



Supplement of

Understanding isoprene photo-oxidation using observations and modelling over a subtropical forest in the Southeast US

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3 **1 Algorithms used in ToFIND**

4 **1.1 Peak shape fitting**

5 The peak generated by the PTR-TOF-MS is asymmetric and can be mathematically
6 approximated by peak-shaped functions. Here we choose an empirically transformed
7 Gaussian function (ETG) model to describe a single peak produced by the instrument (Li,
8 1997). The ETG is based on the decomposition of Gaussian function into leading and trailing
9 edge functions. The two functions are weakly combined and subsequently modified, making it
10 very suitable for approximating asymmetric peak shapes. The ETG is given by:

$$11 \quad f(t) = \frac{H}{\{1 + \lambda_L \exp[k_L(t_L - t)]\}^\alpha + \{1 + \lambda_T \exp[k_T(t - t_T)]\}^\beta - 1} \quad (S1)$$

12 where $f(t)$ denotes peak function; t denotes time-of-flight; H is related to peak height; λ_L and
13 λ_T are pre-exponential parameters; k_L and k_T are the parameters related to the speeds of
14 the rise and fall of the leading and trailing edges, respectively; t_L and t_T are the inflection
15 times of the leading and trailing edges, respectively, and are fixed values; α and β are the
16 parameters to further modify the shapes of the leading and trailing edges, respectively. The
17 ETG features in a weak link between the leading and trailing parts of a single peak, thus is
18 very flexible and suitable for fitting asymmetric peaks. The values of t_L and t_T do not need
19 to be accurate, here we use the time-of-flight at the full width at half maximum (FWHM) on
20 the left and right side of a single peak to represent those two values, respectively. There are a
21 total of seven parameters in each ETG (H , λ_L , λ_T , k_L , k_T , α , and β), and the rate of
22 convergence is fast. Levenberg-Marquardt algorithm is used to solve the non-linear least-
23 squares curve fitting problem. To achieve a sufficiently high speed of the curve fitting,
24 MINPACK-1 Least Squares Fitting Library in C
25 (<http://www.physics.wisc.edu/~craigm/idl/cmpfit.html>) is used and integrated into the
26 ToFIND toolbox through MATLAB-C interface (<http://www.imm.dtu.dk/~guuru/>).

27 **1.2 Time-of-flight to m/z conversion**

28 In the field free area of the TOF-MS, the time-of-flight of ions show slow drift due to the
29 change in the length of the ions' flight path caused by temperature variations in this area. To

1 gain a high mass accuracy, continuous mass scale correction is performed on the raw data.
2 Four well identified peaks: three internal ions (H₂18OH⁺, m/z=21.022, NO⁺, m/z=29.997,
3 and H₂16OH₂18OH⁺, m/z=39.033), and one externally added compound (diiodomethane,
4 fragmented at ion CH₂I⁺, m/z=140.920) are used for mass scale correction peaks. The time-
5 of-flight *t* and mass scale *m/z* are related to each other through:

$$6 \quad t = a\sqrt{m/z} + b \quad (S2)$$

7 For each raw HDF5 data file, every 3 of the 300 spectra are averaged to produce 100 spectra.
8 For each of the 100 spectra, the peak shape fitting algorithm described above is applied to
9 find the peak centers (*t*) of the four mass scale correction peaks. Since the exact *m/z* values
10 for each peak are known, the values of *a* and *b* in Eqn. A2 are calculated through linear least-
11 squares method. The relative mass error (RME) is less than 10 parts per million (ppm) over
12 the spectrum after mass scale calibration.

13 **1.3 Peak detection**

14 After time-of-flight to *m/z* conversion, the 300 mass scale corrected spectra are aligned with
15 each other, and averaged to produce one single spectrum. A target mass list is predefined
16 which includes a series exact *m/z* of interested VOCs compounds (e.g., methanol, acetone,
17 isoprene, MVK+MACR, and monoterpenes). For one specific target peak, there may be other
18 peaks co-existing in the same nominal *m/z*, a peak searching routine is used to find all the
19 peaks including the target peak within a certain range of one nominal *m/z*. The mass list is
20 then updated and used for peak shape fitting (see next section).

21 **1.4 Signal integration**

22 A few issues exist when performing signal integration on each individual peak: (1), due to the
23 low counts rate in each individual 1 Hz spectrum, the peak shape fitting algorithm often fails
24 to converge or generate desirable fitting results when applied to a single spectrum, (2),
25 interference exists between two neighbouring peaks when they overlap with each other due to
26 the limited mass resolution of the PTR-TOF-MS. To improve the accuracy of the peak fitting
27 result, the updated peak list is used and the peak shape fitting algorithm is applied to the
28 averaged spectrum described above to produce a peak function. Based on this, the FWHM for
29 each peak is recalculated and used to define the peak integration boundaries. For the
30 asymmetrical Gaussian-shaped peak which is typical for the PTR-TOF-MS, the integration

1 boundaries on the left and right sides are processed separately. With the updated FWHM
2 values for all the peaks, the ratios of the portion before the peak center to the portion after the
3 peak center are calculated and averaged (r). The integration boundary on the left side is
4 defined as $a \cdot \text{FWHM} \cdot r$ to the left of the peak center. The integration boundary on the right
5 side is defined as $a \cdot \text{FWHM} \cdot (1-r)$ to the right of the peak center. Add a figure to describe the
6 procedure. The factor a is set to 4 for single peak and 3 for multiple peaks. The factor value is
7 reduced for multiple peaks to lower the interference of neighbouring peaks.

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Table S1. Other observations during the 2013 SAS campaign used in this study. Days used are during June 5, 6, 8, 10-13 unless otherwise specified in the table.

Parameter	Platform	Instrument/technique	Uncertainty	Days used	Institute
VOC	Airborne (NCAR C-130)	Proton Transfer Reaction Mass Spectrometer (PTR-MS)	20%	June 12	NCAR, NOAA
ISOPN	SEARCH site	chemical ionization mass spectrometer (CIMS)	25%	June 5, 6, 8, 10-13	Purdue University
O3, NOx	Airborne (NCAR C-130)	chemiluminescence	10% for NO, 15% for NO2, 5% for O3	June 12	NCAR
O3, NOx	SEARCH tower	ThermoScientific 49i, 42i	5.5% for NO, 15.0% for NO2, 6.1% for O3	June 5, 6, 8, 10-13	ARA
OH	SEARCH tower	laser induced fluorescence (LIF)	32%	June 5, 6, 8, 10-13	Pennsylvania State University
Surface heat flux	AABC tower	Integrated Surface Flux System (ISFS)		June 10-13	NCAR
Potential temperature	AABC tower	ISFS		June 10-13	NCAR
specific humidity	Airborne (NCAR C-130)	onboard sensor		June 12	NCAR
Potential temperature	Airborne (NCAR C-130)	onboard sensor		June 12	NCAR
specific humidity	Airborne (NCAR C-130)	onboard sensor		June 12	NCAR
Photolysis rates	Airborne (NCAR C-130)	onboard sensor		June 12	NCAR
Boundary layer height	SEARCH	Ceilometer, Integrated Sounding System (ISS)	13.0% for ceilometer	June 10-13	ARA, NCAR

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2 Table S2. Chemical species in the complex scheme. See more details at

3 [http:// http://wiki.seas.harvard.edu/geos-chem/index.php/New_isoprene_scheme](http://wiki.seas.harvard.edu/geos-chem/index.php/New_isoprene_scheme)

Species	Formula	Note
A3O2	CH ₃ CH ₂ CH ₂ OO	primary RO ₂ from C ₃ H ₈
ACET	CH ₃ C(O)CH ₃	acetone
ACTA	CH ₃ C(O)OH	acetic acid
ALD2	CH ₃ CHO	acetaldehyde
ALK4	RH	≥C ₄ alkanes
ATO2	CH ₃ C(O)CH ₂ O ₂	RO ₂ from acetone
ATOOH	CH ₃ C(O)CH ₂ OOH	ATO ₂ peroxide
B3O2	CH ₃ CH(OO)CH ₃	secondary RO ₂ from C ₃ H ₈
C ₂ H ₆	C ₂ H ₆	ethane
C ₃ H ₈	C ₃ H ₈	propane
CH ₂ O	CH ₂ O	formaldehyde
CH ₄	CH ₄	methane
CO	CO	carbon monoxide
CO ₂	CO ₂	carbon dioxide
DHMOB	HOCH ₂ C(CH ₃)(OH)C(=O)CHO	See Paulot et al., ACP (2009)
DIBOO		Dibble peroxy radical
EOH	C ₂ H ₅ OH	ethanol
ETHLN	CHOCH ₂ ONO ₂	Ethanal nitrate
ETO ₂	CH ₃ CH ₂ OO	ethylperoxy radical
ETP	CH ₃ CH ₂ OOH	ethylhydroperoxide
GLYC	HOCH ₂ CHO	glycoaldehyde (hydroxyacetaldehyde)
GLYX	CHOCHO	glyoxal
H ₂	H ₂	hydrogen atom
H ₂ O	H ₂ O	water vapor
H ₂ O ₂	H ₂ O ₂	hydrogen peroxide
HAC	HOCH ₂ C(O)CH ₃	hydroxyacetone
HCOOH	HCOOH	formic acid
HC ₅	HOCH ₂ CH=C(CH ₃)CHO	Hydroxycarbonyl with 5C
HC ₅ OO		Peroxy radical from HC ₅ (old IAO ₂ ?)
HNO ₂	HONO	nitrous acid
HNO ₃	HNO ₃	nitric acid
HNO ₄	HNO ₄	pernitric acid
HO ₂	HO ₂	hydroperoxyl radical
IALD	HOCH ₂ C(CH ₃)=CHCHO	hydroxy carbonyl alkenes from isoprene
IAP	HOCH ₂ C(CH ₃)(OOH)CH(OH)CHO	peroxide from IAO ₂
IEPOX		Isoprene epoxide
IEPOXOO		RO ₂ from IEPOX
INO ₂	O ₂ NOCH ₂ C(OO)(CH ₃)CH=CH ₂	RO ₂ from ISOP+NO ₃
INPN	O ₂ NOCH ₂ C(OOH)(CH ₃)CH=CH ₂	peroxide from INO ₂
ISN1		nighttime isoprene nitrate

ISNOOA		peroxy radical from ISN1
ISNOOB		peroxy radical from ISN1
ISNOHOO		peroxy radical from ISN1
ISNP	HOCH ₂ C(OOH)(CH ₃)CH(ONO ₂)CH ₂ OH	peroxide from ISOPNBO ₂ and ISOPNDO ₂
ISOP	CH ₂ =C(CH ₃)CH=CH ₂	isoprene
ISOPNB	C ₅ H ₉ NO ₄	Isoprene nitrate Beta
ISOPND	C ₅ H ₉ NO ₄	Isoprene nitrate Delta
KO ₂	RO ₂ from >3 ketones	RO ₂ from >3 ketones
M		for three body reactions
MACR	CH ₂ =C(CH ₃)CHO	methacrolein
MACRN	HOCH ₂ C(ONO ₂)(CH ₃)CHO	Nitrate from MVK
MAN ₂	HOCH ₂ C(ONO ₂)(CH ₃)CHO	RO ₂ from MACR+NO ₃
MAO ₃	CH ₂ =C(CH ₃)C(O)OO	peroxyacyl from MVK and MACR
MAOP	CH ₂ =C(CH ₃)C(O)OOH	peroxide from MAO ₃
MAOPO ₂	CH ₂ OH-CHOO*CH ₃ C(O)OOH	Peroxy radical from MAOP (addition on the double bond)
MAP	CH ₃ C(O)OOH	peroxyacetic acid
MCO ₃	CH ₃ C(O)OO	peroxyacetyl radical
MEK	RC(O)R	>3 ketones
MGLY	CH ₃ COCHO	methylglyoxal
MNO ₃	CH ₃ ONO ₂	methylnitrate
MOBA	HOC(=O)C(CH ₃)=CHCHO	5C acid from isoprene
MOBAOO		RO ₂ from MOBA
MO ₂	CH ₃ O ₂	methylperoxy radical
MOH	CH ₃ OH	methanol
MP	CH ₃ OOH	methylhydroperoxide
MRO ₂	HOCH ₂ C(OO)(CH ₃)CHO	RO ₂ from MACR+OH
MRP	HOCH ₂ C(OOH)(CH ₃)CHO	peroxide from MRO ₂
MVK	CH ₂ =CHC(=O)CH ₃	methylvinylketone
MVKN	HOCH ₂ CH(ONO ₂)C(=O)CH ₃	Nitrate from MACR
N ₂	N ₂	nitrogen
N ₂ O	N ₂ O	nitrous oxide
N ₂ O ₅	N ₂ O ₅	dinitrogen pentoxide
NH ₂	NH ₂	ammonia radical
NH ₃	NH ₃	ammonia
NO	NO	nitric oxide
NO ₂	NO ₂	nitrogen dioxide
NO ₃	NO ₃	nitrate radical
O ₂	O ₂	molecular oxygen
O ₂ CH ₂ OH	O ₂ CH ₂ OH	produced by CH ₂ O+HO ₂
O ₃	O ₃	ozone
OH	OH	hydroxyl radical
PAN	CH ₃ C(O)OONO ₂	peroxyacetylnitrate
PMN	CH ₂ =C(CH ₃)C(O)OONO ₂	peroxymethacryloyl nitrate (MPAN)
PO ₂	HOCH ₂ CH(OO)CH ₃	RO ₂ from isoprene
PP	HOCH ₂ CH(OOH)CH ₃	peroxide from PO ₂

PPN	CH ₃ CH ₂ C(O)OONO ₂	peroxypropionyl nitrate
PRN1	O ₂ NOCH ₂ CH(OO)CH ₃	RO ₂ from propene + NO ₃
PRPE	C ₃ H ₆	≥C ₄ alkenes
PRPN	O ₂ NOCH ₂ CH(OOH)CH ₃	peroxide from PRN1
PROPNN	CH ₃ C(=O)CH ₂ ONO ₂	Propanone nitrate
PYAC	CH ₃ COOOH	Pyruvic acid
R4N1	RO ₂ from R4N ₂	RO ₂ from R4N ₂
R4N2	RO ₂ NO	≥C ₄ alkyl nitrates
R4O2	RO ₂ from ALK4	RO ₂ from ALK4
R4P	CH ₃ CH ₂ CH ₂ CH ₂ OOH	peroxide from R4O ₂
RA3P	CH ₃ CH ₂ CH ₂ OOH	peroxide from A3O ₂
RB3P	CH ₃ CH(OOH)CH ₃	peroxide from B3O ₂
RCHO	CH ₃ CH ₂ CHO	>C ₂ aldehydes
RCO3	CH ₃ CH ₂ C(O)OO	peroxypropionyl radical
RCOOH	C ₂ H ₅ C(O)OH	>C ₂ organic acids
RIO1	HOCH ₂ C(OO)(CH ₃)CH=CHOH	RO ₂ from isoprene oxidation products
RIO2	HOCH ₂ C(OO)(CH ₃)CH=CH ₂	RO ₂ from isoprene (named as ISOPO ₂ in the literature)
RIP	HOCH ₂ C(OOH)(CH ₃)CH=CH ₂	peroxide from RIO ₂ (named as ISOPOOH in the literature)
ROH	C ₃ H ₇ OH	>C ₂ alcohols
RP	CH ₃ CH ₂ C(O)OOH	peroxide from RCO ₃
VRO2	HOCH ₂ CH(OO)C(O)CH ₃	RO ₂ from MVK+OH
VRP	HOCH ₂ CH(OOH)C(O)CH ₃	peroxide from VRO ₂

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2 Table S3. The complex chemical scheme used in MXLCH.

Reaction	Reaction rate
$O + O_2 + M \rightarrow O_3 + M$	$6E-34*(300/T)^{2.4}$
$O + O_3 \rightarrow 2O_2$	$8.00E-12*\exp(-2060/T)$
$O_1D + N_2 \rightarrow O + N_2$	$2.10E-11*\exp(115/T)$
$O_1D + O_2 \rightarrow O + O_2$	$3.20E-11*\exp(70/T)$
$O_1D + H_2O \rightarrow 2OH$	$2.20E-10$
$H_2 + O_1D \rightarrow HO_2 + OH$	$1.10E-10$
$H_2 + OH \rightarrow H_2O + HO_2$	$5.50E-12*\exp(-2000/T)$
$O + OH \rightarrow HO_2 + O_2$	$2.20E-11*\exp(120/T)$
$HO_2 + O \rightarrow OH + O_2$	$3.00E-11*\exp(200/T)$
$OH + O_3 \rightarrow HO_2 + O_2$	$1.70E-12*\exp(-940/T)$
$HO_2 + O_3 \rightarrow OH + 2O_2$	$1.00E-14*\exp(-490/T)$ $(2.3E-13*\exp(600/T)+1.7E-33*[M]*\exp(1000/T))*(1 + 1.4E-$
$2HO_2 \rightarrow H_2O_2 + O_2$	$21*[H_2O]*\exp(2200/T)$
$H_2O_2 + OH \rightarrow H_2O + HO_2$	$2.90E-12*\exp(-160/T)$
$OH + HO_2 \rightarrow H_2O + O_2$	$4.80E-11*\exp(250/T)$
$2OH \rightarrow H_2O + O$	$4.20E-12*\exp(-240/T)$
$2OH \rightarrow H_2O_2$	$6.29E-12$
$N_2O + O_1D \rightarrow N_2 + O_2$	$4.90E-11$
$N_2O + O_1D \rightarrow 2NO$	$6.70E-11$
$NO + HO_2 \rightarrow NO_2 + OH$	$3.50E-12*\exp(250/T)$
$NO + O_3 \rightarrow NO_2 + O_2$	$3.00E-12*\exp(-1500.0/T)$
$NO_2 + O \rightarrow NO + O_2$	$5.60E-12*\exp(180/T)$
$NO_2 + O_3 \rightarrow NO_3 + O_2$	$1.20E-13*\exp(-2450/T)$
$NO_3 + HO_2 \rightarrow OH + NO_2$	$2.30E-12*\exp(170/T)$
$NO_2 + NO_3 \rightarrow N_2O_5$	$1.16E-12$
$N_2O_5 \rightarrow NO_2 + NO_3$	$4.70E-02$
$NO_2 + OH \rightarrow HNO_3$	$3.50E-12*\exp(340/T)$
$HNO_3 + OH \rightarrow NO_3$	$1.59E-13$
$NO_3 + NO \rightarrow 2NO_2$	$1.50E-11*\exp(170/T)$
$NO_2 + HO_2 \rightarrow HO_2NO_2$	$1.37E-12$
$HO_2NO_2 + OH \rightarrow H_2O + NO_2 + O_2$	$1.30E-12*\exp(380/T)$
$HO_2NO_2 \rightarrow HO_2 + NO_2$	$8.50E-02$
$CH_4 + OH \rightarrow MO_2 + H_2O$	$2.45E-12*\exp(-1775/T)$
$CH_4 + O_1D \rightarrow .75MO_2 + .75OH + .25CH_2O + .4HO_2 + .05H_2$	$1.50E-10$
$MO_2 + NO \rightarrow CH_2O + NO_2 + HO_2$	$2.80E-12*\exp(300/T)$
$2MO_2 \rightarrow 2CH_2O + 2HO_2$	$5.00E-13*\exp(-424/T)$
$2MO_2 \rightarrow CH_2O + MOH$	$1.90E-14*\exp(706/T)$
$MO_2 + HO_2 \rightarrow MP + O_2$	$4.10E-13*\exp(750/T)$
$MP + OH \rightarrow .7MO_2 + .3OH + .3CH_2O + H_2O$	$3.80E-12*\exp(200/T)$
$CH_2O + OH \rightarrow CO + H_2O + HO_2$	$9.00E-12$
$CO + OH \rightarrow (CO_2) + HO_2$	$2.40E-13$

MOH + OH -> HO2 + CH2O	7.30E-12*exp(-620/T)
ISOP + OH -> RIO2	3.1E-11*exp(350/T)
MACR + OH -> 0.53MAO3 + 0.47MRO2	8.0E-12*exp(380/T)
MVK + OH -> VRO2	2.6E-12*exp(610/T)
PMN + OH -> HAC + CO + NO2	2.90E-11
GLYC + OH -> 0.732CH2O + (0.361CO2) + 0.505CO + 0.227OH + 0.773HO2 + 0.134GLYX + 0.134HCOOH	6.54E-12
GLYC + OH -> HCOOH + OH + CO	1.45E-12
GLYX + OH -> HO2 + 2CO	3.1E-12*exp(340/T)
HAC + OH -> MGLY + HO2	1.81E-12*exp(305/T)
HAC + OH -> 0.5HCOOH + OH + 0.5ACTA + (0.5CO2) + 0.5CO + 0.5MO2	3.43E-13*exp(305/T)
PRPN + OH -> 0.209PRN1 + 0.791OH + 0.791PROPNN	8.78E-12*exp(200/T)
ETP + OH -> 0.64OH + 0.36ETO2 + 0.60ALD2	5.18E-12*exp(200/T)
RA3P + OH -> 0.64OH + 0.36A3O2 + 0.64RCHO	5.18E-12*exp(200/T)
RB3P + OH -> 0.791OH + 0.209B3O2 + 0.791ACET	8.78E-12*exp(200/T)
R4P + OH -> 0.791OH + 0.209R4O2 + 0.791RCHO	8.78E-12*exp(200/T)
RP + OH -> RCO3	6.13E-13*exp(200/T)
PP + OH -> 0.791OH + 0.209PO2 + 0.791HAC	8.78E-12*exp(200/T)
RIP + OH -> 0.387RIO2 + 0.613OH + 0.613HC5	4.75E-12*exp(200/T)
RIP + OH -> OH + IEPOX	1.9E-11*exp(390/T)
IEPOX + OH -> IEPOXOO	5.78E-11*exp(-400/T)
IAP + OH -> 0.654OH + 0.654DHMOB + 0.346HC5OO	5.31E-12*exp(200/T)
VRP + OH -> 0.791OH + 0.791MEK + 0.209VRO2	8.78E-12*exp(200/T)
MRP + OH -> MRO2	1.84E-12*exp(200/T)
MRP + OH -> (CO2) + HAC + OH	4.40E-12*exp(380/T)
MAOP + OH -> MAO3	6.13E-13*exp(200/T)
MAOP + OH -> MAOPO2	3.60E-12*exp(380/T)
OH + MAP -> 1.0MCO3	6.13E-13*exp(200/T)
HC5 + OH -> HC5OO	3.35E-11*exp(380/T)
ISOPND + OH -> ISOPNDO2	2.64E-11*exp(380/T)
ISOPNB + OH -> ISOPNBO2	3.61E-12*exp(380/T)
ISNP + OH -> 0.612OH + 0.612R4N1 + 0.193ISOPNBO2 + 0.193ISOPNDO2	4.75E-12*exp(200/T)
MVKN + OH -> 0.650HCOOH + NO3 + 0.650MGLY + 0.350CH2O + 0.350PYAC	1.5E-12*exp(380/T)
MACRN + OH -> 1.0MACRNO2	1.39E-11*exp(380/T)
DHMOB + OH -> 1.5CO + 1.0HO2 + 0.5HAC + 0.5MEK	2.52E-11*exp(410/T)
MOBA + OH -> MOBAOO	2.79E-11*exp(380/T)
ETHLN + OH -> CH2O + (CO2) + NO2	1.00E-11
PROPNN + OH -> NO2 + MGLY	1.00E-15
ATOOH + OH -> ATO2 + H2O	2.66E-12*exp(200/T)
ATOOH + OH -> MGLY + OH + H2O	1.14E-12*exp(200/T)
R4N2 + OH -> R4N1 + H2O	1.60E-12
RIO2 + NO -> 0.883NO2 + 0.783HO2 + 0.660CH2O + 0.400MVK + 0.260MACR + 0.070ISOPND + 0.047ISOPNB + 0.123HC5 + 0.1DIBOO	2.70E-12*exp(350/T)
VRO2 + NO -> 0.88NO2 + 0.35HO2 + 0.35CH2O + 0.53MCO3 + 0.53GLYC + 0.35MGLY + 0.12MVKN	2.70E-12*exp(350/T)

MRO2 + NO -> 0.85NO2 + 0.85HO2 + 0.122MGLY + 0.728HAC + 0.728CO + 0.122CH2O + 0.15MACRN	2.70E-12*exp(350/T)
MAN2 + NO -> 1.5NO2 + 0.5CH2O + 0.5MGLY + 0.5PROPNN + 0.5CO + 0.5OH	2.70E-12*exp(350/T)
IEPOXOO + NO -> 0.725HAC + 0.275GLYC + 0.275GLYX + 0.275MGLY + 0.125OH + 0.825HO2 + (0.200CO2) + 0.375CH2O + 0.074HCOOH + 0.251CO + NO2	2.70E-12*exp(350/T)
MAOPO2 + NO -> 1.0HAC + (1.0CO2) + 1.0OH + 1.0NO2	2.48E-12
MAOPO2 + NO -> 1.0HNO3	2.24E-13
HC5OO + NO -> NO2 + 0.216GLYX + 0.234MGLY + 0.234GLYC + 0.216HAC + 0.290DHMOB + 0.170MOBA + 0.090RCHO + HO2 + 0.090CO	2.48E-12
HC5OO + NO -> HNO3	3.42E-13
ISOPNDO2 + NO -> 0.070MACRN + 0.310HCOOH + 0.440HAC + 0.130ETHLN + 0.650CH2O + 1.34NO2 + 0.150GLYC + 0.310NO3 + 0.150PROPNN + 0.340MEK + 0.350HO2	2.48E-12
ISOPNDO2 + NO -> HNO3	3.42E-13
ISOPNBO2 + NO -> 0.6GLYC + 0.6HAC + 0.4CH2O + 1.6NO2 + 0.26MACRN + 0.4HO2 + 0.14MVKN	2.48E-12
ISOPNBO2 + NO -> HNO3	3.42E-13
MACRNO2 + NO -> 0.08ACTA + 0.08CH2O + 0.15NO3 + 0.07HCOOH + 0.070MGLY + 0.850HAC + 0.85NO2 + (0.93CO2) + 1.0NO2	2.7E-12*exp(350/T)
DIBOO + NO -> HO2 + NO2 + 0.520GLYC + 0.520MGLY + 0.480HAC + 0.480GLYX	2.48E-12
DIBOO + NO -> HNO3	3.42E-13
MOBAOO + NO -> RCHO + (CO2) + HO2 + NO2	2.48E-12
MOBAOO + NO -> HNO3	3.42E-13
MAN2 + NO -> 1.5NO2 + 0.5CH2O + 0.5MGLY + 0.5PROPNN + 0.5CO + 0.5OH	2.70E-12*exp(350/T)
MCO3 + NO -> MO2 + NO2 + (CO2)	8.10E-12*exp(270/T)
RCO3 + NO -> NO2 + ETO2	6.70E-12*exp(340/T)
MAO3 + NO -> NO2 + 0.5CH2O + 0.5CO + (CO2) + 0.5MO2 + 0.5MCO3	6.70E-12*exp(340/T)
ATO2 + NO -> 0.96NO2 + 0.960CH2O + 0.960MCO3 + 0.04R4N2	2.80E-12*exp(300/T)
RIO2 + HO2 -> 0.88RIP + 0.12OH + 0.047MACR + 0.073MVK + 0.12HO2 + 0.12CH2O	2.06E-13*exp(1300/T)
VRO2 + HO2 -> 0.1VRP + 0.68OH + 0.578GLYC + 0.578MCO3 + 0.187MEK + 0.102HO2 + 0.102CH2O + 0.102MGLY + 0.033RCHO	1.82E-13*exp(1300/T)
MRO2 + HO2 -> 1.0MRP	1.82E-13*exp(1300/T)
MAN2 + HO2 -> 0.075PROPNN + 0.075CO + 0.075HO2 + 0.075MGLY + 0.075CH2O + 0.075NO2 + 0.15OH + 0.85ISNP	1.82E-13*exp(1300/T)
IEPOXOO + HO2 -> 0.725HAC + 0.275GLYC + 0.275GLYX + 0.275MGLY + 1.125OH + 0.825HO2 + (0.200CO2) + 0.375CH2O + 0.074HCOOH + 0.251CO	2.06E-13*exp(1300/T)
DIBOO + HO2 -> 0.15HO2 + 0.15OH + 0.078GLYC + 0.078MGLY + 0.072HAC + 0.072GLYX + 0.85R4P	2.06E-13*exp(1300/T)
MAOPO2 + HO2 -> 1.0HAC + (1.0CO2) + 2.0OH	1.82E-13*exp(1300/T)
HC5OO + HO2 -> 0.1IAP + 0.9OH + 0.9MGLY + 0.9GLYC + 0.9HO2	2.06E-13*exp(1300/T)
ISOPNDO2 + HO2 -> 0.5OH + 0.035MACRN + 0.155HCOOH + 0.22HAC + 0.065ETHLN + 0.325CH2O + 0.170NO2 + 0.075GLYC + 0.155NO3 + 0.075PROPNN + 0.170MEK + 0.175HO2 + 0.5ISNP	2.06E-13*exp(1300/T)

ISOPNBO2 + HO2 -> 0.3GLYC + 0.3HAC + 0.2CH2O + 0.13MACRN + 0.07MVKN + 0.3NO2 + 0.2HO2 + 0.5OH + 0.5ISNP	2.06E-13*exp(1300/T)
MACRNO2 + HO2 -> 0.08ACTA + 0.08CH2O + 0.15NO3 + 0.07HCOOH + 0.07MGLY + 0.85HAC + 0.85NO2 + (0.93CO2) + 1.0OH	1.82E-13*exp(1300/T)
MOBAOO + HO2 -> 0.15OH + 0.15HO2 + 0.15RCHO + (0.15CO2) + 0.85R4P	2.06E-13*exp(1300/T)
MCO3 + HO2 -> 0.15ACTA + 0.15O3 + 0.44OH + 0.44MO2 + 0.41MAP	5.20E-13*exp(980/T)
RCO3 + HO2 -> 0.410RP + 0.150RCOOH + 0.150O3 + 0.440OH + 0.440ETO2	4.30E-13*exp(1040/T)
ATO2 + HO2 -> 0.15MCO3 + 0.15OH + 0.15CH2O + 0.85ATOOH	8.60E-13*exp(700/T)
KO2 + HO2 -> 0.15OH + 0.15ALD2 + 0.15MCO3 + 0.85ATOOH	1.82E-13*exp(1300/T)
MAO3 + HO2 -> 0.44OH + 0.15O3 + 0.59CH2O + 0.39MO2 + 0.41MAOP + 0.39CO	4.30E-13*exp(1040/T)
RIO2 + MO2 -> 1.1HO2 + 1.22CH2O + 0.280MVK + 0.180MACR + 0.3HC5 + 0.24MOH + 0.24ROH	8.37E-14
HC5OO + MO2 -> 0.50HO2 + 0.33CO + 0.09H2 + 0.18HAC + 0.13GLYC + 0.29MGLY + 0.25MEK + 0.95CH2O + 0.25MOH + 0.25ROH + 0.5HO2	8.37E-14
MRO2 + MO2 -> 0.595HAC + 0.255MGLY + 0.595CO + 1.255CH2O + 1.7HO2 + 0.150ROH	8.37E-14
VRO2 + MO2 -> 0.14HO2 + 0.14CH2O + 0.36MCO3 + 0.36GLYC + 0.14MGLY + 0.25MEK + 0.75CH2O + 0.25MOH + 0.25ROH + 0.5HO2	8.37E-14
MAN2 + MO2 -> 0.375PROPNN + 0.375CO + 0.375HO2 + 0.375MGLY + 0.375CH2O + 0.375NO2 + 0.250CH2O + 0.250R4N2	8.37E-14
MAOPO2 + MO2 -> 0.7HAC + (0.7CO2) + 0.7OH + 1.0CH2O + 0.7HO2 + 0.3ROH	8.37E-14
2RIO2 -> 1.28HO2 + 0.92CH2O + 0.56MVK + 0.36MACR + 0.48ROH + 0.5HC5	1.54E-13
2MAOPO2 -> 2.0HAC + (2.0CO2) + 2.0OH	8.37E-14
MCO3 + MO2 -> CH2O + MO2 + HO2	1.80E-12*exp(500/T)
MCO3 + MO2 -> ACTA + CH2O	2.00E-13*exp(500/T)
RCO3 + MO2 -> CH2O + HO2 + ETO2	1.68E-12*exp(500/T)
RCO3 + MO2 -> RCOOH + CH2O	1.87E-13*exp(500/T)
MAO3 + MO2 -> CH2O + HO2 + CH2O + MCO3	1.68E-12*exp(500/T)
MAO3 + MO2 -> RCOOH + CH2O	1.87E-13*exp(500/T)
MAOPO2 + MCO3 -> 1.0HAC + (2.0CO2) + OH + MO2	1.68E-12*exp(500/T)
MAOPO2 + MCO3 -> 1.0ACTA + 1.0MEK	1.87E-13*exp(500/T)
R4O2 + MCO3 -> MO2 + 0.32ACET + 0.19MEK + 0.27HO2 + 0.32ALD2 + 0.13RCHO + 0.05A3O2 + 0.18B3O2 + 0.32ETO2	1.68E-12*exp(500/T)
R4O2 + MCO3 -> 1.0ACTA + 1.0MEK	1.87E-13*exp(500/T)
ATO2 + MCO3 -> MCO3 + CH2O + MO2	1.68E-12*exp(500/T)
ATO2 + MCO3 -> MGLY + ACTA	1.87E-13*exp(500/T)
HC5OO + MCO3 -> 0.216GLYX + 0.234MGLY + 0.234GLYC + 0.216HAC + 0.29DHMOB + 0.17MOBA + 0.09RCHO + HO2 + 0.09CO + MO2	1.68E-12*exp(500/T)
HC5OO + MCO3 -> MEK + ACTA	1.87E-13*exp(500/T)
VRO2 + MCO3 -> 0.4HO2 + 0.4CH2O + 0.6MCO3 + 0.6GLYC + 0.4MGLY + 1.0MO2	1.68E-12*exp(500/T)

VRO2 + MCO3 -> MEK + ACTA	1.87E-13*exp(500/T)
MRO2 + MCO3 -> 0.850HO2 + 0.143MGLY + 0.857HAC + 0.857CO + 0.143CH2O + 1.0MO2	1.68E-12*exp(500/T)
MRO2 + MCO3 -> MEK + ACTA	1.87E-13*exp(500/T)
MAN2 + MCO3 -> 0.5PROPNN + 0.5CO + 0.5HO2 + 0.5MGLY + 0.5CH2O + 0.5NO2 + (CO2) + MO2	1.68E-12*exp(500/T)
MAN2 + MCO3 -> RCHO + ACTA + NO2	1.87E-13*exp(500/T)
RIO2 + MCO3 -> 0.887HO2 + 0.747CH2O + 0.453MVK + 0.294MACR + 0.140HC5 + 0.113DIBOO + (CO2) + MO2	1.68E-12*exp(500/T)
RIO2 + MCO3 -> MEK + ACTA	1.87E-13*exp(500/T)
MCO3 + MCO3 -> 2MO2	2.50E-12*exp(500/T)
RCO3 + MCO3 -> MO2 + ETO2	2.50E-12*exp(500/T)
MAO3 + MCO3 -> MO2 + MCO3 + CH2O	2.50E-12*exp(500/T)
MCO3 + NO2 -> PAN	8.59E-12
PAN -> MCO3 + NO2	5.73E-04
RCO3 + NO2 -> PPN	7.40E-12
PPN -> RCO3 + NO2	4.44E-04
MAO3 + NO2 -> PMN	7.40E-12
PMN -> MAO3 + NO2	4.44E-04
MACRNO2 + NO2 -> PMNN	7.40E-12
PMNN -> MACRNO2 + NO2	4.44E-04
ISOP + O3 -> 0.244MVK + 0.325MACR + 0.845CH2O + 0.110H2O2 + 0.522CO + 0.204HCOOH + 0.199MCO3 + 0.026HO2 + 0.270OH + 0.128PRPE + 0.051MO2	1.00E-14*exp(-1970/T)
MVK + O3 -> 0.202OH + 0.202HO2 + 0.352HCOOH + 0.535CO + 0.050ALD2 + 0.950MGLY + 0.050CH2O	8.50E-16*exp(-1520/T)
MACR + O3 -> 0.261OH + 0.202HO2 + 0.326HCOOH + 0.569CO + 0.880MGLY + 0.120CH2O	1.40E-15*exp(-2100/T)
HC5 + O3 -> 0.6MGLY + 0.1OH + 0.12CH2O + 0.28GLYC + 0.3O3 + 0.4CO + 0.2H2 + 0.2HAC + 0.2HCOOH	6.16E-15*exp(-1814/T)
ISOPNB + O3 -> 0.610MVKN + 0.390MACRN + 0.27OH + CH2O	1.06E-16
ISOPND + O3 -> 0.5PROPNN + 0.5ETHLN + 0.27OH + 0.5GLYC + 0.5HAC	5.30E-17
MOBA + O3 -> OH + HO2 + (CO2) + MEK	2.00E-17
PMN + O3 -> NO2 + 0.6CH2O + HO2	8.20E-18
RIO2 -> 2.0HO2 + 1.0CH2O + 0.5MGLY + 0.5GLYC + 0.5GLYX + 0.500HAC + 1.0OH	4.07E+08*exp(-7694/T)
MRO2 -> 1.0CO + 1.0HAC + 1.0OH	2.90E+07*exp(-5297/T)
C10H16 + OH -> TERPO2	1.20E-11*exp(444/T)
C10H16 + O3 -> .7OH + MVK + MACR + HO2	1.00E-15*exp(-732/T)
C10H16 + NO3 -> TERPO2 + NO2	1.20E-12*exp(490/T)
TERPO2 + NO -> .1ACET + MVK + MACR + NO2	4.20E-12*exp(180/T)
TERPO2 + HO2 -> TERPOOH	7.50E-13*exp(700/T)
TERPOOH + OH -> TERPO2	3.80E-12*exp(200/T)
O3 -> O2 + O1D	2.80E-04*exp(-1.75/cos(x))
O3 -> O2 + O	6.76E-04*exp(-0.30/cos(x))
H2O2 -> 2OH	2.10E-05*exp(-0.86/cos(x))
NO2 -> NO + O	1.71E-02*exp(-0.55/cos(x))
NO3 -> NO + O2	3.03E-02*exp(-0.22/cos(x))

NO3 -> NO2 + O	2.49E-01*exp(-0.25/cos(x))
N2O5 -> NO3 + NO + O	1.02E-06*exp(-5.50/cos(x))
N2O5 -> NO3 + NO2	1.25E-04*exp(-0.85/cos(x))
HNO3 -> OH + NO2	2.8E-06*exp(-1.20/cos(x))
HO2NO2 -> HO2 + NO2	2.05E-05*exp(-1.20/cos(x))
CH2O -> 2HO2 + CO	1.05E-04*exp(-1.00/cos(x))
CH2O -> H2 + CO	1.25E-04*exp(-0.80/cos(x))
MP -> CH2O + HO2 + OH	1.65E-05*exp(-0.88/cos(x))
ALD2 -> MO2 + HO2 + CO	2.95E-05*exp(-1.40/cos(x))
ALD2 -> CH4 + CO	1.25E-08*exp(-6.00/cos(x))
PAN -> 0.6MCO3 + 0.6NO2 + 0.4MO2 + 0.4NO3	2.5E-06*exp(-1.00/cos(x))
RCHO -> ETO2 + HO2 + CO	7.35E-05*exp(-1.10/cos(x))
ACET -> MCO3 + MO2	3.76E-06*exp(-1.80/cos(x))
MEK -> 0.85MCO3 + 0.85ETO2 + 0.15MO2 + 0.15RCO3	3.05E-05*exp(-1.20/cos(x))
GLYC -> CH2O + 2.0HO2 + CO	2.20E-05*exp(-1.20/cos(x))
GLYX -> 2.0CO + 2.0HO2	1.16E-04*exp(-0.45/cos(x))
MGLY -> MCO3 + CO + HO2	2.60E-04*exp(-0.60/cos(x))
MVK -> PRPE + CO	6.72E-06*exp(-0.77/cos(x))
MVK -> MCO3 + CH2O + CO + HO2	2.24E-06*exp(-0.77/cos(x))
MVK -> MO2 + MAO3	2.24E-06*exp(-0.77/cos(x))
MACR -> MAO3 + HO2	5.85E-06*exp(-0.60/cos(x))
MACR -> CO + HO2 + CH2O + MCO3	5.85E-06*exp(-0.60/cos(x))
HAC -> MCO3 + CH2O + HO2	4.05E-06*exp(-1.10/cos(x))
INPN -> OH + HO2 + RCHO + NO2	1.65E-05*exp(-0.88/cos(x))
PRPN -> OH + HO2 + RCHO + NO2	1.65E-05*exp(-0.88/cos(x))
ETP -> OH + HO2 + ALD2	1.65E-05*exp(-0.88/cos(x))
RA3P -> OH + HO2 + RCHO	1.65E-05*exp(-0.88/cos(x))
RB3P -> OH + HO2 + ACET	1.65E-05*exp(-0.88/cos(x))
R4P -> OH + HO2 + RCHO	1.65E-05*exp(-0.88/cos(x))
PP -> OH + HO2 + ALD2 + CH2O	1.65E-05*exp(-0.88/cos(x))
RP -> OH + HO2 + ALD2	1.65E-05*exp(-0.88/cos(x))
RIP -> OH + HO2 + 0.710CH2O + 0.425MVK + 0.285MACR + 0.29HC5	1.65E-05*exp(-0.88/cos(x))
IAP -> OH + HO2 + 0.67CO + 0.190H2 + 0.36HAC + 0.26GLYC + 0.580MGLY	1.65E-05*exp(-0.88/cos(x))
ISNP -> OH + HO2 + RCHO + NO2	1.65E-05*exp(-0.88/cos(x))
VRP -> OH + 0.3HO2 + 0.3CH2O + 0.7MCO3 + 0.7GLYC + 0.3MGLY	1.65E-05*exp(-0.88/cos(x))
MRP -> OH + HO2 + HAC + CO + CH2O	1.65E-05*exp(-0.88/cos(x))
MAOP -> OH + CH2O + MCO3	1.65E-05*exp(-0.88/cos(x))
R4N2 -> NO2 + 0.320ACET + 0.190MEK + 0.180MO2 + 0.270HO2 + 0.320ALD2 + 0.130RCHO + 0.050A3O2 + 0.180B3O2 + 0.320ETO2	3.94E-06*exp(-1.20/cos(x))
MAP -> OH + MO2	1.65E-05*exp(-0.88/cos(x))
MACRN -> NO2 + HAC + MGLY + 0.5CH2O + HO2 + 0.5CO	3.50E-4*exp(-1.00/cos(x))
MVKN -> GLYC + NO2 + MCO3	2.50E-06*exp(-1.00/cos(x))
ISOPNB -> HC5 + NO2 + HO2	2.50E-06*exp(-1.00/cos(x))
ISOPND -> HC5 + NO2 + HO2	2.50E-06*exp(-1.00/cos(x))
PROPNN -> CH2O + NO2 + CO + MO2	2.50E-06*exp(-1.00/cos(x))
ATOOH -> OH + CH2O + MCO3	1.65E-05*exp(-0.88/cos(x))

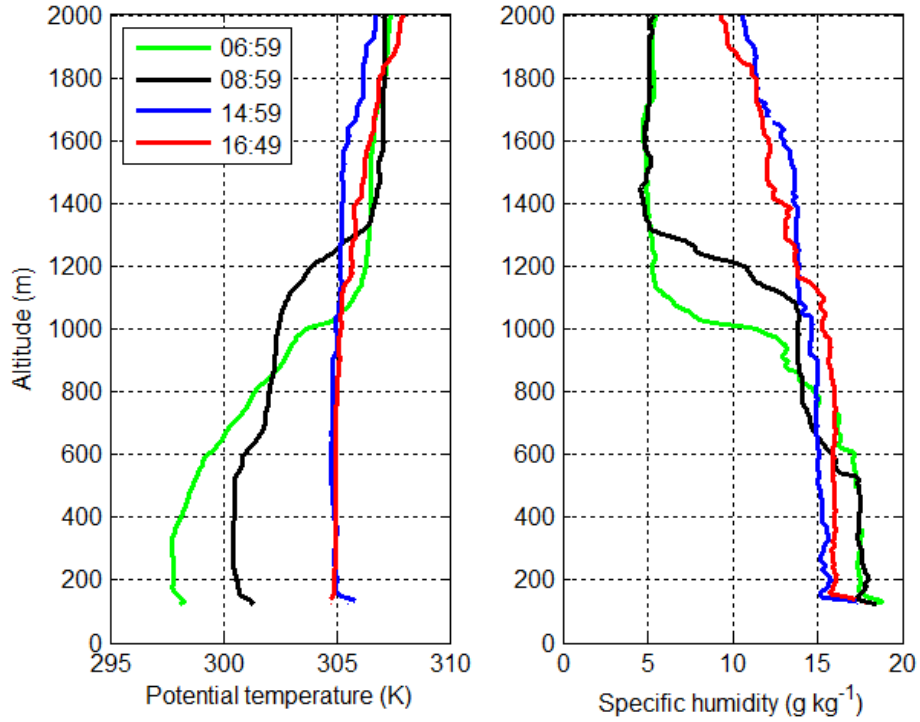
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2 Table S4. The relationship between solar zenith angle (sza) and photolysis rates ($j =$
3 $a \cdot \exp(b/\cos(\text{sza}))$), which are obtained from the TUV radiation model. Adjusted R^2 indicates
4 the adjusted correlation coefficient for each curve fitting result. The last two columns show
5 the photolysis rates obtained from TUV simulations and NCAR C-130 observations at 10:37
6 CST (which corresponds to sza of 0.32), respectively. J-value observations onboard NCAR C-
7 130 over the AABC and SEARCH sites are available on 06/14/2013. The j-values collected
8 below pressure altitude of 1000 m while over the two ground sites are averaged and used here
9 (sampling time ranges between 10:02 CST and 11:11 CST with middle sampling time at
10 10:37 CST). NA indicates no observations were available for the specific reaction.

Number	Reaction	a	b	Adjusted R^2	TUV (s-1)	C-130 (s-1)
R_01	$\text{O}_3 \rightarrow \text{O}_2 + \text{O}(1\text{D})$	2.80E-4	-1.75	0.996	4.47E-05	3.74E-05
R_02	$\text{O}_3 \rightarrow \text{O}_2 + \text{O}(3\text{P})$	6.76E-4	-0.30	0.996	4.92E-04	NA
R_03	$\text{H}_2\text{O}_2 \rightarrow 2 \text{OH}$	2.10E-5	-0.86	0.993	8.42E-06	7.96E-06
R_04	$\text{NO}_2 \rightarrow \text{NO} + \text{O}(3\text{P})$	1.88E-2	-0.50	0.999	1.11E-02	9.48E-03
R_05	$\text{NO}_3 \rightarrow \text{NO} + \text{O}_2$	3.03E-2	-0.22	0.998	2.40E-02	NA
R_06	$\text{NO}_3 \rightarrow \text{NO}_2 + \text{O}(3\text{P})$	2.49E-1	-0.25	0.999	1.91E-01	NA
R_07	$\text{N}_2\text{O}_5 \rightarrow \text{NO}_3 + \text{NO} + \text{O}(3\text{P})$	1.02E-6	-5.50	0.998	3.03E-09	NA
R_08	$\text{N}_2\text{O}_5 \rightarrow \text{NO}_3 + \text{NO}_2$	1.25E-4	-0.85	0.994	5.09E-05	4.73E-05
R_09	$\text{HNO}_2 \rightarrow \text{OH} + \text{NO}$	3.85E-3	-0.50	0.995	2.26E-03	1.54E-03
R_10	$\text{HNO}_3 \rightarrow \text{OH} + \text{NO}_2$	2.80E-6	-1.20	0.993	7.93E-07	6.82E-07
R_11	$\text{HNO}_4 \rightarrow \text{HO}_2 + \text{NO}_2$	2.05E-5	-1.20	0.995	5.80E-06	7.22E-06
R_12	$\text{CH}_2\text{O} \rightarrow \text{H} + \text{HCO}$	1.05E-4	-1.00	0.996	3.68E-05	3.53E-05
R_13	$\text{CH}_2\text{O} \rightarrow \text{H}_2 + \text{CO}$	1.25E-4	-0.80	0.990	5.37E-05	5.38E-05
R_14	$\text{CH}_3\text{CHO} \rightarrow \text{CH}_3 + \text{HCO}$	2.95E-5	-1.40	0.996	6.74E-06	5.79E-06
R_15	$\text{CH}_3\text{CHO} \rightarrow \text{CH}_4 + \text{CO}$	1.25E-8	-6.00	0.992	2.16E-11	NA
R_16	$\text{C}_2\text{H}_5\text{CHO} \rightarrow \text{C}_2\text{H}_5 + \text{HCO}$	7.35E-5	-1.10	0.988	2.28E-05	1.94E-05
R_17	$\text{CH}_2(\text{OH})\text{CHO} \rightarrow \text{Products}$	2.20E-5	-1.20	0.995	6.23E-06	NA
R_18	$\text{CH}_2=\text{C}(\text{CH}_3)\text{CHO} \rightarrow \text{Products}$	1.17E-5	-0.60	0.996	6.21E-06	NA
R_19	$\text{CH}_3\text{COCH}_3 \rightarrow \text{CH}_3\text{CO} + \text{CH}_3$	3.76E-6	-1.80	0.992	5.61E-07	1.20E-06
R_20	$\text{CH}_3\text{COCH}_2\text{CH}_3 \rightarrow \text{Products}$	1.12E-5	-0.77	0.991	4.92E-06	NA
R_21	$\text{CH}_3\text{COCH}_2\text{CH}_3 \rightarrow \text{CH}_3\text{CO} + \text{CH}_2\text{CH}_3$	3.05E-5	-1.20	0.994	8.52E-06	7.26E-06
R_22	$\text{CH}_2(\text{OH})\text{COCH}_3 \rightarrow \text{CH}_3\text{CO} + \text{CH}_2(\text{OH})$	4.05E-6	-1.10	0.990	1.26E-06	NA
R_23	$\text{CH}_3\text{OOH} \rightarrow \text{CH}_3\text{O} + \text{OH}$	1.65E-5	-0.88	0.991	6.44E-06	NA
R_24	$\text{CH}_3\text{CO}(\text{OOH}) \rightarrow \text{Products}$	2.74E-6	-1.00	0.994	9.68E-07	NA
R_25	$\text{CH}_3\text{ONO}_2 \rightarrow \text{CH}_3\text{O} + \text{NO}_2$	3.94E-6	-1.20	0.994	1.12E-06	NA
R_26	$\text{CH}_3\text{CO}(\text{OONO}_2) \rightarrow \text{CH}_3\text{CO}(\text{OO}) + \text{NO}_2$	1.51E-6	-1.00	0.992	5.26E-07	5.54E-07
R_27	$\text{CH}_3\text{CO}(\text{OONO}_2) \rightarrow \text{CH}_3\text{CO}(\text{O}) + \text{NO}_3$	1.00E-6	-1.00	0.993	3.51E-07	2.37E-07
R_28	$\text{CH}_3\text{CH}_2\text{CO}(\text{OONO}_2) \rightarrow \text{CH}_3\text{CH}_2\text{CO}(\text{OO}) + \text{NO}_2$	2.50E-6	-1.00	0.992	8.68E-07	NA

R_29	CHOCHO -> HCO + HCO	1.16E-4	-0.45	0.998	7.22E-05	7.59E-05
R_30	CH3COCHO -> CH3CO + HCO	2.60E-4	-0.60	0.996	1.38E-04	1.45E-04

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5 Figure S1. Vertical profiles of potential temperature and specific humidity collected from the
6 sounding system at the SEARCH site on 2013-06-11. Time series shown in the legend are in
7 CST format.

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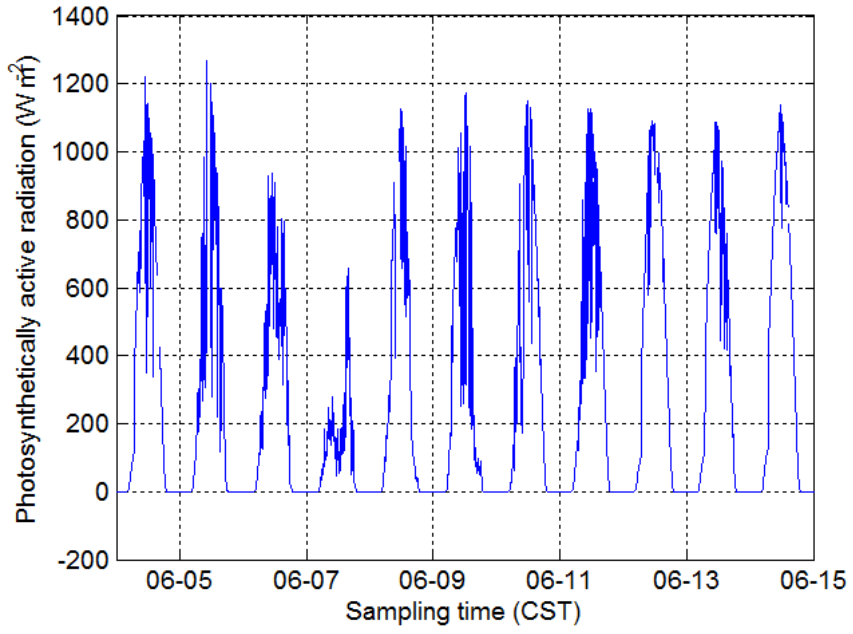
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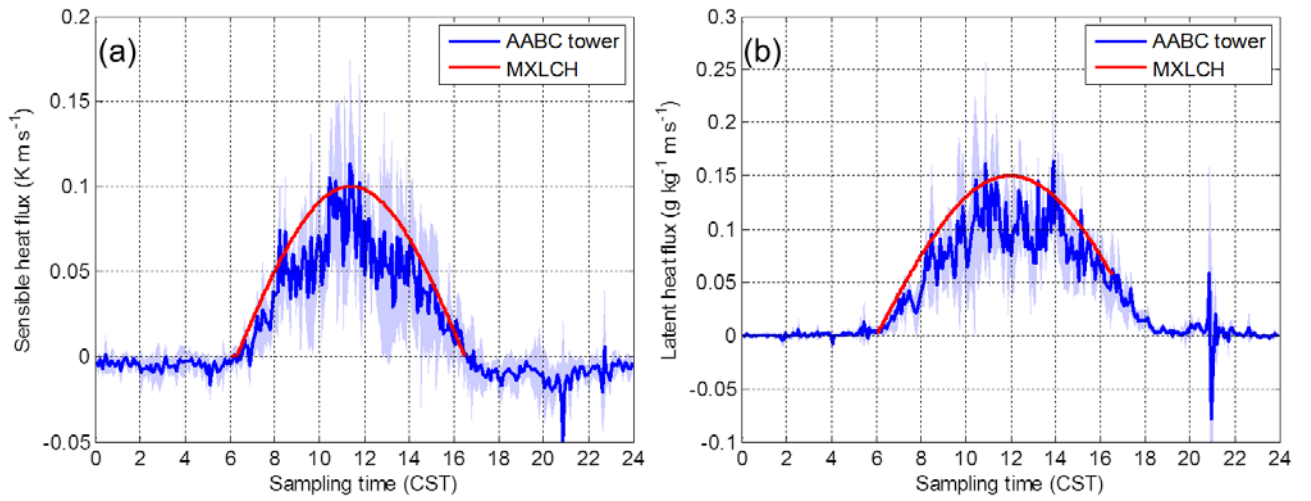
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Figure S2. Photosynthetically active radiation (PAR) measured at 32 m on the AABC flux tower.

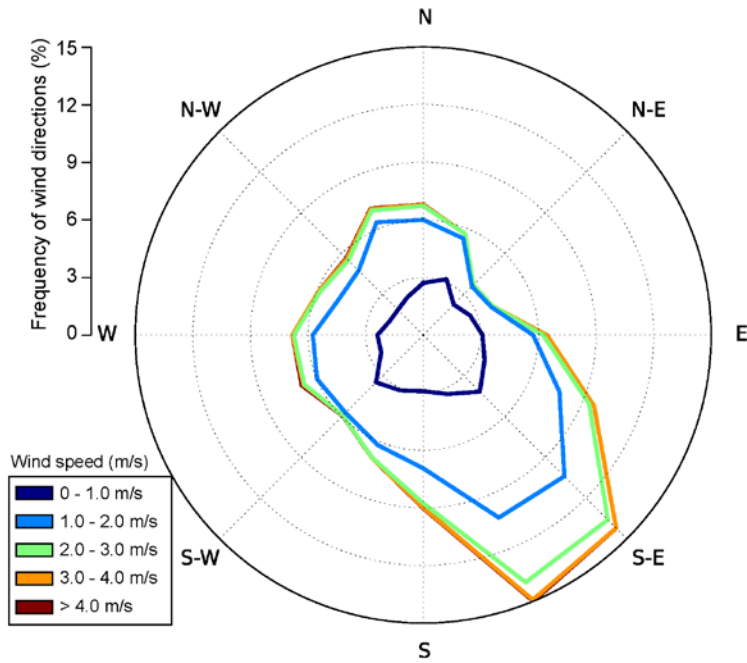
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Figure S3. Diurnal evolution of (a) surface sensible heat flux, (b) surface latent heat flux. The blue lines indicate the averaged data obtained from the AABC tower over the low cloud cover days (June 10-13). The shaded areas indicate 1 standard deviation of the observations. The red line indicates the fitted curve, which is prescribed to MXLCH.

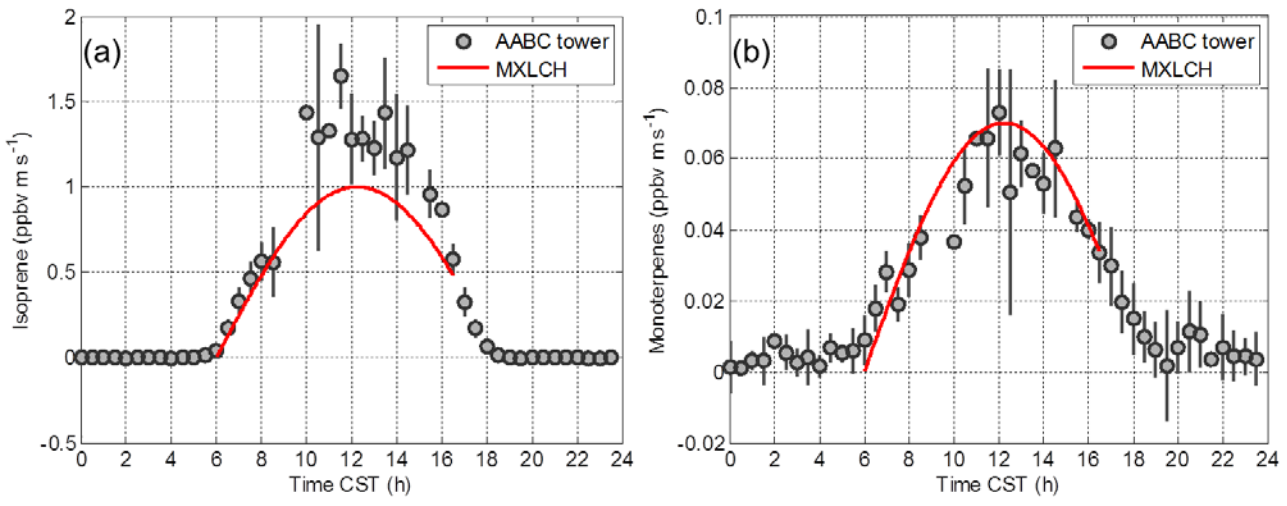
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Figure S4. Summary of the horizontal wind data during 06:00-16:30 CST of the selected days (June 5, 6, 8, 10-13, 2013). The wind data were collected on top of the AABC flux tower at 44 m height above the ground level.

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6 Figure S5. Diurnal evolution of (a) isoprene flux, (b) monoterpenes fluxes. The circles
7 indicate averaged flux during within the time bin (30 min). Error bars indicate 1 standard
8 deviation. The red line indicates the fitted curve, which is prescribed to MXLCH. The
9 isoprene peak flux value prescribed in MXLCH is on the lower bound of the observations.

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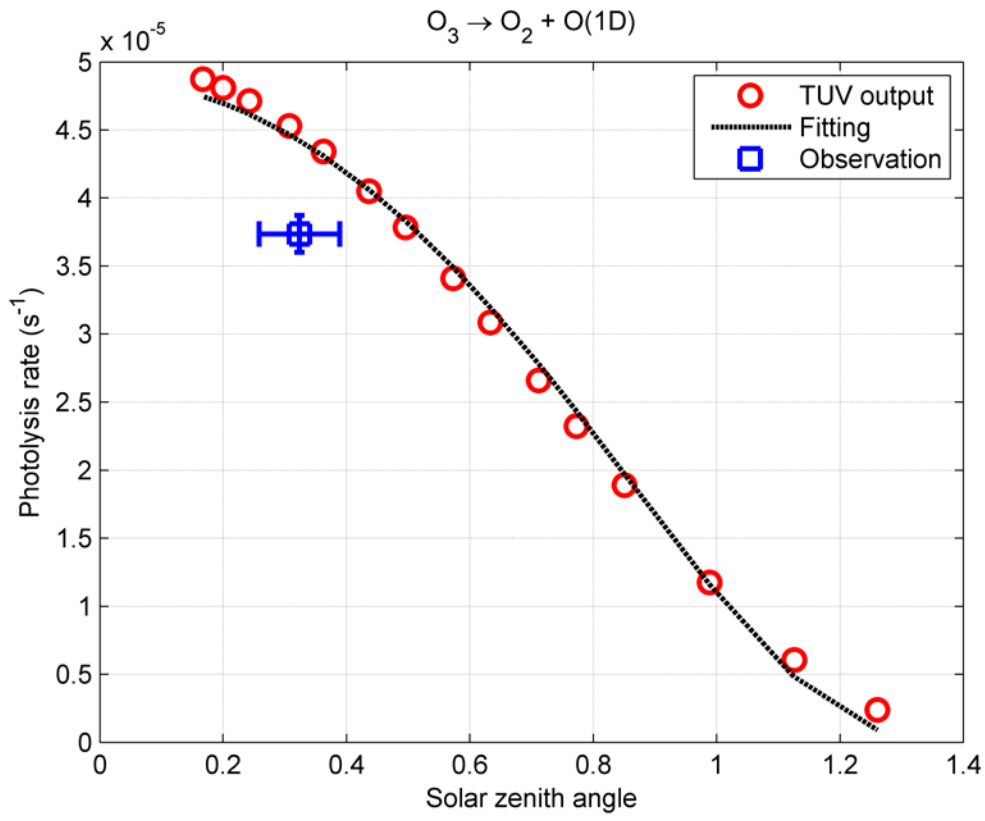
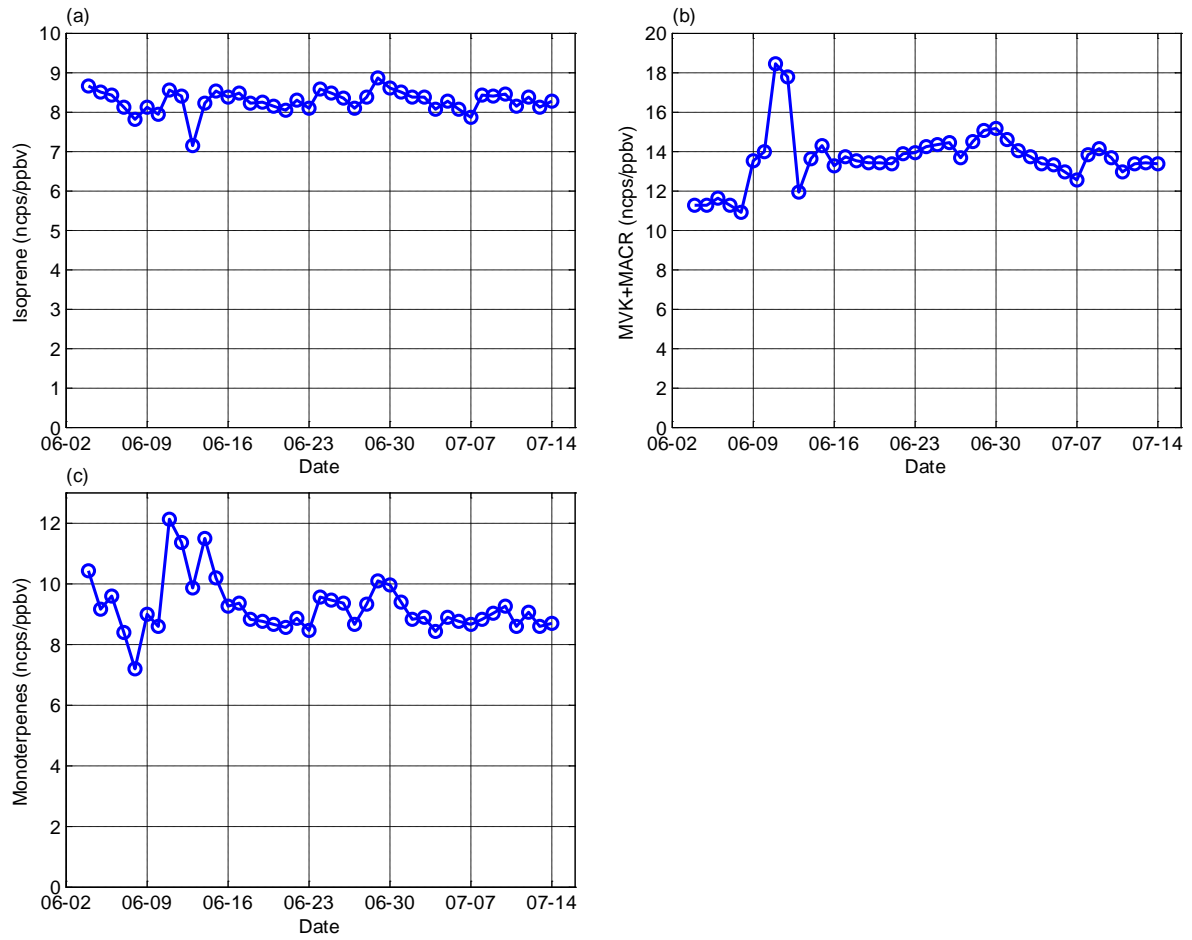


Figure S6. Comparison of TUV outputs with observation for photolysis reaction $O_3 \rightarrow O_2 + O(1D)$. Error bars indicate 1 standard deviation. See Table S4 for curve fitting parameters and methods of processing the observational data.

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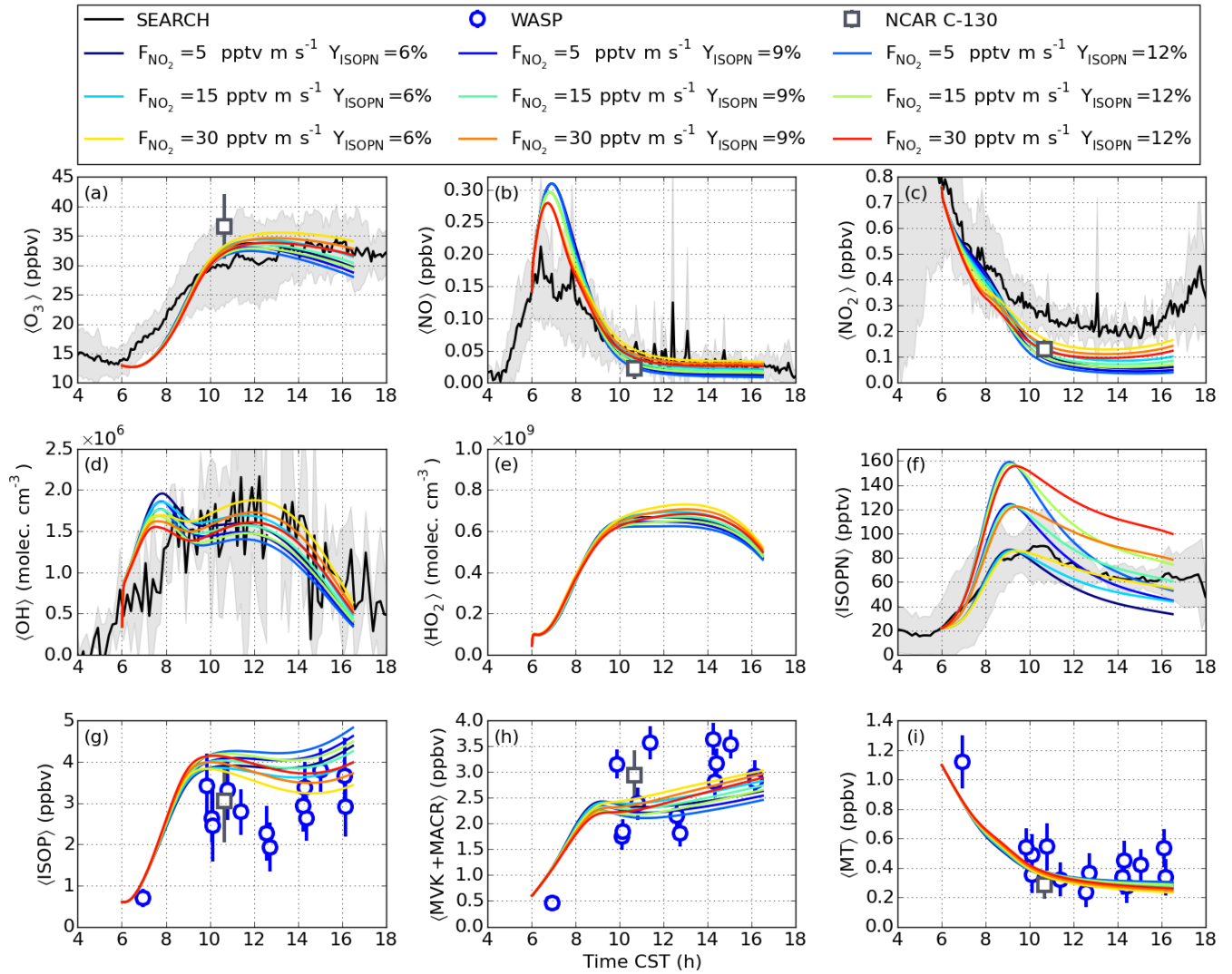


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5 Figure S7. Summary of the sensitivities for (a) isoprene, (b) MVK+MACR, and (c)
6 monoterpenes obtained from standard gas calibrations during the whole campaign period.

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8 Figure S8. Comparison of observations with MXLCH outputs of different NO_x flux levels
 9 and ISOPN yields. For each NO₂ flux level, the NO flux is downward and takes the same
 10 magnitude, i.e., $F_{\text{NO}_2} = 5 \text{ ppbv m s}^{-1}$ corresponds to $F_{\text{NO}} = -5 \text{ ppbv m s}^{-1}$. See Figure 5 for
 11 denotation of observations.

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2 **References**

3 Li, J. W.: Development and evaluation of flexible empirical peak functions for processing
4 chromatographic peaks, Anal Chem, 69, 4452-4462, Doi 10.1021/Ac970481d, 1997.

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