1	Seasonal variation of secondary organic aerosol tracers in Central
2	Tibetan Plateau
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21 Abstract

22 Secondary organic aerosol (SOA) affects the earth's radiation balance and global climate. 23 High-elevation areas are sensitive to global climate change. However, at present, SOA origins and 24 seasonal variations are understudied in remote high-elevation areas. In this study, particulate samples 25 were collected from July 2012 to July 2013 at the remote Nam Co (NC) site, Central Tibetan Plateau 26 and analyzed for SOA tracers from biogenic (isoprene, monoterpenes and β -caryophyllene) and 27 anthropogenic (aromatics) precursors. Among these compounds, isoprene SOA (SOAI) tracers represented the majority (26.6 \pm 44.2 ng m $^{\text{-3}}$), followed by monoterpene SOA (SOA_M) tracers (0.97 \pm 28 0.57 ng m⁻³), aromatic SOA (SOA_A) tracer (2,3-dihydroxy-4-oxopentanoic acid, DHOPA, 0.25 ± 0.18 29 ng m⁻³) and β -caryophyllene SOA tracer (β -caryophyllenic acid, 0.09 ± 0.10 ng m⁻³). SOA_I tracers 30 31 exhibited high concentrations in the summer and low levels in the winter. The similar temperature 32 dependence of SOA_I tracers and isoprene emission suggested that the seasonal variation of SOA_I 33 tracers at the NC site was mainly influenced by the isoprene emission. The ratio of high-NOx to 34 low-NOx products of SOA_I (2-methylglyceric acid to 2-methyltetrols) was the highest in the winter and 35 the lowest in the summer, due to the influence of temperature and relative humidity. The seasonal 36 variation of SOA_M tracers was impacted by monoterpenes emission and gas-particle partitioning. 37 During the summer to the fall, temperature effect on partitioning was the dominant process influencing SOA_M tracers' variation; while the temperature effect on emission was the dominant process 38 39 influencing SOA_M tracers' variation during the winter to the spring. SOA_M tracer levels did not elevate 40 with increased temperature in the summer, probably resulting from the counteraction of temperature 41 effects on emission and partitioning. The concentrations of DHOPA were 1-2 orders of magnitude 42 lower than those reported in the urban regions of the world. Due to the transport of air pollutants from 43 the adjacent Bangladesh and the northeastern India, DHOPA presented relatively higher levels in the 44 summer. In the winter when air masses mainly came from the northwestern India, mass fractions of 45 DHOPA in total tracers increased, although its concentrations declined. The SOA-tracer method was 46 applied to estimate secondary organic carbon (SOC) from these four precursors. The annual average of SOC was $0.22 \pm 0.29 \ \mu gC \ m^{-3}$, with the biogenic SOC (sum of isoprene, monoterpenes and 47 48 β -caryophyllene) accounting for 75%. In the summer, isoprene was the major precursor with its SOC 49 contributions of 81%. In the winter when the emission of biogenic precursors largely dropped, the 50 contributions of aromatic SOC increased. Our study implies that anthropogenic pollutants emitted in

- 51 the Indian subcontinent could be transported to the TP and have impact on SOC over the remote NC.
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53 Keywords: Secondary organic aerosol, Tibetan Plateau, Isoprene, Monoterpenes, Aromatics

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55 1. Introduction

Organic aerosol affects the earth's radiation balance and global climate. As a large fraction of organic aerosol, secondary organic aerosol (SOA) is produced by homogenous (Claeys *et al.* 2004) and heterogeneous (Jang *et al.* 2002) reactions of volatile organic compounds (VOCs) as well as aging of organic aerosol (Robinson *et al.* 2007, Donahue *et al.* 2012). The global emission of biogenic VOCs (BVOCs), such as isoprene and monoterpenes (Guenther *et al.* 1995) were estimated to be one order of magnitude higher than those of anthropogenic sources (Piccot *et al.* 1992). Thus, global SOA is believed to be largely from BVOCs.

63 SOA tracers from specific VOCs can provide insight on processes and sources influencing SOA 64 formation and spatiotemporal distribution. The identification of the isoprene SOA (SOAI) tracers, 65 2-methyltetrols (Claeys et al. 2004) revealed the importance of SOA₁ in global SOA burden. The 66 further studies in high-NOx and low-NOx products of isoprene intermediates (e.g. methacrylic acid 67 epoxide and isoprene epoxydiols) provided more details in the mechanisms of SOA₁ formation under 68 the influence of NOx (Paulot et al. 2009, Froyd et al. 2010, Surratt et al. 2010, Lin et al. 2013). The 69 identification of tracers from aromatic SOA (SOA_A) (Offenberg *et al.* 2007) offered a way to directly 70 evaluate the variation of anthropogenic SOA, particularly in urban regions. In addition, specific tracers 71 have been determined in monoterpene SOA (SOA_M) (Jaoui et al. 2005, Claeys et al. 2007) and 72 β -carvophyllene SOA (SOA_c) (Jaoui *et al.* 2007, van Eijck *et al.* 2013). Based on these SOA tracers, 73 Kleindienst and coworkers further developed an SOA-tracer method to attribute SOA sources in the 74 ambient air. Since it is difficult to directly measure SOA, the SOA-tracer method provides a valuable 75 technique to estimate SOA in the ambient air, and it has been widely used around the world (Hu et al. 76 2008, von Schneidemesser et al. 2009, Guo et al. 2012, Lewandowski et al. 2013, Ding et al. 2014).

High-elevation areas are sensitive to global climate change (Xua *et al.* 2009). Observation of aerosol concentrations and compositions at high elevation sites can provide insight into the influence of natural and anthropogenic aerosols on global climate. The Tibetan Plateau (TP), the largest and highest plateau, is at the juncture of large desert areas and the densely populated Indian subcontinent. Previous

study found the northwesterly winds could bring dust from the western deserts to the TP and lead to high levels of geological aerosols at a site on the southeast TP (Zhao *et al.* 2013). Moreover, anthropogenic pollutants (e.g. sulfate, nitrate, potassium, element carbon, and heavy metals) emitted in the developing countries in South Asia could be transported to the TP by the southerly and southwesterly winds, especially during the summer monsoon season (Cong *et al.* 2007, Ming *et al.* 2010, Li *et al.* 2013, Zhao *et al.* 2013).

87 The observation at the remote central TP site, Nam Co (NC) discovered that the mean ratio of 88 organic carbon (OC) to element carbon (EC) was 31.9 ± 31.1 during July 2006 to January 2007, 89 implying the significant SOA contribution to OC (Ming et al. 2010) in the TP. However, there are only 90 three studies in SOA compositions within the TP. Li et al. (2013) reported biogenic SOA (BSOA) 91 tracers during the summer of 2010 at Qinghai Lake in the northeastern part of the TP. Stone et al., 92 (2012) measured BSOA tracers from August to October 2005 on the south slope of Himalayas in the 93 southwestern part of the TP. Due to the limited samples, it was difficult to examine the seasonal 94 variation of these BSOA tracers in the TP. Moreover, due to the lack of anthropogenic SOA tracers, it 95 was not possible to examine anthropogenic SOA in the TP, although above discussions have 96 demonstrated that air pollutants from South Asia could be transported to the TP. Our recent study 97 provided a snapshot of SOA tracers over China (including the NC and Linzhi sites in the TP) during the 98 summer of 2012 (Ding et al. 2014). In this study, the observation at the remote NC site extended to one 99 year. Seasonal trends of SOA tracers from isoprene, monoterpene, β -caryophyllene and aromatics were 100 determined in the TP. Furthermore, secondary organic carbon (SOC) was estimated by the SOA-tracer 101 method to check the variations of SOA origins at the NC site. To our knowledge, it is the first time that 102 the seasonal trends of SOA tracers and origins are studied in the remote TP.

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104 2. Experiment

105 2.1 Field Sampling

106 Samples were collected at a remote site (4730 meters above sea level) at the southeastern shore of 107 Nam Co Lake in the central TP (Figure 1). Nam Co Lake ($90^{\circ}16' - 91^{\circ}03'$ E and $30^{\circ}30' - 30^{\circ}55'$ N) is 108 located in the Nyainqen Tanglha Mountain Range with a total area of 2017 km² (Zhou *et al.* 2013). The 109 major vegetation in the Nam Co Lake Basin is the high cold alpine meadow.

110 Sampling was undertaken from July 2012 to July 2013. An Anderson sampler equipped with

111 9-stage cascade impactors and pre-baked quartz fiber filters (Whatman, baked at 450 °C for 8 h) was 112 used to get size-segregated particle samples at an air flow rate of 28.3 L min⁻¹. The 50% cutoff sizes are 113 <0.4, 0.4–0.7, 0.7–1.1, 1.1–2.1, 2.1–3.3, 3.3–4.7, 4.7–5.8, 5.8–9.0, and \geq 9.0 µm, respectively. The 114 flow rate was calibrated before and after each sampling episode using an airflow meter to ensure the sampler operated at the specified flow rate. One set of 9 size-fractionated filters were collected for 72 115 116 hours every two weeks. Additionally, four sets of field blanks were collected in the same way as the 117 ambient samples for 5 minutes when the sampler was turned off. All samples were wrapped with 118 aluminum foil and stored at -18°C before analysis.

119

120 2.2 Chemical Analysis

Each set of nine filters were combined together as one sample to meet the analysis requirement. Detailed information on the SOA tracer analysis is described elsewhere (Ding *et al.* 2014). Prior to solvent extraction, isotope-labeled standard mixtures were spiked into samples as internal standards. Samples were extracted twice by sonication with the mixed solvent dichloride methane (DCM)/hexane (1:1, v/v), then three times with the mixed solvent DCM/methanol (1:1, v/v). The extracts of each sample were combined, filtered and concentrated to ~2 mL. Then, the concentrated solution was divided into two parts for methylation and silylation, respectively.

The samples were analyzed by an a gas chromatography/mass spectrometer detector (GC/MSD, 128 129 Agilent 7890/5975C) in the selected ion monitoring (SIM) mode with a 30 m HP-5 MS capillary 130 column (i.d. 0.25 mm, 0.25 µm film thickness). Splitless injection of a 2 µL sample was performed. The GC temperature was initiated at 65 °C, held for 2 min, then increased to 290 °C at 5 °C min⁻¹ and 131 132 held for 20 min. Thirteen SOA tracers were quantified by the GC/MSD coupled with an electron 133 impact (EI) ionization source, including five SOA_M tracers (cis-pinonic acid, pinic acid, 134 3-methyl-1,2,3-butanetricarboxylic acid, 3-hydroxyglutaric acid and 3-hydroxy-4,4-dimethylglutaric 2-methylerythritol, 135 (2-methylthreitol, 2-methylglyceric acid), six SOA tracers acid, 136 trans-2-methyl-1,3,4-trihydroxy-1-butene cis-2-methyl-1,3,4-trihydroxy-1-butene, and 137 3-methyl-2,3,4-trihydroxy-1-butene), one SOA_C tracer (β -caryophyllenic acid) and one SOA_A tracer 138 (2,3-dihydroxy-4-oxopentanoic acid, DHOPA). Figure S1 in supplemental information presents the 139 total ion chromatogram (TIC) of these SOA tracers. cis-Pinonic acid and pinic acid were quantified by 140 authentic standards. Due to the lack of standards, the SOA₁ tracers were quantified using erythritol

141 (Claeys *et al.* 2004, Ding *et al.* 2008). The other SOA_M tracers were quantified using *cis*-pinonic acid. 142 β -Caryophyllenic acid and DHOPA were quantified using octadecanoic acid and azelaic acid, 143 respectively (Ding *et al.* 2012). The EI spectrum of each SOA tracer is shown in Figure S2-S4. The 144 method detection limits (MDLs) for *cis*-pinonic acid, pinic acid, erythritol, octadecanoic acid and 145 azelaic acid were 0.03, 0.05, 0.04, 0.03 and 0.07 ng m⁻³, respectively, at a total volume of 122 m³.

146 2.3 Quality assurance and quality control

Field and laboratory blanks were analyzed in the same manner as the field samples. These SOA tracers were not detected in the field or laboratory blanks. To evaluate the recoveries of the analytical method, six spiked samples (authentic standards spiked into solvent with pre-baked quartz filters) were analyzed. The recoveries were 101 ± 3 % for *cis*-pinonic acid, 70 ± 10 % for pinic acid, 65 ± 14 % for erythritol, 83 ± 7 % for octadecanoic acid, and 89 ± 9 % for azelaic acid. The relative differences for target compounds in samples collected in parallel (n=6) were all below 15%.

153 It should be noted that ketopinic acid was used as the surrogate for the quantification of all SOA 154 tracers by Kleindienst et al. (2007); while different surrogates were used to quantify different SOA 155 tracers in this study. The response factors of internal standard calibration for the 5 surrogates ranged 156 from 0.98 (azelaic acid) to 1.78 (pinic acid), with the average of 1.38 and the relative standard 157 deviation (RSD) of 23%. The response factor of ketopinic acid was also calculated in this study. Its 158 value (1.27) was consistent with the average of the five surrogates.

159 2.4 Estimation of measurement uncertainty

Since there is no commercial standard available for most SOA tracers (except *cis*-pinonic acid and pinic acid), the use of surrogate standards for quantification introduces additional error to measurement. Error in analyte measurement (E_A) is propagated from the standard deviation of the field blank (E_{FB}), error in spike recovery (E_R) and the error from surrogate quantification (E_Q):

164 $E_A = \sqrt{E_{FB}^2 + E_R^2 + E_Q^2}$ (1)

Since SOA tracers were not detected in the field blanks, E_{FB} was 0 in this study. The spike recoveries of surrogate standards were used to estimate the E_R of tracers which ranged from 1% (*cis*-pinonic acid) to 35% (erythritol). Stone et al., (2012) developed an empirical approach to estimate E_0 based on homologous series of atmospherically relevant compounds. The relative error introduced by each carbon atom (E_n) was estimated to be 15 %, each oxygenated functional group (E_f) to be 10%
and alkenes (E_d) to be 60%. The errors introduced from surrogate quantification are treated as additive
and are calculated as:

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$$E_Q = E_n \Delta n + E_f \Delta f + E_d \Delta d \tag{2}$$

173 where Δ n is the difference in carbon atom number between a surrogate and an analyte, Δ f is the 174 difference in oxygen-containing functional group between a surrogate and an analyte, Δ d is the 175 difference in alkene functionality between a surrogate and an analyte.

Table S1 shows the estimated uncertainties in tracer measurement. The errors from surrogate quantification (E_Q) ranged from 15% (2-methyltetrols) to 155% (β -caryophyllenic acid) in this study. Propagated with the error in recovery, the uncertainties in analyte measurement (E_A) were estimated in the range of 38% to 156%.

180 2.5 Backward trajectories

The air masses' transport during each sampling episode was investigated using Hybrid Single Particle Lagrangian Integrated Trajectory Model (HYSPLIT V4.9). Five-day backward trajectories (BTs) were analyzed during each sampling episode with 6-hour step at the height of 500 m above ground level. Then cluster analysis was performed to present the mean trajectory of each cluster, based on all the trajectories during our campaign.

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187 3. Results and Discussions

188 3.1 Seasonal variations of SOA tracers

189 Since the NC site is located in the high elevation TP, the annual temperature was only -1.64 °C with 190 the range of -16.1 °C in January to 10.2 °C in July (Table 1). The annual relative humidity (RH) was 58% 191 with the peak in July (84%) and the lowest in January (30%). The sum of all tracers ranged from 0.78 to 185 ng m⁻³. Among these compounds, SOA_I tracers (26.6 \pm 44.2 ng m⁻³) represented the majority, 192 followed by SOA_M tracers (0.97 \pm 0.57 ng m⁻³), DHOPA (0.25 \pm 0.18 ng m⁻³) and β -caryophyllenic 193 acid $(0.09 \pm 0.10 \text{ ng m}^{-3})$. During the summer (July-September 2012 and June-July 2013), SOA₁ tracers 194 195 presented the majority (> 95%). The mass fractions of SOA_M tracers in all compounds increased during 196 the cold period (October 2012 to May 2013).

198 3.1.1 Isoprene SOA tracers

199 The total concentrations of SOA_I tracers (sum of six tracers) ranged from 0.36 - 184 ng m⁻³. The 200 levels of SOA₁ tracers were 1-2 orders of magnitude higher than those over the global oceans and the 201 Arctic (Table 2). Among the SOA_I traces, 2-methyltetrols (sum of 2-methylthreitol and 2-methylerythritol, MTLs) were the major components (72%), with an annual average of 23.8 ± 40.3 202 ng m⁻³ (0.18 to 165 ng m⁻³). The 2-methylglyceric acid (MGA) averaged 1.95 \pm 2.92 ng m⁻³ and 203 204 C₅-alkenetriols (sum of cis-2-methyl-1,3,4-trihydroxy-1-butene, 205 trans-2-methyl-1,3,4-trihydroxy-1-butene, and 3-methyl-2,3,4-trihydroxy-1-butene) averaged 0.93 \pm 1.39 ng m^{-3} . MTLs are produced through the particle-phase uptake of the epoxydiols that formed in the 206 207 gas-phase photo-oxidation of isoprene under low-NOx or NOx free conditions (Paulot et al. 2009, 208 Surratt et al. 2010). Since the remote TP is a low-NOx environment, it is expected that the low-NOx 209 products, MTLs dominated over other SOA_I tracers. The majority of MTLs at the NC site was 210 consistent with those observed within the TP (Stone et al. 2012, Li et al. 2013) and over most global 211 oceans (Fu et al. 2011, Hu et al. 2013), but different from those over the North Pacific Ocean and the 212 Arctic where MGA was the major SOA_I tracer due to the significant influence of Siberian fires (Fu et 213 al. 2011, Ding et al. 2013). The two MTL isomers exhibited a strong correlation with each other throughout the year (R^2 =0.996, p<0.001) with a slope of 3.7, indicating that the two isomers shared 214 215 similar formation pathways.

Figure 2a presents a typical seasonal trend of SOA_1 tracers that high concentrations all existed in the summer. From October 2012 to April 2013, temperature was below zero, the levels of SOA_1 tracers dramatically decreased as low as 0.38 ng m⁻³ in January.

Isoprene emission rate (E_I) depends on light and temperature (Guenther *et al.* 1993):

220

$$E_I = EF_I \times C_L \times C_T \tag{3}$$

where EF_I is the basal emission rate at 30 °C leaf temperature and 1000 µmol m⁻² s⁻¹ PAR. C_L and C_T are the factors representing the influences of light and temperature, respectively. C_T can be estimated as:

224
$$C_T = \frac{exp^{\frac{C_{T_1}(T-T_s)}{RT_s T}}}{\frac{c_{T_2}(T-T_m)}{RT_s T}}$$
(4)

225 Then the natural logarithm of C_T is calculated as:

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226
$$\operatorname{Ln} C_{T} = \frac{C_{T_{1}}}{RT_{s}} \left(1 - \frac{T_{s}}{T} \right) - \operatorname{Ln} \left[1 + exp^{\frac{C_{T_{2}}}{RT_{s}} \left(1 - \frac{T_{m}}{T} \right)} \right]$$
(5)

where R = 8.314 J K⁻¹ mol⁻¹, C_{T1} = 95000 J mol⁻¹, C_{T2} = 230000 J mol⁻¹, T_s = 303 K, T_m = 314 K, and T is the leaf temperature (Guenther *et al.* 1993). Under the condition of T < T_m , the latter part in Equation (5) is close to zero, and Ln C_T is linearly correlated with 1/T.

230 Figure 3a presents a negative correlation between the natural logarithm of SOA₁ tracer levels and 231 the reciprocal of temperature in Kelvin (p<0.001). Moreover, the temperature dependence of SOA_I 232 tracers was similar to that of C_T , and SOA_I tracers exhibited a significant positive correlation with C_T 233 during our sampling at the NC site (Figure 3b). These results indicated that the seasonal variation of 234 SOA_I tracers at the NC site was mainly influenced by the isoprene emission. Considering the short 235 lifetime (several hours) of isoprene in the air, SOA₁ should be mainly formed from local precursor. In 236 summer, high temperature and intense light could enhance isoprene emission and photo-reactions. 237 Moreover, high temperature in summer could enhance the heterogeneous reactions of isoprene-derived 238 epoxides on particles which play the key roles in SOA_I formation (Lin et al., 2013; Paulot et al., 2009). 239 All these interpreted the high levels of SOA₁ tracers in the summer at the NC site. In the winter, 240 isoprene emission significantly dropped due to the extremely low temperature. Thus, the tracers were 241 only in trace amount at the NC site.

242 It is worth noting that the ratio of MGA to MTLs (MGA/MTLs) was negatively correlated with 243 temperature (Figure 4a) and RH (Figure 4b). Based on chamber results, the formation mechanisms of 244 MGA and MTLs are quite different. MGA is produced under high-NO_x conditions; while MTLs are 245 mainly formed under low-NO_x or NO_x free conditions (Surratt et al. 2010). Moreover, low RH (15 -246 40 %) could enhance the formation of MGA in the particulate phase but not of MTLs (Zhang et al. 247 2011). In addition, high particle acidity would favor the formation of MTLs instead of MGA (Surratt et 248 al. 2007). Although there are few data available in the TP, the aerosols are expected to be neutral at the 249 remote NC site. Thus, the influence of acidity on MGA/MTLs should be not significant. Isoprene 250 emission is apparently high in summer due to high temperature and light intensity, which could 251 enhance the ratio of isoprene to NOx and favor MTLs formation at the NC site. Moreover, high RH 252 (~70%) in the summer (Table 1) could not favor MGA formation. Thus, MGA/MTLs exhibited the 253 lowest values (less than 0.1) in the summer samples (Figure 4). In the winter, both temperature and RH 254 dropped to the lowest of the whole year. Low temperature reduced isoprene emission and low RH

favored MGA formation. Thus, MGA/MTLs increased up to 0.8 in the winter samples (Figure 4).

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257 3.1.2 Terpene SOA tracers

The total concentrations of SOA_M tracers (sum of five tracers) ranged from 0.11 - 2.39 ng m⁻³. The levels of the SOA_M tracers were consistent with those over the global oceans and the Arctic (Table 2). Among these traces, *cis*-pinonic acid was the major compound (54%), with an annual average of 0.49 ± 0.38 ng m⁻³, followed by pinic acid (0.22 ± 0.32 ng m⁻³), 3-methyl-1,2,3-butanetricarboxylic acid (0.18 ± 0.25 ng m⁻³), 3-hydroxyglutaric acid (0.08 ± 0.06 ng m⁻³) and 3-hydroxy-4,4-dimethylglutaric acid (below MDL in the most samples). The monthly variation of SOA_M tracers did not fully follow that of temperature (Figure 2b). From

July to November 2012 (period 1), temperature decreased to -15 °C; while SOA_M tracer levels increased as high as 1.99 ng m⁻³. After that, both temperature and SOA_M tracers dropped to the lowest values in January 2013, and increased concurrently till April 2013 (period 2). During May to July 2013 (period 3), SOA_M tracer levels exhibited slight variation, although temperature kept increasing.

The seasonal variation of SOA_M tracers could be influenced by monoterpenes emission and gas-particle partitioning. Monoterpenes emission rate (E_M) is often assumed to be solely dependent on temperature (Guenther *et al.* 1993):

- $E_M = EF_M \times \gamma_T \tag{6}$
- 273

$$\mathbf{y}_{\tau} = e x p^{\beta(T-T_s)} \tag{7}$$

where EF_M is monoterpenes emission rate at a standard temperature T_s (303 K), γ_T is the activity factor by temperature, β is an empirical coefficient usually taken to be 0.09 K⁻¹ (Guenther *et al.* 1993), T is the leaf temperature.

SOA yield (Y) of precursors could be expressed using an empirical relationship based on
gas-particle partitioning of two semi-volatile products (Odum et al., 1996):

279 $Y = M_0 \sum_{i=1}^{2} \frac{\alpha_i K_i}{1 + M_0 K_i}$ (8)

where $M_0 (\mu g m^{-3})$ is the total concentration of absorbing organic material, α_i is the mass stoichiometric coefficients of the product i, $K_i (m^3 \mu g^{-1})$ is the temperature-dependent partitioning coefficient of the semi-volatile compound i. Assuming a constant activity coefficient and mean molecular weight, the partitioning coefficient, K_i (T) at a certain temperature (T) could be estimated as (Sheehan and 284 Bowman, 2001):

285

$$K_{i}(T) = K_{i}^{*} \frac{T}{T^{*}} \exp\left[\frac{H_{i}}{R}\left(\frac{1}{T} - \frac{1}{T^{*}}\right)\right]$$
(9)

where K_i^* is an experimentally determined partitioning coefficient at a reference temperature, T^{*}. H_i is the vaporization enthalpy, R is the gas constant. To model the temperature-dependent absorptive partitioning, three parameters, α_i , K_i, and H_i, are required for each condensable product.

289 Table S2 lists all the parameters for two-product model of α -pinene SOA which were also used to 290 estimate the temperature effect on SOA partitioning by Sheehan and Bowman (2001). The available data of OC at the NC site were reported in the range of 1.18 to 2.26 µgC m⁻³ during July 2006 to 291 January 2007 with an average of 1.66 µgC m⁻³ (Ming et al., 2010). Thus, M₀ is calculated as 2.32 µg 292 293 m^{-3} by the average OC multiplying 1.4. Figure S5 shows the temperature dependence of α -pinene emission rate (γ_T) and SOA yield within the temperature range at the NC site (-16.7 to 10.2 °C). 294 295 Obviously, decreasing temperature could reduce the emission but enhance the gas to particle 296 partitioning and SOA yield.

297 From July to November 2012 (period 1), high values of SOA_M tracers and SOA yield existed under 298 low temperature, and SOA_M tracers were positively correlated with SOA yield (r=0.647, p<0.05, Figure 299 5a). These suggested that the temperature effect on partitioning was the dominant process influencing 300 SOA_M tracers' variation during the period 1. From December 2012 to April 2013 (period 2), high 301 values of SOA_M tracers and activity factor (γ_T) existed under high temperature, and SOA_M tracers were 302 positively correlated with γ_T (r=0.741, p<0.05, Figure 5b). These suggested that the temperature effect 303 on emission was the dominant process influencing SOA_M tracers' variation during the period 2. The 304 increase of SOA_M tracer concentrations during spring was also observed in the southeastern United 305 States (Ding et al. 2008), resulting from the enhancement of monoterpenes emission in spring (Kim 306 2001). From May to July 2013 (period 3), SOA_M tracer concentrations were relative stable, and there 307 was no correlation of SOA_M tracers with γ_T or SOA yield (p>0.05). These might result from the 308 counteraction of temperature effects on emission and partitioning during the summer.

Previous study proposed that *cis*-pinonic acid and pinic acid (P) were the first-generation products of SOA_M and only formed under low-NOx conditions (Eddingsaas et al. 2012). The dominance of *cis*-pinonic acid and pinic acid among SOA_M tracers at the remote NC site indicated that SOA_M there was mainly formed under low-NOx conditions. Moreover, *cis*-pinonic acid and pinic acid could be 313 further photo-degraded to high-generation products, e.g. 3-methyl-1,2,3-butanetricarboxylic acid (M) 314 (Glasius et al. 2000; Jaoui et al. 2005; Szmigielski, et al., 2007). And the ratio of cis-pinonic acid plus 315 pinic acid to 3-methyl-1,2,3-butanetricarboxylic acid (P/M) could be applied to trace the aging of 316 SOA_M (Ding et al., 2011; Gómez-González et al., 2012). In the fresh chamber produced α -pinene SOA 317 samples, the ratios of P/M were reported in the range of 1.51 to 3.21 (Offenberg, et al., 2007). In this 318 study, the ratio of P/M averaged 16.7 \pm 20.9. Thus, SOA_M was generally fresh at the NC site and 319 should be mainly formed from local precursors. Figure 6 presents a negative correlation between P/M 320 and temperature (r=-0.560, p=0.008). Higher P/M ratios were observed in the fall and the winter, and 321 lower P/M ratios occurred in the spring and the summer. Since temperature has positive influence on 322 photo-reaction rates, the higher temperature during the summer could accelerate the photochemistry in 323 the air and result in P to M conversion being more efficient. Thus, SOA_M in the summer was more aged 324 than that in the winter.

The levels of SOA_C tracer, β -caryophyllenic acid were in the range of below MDL to 0.40 ng m⁻³. As Figure 2c shows, the levels elevated from July to November 2012 and dropped to below MDL in December 2012. Then, the concentrations increased from January to March 2013 and decreased from April to June 2013. β -Caryophyllenic acid was positively correlated with SOA_M tracers (p=0.025), indicating that the seasonal variation of β -caryophyllenic acid was similar with that of the SOA_M tracers.

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332 3.1.3 Aromatic SOA tracer

The levels of SOA_A tracer, DHOPA were in the range of below MDL to 0.61 ng m⁻³. This anthropogenic tracer was not detected or reported in global remote areas (Table 2). Due to few human activity at the remote NC site, the highest concentration of DHOPA was 1-2 orders of magnitude lower than those (up to 52 ng m⁻³) reported in the urban regions of United States (Lewandowski *et al.* 2013) and China (Ding *et al.* 2014). DHOPA exhibited the higher concentrations in the summer and the lower levels in the winter (Figure 2d).

Besides urban emissions from solvent and fossil fuel use, biomass burning is an important source of aromatics in many parts of the world (Lewis et al. 2013). The local dung or biomass burning (Duo et al. 2015; Xiao et al. 2015) may be potential sources of aromatics in the TP. Hence, DHOPA may come from the processing of biomass burning emission. Figure 7 exhibits the monthly variation of biomass burning tracer, levoglucosan during our sampling. The concentrations of levoglucosan ranged from 0.82 ng m⁻³ (October 2012) to 4.55 ng m⁻³ (April 2013) with a mean of 1.87 ± 1.14 ng m⁻³. Apparently, the monthly variation trend of levoglucosan was quite different from that of DHOPA. And there was no correlation between DHOPA and levoglucosan (p>0.05) (Figure S6). These indicated that DHOPA at the NC site was not mainly from the processing of biomass burning emission. Since there was few anthropogenic source near the remote NC site, the SOA_A tracer should be not locally formed but mainly transported from upwind regions.

350 To check the potential source areas of anthropogenic emissions, the satellite data of population 351 (http://sedac.ciesin.columbia.edu/theme/population), aerosol optical thickness (AOT, density 352 http://neo.sci.gsfc.nasa.gov/), tropospheric NO_2 vertical column densities (VCD, 353 http://avdc.gsfc.nasa.gov/), and surface CO (https://www2.acd.ucar.edu/mopitt) were analysis on the 354 global scale. As shown in Figure S7a, the northern Indian subcontinent was the most populated region of the world, with a population density of more than 1000 persons per km². Moreover, the plots of 355 356 global AOT, tropospheric NO2 VCD, and surface CO (Figure S7, b-d) all illustrated that the northern 357 Indian subcontinent, including Bangladesh, Nepal, the northeastern India, and the northwestern India 358 was the global hotspots of these anthropogenic pollutants. Compared with the northern Indian 359 subcontinent, the TP exhibited extremely low population density and low levels of AOT, surface CO, 360 and NO₂ VCD (Figure 8, a-d). Besides these satellite data, a recent study at a site in the northwestern India (Indo-Gangetic plain) witnessed extremely high levels (up to 2065 ng m⁻³) of polycyclic aromatic 361 362 hydrocarbons which were mainly formed from anthropogenic combustion processes (Dubey et al., 363 2015). All these demonstrated that there were high anthropogenic emissions in the northern India 364 subcontinent.

365 The TP features a monsoon climate (Cong et al. 2007, Ming et al. 2010, Zhao et al. 2013). Figure 366 9a presents the average trajectory of each cluster during our sampling in the whole year. The air masses 367 over the NC were primarily from Bangladesh, Nepal and the northeastern India (cluster 1, 32%), the 368 northwestern India (Indo-Gangetic basin) (cluster 3-6, 55%), and the Taklimakan Desert (cluster 2, 369 13%) during the sampling period. In the summer, the prevailing southerly winds (cluster 1, Figure 9b) 370 passed through the heavily polluted areas in the Bangladesh and the northeastern India, and could bring 371 anthropogenic pollutants into the TP. Previous studies in the TP have witnessed the enrichment of 372 anthropogenic metals (Cong et al. 2007) and the enhancement of carbonaceous aerosols (Ming et al.

2010, Zhao *et al.* 2013) under the influence of summer monsoon. Thus, the increase of DHOPA levels
at the NC site in the summer was mainly due to the transport of air pollutants from the upwind
Bangladesh and the northeastern India.

376 In the winter, the air masses over the NC site were mainly originated from the northwestern India 377 (Indo-Gangetic basin) by the westerly winds (Figure 9b). Compared with the summer samples, the winter samples underwent the longer distance transport. Moreover, extremely low temperature in the 378 379 winter could reduce DHOPA formation. Therefore, the levels of DHOPA were lower in the winter. It is 380 worth noting that the mass fractions of DHOPA in all tracers significantly elevated in the winter (less 381 than 2% in the summer but up to 10% in January, Figure 2d), although its levels reduced. As described 382 in equation (3) and (6), temperature is an important factor controlling BVOCs emission. The drop of 383 temperature from the summer (up to 10.2 °C) to the winter (low to -16.7 °C) at the NC site would lead 384 to the emission of isoprene and monoterpenes decreasing by 98% and 90%, respectively. The elevated 385 fractions of DHOPA in the winter samples suggested that the SOA contributions from aromatics would 386 increase in the winter when BVOCs emission largely decreased.

387

388 3.2 Source apportionment

389 The SOA-tracer method developed by Kleindienst and co-workers was applied to attribute SOC at 390 the NC site. The researchers performed chamber experiments to obtain the mass fraction of the tracers 391 in SOC (f_{SOC}) for individual precursor:

392

$$f_{\rm SOC} = \frac{\sum_i [\rm tri]}{[\rm SOC]} \tag{10}$$

393 where \sum_{i} [tri] is the total concentrations of the tracers for a certain precursor. [SOC] is the mass 394 concentration of SOC. With these f_{SOC} values and the measured SOA tracers in the ambient air, SOC from different precursors can be estimated in the atmosphere, with the assumption that the $f_{\rm SOC}$ values 395 396 in the chamber are the same as those in the ambient air. There is some degree of uncertainty in the 397 SOA-tracer method due to the quantification with a single surrogate calibration standard (ketopinic 398 acid) and the simplification of applying SOA tracers and conversion factors to calculate SOC in the 399 ambient samples (Kleindienst et al. 2007). However, this method has been widely applied to attribute 400 SOC from different precursors and proven to be able to provide reasonable results in the United States 401 (Kleindienst et al. 2007, Stone et al. 2009, Lewandowski et al. 2013), and China (Hu et al. 2008, Guo 402 et al. 2012, Peng et al. 2013, Ding et al. 2014). Lewandowski et al. (2008) found that the measured OC 403 in the midwestern United States could be fully explained by primary OC from chemical mass balance 404 (CMB) model plus SOC from the SOA-tracer method, suggesting that the secondary organic tracer 405 technique could be a valuable method for SOC estimation. Kleindienst et al. (2010) further compared 406 the estimated SOC by the SOA-tracer method and other four independent methods (multiple 407 regressions, CMB, carbon isotope and EC-tracer) in the southeastern United States, and found that 408 these five methods matched well. Our previous study in the Pearl River Delta found SOC levels 409 estimated by the SOA-tracer method were not only consistent with but also correlated well with those by EC-tracer method in summer, (Ding et al., 2012). The SOC apportionment results were also 410 411 comparable between the SOA-tracer method and positive matrix factorization (PMF) model in Hong 412 Kong (Hu et al. 2010).

The f_{SOC} were reported as $0.155 \pm 0.039 \ \mu\text{g} \ \mu\text{gC}^{-1}$, $0.023 \pm 0.0046 \ \mu\text{g} \ \mu\text{gC}^{-1}$ and 0.00797 ± 0.0026 413 $\mu g \mu g C^{-1}$ for isoprene (SOC_I), β -caryophyllene (SOC_C) and aromatics (SOC_A), respectively 414 415 (Kleindienst et al. 2007). In this study, the same set of SOA tracers as reported by Kleindienst et al. 416 (2007) were used for SOC estimation, including MGA and MTLs for SOC₁, β -caryophyllenic acid for 417 SOC_{C} and DHOPA for SOC_{A} . For monoterpene SOC (SOC_{M}), nine tracers were involved in the source 418 profile (Kleindienst et al. 2007). However, only five of the nine SOA_M tracers were measured in the 419 current study. Wang et al. (2013) compared the results from model prediction with field observation in 420 the Pearl River Delta and pointed out that the SOA-tracer method would underestimate SOA_M, 421 probably due to the mismatch of tracer compositions in the field and the source profile (Ding et al. 2014). To minimize the uncertainty caused by the mismatch in tracer compositions, the f_{SOC} with the 422 same five SOA_M tracers (0.059 μ g μ gC⁻¹) was computed using the chamber data from another study by 423 424 the same research group (Offenberg et al. 2007). The same f_{SOC} for SOA_M was also applied to estimate 425 SOC_M in our previous study over China (Ding *et al.* 2014).

The uncertainty in the SOA-tracer method is induced from the analysis of organic tracers and the determination of the conversion factors. Based on the E_A values in Table S1, the uncertainties in the tracer analyses were within 40% for SOA_I (only MGA and MTLs involved for SOC estimation), 95% for SOA_M, 156% for SOA_C, and 91% for SOA_A. The uncertainties of f_{SOC} were reported to be 25% for isoprene, 48% for monoterpenes, 22% for β -caryophyllene and 33% for aromatics (Kleindienst *et al.* 2007, Lewandowski *et al.* 2013). Considering these factors, the uncertainties of SOC apportionment 432 were calculated through error propagation. The RSD were 47% for SOC₁, 106% for SOC_M, 157% for 433 SOC_C, and 96% for SOC_A. On average, the RSD of the reconstructed SOC (sum of the four precursors) 434 was $51 \pm 11\%$.

435 Figure 10 presents the monthly variations of the reconstructed SOC. SOC was high in the summer 2012 and declined from October to December. After that, it kept increasing from January to June. The 436 total concentrations of SOC ranged from 0.02 µgC m⁻³ to 0.69 µgC m⁻³ with an annual average of 0.22 437 \pm 0.29 µgC m⁻³. The available data of OC in total suspended particles at the NC site were reported in 438 the range of 1.18 to 2.26 µgC m⁻³ during July 2006 to January 2007 (Ming et al. 2010). Since we did 439 440 not measure OC in our size-segregated samples, the OC data reported by Ming et al. (2010) were used 441 to calculate SOC fraction in OC (SOC/OC) from July to January. The calculated SOC/OC was average 442 38% in the summer and up to 58% in September, suggesting that SOC was an important contributor to 443 OC at the NC site during the summer (Ming et al. 2010). However, from the fall to winter, the elevated 444 OC and decreased SOC led to SOC/OC declining from 11% (in October) to 1% (in January), indicating 445 that SOA from the four precursors had minor contributions to the elevated OC. Since the air masses 446 during the fall to the winter were mostly originated from the northwestern Indo-Gangetic basin (cluster 447 3-6 in Figure 9), primary pollutants emitted there could transport to the TP and have significant impact 448 on the air at the NC site. In addition, SOA from aqueous-phase reactions and primary OA aging could 449 not be captured by the SOA-tracer method. Thus, the current results might underestimate the total 450 amount of SOC, which partly explained the low OC shares of SOC at the NC site during the fall to the 451 winter.

Biogenic SOC (sum of SOC_I, SOC_M, and SOC_C) dominated over SOC_A at the NC site, averagely 452 453 accounting for 75% of the estimated SOC. In the summer, SOC_{I} was the major contributor with the 454 SOC shares of 81%. From the fall to the spring, SOC_M became the major contributor, averagely 455 contributing 38% to SOC. Although SOC_A levels reduced in the winter, SOC_A contributions elevated as 456 high as 53% in January 2013. The elevated OC and the higher SOC_A contributions in the winter 457 samples (Figure 10) implied that the transport of anthropogenic pollutants from the Indian subcontinent 458 might have significant influence on carbonaceous aerosols over the remote NC during winter.

459

460 4. Conclusion

461

Seasonal trends of SOA tracers and origins were studied in the remote TP for the first time. SOA₁

462 tracers represented the majority among these compounds. The significant temperature dependence of 463 SOA₁ tracers suggested that the seasonal variation of SOA₁ tracers at the NC site was mainly influenced 464 by the isoprene emission. Due to the influence of temperature and relative humidity, the ratio of 465 high-NOx to low-NOx products of SOA₁ (MGA/MTLs) was the highest in the winter and the lowest in the summer. The seasonal variation of SOA_M tracers was impacted by monoterpenes emission and 466 467 gas-particle partitioning. Due to the transport of air pollutants from the Indian subcontinent, DHOPA 468 presented relatively higher concentrations in the summer and increased mass fractions in the winter. 469 The SOA-tracer method was applied to estimated SOC from these four precursors. The annual average of SOC was $0.22 \pm 0.29 \ \mu$ gC m⁻³, with the biogenic SOC accounting for 75%. In the summer, isoprene 470 471 was the major precursor with its SOC shares of 81%. In the winter when the emissions of biogenic 472 precursors largely declined, the contributions of SOC_A increased. At present, SOA origins and their 473 seasonal variations are unclear in the remote high-elevation TP. The remote TP is connected to the 474 densely populated Indian subcontinent. Our study implies that anthropogenic pollutants emitted there 475 could be transported to the TP and influence SOC over the remote NC.

476

477

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483 References

Claeys, M., Graham, B., Vas, G., Wang, W., Vermeylen, R., Pashynska, V., Cafmeyer, J., Guyon, P.,
Andreae, M. O., Artaxo, P., and Maenhaut, W.: Formation of secondary organic aerosols through
photooxidation of isoprene, Science, 303, 1173-1176, 2004.

Claeys, M., Szmigielski, R., Kourtchev, I., Van der Veken, P., Vermeylen, R., Maenhaut, W., Jaoui, M.,
Kleindienst, T. E., Lewandowski, M., Offenberg, J. H., and Edney, E. O.: Hydroxydicarboxylic acids:
Markers for secondary organic aerosol from the photooxidation of alpha-pinene, Environ. Sci. Technol.,
41, 1628-1634, 2007.

Cong, Z. Y., Kang, S. C., Liu, X. D., and Wang, G. F.: Elemental composition of aerosol in the Nam Co
region, Tibetan Plateau, during summer monsoon season, Atmos. Environ., 41, 1180-1187, 2007.

- Ding, X., Zheng, M., Yu, L. P., Zhang, X. L., Weber, R. J., Yan, B., Russell, A. G., Edgerton, E. S., and
 Wang, X. M.: Spatial and seasonal trends in biogenic secondary organic aerosol tracers and
 water-soluble organic carbon in the southeastern United States, Environ. Sci. Technol., 42, 5171-5176,
 2008.
- 497 Ding, X., Wang, X., and Zheng, M.: The influence of temperature and aerosol acidity on biogenic
 498 secondary organic aerosol tracers: Observations at a rural site in the central Pearl River Delta region,
 499 South China, Atmos. Environ., 45, 1303-1311, 2011.
- Ding, X., Wang, X. M., Gao, B., Fu, X. X., He, Q. F., Zhao, X. Y., Yu, J. Z., and Zheng, M.:
 Tracer-based estimation of secondary organic carbon in the Pearl River Delta, south China, J. Geophys.
- 502 Res.-Atmos., 117, D05313, DOI: 10.1029/2011jd016596, 2012.
- 503 Ding, X., Wang, X. M., Xie, Z. Q., Zhang, Z., and Sun, L. G.: Impacts of Siberian biomass burning on
 504 organic aerosols over the North Pacific Ocean and the Arctic: Primary and secondary organic tracers,
 505 Environ. Sci. Technol., 47, 3149-3157, 2013.
- 506 Ding, X., He, Q.-F., Shen, R.-Q., Yu, Q.-Q., and Wang, X.-M.: Spatial distributions of secondary 507 organic aerosols from isoprene, monoterpenes, β -caryophyllene, and aromatics over China during 508 summer, J. Geophys. Res.-Atmos., 119, 11877-11891, 2014.
- 509 Donahue, N. M., Henry, K. M., Mentel, T. F., Kiendler-Scharr, A., Spindler, C., Bohn, B., Brauers, T.,
- 510 Dorn, H. P., Fuchs, H., Tillmann, R., Wahner, A., Saathoff, H., Naumann, K.-H., Moehler, O., Leisner,
- 511 T., Mueller, L., Reinnig, M.-C., Hoffmann, T., Salo, K., Hallquist, M., Frosch, M., Bilde, M., Tritscher,
- 512 T., Barmet, P., Praplan, A. P., DeCarlo, P. F., Dommen, J., Prevot, A. S. H., and Baltensperger, U.:
- 513 Aging of biogenic secondary organic aerosol via gas-phase OH radical reactions, Proc. Natl. Acad. Sci.
- 514 U. S. A., 109, 13503-13508, 2012.
- 515 Dubey, J., Maharaj Kumari, K., and Lakhani, A.: Chemical characteristics and mutagenic activity of
 516 PM2.5 at a site in the Indo-Gangetic plain, India, Ecotoxicol. Environ. Saf., 114, 75-83, 2015.
- 517 Duo, B., Zhang, Y., Kong, L., Fu, H., Hu, Y., Chen, J., Li, L., and Qiong, A.: Individual particle
 518 analysis of aerosols collected at Lhasa City in the Tibetan Plateau, J. Environ. Sci., 29, 165-177, 2015.
- Eddingsaas, N. C., Loza, C. L., Yee, L. D., Chan, M., Schilling, K. A., Chhabra, P. S., Seinfeld, J. H.,
 and Wennberg, P. O.: α-pinene photooxidation under controlled chemical conditions Part 2: SOA
 yield and composition in low- and high-NOx environments, Atmos. Chem. Phys., 12, 7413-7427,
 10.5194/acp-12-7413-2012, 2012.
- Froyd, K. D., Murphy, S. M., Murphy, D. M., de Gouw, J. A., Eddingsaas, N. C., and Wennberg, P. O.:
 Contribution of isoprene-derived organosulfates to free tropospheric aerosol mass, Proc. Natl. Acad.
 Sci. U. S. A., 107, 21360-21365, 2010.
- Fu, P. Q., Kawamura, K., Chen, J., and Barrie, L. A.: Isoprene, monoterpene, and sesquiterpene
 oxidation products in the high Arctic aerosols during late winter to early summer, Environ. Sci.
 Technol., 43, 4022-4028, 2009.

- Fu, P. Q., Kawamura, K., and Miura, K.: Molecular characterization of marine organic aerosols
 collected during a round-the-world cruise, J. Geophys. Res.-Atmos., 116, D13302, DOI:
 10.1029/2011jd015604, 2011.
- Fu, P. Q., Kawamura, K., Chen, J., Charrière, B., and Sempéré, R.: Organic molecular composition of
 marine aerosols over the Arctic Ocean in summer: contributions of primary emission and secondary
 aerosol formation, Biogeosciences, 10, 653-667, 2013.
- 535 Glasius, M., Lahaniati, M., Calogirou, A., Di Bella, D., Jensen, N. R., Hjorth, J., Kotzias, D., and
- 536 Larsen, B. R.: Carboxylic acids in secondary aerosols from oxidation of cyclic monoterpenes by ozone,
- 537 Environ. Sci. Technol., 34, 1001-1010, 2000.
- Gómez-González, Y., Wang, W., Vermeylen, R., Chi, X., Neirynck, J., Janssens, I. A., Maenhaut, W.,
 and Claeys, M.: Chemical characterisation of atmospheric aerosols during a 2007 summer field
 campaign at Brasschaat, Belgium: sources and source processes of biogenic secondary organic aerosol,
 Atmos. Chem. Phys., 12, 125-138, 2012.
- Guenther, A., Hewitt, C. N., Erickson, D., Fall, R., Geron, C., Graedel, T., Harley, P., Klinger, L.,
 Lerdau, M., Mckay, W. A., Pierce, T., Scholes, B., Steinbrecher, R., Tallamraju, R., Taylor, J., and
 Zimmerman, P.: A global-model of natural volatile organic-compound emissions, J. Geophys.
 Res.-Atmos., 100, 8873-8892, 1995.
- 546 Guenther, A. B., Zimmerman, P. R., Harley, P. C., Monson, R. K., and Fall, R.: Isoprene and
 547 monoterpene emission rate variability Model evaluations and sensitivity analyses, J. Geophys.
 548 Res.-Atmos., 98, 12609-12617, 1993.
- Guo, S., Hu, M., Guo, Q., Zhang, X., Zheng, M., Zheng, J., Chang, C. C., Schauer, J. J., and Zhang, R.:
 Primary sources and secondary formation of organic aerosols in Beijing, China, Environ. Sci. Technol.,
 46, 9846-9853, 2012.
- Hu, D., Bian, Q., Li, T. W. Y., Lau, A. K. H., and Yu, J. Z.: Contributions of isoprene, monoterpenes,
 β-caryophyllene, and toluene to secondary organic aerosols in Hong Kong during the summer of 2006,
 J. Geophys. Res.-Atmos., 113, D22206, DOI: 10.1029/2008jd010437, 2008.
- Hu, D., Bian, Q., Lau, A. K. H., and Yu, J. Z.: Source apportioning of primary and secondary organic
 carbon in summer PM_{2.5} in Hong Kong using positive matrix factorization of secondary and primary
 organic tracer data, J. Geophys. Res.-Atmos., 115, D16204, doi: 10.1029/2009JD012498, 2010.
- Hu, Q. H., Xie, Z. Q., Wang, X. M., Kang, H., He, Q. F., and Zhang, P.: Secondary organic aerosols
 over oceans via oxidation of isoprene and monoterpenes from Arctic to Antarctic, Sci. Rep., 3, 2280,
 2013.
- Jang, M. S., Czoschke, N. M., Lee, S., and Kamens, R. M.: Heterogeneous atmospheric aerosol
 production by acid-catalyzed particle-phase reactions, Science, 298, 814-817, 2002.
- Jaoui, M., Kleindienst, T. E., Lewandowski, M., Offenberg, J. H., and Edney, E. O.: Identification and quantification of aerosol polar oxygenated compounds bearing carboxylic or hydroxyl groups. 2.

- 565 Organic tracer compounds from monoterpenes, Environ. Sci. Technol., 39, 5661-5673, 2005.
- 566 Jaoui, M., Lewandowski, M., Kleindienst, T. E., Offenberg, J. H., and Edney, E. O.: β-Caryophyllinic
- scid: An atmospheric tracer for β -caryophyllene secondary organic aerosol, Geophys. Res. Lett., 34,
- 568 L05816, DOI: 10.1029/2006gl028827, 2007.
- 569 Kim, J.-C.: Factors controlling natural VOC emissions in a southeastern US pine forest, Atmos.
 570 Environ., 35, 3279-3292, 2001.
- Kleindienst, T. E., Jaoui, M., Lewandowski, M., Offenberg, J. H., Lewis, C. W., Bhave, P. V., and
 Edney, E. O.: Estimates of the contributions of biogenic and anthropogenic hydrocarbons to secondary
- organic aerosol at a southeastern US location, Atmos. Environ., 41, 8288-8300, 2007.
- 574 Kleindienst, T. E., Lewandowski, M., Offenberg, J. H., Edney, E. O., Jaoui, M., Zheng, M., Ding, X.,
 575 and Edgerton, E. S.: Contribution of primary and secondary sources to organic aerosol and PM2.5 at
 576 SEARCH network sites, J. Air Waste Manage. Assoc., 60, 1388-1399, 2010.
- Lewandowski, M., Jaoui, M., Offenberg, J. H., Kleindienst, T. E., Edney, E. O., Sheesley, R. J., and
 Schauer, J. J.: Primary and secondary contributions to ambient PM_{2.5} in the midwestern United States,
 Environ. Sci. Technol., 42, 3303-3309, 2008.
- Lewandowski, M., Piletic, I. R., Kleindienst, T. E., Offenberg, J. H., Beaver, M. R., Jaoui, M., Docherty,
 K. S., and Edney, E. O.: Secondary organic aerosol characterisation at field sites across the United
 States during the spring-summer period, Int. J. Environ. Anal. Chem., 93, 1084-1103, 2013.
- Lewis, A. C., Evans, M. J., Hopkins, J. R., Punjabi, S., Read, K. A., Purvis, R. M., Andrews, S. J.,
 Moller, S. J., Carpenter, L. J., Lee, J. D., Rickard, A. R., Palmer, P. I., and Parrington, M.: The
 influence of biomass burning on the global distribution of selected non-methane organic compounds,
 Atmos. Chem. Phys., 13, 851-867, 2013.
- Li, J. J., Wang, G. H., Wang, X. M., Cao, J. J., Sun, T., Cheng, C. L., Meng, J. J., Hu, T. F., and Liu, S.
 X.: Abundance, composition and source of atmospheric PM2.5 at a remote site in the Tibetan Plateau,
 China, Tellus Ser. B-Chem. Phys. Meteorol., 65, 20281, DOI:10.3402/tellusb.v65i0.20281, 2013.
- Lin, Y. H., Zhang, H. F., Pye, H. O. T., Zhang, Z. F., Marth, W. J., Park, S., Arashiro, M., Cui, T. Q.,
 Budisulistiorini, H., Sexton, K. G., Vizuete, W., Xie, Y., Luecken, D. J., Piletic, I. R., Edney, E. O.,
 Bartolotti, L. J., Gold, A., and Surratt, J. D.: Epoxide as a precursor to secondary organic aerosol
 formation from isoprene photooxidation in the presence of nitrogen oxides, Proc. Natl. Acad. Sci. U. S.
 A., 110, 6718-6723, 2013.
- Ming, J., Xiao, C. D., Sun, J. Y., Kang, S. C., and Bonasoni, P.: Carbonaceous particles in the atmosphere and precipitation of the Nam Co region, central Tibet, J. Environ. Sci., 22, 1748-1756, 2010.
- Odum, J. R., Hoffmann, T., Bowman, F., Collins, D., Flagan, R. C., and Seinfeld, J. H.: Gas/particle
 partitioning and secondary organic aerosol yields, Environ. Sci. Technol., 30, 2580-2585, 1996.

- Offenberg, J. H., Lewis, C. W., Lewandowski, M., Jaoui, M., Kleindienst, T. E., and Edney, E. O.:
 Contributions of toluene and α-pinene to SOA formed in an irradiated toluene/α-pinene/NOx/ air
 mixture: Comparison of results using ¹⁴C content and SOA organic tracer methods, Environ. Sci.
 Technol., 41, 3972-3976, 2007.
- Paulot, F., Crounse, J. D., Kjaergaard, H. G., Kurten, A., St Clair, J. M., Seinfeld, J. H., and Wennberg,
 P. O.: Unexpected epoxide formation in the gas-phase photooxidation of isoprene, Science, 325,
 730-733, 2009.
- 607 Peng, J. L., Li, M., Zhang, P., Gong, S. Y., Zhong, M. A., Wu, M. H., Zheng, M., Chen, C. H., Wang, H.
- 608 L., and Lou, S. R.: Investigation of the sources and seasonal variations of secondary organic aerosols in
- 609 PM2.5 in Shanghai with organic tracers, Atmos. Environ., 79, 614-622, 2013.
- Piccot, S. D., Watson, J. J., and Jones, J. W.: A global inventory of volatile organic-compound
 emissions from anthropogenic sources, J. Geophys. Res.-Atmos., 97, 9897-9912, 1992.
- 612 Robinson, A. L., Donahue, N. M., Shrivastava, M. K., Weitkamp, E. A., Sage, A. M., Grieshop, A. P.,
- 613 Lane, T. E., Pierce, J. R., and Pandis, S. N.: Rethinking organic aerosols: Semivolatile emissions and
- 614 photochemical aging, Science, 315, 1259-1262, 2007.
- Sheehan, P. E., and Bowman, F. M.: Estimated effects of temperature on secondary organic aerosol
 concentrations, Environ. Sci. Technol., 35, 2129-2135, 2001.
- 617 Stone, E. A., Zhou, J., Snyder, D. C., Rutter, A. P., Mieritz, M., and Schauer, J. J.: A comparison of
- 618 summertime secondary organic aerosol source contributions at contrasting urban locations, Environ.
- 619 Sci. Technol., 43, 3448-3454, 2009.
- Stone, E. A., Nguyen, T. T., Pradhan, B. B., and Dangol, P. M.: Assessment of biogenic secondary
 organic aerosol in the Himalayas, Environ. Chem., 9, 263-272, 2012.
- Surratt, J. D., Lewandowski, M., Offenberg, J. H., Jaoui, M., Kleindienst, T. E., Edney, E. O., and
 Seinfeld, J. H.: Effect of acidity on secondary organic aerosol formation from isoprene, Environ. Sci.
 Technol., 41, 5363-5369, 2007.
- Surratt, J. D., Chan, A. W. H., Eddingsaas, N. C., Chan, M. N., Loza, C. L., Kwan, A. J., Hersey, S. P.,
- Flagan, R. C., Wennberg, P. O., and Seinfeld, J. H.: Reactive intermediates revealed in secondary
- organic aerosol formation from isoprene, Proc. Natl. Acad. Sci. U. S. A., 107, 6640-6645, 2010.
- 628 Szmigielski, R., Surratt, J. D., Gómez-González, Y., Veken, P. V. d., Kourtchev, I., Vermeylen, R.,
- 629 Blockhuys, F., Jaoui, M., Kleindienst, T. E., Lewandowski, M., Offenberg, J. H., Edney, E. O., Seinfeld,
- 630 J. H., Maenhaut, W., and Claeys, M.: 3-Methyl-1,2,3-butanetricarboxylic acid: An atmospheric tracer
- for terpene secondary organic aerosol, Geophys. Res. Lett., 34, L24811, doi:10.1029/2007GL031338,
 2007
- van Eijck, A., Opatz, T., Taraborrelli, D., Sander, R., and Hoffmann, T.: New tracer compounds for
 secondary organic aerosol formation from beta-caryophyllene oxidation, Atmos. Environ., 80, 122-130,
 2013.

von Schneidemesser, E., Schauer, J. J., Hagler, G. S. W., and Bergin, M. H.: Concentrations and
sources of carbonaceous aerosol in the atmosphere of Summit, Greenland, Atmos. Environ., 43,
4155-4162, 2009.

Wang, S. Y., Wu, D. W., Wang, X. M., Fung, J. C. H., and Yu, J. Z.: Relative contributions of secondary
organic aerosol formation from toluene, xylenes, isoprene, and monoterpenes in Hong Kong and
Guangzhou in the Pearl River Delta, China: an emission-based box modeling study, J. Geophys.
Res.-Atmos., 118, 507-519, 2013.

Xiao, Q., Saikawa, E., Yokelson, R. J., Chen, P., Li, C., and Kang, S.: Indoor air pollution from burning
yak dung as a household fuel in Tibet, Atmos. Environ., 102, 406-412, 2015.

Kua, B., Cao, J., Hansen, J., Yao, T., Joswia, D. R., Wang, N., Wu, G., Wang, M., Zhao, H., Yang, W.,
Liu, X., and He, J.: Black soot and the survival of Tibetan glaciers, Proc. Natl. Acad. Sci. U. S. A., 106,
22114-22118, 2009.

- Zhang, H., Surratt, J. D., Lin, Y. H., Bapat, J., and Kamens, R. M.: Effect of relative humidity on SOA
 formation from isoprene/NO photooxidation: enhancement of 2-methylglyceric acid and its
 corresponding oligoesters under dry conditions, Atmos. Chem. Phys., 11, 6411-6424, 2011.
- Zhao, Z. Z., Cao, J. J., Shen, Z. X., Xu, B. Q., Zhu, C. S., Chen, L. W. A., Su, X. L., Liu, S. X., Han, Y.
 M., Wang, G. H., and Ho, K. F.: Aerosol particles at a high-altitude site on the Southeast Tibetan
 Plateau, China: Implications for pollution transport from South Asia, J. Geophys. Res.-Atmos., 118, 11360-11375, 2013.

Zhou, S. Q., Kang, S. C., Chen, F., and Joswiak, D. R.: Water balance observations reveal significant
subsurface water seepage from Lake Nam Co, south-central Tibetan Plateau, J. Hydrol., 491, 89-99,
2013.



659 Figure 1 Nam Co site in the Tibetan Plateau, China





662 Figure 2 Monthly variations of SOA tracers.



 $\label{eq:GG5} {\mbox{Figure 3 Correlations of SOA}_{I} \mbox{ tracers with temperature (a) and } C_{T} \mbox{ (b)}.$

Figure 4 Correlations of MGA/MTL with temperature (a) and relative humidity (b). Summer is from
July to September 2012 and from June to July 2013, fall is from October to November 2012, winter is
from December 2012 to February 2013, and spring is from March to May 2013.

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Figure 5 Correlation of SOA_M tracers with SOA yield in period 1 (a) and γ_{T} in period 2 (b)

Figure 6 Negative correlation between P/M ratio and temperature

Figure 7 Monthly variation of biomass burning tracer, levoglucosan

Figure 8 Spatial distribution of population density in 2000 (a), AOT (b), surface CO
(c), and NO₂ VCD (d) in May 2013 over the Indian subcontinent and the TP.

Figure 9 Cluster analyses of air masses at the NC site (a) and seasonal variations of clusters (b), based
on 5-day backward trajectories during the sampling period. Summer 1 is from July to September 2012,
fall is from October to November 2012, winter is from December 2012 to February 2013, spring is
from March to May 2013, summer 2 is from June to July 2013.

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Figure 10 Seasonal variations of estimated SOC. OC data at the NC site during July 2006 to January2007 were reported by Ming et al., (2010) and the error bar means one standard deviation in each

month.

Manth	Temp. °C ^a	RH % ^a	SOA tracers					
Month			Isoprene	Monoterpenes	β-Caryophyllene	Aromatics	Sum	
Jul. 2012	7.78	84	54.1±22.9 ^b	0.45 ± 0.48	0.10±0.13	0.37±0.23	55.0±22.5	
Aug. 2012	7.70	76	66.0±69.3	0.46±0.18	nd ^c	0.49±0.03	67.0±69.1	
Sep. 2012	5.92	66	100±118	1.06±0.43	0.08 ± 0.11	0.35±0.36	102±118	
Oct. 2012	-1.50	70	14.7±19.0	1.79±0.08	0.16±0.01	0.22±0.07	16.8±18.9	
Nov. 2012	-14.9	63	2.04±1.76	1.99±0.56	0.20±0.19	0.25±0.15	4.48 ± 2.66	
Dec. 2012	-13.0	45	0.52	0.73	nd	nd	1.25	
Jan. 2013	-16.1	30	0.38±0.02	0.30±0.04	0.03 ± 0.01	0.08 ± 0.01	0.78 ± 0.01	
Feb. 2013	-9.69	49	0.86 ± 0.45	0.52±0.25	0.09 ± 0.02	0.09 ± 0.01	1.55±0.22	
Mar. 2013	-7.83	41	1.56±1.15	0.74±0.59	0.23±0.25	0.12±0.17	2.65±2.15	
Apr. 2013	-3.42	52	2.82±0.20	1.24±0.15	0.15±0.03	0.20±0.03	4.40±0.11	
May 2013	3.77	54	10.1±9.70	1.11±0.13	0.06 ± 0.06	0.27±0.19	11.5±9.97	
Jun. 2013	7.25	55	54.1±42.9	0.83±0.18	0.03 ± 0.04	0.30 ± 0.02	55.3±42.8	
Jul. 2013	10.2	69	41.9	1.41	0.07	0.49	43.9	
Annual	-1.64	58	26.6±44.2	0.97±0.57	0.09±0.10	0.25±0.18	28.0±44.2	

Table 1 SOA tracers at the NC site (ng m^{-3})

^a Temperature and RH are monthly averages; ^b one standard deviation; ^c "nd" means not detected.

	Landing	Seasons	References	SOA tracers			
	Locations			Isoprene ^a	Monoterpenes ^a	β-Caryophyllene	Aromatics
	Nam Co Lake	Whole year	This study	26.6(0.36-184) ^b	0.97(0.11-2.39)	0.09(nd-0.40)	0.25(nd-0.61)
Tibetan Plateau	uQianghai Lake	Summer	(Li <i>et al.</i> 2013)	2.50(0.13-7.15)	2.95(0.30-10.4)	0.87(0.05-2.41)	na ^c
	Himalayas	Summer-autumn	(Stone et al. 2012)	30.7(5.5-105)	13.2(5.6-31.3)	1.6(1.1-2.3)	na
Arctic	Alert	Winter-Summer	(Fu et al. 2009)	0.3(0.08-0.567)	1.6(0.138-5.3)	0.12(0.01-0.372)	na
	Arctic Ocean	Summer	(Fu et al. 2013)	4.0(0.16-31.8)	4.8(0.44-24.1)	0.017(0.005 - 0.048)	na
Clobal agains	Low- to mid-latitude	Fall-Spring	(Fu et al. 2011)	3.6(0.11-22)	2.7(0.02-15)	0.32(0-2.5)	na
Giobal oceans	Antarctic to Arctic	Summer	(Hu et al. 2013)	8.5(0.018-36)	3.0(0.05-20)	na	na
	North Pacific and Arctic	c Summer	(Ding et al. 2013)	0.62(0.12-1.45)	0.06(0.01-0.25)	0.002(nd-0.03)	nd ^d

Table 2 SOA tracers in remote places on the global range (ng m⁻³)

^a compositions are different in different studies. ^b data range in brackets. ^c "na" means not available. ^d "nd" means not detected.