We thank three referees for their useful comments and suggestions which have helped us to improve the manuscript. Our point-by-point responses are below. The referees' comments are in *Italic* and our responses are in normal font. Changes in the manuscript (followed the response letter) are highlighted by track changes.

Referee 1:

The manuscript attempts to explain the observation of atmospheric ammonia in the urban atmosphere of Shanghai. The results are interesting and worthy of publication.

We thank the referee for his/her favorable comments.

This reviewer has a few minor comments for the authors considering. 1) P34728, line 5, 'the NH₃ concentrations varied between 0.03 and 39.2 μ g m⁻³'. This reviewer didn't believe that the detection limit of the MARGA can be as low as 0.03 μ g m⁻³ because of highly non-linear response of NH₄⁺ in the system.

Good point! In our study, out of a total of 8447 hourly NH₃ data points, the ten lowest NH₃ values (μ g m⁻³) were 0.03, 0.06, 0.07, 0.08, 0.08, 0.09, 0.10, 0.11, 0.12, and 0.13 (all data are available as requested for reviewing purpose).

The performance of the MARGA has been systematically assessed by the US Environmental Protection Agency previously (Rumsey et al., 2014), and the detection limit for NH_3 at hourly resolution was identified as 0.05 μg m⁻³. Strictly speaking, the values of higher than the twice of the detection limit are thought to be valid. Therefore, the lowest NH_3 concentration in the revised manuscript has been changed to 0.1 μg m⁻³. The number of data points has also been corrected accordingly.

Reference:

Rumsey, I. C., Cowen, K. A., Walker, J. T. et al., An assessment of the performance of the Monitor for AeRosols and GAses in ambient air (MARGA): a semi-continuous method for soluble compounds, Atmos. Chem. Phys., 14, 5639-5658, doi:10.5194/acp-14-5639-2014, 2014.

2) P34728, lines 7-10, "This might be expected since Beijing and Xi'an are located in the North China Plain (NCP) and the Guanzhong Plain (GZP), respectively, two of the most intensive agricultural production regions in China." It is very difficult for this reviewer to understand that Beijing is one of the most intensive agricultural production regions in China.

We are sorry for making the reviewer confused and this sentence has been revised (page 34728, line 9). We would like to clarify that the North China Plain (NCP), instead of Beijing, is one of the most intensive agricultural production regions in China. The NCP (Figure 1), also known as "China's granary", provides 40% and 25% of China's wheat and corn production on 3.3% of the national area (Zhang et al.,

2010). Application rates of mineral nitrogen fertilizers in the NCP are up to 600 kg N ha⁻¹ yr⁻¹ (Ju et al., 2009). Less than 30% efficiency in N application introduces about 40% N loss by various routes, including emissions of NH₃ (Ju et al., 2009). In addition, about 30% of national animal products are also from this area, which further increases the ammonia emission (Zhang et al., 2010). The NCP are recognized as a global hot spot for ammonia emissions (Clarisse et al., 2009). Beijing is situated at the northern tip of the NCP (Figure 1), and the southeasterly winds are prevailing in summer seasons (Sun et al., 2015). Therefore, Beijing is regarded as a receptor of agricultural NH₃ from the NCP (Ianniello et al., 2010).

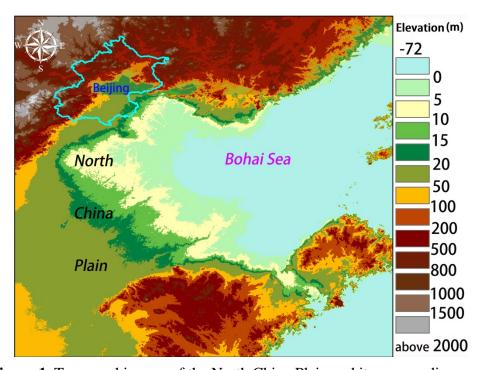


Figure 1. Topographic map of the North China Plain and its surrounding areas

Moreover, Beijing's increase in land area from 4822 km² in 1956 to 16808 km² in 1958 led to the increased adoption of peri-urban agriculture. Such "suburban agriculture" led to more than 70% of non-staple food in Beijing, mainly consisting of vegetables and milk, to be produced by the city itself in the 1960s and 1970s (Cao, 2003). In the late 1990s, recognizing the importance of urban agriculture to sustainable urban development, Beijing's municipal government launched an official program encouraging multi-function urban agriculture in peri-urban areas by supporting the development of "agro-parks", which not only produce food but also attract tourism and are used as educational tools (Cao, 2003). One of the more recent experiments in urban agriculture is the Modern Agricultural Science Demonstration Park in Xiaotangshan Town of Changping District (Cao, 2003). Today, Beijing is leading the way in using smart-city technologies to make urban farming more viable (Cao, 2014).

Reference:

- Cao, J., Periurban agriculture development in China: a new approach in Xiaotangshan, Beijing, UA-Magazine, 2003. Accessible at http://www.ruaf.org/periurban-agriculture-development-china-new-approach-xia otangshan-beijing (2016/1/23).
- Cao, J., Urban agriculture makes China's cities more livable, China Dialogue, 2014 Accessible at https://www.chinadialogue.net/article/show/single/en/7091-Urban-agriculture-makes-China-s-cities-more-liveable (2016/1/23).
- Clarisse, L., Clerbaux, C., Dentener, F., et al. Global ammonia distribution derived from infrared satellite observations. Nature Geosci., 2009, 2(7): 479-483.
- Ju, X. T., Xing, G. X., Chen, X. P., et al. Reducing environmental risk by improving N management in intensive Chinese agricultural systems, PNAS, 2009, 106(9): 3041-3046.
- Ianniello A, Spataro F, Esposito G, et al. Occurrence of gas phase ammonia in the area of Beijing (China), Atmos. Chem. Phys., 2010, 10(19): 9487-9503.
- Sun, Y. L., Wang, Z. F., Du, W., et al. Long-term real-time measurements of aerosol particle composition in Beijing, China: seasonal variations, meteorological effects, and source analysis, Atmos. Chem. Phys., 2015, 15(10): 14549-14591.
- Zhang, Y., Dore, A. J., Ma, L., et al. Agricultural ammonia emissions inventory and spatial distribution in the North China Plain, Environ. Pollut., 2010, 158(2): 490-501.
- 3) P34729, lines 11-12 "Smaller seasonal temperature differences and less agricultural activity in Shanghai could be the contributing factors." This could be reasons, but more direct evidences are needed.

We have to admit that it is very difficult to thoroughly isolate the effects of local emission, regional sources and changes in meteorology on measured ambient NH₃ concentrations. However, our conclusion lies on the well-established fact that almost no one questions the major contribution of agricultural activity to NH₃ emissions, and NH₃ volatilization from fertilizer and livestock waste is highly sensitive to the variations of meteorological parameters (particularly temperature).

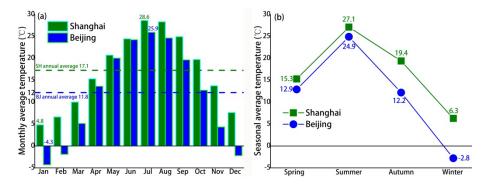


Figure 2. The variations of monthly (**a**) and seasonal (**b**) average temperature (normals 1971-2000) in Shanghai and Beijing.

Specifically, Figure 2 clearly shows that there is a smaller monthly and seasonal temperature differences in Shanghai. Besides, according to Huang et al. (2012), the NH₃ emissions in Beijing were nearly twice than that of Shanghai in 2006. Moreover, as we mentioned above, Beijing is situated at the northern tip of the North China Plain, and the NCP is the most intensive agricultural production region in China. However, located in the mouth of the Yangtze River Delta in East China, Shanghai borders the regions of most developed city clusters in China, and is bounded to the east by the East China Sea.

Collectively, we think at the current stage, it is safe for us to conclude that "smaller seasonal temperature differences and less agricultural activity in Shanghai could be the contributing factors" to its smaller seasonal ambient NH₃ concentration.

Reference:

Huang, X., Song, Y., Li, M., et al. A high-resolution ammonia emission inventory in China, Glob. Biogeochem. Cycl., 2012, 26(1), doi:10.1029/2011GB004161.

4) P34730, lines 3-8, "Although 19.6 mm of rainfall in the July period would be expected to lower NH₃ levels, the temperature on this high concentration date was much higher than on the low concentration March date. Over a longer time frame, even though rainfall in summer was around twice the amount of rainfall in other seasons, other factors such as greater NH₃ emissions at higher temperature outweigh the wet scavenging effects of rainfall yielding higher summertime NH₃ concentrations." The analysis needs to be revised. More and more on-line observations indicated that rainfall enhanced NH₃ emissions.

Taking advantage of fast (1 Hz) measurements of NH₃, we notice that Prof. Murphy's group found the significant increase in NH₃ concentrations followed by precipitation events (Ellis et al., 2011). There is substantial evidence that over certain land types both emission and deposition of NH₃ (known as bi-directional flux or exchange) can occur depending on the NH₃ compensation point (e.g. Farquhar et al., 1980; Sutton et al., 1993; Sutton et al., 1995; Asman et al., 1998; Nemitz et al., 2001). However, we would like to point out the underlying surface in Shanghai is dominated by buildings and roads (hard ground), which could hamper the production of biogenetic-related NH₃ emissions. Besides, given that NH₃ is readily water-soluble, the ambient NH₃ in the urban atmosphere of Shanghai can be expected to be scavenged after rainfall.

Reference:

Asman, W. A. H., Sutton, M. A., Schjørring, J. K., Ammonia: emission, atmospheric transport and deposition, New Phytol., 1998, 139(1): 27-48.

Ellis, R. A., Murphy, J. G., Markovic, M. Z., et al., The influence of gas-particle partitioning and surface-atmosphere exchange on ammonia during BAQS-Met, Atmos. Chem. Phys., 2011, 11(1): 133-145.

Farquhar, G. D., Firth, P. M., Wetselaar, R., et al. On the gaseous exchange of

- ammonia between leaves and the environment: determination of the ammonia compensation point, Plant Physiol., 1980, 66(4): 710-714.
- Nemitz, E., Milford, C., Sutton, M. A., A two-layer canopy compensation point model for describing bi-directional biosphere-atmosphere exchange of ammonia, Q. J. Roy. Meteor. Soc., 2001, 127(573): 815-833.
- Sutton, M. A., Schjorring, J. K., Wyers, G. P., et al., Plant-atmosphere exchange of ammonia, Philos. T. Roy. Soc., 1995, 351(1696): 261-278.
- Sutton, M. A., Flower, D., Moncriefe, J. B., The exchange of atmospheric ammonia with vegetated surfaces. I: Unfertilized vegetation, Q. J. Roy. Meteor. Soc., 1993, 119(513): 1023-1045.
- 5) Section 3.2, to this reviewer, the correlation analysis was valid only if atmospheric NH_3 was derived from local sources. This has to be clarified.

We thank the reviewer for his/her careful review. In this work, the potential source contribution function (PSCF) has been used to identify the possible geographic origins of NH₃ at different seasons. Figure 8c in our manuscript clearly illustrates that for all seasons, the source region of NH₃ in Shanghai is locally-dominated.

6) Section 3.3, in the morning, it has been well demonstrated that dew evaporation can also lead to the elevation of atmospheric NH_3 . This should be added in the revision.

This is a very helpful comment and has been added in our revised manuscript (page 34733, line 10). We also notice that many studies (e.g., Flechard and Fowler, 1998; Wentworth et al., 2014) suggest that pools of surface water (i.e. dew or fog), which form on nights that have a high RH, can act as NH_x (NH₃+NH₄⁺) reservoirs that release NH₃ upon evaporation in the mid-morning, particularly in spring seasons.

Reference:

- Flechard, C. R. and Fowler, D.: Atmospheric ammonia at a moorland site. II: Long-term surface-atmosphere micrometeorological flux measurements, Q. J. Roy. Meteor. Soc., 124, 759-791, doi: 10.1002/qj.49712454706, 1998.
- Wentworth, G. R., Murphy J. G., Gregoire P. K., Cheyne, C. A. L., Tevlin, A. G., and Hems, R.: Soil–atmosphere exchange of ammonia in a non-fertilized grassland: measured emission potentials and inferred fluxes, Biogeosciences, 11, 5675-5686, doi:10.5194/bg-11-5675-2014, 2014.
- 7) Section 3.4, Tunnel is an ideal place to study vehicle emission factors of NO_x , CO and BC, etc. This reviewer has concerns whether it is applicable to study the vehicle emission factor of NH_3 . The potential biogenic emission of NH_3 in the ventilation system could be huge because of huge N sources for bacteria.

The ventilation system for the Handan tunnel was constructed by reinforced concrete.

Based on our visual inspection (Figure 3), the ventilation orifice in Handan tunnel is not a good habitat for the growth and reproduction of bacteria. Besides, the NH_3 concentrations near the ventilation orifice (T_b and T_c) we measured did not support the assumption that "biogenic emission of NH_3 in the ventilation system could be huge".



Figure 3. Photo shows the inner condition of Handan tunnel.

Referee 2:

General comment:

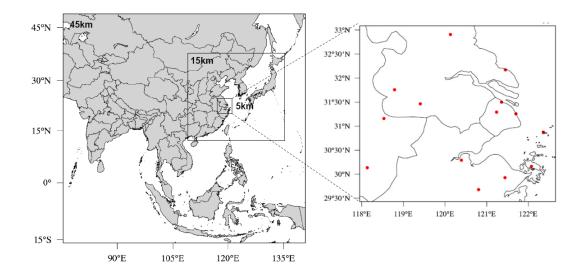
This paper deals with an important topic – the contribution of vehicle emitted NH_3 to the urban atmosphere. The study links a long term measurements with a short-term campaign performed in and out of a major freeway tunnel in Shanghai. Meteorological parameters from the WRF model and the Hysplit model simulations were used to support the analysis of measured NH_3 concentrations. The authors show some interesting results, however some explanations and analysis are not always clear.

We thank the referee for his/her critical comments. One thing we would like to emphasize is that meteorological parameters such as RH, temperature and rainfall we used in this study were from observations except for PBLH which was derived from the WRF modeling results. Based on the specific comments as below, we have responded to all the comments point-by-point and made corresponding changes in the manuscript.

My main comments are: 1. P34730, chapter 3.2. The authors use the boundary layer height (PBLH) from the WRF model simulations and try to interpret the relations between PBLH for different seasons. There is no information on the WRF model domains, spatial resolution, simulation period. Did you verified the WRF modelled PBLH against observations or are there any other studies on this for this region. So, in this context the statement that the boundary layer is similar for four seasons and is

Thanks for the comments. In the revised manuscript, we now added more details about the WRF configuration (page 34724, line 8).

The WRF simulation was performed from a mother domain with a 45×45 km horizontal resolution over Asia, and nested down to a second domain of 15×15 km covering Eastern China, Korea and Japan, and further nested down to a third domain of 5×5 km covering the Yangtze Delta River region. Lambert conformal conic projection was used with true latitude limits of 4° and 44° and standing longitude of 115° . The coverage of three domains is shown in the figure below. We chose the RRTM longwave radiation scheme and the Dudhia shortwave radiation scheme. The Yonsei University scheme was used for the planetary boundary layer option. The simulation period is consistent with the observation, i.e. from April, 2014 to April, 2015. In this study, PBLH derived from the third domain is used.



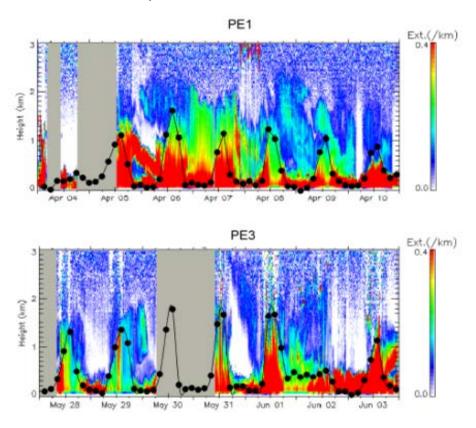
We have performed model evaluations of major meteorological parameters against the NCDC network (National Climate Data Center, http://www7.ncdc.noaa.gov/CDO/cdo), which is limited to the surface observation. The available NCDC sites within the YRD region are marked in the right figure. The evaluation results of wind speed, temperature, and humidity are show in the table below. It could be seen that these meteorological parameters are within the benchmarks (First column in the table) during most of the months, suggesting our WRF modeling results are reliable.

		JAN	FEB	MAR	APR	MAY	JUN	JUL	AUG	SEP	OCT	NOV	DEC
Wind Speed (m/s)	Mean Obs	2.56	2.87	2.98	3.18	3.65	3.22	3.86	3.37	3.34	2.81	2.51	2.43
Benchmark	Mean Sim	3.08	3.38	2.88	3.23	3.92	2.94	3.5	3.45	3.38	2.94	2.84	2.62
\leq ± 0.5	Mean Bias	0.52	0.51	-0.1	0.05	0.27	-0.28	-0.36	0.08	0.04	0.13	0.33	0.19
≤ 2	Gross Error	1.3	1.29	1.19	1.36	1.53	1.16	1.22	1.21	1.14	1.11	1.33	1.22
	RMSE	1.72	1.72	1.65	1.79	1.94	1.51	1.59	1.54	1.44	1.4	1.74	1.6
≥ 0.6	IOA	0.77	0.77	0.81	0.78	0.66	0.73	0.74	0.72	0.79	0.74	0.79	0.78
Temperature (K)	Mean Obs	272.95	278.54	281.18	287.25	293.62	298.5	301.93	302.64	296.17	294.34	281.42	274.53
Benchmark	Mean Sim	273.63	278.3	281.16	287.09	293.8	298.84	301.89	302.27	296.57	294.68	282.45	275.72
\leq \pm 0.5	Mean Bias	0.68	-0.24	-0.02	-0.16	0.18	0.34	-0.04	-0.37	0.4	0.34	1.03	1.19
≤ 2	Gross Error	1.29	1.32	1.41	1.47	1.68	1.71	1.39	1.26	1.28	1.23	1.51	1.69
	RMSE	1.61	1.75	1.83	1.89	2.09	2.19	1.8	1.61	1.59	1.57	1.86	2.06
≥ 0.8	IOA	0.98	0.97	0.96	0.94	0.86	0.87	0.85	0.9	0.89	0.91	0.98	0.97
Humidity (g/kg)	Mean Obs	3.61	4.15	4.84	6.93	11.18	16.35	20.4	19.98	13.97	12.22	5.72	3.72
Benchmark	Mean Sim	4.26	3.67	4.9	7.36	10.2	14.92	19.28	19.21	13.24	11.36	6.68	4.36
≤±1	Mean Bias	0.65	-0.48	0.06	0.43	-0.98	-1.43	-1.12	-0.77	-0.73	-0.86	0.96	0.64
≤ 2	Gross Error	0.51	0.92	0.92	1.45	1.54	1.7	1.06	1.02	0.98	1.34	1.15	0.72
	RMSE	0.63	1.13	1.13	1.76	1.93	2.41	1.56	2.14	1.43	1.7	1.38	0.87
≥ 0.6	IOA	0.86	0.78	0.78	0.73	0.59	0.6	0.71	0.72	0.69	0.58	0.87	0.87

$$MB = \frac{1}{N} \sum_{i=1}^{N} C_m - C_o, GE = \frac{1}{N} \sum_{i=1}^{N} \left| C_m - C_o \right|, RMSE = \sqrt{\frac{1}{N} \sum_{i=1}^{N} (C_m - C_o)^2}, IOA = 1 - \frac{\sum_{i=1}^{N} (C_m - C_o)^2}{\sum_{i=1}^{N} (\left| C_m - \overline{C_o} \right| + \left| C_o - \overline{C_o} \right|)^2}$$

However, there are very rare observations of PBLH in Shanghai as it may require setup of instruments such as Lidar or Ceilometer. In our previous work (Huang et al., 2012), we compared the vertical profile of aerosol extinction measured by a Lidar to the 3-hour resolution PBLH extracted from the NCEP reanalysis data (http://www.atmos-chem-phys-discuss.net/11/C10382/2011/acpd-11-C10382-2011-su pplement.pdf). As shown in the figure below, the temporal variations of NCEP PBLH (black dotted lines) are relatively consistent with the PBLH as visualized from the Lidar measured aerosol extinction at most times during two periods. As the NCEP reanalysis data is used for the inputs for the WRF model, we think the WRF simulated PBLH is capable of reasonably simulating the PBLH.

However, as we didn't have Lidar observations in this study (April 2014-April 2015), we couldn't verify the model simulated PBLH at this moment. We hope the reviewer could understand our difficulty that no observations of PBLH are available.



Reference:

Huang, K., Zhuang, G., Fu, J. S., et al., Typical types and formation mechanisms of haze in an Eastern Asia megacity, Shanghai, Atmos. Chem. Phys., 2012, 12, 105-124.

2. P34728, line 22: "The variations of NH_3 in spring and summer were generally consistent with fluctuations of temperature (Fig. 2a)." The variation of ammonia is also consistent with fluctuations of temperature in winter and autumn. This issue needs to be reconsidered.

Thanks for the comment. In winter, temperature and NH₃ concentration did co-varied but the correlation was much lower than spring (Figure 5). However, in autumn, there was a sustained drop in temperature while the NH₃ concentrations remained quite flat during autumn days (Figure 4).

In the revised manuscript, this sentence has been revised as "The variations of NH₃ in spring and summer were generally consistent with fluctuations of temperature (Fig. 2a). In winter, their correlations turned to be much weaker (Fig. 4d). While in autumn, no significant correlation between temperature and NH₃ was observed (Fig. 4c)."

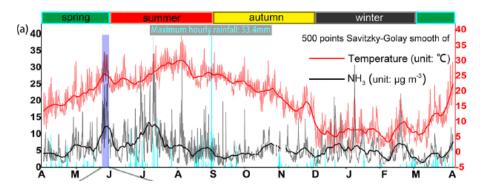


Figure 4. Temporal variations of hourly NH₃ concentrations (gray) and temperature (red), along with 500-point Savitzky-Golay smoothed records in Shanghai from 3 April 2014 to 2 April 2015.

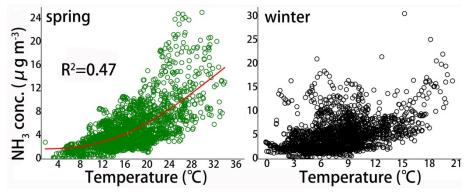


Figure 5. The relationship between hourly NH₃ concentration and hourly temperature in spring (left) and winter during our study period.

3. P34730-31 You have not clarified what is the source of the meteorological data you used, like e.g. temperature, relative humidity, rainfall. Only PBLH data were described in the Methods section. It has to be clarified.

Sorry for our negligence. We have clarified in the revised manuscript (page 34724, line 8) that meteorological parameters including temperature, relative humidity, and rainfall were monitored by an automatic meteorological station (Met One Instruments, US), which was co-located at the rooftop of the Pudong supersite.

4. P34732 "However, it is unexpected that in Shanghai, almost all high NH₃

concentration values in summer are concentrated in the direction of South-Southwest-West (Supplement Fig. S3b), which strongly indicates that the urban area is one of the most important NH_3 emission regions in Shanghai." This is not clear – from the map 1 it is clear that there are rice fields in the south and west from the city. What is the wind direction frequency for this season – it could also influence the results.

We think this question has been well discussed in our manuscript. In terms of emission inventory, the NH₃ emissions in the region of Shanghai (including rural and urban areas) were dominated by agricultural activity and, we agree with the reviewer that there are rice fields in the south and west from the city. However, our PSCF analysis indicated that independent of investigated season, the NH₃ sources in Shanghai were locally dominated during our study period (Figure 6c). This is particularly true in summer: both CO and NH₃ were sourced near the city and therefore yield the highest correlation coefficient (Figure 6b). Moreover, although a much higher temperature in summer tends to enhance the contribution of agricultural NH₃ emissions, almost all high NH₃ concentration values in summer were occurred within the city (indicated by rather low wind speed (lower than 1 m s⁻¹) in Supplement Fig. S3b). This suggests that even in hot summer, there was a strong contribution of NH₃ emissions within the city (e.g., on-road traffic).

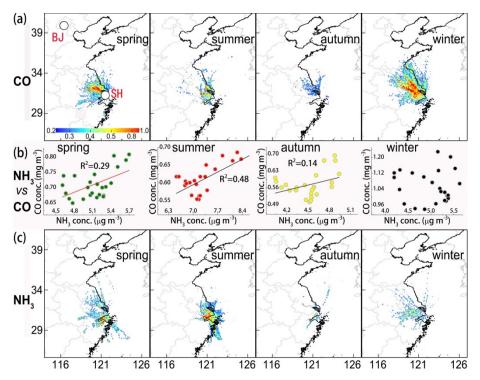


Figure 6. PSCF of CO (**a**) and NH₃ (**c**) during four seasons. The cities marked in each panel are Beijing (BJ) and Shanghai (SH). The color scales indicate the values of PSCF. (**b**) Relationship between hourly NH₃ and CO during four seasons.

The wind rose (wind speed/wind direction) in different season during our study period is shown in Figure 7. It is clear that the winds in summer mainly came from the directions of North and Northwest instead of South-Southwest-West.

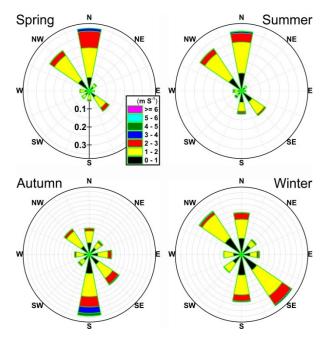


Figure 7. Wind rose plots, color coded by wind speed for each season during our study period. The frequencies are set to the same scales for all seasons.

5. P34733 "NH₃ also displays a clear bimodal profile during all four seasons, similar to the CO diurnal profile, suggesting a significant influence of on-road traffic (with daily commuting) on ambient NH₃ concentrations in the urban environment of Shanghai." — concerns fig.7 Similar profiles were previously observed for agricultural stations (e.g. in (Aas et al., 2012) or (Schaap et al., 2011)). Please compare the profiles you got to other studies. Aas, W., Tsyro, S., Bieber, E., Bergström, R., Ceburnis, D., Ellermann, T., Fagerli, H., Frölich, M., Gehrig, R., Makkonen, U., Nemitz, E., Otjes, R., Perez, N., Perrino, C., Prévôt, A. S. H., Putaud, J.-P., Simpson, D., Spindler, G., Vana, M. and Yttri, K. E.: Lessons learnt from the first EMEP intensive measurement periods, Atmos. Chem. Phys., 12(17), 8073–8094, doi:10.5194/acp-12-8073-2012, 2012. Schaap, M., Otjes, R. P. and Weijers, E. P.: Illustrating the benefit of using hourly monitoring data on secondary inorganic aerosol and its precursors for model evaluation, Atmos. Chem. Phys., 11(21), 11041–11053, doi:10.5194/acp-11-11041-2011, 2011.

Thanks for the comment. As suggested, we bring out the observed and modeled NH₃ diurnal profiles from agricultural stations in Aas et al. (2012) and Schaap et al. (2011) that are shown in Figure 8 and Figure 9, respectively. It is very clear to tell the difference between the urban station (Shanghai Pudong supersite; Figure 10) and the agricultural stations in terms of their diurnal variation pattern of NH₃. In Shanghai urban areas (Figure 10), "NH₃ displays a clear bimodal profile during all four seasons". In Schaap et al. (2011), the variation of NH₃ at Cabauw was clearly characterized of a single modal profile and didn't show any peaks during the evening rush hours (Figure 9). As similar as in Aas et al. (2012), generally there were no high

peaks during the rush hours in different seasons (Figure 8). Specifically, at the Ispra site, high values of NH₃ occurred during the daytime which was clearly related to the agricultural activities.

Overall, we found out significant differences of the NH_3 diurnal profiles between our urban site and the agricultural sites. We thank for the reviewer for providing these references.

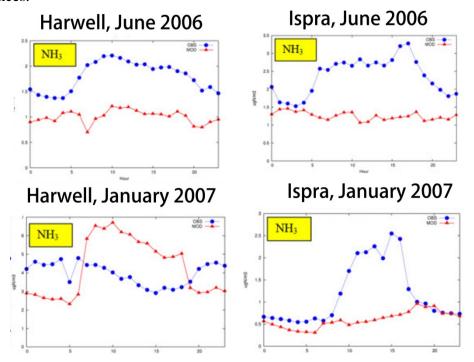


Figure 8. Diurnal variations of ambient NH₃ from some of the intensive measurements compared with the EMEP model. This figure was modified from Figure 8 in Aas et al. (2012).

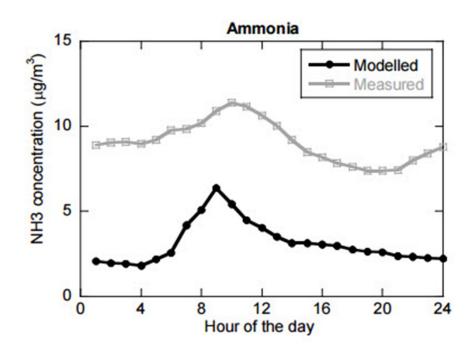


Figure 9. Comparison of the measured (grey) diurnal cycle for NH₃ at Cabauw with 7 km grid resolution estimates (black) of the LOTOS-EUROS model. This figure was modified from Figure 6 in Schaap et al. (2011).

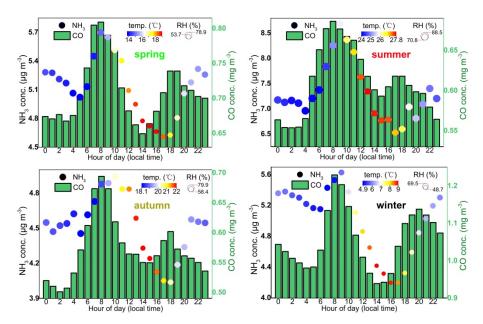


Figure 10. Seasonal diurnal profiles of NH₃ and CO concentrations determined at the Pudong supersite in our study. Color coded by hourly temperature and circle radius coded by hourly relative humidity.

Reference:

Aas, W., Tsyro, S., Bieber, E., et al., Lessons learnt from the first EMEP intensive measurement periods, Atmos. Chem. Phys., 2012, 12(17): 8073-8094.

Schaap, M., Otjes, R. P., Weijers, E. P., Illustrating the benefit of using hourly monitoring data on secondary inorganic aerosol and its precursors for model evaluation, Atmos. Chem. Phys., 2011, 11(21): 11041-11053.

6. P34736 "During spring, ambient NH_3 concentrations appeared to be influenced to some extent by temperature-dependent emissions, likely from agricultural activities including crop fertilization. No such relationship was apparent during other seasons" This is not true. NH_3 concentrations seem to be dependent on temperature also for other seasons. It is visible in fig. 2. Please clarify.

Please refer to our response to *comment 2*.

7. P34736 While mixing height of planetary boundary layer and relative humidity were not the main factors influencing seasonal NH_3 concentrations. The PBL issue has to be reconsidered after the explanation of the PBLH data quality used in this study (please see also comment 1). Compare your results on PBLH with other studies.

Please refer to our response to *comment 1*.

8. The diurnal profile of NH_3 concentrations showed a typical bimodal cycle during four seasons, with maxima in the morning and the evening rush hours, suggesting a persistent influence of on-road traffic (with daily commuting) on ambient NH_3 levels in Shanghai. It has to be reconsidered (please see comment 5).

Please refer to our response to *comment 5*.

Referee 3 (Prof. Xuejun Liu):

General comments

Ammonia is regarded mainly as emissions from agricultural sources (e.g. ammonium-based N fertilizers and animal husbandry). However, ammonia emissions from non-agricultural or urban sources have been paid much less attention currently. This manuscript provides strong evidence that traffic vehicles are significant urban ammonia sources which contribute to ground level NH_3 in megacity of Shanghai. The authors use a one-year continuous monitoring data from a super cite in Pudong (east of Shanghai) and monitoring results from an urban tunnel (west of Shanghai) to support their conclusions. In addition, using bottom-up approach, they have estimated overall annual vehicle emissions of ammonia in the megacity of Shanghai (1300 t yr^{-1}) for the first time, in spite of some uncertainties. Although non-agricultural source NH_3 emission is not new topic, the manuscript systematically proves the non-negligible contribution of traffic vehicles, as an important non-agricultural source, to urban ammonia pollution. The related results may provide implications for haze or $PM_{2.5}$ pollution in megacities worldwide. I strongly support the publication of the manuscript after minor revisions as suggested in the specific comments.

We are grateful to Prof. Liu for his strong support and highly praised to our work.

Specific comments

1. The title of the paper could be slightly modified as "Non-negligible contribution of vehicle emissions to atmospheric ammonia in the megacity of Shanghai".

We think the word "non-negligible" has some sort of ambiguity. After personal communication with Prof. Liu, we would like to keep our original title.

2. In the section of Results and Discussion, I suggest the authors provide a combined analysis of ammonia and ammonium (ion) dynamics in $PM_{2.5}$ at the PD super site.

The analysis of water-soluble traces gases and ions data in PM_{2.5} concentrations by MARGA is not presented in the present manuscript, although models applied to the MARGA database is usefull to investigate the contribution and processess of NH₃ to secondary inorganic aerosol formation. Such work will be addressed in details, with further tests in the near future.

3. Legend (of land use) in Figure 1: using "upland cropland" instead of "drought land".

Revised accordingly.

4. Lines 7-8 in Page 38422: "Despite the focus on ammonia sources mainly from agricultural and rural environments,...".

Revised accordingly.

5. Mileage emission factor for NH_3 in this study is quite different from other reports and it can be discussed for details in the revision.

Both performed in two of China's megacities (Shanghai and Guangzhou), the emission factor (EF) for vehicle-emitted NH_3 in our study is much lower than that work of Liu et al. (2014). In fact, the EF we derived is very similar to that observed for the Gurbrist tunnel in Switzerland (31 \pm 4 mg km⁻¹) (Emmenegger et al., 2004) and the Caldecott tunnel in California (49 \pm 3 mg km⁻¹) (Kean et al., 2000). Yunhua Chang (the first author of the manuscript) had tried to discuss this with one of the co-authors of Liu et al., (2014) via email, but unfortunately, no feedback has yet been received.

In collaboration with Yunting Fang, Professor of the Institute of Applied Ecology, Chinese Academy of Sciences, we recently measured the vehicle-emitted NH₃ direct from vehicular tailpipes. We found that varying at different stages, the NH₃ concentrations ranged from 300 ppb to 500 ppb (unpublished data). We have no intention to judge any work related to the EF of vehicle-emitted NH₃, however, we still think the NH₃ concentrations measured at the exit of the Zhujiang tunnel in Guangzhou by Liu et al. (2014) were unexpectedly high (729±497 ppb). In our manuscript, we call for more research to be undertaken to pinpoint this parameter in order to accurately quantify the level of NH₃ emitted from vehicles.

Reference:

- Emmenegger, L., Mohn, J., Sigrist, M., Marinov, D., Steinemann, U., Zumsteg, F., and Meier, M., Measurement of ammonia emissions using various techniques in a comparative tunnel study, Int. J. Environ. Pollut., 2004, 22, 326–341, doi:10.1504/IJEP.2004.005547.
- Kean, A. J., Harley, R. A., Littlejohn, D., and Kendall, G. R., On-road measurement of ammonia and other motor vehicle exhaust emissions, Environ. Sci. Technol., 2000, 34, 3535–3539, doi:10.1021/es991451q.
- Liu T, Wang X, Wang B, et al., Emission factor of ammonia (NH₃) from on-road vehicles in China: tunnel tests in urban Guangzhou, Environ. Res. Lett., 2014, 9(6), doi:10.1088/1748-9326/9/6/064027.



This discussion paper is/has been under review for the journal Atmospheric Chemistry and Physics (ACP). Please refer to the corresponding final paper in ACP if available.

The importance of vehicle emissions as a source of atmospheric ammonia in the megacity of Shanghai

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Received: 1 December 2015 - Accepted: 2 December 2015 - Published: 10 December 2015

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Published by Copernicus Publications on behalf of the European Geosciences Union.

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Agricultural activities are a major source contributing to NH₃ emissions in Shanghai and most other regions of China; however, there is a long-standing and ongoing controversy regarding the contributions of vehicle-emitted NH₃ to the urban atmosphere. From April 2014 to April 2015, we conducted measurements of a wide range of gases (including NH₃) and the chemical properties of PM_{2.5} at hourly resolution at a Shanghai urban supersite. This large dataset shows NH₃ pollution events, lasting several hours with concentrations four times the annual average of 5.3 µg m⁻³, caused by the burning of crop residues in spring. There are also generally higher NH₃ concentrations (mean $\pm 1\sigma$) in summer (7.3 \pm 4.9 μ g m⁻³; n = 2181) because of intensive emissions from temperature-dependent agricultural sources. However, the NH₃ concentration in summer was only an average of 2.4 µg m⁻³ or 41 % higher than the average NH₂ concentration of other seasons. Furthermore, the NH₂ concentration in winter $(5.0 \pm 3.7 \,\mu\text{g m}^{-3}; n = 2113)$ was similar to that in spring $(5.1 \pm 3.8 \,\mu\text{g m}^{-3}; n = 2204)$ but slightly higher, on average, than that in autumn $(4.5 \pm 2.3 \,\mu\text{g m}^{-3}; n = 1949)$. Moreover, other meteorological parameters like planetary boundary layer height and relative humidity were not major factors affecting seasonal NH3 concentrations. These findings suggest that there may be some climate-independent NH₃ sources present in the Shanghai urban area. Independent of season, the concentrations of both NH₃ and CO present a marked bimodal diurnal profile, with maxima in the morning and the evening. A spatial analysis suggests that elevated concentrations of NH₃ are often associated with transport from regions west-northwest and east-southeast of the city, areas with dense road systems. The spatial origin of NH₃ and the diurnal concentration profile together suggest the importance of vehicle-derived NH₃ associated with daily commuting in the urban environment. To further examine vehicular NH₃ emissions and transport, sampling of the NH₃ concentration was performed in (from the entrance to the exit of the tunnel) and out (along a roadside transect spanning 310 m perpendicular to the tunnel) of a heavily trafficked urban tunnel during the spring 2014. NH₃ concentrations

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in the tunnel exit were over 5 and 11 times higher than those in the tunnel entrance and in the ambient air, respectively. Based on the derived mileage-based NH₃ emission factor of 28 mg km⁻¹, a population of 3.04 million vehicles in Shanghai produced around 1300 t NH_3 in 2014, which accounts for 12 % of total NH_3 emissions in the urban area. Collectively, our results clearly show that vehicle emissions associated with combustion are an important NH₃ source in Shanghai urban areas and may have potential implications for PM_{2.5} pollution in the urban atmosphere.

Introduction

Ammonia (NH₂) is one of the most abundant nitrogen-containing substances and the principal reduced nitrogen component in the atmosphere. It plays a strong role in local and regional scale tropospheric chemistry and air quality by serving as a precursor to particulate ammonium (pNH_4^+) (Seinfeld and Pandis, 2012). Although major efforts have been made towards regulating NO_x and SO₂ emissions to improve air quality in China (Wang et al., 2014; Zhao et al., 2013), a major portion of the nation's population presently lives in environments of non-compliance with national standards for fine particulate matter (PM_{2.5}, representing particles with aerodynamic diameters smaller than 2.5 microns) (Huang et al., 2014; Lin et al., 2010; Ma et al., 2014, 2015). NH₃ emission reduction has been proposed as a cost-effective strategy to lower ambient PM_{2.5} levels (Heald et al., 2012; Pinder et al., 2007; Wang et al., 2011, 2013; Ye et al., 2011). However, the emission sources of NH₃ and their relative contributions to ambient concentrations, especially in urban atmospheres, remain uncertain (Chang, 2014; Felix et al., 2014; Yao et al., 2013).

Emission sources of NH₃ have been previously reviewed (e.g., Asman et al., 1998; Reis et al., 2009; Sutton et al., 2008). Major sources include volatilization of Ncontaining fertilizers and excreta from animal husbandry, which together contribute over 80 % of total global NH₃ emissions (Bouwman et al., 1997; Clarisse et al., 2009; Olivier et al., 1998; Schlesinger and Hartley, 1992). Thus it is not surprising that previous

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investigations of NH₃ emissions were mainly performed adjacent to dairy operations (Mount et al., 2002), animal housing (Gay et al., 2003), livestock facilities (Kawashima and Yonemura, 2001), slurry lagoons (Aneja et al., 2000), pit latrines (Rodhe et al., 2004), and croplands (Yan et al., 2003), where elevated levels of NH₃ are often ob-5 served. Varying significantly in time and space, biomass burning (including agricultural waste, savanna and forest fires) may contribute up to 12% of the global NH₃ emissions flux (Behera et al., 2013; Lamarque et al., 2010). Despite the focus on ammonia sources is typical of agricultural and rural environments, a number of studies reveal that ambient NH₃ concentrations in urban areas can be comparable to (Cao et al., 2009; Stanier et al., 2012) or even higher than (Bettez et al., 2013; Meng et al., 2011; Singh and Kulshrestha, 2014) those in rural areas. These observations strongly suggest that there must be other non-agricultural NH₃ sources present in urban areas.

Starting in the 1980s, the introduction of three-way catalytic converters (TWCs) on automobiles dramatically mitigated pollutant emissions from vehicle tailpipes (Shelef and McCabe, 2000). An unwanted side effect of the use of TWCs for gasoline powered vehicles and selective catalytic reduction (SCR) for control of nitrogen oxides emissions from diesel powered vehicles, has been an increase in NH₃ emissions from motor vehicles, a significant source of non-agricultural NH3 that has been documented directly through laboratory dynamometer studies (Durbin et al., 2002; Heeb et al., 2008, 2006; Huai et al., 2005; Livingston et al., 2009; Suarez-Bertoa et al., 2014, 2015) and on-road measurements (including mobile chase systems and tunnel tests) (Brito et al., 2013; Fraser and Cass, 1998; Kean et al., 2009; Liu et al., 2014; Moeckli et al., 1996; Pierson and Brachaczek, 1983; Pierson et al., 1996; Sun et al., 2014), or indirectly through correlation analysis between ambient NH₃ concentrations and other recognized traffic tracers (e.g., CO, NO_x) (Bishop and Stedman, 2015; Gong et al., 2011, 2013; Ianniello et al., 2010; Nowak et al., 2010; Pandolfi et al., 2012; Phan et al., 2013; Reche et al., 2012). In the US, it is estimated that 5% of the national NH₃ emissions are due to motor vehicles (Kean et al., 2009), while this figure is estimated at 12 % for the UK (Sutton et al., 2000), with almost all the remaining NH₃ coming from agricultural processes. At

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a regional level, motor vehicle emissions make a small contribution to the total. Nevertheless, they are locally concentrated in urban areas where agricultural sources of NH₂ are mostly absent. Therefore, a disproportionately greater impact of motor vehicles on the urban NH₃ budget and subsequent secondary PM_{2.5} formation can be expected (Chang, 2014). On the other hand, we notice that several important studies did not detect evidence of an influence of on-road traffic on ambient NH₃ concentrations (Pryor et al., 2004; Saylor et al., 2010; Yao et al., 2013). Therefore, more efforts needed to be made to elucidate the contribution of vehicle-emitted NH₃ to the urban atmosphere.

Shanghai, like many other cities in eastern China, is suffering severe air pollution problems, such as high PM_{2.5} concentrations and resulting poor visibility (Huang et al., 2013b, 2012). Although there are many studies aimed at understanding PM pollution, little is known about the characteristics of NH₃ in the largest city of China. In an effort to curb its severe air pollution, China recently launched an air pollution monitoring research program (known as the supersite program) in several major cities. In 2014, a new in situ atmospheric station equipped with state-of-the-art instruments was installed in the Shanghai region, allowing comprehensive characterization of PM_{2.5} and associated precursor gases. Here seasonal trends, diurnal variations and pollution episodes retrieved from one year of real-time measurement of NH₃ are presented and interpreted in order to explore the sources and parameters controlling the NH3 concentrations across Shanghai. Meanwhile, an additional source-specific campaign was performed to examine the emission and transport of vehicle-emitted NH₃ from an urban heavily trafficked tunnel in Shanghai.

Methods

Long-term monitoring at Pudong supersite

In situ continuous observations of the chemical and optical properties of atmospheric aerosols and associated precursor gases were made on the rooftop (18 ma.g.l.) of

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the Pudong Environmental Monitoring Center (PEMC; 121.5446° E, 31.2331° N), 5 km east of the Shanghai urban center (the People's Square) (Fig. 1). The site is located in a mixed-use urban area (office, commercial, residential and traffic) east of downtown Shanghai, with no obvious NH₃ point source within 5 km (Zou et al., 2015). As one of the state-controlled sites, Pudong (PD) supersite was designed by the Ministry of Environmental Protection of China and operated by the Shanghai Environmental Monitoring Center, being responsible for the release of hourly air quality data for PM₁₀, PM_{2.5}, and other criteria pollutants (CO, SO₂, NO_x and O₃).

From 3 April 2014 to 2 April 2015, using a MARGA instrument (Measurement of Aerosols and Reactive Gases Analyzer, Metrohm Applikon B.V., NL), water-soluble gases (NH $_3$, HNO $_3$, HONO, HCl and SO $_2$) and PM $_{2.5}$ components (NO $_3^-$, Cl $^-$, SO $_4^{2-}$, Na $^+$, NH $_4^+$, K $^+$, Mg $^{2+}$ and Ca $^{2+}$) were measured with hourly temporal resolution. The MARGA removes soluble gases in a rotating, wet-walled denuder, while a steam-jet aerosol collector is used for fine particle collection. Meanwhile, aerosol light absorption coefficients (b_{abs}) were retrieved every 5 min from an AE31 Aethalometer using seven wavelengths (370, 470, 520, 590, 660, 880 and 950 nm) with a PM $_{2.5}$ cut-off inlet. Black carbon (BC) concentrations for the whole data set were calculated from the absorption coefficient at 880 nm. The measurement process was subjected to rigorous quality assurance and quality control procedures according to the Technical Guideline of Automatic Stations of Ambient Air Quality in Shanghai based on the national specification HJ/T193-2005.

To explore the comparability between on-line and off-line methods for NH $_3$ measurement, an Ogawa passive sampling device (PSD) was co-located with MARGA at PD to passively measure weekly ambient NH $_3$ concentration from May 2014 to June 2015. The Ogawa PSD is a double-sided passive sampler equipped with two 14 mm quartz filters (serving as duplicates) impregnated with phosphoric acid provided by the manufacturer. Following the manufacturer's protocols (http://www.ogawausa.com), exposed filter samples were soaked with 8 mL ultra-pure water (18.2 M Ω cm) and analyzed by an ion chromatography system (883 Basic IC plus, Metrohm Co., Switzerland). The

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detection limit for NH₄⁺ in the passive sampler extracts was $2.8 \,\mu g \, L^{-1}$; this corresponds to an ambient NH₃ concentration detection limit of approximately 0.1 ppb for a 7 day sample. The NH₃ concentrations measured by the MARGA (ppb) were averaged over the same time period as the Ogawa PSDs (ppb). Fig. S1 in the Supplement shows a good correlation ($y = 0.82 \times +0.56$, n = 53, $R^2 = 0.84$, p < 0.001) between the two NH₃ measurement methods, validating the reliability of NH₃ data from the MARGA platform.

2.2 On-road measurement of NH₃ concentration in and out of a tunnel

To complement the information obtained from the main monitoring campaign described above, additional measurements of NH₃ concentration were performed at eight sites in and out the Handan tunnel from 9 April to 21 May 2014. The Handan tunnel is a 720 m long urban freeway in the northeast of Shanghai, separating the campus of Fudan University into two parts (Fig. 9a). It contains an array of ventilation orifices in the middle section of the tunnel, 200 m in total. The tunnel has two traffic bores; each bore has a cross section of 70 m² and four lanes with typically 120 000 vehicles (of which 85 % of are light-duty vehicles) passing per day (Li, 2007). Driven by a group of high power fans, the average wind speed measured at the exit of the tunnel was approximately $5 \pm 1 \,\mathrm{m \, s}^{-1}$. The maximum vehicle speed limit in the tunnel is $80 \,\mathrm{km \, h}^{-1}$, with typical driving speeds of 50-60 km h⁻¹. Inside the northern bore of the tunnel, four sampling points were located at both ends of the tunnel (10 m from the exit and entrance of the tunnel, or T-d and T-a, for short) and the two ends of an array of ventilation orifices located in the middle section of the tunnel (Fig. 9a; the site near the entrance and the exit named as T-b, and T-c for short, respectively). Outside the tunnel, a roadside transect involving four sites perpendicular to the tunnel was established, spanning the distance from 0 m (O_{0m} , for short), 20 m (O_{20m}), 150 m (O_{150m}), to 310 m (O_{310m}). Figure 9a shows the layout of the tunnel and the sampling points.

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Using US EPA Method 207.1 (Determination of Ammonia Emissions from Stationary Sources), the NH₃ concentration at each site was measured. Briefly, for each sample, ambient air was pumped through two fritted glass bubblers (containing 10 mL $0.005\,\text{mol}\,\text{L}^{-1}\,\text{H}_2\text{SO}_4$ absorbing solution in each bubbler) for two hours at a flow rate of $1\,\text{L}\,\text{min}^{-1}$. These two bubblers were connected in series, and the NH₃ collection efficiency of the sampling trains was 95% or better (checked by using four bubblers in series in our pilot study, the collection efficiency = $100\times$ (the sum of the values of the first two bubblers)/the sum of the values of the four bubblers). Measurements were made during the morning (between 08:00 and 11:00 local time) and afternoon (between 14:00 and 19:00 LT). Due to the proximity of the monitoring sites to the laboratory, all samples could be collected and analyzed by IC swiftly to avoid potential contamination, and field blanks were below the detection limit. Due to the dangers to personnel of sampling at the T-a, T-b and T-c sites, six samples were collected synchronously at these three sites. 19 paired samples were successfully collected and determined at the site of T-d, $O_{0.m}$, $O_{20.m}$, $O_{150.m}$ and $O_{310.m}$.

2.3 Planetary boundary layer height simulation

The Weather Research and Forecasting (WRF) model v3.5.1 (Skamarock et al., 2008) is used for simulating the height of planetary boundary layer. The National Centers for Environmental Prediction (NCEP) Final (FNL) Operational Global Analysis dataset (http://rda.ucar.edu/datasets/ds083.2/) with a horizontal resolution of a resolution of 1.0° × 1.0° are incorporated as initial and boundary conditions for the model, An one-way nested approach with four-dimensional data assimilation (FDDA) in WRF is applied. The WRF model configurations can be found elsewhere (Huang et al., 2013a). A Meteorology/Chemistry Interface Processor (MCIP) (Otte and Pleim, 2010) v4.1 is used to postprocess the WRF output. The atmospheric height of the planetary boundary layer field, one of the standard MCIP outputs, is used in this study. Additionally, the planetary boundary layer depths at 3 h resolution were obtained from the US National Oceanic and Atmospheric Administration (NOAA) READY archived Global Data Assim-

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ilation System (GDAS) meteorological data ($1^{\circ} \times 1^{\circ}$) based on Coordinated Universal Time (UTC). All UTC values are converted to local time (UTC + 8).

2.4 Potential source contribution analysis

24 h back trajectories arriving at the PD supersite at a height of 500 m were calculated at 1 h time intervals for each of the four seasons using NOAA Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model with GDAS one degree archive meteorological data (Draxler and Rolph, 1997). An in-depth back trajectory analysis, the potential source contribution function (PSCF), is useful for identifying the possible geographic origin of emission sources; this method calculates the ratio of the number of points with concentration higher than a threshold value (m_{ij}) to the total number of points (n_{ij}) in the ijth grid cell. Higher PSCF values indicate higher potential source contributions to the receptor site. In this study, the domain for the PSCF was set within the range of (26–42° N, 112.5–125.5° E) in 0.1° × 0.1° grid cells. The 75th percentile for CO and NH₃ during the four seasons was used as the threshold value m_{ij} . To reduce the uncertainties of m_{ij}/n_{ij} for those grid cells with a limited number of points, a weighting function recommended by (Polissar et al., 2001) was applied to the PSCF in each season. Visualizations of the PSCF were mapped using ArcMap 10.2.

3 Results and discussion

3.1 Temporal evolution of NH₃ concentrations

The temporal patterns of hourly gaseous NH₃ concentrations determined by the MARGA at the Pudong supersite are reported in Fig. 2. Summary statistics for NH₃ concentrations (μg m⁻³) during 3 April 2014–2 April 2015 are shown in Table 1. Using a variety of chemical, physical and optical techniques, numerous studies have examined ambient NH₃ concentrations over the last three decades; however, few of them

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were conducted in urban areas. As a comparison, we compiled previous work related to the measurement of urban NH_3 concentrations in Table 2.

The one year dataset (n = 8447; data availability 96.4%) in the current study represents one of the longest on-line continuous measurement series of atmospheric NH3 in China. During the study period, the NH₃ concentrations varied between 0.03 and 39.2 μ g m⁻³, with an average ($\pm 1\sigma$) of 5.5 \pm 3.9 μ g m⁻³. Domestically, the annual average NH₃ concentrations in Beijing and Xi'an were much higher than in Shanghai (see Table 2). This might be expected since Beijing and Xi'an are located in the North China Plain (NCP) and the Guanzhong Plain (GZP), respectively, two of the most intensive agricultural production regions in China. Moreover, the NH₃ loss from soil increases with an increase in soil pH value (Ju et al., 2009). Shanghai and its surrounding regions are dominated by the acid soils of paddy fields (Fig. 1) (Zhao et al., 2009), while Beijing and Xi'an are dominated by the alkaline soils of dry land (Wei et al., 2013). Internationally, the NH₃ concencentration level in Shanghai was similar to observations from cities in developed and middle-income countries, but much lower than those cities in emerging countries. This is particularly true when comparing with cities in South Asia (e.g., Delhi in India and Lahore in Pakistan), where there is a lack of basic sanitation facilities (e.g., public flush toilets) and significant animal populations (such as cow) coexist with people in urban areas. The higher NH₃ concentrations measured at surface sites in South Asia are consistent with spatial patterns from recent satellite remote sensing observations (Clarisse et al., 2009; Van Damme et al., 2014).

The variations of NH₃ in spring and summer were generally consistent with fluctuations of temperature (Fig. 2a). Monthly, from March to September, the NH₃ concentration first increased steadily, with the highest value in July, then decreased gradually, along with falling temperature (Fig. 2c). In summer (June to August), high temperatures favor NH₃ volatilization from urea and other N fertilizers applied to croplands (Fu et al., 2013; Huang et al., 2011; Ianniello et al., 2010; Meng et al., 2011). High temperatures in summer also favor NH₃ emission from other sources, such as animal housing, landfills, laystalls and pit latrines, animal manure, natural and fertilized soils, vegetation, and

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municipal solid waste (Fu et al., 2013; Huang et al., 2011). Moreover, given that the equilibrium between ammonium nitrate particles and gaseous ammonia and nitric acid favors the gas phase compounds at higher temperature, warmer summer conditions promote dissociation of ammonium nitrate particles, shifting the ammonium/ammonia partitioning toward the gas phase (Behera et al., 2013). In this study, the average NH₃ concentration in summer $(7.3 \pm 4.9 \,\mu\text{g m}^{-3}; n = 2181)$ was $2.4 \,\mu\text{g m}^{-3}$ or $41 \,\%$ higher than the average of other seasons. The gap between summer and winter in Shanghai was similar to New York, but generally much lower than many other cities. Taking Beijing for example, according to lanniello et al. (2010), the NH₃ concentration in summer was 460 % higher than in winter; this figure was 320 % in Xi'an between 2006 and 2007 (Cao et al., 2009). Smaller seasonal temperature differences and less agricultural activity in Shanghai could be the contributing factors.

Based on the "bottom-up" methodologies, previous emission inventories indicate that livestock feeding and N-fertilizer application contribute around 50 (48-54.9%) and 35% (33.4–40%) of the total NH₃ emissions in the Yangtze River delta region (YRD for short, including Shanghai as well as 24 cities in the provinces of Jiangsu and Zhejiang), respectively (Fu et al., 2013; Huang et al., 2011, 2012). Agricultural production is also the dominant source of NH₃ emissions in most other regions worldwide (Bouwman et al., 1997; Olivier et al., 1998; Reis et al., 2009). However, performed at an urban level, many studies in Table 2 concluded that the concentrations and evolution of ambient NH₃ in urban areas were influenced by traffic emissions. As one of the world's largest megacities, Shanghai might expect contributions of vehicle-emitted NH₃ as well. NH₃ concentrations in the atmosphere, however, are also sensitive to other important factors such as changes in temperature, wind speed or direction, and boundary layer depth; other influential factors might include local or regional NH3 emissions, dry and wet deposition, and gas-to-particle partitioning. The relative importance of such factors in controlling ambient NH₂ concentration may vary seasonally. For example, the highest and lowest daily NH₂ concentrations in Shanghai were observed on 10-11 July 2014 (23.4 \pm 6.7 μ g m⁻³) and 10–11 March 2015 (0.5 \pm 0.4 μ g m⁻³), respectively.

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For the two periods, there was no significant difference between them in terms of wind speed and planetary boundary layer height (the relative humidity data in March period was missed). Although 19.6 mm of rainfall in the July period would be expected to lower NH₃ levels, the temperature on this high concentration date (28.4 °C) was much higher than on the low concentration March date (4.7 °C). Over a longer time frame, even though rainfall in summer was around twice the amount of rainfall in other seasons, other factors such as greater NH₃ emissions at higher temperature outweigh the wet scavenging effects of rainfall yielding higher summertime NH₃ concentrations. High NH₃ concentration episodes during burning of agricultural wheat residues, indicated by a strong and synchronous rise of trace aerosols from biomass burning (e.g., K⁺ and BC), were also evident in late spring (Fig. 2b). The evolution of this pollution episode induced by biomass burning and its influence of on the air quality of Shanghai has been examined in our recent paper (Zou et al., 2015).

3.2 Effects of meteorological parameters

In the following, we will examine the (synergistic) effects of various meteorological parameters on measured NH₃ concentrations in Shanghai, because these factors may mask the effect of emissions on the measured NH₃ concentrations. Summary statistics for meteorological parameters during 3 April 2014–2 April 2015 are shown in Table 3.

Planetary boundary layer (PBL) height plays a vital role in determining the vertical dispersion of air pollutants that are emitted from the Earth surface. Decreasing height of PBL can normally hold the pollutants within the shallow surface layer so as to suppress the vertical atmospheric dilution. In many previous studies, as described above, the NH₃ concentrations in winter were much lower than those in summer. However, the NH₃ concentrations observed here in Shanghai during winter are relatively high. One may argue that weaker vertical mixing and shallow PBL layers in winter could trap NH2 and contribute to elevated concentrations. In Fig. 3a, although the simulated average PBL height in winter is the lowest during our study period, there is no significant difference among different seasons. In Fig. 3b, the average PBL height in winter is even higher

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than that in spring and summer. Therefore, a relatively high NH₃ concentration in winter at PD cannot be fully explained by the strength of vertical mixing or PBL height in this study.

Figure 4a suggests that temperature (*T*) is an important driver of the increase of NH₃ concentration in spring. No clear relationship is seen for other seasons. As the transitional period between winter and summer, springtime in Shanghai has the highest standard deviation of temperature during our study period (Table 3). Additionally, spring is known as the sowing season in South China, with the greatest application of N-containing fertilizers (mainly in the form of urea) of the year. Warming temperature tends to increase the rate of urea hydrolysis and ammonium conversion to NH₃, and therefore volatilization. For example, an increase in temperature from 7.2 to 15.6 °C can double volatilization loss when moisture content is kept the same (Ernst and Massey, 1960). For relative humidity (RH), there is no clear evidence to suggest RH as an important factor controlling the dynamics of NH₃ concentrations in any of the seasons (Fig. 4b). Figure S2 in the Supplement shows the RH and *T* dependent distributions of NH₃ concentration for each season. Given the generally poor relationship between the NH₃ concentration and *T* and RH as discussed above, NH₃ concentrations have no clear dependence on *T* and RH seasonally.

In Fig. 5a and b the distribution of hourly average wind speeds was calculated for values between 0 and $4.0\,\mathrm{m\,s}^{-1}$ (99.5% of occurrence). Figure 5a shows that there is a highly significant relationship between WS and NH₃ concentration (R^2 = 0.91, p < 0.001). The highest average NH₃ concentrations were measured under the lowest wind speeds and the lowest concentrations were measured at the highest wind speeds. There is no clear relationship between wind frequency (the number of wind occurrence) and average NH₃ concentration or WS during the study period (Fig. 5c). Figure S3 in the Supplement shows WS/WD dependence of NH₃ concentrations in different seasons. The distribution of NH₃ concentration showed an obvious concentration gradient as a function of WS. Seasonally, there are different preferential wind directions for the highest NH₃ concentration values. Generally, an overwhelming higher T

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in summer tends to greatly enhance the contribution of temperature-dependent emissions to the urban NH₃ budget from agricultural areas. And the nearby rural areas around Pudong supersite are in the direction of Southeast (Nanhui) and Northeast (Chongming) (Fig. 6d). However, it is unexpected that in Shanghai, almost all high NH₃ concentration values in summer are concentrated in the direction of South-Southwest-West (Supplement Fig. S3b), which strongly indicates that the urban area is one of the most important NH₃ emission regions in Shanghai.

Figure 6a and b shows the RH and T dependent distributions of NH $_3$ and WS for the entire study period, respectively. Although the distribution of higher NH $_3$ favors the condition of high $T(>25\,^{\circ}C)$, low RH ($<60\,\%$) and low WS ($<1.2\,\mathrm{m\,s^{-1}}$), Fig. 6a shows that there is no obvious concentration gradient as a function of RH and T (Note that in Supplement Fig. S2d, higher NH $_3$ concentrations in winter tend to occur at higher T for a given RH range of 60 to 80 %). This can be explained by the dominance of low WS (often lower than $1.2\,\mathrm{m\,s^{-1}}$) during east-southeast and west-northwest wind directions associated with intense local sources for NH $_3$ within the city (Fig. 6c). In brief, our results suggest that there are some temperature-independent and important NH $_3$ sources in the urban area of Shanghai.

3.3 NH₃ diurnal profiles and insight into sources

Hourly observations over long-term periods offer a unique opportunity to provide robust diurnal profiles for each season. Figure 7 shows the average diurnal profiles of $\rm NH_3$ and CO concentrations across seasons. Historically, CO emissions in Shanghai and its surrounding YRD region mainly came from iron and steel manufacturing and onroad vehicles, which contributed 34 and 30 % of the total, respectively in 2007 (Huang et al., 2011). Due to changing economic activity, emission sources of air pollutants in China are changing rapidly. For example, over the past several years, China has implemented a portfolio of plans to phase out its old-fashioned and small steel mills, and raise standards for industrial pollutant emissions (Chang et al., 2012). In contrast, China continuously experienced double digit growth in terms of auto sales during the

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same period, and became the world's largest automobile market since 2009 (Chang, 2014). Consequently, on-road traffic has overtaken industrial sources as the dominant source of CO emissions in Eastern China (Zhao et al., 2012). In Fig. 7, CO shows a well-marked bimodal diurnal profile, with maxima in the morning (starting at 05:00 ₅ local time) and the evening (starting at 16:00 LT), consistent with the variation of traffic flow in Shanghai (Liu et al., 2012). Therefore, CO variation can be utilized as a robust indicator of vehicle emissions. NH3 also displays a clear bimodal profile during all four seasons, similar to the CO diurnal profile, suggesting a significant influence of on-road traffic (with daily commuting) on ambient NH₃ concentrations in the urban environment

Interestingly however, NH₃ shows different degrees of positive relationship with CO as a function of season (Fig. 8b). Specifically, during summertime, NH₃ displays a significant relationship with CO ($R^2 = 0.48$, p < 0.001), while this positive relationship is not observed during the winter season, when heavy traffic volume also occurs. As discussed above, the seasonal variation of NH₃ concentration in Shanghai during our study period was quite flat. The seasonal average CO concentrations at PD were 0.71, 0.61, 0.58, and 1.1 ppmv in spring, summer, autumn, and winter, respectively. And the CO level in wintertime was higher than other seasons. Moreover, Fig. 8c suggests that for all seasons, the source region of NH₃ in Shanghai is local-dominated. However, the atmospheric lifetime of CO is much longer than that of NH₃ (typically several hours depending on meteorology) (Asman et al., 1998). PSCF analysis for CO in winter suggests important contributions from north of Shanghai (Fig. 8a), a region that does not appear as important as a NH₃ source (Fig. 8c). Consequently, elevated regional background levels of CO from long-range transport appear to yield a poorer relationship between NH₃ and CO in wintertime (Fig. 8b).

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Table S1 in the Supplement summarizes statistics of the NH₃ concentrations (µg m⁻³) measured at each sampling point in and out of the Handan tunnel, which have also 34733

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been visualized in Fig. 9b. As expected, the highest average NH₃ concentration occurred at the exit of the tunnel (T-d). Although NH₃ concentration varied temporally, throughout the two months of observations, a large spatial gradient in NH₃ concentration at near-road sites was present in every sampling event, suggesting that an intensive NH₃ source from on-road traffic (not meteorological parameters) is the leading factor in governing the variation of ambient NH₃ concentration in a road-side environment. The NH₃ concentrations in the tunnel were increased with distance from the entrance of the tunnel (T-a). The NH₃ concentration at T-d $(64.9 \pm 11.5 \,\mu g \, m^{-3})$ was over 5 times that at T-a $(12.6 \pm 3.3 \,\mu\text{g m}^{-3})$. Moreover, the lowest NH₃ concentration value obtained at T-d (47.0 μg m⁻³) was still nearly 10 μg m⁻³ or 20 % higher than the highest value of other sites (Supplement Table S1). These observations provide compelling evidence that on-road traffic is an important emission source of NH₂ in the urban atmosphere. Given that there is a significant loss of vehicle exhaust from the tunnel through an array of ventilation orifices in the middle section of the tunnel, the NH₃ concentration at T-b $(29.2 \pm 6.6 \,\mu\text{g m}^{-3})$ was close to that at T-c $(31.5 \pm 5.9 \,\mu\text{g m}^{-3})$. If we taking into account of the Physical Distance (PD; 300 m) and the NH₃ Concentration Gap (CG; $33.4 \pm 11.5 \,\mu\text{g m}^{-3}$) between T-c and T-d, the Cross Section of tunnel bore (CS; 70 m³), the average Wind Speed (WS; 5 m s⁻¹) and Traffic Flow (TF;

 $EF = \frac{CG \times CS \times WS \times 86400}{TF \times PD}$ (1)

120 000 vehiclesday⁻¹), we can obtain an approximate mileage-based NH₃ Emission

Factor (EF) of $28 \pm 5 \,\mathrm{mg\,km}^{-1}$ for a single vehicle using the following equation:

where 86 400 is the number of seconds in a day. This NH_3 emission factor was similar to that observed for the Gurbrist tunnel in Switzerland (31 \pm 4 mg km⁻¹) (Emmenegger et al., 2004) and the Caldecott tunnel in California (49 \pm 3 mg km⁻¹) (Kean et al., 2000), while much lower than that recently measured in Guangzhou, China (230 \pm 14 mg km⁻¹) (Liu et al., 2014). Based on the emission factor we developed, a population of 3.04 million vehicles (average mileage of 15 000 km yr⁻¹) in Shanghai would produce around

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1300 t NH₃ in 2014. This is very close to the "bottom-up" emission inventory in Shanghai for 2010 (1581.1 t) (Chang, 2014). Previous emission inventories in Shanghai (e.g., Huang et al., 2011 and Fu et al., 2012) made a significant underestimation of the NH₃ emissions from city areas. When compared with the NH₃ emissions from city area, the contribution of on-road traffic can reach 12% of the total NH₃ emissions in Shanghai city areas (10742 t) (Chang, 2014). Moreover, model results have shown that over half of agricultural NH₃ emissions would be deposited downwind of its source within 10 km depending on local meteorological conditions (Asman et al., 1998). Therefore, the relative contribution of NH₃ emissions from on-road traffic to urban PM pollution could be higher than the share of its mass contribution. Given that precisely estimating the EF of NH₃ from on-road traffic is beyond the scope of this paper, more research is needed to pinpoint this parameter in order to accurately quantify the amount of NH₃ emissions from vehicles.

From the tunnel exit to the open environment, the average NH $_3$ concentration at T-d (64.9 ± 11.5 µg m $^{-3}$) was 11 times more than that at O $_{310\,\mathrm{m}}$ (5.6 ± 2.5 µg m $^{-3}$), and a general negative relationship was found between distance and ambient NH $_3$ concentration. Over the total measured distance, the maximum percent decrease was observed between the sites of T-d and O $_{0\,\mathrm{m}}$ (50 m apart), indicating a rapid dispersion of vehicle-emitted NH $_3$ from the road tunnel. Still, Fig. 9c clearly shows that 64 % (48 %) of the NH $_3$ concentration we observed at the site of O $_{0\,\mathrm{m}}$ (O $_{20\,\mathrm{m}}$) can be explained by the simultaneous measurements of NH $_3$ concentration at T-d. No significant decrease in the gradients of NH $_3$ concentration was observed between the sites of O $_{150\,\mathrm{m}}$ (5.9 ± 2.5 µg m $^{-3}$; n = 6) and O $_{310\,\mathrm{m}}$ (5.6 ± 2.5 µg m $^{-3}$; n = 6), suggesting that the strongest impact of NH $_3$ emission and transport from local traffic flow on ambient NH $_3$ concentrations in the Shanghai urban area lies within 150 m distance.

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This study linked a long-term and near real-time measurement of NH₃ at one of China's flagship supersites with a vehicle source-specific campaign performed in and out of a major freeway tunnel in Shanghai. The conclusions are shown as below:

- The average NH_3 concentration (mean $\pm 1\sigma$) between April 2014 and April 2015 was $5.5 \pm 3.9 \,\mu\text{g}\,\text{m}^{-3}$. Seasonal NH₃ concentration levels varied in the following sequence: summer $(7.3 \pm 4.9 \,\mu\text{g m}^{-3}) > \text{spring} (5.1 \pm 3.8 \,\mu\text{g m}^{-3}) \approx \text{winter} (5.0 \pm 3.8 \,\mu\text{g m}^{-3}) \approx 1.0 \,\text{m/s}$ $3.4 \,\mu\text{g}\,\text{m}^{-3}$) > fall $(4.5 \pm 2.3 \,\mu\text{g}\,\text{m}^{-3})$.
- During spring, ambient NH₃ concentrations appeared to be influenced to some extent by temperature-dependent emissions, likely from agricultural activities including crop fertilization. No such relationship was apparent during other seasons. Measured NH₃ concentrations were highly dependent on wind speed, while mixing height of planetary boundary layer and relative humidity were not the main factors influencing seasonal NH₃ concentrations.
- The diurnal profile of NH₃ concentrations showed a typical bimodal cycle during four seasons, with maxima in the morning and the evening rush hours, suggesting a persistent influence of on-road traffic (with daily commuting) on ambient NH3 levels in Shanghai.
- The NH₃ concentration in the exit of the Handan tunnel $(64.9 \pm 11.5 \,\mu\text{g m}^{-3})$ was over 5 and 11 times higher than that in the tunnel entrance $(12.6 \pm 3.3 \,\mu\text{g m}^{-3})$ and the ambient air $(5.6 \pm 2.5 \,\mu\text{g m}^{-3})$, respectively, providing further compelling evidence that on-road traffic is an important NH3 source. 1300 t vehicle-emitted NH₃ in 2014 was calculated based on a mileage-based NH₃ emission factor of $28 \pm 5 \,\mathrm{mg\,km}^{-1}$ we developed.
- A negative relationship was found between the distance and ambient NH₃ concentration in our near-road gradient experiment. Up to 64 % of ambient NH₃ con-34736

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Unlike NH₃ emissions in agricultural areas, the NH₃ emissions in urban areas originate from a variety of stationary sources (industrial coal/oil/gas combustion, wastewater, landfill, compost and incineration), mobile sources and area sources (e.g., humans, green land, domestic fuel combustion). As a start, our study is far from fully elucidating the complex origins of urban NH₃ in Shanghai. Vehicle-emitted NH₃, while important, is likely not the only major source of NH₃. Additional useful investigative steps could include:

- Using isotopes as a source apportionment tool. Isotopic techniques have been proven to be useful tools for sources apportionment of gases and PM. Although the δ^{15} N values of NH₄⁺ in rainwater and aerosols have been examined (Xiao et al., 2015; Xiao and Liu, 2002; Xiao et al., 2012), atmospheric NH₃ has received much less attention. According to Felix et al. (2013), NH₃ emitted from volatilized sources has relatively low δ^{15} N values, allowing them to be distinctly differentiated from NH₃ emitted from fuel-related sources (e.g., on-road traffic) that are characterized by relatively high δ^{15} N values.
- Using chemical transport modeling (CTM) as a cost-effective analysis tool. CTM has the potential to capture the complex atmospheric behavior of NH₃. Moreover, NH₃ emission reduction targets are represented as constraints in the optimization problem, and have a major influence on overall costs of a cost-effective solution and their distribution across different sources and economic sectors. Through sensitivity analyses of specific emission source or assuming possible emission control scenarios, CTM can contribute to the setting of effective emission reduction strategies to achieve cost-effective improvements in air quality.

The Supplement related to this article is available online at doi:10.5194/acpd-15-34719-2015-supplement.

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Acknowledgements. Special thanks go to Tony Dore (CEH, UK), Yunting Fang (CAS, CN) and Fujiang Wang (Fudan, CN) for their insightful comments. This work was supported financially by the National Natural Science Foundation of China (Grant Nos. 21377029, 21277030 and 41405115). Yunhua Chang acknowledges the support of the Gao Tingyao scholarship and the Qingyue Environmental Data Center. The authors declare no competing financial interest.

References

- Aneja, V. P., Chauhan, J. P., and Walker, J. T.: Characterization of atmospheric ammonia emissions from swine waste storage and treatment lagoons, J. Geophys. Res., 105, 11535-11545. doi:10.1029/2000JD900066. 2003.
- ¹⁰ Asman, W. A., Sutton, M. A., and Schjørring, J. K.: Ammonia: emission, atmospheric transport and deposition, New Phytol., 139, 27-48, doi:10.1046/j.1469-8137.1998.00180.x, 1998.
 - Bari, A., Ferraro, V., Wilson, L. R., Luttinger, D., and Husain, L.: Measurements of gaseous HONO, HNO₃, SO₂, HCl, NH₃, particulate sulfate and PM_{2.5} in New York, NY, Atmos. Environ., 37, 2825–2835, doi:10.1016/S1352-2310(03)00199-7, 2003.
- 15 Behera, S. N., Sharma, M., Aneja, V. P., and Balasubramanian, R.: Ammonia in the atmosphere: a review on emission sources, atmospheric chemistry and deposition on terrestrial bodies, Environ. Sci. Pollut. R., 20, 8092-8131, doi:10.1007/s11356-013-2051-9, 2013.
 - Bettez, N. D., Marino, R., Howarth, R. W., and Davidson, E. A.: Roads as nitrogen deposition hot spots, Biogeochemistry, 114, 149-163, doi:10.1007/s10533-013-9847-z, 2013.
 - Bishop, G. A. and Stedman, D. H.: Reactive nitrogen species emission trends in three light-/medium-duty United States fleets, Environ. Sci. Technol., 49, 11234-11240, doi:10.1021/acs.est.5b02392, 2015.
 - Biswas, K. F., Ghauri, B. M., and Husain, L.: Gaseous and aerosol pollutants during fog and clear episodes in South Asian urban atmosphere, Atmos. Environ., 42, 7775-7785, doi:10.1016/j.atmosenv.2008.04.056, 2008.
 - Blanchard, C. L., Roth, P. M., Tanenbaum, S. J., Ziman, S. D., and Seinfeld, J. H.: The use of ambient measurements to identify which precursor species limit aerosol nitrate formation. JAPCA J. Air Waste MA, 50, 2073-2084, doi:10.1080/10473289.2000.10464239. 2000.

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Bouwman, A. F., Lee, D. S., Asman, W. A. H., Dentener, F. J., Van Der Hoek, K. W., and Olivier, J. G. J.: A global high-resolution emission inventory for ammonia, Global Biogeochem. Cy., 11, 561–587, doi:10.1029/97GB02266, 1997.

Brito, J., Rizzo, L. V., Herckes, P., Vasconcellos, P. C., Caumo, S. E. S., Fornaro, A., Ynoue, R. Y., Artaxo, P., and Andrade, M. F.: Physical–chemical characterisation of the particulate matter inside two road tunnels in the São Paulo Metropolitan Area, Atmos. Chem. Phys., 13, 12199–12213, doi:10.5194/acp-13-12199-2013, 2013.

Brook, J. R., Wiebe, A. H., Woodhouse, S. A., Audette, C. V., Dann, T. F., Callaghan, S., Piechowski, M., Dabek-Zlotorzynska, E., and Dloughy, J. F.: Temporal and spatial relationships in fine particle strong acidity, sulphate, PM₁₀, and PM_{2.5} across multiple Canadian locations, Atmos. Environ., 31, 4223–4236, doi:10.1016/S1352-2310(97)00248-3, 1997.

Broxton, P. D., Zeng, X., Sulla-Menashe, D., and Troch, P. A.: A global land cover climatology using MODIS data, J. Appl. Meteorol. Clim., 53, 1593–1605, doi:10.1175/JAMC-D-13-0270.1, 2014.

¹⁵ Cadle, S. H., Countess, R. J., and Kelly, N. A.: Nitric acid and ammonia in urban and rural locations, Atmos. Environ., 16, 2501–2506, doi:10.1016/0004-6981(82)90141-X, 1982.

Cao, J. J., Zhang, T., Chow, J. C., Watson, J. G., Wu, F., and Li, H.: Characterization of atmospheric ammonia over Xi'an, China, Aerosol Air Qual. Res., 9, 277–289, doi:10.4209/aaqr.2008.10.0043, 2009.

Cape, J. N., Tang, Y. S., Van Dijk, N., Love, L., Sutton, M. A., and Palmer, S. C. F.: Concentrations of ammonia and nitrogen dioxide at roadside verges, and their contribution to nitrogen deposition, Environ. Pollut., 132, 469–478, doi:10.1016/j.envpol.2004.05.009, 2004.

Chang, Y. H.: Non-agricultural ammonia emissions in urban China, Atmos. Chem. Phys. Discuss., 14, 8495–8531, doi:10.5194/acpd-14-8495-2014, 2014.

²⁵ Chang, Y. H., Liu, X., Dore, A. J., and Li, K.: Stemming PM_{2.5} pollution in China: re-evaluating the role of ammonia, aviation and non-exhaust road traffic emissions, Environ. Sci. Technol., 46, 13035–13036, doi:10.1021/es304806k, 2012.

Clarisse, L., Clerbaux, C., Dentener, F., Hurtmans, D., and Coheur, P. F.: Global ammonia distribution derived from infrared satellite observations, Nat. Geosci., 2, 479–483, doi:10.1038/ngeo551, 2009.

Draxler, R. R. and Rolph, G. D.: HYSPLIT (HYbrid Single-Particle Lagrangian Integrated Trajectory) Model access via NOAA ARL READY Website, Rep., NOAA Air Resources Laboratory,

- Silver Spring, MD, USA, available at: http://ready.arl.noaa.gov/HYSPLIT.php (last access: 15 October 2015), 1997.
- Durbin, T. D., Wilson, R. D., Norbeck, J. M., Miller, J. W., Huai, T., and Rhee, S. H.: Estimates of the emission rates of ammonia from light-duty vehicles using standard chassis dynamometer test cycles, Atmos. Environ., 36, 1475–1482, doi:10.1016/S1352-2310(01)00583-0, 2002.
- Emmenegger, L., Mohn, J., Sigrist, M., Marinov, D., Steinemann, U., Zumsteg, F., and Meier, M.: Measurement of ammonia emissions using various techniques in a comparative tunnel study, Int. J. Environ. Pollut., 22, 326–341, doi:10.1504/IJEP.2004.005547, 2004.
- Ernst, J. W. and Massey, H. F.: The effects of several factors on volatilization of ammonia formed from urea in the soil, Soil Sci. Soc. Am. Pro., 24, 87–90, doi:10.2136/sssaj1960.03615995002400020007x, 1960.
- Felix, J. D., Elliott, E. M., Gish, T. J., McConnell, L. L., and Shaw, S. L.: Characterizing the isotopic composition of atmospheric ammonia emission sources using passive samplers and a combined oxidation-bacterial denitrifier approach, Rapid Commun. Mass Sp., 27, 2239–2246, doi:10.1002/rcm.6679, 2013.
- Felix, J. D., Elliott, E. M., Gish, T. J., Maghirang, R., Cambal, L., and Clougherty, J.: Examining the transport of ammonia emissions across landscapes using nitrogen isotope ratios, Atmos. Environ., 95, 563–570, doi:10.1016/j.atmosenv.2014.06.061, 2014.
- Fountoukis, C., Nenes, A., Sullivan, A., Weber, R., Van Reken, T., Fischer, M., Matías, E., Moya, M., Farmer, D., and Cohen, R. C.: Thermodynamic characterization of Mexico City aerosol during MILAGRO 2006, Atmos. Chem. Phys., 9, 2141–2156, doi:10.5194/acp-9-2141-2009, 2009.
- Fraser, M. P. and Cass, G. R.: Detection of excess ammonia emissions from in-use vehicles and the implications for fine particle control, Environ. Sci. Technol., 32, 1053–1057, doi:10.1021/es970382h, 1998.
- Fu, X., Wang, S., Zhao, B., Xing, J., Cheng, Z., Liu, H., and Hao, J.: Emission inventory of primary pollutants and chemical speciation in 2010 for the Yangtze River Delta region, China, Atmos. Environ., 70, 39–50, doi:10.1016/j.atmosenv.2012.12.034, 2013.
- Gay, S. W., Schmidt, D. R., Clanton, C. J., Janni, K. A., Jacobson, L. D., and Weisberg, S.: Odor, total reduced sulfur, and ammonia emissions from animal housing facilities and manure storage units in Minnesota, Appl. Eng. Agric., 19, 347–360, doi:10.13031/2013.13663, 2003.

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- Giroux, M., Esclassan, J., Arnaud, C., and Chalé, J. J.: Analysis of levels of nitrates and derivatives of ammonia in an urban atmosphere, Sci. Total Environ., 196, 247–254, doi:10.1016/S0048-9697(96)05423-X, 1997.
- Gong, L., Lewicki, R., Griffin, R. J., Flynn, J. H., Lefer, B. L., and Tittel, F. K.: Atmospheric ammonia measurements in Houston, TX using an external-cavity quantum cascade laser-based sensor, Atmos. Chem. Phys., 11, 9721–9733, doi:10.5194/acp-11-9721-2011, 2011.
- Gong, L., Lewicki, R., Griffin, R. J., Tittel, F. K., Lonsdale, C. R., Stevens, R. G., Pierce, J., Malloy, Q., Travis, S., Bobmanuel, L., Lefer, B., and Flynn, J. H.: Role of atmospheric ammonia in particulate matter formation in Houston during summertime, Atmos. Environ., 77, 893–900, doi:10.1016/j.atmosenv.2013.04.079, 2013.
- Heald, C. L., Collett Jr., J. L., Lee, T., Benedict, K. B., Schwandner, F. M., Li, Y., Clarisse, L., Hurtmans, D. R., Van Damme, M., Clerbaux, C., Coheur, P.-F., Philip, S., Martin, R. V., and Pye, H. O. T.: Atmospheric ammonia and particulate inorganic nitrogen over the United States, Atmos. Chem. Phys., 12, 10295–10312, doi:10.5194/acp-12-10295-2012, 2012.
- Heeb, N. V., Forss, A. M., Brühlmann, S., Lüscher, R., Saxer, C. J., and Hug, P.: Three-way catalyst-induced formation of ammonia-velocity-and acceleration-dependent emission factors, Atmos. Environ., 40, 5986–5997, doi:10.1016/j.atmosenv.2005.12.035, 2006.
 - Heeb, N. V., Saxer, C. J., Forss, A. M., and Brühlmann, S.: Trends of NO⁻, NO₂⁻, and NH₃-emissions from gasoline-fueled Euro-3-to Euro-4-passenger cars, Atmos. Environ., 42, 2543–2554, doi:10.1016/j.atmosenv.2007.12.008, 2008.
 - Huai, T., Durbin, T. D., Younglove, T., Scora, G., Barth, M., and Norbeck, J. M.: Vehicle specific power approach to estimating on-road NH₃ emissions from light-duty vehicles, Environ. Sci. Technol., 39, 9595–9600, doi:10.1021/es050120c, 2005.
 - Huang, C., Chen, C. H., Li, L., Cheng, Z., Wang, H. L., Huang, H. Y., Streets, D. G., Wang, Y. J., Zhang, G. F., and Chen, Y. R.: Emission inventory of anthropogenic air pollutants and VOC species in the Yangtze River Delta region, China, Atmos. Chem. Phys., 11, 4105–4120, doi:10.5194/acp-11-4105-2011, 2011.
 - Huang, K., Zhuang, G., Lin, Y., Fu, J. S., Wang, Q., Liu, T., Zhang, R., Jiang, Y., Deng, C., Fu, Q., Hsu, N. C., and Cao, B.: Typical types and formation mechanisms of haze in an Eastern Asia megacity, Shanghai, Atmos. Chem. Phys., 12, 105–124, doi:10.5194/acp-12-105-2012, 2012.

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15, 34719–34763, 2015

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- Interactive Discussion

Huang, K., Fu, J. S., Hsu, N. C., Gao, Y., Dong, X., Tsay, S. C., and Lam, Y. F.: Impact assessment of biomass burning on air quality in Southeast and East Asia during BASE-ASIA, Atmos. Environ., 78, 291–302, doi:10.1016/j.atmosenv.2012.03.048, 2013a.

Huang, K., Zhuang, G., Lin, Y., Wang, Q., Fu, J. S., Fu, Q., Liu, T., and Deng, C.: How to improve the air quality over megacities in China: pollution characterization and source analysis in Shanghai before, during, and after the 2010 World Expo, Atmos. Chem. Phys., 13, 5927-5942, doi:10.5194/acp-13-5927-2013, 2013b.

Huang, R. J., Zhang, Y., Bozzetti, C., Ho, K. F., Cao, J. J., Han, Y., Daellenbach, K., Slowik, J., Platt, S., Canonaco, F., Zotter, P., Wolf, R., Pieber, S., Bruns, E., Crippa, M., Ciarelli, G., Piazzalunga, A., Schwilowski, M., Abbaszade, G., Schnelle-Kreis, J., Zimmermann, R., An, Z., Szidat, S., Baltensperger, U., El Haddad, L., and Prévôt, A. S.: High secondary aerosol contribution to particulate pollution during haze events in China, Nature, 514, 218-222, doi:10.1038/nature13774, 2014.

Ianniello, A., Spataro, F., Esposito, G., Allegrini, I., Rantica, E., Ancora, M. P., Hu, M., and Zhu, T.: Occurrence of gas phase ammonia in the area of Beijing (China), Atmos. Chem. Phys., 10, 9487–9503, doi:10.5194/acp-10-9487-2010, 2010.

Ju, X. T., Xing, G. X., Chen, X. P., Zhang, S. L., Zhang, L. J., Liu, X. J., Cui, Z. L., Yin, B., Christie, P., Zhu, Z. L., and Zhang, F. S.: Reducing environmental risk by improving N management in intensive Chinese agricultural systems, P. Natl. Acad. Sci. USA, 106, 3041-3046, doi:10.1073/pnas.0813417106, 2009.

Kawashima, S. and Yonemura, S.: Measuring ammonia concentration over a grassland near livestock facilities using a semiconductor ammonia sensor, Atmos. Environ., 35, 3831–3839, doi:10.1016/S1352-2310(01)00145-5, 2001.

Kean, A. J., Harley, R. A., Littlejohn, D., and Kendall, G. R.: On-road measurement of ammonia and other motor vehicle exhaust emissions, Environ. Sci. Technol., 34, 3535-3539, doi:10.1021/es991451q, 2000.

Kean, A. J., Littlejohn, D., Ban-Weiss, G. A., Harley, R. A., Kirchstetter, T. W., and Lunden, M. M.: Trends in on-road vehicle emissions of ammonia, Atmos. Environ., 43, 1565-1570, doi:10.1016/j.atmosenv.2008.09.085, 2009.

Lamarque, J.-F., Bond, T. C., Eyring, V., Granier, C., Heil, A., Klimont, Z., Lee, D., Liousse, C., Mieville, A., Owen, B., Schultz, M. G., Shindell, D., Smith, S. J., Stehfest, E., Van Aardenne, J., Cooper, O. R., Kainuma, M., Mahowald, N., McConnell, J. R., Naik, V., Riahi, K., and van Vuuren, D. P.: Historical (1850-2000) gridded anthropogenic and biomass burn-

- ing emissions of reactive gases and aerosols: methodology and application, Atmos. Chem.
- Leaderer, B. P., Naeher, L., Jankun, T., Balenger, K., Holford, T. R., Toth, C., Sullivan, J., Wolfson, J. M., and Koutrakis, P.: Indoor, outdoor, and regional summer and winter concentrations of PM₁₀, PM₂₅, SO₄²⁻, H⁺, NH₄⁺, NO₃⁻, NH₃, and nitrous acid in homes with and without kerosene space heaters, Environ. Health Persp., 107, 223-231, doi:10.1289/ehp.99107223, 1999.

Phys., 10, 7017-7039, doi:10.5194/acp-10-7017-2010, 2010.

15

- Lee, H. S., Wadden, R. A., and Scheff, P. A.: Measurement and evaluation of acid air pollutants in Chicago using an annular denuder system, Atmos. Environ., 27, 543-553, doi:10.1016/0960-1686(93)90211-G, 1993.
- Lee, H. S., Kang, C. M., Kang, B. W., and Kim, H. K.: Seasonal variations of acidic air pollutants in Seoul, South Korea, Atmos. Environ., 33, 3143-3152, doi:10.1016/S1352-2310(98)00382-3, 1999.
- Li, K., Liu, X., Song, W., Chang, Y., Hu, Y., and Tian, C.: Atmospheric nitrogen deposition at two sites in an arid environment of Central Asia, Plos One, 8, e67018, doi:10.1371/journal.pone.0067018, 2013.
- Li, M.: Engineering design of the Handan Road tunnel in the Middle Ring Road, China Munic. Eng., 1, 34–38, 2007. (in Chinese with English abstract)
- Li, Y. Q., Schwab, J. J., and Demerjian, K. L.: Measurements of ambient ammonia using a tunable diode laser absorption spectrometer: characteristics of ambient ammonia emissions in an urban area of New York City, J. Geophys. Res., 111, D10S02, doi:10.1029/2005JD006275, 2006.
- Lin, J., Nielsen, C. P., Zhao, Y., Lei, Y., Liu, Y., and McElroy, M. B.: Recent changes in particulate air pollution over China observed from space and the ground: effectiveness of emission control, Environ. Sci. Technol., 44, 7771-7776, doi:10.1021/es101094t, 2010.
- Lin, Y. C., Cheng, M. T., Ting, W. Y., and Yeh, C. R.: Characteristics of gaseous HNO₂, HNO₃, NH₃ and particulate ammonium nitrate in an urban city of Central Taiwan, Atmos. Environ., 40, 4725-4733, doi:10.1016/j.atmosenv.2006.04.037, 2006.
- Liu, T., Wang, X., Wang, B., Ding, X., Deng, W., Lü, S., and Zhang, Y.: Emission factor of ammonia (NH₃) from on-road vehicles in China: tunnel tests in urban Guangzhou, Environ. Res. Lett., 9, 064027, doi:10.1088/1748-9326/9/6/064027, 2014.

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15, 34719–34763, 2015

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- **Printer-friendly Version**
- Interactive Discussion
 - © **()**

Liu, Y., Wang, F. Xiao, Y., and Gao, S: Urban land uses and traffic "source-sink areas": evidence from GPS-enabled taxi data in Shanghai, Landscape Urban Plan., 106, 73–87, doi:10.1016/i.landurbplan.2012.02.012.

Livingston, C., Rieger, P., and Winer, A.: Ammonia emissions from a representative in-use fleet of light and medium-duty vehicles in the California South Coast Air Basin, Atmos. Environ., 43, 3326–3333, doi:10.1016/j.atmosenv.2009.04.009, 2009.

Ma, Z., Hu, X., Huang, L., Bi, J., and Liu, Y.: Estimating ground-level PM_{2.5} in China using satellite remote sensing, Environ. Sci. Technol., 48, 7436–7444, doi:10.1021/es5009399, 2014.

Ma, Z., Hu, X., Sayer, A. M., Levy, R., Zhang, Q., Xue, Y., Tong, S., Bi, J., Huang, L., and Liu, Y.: Satellite-based spatiotemporal trends in PM_{2.5} concentrations: China, 2004–2013, Environ. Health Persp., accepted, doi:10.1289/ehp.1409481, 2015.

Matsumoto, M. and Okita, T.: Long term measurements of atmospheric gaseous and aerosol species using an annular denuder system in Nara, Japan, Atmos. Environ., 32, 1419–1425, doi:10.1016/S1352-2310(97)00270-7. 1998.

McCurdy, T., Zelenka, M. P., Lawrence, P. M., Houston, R. M., and Burton, R.: Acid aerosols in the Pittsburgh Metropolitan area, Atmos. Environ., 33, 5133–5145, doi:10.1016/S1352-2310(99)00119-3, 1999.

Meng, Z. Y., Lin, W. L., Jiang, X. M., Yan, P., Wang, Y., Zhang, Y. M., Jia, X. F., and Yu, X. L.: Characteristics of atmospheric ammonia over Beijing, China, Atmos. Chem. Phys., 11, 6139–6151, doi:10.5194/acp-11-6139-2011, 2011.

20

Moeckli, M. A., Fierz, M., and Sigrist, M. W.: Emission factors for ethene and ammonia from a tunnel study with a photoacoustic trace gas detection system, Environ. Sci. Technol., 30, 2864–2867, doi:10.1021/es960152n, 1996.

Mount, G. H., Rumburg, B., Havig, J., Lamb, B., Westberg, H., Yonge, D., Johnson, K., and Kincaid, R.: Measurement of atmospheric ammonia at a dairy using differential optical absorption spectroscopy in the mid-ultraviolet, Atmos. Environ., 36, 1799–1810, doi:10.1016/S1352-2310(02)00158-9, 2002.

Nowak, J. B., Neuman, J. A., Bahreini, R., Brock, C. A., Middlebrook, A. M., Wollny, A. G., Holloway, J., Peischl, J., Ryerson, T., and Fehsenfeld, F. C.: Airborne observations of ammonia and ammonium nitrate formation over Houston, Texas, J. Geophys. Res., 115, D22304, doi:10.1029/2010JD014195, 2010.

ACPD

15, 34719–34763, 2015

The importance of vehicle emissions as a source of atmospheric ammonia

Y. H. Chang et al.

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- Interactive Discussion

- Olivier, J. G. J., Bouwman, A. F., Van der Hoek, K. W., and Berdowski, J. J. M.: Global air emission inventories for anthropogenic sources of NO_x, NH₃ and N₂O in 1990, Environ. Pollut., 102, 135-148, doi:10.1016/S0269-7491(98)80026-2, 1998.
- Pandolfi, M., Amato, F., Reche, C., Alastuey, A., Otjes, R. P., Blom, M. J., and Querol, X.: Summer ammonia measurements in a densely populated Mediterranean city, Atmos. Chem. Phys., 12, 7557-7575, doi:10.5194/acp-12-7557-2012, 2012.
- Parmar, R. S., Satsangi, G. S., Lakhani, A., Srivastava, S. S., and Prakash, S.: Simultaneous measurements of ammonia and nitric acid in ambient air at Agra (27°10'N and 78°05'E) (India), Atmos. Environ., 35, 5979–5988, doi:10.1016/S1352-2310(00)00394-0, 2001.
- Perrino, C., Catrambone, M., Di Bucchianico, A. D. M., and Allegrini, I.: Gaseous ammonia in the urban area of Rome, Italy and its relationship with traffic emissions, Atmos. Environ.. 36. 5385-5394, doi:10.1016/S1352-2310(02)00469-7, 2002.
 - Phan, N. T., Kim, K. H., Shon, Z. H., Jeon, E. C., Jung, K., and Kim, N. J.: Analvsis of ammonia variation in the urban atmosphere. Atmos. Environ., 65, 177-185. doi:10.1016/i.atmosenv.2012.10.049.2013.
 - Pierson, W. R. and Brachaczek, W. W.: Emissions of ammonia and amines from vehicles on the road, Environ. Sci. Technol., 17, 757-760, doi:10.1021/es00118a013, 1983.
 - Pierson, W. R., Gertler, A. W., Robinson, N. F., Sagebiel, J. C., Zielinska, B., Bishop, G. A., Stedman, D., Zweidinger, R., and Ray, W. D.: Real-world automotive emissions-summary of studies in the Fort McHenry and Tuscarora Mountain tunnels, Atmos. Environ., 30, 2233-2256, doi:10.1016/1352-2310(95)00276-6, 1996.
 - Pinder, R. W., Adams, P. J., and Pandis, S. N.: Ammonia emission controls as a cost-effective strategy for reducing atmospheric particulate matter in the eastern United States, Environ. Sci. Technol., 41, 380-386, doi:10.1021/es060379a, 2007.
- 25 Pio, C. A., Santos, I. M., Anacleto, T. D., Nunes, T. V., and Leal, R. M.: Particulate and gaseous air pollutant levels at the Portuguese west coast, Atmos. Environ., 25, 669-680, doi:10.1016/0960-1686(91)90065-F, 1991.
 - Polissar, A. V., Hopke, P. K., and Poirot, R. L.: Atmospheric aerosol over Vermont: chemical composition and sources, Environ. Sci. Technol., 35, 4604-4621, doi:10.1021/es0105865, 2001.
 - Pryor, S. C., Anlauf, K., Boudries, H., Hayden, K., Schiller, C. L., and Wiebe, A.: Spatial and temporal variability of high resolution reduced nitrogen concentrations in the Fraser Valley, Atmos. Environ., 38, 5825-5836, doi:10.1016/j.atmosenv.2003.12.045, 2004.

Reche, C., Viana, M., Pandolfi, M., Alastuey, A., Moreno, T., Amato, F., Ripoll, A., and Querol, X.: Urban NH₃ levels and sources in a Mediterranean environment, Atmos. Environ., 57, 153–

164, doi:10.1016/j.atmosenv.2012.04.021, 2012.

20

Reche, C., Viana, M., Karanasiou, A., Cusack, M., Alastuey, A., Artiñano, B., Revuelta, M., López-Mahía, P., Blanco-Heras, G., Rodríguez, S., Sánchez de la Campa, A., Fernández-Camacho, R., González-Castanedo, Y., Mantilla, E., Tang, S., and Querol, X.: Urban NH₃ levels and sources in six major Spanish cities, Chemosphere, 119, 769–777, doi:10.1016/j.chemosphere.2014.07.097, 2015.

Reis, S., Pinder, R. W., Zhang, M., Lijie, G., and Sutton, M. A.: Reactive nitrogen in atmospheric emission inventories, Atmos. Chem. Phys., 9, 7657–7677, doi:10.5194/acp-9-7657-2009, 2009.

Rodhe, L., Stintzing, A. R., and Steineck, S.: Ammonia emissions after application of human urine to a clay soil for barley growth, Nutr. Cycl. Agroecosys., 68, 191–198, doi:10.1023/B:FRES.0000019046.10885.ee, 2004.

Salem, A. A., Soliman, A. A., and El-Haty, I. A.: Determination of nitrogen dioxide, sulfur dioxide, ozone, and ammonia in ambient air using the passive sampling method associated with ion chromatographic and potentiometric analyses, Air Qual. Atmos. Health, 2, 133–145, doi:10.1007/s11869-009-0040-4, 2009.

Sather, M. E., Mathew, J., Nguyen, N., Lay, J., Golod, G., Vet, R., Cotie, J. Hertel, T., Aaboe, E., Callison, R., Adam, J., Keese, D., Freise, J., Hathcoat, A., Sakizzie, B., King, M., Lee, C., Oliva, S., San Miguel, G., Crow, L., and Geasland, F.: Baseline ambient gaseous ammonia concentrations in the Four Corners area and eastern Oklahoma, USA, J. Environ. Monitor., 10, 1319–1325, doi:10.1039/B807984F, 2008.

Saylor, R. D., Edgerton, E. S., Hartsell, B. E., Baumann, K., and Hansen, D. A.: Continuous gaseous and total ammonia measurements from the southeastern aerosol research and characterization (SEARCH) study, Atmos. Environ., 44, 4994–5004, doi:10.1016/j.atmosenv.2010.07.055, 2010.

Schlesinger, W. H. and Hartley, A. E.: A global budget for atmospheric NH₃, Biogeochemistry, 15, 191–211, doi:10.1007/BF00002936, 1992.

Seinfeld, J. H. and Pandis, S. N.: Atmospheric Chemistry and Physics: From Air Pollution to Climate Change, Wiley, New York, USA, 1998.

ACPD

15, 34719–34763, 2015

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 - © **()**

- Sharma, S., Pathak, H., Datta, A., Saxena, M., Saud, T., and Mandal, T.: Study on mixing ratio of atmospheric ammonia and other nitrogen components, Proc. Int. Acad. Ecol. Environ. Sci., 1, 26–35, 2011.
- Shelef, M. and McCabe, R. W.: Twenty-five years after introduction of automotive catalysts: what next?, Catal. Today, 62, 35–50, doi:10.1016/S0920-5861(00)00407-7, 2000.
- Shon, Z. H., Ghosh, S., Kim, K. H., Song, S. K., Jung, K., and Kim, N. J.: Analysis of water-soluble ions and their precursor gases over diurnal cycle, Atmos. Res., 132, 309–321, doi:10.1016/j.atmosres.2013.06.003, 2013.
- Singh, S. and Kulshrestha, U. C.: Abundance and distribution of gaseous ammonia and particulate ammonium at Delhi, India, Biogeosciences, 9, 5023–5029, doi:10.5194/bg-9-5023-2012, 2012.
- Singh, S. and Kulshrestha, U. C.: Rural vs. urban gaseous inorganic reactive nitrogen in the Indo-Gangetic plains (IGP) of India, Environ. Res. Lett., 9, 125004, doi:10.1088/1748-9326/9/12/125004. 2014.
- Skamarock, W. C., Klemp, J. B., Dudhia, J., Gill, D. O., Barker, D. M., Duda, M. G., Huang, X., Wang, W., and Powers, J. G.: A description of the advanced research WRF version 3, NCAR Tech. Note, NCAR/TN-475+STR, 8 pp., Natl. Cent. for Atmos. Res., Boulder, Colo., USA, available at: http://opensky.ucar.edu/islandora/object/technotes:500 (last access: 12 November 2015), 2008.
 - Stanier, C., Singh, A., Adamski, W., Baek, J., Caughey, M., Carmichael, G., Edgerton, E., Kenski, D., Koerber, M., Oleson, J., Rohlf, T., Lee, S. R., Riemer, N., Shaw, S., Sousan, S., and Spak, S. N.: Overview of the LADCO winter nitrate study: hourly ammonia, nitric acid and PM_{2.5} composition at an urban and rural site pair during PM_{2.5} episodes in the US Great Lakes region, Atmos. Chem. Phys., 12, 11037–11056, doi:10.5194/acp-12-11037-2012, 2012.
 - Suarez-Bertoa, R., Zardini, A., and Astorga, C.: Ammonia exhaust emissions from spark ignition vehicles over the New European driving cycle, Atmos. Environ., 97, 43–53, doi:10.1016/j.atmosenv.2014.07.050, 2014.
 - Suarez-Bertoa, R., Zardini, A. A., Lilova, V., Meyer, D., Nakatani, S., Hibel, F., Ewers, J., Clairotte, M., Hill, L., and Astorga, C.: Intercomparison of real-time tailpipe ammonia measurements from vehicles tested over the new world-harmonized light-duty vehicle test cycle (WLTC), Environ. Sci. Pollut. R., 22, 7450–7460, doi:10.1007/s11356-015-4267-3, 2015.

- ACPD
- 15, 34719-34763, 2015
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 - l∢ ≻l

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→

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- Printer-friendly Version
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 - © **()**

- Sun, K., Tao, L., Miller, D. J., Khan, M. A., and Zondlo, M. A.: On-road ammonia emissions characterized by mobile, open-path measurements, Environ. Sci. Technol., 48, 3943–3950, doi:10.1021/es4047704, 2014.
- Sutton, M., Dragosits, U., Y. Tang, and Fowler, D.: Ammonia emissions from non-agricultural sources in the UK, Atmos. Environ., 34, 855–869, doi:10.1016/S1352-2310(99)00362-3, 2000.
- Sutton, M. A., Erisman, J. W., Dentener, F., and Möller, D.: Ammonia in the environment: from ancient times to the present, Environ. Pollut., 156, 583–604, doi:10.1016/j.envpol.2008.03.013, 2008.
- Tanner, P. A.: Vehicle-related ammonia emissions in Hong Kong, Environ. Chem. Lett., 7, 37–40, doi:10.1007/s10311-007-0131-0, 2009.
- Toro, R. A., Canales, M., Flocchini, R. G., and Morales, R. G.: Urban atmospheric ammonia in Santiago city, Chile, Aerosol Air Qual. Res., 14, 33–44, doi:10.4209/aaqr.2012.07.0189, 2014.
- Van Damme, M., Clarisse, L., Heald, C. L., Hurtmans, D., Ngadi, Y., Clerbaux, C., Dolman, A. J., Erisman, J. W., and Coheur, P. F.: Global distributions, time series and error characterization of atmospheric ammonia (NH₃) from IASI satellite observations, Atmos. Chem. Phys., 14, 2905–2922, doi:10.5194/acp-14-2905-2014, 2014.
 - Vogt, E., A. Held, and Klemm, O.: Sources and concentrations of gaseous and particulate reduced nitrogen in the city of Münster (Germany), Atmos. Environ., 39, 7393–7402, doi:10.1016/i.atmosenv.2005.09.012, 2005.
 - Walker, J. T., Whitall, D. R., Robarge, W., and Paerl, H. W.: Ambient ammonia and ammonium aerosol across a region of variable ammonia emission density, Atmos. Environ., 38, 1235–1246, doi:10.1016/j.atmosenv.2003.11.027, 2004.
- Wang, J., Xie, P., Qin, M., Ling, L., Ye, C., Liu, J., and Liu, W.: Study on the measurement of ambient ammonia in urban area based on open-path DOAS technique, Spectrosc. Spect. Anal., 32, 476–480, doi:10.3964/j.issn.1000-0593(2012)02-0476-05, 2012.
 - Wang, S. X., Xing, J., Jang, C. R., Zhu, Y., Fu, J. S., and Hao, J. M.: Impact assessment of ammonia emissions on inorganic aerosols in East China using response surface modeling technique, Environ. Sci. Technol., 45, 9293–9300, doi:10.1021/es2022347, 2011.
 - Wang, S. X., Zhao, B., Cai, S. Y., Klimont, Z., Nielsen, C. P., Morikawa, T., Woo, J. H., Kim, Y., Fu, X., Xu, J. Y., Hao, J. M., and He, K. B.: Emission trends and mitigation options for air pol-

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- lutants in East Asia, Atmos. Chem. Phys., 14, 6571–6603, doi:10.5194/acp-14-6571-2014, 2014.
- Wang, W., Wang, S., Xu, J., Zhou, R., Shi, C., and Zhou, B.: Gas-phase ammonia and PM ammonium in a busy traffic area of Nanjing, China, Environ. Sci. Pollut. R., 9, 1–12, doi:10.1007/s11356-015-5397-3, 2015.
- Wang, Y., Zhang, Q. Q., He, K., Zhang, Q., and Chai, L.: Sulfate-nitrate-ammonium aerosols over China: response to 2000–2015 emission changes of sulfur dioxide, nitrogen oxides, and ammonia, Atmos. Chem. Phys., 13, 2635–2652, doi:10.5194/acp-13-2635-2013, 2013.
- Wei, S., Dai, Y., Liu, B., Zhu, A., Duan, Q., Wu, L., Ji, D., Ye, A., Yuan, H., Zhang, Q., Chen, D., Chen, M., Chu, J., Dou, Y., Guo, J., Li, H., Li, J., Liang, L., Liang, X., Liu, H., Liu, S., Miao, C., and Zhang, Y.: A China data set of soil properties for land surface modeling, J. Adv. Model. Earth Sy., 5, 212–224, doi:10.1002/jame.20026, 2013.
- Xiao, H. W., Xiao, H. Y., Long, A. M., and Wang, Y. L.: Who controls the monthly variations of NH₄⁺ nitrogen isotope composition in precipitation?, Atmos. Environ., 54, 201–206, doi:10.1016/j.atmosenv.2012.02.035, 2012.
- Xiao, H. W., Xie, L. H., Long, A. M., Ye, F., Pan, Y. P., Li, D. N., Long, Z. H., Chen, L., Xiao, H. Y., and Liu, C. Q.: Use of isotopic compositions of nitrate in TSP to identify sources and chemistry in South China Sea, Atmos. Environ., 109, 70–78, doi:10.1016/j.atmosenv.2015.03.006, 2015.
- Xiao, H. Y. and Liu, C. Q.: Sources of nitrogen and sulfur in wet deposition at Guiyang, southwest China, Atmos. Environ., 36, 5121–5130, doi:10.1016/S1352-2310(02)00649-0, 2002.
- Yamamoto, N., Nishiura, H., Honjo, T., Ishikawa, Y., and Suzuki, K.: A long-term study of atmospheric ammonia and particulate ammonium concentrations in Yokohama, Japan, Atmos. Environ., 29, 97–103, doi:10.1016/1352-2310(94)00226-B, 1995.
- Yan, X., Akimoto, H., and Ohara, T.: Estimation of nitrous oxide, nitric oxide and ammonia emissions from croplands in East, Southeast and South Asia, Glob. Change Biol., 9, 1080–1096, doi:10.1046/j.1365-2486.2003.00649.x, 2003.
 - Yao, X., Hu, Q., Zhang, L., Evans, G. J., Godri, K. J., and Ng, A. C.: Is vehicular emission a significant contributor to ammonia in the urban atmosphere?, Atmos. Environ., 80, 499–506, doi:10.1016/j.atmosenv.2013.08.028, 2013.
 - Ye, X., Ma, Z., Zhang, J., Du, H., Chen, J., Chen, H., Chen, H., Yang, X., Gao, W., and Geng, F.: Important role of ammonia on haze formation in Shanghai, Environ. Res. Lett., 6, 024019, doi:10.1088/1748-9326/6/2/024019, 2011.

- Zhao, X., Xie, Y., Xiong, Z., Yan, X., Xing, G., and Zhu, Z.: Nitrogen fate and environmental consequence in paddy soil under rice-wheat rotation in the Taihu lake region, China. Plant Soil, 319, 225–234, doi:10.1007/s11104-008-9865-0, 2009.
- Zhao, Y., Nielsen, C. P., McElroy, M. B., Zhang, L., and Zhang, J.: CO emissions in China: uncertainties and implications of improved energy efficiency and emission control, Atmos. Environ., 49, 103–113, doi:10.1016/j.atmosenv.2011.12.015, 2012.
- Zhao, Y., Zhang, J., and Nielsen, C. P.: The effects of recent control policies on trends in emissions of anthropogenic atmospheric pollutants and CO₂ in China, Atmos. Chem. Phys., 13, 487–508, doi:10.5194/acp-13-487-2013, 2013.
- Zheng, J., Ma, Y., Chen, M., Zhang, Q., Wang, L., Khalizov, A. F., Yao, L., Wang, Z., Wang, X., and Chen, L.: Measurement of atmospheric amines and ammonia using the high resolution time-of-flight chemical ionization mass spectrometry, Atmos. Environ., 102, 249–259, doi:10.1016/j.atmosenv.2014.12.002, 2015.
- Zou, Z., Shan, F., Cai, Y., Chang, Y. H., and Zhao, Q.: Evolution of biomass burning on urban air quality: a case study of spring 2014 in Shanghai, Environ. Monit. Forewarn., 4, 4–8, 2015. (in Chinese with English abstract)

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Table 1. Summary statistics of the NH_3 concentrations ($\mu g \, m^{-3}$) measured in Shanghai during 3 April 2014–2 April 2015. P10 and P90 represent the 10th and 90th concentration percentile, respectively.

	N	Mean	SD	Minimum	P10	Medium	P90	Maximum
All	8447	5.5	3.9	0.03	2.0	4.6	10.2	39.2
Spring	2204	5.1	3.8	0.03	1.7	4.1	9.6	25.1
Summer	2181	7.3	4.9	0.65	2.6	6.3	12.7	39.2
Autumn	1949	4.5	2.3	0.57	2.3	3.9	7.2	19.7
Winter	2113	5.0	3.4	0.43	1.8	4.3	9.3	30.7

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Table 2. Ambient NH₃ concentration measurements in the urban atmosphere of China and other countries/regions.

Location	Period	Methodology	Time resolution	Concentration (µm ⁻³)	Reference
East Asia					
Shanghai, CN	Apr 2014-Apr 2015	MARGA online monitor	hourly	5.5 ± 3.9	This study
Beijing, CN	Feb 2008–Jul 2010	Ogawa passive sampler	weekly	14.2 ± 10.6 (2008), 18.1 ± 13.8 (2009)	Meng et al. (2011)
Beijing, CN	Jan-Feb 2007, 8	Annular diffusion denuder	daily	5.5 ± 3.8 (winter), 25.4 ± 6.9 (summer)	lanniello et al. (2010)
Xi'an, CN	Apr 2006-Apr 2007	Ogawa passive sampler	weekly	12.9/6.4/20.3 (annual/winter/summer)	Cao et al. (2009)
Nanjing, CN	Aug-Sep 2012	HRToF-CIMS ^a	1 Hz	1.3 ± 1.8 (industrial area)	Zheng et al. (2015)
Nanjing, CN	Jul-Aug 2013	Portable NH ₃ online detector	hourly	6.7 (near road)	Wang et al. (2015)
Guangzhou, CN	Nov 2010	OP-DOAS b	2.5 min	1.6	Wang et al. (2012)
Urumgi, CN	Sep 2009-Aug 2010	Radiello passive sampler	biweekly	6.5	Li et al. (2013)
Hongkong, CN	Oct 2003-May 2006	Ogawa passive sampler	weekly	0.7 (rooftop) -7.1 (near road)	Tanner (2009)
Taichung, TW	Jan-Dec 2002	Annular diffusion denuder	12 h	8.5 ± 3.0	Lin et al. (2006)
Yohokama, JP	Jan 1987-Dec 1991	Glass flask sampling	3 h	2.5 ± 1.4 (winter), 8.7 ± 3.1 (summer)	Yamamoto et al. (1995)
Nara, JP	Jun 1994-May 1995	Annular diffusion denuder	12 h	1.7 (winter), 1.6 (summer)	Matsumoto and Okita (1998
Seoul, KP	Oct 1996-Sep 1997	Annular diffusion denuder	daily;	4.3/0.7/38.6 (annual/winter/summer)	Lee et al. (1999)
Seoul, KP	Jan-Dec 2010	MARGA online monitor	hourly	6.8 ± 3.3 (spring), 11.2 ± 3.9 (summer)	Shon et al. (2013)
Seoul, KP	Sep 2010-Aug 2011	MARGA online monitor	hourly	8.4 ± 3.3	Phan et al. (2013)
North America					
New York, US	Jan 1999-Jun 2000	Annular diffusion denuder	daily	5.0/4.1/6.1 (annual/winter/summer)	Bari et al. (2003)
New York, US	Jan-Feb 2004	TDLAS ^c	< 1 min	0.6 (winter)	Li et al. (2006)
Chicago, US	Apr 1990-Mar 1991	Annular diffusion denuder	12 h	1.6 ± 1.7	Lee et al. (1993)
Pittsburgh, US	Jul-Sep 1993	Annular diffusion denuder	daily	3.9 ± 4.4 (summer)	McCurdy et al. (1999)
Los Angeles, US	May 1988-Sep 1994	Annular diffusion denuder	12 h	8.3	Blanchard et al. (2000)
Sacramento, US	Oct 1988-Sep 1994	Annular diffusion denuder	12 h	9.5	Blanchard et al. (2000)
Santa Barbara, US	May 1988-Sep 1994	Annular diffusion denuder	12 h	2.7	Blanchard et al. (2000)
Farmington, US	Dec 2006-Dec 2007	Ogawa passive sampler	3 week	1.2 ± 0.4	Sather et al. (2008)
Clinton, US	Jan-Dec 2000	Annular diffusion denuder	12 h	2.6 (winter), 6.2 (summer)	Walker et al. (2004)
Kinston, US	Jan-Dec 2000	Annular diffusion denuder	12 h	0.5 (winter), 2.7 (summer)	Walker et al. (2004)
Morehead, US	Jan-Dec 2000	Annular diffusion denuder	12 h	0.3 (winter), 0.7 (summer)	Walker et al. (2004)
Houston, US	Aug 2010	Quantum laser spectrometer	10 min	2.3 ± 1.9 (summer)	Gong et al. (2011)
Commerce, US	Nov-Dec 1978	Customed passive sampler	two days	2.6 (winter)	Cadle et al. (1982)
Vinton, US	May-Sep 1995	Ogawa passive sampler	biweekly	1.3 ± 0.4 (summer)	Leaderer et al. (1999)
Mexico city, MX	Mar 2006	Quantum laser spectrometer	6 min	17.7 ± 11.0 (spring)	Fountoukis et al. (2009)
Hamilton,CA	1992-1994	Annular diffusion denuder	daily	4.3	Brook et al. (1997)

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Table 2. Continued.

Location	Period	Methodology	Time resolution	Concentration (µm ⁻³)	Reference
Europe					
Edinburgh, UK	Apr-May 2002	ALPHA passive sampler	bimonthly	4.8 (spring)	Cape et al. (2004)
Münster, DE	Mar-Jul 2004	AMANDA ^d	10 min	3.9 (spring-summer)	Vogt et al. (2005)
Toulouse, FR	Mar 1985-Mar 1986	Nylon filter pack method	daily	3.8 (near road) - 19.8 (residential)	Giroux et al. (1997)
Rome, IT	May 2001-Mar 2002	Annular diffusion denuder	30 min	17.2 ± 2.7 (near road)	Perrino et al. (2002)
Al-Ain, AE	Apr 2005–Apr 2006	Ogawa passive sampler	biweekly	9.7 ± 4.8	Salem et al. (2009)
Barcelona, ES	May-Sep 2011	AiRRmonia online analyzer	1 min	2.2 ± 1.0 (near road), 5.6 ± 2.1 (mixed)	Pandolfi et al. (2012)
Barcelona, ES	Jan, Jul 2010; Jan 2011	ALPHA passive sampler	biweekly	4.4 (winter), 9.5 (summer)	Reche et al. (2012)
Barcelona, ES	Jan, Jul 2010; Jan 2011	ALPHA passive sampler	biweekly	4.5 ± 2.1 (winter), 9.2 ± 6.6 (winter)	Reche et al. (2015)
Madrid, ES	Mar, Jul 2010; Jul 2011	ALPHA passive sampler	biweekly	2.3 ± 1.3 (winter), 2.6 ± 1.8 (summer)	Reche et al. (2015)
Valencia, ES	Feb-Mar, Jun 2010; Feb-Mar 2011	ALPHA passive sampler	biweekly	1.5 ± 0.9 (winter), 0.5 ± 0.4 (summer)	Reche et al. (2015)
Huelva, ES	May-Jun, Nov 2010; May-Jun 2011	ALPHA passive sampler	biweekly	2.8 ± 3.8 (winter), 1.2 ± 0.9 (summer)	Reche et al. (2015)
Aveiro, PT	Aug 1988–May 1989	Nylon filter pack method	daily	3.5 ± 1.9	Pio et al. (1991)
South America					
Santiago, CL	Apr–Jun 2008	Ogawa passive sampler	monthly	15.0 ± 3.8 (spring)	Toro et al. (2014)
Sourth Asia					
Lahore, PK	Dec 2005-Feb 2006	Annular diffusion denuder	12 h	50.1 ± 16.9	Biswas et al. (2008)
Dayalbagh, IN	Feb, Jun 1997; Feb 1998	Annular diffusion denuder	3 h	12.5 ± 2.2	Parmar et al. (2001)
Delhi, IN	Sep-Oct 2008; Sep-Oct 2009	Chemiluminescence analyzer	1 h	13.5 ± 2.5 (2008), 14.4 ± 3.7 (2009)	Sharma et al. (2011)
Delhi, IN	Apr 2010–Nov 2011	Glass flask sampling	5 h	35.0 ± 16.8	Singh and Kulshrestha (2012)
Delhi, IN	Oct 2012-Sep 2013	Glass flask sampling	8 h	40.7 ± 16.8	Singh and Kulshrestha (2014)

a High resolution time-of-flight chemical ionization mass spectrometry.

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b Open-path differential optical absorption spectroscopy.

^c Tunable diode laser absorption spectrometer.

^d Horizontal continuous-flow wet denuder.

Table 3. The average of temperature (°C), relative humidity (%), accumulated rainfall (mm) and simulated planetary boundary layer (PBL) height (m) in Shanghai (mean \pm 1 σ) during 3 Apr 2014–2 Apr 2015.

	Temperature	Relative humidity	Accumulated rainfall	Simulated PBL height
All	17.1 ± 8.2	72.4 ± 16.1	1271.5	454.0 ± 309.2
Spring	16.1 ± 6.1	63.7 ± 19.5	298.3	448.8 ± 311.9
Summer	25.7 ± 3.4	78.4 ± 12.5	550.7	460.6 ± 293.2
Autumn	19.8 ± 4.4	76.5 ± 13.0	221.4	482.6 ± 321.6
Winter	6.7 ± 3.3	67.0 ± 15.6	192.1	428.0 ± 307.4

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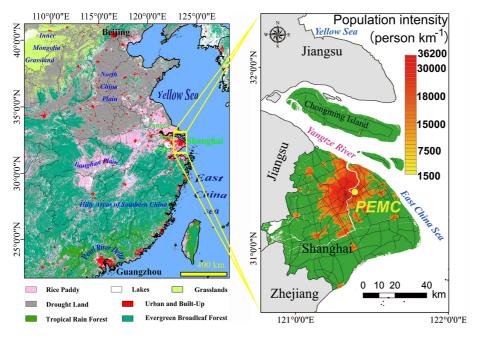


Figure 1. Aocation of the Pudong Environmental Monitoring Center (PEMC) supersite in Shanghai. The left panel shows various types of land use in eastern and southern China (adopted from, Broxton et al., 2014). The red areas and black lines in the right panel represent the urban areas and main roads in Shanghai, respectively.



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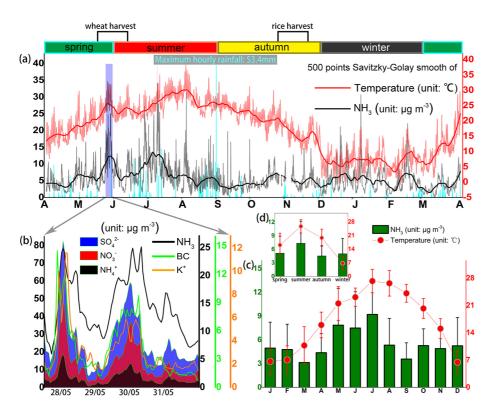


Figure 2. (a) Temporal variations of hourly NH₃ concentrations (gray) and temperature (red), along with 500-point Savitzky-Golay smoothed records in Shanghai from 3 April 2014 to 2 April 2015. Rainfall is shown in cyan. The vertical blue rectangle highlights NH₂ pollution episodes that occured during the wheat harvest season. (b) Time series of NH₃, BC, SO₄², NO₃, NH₄, and K⁺ concentrations during periods of pollution associated with biomass burning. Monthly (c) and seasonal (d) variations of NH₃ average concentrations and temperature.

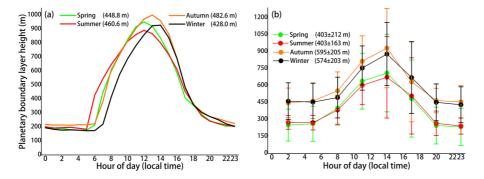


Figure 3. (a) Simulated diurnal profiles of the planetary boundary layer height in Shanghai during 3 April 2014–2 April 2015. **(b)** Daily evolution of the planetary boundary layer height (NOAA READY achieved GDAS data) in Shanghai from 12 April 2014 to 11 April 2015. The number in the legend represents the average planetary boundary layer height, by time of day, in different seasons.

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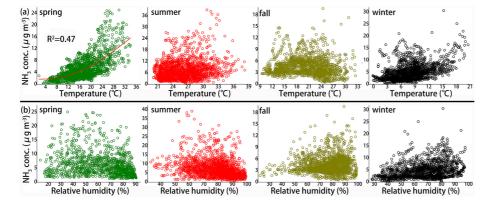


Figure 4. The relationship between hourly NH₃ concentration and hourly temperature **(a)** and hourly relative humidity **(b)** in four seasons at Pudong supersite between during 3 April 2014-2 April 2015.

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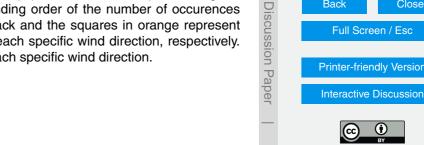


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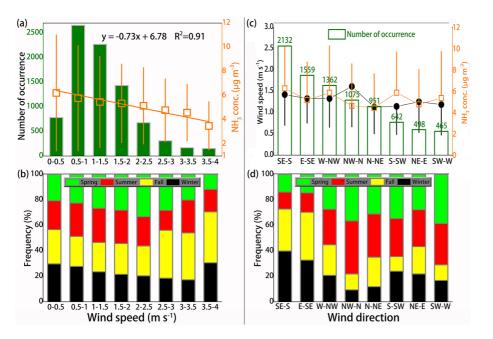


Figure 5. (a) Linear fitting of average NH₃ concentrations at different ranges of wind speed. The number of occurrences of wind (NOW) within each specific range of wind speed is shown as green columns. (b) Seasonal frequency distribution (%) of NOW at each specific range of wind speed. (c) The green boxes showing a descending order of the number of occurences of wind at different wind directions. The points in black and the squares in orange represent the average wind speed and NH₃ concentration for each specific wind direction, respectively. (d) Seasonal frequency distribution (%) of NOW at each specific wind direction.

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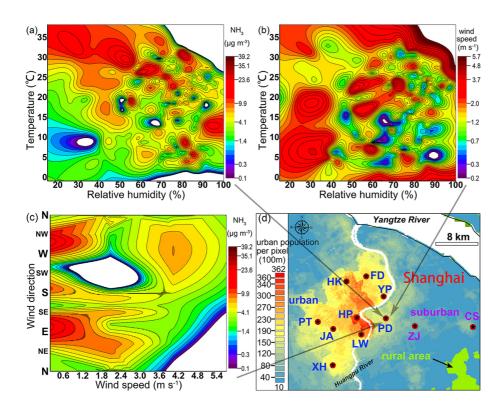


Figure 6. RH/T dependence of (a) NH₃ mass concentration and (b) WS, and (c) WS/WD dependence of NH₃ mass concentration at Pudong (PD) supersite for the year sampled. (d) The spatial distribution of environemtal monitoring network in Shanghai. FD represents Fudan university. The base map is the 2010 urban population density, derived from a newly released highresolution (100 m × 100 m per pixel) population map of China (http://www.worldpop.org.uk/).

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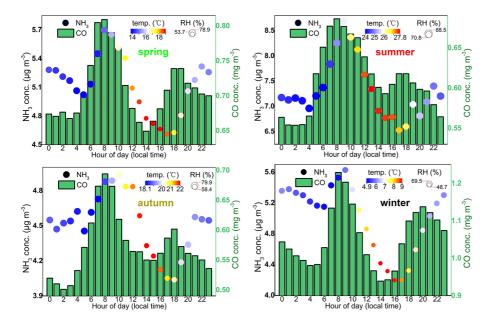


Figure 7. Seasonal diurnal profiles of NH₃ and CO concentrations in Shanghai. Color coded by hourly temperature and circle radius coded by hourly relative humidity.

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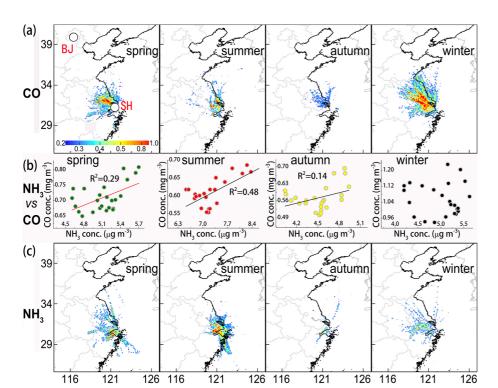


Figure 8. PSCF of CO **(a)** and NH $_3$ **(c)** during four seasons. The cities marked in each panel are Beijing (BJ) and Shanghai (SH). The color scales indicate the values of PSCF. **(b)** Relationship between hourly NH $_3$ and CO during four seasons.

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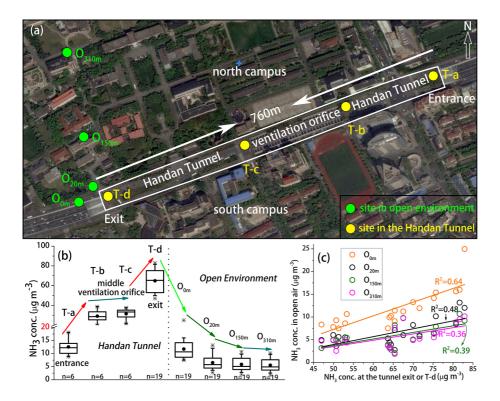


Figure 9. (a) Location of the eight sampling points in (labeled in yellow; inside the tunnel from the entrance to the exit) and out (labeled in green; varying in distance from the tunnel) of the Handan tunnel where atmospheric concentrations of NH_3 were measured using fritted glass bubblers. The campus of Fudan University was separated into north and south parts by the tunnel. **(b)** Box-whisker plots of the NH_3 concentration sampled at each site, setting 20 as the breaking point of y axis. The box boundaries represent the 25th and 75th percentile, the horizontal line is the median, and the whiskers mark the 10th and 90th percentiles. **(c)** Relationship between the NH_3 concentration at T-d (the exit of the Handan tunnel) and the other four sites varying in distance from the Handan road in the open environment.

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