

1 Summertime nitrate aerosol in the upper troposphere and lower stratosphere
2 over the Tibetan Plateau and the South Asian summer monsoon region

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26 **Abstract**

27 We use the global three-dimensional Goddard Earth Observing System
28 chemical transport model (GEOS-Chem) to examine the contribution of nitrate
29 aerosol to aerosol concentrations in the upper troposphere and lower
30 stratosphere (UTLS) over the Tibetan Plateau and the South Asian summer
31 monsoon (TP/SASM) region during summertime of year 2005. Simulated
32 surface-layer aerosol concentrations are compared with ground-based
33 observations, and simulated aerosols in the UTLS are evaluated by using the
34 Stratospheric Aerosol and Gas Experiment II satellite data. Simulations show
35 elevated aerosol concentrations of sulfate, nitrate, ammonium, black carbon,
36 organic carbon, and PM_{2.5} (particles with diameter equal or less than 2.5 μm,
37 defined as the sum of sulfate, nitrate, ammonium, black carbon, and organic
38 carbon aerosols in this study) in the UTLS over the TP/SASM region
39 throughout the summer. Nitrate aerosol is simulated to be of secondary
40 importance near the surface but the most dominant aerosol species in the
41 UTLS over the studied region. Averaged over summertime and over the
42 TP/SASM region, C_{NIT} (the ratio of nitrate concentration to PM_{2.5} concentration)
43 values are 5–35% at the surface, 25–50% at 200 hPa, and could exceed 60%
44 at 100 hPa. The mechanisms for the accumulation of nitrate in the UTLS over
45 the TP/SASM region include vertical transport and the gas-to-aerosol
46 conversion of HNO₃ to form nitrate. The high relative humidity and low
47 temperature associated with the deep convection over the TP/SASM region
48 are favorable for the gas-to-aerosol conversion of HNO₃.

49 **1 Introduction**

50 Aerosols in the upper troposphere and lower stratosphere (UTLS) have much
51 longer residence times than those in the lower troposphere, influencing
52 atmospheric chemistry and the Earth's climate with large spatial and temporal
53 coverage (Rasch et al., 2008). Aerosols in the UTLS influence the
54 concentrations of chemical species via changes in photolysis rates and
55 heterogeneous reactions (Pitari et al., 2014). For example, heterogeneous
56 reactions on sulfate aerosol can perturb the chemical partitioning in the lower
57 stratosphere, leading to significant O₃ depletion through enhanced chlorine,
58 bromine, and odd-hydrogen catalytic cycle (Zhao et al., 1997; Considine et al.,
59 2001; Talukdar et al., 2012; Tang et al., 2014; Pitari et al., 2014). Aerosols in
60 the UTLS also influence climate by altering properties of cirrus clouds via
61 homogeneous or heterogeneous ice nucleation (Li et al., 2005; Liu et al., 2009;
62 Yin et al., 2012; Fadnavis et al., 2013). Injection of aerosols into the UTLS has
63 been reported to induce complex responses in circulation, temperature, and
64 water vapor (Liu et al., 2009; Wu et al., 2011; Su et al., 2011; Fadnavis et al.,
65 2013).

66 Aerosols over the Tibetan Plateau (TP) and the Asian summer monsoon
67 region are especially important. The TP is surrounded by countries with large
68 anthropogenic emissions (Li et al., 2005; Lau et al., 2006). Aerosols from India,
69 Southeast Asia, and southern China can be transported to the TP by prevailing
70 winds in the premonsoon and monsoon seasons (Lawrence and Lelieveld,
71 2010; Xia et al., 2011). Observational and modeling studies have shown that
72 persistent maxima of atmospheric constituents, such as water vapor
73 (Gettelman et al., 2004; Randel and Park, 2006; Park et al., 2007), CO (Kar et

74 al., 2004; Li et al., 2005; Park et al., 2007, 2008, 2009), CH₄ (M. Park et al.,
75 2004; Xiong et al., 2009), NO_x (M. Park et al., 2004), HCN (Park et al., 2008;
76 Randel et al., 2010), C₂H₆ and C₂H₂ (Park et al., 2008), exist in the UTLS
77 above the TP and the South Asian summer monsoon (SASM) region because
78 of the deep convection during boreal summer. Satellite observations
79 suggested that the convection associated with the SASM is a vital pathway to
80 transport air mass from the lower troposphere into the stratosphere (Chen et
81 al., 2006; Randel and Park, 2006; Randel et al., 2010; Bian et al., 2011a). The
82 heating associated with the persistent deep convection during summertime
83 leads to the formation of the Tibetan anticyclone in the UTLS, which acts to
84 isolate air within the anticyclone and traps the uplifted pollutants at that altitude
85 (Park et al., 2007; Vernier et al., 2011; Bourgeois et al., 2012; Fadnavis et al.,
86 2013; He et al., 2014). The stratosphere-troposphere exchange (STE) over the
87 TP contributes largely to the global STE (Chen et al., 2006).

88 Previous studies have reported that aerosols exist in the UTLS over the
89 TP/SASM region. Kim et al. (2003) carried out optical measurements with a
90 ground-based lidar in Lhasa from August to October of 1999, and found an
91 enhancement in aerosol concentration near the local tropopause with
92 scattering ratio (SR, the ratio of aerosol plus molecular backscatter to
93 molecular backscatter alone) of 1.1–1.2. Tobo et al. (2007) reported an
94 enhancement of sub-micron aerosols (effective radius $r = 0.15\text{--}0.6\ \mu\text{m}$) near
95 the summertime tropopause (about 130 to 70 hPa), on the basis of in situ
96 balloon measurements from an Optical Particle Counter at the same location in
97 August of 1999. Vernier et al. (2009) examined satellite measurements from
98 the Cloud-Aerosol Lidar with Orthogonal Polarization (CALIOP) onboard

99 Cloud-Aerosol Lidar and Infrared Pathfinder Satellite Observation (CALIPSO)
100 and reported the presence of small depolarizing particles with high SR values
101 (about 1.20 at 532 nm) at 16–17 km altitude over South Asia in July and
102 August of 2007 and 2008. Bourgeois et al. (2012) found that an aerosol layer
103 existed at 16–18 km altitude over the Asian continent and Indian Ocean
104 (20°S–30°N, 5–105°E) on the basis of the CALIOP observations. Recently, He
105 et al. (2014) examined the vertical profiles of aerosol extinction coefficients
106 measured with a Micro Pulse Lidar at Naqu, a meteorological station located in
107 the central part of the TP, and also showed a maximum in aerosol extinction
108 coefficient ($\sim 2.10^{-3} \text{ km}^{-1}$) in the UTLS (18–19 km) during the summer of 2011.

109 A number of previous studies have attempted to understand the chemical
110 composition of aerosols in the UTLS. Froyd et al. (2009) measured aerosol
111 composition with the National Oceanic and Atmospheric Administration (NOAA)
112 single-particle mass spectrometer aboard the National Aeronautics and Space
113 Administration (NASA) WB-57 high altitude aircraft platform, and reported that
114 particles in the tropical tropopause layer were rich in nitrogen. Vernier et al.
115 (2011) suggested that aerosol layer at the tropopause of Asia could be sulfur
116 and/or organics, considering that Asian pollutants consisted of black carbon,
117 organic carbon, SO₂, and NO_x (Park et al., 2009; Randel et al., 2010). Weigel
118 et al. (2011) analyzed the volatility of aerosols obtained from in situ airborne
119 measurements and reported that about 75–90 % of the particles in the tropical
120 tropopause layer were volatile, but this study did not give any detailed
121 analyses of chemical composition of aerosols. Bourgeois et al. (2012) showed,
122 by using the ECHAM5.5-HAM2 model, that sulfate, water, and OC contributed,
123 respectively, 53%, 29%, and 11% to aerosol extinction in the vicinity of the

124 tropical tropopause layer. The ECHAM5.5-HAM2 model used by Bourgeois et
125 al. (2012) simulated all major aerosol species in the atmosphere except for
126 nitrate.

127 Few previous studies have examined nitrate aerosol in the UTLS, although
128 nitrate is expected to be important for the following reasons. First, emissions of
129 precursors of nitrate, such as NO_x and NH_3 , are high over India, Southeast
130 Asia, and China (Streets et al., 2003; Datta et al., 2012; Huang et al., 2012).
131 Second, simulated nitrate concentrations are high over those regions (Liao and
132 Seinfeld, 2005; Mu and Liao, 2014; Lou et al., 2014). Third, measured
133 concentrations of nitrate are comparable to or larger than those of sulfate at
134 rural and urban sites in the SASM region. Shrestha et al. (2000) carried out
135 measurements of aerosols at Phortse, Nepal, during September
136 1996–November 1997, and showed that the average concentration of nitrate
137 during the monsoon season (June–September) was $0.34 \mu\text{g m}^{-3}$, higher than
138 that of sulfate ($0.17 \mu\text{g m}^{-3}$). Decesari et al. (2010) reported, on the basis of
139 measurements at the Nepal Climate Observatory–Pyramid from 2006 to 2008,
140 that the concentrations of nitrate and sulfate were $0.37 \mu\text{g m}^{-3}$ and $0.50 \mu\text{g m}^{-3}$,
141 respectively, during the monsoon season. Chatterjee et al. (2010) measured
142 aerosols at a high altitude station in northeastern Himalayas during
143 January–December 2005. They found that the average concentrations of
144 fine-mode nitrate and sulfate were $3.31 \pm 2.25 \mu\text{g m}^{-3}$ and $3.80 \pm 2.9 \mu\text{g m}^{-3}$,
145 respectively. At Lahore, an urban site in Pakistan, the observed daytime nitrate
146 concentration of $21.8 \mu\text{g m}^{-3}$ was also higher than sulfate concentration of 12.6
147 $\mu\text{g m}^{-3}$ (Lodhi et al., 2009), as the observations were averaged over November
148 2005 to March 2006. Fourth, the low temperatures in the UTLS would favor

149 nitrate formation (Seinfeld and Pandis, 2006). Therefore, it is of interest to take
150 nitrate aerosol into consideration when we examine aerosols in the UTLS.

151 In this work we simulate nitrate aerosol and its contribution to aerosol
152 concentrations in the UTLS over the TP (70–105°E, 25–40°N) and the SASM
153 region (70–105°E, 10–25°N) by using the global chemical transport model
154 GEOS-Chem driven by the assimilated meteorological fields. These regions of
155 interest are shown in Fig. 1. Simulated surface-layer aerosol concentrations
156 are compared with ground-based observations, and simulated aerosols in the
157 UTLS are evaluated by using the Stratospheric Aerosol and Gas Experiment
158 II (SAGE II) satellite data. Section 2 is a brief description of the GEOS-Chem
159 model and numerical experiment. Section 3 presents the simulation and
160 evaluation of distributions and concentrations of HNO₃ and O₃ to show model's
161 capability in simulating the NO_x-O₃-HNO₃ cycle over the studied regions.
162 Section 4 shows simulated aerosols and Section 5 presents the simulated
163 contribution of nitrate to aerosol concentrations in the UTLS over the TP and
164 the SASM region. Section 6 discusses the mechanisms for high concentrations
165 of nitrate in the UTLS. Section 7 discusses the impacts of uncertainties in
166 surface-layer aerosol concentrations on simulated nitrate in the UTLS.

167

168 **2 Model description and numerical experiment**

169 **2.1 GEOS-Chem model**

170 We simulate gas-phase species and aerosols using the global chemical
171 transport model GEOS-Chem (version 9-01-03,
172 <http://acmg.seas.harvard.edu/geos/index.html>) driven by the GEOS-5
173 assimilated meteorological fields from the Goddard Earth Observing System of

174 the NASA Global Modeling and Assimilation Office. The version of the model
175 used here has a horizontal resolution of 2° latitude by 2.5° longitude and 47
176 vertical layers extending from the surface to 0.01 hPa. Over the TP and the
177 SASM region, the model has about 34 layers in the troposphere and 12 layers
178 in the stratosphere.

179 The GEOS-Chem model has a fully coupled treatment of tropospheric
180 NO_x-CO-hydrocarbon-aerosol chemistry and aerosols including sulfate (SO₄²⁻),
181 nitrate (NO₃⁻), ammonium (NH₄⁺), organic carbon (OC), black carbon (BC) (R. J.
182 Park et al., 2003; 2004; Pye et al., 2009), mineral dust (Fairlie et al., 2007), and
183 sea salt (Alexander et al., 2005; Jaeglé et al., 2011). Anthropogenic aerosols
184 are treated as bulk mass concentrations (particles of SO₄²⁻, NO₃⁻, NH₄⁺, BC,
185 and OC are not size-resolved). Sea Salt mass is simulated for two size bins
186 (0.1–0.5 and 0.5–8 μm) and mineral dust is simulated for four size bins
187 (0.1–1.0, 1.0–1.8, 1.8–3.0, and 3.0–6.0 μm). Both BC and OC consist of
188 hydrophilic and hydrophobic fractions in the model. It is assumed that 80% of
189 BC and 50% of OC emitted from all primary sources are hydrophobic (Cooke
190 et al., 1999; Chin et al., 2002; Chung and Seinfeld, 2002), which become
191 hydrophilic with an e-folding time of 1.2 days following Cooke et al. (1999) and
192 Chin et al. (2002). All secondary OC is assumed to be hydrophilic. Hydrophilic
193 fractions of both BC and OC aerosols are assumed to be fully soluble.

194 The gas-aerosol partitioning of nitric acid and ammonium is calculated
195 using the ISORROPIA II thermodynamic equilibrium module (Fountoukis and
196 Nenes, 2007). In the version of the GEOS-Chem model used in this work, ions
197 considered in ISORROPIA II include H⁺/Na⁺/NH₄⁺/Cl⁻/SO₄²⁻/HSO₄⁻/NO₃⁻/OH⁻. The
198 two-way coupling between aerosols and gas phase chemistry provides

199 consistent chemical fields for aerosol simulation and aerosol mass for
200 heterogeneous processes and calculations of gas-phase photolysis rates.
201 Heterogeneous reactions include hydrolysis of N_2O_5 (Evans and Jacob, 2005),
202 irreversible absorption of NO_3 and NO_2 on wet aerosols (Jacob, 2000), and the
203 uptake of HO_2 by aerosols (Liao and Seinfeld, 2005; Thornton et al., 2008).
204 Aerosol species are treated as an external mixture in the calculation of aerosol
205 optical properties.

206 With respect to chemistry in the stratosphere, stratospheric O_3
207 concentrations are calculated using the linearized parameterization scheme
208 (McLinden et al., 2000). The monthly mean production rates and loss
209 frequencies of other stratospheric species (including long-lived species such
210 as CFCs and N_2O) use those from NASA Global Modeling Initiative (GMI)
211 Combo simulations (Duncan et al., 2007; Considine et al., 2008; Murray et al.,
212 2012).

213 Convective transport in GEOS-Chem mimics that in the parent GEOS
214 general circulation model (GCM) (Hack, 1994; Zhang and McFarlane, 1995),
215 which accounts for updraft, downdraft, and entrainment mass fluxes for deep
216 and shallow convection (Wu et al., 2007). The aerosol wet deposition scheme
217 in the GEOS-Chem follows that of Liu et al. (2001). For the scavenging of
218 aerosols, SO_4^{2-} , NO_3^- , NH_4^+ , and hydrophilic OC and hydrophilic BC aerosols
219 are assumed to be fully soluble. Dry deposition follows the standard
220 resistance-in-series model of Wesely (1989).

221 Global emissions of aerosols and their precursors in the GEOS-Chem
222 follow R. J. Park et al. (2003, 2004), with anthropogenic emissions of NO_x , CO,
223 SO_2 , and non-methane volatile organic compounds (NMVOC) in Asia

224 overwritten by David Streets' 2006 emission inventory
225 (<http://mic.greenresource.cn/intex-b2006>). Emissions of NH₃ in Asia are taken
226 from Streets et al. (2003). Since NH₃ emissions in China showed large
227 uncertainties in previous studies (Streets et al., 2003; Kim et al., 2006; Y.
228 Zhang et al., 2010; Huang et al., 2011, 2012), we use the most recent estimate
229 of NH₃ emissions in China by Huang et al. (2012), which is 9.8 Tg yr⁻¹, instead
230 of 13.5 Tg yr⁻¹ from Streets et al. (2003). Table 1 summarizes the annual
231 emissions of NO_x, SO₂, NH₃, OC, and BC in Asia domain (60–155°E,
232 10–55°N).

233 Natural NO_x emissions from lightning are calculated using the scheme
234 described by Sauvage et al. (2007) and Murray et al. (2012), and those from
235 soil are simulated following Wang et al. (1998). Natural NH₃ emissions from
236 soil, vegetation, and the oceans are taken from the Global Emissions Inventory
237 Activity inventory (Bouwman et al., 1997). Biomass burning emissions are from
238 the monthly Global Fire Emissions Database (GFED v3) driven by satellite
239 observations of fire activity (van der Werf et al., 2010). Biogenic VOC (volatile
240 organic compounds) emissions are calculated from the Model of Emissions of
241 Gases and Aerosols from Nature (Guenther et al., 2006).

242 The monthly variations of emissions of SO₂ and NO_x follow Wang et al.
243 (2013) and those of BC and OC follow Lou et al. (2014). The monthly scaling
244 factors for NH₃ emissions follow the global inventory compiled by Marcel
245 Meinders and Lex Bouwman (Fisher et al., 2011). Monthly variations of
246 emissions (anthropogenic plus natural emissions) of NO_x, SO₂, NH₃, OC, and
247 BC over Asia are displayed in Fig. 2. The emissions of NH₃ are the highest in
248 June as a result of the agriculture practice and high temperatures (Wang et al.,

249 2013).

250 **2.2 Numerical experiment**

251 To examine the contribution of nitrate to aerosol concentrations in the UTLS
252 over the TP/SASM region, we simulate aerosol concentrations by using the
253 emissions of and meteorological fields of year 2005. Year 2005 is chosen so
254 that we can use the observational datasets for this year from SAGE II and MLS,
255 as described in Sects. 3 and 4. Following Rasch et al. (2008), we perform a
256 10-year spin-up run to generate the initial conditions (to allow the stratospheric
257 species to reach quasi-steady state conditions). We would consider that the
258 tropospheric simulation can be representative of year 2005 but stratosphere
259 simulation should represent a multi-year average, because the production
260 rates and loss frequencies in the stratosphere are the averages over years of
261 2004–2010 ([http://wiki.seas.harvard.edu/geos-chem/index.php/Stratospheric_](http://wiki.seas.harvard.edu/geos-chem/index.php/Stratospheric_chemistry)
262 [chemistry](http://wiki.seas.harvard.edu/geos-chem/index.php/Stratospheric_chemistry)).

263

264 **3 Simulated concentrations of HNO₃ and O₃ and model evaluation**

265 Nitrate aerosol forms when nitric acid (HNO₃) reacts with alkaline gases (for
266 example, ammonia) in the atmosphere (Seinfeld and Pandis, 2006). HNO₃, as
267 the important precursor of nitrate, is the major oxidation product of nitrogen
268 oxides (NO_x = NO+NO₂) (Seinfeld and Pandis, 2006). To show the model's
269 capability in simulating the NO_x-O₃-HNO₃ cycle over the studied regions, we
270 present and evaluate the simulated HNO₃ and O₃ in this section.

271 Simulated mixing ratios of HNO₃ and O₃ in the UTLS are evaluated by
272 using datasets from the limb viewing satellite instrument of Microwave Limb
273 Sounder (MLS, version 3.3, level 2,

274 ftp://acdisc.gsfc.nasa.gov/data/s4pa///Aura_MLS_Level2/). The MLS datasets
275 provide valuable information on atmospheric compositions in the UTLS
276 (Waters et al., 2006). For HNO₃, the MLS provides datasets for 215 to 1.5 hPa,
277 with a vertical resolution of 3–4 km and a horizontal resolution of 400–500 km.
278 Since further evaluations are needed for datasets at altitudes with pressures
279 higher than 215 hPa (Livesey et al., 2011), we use only datasets for pressures
280 lower than that. For O₃, the MLS provides datasets for 261 to 0.02 hPa, with a
281 vertical resolution of 2.5–3 km and a horizontal resolution of 300–400 km in the
282 UTLS (Santee et al., 2007; Livesey et al., 2011). The uncertainties of the MLS
283 HNO₃ and O₃ datasets in the UTLS are about ±0.5–1 ppbv and 0.02–0.04
284 ppmv, respectively (Livesey et al., 2011).

285 **3.1 HNO₃**

286 Figure 3(a) shows the simulated global distribution of HNO₃ concentrations
287 averaged over June-August of 2005. Concentrations of HNO₃ exceed 1 ppbv
288 over the industrialized areas such as Europe, North America, central and
289 eastern Asia, and over biomass burning regions in the tropics, in agreement
290 with the distributions and magnitudes reported in Liao et al. (2003). Over South
291 Asia, simulated HNO₃ concentrations are high (0.3–1 ppbv) in the northern
292 Indian subcontinent, because the emissions of NO_x and NH₃ are high in this
293 region (Streets et al., 2003; Zhang et al., 2009; Datta et al., 2012).

294 Figures 4(a)-4(b) show the simulated HNO₃ concentrations in the UTLS
295 averaged over June-August of 2005. Since the tropopause is located at
296 70–150 hPa (12–15 km) over the TP/SASM region (Li et al., 2005; Bian et al.,
297 2011b; Fadnavis et al., 2014), we choose the vertical layers of 200 hPa and
298 100 hPa to represent the UTLS. At both 200 hPa and 100 hPa, the highest

299 HNO₃ concentrations are simulated to occur in the high latitude regions in the
300 Northern Hemisphere (NH) (Fig. 4(a) and Fig. 4(b)). Simulated HNO₃
301 concentrations at 100 hPa are low over the region of 40–100°E and 10–30°N,
302 which is part of the anticyclone region defined in Fig. 1. Figure 4(c) shows the
303 latitude-altitude cross section of simulated seasonal mean HNO₃ mixing ratios
304 averaged over 70–105°E. In boreal summer, the highest HNO₃ mixing ratios
305 are simulated to occur at 30 hPa over the Polar Regions in both hemispheres.
306 Over high latitudes, HNO₃ concentrations in the Southern Hemisphere (SH)
307 are simulated to be higher than those in the NH.

308 To evaluate the simulated HNO₃, Figures 4(d)-4(f) show HNO₃
309 concentrations in the UTLS from MLS that are averaged over June-August of
310 2005. At 200 and 100 hPa altitudes, the observed HNO₃ mixing ratios are high
311 in the high latitudes in the NH, which are captured by the GEOS-Chem model.
312 The observed HNO₃ at 100 hPa exhibits low values of less than 400 pptv over
313 30–100°E and 10–30°N in the Asian monsoon anticyclone region (Fig. 4(e)). At
314 100 hPa, the observed HNO₃ mixing ratio averaged over the TP/SASM region
315 (70–105°E, 10–40°N) is 301.3 pptv, which is lower than the simulated value of
316 349.1 pptv. The difference between the simulated and observed HNO₃ mixing
317 ratio lies within the confidence range of ±500–1000 pptv of the MLS
318 instruments (Livesey et al., 2011). Considering all the grid cells with MLS
319 HNO₃ data available, the simulated seasonal mean HNO₃ concentrations show
320 normalized mean bias (NMB) of +15.9% at 100 hPa over the TP/SASM region
321 in summer of year 2005. The observed pattern of the HNO₃ vertical distribution
322 (Fig. 4(f)) is also captured by the GEOS-Chem model (Fig. 4(c)). The
323 distributions of HNO₃ in the UTLS are associated with the Brewer-Dobson

324 circulation proposed by Brewer (1949) and Dobson (1956), traveling upwards
325 across the tropopause to the stratosphere at the equator and downwards to
326 the troposphere near the Polar region.

327 **3.2 O₃**

328 Figure 3(b) shows the global distribution of simulated summertime
329 surface-layer O₃ concentrations. Simulated O₃ concentrations are in a range of
330 40–70 ppbv over Europe, North America, China, and the biomass burning
331 region of South Africa. Our model results agree closely with the simulated
332 distributions and magnitudes reported in Mickley et al. (1999), Collins et al.,
333 (2000), Liao et al. (2003), Wu et al., (2008), Zeng et al. (2008), and Fadnavis et
334 al. (2014). Fadnavis et al. (2014) also presented aircraft measurements over
335 India in September of 2010 during the Cloud Aerosol Interaction and
336 Precipitation Enhancement Experiment (CAIPEEX). Our simulated O₃
337 concentrations of 30–40 ppbv over India agree with the CAIPEEX
338 measurements.

339 Figures 5(a)-5(b) show the simulated O₃ concentrations in the UTLS
340 averaged over June-August of 2005. The distributions of O₃ concentrations in
341 the UTLS are similar to those of HNO₃, with elevated values in the high
342 latitudes of the NH. Relatively low O₃ mixing ratios of less than 200 ppbv are
343 simulated at 100 hPa over 10–30°N, 20–110°E, within the anticyclone region
344 defined in Fig. 1. Our simulated distributions and magnitudes of O₃ agree with
345 those reported in Bian et al. (2011b), which examined the summertime
346 distributions of O₃ in the UTLS during 2005–2009 by using the MLS version 2.2
347 level 2 products (Livesey et al., 2008). Because the background O₃
348 concentrations are generally high in the UTLS and the stratosphere, the low O₃

349 concentrations in the UTLS over the TP/SASM region are caused by the deep
350 convection that transports O₃-poor air upward (Fu et al., 2006; Randel and
351 Park, 2006; Park et al., 2007; Bian et al., 2011b). Figure 5(c) displays the
352 latitude-altitude cross section of seasonal mean O₃ mixing ratios averaged
353 over 70–105°E. As a result of the Brewer-Dobson circulation, O₃
354 concentrations in the UTLS are lower over the tropics than in the Polar
355 Regions, even though the maximum O₃ concentrations are located around 10
356 hPa over the tropics (Brewer, 1949). Our simulated O₃ concentrations in the
357 UTLS agree well with the measurements from MLS (Fig. 5(d)-5(f)). At 100 hPa,
358 simulated and MLS observed O₃ mixing ratios averaged over the TP/SASM
359 region (70–105°E, 10–40°N) are 190.6 and 145.1 ppbv, respectively.
360 Compared to MLS observations, simulated O₃ concentrations at 100 hPa have
361 a NMB of +31.4% over the TP/SASM region in summer of 2005. Our simulated
362 global STE of O₃ is 420 Tg yr⁻¹, which is within the range reported in previous
363 studies (475±120 Tg yr⁻¹ in McLinden et al. (2000), 420 Tg yr⁻¹ in Škerlak et al.
364 (2014), 556±154 Tg yr⁻¹ in Stevenson et al. (2006), and 550±140 Tg yr⁻¹ in
365 Solomon et al. (2007)).

366 In addition to the comparisons against MLS products, the simulated O₃
367 profiles are compared with balloon-borne sonde measurements in Fig. 6. The
368 measurements were carried out at Kunming (KM, 102.7°E, 25.0°N) in August
369 of 2009 and 2012, and at Lhasa (LH, 91.1°E, 29.7°N) in August of 2010 and
370 2013. The uncertainties of the observed O₃ mixing ratios were estimated to be
371 within 5–10% (Bian et al. 2012). The comparisons with multi-year observations
372 show that the model can reproduce the vertical distributions of O₃ above 12 km
373 in Kunming and Lhasa. At 100 hPa, the simulated monthly mean O₃ mixing

374 ratio in KM is 112.6 ppbv, and the observed value is 124.2 ppbv in 2009 and
375 113.5 ppbv in 2012. In LH, the simulated monthly O₃ mixing ratio at 100 hPa is
376 152.6 ppbv, and the observed O₃ mixing ratio at that altitude is 142.4 ppbv in
377 2010 and 167.9 ppbv in 2013. The magnitudes of O₃ mixing ratios from these
378 balloon-borne sonde measurements support those from MLS; O₃ mixing ratios
379 in the UTLS are less than 200 ppbv over the TP/SASM region.

380

381 **4 Simulated aerosols and model evaluation**

382 **4.1 Simulated aerosols**

383 Figure 7 (a) shows the simulated surface-layer concentrations of SO₄²⁻, NO₃⁻,
384 NH₄⁺, OC, BC, and PM_{2.5} (the sum of the mass of SO₄²⁻, NO₃⁻, NH₄⁺, BC, and
385 OC aerosols) averaged over June-August of year 2005. As expected,
386 simulated aerosol concentrations are high over polluted regions such as India
387 and eastern China as a result of the high anthropogenic emissions of aerosol
388 precursors and aerosols (Streets et al., 2003; Huang et al., 2012). Over the
389 TP/SASM region (70–105°E, 10–40°N), the average concentrations of SO₄²⁻,
390 NO₃⁻, NH₄⁺, BC, and OC are 1.70, 0.94, 0.85, 0.30, and 0.94 μg m⁻³,
391 respectively. NO₃⁻ is simulated to be of secondary importance at the surface
392 over the region of our interest. The simulated distributions and magnitudes of
393 these aerosol species are similar to those reported in Wang et al. (2013) and
394 Mu and Liao (2014).

395 Figures 7(b) and 7(c) also show the simulated concentrations of SO₄²⁻,
396 NO₃⁻, NH₄⁺, OC, BC, and PM_{2.5} in the UTLS. Elevated concentrations of SO₄²⁻,
397 NO₃⁻, NH₄⁺, OC, BC and PM_{2.5} are simulated over the TP and Plateau south
398 slope at 200 hPa altitude, and extend from eastern Mediterranean to western

399 China at 100 hPa. The simulated enhanced concentrations of SO_4^{2-} , OC, and
400 BC at 100 hPa over the anticyclone region (20–120°E, 10–40°N) agree with
401 previous observational and modeling studies (Lelieveld et al., 2001; Li et al.,
402 2005; Fadnavis et al., 2013). Li et al. (2005) reported elevated CO
403 concentrations in the upper troposphere over the TP, on the basis of both MLS
404 measurements and the GEOS-Chem simulation for September 2004.
405 Fadnavis et al. (2013) also simulated maximum concentrations of SO_4^{2-} , OC,
406 BC, and mineral dust aerosols in the UTLS during the Asian summer monsoon
407 season owing to convective uplifting of the boundary layer pollutants. With
408 NO_3^- aerosol accounted for in our simulation, NO_3^- is simulated to be the most
409 dominant aerosol species in the UTLS over the TP/SASM region, followed
410 by SO_4^{2-} , NH_4^+ , OC, and BC. At 100 hPa, the averaged concentrations of SO_4^{2-} ,
411 NO_3^- , NH_4^+ , OC, and BC over the TP/SASM region (70–105°E, 10–40°N)
412 region are 0.026, 0.069, 0.014, 0.011, and 0.002 $\mu\text{g m}^{-3}$, respectively.

413 **4.2 Comparisons of simulated aerosol concentrations with in-situ** 414 **observations**

415 The simulated aerosol concentrations in East Asia in the GEOS-Chem model
416 have been evaluated in previous studies (L. Zhang et al., 2010; Fu et al., 2012;
417 Jeong and Park, 2013; Jiang et al., 2013; Wang et al., 2013; Lou et al., 2014).
418 Here we are focused on the evaluation of aerosols in the South Asian
419 monsoon region. For lack of publicly accessible in situ measurements of
420 summertime aerosols in South Asia monsoon area, we compiled monthly or
421 seasonal mean measured concentrations of each aerosol species based on
422 measurements reported in the literature (see Table S1 in the Supplementary
423 Material). These measurements were carried out over years of 1992–2010.

424 The locations of sites with measurements available are shown in Fig. 8(a).
425 Most sites are located in the upwind directions of the TP, with pollutants that
426 can be transported to the UTLS during the South Asian summer monsoon
427 season. The observed PM₁₀ concentrations listed in Table S1 are multiplied by
428 0.6 to convert to PM_{2.5} for model evaluation, following the suggestions in
429 Zhang et al. (2002) and Chatterjee et al. (2010).

430 Figures 8(b)–8(f) show the scatterplots of simulated versus observed
431 seasonal mean aerosol concentrations. Compared with measurements,
432 simulated SO₄²⁻, NO₃⁻, NH₄⁺, OC and BC have NMBs of -17.0%, +38.8%,
433 +42.0%, -69.7% and -41.0%, respectively, as the concentrations of all
434 seasons are considered. The correlations between model results and
435 observations have *R* values of 0.49–0.85 for all aerosol species, indicating that
436 the model is capable of capturing the spatial distributions and seasonal
437 variations of each aerosol species in the South Asian monsoon region despite
438 the biases in concentrations. If we consider simulated and measured
439 concentrations for JJA alone, the simulated concentrations of SO₄²⁻, NO₃⁻,
440 NH₄⁺, OC and BC exhibit seasonal NMBs of -14.7%, +51.5%, +74.9%, -57.2%
441 and -32.2%, respectively, and the values of *R* are in the range of 0.24–0.85.
442 Note that the measurements of NO₃⁻ and NH₄⁺ are quite limited in terms of the
443 number of samples, and the discrepancies between model results and
444 measurements may also arise from the mismatch of the model year 2005 with
445 the years of 1992–2010 with observations available.

446 **4.3 Comparisons of simulated aerosol extinction coefficients with SAGE** 447 **II datasets**

448 Satellite datasets from the Stratospheric Aerosol and Gas Experiment II

449 (SAGE II, https://eosweb.larc.nasa.gov/project/sage2/sage2_v620_table) are
450 used to evaluate the simulated aerosol extinction in the UTLS. The SAGE II
451 instrument was launched in October 1984 aboard the Earth Radiation Budget
452 Satellite (ERBS) and terminated on 8 September 2005 (McCormick et al. 1987;
453 Chu et al. 1989). The datasets used here are aerosol extinction coefficients at
454 525 nm from the version 6.20 SAGE retrievals, covering from 0.5 to 40 km with
455 a vertical resolution of 0.5 km. Many validation studies have been conducted
456 on the SAGE II aerosol data (Russell and McCormick, 1989; Oberbeck et al.,
457 1989; Wang et al., 1989), which indicated that extinction coefficients have
458 uncertainties of 20–30%. The extinction coefficients of aerosols in the
459 GEOS-Chem model are calculated using aerosol mass concentration,
460 extinction efficiency, effective radius, particle mass density, and the assumed
461 aerosol size distribution (Drury et al., 2010). The hygroscopic growth of each
462 aerosol species with relative humidity is accounted for, using the hygroscopic
463 growth factors listed in Martin et al. (2003).

464 Figure 9(a) presents the simulated monthly mean distribution of aerosol
465 extinction coefficients at 100 hPa for July of 2005. At 100 hPa, the simulated
466 aerosol extinction coefficients are relatively high over the anticyclone region,
467 where anthropogenic aerosol species (Fig. 7) and natural aerosols such as
468 mineral dust and sea salt contribute to aerosol extinction coefficients in
469 summer. Note that the contributions of sulfate, nitrate, ammonium, OC, sea
470 salt, and mineral dust are all considered when we calculate aerosol extinction
471 coefficients. Aerosol extinction coefficients are simulated to be $1.2\text{--}2\times 10^{-3}$
472 km^{-1} at 100 hPa over the Asian continent and Indian Ocean ($20^{\circ}\text{S}\text{--}30^{\circ}\text{N}$,
473 $30^{\circ}\text{--}105^{\circ}\text{E}$). These values agree closely with aerosol extinction coefficients

474 measured at Naqu during August of 2011 for the same altitude, the maximum
475 of which was $2.4 \times 10^{-3} \text{ km}^{-1}$ (He et al., 2014). Vernier et al. (2011) also
476 identified this Asian aerosol layer with high SR at 100 hPa by observations of
477 CALIPSO for JJA of 2006–2008.

478 Figure 9(b) displays the monthly mean vertical profiles of aerosol extinction
479 coefficients averaged over the Asian monsoon anticyclone region ($20\text{--}120^\circ\text{E}$
480 $10\text{--}40^\circ\text{N}$) (Fig. 1) for July of 2005. The SAGE II datasets are available for July
481 only in 2005. The profiles from SAGE II and the GEOS-Chem simulation are all
482 shown. The vertical distributions of aerosol extinction coefficients “with nitrate”
483 and “without nitrate” are both from the baseline run with full chemistry. The
484 vertical distribution of aerosol extinction coefficient “with nitrate” (or “without
485 nitrate”) indicates that the contribution of nitrate aerosol to aerosol extinction is
486 (or is not) accounted for. Accounting for all aerosol species, the GEOS-Chem
487 model reproduces well the aerosol extinction coefficients above 10 km, but the
488 discrepancies are rather large in altitudes less than 10 km. Note that the
489 uncertainties in satellite datasets increase as the altitude decreases
490 (Vanhellemont et al., 2008; Kulkarni and Ramachandran, 2015), and the
491 missing data in the lower troposphere along the satellite trajectories over the
492 region of our interest also contribute to the discrepancies

493 Comparisons of profiles of aerosol extinction coefficients with and without
494 nitrate aerosol indicate that the profiles show small differences in altitudes less
495 than 6 km but large discrepancies from 6 km to the tropopause. With nitrate
496 aerosol accounted for, the simulated aerosol extinction coefficients agree
497 closely with SAGE II datasets in the UTLS (averaged over 14–16 km, the
498 simulated value is $8.6 \times 10^{-4} \text{ km}^{-1}$ while the observed value is $8.0 \times 10^{-4} \text{ km}^{-1}$).

499 Without nitrate aerosol, the simulated aerosol extinction coefficient at 14–16
500 km altitude is $1.5 \times 10^{-4} \text{ km}^{-1}$, which underestimates the aerosol extinction
501 coefficient by 82.6% compared to that calculated with all the aerosol species.
502 These comparisons of extinction coefficients with and without nitrate aerosol
503 suggest that nitrate aerosol plays an important role in aerosol extinction in the
504 UTLS over the region of our interest.

505

506 **5 Contribution of nitrate to aerosol concentrations in the UTLS**

507 Since nitrate aerosol is simulated to be the most abundant aerosol species in
508 the UTLS over the TP/SASM region, we analyze the contribution of nitrate to
509 $\text{PM}_{2.5}$ concentration ($C_{\text{NIT}} = \text{nitrate concentration} / \text{PM}_{2.5} \text{ concentration}$) in this
510 section. Figure 10 shows the simulated seasonal mean distributions of C_{NIT} for
511 June–August of year 2005. At the surface layer (Fig. 10(a)), simulated high C_{NIT}
512 values are located over the areas with high nitrate concentrations (India and
513 eastern China) as well as the oceans where NO_3^- also forms on sea salt and
514 mineral dust particles (Arimoto et al., 1996; Nakamura et al., 2005; George and
515 Nair, 2008). Over the TP/SASM region, the C_{NIT} values in JJA are 5–35% at the
516 surface, 25–50% at 200 hPa (Fig. 10(b)), and could exceed 60% at 100 hPa
517 (Fig. 10(c)). The latitude–altitude cross section of C_{NIT} (Fig. 10(d)) shows that
518 C_{NIT} over 20–40°N increases with altitude and reaches maximum values
519 around the extratropical tropopause.

520 Table 2 lists the mean concentrations of SO_4^{2-} , NO_3^- , NH_4^+ , BC and OC, and
521 their contributions to $\text{PM}_{2.5}$ during summertime of 2005 over the TP/SASM, TP,
522 and SASM regions. Over the TP/SASM region, SO_4^{2-} , NO_3^- , NH_4^+ , BC and OC
523 are simulated to contribute 35.9%, 19.8%, 18.1%, 6.4%, and 19.8%,

524 respectively, to $PM_{2.5}$ mass concentration at the surface layer. The
525 contributions increase significantly in the UTLS. The largest C_{NIT} is simulated
526 in the SASM region at 100 hPa, where NO_3^- accounts for 60.5% of $PM_{2.5}$ mass
527 concentration. The high C_{NIT} values indicate that NO_3^- plays an important role
528 in the aerosol layer in the UTLS over the TP/SASM region.

529 Considering the large uncertainties in simulated sea salt (Jaeglé et al.,
530 2011) and mineral dust (Fairlie et al., 2007) aerosols, we tend to be focused on
531 anthropogenic aerosol species (SO_4^{2-} , NO_3^- , NH_4^+ , BC, and OC) in this work. In
532 our model, concentrations of sea salt (or mineral dust) are simulated to be
533 $1.0\text{--}1.7\text{ ng m}^{-3}$ (or $5.0\text{--}7.0\text{ ng m}^{-3}$) over the studied region in the summer of
534 2005, which contribute less than 1.2% (or 5.0%) to total aerosol mass at 100
535 hPa. Therefore the consideration of sea salt and mineral dust can slightly
536 reduce C_{NIT} values, but C_{NIT} values at 100 hPa are still as high as 45-65% over
537 the TP/SASM region in summer.

538

539 **6 Mechanisms for high nitrate concentrations in the UTLS**

540 **6.1 Upward transport of nitrate from the lower troposphere**

541 The intense convective transport of chemical species into the UTLS over the
542 TP/SASM region during summertime has been widely discussed in previous
543 studies (Randel et al., 2010; Bian et al., 2011a; Fadnavis et al., 2013, 2014;
544 Qie et al., 2014; He et al., 2014). Since nitrate aerosol is simulated to be of
545 secondary abundant aerosol species in the surface layer over the TP/SASM
546 region (Fig. 7), the vertical mass transport through the deep convection in this
547 region contributes to the accumulation of NO_3^- in the UTLS. Figure 11 shows
548 the latitude-altitude cross sections of simulated concentrations of SO_4^{2-} and

549 NO_3^- averaged over 70–105°E in June–August of 2005, together with the
550 wind vectors obtained from the European Centre for Medium-Range Weather
551 Forecasts (ECMWF) ERA-Interim Reanalysis data. Note that the assimilated
552 GEOS-5 meteorological fields do not have vertical winds
553 (http://wiki.seas.harvard.edu/geos-chem/index.php/List_of_GEOS-5_met_fields
554 s), so we use the ECMWF reanalysis wind fields to do the analysis here. High
555 values of aerosol concentrations are found on the south slope of the
556 Himalayas, where the deep convection exists. Although both SO_4^{2-} and NO_3^-
557 are transported upward to the extratropical tropopause, the details of the
558 vertical distributions are different. At altitudes higher than 8 km, the
559 concentrations of NO_3^- do not decrease with altitude as quickly as those of
560 SO_4^{2-} , and the concentrations of NO_3^- over 10–40°N are higher than those of
561 SO_4^{2-} .

562 The chemical mechanisms for the formation of SO_4^{2-} , NO_3^- , and NH_4^+
563 aerosols in the GEOS-Chem model were described in R. J. Park et al. (2004),
564 which are comprehensive and have been used extensively in previous studies
565 to simulate these three aerosol species (R. J. Park et al., 2004; Pye et al.,
566 2009; L. Zhang et al., 2010; Zhu et al., 2012; Jiang et al., 2013; Lou et al.,
567 2014). Sulfate aerosol forms from gas-phase oxidation of SO_2 by OH and from
568 in-cloud oxidation of SO_2 by O_3 and H_2O_2 . Nitrate forms from the partitioning of
569 HNO_3 between gas and aerosol phases, which is calculated by the
570 ISORROPIA II thermodynamic equilibrium module (Fountoukis and Nenes,
571 2007) in the GEOS-Chem model. Major reactions for the production and loss
572 of HNO_3 were listed in Liao and Seinfeld (2005). HNO_3 is produced by the
573 reaction of NO_2 with OH during daytime and by hydrolysis of N_2O_5 on aerosol

574 surfaces at night. The chemical mechanisms for SO_4^{2-} and NO_3^- have
575 different sensitivity to meteorological conditions. During the vertical transport,
576 temperature decreases, which reduces the gas-phase oxidation of SO_2 (Yao et
577 al., 2002; Seinfeld and Pandis 2006) but promotes the formation of NO_3^- by
578 shifting gas-particle equilibria (Dawson et al., 2007; Liao et al., 2009). Dawson
579 et al. (2007) examined the sensitivities of sulfate and nitrate concentrations to
580 temperature by using the Particulate Matter Comprehensive Air Quality Model
581 with extensions (PMCAMx). The sensitivity test was performed by fixing all
582 meteorological parameters but perturbing temperature. Their sensitivity
583 simulations showed that the increases in temperature led to increases in
584 sulfate concentrations and decreases in nitrate concentrations. Compared to
585 nitrate, sulfate concentrations showed smaller sensitivity to temperature
586 changes (Dawson et al., 2007); as temperature increased, nitrate
587 concentrations decreased by $19\% \text{ K}^{-1}$ and $17\% \text{ K}^{-1}$ in January and July
588 respectively, while sulfate concentration increased by $0.12\% \text{ K}^{-1}$ and $1.3\% \text{ K}^{-1}$
589 in January and July, respectively. Therefore the different chemical mechanisms
590 for SO_4^{2-} and NO_3^- formation contribute to the differences in their vertical
591 distributions.

592 **6.2 The gas-to-aerosol conversion of HNO_3 to form nitrate**

593 As mentioned above, the formation of gas-phase HNO_3 and the partitioning of
594 HNO_3 between gas and aerosol phases are the two major chemical processes
595 that influence NO_3^- concentrations. We have evaluated the ability of the
596 GEOS-Chem model to simulate gas-phase HNO_3 in Section 3.1 (by
597 comparisons of our model results with MLS observations and concentrations
598 from previous modeling studies), so we quantify here NO_3^- formation from

599 gas-to-aerosol conversion of HNO_3 based on the ISORROPIA II
600 thermodynamic equilibrium module (Fountoukis and Nenes, 2007). The
601 gas-to-aerosol conversion of HNO_3 to form NO_3^- is very sensitive to relative
602 humidity (RH) and temperature (Fountoukis and Nenes 2007; Dawson et al.,
603 2007). Low temperature and high RH are favorable for NO_3^- formation. Figure
604 12 shows the seasonal mean horizontal distributions of RH and temperature at
605 100 hPa and the latitude-altitude cross sections of these two parameters
606 averaged over 70–105°E. RH exhibits high values in the TP/SASM region,
607 which are consistent with the high H_2O mixing ratios in this area reported in
608 Gettelman et al. (2004), M. Park et al. (2004), and Fu et al. (2006). At 100 hPa,
609 the locations with high RH of exceeding 45% correspond well with those with
610 high C_{NIT} values (Fig. 10(c)). The latitude-altitude cross section of RH (Fig.
611 12(c)) shows that RH has high values over the places with intense upward
612 transport (Fig. 11). For temperature, as Fig. 12(b) and 12(d) show, summertime
613 temperatures are cold (190–200 K) at 100 hPa in the TP/SASM region,
614 consistent with the distribution and magnitude reported for August, 2011, in He
615 et al. (2014) on the basis of the NCEP Reanalysis data. The low temperatures
616 over the TP/SASM region are associated with the adiabatic expansion of
617 ascending air mass of the deep convections (Yanai et al., 1992; Park et al.,
618 2007; He et al., 2014).

619 Because of the favorable conditions of RH and temperature, the
620 gas-to-aerosol conversion of HNO_3 to form nitrate can occur during the upward
621 transport and in the UTLS. Figure 13 shows the mass budget for nitrate
622 aerosol within the selected box of (70–105°E, 10–40°N, 8–16 km) to see the
623 role of nitrate formation over the TP/SASM region. The horizontal mass fluxes

624 have a net negative value of $0.10 \text{ Tg season}^{-1}$, reducing nitrate aerosol in the
625 selected box. The vertical transport and the gas-to-aerosol conversion of
626 HNO_3 increase nitrate mass in the selected box, with values of 0.09 Tg
627 season^{-1} and $0.11 \text{ Tg season}^{-1}$, respectively, indicating that the gas-to-aerosol
628 conversion plays an important role in the enhancement of nitrate in the UTLS
629 over the TP/SASM region. Although relatively high RH exists near the
630 tropopause of the TP/SASM region, the air near the tropopause is still dryer
631 compared to that in the lower altitudes. Model results show that the
632 gas-to-aerosol partition of HNO_3 decreases with altitude over 8–16 km,
633 indicating that the gas to aerosol conversion contributes to nitrate
634 accumulation in the UTLS mainly during the process of upward transport.

635 Previous studies have also reported that nitric acid trihydrates (NAT,
636 $\text{HNO}_3 \cdot (\text{H}_2\text{O})_3$) could form in the polar and tropical stratosphere at low
637 temperatures through two mechanisms: (1) the homogeneous nucleation out
638 of supercooled ternary solutions, and (2) the heterogeneous formation on ice
639 particles (Hofmann et al., 1989; Carslaw et al., 1998; Voigt et al., 2000; Popp et
640 al., 2006; Kirner et al., 2011). A typical NAT condensation temperature is
641 approximate 193 K (Kirner et al., 2011). As shown in Fig. 12, the temperatures
642 around 100 hPa over the TP/SASM region are in the range of 190–200 K,
643 which are low enough to produce some NAT particles. However, balloon-borne
644 measurements of depolarization ratio and backscattering ratio of aerosols at
645 Lhasa during August-October of 1999 by Kim et al. (2003) and Tobo et al.
646 (2007) suggested that coarse and aspherical particles such as NAT are scarce
647 in the UTLS of the TP/SASM.

648

649 **7 Sensitivities of simulated nitrate in the UTLS to anthropogenic NO_x,**
650 **NH₃, and SO₂ emissions in Asia**

651 Since simulated SO₄²⁻, NO₃⁻ and NH₄⁺ concentrations have, respectively,
652 NMBs of -17.0%, +38.8%, and +42.0% on an annual mean basis and of
653 -14.7%, +51.5%, and +74.9% in summer (Section 4.2), we perform four
654 sensitivity simulations to examine the impacts of uncertainties in surface-layer
655 aerosol concentrations on simulated nitrate in the UTLS. In the first three
656 cases, anthropogenic emissions of NO_x, NH₃, and SO₂ in Asia are changed by
657 -50%, -50%, and +20%, respectively, relative to those in our standard
658 simulation. In the last case, anthropogenic emissions of all these three species
659 are changed simultaneously, with NO_x reduced by 50%, NH₃ reduced by 50%,
660 and SO₂ increased by 20% in Asia relative to the standard case. The purpose
661 of these sensitivity studies is to reduce NMBs of simulated surface-layer
662 concentrations of SO₄²⁻, NO₃⁻ and NH₄⁺ and see whether NO₃⁻ is still the
663 most dominant aerosol species in the UTLS. Model results from these
664 sensitivity studies for summer of 2005 are presented in Table 3.

665 As anthropogenic emissions of SO₂ in Asia are increased by 20%, the
666 NMB of simulated surface-layer SO₄²⁻ concentrations is -4.4%, which is an
667 improvement compared to the NMB of -14.7% in the standard simulation.
668 However, the increases in SO₂ emissions lead to larger NMBs of surface-layer
669 NO₃⁻ and NH₄⁺ because of the increased formation of ammonium sulfate or
670 ammonium bisulfate. The percentage contributions of SO₄²⁻ to total aerosol
671 mass in the UTLS increase slightly by 2.7% at 200 hPa and by 1.6% at 100
672 hPa, and nitrate in the UTLS also shows small sensitivity to the change in SO₂
673 emissions.

674 With anthropogenic emissions of NO_x in Asia reduced by 50%, the NMB of
 675 simulated surface-layer NO_3^- concentrations changes from +51.5% in the
 676 standard simulation to -11.7% in this sensitivity run. The contribution of each
 677 of SO_4^{2-} , NO_3^- and NH_4^+ aerosols to total aerosol mass in the UTLS is not
 678 sensitive to this reduction in NO_x emissions at the surface; the percentage
 679 contribution obtained from this sensitivity run is very close to the value
 680 obtained in the standard simulation (Table 3). Similarly, in the sensitivity study
 681 with NH_3 emissions reduced by 50% in Asia, simulated surface-layer
 682 concentrations of NO_3^- and NH_4^+ are improved in terms of the values of NMBs,
 683 but the improvement in simulated aerosol concentrations at the surface-layer
 684 does not influence our conclusion of high nitrate aerosol concentration in the
 685 UTLS.

686 As shown in Table 3, for the surface layer, simulated nitrate concentration
 687 over the TP/SASM region decreases by 46.8% with a 50% reduction in
 688 anthropogenic NO_x emissions in Asia, and it decreases by 22.3% when
 689 anthropogenic NH_3 emissions are reduced by the same percentage, indicating
 690 that surface-layer nitrate aerosol is more sensitive to anthropogenic emissions
 691 of NO_x than to those of NH_3 . Relative to the baseline simulation, simulated
 692 nitrate concentrations at 200 hPa and 100 hPa decrease, respectively, by 49.0%
 693 and 17.7% with a 50% reduction in NH_3 emissions, whereas only by 2.1% and
 694 1.3% with a 50% reduction in NO_x emissions. Over the studied region, the role
 695 of NH_3 in the sulfate-nitrate-ammonium aerosol system can be quantified by
 696 the gas ratio of $\text{GR} = \frac{\text{free ammonia}}{\text{total nitrate}} = \frac{\text{TA} - 2 \times \text{TS}}{\text{TN}}$ (Ansari and Pandis, 1998), where
 697 $\text{TA} = \text{NH}_3 + \text{NH}_4^+$, $\text{TS} = \text{SO}_4^{2-}$, and $\text{TN} = \text{HNO}_3 + \text{NO}_3^-$. Over the TP/SASM
 698 region, GR is generally positive both at the surface and in the UTLS, especially

699 over 20–40°N where deep convection exits (Fig. 11), indicating that S(VI) is in
700 from of sulfate and free ammonia is available to react with nitrate (Seinfeld and
701 Pandis 2006). However, GR is generally less than 1.0 above 400 hPa in
702 summer over the TP/SASM region, which indicates nitrate concentrations are
703 most sensitive to changes in NH₃ and explains the small sensitivity of nitrate
704 aerosol to NO_x emissions in the UTLS.

705 In the sensitivity study with emissions of NO_x, NH₃, and SO₂ in Asia
706 changed simultaneously, simulated surface-layer concentrations of SO₄²⁻, NO₃⁻
707 and NH₄⁺ have NMBs of -8.3%, -27.0% and +55.4%, respectively, which are
708 all improved compared to those in the standard simulation. Even though nitrate
709 aerosol is now underestimated at the surface, it still accounts for 53.3% of the
710 PM_{2.5} concentration at 100 hPa over the TP/SASM region in summer.

711 It should be noted that the concentrations of OC and BC are also
712 underestimated, with NMBs of -57.2% and -32.2%, respectively, in summer
713 (Section 4.2). We have done a simple calculation with the concentrations of
714 OC and BC in the UTLS multiplied by 2.3 and 1.5, respectively, and nitrate is
715 still the most dominant aerosol species in summertime in the UTLS over the
716 TP/SASM region (not shown in Table 3). Therefore the uncertainties in surface
717 aerosol concentrations do not compromise the conclusion of this study.

718

719 **8 Conclusions**

720 In this work we simulate nitrate aerosol and its contribution to aerosol
721 concentrations in the UTLS over the TP/SASM region (70–105°E, 10–40°N)
722 for summertime of year 2005, using the global chemical transport model
723 GEOS-Chem driven by the assimilated meteorological fields.

724 Simulated HNO₃ and O₃ are evaluated to show the model's ability to
725 simulate the NO_x-O₃-HNO₃ cycle over the studied region. In the UTLS, both
726 the horizontal and vertical distributions of simulated HNO₃ and O₃ agree well
727 with the MLS observations. At 100 hPa, simulated seasonal mean HNO₃ and
728 O₃ mixing ratios show NMBs of +15.9% and +31.4%, respectively, over the
729 TP/SASM region (70–105°E, 10–40°N) in summer of year 2005, and the model
730 biases lie within the confidence range of the MLS instruments. Both simulated
731 and observed O₃ concentrations show relatively low values of less than 200
732 ppbv at 100 hPa over the TP/SASM region.

733 Averaged over the TP/SASM region, the surface-layer concentrations of
734 SO₄²⁻, NO₃⁻, NH₄⁺, BC, and OC are simulated to be 1.70, 0.94, 0.85, 0.30, and
735 0.94 μg m⁻³, respectively. Nitrate aerosol is simulated to be of secondary
736 importance near the surface over the region of our interest. Comparisons of
737 simulated aerosol concentrations with ground-based observations show that
738 simulated summertime concentrations of SO₄²⁻, NO₃⁻, NH₄⁺, OC and BC have
739 NMB of -14.7%, +51.5%, +74.9%, -57.2% and -32.2%, respectively. Note that
740 the measurements of NO₃⁻ and NH₄⁺ are quite limited in terms of the number
741 of samples.

742 Model results show elevated concentrations of SO₄²⁻, NO₃⁻, NH₄⁺, OC, BC
743 and PM_{2.5} in the UTLS over the TP/SASM region throughout the summer. NO₃⁻
744 is simulated to be the most dominant aerosol species in the UTLS of the
745 TP/SASM region. Accounting for NO₃⁻ aerosol, the GEOS-Chem model
746 reproduces well the magnitude of aerosol extinctions above 10 km, as model
747 results are compared with the SAGE II measurements. The discrepancies
748 between the simulated and observed aerosol extinction coefficient are within 8%

749 in the UTLS (averaged over 14–16 km). Simulated vertical profiles of aerosol
750 extinction coefficients with and without nitrate aerosol show large
751 discrepancies from 6 km to tropopause, indicating the important role of nitrate
752 in aerosol layer in the UTLS over the TP/SASM region.

753 The contribution of NO_3^- to aerosols in the TP/SASM region is quantified
754 by C_{NIT} (the ratio of nitrate concentration to $\text{PM}_{2.5}$ concentration). Over the
755 TP/SASM region, the C_{NIT} values in summer are 5–35% at the surface, 25–
756 50% at 200 hPa, and could exceed 60% at 100 hPa. The mechanisms for the
757 accumulation of nitrate in the UTLS over the TP/SASM region include vertical
758 transport and the gas-to-aerosol conversion of HNO_3 to form nitrate. Such
759 gas-to-aerosol conversion occurs during the upward transport and in the UTLS.
760 The high relative humidity and low temperature associated with the deep
761 convection over the TP/SASM region are favorable for nitrate formation.

762 Results from the present study indicate that nitrate is an important aerosol
763 species in the UTLS over the ASM/TP region. Considering the scarce
764 measurements of nitrate in the UTLS and the model uncertainties, more
765 observational and modeling studies are needed to further explore the aerosol
766 composition in the Asian tropopause aerosol layer. Further simulations of
767 nitrate aerosol in the UTLS also need to account for NAT formation at low
768 temperatures (Kirner et al., 2011) and the roles of natural aerosols, including
769 the transport of mineral dust and sea salt to the UTLS as well as nitrate
770 formation on these natural particles (Ma et al., 2003).

771

772

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Table 1. Summary of Annual Emissions of Aerosols and Aerosol Precursors in Asia (60°E-155°E, 10°N-55°N)

Species	Global	Asia
NO_x (Tg N yr⁻¹)		
Aircraft	0.5	0.08
Anthropogenic	28.6	9.96
Biomass burning	4.7	0.27
Fertilizer	0.7	0.31
Lightning	5.9	0.87
Soil	5.9	0.96
Total	46.3	12.45
SO₂ (Tg S yr⁻¹)		
Aircraft	0.1	0.01
Anthropogenic	52.6	23.46
Biomass burning	1.2	0.07
Volcanoes	4.4	1.04
No_eruption	8.9	1.78
Ship	7.4	0.94
Total	74.6	27.30
NH₃ (Tg N yr⁻¹)		
Anthropogenic	34.9	17.83
Natural	14.2	2.01
Biomass burning	3.5	0.21
Biofuel	1.6	0.71
Total	54.2	20.76
OC (Tg C yr⁻¹)		
Anthropogenic	3.1	1.42
Biomass burning	18.7	1.10
Biofuel	6.3	3.28
Biogenic	9.7	1.22
Total	37.8	7.02
BC (Tg C yr⁻¹)		
Anthropogenic	3.0	1.43
Biomass burning	2.2	0.12
Biofuel	1.6	0.86
Total	6.8	2.41

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Table 2. Simulated seasonal mean concentrations of aerosols and their contributions to PM_{2.5} (in percentages in parentheses) during summertime (June-August) of 2005 for the TP/SASM, TP, and SASM regions. The unit is $\mu\text{g m}^{-3}$ for concentrations at the surface, and $10^{-2} \mu\text{g m}^{-3}$ for concentrations at 200 hPa and 100 hPa.

	PM _{2.5}	SO ₄ ²⁻	NO ₃ ⁻	NH ₄ ⁺	OC	BC
TP/SASM						
Surface	4.73	1.70(35.9%)	0.94(19.8%)	0.85(18.1%)	0.94(19.8%)	0.30(6.4%)
200 hPa	16.19	3.27(20.2%)	7.57(46.8%)	2.67(16.5%)	2.22(13.7%)	0.44(2.7%)
100 hPa	12.14	2.60(21.4%)	6.90(56.8%)	1.43(11.8%)	1.05(8.6%)	0.16(1.3%)
TP						
Surface	5.44	2.12(39.0%)	1.05(19.3%)	1.08(19.9%)	0.88(16.1%)	0.31(5.7%)
200 hPa	19.80	4.16(21.0%)	9.43(47.6%)	3.25(16.4%)	2.49(12.6%)	0.47(2.4%)
100 hPa	10.58	2.60(24.6%)	5.51(52.0%)	1.35(12.7%)	0.99(9.4%)	0.14(1.3%)
SASM						
Surface	4.02	1.28(31.8%)	0.83(20.5%)	0.63(15.6%)	1.00(24.8%)	0.29(7.2%)
200 hPa	12.57	2.38(18.9%)	5.72(45.5%)	2.10(16.7%)	1.95(15.5%)	0.41(3.3%)
100 hPa	13.71	2.60(19.0%)	8.30(60.5%)	1.52(11.1%)	1.11(8.1%)	0.18(1.3%)

1379 Table 3. Sensitivity simulations to examine the impacts of uncertainties in surface-layer aerosol concentrations on simulated NO_3^-
 1380 in the UTLS. “Conc” and “Ctr” denote, respectively, simulated seasonal mean concentrations of SO_4^{2-} , NO_3^- , NH_4^+ , OC, BC and
 1381 their contributions to $\text{PM}_{2.5}$ (in percentages) during summertime (June-August) of 2005. The mass concentrations are averaged
 1382 over the TP/SASM region, with unit of $\mu\text{g m}^{-3}$ at the surface layer and of $10^{-2} \mu\text{g m}^{-3}$ at 200 hPa and 100 hPa. Also shown are the
 1383 NMBs, as the simulated surface-layer concentrations are compared with measurements described in Section 4.2.

Species	Baseline Case			SO_2 (+20%)			NO_x (-50%)			NH_3 (-50%)			All Change		
	Conc.	Ctri.	NMB	Conc.	Ctri.	NMB	Conc.	Ctri.	NMB	Conc.	Ctri.	NMB	Conc.	Ctri.	NMB
Surface															
SO_4^{2-}	1.70	35.9%	-14.7%	1.92	38.1%	-4.4%	1.58	39.5%	-18.1%	1.70	38.1%	-14.7%	1.78	43.2%	-8.3%
NO_3^-	0.94	19.8%	+51.5%	0.94	18.7%	+53.5%	0.50	12.5%	-11.7%	0.73	16.4%	+24.1%	0.39	9.5%	-27.0%
NH_4^+	0.85	18.1%	+74.9%	0.94	18.6%	+93.8%	0.68	17.1%	+44.1%	0.78	17.6%	+64.6%	0.71	17.3%	+55.4%
OC	0.94	19.8%	-57.2%	0.94	18.6%	-57.2%	0.94	23.4%	-57.2%	0.94	21.0%	-57.2%	0.94	22.7%	-57.2%
BC	0.30	6.4%	-32.2%	0.30	6.0%	-32.2%	0.30	7.5%	-32.2%	0.30	6.8%	-32.2%	0.30	7.3%	-32.2%
200 hPa															
SO_4^{2-}	3.27	20.2%		3.67	22.9%		3.31	20.6%		3.29	29.1%		3.74	33.7%	
NO_3^-	7.57	46.8%		7.05	43.9%		7.41	46.0%		3.86	34.2%		3.19	28.7%	
NH_4^+	2.67	16.5%		2.67	16.6%		2.71	16.8%		1.49	13.2%		1.50	13.5%	
OC	2.22	13.7%		2.22	13.8%		2.22	13.8%		2.22	19.7%		2.22	20.0%	
BC	0.44	2.7%		0.44	2.7%		0.44	2.7%		0.44	3.9%		0.44	4.0%	
100 hPa															
SO_4^{2-}	2.60	21.4%		2.80	23.0%		2.66	21.9%		2.60	25.2%		2.87	27.2%	
NO_3^-	6.90	56.8%		6.72	55.3%		6.81	56.1%		5.68	55.0%		5.62	53.3%	
NH_4^+	1.43	11.8%		1.43	11.7%		1.45	12.0%		0.84	8.1%		0.84	8.0%	
OC	1.05	8.6%		1.05	8.6%		1.05	8.7%		1.05	10.2%		1.05	10.0%	
BC	0.16	1.3%		0.16	1.3%		0.16	1.3%		0.16	1.6%		0.16	1.5%	

1384 **Figure Captions**

1385

1386 **Figure. 1.** Regions examined in this study: the Tibetan Plateau region (TP,
1387 70–105°E, 25–40°N), the SASM region (SASM, 70–105°E, 10–25°N), and the
1388 anticyclone region of (20–120°E, 10–40°N).

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1390 **Figure. 2.** Monthly variations in emissions of NO_x (Tg N month⁻¹), SO₂ (Tg S
1391 month⁻¹), NH₃ (Tg N month⁻¹), OC (Tg C month⁻¹), and BC (Tg C month⁻¹)
1392 over Asia. Values shown are the total emissions (anthropogenic plus natural
1393 emissions listed in Table 1).

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1395 **Figure. 3.** Simulated global distributions of surface-layer HNO₃ (pptv) and O₃
1396 (ppbv) averaged over June-August, 2005.

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1398 **Figure. 4.** Comparisons of simulated HNO₃ concentrations (pptv) with
1399 observations (pptv) from MLS. (a) and (b) are simulated concentrations at 200
1400 hPa and 100 hPa, respectively. (c) is the latitude-altitude cross section of
1401 simulated HNO₃ concentrations averaged over 70–105°E. (d)-(f) are the same
1402 as (a)-(c), except that (d)-(f) are observations from MLS. The white areas in (d)
1403 and (f) have no datasets available from MLS. All the datasets are averaged
1404 over June-August of 2005.

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1406 **Figure. 5.** Comparisons of simulated O₃ concentrations (ppbv) with
1407 observations (ppbv) from MLS. (a) and (b) are simulated concentrations at 200
1408 hPa and 100 hPa, respectively. (c) is the latitude-altitude cross section of
1409 simulated O₃ concentrations averaged over 70–105°E. (d)-(f) are the same as
1410 (a)-(c), except that (d)-(f) are observations from MLS. All the datasets are
1411 averaged over June-August of 2005.

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1413 **Figure. 6.** The simulated and observed vertical profiles of monthly mean O₃
1414 mixing ratios at (a) Kunming and (b) Lhasa in August. The model results are
1415 from the simulation of year 2005. The observations in Kunming were
1416 conducted during August 7–13 (11 profiles of O₃ collected) in 2009 and during
1417 August 12–31 in 2012 (daily observations). The observations in Lhasa were
1418 conducted during August 22–28 in 2010 (12 profiles of O₃ collected) and
1419 during August 4–26 in 2013 (daily observations).

1420

1421 **Figure. 7.** Simulated seasonal mean concentrations (μg m⁻³) of sulfate, nitrate,
1422 ammonium, organic carbon, black carbon, and PM_{2.5} at (a) the surface layer, (b)
1423 200 hPa, and (c) 100 hPa, during summer (June-August) of year 2005. Note
1424 that color bars are different for concentrations at the surface, 200 hPa, and 100
1425 hPa.

1426

1427 **Figure. 8.** (a) Locations with measured aerosol concentrations from previous
1428 studies. Also shown are surface winds during summertime. (b)–(f) show the
1429 comparisons of simulated seasonal mean concentrations of sulfate, nitrate,
1430 ammonium, OC, and BC with measured values, respectively. Also shown in
1431 (b)–(f) are the 1:1 line (dashed), linear fit (solid line and equation), correlation
1432 coefficient between simulated and measured concentrations (R), and

1433 normalized mean bias (NMB) (defined as $NMB = \frac{\sum_{i=1}^n (P_i - O_i)}{\sum_{i=1}^n O_i} \times 100\%$, where P_i
1434 and O_i are predicted and observed concentrations at station i for each aerosol
1435 species).

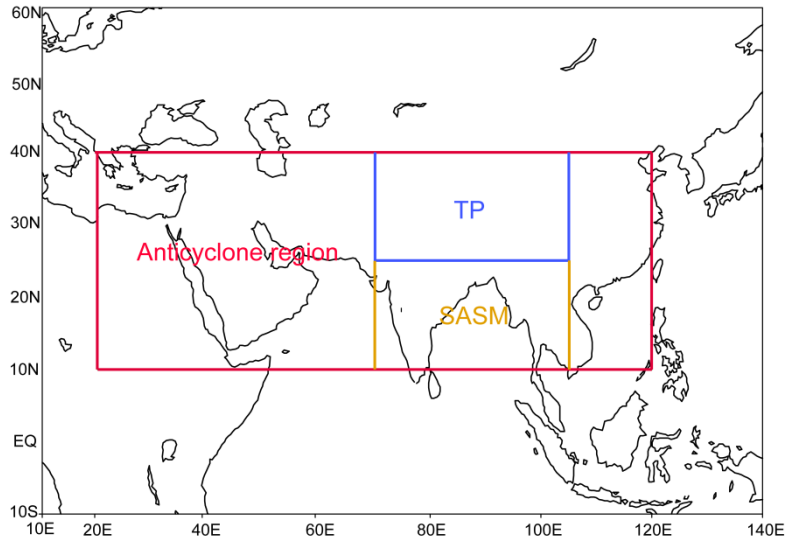
1436
1437 **Figure. 9.** (a) Monthly mean distribution of aerosol extinction coefficients (km^{-1})
1438 at 100 hPa for July of 2005. (b) Monthly mean vertical distributions of aerosol
1439 extinction coefficients (at 525 nm for SAGE II and 550 nm for GEOS-Chem)
1440 (km^{-1}) averaged over the Asian monsoon anticyclone region ($20\text{--}120^\circ\text{E}$,
1441 $10\text{--}40^\circ\text{N}$) for July of 2005. The horizontal dashed line represents the
1442 tropopause averaged over the Asian monsoon anticyclone region simulated by
1443 the GEOS-Chem model.

1444
1445 **Figure. 10.** Simulated contributions of nitrate to $\text{PM}_{2.5}$ ($C_{\text{NIT}} = [\text{NIT}] / [\text{PM}_{2.5}]$
1446 $\times 100\%$) averaged over summer (June-August) of year 2005 at (a)
1447 surface-layer, (b) 200 hPa, and (c) 100 hPa. (d) The latitude-altitude cross
1448 section of simulated C_{NIT} (%) averaged over $70\text{--}105^\circ\text{E}$.

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1450 **Figure. 11.** Latitude-altitude cross sections of simulated concentrations (color
1451 shades, $\mu\text{g m}^{-3}$) of SO_4^{2-} and NO_3^- averaged over $70\text{--}105^\circ\text{E}$ in June-August
1452 of 2005, together with the wind vectors obtained from the European Centre for
1453 Medium-Range Weather Forecasts (ECMWF) ERA-Interim Reanalysis data.
1454 The black line is the tropopause simulated by the GEOS-Chem model.

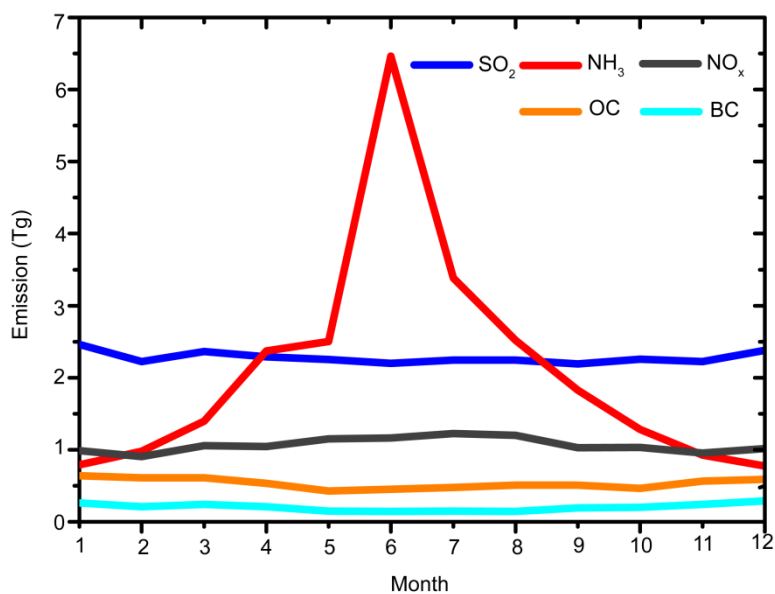
1455
1456 **Figure. 12.** (a)-(b) Distributions of RH (%) and temperature (K) at 100 hPa.
1457 (c)-(d) The latitude-altitude cross sections of RH (%) and temperature (K)
1458 averaged over $70\text{--}105^\circ\text{E}$. RH and temperature are from the GEOS5
1459 assimilated meteorological fields, and all the values are the averages over
1460 June-August of year 2005.

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1462 **Figure. 13.** Mass budget for nitrate aerosol within the selected box of
1463 ($70\text{--}105^\circ\text{E}$, $10\text{--}40^\circ\text{N}$, $8\text{--}16$ km). E/W transport indicates net mass flux through
1464 the east and west lateral boundaries, N/S transport indicates net mass flux
1465 through the north and south lateral boundaries, and upward transport is the net
1466 mass flux through the top and bottom sides of the box. The mass flux is
1467 positive if it increases nitrate mass within the box. Unit of fluxes is Tg season^{-1} .
1468 All the values are the averages over June-August of 2005.



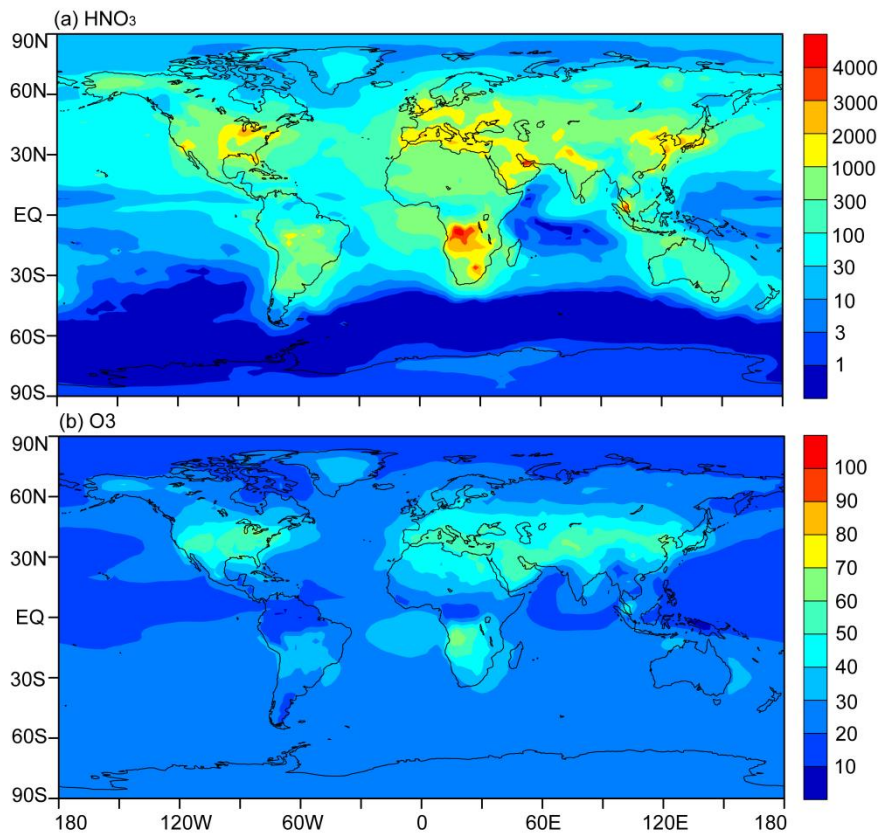
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Figure. 1. Regions examined in this study: the Tibetan Plateau region (TP, 70–105°E, 25–40°N), the SASM region (SASM, 70–105°E, 10–25°N), and the anticyclone region of (20–120°E, 10–40°N).



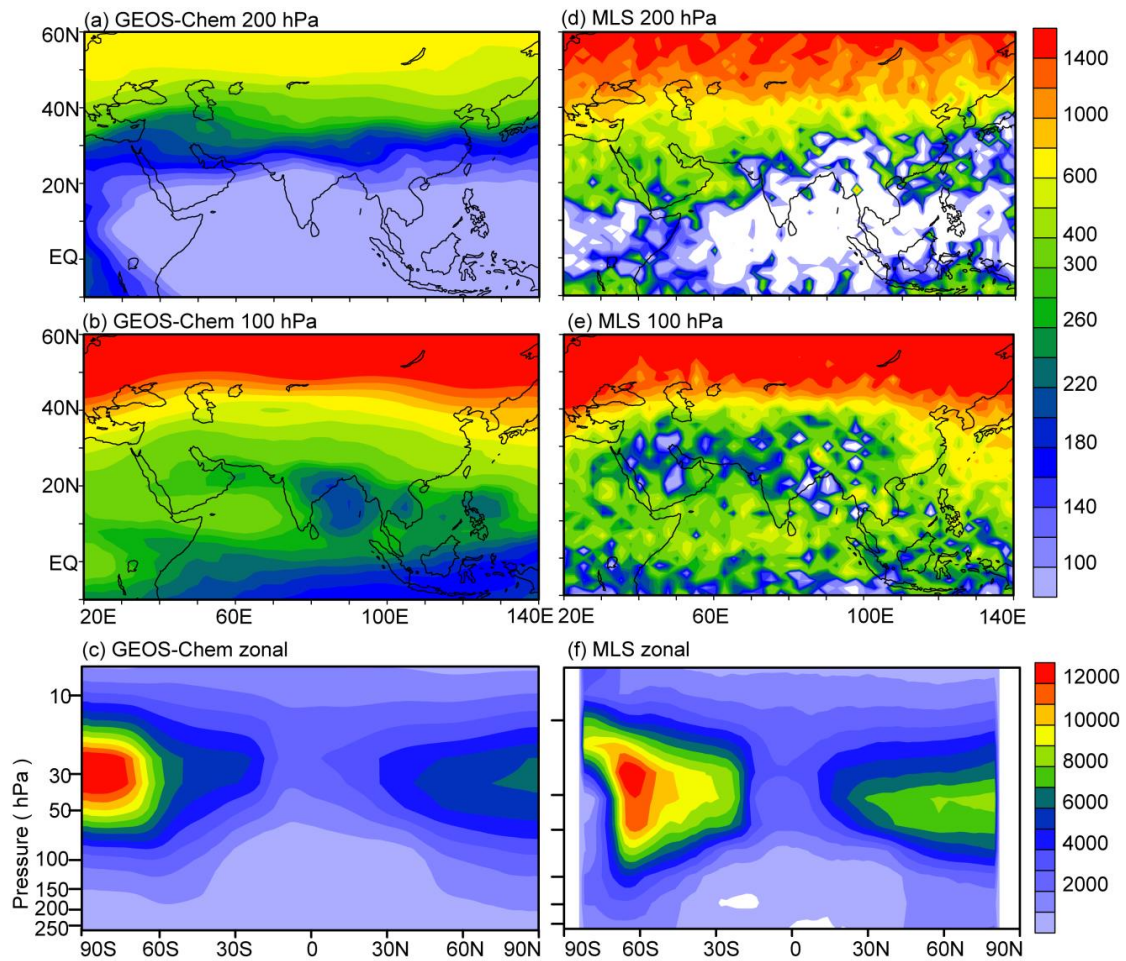
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Figure. 2. Monthly variations in emissions of NO_x (Tg N month⁻¹), SO₂ (Tg S month⁻¹), NH₃ (Tg N month⁻¹), OC (Tg C month⁻¹), and BC (Tg C month⁻¹) over Asia. Values shown are the total emissions (anthropogenic plus natural emissions listed in Table 1).



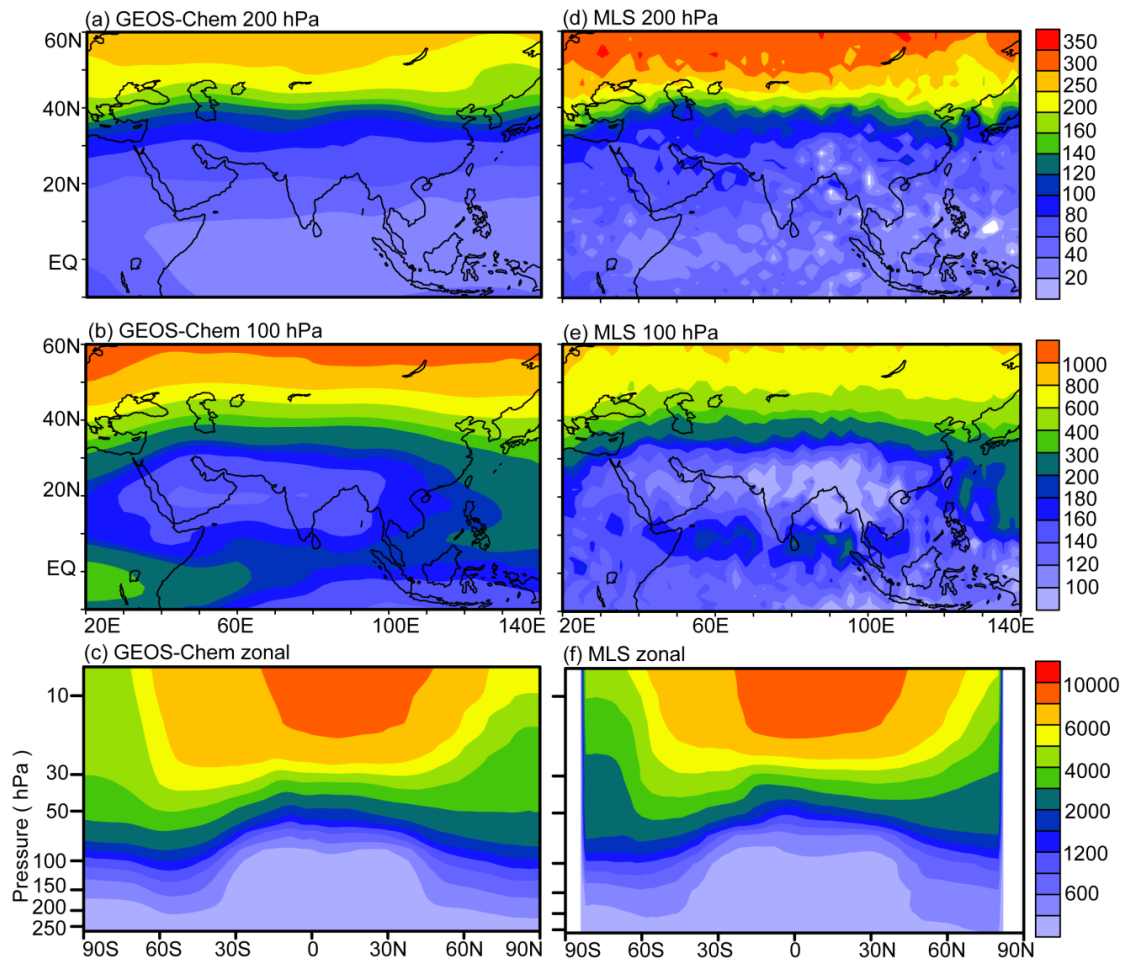
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Figure. 3. Simulated global distributions of surface-layer HNO_3 (pptv) and O_3 (ppbv) averaged over June-August, 2005.



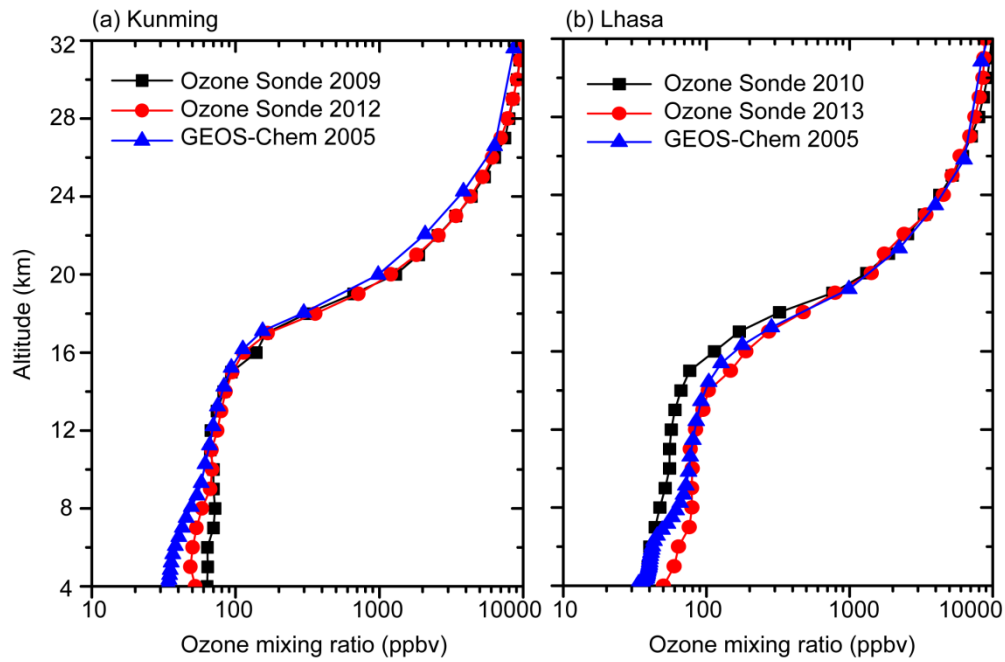
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Figure. 4. Comparisons of simulated HNO_3 concentrations (pptv) with observations (pptv) from MLS. (a) and (b) are simulated concentrations at 200 hPa and 100 hPa, respectively. (c) is the latitude-altitude cross section of simulated HNO_3 concentrations averaged over $70\text{--}105^\circ\text{E}$. (d)-(f) are the same as (a)-(c), except that (d)-(f) are observations from MLS. The white areas in (d) and (f) have no data available from MLS. All the datasets are averaged over June-August of 2005.



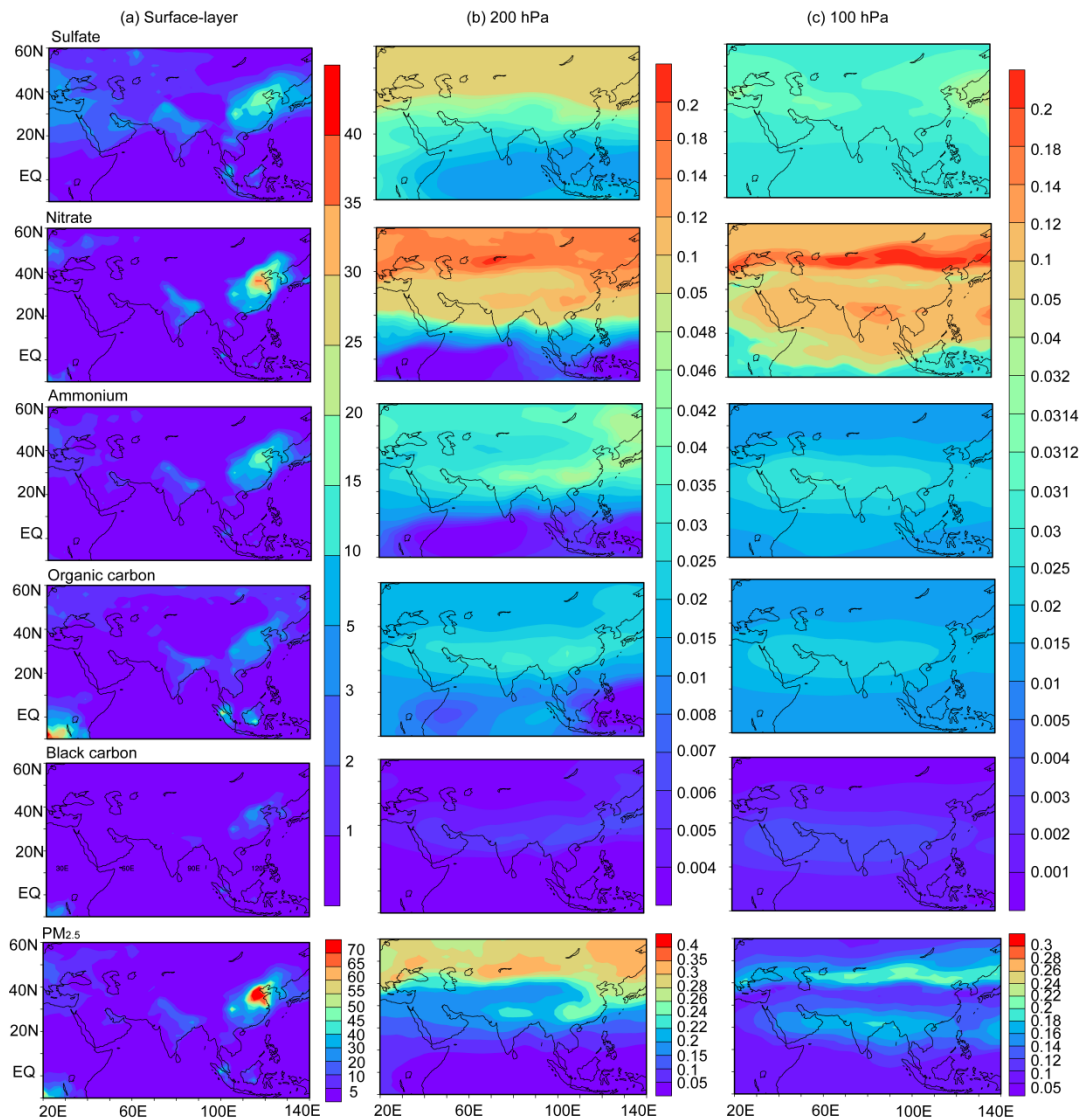
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Figure. 5. Comparisons of simulated O_3 concentrations (ppbv) with observations (ppbv) from MLS. (a) and (b) are simulated concentrations at 200 hPa and 100 hPa, respectively. (c) is the latitude-altitude cross section of simulated O_3 concentrations averaged over 70–105°E. (d)-(f) are the same as (a)-(c), except that (d)-(f) are observations from MLS. All the datasets are averaged over June-August of 2005.



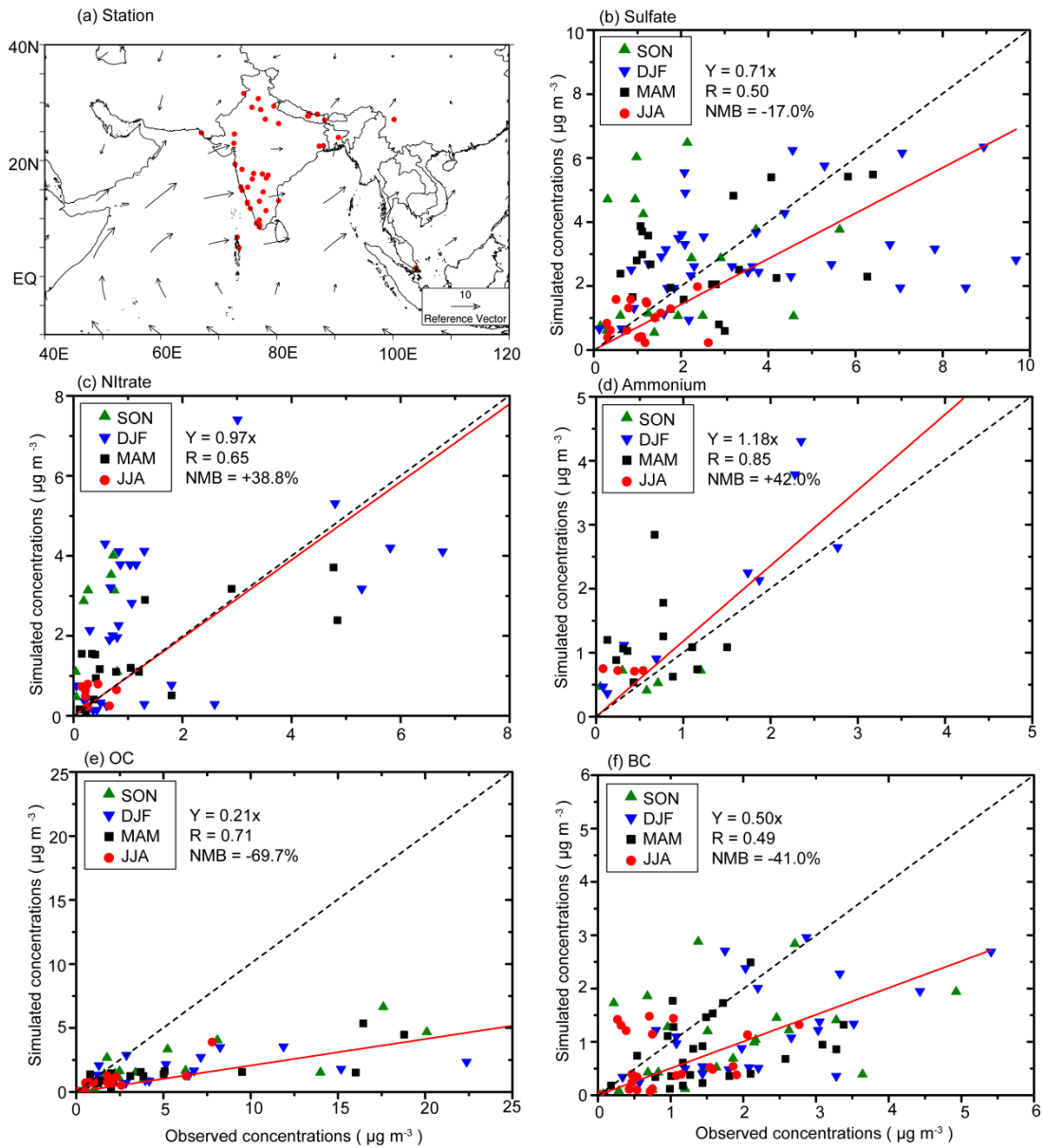
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Figure 6. The simulated and observed vertical profiles of monthly mean O_3 mixing ratios at (a) Kunming and (b) Lhasa in August. The model results are from the simulation of year 2005. The observations in Kunming were conducted during August 7–13 (11 profiles of O_3 collected) in 2009 and during August 12–31 in 2012 (daily observations). The observations in Lhasa were conducted during August 22–28 in 2010 (12 profiles of O_3 collected) and during August 4–26 in 2013 (daily observations).



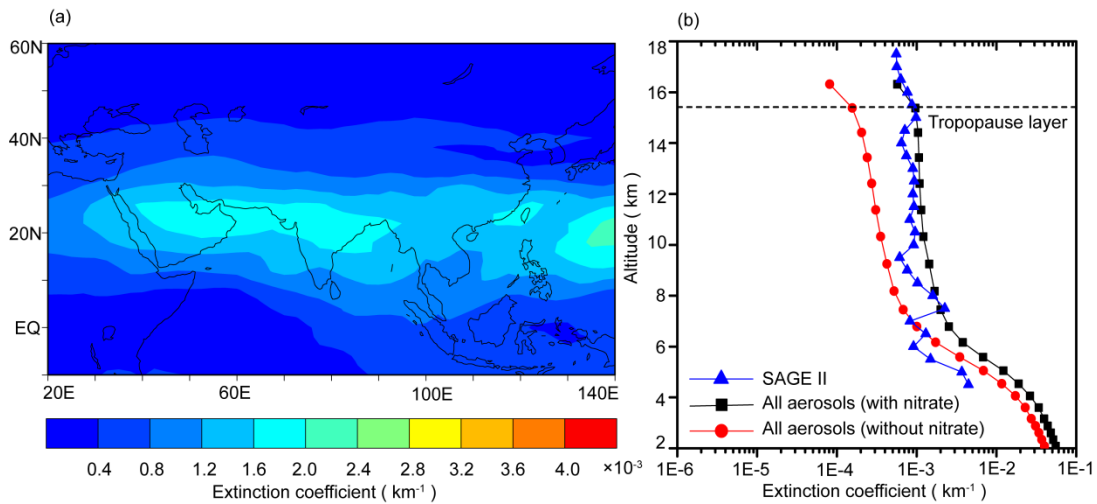
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 1513 ammonium, organic carbon, black carbon, and $\text{PM}_{2.5}$ at (a) the surface layer, (b)
 1514 200 hPa, and (c) 100 hPa, during summer (June-August) of year 2005. Note
 1515 that color bars are different for concentrations at the surface, 200 hPa, and
 1516 hPa.



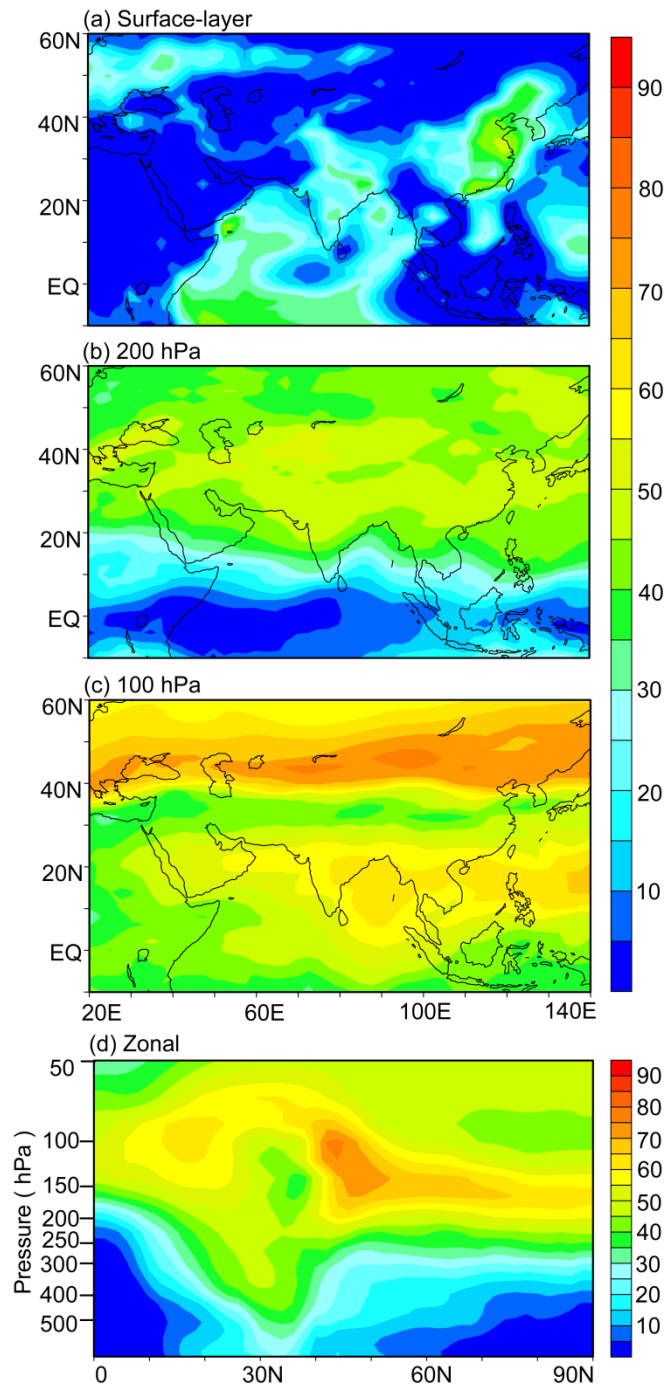
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1525 normalized mean bias (NMB) (defined as $NMB = \frac{\sum_{i=1}^n (P_i - O_i)}{\sum_{i=1}^n O_i} \times 100\%$, where P_i
1526 and O_i are predicted and observed concentrations at station i for each aerosol
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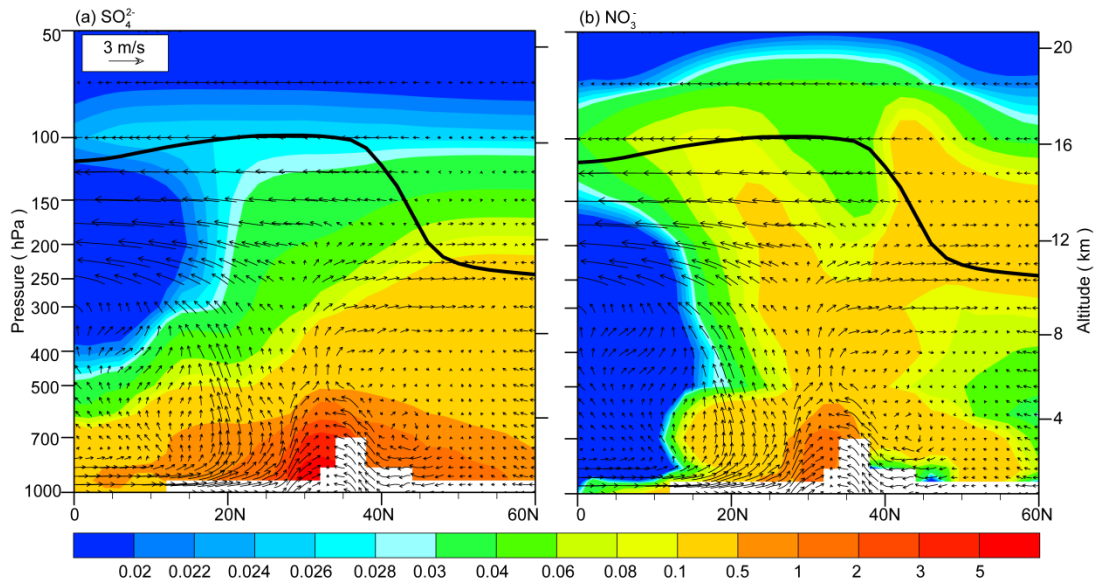
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 1535 tropopause averaged over the Asian monsoon anticyclone region simulated by
 1536 the GEOS-Chem model.



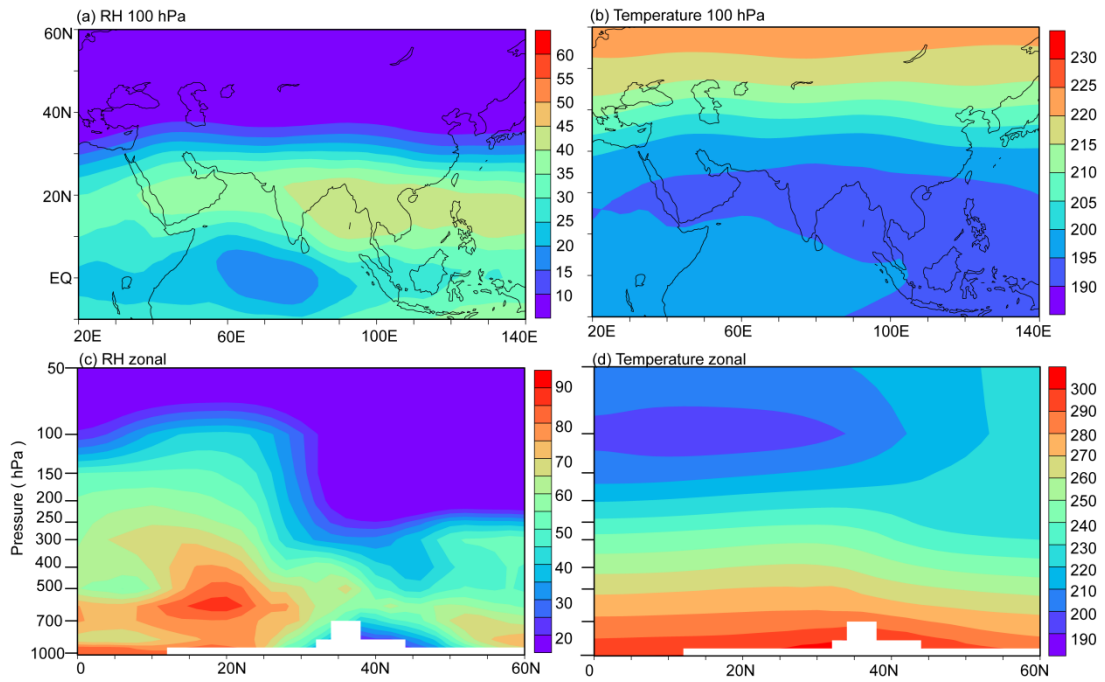
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Figure. 10. Simulated contributions of nitrate to $PM_{2.5}$ ($C_{NIT} = [NIT] / [PM_{2.5}] \times 100\%$) averaged over summer (June-August) of year 2005 at (a) surface-layer, (b) 200 hPa, and (c) 100 hPa. (d) The latitude-altitude cross section of simulated C_{NIT} (%) averaged over 70–105°E.



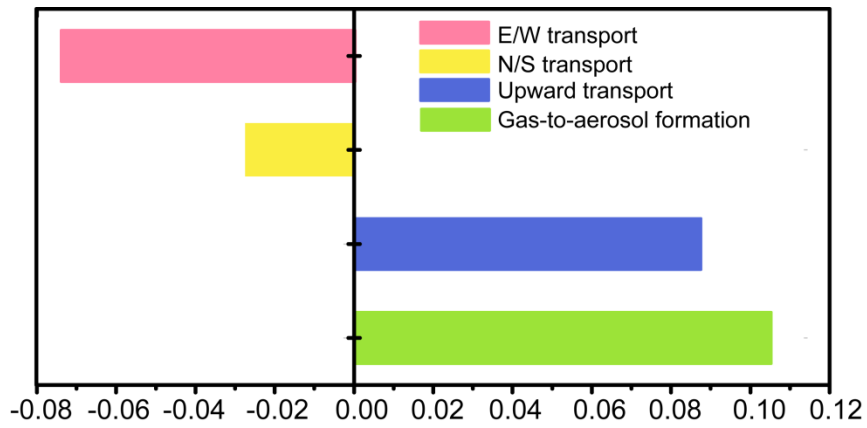
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Figure. 11. Latitude-altitude cross sections of simulated concentrations (color shades, $\mu\text{g m}^{-3}$) of SO_4^{2-} and NO_3^- averaged over $70\text{--}105^\circ\text{E}$ in June-August of 2005, together with the wind vectors obtained from the European Centre for Medium-Range Weather Forecasts (ECMWF) ERA-Interim Reanalysis data. The black line is the tropopause simulated by the GEOS-Chem model.



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Figure. 12. (a)-(b) Distributions of RH (%) and temperature (K) at 100 hPa. (c)-(d) The latitude-altitude cross sections of RH (%) and temperature (K) averaged over 70–105°E. RH and temperature are from the GEOS5 assimilated meteorological fields, and all the values are the averages over June-August of year 2005.



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Figure. 13. Mass budget for nitrate aerosol within the selected box of (70–105°E, 10–40°N, 8–16 km). E/W transport indicates net mass flux through the east and west lateral boundaries, N/S transport indicates net mass flux through the north and south lateral boundaries, and upward transport is the net mass flux through the top and bottom sides of the box. The mass flux is positive if it increases nitrate mass within the box. Unit of fluxes is Tg season⁻¹. All the values are the averages over June-August of 2005.