



Supplement of

Atmospheric methane evolution the last 40 years

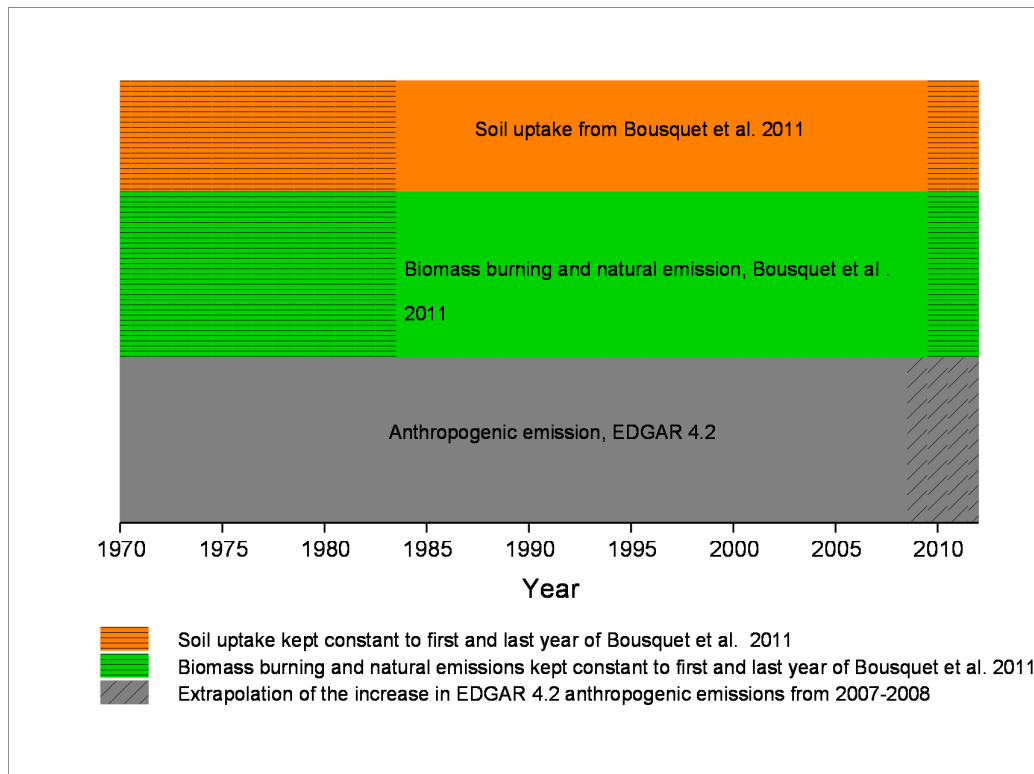
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1 **S1 Setup of emission inventories in model simulations**

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4 **Figure S1.** Overview of how emission inventories are included in the model for different time
5 periods.

6

7 **S2 Emission sectors and tracers**

8 **Table S1.** List of CH₄ emission sectors and emission tracers used in the model simulations.

9 The text colours in column 1 and 2 shows the legend colours in Fig. 1 and Fig. 6-9.

Wetlands	Emissions shown in Fig. 1	Tracer from these sectors shown in Fig. 6-9, 11-13, named "Natural"
Biomass burning	Emissions shown in Fig. 1	
Oceans+Termites+other natural	Emissions shown in Fig. 1	
Enteric fermentation	Emissions shown in Fig. 1	Tracer from this sector shown in Fig. 6-9, 11-13
Agricultural soils	Emissions shown in Fig. 1	Tracer from this sector shown in Fig. 6-9, 11-13
Solid fuels: Fugitive from solids	Emissions shown in Fig. 1	Tracer from this sector shown in Fig. 6-9, 11-13
Gas production and distribution	Emissions shown in Fig. 1	Tracer from this sector shown in Fig. 6-9, 11-13
Sum all other anthropogenic (those listed below)	Emissions shown in Fig. 1	Tracer from this sector shown in Fig. 6-9, 11-13
Solid waste		
Waste water		
Residential		
Manure management		
Energy manufacturing transformation		

Agricultural waste burning		
Road transportation		
Fossil fuel fires		
Non-road transportation		
Oil production and refineries		
Industrial process and product use		

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2 **S3 Theoretical foundation of the use of the fictitious tracers**

3 As indicated in Sect. 2.2 of the main paper, the simulations used 18 passive fictitious tracers
4 for each of the CH₄ emission sectors listed in Table S1. The tracers were continuously emitted
5 and then given an e-folding lifetime of 1 month undergoing transport but not interacting
6 chemically. These tracers were used as a proxy for the different sector's recent contribution to
7 monthly mean surface CH₄ concentrations, with the aim of revealing key sectors and regions
8 behind recent changes in spatial distribution or temporal evolution of CH₄. In this section, we
9 provide the theoretical foundation that justifies the use of these fictitious tracers.

10

11 Firstly, we summarize the results obtained in this section. We split the CH₄ mole fraction into
12 two components: a quite uniform background component (r_B) and an inhomogeneous recently
13 emitted component (r_R); being the fictitious tracer a proxy for the second component. The
14 CH₄ surface emissions act as the sources for r_R (not for r_B), then this component is advected
15 and mixed, and when achieving a good mixing (after 1-2 months) it is converted into r_B . Since
16 the life time of r_R is of around 1 month (much smaller than the mean CH₄ lifetime), the
17 chemical destruction acting on r_R is almost negligible (only acts on r_B). The same reason
18 makes $|r_B| \gg |r_R|$, except very near strong CH₄ emission sources.

19

20 We start with the continuity equation for the CH₄ mole fraction (r) in dry air:

$$21 \quad \frac{Dr}{Dt} = -\frac{1}{n} \nabla \cdot \bar{F}_D(r) - \sum_i k_i \cdot c_i \cdot r \quad (1),$$

22 where D/Dt is the Lagrangian time derivate, n is the number density (mol/m³) of the dry air,
23 the vector F_D is the diffusive flux due to turbulence (i.e., unresolved flow by the spatial scale
24 of the model), k_i is the reaction rate with the trace gas i and c_i is the number density of the
25 trace gas i . Note that the diffusive flux is linear in r (e.g., proportional to the gradient of r),
26 even in the case in which the flux is non local (e.g., see Holtslag & Boville, 1993), as well as
27 the rest of the terms of Eq. (1). The surface sources and sinks enter as the boundary conditions
28 of Equation (1).

1

2 Now, we split Eq. (1) into two equations (this is our definition for the components r_B and r_R ;
3 we do not base our definition in spatial averages):

$$4 \quad \frac{Dr_B}{Dt} = -\frac{1}{n} \nabla \cdot \bar{F}_D(r_B) - \sum_i k_i \cdot c_i \cdot r_B + P(r_R) \quad (2),$$

$$5 \quad \frac{Dr_R}{Dt} = -\frac{1}{n} \nabla \cdot \bar{F}_D(r_R) - \sum_i k_i \cdot c_i \cdot r_R - P(r_R) \quad (3),$$

6 where the chemical losses in Eq. (3) will be almost negligible except very near strong CH₄
7 sources (as explained in the second paragraph of this section), and P is a linear projector onto
8 a complete set of vectors (spectral components) for wavelengths larger than around 6,000 km.
9 This projector continuously removes the smoothed part of r_R , which is continuously created as
10 the emitted CH₄ becomes well mixed, and converts it into r_B . The CH₄ surface emissions are
11 only included as boundary conditions for Eq. (3) (not for Eq. (2)). The CH₄ surface sink is
12 only included as boundary condition for Eq. (2). Note that summing Eqs. (2) and (3), Eq. (1)
13 is obtained. The reason for chosen 6,000 km as threshold wavelength is: 1) The mid-latitude
14 synoptic scale motions have a characteristic variation length, L, of around 1,000 km (e.g., see
15 Holton, 1992), and their associated wavelength is therefore of around 6,000 km. 2) Synoptic
16 scale latitudinal motions are able to build up mole fraction inhomogeneities by advection of
17 the CH₄ climatological latitudinal pattern. 3) What makes more sense is to convert r_R into r_B
18 around the smaller scale in which the background by itself can build up inhomogeneities.
19 Anyway, in the main article we plot and analyse monthly averages, therefore the mole
20 fraction mark due to synoptic motions will be smoothed.

21

22 In the main paper, instead of using the no local projector operator for coupling r_R and r_B , we
23 use a simpler local proxy for this purpose: a volumetric sink for r_R with an e-folding lifetime
24 of 1 month ($A \times r_R$ instead of $P(r_R)$, with $A=1 \text{ month}^{-1}$). This process can approximately
25 mimic the projector behaviour: it transforms r_R into r_B at a rate similar to that in which the
26 projector acts (i.e., the rate at which mixing is able to smooth the emitted CH₄ till the 1,000
27 km characteristic variation length at which the projector starts to act). Indeed, r_R will be
28 underestimated a bit because part of it will be removed by the 1-month lifetime e-folding sink
29 before being smoothed to the 1,000 km characteristic variation length. The time needed to
30 mix a species throughout a hemisphere is about 1 to 2 months, whereas 1 to 2 years are
31 needed to mix a species through the entire Earth troposphere (Seinfeld & Pandis, 1998).

32

1 Now, we introduce the following notation: (Eulerian) annual means are denoted as $\langle \rangle$ (in the
 2 main paper we use annual running means since we are interested in the inter-annual variation
 3 of CH₄), whereas longitudinal means along a whole terrestrial parallel are denoted as $[]$. Each
 4 variable can be decomposed in two components: the mean and the deviation from the average,
 5 for instance:

$$6 \quad r = \langle r \rangle + r' \quad (4)$$

$$7 \quad r = [r] + r^* \quad (5)$$

$$8 \quad \langle r \rangle = [\langle r \rangle] + \langle r \rangle^* \quad (6)$$

9 We have found (see the main paper) that there is a high correlation between $\langle r \rangle^*$ and $\langle r_R \rangle^*$
 10 for most of the stations, with r_R defined using the local 1-month e-folding sink. Writing Eqs.
 11 (1) and (3) in conservative form, and expanding each variable simultaneously into its
 12 longitudinal and time components, in a similar fashion than Sect. 4.1.1 of Peixoto and Oort
 13 (1992), we have obtained quite complex PDEs (Partial differential Equations) linear in $\langle r \rangle^*$
 14 and $\langle r_R \rangle^*$, respectively. Both equations are very similar (it is out of the scope of this paper to
 15 present such equations), and the main differences are:

- 16 a) a term containing the flux $[\langle r \rangle] \langle v \rangle^*$ appears in the first equation, whereas $[\langle r_R \rangle]$
 17 $\langle v \rangle^*$ appears in the second equation (they are no homogeneous terms of the
 18 corresponding PDEs).
- 19 b) a chemical term containing $\langle k_i c_i \rangle^* [\langle r_B \rangle]$ appears in the first equation, whereas the
 20 term $-A_n \langle r_R \rangle^*$ appears in the second equation.

21 We expect these terms are usually small (because we expect $\langle v \rangle^*$ and $\langle k_i c_i \rangle^*$ are usually
 22 small; these are the prerequisites mentioned in the main paper), except the term $-A_n \langle r_R \rangle^*$
 23 that we think it is compensated in the other PDE by a larger mixing (due to differential
 24 advection and turbulent diffusion), which tends to convert $\langle r \rangle^*$ into $[\langle r \rangle^*]$. If both PDEs
 25 were identical, there would be a linear relation between their solutions, and the time linear
 26 correlation coefficient between the solutions would be exactly 1. However, the few small
 27 differences between both PDEs make the time correlation coefficient between the solutions
 28 smaller than one. Note that $\langle r \rangle^*$ changes along the corresponding parallel. For locations with
 29 $|\langle r \rangle^*|$ small (i.e., $\langle r \rangle$ closes to $[\langle r \rangle]$) compared to the maximum and minimum within the
 30 parallel, the relative contribution of the terms different (between the PDEs) may be larger and
 31 therefore the correlation coefficient smaller for these locations (this might explain the case of
 32 the Wendover station). As mentioned in the previous paragraph, r_R will be underestimated a

1 bit when using the 1-month e-folding sink term, and this fact can contribute to the offset
2 between $\langle r \rangle^*$ and $\langle r_{\text{R}} \rangle^*$, and might explain why for most of the stations $|\langle r \rangle^*| > |\langle r_{\text{R}} \rangle^*|$.

4 **S4 Scaling procedure**

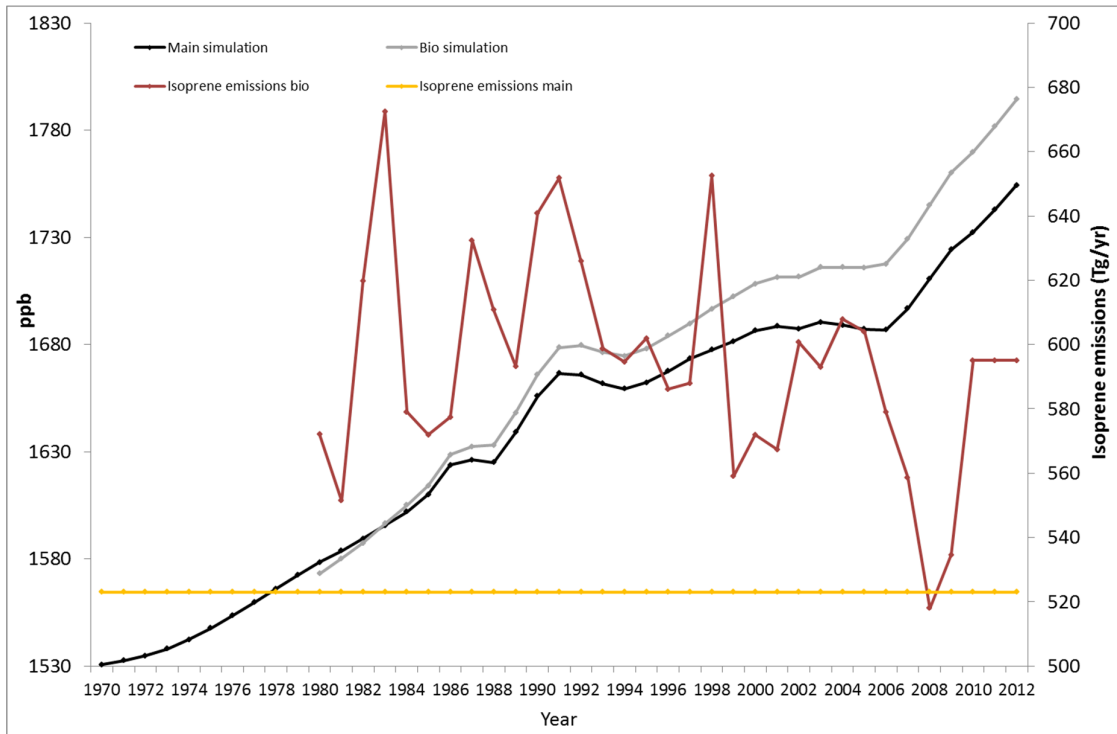
5 As noted in the main text the model in general underestimates the observed surface methane
6 levels and likely reasons are discussed there. In Fig. 6-9 in the main article, the model results
7 are scaled to the observed mean over the periods of measurements to better discern
8 differences in trends between observations and model. To do this the absolute difference
9 between the model output and the measurements is calculated for each year in the period
10 1970-2012. The mean of these differences is then added to the modelled values for all years.
11 The model values are sampled from the gridbox with the closest location to the geographical
12 position of the stations. Likewise, the model layer best corresponding to the station altitude is
13 used.

15 **S5 Discussion of sensitivity studies**

16 **S5.1 Influence of inter-annual variation in emissions from vegetation**

17 In the “main” simulation discussed in the main article, natural emission data for 2000 were
18 used for all years and all components except CH₄. The emissions from vegetation of CO and
19 NMVOCs are from MEGAN (Guenther et al., 2006). Recently a new dataset (Sindelarova et
20 al., 2014) with MEGAN emissions covering the period 1980-2010 became available. This
21 dataset was used in the “bio” simulation to investigate whether inter-annual variations in CO
22 and NMVOCs emissions from vegetation are important for the CH₄ evolution. Variations in
23 these emissions affect OH levels which in turn influence the atmospheric CH₄ loss. Fig. S2
24 shows that surface CH₄ levels are higher in the “bio” simulation. Due to the long response
25 time of CH₄ the difference between the two simulations grows over the first two decades. The
26 higher CH₄ level is expected since the emissions (illustrated by the isoprene emission curves
27 in Fig. S2) in the new inventory are higher for most years compared to the constant year 2000
28 emissions in the old inventory. Larger emissions of components like isoprene and CO results
29 in lower OH values and reduced CH₄ loss. However, accounting for inter-annual variation of
30 vegetation emissions of CO and NMVOCs does not shift the periods of growth and
31 stagnation. Neither does it lead to larger year to year fluctuations in CH₄ levels. Compared to
32 surface measurements (discussed in section 3.2 of the main article) the underestimation of
33 CH₄ levels is less in the “bio” simulation, except from that there is no improvement in model
34 performance.

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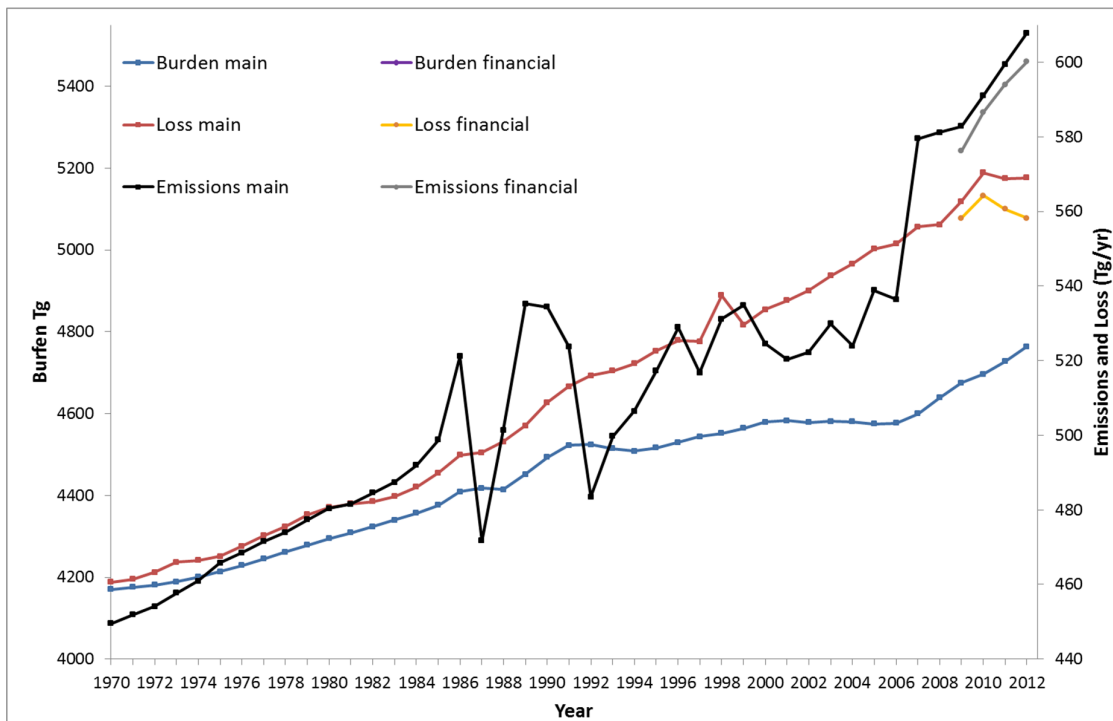
3 **Figure S2.** Surface CH₄ levels and isoprene emissions in main simulation and bio simulation.

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5 **S5.2 Influence of financial crisis.**

6 The period 2009-2012 was rerun with slightly different emissions evaluating whether the
7 recent financial crisis had any significant impact on CH₄ levels. Here, the emissions from
8 petroleum and solid fuel production and distribution were scaled with BP Statistical Review
9 of World Energy (bp.com/statisticalreview) numbers for gas production, oil and coal
10 consumption resulting in a drop in total emissions in 2009 (**Error! Reference source not found.**
11 main article). However, the evolution in emissions from 2010 with this alternative
12 extrapolation is rather similar to that for the standard extrapolation. Due to the drop in
13 emissions in 2009 in the “financial” run methane loss after 2009 is lower than for the “main”
14 simulation (Fig. S3). The emission growth in 2011 and 2012 is also slightly lower in the
15 financial simulation. This results in declining methane loss for these years in the “financial”
16 simulation. In contrast the methane loss in 2011 and 2012 is rather stable in the “main”
17 simulation. Despite differences for the methane loss the methane burden is very similar in the
18 two simulations. Therefore, it seems likely that the financial crisis had small impact on the
19 methane burden, but due to the long methane lifetime some of the difference in methane loss
20 could manifest as burden changes after 2012, which is the end-year of our simulations.

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Figure S3. Methane budget in main and financial simulations.

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S6 Comparison with CO observations

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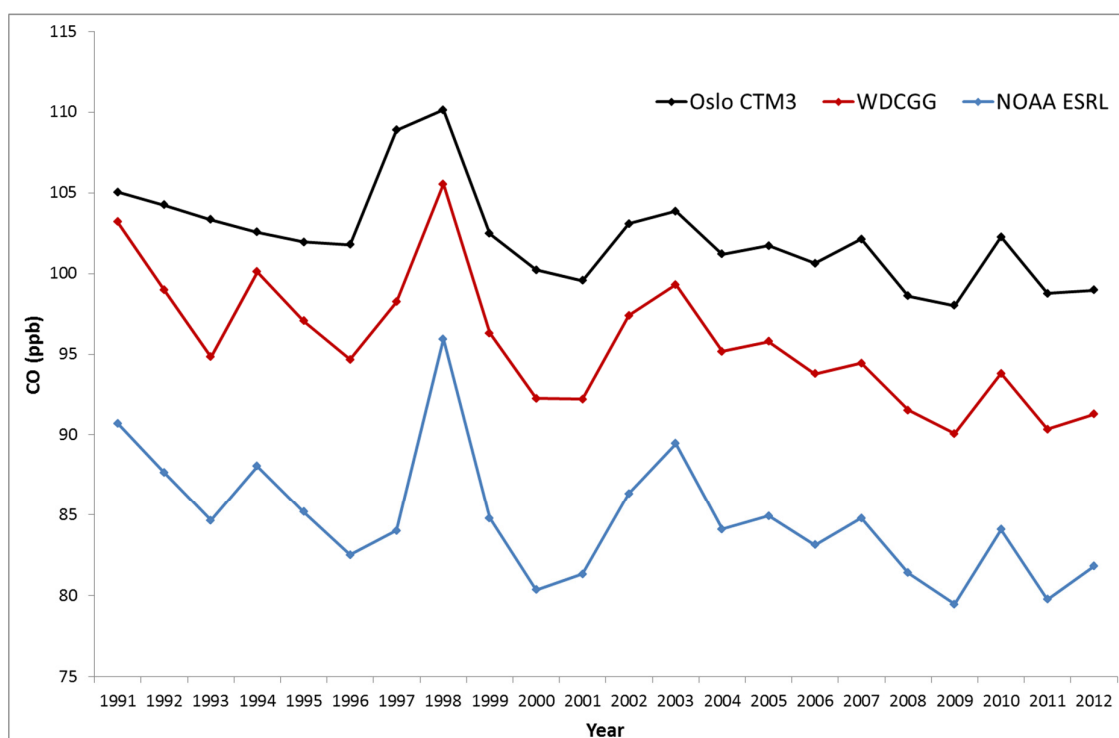
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Since reaction with OH is the major loss of CO from the atmosphere a comparison with CO measurements indicates whether modelled changes of OH are realistic. In this section, a first elementary evaluation is made based on comparison between our model results and observation-based estimates of global mean surface CO levels (Fig. S4). There is very good agreement for the long term evolution (years-decades). The same is the case for short term (year to year) variations, especially after 1996 when our simulations include inter-annual variability in meteorology and biomass burning emissions (CO, NO_x and NMVOCs). Our simulations do not fully account for the effect of the Pinatubo eruption and this also explains parts of the model discrepancy for the early nineties. Fig. S5 shows that the modeled gradual decline in CO levels over the period 1991-2012 is caused by stable to moderately declining CO emissions over the period combined with increasing OH for most of the period. Much of the large year to year fluctuations in CO levels (black line) are due to variation in emissions (purple line) caused by irregular occurrence and extent of vegetation fires. To summarize, the good agreement between our model and the observation based estimates supports that both the applied CO emission inventory and calculated OH changes are realistic.

23

1 The WDCGG global mean estimate is higher than the NOAA ESRL (Fig. S4) since NOAA
 2 ESRL is based on relatively unpolluted marine boundary layer stations while WDCGG
 3 include inland stations in their calculations. CO is more unevenly mixed than CH₄ due to its
 4 shorter lifetime. How polluted versus un-polluted air masses go in the global mean
 5 calculations therefore matters. The modeled global mean is based on all grid-boxes in the
 6 lowest model layer and one reason for its higher value might be the above mentioned
 7 sensitivity. However, comparing seasonal distributions (not shown) the model seems to
 8 overestimate CO levels throughout the Northern Hemisphere summer season and that is likely
 9 the main reason for its higher global mean.

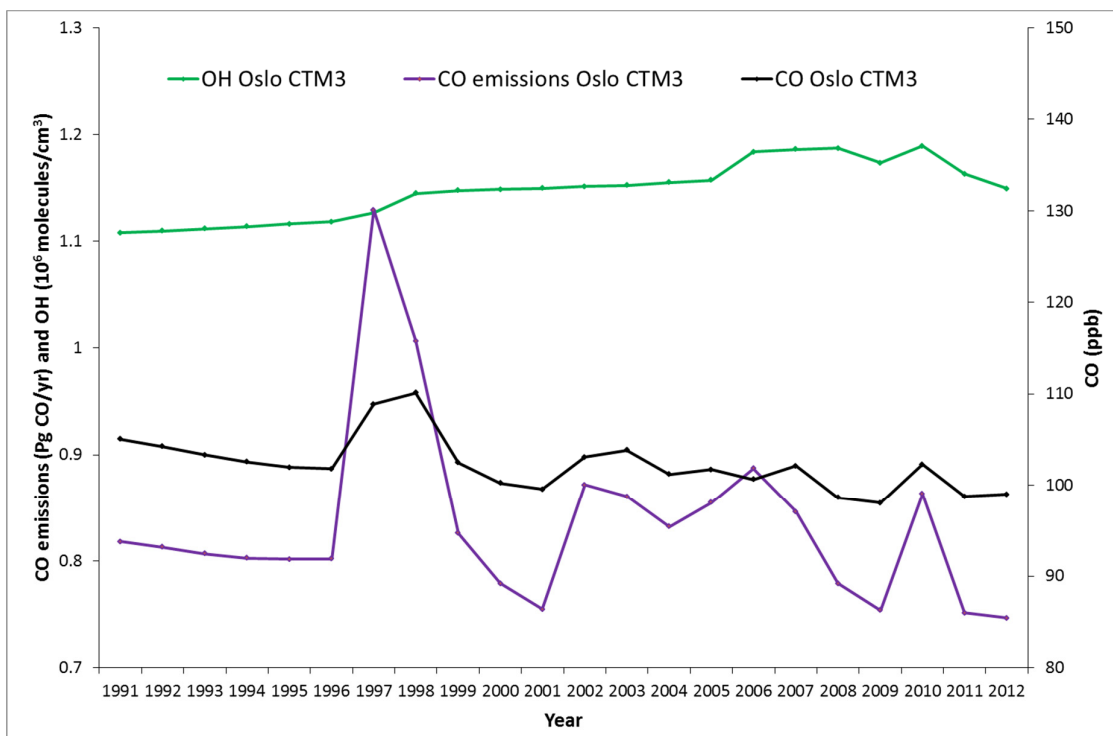
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12 **Figure S4.** Comparison of model and observation based yearly global mean surface CO for
 13 the period 1991-2012. NOAA ESRL data set provided by Paul Novelli, personal
 14 communication. WDCGG data set (WMO/WDCGG/GAW (2015) provided by WDCGG,
 15 personal communication.

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3 **Figure S5.** Yearly global average atmospheric OH concentration in the main simulation using
4 the reaction rate with CO as averaging kernel and yearly total global CO emissions (left y-
5 axis). Modeled global mean surface CO (right y-axis).

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7 **Acknowledgements**

8 We thank WDCGG and Paul Novelli for providing and sharing CO datasets used in figure
9 Figure S4.

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