

## 1 **Responses to anonymous referee #1**

2 *Received and published: 18<sup>th</sup> January 2016*

3 We thank the reviewer for the careful review of our manuscript; the comments and  
4 suggestions are greatly appreciated. All the comments have been addressed. In the following  
5 please find our responses to the comments one by one and the corresponding changes made to  
6 the manuscript. The original comments are shown in italics.

7

8 *1) The scientific quality and presentation of the manuscript are good. The manuscript will*  
9 *need copy-editing because there are numerous grammatical errors, although the text is still*  
10 *clear. There is agreement between the various instruments utilised at the site (ACSM vs SMPS*  
11 *vs MARGA), and the PMF analysis follows established procedures. I do have questions about*  
12 *the determination of organic nitrates and about the correlations between the OA components*  
13 *and tracers, and these questions are given below. On the other hand, the scientific*  
14 *significance of the manuscript is not clear. Aerosol mass spectrometry (AMS) including*  
15 *ACSM measurements are fairly standard now, and AMS measurements were previously*  
16 *carried out at this site, as indicated in the manuscript. I acknowledge that the long duration of*  
17 *the measurements in this manuscript is somewhat novel, but it is not clear how having simply*  
18 *more measurements provides new significant insight into atmospheric chemistry. This*  
19 *concern will need to be addressed before final publication, in addition to the comments*  
20 *provided below.*

### 21 **Response:**

22 Fortunately, copy-editing is standard in ACP. We are relieved though, that despite some  
23 language issues the manuscript is understandable and clear.

24 The reviewer is right that a publication just presenting longer records obtained with AMS or  
25 ACSM is not per se novel enough to warrant publication in a quality journal such as ACP,  
26 also considering that two AMS campaigns were previously performed in Cabauw. We  
27 emphasize here that the focus of this work extends beyond the presentation of a larger data set  
28 for this site:

29 In this study, we showed that the major fraction of atmospheric aerosol at this site is  
30 comprised of secondary aerosol (SA) which is chemically formed in the atmosphere. SA also  
31 dominates the high mass loadings during the pollution episodes exceeding air quality  
32 standards. The long term aerosol mass spectrometric data provided by this work is important

1 for the understanding of the chemical processes leading to such high aerosol masses which are  
2 unexpected for a rural site like Cabauw. This information is necessary for potential mitigation  
3 of particle masses. The findings presented here were not seen in previous studies due to their  
4 limited time of sampling.

5

6 **General Comments:**

7 *2) Organic nitrates: The difference in the concentration of nitrate measured by the ACSM and*  
8 *MARGA is not a direct means of identifying and quantifying the presence of organic nitrates.*  
9 *Is there other evidence from the ACSM or other instruments to support the conclusion that*  
10 *these compounds are present and account for 9% of the total ACSM NO<sub>3</sub>? If not, then this*  
11 *conclusion is rather weakly supported by the data.*

12 **Response:**

13 We followed a procedure given by Xu et al. (2015) who calculated the organic nitrate fraction  
14 by subtracting the inorganic nitrate concentrations measured by a particle-into-liquid sampler  
15 (PILS, see Orsini et al. (2003)) from ToF-AMS total nitrate concentrations. Those instruments  
16 have similar uncertainties as the respective instruments reported here. In the revised  
17 manuscript we provide following additional evidence for the presence of OrgNO<sub>3</sub> in the  
18 aerosol: the use of external data from the MARGA to determine the inorganic nitrate in the  
19 ACSM data set improved the agreement of measured against predicted ACSM-NH<sub>4</sub>, resulting  
20 in a nearly 1:1 regression line without a significant offset. This is in agreement with the  
21 MARGA internal ion balance which also indicates neutralized inorganic aerosols. All  
22 evidence together makes a strong case that the difference of AMS total nitrate to MARGA  
23 nitrate is most likely due to the presence of organic nitrates. These findings and the previously  
24 reported relatively high AMS organic nitrate fractions by Mensah (2011) in May 2008  
25 (0.5 μg m<sup>-3</sup>, 35% of total nitrate, 5.2% of total aerosol mass) and March 2009 (0.2 μg m<sup>-3</sup>,  
26 10% of total nitrate, 3.6% of total aerosol mass) in Cabauw show strong evidence of the  
27 presence of organic nitrates during this campaign and the reliability of its estimation as  
28 presented in the manuscript.

29 To provide a more detailed description this part was changed in the revised manuscript as  
30 follows:

31 "This assumption is acceptable, as shown by using the MARGA-NO<sub>3</sub> instead of the ACSM-  
32 total-NO<sub>3</sub> for the ion balance of ACSM data (including ACSM-SO<sub>4</sub>, -Chl and -NH<sub>4</sub>),

1 following a procedure given by Xu et al. (2015) who calculated the organic nitrate fraction by  
2 subtracting the inorganic nitrate concentrations measured by a particle-into-liquid sampler  
3 (PILS, see Orsini et al. (2003)) from ToF-AMS total nitrate concentrations. In the Cabauw  
4 data set, the correlation of measured against predicted  $\text{NH}_4$  resulted in a nearly 1:1 regression  
5 line without a significant offset (Fig. S10). This is in agreement with the MARGA internal ion  
6 balance which also indicates neutralized inorganic aerosols. Therefore, the mass concentration  
7 of nitrate groups associated with organic molecules (hereafter called organic nitrate or  
8  $\text{OrgNO}_3$ ), can be estimated by subtracting the MARGA-nitrate from the ACSM-nitrate  
9 concentration. The  $\text{OrgNO}_3$  time series using this approach is plotted in Fig. S11, the  
10 respective diurnal variation averaged over the whole campaign in Fig. S12. An average mass  
11 fraction of 9% was calculated for  $\text{OrgNO}_3$  (average concentration:  $0.43 \mu\text{g m}^{-3}$ ) in respect to  
12 total ACSM- $\text{NO}_3$ . The organic nitrate fraction shows a maximum concentration in the night,  
13 followed by a decrease during the day. These findings are in agreement with previously  
14 reported relatively high AMS organic nitrate fractions by Mensah (2011) in May 2008  
15 ( $0.5 \mu\text{g m}^{-3}$ , 35% of total nitrate, 5.2% of total aerosol mass) and March 2009 ( $0.2 \mu\text{g m}^{-3}$ ,  
16 10% of total nitrate, 3.6% of total aerosol mass) in Cabauw.”

17

18 3) *Correlation of PMF time series with tracers: The correlations of the PMF factors with the*  
19 *tracer time series seems very low for HOA, BBOA and HULIS, and the correlation*  
20 *coefficients ( $R^2$ ) are all below 0.5 for all the tracers used in this study. For example, the*  
21 *HULIS factor, which is presumably secondary in nature, exhibits the highest correlation with*  
22 *BC rather than one of the primary combustion factors (i.e. HOA and BBOA). This lack of*  
23 *correlation indicates that PMF is not identifying OA components that can be unambiguously*  
24 *associated with specific sources. The lack of correlation should be examined in more detail as*  
25 *it calls into question the validity of the PMF results.*

26 **Response:**

27 As mentioned in the manuscript the HULIS factor was found as high oxygenized background  
28 aerosol in Cabauw, which is characterized by the low temporal variation and therefore  
29 reasonably low correlation coefficients with most of the tracers. On the other hand the  
30 reviewer noticed correctly that the correlation results given as Pearson- $R^2$  in case of the POA  
31 factors with BC data don't seem to be as high as seen in previous studies showing PMF/ME-2  
32 data. This has several reasons:

1 The black carbon data used in this study derived from a MAAP, which measures  
2 simultaneously the radiation penetrating through and scattered back from a particle-loaded  
3 fiber filter. Andreae and Gelencser (2006) pointed out that non-BC light-absorbing  
4 carbonaceous matter like brown carbon can interfere with BC measurements using this  
5 instrument. They also stated that filters loaded with macroscopic amounts of humic like  
6 substances (HULIS) samples look very dark brown or nearly black. This would imply a  
7 possible interference of HULIS with BC. Consequently this may explain why the HULIS  
8 factor has a relatively high correlation with BC. In summary, black carbon measured by the  
9 MAAP can have three main sources: traffic and biomass burning, represented by the HOA  
10 and BBOA factors, respectively, and the interference with HULIS. All three sources are  
11 independent from each other, leading to the observed correlation results. The use of  
12 instruments like an aethalometer or a particle soot absorption photometer as used in other  
13 studies would reduce these interferences (Andreae and Gelencser, 2006; Petzold et al., 2013),  
14 but were not available during the campaign.

15 As expected, the gaseous tracers  $\text{NO}_x$  and CO exhibited the highest  $R^2$  values with HOA, and  
16 higher correlations with BBOA than with both SOA factors.

17 In the revised manuscript the respective part was changed to:

18 “Note that the correlation (Pearson- $R^2$ ) of the POA factors with eBC data is relatively low  
19 ( $R^2 = 0.38$  and  $0.39$  with HOA and BBOA, respectively), while in turn HULIS shows a higher  
20 correlation with eBC,  $R^2 = 0.47$ . This can be understood in the light of a study by Andreae  
21 and Gelencser (2006) who pointed out that the fiber filters, as used by the MAAP, loaded with  
22 macroscopic amounts of humic like substances samples look very dark brown or nearly black.  
23 This implies an interference of HULIS with BC measurements and in consequence explains  
24 why the HULIS factor, in contrast to the POA factors, has a relatively high correlation  
25 coefficient with BC data.”

26

27 **Specific Comments:**

28 4) P35121, L20: *What is the difference between intensively and extensively managed*  
29 *grassland?*

30 **Response:**

31 The phrase “intensively managed” agricultural land refers to the extended use of fertilization,  
32 irrigation or drainage techniques. In case of grasslands this definition is also applied to fields

1 where, with respect to the agricultural area, a relatively high amount of farm animals are hold  
2 in order to optimize the respective production. Consequently, “extensively managed” areas  
3 refer to the opposite: keeping the pasture area in their natural conditions and using them with  
4 less numbers of animals. Since these phrases are common agricultural definitions, the authors  
5 decided to not explain them in detail.

6

7 5) P35122, L2: *Remote sensing is a very general term. The manuscript should specify exactly*  
8 *which remote sensing measurements are made at the tower site.*

9 **Response:**

10 In the revised manuscript, we have modified this sentence as follows:

11 “In addition, other meteorological data like precipitation, radiation and remote sensing,  
12 including lidar, radar and radiometer techniques, are acquired at the tower and submitted to  
13 the CESAR data base.”

14

15 6) Page 35123, L27: *In contrast with what is implied in the text, I believe that the*  
16 *Middlebrook et al. (2012) CE correction algorithm also accounts for high ammonium nitrate.*

17 **Response:**

18 We agree with the reviewer that the ammonium nitrate mass fraction (ANMF) is used as a  
19 parameter within the algorithm published by Middlebrook et al. (2012). Specifically ANMF is  
20 used when the ratio of measured to predicted  $\text{NH}_4$  is higher than 0.75. Otherwise a different  
21 equation using just this ratio is applied to determine the CE in the Middlebrook algorithm. For  
22 the Cabauw ACSM data set, where  $\text{OrgNO}_3$  could only be determined from the combined  
23 ACSM and MARGA datasets, the ratio of measured to predicted  $\text{NH}_4$  varies around 0.75,  
24 which lead to unreasonable discontinuities of CE values when applying the Middlebrook  
25 algorithm. The authors agree that the current explanation is misleading. In the revised  
26 manuscript the respective part was changed to:

27 “In contrast to the commonly used constant value of 0.5 this CE correction accounts for the  
28 high ammonium nitrate mass fraction (ANMF) found at this site and is thus more suitable for  
29 the data presented here. Another algorithm for composition dependent CE determination  
30 (Middlebrook et al., 2012) was also tested for its validity. It uses a threshold ratio of measured  
31 to predicted  $\text{NH}_4$  to switch between two different equations to determine the CE. The  
32 threshold value of 0.75 is close to the observed ratio of measured over predicted  $\text{NH}_4$  of this

1 data set, resulting in large discontinuities of CE values and in consequence, discontinuous  
2 changes in aerosol mass concentrations. In other words, the Middlebrook algorithm is not  
3 suitable for data sets showing at the same time low ratios of measured to predicted  $\text{NH}_4$  and  
4 high AMNF's."

5  
6 7) Page 35124, L5 – L8: *Why was the detection limit not calculated during the measurement*  
7 *period or afterwards using data from the measurement period? Data from periods when the*  
8 *instrument was sampling behind a particle filter could be used for this analysis.*

9 **Response:**

10 Unfortunately, there were no ACSM measurements done during the presented campaign  
11 where a separate particle filter was introduced in the sample line, in addition to the filter  
12 included in the gas-phase background filter cycle as described by Ng et al. (2011).  
13 Furthermore, the ACSM software version used in this study could not show data acquired  
14 during the filter cycle measurements (e.g. closed mass spectra/time series), which would be  
15 needed for the determination of the detection limits. The software could only show the  
16 differential mass spectra/time series.

17  
18 8) Page 35125, L23 – 29: *The correction for the wall losses in the 60 m sampling line is a*  
19 *critical point for the manuscript. However, the description of how the losses were calculated*  
20 *or estimated is not sufficient. Only a reference to a personal communication is provided.*  
21 *Given that this correction can impact the aerosol measurements substantially – by 33% as*  
22 *indicated by the authors – the manuscript must contain a detailed explanation of how the*  
23 *aerosol losses in the sampling line are determined.*

24 **Response:**

25 We agree with the reviewer that the correction and explanation for these losses is not  
26 sufficient enough.

27 In the revised manuscript we used now the particle density deriving from the chemical  
28 composition not as the campaign average but time resolved for each data point. In addition we  
29 introduced SMPS data which is now size dependently corrected as published by Henzing  
30 (2011). The description of the sampling losses of the 60 m inlet for eBC given in the  
31 manuscript derived from a series of measurements at the Cabauw tower performed in a  
32 previous campaign. Unfortunately these results are not published yet. To clarify how these

1 losses were determined in the revised manuscript and to account for a comment from referee  
2 #2, the loss description was changed as follows:

3 “SMPS data was corrected size dependently for (diffusional) losses in the inlet system and  
4 SMPS system itself according to Henzing (2011) who compared theoretical findings with  
5 measured losses that are obtained by measuring simultaneously before and after the various  
6 parts of the inlet system at the CESAR tower. In addition, particles of different compositions  
7 were measured in 2013 simultaneously at the pipe entrance at 60 m height and in the  
8 basement (J. S. Henzing, personal communication). For more than 8000 simultaneous  
9 observations, the results showed that aerosol measurements through this 60 m sampling line  
10 underestimate  $PM_{10}$ -eBC by approximately 33% with an uncertainty of 7%. Therefore, eBC  
11 obtained from the MAAP are divided by a factor of 0.66 to account for these losses. For the  
12 inorganic species penetrations through this inlet line were reported to be 62-73% for nitrate,  
13 55-64% for sulfate, and 54-56% for ammonium. However these results were not used for  
14 corrections in this work”

15 Please note that these losses influenced only data acquired by the MAAP and SMPS. Since  
16 the contribution of eBC is rather low (average: 5%) a potential overall error for total aerosol  
17 masses is low and would not significantly alter one quintessence of the paper, namely total  
18 mass concentrations above the air quality limits.

19 As a consequence of the newly evaluated SMPS data the correlation values between  
20 ACSM+MAAP data with SMPS data changed as seen in Fig. S3 and S4 in the revised  
21 supplement. Nevertheless, the overall qualitative and quantitative agreement is still given  
22 except that the ACSM+MAAP data is now overestimating the total  $PM_1$  mass by 16%,  
23 excluding the eBC data the ACSM overestimates total mass by 12%. As seen in Fig. S3 the  
24 difference between both systems is significantly higher during the pollution events 16 to 27  
25 January 2013 and 5 to 8 May 2013. Since the quantitative agreement with the MARGA is  
26 much higher at these times the discrepancy to the SMPS is likely due to the fact that the losses  
27 within the 60 m inlet could not be corrected for individual species as mentioned above.  
28 Therefore the following paragraph was added at the end of the cross validation chapter in the  
29 revised manuscript:

30 “Major discrepancies to the SMPS especially during some of the pollution events like 16 to 27  
31 January 2013 and 5 to 8 May 2013 (see below) can be explained by the correction of losses  
32 through the 60 m inlet line which was done size dependently and did not account for losses of  
33 individual species as mentioned in chapter 2.3. As the quantitative agreements of individual

1 inorganic species as well as of total inorganics between the ACSM and the MARGA during  
2 these periods are much higher, the mass loadings determined from these instruments are more  
3 reliable than the SMPS data.”

4  
5 9) Page 35126, L13 – L16: During the measurements with the MARGA system, a polyethylene  
6 tube was used as the sampling line. Could the use of a non-conducting material for the line  
7 lead to substantial electrostatic deposition of the particles to the walls of the tube?

8 **Response:**

9 Indeed, a polyethylene (PE) tube can potentially enhance wall losses of particles comparing to  
10 stainless steel tubes. The MARGA-inlet system at the Cabauw tower as used for this study  
11 was previously described by Schaap et al. (2011). It actually did not only consist of PE tubes  
12 but of a series of components reducing particle losses. They investigated wall losses on a  
13 similar system and found only minor concentration losses for several compounds of 2% and  
14 less. A more detailed description was added in the revised manuscript:

15 “The sample air was transferred into the instrument within a polyethylene („Polyflo“) tube  
16 with an inner diameter of 0.5” (= 1.27 cm) and a sample flow of 16.7 L min<sup>-1</sup>, which is either  
17 directed through a PM<sub>1</sub> or a PM<sub>2.5</sub> size selective head. A detailed description of the MARGA  
18 inlet system at the Cabauw tower was previously described by Schaap et al. (2011). There,  
19 wall losses were investigated and found to be less than 2% for several gaseous and particulate  
20 compounds.”

21  
22 10) Page 35129, L7: Why is SO<sub>2</sub> included in the MARGA PM<sub>2.5</sub> mass? In the atmosphere,  
23 this compound is found in the gas phase.

24 **Response:**

25 We thank the reviewer for pointing out this mistake. Not only SO<sub>2</sub> but all gas phase  
26 compounds HNO<sub>2</sub>, HNO<sub>3</sub> and HCl were mistakenly added to total aerosol masses.  
27 Nevertheless, the conclusions deriving from these calculations did not change. In the revised  
28 manuscript, the resulting time series “MARGA PM<sub>2.5</sub> + ACSM-Org + eBC” in Fig. 1 was  
29 exchanged and the sentence was changed to:

30 Nevertheless, the average total mass derived from the combination of the MARGA PM<sub>2.5</sub> data  
31 (including all water soluble inorganic components NO<sub>3</sub>, NH<sub>4</sub>, SO<sub>4</sub>, Cl, Na, K, Mg, and Ca),



1 MAAP eBC and ACSM PM<sub>1</sub> organics resulted in 12.5 µg m<sup>-3</sup> during this time, clearly  
2 exceeding the WHO PM<sub>2.5</sub> annual mean limit (10 µg m<sup>-3</sup>) by 25%.

3 The reader may now notice that the concentration values of MARGA PM<sub>2.5</sub> + MAAP eBC +  
4 ACSM PM<sub>1</sub> Organics (light blue time series in Fig. 1) decreased reasonably because the  
5 mentioned gas phase data was excluded. On the other hand, more data points are now seen.  
6 This is due to the fact that data coverage of the MARGA PM<sub>2.5</sub> data is better than for the  
7 MARGA gas phase data.

8

9 *11) Page 35129, L9: It appears that there is an error on this line and the concentration for*  
10 *the annual mean limit should be 10 µg m<sup>-3</sup> rather than 25 µg m<sup>-3</sup> as is currently written.*

11 **Response:**

12 The value is indeed wrong. In the revised manuscript it was changed from 25 to 10 µg m<sup>-3</sup>.

13

14 *12) Page 35130, L6 – L9: This sentence is confusing. First, it is written that the pollution*  
15 *events are highlighted in green, but (to my eye) the highlighting is grey. Second, there are*  
16 *four periods that are highlighted in Figure 2, but only three periods are listed in the text. The*  
17 *dates for the missing period should be given as well.*

18 **Response:**

19 The authors agree that the color used for highlighting cannot be designated unambiguously.  
20 To keep it simple, we renamed it as just “shaded”. Figure 2 is the only figure where two  
21 different colors were used for shaded areas. Therefore, the shading for the period where  
22 ACSM and AMS data were overlapping was removed in this figure in the revised manuscript.  
23 Consequently, the captions from Figures 1 and 4 as well as from several figures in the revised  
24 supplement file were changed by calling the “green shaded” areas as just “shaded” instead of  
25 giving them a certain color. Furthermore and accounting for a comment from referee #2, the  
26 sentences on page 35130, lines 6-8, were changed in the revised manuscript to:

27 “The most significant pollution events (17 to 21 August 2012, 21 to 25 October 2012, 16 to  
28 27 January 2013, and 5 to 8 May 2013) are highlighted with shaded backgrounds in Fig. 2.  
29 During the last three periods, northerly and north-easterly winds dominated.”

30

1 13) Page 35132, L21: From Figure S10, it appears that the intercept is negative rather than  
2 positive (i.e.  $-0.80$  versus  $0.80$ ). The sign is important as there is likely a small amount of dust  
3 and perhaps sea salt which is measured by the SMPS and not the ACSM that is responsible  
4 for the non-zero intercept value.

5 **Response:**

6 Indeed, the value in the main text has a wrong algebraic sign. As described in the respond to  
7 comment #8 we introduced newly evaluated SMPS data in the revised manuscript by which  
8 the values changed. The intercept is now  $-1.05 \pm 0.06$  but still negative. We thank the  
9 reviewer for pointing out that a negative offset infers influence from sea salt or dust aerosols,  
10 which can be detected well by the SMPS and MARGA but not by the ACSM with a sufficient  
11 sensitivity. Therefore we added a small discussion to this section in the revised manuscript:

12 “The negative offset can be explained by minor influences of sea salt and dust particles,  
13 which can be detected well by the SMPS and MARGA but not by the ACSM with a sufficient  
14 sensitivity. But the low value of the intercept shows already that the uncertainty introduced by  
15 these aerosol components is rather low in general. This can also be explained by the low  
16 concentrations of Mg, Na, K and Ca as measured by the MARGA (see below) and the  
17 assumption that the majority of dust particles is most likely found in particles with diameters  
18 larger than 1 or even  $2.5 \mu\text{m}$  (Finlayson-Pitts and Pitts (2000) and references therein).”

19

20 14) Page 35133, L18 – 20: Why couldn't the fragmentation table be adjusted? Adjustment of  
21 the fragmentation table is fairly standard during analysis of AMS data. If it is believed that  
22 the fragmentation table may be responsible for the underestimation of  $\text{SO}_4$ , then the authors  
23 should explore making possible adjustments of the values in the table.

24 **Response:**

25 We agree with the reviewer that for AMS data a proper adjustment of the fragmentation table  
26 is recommended especially for unit mass resolution data, which is produced by the ACSM,  
27 too. Unfortunately, the ACSM software used in this study did not allow plotting time series  
28 for specific fragments apportioned by the frag table, e.g. “frag\_sulphate[48]” like it is  
29 commonly done in the AMS analysis toolkits SQUIRREL or PIKA. Without exploring the  
30 resulting changes of these time series by adjusting the fragmentation table it did not make  
31 sense to vary the fragmentation table entries. To clarify that the revised manuscript was  
32 adjusted to:

1 Additionally, the ACSM fragmentation table could not be adjusted for interferences of ions  
2 from different aerosol species on the same m/z properly because ACSM software used in this  
3 study did not allow plotting time series for specific fragments apportioned by the  
4 fragmentation table. Thus, the standard table had to be used.

5

6 *15) Figure 1: Using zeros to fill in the missing values for black carbon for periods when*  
7 *measurements are not available is not a standard approach (to my knowledge). It would be*  
8 *preferable if the existing BC data was simply just averaged. It is already very clear from the*  
9 *manuscript that data is missing in certain periods, so a reader can take this into account*  
10 *when evaluating the data. Alternatively, the pie chart could be modified to include only the*  
11 *ACSM data for when the MAAP was functional.*

12 **Response:**

13 Unfortunately, the figure caption here was misleading: Missing values of eBC data was only  
14 filled with zeros for the calculation of the time series of the daily means. For the calculation  
15 of the pie chart values eBC data was just averaged as suggested by the reviewer. In the  
16 revised manuscript, the caption for this figure was changed to:

17 **Figure 1: Time series of the daily mean. The black line represents the sum of eBC and all**  
18 **ACSM species, the blue line the sum of eBC, ACSM organics and all MARGA-PM<sub>2.5</sub>**  
19 **species. The pie chart shows the fractional abundances of individual eBC and ACSM species**  
20 **averaged over the whole campaign. For the determination of the daily means missing eBC**  
21 **data was filled with zero values, thus deriving lower concentration limits.**

22 We thank the reviewer to bring up this ambiguity.

23

24 *16) Figure 4: I assume the authors mean to say “grey shaded areas” rather than “green*  
25 *shaded areas”.*

26 **Response:**

27 Please see the response for author’s comment #12 above.

28

29

30

1 **Supporting Information:**

2 17) Page 2, L3: I think this equation should be  $MFNO3 \geq 0.78$ .

3 **Response:**

4 Indeed. In the revised manuscript, this equation was corrected.

5

6 18) Figure S1: The figure legend text is very small and hard to read. The font size should be  
7 increased to improve readability.

8 **Response:**

9 In the revised supplement material, this picture is enlarged to enhance the readability of the  
10 legend. The picture of the tower itself was removed.

11

12 19) Figure S2: I assume the authors mean to say “gray shaded areas” rather than “green  
13 shaded areas”.

14 **Response:**

15 Please see the response for author’s comment #12 above.

16

17 20) Figure S6: Similar to the previous comments, the shaded areas look grey to me rather  
18 than green. Perhaps there is a difference in how the colors are displayed by the author’s  
19 monitor and my monitor? There are subsequent figures where this comment applies as well,  
20 but I will not repeat it to avoid being overly repetitive.

21 **Response:**

22 Please see the response for author’s comment #12 above.

23

24 21) Figure S13: It would improve the presentation of the results if all the bar graphs were  
25 plotted with the same format. The size of the bars is not consistent.

26 **Response:**

27 In the revised manuscript, the format of all four graphs is similar.

## 1 **References**

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17 concentrations of ambient aerosol. *Aerosol Science and Technology* 45, 780-794.
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29 VanWertz, G., Kreisberg, N.M., Knote, C., Olson, K., Koss, A., Goldstein, A.H., Hering,  
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31 of anthropogenic emissions on aerosol formation from isoprene and monoterpenes in the  
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- 33
- 34

1 **Responses to anonymous referee #2**

2 *Received and published: 19<sup>th</sup> February 2016*

3 We thank the reviewer for the careful review of our manuscript; the comments and  
4 suggestions are greatly appreciated. All the comments have been addressed. In the following  
5 please find our responses to the comments one by one and the corresponding changes made to  
6 the manuscript. The original comments are shown in italics.

7

8 **General Comments:**

9 *1) Order of sections: Please exchange 3.1. and 3.2 (data quality assurance should be before*  
10 *the results). You might even consider moving section 3.2 to chapter 2. Also in the PMF results*  
11 *section, the applied constraints should be described first, and then the results should be*  
12 *discussed. See also my comments below. With respect to the discussion on organic nitrate, I*  
13 *am very sceptical that the data support the conclusions. The uncertainties of both instruments*  
14 *are too large for an estimation of organic nitrate from the difference between ACSM and*  
15 *MARGA.*

16 **Response:**

17 The authors agree with the reviewer that the cross validation chapter should be placed before  
18 the presentation of the results themselves. In the revised manuscript chapter 3.1 and 3.2 were  
19 exchanged.

20 We also accept the suggestion from the reviewer to move the description for the applied a  
21 values to the beginning of chapter 3.3 in the revised manuscript. Referring to comments #12  
22 and #14 from the reviewer, we added a sentence describing that the choice for the a-values  
23 resulted from a series of sensitivity tests:

24 "Figure 4 summarizes the time series of PMF factor classes (two POA factors: HOA and  
25 BBOA; and two SOA factors: OOA and a humic-like substances (HULIS) related factor, see  
26 below for descriptions) found for all seasons, except for BBOA in Summer 2012. A  
27 corresponding graph dividing these PMF results into the five periods according to Fig. 2 is  
28 shown in Fig. S13. The POA profiles were constrained within ME-2 using the HOA and  
29 BBOA mass spectra found by ME-2 operated PMF analyses by Crippa et al. (2014) at the  
30 CESAR tower in Cabauw in March 2009. For the HOA profile, a constant a-value of 0.1 was  
31 found to be most suitable for every season. If observed, the BBOA a-value was set to either

1 0.2 (Autumn 2012) or 0.3 (Winter and Spring 2013). The a-values based on different  
2 sensitivity tests for each season as described by Canonaco et al. (2013).”

3 Organic nitrate related aspects are answered in the specific comments below.

4  
5 **Specific Comments:**

6 2) *Abstract: Please mention the height above ground of the ACSM. Until the end of section*  
7 *2.2 the reader thinks that the ACSM was located on top of the tower.*

8 **Response:**

9 The first sentence of the abstract was changed in the revised manuscript to:

10 “Intensive measurements of submicron aerosol particles and their chemical composition were  
11 performed with an Aerosol Chemical Speciation Monitor (ACSM) at the Cabauw  
12 Experimental Site for Atmospheric Research (CESAR) in Cabauw, NL, sampling at 5 m  
13 height above ground.”

14  
15 *Section 2.2: Instrument description (second paragraph of this section) should be placed*  
16 *before the statements on the agreement between different ACSMs and the ACTRIS*  
17 *comparisons.*

18 **Response:**

19 In the revised manuscript we accepted the suggestion of the reviewer and placed the  
20 instrument description before the statements on the agreement between different ACSMs and  
21 the ACTRIS comparisons.

22  
23 3) *Page 35124, line 5-8: Detection limits? It is not clear to me why they were not determined?*  
24 *As far as I understand, a blank filter measurement is done on a routine time schedule by an*  
25 *ACSM, and then the detection limit can be calculated as three times the average noise level.*

26 **Response:**

27 As responded to a similar comment from referee #1, there were no measurements done with  
28 this ACSM during the presented campaign, where a separate particle filter was introduced in  
29 the sample line, in addition to the filter including in the gas-phase background filter cycle as  
30 described by Ng et al. (2011). Furthermore, the ACSM Local software version used in this

1 study could not show data acquired during the filter cycle measurements (e.g. closed mass  
2 spectra/time series), which would be needed for the determination of the detection limits. The  
3 software could only show the differential mass spectra/time series.

4  
5 4) Page 35124, line 12-15: Particle losses: I recommend using the PLC by von der Weiden et  
6 al (AMT, 2009) to estimate the losses in such long a sampling line (all vertical or horizontal?)  
7 [www.atmos-meas-tech.net/2/479/2009/](http://www.atmos-meas-tech.net/2/479/2009/) <http://www.mpchmainz.mpg.de/drewnick/PLC/>

8 **Response:**

9 As responded to a similar comment from referee #1, we used now the particle density  
10 deriving from the chemical composition not as the campaign average but time resolved for  
11 each data point in the revised manuscript. In addition we introduced SMPS data which is now  
12 size dependently corrected as published by Henzing (2011). The description of the sampling  
13 losses of the 60 m inlet for eBC given in the manuscript derived from a series of  
14 measurements at the Cabauw tower performed in a previous campaign. Unfortunately these  
15 results are not published yet. To clarify how these losses were determined in the revised  
16 manuscript and to account for a comment from referee #2, the loss description was changed as  
17 follows:

18 “SMPS data was corrected size dependently for (diffusional) losses in the inlet system and  
19 SMPS system itself according to (Henzing, 2011) who compared theoretical findings with  
20 measured losses that are obtained by measuring simultaneously before and after the various  
21 parts of the inlet system at the CESAR tower. In addition, particles of different compositions  
22 were measured in 2013 simultaneously at the pipe entrance at 60 m height and in the  
23 basement (J. S. Henzing, personal communication). For more than 8000 simultaneous  
24 observations, the results showed that aerosol measurements through this 60 m sampling line  
25 underestimate PM<sub>10</sub>-eBC by approximately 33% with an uncertainty of 7%. Therefore, eBC  
26 obtained from the MAAP are divided by a factor of 0.66 to account for these losses. For the  
27 inorganic species penetrations through this inlet line were reported to be 62-73% for nitrate,  
28 55-64% for sulfate, and 54-56% for ammonium. However these results were not used for  
29 corrections in this work”

30 Please note that these losses influenced only data acquired by the MAAP and SMPS. Since  
31 the contribution of eBC is rather low (average: 5%) a potential overall error for total aerosol  
32 masses is low and would not significantly alter one quintessence of the paper, namely total



1 mass concentrations above the air quality limits. We think theoretical particle loss calculations  
2 would not add additional information in the context of this paper.

3 As a consequence of the newly evaluated SMPS data the correlation values between  
4 ACSM+MAAP data with SMPS data changed as seen in Fig. S3 and S4 in the revised  
5 supplement. Nevertheless, the overall qualitative and quantitative agreement is still given  
6 except that the ACSM+MAAP data is now overestimating the total PM<sub>1</sub> mass by 16%,  
7 excluding the eBC data the ACSM overestimates total mass by 12%. As seen in Fig. S3 the  
8 difference between both systems is significantly higher during the pollution events 16 to 27  
9 January 2013 and 5 to 8 May 2013. Since the quantitative agreement with the MARGA is  
10 much higher at these times the discrepancy to the SMPS is likely due to the fact that the losses  
11 within the 60 m inlet could not be corrected for individual species as mentioned above.  
12 Therefore the following paragraph was added at the end of the cross validation chapter in the  
13 revised manuscript:

14 “Major discrepancies to the SMPS especially during some of the pollution events like 16 to 27  
15 January 2013 and 5 to 8 May 2013 (see below) can be explained by the correction of losses  
16 through the 60 m inlet line which was done size dependently and did not account for losses of  
17 individual species as mentioned in chapter 2.3. As the quantitative agreements of individual  
18 inorganic species as well as of total inorganics between the ACSM and the MARGA during  
19 these periods are much higher, the mass loadings determined from these instruments are more  
20 reliable than the SMPS data.”

21  
22 5) Page 35124, lines 24 ff: MAAP: Please mention that a MAAP reports equivalent black  
23 carbon (eqBC), see Petzold et al., *Atmos. Chem. Phys.*, 13, 8365-8379, doi:10.5194/acp-13-  
24 8365-2013, 2013

25 **Response:**

26 We thank the reviewer for the supportive remark and pointing out the reference. In the revised  
27 manuscript, the MAAP description was written in more detail as follows:

28 “The MAAP instrument has been introduced by Petzold and Schönlinner (2004) and Petzold  
29 et al. (2005). It is designed for the determination of the black carbon (BC), which is a product  
30 of incomplete combustion. There is in the scientific community a general consensus over  
31 what black carbon is in terms of properties (Bond et al., 2013) The MAAP measures the  
32 strong visible light absorption property of BC by simultaneous measurements of the radiation

1 penetrating through and scattered back from a particle-loaded fiber filter. According to  
2 Petzold et al. (2013), optical BC determined by MAAP is to be referred to as equivalent black  
3 carbon (eBC). One property of BC is that it is highly refractory with a vaporization  
4 temperature near 4000K (Schwarz et al., 2006), thus BC is not vaporized at 600°C and cannot  
5 be measured by the ACSM. The MAAP achieves a time resolution of 5 minutes with an  
6 uncertainty of 12% (Petzold and Schönlinner, 2004).”

7 In addition, we used the term equivalent black carbon or its abbreviation (eBC) in the revised  
8 manuscript and supplement including tables and figures, whenever its measured values are  
9 given, since eBC represents the actual quantitative the MAAP is acquiring.

10  
11 6) Page 35125, lines 23-29: *For the 60 m inlet the sampling losses were calculated. So why*  
12 *not for the ACSM inlet?*

13 **Response:**

14 As pointed out in the response to the reviewer’s comment #4, the losses in the 60 m sampling  
15 line were measured on site in a previous campaign. These measurements could reasonably not  
16 be done for the ACSM inlet within the scope of the presented study.

17  
18 7) Page 35127, line 15: *38Cl should read H37Cl, I assume? There is no 38Cl isotope*

19 **Response:**

20 Indeed, the interference on m/z 38 is due to the H<sup>37</sup>Cl ion. The respective sentence was  
21 changed in the revised manuscript to:

22 “This was done because the signal at these masses showed high interferences with the  
23 chloride related ions <sup>37</sup>Cl and H<sup>37</sup>Cl.”

24  
25 8) Page 35128, line 9-11: *"Even when only considering the ACSM + MAAP PM1*  
26 *concentration, where a campaign average of 9.5 µg m<sup>-3</sup> was determined, the WHO PM2.5*  
27 *limit was approximated." I suggest rephrasing as: Even the PM1 concentration inferred from*  
28 *ACSM + MAAP data (campaign average 9.5 µg m<sup>-3</sup>) approached the WHO PM2.5 limit.*

1 **Response:**

2 We accepted the suggestion of the reviewer and changed this part accordingly in the revised  
3 manuscript.

4

5 9) Page 35131 line 29 – page 35132, line 8: You calculate organic nitrate from the difference  
6 of ACSM and MARGA. I think that bears quite some uncertainty considering the large  
7 uncertainties of the instruments. At least you should provide an error estimation. What about  
8 looking at ion ratios, e.g., 30/46? This is known to differ between inorganic and organic  
9 nitrates? Can you plot 30/46 vs. the difference between MARGA-nitrate and ACSM-nitrate?  
10 Later on, on page 35133, lines 6-7, you state that the uncertainties are 30% for ACSM-nitrate  
11 and 10% for MARGA-nitrate. Thus, how can you infer 9% organic NO<sub>3</sub>?

12 **Response:**

13 As responded to the similar comment from referee #1, there are strong evidences for the  
14 existence of particulate organic nitrates during this campaign despite the uncertainties of both  
15 instruments. We agree with the reviewer that these explanations are contradictory to those  
16 given in the cross-validation chapter on the comparison of ACSM and MARGA total nitrates.  
17 The latter part (page 35133, lines 4-12) was changed in the revised manuscript to:

18 “The quantitative difference to the AMS-organics is also very low, and the discrepancies in  
19 case of ammonium and nitrate are within the stated  $\pm 30\%$  accuracy of the AMS and ACSM  
20 (Ng et al., 2011) and the  $\pm 10\%$  for the MARGA-NO<sub>3</sub>, respectively (Makkonen et al., 2012).  
21 Similar variations were also found by Crenn et al. (2015) as well as Budisulistiorini et al.  
22 (2014). The latter reported of a comparison between two collocated ACSM’s ( $\pm 27\%$ ,  $R^2 =$   
23  $0.21$  for Chl,  $R^2 > 0.8$  for the other species) and between these ACSM’s and a continuous  
24 Tapered Element Oscillating Microbalance (TEOM, PM<sub>2.5</sub>) instrument. Note that the major  
25 ions used for nitrate detection in AMS and ACSM (NO<sup>+</sup> and NO<sub>2</sub><sup>+</sup>) are produced from both  
26 inorganic and organic nitrate (e.g. Farmer et al. (2010)). The higher ACSM nitrate compared  
27 to MARGA nitrate can therefore also be explained by the presence of organic nitrates. In case  
28 of nitrate the ASCM and MARGA comparison cannot be seen as independent”

29 As the reviewer pointed out correctly, the ratio of m/z46 (NO<sub>2</sub><sup>+</sup>) to m/z30 (NO<sup>+</sup>) can be used  
30 as a parameter to determine to organic and inorganic nitrate fractions. To do that, the 46/30  
31 ratios of purely inorganic and pure organic nitrates need to be known. The value of purely  
32 inorganic nitrate for individual instruments can be taken from ammonium nitrate

1 measurements during IE calibrations. For AMS instruments this ratio was commonly seen to  
2 be around 0.4 (Fry et al., 2011; Fry et al., 2009; Mensah, 2011). The corresponding ratio of  
3 purely organic nitrates ( $R_{\text{OrgNO}_3}$ ) was found to be approximately 0.1 in chamber experiments,  
4 measuring SOA from the reactions of BVOC's with  $\text{NO}_3$  radicals with an HR-ToF-AMS (Fry  
5 et al., 2011; Fry et al., 2009; Mensah, 2011). Analogous experiments were not yet published  
6 for ACSM instruments. During the presented campaign in Cabauw, a 46/30 ratio of around  
7 0.2 was observed during the response factor calibrations of the ACSM. Crenn et al. (2015)  
8 reported a similar value (0.23) for this particular ACSM and that this ratio is strongly varying  
9 between the investigated ACSM's (0.20-0.58). That implies that it is highly uncertain that the  
10 46/30 ratio of 0.1 for organic nitrates should be used for ACSM data, at least for the campaign  
11 presented here. Furthermore, as stated in the response to a comment from referee #1, the  
12 fragmentation table of the ACSM Local software used here could not be user defined. Thus  
13 possible interferences on  $m/z$ 30 and  $m/z$ 46 between nitrate and organic fragments could not  
14 be corrected. Those interferences would in turn increase the uncertainty of the  $\text{NO}_2^+/\text{NO}^+$  ratio  
15 determination from unit mass resolution data.

16

17 *10) Page 35132, line 10-12: What value was assumed for the density of organics?*

18 **Response:**

19 As mentioned on page 35125, line 12, a density of  $1.4 \text{ g cm}^{-3}$  for organics was assumed.

20

21 *11) Page 35133, line 12 ff: There seems to be a problem with sulfate here. The ACSM*  
22 *appears to underestimate  $\text{SO}_4$  by a factor of two compared to the HR-ToF-AMS. In the paper*  
23 *by Crenn et al (Atmos. Meas. Tech., 8, 5063–5087, 2015, [www.atmos-](http://www.atmos-meastech.net/8/5063/2015/)*  
24 *[meastech.net/8/5063/2015/](http://www.atmos-meastech.net/8/5063/2015/) doi:10.5194/amt-8-5063-2015) it looked much better. How were*  
25 *the RIEs for the ACSM and for the HR-ToF-ASM determined? How often for the ACSM? Did*  
26 *all sulfate calibrations suffer from high background signals?*

27 **Respond:**

28 The RIE of sulfate was calculated using a common procedure for AMS and ACSM  
29 instruments (e.g. Budisulistiorini et al. (2014) and Elser et al. (2015)). Monodisperse  
30 ammonium sulfate particles were measured right after the measurement of monodisperse  
31 ammonium nitrate. Using the RIE of ammonium deriving from the ammonium nitrate data,  
32 the RIE of sulfate was determined by an ion balance of data derived by the ammonium sulfate

1 measurements. For the ACSM, this was done every 1-2 months (AMS: once per week) with  
2 relatively stable results ( $RIE_{SO_4} = 0.81 \pm 0.10$ ). Close after the sampling of ammonium sulfate  
3 (up to  $120 \mu\text{g m}^{-3}$ ), a largely increased background of sulfate was observed during all  
4 calibrations, leading to the conclusions mentioned in the manuscript.

5 As mentioned in Crenn et al. (2015), the accuracy of AMS- and ACSM-sulfate measurements  
6 is still under debate. In their work, RIE calibrations were done for all instruments, including  
7 the particular ACSM used here. For the inter-comparison of the ACSM data sets with the  
8 collocated HR-ToF-AMS, Crenn et al. (2015) took the default  $RIE_{SO_4}$  of 1.2 for all ACSM's,  
9 although the calibration results were partly much lower with strongly varying values of  
10  $RIE_{SO_4}$  between 0.48 and 0.97, while for the AMS the experimental  $RIE_{SO_4}$  was applied  
11 (Fröhlich et al., 2015). Unfortunately, Crenn et al. (2015) did not publish comparison results  
12 with the AMS for individual ACSM's. Overall we think that in the case of sulfate it is  
13 difficult to generalize from Crenn et al. (2015) on AMS-ACSM comparisons for the presented  
14 work.

15 The reader should also take into account that the overlap period of the AMS and ACSM  
16 comprises only one week (294 common data points) in the beginning of the ACSM campaign  
17 during a time with low mass concentrations, while the comparison with the MARGA  $PM_{10}$   
18 data could be done over the whole ACSM data set (1943 common data points), leading to  
19 more representative correlation results. There, a better quantitative and qualitative agreement  
20 in case of sulfate was found comparing to the ACSM-AMS cross-validation.

21

22 *12) Section 3.3 Factor analysis:*

23 *You prefer the 4-piece separate analysis over the one-year analysis, but I don't see an*  
24 *objective reason for this. To my opinion this discrepancy reflects the uncertainty of the factor*  
25 *analysis. I appreciate that you openly show and discuss the differences, but the conclusion*  
26 *that the 1-year analysis "overestimates" something is not backed up by any data, or am I*  
27 *wrong? If you argue that BBOA is overestimated because of the high degree of freedom ( $a =$*   
28 *0.3), why didn't you restrict it more ( $a = 0.1$ ) as the HOA? You should state the reasons for*  
29 *choosing a certain a value at the beginning for the section. If the 1-year analysis would be*  
30 *run with  $a=0.1$ , would it be better? In general I would describe first the constraints, then the*  
31 *factors, then the alternative 1-year solution, and then conclude that latter cannot be correct*  
32 *(no  $m/z$  60 in summer).*

33

1 **Response:**

2 As described in the response to comment #1 of the reviewer, we added an explanation that the  
3 a-values were determined by a standardized procedure described in detail by Canonaco et al.  
4 (2013) and Crippa et al. (2014), in addition to the description of the source apportionment as  
5 applied in this work in chapter 2.4 (specifically page 35128, lines 1-13). In contrast to  
6 unconstrained PMF commonly published for AMS data, a higher amount of exploration runs  
7 including sensitivity tests are necessary to choose the most suitable a-values for constrained  
8 factors and overall the most suitable solution for a data set within a ME-2 analysis. It is not  
9 feasible to report all these runs with individual explanations on choices of solutions.

10 We agree with the reviewer that the conclusion of a possible overestimation of BBOA in  
11 summer due to the high degree of freedom cannot be sufficiently proven. Further PMF  
12 analyses with a constrained BBOA and a-values lower than 3 gave the same results with only  
13 very minor differences regarding profiles and time series. As suggested by the reviewer, the  
14 description of the 1-year PMF was moved to the end chapter 3.3 in the revised manuscript and  
15 changed as follows:

16 “The source apportionment as described here used a data set which was subdivided into the  
17 four seasons prior to PMF analysis. A single PMF analysis of the whole data set with  
18 constrained HOA and BBOA profiles lead to solutions with a highly overestimated BBOA  
19 factor in the summer, compared to the results when the seasons were explored individually  
20 (see Fig. S17 and S18 in the supplement). Furthermore, the contributions of individual factors  
21 change significantly in some periods, especially for the OOA factor during pollution events.  
22 This is mainly driven by the different OOA-*f*<sub>44</sub> and -*f*<sub>43</sub> values. This behavior is independent  
23 from applied a-values for BBOA and may result from the uncertainty of this statistical tool.  
24 Since there was no evidence of BBOA seen in the separate analysis of the summer period  
25 (e.g. low fraction of *m/z* 60 in the organic mass spectrum and no correlation of the  
26 constrained BBOA factor with POA tracers, no matter which a-value was used), the solution  
27 derived from the single PMF analysis was reasonably rejected.”

28

29 *13) Page 35136 line 16: Typo: it's -> its*

30 **Response:**

31 This part was corrected in the revised manuscript.

32

1 14) Page 35137 line 10: Explain humic-like substances (HULIS) on first occurrence

2 **Response:**

3 To clarify the HULIS term, the introductory sentence on page 35134, line 18, was changed in  
4 the revised manuscript to:

5 “Figure 4 summarizes the time series of PMF factor classes (two POA factors: HOA and  
6 BBOA; and two SOA factors: OOA and a humic-like substances (HULIS) related factor, see  
7 below for descriptions) found for all seasons, except for BBOA in Summer 2012.”

8

9 15) Conclusions: This section is too short and too weak and should be expanded.

10 **Response:**

11 To meet this suggestion by the reviewer we restructured the conclusions section and moved  
12 some concluding statements from previous sections to this part. The conclusions now  
13 emphasize the important finding that NH<sub>3</sub> emission control can significantly decrease PM  
14 loading at this rural site. Furthermore the high SOA fraction in the aerosol challenges  
15 mitigation strategies in particular when large background fractions are observed as is the case  
16 with HULIS here. To account for comments from other reviewers as well the revised section  
17 now reads:

18 This work provides chemical composition data of atmospheric aerosols acquired during one  
19 year at the CESAR tower in Cabauw, the Netherlands, which is a representative rural site for  
20 North Western Europe. The concentration of submicron particles from combined ACSM and  
21 MAAP data showed 12 exceedances from the WHO PM<sub>2.5</sub> daily mean limit. The respective  
22 campaign average of 9.5 µg m<sup>-3</sup> approached the WHO PM<sub>2.5</sub> annual mean limit. Taking  
23 MARGA PM<sub>2.5</sub> into account the number and proportions of these exceedances are even  
24 higher, emphasizing the importance of these high ACSM PM<sub>1</sub> results shown here which  
25 represent lower limits. As carbonaceous compounds are estimated to be five times more toxic  
26 than inorganic particles (Lelieveld et al., 2015) MARGA data alone would not give  
27 sufficiently possible implications regarding adverse health effects. While few people live in  
28 the direct vicinity of the measurement site, the high aerosol concentration measured at the site  
29 can be considered to represent the regional background. This regional background is adding to  
30 local aerosol contributions in high populated urban sites (Pandis et al., 2013), namely the 4  
31 largest cities of the Netherlands which have a distance of 40 km or less from the CESAR  
32 tower.

1 Particulate mass loadings found at this rural site are dominated by secondary aerosol  
2 formation through atmospheric gas phase chemistry and particle phase aging. It is shown that  
3 particulate ammonium nitrate is the major aerosol component (39% on average) and  
4 represents the more hygroscopic aerosol fraction. Since the human respiratory system is  
5 characterized by high humidity more hygroscopic aerosols have a higher deposition tendency  
6 within the human lung than less water soluble particle compounds (Asgharian, 2004; Broday  
7 and Georgopoulos, 2001). With regard to adverse health effects this is very crucial because  
8 Asgharian (2004) also found that especially hygroscopic submicron particles can deposit in  
9 the entire lung. The high ammonium nitrate fraction also implies that inorganic SA reduction  
10 in Cabauw can be most efficiently achieved through the reduction of gaseous ammonia  
11 emissions in the area.

12 As indicated by model results for the South Western United States from Zhang and Wu  
13 (2013), the reduction of  $\text{NH}_3$  emissions, in conjunction with already implemented  $\text{SO}_2$  and  
14  $\text{NO}_x$  emission reductions, can further reduce  $\text{PM}_{2.5}$  than reducing  $\text{SO}_2$  and  $\text{NO}_x$  emissions  
15 alone, particularly for regions with high emissions of  $\text{NH}_3$  from agricultural sources. The  
16 latter is clearly the case for Cabauw with its high number of animal husbandry and the use of  
17 nitrogen containing fertilizers around the CESAR tower. Banzhaf et al. (2013) derived similar  
18 conclusions for  $\text{PM}_{10}$  using different emission scenarios within domains covering Germany  
19 and Europe. Applying a 3d chemical transport model over Europe Megaritis et al. (2014)  
20 found that a reduction of  $\text{NH}_3$  emissions by 50% would have a much higher effect on  
21 reducing  $\text{PM}_{2.5}$  than decreasing  $\text{NO}_x$  emissions by 50%. The latter scenario would even result  
22 in negative side effects such as higher tropospheric ozone concentrations (especially in  
23 summertime 4% over Western Europe and up to 40% in major urban areas) and higher  
24 amounts of particulate sulfate and OA by 8% and 12%, respectively, in winter.

25 The local mitigation of organic aerosol mass (29% contribution on average) is more  
26 challenging, as secondary organic aerosols are highly abundant at the Cabauw site (74% and  
27 22% of OA and total  $\text{PM}_1$  on average, respectively). The presented data set shows a large and  
28 ubiquitous HULIS fraction (37%) which based on diurnal patterns and a lack of correlation  
29 with wind direction can be considered as long-range background aerosol formed from  
30 atmospheric aging processes. In turn, primary organic aerosols emitted mainly from traffic  
31 and biomass burning (12% and 13% of OA on average) have only minor importance. For a  
32 more detailed identification of the SOA sources compound specific measurements of OA as  
33 well as routine VOC monitoring are needed.



1 Finally, the presented data set and interpretations provide an important contribution to the  
2 EU-FP7-ACTRIS project which supported building of new knowledge as well as policy  
3 issues on climate change, air quality, and long-range transport of pollutants on a European  
4 scale.”

5

6 *Figures:*

7 *16) Figure S2: Wind speed is hard to see. Please make extra graph.*

8 **Response:**

9 As suggested by the reviewer, the wind speed is now shown in a separate graph in Fig. S2 in  
10 the revised supplement material

11

12 *17) Figure S11: Mention in figure caption that MARGA data are PM1.*

13 **Response:**

14 In the revised supplement material the figure caption was changed to:

15 “Figure S11: Correlation graphs of Chl, NH<sub>4</sub>, SO<sub>4</sub>, NO<sub>3</sub>, and total inorganic mass  
16 concentrations from ACSM and MARGA PM<sub>1</sub> data”

17

## 1 **References**

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20

21

1 **Responses to anonymous referee #3**

2 *Received and published: 23<sup>th</sup> March 2016*

3 We thank the reviewer for the careful review of our manuscript; the comments and  
4 suggestions are greatly appreciated. All the comments have been addressed. In the following  
5 please find our responses to the comments one by one and the corresponding changes made to  
6 the manuscript. The original comments are shown in italics.

7

8 *1) This paper is well written. Figures are clear and provide important information. The*  
9 *supporting information is also well presented. The source apportionment study is very well*  
10 *conducted in terms of methodology and scientific interpretation. Out of this, the scientific*  
11 *interest for this paper remains (too) limited with no clear addedvalue compare to previous*  
12 *AMS studies performed at Cabauw. Providing a 1-year continuous observation of NR-PM1*  
13 *with Q-ACSM (with source apportionment) is not any more self-sufficient for a scientific*  
14 *paper. Scientific interpretation is often set here at the minimum with poor perspectives; lack*  
15 *of comparison neither with previous studies related to Cabauw nor Q-ACSM measurements*  
16 *and source apportionment studies. The scientific motivation is not convincing enough. Air*  
17 *Quality issues related to PM are important in densely populated regions (with high exposure*  
18 *to particulate pollution), much less in a rural area like Cabauw. In this AQ context, it would*  
19 *have been more meaningful to use this background (Cabauw) PM dataset to interpret the*  
20 *contribution of regional pollution to PM in Dutch cities. You missed the point that this dataset*  
21 *remains an important contribution to a larger one obtained at the European scale within*  
22 *theEU-FP7-ACTRIS project (ACTRIS Q-ACSM network).*

23 **Response:**

24 As responded to a similar comment from referee #1, it is correct that a publication just dealing  
25 with AMS or ACSM measurements as well as the presentation of their data sets are not novel  
26 enough for ACP, also considering that two AMS campaigns were previously performed in  
27 Cabauw. We emphasize here that the focus of this work extends beyond the presentation of a  
28 larger data set for this site:

29 In this study, we showed that the major fraction of atmospheric aerosol at this site is  
30 comprised of secondary aerosol (SA) which is chemically formed in the atmosphere. SA also  
31 dominates the high mass loadings during the pollution episodes greatly exceeding air quality  
32 standards. The long term aerosol mass spectrometric data provided by this work is important

1 for the understanding of the chemical processes leading to such high aerosol masses which are  
2 unexpected for a rural site like Cabauw. This information is necessary for potential mitigation  
3 of particle masses. The findings presented here were not seen in previous studies due to their  
4 limited time of sampling.

5 It is true that Cabauw is categorized as rural. However one should note that this term is not  
6 indicative that is has not significant impact on the population. The Netherlands is the 24<sup>th</sup>  
7 most densely populated country in the world (average population density 502 persons per  
8 km<sup>2</sup>, UN (2015)), exceeding e.g. population densities of India (391) or Japan (335). The  
9 measurement site is located in the part of the municipality of Lopik which lies within or near  
10 the three most populated provinces: Noord-Holland, Zuid-Holland and Utrecht (976, 1227,  
11 851 persons per km<sup>2</sup>). The population of the statistical district Cabauw, covering the village  
12 center and the farms along the canal, has a population of around 690 (CBS, 2004). This  
13 number is indeed small, but on the other hand the centers of the four largest and most  
14 populated cities of the Netherlands are located within a surrounding of maximum 40 km air-  
15 line from the CESAR tower (Amsterdam and The Hague ca. 40 km, Rotterdam 25 km, and  
16 Utrecht 20 km), covering around 2.3 million inhabitants. Pandis et al. (2013) stated that high  
17 organic mass loadings found in rural sites clearly influence the background of urban areas, as  
18 mentioned by the reviewer. Skyllakou et al. (2014) applied the particulate matter source  
19 apportionment technology (PSAT) together with PMCAMx, a regional chemical transport  
20 model for the Megacity of Paris. They showed that only 13% of the PM<sub>2.5</sub> fraction originated  
21 from the urban agglomeration (an area of around 50 km around the city center), 36% coming  
22 from mid-range (50–500 km from the center of the Paris) sources and 51% from long range  
23 transport (more than 500 km from Paris). In this context the high aerosol mass loadings found  
24 in this work have a strong impact on the air quality not only for the area close to the  
25 measurement site but also for a huge number of inhabitants within a surrounding of less than  
26 40 km, justifying the comparison with the WHO air quality limits.

27 To emphasize the influence of the findings from this site to urban areas we added the  
28 following sentence to first paragraph of the conclusions chapter:

29 “While few people live in the direct vicinity of the measurement site, the high aerosol  
30 concentration measured at the site can be considered to represent the regional background.  
31 This regional background is adding to local aerosol contributions in high populated urban  
32 sites (Pandis et al., 2013), namely the 4 largest cities of the Netherlands which have a distance  
33 of 40 km or less from the CESAR tower.”

1 We thank the reviewer to point out that this work is also an important contribution within the  
2 framework of ACTRIS. As suggested by the reviewer we added the following paragraph into  
3 the conclusion chapter:

4 “Finally, the presented data set and interpretations provide an important contribution to the  
5 EU-FP7-ACTRIS project which supported building of new knowledge as well as policy  
6 issues on climate change, air quality, and long-range transport of pollutants on a European  
7 scale.”

8

9 **Specific Comments:**

10 2) *Hyphen is often missing (gas-to-particle, long-term, time-resolved, water-soluble, etc)*

11 **Response:**

12 Fortunately, copy-editing is standard in ACP. Thus minor errors will be eliminated later on.

13

14 3) *Page 35119, line 10: Why are you focusing only on WHO and not on EU-regulated*  
15 *PM2.5? Because PM2.5 at Cabauw may exceed WHO AQ guidelines? (and not EUregulated*  
16 *PM2.5)*

17 **Response:**

18 The air quality guideline of the European Union (EU) only provides a “target value” for the  
19 annual PM<sub>2.5</sub> limit of 25 µg m<sup>-3</sup> (EU, 2008), which is indeed higher than the respective WHO  
20 limit. As the WHO gives additionally a daily mean limit the conclusions of this work with  
21 respect to air quality could be aimed with a much higher time resolution. It could be shown  
22 which conditions, compounds and sources are responsible for high particle mass loading  
23 during times of exceedances. That is why the authors focused on the WHO limits.

24 Furthermore, the WHO limits are based on the now extensive body of scientific evidence  
25 relating to air pollution and its health consequences. Although this information base has gaps  
26 and uncertainties, it offers a strong foundation for the recommended guidelines (WHO, 2006).  
27 EU standards in contrast vary according to the approach adopted for balancing health risks,  
28 technological feasibility, economic considerations and various other political and social  
29 factors. This in turn will depend on, among other things, the level of development and  
30 national capability in air quality management in each member state. Since it is not in the  
31 scope of the manuscript to discuss these issues in detail, the authors decided to just provide

1 the target value for the annual PM<sub>2.5</sub> limit according to the EU in the introduction chapter of  
2 the revised manuscript as follows:

3 “The European Union Air Quality Directive 2008/50/EC provides only a target value of the  
4 annual mean limit of PM<sub>2.5</sub> of 25 µg m<sup>-3</sup> (EU, 2008).”

5

6 4) Page 35120, line 16: it should be NR-PM1 (and not ambient aerosol).

7 **Response:**

8 In the revised manuscript, this part was changed to:

9 “The Aerosol Mass Spectrometer (AMS, Aerodyne Research Inc., Billerica, MA, USA) is a  
10 powerful instrument to quantitatively measure the chemical composition of the Non-  
11 Refractory fraction of PM<sub>1</sub> (NR-PM<sub>1</sub>) with high time and mass resolution.”

12

13 5) Page 35120, line 25: Please state that it is a Q-ACSM (Q=Quadripole).

14 **Response:**

15 In the revised manuscript, this part was changed to:

16 “The Quadrupol-ACSM (Q-ACSM) is specially designed for long-term continuous and real-  
17 time measurements of mass concentrations and composition of NR-PM<sub>1</sub> species (Ng et al.,  
18 2011).”

19

20 6) Page 35122, line 20: Remove Fröhlich et al. (2015) which is focused on Source  
21 apportionment intercomparison only.

22 **Response:**

23 In the revised manuscript, the reference Fröhlich et al. (2015) was removed.

24

25 7) Page 35123, line 23: Did you install a dryer at the inlet? If so, please specify. Otherwise,  
26 you should address the impact of sampling aerosols at ambient RH.

27 **Response:**

28 On page 35123, line 23 in the manuscript, there is no description of any inlet so far. On page  
29 35124, line 14 and page 35125, line 17 the nafion dryers for the ACSM and for the



1 MAAP/SMPS inlet are mentioned, respectively. As described in line 6 on page 35126, the  
2 MARGA samples at ambient RH.

3

4 8) Page 35124, line 15: 3LPM is supposed to be the nominal flow in the Q-ACSM for  
5 isokinetic sampling. Why did you use 1LPM. Impact?

6 **Response:**

7 The flow rate of 1 liter per minute at this part of the inlet was set due to a restriction of the  
8 nafion dryer. At higher flow rates this dryer may not dry the sample air sufficient enough.  
9 Since the length of this inlet section was rather small (less than 2 m) we don't expect a  
10 significant impact, e.g. enhanced wall losses, comparing to a flow of 3 liter per minute.

11

12 9) Page 35124, 17: "The" before "following"

13 **Response:**

14 In the revised manuscript, this part was changed to:

15 "The following collocated aerosol instruments were used for cross-validation of the ACSM  
16 data:"

17

18 10) Page 35125, line 1-2: MAAP provides absorption data and uncalibrated BC. You did not  
19 mention here how you address this issue.

20 **Response:**

21 In this work, the factory default inversion and calibration was used to determine the eBC  
22 concentrations from MAAP data. We agree that this information should be added in the  
23 revised manuscript. In addition to other changes due to comments from other referees, the  
24 MAAP description was changed as follows:

25 "The MAAP instrument has been introduced by Petzold and Schönlinner (2004) and Petzold  
26 et al. (2005). It is designed for the determination of the black carbon (BC), which is a product  
27 of incomplete combustion. There is in the scientific community a general consensus over  
28 what black carbon is in terms of properties (Bond et al., 2013) The MAAP measures the  
29 strong visible light absorption property of BC by simultaneous measurements of the radiation  
30 penetrating through and scattered back from a particle-loaded fiber filter. According to

1 Petzold et al. (2013), optical BC determined by MAAP is to be referred to as equivalent black  
2 carbon (eBC). One property of BC is that it is highly refractory with a vaporization  
3 temperature near 4000K (Schwarz et al., 2006), thus BC is not vaporized at 600°C and cannot  
4 be measured by the ACSM. The MAAP achieves a time resolution of 5 minutes with an  
5 uncertainty of 12% (Petzold and Schönlinner, 2004). A mass absorption cross section (MAC)  
6 of  $6.6 \text{ m}^2 \text{ g}^{-1}$  for a wavelength of 637 nm (Müller et al., 2011) was chosen to convert the  
7 measured particle absorption coefficient to eBC mass concentrations.”

8  
9 *11) Page 35125, line 23: The number here (33% of particle loss) is very high and not address*  
10 *correctly in the paper. It should be a function of size. Brownian diffusion (and associated*  
11 *losses) may be an important issue for number concentration in the nanometer range; not sure*  
12 *it is an important issue for PM calculation using SMPS. More information should be provided*  
13 *here.*

14 **Response:**

15 As responded to a similar comment from referee #1, we used now the particle density  
16 deriving from the chemical composition not as the campaign average but time resolved for  
17 each data point in the revised manuscript. In addition we introduced SMPS data which is now  
18 size dependently corrected as published by Henzing (2011) The description of the sampling  
19 losses of the 60 m inlet for eBC given in the manuscript derived from a series of  
20 measurements at the Cabauw tower performed in a previous campaign. Unfortunately these  
21 results are not published yet. To clarify how these losses were determined in the revised  
22 manuscript and to account for a comment from referee #2, the loss description was changed as  
23 follows:

24 “SMPS data was corrected size dependently for (diffusional) losses in the inlet system and  
25 SMPS system itself according to (Henzing, 2011) who compared theoretical findings with  
26 measured losses that are obtained by measuring simultaneously before and after the various  
27 parts of the inlet system at the CESAR tower. In addition, particles of different compositions  
28 were measured in 2013 simultaneously at the pipe entrance at 60 m height and in the  
29 basement (J. S. Henzing, personal communication). For more than 8000 simultaneous  
30 observations, the results showed that aerosol measurements through this 60 m sampling line  
31 underestimate  $\text{PM}_{10}$ -eBC by approximately 33% with an uncertainty of 7%. Therefore, eBC  
32 obtained from the MAAP are divided by a factor of 0.66 to account for these losses. For the  
33 inorganic species penetrations through this inlet line were reported to be 62-73% for nitrate,

1 55-64% for sulfate, and 54-56% for ammonium. However these results were not used for  
2 corrections in this work”

3 Please note that these losses influenced only data acquired by the MAAP and SMPS. Since  
4 the contribution of eBC is rather low (average: 5%) a potential overall error for total aerosol  
5 masses is low and would not significantly alter one quintessence of the paper, namely total  
6 mass concentrations above the air quality limits.

7 As a consequence of the newly evaluated SMPS data the correlation values between  
8 ACSM+MAAP data with SMPS data changed as seen in Fig. S3 and S4 in the revised  
9 supplement. Nevertheless, the overall qualitative and quantitative agreement is still given  
10 except that the ACSM+MAAP data is now overestimating the total PM<sub>1</sub> mass by 16%,  
11 excluding the eBC data the ACSM overestimates total mass by 12%. As seen in Fig. S3 the  
12 difference between both systems is significantly higher during the pollution events 16 to 27  
13 January 2013 and 5 to 8 May 2013. Since the quantitative agreement with the MARGA is  
14 much higher at these times the discrepancy to the SMPS is likely due to the fact that the losses  
15 within the 60 m inlet could not be corrected for individual species as mentioned above.  
16 Therefore the following paragraph was added at the end of the cross validation chapter in the  
17 revised manuscript:

18 “Major discrepancies to the SMPS especially during some of the pollution events like 16 to 27  
19 January 2013 and 5 to 8 May 2013 (see below) can be explained by the correction of losses  
20 through the 60 m inlet line which was done size dependently and did not account for losses of  
21 individual species as mentioned in chapter 2.3. As the quantitative agreements of individual  
22 inorganic species as well as of total inorganics between the ACSM and the MARGA during  
23 these periods are much higher, the mass loadings determined from these instruments are more  
24 reliable than the SMPS data.”

25

26 12) Page 35126, line 23: “Source apportionment : : : was performed” (and not “were  
27 performed”).

28 **Response:**

29 In the revised manuscript, this part was changed to:

30 “Source apportionment of organic aerosol components was performed using Positive Matrix  
31 Factorization...”

32

1 13) Page 35127, line 23: BBOA is not 100% POA. But if you want to go in this direction, you  
2 may specify here “pBBOA” (primary BBOA).

3 **Response:**

4 We agree with the reviewer that some secondary organic aerosols originate from biomass  
5 burning. Nevertheless, BBOA is related to the profiles found by PMF analysis using a BBOA  
6 mass spectrum found by Crippa et al. (2014) at the same site. In this and a number of other  
7 publications BBOA factors as found by PMF explorations from AMS or ACSM  
8 measurements are described as POA (Canonaco et al., 2015; Elser et al., 2016; Fröhlich et al.,  
9 2015; Ripoll et al., 2015; Sun et al., 2012) as well as this particular section of this manuscript  
10 does. This is mainly due to the fact that this factor is usually partly identified by the  
11 correlation with external primary emitted tracers like BC or CO. That is why BBOA as found  
12 by the PMF analysis using ACSM data is considered here as POA.

13

14 14) Page 35129, line 2: You want to address here risk assessment (WHO air quality  
15 guideline); but for a rural site having few inhabitants. This is not very relevant.

16 **Response:**

17 We hope that this comment is answered in detail by the responds to the referee’s comments  
18 #1 and #3.

19

20 15) Page 35129, line 7: You state here that MARGA PM<sub>2.5</sub> is calculated as the sum of  
21 chemical compounds in the aerosol phase (NO<sub>3</sub>, NH<sub>4</sub>, SO<sub>4</sub>, Cl, Na, K, Mg, Ca) AND in the  
22 gas phase (SO<sub>2</sub>, HCl, HNO<sub>3</sub>, HNO<sub>2</sub>). Are you sure? (if so MARGA PM<sub>2.5</sub> data is not  
23 correct).

24 **Response:**

25 We thank the reviewer for pointing out this mistake as well as referee #1 did in a similar  
26 comment. The gas phase compounds SO<sub>2</sub>, HNO<sub>2</sub>, HNO<sub>3</sub> and HCl were mistakenly added to  
27 total aerosol masses. Nevertheless, the conclusions deriving from these calculations did not  
28 change. In the revised manuscript, the resulting time series “MARGA PM<sub>2.5</sub> + ACSM-Org +  
29 eBC” in Fig. 1 was exchanged and the sentence was changed to:

30 **Nevertheless, the average total mass derived from the combination of the MARGA PM<sub>2.5</sub> data**  
31 **(including all water soluble inorganic components NO<sub>3</sub>, NH<sub>4</sub>, SO<sub>4</sub>, Cl, Na, K, Mg, and Ca),**

1 MAAP eBC and ACSM PM<sub>1</sub> organics resulted in 12.5 μg m<sup>-3</sup> during this time, clearly  
2 exceeding the WHO PM<sub>2.5</sub> annual mean limit (10 μg m<sup>-3</sup>) by 25%.

3 The reader may now notice that the concentration values of MARGA PM<sub>2.5</sub> + MAAP eBC +  
4 ACSM PM<sub>1</sub> Organics (light blue time series in Fig. 1) decreased reasonably because the  
5 mentioned gas phase data was excluded. On the other hand, more data points are now seen.  
6 This is due to the fact that data coverage of the MARGA PM<sub>2.5</sub> data is better than for the  
7 MARGA gas phase data.

8

9 *16) Page 35129 : You have 30-min time resolution data and you interpret seasonal variability*  
10 *(averaging data over typically 3 months). Why don't you go for monthly mean variability?*  
11 *You may better see some trends here with 12 points instead of 4.*

12 **Response:**

13 The presentation of 3-months-averaged data was mainly due to the season-wisely separation  
14 of the data set prior to PMF analysis. Furthermore, some months have larger lacks of data  
15 coverage either from the ACSM or MAAP, especially in periods 3 and 4. That is why it was  
16 more representative to average the data over 2-3 months than for each month. We agree with  
17 the reviewer that this sentence is misleading since there was no actual further analysis  
18 performed using 3-months-averaged data of the aerosol species apart from the calculation of  
19 the pie charts in Fig. 2a. The variation of each species across the whole campaign can be  
20 much better observed by the 30 minutes resolution data shown in Fig. 2b. Therefore this part  
21 was changed in the revised manuscript to:

22 **“To determine properly average species contributions for each pie chart the campaign was**  
23 **therefore not divided strictly season-wise, but into five periods.”**

24 In addition, Table S2 shows now the contribution and total mass concentrations averaged over  
25 each month instead over the whole season/period. Another column was introduced to show  
26 the number of data points acquired each month and used for averaging.

27

28 *17) Page 35130, line 10: You state here that you may observe some inversions a low altitudes*  
29 *(typically below 60m). How often? Then you would assume that concentrations below/above*  
30 *the inversion layer are disconnected. If so, how are you dealing with SMPS and BC data*  
31 *which are sampled at 60m height?*

1 **Response:**

2 We agree with the reviewer that a temperature inversion between the inlet of the ACSM (at  
3 5 m) and the inlet of the MAAP and SMPS (at 60 m) would infer that the instruments sample  
4 from different atmospheric layers. In fact, such distinct inversions between the temperature  
5 sensors at 2 m and 80 m height happened only few times across the whole campaign (in total  
6 18 times with a duration of around 1-6 hours) and during periods with aerosol mass loadings  
7 not higher than  $10 \mu\text{g m}^{-3}$ . An exception is the inversion of the morning on 25 January  
8 between 3 and 9 am where total mass concentrations of around  $20 \mu\text{g m}^{-3}$  from both SMPS  
9 and ACSM+MAAP was determined, meaning that there no significant difference was seen  
10 from both instruments. On the other hand it is true that in some other cases the total mass  
11 concentrations of the SMPS and the ACSM differ from each other. But these differences are  
12 not higher than in some periods without a temperature inversion. That means that if the data  
13 points during temperature inversions would be cancelled out, it would not alter the correlation  
14 results from both instruments significantly as well as the main conclusions of this work.  
15 Therefore, the SMPS and BC data was untreated with respect to possible samplings from  
16 different atmospheric layers. In the morning of 16 January the temperature inversion is not  
17 relevant since there is no ACSM data existing, but is only beginning from the evening of 16  
18 January. We admit that the current mention of this morning in the manuscript is unnecessary  
19 and misleading. Therefore it was deleted from this part in the revised manuscript to:

20 “The period 16 to 27 January 2013 showed also the lowest temperatures (average:  $-4^{\circ}\text{C}$ ) with  
21 respect to the whole campaign and a temperature inversion between 2 m and 40 m height in  
22 the morning 25 January 2013, where both instruments showed total mass concentrations of ca.  
23  $20 \mu\text{g m}^{-3}$ .”

24

25 *18) Page 35130, line 25: You are located at a rural site and you report diurnal BC variations*  
26 *with maxima at rush hours (evening/morning). What does it mean? You are locally influenced*  
27 *by traffic? Are you sure?*

28 **Response:**

29 As mentioned in the respond to the referee’s comments #1, this site is categorized as rural but  
30 should not be considered as very low dense populated. Thus it has significant influence from  
31 anthropogenic sources including traffic. With a low air-line distance of just 600 m from the  
32 CESAR Tower there is the nearest expressway (Dutch: Autoweg) N210, which connects the  
33 cities of Rotterdam and IJsselstein and is only allowed for vehicles with a velocity of at least

1 50 km h<sup>-1</sup>. Beside that the highway (Dutch: Autosnelweg) A27 has a distance of just 6 km in  
2 addition to the highly frequented highways A2 and A12, which have a distance of around  
3 10 km each. Together with the observed clear traffic related diurnal pattern of the HOA factor  
4 and NO<sub>x</sub> data, the authors are confident that this site is influenced by traffic emission.

5

6 *19) Page 35131: the discussion on organic nitrates is not convincing at all. You show in SI*  
7 *that you have strong differences between SO<sub>4</sub> (and NO<sub>3</sub>) between ACSM and MARGA. For*  
8 *me, it clearly shows that IENO<sub>3</sub>, RIESO<sub>4</sub> and RIENH<sub>4</sub> are not properly determined. As a*  
9 *result, I am not surprised to see that NH<sub>4</sub> (measured vs predicted) is not matching well. And*  
10 *for that reason, I am not convinced that the difference is due to organic nitrates.*

11 **Response:**

12 While comparisons between the ACSM and MARGA data show discrepancies which are  
13 interpreted in the manuscript, we cannot see how these could be explained based on wrong  
14 RIE's. From the way OrgNO<sub>3</sub> is derived (subtraction of MARGA inorganic nitrate from  
15 ACSM total nitrate) it is independent of RIE's. If the RIE<sub>NH<sub>4</sub></sub> were wrong, it would be wrong  
16 by a constant factor. The authors see no way how a time dependent correction to the total  
17 ACSM nitrate (which the OrgNO<sub>3</sub> is) would "happen by chance" to arrive at a closed ion  
18 balance.

19 As responded to a similar comment by referee #1, the presented determination of organic  
20 nitrates, which is in agreement with the procedure of Xu et al. (2015), improved the  
21 agreement of measured against predicted ACSM-NH<sub>4</sub>, resulting in a nearly 1:1 regression line  
22 without a significant offset.

23 This is in agreement with the MARGA internal ion balance which also indicates neutralized  
24 inorganic aerosols. All evidence together makes a strong case that the difference of AMS total  
25 nitrate to MARGA nitrate is most likely due to the presence of organic nitrates. These  
26 findings and the previously reported relatively high AMS organic nitrate fractions by Mensah  
27 (2011) in May 2008 (0.5 μg m<sup>-3</sup>, 35% of total nitrate, 5.2% of total aerosol mass) and March  
28 2009 (0.2 μg m<sup>-3</sup>, 10% of total nitrate, 3.6% of total aerosol mass) in Cabauw show strong  
29 evidence of the presence of organic nitrates during this campaign and the reliability of its  
30 estimation as presented in the manuscript.

31 To provide a more detailed description this part was changed in the revised manuscript as  
32 follows:

1 “This assumption is acceptable, as shown by using the MARGA-NO<sub>3</sub> instead of the ACSM-  
2 total-NO<sub>3</sub> for the ion balance of ACSM data (including ACSM-SO<sub>4</sub>, -Chl and -NH<sub>4</sub>),  
3 following a procedure given by Xu et al. (2015) who calculated the organic nitrate fraction by  
4 subtracting the inorganic nitrate concentrations measured by a particle-into-liquid sampler  
5 (PILS, see Orsini et al. (2003)) from ToF-AMS total nitrate concentrations. In the Cabauw  
6 data set, the correlation of measured against predicted NH<sub>4</sub> resulted in a nearly 1:1 regression  
7 line without a significant offset (Fig. S10). This is in agreement with the MARGA internal ion  
8 balance which also indicates neutralized inorganic aerosols. Therefore, the mass concentration  
9 of nitrate groups associated with organic molecules (hereafter called organic nitrate or  
10 OrgNO<sub>3</sub>), can be estimated by subtracting the MARGA-nitrate from the ACSM-nitrate  
11 concentration. The OrgNO<sub>3</sub> time series using this approach is plotted in Fig. S11, the  
12 respective diurnal variation averaged over the whole campaign in Fig. S12. An average mass  
13 fraction of 9% was calculated for OrgNO<sub>3</sub> (average concentration: 0.43 μg m<sup>-3</sup>) in respect to  
14 total ACSM-NO<sub>3</sub>. The organic nitrate fraction shows a maximum concentration in the night,  
15 followed by a decrease during the day. These findings are in agreement with previously  
16 reported relatively high AMS organic nitrate fractions by Mensah (2011) in May 2008  
17 (0.5 μg m<sup>-3</sup>, 35% of total nitrate, 5.2% of total aerosol mass) and March 2009 (0.2 μg m<sup>-3</sup>,  
18 10% of total nitrate, 3.6% of total aerosol mass) in Cabauw.”

19  
20 20) Page 35133, line 25: You completely skip sea salt chloride (measured by MARGA and not  
21 by ACSM). Please revise your conclusions here which are not correct.

22 **Response:**

23 Sea salt chloride was meant to be included in the term “inorganic chloride”. In the manuscript  
24 it is described as component which can be detected well by the MARGA but not by the  
25 ACSM with a sufficient sensitivity as mentioned by the reviewer. That means that to our best  
26 knowledge our conclusions remain correct with respect to the inter-comparison between  
27 MARGA- and ACSM-chloride.

28 For clarification, this part was changed in the revised manuscript to:

29 “It should also be noted, that chloride concentrations can originate from particulate organic  
30 and inorganic chloride components originating from e.g. sea salt. For the latter, the ACSM is  
31 much less sensitive than the MARGA. As described above, influences from sea salt can be  
32 considered rather low. In turn, the MARGA might be less sensitive to organic chlorides, as  
33 they are likely less water soluble than inorganic chlorides.”



1 The changes refer now to an added discussion about a possible influence of sea salt and dust  
2 particles as suggested by referee #1 to the section of the inter-comparison between the SMPS  
3 and ACSM+BC:

4 “The negative offset can be explained by minor influences of sea salt and dust particles,  
5 which can be detected well by the SMPS and MARGA but not by the ACSM with a sufficient  
6 sensitivity. But the low value of the intercept shows already that the uncertainty introduced by  
7 these aerosol components is rather low in general. This can also be explained by the low  
8 concentrations of Mg, Na, K and Ca as measured by the MARGA (see below) and the  
9 assumption that the majority of dust particles is most likely found in particles with diameters  
10 larger than 1 or even 2.5  $\mu\text{m}$  (Finlayson-Pitts and Pitts (2000) and references therein).”

11  
12 21) Page 35134, line 3: “PM<sub>2.5</sub> values were up to 33% higher than PM<sub>1</sub>”. This is a very  
13 interesting statement that points out the lack of efficiency of Q-ACSM to properly characterize  
14 PM<sub>2.5</sub> pollution events. This should be highlighted at least when addressing these pollution  
15 events. How far is the Q-ACSM from MARGA PM<sub>2.5</sub> during these episodes? Consequences  
16 on the conclusions of the paper?

17 **Response:**

18 The reviewer is right that the ACSM is not able to properly characterize PM<sub>2.5</sub> pollution  
19 events due to its inlet design. The differences between the data sets are implicated in Fig. 1,  
20 where MARGA PM<sub>2.5</sub> data. This is why it is considered as PM<sub>1</sub> instrument as stated in the  
21 manuscript. We also pointed out that even the presented PM<sub>1</sub> data exceeded the WHO PM<sub>2.5</sub>  
22 daily mean limit several times and approached the respective annual mean limit. In our  
23 opinion the conclusions of the paper do not change due to the reasonable underestimation of  
24 the ACSM concentrations with respect to PM<sub>2.5</sub>. But to emphasize that this data represents the  
25 lower limits and to highlight the importance of the ACSM measurements in this context, we  
26 changed the description of the pollution events in the revised manuscript as follows:

27 “The most significant pollution events (17 to 21 August 2012, 21 to 25 October 2012, 16 to  
28 27 January 2013, and 5 to 8 May 2013) are highlighted with shaded backgrounds in Fig. 2.  
29 During these events, MARGA PM<sub>2.5</sub> data showed up to 33% higher (e.g. on 17 to 21 August  
30 2012) total inorganic masses than ACSM PM<sub>1</sub> inorganics as implicated in Fig. 1. As  
31 mentioned above the number and proportions of exceedances of the WHO PM<sub>2.5</sub> daily mean  
32 limit is therefore even higher than when only PM<sub>1</sub> data is considered. During the last three  
33 events, northerly and north-easterly winds dominated.”

1 Furthermore, we added a paragraph into the conclusion chapter when addressing the  
2 exceedances. The first paragraph is now written as follows:

3 “This work provides chemical composition data of atmospheric aerosols acquired during one  
4 year at the CESAR tower in Cabauw, the Netherlands, which is a representative rural site for  
5 North Western Europe. The concentration of submicron particles from combined ACSM and  
6 MAAP data showed 12 exceedances from the WHO PM<sub>2.5</sub> daily mean limit. The respective  
7 campaign average of 9.5 µg m<sup>-3</sup> approached the WHO PM<sub>2.5</sub> annual mean limit. Taking  
8 MARGA PM<sub>2.5</sub> into account the number and proportions of these exceedances are even  
9 higher, emphasizing the importance of these high ACSM PM<sub>1</sub> results shown here which  
10 represent lower limits. As carbonaceous compounds are estimated to be five times more toxic  
11 than inorganic particles (Lelieveld et al., 2015) MARGA data alone would not give  
12 sufficiently possible implications regarding adverse health effects.”

13

14 22) Page 35136; line 18: *I am a little bit loss. You are using BC and CO as external tracers*  
15 *to constrain both HOA and BBOA. Based on HOA and BBOA concentration levels and using*  
16 *literature data, you may be able to give a range of BC originating from both sources. I am*  
17 *pretty sure you will find that BC is coming mainly from traffic and thus cannot be used to*  
18 *constrain BBOA as done in the paper.*

19 **Response:**

20 We think that there is a misunderstanding regarding the use of BC data in this work due to a  
21 misleading expression in the manuscript. Source apportionment was only performed for  
22 organic aerosols, not for CO and BC data. HOA and BBOA profiles were constrained using  
23 reference mass spectra from the literature. The resulting time series were then compared with  
24 the external tracers CO and BC to verify their origins as primary organic aerosols. The  
25 respective sentence in the manuscript should not imply that in Winter these tracers are mainly  
26 emitted by biomass burning comparing to traffic emission but the contribution from biomass  
27 burning to these tracer concentrations are higher in Winter than from biomass burning  
28 averaged over the whole campaign. This is a result of reasonably enhanced amount of  
29 biomass burning due to enhanced domestic heating in winter. To clarify this issue, the  
30 respective part was changed in the revised manuscript as follows:

31 “In Winter, the correlations with eBC and CO were higher ( $R^2 = 0.64$  and  $0.57$ , respectively)  
32 than over the whole campaign ( $R^2 = 0.39$  and  $0.49$ , respectively), meaning that these  
33 compounds are reasonably more attributed to domestic heating during the colder periods

1 comparing to the contribution of heating to BC and CO during the other seasons in this  
2 region.”

3

4 23) Page 35137, line 19: It should be  $R^2=0.39$  for  $NO_3$  (not 0.47).

5 **Response:**

6 In the revised manuscript, this part was changed to:

7 “Also the comparison to the most important tracers (Table S4 in the supplement) showed no  
8 preferential attribution, either to a low-volatile ( $R^2 = 0.41$  with  $SO_4$ ), semi-volatile ( $R^2 = 0.39$   
9 with  $NO_3$ ) or to primary organic aerosol ( $R^2 = 0.47$  with eBC).”

10

11 24) Page 35138, line 15: BC is not POA!

12 **Response:**

13 We thank the reviewer for pointing out this mistake. In the revised manuscript the term “POA  
14 components” was replaced by the term “primary aerosol components”

15

16 25) Page 35140, line 6: Why do you want to speak about hygroscopicity ? It has never been  
17 addressed before in the paper and has nothing to do with AQ-regulated PM.

18 **Response:**

19 It is true that hygroscopicity was not discussed as an aerosol property throughout the  
20 manuscript. Here it was mentioned to highlight the importance of the findings with regard to  
21 adverse health effects of aerosols, as the manuscript focuses on the WHO air quality  
22 guidelines. To clarify and emphasize this issue the respective part was changed to:

23 “It is shown that particulate ammonium nitrate is the major aerosol component (39% on  
24 average) and represents the more hygroscopic aerosol fraction. Since the human respiratory  
25 system is water vapor saturated more hygroscopic aerosols have a higher deposition tendency  
26 within the human lung than less water soluble particle compounds (Asgharian, 2004; Broday  
27 and Georgopoulos, 2001). With special regard to adverse health effects this is very crucial  
28 because Asgharian (2004) also found that especially hygroscopic submicron particles can  
29 deposit in the entire lung. The high ammonium nitrate fraction also implies that inorganic SA

1 reduction in Cabauw can be most efficiently achieved through the reduction of gaseous  
2 ammonia emissions in the area.”

3

4 26) Page 35140, line 23: *I think P. Croteau is from Aerodyne Research Inc., not from PSI.*

5 **Response:**

6 In the revised manuscript, this part was changed to:

7 “We also thank Philip Croteau (Aerodyne Research) for his support during the measurements  
8 regarding the data acquisition and evaluation.”

9

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## 1 **Responses to anonymous referee #4**

2 *Received and published: 23<sup>th</sup> March 2016*

3 We thank the reviewer for the careful review of our manuscript; the comments and  
4 suggestions are greatly appreciated. All the comments have been addressed. In the following  
5 please find our responses to the comments one by one and the corresponding changes made to  
6 the manuscript. The original comments are shown in italics.

7

### 8 **Specific Comments:**

9

10 *1) P35123, L27: Contrary to what is implied, the Middlebrook parameterisation does take*  
11 *account of the nitrate mass fraction. More justification should be given in the main article on*  
12 *why it was not suitable here because it would be preferable to use the SMPS data for an*  
13 *independent validation, rather than informing the CE.*

### 14 **Response:**

15 The SMPS was sampling from the inlet located at 60 m height while the ACSM sampled from  
16 the roof at 5 m height, resulting in a higher uncertainty comparing to the usual set-up where  
17 both instruments would sampling from inlets close together. In addition and as responded to  
18 the similar comment from referee #1, the application of the Middlebrook algorithm lead to  
19 unreasonable discontinuities of CE value. The algorithm used in the work originated from the  
20 comparison of a ToF-AMS with the same SMPS, both sampling from the 60 m-inlet, during a  
21 campaign in Cabauw in May 2008 (Mensah et al., 2012). But the authors agree that the  
22 current explanation is misleading. In the revised manuscript the respective part was changed  
23 to:

24 **“In contrast to the commonly used constant value of 0.5 this CE correction accounts for the**  
25 **high ammonium nitrate mass fraction (ANMF) found at this site and is thus more suitable for**  
26 **the data presented here. Another algorithm for composition dependent CE determination**  
27 **(Middlebrook et al., 2012) was also tested for its validity. It uses a threshold ratio of measured**  
28 **to predicted NH<sub>4</sub> to switch between two different equations to determine the CE. The**  
29 **threshold value of 0.75 is close to the observed ratio of measured over predicted NH<sub>4</sub> of this**  
30 **data set, resulting in large discontinuities of CE values and in consequence, discontinuous**  
31 **changes in aerosol mass concentrations. In other words, the Middlebrook algorithm is not**



1 suitable for data sets showing at the same time low ratios of measured to predicted NH<sub>4</sub> and  
2 high AMNF's."

3 2) P34124, L25: Was the factory default inversion and calibration of the MAAP used? If  
4 so, this should be specified.

5 **Response:**

6 Indeed, the factory default inversion and calibration was used to determine the eBC  
7 concentrations from MAAP data. We agree that this information should be added in the  
8 revised manuscript. In addition to other changes due to comments from other referees, the  
9 MAAP description was changed as follows:

10 "The MAAP instrument has been introduced by Petzold and Schönlinner (2004) and Petzold  
11 et al. (2005). It is designed for the determination of the black carbon (BC), which is a product  
12 of incomplete combustion. There is in the scientific community a general consensus over  
13 what black carbon is in terms of properties (Bond et al., 2013) The MAAP measures the  
14 strong visible light absorption property of BC by simultaneous measurements of the radiation  
15 penetrating through and scattered back from a particle-loaded fiber filter. According to  
16 Petzold et al. (2013), optical BC determined by MAAP is to be referred to as equivalent black  
17 carbon (eBC). One property of BC is that it is highly refractory with a vaporization  
18 temperature near 4000K (Schwarz et al., 2006), thus BC is not vaporized at 600°C and cannot  
19 be measured by the ACSM. The MAAP achieves a time resolution of 5 minutes with an  
20 uncertainty of 12% (Petzold and Schönlinner, 2004). The eBC mass concentration was  
21 determined using the factory default inversion and calibration."

22

23 3) P35125, L3: The model numbers of the SPMS and CPC should be given.

24 **Response:**

25 The SMPS used in this work is actually a combination of a DMA and a CPC in one  
26 instrument. We agree that the model number of this instrument should be given. In the revised  
27 manuscript, this part was therefore changed to:

28 "The SMPS (TSI, Model 3034) is a sequential combination of several integrated components:  
29 an impactor, a neutralizer, a differential mobility analyzer and a condensation particle counter.  
30 It determines the size distribution of particles in a range of 10 nm to 487 nm (electro-mobility  
31 diameter)."

1 4) P35125, L23: More detail should be given regarding how the losses down the inlet pipe  
2 were calculated, given the magnitude of the correction. In particular, if diffusional losses  
3 were significant, whether this correction should be size-dependent should be commented on.

4 **Response:**

5 As responded to a similar comment from referee #1, we used now the particle density  
6 deriving from the chemical composition not as the campaign average but time resolved for  
7 each data point in the revised manuscript. In addition we introduced SMPS data which is now  
8 size dependently corrected as published by Henzing (2011) The description of the sampling  
9 losses of the 60 m inlet for eBC given in the manuscript derived from a series of  
10 measurements at the Cabauw tower performed in a previous campaign. Unfortunately these  
11 results are not published yet. To clarify how these losses were determined in the revised  
12 manuscript and to account for a comment from referee #2, the loss description was changed as  
13 follows:

14 “SMPS data was corrected size dependently for (diffusional) losses in the inlet system and  
15 SMPS system itself according to (Henzing, 2011) who compared theoretical findings with  
16 measured losses that are obtained by measuring simultaneously before and after the various  
17 parts of the inlet system at the CESAR tower. In addition, particles of different compositions  
18 were measured in 2013 simultaneously at the pipe entrance at 60 m height and in the  
19 basement (J. S. Henzing, personal communication). For more than 8000 simultaneous  
20 observations, the results showed that aerosol measurements through this 60 m sampling line  
21 underestimate PM<sub>10</sub>-eBC by approximately 33% with an uncertainty of 7%. Therefore, eBC  
22 obtained from the MAAP are divided by a factor of 0.66 to account for these losses. For the  
23 inorganic species penetrations through this inlet line were reported to be 62-73% for nitrate,  
24 55-64% for sulfate, and 54-56% for ammonium. However these results were not used for  
25 corrections in this work”

26 Please note that these losses influenced only data acquired by the MAAP and SMPS. Since  
27 the contribution of eBC is rather low (average: 5%) a potential overall error for total aerosol  
28 masses is low and would not significantly alter one quintessence of the paper, namely total  
29 mass concentrations above the air quality limits. We think theoretical particle loss calculations  
30 would not add additional information in the context of this paper.

31 As a consequence of the newly evaluated SMPS data the correlation values between  
32 ACSM+MAAP data with SMPS data changed as seen in Fig. S3 and S4 in the revised  
33 supplement. Nevertheless, the overall qualitative and quantitative agreement is still given

1 except that the ACSM+MAAP data is now overestimating the total PM<sub>1</sub> mass by 16%,  
2 excluding the eBC data the ACSM overestimates total mass by 12%. As seen in Fig. S3 the  
3 difference between both systems is significantly higher during the pollution events 16 to 27  
4 January 2013 and 5 to 8 May 2013. Since the quantitative agreement with the MARGA is  
5 much higher at these times the discrepancy to the SMPS is likely due to the fact that the losses  
6 within the 60 m inlet could not be corrected for individual species as mentioned above.  
7 Therefore the following paragraph was added at the end of the cross validation chapter in the  
8 revised manuscript:

9 “Major discrepancies to the SMPS especially during some of the pollution events like 16 to 27  
10 January 2013 and 5 to 8 May 2013 (see below) can be explained by the correction of losses  
11 through the 60 m inlet line which was done size dependently and did not account for losses of  
12 individual species as mentioned in chapter 2.3. As the quantitative agreements of individual  
13 inorganic species as well as of total inorganics between the ACSM and the MARGA during  
14 these periods are much higher, the mass loadings determined from these instruments are more  
15 reliable than the SMPS data.”

16

17 5) P35126, L13: Polyethylene is not a conductive polymer, so electrostatic losses of particles  
18 should be expected. Has this inlet line been characterised for this?

19 **Response:**

20 As responded to the similar comment from referee #1 a polyethylene (PE) tube can indeed  
21 potentially enhance wall losses of particles comparing to stainless steel tubes. The MARGA-  
22 inlet system at the Cabauw tower as used for this study was previously described by Schaap et  
23 al. (2011). It actually did not only consist of PE tubes but of a series of components reducing  
24 particle losses. They investigated wall losses on a similar system and found only minor  
25 concentration losses for several compounds of 2% and less. A more detailed description was  
26 added in the revised manuscript:

27 “The sample air was transferred into the instrument within a polyethylene („Polyflo“) tube  
28 with an inner diameter of 0.5” (= 1.27 cm) and a sample flow of 16.7 L min<sup>-1</sup>, which is either  
29 directed through a PM<sub>1</sub> or a PM<sub>2,5</sub> size selective head. A detailed description of the MARGA  
30 inlet system at the Cabauw tower was previously described by Schaap et al. (2011). There,  
31 wall losses were investigated and found to be less than 2% for several gaseous and particulate  
32 compounds.”

1 6) P35126, L15: *The method of size selection (e.g. impaction, cyclone) should be specified.*

2 **Response:**

3 The size selection was done using size selective cyclones. We agree that this should be  
4 specified. This part was changed in the revised manuscript to:

5 “The sample air was transferred into the instrument within a polyethylene („Polyflo“) tube  
6 with an inner diameter of 0.5” (= 1.27 cm) and a sample flow of 16.7 L min<sup>-1</sup>, which is either  
7 directed through a PM<sub>1</sub> or a PM<sub>2.5</sub> size selective cyclone.”

8

9 7) P35133, L3: *I don't see how the sulphate comparison can be regarded as “high*  
10 *quantitative agreement” given that it is of the order of 50% out. Given that historically,*  
11 *comparisons regarding sulphate generally tend to be quite favourable, this is quite surprising.*  
12 *It is also a little worrying that the ACSM measures more than both the AMS and the MARGA.*  
13 *The authors should investigate this further.*

14 **Response:**

15 The “very high qualitative and quantitative agreements” is just referring to the comparison of  
16 the total inorganic masses from both instruments as shown in the last plot of Fig. S11 (which  
17 changed to Fig. S5 in the revised supplement). The authors agree that the current description  
18 is misleading because the word “agreements” is written in the plural form. In the revised  
19 manuscript, this word was changed to its singular form “agreement”.

20 As seen by the last plots in Fig. S11 and S12 (which changed to Fig. S5 and S6, respectively,  
21 in the revised supplement), the ACSM actually measures the same or even little less than the  
22 MARGA and AMS, respectively, in terms of total concentrations (slopes of the regression  
23 lines are 1.05 and 0.90, respectively). The ACSM showed higher concentrations only in case  
24 of nitrate. This issue and the discrepancies regarding other aerosol species is explained in  
25 detail in the manuscript and in the responds to similar comments from referees #1 and #2.

26

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36 carbon and light-scattering aerosols from the boundary layer to the lower stratosphere. *Journal*  
37 *of Geophysical Research: Atmospheres* 111, n/a-n/a.

38

39

1 **Aerosol source apportionment from 1-year-measurements**  
2 **at the CESAR tower at Cabauw, NL**

3

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## 1 **Abstract**

2 Intensive measurements of submicron aerosol particles and their chemical composition were  
3 performed with an Aerosol Chemical Speciation Monitor (ACSM) at the Cabauw  
4 Experimental Site for Atmospheric Research (CESAR) in Cabauw, NL, **sampling at 5 m**  
5 **height above ground.** The campaign lasted nearly one year from July 2012 to June 2013 as  
6 part of the **EU-FP7-ACTRIS project (Q-ACSM Network).** Including equivalent black carbon  
7 an average particulate mass concentration of  $9.50 \mu\text{g m}^{-3}$  was obtained during the whole  
8 campaign with dominant contributions from ammonium nitrate (45%), organic aerosol (OA,  
9 29%), and ammonium sulfate (19%). 12 exceedances of the World Health Organization  
10 (WHO)  $\text{PM}_{2.5}$  daily mean limit ( $25 \mu\text{g m}^{-3}$ ) were observed at this rural site using  $\text{PM}_{10}$   
11 instrumentation only. Ammonium nitrate and OA represented the largest contributors to total  
12 particulate matter during periods of exceedance.

13 Source apportionment of OA was performed season-wise by Positive Matrix Factorization  
14 (PMF) using the Multilinear Engine 2 (ME-2) controlled via the source finder (SoFi). Primary  
15 organic aerosols were attributed mainly to traffic (8% - 16% contribution to total OA,  
16 averaged season-wise) and biomass burning (0% - 23%). Secondary organic aerosols (SOA,  
17 61% - 84%) dominated the organic fraction during the whole campaign, particularly on days  
18 with high mass loadings. A SOA factor which is attributed to humic-like substances (HULIS)  
19 was identified as a highly oxidized background aerosol in Cabauw. This shows the  
20 importance of atmospheric ageing processes for aerosol concentration at this rural site. Due to  
21 the large secondary fraction, the reduction of particulate mass at this rural site is challenging  
22 on a local scale.

# 1 1 Introduction

2 Atmospheric aerosols have large impacts on the climate directly by scattering and absorbing  
3 short wave radiation. Besides the resulting influence on the visibility (Ramanathan et al.,  
4 2007; Romanou et al., 2007), this can have a cooling or heating effect on the atmosphere  
5 (IPCC, 2013). The indirect effect refers to the impact of particles on cloud formation and their  
6 properties.

7 In addition, particles can impact adversely on human health by e.g. increasing the probability  
8 of cardiopulmonary and lung cancer mortality (Pope et al., 2002). The World Health  
9 Organization (WHO) recently estimated that globally, 3.7 million deaths were attributable to  
10 ambient air pollution in both cities and rural areas in 2012 (EU, 2008). This mortality is  
11 reported to be due to exposure to small particulate matter (PM<sub>10</sub>), which can cause  
12 cardiovascular and respiratory disease, and cancers. Particles with lower diameters such as  
13 PM<sub>2.5</sub> or PM<sub>1</sub> are reported to have enhanced toxicological effects since they can deposit more  
14 deeply in the respiratory system and remain suspended for longer periods of time (Pope and  
15 Dockery, 2006). Therefore, a number of institutions established several air quality standards  
16 for different particle sizes to limit aerosol mass. The WHO air quality guideline (global  
17 update 2005, WHO (2006)) defines a PM<sub>2.5</sub> daily mean limit of 25 µg m<sup>-3</sup> and a PM<sub>2.5</sub> annual  
18 mean limit of 10 µg m<sup>-3</sup>. **The European Union Air Quality Directive 2008/50/EC provides**  
19 **only a target value of the annual mean limit of PM<sub>2.5</sub> of 25 µg m<sup>-3</sup> (EU, 2008).**

20 Air quality and climate effects are not only depending on the particle number concentration  
21 and size, but also on their chemical composition. This information is not only relevant to  
22 investigate the nature and magnitude of each effect, but also for the identification and  
23 quantification of aerosol sources and mitigation strategies for a potential reduction of aerosol  
24 mass concentrations. Major inorganic components of PM<sub>1</sub> consist mainly of ammonium  
25 nitrate (NH<sub>4</sub>NO<sub>3</sub>) and ammonium sulfate ((NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>), formed in the presence of ammonia  
26 (NH<sub>3</sub>), nitrogen oxides (NO<sub>x</sub> = NO + NO<sub>2</sub>) and sulfur dioxide (SO<sub>2</sub>), respectively (Seinfeld  
27 and Pandis, 2006). Therefore NH<sub>4</sub>NO<sub>3</sub> and (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> are strongly attributed to  
28 anthropogenic sources (Finlayson-Pitts and Pitts, 2000). Since the reactions leading to  
29 inorganic aerosol happen from gaseous precursors in the atmosphere, the condensed products  
30 are considered as secondary aerosols, while primary aerosols like black carbon (BC) are  
31 emitted directly.

32 In contrast, organic aerosols (OA), which can also be of primary (POA) or secondary (SOA)  
33 origin, consist of up to hundreds of thousands of different molecules (Goldstein and Galbally,



1 2007), where SOA contributes on average 70% to organic aerosol mass (Hallquist et al.,  
2 2009). SOA is formed by gas to particle conversion of atmospherically oxidized semi- and  
3 low-volatile organic compounds (VOC's). Guenther et al. (1995) estimated a global VOC  
4 budget in the order of 1150 Tg carbon per year. Biogenic VOC's (BVOC's) contribute  
5 approximately 90% of total VOC, including isoprene (50% of total BVOC's), monoterpenes  
6 (15%), and sesquiterpenes (3%) (Guenther et al., 2012). In turn, 10% of emitted VOC's are of  
7 anthropogenic origin, including e.g. alkanes, alkenes, benzene and toluene.

8 The investigation of the aerosol composition is critical for the development of climate models,  
9 since the composition influences important particle properties. However, its determination is  
10 still challenging, especially in case of OA, which contribute significantly to atmospheric  
11 particulate matter (Jimenez et al., 2009). In fact, the lack of knowledge on particle  
12 composition is a key contribution to the large uncertainty for the determination of the total  
13 anthropogenic radiative forcing (IPCC, 2013).

14 The development of online aerosol mass spectrometric techniques during the last decades  
15 enhanced the possibilities to investigate aerosol chemical composition in real-time (DeCarlo  
16 et al., 2006; Jayne et al., 2000; Jimenez et al., 2003). The Aerosol Mass Spectrometer (AMS,  
17 Aerodyne Research Inc., Billerica, MA, USA) is a powerful instrument to quantitatively  
18 measure the chemical composition of the non-refractory fraction of PM<sub>1</sub> (NR-PM<sub>1</sub>) with high  
19 time and mass resolution. Due to the high amount of maintenance from skilled and trained  
20 personnel needed for continuous operating measurement campaigns using an AMS are usually  
21 not exceeding one or two months (Sun et al., 2012). Since the variation of aerosol  
22 composition is very high depending on measurement site and season (Jimenez et al., 2009),  
23 long term measurements ( $\geq 1$  year) are clearly needed. At a European level, this effort is  
24 supported by the Aerosols, Clouds and Traces gases Research InfraStructure network  
25 (ACTRIS) program that aims at pooling high-quality data from state-of-the-art  
26 instrumentation such as the Aerodyne Aerosol Chemical Speciation Monitor (ACSM). The  
27 Quadrupol-ACSM (Q-ACSM) is specially designed for long-term continuous and real-time  
28 measurements of mass concentrations and composition of NR-PM<sub>1</sub> species (Ng et al., 2011b).

29 In this study, an ACSM was used to measure the submicron aerosol chemical composition  
30 from 11.07.2012 to 03.06.2013 at the CESAR tower in Cabauw, NL, as part of the EU-FP7-  
31 ACTRIS project. A collocated Multi-Angle Absorption Photometer (MAAP, Thermo  
32 Scientific Model 5012), provided equivalent black carbon (eBC) data. Organic aerosol data  
33 was further analyzed by Positive Matrix Factorization using the Multilinear Engine 2

1 (Paatero, 1999) via the source finder (SoFi, Canonaco et al. (2013)). This data set shows the  
2 long-term variability of particle composition and is used for source apportionment of  
3 atmospheric aerosols at this North Western European rural site, with the focus on periods  
4 where air quality standards were violated. This information can be further used to establish  
5 strategies for the reduction of particulate matter.

## 1   **2   Methodology**

### 2   **2.1   Site description: CESAR**

3   The CESAR tower is 220 m high and managed and operated by the Royal Netherlands  
4   Meteorological Institute (KNMI, The Netherlands). It is located in a rural site (51.970°N,  
5   4.926°E) near Cabauw, the Netherlands, about 20 km south-west of the city of Utrecht and  
6   about 45 km south east of the Dutch North Sea coast. The site conditions are typical for North  
7   Western Europe. They can either be maritime or continental, depending on the wind direction.  
8   The surface elevation changes in the surrounding are at most a few meters over 20 km. The  
9   tower ground is approximately 0.7 m below sea level, the diurnal variation of the temperature  
10   is relatively stable (Vermeulen et al., 2011). The direct surroundings of the tower have a  
11   relatively low population density. The nearby region is used mainly by agriculture, with a  
12   mixture of intensively and extensively managed grassland. These are used also for animal  
13   keepings like cattle and sheep, besides nearby located chicken farms.

14   The tower is equipped with external platforms and booms at 2 m, 10 m, 20 m, 40 m, 80 m,  
15   140 m, and 200 m. At all these levels, meteorological observations of standard parameters like  
16   wind speed, wind direction, dew point temperature, and ambient temperature are routinely  
17   performed (Ulden and Wieringa, 1996). These data sets are available at the CESAR data base  
18   (KNMI, 2013). **In addition, other meteorological data like precipitation, radiation and remote**  
19   **sensing, including lidar, radar and radiometer techniques, are acquired at the tower and**  
20   **submitted to the CESAR data base.** Also concentrations of gaseous compounds, including  
21   greenhouse gases, are monitored at CESAR (Russchenberg et al., 2005; Vermeulen et al.,  
22   2011). The National Institute for Public Health and the Environment (RIVM, the Netherlands)  
23   provides hourly data from gaseous CO, NO, NO<sub>2</sub>, SO, SO<sub>2</sub>, and O<sub>3</sub>, but also daily PM<sub>10</sub>  
24   measurements. Data of ambient CO<sub>2</sub>, CH<sub>4</sub> (both measured at 20 m, 60 m, 120 m, and 200 m  
25   height), and Radon 222 (<sup>222</sup>Rn, measured at 20 m and 200 m height) is determined by ECN on  
26   a time scale of 30 minutes. A map indicating the land use of the Netherlands is given in the  
27   supplement (Fig. S1).

28

## 1 2.2 ACSM sampling and data analysis

2 The ACSM measures the NR-PM<sub>1</sub> fraction, including the organic fraction (Org), ammonium  
3 (NH<sub>4</sub>), nitrate (NO<sub>3</sub>), sulfate (SO<sub>4</sub>), and chloride (Chl), using mass spectrometric information  
4 on a time base of approximately 30 minutes. The ACSM used in this work is equipped with a  
5 commercial grade Quadrupole Residual Gas Analyzer (RGA), thus it is also called a Q-  
6 ACSM. This instrument is described in detail by Ng et al. (2011b). The RGA provides unit  
7 mass resolution (UMR) mass spectra. Fractions of measured UMR signals were assigned to  
8 individual aerosol species using the fragmentation table introduced by Allan et al. (2004). Due  
9 to the automated zeroing system used for the ACSM, adjustments of the fragmentation table  
10 entries concerning interferences from air beam molecules are not needed. All ACSM data  
11 processing and analysis within this work was performed using software provided by  
12 Aerodyne Research (ACSM Local, version 1.531, ARI (2012) ) within IGOR Pro version  
13 6.2.3.

14 Recent studies showed good agreements of ACSM data with data from other aerosol  
15 instruments like the High Resolution Time-of-Flight AMS (HR-ToF-AMS) in ambient  
16 measurement campaigns, even in highly polluted areas such as Beijing (Sun et al., 2012; Sun  
17 et al., 2011). An intercomparison of two collocated ACSM's resulted in strong correlations  
18 ( $R^2 = 0.8$ ) and agreements within 27% (Budisulistiorini et al., 2014). Crenn et al. (2015)  
19 reported similar results from the intercomparison of 13 ACSM's, a ToF-ACSM's, a ToF-  
20 AMS, and other collocated instruments in the region of Paris, France. There, the same ACSM  
21 instrument (S/N A140-145) as the one used for this study was tested. Those results indicate  
22 that the ACSM can be used as a suitable and cost-effective alternative to the AMS for aerosol  
23 composition measurements due to its capability of stable and reproducible operation.

24 Mass calibrations were performed approximately every month and were based on determining  
25 the instrument response factor (RF) (Ng et al., 2011b) using monodisperse NH<sub>4</sub>NO<sub>3</sub> (320 nm)  
26 as calibration substance and a Condensation Particle Counter (CPC, TSI 3022a) as reference  
27 instrument (Jayne et al., 2000; Jimenez et al., 2003). An average RF<sub>NO<sub>3</sub></sub> of  $2.74 \pm 0.45 \cdot 10^{-11}$   
28 was obtained and used for the calculation of aerosol mass concentrations. Instead of  
29 performing a mass calibration for every aerosol species, relative ionization efficiencies  
30 (RIE's), compared to that of nitrate, were used. The RIE's of NH<sub>4</sub> and SO<sub>4</sub> were determined  
31 directly during the mass calibrations by measuring dry NH<sub>4</sub>NO<sub>3</sub> and (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> particles after  
32 another. Averaged over all calibrations, a RIE<sub>NH<sub>4</sub></sub> of  $7.53 \pm 0.21$  and a RIE<sub>SO<sub>4</sub></sub> of  $0.81 \pm 0.10$   
33 were found and used for the whole data set. Calibration results gained from this particular

1 instrument during an intercomparison in Paris as described by Crenn et al. (2015) were within  
2 the uncertainty or only slightly different ( $RF_{NO_3}=2.34 \cdot 10^{-11}$ ,  $RIE_{NH_4}=6.54$ ,  $RIE_{SO_4}=0.62$ )  
3 considering the large differences between individual instruments at the intercomparison. RIE  
4 values of 1.4 and 1.3 for organics and chloride, respectively, were taken from the literature  
5 (Alfarra et al., 2004; Canagaratna et al., 2007).

6 A site specific, time resolved particle collection efficiency (CE) correction algorithm  
7 (equations are given in the supplement) was applied, which was developed by Mensah et al.  
8 (2012), using SMPS data as reference. In contrast to the commonly used constant value of 0.5  
9 this CE correction accounts for the high ammonium nitrate mass fraction (ANMF) found at  
10 this site and is thus more suitable for the data presented here. Another algorithm for  
11 composition dependent CE determination (Middlebrook et al., 2012) was also tested for its  
12 validity. It uses a threshold ratio of measured to predicted  $NH_4$  to switch between two  
13 different equations to determine the CE. The threshold value of 0.75 is close to the observed  
14 ratio of measured over predicted  $NH_4$  of this data set, resulting in large discontinuities of CE  
15 values and in consequence, discontinuous changes in aerosol mass concentrations. In other  
16 words, the Middlebrook algorithm is not suitable for data sets showing at the same time low  
17 ratios of measured to predicted  $NH_4$  and high AMNF's. According to Ng et al. (2011b) and  
18 Sun et al. (2012), the variability of the instrument performance was corrected based on the  
19 inlet pressure and  $N_2$  signal, as well as the signals derived from the internal naphthalene  
20 source were taken to correct for the mass dependent ion transmission efficiency of the RGA.

21 The detection limits for each species were not determined within this work, but are reported to  
22 be 0.148, 0.284, 0.012, 0.024, and  $0.011 \mu g m^{-3}$  ( $3\sigma$ ) for organics, ammonium, nitrate, sulfate,  
23 and chloride, respectively, at an averaging time of 30 minutes (Ng et al., 2011b).

24 During the whole campaign, the ACSM was located inside the CESAR tower building. Its  
25 inlet was placed on the roof of the building at approximately 5 m height above ground. The  
26 inlet head was equipped with a  $PM_{2.5}$  cyclone (URG-2000-30EN, URG Corporation, Chapel  
27 Hill, USA). The sample air was pulled through a 10 m copper line (ID = 6.5 mm) at a flow  
28 rate of  $9 L min^{-1}$ . From this flow, a subsample of  $1 L min^{-1}$  was diverted to a Nafion dryer  
29 (RH < 40%) of which approximately  $80 mL min^{-1}$  entered the ACSM. This resulted in a total  
30 residence time of 18 seconds.

31

## 1 2.3 Collocated aerosol measurements

2 The following collocated aerosol instruments were used for cross-validation of the ACSM  
3 data: (i) a Scanning Mobility Particle Sizer (SMPS, TSI 3034), operated by the Netherlands  
4 Organization for Applied Scientific Research (TNO, The Netherlands), (ii) a Monitor for  
5 Aerosol and Gases (MARGA, Applikon Analytical BV), operated by ECN, and (iii) a HR-  
6 ToF-AMS, which was operated by Forschungszentrum Juelich during the first 6 days of the  
7 ACSM campaign. In addition, BC data obtained by a MAAP instrument (TNO, The  
8 Netherlands) was included into the analysis.

9 The MAAP instrument has been introduced by Petzold and Schönlinner (2004) and Petzold et  
10 al. (2005). It is designed for the determination of the black carbon (BC), which is a product of  
11 incomplete combustion. There is in the scientific community a general consensus over what  
12 black carbon is in terms of properties (Bond et al., 2013) The MAAP measures the strong  
13 visible light absorption property of BC by simultaneous measurements of the radiation  
14 penetrating through and scattered back from a particle-loaded fiber filter. According to  
15 Petzold et al. (2013), optical BC determined by MAAP is to be referred to as equivalent black  
16 carbon (eBC). One property of BC is that it is highly refractory with a vaporization  
17 temperature near 4000K (Schwarz et al., 2006), thus BC is not vaporized at 600°C and cannot  
18 be measured by the ACSM. The MAAP achieves a time resolution of 5 minutes with an  
19 uncertainty of 12% (Petzold and Schönlinner, 2004). A mass absorption cross section (MAC)  
20 of  $6.6 \text{ m}^2 \text{ g}^{-1}$  for a wavelength of 637 nm (Müller et al., 2011) was chosen to convert the  
21 measured particle absorption coefficient to eBC mass concentrations. Although the MAAP  
22 has no size selective inlet beside the  $\text{PM}_{10}$  heads described below, it can be assumed that eBC-  
23 containing aerosol generally fall into the submicron size range (Bond et al., 2013). Thus eBC  
24 mass concentrations are considered as part of the  $\text{PM}_1$  fraction from hereon. As seen later the  
25 eBC fraction is rather low throughout the campaign, meaning that the overall error of this  
26 assumption is not significant.

27 The SMPS (TSI, Model 3034) is a sequential combination of several integrated components:  
28 an impactor, a neutralizer, a differential mobility analyser and a condensation particle counter.  
29 It determines the size distribution of particles in a range of 10 nm to 487 nm (electromobility  
30 diameter). The SMPS aerosol mass concentration was calculated from the measured volume  
31 distributions using the particle density determined by the aerosol composition information  
32 derived from the ACSM and the MAAP. Assuming spherical particles, the total density is  
33 computed by using the densities of the aerosol species, weighted by their mass fractions. Bulk

1 densities of  $\text{NH}_4\text{NO}_3$  ( $1.72 \text{ g cm}^{-3}$ ) and  $(\text{NH}_4)_2\text{SO}_4$  ( $1.77 \text{ g cm}^{-3}$ ), and densities of  $1.8 \text{ g cm}^{-3}$   
2 for BC (Bond and Bergstrom, 2006; Park et al., 2004) of organics ( $1.4 \text{ g cm}^{-3}$ , Hallquist et al.  
3 (2009)) were taken into account. Considering its low influence on the total particle density at  
4 this site, it is acceptable to set the density for chloride to  $1 \text{ g cm}^{-3}$  (Mensah et al., 2012).

5 During the presented campaign, the MAAP and the SMPS were connected to the common  
6 aerosol inlet which sampled at 60 m height. This inlet consisted of four  $\text{PM}_{10}$  size selective  
7 heads at the top, followed by a Nafion dryer to keep the relative humidity (RH) of the sample  
8 air below 40%. The stainless steel pipe, ranging from the aperture at 60 m to the basement,  
9 has an inner diameter of 0.5" (= 1.27 cm) and ends in a manifold, where the sampled air is  
10 distributed to a variety of different instruments, including the MAAP and the SMPS, each  
11 with its own sample flow. An overall sample flow of  $60 \text{ L min}^{-1}$  was adjusted inside the 60 m  
12 pipe, assuring laminar conditions.

13 SMPS data was corrected size dependently for (diffusional) losses in the inlet system and  
14 SMPS system itself according to Henzing (2011) who compared theoretical findings with  
15 measured losses that are obtained by measuring simultaneously before and after the various  
16 parts of the inlet system at the CESAR tower. In addition, particles of different compositions  
17 were measured in 2013 simultaneously at the pipe entrance at 60 m height and in the  
18 basement (J. S. Henzing, personal communication). For more than 8000 simultaneous  
19 observations, the results showed that aerosol measurements through this 60 m sampling line  
20 underestimate  $\text{PM}_{10}\text{-eBC}$  by approximately 33% with an uncertainty of 7%. Therefore, eBC  
21 obtained from the MAAP are divided by a factor of 0.66 to account for these losses. For the  
22 inorganic species penetrations through this inlet line were reported to be 62-73% for nitrate,  
23 55-64% for sulfate, and 54-56% for ammonium. However these results were not used for  
24 corrections in this work.

25 The MARGA measures water soluble components of both the gas and aerosol phase  
26 simultaneously, including the aerosol species nitrate, ammonium, sulfate, and chloride. It uses  
27 a Wet Annular Denuder (WAD) followed by a Steam Jet Aerosol Collector (SJAC). The  
28 operational and calibration procedures, as well as details of the data analysis are given by  
29 Trebs et al. (2004). During the campaign the MARGA collected alternately the  $\text{PM}_1$  and  
30  $\text{PM}_{2.5}$  fraction of ambient particles at ambient RH and temperature, each fraction sampled  
31 hourly with a measurement error smaller than 10% (Schaap et al., 2011). The detection limits  
32 for the most abundant aerosol species were 0.05, 0.10, 0.08,  $0.01 \mu\text{g m}^{-3}$  ( $4\sigma$ , Rumsey et al.  
33 (2014)) for ammonium, nitrate, sulfate, and chloride, respectively.

1 The MARGA inlet was equipped with a PM<sub>10</sub> size selective head (Rupprecht and Pataschnick,  
2 R&P), placed on the roof of the tower building next to the ACSM inlet aperture at 5 m height.  
3 The sample air was transferred into the instrument within a polyethylene („Polyflo“) tube with  
4 an inner diameter of 0.5” (= 1.27 cm) and a sample flow of 16.7 L min<sup>-1</sup>, which is either  
5 directed through a PM<sub>1</sub> or a PM<sub>2.5</sub> size selective cyclone. A detailed description of the  
6 MARGA inlet system at the Cabauw tower was previously described by Schaap et al. (2011).  
7 There, wall losses were investigated and found to be less than 2% for several gaseous and  
8 particulate compounds. To compare with the ACSM, only MARGA data containing PM<sub>1</sub>  
9 composition is considered within this work.

10 An Aerodyne HR-ToF-AMS was connected to the MARGA inlet during the first 6 days of the  
11 campaign. As the ACSM is built upon AMS, the latter measures likewise the chemical  
12 composition of the PM<sub>1</sub> non-refractory (NR) fraction. Instrumental details are available in  
13 Jayne et al. (2000), Jimenez et al. (2003), and DeCarlo et al. (2006).

14

## 15 **2.4 Statistical methods of organic aerosol data analysis**

16 Source apportionment of organic aerosol components was performed using Positive Matrix  
17 Factorization (PMF, Paatero (1997); Paatero and Tapper (1994)) via the ME-2 solver  
18 (Paatero, 1999). PMF is a bilinear model and assumes that the original data set, containing  
19 variable mass spectra over time, is a linear combination of a given number of factors, each  
20 with a constant mass spectrum and its contributions over time. It has been successfully used in  
21 AMS ambient studies apportioning the measured organic mass spectra in terms of  
22 source/process-related components (Zhang et al., 2011). With the ME-2 solver it is possible to  
23 introduce a priori mass spectral information and hence to reduce the rotational ambiguity, i.e.  
24 similar PMF results with the same goodness of fit, of PMF solutions (Paatero and Hopke,  
25 2003).

26 The extraction of OA data and error matrices as mass concentrations in  $\mu\text{g m}^{-3}$  over time, as  
27 well as their preparation for PMF/ME-2 according to Ulbrich et al. (2009), was done within  
28 the ACSM software. Only  $m/z$ 's  $\leq 100$  were considered here since they represented nearly the  
29 whole OA mass (around 98%) and did not interfere with ion fragments originating from  
30 naphthalene (e.g.,  $m/z$  127, 128, and 129, see also Sec. 2.2). From these matrices, the  $m/z$  12  
31 was discarded because it showed negative signals, probably due to too short delay time of the  
32 quadrupole scan (125 ms) after a valve switch (Fröhlich et al., 2015). In addition, the  $m/z$ 's 37



1 and 38 were also removed from the organic matrices of the whole campaign except for winter  
2 2013. This was done because the signal at these masses showed high interferences with **the**  
3 **chloride related ions  $^{37}\text{Cl}$  and  $\text{H}^{37}\text{Cl}$** . Including these ions lead to unreasonable PMF factors  
4 which mainly contained only these two masses and represented the chloride time series,  
5 whereas during Winter 2013 no such interferences were observed.

6 The interface source finder (SoFi, Canonaco et al. (2013)), version 4.8, was used to control  
7 ME-2 for the PMF runs of the ACSM OA data adopting the source apportionment strategy  
8 developed by Crippa et al. (2014). Briefly, unconstrained PMF runs were first investigated  
9 with 1 to 10 factors and a moderate number of seeds (10 to 15) for each factor number. If  
10 Primary Organic Aerosol (POA) factor profiles like hydrocarbon-like OA (HOA) or OA from  
11 biomass burning (BBOA) were found, site specific POA mass spectra or spectra derived from  
12 the data base were constrained and a sensitivity analysis performed on the tightness of  
13 constraint (a-value, Lanz et al. (2008)). Since aged OA (or oxidized OA, OOA) factors show  
14 more variability between measurement sites in terms of their mass spectra (MS) than POA, it  
15 is not appropriate to constrain SOA factor profiles using reference spectra derived from  
16 different locations (Canonaco et al., 2015). According to Crippa et al. (2014), HOA MS  
17 should be more constrained (a-value between 0 and 0.2) than BBOA MS ( $a \approx 0.3$ ), since the  
18 BBOA fingerprint shows higher variations depending on the burning type and conditions  
19 (Alfarra et al., 2007). In each step, either in unconstrained or constrained approaches, the  
20 following criteria for finding a proper solution were used based on the recommendations from  
21 Ulbrich et al. (2009) and Canonaco et al. (2013):

- 22 • The quality parameter  $Q/Q_{\text{exp}}$  was minimized.
- 23 • Factor profiles have reasonable mass spectra, as expected for the measurement site.
- 24 • Factor time series have high correlations with respective external data sets such as  
25 gaseous  $\text{CO}$ ,  $\text{CO}_2$ ,  $\text{NO}_x$ , and particulate nitrate, sulfate and black carbon.
- 26 • When a proper solution is found, 50 seed runs were used to find the global minimum  
27 for  $Q/Q_{\text{exp}}$ .
- 28 • Investigation of the rotational ambiguity of the solution space is carried out using the  
29 a-value approach for the constrained factor profiles

30

## 1 **3 Results and discussion**

### 2 **3.1 Cross-validation of particulate total mass and chemical species** 3 **concentrations**

4 The particle density during the ACSM campaign was determined using the chemical  
5 composition data from the ACSM and the MAAP and resulted in an average of  $1.63 \pm$   
6  $0.12 \text{ g cm}^{-3}$ . The time series of the particle density is given in the supplement. It was used to  
7 calculate the SMPS total mass concentration from its measured volume concentration  
8 throughout the campaign. Due to the relatively low signal-to-noise ratio of the ACSM, the  
9 density shows scattering only during periods with low mass loadings. Therefore it does not  
10 influence the cross-validation with the SMPS mass significantly. Figures S3 and S4 show the  
11 time series of the SMPS mass and the combined mass concentrations measured by ACSM and  
12 MAAP and the correlation plot of both data sets, respectively. Using 12275 common data  
13 points for the linear fit, a good qualitative and quantitative agreement (Slope:  $1.16 \pm 0.01$ ,  
14 intercept:  $-1.05 \pm 0.06$ ,  $R^2 = 0.78$ ) was observed. Excluding eBC data resulted in a slope of  
15  $1.13 \pm 0.01$ , an intercept of  $-1.14 \pm 0.06$ , and a  $R^2$  of 0.78. The negative offset can be  
16 explained by minor influences of sea salt and dust particles, which can be detected well by the  
17 SMPS and MARGA but not by the ACSM with a sufficient sensitivity. But the low value of  
18 the intercept shows already that the uncertainty introduced by these aerosol components is  
19 rather low in general. This can also be explained by the low concentrations of Mg, Na, K and  
20 Ca as measured by the MARGA (see below) and the assumption that the majority of dust  
21 particles is most likely found in particles with diameters larger than 1 or even  $2.5 \mu\text{m}$   
22 (Finlayson-Pitts and Pitts (2000) and references therein).

23 Since the MARGA measures routinely the water soluble inorganic aerosol compounds, data  
24 from ACSM inorganic species were synchronized and compared to corresponding MARGA  
25  $\text{PM}_{10}$  data for the whole measurement period. The temporal overlap with the collocated high  
26 resolution AMS was between 11 and 17 July 2012. The correlation parameters of individual  
27 aerosol species and respective total masses between the ACSM data and the data sets from the  
28 MARGA and AMS are given in Table 1, using 1943 and 294 common data points,  
29 respectively. The corresponding correlation graphs are shown in the supplement (Fig. S5 and  
30 S6). Except for chloride, high correlation coefficients were achieved. Furthermore, the  
31 comparison to both total mass time evolutions shows very high qualitative and quantitative  
32 agreement. The quantitative difference between ACSM- and AMS-organics is also very low,  
33 and the discrepancies in case of ammonium and nitrate are within the stated  $\pm 30\%$  accuracy

1 of the AMS and ACSM (Ng et al., 2011b) and the  $\pm 10\%$  for the MARGA-NO<sub>3</sub>, respectively  
2 (Makkonen et al., 2012). Similar variations were also found by Crenn et al. (2015) as well as  
3 Budisulistiorini et al. (2014). The latter reported of a comparison between two collocated  
4 ACSM's ( $\pm 27\%$ ,  $R^2 = 0.21$  for Chl,  $R^2 > 0.8$  for the other species) and between these  
5 ACSM's and a continuous Tapered Element Oscillating Microbalance (TEOM, PM<sub>2.5</sub>)  
6 instrument. Note that the major ions used for nitrate detection in AMS and ACSM (NO<sup>+</sup> and  
7 NO<sub>2</sub><sup>+</sup>) are produced from both inorganic and organic nitrate (e.g. Farmer et al. (2010)). The  
8 higher ACSM nitrate compared to MARGA nitrate can therefore also be explained by the  
9 presence of organic nitrates. In case of nitrate the ACSM and MARGA comparison cannot be  
10 seen as independent. The underestimation of the ACSM in case of sulfate exceeding the  
11 uncertainties may arise from calibration issues. The mass calibration procedure used in this  
12 work was mainly adopted from AMS procedures which may not be directly suitable for the  
13 ACSM. The RIE of sulfate might be overestimated due to high observed background signals  
14 during the calibration using (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> particles. An overestimated RIE<sub>SO<sub>4</sub></sub> results in  
15 underestimated mass concentrations. This would in turn explain the low SO<sub>4</sub> mass  
16 concentrations comparing to the MARGA and AMS. Additionally, the ACSM fragmentation  
17 table could not be adjusted for interferences of ions from different aerosol species on the same  
18 m/z properly because ACSM software used in this study did not allow plotting time series for  
19 specific fragments apportioned by the fragmentation table. Thus, the standard table had to be  
20 used. Furthermore, the MARGA measured only low concentrations of Mg, Na, K and Ca  
21 (average sum: 0.08  $\mu\text{g m}^{-3}$ ). Thus, contributions of their corresponding sulfate salts, for which  
22 the ACSM is less sensitive, are negligible in this context. It should also be noted, that chloride  
23 concentrations can originate from particulate organic and inorganic chloride components  
24 originating from e.g. sea salt. For the latter, the ACSM is much less sensitive than the  
25 MARGA. As described above, influences from sea salt can be considered rather low. In turn,  
26 the MARGA might be less sensitive to organic chlorides, as they are likely less water soluble  
27 than inorganic chlorides. These explanations would explain the low agreement between the  
28 two instruments in case of chloride.

29 Overall, the comparison of the data measured by the ACSM and MAAP with collocated  
30 aerosol instruments showed a good reliability, precision, and in most cases a good accuracy  
31 over the whole campaign, including periods with high and low mass loadings. Therefore, the  
32 here obtained chemical composition can be used qualitatively and quantitatively for the  
33 source apportionment of aerosol components at this rural site. Major discrepancies to the  
34 SMPS especially during some of the pollution events like 16 to 27 January 2013 and 5 to 8

1 May 2013 (see below) can be explained by the correction of losses through the 60 m inlet line  
2 which was done size dependently and did not account for losses of individual species as  
3 mentioned in chapter 2.3. As the quantitative agreements of individual inorganic species as  
4 well as of total inorganics between the ACSM and the MARGA during these periods are  
5 much higher, the mass loadings determined from these instruments are more reliable than the  
6 SMPS data.

### 8 3.2 Aerosol chemical composition

9 A meteorological overview of this campaign, including wind direction, precipitation, Radon-  
10 222 measurements and ambient temperature and relative humidity (RH) is provided in Fig.  
11 S7. Table S1 shows temperature and RH values averaged over selected periods (see below). If  
12 not stated else, all data shown in this chapter is in UTC (local time minus 1 or 2 hours,  
13 respectively) and averaged and synchronized to the ACSM data resolution of 30 minutes.  
14 Summing up the ACSM and MAAP data, a total mass concentration of  $9.5 \mu\text{g m}^{-3}$  was  
15 measured on average, with a maximum of  $78.4 \mu\text{g m}^{-3}$  and a minimum of  $0.2 \mu\text{g m}^{-3}$ .

16 Data coverage of 75% for one day and 90% for one calendar year, respectively, is defined  
17 mandatory for a proper risk assessment according to the WHO air quality guideline. The last  
18 requirement could not be achieved either with the ACSM or with the MARGA between July  
19 2012 and June 2013 (70% and 71% coverage, respectively). Nevertheless, the average total  
20 mass derived from the combination of the MARGA  $\text{PM}_{2.5}$  data (including all water soluble  
21 inorganic components  $\text{NO}_3$ ,  $\text{NH}_4$ ,  $\text{SO}_4$ ,  $\text{Cl}$ ,  $\text{Na}$ ,  $\text{K}$ ,  $\text{Mg}$ , and  $\text{Ca}$ ), MAAP eBC and ACSM  $\text{PM}_1$   
22 organics resulted in  $12.5 \mu\text{g m}^{-3}$  during this time, clearly exceeding the WHO  $\text{PM}_{2.5}$  annual  
23 mean limit ( $10 \mu\text{g m}^{-3}$ ) by 25%. Even the  $\text{PM}_1$  concentration inferred from ACSM + MAAP  
24 data (campaign average  $9.5 \mu\text{g m}^{-3}$ ) approached the WHO  $\text{PM}_{2.5}$  limit.

25 The time series of the daily mean total mass derived from both combinations (MARGA- $\text{PM}_{2.5}$   
26 + ACSM-Org + eBC and ACSM-total + eBC), where the required data coverage of 75% was  
27 reached, are shown in Fig. 1. The WHO  $\text{PM}_{2.5}$  daily mean limit of  $25 \mu\text{g m}^{-3}$  was exceeded on  
28 17 and 12 days, respectively. 8 and 7 exceedances were observed during a period with high  
29 pollution from 16 to 27 January 2013, respectively.

30 In Fig. 2 the average contributions of individual species are shown as pie charts (a) and  
31 stacked time series (b). A technical problem of the MAAP instrument was responsible for the  
32 lack of BC data from 15 February 2013 to 25 April 2013. To determine properly average

1 species contributions for each pie chart the campaign was therefore not divided strictly  
2 season-wise, but into five periods. The first two sections represent the summer (July-August-  
3 September) and autumn (October-November-December) 2012, while the first half of 2013  
4 was divided into periods with and without BC data. Larger gaps in ACSM data occurred  
5 mainly due to problems with the RGA detector, in addition to minor measurement gaps for  
6 maintenance and calibrations.

7 Overall, particulate nitrate and organics were the dominant species, representing 39% and  
8 29% of the total aerosol, respectively. Both compounds show similar contributions in summer  
9 (period 1) and autumn 2012 (period 2), whereas in winter (January-February-March, period 3  
10 and beginning of 4) and spring (April-May-June, periods 4 and 5) the  $\text{NO}_3$  fraction increased  
11 up to an average of 46% of the total particulate mass, and the organic and eBC fractions  
12 decreased. The contributions of the other components showed only small variations between  
13 the seasons (see Table S2).

14 The most significant pollution events (17 to 21 August 2012, 21 to 25 October 2012, 16 to 27  
15 January 2013, and 5 to 8 May 2013) are highlighted with shaded backgrounds in Fig. 2.  
16 During these events, MARGA  $\text{PM}_{2.5}$  data showed up to 33% higher (e.g. on 17 to 21 August  
17 2012) total inorganic masses than ACSM  $\text{PM}_1$  inorganics as implicated in Fig. 1. As  
18 mentioned above the number and proportions of exceedances of the WHO  $\text{PM}_{2.5}$  daily mean  
19 limit is therefore even higher than when only  $\text{PM}_1$  data is considered. During the last three  
20 events, northerly and north-easterly winds dominated. The period 16 to 27 January 2013  
21 showed also the lowest temperatures (average:  $-4^\circ\text{C}$ ) with respect to the whole campaign and  
22 a temperature inversion between 2 m and 40 m height in the in the morning 25 January 2013,  
23 where both instruments showed total mass concentrations of ca.  $20 \mu\text{g m}^{-3}$ . On the other hand,  
24 no temperature inversion was seen during times when very high ( $> 40 \mu\text{g m}^{-3}$ ) aerosol mass  
25 loadings were observed, even in winter times. Many sudden drops of the particulate mass can  
26 be either explained by changes in wind direction and/or precipitation events, like in case of  
27 the two latter pollution events (16 to 27 January 2013, and 5 to 8 May 2013).

28 Figure 3 shows the diurnal patterns of each individual species and the total particulate mass  
29 for the whole campaign. Corresponding plots with data averaged separately for the five  
30 chosen periods can be found in the supplement (Fig. S8). Overall,  $\text{NO}_3$  showed the largest  
31 diurnal variation, with a maximum during the night/morning hours, reflecting its nighttime  
32 production and a minimum during the day due to the volatility of  $\text{NH}_4\text{NO}_3$ . This pattern is  
33 more pronounced in the warmer periods 1 and 5. Since the majority of ammonium is

1 originated from  $\text{NH}_4\text{NO}_3$ ,  $\text{NH}_4$  has a similar pattern to that of  $\text{NO}_3$ .  $\text{SO}_4$ , which is mainly  
2 formed photochemically during the day from gaseous  $\text{SO}_2$ , showed peaks during daytime,  
3 although its overall variation is rather low. The maxima of BC can be attributed to direct  
4 emissions from traffic (morning and evening rush hours) and biomass burning events  
5 (domestic heating in the evenings/nights). Finally, OA showed peaks at the evening hours  
6 during the colder periods and a daytime minimum during the summer. More detailed  
7 discussion of the diurnal patterns of individual OA factors is given below.

8 An ion balance of all inorganic compounds indicates that too less  $\text{NH}_4$  was measured to  
9 neutralize all  $\text{NO}_3$  and  $\text{SO}_4$  to their corresponding ammonium salts. The measured  $\text{NH}_4$  mass  
10 concentration against the predicted  $\text{NH}_4$  from the ion balance is plotted in the supplement  
11 (Fig. S9) and resulted in a slope of  $0.83 \pm 0.00$ . One uncertainty of the ion balance is  
12 introduced by the use of RIE's for the inorganic species. As shown below we consider  
13 uncertainties of RIE to be of minor importance in explaining the observed low particulate  
14  $\text{NH}_4$  concentration. It should be noted that for the calculation of the ion balance all measured  
15  $\text{NO}_3$  is considered. In addition to  $\text{NH}_4\text{NO}_3$  organic nitrates give rise to nitrate signal in the  
16 AMS, albeit with distinct difference in relative ion abundance (Alfarra et al., 2004; Fry et al.,  
17 2011; Fry et al., 2009). It was not possible to distinguish the organic and inorganic nitrate  
18 fraction based on their mass spectra with the ACSM. Since a relatively high amount of  
19 particulate organic nitrates was found in previous campaigns at CESAR using an AMS  
20 (Mensah, 2011), and the region is characterized by high  $\text{NH}_3$  emissions (Derksen et al., 2011),  
21 the potential contribution of organic nitrates to the gap in the ion balance was explored as  
22 follows.

23 In contrast to the ion balance from the ACSM data, MARGA  $\text{PM}_{10}$  measurements during the  
24 whole campaign showed a nearly 1:1 correlation of measured against predicted  $\text{NH}_4$  (slope of  
25 the linear regression line:  $1.03 \pm 0.00$ , Pearson- $R^2 = 0.97$ ), but with a negative offset of ca.  
26  $0.30 \pm 0.01 \mu\text{g m}^{-3}$ . This offset, which is at least 3 times higher than the detection limits of the  
27 MARGA, cannot be explained by including positive metal ions to the ion balance, since the  
28 sum of Mg, Na, K and Ca mass concentrations had low contribution to particulate mass as  
29 mentioned in section 3.1. Thus, significant influence of their nitrate salts to total nitrate can be  
30 excluded. In addition, as the MARGA is measuring the water-soluble nitrate fraction, the  
31 MARGA- $\text{NO}_3$  can be considered to be exclusively  $\text{NH}_4\text{NO}_3$ . This assumption is acceptable,  
32 as shown by using the MARGA- $\text{NO}_3$  instead of the ACSM-total- $\text{NO}_3$  for the ion balance of  
33 ACSM data (including ACSM- $\text{SO}_4$ , -Chl and - $\text{NH}_4$ ), following a procedure given by Xu et al.

1 (2015) who calculated the organic nitrate fraction by subtracting the inorganic nitrate  
2 concentrations measured by a particle-into-liquid sampler (PILS, see Orsini et al. (2003))  
3 from ToF-AMS total nitrate concentrations. In the Cabauw data set, the correlation of  
4 measured against predicted  $\text{NH}_4$  resulted in a nearly 1:1 regression line without a significant  
5 offset (Fig. S10). This is in agreement with the MARGA internal ion balance which also  
6 indicates neutralized inorganic aerosols. Therefore, the mass concentration of nitrate groups  
7 associated with organic molecules (hereafter called organic nitrate or  $\text{OrgNO}_3$ ), can be  
8 estimated by subtracting the MARGA-nitrate from the ACSM-nitrate concentration. The  
9  $\text{OrgNO}_3$  time series using this approach is plotted in Fig. S11, the respective diurnal variation  
10 averaged over the whole campaign in Fig. S12. An average mass fraction of 9% was  
11 calculated for  $\text{OrgNO}_3$  (average concentration:  $0.43 \mu\text{g m}^{-3}$ ) in respect to total ACSM- $\text{NO}_3$ .  
12 The organic nitrate fraction shows a maximum concentration in the night, followed by a  
13 decrease during the day. These findings are in agreement with previously reported relatively  
14 high AMS organic nitrate fractions by Mensah (2011) in May 2008 ( $0.5 \mu\text{g m}^{-3}$ , 35% of total  
15 nitrate, 5.2% of total aerosol mass) and March 2009 ( $0.2 \mu\text{g m}^{-3}$ , 10% of total nitrate, 3.6% of  
16 total aerosol mass) in Cabauw.

17

### 18 3.3 Factor analysis of organic aerosols

19 Prior to PMF analysis, the ACSM data set was subdivided into four data sets, which were  
20 explored by PMF separately, mainly due to the operational time of the ACSM that are in the  
21 following referred as: (i) July-August-September 2012: Summer 2012; (ii) October-  
22 November-December 2012: Autumn 2012; (iii) January-February-March 2013: Winter 2013;  
23 (iv) April-May-beginning of June 2013: Spring 2013. A more detailed overview of the  
24 selected seasons is given in Table S3.

25 Figure 4 summarizes the time series of PMF factor classes (two POA factors: HOA and  
26 BBOA; and two SOA factors: OOA and a humic-like substances (HULIS) related factor, see  
27 below for descriptions) found for all seasons, except for BBOA in Summer 2012. A  
28 corresponding graph dividing these PMF results into the five periods according to Fig. 2 is  
29 shown in Fig. S13. The POA profiles were constrained within ME-2 using the HOA and  
30 BBOA mass spectra found by ME-2 operated PMF analyses by Crippa et al. (2014) at the  
31 CESAR tower in Cabauw in March 2009. For the HOA profile, a constant  $a$ -value of 0.1 was  
32 found to be most suitable for every season. If observed, the BBOA  $a$ -value was set to either

1 0.2 (Autumn 2012) or 0.3 (Winter and Spring 2013). The a-values based on different  
2 sensitivity tests for each season as described by Canonaco et al. (2013). Factor profiles  
3 including fractional contributions of marker ions ( $f_{44}$ ,  $f_{43}$ , and  $f_{60}$ ) and diurnal variations for  
4 Winter 2013 are displayed in Fig. 5 as a representative example. Corresponding graphs for  
5 every season are shown in Fig. S14 and S15, respectively. Table S4 gives an overview of the  
6 temporal correlations of each factor with external data sets, while scatter plots from Winter  
7 2013 profiles against profiles found in May 2008 at CESAR by Crippa et al. (2014) are given  
8 in Fig. S16.

9 As mentioned, no significant BBOA influence was seen in Summer 2012. Thus, a three-factor  
10 solution was chosen for this season and consequently, no BBOA time series exists at that  
11 time. For all other factors the time series are continuous on 1 October 2012, i.e. analysis of  
12 separate data files leads to a consistent result in terms of both factor profiles and  
13 concentrations. The SOA factors showed always higher contribution (54% - 84%, averaged  
14 season-wise) to total organics compared to POA (16% - 46%). For all PMF factors, no  
15 preferential wind direction was observed over the entire campaign. During the pollution  
16 events mentioned above, OOA originated mostly from the directions between 20° and 180° in  
17 respect to the tower. This is not the case for HULIS, which origins varied throughout all  
18 directions, also during pollution events.

19 The seasonal average HOA contribution to total organic mass was highest in Summer 2012  
20 and lowest in Spring 2013 (16% and 8%, respectively). All HOA diurnal patterns (Fig. 5b)  
21 showed a maximum at 7 and 11 am (LT) and a slight increase in the evening, emphasizing  
22 that its main source is related to traffic likewise to BC (see Sec. 3.1). In Winter 2013, these  
23 maxima were less distinctive comparing to the other seasons. HOA Highest temporal  
24 agreements with HOA were seen by the POA tracers BC, NO<sub>x</sub> and CO ( $R^2 = 0.38, 0.47,$  and  
25 0.47, respectively) over the entire campaign.

26 The BBOA profile showed a very high contribution of m/z 60, which is dominated by the  
27 C<sub>2</sub>H<sub>4</sub>O<sub>2</sub><sup>+</sup> ion. This fragment is characteristic for anhydrosugars such as levoglucosan (Alfarra  
28 et al., 2007) which are established markers of wood combustion processes (Simoneit and  
29 Elias, 2001; Simoneit et al., 1999). The fractions of m/z 60 to the BBOA profile in Autumn  
30 2012 (3.7%) and Winter 2013 (3.2%) are higher than in Spring 2013 (2.4%). As mentioned,  
31 BBOA was not found in Summer 2012. This was verified by the fact that the contribution of  
32 m/z 60 to the BBOA profile decreases for higher a-values in that season, which is an  
33 indication for the non-existence of BBOA. The highest contributions of BBOA to total



1 organics were seen in the colder Autumn (23%) and Winter seasons (15%). This and the  
2 diurnal maximum during the evenings and nights match the expectations for a factor linked  
3 with domestic heating activities, together with the fact, that this factor was not seen during the  
4 warmer summer season. Averaged over the whole campaign, the contribution to total organics  
5 was 13%, including Summer 2012, where **its** fraction was set to zero. In Winter, the  
6 correlations with eBC and CO were higher ( $R^2 = 0.64$  and  $0.57$ , respectively) than over the  
7 whole campaign ( $R^2 = 0.39$  and  $0.49$ , respectively), meaning that these compounds are  
8 reasonably more attributed to domestic heating during the colder periods **comparing to the**  
9 **contribution of heating to BC and CO during the other seasons** in this region.

10 The OOA profile is similar to a MS pattern as expected for a low volatile OOA (LVOOA)  
11 factor. The correlation coefficients (Pearson- $R^2$ ) with the OOA and LVOOA spectra given by  
12 Ng et al. (2011a) are 0.94 and 0.97, respectively. Similar agreement was found compared to  
13 the LVOOA factor observed by Mensah (2011) and Crippa et al. (2014) ( $R^2 = 0.97$  and  $0.94$ ,  
14 respectively) at the CESAR tower in May 2008. The OOA factor showed a night-time  
15 maximum and a day-time minimum. This is rather characteristic for a semi volatile OOA  
16 (SVOOA) behavior, as well as the high agreement with  $\text{NO}_3$  over the whole year ( $R^2 = 0.63$ ),  
17 as described by Lanz et al. (2007). The correlation to the LVOOA associated compound  $\text{SO}_4$   
18 is less significant ( $R^2 = 0.48$ ). OOA dominated the organic fraction in Winter and Spring 2013  
19 (47% and 48% contribution, respectively; 33% over the whole year). During the defined  
20 pollution events the increase of the OOA mass concentration (up to  $11.8 \mu\text{g m}^{-3}$ ), relative to  
21 the campaign average ( $1.05 \mu\text{g m}^{-3}$ ), is much stronger compared to the other PMF factors,  
22 demonstrating that this factor is mainly responsible the high total OA mass during these  
23 periods.

24 The so called HULIS factor showed the highest  $f_{44}$  values of all factors, increasing from  
25 Summer 2012 to Spring 2013 from 0.23 to 0.35. The HULIS factor class was first observed  
26 by Mensah (2011) in previous AMS campaigns at the CESAR tower in May 2008 and March  
27 2009. The identification and characterization of this factor class was done in the 2008  
28 campaign by the comparison with data from an ion-exchange chromatographic method for  
29 direct quantification of humic-like substances (HULIS) and from water-soluble organic  
30 carbon (WSOC) analyzed offline on a set of filters collected in parallel (Paglione et al., 2014).  
31 The HULIS factor provided the highest contribution to the total organic mass over the entire  
32 campaign (41%) and was the dominant factor in Summer and Autumn 2012. Since it had no  
33 distinct diurnal variation and preferential wind direction, it can be considered as regional

1 background aerosol at this rural site. Additionally, the variation between the seasonal average  
2 concentrations of HULIS within the ACSM data set is less than  $\pm 10\%$ . Also the comparison  
3 to the most important tracers (Table S4 in the supplement) showed no preferential attribution,  
4 either to a low-volatile ( $R^2 = 0.41$  with  $\text{SO}_4$ ), semi-volatile ( $R^2 = 0.39$  with  $\text{NO}_3$ ) or to primary  
5 organic aerosol ( $R^2 = 0.47$  with eBC). These characteristics were also reported for the HULIS  
6 factor found at CESAR in May 2008 (Crippa et al., 2014; Mensah, 2011). The correlation of  
7 the sum of the secondary inorganic species  $\text{NO}_3$  and  $\text{SO}_4$  with the sum of the OOA and  
8 HULIS time series gives a coefficient of  $R^2 = 0.70$ , which is slightly higher than with OOA  
9 only ( $R^2 = 0.67$ ). This might confirm the SOA character of the HULIS factor. The reason why  
10 HULIS was the dominant factor in Summer and Autumn 2012, is due to the lower mass  
11 concentrations of the other factors compared to the remaining periods. In turn, OA mass  
12 increased during pollution events mainly due to the increase of the other SOA factor, namely  
13 OOA. A number of studies are published with different theories on the formation and sources  
14 of atmospheric HULIS. It shows similarities to terrestrial humic acids (HA) and fulvic acids  
15 (FA) due to their (poly-) acidic nature, but with lower molecular weight ( $< 1000$  amu) than  
16 HA or FA (Graber and Rudich, 2006; Kiss et al., 2003). On the other hand, solid-phase  
17 extraction protocols for HULIS associate them with the more hydrophobic fraction of water-  
18 soluble aerosols (e.g. Varga et al. (2001)). Graber and Rudich (2006) suggested that the  
19 formation of HULIS happens by oligomerization of lighter organic acids in liquid droplets in  
20 the atmosphere within time scales of hours to days. This emphasizes, that HULIS is an  
21 ubiquitous background factor at Cabauw, characterized as regionally well mixed aerosol of  
22 long atmospheric lifetime.

23 Note that the correlation (Pearson- $R^2$ ) of the POA factors with eBC data is relatively low  
24 ( $R^2 = 0.38$  and  $0.39$  with HOA and BBOA, respectively), while in turn HULIS shows a higher  
25 correlation with eBC,  $R^2 = 0.47$ . This can be understood in the light of a study by Andreae  
26 and Gelencser (2006) who pointed out that the fiber filters, as used by the MAAP, loaded with  
27 macroscopic amounts of humic like substances samples look very dark brown or nearly black.  
28 This implies an interference of HULIS with BC measurements and in consequence explains  
29 why the HULIS factor, in contrast to the POA factors, has a relatively high correlation  
30 coefficient with BC data.

31 The source apportionment as described here used a data set which was subdivided into the  
32 four seasons prior to PMF analysis. A single PMF analysis of the whole data set with  
33 constrained HOA and BBOA profiles lead to solutions with a highly overestimated BBOA

1 factor in the summer, compared to the results when the seasons were explored individually  
2 (see Fig. S17 and S18 in the supplement). Furthermore, the contributions of individual factors  
3 change significantly in some periods, especially for the OOA factor during pollution events.  
4 This is mainly driven by the different OOA-*f*<sub>44</sub> and -*f*<sub>43</sub> values. This behavior is independent  
5 from applied *a*-values for BBOA may result from the uncertainty of this statistical tool. Since  
6 there was no evidence of BBOA seen in the separate analysis of the summer period (e.g. low  
7 fraction of *m/z* 60 in the organic mass spectrum and no correlation of the constrained BBOA  
8 factor with POA tracers, no matter which *a*-value was used), the solutions derived from the  
9 single PMF analysis was reasonably rejected.

10

### 11 3.4 Composition and sources of aerosols during pollution events

12 The investigation of the aerosol composition during the pollution events showed that the  
13 majority was contributed from secondary aerosols. As an example, Fig. 6 displays average  
14 abundances during two selected periods (17 to 21 August 2012 and 16 to 27 January 2013),  
15 where the organic fraction is further distinguished into the PMF factors. Nitrate and organics  
16 are still the dominant species, while the primary aerosol components (BC, HOA, BBOA) have  
17 even less influence than averaged over the respective season. An exception is seen from  
18 BBOA, which contribution is reasonably higher during the latter pollution event with very  
19 low temperatures, when domestic heating sources most likely increased. But even at this time,  
20 primary organics do not exceed 11% of total mass on average.

21 Since the major inorganic components  $\text{NH}_4\text{NO}_3$  and  $(\text{NH}_4)_2\text{SO}_4$  are produced by atmospheric  
22 processes, they are considered as secondary aerosols. Together with the high abundance of  
23 SOA (OOA, HULIS), chemically aged aerosol components have the largest impact on total  
24 particulate mass (up to 90%) at Cabauw. The largest effect on a potential reduction of  
25 particulate mass can therefore be achieved by reducing  $\text{NO}_x$  and/or  $\text{NH}_3$  emissions, the  
26 anthropogenic precursors of particulate  $\text{NH}_4$  and  $\text{NO}_3$ . Indeed, on 16 January 2013 highest  
27  $\text{NO}_x$  values were observed compared to the whole campaign (daily mean:  $96 \mu\text{g m}^{-3}$ ; seasonal  
28 average:  $29 \mu\text{g m}^{-3}$ ). Also the highest  $\text{NH}_3$  values were obtained between 17 and 19 August  
29 2012 (daily means:  $26\text{--}28 \mu\text{g m}^{-3}$ ; seasonal average:  $10 \mu\text{g m}^{-3}$ ). These ammonia values are  
30 higher than previously reported for Cabauw, e.g. by Derksen et al. (2011) in May 2008  
31 (approximately  $20 \mu\text{g m}^{-3}$ ). Lolkema et al. (2015) gave an annual average of  $5.3 \mu\text{g m}^{-3}$  for  
32 2013 in this region.

## 4 Conclusions

This work provides chemical composition data of atmospheric aerosols acquired during one year at the CESAR tower in Cabauw, the Netherlands, which is a representative rural site for North Western Europe. The concentration of submicron particles from combined ACSM and MAAP data showed 12 exceedances from the WHO PM<sub>2.5</sub> daily mean limit. The respective campaign average of 9.5 µg m<sup>-3</sup> approached the WHO PM<sub>2.5</sub> annual mean limit. Taking MARGA PM<sub>2.5</sub> into account the number and proportions of these exceedances are even higher, emphasizing the importance of these high ACSM PM<sub>1</sub> results shown here which represent lower limits. As carbonaceous compounds are estimated to be five times more toxic than inorganic particles (Lelieveld et al., 2015) MARGA data alone would not give sufficiently possible implications regarding adverse health effects. While few people live in the direct vicinity of the measurement site, the high aerosol concentration measured at the site can be considered to represent the regional background. This regional background is adding to local aerosol contributions in high populated urban sites (Pandis et al., 2013), namely the 4 largest cities of the Netherlands which have a distance of 40 km or less from the CESAR tower.

Particulate mass loadings found at this rural site are dominated by secondary aerosol formation through atmospheric gas phase chemistry and particle phase aging. It is shown that particulate ammonium nitrate is the major aerosol component (39% on average) and represents the more hygroscopic aerosol fraction. Since the human respiratory system is characterized by high humidity more hygroscopic aerosols have a higher deposition tendency within the human lung than less water soluble particle compounds (Asgharian, 2004; Broday and Georgopoulos, 2001). With special regard to adverse health effects this is very crucial because Asgharian (2004) also found that especially hygroscopic submicron particles can deposit in the entire lung. The high ammonium nitrate fraction also implies that inorganic SA reduction in Cabauw can be most efficiently achieved through the reduction of gaseous ammonia emissions in the area.

As indicated by model results for the South Western United States from Zhang and Wu (2013), the reduction of NH<sub>3</sub> emissions, in conjunction with already implemented SO<sub>2</sub> and NO<sub>x</sub> emission reductions, can further reduce PM<sub>2.5</sub> than reducing SO<sub>2</sub> and NO<sub>x</sub> emissions alone, particularly for regions with high emissions of NH<sub>3</sub> from agricultural sources. The latter is clearly the case for Cabauw with its high number of animal husbandry and the use of nitrogen containing fertilizers around the CESAR tower. Banzhaf et al. (2013) derived similar

1 conclusions for PM<sub>10</sub> using different emission scenarios within domains covering Germany  
2 and Europe. Applying a 3d chemical transport model over Europe Megaritis et al. (2014)  
3 found that a reduction of NH<sub>3</sub> emissions by 50% would have a much higher effect on  
4 reducing PM<sub>2.5</sub> than decreasing NO<sub>x</sub> emissions by 50%. The latter scenario would even result  
5 in negative side effects such as higher tropospheric ozone concentrations (especially in  
6 summertime 4% over Western Europe and up to 40% in major urban areas) and higher  
7 amounts of particulate sulfate and OA by 8% and 12%, respectively, in winter.

8 The local mitigation of organic aerosol mass (29% contribution on average) is more  
9 challenging, as secondary organic aerosols are highly abundant at the Cabauw site (74% and  
10 22% of OA and total PM<sub>1</sub> on average, respectively). The presented data set shows a large and  
11 ubiquitous HULIS fraction (37%) which based on diurnal patterns and a lack of correlation  
12 with wind direction can be considered as long-range background aerosol formed from  
13 atmospheric aging processes. In turn, primary organic aerosols emitted mainly from traffic  
14 and biomass burning (12% and 13% of OA on average) have only minor importance. For a  
15 more detailed identification of the SOA sources compound specific measurements of OA as  
16 well as routine VOC monitoring are needed.

17 Finally, the presented data set and interpretations provide an important contribution to the  
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19 issues on climate change, air quality, and long-range transport of pollutants on a European  
20 scale.

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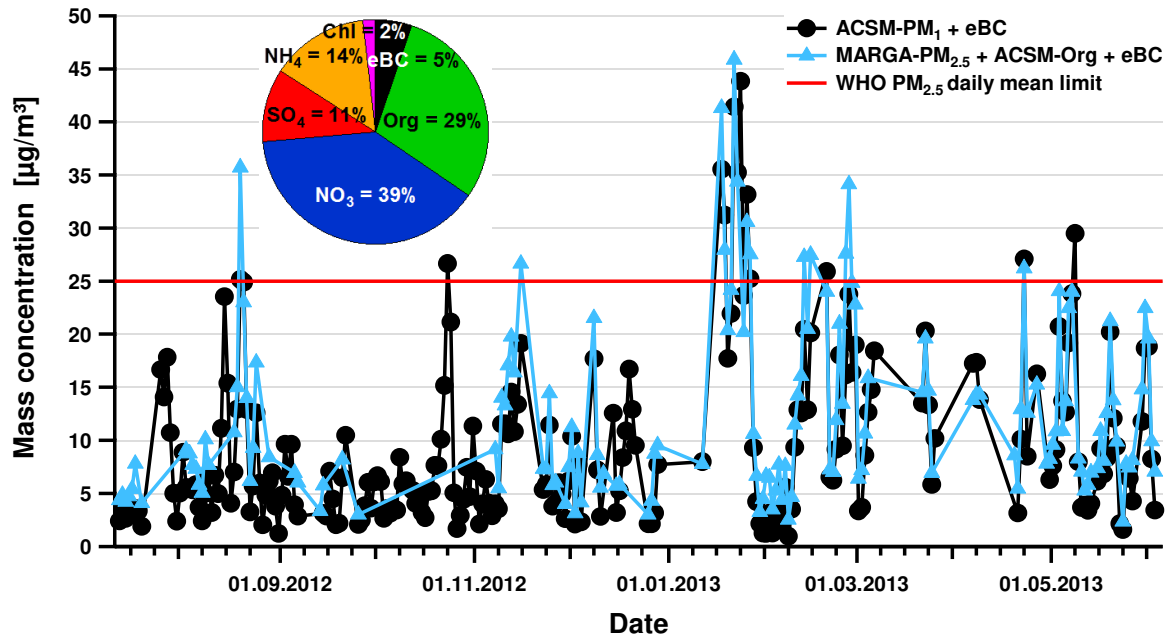
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1 Table 1: Results of the ACSM to MARGA and ACSM to AMS comparisons for individual  
 2 species and the respective total masses. Note that for comparison with the MARGA total mass  
 3 concentrations, only the ACSM inorganic species were considered.

		Slope	intercept [ $\mu\text{g m}^{-3}$ ]	R <sup>2</sup>
Chl	MARGA	$0.49 \pm 0.02$	$0.11 \pm 0.01$	0.24
	AMS	$0.67 \pm 0.01$	$-0.01 \pm 0.01$	0.31
NH <sub>4</sub>	MARGA	$0.88 \pm 0.01$	$0.07 \pm 0.01$	0.93
	AMS	$0.82 \pm 0.03$	$0.01 \pm 0.02$	0.71
SO <sub>4</sub>	MARGA	$0.63 \pm 0.01$	$-0.08 \pm 0.01$	0.86
	AMS	$0.49 \pm 0.02$	$-0.18 \pm 0.02$	0.76
NO <sub>3</sub>	MARGA	$1.23 \pm 0.01$	$-0.37 \pm 0.03$	0.96
	AMS	$1.17 \pm 0.02$	$-0.04 \pm 0.02$	0.89
Organics	AMS	$1.03 \pm 0.04$	$0.07 \pm 0.04$	0.73
Total	MARGA	$1.05 \pm 0.01$	$-0.70 \pm 0.06$	0.93
	AMS	$0.90 \pm 0.02$	$-0.02 \pm 0.07$	0.84

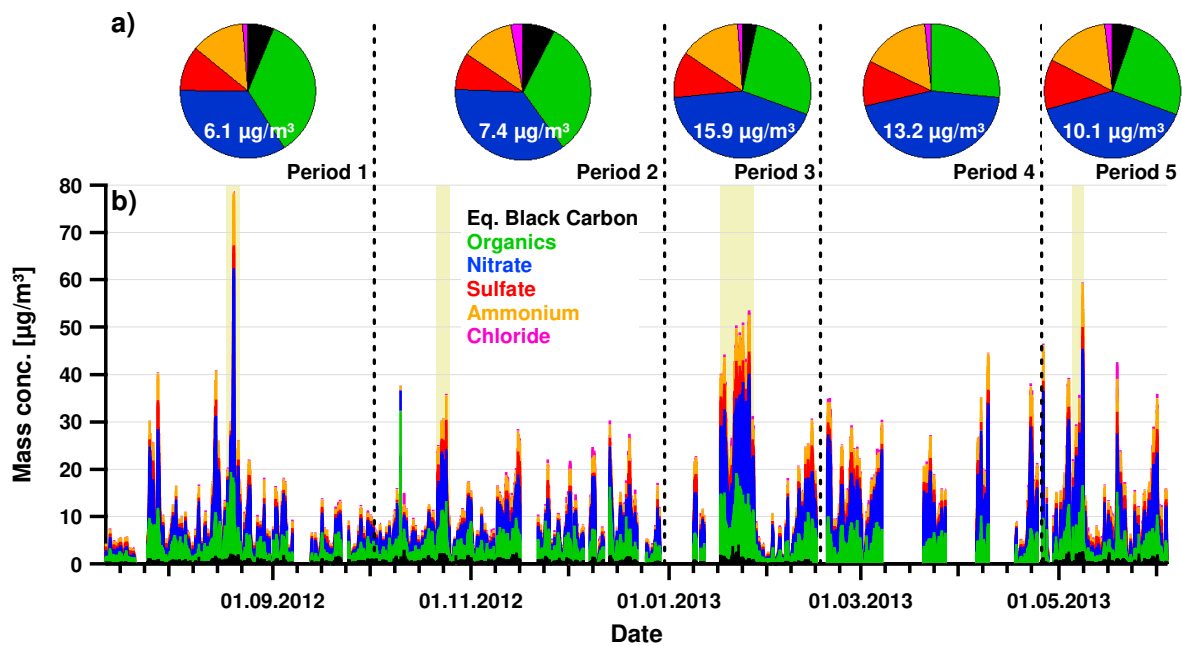
4



1

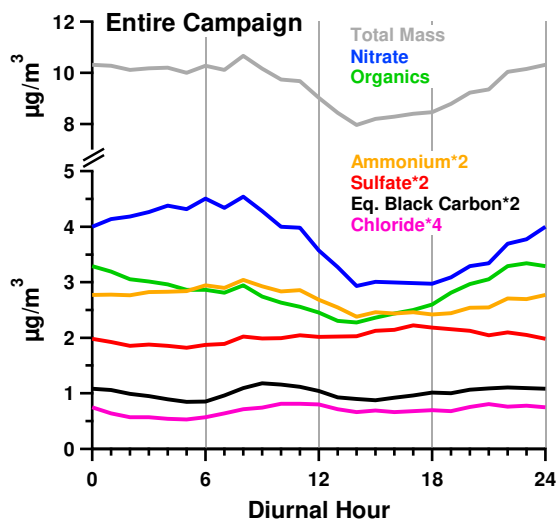
2 Figure 1: Time series of the daily mean. The black line represents the sum of eBC and all  
 3 ACSM species, the blue line the sum of eBC, ACSM organics and all MARGA-PM<sub>2.5</sub>  
 4 species. The pie chart shows the fractional abundances of individual eBC and ACSM species  
 5 averaged over the whole campaign. For the determination of the daily means missing eBC  
 6 data was filled with zero values, thus deriving lower concentration limits.

7



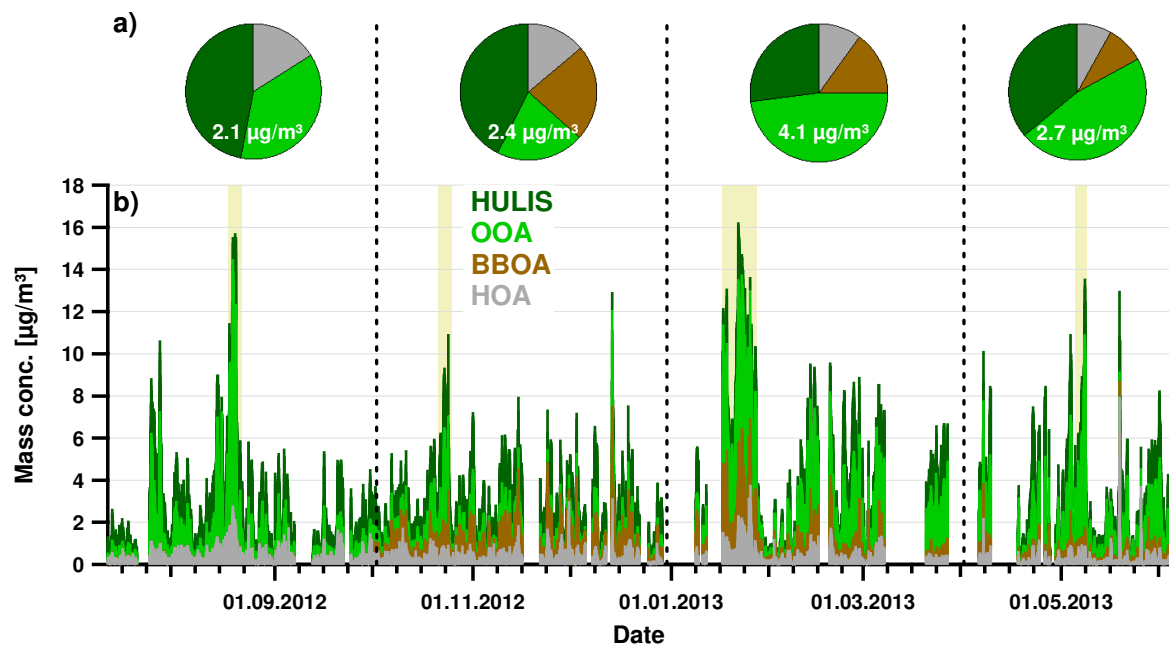
1  
 2 Figure 2: Overview of the ACSM campaign: a) Pie charts of average fractional abundances of  
 3 aerosol species, separated in five periods. The respective average total mass concentration is  
 4 written inside the pie chart. b) Stacked time series of mass concentrations of aerosol species.  
 5 Pollution events are indicated by shaded areas.

6



7  
 8 Figure 3: Diurnal variation (local time, LT) of individual species and the total mass, averaged  
 9 over the whole ACSM campaign

10



1

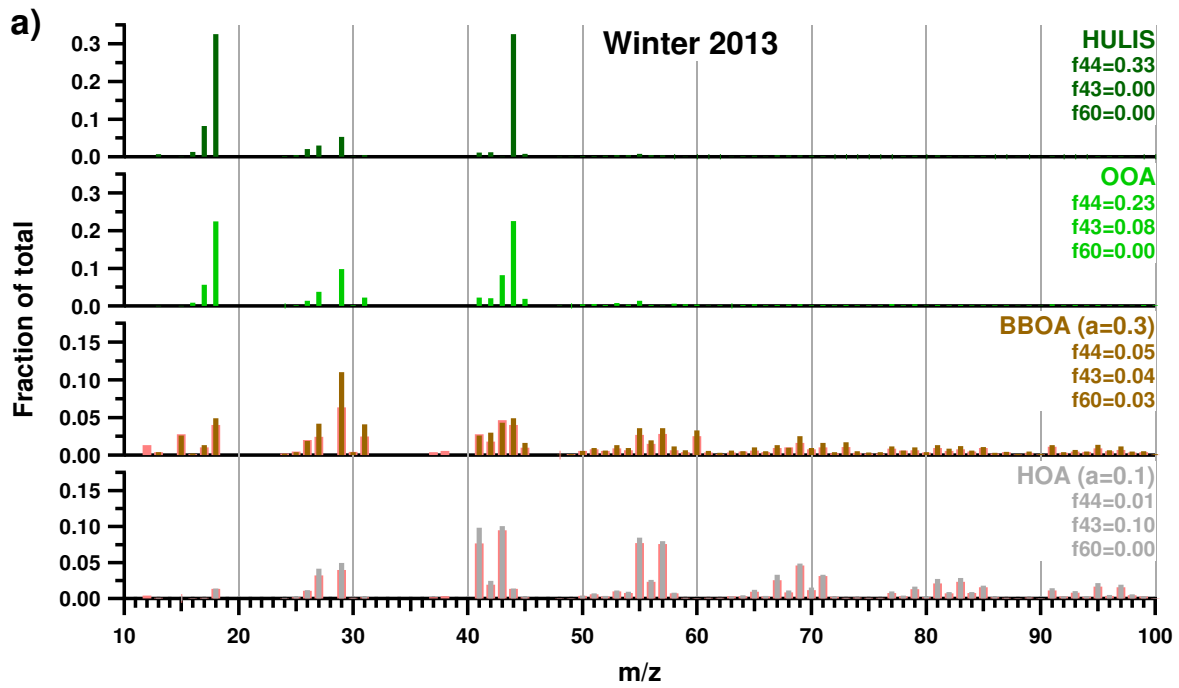
2 Figure 4: Overview of ACSM PMF factors: a) Seasonal pie charts of the factor contributions.

3 The respective average total organic concentration is written inside the pie chart. b) Stacked

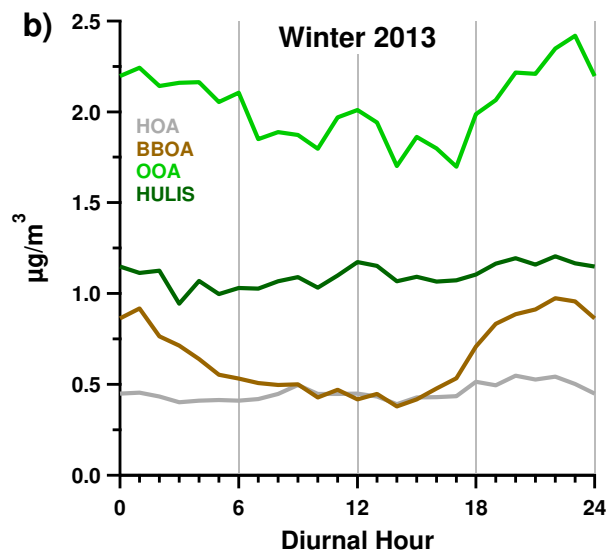
4 time series of mass concentrations. Shaded areas represent highly polluted events.

5





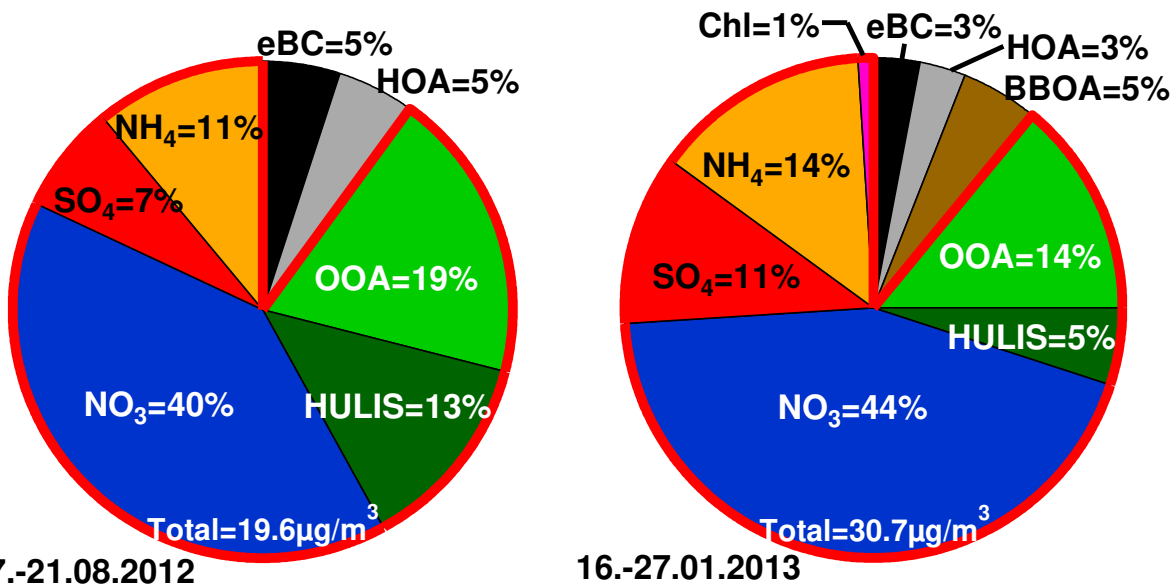
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2

3 Figure 5: a) Mass spectra of ACSM PMF factors and b) average diurnal variations (LT) of  
 4 ACSM factors found in Winter 2013. For the constrained profiles HOA and BBOA, the  
 5 applied a-value is written in brackets. Corresponding reference spectra are shown by red bars.  
 6 Note that the y-axis scales of the POA profiles are zoomed by a factor of 2 comparing to SOA  
 7 profiles.

8



1 17.-21.08.2012 16.-27.01.2013

2 Figure 6: Average contributions of individual aerosol species and PMF factors during selected

3 pollution events. Surrounded red regions represent the secondary aerosol fraction.