



Supplement of

Discrimination of water, ice and aerosols by light polarisation in the CLOUD experiment

L. Nichman et al.

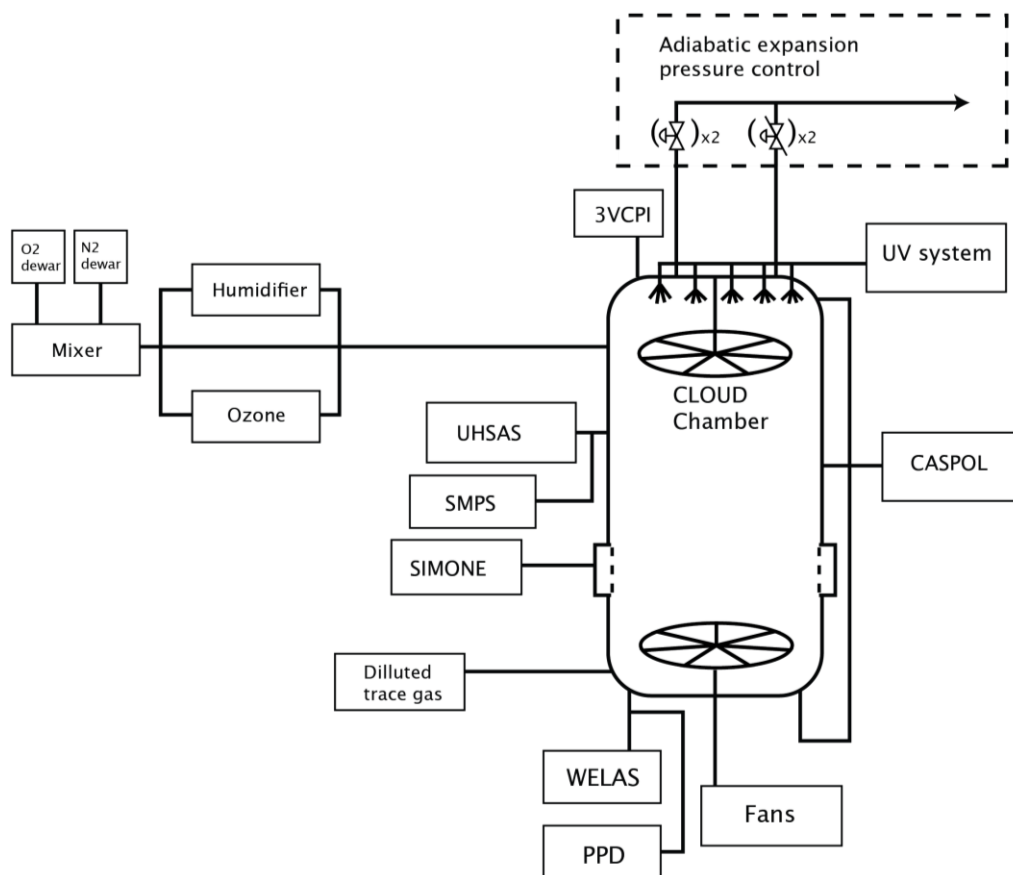
Correspondence to: L. Nichman (leonid.nichman@manchester.ac.uk)

The copyright of individual parts of the supplement might differ from the CC-BY 3.0 licence.

1 **Supplementary materials**

2 **The CLOUD Chamber**

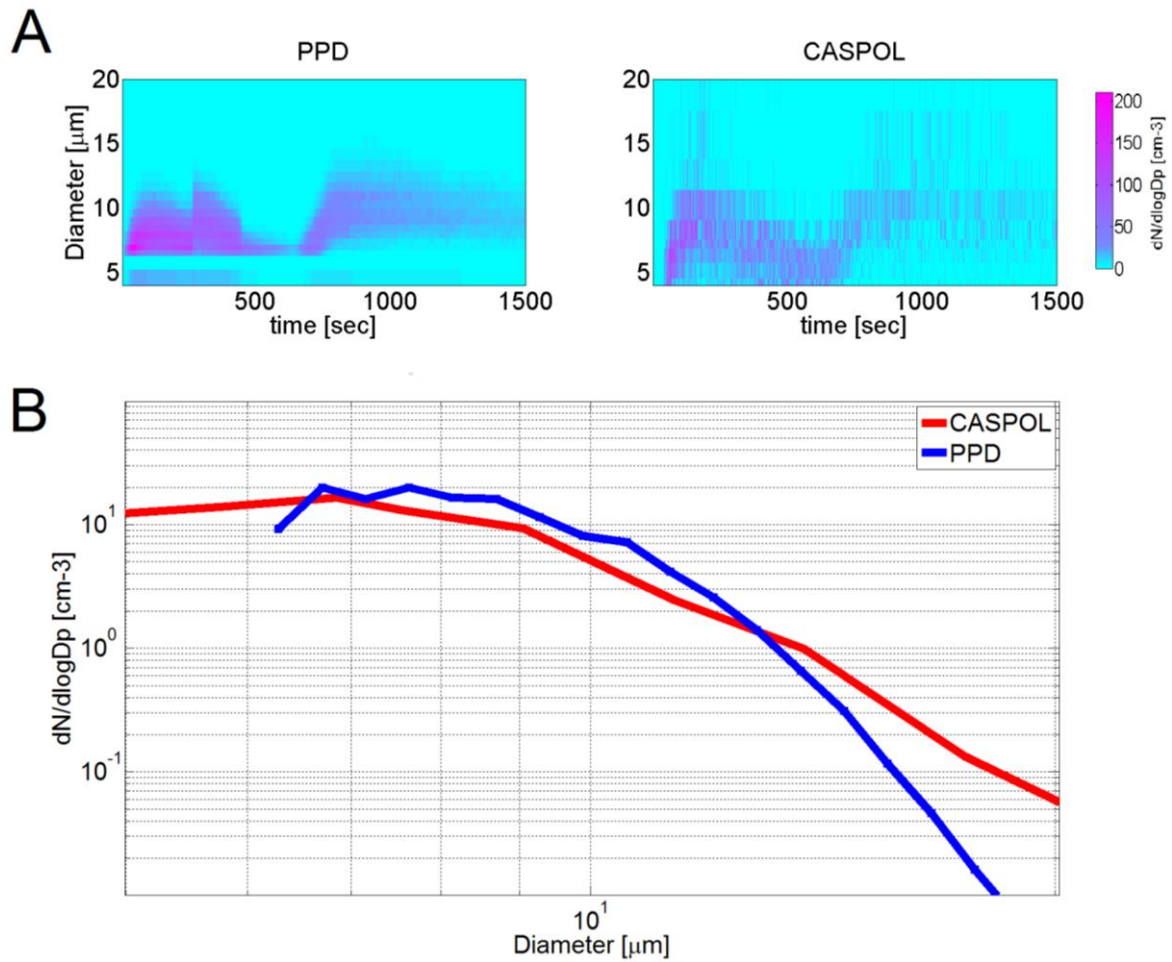
3 The CLOUD chamber is a 3 m-diameter electropolished stainless-steel cylinder (26.1 m³). An
4 insulated thermal housing surrounds the chamber. The temperature is controlled by precisely
5 regulating the temperature of the air circulating in the space between the chamber and the
6 thermal housing. Experimental runs can be performed at highly stable temperatures (near
7 0.01 °C) between +40 °C and -70 °C. Ultra-pure synthetic air is obtained from the
8 evaporation of cryogenic liquid N₂ and liquid O₂, mixed in the ratio 79:21 (Fig. S1),
9 respectively. The air is humidified using ultra-pure water from a filtered re-circulation
10 system. Ozone is added to the chamber by UV irradiation of a small inlet flow of dry air.
11 Magnetically coupled stainless steel fans on both manhole covers serve to mix the fresh gases
12 and beam ions, and ensure uniformity inside the chamber (Voigtlander et al., 2012). Volatile
13 trace gases such as SO₂ or NH₃ are supplied from concentrated gas cylinders pressurised with
14 N₂ carrier gas. The trace gas mixtures are highly diluted using synthetic air before injection
15 into the chamber. Less volatile trace gases such as alpha-pinene (C₁₀H₁₆) are supplied from
16 temperature-controlled stainless steel evaporators using ultrapure N₂ carrier gas. In order to
17 compensate for sampling losses, there is a continuous flow of fresh gases into the chamber of
18 about 150-250 L/min, resulting in a dilution lifetime of 2-3 h. The chamber and gas system
19 are designed to operate at up to +220 mb relative pressure and to make controlled adiabatic
20 expansions down to +5 mb. In this way, starting from relative humidity near 100 %, the
21 chamber can be operated as a classical Wilson cloud chamber for studies of ion-aerosol
22 interactions with cloud droplets and ice particles. The chamber can be evacuated from +200
23 mb to +5 mb over any chosen time interval above 10 sec, in order to simulate the adiabatic
24 cooling in ascending air masses that form clouds. Multistep programmed variations of
25 pressure drop are available for cloud lifetime extension or regrowth. Two 60 cm in diameter
26 fans rotating at speeds up to 400 RPM are responsible for uniform mixing in the chamber.
27 (For more details see Duplissy et al., 2015, and Kirkby et al., 2011)



1
2 Fig. S1 Simplified diagram of the CLOUD chamber.

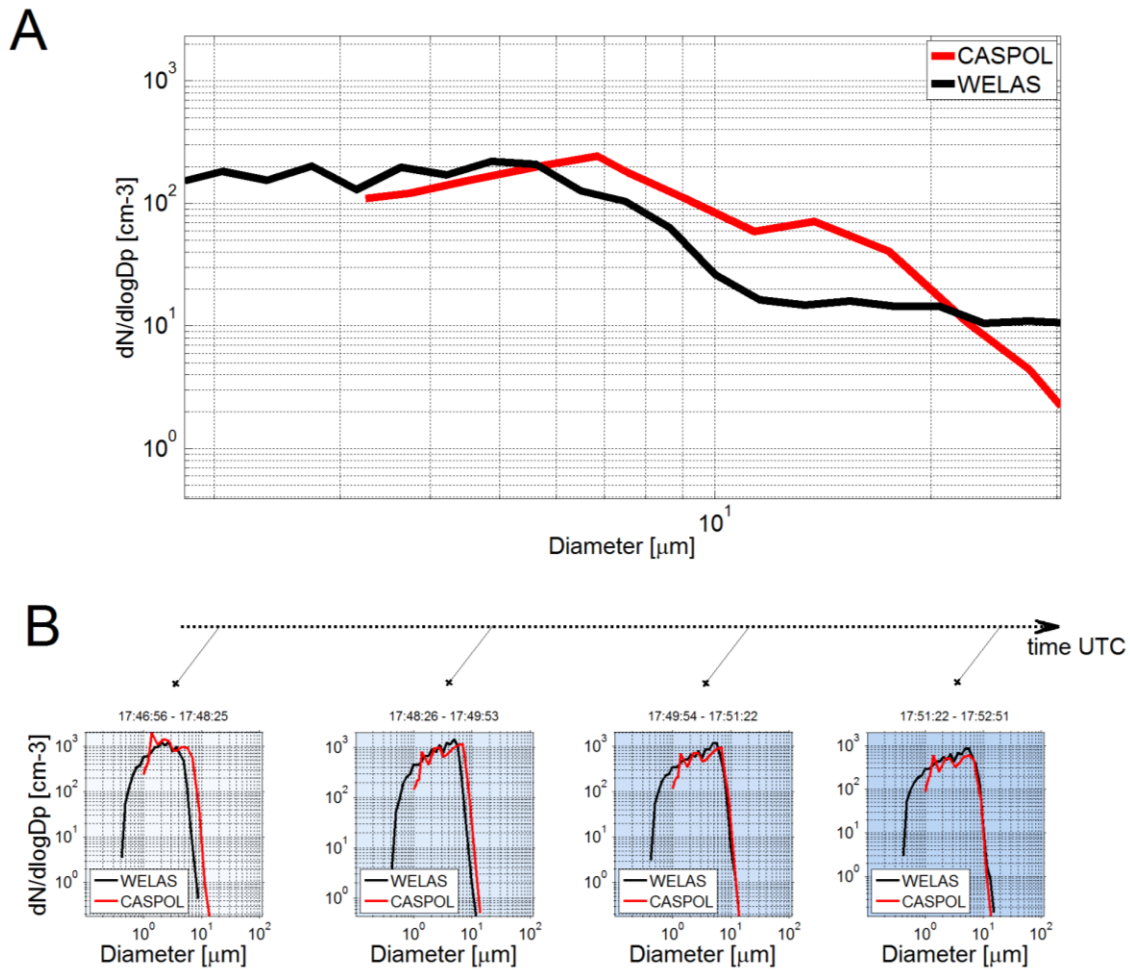
3
4 Table S2. CASPOL detectors have 3 gain stages in the forward scattering direction and 2 in
5 the backward. Signal to size conversion requires the adjusted linearly scaled reading of PBP
6 data. Corrections to the Forward, Backward and the Dpol signals are summarized.

Forward signal	Adjusted Forward scattering signal
20 – 3071	20 – 3071
3072 – 6143	$([Forward\ Size] - 3071) \times 22 + 3072$
6143 – 9216	$([Forward\ Size] - 6143) \times 506 + (6143 - 3071) \times 22 + 3072$
Backward signal	Adjusted Backward scattering signal
0 – 2000	0 – 1536
2001 – 3071	$([Backward\ Signal] - 2000) \times 22 + 3072$
Dpol signal	Adjusted Dpol signal
> 2730	$([Dpol\ signal] - 2730) \times 22 + 2731$



1

2 Fig. S3 Ice measurements (-50°C) PPD-CASPOL comparison (Run # 1298.20), Represented
 3 as 'Ice - CLOUD 8' in Fig. 8 (A) PSD plots: PPD, CASPOL. (B) Total PSD for the whole
 4 run.



1

2 Fig. S4 Super-cooled water droplets (-10^0C) (Run # 1311.03). Represented as ‘Supercooled,
 3 frozen droplets – CLOUD 8’ in Fig. 8 (A) CASPOL WELAS, total PSD comparison for the
 4 whole run (B) Comparison of sequential time frames.

5 Duplissy, J., Merikanto, J., Franchin, A., Tsagkogeorgas, G., Kangasluoma, J., Wimmer, D.,
 6 Vuollekoski, H., Schobesberger, S., Lehtipalo, K., Flagan, R. C., Brus, D., Donahue, N. M.,
 7 Vehkämäki, H., Almeida, J., Amorim, A., Barmet, P., Bianchi, F., Breitenlechner, M.,
 8 Dunne, E. M., Guida, R., Henschel, H., Junninen, H., Kirkby, J., Kürten, A., Kupc, A.,
 9 Määttänen, A., Makhmutov, V., Mathot, S., Nieminen, T., Onnela, A., Praplan, A. P.,
 10 Riccobono, F., Rondo, L., Steiner, G., Tome, A., Walther, H., Baltensperger, U., Carslaw, K.
 11 S., Dommen, J., Hansel, A., Petäjä, T., Sipilä, M., Stratmann, F., Vrtala, A., Wagner, P. E.,
 12 Worsnop, D. R., Curtius, J., and Kulmala, M.: Effect of ions on sulfuric acid-water binary
 13 particle formation II: Experimental data and comparison with QC-normalized classical
 14 nucleation theory, *J. Geophys. Res. Atmos.*, 10.1002/2015JD023539, 2015.

1 Kirkby, J., Curtius, J., Almeida, J., Dunne, E., Duplissy, J., Ehrhart, S., Franchin, A., Gagne,
2 S., Ickes, L., Kurten, A., Kupc, A., Metzger, A., Riccobono, F., Rondo, L., Schobesberger, S.,
3 Tsagkogeorgas, G., Wimmer, D., Amorim, A., Bianchi, F., Breitenlechner, M., David, A.,
4 Dommen, J., Downard, A., Ehn, M., Flagan, R. C., Haider, S., Hansel, A., Hauser, D., Jud,
5 W., Junninen, H., Kreissl, F., Kvashin, A., Laaksonen, A., Lehtipalo, K., Lima, J., Lovejoy,
6 E. R., Makhmutov, V., Mathot, S., Mikkila, J., Minginette, P., Mogo, S., Nieminen, T.,
7 Onnela, A., Pereira, P., Petaja, T., Schnitzhofer, R., Seinfeld, J. H., Sipila, M., Stozhkov, Y.,
8 Stratmann, F., Tome, A., Vanhanen, J., Viisanen, Y., Vrtala, A., Wagner, P. E., Walther, H.,
9 Weingartner, E., Wex, H., Winkler, P. M., Carslaw, K. S., Worsnop, D. R., Baltensperger, U.,
10 and Kulmala, M.: Role of sulphuric acid, ammonia and galactic cosmic rays in atmospheric
11 aerosol nucleation, *Nature*, 476, 429-433,
12 <http://www.nature.com/nature/journal/v476/n7361/abs/nature10343.html#supplementary->
13 [information](http://www.nature.com/nature/journal/v476/n7361/abs/nature10343.html#supplementary-), 2011.

14 Voigtländer, J., Duplissy, J., Rondo, L., Kürten, A., and Stratmann, F.: Numerical simulations
15 of mixing conditions and aerosol dynamics in the CERN CLOUD chamber, *Atmos. Chem.*
16 *Phys.*, 12, 2205-2214, 10.5194/acp-12-2205-2012, 2012.