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2     **Chemical characterization of submicron regional background aerosols in the Western**  
3         **Mediterranean using an Aerosol Chemical Speciation Monitor**

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17

18 **ABSTRACT**

19 An Aerosol Chemical Speciation Monitor (ACSM, Aerodyne Research Inc.) was  
20 deployed at Montseny (MSY, 720 m a.s.l.) regional background site in the Western  
21 Mediterranean from June 2012 to July 2013 to measure real-time inorganic (nitrate, sulphate,  
22 ammonium and chloride) and organic submicron aerosol concentrations. Co-located  
23 measurements were also carried out including real-time submicron particulate matter (PM<sub>1</sub>)  
24 and black carbon (BC) concentrations, and off-line PM<sub>1</sub> chemical analysis. This is one of the few  
25 studies that compare ACSM data with off-line PM<sub>1</sub> measurements, avoiding the tail of the  
26 coarse mode included in the PM<sub>2.5</sub> fraction. The ACSM + BC concentrations agreed with the  
27 PM<sub>1</sub> measurements, and strong correlation was found between the concentrations of ACSM  
28 species and the off-line measurements, although some discrepancies remain unexplained.  
29 Results point to a current underestimation of the relative ionization efficiency (RIE) established  
30 for organic aerosol (OA), which should be revised in the future. The OA was the major  
31 component of submicron aerosol (53% of PM<sub>1</sub>), with a higher contribution in summer (58% of  
32 PM<sub>1</sub>) than in winter (45% of PM<sub>1</sub>). Source apportionment of OA was carried out by applying  
33 Positive Matrix Factorization (PMF) using the Multilinear Engine (ME-2) to the organic mass  
34 spectral data matrix. Three sources were identified in summer: hydrocarbon-like OA (HOA),  
35 low-volatile oxygenated OA (LV-OOA), and semi-volatile oxygenated OA (SV-OOA). The  
36 secondary OA (SOA, 4.8  $\mu\text{g m}^{-3}$ , sum of LV-OOA and SV-OOA) accounted for 85% of the total  
37 OA and its formation during daytime (mainly SV-OOA) was estimated to be 1.1  $\mu\text{g m}^{-3}$ . In  
38 winter, HOA was also identified (12% of OA), a contribution from biomass burning OA was  
39 included, and it was not possible to differentiate two different SOA factors but a single OOA  
40 factor was resolved. The OOA contribution represented the 60% of the total OA, with a degree  
41 of oxidation higher than both OOA summer factors. An intense wildfire episode was studied  
42 obtaining a region-specific BBOA profile.

43

44 **KEYWORDS:** ACSM, PM<sub>1</sub>, organics, chemical composition, Mediterranean, air quality.

45

46 **1 INTRODUCTION**

47       Ambient aerosols have adverse effects on human health (Pope III and Dockery, 2006),  
48 and affect climate (IPCC, 2013), ecosystems, crops, and regional visibility. Fine particulate  
49 matter (PM<sub>1</sub>, particles with an aerodynamic diameter <1  $\mu\text{m}$ ) contains substantial fractions of  
50 inorganic compounds and carbonaceous aerosols, the latter reaching up to 90% of the mass  
51 (Jimenez et al., 2009). Carbonaceous aerosols are comprised of organic compounds,  
52 collectively referred to as organic aerosol (OA), elemental carbon (EC), and carbonates (from  
53 mineral dust), although the latter can be considered negligible in submicron aerosols.

54       The Western Mediterranean Basin (WMB) has special atmospheric and geographic  
55 characteristics that imply the interest of the detailed study of the ambient aerosols in this area  
56 (Querol et al., 2009). The regional background has been investigated through long data series  
57 of measurements in previous studies available at Montseny (representative of the regional  
58 background in the WMB). Pérez et al. (2008) found average particulate matter concentrations  
59 at Montseny of 17, 13 and 11  $\mu\text{g m}^{-3}$  of PM<sub>10</sub>, PM<sub>2.5</sub> and PM<sub>1</sub>, respectively, in the 2002-2007  
60 period. Cusack et al. (2012) and Querol et al. (2014) found a decreasing trend in PM<sub>2.5</sub>  
61 concentrations from 2001 to 2012 of  $-0.39 \mu\text{g m}^{-3}$  per year. PM<sub>2.5</sub> concentrations were found  
62 higher in the WMB than at other rural background sites across Spain, Portugal, Germany and  
63 Scandinavia but lower than those measured in Switzerland, Italy and Austria (Cusack et al.,  
64 2012). The prevailing daily evolution is driven by the breeze circulation (mountain and sea  
65 breezes), with lower PM<sub>x</sub> concentrations at night owing to the nocturnal drainage flows, and  
66 higher PM<sub>x</sub> concentrations at midday owing to the transport of atmospheric pollutants  
67 accumulated in the pre-coastal depression upwards by the breeze (Pérez et al., 2008).  
68 Maximum PM<sub>10</sub> concentrations were found in summer, February-March and November, and  
69 sporadic PM<sub>x</sub> increases may be recorded under anticyclonic conditions (Pey et al., 2010). The  
70 chemical composition of PM<sub>2.5</sub> is characterised by high concentrations of organic aerosol and  
71 sulphate, followed by crustal material, nitrate and ammonia, with sea spray and elemental  
72 carbon being a minor part of the total PM<sub>2.5</sub> mass (Cusack et al., 2012). Compared to other  
73 central European sites, the Western Mediterranean aerosol is characterised by higher  
74 concentrations of crustal material but lower concentrations of organic aerosol, elemental  
75 carbon and ammonium nitrate (Pey et al., 2009). Nevertheless, relatively high PM<sub>2.5</sub>  
76 concentrations of carbonaceous aerosol and sulphate transported from populated coastal  
77 areas are regularly recorded, especially during winter anticyclonic episodes and summer  
78 midday PM highs (Pey et al., 2009;Pey et al., 2010). A organic carbon (OC) to elemental carbon  
79 (EC) ratio (14 in summer, 10 in winter) was detected, pointing to the influence of biogenic

80 emissions, secondary organic aerosol (SOA) formation favoured by high ozone concentrations  
81 and insolation, and intensive recirculation of aged air masses (Pey et al., 2009;Querol et al.,  
82 2013).

83 The sources of organic aerosol in the regional background site of Montseny were  
84 studied in two intensive campaigns, using off-line  $^{14}\text{C}$  analysis (Minguillón et al., 2011), Aerosol  
85 Mass Spectrometers (AMS) (Minguillón et al., 2011;Crippa et al., 2014), and organic tracers  
86 (Alves et al., 2012;Van Drooge et al., 2012). Minguillón et al. (2011) found that the contribution  
87 of fossil fuel combustion sources (mainly road traffic emissions) to OC at Montseny was 31%  
88 and 25%, in winter and summer, respectively, and that 85% of this fossil OC was secondary.  
89 The contribution of biomass burning emissions was relatively low when compared with other  
90 regional background sites in Europe, and was estimated to be 21% and 12% of the total OC in  
91 winter and summer, respectively. Alves et al. (2012) concluded that the anthropogenic input  
92 may be associated with the transport of aged air masses from the surrounding industrial/urban  
93 areas, which superimpose the locally originated biogenic hydrocarbons.

94 Besides these studies, a long time series of organic aerosol data has not been analysed  
95 in Montseny. To this end, the newly-developed Aerosol Chemical Speciation Monitor (ACSM)  
96 would be suitable (Ng et al., 2011b), as opposed to the use of AMS, which cannot work  
97 unattended and therefore is usually employed around the world for periods of about one  
98 month. Nevertheless, due to its recent implementation, some studies based on ACSM data are  
99 found in the literature (Ng et al., 2011b;Shaw et al., 2012;Sun et al., 2012;Budisulistiorini et al.,  
100 2013;Canonaco et al., 2013;Carbone et al., 2013;Sun et al., 2013a;Sun et al., 2013b;Takahama  
101 et al., 2013;Bougiatioti et al., 2014;Canonaco et al., 2014;Petit et al., 2014;Ripoll et al., 2014a).

102 The present study aims at interpreting a one-year time series of inorganic and organic  
103 compounds in the submicron aerosol in the regional WMB, with special focus on their  
104 evolution throughout the year as a function of the concatenation of different atmospheric  
105 scenarios. The different types and origin of organic aerosol (OA) are also investigated. To this  
106 end, an ACSM was deployed for a year in the regional background site of Montseny (MSY),  
107 according to the schedule planned within the Aerosols, Clouds, and Trace gases Research  
108 InfraStructure (ACTRIS) Network project. Moreover, a validation of the ACSM data is carried  
109 out by comparison with co-located instruments both real-time and off-line.

110

111 **2 METHODOLOGY**

112 **2.1 Sampling site**

113 The MSY station (41°46'46"N, 02°21'29"E, 720 m a.s.l.) is located in the Montseny  
114 natural park, in a densely forested area, 50 km to the N-NE of the Barcelona urban area, and  
115 25 km from the Mediterranean coast. The station is located on the upper walls of a valley  
116 extending perpendicularly from the Catalan Pre-Coastal ranges to the coast. The site is  
117 relatively far from urban and industrial areas, but it can be affected by anthropogenic  
118 emissions transported from populated and industrialised areas under specific meteorological  
119 conditions. The MSY station is in the ACTRIS Network (formerly EUSAAR, European Supersites  
120 for Atmospheric Aerosol Research), is a Global Atmosphere Watch (GAW) site, and is part of  
121 the IDAEA-CSIC and the Department of Environment of the Autonomous Government of  
122 Catalonia air quality monitoring network.

123 The prevailing atmospheric dynamics has been described elsewhere (Pérez et al.,  
124 2008;Pey et al., 2009). Briefly, in winter the location of the Azores high pressure system  
125 favours the entry of clean Atlantic air masses into the WMB which replace the existing air  
126 masses leading to a decrease of pollutants. In summer, the very weak pressure gradients result  
127 in local circulations dominating the atmospheric dynamics with the consequent accumulation  
128 of pollutants (Millán et al., 1997). The climate is typical Mediterranean with warm summers,  
129 temperate winters and irregular precipitation rates during the year.

130 The daily classification of meteorological episodes affecting MSY during the study  
131 period was made as described in Pérez et al. (2008), leading to the following types of scenario:  
132 Atlantic Advection, North African, Mediterranean, European, Regional, and Winter Anticyclonic  
133 Episodes. The frequency of each type of scenario for each of the months of the study period is  
134 shown in [Figure S1](#).

135 Eliminado: Figure S1

136 **2.2 ACSM settings, calibrations and data processing**

137 An ACSM was deployed from June 2012 to July 2013, according to the ACTRIS  
138 schedule, to measure non-refractory submicron aerosol species (organics, nitrate, sulphate,  
139 ammonium and chloride) in real-time (Ng et al., 2011b). Briefly, the instrument uses an  
140 aerodynamic lens to sample and focus submicron particles (75-650 nm) into a narrow particle  
141 beam (Liu et al., 2007), with a flow of approximately 85 cc/min. The beam is transmitted into  
142 the final of three vacuum chambers, where particulate matter is flash-vaporized on a hot oven  
143 (600 °C), ionized by hard electron impact ionization (70eV) and subsequently detected using a

145 commercial quadrupole mass spectrometer. The concentration of the aforementioned species  
146 is calculated based on the measured aerosol mass spectra. For a given species, its  
147 concentration is calculated based on the addition of the ion signals at each of its mass spectral  
148 fragments and its ionization efficiency (IE) (Canagaratna et al., 2007). Since calibration of IEs  
149 for all ambient species is not feasible, the relative ionization efficiency (RIE) (compared to that  
150 of nitrate) is used for different species.

151 Thus, mass calibration of the ACSM is based on determining the instrument response  
152 factor (RF) using ammonium nitrate calibration aerosol (Ng et al., 2011b). In this study, an  
153 atomizer (TSI, Constant Output Atomizer Model 3076) was used for primary aerosol  
154 generation, followed by a silica gel diffusion dryer, an SMPS system (model TSI 3936),  
155 comprised of an electrostatic classifier (model TSI 3080) with a differential mobility analyzer  
156 (DMA, model TSI 3081) and a condensation particle counter (CPC, TSI 3772). Monodisperse  
157 300 nm ammonium nitrate aerosol particles were generated for the calibration. The calibration  
158 comprised a range of nitrate concentrations from 0 to 15  $\mu\text{g m}^{-3}$ , which were achieved by  
159 diluting the generated aerosol. RIE for ammonium was directly determined from the  
160 ammonium nitrate calibration.

161 Several calibrations were carried out through the sampling period, and average values  
162 for nitrate IE and RIE for ammonium were used for the whole dataset. After several tests  
163 around the world, more experience has been gained regarding the performance of the ACSM.  
164 Hence, RIE for sulphate has been shown to vary from instrument to instrument and therefore  
165 the default value (1.2) (Ng et al., 2011b) cannot be directly used. Nevertheless, this  
166 information was known when our ACSM was no longer at the MSY station, and hence,  
167 sulphate RIE was determined by doing the aforementioned calibration exercise with  
168 ammonium sulphate monodisperse aerosol in Barcelona. The sulphate RIE value found was  
169 very close to the default value and hence 1.2 was used for the current dataset. The default RIE  
170 for organics (1.4) (Ng et al., 2011b) has been used, although some discussion about this can be  
171 found in section 3.1.

172 The ACSM was connected to a general inlet equipped with a nafion drier to maintain  
173 the RH below 40%, although technical problems resulted in some periods (about 50% of the  
174 data points) with uncontrolled RH. The ACSM was set to measure with a time resolution of  
175 approximately 30 minutes, resulting from setting it to work with 24 scans (alternatively 1  
176 sample and 1 filtered) per data point with a scan speed of 500 ms/amu. The data acquisition  
177 software provided by Aerodyne Research (version 1.4.2.5 from the beginning to 18<sup>th</sup> December  
178 2012, and version 1.4.3.8 for the rest of the period) was used to process the measurements.  
179 The data were analyzed with the ACSM data analysis software version 1.5.3.2 (Aerodyne

180 Research Inc.) written in Igor Pro (WaveMetrics, Inc., Lake Oswego, OR, USA). A correction for  
181 the instrument performance limitations was applied to the dataset based on the inlet pressure  
182 and N<sub>2</sub> signal. The aerosol mass concentrations were then corrected for particle collection  
183 efficiency (CE) following the Middlebrook approach (Middlebrook et al., 2012). The aerosols at  
184 MSY are assumed to be internally mixed and thus the CE was assumed to be the same for  
185 different components in contrast to e.g. Hawkins et al. (2010).

186

187 **2.3 Additional measurements and instrumentation**

188 Submicron particulate matter (PM<sub>1</sub>) 24-h samples were collected on quartz fibre filters  
189 (Pallflex 2500QAT-UP) using DIGITEL (DH-80) high volume (30 m<sup>3</sup> h<sup>-1</sup>) samplers with a PM<sub>1</sub>  
190 impactor inlet. The sampler, and therefore the collected samples, was kept inside a container  
191 with controlled temperature (between 24 and 26°C). Samples were collected every 4 days.  
192 Gravimetric PM<sub>1</sub> determination was carried out by weighing the filters before and after  
193 sampling, after stabilization in a conditioned room (20°C and 50% relative humidity). Chemical  
194 off-line analyses were carried out. A quarter of the filter was acid digested (HNO<sub>3</sub>:HF:HClO<sub>4</sub>),  
195 and the resulting solution was analysed by Inductively Coupled Plasma Atomic Emission  
196 Spectroscopy (ICP-AES) for major elements determination, including S, from which the  
197 sulphate concentration was calculated. Another quarter of the filter was water extracted to  
198 determine the nitrate, sulphate and chloride concentrations by Ion Chromatography and the  
199 ammonium concentrations by an ion selective electrode. OC concentrations were determined  
200 by thermal-optical methods using a Sunset instrument following the EUSAAR2 thermal  
201 protocol (Cavalli et al., 2010). Blank filters were analysed together with the samples and  
202 concentrations were subtracted from those found in the samples in order to calculate the  
203 ambient concentrations.

204 PM<sub>1</sub> hourly concentrations were measured using an optical particle counter (GRIMM,  
205 model 180) and corrected with the simultaneous 24-h gravimetric measurements (Alastuey et  
206 al., 2011). Equivalent Black Carbon (BC) mass concentrations (Petzold et al., 2013) were  
207 measured with a 1-minute time resolution by a multi-angle absorption photometer (MAAP,  
208 model 5012, Thermo) using a PM<sub>10</sub> inlet, and using the default mass absorption cross section  
209 (MAC) from the instrument software (6.6 m<sup>2</sup> g<sup>-1</sup>). Particle number size distributions (9-820 nm)  
210 were measured by a Scanning Mobility Particle Sizer (SMPS), comprising a DMA connected to a  
211 CPC (TSI 3772), with a system designed and manufactured at the Leibniz Institute for  
212 Tropospheric Research (Wiedensohler et al., 2012). The mass concentration from SMPS data

213 was calculated from the total volume of particles and the composition-dependent density  
214 calculated based on the ACSM chemical composition.

215 Wind direction and speed, solar radiation, temperature, relative humidity and  
216 precipitation were recorded using conventional instruments and hourly data can be seen in

217 | [Figure S2](#)

218

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219 **2.4 Source apportionment of OA**

220 The source apportionment to the organic fraction can be investigated by applying  
221 Positive Matrix Factorization (PMF) (Paatero and Tapper, 1994) using the Multilinear Engine  
222 (ME-2) (Paatero, 1999) to the organic mass spectra. Both methods describe the measurements  
223 with a bilinear factor model:

224 
$$x_{ij} = \sum_{k=1}^p g_{ik} f_{kj} + e_{ij} \quad (1)$$

225 where  $x_{ij}$  is the  $j^{\text{th}}$  species ( $m/z$ ) concentration measured in the  $i^{\text{th}}$  sample,  $p$  is the  
226 number of sources,  $g_{ik}$  is the contribution of the  $k^{\text{th}}$  source to the  $i^{\text{th}}$  sample,  $f_{kj}$  is the  
227 concentration of the  $j^{\text{th}}$  species in the  $k^{\text{th}}$  source (mass spectra) and  $e_{ij}$  is the residual associated  
228 with the  $j^{\text{th}}$  species concentration measured in the  $i^{\text{th}}$  sample. The values  $g_{ik}$  and  $f_{kj}$  are adjusted  
229 until a minimum for the objective function  $Q$  for a given number of factors  $p$  is found:

230 
$$Q = \sum_{i=1}^n \sum_{j=1}^m \left( \frac{e_{ij}}{\sigma_{ij}} \right)^2 \quad (2)$$

231 where  $\sigma_{ij}$  is the user defined uncertainty for the  $j^{\text{th}}$  species in the  $i^{\text{th}}$  sample.

232 With the ME-2, the user can introduce a priori information about sources e.g. using the  
233 so-called a-value approach. Hence, the user inputs one or more factor profiles and a constraint  
234 defined by the a-value, which determines the extent to which the output profile can differ  
235 from the profile fed to the model.

236 In the present study the source apportionment to OA was performed applying ME-2  
237 using the toolkit SoFi (Source Finder) version 4.7 described in Canonaco et al. (2013). The ME-2  
238 was applied separately for the warm and cold periods in this study, given the expected  
239 differences among them. The warm period was defined as a period with >70% of the days with  
240 average  $T > 19^{\circ}\text{C}$ , hourly max  $T > 24^{\circ}\text{C}$  and hourly min  $T > 15^{\circ}\text{C}$ , which includes 14<sup>th</sup> June to 9<sup>th</sup>  
241 October 2012. The cold period was defined as a period with >70% of the days with average  
242  $T < 10^{\circ}\text{C}$ , hourly max  $T < 13^{\circ}\text{C}$  and hourly min  $T < 8^{\circ}\text{C}$  and includes 28<sup>th</sup> October 2012 to 7<sup>th</sup> April  
243 2013. Only  $m/z \leq 100$  were used for several reasons: a) the signals of  $m/z > 100$  account for a

245 minor fraction of the total signal (2% on average), b) the  $m/z$ >100 have larger uncertainties,  
246 and c) the large interference of naphthalene signals (at  $m/z$  127, 128, and 129) is avoided. The  
247 error matrix was calculated by the aforementioned customized software, which downweights  
248 the  $m/z$  masses calculated from the  $m/z$  44 signal. Moreover,  $m/z$  with signal to noise ratio  
249 (S/N) below 0.2 were downweighted by a factor of 10, and those with S/N between 0.2 and 1  
250 were downweighted by a factor of 2.

251

252 **3 RESULTS AND DISCUSSION**

253 **3.1 Comparison of ACSM data with other measurements**

254 This is one of the few studies, together with Ripoll et al. (2014a), that compare ACSM  
255 data with off-line  $PM_1$  measurements. Most of the studies found in the literature comparing  
256 ACSM data with off-line measurements are based in the  $PM_{2.5}$  fraction for the off-line  
257 measurements. In this study we use  $PM_1$  measurements, avoiding the tail of the coarse mode  
258 that the  $PM_{2.5}$  fraction includes, and hence being closer to the size range measured by the  
259 ACSM (75-650nm).

260 The sum of the ACSM components concentrations and the BC concentrations  
261 measured by the MAAP was compared with  $PM_1$  concentrations determined by the optical  
262 particle counter, resulting in a strong correlation (squared Pearson correlation coefficient,  
263  $R^2=0.66$ ) and a slope very close to unity (1.005) (Figure 1). The application of a time-dependent  
264 collection efficiency (CE) to the ACSM data based on the Middelbrook approach (Middlebrook  
265 et al., 2012) resulted in a better fit compared to the use of a constant CE=0.5 used in several  
266 studies (which resulted in a slope of 0.913 and a  $R^2=0.65$ ). Hence the time-dependent CE  
267 application is considered more suitable for the present study. The time-dependent CE equalled  
268 the default value of 0.45 for most of the period, and increased up to 0.65 during the colder  
269 period (Figure S3). The sum of ACSM components and BC concentrations was also compared to  
270 the mass concentration calculated from SMPS data, resulting in a strong correlation (squared  
271 Pearson correlation coefficient,  $R^2=0.77$ ) and a slope very close to unity (0.997) (Figure S4).  
272 Moreover, ACSM components concentrations were daily averaged and compared to off-line  
273 measurements from 24-h  $PM_1$  samples (Figure 2). All the species, except for chloride, showed  
274 strong correlations ( $R^2$  of 0.68, 0.82 and 0.94 for ammonium, nitrate and sulphate,  
275 respectively). Chloride concentrations were below or close to detection limits for both ACSM  
276 and off-line analysis, which may be the cause for the discrepancies found. Such discrepancies  
277 were also found in other studies (Budisulistiorini et al., 2014). For the strongly-correlated

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282 species, the slopes (ACSM vs off-line measurements) were different for each of them. Whereas  
283 it was close to unity for sulphate (1.15), it was higher for ammonium (1.72), and much higher  
284 for nitrate (2.80). The final reasons for this discrepancy remain unexplained, although a  
285 possible cause is the volatilization of ammonium nitrate from the filters. Nevertheless, the  
286 volatilization of ammonium nitrate is expected to be low given that the samples are kept at  
287 controlled conditions (24-26°C) as described in the methods section. Moreover, if random  
288 volatilization occurred, the correlation coefficients found between ACSM and filters would be  
289 lower. The apparent discrepancy between the slope for total PM<sub>1</sub> (ACSM+BC vs PM<sub>1</sub> from OPC  
290 corrected with gravimetric measurements, close to unity) and the slopes for the different  
291 components (>1) is attributed to the undetermined fraction of PM<sub>1</sub> mass in the filters. Thus,  
292 whereas the ACSM+BC concentrations are strictly the sum of the components, the PM<sub>1</sub>  
293 gravimetric concentrations include a fraction of undetermined mass, partially attributed to  
294 water ([Figure S5](#)).

295 For organic aerosol a strong correlation was found ( $R^2=0.82$ ), and the high slope  
296 obtained (4.25) may be interpreted as the OM-to-OC ratio, since the ACSM measures OA and  
297 the off-line measurements determined OC. This large OM-to-OC ratio suggests  
298 photochemically well-aged organics, but it is too high even for a pure SOA (Aiken et al., 2008),  
299 which is expected to have an important contribution at MSY as will be discussed later (section  
300 3.5), and it is higher than the OM-to-OC ratio determined in March 2009 at Montseny (2.0)  
301 (Minguillón et al., 2011). This extremely large OM-to-OC ratio might be attributed to a)  
302 underestimation of OC due to loss of semi-volatile organic compounds from the filters, and b)  
303 overestimation of OM by the ACSM due to an underestimation of the RIE for organics. The first  
304 reason is expected to be less likely given the strong correlation found between OA and OC  
305 (which would not be so if random volatilization occurred) and given that the samples are kept  
306 at controlled conditions, as formerly explained, hence reducing the possible volatilization.  
307 Previous studies also found higher than expected OM-to-OC ratios when comparing ACSM OA  
308 with off-line OC measurements. Budisulistiorini et al. (2014) found OM-to-OC ratios of 4.85  
309 and 3.85 in summer and fall, respectively. Ripoll et al. (2014a) found an OM-to-OC ratio of 3.39  
310 for a one year sampling period. This topic is currently being investigated by the ACSM  
311 manufacturer. The calculation of OM-to-OC ratio from the f44 based on Aiken et al. (2008) was  
312 not carried out given that it is not suitable for ACSM instruments, as recently learnt from an  
313 intercomparison of 13 Q-ACSM instruments (Fröhlich et al., 2015).

314

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316 **3.2 Time series and average composition of submicron aerosol. Seasonal variation**

317 The average concentration (P25, P75) of the ACSM components plus BC concentrations  
318 during the study period was  $7.3 \mu\text{g m}^{-3}$  (3.1, 10.2). The highest concentrations were measured  
319 during the warm periods (average  $10.3 \mu\text{g m}^{-3}$ ), defined as the periods with most of the days  
320 with average  $T > 20^\circ\text{C}$  (from 14<sup>th</sup> June to 9<sup>th</sup> October 2012 and from 13<sup>th</sup> June to 9<sup>th</sup> July 2013).  
321 The lowest concentrations were recorded during the cold period (average  $5.8 \mu\text{g m}^{-3}$ ), which  
322 includes a period with most of the days with average  $T < 13^\circ\text{C}$  (from 28<sup>th</sup> October 2012 to 7<sup>th</sup>  
323 April 2013) (Figure 3). The average monthly concentrations, following the described variation,  
324 can be seen in Figure 4. This is in agreement with the seasonal variations observed during a  
325 long time period (2002-2010) by Cusack et al. (2012). The summer increase is associated with  
326 the recirculation of air masses that prevent air renovation, the low precipitation (Figure S2),  
327 and the formation of secondary aerosols enhanced by the maximum solar radiation (Figure  
328 S2). The lower winter concentrations can be explained by the high frequency of Atlantic  
329 advections (Figure S1) and the higher precipitation rates, although occasional high  
330 concentrations are attributed to winter anticyclonic scenarios (Pey et al., 2010). The seasonal  
331 variation of  $\text{PM}_1$  concentrations at MSY is also influenced by the evolution of the boundary  
332 layer height, which is lower during wintertime and increases during summertime, especially  
333 during the central hours of the day. Changes in the origin of air masses also determined the  
334 seasonal variation of  $\text{PM}_1$  concentrations.

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Eliminado: Figure 4

Eliminado: Figure S2

Eliminado: Figure S2

Eliminado: Figure S1

335 On average, the most abundant component was OA ( $3.8 \mu\text{g m}^{-3}$ ), followed, in this  
336 order, by sulphate ( $1.3 \mu\text{g m}^{-3}$ ), ammonium ( $0.8 \mu\text{g m}^{-3}$ ), nitrate ( $0.8 \mu\text{g m}^{-3}$ ), BC ( $0.4 \mu\text{g m}^{-3}$ )  
337 and chloride ( $<0.1 \mu\text{g m}^{-3}$ ). The OA contribution varied throughout the year, reaching 60% of  
338 the total  $\text{PM}_1$  in the summer period (June, July and August) and decreasing progressively down  
339 to 43% in February (Figure 4). The contribution of sulphate followed the same seasonal  
340 variation, from about 20% in the warmer months to about 8% in the colder months. The  
341 nitrate contribution showed an inverse trend, with higher relative contributions in the winter  
342 and much lower in summer. These seasonal variations were already observed in previous  
343 studies using off-line filter sampling (Pey et al., 2009; Ripoll et al., 2014b) and can be attributed  
344 to a higher SOA contribution, favoured formation of sulphate, and nitrate gas/aerosol  
345 partitioning leading to vaporization of ammonium nitrate during the warmer period.

Eliminado: Figure 4

346 When investigating the diurnal patterns, it is observed that OA, nitrate and BC  
347 concentrations reach the maximum at around 14h UTC in summer, whereas sulphate and  
348 ammonium show a delayed increase in their concentrations, peaking at around 16h UTC  
349 (Figure 5). The reasons for this shift may obey to the different origin of each component.

Eliminado: Figure 5

357 Whereas the OA, nitrate and BC are transported with the breeze from the populated areas and  
358 the valley towards the regional background site, the sulphate can also be transported from  
359 further away, i.e. from over the Mediterranean Sea due to shipping emissions. Later in the day,  
360 when the breeze is developed in the opposite direction (from inland towards the coast), the  
361 concentrations of OA, BC and nitrate decrease, whereas the sulphate and ammonium  
362 concentrations remained high for longer time (until about 19h UTC). This is due to the more  
363 regional character of ammonium sulphate, which is present in a wider area due to its longer  
364 lifetime in the atmosphere (Seinfeld and Pandis, 2006) and hence remains longer at MSY. In  
365 addition to the transport of pollutants, local SOA can be formed (see section 3.5 for  
366 discussion). Specific episodes may differ from this average behaviour owing to specific  
367 atmospheric characteristics, for which sulphate concentrations increase simultaneously with  
368 OA, but the most common variation is the one described here. On the other hand, in winter, all  
369 the components show an increase at around 15h UTC, and concentrations remain high until  
370 around 22h UTC, when they start to decrease to reach a minimum around 9h UTC ([Figure 5](#)).  
371 This simultaneous variation indicates that the pollutants are transported from the nearby  
372 polluted areas to MSY with the breeze.

### Eliminado: Figure 5

### 374 3.3 Influence of the type of scenario on submicron aerosol

The total PM<sub>1</sub> concentrations were investigated as a function of the type of scenario, finding the lowest concentrations during Mediterranean episodes and Atlantic advections, and the highest during North African outbreaks, European episodes and Winter Anticyclonic episodes (Figure 6). Some differences in the relative chemical composition as a function of the type of scenario were found (Figure 6). OA and sulphate relative contributions were higher under regional and North African episodes. This may be due to the higher formation of secondary aerosols enhanced by the higher temperature and solar radiation during these episodes. See additional discussion about formation of SOA in section 3.5. Moreover, the higher sulphate concentrations under regional episodes may also be due to the enhanced regional mixing, as shown by the flatter diurnal pattern shown for this pollutant (Figure S6). Sulphate relative contribution was also high when Mediterranean air masses affected MSY probably owing to the impact of shipping emissions. On the other hand, nitrate relative contribution was found to be higher for Winter Anticyclonic and European episodes. For both cases the colder weather compared to the rest of the year is partly responsible for the higher nitrate concentrations. During Winter Anticyclonic episodes, the stagnant conditions favouring the accumulation of polluted air masses that are transported from the Barcelona metropolitan

### Eliminado: Figure 6

### Eliminado: Figure 6

### Eliminado: Figure S6

395 area towards MSY may also be responsible for the high nitrate concentrations (Pey et al.,  
396 2010). Note that this transport takes place later in the day than in warm conditions, thus  
397 reaching the maximum concentrations between 15h UTC and 22h UTC, and that the day-night  
398 difference is much higher than for other scenarios ([Figure S6](#)). During European episodes, the  
399 higher nitrate concentration can be attributed to the long range transport of nitrate from  
400 Europe to the study area, although this type of episodes often take place under anticyclonic  
401 conditions and hence the nitrate may have a local origin at lower heights, whereas European  
402 nitrate is transported at higher altitudes, as it was seen by Ripoll et al. (2014b).

Eliminado: Figure S6

403 The relative chemical composition as a function of the total concentration was also  
404 investigated, but no clear patterns were identified, meaning that there is not a prevalent  
405 component for low or for high concentrations ([Figure S7](#)).  
406

Eliminado: Figure S7

#### 407 3.4 Wildfire episode

408 A wildfire episode took place from 22 to 26 July 2012 ([Figure 7](#)) at 100 km to the NE of  
409 MSY, affecting a wide area ([Figure S8](#)). It resulted in an average ACSM components + BC  
410 concentrations of  $16.5 \mu\text{g m}^{-3}$  over the five-days period. Mainly, the components whose  
411 concentrations increased significantly were OA, nitrate and BC, reaching 30-min values of  $50 \mu\text{g m}^{-3}$ ,  
412  $4.5 \mu\text{g m}^{-3}$  and  $3.6 \mu\text{g m}^{-3}$ , respectively, which are 9, 8 and 7 times higher than their  
413 respective summer averages. The average relative concentration during this episode was  
414 dominated by OA (73%).

Eliminado: Figure 7

Eliminado: Figure S8

415 In order to investigate the wildfire source, an unconstrained source apportionment  
416 (PMF) of the organic aerosol fraction during this episode was carried out. The PMF resulted in  
417 two factors, one representing the fresh biomass burning organic aerosol (named as  
418 BBOA\_MSY) and another one interpreted as the mix of other OA sources and aged BBOA,  
419 named as OOAm (where the m stands for mix). The interpretation of the factors is based on  
420 their mass spectral source profiles and the time series of their contributions. The BBOA\_MSY  
421 shows higher peaks for the specific tracers of biomass burning ( $m/z$  60 and  $m/z$  73) (Alfarra et  
422 al., 2007) than the OOAm, which indicates that the primary BBOA is well represented by this  
423 factor. Moreover, the f44 (ratio of  $m/z$  44 (mostly  $\text{CO}_2^+$ ) to total signal in the component mass  
424 spectra), an indicator of oxygenated organic species (Alfarra et al., 2007), was higher for the  
425 OOAm than for the BBOA\_MSY factor, which indicates that this factor corresponds to a more  
426 oxidized aerosol. On the other hand, the f43 is higher than the f44 in the BBOA\_MSY factor,  
427 whereas it is the other way around (f44 much higher than f43) in the OOAm. These differences  
428 in relative intensities indicate the differences in the age of the aerosol (Ng et al., 2010) and

433 further lead to differentiate the factors as fresh BBOA and OOAm. The SOA formation from  
434 biomass burning has been reported to be quick (Heringa et al., 2011), and hence part of the  
435 OOAm factor is formed of aged BBOA, which also explains that the time series of the OOAm  
436 factor partially tracks that of the BBOA\_MSY. The BBOA\_MSY profile found here is very similar  
437 to a BBOA profile found for Montseny in March 2009 (Minguillón et al., 2011)  
438 (<https://sites.google.com/site/amsglobaldatabase>) and to an average profile for BBOA from  
439 various datasets (Ng et al., 2011a) (Figure S9). The f60 in the BBOA\_MSY factor is 0.014, similar  
440 to the f60 in these other two BBOA profiles (0.017 and 0.024). It has been also compared to  
441 the BBOA found in the background of Paris (Crippa et al., 2013), with which some more  
442 differences were found, mainly our profile has higher  $m/z$  43 and  $m/z$  41 signals and lower  $m/z$   
443 60 (Figure S9). This BBOA\_MSY mass spectrum is considered specific for the study area and  
444 hence it can be later used for other studies in the region, to be fed to the ME-2 model in order  
445 to quantify the BBOA contribution. We have done so in the present study for the winter  
446 period. Whereas the time series of both factors were similar, the BBOA\_MSY contribution  
447 showed more intense peaks, and the increase in the OOAm was slightly higher for the second  
448 part of the main peak on the 23 July.

449

### 450 3.5 Source apportionment of organic aerosol

451 The source apportionment of organic aerosol was carried out separately for the  
452 warmer period (14 June to 9 October 2012) and the colder period (28 October 2012 to 7 April  
453 2013). The days of the wildfire event were excluded from the warmer period dataset. The  
454 separation in two seasons was done to better characterize the source profiles of the different  
455 sources, especially the different types of OOA, given that it is expected to vary throughout the  
456 year.

457 The application of ME-2 to the warmer period resulted in a solution with 3 factors: a  
458 hydrocarbon-like OA (HOA), a semi-volatile oxygenated OA (SV-OOA) and a low-volatile  
459 oxygenated OA (LV-OOA). This solution was chosen based on several tests with different  
460 number of factors and different a-values for the constrained factors, taking into account the  
461 correlations with external data, the diurnal patterns and the residuals, following the strategy  
462 described by Crippa et al. (2014) and Canonaco et al. (2013). The HOA factor was constrained  
463 using an average HOA factor (HOA\_avg) from different datasets (Ng et al., 2011a). An a-value  
464 range from 0.05 to 0.3 was explored and an a-value of 0.2 was finally selected, which was a  
465 compromise between a higher squared Pearson correlation coefficient between HOA and BC  
466 (which increased when increasing the a-value) and the physically meaningful profiles of the

Eliminado: Figure S9

Eliminado: Figure S9

469 whole solution (i.e. assessing the profiles of the LV-OOA and SV-OOA factors). Hence, the a-  
470 value 0.05 did not reach converge, and the a-value 0.1 resulted in not-well-resolved OOA  
471 factors. BC concentrations correlated moderately with HOA of the chosen solution (squared  
472 Pearson coefficient  $R^2= 0.51$ , Table S1). The SV-OOA shows higher 43-to-44 ratio compared to  
473 the LV-OOA, together with a lower f44, which are the main differences between these two  
474 profiles (Figure 8a). The BBOA contribution in summer is expected to be low based on previous  
475 studies carried out in July 2009 (Minguillón et al., 2011) and on the low f60 registered in the  
476 present study in summer (Figure S10), which is below the background threshold (0.003) Eliminado: Figure 8  
477 established by Cubison et al. (2011). Hence, the BBOA factor was not identified and it was not  
478 constrained by the ME-2 in summer. The correlations of the two OOA factors with external  
479 secondary pollutants is not very high (LV-OOA with ammonium shows  $R^2=0.43$ , LV-OOA with  
480 sulphate shows  $R^2=0.34$ , and SV-OOA with nitrate shows  $R^2=0.16$ , Table S1). Nevertheless, the  
481 source profiles are well defined. The 4-factors solution was investigated and resulted in a split  
482 of the LV-OOA factor. The residuals of the 3-factors solution showed an increase between 12h  
483 and 15h (which did not disappear in the 4-factors solution), hence indicating the difficulty in  
484 explaining the variation of the OA during the middle hours of the day. This pattern was also  
485 observed for the  $m/z$  43 variable residuals, whereas it was not observed for other variables.  
486 Therefore, the difficulty in the OA explanation may be more related to the SOA formation  
487 processes rather than to primary OA sources. Further details on the chosen solution can be  
488 found in the supplementary material (Figures S11 to S15).

489 During the warmer period, the HOA accounted for 13% ( $0.7 \mu\text{g m}^{-3}$ ), whereas the LV-  
490 OOA and the SV-OOA accounted for 44% and 41% of the total OA ( $2.5 \mu\text{g m}^{-3}$  and  $2.3 \mu\text{g m}^{-3}$ ),  
491 respectively, hence remaining 2% of unexplained OA mass (Figure 8c). The time series is shown Eliminado: Figure 8  
492 in Figure S16. As explained before, the location and meteorological conditions at MSY result in Eliminado: Figure S16  
493 an increase of pollutants concentrations starting at mid-morning, caused by the breeze  
494 transport from populated areas to the regional site. This variation is clearly observed for BC  
495 (Figure 9), which showed a moderate correlation with HOA ( $R^2= 0.51$ ). Nevertheless, the Eliminado: Figure 9  
496 midday increase in the concentration of SV-OOA is larger than that of BC, and therefore it  
497 cannot be only explained by the transport of pollutants, including the SOA formed during the  
498 transport, but it is attributed to the formation of SOA during these hours in MSY. Hence, the  
499 SOA formation can be estimated as the additional increase with respect to that of BC  
500 (considered in % of the average concentration during the night hours), which results in a local  
501 SOA formation of  $1.1 \mu\text{g m}^{-3}$ . This SOA may result mainly from biogenic precursors, in  
502 agreement with the 70% of non-fossil SOA found in March 2009 (Minguillón et al., 2011). The  
503 flatter diurnal pattern of LV-OOA (Figure 9) points to a more regional and well-oxidized Eliminado: Figure 9

510 aerosol, which could be interpreted as the regional background SOA. The difference in the  
511 diurnal patterns between LV-OOA and SV-OOA is more evident during the regional episodes,  
512 evidencing the mixing of the background pollutants (LV-OOA) and the formation of SOA on site  
513 (part of the SV-OOA) ([Figure S17](#)). This SOA formation during warm periods was also observed  
514 by Cusack et al. (2013), who studied nucleation and particle growth events, identifying both of  
515 them even under polluted conditions at MSY.

Eliminado: Figure S17

516 In the colder period, the application of ME-2 resulted in a solution with 3 factors:  
517 hydrocarbon-like OA (HOA), biomass burning OA (BBOA) and oxygenated OA (OOA) ([Figure](#)  
518 [8b](#)). A solution with two OOA factors was investigated and it was not meaningfully  
519 interpretable. Probably the small temperature range variation in winter results in not enough  
520 diurnal variation in f43 and f44 for a split of the OOA in SV-OOA and LV-OOA. As per the  
521 warmer period, the final solution was chosen based on the strategy described by Crippa et al.  
522 (2014) and Canonaco et al. (2013). For coherence with the warmer period, the HOA factor was  
523 based in an average HOA factor (HOA\_avg) from different datasets (Ng et al., 2011a), and it  
524 was constrained with an a-value of 0.1. This a-value was chosen based on the correlation  
525 between the HOA contribution and the BC concentrations found for different a-values tests,  
526 the source profiles obtained for the rest of the sources (i.e. assuring that they were  
527 meaningful), and a preference for relatively low a-values to avoid mixing of OOA sources into  
528 the HOA source. The HOA contribution of the chosen solution shows a relatively strong  
529 correlation with BC concentrations (squared Pearson coefficient  $R^2=0.70$ , Table S1). The HOA  
530 spectral profiles found for summer and winter are quite similar, and hence the HOA  
531 contributions in summer and winter can be compared directly. The BBOA factor was decided  
532 to be constrained based on the f60 signal, which was above the aforementioned threshold of  
533 0.003 ([Figure S10](#)). It was based in the BBOA\_MSY profile found for the wildfire episode that

Eliminado: Figure S10

534 took place during this study, constrained with an a-value of 0.1. The a-value was chosen with  
535 the following criteria: preference for a low a-value given that the anchor profile used was site-  
536 specific, residuals for the  $m/z$  60 not showing any diurnal pattern, contribution of the BBOA  
537 factor to the total  $m/z$  60 (which reached 64% for the chosen solution). The resulting BBOA  
538 profile has a higher  $m/z$  44 signal than the BBOA\_MSY, which may indicate differences in the  
539 biomass burning emissions from the wildfire event compared to the emissions from regular  
540 biomass burning, or it could indicate that the BBOA contribution identified here may be  
541 partially mixed with some oxidized OA. The single winter OOA factor identified shows higher  
542 f44 than both LV-OOA and SV-OOA in summer. This higher degree of oxidation of the OA in  
543 winter indicates that there is less newly-formed SOA during winter compared to summer. A  
544 similar variation was observed in Zurich (Canonaco et al., 2014). The OOA contribution

548 correlates moderately with sulphate ( $R^2=0.49$ ), relatively strongly with nitrate ( $R^2=0.73$ ) and  
549 more strongly with ammonium ( $R^2=0.79$ , Table S1). Further details on the chosen solution can  
550 be found in the supplementary material (Figures S11 to S15).

551 The major OA constituent in winter was the OOA, with 59% ( $1.5 \mu\text{g m}^{-3}$ ), whereas the  
552 HOA and BBOA accounted for 12% ( $0.3 \mu\text{g m}^{-3}$ ) and 28% ( $0.7 \mu\text{g m}^{-3}$ ) of the total OA,  
553 respectively, with only 1% of the OA mass remaining unexplained (Figure 8d). Note that the  
554 BBOA contribution may be mixed with some OOA as stated before, given the relatively high  
555 signal at  $m/z$  44 and hence the pure BBOA contribution would be lower than that determined.  
556 Actually it accounts for 6% of the total signal at  $m/z$  44. Nevertheless, strong correlation  
557 ( $R^2=0.77$ ) was found between the BBOA contribution and the potassium concentrations  
558 determined in 24-h  $\text{PM}_1$  samples (Figure S18), which further confirms the existence of this  
559 source at MSY in winter. The relative BBOA contributions found in the present study are similar  
560 to those found in a previous study in March 2009 using a HR-ToF-AMS, where the HOA  
561 represented 7% of the total OA, the BBOA contributed with 9% and the rest was attributed to  
562 OOA (Minguillón et al., 2011; Crippa et al., 2014). The discrepancy in the BBOA contribution  
563 (29% vs 9%) may be due to the different sampling periods (the current study included Nov  
564 2012–March 2013 whereas the previous study only included March 2009), to the mixture of  
565 some OOA in the BBOA factor for the present study, and/or to the possible increase of biomass  
566 burning due to the climate and energy policies in the last five years.

567 The average daily pattern shown by the different OA sources in winter (Figure 9)  
568 resembles that of BC, nitrate, sulphate and ammonium (Figure 5), with an increase of  
569 pollutants concentrations starting at around 10h UTC and reaching high concentrations at  
570 around 13h UTC. No significant differences in the daily pattern were observed for different  
571 meteorological episodes, other than the different concentrations (Figure S17). This daily  
572 increase is attributed to the transport from populated areas to the mountain site with the  
573 breeze. This variation is observed for all the components and therefore the local formation of  
574 SOA is deduced to be low in winter.

575

#### 576 4 CONCLUSIONS

577 The deployment of an ACSM at the regional background site of Montseny during one  
578 year allowed for the characterization of  $\text{PM}_1$  composition and its variation as a function of time  
579 of the year and atmospheric scenarios. The OA sources were also identified and studied.

580 Strong correlation ( $R^2=0.66$ ) was found between total mass determined by ACSM  
581 components + BC and  $\text{PM}_1$  determined by an optical particle counter with a slope near to

Eliminado: Figure 8

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587 unity. The suitability of the application of a composition-dependent collection efficiency (CE)  
588 was confirmed.

589 Strong correlations were found between the ACSM measurements and off-line  
590 measurements (filters) for sulphate ( $R^2=0.93$ ), ammonium ( $R^2=0.68$ ) and nitrate ( $R^2=0.82$ ).  
591 Nevertheless the slopes differ more than 20% from the unity for nitrate and ammonium.

592 The comparison of the OA measured by the ACSM with the OC measured in filter  
593 samples points to a current underestimation of the RIE established for OA.

594 A wildfire episode affected significantly the organic aerosol concentrations. The source  
595 profile of fresh BBOA for this specific episode was characterized and it resembles those from  
596 other studies. The BBOA contribution in winter correlates with the  $PM_{1}$  potassium  
597 concentrations.

598 OA was the major component of submicron aerosol on average and especially during  
599 the warm periods. Three organic sources were identified by PMF in summer: HOA, SV-OOA  
600 and LV-OOA; and three sources in winter: HOA, BBOA and OOA. SOA was the major  
601 constituent of the OA at MSY, being more than 80% of total OA in summer and about 60% in  
602 winter. The in-situ formation of SOA in summer, happening around midday, was estimated to  
603 be  $1.1 \mu g m^{-3}$  on average (20% of OA).

604 Sulphate concentrations were higher in summer, while nitrate concentrations were  
605 higher in winter due to environmental conditions (temperature, relative humidity and solar  
606 radiation, among others). Sulphate originates from a wider area and is affected by the shipping  
607 emissions from the Mediterranean, while the rest of the components may have a nearer  
608 origin.

609 As typical for mountain sites, all the pollutants were affected by the general breeze  
610 regime, leading to an increase from mid-morning until the afternoon, and a decrease until the  
611 evening.

612

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621

622 **FIGURE CAPTIONS**

623 **Figure 1.** ACSM components + BC concentrations vs PM<sub>1</sub> measured by the optical counter coloured by  
624 the sampling time (dd/mm/aaaa). Line and parameters correspond to least orthogonal distance fit  
625 (y=a+bx). The wild fire period is excluded from the fit.

626 **Figure 2.** ACSM components concentrations vs 24-h samples concentrations. Lines and parameters  
627 correspond to least orthogonal distance fits.

628 **Figure 3.** Time series of ACSM components and BC concentrations during the whole study period.

629 **Figure 4.** Monthly relative chemical composition of submicron aerosol. The numbers on top of each  
630 bar represent the average monthly concentration. n is the number of data points for each month  
631 (right axis).

632 **Figure 5.** Average daily pattern for (a) the warmer and (b) the colder periods. Note that OA is plotted  
633 in the right axis.

634 **Figure 6.** Relative chemical composition of submicron aerosol as a function of the type of scenario.  
635 The numbers on top of each bar represent the average concentration for each type of scenario. n is  
636 the number of data points for each type of scenario (right axis).

637 **Figure 7.** (a) Time series of ACSM components and BC concentrations and pie chart of the average  
638 chemical composition during the wildfire episode from 22<sup>nd</sup> to 26<sup>th</sup> July 2012. (b) Time series of the  
639 contribution of the BBOA\_MSY and Aged BBOA sources identified by PMF. (c) Source profile of the  
640 BBOA\_MSY and Aged BBOA sources.

641 **Figure 8.** Mass spectral profiles of the organic sources identified for (a) summer and (b) winter.  
642 Average contribution of the organic sources to total OA for (c) summer and (d) winter. The white  
643 fraction of the pie charts corresponds to unexplained mass.

644 **Figure 9.** Average daily patterns of the organic sources contributions and BC concentrations for (a)  
645 summer and (b) winter. Error bars represent standard deviations.

646

647 **SUPPLEMENTARY MATERIAL**

648 **Figure S1.** Frequency of type of scenario for each of the months of the study period.

649 **Figure S2.** Temperature, relative humidity, wind direction and speed, solar radiation and precipitation  
650 hourly data at MSY during the study period.

651 **Figure S3.** Time-dependent CE calculated with the Middlebrook approach (Middlebrook et al., 2012).

652 **Figure S4.** ACSM components + BC concentrations vs mass concentration calculated from Scanning  
653 Mobility Particle Sizer (SMPS) data coloured by the sampling time (dd/mm/aaaa). Data availability for  
654 SMPS data covered only 2012 period. Line and parameters correspond to least orthogonal distance fit  
655 (y=a+bx). The wild fire period is excluded from the fit.

656 **Figure S5.** Schematic comparison of ACSM components + BC concentrations vs PM<sub>1</sub> concentrations  
657 from OPC corrected with gravimetric determinations. The numbers indicate the slopes found for  
658 experimental data for Montseny during June 2012 to July 2013. The 2.25 corresponds to the slope of  
659 OA (ACSM) vs OM estimated from OC (filters) as 2\*OC.

660 **Figure S6.** Average daily pattern for (a) Atlantic advective (ATL), (b) North African episodes (NAF), (c)  
661 regional episodes (REG), and (d) Winter Anticyclonic episodes (WAE). Note that OA is plotted in the  
662 right axis.

663 **Figure S7.** Relative chemical composition as a function of average concentration and number of data  
664 points for each range of concentrations. No clear differences are observed.

665 **Figure S8.** Total optical depth, sulphate surface concentration, dust surface concentration, and smoke  
666 surface concentration from the NAAPS model for 23, 24 and 25 July 2012 (wildfire event) (a-c), and  
667 satellite images from 22 and 23 July 2012 from The Earth Observing System Data and Information  
668 System (EOSDIS), NASA's Earth Science Data Systems Program (d, e).

669 **Figure S9.** Comparison of the BBOA factor found for the wildfire episode (BBOA\_MSY) with other  
670 BBOA profiles found in the literature (Ng et al., 2011a; Crippa et al., 2014).

671 **Figure S10.** Time series of f60 (unitless) and OA concentration ( $\mu\text{g m}^{-3}$ ) throughout the study period at  
672 MSY. Dashed line corresponds to the 0.5% threshold for the f60 determined by Cubison et al. (2011).

673 **Figure S11.** Q/Qexp vs number of factors for the warmer and the colder periods. The grey dot in the  
674 colder period corresponds to a solution with an a-value of 0.2 for the BBOA factor, given that there  
675 was no convergence with an a-value of 0.1 for the 2-factors solution.

676 **Figure S12.** Weighted residuals vs  $m/z$  for the warmer and the colder periods.

677 **Figure S13.** Time series of the measured and reconstructed OA concentrations for the warmer and the  
678 colder periods.

679 **Figure S14.** Time series of Q/Qexp for the warmer and the colder periods.

680 **Figure S15.** Q/Qexp vs  $m/z$  for the warmer and the colder periods.

681 **Figure S16.** Time series of the OA sources in the warmer (top) and the colder (bottom) periods.

682 **Figure S17.** Average daily pattern for Atlantic advections (ATL), for the warmer and the colder periods,  
683 regional episodes (REG) (only warmer period), and Winter Anticyclonic episodes (WAE) (only colder  
684 period).

685 **Figure S18.** Contribution of BBOA in winter (averaged to 24-h periods matching the filter sampling) vs  
686 potassium concentrations in  $\text{PM}_{1}$ .

687 **Table S1.** Squared Pearson correlation coefficients between OA sources/types and BC, sulphate,  
688 nitrate and ammonium for the warmer and the colder periods

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690

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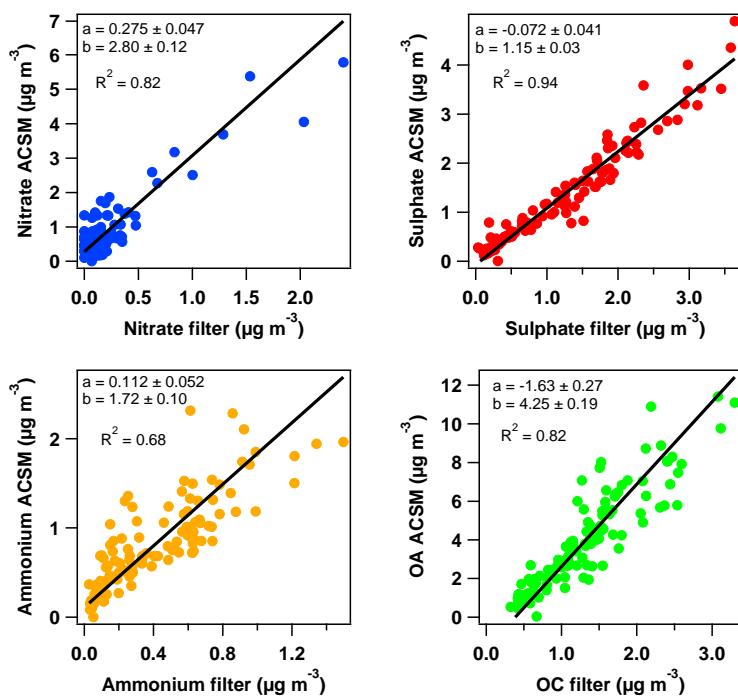
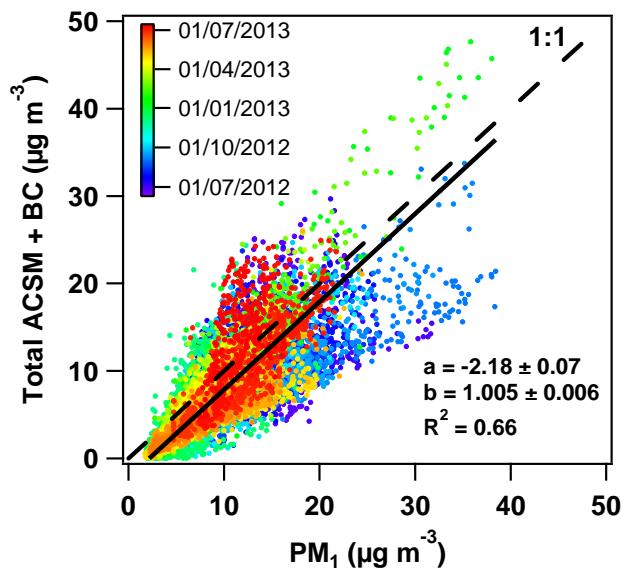
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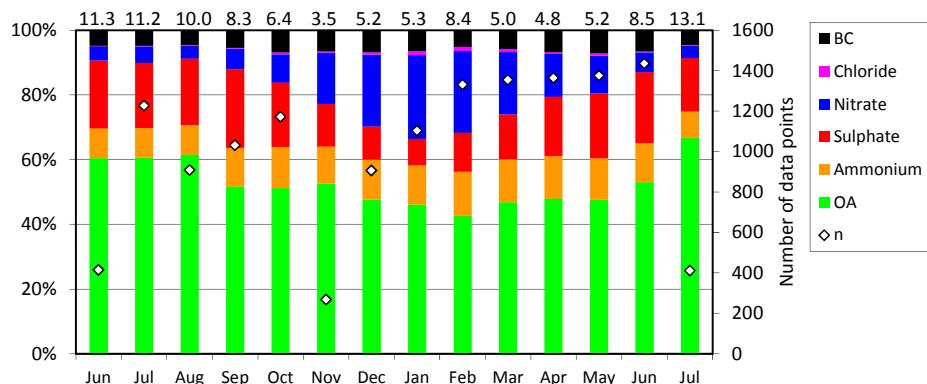
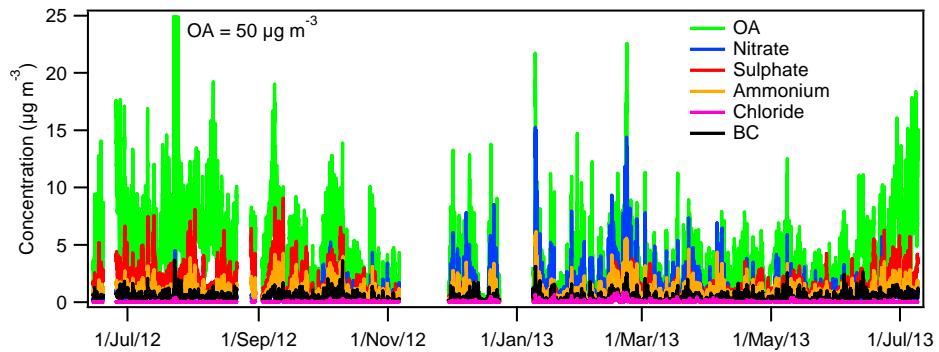
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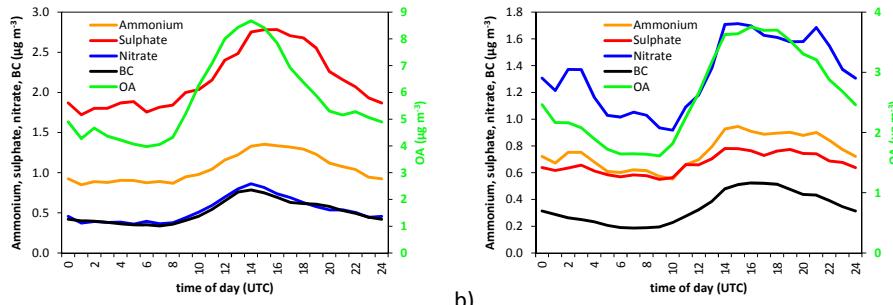
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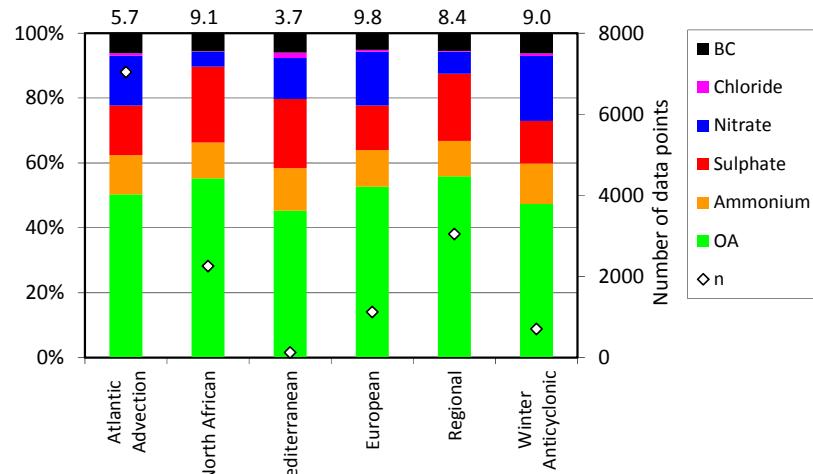


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933 **Figure 5**

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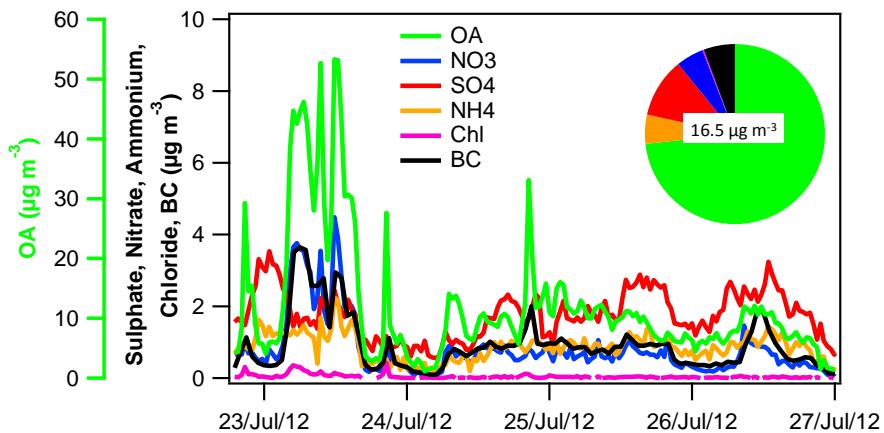
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937 **Figure 6**

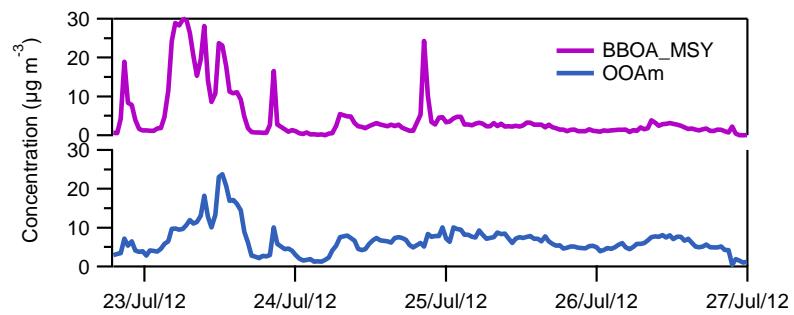
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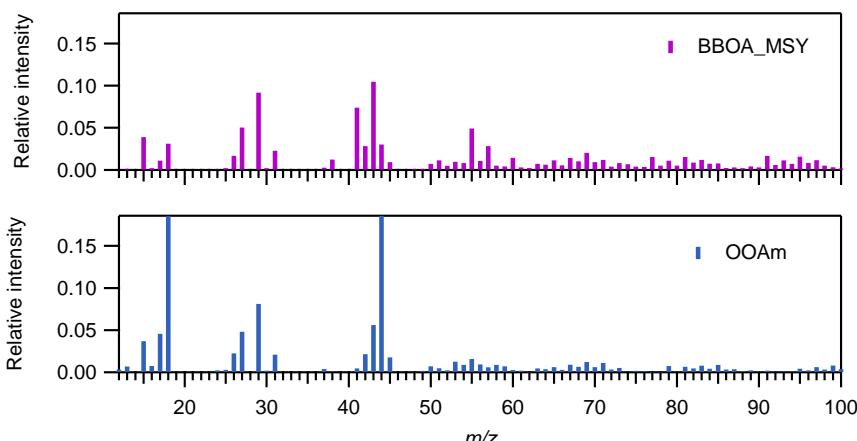
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941 a)



942 b)



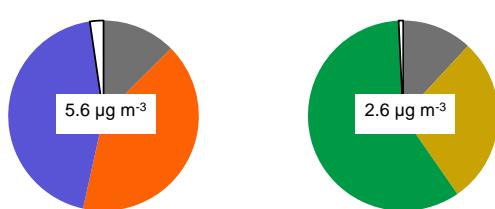
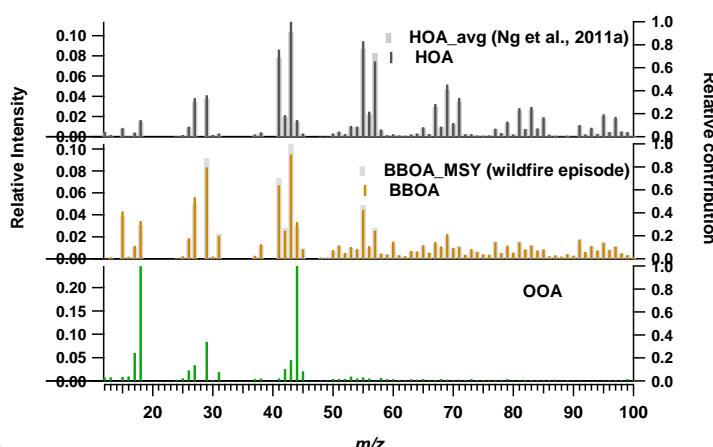
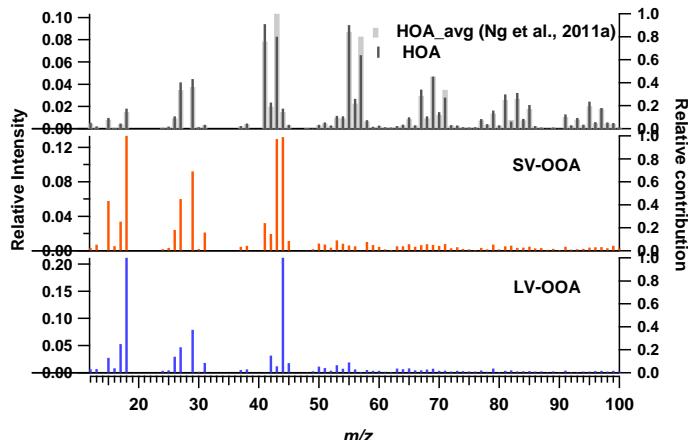
943 c)

944 **Figure 7**

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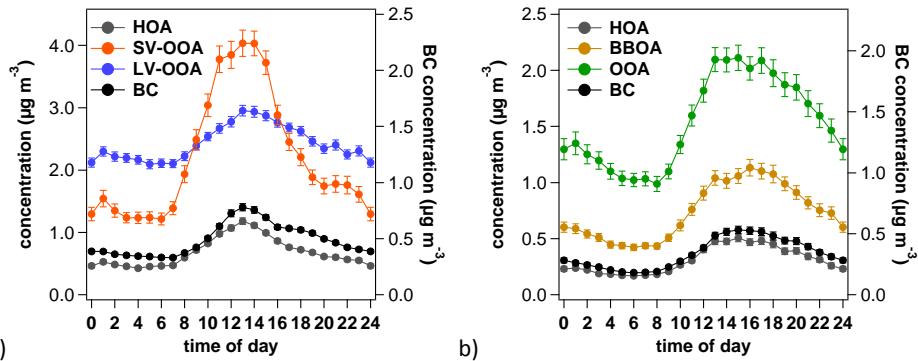
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952 **Figure 8**

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955 a)

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957 **Figure 9**

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