

1 **Source apportionment and dynamic changes of carbonaceous aerosols during the haze**
2 **bloom-decay process in China based on radiocarbon and organic molecular tracers**

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16

17 **Abstract**

18 Fine carbonaceous aerosols (CAs) is the key factor influencing the currently filthy air in megacities of
19 China, yet seldom study simultaneously focuses on the origins of different CAs species using specific and
20 powerful source tracers. Here, we present a detailed source apportionment for various CAs fractions,
21 including organic carbon (OC), water-soluble OC (WSOC), water-insoluble OC (WIOC), elemental carbon
22 (EC) and secondary OC (SOC) in the largest cities of North (Beijing, BJ) and South China (Guangzhou,
23 GZ), respectively, using the measurements of radiocarbon and anhydrosugars. Results show that non-fossil
24 fuel sources such as biomass burning and biogenic emission make a significant contribution to the total
25 CAs in Chinese megacities: $56\pm 4\%$ in BJ and $46\pm 5\%$ in GZ, respectively. The relative contributions of
26 primary fossil carbon from coal and liquid petroleum combustions, primary non-fossil carbon and
27 secondary organic carbon (SOC) to total carbon are 19%, 28% and 54% in BJ, and 40%, 15% and 46% in
28 GZ, respectively. Non-fossil fuel sources account for 52% in BJ and 71% in GZ of SOC, respectively. These
29 results suggest that biomass burning has a greater influence on regional particulate air pollution in North
30 China than in South China. We observed an unabridged haze bloom-decay process in South China, which
31 illustrates that both primary and secondary matter from fossil sources played a key role in the blooming
32 phase of the pollution episode, while haze phase is predominantly driven by fossil-derived secondary
33 organic matter and nitrate.

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35

36 **1 Introduction**

37 Particulate matter with an aerodynamic diameter of $<2.5 \mu\text{m}$ ($\text{PM}_{2.5}$) is either directly emitted from emission
38 sources (chemical industry, power plant, vehicle, biomass burning, soil dust, etc.) or formed as secondary
39 particles via the conversion of volatile organic compounds (VOCs) and inorganic gases ($\text{SO}_2/\text{NO}_x/\text{NH}_3$) to
40 the particulate phase. These tiny particles have been shown to cause numerous environmental, health and
41 climate problems that closely link humans and the global climate system. (Brunekreef et al., 2002; Dockery
42 et al., 1993; Huang et al., 2014a, Wang et al., 2011; Wang et al., 2014). Many countries or regions have set
43 strict standards for the ambient concentration of $\text{PM}_{2.5}$ in an attempt to improve public health and protect
44 air quality. For example, the annual and 24-hour $\text{PM}_{2.5}$ standards are regulated at $35 \mu\text{g}/\text{m}^3$ and $75 \mu\text{g}/\text{m}^3$,
45 respectively, in China.

46 As the world's largest contributor of $\text{PM}_{2.5}$ (Huang et al., 2014b), China is currently facing the challenge
47 of severe air pollution (i.e., haze episodes) (Chan et al., 2008; Zhang et al., 2012), which has already led to
48 numerous negative impacts on the atmospheric environment and public health. Up to 1.2 million premature
49 deaths in China were directly or indirectly correlated with air pollution in 2010 (Lim et al., 2013).
50 Furthermore, haze events in Chinese urban areas, especially in megacities, have become a common
51 phenomenon that can occur during any season owing to the intensive emissions of pollutants and
52 unfavorable meteorological conditions (He et al., 2014; Liu et al., 2013a). Better understanding of $\text{PM}_{2.5}$
53 sources and formation processes, which remain unclear due to the complicated chemical constituents, is
54 urgently needed and would greatly facilitate the development of steps to mitigate the serious haze pollution
55 in China.

56 Carbonaceous aerosols (CAs) in $\text{PM}_{2.5}$ have been shown to be crucial factors in the haze episodes. In a
57 highly polluted region, ~40% of $\text{PM}_{2.5}$ can be explained by CAs (Cao et al. 2003), exerting remarkable
58 impacts on the atmospheric visibility (Deng et al. 2008). Traditionally, CAs, of which classification

59 approach is method-dependent, are categorized as organic carbon (OC) and elemental carbon (EC). OC
60 represents the less refractory CAs that contain thousands of organic molecules either emitted by primary
61 emission sources (primary organic carbon, POC) or formed by the conversion process of gas-to-particle
62 (secondary organic carbon, SOC). In addition, OC also can be further classified into water-soluble organic
63 carbon (WSOC) and water-insoluble organic carbon (WIOC). EC is formed only during incomplete
64 combustion processes. TC is the sum of OC and EC. In recent years, source apportionments for these carbon
65 species have yielded useful information in China through radiocarbon (^{14}C)-based top-down studies. At the
66 early period, studies related the ^{14}C of atmospheric CAs were focusing on the TC fraction in China. It was
67 reported that ~30-50% of $\text{PM}_{2.5}$ TC is contributed from modern/non-fossil sources in Beijing during the
68 year of 2001 (Yang et al., 2005). In Lhasa, a remote city of China, non-fossil sources accounted ~for 36-
69 70% of TC (Huang et al., 2010). In a background site of Southeast China, Niu et al. (2013) observed the
70 obvious seasonality of percentages of non-fossil sources in TC: ~45% in summer and ~95% in winter,
71 respectively. Nowadays, studies are beginning to concern the origins of sub-fractions of CAs, i.e., EC and
72 OC in the filthy air of Chinese cities. Chen et al. (2013) first systematically studied the ^{14}C levels of EC,
73 also known as black carbon, in East Asia and found that fossil-fuel combustion contributed $80\pm 6\%$ of the
74 EC emitted from China, which is confirmed by the studies of Liu et al. (2014) and Zhang et al. (2015).
75 Using the combined measurements of ^{14}C and stable carbon isotope (^{13}C), Andersson et al. (2015) further
76 pointed out that the sources of EC covering China are highly region-specific, probably due to the big
77 difference of energy consumption among regions. Concerning OC, Zhang et al. (2015) found that the
78 averaged contribution of fossil sources is 35-58% in Chinese cities with the rest come from non-fossil
79 sources such as biomass burning and biogenic emissions. Through combining a series of analytical methods
80 with ^{14}C , Huang et al. (2014a) believed that haze enveloping Chinese cities are subject to secondary aerosols,
81 which accounting for 30-77% of $\text{PM}_{2.5}$ and 41-71% of organic aerosols, respectively. However, most of

82 these studies have emphasized only the ^{14}C levels of one or two carbon species, and our understanding of
83 haze formation remains limited. In this study, ^{14}C levels for the various carbon species (WIOC, WSOC,
84 OC, EC, and TC) are reported simultaneously in two cities located in North and South China, respectively.
85 To further constrain the atmospheric behavior of $\text{PM}_{2.5}$, secondary inorganic ions (SO_4^{2-} , NO_3^- and NH_4^+),
86 primary inorganic ions (K^+ , Ca^{2+} , Mg^{2+} , Cl^- and Na^+) and biomass burning-specific organic tracers
87 (Simoneit et al. 2001) (levoglucosan, Lev; Galactosan, Gal; Mannosan, Mann) were also measured. Finally,
88 a detailed source apportionment of CAs and $\text{PM}_{2.5}$ was achieved in the largest city of North (Beijing) and
89 South China (Guangzhou) using the measurements of ^{14}C and other organic/inorganic chemicals, and the
90 source dynamics of individual primary and secondary aerosols during the haze bloom-decay process in
91 Guangzhou basing day-to-day time serials and Beijing basing low-to-high $\text{PM}_{2.5}$ concentrations were
92 investigated as well.

93

94 **2 Materials and Methods**

95 **2.1 Field Sampling Campaign**

96 Twenty-four-hour $\text{PM}_{2.5}$ samples (9:00 a.m. to 9:00 a.m. the following day) were collected continuously on
97 pre-baked quartz fiber filters (8×10 inches, Pall) through a high-volume sampler that equipped with a $\text{PM}_{2.5}$
98 inlet (1 m^3/min , XT Instruments, Shanghai, China) in China's two largest megacities, located in North
99 (Beijing, BJ, 39.9°N, 116.4°E, ~20 million inhabitants; 21 samples) and South China (Guangzhou, GZ,
100 23.1°N, 113.3°E, ~10 million inhabitants; 14 samples) during March and April, 2013, respectively (Fig. 1).
101 The meteorological parameters during sampling are shown in Fig. S1. After sampling, the filters were
102 folded, wrapped in aluminum foil, sealed in airtight plastic bags, and stored in a refrigerator at -20 °C until
103 analysis. Three field blanks were collected from both sampling sites.

104

2.2 Separation for Carbon Species

A punch of filter was cut and sandwiched by a filtration unit equipped with a quartz cartridge, and subsequently extracted by 100-mL ultra-pure water (18.2 M Ω) carefully. Only ~5% of carbon in original filter was lost during this water-extraction and thus the resulting bias towards the measurement of WIOC and EC and the source apportionment of TC in the following text is marginal and can be neglected. WSOC species were quantified using a TOC analyzer (Shimadzu TOC_VCPH, Japan). The washed filter was dried in a desiccator, wrapped in aluminum foil and stored in a refrigerator. Contribution of carbonate carbon (CC) to fine aerosols generally can be neglected if studies focusing on the CAs (Chow and Watson, 2002). While, some early investigations showed that CC-rich dusts derived from deserts may exert a substantial influence on the air quality during the spring season in North China (He, et al., 2001; Zheng, et al., 2005). Thus, hydrochloric acid (1 M) was used to remove the potential CC in the samples collected in BJ in this study. WIOC and EC were obtained from the water-filtered sample using an off-line carbon analyzer (Sunset Laboratory, Inc., US) by the thermo-optical transmittance method (NIOSH 870). The average WIOC contents in the field blanks from BJ and GZ were 0.25 ± 0.02 $\mu\text{g}/\text{cm}^2$ and 0.26 ± 0.03 $\mu\text{g}/\text{cm}^2$, respectively. No EC and WSOC were detected in any of the field blanks. In this study, the reproducibility of the measurement of WIOC, EC, and WSOC is 5%, 7%, and 9%, respectively (n=4).

2.3 Radiocarbon Measurements

Isolation procedures for the ^{14}C measurements of WIOC, EC and WSOC have been described previously (Liu et al., 2013b; Liu et al., 2014; Zhang et al. 2010a). In brief, WIOC and EC were combusted in a stream of pure oxygen at 340 $^{\circ}\text{C}$ for 15 min and 650 $^{\circ}\text{C}$ for 10 min, respectively. Prior to combustion at 650 $^{\circ}\text{C}$, EC was placed in a tube furnace at 375 $^{\circ}\text{C}$ for 4 h with air. WSOC solution was frozen, freeze-dried, and then combusted at 850 $^{\circ}\text{C}$. Graphite target preparations and accelerator mass spectrometry (AMS) measurements

128 were performed at the Guangzhou Institute of Geochemistry, Chinese Academic Sciences (GIGCAS) and
129 Peking University NEC compact AMS facility, respectively. All ^{14}C values were reported as the fraction of
130 modern carbon (f_m) after correction with $\delta^{13}\text{C}$ for fractionation. f_m was converted into the fraction of
131 contemporary carbon (f_c) to eliminate the effect of nuclear bomb tests through conversion factors (Mohn et
132 al., 2008), which were 1.10 ± 0.05 for EC and 1.06 ± 0.05 for OC in 2013 (Liu et al., 2014), respectively. Both
133 f_c values for TC and OC were calculated by isotopic mass balance. No blank corrections were performed
134 owing to the low carbon amount in the filed blanks in this study, which accounted for only $<2\%$ of samples.

135

136 **2.4 Levoglucosan, Galactosan and Mannosan**

137 Levoglucosan (Lev), Galactosan (Gal) and Mannosan (Mann) are regarded as excellent tracers for biomass
138 burning activities (Simoneit et al. 2001). Detailed analytical procedure has been described in Liu et al.
139 (2013b, 2014). In brief, a section of filter was removed, spiked with 500 ng of methyl- β -L-
140 xylanopyranoside (m-XP) as internal standards, extracted with methanol, reduced using a rotary evaporator,
141 filtered through a Teflon syringe filter, dried in a stream of gentle nitrogen, and then reacted with a mixture
142 of 40- μL BSTFA (1% TMCS) and pyridine at 70 $^{\circ}\text{C}$ for 1 h. Subsequently, this derivatization solution was
143 injected into a gas chromatograph-mass spectrometer (GC-MS, Agilent 7890-5975) with a capillary column
144 (DB-5MS, 30 m, 0.25 mm, 0.25 μm).

145

146 **2.5 Inorganic Ions**

147 A 2.54 cm^2 filter was punched out, extracted twice in ultra-pure water (18.2 $\text{M}\Omega$) with a centrifuge,
148 sonicated in an ice-water bath, and filtrated using a Teflon syringe filter (0.22 μm). Subsequently, the
149 filtrates were combined and analyzed for anions and ions (Na^+ , Cl^- , Ca^{2+} , Mg^{2+} , K^+ , NH_4^+ , SO_4^{2-} and NO_3^-)
150 using an ion chromatography (Metrohm 883 Basic IC plus, Switzerland). Small amounts of Mg^{2+} , Ca^{2+} , Cl^-

151 and SO_4^{2-} were detected in the field blanks, and their corresponding concentrations in samples were
152 corrected. The other ions were not be detected in the field blank filters. The reproducibility of these ions
153 ranged between 5% and 11% (n=4) in this study.

154

155 **2.6 Methodology for the Source Apportionment of Carbons**

156 **2.6.1 The Separation of Fossil and Non-fossil Carbon Species**

157 The concentrations of non-fossil/fossil carbon species were directly calculated using their corresponding f_c
158 values as mentioned above. For example, the concentrations of fossil WIOC (WIOC_f) and non-fossil WIOC
159 (WIOC_{nf}) were calculated as follows:

$$160 \quad \text{WIOC}_f = (1 - f_c(\text{WIOC})) \times \text{WIOC}$$

$$161 \quad \text{WIOC}_{nf} = f_c(\text{WIOC}) \times \text{WIOC}$$

162 Thus, all carbon species could be separated into non-fossil and fossil fractions. EC is formed directly by
163 incomplete combustion processes, with non-fossil source EC emitted from biomass burning activities.

164

165 **2.6.2 Primary Organic Carbon**

166 OC particles existing in the present atmosphere are actually a mixture of POC and SOC. POC is the sum of
167 primary biomass burning OC (POC_{bb}), primary fossil fuel combustion-derived OC (POC_f) and the
168 carbonaceous material that exists in vegetation debris, bioaerosols and resuspended soil organic matter.
169 Since vegetation debris, bioaerosols and soil dust in the air is generally much larger than $2.5 \mu\text{m}$, its portion
170 in $\text{PM}_{2.5}$ is small and can be neglected, especially in highly polluted air. For example, it was reported that
171 vegetation debris accounted for only ~1% of $\text{PM}_{2.5}$ OC in Beijing, China (Guo et al., 2012). Moreover,
172 although soil dust may contribute to the primary inorganic aerosol at Beijing (Huang et al., 2014a), its
173 impact on CAs is negligible due to the low carbon content of soil. Thus, this fraction of OC was ignored in

174 our study. The fraction of POC_{bb} can be estimated using the concentration of Lev, which is an excellent
175 molecular marker for biomass burning, with the hypothesis that the $\text{POC}_{\text{bb}}/\text{Lev}$ value is maintained at a
176 stable level during transportation from emission sources to the ambient environment. In the real atmosphere,
177 Lev is gradually oxidized and degraded (Hoffmann, et al., 2009). However, this decay occurs mainly in
178 typical summer conditions, with 1×10^6 molecules cm^{-3} of hydroxyl radicals and independent of relative
179 humidity, according to smog chamber results (Hennigan et al., 2010). Our study, however, was performed
180 in spring with lower temperatures (lower hydroxyl radical levels), and air masses containing biomass
181 burning-derived particles were derived mainly from the sub-urban/rural areas around the sampling locations
182 (relatively shorter air transportation distances). Thus, it was appropriate to use Lev to estimate POC_{bb} in
183 this study.

$$184 \quad \text{POC}_{\text{bb}} = \text{Lev} \times (\text{POC}/\text{Lev})_{\text{bb}}$$

185 $(\text{POC}/\text{Lev})_{\text{bb}}$, the ratio of primary OC to Lev in the pure biomass burning particles, is closely related to
186 the type of biomass burning (Fine et al., 2002). According to information from previously reported emission
187 inventories, the mean values of $(\text{POC}/\text{Lev})_{\text{bb}}$ for hardwood, softwood, and annual plants are 7.76, 14.2 and
188 13.5 with uncertainties of 19%, 22%, and 22%, respectively (Fig. S3).

189 OC directly emitted from fossil fuel combustion is, in principle, water-insoluble (Weber et al., 2007; Zhang
190 et al., 2014). This was further confirmed in a recent study performed by Dai et al. (2015), who found >90%
191 of traffic tunnel-emitted OC is water-insoluble, implying that primary fossil organic carbon (POC_{f}) should
192 be nearly exclusively water-insoluble, if considering the limited but existing SOC in the traffic tunnel.
193 Therefore, POC_{f} is estimated by the fraction of WIOC_{f} .

$$194 \quad \text{POC}_{\text{f}} = \text{WIOC} \times (1 - f_{\text{c}}(\text{WIOC}))$$

195

196 **2.6.3 Secondary Organic Carbon**

197 Non-fossil OC is composed of POC_{bb} and non-fossil SOC (SOC_{nf}) neglecting any contribution from plant
198 debris. Thus, SOC_{nf} can be estimated as follows:

$$199 \quad \text{SOC}_{\text{nf}} = \text{OC}_{\text{nf}} - \text{POC}_{\text{bb}}$$

200 It should be noted that the VOC precursors of SOC_{nf} can originate from both biogenic emissions and
201 biomass burning.

202 Studies have found that fossil WSOC can serve as an accurate proxy for fossil SOC (SOC_{f}) (Liu et al., 2014;
203 Zhang et al., 2014), which was therefore calculated by the following:

$$204 \quad \text{SOC}_{\text{f}} = \text{WSOC} \times (1 - f_{\text{c}}(\text{WSOC}))$$

205 Some fossil SOC may exist in water-insoluble phase. In Tokyo, Japan, Miyazaki et al. (2006) estimated that
206 <30% of SOC is water-insoluble, mainly representing fresh SOC (Favet et al., 2008).. It is very likely that
207 practically all SOC species would exist in water-soluble phases, if they underwent sufficient atmospheric
208 reaction. Note that a small part of fresh SOC that exists in the water-insoluble phase is not taken into account
209 for the estimation of fossil SOC, although samples were collected every 24 hours in this study.

210

211 **3 Results and Discussion**

212 **3.1 PM_{2.5} and Chemical Composition**

213 Extremely high PM_{2.5} mass concentrations are detected in both BJ (74.7–418 $\mu\text{g}/\text{m}^3$, mean 218 $\mu\text{g}/\text{m}^3$) and
214 GZ (46.1–145 $\mu\text{g}/\text{m}^3$, mean 90.6 $\mu\text{g}/\text{m}^3$) (Fig. 1), compared to the Chinese first-grade air quality standard
215 (35 $\mu\text{g}/\text{m}^3$, 24-hour) and the World Health Organization (25 $\mu\text{g}/\text{m}^3$, 24-hour). Such elevated PM_{2.5} loadings
216 can be vividly observed from space (Fig. S2). In this study, the average PM_{2.5} concentration in BJ is 2.4-
217 fold higher than that in GZ, indicating poorer air quality over the North China region, which has been found
218 in previous investigations (Cao et al., 2012) and is also consistent with the fact that BJ and the adjacent
219 suburban/rural areas (North China Plain) have the world's highest population density. Organic matter (OM)

220 is the largest contributor to $PM_{2.5}$ in both cities (Fig. 1), followed by NO_3^- (13%), SO_4^{2-} (9.5%), NH_4^+
221 (5.8%), Ca^{2+} (2.6%), Cl^- (2.6%), and EC (1.1%) in BJ and SO_4^{2-} (16%), NO_3^- (10%), NH_4^+ (7.6%), EC
222 (3.2%), and Cl^- (2.0%) in GZ. OM calculations are based on the relative contributions of WSOC and WIOC
223 in OC, due to the different conversion factors for WIOC to WIOM (1.3) and WSOC to WSOM (2.1) (Yttri
224 et al., 2007; Favez et al. 2009; Sun et al. 2011; Chen et al. 2014). OM is the sum of WIOM and WSOM.
225 The proportions of the other measured chemicals are generally lower than 1%. In this study, ~40-50% of
226 $PM_{2.5}$ are unidentified, which is relatively higher than that (~10-36%) from a recent study performed in
227 China (Huang et al. 2014a) and probably because we didn't measure elementals and their oxidants in $PM_{2.5}$.
228 For example, Chen et al. (2001) found ~35-40% of $PM_{2.5}$ can't be identified in Taiwan without measuring
229 some constituents such as Cu, Cd, and As. In addition, the water absorbed by organics would also influence
230 the proportion of unidentified fraction in particles (Andrews et al. 2000). Different chemical compositions
231 are largely reflective of their different source characteristics and the corresponding atmospheric processes.
232 For example, the higher concentrations of Ca^{2+} recorded in BJ in this study are due to the extensive dust
233 storms originating from deserts in northwestern regions that strongly impact North China in spring (He et
234 al., 2001; Zheng et al., 2005). Road and construction fugitive dust also can lead to a relative high Ca^{2+} in
235 $PM_{2.5}$, while this source may mainly influences Beijing in summer and is insignificant in spring (Zheng et
236 al. 2005). In addition, biomass burning activities, using domestic biofuel, are generally more widespread in
237 North China (Liu et al., 2007), which is reflected by the higher levels of biomass burning markers (Lev and
238 K^+) in BJ (Table 1).

239

240 **3.2 Radiocarbon Results**

241 Although the chemical compositions of $PM_{2.5}$ can largely be identified by state-of-the-art analytical
242 technology (Huang et al., 2014a), no a robust method of performing direct and precise $PM_{2.5}$ source

243 apportionment exists due to the complexity of the emission sources and atmospheric processes.
244 Carbonaceous aerosols are not only the most important fractions of PM_{2.5}, but also the main factors that
245 significantly lead to severe air pollution and deterioration in atmospheric visibility (Cao et al., 2003; Cao
246 et al., 2007; Deng et al., 2008; Yang et al., 2011) owing to their strong ability for scattering and absorbing
247 visible light (Bond et al., 2006; Kanakidou et al., 2005). Source apportionment based on ¹⁴C analysis has
248 the potential to unambiguously separate the various carbon species into fossil (¹⁴C-free) and non-fossil
249 fractions (modern ¹⁴C level) (Gustafsson et al., 2009; Huang et al., 2014a; Kirillova et al., 2013; Liu et al.,
250 2013b; Szidat et al., 2004; Weber et al., 2007; Wozniak et al., 2012; Lewis et al., 2004), providing significant
251 information on the PM_{2.5} sources and corresponding atmospheric processes that they have undergone. Our
252 results show that non-fossil emissions represent a significant portion of the TC in both North and South
253 China: 56±4% in BJ and 46±5% in GZ, respectively (Table 2). Similar to previous studies conducted in
254 cities around the world (Andersson et al., 2015; Bernardoni et al., 2013; Liu et al., 2013b; Szidat et al.,
255 2009), EC is derived mainly from fossil-fuel combustion in urban regions. In the remote areas, biomass
256 burning plays a more role in EC loading. For example, ~50-70% of EC were found come from the burning
257 of biomass materials in Arctic (Barrett et al. 2015) and the nature protection area of China (Zhang et al.
258 2014). On average, a larger proportion of biomass burning in the EC in BJ (33±7%) relative to GZ (20±5%)
259 indicates that the emissions of biomass burning, e.g., biofuel burning and outdoor fires in North China, are
260 higher than those in South China. This result is consistent with the other observations in this study: the
261 mean concentrations of Lev and K⁺ in BJ are higher than in GZ by factors of 1.4 and 1.7, respectively
262 (Table 1). More than half of the OC fraction is contributed by non-fossil sources in both BJ (59±4%) and
263 GZ (54±6%). Although the ¹⁴C levels in the OC in BJ and GZ have similar ranges, the cities differ
264 considerably in the species of WIOC and WSOC. On average, the percentage of non-fossil carbon in WIOC
265 in BJ (74±8%) is 23% higher than that in GZ (51±2%). This result is probably explained by the large

266 difference between North and South China in the types of biomass burned. In Beijing and its adjacent
267 regions, annual plants, e.g., crop residues and agricultural grass, are the main materials involved in biomass
268 burning activities (Cheng et al., 2013; Duan et al., 2004; Li et al., 2008; Li et al., 2010; Yan et al., 2006),
269 Whereas, 2000 km south from BJ, in GZ, the types of biomass are marked by hardwood (Liu et al., 2014),
270 which is further confirmed by the compositions of anhydrosugars in this study (Fig. S3). Compared to the
271 hardwood burning (WIOC/OC = 79%), OC aerosols emitted directly from combustion of annual plants
272 appeared to be more enriched in WIOC species (WIOC/OC = 97%) (Iinuma et al., 2007), which is probably
273 the most important factor in the markedly higher ^{14}C levels in WIOC in BJ compared to those in GZ. In
274 contrast, WSOC in BJ is more depleted in ^{14}C than that in GZ: $46\pm 13\%$ of WSOC in BJ and $60\pm 11\%$ in GZ
275 is directly correlated with non-fossil sources. WSOC is regarded as a mixture of mainly SOC and POC_{bb}
276 (Ding et al., 2008; Weber et al., 2007). Such a difference in the WSOC- ^{14}C levels between the two cities
277 could be attributed mainly to the origins and formation processes of the SOC, of which will be discussed
278 below.

279

280 **3.3 Source Apportionment**

281 A source apportionment model for carbonaceous aerosols using the combined measurements of ^{14}C and
282 biomass burning tracers was recently reported (Liu et al., 2014). Detailed information with respect to this
283 model provided in the Methodology section of the paper. Significantly distinct characteristics between BJ
284 and GZ are observed for the source apportionments of TC (Fig. 3). The largest contributor to TC in BJ is
285 SOC_{nf} ($28\pm 2\%$), followed by SOC_{f} ($26\pm 5\%$), POC_{bb} ($24\pm 1\%$), POC_{f} ($10\pm 4\%$), EC_{f} ($8\pm 2\%$), and EC_{bb}
286 ($4\pm 1\%$), whereas in GZ, an order of SOC_{nf} ($31\pm 2\%$) > POC_{f} ($22\pm 4\%$) > EC_{f} ($18\pm 3\%$) > SOC_{f} ($15\pm 9\%$) >
287 POC_{bb} ($10\pm 4\%$) > EC_{bb} ($4\pm 1\%$) is observed. Consequently, the fraction of primary sources, i.e., the sum of
288 POC and EC, was higher in GZ ($54\pm 9\%$) than that in BJ ($46\pm 4\%$). The variability of the source

289 apportionment results represent the 1σ standard deviation of the individual sources during the study. It
290 should be noted that model uncertainties of the contributors mainly depend on correction factors such as
291 $(\text{POC}/\text{Lev})_{\text{bb}}$ emission ratios of wood-burning markers and conversion factors for determination of the
292 fraction of contemporary carbon from ^{14}C analysis. Typical relative model uncertainties were recently
293 estimated by Zhang et al. (2015) using a similar model approach as 20-25% for SOC_{nf} , SOC_{f} , POC_{bb} , and
294 POC_{f} as well as $\sim 13\%$ for EC_{f} , and EC_{bb} .

295 POC and EC aerosols are independent of the conditions of atmospheric gas reactions and thus can directly
296 reflect the characteristics of local emission sources. The total proportions of EC_{f} and POC_{f} in GZ ($40\pm 6\%$)
297 are significantly higher than those in BJ ($19\pm 5\%$), indicating high emissions from fossil-fuel combustion in
298 GZ. The ratios of POC_{f} to EC_{f} are similar in both cities: 1.3 ± 0.42 in BJ and 1.2 ± 0.26 in GZ. These fossil
299 fuel-derived POC/EC ratios in ambient air are considerably lower than the ratios directly derived from
300 industrial coal combustion (2.7-6.1) (Zhang et al., 2008) but close to those of traffic exhaust (0.5-1.3) (He
301 et al., 2008; Zhou et al., 2014). The total proportions of EC_{bb} and POC_{bb} in the TC accounted for $28\pm 1\%$
302 and $15\pm 5\%$ in BJ and GZ, respectively, which confirmed the greater impact of biomass burning on regional
303 air pollution in North China.

304 Of SOC, $52\pm 5\%$ and $71\pm 11\%$ are derived from non-fossil sources in BJ and GZ, respectively. Using
305 multi-technique integrated methods, Huang et al. (2014a) found that 35-54% and 65-85% of the SOC were
306 derived from non-fossil sources in BJ and GZ, respectively, which is consistent with our results. These
307 findings underline the importance of the non-fossil contribution to SOC formation in China's megacities.
308 The considerable difference in SOC composition between the two cities might be due to the significant
309 difference in SOC precursors and corresponding atmospheric reaction processes. In spring (from March to
310 early April), in North China, the land is bare and trees still leafless, whereas, in South China there is lush
311 vegetation, which release non-fossil VOCs (isoprene, α -pinene, β -caryophyllene, etc.), which are SOC

312 precursors. Additionally, a close relationship is observed between POC_{bb} and SOC_{nf} in BJ ($R^2 = 0.99$) but
313 not in GZ ($R^2 = 0.50$) (Fig. 2), indicating the predominant role of biomass burning-derived VOCs in SOC_{nf}
314 formation in North China. Combined with the fact that the total proportion of EC_{bb} and POC_{bb} in TC in BJ
315 is almost twofold higher than that in GZ, efforts to control biomass burning, such as fugitive open fires,
316 burning of agricultural residue, and domestic cooking/heating, are likely the most means of mitigating haze
317 pollution in the North China region. In addition to the difference in the SOC precursors, the difference in
318 meteorological conditions between the two cities is another likely reason for the difference in SOC
319 composition. Compared to fossil-derived SOC, non-fossil SOC would preferentially formed in a higher
320 humidity environment (Liu et al., 2014; Favez et al., 2008). The relative humidity in GZ (79%, 66-95%) is
321 much higher than that in BJ (48%, 19-79%) during the sampling campaign.

322

323 **3.4 Source Dynamics of Carbonaceous Aerosols during Haze Process**

324 Particulate-derived haze pollution is characterized by an elevated $\text{PM}_{2.5}$ mass concentration due to the rapid
325 physical accumulation and intensive atmospheric reactions. Knowledge of the source dynamics of the fine
326 particles is crucial to understanding haze pollution.

327 An integrated haze process is observed in GZ from Apr.11–17 (Fig. S4). From Apr. 11 ($65 \mu\text{g}/\text{m}^3$), the
328 $\text{PM}_{2.5}$ concentration increases sharply at an average rate of $34 \mu\text{g}/\text{m}^3/\text{day}$ for two days, then reaches a
329 plateau from Apr. 13-15 ($132\text{-}145 \mu\text{g}/\text{m}^3$), and fall sharply to $76 \mu\text{g}/\text{m}^3$ on Apr. 17 at a rate of $35 \mu\text{g}/\text{m}^3/\text{day}$.
330 To illustrate this haze bloom-decay process, all component concentrations (C) are normalized to the
331 concentration on the first day (C_0). The concentrations for these components are listed in Table S1. As
332 shown in Fig. 4C, both total primary and secondary matter concentrations increase by a factor of ~ 1.5 from
333 the first day to the second day (blooming stage), indicating that direct emissions and atmospheric reactions
334 played the similar roles in $\text{PM}_{2.5}$ growth during this phase. Notably, primary fossil organic matter (POM_{f}),

335 EC_f and NO₃⁻ have the highest formation rates, with the values of C/C₀ > 2.0 (Fig. 4A, B), implying greater
336 contribution of vehicle exhaust to the fine particles since NO₃⁻ is an indicator of traffic emissions. At the
337 haze outburst (the third day also the initial day of haze), the growth rates of total primary and secondary
338 matter diverge significantly, and the C/C₀ values are 1.6 and 2.6, respectively, suggesting that atmospheric
339 reactions started to play a more important role in the particle growth than direct emission-derived particles.
340 The growth of POM_f, EC_f, secondary fossil organic matter (SOM_f) and NO₃⁻ (C/C₀ > 3.0) are more rapid
341 than that of the other components, again indicating the important role of vehicle emissions. On the last day
342 of the haze episode, all primary matter C/C₀ values markedly decrease, with the exception of unidentified
343 materials. This phenomenon could be explained by the rapid shift in the pollutant sources transported by
344 the air masses. As shown in Fig.1, air masses collected at the fifth day originated from southern GZ, a major
345 area of anthropogenic pollutant emission (Zheng et al., 2009) incorporating large-scale industry city
346 districts, such as Foshan, Zhongshan and Dongguan. These industry-derived air masses are likely enriched
347 with those materials that are not identified in this study, such as mineral dust. Furthermore, the C/C₀ values
348 for industry-related secondary matter such as SO₄²⁻ (an indicator for industrial emissions) and SOM_f showed
349 a clear increase on the same day. The reason for the decline in the C/C₀ values of SOM_{nf} is probably due to
350 the dilution of biogenic/biomass burning VOCs with the invasion of these industry-related air masses. All
351 of the C/C₀ values of the secondary aerosols decrease markedly on the seventh day (Fig. 4A, B), indicating
352 that the atmospheric reactions has weakened in the post-haze phase, due in part to the scavenging effect of
353 the precipitation that occurred on Apr. 16 on the aged aerosols (Fig. S1).

354 In BJ, PM_{2.5} concentrations remain high during the sampling period, and no clear bloom/decay haze
355 process similar to that observed in GZ is captured. It should be noted, however, that samples for the ¹⁴C
356 measurements are not collected on consecutive days in BJ, as is case for GZ. Therefore, C/C₀ values are
357 plotted in Fig. 4 along the increasing trend of PM_{2.5} concentrations. After comparing the characteristics of

358 relative lower and higher $PM_{2.5}$ loadings, a recent study revealed that severe haze pollution in North China
359 is controlled by the secondary matters (Huang et al., 2014a). Our results confirm this conclusion and find
360 an anomalous relationship between $PM_{2.5}$ level and secondary matter (Fig. 4F), pointing out that
361 atmospheric reactivity is not sufficient for the initiation of strong haze events. To a large extent, $PM_{2.5}$ and
362 secondary aerosols in North China is depending on meteorological conditions and the origins of air masses
363 (Guo et al., 2014). $PM_{2.5}$ pollution on Mar. 25 is more severe than that on Mar.16, while the C/C_o of total
364 secondary matter Mar. 16 is much higher than Mar. 25 (Fig. 4F). Most air masses reaching on Mar.16 come
365 from the most polluted region in North China, i.e. southern BJ (Fig. 1 and Fig. S2). These migrated aerosols
366 would underwent lots of atmospheric reactions and aged process. On other hand, NO_3^- displays the highest
367 growth rate among all of the types of matter, indicating that traffic emissions contribute most to the air
368 pollution. However, SOM_{nf} , which in this study is mostly derived from biomass burning (section 3.3), plays
369 a more important role than SOM_f (Fig. 4E), implying the importance of biomass burning in haze formation
370 in BJ. This is much different from that in South China (Fig. 4B). Furthermore, the NH_4^+ C/C_o along the
371 increase of $PM_{2.5}$ loading in BJ is ~5-10 times as high as that in GZ. This is probably because North China,
372 especially those regions in southern BJ (Heibei, Henan and Shandong), has the most intensive NH_4^+
373 emissions from fertilizer and livestock in China (Huang et al., 2012; Zhang et al., 2010b). In addition, it is
374 reported that the emission factor of NH_4^+ from annual plant burning (47 mg/kg) is found to be
375 approximately fivefold that of hardwood burning (10 mg/kg) (Iinuma et al., 2007). Therefore, the much
376 higher NH_4^+ growth rate in BJ compared to that of GZ may partly be attributed to the importance of
377 agricultural residue burning in the North China.

378

379 **4 Conclusions**

380 Severely high loadings of carbonaceous aerosols (CAs) regarding the deterioration of air quality, risk of

381 human health, and abnormal change of climate system in Chinese megacities has drawn a lot of scientific
382 and public attentions. Through the combined measurements of powerful sources tracers (radiocarbon and
383 anhydrosugars), this study reveals the significant differences of the origins of various CAs in the megacities
384 of North (Beijing) and South China (Guangzhou). The contribution of non-fossil sources (e.g., domestic
385 heating and cooking) to total carbon (TC), organic carbon (OC), water-soluble OC (WSOC), water-
386 insoluble OC (WIOC), and elemental carbon (EC) is $56\pm4\%$, $59\pm4\%$, $46\pm13\%$, $74\pm8\%$, and $33\pm7\%$ in
387 Beijing, and $46\pm5\%$, $54\pm6\%$, $60\pm11\%$, $51\pm2\%$, and $20\pm5\%$ in Guangzhou, respectively. Overall, non-fossil
388 sources play a more important role in CAs in North China than South China. Lower contribution of non-
389 fossil sources to secondary OC in Beijing than Guangzhou is largely because the much lower humidity and
390 limited biogenic volatile organic compounds in North China during the sampling campaign. The air
391 pollution controls in China probably should be enacted and performed according to the local circumstances.
392 Finally, we find that primary aerosols play an equal important role on the haze blooming phase as secondary
393 aerosols in South China, yet nitrate and fossil secondary organic matter predominate in the haze stage.

394

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580 Table 1 Dataset (average values with standard deviation) for the measured components in this study. Units
 581 for the carbon fractions, anhydrosugars and ions are $\mu\text{g C/m}^3$, ng/m^3 , $\mu\text{g/m}^3$, respectively.

	Beijing (North China) (Mar. 15 – Apr. 12, 2013)		Guangzhou (South China) (Apr. 04 – Apr. 18, 2013)	
	Average	Std.	Average	Std.
WIOC	9.07	4.88	5.99	3.37
WSOC	10.2	5.76	4.31	2.25
OC	19.2	10.3	10.3	5.13
EC	2.56	1.68	3.04	1.42
TC	21.8	11.9	13.3	6.47
Gal	11.7	7.78	5.75	2.51
Mann	10.3	6.24	11.5	5.47
Lev	369	249	259	172
Na ⁺	0.83	0.32	0.34	0.15
NH ₄ ⁺	13.0	10.6	7.01	3.05
K ⁺	1.43	0.90	0.82	0.34
Mg ²⁺	0.48	0.19	0.07	0.03
Ca ²⁺	5.63	2.54	0.60	0.36
Cl ⁻	5.62	3.30	1.91	1.68
NO ₃ ⁻	31.1	27.9	10.2	6.99
SO ₄ ²⁻	20.8	14.1	13.9	3.87

582 WIOC: water-insoluble organic carbon; WSOC: water-soluble organic carbon; OC: organic carbon; EC:
 583 elemental carbon; TC: total carbon; Gal: galactosan; Mann: mannosan; Lev: levoglucosan;

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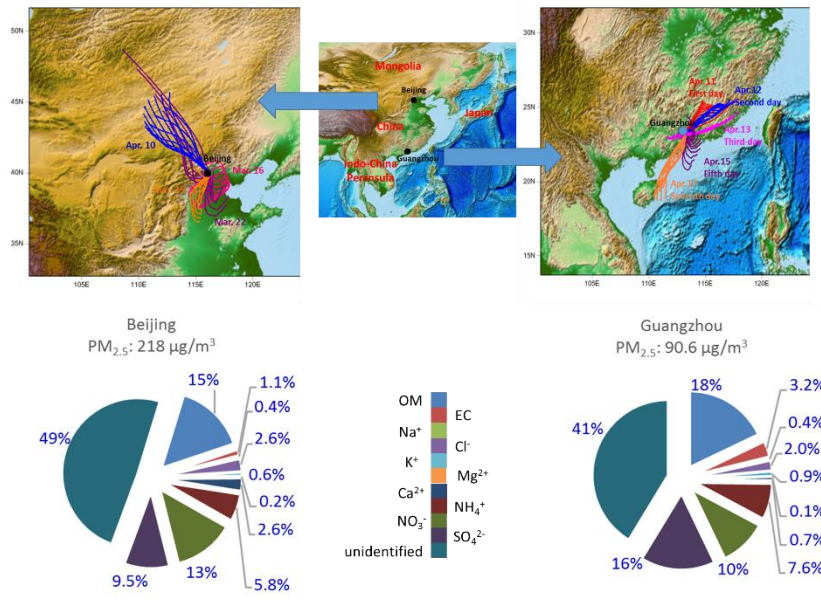
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586 Table 2 Percentages of non-fossil sources in various carbon species (%)

Location	Sampling Data	OC	WIOC	WSOC	EC	TC	SOC
Beijing (BJ, North China)	Mar.16	58±4	67±3	52±3	34±2	55±4	52±11
	Mar.22	54±5	83±4	53±3	33±2	60±5	55±11
	Mar.25	59±4	65±3	52±2	24±1	54±4	58±13
	Apr.10	54±4	82±4	22±1	43±2	53±4	43±11
	Average	59±4	74±8	46±13	33±7	56±4	52±5
Guangzhou (GZ, South China)	Apr.11	59±4	52±2	68±3	28±1	52±4	74±10
	Apr.12	53±4	50±2	60±3	21±1	46±4	77±10
	Apr.13	57±4	53±3	63±3	19±1	49±7	75±8
	Apr.15	42±3	48±2	39±2	18±1	38±3	49±5
	Apr.17	58±4	50±2	68±3	12±1	45±4	78±6
Average	54±6	51±2	60±11	20±5	46±5	71±11	

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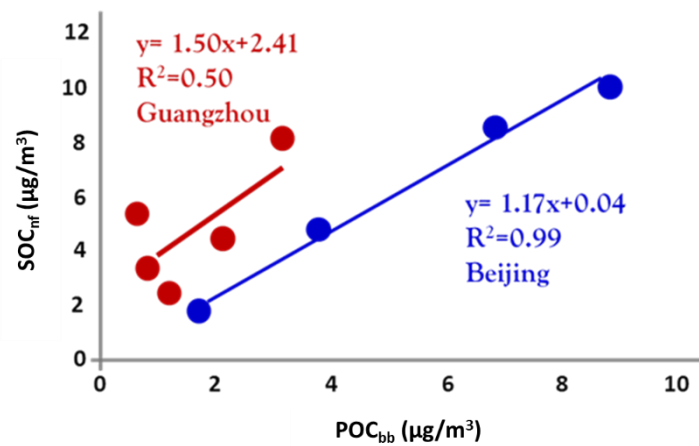


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590 Fig. 1 Geographic locations of Beijing and Guangzhou as well as their PM_{2.5} chemical compositions. Air
 591 mass back trajectories within 24 hours (run every 2 hours from the end of sampling) for the selected samples
 592 are modeled at 100 m above ground level by Air Resources Laboratory, National Oceanic and Atmospheric
 593 Administration (Hybrid Single Particle Lagrangian Integrated Trajectory Model).

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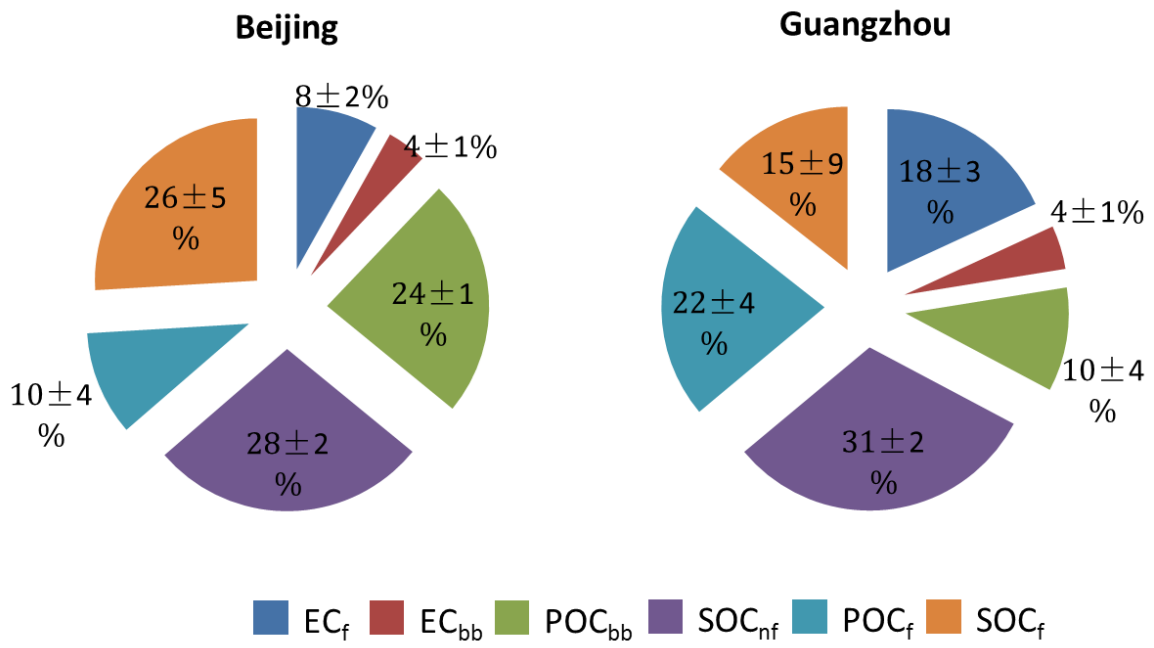


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597 Fig.2 Correlations between the primary biomass burning OC (POC_{bb}) and the non-fossil secondary organic
 598 carbon (SOC_{nf})

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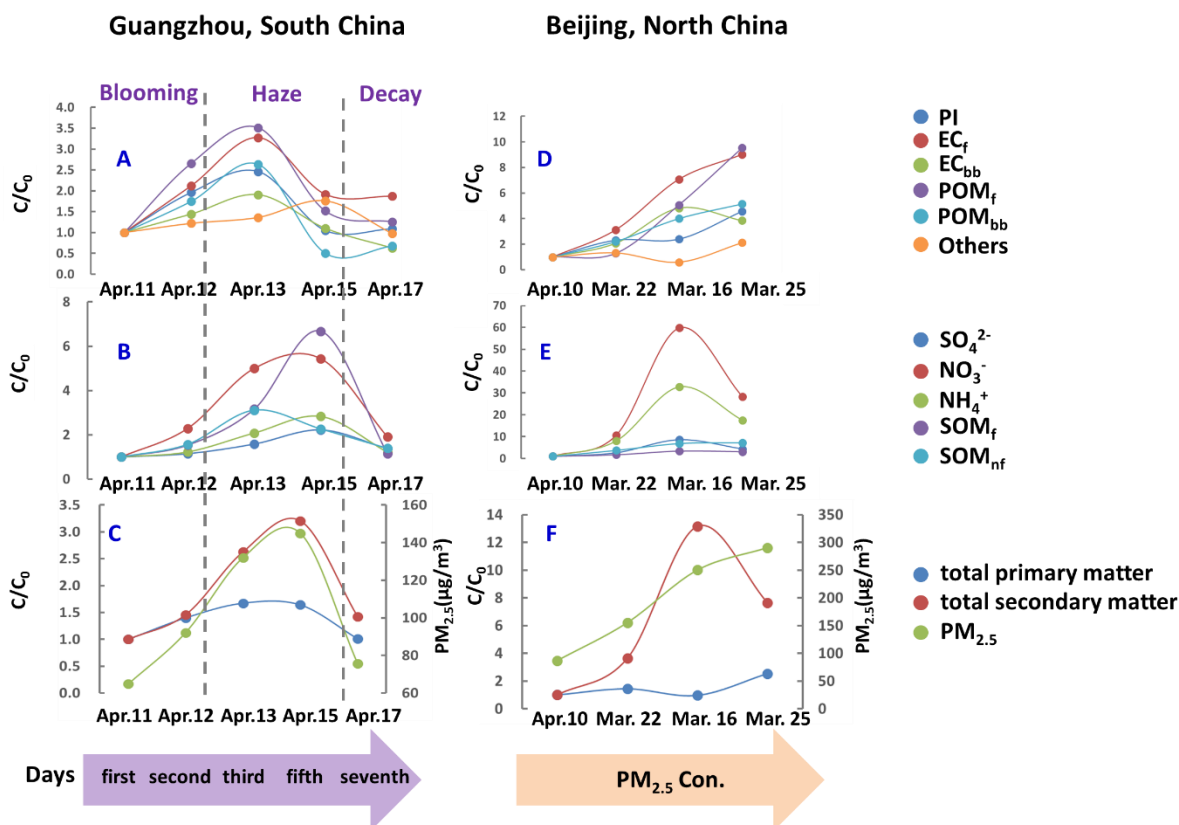
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Fig.3 Relative contributions (%) of different carbon species to total carbon with variability of the individual sources during the study. For model uncertainties of the individual fractions see Chapter 3.3.



606

607 Fig.4 Dynamic changes for the primary and secondary matters in Chinese cities. Detailed concentrations
 608 and the calculation methods are shown in Table S2. The first day is Apr. 11 in Guangzhou. For Beijing,
 609 samples are ordered by their $PM_{2.5}$ concentrations. All concentrations (C) are normalized to that of a
 610 reference day (C_0), i.e. the first day and the day with the lowest $PM_{2.5}$ concentration for Guangzhou and
 611 Beijing, respectively. PI: primary ions; POM_f : primary fossil organic matter; POM_{bb} : biomass burning
 612 organic matter; SOM_f : secondary fossil organic matter; SOM_{nf} : secondary non-fossil organic matter.