# 1 Sources and Chemical Characterization of Organic Aerosol during the Summer in the 2 **Eastern Mediterranean** 3 4 Evangelia Kostenidou<sup>1,2</sup>, Kalliopi Florou<sup>1,2</sup>, Christos Kaltsonoudis<sup>1,2</sup>, Maria 5 Tsiflikiotou<sup>1,2</sup>, Stergios Vratolis<sup>3</sup>, Konstantinos Eleftheriadis<sup>3</sup>, and Spyros N. Pandis<sup>1,2,4</sup> 6 7 8 <sup>1</sup>Institute of Chemical Engineering Sciences, ICE-HT, Patras, Greece 9 <sup>2</sup>Department of Chemical Engineering, University of Patras, Patras, Greece 10 <sup>3</sup> ERL Institute of Nuclear and Radiological Science & Technology, Energy & Safety, 11 12 NCRS Demokritos, Attiki, Greece <sup>4</sup>Department of Chemical Engineering, Carnegie Mellon University, Pittsburgh, USA 13 14 15 16 17 **Abstract** 18 The concentration and chemical composition of non-refractory fine particulate 19 matter (NR-PM<sub>1</sub>) and black carbon (BC) levels were measured during the summer of 20 2012 in the suburbs of two Greek cities, Patras and Athens, in an effort to better 21 understand the chemical processing of particles in the high photochemical activity 22 environment of the Eastern Mediterranean. The composition of PM<sub>1</sub> was surprisingly 23 similar in both areas demonstrating the importance of regional sources for the corresponding pollution levels. The PM<sub>1</sub> average mass concentration was 9-14 µg m<sup>-3</sup>. 24 The contribution of sulphate was around 38%, while organic aerosol (OA) contributed 25 26 approximately 45% in both cases. PM<sub>1</sub> nitrate levels were low (2%). The oxygen to 27 carbon (O:C) atomic ratio was 0.50±0.08 in Patras and 0.47±0.11 in Athens. In both cases 28 PM<sub>1</sub> was acidic. 29 Positive matrix factorization (PMF) was applied to the high resolution organic 30 aerosol mass spectra obtained by an Aerodyne High Resolution Time-of-Flight Aerosol

Mass Spectrometer (HR-ToF-AMS). For Patras five OA sources could be identified: 19%

very oxygenated OA (V-OOA), 38% moderately oxygenated OA (M-OOA), 21% biogenic oxygenated OA (b-OOA), 7% hydrocarbon-like OA (HOA-1) associated with traffic sources and 15% hydrocarbon-like OA (HOA-2) related to other primary emissions (including cooking OA). For Athens the corresponding source contributions were: V-OOA (35%), M-OOA (30%), HOA-1 (18%) and HOA-2 (17%). In both cities the major component was OOA, suggesting that under high photochemical conditions most of the OA in the Eastern Mediterranean is quite aged. The contribution of the primary sources (HOA-1 and HOA-2) was important (22% in Patras and 35% in Athens) but not dominant.

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#### 1. Introduction

Atmospheric aerosols can affect human health by causing cardiovascular and respiratory problems (Davidson et al., 2005; Pope and Dockery, 2006), they reduce visibility (Watson, 2002) and influence the energy balance of our planet (IPCC, 2013) by scattering or absorbing radiation and changing cloud reflectivity and lifetime. Submicrometer atmospheric particles mainly consist of sulphates, ammonium, organic matter, nitrates, elemental carbon and metals. Organic aerosol (OA) represents a major fraction of the submicron aerosol mass (Kanakidou et al., 2005; Zhang et al., 2007). The recent development of the Aerodyne HR-ToF-AMS (DeCarlo et al., 2006) allows high time resolution size-resolved measurements of the fine non-refractory inorganic and organic aerosol components. In addition, several techniques have been developed for the deconvolution of the AMS organic mass spectra (Zhang et al., 2011) including custom principal component analysis (Zhang et al., 2005), multiple component analysis (Zhang et al., 2007), and positive matrix factorization (PMF) (Paatero and Tapper 1994; Lanz et al., 2007). The most recent algorithm, the multilinear engine, (ME-2) (Lanz et al., 2008, Canonaco et al., 2013) is a hybrid of chemical mass balance (CMB) and bilinear models (e.g., PMF). The combination of AMS measurements with other instrumentation and the use of the corresponding source apportionment techniques can provide valuable information about the ambient aerosol sources and their chemical characterization. In this study the technique of PMF analysis on HR-ToF-AMS mass spectra is used.

Zhang et al. (2005) showed that summertime OA in a major urban area of the US (Pittsburgh) consisted of oxygenated OA (OOA) and hydrocarbon-like OA (HOA). Lanz et al. (2007) further deconvolved OOA into a more oxygenated OA (OOA-1) and a less oxygenated (OOA-2) component during a summer period in Zurich. They also reported for the fist time a wood burning and a charbroiling factor. Marine OA (MOA) was identified by Crippa et al. (2013a) in Paris during summertime, while Schmale et al. (2013) suggested the existence of an amino-acids/amine OA factor (AA-OA), a methanesulfonic acid OA factor (MSA-OA), a marine oxygenated OA factor (M-OOA) and a sea spray OA factor (SS-OA) on Bird Island in the South Atlantic. The major OA components found usually are OOA and HOA (Ng et al., 2010). In our work the OA sources are investigated, in order to characterize two urban environments in the Eastern Mediterranean.

Most of the studies on air quality in the Eastern Mediterranean and Greece have been based on filter measurements (Manoli et al., 2002; Grivas et al., 2004; Papaefthymiou et al., 2005; Karanasiou et al., 2007; Chrysikou and Samara, 2009) or monitoring PM mass concentrations (Chaloulakou et al., 2003; Vardoulakis and Kassomenos, 2008; Yannopoulos, 2008) in major Greek cities (Athens, Thessaloniki, Patras). During the past years a few field campaigns using continuous PM composition measurement techniques have been conducted in rural/remote areas and urban centres in Greece. At the remote site of Finokalia (Crete) factor analysis on filter samples revealed three sources of coarse particles (crustal, photochemical, and marine aerosols) and two additional factors in the fine mode (residual oil and secondary/combustion aerosols) (Koulouri et al., 2008). For the same site no HOA was detected by PMF analysis of Quadropole AMS data either in the summer or in the winter (Hildebrandt et al., 2010; 2011). For the city of Patras Pikridas et al. (2013) found that the transported pollution accounted for 50% of the PM<sub>2.5</sub> during the winter and more than 70% during the rest of the year. However, the Eastern Mediterranean urban environment remains not well characterized.

In the Western Mediterranean there have been more measurement campaigns (e.g., Viana et al., 2005; Pérez et al., 2008; Pey et al., 2010; Pandolfi et al., 2014). In Barcelona primary OA accounted for 59% of the OA in the late winter (Mohr et al., 2012). El

93 Haddad et al. (2013) reported that OOA contributed 80% of the OA mass in Marseille 94 during the summer, while only 5% of the OA was of industrial origin. Nicolas (2013) 95 found that OOA ranged from 70% to 85% of the total OA over a year at the Cape Corsica 96 station. For the same site the oxidation state of the OA during the summer reached 97 extremely high values with  $f_{44}$  (the fraction of the AMS m/z 44, which is an indicator of 98 the oxidation state) higher than 0.25. Thus the measurements in Western Mediterranean 99 have covered better the different types of environments compared to Eastern 100 Mediterranean.

Despite the previous efforts, there are still several knowledge gaps related to the characterization and OA sources in the Mediterranean Basin. For example the contribution of the primary sources is still uncertain. Fresh OA from diesel exhaust initially resembles HOA but after aging may resemble semi-volatile OOA (SV-OOA) (Jimenez et al., 2009; Chirico et al., 2010). Aged gas phase emissions from diesel engines, scooters and biomass burning may also produce secondary OA (SOA) with mass spectra similar to SV-OOA (Heringa et al., 2012). In addition, volatile organic compounds (VOCs) in chamber experiments form SOA with mass spectra close to SV-OOA (Ng et al., 2010). The conversion of SV-OOA to more oxidized OA is still quite difficult to reproduce in the laboratory. Only recently Platt et al. (2013) showed that after 12 hours of aging of gasoline Euro 5 car emissions, SOA with an O:C ratio of 0.7 was produced resembling LV-OOA. In the field, Hildebrandt et al. (2010) demonstrated that primary emissions are becoming highly oxygenated within 1-2 days of transport under intense photochemical conditions. Recently Bougiatioti et al. (2014) showed that a large fraction of biomass burning OA (BBOA) is transformed to OOA even in the dark in less than a day during the summer in Eastern Mediterranean. The objectives of this work are to:

- 1) characterize the chemical composition of the PM<sub>1</sub> in two Greek cities in the Eastern
- 118 Mediterranean using high time resolution instrumentation
- 119 2) determine the corresponding OA sources and evaluate the relative significance of the
- primary versus the oxygenated factors
- 121 3) provide insights about the chemical processing of the primary PM
- 4) compare the mass spectra profiles to the corresponding profiles extracted in previous
- studies and to

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124 5) assess the overall PM<sub>1</sub> pollution in the Eastern Mediterranean in comparison with other

125 European cities.

#### 2 Experimental

### 2.1 Measurement campaign

The measurements presented here were part of a larger study, which involved measurements in several areas in Greece (Patras, Athens, Thessaloniki, and Finokalia) both in summer and winter. In this work we will focus on the summer HR-ToF-AMS measurements in Patras and Athens. More information about the summer study is provided in Tsiflikiotou et al. (in preparation).

# 2.2 Sampling Sites in Patras

Patras has approximately 300,000 inhabitants and is located at the Gulf of Patras, at the foothills of a 2 km high mountain. The major anthropogenic activities include a small industrial zone about 16 km southwest of the city center and a harbor around 2.5 km southwest of the city. The nearest major city is Athens, around 220 km to the east. Measurements in Patras were performed simultaneously at two locations: in the center of the city (38° 14' 46" N, 21° 44' 08" E) and at the Institute of Chemical Engineering Sciences, ICE-HT (38° 17' 52" N, 21° 48' 31" E), which is 8 km (north-east) away from the city center and 1 km (south) from the Patras-Athens highway (Figure S1). The site is surrounded by olive tree fields. A few small settlements are located in a distance of 1 km (southwest and north-east). This paper focuses only on the ICE-HT measurements.

#### 2.3 Instrumentation in Patras

A HR-ToF-AMS from Aerodyne Research Inc. (DeCarlo et al., 2006) was measuring the size-resolved chemical composition of the NR-PM<sub>1</sub> aerosol species. The tungsten filament for electron ionization was run at an accelerating voltage of 70 eV, while the vaporizer temperature was set at 600°C. Alternative runs between V-mode (single stage reflectron) and W-mode (double stage reflectron) were performed with 3 min of measuring time for each mode. In this paper the V-mode data are presented.

A Proton Transfer Reaction Mass Spectrometer (PTR-MS, Ionicon Analytik) was used for the characterization of the volatile organic compounds (VOCs). More details about the VOC measurements are presented in Kaltsonoudis et al. (in preparation). A Scanning Mobility Particle Sizer, (SMPS, classifier model 3080, DMA model 3081, CPC model 3787, TSI) was operated at a sheath flow rate of 5 L min<sup>-1</sup> and a sample flow rate of 1 L min<sup>-1</sup>. The SMPS measured the number size distribution in the 10 - 500 nm range. A Multiple-Angle Absorption Photometer (MAAP, Thermo Scientific Inc.) (Petzold and Schönlinner, 2004) was used for the BC measurement. NO<sub>x</sub>, SO<sub>2</sub>, O<sub>3</sub> and CO concentrations were measured by the corresponding monitors (Teledyne, models T201, 100E, 400E and 300E respectively).

A filter sampler (MetOne SAASS) was used to collect PM<sub>2.5</sub> samples for inorganic and organic chemical composition analysis. The sampling resolution was 24 h and the flow rate of each filter was 6.7 L min<sup>-1</sup>. Teflon filters (Whatman 7582 004, 0.2 μm pore size) were used for the measurement of the inorganic anion and cation mass concentrations, using two ion chromatography systems (Metrohm 761 Compact IC). Prebaked Quartz filters were used for the EC/OC measurement using a laboratory EC/OC analyzer (Sunset Laboratory Inc). More details about the filter extraction procedure are given by Pikridas et al. (2013) and Tsiflikiotou et al. (in preparation).

The HR-ToF-AMS and the PTR-MS measurements covered the period from 8 to 27 of June 2012, while the rest of the instrumentation provided measurements from June 8 to July 26, 2012. All the above mentioned instruments were located at the ICE-HT campus.

# 2.4 Sampling Site in Athens

Athens is the most densely populated city in Greece with around 4 million inhabitants. The sampling site was at Demokritos National Center for Scientific Research, NCRS (37° 59' 43" N, 23° 48' 57" E), at the municipality of Agia Paraskevi. This is a suburban background site 10 km from the city center located at the foothills of Ymittos mountain and is surrounded by pine trees. This site is close (0.5 km) to the ring highway of Ymittos and approximately 1.5 km away from the Mesogion highway to the north-

west. The closest residences are 0.5 km away. The relative position between Patras and Athens sampling site is illustrated in Figure S2.

#### 2.5 Instrumentation in the Athens site

For the measurements in Athens we used our mobile laboratory as a fixed station, in which the HR-ToF-AMS, PTR-MS and SMPS (same models as described in session 2.3) were placed. An aethalometer (Magee Scientific, AE31) provided the BC concentrations at 880 nm. An SMPS (classifier model 3080 TSI, custom DMA, CPC model 3022, TSI) was operated at a sheath flow rate of 3 L min<sup>-1</sup> and a sample flow rate of 0.3 L min<sup>-1</sup> and measured the number size distribution in the 10 - 660 nm range. PM<sub>2.5</sub> was collected on Teflon filters every 24 h for the inorganic composition characterization, while a semi-continuous OC/EC analyzer (Field Instrument, Model 4F, Sunset Laboratory Inc) equipped with a PM<sub>2.5</sub> inlet and an activated carbon denuder was used for the PM<sub>2.5</sub> EC/OC measurements with a 3 hr resolution.

The sampling period was from 8 to 26 of July 2012. Due to technical problems the HR-ToF-AMS started measuring on the July 12, 2012.

# 2.6 Data Analysis

For HR-ToF-AMS data analysis SQUIRREL v1.51C and PIKA v1.10C (Sueper, 2014) with Igor Pro 6.22A (Wavemetrics) were used. For the HR-ToF-AMS organic mass spectra, we used the fragmentation table of Aiken et al. (2009).

PMF and ME-2 analysis (Paatero and Tapper, 1994; Lanz et al., 2007; Lanz et al., 2008; Ulbrich et al., 2009; Canonaco et al., 2013) were performed using the HR-ToF-AMS organic mass spectra in order to investigate the different organic aerosol sources. The m/z's 12-200 were used as inputs following the procedure of Ulbrich et al. (2009).

The Patras HR-ToF-AMS data were corrected for the collection efficiency CE using the algorithm of Kostenidou et al. (2007) with a 2-hour resolution throughout the campaign. A shape factor ( $\chi$ ) of 1 was found to be the most appropriate for this data set (Supplement section 2.A, Figures S3-S4). The average CE was  $0.91\pm0.10$  and it was quite higher compared to other studies, which often use a CE of 0.5. One reason could be that the particles entering the HR-ToF-AMS were not dried. Drying the particles usually

decreases the CE because the particles bounce on the vaporizer (Matthew et al., 2008). The  $R^2$  between CE and the ambient relative humidity (RH) was very low (0.02). This low correlation and the high CE values can be explained by the fact that the particles were acidic most of the time (see section 3.1) and probably contained water during all the campaign. Comparing the PM<sub>1</sub> HR-ToF-AMS sulphate (after CE corrections) to the PM<sub>2.5</sub> filter sulphate measurements a slope of 1.05 and a high correlation ( $R^2$ =0.98) were found as shown in Figure S5. This could indicate that most of the PM<sub>2.5</sub> sulfate was in the submicrometer particles. The average OA density estimated from the above algorithm was  $1.34\pm0.21$  g cm<sup>-3</sup>, which is very close to the organic density calculated for Finokalia during the summer of 2008 ( $1.35\pm0.22$  g cm<sup>-3</sup>, Lee et al., 2010).

For Athens the aerosol was not dried before it entered the HR-ToF-AMS, but the particles entering the SMPS were dried to maintain compatibility with long term measurements performed in the site. Thus we modified the algorithm of Kostenidou et al. (2007) converting the ambient HR-ToF-AMS mass distributions to dry mass distributions using the relative humidity (RH) inside the sampling line and calculating the inorganic aerosol water content using the Extended Aerosol Inorganic Model II (E-AIM, Carslaw et al., 1995; Clegg et al., 1998; Massucci et al., 1999). As inputs to the E-AIM model we used the inorganic concentrations of sulphate, ammonium and nitrate and the temperature and RH at the entrance of the HR-ToF-AMS line. The algorithm was run for shape factors 1-1.4 and the optimum solution was selected (i.e. the minimum of the minimum error scores) which corresponded to the optimum CE, organic density, and shape factor (Supplement section 2.B, Figures S6-S7). The average CE was 0.63±0.13. After applying CE corrections the PM<sub>1</sub> HR-ToF-AMS sulphate correlated well with the PM<sub>2.5</sub> filter sulphate (Figure S8) (R<sup>2</sup>=0.91, slope=0.98), once more suggesting that the sulphate in the 1-2.5 µm range was a small fraction of the PM<sub>2.5</sub> sulfate. The average organic density was 1.15±0.36 g cm<sup>-3</sup>, lower than in Patras, and the average shape factor was 1.16±0.1 suggesting that a lot of the particles were not spherical.

The angle theta  $(\theta)$  between the mass spectra vectors was used as a measure of their similarity (Kostenidou et al., 2009). The mass spectra are treated as vectors and the angle  $\theta$  is calculated by using their internal product. The lower the angle  $\theta$  is the higher the similarity between the two spectra.

#### 3 Results and Discussion

#### 3.1 Patras

The time series of the mass concentration of the NR-PM $_1$  components measured by the HR-ToF-AMS and the BC in Patras are shown in Figure 1a. The average PM $_1$  mass concentration (not including dust) was 8.6  $\mu g$  m $^{-3}$ . On average, the organic mass concentration was 3.8  $\mu g$  m $^{-3}$ , sulphate 3.3  $\mu g$  m $^{-3}$ , ammonium 0.9  $\mu g$  m $^{-3}$  and BC 0.5  $\mu g$  m $^{-3}$ . Nitrate was very low around 0.1  $\mu g$  m $^{-3}$ .

The average OM:OC ratio was  $1.80\pm0.10$ . The average diurnal profile of O:C is depicted in Figure 2. O:C increased in the early morning hours (2:00-7:00 LT) and during the afternoon (15:00-17:00). The OA average diurnal profile exhibited 3 peaks (Figure S9a), 2 of them (around 8:00-10:00 and 21:00-23:00) were associated with primary sources as the BC increased the same time as well. The OA increase at 11:00-16:00 is related to the formation of secondary species as the solar intensity peaks the same time. The fragments of m/z 44 and 57 represented 0.14 and 0.01 of the organic signal correspondingly (Figure S10a).

Combining the OM:OC ratio provided by the high resolution HR-ToF-AMS data and the organic carbon (OC) filter  $PM_{2.5}$  measurements we estimated the organic mass (OM) based on the filter samples. Figure S11a shows the correlation between the HR-ToF-AMS and the filters for organics which resulted in an  $R^2$ =0.57. This probably implies that the majority of the  $PM_{2.5}$  organics were in the  $PM_1$  range.

The  $PM_1$  aerosol as measured by the HR-ToF-AMS was always acidic with an average inorganic cations/anions equivalent ratio of  $0.75\pm0.07$ . This indicates that there was no available ammonia to fully neutralize the sulphate and thus the formation of  $PM_1$  ammonium nitrate (from ammonia and nitric acid) in the  $PM_1$  was not thermodynamically favorable.

In order to estimate the organic nitrate mass (ONit) contribution to the total nitrate mass (TotNit) we followed the procedure of Farmer et al. (2010):

$$\frac{ONit}{TotNit} = \frac{\left(1 + R_{ONit}\right) \times \left(R_{[NO_{2}^{+}/NO^{+}]meas} - R_{[NO_{2}^{+}/NO^{+}]cal}\right)}{\left(1 + R_{[NO_{2}^{+}/NO^{+}]meas}\right) \times \left(R_{ONit} - R_{[NO_{2}^{+}/NO^{+}]cal}\right)}$$
(1)

where  $R_{NO_{1}^{+}/NO_{1}^{+}|meas}$  is the measured ratio of  $NO_{2}^{+}/NO_{1}^{+}$  ions as a function of time,

 $R_{NO_2^+/NO_1^+|cal|}$  is the ratio of  $NO_2^+/NO_1^+$  ions obtained during  $NH_4NO_3$  calibrations (0.58 on

average) and  $R_{ONit}$  is a fixed value set to 0.05, as the minimum ratio of  $NO_2^+/NO^+$ 

observed in this campaign was 0.05. The average organic nitrate fraction was 0.91±0.05,

which suggests that most of the nitrate was in the form of organic nitrate (Figure S12a).

The correlation between OA and nitrate was moderate ( $R^2=0.38$ ).

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#### 3.2 Athens (Demokritos Station)

On average the PM<sub>1</sub> mass concentration in Athens (not including dust) was 14.2 μg m<sup>-3</sup>.

284 This is a different period than the one in Patras so direct comparisons could result in

285 erroneous conclusions. A detailed description and intercomparison of the various

286 measurements (including filter samples) performed in Athens and Patras is given in the

companion paper by Tsiflikiotou et al. (in preparation). Briefly, these authors found that

for the July period the mean concentrations of PM<sub>2.5</sub> sulphate in Patras and Athens were

similar, while PM<sub>2.5</sub> organics were higher in Patras than in Athens. Figure 1b shows the

290 time series of the concentration of the NR-PM<sub>1</sub> components measured by the HR-ToF-

291 AMS in Athens and the BC measured by the aethalometer. The average organic

concentration was 6.6 μg m<sup>-3</sup>, sulphate 5.3 μg m<sup>-3</sup>, ammonium 1.4 μg m<sup>-3</sup> and BC 0.7 μg

m<sup>-3</sup>. Similarly to Patras nitrate levels were low, 0.2 μg m<sup>-3</sup> on average.

The average O:C mass ratio was 0.47±0.11, lower than in Patras (0.50), because of the higher contribution of primary emissions (Section 3.3.1 and 3.3.2). The average OM:OC ratio was 1.76±0.14. The diurnal cycle of O:C is shown in Figure 2 and is similar

to the one in Patras. The O:C exhibited two peaks one in the early morning around 3:00-

5:00 LT and a second in the late afternoon 17:00-20:00 LT. The OA profile is

characterized by three peaks (Figure S9b). Two of them, around 7:00-12:00 and 20:00-

24:00 are related to fresh emissions since the BC follows the same trend. A smaller

increase around 14:00-17:00 is probably due to photochemistry. On average  $f_{44}$  and  $f_{57}$ 

were 0.13 and 0.02 respectively (Figure S10b).

The OM based on the  $PM_{2.5}$  semi-continuous OC/EC measurements was calculated using the OM:OC ratio provided by the HR-ToF-AMS. Figure S11b illustrates

the correlation for the organics between the HR-ToF-AMS and the semi-continuous OC/EC samples. The measurements are in reasonable agreement ( $R^2$ =0.48), with the HR-ToF-AMS providing slightly higher concentrations. The scatter observed when comparing the corresponding measurements is due to a number of factors including the positive and negative artifacts of filter measurements, the uncertainties of the corresponding measurements (e.g., due to HR-ToF-AMS CE), and the different size ranges. The negative artifacts of the filter measurements due to evaporation of the collected OA can explain the occurrence of some measurements in which the PM<sub>2.5</sub> filter-based OA is less than the PM<sub>1</sub> HR-ToF-AMS OA.

The equivalent ratio of  $PM_1$  cations to anions (measured by the HR-ToF-AMS) was on average  $0.70\pm0.09$  and thus that the aerosol was acidic. Again free ammonia was not enough to neutralize the sulphate and thus there was no ammonium nitrate observed in the  $PM_1$  fraction.

Applying equation (1) to the Athens data set (with an average  $R_{[NO_2^+/NO^+]cal} = 0.68$ , and  $R_{ONit} = 0.12$ ), the organic nitrate fraction was  $0.89 \pm 0.08$ , indicating that most of the nitrate was actually organic nitrate (Figure S12b). The coefficient of determination between OA and nitrate was  $R^2 = 0.62$ .

# 3.3 OA sources

#### **3.3.1 Patras**

For the PMF analysis both PMF Evaluation Tool, PET, (Lanz et al., 2008; Ulbrich et al., 2009) and ME-2 (Lanz et al., 2008; Canonaco et al., 2013) solutions were examined and evaluated using the HR organic mass spectra. A five factor solution was chosen as the best to describe the major sources of OA in Patras. More information about the choice of the solution is provided in the supplementary information (section 7.A, Figures S13-S21). The factors corresponded to very oxygenated OA (V-OOA, 19%), moderately oxygenated OA (M-OOA, 38%), biogenic oxygenated OA (b-OOA, 21%), HOA-1 (7%) and HOA-2 (15%). The assignment of the factors to specific sources was based on their mass spectra and diurnal profile characteristics: V-OOA had a high contribution of m/z 44, M-OOA had a moderate m/z 44 contribution, b-OOA was characterized by both biogenic and oxidized OA signatures, HOA-1 was similar to the

literature HOA factor related to traffic emissions and HOA-2 included primary cooking and aged traffic emissions. Overall, the oxygenated OA was the dominant component (78%) and the primary OA accounted for only 22%. The HR mass spectrum of each factor is illustrated in Figure 3a, while their time series are shown in Figure 4a and their average diurnal profiles in Figure 5a.

V-OOA was characterized by a high m/z 44 (~22%) with an O:C= 0.81 and is related to aged aerosol (Figure 3a). The V-OOA factor correlated well with sulphate (R<sup>2</sup>=0.48) and ammonium (R<sup>2</sup>=0.51), which is typical for highly oxygenated OA. Its correlation with nitrate was low (R<sup>2</sup>=0.09). The V-OOA diurnal profile was almost flat (Figure 5a). R<sup>2</sup> between V-OOA and individual VOCs measured by the PTR-MS were all less than 0.13.

M-OOA had a pronounced but of lower intensity m/z 44 (~16%) and an O:C=0.54 implying aged but less oxidized particles (Figure 3a). Nitrate has been suggested as a tracer for the less oxygenated species (e.g. Lanz et al., 2007; Mohr et al., 2012), however there was very little nitrate during our measurements and most of it was organic nitrate resulting in a weak correlation between the M-OOA and nitrate (R<sup>2</sup>=0.04). M-OOA had a very similar pattern with the solar radiation peaking at around 14:00 (local time). M-OOA had low correlation with methyl vinyl ketone (MVK) and methacrolein (MACR) (m/z 71, R<sup>2</sup>=0.31), hydroxyacetone (m/z 75, R<sup>2</sup>=0.29), C<sub>5</sub> carbonyls/2-methyl-butene-2-ol (MBO)/methacrylic acid (m/z 87, R<sup>2</sup>=0.29), terpene oxidation products (m/z 113, R<sup>2</sup>=0.30) and nopinone (m/z 139, R<sup>2</sup>=0.28), (Table 1 and Figure S22). Comparing the M-OOA spectrum to the  $\alpha$ -pinene SOA spectrum of Heringa et al. (2012) the angle theta was quite high ( $\theta$ =35°), while the similarity between the M-OOA and the toluene photooxidation SOA spectrum (Kostenidou et al., not published data) was greater ( $\theta$ =16°). Thus this factor may include contributions by both anthropogenic and biogenic sources.

The biogenic OOA factor had an  $f_{44}$ =0.13 and O:C=0.48 (Figure 3a) which indicates a relatively moderate degree of oxygenation. It was characterized by a distinct m/z 82 (mainly composed of  $C_5H_6O^+$ ), an elevated m/z 53 (mostly  $C_4H_5^+$ ) and a significant contribution at m/z 39 (5%, mainly  $CH_3O^+$ ). These characteristics were similar to those found at a rural area in Ontario, Canada (Slowik et al., 2011), downtown Atlanta, Georgia, USA (Budisulistiorini et al., 2013), in tropical rainforests in the central Amazon

Basin (Chen et al., 2014) and in Borneo, Malaysia (Robinson et al., 2011) and in Centreville in rural Alabama (Xu et al., 2014) which were associated to secondary OA produced by isoprene photooxidation. Using the HYSPLIT back trajectory model (Draxler and Rolph, 2013), the mass concentration of b-OOA was almost zero when the air masses were coming from the west (e.g. Ionian Sea, 11-15 June, Figure S23), while M-OOA and V-OOA were still high. However, the b-OOA increased in periods (e.g. 10-11 June, 16-17 and 24-26 June) when the air masses passed over the forested mountains of Central Greece (Figure S23), which is an area characterized by high terpene and isoprene emissions (Karl et al., 2009). This supports the biogenic character of this factor. The highest concentration of b-OOA was at 6:00 in the morning on June 16<sup>th</sup> (5.4 µg m<sup>-3</sup>). b-OOA exhibited low correlation with isoprene (m/z 69,  $R^2$ =0.13), isoprene peroxides  $(m/z 101, R^2=0.28)$  and the first generation isoprene products such as MVK and MACR  $(m/z 71, R^2=0.21)$  (Table 1). However, it correlated better with acetone  $(m/z 59, R^2=0.35)$ , hydroxyl-acetone (m/z 75,  $R^2=0.41$ ), PAN (m/z 77,  $R^2=0.37$ ), nopinope (m/z 139,  $R^2=0.39$ ) and pinonaldehyde (m/z 151,  $R^2=0.30$ ), Figure S24, which are products of terpenes ozonolysis (Matsunaga et al., 2003; Holzinger et al., 2005; Lee et al., 2006). HOA-1 was characterized mainly by the m/z's 39, 41, 43, 55, 57, 67, 69 and 81 

HOA-1 was characterized mainly by the m/z's 39, 41, 43, 55, 57, 67, 69 and 81 (Figure 3a) which are typical hydrocarbon fragments of fresh traffic emissions (Zhang et al., 2005; Aiken et al., 2009). Its O:C was 0.1 which is in the range found in literature (the HOA factor of Mohr et al. (2012) had an O:C=0.03, while the HOA factor of Ulbrich et al. (2009) had an O:C=0.18). HOA-1 had a medium correlation with BC ( $R^2$ =0.38) and nitrate ( $R^2$ =0.27). The HOA-1 diurnal profile was characterized by two peaks during the morning (8:00) and evening rush hours (21:00) (Figure 5a). HOA-1 correlated moderately with benzene (m/z 79,  $R^2$ =0.36), toluene (m/z 93,  $R^2$ =0.35), xylenes (m/z 107,  $R^2$ =0.37),  $R^2$ =0.37),  $R^2$ =0.36). The correlation with NO<sub>x</sub> was relatively low ( $R^2$ =0.26).

The HOA-2 O:C was 0.21. The HOA-2 spectrums was characterized among others by the m/z's 39, 41, 43, 44, 55, 57 and 67 (Figure 3a) which are features of cooking organic aerosol, COA (Ge et al., 2011a, b; Crippa et al., 2013a, b). However, the reported COA spectra (e.g. Ge et al., 2011a, b; Crippa et al., 2013a, b) have lower contributions at m/z 44 (0.011-0.022), indicating less oxygenation. The  $f_{44}$  in the HOA-2

mass spectrum was 0.034 which implies that this factor may also contain species that have been oxidized to some degree. The HOA-2 diurnal profile had a small peak around 14:00 and a higher one around 22:00, which are consistent with the Greek lunch and dinner periods (Figure 5a). Thus, the HOA-2 in Patras is mainly due to charbroiling of meat cooking OA. The correlation between HOA-2 and BC, nitrate, benzene, toluene, xylenes and C<sub>9</sub> aromatic compounds was moderate (R<sup>2</sup>=0.38, 0.38, 0.33, 0.30, 0.33 and 0.43 correspondingly).

During June 16-23 the wind speed was relatively high (average 5.3 m s<sup>-1</sup>) compared to the rest of the sampling days (average 2.5 m s<sup>-1</sup>). The contribution of the local sources (HOA-1 and HOA-2) was less than 9% of the OA during that windy period compared to 28% during the remaining days (Figure 4a).

# **3.3.2** Athens

Four OA factors could be identified in the Athens HR-ToF-AMS data set: V-OOA (35%), M-OOA (30%), HOA-1 (18%) and HOA-2 (17%). A detailed description of the reasons for this selection can be found in the supplement (section 7.B, Figures S25-S33). The corresponding mass spectra are provided in Figure 3b. The time series of the four PMF factors are shown in Figure 4b, while their diurnal cycles in Figure 5b. Similarly to Patras the contribution of the oxygenated OA in Athens was high (65%), while the primary sources contributed 35%.

The very oxidized OA (V-OOA) had an  $f_{44}$ =0.18 and O:C=0.68 and showed a good correlation with sulphate (R<sup>2</sup>=0.53) and ammonium (R<sup>2</sup>=0.50) consistent with the aged character of this factor. The diurnal profile of the V-OOA (Figure 5b) was characterized by a peak around 15:00-18:00, which is probably associated with production of this component in the afternoon over the region. At the same time M-OOA decreased, which indicates that V-OOA is a product of photochemical processing. V-OOA correlated with formic acid (m/z 47, R<sup>2</sup>=0.47), hexenal (m/z 99, R<sup>2</sup>=0.42), isoprene peroxides (m/z 101, R<sup>2</sup>=0.35), terpene oxidation products (m/z 113, R<sup>2</sup>=0.40), and heptanal (m/z 115, R<sup>2</sup>=0.42) (Table 1 and Figure S34).

The moderately oxidized OA (M-OOA) was characterized by an  $f_{44}$ =0.14 and O:C=0.56 and exhibited a weak correlation with sulphate, ammonium and nitrate

 $(R^2=0.17, 0.17 \text{ and } 0.13 \text{ respectively})$ . M-OOA increased during the day with a maximum at 12:00-14:00, following the diurnal profile of solar radiation, which implies relatively fast photochemical reactions. M-OOA had a second peak during the night around midnight, which could be an indication of nighttime production or condensation due to the change in temperature between day and night (the average temperature at noon was  $32^{\circ}$ C, while during the night it decreased to around  $21^{\circ}$ C). M-OOA did not show any correlation ( $R^2$  less than 0.07) with the measured VOCs.

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The HOA-1 O:C was 0.07. Surprisingly it had a rather weak correlation with BC  $(R^2=0.05)$  and it also showed a low correlation with nitrate  $(R^2=0.13)$ . The HOA-1 diurnal profile had 2 peaks, a small increase during the morning (7:00) and a larger peak in the evening hours (20:00), consistent with the contribution of traffic emissions. HOA-1 did not correlate with VOCs characteristic of traffic such as benzene, toluene and xylenes (the  $R^2$  values were correspondingly 0.15, 0.13 and 0.16) and inorganic gases as  $NO_x$  $(R^2 = 0.06)$ . The HOA-1 mass spectrum and time series were very stable for 2 to 5 factorial solutions and for  $f_{peaks}$  values from -1 to 1. The  $R^2$  between the HOA-1 of the selected solution and the HOA-1 of the solutions of 2, 3 and 5 factors was always greater than 0.973 (both for the time series and the mass spectra). This weak correlation between the HOA-1 factor and other primary organic pollutants indicates that their concentrations were not dominated by the same sources. For example HOA-1 possibly originated mostly from passenger cars while BC originated from diesel vehicles. The location of the sampling site and the inhomogeneity of the surrounding areas, in combination with the wind direction changes have confounded these effects. The rose plots of HOA-1, BC, NO<sub>x</sub> and benzene for wind speeds greater than 1 m s<sup>-1</sup> indicated that BC, NO<sub>x</sub> and benzene had the same origin, while HOA-1 did not. For example at 8:00 LT the HOA-1 was on average coming from southwest (Figure S35-S38) likely from the ring highway of Ymittos, while BC, NO<sub>x</sub> and benzene from the north probably influenced by the Mesogion highway.

HOA-2 had an O:C=0.24 and exhibited a good correlation with BC ( $R^2$ =0.57). HOA-2 also correlated well with nitrate ( $R^2$ =0.75) implying that organic nitrate compounds were possibly emitted or produced along with this OA type. The HOA-2 profile had 2 peaks at around 11:00 and 22:00. The second peak is characteristic of Greek dinner period, thus part of HOA-2 could be attributed to meat cooking OA. However, the first peak can not be explained by the Greek lunch period. The correlation with the BC implies that HOA-2 and BC had the same origin. Comparing the HOA-2 mass spectrum with aged POA or SOA emissions from other sources such as  $\alpha$ -pinene, wood burning, scooter and diesel (Heringa et al., 2012) the angles  $\theta$  were 19, 19, 28, 17° respectively. This indicates that HOA-2 could include aged diesel emissions since  $\alpha$ -pinene is not associated with BC and there were no observed biomass burning events during the sampling period. The HOA-2 factor correlated with the m/z 43 ( $R^2$ =0.43), acetone (m/z 59,  $R^2$ =0.44), methyl ethyl ketone (MEK, m/z 73,  $R^2$ =0.49), benzene (m/z 79,  $R^2$ =0.62), toluene (m/z 93,  $R^2$ =0.54), xylenes (m/z 107,  $R^2$ =0.56), C<sub>9</sub> aromatic compounds (m/z 121,  $R^2$ =0.58) and C<sub>10</sub> aromatic compounds (m/z 135,  $R^2$ =0.55). It also had a good correlation with NO<sub>x</sub> ( $R^2$ =0.58).

Trying rotations in the  $f_{peak}$  range -1 to 1 the correlations between the factors HOA-1 and HOA-2 and traffic markers such as BC and toluene practically did not change. In addition using a fixed HOA spectrum (the HOA mass spectrum of Aiken et al. (2009) and the HOA mass spectrum found in Athens during winter, unpublished results) with a=0.1 the correlation between HOA-1 and BC did not improve. All these suggest that our conclusions are robust to the details of the PMF process. Our explanation is that the correlations are affected also by the locations of the various sources around the receptor site and the corresponding wind directions. The fact that these area sources (gasoline cars, diesel cars, cooking activities) did not have a homogeneous spatial distribution can lead to these rather unexpected results.

#### 3.3.3 Comparison of the PMF factors in the two cities

The mass spectra of the V-OOA factors in the two cities were almost the same  $(\theta=6.7^{\circ})$ . However, Athens V-OOA exhibited lower  $f_{44}$  and O:C ratio (0.18 and 0.68 correspondingly) compared to Patras V-OOA (0.22 and 0.81 respectively). This difference could be partially due to the different periods of the two measurements.

The two M-OOA mass spectra were even more similar to each other ( $\theta$ =5.4°). A high correlation was also observed between the two HOA-2 mass spectra ( $\theta$ =5.9°). The Athens HOA-2 was slightly more oxygenated (O:C=0.24), as the oxygenated part of m/z

491 43 was more elevated, 65%, compared to 51% observed in Patras HOA-2 and the  $f_{44}$  was 492 just slightly higher (0.07), while in Patras HOA-2 was 0.06.

The two HOA-1 factors exhibited a lower correlation with each other ( $\theta$ =22.8°). The main differences were at m/z 41, 43, 55, 57, 69, 71, 81, 83 and 85 which were more abundant in the Athens HOA-1 spectrum, while in the Patras spectrum the  $f_{44}$  was higher (0.04) compared to Athens (0.02). In Patras HOA-1 spectrum m/z 57 was lower in comparison with m/z 55. The O:C in Athens HOA-1 was a little lower (0.07) than in Patras HOA-1 (0.10).

### 3.3.4 Comparing the PMF factors with other studies

Table 2 shows comparisons of the mass spectra of the factors found in the two cities with selected PMF factors from the literature (Aiken et al., 2009; Docherty et al., 2011; Sun et al., 2011; Robinson et al., 2011; Heringa et al., 2012; Mohr et al., 2012, Ge et al., 2012a, b; Crippa et al., 2013a, b; Budisulistiorini et al., 2013; Xu et al., 2014) that have all been extracted using the fragmentation table of Aiken et al. (2009).

The two V-OOA factors were quite similar with LV-OOA at Riverside (Docherty et al., 2011),  $\theta$ =8-10°, and with LV-OOA at New York City (Sun et al., 2011),  $\theta$ =11-14°. They also showed good correlation with LV-OOA measured in Barcelona (Mohr et al., 2012) and Paris (Crippa et al., 2013a, b)  $\theta$ =13-19°.

M-OOA had moderate to low agreement with the majority of the literature profiles. The lowest angle  $\theta$  (around 22-24°) corresponded to the comparison with the SV-OOA found in Mexico City (Aiken et al., 2009). It did however resemble ( $\theta$ =13-16°) the toluene SOA spectrum (Kostenidou et al., not published data).

The b-OOA factor in Patras correlated moderately with  $\alpha$ -pinene SOA,  $\theta$ =29°, (Heringa et al., 2012) and the isoprene factor found in Alabama,  $\theta$ =28°, (Xu et al., submitted). However, it had low correlation with the Factor 82,  $\theta$ =47°, found in Malaysia (Robinson et al., 2011) and the IEPOX OA factor extracted in Atlanta,  $\theta$ =76°, (Budisulistiorini et al., 2013).

The HOA-1 factors correlated well with most of the literature HOA profiles,  $\theta$ =10-25°. The highest angles (24 and 25°) corresponded to the comparison with Paris during the winter (Crippa et al., 2013b). The major differences were in the m/z's 44 and

- 28 which were higher than in Paris, m/z's 29 and 43 which were lower compared to Paris and m/z 39 which was absent in Paris probably due to the unit mass resolution spectra used as input for the Crippa et al. (2013) PMF analysis.
- The HOA-2 mass spectra resembled the COA factor extracted in New York City,  $\theta$ =11-14°, (Sun et al., 2011) and the SOA from diesel VOCs emissions,  $\theta$ =17°, (Heringa et al., 2012).
- Figure S39 summarizes Patras and Athens OA measurements and PMF factors in the Ng triangle (Ng et al., 2010). All the data fall within the triangle.

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#### 3.4 Discussion

- 532 In both cities the composition of NR-PM<sub>1</sub> was surprisingly similar. Organics' 533 contribution in Patras and Athens was around 45%, which is similar to other areas in 534 Europe: London (UK) 46%, mountain Taunus (Germany) 59%, Melpitz (Germany) 59%, Mace Head (Ireland) 39%, Po Valley (Italy) 33%, and Paris (France) 50% (Cubison et al., 535 536 2006; Hings et al., 2007; Poulain et al., 2011; Dall'Osto et al., 2010; Saarikoski et al., 537 2012; Crippa et al., 2013a). With 38%, sulphate made a larger contribution compared to other studies in Europe during the summer: London 31%, mountain Taunus 24%, Mace 538 539 Head 32%, Melpitz 22%, Po Valley 9%, and Paris 25% (Cubison et al., 2006; Hings et al., 540 2007; Dall'Osto et al., 2010; Poulain et al., 2011; Saarikoski et al., 2012; Crippa et al., 541 2013a). Nitrate contributed very little (less than 2%) and was mostly attributed to 542 organonitrate compounds, in contrast with other European studies where nitrate ranged 543 from 6% (Melpitz) to 39% (Po Valley) (Poulain et al., 2011; Saarikoski et al., 2012) and 544 mainly was ammonium nitrate. The absence of particulate nitrate in PM<sub>1</sub> was also 545 observed at Finokalia (Hildebrandt et al., 2010) and is characteristic of the Eastern 546 Mediterranean. Ammonia levels in this region are quite low (Wichink Kruit et al., 2012) 547 to fully neutralize the existing relatively high sulphate.
- The O:C ratios (0.50 for Patras and 0.47 for Athens) were moderately high. The O:C ratio at Finokalia was 0.8 (Hildebrandt et al., 2010), at Cape Corsica 0.9 (Nicolas, 2013), while in Paris 0.38 (Crippa et al., 2013a).
- In both cities the OOA was the dominant OA component (78% in Patras and 65% in Athens). This fraction is within the range that has been measured in previous summer

studies in the Mediterranean. For example, the OOA at Finokalia was 100% of the OA (Hildebrandt et al., 2010), in Marseille accounted for 80% (El Haddad et al., 2013), at Cape Corsica 80-85% (Nicolas, 2013), while in Po Valley 61% (Saarikoski et al., 2012).

In Athens 35% of the OA was V-OOA, while in Patras only 19% of the OA was attributed to V-OOA. In addition the V-OOA in Athens was increasing in the afternoon, suggesting its production either locally or regionally. In Patras V-OOA had an almost flat diurnal profile. This difference could be attributed to the different air masses that arrive to each site. According to back trajectory analysis, based on FLEXPART (Stohl et al., 2005) (Figure S40) and HYSPLIT (Draxler and Rolph 2013) during most of the sampling days in Athens the air masses had spent considerable time over the source-poor Aegean Sea, while the majority of the air masses that arrived in Patras were continental and passed over the mountains of Central Greece.

Given the location of the two sites, one would expect to find a marine OA (MOA) factor. The S:C ratio estimated by the AMS is often underestimated in the presence of organosulfates (Farmer et al., 2010; Docherty et al., 2011). To avoid such artifacts we investigated the contribution of MOA applying a constrained solution in the ME-2 using the MOA mass spectrum of Crippa et al. (2013) with a=0.1. For Patras the average MOA concentration for 4, 5, and 6 factors was around 0.04  $\mu$ g m<sup>-3</sup> corresponding to 1 percent of the OA mass. For Athens the MOA concentration for the 3, 4 and 5 factor solutions was approximately 0.25  $\mu$ g m<sup>-3</sup> (3.7 percent of the OA mass). So if MOA was indeed present its contribution to OA was quite low.

#### 4 Conclusions

During the summer of 2012 the air pollution in Patras (June) and Athens (July) was monitored continuously. The sum of the NR-PM<sub>1</sub> and BC concentration was on average  $8.6~\mu g$  m<sup>-3</sup> in Patras and  $14.2~\mu g$  m<sup>-3</sup> in Athens. However, the aerosol composition was quite similar in both areas: 45% OA, 38% sulphate, 11% ammonium, 1% nitrate (mostly organic) and 5% BC indicating the importance of regional sources. In both cities the fine aerosol was acidic, which is consistent with the low ammonia levels in the Eastern Mediterranean.

For Patras the average O:C ratio was 0.50±0.08, while in Athens 0.47±0.11. In both cities oxygenated OA was the major component of organic aerosol (78% in Patras and 65% in Athens), indicating the impact of regional pollution in Mediterranean. OOA included a moderately oxygenated OA and a very oxygenated OA component. In Patras a biogenic oxidized OA factor could be identified, which was related to air masses passing over the forests of Central Greece. A primary OA factor (HOA-2) was found in both cities attributed to primary emissions such as meat cooking. This factor may also contain oxygenated primary emissions (e.g., aged diesel emissions). Hydrocarbon-like OA mainly from traffic emissions was also indentified, accounting only for 7-18% of the OA and indicating that new emission control technologies applied to vehicles during the last decade have reduced dramatically the levels of the corresponding primary OA.

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Table 1. Correlations of the M-OOA and b-OOA factors (Patras) and the V-OOA factor(Athens) with various VOCs measured by the PTR-MS.

$\mathbb{R}^2$	Patras M-OOA	Patras b-OOA	Athens V-OOA
m/z 43	0.29	0.13	0.39
m/z 47 (formic acid)	0.21	0.09	0.47
m/z 59 (acetone, glyoxal)	0.21	0.35	0.32
<i>m/z</i> 71 (MVK, MACR)	0.31	0.21	0.20
<i>m/z</i> 73 (MEK)	0.24	0.25	0.29
<i>m/z</i> 75 (hydroxyacetone)	0.29	0.41	0.30
m/z 77 (PAN)	0.16	0.37	0.16
m/z 81 (terpenes)	0.19	0.26	0.12
<i>m/z</i> 87 (MBO, C5, methacrylic acid)	0.29	0.31	0.38
m/z 95 (2 vinyl furan, phenol)	0.17	0.31	0.36
<i>m/z</i> 99 (hexenal)	0.19	0.30	0.42
<i>m/z</i> 101 (isoprene hyperoxides,hexanal)	0.17	0.28	0.35
m/z 103 (MPAN)	0.23	0.28	
$m/z$ 113 (chlorobenzene, terpenes $+O_3$ )*	0.30	0.37	0.40
<i>m/z</i> 115 (heptanal)	0.16	0.32	0.42
<i>m/z</i> 137 (monoterpenes)	0.20	0.24	0.1
<i>m/z</i> 139 (nopinone)	0.28	0.39	0.19
<i>m/z</i> 151 (pinonaldehyde)	0.17	0.30	0.17

Table 2. Correlations between PMF factors from Patras and Athens and PMF factors from selected studies.

	Angle (degrees) with V-OOA Patras	Angle (degrees) with V-OOA Athens
LV-OOA Mexico City <sup>1</sup>	21	20
LV-OOA Riverside <sup>2</sup>	10	8
LV-OOA Barcelona <sup>3</sup>	18	16
LV-OOA Paris (SIRTA) summer <sup>4</sup>	19	15
OOA Paris (SIRTA), winter <sup>5</sup>	13	15
LV-OOA New York City <sup>6</sup>	14	11
LV-OOA Fresno <sup>7,8</sup>	36	30

	Angle (degrees) with M-OOA Patras	Angle (degrees) with M-OOA Athens
SV-OOA Mexico City <sup>1</sup>	24	22
SV-OOA Riverside <sup>2</sup>	42	39
SV-OOA Barcelona <sup>3</sup>	33	31
SV-OOA Paris (SIRTA) summer <sup>4</sup>	42	39
SV-OOA New York City <sup>6</sup>	30	27
α-pinene ozonolysis SOA aged <sup>9</sup>	35	32
Toluene photooxidation (HONO) SOA <sup>10</sup>	16	13

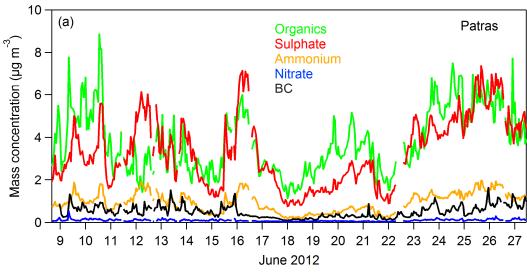
	Angle (degrees) with HOA-1 Patras	Angle (degrees) with HOA-1 Athens
HOA Mexico City <sup>1</sup>	13	14
HOA Riverside <sup>2</sup>	20	10
HOA Barcelona <sup>3</sup>	22	11
HOA Paris (SIRTA) summer <sup>4</sup>	16	18
HOA Paris (SIRTA), winter <sup>5</sup>	25	24
HOA New York City <sup>6</sup>	12	10
HOA Fresno <sup>7,8</sup>	11	11

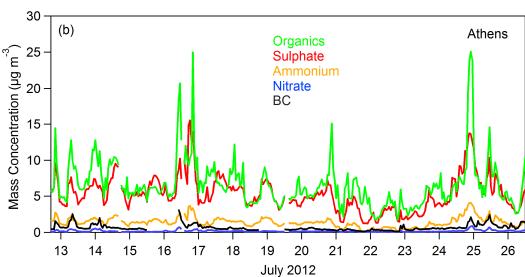
	Angle (degrees) with HOA-2 Patras	Angle (degrees) with HOA-2 Athens
COA Barcelona <sup>3</sup>	77	77
COA Paris (SIRTA), summer <sup>4</sup>	30	34
COA Paris (SIRTA), winter <sup>5</sup>	30	28
COA Paris (LHVP), winter <sup>5</sup>	27	31
COA New York City <sup>6</sup>	11	14
COA Fresno <sup>7,8</sup>	28	33
Aged VOCs diesel emissions <sup>9</sup>	17	17
α-pinene ozonolysis SOA aged <sup>9</sup>	20	19
Toluene photooxidation (HONO) SOA <sup>10</sup>	26	23

	Angle (degrees) with b-OOA Patras
α-pinene ozonolysis SOA aged <sup>9</sup>	29
Factor 82 <sup>11</sup>	47
IEPOX OA <sup>12</sup>	76
Isoprene-OA <sup>13</sup>	28

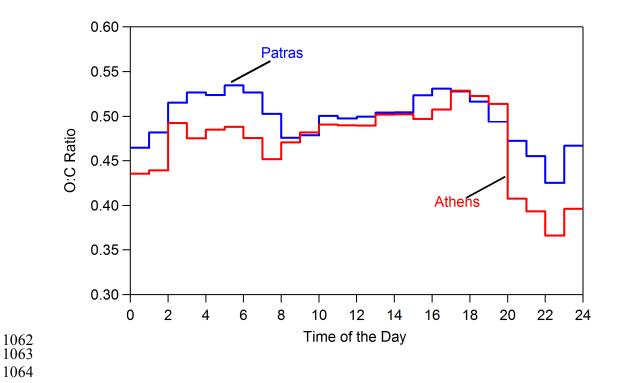
 $\begin{array}{c} 1012 \\ 1013 \end{array}$ 

 $^{1}\text{Aiken et al. (2009), }^{2}\text{Docherty et al. (2011), }^{3}\text{Mohr et al. (2012), }^{4}\text{Crippa et al. (2013a), }^{5}\text{Crippa et al. (2013b), }^{6}\text{Sun et al. (2011), }^{7,8}\text{Ge et al. (2012a,b), }^{9}\text{Heringa et al. (2012), }^{10}\text{Kostenidou et al., (not published data), }^{11}\text{Robinson et al. (2011), }^{13}\text{Budisulistiorini et al. (2013), }^{13}\text{Xu et al., (2014), }^{13}\text{Nu et al.,$ 





**Figure 1**. Time series of organics, sulphate, ammonium and nitrate mass concentration measured by the HR-ToF-AMS (corrected for the CE) and BC a) for Patras and b) for Athens. The BC was provided by MAAP for Patras measurements and by an aethalometer for Athens.



**Figure 2**. Diurnal profile of O:C ratio for Patras (blue line) and Athens (red line) data set. The O:C was calculated with the Aiken et al. (2009) fragmentation table.

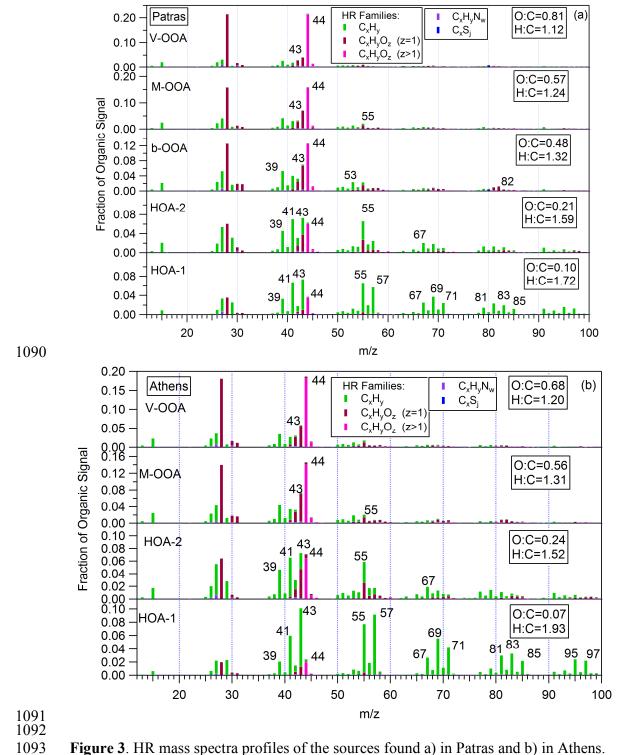
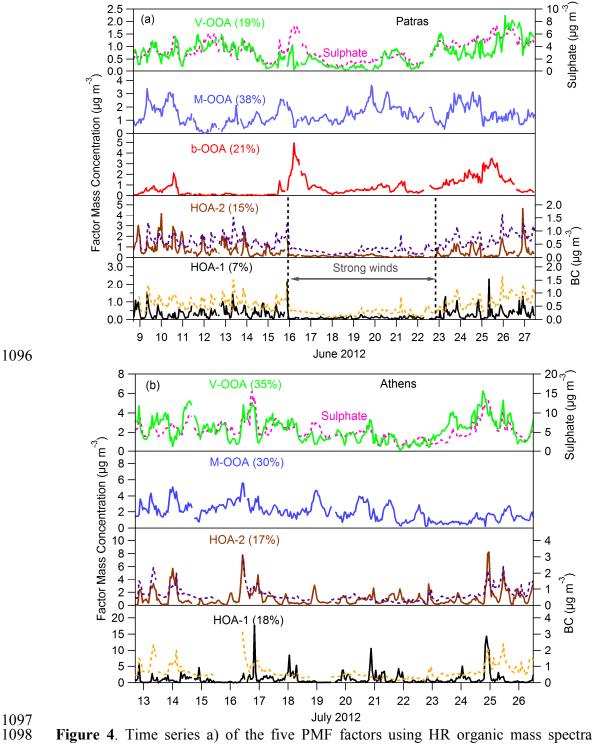


Figure 3. HR mass spectra profiles of the sources found a) in Patras and b) in Athens.



**Figure 4**. Time series a) of the five PMF factors using HR organic mass spectra for Patras and b) of the four PMF factors found in Athens. For the Patras measurements the HOA-1 and HOA-2 contribution was very low between the 16<sup>th</sup> and 23<sup>rd</sup> of June 2012 due to the high wind speed during that period.

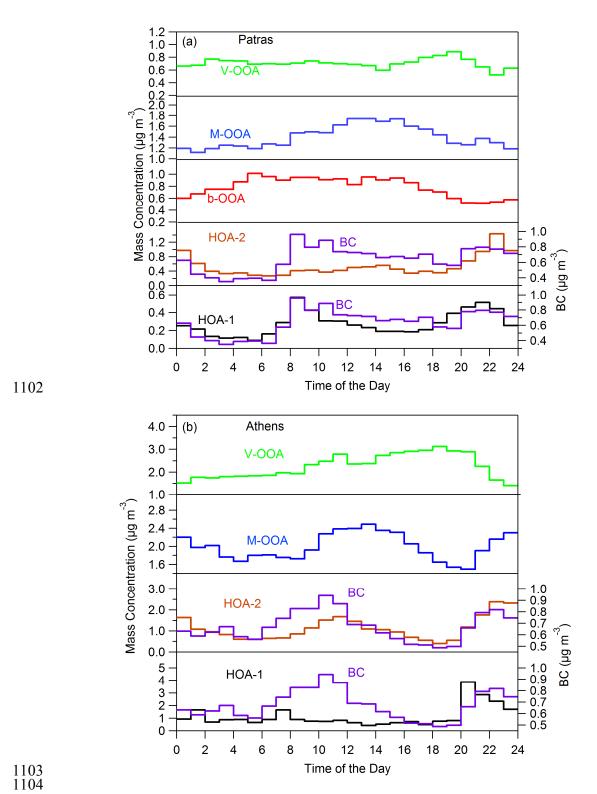


Figure 5. Diurnal cycles of the PMF factors a) in Patras and b) in Athens.