

We would like to thank reviewer 2 for their extensive comments. We have made an initial statement within the general comments as to how we have now changed the model in this study, which answers a few of the subsequent points. We also try to address all of the specific points below, along with how we have changed the manuscript.

Referee #2 comments:

General comments:

The Authors present an observationally-constrained model analysis of the unknown daytime HONO source. Measurements of HONO from an urban environment during the ClearLo experiment are presented, showing levels similar to those observed in other urban environments. The Authors use these observations in addition to many supporting measurements to constrain the MCM to interrogate the potential source(s) of daytime HONO at this location. In particular, the Authors provide interesting commentary and analysis on the limitations of modeling this daytime source using the simple PSS approach instead of a photochemical box model such as the MCM, consistent with previous reports for urban environments (Lee et al., 2013). Mechanisms for the photolysis of *o*-nitrophenols, HO<sub>2</sub>+NO<sub>2</sub> and heterogeneous conversion of NO<sub>2</sub> to HONO on the ground and on aerosols have been added to the model to more fully represent the current state of knowledge in HONO production. The correlational analysis of a variety of collocated measurements with the modeled unknown daytime HONO source provides discussion consistent with previous reports in the literature, with a strong relationship observed with  $j(\text{NO}_2) \times \text{NO}_2$ . The Authors then use the model to assess the impact of using observations versus model-generated HONO on the oxidative capacity of the atmosphere, finding that major discrepancies can arise.

One issue with the paper is the discussion is focused mainly on a single HONO source at the ground surface dependent on the conversion of NO<sub>2</sub>. This is done despite the clear evidence in the literature over the past 5 years that multiple mechanisms underpinning the HONO daytime source are likely at work and are variable depending on the observation location (e.g. HNO<sub>3</sub> or particle nitrate photolysis, physisorption of HONO at mineral interfaces, soil nitrite partitioning from soil pore water, acid displacement of surface nitrite produced from deposited HONO, and microbial production of nitrite followed by soil emission of HONO). These mechanisms are all expected to have different diurnal trends and are not dependent on NO<sub>2</sub> for the release of HONO. This suggests that correlational analyses, such as that presented, are biased from being able to identify phenomena which may have inconsistent temporal variability over the course of a day and between days.

The Authors clearly understand this limitation based on their discussion surrounding the inability to unequivocally confirm surface photoenhanced conversion of NO<sub>2</sub> on sensitized organics with such an approach. Yet, the mechanism assigned to be the dominant daytime HONO source in discussion and in their concluding statements is the 'photosensitized heterogeneous conversion of NO<sub>2</sub> on organic substrates discovered in laboratory studies', which is decidedly at odds with the current state of knowledge. Further, many variables in the correlational analysis that return moderate to strong correlations with the unknown daytime HONO source are not discussed in light of supporting mechanisms where they exist. This major issue certainly warrants discussion and, potentially, inclusion to a reasonable extent in the model analyses as these non-NO<sub>2</sub> mechanisms are where modern lab and field measurements indicate significant daytime HONO may be generated.

In addition, the impact of vertical structure in radical reservoirs on the oxidative capacity of the atmosphere have certainly been presented in the literature, including using the MCM to assess the impact of HONO on OH levels. The lack of a 1D vertical transport component to this analysis, coupled to the use of HONO measurements made presumably within 10 m of the ground surface, mean that the impact on oxidative chemistry is biased by the proximity of the

HONO measurements to the ground surface and do not apply throughout the daytime boundary layer.

#### Response to general comments

We are aware of also other HONO sources proposed for certain atmospheric conditions, however, our experimental data do not confirm most of them for the urban conditions in London (see discussion below). In addition, most of the sources listed by the referee were already cited in the introduction. As already stated in the text, we have not considered soil emissions here in detail, since the different postulated sources are a) still speculative b) depend on many uncertain variables (soil pH, bacterial activity, soil humidity, etc.) and c) most probably have a very minor contribution under our highly urban conditions (low soil coverage), even if the conditions would be optimal for that source (and they are often not, see e.g. Oswald et al., 2015, ACP, 15, 799). We have updated the reason for not including this in the manuscript.

The photolysis of  $\text{HNO}_3$  using an upper limit deposition velocity and the high photolysis frequency based on several lab studies is considered in our model.

Further, we have reduced the yield of HONO from  $\text{HO}_2 \times \text{H}_2\text{O} + \text{NO}_2$  to 3% in light of recent work by Ye et al. (2015) showing that this reaction is not as important as had previously been postulated.

We have reduced the effective boundary layer height in the model as we have estimated that for a HONO lifetime of 15 minutes HONO will on average only reach ~ 75 m height over ground (1/e). This “effective mixing height” is now considered for ground surface sources of HONO and also for its deposition losses.

Besides these modifications, we disagree with the statement that the photosensitized conversion of  $\text{NO}_2$  “decidedly at odds with the current state of knowledge” for the following reasons:

- a) There are several lab studies on different organic substrates now available confirming the first studies by George et al. (2005) and Stemmler et al. (2006). We feel that they are certainly important sources in the atmosphere, due to their fast uptake kinetics.
- b) There are now several field studies available (including flux measurements, and detailed budget analysis studies) where similar findings (HONO source correlates with  $\text{NO}_2 \times$  radiation, see below) confirm our proposed major source mechanism.
- c) Results from recent papers mentioned below by the referee in which other sources were proposed based on field measurements are still under discussion. For example the results from the study of Pusede et al. (see below) can be easily explained by the well-known non-linear HONO formation by  $\text{NO}_2$  conversion on different substrates. Laboratory studies show that  $\text{HONO}/\text{NO}_x$  is higher at low  $\text{NO}_x$  levels (Langmuir-Hinshelwood type kinetics). In addition, many field studies also confirm these lab observation ( $\text{HONO}/\text{NO}_x(\text{rural})$ : typically 10-20 %, much higher compared to  $\text{HONO}/\text{NO}_x(\text{urban})$ : only ca. 3-5 %). These observations are however not in contrast with  $\text{NO}_2$  being a HONO precursor, for further details on other proposed reactions, see discussion below.

We have now added a parameterisation of light induced conversion of  $\text{NO}_2$  to HONO on aerosol and ground sources in our model and carried out a sensitivity study into their effects on the ‘missing’ HONO source (section 4.2 of the revised manuscript).

In addition, also in contrast to the statement by the referee we have discussed some other correlations besides that with  $\text{NO}_2 \times J(\text{NO}_2)$ , (e.g.  $k(\text{OH})$ ), However, we have added a few more discussions on some other correlations in the revised manuscript (section 4.2).

The missing 1D vertical model analysis would be indeed a preferable approach when ground surface processes are studied and when the whole boundary layer should be considered. In contrast, the results presented here describe only the radical budget at the measurements site (no vertical resolved measurement data available in the present study). This will be further clarified in the revised manuscript.

We now respond to each point in turn with details of how we have updated the manuscript.

Overall, a number of major revisions should be made prior to this manuscript being considered for publication in Atmospheric Chemistry and Physics.

Major comments:

1. Surface processes considered in the model and discussion are not comprehensive and should be updated.

Firstly, Sörgel and coauthors have demonstrated on a number of occasions that surface conversion of NO<sub>2</sub> to HONO on photoexcited organics can only account for a fraction of observed daytime HONO in environments where humic-like substances are modelled to cover the entire ground surface (Sörgel et al., 2011a, 2015). In built environments such as London, other surface processes (e.g. nitrate photolysis in aqueous solution (Scharko et al., 2014), metal/mineral surface sorption (Donaldson et al., 2014), reduction of nitrate on organic aerosols (Rutter et al., 2014; Ziemba et al., 2010), photolysis of nitrate in urban grime (Baergen and Donaldson, 2013) and acid displacement of nitrite (VandenBoer et al., 2015)) should be considered plausible and included in the model if possible. These mechanisms and their potential importance must be presented in the discussion even if they cannot be explicitly represented in the model as they provide much needed context.

We thank the reviewer for the comprehensive listing of HONO sources, however we did not intend this paper to be a review concerning all HONO source studies related to any atmospheric conditions, but mainly only major studies related to the urban environment in this study were referred to in the introduction. However, we will add some more postulated HONO sources to the introduction in the revised manuscript.

We do not say here that the photosensitized conversion only takes place on humic like substrates. Other organic substrates like organic grime typically prevailing in urban environments may be important. However, the exact identification of the organics adsorbed on the urban surfaces is out of the scope of the present study. We just say, the source correlates with J(NO<sub>2</sub>)xNO<sub>2</sub> (along with other things) and postulate the photosensitized conversion of NO<sub>2</sub> on a generic organic substrate (based on the only available known lab studies explaining that observation). In addition, in Sörgel et al. 2011, it was only shown that the results presented by Stemmler et al. 2007 on an artificial humic acid are not able to describe their field observation. The heterogeneous NO<sub>2</sub> uptake kinetics and HONO yields of real urban organic substrates are not known and maybe much faster compared to the artificial surfaces studied in the lab. Detailed lab studies on real surfaces collected from the surrounding of the field site in London would be necessary, which is again out of the scope of our study.

In addition, aqueous solutions (Scharko et al., 2014) in which HONO yields from nitrate photolysis may be enhanced by organics will be not important for the present field site, since there are no aqueous surfaces in the surrounding. Besides, the low photolysis frequency of aqueous nitrate (ca.  $4 \times 10^{-7} \text{ s}^{-1}$ , 0° SZA) was not enhanced in Scharko et al. (2014) making that source less important independent on any HONO yield. Finally, the much faster photolysis of HNO<sub>3</sub> adsorbed on surfaces ( $10^{-5} \text{ s}^{-1}$ ) with a 100 % upper limit HONO yield is already included in our model. However even this fast source is contradicted by its small contribution in the

model and by the correlation analysis (correlation of the missing HONO source with  $\text{HNO}_3(\text{ads})$  or  $\text{HNO}_3(\text{ads}) \times \text{J}(\text{NO}_2)$  is weaker compared to  $\text{NO}_2 \times \text{J}(\text{NO}_2)$ ).

In the study of Donaldson et al. (2014), only HONO adsorption/desorption on soil substrates as a function of the pH is studied, independent on the HONO source active. Thus, we feel these results are not in contrast to our proposed mechanism. In addition, since the pH and composition of the surrounding surfaces are not known, any parameterization of these pH-dependent physical soil processes would be completely speculative and cannot be considered here.

In contrast to the statement by the referee, in the study of Rutter et al. (2014),  $\text{HNO}_3$  reduction on organic aerosols was explicitly excluded and a gas phase reaction was proposed. However, since the conditions of that lab study were far away from any atmospheric relevant situation (reaction in the presence of a saturated steam of a high molecular motor oil, ca. 200 ppb), we have not considered this source for the analysis of our field study. In addition, this is a dark reaction, while we have mainly considered the more important daytime HONO chemistry in the present manuscript caused by the discrepancy between known HONO sources and measurements only during daytime.

In the study of Ziemba et al. (2010) indeed a conversion of  $\text{HNO}_3$  on organic aerosols was proposed based solely on field observations (which actually could be alternatively explained by air mass changes). However, HONO formation was only observed in the dark in that study, which is out of the scope of the present study (see above). In addition the completely absent correlation of the missing HONO source with aerosol nitrate (0.0006) also does not support this mechanism.

The lab study by Baergen and Donaldson is on the renoxification by  $\text{HNO}_3$  on organic grime surfaces, in which however HONO was not detected. Thus, we might introduce that as a loss process for surface  $\text{HNO}_3$ , but not as a HONO source (HONO yield would be completely speculative). In addition, this source is not supported by our analysis for London (correlation of the missing HONO source with  $\text{HNO}_3(\text{ads})$  or  $\text{HNO}_3(\text{ads}) \times \text{J}(\text{NO}_2)$  is much weaker compared to  $\text{NO}_2 \times \text{J}(\text{NO}_2)$ ) even if HONO would be a product in this reaction.

In VandenBoer et al. (2015) a so called “acid displacement mechanism” is presented, which is based on the more than 100 years old inorganic chemistry rule that a “weak acid (here HONO) is displaced by a strong acid (e.g.  $\text{HNO}_3$ )”. They studied this for soil and similar substrates and also discuss the source for urban surfaces like concrete, which could be of importance for the field conditions in London. However, the “acid displacement efficiencies” were found to be highly substrate dependent (average 9% of adsorbed HONO was displaced) and accordingly, this source would be highly uncertain for London. Also, true “acid displacement efficiencies” in the real atmosphere will be by definition much lower than those determined in the clean lab experiments of VandenBoer et al. since nitrite ( $\text{NO}_2^-$ ) is a very unstable salt, which is a) oxidized by any surface oxidant ( $\text{O}_2$ ,  $\text{O}_3$ ,  $\text{H}_2\text{O}_2$ ,  $\text{OH}$ ) and b) photolyzes to  $\text{NO}$  in the daytime atmosphere. These loss processes were not considered in VandenBoer et al. leading to expected “acid displacement efficiencies” much lower than 9% in the real atmosphere. VandenBoer et al. also calculated theoretical HONO fluxes based on their mechanism and compared these results with measured HONO fluxes during the CalNex campaign (Ren et al., 2011). While the magnitudes of the theoretical fluxes (which will likely be lower in reality - see above) were on the same order like the measured fluxes, the diurnal behaviour of the two fluxes was completely different (see their Fig. 4 c). Reasons for this are:

a) In Ren et al. the HONO flux correlated perfectly with  $\text{NO}_2 \times$  radiation (in excellent agreement with our results), leading to an asymmetric shape of the flux with higher values in the morning compared to the afternoon (higher  $\text{NO}_2$  in the morning).

b) In contrast, calculated fluxes by the “acid displacement mechanism” will maximize in the afternoon caused by the delayed formation of  $\text{HNO}_3$  by  $\text{NO}_2 + \text{OH}$  during daytime ( $\text{HNO}_3$  is highest in the afternoon...) and subsequent acid deposition.

Thus, the different temporal shapes of measured and speculated fluxes clearly demonstrate that the “acid displacement mechanism” had no significant impact on HONO formation during CalNex and is also not supported by the experimental data of the present study. Here, the correlation of the missing HONO source with  $\text{HNO}_3(\text{ads})$  is much weaker (0.096) compared to  $\text{NO}_2 \times \text{J}(\text{NO}_2)$  (0.696).

Because of the high uncertainty of the “acid displacement efficiencies” and the contradiction of the propose mechanism to field data we have not included this mechanism in our model. The recent paper by VandenBoer et al. (2015) will be added to the references cited in the revised manuscript.

We have now added brief discussions on these other potential HONO sources to the manuscript (section 4.2).

Secondly, the implemented mechanism for the photolysis of surface nitrate used is for leaf surfaces using rates that have not formally been published. There are a number of recent literature reports that probe this mechanism specifically for surfaces more representative of urban environments, along with rates, and the analysis and discussion should be modified to reflect the current state of knowledge (Baergen and Donaldson, 2013; Ma et al., 2013; Nanayakkara et al., 2014; Scharko et al., 2014). Assumption of 100 % HONO yield is also not consistent with reports of product ratios in this literature and the citations within.

In contrast to the statement of the referee, the values which we used for the photolysis of adsorbed  $\text{HNO}_3$  in our model (daytime maximum used is  $1.2 \times 10^{-5} \text{ s}^{-1}$ ) is not based on the unpublished lab results for leaf surfaces from the group of Xianliang Zhou, but is based on several of their former studies in which this photolysis frequency was determined on glass surfaces (e.g. Zhou et al., 2003). Using this photolysis frequency, atmospheric HONO levels (Zhou et al. 2003) and HONO fluxes (Zhou et al., 2011) could be explained for rural environments.

In contrast, a much higher photolysis frequency ( $10^{-3} \text{ s}^{-1}$ ) of  $\text{HNO}_3$  adsorbed on urban grime was determined in Baergen and Donaldson (2013). However, they not measured HONO in their study (see above). Also, based on calculations by Zhou et al. (2003, 2011) in which good agreement with measurements was obtained for  $\text{J}(\text{HNO}_3 \Rightarrow \text{HONO}) = 10^{-5} \text{ s}^{-1}$ , it is expected that such fast photolysis would strongly overestimated HONO near to the ground surface. Thus, we not include this highly uncertain HONO source in the model.

The paper of Ma et al. is a review, which we do not feel should be the basis of a photolysis frequency that we use in our model.

In Nanayakkara et al. (2014) no photolysis frequencies of  $\text{HNO}_3$  are specified and thus, can also not be considered here.

In the study of Scharko et al. (2014) the photolysis was studied in the liquid phase. Besides the fact that there are no significant liquid surfaces near the field site, the small photolysis frequencies of nitrate in the liquid phase ( $4 \times 10^{-7} \text{ s}^{-1}$  at  $0^\circ$  SZA) were considered there, too low to be of importance even if the HONO yield was increased by the addition of organics. Thus, we also not considered this source.

And finally, in the studies of Zhou et al. HONO was a major product in the  $\text{HNO}_3$  photolysis on glass surfaces and the photolysis frequency used ( $\text{J}(\text{HNO}_3 \Rightarrow \text{HONO})$ ) reflects only the HONO formation (here the yield is 100 % per definition). Any lower yield would even reduce the significance of this reaction. In the manuscript we already stated that the 100 % yield was used as upper limit and even with this upper limit the contribution was found to be minor.

We have made it clearer in the manuscript what photolysis frequency was used and that our model will be an upper limit for this source.

2. Model is constrained by or compared to HONO measured at an unspecified height near the ground surface, but applied throughout the depth of the PBL.

The model simulations of the unknown daytime source and nighttime production are using HONO measured from a height not presented in the manuscript. Presumably this measurement was made within 10 m of the ground surface? Numerous measurements demonstrate that near-surface vertical structure in HONO can be significant at night and during the day (Oswald et al., 2015; Stutz et al., 2002; Villena et al., 2011; Wong et al., 2012; Young et al., 2012; Zhang et al., 2009) and that a model using a near-surface value distributed throughout the PBL or into a stable nocturnal boundary layer produces results inconsistent with observations (Kim et al., 2014; Sörgel et al., 2015; Vandenboer et al., 2013; Wong et al., 2013). Thus, some of the discrepancy between the model and measurements, particularly in the early morning when thermal inversions can persist, could be ascribed to biases from vertical stratification in HONO.

The influence of vertical structure in radical reservoirs has also been demonstrated to have similar impact on collocated production of atmospheric oxidants (Young et al., 2012). In particular, the MCM has previously been shown to underestimate HONO contributions to OH production when the observation heights have not been confined to the same atmospheric layer (Kim et al., 2014).

This is indeed a good point and a 1D model would be clearly preferable when vertical transport is considered. Thus, our model results only represent the contribution of different HONO sources at the measurement height of all instruments (OH, HONO, NO<sub>x</sub>) of ca. 5 m (now stated in the manuscript in section 2.1) and should not be used for the whole boundary layer. This is now clarified in the revised manuscript (section 4.1). However, since we have no information of the vertical structure of important trace species and of the vertical mixing for the present field campaign, 1D model calculations are out of the scope of the present study.

3. HONO/NO<sub>x</sub> has been demonstrated in numerous recent studies to be a questionable metric for identifying the activity of daytime HONO sources due to HONO production not being dependent on NO<sub>2</sub>.

Examples include: nitrate photolysis in aqueous solution (Scharko et al., 2014), on urban grime (Baergen and Donaldson, 2013), and on solid/frozen substrates (Anastasio and Liang, 2009; Honrath et al., 2002; Zatko et al., 2015), acid displacement (VandenBoer et al., 2015), mineral/metal sorption (Donaldson et al., 2014), soil nitrite pore water partitioning (Su et al., 2011), microbial production and emission (Maljanen et al., 2013; Oswald et al., 2013), and through weekend-weekday analyses (Pusede et al., 2015)).

Urban environments also pose a reasonable possibility that a significant loss of NO<sub>2</sub> takes place in the formation of NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub>, followed by reactive uptake and loss of these compounds to aerosol surfaces at night.

How do the Authors justify that the HONO/NO<sub>x</sub> analysis used represents a period of HONO production given that so much published evidence contradicts such an approach?

Although the HONO/NO<sub>x</sub> ratio is indeed depending on many variables (e.g. WS, BLH, general pollution level: urban/rural/remote) it can nicely indicate the daytime HONO production for a single field site (see maxima in Fig. 2). In addition, by the increase of the HONO/NO<sub>x</sub> ratio during all the night (until morning when NO<sub>x</sub> emissions and photolysis start) night-time formation by NO<sub>2</sub> conversion can also be nicely demonstrated (see Fig 2 and also the discussion on Fig. 5 in Kleffmann et al., 2002). Thus, we disagree with the statement by the referee that HONO production is not dependant on NO<sub>2</sub>. The NO<sub>2</sub> and irradiance dependence of the daytime HONO formation was for example confirmed by the above cited flux measurements by Ren et al. (2011). Flux measurements over irradiated surfaces are the most direct method to prove a surface source mechanism in the atmosphere. These results were also confirmed by recent flux measurements over soil surfaces of one of the co-authors here (Kleffmann) in the German/French PHOTONA project (manuscript in preparation). A light and NO<sub>2</sub> dependent HONO formation was also proposed in many other field studies from urban to remote conditions (e.g. Wong et al., 2012; Sörgel et al., 2011; Villena et al., 2011) and should definitely be considered. We do not say that other mechanisms as identified by lab studies are

not important under certain conditions, but the experimental data of the present study do not confirm most of them for the field conditions in London.

Short comments to the cited references in the reviewer comment:

Scharko et al. (2014): aqueous nitrate photolysis, not important here (see above).

Baergen and Donaldson (2013): no HONO detected, not confirmed as a major source in London based on the correlation analysis (see above).

References to the frozen samples: We had no snow or ice during the field campaigns in London. Besides that, even over polar snow surfaces, the daytime source of HONO nicely correlated with  $\text{NO}_2 \times J(\text{NO}_2)$  in Villena et al., 2011.

VandenBoer et al. (2015): acid displacement not confirmed by present and other field data, see detailed answer above.

Donaldson et al. (2014): see above, not in contradiction with the present study.

References to the soil mechanisms: not considered here (urban field site).

Pusede et al. (2015): see above - results are not in contradiction with the proposed photosensitized  $\text{NO}_2$  conversion.

Finally, with respect to the losses of  $\text{NO}_x$  and its impact on the HONO/ $\text{NO}_x$  ratio, the losses of HONO by photolysis are much faster than the chemical losses of  $\text{NO}_x$ . Thus, this ratio is not expected to be significantly affected by  $\text{NO}_x$  losses during daytime. For the nighttime this certainly has to be considered, but the night-time chemistry was not the focus of the present study.

Thus, for these reasons we believe our discussion using the HONO /  $\text{NO}_x$  ratio should remain in the manuscript.

4. Given that many hypothesized daytime HONO sources are present in the literature, the Authors' decision in making the assignment of a single daytime HONO source as the only one that matters based on a correlational analysis is questionable. The choosing of the photoexcited organic reduction of  $\text{NO}_2$  mechanism as the dominant daytime HONO source is biased by the chosen analysis.

The correlation approach taken by the Authors and others (e.g. (Michoud et al., 2014)) between HONO and other co-located measurements relies on the assumption that there is a single source responsible for the majority of the unknown daytime HONO source. This is inconsistent with the literature where up to six separate mechanisms (photoenhanced  $\text{NO}_2$ , nitrate photolysis, acid displacement,  $\text{NO}_2+\text{HO}_2$ , soil partitioning, microbial production) have been presented that may account for more than 10 % of the unknown daytime HONO source. The Authors should be clear in that their approach is looking for the source(s) that have HONO production terms that most closely match the sum of the mechanisms and may allow for a tentative identification of mechanisms that have a more important role to play in this observational dataset. A correlational analysis throughout the daytime will only identify HONO production mechanisms that have a consistent diel cycle, such as the photoenhanced conversion of  $\text{NO}_2$  on organics.

Given the number of surface processes that are independent of  $\text{NO}_2$  in their production of HONO, it seems plausible that multiple mechanisms may be contributing to the HONO daytime source, but at different times of day and with differences between days (e.g. microbial activity, and evaporation of surface or soil water) and the discussion of this work should more appropriately reflect this.

Again, we do not say that other mechanisms than the photosensitized conversion of  $\text{NO}_2$  could not be important under certain conditions (e.g.  $\text{HNO}_3$  photolysis under rural conditions, see studies by Zhou et al.) but are expected to be of minor importance for the field site in London by the following reasons.

a) nitrate photolysis: This source is included in the model with upper limit kinetics, but is not significant. In addition, much lower correlations of the missing HONO source with  $\text{HNO}_3$  and  $\text{HNO}_3 \times$  radiation compared to analogue correlations with  $\text{NO}_2$  are observed.

b) Acid displacement: see long discussion above; completely different diurnal profile expected for that source compared to the experimental data. In addition, much lower acid displacement efficiencies are expected in the atmosphere compared to the lab results caused by additional nitrite losses.

c)  $\text{NO}_2 + \text{HO}_2$ : Reaction is included in the model. In addition to that, it turned out that even this source (which was speculated solely based on field data, not confirmed by any lab experiments) is not of importance in the atmosphere (Ye et al., 2015) and that HONO yields are much lower (<3 %) compared to the unity yield proposed in Li et al. (2014). These recent findings will be implemented in the revised manuscript, making this source completely unimportant for the present field site.

d) Microbial production: will be not important for the present field site (minor soil coverage). In conclusion, we feel that our postulated source mechanism is the most reasonable one for the present field site and is confirmed by the correlation analysis. We have added to our discussion the above mechanisms along with reasons for not including them in our analysis (section 4.2).

The most thorough investigation of the unknown daytime HONO source dependence on light showed that total irradiance is a better predictor than  $j(\text{NO}_2)$  or other photolysis rates (Wong et al., 2012, 2013). Why have the Authors used  $j(\text{NO}_2)$  as their proxy in the correlational investigation? There are also other multiple or single parameter terms with significant correlation coefficients in Table 1 that are not discussed (e.g.  $\text{OH} \times \text{NO}_2$ , temperature) which would be consistent with other proposed mechanisms (i.e. production, deposition, and photolysis of  $\text{HNO}_3$  or displacement of HONO from a reservoir at the surface; soil emissions by bacterial processes or temperature-driven partitioning from surface water films after nocturnal deposition). Other strong correlation coefficients (e.g.  $j(\text{NO}_2) \times \text{NH}_4^+$ ) are also not discussed and demonstrate that such relationships may be spurious or that unexpected mechanisms (Kebede et al., 2013) may be identified in urban environments.

We agree with the referee, that the two studies by Wong et al. are very nice pieces of work, especially since the vertical structure of the atmosphere was considered (1D approach). Unfortunately, this was not possible in the present study (see above).

In addition, the irradiance was used in Wong et al. instead of  $J(\text{NO}_2)$  (the latter is a measure for the actinic flux) to parameterize the photosensitized  $\text{NO}_2$  conversion. However, we used here  $J(\text{NO}_2)$  for two reasons:

a) there were simply no UV-irradiance measurements available during the campaign and any conversion of actinic fluxes to the irradiance are highly uncertain, especially in the case of cloud coverage;

b) when considering the photochemistry on surfaces, only for completely horizontal, flat surfaces, the irradiance is a correct measure to describe the photons flux densities. However, for the highly urban situation in London, with surfaces orientated in all upwards directions (vertical walls, tilted roof tops, etc.) on which photosensitized HONO formation is expected, the  $180^\circ$  measured actinic flux is still considered as a reasonable measure for parameterization. In addition, the long-wavelength UV  $J(\text{NO}_2)$  was used instead of e.g.  $J(\text{O}^1\text{D})$ , since this was shown to be a good measure to describe the photosensitized conversion of  $\text{NO}_2$  on organic substrates in lab experiments (Stemmler et al., 2007) in agreement with field studies, see e.g. Elshorbany et al. (2009).

In the revised manuscript we have added further discussions on other (weaker) correlations observed, e.g. with temperature and OH. However, we do not expect significant contribution of the photocatalytic conversion of  $\text{NH}_3$  into HONO on  $\text{TiO}_2$  containing urban surfaces (Kebede et al., 2013), caused by the still very limited use of this air remediation technique in the urban atmosphere.



5. Referencing throughout the introduction and results and discussion should provide a more comprehensive survey of the recent literature. Citations provided do not represent first, best or most recent examples in many cases (see references provided above and in the following minor comments).

We have added some more references of importance for the present urban field and model study in the introduction and discussion to the revised manuscript. However, this is not a review on all postulated HONO sources and thus, e.g. studies on snow or liquid surface are not considered here. Finally, we also think it is difficult to judge what is the 'best' example of the work done as this is quite a subjective term.

Minor comments:

Unfortunately, it appears the reviewer gave page and line numbers from the original submission during the pre-review process and not for the published ACPD manuscript. To better follow the discussion we have added references to the line and page numbers of the final ACPD document.

Page 2, Lines 4-7 (ACPD: page 22099, lines 14-17): First instance where surface processes are clearly not considered and should be mentioned for their implementation or lack thereof in the model.

We have added the most important surface processes for the urban field site in London to our model (dark conversion of  $\text{NO}_2$  on surfaces,  $\text{HNO}_3(\text{ad})$  photolysis on ground and aerosol surfaces, HONO deposition). Because information on surfaces types in London is lacking, we will run a sensitivity analyses (varying the reactive uptake of  $\text{NO}_2$ ) to assess how photosensitised  $\text{NO}_2$  conversion on ground and aerosol surfaces can impact modelled HONO. Others sources were not considered for the reasons discussed above. We do now discuss the reasons for not including them in the discussion.

Page 2, Line 9 (ACPD: page 22099, lines 18-20): The product of  $\text{NO}_2$  with OH reactivity is essentially a proxy for the production and deposition rates of nitric acid. There are surface mechanisms in the literature that would support such a correlation, yet the Authors conclude that only  $\text{NO}_2$  and sunlight are good predictors. This needs to be revisited throughout the manuscript.

We do not think this statement is correct, since:

a) At this site  $\text{NO}_2 \times k(\text{OH})$  is not a good proxy for the production of  $\text{HNO}_3$ , since VOCs play an important role to describe  $k(\text{OH})$ . Whalley et al., ACPD, 15, 31247–31286, 2015, show that only around 20% of  $k(\text{OH})$  is due to reaction with  $\text{NO}_2$  during daytime during this study. This is now stated in the discussion section of the revised manuscript.

b) Even if  $\text{NO}_2$  completely controlled  $k(\text{OH})$ , this would be only a good proxy for the production rate of  $\text{HNO}_3$ , but not for its concentration, which shows a maximum completely different to  $\text{NO}_2$  ( $\text{HNO}_3$ : early afternoon,  $\text{NO}_2$ : morning). That is the reason for the much worse correlations of the missing HONO source with  $\text{HNO}_3$  compared to  $\text{NO}_2$  (see also the flux study by Ren et al., 2011), by which any  $\text{HNO}_3$  dependent sources are highly unreasonable as major contributors to the daytime HONO formation in London. We definitely not say here that  $\text{HNO}_3$  could not be a minor contributor in London (see also the  $\text{HNO}_3$  photolysis included in the model) or a major contributor at other field sites.

Page 3, Lines 15-17 (ACPD: page 22101 lines 1-3): Tower or similar gradient measurements (Harrison and Kitto, 1994; Kleffmann et al., 2003; Oswald et al., 2015; Sörgel et al., 2011b, 2015; Stutz et al., 2002; Vandenboer et al., 2013; Villena et al., 2011; Vogel et al., 2003; Wong et al., 2012; Young et al., 2012), and aircraft (Li et al., 2014; Zhang et al., 2009) observations also have demonstrated that major HONO sources exist at canopy or ground surfaces through the measurement of vertical gradients. They should be mentioned here and cited appropriately.

Since the aim of the present study was not the description of the vertical structure of HONO sources, we only added here references to direct surface flux studies, by which potential source reactions can be best identified. Simple gradient studies are much more indirect and can only qualitatively describe source processes. Nevertheless, we have added some references to the gradient studies mentioned, which actually partially confirm our results (photosensitized conversion of NO<sub>2</sub> is a ground surface HONO source).

Page 3, Line 26 (ACPD: page 22101 line 12): The observations made in France (Michoud et al., 2014) do not constitute a review of the daytime sources. More comprehensive surveys of the literature include: (Ma et al., 2013; Pöschl and Shiraiwa, 2015; Spataro and Ianniello, 2014).

Although the paper Michoud et al., 2014 is not a direct review, we found that their introduction nicely covered almost all aspects of the daytime formation of HONO, even more precise compared to some reviews. It also described measurements in a city likely to have similar composition to London (no comparable study in London itself is available), so we felt it was important for it to be a prominent reference in the introduction. The review by Pöschl and Shiraiwa (2015), although highly complex and detailed (18 pages of references) focuses mostly on the interaction with the biosphere, which was thus not considered for the highly urban conditions of the present study. We also believe the use of references (when hundreds are available) is the subjective choice of the authors of a manuscript. We have changed the text so we don't describe the Michoud paper as a review.

Page 4, Lines 13-14 (ACPD: page 22102 lines 3-4): 'detailed' occurs twice in this sentence. Consider alternate phrasing.

Corrected.

Page 4, Line 30 (ACPD: page 22101 line 18): At what height above ground level is the HONO measurement made? What about all the supporting measurements used to constrain the MCM model? These missing details influence the subsequent ability to assess how the model may be limited in addressing the issue of daytime HONO formation. For example, how would vertical gradients in any of these species, particularly HONO in the early morning when stable surface layers can persist, bias the model results? What assurances can the Authors provide that the data they are using in their model runs is consistent with the assumptions being made between the model and the variety of measurements constraining it?

The sampling height of most measured species was around 5 m above the ground. In addition, due to the missing vertical resolved measurement data, the results of the present study reflect only the situation at the sampling height and should not be used to describe the chemistry at higher altitudes. This is now clarified in the revised manuscript.

Page 5, Line 1 (ACPD: page 22102 line 20): Remove 'a highly sensitive'. The sensitivity of the LOPAP is given explicitly by the LOD later in the paragraph.

Done

Page 5, Lines 10-12 (ACPD: page 22102 lines 3-5): There is a potential for particulate matter to interfere with the tandem stripping coil setup used by the LOPAP. What if artifact nitrite was present in coarse particles that were stripped in the primary channel, but not in the secondary (Bröske et al., 2003; Kleffmann et al., 2006)? This has been demonstrated to be a problem, particularly with fog droplets (Sörgel et al., 2011b), which may have been present during this observation time period. Have such possibilities been considered and removed from the data used to drive the model?

For submicrometer particles we can definitely exclude any interferences by particle nitrite, since their sampling efficiency is <2 % in the very short stripping coil (4 coil sampler), see e.g. the cited study by Bröske et al. And even if that increased to values of 10 % for larger coarse particles, such an interference would be almost perfectly corrected for by the two channel approach (=> 10 % interference in the first coil = 9 % interferences in the second coil => error by an incomplete interference correction = 1%, in addition typically: [nitrite] < [HONO]...). However, for much larger fog particles (which were not present during the campaign during daytime) interferences would be only expected in the case of high fog pH values of >5. For lower pH, expected for the urban conditions in London, the effective solubility of HONO (HONO+nitrite) would be too low to significantly influence the HONO data, even for high uptake efficiency of fog particles. Accordingly, we do not consider particle interferences as an important issue, which is confirmed by excellent intercomparison exercises of the instrument with the DOAS technique (Kleffmann et al., 2006). We have stated this in the revised manuscript (section 4.1)

Page 5, Lines 14-16 (ACPD: page 22103 line 8-10): An intercomparison was not made during this campaign, and it is well documented that interferences are location-dependent, so it seems unnecessary to validate the performance of the LOPAP in this way. Suggest removing this sentence.

First, interferences are not location-dependent, since they depend only on the interfering trace species and not on the location. Second, the LOPAP instrument used here was intercompared several times to the DOAS technique and to a PTRMS covering a wide range of conditions, from smog chambers (pure and complex mixtures including photo-smog experiments in the presence of soot particles) over semi-urban conditions, to a highly urban situation (Milan). Caused by the similarity of the latter to London (high NO<sub>x</sub> conditions), there is no reason to expect any significant interferences for the present field conditions. Reasons for former successful intercomparisons are:

- a) the used external sampling unit (no sampling lines used for the LOPAP);
- b) the extremely short gas/liquid contact time of only ca. 10 ms,
- c) the acidic sampling conditions (most known interferences are important under alkaline conditions (e.g. NO<sub>2</sub>+SO<sub>2</sub>, NO<sub>2</sub>+phenols, PAN, ...)) and
- d) the two-channel concept of the instrument (correction of interferences).

The reliability of the HONO data is of high importance here, and thus we would like to leave that sentence.

Page 5, Line 26 (ACPD: page 22103 line 20): Delete ‘?’

Already done in the ACPD version.

Page 5, Lines 27-30 (ACPD: [page 22103 line 23-25](#)): A detection limit is usually defined as three times the signal to noise for a data acquisition cycle, at minimum. This should be corrected here unless there is precedent for this approach?

This has been changed in the revised manuscript.

Page 6, Line 16 (ACPD: [page 22104 line 18](#)): Start a new paragraph at 'VOC' to help separate the different types of measurements being made.

Done.

Page 6, Line 19 (ACPD: [page 22101 line 21](#)): FID is not yet defined in the manuscript. Page 6, Line 21: Start new paragraph at 'measurements' again to help separate different classes of measurements.

Done.

Page 6, Line 28 (ACPD: [page 22105 line 2](#)): Start new paragraph at 'non-refractory'. Also, what was the size selection of the cToF-AMS? PM1.0? Were sulfate and organics measured as well? If so, consider alternative phrasing here to reflect: the full suite of non-refractory compounds that were measured, what particle size the instrument cut off its measurements at, and mention what compounds in particular were useful for this work.

The size selection of the inlet is approximately PM1.0 (Zhang et al., 2004), sulphate and organics were measured as well (Young et al., 2015) and it is specifically nitrate that is of interest here because it pertains to the working hypothesis.

This has been updated in the manuscript.

Non-refractory PM1.0 nitrate, sulphate, organic matter, chloride and ammonium were quantified. This is reflective of the overall ammonium nitrate because ammonium nitrate is both non-refractory and tends to be in the submicron fraction. While there is supermicron nitrate, it is overwhelmingly in the form of sodium nitrate, which is refractory and not measured by the AMS (see Young et al., 2015),

Were there any particle number and size distribution measurements available to include in the data analysis? A correlation coefficient is presented in Table 1, so the source of this data should be included. There is extensive precedent showing that aerosols of atmospherically relevant composition, particularly those found in urban environments, are capable of converting NO<sub>2</sub> to HONO. Is there any ability in this work to constrain such mechanisms against a ground surface source? Such comparisons have been limited and would be of great utility in guiding the focus of future field measurements.

We calculate total surface area using data from an aerodynamic particle sizer (APS) instrument by using the mean diameter of particles in each size bin (assume spherical) multiplied by the number of particles in that bin. In total there were 53 size bins ranging from 0.53 to 21.29 μm. Details of this have been added to the manuscript.

The discrimination between heterogeneous HONO formation on ground vs. particle surfaces in field campaigns needs gradient measurements not only of HONO, but also of all precursors and the particle surface area (for discussion, see Kleffmann et al., 2003), which was out of the scope of the present study. However, at least up to now, no heterogeneous reaction on particle

surfaces have been identified which would be fast enough to explain near ground HONO levels in the urban atmosphere, the reason being the much higher S/V(ground) compared to S/V(particles) and the similar uptake kinetics on ground and particle surfaces (e.g. humic, organic, aqueous, soot, surfaces, etc.). This would be also in agreement with the low correlation with the particle surface in Table 1.

Page 7, Line 15 (ACPD: page 22105 line 13): There may be the possibility that NO<sub>2</sub> from London is being transported over the open ocean to form HONO at night, only to be returned the following day with the sea to land breeze. Such formation has been shown before (Wojtal et al., 2011) and suggested that the surface layer of the ocean could act as a reservoir for HONO. Is there any evidence here that such partitioning and transport processes may contribute to the unknown daytime HONO source? Also, from here forward the referral to figures should be capitalized as 'Figure'

We do not believe this to be a source of HONO in London. Sea breeze effect observed in central London are very rare and the distance from the ocean to the measurement site is significant (~60 miles), so any HONO produced would not live long enough to be a significant source at this site.

Page 7, Line 23 (ACPD: page 22106 line 4): 'exception' should be plural

done

Page 8, Lines 3-29 (ACPD: page 22106 line 15 onwards): HONO to NO<sub>x</sub> ratios operate on the assumption that HONO only can be produced from NO<sub>2</sub> as a precursor (see major comment above), so using this ratio to assign periods when there is unknown HONO production is biased to an NO<sub>2</sub>-centric hypothesis. The Authors should be clear that their approach is biased or consider removing this part of their analysis from the manuscript and replacing it with a more representative analysis of NO<sub>2</sub> and non-NO<sub>2</sub> daytime HONO formation mechanisms. Further, there are no error bars on the HONO/NO<sub>x</sub> figures. Are the daytime and nighttime values in HONO/NO<sub>x</sub> actually statistically different based on the variability in and accuracy of the measurements?

We feel that the presentation of error bars in Fig. 2 would completely overload at least the top figure (6 data lines). For the significance of the day and nighttime data only the precision errors and not the accuracy matters. This is due to the low precision errors of the HONO and NO<sub>x</sub> data and thus we do consider the difference to be significant.

Although the HONO/NO<sub>x</sub> ratio is indeed dependant on many variables (e.g. WS, BLH, general pollution level: urban/rural/remote) it can nicely show the daytime HONO production for a single field site (see max. in Fig. 2). By the increase of the HONO/NO<sub>x</sub> ratio throughout the night (until morning NO<sub>x</sub> emissions and photolysis start) also night-time formation by NO<sub>2</sub> conversion can be nicely demonstrated (see Fig 2 and cf. also discussion on Fig. 5 in Kleffmann et al., 2002). In contrast HONO levels often stagnate to the end of the night, caused by decreasing precursor (NO<sub>2</sub>) levels (cf. Fig. 4 and 5 in Kleffmann et al., 2002). Thus, we disagree to the reviewer's statement that NO<sub>2</sub> is not a precursor of HONO (see also discussion above). If that would be the case, why are the HONO/NO<sub>x</sub> ratios (a few % in urban regions) so constant all over the world? For the reasons presented above, we do not feel we should remove the discussion and presentation on the HONO/NO<sub>x</sub> ratio.

Page 8, Lines 28-29 (ACPD: page 22107 lines 12-14): 'suggests a significant secondary and probably photo-enhanced, HONO source'. This statement is unjustified speculation and should be removed. While it is a valid consideration, such an assertion that one mechanism is the

dominant daytime HONO source before the model results are presented is premature. The data subsequently demonstrate a variety of mechanisms may be at work. Caution in revising conclusions after all revisions are made should be taken.

We agree with the reviewer that it is too early in the manuscript to make such a statement. We have changed the manuscript so that it now presents more of a discussion, stating:

“If the HONO sources which are active during night-time are the only active sources also during daytime, the HONO/NO<sub>x</sub> ratio should show a deep minimum around noon. In contrast, in Figure 2 a maximum is observed which is a clear hint to an additional daytime source. In addition, the maximum of HONO/NO<sub>x</sub> during daytime coincidences well with the one for radiation, which is again a hint for a photochemical process.”

Page 8, Line 32 to Page 9, Line 9 (ACPD: page 22107 line 15): The Authors should summarize the range of urban HONO values from the literature and cite the appropriate references. The four following sentences is an inappropriate comparison without knowing the vertical structure of HONO near the surface. HONO mixing ratios can vary by differences greater than those discussed within a few tens of meters of the ground surface at a single location. Furthermore, there are data spanning decades which are more comprehensive (i.e. greater instrumental diversity for HONO measurement and intercomparison) urban observations of HONO for Los Angeles and Houston that would likely provide better contrasts to the presented dataset.

In the original manuscript, we used references to very different urban conditions ranging from Santiago/Chile, Paris/France to two urban Chinese locations. We agree that this is probably insufficient and so have added a couple more references, also including urban DOAS observations in Milan, Italy (Kleffmann et al., 2006); and Houston, US (Wong et al., 2011). However, it is not a review of all HONO studies so we feel that comparing to a few relevant other measurements is sufficient.

Page 9, Lines 9-13 (ACPD: page 22107 line 26): This suggests even further that HONO/NO<sub>x</sub> is a poor proxy for understanding daytime HONO production.

We do not understand this comment. The similar HONO/NO<sub>x</sub> ratios for very different urban conditions and very different daytime HONO levels are clear indication for NO<sub>x</sub> being precursor of HONO (cf. again also Fig. 4 and 5 in Kleffmann et al., 2002).

Page 9, Lines 13-16 (ACPD: page 22108 line 2): It seems amiss to say that the range of HONO mixing ratios is the motivation for this modeling study. The consistent identification of daytime HONO above levels predicted from easily modeled mechanisms seems to be the true reason based on the work presented. The Authors should rework this transition to reflect exactly what the purpose of comparing the MCM to a PSS approach is.

We agree and have reworded this sentence in the revised manuscript.

Page 9, Lines 23-26 (ACPD: page 22108 lines 14-18): This is a fair critique, but misses the fact that vertical structure is also often not considered in unknown daytime HONO PSS calculations, with the exception of (Wong et al., 2012). The magnitude of the unknown source, in an environment where there is vertical structure in HONO through part or all of a day, is therefore dependent on the height above the ground surface that the measurements are being made. Further, the issue of using PSS for HONO has been raised previously and should be used as a comparison for this work (Lee et al., 2013).

We already cite the study of Lee et al. (2013) as an example of the issues surrounding the use of the PSS. We now mention the vertical structure in the text, however our work does not include any vertical structure data (see previous comments).

Page 10, Line 9 (ACPD: page 22109 line 3): Fix reference formatting.

Already done in the ACPD version.

Page 10, Lines 10-16 (ACPD: page 22110 lines 6-11): This is some nice commentary that is also consistent with the observational constraints of HONO vertical structure that the MCM would otherwise not capture. It would improve the argument here and the Authors should consider adding a sentence with this context.

Certainly, a 1D model approach would be preferable, however not possible caused by the missing experimental data (see above). Thus, here we only tried to account for the vertical transport, whenever this was important for the description of the near ground surface HONO concentration and its contribution to the OH chemistry at the measurement site. We now state this throughout the revised manuscript.

In contrast to the statement by the referee we addressed that issue also elsewhere in the document (e.g. for the NO<sub>2</sub> conversion on ground surfaces, see page 22111, lines 4-15). In addition, in the revised manuscript, we also stress that issue for the HNO<sub>3</sub> source description (see major issue point 4).

Page 10, Lines 21-23 (ACPD: page 22109 line 16): This value is not 'virtually zero'. It is 50 times the LOPAP detection limit, which is determined at three times the signal to noise (i.e. S/N = 150). Consider rewording this sentence to '. . . decreases to < 0.05 ppbV by midday.'

We agree and now specify the lower boarder HONO PSS levels during daytime more exactly in the revised manuscript.

Page 11, Lines 15-17 (ACPD: page 22110 lines 12-13): This value for deposition velocity is three times less than what was used for HONO (unless the HONO deposition velocity was for a boundary layer of different depth?). This blanket approach likely isn't representative of the suite of compounds in the model that are central to the chemistry being probed (e.g. NO<sub>2</sub> vs HNO<sub>3</sub>). How do the Authors justify this approach?

We agree that this was not a good approach. We now use surface deposition velocities ( $1/R_c$ ) of  $3 \text{ cm s}^{-1}$  for HNO<sub>3</sub> and  $2 \text{ cm s}^{-1}$  for HONO and  $1 \text{ cm s}^{-1}$  for NO<sub>2</sub>. For the total deposition velocities used in the model parameterization of the turbulent (Ra) and diffusive (Rb) mixing with the wind speed were used (VDI 3782, 2006). In addition, in order to better describe HONO source and sink prosses on ground surfaces, we now use an effective HONO mixing layer height of 75 m, calculated as the likely height to which HONO will reach ( $1/e$ ), given a daytime lifetime of 15 minutes. The deposition velocity of all other species will not affect the results of the present study (HONO chemistry) to a significant extent. The model description has been updated in the manuscript.

Page 12, Lines 3-6 (ACPD: page 22111 lines 1-4): There are many published reactive uptake values for NO<sub>2</sub>, most are smaller than 0.03. How do the Authors justify using this value? Presumably this produces the best match of the modeled HONO to that observed at night, but the argument is not made and the data is not shown.

The value specified is not a “reactive uptake value”, but the effective HONO yield of the dark NO<sub>2</sub> conversion on ground surface. We have used an effective yield measured directly in the atmosphere, which is in contrast to lab studies on pure substrates for which higher yields have been observed. Certainly - and as already discussed in that section – the number is highly uncertain, but will not affect our daytime results (the topic of the manuscript) to any significant extent, caused by the slow uptake kinetics of the dark uptake vs. the photoenhanced uptake (minimum one order of magnitude difference). So even with a 100 % HONO yield, the dark reaction would be not significant during daytime. In the revised manuscript we have in addition scaled the effective HONO yield to better describe nighttime formation of HONO, since the literature value (0.03) was determined over grass land, which is not representative for our field site.

Page 12, Lines 6-9 (ACPD: page 22111 lines 4-7): There is certainly a lot of uncertainty in this assumption since the model is trying to match HONO mixing ratios observed at the surface, while immediately diluting them throughout the boundary layer. If the HONO mixing ratios at the surface can be matched, then the sources of daytime HONO are over-represented by the difference in the HONO vertical gradient between the measurement height and the top of the mixed layer. Similar issues arise when investigating the influence of HONO on the local OH production, which should be discussed in more detail in the appropriate section.

The point by the referee is already discussed in detail below this (ACPD: page 22111 line 9). Again a 1D model would be preferable, but it is not possible here.

Page 12, Line 9 (ACPD: page 22111 line 8): Delete ‘in contrast’ and start the sentence with ‘Strong HONO’. This sentence presents information that is consistent with the consequence of the prior assumption. It is not a contrast.

Done.

Page 12, Line 13 (ACPD: page 22110 lines 12-13): Is this why nighttime HONO is not presented in Figure 3?

The PSS approach would not work at night and so this is why we confined this study to the daytime. This is now made clear in the manuscript (see sections 3.1 and 3.2).

Page 12, Lines 19-25 (ACPD: page 22111 lines 19-26): The experimental data for the photolysis rates of HNO<sub>3</sub> on leaf surfaces, to the knowledge of this Reviewer, have still not been published. In any case, leaf surfaces are not truly representative of urban environments and lab studies using better urban proxies have demonstrated that the HONO yield is not 100 % (Baergen and Donaldson, 2013; Nanayakkara et al., 2014; Scharko et al., 2014). The Authors should revisit the literature to constrain these model runs with more realistic HONO yields and HNO<sub>3</sub> photolysis rates.

See answer above, the statement is not correct and we feel that we do use realistic values.

Page 12, Lines 26-32 (ACPD: page 22111 line 27 – page 22112 line 9): If this is the case, then the same argument is relevant for the previous section on HNO<sub>3</sub> photolysis. Approximations of this have been presented and should be implemented in the model runs presented (Oswald et al., 2015; Sörgel et al., 2011a, 2015).



The approach used by the mentioned former studies were already applied here for the ground surfaces sources used in our model (homogeneous mixed surface layer), resulting only in a minor contribution to the measured HONO levels. From the difference to the measured HONO a major extra source was quantified, which correlates with  $\text{NO}_2 \times J(\text{NO}_2)$  and the photosensitized conversion of  $\text{NO}_2$  was inferred based on known lab studies. However, in contrast to the considered minor sources (for which any description in the model will not significantly affect the results) we have not explicitly added the photosensitized conversion of  $\text{NO}_2$  into the base model. Because information on surface types in London is uncertain and  $\text{NO}_2$  uptake kinetics on different surface types e.g. urban grime is also uncertain, we will run a sensitivity analyses (varying the reactive uptake of  $\text{NO}_2$ ) to assess how photosensitized  $\text{NO}_2$  conversion on ground and aerosol surfaces can impact modelled HONO. In contrast, the kinetics of an artificial humic acid (Aldrich), see Stemmler et al. (2006; 2007), as used in other studies, is not expected to represent the reality for an urban measurement site. Thus, our results give only an indication for the missing daytime source and further studies on real urban grime (etc.) surfaces are necessary for the future. Only based on such kinetic results could the photosensitized conversion of  $\text{NO}_2$  be implemented correctly into a model. However we do now include a sensitivity study into this source in the discussion of the revised manuscript (section 4.2).

Page 14, Lines 4-5 (ACPD: page 22113 lines 15-17): What is the implication of this statement? If the model cannot reproduce nighttime HONO, then how is it initializing each daytime calculation when not constrained to HONO? How does this affect the performance of the model with respect to daytime HONO chemistry?

We agree that it is not clear what we have done. We have now removed the statement about nighttime chemistry and state that we only consider what is happening during the day. We now only consider data from 08:00 UTC, a time at which all HONO produced during the night will have been lost. This has now been explicitly stated in the manuscript.

Page 14, Line 8 (ACPD: page 22113 line 19): 'significantly' Which statistical test was performed that substantiates this word choice?

We have removed the word 'statistical' as we did not carry out a full statistical analysis. We do now state that the discrepancy is outside the 10% error of the LOPAP instrument.

Page 14, Lines 14-15 (ACPD: page 22113 lines 26-27): If it is possible to ballpark the daytime contributions of onitrophenols to the daytime HONO budget without measurements, then it does not seem unreasonable to also consider the  $\text{NO}_2$ -independent surface mechanisms from the literature that have been shown to have a greater potential significance on daytime HONO production. Addition of aerosol conversion of  $\text{NO}_2$  or photolysis of particulate  $\text{HNO}_3$  would also raise the impact in testing hypotheses contrasting the different surfaces present for production of daytime HONO.

The considered HONO source by nitroaromatics certainly represents an upper limit, and will be not important even if the description is erroneous. For the other  $\text{NO}_2$  independent sources, not used in the model and not expected to be a significant importance for the present measurement site, see extended discussion above. In contrast to the statement by the referee an  $\text{NO}_2$  conversion on aerosols was used in the model, see page 22111, point 3. In addition, we have added  $\text{HNO}_3$  photolysis in the particle phase to the model and this is now discussed in the model description and shown on the figure.

Page 14, Lines 23-25 (ACPD: page 22114 lines 7-9): This term is dependent on the measurement height which has already been stated as having bias. Daytime HONO gradients have been reported previously (Vandenboer et al., 2013; Wong et al., 2012, 2013), so is this further analysis truly giving new insight into daytime HONO production?

While the magnitude of the missing HONO daytime source may be indeed depending on possible vertical gradients, its correlation with potential precursors will be not affected. In addition also the major contribution of HONO as a daytime source of OH radicals near to ground surfaces will be also not affected. Thus, we consider the results still important, although we are aware of the shortcomings compared to a 1D model analysis.

Page 14, Lines 28-31 (ACPD: page 22114 lines 13-15): In urban areas of California, NO<sub>2</sub> dependence has been recently shown to not underlie daytime HONO production by using statistical analyses of weekend and weekday data from the CalNex campaigns (Pusede et al., 2015). Photolysis of nitrate deposited the previous day (Zhou et al., 2011) could also be consistent with the production of HONO from urban areas where NO<sub>x</sub> emissions are greater, but not be directly dependent on NO<sub>x</sub>. Assigning the likelihood of the daytime HONO production to NO<sub>2</sub> is not unreasonable, but without putting the assertion in the context of other hypotheses is neglecting a large body of recent literature.

The term 'significantly' is used here again and should be addressed as with previous instances of this word choice.

The results by the study of Pusede et al. (2015) can be explained by the non-linear kinetics of heterogeneous NO<sub>2</sub> conversion reactions (NO<sub>2</sub>+H<sub>2</sub>O, NO<sub>2</sub>+ organics, NO<sub>2</sub>+TiO<sub>2</sub>), which was not considered in that study (see also answer above: HONO/NO<sub>x</sub> is expected to be higher on weekend, with low NO<sub>x</sub> levels compared to weekdays. This is not a contrast to the proposed photosensitized conversion of NO<sub>2</sub>). In addition, Ren et al. (2011) showed a high correlation of measured HONO fluxes with NO<sub>2</sub> x radiation for the CalNex campaign. Flux measurements are the most direct way to identify a ground surface source mechanism, much better than any statistical analysis.

Nitrate photolysis by deposited HNO<sub>3</sub> was considered in the model and is not able to describe measured HONO levels.

We have left this section unchanged as it is merely pointing to the potential source of the missing HONO, however our updated discussion section (see earlier comments) now more fully discusses other sources in the literature.

We again remove the word 'significantly'.

Page 15, Lines 1-5 (ACPD: page 22114 lines 18-21): Is the photochemical model really more complete than the PSS for understanding daytime HONO production? PSS models can capture the underlying nighttime formation mechanisms of HONO fairly well and the magnitude of the unknown daytime source. The photochemical model does not have any vertical resolution or transport processes and this was used despite the fact that HONO is well documented to be formed from surface chemistry. Further to this point, statements here about the postulated HONO sources are simply not true as per the comments made in the major comments above, specifically for surface NO<sub>2</sub> conversion on photosensitized surfaces (Sörgel et al., 2015). It seems that the point of using the MCM here is to get at the impact of near-surface HONO on radical reservoirs, since this would be the major reason for including all of the supporting measurements mentioned, but the issue is not presented clearly between the stated objectives at the outset of the manuscript, nor in the discussion.

If the major HONO source is missing, neither a PSS nor the MCM model can describe HONO daytime levels properly. Since the missing source correlates with NO<sub>2</sub> x J(NO<sub>2</sub>) a photosensitized NO<sub>2</sub> conversion – as identified in the lab – is proposed here. To the uncertain description of this source in other models, see answers above (nobody knows the kinetics for

the urban surfaces of London). We have updated the manuscript introduction to make clearer the objectives of using the MCM study.

Page 15, Lines 23-25 (ACPD: page 22115 lines 12-15): This sentence is confusing. This seems to be saying that the LOPAP is not measuring all of the HONO, but if that was the case then the bias would be a higher HONO signal. Maybe rephrase to be clear that 2 ppb HO<sub>2</sub>NO<sub>2</sub> at 15 % interference would explain the difference between measured and modeled HONO.

We agree and have updated the sentence accordingly.

Page 15, Lines 27-29: An intercomparison was not done during the ClearfLo study though, so this statement has marginal relevance. Consider removing. More suitable to the discussion would be statements summarizing the known maximum error in the LOPAP measurements of HONO from the literature, specifically those that cannot be easily corrected for, such as coarse particulate matter that is collected with bias in the measurement channel over the background channel (Bröske et al., 2003; Kleffmann and Wiesen, 2008; Sörgel et al., 2011b).

We feel this sentence is important, since we do not expect any other interferences in London compared to the similar urban conditions in Milan. All identified interferences are marginal including particle matter. For further details see answer above.

Page 16, Lines 3-8 (ACPD: page 22115 lines 22-24):  $j(\text{anything}) \times \text{NO}_2$  that is relevant in the troposphere will give this relationship. Irradiance has been shown to be the best measure of unknown daytime HONO production through correlational analysis (Wong et al., 2012, 2013). This style of analysis may indicate that NO<sub>2</sub> conversion on photoexcited organics is taking place, but should not be constrained to that interpretation as the production could equally be interpreted as any process related to photochemistry (e.g. Table 1 shows a stronger correlation with OH than with  $j(\text{NO}_2)$ ) and not dependent on NO<sub>2</sub> (Pusede et al., 2015).

The use of irradiance would have been a better parameter to describe photolytic surface reactions for perfectly flat and horizontal surfaces, which we however do not have at our field site. In addition, we did not have such data available and calculating it from actinic fluxes has potentially large errors. So we use  $J(\text{NO}_2)$  as a 'proxy' for radiation and now state this in the manuscript. However we do not feel this changes our conclusions. The correlations with  $J(\text{NO}_2)$  (0.539) is smaller than when NO<sub>2</sub> is included ( $J(\text{NO}_2) \times \text{NO}_2$ : 0.696) in contrast to the inclusion of HNO<sub>3</sub> ( $J(\text{NO}_2) \times \text{HNO}_3(\text{ads})$ : 0.435). All these results support our proposed mechanism. For answers to the statement that the source is not NO<sub>2</sub> dependent and to the Pusede et al. study, see above.

Page 16, Lines 12-20 (ACPD: page 22116 lines 6-14): This is, in effect, confirming that additional degrees of freedom allow for higher explanatory power in the variance of daytime HONO production. Maybe explain why such an approach is technically sound for isolating mechanisms of HONO production and how the variables explored may cause bias. Can this approach be used to soundly assign the dominant daytime HONO source? What about sources that have a more spurious, yet significant, nature within or between days? How does the error in the product of the two term investigations get taken into account when performing linear regression? Are the linear regressions weighted by the error in all measurements and/or the propagated error where two measurements are being combined? Is the regression utilizing an adjusted  $r^2$  approach to account for the number of terms in the model?

All we can say about this analysis is that if  $r^2$  increases when adding another precursor parameter it is likely that the source is more relevant. We do not attempt a full statistical

analysis of this, it is merely to suggest potential parameters that may have an influence on HONO production that are not currently in our model.

Page 16, Lines 32-33 (ACPD: page 22116 lines 28-29): The Authors should explain how  $k(\text{OH})$  represents surface organic matter in brief here. Some expansion is certainly warranted and may reduce the speculative tone of the photolytic  $\text{NO}_2$  conversion arguments.

If organic matter results from the uptake of semivolatile organics on surfaces, then it is reasonable that the postulated source correlates with  $k(\text{OH})$  which represent, at least in part, VOC levels in the atmosphere. Whalley et al. (2015) showed that  $k(\text{OH})$  was dominated by VOCs during the daytime at the measurement site (~80%). We have added this statement to the manuscript.

Page 17, Section 4.3 (ACPD: section 4.3): Vertical gradients in measured radical reservoirs have been demonstrated in the literature (Kim et al., 2014; Young et al., 2012) and, like HONO, not accounting for this vertical structure in a photochemical model may lead to biases. How might these previous studies affect the interpretation of the MCM results in this work?

We stress again, that the results of the present study are strictly only valid for the measurement site, i.e. close to the ground surfaces. For the analysis of the vertical structure of the HONO contribution to the OH initiation, our measurement data is not sufficient. Here further gradient studies would be necessary. We now state this in the revised manuscript.

Page 17, Lines 21-32 (ACPD: page 22117 lines 21-28): If the photochemical model is diluting the surface HONO throughout the boundary layer, but the HONO measured at the surface is part of a vertical gradient due to surface production and subsequent transport, would constraining the model to surface HONO measurements be appropriate for assessing HONO impacts on boundary layer OH production rates? Would it be more accurate to say that the model is being used to understand OH production at the HONO measurement height even though it is doing the chemistry in a dynamic boundary layer?

We agree with this and the manuscript has been updated accordingly.

Page 18, Lines 8-10 (ACPD: page 22118 lines 13-15): As with the previous comment, the vertical structure in HONO and radical reservoirs means that the model is heavily biased to the surface observations and that reproduction of observed OH is only valid near the measurement heights of the HONO and OH instrumentation, and not throughout the boundary layer. This should be clearly stated.

We agree with this and the manuscript has been updated accordingly.

Page 18, Lines 27-30 (ACPD: page 22119 lines 4-5): The result is still only a correlation, not a confirmation. Maybe if this gave the only high correlation value, but that is not the case.

We agree with this and have updated the text in the revised manuscript accordingly.

Page 19, Lines 4-6 (ACPD: page 22119 lines 14-16): The significance of these results is certainly that models need to get HONO formation chemistry and number densities right to understand OH production. However, vertical structure is not captured by this model and so the results only apply to the measurement height and caution should be used in the breadth of the conclusions.

We agree with this and have updated the text in the revised manuscript accordingly.

Table 1: There are a number of 'species' here that have published 'mechanisms' where correlations greater than 0.3 have been found, but are not discussed (e.g.  $j(\text{NO}_2)^*T$ ;  $j(\text{NO}_2)^*\text{NH}_4^+$ ,  $j(\text{NO}_2)^*k(\text{OH})$ , etc.). Certainly these warrant some expansion in the discussion since they are as important as the  $j(\text{NO}_2)^*\text{NO}_2$  finding and were included in this table for a reason. There are a number of other variables that were explored, that returned correlations greater than 0.3 that are consistent with other hypotheses presented for surface processes (e.g. photochemistry, partitioning from soil pore water as a function of temperature, production of  $\text{HNO}_3$ ). Further, this approach is looking only for a persistent daytime production mechanism and would miss any HONO production that has temporal variability that is not captured by the included terms (e.g. soil microbial activity, changes in surface acidity and/or water leading to release of HONO).

We have added some discussion to other correlations in the revised manuscript.

Figure 2: Error bars in the measurements here would be useful in justifying the separation of easterly and westerly flows for further analysis. The LOPAP accuracy at 400 pptV is  $\pm 40$  pptV, so are the daytime HONO values between the two transport conditions actually statistically different? If so, is it possible to exclude known LOPAP interferences from direct bias in these air masses (e.g. aerosols (Bröske et al., 2003), fog (Sörgel et al., 2011b))  
What is the error in HONO/NO<sub>x</sub>? When the error is considered is there any statistically significant change in HONO between maximum and minimum HONO/NO<sub>x</sub> average values?

For the error analysis discussed by the referee only the precision errors and not the accuracy has to be considered. For example, the main contribution to the accuracy errors of the LOPAP results from the calibration (pipettes, flasks, standard, etc.). However, these errors are independent e.g. on the wind direction. Precision errors of the instruments used are much lower than stated here (e.g. only 1-2 % for the LOPAP and NO<sub>x</sub> at these concentrations). Since we also do not consider interferences to be of significant importance for the highly urban conditions in London (see above), differences between the data traces are significant. We feel that error bars on the figure would make it overly busy and would not add to the discussion for the reasons described above.

Figure 6: What explanatory power would an error of a factor of 2 in direct emissions of HONO have on this figure? Is the current data available on this satisfactory or is there a need for greater constraints so that better estimates of the unknown daytime source can be made? It would be interesting to see how the missing HONO term changes by constraining the model to the upper and lower limits of HONO/NO<sub>x</sub> from primary emissions.

We believe we have already overestimated the emissions as stated in the MCM model description text (there is significant HONO photolysis since emission of HONO/NO<sub>x</sub> but we used a constant emission ratio). We have now carried out a sensitivity analysis in the model by increasing the direct emissions by a factor of 2 and we see a 4% increase in the modelled HONO. Hence we do not believe direct emissions to be the source of the missing HONO. We now stated this in the text (section 4.2).

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