

We greatly appreciate the reviewers for their valuable comments, which have helped us improve the paper significantly. We have addressed all of the comments carefully, as detailed below. The original comments are in black, and our responses are in blue.

Anonymous Referee #1

The authors presented field observation results of PM_{2.5} mass concentration, water soluble inorganic ions in PM_{2.5}, OC/EC analysis of PM_{2.5}, as well as reconstructed light extinction coefficient during haze events in January 2013 in the city of Suzhou, China. (1) Sources and processes leading to PM components were discussed with both back-trajectory analysis and the potential source contribution function (PSCF) method. (2) To investigate the contribution of PM species to visibility reduction, the authors attempted to link the two with the revised IMPROVE algorithm and the Koschmieder equation. (3) Finally, gas-to-particle conversion in secondary inorganic species (sulfate and nitrate) formation was briefly discussed.

The dataset itself is definitely interesting and worth exploring, and the methodologies employed by the authors were also scientifically sound. There are two reasons I don't see this version of the manuscript is publishable yet. First, for the three points summarized in the first paragraph, point (1) is just standard and should be in a short description only; point (2) is totally unnecessary since it is a two-step linkage from chemical composition to visibility (to me, it looks more like a number game if no measured extinction coefficient is shown); point (3) should be explored in greater detail, together with the distinct feature of the third haze event (30, January). This first reason will be elaborated point-by-point below in the major comments. Second, the language of the current form of the manuscript needs lots of work to enhance its readability. This second reason will be supplemented with some minor comments below (for those I have spotted). With these reasons, I suggest a major revision before the manuscript can be published in ACP.

We have made major revisions of the paper based on the reviewer's comments. We'd like to illustrate a few points here:

(1) The discussion on the backward trajectory analysis results has been simplified and improved for easy reading. We kept this discussion as one subsection due to the following considerations. Firstly, we think it's important to investigate the effects of the regional sources to the aerosol pollution in Suzhou as it is located in the typical monsoon region. Secondly, we think the results of back trajectory analysis and PSCF do not conflict with the assertion that secondary formation was responsible for high PM level. The impact of regional areas to the studied site is multifaceted. The regional transport might bring precursors or catalytic agents which could promote the secondary formation and/or the generation of PM instead of directly bring PM or secondary components to the studied site. Besides, the results of PSCF in the present study showed that Suzhou was mainly affected by local and nearby areas. Thirdly, the analysis helped the understanding of the distinct feature of the third haze event.

(2) In the revised manuscript, the reconstructed light extinction coefficients were compared

with those derived from visibility and calculated using another model. Strong correlations were observed, confirming that the reconstructed b_{ext} from IMPROVE algorithm was reliable. It's necessary to investigate the dominant contributors to the light extinction which is directly linked to the visibility. The contribution of chemical specie to light extinction might be different from that to aerosol.

(3) The discussions about the formation of secondary inorganic species and the distinct feature of the third haze event were strengthened in the revised manuscript. We have done more comprehensive analysis of the data to explore the SIA formation mechanism in the revised manuscript, such as the roles of liquid water content in aerosol, meteorological parameters and gaseous pollutants. The results suggested that the gas-phase homogeneous reaction and the heterogeneous process both responsible for the formation of nitrate and sulfate. The discussion of the distinct characteristics of the third haze was also strengthened by making comparisons with the other two haze events from more aspects such as $\text{PM}_{2.5}$ composition, light extinction coefficient, sources, etc.. The language has also been modified by professionals to enhance its readability.

Major:

1. Section 2. (a) The authors used data of SO_2 , CO , and NO_x quite a bit later in the Results and discussion section. Measurement of these criteria pollutants should be mentioned here as part of the methodology. (b) A brief description on URG 9000 IC would be beneficial to readers since it is not as standard as TOEM. (c) A brief description on OC/EC analyzer and what method of OC/EC splitting was used (thermal OC/EC or optical OC/EC)?

The instruments used for the measurements of gaseous pollutants and meteorological parameters have been described in the revised manuscript. A brief description of instruments applied for PM, water soluble ions and OC/EC determinations has been provided in the revised manuscript.

2. Section 3.4. In this section, the authors tried to link the sources with both chemical composition and light extinction by back-trajectory analysis and PSCF. I don't see a great value of this sub-section for the following two reasons. First, if this analysis is useful, then the results just basically invalidated the authors early assertion that secondary formation was the dominant "source" for high PM levels observed (which I believe in). Second, the discussion of this sub-section is just too confusing and difficult to follow. I would suggest to talk generally about air mass origins in the general characteristics sub-section while not pushing too far to pin-point sources of those mainly secondary species.

As mentioned earlier, we think this discussion is helpful and kept this sub-section due to the following considerations. Firstly, we think it's important to investigate the effects of the regional sources to the aerosol pollution in Suzhou as it is located in the typical monsoon region. Secondly, we think the results of back trajectory analysis and PSCF do not conflict with the assertion that secondary formation was responsible for high PM level. The impact of regional areas to the studied site is multifaceted. The regional transport might bring

precursors or catalytic agents which could promote the secondary formation and/or the generation of PM instead of directly bring PM or secondary components to the studied site. Besides, the results of PSCF in the present study showed that Suzhou was mainly affected by local and nearby areas. Thirdly, the analysis helped the understanding of the distinct feature of the third haze event. For your second reason, the discussion on the backward trajectory analysis results has been simplified and improved for easy reading.

3. Sub-section 3.2.3. The authors used revised IMPROVE algorithm to reconstruct light extinction coefficient and used Koshmieder equation to “reconstruct” visibility. The discrepancy is large (a factor of two), owing (in my opinion) to this two-step linkage with both steps involving a number of assumptions and uncertainties. If there is no measured extinction coefficient to support, I do not see what value this analysis would add to the manuscript.

It's meaningful to investigate the contributions of PM_{2.5} component to light extinction, because the results directly manifest the dominant contributor to visibility reduction, and have policy implications. Additionally, the contribution of chemical specie to light extinction might be different from that to aerosol loading. Although the discrepancy in light extinction between those reconstructed using IMPROVE and those derived from visibility is large, the results provide useful knowledge in improving our standing of the topic. Therefore, in the revised manuscript, we have made comprehensive comparisons of light extinction coefficients calculated from using different models, and with added uncertainty analysis. The results confirmed that the reconstructed b_{ext} from the IMPROVE algorithm was reliable to a large extent.

4. Section 3.3. The discussion of secondary inorganic species formation is informative but need some cautions. (a) correlation between RH and SOR does point to the importance of aqueous-phase formation of sulfate, but RH is an indication of gas-phase water after all. It is suggested that liquid water content in PM to be estimated using E-AIM or ISORROPIA.

We agree that to get the conclusion about the importance of aqueous-phase formation to sulfate only based on the correlation between RH and SOR might be hasty. So in the revised manuscript, we estimated the liquid water content in PM by using E-AIM and conducted more detailed analysis to explore the formation mechanism of nitrate and sulfate.

(b) the authors cited Pathak et al., 2004, Pathak and Chan, 2005 to back the statement that homogeneous reaction between HNO₃ and NH₃ was important in nitrate formation. But these two papers talked about sampling artefacts for filter sampling when particles and gases can interact for 24 hours, while the authors used continuous measurement technique to measure SNA. I don't see that is relevant.

We agree with the reviewer that it's inappropriate. We cited other references (Pathak et al., 2009; Jansen et al., 2014; He et al., 2012) to back that statement in the revised manuscript.

(c) excess ammonium is of course one way to look at nitrate formation, but partitioning

equilibrium between NH_4NO_3 and HNO_3 and NH_3 is also important to be considered given the low temperature and high RH in the studied period.

We explored nitrate formation mechanism from more aspects and conducted more detailed analysis. The results indicated that heterogeneous chemistry, such as equilibrium partitioning between NH_4NO_3 and HNO_3 and NH_3 ($\text{HNO}_3(\text{g}) + \text{NH}_3(\text{g}) \leftrightarrow \text{NH}_4^+(\text{aq}) + \text{NO}_3^-(\text{aq})$) also contributed to the formation of nitrate just as the reviewer expected.

5. The authors briefly mentioned the uniqueness of the third haze events compared to the first two in a number of places, but did not elaborate them. From the high POC and dominated air mass origins of C2 (short circulating trajectories), I believe this event was mainly contributed by primary emission from the local (surrounding) areas. It is suggested that this episode to be discussed in contrast to the other two with respect to primary/secondary fractions and meteorological parameters.

Thanks for the useful suggestion. According to the reviewer's suggestion, the discussion about the distinct feature of the third haze event has been strengthened in the revised manuscript. The characteristics of the third haze event has been compared with the other two haze events from various aspects, such as weather conditions, aerosol compositions, light extinction contributions oxidation ratio, air mass origination and etc. The results indeed indicated that the carbon components from the primary emission in local and/or nearby areas instead of secondary formation inorganic ions might be relatively more important for the visibility reduction for the third haze event.

Minor

P33409, L3: change "public" to "the public".

It has been revised accordingly.

P33409, L15: what is "artificial sampling"? should be "filter sampling"?

The phrase "artificial sampling" has been changed to "filter sampling": "Most existing studies were based on filter sampling and off-line analysis and had limitations in providing detailed insight into the roles the major chemical species played during shorter haze periods."

P33409, L26: change "close" to "similar".

It has been revised accordingly.

P33410, L3: change "salt" to "salts".

It has been revised accordingly.

P33410, L22: change "investigate the" to "investigation of".

The sentence has been revised.

P33411, L1: suffered should be suffered from.

It has been revised accordingly.

P33414, L20: change little to slightly (also in P33416, L6).

It has been revised accordingly.

P33417, L14-16: this sentence is not convincing to me. It is stated that there is difference between Suzhou and Beijing. But here it is asserted that nitrate formation may be also affected by re-volatilization of NH_4NO_3 as that in Beijing (similarity?).

There are both difference and similarity between Suzhou and Beijing. The $\text{NO}_3^-/\text{SO}_4^{2-}$ ratio was higher under worst visibility conditions in Suzhou while higher under better visibility in Beijing. But similar to Beijing, the ratios of ($\text{NO}_3^-/\text{SO}_4^{2-}$) to (NO_x/SO_2) were lower for worse visibility period, which could indicate the effect of the re-volatilization of NH_4NO_3 on the nitrate concentrations. We have modified this paragraph hopefully to make it more understandable.

P33418, L16: change migh related to might be related.

It has been revised accordingly.

P33418, L28: change “similar profile” to a profile similar”.

It has been revised accordingly.

L33149, L3: I don't agree that all the secondary aerosol species were “affected” by O_3 . O_3 is just one of those oxidants that can oxidize precursors and lead to SIA formation. In fact, it is the secondary nature of O_3 that makes its diurnal profile some similar to those of the secondary aerosol species.

We agree with the reviewer. We now rephrase this sentence: “The diurnal profiles of the secondary species were similar to their precursors but obviously affected by other factors such as solar radiation, which could promote the oxidation of the precursors.”

P33149, L11: also responsible should be also be responsible.

It has been revised accordingly.

L33419, L12: level should be levels.

It has been revised accordingly.

L33419, L19: “because of” should be “be due to”.

It has been revised accordingly.

P33419, L21: favored for should be favored.

It has been revised accordingly.

P33419, L22: always southwest wind? The discussion later for the first and second haze events suggests otherwise.

We assume that the later discussion the reviewer refers to is the back-trajectory analysis. We think there is no certain link between wind direction and back-trajectory results. Wind direction refers to instantaneous local weather condition. But back-trajectory results reflect the migration of air mass on a much larger temporal and spatial scale. So we don't think the results conflict with each other.

P33420, L2-7: I don't see it is necessary to repeat the equation here. In fact, I don't see it is necessary to have this analysis of reconstructed light extinction coefficient, as shown above.

We agree with the reviewer that it's unnecessary to repeat the equation. Therefore, the revised equation was not showed in the new version of manuscript. But we do think it's meaningful to have the discussion about light extinction for aforementioned reasons.

P33420, L14: "similar temporary trend" and "significantly correlated" are referring to the same thing.

We agree with the reviewer. The sentence has been rephrased in the revised manuscript: "Nevertheless, they were significantly correlated with each other ($r = 0.71, p < 0.001$)."

P33420, L16: were should be was.

It has been revised accordingly.

P33420, L26: were reduced should be reduced.

It has been revised accordingly.

P33421, L28: how come only NO₂ (not NO + NO₂) was used in calculation of NOR?

The calculation of NOR by NO₂ and NO₃⁻ has been used to estimate the secondary formation of nitrogen in many studies (Wang et al., 2016; Ji et al., 2014; Jansen et al., 2014; Zhao et al., 2013; Squizzato et al., 2013; Fu et al., 2008). The pathways for nitrate formation mainly include heterogeneous hydrolysis of N₂O₅ and gas-phase photochemical oxidation of NO₂. N₂O₅ is mainly produced from NO₂ (NO₂ + O₃ → NO₃ + O₂, NO₂ + NO₃ + M ↔ N₂O₅ + M). So we think it's reasonable to only use NO₂ in the calculation of NOR.

P33422, L4-6: should be two sentences.

This sentence has been split into two sentences: "The daily variation of NOR showed similar pattern as NH₄⁺ and NO₃⁻. Likewise, SOR had similar diurnal change as SO₄²⁻."

P33422, L11-14: show the correlation.

The correlation coefficient has been provided in the revised manuscript.

P33422, L27: a variety of cities (change to different cities?); a little (change to slightly?)

The phrase "a variety of cities" should be "several" and "a little" has been changed to "slightly". The whole sentence now is "Pathak et al. (2009) also reported an intercept value of 1.5 for several cities worldwide while Jansen et al. (2014) found a slightly smaller intercept value of 1.38 for Hangzhou."

P33424, L27: remove "differently".

It has been revised accordingly.

P33425, L13: suggest should be suggests.

It has been revised accordingly.

P33425, L14: but not should be but does not.

It has been revised accordingly.

P33426, L10: severe should be efficient.

It has been revised accordingly.

P33426, L18: were should be was.

There are many changes in this section in order to highlight the main conclusions of this article. This sentence is not included in the revised manuscript.

P33426, L24: But distinctively should be However.

There are many changes in this section in order to highlight the main conclusions of this article. This sentence is not included in the revised manuscript.

P33435: what are the shaded areas with different colors?

The shaded areas in orange refers to periods when visibility lower than 10 km and accompanied by precipitation. The shaded areas in grey refers to haze periods. We have clarified this in the revised manuscript.

P33437: it is difficult to be convinced that CO is a precursor of aerosol species; O₃ might be one type of oxidants in secondary aerosol formation, but not a precursor.

We agree with the reviewer that it might be inappropriate to refer to these species as precursors. So the title of this figure has been rephrased: "Figure 6. Diurnal profiles of meteorological variables, aerosol precursors (NO_x, SO₂), CO, O₃, PM_{2.5}, and major aerosol compounds (NO₃⁻, SO₄²⁻, NH₄⁺, EC, POC, SOC) under different visibility conditions."

P33439: if it is a ratio, then show a ratio, not a percentage.

It has been revised accordingly in the revised manuscript.

Anonymous Referee #2

Manuscript on "Characteristics of aerosol pollution during heavy haze events in Suzhou, China" evaluates the atmospheric concentrations of PM, its chemical composition, and light extinction properties to understand the characteristics of haze events at Suzhou. Their results showed that haze events are characterized by the presence of enhanced concentration of secondary aerosol species and their precursors; further, abundance of OM, (NH₄)₂SO₄, and NH₄NO₃ led to visibility impairment during the haze events. The study is useful towards understanding the role of pollutants in haze formation and visibility degradation and could help policy makers in specific control measures. This is an interesting piece of work, but in

my view, the manuscript needs to be strengthened in terms of scientific analysis and its novelty. Authors should highlight in the “Introduction” and in their Results and Discussion, the gaps in understanding (what is known and what we still need to understand through this work) related to aerosol pollution during haze events and how objectives and analyses presented in the manuscript for a specific location of Suzhou is different than reported information in literature for other locations influenced by haze events, e.g. Beijing etc. Also, there is a difficulty in the readability of the manuscript, this needs to be improved checking sentences throughout.

We have made significant revisions based on the reviewer’s comments to strengthen the paper. The novelty, originality and importance of this study can be summarized below. (1) Suzhou, an important city in the YRD, with annual haze days increased from only two days to more than 150 days from 1956 to 2011, has experienced the extremely serious aerosol pollution in Jan. 2013, but little is known about the chemical characteristics and sources of fine particles in this city. This paper provide basic data for implementing effective pollution control measures in Suzhou. (2) The discussion in the present study was based on high time resolution data which could provide detailed insight into the role the major chemical species in PM_{2.5} played during shorter haze periods. (3) The dominant species in PM_{2.5} and those responsible for the visibility reduction were identified. (4) The formation mechanism of sulfate and nitrate were explored as high secondary aerosol contributions to particulate pollution during haze events. (5) The impact of local and outside sources on aerosol pollution in urban Suzhou was discussed. We now highlighted these points in the manuscript and we have also strengthened the comparison of the haze characteristics between Suzhou and other locations such as Beijing etc. The language has also been proofread by professionals to enhance its readability.

Specific comments:

Section 2.1: Please discuss uncertainty involved in measurements using each of the equipments.

The uncertainties have been discussed for the measurement of each equipment in the revised manuscript.

Section 2.2.1: It would be helpful to the reader if you can include information on basis of “numbers” (such as $2.2 \times f_s(\text{RH})$ and so on, what does 2.2 indicate?) used in the equation 1 (IMPROVE algorithm).

The information of the numbers have been included in the revised manuscript: “The constant numbers in the above equation are extinction efficiencies for each chemical species under dry condition.”

Section 2.2.2: please provide reason for 48 h back trajectories calculation; typically, taking into account lifetime of aerosols of the order of seven-days in the lower troposphere, should you perform 7-day back trajectory calculations?

48-h back trajectory analysis has been used in many studies (Yu et al., 2014; Ji et al., 2014; Behera and Balasubramanian, 2014; Zhang et al., 2013; Huang et al., 2012; Huang et al.,

2010; Wang et al., 2005). We think the back trajectory analysis is mainly used to explore the origination of the air mass in the studied area. The transport of air mass might bring aerosols and/or precursors to the target area and consequently influence the aerosol pollution. The results of this analysis basically could imply the impact of regional transport on the existing aerosol in the studied area. So we think the back trajectory analysis don't mean to calculate the lifetime trajectory of the aerosol.

Section 3.2.1: How does the chemical composition observed at Suzhou compares with that at Beijing or other places where haze events are frequent. Please discuss if sources of aerosol species are distinct between Suzhou and others.

The chemical composition observed at Suzhou has been compared with that at Beijing or other places in the manuscript, such as “These secondary inorganic components in total constitute 93% of total WSIs, close to the result in Beijing (Gao et al., 2015; Tao et al., 2015).” and “The carbonaceous species, constituting 22% of PM_{2.5}, were dominated by organic carbon, which was $22.8 \pm 10.6 \text{ mg m}^{-3}$ and 3 to 29 times of that of elemental carbon ($2.79 \pm 2.58 \text{ mg m}^{-3}$), similar to those in Beijing (Tao et al., 2015).”. Other results in present study have also been compared to other places, for instance, “The relatively high ratios of OC/EC (10.6 ± 4.29), which were higher than the ratios in Beijing (7.1 ± 0.5) and Jinan (7.15 ± 1.78)”, “The ratio of $\text{NO}_3^-/\text{SO}_4^{2-}$ in this study was lower than that in Beijing, but higher than those in Shanghai (0.43), Qingdao (0.35), Taiwan (0.20), and Guiyang (0.13) (Wang et al., 2006; Yao et al., 2002; Hu et al., 2002; Fang et al., 2002; Xiao and Liu, 2004).”, etc.. We agree with the reviewer that to compare the sources of aerosol species between Suzhou and other places might be interesting. However, specific emission sources could not be identified only based on the existing data of species (water soluble ion, OC, EC) in PM_{2.5}. Some tracers of emission sources would be needed other than existing components.

Section 3.2.2: Please replace ‘AM peak’ and ‘PM peak’ by other relevant words. “It seemed that low visibility: : southwest wind” please explain the possible reason for this.

The “AM peak” and “PM peak” have been replaced by “morning peak” and “afternoon peak”. The possible explanation has been provided in the revised manuscript: “This might related to the topography. There are mountains located on the southwest which is not conducive to the diffusion of pollutants.”

Section 3.2.3: “In order to appoint: :” please change the word “appoint”. Your analysis show that while OM is the largest contributor to light extinction, but its contribution to PM during haze events is lower than rest of the water soluble constituents. Please discuss the possible reason for this.

(1) The word “appoint” has been changed to “determine” in the revised manuscript: “In order to determine the contribution of PM_{2.5} constituents to the visibility degradation, light extinction (b_{ext}) was reconstructed based on the revised IMPROVE algorithm.” (2) There might be two possible reasons. Firstly, OM refers to organic matter which is derived from multiplying OC concentrations by a factor of 1.8 to account for unmeasured atoms. Secondly, the extinction efficiencies for each species under dry conditions are different. The extinction efficiencies of OM (2.8 for Small OM and 6.1 for Large OM) are higher than $(\text{NH}_4)_2\text{SO}_4$ (2.2

and 4.8) and NH_4NO_3 (2.4 and 5.1).

Section 3.4: Please provide insights on emission sources corresponding to paths of clusters and PSCF analysis. Please discuss why a set of clusters are found to be different than the rest.

We agree with the reviewer that discussing the emission sources (such as biomass burning, coal combustion, vehicle emission, etc.) corresponding to paths of clusters and PSCF analysis would be of great interest. However, specific emission sources could not be identified only based on the existing data of species (water soluble ion, OC, EC) in $\text{PM}_{2.5}$. Some tracers of emission sources would be needed other than existing data. Besides, the current paper is already very long and focuses on the chemical contribution, formation mechanism and regional impact. Introducing additional analysis of emission sources would weaken the overall focus, increase the length, and decrease the readability. Trajectories are calculated on basis of meteorological parameters such as wind direction and wind speed. Different meteorological parameters would result in different trajectory. The pollution characteristics (such as aerosol pollution level, chemical composition, light extinction, etc.) in clusters might differ from each other. There might be many reasons, for instance, the trajectories pass over different areas and/or the local emission sources varied.

Anonymous Referee #3

Reviewer's comments for "Characteristics of aerosol pollution during heavy haze events in Suzhou, China" by Tian et al. The manuscript by Tian et al. presented a field study for $\text{PM}_{2.5}$ mass concentration, $\text{PM}_{2.5}$ chemical composition, and associated gaseous precursors during haze events in a major Chinese city located in the Yangtze River Delta. Like many other studies in China, this work highlighted the large contributions of secondary aerosol species in $\text{PM}_{2.5}$. Source regions were analyzed based on the back trajectory calculations.

The new perspective (in my opinion) this paper brought to us is the results on the main contributors of the light extinction in $\text{PM}_{2.5}$ components. These results are interesting and potentially have policy implications, because light extinction is directly linked to the visibility, which is one of the major public concerns in China. In the current manuscript, this assessment was based on the measured concentrations of chemical species weighted by the mass-extinction coefficients derived from the revised IMPROVE algorithm. The authors identified OM, sulfate, and nitrate as the major contributors of visibility impairment. However, I do agree with reviewer #1 that such analysis was not convincing without appropriate comparisons with other measurements. I would suggest the authors to strengthen this analysis before this paper can be published in ACP, since it seems to be crucial for the main conclusions.

A possible way to do this can be including a more comprehensive comparison between the light extinction reconstructed from chemical species with that derived from the visibility. A previous study (Chen et al. ACP, 2012) in North China has shown that the visibility-derived ambient light extinction can be well reconstructed by an optical model with measured number-size distribution, hygroscopicity, and RH. For the revision, a scatter plot and/or a time series plot could be included for comparison of the light extinction coefficients derived from both methods. Uncertainties should also be included. In the case of poor agreement,

several hypotheses can be tested, including: 1) Coefficients developed based the IMPROVE data are not suitable for the aerosol populations in China, e.g., due to the differences in size distributions. In this case better parameterizations are needed for the haze over China. 2) The RH (or other key parameters) measurement may not be accurate; e.g., Fig. 2 shows severe haze events are associated with high RH ($\sim 90\%$), where the RH sensor may have a large error. In this case the assessment can be biased for contributions between hygroscopic inorganic species and hydrophobic OM. Some caveats should be discussed.

We have compared the light extinction coefficients reconstructed by IMPROVE algorithm with that calculated by the optical model as the reviewer suggested. Strong correlations were observed ($R^2 = 0.952$) with a slope of 0.837, confirming that the reconstructed b_{ext} from IMPROVE algorithm were basically reliable. We also discussed the possible reasons for the deviations between the results.

Another suggestion is that the authors can also present the contributions of different species to PM_{2.5} mass concentration in addition to light extinction (e.g., in Fig. 5 and 9, Table 1). This analysis will be based on measured variables and thus less ambiguous. Such results can be useful in the context of aerosol health effect, which is another major concern related to the air pollution.

Thanks for the valuable suggestion. The aerosol compositions in the three haze events have been illustrated in the pie chart in figure 3 in the revised manuscript.

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1 **Characteristics of aerosol pollution during heavy haze events**
2 **in Suzhou, China**

3

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15

16 **Abstract**

17 ~~A comprehensive measurement was carried out to analyze the heavy haze events in~~
18 ~~Suzhou in January 2013 when extremely severe haze weather events~~~~pollution~~
19 occurred in many cities in China, especially in the ~~East part of the country, in~~
20 ~~January 2013. Comprehensive measurements including hourly concentrations of~~
21 ~~PM_{2.5}, and its major chemical components~~~~sition (including water-soluble inorganic~~
22 ions, OC, and EC), and ~~related~~ gas-phase precursors were ~~obtained~~~~conducted~~ via
23 on-line monitoring system ~~in Suzhou, a medium size city of Jiangsu province, just~~
24 ~~east of Shanghai. Based on these data, detailed aerosol composition, light extinction~~
25 ~~and gas-phase precursors were analyzed to understand the characteristics of the haze~~
26 ~~events, moreover, the formation mechanism of nitrate and sulfate in PM_{2.5} and the~~
27 ~~regional sources deduced from trajectory and PSCF were discussed to explore the~~
28 ~~origin of the heavy aerosol pollution. The results showed that frequent haze events~~
29 ~~were occurred on Jan. 2013 and the concentrations of PM_{2.5} frequently~~~~often~~
30 150 $\mu\text{g m}^{-3}$ ~~on hazy days~~~~during the haze occurrence~~, with ~~the~~ maximum
31 ~~reaching~~~~concentration of~~ 324 $\mu\text{g m}^{-3}$ on Jan. 14, 2013. Unfavorable weather
32 conditions (high RH, and low rainfall, wind speed and atmospheric pressure) ~~were~~
33 ~~conducive for the haze formation.~~ ~~High concentrations~~ of secondary aerosol species
34 (including SO_4^{2-} , NO_3^- , NH_4^+ , and SOC) and ~~gaseous~~ precursors were observed during
35 the ~~first two~~ haze events, ~~while elevated primary carbonaceous species emissions~~
36 ~~were found during the third haze period, pointing to different haze formation~~
37 ~~mechanisms.~~ ~~Additionally~~, OM, $(\text{NH}_4)_2\text{SO}_4$, NH_4NO_3 were ~~demonstrated~~~~found~~ to be
38 the major contributors to ~~the~~ visibility impairment ~~but the share differed from haze~~
39 ~~events. This study also found that the~~ ~~High concentrations~~ of sulfate ~~and nitrate~~
40 might be explained by ~~the~~ ~~homogeneous gas-phase reactions under low RH conditions~~
41 ~~and by heterogeneous processes under relatively high RH conditions.~~ ~~heterogeneous~~
42 ~~reactions in the aqueous surface layer of pre-existing particles or in cloud processes~~
43 ~~while nitrate might be mainly formed through homogeneous gas-phase reactions. The~~
44 ~~results of~~ ~~Analysis of air mass~~ trajectory clustering and ~~potential source contribution~~
45 ~~function~~ ~~the PSCF method~~ manifested that aerosol pollutions in the studied areas were

46 mainly ~~caused~~~~affected~~ by local activities and surrounding sources transported from
47 nearby cities.

48

49 1. Introduction

50 ~~High occurrence of haze events (is defined as~~ visibility lower than 10 km
51 ~~when under the conditions of~~ relative humidity < 80%.) Rapidly increased air
52 pollution in China in the past several decades have resulted in frequent occurrences of
53 haze events, which have caused~~is of~~ great concern to the scientific community as well
54 as both scientists and the public (Zhang et al., 2012). ~~in China in recent years because~~
55 ~~of its~~ Haze events have great adverse effects on ~~the people's~~human health, traffic,
56 climate, and other important aspects (Zhang et al., 2015; Charlson et al., 1987;
57 Ramanathan and Vogelmann, 1997; Tegen et al., 2000; Yu et al., 2002; Tie et al.,
58 2009a; Tie et al., 2009b). Fine particles (PM_{2.5}, aerosols with an aerodynamic
59 diameter of 2.5 microns or less) ~~are~~has a largely responsible for haze formation due to
60 their ability in impact on visibility by light extinction including scattering and
61 absorbing solar and infrared radiation ~~and is mainly responsible for the formation of~~
62 ~~haze~~(Yu et al., 2014). ~~The~~Light extinction of PM_{2.5} is highly associated with ~~its~~the
63 chemical compositions (Tao et al., 2014). Water-soluble inorganic ions and
64 carbonaceous species often account for major fractions of PM_{2.5} and are important
65 contributors to visibility impairment (Tan et al., 2009; Pathak et al., 2009), and thus
66 have been. ~~Therefore, these species were emphatically~~ investigated extensively in
67 researches related to haze occurrence (Yang et al., 2005; Jansen et al., 2014; Pathak et
68 al., 2009). ~~However, most of these existing~~ studies were based on ~~artificial filter~~
69 sampling and off-line analysis ~~and had~~which has its limitations ~~of in~~ providing
70 detailed insight into the roles ~~of the major chemical~~these species played during shorter
71 haze periods.

72 High contributions of secondary inorganic aerosols (SIA, including sulfate,
73 nitrate and ammonium), the predominant water-soluble ionic species in PM_{2.5}, to
74 visibility reduction have been observed in many cities in China (Huang et al., 2014).

75 Gas-phase or liquid-phase reactions of sulfur dioxide and nitrogen oxides ~~are~~ the
76 primary ~~mechanisms forming source of~~ aerosol sulfate and nitrate. For the formation
77 of sulfate, homogeneous gas phase reaction of SO₂ with OH radical, ~~and~~
78 heterogeneous reactions in the aqueous surface layer of pre-existing particles, and
79 in-cloud processes are the primary mechanisms (Wang et al., 2006). The rates of
80 gas-phase and liquid-phase reactions of SO₂ were ~~close~~ similar in summer while the
81 heterogeneous processes were responsible for the oxidation in winter (Hewitt, 2001).
82 Nitric acid can be formed from homogeneous gas-phase reactions of NO₂ with OH or
83 O₃ and from heterogeneous hydrolysis of N₂O₅, which occurred predominantly during
84 daytime and nighttime, respectively (Khoder, 2002). Both sulfuric acid and nitric acid
85 ~~will~~ react with alkaline substance in the atmosphere, mostly ammonia under ambient
86 conditions to produce salts (Hewitt, 2001). The neutralization of sulfuric acid by
87 ammonia has been found to be preferred over the formation of ammonium nitrate
88 (Warneck, 1999). Thus, the formation of ammonium nitrate in fine particles is usually
89 under significantly neutralized or ammonium-rich ~~sulfate~~ conditions (Pathak et al.,
90 2009). There are various factors influencing the formation of aerosol sulfate and
91 nitrate, such as the levels of gaseous precursors (SO₂, NH₃, NO_x) and oxidants, the
92 characteristics of pre-existing aerosols, and meteorological conditions. These factors
93 may vary by location, ~~this may result~~ in different formation mechanism in different
94 areas. For instance, different formation pathways had been reported for nitrate in
95 ammonia-rich and ammonia-deficient areas (Pathak et al., 2009).

96 ~~Due to the rapid economic and industrial developments and urbanization in past~~
97 ~~few decades, the visibility reduction become increasingly serious in China, especially~~
98 ~~in the East (Zhang et al., 2012).~~ In January 2013, extremely severe, persistent and
99 widespread haze weather occurred in 10 provinces, ~~regions and cities~~ in central and
100 eastern China. These serious pollution events not only ~~had~~ great adverse effects on
101 human health, ~~as seen in~~ a sharp increase in respiratory diseases, but also
102 ~~caused~~ immeasurable economic loss (Huang et al., 2014; Chen et al., 2013). High
103 secondary inorganic and organic aerosol contributions to particulate pollution during

104 these haze events were reported in a recent study based on the measurements at urban
105 sites in Beijing, Shanghai, Guangzhou and Xi'an, which located in the northern,
106 eastern, southern and western regions of China, respectively (Huang et al., 2014).
107 ~~This result indicated that i~~In addition to ~~investigating~~investigate the primary particulate
108 emissions, ~~it's also important to explore~~the formation mechanisms ~~and effect factors~~
109 of these secondary species and related affecting factors also need to be understood in
110 order to controlling PM_{2.5} levels in China.

111 The Yangtze River Delta (YRD), ~~together with~~the Pearl River Delta,
112 Beijing–Tianjin–Tangshan, and the Sichuan Basin are ~~of most interesting as they are~~
113 the four regions with heaviest haze influenceregions in China. The characteristics and
114 formation mechanisms of haze in the YRD are different from other haze regions, such
115 as Beijing and the Pearl River Delta (Fu et al., 2008). Suzhou is located in the
116 heartland of YRD region and is an important city ~~in the YRD~~. ~~Suzhou~~It suffered from
117 the extremely serious aerosol pollution in Jan. 2013. With the tremendous economic
118 growth over the past 30 years, Suzhou has experienceds high levels of air pollution as
119 reflected in the frequency of haze occurrence. The annually haze days in Suzhou
120 increased from only two days to more than 150 days from 1956 to 2011, i.e. over 40%
121 of days were hazy in 2011. The lower visibility, particularly the haze, has become a
122 major concern of the citys. ~~H~~However, only a few ~~studies~~researches have focused
123 ~~on been carried out to study the~~ haze events in Suzhou and. ~~L~~ittle is known about the
124 chemical characteristics and sources of fine particles in this city~~Suzhou~~. To fill these
125 knowledge gaps~~Therefore~~, an intensified monitoring campaign was launched from
126 December 2012 to January 2013 to ~~collect~~get insight into the haze occurrence in
127 ~~urban Suzhou. On the basis of~~ high temporal resolution chemical and ~~measurements~~
128 ~~combined with~~ meteorological datas. ~~T~~The objectives of this study are to (1) identify
129 the dominant species in PM_{2.5} and responsible for the visibility reduction; (2) explore
130 the formation mechanism of the aerosol pollution; (3) study the impact of local,
131 nearby and remote~~transport~~ sources on the formation of haze in urban Suzhou.

132

133 2. Methodology

134 2.1 Field observations

135 The sampling station was set up at the roof of ~~one~~the building in Suzhou Institute of
136 Environmental Sciences (31°20'N, 120°36'E) (Fig. 1), about 300 m west to Nanyuan
137 South Road and 360 m north to S Ring Road Elevated Bridge. There is no industrial
138 source nearby and ~~the~~is site is representative of an urban residential and commercial
139 ~~environment~~-area. Suzhou is located in the center of Yangtze River Delta (YRD) and
140 about 80 km east ~~of~~ Shanghai and 200 km west ~~of~~ Nanjing.

141 ~~On-line hourly PM_{2.5} mass concentrations, ionic species and OC/EC were~~
142 ~~measured by tapered element oscillating microbalance (TOEM), URG 9000 Ion~~
143 ~~Chromatography, Sunset semi-continuous OC/EC analyzer, respectively. On-line~~
144 hourly PM_{2.5} mass concentrations were measured using tapered element oscillating
145 microbalance (TEOM1405, Thermo Scientific Corp., MA, US) with the heating
146 temperature of 50 °C. Some of the volatile particulate matter might be lost at 50 °C,
147 but comparisons with collocated filters showed that the loss was less than 10%-20%
148 of the gravimetric mass (Chow et al., 2008).

149 Hourly real-time concentrations of five cations (Na⁺, K⁺, NH₄⁺, Ca²⁺, and Mg²⁺)
150 and four anions (F⁻, Cl⁻, NO₃⁻, and SO₄²⁻) in PM_{2.5} were determined by URG Series
151 9000 Ambient Ion Monitor (AIM, URG Corporation, Chapel Hill, NC). The system
152 consists of a particle collection unit and two ion chromatograph analyzers for cation
153 and anion analyses. PM_{2.5} was separated by a sharp-cut cyclone inlet operating at a
154 flow rate of 3 L/min. The air was drawn through a liquid diffusion parallel-plate
155 denuder to remove the interfering acidic and basic gases. A Steam-Jet Aerosol
156 Collector was placed downstream of the denuder for collection and extraction of
157 particles. The water extract was subsequently injected into the two ICs once an hour.
158 The estimated uncertainties of the AIM measurements were approximately less than
159 15% (Trebs et al., 2004; Pathak et al., 2011). Some measures were taken out to reduce
160 error, for instance, standards solutions were periodically injected to check the

161 consistency of sensitivity of the detectors and air flow rate is frequently checked using
162 a calibrated flow meter.

163 A semi-continuous OC/EC analyzer (Sunset Laboratory, Forest Grove, Oregon,
164 USA) was applied to determine the carbonaceous species in PM_{2.5}. This instrument
165 used the thermal-optical transmittance method based on NIOSH Method 5040.
166 Organic compounds were vaporized in pure helium and then oxidized to CO₂ in a
167 manganese dioxide oxidizing oven. CO₂ was then quantified by non-dispersive
168 infrared detector. EC was then desorbed in an oxygen blend carrier gas and then
169 oxidized and quantified using the same method as for OC. The split point between the
170 pyrolyzed carbon formed from the organic carbon during the heating and EC that was
171 originally in the sample was determined by measuring the transmission of a laser
172 beam through the filter. Known volume of methane was injected, oxidized and
173 quantified as an internal standard. Good correlations were found between the data
174 measured by this instrument and filter-based laboratory analyses (Bae et al., 2004).

175 Visibility was monitored using the Belfort Model 6000 Visibility Sensor (Belfort
176 Instrument Corp., MD, US). Trace O₃, SO₂, NO-NO₂-NO_x and CO gases were
177 obtained with a resolution of 1 h by applying online analyzers (Thermo Instruments,
178 TEI 49i, 43i, 42i and 48i, respectively) (Wang, 2016 #343). Meteorological
179 parameters were collected using Met Station One (Met One Corp., OR, US).

181 **2.2 Data analysis methods**

182 **2.2.1 Reconstruction of the light extinction coefficient**

183 The light extinction (b_{ext}) which is the sum of light scattering by particle ($b_{\text{s,p}}$),
184 absorption by particle ($b_{\text{a,p}}$), scattering by gas ($b_{\text{s,g}}$), and absorption by gas ($b_{\text{a,g}}$), is
185 reconstructed according to the revised IMPROVE algorithm as following (Pitchford et
186 al., 2007):

$$187 \quad b_{\text{ext}} = b_{\text{s,p}} + b_{\text{a,p}} + b_{\text{a,g}} + b_{\text{s,g}}$$

$$\begin{aligned}
188 \quad & \approx 2.2 \times f_s(\text{RH}) \times [\text{Small } (\text{NH}_4)_2\text{SO}_4] + 4.8 \times f_L(\text{RH}) \times [\text{Large } (\text{NH}_4)_2\text{SO}_4] \\
189 \quad & + 2.4 \times f_s(\text{RH}) \times [\text{Small } \text{NH}_4\text{NO}_3] + 5.1 \times f_L(\text{RH}) \times [\text{Large } \text{NH}_4\text{NO}_3] \\
190 \quad & + 2.8 \times [\text{Small OM}] + 6.1 \times [\text{Large OM}] \\
191 \quad & + 1 \times [\text{Fine Soil}] + 1.7 \times f_{ss}(\text{RH}) \times [\text{Sea Salt}] \\
192 \quad & + 0.6 \times [\text{Coarse Mass}] + 10 \times [\text{EC Mass}] \\
193 \quad & + 0.33 \times [\text{NO}_2 \text{ (ppb)}] + \text{Rayleigh Scattering} \quad (1)
\end{aligned}$$

194 where $f_s(\text{RH})$ and $f_L(\text{RH})$ are the water growth factors for small- and large-sized
195 distribution of sulfate and nitrate, respectively; and $f_{ss}(\text{RH})$ is the water growth factor
196 for sea salt. Water growth factors are adopted according to Pitchford~~Isakov~~ et al.
197 (Pitchford et al., 2007)~~(Isakov et al., 2007)~~. The constant numbers in the above
198 equation are extinction efficiencies for each chemical species under dry condition.
199 $(\text{NH}_4)_2\text{SO}_4$ mass is estimated as 1.38 times of~~by the~~ SO_4^{2-} mass ~~multiplied by a factor~~
200 ~~of 1.38~~, and ~~the~~ NH_4NO_3 mass is estimated by the 1.29 times of NO_3^- mass ~~multiplied~~
201 ~~by a factor of 1.29 based on the assumption~~ that SO_4^{2-} and NO_3^- are fully
202 neutralized by NH_4^+ in the forms of $(\text{NH}_4)_2\text{SO}_4$ and NH_4NO_3 , respectively, according
203 to the revised IMPROVE method. Organic matter (OM) is estimated as 1.8 times
204 ~~of derived from multiplying~~ OC concentrations ~~by a factor of 1.8~~ to account for
205 unmeasured fractions~~atoms~~.

206 The concentrations of sulfate, nitrate, and OM are divided into small- and
207 large-sized fractions in this algorithm. The size modes are described by log-normal
208 mass size distributions with geometric mean diameter and geometric standard
209 deviations. Concentrations of sulfate, nitrate, and OM in the large- and small-mode
210 are estimated ~~by~~ using the following equations (taking sulfate as an example):

$$211 \quad [\text{Large } (\text{NH}_4)_2\text{SO}_4] = [\text{Total } (\text{NH}_4)_2\text{SO}_4]^2/20, \text{ for } [\text{Total } (\text{NH}_4)_2\text{SO}_4] < 20 \mu\text{g m}^{-3} \quad (2)$$

$$212 \quad [\text{Large } (\text{NH}_4)_2\text{SO}_4] = [\text{Total } (\text{NH}_4)_2\text{SO}_4], \text{ for } [\text{Total } (\text{NH}_4)_2\text{SO}_4] > 20 \mu\text{g m}^{-3} \quad (3)$$

$$213 \quad [\text{Small } (\text{NH}_4)_2\text{SO}_4] = [\text{Total } (\text{NH}_4)_2\text{SO}_4] - [\text{Large } (\text{NH}_4)_2\text{SO}_4] \quad (4)$$

214

215 **2.2.2 Air mass back trajectory**

216 To study the impact of local and regional sources on the aerosol pollution in Suzhou,
217 ~~the~~ 48-h back trajectories starting at 100 m from the sampling site were calculated ~~by~~
218 using the NOAA HYSPLIT model (http://ready.arl.noaa.gov/HYSPLIT_traj.php). The
219 back trajectories were calculated four times per day at starting times of 04:00, 10:00,
220 16:00, and 22:00 UTC, i.e. 12:00, 18:00, 00:00, and 06:00 local times, respectively.
221 The trajectory cluster analysis was based on the GIS-based software TrajStat (Wang et
222 al., 2009).

223

224 **2.2.3 Potential source contribution function**

225 The potential source contribution function (PSCF) method is based on the results of
226 HYSPLIT model and can be used to identify the regional sources. The zone of
227 concern is divided into $i \times j$ small equal grid cells. The PSCF value for the ij th grid
228 cell is calculated as: $PSCF_{ij} = m_{ij}/n_{ij}$, where n_{ij} is designated as the number of
229 trajectory segment endpoints that fall in the ij th cell and m_{ij} is defined as the number
230 of trajectory endpoints with pollutants concentrations higher than an set criterion
231 (Ashbaugh et al., 1985; Wang et al., 2009). In present study, the average
232 concentrations were treated as the criterion (Hsu et al., 2003). The PSCF values were
233 multiplied by a weighting function W_{ij} to reduce the effect of small values of n_{ij} and to
234 better reflect the uncertainty in the values for the cells with small n_{ij} values. The
235 weighting function W_{ij} is defined as follows (Polissar et al., 1999):

$$236 \quad W_{ij} = \begin{cases} 1.00, & 80 < n_{ij} \\ 0.70, & 20 < n_{ij} \leq 80 \\ 0.42, & 10 < n_{ij} \leq 20 \\ 0.05, & n_{ij} \leq 10 \end{cases}$$

237

238 The PSCF value can be interpreted as the conditional probability that ~~the~~-air
239 masses with pollutants concentration greater than the set criterion pass through the ij th

240 cell during transport to the receptor site (Wang et al., 2009). That is, cells with high
241 PSCF values are indicative of regions having high potential contributions to the
242 pollution at the receptor site.

243

244 3. Results and discussion

245 3.1 General characteristics of haze events

246 As illustrated in Fig. 2, the visibility varied from a few hundred meters to more than
247 50 km with a minimum value of only 322 m on Jan. 15, 2013, which was
248 accompanied by high RH (82%). During the 2-month observation period, there were a
249 ~~totally of~~ ten periods when visibility was below 10 km. ~~Excluding~~~~cept for~~ the five
250 periods ~~which were~~ accompanied by precipitation, the other five periods were
251 identified as haze events ~~were identified~~ and all of these events occurred in January
252 2013. During the haze occurrence, ~~the~~ hourly concentrations of PM_{2.5} often exceeded
253 150 µg m⁻³, with a maximum concentration of 324 µg m⁻³ observed on Jan. 14, 2013.
254 These concentrations were generally higher than those in normal periods. The daily
255 concentrations of PM_{2.5} on haze days varied from 148 to 196 µg m⁻³, which were 1.97
256 to 2.61 times the Grade II criteria of the national ambient air quality standard (75 µg
257 m⁻³). ~~These values is~~ ~~were~~ comparable to that observed~~the PM_{2.5} concentrations~~ in
258 Nanjing ~~where the with~~ average PM_{2.5} value of was 175.6 µg m⁻³, but ~~slightly~~~~little~~
259 higher than those in some other cities in YRD ~~where with the~~ mean values were
260 generally lower than 147.3 µg m⁻³ when haze occurred in Jan. 2013 (Wang et al.,
261 2014a; Wang et al., 2014d). The aerosol pollution happened in northeast China such
262 as in Beijing, Tianjin, and Shijiazhuang were much severer, for instance, the daily and
263 hourly concentrations of PM_{2.5} were up to 368 µg m⁻³ and 462 µg m⁻³ in Tianjin in
264 January 9 to 13, 2013, ~~and~~ the ~~hourly~~-maximum hourly values of approximately 1000
265 µg m⁻³ ~~was~~~~ere~~ recorded in Beijing and Shijiazhuang in Jan. 2013 (Ji et al., 2014; Han
266 et al., 2014; Wang et al., 2015).

267 ~~The duration of haze events comprised a~~ approximately 40% of the time in~~whole~~

268 January 2013 met haze weather criteria, whereas no haze appeared in December 2012.
269 LessLow amount of rainfall in ~~the~~ January might be one of the factors causing ~~the~~
270 long duration of haze. ~~The~~Relative humidity (RH) was reported to be an important
271 contributor to the visibility reduction. In ~~the~~ present study, ~~the RH increased with the~~
272 ~~reduction of~~visibility decreased with increasing RH, e.g. when RH increased from
273 42% to 78%, ~~the~~visibility decreased from 42 km at 2:00 p.m. on 17 January to 4 km
274 at 7:00 a.m. on 19 January. Statistically, ~~the~~RH was relatively higher during haze
275 occurrence than clear periods. Low wind speed, smaller than 5 m s⁻¹, ~~was~~
276 frequently observed during this winter. Furthermore, ~~the~~wind speeds ~~was~~ mostly
277 less than 1 m s⁻¹ during the haze events, lower than those in Beijing (Yang et al.,
278 2015). Besides, atmospheric pressure was also found to be relatively low during the
279 haze occurrences. The stagnant air, due to low wind speed and pressure, was
280 unfavorable for ~~the aerosol~~ horizontal transport or vertical diffusion ~~of aerosols~~, and
281 therefore resulted in~~leading to the increase of~~ aerosol accumulation~~concentration~~.
282 Therefore, unfavorable weather conditions (high RH, and low rainfall, wind speed and
283 atmospheric pressure) were among the causes ~~might provide external caused~~
284 beneficially for the ~~formation of~~ haze in January 2013 in Suzhou as well as
285 many other cities (Wang et al., 2014b; Wang et al., 2014c; Wang et al., 2014d; Han et
286 al., 2014; Yang et al., 2015).

287 In order to get more insights of the haze formation in this region, three haze
288 events, which occurred on Jan. 19, from Jan. 21 to 26, and on Jan. 30, respectively,
289 were further discussed below. Generally, the meteorological parameters and aerosol
290 pollution level were comparable during these three haze occurrences, except for the
291 relatively lower RH and higher temperature in the third haze events.–

292

293 **3.2 PM_{2.5} chemical composition and light extinction**

294 **3.2.1 PM_{2.5} chemical composition**

295 The temporal variations of the concentrations of water-soluble inorganic ions (WSIIs)
296 ~~we~~are illustrated in Fig. 3. The mean concentration of WSIs (including four anions
297 and five cations) was $48.8 \pm 24.6 \mu\text{g m}^{-3}$, accounting for 40% of $\text{PM}_{2.5}$ mass
298 concentration, ~~slightly little~~ lower than that in Beijing which was $69.4 \pm 55.8 \mu\text{g m}^{-3}$
299 and accounted for 43% of $\text{PM}_{2.5}$ (Tao et al., 2015). SO_4^{2-} was the most abundant
300 species in WSIs, with averaged value of $21.1 \pm 13.5 \mu\text{g m}^{-3}$, followed by NH_4^+ (13.9
301 $\pm 5.69 \mu\text{g m}^{-3}$) and NO_3^- ($10.7 \pm 6.75 \mu\text{g m}^{-3}$), accounting for 43%, 29% and 21% of
302 WSIs, respectively. These secondary inorganic components ~~in totally~~ constitute 93%
303 of total WSIs, close to the result in Beijing (Gao et al., 2015; Tao et al., 2015). The
304 rest of ions, Na^+ ($1.36 \pm 0.43 \mu\text{g m}^{-3}$), K^+ ($0.85 \pm 0.45 \mu\text{g m}^{-3}$), Cl^- ($0.54 \pm 1.28 \mu\text{g}$
305 m^{-3}), Ca^{2+} ($0.34 \pm 0.27 \mu\text{g m}^{-3}$), F^- ($0.06 \pm 0.72 \mu\text{g m}^{-3}$), Mg^{2+} ($0.05 \pm 0.07 \mu\text{g m}^{-3}$),
306 each had minor contribution ($< 3\%$) to WSIs.

307 NO_3^- and SO_4^{2-} are mainly formed from the transformation of their ~~respective~~
308 ~~gaseous~~ precursors of NO_x and SO_2 (Wang et al., 2005). The emission ratio of NO_x to
309 SO_2 was 17.2–52.6 for motor vehicles and 0.527–0.804 for stationary sources in the
310 Yangtze River Delta, which means that the emissions of SO_2 from motor vehicles
311 were much less than NO_x , but the emissions of SO_2 from stationary sources such as
312 power plants, industrial boilers and furnaces were relatively higher than NO_x (Fu et
313 al., 2008). Thus, the mass ratio of $\text{NO}_3^-/\text{SO}_4^{2-}$ could be used as an indicator of the
314 relative importance of mobile and stationary sources of sulfur and nitrogen in the
315 atmosphere (Arimoto et al., 1996). In ~~the~~ present study, the averaged ratios of
316 $\text{NO}_3^-/\text{SO}_4^{2-}$ and NO_x/SO_2 were 0.59 and 5.68, respectively, indicating that emissions
317 from vehicles and stationary sources were both important in Suzhou. The ratios of
318 $\text{NO}_3^-/\text{SO}_4^{2-}$ in this study ~~wasere~~ lower than ~~thatthe ratio~~ in Beijing, but higher than
319 those ~~reported~~ in Shanghai (0.43), Qingdao (0.35), Taiwan (0.20), and Guiyang (0.13)
320 (Wang et al., 2006; Yao et al., 2002; Hu et al., 2002a; Fang et al., 2002; Xiao and Liu,
321 2004).

322 The $\text{NO}_3^-/\text{SO}_4^{2-}$ ratio was relatively higher for 20% worst visibility hours (0.58)
323 than 20% best visibility hours (0.54), ~~which were 0.58 and 0.54, respectively,~~

324 suggesting that vehicle emission might play an important role in haze pollution. This
325 was in agreement with the result in Guangzhou, where the $\text{NO}_3^-/\text{SO}_4^{2-}$ ratio was 1.02
326 under stagnation and 0.55 in normal days, but contrary to that in Beijing, where the
327 ratio in haze days (0.6489) was lower than in normal days (0.9683) (Tan et al., 2009;
328 Wang et al., 2006). In present study, NO_x concentration greatly exceeded that of SO_2
329 during haze period, coincided with the result in Guangzhou, but disagreed with that in
330 Beijing (Tan et al., 2009). Previous studies have indicated that high NO_x emission
331 may reduce the formation of OH and H_2O_2 , and further decrease the possibility of
332 SO_4^{2-} formation (Tan et al., 2009). Thus, the elevation of NO_3^- concentration under
333 worse visibility conditions was greater than that of SO_4^{2-} in both Suzhou and
334 Guangzhou. Besides of lower $\text{NO}_3^-/\text{SO}_4^{2-}$ ratio, Wang et al. also found lower NO_2/SO_2
335 ratio and lower ratio of ($\text{NO}_3^-/\text{SO}_4^{2-}$) to (NO_2/SO_2) in haze days than that in clear days
336 in Beijing, and summarized that the formation rate of nitrate might not be the
337 controlling factor for the nitrate concentrations in $\text{PM}_{2.5}$ (Wang et al., 2006). The low
338 $\text{NO}_3^-/\text{SO}_4^{2-}$ ratios found in haze days in Beijing was considered to be related to the
339 thermodynamic characteristic of NH_4NO_3 (Wang et al., 2006). The ratios of NO_x/SO_2
340 in present study were 6.89 for 20% worst visibility period higher than 4.30 for 20%
341 best visibility period. The ratios of ($\text{NO}_3^-/\text{SO}_4^{2-}$) to (NO_x/SO_2) were also lower for
342 worse visibility period in present study, in accordance with that in Beijing, suggesting
343 showing that the nitrate concentrations may be also greatly affected by the
344 re-volatilization of NH_4NO_3 (Tan et al., 2009) ~~as those in Beijing.~~

345 The carbonaceous species, constituting 22% of $\text{PM}_{2.5}$, were dominated by
346 organic carbons, which were $22.8 \pm 10.6 \mu\text{g m}^{-3}$ and 3 to 29 times of that of
347 elemental carbon ($2.79 \pm 2.58 \mu\text{g m}^{-3}$), similar to those in Beijing (Tao et al., 2015).
348 The relatively high ratios of OC/EC (10.6 ± 4.29), which were higher than the ratios
349 in Beijing (7.1 ± 0.5) and Jinan (7.15 ± 1.78), demonstrated the existence of
350 secondary organic carbon (SOC) (Ji et al., 2014; Zhang et al., 2014). The
351 concentrations of SOC were estimated by applying the EC tracer method, which has
352 been widely used to estimate the secondary organic aerosol contribution to $\text{PM}_{2.5}$

353 concentrations (Castro et al., 1999; Yang et al., 2005). The minimum ratio of OC/EC
354 was 3.09 in the present study. ~~So t~~The estimated ~~concentrations of~~ SOC ~~wasere~~ $14.2 \pm$
355 $5.69 \mu\text{g m}^{-3}$, contributing 65% on average to OC. The ~~ratios of~~ SOC/OC ~~wasere~~
356 higher than 0.5 during almost the whole sampling ~~periodtime~~ except ~~onfor the periods~~
357 ~~around~~ Jan. 30, when the third haze event occurred. This ratio was higher than most of
358 the results found in other areas such as in Beijing and Guangzhou (Yang et al., 2005;
359 Tan et al., 2009).

360 ~~Overall, according to the percentage of each species in PM_{2.5} mass, the m~~Major
361 components in PM_{2.5} were found to be SO_4^{2-} (17%), SOC (14%), NH_4^+ (12%), NO_3^-
362 (8%), and POC (6%) ~~with total percentage of 57%. It is noted that t~~The topfirst four
363 ~~componentspecies~~ were mainly from secondary sources. ~~In addition, the~~
364 ~~concentrations of~~ PM_{2.5} ~~wasere~~ significantly correlated with these secondary species,
365 revealing that gas to particle conversion ~~during winter in this region~~ was severe in
366 winter and had great impact on aerosol pollution in this region. It's worth noting that
367 the aerosol composition in the third haze event was distinct from the other two (Fig. 3)
368 as seen from the higher proportion of carbon species from primary emissions (POC
369 and EC) and lower fraction of secondary formation components (SIA and SOC),
370 indicating different haze formation mechanism in the third haze event.

371

372 ~~3.2.2 Variations of aerosol particles and precursors~~

373 ~~Fig. 4 diagrammed the diurnal variation of meteorological parameters, various aerosol~~
374 ~~components and, the precursors and some other important gaseous species under three~~
375 ~~different visibility conditions (i.e., (1) all data, (2) visibility ≤ 10 km, (3) visibility $>$~~
376 ~~10 km). The daily variation of gas phase compounds were different between species~~
377 ~~and were mainly controlled by the direct surface emissions (such as NO_x , SO_2 , and~~
378 ~~CO) or photochemical process (O_3). There were a distinct AM morning peak and a~~
379 ~~less distinct afternoon PM peak, consistent with morning and afternoon PM rush hours~~
380 ~~for NO_x and CO . This might related to the heavy traffic emission in the rush hours~~

381 and the strong elevation of the Planetary Boundary Layer heights at noon. In contrast,
382 there was only one mid-day peak for SO₂. This diurnal profiles were similar to those
383 observed in Guangzhou (Hu et al., 2002b) and Maryland (Antony Chen et al., 2001).
384 In the latter study, the dominant source of SO₂ was considered to be the long range
385 transport from the industrialized Midwest and with the deep boundary layer around
386 noon, SO₂ aloft mixed more effectively down to the surface and thus caused the
387 mid-day peak of SO₂. The reasons for the diurnal variation of SO₂ observed in present
388 study need further investigation. Similar to the diurnal distribution of SO₂, O₃ also
389 showed one distinct peak around noon due to the strong photochemistry at that time
390 (Quan et al., 2014).

391 For the aerosol components, EC which was also produced by the surface
392 emissions showed similar profile to NO_x and CO. Furthermore, EC had significantly
393 positive correlation with NO_x and CO, demonstrating that they had common sources,
394 mainly from vehicular exhaust. However, The diurnal profiles of the secondary
395 species were similar to their precursors but obviously affected by O₃
396 concentrations other factors such as solar radiation, which could promote the oxidation
397 of the precursors, as these species were mainly produced by chemical processes. For
398 instance, there was a 2-hour delay for sulfate to reach its peak compared to SO₂ due to
399 the transformation. This pattern was also observed in Guangzhou (Hu et al., 2002b).
400 NO₃⁻ and SOC exhibited similar diurnal variation as their precursors had common
401 sources and they both formed from secondary photochemical oxidation. The daily
402 profiles of NO₃⁻, NH₄⁺ and SOC showed lower concentrations around 15:00 (local
403 time) probably due to the high boundary layer and/or low concentration of precursors.
404 Besides, for NO₃⁻ and NH₄⁺, high temperature, which enhanced the evaporative loss,
405 and low relative humidity may also responsible for the low level.

406 Fig. 4 also suggested that both gas phase compounds and aerosol components all
407 showed similar pattern of diurnal variation but had different magnitudes of
408 concentrations for different visibility levels. These components except for O₃ all
409 showed relatively higher concentrations under low visibility especially for the

410 ~~secondary inorganic species, indicating the important impact of the formation of~~
 411 ~~secondary components on the visibility reduction. The relatively low levels of O₃~~
 412 ~~under low visible conditions might because of the decreased photochemical~~
 413 ~~production and the chemical conversions of SO₂ and NO_x to sulfate and nitrate. It is~~
 414 ~~worth noting that relatively high humidity which favored for the formation of sulfate~~
 415 ~~and nitrate was observed under low visibility conditions. In addition, it seemed that~~
 416 ~~low visibility was associated with southwest wind. This might related to the~~
 417 ~~topography. There are mountains located on the southwest which is not conducive to~~
 418 ~~the diffusion of pollutants.~~

419

420 **3.2.23 Light extinction coefficient**

421 In order to ~~appoint~~ determine the contribution of PM_{2.5} constituents to the visibility
 422 degradation, light extinction (b_{ext}) was reconstructed based on the revised IMPROVE
 423 algorithm. In the present study, the impact of fine soil and coarse mass were not
 424 included because of the lack of metal elements and coarse matter concentrations. ~~Thus,~~
 425 ~~the revised IMPROVE algorithm was modified as following:~~

$$\begin{aligned}
 426 \quad b_{ext} &= b_{s,p} + b_{a,p} + b_{a,g} + b_{s,g} \\
 427 \quad &\approx 2.2 \times f_s(RH) \times [\text{Small } (NH_4)_2SO_4] + 4.8 \times f_L(RH) \times [\text{Large } (NH_4)_2SO_4] \\
 428 \quad &+ 2.4 \times f_s(RH) \times [\text{Small } NH_4NO_3] + 5.1 \times f_L(RH) \times [\text{Large } NH_4NO_3] \\
 429 \quad &+ 2.8 \times [\text{Small OM}] + 6.1 \times [\text{Large OM}] \\
 430 \quad &+ 1.7 \times f_{ss}(RH) \times [\text{Sea Salt}] + 10 \times [\text{EC Mass}] + 0.33 \times [\text{NO}_2 \text{ (ppb)}] + \text{Rayleigh Scattering}
 \end{aligned}$$

431 The estimated b_{ext} ~~in present study wasere~~ $664 \pm 288 \text{ Mm}^{-1}$; and was significantly
 432 correlated with PM_{2.5} concentrations ($r = 0.94$, $p < 0.001$), demonstrating the strong
 433 influence of fine aerosols on visibility degradation.

434 The reconstructed light extinction coefficient was compared with that derived
 435 from visibility and that calculated using a regression model developed by Chen.
 436 Extinction coefficient is inversely correlated with visibility according to the

437 Koschmieder equation ($Vis = K/b_{ext}$) (Seinfeld and Pandis, 2012). By using a K value
438 of 3.912, a comparison of the light extinction coefficients derived from different
439 methods. he calculated b_{ext} . Visibility is inversely correlated with the extinction
440 coefficient according to the Koschmieder equation ($Vis = K/b_{ext}$). By using a K value
441 of 3.912, we further calculated the visibility based on the reconstructed b_{ext} . The
442 estimated visibilities were $7.47371 \pm 4.12234 \text{ Mm}^{-1}\text{-km}$, ranged from 2.57 km to
443 23.41 km. Although this was much lower than the measured visibility, which were
444 $15.0 \pm 8.50 \text{ km}$, coefficients obtained from IMPROVE algorithm. Nevertheless, the
445 estimated and measured visibility they had similar temporary trend and were
446 significantly correlated with each other ($r = 0.71, p < 0.001$). Another method applied
447 here to estimate b_{ext} was a 2-factor parameterization regression model based on RH
448 and aerosol volume concentration (Chen et al., 2012). The volume concentration can
449 be acquired from the mass concentration divided by an average particle density of 1.7
450 g cm^{-3} (Wehner et al., 2008). A comparison of b_{ext} reconstructed by IMPROVE
451 algorithm and the regression model is presented in Fig. 4. Generally, a strong
452 correlation was evident with a correlation coefficient higher than 0.97 ($R^2 = 0.952$),
453 confirming that the reconstructed b_{ext} from the IMPROVE algorithm was reliable. The
454 majority of b_{ext} was clustered near the 1:1 line for $b_{ext} < 500 \text{ Mm}^{-1}$, with the
455 corresponding RHs mainly below 75%. However, for $b_{ext} > 500 \text{ Mm}^{-1}$, the dispersion
456 of b_{ext} gradually increased, and most of the corresponding RH was higher than 75%.
457 There are two possibilities causing these deviations. The first one is the ignorance of
458 the impact of particle size distribution on light extinction in the 2-factor
459 parameterization model applied here, as the variation of aerosol mass or volume
460 fractions of different size particles can influence b_{ext} value especially under high RH
461 (Chen et al., 2012). The second one is from the uncertainties of in situ measurements
462 since RH sensor may have large errors under high RH condition. The regression
463 model result was directly controlled by the RH value, and the hygroscopic growth
464 factor in the IMPROVE algorithm depends on RH.

465

466 The light extinction ~~was~~ were mostly influenced by aerosol light scattering as the
467 estimated $b_{s,p}$ ~~was~~ were $609 \pm 277 \text{ Mm}^{-1}$, accounting for 91% of ~~the~~ b_{ext} ~~(at least 75%)~~,
468 while $b_{a,p}$ and the extinction coefficient by gaseous were only $27.9 \pm 25.8 \text{ Mm}^{-1}$ and
469 $26.6 \pm 4.87 \text{ Mm}^{-1}$, respectively. The largest contributor to b_{ext} from the
470 reconstructed chemical species in fine particles ~~to b_{ext}~~ was organic matter (OM),
471 accounting for 40%, followed by $(\text{NH}_4)_2\text{SO}_4$, 34%, NH_4NO_3 , 16%, and EC , with their
472 shares of 34%, 16% and 4%, respectively. Fractions of these ~~However, the percentage~~
473 ~~contributors~~ varied greatly during over the study period, e.g. the contributions of
474 NH_4NO_3 ranged from only 3% to up to 40%. Generally, the contributions of
475 $(\text{NH}_4)_2\text{SO}_4$ and NH_4NO_3 were higher under low visibility period, increased from 30%
476 and 11%, respectively, during the ~~under~~ 20% best visibility periods to 39% and 19%,
477 respectively, during the ~~under~~ 20% worst visibility period, increased 1.3 and 1.7 times,
478 respectively. While ~~Correspondingly~~, the contributions of OM and EC ~~were~~ reduced
479 from 46% and 5%, respectively, under 20% best visibility period to 35% and 4%,
480 respectively, during the same periods ~~under 20% worst visibility period~~. These results
481 indicated the important role ~~of~~ sulfate and nitrate played on haze formation.

482 The percentages contribution to light extinction from individual ~~of the~~ aerosol
483 components ~~contribute to the light extinction were~~ also varied with ~~experienced~~
484 different variations in their fractions during different haze events and visibility
485 conditions. ~~We compared~~ ~~the~~ percentage contributions of individual ~~these~~
486 components during the best and worst ~~under~~ 20% best-visibility hours in each of the
487 three haze events are compared and shown in Fig. 5 ~~conditions to those under 20%~~
488 worst visibility conditions to investigate the controlling factor for the haze formation
489 (Fig. 5). During the first haze event (on Jan. 19), the contributions of NH_4NO_3
490 was ~~increased from only 8% under 20% best visibility to~~ and 24% during the ~~under~~
491 20% best and worst visibility hours, respectively, while the corresponding numbers
492 are ~~percentage of OM decreased from 48% and~~ to 37% for OM. ~~For~~ There were no
493 significant differences between the two visibility categories for the contributions of
494 $(\text{NH}_4)_2\text{SO}_4$ and EC , there was no significant change. During ~~For~~ the second haze

495 event (from Jan. 21 to 26), the fractions were 1.8, 1.5 and 1.3 times higher for
496 NH_4NO_3 , $(\text{NH}_4)_2\text{SO}_4$ and EC respectively but 1.2 times lower for OM under ~~worse~~
497 20% worst visibility condition than those under 20% best visibility condition during
498 this time. Overall, $(\text{NH}_4)_2\text{SO}_4$ made great contribution to the light extinction and
499 NH_4NO_3 had largest difference between 20% best and worst visibility conditions
500 during these two haze events. Therefore, secondary inorganic aerosols especially
501 NH_4NO_3 was likely the key component for the impaired visibility for these two haze
502 events. The elevated proportion of $(\text{NH}_4)_2\text{SO}_4$ and NH_4NO_3 during the heavy polluted
503 period was also observed in Beijing (Tao et al., 2015; Wang et al., 2015; Zheng et al.,
504 2015). A different trend of comparison between the best and worst visibility periods
505 was found in the third haze event (on Jan. 30) than in the first two. In the third event,
506 the percentage contributions of ~~Contrarily, during the third haze (on Jan. 30)~~
507 ~~increasing proportions of~~ OM and EC increased during the worst visibility period
508 compared to the best visibility period (from 40% to 49% and 6.8% to 11%,
509 respectively), while that of ~~accompanied with decreasing percentage of~~ $(\text{NH}_4)_2\text{SO}_4$
510 decreased (from 28% to 19%), ~~were found under worse visibility period, indicating~~
511 ~~that the e~~Carbonaceous components might be relatively played a more important role
512 for ~~the~~ visibility reduction in the third event. Therefore, there seems to be different
513 formation mechanisms for haze ~~eventsformation~~ in Suzhou.

514

515 3.3 Conversion from gas to particle phase

516 3.3.1 Variations of aerosol particles and precursors

517 Fig. 64 diagrammed the diurnal variations of meteorological parameters, various
518 aerosol components, the gaseous precursors, and some other important gaseous species
519 under three different visibility conditions: (i.e., (1) all data, (2) visibility ≤ 10 km, and
520 (3) visibility > 10 km). The daily variations of gas-phase compounds were different
521 between species and were mainly controlled by the direct surface emissions (such as
522 NO_x , SO_2 , and CO) or photochemical process (O_3). There were a distinct morning

523 peak and a less distinct afternoon peak, consistent with morning and afternoon rush
524 hours for NO_x and CO. This might be related to the heavy traffic emissions in the
525 rush hours and the strong elevation of the Planetary Boundary Layer heights at noon.
526 In contrast, there was only one mid-day peak for SO₂. This diurnal profiles were
527 similar to those observed in Guangzhou (Hu et al., 2002b) and Maryland (Antony
528 Chen et al., 2001). In the latter study, the dominant source of SO₂ was considered to
529 be the long range transport from the industrialized Midwest and with the deep
530 boundary layer around noon; SO₂ aloft mixed more effectively down to the surface
531 and thus caused the mid-day peak of SO₂. The reasons for the diurnal variation of SO₂
532 observed in the present study need further investigation. Similar to the diurnal
533 distribution of SO₂, O₃ also showed one distinct peak around noon due to the strong
534 photochemistry at that time (Quan et al., 2014).

535 For the aerosol components, EC which was also produced by the surface
536 emissions showed a profile similar-profile to NO_x and CO. Furthermore, EC had
537 significantly positive correlation with NO_x and CO, demonstrating that they had
538 common sources, mainly from vehicular exhaust. The diurnal profiles of the
539 secondary species were similar to their precursors but obviously affected by other
540 factors such as solar radiation, which could promote the oxidation of the precursors.
541 For instance, there was a 2-hour delay for sulfate to reach its peak compared to SO₂
542 due to the transformation processes. This pattern was also observed in Guangzhou
543 (Hu et al., 2002b). NO₃⁻ and SOC exhibited similar diurnal variation as their
544 precursors had common sources and they both formed from secondary photochemical
545 oxidation. The daily profiles of NO₃⁻, NH₄⁺ and SOC showed lower concentrations
546 around 15:00 (local time) probably due to the high boundary layer and/or low
547 concentration of precursors. Besides, for NO₃⁻ and NH₄⁺, high temperature, which
548 enhanced the evaporative loss, and low relative humidity may also be responsible for
549 the low levels.

550 Fig. 64 also suggested that both gas-phase compounds and aerosol components
551 all showed similar patterns of diurnal variations but had different magnitudes of

552 concentrations for different visibility levels. These components except for O₃ all
553 showed relatively higher concentrations under low visibility especially for the
554 secondary inorganic species, indicating the important impact of the formation of
555 secondary components on the visibility reduction. The relatively low levels of O₃
556 under low visible conditions might be due to ~~because of~~ the decreased photochemical
557 production and the chemical conversions of SO₂ and NO_x to sulfate and nitrate. It is
558 worth noting that the relatively high humidity which favored ~~for~~ the formation of
559 sulfate and nitrate was observed under low visibility conditions. In addition, it seemed
560 that low visibility was associated with southwest wind. This might related to the
561 topography. There are mountains located on the southwest which is not conducive to
562 the diffusion of pollutants.

563 In consideration of the distinct aerosol composition during the third haze, the
564 comparison of gaseous pollutants between the third and the first two haze episodes
565 were made. The concentrations of SO₂ and O₃ were comparable for these three haze
566 event. Contrarily, much higher levels of NO, CO and NO₂ were evident during the
567 third haze, in accordance with the high concentrations of POC and EC. These species
568 (NO, CO, NO₂, POC and EC) ~~were~~ had good correlations with each other.
569 Furthermore, they had similar diurnal variations and exhibited extremely high levels
570 in the morning rush hours on January 30 when the third haze occurred. These results
571 implied that there were common sources for these species, mainly from vehicle
572 exhaust emission. —

574 **3.3.2 Formation mechanisms of sulfate and nitrate**

575 As discussed earlier, the chemical formation of sulfate and nitrate from SO₂ and NO₂
576 respectively, should play important roles for visibility reduction, especially for the
577 first two haze events. The sulfur oxidation ratio, defined as $SOR = n\text{-SO}_4^{2-} / (n\text{-SO}_4^{2-} +$
578 $n\text{-SO}_2)$ and the nitrogen oxidation ratio, defined as $NOR = n\text{-NO}_3^- / (n\text{-NO}_3^- + n\text{-NO}_2)$
579 were used as indicators of the secondary transformation processes. The daily

580 variations of NOR showed similar patterns as those of NH₄⁺ and NO₃⁻. Likewise,
581 SOR had similar diurnal changes as SO₄²⁻. The values of SOR and NOR increased
582 more than 1.3 and 2.0 times, respectively, during the first two haze periods compared
583 to clear periods. Both SOR and NOR were higher with lower visibility (Fig. 6),
584 implying greater oxidation of gaseous species and more elevated secondary aerosols.
585 This was supported by the evidently higher concentrations of SO₄²⁻, NH₄⁺, and NO₃⁻
586 under worse visibility conditions in the first two haze events. Almost no elevating
587 levels of SOR, NOR or SIA were observed in the third haze, again confirming that the
588 SIA formation may not be the predominant factor controlling the occurrence of this
589 haze event.

590 The formation of SO₄²⁻ from SO₂ was mainly ascribed to the gas-phase oxidation
591 withby OH and H₂O₂ radical or heterogeneous oxidation (Wang et al., 2006; Zhao et
592 al., 2013). The gas-phase reaction is a strong function of temperature and
593 heterogeneous reactions always associated with high RH (Sun et al., 2006; Sun et al.,
594 2014; Sun et al., 2013a). However, weak correlations were found between SOR and
595 either temperature ($r = 0.174$, $p < 0.01$) or RH ($r = 0.150$, $p < 0.01$) in the present
596 study, indicating the complex formation mechanism of sulfate.

597 Many studies suggested that sulfate from aqueous SO₂ oxidation catalyzed by
598 transition metals was more significant during winter haze rather than gas-phase
599 oxidation (Li et al., 2011; Sun et al., 2013b; Zhao et al., 2013). Our measurement also
600 found that the heterogeneous oxidation was an important sulfate formation pathway in
601 this study area. As shown in Fig. 7, high concentrations of NO₂ accompanied with
602 ultralow level of O₃ less than 10 ppb and low amount of solar radiation were observed
603 during most of the time in the haze period, such as from 12 PM to 10 AM on 19
604 January and from 21 to 34 January. These results revealed the rather weak
605 photochemical activities during these time windows. The high levels of NO₂ and
606 weak photochemical activities could result in insufficient production of oxidants (OH
607 and H₂O₂ radicals) for gas-phase oxidation (Hua et al., 2008). Thus, other oxidation
608 reactions other than gas-phase oxidation likely explained the formation of abundant

609 secondary sulfates during the haze episode. The high RH (> 70%) during these haze
610 period was a beneficial factor for aqueous-phase oxidation of SO₂ to sulfate. Besides,
611 the calculation results using the Extended AIM Aerosol Thermodynamic Model
612 (E-AIM, Model II) (Clegg et al., 1998) (<http://www.aim.env.uea.ac.uk/aim/aim.php>)
613 showed a significant increase of the liquid water content when RH > 70%. These
614 results highlight the importance of aqueous-phase reaction to the secondary
615 transformation of SO₂.

616 The concentrations of O₃ were not extremely low during the whole haze period,
617 such as during daytime on January 23 and 25; the O₃ levels were as usual and had
618 obvious diurnal variations with one distinct peak around noon. RH was lower than
619 70% and sulfate was primarily in the solid phase at these times, indicating that
620 gas-phase oxidation was probably the dominant pathway for sulfate formation.
621 Aqueous-phase oxidation likely became predominant at night. This could explain the
622 high SOR in the afternoon and the sustained high level at night.

623 ~~The daily variation of NOR showed similar pattern as NH₄⁺ and NO₃⁻; likewise,~~
624 ~~SOR had similar diurnal change as SO₄²⁻, indicating the influence from NO_x or SO₂~~
625 ~~and O₃. Additionally, it is interesting to notice that under low visibility conditions~~
626 ~~during nighttime when O₃ concentrations were extremely low, there was still a rapid~~
627 ~~chemical conversion from gaseous to particle phase especially for sulfate particles.~~
628 ~~This conversion might mainly be produced through the processes other than the~~
629 ~~photochemical activities such as heterogeneous reactions in the aqueous surface layer~~
630 ~~of pre-existing particles or in cloud processes. The higher humidity during that time~~
631 ~~and the significant positive correlations between SOR and humidity through the~~
632 ~~whole study period again seem to validate that the heterogeneous process likely~~
633 ~~dominate the sulfate formation. This was consistent with the finding by Hewitt that~~
634 ~~liquid phase chemical conversion process was responsible for the formation of sulfate~~
635 ~~partieles in winter (Hewitt, 2001).~~

636 ~~Compared to the formation of sulfate, the contributions of various conversion~~
637 ~~pathways to nitrate formation were less known (Pathak et al., 2009). In this section,~~

638 ~~we examined in detail the possible causes of nitrate in PM_{2.5}.~~ Nitrate formation is
639 mainly through gas-phase oxidation of NO₂ by OH during daylight and the
640 heterogeneous reactions of nitrate radical during nighttime (Seinfeld and Pandis,
641 2012). ~~Figure-Fig. 7-8~~ showed the nitrate-to-sulfate molar ratio ($[\text{NO}_3^-]/[\text{SO}_4^{2-}]$) as a
642 function of the ammonium-to-sulfate molar ratio ($[\text{NH}_4^+]/[\text{SO}_4^{2-}]$), which can provide
643 an insight into the formation pathway of the secondary species (Jansen et al., 2014;
644 Pathak et al., 2009; He et al., 2012).

645 The relative abundance of nitrate linearly increased with the increasing
646 ammonium-to-sulfate molar ratio. Fitting a linear regression line resulted in an
647 intercept of $[\text{NH}_4^+]/[\text{SO}_4^{2-}]$ -axis of 1.51, indicating that nitrate formation via
648 homogeneous reaction of HNO₃ with NH₃ became significant at $[\text{NH}_4^+]/[\text{SO}_4^{2-}] >$
649 1.51 (Pathak et al., 2009; Jansen et al., 2014; He et al., 2012) ~~(Pathak et al., 2004;~~
650 ~~Pathak and Chan, 2005)~~. Pathak et al. (2009) also reported an intercept value of 1.5
651 for ~~a variety of several~~ cities worldwide while Jansen et al. (2014) found a ~~little~~
652 ~~slightly~~ smaller intercept value of 1.38 for Hangzhou. The ammonium concentration
653 in excess of the amount at which nitrate formation became evident was defined as
654 excess ammonium ($[\text{NH}_4^+]_{\text{exc}} = ([\text{NH}_4^+]/[\text{SO}_4^{2-}] - 1.51) \times [\text{SO}_4^{2-}]$). The concentrations of
655 excess ammonium were always higher than 0 and linearly correlated with nitrate
656 concentration, as shown in ~~Fig. ure~~ 8. This indicated that the formation of nitrate was
657 strongly associated with ammonium formation. In other words, when the excess
658 ammonium was > 0 , the gas-phase homogeneous reaction between the ambient
659 ammonia and nitric acid was responsible for forming nitrate (Pathak et al., 2009;
660 Jansen et al., 2014). The slope of 0.37 for the regression and the scattering of the data
661 indicated that the excess ammonium was bound to species other than nitrate, such as
662 chloride, bisulfate, etc. The significance of gas-phase homogeneous reaction to nitrate
663 formation has been reported for many cities (Jansen et al., 2014; Pathak et al., 2009).
664 However, as mentioned above, in some cases during haze period, the conditions
665 (ultralow ozone concentrations, low solar radiation and high NO₂) were not favorable
666 for the gas-phase oxidation. Relatively high RH were often observed in those cases,

667 which may have favored the gas to particle partitioning of nitrate acid and ammonia
668 (Sun et al., 2011). Furthermore, the E-AIM calculation results manifested that the
669 nitrate partly or completely existed in the aqueous phase during those conditions.
670 Therefore, we assumed that heterogeneous chemistry, such as heterogeneous
671 hydrolysis of N₂O₅ (N₂O₅ + H₂O (aq) → 2HNO₃) or equilibrium partitioning (HNO₃
672 (g) + NH₃ (g) ↔ NH₄⁺ (aq) + NO₃⁻ (aq)) also contributed to the formation of nitrate
673 under high RH conditions. This was supported by the fact that in Fig. 8 more plots
674 were deviated from the regression line when RH were relatively high. The importance
675 of Besides, a model study also verified that aqueous-phase oxidation of NO_x was of
676 minor importance for nitrate production in the atmosphere (Pandis and Seinfeld,
677 1989). But contrarily, heterogeneous reactions for nitrate formation were considered
678 to be significantly important in the production of both sulfate and nitrate also reported
679 in other studies (Sun et al., 2011; Zheng et al., 2015) ~~(Sun et al., 2011).~~

680

681 **3.4 Regional sources deduced from trajectory and PSCF analyses**

682 The regional sources and transport of air pollutants exert a profound impact on local
683 air quality in YRD region as it is located in the typical monsoon region (Ding et al.,
684 2013). Therefore, trajectory clustering method was employed to examine the pathway
685 of air masses and to look into the chemical compositions and light extinction
686 coefficients among the air masses with different origination. The calculated 48-h
687 back trajectories were clustered into six clusters (Fig. 9 and Table 1), i.e. six air mass
688 transport pathways. As can be seen, Table 1 summarized the percentage of trajectories
689 for each trajectory cluster and the corresponding mean concentrations of PM_{2.5} and
690 major aerosol species. air masses reaching at Suzhou mainly came from local areas,
691 the nearby provinces (cluster 2, accounting 31.7%), and the northwestern areas
692 (cluster 1 and 3, both accounted for 20.6%).

693 High aerosol concentrations were associated with these trajectories. This was
694 reasonable considering that these air masses passed over some highly industrialized

695 cities, such as Hefei, Nanjing, and Hangzhou (Fig. 1). ~~The trajectory clusters were~~
696 ~~dominated by cluster 2, accounting for 31.7%, followed by cluster 1 and 3, both~~
697 ~~accounted for 20.6%, and the rest clusters, cluster 4, 5 and 6 accounted for 12.7%,~~
698 ~~6.3% and 7.9%, respectively. Obviously, air masses mainly originated from the north~~
699 ~~and southwest (cluster 1 to 4).~~

700 The mean concentrations of PM_{2.5} and main aerosol species varied with certain
701 types of air masses. The highest mean concentration of PM_{2.5} (167 $\mu\text{g m}^{-3}$) was
702 associated with C3 cluster with the air masses originating from northwest and passed
703 over Anhui and Jiangsu provinces. The air from the C1, C2 and C4 areas were also
704 related to relatively high PM_{2.5} concentrations, which were 129 $\mu\text{g m}^{-3}$, 117 $\mu\text{g m}^{-3}$
705 and 103 $\mu\text{g m}^{-3}$ respectively. These air masses all passed over some highly
706 industrialized cities, such as Qingdao, Nanjing, and Hangzhou (Fig. 1). Relatively low
707 concentrations levels of PM_{2.5} were associated with related to clusters C5 (84 $\mu\text{g m}^{-3}$)
708 and C6 (87 $\mu\text{g m}^{-3}$). The trajectories of C5 and C6 mostly originated from southeast
709 ~~and northeast respectively.~~ Based on the pathways and origins, these air masses were
710 expected to bring in relatively clean air from the East China Sea and the Yellow Sea
711 and consequently reduced the aerosol pollution in Suzhou.

712 Aerosols had high fractions of secondary inorganic ions (39%-42%) and
713 relatively low contributions of OC (15%-16%) when the air masses fell in the C1, C3
714 and C4, while had relatively high percentage of OC (36%) when the air masses fell
715 into the C5. These differences of aerosol concentrations and composition in different
716 clusters may result in distinct light extinction coefficients and species contributions to
717 visibility reduction when air masses originated from different directions (Fig. 9).
718 Similar to PM_{2.5}, the reconstructed b_{ext} was the highest when air masses originated
719 from the Northwest area (C3) and was relatively low when air masses fell into C5 and
720 C6 areas. However, the lowest value of b_{ext} was in the C6 instead of C5 for the lowest
721 PM_{2.5} level, because of the higher contribution of OM in the C5. For the major
722 components in PM_{2.5}, the highest averaged concentrations were associated with C3
723 cluster for sulfate, nitrate and ammonium but with C5 cluster for both OC and EC.

724 Accordingly, the composition profiles of these major species were different between
725 clusters. C1, C2, C3 and C4 trajectory clusters had similar composition with relatively
726 higher fractions of secondary inorganic ions (secondary inorganic ions > 35%, OC <
727 19%) compared to C5 and C6 clusters which both had relatively high contribution of
728 OC (secondary inorganic ions < 30%, OC > 23%).

729 These differences of PM_{2.5} concentrations and compositions in different clusters
730 may result in distinct light extinction coefficients and species contributions to
731 visibility reduction when air mass originated from different directions (Fig. 9).
732 Similar to PM_{2.5}, b_{ext} values were higher when the air masses originated from the
733 Northwest area (C3), whereas b_{ext} were relatively low when air masses fell into C5
734 and C6 areas. AS and OM were the dominant species determining the light
735 extinction for all clusters. However, differently, AS was the predominant contributor
736 to light extinction for trajectories from north and northwest (C1, C3, C4), while in
737 other cases the light extinction was primarily affected by OM. AN was the third
738 highest contributor in all trajectory clusters with the largest contribution when air
739 masses originated from northwest.

740 The origins of air masses in different haze events were further analyzed to
741 interpret the relative contributions of chemical species to visibility reduction that
742 differed between haze events. Most air masses fell into C1 and C3 (air masses from
743 north and northwest, respectively) in the first two haze occurrence while all air mass
744 trajectories were in C2 (air masses from south local and nearby areas) for the third
745 haze event. The contribution of OM to the total light extinction was higher in the third
746 haze event than in the first two as discussed early, consistent with the results for
747 cluster analysis that the light extinction was primarily impacted by AS for C1 and C3
748 but by OM for C2. These results manifested that the third haze event was mainly
749 contributed by the primary emission of carbon species from the local and/or
750 surrounding areas.

751 the air mass originated from southwest and passed over Zhejiang province might
752 play a key role in the high contribution of OM in the third haze event.

753 It should be noted that air mass back trajectory analysis only suggests the
754 originations and pathways of air masses but does not directly reveal the exact sources.
755 Based on the results of trajectory analysis, the PSCF method was applied to explore
756 the likely regional sources of major components in PM_{2.5}, including sulfate, nitrate,
757 OC, and EC, as illustrated in Fig. 10. Generally, PM_{2.5} and the five aerosol species in
758 Suzhou were mainly affected by local sources and nearby cities. Specifically, the
759 higher value for PM_{2.5} and the aerosol components were all localized in northwest to
760 the south, covering surrounding cities in Jiangsu and near the border of Anhui and
761 Zhejiang provinces. Additionally, these species were all affected by pollutions from
762 Anhui province. Sulfate, nitrate and ammonium had similar spatial distributions, and
763 relatively more affected by the north and northwest cities in Shandong, Jiangsu and
764 Anhui provinces while pollutions from south cities in Zhejiang province had more
765 impact on OC and EC in studied area than sulfate, nitrate and ammonium.

767 4. Conclusions

768 Heavy aerosol pollution occurred in Suzhou in January 2013 with daily PM_{2.5}
769 concentrations on haze days 1.97 to 2.61 times higher than Grade II criteria of the
770 national ambient air quality standard (75 $\mu\text{g m}^{-3}$) and maximum value of 324 $\mu\text{g m}^{-3}$
771 on Jan. 14, 2013. Unfavorable weather conditions (high RH, low rainfall, wind speed
772 and atmospheric pressure) especially high RH together with increased air pollutants
773 produced from local and nearby sources were responsible might provide beneficial
774 conditions for these haze formation.

775 During the first two haze periods, WSH and carbonaceous species both increased
776 during the haze events and the major aerosol components were SO_4^{2-} , NO_3^- ,
777 NH_4^+ , and SOC, which were mainly from secondary sources. Furthermore, SOR and
778 NOR both increased under worst visibility conditions, revealing efficient severe gas to
779 particle conversion during winter in this region. Additionally, The conversion
780 mechanisms were further analyzed for sulfate and nitrate. Rapid chemical conversion
781 from gas to particle phase for sulfate particles under extremely low O_3 concentrations

782 ~~and significant correlations between SOR and humidity demonstrated that~~
783 ~~heterogeneous process might dominate the sulfate formation. However, the result of~~
784 ~~($[\text{NO}_3^-]/[\text{SO}_4^{2-}]$) as a function of ($[\text{NH}_4^+]/[\text{SO}_4^{2-}]$) showed that the formation of~~
785 ~~ammonium nitrate via the homogeneous gas-phase reaction was favored.~~

786 ~~The reconstructed light extinction coefficients based on the revised IMPROVE~~
787 ~~algorithm were $664 \pm 288 \text{ Mm}^{-1}$, and mainly contributed by OM (40%), $(\text{NH}_4)_2\text{SO}_4$~~
788 ~~(34%), NH_4NO_3 (16%), and EC (4%). The contributions of these species experienced~~
789 ~~different variations in their fractions under different visibility conditions. Generally,~~
790 ~~the contributions share of $(\text{NH}_4)_2\text{SO}_4$ and NH_4NO_3 to the reconstructed b_{ext} based on~~
791 ~~IMPROVE were higher under low visibility conditions while those the percentages of~~
792 ~~OM and EC were higher increased under high visibility conditions, indicating that~~
793 ~~secondary inorganic aerosols especially NH_4NO_3 seemed to be very important for the~~
794 ~~impaired visibility. Gas-phase homogeneous reaction might dominate the formation of~~
795 ~~sulfate and nitrate under low RH conditions while heterogeneous process might be~~
796 ~~responsible when RH were relatively high.~~

797 ~~But d~~Distinctively, ~~high proportion of carbon species from primary emission and~~
798 ~~lower fraction of secondary formation components were observed in the third haze.~~
799 ~~The SOR and NOR during the third haze episode were comparable to clean days.~~
800 ~~Moreover,~~ increasing proportions of OM and EC accompanied with decreasing
801 percentage of $(\text{NH}_4)_2\text{SO}_4$ were found under worst visibility conditions when the third
802 haze occurred. ~~These results,~~ suggested suggesting that the carbon components from
803 the primary emission might be relatively important for the visibility reduction for this
804 haze event.

805 ~~These differences in different haze events might be greatly affected by the~~
806 ~~pathways of air masses according to trajectory clustering analysis.~~

807 ~~The result of t~~Trajectory clustering analysis showed that the air quality in Suzhou
808 was mostly affected by air masses originating from North and Southwestern areas
809 which were associated with high aerosol concentrations. Distinct aerosol composition
810 profiles, light extinction coefficients and species contributions to visibility reduction

811 were observed when air masses originated from different directions, e.g. AS was the
812 predominant contributor to light extinction for trajectories from north and northwest,
813 while in other cases the light extinction was primarily affected by OM.

814 The likely sources of aerosol and the major species based on the PSCF method
815 were mainly from local anthropogenic activities and source emissions transported
816 from nearby cities. The northwestern to southern regions may be important sources of
817 aerosols and the major components. The northern and northwestern areas were
818 predominant source regions for sulfate, nitrate and ammonium aerosols, whereas the
819 southern area could be the common source region for carbonaceous species. This
820 information has the implications for the importance of collaborative air pollution
821 control strategy in the Yangtze River Delta Region.

822

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829

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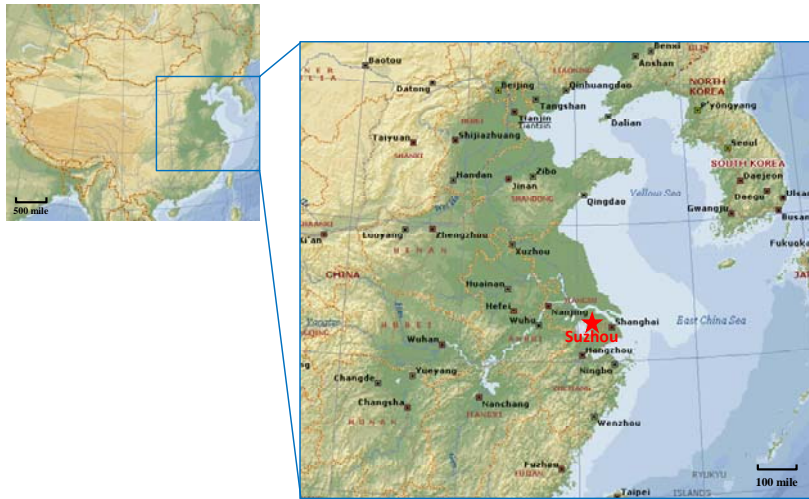
1077

1078 Table 1. The percentages of air masses from each trajectory cluster and associated~~the~~
 1079 mean b_{ext} (Mm^{-1}) and mean concentrations ($\mu\text{g m}^{-3}$) of $\text{PM}_{2.5}$ and its major chemical
 1080 components, and mean b_{ext} (Mm^{-1})~~selected aerosol species in the identified trajectory~~
 1081 clusters.

	1	2	3	4	5	6
Percent (%)	20.6	31.7	20.6	12.7	6.3	7.9
$\text{PM}_{2.5}$	129	117	167	103	84.0	87.4
OC	21.1	22.8	25.3	16.4	30.5	20.3
EC	2.05	3.87	2.21	1.45	3.90	1.86
Sulfate	22.9	18.2	32.4	23.1	8.85	8.39
Nitrate	11.5	10.0	16.9	7.84	3.26	7.91
Ammonium	15.2	12.3	18.9	12.4	7.93	9.88
b_{ext}	675	597	921	556	548	463

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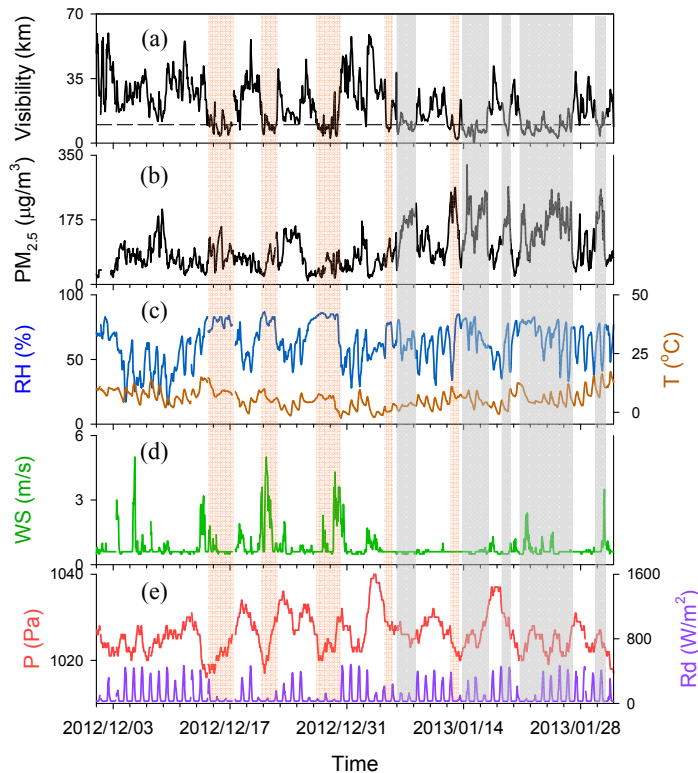
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1085 Figure 1. The sampling site in Suzhou. The locations of ~~some~~ major cities with a
 1086 population of more than 1 million (such as Qingdao, Nanjing and Hangzhou) are
 1087 marked with a square symbol. The topographical map was derived from Microsoft®
 1088 Encarta® 2009 © 1993 - 2008.

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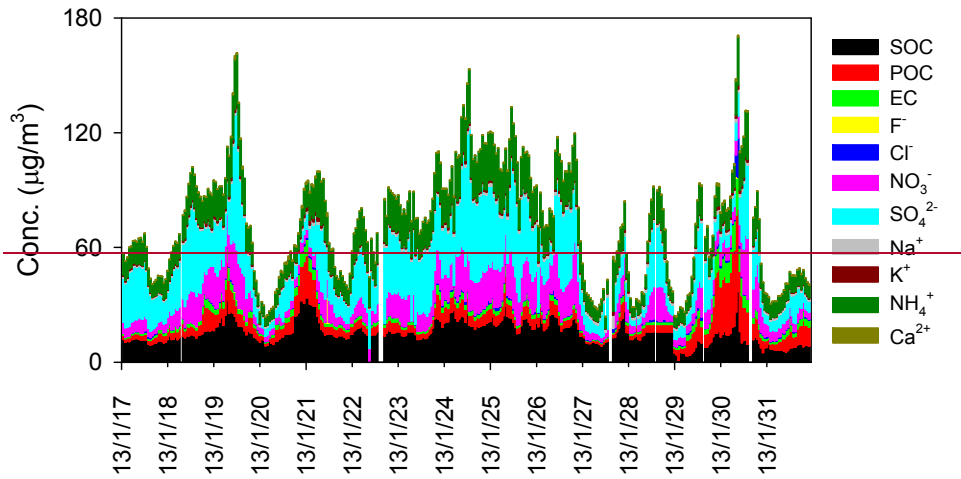


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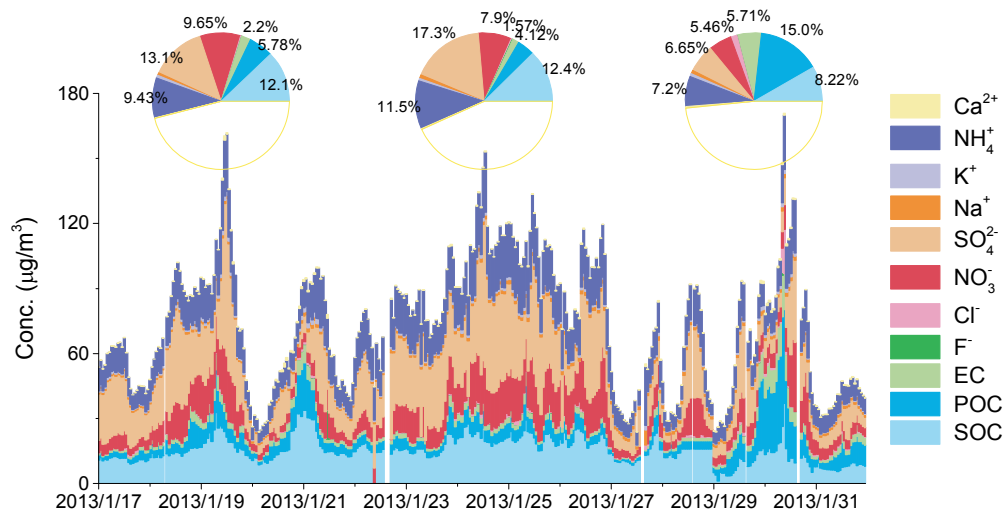
1091 Figure 2. Time series of (a) visibility; (b) $PM_{2.5}$ concentration; (c) relative humidity
 1092 (RH) and temperature (T); (d) wind speed (WS) and pressure (P); and (e) solar
 1093 radiation (Rd). The shaded areas in orange represent periods when visibility were

1094 lower than 10 km and accompanied by precipitation. The shaded areas in grey
 1095 represent haze periods.

1096



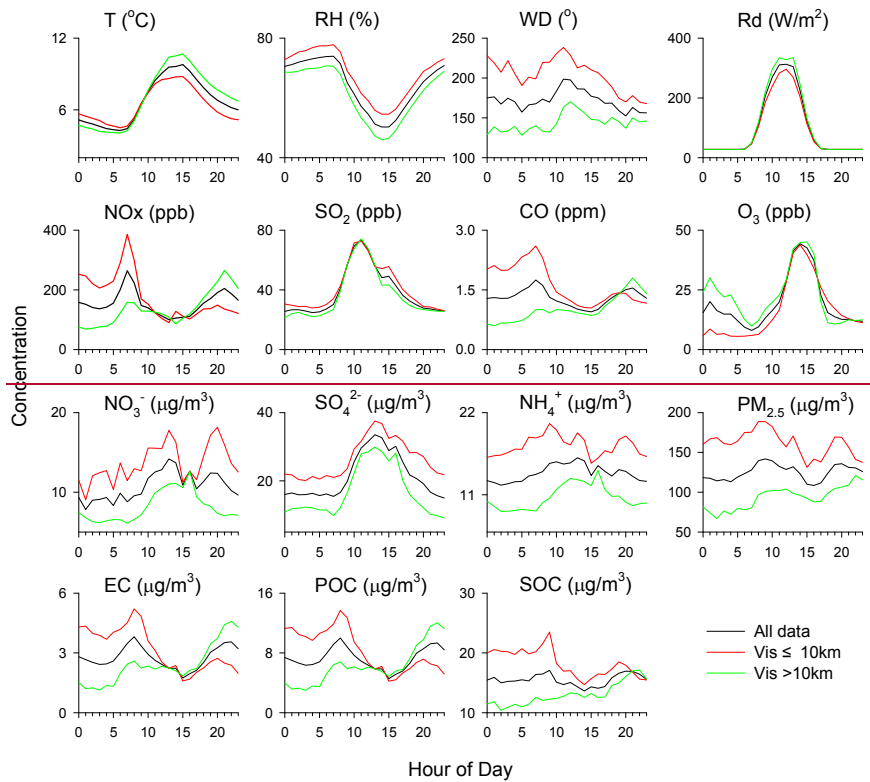
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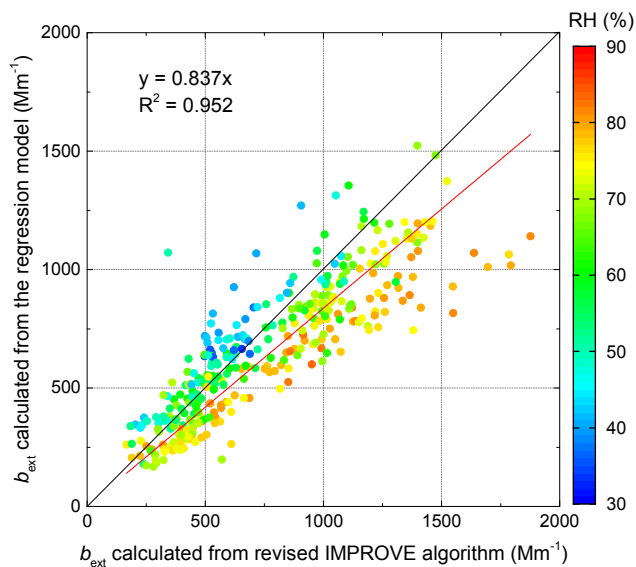
1099 Figure 3. Temporal distribution of water soluble inorganic ions and carbonaceous
 1100 species. The aerosol composition in the three haze events was also illustrated in the
 1101 pie chart.

1102



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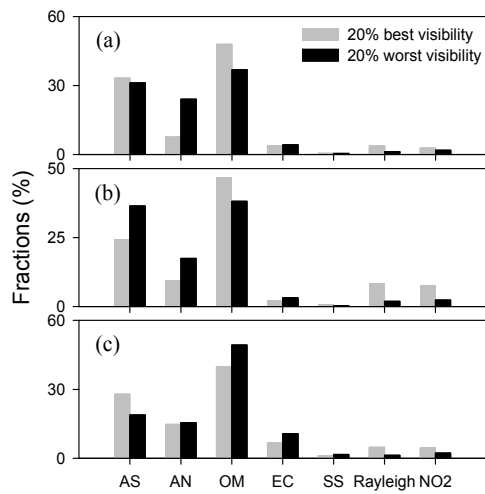
1104 Figure 4. Diurnal profiles of meteorological variables, aerosol precursors (NO_x , SO_2),
 1105 CO , O_3), $\text{PM}_{2.5}$, and major aerosol compounds (NO_3^- , SO_4^{2-} , NH_4^+ , EC , POC , SOC)
 1106 under different visibility conditions.



1107

1108 Figure 4. Comparison of the calculated b_{ext} between those obtained from the
 1109 regression equation and those from the IMPROVE algorithm.

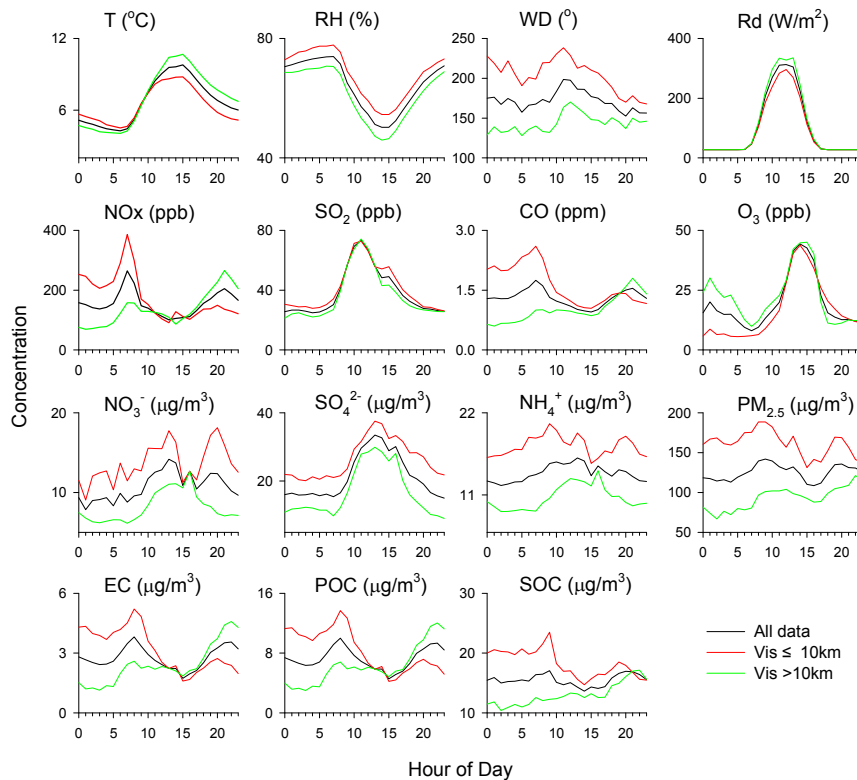
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1111

1112 Figure 5. Relative contributions of various chemical components in PM_{2.5}
 1113 (ammonium sulfate (AS), ammonium nitrate (AN), OM, and EC) to the total light
 1114 extinction under 20% best and 20% worst visibility conditions during the first (a),
 1115 second (b), and third (c) haze events.

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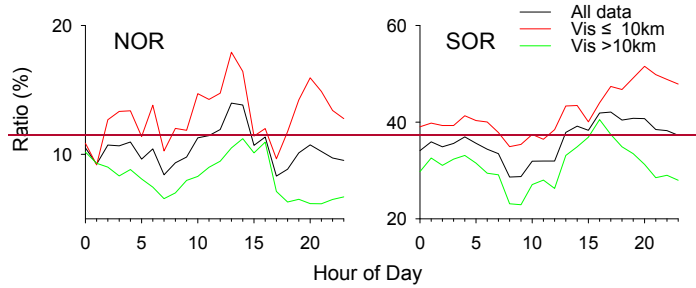


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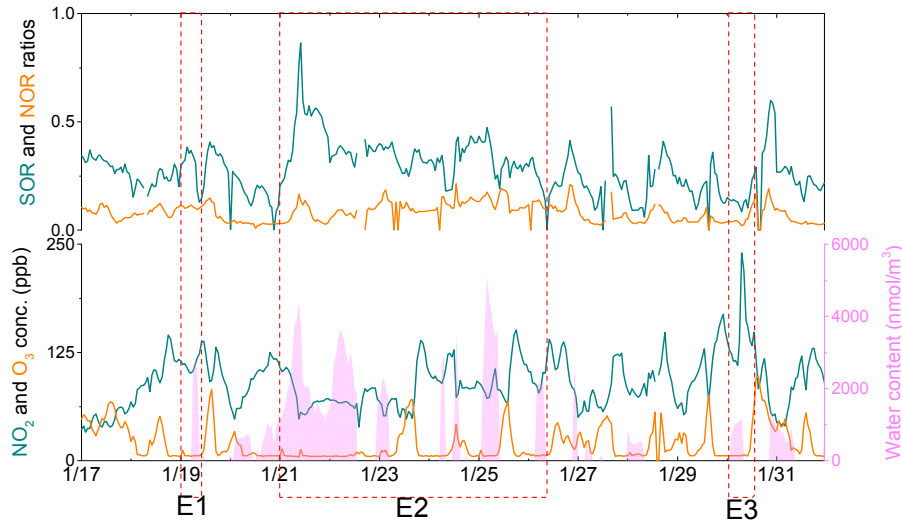
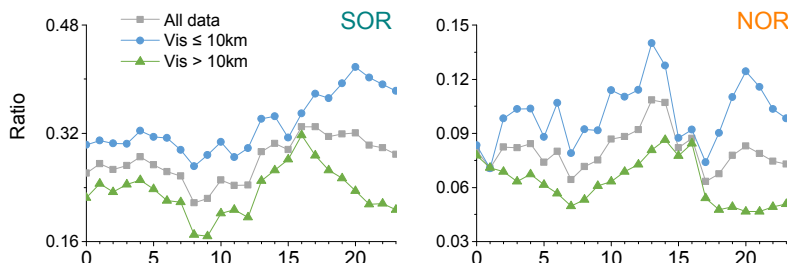
1118 Figure 6. Diurnal profiles of meteorological variables, aerosol precursors (NO_x, SO₂),

1119 CO, O₃, PM_{2.5}, and major aerosol compounds (NO₃⁻, SO₄²⁻, NH₄⁺, EC, POC, SOC)
 1120 under different visibility conditions.

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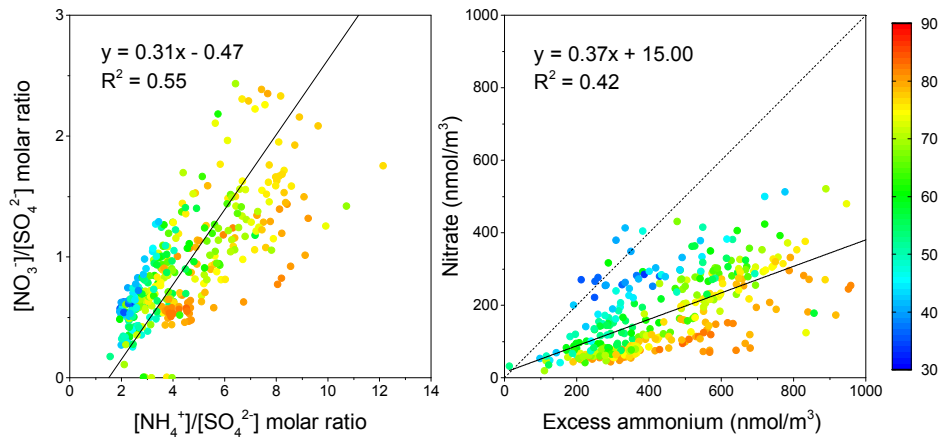


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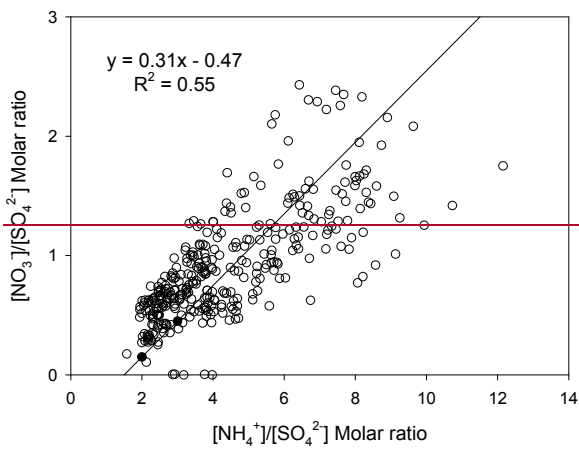
1124 Figure 76. Temporary distributions of SOR, NOR, NO₂, O₃ and aerosol water content.

1125 Diurnal profiles of NOR and SOR ratios under different visibility conditions.

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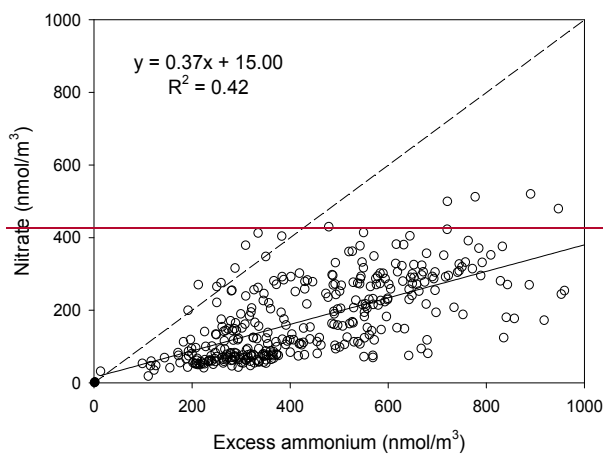
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1129 Figure 78. Nitrate to sulfate molar ratio as a function of ammonium to sulfate molar
 1130 ratio- (left) and relationship

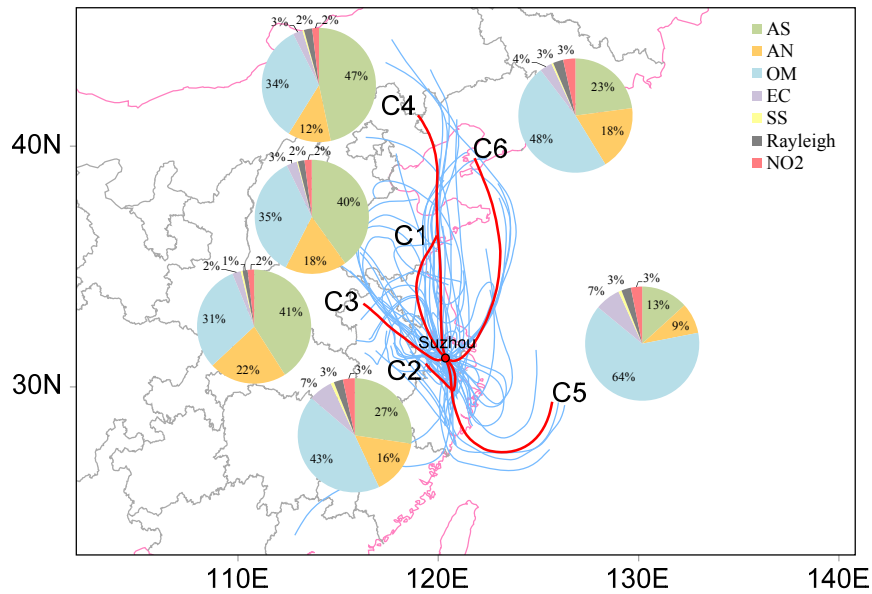
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1133 Figure 8. Relationship between molar concentrations of nitrate and excess ammonium
 1134 (right).

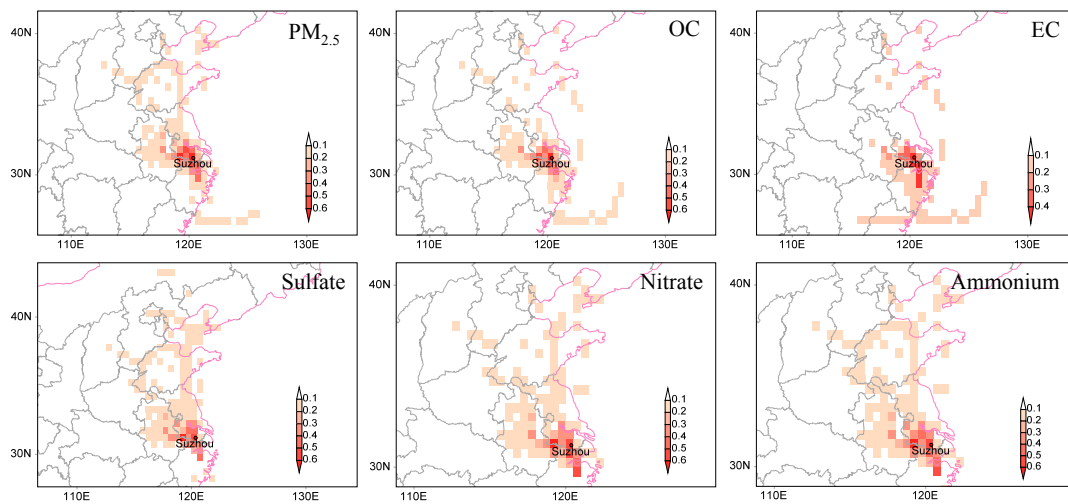
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1137 Figure 9. Backward air mass trajectories and six mean trajectories after the cluster
 1138 analysis at the sampling site during Jan. 17 to 31. Relative contributions of various
 1139 chemical components to the total light extinction in different clusters are illustrated.

1140



1141

1142 Figure 10. The PSCF maps for PM_{2.5}, OC, EC, sulfate, nitrate, and ammonium.

1143