

# 1    **Production of peroxy nitrates in boreal biomass burning plumes over** 2    **Canada during the BORTAS campaign**

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4    **Marcella Busilacchio<sup>1</sup>, Piero Di Carlo<sup>1,2</sup>, Eleonora Aruffo<sup>1,2</sup>, Fabio Biancofiore<sup>1,2</sup>, Cesare Dari**  
5    **Salisburgo<sup>1</sup>, Franco Giammaria<sup>2</sup>, Stephane Bauguitte<sup>3</sup>, James Lee<sup>4</sup>, Sarah Moller<sup>4</sup>, James**  
6    **Hopkins<sup>4</sup>, Shalini Punjabi<sup>4</sup>, Stephen Andrews<sup>4</sup>, Alistair C. Lewis<sup>4</sup>, Mark Parrington<sup>5,\*</sup>, Paul I.**  
7    **Palmer<sup>5</sup>, Edward Hyer<sup>6</sup>, Glenn M. Wolfe<sup>7,8</sup>**

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9    [1] Center of Excellence CETEMPS, Universita' dell'Aquila, Via Vetoio, Coppito, L'Aquila, Italy,

10    [2] Department of Physical and Chemical Sciences, University of L'Aquila, Coppito L'Aquila, Italy,

11    [3] Facility for Airborne Atmospheric Measurements, Bedfordshire, UK,

12    [4] Department of Chemistry, University of York, York, UK,

13    [5] School of GeoSciences, University of Edinburgh, UK.

14    [6] Marine Meteorology Division, Naval Research Laboratory, Monterey, California, USA.

15    [7] Atmospheric Chemistry and Dynamics Laboratory, NASA Goddard Space Flight Center, Greenbelt,  
16    Maryland, USA

17    [8] Joint Center for Earth Systems Technology, University of Maryland Baltimore County, Baltimore, MD,  
18    USA

19    [\*] now at European Centre for Medium-Range Weather Forecasts (ECMWF), Reading, UK.

20  
21    Correspondence to: P. Di Carlo (piero.dicarlo@aquila.infn.it)

## 22 23 24    **Abstract**

25  
26    The observations collected during the BORTAS campaign in summer 2011 over Canada are  
27    analysed to study the impact of forest fire emissions on the formation of ozone (O<sub>3</sub>) and total  
28    peroxy nitrates ( $\Sigma$ PNs,  $\Sigma$ ROONO<sub>2</sub>). The suite of measurements onboard the BAe-146 aircraft,  
29    deployed in this campaign, allows us to calculate the production of O<sub>3</sub> and of  $\Sigma$ PNs, a long lived

1 NO<sub>x</sub> reservoir whose concentration is supposed to be impacted by biomass burning emissions. In  
2 fire plumes, profiles of carbon monoxide (CO), which is a well-established tracer of pyrogenic  
3 emission, show concentration enhancements that are in strong correspondence with a significant  
4 increase of ΣPNs concentrations, whereas minimal increase of the concentrations of O<sub>3</sub> and NO<sub>2</sub> are  
5 observed. The ΣPNs and O<sub>3</sub> productions have been calculated using the rate constants of the first  
6 and second order reactions of Volatile Organic Compounds (VOCs) oxidation. The ΣPNs and O<sub>3</sub>  
7 productions have also been quantified by 0-D model simulation based on the Master Chemical  
8 Mechanism. Both methods show that in fire plumes the average production of ΣPNs and O<sub>3</sub> are  
9 greater than in the background plumes, but the increase of ΣPNs production is more pronounced  
10 than the O<sub>3</sub> production. The average ΣPNs production in fires plumes is from 7 to 12 times greater  
11 than in the background, whereas the average O<sub>3</sub> production in fires plumes is from 2 to 5 times  
12 greater than in the background. These results suggest that, at least for boreal forest fires and for the  
13 measurements recorded during the BORTAS campaign, fire emissions impact both the oxidized  
14 NO<sub>y</sub> and O<sub>3</sub>, but: 1) ΣPNs production is amplified significantly more than O<sub>3</sub> production and 2) in  
15 the forest fire plumes the ratio between the O<sub>3</sub> production and the ΣPNs production is lower than the  
16 ratio evaluated in the background air masses, thus confirming that the role played by the ΣPNs  
17 produced during biomass burning is significant in the O<sub>3</sub> budget. These observations are consistent  
18 with elevated production of PAN and concurrent low production (or sometimes loss) of O<sub>3</sub> observed  
19 in some another campaigns (i.e. ARCTAS-B) focused on forest fire emissions. Moreover our  
20 observations extend ARCTAS-B results since PAN is one of the compounds included in the ΣPNs  
21 family detected during BORTAS. The implication of these observations is that fire emissions in  
22 some cases, for example Boreal forest fires and in the conditions reported here, may influence more  
23 long lived precursors of O<sub>3</sub> than short lived pollutants, which in turn can be transported and  
24 eventually diluted in a wide area. These observations provide additional indirect evidence that O<sub>3</sub>  
25 production may be enhanced as plumes from forest fires age.

26

## 1. Introduction

Biomass burning emissions are an important atmospheric source of fine carbonaceous particles, trace gases and aerosols that significantly affect the chemical composition of the atmosphere and the radiation balance of the Earth-atmosphere system (Crutzen et al., 1979; Crutzen and Andreae, 1990; Andreae and Merlet, 2001; Bond et al., 2004; Langmann et al., 2009; Bowman et al., 2009).

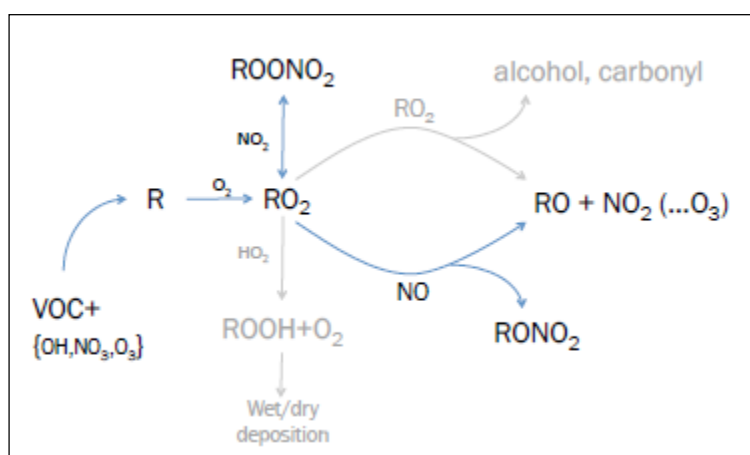
Biomass burning generates large quantities of carbon monoxide (CO), nitrogen oxides ( $\text{NO}_x = \text{NO} + \text{NO}_2$ ) and VOCs which are the major precursors involved in the photochemical production of tropospheric ozone ( $\text{O}_3$ ) (Goode et al., 2000, Chan et al., 2003). Moreover, biomass burning emissions include some greenhouse gases ( $\text{CO}_2$ ,  $\text{CH}_4$ ,  $\text{N}_2\text{O}$ ) that alter the climate and air quality (Langmann et al., 2009; Lapina et al., 2006; Simpson et al., 2006). Quantification of the influence of boreal forest fires on the Earth-atmosphere system and on the climate has become one of the key topics for the scientific community.

Forest fires in the boreal regions of Siberia, Canada and Alaska peak during the period from May to October (Lavoue et al., 2000). Some studies highlight the increase in the number of boreal forest fires and the total forested area burned over Canada during the past three decades, corresponding to increasing temperatures and reduced moisture in this area (Gillett et al., 2004; Rinsland 2007; Marlon 2008). Wotton et al. (2010) estimate an increase of 30% in boreal forest fire occurrence by 2030, causing a possible growth of 30% in the emission of  $\text{CO}_2$  and other greenhouse gases (Amiro et al., 2009). The effects of boreal biomass burning emissions on the  $\text{O}_3$  concentration has been investigated by several authors with some studies showing situations where  $\text{O}_3$  concentrations increase and others where it was unaffected (e.g., Wofsy et al., 1992; Jacob et al., 1992; Mauzerall et al., 1996; Wotawa and Trainer, 2000; Val Martin et al., 2006; Real et al., 2007; Leung et al., 2007, Jaffe and Wigder, 2012, Parrington et al., 2012 ). The analysis of the ARCTAS-B (NASA Arctic Research of the Composition of the Troposphere from Aircraft and Satellites) aircraft measurements of biomass burning plumes in central Canada in the spring and summer of 2008 showed consistent production of peroxyacetyl nitrate (PAN), with little evidence for  $\text{O}_3$  formation and, in some

1 plumes, the O<sub>3</sub> mixing ratios measured within boreal biomass burning plumes were  
 2 indistinguishable from measurements outside of the plumes (Alvarado et al., 2010). The production  
 3 of ozone  $P(O_3)$  measured in boreal fire plumes has been reported to be a function of the plume age  
 4 (Parrington et al., 2013), but with mixed, non-conclusive results. For example, boreal fire plumes  
 5 transported over the Azores and measured between 1 and 2 weeks after emission showed an O<sub>3</sub>  
 6 increase between 40% and 90% (Val Martin et al., 2006; Pfister et al., 2006). On the other hand,  
 7 observations over Siberia in 2006 of aged boreal fire plumes (up to a week) showed some plumes  
 8 with O<sub>3</sub> enhanced and others with O<sub>3</sub> depletion; on average, the O<sub>3</sub> in the fire plumes was not  
 9 significantly different from that in the background atmosphere (Verma et al., 2009). In earlier  
 10 studies of relatively fresh plumes (1-2 days), O<sub>3</sub> was reported to be enhanced in one third of the  
 11 boreal fire plumes with concentrations in the remaining plumes being unaffected (Wofsy et al.  
 12 1992; Mauzerall et al. 1996).

13 In the atmosphere, volatile organic compounds (VOCs) are oxidized by OH, NO<sub>3</sub> or O<sub>3</sub> producing  
 14 an alkyl radical R that rapidly reacts with molecular oxygen O<sub>2</sub> to form peroxy radicals (HO<sub>2</sub>, RO<sub>2</sub>)  
 15 (reaction R1). The RO<sub>2</sub>, then, can proceed in different ways: 1) reacting with NO and producing a  
 16 molecule of alkyl nitrate (ΣANs, ΣRONO<sub>2</sub>) (R2) or an alkoxy radical RO (R4) or 2) reacting with  
 17 NO<sub>2</sub> and producing peroxy nitrates (ΣPNs, ΣROONO<sub>2</sub>) (R3). Reactions (R4) and (R3) have  
 18 opposite effects on the O<sub>3</sub> budget, propagating or terminating radical cycles, respectively. Thus,  
 19 peroxy nitrate formation competes with the O<sub>3</sub> production resulting from reactions (R4)-(R8). Alkyl  
 20 nitrate formation via (R2) can also affect the O<sub>3</sub> budget. The reaction cycles that are of interest  
 21 when considering Nitrogen oxides (NO<sub>x</sub>) and odd-hydrogen radicals (HO<sub>x</sub>) (R1-R8) are illustrated  
 22 schematically in Figure 1 and listed below:





**Figure 1.** A schematic of the atmospheric chemical system (Atkinson and Arey, 2003, Palmer et al.2013).

In July and August 2011, the BOREal forest fires on Tropospheric oxidants over the Atlantic using Aircraft and Satellites (BORTAS) measurement campaign was carried out in order to quantify the impact of boreal biomass burning on the composition and distribution of tropospheric oxidants. The BORTAS project involved several international institutions with the support of the UK Facility for Airborne Atmospheric Measurements (FAAM). The instruments were installed on board the FAAM BAe146 research aircraft and the campaign was based at Halifax airport (Nova Scotia, Canada). During the campaign, fifteen flights were carried out (nominally referenced as flights B618 to B632) in Eastern Canada that were planned to maximize the probability of sampling air masses produced from forest fires in Canada (Ontario) or the USA. More detailed information about the

1 BORTAS campaign objectives and preliminary results are presented by Palmer et al. (2013). The  
2 primary aim of this study is to evaluate and understand the impact of the boreal fire emissions  
3 during the BORTAS campaign on the formation of  $O_3$  and  $\Sigma PNs$  within biomass burning plumes  
4 and, in particular, to estimate the balance between the production of ozone  $P(O_3)$  and the  
5 production of total peroxy nitrates  $P(\Sigma PNs)$  in this specific environment.

6

## 7 **2. Instrumental**

8 A comprehensive description of the BORTAS experiment and of the overall instrumentations  
9 involved can be found in Palmer et al. (2013). Measurements included in this analysis are  
10 summarized in Table 1.  $NO_2$ ,  $\Sigma PNs$  and  $\Sigma ANs$  were measured using the TD-LIF (Thermal  
11 Dissociation – Laser Induced Fluorescence) instrument developed at the University of L'Aquila  
12 (Italy) (Dari-Salisburgo et al., 2009; Di Carlo et al., 2013). Briefly, this technique permits direct  
13 measurement of  $NO_2$  molecules excited by laser radiation. The  $\Sigma PNs$  and  $\Sigma ANs$  are measured after  
14 thermal-dissociation into  $NO_2$  by heating the air sample at 200°C and 400°C, respectively (Day et  
15 al., 2002; Di Carlo et al., 2013). Nault et al. (2015) found that methyl peroxy nitrate ( $CH_3O_2NO_2$ ),  
16 which can be abundant in particular conditions (very low temperature, below 240K, typical of the  
17 high atmosphere), may contribute interference to high altitude  $NO_2$  measurements resulting from  
18 thermal decomposition occurring in the sample intake system. This interference is a function of the  
19 intake system temperature and increases from 280 K in which the interference is negligible up to  
20 300 K in which it can be on the order of 10%. During all the BORTAS flights analysed in this  
21 paper, the cabin temperature has been kept at about 280 K and, as a consequence, the impact on the  
22  $NO_2$  of the  $CH_3O_2NO_2$  dissociation is negligible. Moreover, this species is not expected to be  
23 significant in our study, since the ambient temperatures of the air masses sampled during the period  
24 in analysis range between 250 K and 280 K and the  $CH_3O_2NO_2$  concentration is significant only for  
25 temperatures lower than 240 K. The measurements of  $O_3$  were carried out with an UV absorption

system Model 49C (Thermo environmental Corp.) (Wilson and Birks, 2006). CO was measured using a VUV resonance/fluorescence system (Gerbig et al. (1999). A chemiluminescence instrument equipped with a photolytic converter was also used to measure NO and NO<sub>2</sub> (Lee et al. 2009; Reidmiller et al. 2010). VOC concentrations were measured by the University of York using a WAS (Whole Air Sampling) system coupled to an offline GC-FID (Gas Chromatography with Flame Ionization Detector) (Hopkins et al. 2003; Purvis et al. 2013) and by the University of East Anglia using a PTR-MS (Murphy et al. 2010).

**Table 1.** Observed compounds and instruments on board the BAe-146 aircraft during BORTAS campaign, used in the analysis in this paper. A complete list of the instruments with accuracy and detection limit, is reported in Palmer et al. (2013).

| Species   | Method                     | Reference   |
|---|----------------------------|---|
| CO  | VUV resonance/fluorescence | Gerbig et al. (1999)                                  |
| O <sub>3</sub>  | UV absorption              | Wilson and Birks (2006)                               |
| NO <sub>2</sub> , ΣRO <sub>2</sub> NO <sub>2</sub> , ΣRONO <sub>2</sub> , NO <sub>y</sub>   | TD-LIF                     | Dari-Salisburgo et al. (2008); Di Carlo et al. (2013) |
| C <sub>5</sub> –C <sub>12</sub> VOCs  | GC-MS                      | Purvis et al. (2013)                                  |
| C <sub>2</sub> –C <sub>7</sub> NMHCs, acetone<br>CH <sub>3</sub> OH   | WAS-GC-FID                 | Hopkins et al. (2003)                                 |
| CH <sub>3</sub> CN, C <sub>3</sub> H <sub>6</sub> O, C <sub>5</sub> H <sub>8</sub> ,<br>MVK+MACR, C <sub>4</sub> H <sub>8</sub> O,<br>C <sub>6</sub> H <sub>6</sub> , C <sub>7</sub> H <sub>8</sub> , C <sub>10</sub> H <sub>16</sub> | PTR-MS                     | Murphy et al. (2010)                                  |

### 3. Data analysis

#### 3.1 Geographical location and meteorological situation

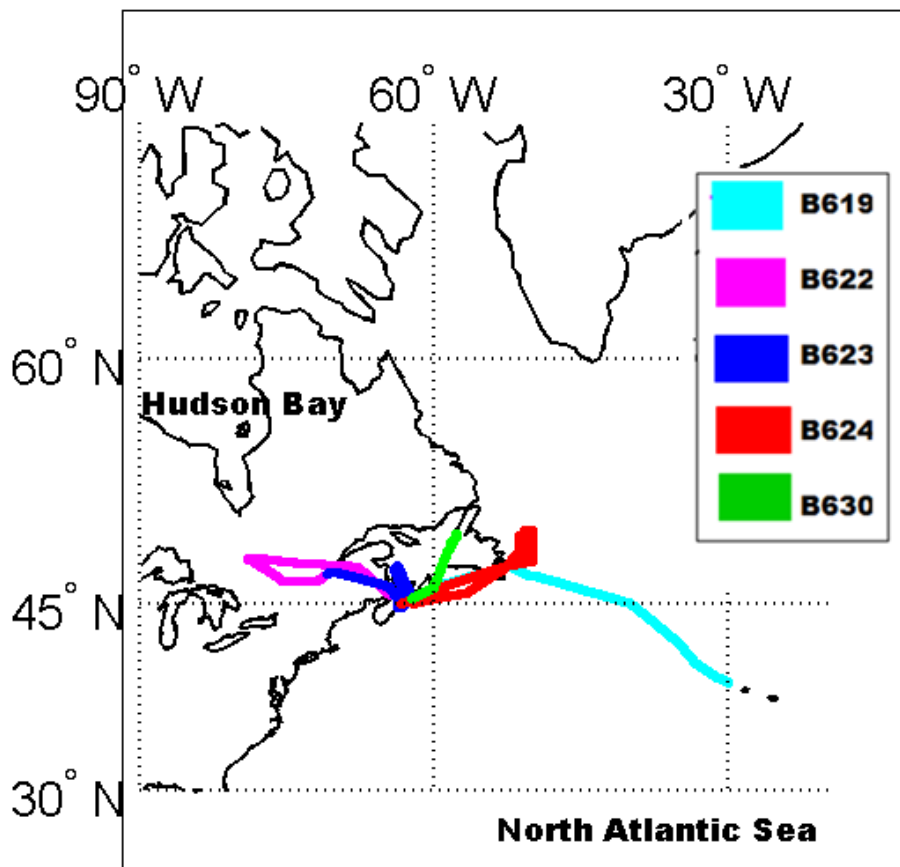
Fig. 2 shows the geographic coverage of the five flights selected for our analysis. The flights were carried out between 12<sup>th</sup> July and 3<sup>rd</sup> August 2011 over Canada and, in particular, above the North

1 Atlantic Ocean, Nova Scotia, Maine and Québec. The altitude during the flights exceeded a typical  
 2 planetary boundary layer depth of 2000 m a.s.l. so that local emissions do not affect the  
 3 measurements, especially those carried out in the fire plumes. Table 2 summarizes some specific  
 4 features of each flight BORTAS selected in this analysis and provides a brief description of the  
 5 meteorology associated with them. Other details about the BORTAS flights can be found in Palmer  
 6 et al. (2013). From these descriptions, it can be seen that the synoptic situation of the fire plume  
 7 flights are similar to those of background flights.

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12 **Figure 2.** FAAM146 flight tracks during July 2011. The different colours are the tracks of each  
 13 different flight: during the B623 and B624 fire plumes were observed, during B619 and B630



1 background air was measured, whereas in the B622 flight fire plume and background air were  
2 detected. See Table 2 for details of individual flights  
3

1 **Table 2.** Synoptic meteorology and weather associated with the five BORTAS flights considered in this analysis.  
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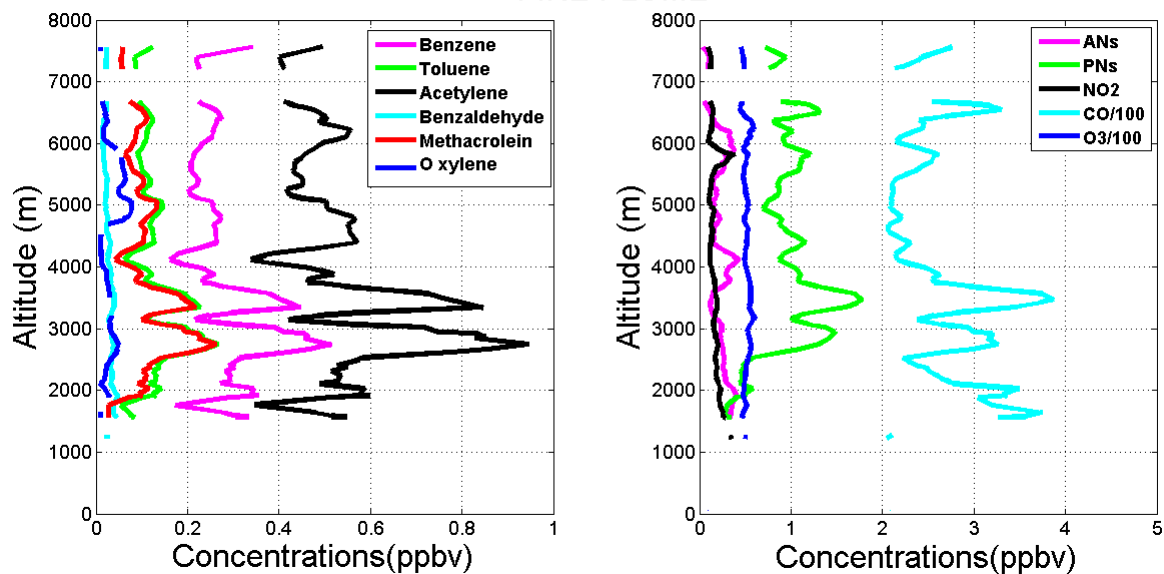
| Flight                                    | Date       | Trajectory                          | Flight Altitude<br>(Max-Min-Mean) | Synoptic meteorology  |
|---|------------|-------------------------------------|-----------------------------------|---|
| B619<br>(background)                      | 13/07/2011 | St John's-<br>Halifax               | 7257<br>100<br>4594               | Skies mostly cloudy.  |
| B622<br>(fire plume<br>and<br>background) | 20/07/2011 | Halifax –<br>Québec City            | 7575<br>1892<br>4699              | Low from surface to 500 hPa S Ungava Bay. Surface low and frontal wave moving E from mouth of St Lawrence. Flight in “warm” sector – Mainly clear to 21:00 then cloudy.   |
| B623<br>(fire plume)                      | 20/07/2011 | Québec City–<br>Halifax             | 6173<br>1888<br>4451              | Low from surface to 500 hPa S Ungava Bay. Surface low and frontal wave just N of Anticosti Island and cold front west. Showers and thundershowers along and in advance of front. Aircraft may have encountered showers over Prince Edward Island (PEI). |
| B624<br>(fire plume)                      | 21/07/2011 | Halifax - St<br>John's –<br>Halifax | 2826<br>1743<br>2069              | Low from surface to 500 hPa over extreme N Labrador. Cold front from NB to S of NF (Newfoundland). Weak low crossing NB late day. Cloud moved into flight zone from the west. Precipitation for return flight from S of NF to Halifax.                  |
| B630<br>(background)                      | 31/07/2011 | Halifax,<br>Nova Scotia             | 7616<br>5076<br>6704              | Trough from surface through to 500 hPa along Labrador coast to low centre off south coast NF. Cooler air mass over region. Weak ridge building over NB to W Labrador. Generally clear skies for flight route and level.                                 |

### 3.2 Identification of the plumes: vertical profiles and back trajectories

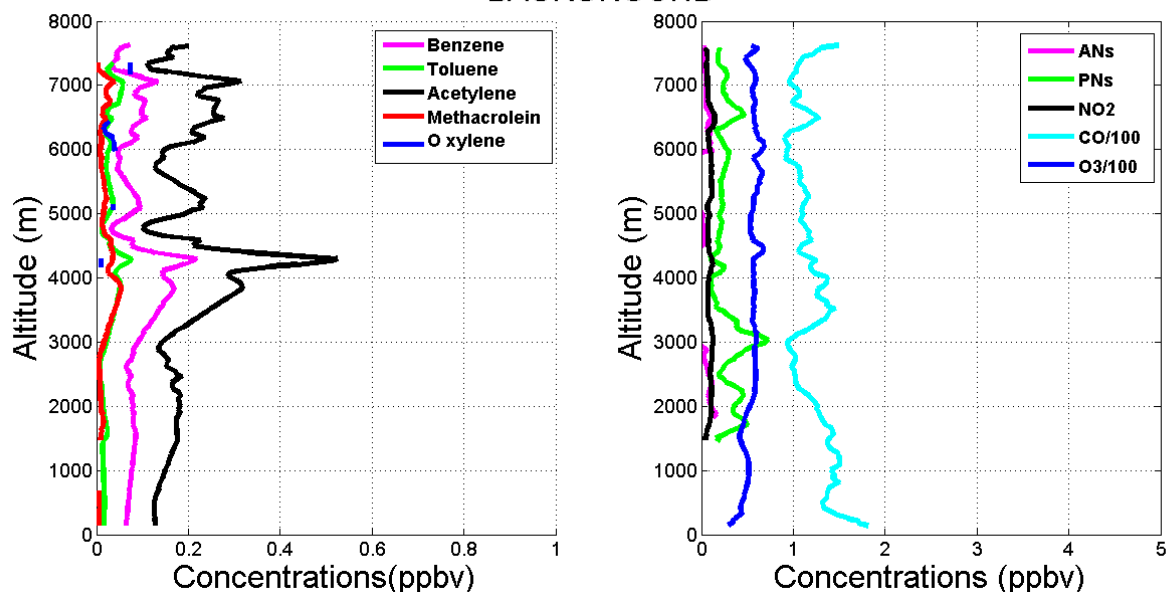
CO is a product of incomplete combustion (Crutzen et al., 1979; Andreae and Merlet, 2001; Lewis et al., 2013), therefore it is one of the tracers used to classify the plumes emitted by boreal fires. However, it is necessary to discriminate between anthropogenic and biomass burning CO emissions; for this purpose, following Lewis et al. (2013), we defined a CO threshold of 200 ppbv and we verified at the same time the presence of other pyrogenics such as furfural or camphor to confirm the fire origin of the plume. In conclusion, we classify the air masses in three classes: 1) those sampled within boreal biomass burning plumes ( $\text{CO} \geq 200$  ppbv with significant presence of other pyrogenics species such as furfural or camphor (Andreae and Merlet, 2001); 2) those impacted by anthropogenic emissions ( $\text{CO} \geq 200$  ppb without the presence of furfural or camphor) and 3) those sampled in background conditions ( $\text{CO} < 200$  ppb). Using the above criteria to distinguish between flights where we sampled fire plumes and those when we sampled background air we analysed the vertical profiles of species known to have a significant biomass burning source, such as  $\text{NO}_2$ ,  $\Sigma\text{PNs}$ ,  $\Sigma\text{ANs}$ ,  $\text{CO}$ ,  $\text{O}_3$  and some  $\text{VOCs}$  (i.e., propene, methacrolein, acetylene, benzene, ethyl-benzene, toluene, o-xylene, benzaldehyde, furfural and camphor). The CO and pyrogenic species analysis allows us to select five flights in which we distinguish between those where we sampled boreal fire emissions (B622, B623 and B624 – labelled henceforth “plume” flights) and those in which we measured background air (B619, B622 and B630 – labelled henceforth “background” flights). Flight B622 is a particular case in which both conditions are met, and we split this flight into two different parts: plume and background. Figure 3 shows profiles of the species indicated above as a function of the altitude for the plume flights (upper panels) and for the background flights (lower panels). It is possible to observe in Fig. 3 that the vertical structures are different in the two conditions. In the upper panels (plume flights) the concentrations of some species, especially  $\text{CO}$ ,  $\Sigma\text{PNs}$ , Acetylene and Benzene, show significant and concomitant increases at 3500 m above sea level (a.s.l.) and 6000 m a.s.l.. Moreover, in the plume measurements at 2000 m a.s.l. a large increase in the CO levels is measured concurrent with an increase in the  $\Sigma\text{PNs}$

1 smaller than at the other altitudes. This suggests that the conditions of the air masses at 2000 m  
 2 a.s.l. are more complex and that it potentially has various origins, i.e., impacted both by  
 3 anthropogenic and boreal biomass burning emissions. The  $\Sigma ANs$  concentrations are lower than the  
 4  $\Sigma PNs$  and do not show significant structures. The  $O_3$  profile shows little variability between 1000  
 5 and 7000 m of altitude with no concentration changes that coincide with variations in CO. In the  
 6 background flights, as expected, the concentrations of the species analysed do not show strong  
 7 vertical structures such as in the plume flights, with the exception of VOCs that show a peak at  
 8 about 4 Km

### FIRE PLUME

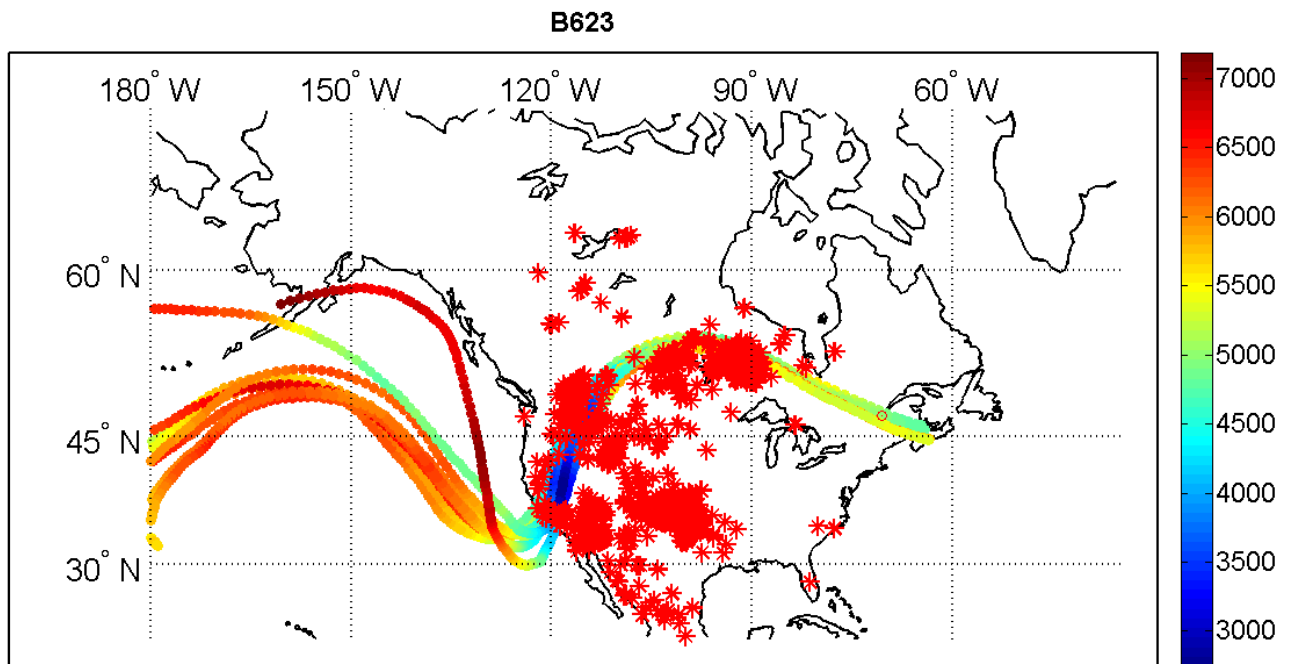
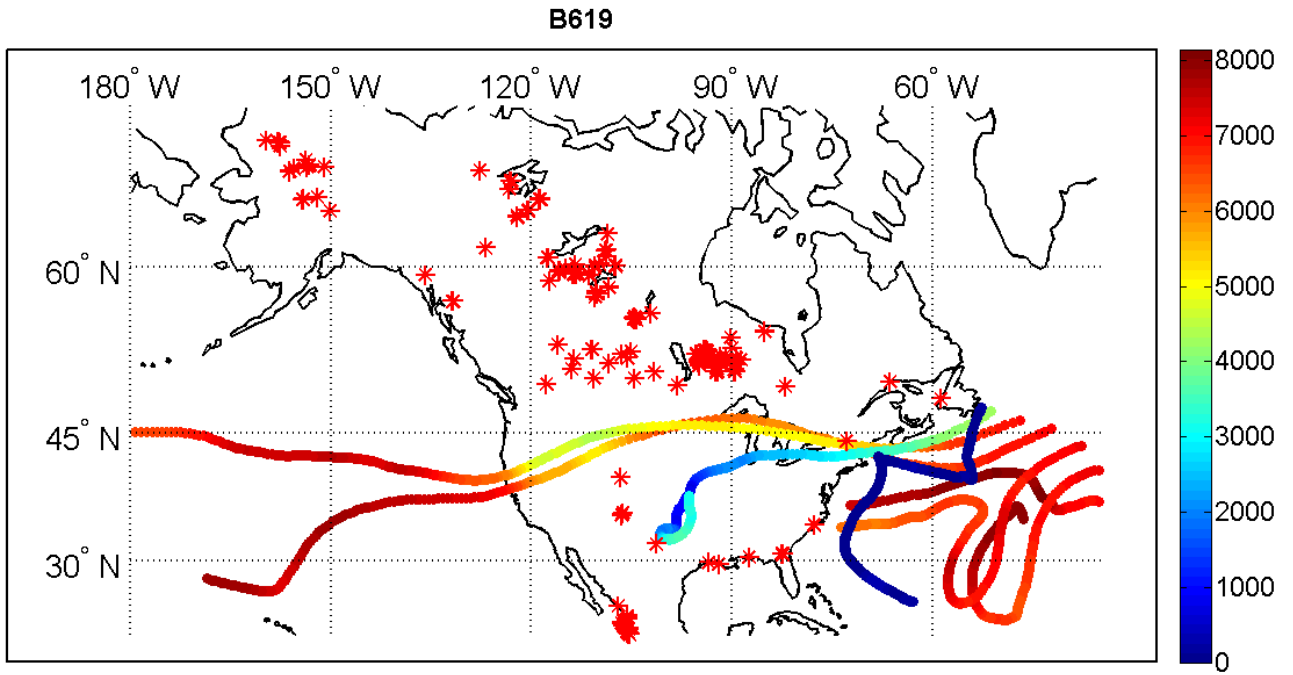


### BACKGROUND



**Figure 3.** Vertical profiles of Benzene, Toluene, Acetylene, Methacrolein O-Xylene (panels on the left) and  $\Sigma$ ANs,  $\Sigma$ PNs,  $\text{NO}_2$ , CO and  $\text{O}_3$  (panels on the right) concentrations averaged for the plume flights (upper panels: B622, B623, B624 flights) and the background flights (lower panels: B619, B622, B630 flights).

To facilitate the determination of the sources of the biomass burning plumes (Tereszchuk et al., 2011; Parrington et al., 2012), we calculated Lagrangian back trajectories using the Hysplit model (Draxler et al., 2003) to verify the origin of the air masses. The Fire Locating And Monitoring of Burning Emissions (FLAMBE) archive provides fires emissions data from 2000 to the present worldwide (Reid et al., 2009) incorporating active fire detection data from geostationary and polar-orbiting satellites. To locate the sources of the boreal biomass burning plumes measured during the BORTAS campaign, the FLAMBE inventory data have been used in conjunction with the Hysplit back trajectories. In Fig. 4, 8 day back-trajectories are evaluated starting from points along the flight track and the corresponding fires (red asterisks) from the FLAMBE archive are shown for the plume flight B619 (upper panel) and for the background flight B623 (lower panel). The same analysis has been done for all the flights of the campaign, although here we report only the results of flights B619 and B623 since they are representative of all the other flights. Parrington et al. (2013) evaluated the photochemical age of the air masses for each flight using the ratio of  $\log(n\text{-butane/ethane})$  and assuming an  $\text{OH}$  concentration of  $2 \times 10^6$  molecules/ $\text{cm}^3$ . They found that the age calculated for the air masses sampled within the boreal biomass burning emissions ranges between 1 and 5 days and the background air is older than 6 days.



**Figure 4.** Location of the boreal biomass burning activity during the BORTAS campaign recorded by the FLAMBE inventory (red asterisks) and air mass backward trajectory analysis starting from location along the flight trajectories. The flight B623 (lower panel) sampled multiple fire plumes, whereas the flight B619 (upper panel) was representative of background conditions.

1 Their results are in agreement with the back-trajectories analysis, confirming that the air masses  
2 sampled during the plume flights crossed biomass fires during the previous 8 days and, conversely,  
3 the background air masses do not overlap fires up to 8 days before. In addition, Griffin et al.(2013)  
4 investigates boreal fire plumes during the BORTAS campaign using back trajectories calculated by  
5 the Canadian Meteorological Centre (CMC) and shows that the boreal fire plume originated from  
6 forest fires is approximately 1.5 days old, which is in agreement with the age calculated for the air  
7 masses sampled within the boreal biomass burning emissions.

### 8 9 **3.3 Chemical signatures of plumes**

10 In Figure 5 the time series of NO<sub>2</sub>, ΣPNs, ΣANs, O<sub>3</sub>, CO and furfural (when measured) for the B619  
11 flight (top panel) and the B630 flight (bottom panel) are shown. During these background flights,  
12 the concentrations of all the species measured remain quite stable. The ΣPNs concentrations are  
13 significantly greater than the ΣANs but lower compared to those measured in the plume flights (less  
14 than 0.5 ppb). Moreover, ΣPNs do not show the significant structure that is shown in the O<sub>3</sub>  
15 measurements. CO is substantially lower than the 200 ppb threshold with the exception of one peak  
16 measured during B619 during a period spent in the airport for refueling (at ground level) where the  
17 CO level is affected by anthropogenic emissions and increases, reaching a maximum of about 300  
18 ppb during take off.

19 The B622 flight shows two regimes, as indicated by the CO concentrations and by the furfural  
20 measurements. In the first part of the flight (between 2000 m a.s.l. and 4000 m a.s.l., highlighted by  
21 a grey box in Fig.5b) the CO levels (cyan line) exceed 150 ppb and the furfural (yellow line) shows  
22 three big plumes (up to 1.2 ppb) in which the ΣPNs also increase (reaching the maximum value of  
23 3.5 ppb). On the other hand, in the second part of the flight the CO and ΣPNs decrease and the  
24 furfural is below the detection limit indicating that the air sampled is not affected by biomass  
25 burning. It is interesting to observe that O<sub>3</sub> and NO<sub>2</sub> concentrations are quite stable flying within or  
26 outside of the fire plume. Flight B623 (Fig. 5c) represents a case in which the air masses sampled

1 for most of the flight were impacted by biomass burning emissions and the remaining air masses  
2 show influence from human activities. In fact, CO levels are also always greater than 200 ppb and  
3 the furfural is below the detection limit during the whole flight, indicating an anthropogenic origin  
4 of the air masses. The fire plumes (highlighted by grey boxes) are characterized by sharp increases  
5 in the CO concentrations (maximum value of 552 ppb) and in the  $\Sigma$ PNs concentrations (maximum  
6 value of 1.5 ppb) measured while flying at constant altitude of about 4000 m a.s.l.. In the final part  
7 of flight B623 (between 00:26 and 01:00 UTC) a vertical spiral was carried out flying from 2000 m  
8 a.s.l. up to 8000 m a.s.l.. In this leg, plumes originating from different fires (identified analyzing the  
9 Hysplit back trajectories) were sampled. At about 4000 m a.s.l., back trajectories showed that the air  
10 masses sampled had the same origin of the fires plumes sampled at the same altitude in the first part  
11 of the flight. Both plumes were characterized by high levels of  $\Sigma$ PNs (up to 1.7 ppb). At the top of  
12 the spiral (8000 m a.s.l.), an aged plume was encountered with low  $\Sigma$ PNs and O<sub>3</sub> concentrations  
13 quite high (about 60 ppb). This high O<sub>3</sub> concentration represents the highest value measured during  
14 the whole flight.

15 According to the back-trajectories, this air mass originated from fires in the Western States of the  
16 U.S.A. (Oregon, Montana, Washington, Idaho, California, Nevada).

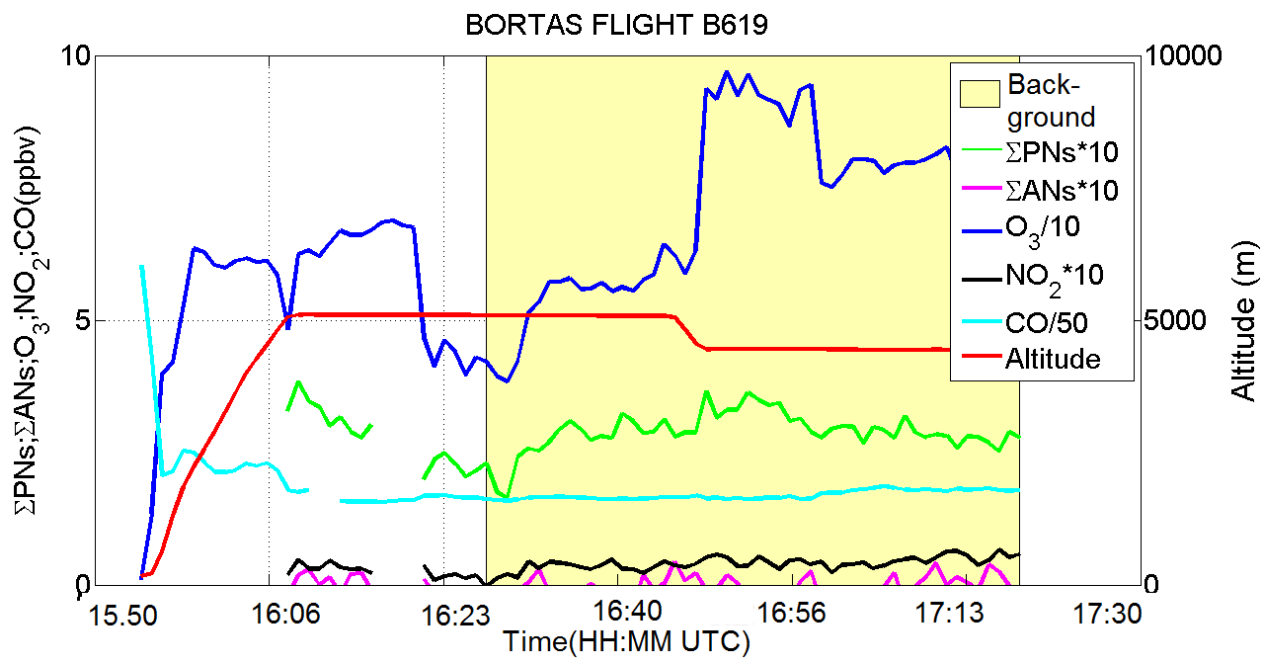
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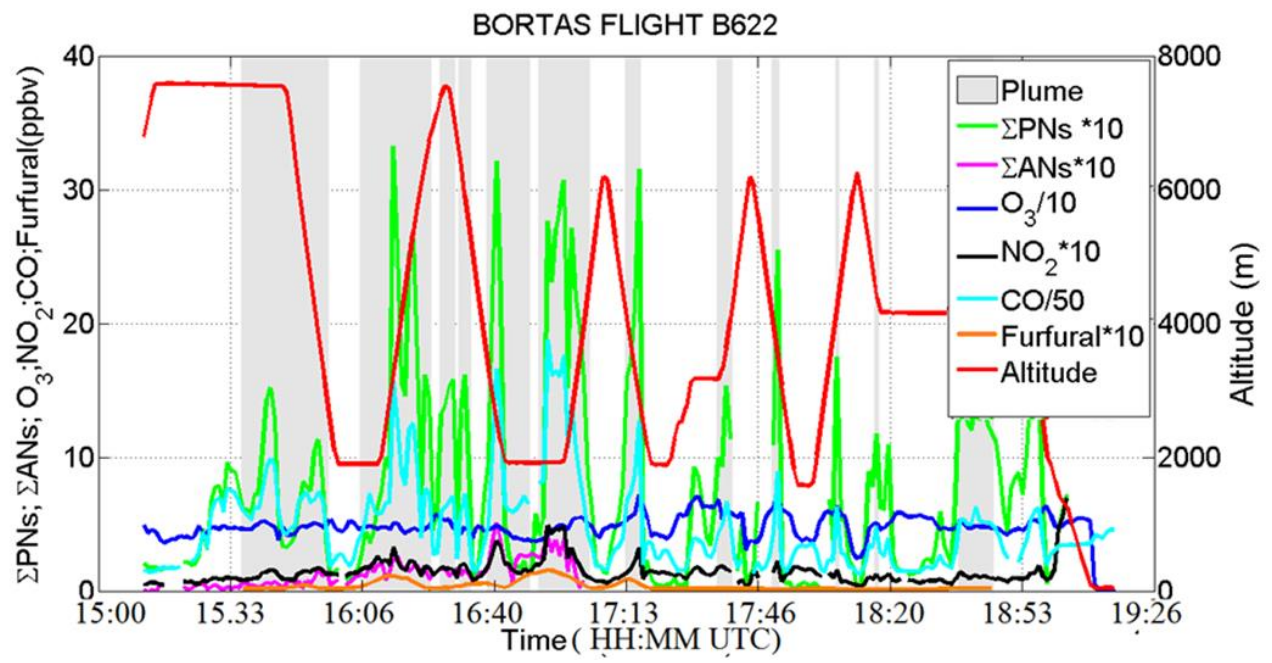


a)

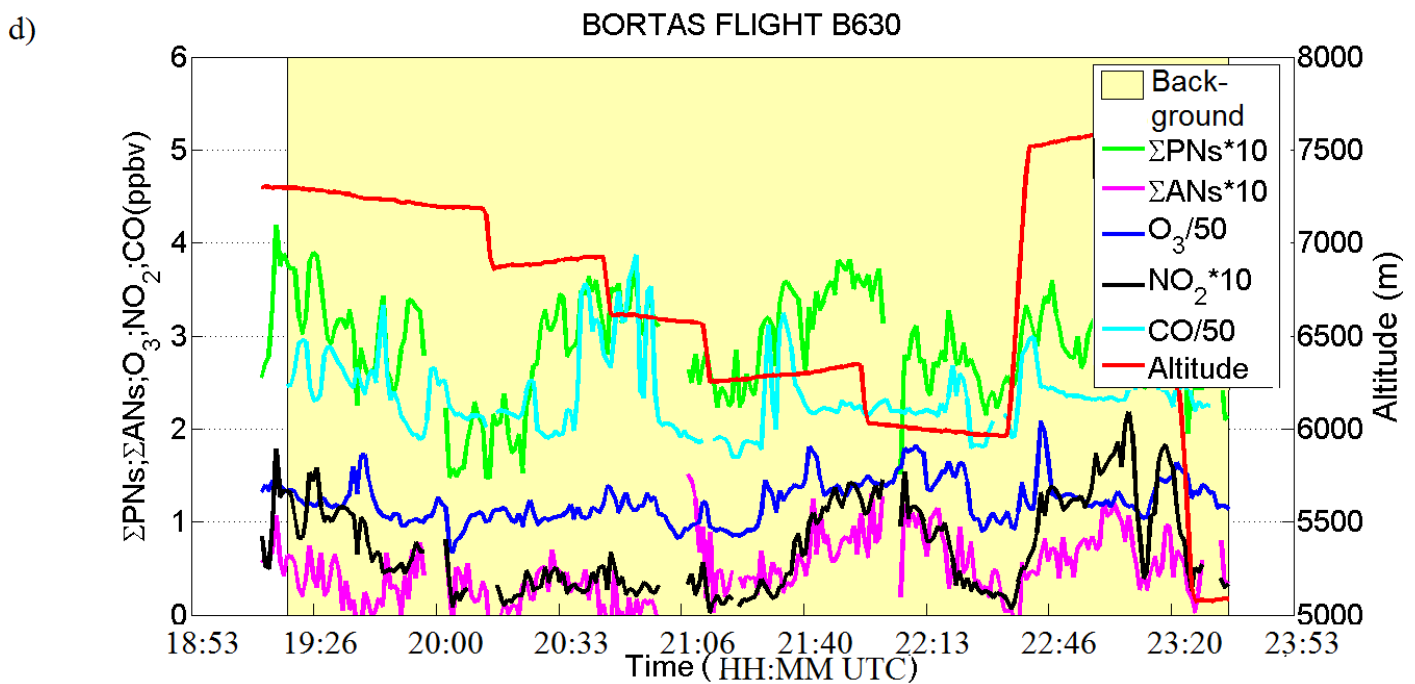
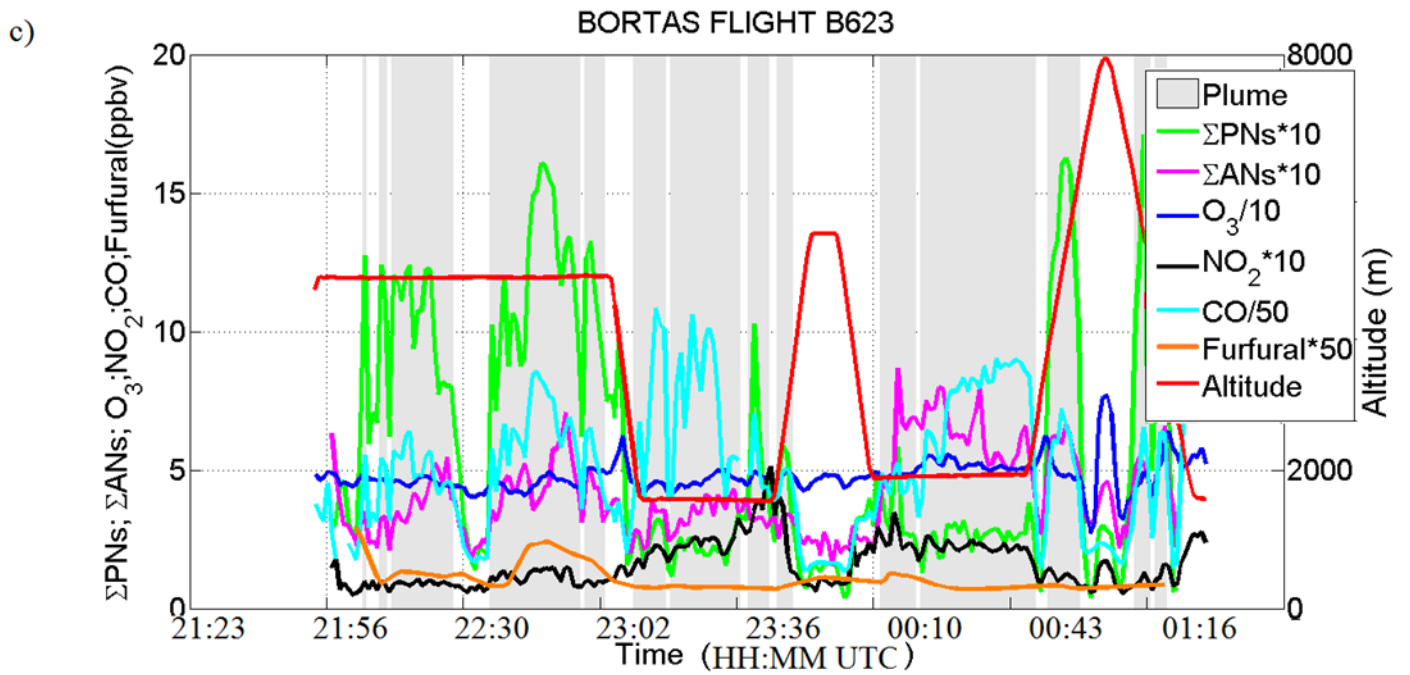


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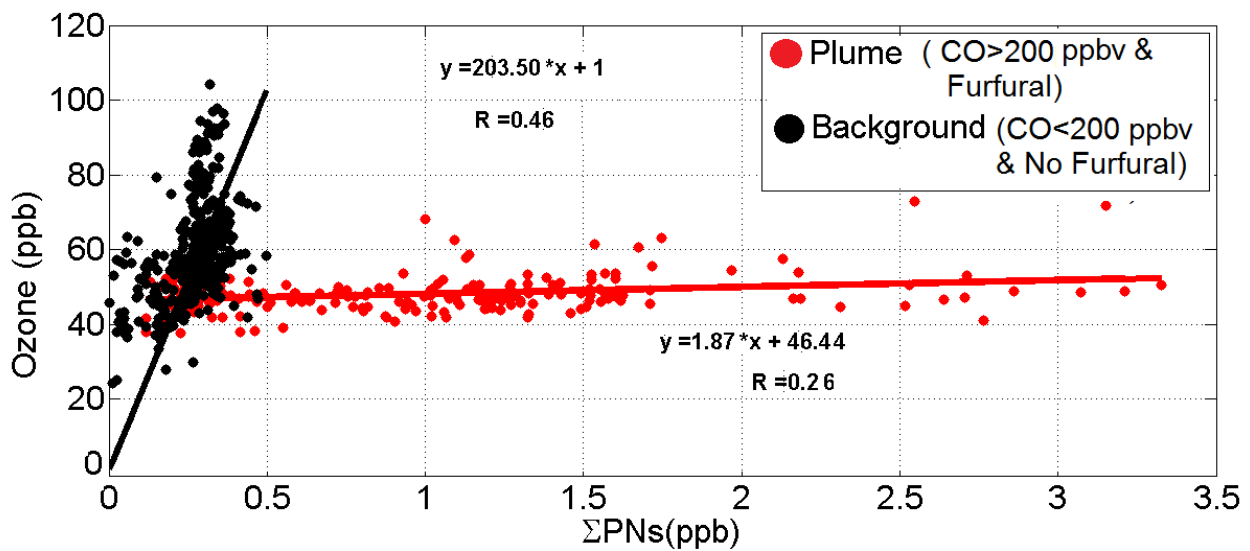
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**Figure 5.** Time series of the  $\Sigma\text{PNs}$ ,  $\Sigma\text{ANs}$ ,  $\text{NO}_2$ ,  $\text{O}_3$ ,  $\text{CO}$ , Furfural (ppbv) measured during the flights in this analysis: the flights B619 (panel a) and B630 (panel d) were background plumes, the flight B622 was in part impacted by fire plume and part by no-fire (panel b), the flight B623 (panel c) was affected by fire plume. The time is reported in Coordinated Universal Time (UTC).

### 3.4 $\Sigma\text{PNs}$ and Ozone

1  
2 The connection between  $O_3$  and  $\Sigma PNs$  is highlighted by the scatterplot of ozone vs  $\Sigma PNs$  mixing  
3 ratios in Fig. 6. Two different dependences can be identified distinguishing the air masses that are  
4 representative of the background environment (flights B619, part of the B622 and B630) and those  
5 emitted or influenced by emissions from biomass burning (flights B623, B624 and part of B622).  
6 We distinguished between the “plume” and the “background” flights as described in Sect. 3.2: that  
38 ppb) and the pyrogenic species analysis. The linear fit of the data influenced by biomass burning  
39 emissions has a slope of  $\sim 1.87$  ppb  $O_3$ /ppb  $\Sigma PNs$  compared to  $\sim 203.5$  for the slope of the linear fit  
40 of background data, which indicates the important role played by the  $\Sigma PNs$  in the sequestration of  
41 ozone precursors in air masses influenced by fire emissions. This can be quantified by calculating  
42 the productions of  $O_3$  and  $\Sigma PNs$ , following the  $\Sigma ANs$  production schemes introduced by Atkinson  
43 (1985) and applied in other studies (Perring et al. 2010). Here we applied the same technique for the  
44 calculation of the  $PNs$  production defining the branching ratio for the peroxy nitrates as  
45  $\alpha = k_{R3}/(k_{R3} + k_{R4})$ . Therefore, the net  $\Sigma PNs$  production is given by  
46  $\alpha(OH + RH + O_2 + NO_2 \rightarrow H_2O + RO_2NO_2)$  and the net  $O_3$  production is described as  
47  $(1 - \alpha)(RH + 4O_2 + h\nu \rightarrow H_2O + R'C(O) + 2O_3)$ .  
48



49

**Figure 6.** Scatter plot between measured O<sub>3</sub> and measured ΣPNs for the flights B619, B622, B623, B624 and B630. Straight line is best fit linear regression. Plume identification follows the methodology and the analysis described in Sect. 3.2.

The production terms can be written as:

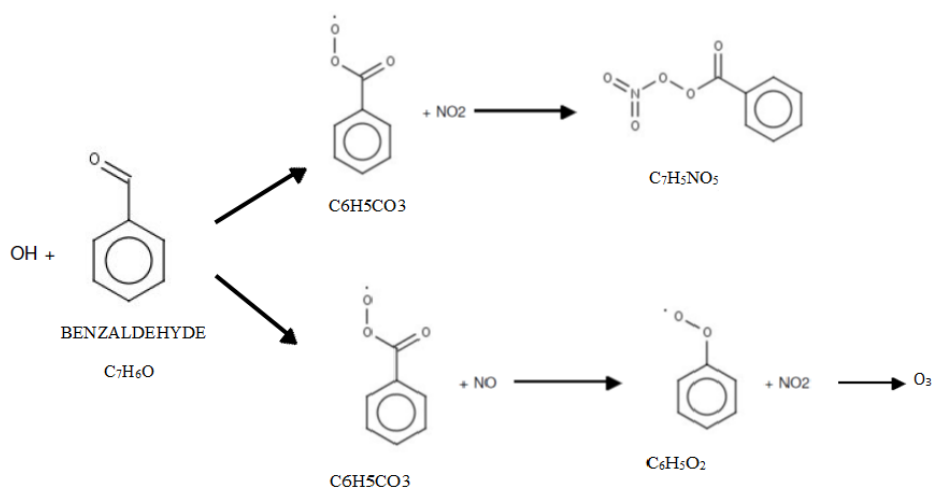
$$P(\sum PNs) = \sum_i \alpha_i k_{OH+RH_i} [OH][VOCs] \quad (1)$$

$$P(O_3) = \sum_i 2(1 - \alpha_i) k_{OH+RH_i} [OH][VOC] + k_{OH+CO} [OH][CO] \quad (2)$$

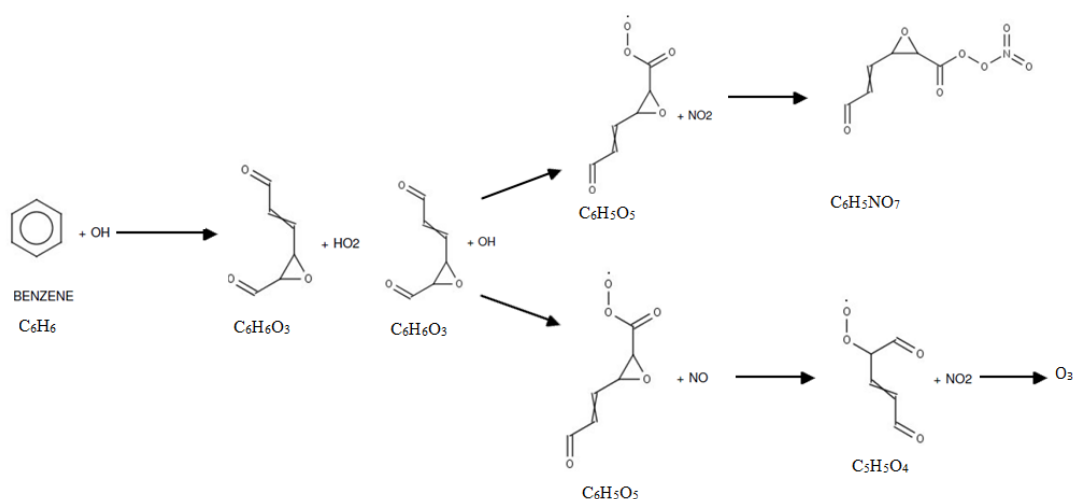
where we considered the weighted sum of the contribution of each VOC to the ΣPNs and to the O<sub>3</sub> production. For the O<sub>3</sub> we take into account also the CO contribution on the  $P(O_3)$  because of significant emissions associated with biomass burning.

In our analysis, we use two approaches to estimate the production of the ΣPNs and O<sub>3</sub>: 1) a direct calculation considering the contribution to the PNs and O<sub>3</sub> production of all the VOCs, among those measured during BORTAS, that produce a PN species after first or second order reactions of the VOCs oxidation by OH; 2) a simulation using a box-model based on the Master Chemical Mechanism (MCM) where all the measured VOCs are used as input compounds to evaluate in output the production of PNs and O<sub>3</sub>. The mechanism used to calculate directly the ΣPNs and O<sub>3</sub> production is similar for all the VOCs therefore we illustrate as an example the production mechanism of the perbenzoyl nitrate (C<sub>7</sub>H<sub>5</sub>NO<sub>5</sub>), derived from the first-order oxidation of the benzaldehyde (C<sub>7</sub>H<sub>6</sub>O) (Figure 7, upper panel) and the production of the PN (C<sub>6</sub>H<sub>5</sub>NO<sub>7</sub>), generated by the second-order oxidation of the benzene (Figure 7, lower panel). In the first case, abstraction of the aldehydic hydrogen by OH followed by O<sub>2</sub> addition forms an acyl peroxy radical (C<sub>7</sub>H<sub>5</sub>O<sub>3</sub>). The acyl peroxy radical can react either with NO<sub>2</sub> forming the perbenzoyl nitrate or with NO producing C<sub>6</sub>H<sub>5</sub>O<sub>2</sub> and NO<sub>2</sub> (Figure 7, upper panel). In the second case, the production of PN starts with the benzene oxidation by OH forming three different products: 11.8% of these reactions generate

1 benzene-1,3,5-triol ( $C_6H_6O_3$ ) and  $HO_2$ . The benzene-1,3,5-triol oxidation by OH, in turn, produces  
 2 a molecule of  $C_6H_5O_5$  in 31% of cases, that finally, reacts with  $NO_2$  to form the peroxy nitrate  
 3  $C_6H_5NO_7$  or with NO generating  $C_5H_5O_4$  plus  $NO_2$  (Figure 7, lower panel). For the branch of  
 4 benzene oxidation that produces PN it is necessary to weight the contribution of the VOC oxidation  
 5 to the PN formation by applying a branching ratio of 0.118 to the reaction constant for the initial  
 6 benzene oxidation by OH and of 0.31 for the following benzene-1,3,5-triol oxidation: hereinafter  
 7 we indicate the OH reaction constant weighted following this method as  $k^*$ . The same procedure  
 8 has been applied also to the other VOCs that do not directly produce peroxy nitrates. Table 3  
 9 summarizes all the species involved in the evaluation of the  $\Sigma$ PNs and  $O_3$  production during all the  
 10 flights, indicating for each of them the OH reaction constant  $k^*$  and the branching ratio calculated  
 11 as  $\alpha = k_{R3} / (k_{R3} + k_{R4})$ .



1



2

3 **Figure 7.** Examples of oxidation schemes that are common to all the VOCs that have as products  
 4 PNs and  $O_3$ . Upper panel: structural formula of the oxidation of benzaldehyde that produces directly  
 5 perbenzoyl nitrate ( $C_7H_5NO_5$ ) and  $O_3$ . Lower panel: structural formula of the oxidation of benzene  
 6 that produce  $O_3$  and indirectly the PN ( $C_6H_5NO_7$ ).

7

8 **Table 3.** Species involved in the calculation of peroxy nitrate and ozone production, their weighted  
 9 reaction constant with OH ( $k^*$  expressed in  $cm^3s^{-1}$ , see the text on how it is calculated) and the  
 10  $\Sigma$ PNs branching ratio ( $\alpha$ ).

| Species      | $k^*$                  | $\alpha$ |
|--------------|------------------------|----------|
| Methacrolein | $1.48 \times 10^{-11}$ | 0.2777   |
| Acetylene    | $2.37 \times 10^{-13}$ | 0.3084   |
| Benzene      | $4.16 \times 10^{-14}$ | 0.3084   |
| Ethylbenzene | $1.82 \times 10^{-13}$ | 0.3084   |
| Toluene      | $1.97 \times 10^{-13}$ | 0.3084   |
| O-Xylene     | $7.29 \times 10^{-12}$ | 0.3084   |
| Benzaldehyde | $1.36 \times 10^{-11}$ | 0.3084   |
| CO           | $2.39 \times 10^{-13}$ | 0        |

1

2 The reaction constants were extracted from the MCM model data or the references therein, and

3 from this, the branching ratios ( $\alpha = k_{R3}/(k_{R3} + k_{R4})$ ) were calculated. For the branching ratio of

4 Methacrolein, the value of  $k_{R4}$  is  $(8.70 \times 10^{-12})\exp(290/T)$ , where T is the temperature, and  $k_{R3}$  was

5 evaluated following the MCM model procedure that takes into account the ambient pressure. For

6 the other species, the  $k_{R4}$  reaction constant is  $(7.50 \times 10^{-12})\exp(290/T)$ , where T is the ambient

7 temperature, and  $k_{R3}$  was evaluated as for methacrolein.

8 The simulation to retrieve the production of  $\Sigma$ PNs and  $O_3$  were carried out using a 0-D

9 Photochemical Box Model (UW Chemical Model, UWCM) that is based on the Master Chemical

10 Mechanism (MCM) version v3.2 (<http://mcm.leeds.ac.uk/MCM/>) into a MATLAB-based source

11 code (Wolfe and Thornton 2011). The MCM is a nearly-explicit reaction set including primary,

12 secondary and radical species and about 17000 reactions to tracks all oxidation processes and

13 products throughout the photochemical degradation of VOCs. The inorganic chemistry has been

14 also included in the simulations. The photolysis reactions constants have been estimated from the

15 TUV model (<http://cprm.acd.ucar.edu/Models/TUV/>). The model has been initialized using both the

16 meteorological parameters (T, P, RH and J-values) and the chemical concentrations of NO, NO<sub>2</sub>,

OH (fixed at  $2 \times 10^6$  molecules/cm<sup>3</sup>, as for the direct calculation), CO, O<sub>3</sub> and all the VOCs (see Table 4) measured during BORTAS campaign. As no OH measurements were made during the BORTAS campaign, its value was chosen to be representative of a northern mid-latitude summertime OH concentration (Spivakovsky et al., 2000). This assumption was validated by Parrington et al. (2013) carrying out several tests in order to compare the photochemical ages using different OH concentrations with the transport timescales from the emission source determined by back trajectory calculations. Table 4 summarizes the mean concentrations of the VOCs and other species used in the simulations, the  $\Sigma$ PNs and O<sub>3</sub> production and their ratio for each flight analysed. The species highlighted with one asterisk have been used also for the direct calculation of  $\Sigma$ PNs and O<sub>3</sub> production terms. The quantities highlighted with two asterisks are the production of PNs and O<sub>3</sub> calculated directly, while those without asterisks are the  $\Sigma$ PNs and O<sub>3</sub> production retrieved from the model simulations.

**Table 4.** Concentrations of each species involved in the  $\Sigma$ PNs and O<sub>3</sub> production (all reported in ppt), the production terms  $P(O_3)$  and  $P(\Sigma PNs)$  (expressed in ppt/s), their ratios  $P(O_3)/P(\Sigma PNs)$  for all the flights analysed. While all the species reported in this table are used for the MCM model calculation of  $P(O_3)$  and  $P(\Sigma PNs)$ , those with \* are species used for the direct calculation of the production using the product between reaction constants and concentrations of the single species. The  $\Sigma$ PNs and O<sub>3</sub> production quantified with the model simulation are signed in this table with \*\*. The selected flights are distinguished between the flights where we sampled boreal fire emissions (part of B622, B623 and B624 – labelled “plume” flights) and those in which we measured background air (B619, part of B622 and B630 – labelled “background” flights).

|   | Parameters | B619   | B622   | B630  | B622   | B623   | B624   |
|---|------------|--------|--------|-------|--------|--------|--------|
| 1 | Ethane     | 1094.0 | 1209.8 | 975.1 | 4705.0 | 2407.5 | 1919.6 |
| 2 | Propane    | 225.0  | 270.4  | 186.0 | 1141.2 | 563.4  | 432.3  |



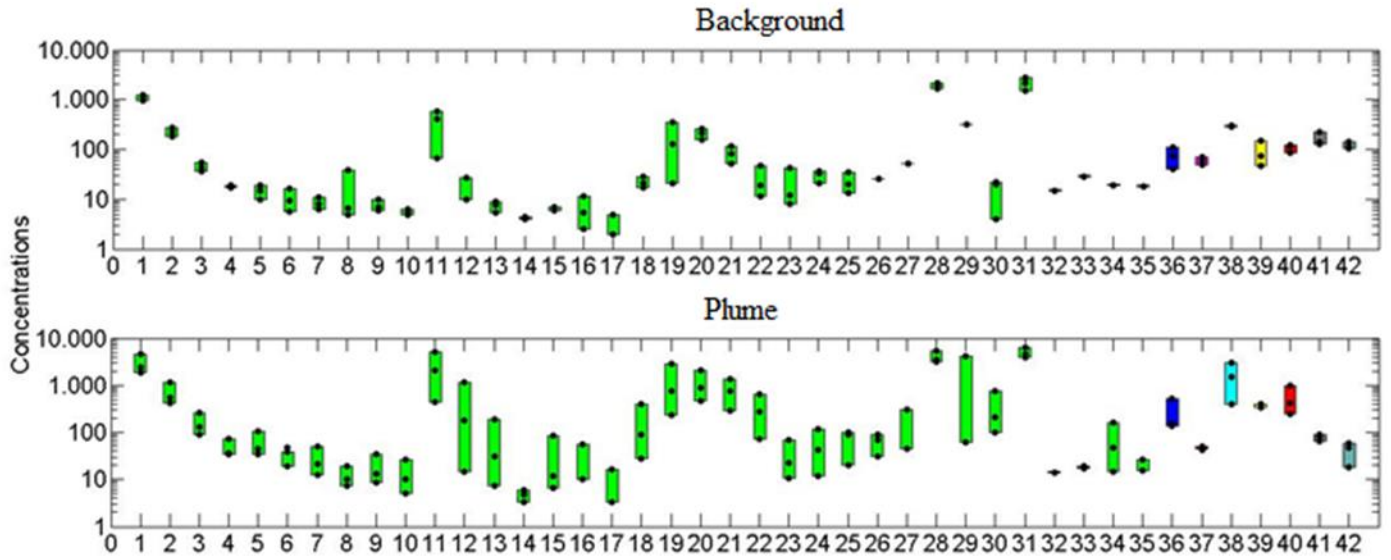
|    |                     |        |        |        |        |        |        |
|----|---------------------|--------|--------|--------|--------|--------|--------|
| 3  | n-Butane            | 42.9   | 53.7   | 36.9   | 258.7  | 133.4  | 89.8   |
| 4  | i-Butane            | 16.8   | 17.9   | 18.6   | 73.3   | 36.7   | 33.8   |
| 5  | n-Pentane           | 14.5   | 18.7   | 10.1   | 106.2  | 46.1   | 34.7   |
| 6  | i-Pentane           | 9.6    | 16.7   | 5.6    | 37.6   | 19.3   | 47.7   |
| 7  | n-Hexane            | 11.0   | 8.0    | 6.3    | 49.4   | 21.0   | 12.7   |
| 8  | 2+3-Methylpentane   | 5.0    | 6.6    | 39.4   | 19.4   | 7.5    | 10.4   |
| 9  | n-Heptane           | 6.0    | 9.9    | 6.8    | 35.1   | 13.5   | 8.8    |
| 10 | n-Octane            | 4.8    | 5.4    | 6.2    | 26.0   | 10.3   | 5.1    |
| 11 | Ethene              | 419.0  | 585.4  | 67.2   | 5115.2 | 2038.4 | 452.5  |
| 12 | Propene             | 27.1   | 27.4   | 10.1   | 1127.6 | 179.8  | 14.7   |
| 13 | 1-Butene            | 7.7    | 9.1    | 5.3    | 185.0  | 31.4   | 7.3    |
| 14 | Trans-2-butene      | 4.0    | 4.3    | 4.5    | 3.3    | 4.8    | 6.1    |
| 15 | i-Butene            | 6.0    | 6.1    | 6.8    | 84.1   | 12.2   | 6.5    |
| 16 | 1-Pentene           | 5.3    | 11.4   | 2.6    | 56.7   | 10.0   | -      |
| 17 | Trans-2-pentene     | 2.0    | 4.8    | 4.9    | 16.1   | 3.4    | -      |
| 18 | 1,3-Butadiene       | 28.3   | 17.1   | 21.4   | 399.1  | 88.9   | 27.5   |
| 19 | Isoprene            | 20.5   | 347.5  | 130.4  | 2796.3 | 763.0  | 231.0  |
| 20 | Acetylene *         | 256.3  | 208.8  | 156.6  | 2053.6 | 887.8  | 480.4  |
| 21 | Benzene *           | 115.5  | 81.1   | 51.6   | 1387.0 | 776.0  | 291.4  |
| 22 | Toluene *           | 46.4   | 18.7   | 11.6   | 636.2  | 282.0  | 72.6   |
| 23 | O-Xylene *          | 12.3   | 7.9    | 43.2   | 68.6   | 22.5   | 10.8   |
| 24 | m+p-Xylene          | 33.6   | 20.6   | 36.0   | 117.8  | 42.8   | 12.2   |
| 25 | E-Benzene *         | 19.9   | 13.1   | 35.3   | 90.6   | 97.6   | 19.9   |
| 26 | Benzaldehyde *      | -      | 26.0   | -      | 68.0   | 30.5   | 88.6   |
| 27 | Acetophenone        | -      | 51.8   | -      | 44.0   | 46.2   | 312.3  |
| 28 | Acetone             | 1692.1 | 1959.9 | 2144.8 | 5561.7 | 3166.5 | 3594.0 |
| 29 | Methyl vinyl ketone | -      | 319.7  | -      | 4126.0 | -      | 62.2   |

|    |                         |                             |                             |                             |         |        |        |
|----|-------------------------|-----------------------------|-----------------------------|-----------------------------|---------|--------|--------|
| 30 | Methacrolein *          | 22.5                        | 20.4                        | 4.0                         | 754.5   | 213.3  | 100.6  |
| 31 | Methanol                | 2119.0                      | 2731.7                      | 1549.9                      | 6369.9  | 3950.8 | 4677.3 |
| 32 | Limonene                | -                           | 15.0                        | -                           | 14.3    | -      | 14.3   |
| 33 | $\alpha$ -Pinene        | -                           | 29.1                        | -                           | 18.5    | 17.5   | 19.3   |
| 34 | Furfural                | -                           | 19.4                        | -                           | 157.5   | 46.5   | 14.4   |
| 35 | Camphor                 | -                           | 18.5                        | -                           | 26.2    | 15.5   | 15.3   |
| 36 | NO <sub>2</sub>         | 40.2                        | 108.8                       | 73.0                        | 507.3   | 137.1  | 153.9  |
| 37 | O <sub>3</sub>          | 71824.8                     | 48217                       | 61195                       | 42431.0 | 45425  | 50858  |
| 38 | $\Sigma$ PNs (ppt)      | 288.5                       | 281.9                       | 298.2                       | 2981.2  | 1543.2 | 407.8  |
| 39 | $\Sigma$ ANs (ppt)      | 148.9                       | 72.3                        | 46.9                        | 404.8   | 399.8  | 335.0  |
| 40 | CO (ppt)                | 84887.4                     | 119559.0                    | 119040                      | 984590  | 419000 | 251540 |
|    | $P(O_3)$ (ppt/s) **     | 0.0420                      | 0.0593                      | 0.0581                      | 0.5082  | 0.2120 | 0.1379 |
|    | $P(\sum PNs)$ (ppt/s)** | 2.9719*<br>10 <sup>-4</sup> | 4.6631*<br>10 <sup>-4</sup> | 2.5807*<br>10 <sup>-4</sup> | 0.0078  | 0.0023 | 0.0017 |
| 41 | $P(O_3)/P(\sum PNs)$ ** | 141.3                       | 127.2                       | 225.0                       | 65.0    | 90.3   | 78.9   |
|    | $P(O_3)$ (ppt/s)        | 0.5133                      | 1.8446                      | 0.5554                      | 5.5643  | 0.6263 | 0.2432 |
|    | $P(\sum PNs)$ (ppt/s)   | 0.0035                      | 0.0163                      | 0.0053                      | 0.1182  | 0.0341 | 0.0041 |
| 42 | $P(O_3)/P(\sum PNs)$    | 145.6                       | 113.5                       | 105.4                       | 47.1    | 18.3   | 58.8   |

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2

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**Figure 8.** Average concentrations of the species involved in the  $O_3$  and  $\Sigma PNs$  production. VOCs are in green, CO in red,  $NO_2$  in blue,  $O_3$  in magenta,  $\Sigma PNs$  in cyan and  $\Sigma ANs$  in yellow. In grey is reported the ratio between the  $P(O_3)$  and  $P(\sum PNs)$  evaluated using the direct calculation (see section 3.3); in teal blue is reported the ratio between the  $P(O_3)$  and  $P(\sum PNs)$  evaluated using the model simulation. The upper shows data measured during background flights (B619, part of B622, B630); the lower panel shows data from fire plume flights (part of B622, B623, B624). The parameters showed in Figure 8 are numbered according to Table 4.

Figure 8 shows graphically the results summarized in Table 4. It is evident that during the background flights both the VOC (in green) and CO (in red) concentrations are significantly lower with respect to those measured during the plume flights, as expected. At the same time, however, the  $O_3$  does not show significantly different concentrations in the biomass burning plumes. Conversely  $\Sigma PNs$  concentrations in the fire plumes increase to a level three times higher than the measurements in background air masses and the alkyl nitrates double. Analysing the measured concentrations of  $O_3$  and  $\Sigma PNs$ , we deduced that the boreal biomass burning emissions affect the  $\Sigma PNs$  production more (on average 12 times higher in the fire plume compared with the background air) than the production, which increase by only 5 times in the fire plume. Using the MCM

1 simulation we got a slightly different increase of  $\Sigma$ PNs production in the fire plume (on average 7  
 2 times), whereas the  $O_3$  production in the fire plume on average increases 2 times. Therefore in the  
 3 fire plumes sampled during the BORTAS campaign, with both methods we observed more  
 4 production of  $NO_x$  reservoir species, which can be transported and potentially impact the  $O_3$   
 5 concentrations in other locations. Alvarado et al. (2010), using a global chemical-transport model,  
 6 estimated that 40% of the initial  $NO_x$  emission from boreal forest fires were converted into PAN.  
 7 Since PAN is one of the compounds included in  $\Sigma$ PNs family, our results show that more  
 8 production of  $\Sigma$ PNs in fire plumes compared with background air is plausible. Moreover,  
 9 calculating the ozone and peroxy nitrate production ratio (Fig. 6), we found that it is lower in the  
 10 fires plumes than in the background samples. This suggests that the production of peroxy nitrates  
 11 during the boreal biomass burning becomes a significant process compared with the ozone  
 12 production, at least in cold air when the thermal dissociation of  $\Sigma$ PNs is not efficient. For example  
 13 PAN, which is usually the most abundant  $\Sigma$ PNs, has a lifetime strongly dependent on temperature: 1  
 14 hr at 300 K, 2 days at 273 K and 118 days at 250 K (Isaksen, 1985). In order to understand the  
 15 impact of a specific category of VOCs, we calculated the contribution of each VOC species and CO  
 16 on the  $\Sigma$ PNs and  $O_3$  production for the fire plume flights (B622, B623 and B624). We find that the  
 17 ozone production, as expected, is dominated by CO (with percentages exceeding 93% for all the  
 18 flights). Moreover, the production of peroxy nitrates is dominated by methacrolein (with  
 19 percentages ranging between 38% and 86%), followed by benzaldehyde (47%-7%) and o-xylene  
 20 (19%-3%). An unusual case, in terms of the peroxy nitrates production, is the background flight  
 21 (B630) during which 75% of  $P(\sum PNs)$  is derived from o-xylene and only 13% from  
 22 methacrolein, which dominates on all the other flights analysed in this study. At first look this is  
 23 strange because methacrolein is one of the major products of isoprene oxidation and it is expected  
 24 that air masses coming from boreal forests (burning or not) would be characterized by high  
 25 concentrations of biogenic VOCs rather than o-xylene which is an anthropogenic VOC. Lai et al.

(2013) found that at the Taipei International Airport (Taiwan) the most abundant VOCs produced by the aircraft exhaust emissions is o-xylene. During the B630 flight the altitude was of about 7000 m a.s.l. (ranging between 7500-6000 m.a.s.l.), higher than the other flights (1700-6000 m.a.s.l.), and the flight track was around the eastern coast of Canada: Nova Scotia and Newfoundland Island. At the flight altitude of B630 it is possible to sample air masses affected by aircraft emissions and, so it is likely that the o-xylene dominance on the  $\Sigma$ PNs production can be explained due to emissions from aircraft traffic.

Finally, the analysis of the  $O_3$  and  $\Sigma$ PNs production in different environments (background and boreal biomass burning plumes) indicates the impact on the tropospheric  $O_3$  budget of the fire emissions. In fact, the air masses influenced by biomass burning emissions show a lower (about 90 with the direct method and about 40 with the model)  $P(O_3)/P(\sum PNs)$  ratio with respect to that for the background air masses (about 180 with the direct method and about 120 with the model) suggesting that the ozone production in the fire plumes is less significant than the peroxy nitrate formation, on the contrary of what occurs in the background air masses. The difference between the calculate ratios and the measured  $O_3/\Sigma$ PNs (see Fig. 6) can be explained considering that: 1) the air masses are not fresh emissions; 2) the  $\Sigma$ PNs production (term at the denominator) is underestimated, as expected since we are not considering all the possible VOCs precursors but only the available for the BORTAS campaign. Moreover, the higher VOCs and  $\Sigma$ PNs concentrations measured during the fire plume flights, associated with stable  $O_3$  levels in the two environments, are indicative of processed air masses (produced 4-5 days before) and suggest that  $NO_2$  reservoir species are produced in these plumes and transported to other regions.

#### 4. Conclusions

1 In July and August 2011 the BORTAS aircraft campaign was carried out in Canada investigating  
2 the impact of the emissions of boreal biomass burning on tropospheric chemistry. We analysed the  
3  $\Sigma$ PNs and  $O_3$  production in two different environments (air masses affected by fire emissions and  
4 those representative of background air) and using different approaches: 1) a direct calculation in  
5 which we considered the VOCs oxidation rate constant and the  $\Sigma$ PNs branching ratios for all the  
6 VOCs species that produce PN after the first or second order reaction of their oxidation by OH; 2)  
7 using a 0-D photochemical model based on MCM that includes a detailed chemistry of all the  
8 VOCs measured. Comparing the production of  $\Sigma$ PNs and  $O_3$  in plumes impacted by fire emissions  
9 with that in background air, we found that, on average,  $\Sigma$ PNs production is more strongly enhanced  
10 than  $O_3$  production: 5 - 12 times versus 2 - 7 times. Boreal biomass burning plumes observed during  
11 BORTAS campaign show minimal enhancement of the  $O_3$  and  $NO_2$  concentrations and slight  
12 enhancement of the  $O_3$  production. However, they show significant enhancement in both  
13 concentration and production of  $\Sigma$ PNs, which can act as a reservoir and enhance ozone production  
14 downwind of the plume.

15

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20

## 21 **References**

22 Alvarado, M. J., J. A. Logan, J. Mao, E. Apel, D. Riemer, D. R. Blake, R. C. Cohen, K.-E. Min, A.  
23 Perring, E. C. Browne, P. J. Wooldridge, G. S. Diskin, G. Sachse, H. Fuelberg, W. R. Sessions,  
24 D. L. Harrigan, L. G. Huey, J. Liao, A. Case-Hanks, J. Jimenez-Palacios, M. J. Cubison, S. A.  
25 Vay, A. Weinheimer, D. J. Knapp, D. D. Montzka, F. Flocke, I. B. Pollack, P. Wennberg, A.  
26 Kurten, J. D. Crounse, J. M. St. Clair, A. Wisthaler, T. Mikoviny, R. M. Yantosca, C. C.  
27 Carouge, and P. Le Sager :Nitrogen oxides and PAN in plumes from boreal fires during

1 ARCTAS-B and their impact on ozone: an integrated analysis of aircraft and satellite  
2 observations, *Atmos. Chem. Phys.*, 10, 9739-9760, doi:10.5194/acp-10-9739-2010,2010.

3 Amiro, B. D., Cantin, A., Flannigan, M. D. and De Groot, W. J.: Future emissions from Canadian  
4 boreal forest fires, *Canadian Journal of Forest Research*, 39(2), 383–395, doi:10.1139/X08-154,  
5 2009.

6 Andreae, M. O. and Merlet, P.: Emission of trace gases and aerosols from biomass burning, *Global*  
7 *Biogeochem. Cy.*, 15, 955–966, 2001.

8 Atkinson, R., Kinetics and Mechanisms of the Gas-Phase Reactions of the Hydroxyl Radical with  
9 Organic Compounds under Atmospheric Conditions *Chem. Rev.* 85, 89-201, 1985.

10 Bowman, D.M.J.S., J.K. Balch, P. Artaxo, W.J. Bond, J.M. Carlson, M.A. Cochrane, C.M. D'Antonio,  
11 R.S. DeFries, J.C. Doyle, S.P. Harrison, F.H. Johnston, J.E. Keeley, M.A. Krawchuk, C.A. Kull,  
12 J.B. Marston, M.A. Moritz, I.C. Prentice, C.I. Roos, A.C. Scott, T.W. Swetnam, G.R. van der  
13 Werf, and S.J. Pyne: Fire in the earth system, *Science*, 324, 481-484, 2009.

14 Chan, C. Y., Chan, L. Y., Harris, J. M., Oltmans, S. J., Blake, D. R., Qin, Y., Zheng, Y. G., and  
15 Zheng, X. D.: Characteristics of biomass burning emission sources, transport, and chemical  
16 speciation in enhanced springtime tropospheric ozone profile over Hong Kong, *J. Geophys. Res.*,  
17 108, 4015, doi:10.1029/2001JD001555, 2003.

18 Chia-Hsiang Lai, Kuen-Yuan Chuang, Jin-Wei Chang: Source Apportionment of Volatile Organic  
19 Compounds at an International Airport, *Aerosol and Air Quality Research*, 13: 689–698, 2013  
20 Copyright © Taiwan Association for Aerosol Research ISSN: 1680-8584 print / 2071-1409  
21 online doi: 10.4209/aaqr.2012.05.0121

22 Crutzen, P.J., L.E. Heidt, J.P. Krasnec, W.H. Pollock and W. Seiler: Biomass burning as a source of  
23 atmospheric gases CO, H<sub>2</sub>, N<sub>2</sub>O, NO, CH<sub>3</sub>Cl and COS. *Nature*, 282, 253-256, 1979.

24 Dari-Salisburgo, C., Carlo, P. D., Giammaria, F., Kajii, Y., and D'Altorio, A.: Laser induced  
25 fluorescence instrument for NO<sub>2</sub> measurements: Observations at a central Italy background  
26 site, *Atmos. Environ.*, 43, 970–977, 2008.

27 Day, D. A., P. J. Wooldridge, M. B. Dillon, J. A. Thornton, and R. C. Cohen: A thermal  
28 dissociation laser-induced fluorescence instrument for in-situ detection of NO<sub>2</sub>, peroxy nitrates,  
29 alkyl nitrates, and HNO<sub>3</sub>, *J. Geophys. Res.*, 107(D6), 4046, doi:10.1029/2001JD000779, 2002.

30 Di Carlo, P., Aruffo, E., Busilacchio, M., Giammaria, F., Dari-Salisburgo, C., Biancofiore, F.,  
31 Visconti, G., Lee, J., Moller, S., Reeves, C. E., Bauguitte, S., Forster, G., Jones, R. L., and  
32 Ouyang, B.: Aircraft based four-channel thermal dissociation laser induced fluorescence  
33 instrument for simultaneous measurements of NO<sub>2</sub>, total peroxy nitrate, total alkyl nitrate, and  
34 HNO<sub>3</sub>, *Atmos. Meas. Tech.*, 6, 971–980, doi:10.5194/amt-6-971-2013, 2013.

1 Draxler, R. R.: HYSPLIT4 user's guide, Tech. Rep. NOAA Tech. Memo. ERL ARL-230, NOAA  
2 Air Resources Laboratory, Silver Spring, MD, 1999.

3 Gerbig, C., S. Schmitgen, D. Kley, A. Volz-Thomas, K. Dewey, and D. Haaks: An improved fast-  
4 response vacuum-UV resonance fluorescence CO instrument, *J. Geophys. Res.*, 104 (D1), 1699–  
5 1704, 1999.

6 Gillett, N., A. J. Weaver, F. W. Zwiers, and M. D. Flannigan : Detecting the effect of climate  
7 change on Canadian forest fires, *Geophys. Res. Lett.*, 31, L18211,  
8 doi:10.1029/2004GL020876, 2004.

9 Goode, J. G., Yokelson, R. J., Ward, D. E., Susott, R. A., Babbitt, R. E., Davies, M. A., and Hao,  
10 W. M.: Measurements of Excess O<sub>3</sub>, CO<sub>2</sub>, CO, CH<sub>4</sub>, C<sub>2</sub>H<sub>4</sub>, C<sub>2</sub>H<sub>2</sub>, HCN, NO, NH<sub>3</sub>, HCOOH,  
11 CH<sub>3</sub>COOH, HCHO and CH<sub>3</sub>OH in 1997 Alaskan Biomass Burning Plumes by Airborne Fourier  
12 Transform Infrared Spectroscopy (AFTIR), *J. Geophys. Res.*, 105, 22147–22166, 2000.

13 Griffin D., Walker K. A., Franklin J. E. , Parrington M., Whaley C. , Hopper J., Drummond J.  
14 R., Palmer P. I. , Strong K., Duck T. J. , Abboud I., Bernath P. F., Clerbaux C. , Coheur P.F. ,  
15 Curry K. R. , Dan L. , Hyer E. , Kliever J. , Lesins G., Maurice M., Saha A. , Tereszchuk K., and  
16 Weaver D. Investigation of CO, C<sub>2</sub>H<sub>6</sub> and aerosols in a boreal fire plume over eastern Canada  
17 during BORTAS 2011 using ground- and satellite-based observations and model simulations  
18 *Atmos. Chem. Phys.*, 13, 10227–10241, 2013

19 Hopkins, J. R., Read, K. A., and Lewis, A. C.: Two column method for long-term monitoring of  
20 non-methane hydrocarbons (NMHCs) and oxygenated volatile organic compounds, *J.*  
21 *Environ. Monitor.*, 5, 8–13, 2003.

22 Jacob, D. J., Wofsy, S. C., Bakwin, P. S., Fan, S.-M., Harriss, R.C., Talbot, R.W., Bradshaw, J.,  
23 Sandholm, S., Singh, H. B., Gregory, G. L., Browell, E. V., Sachse, G. W., Blake, D. R., and  
24 Fitzjarrald, D. R.: Summertime photochemistry at high northern latitudes, *J. Geophys. Res.*, 97,  
25 16421–16431, 1992.

26 Jaffe, D.A., Wigder, N.L.: Ozone production from wildfires: A critical review. *Atmospheric*  
27 *Environment* 51, 1–10, doi:10.1016/j.atmosenv.2011.11.063, 2012.

28 Jenkin, M. E., Saunders, S. M., Wagner, V., and Pilling, M. J.: Protocol for the development of the  
29 Master Chemical Mechanism, MCM v3 (Part B): tropospheric degradation of aromatic volatile  
30 organic compounds, *Atmos. Chem. Phys.*, 3, 181–193, doi:10.5194/acp-3-181-2003, 2003.

31 Isaksen, I. S. A., ed., *Tropospheric Ozone: Regional and Global Scale Interactions*, D. Reidel Pub.  
32 Co., Dordrecht, NATO ASI Series C, Vol. 227, 1988.

33 Langmann, B., Duncan, B., Textor, C., Trentmann, J., and van der Werf, G. R.: Vegetation fire  
34 emissions and their impact on air pollution and climate, *Atmos. Environ.*, 43, 107–116, 2009.



1 Lapina, K., Honrath, R.E., Owen, R.C., Val Martin, M., and Pfister, G.: Evidence of significant  
2 large-scale impacts of boreal fires on ozone levels in the midlatitude Northern Hemisphere free  
3 troposphere. *Geophys. Res. Lett.* 33, L10815, 2006.

4 Lavoué D, Lioussé C, Cachier H, Stocks BJ, Goldammer JG. :Modeling of carbonaceous particles  
5 emitted by boreal and temperate wildfires at northern latitudes. *J. Geophys. Res.: Atmos.*  
6 105(D22): 26871-26890,2000.

7 Lee, J. D., Moller, D. J., Read, K. A., Lewis, A. C., Mendes, L., and Carpenter, L. J.: Year-round  
8 measurements of nitrogen oxides and ozone in the tropical North Atlantic marine boundary  
9 layer, *J. Geophys. Res.*, 114, D21302, doi:10.1029/2009JD011878, 2009.

10 Leung, F.-Y. T., Logan, J. A., Park, R., Hyer, E., Kasischke, E., Streets, D., and Yurganov, L.:  
11 Impacts of enhanced biomass burning in the boreal forests in 1998 on tropospheric chemistry  
12 and the sensitivity of model results to the injection height of emissions, *J. Geophys. Res.*, 112,  
13 D10313, doi:10.1029/2006JD008132, 2007.

14 Lewis, A. C., Evans, M. J., Hopkins, J. R., Punjabi, S., Read, K.A., Purvis, R. M., Andrews, S. J.,  
15 Moller, S. J., Carpenter, L.J., Lee, J. D., Rickard, A. R., Palmer, P. I., and Parrington, M.:The  
16 influence of biomass burning on the global distribution of selected non-methane organic  
17 compounds, *Atmos. Chem. Phys.*, 13, 851–867, doi:10.5194/acp-13-851-2013, 2013.

18 Marlon, J. R., Bartlein, P. J., Carcaillet, C., Gavin, D. G., Harrison, S. P., Higuera, P. E., Joos,  
19 F., Power, M. J., and Prentice, I. C.: Climate and human influences on global biomass burning  
20 over the past two millennia, *Nature Geoscience* , 1, 69–702, 2008.

21 Mauzerall, D., Jacob, D. J., Fan, S.-M., Bradshaw, J., Gregory, G., Sachse, G., and Blake, D.:  
22 Origin of tropospheric ozone at remote high northern latitudes in summer, *J. Geophys. Res.*,  
23 101, 4175–4188, 1996.

24 Murphy, J. G., Oram, D. E., and Reeves, C. E.: Measurements of volatile organic compounds over  
25 West Africa, *Atmos. Chem. Phys.*, 10, 5281–5294, doi: 10.5194/acp-10-5281-2010 , 2010.

26 Nault, B. A., Garland, C., Pusede, S. E., Wooldridge, P. J., Ullmann, K., Hall, S. R., Cohen, and R.  
27 C.: Measurements of CH<sub>3</sub>O<sub>2</sub>NO<sub>2</sub> in the upper troposphere, *Atmos. Meas. Tech.*, 8, 987–997,  
28 2015.

29 Palmer, P. I., Parrington, M., Lee, J. D., Lewis, A. C., Rickard, A.R., Bernath, P. F., Duck, T. J.,  
30 Waugh, D. L., Tarasick, D. W., Andrews, S., Aruffo, E., Bailey, L. J., Barrett, E., Bauguitte, S.  
31 J.-B., Curry, K. R., Di Carlo, P., Chisholm, L., Dan, L., Forster, G., Franklin, J. E., Gibson, M.  
32 D., Griffin, D., Helmig, D., Hopkins, J. R., Hopper, J. T., Jenkin, M. E., Kindred, D., Kliever,  
33 J., Le Breton, M., Matthiesen, S., Maurice, M., Moller, S., Moore, D. P., Oram, D. E., O’Shea, S.  
34 J., Owen, R. C., Pagniello, C.M. L. S., Pawson, S., Percival, C. J., Pierce, J. R., Punjabi,

S., Purvis, R. M., Remedios, J. J., Rotermund, K. M., Sakamoto, K. M., da Silva, A. M.,  
 Strawbridge, K. B., Strong, K., Taylor, J., Trigwell, R., Tereszchuk, K. A., Walker, K. A.,  
 Weaver, D., Whaley, C., and Young, J. C.: Quantifying the impact of BOREal forest fires on  
 Tropospheric oxidants over the Atlantic using Aircraft and Satellites (BORTAS) experiment:  
 design, execution and science overview, *Atmos. Chem. Phys.*, 13, 6239–6261, doi:10.5194/acp-  
 13-6239-2013, 2013.

Parrington, M., Palmer, P. I., Henze, D. K., Tarasick, D. W., Hyer, E. J., Owen, R. C., Helmig, D.,  
 Clerbaux, C., Bowman, K. W., Deeter, M. N., Barratt, E. M., Coheur, P.-F., Hurtmans, D., Jiang,  
 Z., George, M., and Worden, J. R.: The influence of boreal biomass burning emissions on the  
 distribution of tropospheric ozone over North America and the North Atlantic during  
 2010, *Atmos. Chem. Phys.*, 12, 2077–2098, doi:10.5194/acp-12-2077-2012, 2012.

Parrington, M., Palmer, P. I., Lewis, A. C., Lee, J. D., Rickard, A. R., Di Carlo, P., Taylor, J. W.,  
 Hopkins, J. R., Punjabi, S., Oram, D. E., Forster, G., Aruffo, E., Moller, S. J., Bauguutte, S. J.-  
 B., Allan, J. D., Coe, H., and Leigh, R. J.: Ozone photochemistry in boreal biomass burning  
 plumes, *Atmos. Chem. Phys.*, 13, 7321–7341, doi:10.5194/acp-13-7321-2013, 2013.

Pfister, G., Emmons, L. K., Hess, P. G., Honrath, R., Lamarque, J.-F., Val Martin, M., Owen, R. C.,  
 Avery, M., Browell, E. V., Holloway, J. S., Nedelec, P., Purvis, R., Rywerson, T. B.,  
 Sachse, G. W., and Schlager, H.: Ozone production from the 2004 North American boreal fires, *J.*  
*Geophys. Res.*, 111, D24S07, doi:10.1029/2006JD007695, 2006.

Perring A. E., Bertram T. H., Farmer D. K., Wooldridge P. J., Dibb J., Blake N. J., Blake D. R., Singh  
 H. B., Fuelberg H., Diskin G., Sachse G., and Cohen R. C. The production and persistence of  
 $\Sigma\text{RONO}_2$  in the Mexico City plume, *Atmos. Chem. Phys.*, 10, 7215–7229, 2010.

Purvis, R. M., Lewis, A. C., Hopkins, J. R., Andrews, S., and Minaeian, J.: Functionalized aromatic  
 compounds within middle troposphere boreal biomass burning plumes, in preparation, 2013.

Real, E., Law, K. S., Weinzierl, B., Fiebig, M., Petzold, A., Wild, O., Methven, J., Arnold, S.,  
 Stohl, A., Huntrieser, H., Roiger, A., Schlager, H., Stewart, D., Avery, M., Sachse, G., Browell,  
 E., Ferrare, R., and Blake, D.: Processes influencing ozone levels in Alaskan forest fire plumes  
 during long-range transport over the North Atlantic, *J. Geophys. Res.*, 112,  
 D10S41, doi:10.1029/2006JD007576, 2007.

Reid, J. S., Hyer, E. J., Prins, E. M., Westphal, D. L., Zhang, J., Wang, J., Christopher, S. A., Curtis,  
 C. A., Schmidt, C. C., Eleuterio, D. P., Richardson, K. A., and Hoffman, J. P.: Global monitoring  
 and forecasting of biomass burning smoke: Description of and lessons from the Fire Locating  
 and Modeling of Burning Emissions (FLAMBE) program, *IEEE J. Sel. Top. Appl.*, 2, 144–162,  
 2009.

- Reidmiller, D. R., Jaffe, D. A., Fischer, E. V., and Finley, B.: Nitrogen oxides in the boundary layer and free troposphere at the Mt. Bachelor Observatory, *Atmos. Chem. Phys.*, 10, 6043–6062, doi:10.5194/acp-10-6043-2010, 2010.
- Rinsland, C. P., Dufour, G., Boone, C. D., Bernath, P. F., Chiou, L. Coheur, P.-F., Turquety, S., and Clerbaux, C. :Satellite boreal measurements over Alaska and Canada during June–July 2004: Simultaneous measurements of upper tropospheric CO, C<sub>2</sub>H<sub>6</sub>, HCN, CH<sub>3</sub>Cl, CH<sub>4</sub>, C<sub>2</sub>H<sub>2</sub>, CH<sub>3</sub>OH, HCOOH, OCS, and SF<sub>6</sub> mixing ratios, *Global Biogeochemical Cycles*, Vol.21, GB3008, doi:10.1029/2006GB002795, 2007.
- Saunders, S. M., Jenkin, M. E., Derwent, R. G., and Pilling, M. J.: WWW site of a master chemical mechanism (MCM) for use in tropospheric chemistry models, *Atmos. Environ.*, (Report Summary), 31, 1249, 1997.
- Saunders, S. M., Jenkin, M. E., Derwent, R. G., and Pilling, M. J.: Protocol for the development of the Master Chemical Mechanism, MCM v3 (Part A): tropospheric degradation of nonaromatic volatile organic compounds, *Atmos. Chem. Phys.*, 3, 161–180, doi:10.5194/acp-3-161-2003, 2003.
- Simpson, I. J., Rowland, F. S., Meinardi, S., and Blake, D. R. :Influence of biomass burning during recent fluctuations in the slow growth of global tropospheric methane, *Geophys. Res. Lett.*, 33, L22808, doi:10.1029/2006GL027330, 2006.
- Spivakovsky, C. M., Logan, J. A., Montzka, S. A., Balkanski, Y. J., Foreman-Fowler, M., Jones, D. B. A., Horowitz, L. W., Fusco, A. C., Brenninkmeijer, C. A. M., Prather, M. J., Wofsy, S. C., and McElroy, M. B.: Three-dimensional climatological distribution of tropospheric OH: Update and evaluation, *J. Geophys. Res.*, 105, 8931–8980, 2000.
- Tereszczuk, K. A., Gonz´alez Abad, G., Clerbaux, C., Hurtmans, D., Coheur, P.-F., and Bernath, P. F.: ACE-FTS measurements of trace species in the characterization of biomass burning plumes, *Atmos. Chem. Phys.*, 11, 12169–12179, doi:10.5194/acp-11-12169-2011, 2011.
- Val Martin, M., Honrath, R., Owen, R. C., Pfister, G., Fialho, P., and Barata, F.: Significant enhancements of nitrogen oxides, ozone and aerosol black carbon in the North Atlantic lower free troposphere resulting from North American boreal wildfires, *J. Geo-phys. Res.*, 111, D23S60, doi:10.1029/2006JD007530, 2006.
- Verma, S., Worden, J., Pierce, B., Jones, DBA., Al-Saadi, J., Boersma, F., Bowman, K., Eldering, A., Fisher, B., Jourdain, L., Kulawik, S., Worden, H. :Ozone production in boreal fire smoke plumes using observations from the Tropospheric Emission Spectrometer and the Ozone Monitoring Instrument, *J. Geophys. Res.*, 114, 0.1029/2008JD010108, 2009.

1 Wilson, K. L. and J. W. Birks: Mechanism and Elimination of a Water Vapor Interference in the  
2 Measurement of Ozone by UV Absorbance, *Environmental Science and Technology* 40, 6361-  
3 6367,2006.

4 Wofsy, S.C., Sachse, G.W., Gregory, G.L., Blake, D.R., Bradshaw, J.D., Sandholm, S.T., Singh,  
5 H.B., Barrick, J.A., Harriss, R.C., Talbot, R.W., Shipham, M.A., Browell, E.V., Jacob, D.J. and  
6 Logan, J.A.:Atmospheric chemistry in the Arctic and subarctic: Influence of natural fires,  
7 industrial emissions, and stratospheric inputs. *Journal of Geophysical Research* 97: doi:  
8 10.1029/92JD00622. issn: 0148-0227, 1992.

9 Wolfe, G.M. and Thornton, J.A.: The Chemistry of Atmosphere Forest Exchange (CAFE) Model -  
10 Part 1:Model Description and Characterization, *Atmospheric Chemistry and Physics*, 11, 77-101,  
11 2011.

12 Wotawa, G., and M. Trainer :The influence of Canadian forest fires on pollutant concentrations in  
13 the United States. *Science* 288(5464):324-328,2000.

14 Wotton, B. M., Nock, C. A., and Flannigan, M. D.: Forest fire occurrence and climate change in  
15 Canada, *International Journal of Wildland Fire*, 19, 253–271, 2010.

16  
17  
18  
19