1	Microphysics-based Black Carbon Aging in a global CTM: Constraints from
2	HIPPO Observations and Implications for Global Black Carbon Budget
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4	Cenlin He ¹ , Qinbin Li ¹ , Kuo-Nan Liou ¹ , Ling Qi ¹ , Shu Tao ² , Joshua P. Schwarz ³
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6	¹ Department of Atmospheric and Oceanic Sciences and Joint Institute for Regional Earth
7	System Science and Engineering, University of California, Los Angeles (UCLA), Los
8	Angeles, CA 90095, USA
9	² Laboratory for Earth Surface Processes, College of Urban and Environmental Sciences,
10	Peking University, Beijing 100871, China
11	³ Chemical Sciences Division, Earth System Research Laboratory, NOAA, Boulder, CO
12	80305, USA
13	
14	
15	Correspondence to: Cenlin He (cenlinhe@atmos.ucla.edu)
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18 Abstract

We develop and examine a microphysics-based black carbon (BC) aerosol aging scheme 19 20 that accounts for condensation, coagulation, and heterogeneous chemical oxidation 21 processes in a global 3-D chemical transport model (GEOS-Chem) by interpreting the BC measurements from the HIAPER Pole-to-Pole Observations (HIPPO, 2009–2011) using 22 the model. We convert aerosol mass in the model to number concentration by assuming 23 24 lognormal aerosol size distributions and compute the microphysical BC aging rate 25 (excluding chemical oxidation aging) explicitly from the condensation of soluble materials onto hydrophobic BC and the coagulation between hydrophobic BC and 26 preexisting soluble particles. The chemical oxidation aging is tested in the sensitivity 27 simulation. The microphysical aging rate is ~4 times higher in the lower troposphere over 28 29 source regions than that from a fixed aging scheme with an e-folding time of 1.2 days. The higher aging rate reflects the large emissions of sulfate-nitrate and secondary organic 30 aerosol precursors hence faster BC aging through condensation and coagulation. In 31 contrast, the microphysical aging is more than fivefold slower than the fixed aging in 32 remote regions, where condensation and coagulation are weak. Globally BC 33 microphysical aging is dominated by condensation, while coagulation contribution is 34 largest over East China, India, and Central Africa. The fixed aging scheme results in an 35 overestimate of HIPPO BC throughout the troposphere by a factor of 6 on average. The 36 microphysical scheme reduces this discrepancy by a factor of ~3, particularly in the 37 middle and upper troposphere. It also leads to a threefold reduction in model bias in the 38 latitudinal BC column burden averaged along the HIPPO flight tracks, with largest 39 improvements in the tropics. The resulting global annual mean BC lifetime is 4.2 days 40 and BC burden is 0.25 mg m^{-2} , with 7.3% of the burden at high altitudes (above 5 km). 41 Wet scavenging accounts for 80.3% of global BC deposition. We find that in source 42 regions the microphysical aging rate is insensitive to aerosol size distribution, 43 condensation threshold, and chemical oxidation aging, while it is the opposite in remote 44

- 45 regions, where the aging rate is orders of magnitude smaller. As a result, global BC
- 46 burden and lifetime show little sensitivity (<5% change) to these three factors.

48 **1. Introduction**

Black carbon (BC) aerosol is one of the most important contributors to current global and 49 50 regional warming (Bond et al., 2013). BC directly absorbs solar radiation, leading to significant atmospheric warming (Ramanathan and Carmichael, 2008). It also acts as 51 cloud condensation nuclei (CCN), affecting cloud formation and distribution (Jacobson, 52 2014). Additionally, BC reduces snow albedo after deposition on snow, resulting in 53 54 accelerated snow melting (Painter et al., 2013; Liou et al., 2014). The assessment by Bond et al. (2013) pointed out that current estimates of BC climatic effects involve large 55 uncertainties. One of the critical uncertainty sources is BC atmospheric aging through the 56 physical and chemical transformation of BC from hydrophobic to hydrophilic particles. 57

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59 BC is emitted mostly as hydrophobic particles externally mixed with other aerosol constituents (Zhang et al., 2008). Recent studies showed that BC can also be frequently 60 mixed with organics even at emission under specific conditions (Willis et al., 2015 and 61 references therein). Hydrophobic BC becomes hydrophilic due to increasing internal 62 mixing with water-soluble materials through condensation (Moteki et al., 2007), 63 coagulation (Johnson et al., 2005), and heterogeneous oxidation (Khalizov et al., 2010) 64 during atmospheric aging. Hereinafter we refer to these internal mixtures generically as 65 "coatings" without making any specific reference to mixing morphology. Coating 66 67 enhances BC absorption and scattering capacities (Bond et al., 2006), which depends on coating properties and particle morphology (Scarnato et al., 2013; He et al., 2015). 68 Coated BC particles typically have a higher hygroscopicity (Zhang et al., 2008) and 69 hence more efficient wet scavenging, which further affects BC atmospheric lifetime 70 71 (Zhang et al., 2015). Thus, BC aging is expected to play a critical role in affecting both BC optical properties and global distribution. 72

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74 In global chemical transport models (CTMs), BC aging is typically parameterized by a

fixed e-folding time of 1–2 days for the hydrophobic-to-hydrophilic BC conversion (e.g., 75 Chung and Seinfeld, 2002; Koch et al., 2009; Q. Q. Wang et al., 2011, 2014). However, 76 77 Koch et al. (2009) and Schwarz et al. (2013) showed that most global models significantly deviate from observed global BC distributions. It is likely that the 78 prescription of uniform BC aging timescales may be partially contributing to such biases. 79 In this vein, Shen et al. (2014) optimized the fixed e-folding aging time of BC by fitting a 80 81 global CTM results to HIPPO observations. They suggested that anthropogenic BC from 82 East Asia ages much faster than one day, while the aging of biomass burning BC from Southeast Asia is much slower. Using another global CTM constrained by HIPPO 83 observations, Zhang et al. (2015) pointed out that the optimized e-folding aging time 84 following Shen et al. (2014) varies significantly for BC emitted from different source 85 86 regions, with less than half a day for BC emitted from the tropics and mid-latitudes and ~1 week for BC emitted from high latitudes. Thus, a uniform BC aging time is likely not 87 representative and can conceivably lead to large uncertainties. 88

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90 In addition, field measurements have shown that a fixed e-folding time of 1-2 days underestimates the BC aging rate in polluted areas and is unrepresentative under complex 91 92 atmospheric conditions. For example, Johnson et al. (2005) found that ambient BC particles that underwent aging for less than a few hours were heavily coated in Mexico 93 94 City, primarily with ammonium sulfate. Schwarz et al. (2008) showed that 60–80% of BC particles are coated in fresh emissions from biomass burning sources. Moffet and Prather 95 (2009) observed a BC aging time of ~3 hours in Mexico City under photochemically 96 active conditions. Akagi et al. (2012) showed that the fraction of coated BC particles 97 98 produced from a chaparral fire in California increased up to ~85% over a 4-hour period. Therefore, it is imperative to better capture BC aging rate under different atmospheric 99 conditions in order to accurately estimate BC spatiotemporal distribution and 100 consequently its radiative effects. 101

To that end, several global models have treated BC aging with size-resolved aerosol 103 104 microphysics (e.g., Jacobson, 2010; Aquila et al., 2011). Many global models still rely on 105 relatively simple parameterizations for BC aging (e.g., Riemer et al., 2004; Liu et al., 2011; Oshima and Koike, 2013) in part for computational efficiency consideration. For 106 example, Riemer et al. (2004) developed a BC aging parameterization in which aging rate 107 108 was a function of total number concentration of secondary inorganic particles and 109 internally mixed BC particles, representing the effects of condensation and coagulation processes. Liu et al. (2011) proposed another parameterization in which BC aging rate 110 was a linear function of hydroxyl radical (OH) concentration (i.e., a fast-aging term 111 representing condensation of sulfuric acid) with a constant slow-aging term (e.g., 112 113 coagulation). Croft et al. (2005) and Huang et al. (2013) further employed these two parameterizations in a global model to estimate aging effects on BC budget and lifetime. 114 However, there are limitations in these simplified aging parameterizations. The Riemer et 115 al. (2004) parameterization was designed specifically for domains dominated by fossil 116 117 fuel emissions, which may not be suitable for application to a global scale. The Liu et al. (2011) formulation neglected the dependencies of BC aging on different condensable 118 materials and their precursors (e.g., SO₂). Furthermore, under complex atmospheric 119 120 conditions, these parameterizations likely introduce large uncertainties by lumping microphysical details of BC aging into a few parameters. Incorporating explicit 121 microphysical representations of the BC aging process in global models may partially 122 rectify and reduce the uncertainties. 123

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In this study, we develop a "hybrid" microphysics-based BC aging scheme that accounts
for condensation and coagulation processes in the GEOS-Chem global 3-D CTM.
GEOS-Chem has been widely used to simulate BC in source regions (Li et al., 2015),
continental outflows (Park et al., 2005), remote mountainous regions (He et al., 2014a),

the Arctic (Wang et al., 2011), and remote oceans (Q. Q. Wang et al., 2014). Presently 129 GEOS-Chem employs a fixed e-folding time of 1.2 days for the BC aging (Park et al., 130 131 2003). We convert aerosol mass in the model to number concentration by assuming 132 lognormal aerosol size distributions and explicitly compute the microphysical BC aging rate from the condensation of soluble materials onto hydrophobic BC and the coagulation 133 between hydrophobic BC and preexisting hydrophilic particles. The 'hybrid' 134 135 microphysical aging scheme thus not only takes account of the microphysical aging processes but also avoids the use of full-fledged dynamic aerosol microphysics thereby 136 retains the computational efficiency of the fixed e-folding time aging scheme. The 137 'hybrid' microphysical aging scheme can be similarly applied in other CTMs. We 138 systematically examine BC simulations using the aging scheme by comparison with the 139 140 HIAPER Pole-to-Pole Observations (HIPPO) of BC during 2009-2011. We further analyze the effects of the aging scheme on global BC lifetime and budget. Finally, we 141 quantify the uncertainties associated with key parameters in the aging scheme and the 142 effects of chemical oxidation on BC aging. 143

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145 **2. Methods**

146 **2.1 Observations**

147 We use the HIPPO aircraft measurements (Wofsy et al., 2011) of BC in this study. HIPPO 148 (http://hippo.ornl.gov/) provides unique constraints on BC distributions from the surface up to ~14 km across the Pacific from 67°S to 85°N latitudes. There were five 149 deployments during 2009–2011 (Fig.1, HIPPO 1: January 8–30, 2009; HIPPO 2: October 150 30 – November 22, 2009; HIPPO 3: March 24 – April 16, 2010; HIPPO 4: June 14 – July 151 11, 2011; HIPPO 5: August 9 - September 9, 2011). The refractory BC (rBC) mass 152 concentration was measured by a single-particle soot photometer (SP2) that detects 153 individual particles (Schwarz et al., 2010, 2013). SP2 measures rBC in a mass range 154 corresponding to volume-equivalent diameter range of ~90-550 nm assuming 1.8 g/cc 155

void free density. This range contains about 90% of the total BC mass in the 156 accumulation mode. The observed rBC concentration was scaled upwards by 10% to 157 158 account for BC particles undetected by SP2 (Schwarz et al., 2010) in this mode. rBC is 159 experimentally equivalent to elemental carbon at the 15 % level (Kondo et al., 2011), and hence is equivalent to BC in the model. The effective detection limit (2σ level) is 0.01 ng 160 kg⁻¹ (0.1 ng kg⁻¹) for 15-minute (1-minute) sampling at low altitudes and increase to 0.05 161 (0.5 ng kg⁻¹) at the higher altitudes (Schwarz et al., 2013). The SP2 measurement of rBC 162 163 mass is insensitive to non-BC mass and not influenced by other absorbing particles such as dust or non-absorbing species including salt and sulfate. Schwarz et al. (2013) 164 determined a total systematic uncertainty of 30% associated with rBC mass concentration 165 measured with the SP2. Schwarz et al. (2010, 2013) provided details of the BC 166 167 measurement during HIPPO. We average the BC observations that are located within each model grid and over the model transport time step (15 minutes), thus ensuring a 168 consistent spatiotemporal resolution for direct comparison with the model results. 169

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171 **2.2 Model description and simulations**

In this study, we use the GEOS-Chem model (version 9-01-03) driven by assimilated 172 meteorological fields from the Goddard Earth Observing System (GEOS-5) of the NASA 173 174 Global Modeling and Assimilating Office (GMAO). GEOS-5 meteorological fields have 175 a 6-hour temporal resolution (3 hours for surface variables and mixing depths), a native horizontal resolution of 0.5°×0.667°, and 72 vertical layers (from the surface to 0.01 hPa). 176 The spatial resolution is degraded to $2^{\circ} \times 2.5^{\circ}$ horizontally and 47 layers vertically for 177 computational efficiency. GEOS-Chem includes a fully coupled treatment of tropospheric 178 O₃-NO_x-VOC chemistry, sulfate-nitrate-ammonia and carbonaceous aerosols. Park et al. 179 (2003) presented the first GEOS-Chem simulation of carbonaceous aerosols including 180 BC and organic carbon (OC). The model also accounts for other aerosols including 181 secondary organic aerosol (SOA), dust, and sea salt. GEOS-Chem uses a bulk aerosol 182

scheme that separately tracks mass concentrations of different aerosol species (i.e., 183 externally mixed). The model resolves hydrophobic and hydrophilic BC and OC, 184 fine-mode (0.01–0.5 µm) and coarse-mode (0.5–8.0 µm) sea salt, dust in four size bins 185 186 (0.1–1.0, 1.0–1.8, 1.8–3.0, and 3.0–6.0 µm), and five types of lumped SOA formed from different precursors. Aerosol and gas phase simulations are coupled through formation of 187 sulfate, nitrate, and SOA, heterogeneous chemistry, and aerosol effects on photolysis 188 189 rates. Details on the GEOS-Chem aerosol simulations are provided, respectively, by Park 190 et al. (2003) for BC and OC, Park et al. (2004) for sulfate-nitrate-ammonia, Liao et al. (2007) for SOA, Fairlie et al. (2007) for dust, and Alexander et al. (2005) for sea salt. 191

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193 2.2.1 A microphysics-based BC aging scheme

194 We assume that 80% of freshly emitted BC particles are hydrophobic (Cooke et al., 1999; Park et al., 2003). Recent observations (e.g., Johnson et al., 2005; Willis et al., 2015) 195 showed that BC particles can be often coated with organics at emissions, suggesting that 196 the hydrophobic fraction of freshly emitted BC particles depends on the amount and type 197 198 of coated organics. Thus, the assumed fresh hydrophobic BC fraction in this study could involve uncertainty, which requires further investigation. Presently BC aging process is 199 not explicitly represented in GEOS-Chem. Instead, a fixed e-folding time (τ) of 1.2 days 200 201 is assumed for the hydrophobic-to-hydrophilic BC conversion in the forms (Park et al., 202 2003):

203
$$\left(\frac{dm_{BCPO}}{dt}\right) = -\frac{m_{BCPO}}{\tau}$$
(1)

204
$$\left(\frac{dm_{BCPI}}{dt}\right) = -\left(\frac{dm_{BCPO}}{dt}\right)$$
(2)

where m_{BCPO} and m_{BCPI} are the mass concentration of hydrophobic (BCPO) and hydrophilic BC (BCPI), respectively. In this study, we develop a microphysics-based BC aging scheme in the model by explicitly accounting for both condensation and coagulation processes. This microphysical aging scheme can be similarly applied in other
CTMs. Additionally, we incorporate an experiment-based parameterization for BC aging
through heterogeneous chemical oxidation (Pöschl et al., 2001) for comparison and
contrast.

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213 **2.2.1.1 Condensation**

The condensation rate of a gaseous species (*A*) onto an individual particle can be expressed as follows (Seinfeld and Pandis, 2006):

216
$$J_{A,p} = 4\pi f(K_n, \alpha) R_p D_{f_A}(c_{\infty} - c_s)$$
(3)

217
$$f(K_n, \alpha) = \frac{1 + K_n}{1 + 2K_n (1 + K_n) / \alpha}$$
(4)

where $J_{A,p}$ is the condensation rate (mol s⁻¹) of A onto a particle, R_p represents the particle 218 radius (m), D_{f_A} is the gas-phase diffusivity (m² s⁻¹) of A, c_{∞} and c_s are gas-phase 219 concentrations (mol m^{-3}) of A far from the particle and at the particle surface, respectively. 220 K_n is the Knudsen number (i.e., ratio of air mean free path to particle radius), α is the 221 accommodation coefficient ($\alpha = 1$ in this study), and $f(K_n, \alpha)$ is the correction factor for 222 223 non-continuum effects and imperfect surface accommodation based on the mathematical expression from Dahneke (1983). The mass of A condensing onto BCPO in a model grid 224 per unit time, k_A , can be calculated by the product of the total available mass of A for 225 226 condensing onto all pre-existing aerosols and the fraction of condensed mass partitioned 227 to BCPO, which depends on condensation rate (J) and particle number concentration (N)as follows: 228

229
$$k_{A} = \frac{J_{A,BCPO,tot}}{\sum_{p_{i=1}}^{p_{i=7}} J_{A,p_{i},tot}} M_{A,cond} = \frac{\int_{0}^{\infty} n_{BCPO}(R_{BCPO}) f_{BCPO}(K_{n},\alpha) R_{BCPO} dR_{BCPO}}{\sum_{p_{i=1}}^{p_{i=7}} \int_{0}^{\infty} n_{p_{i}}(R_{p_{i}}) f_{p_{i}}(K_{n},\alpha) R_{p_{i}} dR_{p_{i}}} M_{A,cond}$$
(5)

where p_i (i = 1-7) represents seven types of pre-existing aerosols (i.e., BCPO, BCPI, hydrophobic OC, hydrophilic OC, sulfate, fine-mode and coarse-mode sea salt) available for condensation, $J_{A,p_i,tot}$ is the condensation rate of *A* onto particle p_i , $M_{A,cond}$ is the total condensed mass of *A* in a model grid per unit time, R_{p_i} and n_{p_i} (= dN_{p_i}/dR_{p_i}) are the radius and number concentration distribution function of pre-existing particles, respectively. We account for condensation of gaseous sulfuric acid (H₂SO₄), nitric acid (HNO₃), ammonia (NH₃), and SOA onto pre-existing BC, OC, sulfate (SO₄²⁻, NO₃⁻, and NH₄⁺), and sea salt aerosols. We do not include the condensation of soluble materials on dust particles, which may introduce some uncertainty.

239

GEOS-Chem tracks only aerosol mass concentration rather than number concentration that is required in Eq. (4). We convert aerosol mass concentration (m_{p_i}) to number concentration (N_{p_i}) , assuming lognormal distributions for different aerosols following Croft et al. (2005) in the form:

244
$$N_{p_i} = \frac{m_{p_i}}{\rho_{p_i}} (\frac{\pi}{6} D_{p_i}^3 \exp(\frac{9}{2} \ln^2 \sigma_{p_i}))^{-1}$$
(6)

where ρ_{p_i} is the particle density (1.8 g cm⁻³ for BC and OC, 1.7 g cm⁻³ for sulfate, and 245 2.2 g cm⁻³ for sea salt), D_{p_i} and σ_{p_i} are the geometric mean diameter and standard 246 deviation of number size distribution, respectively. Following Dentener et al. (2006) and 247 Yu and Luo (2009), we assume $D_p = 60$ nm and $\sigma_p = 1.8$ for BCPO and hydrophobic OC, 248 and $D_p = 150$ nm and $\sigma_p = 1.8$ for BCPI and hydrophilic OC (Table 1). We use size 249 distributions from the Global Aerosol Dataset (GADS) (Koepke et al., 1997) for sulfate 250 $(D_p = 150 \text{ nm}, \sigma_p = 1.6)$, fine sea salt $(D_p = 200 \text{ nm}, \sigma_p = 1.5)$, and coarse sea salt $(D_p = 1.6)$ 251 800 nm, $\sigma_p = 1.8$). In order to analytically compute the integral in Eq. (5), we have 252 assumed a constant correction factor $f(K_n, \alpha)$ for each type of aerosols with different sizes, 253 which may introduce uncertainty in the computation. Under this assumption and using a 254 lognormal aerosol size distribution, the integral can now be computed by following the 255

256 mathematical identity:

257
$$\int_{0}^{\infty} n_{p_{i}}(R_{p_{i}})R_{p_{i}}dR_{p_{i}} = N_{p_{i}}\frac{D_{p_{i}}}{2}\exp(\frac{1}{2}\ln^{2}\sigma_{p_{i}})$$
(7)

The hydrophobic-to-hydrophilic BC conversion rate (kg m⁻³ s⁻¹) due to condensation can be written as

260
$$\left(\frac{dm_{BCPO}}{dt}\right)_{cond} = -\frac{F_{BCPO->BCPI}m_{BCPO}}{\Delta t}$$
(8)

261
$$F_{BCPO->BCPI} = \frac{\sum_{A_{i=1}}^{A_{i=4}} k_{A_i} \cdot \Delta t}{\beta M_{BCPO}}$$
(9)

where $F_{BCPO->BCPI}$ represents the fraction of BCPO becoming BCPI through condensation 262 of four types of soluble species A_i (i = 1-4, i.e., H_2SO_4 , HNO_3 , NH_3 , and SOA) in a 263 model time step (Δt). This implicitly assumes that different secondary aerosol species 264 have the same hygroscopicity. We note that some SOA species could be less hygroscopic 265 than ammonia sulfate (Prenni et al., 2007). m_{BCPO} is BCPO mass concentration (kg m⁻³), 266 M_{BCPO} is the total BCPO mass (kg) in a model grid, and β is the condensation threshold 267 (i.e., the mass fraction of condensed soluble materials on BCPO required for the 268 hydrophobic-to-hydrophilic conversion). Following Riemer et al. (2004), we set $\beta = 5\%$ 269 in the standard simulation based on hygroscopic growth behavior of aerosols 270 271 (Weingartner et al., 1997). After the hydrophobic-to-hydrophilic BC conversion, we lump 272 the mass of secondary aerosol material coated on BC with those not mixed with BC in order to be compatible with the bulk aerosol scheme in GEOS-Chem, where the mass 273 concentration of different aerosol species are separately tracked. The lumping, instead of 274 treating coating materials and hydrophilic BC together, only introduces small uncertainty, 275 276 considering that the size distribution of hydrophilic BC is similar to that of SOA and sulfate in this study (see Table 1). Further sensitivity analysis also show minor effects of 277 the hydrophilic aerosol size distribution on global BC concentration and lifetime (see 278 Sect. 4.3). The use of global uniform particle size distributions and β value can 279

conceivably introduce large uncertainties. To quantify the uncertainties, we conduct additional model simulations by varying the size distribution and β value (see Table 1 and Sect. 4).

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284 **2.2.1.2 Coagulation**

The coagulation rate $(J_{BCPO,X}, m^{-3} s^{-1})$ between BCPO and hydrophilic particles (*X*) can be expressed by (Seinfeld and Pandis, 2006)

$$J_{BCPO,X} = \gamma K_{BCPO,X} N_{BCPO} N_X \tag{10}$$

288
$$K_{BCPO,X} = 4\pi (R_{BCPO} + R_X) (D_{f_{BCPO}} + D_{f_X})$$
(11)

where N_{BCPO} and N_X are number concentrations (m⁻³) of BCPO and particle X computed 289 from Eq. (6), $K_{BCPO,X}$ is the coagulation coefficient (m³ s⁻¹) that depends on particle radius 290 $(R_{BCPO} \text{ and } R_X)$ and Brownian diffusivities $(D_{f_{BCPO}} \text{ and } D_{f_X})$, and γ (from 0.014 for 291 0.001 µm particles to 1.0 for 1 µm particles) is a correction factor that accounts for 292 kinetic effects of small particles. We use geometric mean radii for R_{BCPO} and R_X , which 293 could introduce uncertainty for particle sizes largely deviating from the mean value. We 294 note that the resulting uncertainty in BC concentration and lifetime is likely small, 295 because model results show that coagulation only makes a small contribution to the total 296 BC aging rate over the globe (see Sect. 3.1) and the global BC distribution is insensitive 297 to aerosol size distribution in this study (see Sects. 4.2 and 4.3). The 298 hydrophobic-to-hydrophilic BC conversion rate through coagulation can be written in the 299 form: 300

301
$$\left(\frac{dN_{BCPO}}{dt}\right)_{coag} = -\sum_{p_{j=1}}^{p_{j=6}} \gamma_{p_j} K_{BCPO, p_j} N_{p_j} N_{BCPO}$$
(12)

where p_j (j = 1-6) represents six types of hydrophilic particles, including sulfate-nitrate-ammonia, BCPI, hydrophilic OC, SOA, fine-mode and coarse-mode sea salt. We assume that BCPO is converted to BCPI upon coagulating with these hydrophilic aerosols. The resulting BCPO mass change per unit time (kg $m^{-3} s^{-1}$) is given by:

306
$$\left(\frac{dm_{BCPO}}{dt}\right)_{coag} = \frac{\pi}{6}\rho_{BCPO}D_{BCPO}^3 \exp(\frac{9}{2}\ln^2\sigma_{BCPO})\left(\frac{dN_{BCPO}}{dt}\right)_{coag}$$
(13)

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308 2.2.1.3 Chemical oxidation

To account for the BC aging through ozone oxidation on the BC surface, we follow an experiment-based parameterization by Pöschl et al. (2001). The same parameterization has been used in previous studies (Croft et al., 2005; Huang et al., 2013). The chemical hydrophobic-to-hydrophilic BC conversion rate (kg m⁻³ s⁻¹) can be expressed by

313
$$\left(\frac{dm_{BCPO}}{dt}\right)_{chem} = -k_{chem}m_{BCPO}$$
(14)

314 where k_{chem} is the reaction rate coefficient (s⁻¹) given by

315
$$k_{chem} = \frac{\lambda K_{\infty} K_{O_3}[O_3]}{1 + K_{O_3}[O_3] + K_{H_2O}[H_2O]}$$
(15)

where K_{∞} (= 0.015 s⁻¹) is the pseudo-first-order decay rate coefficient in the presence of high ozone concentrations, K_{O_3} (= 2.8×10⁻¹³ cm³) and K_{H_2O} (= 2.1×10⁻¹⁷ cm³) are the adsorption rate coefficients of O₃ and H₂O, which are a function of available adsorption sites residence time, and sticking coefficients of O₃ and H₂O on BC surface. λ (= 0.01) is the physical shielding factor that accounts for the fact that the oxidized coating material is not distributed homogenously over the BC particle surface (Croft et al., 2005). [O₃] and [H₂O] are atmospheric concentrations (molec cm⁻³) of O₃ and H₂O, respectively.

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Recent experimental studies also confirmed that BC can be aged through heterogeneous chemical oxidation by O_3 (Decesari et al., 2002; Zuberi et al., 2005) and NO₂ (Khalizov et al., 2010), which results in the formation of soluble organic compounds on BC particle surface. However, none of these recent studies have given explicit quantitative parameterizations for BC chemical aging, which can be applied in modeling studies. Moreover, experimental results presented in these studies vary substantially, suggesting that BC chemical aging process could involve large uncertainty. For this reason, the parameterization scheme developed by Pöschl et al. (2001) should be used with caution.

333 **2.2.1.4 Total BC aging rate**

We assume a linear combination of the condensation, coagulation, and chemical oxidation processes, following Croft et al. (2005) and Huang et al. (2013). The total BC aging rate can thus be expressed in the forms:

337
$$\left(\frac{dm_{BCPO}}{dt}\right)_{mic} = \left(\frac{dm_{BCPO}}{dt}\right)_{cond} + \left(\frac{dm_{BCPO}}{dt}\right)_{coag}$$
(16)

338
$$\left(\frac{dm_{BCPO}}{dt}\right)_{mic+chem} = \left(\frac{dm_{BCPO}}{dt}\right)_{mic} + \left(\frac{dm_{BCPO}}{dt}\right)_{chem}$$
(17)

where the subscripts *mic* and *chem* represent microphysical and chemical aging, respectively. Such linear combination may overestimate BC aging rate, because these processes likely compete with each other rather than occur independently (Croft et al., 2005). However, no observational evidence is currently available to quantify interactions among these processes.

344

345 **2.2.2 BC emissions**

346 We use a global anthropogenic BC emission inventory developed by R. Wang et al. (2014) (hereinafter the PKU-BC inventory), with an annual emission of 8.5 TgC for 2008 (see 347 Fig. S1). PKU-BC incorporates a recent global high-resolution $(0.1^{\circ} \times 0.1^{\circ})$ fuel 348 combustion dataset (Wang et al., 2013) that covers 64 types of combustion based on local 349 350 or national fuel statistics. The dataset significantly improves the spatial resolution of emission distribution for large countries. In addition, the inventory uses updated BC 351 emission factors based on up-to-date measurements, particularly for developing countries 352 (Wang et al., 2012a, b). The use of local and subnational fuel data and updated emission 353

factors results in 70% higher global anthropogenic BC emissions than previous 354 bottom-up inventories (4.9±0.4 TgC) (Dentener et al., 2006; Bond et al., 2007; Lamarque 355 356 et al., 2010; Granier et al., 2011) yet still 20% lower than top-down estimates (Bond et al., 357 2013). The PKU-BC emissions are 58% higher than the global anthropogenic emissions used in our previous studies (He et al., 2014a, b). R. Wang et al. (2014) found that using 358 PKU-BC reduces the bias in modeled surface BC concentrations by up to 25% in Asia, 359 360 Africa, and Europe. However, Bond et al. (2013) pointed out that current anthropogenic 361 BC emission estimates are associated with large uncertainties (more than a factor of 2 across different inventories). Based on a Monte Carlo estimation, R. Wang et al. (2014) 362 showed an uncertainty range (interquartile) of -40% to +70% for global annual 363 anthropogenic BC emissions in PKU-BC, where errors in emission factors dominate the 364 365 overall uncertainty.

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We use the Global Fire Emissions Database version 3 (GFED3) (van der Werf et al., 2010) for global biomass burning emissions (see Fig. S1), which now includes small fire emissions (Randerson et al., 2012). Carbon emissions in GFED3 increase by 35% globally when small fires are included. In this study, the GFED3 emissions with a 3-hour temporal resolution are used. The uncertainty in GFED3 is \geq 20% globally and highest in boreal regions and Equatorial Asia, primarily because of insufficient data on fuel load, emission factor, and burned area (van der Werf et al., 2010; Randerson et al., 2012).

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2.2.3 BC deposition

Aerosol dry deposition follows a standard resistance-in-series scheme (Wesely, 1989), which depends on local surface type and meteorological conditions, as implemented by Wang et al. (1998). Wang et al. (2011) further updated aerosol dry deposition velocity over snow- and ice-covered regions for improved BC simulations in GEOS-Chem. They applied a constant value of 0.03 cm s⁻¹, within the range (0.01–0.07 cm s⁻¹) employed in previous studies (Liu et al., 2011). In the GEOS-Chem simulations presented here, dry deposition accounts for ~20% of global BC deposition, consistent with the results (17-23%) from previous studies using GEOS-Chem (He et al., 2014a; Q. Q. Wang et al., 2014) and that (21.4±17%) from the AeroCom multi-model study (Textor et al., 2006).

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Liu et al. (2001) first described aerosol wet scavenging in GEOS-Chem. Wang et al. 386 387 (2011) updated in the model the below-cloud scavenging parameterization for fine and coarse aerosol modes by distinguishing between aerosol removals by snow and by rain. 388 Different in-cloud scavenging schemes have also been applied to cold and warm clouds, 389 with an improved areal fraction of model grids that experience precipitation (Wang et al., 390 2011). Following Q. Q. Wang et al. (2014), we further update in-cloud scavenging of 391 392 water-soluble aerosols by accounting for homogeneous and heterogeneous freezing nucleation in cold clouds. 393

394

395 2.2.4 Model simulations

396 To investigate the effects of the microphysics-based BC aging scheme, we first conduct two GEOS-Chem BC simulations with the fixed aging (e-folding time of 1.2 days) 397 scheme (FIX, Table 1; see also Eq. (1)) and the standard microphysics-based scheme 398 399 (MP_{STD}, Table 1; see also Eq. (16)). In addition, we conduct 11 sensitivity simulations for 400 the microphysics-based scheme to quantify the effects of chemical oxidation on BC aging and the uncertainty associated with aerosol size distribution and condensation threshold 401 microphysics-based scheme. Specifically, we combine 402 in the the standard 403 microphysics-based scheme with the Pöschl et al. (2001) parameterization for chemical oxidation to examine the effects of chemical oxidation on BC aging (MP_{chem}, Table 1; see 404 also Eq. (17)). We use geometric mean diameters of 30 nm and 90 nm for BCPO number 405 size distribution as lower and upper bounds ($D_{BCPO}30$ and $D_{BCPO}90$, Table 1), following 406 Bond et al. (2006), instead of 60 nm in the standard simulation (i.e., MP_{STD}). We use 407

geometric standard deviations of 1.4 and 2.0 for BCPO number size distribution as lower 408 and upper bounds (SD_{BCPO}1.4 and SD_{BCPO}2.0, Table 1), following Bond et al. (2006), 409 410 instead of 1.8 in the standard simulation. We vary the geometric mean diameters and standard deviations of all hydrophilic particles' number size distribution in the standard 411 simulations by $\pm 50\%$ (D_{PI}+50% and D_{PI}-50%, Table 1) and ± 0.2 (SD_{PI}+0.2 and SD_{PI}-0.2, 412 Table 1). We vary the condensation threshold (β in Eq. (7)) from 5% in the standard 413 414 simulation to 2.5% (BETA2.5, Table 1) and to 10% (BETA10, Table 1). Model results used for comparison with HIPPO observations are sampled along the HIPPO flight 415 416 tracks.

417

418 **3. Results and discussions**

419 **3.1 BC aging rate**

Figure 2 shows model simulated surface-layer and zonal distributions of annual mean 420 aging rate with the fixed (r_{fix} , calculated by Eq. 1) and microphysics-based (r_{mic} , 421 calculated by Eq. 16) aging schemes. The maximum rates are in the surface layer over 422 major BC source regions such as Eastern China, India, Europe, eastern United States, and 423 tropical Africa, ranging from less than 100 ngC $m^{-3} h^{-1}$ for the fixed scheme (Fig. 2a) to 424 100-500 ngC m⁻³ h⁻¹ for the microphysics-based scheme (Fig. 2b). r_{mic} is 2-6 times 425 higher than r_{fix} throughout the year over both continents and oceanic areas with heavy 426 427 marine traffic (Fig. S2). This is primarily because of the strong anthropogenic emissions of SO₂, NO_x, and NH₃ in major continents and along shipping corridors in the Northern 428 Hemisphere (Park et al., 2004) and biomass burning emissions of SOA precursors in 429 tropical continents (Guenther et al., 2006). These emissions lead to fast BC aging through 430 both condensation of soluble materials (i.e., H₂SO₄-HNO₃-NH₃ and SOA) and 431 coagulation with hydrophilic particles. 432

433

434 Figure 3 shows the probability density function (PDF) of simulated annual mean BC

e-folding aging time (τ in Eq. 1) over the globe. The first PDF peak around $\tau = 8$ h 435 represents the fast aging near source regions, while the second bump corresponds to $\tau =$ 436 437 ~ 1 day, which reflects the aging over rural areas and in the middle troposphere. The third 438 small bump is around $\tau = 100$ h, mainly representing the very slow aging in remote regions (e.g., Polar regions and the upper troposphere). The microphysics-based scheme 439 results in an e-folding BC aging time of 0.6-2 hours in summer and 1-3 hours in winter 440 441 when averaged within the boundary layer over major anthropogenic source regions 442 including Eastern China, eastern United States, and Europe. These are much faster than the timescale of 1–2 days typically assumed in many global models (Chung and Seinfeld, 443 2002; Koch et al., 2009; He et al., 2014a). Such fast aging for anthropogenic BC has been 444 reported in previous studies. For example, Riemer et al. (2004) used a size-resolved 445 446 aerosol model to show an e-folding BC aging time of ~2-hour during the day from 250 to 1500 m above source (urban) regions for both summer and winter. Jacobson (2010) also 447 reported a ~3-hour e-folding aging time for fossil-fuel BC based on a global size-resolved 448 aerosol model. Shen et al. (2014) found that the e-folding aging timescale of 449 450 anthropogenic BC from East Asia is several hours, based on constraints from the HIPPO observations. In addition, Akagi et al. (2012) observed that ~20% of BC ages within one 451 hour after emission in a biomass-burning plume over California in November 2009, 452 whereas our microphysics-based scheme shows a mean BC aging rate of 24% h⁻¹ within 453 the same region and time period. Moteki et al. (2007) measured a BC aging rate of 2.3% 454 h⁻¹ downwind of an urban area in Japan in March 2004. Though not a direct comparison, 455 our microphysics-based scheme results in a BC aging rate of 10% h⁻¹ averaged over the 456 457 same region for March 2009.

458

459 r_{mic} decreases by several orders of magnitude with increasing altitude and from 460 continents to remote oceans and the polar regions (Fig. 2). r_{fix} is a factor of 1.5–4 lower 461 than r_{mic} below 900 hPa in the tropics and middle latitudes, but more than 5 times higher in the polar regions, remote oceans, and above 800 hPa throughout the year (Figs. S2–S3). The annual mean r_{mic} is 50% higher than r_{fix} in the middle to lower troposphere but more than fivefold lower in the middle to upper troposphere (Table 2). Because the aging rates in remote regions are vanishingly small, the difference in global BC distribution between the two aging schemes is thus dominated by the difference in the significantly larger aging rates over the source regions (see Sect. 3.3).

468

Figures 4a and 4b show that condensation dominates (>70%) BC aging globally, 469 particularly in the middle to upper troposphere (500–200 hPa) and in the southern high 470 latitudes. In contrast, the contribution from coagulation is <15% throughout much of the 471 globe, with the exception of ~30% over Eastern China, India, and central Africa, where 472 473 relatively abundant hydrophilic aerosols are available. This is primarily because hydrophilic aerosols are substantially removed by wet scavenging during the transport to 474 remote regions, thus reducing the coagulation between hydrophobic BC and hydrophilic 475 particles. Figures 4c and 4d show that the H₂SO₄-HNO₃-NH₃ condensation dominates 476 477 (>70%) the total condensation in the northern low and middle latitudes throughout the troposphere and in the southern middle latitudes below 800 hPa. This is largely explained 478 by strong anthropogenic emissions of SO_2 and NO_x in the Northern Hemisphere (Park et 479 al., 2004) and global oceanic emissions of dimethylsulfide (DMS) (Lana et al., 2011). 480 481 The SOA condensation, however, contributes to more than 80% of the total condensation over tropical continents dominated by biomass burning emissions and the southern 482 extratropics above 700 hPa, where rather limited sulfuric acid is produced. 483

484

485 **3.2 Model evaluation with HIPPO observations**

Figure 5 shows the PDF of observed and simulated BC concentrations during the HIPPO measurements. Model results from the fixed aging scheme significantly overestimate BC concentrations, with a PDF peak at ~15 ng kg⁻¹ (corresponding to BC in the northern

extratropics) that is an order of magnitude higher than the observations. The 489 microphysics-based aging scheme substantially reduces the discrepancy between 490 modeled and observed PDFs, particularly at BC concentrations of <1 ng kg⁻¹ and 10–50 491 ng kg⁻¹. The remaining model bias at extremely low BC concentrations (<0.1 ng kg⁻¹), 492 corresponding to regions remote from combustion influence and highly influenced by 493 tropical convection, likely reflects the inefficient BC wet scavenging in the model (Q. Q. 494 Wang et al., 2014). Statistical analysis shows that compared with the fixed aging scheme, 495 496 the microphysics-based aging scheme reduces the percentages of modeled BC concentrations that are more than a factor of 10 and 4 higher/lower than the observations 497 from 28% to 16% and from 53% to 37%, respectively, with a better model-observation 498 correlation as well (Fig. 6). The median BC concentration (6.6 ng kg⁻¹) from the fixed 499 aging scheme is biased high by a factor of four compared with the observations (1.6 ng 500 kg^{-1}), while the microphysics-based median concentration (3.5 ng kg^{-1}) reduces the bias 501 by a factor of two. 502

503

Figure 7 shows the median vertical profiles of BC concentration in different latitude 504 bands from HIPPO observations and model simulations. The median in the northern 505 extratropics (20 $^{\circ}$ -90 N) is 5–10 ng kg⁻¹ near the surface and decreases to <1 ng kg⁻¹ 506 above 200 hPa, while the concentration is a factor of 2-10 lower in tropical regions and 507 508 Southern Hemisphere throughout the troposphere. This is because of strong BC emissions in the Northern Hemisphere and strong wet scavenging of BC during transport to the 509 tropics and Southern Hemisphere, particularly by tropical deep convection (Q. Q. Wang 510 511 et al., 2014). Model results using the fixed aging scheme generally capture the spatiotemporal pattern of BC median vertical profiles in the northern extratropics, but 512 overestimate the magnitude by a factor of ~5 on average (Figs. 7 and S4). The largest 513 model-observation absolute discrepancies (>20 ng kg⁻¹) are at 900-400 hPa during 514 non-winter seasons (Fig. S4). The microphysics-based aging scheme reduces that 515

discrepancy by 2–3 times across different altitudes in the northern extratropics, particularly at 900–300 hPa (Figs. 7 and S4). This is a result of the faster BC aging from the microphysics-based scheme over source regions in the Northern Hemisphere (Fig. 2), which increases the amount of hydrophilic BC removed by wet scavenging during the transport to the free troposphere.

521

In the tropics (20 S-20 N), the median BC concentration is generally less than 1 ng kg⁻¹, 522 523 with small variations across different altitudes and seasons (Figs. 7 and S4). Model results using the microphysics-based aging scheme reproduce the observed BC vertical 524 distributions in different seasons with discrepancies of <0.5 ng kg⁻¹, except for a 50% 525 underestimate in the lower troposphere (Figs. 7 and S5). This could be due to the 526 527 overestimate in GEOS-5 precipitation fields over the tropics (Molod et al., 2012). Compared with the microphysics-based aging scheme, model results from the fixed aging 528 scheme overestimate BC concentration by more than fivefold above 800 hPa in the 529 tropics, particularly in summer and fall (Fig. S5). This is primarily because of the slow 530 531 BC aging over tropical source regions in the fixed aging scheme (Fig. S2), resulting in insufficient BC wet removal and thus excessive BC transported to the free troposphere. 532

533

534 In the Southern Hemisphere $(20^{\circ}-70^{\circ}S)$, the microphysics-based aging scheme improves the modeled median BC vertical profiles by a factor of 2-4 throughout all altitudes and 535 seasons (Figs. 7 and S4) with the largest improvement at 600–200 hPa (Fig. S5), 536 compared with the fixed aging scheme. However, the microphysics-based model results 537 still overestimate the BC concentration above 300 hPa by ~3 times in the Southern 538 539 Hemisphere, probably due to inefficient in-cloud scavenging of BC in the model for ice clouds or mixed-phase clouds at such high altitudes. However, we note that the ability of 540 BC to be ice nuclei is largely uncertain. For example, Cozic et al. (2008) suggested that 541 BC particles can act as efficient ice nuclei based on observations in mixed-phase clouds, 542

whereas Friedman et al. (2011) showed that BC particles are unlikely to serve as icenuclei efficiently in cold clouds based on laboratory experiments.

545

546 Figure 8 shows the observed and model simulated latitudinal and seasonal BC column burden averaged along the HIPPO flight tracks. The lowest burden is in the tropics due to 547 strong wet scavenging by deep convection. It is two orders of magnitude lower than that 548 549 in the extratropical Northern Hemisphere. The burden in the Southern Hemisphere is 550 minimum in spring because of the wet season in the southern low latitudes (Wang and Ding, 2008), while the burden in the Northern Hemisphere is maximum during the same 551 period when the impact of Asian outflow on the Pacific peaks (Schwarz et al., 2013). 552 Model results using the fixed aging scheme capture the latitudinal and seasonal trend for 553 554 the observations, but significantly overestimate the magnitude with a mean positive bias of 194%, particularly in the Northern Hemisphere during summer and fall, as a result of 555 the model overestimate of BC vertical profiles in the Northern Hemisphere (Fig. 7). The 556 microphysics-based aging scheme reduces the model overestimate at most latitudes 557 throughout the year, with a mean positive bias of 60% and a correlation coefficient (r) of 558 0.85 with observations. The largest improvement from the microphysics-based scheme is 559 in the tropics (Fig. 8), except in January. 560

561

562 Therefore, the microphysics-based aging scheme substantially improves BC simulations pole-to-pole over the Pacific during HIPPO, both spatially and temporally, compared 563 with the fixed aging scheme. The remaining model-observation discrepancy, particularly 564 565 in the northern extratropics and the upper troposphere, is likely a result of the uncertainty associated with BC emissions, wet scavenging, and model meteorological fields (Molod 566 et al., 2012; Bond et al., 2013; Q. Q. Wang et al., 2014). We note that it is important to 567 quantify the contribution of these factors to the model-observation discrepancies, which 568 will be investigated in our future study. Additionally, the spatiotemporal variability of the 569

observed BC concentration is significantly large within each altitude and latitude band
(Fig. 7), suggesting a strong dependence of BC vertical profile on sampling location and
time during the HIPPO aircraft measurements.

573

574 **3.3 Global BC distribution and budget**

Figures 9a and 9b show the annual mean global BC column burden from model results 575 using the fixed and microphysics-based BC aging schemes. Both schemes result in a 576 similar spatial distribution of BC burden, with highest values over source regions and 577 lowest values in the southern hemispheric oceans. However, BC burden from the 578 microphysics-based scheme is much smaller than that from the fixed scheme globally, 579 with the major difference over source regions throughout the year (Fig. S6). This is 580 581 because the maximum enhancement of BC aging rate from the microphysics-based scheme is over source regions (Fig. 2), where there is a 20-60% increase of the 582 hydrophilic BC fraction in total BC particles relative to the fixed aging scheme (Fig. S7). 583 The faster hydrophobic-to-hydrophilic conversion leads to a stronger BC wet removal for 584 the microphysics-based aging scheme. As a result, the global annual mean BC load is 585 0.24 mg m⁻² in the microphysics-based scheme, consistent with the result (0.23 \pm 0.07 mg 586 m⁻²) from the AeroCom multi-model study (Schulz et al., 2006). It is also comparable to 587 the value (0.25 mg m^{-2}) reported by Jacobson (2010), which accounted for size-resolved 588 aerosol microphysical processes. Bond et al. (2013) reported a much higher BC load 589 (0.55 mg m⁻²) based on top-down estimated BC emissions that are 70% larger than the 590 emissions used in this study. The resulting global BC lifetime of 4.2 days from the 591 microphysics-based scheme is within the range (3.2–9.9 days) from previous studies 592 (Table 2). 593

594

595 Figures 9c and 9d show model simulated annual zonal mean BC concentration. The 596 concentration decreases substantially from the surface in the northern middle latitudes to

high altitudes and the polar regions, varying by three orders of magnitude. The 597 concentration from the microphysics-based aging scheme is smaller than that from the 598 fixed aging scheme at different latitudes and altitudes, with the largest difference (>400 599 600 ng m^{-3}) in the northern middle latitudes (Fig. S8). Although the microphysical aging rate is much lower than the rate from the fixed aging scheme in the middle and upper 601 troposphere and the polar regions (Figs. 2 and S3), the faster microphysical aging over 602 603 source regions dominates the aging effects on global BC distribution, resulting in a 604 substantial reduction of BC concentration globally. As a result, the fraction of global BC load above 5 km is 7.3% for the microphysics-based scheme, which is close to the lower 605 bound of the range (6.1–40%) from previous studies (e.g., Schulz et al., 2006; He et al., 606 2014b; Q. Q. Wang et al., 2014). The relatively low BC load at high altitudes in the 607 608 microphysics-based scheme has a significant implication for global BC radiative effects. 609 Samset et al. (2013) showed that more than 40% of global BC direct radiative forcing is contributed by BC particles above 5 km, whereas this estimate has large uncertainties 610 across various models. BC particles at high altitudes could also affect the formation and 611 612 distribution of cirrus clouds and thus BC indirect radiative effects (e.g., Liu et al., 2009).

613

614 **4. Uncertainty analysis**

615 **4.1 Heterogeneous chemical oxidation**

616 Figure 10 shows the annual mean contribution of chemical oxidation to the total BC aging rate. The contribution is 10–30% below 900 hPa over most regions at 60 S–60 N 617 latitudes, particularly at low latitudes and over source regions, due partly to the strong 618 619 condensation-coagulation in these areas. Additionally, the relatively high humidity and low ozone concentration also contribute to less efficient chemical oxidation aging in the 620 lower troposphere and the tropics, compared with the middle troposphere and the polar 621 regions (Fig. 10b). There is efficient chemical oxidation aging over the Arctic and 622 Antarctic in the middle troposphere, where water vapor is scarce and relatively abundant 623

ozone is available. This is consistent with the conclusion from Huang et al. (2013), which
used the same chemical oxidation scheme and showed the lowest chemical aging rate
over the tropics at lower altitudes and faster aging at higher altitudes including the polar
regions.

628

Compared with the standard microphysics-based aging scheme (MP_{STD}), incorporating 629 630 chemical oxidation (MP_{chem}) slightly (<5%) increases the total BC aging rate at 60 S-60 N latitudes in the lower troposphere but more than halves the aging rate over 631 the polar regions (Figs. 11 and 12). This is because faster BC aging over non-polar 632 regions reduces the amount of hydrophobic BC transported to remote areas. However, the 633 absolute aging rate over the polar regions is several orders of magnitude smaller than that 634 635 in non-polar regions. Chemical oxidation aging thus has a vanishingly small (<0.5%) impact on the global annual mean BC aging rate (Table 2). It leads to only a small (<1%) 636 reduction in BC column burden and zonal mean concentration globally (Figs. 13 and 14) 637 as well as global BC lifetime (Table 2). Croft et al. (2005) showed a comparably small 638 $(\sim 5\%)$ decrease in global BC burden and lifetime when the chemical oxidation process is 639 included, while Huang et al. (2013) found a ~10% decrease in BC burden and lifetime 640 with the incorporation of chemical oxidation aging. 641

642

643 **4.2 Size distribution of hydrophobic BC**

Reducing the geometric mean diameter of hydrophobic BC from 60 to 30 nm ($D_{BCPO}30$) and the standard deviation from 1.8 to 1.4 ($SD_{BCPO}1.4$) increases hydrophobic BC number concentration. As a result, more hydrophobic BC particles are available for condensation-coagulation aging, leading to a 5–50% increase of condensation-induced aging rate within 60 S–60 N below 900 hPa (Figs. S9 and S10). The enhancement of aging rate is stronger for $SD_{BCPO}1.4$ than for $D_{BCPO}30$, suggesting that the condensation-induced aging is more sensitive to the change in geometric standard deviation than geometric mean diameter. Interestingly, the largest enhancement for both simulations is over the oceans, while only less than 10% increase occurs in the source regions. This is probably because the BC aging rate over the oceans is much smaller than that over the source regions (Fig. 2), making it more sensitive to the change in the size distribution of hydrophobic BC.

656

657 $D_{BCPO}30$ increases the coagulation-induced aging rate by up to a factor of 2 within 658 60 S-60 N near the surface (Fig. S11), particularly in source regions, whereas SD_{BCPO} 1.4 only increases the coagulation-induced surface-layer aging rate along shipping corridors 659 over the oceans, with up to 50% decrease in the rest of non-polar regions. In the middle 660 to upper troposphere and the polar regions, both $D_{BCPO}30$ and $SD_{BCPO}1.4$ result in a more 661 than 50% decrease in coagulation-induced and condensation-induced BC aging rates 662 (Figs. S10 and S12), because of the reduction in the amount of hydrophobic BC 663 transported to remote regions. The change in total BC aging rate shows a very similar 664 spatial pattern with that of condensation (Figs. 11 and 12), due to the dominant role of 665 condensation-induced aging globally. However, because of the rather low BC aging rate 666 over the oceans and at high altitudes relative to that over source regions, the impact of 667 hydrophobic BC size distribution on global BC distribution is dominated by the change 668 of aging rate in source regions for both $D_{BCPO}30$ and $SD_{BCPO}1.4$, which results in less 669 670 than 5% reductions in BC column burden and zonal mean concentration globally, except a 10–20% reduction in the tropics at 600–200 hPa (Figs. 13 and 14). The resulting global 671 annual mean BC load and lifetime show negligible (<1%) increases (Table 2). 672

673

Increasing the geometric mean diameter of hydrophobic BC from 60 to 90 nm ($D_{BCPO}90$) and the standard deviation from 1.8 to 2.0 ($SD_{BCPO}2.0$) decreases hydrophobic BC number concentration. Both simulations exhibit opposite patterns of the change in BC aging rate compared with those from $D_{BCPO}30$ and $SD_{BCPO}1.4$ (Figs. S13–18), because of lower hydrophobic BC number concentrations in the former two. Nevertheless, similar to D_{BCPO}30 and SD_{BCPO}1.4, D_{BCPO}90 and SD_{BCPO}2.0 also result in less than 10% change in global BC column burden and zonal mean concentration (Figs. S19–20). We note that the observationally constrained accumulation mode BC mass size distributions for HIPPO have a geometric mean diameter of ~180 nm (Schwarz et al., 2010), which is the upper bound value used in this study for the geometric mean diameter of mass size distribution (~ 60–180 nm) converted from that of number size distribution (30–90 nm).

685

686 **4.3 Size distribution of hydrophilic aerosols**

Increasing the geometric mean diameters of all hydrophilic aerosols by 50% (D_{PI}+50%) 687 and the standard deviations by 0.2 (SD_{PI}+0.2) reduces the number concentration of 688 689 hydrophilic particles. This results in an enhanced condensation-induced aging rate due to 690 fewer hydrophilic particles competing for condensed soluble materials, but a reduced coagulation-induced aging rate due to fewer hydrophilic particles available for 691 coagulating with hydrophobic BC. Both simulations show up to 50% increase in the 692 693 annual mean condensation-induced BC aging rate at 60 S-60 N below 900 hPa (Figs. S9–10), particularly over the oceans where BC aging rate is relatively small. The faster 694 aging through condensation in the lower troposphere reduces the amount of hydrophobic 695 696 BC transported to higher altitudes and the polar regions, resulting in more than 50% 697 slower condensation-induced aging in remote regions. In contrast, the coagulation-induced aging rate decreases globally, by 20-50% at 60 S-60 N near the 698 surface and more than 50% in elsewhere (Figs. S11 and S12). The change of total BC 699 700 aging rate follows the spatial pattern of the condensation-induced aging rate (Figs. 11 and 12). The global annual mean BC aging rate is slightly higher (~1.5%) in both D_{PI} +50 and 701 SD_{PI} +0.2. Figures 11 and 12 show that increasing geometric mean diameters and standard 702 deviations of hydrophilic particles only reduces global BC column burden and zonal 703 mean concentration by less than 10% and BC lifetime by less than 1% (Table 2). 704

Decreasing the geometric mean diameters of hydrophilic aerosols by 50% (D_{PI}-50%) and 706 707 the standard deviations by 0.2 (SD_{PI}-0.2) increases the number concentration of 708 hydrophilic particles, resulting in an opposite spatial pattern of the change in BC aging 709 rate (Figs. S13–18) than that from D_{PI} +50 and SD_{PI} +0.2. Both condensation-induced and coagulation-induced aging rates are more sensitive to the decrease of geometric mean 710 711 diameters (D_{PI} -50%) than geometric standard deviations (SD_{PI} -0.2) globally. The 712 resulting BC concentration change is less than 10% over much of the globe, with a <2%increase in global BC load and lifetime (Table 2). 713

714

715 **4.4 Condensation threshold**

716 The condensation-induced BC aging rate is critically dependent on the condensation threshold β (see Eq. (9)), which represents the mass fraction of condensed soluble 717 materials on hydrophobic BC required for hydrophobic-to-hydrophilic BC conversion. 718 Reducing β by a factor of 2 (BETA2.5) increases the condensation-induced aging rate by 719 720 5–10% near source regions and by up to 50% over remote oceans at 60 S–60 N (Fig. S9), while the condensation-induced aging rate decreases by 20-50% in the middle and upper 721 troposphere and more than 80% over the polar regions (Fig. S10), because of the reduced 722 723 amount of hydrophobic BC transported to remote areas. The change in total BC aging 724 rate follows the pattern of the change in condensation-induced aging rate, with major increases at 60 S-60 N below 900 hPa (Figs. 11 and 12). This results in a decrease 725 (<20%) of BC concentration globally, particularly in the tropics at 600–200 hPa (Figs. 13 726 and 14). In contrast, doubling β (BETA10) shows the opposite spatial pattern of the 727 change in BC aging rate compared with BETA2.5 (Figs. S13 and S14), where the aging 728 rate decreases by up to 20% in the non-polar regions below 900 hPa but increases by up 729 to more than twofold at high altitudes and polar regions. The resulting global BC load 730 and lifetime, however, change by less than 2% for both BETA10 and BETA2.5 (Table 2). 731

There is strong spatial heterogeneity in the sensitivity of microphysical BC aging to 733 aerosol size distribution, condensation threshold, and chemical oxidation, with little 734 735 sensitivity over the source regions but rather large sensitivity in remote regions. However, the BC aging rate in remote areas is several orders of magnitude smaller than that over 736 source regions. As a result, the global BC column burden and zonal mean concentration 737 738 are only slightly affected by the change in the above-mentioned factors, with less than 5% 739 change over much of the globe (Figs. 13 and 14). Very small changes are also seen in global BC lifetime (Table 2). Nevertheless, using a uniform aerosol size distribution and 740 condensation threshold may not be realistic or representative, particularly on a regional 741 scale with complex atmospheric conditions. The two-/three-moment aerosol scheme (e.g., 742 743 Li et al., 2008), which predicts aerosol size distribution from simulated aerosol mass, 744 number, and/or surface area depending on atmospheric conditions, could be a potential improvement to represent and understand BC aging, interaction with cloud, and 745 deposition compared with the microphysical scheme developed in this study. 746

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732

748 **5.** Conclusions

We have developed and examined a microphysics-based BC aging scheme that explicitly 749 750 accounts for condensation and coagulation processes in GEOS-Chem global CTM. We 751 analyzed the difference in BC aging rate between the microphysics-based scheme and a fixed aging scheme with an e-folding time of 1.2 days, followed by a systematic 752 evaluation of BC simulations using HIPPO observations from 2009 to 2011. We further 753 754 analyzed the effects of the microphysics-based aging scheme on global BC distribution and lifetime. Finally, we quantified the uncertainty associated with aerosol size 755 distribution and condensation threshold in the microphysics-based aging scheme and the 756 impact of heterogeneous chemical oxidation on BC aging. 757

In the microphysics-based BC aging scheme, we converted aerosol mass to number 759 concentration by assuming lognormal aerosol size distributions. We computed the 760 condensation-induced aging rate from the condensation of soluble materials onto 761 762 hydrophobic BC, which converted hydrophobic to hydrophilic BC when the condensed mass reached the condensation threshold (β in Eq. (9)). We computed the 763 coagulation-induced aging rate from the coagulation between hydrophobic BC and 764 hydrophilic particles, assuming a hydrophobic-to-hydrophilic BC conversion upon 765 766 coagulating. This microphysics-based scheme thus incorporated an explicit microphysical representation of BC aging and retained the computational efficiency of the fixed aging 767 scheme. The microphysical aging scheme can also be applied in other CTMs. 768

769

770 The microphysical aging rate was a factor of 2–6 higher than that from the fixed aging 771 scheme in the lower troposphere over such source regions as East China, India, Europe, United States, tropical continents, and marine shipping corridors, because of strong 772 773 emissions of sulfate-nitrate and SOA precursors, which resulted in faster BC aging through condensation and coagulation. The microphysical aging rate is more than 774 fivefold lower than that from the fixed aging scheme in remote regions, where 775 condensation and coagulation are rather weak. We found that condensation dominated 776 777 (>70%) BC aging globally, particularly in the Southern Hemisphere and above 5 km, while the largest coagulation contribution (~30%) was over East China, India, and 778 Central Africa, primarily because the hydrophilic aerosols required for coagulation were 779 substantially removed by wet scavenging during transport to remote regions. 780

781

Compared with the fixed aging scheme, the microphysical scheme substantially reduced the discrepancy between modeled and observed probability density functions of BC concentrations during HIPPO, particularly at BC concentrations of $< 1 \text{ ng kg}^{-1}$ and 10–50 ng kg⁻¹. Model results using the fixed aging scheme overestimated BC median vertical

profiles in the northern extratropics by ~ 5 times on average, while the microphysical 786 scheme improved the modeled BC vertical profiles by a factor of 2-3 throughout the 787 788 troposphere, particularly at 900–300 hPa. Model results using the microphysical aging 789 scheme reproduced the observed BC vertical distribution in the tropics. In the Southern Hemisphere, the microphysical aging scheme reduced the model bias in BC vertical 790 profiles by a factor of 2–4, with largest improvements at 600–200 hPa, compared with the 791 792 fixed aging scheme. The model bias in latitudinal BC column burden along the HIPPO 793 flight tracks was reduced from +194% for the fixed aging scheme to +60% for the microphysics-based scheme, with largest improvements in the tropics. The remaining 794 model-observation discrepancy for the microphysics-based simulation was likely due to 795 the uncertainty associated with BC emissions, wet scavenging, and meteorological fields 796 797 in the model. We note that it is also very important to evaluate BC simulations from other 798 perspectives in addition to atmospheric concentration, such as aerosol optical depth, a subject requiring further investigation. 799

800

801 We found that the faster BC aging over the source regions from the microphysics-based scheme dominated the aging effects on global BC distribution, resulting in a much lower 802 BC column burden and zonal mean concentration globally, compared with the fixed 803 804 aging scheme. The global annual mean BC lifetime was 4.2 days in the microphysics-based scheme, where wet scavenging accounts for 80.3% of global BC 805 deposition. The resulting global BC burden was 0.25 mg m⁻², with 7.3% of the burden 806 above 5 km. The relatively low BC load at high altitudes had important implications on 807 808 the estimate of global BC radiative effects.

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Furthermore, we found that BC aging rate was insensitive (<10% change) to aerosol size distribution, condensation threshold, and chemical oxidation over source regions, while it was the opposite (more than twofold change) in the polar regions and at high altitudes. However, the BC aging rate in remote regions was orders of magnitude lower than that in source regions. Thus, the global BC burden and lifetime showed little sensitivity (<5% change) to the above-mentioned three factors. Nevertheless, assuming global uniform aerosol size distribution and condensation threshold may not be representative or accurate, particularly for regions with complex atmospheric conditions. Further improvements require the incorporation of a dynamic size-resolved microphysical aging scheme.

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Simulations		Hydro	phobic ^a	Hydro	Condensation			
		D_p (nm)	σ_p	D_p (nm)	σ_p	$(\beta)^{b}$		
Fixed aging (FIX)		<i>e</i> -folding BC aging time $\tau_{\text{fix}} = 1.2$ days						
Standard	Microphysics-based (MP _{STD}) ^c	$D_{BC} = 60$ $D_{OC} = 60$	$\sigma_{BC} = 1.8$ $\sigma_{OC} = 1.8$	$D_{BC} = 150$ $D_{OC} = 150$ $D_{sulf} = 150$ $D_{fss} = 200$ $D_{css} = 800$	$\sigma_{BC} = 1.8$ $\sigma_{OC} = 1.8$ $\sigma_{sulf} = 1.6$ $\sigma_{fss} = 1.5$ $\sigma_{css} = 1.8$	5% of hydrophobic BC mass		
	D _{BCPO} 30	Hydrophobic $D_{BC} = 30 \text{ nm}^d$						
	D _{BCPO} 90							
	$SD_{BCPO}1.4$	Hydrophobic $\sigma_{BC} = 1.4^{d}$						
suo	$SD_{BCPO}2.0$	Hydrophobic $\sigma_{BC} = 2.0^{d}$						
ulati	D _{PI} +50%	All hydrophilic particles D_p increased by 50%						
sim	D _{PI} -50%	All hydrophilic particles D_p decreased by 50%						
ivity	SD _{PI} +0.2	All hydrophilic particles σ_p increased by 0.2						
ensit	SD _{PI} -0.2	All hydrophilic particles σ_p decreased by 0.2						
S	BETA2.5	eta=2.5%						
	BETA10			$\beta = 10\%$)			
	MP _{chem}	Stan	oxidation					

Table 1. Key aging parameters used in GEOS-Chem simulations of BC.

1095 ${}^{a}D_{p}$ and σ_{p} are geometric mean diameter and geometric standard deviation for particle number size 1096 distribution, respectively.

1097 ^bCondensation threshold (β in Eq. (7)) represents the mass fraction of condensed soluble materials on 1098 hydrophobic BC required for hydrophobic-to-hydrophilic BC conversion. A value of 5% is used in the 1099 standard microphysics-based scheme following Riemer et al. (2004).

1100 ^cGeometric mean diameters (D_p) and standard deviations (σ_p) are followling Dentener et al. (2006) and

1101 Yu and Luo (2009) for hydrophobic and hydrophilic BC and OC, and the Global Aerosol Dataset

(GADS) (Koepke et al., 1997) for sulfate-nitrate-ammonia ("sulf"), fine-mode sea salt ("fss"), and
coarse-mode sea salt ("css"). SOA is treated as hydrophilic OC.

^dLower and upper bounds of geometric mean diameters (30 nm and 90 nm) and standard deviations

1105 (1.4 and 2.0) for hydrophobic BC are following Bond et al. (2006).

References		BC aging rate (ngC m ⁻³ h ⁻¹)				T 10 /		Loading	Wet	
		< 5 km	> 5 km	mean	Emissions (Tg yr ⁻¹) ^c	(day)	(mg m ⁻²)	above 5 km (%)	deposition (%)	Hydrophilic fraction (%)
	FIX	0.378	0.005	0.221	10.8	13.7	0.795	26.2	78.9	85.8
	MP _{STD}	0.560	< 0.001	0.324		4.21	0.244	7.32	80.3	98.8
	MP _{chem}	0.561		0.325		4.21	0.244	7.31	80.3	98.9
	D _{BCPO} 30	0.570		0.330		4.18	0.242	7.16	80.2	99.3
	D _{BCPO} 90	0.551		0.319		4.25	0.246	7.53	80.3	98.1
	SD _{BCPO} 1.4	0.571		0.331		4.18	0.242	7.14	80.2	99.4
This	SD _{BCPO} 2.0	0.550		0.319		4.25	0.246	7.55	80.3	98.1
study	D _{PI} +50%	0.568		0.329		4.19	0.243	7.19	80.3	99.2
	D _{PI} -50%	0.548		0.317		4.27	0.247	7.68	80.2	97.8
	SD _{PI} +0.2	0.567		0.328		4.19	0.243	7.20	80.3	99.2
	SD _{PI} -0.2	0.552		0.320		4.24	0.246	7.50	80.3	98.2
	BETA2.5	0.569		0.330		4.19	0.242	7.17	80.3	99.3
	BETA10	0.547		0.317		4.27	0.247	7.65	80.2	97.7
He et al. (2014a, b)		Fixed aging ($\tau = 1.2$ days)			8.1	6.6	0.29	18	83	-
Q. Q. Wang et al. (2014)		Fixed aging ($\tau = 1.2$ days)			6.5	4.2	0.15	8.7	77	-
Bond et a	al. (2013) ^a	Model ensemble mean			17	6.1	0.55	-	-	-
Jacobson (2012)		Aging microphysics			9.3	3.2	0.18	-	94	-
Chung et al. (2012)		Fixed aging			6.3	5.5	0.19	-	-	-
Jacobson (2010)		Aging microphysics			4.7	9.9	0.25	-	92	-
Schulz et al. (2006) ^b		Model ensemble mean		6.3	6.8±1.8	0.23±0.07	21±11	-	-	

1107 **Table 2**. GEOS-Chem simulated global annual mean BC budget.

^aBased on AeroCom Phase I simulations after scaling to match AERONET BC absorption optical
depth (AAOD).

1110 ^bMean and standard deviations for eight models from AeroCom Phase I simulations. All AeroCom

1111 models use the same emissions.

1112 ^cGlobal total BC emissions including anthropogenic and biomass burning sources.



Figure 1. HIPPO aircraft flight tracks in January 2009 (black), October – November
2009 (blue), March – April 2010 (pink), June – July 2011 (red), and August – September
2011 (green).



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Figure 2. GEOS-Chem simulated annual mean BC aging rates $(ngC m^{-3} h^{-1})$ in the surface layer (top panels) and averaged zonally (lower panels) from a fixed BC aging scheme (left panels) and a microphysics-based aging scheme (right panels). See text for details. Model results are for 2009.





Figure 3. Probability density function of simulated annual mean BC e-folding aging time
(h) over the globe for a microphysics-based scheme (solid line) and a fixed aging scheme
(dashed line). Also shown is 95% uncertainty range (in grey) of the microphysics-based
aging time estimated from a Monte Carlo method.



1138Figure 4. GEOS-Chem simulated annual mean contribution of condensation to total BC1139aging rate (sum of condensation and coagulation) averaged within 0-1 km above the1140surface (top left) and zonally (top right) and annual mean contribution of1141 H_2SO_4 -HNO₃-NH₃ condensation to the BC aging rate through total condensation (sum of1142 H_2SO_4 -HNO₃-NH₃ and SOA condensation) averaged within 0-1 km above the surface1143(lower left) and zonally (lower right). Model results are for 2009.



Figure 5. Probability density functions of HIPPO observed (black) and GEOS-Chem simulated BC concentrations. Model results using a fixed BC aging (blue) and a microphysics-based aging (red) are shown. The 95% uncertainty range of the HIPPO observations, estimated from a Monte Carlo method, is shown (in grey). Also shown (in light red) is the range of model results from microphysics-based sensitivity simulations (see Table 1 and text for details). Dashed lines show the median of observations (1.56 ng kg⁻¹, black), fixed aging (6.55 ng kg⁻¹, blue), and microphysics-based aging (3.52 ng kg⁻¹, red). About 5% of the observed BC concentrations are below 0.01 ng kg⁻¹.





Figure 6. HIPPO observed and GEOS-Chem simulated BC concentrations sampled along the HIPPO flight tracks (HIPPO 1: pink, HIPPO 2: red, HIPPO 3: blue, HIPPO 4: orange, HIPPO 5: green; see Fig. 1), with model results using a fixed BC aging scheme (left panel) and a microphysics-based BC aging scheme (right panel). Also shown are the 1:1 and 1:10 (or 10:1) ratio lines, percentages of data points outside 1:10 (or 10:1) and 1:2 (or 2:1) ratio lines, and slopes and correlation coefficients (*r*) of the regression lines between model results and observations.



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1171 **Figure 7.** Median vertical profiles of HIPPO observed (black) and GEOS-Chem 1172 simulated BC concentrations at different latitudes. Results are averaged over 50 hPa 1173 altitude bins. Model results using a fixed BC aging (blue) and a microphysics-based 1174 aging (red) are shown. Also shown are the 1- σ uncertainties of observations (grey) and 1175 model results (light red) from microphysics-based sensitivity simulations (see Table 1 and 1176 text for details).





Figure 8. Latitudinal distributions of HIPPO observed (black) and GEOS-Chem simulated BC column burden (μ g m⁻²) in different seasons. The column burden is computed by integrating vertical profiles from the surface to 250 hPa (~10 km) in 10° latitude bins. Model results using a fixed BC aging (blue) and a microphysics-based aging (red) are shown. Also shown are the 1- σ uncertainties of observations (grey) and model results (light red) from microphysics-based sensitivity simulations (see Table 1 and text for details).

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Figure 9. GEOS-Chem simulated annual mean BC column burden (mg m⁻²) (top panels) and zonal mean BC concentrations (ng m⁻³) (lower panels) using a fixed BC aging scheme (left panels) and a microphysics-based aging scheme (right panels). Model results are for 2009.



Figure 10. GEOS-Chem simulated annual mean contribution of chemical oxidation to the
 total BC aging rate (sum of condensation, coagulation, and chemical oxidation) averaged

1202 within 0-1 km above the surface (left) and zonally (right). Model results are for 2009.



Figure 11. GEOS-Chem simulated ratios of annual mean BC aging rate from six microphysics-based simulations (Table 1) to that from the standard microphysics-based simulation in the surface layer. See Table 1 and text for details. Model results are for 2009.



Figure 12. Same as Fig. 11, but for zonal mean BC aging rate.



Figure 13. Same as Fig. 11, but for BC column burden.



1225 Figure 14. Same as Fig. 12, but for zonal mean BC concentrations.