

1 **In situ vertical profiles of aerosol extinction, mass, and**
2 **composition over the southeast United States during SENEX**
3 **and SEAC⁴RS: Observations of a modest aerosol enhancement**
4 **aloft**

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30
31 **Abstract:**

32

1 Vertical profiles of submicron aerosol from in situ aircraft-based measurements
2 were used to construct aggregate profiles of chemical, microphysical, and optical
3 properties. These vertical profiles were collected over the southeastern United States
4 (SEUS) during the summer of 2013 as part of two separate field studies: the Southeast
5 Nexus (SENEX) study and the Study of Emissions and Atmospheric Composition,
6 Clouds, and Climate Coupling by Regional Surveys (SEAC⁴RS). Shallow cumulus
7 convection was observed during many profiles. These conditions enhance vertical
8 transport of trace gases and aerosol and create a cloudy transition layer on top of the sub-
9 cloud mixed layer. The trace gas and aerosol concentrations in the transition layer were
10 modeled as a mixture with contributions from the mixed layer below and the free
11 troposphere above. The amount of vertical mixing, or entrainment of air from the free
12 troposphere, was quantified using the observed mixing ratio of carbon monoxide (CO).
13 Although the median aerosol mass, extinction, and volume decreased with altitude in the
14 transition layer, they were ~10% larger than expected from vertical mixing alone. This
15 enhancement was likely due to secondary aerosol formation in the transition layer.
16 Although the transition layer enhancements of the particulate sulfate and organic aerosol
17 (OA) were both similar in magnitude, only the enhancement of sulfate was statistically
18 significant. The column integrated extinction, or aerosol optical depth (AOD), was
19 calculated for each individual profile, and the transition layer enhancement of extinction
20 typically contributed less than 10% to the total AOD. Our measurements and analysis
21 were motivated by two recent studies that have hypothesized an enhanced layer of
22 secondary aerosol aloft to explain the summertime enhancement of AOD (2-3 times
23 greater than winter) over the southeastern United States. The first study attributes the
24 layer aloft to secondary organic aerosol (SOA) while the second study speculates that the
25 layer aloft could be SOA or secondary particulate sulfate. In contrast to these hypotheses,
26 the modest enhancement we observed in the transition layer was not dominated by OA
27 and was not a large fraction of the summertime AOD.

28

29 **1. Introduction**

30 Shallow cumulus convection is common over the southeastern United States
31 (SEUS) during the summer. It enhances the vertical transport of trace gases and aerosol,

1 and creates a transition layer between the mixed layer and free troposphere [*Siebesma,*
2 1998]. Due to the presence of clouds and entrainment in the transition layer, it has also
3 been referred to as the cloud layer and the entrainment zone. The transition layer is
4 intermittently mixed by thermal plumes that originate in the mixed layer and form
5 cumulus clouds that release latent heat within the layer. There have been several
6 observations of vertical transport and redistribution of trace gases by shallow cumulus
7 convection [*Angevine, 2005; Ching and Alkezweeny, 1986; Ching et al., 1988; Greenhut,*
8 1986], and a few studies have investigated the vertical transport and aerosol formation
9 during cumulus convection [*Ching et al., 1988; Sorooshian et al., 2007; Sorooshian et*
10 *al., 2006; Wonaschuetz et al., 2012*].

11 Based on the seasonality of the surface aerosol-AOD relationship in the SEUS
12 and the spatial similarity of biogenic emissions and enhanced AOD, Goldstein et al.
13 [2009] and Ford and Heald [2013] have hypothesized the existence of a layer of enhanced
14 secondary aerosol aloft in the summer which contributes to AOD but not to surface
15 measurements of aerosol mass. Goldstein et al. hypothesize that the aerosol layer is
16 primarily SOA, while Ford and Heald speculate that the layer aloft could be either SOA
17 or particulate sulfate. Although neither study speculates about meteorological
18 mechanisms that would lead to the formation of this layer, aerosol production in the
19 transition layer of shallow cumulus convection is a plausible mechanism that could
20 produce the hypothesized layer. More generally, the vertical distribution of aerosol and
21 aerosol formation are integral to understand the relationship between aerosol mass
22 ($PM_{2.5}$) at the surface and AOD [*Hoff and Christopher, 2009*].

23 Submicrometer aerosol particles, which dominate aerosol mass, are largely
24 secondary and composed of OA and sulfates during the summer in the SEUS [*Edgerton*
25 *et al., 2005; Weber et al., 2007*]. While the formation mechanisms of secondary
26 particulate sulfate are well understood (e. g. Seinfeld and Pandis [1998]), the formation
27 of SOA is more complex and uncertain. Both biogenic and anthropogenic precursor
28 emissions are thought to be important [*de Gouw and Jimenez, 2009*]. The relative
29 importance of the homogenous and aqueous oxidation pathways for both sulfate and OA
30 is also uncertain [*Carlton and Turpin, 2013; Carlton et al., 2008; Eatough et al., 1994;*
31 *Ervens et al., 2011; Luria and Sievering, 1991; McKeen et al., 2007*]. Based on the

1 abundance of aerosol water and cumulus convection, aqueous processing is expected to
2 be an important aerosol formation pathway in the SEUS [He *et al.*, 2013], and processing
3 in cloud droplets would occur primarily in the transition layer.

4 In this analysis, aircraft-based in situ measurements of aerosol chemical, physical,
5 and optical properties are used to examine the vertical structure of aerosol in the SEUS
6 during shallow cumulus convection and to quantify aerosol enhancements in the
7 transition layer and its contribution to summertime AOD. We use measurements made
8 aboard the National Oceanic and Atmospheric Administration (NOAA) WP-3D aircraft
9 during the SENEX study in June and July of 2013 and the National Aeronautic and Space
10 Administration (NASA) DC-8 aircraft during SEAC⁴RS in August and September of
11 2013 to construct aggregate vertical profiles of aerosol extinction, mass, and composition
12 as a function of altitude over the SEUS. The transition layer aerosol and trace gas
13 concentrations are modeled as a mixture with contributions from the free troposphere and
14 mixed layer. The in situ measurements of the extinction coefficient are used to calculate
15 the AOD and contributions to the AOD from aerosol water, from the mixed layer, and
16 from the transition layer.

17 18 **2. Methods and Measurements**

19 In this analysis we combine data collected during two aircraft field studies that were
20 partially conducted over the SEUS in the spring and summer of 2013. Although the
21 SENEX study collected measurements in late spring and early summer while SEAC⁴RS
22 collected measurements in the late summer, both studies encountered shallow cumulus
23 convection. Additionally, both aircraft hosted a similar set of in situ instruments which
24 was conducive to a combined analysis.

25 26 **2.1 SENEX**

27 The SENEX study was part of the larger Southeast Atmosphere Study (SAS) in the
28 SEUS during the June and July of 2013. The NOAA WP-3D aircraft flew 18 research
29 flights based out of Smyrna, Tennessee during June and July 2013 for SENEX with a
30 payload of instruments measuring atmospheric trace gases, aerosol properties, and
31 meteorological parameters. This analysis uses measurements of carbon monoxide (CO),

1 carbon dioxide (CO₂), methane (CH₄), sulfur dioxide (SO₂), isoprene, aerosol
 2 composition, and aerosol optical properties (Table 1). The aerosol was sampled
 3 downstream of a low turbulence inlet [Wilson *et al.*, 2004] and an impactor with a 1 μm
 4 aerodynamic diameter size cut. Before impaction, the sampled aerosol was initially dried
 5 by ram heating when sampling into the aircraft. The sampled aerosol was then dried
 6 further in each instrument.

7 The non-refractory aerosol composition was measured by a compact time-of-flight
 8 aerosol mass spectrometer (AMS) downstream of a pressure controlled inlet [Bahreini *et al.*
 9 *et al.*, 2008] and most (97%) of the submicron volume measured by the aerosol sizing
 10 instruments was transmitted into the AMS during SENEX. The collection efficiency for
 11 the AMS was determined by the composition for each data point using the algorithm
 12 described by Middlebrook *et al.* [2012]. When comparing the volume derived from
 13 composition (AMS plus black carbon mass) to the volume measured by the aerosol sizing
 14 instruments in the manner outlined by Bahreini *et al.* [2009], 87% of the aerosol
 15 composition and sizing data from the entire SENEX study are within the combined
 16 uncertainties.

17
 18 **Table 1:** Measurements aboard the NOAA WP-3D used in this analysis. The sample
 19 interval corresponds to the rate at which data is reported and is the integration time for
 20 the lower limit of detection.

Measurement	Technique	Sample Interval	Lower Limit of Detection	Accuracy	Reference
Dry Sub-micrometer Aerosol Extinction (532 nm)	Cavity Ringdown Spectrometer	1 s	0.1 Mm ⁻¹	5% (RH < 30%)	[Langridge <i>et al.</i> , 2011]
Humidified Sub-micrometer Aerosol Extinction (532 nm)	Cavity Ringdown Spectrometer	1 s	0.1 Mm ⁻¹	11% (RH = 90%)	[Langridge <i>et al.</i> , 2011]
Sub-micrometer Aerosol Composition	Compact – Time of Flight – Aerosol Mass Spectrometer	10 s	OA < 0.4 μg m ⁻³ SO ₄ < 0.05 μg m ⁻³	38% OA 34% Inorg.	[Drewnick <i>et al.</i> , 2005], [Canagaratna <i>et al.</i> , 2007]
Sub-micrometer Aerosol Volume	Optical particle counter	1 s	0.03 μm ³ cm ⁻³	+26%, -12 %	[Cai <i>et al.</i> , 2008]
CO	Vacuum Ultraviolet fluorescence	1 s	0.5 ppbv	5%	[Holloway <i>et al.</i> , 2000]
Isoprene	Proton Transfer Reaction Mass Spectrometer	14 s	< 32 pptv	20%	[de Gouw and Warneke, 2007]
Dew Point (RH)	Chilled Mirror Hygrometer	1 s	-	0.2°C	-
Black Carbon Mass	Single Particle Soot Photometer	1 s	12 ng m ⁻³	30%	[Schwarz <i>et al.</i> , 2008]
SO ₂	Pulsed Ultraviolet Fluorescence	1 s	250 pptv	20%	[Ryerson <i>et al.</i> , 1998]

CH ₄	Cavity Ringdown Spectrometer	1 s	-	1.2 ppbv	[Peischl et al., 2012]
CO ₂	Cavity Ringdown Spectrometer	1 s	-	0.15 ppmv	[Peischl et al., 2012]

1

2 2.2 SEAC⁴RS

3 SEAC⁴RS consisted of measurements aboard three aircraft based in Houston,
 4 Texas during August and September of 2013. In this analysis, we focus on in situ
 5 measurements from the NASA DC-8 aircraft, which conducted 19 research flights. The
 6 measurements that were collected onboard the DC-8 and used in this analysis are
 7 summarized in Table 2. Unlike the SENEX study, there was no continuous measurement
 8 of methane during SEAC⁴RS. The aerosol extinction and black carbon instruments
 9 flown on the DC-8 during SEAC⁴RS were the same instruments used onboard the NOAA
 10 WP-3D aircraft during SENEX. Measurements of aerosol extinction, volume, black
 11 carbon mass sampled aerosol through a shrouded diffuser inlet described by McNaughton
 12 et al. [2007]. The aerosol extinction was measured downstream of 1 μ m impactor.

13 The non-refractory aerosol composition was measured by a high resolution time-
 14 of-flight AMS similar to the compact time-of-flight AMS used during the SENEX study.
 15 The two instrument differed in the resolution of the mass spectrometer. The higher
 16 resolution AMS used during SEAC⁴RS was operated with a 1 s sample interval. This
 17 AMS was operated similar to Dunlea et al. [2009] and also used a pressure-controlled
 18 inlet [Bahreini et al., 2008]. The composition-dependent formulation of Middlebrook et
 19 al. [2012] was used to estimate the collection efficiency. The AMS sampled aerosol
 20 downstream of a HIMIL inlet.

21 (<http://www.eol.ucar.edu/homes/dcrogers/Instruments/Inlets/>). In both the HIMIL inlet
 22 and the shrouded diffuser inlet, the sampled aerosol was initially dried by ram heating
 23 and then further dried in each instrument.

24

25 **Table 2:** Measurements aboard the NASA DC-8 used in this analysis. The sample
 26 interval corresponds to the rate at which data is reported and is the integration time for
 27 the lower limit of detection.

Measurement	Technique	Sample Interval	Lower Limit of Detection	Accuracy	Reference
Dry Sub-micrometer Aerosol Extinction (532 nm)	Cavity Ringdown Spectrometer	1 s	0.1 Mm ⁻¹	5% (RH < 30%)	[Langridge et al., 2011]

Humidified Sub-micrometer Aerosol Extinction (532 nm)	Cavity Ringdown Spectrometer	1 s	0.1 Mm ⁻¹	11% (RH = 90%)	[Langridge <i>et al.</i> , 2011]
Sub-micrometer Aerosol Composition	High Resolution - Time of Flight – Aerosol Mass Spectrometer	1 s	0.6 μg m ⁻³ OA 0.06 μg m ⁻³ SO ₄ 0.06 μg m ⁻³ NO ₃ 0.01 μg m ⁻³ NH ₄	38% OA 34% Inorg.	[Canagaratna <i>et al.</i> , 2007]
Sub-micrometer Aerosol Volume	Optical particle counter	1 s	0.03 μm ³ cm ⁻³	+26%, -12 %	[Cai <i>et al.</i> , 2008]
CO	Infrared Absorption	1 s	0.5 ppbv	5%	[Sachse <i>et al.</i> , 1987]
Isoprene	Proton Transfer Reaction Mass Spectrometer	14 s	25 pptv	10%	[de Gouw and Warneke, 2007]
Dew Point (RH)	Chilled Mirror Hygrometer	1 s	-	0.2°C	-
Black Carbon Mass	Single Particle Soot Photometer	1 s	12 ng m ⁻³	30%	[Schwarz <i>et al.</i> , 2008]
SO ₂	Chemical Ionization Mass Spectrometer	1 s	9 pptv	15%	[S Kim <i>et al.</i> , 2007]
CO ₂	Infrared Absorption	1 s	-	0.2 ppm	[Vay <i>et al.</i> , 2011]

1

2 2.3 Surface Measurements

3 The Southeastern Aerosol Research and Characterization (SEARCH) Network
4 consists of 8 continuous monitoring ground sites in Georgia and Alabama hosting several
5 gas-phase and aerosol measurements [Edgerton *et al.*, 2005; 2006; Hansen *et al.*, 2003].
6 During SENEX the NOAA WP-3D flew over four of these sites a total of 15 times, and
7 extinction near the surface is calculated using measurements of aerosol scattering
8 (Radiance Research Model M903 nephelometer, Tempe, Arizona, USA) and absorption
9 (Magee Scientific Model AE-16 Aethalometer, Berkeley, California, USA) at each SEARCH
10 site to compare with the extinction measured onboard the NOAA WP-3D aircraft in the
11 mixed layer.

12

13 2.4 Aerosol Water

14 The enhancement of extinction due to condensation of water onto the aerosol is
15 modeled using an empirical parameterization (shown in Eq. 1), hereafter referred to as
16 the kappa parameterization [Brock *et al.*, in preparation]. The hygroscopic growth of
17 particle diameter is described by kappa-Kohler theory [Petters and Kreidenweis, 2007].
18 A particle size distribution and a Mie scattering calculation would be necessary to
19 rigorously extend the kappa-Kohler theory to the hygroscopic enhancement of optical
20 properties. However, Brock *et al.* [in preparation] shows that if atmospheric

1 accumulation mode size distributions typical of the SEUS are used, the functional form of
2 kappa-Kohler theory can be applied directly to the optical extinction (Eq. 1).

$$\sigma_{ext}(RH) = \sigma_{ext}(dry) \times \left(1 + \kappa_{opt} \times \left(\frac{RH}{100 - RH} \right) \right) \quad (1)$$

3
4
5
6 The humidified extinction coefficient $\sigma_{ext}(RH)$ is a function of the dry extinction $\sigma_{ext}(dry)$
7 and the hygroscopicity parameter κ_{opt} . We note that κ_{opt} is based on the measurement of
8 humidified extinction rather than the direct measurement of the diameter growth factor or
9 activation fraction, i. e. humidified tandem differential mobility analyzers (HTDMA) and
10 cloud condensation nuclei counters (CCNc). The aerosol extinction is measured in three
11 separate constant RH channels: in dry conditions (RH less than 30%), medium RH
12 typically 70%, and high RH greater than 80%. The hygroscopicity parameter (κ_{opt}) is
13 determined by fitting the three measurements of extinction to Eq. 1. The resulting
14 κ_{opt} and $\sigma_{ext}(dry)$ are then used to estimate the extinction at ambient RH. Lower values
15 of aerosol hygroscopicity generally correspond to mineral dust, aerosol with high soot
16 fraction, or primary OA such as fresh biomass burning emissions or automotive
17 emissions [Massoli *et al.*, 2009; Quinn *et al.*, 2005]. High hygroscopicity usually
18 corresponds to an oxidized, aged aerosol, large sulfate mass fractions and/or sea salt
19 aerosol.

20 The calculated ambient extinction will differ from the actual ambient extinction in
21 three cases. First, if the hygroscopic growth exhibits hysteresis and the ambient RH is
22 below the deliquesce RH, ambient particles may be on the deliquescing (lower) or
23 efflorescing (upper) branch of the hysteresis curve [Santarpia *et al.*, 2004]. Our
24 extinction measurements cannot distinguish between these two states, because the sample
25 aerosol is first dried. Then the aerosol is humidified to RH greater than 90% in a cooled
26 Nafion humidifier. The temperature of the humidifier (10-15 K below instrument
27 temperature for 70% RH and 1-3 K below for 90% RH) is varied to control amount of
28 water vapor added to the sample and maintain a constant RH in the sample cell. Finally
29 the sample aerosol is reheated to the instrument temperature and measured in the sample
30 cell. (The sample aerosol is exposed to elevated RH for a duration of 4 s before

1 measurement.) Hence, the measured extinction at high RH and the subsequently
2 calculated hygroscopicity parameter κ_{opt} represent the hygroscopic growth of an
3 efflorescing aerosol on the upper branch. If the aerosol undergoes hysteresis, we
4 expected most aerosol in the summertime SEUS to be on the upper branch because the
5 aerosol regularly pass through clouds and are exposed to high RH conditions. The aerosol
6 also rarely experience dry conditions (RH less than 30%). Second, because the kappa
7 parameterization produces an ambient extinction that asymptotically approaches infinity
8 as RH approaches 100%, we used ambient RH to calculate the ambient extinction only
9 when RH was less than 95%, and assumed a constant RH of 95% when ambient RH was
10 greater than or equal to 95%. Therefore, the calculated ambient extinction is a lower
11 limit of the ambient extinction when RH was greater than 95%. Third, in the case that
12 super-micrometer particles (which are not sampled by instruments used in this analysis)
13 make a significant contribution, the ambient extinction is underestimated. This is
14 typically the case during dust events and during in-cloud sampling which were either not
15 observed or excluded from this analysis, respectively.

16

17 **2.5 Aggregation of Vertical Profiles**

18 Individual profiles are affected by horizontal advection which couples spatially
19 inhomogeneous emissions to the vertical profiles. Because of vertical wind shear and
20 spatial variability during slant profiles, the vertical layers in an individual profile are not
21 always directly comparable. The aggregation the individual profiles is used to reduce the
22 influence of this variability and resolve the typical vertical structure and mixing over the
23 SEUS. Individual vertical profiles were selected from the research flights by inspection
24 of the altitude time series and are generally included for all ascents and descents with an
25 altitude difference greater than 1 km. Measurements during level flight legs were not
26 used in this analysis. Although cloud penetration was mostly avoided, aerosol data
27 sampled during cloud penetration events were excluded due to the effects of particle
28 shattering in the inlets. Cloud penetration events were identified using the video from the
29 nose of the aircraft and cloud particle imaging probes mounted near the wingtips.
30 Transects of biomass burning plumes were identified using tracers such as the acetonitrile
31 mixing ratio, were typically during level flight legs, and were not found during any of the

1 profiles used in this analysis. Extensive aerosol parameters (mass, extinction, volume)
2 have been corrected to standard temperature (273.15 K) and pressure (1013 hPa). All
3 calculated quantities such as ambient extinction and transition layer enhancements were
4 determined before aggregation and then were aggregated in the same manner as the
5 observations.

6 For the SENEX campaign, the vertical profiles (which were located primarily over
7 northern Georgia and Alabama) were generally included in the flight plans for three
8 purposes: to characterize the background boundary layer structure before and after urban
9 and power plant plume intercepts, to characterize the vertical structure over surface
10 measurement sites, and as enroute ascents and descents into and out of the region of
11 interest. The SEAC⁴RS profiles that we use were distributed through Mississippi and
12 Alabama and were conducted to characterize inflow and outflow near convective
13 systems, to examine boundary layer chemistry over the SEUS, and as enroute ascents and
14 descents. The individual vertical profiles used here include both spiral and slant ascents
15 and descents and were typically between 5 and 15 min. in duration.

16 In this analysis we construct two types of aggregate profiles, the first of which
17 includes all of the afternoon vertical profiles over the SEUS and is binned according to
18 altitude above ground level ("altitude-binned" aggregate profiles). The second type of
19 aggregate includes only the subset of profiles during which shallow cumulus convection
20 was present and is binned according to a normalized altitude described in Sec. 3.2
21 ("normalized" aggregate profiles). The profiles not included in the normalized
22 aggregates were either collected during deeper convection and/or had a more complicated
23 structure.

24 For the altitude-binned profiles, we have chosen all of the available vertical
25 profiles from both SENEX and SEAC⁴RS over Mississippi, Alabama, and Georgia in the
26 afternoon between 12:00 pm and 6:30 pm Central Daylight Time (CDT) when we expect
27 the boundary layer structure to be well developed and no residual layers left over from
28 the previous day. The aggregate includes 74 profiles of which 41 profiles are from 6
29 research flights during SENEX and 33 profiles from 6 research flights during SEAC⁴RS.
30 The locations of the profiles used in the altitude-binned aggregate are shown in Fig. 1.
31 The data from individual vertical profiles were aggregated into 150 m vertical bins from

1 the surface to 4.5 km based on the altitude above ground level. The vertical bin height of
2 150 m was chosen such that the slower measurements (aerosol mass and isoprene)
3 typically contributed at least one datum to each bin for each individual profile. In each
4 bin with data from 5 or more individual profiles, the median, interquartile range, and
5 interdecile range were calculated. The median and percentiles were used because these
6 statistics are more robust when outliers are present. The number of vertical profiles
7 which contribute to each aggregated altitude bin varies with altitude because of the
8 differences of the starting and ending altitudes of each individual profile (Fig. 2a).
9 During some profiles or some portions of a profile, individual measurements of trace
10 gases and aerosol properties did not report data (e. g. due to zeroing or calibrations).

11 The normalized aggregate profiles were calculated using only those individual
12 profiles obtained during shallow cumulus convection and were altitude-normalized as
13 described below. Shallow cumulus convection is common over the SEUS. Warren et al.
14 [2007] have compiled a global cloud climatology based on surface observations.
15 According to their work, the mean frequency of daytime cumulus clouds is 49% over
16 Alabama during June, July, and August, and the mean cloud coverage when cumulus
17 clouds are present is 35%. The presence of shallow cumulus convection during
18 individual profiles was determined by inspection of visible images from the GOES
19 satellite and the presence of a three layer structure (mixed layer, transition layer, and free
20 troposphere), which is expected during shallow cumulus convection. The second
21 aggregate includes 37 of the 74 SEUS profiles from the first aggregate. The locations of
22 profiles in the second aggregate are show in Fig. 1, and Fig. 2b shows the distributions of
23 the mixed layer and transition layer heights determined from individual profiles which
24 had medians of 1.2 km and 2.2 km respectively. For cumulus convection the height of
25 the planetary boundary layer is defined as the cloud base or the top of the mixed layer;
26 however, we find the term planetary boundary layer confusing in the context of shallow
27 cumulus convection and have avoided using it. Determination of the mixed layer height
28 and transition layer height is described in Sec. 3.2. For the normalized aggregate profiles,
29 there are 25 bins assigned to each layer. Figure 2c shows the number of profiles that
30 contribute to the second aggregate at each normalized altitude. The number of profiles
31 varies with normalized altitude due to variability in the starting and ending altitudes of

1 each profile and because the second aggregate is limited to portions of the individual
2 profiles when aerosol mass, extinction, volume and CO measurements all report data.
3 Limiting data as such facilitates quantitative comparison of the aggregate profiles. In
4 contrast the first aggregate is constructed using all available data.

6 **3. Results**

8 **3.1 Altitude-Binned Aggregate Profile**

9 Altitude-binned aggregate profiles of dry and ambient aerosol extinction show
10 several characteristics of note (Fig. 3). The median 532 nm dry extinction coefficient
11 (Fig. 3a) is approximately independent of altitude below 1.5 km with median value of 50
12 Mm^{-1} , and the interquartile range is 27 Mm^{-1} to 73 Mm^{-1} . The interquartile (25th -75th
13 percentiles) and interdecile (10th – 90th percentiles) range are due largely to variation
14 between individual profiles rather intra-profile point-to-point variation. Above 1.5 km
15 the extinction coefficient decreases with altitude to a median value of 6 Mm^{-1} above 3.0
16 km. The gradual decrease in extinction with altitude from 1.2 to 2.5 km is due to the
17 variation of mixed layer and transition layer heights in the individual profiles. Figure 3b
18 shows the RH increasing with increasing altitude below 1.2 km. Above this level the RH
19 has a slight decreasing trend with altitude and a large interquartile range spanning 30-
20 70%. The relative humidity of the aggregate profile could be biased low, because during
21 SENEX the flight dates were chosen to avoid precipitation and cloud penetration was
22 mostly avoided during flights however this was not the case during SEAC⁴RS.

23 The median hygroscopicity parameter (κ , Fig. 3c) increases from 0.11 at the
24 bottom of the profile to 0.18 at 3 km and is more variable above 3 km. The hygroscopic
25 growth of the aerosol enhances the ambient extinction (Fig. 3d) throughout the profile
26 and significantly between 0.7 and 1.7 km. Below 1.2 km, the ambient extinction
27 coefficient increases with altitude due to increasing RH, and above 1.2 km decreases with
28 altitude due to a combination of decreasing RH and decreasing dry extinction. The
29 hygroscopic growth of aerosol and the subsequent enhancement of extinction aloft could
30 explain some of the enhancement of AOD noted by Goldstein et al. [2009] and Ford and
31 Heald [2013].

1 The minimum altitude of individual aircraft profiles ranged from 300 - 700 m
2 above the surface. We estimate the profile of dry extinction between the surface and the
3 minimum altitude of the profiles by combining aircraft measurements made in the mixed
4 layer in the vicinity of surface monitoring sites using ground data from those sites.
5 During the SENEX study, there were 15 overflights in the mixed layer within 10 km of
6 four SEARCH monitoring sites. The surface aerosol extinction at each SEARCH site
7 was calculated using the aerosol scattering coefficient measured by a nephelometer with a
8 center wavelength of 530 nm and the aerosol absorption coefficients measured by an
9 aethalometer at 880 nm. Because the optical absorption at this wavelength was likely due
10 to black carbon aerosol, we corrected the absorption coefficient to 530 nm using an
11 Ångstrom exponent of 1 which is conventionally used for black carbon [Bergstrom *et al.*,
12 2002; Lack and Langridge, 2013]. Absorption typically accounted for less than 5% of
13 the extinction. The calculated 530 nm surface extinction was not corrected to the 532 nm
14 aircraft extinction because the correction would be less than 1%. The surface and aircraft
15 extinction coefficients are correlated ($R^2 = 0.91$), and the slope of a orthogonal distance
16 regression (ODR) fit to the data indicates that the aircraft data are ~6% lower than the
17 surface measurements (Fig. 4), which is within the combined uncertainty in the
18 measurements. We conclude that the dry extinction is roughly independent of altitude
19 from the surface to the top of the well-mixed layer. Crumeyrolle *et al.* [2014] found
20 similar agreement between surface and aircraft-based boundary layer measurements of
21 ozone in the Baltimore-Washington metropolitan area.

22 The altitude-binned aggregate vertical profile of aerosol mass (Fig. 5a) is similar
23 in shape to the dry extinction profile. The median mass is $13.7 \mu\text{g m}^{-3}$ at the bottom of
24 the profile and decreases to $2.1 \mu\text{g m}^{-3}$ above 3 km. The aerosol mass is the total of all
25 ions measured by the AMS, and these ions are typically classified as SO_4 , NH_4 , NO_3 , and
26 OA. The inorganic ions are typically formed by ionization of simple salts such as
27 ammonium sulfate and ammonium nitrate but may be formed from more complex
28 compounds (i. e. organosulfates, organonitrates, and amines) that produce both inorganic
29 and organic ions when ionized. To indicate this complexity, we have omitted ionic
30 charges from the notation (i. e. SO_4 , NH_4 , NO_3 ,). In this classification scheme, the
31 composition (Fig. 5b) of the submicron aerosol is primarily OA, sulfate, and ammonium.

1 The mass fraction of the inorganic components (NO_3 , NH_4 , and SO_4) increase with
2 altitude up to 3 km, while the OA mass fraction decreases with altitude up to 3 km.
3 Above 3 km, the OA fraction increases; however, at this altitude the median aerosol mass
4 is only $2 \mu\text{g m}^{-3}$. The increase of aerosol hygroscopicity with altitude up to 3 km
5 corresponds with the increasing inorganic fraction of the aerosol. In particular, sulfate is
6 typically more hygroscopic than OA, is 20% of the aerosol mass at the bottom of the
7 profile, and is 28% of the aerosol mass at 3 km.

9 **3.2 Normalized Aggregate Profiles**

10 The heights of the mixed and transition layers varies among individual vertical
11 profiles and this variation obscures the transition layer in the altitude-binned aggregate
12 profile presented in Figs. 3 and 5. For example, the layer structure is clear in a semi-rural
13 profile measured in the vicinity of shallow cumulus convection over central Georgia on
14 the afternoon of 16 June (Fig. 6). The mixed layer is closest to the surface, a transition
15 layer is formed above the mixed layer, and the free troposphere is on top.

16 The layered structure is evident in both the physical parameters such virtual
17 potential temperature (θ_v) and ambient temperature as well as chemical mixing ratios
18 such as carbon monoxide (CO) and isoprene. In the mixed layer, adiabatically conserved
19 parameters such as virtual potential temperature (Fig. 6a) are independent of altitude.
20 However, in the transition the virtual potential temperature increases with altitude until
21 the top of the transition layer is reached where a capping temperature inversion is present.
22 In the mixed layer, the mixing ratio of water vapor is also independent of altitude;
23 however, RH increases with altitude as temperature decreases (Fig. 6b). Relative
24 humidity is high in the transition layer, and video from the nose of the aircraft confirms
25 the presence of clouds in this layer.

26 The transition layer is also evident in the comparison of long-lived trace gases
27 such as CO (Fig. 6c) with short lifetime trace gases such as isoprene (Fig. 6d). Carbon
28 monoxide is directly emitted during combustion, produced by oxidation of hydrocarbons,
29 lost to oxidation by OH, and typically has an atmospheric lifetime of 1 - 4 months which
30 varies seasonally and regionally [*Seinfeld and Pandis, 1998*]. In this profile, the CO
31 mixing ratio is greater than 110 ppbv in both the mixed and transition layers and

1 decreases to less than 100 ppbv in the free troposphere. In the mixed layer the CO
 2 mixing ratio is independent of altitude and decreases modestly with altitude in the
 3 transition layer. Isoprene is a short-lived trace gas that typically has an atmospheric
 4 lifetime less than 2 hr. [Seinfeld and Pandis, 1998] and in the summertime is emitted by
 5 vegetation common in the SEUS. In the mixed layer, the isoprene mixing ratio is greater
 6 than 1 ppbv and variable due to heterogeneous surface emissions (Fig. 6d). The isoprene
 7 mixing ratio in the transition layer is always less than 500 ppbv and typically ~10% of the
 8 mixed layer value. In the free troposphere, the isoprene mixing ratio is below the
 9 detection limit of the measurement.

10 To examine vertical structure in more detail, altitude-normalized aggregate
 11 profiles were calculated. Altitude normalization is commonly done by dividing the
 12 altitude by the height of the mixed layer. However, because of the more complex vertical
 13 structure often encountered during shallow cumulus convection, we have defined a
 14 normalized altitude, h_{norm} , for each profile such that the top of the mixed layer, h_{ML} , is
 15 assigned a normalized altitude of 1, and the top of the transition layer, h_{TL} , is assigned a
 16 normalized altitude of 2:

$$\begin{aligned}
 18 \quad 0 < h < h_{ML} & \quad h_{norm} = h / h_{ML} \\
 19 \quad h_{ML} < h < h_{TL} & \quad h_{norm} = 1 + (h - h_{ML}) / (h_{TL} - h_{ML}) \\
 20 \quad h > h_{TL} & \quad h_{norm} = 1 + h / h_{TL}
 \end{aligned} \tag{2}$$

21
 22 For individual profiles, the mixed layer height was determined by inspection of each
 23 profile as the highest altitude at which the virtual potential temperature (Θ_v) was constant
 24 (typical variation in the mixed layer was less than 0.5 K) and there was a reduction in the
 25 isoprene concentration. The top of the transition layer was defined by a temperature
 26 inversion and a rapid decrease in the CO mixing ratio.

27 The altitude-normalized aggregate profiles of CO (Fig. 7a) and isoprene (Fig. 7b)
 28 mixing ratios demonstrate the contrast between the mixed layer and transition layer.
 29 During shallow cumulus convection, CO is transported out of the mixed layer into the
 30 transition layer due to its longer lifetime relative to isoprene. The modest decrease of CO

1 with altitude in the mixed layer is likely due to the influence of near source emissions in
2 some profiles. In the mixed layer the isoprene profile is variable, and the median is only
3 modestly dependent on altitude with a median mixing ratio of 1 ppbv. However, the
4 median isoprene mixing ratio decreases to ~10% of this value in the transition layer. The
5 isoprene observed above the mixed layer is consistent with large eddy simulations
6 performed by Kim et al. [2012] who found that cumulus clouds can transport some
7 isoprene out of the mixed layer into the cloud layer.

9 **4. Analysis**

11 **4.1 Aerosol enhancements in the transition layer**

12 During shallow cumulus convection, the air in the transition layer is a mixture of
13 air from the mixed layer below and the free troposphere above. The concentrations of
14 trace gases and extensive aerosol parameters $C(h)$ in the transition layer are described in
15 this analysis by a vertical mixing model consisting of three terms (Eq. 3): a contribution
16 from the mixed layer, a contribution from the free troposphere, and any enhancement
17 $E(h)$ relative to concentration expected from the vertical mixing alone, as

$$19 \quad C(h) = C_{ML}f_m(h) + C_{FT}(1 - f_m(h)) + E(h), \quad (3)$$

20
21 where $C(h)$ is the aerosol or trace gas concentration, C_{ML} and C_{FT} are the aerosol or trace
22 gas concentrations in the mixed layer and the free troposphere. Positive enhancements
23 could be due to local production, or direct emissions to the transition layer from buoyant
24 plumes, e. g. large biomass burning sources, and negative enhancements represent losses.
25 The fraction of air from the mixed layer (f_m) present in the transition layer is determined
26 by using the CO mixing ratio as

$$28 \quad f_m(h) = \frac{CO(h) - CO_{FT}}{CO_{ML} - CO_{FT}}, \quad (4)$$

1 for which the enhancement $E(h)$ due to local production and losses is assumed to be
2 zero. For each profile, the mixing ratio of CO in the mixed layer CO_{ML} and the free
3 troposphere CO_{FT} were determined using the mean between normalized altitudes of 0.5
4 and 0.9 for the mixed layer and 2.0 and 2.5 for the free troposphere. To investigate
5 transition layer enhancements of chemical concentrations and aerosol extensive
6 properties, we calculate a concentration expected from vertical mixing alone using Eq. 3
7 and setting $E(h)$ to zero. The concentration expected from vertical mixing alone is
8 calculated for each profile and aggregated in the same manner as the observations. In the
9 Figs. 8-10, the median concentration expected from vertical mixing alone is shown as a
10 dashed line on top of the observed concentrations. The interquartile and interdecile
11 ranges of the value expected from mixing are not shown.

12 CO is produced through the oxidation of VOCs, and this CO production likely
13 accounts for a significant fraction of the CO budget during the summer in the SEUS
14 [Hudman *et al.*, 2008]. If CO production in the transition layer is significant, the fraction
15 of air from the mixed layer (f_m , determined using CO) would be biased high and any
16 transition layer enhancements of other species ($E(h)$, determined using the CO
17 concentration and Eqs. 3 and 4) would be biased low. By comparing the observed
18 concentration and the concentration expected from vertical mixing alone of several long
19 lifetime species, the importance of CO production in the transition layer can be assessed.

20 Fig. 8a shows the altitude normalized profile of the fraction of air from the mixed
21 layer (f_m , calculated using Eq. 4). The median is 1 in the mixed layer, 0 in the free
22 troposphere, and decreases from 1 to 0.6 in the transition layer due to entrainment of air
23 from the free troposphere. Figs. 8b-e show the altitude-normalized aggregate profiles
24 and the median concentration expected from vertical mixing alone (dashed line) of
25 methane (CH₄), carbon dioxide (CO₂), water vapor (H₂O), and black carbon aerosol
26 mass, respectively. Both CH₄ and CO₂ have atmospheric lifetimes greater than a year
27 and no significant production or losses in the transition layer on the time scale of
28 atmospheric mixing. (The aggregate profile in Fig. 8b only includes data from the
29 SENEX study because CH₄ was not measured during SEAC⁴RS.) H₂O and black carbon
30 aerosol are also not produced in the transition layer and are not lost except in the presence

1 of precipitating clouds. Profiles in precipitating clouds are mostly excluded from the
2 aggregate. Based on the agreement between the observed vertical profiles of CH₄, CO₂,
3 H₂O, and black carbon mass (Fig. 8) and their expected concentration from vertical
4 mixing alone, we conclude the CO production in the transition layer is not significant.

5 In contrast, the altitude-normalized profiles of submicrometer aerosol mass (Fig.
6 9a), extinction (Fig. 9b), and volume (Fig. 9c) are greater than the value expected from
7 vertical mixing alone (dashed lines) in the transition layer. This indicates that $E(h)$ is
8 positive for these aerosol properties. These transition layer enhancements are quantified
9 for individual profiles using the difference between the observed value and the value
10 expected from vertical mixing alone. The difference is expressed as a percentage of the
11 observed value and averaged over the transition between normalized altitudes of 1.1 and
12 1.9. The mean transition layer enhancements of aerosol mass, extinction, and volume
13 were +8.6%, +11.3%, and +9.3% respectively. The difference in the enhancements of
14 mass, extinction, and volume may reflect actual changes in the aerosol density and
15 extinction cross-section or could be due to imperfections in the measurements and data
16 aggregation.

17 Altitude-normalized aggregate profiles of aerosol composition are shown in Fig.
18 10. The enhancement of each aerosol component is quantified in the same manner as
19 aerosol mass, extinction, and volume. The observed median is greater than the value
20 expected from vertical mixing alone by +6% for OA mass, +18% for SO₄, +25% for
21 NH₄, and +15% for NO₃. Although enhancement of sulfate is larger than OA as a
22 percentage, the absolute enhancement is a similar magnitude for both SO₄ and OA, ~0.5
23 $\mu\text{g m}^{-3}$.

24 The transition layer enhancements can be further investigated by examining the
25 distribution of enhancements for individual profiles (Fig. 11). For each profile, the
26 enhancement is calculated using the absolute difference between the observed value and
27 that expected from vertical mixing alone. The difference is averaged between normalized
28 altitudes of 1.1 and 1.9. Because the distributions of enhancements range from negative
29 to positive values, the Student's T-test is used to assess if the enhancement distributions
30 are statistical different from zero, or no enhancement. Enhancement distributions with p-
31 values less than 0.05 are considered statistically significant. As expected, Figs. 11a-d

1 shows conserved species that do not have statistically significant enhancements, CH₄,
2 CO₂, H₂O, and black carbon mass. The enhancement distributions of aerosol mass,
3 extinction, and volume (Fig. 11e-g) are all statistically significant. Although both OA
4 and inorganic aerosol components are enhanced in the transition layer, the enhancement
5 distribution of OA is not statistically significant while the enhancement distribution of
6 inorganic components is significant.

7 Secondary aerosol formation in the transition layer is the likely mechanism that
8 would lead to the observed enhancement of aerosol mass, volume, and extinction. The
9 enhancement of aerosol loading is the net result of production and loss in the transition
10 layer; however, profiles of black carbon and total sulfur (see section 4.2) suggest that the
11 aerosol losses are small. Secondary aerosol formation in the transition layer is a
12 combination of aqueous production (both in clouds and in aerosol water), homogenous
13 oxidation followed by condensation on existing particles, and condensation of semi-
14 volatile species such as NH₄NO₃. The presence of clouds within the transition layer
15 suggests a large role for aqueous production; however, our dataset does not allow us to
16 determine the relative importance of each pathway.

17 Biomass burning emits aerosol in buoyant plumes that, if large enough, could
18 contribute to the observed enhancement of aerosol loading in the transition layer and
19 would not be consistent with the simple vertical mixing model used here to describe the
20 transition layer concentrations. Biomass burning is common in the SEUS during the fall,
21 winter, and spring, but is less common during the summer. Zhang et al. [2010] found that
22 biomass burning contributed between 2 and 10% to measurements of PM_{2.5} in the
23 summer of 2007 and significantly more in other seasons. Although, biomass burning
24 plumes from agricultural fires were transected on level flight legs during both SENEX
25 and SEAC⁴RS, none of the profiles used in this analysis included fresh biomass burning
26 plumes. However, the possibility remains that the aggregate profiles are contaminated by
27 aged and diluted biomass burning plumes which have not been identified. To address this
28 we considered biomass burning emission factors of black carbon, sulfate, and sulfur
29 dioxide (SO₂, which is oxidized in the atmosphere to sulfate) reported by Akagi et al.
30 [2011]. The emission factors range from 0.20 g/kg to 0.91 g/kg for black carbon mass
31 and 0.45 g/kg to 0.87 g/kg for the combination of SO₂ and sulfate. Based on these

1 emission factors, we would expect the ratio of the combination of SO₂ and sulfate mass to
2 black carbon mass in biomass burning plumes to range from 0.5 to 4.35. If the observed
3 enhancement of sulfate (~0.5 μg/m³), were due exclusively to biomass burning, we would
4 expect a concomitant enhancement of black carbon (based on the ratio of emission
5 factors for black carbon and sulfate) in the range of 100 to 1000 ng/m³, which is not
6 observed in the profile of black carbon mass (Fig. 8e). Hence, we conclude the
7 enhancement observed in the altitude-normalized aggregate profile is not due to biomass
8 burning.

10 **4.2 Sulfur budget**

11 Further evidence for the transition layer enhancement of particulate sulfate comes
12 from the reduction of the concentration of gas-phase SO₂ in the transition layer.
13 Particulate sulfate is produced through gas-phase and aqueous oxidation of SO₂ [*Seinfeld*
14 *and Pandis*, 1998]. We expect that mixing in the transition layer would conserve total
15 sulfur which we define as the sum of particulate sulfate and gas phase SO₂. While
16 particulate sulfate is enhanced in the transition layer as described in Sect. 4.1, there is
17 also a reduction in the mixing ratio of gas phase SO₂ in the transition layer. Figures 12a-c
18 shows altitude-normalized aggregate profiles and values expected from vertical mixing
19 alone for particulate sulfate, gas phase SO₂, and the total sulfur. Particulate sulfate (Fig
20 12a) is enhanced by approximately the same amount as the reduction of SO₂ (Fig. 12b),
21 ~0.1 ppbv in the transition layer. Consequently, the median value of total sulfur agrees
22 well with the value expected from vertical mixing alone.

23 The enhancement distributions for particulate sulfate, SO₂, and total sulfur are
24 shown in Figs. 12 d-f. While the transition layer enhancement of particulate sulfate is
25 significant with a p-value of 3×10^{-5} (Fig. 12d), the reduction of SO₂ in the transition
26 layer (Fig. 12e) is not. The lack of statistical significance in SO₂ reduction is due to
27 positive outliers in the enhancement distribution. The enhancement distribution of total
28 sulfur indicates a small enhancement that is not statistically significant (Fig. 12f). We
29 note that the conservation of sulfate and SO₂ is only apparent when mixing in the
30 transition layer is taken into account. If biomass burning were the source of the transition

1 layer enhancement of particulate sulfate, we expect that total sulfur would be enhanced a
2 similar magnitude to particulate sulfate in the transition layer.

4 4.3 Aerosol Optical Depth

5 Aerosol optical depth (AOD) is typically measured remotely from space-based
6 satellites [King *et al.*, 1999] and by ground-based sun photometer networks [Holben *et*
7 *al.*, 2001]. These remote measurements of AOD have been complemented by AOD
8 calculated from aircraft-based in situ measurements of extinction which have the ability
9 to quantify contributions to the AOD from individual layers and aerosol water
10 [Crumeyrolle *et al.*, 2014; Esteve *et al.*, 2012]. Calculated from in situ measurements,
11 AOD is the integral of the ambient aerosol extinction coefficient (σ_{ext}), Eq. 5

$$12 \quad AOD = \int_{surface}^{TOA} \sigma_{ext}(z) dz. \quad (5)$$

14
15 where σ_{ext} is a function of altitude z and the integration extends to the top of the
16 atmosphere (TOA). The extinction coefficient in Eq. 5 is at ambient relative humidity,
17 pressure and temperature. Several assumptions are necessary to calculate ambient
18 extinction and subsequently AOD. First, the aircraft profiles of dry extinction, relative
19 humidity, pressure, and temperature must be extrapolated to the surface. Because
20 extinction at the surface and aloft in the mixed layer are correlated (Fig. 4), the dry
21 extinction is extrapolated as a constant to the surface based on the mean extinction
22 measured in the lowest 200 m of each profile. Relative humidity is extrapolated to the
23 surface using the linear trend in the lowest 200 m of each profile if the trend is positive
24 (RH increases with increasing altitude); otherwise, it is extrapolated as a constant based
25 on the mean RH of the lowest 200 m of each profile. Pressure and temperature are both
26 extrapolated using the linear trend in the lowest 200 m of each profile. The second
27 assumption is that the contributions to AOD from aerosol layers above top of the aircraft
28 profile are negligible. For example, smoke from large forest fires in the western US can
29 be lofted high into the troposphere and transported over the SEUS [Peltier *et al.*, 2007].
30 This contribution to AOD cannot be included if the smoke layer were above the

1 maximum altitude of the profile. In this case the AOD calculated from the in situ profiles
2 is a lower limit. Third, we neglect the contribution to the AOD from supermicroeter
3 particles, which we estimate to be less than 10% of the sub-micrometer AOD based on
4 coarse particle size distribution measurements made during both SENEX and SEAC⁴RS.
5 Fourth, because we have restricted calculated aerosol hygroscopic growth to RH values
6 less than 95%, the AOD calculated here is only a lower limit.

7 In addition to the AOD for each profile in the altitude-normalized aggregate, we
8 have also calculated the contributions to AOD from the mixed layer, the transition layer,
9 aerosol water, and the enhancement of aerosol extinction in the transition layer. The
10 median calculated AOD was 0.14 and the interquartile range (IQR) spanned 0.10 to 0.20
11 (Fig. 13a). An idealized profile of extinction during shallow cumulus convection is used
12 to show the contributions to AOD from the transition layer enhancement of extinction
13 (Fig. 13b), aerosol water (Fig. 13c), the transition layer (Fig. 13d), and the mixed layer
14 (Fig 13e). The contribution of the transition layer enhancement of ambient extinction
15 (median: 7%, IQR: 4%-10%) is split between the enhancement of dry extinction and the
16 aerosol water associated with the additional aerosol loading. The contribution of aerosol
17 water to the whole profile (median: 33%, IQR: 24%-38%) is sensitive to the aerosol
18 hygroscopicity parameter and ambient RH encountered. The transition layer contribution
19 (median: 45%, IQR: 33%-55%) was slightly smaller than the mixed layer contribution
20 (median: 48%, IQR: 38%-57%). The mixed layer's slightly greater vertical extent and
21 higher average dry extinction favor a larger contribution to AOD; however, the transition
22 layer also provides a substantial contribution to AOD because of the aerosol water
23 associated with the higher mean RH in the transition layer. The contributions to AOD
24 presented in Fig. 13 have substantial overlap (i. e. aerosol water also contributes to mixed
25 and transition layer AOD contributions); hence, the contributions do not add to unity.

26 The altitude-normalized aggregate profiles used in this analysis are drawn from
27 37 vertical profiles; however, they represent only eight afternoons during the summer of
28 2013. For comparison and context, Fig. 14 shows an extended time series of 532 nm
29 AOD (level 2 data) measured by AERONET sun photometers [Holben *et al.*, 2001] at the
30 Centreville SEARCH site and at the Georgia Tech site in Atlanta, Georgia. The Georgia
31 Tech site is in an urban area and is perhaps biased toward larger AOD from urban

1 emissions, while the Centreville site is rural. The sun photometers only report data
2 during cloud-free conditions. Plotted on top of these data from the sun photometers are
3 the AOD from the profiles used in the altitude-normalized aggregate. These data are
4 grouped into the profiles from the SENEX and SEAC⁴RS studies. Aircraft profiles
5 during the SENEX study did not sample AOD greater than 0.3 while the maximum of
6 AOD observed by the sun photometers was greater than 0.4. Profiles during the
7 SEAC⁴RS study, although limited in number, cover a range of AOD similar to the sun
8 photometers. Because the majority of the profiles in the altitude-normalized aggregate
9 are from the SENEX study, the aggregate may be biased toward cleaner conditions. The
10 range of AOD observed during summers of 2011-2013 at the Georgia Tech site indicate
11 that the summer of 2013 is not an outlier with AOD higher or lower than typical
12 summers. This is consistent with the analysis of Kim et al. [2015] who has compared
13 satellite measurements of AOD during the summers 2006-2013.

14 Several SEUS studies have noted decreases in anthropogenic emissions (sulfur,
15 nitrogen oxides, and volatile organic compounds) in the first decade of the 21st century
16 [Alston et al., 2012; Attwood et al., 2014; Hand et al., 2012; Hand et al., 2013].
17 Concurrently, particulate sulfate, OA, and AOD have also decreased. Alston et al. [2012]
18 have shown that the summertime mean AOD over Georgia reported by the MISR
19 instrument on the Terra spacecraft decreased from ~0.3 in the summer of 2000 to less
20 than 0.2 in the summer of 2009, which is in the range of AOD calculated in this work for
21 the summer 2013.

22

23 **5. Conclusions**

24 Several preceding studies have observed vertical transport of trace gases and
25 aerosol from the mixed layer into the cloud-influenced transition layer during shallow
26 cumulus convection [Angevine, 2005; Ching and Alkezweeny, 1986; Greenhut, 1986;
27 Langford et al., 2010]. Our observations are consistent with this earlier work. In
28 addition to vertical transport and redistribution of aerosol, we observed a modest
29 enhancement of aerosol loading in the transition layer and conclude that secondary
30 aerosol formation in the transition layer is the likely source of the enhancement.

31 Although we cannot distinguish between condensational and aqueous aerosol formation

1 pathways, the presence of clouds and elevated relative humidity in the transition layer
2 suggests a potential role for aqueous reactions. Using measurements of particulate
3 oxalate as a tracer for aqueous processing, Wonaschuetz et al. [2012] and Sorooshian et
4 al. [2007] have also observed evidence for secondary aerosol formation in the transition
5 layer during cumulus convection over Texas and near the coast of California.
6 Wonaschuetz et al. [2012] show no trends in the OA and particulate sulfate mass
7 fractions with altitude in the mixed and transition layers, which could occur if the
8 production was sufficiently small or if the additional aerosol mass in the transition layer
9 were produced with the same ratio of OA and particulate sulfate that was originally
10 present in the mixed layer. In contrast, our measurements show a distinct difference in
11 composition between the mixed and transition layers and imply a similar magnitude of
12 secondary sulfate and OA production in the transition layer, although the production of
13 OA was not statistically significant.

14 Goldstein et al. [2009] and Ford and Heald [2013] hypothesized a layer of
15 aerosol, that would be sufficient to explain a significant fraction of the observed
16 summertime enhancement of AOD (2-3 times greater than winter), and that does not
17 contribute to aerosol mass at the surface. The hypothesis is partially supported by the
18 spatial similarity of summertime biogenic emission and summertime AOD over the
19 SEUS. Although, Alston et al. [2012] found that the spatial similarity depended on the
20 spatial resolution AOD of the data used in the analysis, and their analysis of AOD and
21 surface aerosol mass over Georgia did not fully support the hypothesis. Kim et al. [2015]
22 found that the increase of the planetary boundary layer height during the summer could
23 bring the seasonality of the AOD and surface aerosol mass into agreement without the
24 need for an enhanced aerosol layer aloft.

25 Here, we have examined in situ vertical profiles of aerosol and found the dry
26 aerosol to be well mixed in the lowest layer. Above the lowest layer, the aerosol mass
27 and extinction decreased with increasing altitude above that layer (Figs. 3-5). The
28 hygroscopic growth of aerosol at high RH resulted in a layer of enhanced extinction near
29 the top of the mixed layer. The aerosol water accounted for approximately a third of the
30 AOD which would explain a portion of the summertime AOD enhancement. The
31 hypothesized, large enhancement of secondary aerosol aloft was not apparent in these

1 aggregate profiles. However, after normalizing the altitude to the vertical structure and
2 using the CO concentration to quantify the vertical mixing (Figs. 7-10), we were able to
3 resolve a modest enhancement of aerosol in the transition layer. This layer was not
4 consistent with the hypothesized layer in magnitude, and the observed composition was
5 not consistent with the SOA dominated layer hypothesized by Goldstein et al. (2009).
6 We observed enhancements that were less than 10% of AOD, and sulfate and OA were
7 enhanced by similar magnitude although the OA enhancement was not statistically
8 significant. The seasonality of the enhancement of surface aerosol mass (less than 1.6
9 times greater in summer than winter) compared to the AOD enhancement (2-3 times) was
10 the primary evidence for the hypothesized layer. Given the absence of such a layer, our
11 observations suggest that other factors such as meteorology and transport may influence
12 the seasonality of the relationship of AOD to surface aerosol mass and warrant further
13 investigation.

14

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28

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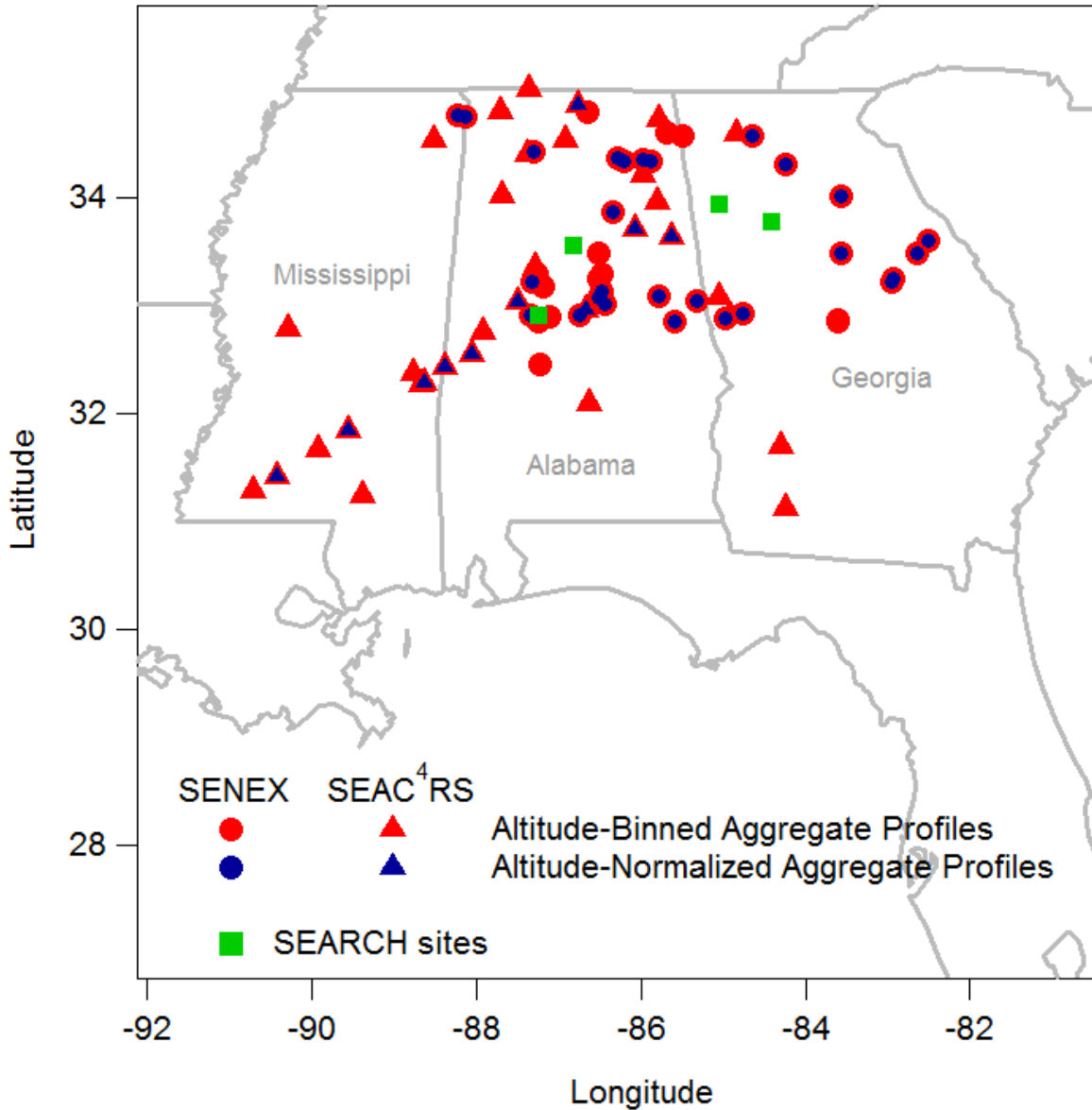
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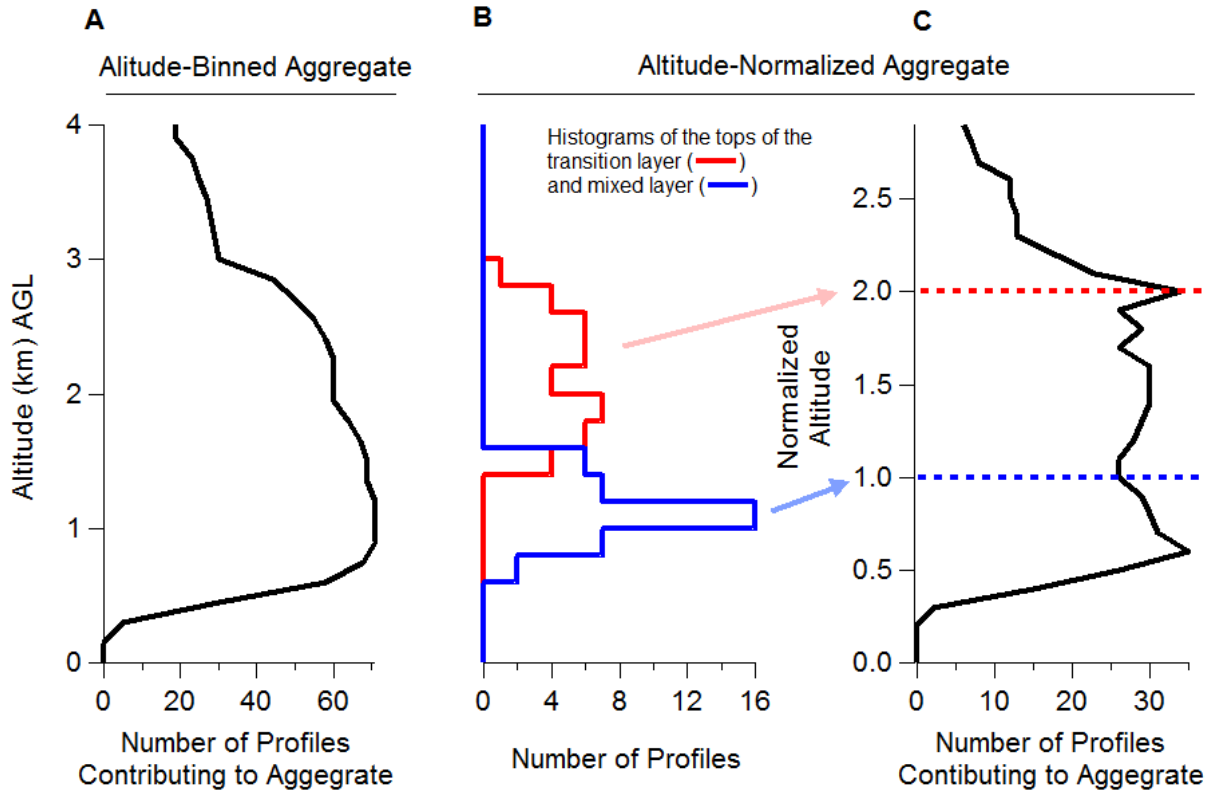
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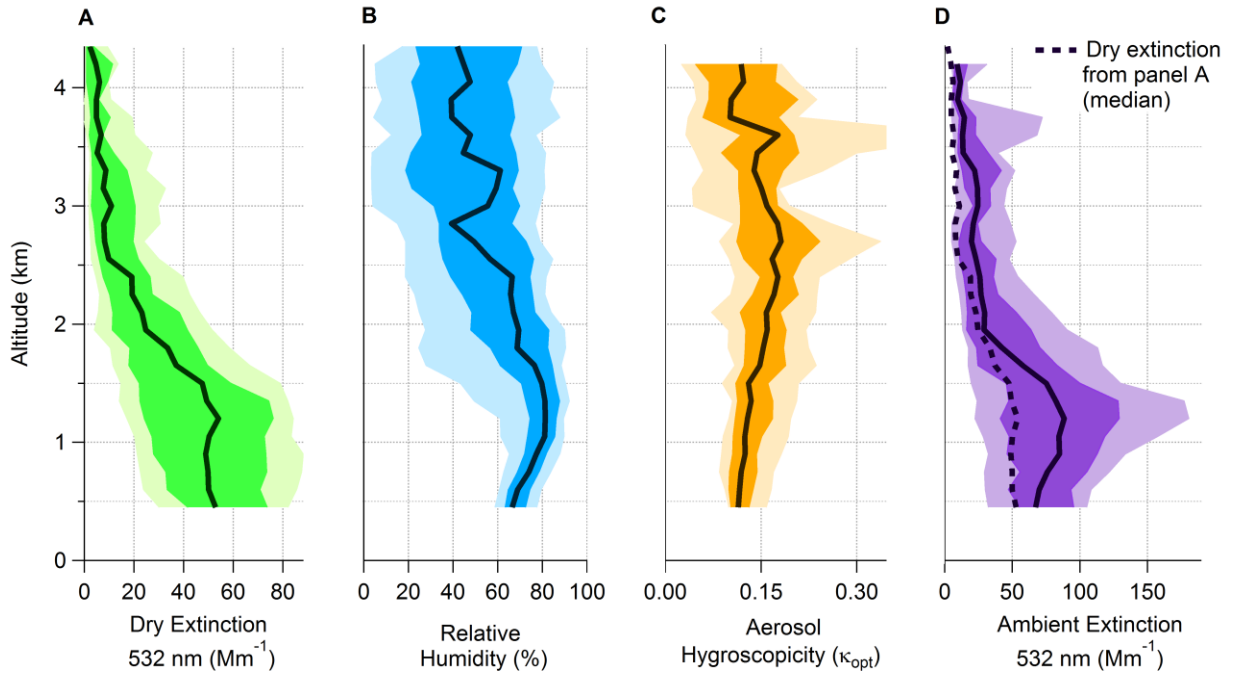


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4 **Figure 1:** The locations of the vertical profiles from the SENEX (circles) and SEAC⁴RS
5 (triangles) and SEARCH monitoring sites (green squares). The markers (both red and
6 blue) are the locations of afternoon profiles used to construct the altitude-binned
7 aggregate profile that includes 74 profiles: 41 from SENEX and 33 from SEAC⁴RS. The
8 blue markers show the location of the profiles used to construct the altitude-normalized
9 aggregate profile that includes 37 profiles: 27 from SENEX and 10 from SEAC⁴RS.
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2 **Figure 2:** (A) The number of profiles that contribute to the altitude-binned aggregate,
3 (B) histograms of the altitude of the tops of the transition and mixed layers, and (C) the
4 number of profiles that contribute to the altitude-normalized aggregate.
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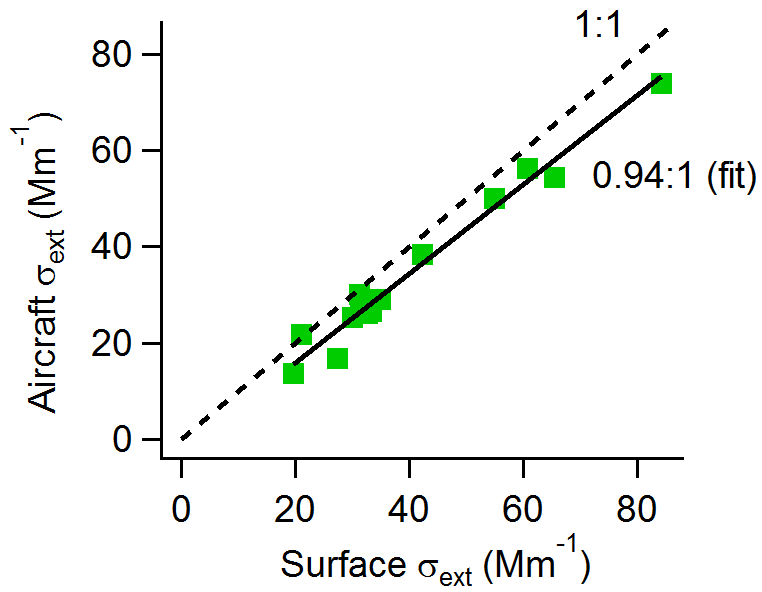
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Figure 3: Altitude-binned aggregate profiles of (A) the 532 nm dry aerosol extinction, (B) relative humidity (calculated from dew point measurements), (C) aerosol hygroscopicity (humidified extinction measurements fit to Eq. 1), and (D) the calculated ambient extinction (Eq. 1). The shaded regions show the interdecile range (light) and the interquartile range (medium), and the solid lines are the median (dark). The dashed line in panel D shows the median dry extinction for comparison.

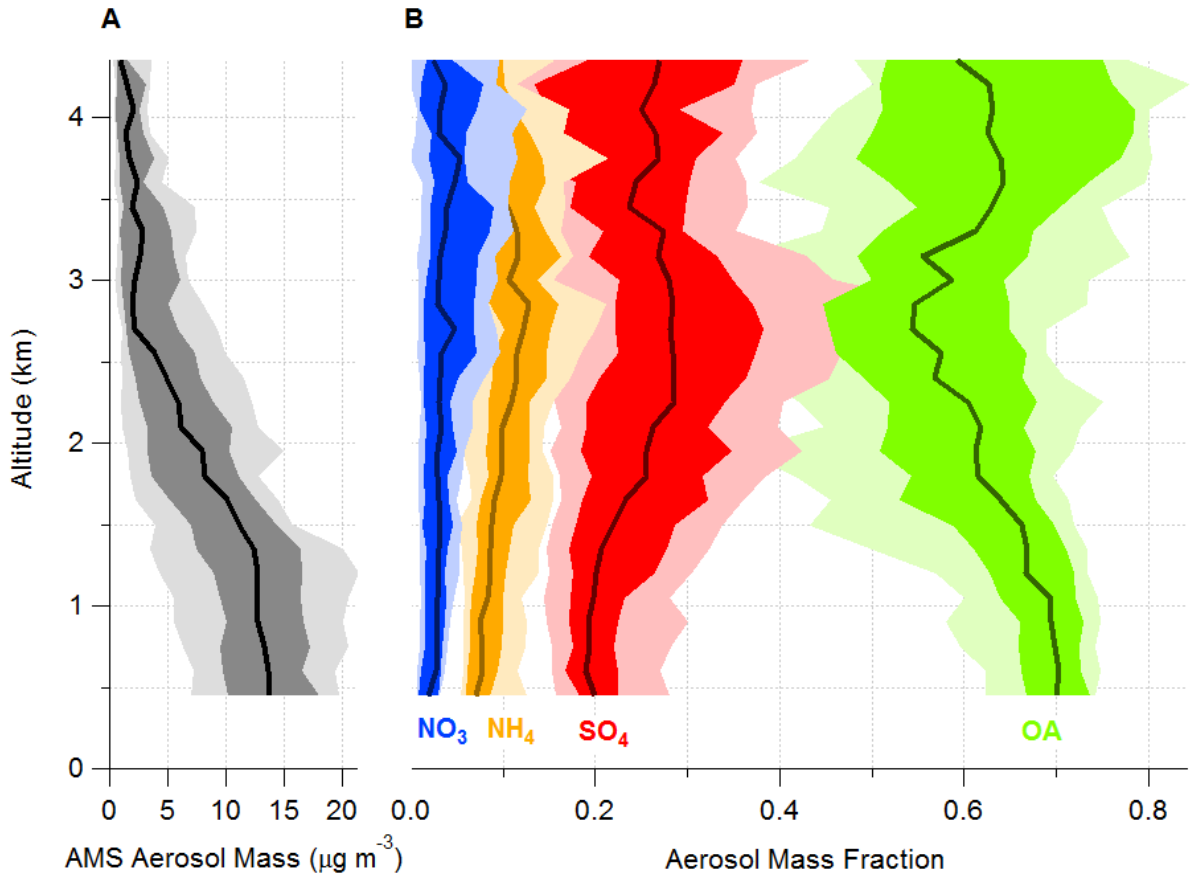
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Figure 4: Comparison of airborne and surface measurements of the dry aerosol extinction coefficient. The airborne measurements are aboard the NOAA WP-3 aircraft. The surface measurements are from the SEARCH monitoring sites.

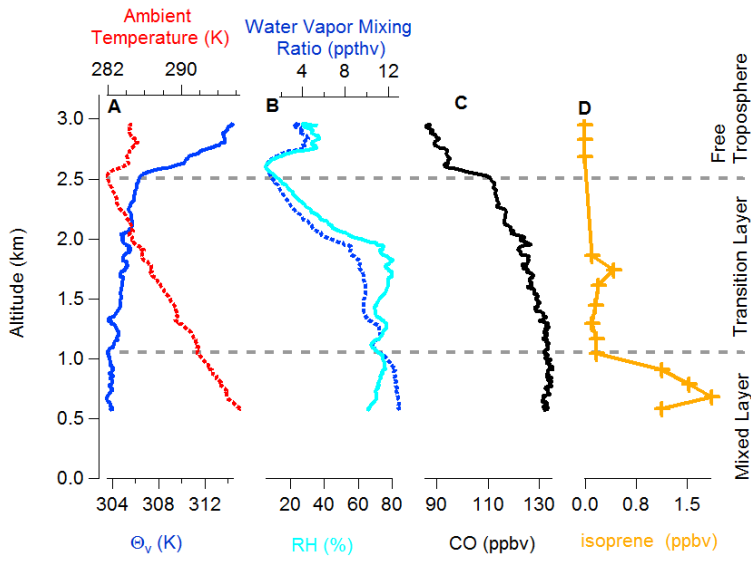
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Figure 5: Aggregate profiles of (A) the aerosol mass and (B) the mass fractions of nitrate, ammonium, sulfate, and OA. The shaded regions show the interdecile range (light) and the interquartile range (medium), and the solid lines are the median (dark).

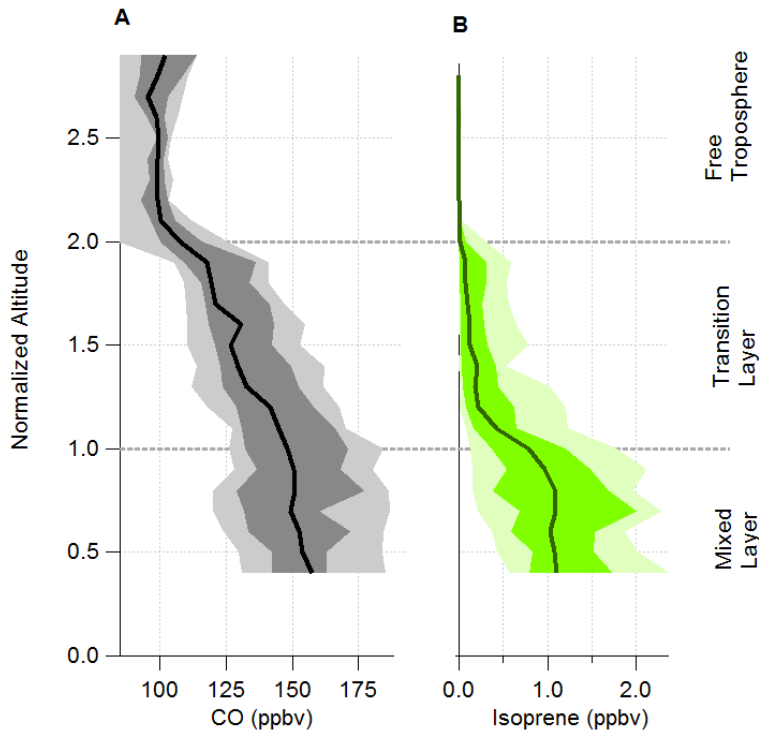
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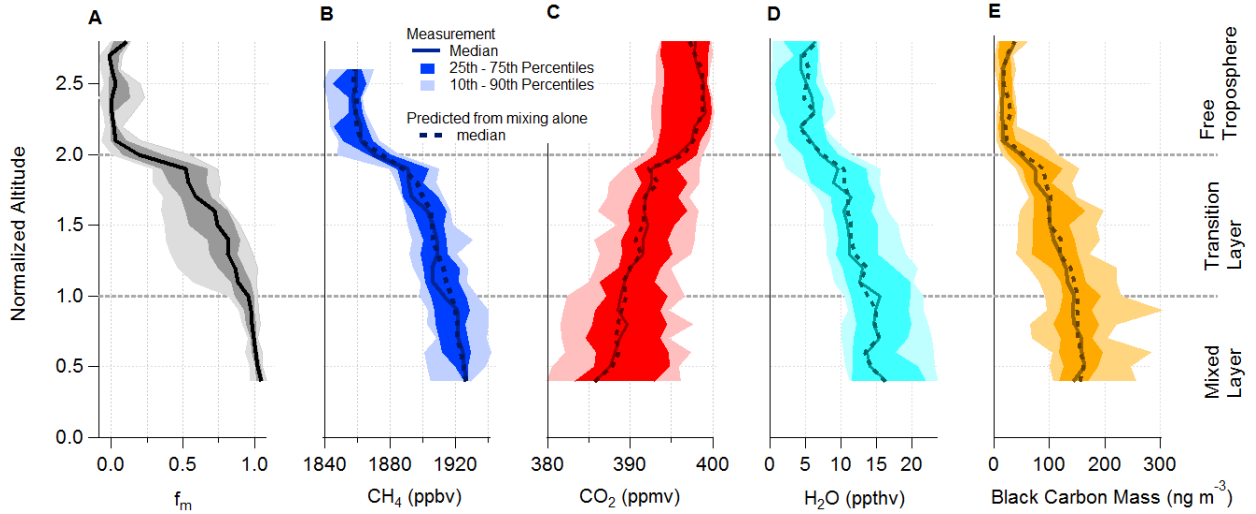
Figure 6: An example profile collected over central Georgia in the afternoon of 16 June showing (A) the temperature and virtual potential temperature, (B) the relative humidity and the water vapor mixing ratio, (C) the mixing ratio of CO that has a long lifetime, and (D) the mixing ratio of isoprene that has a short lifetime.

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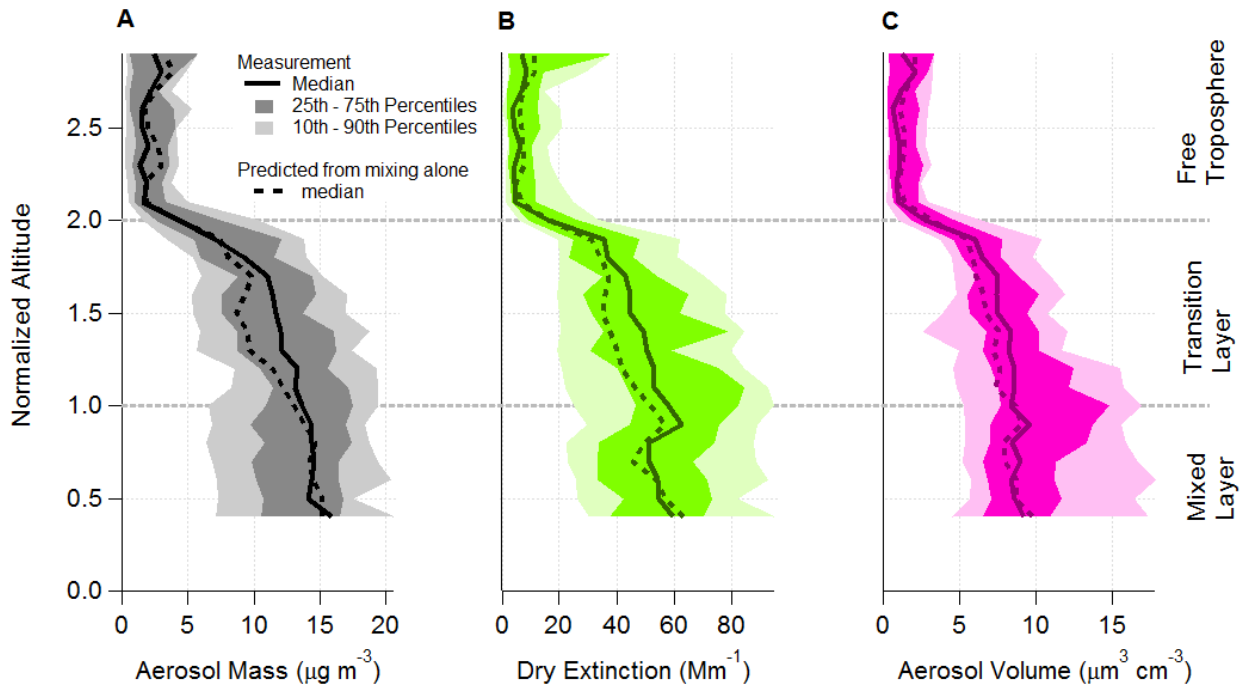
Figure 7: Altitude-normalized aggregate profiles of (A) CO, and (B) isoprene mixing ratios. The shaded regions show the interdecile range (light) and the interquartile range (medium), and the solid lines are the median (dark).



1 **Figure 8:** Altitude-normalized aggregate profile of (A) fraction of mixed layer air (Eq.
 2 4), (B) CH₄, (C), CO₂, (D) H₂O, and (E) black carbon aerosol mass. The dashed line
 3 shows the concentrations expected from vertical mixing alone (Eq. 3). The shaded
 4 regions show the interdecile range (light) and the interquartile range (medium), and the
 5 solid lines are the median (dark). These trace gases and black carbon aerosol mass are
 6 not expected to be enhanced or reduced in the transition layer. The agreement between
 7 the observations and the concentration expected from vertical mixing alone demonstrates
 8 that CO can be used to quantify the fraction air from the mixed layer.
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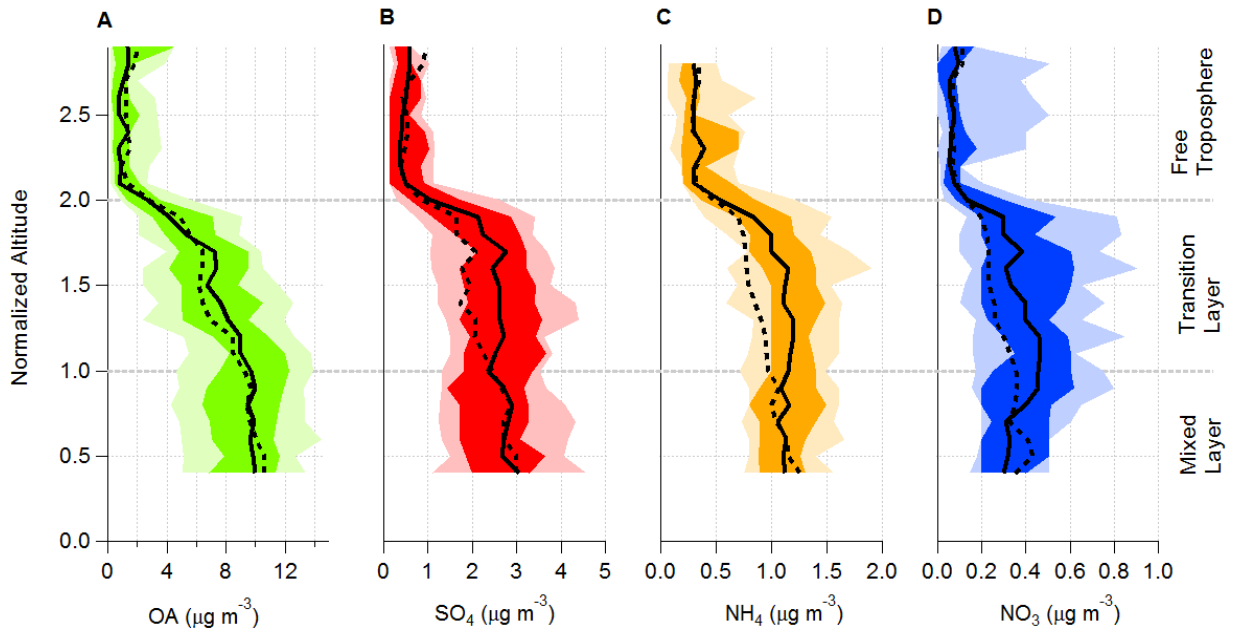
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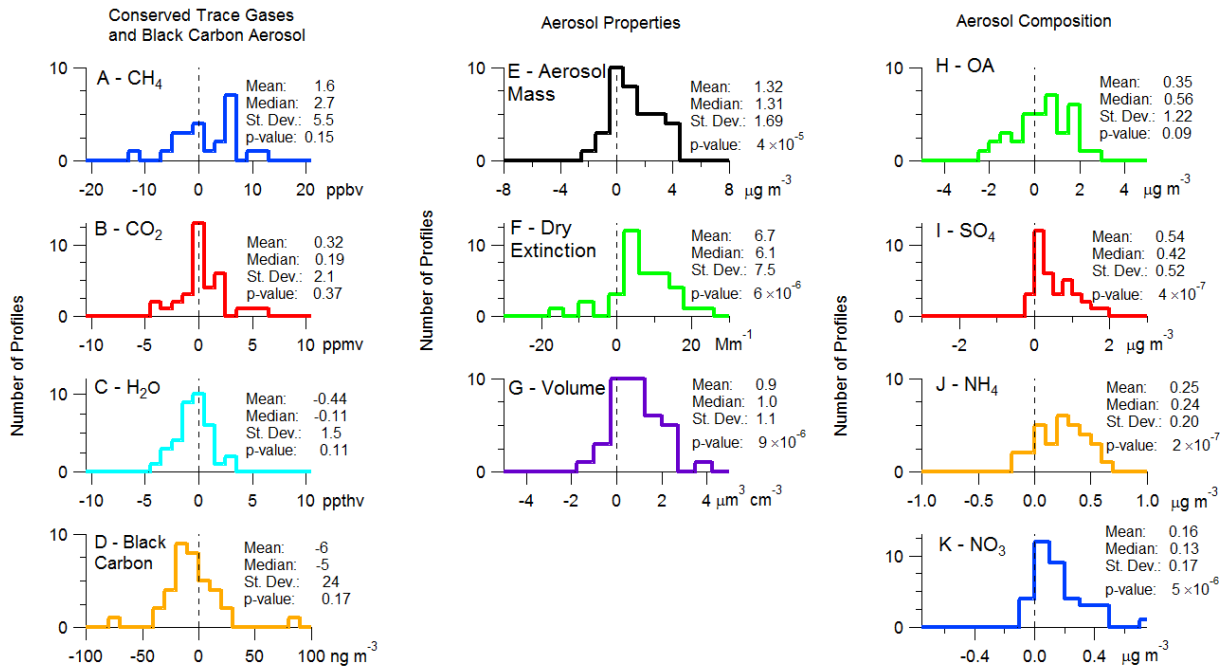
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Figure 9: Altitude-normalized aggregate profiles of aerosol mass (A), extinction (B), and volume (C). The aerosol volume was calculated from measured particle size distributions. The shaded regions show the interdecile range (light) and the interquartile range (medium), and the solid lines are the median (dark). The dashed line shows the median value expected from mixing alone (Eq. 3). The difference between the observed median value and the median value expected from mixing alone indicates an enhancement of aerosol in the transition layer.

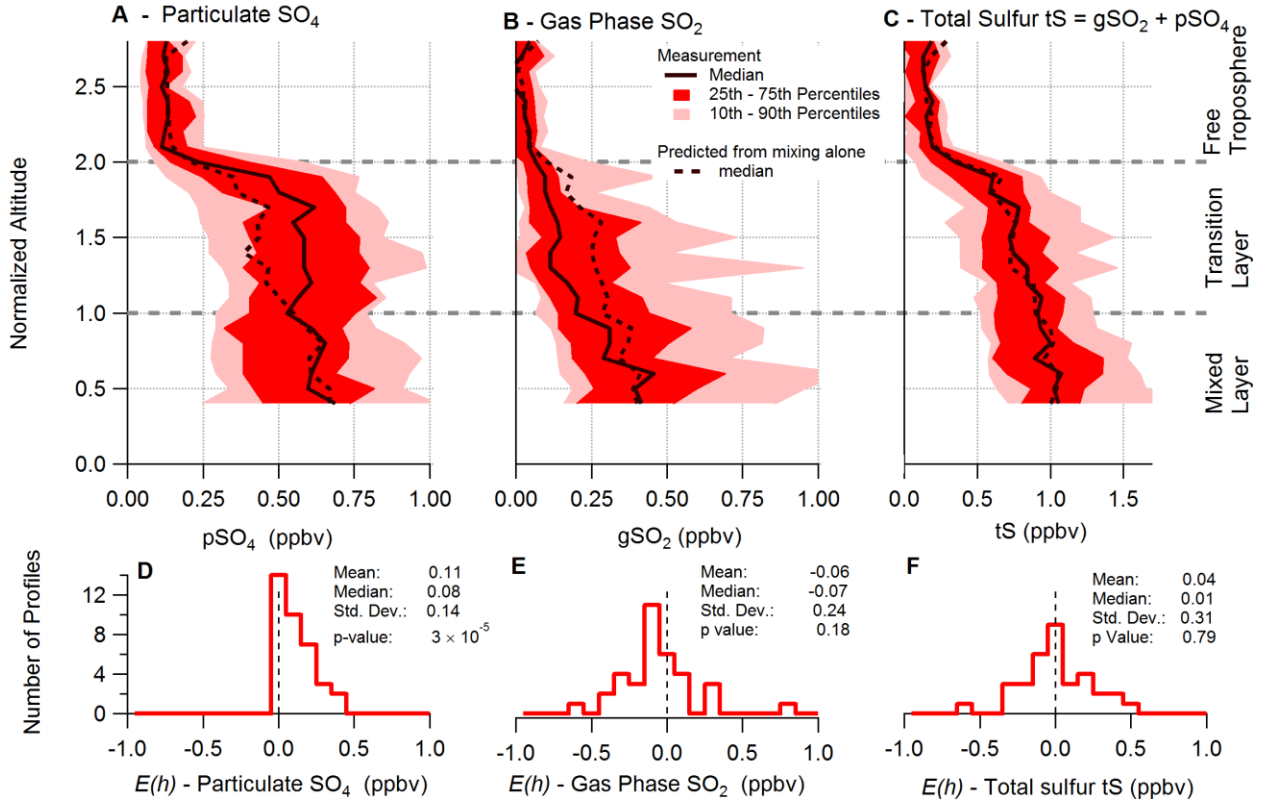


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 2 **Figure 10:** Altitude-normalized profiles of the aerosol composition: (A) OA, (B) SO₄,
 3 (C) NH₄, and (D) NO₃. The shaded regions show the interdecile range (light) and the
 4 interquartile range (medium), and the solid lines are the median (dark). The dashed line
 5 shows the median expected concentration from vertical mixing alone (Eq. 3).
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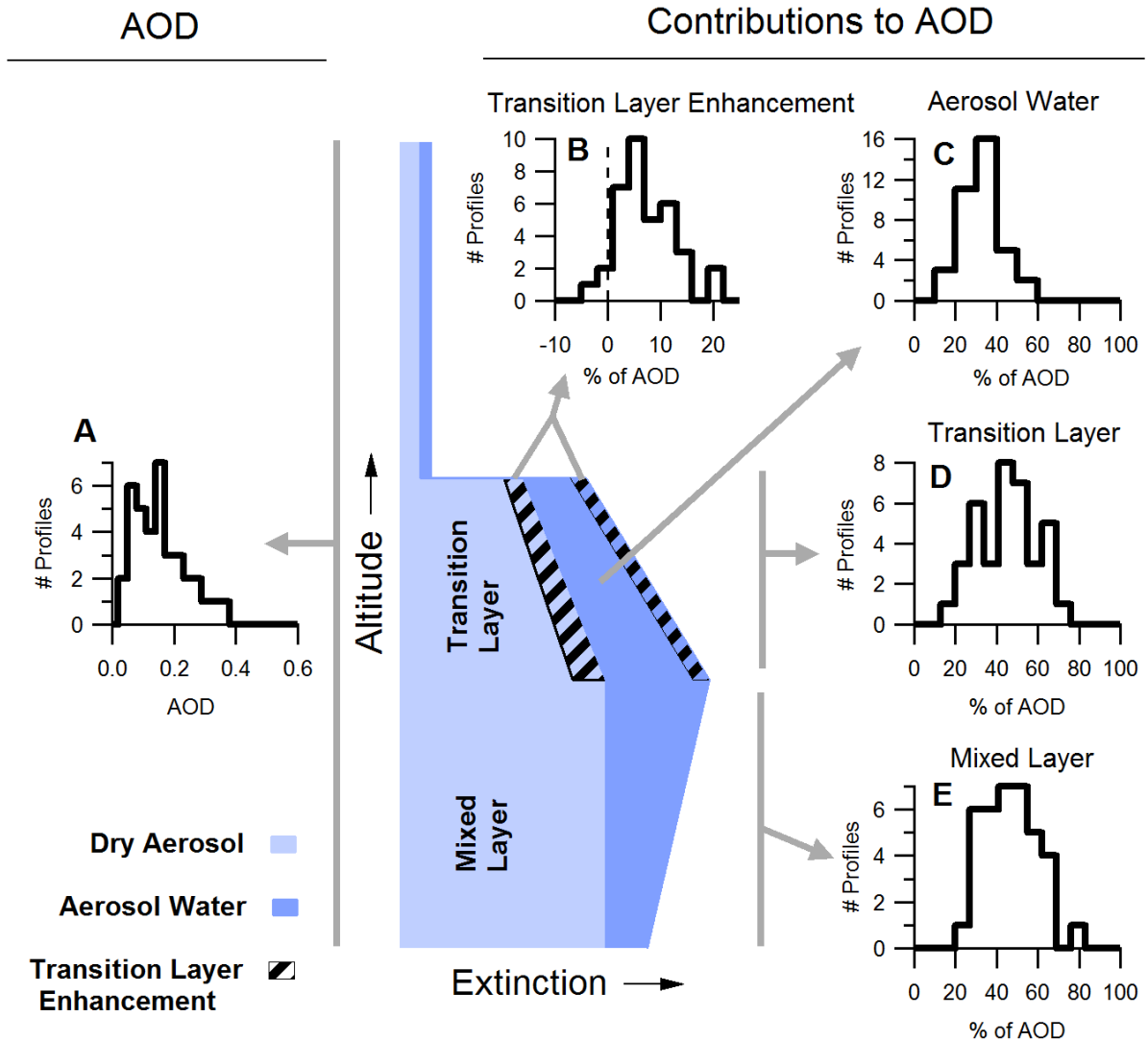


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4 **Figure 11:** Histograms of the transition layer enhancement ($E(h)$ – see Eq. 4) for several
5 trace gases and aerosol properties. The first column shows conserved species and black
6 carbon: (A) CH₄, (B) CO₂, (C) H₂O, and (D) black carbon mass. The second column
7 shows the aerosol extensive properties: (E) aerosol mass, (F) dry extinction, and (G)
8 aerosol volume. The third column shows the aerosol composition: (H) OA, (I) SO₄, (J)
9 NH₄, and (K) NO₃. The Student's T-test and resulting p-value (noted in each histogram)
10 were used to test if the mean of each distribution was statistically different from zero.



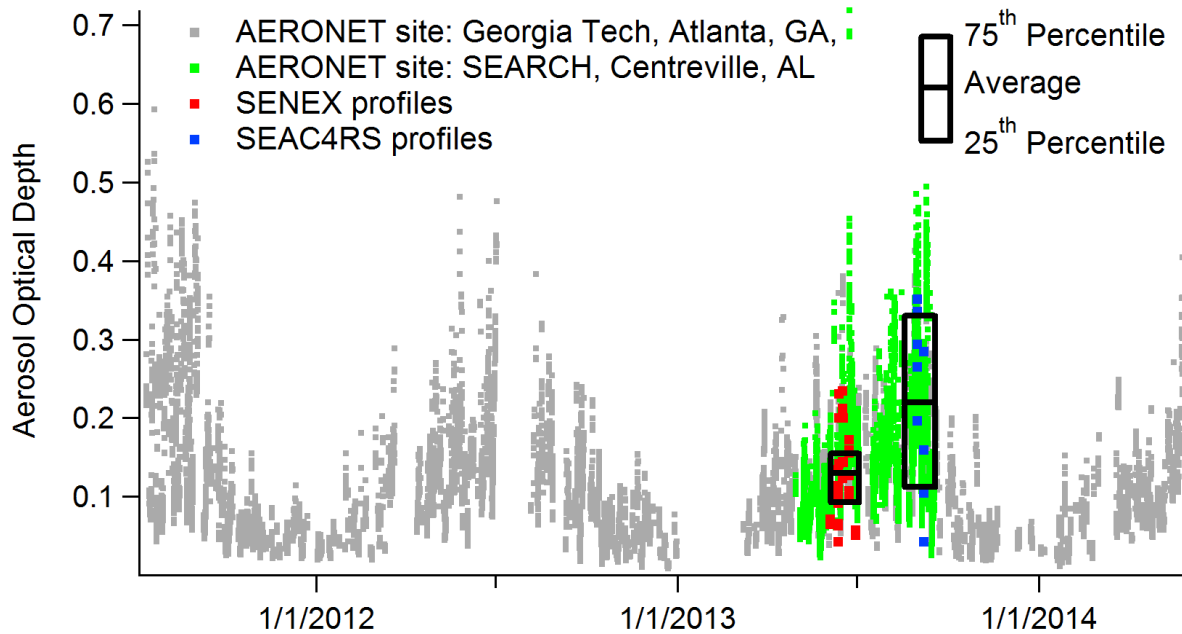
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Figure 12: Altitude-normalized profiles of (A) particulate sulfate, (B) gas phase SO_2 , and (C) total sulfur $t\text{S}$. The shaded regions show the interdecile range (light) and the interquartile range (medium), and the solid lines are the median (dark). The dashed line shows the median value expected from mixing alone. Histograms of the transition layer enhancement ($E(h)$) and the results of the T-test for (D) particulate sulfate, (E) SO_2 , and (F) total sulfur are shown.



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2 **Figure 13:** The AOD contributions of dry aerosol, aerosol water, and enhanced
3 extinction in the transition layer are illustrated in an idealized profile. The idealized
4 profile of extinction (blue) at the center of the figure shows the vertical location of each
5 contribution to AOD. The light blue area represents the extinction of dry aerosol, and the
6 darker blue area shows the enhancement to aerosol water. The subpanels show (A)
7 histograms of AOD calculated from individual profiles (solid), and the contributions to
8 AOD from the (B) transition layer enhancement of extinction, (C) aerosol water, (D)
9 the transition layer, and (E) the mixed layer. The calculated AOD assumes no contribution
10 from aerosol above the top of the profile and extrapolates the dry extinction and RH to
11 the surface.

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Figure 14: The AOD measured by AERONET sun photometers in Atlanta, GA (gray) and Centreville, Al (green) and the AOD from the SENEX (red) and SEAC⁴RS (blue) profiles included in the altitude-normalized aggregate are shown. The black boxes show the average, 25th, and 75th percentiles of AOD from both the SENEX and SEAC⁴RS profiles.