We would like to thanks the reviewers for their time and constructive comments. Below we have copied the reviewer comments in plain text and included our responses in **bold** text.

5 Anonymous Referee #1 Received and published: 10 March 2015

6 This paper uses aircraft observations from SENEX and SEAC4RS to investigate what 7 fraction of aerosol mass and extinction in the transition layer over the southeastern U.S. 8 can be attributed to mixing vs. production. It represents a clear and careful analysis of a 9 very nice dataset. However I was disappointed that the study didn't really go the extra 10 step to connect to the big picture question of aerosol sources and seasonality in the 11 southeastern U.S. which served as motivation for the study (i.e. Goldstein et al.). The 12 results beg several questions: (1) how representative was the summer of 2013? Are these results generalizable for the southeast? (2) what is the impact of this highly selective 13 analysis (sub-micron only, afternoon only, no biomass burning, no in-cloud) and how 14

15 does this relate to average observed aerosol extinction in the region? The AERONET 16 observation of Figure 13 suggest that the sampling may bias this dataset to low AOD, 17 limiting it relevance to mean conditions. While these airborne campaigns provide a

18 limited snapshot (which nevertheless should be mined further to identify the biases

19 associated with the sub-sampling used here), there are longer term ground and satellite

20 observations which should have been used to provide a wider context. Without such work 21 to contextualize the results, the conclusions (line 3150 lines 17 onwards) are overstated.

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The reviewer has raised several important points which we respond to in order:

1) We agree with the reviewer that adding some additional comparison with
 other AOD datasets would give the conclusions more context; however, we
 also would like to limit the scope of this manuscript to focus on the in situ
 aerosol profiles. To that end, we have added a figure and description in the
 text comparing AOD from the in situ profiles to AOD measured by
 AERONET sun photometers at two locations in the SEUS. Both the figure
 and additional description in the text are copied below.

2) The focused analysis employed here is necessary to 'resolve' the small
enhancement in the transition layer. A larger enhancement (one that would
significantly increase AOD) would be clearly apparent in the profile of
extinction (Fig. 3a). In the case of a large enhancement aloft, the extinction
would increase with altitude. Because the extinction decreases with altitude,
an analysis limited to profiles during well-developed shallow cumulus
convection was needed to resolve the smaller enhancement.

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41 **3**) Our conclusions about the hypothesized layer of aloft are based only on

- 42 observations we made during research flights during the summer of 2013.
- 43 However, our conclusions *suggest* that further study of the seasonality of
- 44 AOD, meteorology affecting AOD are needed, and that the hypothesis of an
- 45 aerosol layer aloft should be re-examined. To provide more context for these

3 4 "The altitude-normalized aggregate profiles used in this analysis are drawn from 37 vertical profiles: however, they represent only eight afternoons 5 during the summer of 2013. For comparison and context, Fig. 14 shows an 6 extended time series of 532 nm AOD (level 2 data) measured by AERONET 7 sun photometers [Holben et al., 2001] at the Centreville SEARCH site and at 8 9 the Georgia Tech site in Atlanta, Georgia. The Georgia Tech site is in an 10 urban area and is perhaps biased toward larger AOD from urban emissions, while the Centreville site is rural. The sun photometers only report data 11 12 during cloud-free conditions. Plotted on top of these data from the sun 13 photometers are the AOD from the profiles used in the altitude-normalized 14 aggregate. These data are grouped into the profiles from the SENEX and SEAC4RS studies. Aircraft profiles during the SENEX study did not sample 15 AOD greater than 0.3 while the maximum of AOD observed by the sun 16 17 photometers was greater than 0.4. Profiles during the SEAC4RS study, 18 although limited in number, cover a range of AOD similar to the sun 19 photometers. Because the majority of the profiles in the altitude-normalized 20 aggregate are from the SENEX study, the aggregate may be biased toward 21 cleaner conditions. The range of AOD observed during summers of 2011-22 2013 at the Georgia Tech site indicate that the summer of 2013 is not an 23 outlier with AOD higher or lower than typical summers. This is consistent 24 with the analysis of Kim et al. [2015] who has compared satellite 25 measurements of AOD during the summers 2006-2013."

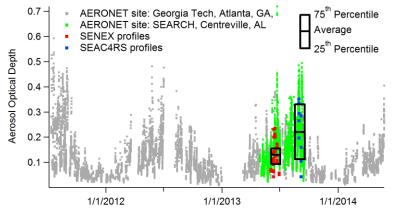
conclusions we have added references (Alston et al. (2012) and Kim et al.

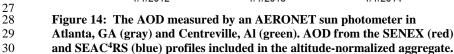
(2015)) that discuss the hypothesis of an aerosol aloft over the SEUS.

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1	The black boxes show the average, 25 th , and 75 th percentiles of AOD from
2	both the SENEX and SEAC ⁴ RS profiles.
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4	1. Abstract: should state year of observations as well as campaign names
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6	We have added a sentence to the abstract stating the year and campaign
7	names.
8	
9	2. Page 3129, line 20; page 3130 line 17; page 3150, line 17: Goldstein et al., 2009 linked
10	the seasonality of AOD to biogenic SOA. Ford and Heald concludes only that the
11	observations support a significant summertime aerosol source aloft. They speculate that
12	aqueous sources of SOA or H2SO4 from Criegee chemistry are possible sources, so it's
13	erroneous to suggest that SOA was the hypothesis of this study.
14	
15	We thank the reviewer for pointing out the lack of nuance in our description
16	of the conclusions of Ford and Heald (2013). To correct this we have
17	rewritten sections of the abstract, introduction, and conclusion to make it
18	clear that the Goldstein et al. hypothesize a layer of SOA while Ford and
19	Heald hypothesize an aerosol layer that would be composed of organics
20	and/or particulate sulfate.
21	
22	3. Sections 2.1 & 2.2: Please indicate the collection efficiency of the AMS instruments
23 24	used and how these two instruments differ on the two aircraft.
24 25	We have added the following tout to Section 2.1
23 26	We have added the following text to Section 2.1.
20 27	"The non-refractory aerosol composition was measured by a compact time-
28	of-flight aerosol mass spectrometer (AMS) downstream of a pressure
20 29	controlled inlet [Bahreini et al., 2008] and most (97%) of the submicron
30	volume measured by the aerosol sizing instruments was transmitted into the
31	AMS during SENEX. The collection efficiency for the AMS was determined
32	by the composition for each data point using the algorithm described by
33	Middlebrook et al. [2012]. When comparing the volume derived from
34	composition (AMS plus black carbon mass) to the volume measured by the
35	aerosol sizing instruments in the manner outlined by Bahreini et al. [2009],
36	87% of the aerosol composition and sizing data from the entire SENEX study
37	are within the combined uncertainties."
38	
39	We have added the following text to Section 2.2.
40	
41	"The non-refractory aerosol composition was measured by a high resolution
42	time-of-flight AMS similar to the compact time-of-flight AMS used during
43	the SENEX study. The two instrument differed in the resolution of the mass
44	spectrometer. The higher resolution AMS used during SEAC4RS was
45	operated with a 1 s sample interval. This AMS was operated similar to
46	Dunlea et al. [2009] and also used a pressure-controlled inlet [Babreini et al.

46 Dunlea et al. [2009] and also used a pressure-controlled inlet [Bahreini et al.,

1	2008]. The composition-dependent formulation of Middlebrook et al. [2012]
2	was used to estimate the collection efficiency. The AMS sampled aerosol
3	downstream of a HIMIL inlet.
4	(http://www.eol.ucar.edu/homes/dcrogers/Instruments/Inlets/). In both the
5	HIMIL inlet and the shrouded diffuser inlet, the sampled aerosol was
6	initially dried by ram heating and then further dried in each instrument."
7	
8	4. Page 3134, lines 1-5: state what wavelengths were used for the extinction
9	measurements (here or in table)
10	
11	We have add sentences to the text stating the wavelength used for the
12	extinction measurements.
13	"The extinction of dry aerosol was measured at three wavelengths (405 nm,
14	532 nm, and 662 nm), and humidified extinction was measured at 532 nm.
15	Only 532 nm extinction measurements are used in this analysis."
16	
17	5. Table 1 and 2: aerosol extinction is listed as "dry" although the text indicates
18	measurements were taken at 3 RH conditions. Perhaps "dry" is meant to indicate the
19	uncertainty under "dry" conditions? If so, uncertainties for "wet" should also be stated.
20	
21	"Dry" was intended to indicate that the accuracy was for dry conditions. We
22	have added a line to Tables 1 and 2 stating the accuracy of the extinction
23	measurements at 90% RH. The additional uncertainty at elevated RH is due
24	to uncertainty in the RH measurement (+/-1%) which is translated into an
25	uncertainty in extinction using the kappa parameterization ($\Box = 0.15$ typical
26	see Fig. 3c) and added directly, because these are likely systematic (opposed
27	to random) uncertainties.
28	(Table 1 and 2. Discon homeonics instrument labelings some one since in full (s. s.
29	6. Table 1 and 2: Please harmonize instrument labeling: some are given in full (e.g.
30	Aerosol Mass Spectrometer) and some are given as acronyms (e.g. PTR-MS)
31	We have showed the concurrence in the table to full names
32	We have changed the acronyms in the table to full names.
33 34	7. Table 1 and 2: 5th entry should read "Dew Point (RH)" since the accuracy is given in
34 35	units of dew point, not RH.
35 36	units of dew point, not Kri.
30 37	We have changed the 5 th entry to 'Dew point (RH)' as suggested by the
38	reviewer.
38 39	leviewei.
39 40	8. Section 2.4: would have been insightful for the authors to compare their extinction or
40 41	total AOD values to other available datasets (CALIPSO, MODIS, or AERONET).
41	total AOD values to other available datasets (CALILSO, WODDS, OF AERONET).
42 43	As suggest by the reviewer, we have added a comparison to AERONET AOD
43 44	at two sites in the SEUS. Please see our response to the overview comments of
44 45	the first reviewer above.
45	

9. Section 2.4: Did the authors check if the fitted kappa is consistent with the measured 1 2 composition? 3 4 Another manuscript, Brock et al. (in preparation), looks at the aerosol 5 hygroscopicity during SENEX and SEAC⁴RS in more detail and has found 6 that the volume-weighted kappa from calculated from the aerosol 7 composition and the optical fitted kappa agree within uncertainties if we 8 assume the kappa from the organic portion of the aerosol is $k_{org} < 0.1$, which 9 is consistent with other work such as Good et al., Atmos. Chem. Phys., 10(7), 10 3189-3203, 2010. 11 12 10. Page 3136, line 27-28: awkward phrasing "globally and over the SEUS" suggests that the SEUS is not part of the globe. Suggest that this is re-phrased 13 14 15 We have removed globally from the sentence it now reads: "Shallow cumulus 16 convection is common over the SEUS." 17 18 11. Page 3137, line 10-11: Does "mixed layer height" mean top of the mixed layer? 19 Please clarify. 20 21 We have changed "mixed layer height" to "the top of the mixed layer" to 22 avoid any confusion. 23 24 12. Figure 3 caption (and generally all Figures): please specify which variables are 25 measured and which are calculated. 26 27 We have added more description to Figs. 3, 8, 9, and 10 so that it is clear to 28 the reader which quantifies are directly measured and which are calculated 29 from precursor measurements. 30 31 13. Figure 4: It would be appropriate to use reduced major axis fitting here. Please 32 specify in the caption/text if this is the case or correct if otherwise. 33 34 For this fit we have used orthogonal distance regression (ODR) and have 35 added this detail to the text. Using the ODR fit method is appropriate because both independent and dependent variable are expected to have 36 37 similar uncertainty and the purpose of the fitting is to determine the 38 relationship between extinction aloft and on the surface. 39 40 14. Page 3139, lines 21-22: ion charges missing on nitrate (NO3-), ammonium (NH4+), 41 and sulfate (SO42-) 42 43 In the literature, we have found examples of AMS data presented with and 44 without the ion charges. We normally don't use the charges when referring to AMS data, since those signals can originate from organic species in 45

46 addition to inorganic salts. For an example of AMS data presented without

1 2 3	ion charges see Ng, et al., Aerosol Science and Technology 45(7), 770-784, Figs. 7 and 9. Below is a passage from the text illustrating this point. We have added the sentence in italics to inform the reader of the notation.
4	"The aerosol mass is the total of all ions measured by the AMS, and these
5	ions are typically classified as SO4, NH4, NO3, and OA. The inorganic ions
6	are typically formed by ionization of simple salts such as ammonium sulfate
7	and ammonium nitrate or may be formed from more complex compounds (i.
8 9	e. organosulfates, organonitrates, and amines) that produce inorganic and
9 10	organic ions when ionized. To indicate this complexity, we have omitted ionic charges from the notation (i. e. SO4, NH4, NO3,)."
10	charges from the holation (i. e. 504, 14114, 1403,).
12	15. Page 3141, line 14: "virtual potential temperature was constant". With what
12	tolerance?
14	
15	We have added the tolerance to the text. The relevant sentence is copied
16	below.
17	
18	"For individual profiles, the mixed layer height was defined by inspection of
19	each profile as the highest altitude at which the virtual potential temperature
20	(Θ_{v}) was constant (typical variation in the mixed layer was less than 0.5 K)
21	and there was a reduction in the isoprene concentration."
22	
23	16. Section 4.1: Please comment on the impact of neglecting horizontal advection.
24	
25	We have added the following sentences to the Section 2.5 motivating the
26	aggregation of individual profiles.
27	(21) · · · · · · · · · · · · · · · · · ·
28 29	"Individual profiles are affected by horizontal advection which couples spatially inhomogeneous emissions to the vertical profiles. Because of
29 30	vertical wind shear and spatial variability during slant profiles, the vertical
30	layers in an individual profile are not always directly comparable. The
32	aggregation the individual profiles is used to reduce the influence of this
33	variability and resolve the typical vertical structure and mixing over the
34	SEUS."
35	
36	17. Section 4.1: Could you comment on how/whether this analysis of the fraction of air is
37	affected by the lifetime of the assumed compound (i.e. CO) relative to the applied species
38	(i.e. aerosols)?
39	
40	For this analysis, the important time scale is that of atmospheric mixing. In
41	the mixed layer this time scale is approximately one half hour. In the
42	transition layer, the typical lifetime of a cumulus cloud is ~ 1 hour, so the
43	timescale for mixing in transition layer might be a few hours. Regional
44	subsidence (typical velocity = 500 meters /day) mixes air from the free
45	troposphere into the transition layer and mixed layer below. The lifetime of
46	both aerosol and CO are much longer than all of these lower troposphere

46 both aerosol and CO are much longer than all of these lower troposphere

1 2 3 4	mixing processes. We could use other long lived species (CH4, CO2, non- precipitating H2O) to determine the fraction of air from the mixed layer, $f_m(h)$, and Fig. 8 shows that the vertical profile of all of these long live species behave in a similar manner.
5 6 7 8	18. Page 3145, lines 23-24: How does the vertical profile of aerosol mass/extinction differ when including biomass burning?
9 10 11	During the SENEX study, the plumes from several small agricultural fires were transected over the SEUS. These transects intentionally targeted these plumes, and the plumes were transected during level flight legs. During
12 13 14	SEAC ⁴ RS, several large western wildfire plumes were sampled over the western US (in research flights that were not used here) and several small agricultural fires plumes were similarly sampled over the SEUS again largely device for the several flight here.
15 16 17 18	during level flight legs. Because the analysis presented here only considered the ascending and descending portions of the flights, these biomass burning plumes were <i>implicitly</i> not included in this analysis. Hence <i>explicitly</i> including/excluding these plumes has no effect on the analysis presented
19 20 21	here.
22 23 24 25	transects from the data from each flight and then extracted the profiles. However we subsequently realized that the biomass burning didn't occur during the profiles. We have modified (copied below) the description of the data filtering in Section 2.5 to indicate this point.
26 27 28 29 30	"Transects of biomass burning plumes were identified using tracers such as the acetonitrile mixing ratio, were typically during level flight legs, and were not found in any of the profiles used in this analysis."
31 32	19. Page 3147, line 15: typo: eq (5)
33 34	We have corrected the equation referenced in this sentence to Eq. 5.
35 36 37	20. Page 3147, line19-20: The authors might want to note that this approach is supported by their own analysis in Figure 4.
38 39 40	We have added a note that the assumption is supported by the correlation in Fig. 4.
41 42 43 44 45	<i>"Because extinction at the surface and aloft in the mixed layer are well correlated as shown in Fig. 4, the dry extinction is extrapolated as a constant to the surface based on the mean extinction measured in the lowest 200 m of each profile."</i>
46	21. Figure 11: caption missing (caption is for Figure 12).

1	
2	We have corrected this error. The caption for Fig. 11 follows below:
3	
4	"Figure 11: Histograms of the transition layer enhancement $(E(h) - \text{see Eq.})$
5	4) for several trace gases and aerosol properties. The first column shows
6	conserved species and black carbon: (A) CH4, (B) CO2, (C) H2O, and (D)
7	black carbon mass. The second column shows the aerosol extensive
8	properties: (E) aerosol mass, (F) dry extinction, and (G) aerosol volume. The
9	third column shows the aerosol composition: (H) OA, (I) SO4, (J) NH4, and
10	(K) NO ₃ . The Student's T-test and resulting p-value (noted in each
11	histogram) were used to test if the mean of each distribution was statistically
12	different from zero."
13	
14	22. Figure 12: typo "total sulfur (tS)."
15	
16 17	We have corrected the labels on Fig. 12.
17	Anonymous Deferse #2 Deserved and published 7 April 2015
18 19	Anonymous Referee #2 Received and published: 7 April 2015
20	This manuscript uses aircraft in-situ data from two different campaigns (SEAC4RS and
20	SENEX) during the summer and early fall of 2013. The aircraft platform has advantages
22	to other platforms (space and ground) in that vertical layers can be probed directly. The
23	analyses presented though representative of a small time period, are methodical and
24	logical. However, these results seem to be disconnected to other research in this area.
25	This reviewer kept waiting to read about how these results tie in (or do not tie in) with
26	other contextual works. The conclusion could potentially be quite insightful, if given the
27	proper context in which to state it. Overall, the work is sound and could be improved by
28	minor changes.
29	
30	We have made a few changes to the manuscript to provide the reader with
31	more context. First, we have added a figure which compares the AOD of
32	aircraft profiles to the AOD measured by sun photometer at two sites in the
33	SEUS. Second, we have added a brief discussion of two recent articles
34	relevant to the hypothesis of the layer aloft. The following sentences were
35	added to the conclusion.
36 37	"The hypothesis is partially supported by the spatial similarity of
38	summertime biogenic emission and summertime AOD over the SEUS.
39	Although, Alston et al. [2012] found that the spatial similarity depended on
40	the spatial resolution AOD of the data used in the analysis, and their analysis
41	of AOD and surface aerosol mass over Georgia did not fully support the
42	hypothesis. Kim et al. [2015] found that the increase of the planetary
43	boundary layer height during the summer could bring the seasonality of the
44	AOD and surface aerosol mass into agreement without the need for an
45	aerosol layer aloft. "
46	

1 Specific points:

1 - Section 3.1: Were there any other measurements that could be related to groundbased
measurements similar to Fig. 4 in Section 3.1. This could open up an avenue of
comparison with older field intensive studies in the same region, e.g., Atlanta Supersite

5 Experiment in 1999.

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Although detailed comparison of aircraft and surface measurements and historical measurements are scientifically interesting and there is potentially a lot that can be learn from them, aside from surface measurements presented in section 3.1 (included to support assumptions used to calculate AOD) they are beyond the scope of this analysis.

2 - Typo: Line 9 3143 change expecting to expected

We have corrected this typo.

17 3 - Section 4.3: It would have been more conceptually thorough if the authors had 18 brought in some satellite (CALIPSO for vertical profile comparison, OMI as another indicator of the lack of biomass burning aerosols) observation comparisons in Section 19 20 4.3. Even if that work is being done by another author, a reference to those results would 21 have been helpful to understand what dynamics the entire region was undergoing at the 22 time of these flights. Or the authors could have compared their results with time averaged 23 satellite information, which could open up the avenue for discussing how similar or 24 dissimilar summer 2013 was from other summers. 25

We have added an additional figure and references to some complementary work (see the response to the overview comments of the first and second reviewers).

4 - Section 4.3: There did not appear to be much summarization at the end of each
respective section. Is that on purpose? The only section that does an adequate job in
relating that section's work to the field at large is Section 4.3.

This comment makes largely the same point as comment 5 and is addressed in the response to that comment.

5 - Conclusions: Most of this section speaks to the hypothesis of an elevated layer of
aerosols which account for changes in AOD and the author's perspective on that. What
about the other results? How do these other results shown in Figs 1-12 support the
author's thesis statement? There was a lot of work done to create these different analyses,
yet at the end there's scant mention of them.

Many of the other results are included to support the conclusion of a modest
 enhancement in the transition layer. For example Fig. 4, demonstrates that
 aerosol is well mixed down to the surface below the minimum altitude of the
 aircraft and that extinction can be extrapolated to the surface in the AOD

1 calculation. Fig. 11 shows the statistical significance of the observed 2 enhancements. Figs. 8 and 12 are included to demonstrate that other long 3 lived species have a profile similar to CO. Fig. 12 also shows a reduction in 4 the precursor of particulate sulfate, SO₂, further supporting the observed 5 enhancement of particulate sulfate in the transition layer. In the conclusion, 6 rather than reiterating these supporting agreements we discuss the 7 disagreement between our observations and the hypothesized layer. To 8 explain how the absence of a large layer of enhanced aerosol in Fig. 3 9 motivated a more refined analysis, we have added a paragraph to the 10 conclusion to tie together the altitude-binned aggregate in Figs 3-5 with the 11 altitude-normalize aggregate presented in Figs. 7-10. 12

13 "Here, we have examined in situ vertical profiles of aerosol and found the 14 dry aerosol to be well mixed in the lowest layer. Above the lowest layer the 15 aerosol mass and extinction decreased with increasing altitude above that 16 layer (Figs. 3-5). The hygroscopic a growth of aerosol resulted in a layer of 17 enhanced extinction near the top of the mixed layer. The aerosol water 18 accounted for approximately a third of the AOD which would account for a 19 portion of the summertime AOD enhancement. The hypothesized, large 20 enhancement of secondary aerosol aloft was not apparent in these aggregate 21 profiles. However, after normalizing the altitude to the vertical structure 22 and using the CO concentration to quantify the vertical mixing during 23 shallow cumulus convection, we were able to resolve a modest enhancement 24 of aerosol in the transition layer."

6 - Figures: Fig 13 is somewhat confusing. What does the blue portion of the figure
represent? A couple of sentences added to the caption would help orient the reader.

We have added the following sentences to the caption for Fig. 13. "The idealized blue profile of extinction at the center of the figure shows the vertical location of each contribution to AOD. The light blue area represents the extinction of dry aerosol, and the darker blue area shows the enhancement to aerosol water."

35 Additional changes:

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36	1) We changes the notation of the hygroscopicity parameter from κ to
37	κ _{opt} to avoid confusion with hygroscopicity derived from more direct
38	measurements of the diameter growth factor i. e. HTDMA.
39	

- We have added some more background detail on biomass burning in
 the SEUS to section 3.2 as suggest by B. Yokelson (through personal communication).
- 43 "Biomass burning is common in the SEUS during the fall, winter, and 44 spring, but is less common during the summer. Zhang et al. [2010]

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found that in the summer of 2007 biomass burning contributed between 2 and 10% to measurements of $PM_{2.5}.^{\prime\prime}$

- 1 In situ vertical profiles of aerosol extinction, mass, and
- 2 composition over the southeast United States during SENEX
- **3 and SEAC⁴RS: Observations of a modest aerosol enhancement**
- 4 **aloft** 5
- 6 N. L. Wagner^{1, 2}, C. A. Brock¹, W. M. Angevine^{1,2}, A. Beyersdorf³, P. Campuzano-
- 7 Jost^{2,4}, D. Day^{2,4}, J. A. de Gouw^{1,2}, G. S. Diskin³, T. D. Gordon^{1,2}, M. G. Graus^{1,2*}, J. S.
- 8 Holloway^{1,2}, G. Huey⁵, J. L. Jimenez^{2,4}, D. A. Lack^{1,2}, J. Liao^{1,2}, X. Liu⁵, M. Z.
- 9 Markovic^{1,2**}, A. M. Middlebrook¹, T. Mikoviny⁶, J. Peischl^{1,2}, A. E. Perring^{1,2}, M. S.
- 10 Richardson^{1,2}, T. B. Ryerson¹, J. P. Schwarz^{1,2}, C. Warneke^{1,2}, A. Welti^{1,2,7}, A.
- 11 Wisthaler⁸, L. D. Ziemba³, D. M. Murphy¹
- 12
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- 14 [2] Cooperative Institute for Research in Environmental Sciences, University of
- 15 Colorado, Boulder, CO 80309, USA
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- 23 Zürich, Switzerland
- 24 [8] Institute for Ion Physics and Applied Physics, University of Innsbruck,
- 25 Technikerstrasse 25, 6020 Innsbruck, Austria
- 26 [*] Now at Institute of Meteorology and Geophysics, University of Innsbruck, Austria
- 27 [**] Now at Air Quality Research Division, Environment Canada, Toronto, ON, Canada
- 28
- 29 Correspondence to: N. L. Wagner (<u>nick.wagner@noaa.gov</u>)
- 30
- 31 Abstract:
- 32

1 Vertical profiles of submicron aerosol over the southeastern United States (SEUS) 2 during the summertime from in situ aircraft-based measurements were used to construct 3 aggregate profiles of chemical, microphysical, and optical properties. These vertical 4 profiles were collected over the southeastern United States (SEUS) during the summer of 5 2013 as part of two separate field studies: the Southeast Nexus (SENEX) study and the Study of Emissions and Atmospheric Composition, Clouds, and Climate Coupling by 6 7 Regional Surveys (SEAC⁴RS). Shallow cumulus convection was observed during many 8 profiles. These conditions enhance vertical transport of trace gases and aerosol and 9 create a cloudy transition layer on top of the sub-cloud mixed layer. The trace gas and 10 aerosol concentrations in the transition layer were modeled as a mixture with 11 contributions from the mixed layer below and the free troposphere above. The amount of 12 vertical mixing, or entrainment of air from the free troposphere, was quantified using the 13 observed mixing ratio of carbon monoxide (CO). Although the median aerosol mass, 14 extinction, and volume decreased with altitude in the transition layer, they were ~10% 15 larger than expected from vertical mixing alone. This enhancement was likely due to secondary aerosol formation in the transition layer. Although the transition layer 16 enhancements of the particulate sulfate and organic aerosol (OA) were both similar in 17 18 magnitude, only the enhancement of sulfate was statistically significant. The column 19 integrated extinction, or aerosol optical depth (AOD), was calculated for each individual 20 profile, and the transition layer enhancement of extinction typically contributed less than 21 10% to the total AOD. Our measurements and analysis were motivated by two recent 22 studies that have hypothesized an enhanced layer of secondary organic-aerosol-(SOA) 23 aloft to explain the summertime enhancement of AOD (2-3 times greater than winter) 24 over the southeastern United States. The first study attributes the layer loft to secondary 25 organic aerosol (SOA) while the second study speculates that the layer aloft could be 26 SOA or secondary particulate sulfate. In contrast to this these hypotheses hypotheses, the 27 modest enhancement we observed in the transition layer was not dominated by OA and 28 was not a large fraction of the summertime AOD. 29

30 1. Introduction

1 Shallow cumulus convection is common over the southeastern United States 2 (SEUS) during the summer. It enhances the vertical transport of trace gases and aerosol, 3 and creates a transition layer between the mixed layer and free troposphere [Siebesma, 4 1998]. Due to the presence of clouds and entrainment in the transition layer, it has also 5 been referred to as the cloud layer and the entrainment zone. The transition layer is intermittently mixed by thermal plumes that originate in the mixed layer and form 6 7 cumulus clouds that release latent heat within the layer. There have been several 8 observations of vertical transport and redistribution of trace gases by shallow cumulus 9 convection [Angevine, 2005; Ching and Alkezweeny, 1986; Ching et al., 1988; Greenhut, 10 1986], and a few studies have investigated the vertical transport and aerosol formation 11 during cumulus convection [Ching et al., 1988; Sorooshian et al., 2007; Sorooshian et 12 al., 2006; Wonaschuetz et al., 2012]. 13 Based on the seasonality of the surface aerosol-AOD relationship in the SEUS 14 and the spatial similarity of biogenic emissions and enhanced AOD, Goldstein et al. 15 [2009] and Ford and Heald [2013] have hypothesized the existence of a layer of enhanced SOAsecondary aerosol aloft in the summer which contributes to AOD but not to surface 16 17 measurements of aerosol mass. Goldstein et al. hypothesize that the aerosol layer is 18 primarily SOA, while Ford and Heald speculate that the layer aloft could be either SOA 19 or particulate sulfate. Although, neither study speculates about meteorological and 20 chemical-mechanisms that would lead to its the formation of this layer, aerosol 21 production in the transition layer of shallow cumulus convection is a plausible 22 mechanism that could produce the hypothesized layer. More generally, the vertical 23 distribution of aerosol and aerosol formation are integral to understand the relationship 24 between aerosol mass (PM_{2.5}) at the surface and AOD [Hoff and Christopher, 2009]. 25 Submicrometer aerosol particles, which dominate aerosol mass, are largely 26 secondary and composed of OA and sulfates during the summer in the SEUS [Edgerton 27 et al., 2005; Weber et al., 2007]. While the formation mechanisms of secondary 28 particulate sulfate are well understood (e. g. Seinfield and Pandis [1998]), the formation 29 of SOA is more complex and uncertain. Both biogenic and anthropogenic precursor 30 emissions are thought to be important [de Gouw and Jimenez, 2009]. The relative 31 importance of the homogenous and aqueous oxidation pathways for both sulfate and OA

is also uncertain [Carlton and Turpin, 2013; Carlton et al., 2008; Eatough et al., 1994; 1 2 Ervens et al., 2011; Luria and Sievering, 1991; McKeen et al., 2007]. Based on the 3 abundance of aerosol water and cumulus convection, aqueous processing is expected to 4 be an important aerosol formation pathway in the SEUS [He et al., 2013], and processing 5 in cloud droplets would occur primarily in the transition layer. 6 In this analysis, aircraft-based in situ measurements of aerosol chemical, physical, 7 and optical properties are used to examine the vertical structure of aerosol in the SEUS 8 during shallow cumulus convection and to quantify aerosol enhancements in the 9 transition layer and its contribution to summertime AOD. We use measurements made 10 aboard the National Oceanic and Atmospheric Administration (NOAA) WP-3D aircraft 11 during the NOAA Southeast Nexus (SENEX)X study in June and July of 2013 and the 12 National Aeronautic and Space Administration (NASA) DC-8 aircraft during the Study of Emissions and Atmospheric Composition, Clouds, and Climate Coupling by Regional 13 14 Surveys (SEAC⁴RS) in August and September of 2013 to construct aggregate vertical 15 profiles of aerosol extinction, mass, and composition as a function of altitude over the 16 SEUS. The transition layer aerosol and trace gas concentrations are modeled as a 17 mixture with contributions from the free troposphere and mixed layer. The in situ 18 measurements of the extinction coefficient are used to calculate the AOD and 19 contributions to the AOD from aerosol water, from the mixed layer, and from the 20 transition layer. 21 22 2. Methods and Measurements 23 In this analysis we combine data collected during two aircraft field studies that were 24 partially conducted over the SEUS in the spring and summer of 2013. Although the 25 SENEX study collected measurements in late spring and early summer while SEAC⁴RS 26 collected measurements in the late summer, both studies encountered shallow cumulus 27 convection. Additionally, both aircraft hosted a similar set of in situ instruments which 28 was conducive to a combined analysis. 29

30 2.1 SENEX

1	The SENEX study was part of the larger Southeast Atmosphere Study (SAS) in the
2	SEUS during the June and July of 2013. The NOAA WP-3D aircraft flew 18 research
3	flights based out of Smyrna, Tennessee during June and July 2013 for SENEX with a
4	payload of instruments measuring atmospheric trace gases, aerosol properties, and
5	meteorological parameters. This analysis uses measurements of carbon monoxide (CO),
6	carbon dioxide (CO ₂), methane (CH ₄), sulfur dioxide (SO ₂), isoprene, aerosol
7	composition, and aerosol optical properties (Table 1). The aerosol was sampled
8	downstream of a low turbulence inlet [Wilson et al., 2004] and an impactor with a 1 μ m
9	aerodynamic diameter size cut. Before impaction, the sampled aerosol was initially dried
10	by ram heating when sampling into the aircraft. The sampled aerosol was then dried
11	further in each instrument.
12	The non-refractory aerosol composition was measured by a compact time-of-flight
13	aerosol mass spectrometer (AMS) downstream of a pressure controlled inlet [Bahreini et
14	al., 2008] and most (97%) of the submicron volume measured by the aerosol sizing
15	instruments was transmitted into the AMS during SENEX. The collection efficiency for
16	the AMS was determined by the composition for each data point using the algorithm
17	described by Middlebrook et al. [2012]. When comparing the volume derived from
18	composition (AMS plus black carbon mass) to the volume measured by the aerosol sizing
19	instruments in the manner outlined by Bahreini et al. [2009], 87% of the aerosol
20	composition and sizing data from the entire SENEX study are within the combined
21	uncertainties.

22

23 **Table 1:** Measurements aboard the NOAA WP-3D used in this analysis. <u>The sample</u>

- 24 interval corresponds to the rate at which data is reported and is the integration time for
- 25 <u>the lower limit of detection.</u>

Measurement	Technique	Sample Interval	Lower Limit of Detection	Accuracy	Reference
Dry_Sub-micrometer Aerosol Extinction_(532 n m)-(dry)	Cavity Ringdown Spectrometer	1 s	0.1 Mm ⁻¹	5% (RH < 30%)	[Langridge et al., 2011]
Humidified Sub-micrometer Aerosol Extinction (532 nm)	Cavity Ringdown Spectrometer	<u>1 s</u>	<u>0.1 Mm⁻¹</u>	<u>11% (RH = 90%)</u>	[Langridge et al., 2011]
Sub-mircometer Aerosol	Compact - Time of Flight -	10 s	$OA < 0.8 - 4 \ \mu g \ m^{-1}$	<u>38% OA</u>	[Drewnick et al., 2005],
Composition	Aerosol Mass Spectrometer		3	34% Inorg.35%	[Canagaratna et al., 2007]
			$SO_4 < 0.1 - 0.05 \ \mu g$		
			m ⁻³		

Sub-micrometer Aerosol Volume	Optical particle counter	1 s	0.03 um ³ cm ⁻³	+26%, -12 %	[Cai et al., 2008]
СО	Vacuum UV- <u>Ultraviolet</u> fluorescence	1 s	0.5 ppbv	5%	[Holloway et al., 2000]
Isoprene	PTR-MSProton Transfer Reaction Mass Spectrometer	14 s	< 32 pptv	20%	[de Gouw and Warneke, 2007]
RH (Dew Point (RH))	Chilled Mirror Hygrometer	1 s	-	0.2°C	-
Black Carbon Mass	SP2Single Particle Soot Photometer	1 s	12 ng m ⁻³	30%	[Schwarz et al., 2008]
SO_2	Pulsed <u>UV-Ultraviolet</u> Fluorescence	1 s	250 pptv	20%	[Ryerson et al., 1998]
CH ₄	Cavity Ringdown Spectrometer	1 s		1.2 ppbv	[Peischl et al., 2012]
CO ₂	Cavity Ringdown Spectrometer	1 s	-	0.15 ppmv	[Peischl et al., 2012]

1

2 2.2 SEAC⁴RS

3	SEAC ⁴ RS consisted of measurements aboard three aircraft based in Houston,
4	Texas during August and September of 2013. In this analysis, we focus on in situ
5	measurements from the NASA DC-8 aircraft, which conducted 19 research flights. The
6	measurements that were collected onboard the DC-8 and used in this analysis are
7	summarized in Table 2. Unlike the SENEX study, there was no continuous measurement
8	of methane during SEAC ⁴ RS. The aerosol extinction and black carbon instruments
9	flown on the DC-8 during SEAC ⁴ RS were the same instruments used onboard the NOAA
10	WP-3D aircraft during SENEX. Measurements of aerosol extinction, volume, black
11	carbon mass sampled aerosol through a shrouded diffuser inlet described by McNaughton
12	et al. [2007]. The aerosol extinction was measured downstream of 1 μ m impactor.
13	The non-refractory aerosol composition was measured by a high resolution time-
14	of-flightThe aerosol mass spectrometer (AMS) similar to the compact time-of-flight
15	AMS used during the SENEX study. The two instrument differed in the resolution of the
16	mass spectrometer. The higher resolution AMS used during SEAC ⁴ ₂ RS was operated
17	with a 1 s sample interval. This AMS was operated similar to Dunlea et al. [2009] and
18	also used a pressure-controlled inlet [Bahreini et al., 2008]. The composition-dependent
10	formulation of Middlehungle at al. [2012] was used to estimate the collection officiance

- 19 formulation of Middlebrook et al. [2012] was used to estimate the collection efficiency.
- 20 <u>The AMS sampled aerosol downstream of sampled aerosol through aa HIMIL inlet.</u>
- 21 inlet (http://www.eol.ucar.edu/homes/dcrogers/Instruments/Inlets/). In both inletsthe

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1 <u>HIMIL inlet and the shrouded diffuser inlet</u>, the sampled aerosol was initially dried by

- 2 ram heating and then further dried in each instrument.
- 3

4 Table 2: Measurements aboard the NASA DC-8 used in this analysis. The sample

5 interval corresponds to the rate at which data is reported and is the integration time for

6 the lower limit of detection.

Measurement	Technique	Sample Interval	Lower Limit of Detection	Accuracy	Reference
Dry_Sub-micrometer Aerosol Extinction <u>(532 nm)</u> (dry)	Cavity Ringdown Spectrometer	1 s	0.1 Mm ⁻¹	5% (RH < 30%)	[Langridge et al., 2011]
Humidified Sub-micrometer Aerosol Extinction (532 nm)	Cavity Ringdown Spectrometer	<u>1 s</u>	<u>0.1 Mm⁻¹</u>	<u>11% (RH = 90%)</u>	[Langridge et al., 2011]
Sub-mircometer Aerosol	High Resolution - Time of Flight	1 s	ΟΑ- 0.6 μg m ⁻³ <u>ΟΑ</u>	35% <u>38% OA</u>	[Canagaratna et al., 2007
Composition	- Aerosol Mass Spectrometer		SO4- 0.06 µg m ⁻³ <u>SO</u> 4	<u>34% Inorg.</u>	
			NO 3-0.06 μg m ⁻³ <u>NO3</u>		
			ΝΗ4 -0.01 μg m ⁻³ <u>ΝΗ4</u>		
Sub-micrometer Aerosol Volume	Optical particle counter	1 s	0.03 um ³ cm ⁻³	+26%, -12 %	[Cai et al., 2008]
со	Infrared Absorption	1 s	0.5 ppbv	5%	[Sachse et al., 1987]
Isoprene	Proton Transfer Reaction Mass SpectrometerPTR-MS	14 s	25 pptv	10%	[de Gouw and Warneke, 2007]
RH (dew point)Dew Point (RH)	Chilled Mirror Hygrometer	1 s	-	0.2°C	
Black Carbon Mass	SP2Single Particle Soot	1 s	12 ng m ⁻³	30%	[Schwarz et al., 2008]
SO ₂	CIMSChemical Ionization Mass Spectrometer	1 s	9 pptv	15%	[S Kim et al., 2007]
CO ₂	Infrared Absorption	1 s	-	0.2 ppm	[Vay et al., 2011]

7

8 2.3 Surface Measurements

9 The Southeastern Aerosol Research and Characterization (SEARCH) Network

10 consists of 8 continuous monitoring ground sites in Georgia and Alabama hosting several

11 gas-phase and aerosol measurements [*Edgerton et al.*, 2005; 2006; *Hansen et al.*, 2003].

12 During SENEX the NOAA WP-3D flew over four of these sites a total of 15 times, and

13 extinction near the surface is calculated using measurements of aerosol scattering

14 (Radiance Research Model M903 nephelometer, Tempe, Arizona, USA) and absorption

15 (Magee Scientific Model AE-16 Aethalometer, Berkeley, California, USA) at each SEARCH

site to compare with the extinction measured onboard the NOAA WP-3D aircraft in the
 mixed layer.

3

4 2.4 Aerosol Water

5 The enhancement of extinction due to condensation of water onto the aerosol is

6 modeled using an empirical parameterization (shown in Eq. 1), hereafter referred to as

7 the kappa parameterization [*Brock et al.*, in preparation]. The hygroscopic growth of

8 particle diameter is described by kappa-Kohler theory [Petters and Kreidenweis, 2007].

9 A particle size distribution and a Mie scattering calculation would be necessary to

10 rigorously extend the kappa-Kohler theory to the hygroscopic enhancement of optical

11 properties. However, Brock et al. [in preparation] shows that if atmospheric

12 accumulation mode size distributions typical of the SEUS are used, the functional form of

13 kappa-Kohler theory can be applied directly to the optical extinction (Eq. 1).

14

$$\sigma_{ext}(RH) = \sigma_{ext}(dry) \times \left(1 + \kappa_{opt} \times \left(\frac{RH}{100 - RH}\right)\right)$$
(1)

16

17 The humidified extinction coefficient $\sigma_{ext}(RH)$ is a function of the dry extinction $\sigma_{ext}(dry)$ 18 and the hygroscopicity parameter κ_{opt} . We note that κ_{opt} is based on the measurement of 19 humidified extinction rather than the direct measurement of the diameter growth factor or 20 activation fraction, i. e. humidified tandem differential mobility analyzers (HTDMA) and 21 cloud condensation nuclei counters (CCNc). The aerosol extinction is measured in three 22 separate constant RH channels: in dry conditions (RH less than 30%), medium RH 23 typically 70%, and high RH greater than 80%. The hygroscopicity parameter (Koptk) is 24 determined by fitting the three measurements of extinction to Eq. 1. The resulting 25 $\kappa_{opt} \star$ and $\sigma_{ext}(dry)$ are then used to estimate the extinction at ambient RH. Lower values 26 of aerosol hygroscopicity generally correspond to mineral dust, aerosol with high soot 27 fraction, or primary OA such as fresh biomass burning emissions or automotive 28 emissions [Massoli et al., 2009; Quinn et al., 2005]. High hygroscopicity usually 29 corresponds to an oxidized, aged aerosol, large sulfate mass fractions and/or sea salt 30 aerosol.

1 The calculated ambient extinction is a lower limit of will differ from the actual 2 ambient extinction in three cases. First, if the hygroscopic growth exhibits hysteresis and 3 the ambient RH is below the deliquesce RH, ambient particles may be on the 4 deliquescing (lower) or efflorescing (upper) branch of the hysteresis curve [Santarpia et al., 2004]. Our extinction measurements -cannot distinguish between these two states, 5 because- the sample aerosol is first dried. and then re-Then the aerosol is humidified to 6 7 RH greater than 90% in a cooled Nafion humidifier. The temperature of the humidifier 8 (10-15 K below instrument temperature for 70% RH and 1-3 K below for 90% RH) is 9 varied to control amount of water vapor added to the sample and maintain a constant RH 10 in the sample cell. Finally the sample aerosol is reheated to the instrument temperature 11 and measured in the sample cell. in our instrument before measurement at high RH. (The 12 sample aerosol is exposed to elevated RH for a duration of 4 s before measurement.) 13 Hence, the measured extinction at high RH and the subsequently calculated 14 hygroscopicity parameter $\kappa_{opt_{a}}$ represent the hygroscopic growth of an 15 efflorescingdeliquescing aerosol on the lower-upper branch. If the aerosol undergoes hysteresis, we expected most aerosol in the summertime SEUS to be on the upper branch 16 17 because the aerosol regularly pass through clouds and are exposed to high RH conditions. 18 The aerosol also rarely experience dry conditions (RH less than 30%). Second, because 19 the kappa parameterization produces an ambient extinction that asymptotically 20 approaches infinity as RH approaches 100%, we used ambient RH to calculate the 21 ambient extinction only when RH was less than 95%, and assumed a constant RH of 95% 22 when ambient RH was greater than or equal to 95%. Therefore, the calculated ambient 23 extinction is a lower limit of the ambient extinction when RH was greater than 95%. 24 Third, in the case that super-micrometer particles (which are not sampled by instruments 25 used in this analysis) make a significant contribution, the ambient extinction is 26 underestimated. This is typically the case during dust events and during in-cloud 27 sampling which were either not observed or excluded from this analysis, respectively. 28 29 2.5 Aggregation of Vertical Profiles 30 Individual profiles are affected by horizontal advection which couples spatially 31 inhomogeneous emissions to the vertical profiles. Because of vertical wind shear and 20

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spatial variability during slant profiles, the vertical layers in an individual profile are not 1 2 always directly comparable. The aggregation the individual profiles is used to reduce the 3 influence of this variability and resolve the typical vertical structure and mixing over the 4 SEUS. For this analysis, individual Individual vertical profiles were selected from the 5 research flights by inspection of the altitude time series and are generally included for all ascents and descents with an altitude difference greater than 1 km. Measurements during 6 7 level flight legs were not used in this analysis. Although cloud penetration was mostly 8 avoided, aerosol data sampled during cloud penetration events were excluded due to the 9 effects of particle shattering in the inlets. Cloud penetration events were identified using 10 the video from the nose of the aircraft and cloud particle imaging probes mounted near 11 the wingtips. Transects of biomass burning plumes were identified using tracers such as 12 the acetonitrile mixing ratio-and, were typically during level flight legs, and were not 13 found during any of the profiles used in this analysis were excluded from the analysis. 14 Extensive aerosol parameters (mass, extinction, volume) have been corrected to standard 15 temperature (0 °C) and pressure (1 atmosphere). All calculated quantities such as ambient extinction and transition layer enhancements were determined before 16 17 aggregation and then were aggregated in the same manner as the observations. 18 For the SENEX campaign, the vertical profiles (which were located primarily over 19 northern Georgia and Alabama) were generally included in the flight plans for three 20 purposes: to characterize the background boundary layer structure before and after urban 21 and power plant plume intercepts, to characterize the vertical structure over surface 22 measurement sites, and as enroute ascents and descents into and out of the region of 23 interest. The SEAC⁴RS profiles that we use were distributed through Mississippi and 24 Alabama and were conducted to characterize inflow and outflow near convective 25 systems, to examine boundary layer chemistry over the SEUS, and as enroute ascents and 26 descents. The individual vertical profiles used here include both spiral and slant ascents 27 and descents and were typically between 5 and 15 min. in duration. 28 In this analysis we construct two types of aggregate profiles, the first of which 29 includes all of the afternoon vertical profiles over the SEUS and is binned according to 30 altitude above ground level ("altitude-binned" aggregate profiles). The second type of

31 aggregate includes only the subset of profiles during which shallow cumulus convection

was present and is binned according to a normalized altitude described in Sec. 3.2
 ("normalized" aggregate profiles). The profiles not included in the normalized
 aggregates were either collected during deeper convection and/or had a more complicated
 structure.

5 For the altitude-binned profiles, we have chosen all of the available vertical profiles from both SENEX and SEAC⁴RS over Mississippi, Alabama, and Georgia in the 6 7 afternoon between 12:00 pm and 6:30 pm Central Daylight Time (CDT) when we expect 8 the boundary layer structure to be well developed and no residual layers left over from 9 the previous day. The aggregate includes 74 profiles of which 41 profiles are from 6 10 research flights during SENEX and 33 profiles from 6 research flights during SEAC⁴RS. 11 The locations of the profiles used in the altitude-binned aggregate are shown in Fig. 1. 12 The data from individual vertical profiles were aggregated into 150 m vertical bins from the surface to 4.5 km based on the altitude above ground level. The vertical bin height of 13 14 150 m was chosen such that the slower measurements (aerosol mass and isoprene) 15 typically contributed at least one datum to each bin for each individual profile. In each bin with data from 5 or more individual profiles, the median, interquartile range, and 16 interdecile range were calculated. The median and percentiles were used because these 17 18 statistics are more robust when outliers are present. The number of vertical profiles 19 which contribute to each aggregated altitude bin varies with altitude because of the 20 differences of the starting and ending altitudes of each individual profile (Fig. 2a). 21 During some profiles or some portions of a profile, individual measurements of trace 22 gases and aerosol properties did not report data (e. g. due to zeroing or calibrations). 23 The normalized aggregate profiles were calculated using only those individual 24 profiles obtained during shallow cumulus convection and were altitude-normalized as 25 described below. Shallow cumulus convection is common globally and over the SEUS. 26 Warren et al. [2007] have compiled a global cloud climatology based on surface 27 observations. According to their work, the mean frequency of daytime cumulus clouds is 28 49% over Alabama during June, July, and August, and the mean cloud coverage when 29 cumulus clouds are present is 35%. The presence of shallow cumulus convection during 30 individual profiles was determined by inspection of visible images from the GOES 31 satellite and the presence of a three layer structure (mixed layer, transition layer, and free

troposphere), which is expected during shallow cumulus convection. The second 1 2 aggregate includes 37 of the 74 SEUS profiles from the first aggregate. The locations of 3 profiles in the second aggregate are show in Fig. 1, and Fig. 2b shows the distributions of 4 the mixed layer and transition layer heights determined from individual profiles which 5 had medians of 1.2 km and 2.2 km respectively. For cumulus convection the height of the planetary boundary layer is defined as the cloud base or the top of the mixed 6 7 layermixed layer height,; however, we find the term planetary boundary layer confusing 8 in the context of shallow cumulus convection and have avoided using it. Determination 9 of the mixed layer height and transition layer height is described in Sec. 3.2. For the 10 normalized aggregate profiles, there are 25 bins assigned to each layer. Figure 2c shows 11 the number of profiles that contribute to the second aggregate at each normalized altitude. 12 The number of profiles varies with normalized altitude due to variability in the starting 13 and ending altitudes of each profile and because the second aggregate is limited to 14 portions of the individual profiles when aerosol mass, extinction, volume and CO 15 measurements all report data. Limiting data as such facilitates quantitative comparison of the aggregate profiles. In contrast the first aggregate is constructed using all available 16 17 data. 18 19 3. Results 20 21 **3.1 Altitude-Binned Aggregate Profile** 22 Altitude-binned aggregate profiles of dry and ambient aerosol extinction show

23 several characteristics of note (Fig. 3). The median 532 nm dry extinction coefficient 24 (Fig. 3a) is approximately independent of altitude below 1.5 km with median value of 50 Mm⁻¹, and the interquartile range is 27 Mm⁻¹ to 73 Mm⁻¹. The interquartile (25th -75th 25 percentiles) and interdecile $(10^{th} - 90^{th} \text{ percentiles})$ range are due largely to variation 26 between individual profiles rather intra-profile point-to-point variation. Above 1.5 km 27 the extinction coefficient decreases with altitude to a median value of 6 Mm⁻¹ above 3.0 28 29 km. The gradual decrease in extinction with altitude from 1.2 to 2.5 km is due to the 30 variation of mixed layer and transition layer heights in the individual profiles. Figure 3b 31 shows the RH increasing with increasing altitude below 1.2 km. Above this level the RH

has a slight decreasing trend with altitude and a large interquartile range spanning 30-1 2 70%. The relative humidity of the aggregate profile could be biased low, because during 3 SENEX the flight dates were chosen to avoid precipitation and cloud penetration was 4 mostly avoided during flights however this was not the case during SEAC⁴RS. 5 The median hygroscopicity parameter (κ , Fig. 3c) increases from 0.11 at the 6 bottom of the profile to 0.18 at 3 km and is more variable above 3 km. The hygroscopic 7 growth of the aerosol enhances the ambient extinction (Fig. 3d) throughout the profile 8 and significantly between 0.7 and 1.7 km. Below 1.2 km, the ambient extinction 9 coefficient increases with altitude due to increasing RH, and above 1.2 km decreases with 10 altitude due to a combination of decreasing RH and decreasing dry extinction. The 11 hygroscopic growth of aerosol and the subsequent enhancement of extinction aloft could explain some of the enhancement of AOD noted by Goldstein et al. [2009] and Ford and 12 13 Heald [2013]. 14 The minimum altitude of individual aircraft profiles ranged from 300 - 700 m 15 above the surface. We estimate the profile of dry extinction between the surface and the 16 minimum altitude of the profiles by combining aircraft measurements made in the mixed 17 layer in the vicinity of surface monitoring sites using ground data from those sites. 18 During the SENEX study, there were 15 overflights in the mixed layer within 10 km of 19 four SEARCH monitoring sites. The surface aerosol extinction at each SEARCH site 20 was calculated using the aerosol scattering coefficient measured by a nephelometer with a 21 center wavelength of 530 nm and the aerosol absorption coefficients measured by an 22 aethalometer at 880 nm. Because the optical absorption at this wavelength was likely due 23 to black carbon aerosol, we corrected the absorption coefficient to 530 nm using an 24 Ångstrom exponent of 1 which is conventionally used for black carbon [Bergstrom et al., 25 2002; Lack and Langridge, 2013]. Absorption typically accounted for less than 5% of the extinction. The calculated 530 nm surface extinction was not corrected to the 532 nm 26 27 aircraft extinction because the correction would be less than 1%. The surface and aircraft 28 extinction coefficients are correlated ($R^2 = 0.91$), and the slope of a linear least 29 squares orthogonal distance regression (ODR) fit to the data indicates that the aircraft data 30 are $\sim 6\%$ lower than the surface measurements (Fig. 4), which is within the combined

31 uncertainty in the measurements. We conclude that the dry extinction is roughly

independent of altitude from the surface to the top of the well-mixed layer. Crumeyrolle 1 2 et al. [2014] found similar agreement between surface and aircraft-based boundary layer 3 measurements of ozone in the Baltimore-Washington metropolitan area. 4 The altitude-binned aggregate vertical profile of aerosol mass (Fig. 5a) is similar in shape to the dry extinction profile. The median mass is 13.7 μ g m⁻³ at the bottom of 5 the profile and decreases to 2.1 μ g m⁻³ above 3 km. The aerosol mass is the total of all 6 7 ions measured by the AMS, and these ions are typically classified as SO₄, NH₄, NO₃, and 8 OA. The inorganic ions may are typically represent formed by ionization of simple 9 inorganic salts such as ammonium sulfate and ammonium nitrate in the aerosol orbut may 10 be formed from more complex compounds with both an inorganic and organic character 11 (i. e. organosulfates, organonitrates, and amines) that produce both inorganic and organic 12 ions when ionized. To indicate this complexity, we have omitted ionic charges from the 13 notation (i. e. SO₄, NH₄, NO₃). In this classification scheme, the composition (Fig. 5b) 14 of the submicron aerosol is primarily OA, sulfate, and ammonium. The mass fraction of 15 the inorganic components (NO₃, NH₄, and SO₄) increase with altitude up to 3 km, while 16 the OA mass fraction decreases with altitude up to 3 km. Above 3 km, the OA fraction 17 increases; however, at this altitude the median aerosol mass is only 2 μ g m⁻³. The

18 increase of aerosol hygroscopicity with altitude up to 3 km corresponds with the

19 increasing inorganic fraction of the aerosol. In particular, sulfate is typically more

20 hygroscopic than OA, is 20% of the aerosol mass at the bottom of the profile, and is 28%

- 21 of the aerosol mass at 3 km.
- 22

23 3.2 Normalized Aggregate Profiles

The heights of the mixed and transition layers varies among individual vertical profiles and this variation obscures the transition layer in the altitude-binned aggregate profile presented in Figs. 3 and 5. For example, the layer structure is clear in a semi-rural profile measured in the vicinity of shallow cumulus convection over central Georgia on the afternoon of 16 June (Fig. 6). The mixed layer is closest to the surface, a transition layer is formed above the mixed layer, and the free troposphere is on top. The layered structure is evident in both the physical parameters such virtual

31 potential temperature (Θ_v) and ambient temperature as well as chemical mixing ratios

such as carbon monoxide (CO) and isoprene. In the mixed layer, adiabatically conserved 1 2 parameters such as virtual potential temperature (Fig. 6a) are independent of altitude. 3 However, in the transition the virtual potential temperature increases with altitude until 4 the top of the transition layer is reached where a capping temperature inversion is present. 5 In the mixed layer, the mixing ratio of water vapor is also independent of altitude; however, RH increases with altitude as temperature decreases (Fig. 6b). Relative 6 7 humidity is high in the transition layer, and video from the nose of the aircraft confirms 8 the presence of clouds in this layer. 9 The transition layer is also evident in the comparison of long-lived trace gases 10 such as CO (Fig. 6c) with short lifetime trace gases such as isoprene (Fig. 6d). Carbon 11 monoxide is directly emitted during combustion, produced by oxidation of hydrocarbons, 12 lost to oxidation by OH, and typically has an atmospheric lifetime of 1 - 4 months which 13 varies seasonally and regionally [Seinfeld and Pandis, 1998]. In this profile, the CO 14 mixing ratio is greater than 110 ppby in both the mixed and transition layers and 15 decreases to less than 100 ppbv in the free troposphere. In the mixed layer the CO 16 mixing ratio is independent of altitude and decreases modestly with altitude in the 17 transition layer. Isoprene is a short-lived trace gas that typically has an atmospheric 18 lifetime less than 2 hr. [Seinfeld and Pandis, 1998] and in the summertime is emitted by 19 vegetation common in the SEUS. In the mixed layer, the isoprene mixing ratio is greater 20 than 1 ppbv and variable due to heterogeneous surface emissions (Fig. 6d). The isoprene 21 mixing ratio in the transition layer is always less than 500 ppby and typically ~10% of the 22 mixed layer value. In the free troposphere, the isoprene mixing ratio is below the 23 detection limit of the measurement. 24 To examine vertical structure in more detail, altitude-normalized aggregate 25 profiles were calculated. Altitude normalization is commonly done by dividing the 26 altitude by the height of the mixed layer. However, because of the more complex vertical 27 structure often encountered during shallow cumulus convection, we have defined a 28 normalized altitude, h_{norm} , for each profile such that the top of the mixed layer, h_{ML} , is 29 assigned a normalized altitude of 1, and the top of the transition layer, h_{TL} , is assigned a 30 normalized altitude of 2:

31

1
$$0 < h < h_{ML}$$
 $h_{norm} = h / h_{ML}$

2
$$h_{ML} < h < h_{TL}$$
 $h_{norm} = 1 + (h - h_{ML})/(h_{TL} - h_{ML})$

3
$$h > h_{TL}$$
 $h_{norm} = 1 + h / h_{TL}$

4	
5	For individual profiles, the mixed layer height was defined determined by inspection of
6	<u>each profile</u> as the highest altitude at which the virtual potential temperature (Θ_v) was
7	constant (typical variation in the mixed layer was less than 0.5 K) and there was a
8	reduction in the isoprene concentration. The top of the transition layer was defined by a
9	temperature inversion and a rapid decrease in the CO mixing ratio.
10	The altitude-normalized aggregate profiles of CO (Fig. 7a) and isoprene (Fig. 7b)
11	mixing ratios demonstrate the contrast between the mixed layer and transition layer.
12	During shallow cumulus convection, CO is transported out of the mixed layer into the
13	transition layer due to its longer lifetime relative to isoprene. The modest decrease of CO
14	with altitude in the mixed layer is likely due to the influence of near source emissions in
15	some profiles. In the mixed layer the isoprene profile is variable, and the median is only
16	modestly dependent on altitude with a median mixing ratio of 1 ppbv. However, the
17	median isoprene mixing ratio decreases to $\sim 10\%$ of this value in the transition layer. The
18	isoprene observed above the mixed layer is consistent with large eddy simulations
19	performed by Kim et al. [2012] who found that cumulus clouds can transport some
20	isoprene out of the mixed layer into the cloud layer.
21	
22	4. Analysis
23	
24	4.1 Aerosol enhancements in the transition layer
25	During shallow cumulus convection, the air in the transition layer is a mixture of
26	air from the mixed layer below and the free troposphere above. The concentrations of
27	trace gases and extensive aerosol parameters $C(h)$ in the transition layer are described in
28	this analysis by a vertical mixing model consisting of three terms (Eq. 3): a contribution
29	from the mixed layer, a contribution from the free troposphere, and any enhancement
30	E(h) relative to concentration expected from the vertical mixing alone, as

(2)

2
$$C(h) = C_{ML} f_m(h) + C_{FT} (1 - f_m(h)) + E(h),$$
 (3)

3

1

4 where C(h) is the aerosol or trace gas concentration, C_{ML} and C_{FT} are the aerosol or trace 5 gas concentrations in the mixed layer and the free troposphere. Positive enhancements 6 could be due to local production, or direct emissions to the transition layer from buoyant 7 plumes, e. g. large biomass burning sources, and negative enhancements represent losses. 8 The fraction of air from the mixed layer (f_m) present in the transition layer is determined 9 by using the CO mixing ratio as

10

$$f_m(h) = \frac{CO(h) - CO_{FT}}{CO_{ML} - CO_{FT}},$$
(4)

12

11

13 for which the enhancement E(h) due to local production and losses is assumed to be 14 zero. For each profile, the mixing ratio of CO in the mixed layer $CO_{\rm ML}$ and the free 15 troposphere CO_{FT} were determined using the mean between normalized altitudes of 0.5 16 and 0.9 for the mixed layer and 2.0 and 2.5 for the free troposphere. To investigate 17 transition layer enhancements of chemical concentrations and aerosol extensive 18 properties, we calculate a concentration expected from vertical mixing alone using Eq. 3 19 and setting E(h) to zero. The concentration expected from vertical mixing alone is 20 calculated for each profile and aggregated in the same manner as the observations. In the 21 Figs. 8-10, the median concentration expected from vertical mixing alone is shown as a 22 dashed line on top of the observed concentrations. The interquartile and interdecile 23 ranges of the value expecting expected from mixing are not shown. 24 CO is produced through the oxidation of VOCs, and this CO production likely 25 accounts for a significant fraction of the CO budget during the summer in the SEUS 26 [Hudman et al., 2008]. If CO production in the transition layer is significant, the fraction 27 of air from the mixed layer (f_m , determined using CO) would be biased high and any 28 transition layer enhancements of other species (E(h), determined using the CO 29 concentration and Eqs. 3 and 4) would be biased low. By comparing the observed

concentration and the concentration expected from vertical mixing alone of several long 1 2 lifetime species, the importance of CO production in the transition layer can be assessed. 3 Fig. 8a shows the altitude normalized profile of the fraction of air from the mixed 4 layer (f_m , calculated using Eq. 4). The median is 1 in the mixed layer, 0 in the free 5 troposphere, and decreases from 1 to 0.6 in the transition layer due to entrainment of air from the free troposphere. Figs. 8b-e show the altitude-normalized aggregate profiles 6 7 and the median concentration expected from vertical mixing alone (dashed line) of 8 methane (CH₄), carbon dioxide (CO₂), water vapor (H₂O), and black carbon aerosol 9 mass, respectively. Both CH_4 and CO_2 have atmospheric lifetimes greater than a year 10 and no significant production or losses in the transition layer on the time scale of 11 atmospheric mixing. (The aggregate profile in Fig. 8b only includes data from the 12 SENEX study because CH₄ was not measured during SEAC⁴RS.) H₂O and black carbon 13 aerosol are also not produced in the transition layer and are not lost except in the presence 14 of precipitating clouds. Profiles in precipitating clouds are mostly excluded from the 15 aggregate. Based on the agreement between the observed vertical profiles of CH₄, CO₂, 16 H₂O, and black carbon mass (Fig. 8) and their expected concentration from vertical 17 mixing alone, we conclude the CO production in the transition layer is not significant. 18 In contrast, the altitude-normalized profiles of submicrometer aerosol mass (Fig. 19 9a), extinction (Fig. 9b), and volume (Fig. 9c) are greater than the value expected from 20 vertical mixing alone (dashed lines) in the transition layer. This indicates that E(h) is 21 positive for these aerosol properties. These transition layer enhancements are quantified 22 for individual profiles using the difference between the observed value and the value 23 expected from vertical mixing alone. The difference is expressed as a percentage of the 24 observed value and averaged over the transition between normalized altitudes of 1.1 and 25 1.9. The mean transition layer enhancements of aerosol mass, extinction, and volume 26 were +8.6%, +11.3%, and +9.3% respectively. The difference in the enhancements of 27 mass, extinction, and volume may reflect actual changes in the aerosol density and 28 extinction cross-section or could be due to imperfections in the measurements and data 29 aggregation. 30 Altitude-normalized aggregate profiles of aerosol composition are shown in Fig.

31 10. The enhancement of each aerosol component is quantified in the same manner as

1 aerosol mass, extinction, and volume. The observed median is greater than the value

2 $\hfill expected from vertical mixing alone by +6% for OA mass, +18% for SO4, +25% for$

3 NH₄, and +15% for NO₃. Although enhancement of sulfate is larger than OA as a

4 percentage, the absolute enhancement is a similar magnitude for both SO₄ and OA, ~0.5 5 μ g m⁻³.

6 The transition layer enhancements can be further investigated by examining the 7 distribution of enhancements for individual profiles (Fig. 11). For each profile, the 8 enhancement is calculated using the absolute difference between the observed value and 9 that expected from vertical mixing alone. The difference is averaged between normalized 10 altitudes of 1.1 and 1.9. Because the distributions of enhancements range from negative 11 to positive values, the Student's T-test is used to assess if the enhancement distributions 12 are statistical different from zero, or no enhancement. Enhancement distributions with p-13 values less than 0.05 are considered statistically significant. As expected, Figs. 11a-d 14 shows conserved species that do not have statistically significant enhancements, CH₄, 15 CO₂, H₂O, and black carbon mass. The enhancement distributions of aerosol mass, 16 extinction, and volume (Fig. 11e-g) are all statistically significant. Although both OA 17 and inorganic aerosol components are enhanced in the transition layer, the enhancement 18 distribution of OA is not statistically significant while the enhancement distribution of 19 inorganic components is significant. 20 Secondary aAerosol formation in the transition layer is the likely mechanism that 21 would lead to the observed enhancement of aerosol mass, volume, and extinction. The 22 enhancement of aerosol loading is the net result of production and loss in the transition 23 layer; however, profiles of black carbon and total sulfur (see section 4.2) suggest that the 24 aerosol losses are small. Secondary Agerosol formation in the transition layer is a 25 combination of aqueous production (both in clouds and and in aerosol water), 26 homogenous oxidation followed by condensation on existing particles, and condensation 27 of semi-volatile species such as NH₄NO₃. The presence of clouds within the transition 28 layer suggests a large role for aqueous production; however, our dataset does not allow us 29 to determine the relative importance of each pathway. 30 Biomass burning, common during the summer in the SEUS, emits aerosol in

31 buoyant plumes that, if large enough, could contribute to the observed enhancement of

1 aerosol loading in the transition layer and would not be consistent with the simple vertical

2 mixing model used here to describe the transition layer concentrations. Biomass burning

3 is common in the SEUS during the fall, winter, and spring-, but is less common during

4 the summer. Zhang et al. [2010] found that biomass burning contributed between 2 and

5 10% to measurements of $PM_{2.5}$ in the summer of 2007 and significantly more in other

6 seasons. Although, biomass burning plumes from agricultural fires were transected on

7 level flight legs during both SENEX and SEAC⁴RS, none of the profiles used in this

8 <u>analysis Profiles during whichincluded</u> fresh biomass burning plumes. were sampled

9 have been excluded from this analysis. However, the possibility remains that the

10 aggregate profiles are contaminated by aged and diluted biomass burning plumes which

11 have not been identified. To address this we considered biomass burning emission factors

12 of black carbon, sulfate, and sulfur dioxide (SO₂, which is oxidized in the atmosphere to

13 sulfate) reported by Akagi et al. [2011]. The emission factors range from 0.20 g/kg to

14 0.91 g/kg for black carbon mass and 0.45 g/kg to 0.87 g/kg for the combination of SO₂

15 and sulfate. Based on these emission factors, we would expect the ratio of the

16 combination of SO₂ and sulfate mass to black carbon mass in biomass burning plumes to

17 range from 0.5 to 4.35. If the observed enhancement of sulfate ($\sim 0.5 \, \mu g/m^3$), were due

18 exclusively to biomass burning, we would expect a concomitant enhancement of black

19 carbon (based on the ratio of emission factors for black carbon and sulfate) in the range

20 of 100 to 1000 ng/m³, which is not observed in the profile of black carbon mass (Fig. 8e).

21 Hence, we conclude the enhancement observed in the altitude-normalized aggregate

22 profile is not due to biomass burning.

23

24 **4.2 Sulfur budget**

Further evidence for the transition layer enhancement of particulate sulfate comes from the reduction of the concentration of gas-phase SO₂ in the transition layer.

27 Particulate sulfate is produced through gas-phase and aqueous oxidation of SO₂ [*Seinfeld*

and Pandis, 1998]. We expect that mixing in the transition layer would conserve total

29 sulfur which we define as the sum of particulate sulfate and gas phase SO₂. While

30 particulate sulfate is enhanced in the transition layer as described in Sect. 4.1, there is

also a reduction in the mixing ratio of gas phase SO_2 in the transition layer. Figures 12a-c

shows altitude-normalized aggregate profiles and values expected from vertical mixing 1

2 alone for particulate sulfate, gas phase SO₂, and the total sulfur. Particulate sulfate (Fig

3 12a) is enhanced by approximately the same amount as the reduction of SO₂ (Fig. 12b),

4 ~0.1 ppbv in the transition layer. Consequently, the median value of total sulfur agrees

- 5 well with the value expected from vertical mixing alone.
- 6 The enhancement distributions for particulate sulfate, SO₂, and total sulfur are

7 shown in Figs. 12 d-f. While the transition layer enhancement of particulate sulfate is

8 significant with a p-value of 3×10^{-5} (Fig. 12d), the reduction of SO₂ in the transition

9 layer (Fig. 12e) is not. The lack of statistical significance in SO₂ reduction is due to

10 positive outliers in the enhancement distribution. The enhancement distribution of total

11 sulfur indicates a small enhancement that is not statistically significant (Fig. 12f). We

note that the conservation of sulfate and SO₂ is only apparent when mixing in the 12

transition layer is taken into account. If biomass burning were the source of the transition 13

14 layer enhancement of particulate sulfate, we expect that total sulfur would be enhanced as 15

wella similar magnitude to particulate sulfate in the transition layer.

17 4.3 Aerosol Optical Depth

18 Aerosol optical depth (AOD) is typically measured remotely from space-based 19 satellites [King et al., 1999] and by ground-based sun photometer networks [Holben et 20 al., 2001]. These remote measurements of AOD have been complemented by AOD 21 calculated from aircraft-based in situ measurements of extinction which have the ability 22 to quantify contributions to the AOD from individual layers and aerosol water 23 [Crumeyrolle et al., 2014; Esteve et al., 2012]. Calculated from in situ measurements, 24 AOD is the integral of the ambient aerosol extinction coefficient (σ_{ext}), Eq. 65

26

16

$$AOD = \int_{surface}^{TOA} \sigma_{ext}(z) dz \,. \tag{65}$$

27

28 where σ_{ext} is a function of altitude z and the integration extends to the top of the

29 atmosphere (TOA). The extinction coefficient in Eq. 6-5 is at ambient relative humidity,

30 pressure and temperature. Several assumptions are necessary to calculate ambient

extinction and subsequently AOD. First, the aircraft profiles of dry extinction, relative 1 2 humidity, pressure, and temperature must be extrapolated to the surface. Because 3 extinction at the surface and aloft in the mixed layer are correlated (Fig. 4), T the dry 4 extinction is extrapolated as a constant to the surface based on the mean extinction 5 measured in the lowest 200 m of each profile. Relative humidity is extrapolated to the surface using the linear trend in the lowest 200 m of each profile if the trend is positive 6 7 (RH increases with increasing altitude); otherwise, it is extrapolated as a constant based 8 on the mean RH of the lowest 200 m of each profile. Pressure and temperature are both 9 extrapolated using the linear trend in the lowest 200 m of each profile. The second 10 assumption is that the contributions to AOD from aerosol layers above top of the aircraft 11 profile are negligible. For example, smoke from large forest fires in the western US can 12 be lofted high into the troposphere and transported over the SEUS [Peltier et al., 2007]. 13 This contribution to AOD cannot be included if the smoke layer were above the 14 maximum altitude of the profile. In this case the AOD calculated from the in situ profiles 15 is a lower limit. Third, we neglect the contribution to the AOD from supermicroeter particles, which we estimate to be less than 10% of the sub-micrometer AOD based on 16 coarse particle size distribution measurements made during both SENEX and SEAC⁴RS. 17 18 Fourth, because we have restricted calculated aerosol hygroscopic growth to RH values 19 less than 95%, the AOD calculated here is only a lower limit. 20 In addition to the AOD for each profile in the altitude-normalized aggregate, we 21 have also calculated the contributions to AOD from the mixed layer, the transition layer, 22 aerosol water, and the enhancement of aerosol extinction in the transition layer. The 23 median calculated AOD was 0.14 and the interquartile range (IQR) spanned 0.10 to 0.20 24 (Fig. 13a). An idealized profile of extinction during shallow cumulus convection is used 25 to show the contributions to AOD from the transition layer enhancement of extinction 26 (Fig. 13b), aerosol water (Fig. 13c), the transition layer (Fig. 13d), and the mixed layer 27 (Fig 13e). The contribution of the transition layer enhancement of ambient extinction 28 (median: 7%, IQR: 4%-10%) is split between the enhancement of dry extinction and the 29 aerosol water associated with the additional aerosol loading. The contribution of aerosol 30 water to the whole profile (median: 33%, IQR: 24%-38%) is sensitive to the aerosol 31 hygroscopicity parameter and ambient RH encountered. The transition layer contribution

(median: 45%, IQR: 33%-55%) was slightly smaller than the mixed layer contribution 1 2 (median: 48%, IQR: 38%-57%). The mixed layer's slightly greater vertical extent and 3 higher average dry extinction favor a larger contribution to AOD; however, the transition 4 layer also provides a substantial contribution to AOD because of the aerosol water 5 associated with the higher mean RH in the transition layer. The contributions to AOD 6 presented in Fig. 13 have substantial overlap (i. e. aerosol water also contributes to mixed 7 and transition layer AOD contributions); hence, the contributions do not add to unity. 8 The altitude-normalized aggregate profiles used in this analysis are drawn from 9 37 vertical profiles; however, they represent only eight afternoons during the summer of 10 2013. For comparison and context, Fig. 14 shows an extended time series of 532 nm 11 AOD (level 2 data) measured by AERONET sun photometers [Holben et al., 2001] at the 12 Centreville SEARCH site and at the Georgia Tech site in Atlanta, Georgia. The Georgia Tech site is in an urban area and is perhaps biased toward larger AOD from urban 13 14 emissions, while the Centreville site is rural. The sun photometers only report data 15 during cloud-free conditions. Plotted on top of these data from the sun photometers are the AOD from the profiles used in the altitude-normalized aggregate. These data are 16 grouped into the profiles from the SENEX and SEAC⁴RS studies. Aircraft profiles 17 18 during the SENEX study did not sample AOD greater than 0.3 while the maximum of 19 AOD observed by the sun photometers was greater than 0.4. Profiles during the 20 SEAC⁴RS study, although limited in number, cover a range of AOD similar to the sun 21 photometers. Because the majority of the profiles in the altitude-normalized aggregate 22 are from the SENEX study, the aggregate may be biased toward cleaner conditions. The 23 range of AOD observed during summers of 2011-2013 at the Georgia Tech site indicate 24 that the summer of 2013 is not an outlier with AOD higher or lower than typical 25 summers. This is consistent with the analysis of Kim et al. [2015] who has compared 26 satellite measurements of AOD during the summers 2006-2013. 27 For comparison, Fig 13a (dashed) shows a histogram of 500 nm AOD 28 (measured by an AERONET sun photometer [Holben et al., 2001] at the Centreville 29 SEARCH site) constructed from all available data between 1 May and 30 September 30 2013. The AOD from the sun photometer is higher than the AOD from the profiles in the 31 normalized aggregate, and this difference likely reflects bias in sampling shallow

1 cumulus convection during the in situ profiles compared to more uniform sampling by 2 the sun photometer. The in situ profiles presented here are weighted toward relatively 3 clean early summer conditions in the presence of shallow cumulus convection, and may 4 be biased against late summer biomass burning influence from the western U.S. and 5 Canada. 6 Several SEUS studies have noted decreases in anthropogenic emissions (sulfur, nitrogen 7 oxides, and volatile organic compounds) in the first decade of the 21st century [Alston et 8 al., 2012; Attwood et al., 2014; Hand et al., 2012; Hand et al., 2013]. Concurrently, 9 particulate sulfate, OA, and AOD have also decreased. Alston et al. [2012] have shown 10 that the summertime mean AOD over Georgia reported by the MISR instrument on the 11 Terra spacecraft decreased from ~0.3 in the summer of 2000 to less than 0.2 in the 12 summer of 2009, which is in the range of AOD calculated in this work for the summer 13 2013. 14 15 5. Conclusions 16 Several preceding studies have observed vertical transport of trace gases and 17 aerosol from the mixed layer into the cloud-influenced transition layer during shallow 18 cumulus convection [Angevine, 2005; Ching and Alkezweeny, 1986; Greenhut, 1986; 19 Langford et al., 2010]. Our observations are consistent with this earlier work. In 20 addition to vertical transport and redistribution of aerosol, we observed a modest

21 enhancement of aerosol loading in the transition layer and conclude that secondary

22 aerosol formation in the transition layer is the likely source of the enhancement.

23 Although we cannot distinguish between condensational and aqueous aerosol formation

24 pathways, the presence of clouds and elevated relative humidity in the transition layer

- 25 suggests a potential role for aqueous reactions. Using measurements of particulate
- 26 oxalate as a tracer for aqueous processing, Wonaschuetz et al. [2012] and Sorooshian et

27 al. [2007] have also observed evidence for secondary aerosol formation in the transition

28 layer during cumulus convection over Texas and near the coast of California.

- 29 Wonaschuetz et al. [2012] show no trends in the OA and particulate sulfate mass
- 30 fractions with altitude in the mixed and transition layers, which could occur if the
- 31 production was sufficiently small or if the additional aerosol mass in the transition layer

were produced with the same ratio of OA and particulate sulfate that was originally 1 2 present in the mixed layer. In contrast, our measurements show a distinct difference in 3 composition between the mixed and transition layers and imply a similar magnitude of 4 secondary sulfate and OA production in the transition layer, although the production of 5 OA was not statistically significant. Goldstein et al. [2009] and Ford and Heald [2013] hypothesized a layer of aerosol 6 7 that is primarily composed of SOA, that would be sufficient to explain a significant 8 fraction of the observed summertime enhancement of AOD (2-3 times greater than 9 winter), and that does not contribute to aerosol mass at the surface. The hypothesis is 10 partially supported by the spatial similarity of summertime biogenic emission and summertime AOD over the SEUS. Although, Alston et al. [2012] found that the spatial 11 12 similarity depended on the spatial resolution AOD of the data used in the analysis, and their analysis of AOD and surface aerosol mass over Georgia did not fully support the 13 14 hypothesis. Kim et al. [2015] found that the increase of the planetary boundary layer 15 height during the summer could bring the seasonality of the AOD and surface aerosol 16 mass into agreement without the need for an enhanced aerosol layer aloft. 17 Here, we have examined in situ vertical profiles of aerosol and found the dry 18 aerosol to be well mixed in the lowest layer. Above the lowest layer, the aerosol mass 19 and extinction decreased with increasing altitude above that layer (Figs. 3-5). The 20 hygroscopic growth of aerosol at high RH resulted in a layer of enhanced extinction near 21 the top of the mixed layer. The aerosol water accounted for approximately a third of the 22 AOD which would explain a portion of the summertime AOD enhancement. The 23 hypothesized, large enhancement of secondary aerosol aloft was not apparent in these 24 aggregate profiles. However, after normalizing the altitude to the vertical structure and 25 using the CO concentration to quantify the vertical mixing (Figs. 7-10), we were able to 26 resolve a Although we have observed modestan enhancement of aerosol in the transition 27 layer. -This layer wasit is not consistent with the hypothesized layer in magnitude-or-, 28 and the observed composition was not consistent with the SOA dominated layer 29 hypothesized by Goldstein et al. (2009). We observed enhancements that were less than 30 10% of AOD, and sulfate and OA were enhanced by similar magnitude although the OA 31 enhancement was not statistically significant. The seasonality of the enhancement of

- 1 surface aerosol mass (less than 1.6 times greater in summer than winter) compared to the
- 2 AOD enhancement (2-3 times) was the primary evidence for the hypothesized layer.
- 3 Given the absence of such a layer, our observations suggest that other factors such as
- 4 meteorology and transport may influence the seasonality of the relationship of AOD to
- 5 surface aerosol mass and warrant further investigation.
- 6

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- 20

21 References

- Akagi, S. K., R. J. Yokelson, C. Wiedinmyer, M. J. Alvarado, J. S. Reid, T. Karl, J. D.
 Crounse, and P. O. Wennberg (2011), Emission factors for open and domestic
 biomass burning for use in atmospheric models, *Atmospheric Chemistry and Physics*, 11(9), 4039-4072, doi: 10.5194/acp-11-4039-2011.
- 26 Alston, E. J., I. N. Sokolik, and O. V. Kalashnikova (2012), Characterization of
- atmospheric aerosol in the US Southeast from ground- and space-based
 measurements over the past decade, *Atmospheric Measurement Techniques*, 5(7),
 1667-1682, doi: 10.5194/amt-5-1667-2012.
- Angevine, W. M. (2005), An integrated turbulence scheme for boundary layers with shallow cumulus applied to pollutant transport, *J. Appl. Meteorol.*, 44(9), 1436-1452,
- doi: 10.1175/jam2284.1.
 Attwood , A. R., et al. (2014), Trends in sulfate and organic aerosol mass in the Southeast
- U.S.: Impact on aerosol optical depth and radiative forcing, *Geophys. Res. Lett.*, 41,
 7701–7709, doi: 10.1002/2014GL061669.

- Bahreini, R., E. J. Dunlea, B. M. Matthew, C. Simons, K. S. Docherty, P. F. DeCarlo, J.
 L. Jimenez, C. A. Brock, and A. M. Middlebrook (2008), Design and operation of a
 pressure-controlled inlet for airborne sampling with an aerodynamic aerosol lens, in
 Aerosol Science and Technology, edited, pp. 465-471.
- Bahreini, R., et al. (2009), Organic aerosol formation in urban and industrial plumes near
 Houston and Dallas, Texas, *Journal of Geophysical Research-Atmospheres*, 114,
 doi: 10.1029/2008jd011493.
- Bergstrom, R. W., P. B. Russell, and P. Hignett (2002), Wavelength dependence of the
 absorption of black carbon particles: Predictions and results from the TARFOX
 experiment and implications for the aerosol single scattering albedo, *J. Atmos. Sci.*,
 59(3), 567-577, doi: 10.1175/1520-0469(2002)059<0567:wdotao>2.0.co;2.
- Brock, C. A., et al. (in preparation), Sensitivity of Aerosol Optical Depth to Aerosol and
 Meteorological Parameters in the Southeastern United States in Summer.
- Cai, Y., D. C. Montague, W. Mooiweer-Bryan, and T. Deshler (2008), Performance
 characteristics of the ultra high sensitivity aerosol spectrometer for particles between
 55 and 800 nm: Laboratory and field studies, *Journal of Aerosol Science*, *39*(9), 759769, doi: 10.1016/j.jaerosci.2008.04.007.
- Canagaratna, M. R., et al. (2007), Chemical and microphysical characterization of
 ambient aerosols with the aerodyne aerosol mass spectrometer, *Mass Spectrometry Reviews*, 26(2), 185-222, doi: 10.1002/mas.20115.
- Carlton, A. G., and B. J. Turpin (2013), Particle partitioning potential of organic
 compounds is highest in the Eastern US and driven by anthropogenic water,
 Atmospheric Chemistry and Physics, 13(20), 10203-10214, doi: 10.5194/acp-13 10203-2013.
- Carlton, A. G., B. J. Turpin, K. E. Altieri, S. P. Seitzinger, R. Mathur, S. J. Roselle, and
 R. J. Weber (2008), CMAQ Model Performance Enhanced When In-Cloud
 Secondary Organic Aerosol is Included: Comparisons of Organic Carbon Predictions
 with Measurements, *Environ. Sci. Technol.*, 42(23), 8798-8802, doi:
 10.1021/es801192n.
- Ching, J. K. S., and A. J. Alkezweeny (1986), Tracer Study of Vertical Exchange by
 Cumulus Clouds, *Journal of Climate and Applied Meteorology*, 25(11), 1702-1711,
 doi: 10.1175/1520-0450(1986)025<1702:tsoveb>2.0.co;2.
- Ching, J. K. S., S. T. Shipley, and E. V. Browell (1988), Evidence for Cloud Venting of
 Mixed Layer Ozone and Aerosols, *Atmospheric Environment*, 22(2), 225-242, doi:
 10.1016/0004-6981(88)90030-3.
- Crumeyrolle, S., G. Chen, L. Ziemba, A. Beyersdorf, L. Thornhill, E. Winstead, R. H.
 Moore, M. A. Shook, C. Hudgins, and B. E. Anderson (2014), Factors that influence
 surface PM2.5 values inferred from satellite observations: perspective gained for the
 US Baltimore-Washington metropolitan area during DISCOVER-AQ, *Atmospheric Chemistry and Physics*, 14(4), 2139-2153, doi: 10.5194/acp-14-2139-2014.
- de Gouw, J. A., and C. Warneke (2007), Measurements of volatile organic compounds in
 the earths atmosphere using proton-transfer-reaction mass spectrometry, *Mass*
- 43 Spectrometry Reviews, 26(2), 223-257, doi: 10.1002/mas.20119.
- de Gouw, J. A., and J. L. Jimenez (2009), Organic Aerosols in the Earth's Atmosphere, *Environ. Sci. Technol.*, 43(20), 7614-7618, doi: 10.1021/es9006004.

- Drewnick, F., et al. (2005), A new time-of-flight aerosol mass spectrometer (TOF-AMS)
 Instrument description and first field deployment, *Aerosol Science and Technology*,
 39(7), 637-658, doi: 10.1080/02786820500182040.
- Dunlea, E. J., et al. (2009), Evolution of Asian aerosols during transpacific transport in
 INTEX-B, *Atmospheric Chemistry and Physics*, 9(19), 7257-7287.
- Eatough, D. J., F. M. Caka, and R. J. Farber (1994), The Conversion of SO₂ to Sulfate in
 the Atmosphere, *Isr. J. Chem.*, *34*(3-4), 301-314, doi: 10.1002/ijch.199400034.
- Edgerton, E. S., B. E. Hartsell, R. D. Saylor, J. J. Jansen, D. A. Hansen, and G. M. Hidy
 (2005), The Southeastern Aerosol Research and Characterization Study: Part II.
 Filter-based Measurements of Fine and Coarse Particulate Matter Mass and
 Composition, *Journal of the Air & Waste Management Association*, *55*(10), 1527-
- 1542, doi: 10.1080/10473289.2005.10464744.
 Edgerton, E. S., B. E. Hartsell, R. D. Saylor, J. J. Jansen, D. A. Hansen, and G. M. Hidy
- Edgerton, E. S., B. E. Hartsell, R. D. Saylor, J. J. Jansen, D. A. Hansen, and G. M. Hidy
 (2006), The Southeastern Aerosol Research and Characterization Study, part 3:
 Continuous Measurements of Fine Particulate Matter Mass and Composition, *Journal of the Air & Waste Management Association*, 56(9), 1325-1341, doi:
 10.1080/10473289.2006.10464585.
- Ervens, B., B. J. Turpin, and R. J. Weber (2011), Secondary organic aerosol formation in
 cloud droplets and aqueous particles (aqSOA): a review of laboratory, field and
 model studies, *Atmospheric Chemistry and Physics*, 11(21), 11069-11102, doi:
 10.5194/acp-11-11069-2011.
- Esteve, A. R., J. A. Ogren, P. J. Sheridan, E. Andrews, B. N. Holben, and M. P. Utrillas
 (2012), Sources of discrepancy between aerosol optical depth obtained from
 AERONET and in-situ aircraft profiles, *Atmospheric Chemistry and Physics*, 12(6),
 2987-3003, doi: 10.5194/acp-12-2987-2012.
- Ford, B., and C. L. Heald (2013), Aerosol loading in the Southeastern United States:
 reconciling surface and satellite observations, *Atmospheric Chemistry and Physics*, *13*(18), 9269-9283, doi: 10.5194/acp-13-9269-2013.
- Goldstein, A. H., C. D. Koven, C. L. Heald, and I. Y. Fung (2009), Biogenic carbon and
 anthropogenic pollutants combine to form a cooling haze over the southeastern
- United States, *Proceedings of the National Academy of Sciences of the United States* of America, 106(22), 8835-8840, doi: 10.1073/pnas.0904128106.
- Greenhut, G. K. (1986), Transport of Ozone Between Booundary-Layer and Cloud Layer
 by Cumulus Clouds, *Journal of Geophysical Research-Atmospheres*, *91*(D8), 8613 8622, doi: 10.1029/JD091iD08p08613.
- Hand, J. L., B. A. Schichtel, W. C. Malm, and M. L. Pitchford (2012), Particulate sulfate
 ion concentration and SO₂ emission trends in the United States from the early 1990s
 through 2010, *Atmospheric Chemistry and Physics*, *12*(21), 10353-10365, doi:
 10.5194/acp-12-10353-2012.
- Hand, J. L., B. A. Schichtel, W. C. Malm, and N. H. Frank (2013), Spatial and Temporal
 Trends in PM2.5 Organic and Elemental Carbon across the United States, *Advances in Meteorology*, doi: 10.1155/2013/367674.
- Hansen, D. A., E. S. Edgerton, B. E. Hartsell, J. J. Jansen, N. Kandasamy, G. M. Hidy,
 and C. L. Blanchard (2003), The Southeastern Aerosol Research and
- 45 Characterization Study: Part 1-Overview, *Journal of the Air & Waste Management*46 *Association*, 53(12), 1460-1471, doi: 10.1080/10473289.2003.10466318.

He, C., J. Liu, A. G. Carlton, S. Fan, L. W. Horowitz, H. Levy, II, and S. Tao (2013), Evaluation of factors controlling global secondary organic aerosol production from cloud processes, *Atmospheric Chemistry and Physics*, 13(4), 1913-1926, doi: 10.5194/acp-13-1913-2013.

1 2

3

- Hoff, R. M., and S. A. Christopher (2009), Remote Sensing of Particulate Pollution from
 Space: Have We Reached the Promised Land?, *Journal of the Air & Waste Management Association*, 59(6), 645-675, doi: 10.3155/1047-3289.59.6.645.
- Holben, B. N., et al. (2001), An emerging ground-based aerosol climatology: Aerosol optical depth from AERONET, *Journal of Geophysical Research-Atmospheres*, 106(D11), 12067-12097, doi: 10.1029/2001jd900014.
- Holloway, J. S., R. O. Jakoubek, D. D. Parrish, C. Gerbig, A. Volz-Thomas, S.
 Schmitgen, A. Fried, B. Wert, B. Henry, and J. R. Drummond (2000), Airborne
 intercomparison of vacuum ultraviolet fluorescence and tunable diode laser
 absorption measurements of tropospheric carbon monoxide, *Journal of Geophysical*
- *Research-Atmospheres*, *105*(D19), 24251-24261, doi: 10.1029/2000jd900237.
 Hudman, R. C., L. T. Murray, D. J. Jacob, D. B. Millet, S. Turquety, S. Wu, D. R. Blake,
- A. H. Goldstein, J. Holloway, and G. W. Sachse (2008), Biogenic versus
 anthropogenic sources of CO in the United States, *Geophys. Res. Lett.*, 35(4), doi:
 10.1029/2007gl032393.
- Kim, P. S., et al. (2015), Sources, seasonality, and trends of Southeast US aerosol: an
 integrated analysis of surface, aircraft, and satellite observations with the GEOS Chem chemical transport model, *ACPD*, (*submitted*).
- Kim, S., et al. (2007), Measurement of HO₂NO₂ in the free troposphere during the
 intercontinental chemical transport experiment North America 2004, *Journal of Geophysical Research-Atmospheres*, *112*(D12), doi: 10.1029/2006jd007676.
- Kim, S. W., M. C. Barth, and M. Trainer (2012), Influence of fair-weather cumulus
 clouds on isoprene chemistry, *Journal of Geophysical Research-Atmospheres*, *117*,
 doi: 10.1029/2011jd017099.
- King, M. D., Y. J. Kaufman, D. Tanre, and T. Nakajima (1999), Remote sensing of
 tropospheric aerosols from space: Past, present, and future, *Bulletin of the American Meteorological Society*, 80(11), 2229-2259, doi: 10.1175/15200477(1999)080<2229:rsotaf>2.0.co;2.
- Lack, D. A., and J. M. Langridge (2013), On the attribution of black and brown carbon
 light absorption using the Angstrom exponent, *Atmospheric Chemistry and Physics*,
 13(20), 10535-10543, doi: 10.5194/acp-13-10535-2013.
- Langford, A. O., S. C. Tucker, C. J. Senff, R. M. Banta, W. A. Brewer, R. J. Alvarez, R.
 M. Hardesty, B. M. Lerner, and E. J. Williams (2010), Convective venting and
 surface ozone in Houston during TexAQS 2006, *Journal of Geophysical Research- Atmospheres*, *115*, doi: 10.1029/2009jd013301.
- Langridge, J. M., M. S. Richardson, D. Lack, D. Law, and D. M. Murphy (2011), Aircraft
 Instrument for Comprehensive Characterization of Aerosol Optical Properties, Part I:
 Wavelength-Dependent Optical Extinction and Its Relative Humidity Dependence
- 42 wavelength-Dependent Optical Extinction and its Relative Humany Dependence 43 Measured Using Cavity Ringdown Spectroscopy, *Aerosol Science and Technology*,
- 44 45(11), 1305-1318, doi: 10.1080/02786826.2011.592745.

- Luria, M., and H. Sievering (1991), Heterogeneous and Homogeneous Oxidation of SO₂
 in the Marine Atmosphere, *Atmospheric Environment Part a-General Topics*, 25(8),
 1489-1496, doi: 10.1016/0960-1686(91)90008-u.
- Massoli, P., T. S. Bates, P. K. Quinn, D. A. Lack, T. Baynard, B. M. Lerner, S. C.
 Tucker, J. Brioude, A. Stohl, and E. J. Williams (2009), Aerosol optical and
 hygroscopic properties during TexAQS-GoMACCS 2006 and their impact on
 aerosol direct radiative forcing, *Journal of Geophysical Research-Atmospheres*, 114,
- 8 doi: 10.1029/2008jd011604.
- McKeen, S., et al. (2007), Evaluation of several PM2.5 forecast models using data
 collected during the ICARTT/NEAQS 2004 field study, *Journal of Geophysical Research-Atmospheres*, *112*(D10), doi: 10.1029/2006jd007608.
- 12 McNaughton, C. S., et al. (2007), Results from the DC-8 Inlet Characterization
- 13 Experiment (DICE): Airborne versus surface sampling of mineral dust and sea salt 14 aerosols, *Aerosol Science and Technology*, 41(2), 136-159, doi:
- 15 10.1080/02786820601118406.
- Middlebrook, A. M., R. Bahreini, J. L. Jimenez, and M. R. Canagaratna (2012),
 Evaluation of Composition-Dependent Collection Efficiencies for the Aerodyne
 Aerosol Mass Spectrometer using Field Data, *Aerosol Science and Technology*,
 46(3), 258-271, doi: 10.1080/02786826.2011.620041.
- Peischl, J., et al. (2012), Airborne observations of methane emissions from rice
 cultivation in the Sacramento Valley of California, *Journal of Geophysical Research-Atmospheres*, 117, doi: 10.1029/2012jd017994.
- Peltier, R. E., A. P. Sullivan, R. J. Weber, C. A. Brock, A. G. Wollny, J. S. Holloway, J.
 A. de Gouw, and C. Warneke (2007), Fine aerosol bulk composition measured on
 WP-3D research aircraft in vicinity of the Northeastern United States results from
 NEAQS, *Atmospheric Chemistry and Physics*, 7(12), 3231-3247, doi: 10.5194/acp7-3231-2007.
- Petters, M. D., and S. M. Kreidenweis (2007), A single parameter representation of
 hygroscopic growth and cloud condensation nucleus activity, *Atmospheric Chemistry and Physics*, 7(8), 1961-1971.
- Quinn, P. K., et al. (2005), Impact of particulate organic matter on the relative humidity
 dependence of light scattering: A simplified parameterization, *Geophys. Res. Lett.*,
 32(22), doi: 10.1029/2005gl024322.
- Ryerson, T. B., et al. (1998), Emissions lifetimes and ozone formation in power plant
 plumes, *Journal of Geophysical Research-Atmospheres*, *103*(D17), 22569-22583,
 doi: 10.1029/98jd01620.
- Sachse, G. W., G. F. Hill, L. O. Wade, and M. G. Perry (1987), Fast Response, High
 Precision Carbon Monoxide Sensor Using a Tunable Diode-Laser Absorption
 Technique, *Journal of Geophysical Research-Atmospheres*, 92(D2), 2071-2081, doi:
 10.1029/JD092iD02p02071.
- Santarpia, J. L., R. J. Li, and D. R. Collins (2004), Direct measurement of the hydration
 state of ambient aerosol populations, *Journal of Geophysical Research-Atmospheres*,
 109(D18), doi: 10.1029/2004jd004653.
- Schwarz, J. P., et al. (2008), Measurement of the mixing state, mass, and optical size of
 individual black carbon particles in urban and biomass burning emissions, *Geophys. Res. Lett.*, 35(13), doi: 10.1029/2008gl033968.

l	Seinfeld, J. H., and S. N. Pandis (1998), Atmospheric Chemistry and Physics, John Wiley
2	& Sons, New York.

- Siebesma, A. P. (1998), Shallow Cumulus Convection, in *Buoyant Convection in Geophysical Flows*, edited by E. J. Plate, E. E. Fedorovich, D. X. Viegas and J. C.
 Wyngaard, pp. 441-486, Kluwer Academic Publishers, Dordrecht.
- Sorooshian, A., M.-L. Lu, F. J. Brechtel, H. Jonsson, G. Feingold, R. C. Flagan, and J. H.
 Seinfeld (2007), On the source of organic acid aerosol layers above clouds, *Environ*. *Sci. Technol.*, 41(13), 4647-4654, doi: 10.1021/es0630442.
- Sorooshian, A., et al. (2006), Oxalic acid in clear and cloudy atmospheres: Analysis of
 data from International Consortium for Atmospheric Research on Transport and
 Transformation 2004, *Journal of Geophysical Research-Atmospheres*, *111*(D23),
 doi: 10.1029/2005jd006880.
- Vay, S. A., et al. (2011), Patterns of CO₂ and radiocarbon across high northern latitudes
 during International Polar Year 2008, *Journal of Geophysical Research- Atmospheres, 116*, doi: 10.1029/2011jd015643.
- Warren, S. G., R. M. Eastman, and C. J. Hahn (2007), A survey of changes in cloud
 cover and cloud types over land from surface observations, 1971-96, *Journal of Climate*, 20(4), 717-738, doi: 10.1175/jcli4031.1.
- Weber, R. J., et al. (2007), A study of secondary organic aerosol formation in the
 anthropogenic-influenced southeastern United States, *Journal of Geophysical Research-Atmospheres*, 112(D13), doi: 10.1029/2007jd008408.
- Wilson, J. C., B. G. Lafleur, H. Hilbert, W. R. Seebaugh, J. Fox, D. W. Gesler, C. A.
 Brock, B. J. Huebert, and J. Mullen (2004), Function and performance of a low
 turbulence inlet for sampling supermicron particles from aircraft platforms, *Aerosol Science and Technology*, *38*(8), 790-802, doi: 10.1080/027868290500841.
- Wonaschuetz, A., A. Sorooshian, B. Ervens, P. Y. Chuang, G. Feingold, S. M. Murphy, J.
 de Gouw, C. Warneke, and H. H. Jonsson (2012), Aerosol and gas re-distribution by
 shallow cumulus clouds: An investigation using airborne measurements, *Journal of Geophysical Research-Atmospheres*, *117*, doi: 10.1029/2012jd018089.
- Zhang, X., A. Hecobian, M. Zheng, N. H. Frank, and R. J. Weber (2010), Biomass
 burning impact on PM2.5 over the southeastern US during 2007: integrating
 chemically speciated FRM filter measurements, MODIS fire counts and PMF
 analysis, *Atmospheric Chemistry and Physics*, 10(14), 6839-6853, doi: 10.5194/acp10-6839-2010.
- 35

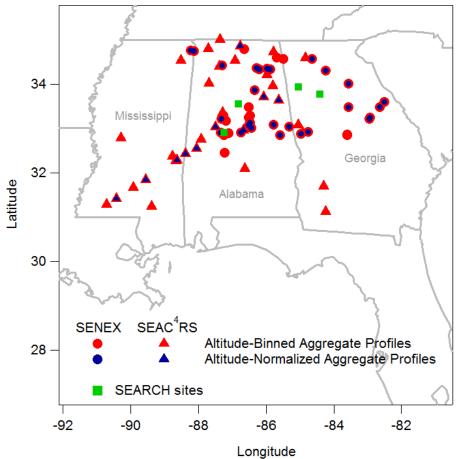


Figure 1: The locations of the vertical profiles from the SENEX (circles) and SEAC⁴RS 3 (triangles) and SEARCH monitoring sites (green squares). The markers (both red and blue) are the locations of afternoon profiles used to construct the altitude-binned aggregate profile that includes 74 profiles: 41 from SENEX and 33 from SEAC⁴RS. The blue markers show the location of the profiles used to construct the altitude-normalized aggregate profile that includes 37 profiles: 27 from SENEX and 10 from SEAC⁴RS. 8

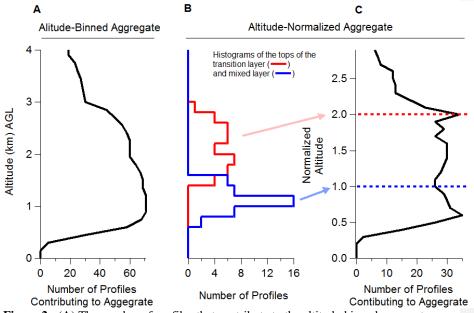
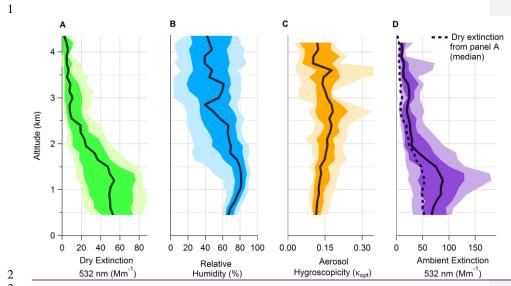


Figure 2: (**A**) The number of profiles that contribute to the altitude-binned aggregate, (**B**) histograms of the altitude of the tops of the transition and mixed layers, and (**C**) the

number of profiles that contribute to the altitude-normalized aggregate.



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Figure 3: Altitude-binned aggregate profiles of (A) the 532 nm dry aerosol extinction,

(B) relative humidity (calculated from dew point measurements), (C) aerosol

6 hygroscopicity (humidified extinction measurements fit to Eq. 1), and (**D**) the calculated

7 ambient extinction (Eq. 1). The shaded regions show the interdecile range (light) and the

8 interquartile range (medium), and the solid lines are the median (dark). The dashed line
9 in panel D shows the median dry extinction for comparison.

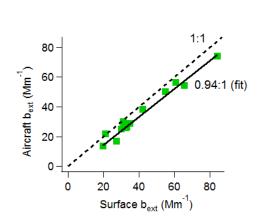
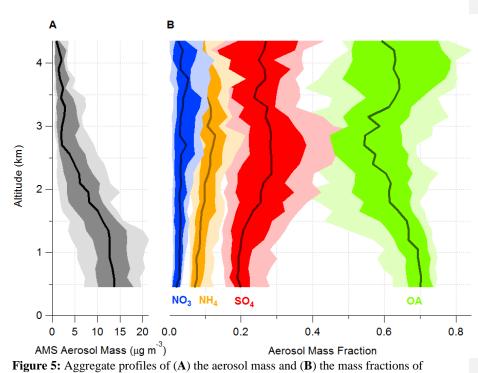


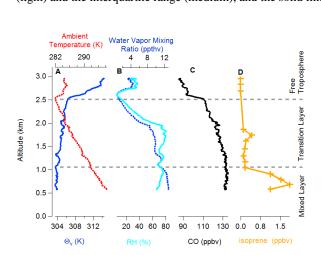


Figure 4: Comparison of airborne and surface measurements of the dry aerosol

- extinction coefficient. The airborne measurements are aboard the NOAA WP-3 aircraft.
- The surface measurements are from the SEARCH monitoring sites.



nitrate, ammonium, sulfate, and OA. The shaded regions show the interdecile range (light) and the interquartile range (medium), and the solid lines are the median (dark).



- 1 Figure 6: An example profile collected over central Georgia in the afternoon of 16 June
- showing (**A**) the temperature and virtual potential temperature, (**B**) the relative humidity and the water vapor mixing ratio, (**C**) the mixing ratio of CO that has a long lifetime, and (**D**) the mixing ratio of isoprene that has a short lifetime.

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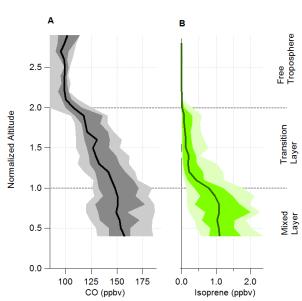
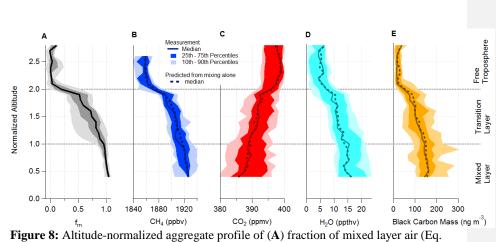
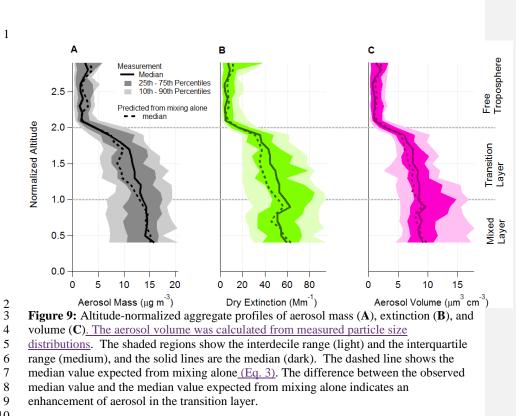


Figure 7: Altitude-normalized aggregate profiles of (**A**) CO, and (**B**) isoprene mixing ratios. The shaded regions show the interdecile range (light) and the interquartile range (medium), and the solid lines are the median (dark).



 f_m CH_4 (ppbv) CO_2 (ppmv) H_2O (ppthv)Black Carbon Mass2Figure 8: Altitude-normalized aggregate profile of (A) fraction of mixed layer air (Eq.34), (B) CH4, (C), CO2, (D) H2O, and (E) black carbon aerosol mass. The dashed line4shows the concentrations expected from vertical mixing alone (Eq. 3). The shaded5regions show the interdecile range (light) and the interquartile range (medium), and the6solid lines are the median (dark). These trace gases and black carbon aerosol mass are7not expected to be enhanced or reduced in the transition layer. The agreement between8the observations and the concentration expected from vertical mixing alone demonstrates9that CO can be used to quantify the fraction air from the mixed layer.



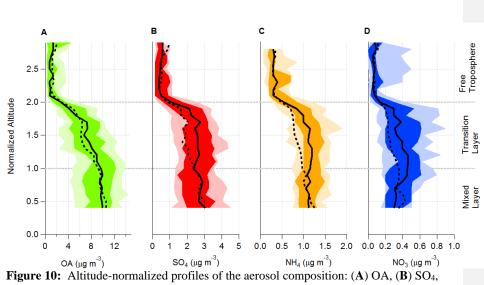
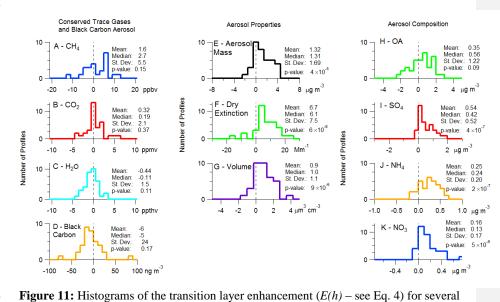


Figure 10: Altitude-normalized profiles of the aerosol composition: (**A**) OA, (**B**) SO₄, (**C**) NH₄, and (**D**) NO₃. The shaded regions show the interdecile range (light) and the interquartile range (medium), and the solid lines are the median (dark). The dashed line shows the median expected concentration from vertical mixing alone (Eq. 3).





trace gases and aerosol properties. The first column shows conserved species and black carbon: (A) CH_4 , (B) CO_2 , (C) H_2O , and (D) black carbon mass. The second column

shows the aerosol extensive properties: (E) aerosol mass, (F) dry extinction, and (G)

8 aerosol volume. The third column shows the aerosol composition: (H) OA, (I) SO_4 , (J)

9 NH4, and (K) NO3. The Student's T-test and resulting p-value (noted in each histogram)

10 were used to test if the mean of each distribution was statistically different from zero.

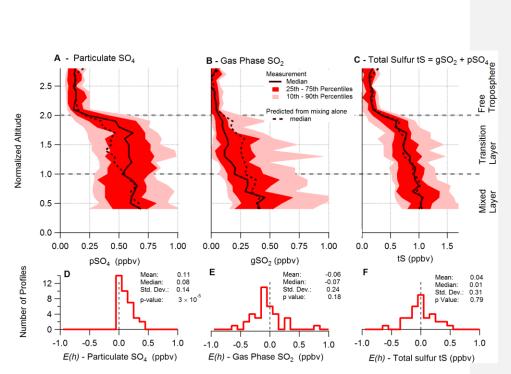
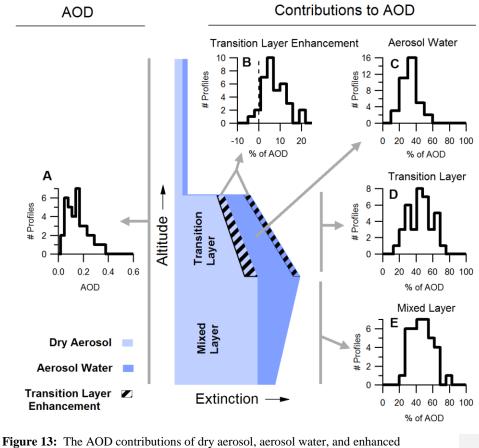
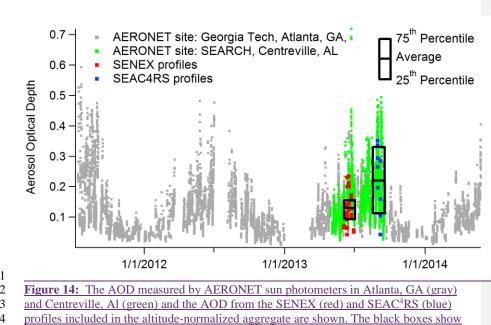


Figure 12: Altitude-normalized profiles of (**A**) particulate sulfate, (**B**) gas phase SO₂, and (**C**) total sulfur tS. The shaded regions show the interdecile range (light) and the interquartile range (medium), and the solid lines are the median (dark). The dashed line shows the median value expected from mixing alone. Histograms of the transition layer enhancement (E(h)) and the results of the T-test for (**E**) particulate sulfate, (**F**) SO₂, and (**G**) total sulfur are shown.



 $\frac{1}{2}$ 3 extinction in the transition layer are illustrated in an idealized profile. The idealized 4 profile of extinction (blue) at the center of the figure shows the vertical location of each 5 contribution to AOD. The light blue area represents the extinction of dry aerosol, and the 6 darker blue area shows the enhancement to aerosol water. The subpanels show (A) 7 histograms of AOD calculated from individual profiles (solid) and AOD measured with a 8 sun photometer (dashed) at the Centreville SEARCH site, and the contributions to AOD 9 from the (B) transition layer enhancement of extinction, (C) aerosol water, (D) the 10 transition layer, and (E) the mixed layer. The calculated AOD assumes no contribution 11 from aerosol above the top of the profile and extrapolates the dry extinction and RH to 12 the surface. 13



the average, 25th, and 75th percentiles of AOD from both the SENEX and SEAC⁴RS profiles.