



Source
apportionment of
methane and nitrous
oxide in California's
San Joaquin Valley

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Source apportionment of methane and nitrous oxide in California's San Joaquin Valley at CalNex 2010 via positive matrix factorization

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Abstract

Sources of methane (CH_4) and nitrous oxide (N_2O) were investigated using measurements from a site in southeast Bakersfield as part of the CalNex (California at the Nexus of Air Quality and Climate Change) experiment from 15 May to 30 June 2010. Typical daily minimum mixing ratios of CH_4 and N_2O were higher than daily averages that were simultaneously observed at a similar latitude background station (NOAA, Mauna Loa) by approximately 70 and 0.5 ppb, respectively. Substantial enhancements of CH_4 and N_2O (hourly averages > 500 ppb and > 7 ppb, respectively) were routinely observed suggesting the presence of large regional sources. Collocated measurements of carbon monoxide (CO) and a range of volatile organic compounds (VOCs) (e.g. straight-chain and branched alkanes, cycloalkanes, chlorinated alkanes, aromatics, alcohols, isoprene, terpenes and ketones) were used with a Positive Matrix Factorization (PMF) source apportionment method to estimate the contribution of regional sources to observed enhancements of CH_4 and N_2O .

The PMF technique provided a “top-down” deconstruction of ambient gas-phase observations into broad source categories, yielding a 7-factor solution. We identified these source factors as emissions from evaporative and fugitive; motor vehicles; livestock and dairy; agricultural and soil management; daytime light and temperature driven; non-vehicular urban; and nighttime terpene biogenics and anthropogenics. The dairy and livestock factor accounted for a majority of the CH_4 (70–90 %) enhancements during the duration of the experiments. Propagation of uncertainties in the PMF-derived factor profiles and time series from bootstrapping analysis resulted in a 29 % uncertainty in the CH_4 apportionment to this factor. The dairy and livestock factor was also a principal contributor to the daily enhancements of N_2O (60–70 %) with an uncertainty of 33 %. Agriculture and soil management accounted for ~ 20 –25 % of N_2O enhancements over the course of a day, not surprisingly given that organic and synthetic fertilizers are known to be a major source of N_2O . The evaporative/fugitive source profile resembles a mix of petroleum operation and non-tailpipe evaporative gasoline

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processes including fossil fuel combustion have been estimated to account for 15 % of total global anthropogenic N₂O emissions (Denman et al., 2007).

In 2006, the state of California adopted Assembly Bill 32 (AB32) into a law known as the Global Warming Solutions Act, which committed the state to cap and reduce anthropogenic GHG emissions to 1990 levels by 2020. A statewide GHG emission inventory (CARB, 2013) maintained by the Air Resources Board of California (CARB) is used to report, verify and regulate emissions from GHG sources. In 2011, CH₄ accounted for 32.5 million metric tonnes (MMT) CO₂-eq representing 6.2 % of the statewide GHG emissions, while N₂O emissions totaled 6 MMT CO₂-eq representing about 3 % of the GHG emissions inventory (Fig. 1). CARB's accurate knowledge of GHG sources and statewide emissions is key to the success of any climate change mitigation strategy under AB32. CARB's GHG inventory is a "bottom-up" summation of emissions derived from emission factors and activity data. The bottom-up approach is reasonably accurate for estimation and verification of emissions from mobile and point sources (vehicle tailpipes, power plant stacks etc.) where the input variables are well-understood and well-quantified. The main anthropogenic sources of CH₄ in the CARB inventory include ruminant livestock and manure management, landfills, wastewater treatment, fugitive and process losses from oil and gas production and transmission, and rice cultivation while the major N₂O sources are agricultural soil management, livestock manure management and vehicle fuel combustion (CARB, 2013). The emission factors for many of these sources have large uncertainties as they are biological and their production and release mechanisms are inadequately understood thus making these sources unsuitable for direct measurements (e.g. emissions of N₂O from farmlands). Many of these sources (e.g. CH₄ from landfills) are susceptible to spatial heterogeneity and seasonal variability. Unfortunately, a more detailed understanding of source characteristics is made difficult because CH₄ and N₂O are often emitted from a mix of point and area sources within the same source facility (e.g. dairies in the agricultural sector) making bottom-up estimation uncertain. There is a lack of direct measurement data or "top-down" measurement-based approaches to independently validate seasonal trends and

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inventory estimates of CH₄ and N₂O in California's Central Valley, which has a mix of several agricultural sources and oil and gas operations, both of which are known major sources of GHGs.

In the recent past, regional emission estimates derived from measurements from a tall tower at Walnut Grove in Central California coupled with inverse dispersion techniques (Fischer et al., 2009) reported underestimation of CH₄ and N₂O emissions especially in the Central Valley. Comparison of regional surface footprints determined from WRF-STILT algorithm between October–December 2007 indicate posterior CH₄ emissions to be higher than California-specific inventory estimates by 37 ± 21 % (Zhao et al., 2009). Predicted livestock CH₄ emissions are 63 ± 22 % higher than a priori estimates. A study over a longer period (December 2007–November 2008) at the same tower (Jeong et al., 2012a) generated posterior CH₄ estimates that were 55–84 % larger than California-specific prior emissions for a region within 150 km from the tower. For N₂O, inverse estimates for the same sub-regions (using either EDGAR32 and EDGAR42 a priori maps) were about twice as much as a priori EDGAR inventories (Jeong et al., 2012b). Recent studies have incorporated WRF-STILT inverse analysis on airborne observations across California (Santoni et al., 2012). The authors conclude that CARB CH₄ budget is underestimated by a factor of 1.64 with aircraft-derived emissions from cattle and manure management, landfills, rice, and natural gas infrastructure being around 75, 22, 460, and 430 % more than CARB's current estimates for these categories, respectively. Statistical source footprints of CH₄ emissions generated using FLEXPART-WRF modeling and CalNex-Bakersfield CH₄ concentration data are consistent with locations of dairies in the region (Gentner et al., 2014a). The authors conclude that the majority of CH₄ emissions in the region originate from dairy operations. Scaled-up CH₄ rice cultivation estimates derived from aircraft CH₄/CO₂ flux ratio observations over rice paddies in the Sacramento valley during the growing season when emissions are at their strongest (Peischl et al., 2012) are around three times larger than inventory estimates. CH₄ budgets derived for the Los Angeles (LA) basin from aircraft observations (Peischl et al., 2013) and studies involving comparison

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with CO enhancements and inventory at Mt. Wilson (Hsu et al., 2010; Wunch et al., 2009) indicate higher atmospheric CH₄ emissions in the LA basin than expected from bottom-up accounting.

Recent literature seems to suggest that the CARB bottom-up inventory is underestimating CH₄ and N₂O sources, especially from the livestock sector and perhaps from the oil and gas industry as well. Source apportionment studies of non-CO₂ GHGs over the Central Valley can provide critical information about under-inventoried or unknown sources that seek to bridge the gap between “bottom-up” and “top-down” methods. GHG emission inventories can potentially be constrained through simultaneous measurements of GHGs and multiple gas species (VOCs) that are tracers of various source categories. This study provides CH₄ and N₂O source attribution during a six week study involving a complete suite of continuous GHG and VOC tracer measurements during the CalNex 2010 campaign in Bakersfield, located in the southern part of the Central Valley (May–June 2010). The objective of this study is to partition the measured CH₄, N₂O and VOC enhancements into statistically unique combinations using Positive Matrix Factorization (PMF) apportionment technique. We classify these combinations as plausible source factors based on our prior knowledge of the chemical origin of mutually co-varying groups of VOC tracers found in each statistical combination. We examine the source categorization using observations from source-specific, ground site and airborne measurements and results from other source apportionment studies. We also compare the relative abundance of CH₄ and N₂O enhancements in each source factor with the CARB inventory estimates in order to assess the inventory.

2 Experimental Setup

2.1 Field Site and Meteorology

Measurements were conducted from 19 May to 25 June 2010 at the Bakersfield CalNex supersite (35.3463° N, 118.9654° W) (Fig. 2) in the southern San Joaquin Valley (SJV)

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(Ryerson et al., 2013). The SJV represents the southern half of California's Central Valley. It is 60 to 100 km wide, surrounded on three sides by mountains, with the Coastal Ranges to the west, the Sierra Nevada Mountains to the east, and the Tehachapi Mountains to the southeast.

The measurement site was located to the southeast of the Bakersfield urban core in Kern County (Fig. 2). The east–west Highway 58 is located about 0.8 km to the north; the north–south Highway 99 about 7 km to the west. The city's main waste water treatment plant (WWTP) and its settling ponds are located to the east and south of the site (< 2.5 km), respectively. Numerous dairy and livestock operations are located to the south-southwest of the site at 10 km distance or farther. The metropolitan region has three major oil refineries located within 10 km from the site (two to the northwest; one to the southeast). A majority of Kern County's high-production active oil fields (> 10 000 barrels (bbl) per day) (CDC, 2013) are located to the west/northwest and are distant (~ 40–100 km). Kern River oilfield (~ 60 000 bbl day⁻¹), one of the largest in the country, and Kern Front (~ 11 000 bbl day⁻¹) are located about 10–15 km to the north. There are several other oil fields dotted within the urban core (5–20 km) which are less productive (< 2000 bbl day⁻¹) or not active (< 100 bbl day⁻¹). The whole region is covered with agricultural farmlands with almonds, grapes, citrus, carrots and pistachios amongst the top commodities by value and acreage (KernAg, 2010).

The meteorology and transport of air masses in the southern SJV is complex and has been addressed previously (Bao et al., 2007; Beaver and Palazoglu, 2009). The wind rose plots (Fig. 3) shown here present a simplified distribution of microscale wind speed and direction at the site, the latter often being non-linear over larger spatial scales. The plots depict broad differences in meteorology during daytime and nighttime. A mesoscale representation of the site meteorology during this study period was evaluated through back-trajectory footprints generated from each hourly sample using FLEXPART Lagrangian transport model with WRF meteorological modeling (Gentner et al., 2014a). The 6 and 12 h back trajectory footprints are generated on a 4 km × 4 km resolution with simulations originating from top of the 18 m tall tower. The site experi-

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mally insulated temperature controlled 7 foot wide cargo wagon trailer developed by the GHG instrument manufacturers (LGR Inc.). CO was coincidentally measured using another instrument (Teledyne API, USA, Model # M300EU2) with a precision of 0.5 % of reading and output as 1 min averages. The mixing ratios from the two collocated CO instruments correlated well ($r \sim 0.99$) and provided a good stability check for the LGR instrumentation. Scaled Teledyne CO data was used to gap-fill the LGR CO data. The coincident gas-phase VOC measurements were made using a gas chromatograph (GC) with a quadrupole mass selective detector and a flame ionization detector (Gentner et al., 2012).

Hourly calibration checks of the three GHGs and CO were performed using near-ambient level scuba tank standards through the entire campaign. During data processing, final concentrations were generated from the raw data values using scaling factors obtained from comparison of measured and target concentrations during calibration checks. Diurnal plots of measured species are generated from 1 min averages. PMF analyses in the following sections are based on 30 min averages to match the time resolution of VOC measurements. The metrological data measured at the top of the tower included relative humidity (RH), temperature (T), and wind speed (WS) and direction (WD).

3.2 Positive Matrix Factorization (PMF)

Source apportionment techniques like PMF have been used in the past to apportion ambient concentration datasets into mutually co-varying groups of species. PMF is especially suitable for studies where a priori knowledge of number of sources impacting the measurements, chemical nature of source profiles and relative contribution of each source to the concentration time series of a measured compound are unknown or cannot be assumed. PMF has been applied to ambient particulate matter studies (Kim et al., 2004; Lee et al., 1999); in determining sources of atmospheric organic aerosols (OA) (Ulbrich et al., 2009; Slowik et al., 2010; Williams et al., 2010); and in gas phase measurements of VOCs in major metropolitan cities (Bon et al., 2011; Brown et al.,

2007). PMF is a receptor-only unmixing model which breaks down a measured data set containing time series of a number of compounds into a mass balance of an arbitrary number of constant source factor profiles (FP) with varying concentrations over the time of the data set (time series *or* TS) (Ulbrich et al., 2009).

5 In real world ambient scenarios, sources of emissions are often not known or well-understood. PMF technique requires no a priori information about the number or composition of factor profiles or time trends of those profiles. The constraint of non-negativity in PMF ensures that all values in the derived factor profiles and their contributions are constrained to be positive leading to physically meaningful solutions. PMF attributes a measure of experimental uncertainty (or weight) to each input measurement. 10 Data point weights allow the level of influence to be related to the level of confidence the analyst has in the measured data (Hopke, 2000). In this way, problematic data such as outliers, below-detection-limit (BDL), or altogether missing data can still be substituted into the model with appropriated weight adjustment (Comero et al., 2009) allowing for 15 a larger input data set, and hence a more robust analysis. PMF results are quantitative; it is possible to obtain chemical composition of sources determined by the model (Comero et al., 2009). PMF is not data-sensitive and can be applied to data sets that are not homogenous and/or require normalization without introducing artifacts.

3.3 Mathematical framework of PMF

20 The PMF model is described in greater detail elsewhere (Paatero and Tapper, 1994; Paatero, 1997; Comero et al., 2009; Ulbrich et al., 2009) and we will briefly mention some concepts relevant to the understanding of the analysis carried out in this study. The PMF input parameters involve a $m \times n$ data matrix \mathbf{X} with i rows containing mixing ratios at sampling time t_i and j columns containing time series of each tracer j . 25 A corresponding uncertainty matrix \mathbf{S} reports measurement precision (uncertainty) of

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of CH₄ vs. ppt level of propene) and improve the visual attributes of PMF output plots to follow. Data points denoting zero enhancement (lower limit) were replaced by a very small positive number (i.e. exp(-5)) to avoid “zeros” in the data matrix **X**.

$$x_{ij} = (\text{Mixing ratio}_{ij} - \text{Background}_j) / (\text{Maximum mixing ratio}_j - \text{Background}_j) \quad (4)$$

5 For the VOCs, guidelines set forth by (Williams et al., 2010) were adopted to calculate the uncertainty estimates. An analytical uncertainty (AU) of 10 % was used; a limit of detection (LOD) of 1 ppt and a limit of quantification (LOQ) of 2 ppt (Gentner et al., 2012) was used to calculate the total uncertainty for each x_{ij} :

$$s_{ij} \equiv 2 \times \text{LOD}, \quad \text{if } x_{ij} \leq \text{LOD}, \quad (5a)$$

$$10 \quad s_{ij} \equiv \text{LOQ}, \quad \text{if } \text{LOD} < x_{ij} \leq \text{LOQ}, \quad (5b)$$

$$s_{ij} \equiv \left((\text{AU} \times x_{ij})^2 + (\text{LOD})^2 \right)^{0.5}, \quad \text{if } x_{ij} > \text{LOQ} \quad (5c)$$

Using this approach, the detection limit dictates the errors for low enhancements (near LOD) while the errors for larger enhancements of VOCs are tied more to the magnitude of the data value (x_{ij}) itself.

15 The GHG and CO measurements have high precision and significantly lower detection limits than ambient levels. The relatively low values of GHGs in the uncertainty matrix, compared to VOCs, is substituted with those calculated using a custom approach. The GHG and CO uncertainties are assumed to be proportional to the square root of the data value and an arbitrary scaling factor determined through trial and error
20 in order to produce lower values of QQ_{exp}^{-1} :

$$s_{ij} \equiv A \times (x_{ij})^{0.5}, \quad \text{where } A = 1 \text{ (for CH}_4\text{)}, 0.25 \text{ (for CO}_2\text{)}, 0.5 \text{ (for CO)}, 0.1 \text{ (for N}_2\text{O)} \quad (6)$$

This method attributes larger percentage uncertainties to smaller enhancements and hence lesser weight in the final solution and vice versa. This approach leads to an

uncertainty matrix that attributes relatively similar percentage errors to both GHGs and VOCs, which should lead to a better fitting of the data through PMF.

Missing values are replaced by geometric mean of the tracer time series and their accompanying uncertainties are set at four times this geometric mean (Polissar et al., 1998) to decrease their weight in the solution. Based on the a priori treatment of the entire input data (scaling) and the corresponding outputs of the PMF analysis, a weighting-approach (for measurements from different instruments) as used in (Slowik et al., 2010) is not found to be necessary.

3.5 PMF source analysis

We use the customized software tool (PMF Evaluation Tool v2.04, PET) developed by Ulbrich et al. (2009) in Igor Pro (Wavemetrics Inc., Portland, Oregon) to run PMF, evaluate the outputs and generate statistics. The PET calls the PMF2 algorithm (described in detail in Ulbrich et al., 2009) to solve the bilinear model for a given set of matrices \mathbf{X} and \mathbf{S} for different numbers of factors p and for different values of FPEAK or SEED (defined and described later). The tool also stores the results for each of these combinations in a user friendly interface that allows simultaneous display of the factor profiles (FP) and time series (TS) of a chosen solution along with residual plots for individual tracers. A detailed explanation of PMF analysis performed in this study is provided in the Supplement. The supplement describes the PMF methodology of how the final number of user-defined factors was chosen (Sect. S1), the outcomes of linear transformations (rotations) of various PMF solutions (Sect. S2) and how uncertainties in the chosen solution were derived (Sect. S3). The standard deviations in the mass fractions of individual tracers in each factor profile and time series of each factor mass is evaluated using a bootstrapping analysis (Norris et al., 2008; Ulbrich et al., 2009) and described in Sect. S3. These error estimates are combined and propagated to derive PMF-based uncertainties for each factor's contribution to source-apportioned diurnal enhancements for a specific compound (Sect. 4).

4 Results and discussion

In Bakersfield, there are a multitude of pollutant sources, ranging from local to regional, from biogenic to anthropogenic, and from primary to secondary. We recognize that PMF analysis is not capable of precise separation of all sources. In PMF analysis, the analyst chooses the number of factor profiles to include in the solution and assigns a source category interpretation for each identified factor. The PMF factors are not unique sources but really statistical combinations of coincident sources. The chemical profile of each factor may contain some contributions from multiple sources that are co-located, or have a similar diurnal pattern. Such limitations have been observed previously by Williams et al. (2010) while applying PMF in an urban-industrial setting like Riverside, California. The user must infer the dominant source contributions to these individual factors. Our factor profile (FP) nomenclature is based on the closest explanation of the nature and distribution of emission sources in the region. The source factor names should be treated with caution bearing in mind the physical constraints of the solution and not used to over-explain our interpretation of the region's CH₄ and N₂O inventories.

A seven factor solution has been chosen to optimally explain the variability of the included trace gases. The factors have been named based on our interpretation of the emission "source" categories they represent, with corresponding colors which remain consistent in the discussion across the rest of the paper: evaporative and fugitive (black), dairy and livestock (orange), motor vehicles (red), agricultural + soil management (purple), daytime biogenics + secondary organics (light blue), non-vehicular urban (green) and nighttime anthropogenic + terpene biogenics (navy blue). Figure 4 presents the Factor Profile (FP) plots of each factor. The sum of the normalized contributions of the 50 species in each "source" is equal to 1 in the FP plots. Figures 5a through 5g present the diurnal profiles based on mean hourly concentrations (in normalized units) of each PMF factor with standard deviations explaining the variability. The interpretation of the individual FPs is discussed below (in Sects. 4.2–4.8). Molar

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emission factor (EF) of tracers with respect to (w.r.t) one another can be derived for each FP. These EFs can then be compared to those from previous source-specific and apportionment studies (Table 2 through 5). The ratio of PMF-derived total CH₄ enhancement to the input measured CH₄ enhancement ranges from 0.90 to 0.95 through the whole time series except outliers with really high values (> 500 ppb). For N₂O, the ratio is somewhat lower (0.82–0.92) and this is reflected in the PMF-derived uncertainties. The apportionment of some N₂O mass into a statistically weak and time-varying factor is discussed in Sect. 4.5. The general assessment is that PMF analysis is able to reconstruct a majority of the measured enhancements for both CH₄ and N₂O.

4.1 Time trends of measured CH₄, CO₂, CO, and N₂O

The time series of CH₄, CO₂, CO, and N₂O mixing ratios have been plotted in Fig. 6a through d while the diurnal variations have been plotted in Fig. 6e–h, respectively. The color markers in each plot indicate the median wind direction. The daily minima for the three GHGs and CO occur during the late afternoon period when daytime heating, mixing and subsequent dilution occurs rapidly. The daily minimum values of CH₄ and N₂O were larger than that observed at National Oceanic and Atmospheric Administration’s (NOAA) Mauna Loa station (Dlugokencky et al., 2014) by at least 70 and 0.5 ppb, respectively, for this period. This indicates that there are large regional sources of these two GHGs that keep the mixing ratio levels high. Winds during the highest temperature period between noon and evening (12:00–20:00 LT) almost always arrive through the urban core in the northwest. Any PMF factor whose dominant source direction is northwest is likely to contain contributions from VOCs emitted from urban sources, regional sources further upwind or contain contributions from secondary tracers generated from photochemical processing during the day. The three GHGs show a sharp increase during the nighttime when the inversion layer builds up and traps primary emissions close to the ground. For CO, measured concentrations show two distinct peaks in the diurnal plot (Fig. 6g). The observed early morning peak in the concentration is a combination of decreased dilution and fresh emissions from the morning motor vehicle traffic. The

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late evening peak in CO concentrations is not coincident with rush hour and is a result of build-up of evening emissions in the boundary layer that is getting shallower as the night progresses. Figure 6a indicates CH₄ enhancements of 500 ppb or more on almost every night with peak mixing ratios exceeding 3000 ppb on several occasions indicating an active methane source(s) in the region. Figure 6d shows that peak N₂O mixing ratios rise above 330 ppb on almost every night suggesting large sources in the region. Huge enhancements of CH₄, CO₂ and N₂O (on DOY 157, 164, and 165) (in Fig. 6a, b and d, respectively) may appear well-correlated to each other due to regional sources emitting into the inversion layer. However, the shapes of the diurnal cycles differ indicating different emission distributions, with the early morning maximum in CH₄ occurring before the maxima for CO₂ and N₂O, and the morning maximum for CO occurring slightly later. These differences in timing allow PMF analysis to differentiate their contributions into separate factors.

4.2 Factor 1: evaporative and fugitive emissions

Factor 1 has a chemical signature indicative of evaporative and fugitive losses of VOCs. The FP of this source is dominated by C₃ to C₆ straight-chain and branched alkanes and some cycloalkanes (Fig. 4). The average diurnal cycle of Factor 1 (Fig. 5a) shows a broad peak during late night and early morning hours after which the concentrations begin to decrease as the day proceeds reaching a minimum at sunset before beginning to rise again. This is strong indication of a source containing primary emissions that build up in the shallow pronounced nighttime inversions of southern SJV. The subsequent dilution of primary emissions as the mixed layer expands leads to low concentrations during the daytime.

Most of the propane, n-butane and pentanes signal is apportioned to this factor, but not the typical vehicle emission tracers like isooctane or CO or any of the alkenes or aromatics. Absence of these tracers in the FP suggests this factor is not related to vehicular exhaust and is a combination of non-tailpipe emissions and fugitive losses from petroleum operations. None of the CH₄ signal at the SJV site is apportioned to this fac-

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a PMF derived $\text{N}_2\text{O}/\text{CO}$ EF of 0.00015 (mol mol^{-1}). The EMFAC generated $\text{N}_2\text{O}/\text{CO}$ EF in SJV during summer of 2010 is more than 20 times higher at 0.0034 (mol mol^{-1}). The PMF derived “vehicle emissions” contribution to N_2O is in stark contrast to the inventory and is an important outcome suggesting a significant error in the statewide inventory for N_2O .

4.4 Factor 3: dairy and livestock emissions

Factor 3 has a chemical signature indicative of emissions from dairy operations. This source factor is the largest contributor to CH_4 enhancements (Fig. 7a) and a significant portion of the N_2O signal (Fig. 7c). The FP also has major contributions from methanol (MeOH) and ethanol (EtOH), with minor contributions from aldehydes and ketones (Fig. 4). A separate PMF analysis with a broader set of VOC measurements at the same site showed that most of the acetic acid (CH_3COOH) and some formaldehyde (HCHO) signal attributed to this factor as well (Allen Goldstein, personal communication, 2014). All the above-mentioned VOCs are emitted in significant quantities from dairy operations and cattle feedlots (Filipy et al., 2006; Shaw et al., 2007; Ngwabie et al., 2008; Chung et al., 2010). About 70–90 % of the diurnal CH_4 signal is attributed to this factor (Fig. 7a) depending on the time of day. From propagation of errors, an uncertainty of 29 % is determined in the diurnal CH_4 enhancements in Factor 3. This source factor contributes about 60–70 % of the total N_2O daily enhancements as seen in Fig. 7c with an uncertainty of 33 %.

Comparing the Factor 3 profile to dairy source profiles from various studies is challenging. A dairy is, in essence, a collection of area sources with distinct emission pathways and chemical characteristics. Hence, a lot of dairy studies do not look at facility-wide emissions instead focusing on specific area sources within the facility. In contrast, PMF captures the covariance of CH_4 , N_2O , and VOCs emitted from the ensemble source as downwind plumes from dairies arrive at the site. Table 4 compares the PMF derived EFs of CH_4 w.r.t MeOH and EtOH with those from other studies. Pre-

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viously, cow chamber experiments (Shaw et al., 2007; Sun et al., 2008) have measured emissions from ruminants and their fresh manure; emissions have also been studied in a German cowshed (Ngwabie et al., 2008) and EFs have been derived from SJV dairy plumes sampled from aircrafts (Gentner et al., 2014a; Guha et al., 2014). Since enteric fermentation and waste manure is the predominant CH_4 source in dairies, CH_4 emission rates calculated by Shaw et al. (2007) are representative of a whole facility. However, their MeOH/CH_4 ratios are lower than those measured by PMF and aircraft studies. Animal feed and silage are the dominant source of many VOCs including MeOH and EtOH (Alanis et al., 2010; Howard et al., 2010) and the ratios in (Shaw et al., 2007) do not reflect these emissions. In (Ngwabie et al., 2008), experiments were performed in cold winter conditions (-2 to 8°C) when temperature dependent VOC emissions from silage and feed are at a minimum. The authors comment that MeOH emissions from California dairies is likely higher, as the alfalfa-based feed are a big source of MeOH owing to its high pectin content (Galbally and Kirstine, 2002). These observations explain why MeOH/CH_4 ratios in these studies are lower than PMF derived ratios. The PMF range for EtOH/CH_4 EF for Factor 3 agrees with the slope derived from ground-site data (Gentner et al., 2014a) and is similar, but somewhat larger than the German dairy study (Ngwabie et al., 2008). Miller and Varel (2001) and Filipuy et al. (2006) did not measure CH_4 emission rates so a direct derivation of EF w.r.t CH_4 is not possible. These studies, however, reported EtOH emission rates (from dairies and feedlots in United States) which are used to derive EFs w.r.t to CH_4 using an averaged CH_4 emission rate from (Shaw et al., 2007). Using this method, we get EFs that are comparable to PMF derived EF of CH_4/EtOH (Table 4). Hence, we demonstrate within reasonable terms that the relative fractions of masses in Factor 3 are consistent with CH_4 and VOC emissions from dairies.

Enteric fermentation is a part of the normal digestive process of livestock such as cows, and is a large source of CH_4 while the storage and management of animal manure in lagoons or holding tanks is also a major source of CH_4 . According to the state GHG inventory (CARB, 2013), $\sim 58\%$ of the statewide CH_4 emissions results

factor has uncertainties greater than those for other factors (Fig. S4). This is potentially because not all crops emit the same combination of VOCs nor are all agricultural fields fertilized at the same time. The existence of this statistically weak factor is confirmed by bootstrapping runs (Sect. S3) and numerous PMF trials all of which produce a distinct factor with N₂O as a dominant contributor along with certain biogenic VOCs, though often in varying proportions. CO₂ is not included in the PMF analysis reported in the paper, but PMF runs involving CO₂ indicate that most of the CO₂ is apportioned to this factor. Plant and soil respiration (especially during the night) is a major source of CO₂ and the apportionment of CO₂ to Factor 4 confirms the nature of this source. The temporal correlation between CO₂ and N₂O is also evident in their average diurnal cycles (Fig. 6f and h), which have a coincident early morning peak. The absence of monoterpenes from the FP of this factor can be explained by their shorter atmospheric lifetimes compared to VOCs like acetone and MeOH and the rapid daytime mixing which dilutes the terpenoid emissions arriving at the site during the day. At nights, when the atmospheric dilution has been reduced to a low, monoterpene emissions from agriculture are more likely to get apportioned into a separate source factor dominant during nighttime, when temperature-sensitive biogenic emissions of MeOH and acetone can be expected to be a minor constituent in the FP (see Sect. 4.8).

Factor 4 is a significant source of GHGs contributing about 20–25 % of the total N₂O enhancements in the diurnal cycle (Fig. 7c) with a relatively larger uncertainty of 70 %. Kern County is one of the premier agricultural counties of California accounting for USD 4.2 billion (about 18 %) of the total agricultural revenue from fruits and nuts, vegetables and field crops (KernAg, 2011; CASR, 2013) and is also the biggest consumer of synthetic fertilizers. Agricultural soil management accounts for about 60 % of the statewide N₂O emission inventory (CARB, 2013). Our assessment of diurnal source distribution of N₂O emissions from the agriculture source factor (Fig. 7c) is consistent with the inventory estimates from agricultural and soil management sector.

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4.6 Factor 5: daytime biogenics and secondary organics

The chemical composition and diurnal profile of Factor 5 points to a source whose emissions are either primary biogenic VOCs with temperature-dependent emissions (e.g. isoprene), or products of photochemical oxidation of primary VOCs (e.g. acetone) (Fig. 4). Isoprene is a dominant component of the source FP and is mostly apportioned to Factor 5. Figure 5c shows a steady increase in the PMF factor mass concentration during the daytime hours that hits a peak during afternoons indicating that this source is dependent on sunlight and temperature. Potential source contributions come from oak forests on the foothills of the western edge of the SJV or scattered isoprene producing plants in the SJV (note that most crops do not emit significant amounts of isoprene). Factor 5 includes contribution from VOCs that have primary light and temperature driven (crops), as well as secondary sources in the Central Valley e.g. acetone (Goldstein and Schade, 2000), methanol (Gentner et al., 2014b) and aldehydes. A similar PMF analysis with a different objective (Goldstein et al., 2014) shows that secondary organics like glyoxal, formaldehyde and formic acid mostly apportion to Factor 5. The CO apportioned to this factor could potentially be a product of mobile and/or stationary combustion co-located or up/downwind of the biogenic VOC source. CO can also come from coincident isoprene oxidation (Hudman et al., 2008). This daytime source is not responsible for any of the observed CH₄ and N₂O enhancements.

4.7 Factor 6: non-vehicular/miscellaneous urban emissions

The chemical signature of Factor 6 is composed of VOCs associated with an array of applications and processes, including solvents, fumigants, industrial-byproducts, etc. The diurnal profile of Factor 6 (Fig. 5e) is somewhat different from that of evaporative and fugitive source (Fig. 5a) and dairies (Fig. 5c) in that even during the middle of the day when vertical mixing is at its strongest, the enhancements contributing to the factor are substantial. This suggests that the source(s) is in close proximity to the site and hence most likely located within the urban core. The FP has CO as an im-

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direction of the site for only a certain part of the diurnal cycle, is expected to be more directionally constrained and emissions profile from such a source will look similar to the diurnal profile of Factor 7. Among source factors which contain non-negligible fractional contribution of both CH₄ and N₂O (i.e. dairies, agriculture and soil management, and Factor 7), the PMF derived CH₄/N₂O EF of 42 ± 20 (gC(gC)⁻¹) from Factor 7 is most similar to the bottom-up inventory EF of 56 (gC(gC)⁻¹) for waste water treatment in Kern County (KernGHG, 2012). Given the proximity of the WWTP and previous observations of GHGs from them, it is possible that there is a minor but noticeable contribution (~ 5 %) to CH₄ and N₂O enhancements from this nighttime source (Fig. 7a and c).

5 Implications

This study demonstrates the potential of the PMF technique to apportion atmospheric gas-phase observations of CH₄ and N₂O into source categories using a broad array of tracers. PMF is not commonly employed to perform for source attribution of these GHGs because studies generally lack simultaneous measurements of specific source-markers. Applying this statistical technique on a GHG-VOC unified data set, well-represented by a broad suite of VOC classes, allows a set of compounds acting as source markers to be partitioned into separate profiles leading to easier identification of their sources.

We provide clear analysis that dairy and livestock operations are the largest sources of emissions in the Bakersfield region accounting for a majority of the CH₄ (70–90 %) and N₂O (50–60 %) emissions. As per the CARB inventory (Fig. 1), dairy operations are the dominant source of non-CO₂ GHGs in the state and our analysis agrees with that broad trend. However, in the recent past, a number of top-down CH₄ and N₂O emission studies in the Central Valley have reported underestimation of the non-CO₂ GHG inventory (Zhao et al., 2009; Santoni et al., 2012; Jeong et al., 2012a, b; Miller et al., 2013). These studies attribute a majority of this underestimation to the dairy

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cline in tailpipe N₂O emissions. Bakersfield is a fairly large population urban region (~ 500 000) and the essentially non-existent contribution of the PMF vehicle emissions source to the N₂O apportionment and large divergence of the PMF derived N₂O/CO EF from the state inventory EF for motor vehicles is a significant outcome pointing to overestimation of N₂O from motor vehicles in the inventory.

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Table 2. Comparison of light alkane ratios to propane ($\text{gC}(\text{gC})^{-1}$) from PMF fugitive and evaporative factor with those from other PMF studies and oil and gas operations.

Study	Source	propane	<i>n</i> -butane	<i>n</i> -pentane	<i>n</i> -hexane	isopentane
Bakersfield PMF evaporative and fugitive factor ^a	This study	1	0.52 ± 0.02	0.18 ± 0.01	0.06 ± 0.01	0.33 ± 0.02
Bakersfield petroleum operations source profile ^b	Gentner et al. (2014)	1	0.53 ± 0.1	0.09 ± 0.02	0.04 ± 0.01	0.08 ± 0.02
Mexico city PMF LPG factor ^c	Bon et al. (2011)	1	0.5 (0.4–0.7)	0.05 (0.04–0.07)	0.02 (0.02–0.03)	0.07 (0.06–0.1)
Wattenberg field BAO, Colorado ^d	Gilman et al. (2013)	1	0.75 ± 1.37	0.32 ± 0.6	0.08 ± 0.13	0.28 ± 0.52
Wattenberg field BAO, Colorado ^e	Petron et al. (2012)	1	0.58–0.65	0.22–0.31	NA	0.22–0.31
PMF natural gas and evaporation factor, Houston Ship Channel ^f	Leuchner and Rappengluck (2010)	1	0.33	0.27	0.12	0.37
PMF natural gas factor, Houston Ship Channel ^h	Buzcu and Fraser (2006)	1	0.67 ± 0.16	0.07 ± 0.18	NA	NA

^a Uncertainties calculated from propagation of errors (SDs) over FPEAK range of –1.6 to 0.4.

^b Ratios calculated from Table 4, Gentner et al. (2014); uncertainties defined as ±20% to account for variability in oil well data.

^c Uncertainties calculated from propagation of uncertainties over FPEAK range of –3 to 3.

^d Emission ratios derived from multivariate regression analysis; error bars derived from propagation of uncertainty using mean and SD of samples.

^e Range over 5 regressions conducted over data collected in different seasons and from mobile lab samples.

^f Ratios derived from mean and SDs, with propagation of uncertainty.

^g Estimated from Fig. 2, Leuchner and Rappengluck (2010).

^h Estimated from Fig. 2, Buzcu and Fraser (2006).

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Table 3. Comparison of hydrocarbon ratios to toluene ($\text{gC}(\text{gC})^{-1}$) from PMF vehicle emission factor with similar ratios from other California specific studies.

Study	Bakersfield PMF vehicle emissions factor ^a	Bakersfield gasoline source profile ^{b, c}	Riverside liquid gasoline profile ^e	CalNex Los Angeles ambient emission ratios ^g
Source	This study	Gentner et al. (2014)	Gentner et al. (2009)	Borbon et al. (2013)
CH ₄	8.1 ± 2.1	NA	NA	NA
CO	14.0 ± 0.4	NA	NA	22.26
toluene	1	1	1	1
isopentane	0.69 ± 0.01	0.77 ± 0.04	0.64–0.84	1.95
isooctane	0.29 ± 0.03	0.34 ± 0.02	0.64–0.80	NA
<i>n</i> -dodecane	0.03 ± 0.001	(0.04 ± 0.004) ^d	NA	NA
methylcyclopentane	0.24 ± 0.01	0.32 ± 0.02	NA	NA
ethyl benzene	0.17 ± 0.01	0.14 ± 0.01	NA	0.2
<i>m/p</i> -xylene	0.65 ± 0.01	0.65 ± 0.03	(0.45–0.52) ^f	0.64
<i>o</i> -xylene	0.22 ± 0.01	0.23 ± 0.01	NA	0.24

^a Errors are SD of 12 unique PMF solutions between FPEAK = −1.6 to +0.4; see Sect. S2.

^b Derived from liquid gasoline fuel speciation profile (Table S9; Gentner et al., 2012).

^c Errors bars derived from propagation of uncertainties.

^d Derived by combining diesel fuel and gasoline speciation profile (Tables S9 and S10; Gentner et al., 2012).

^e Summer data.

^f Only *m*-xylene.

^g Derived from Linear Regression Fit slope of scatterplot from CalNex Pasadena supersite samples.

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Table 5. Comparison of PMF agricultural and soil management emission factor for acetone vs. methanol ($\text{gC}(\text{gC})^{-1}$) with ratios of basal emission factors generated for major crops grown in the Kern County. Errors denote SDs computed by propagation of uncertainty.

Bakersfield PMF agricultural and soil management factor	Almond greenhouse summer 2008	Table grape greenhouse summer 2008	Pistachio greenhouse summer 2008	Navel oranges greenhouse summer 2008*	Valencia oranges greenhouse summer 2008
This study 0.58 ± 0.37	Gentner et al. (2014b) 0.14 ± 0.2	Gentner et al. (2014b) 0.04 ± 0.02	Gentner et al. (2014b) 0.5 ± 0.6	Fares et al. (2011) 0.57 ± 0.1	Fares et al. (2012) 0.5 ± 0.3

* branch with flowers not removed.



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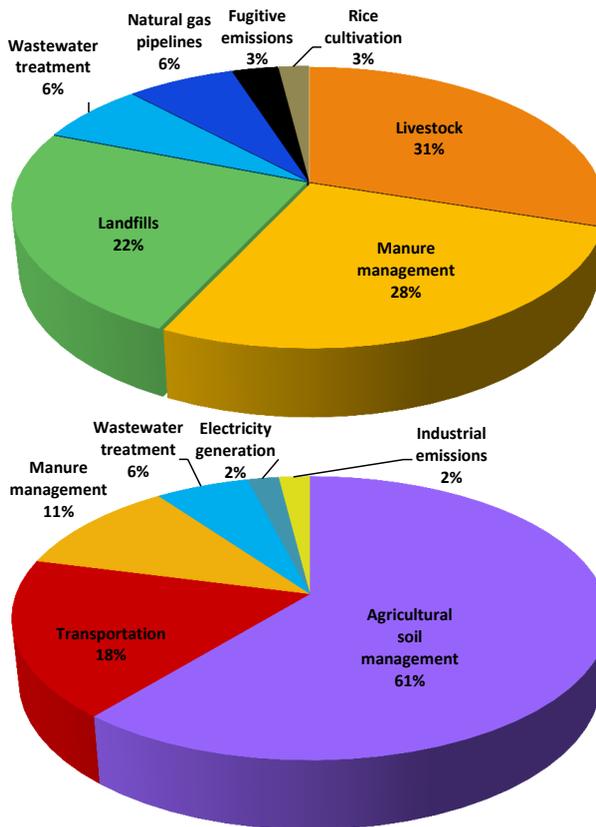


Figure 1. 2011 California emission inventory for (top) methane (CH₄) – 32.5 million tCO₂ eq at GWP = 25; and (bottom) nitrous oxide (N₂O) – 13.4 million tCO₂ eq at GWP = 298. (Source: CARB GHG Inventory Tool, August 2013).

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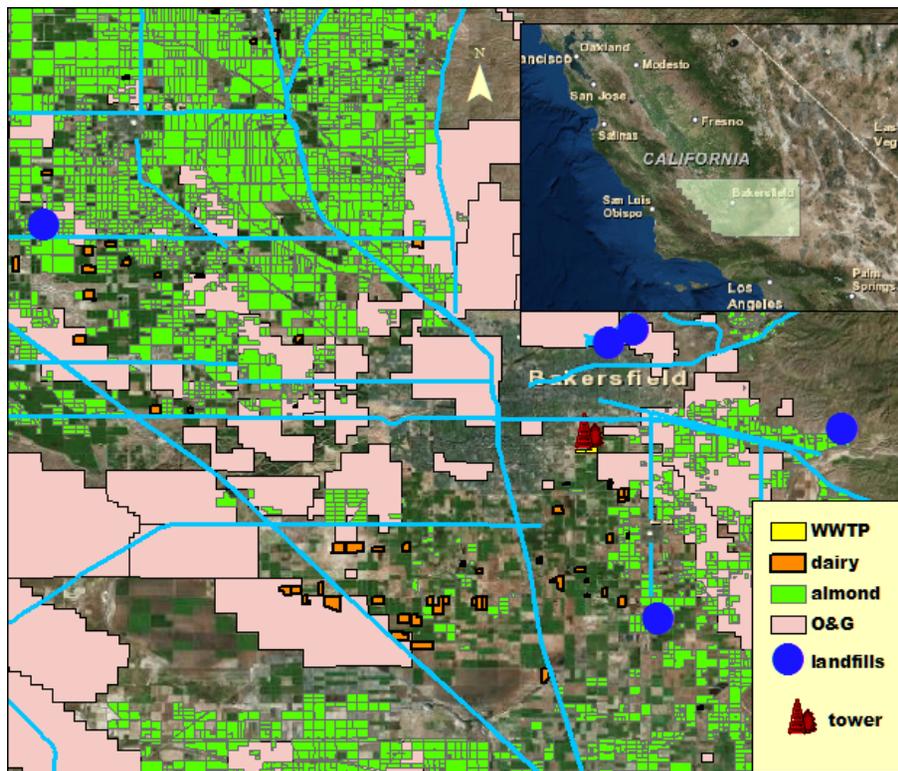


Figure 2. Map of potential sources of methane and nitrous oxide in/around the city of Bakersfield and the surrounding parts of the valley. The inset map is a zoomed out image of the southern part of San Joaquin Valley (SJV) with location of Kern County superimposed. The light blue lines mark the highways, WWTP stands for waste water treatment plant, and O&G stands for oil and gas fields. The location of the CalNex experiment site is marked by the “tower” symbol.

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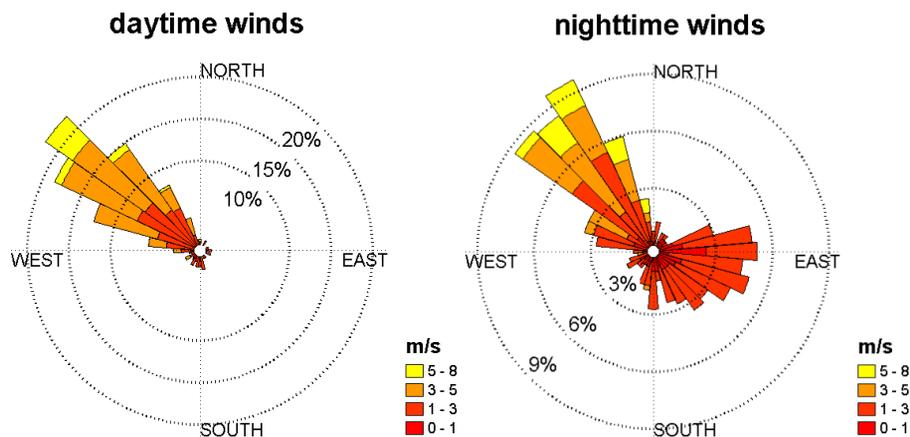


Figure 3. Wind rose plots showing mean wind direction measured at the site during (left) daytime (07:00–16:00 h), and (right) nighttime (17:00–06:00 h). The concentric circles represent the percentage of total observations; each colored pie represents a range of 10° while the colors denote different wind speed ranges.

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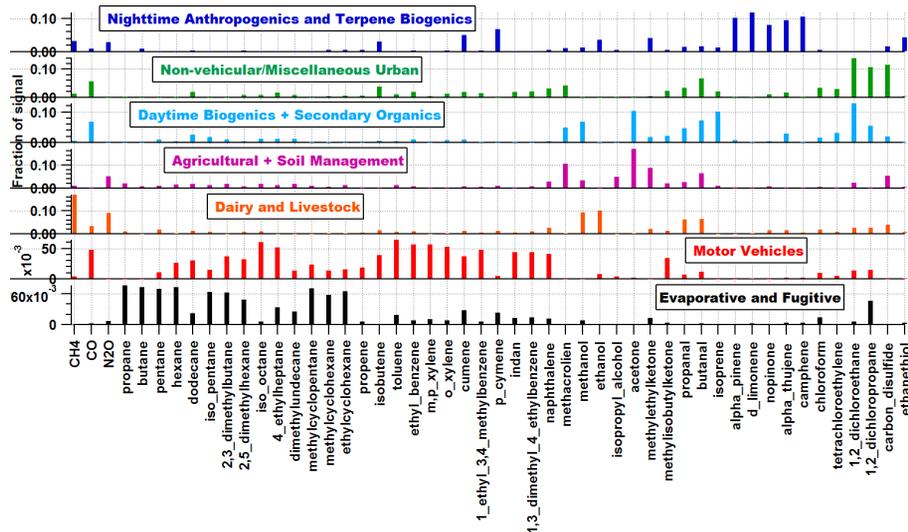


Figure 4. Source profile of the seven factors derived using PMF. The source factors are evaporative and fugitive, motor vehicles, dairy and livestock, agricultural + soil management, daytime biogenics + secondary organics, urban, and nighttime anthropogenics + terpene biogenics. The x axis represents the normalized fraction of mass in each source factor, while the y axis lists all the chemical species included in the PMF analysis.

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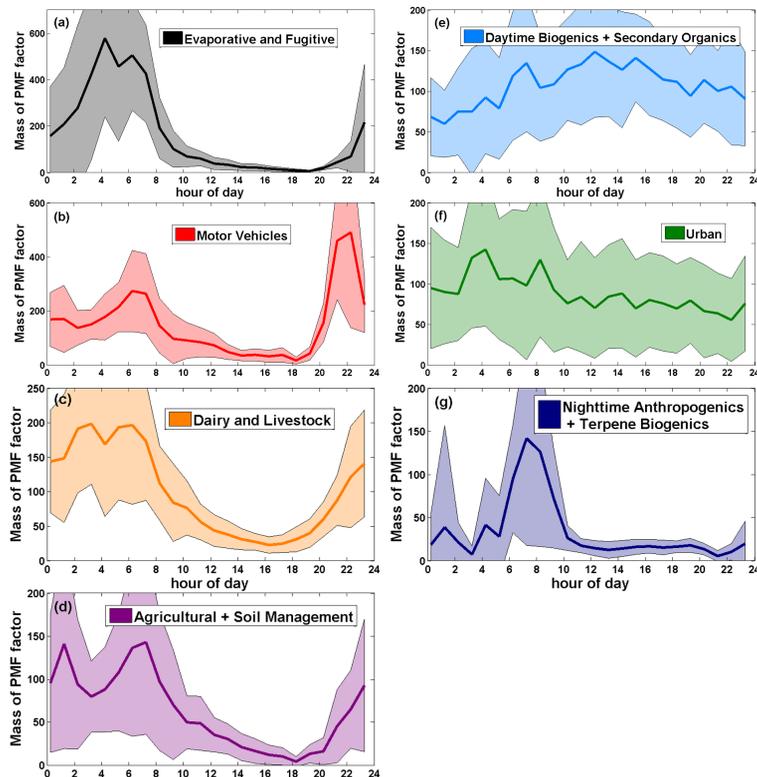


Figure 5. Mean hourly diurnal plots of PMF source factor concentration enhancements for (a) evaporative and fugitive, (b) motor vehicles, (c) dairy and livestock, (d) agricultural + soil management, (e) daytime biogenics and secondary organics, (f) non-vehicular/miscellaneous urban and (g) nighttime anthropogenics + terpene biogenics. The x axis represents sum of normalized mass concentrations from all tracers contributing to the factor. The y axis is hour of day (local time). The solid lines represent the mean and the shaded area represents the standard deviation (variability) at each hour.

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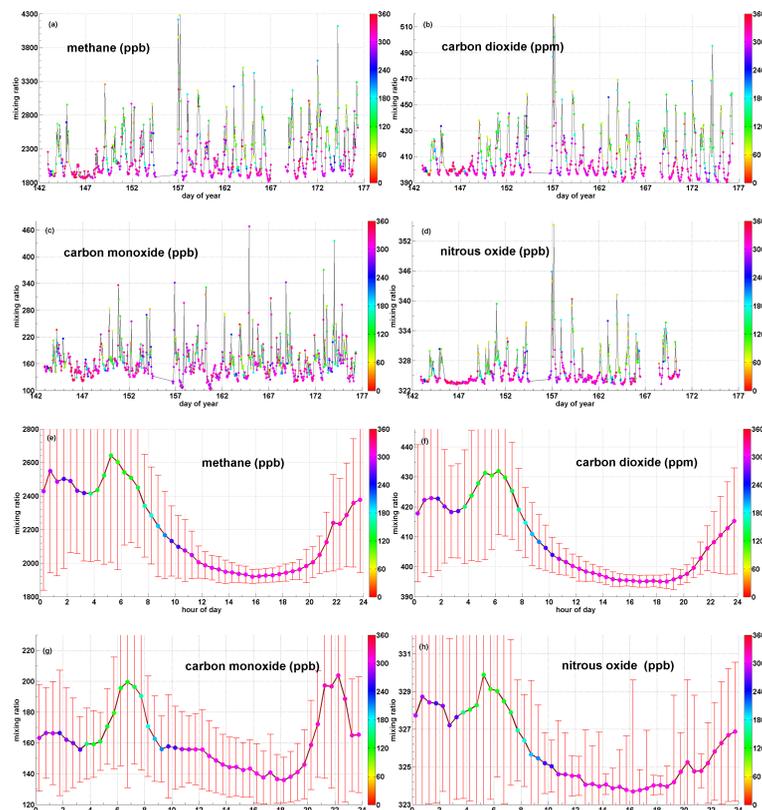


Figure 6. Time series of (a) CH₄, (b) CO₂, (c) CO, and (d) N₂O obtained from 30 min averages over the entire sampling period. The color bar indicates the average wind direction during each 30 min period. Mixing ratios plotted as average diurnal cycles for (e) CH₄, (f) CO₂, (g) CO and (h) N₂O along with wind direction. The curve and the red whiskers represent the mean and the standard deviations about the mean, respectively.

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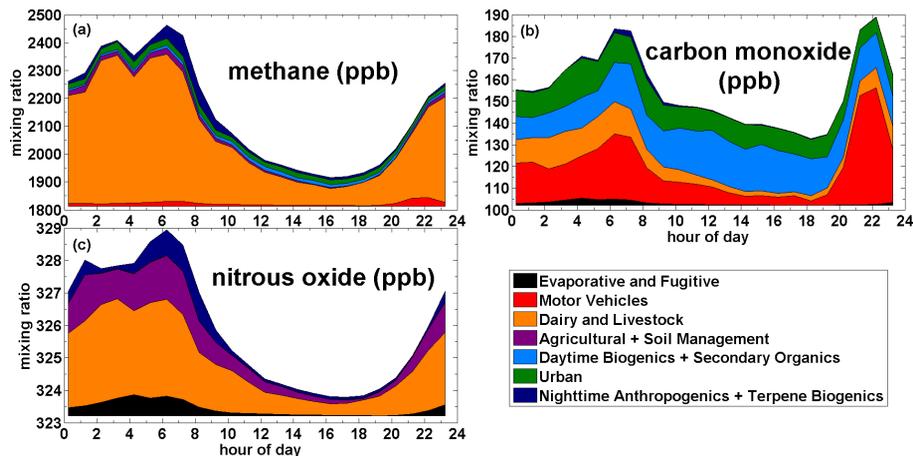


Figure 7. Diurnal plot of PMF derived (a) CH_4 , (b) CO , and (c) N_2O concentrations sorted by PMF source category. The legend on the bottom right shows the names of the PMF source factor which each color represents. The PMF derived enhancements from each source have been added to the background concentrations.

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