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Atmospheric speciated mercury concentrations on an island between China and Korea: sources and transport pathways

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Abstract

As a global pollutant, mercury (Hg) is of particular concern in East Asia where anthropogenic emissions are the largest. In this study, speciated Hg concentrations were measured in the western most island in Korea, located between China and the Korean

- ⁵ mainland to identify the importance of local, regional and distant Hg sources. Various tools including correlations with other pollutants, conditional probability function, and back-trajectory based analysis consistently indicated that Korean sources were important for gaseous oxidized mercury (GOM) whereas, for total gaseous mercury (TGM) and particulate bound mercury (PBM), long-range and regional transport were also
- ¹⁰ important. A trajectory cluster based approach considering both Hg concentration and the fraction of time each cluster was impacting the site was developed to quantify the effect of Korean sources and out-of-Korean source. This analysis suggests that Korean sources contributed approximately 55% of the GOM and PBM while there were approximately equal contributions from Korean and out-of-Korean sources for the TGM
- ¹⁵ measured at the site. The ratio of GOM / PBM decreased when the site was impacted by long-range transport, suggesting that this ratio may be a useful tool for identifying the relative significance of local sources vs. long-range transport. The secondary formation of PBM through gas-particle partitioning with GOM was found to be important at low temperatures and high relative humidity.

20 **1** Introduction

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Mercury (Hg) is the only metal that exists as a liquid at standard conditions (US EPA, 1997) which results in it having a significant vapor pressure and presence in the atmosphere. In the atmosphere, Hg generally does not constitute a direct public health risk at the level of exposure usually found (Driscoll et al., 2007). However, once Hg is deposited into aquatic systems, it can be transformed into methyl-mercury (MeHg)



which is very toxic and readily bioaccumulates through the food web (Mason et al.,

1995). Many studies show that one of the major sources of MeHg in aquatic system is atmospheric deposition of inorganic Hg (Landis and Keeler, 2002; Mason et al., 1997).

Atmospheric mercury exists in three major inorganic forms, including gaseous elemental mercury (GEM, Hg⁰), gaseous oxidized mercury (GOM, Hg²⁺) and particulate

- ⁵ bound mercury (PBM, Hg(p)). The sum of the GEM and GOM is often called as total gaseous mercury (TGM). Due to its high water solubility and deposition velocity, GOM has short atmospheric residence times (~ day) and, consequently, its ambient concentration is mainly affected by local sources. In contrast, GEM, which comprises more than 95% of the total Hg in ambient air, can be transported long distances because it is
- relatively inert and has low water solubility and deposition velocity (Lin and Pehkonen, 1999). The residence time of PBM is dependent on the size of associated particles, but generally, it has been assumed to be a few days (Fang et al., 2012; Zhang et al., 2001). Measurements of GOM and PBM are challenging and uncertain due to their extremely low concentrations and complex chemical reactivity, and because their chemical forms
- ¹⁵ are not actually known (Pirrone et al., 2013). In most studies, GOM and PBM have operational definitions for the mercury species collected by a KCI coated denuder and by a quartz filter downstream of a KCI denuder, respectively.

In the atmosphere, Hg species can be interconverted through various redox reactions. It is known that GOM can be produced by homogeneous and heterogeneous re-

- ²⁰ actions of GEM with O₃, OH, and Br/BrO (Hedgecock and Pirrone, 2004; Obrist et al., 2011), but there is no consensus on which oxidants are most important. GEM can also be formed through reduction of GOM predominantly in cloud water (Subir et al., 2011, 2012). GOM can also be converted to PBM through gas-particle partitioning, with the partition coefficient, K_p , inversely correlated with temperature and positively correlated
- ²⁵ with particle surface area (Lyman and Keeler, 2005; Liu et al., 2010). Since GEM makes up the bulk of the total Hg in ambient air its formation through reduction processes of divalent Hg may not be important. However, the secondary formations of GOM and PBM through the oxidation of Hg⁰ followed by gas-particle partitioning can contribute significantly to their ambient concentrations.



The region of largest anthropogenic Hg emissions is East and Southeast Asia, contributing 39.7% of the total anthropogenic emissions (UNEP, 2013). In Korea, atmospheric Hg emissions have generally decreased since 1990. However, Hg levels in Korea are likely to be highly susceptible to Chinese emissions because China alone
accounts for about one third of the global total (UNEP, 2013) and Korea is situated just west (and generally downwind) of China. According to the recent studies, Hg concentrations in blood of Koreans are more than 4–8 times higher than those found in US and Germany, and approximately 26% of Koreans have higher blood mercury concentrations than a USA guideline level (http://envhealth.nier.go.kr), indicating that there is an urgent need to identify the Hg sources and pathways controlling Hg concentrations in Korea.

This study was designed to identify the contribution of various Hg sources including direct emissions from anthropogenic and natural sources and indirect secondary formation processes to atmospheric Hg concentrations in Korea. In order to achieve these

- objectives, Hg concentrations were measured in the western most island in Korea, located in between eastern China and the Korean mainland, so that, depending on wind patterns, the effects of Chinese and Korean Hg emissions could be evaluated. Previously, our group qualitatively evaluated the impact of local Korean sources and regional Chinese sources on TGM concentrations at the same sampling site (Lee et al., 2014).
- However, that work was unable to identify the effect of sources on Hg levels in Korea because only TGM was measured whereas all three Hg species are needed since they have very different physical and chemical properties. In this study, the importance of sources and pathways were both qualitatively and quantitatively evaluated using all three Hg species' concentrations measured throughout the extended sampling period.



2 Materials and methods

2.1 Site description

TGM, GOM and PBM were measured on the roof of a three-story building on Yongheung Island (YI), the westernmost island in Korea (Fig. 1). YI is a small island located about 15–20 km west from mainland Korea with a population of 5815. The Yongheung Coal-fired Power Plant (YCPP), located approximately 4.5 km southwest of the sampling site (Fig. 1c), emits about 0.11 tyr⁻¹ of Hg. To the east of the sampling site, industrial (the Incheon industrial complex shown as light violet color in Fig. 1b and c) and metropolitan (Seoul shown as a pink color in Fig. 1b and c) areas are located in mainland Korea, and, in the southern direction, there are three large coal fired power plants (Fig. 1b). The Hg emission rate of anthropogenic sources in Korea was estimated to be 8.04 tyr⁻¹ in 2010, with cement production being the largest source type (AMAP/UNEP, 2013).

2.2 Sampling and analysis

- ¹⁵ From January 2013 to August 2014, three atmospheric mercury species: TGM (GEM + GOM), GOM, and PBM (≤ 2.5 μm) were measured during eight intensive sampling periods (Table 1). TGM concentrations were measured every 5 min using a mercury vapor analyzer (Tekran 2537B). This instrument contains two gold cartridges which collect and thermally desorb Hg alternately. Desorbed Hg is quantifed using
 ²⁰ a cold vapor atomic fluorescence spectrometry (CVAFS). Outdoor air at a flow rate of 1.0 L min⁻¹ was transported through a 3 m-long heated sampling line (1/4″ OD Teflon) into the analyzer. The Tekran 2537B was automatically calibrated daily using an internal permeation source. Manual injections were also used to evaluate these automated cal-
- ibrations before each sampling campaign using a saturated mercury vapor standard. The relative percent difference between manual injection and automated calibration was < 2 %. Five-point manual calibration was also performed by injecting Hg vapor



into the sampling line two times during the study period. The R^2 ranged from 0.9991 to 0.9997 between mass injected and Tekran reported area, and the average relative percent difference between the mass injected and the mass calculated was 5.5%. The method detection limit (0.04 ng m⁻³) was calculated as three times the standard deviation obtained after injecting 1 pg of the mercury vapor seven times. The recovery rate (96 ± 3%) was obtained by directly injecting Hg vapor into the sampling line between

the sample inlet and the Tekran 2537B in a zero-air stream.

GOM and PBM were collected manually using an annular denuder coated with KCI followed by a quartz filter, respectively, at a flow rate of 10 Lmin⁻¹. To identify any

- diurnal variations, all samples were separately collected during the daytime (07:00– 19:00 LT) and nighttime (19:00–07:00 LT) except during the 7th sampling period when they were measured every 2 h. The sampling system including an elutriator, an impactor, a KCI-coated denuder and a filter pack was housed in a custom-made sampling box maintained at 45 °C to prevent hydrolysis of KCI. After sampling, the denuder
- ¹⁵ and quartz filter were thermally desorbed using a tube furnace at 525 and 900 °C, respectively, to convert Hg^{2+} to Hg^{0} in a carrier gas of zero air. The heated air was then transported into a Tekran 2537B for quantification. Field blanks for GOM and PBM were collected once for each sampling period, and their average values were $0.23 \pm 0.12 \text{ pgm}^{-3}$ and $0.25 \pm 0.09 \text{ pgm}^{-3}$, respectively.
- ²⁰ The sampling methods used in this study are currently the most accepted methods for the measurement of atmospheric GOM and PBM, however there are many studies reporting that these methods are subject to interferences from ozone, water vapor and possibly other compounds (Lyman et al., 2010; Talbot et al., 2011; Jeff et al., 2014; Finley et al., 2013; Gustin et al., 2013; Huang et al., 2013; McClure et al., 2014) although
- recent side-by-side measurements with two Tekran systems showed good agreement and no impact from added ozone and increasing relative humidity (Edgerton, 2015). Therefore, it should be noted that the GOM and PBM measurements reported in this study may be somewhat biased even though, at present, it is not possible to quantify the magnitude of these uncertainties.



Meteorological data including temperature, wind speed, wind direction, relative humidity and solar radiation were also measured every 5 min at the sampling site using a meteorological tower (DAVIS Inc weather station, Vintage Pro2TM).

Hourly concentrations of SO₂, NO₂, CO, O₃ and PM₁₀ were obtained from the national air quality (NAQ) monitoring station (http://www.airkorea.or.kr/) located approximately 8 km east from the sampling site. These concentrations were compared with those measured at another national air quality monitoring station located approximately 24 km west of the Hg sampling site, and there were no statistical differences between sites (p value < 0.001), indicating that the spatial distribution of these pollutants was relatively uniform across the area.

2.3 Backward trajectory and cluster analysis

Three-day backward trajectories were calculated using the NOAA HYSPLIT 4.7 with GDAS (Global Data Assimilation System) meteorological data which supplies 3 h, global 1° latitude–longitude datasets of the pressure surface. Hourly 3 day back-trajectories were calculated for each hour of sampling, and the arrival heights of both 200 and 500 m were used to describe the local and the regional transport meteorolog-ical pattern, respectively.

The backward trajectories were clustered into groups with similar transport patterns using NOAA HYSPLIT 4.7. This method minimizes the intra-cluster differences among trajectories while maximizing the inter-cluster differences. The clustering of trajectories is based on the total spatial variance (TSV) method. A more detailed description of the clustering process can be obtained in Draxler et al. (2011, 2014).



2.4 Conditional Probability Function (CPF)

The conditional probability that a given concentration from given wind direction will exceed a predetermined threshold criterion, was calculated using the following equation.

$$\mathsf{CPF}_{\Delta\theta} = \frac{m_{\Delta\theta}}{n_{\Delta\theta}}$$

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⁵ where $m_{\Delta\theta}$ is the number of occurrences from wind sector $\Delta\theta$ where the concentration is higher than a criterion value, and $n_{\Delta\theta}$ is the total number of occurrence from this wind sector.

2.5 Potential Source Contribution Function (PSCF)

The PSCF model counts each trajectory segment endpoint that terminates within given grid cell. A high PSCF value signifies a potential source location. The PSCF value was calculated as:

PSCF value =
$$\frac{P[B_{ij}]}{P[A_{ij}]} = \frac{m_{ij}}{n_{ij}}$$

where m_{ij} is the number of endpoints associated with a concentration higher than a criterion value in *ij*th cell, and n_{ij} is total number of endpoints in *ij*th cell. The criterion ¹⁵ value was the top 25% concentration and the cell size of 0.5° by 0.5° was used for tracing sources. To reduce the uncertainty in a grid cell with a small number of endpoints, an arbitrary weight function W_{ij} was applied when the number of the endpoints in a particular cell was less than three times the average number of endpoints (N_{ave}) for all cells (Fu et al., 2011; Han et al., 2007; Polissar et al., 2001a, b).

$${}_{20} \quad W_{ij} = \begin{pmatrix} 1.0 & N_{ij} > 3N_{ave} \\ 0.70 & 3N_{ave} > N_{ij} > 1.5N_{ave} \\ 0.40 & 1.5N_{ave} > N_{ij} > N_{ave} \\ 0.20 & N_{ave} > N_{ij} \end{pmatrix}$$

$$32938$$



(1)

(2)

(3)

3 Results and discussion

3.1 General trends of three Hg species

The average TGM, GOM, and PBM concentrations were $2.8 \pm 1.1 \text{ ngm}^{-3}$, $9.8 \pm 9.9 \text{ pgm}^{-3}$, and $10.6 \pm 12.0 \text{ pgm}^{-3}$, respectively (Table 1). Since the GOM concentration was much lower than TGM the reported TGM concentration can be considered a good approximation of the GEM concentration. TGM varied from 0.1 to 18.8 ngm^{-3} ; the highest concentration was observed around 2 a.m. on 18 March 2014 (Fig. 2). GOM and PBM concentrations peaked at 50.9 pgm^{-3} during the daytime on 19 March 2014 and 56.5 pgm^{-3} in the early morning (06:00-08:00 LT) of 28 May 2014, respectively (Fig. 2). The various Hg species did not follow similar concentration patterns although PBM was statistically significantly correlated with TGM (Pearson correlation coefficient, r = 0.235, p value = 0.03).

The data were grouped into four seasons including spring (March, April, May), summer (June, July, August), fall (September, October, November) and winter (Decem-¹⁵ ber, January, February). Both TGM (ANOVA/Tukey test, *p* value < 0.001) and PBM (*p* value = 0.024, Kruska–Wallis test) had the highest concentrations in winter, followed by spring and summer while there was no statistical difference in GOM concentrations among different seasons (*p* value = 0.288, Kruska–Wallis test) (Fig. 3). Observed TGM concentrations were substantially lower than those measured in a suburban and re-

²⁰ mote site in China and metropolitan areas of Korea (Seoul), but higher than at most North American sites and at a rural site of Korea (Chuncheon) (Table 2). GOM and PBM concentrations were in between those typically found at urban locations and at a rural site in Korea and were much lower than those typically measured in China.

The TGM concentration varied diurnally, generally showing morning maximums (07:00–12:00 LT) and minimums during the nighttime. In urban areas, TGM concentrations are typically higher during the nighttime due to a combination of decreased GEM loss by daytime oxidation, increased use of household heating systems and decreased mixing heights at night (Kim et al., 2012; Han et al., 2014). In contrast daytime



peaks have been observed in rural and remote areas, likely due to increased volatilized of Hg⁰ from natural sources (Choi et al., 2008; Cheng et al., 2014). Overall these results suggest that TGM concentrations at this site are elevated due to the proximity of regional sources and daily variations are controlled by natural emissions from the ocean and soil surfaces.

GOM concentrations were the highest in spring $(10.7 \pm 10.1 \text{ pg m}^{-3})$ and the lowest in summer $(6.2 \pm 4.9 \text{ pg m}^{-3})$ with statistically insignificant seasonal variation (Fig. 3). The lack of a GOM seasonal variation for could be an indicator of insignificant secondary formation through photochemical oxidation reactions, but it might be also be due to the small sample numbers and/or relatively long sampling duration (12 h). PBM concentrations did have statistically significant seasonal variations with the highest average concentration in winter (17.8 ± 16.7 pg m⁻³) and the lowest average concentration in summer (5.8 ± 4.1 pg m⁻³) (Fig. 3). Higher PBM concentrations in winter were likely caused by increased biomass burning and residential heating, decreased removal from

the atmosphere due to the lower precipitation depth, and/or lower temperatures which favor partitioning to the aerosol phase. Previous studies also often observed the highest PBM concentrations in winter (Mao et al., 2012; Amos et al., 2012; Lan et al., 2012).

In Korea emissions of PBM from anthropogenic sources are much smaller then gaseous emissions (the proportion of GEM, GOM, and PBM released are 64.4, 28.8,

²⁰ and 6.8%) (Kim et al., 2010). The fact that PBM concentrations are similar to GOM even though significantly less PBM is released suggests that a significant portion of atmospheric PBM may be due to secondarily formation through gas-particle partitioning. This process is characterized by a partition coefficient, K_p , which is inversely correlated with temperature (Rutter and Schauer, 2007), possibly causing the distinct seasonal variation in PBM concentrations at the sampling site.

The relationship between K_p , defined as:

 $K_{\rm p} = \frac{\rm PBM/PM}{\rm Hg_{\rm gas}}$



(4)

where PM represents the particle mass, and Hg_{gas} is the concentration of gaseous Hg and relative humidity (RH) was examined. RH was included since in recent studies K_p was found to increase at high relative humidity in colder seasons (Lyman and Keeler, 2005; Liu et al., 2010). Note that the sampling site was located in a coastal area with generally high RH.

Some previous studies suggested that all gaseous mercury species including Hg^0 may deposit on particles (Xiu et al., 2005, 2009); however, others suggested that the gas-particle partitioning of GOM occurred but assumed that the adsorption of Hg^0 on particles was negligible due to its high vapor pressure (Amos et al., 2012; Rutter and Schauer, 2007). Consistent with this hypothesis, we found a statistically significant multiple linear relationship between the ratio of PBM / GOM with temperature and relative humidity (Fig. 4):

 $\log(PBM/GOM) = -538 - 0.029(T) + 0.014(RH)$

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where *T* and RH indicate atmospheric temperature (°C) and relative humidity (%), re-¹⁵ spectively. The multiple linear equation fit the data well (*R* = 0.49, *p* value < 0.001), and both variables of temperature and relative humidity were statistically significant (*p* value < 0.001). When the temperature was used as a single independent variable the log(PBM / GOM) regression equation was still significant with a Pearson correlation coefficient of -0.37 (*p* value < 0.001), somewhat lower than that from the multiple re-²⁰ gression. However, the relative humidity as a sole independent variable was not related with the ratio of PBM / GOM indicating that relative humidity affected the gas-particle partitioning only in conjunction with temperature. Han et al. (2014) also found a signifi-

- cant multiple linear relationship between the ratio of PBM / GOM with temperature and relative humidity at a rural site ($R^2 = 0.613$, beta for T = -0.774, beta for RH = 0.33)
- ²⁵ but not at an urban site. The lower correlation coefficient and the beta values found in this study compared to those from Han et al. (2014) is probably due to the greater impact of primary anthropogenic sources around the sampling site.



(5)

3.2 Tracing sources of Hg species

Correlations between Hg and other pollutant concentrations are often used to identify sources. For example good correlations with SO₂ and CO typically indicate the impact of coal combustion (Pirrone et al., 1996; Han et al., 2014), and a strong correlation ⁵ between Hg and CO has often been used as an indicator for long-range transport because both pollutants have similar sources and do not easily decompose by reaction and deposition during transport (Weiss-Penzias et al., 2003, 2006; Kim et al., 2009). A good correlation between Hg and NO₂ suggests the site is being impacted by local sources because the lifetime of NO₂ is relatively short compared with that of CO (Se-¹⁰ infeld and Pandis, 2006). In this study TGM concentrations were well correlated with SO₂, CO, and PM₁₀ concentrations but not with NO₂ concentrations (Table 3), indicating that long-range transport of TGM emitted from coal combustion was impacting the site throughout much of the sampling period. PBM concentrations also had a statistically significant relationship with TGM and CO suggesting long range transport is also important for PBM, but GOM was not correlated with any other pollutant suggesting it

Important for PBM, but GOM was not correlated with any other pollutant suggesting it is impacted to a greater extend by local sources (see additional discussion below).

CPF plots shows that the top 25 % TGM concentrations were associated with winds from the NNW and eastern direction, pointing towards northeastern China and inland Korean sources; however, when the criterion was set to the top 10 % regional transport

- from China became less important and the sources located in southern and eastern areas of Korea were identified as important source areas (Fig. 5). The CPF plot for GOM is significantly different from the one for PBM. High PBM concentrations were associated with northern winds while GOM concentrations were enhanced during southeastern winds.
- These results suggest that for PBM regional transport from Chinese and North Korea sources were more important than Korean sources; in contrast coal fired power plants located in the southern direction rather than regional transport impacted GOM concentration. It should be noted that this result is in apparent conflict with the finding that



there was no relationship between GOM and SO₂ concentrations. Total SO₂ emissions from power plants in China (18.6 Tgyr⁻¹) are much larger than in Korea (0.09 Tgyr⁻¹), and SO₂ emission rates per capita and per area in China also greatly surpass those in Korea (Lu et al., 2010). Much larger SO₂ emissions in China raise the background SO₂ concentration in the region and may mask any correlation between GOM and 5 SO₂ even if coal fired power plants located south of the sampling site impacted GOM concentrations. In support of this hypothesis the pollution rose indicates that high SO_2 concentrations are associated with westerly winds while high GOM concentrations are associated with southerly winds. It should be noted that only the top 25% of GOM and PBM concentrations were used as the criteria for the CPF plot because the numbers of samples for both species were significantly less than for TGM due to their longer

sampling duration (12 h).

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Among the eight sampling periods, the second period (April 2013) had the highest TGM, PBM and the second highest GOM average concentration, and SO₂, NO₂, CO, and PM₁₀ were also quite high (Table 1). During this period, TGM was statistically well correlated with SO₂ (r = 0.55), NO₂ (r = 0.56), and CO (r = 0.36), with the highest

Pearson correlation coefficient with NO₂, the characteristic local pollutant. In addition, the CPF plot (for TGM) and the back-trajectories were also associated with easterly winds transporting air masses from major Korean urban areas, supporting the previous

suggestion that inland sources enhanced all three Hg concentrations during the second 20 sampling period.

In contrast the fifth sampling period had the lowest GOM, PBM and the second lowest TGM concentrations (Table 1). Note however that the TGM concentrations for the first couple of days reached approximately 5 ngm^{-3} and gradually decreased to about 1 ng m⁻³ during the last days of sampling (Fig. 2), indicating that there was likely 25 two different sources affecting Hg concentrations during this period. Back-trajectories associated with high TGM concentrations passed through northeastern China, North Korea and the industrial/metropolitan areas of Korea before arriving at the sampling site whereas trajectories during the low TGM concentrations spent long periods within



the ocean boundary layer. Although Hg can be emitted from ocean surface (Han et al., 2007; UNEP, 2013) heavy rain and low solar radiation occurring during the last two days of this period probably inhibited emissions of Hg from the ocean surface.

3.2.1 GOM / PBM ratio

- ⁵ According to the CFP results, regional sources in China located NW and NE of the sampling site were responsible for the elevated PBM concentrations while inland Korean sources were important contributors to increased GOM concentrations (Fig. 5). The finding that long-range transport of TGM and PBM to the site is important is supported by their significant correlation with CO (Table 3). In order to identify the relative importance of local sources relative to long-range transport, the ratio of GOM / PBM was used as an indicator because the atmospheric residence time of GOM is widely regarded to be shorter than that of PBM (even though there is no consensus on what
- specific chemical forms are collected by KCI-coated denuders). The GOM / PBM ratio should be higher if local sources are more important, and the GOM / PBM ratio
- ¹⁵ is likely to decrease as long-range transport becomes more important. In this study the GOM / PBM ratios were categorized into three groups: low (0–2), middle (2–8), and high (> 8) and the frequency of wind direction was compared (Fig. 6). The result clearly indicates that the southerly and southeasterly winds were associated with high GOM / PBM ratios and that the westerly and northerly winds indicative of long-range
- ²⁰ transport from China prevailed at lower GOM / PBM ratios. There was a weak negative correlation between the ratio of GOM / PBM and CO concentration at a significance level of 0.1 (p value = 0.089), supporting the assertion that the GOM / PBM ratio decreased with the increased effect of long-range transport.

3.2.2 PSCF results

²⁵ In order to locate potential source areas in more detail, PSCF was used. For TGM, potential sources were located in Liaoning, Shandong, and Henan provinces of China



along with the southern area of Korea (Fig. 7). Liaoning province, where large nonferrous smelters are situated, is the province with the largest Hg emission inventory in China; Shandong and Henan provinces are also large Hg emission areas, emitting about 30–40 tyr⁻¹ (Fu et al., 2012) in part due to a large lead smelter (Wang et al., 2014), and biomass burning (Huang et al., 2011).

The probable source areas of PBM identified by PSCF were similar to those for TGM, indicating that both Chinese and inland Korean sources enhanced PBM concentrations, with the exception of metropolitan (Seoul) and industrial (Incheon) areas located in northwestern South Korea which emerged as a more prominent source areas for PBM than for TGM (Fig. 7). Only Korean sources including metropolitan (Seoul)

- and the industrial areas in southern Korea were identified as probable source areas for GOM (Fig. 7); long-range transport of GOM from China was not important. The Yellow Sea between China and Korea was also associated with high PSCF values; however, it is not certain whether this is a trailing effect derived by relatively short sampling
- ¹⁵ duration or a real source area. A trailing effect is often observed, especially with a limited number of measurements or short sampling period, since PSCF evenly distributes weight along the path of trajectories so that PSCF results often identify areas upwind and downwind of real sources as a source area (Han et al., 2007). However, it should be also noted that the marine boundary layer provides good conditions for active Hg
- oxidation reactions due to an abundance of oxidants (Auzmendi-Murua et al., 2014); therefore, the possibility of areas over the ocean being a GOM source should not be excluded.

3.2.3 Source attribution based on cluster analysis

In an effort to quantify the contribution of national and foreign sources to the measured Hg concentrations the back trajectories were grouped into five clusters using the trajectory cluster analysis feature of HYSPLIT. Among the five clusters, clusters 1 and 4 represent trajectories originating from outside Korea whereas the trajectories grouped in the cluster 5 originated and passed through the Korean peninsula (Fig. 8). Clusters



2 and 3 contain trajectories from China and the Korean peninsula, but cluster 2 was more associated with Liaoning province and North Korea while cluster 3 originated more from Shandong and Henan provinces. Clusters 1 through 5 contributed 12, 25, 24, 13, and 25% of the total time, respectively, and the associated concentrations with
⁵ each cluster are shown in Table 4.

The TGM concentration was the highest for cluster 4; however, GOM and PBM concentrations had the lowest averages for this cluster. Cluster 4 contains the back-trajectories originating from Mongolia and Russia and passing through northeastern China before arriving at the sampling site, which suggests long-range transport was important for this cluster (Fig. 8). Average CO concentrations were pretty similar for all clusters, but it was the second highest for cluster 4 (cluster 2 was highest). The highest total average GOM and PBM concentrations were associated with cluster 5 which includes trajectories distributed over the Korean peninsula, suggesting that Korean sources were responsible for the enhanced GOM and PBM concentrations. For

¹⁵ cluster 5, the highest Pearson correlation coefficient between GOM and PBM concentrations (r = 0.721) was observed, indicating that both Hg species were emitted from similar sources. For other clusters, there were no statistically significant correlations between GOM and PBM except for cluster 2 (r = 0.209, p value < 0.001). In addition, both average NO₂ concentration (19.4 ± 14.9 ppb) and the correlation coefficient between NO₂ and TGM (r = 0.688) were the highest for cluster 5, supporting the finding of impact from Korean sources.

In order to consider both Hg concentration and the fraction of time for each cluster, the following equation was used to quantify the effect of Korean and out-of- Korean sources to the Hg concentration at the receptor site.

source contribution of cluster, $i = \frac{\left(\frac{N_i}{N_{\text{total}}}\right) \times C_j}{\sum_{i=1}^n \left\{ \left(\frac{N_i}{N_{\text{total}}}\right) \times C_j \right\}}$

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(6)

where (N_i/N_{total}) indicates the percentage of time associated with the cluster, *i* is the number of trajectories, and C_j indicates the average Hg concentration associated with the cluster, *i*. Compared to the other clusters, the source contributions of clusters 1 and 4, which represent long-range transport, were relatively low for all Hg species (Table 4).

- ⁵ Cluster 5 contributed more significantly, especially for GOM and PBM, indicating the importance of Korean sources. The source contribution of cluster 2 was the highest for PBM compared to other Hg species, suggesting that North Korean sources were an important contributor to the high PBM concentrations measured, likely due to coal and biomass burning in North Korea (IEA, http://www.iea.org/stats/countryresults.asp.
 10 COUNTRY CODE=KP&Submit=Submit:2012; NI, 2003).
 - In order to quantify the contribution of Korean vs. out-of-Korean sources, the source contributions of the clusters were used. Clusters 1 and 4 were used to represent the effect of sources outside of Korea and the cluster 5 was used to indicate the effect of sources in Korea. Since clusters 2 and 3 contain mixed trajectories from Korea and out-
- of-Korea their contribution was divided evenly between in and out of Korea. The results indicate that the sources in Korea and outside Korea contributed about 50 % each to the TGM mass collected at the sample site during the sampling period while the Korean sources affected GOM and PBM more significantly, accounting for approximately 55 and 56 %, respectively (Table 4). These results augment the CPF and PSCF results
- which only use concentrations that are in the top 25th percentile. While CPF and PSCF found that for high concentration events Korean sources were most important for GOM while for TGM and PBM long-range and regional transport from China and North Korea were also important, the cluster based approach suggests that for all 3 species on average approximately 50% of the mass originates in Korea and 50% of the mass originates outside of Korea.



4 Conclusion and implications

This study was initiated to identify the sources affecting speciated mercury concentrations measured on an island located between mainland Korea and Eastern China. Various tools were used to locate and quantify the sources, including correlations with other

- ⁵ pollutants, CPF, and the back-trajectory based analysis (PSCF and cluster analysis). The results consistently show that Korean sources are most important for GOM while for other Hg species (TGM and PBM) long-range and regional transport from China and North Korea were also important. Existing methods including PSCF and CPF are able to locate the source direction and areas, but do not consider the frequency of the
- wind directions which can affect the long-term concentrations at the receptor site. In this study, a new approach considering both the cluster frequency and the Hg concentration associated with each cluster was used to quantify the source contribution at the sampling site. On average contributions from out-of-Korean sources were similar to Korean sources for TGM whereas Korean sources contributed approximately 55–56 %
- ¹⁵ of the GOM and PBM mass compared to the out-of-Korea sources.

The ratio of GOM / PBM proved to be a useful tool for identifying the relative significance of local sources vs. long-range transport. The GOM / PBM ratio decreased as the effect of long-range transport increased and vice versa since GOM has a shorter atmospheric residence time than PBM. The reciprocal of the PBM / GOM ratio was

²⁰ negatively correlated with atmospheric temperature and positively correlated with relative humidity, suggesting that the secondary formation of PBM through gas-particle partitioning of GOM was an important input for atmospheric PBM concentration at low temperature and high relative humidity. This result also suggests that the secondary formation of PBM becomes more important as the significance of long-range transport increased.

Author contributions. The work presented here was carried out in collaboration between all authors. Gang S. Lee analyzed data and wrote the paper. Pyung R. Kim performed the experiments and interpreted the results. Young J. Han defined the research theme, interpreted



the results, and wrote the paper. Yong S. Seo, Seung. M. Yi, and Thomas M. Holsen also interpreted the results and approved the final paper.

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Disclaimer. The authors declare no conflict of interest.

References

20

- AMAP/UNEP: Technical Background Report on the Global Anthropogenic Mercury Assessment 2013, Arctic Monitoring and Assessment Programme, Oslo, Norway/UNEP Chemicals Branch, Geneva, Switzerland, vi + 263 pp., 2013.
 - Amos, H. M., Jacob, D. J., Holmes, C. D., Fisher, J. A., Wang, Q., Yantosca, R. M., Corbitt, E. S., Galarneau, E., Rutter, A. P., Gustin, M. S., Steffen, A., Schauer, J. J., Gray-
- don, J. A., Louis, V. L. St., Talbot, R. W., Edgerton, E. S., Zhang, Y., and Sunderland, E. M.: Gas-particle partitioning of atmospheric Hg(II) and its effect on global mercury deposition, Atmos. Chem. Phys., 12, 591–603, doi:10.5194/acp-12-591-2012, 2012.
 - Auzmendi-Murua, I., Castillo, A., and Bozzelli, J. W.: Mercury oxidation via chlorine, bromine, and iodine under atmospheric conditions: thermochemistry and kinetics, J. Phys. Chem., 118, 2959–2975, 2014.
 - Baya, A. P. and Van Heyst, B.: Assessing the trends and effects of environmental parameters on the behaviour of mercury in the lower atmosphere over cropped land over four seasons, Atmos. Chem. Phys., 10, 8617–8628, doi:10.5194/acp-10-8617-2010, 2010.

Cheng, I., Zhang, L., Mao, H., Blanchard, P., Tordon, R., and John, D.: Seasonal and diurnal

- patterns of speciated atmospheric mercury at a coastal-rural and coastal-urban site, Atmos. Environ., 82, 193–205, 2014.
 - Choi, H. D., Holsen, T. M., and Hopke, P. K.: Atmospheric mercury in the Adirondacks: concentrations and sources, Environ. Sci. Technol., 42, 5644–5653, 2008.



- potential source in Guivang, China, Atmos. Environ., 45, 4205–4212, 2011.
- Gratz, L. E., Keeler, G. J., Marsik, F. J., Barres, J. A., and Dvonch, T.: Atmospheric transport of speciated mercury across southern Lake Michigan: influence from emission sources in the 25 Chicago/Gary urban area, Sci. Total Environ., 448, 84–95, 2013.
 - Gustin, M. S., Huang, J., Miller, M. B., Peterson, C., Jaffe, D. A., Ambrose, J., Finley, B. D., Lyman, S. N., Call, K., Talbot, R., Feddersen, D., Mao, H., and Lindberg, S. E.: Do we understand what the mercury speciation instruments are actually measuring? Results of RAMIX,
- Environ, Sci. Technol., 47, 7295–7306, 2013, 30

10

Han, Y. J., Holsen, T. M., and Hopke, P. K.: Estimation of source location of total gaseous mercury measured in New York State using trajectory-based models, Atmos. Environ., 41, 6033-6047, 2007.

workshop/Spring2011/HYSPLIT_Tutorial.pdf (last access: 1 February 2015), 2011. Driscoll, C. T., Han, Y. J., Chen, C. Y., Evers, D. C., Lambert, K. F., Holsen, T. M., Kamman, N. C., dual Tekran analyzers, in: ICMGP 2015, Jeju, Korea, 19 June 2015, 19M-S07-O-3, 2015.

- Feng, X., Lu, J. Y., Conrad, D. G., Hao, Y., Banic, C. M., and Schroeder, W. H.; Analysis of 15 inorganic mercury species associated with airborne particulate matter/aerosols: method development, Anal. Bioanal. Chem., 380, 683-689, 2004.
 - Finley, B. D., Jaffe, D. A., Call, K., Lyman, S., Gustin, M. S., Peterson, C., Miller, M., and Lyman, T.: Development, testing, and deployment of an air sampling manifold for spiking ele-
- mental and oxidized mercury during the Reno atmospheric mercury intercomparison experi-20 ment (RAMIX), Environ. Sci. Technol., 47, 7277-7284, 2013.
 - Fu, X., Feng, X., Qiu, G., Shang, L., and Zhang, H.: Speciated atmospheric mercury and its

- Fang, G. C., Zhang, L., and Huang, C. S.: Measurement of size-fractionated concentration and bulk dry deposition of atmospheric particulate bound mercury, Atmos. Environ., 61, 371-377, 2012.
- and Munson, R. K.: Mercury contamination in forest and freshwater ecosystems in the Northeastern United States, Bioscience, 57, 17-28, 2007. Edgerton, E. S.: Field investigations of RGM and fine particulate Hg measurement bias using
- NOAA Air Resour. Lab., Silver Spring, Maryland, USA, available at: http://www.arl.noaa.gov/ 5 documents/reports/hysplit user guide.pdf (last access: 1 February 2015), 2014.
- Draxler, R. R., Stunder, B., Rolph, G., Stein, A., and Taylor, A.: HYSPLIT_4 User's Guide,

Draxler, R. R., Stunder, B., Rolph, G., Stein, A., and Taylor, A.: HYSPLIT Tutorial, NOAA Air Re-

sour. Lab., Silver Spring, Maryland, USA, available at: http://www.arl.noaa.gov/documents/

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- Hedgecock, I. M. and Pirrone, N.: Chasing quicksilver: modeling the atmospheric lifetime of $Hg^{0}(g)$ in the marine boundary layer at various latitudes, Environ. Sci. Technol., 38, 69–76, 2004.
 - Huang, J., Choi, H. D., Hopke, P. K., and Holsen, T. M.: Ambient mercury sources in Rochester, NY: results from principal components analysis (PCA) of mercury monitoring network data, Environ. Sci. Technol., 44, 8441–8445, 2010.
- Huang, J., Miller, M. B., Weiss-Penzias, P., and Gustin, M. S.: Comparison of gaseous oxidized Hg measured by KCI-coated denuders, and nylon and cation exchange membranes, Environ. Sci. Technol., 47, 7307–7316, 2013.
 - Huang, X., Li, M., Friedli, H. R., Song, Y., Chang, D., and Zhu, L.: Mercury emissions from biomass burning in China, Environ. Sci. Technol., 45, 9442–9448, 2011.
- Jaffe, D. A., Lyman, S., Amos, H. M., Gustin, M. S., Huang, J., Selin, N. E., Levin, L., Ss-chure, A. T., Mason, R. P., Talbot, R., Rutter, A., Finley, B., Jaeglé, L., Shah, V., McClure, C., Ambrose, J., Gratz, L., Lindberg, S., Weiss-Penzias, P., Sheu, G. R., Fedddersen, D. D., Horvat, M., Dastoor, A., Hynes, A. J., Mao, H., Sonke, J. E., Slemr, F., Fisher, J. A., Ebinghaus, R., Zhang, Y., and Edwards, G.: Progress on understanding atmospheric mercury hampered by uncertain measurements, Environ. Sci. Technol., 48, 7204–7206, 2014.
- Kim, J. H., Park, J. M., Lee, S. B., Pudasainee, D., and Seo, Y. C.: Anthropogenic mercury emission inventory with emission factors and total emission in Korea, Atmos. Environ., 44, 2714–2721, 2010.

Kim, P. R., Han, Y. J., Holsen, T. M., and Yi, S. M.: Atmospheric particulate mercury: concentrations and size distributions, Atmos. Environ., 61, 94–102, 2012.

25

Kim, S. H., Han, Y. J., Holsen, T. M., and Yi, S. M.: Characteristics of atmospheric speciated mercury concentrations (TGM, Hg(II), and Hg(*p*)) in Seoul, Korea, Atmos. Environ., 43, 3267– 3274, 2009.

Lan, X., Talbot, R., Castro, M., Perry, K., and Luke, W.: Seasonal and diurnal variations of

- atmospheric mercury across the US determined from AMNet monitoring data, Atmos. Chem. Phys., 12, 10569–10582, doi:10.5194/acp-12-10569-2012, 2012.
 - Landis, M. S. and Keeler, G. J.: Atmospheric mercury deposition to Lake Michigan during the Lake Michigan mass balance study, Environ. Sci. Technol., 36, 4518–4524, 2002



- Lee, G. S., Kim, P. R., Han, Y. J., Holsen, T. M., and Lee, S. H.: Tracing sources of total gaseous mercury to Yongheung Island off the coast of Korea, Atmosphere, 5, 273–291, 2014.
- Lin, C. J. and Pehkonen, S. O.: The chemistry of atmospheric mercury: a review, Atmos. Environ., 33, 2067–2079, 1999.
- Liu, B., Keeler, G. J., Timothy Dvonch, J., Barres, J. A., Lynam, M. M., Marsik, F. J., and Morgan, J. T.: Urban-rural differences in atmospheric mercury speciation, Atmos. Environ., 44, 2013–2023, 2010.
 - Lu, W., Zhu, Z. Y., and Liu, W. P.: Salt water intrusion numerical simulation on application based on FEFLOW, Ground Water, 32, 19–21, 2010.
- ¹⁰ Lyman, S. N. and Gustin, M. S.: Determinants of atmospheric mercury concentrations in Reno, Nevada, USA, Sci. Total. Environ., 408, 431–438, 2009.
 - Lyman, S. N., Jaffe, D. A., and Gustin, M. S.: Release of mercury halides from KCI denuders in the presence of ozone, Atmos. Chem. Phys., 10, 8197–8204, doi:10.5194/acp-10-8197-2010, 2010.
- ¹⁵ Lynam, M. M. and Keeler, G. J.: Automated speciated mercury measurements in Michigan, Environ. Sci. Technol., 39, 9253–9562, 2005.
 - Malcolm, E. G. and Keeler, G. J.: Evidence for a sampling artifact for particulate-phase mercury in the marine atmosphere, Atmos. Environ., 41, 3352–3359, 2007.

Mao, H., Talbot, R., Hegarty, J., and Koermer, J.: Speciated mercury at marine, coastal, and

- ²⁰ inland sites in New England Part 2: Relationships with atmospheric physical parameters, Atmos. Chem. Phys., 12, 4181–4206, doi:10.5194/acp-12-4181-2012, 2012.
 - Mason, R. P. and Sullivan, K. A.: Mercury in Lake Michigan, Environ. Sci. Technol., 31, 942–947, 1997.

Mason, R. P., Morel, F. M. M., and Hemond, H. F.: The role of microorganisms in elemental mercury formation in natural water, Water Air Soil Pollut., 80, 775–787, 1995.

25

McClure, C. D., Jaffe, D. A., and Edgerton, E. S.: Evaluation of the KCl denuder method for gaseous oxidized mercury using HgBr₂ at an In-Service AMNet site, Environ. Sci. Technol., 48, 11437–11444, 2014.

NI, Nautilus Institute for Security and Sustainable Development: The DPRK Energy Sector: Es-

- timated Year 2000 Energy Balance and Suggested Approaches to Sectoral Redevelopment, Berkeley, CA, USA, 2003
 - Obrist, D., Johnson, D. W., Lindberg, S. E., Luo, Y., Hararuk, O., Bracho, R., Battles, J. J., Dail, D. B., Edmonds, R. L., Monson, R. K., Ollinger, S. V., Pallardy, S. G., Pregitzer, K. S.,



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⁵ Pirrone, N., Aas, W., Cinnirella, S., Ebinghaus, R., Hedgecock, I. M., Pacyna, J., Sprovieri, F., and Sunderland, E. M.: Toward the next generation of air quality monitoring: mercury, Atmos. Environ., 80, 599–611, 2013.

mercury to the atmosphere, Atmos. Environ., 30, 2981–2987, 1996.

15

20

centrations in biomass, litter, and soils, Environ. Sci. Technol., 45, 3974–3981, 2011.

Polissar, A. V., Hopke, P. K., and Harris, J. M.: Source regions for atmospheric aerosol measured at Barrow, Alaska, Environ. Sci. Technol., 35, 4214–4226, 2001a.

and Todd, D. E.: Mercury distribution across 14 US forests. Part I: Spatial patterns of con-

Pirrone, N., Keeler, G. J., and Nriagu, J. O.: Regional differences in worldwide emissions of

- ¹⁰ Polissar, A. V., Hopke, P. K., and Poirot, R. L.: Atmospheric aerosol over Vermont: chemical composition and sources, Environ. Sci. Technol., 35, 4604–4621, 2001b.
 - Rutter, A. P. and Schauer, J. J.: The effect of temperature on the gas-particle partitioning of reactive mercury in atmospheric aerosols, Atmos. Environ., 41, 8647–8657, 2007.

Seinfeld, J. H. and Pandis, S. N.: Atmospherical Chemistry and Physics: from Air Pollution to Climate Change, 2nd edn., John Wiley & Sons, Inc., Hoboken, NJ, USA, 2006.

Subir, M., Ariya, P. A., and Dastoor, A. P.: A review of uncertainties in atmospheric modeling of mercury chemistry. I. Uncertainties in existing kinetic parameters – fundamental limitations and the importance of heterogeneous chemistry, Atmos. Environ., 45, 5664–5676, 2011.

Subir, M., Ariya, P. A., and Dastoor, A. P.: A review of the sources of uncertainties in atmospheric mercury modeling. II. Mercury surface and heterogeneous chemistry – a missing link, Atmos. Environ., 46, 1–10, 2012.

- Talbot, R., Mao, H., Feddersen, D., Smith, M., Kim, S. Y., Sive, B., Haase, K., Ambrose, J., Zhou, Y., and Russo, R.: Comparison of particulate mercury measured with manual and automated methods, Atmosphere, 2, 1–20, 2011.
- ²⁵ UNEP: The global atmospheric mercury assessment, UNEP Chemicals Branch, Geneva, Switzerland, 2013.
 - US EPA: Mercury Study Report to Congress. Office of Air Quality Planning and Standards, Research Triangle Park, NC, USA, and Office of Research and Development, Washington, DC, USA, EPA-452/R-97-005, 1997.
- ³⁰ Wan, Q., Feng, X. B., Julia, L., Zheng, W., Song, X. J., Han, S. J., and Xu, H.: Atmospheric mercury in Changbai Mountain area, northeastern China – Part 1: The seasonal distribution pattern of total gaseous mercury and its potential sources, Environ. Res., 109, 201–206, 2009a.



- Wan, Q., Feng, X. B., Julia, L., Zheng, W., Song, X. J., Han, S. J., and Xu, H.: Atmospheric mercury in Changbai Mountain area, northeastern China. II. The distribution of reactive gaseous mercury and particulate mercury and mercury deposition fluxes, Environ. Res., 109, 721– 727, 2009b.
- ⁵ Wang, L., Wang, S., Zhang, L., Wang, Y., Zhang, Y., Nielsen, C., McElroy, B. M., and Hao, J.: Source apportionment of atmospheric mercury pollution in China using the GEOS-Chem model, Environ. Pollut., 190, 166–175, 2014.
- Wang, S., Holsen, T. M., Huang, J., and Han, Y.-J.: Evaluation of various methods to measure particulate bound mercury and associated artifacts, Atmos. Chem. Phys. Discuss., 13, 8585–
 8614, doi:10.5194/acpd-13-8585-2013, 2013.
- Weiss-Penzias, P., Jaffe, D. A., Mcclintick, A., Prestbo, E. M., and Landis, M. S.: Gaseous elemental mercury in the marine boundary layer: evidence for rapid removal in anthropogenic pollution, Environ. Sci. Technol., 37, 3755–3763, 2003.
- Weiss-Penzias, P., Jaffe, D. A., Swartzendruber, P., Dennison, J. B., Chand, D., Hafner, W., and
 Prestbo, E.: Observations of Asian air pollution in the free troposphere at Mount Bachelor
 Observatory during the spring of 2004, J. Geophys. Res., 111, 1–15, 2006.
 - Xiu, G. L., Jin, Q. X., Zhang, D. N., Shi, S. Y., Huang, X. J., Zhang, W. Y., Bao, L., Gao, P. T., and Chen, B.: Characterization of size-fractionated particulate mercury in Shanghai ambient air, Atmos. Environ., 39, 419–427, 2005.
- Xiu, G. L., Cai, J., Zhang, W. Y., Zhang, D. N., Bueler, A., Lee, S. C., Shen, Y., Xu, L. H., Huang, X. J., and Zhang, P.: Speciated mercury in size fractionated particles in Shanghai ambient air, Atmos. Environ., 43, 3145–3154, 2009.
- Xu, L. L., Chen, J. S., Yang, L. M., Niu, Z. C., Tong, L., Yin, L. Q., and Chen, Y. T.: Characteristics and sources of atmospheric mercury speciation in a coastal city, Xiamen, China, Chemosphere, 119, 530–539, 2015.
 - Zhang, L., Gong, S., Padro, J., and Barrie, L.: A size-segregated particle dry deposition scheme for and atmospheric aerosol module, Atmos. Environ., 35, 549–560, 2001.



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Table 1. Summarized concentrations of speciated Hg and other typical pollutants for each sampling period.

| Sampling periods T | | GOM | PBM | SO ₂ | NO ₂ | CO | O ₃ | PM ₁₀ |
|------------------------|-----------------------|----------------------|-----------------------|-----------------|-----------------|---------------|-----------------|-----------------------|
| | (ng m ⁻³) | (pgm ⁻³) | (pg m ⁻³) | (ppb) | (ppb) | (ppm) | (ppb) | (µg m ⁻³) |
| 1st: 17–23 Jan 2013 | 3.5 ± 0.8 | 5.8 ± 8.8 | 17.0 ± 16.5 | 5.8 ± 4.2 | 26.7 ± 12.4 | 0.5 ± 0.2 | 21.7 ± 13.5 | 55.9 ± 38.5 |
| 2nd: 25 Feb–1 Mar 2013 | 3.7 ± 0.9 | 13.2 ± 14.8 | 19.5 ± 19.6 | 7.3 ± 3.0 | 28.8 ± 19.1 | 0.7 ± 0.2 | 28.3 ± 14.7 | 83.6 ± 28.6 |
| 3rd: 8–13 Apr 2013 | 2.1 ± 0.4 | 4.3 ± 5.6 | 15.6 ± 13.8 | 4.9 ± 1.7 | 8.6 ± 3.3 | 0.4 ± 0.1 | 46.8 ± 6.0 | 45.8 ± 21.4 |
| 4th: 20-25 May 2013 | 2.8 ± 1.0 | 4.2 ± 5.9 | 6.7 ± 7.3 | 5.7 ± 2.5 | 17.2 ± 5.9 | 0.5 ± 0.1 | 36.2 ± 19.4 | 51.9 ± 21.6 |
| 5th: 19-24 Aug 2013 | 2.3 ± 0.9 | 3.2 ± 3.0 | 5.4 ± 4.6 | 4.7 ± 1.6 | 10.2 ± 3.8 | 0.5 ± 0.1 | 40.9 ± 29.6 | 34.2 ± 13.7 |
| 6th: 17–21 Mar 2014 | 2.6 ± 1.2 | 12.8 ± 15.7 | 7.3 ± 3.5 | 7.5 ± 0.1 | 32.9 ± 75.1 | 0.4 ± 0.2 | 39.0 ± 12.5 | 66.7 ± 44.4 |
| 7th: 26–31 May 2014 | 2.8 ± 0.7 | 13.5 ± 9.0 | 9.8 ± 12.2 | 5.8 ± 2.3 | 12.4 ± 5.6 | 0.5 ± 0.1 | 85.5 ± 23.3 | 124.5 ± 44.9 |
| 8th: 19-23 Aug 2014 | 2.4 ± 1.1 | 10.7 ± 3.7 | 6.3 ± 3.6 | 3.4 ± 1.2 | 8.7 ± 6.0 | 0.4 ± 0.1 | 38.7 ± 16.5 | 32.0 ± 22.3 |
| Average | 2.8 ± 1.1 | 9.8 ± 9.9 | 10.6 ± 12.0 | 5.6 ± 5.0 | 18.2 ± 28.7 | 0.5 ± 0.2 | 42.1 ± 25.8 | 61.9 ± 42.8 |

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Table 2. Comparisons of measured Hg concentrations with those reported in other studies.

| Country | Site | Remarks | Year | TGM (ngm ⁻³) | GOM (pgm ⁻³) | PBM (pgm ⁻³) | Reference |
|---------|---------------|----------|-----------|--------------------------|--------------------------|--------------------------|---------------------------|
| Korea | Seoul | Urban | 2005–2006 | 3.22 ± 2.10 | 27.2 ± 19.3 | 23.9 ± 19.6 | Kim et al. (2009) |
| | Seoul | Urban | 2006-2009 | 3.72 ± 2.96 | 11.3 ± 9.5 | 13.4 ± 12.0 | Han et al. (2014) |
| | Chuncheon | Rural | 2006-2009 | 2.12 ± 1.47 | 2.7 ± 2.7 | 3.7 ± 5.7 | Han et al. (2014) |
| | Yongheung | Island | 2013–2014 | 2.8 ± 1.1 | 9.8 ± 9.9 | 10.6 ± 12.0 | This study |
| China | Guiyang | Urban | 2009 | 9.7 ± 10.2 | 35.7 ± 43.9 | 368.0 ± 676.0 | Fu et al. (2011) |
| | Xiamen | Suburban | 2012-2013 | 3.5 | 61.05 | 174.41 | Xu et al. (2015) |
| | Mt. Changbai | Remote | 2005–2006 | 3.58 ± 1.78 | 65 ± 84 | 77 ± 136 | Wan et al. (2009a, b) |
| USA | Chicago | Urban | 2007 | 2.5 ± 1.5 | 17 ± 87 | 9 ± 20 | Gratz et al. (2013) |
| | Reno, Nevada | Suburban | 2007–2009 | 2.0 ± 0.7 | 18 ± 22 | 7±7 | Lyman and Gustin (2009) |
| | Rochester, NY | Urban | 2008–2009 | 1.49 | 4.08 | 6.57 | Huang et al. (2010) |
| Canada | Nova Scotia | Urban | 2010-2011 | 1.67 ± 1.01 | 2.07 ± 3.35 | 2.32 ± 3.09 | Cheng et al. (2014) |
| | Ontario | Rural | 2006–2007 | 1.17 | 15.10 | 16.40 | Baya and Van Heyst (2010) |

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Table 3. Correlation coefficients and *p* values (in parenthesis) for speciated Hg with other pollutants for the whole sampling period. Correlation coefficients with an asterisk indicate a statistically significant relationship at $\alpha = 0.05$.

| | TGM | GOM | SO ₂ | NO ₂ | CO | O ₃ | PM ₁₀ |
|-----|----------------|----------------|-----------------|-----------------|------------------|----------------|------------------|
| TGM | - | -0.132 (0.233) | 0.115* (0.001) | 0.063 (0.074) | 0.571* (< 0.001) | -0.055 (0.115) | 0.401* (< 0.001) |
| GOM | -0.132 (0.233) | - | 0.025 (0.822) | 0.022 (0.846) | -0.149 (0.180) | 0.143 (0.197) | 0.128 (0.248) |
| PBM | 0.235* (0.030) | 0.021 (0.855) | -0.006 (0.954) | 0.008 (0.941) | 0.215* (0.048) | 0.029 (0.794) | 0.139 (0.206) |

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Table 4. Estimated contribution of Korean and out-of-Korean sources on variations of speciated Hg concentration.

| Cluster | Cluster | Average concentration* | | | Source contribution (%) | | | Korean (%) | | | Out-of-Korean (%) | | |
|---------------|---------------|------------------------|-----|------|-------------------------|------|------|------------|------|------|-------------------|------|------|
| | frequency (%) | TGM | GOM | PBM | TGM | GOM | PBM | TGM | GOM | PBM | TGM | GOM | PBM |
| 1 | 12 | 2.2 | 6.9 | 7.7 | 9.7 | 10.6 | 8.8 | | | | 9.7 | 10.6 | 8.8 |
| 2 | 25 | 2.8 | 8.3 | 12.6 | 25.7 | 26.6 | 30.0 | 12.9 | 13.3 | 15.0 | 12.9 | 13.3 | 15.0 |
| 3 | 24 | 2.9 | 8.1 | 10.0 | 25.5 | 24.9 | 22.8 | 12.8 | 12.5 | 11.4 | 12.8 | 12.5 | 11.4 |
| 4 | 13 | 3.2 | 5.5 | 7.4 | 15.3 | 9.2 | 9.2 | | | | 15.3 | 9.2 | 9.2 |
| 5 | 25 | 2.6 | 9.0 | 12.3 | 23.8 | 28.8 | 29.3 | 23.8 | 28.8 | 29.3 | | | |
| Korean | | | | | | | | 49.4 | 54.6 | 55.7 | | | |
| Out-of-Korean | | | | | | | | | | | 50.6 | 45.6 | 44.4 |

 * TGM is shown in ng m $^{-3}$ while for both GOM and PBM the units are pg m $^{-3}$.



Figure 1. (a) The sampling site in Yongheung Island (the star mark). **(b)** Anthropogenic mercury emission sources in Korea. Blue star indicates the sampling site, and the green circle indicates the area where major Korean coal-fired power plants are located. **(c)** The enlarged image of the area near the sampling site.





Figure 2. TGM, GOM, and PBM concentrations measured during the eight sampling periods. TGM was measured every 5 min while GOM and PBM were measured during 12 h except when measured every 2 h in May 2014.





Figure 3. Seasonally averaged concentrations of TGM, GOM, and PBM shown as a box-andwhisker plot. The red dash lines indicate the arithmetic mean.

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Figure 5. CPF plots for TGM using the top 25 % (left upper panel) and the top 10 % (right upper panel) as a criterion, and for GOM (left bottom panel) and for PBM (right bottom panel). For both GOM and PBM, the criterion of the top 25% concentration was used.





Figure 6. Frequency of wind direction with different GOM / PBM ratios. Southerly and easterly winds prevailed for the samples with high GOM / PBM ratio whereas the percentage of westerly winds increased as the GOM / PBM ratio decreased.





Figure 7. PSCF results for TGM (left), PBM (middle), and GOM (right) using the top 25 % of concentrations as criteria.





Figure 8. Back-trajectories for clusters 1 through 5. The left top panel indicates the mean back-trajectory and contribution for each cluster.

