

1 **Measurements of non-volatile aerosols with a VTDMA and**
2 **their correlations with carbonaceous aerosols in**
3 **Guangzhou, China**

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17

1 **Abstract**

2 Simultaneous measurements of aerosol volatility and carbonaceous matters were conducted at
3 a suburban site in Guangzhou, China in February and March 2014 using a volatility tandem
4 differential mobility analyzer (VTDMA) and an organic carbon/ elemental carbon (OC/EC)
5 analyzer. Low volatility (LV) particles, with a volatility shrink factor (*VSF*) at 300°C exceeding
6 0.9, contributed to 5% of number concentrations of the 40 nm particles and 11–15% of the 80–
7 300 nm particles. They were composed of non-volatile material externally mixed with volatile
8 material, and therefore did not evaporate significantly at 300°C. Non-volatile material mixed
9 internally with the volatile material were referred to as medium volatility (MV, $0.4 < VSF <$
10 0.9) and high volatility (HV, $VSF < 0.4$) particles. The MV and HV particles contributed to 57–
11 71% of number concentration for the particles between 40 nm and 300 nm in size. The average
12 EC and OC concentrations measured by the OC/EC analyzer were $3.4 \pm 3.0 \mu\text{g m}^{-3}$ and $9.0 \pm$
13 $6.0 \mu\text{g m}^{-3}$, respectively. Non-volatile OC evaporating at 475°C or above, together with EC,
14 contributed to 67% of the total carbon mass. In spite of the daily maximum and minimum, the
15 diurnal variations in the volume fractions of the volatile material, HV, MV and LV residuals
16 were less than 15% for the 80–300 nm particles. Back trajectory analysis also suggests that over
17 90% of the air masses influencing the sampling site were well-aged as they were transported at
18 low altitudes (below 1500 m) for over 40 h before arrival. Further comparison with the diurnal
19 variations in the mass fractions of EC and the non-volatile OC in $\text{PM}_{2.5}$ suggests that the non-
20 volatile residuals may be related to both EC and non-volatile OC in the afternoon, during which
21 the concentration of aged organics increased. A closure analysis of the total mass of LV and
22 MV residuals and the mass of EC or the sum of EC and non-volatile OC was conducted. It
23 suggests that non-volatile OC, in addition to EC, was one of the components of the non-volatile
24 residuals measured by the VTDMA in this study.

25

1 **1 Introduction**

2 Carbonaceous aerosols comprising organic carbon (OC) and elemental carbon (EC) or black
3 carbon (BC) are one of the major light absorption constituents and are abundant in particulate
4 matter (PM) (Rosen et al., 1978; Hansen et al., 1984; Japar et al., 1986; Chow et al., 1993;
5 Horvath, 1993; Liousse et al., 1993; Fuller et al., 1999; Putaud et al., 2010; Tao et al., 2014;
6 Zhang et al., 2015). In China, the worsening of visibility degradation associated with PM is of
7 increasing concern in recent years. In particular, numerous studies on air pollution were carried
8 out in different cities in China including the Pearl River Delta (PRD) region which is a fast-
9 developing economic zone (Cheng et al., 2006; Wu et al., 2007; Andreae et al., 2008; Chan and
10 Yao, 2008; Gnauk et al., 2008; Tan et al., 2013a). In 2007, the mass concentrations of EC and
11 OC measured at an urban Guangzhou (GZ) site were reported to vary from 6.8 to 9.4 $\mu\text{g m}^{-3}$
12 and from 13.4 to 22.5 $\mu\text{g m}^{-3}$ respectively (Yu et al., 2010).

13 Soot particles are often characterized in terms of EC and BC, depending on whether they are
14 measured thermally or optically (Penner and Novakov, 1996; Lavanchy et al., 1999; Cheng et
15 al., 2011 and references therein). Their optical properties are distinct when they are freshly
16 produced (Novakov et al., 2003). After aging processes such as cloud processing, chemical
17 reactions and coagulation, their structure, shape, size, mixing state and thus optical properties
18 change (Horvath, 1993; Liousse et al., 1993; Ghazi and Olfert, 2012). EC is typically measured
19 by thermal method such as the OC/EC analyzer (Chow et al., 2007), whereas BC is optically
20 measured using instruments such as aethalometer (Hansen et al., 1984), multi-angle absorption
21 photometer (Petzold and Schönlinner, 2004) and particle soot absorption photometer (Virkkula
22 et al., 2005). However, it is not possible to retrieve the mixing state of soot particles with these
23 techniques. To determine the mixing state of soot particles, single particle soot photometer
24 (Stephens et al., 2003), soot particle aerosol mass spectrometer (Onasch et al., 2012) and
25 Volatility Tandem Differential Mobility Analyzer (VTDMA) (Philippin et al., 2004) have been
26 used.

27 Ambient aerosols have varying volatility properties based on their chemical compositions.
28 VTDMA was first introduced by Rader and McMurry (1986) to study the behavior of aerosols
29 upon thermal treatment. A volatility shrink factor (*VSF*) is defined as the ratio of the particle
30 size after exposed to elevated temperature to the original particle size. Later Philippin et al.
31 (2004) developed a VTDMA which was capable of evaporating volatile material in aerosols at
32 temperatures up to 300°C. Non-volatile compounds at 300°C, such as EC, non-volatile organics

1 and sea salt, can internally mix with (or be coated with) volatile material. Note that the terms
2 “volatile” and “non-volatile” are here defined based on the operational parameters and how the
3 aerosol behave, when heated to 300°C in the VTDMA. They are different from the volatilities
4 defined under ambient conditions (Donahue et al., 2009; Murphy et al., 2014) or in other
5 measurement techniques (Twomey, 1968; Pinnick et al., 1987; Huffman et al., 2009). The
6 composition of these non-volatile residuals can vary spatially and temporally. Previous studies
7 have demonstrated a good agreement between the mass concentration of BC and the mass
8 concentration of non-volatile particles that experienced size reductions of 5 to 10% upon
9 heating at 300°C in the VTDMA (Frey et al., 2008). Various studies have also used the VTDMA
10 to estimate the mixing states of soot particles (Philippin et al., 2004; Rose et al., 2011; Levy et
11 al., 2014; Zhang et al., 2016). Particles with small volatile fractions, i.e. $VSF > 0.9$ at 300°C,
12 are often assumed to be soot particles externally mixed with particles with volatile material at
13 300°C. Particles with larger volatile fractions, i.e. $VSF < 0.9$ at 300°C, were assumed to
14 represent soot particles internally mixed (coated) with the volatile material (Cheng et al., 2006;
15 Wehner et al., 2009).

16 Organics also contribute to light absorption by atmospheric particles (Bond, 2001; Kirchstetter
17 et al., 2004; Chen and Bond, 2010). Laboratory studies have shown that organic aerosols may
18 form low volatility oligomers after aging for a long time (e.g. Kalberer et al., 2004). Huffman
19 et al. (2009) showed that highly oxygenated, aged organic aerosols exhibited similar or lower
20 volatility than the primary organic aerosols or the less oxygenated particles. Recently, Häkkinen
21 et al. (2012) compared the residual mass derived from a volatility differential mobility particle
22 sizer (VDMPS) at 280°C with BC measured by an aethalometer and organics measured by an
23 Aerodyne aerosol mass spectrometer (AMS). It was found that the mass fraction remaining of
24 non-BC residuals in VDMPS measurements was positively correlated with the mass fraction of
25 organics in AMS measurements.

26 In this study, simultaneous measurements of aerosols volatility and carbonaceous matter were
27 made at a suburban site in Guangzhou, China during wintertime in February and March 2014
28 using a VTDMA and a semi-continuous OC/EC analyzer, respectively. The volatility
29 measurements were made for ambient aerosols ranging from 40 nm to 300 nm in diameter. Here
30 residuals remaining after heating at 300°C in the VTDMA are referred to as non-volatile in this
31 study. We report the average values, time series and diurnal variations in the number and
32 volume fractions of the volatile and non-volatile material, as well as the OC and EC

1 concentrations. We examine the relationships of the non-volatile material upon heating at
2 300°C to EC and to the non-volatile OC, based on analyses of the diurnal patterns and mass
3 closures of the OC/EC and VTDMA data. Finally, we discuss the influence of air mass origins
4 on the volatility of the sampled aerosols and concentrations of OC and EC based on back
5 trajectory analysis.

6

7 **2 Methodology**

8 **2.1 Experimental**

9 **2.1.1 Measurement details**

10 The campaign took place at the China Meteorological Administration (CMA) Atmospheric
11 Watch Network (CAWNET) Station in Panyu, Guangzhou, China in summer from July to
12 September 2013 and in winter from February to March 2014. The station is operated by the
13 Institute of Tropical and Marine Meteorology (ITMM) of the CMA. The Panyu station is
14 located at the center of the PRD region and on the top of Dazhengang Mountain (23°00' N, 113
15 °21' E) with an altitude of about 150 m (Fig. S1 in Supplemental Information) (Tan et al., 2013a).
16 It is about 120 m above the city and is surrounded by residential neighborhoods with no
17 significant industrial pollution sources nearby. Measurements of particle number size
18 distributions, volatility, mass concentrations of EC and OC in winter were made from 6
19 February to 21 March 2014. Some of the measurements were not made continuously due to
20 maintenance work and hence only the periods with concurrent VTDMA and OC/EC
21 measurements were analyzed.

22 **2.1.2 VTDMA measurements**

23 We used a custom-made VTDMA based on a Hygroscopic TDMA system developed in ITMM
24 (Tan et al., 2013b), with the humidifier between the two DMAs replaced by a heated tube which
25 induce evaporation of volatile material. In our VTDMA system shown in Fig. 1, ambient
26 aerosols were sampled by a PM_{2.5} inlet and subsequently passed through a dryer at relative
27 humidity below 20%. The dry aerosols were then directed through a neutralizer and entered the
28 first differential mobility analyzer (DMA₁) (Stream 1) to produce mono-disperse aerosols of
29 diameter between 40 nm and 300 nm, D_0 . The mono-disperse aerosols went either via path (a)
30 or (b) in Fig. 1 after leaving the DMA₁. In path (a), they (Stream 2) were directed to a

1 condensation particle counter (CPC, TSI Model 3772) to obtain particle counts, N_{D_0} . The
2 particle number size distribution of the ambient aerosols, $dN/d\log D_p$, was also measured by
3 varying the DMA₁ voltage (SMPS scan). Afterwards, the mono-disperse aerosols were directed
4 via path (b) to a heated tube for volatility measurement (V-Mode) sequentially at 25°C, 100°C
5 and 300°C. The heating tube was a 1/2'', 80 cm long stainless steel tube with an inner diameter
6 of 8 mm. With a sample flow rate of 1 L min⁻¹, the resulting residence time in the heated section
7 of the VTDMA was 2.4 s. The estimated aerosol velocity on the center line was 0.33 m s⁻¹.
8 Compared to the residence time of 0.3 s to 1 s in other VTDMA systems (e.g. Brooks et al.,
9 2002; Philippin et al., 2004; Villani et al., 2007), the residence time in our VTDMA is assumed
10 to be long enough for the volatile material to be effectively vaporized. After leaving the heating
11 tube, the flow entered a heat exchanger measuring 30 cm in length to ensure sufficient cooling
12 before entering DMA₂.

13 Upon heating, volatile components of particles such as sulfate, nitrate and volatile organics
14 vaporize at different temperatures depending on their volatilities. As mentioned in Section 1,
15 the volatility shrink factor, VSF , is defined as the ratio of particle diameter after heating at
16 temperature T , $D_{p,T}$, to the diameter before heating, D_0 :

$$17 \quad VSF(T) = \frac{D_{p,T}}{D_0} \quad (1)$$

18 The VSF indicates the size reduction of the ambient particles upon heating. The value of VSF
19 is always smaller than or equal to one, depending on the amount of volatile material vaporized
20 at the heating temperature T . The VSF is used to divide the particles into three groups, namely
21 low volatility (LV), medium volatility (MV) and high volatility (HV) particles. In this study,
22 we focus on the measurements made at 300°C. The VSF ranges for the LV, MV and HV
23 particles upon heating at 300°C are defined as follows: above 0.9, between 0.4 and 0.9 and
24 below 0.4, respectively (Fig. 2) (Wehner et al., 2004; Wehner et al., 2009).

25 The size distribution, $dN'/d\log D_p$ of the remaining particles (hereafter the residuals) were
26 measured by DMA₂ and CPC (Fig. 2b). It can provide information of the mixing state of the
27 sampled aerosols. A uni-modal distribution indicates the presence of internally-mixed particles
28 exhibiting uniform size reduction upon heating, whereas a multi-modal distribution indicates
29 externally-mixed particles of different composition and volatilities. In the multi-modal
30 distribution, each mode represents particles of similar composition and volatility. In this study,
31 multiple modes of LV, MV and HV were observed in the distribution after heating. The LV
32 particles were assumed to represent EC and non-volatile OC externally mixed with the volatile

1 material, while MV and HV particles were assumed to represent EC and non-volatile OC
2 internally mixed with volatile material. While the volatile material in the MV and HV particles
3 were referred to as VM, those exist as external mixtures with the LV, MV and HV particles
4 were referred to as completely vaporized (CV) particles. The CV particles evaporated
5 completely without leaving behind any residuals at 300°C. Excluding particle diffusional and
6 thermophoretic losses, and assuming that the residual material did not evaporate to the sizes
7 below the detection limit of the CPC (here 10 nm), the evaporation of VM and CV did not
8 change the number concentrations of LV, MV and HV particles.

9 Overall it took around one and a half to two hours to complete a cycle of measurements which
10 consisted of SMPS scans and V-Mode measurements at 25°C, 100°C and 300°C. At each
11 temperature, the sampling time for six selected diameters from DMA₁ (40 nm, 80 nm, 110 nm,
12 150 nm, 200 nm and 300 nm) took about half an hour and SMPS scans were made in-between.
13 Hereafter, notations with the superscript prime refer to the LV, MV or HV residuals measured
14 by DMA₂ and CPC after heating, while the corresponding ones without the prime refer to the
15 LV, MV or HV residuals in ambient aerosols prior to heating.

16 **2.1.3 OC/EC measurements**

17 A semi-continuous Sunset OC/EC Analyzer (Model 4) was used to measure PM_{2.5} mass
18 concentrations of organic carbon and elemental carbon, m_{OC} and m_{EC} respectively, on an hourly
19 basis (Turpin et al., 1990; Birch and Cary, 1996; Wu et al., 2012). With the OC/EC Analyzer
20 the ACE-Asia protocol (a NIOSH-derived protocol) was adopted, in which OC was evaporated
21 at four set temperatures of 310°C, 475°C, 615°C and 870°C with pure helium (He) as a carrier
22 gas, whereas EC was combusted at temperatures between 550°C and 870°C under He and 2%
23 oxygen (O₂) (Schauer et al., 2003; Wu et al., 2012). The OC contents were named OC₁ to OC₄
24 based on the temperature protocol of the OC/EC analyzer (Table 1). The mass of EC determined
25 at different temperatures was grouped together in subsequent analysis.

26 It is plausible that in the VTDMA measurements, there were volatile or semi-volatile OC that
27 vaporize at 300°C or below. This vaporized OC is assumed to correspond to OC₁, which was
28 vaporized at 310°C although this OC/EC temperature was slightly higher than the temperature
29 of 300°C in the VTDMA. With this assumption, the residual particles of the VTDMA at 300°C
30 (LV and MV residuals) are postulated to consist of (1) OC₂ to OC₄, which were vaporized at
31 475°C and above, and (2) EC and other refractory PM components. We have ignored the HV
32 residuals as their contributions to the total volume of the particles were insignificant in

1 comparison with LV and MV residuals (Section 3.1). In Section 3.5, we will conduct a mass
2 closure analysis based on the VTDMA and OC/EC measurements to examine this assumption.

3 **2.2 Data analysis**

4 **2.2.1 Number fractions**

5 The number fractions of LV, MV and HV residuals ($\Phi'_{N,LV}$, $\Phi'_{N,MV}$ and $\Phi'_{N,HV}$, with their sum
6 being equal to unity) in Stream 2 on Fig. 1 were obtained from $dN'/d\log D_p$ measured by the
7 VTDMA. However, these fractions do not represent the actual number fractions of LV, MV
8 and HV particles ($\Phi_{N,LV}$, $\Phi_{N,MV}$ and $\Phi_{N,HV}$) before heating because some of the particles can
9 evaporate completely (CV) and due to diffusion and thermophoretic losses. The number
10 fraction of CV ($\Phi_{N,CV}$) was first obtained by considering the number fractions of the residuals
11 ($1-\Phi_{N,CV}$) and the number concentrations at a selected diameter D_0 before heating (N_{D_0}) and
12 after heating (N'):

$$13 \quad N_{D_0} \cdot \eta_{D_0} \cdot (1 - \Phi_{N,CV}) = N' \quad (2)$$

14 where η_{D_0} is the transport efficiency of particles.

15 In Eq. (2) we assume that η is the same for LV, MV and HV particles. η accounts for particle
16 losses between DMA₁ and DMA₂ due to diffusion and thermophoretic forces (Philippin et al.,
17 2004), and it varies as a function of particle size and heating temperature. η at each particle
18 diameter and VTDMA temperature was determined in laboratory calibrations with sodium
19 chloride (NaCl) particles, which do not evaporate (i.e., $\Phi_{N,CV} = 0$) at the temperatures used in
20 our experiments. The transmission efficiency of NaCl of several selected diameters in
21 temperatures between 50°C and 300°C is provided in the supplemental information (Fig. S2).
22 From the known η and observational data obtained with the VTDMA providing N_{D_0} and N' ,
23 $\Phi_{N,CV}$ can be calculated from Eq. (2). Afterwards, $\Phi_{N,LV}$, $\Phi_{N,MV}$ and $\Phi_{N,HV}$ were obtained by
24 renormalizing $\Phi'_{N,LV}$, $\Phi'_{N,MV}$ and $\Phi'_{N,HV}$ with $(1-\Phi_{N,CV})$ so that the sum of $\Phi_{N,LV}$, $\Phi_{N,MV}$, $\Phi_{N,HV}$
25 and $\Phi_{N,CV}$ equaled unity.

26 **2.2.2 Volume fractions**

27 The volume fractions of LV, MV, HV residuals and CV ($\Phi_{V,LV}$, $\Phi_{V,MV}$, $\Phi_{V,HV}$ and $\Phi_{V,CV}$) at each
28 selected diameter D_0 are defined as the ratios of the volume of LV, MV, HV residuals and CV

1 to the total volume of the mono-disperse particles before heating. By assuming that the residuals
 2 are spherical in shape, $\Phi_{V,LV}$, $\Phi_{V,MV}$ and $\Phi_{V,HV}$ can be calculated by:

$$3 \quad \Phi_{V,i} = \frac{N_i \times \frac{\pi}{6} D_{p,i}^3}{N_{D_0} \times \frac{\pi}{6} D_0^3} = \Phi_{N,i} \cdot \frac{D_{p,i}^3}{D_0^3} \quad (3)$$

4 where N_i and $D_{p,i}$ are the number concentration and mean residual diameter of $i = LV, MV$ or
 5 HV residuals.

6 For LV particles, it is assumed that D_0 and mean D_p are the same and hence $\Phi_{V,LV}$ is the same
 7 as $\Phi_{N,LV}$. For MV and HV particles, the mean D_p is smaller than D_0 due to the evaporation of
 8 volatile material. The number weighted mean residual diameter (D_p) was calculated by:

$$9 \quad D_{p,i} = \frac{\sum_j D_{p,i} \cdot N_{i,j}}{N_i} \quad (4)$$

10 where $D_{p,i}$ and $N_{i,j}$ are the residual diameter and number concentration of $i = MV$ or HV at the
 11 75 diameter bins (j) of *VSF*, respectively.

12

13 The volume fractions of the evaporated material were calculated from the volume fractions of
 14 the residuals. The calculation for $\Phi_{V,CV}$ was similar to that for $\Phi_{V,LV}$. Since the particle has
 15 completely vaporized, the vaporized volume is equivalent to the volume of the original particle.
 16 Hence, $\Phi_{V,CV}$ is the same as $\Phi_{N,CV}$:

$$17 \quad \Phi_{V,CV} = \frac{N_{CV} \cdot \frac{\pi}{6} D_{p,CV}^3}{N_{D_0} \cdot \frac{\pi}{6} D_0^3} = \Phi_{N,CV} \quad (5)$$

18 where $D_{p,CV} = D_0$. Since the sum of the total volume fraction of CV, VM and the residuals of
 19 LV, MV, HV equaled unity, $\Phi_{V,VM}$ was obtained after the above volume fractions were
 20 calculated. Furthermore, we also calculated the volume fraction remaining (*VFR*), defined as
 21 the volume ratio of the residual to its host particle, to aid our discussions later:

$$22 \quad VFR_i = \frac{N_i \cdot \frac{\pi}{6} D_{p,i}^3}{N_i \cdot \frac{\pi}{6} D_0^3} = \frac{D_{p,i}^3}{D_0^3} \quad (6)$$

23 where N_i and $D_{p,i}$ are the number concentration and mean residual diameter of $i = MV$ or HV
 24 after heating, respectively.

2.2.3 Particle size distributions of number, volume and mass concentrations of LV, MV and HV residuals

Due to the differences in the size cuts of the VTDMA and the OC/EC analyzer, log-normal fits extrapolated to 5 μm were applied to the particle number size distributions of the residuals of LV, MV and HV ($dN/d\log D_{p,i}$, where $i = \text{LV, MV or HV}$) to estimate the volume and then mass concentrations of the ambient aerosols for comparison with $\text{PM}_{2.5}$ OC/EC measurements. The volume size distributions ($dV/d\log D_{p,i}$) were calculated by:

$$\frac{dV}{d\log D_{p,i}} = \frac{dN}{d\log D_{p,i}} \cdot \frac{\pi}{6} D_{p,i}^3 \quad (7)$$

where $D_{p,i}$ is the mean residual diameter as defined in Section 2.2.2.

Volume (V) concentrations of LV, MV and HV residuals can then be calculated by integrating the area under the fitted curves. As we only focus on LV and MV, densities of 1.0 g cm^{-3} (Hitzenberger et al., 1999) and 1.5 g cm^{-3} are applied to V_{LV} and V_{MV} to obtain mass (m) concentrations of LV and MV residuals, respectively. The choice of the densities is based on the assumption that LV and MV residuals are dominated by soot and non-volatile OC, respectively.

3 Results and Discussions

3.1 Overview

The time series of meteorological conditions, particle number size distribution, $\text{PM}_{2.5}$, OC and EC concentrations during the campaign are presented in Fig. 3. Overall, the campaign was under the influence of the prevailing northerly wind with an average wind speed and temperature (\pm one standard deviation) of $1.73 \pm 0.95 \text{ m s}^{-1}$ and $14.8 \pm 5.1^\circ\text{C}$, respectively. The average $\text{PM}_{2.5}$ concentration was $48 \pm 26 \mu\text{g m}^{-3}$. A few colder periods were observed, during which the wind speed increased and the temperature decreased. In general, the low wind speed favored the accumulation of $\text{PM}_{2.5}$. During the campaign OC concentrations ranged from 0.5 to $47.0 \mu\text{g m}^{-3}$ with an average of $9.0 \pm 6.0 \mu\text{g m}^{-3}$, while EC concentrations ranged from 0.2 to $23.0 \mu\text{g m}^{-3}$ with an average of $3.4 \pm 3.0 \mu\text{g m}^{-3}$. OC_1 , the most volatile group among OC_1 to OC_4 in OC/EC analysis, accounted for one-third of the total carbon mass (Fig. 4).

1 On 17 Feb, 12 and 17 Mar 2014, the daily-averaged $\text{PM}_{2.5}$ concentrations exceeded $95 \mu\text{g m}^{-3}$;
2 and they were nearly twice the daily-averaged values of the other days (Fig. 3, shaded area in
3 grey). Results of 72 h back trajectories (Stein et al., 2015; Rolph, 2016) showed that air masses
4 arriving at the site on or before these three days mostly originated from the continental or
5 oceanic area close to Eastern China (Fig. S3). The SMPS data also showed a mode near 100
6 nm with a high particle number concentration (Fig. 3).

7 The temporal variation of the number concentration of MV particles having an initial diameter
8 of 80 nm or above tracked reasonably well with the accumulation of $\text{PM}_{2.5}$ as the particles aged
9 and became more internally mixed (Fig. 3 and S4). Furthermore, the number concentration of
10 MV particles showed a size dependence in the 80–300 nm particles. There were days, e.g.,
11 from 24 Feb to 10 Mar 2014, when the number concentration of 300 nm MV particles did not
12 track well with $\text{PM}_{2.5}$ mass concentration. The mode of the total particle number size
13 distribution was below 100 nm and the number concentrations of 300 nm particles were low
14 (Fig. 3). $\text{PM}_{2.5}$ mass concentration tracked better the number concentrations of 80 nm to 150
15 nm MV particles (Fig. 4a to S4c) than those of 200 nm and 300 nm MV particles (Fig. S4d and
16 S4e).

17 The average number and volume fractions of CV, HV, MV and LV in VTDMA measurements
18 at 300°C are summarized in Table 2. VM is internally mixed with MV and HV residuals, and
19 hence does not have a separate contribution to number concentrations. Overall, HV and MV
20 particles, indicator for aged aerosols with internally mixed non-volatile and volatile material,
21 accounted for 57% to 71% of the total particle number concentration. Non-volatile material
22 (LV, MV and HV residuals) accounted for 15% to 26% of the total volume of selected particles
23 before heating. While the CV and HV fractions were larger in the finest particles selected (D_0
24 = 40 nm), MV and LV were more abundant in larger particles ($D_0 > 80$ nm). As in Rose et al.
25 (2006), fresh emissions like soot adsorb or absorb volatile material during atmospheric
26 processing. The smaller particles grew faster than the larger ones because of their higher ratio
27 of surface area to volume. When they were heated in the VTDMA at 300°C , the smaller
28 particles reduced more substantially in size, as reflected in the higher CV and HV fractions and
29 lower MV and LV fractions. The higher abundance of MV and LV in the larger size particles
30 could also be explained by the aged particles arriving at the sampling site. Since the sampling
31 site is located on top of a mountain with an altitude of 150 m, the particles were likely aged
32 upon arrival. Non-volatile particles in the ultrafine modes from fresh emissions can be aged

1 with both non-volatile and volatile material, and became larger in size. Nevertheless, the
2 detection limit of the downstream DMA and CPC in the VTDMA system is 10 nm. The particles
3 with diameters below the detection limit leads to an overestimation of CV and an
4 underestimation of the non-volatile residuals for the finest particles selected (with an initial
5 diameter of 40 nm).

6 **3.2 Diurnal variations**

7 Figure 5 shows diurnal variation in the fraction of CV, HV residual, MV residual, LV residual
8 and VM in the total volume of particles of dry initial diameters of 40, 150 and 300 nm. For 40
9 nm particles, a clear maximum and minimum of the fraction of CV, VM and HV residuals are
10 observed at 08:00 and 13:00, respectively. The diurnal variation of the HV and MV particles in
11 the 40 nm particles was clearer in terms of number fraction (Fig. S5). Furthermore, the trend of
12 CV was opposite to those of VM, HV and MV. The increase of CV in the 40 nm particles and
13 to a lesser extent of LV in the 150 nm and 300 nm particles in the morning is consistent with
14 traffic pattern. Fresh emissions of volatile and non-volatile material, likely OC and EC, were
15 externally mixed and contributed to CV and LV, respectively. As time progresses in a day, the
16 highly volatile species (CV) which were freshly emitted in the morning, may evaporate and
17 react to form less volatile particles and become VM instead of CV (Robinson et al., 2007).
18 Alternatively, these CV particles could also coagulate with smaller particles to form VM
19 containing particles. Less fresh emissions with more CV particles turning into VM on MV and
20 HV particles could explain the trend that the number and volume fractions of CV decreased
21 while those of MV and HV increased (Fig. 5 and Fig. S5).

22 We also used the diurnal variations in the volume fraction remaining (*VFR*), again defined as
23 the volume ratio of the residual to its *host* particle (not to the total volume of all particles), to
24 examine the size changes of the non-volatile residuals of HV and MV particles. The *VFR* of
25 HV did not exhibit any obvious diurnal variations but the *VFR* of MV peaked near 18:00. The
26 *VFR* of the 40 nm MV particles increased after 14:00 while those of the 150 nm and 300 nm
27 MV particles increased after 15:00. Since the *VFR* of HV and MV were relatively constant
28 during the day, the increase in the VM fraction after the morning rush hours could be attributed
29 to the increase in the number concentrations of the HV and MV particles instead of changes in
30 the amount of VM on the MV or HV residuals.

1 The diurnal variations for particles larger than 80 nm were much less obvious than those for 40
2 nm particles in this study and in others (Rose et al., 2011; Cheng et al., 2012; Zhang et al.,
3 2016). In winter, the atmosphere is more stable, resulting in a poorer dilution of aged particles
4 with the less polluted aerosols from higher up (Rose et al., 2006). When the aged pollutants
5 were trapped near the ground, the effect of aging of fresh emissions weakened. Therefore,
6 although a daily maximum and a daily minimum were still observed for particles larger than 80
7 nm, the variation was mostly within 15%.

8 The diurnal variations in the mass fractions of OC and EC in $PM_{2.5}$ provided further insights to
9 the observations above (Fig. 6). The OC and EC data on Mar 12 and 17 were excluded since
10 they were more than two standard deviations higher than those on other days. Subtle morning
11 peaks between 06:00 and 10:00 were observed for the volume fraction of LV residuals (Fig. 5).
12 A similar peak was observed for the mass fraction of EC in $PM_{2.5}$ in the morning (Fig. 6). This
13 suggests that the LV particles may be related to the EC from vehicle emissions in the morning.
14 This EC was relatively less aged and externally mixed with the other volatile material. In the
15 late afternoon, the LV residuals showed another peak between 17:00 and 19:00 whereas the
16 mass fraction of EC in $PM_{2.5}$ exhibited a minimum at 15:00, after which it increased
17 continuously. The continuous increase in EC at night is likely related to the increase of heavy-
18 duty diesel traffic (Zhang et al., 2015), which was restricted during daytime (Bradsher, 2007).

19 Although OC_1 contributed to about half of the total OC mass, the diurnal variation in the mass
20 fraction of OC in $PM_{2.5}$ was driven by the total mass of OC_2 , OC_3 and OC_4 (OC_{2-4}), which
21 reached a minimum between 05:00 and 09:00 and increased until 19:00. OC can be attributed
22 to both primary and secondary sources. The increased mass fraction of OC in $PM_{2.5}$ and OC-
23 to-EC ratio in the afternoon suggest that the sources of OC were less related to traffic but more
24 to the aging and formation of secondary organic aerosols (Turpin et al., 1990; Chow et al.,
25 1996). These OC_2 , OC_3 and OC_4 may be highly oxygenated species or oligomers that are less
26 volatile than primary or less oxygenated organics (Kalberer et al., 2004; Huffman et al., 2009).

27 It is interesting to note that the volume fraction of the LV residuals and the *VFR* of MV particles
28 at different sizes showed a dip in the afternoon (Fig. 5, third column from the left). The *VFR* of
29 40 nm MV particles showed a dip at 14:00 while those in 150 nm and 300 nm particles showed
30 a dip at 15:00. The volume fraction of LV residuals in 150 nm and 300 nm particles reached a
31 minimum at 13:00 and 15:00, respectively. Because EC decreased between 12:00 and 15:00,
32 the increase in the volume fraction of LV residuals in 150 nm particles since 13:00 and the *VFR*

1 of 40 nm MV particles since 14:00 may be related to the increased presence of aged organics
2 as well as the EC particles which aged via coagulation and condensation.

3 **3.3 Back trajectory analyses**

4 We calculated 72 h back trajectories of air masses arriving at the sampling site (23°00 N,
5 113°25' E) at 4 h intervals (at 00:00, 04:00, 08:00, 12:00, 16:00 and 20:00 local time, UTC +8)
6 using the PC version of the HYSPLIT4 (Hybrid Single Particle Lagrangian Integrated
7 Trajectory, version 4) model (Stein et al., 2015; Rolph, 2016). Archived meteorological data
8 from the Global Data Assimilation System (GDAS) 1-deg was employed and the receptor
9 height was set at 500 m above ground level (a.g.l.). The 191 back trajectories calculated were
10 grouped into six clusters based on their spatial distribution (Fig. 7).

11 Overall, the sampling site was mostly affected by northwesterly and northeasterly air masses.
12 Clusters 1 and 3 are coastal and continental air masses, respectively, although both originated
13 from the northeast. Clusters 4, 5 and 6 represent continental air masses originating from the
14 northwest. Cluster 2 is a group of maritime air masses originating from the East China Sea
15 northeast or east of Guangzhou. While the air masses in cluster 6 were transported at relatively
16 high speeds and altitudes (over 3000 m a.g.l.), the air masses in all the other clusters were
17 transported at an altitude below 1500 m a.g.l. for over 40 h before arriving at the site. As the air
18 masses in cluster 6 only occurred for less than three days and since the corresponding VTDMA
19 and OC/EC data were sometimes unavailable, cluster 6 will be excluded from the following
20 discussion.

21 The average PM_{2.5}, OC and EC concentrations associated with the air masses from the northeast
22 of Guangzhou (clusters 1, 2 and 3) were higher than those from the northwest (clusters 4 and 5,
23 Table 3). Days associated with the coastal and maritime air masses were more polluted than
24 days associated with continental air masses for several reasons. First, south China as a region
25 is often affected by the high pressure system moving eastward or southward from the continent
26 out to sea in winter. When the maritime or coastal air streams entered from the southeast of the
27 sampling site at Panyu, the atmosphere at the sampling site became more stable with low local
28 wind speeds (e.g. the polluted days on Feb 17 and Mar 12, 16 and 17, Fig. 3 and S3). The local
29 pollutants accumulated and the city was also affected by pollutants from the southeastern areas
30 of the site (e.g. Shenzhen, Nansha and Dongguan). Second, land-sea breeze circulation were
31 observed when the sampling site was under the influence of maritime air masses from Mar 18

1 to 20. During the day, southeasterly wind prevailed and the wind speed was higher. In the
2 evening, the southeasterly wind was gradually replaced by a southwesterly or northwesterly
3 wind and the wind speed decreased (Fig. 3). The cycle started again in the morning when the
4 westerly wind was gradually replaced by southeasterly wind. Such land-sea breeze effects can
5 result in an effective redistribution and accumulation of air pollutants within the PRD region
6 (Lo et al., 2006).

7 Furthermore, $PM_{2.5}$ in the northeastern parts of China can exceed $200 \mu g m^{-3}$ due to both
8 enhanced emissions from coal combustion for heating and poor dispersion during wintertime
9 (Gu et al., 2014). Under the influence of the prevailing northerly or northeasterly wind in China,
10 these pollutants were often transported to southern China and the East China Sea (Chen et al.,
11 2012). The pollutants might also have accumulated when the maritime air masses spent about
12 two days across Taiwan and the coast of south China. In contrast, continental air masses in
13 cluster 5 moved slightly faster, and were often associated with the cold front period during
14 which the local wind speed and pressure increased but the temperature decreased (Fig. 3). As
15 the cold air masses passed through the city, dispersion and clearance of pollutants were
16 promoted, resulting in lower $PM_{2.5}$ concentrations (Tan et al., 2013a). Therefore, unlike in other
17 coastal cities like Hong Kong (Lee et al., 2013), in Panyu the maritime air masses could lead to
18 more severe pollution than the continental ones in winter.

19 The five clusters were further analyzed to study the influence of air mass history on aerosol
20 volatility. The number fractions of CV, HV, MV and LV of the six selected diameters in
21 VTDMA measurements are regrouped based on the clusters as shown in Fig. 8. The total
22 number fractions of the non-volatile residuals (sum of HV, MV and LV) were similar in all
23 clusters. The maritime air masses (cluster 2) had a slightly higher fraction of LV particles while
24 the continental air masses originating from the northwest of the site (clusters 4 and 5) had a
25 higher fraction of HV particles. Although the air masses in clusters 1 and 5 originated from
26 further away and traveled at relatively higher speeds than those in clusters 2, 3 and 4, all the
27 clusters involved transport at low altitudes (below 1500 m) for over 40 h, likely due to the
28 generally lower mixing heights in winter. Therefore, it is plausible that the aerosol particles in
29 these air masses were all well-aged upon arrival. Similar results were observed in Beijing by
30 Wehner et al. (2009). This could be another reason for the lack of size dependence of the
31 number, volume fractions and diurnal variation for the particles larger than 80 nm. When the
32 transported air masses mixed with the local pollutants, the size dependence of the number

1 fractions of different volatility groups as well as the aging of local emissions was further
2 reduced.

3 We also examined the volatility shrink factor (*VSF*) distributions of 40 nm, 110 nm and 300 nm
4 particles upon heating at 300°C (Fig. 9). Log-normal fittings with a three-peak solution were
5 applied to the distributions. The average *VSF* modes of the peaks were located at 0.38 ± 0.021
6 (peak 1), 0.60 ± 0.066 (peak 2) and 0.95 ± 0.007 (peak 3), respectively. The standard deviation
7 of the corresponding normal distribution (σ) of peak 3 was the smallest among the three peaks
8 ($\sigma < 0.1$). For the same particle size, the *VSF* distributions in the *VSF* range between 0.3 and
9 0.8 in cluster 5 was relatively more uni-modal than those of other clusters (Fig. 9b and 9c). This
10 suggests that the composition in cluster 5 was more homogeneous. Cluster 1 also consisted of
11 long-range transported air masses but they likely passed through areas that are more polluted
12 and mixed with different types of pollutants.

13 **3.4 New particle formation**

14 Two new particle formation (NPF) events were observed in the campaign on 20 Feb and 13
15 Mar 2014 (Fig. 3). Since VTDMA data were not available during the NPF event on 13 Mar
16 2014, we only focus on the NPF event on 20 Feb 2014 which happened after a cold front under
17 a low PM_{2.5} concentration. On 20 Feb 2014, a sub-20 nm particle mode was first observed at
18 12:00. This particle mode grew continuously until it reached 120 nm at 02:00 on 21 Feb 2014.
19 In the VTDMA data, a sharp increase in the number concentration of HV particles having an
20 initial diameter of 40 nm was observed at 17:00 on 20 Feb 2014 (Fig. 10). This event is likely
21 related to the growth of the newly formed particles when they mixed with the volatile material
22 accumulated via condensation or adsorption. The volatile material which extensively condensed
23 on the pre-existing particles could be sulfate, ammonium and organics. They were found to be
24 the major species contributing to particle growth in the NPF events at different locations (Zhang
25 et al., 2004; Smith et al., 2008; Zhang et al., 2011; Yue et al., 2016). Zhang et al. (2004)
26 observed that sulfate was always the first and the fastest species to increase in concentration
27 during an NPF event. They also suggested that photochemically formed secondary organics
28 contributed significantly to the growth of the ultrafine particles. Recently, Yue et al. (2016)
29 reported that sulfate, ammonium and organics were the main contributors to particle growth in
30 the NPF events in Taoyuan of the PRD region. As these particles aged further, they grew larger
31 as reflected in the increase in number concentrations of larger MV particles and the increase in
32 PM_{2.5} mass (Fig. 10). Similar results were also observed in the study in Beijing by Wehner et

1 al. (2009). Furthermore, the growth of the newly formed particles can also be observed from
2 the number size distributions of the HV, MV and LV particles at different times on 20 and 21
3 Feb 2014 (Fig. 11). The mode of the HV particles increased from 40 nm at 17:00 to 80 nm at
4 21:00 on 20 Feb 2014. The mode stayed at 80 nm while the corresponding number
5 concentration decreased at 02:00 on 21 Feb 2014. In contrast, the number concentration and
6 diameter mode of the MV particles grew continuously. The HV and MV particle concentrations
7 and diameter modes underwent much smaller changes on the non-event day of 28 Feb 2014
8 (Fig. 11).

9 **3.5 Closure analysis for LV and MV residuals at 300°C, OC and EC**

10 The closure analysis of EC or the sum of EC, OC₂, OC₃, and OC₄ and the total mass of LV and
11 MV residuals was conducted (Fig. 12). Good correlations ($R^2 > 0.9$) for both EC and the sum
12 of EC, OC₂, OC₃, and OC₄ with the total mass of LV and MV residuals were obtained.
13 Nonetheless, the slope for the total mass of LV and MV residuals to the mass of EC (2.94) is
14 more than two times of that for the total mass of LV and MV residuals to the sum of EC, OC₂,
15 OC₃, and OC₄ (1.22), indicating that EC alone cannot account for the total mass of LV and MV
16 residuals. Including non-volatile OC (sum of OC₂ to OC₄) gave a better mass closure with the
17 total of LV and MV residuals. This further supports our initial postulation that the non-volatile
18 residuals which remained intact upon heating at 300°C in the VTDMA may contain a
19 significant amount of non-volatile OC. However, the total mass of EC, OC₂, OC₃, and OC₄ did
20 not explain all the mass of LV and MV residuals. A possible explanation could be that the
21 vaporizing temperatures of some OC₁ are close to the upper limit (310°C), hence they were not
22 completely vaporized in the heated tube and remained in the non-volatile residuals. The
23 presence of other refractory material and the assumption made about the density of LV and MV
24 are two other possible explanations.

25 Other possible errors for the closure could be related to the different heating environments in
26 the VTDMA and the OC/EC analyzer. In the OC/EC analyzer, OC was measured when the
27 samples were heated in the presence of a non-oxidative carrier gas (He). In the VTDMA,
28 aerosols were heated in air which contained O₂. Therefore, some “OC₂₋₄” that evaporated at
29 475°C or above in the OC/EC analyzer may have been oxidized at 300°C in the VTDMA.
30 Charring of organic matter could also occur (Philippin et al., 2004). Further study is needed to
31 quantify the effect of oxygen on the oxidation of OC in the VTDMA. The extrapolated
32 lognormal fitting of the size distribution of non-volatile particles can also cause errors if the

1 mode diameter of the fitting is beyond the VTDMA's range of measurements. While the
2 VTDMA measured the size distribution of particles between 10 nm and 400 nm in diameter,
3 the OC/EC analyzer took into account particles up to 2.5 μm in diameter. Yu et al. (2010)
4 reported three EC and OC modes between 400 nm and 10 μm in ambient aerosols in Guangzhou:
5 400 nm, 900 nm and 5 μm . The 400 nm mode accounted for 44% to 49% of the measured EC
6 but only 17% to 20% of the measured OC.

7

8 **4 Conclusions**

9 This study presents the first VTDMA measurements in a suburban area of Guangzhou in the
10 Pearl River Delta region, China during wintertime. The non-volatile material at 300°C in
11 VTDMA measurements was assumed to be EC and non-volatile OC. The LV particles,
12 representing non-volatile material externally mixed with the volatile material,, contributed to
13 less than 20% of the total particle number concentration at the sampling site. The diurnal
14 variations in the number and volume fractions of LV, MV and HV were much less obvious in
15 this study than in other studies (e.g. Rose et al., 2011; Cheng et al., 2012; Zhang et al., 2016)
16 likely because of the more stable atmosphere and poorer dilution of aged aerosols in winter.
17 The back trajectory analyses showed that the measured $\text{PM}_{2.5}$, EC and OC concentrations were
18 higher when the sampling site came under the influence of maritime and coastal air masses
19 originating from the east or northeast of the site. These observations were attributed to the high
20 pressure system on continent, the prevailing northerly wind and the enhanced pollution from
21 north China in winter. The long-range transport continental trajectories were often associated
22 with the cold front periods during which the dispersion of pollutants was promoted. The number
23 fractions of LV, MV and HV particles did not show much variations among the trajectory
24 clusters, likely because the air masses in all the clusters were transported at low altitudes (below
25 1500 m) for over 40 h. They were therefore well-aged upon arrival at the site.

26 While previous studies have indicated soot as a major component of the non-volatile residuals
27 at 300°C measured by the VTDMA (e.g. Philippin et al., 2004; Frey et al., 2008), Häkkinen et
28 al. (2012) and this work identified non-volatile organics as another possible component. The
29 diurnal variations in the LV fractions and the size of the MV residuals may be related to the
30 variation in the abundance of both EC and non-volatile OC, which evaporated at 475°C and
31 above in the OC/EC analyzer. The analyses of the diurnal variations in the LV fractions and the
32 *VFR* of MV particles, the latter of which reflects the change in size of the non-volatile material

1 in the MV particles, suggest that the increase in the non-volatile fractions and in the size in the
2 early afternoon may be related to the increase in non-volatile OC in addition to the effects of
3 coagulation and condensation. The mass closure analysis of EC and the total mass of LV and
4 MV residuals also indicated that EC alone cannot account for the mass of the non-volatile
5 residuals. The total mass of EC and and non-volatile OC gave a better closure with the total
6 mass of the LV and MV residuals, suggesting that the non-volatile OC may have contributed
7 to the non-volatile residuals in our VTDMA measurements.

8

9

10

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1 References

- 2 Andreae, M. O., Schmid, O., Yang, H., Chand, D., Zhen Yu, J., Zeng, L.-M., and Zhang, Y.-
3 H.: Optical properties and chemical composition of the atmospheric aerosol in urban
4 Guangzhou, China, *Atmos. Environ.*, 42, 6335-6350, 2008.
- 5 Birch, M. E. and Cary, R. A.: Elemental carbon-based method for monitoring occupational
6 exposures to particulate diesel exhaust, *Aerosol Sci. Technol.*, 25, 221-241, 1996.
- 7 Bond, T. C.: Spectral dependence of visible light absorption by carbonaceous particles emitted
8 from coal combustion, *Geophys. Res. Lett.*, 28, 4075-4078, 2001.
- 9 Bradsher, K.: Trucks power China's economy, at a suffocating cost.: available at:
10 <http://www.nytimes.com/2007/12/08/world/asia/08trucks.html>, 2007.
- 11 Brooks, B. J., Smith, M. H., Hill, M. K., and O'Dowd, C. D.: Size-differentiated volatility
12 analysis of internally mixed laboratory-generated aerosol, *J. Aerosol Sci*, 33, 555-579,
13 2002.
- 14 Chan, C. K. and Yao, X.: Air pollution in mega cities in China, *Atmos. Environ.*, 42, 1-42, 2008.
- 15 Chen, B., Du, K., Wang, Y., Chen, J., Zhao, J., Wang, K., Zhang, F., and Xu, L.: Emission and
16 transport of carbonaceous aerosols in urbanized coastal areas in China, *Aerosol Air Qual.*
17 *Res.*, 12, 371-378, 2012.
- 18 Chen, Y. and Bond, T. C.: Light absorption by organic carbon from wood combustion, *Atmos.*
19 *Chem. Phys.*, 10, 1773-1787, 2010.
- 20 Cheng, Y., He, K., Duan, F., Zheng, M., Du, Z., Ma, Y., and Tan, J.: Ambient organic carbon
21 to elemental carbon ratios: Influences of the measurement methods and implications,
22 *Atmos. Environ.*, 45, 2060-2066, 2011.
- 23 Cheng, Y. F., Eichler, H., Wiedensohler, A., Heintzenberg, J., Zhang, Y. H., Hu, M., Herrmann,
24 H., Zeng, L. M., Liu, S., Gnauk, T., Brüggemann, E., and He, L. Y.: Mixing state of
25 elemental carbon and non-light-absorbing aerosol components derived from in situ
26 particle optical properties at Xinken in Pearl River Delta of China, *J. Geophys. Res.-*
27 *Atmos.*, 111, D20204, 2006.
- 28 Cheng, Y. F., Su, H., Rose, D., Gunthe, S. S., Berghof, M., Wehner, B., Achtert, P., Nowak, A.,
29 Takegawa, N., Kondo, Y., Shiraiwa, M., Gong, Y. G., Shao, M., Hu, M., Zhu, T., Zhang,
30 Y. H., Carmichael, G. R., Wiedensohler, A., Andreae, M. O., and Pöschl, U.: Size-
31 resolved measurement of the mixing state of soot in the megacity Beijing, China: diurnal
32 cycle, aging and parameterization, *Atmos. Chem. Phys.*, 12, 4477-4491, 2012.
- 33 Chow, J. C., Watson, J. G., Lowenthal, D. H., Solomon, P. A., Magliano, K. L., Ziman, S. D.,
34 and Richards, L. W.: PM10 and PM2.5 compositions in California's San Joaquin Valley,
35 *Aerosol Sci. Technol.*, 18, 105-128, 1993.
- 36 Chow, J. C., Watson, J. G., Lu, Z., Lowenthal, D. H., Frazier, C. A., Solomon, P. A., Thuillier,
37 R. H., and Magliano, K.: Descriptive analysis of PM2.5 and PM10 at regionally
38 representative locations during SJVAQS/AUSPEX, *Atmos. Environ.*, 30, 2079-2112,
39 1996.
- 40 Chow, J. C., Yu, J. Z., Watson, J. G., Hang Ho, S. S., Bohannon, T. L., Hays, M. D., and Fung,
41 K. K.: The application of thermal methods for determining chemical composition of
42 carbonaceous aerosols: A review, *J. Environ. Sci. Heal. A*, 42, 1521-1541, 2007.

- 1 Donahue, N. M., Robinson, A. L., and Pandis, S. N.: Atmospheric organic particulate matter:
2 From smoke to secondary organic aerosol, *Atmos. Environ.*, 43, 94-106, 2009.
- 3 Frey, A., Rose, D., Wehner, B., Müller, T., Cheng, Y., Wiedensohler, A., and Virkkula, A.:
4 Application of the Volatility-TDMA Technique to Determine the Number Size
5 Distribution and Mass Concentration of Less Volatile Particles, *Aerosol Sci. Technol.*,
6 42, 817-828, 2008.
- 7 Fuller, K. A., Malm, W. C., and Kreidenweis, S. M.: Effects of mixing on extinction by
8 carbonaceous particles, *J. Geophys. Res.-Atmos.*, 104, 15941-15954, 1999.
- 9 Ghazi, R. and Olfert, J. S.: Coating Mass Dependence of Soot Aggregate Restructuring due to
10 Coatings of Oleic Acid and Dioctyl Sebacate, *Aerosol Sci. Technol.*, 47, 192-200, 2012.
- 11 Gnauk, T., Müller, K., van Pinxteren, D., He, L.-Y., Niu, Y., Hu, M., and Herrmann, H.: Size-
12 segregated particulate chemical composition in Xinken, Pearl River Delta, China:
13 OC/EC and organic compounds, *Atmos. Environ.*, 42, 6296-6309, 2008.
- 14 Gu, J., Du, S., Han, D., Hou, L., Yi, J., Xu, J., Liu, G., Han, B., Yang, G., and Bai, Z.-P.: Major
15 chemical compositions, possible sources, and mass closure analysis of PM_{2.5} in Jinan,
16 China, *Air Qual. Atmos. Heal.*, 7, 251-262, 2014.
- 17 Häkkinen, S. A. K., Äijälä, M., Lehtipalo, K., Junninen, H., Backman, J., Virkkula, A.,
18 Nieminen, T., Vestenius, M., Hakola, H., Ehn, M., Worsnop, D. R., Kulmala, M., Petäjä,
19 T., and Riipinen, I.: Long-term volatility measurements of submicron atmospheric
20 aerosol in Hyytiälä, Finland, *Atmos. Chem. Phys.*, 12, 10771-10786, 2012.
- 21 Hansen, A. D. A., Rosen, H., and Novakov, T.: The Aethalometer - An Instrument for the Real-
22 Time Measurement of Optical-Absorption by Aerosol-Particles *Sci. Total Environ.*, 36,
23 191-196, 1984.
- 24 Hitzenberger, R., Jennings, S. G., Larson, S. M., Dillner, A., Cachier, H., Galambos, Z., Rouc,
25 A., and Spain, T. G.: Intercomparison of measurement methods for black carbon
26 aerosols, *Atmos. Environ.*, 33, 2823-2833, 1999.
- 27 Horvath, H.: Atmospheric light absorption—A review, *Atmos. Environ.*, 27, 293-317, 1993.
- 28 Huffman, J. A., Docherty, K. S., Aiken, A. C., Cubison, M. J., Ulbrich, I. M., DeCarlo, P. F.,
29 Sueper, D., Jayne, J. T., Worsnop, D. R., Ziemann, P. J., and Jimenez, J. L.: Chemically-
30 resolved aerosol volatility measurements from two megacity field studies, *Atmos. Chem.*
31 *Phys.*, 9, 7161-7182, 2009.
- 32 Japar, S. M., Brachaczek, W. W., Gorse Jr, R. A., Norbeck, J. M., and Pierson, W. R.: The
33 contribution of elemental carbon to the optical properties of rural atmospheric aerosols,
34 *Atmos. Environ.*, 20, 1281-1289, 1986.
- 35 Kalberer, M., Paulsen, D., Sax, M., Steinbacher, M., Dommen, J., Prevot, A. S. H., Fisseha, R.,
36 Weingartner, E., Frankevich, V., Zenobi, R., and Baltensperger, U.: *Science*, 303, 1659,
37 2004.
- 38 Kirchstetter, T. W., Novakov, T., and Hobbs, P. V.: Evidence that the spectral dependence of
39 light absorption by aerosols is affected by organic carbon, *J. Geophys. Res.-Atmos.*, 109,
40 D21208, 2004.
- 41 Lavanchy, V. M. H., Gaggeler, H. W., Nyeki, S., and Baltensperger, U.: Elemental carbon (EC)
42 and black carbon (BC) measurements with a thermal method and an aethalometer at the
43 high-alpine research station Jungfraujoch, *Atmos. Environ.*, 33, 2759-2769, 1999.

- 1 Lee, B. P., Li, Y. J., Yu, J. Z., Louie, P. K. K., and Chan, C. K.: Physical and chemical
2 characterization of ambient aerosol by HR-ToF-AMS at a suburban site in Hong Kong
3 during springtime 2011, *J. Geophys. Res.-Atmos.*, 118, 8625-8639, 2013.
- 4 Levy, M. E., Zhang, R., Zheng, J., Tan, H., Wang, Y., Molina, L. T., Takahama, S., Russell, L.
5 M., and Li, G.: Measurements of submicron aerosols at the California–Mexico border
6 during the Cal–Mex 2010 field campaign, *Atmos. Environ.*, 88, 308-319, 2014.
- 7 Liousse, C., Cachier, H., and Jennings, S. G.: Optical and thermal measurements of black
8 carbon aerosol content in different environments: Variation of the specific attenuation
9 cross-section, σ , *Atmos. Environ.*, 27, 1203-1211, 1993.
- 10 Lo, J. C. F., Lau, A. K. H., Fung, J. C. H., and Chen, F.: Investigation of enhanced cross-city
11 transport and trapping of air pollutants by coastal and urban land-sea breeze circulations,
12 *J. Geophys. Res.-Atmos.*, 111, D14104, 2006.
- 13 Murphy, B. N., Donahue, N. M., Robinson, A. L., and Pandis, S. N.: A naming convention for
14 atmospheric organic aerosol, *Atmos. Chem. Phys.*, 14, 5825-5839, 2014.
- 15 Novakov, T., Ramanathan, V., Hansen, J. E., Kirchstetter, T. W., Sato, M., Sinton, J. E., and
16 Sathaye, J. A.: Large historical changes of fossil-fuel black carbon aerosols, *Geophys.*
17 *Res. Lett.*, 30, 4, 2003.
- 18 Onasch, T. B., Trimborn, A., Fortner, E. C., Jayne, J. T., Kok, G. L., Williams, L. R., Davidovits,
19 P., and Worsnop, D. R.: Soot Particle Aerosol Mass Spectrometer: Development,
20 Validation, and Initial Application, *Aerosol Sci. Technol.*, 46, 804-817, 2012.
- 21 Penner, J. E. and Novakov, T.: Carbonaceous particles in the atmosphere: A historical
22 perspective to the Fifth International Conference on Carbonaceous Particles in the
23 Atmosphere, *J. Geophys. Res.-Atmos.*, 101, 19373-19378, 1996.
- 24 Petzold, A. and Schönlinner, M.: Multi-angle absorption photometry—a new method for the
25 measurement of aerosol light absorption and atmospheric black carbon, *J. Aerosol Sci.*,
26 35, 421-441, 2004.
- 27 Philippin, S., Wiedensohler, A., and Stratmann, F.: Measurements of non-volatile fractions of
28 pollution aerosols with an eight-tube volatility tandem differential mobility analyzer
29 (VTDMA-8), *J. Aerosol Sci.*, 35, 185-203, 2004.
- 30 Pinnick, R., Jennings, S., and Fernandez, G.: Volatility of aerosols in the arid southwestern
31 United States, *J. Atmos. Sci.*, 44, 562-576, 1987.
- 32 Putaud, J. P., Van Dingenen, R., Alastuey, A., Bauer, H., Birmili, W., Cyrys, J., Flentje, H.,
33 Fuzzi, S., Gehrig, R., Hansson, H. C., Harrison, R. M., Herrmann, H., Hitzenberger, R.,
34 Hüglin, C., Jones, A. M., Kasper-Giebl, A., Kiss, G., Koussa, A., Kuhlbusch, T. A. J.,
35 Löschau, G., Maenhaut, W., Molnar, A., Moreno, T., Pekkanen, J., Perrino, C., Pitz, M.,
36 Puxbaum, H., Querol, X., Rodriguez, S., Salma, I., Schwarz, J., Smolik, J., Schneider,
37 J., Spindler, G., ten Brink, H., Tursic, J., Viana, M., Wiedensohler, A., and Raes, F.: A
38 European aerosol phenomenology – 3: Physical and chemical characteristics of
39 particulate matter from 60 rural, urban, and kerbside sites across Europe, *Atmos.*
40 *Environ.*, 44, 1308-1320, 2010.
- 41 Rader, D. J. and McMurry, P. H.: Application of the tandem differential mobility analyzer to
42 studies of droplet growth or evaporation, *J. Aerosol Sci.*, 17, 771-787, 1986.

- 1 Robinson, A. L., Donahue, N. M., Shrivastava, M. K., Weitkamp, E. A., Sage, A. M., Grieshop,
2 A. P., Lane, T. E., Pierce, J. R., and Pandis, S. N.: Rethinking organic aerosols:
3 Semivolatile emissions and photochemical aging, *Science*, 315, 1259-1262, 2007.
- 4 Rolph, G. D.: Real-time Environmental Applications and Display sYstem (READY) Website
5 (<http://www.ready.noaa.gov>). NOAA Air Resources Laboratory, College Park, MD.,
6 2016.
- 7 Rose, D., Wehner, B., Ketzler, M., Engler, C., Voigtländer, J., Tuch, T., and Wiedensohler, A.:
8 Atmospheric number size distributions of soot particles and estimation of emission
9 factors, *Atmos. Chem. Phys.*, 6, 1021-1031, 2006.
- 10 Rose, D., Gunthe, S. S., Su, H., Garland, R. M., Yang, H., Berghof, M., Cheng, Y. F., Wehner,
11 B., Achtert, P., Nowak, A., Wiedensohler, A., Takegawa, N., Kondo, Y., Hu, M., Zhang,
12 Y., Andreae, M. O., and Pöschl, U.: Cloud condensation nuclei in polluted air and
13 biomass burning smoke near the mega-city Guangzhou, China – Part 2: Size-resolved
14 aerosol chemical composition, diurnal cycles, and externally mixed weakly CCN-active
15 soot particles, *Atmos. Chem. Phys.*, 11, 2817-2836, 2011.
- 16 Rosen, H., Hansen, A. D. A., Gundel, L., and Novakov, T.: Identification of the optically
17 absorbing component in urban aerosols, *Appl. Opt.*, 17, 3859-3861, 1978.
- 18 Schauer, J. J., Mader, B. T., Deminter, J. T., Heidemann, G., Bae, M. S., Seinfeld, J. H., Flagan,
19 R. C., Cary, R. A., Smith, D., Huebert, B. J., Bertram, T., Howell, S., Kline, J. T., Quinn,
20 P., Bates, T., Turpin, B., Lim, H. J., Yu, J. Z., Yang, H., and Keywood, M. D.: ACE-
21 Asia intercomparison of a thermal-optical method for the determination of particle-
22 phase organic and elemental carbon, *Environ. Sci. Technol.*, 37, 993-1001, 2003.
- 23 Smith, J. N., Dunn, M. J., VanReken, T. M., Iida, K., Stolzenburg, M. R., McMurry, P. H., and
24 Huey, L. G.: Chemical composition of atmospheric nanoparticles formed from
25 nucleation in Tecamac, Mexico: Evidence for an important role for organic species in
26 nanoparticle growth, *Geophys. Res. Lett.*, 35, L04808, 2008.
- 27 Stein, A. F., Draxler, R. R., Rolph, G. D., Stunder, B. J. B., Cohen, M. D., and Ngan, F.:
28 NOAA's HYSPLIT Atmospheric Transport and Dispersion Modeling System, *B. Am.*
29 *Meteorol. Soc.*, 96, 2059-2077, 2015.
- 30 Stephens, M., Turner, N., and Sandberg, J.: Particle identification by laser-induced
31 incandescence in a solid-state laser cavity, *Appl. Opt.*, 42, 3726-3736, 2003.
- 32 Tan, H. B., Yin, Y., Gu, X. S., Li, F., Chan, P. W., Xu, H. B., Deng, X. J., and Wan, Q. L.: An
33 observational study of the hygroscopic properties of aerosols over the Pearl River Delta
34 region, *Atmos. Environ.*, 77, 817-826, 2013a.
- 35 Tan, H. B., Xu, H. B., Wan, Q. L., Li, F., Deng, X. J., Chan, P. W., Xia, D., and Yin, Y.: Design
36 and Application of an Unattended Multifunctional H-TDMA System, *J. Atmos. Ocean*
37 *Tech.*, 30, 1136-1148, 2013b.
- 38 Tao, J., Zhang, L., Ho, K., Zhang, R., Lin, Z., Zhang, Z., Lin, M., Cao, J., Liu, S., and Wang,
39 G.: Impact of PM_{2.5} chemical compositions on aerosol light scattering in Guangzhou
40 — the largest megacity in South China, *Atmos. Res.*, 135–136, 48-58, 2014.
- 41 Turpin, B. J., Cary, R. A., and Huntzicker, J. J.: An In Situ, Time-Resolved Analyzer for
42 Aerosol Organic and Elemental Carbon, *Aerosol Sci. Technol.*, 12, 161-171, 1990.
- 43 Twomey, S.: On the composition of cloud nuclei in the northeastern United States, *J. Rech.*
44 *Atmos.*, 3, 281-285, 1968.

- 1 Villani, P., Picard, D., Marchand*, N., and Laj, P.: Design and Validation of a 6-Volatility
2 Tandem Differential Mobility Analyzer (VTDMA), *Aerosol Sci. Technol.*, 41, 898-906,
3 2007.
- 4 Virkkula, A., Ahlquist, N. C., Covert, D. S., Arnott, W. P., Sheridan, P. J., Quinn, P. K., and
5 Coffman, D. J.: Modification, Calibration and a Field Test of an Instrument for
6 Measuring Light Absorption by Particles, *Aerosol Sci. Technol.*, 39, 68-83, 2005.
- 7 Wehner, B., Philippin, S., Wiedensohler, A., Scheer, V., and Vogt, R.: Variability of non-
8 volatile fractions of atmospheric aerosol particles with traffic influence, *Atmos.*
9 *Environ.*, 38, 6081-6090, 2004.
- 10 Wehner, B., Berghof, M., Cheng, Y. F., Achtert, P., Birmili, W., Nowak, A., Wiedensohler, A.,
11 Garland, R. M., Pöschl, U., Hu, M., and Zhu, T.: Mixing state of nonvolatile aerosol
12 particle fractions and comparison with light absorption in the polluted Beijing region, *J.*
13 *Geophys. Res.-Atmos.*, 114, D00G17, 2009.
- 14 Wu, C., Ng, W. M., Huang, J. X., Wu, D., and Yu, J. Z.: Determination of Elemental and
15 Organic Carbon in PM_{2.5} in the Pearl River Delta Region: Inter-Instrument (Sunset vs.
16 DRI Model 2001 Thermal/Optical Carbon Analyzer) and Inter-Protocol Comparisons
17 (IMPROVE vs. ACE-Asia Protocol), *Aerosol Sci. Technol.*, 46, 610-621, 2012.
- 18 Wu, D., Bi, X. Y., Deng, X. J., Li, F., Tan, H. B., Liao, G. L., and Huang, J.: Effect of
19 atmospheric haze on the deterioration of visibility over the Pearl River Delta, *Acta*
20 *Meteorol. Sin.*, 21, 215-223, 2007.
- 21 Yu, H., Wu, C., Wu, D., and Yu, J. Z.: Size distributions of elemental carbon and its
22 contribution to light extinction in urban and rural locations in the pearl river delta region,
23 China, *Atmos. Chem. Phys.*, 10, 5107-5119, 2010.
- 24 Yue, D. L., Zhong, L. J., Zhang, T., J., S., L., Y., Q., Y. S., Y., Z., and Zeng, L. M.: Particle
25 Growth and Variation of Cloud Condensation Nucleus Activity on Polluted Days with
26 New Particle Formation: A Case Study for Regional Air Pollution in the PRD Region,
27 China, *Aerosol Air Qual. Res.*, 16, 323-335, 2016.
- 28 Zhang, Q., Stanier, C. O., Canagaratna, M. R., Jayne, J. T., Worsnop, D. R., Pandis, S. N., and
29 Jimenez, J. L.: Insights into the Chemistry of New Particle Formation and Growth
30 Events in Pittsburgh Based on Aerosol Mass Spectrometry, *Environ. Sci. Technol.*, 38,
31 4797-4809, 2004.
- 32 Zhang, S. L., Ma, N., Kecorius, S., Wang, P. C., Hu, M., Wang, Z. B., Größ, J., Wu, Z. J., and
33 Wiedensohler, A.: Mixing state of atmospheric particles over the North China Plain,
34 *Atmos. Environ.*, 125, Part A, 152-164, 2016.
- 35 Zhang, Y., Wang, X., Li, G., Yang, W., Huang, Z., Zhang, Z., Huang, X., Deng, W., Liu, T.,
36 Huang, Z., and Zhang, Z.: Emission factors of fine particles, carbonaceous aerosols and
37 traces gases from road vehicles: Recent tests in an urban tunnel in the Pearl River Delta,
38 China, *Atmos. Environ.*, 122, 876-884, 2015.
- 39 Zhang, Y. M., Zhang, X. Y., Sun, J. Y., Lin, W. L., Gong, S. L., Shen, X. J., and Yang, S.:
40 Characterization of new particle and secondary aerosol formation during summertime
41 in Beijing, China, *Tellus B*, 63, 382-394, 2011.

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1 Table 1. Temperature (T) and residence time (RT) protocol of the semi-continuous Sunset
 2 OC/EC analyzer (Wu et al., 2012)

Carbon Fraction	Carrier Gas	T (°C)	RT (s)
OC ₁	He	310	80
OC ₂		475	60
OC ₃		615	60
OC ₄		870	90
EC ₁	He and 2% O ₂	550	45
EC ₂		625	45
EC ₃		700	45
EC ₄		775	45
EC ₅		850	45
EC ₆		870	45

3

4

1 Table 2. Summary of average number and volume fractions in VTDMA measurements at 300°C.

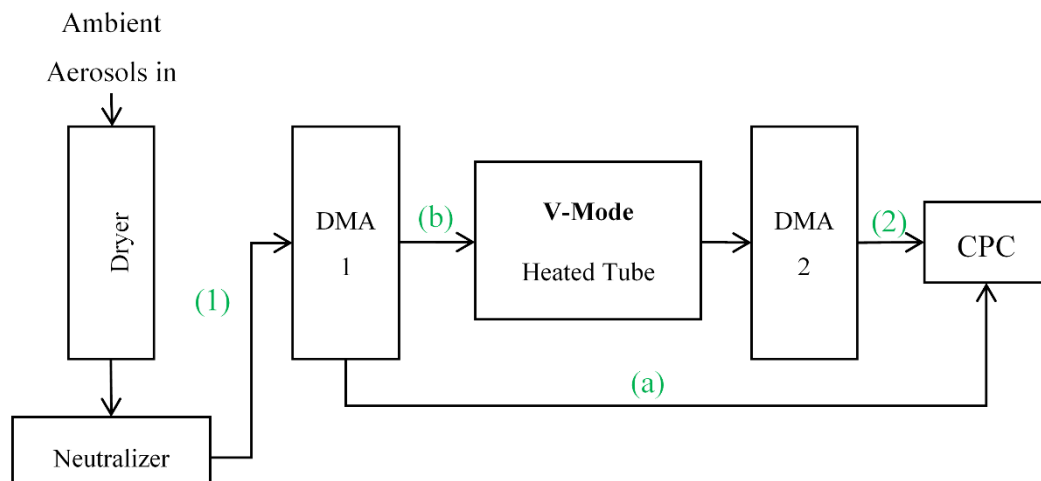
Diameter (nm)	40	80	110	150	200	300
Number fraction						
CV	0.380 ± 0.153	0.174 ± 0.097	0.188 ± 0.081	0.167 ± 0.074	0.153 ± 0.070	0.141 ± 0.065
HV	0.255 ± 0.097	0.198 ± 0.052	0.165 ± 0.055	0.163 ± 0.064	0.178 ± 0.081	0.214 ± 0.097
MV	0.314 ± 0.097	0.513 ± 0.089	0.515 ± 0.098	0.530 ± 0.105	0.523 ± 0.116	0.497 ± 0.125
LV	0.051 ± 0.026	0.113 ± 0.040	0.132 ± 0.041	0.140 ± 0.041	0.146 ± 0.044	0.148 ± 0.047
Volume fraction						
VM	0.503 ± 0.131	0.600 ± 0.082	0.580 ± 0.073	0.590 ± 0.066	0.602 ± 0.064	0.627 ± 0.064
CV	0.361 ± 0.168	0.163 ± 0.105	0.166 ± 0.098	0.148 ± 0.086	0.134 ± 0.080	0.127 ± 0.073
HV	0.014 ± 0.005	0.011 ± 0.003	0.008 ± 0.002	0.007 ± 0.003	0.007 ± 0.003	0.007 ± 0.003
MV	0.070 ± 0.025	0.112 ± 0.024	0.112 ± 0.025	0.115 ± 0.026	0.109 ± 0.027	0.091 ± 0.025
LV	0.052 ± 0.026	0.114 ± 0.040	0.134 ± 0.044	0.140 ± 0.042	0.148 ± 0.048	0.148 ± 0.047

2

1 Table 3. Summary of concentrations of PM_{2.5}, OC, EC and the ratio of OC to EC (OC/EC) in
 2 the five clusters.

	Cluster				
	Coastal	Maritime		Continental	
	1	2	3	4	5
Origin (to the site)	NE	NE/E	NE	NW	NW
PM _{2.5} (µg m ⁻³)	58.5 ± 24.4	58.9 ± 30.9	47.5 ± 28.4	33.9 ± 15.9	33.8 ± 19.3
OC (µg m ⁻³)	10.8 ± 6.01	10.84 ± 7.22	10.13 ± 6.89	5.51 ± 3.3	7.32 ± 2.75
EC (µg m ⁻³)	4.38 ± 2.97	4.98 ± 4.21	3.43 ± 3.12	1.8 ± 0.98	2.46 ± 0.59
OC/EC	2.83 ± 1.05	2.62 ± 1.03	3.65 ± 1.6	3.18 ± 1.26	2.94 ± 0.73

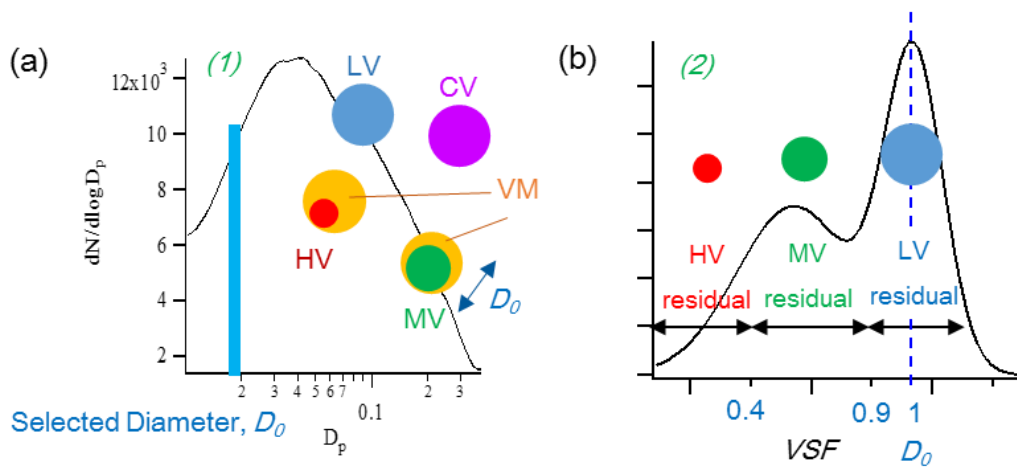
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2 Fig. 1. A Schematic diagram of the volatility tandem differential mobility analyzer (VTDMA).

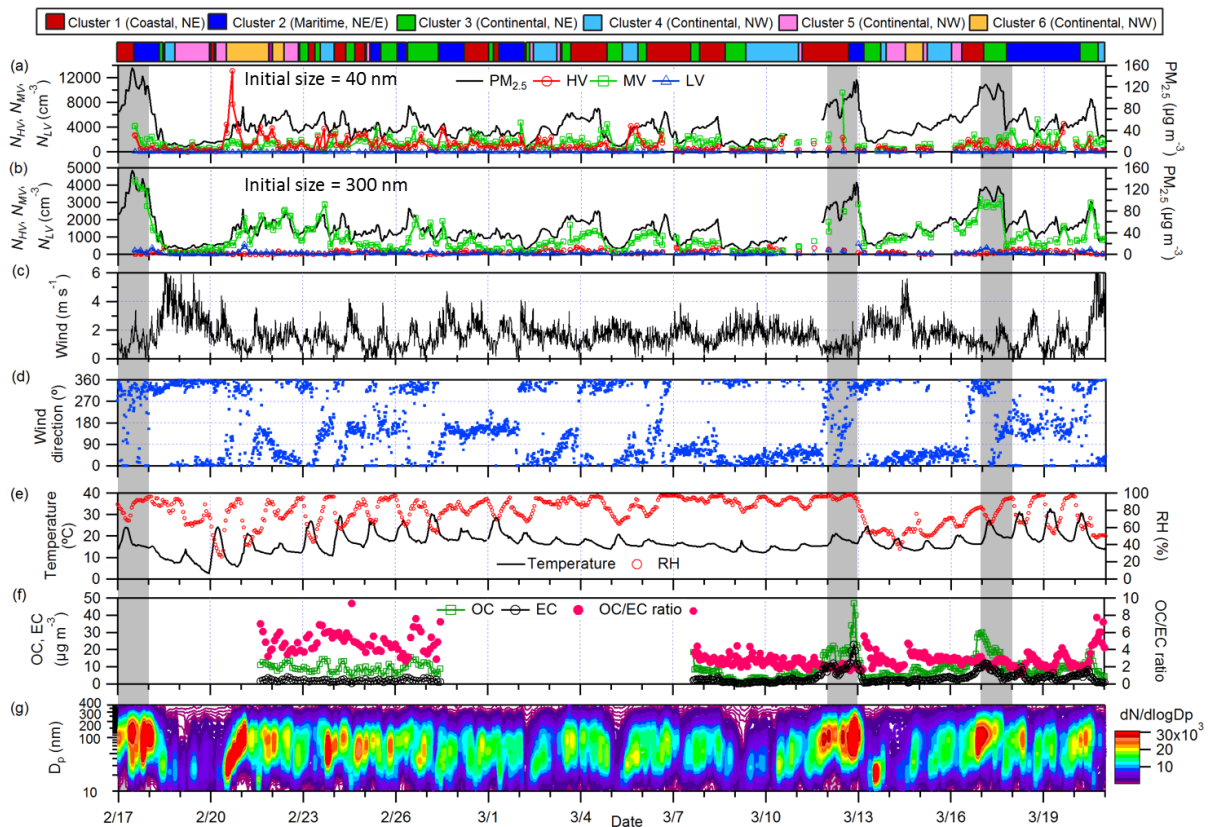
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2 Fig. 2. Examples of particle size distributions of (a) ambient aerosols before entering DMA₁
 3 and (b) residuals of the size-selected particles (D_0) after heating. at 300°C. The distributions in
 4 Fig. 2a and 2b correspond to (1) and (2) in Figure 1 respectively. Residuals are divided into
 5 three groups—LV (blue), MV (green) and HV (red)—based on their *VSF*. CV (purple) and VM
 6 (orange) are vaporized and hence not measured as residuals. VM appears as coating for
 7 illustration purposes only. It does not necessarily reflect the morphology of the particles.

8



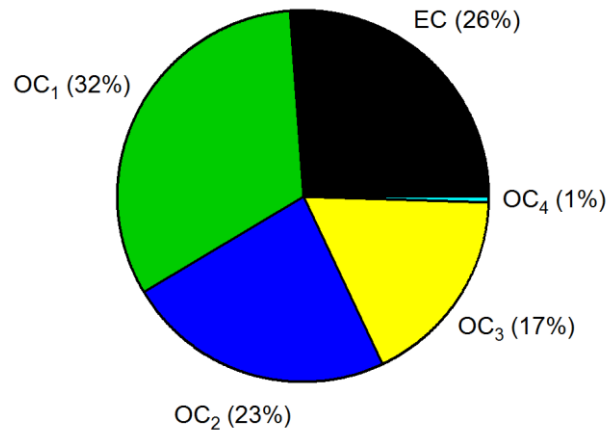
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2 Fig. 3. Temporal variation of number concentrations of HV, MV and LV in 40 nm
 3 particles, $PM_{2.5}$, major meteorological parameters, OC and EC concentrations, OC-to-EC ratio
 4 and particle number size distributions in the campaign. Air mass clusters are depicted at the top
 5 and the shaded areas indicate days with daily-averaged $PM_{2.5}$ concentrations exceeding $95 \mu g$
 6 m^{-3} .

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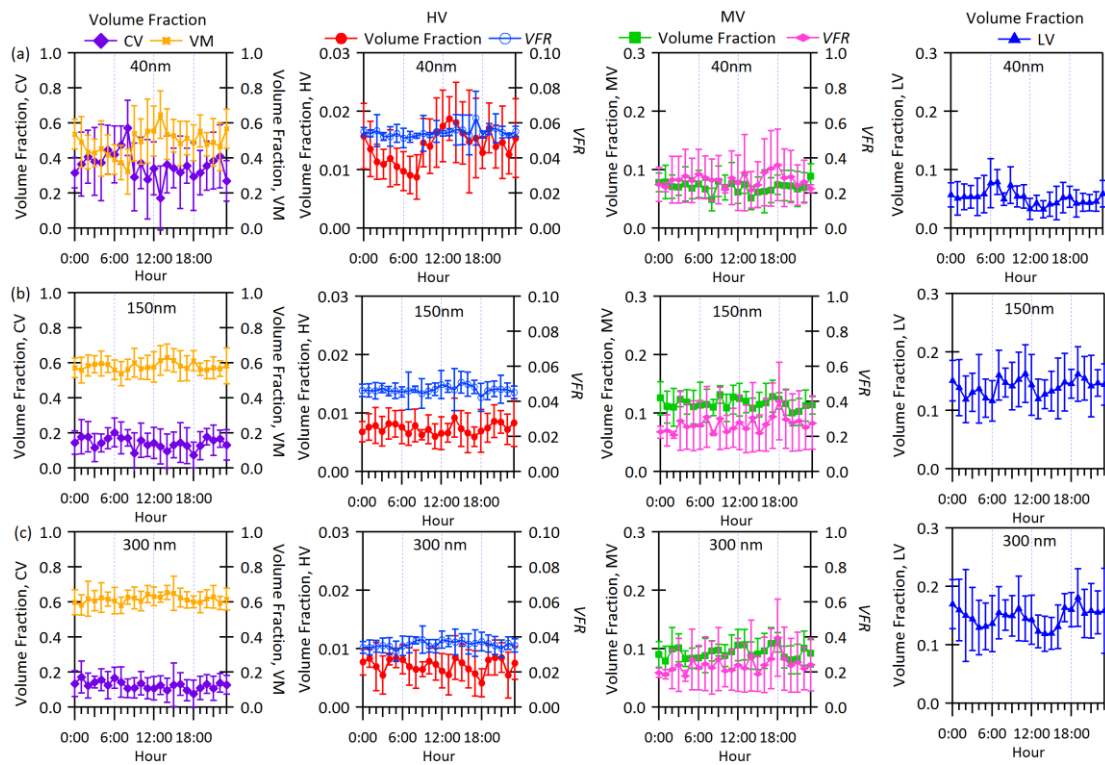
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2 Fig. 4. Average mass fractions of EC, OC₁, OC₂, OC₃ and OC₄ in PM_{2.5}.

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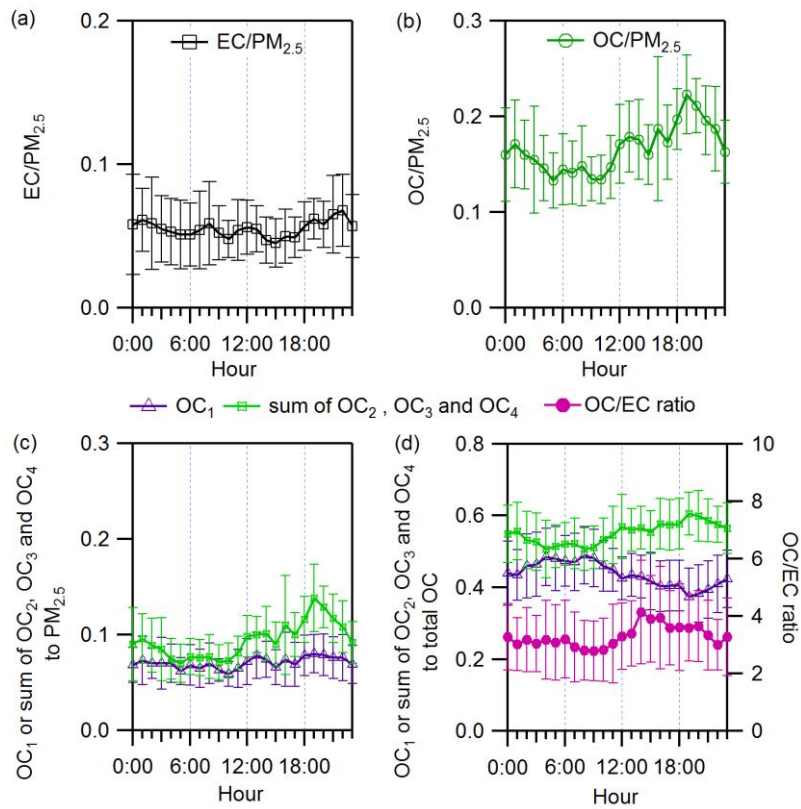


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2 Fig. 5. Diurnal variations in the volume fractions of (columns from left to right) CV, VM, HV
 3 residuals, MV residuals and LV residuals in (a) 40 nm, (b) 150 nm and (c) 300 nm particles in
 4 February and March 2014. Diurnal variations in the volume fraction remaining (*VFR*) of HV
 5 and MV particles are plotted on the right axis. Error bars represent one standard deviation.

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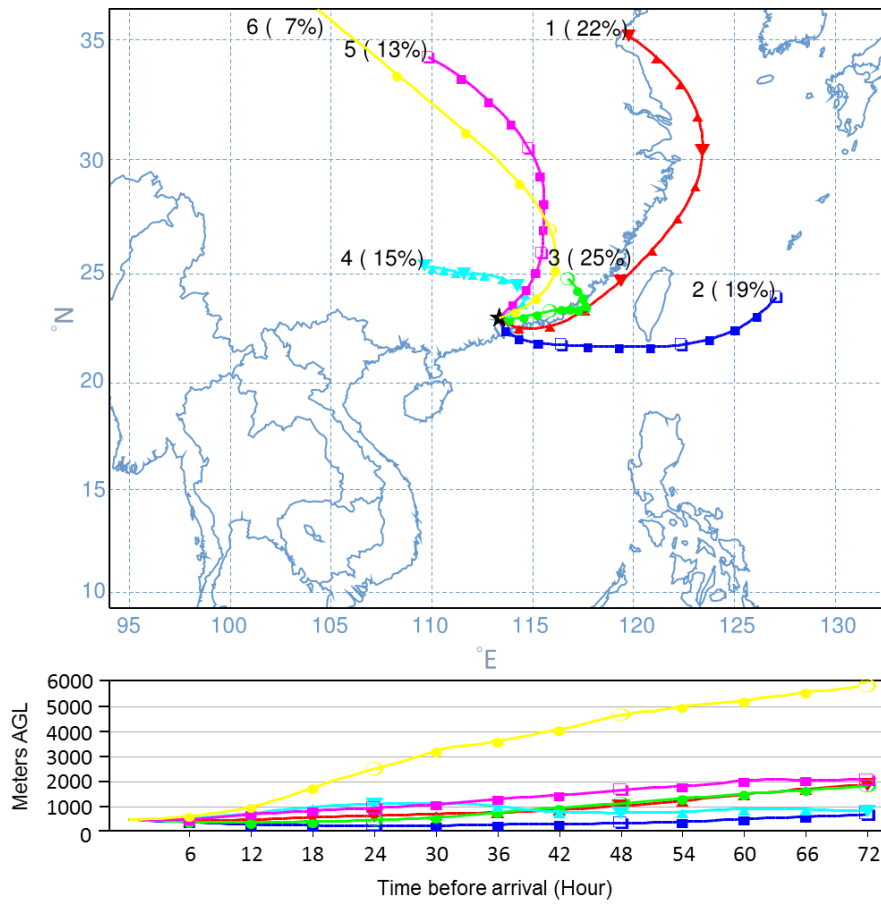


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2 Fig. 6. Diurnal variations in the mass fractions of EC, OC, OC₁ and the sum of OC₂, OC₃ and
 3 OC₄ in PM_{2.5}, the ratio of OC to EC, mass fractions of OC₁ and the sum of OC₂, OC₃ and OC₄
 4 to total OC in February and March 2014. Error bars represent one standard deviation.

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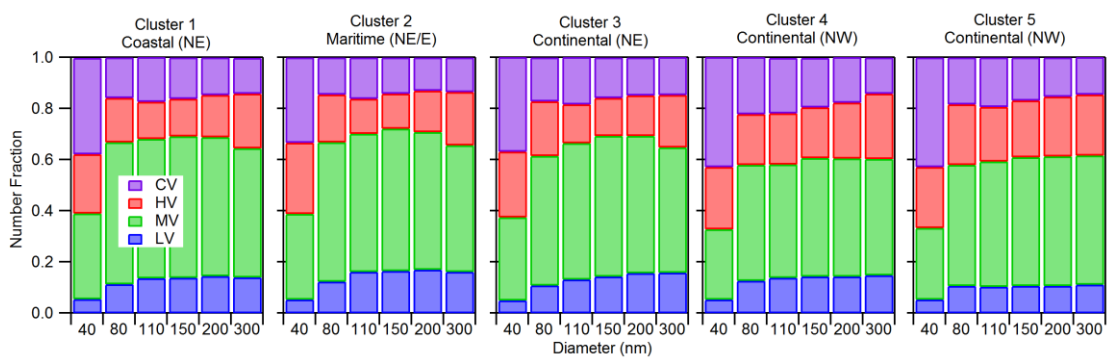
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2 Fig. 7. Mean back trajectories of the six types of air masses arriving at the sampling site.

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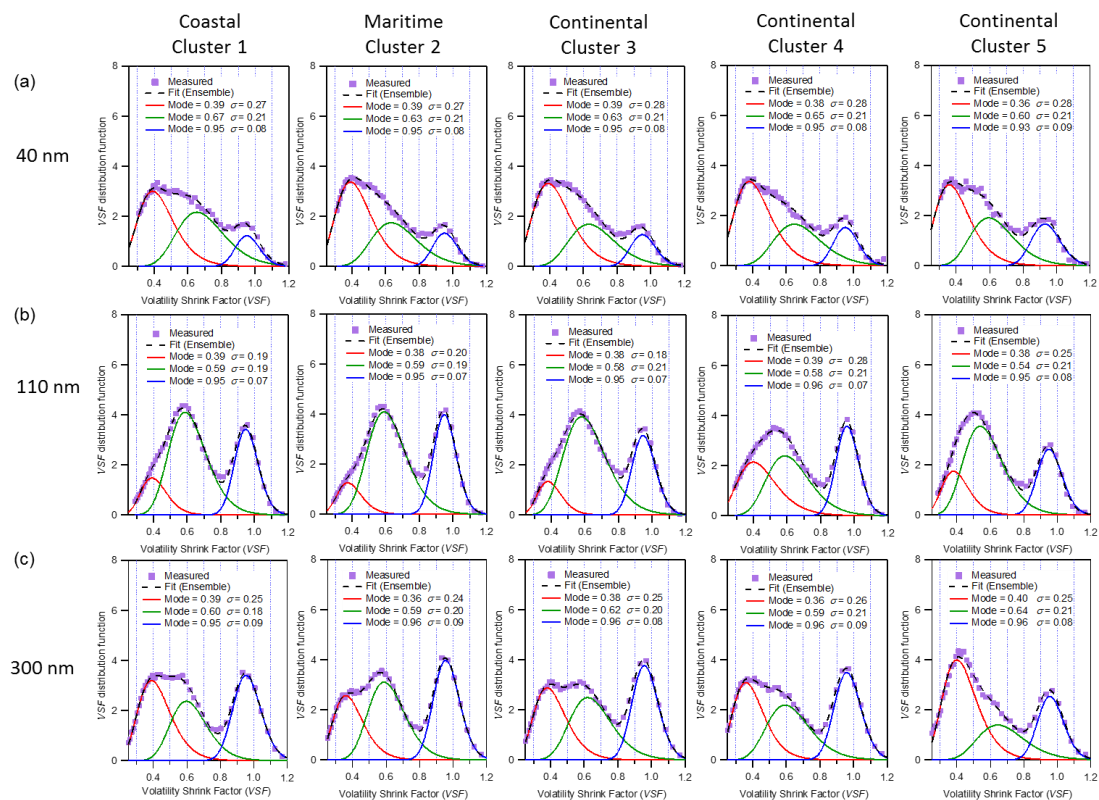
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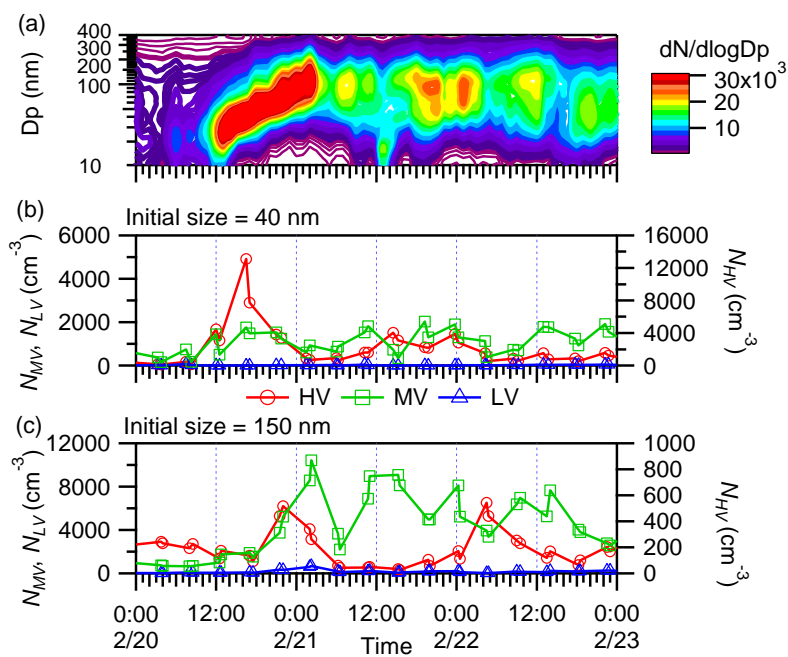


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 2 Fig. 8. Average number fractions of CV, HV, MV and LV in clusters 1 to 5 at different selected
 3 diameters.

4

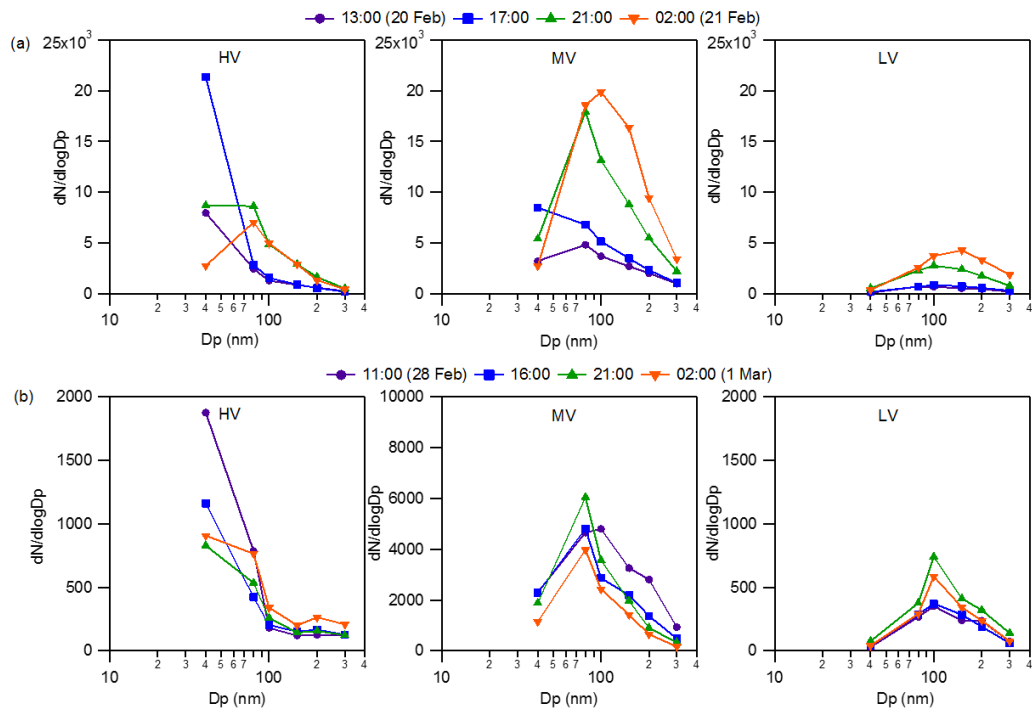


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 2 Fig. 9. Volatility shrink factor (*VSF*) distribution function in different clusters. Solid and dotted
 3 lines are the peaks fitted with log-normal function and the ensemble distributions, respectively.
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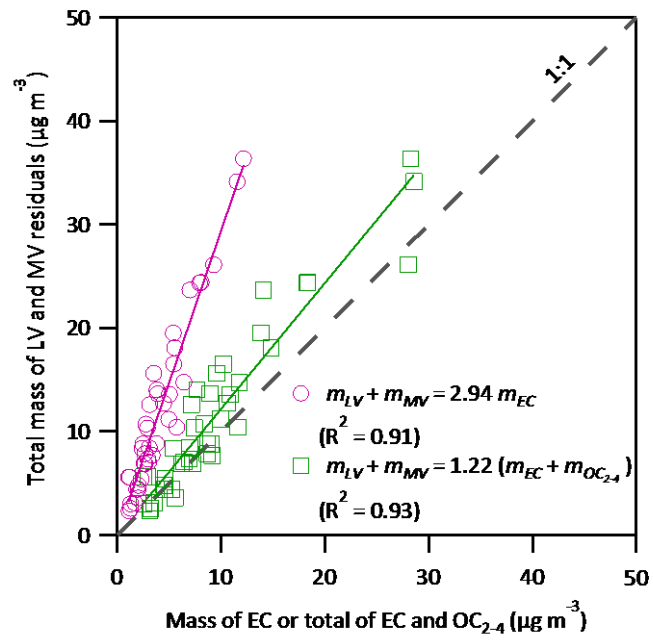
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 2 Fig. 10. Time series of (a) particle number size distribution, (b) number concentrations of HV,
 3 MV and LV in 40 nm particles and (c) number concentrations of HV, MV and LV in 150 nm
 4 particles during a new particle event day on 20 Feb 2014.

5



1
 2 Fig. 11. Particle number size distribution of (columns from left to right) HV, MV and LV
 3 particles (a) during a new particle formation event at 13:00, 17:00, 21:00 on 20 Feb and 02:00
 4 on 21 Feb 2014 and (b) during non-event days at 11:00, 16:00, 21:00 on 28 Feb and 02:00 on
 5 1 Mar 2014.

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Fig. 12. A closure analysis of the total mass of LV and MV residuals from VTDMA at 300°C and measured mass of EC or total of EC and OC₂₋₄ from the OC/EC analyzer.