS}{dz} = a \tag{24}$$

After linearization of its solution Eq.(24) yields the distance between the level with  $S_0$  and saturation level (S = 1):

$$\Delta z = \frac{1 - S_0}{a} \tag{25}$$

According to Eq.25 gives the changing humidity in a vertically ascending parcel. Thus in droplet free entrained air relative humidity changes increases approximately 10% for  $\Delta z \approx \Delta z = 200 \text{m}$  at T = 0 °C. After reaching saturation the mixing turns into a degenerate case, which will appear as extreme inhomogeneous mixing. Joint effects of evaporating droplets and an increase in  $\frac{1}{2}$   $\frac{1}{2}$ 

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during the vertical ascent may facilitate reaching saturation state. This case may <u>also be specifically</u> relevant to the convective cloud <u>measurements</u> described <u>above in Sect.5.2</u>.

5. Characteristic time scales

The relative roles of mixing and evaporation is characterized by the Damkölher number (Dimotakis, 2005)

$$Da = \frac{\tau_{mix}}{\tau_{react}}$$
 (26)

where  $\tau_{mix}$  is the characteristic time of turbulent mixing;  $\tau_{react}$  is the characteristic time of interaction of droplets with the environment through a molecular diffusion. The two extremes with -Da >> 1 and -Da << 1, correspond to homogeneous and inhomogeneous mixing.

Over the years, the definition of the reaction time scale  $\tau_{react}$  was debated in literature on mixing. In some studies  $\tau_{react}$  was associated with the evaporating time of a droplet (e.g. Latham and Reed, 1977; Baker and Latham, 1979; Burnet and Brenguier, 2007; Andejchuk et al. 2009).

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$$\frac{\tau_{ev} - \frac{Fr_0^2}{S - 1}}{(27)}$$

Another group of works considered that  $\tau_{react}$  is equal to time of phase relaxation  $\tau_p$  (Eq.15) (e.g. Baker and Latham, 1982; Jeffery and Reisen, 2006; Kumar et. al. 2013). To compromise these two alternatives Jensen and Baker (1989), Jeffery (2007) considered both  $\tau_{ev}$  and  $\tau_p$ .

In the aforementioned studies in the estimations of  $\tau_{react}$  the values of S (Eq.27)  $\tau$  and N (Eq.15) were assumed to be constant. However, none of these parameters remain constant during mixing. Thus, Lehman et al. (2009) pointed out that during evaporation the population of droplets is interacting with the environment, and therefore S(t) does not remain constant. They proposed considering  $\tau_{react}$  as a "dominant" time of either evaporation of population of droplets ( $\tau_{ev}$ ) or time required for saturation of the cloud environment ( $\tau_{ev}$ ).

In reality the complete evaporation of droplets will be characterized by an ensemble of different  $\tau_{cr}$ , rather than by a single value. The following two conditions should be satisfied for complete evaporation of a droplet: (1) the volume associated the droplet should be  $\Delta V > v$ , where

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$$v = \frac{4\pi\rho_{w}r^{3}}{3\rho_{a}\delta q^{*}} \tag{28}$$

is the volume of the air, which will be saturated during a complete droplet evaporation; (2) the residence time of the droplet in the volume  $\Delta V$ -should be  $\Delta t > \tau_{ev}$ . The volume v-in fact is a minimum volume required for droplet evaporation, which means that the droplet evaporating time  $t_{ev}$ -will be overly long. If the local concentration of the droplets is sparse enough, so that  $\Delta V >> v$ , then the effect of the evaporating water on the humidity in  $\Delta V$ -can be neglected and it can be assumed that the local saturation ratio S-const during  $t_{ev} \approx \tau_{ev}$ .

**Table 1.** Evaporating time  $(t_{ev})$ , evaporating distance of a free falling droplet  $(\lambda_{ev})$ , size of the volume  $(\lambda_{e})$  saturated by a completely evaporated droplet with radius r and saturation ratio S. T =0C, P =687mb.

<del>r</del>	<del>2µm</del>			<del>2µт</del> 5 <del>µт</del>		<del>10μm</del>			
S	ŧ <sub>ev</sub> <del>(s)</del>	ત્રે <sub>લ્પ</sub> <del>(mm)</del>	ત્રે⊹ <del>(mm)</del>	ŧ <sub>ev</sub> ( <del>s)</del>	र्ने <sub>टर</sub> <del>(mm)</del>	<del>A</del> ⊕ <del>(mm)</del>	ŧ <sub>e</sub> , <del>(s)</del>	ने <sub>ev</sub> <del>(mm)</del>	<del>A</del> ⊕ <del>(mm)</del>
0.9	0.32	0.10	0.51	1.8	3.8	1.3	<del>6.9</del>	<del>59.4</del>	<del>2.6</del>
0.7	0.11	0.03	0.35	0.60	1.3	0.87	2.3	<del>19.8</del>	1.7
<del>0.5</del>	0.06	0.02	0.29	0.36	0.76	0.72	1.4	<del>11.9</del>	1.4
0.2	0.04	0.01	0.24	0.23	0.48	0.59	<del>0.9</del>	7.4	1.2

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 905 7. Table 1 shows the calculated evaporating time (t<sub>ev</sub>) and evaporating distance of the free falling

droplet  $(\lambda_{ev})$ , size of the volume  $(\lambda_{tr})$  saturated by a completely evaporated droplet with radius r at saturation ratio S. The size of the volume  $\lambda_{\Delta V}$  required for evaporation of a single droplet should exceed its evaporating distance and saturating droplet volume, i.e.

$$\lambda_{\Delta V} > \lambda_{ev}$$
 and  $\lambda_{\Delta V} >> \lambda_{v}$ . (29)

Table 1 demonstrates that, depending on-S- and-r, the value of  $\lambda_{\Delta V}$ -may vary from  $10^2$ cm to few cm. It appears that high humidity-S- and large-r- are strong factors limiting complete droplet evaporation, and therefore, declining inhomogeneous mixing.

Figure 16 shows  $\tau_{ev}$  (black solid line) calculated for  $S=S_{m0}$  assuming that S-remains constand during its evaporation (i.e. single droplet evaporation). This case corresponds to homogeneous mixing i.e. when turbulent eddies mixed the entrained environment prior droplets start evaporating. As seen

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 $au_{ev}$  is a strong function of  $\mu$  . The smallest value of the evaporating time  $au_{ev0}$  corresponds to unmixed entrained air  $S = S_2$  at  $\mu = 0$ . Due to the fact that during mixing the saturation ratio varies  $S_2 < S < 1$ , the evaporating time will change in the range  $\tau_{ev0} < \tau_{ev} < \infty$ . In other words  $\tau_{ev0}$  presents the shortest evaporation time of a single droplet corresponding to  $S_2$  of the entrained air.  $\mu \leq \mu_{cr}$  regardless type of mixing. The region of complete evaporation calculated for a population df droplets is shown on the left side of Fig.16 in section  $\mu \leq \mu_{cr}$ . During mixing the droplet sizes-r and their concentration-N-are changing. Therefore, as follows from Eq.16  $au_p$  will be changing as well. Figure 16 also shows the dependence of time of phase relaxation versus  $\mu$  -calculated for the final stage of inhomogeneous  $au_p^{(i)}$  (dashed blue) and homogeneous  $au_p^{(h)}$  (solid blue) mixing, and the initial stage homogeneous mixing before droplet evaporation  $au_{pm}^{(h)}$  (blue thin line). As  $\frac{\text{from Fig.16-}\tau_{p0} \leq \tau_{pm}^{(h)} \leq \tau_{p}^{(h)} \leq \tau_{p}^{(i)}}{\tau_{p}} \leq \tau_{p}^{(i)}.$ calculated as a time when saturation ratio changed from  $S_{m0}$  to 0.999. The calculations were  $\wp$ with the help of a system of differential equations. As seen from Fig.16  $au_{eq}^{(h)} > au_{pm}^{(h)} > au_{pm}^{(h)}$  . The time of characteristic time of approaching supersaturation to its quasi-state value  $S_{qs}$  , and in general cash  $au_p^{(h)} 
eq au_{eq}^{(h)}$ . For the case in Fig.16 the values of  $au_{eq}^{(h)}$  and  $au_{pm}^{(h)}$  may differ by more than an order of magnitude, and for this specific case their ratio varies in the range  $0 < \tau_{eq}^{(h)} / \tau_{pm}^{(h)} < 15$ . Note, that it is possible that  $\tau_{eq}^{(h)} \leq \tau_{pm}^{(h)}$ , when  $S \to 1$  the equilibrating time  $\tau_{eq}^{(h)} \to 0$ . This presents a degenerate case when The equilibrating time for inhomogeneous mixing  $au_{eq}^{(i)}$  was calculated for extreme inhomogeneous mixing: at the moment of time  $t = \tau_{eq}^{(i)}$  (a) the entrained volume is saturated  $S_2(\tau_{eq}^{(i)}) \approx 0$  and (b) the population of droplets with concentration  $n_{ev}$  undergoes complete evaporation, i.e.  $r( au_{eq}^{(i)})$  pprox0. As seen from Fig.16,  $au_{eq}^{(i)}$  remains constant at  $\mu > \mu_{cr}$ . Basically in the frame of the inhomogeneous mixing

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concept  $au_{eq}^{(i)}$  coincides with the time of evaporation of the population of droplets, which were transported

into the entrained air. Obviously such a consideration is overly idealized and unlikely to happen in nature. This is suggestive that the consideration of  $au^{(i)}_{eq}$  has significant limitations.

Comparisons of different characteristic times in Fig.16 allow few conclusions about mixing. Thus, for  $\mu_A < \mu \le 1$ - $\tau_{eq}^{(h)} < \tau_{ev0}$ , i.e. the saturation of the environment occurs faster than the fastest possible evaporation time of a single droplet. This suggests that when  $\mu_A < \mu \le 1$  the feasibility of inhomogeneous mixing is low. For  $\mu_{cr} < \mu < \mu_B$   $\tau_{ev} < \tau_{eq}^{(h)}$ , i.e. saturation of the entire volume takes longer due to evaporation of the droplet population than the complete evaporation of a single droplet. Compete evaporation of some droplets may occur, if the conditions in Eq.29 are satisfied. This suggests that when  $\mu_{cr} < \mu < \mu_B$  the feasibility of inhomogeneous mixing is increasing. Since during mixing elementary volumes are mixing at any  $\mu$  homogeneous and inhomogeneous mixing may occur simultaneously in different parts of the mixing volume.

In the conclusion of this section it could be stated that  $\tau_{react}$  can be characterized by different characteristic times (i.e.  $\tau_{ev}$ ,  $\tau_{pm}$ ,  $\tau_p^{(h)}$ ,  $\tau_p^{(i)}$ ,  $\tau_{eq}^{(h)}$ ), each of which is a function of  $\mu$ . The difference between these times can reach two orders of magnitude. This brings up a warning on a proper selection of  $\tau_{react}$  in Da, and requirements for more studies to address this question.

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### 6. Conclusions

This study analyses analyzes dependences of different moments of f(r)f(D) in the frame of formalism of homogeneous and extremely inhomogeneous mixing. The analysis was performed for the final stage of mixing based on the mass balance of vapor and liquid water and assumption of adiabatic process of mixing. and energy conservation consideration. The following results were obtained in the frame of this study:

1) It is shown that for the case of extreme inhomogeneous mixing all moments of f(r) are linearly related as  $M_n = \alpha_{nm} M_m$  where coefficients  $\alpha_{nm}$  are determined by adiabatic valued of  $M_n$  and  $M_m$ . This relation remains valid for progressive mixing.

Analytical 1. Simple analytical relationships  $M_n = F_{nm}(M_m)$  between the moments  $M_n$  and  $M_m$  main microphysical moments were found obtained for the case of final state homogeneous mixing.

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2) 2. It was demonstrated shown that the functional relationships between  $M_n$  and  $M_m$  the moments exist only for the first stage of homogeneous mixing. The following, when equilibrium is reached. Subsequent progressive homogeneous mixing breaks the functional relationship between the moments. It was shown that the scattering of the moments  $M_n$  and  $M_m$  are limited by the lines for inhomogeneous mixing  $M_n = \alpha_{nm} M_m$  and homogeneous mixing  $M_n = F_{nm} (M_m)$ .

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3. It was demonstrated that consideration of scattering N-LWC,  $N-\beta$  diagrams facilitates identification of type of mixing from in-situ measurements. For extreme inhomogeneous mixing the scattering of the data points N-LWC,  $N-\beta$  will be limited by a sector originating at zero point (Fig.12a). However, for homogeneous mixing the scattering data points will be limited by a sector originating at  $(N_1, LWC_1)$  and  $(N_1, \beta_1)$  (Fig.12b). Utilizing a stand-alone conventional  $N-D_v$  mixing diagram may not provide unambiguous answer about type of mixing.

3) 4. The developed approach was applied to a set of in-situ measurements collected in convective clouds. The analysis of the dependences between N,  $\beta$ , LWC and  $\overline{D}_3$  (i.e.  $M_0$ ,  $M_2$ ,  $M_3N$ ,  $\beta$ , LWC and  $M_3/M_0$ ) obtained from in situ measurements of convective clouds  $D_v$  suggests that the interaction between entrained and cloudy environments is for the studied clouds was dominated by inhomogeneous mixing. Homogeneous mixing may to become active, when the mixing fraction  $\mu$  is close to critical  $\mu_{cr}$ .

4) Analysis of different characteristic times related to evaporation and equilibrating of the mixing environment showed their strong dependence on the mixing fraction  $\mu$ . The difference The present study considers relationships between different times may reach two orders of magnitude, which makes the selection of  $\tau_{react}$  ambiguous. This raises a requirement of further studies on characterization moments of  $\tau_{react}$  in Damköhler number. Comparisons of different characteristic times suggest that within f(D) for the same mixing environment depending on  $\mu$  some volumes may be dominated by homogeneous mixing whereas others by inhomogeneous mixing.

Due to the various limitations and idealizations of the mixing process explored in the above discussion, this study acknowledges that the current description of the mixing process is far from complete. Since the process of mixing covers a wide range of time and spatial scales a comprehensive simulating efforts are required to achieve a better understanding of the process of mixing of colloidal

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Formatted: Indent: First line: 0.75 cm, Don't adjust space between Latin and Asian text, Don't adjust space between Asian text and numbers systems. final stage of mixing. The following two works Pinsky et al. (20152, 2016a, b) in this series provide a more detailed analysis of evolution time dependences of droplet size distribution distributions and its moments during homogeneous and inhomogeneous mixing.

#### **Acknowledgements**

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## Appendix A

**List of Symbols** 

Symbol	Description	Units	1
<del>b</del>	$\frac{\frac{1}{q_{_{_{_{_{_{_{_{_{_{_{_{_{_{_{_{_{_{_$	m²-s <sup>-1</sup>	
$\frac{c_{pa}}{}$	specific heat capacity of dry air at constant pressure	<del>J kg⁻¹</del> k	-1
$c_{pv}$	specific heat capacity of water vapor at constant pressure	<del>J kg⁻¹</del> k	4
Ð	<del>droplet diameter</del>	m	
<del>-Da</del>	<del>Damköhler number</del>	-	
$D_{\!$	coefficient of water vapor diffusion in the air	m²-s-1	
e	water vapor pressure	N-m <sup>-2</sup>	
$e_1$	initial water vapor pressure in the cloud parcel	N-m <sup>-2</sup>	
$e_{2}$	initial water vapor pressure in the entrained sub-saturated parcel	N-m <sup>-2</sup>	
$E_{s}$	saturation vapor pressure above flat surface of water	N m <sup>-2</sup>	
<del>F</del>	$\frac{\left(\frac{\rho_{_{W}}L_{_{W}}^{2}}{k_{_{a}}R_{_{v}}T^{2}} + \frac{\rho_{_{W}}R_{_{v}}T}{E_{_{w}}(T)D_{_{v}}}\right)^{-1}}{\text{coefficient in the equation droplet growth}}$	N m <sup>-2</sup>	
f(r)	size distribution of cloud droplets normalized on unity	m <sup>-1</sup>	
$\frac{k_a}{}$	coefficient of air heat conductivity	<del>J m<sup>-1</sup>s</del>	<sup>1</sup> ₭ <sup>-1</sup>
$L_{w}$	latent heat for liquid water	<del>J kg⁻¹</del>	
<del>LWC</del>	liquid water content	<del>kg m</del>	
$\frac{1}{t_p}$	<del>characteristic spatial phase scale</del>	m	
$\frac{m_1}{m_1}$	mass of the cloud parcel (1kg)	<del>kg</del>	
$m_2$	mass of the entrained air (1kg)	<del>kg</del>	
$M_{\overline{n}}$	$r$ th moment of the droplet size distribution $\int\limits_0^\infty f(r)r^n dr$ $\int\limits_0^\infty f(r)dr$	<del>m</del> ª	
<del>N</del>	concentration of droplets	m <sup>-3</sup>	

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$N_0$	concentration of droplets before mixing	m <sup>-3</sup>	
<del>-p-</del>	<del>pressure of moist air</del>	N-m <sup>-2</sup>	
$\overline{P_a}$	pressure of dry air	N m <sup>-2</sup>	
<del>*</del>	<del>droplet radius</del>	m	
$\bar{\tau}$	mean droplet radius	m	
<del>"</del> -	mean square droplet radius	m	
$\bar{r}_3$	mean cubic droplet radius	m	
$R_a$	specific gas constant of moist air	<del>J kg⁻¹</del> ⊭	-1
$R_{\nu}$	specific gas constant of water vapor	<del>J kg ¹</del> k	-1
4	cloud liquid water mixing ratio (mass of liquid water per 1kg of dry air)	-	
$\overline{q_0}$	cloud liquid water mixing ratio before mixing	-	
$\overline{q_{v}}$	water vapor mixing ratio (mass of water vapor per 1kg of dry air)	-	
<del>S</del>	$e/E_{\scriptscriptstyle S}$ , relative humidity over water (saturation ratio)	-	
$S_m$	relative humidity after mixing is completed	-	
$S_{m0}$	relative humidity after instant mixing of cloudy and entrained air but before droplets	-	
	evaporation		
T	temperature	K	
$T_{1}$	temperature of the cloud parcel before mixing	K	
$\overline{T_2}$	temperature of the entrained sub-saturated parcel before mixing	K	
$T_{m0}$	temperature of the parcel after vapor mixing, but before droplet evaporation	K	
+	Time	S	
$\frac{u_z}{}$	vertical velocity	m-s <sup>-1</sup>	
B	extinction coefficient	m <sup>-1</sup>	
$\beta_0$	extinction coefficient before mixing	₽0	
<del>Sq</del>	amount of liquid water per 1kg of sub-saturated air required to saturate the air in the	-	
	mixed parcel before droplet evaporation Eq.5		
$\frac{\delta q^*}{}$	amount of liquid water per 1kg of dry air required to saturate the air in the entrained	-	
	cloud free volume Eq.8		
<del>C</del>	turbulent energy dissipating rate	m²-s³	

#	mixing fraction of cloud and dry parcels $0 \le \mu \le 1$	-
$\mu_{cr}$	critical mixing fraction, such that for $\mu \leq \mu_{cr}$ all droplets evaporate	-
$\rho_a$	density of the dry air	kg m
$\overline{oldsymbol{ ho}_w}$	density of liquid water	<del>kg m</del>
$\tau_{ev}$	time of complete evaporation	s
$\tau_{ev}$	time of complete evaporation at constant $S = S_{m0}$	S
$\tau_{ev0}$	time of complete evaporation at constant $S = S_2$	S
$ au_{eq}^{(h)}$	time of saturating environment during evaporation of a population of homogeneously	s
	mixed droplets	
$ au_{eq}^{(i)}$	time of saturating environment for a population of inhomogeneously mixed droplets	s
$\frac{\tau_p}{p}$	time of phase relaxation	s
$\tau_p^{(h)}$	time of phase relaxation after completing homogeneous mixing	s
$\tau_p^{(h)}$	time of phase relaxation after completing inhomogeneous mixing	S

#### Appendix B 1014

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: Liquid water deficit 1015

> The objective of this section is to find the amount of liquid water, which is required to evaporate be evaporated in order to saturate the parcel formed after mixing of a cloud volume with sub-saturate entrained air.

Assume that  $\frac{q_{v1}}{q_{v2}}q_{v1}q_{v2}$  are the mixing vapor ratios in the cloudy and entrained parcels, 1019 respectively, and  $\frac{T_1}{T_1}$ ,  $\frac{T_2}{T_2}$  are their respective initial temperatures. First, we find the saturation 1020 ratio  $\frac{S_{m0}}{S_{m0}}$  formed after instant mixing of the cloud and entrained before the cloud droplets start 1021 1022 evaporating.

The vapor mixing ratio  $\frac{q_{vm}}{q_{vm}}$  formed in the mixed volume will be 1023

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$$q_{vm} = \mu q_{v1} + (1 - \mu)q_{v2}$$
 (B1)

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$$q_{vm} = \mu q_{v1} + (1 - \mu) q_{v2}$$
 (A1)

The vapor pressure  $\frac{e_m}{e_m}e_m$  in the mixed volume can be derived from Eq.  $\frac{84(A1)}{e_m}$  by substituting 1026

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$$\frac{e}{q_v} = \frac{e}{p - e} \frac{R_a}{R_v} q_v = \frac{e}{p - e} \frac{R_a}{R_v}$$
, i.e.

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$$e_{m} = \frac{\mu + \frac{e_{2}(p - e_{1})}{p(e_{1} - e_{2})}}{\mu + \frac{(p - e_{1})}{(e_{1} - e_{2})}}$$
(B2)

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$$e_{m} = p \frac{\mu + \frac{e_{2}(p - e_{1})}{p(e_{1} - e_{2})}}{\mu + \frac{(p - e_{1})}{(e_{1} - e_{2})}}$$
(A2)

The temperature of the mixed volume  $T_{m0}$  can be found from the energy conservation law 1030

1031 
$$\mu(q_{v1}c_{pv} + c_{pa})(T_1 - T_{m0}) = (1 - \mu)(q_{v2}c_{pv} + c_{pa})(T_{m0} - T_2)$$
 (B3)

1032 
$$\mu(q_{v1}c_{pv} + c_{pa})(T_1 - T_{m0}) = (1 - \mu)(q_{v2}c_{pv} + c_{pa})(T_{m0} - T_2)$$
(A3)

1033 here  $\frac{c_{pv}, c_{pa}, c_{pv}, c_{pa}}{c_{pv}, c_{pa}}$  are the specific heat capacitance of water vapor and dry air at constant

pressure, respectively,  $T_1$ ,  $T_2$ ,  $T_1$ ,  $T_2$  are the initial temperatures in the first and second parcel 1034

1035 before mixing:  $T_{m0}$ : Substituting  $q_{v1}$ :  $q_{v2}q_{v1}$ .  $q_{v2}$  yields the temperature in the mixed volume Formatted: Font: Times New Roman, 12 pt, Font color:

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1036 
$$T_{m} = \frac{\mu T_{1} + a(1-\mu)T_{2}}{\mu + a(1-\mu)}$$
 (B4)

1037 
$$T_{m0} = \frac{\mu T_1 + \alpha (1 - \mu) T_2}{\mu + \alpha (1 - \mu)}$$
(A4)

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$$\frac{1 + \frac{c_{pv}R_{a}e_{2}}{c_{pa}R_{v}(p - e_{2})}}{1 + \frac{c_{pv}R_{a}e_{1}}{c_{pa}R_{v}(p - e_{1})}}$$
(B5)

1040 
$$\alpha = \frac{1 + \frac{c_{pv}R_a e_2}{c_{pa}R_v(p - e_2)}}{1 + \frac{c_{pv}R_a e_1}{c_{pa}R_v(p - e_1)}}$$
(A5)

With a good accuracy  $\alpha \cong 1$ - $\alpha \cong 1$ . The resulting saturation ratio relative humidity after mixing

the two volumes will be

$$\frac{S_{m0} - \frac{e_m}{E_S(T_{m0})}$$
 (B6)

1044 
$$RH_{m0} = \frac{e_{m0}}{e_{S}(T_{m0})}$$
 (A6)

where  $\frac{E_s(T_{m0})e_S(T_{m0})}{e_S(T_{m0})}$  is the saturated vapor pressure at temperature  $\frac{T_{m0}}{t_{m0}}$ .

If the initial saturation ratio is  $S_{m0}$ <1, then the cloud droplets start evaporating. The process of

evaporation is accompanied by changing humidity and temperature due to latent heat of vaporization. This process is described by the Eq. (C2) in Korolev and Mazin (2003). Assuming

the process to be isobaric (i.e. vertical velocity  $u_z = u_z = 0$ ) and absence of ice  $(dq_i = (dq_i) = 0)$ ,

1050 Eq. (C2) (Korolev and Mazin, 2003) yields

$$\frac{dS}{S} = \left(\frac{1}{S} \frac{pR_v}{R_v} + \frac{L^2}{c_p R_v T^2}\right) dq \tag{B7}$$

1052 
$$\frac{dS}{-S+1} = \left(\frac{1}{S+1} \frac{pR_v}{e_S R_a} + \frac{L^2}{c_{pa} R_v T^2}\right) dq$$
 (A7)

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Integrating Eq.81 (A7) from initial humidity  $S_{m0}$  (Eq.6) $S_{m0}$  to saturation state, when S=1S=1054 0, and taking into account that RH=S+1, gives an expression  $\begin{pmatrix} & & & & & & & & \\ & & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & &$ 

$$Sq = \frac{L^2}{c_p R_v T_2^2} \ln \left( \frac{1 + \frac{E_S(T_{m0}) R_a L^2}{p c_p R_v^2 T_{m0}^2}}{1 + S_{m0} \frac{E_S(T_{m0}) R_a L^2}{p c_p R_v^2 T_{m0}^2}} \right)$$
(B8)

It should be noted that in Eq.B8-  $\delta q_m = -b \ln \left( \frac{1 + aRH_{m0}}{1 + a} \right)$ 

1057 <u>(A8</u>

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the changes of temperature during droplet evaporation and related changes mixing ratio of liquid water required to evaporate in order to saturate 1kg of the cloud volume were neglected formed after mixing

1060 with the entrained air, but before droplet start evaporating. Here  $a = \frac{E_S R_a L^2}{p c_n R_a^2 T_m^2}$ ,  $b = \frac{c_p R_v T_m^2}{L^2}$ .

1061  $\frac{\text{Since}}{1+A} \left| \frac{A(RH_{m0}-1)}{1+A} \right| < 1, \text{ Eq.}(A8) \text{ can be simplified as}$ 

1062 
$$\delta q_m = ab \frac{1 - RH_{m0}}{1 + a} = -\frac{S_{m0}}{A_2}$$
 (A9)

where  $A_2 = \frac{ab}{1+a^2}$ . The analysis of the behavior of Eq.B8Eqs. (A8)-(9) shows that for wide range of temperatures  $-30^{\circ}\text{C} < T < 30^{\circ}\text{C}$  the cases when  $30^{\circ}\text{C} < T < 30^{\circ}\text{C}$ , both equations hold with high accuracy as long as the temperatures of the sub-saturated and cloud parcels  $|T_1 - T_2| < |T_1 - T_2| < |T_1 - T_2| < |T_2 - T_2| < |T_2$ 

Figure B1A1 shows comparisons of modeled  $\delta q \delta q_m$  and that calculated from Eq.B8Eqs. (A8) and (A9) for three different temperatures. The model consisted in solvingsolved a system of differential equation with decremental evaporation of liquid water until saturation is reached.

As seen from Fig. 81A1 the agreement between modeled  $\partial_T \delta q_m$  and that calculated from Eq.88 (A8)-(A9) is quite good and does not exceed few percent at  $S_0$ =RH<sub>m0</sub> = \_0.5. This discrepancy results from disregarding the effect of changing T on  $E_s$  during evaporation, i.e. in Eq. 88  $E_s$  and  $E_s$  are assumed to be constant. The changes of the air temperature due to the latent heat of vaporization increase with the increase of the amount evaporated water, when

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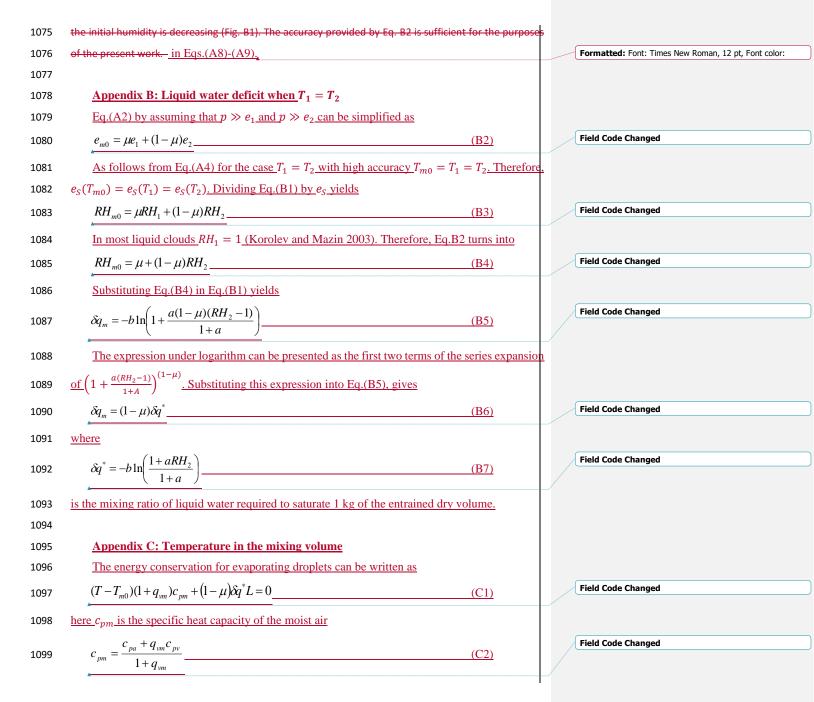
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mixing
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            T = T_{m0} - \frac{(1-\mu)\delta q^* L}{c_{pa}}
1102
                                                                                           (C3)
1103
            For the case when T_1 \neq T_2 Eq. (C3) should be replaced by
            T = T_{m0} - \frac{\delta q_m L}{c_{pa}} 
1104
                                                                                           (C4)
            Eqs. (C3) and (C4) are valid for the mixing fraction \mu > \mu_{cr}. For \mu \le \mu_{cr} all entrained liquid
1105
        water \mu q_0 evaporates, and the final temperature will be
1106
            T = T_{m0} - \frac{\mu q_0 L}{c_{pa}}
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Since  $q_{vm} \ll 1$  and,  $c_{pa} \cong c_{pm}$  Eq.(C1) may be simplified, so that the final temperature after

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2 Table 1. Evaporating time  $(t_{ev})$ , evaporating distance of a free falling droplet  $(\lambda_{ev})$ , size of the volume

3 ( $\lambda_{v}$ ) saturated by a completely evaporated droplet with radius r and saturation ratio S. T =0C, P

### 4 <del>=687mb.</del>

1

F	<del>2μm</del>			<del>5μm</del>			<del>10μm</del>		
\$	ŧ <sub>ev</sub>	Aev	Au	ŧ <sub>ev</sub>	Aev	<del>A</del> ⊎	ŧ <sub>ev</sub>	Aev	<del>A</del> v
	<del>(s)</del>	<del>(mm)</del>	<del>(mm)</del>	<del>(s)</del>	<del>(mm)</del>	<del>(mm)</del>	<del>(s)</del>	<del>(mm)</del>	<del>(mm)</del>
0.9	0.32	0.10	0.51	1.8	3.8	1.3	<del>6.9</del>	59.4	<del>2.6</del>
0.7	0.11	0.03	0.35	0.60	1.3	0.87	2.3	<del>19.8</del>	<del>1.7</del>
0.5	0.06	0.02	0.29	0.36	0.76	0.72	1.4	11.9	1.4
0.2	0.04	0.01	0.24	0.23	0.48	0.59	0.9	7.4	1.2

## List of Symbols

Symbol	Description	Unit
$A_2$	$\frac{pR_{v}}{e_{s}R_{a}} + \frac{L^{2}}{c_{pa}R_{v}T^{2}}$	Ξ
а	$rac{e_{_s}R_aL^2}{pc_{_{pa}}R_{_v}^2T^2}$	Ξ

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b	$\frac{c_{pa}R_{v}T^{2}}{L_{c}^{2}}$	=
	· <del></del>	11 117-1
$c_{pa}$	specific heat capacity of dry air at constant pressure	J kg <sup>-1</sup> K <sup>-1</sup>
$c_{pv}$	specific heat capacity of water vapor at constant pressure	J kg <sup>-1</sup> K <sup>-1</sup>
$\overline{D}$	mean droplet diameter	<u>m</u>
$D_2$	mean square droplet diameter	<u>m</u>
$D_v$	mean volume droplet diameter	<u>m</u>
e	water vapor pressure	<u>N m<sup>-2</sup></u>
$e_1$	initial water vapor pressure in the cloud parcel	<u>N m<sup>-2</sup></u>
$e_2$	initial water vapor pressure in the entrained sub-saturated parcel	<u>N m<sup>-2</sup></u>
$e_s$	saturation vapor pressure above flat surface of water	<u>N m<sup>-2</sup></u>
f(D)	size distribution of cloud droplets normalized on unity	<u>m<sup>-1</sup></u>
<u>L</u>	latent heat for liquid water	J kg <sup>-1</sup>
$M_n$	n-th moment of the droplet size distribution $\int_{0}^{\infty} f(r)r^{n}dr$ $\int_{0}^{\infty} f(r)dr$	m <sup>n</sup>
N	concentration of droplets	<u>m<sup>-3</sup></u>
$N_1$	concentration of droplets before mixing	<u>m<sup>-3</sup></u>
p	pressure of moist air	<u>N m<sup>-2</sup></u>
$R_a$	specific gas constant of moist air	J kg <sup>-1</sup> K <sup>-1</sup>
$R_v$	specific gas constant of water vapor	J kg <sup>-1</sup> K <sup>-1</sup>
RH	$e/E_s$ , relative humidity over water (saturation ratio)	=
$RH_1$	initial relative humidity in the cloud volume ( $RH_1=1$ )	=
$RH_2$	relative humidity in the entrained sub-saturated parcel	=
$RH_{m0}$	relative humidity after instant mixing of cloudy and entrained air but before	=
	droplets evaporation	
q	cloud liquid water mixing ratio (mass of liquid water per 1kg of dry air)	=
$q_1$	cloud liquid water mixing ratio before mixing	=
$q_v$	water vapor mixing ratio (mass of water vapor per 1kg of dry air)	=

S	$e/e_s - 1$ , supersaturation	<u>-</u>
$S_2$	supersaturation of the dry out-of-cloud air	Ξ
$S_{m0}$	supersaturation after instant mixing of cloudy and entrained air, but before	Ξ
	droplets start evaporating	
T	temperature	<u>K</u>
$T_1$	temperature of the cloud parcel before mixing	<u>K</u>
$T_2$	temperature of the entrained sub-saturated parcel before mixing	<u>K</u>
$T_{m0}$	temperature of the parcel after vapor mixing, but before droplet evaporation	<u>K</u>
β	extinction coefficient	<u>m<sup>-1</sup></u>
$oldsymbol{eta_1}$	extinction coefficient before mixing	<u>m<sup>-1</sup></u>
$\delta q_m$	mixing ratio of liquid water required to saturate 1kg of the cloud volume after	Ξ
	instant mixing, but before droplet evaporation.	
$\delta q^*$	mixing ratio of liquid water required to saturate 1kg of the dry out-of-cloud air	Ξ
μ	cloud fraction of mixing air, $0 \le \mu \le 1$	=
$\mu_{cr}$	critical cloud fraction, such that for $\mu \leq \mu_{cr}$ all droplets evaporate	=
$ ho_a$	density of the dry air	kg m <sup>-3</sup>
$ ho_w$	density of liquid water	kg m <sup>-3</sup>
ξ	$\underline{\text{coefficient}}\ 0 \leq \xi \leq \underline{1}\ \underline{\text{characterizing proximity of homogeneous mixing to}}$	Ξ
	inhomogeneous ( when $\xi \to 0$ ).	

# 8 Figure Captions

11 12

Figure 1. Classical conceptual diagram of (a) inhomogeneous and (b) homogeneous mixing.

initial state; 2 mixing state; 3 final state.

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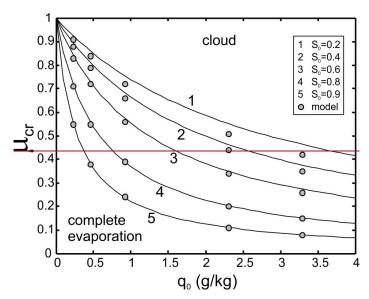


Figure 2. Dependence of initial ratio of mixing  $q_0$  versus critical ratio of mixing fraction  $\mu_{cr}$  versus mixing  $\mu_{cr}$  ratio  $q_0$  calculated from Eq.23...(7). Circles indicate modeled points. The calculations were performed for T=0C and HH=3000m.

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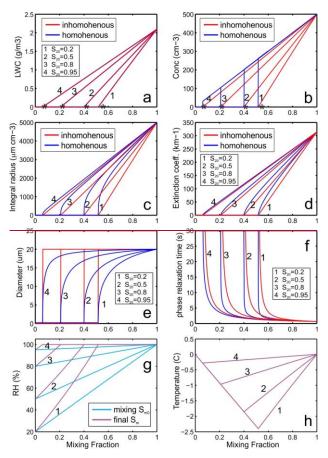


Figure 3. Dependence of  $\xi$  versus  $\mu$ . Numbers are the dimensionless ratios  $\delta q^*/q_1$ . Critical mixing ratios  $\mu_{cr}$  are indicated by stars. Grey area indicates area where the moments of homogeneous and extreme inhomogeneous mixing may not be segregated from in-situ measurements. Dashed line was calculated for the cloud in Figs.13-14.

Figure 4. Simulation of (a) liquid water mixing ratio, (b) droplet number concentration—and (b) liquid water mixing ratio, (, (c) integral droplet radius diameter, (d) extinction coefficient, (e) mean cube volume diameter, (f) time of phase relaxation, (g) relative humidity in the mixed volume before droplet evaporation  $S_{m0}RH_{m0}$  and final  $S_{m}$  at the equilibrium state  $RH_{m2}$  (h) final temperature  $T_mT_{m0}$  versus ratio of mixing  $\mu$  formed after homogeneous

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and extreme inhomogeneous mixing between dry and cloudy parcel with monodispers
droplets. Black stars on (a) and (b) indicate critical ratio of mixing $\mu_{cr}$ fraction $\mu_{cr}$
calculated from Eq.23(7). The calculations were performed for $RH_{20}$ =20%, 50%, 80%
$RH_2 = 0.2, 0.5, 0.8, 0.95\%; r_0 = 10; D_1 = 20 \mu m, N_0 = N_1 = 500 \text{cm}^{-3}; T_{10} = T_{20} = T_1 = T_2 = 00$
# <i>H</i> =1000m.

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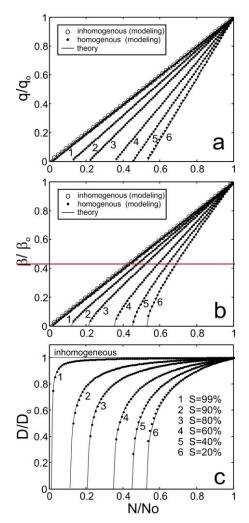


Figure 45. Dependence of normalized <u>liquid water</u> mixing ratio  $\frac{q}{q_0}$ ,  $q/q_1$  (a,d,g), extinction coefficient  $\frac{R}{R}$  (b,e,h) and mean volume diameter  $\frac{r_3}{r_0}$  (c,f,j) versus normalized number concentration  $\frac{R}{R}$  (b,e,h) and for various humidity of the entrained air (a,b,c), for various liquid water mixing ratios (d,e,f) and for various temperatures (g,h,j). The calculations were performed for different relative humidity of the entrained air  $\frac{R}{R}$  (1) 99%; (2) 90%; (3) 80%; (4) 60%; (5) 40%; (6) 20%.

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7 The the initial conditions used for the calculations were:  $H_0$ : H=1000m,  $T_{\pm}(0)=T_{\pm}(0)=0$ C;  $r_0=10D_1=20\mu m_7$ 

8  $N_0 = \frac{\text{for (a-c; g-j)}}{N_1} N_1 = 500 \text{cm}^{-3}$ .

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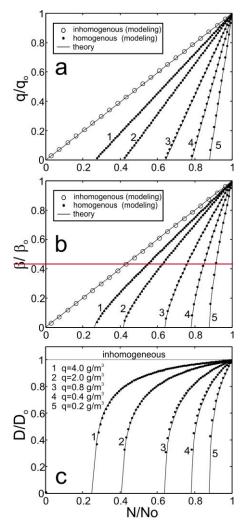


Figure 5. Effect of the water content:  $q_0$ -on mixing. Same as in Fig.4. The calculations were performed for different initial liquid water content: (1) 4.0g/kg; (2) 2.0g/kg; (3) 0.8g/kg; (4) 0.4g/kg; (5) 0.2g/kg. The initial conditions used for the calculations were:  $H_0$ =1000m, RH(0)=50%;  $T_4$ (0)= $T_2$ (0)= $T_2$ (0)= $T_3$ (0)= $T_4$ (0)= $T_$ 



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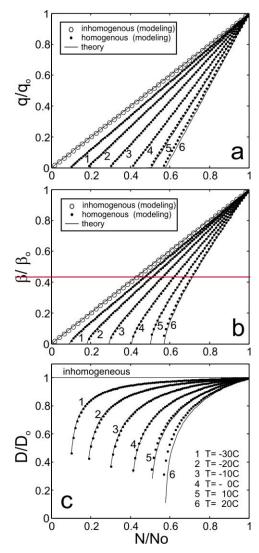


Figure 6. Effect of temperature on mixing. Same as in Fig.4. The calculations were performed; for different initial temperatures T: (1) –30C; (2) –20C; (3) –10C; (4) 0C; (5) 10C; (6) 20C. The initial conditions used for the calculations were:  $H_0$ =1000m, RH(0)=50%;  $r_0$ =10 $\mu$ m,  $N_0$ =500cm<sup>-3</sup>(a-f)  $T_1 = T_2 = 0$ C.

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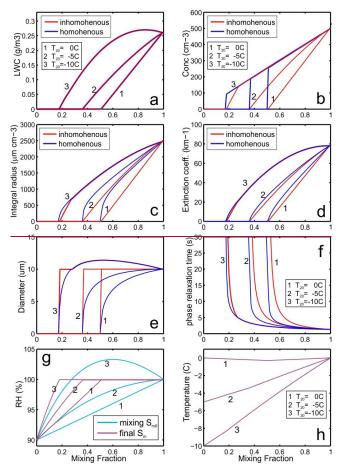


Figure 76. Simulation of (a) droplet number concentration and (b) liquid water mixing ratio, (c) integral droplet radiusdiameter, (d) extinction coefficient, (e) mean cubevolume diameter, (f) time of phase relaxation, (g) relative humidity in the mixed volume before droplet evaporation  $S_{m0}RH_{m0}$  and final  $S_{m7}$  at the equilibrium state  $RH_{m2}$  (h) final temperature  $T_mT_m$  versus ratio of mixing  $\mu$  formed after homogeneous and extreme inhomogeneous mixing between dry and cloudy parcel with monodisperse droplets. Black stars on (a) and (b) indicate critical ratio of mixing  $\mu_{cr}$  calculated from Eq.22. The calculations were

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9 performed for RH = 90%;  $r_0 = 5RH_2 = 0.9$ ;  $D_1 = 10 \mu \text{m}$ ,  $N_0 = N_1 = 500 \text{cm}^{-3}$ ;  $T_{10} = T_1 = 0 \text{C}$ ;  $T_{20} = 0.9$ ; T

 $-T_2 = -10$ C, -5C, 0C; HH = 1000m.

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0

1.2

0.8

0.4

0.2

0

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0.2

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1

Figure 7. Figure 8. Effect of temperature difference between cloud and entrained air on mixing.

Same as in Fig.4. The calculations were performed for different initial temperatures AT: T<sub>2</sub>:

(1) -10C; (2) -5C; (3) -0C.0C. Grey circles indicate extremely inhomogeneous mixing on line 1 at the AB interval. The rest cases on extremely inhomogeneous mixing are indicated

0.4 0 N/No

0.6

b

0.8

1 ΔT= -10C 2 ΔT= -5C 3 ΔT= 0C

0.8

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by open circles. The initial conditions used for the calculations were: H_0H=1000 \,\mathrm{m},

RH(0)=RH_2=90\%; F_0=5D_1=10 \,\mathrm{mm}, N_0=N_1=500 \,\mathrm{cm}^{-3}; T_{10}=T_1=0 \,\mathrm{C}.
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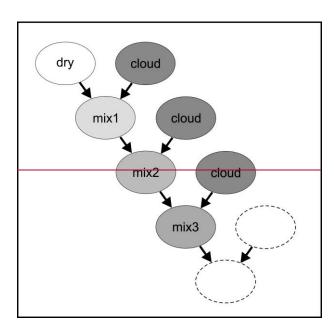


Figure 8. Conceptual diagram of cascade mixing of the out-of-cloud entrained parcel with the cloudy environment

Figure 9. Simulation of stochastic mixing corresponding to stages 1-4 as indicated in Fig.8. Solid red lines indicate the normalized dependences q,  $\beta$ ,  $D_{\nu}$  vs. N for the primary stage of homogeneous mixing. Dashed red lines indicate the same dependences for inhomogeneous mixing. The initial conditions used for the simulations were: H=1000m.  $T_1 = T_2 = 0$ C;  $RH_2=0.5$ ;  $D_1 = 10$  µm,  $N_1 = 500$  cm<sup>-3</sup>.

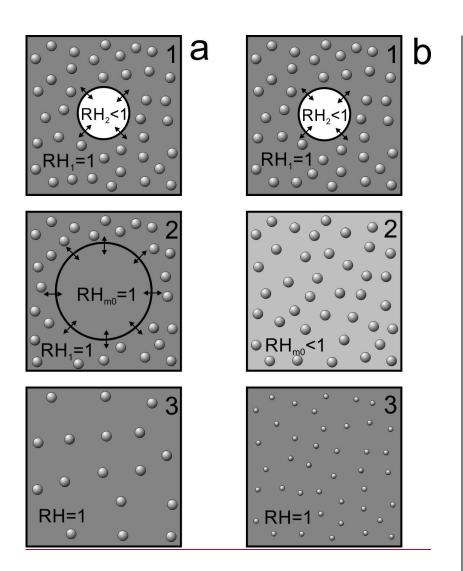
Figure 10. Conceptual diagram explaining breaking the functional relationships between the microphysical moment during progressive missing (see text).

Figure 11. Droplet size distributions formed during the progressive homogeneous mixing corresponding to the (a,e) primary stage; (b,f)  $2^{nd}$  stage; (c,g)  $3^{rd}$  stage; (d,h)  $4^{th}$  stage. Left column (a,b,c,d) corresponds to the case, when the cloud temperature is equal to the dry air temperature  $T_1 = T_2 = 0^{\circ}\text{C}$ ; right column (e,f,g,h) corresponds to the case when  $T_1 = 0^{\circ}\text{C}$ . For both cases the simulation was performed for  $D_1 = 10\mu\text{m}$ ;  $N = 500\text{cm}^{-3}$ ;  $RH_2 = 0.9$ .

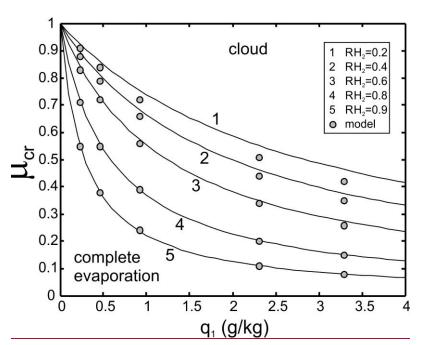
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**Figure 12.** Conceptual diagrams of scattering of measurements of q versus N for (a) extreme 18 19 inhomogeneous and (b) homogeneous mixing. 20 Figure 13. Spatial changes of particle concentration (a), extinction coefficient (b), liquid water 21 content (c) and average and mean mass diameter (d) during transit through one of the convective clouds measured by CDP. The measurements were conducted during the 22 COPE-MED project on 18 July, 2015. The sampling rate 10Hz (~10m spatial resolution) 23 H = 5500 m, T = -12 C, RH = 0.2.24 **Figure 14.** Relatonships between (a) LWC(N); (b)  $\beta(N)$ ; (c)  $D_v(N)$ ; (d)  $LWC(\beta)$  calculated from 25 the CDP measurements obtained during sampling several convective clouds. The 26 meadurements were conducted during the COPE-MED project on 18 July, 2015 27 H=5500m, T=-12C, RH=0.2. The measurements were sampled at 10Hz (~10m spatial 28 resolution). Dashed lines are linear regressions. Red lines indicate primary 29 30 inhomogeneous mixing dependencies calculated for the same environmental conditions Figure 15. Relationships between (a) LWC(N); (b)  $\beta(N)$ ; (c)  $D_v(N)$ ; (d)  $LWC(\beta)$  calculated from 31 the CDP measurements sampled during traverse through 45 convective clouds. The 32 meadurements were conducted during the COPE-MED project on 02 August, 2015 33 34 Dashed lines indicate (a), (b) and (d) indicate the sectors, where the majority of the points are scattered. Figure 9-The altitude of sampling varied in the range 3000m <H < 4500m, 35 temperature -11C<T<0C, relative humidity in the vicinity of clouds 15%<RH<65%. The 36 37 measurements were sampled at 10Hz (~10m spatial resolution). 38 Figure A1. Amount of evaporated liquid water  $\delta q_m$  required for saturation of a cloud volume with 39 initial humidity  $RH_{\rm m}$ . Comparisons of the modeled  $\delta q_m$  and that calculated from Eqs. (A8) and (A9) for three temperatures  $T_{m0} = -20$ C, 0C and 20C. Calculations were 40 41 performed for P=880mb.

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**Figure 1.** Classical conceptual diagram of (a) inhomogeneous and (b) homogeneous mixing. 1 initial state: 2 mixing state; 3 final state.



**Figure 2**. Dependence of critical mixing fraction  $\mu_{cr}$  versus mixing ratio  $q_0$  calculated from Eq.(7). Circles indicate modeled points. The calculations were performed for T=0C and H=3000m.

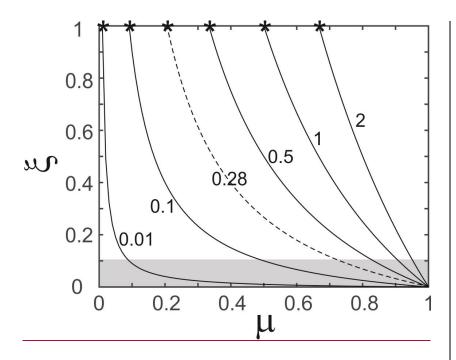


Figure 3. Dependence of  $\xi$  versus  $\mu$ . Numbers are the dimensionless ratios  $\delta q^*/q_1$ . Critical mixing ratios  $\mu_{cr}$  are indicated by stars. Grey area indicates area where the moments of homogeneous and extreme inhomogeneous mixing may not be segregated from in-situ measurements. Dashed line was calculated for the cloud in Figs.13-14.

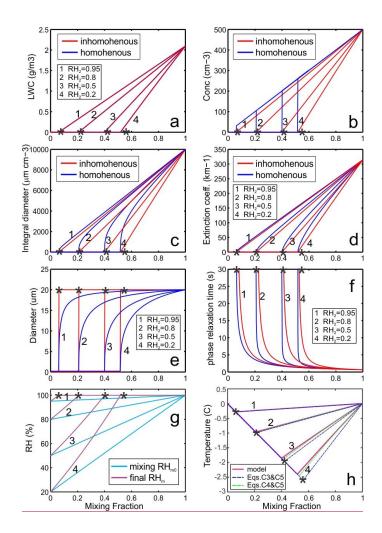


Figure 4. Simulation of (a) liquid water mixing ratio, (b) droplet number concentration, (c) integral droplet diameter, (d) extinction coefficient, (e) mean volume diameter, (f) time of phase relaxation, (g) relative humidity in the mixed volume before droplet evaporation  $RH_{m0}$  and at the equilibrium state  $RH_m$ , (h) final temperature  $T_{m0}$  versus ratio of mixing  $\mu$  formed after homogeneous and extreme inhomogeneous mixing between dry and cloudy parcel with monodisperse droplets. Black stars indicate critical mixing fraction  $\mu_{cr}$  calculated from Eq.(7). The calculations were performed for  $RH_2$  =0.2, 0.5, 0.8, 0.95;  $D_1$ =20 $\mu$ m.

 $N_1 = 500 \text{cm}^{-3}$ ;  $T_1 = T_2 = 0 \text{C}$ ; H = 1000 m.

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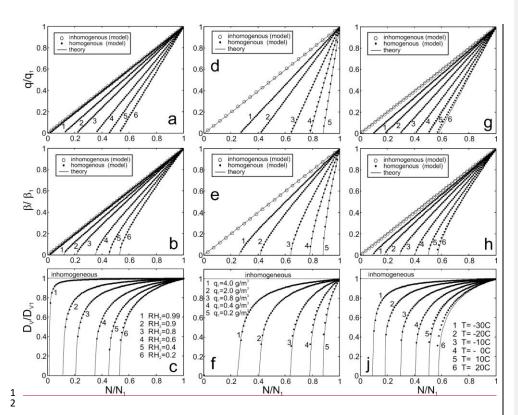


Figure 5. Dependence of normalized liquid water mixing ratio  $q/q_1$  (a,d,g), extinction coefficient  $\beta/\beta$  (b,e,h) and mean volume diameter  $D_v/D_{v1}$  (c,f,j) versus normalized number concentration  $N/N_1$  for various humidity of the entrained air (a,b,c), for various liquid water mixing ratios (d,e,f) and for various temperatures (g,h,j). The calculations were performed the initial conditions: H=1000m,  $D_1=20$  µm; for (a c; g-j)  $N_1=500$ cm<sup>-3</sup>; for (a-f)  $T_1=T_2=0$ C.

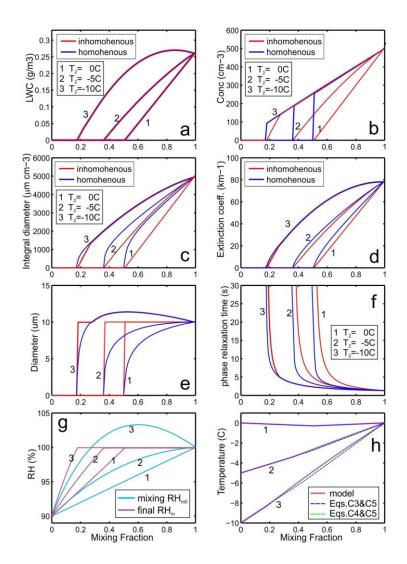


Figure 6. Simulation of (a) droplet number concentration and (b) liquid water mixing ratio, (c) integral droplet diameter, (d) extinction coefficient, (e) mean volume diameter, (f) time of phase relaxation, (g) relative humidity in the mixed volume before droplet evaporation  $RH_{m0}$  and at the equilibrium state  $RH_{m0}$  (h) final temperature  $T_m$  versus ratio of mixing  $\mu$  formed after homogeneous and extreme inhomogeneous

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- 7 mixing between dry and cloudy parcel with monodisperse droplets. The calculations were performed for
- $RH_2 = 0.9$ ;  $D_1 = 10 \mu \text{m}$ ,  $N_1 = 500 \text{cm}^{-3}$ ;  $T_1 = 0 \text{C}$ ;  $T_2 = -10 \text{C}$ , -5 C, 0 C; H = 1000 m.

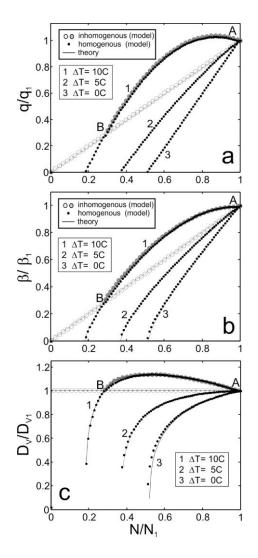


Figure 7. Effect of temperature difference between cloud and entrained air on mixing. The calculations were performed for initial temperatures  $T_2$ : (1) -10C; (2) -5C; (3) 0C. Grey circles indicate extremely inhomogeneous mixing on line 1 at the AB interval. The rest cases on extremely inhomogeneous mixing are indicated by open circles. The initial conditions used for the calculations were: H=1000m,  $RH_2=90\%$ ;  $D_1=10\mu m$ ,  $N_1=500cm^{-3}$ ,  $T_1=0C$ .

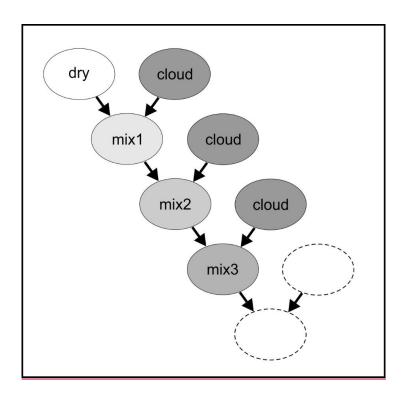
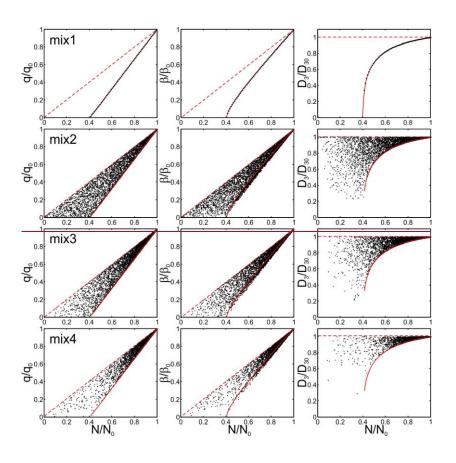


Figure 8, Conceptual diagram of cascade mixing of the out-of-cloud entrained parcel with the cloudy environment

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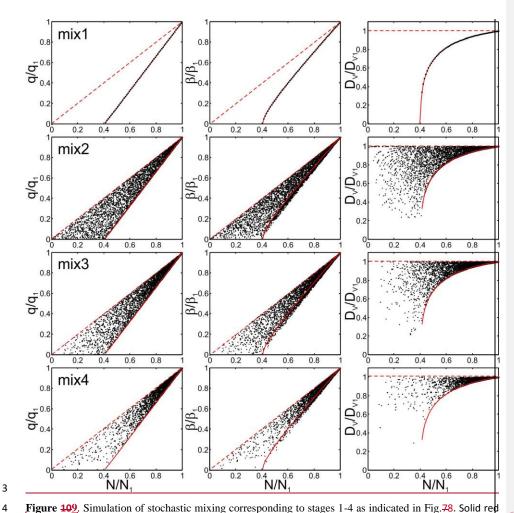


Figure 109. Simulation of stochastic mixing corresponding to stages 1-4 as indicated in Fig. 78. Solid red lines indicate the <u>normalized</u> dependences  $q_1$ ,  $p_2$ ,  $p_3$ ,  $p_4$ ,  $p_5$ ,  $p_6$ ,  $p_8$ 

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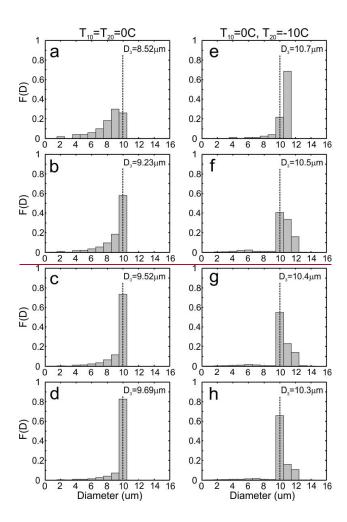
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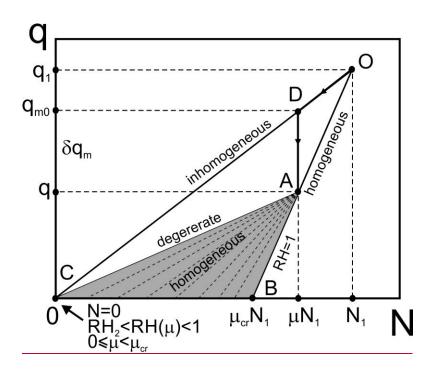


Figure 10. Conceptual diagram explaining breaking the functional relationships between the microphysical moment during progressive missing (see text).

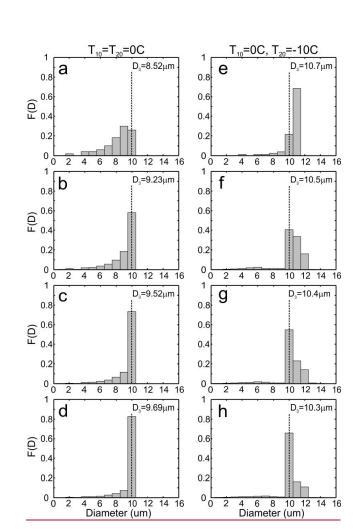
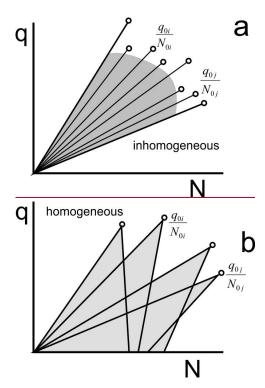


Figure 11. Droplet size distributions formed during the progressive homogeneous mixing corresponding to the (a,e) primary stage; (b,f)  $2^{nd}$  stage; (c,g)  $3^{rd}$  stage; (d,h)  $4^{th}$  stage. Left column (a,b,c,d) corresponds to the case, when the cloud temperature is equal to the dry air temperature  $T_{10} = T_{20} = 0$ C,  $T_{10} = 0$ C,  $T_{10} = 0$ C,  $T_{20} = 1$ C. For both cases the simulation was performed for  $T_{10} = 0$ C,  $T_{20} = 1$ C,  $T_$ 

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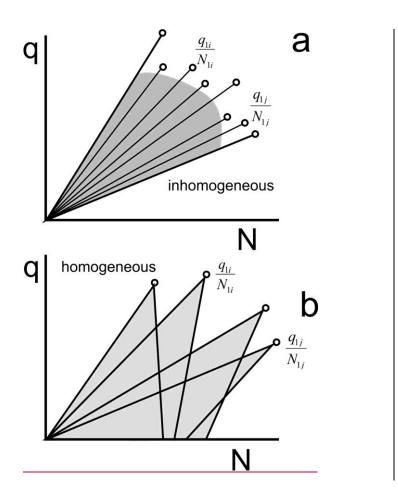


Figure 12. Conceptual diagrams of scattering of measurements of q versus N for (a) extreme inhomogeneous and (b) homogeneous mixing.

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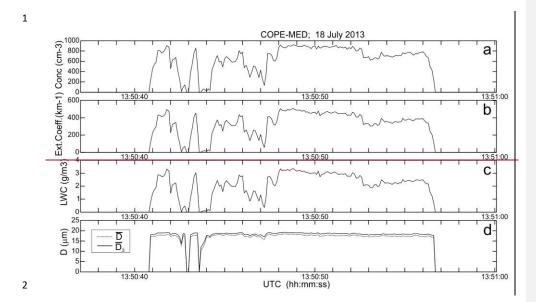
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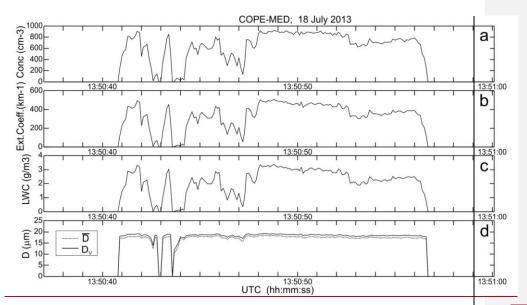
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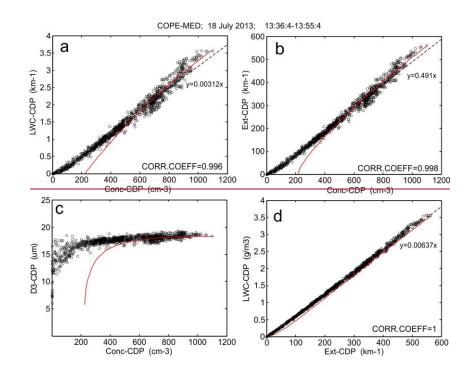




**Figure 13**. Spatial changes of particle concentration (a), extinction coefficient (b), liquid water content (c) and average and mean mass diameter (d) during transit through one of the convective clouds measured by CDP. The measurements were conducted during the COPE-MED project on 18 July, 2015. The sampling rate 10Hz (~10m spatial resolution). #H=5500m, #T=-12C, RH=20%-RH=0.2.

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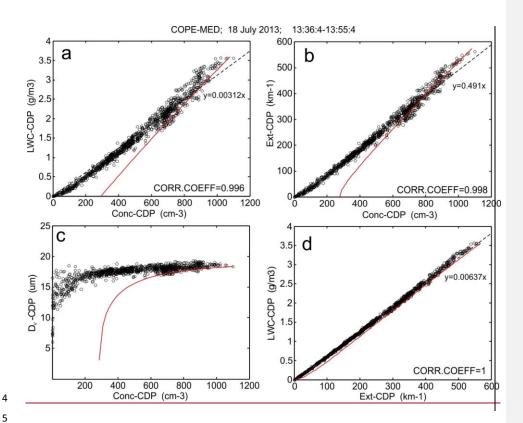


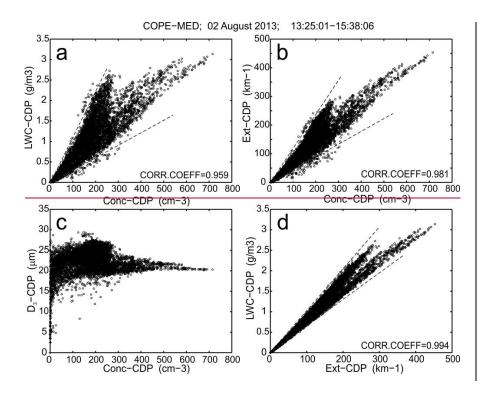
Figure 14. Relatonships between (a)  $\frac{LWC(N);LWC(N);}{LWC(N);LWC(N);}$  (b)  $\frac{B(N);B(N);}{LWC(B);LWC(B)}$  (c)  $\frac{B(N);D_V(N);}{LWC(B);LWC(B)}$  (c) calculated from the CDP measurements obtained during sampling several convective clouds. The meadurements were conducted during the COPE-MED project on 18 July, 2015,  $\frac{B(N);B(N);}{LWC(B)}$  (b)  $\frac{B(N);B(N);}{LWC(B)}$  (c)  $\frac{B(N);B(N);}{LWC(B)}$  (c)  $\frac{B(N);B(N);}{LWC(B)}$  (c)  $\frac{B(N);B(N);}{LWC(B)}$  (d)  $\frac{B(N);B(N);}{LWC(B)}$  (d)  $\frac{B(N);B(N);}{LWC(B)}$  (d)  $\frac{B(N);B(N);}{LWC(B)}$  (d)  $\frac{B(N);B(N);}{LWC(B)}$  (e)  $\frac{B(N);B(N);}{LWC(B)}$  (d)  $\frac{B(N);B(N);}{LWC(B)}$  (e)  $\frac{B(N);B(N);}{LWC(B)}$  (d)  $\frac{B(N);B(N);}{LWC(B)}$  (e)  $\frac{B(N);B(N);}{LWC(B)}$  (d)  $\frac{B(N);B(N);}{LWC(B)}$  (e)  $\frac{B(N);}{LWC(B)}$  (e)  $\frac{B(N);}{LWC(B)}$  (e)  $\frac{B$ 

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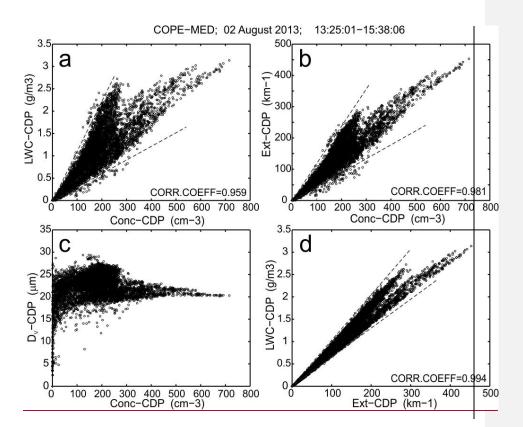


Figure 15. Relatonships between (a)  $\frac{LWC(N);LWC(N)}{LWC(N);LWC(N)}$  (b)  $\frac{LWC(N);LWC(N)}{LWC(N);LWC(N)}$  (c)  $\frac{1}{D_3(N);D_v(N);LWC(N)}$  (d)  $\frac{LWC(B)}{LWC(B)}$  calculated from the CDP measurements sampled during traverse through 45 convective clouds. The meadurements were conducted during the COPE MED project on 02 August, 2015. Dashed lines indicate (a), (b) and (d) indicate the sectors, where the majority of the points are scattered. The meadurements were conducted during the COPE-MED project on 02 August, 2015. Dashed lines indicate (a), (b) and (d) indicate the sectors, where the majority of the points are scattered. The altitude of sampling varied in the range 3000m < HH< 4500m, temperature -11C< T<0C, relative humidity in the vicinity of clouds 15%< RHRH<65%. The measurements were sampled at 10Hz (<10m spatial resolution)

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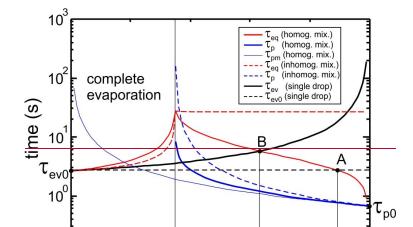
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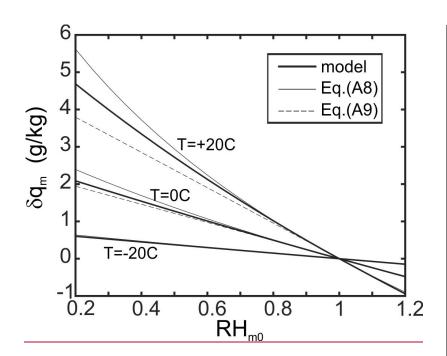


0.4 0.6  $\mu_{B}$ Mixing Fraction 0.8

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**Figure 16.** Dependencies of different characteristic times versus ratio of mixing. The calculations we 8 performed for  $RH_{20}$ =50%;  $r_0$ =10 $\mu$ m,  $N_0$ =500cm<sup>-3</sup>;  $T_{10}$ = $T_{20}$ =00; H=1000m.

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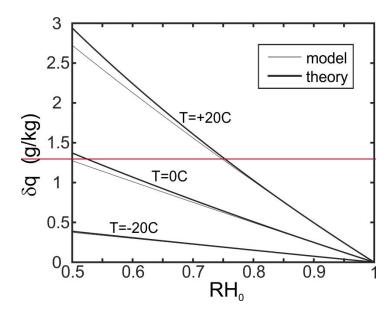


Figure B1 $\underline{A1}$ . Amount of evaporated liquid water  $\delta q \delta q_m$  required for saturation of a cloud volume with initial humidity  $RH_0RH_m$ . Comparisons of the modeled  $\delta q \delta q_m$  and that calculated from  $\underline{Eq.88Eqs.}$  (A8) and (A9) for three different temperatures  $-T_{m0} = 20$ C, 0C and 20C. Calculations were performed for  $\underline{P} = 20$ C, 0C and 20C.

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