Answers to the reviewers

Referee 1

The 2010 cloud experiment at Mt. Schmucke (Germany) was performed with much effort and heavily instrumented with regard to cloudwater collection. This manuscripts describes a part of the data set (8 events, inorganic ions and DOC) and highlights three topics: The correlation between cloud concentrations and the liquid water content, the scavenging of substances by the cloud, and the cloud-droplet-size dependent concentrations of substances. Some parameters were measured with high time resolution, as two AMS systems were employed in the field. The results are manifold and described in great detail in this manuscript. The evaluation and interpretation of data was performed carefully yet clearly, results are embedded into the respective literature results. Overall, this is an interesting and precious data set. Although not breathtaking, this manuscript should be published in ACP. Some comments are intended to help the authors clarify some issues before final publication in ACP:

We thank the reviewer for his or her effort and thoughtful remarks and will reply to all comments in the following.

page 24320, line 26: Schneider (2015) is not a valid reference. Overall in section "2.2 Interstitial and residual particle sampling", there is no information about how efficient the separation between cloud droplets and interstitial aerosol is.

We removed the reference to the not yet submitted work and modified the sentence as follows:
"Details of the AMS measurements will be given in a forthcoming companion paper of this special issue (Schneider et al., in preparation)(Schneider et al., 2015)."

With regard to the separation between cloud droplets and interstitial aerosol: This is given in the Schwarzenböck et al., 2000, reference, where we refer the reader to for more details: Collection efficiency increases from 10% to 90% from approx. 4 to 6 µm.

page 24321, line 5: Poulain et al. (2015) is not a valid reference. The information of the monitor "for continuous (1 h time resolution) determination of water-soluble inorganic trace gases and particulate ions" is too sparse.

We removed this reference as well and added additional information on the MARGA system as follows:
"A full description of the instrumental setup will be given in a forthcoming companion paper of this special issue (Poulain et al., in preparation). In brief, a commercial monitor for aerosols and gases (MARGA 1S, Metrohm Applikon, The Netherlands) was used for continuous (1 h time resolution) determination of water-soluble inorganic trace gases and particulate ions. The MARGA operated at a sampling rate of 1 m^3 h^{-1} and consisted of a PM_{10} inlet, a wet rotating denuder absorbing water-soluble gases into deionised water (10 ppm H_2O_2 added as biocide), a steam jet aerosol collector to grow and collect aerosol particles, and 2 ion chromatography systems for online cation and anion analysis."

page 24322, lines 13-19: What about the stability of H2O2 and S(IV) during cloud water collection? These species may react with each other faster than within one hour, which is the time resolution of the cloudwater collection.

We agree that this can be a potential artefact, which is, however, difficult to address because it is a function of reactant concentrations and cloud water pH over the time of sampling. Observed concentrations of reactive species at the point of sample preservation can thus be more or less biased depending on actual cloud water conditions. To clarify this, we added the following sentence into section 2.5:

"Concentrations of reactive compounds at the time of sample preservation can be biased due to reactions during the collection period. The extent of such artefacts will depend on reactant concentrations and cloud water pH and cannot easily be estimated."

page 24323, line 19: Why three sites, didn't you talk about two sites only so far?

We haven't introduced the third site, because in this manuscript we don't present data from it. We understand this can be confusing, however, and have therefore changed section 2.3 from "Upwind site aerosol sampling" to "Valley sites aerosol sampling" and modified its beginning as follows:

"Next to the Schmücke in-cloud site, two more valley sites upwind and downwind of the Schmücke were installed during HCCT-2010 to characterise air masses before and after their passage through the clouds."

In addition, we added this sentence at the end of section 2.3:

"Data from the downwind site has not been used in the present contribution."

page 24327, line 6: van Pinxteren et al. (2015) is not a valid reference. As "Concentrations of a large number of organic acids were measured from the bulk cloud water samples", they need to be reported here. The ion balance (Fig. 2) should be complemented by the measured organic acid anions.

We removed the reference, but would rather not report the organic acid concentrations in detail in this manuscript. These data will be presented and discussed together with a range of other organics in a forthcoming manuscript. Including them here just for the ion balance would lead to duplicate publication. We agree, however, to give more details on the importance of the organic acids for the ion balance in the text and modified the paragraph as follows:

"Concentrations of a large number of organic acids were measured from the bulk cloud water samples and will be presented elsewhere (van Pinxteren et al., in preparation). Summing up the equivalent concentrations of the most abundant determined acids (formic, acetic, glycolic, oxalic, malonic, succinic, and malic acid) with consideration of their respective dissociation states depending on their pK_a values and sample pH values gives a range of 5 – 82 (average of 23) µeq L^{-1}, which explains 6 – 100% (average of 56%) of the inorganic anion deficit. In about 10% of the samples organic acid equivalent concentrations significantly exceeded the anion deficit (up to 255%), likely related to measurement uncertainties and/or non-determined cations. Considering that the DOC fraction likely contains many more than the analytically resolved organic acids, it can be assumed that the missing anions are predominantly organic in nature…"

Figure 3: Do symbols indicate the maximum concentration (CWL) datapoints?

The dots indicate data outside 1.5*IQR. We realise this information was missing and have now included it into the caption of Figure 3:

"…, whiskers extend to 1.5 IQR (interquartile range), and dots indicate individual data points outside this range."

page 24328, lines 19-23: This sentence is awkward and not clear. It may be deleted or, if considered important, be expanded in order to better explain the involved processes and conditions.

We agree and have deleted the sentence. The paragraph now reads:

"While LWC undoubtedly impacts solute concentrations, the additional variance it introduces to observed concentrations seems to be small, overall, and LWC does not seem to be the main driver in solute concentration differences during this study."

page 24329, lines 4-6: The sentence "Also in our dataset, LWC does therefore rather control the range of observable TICs than the actual TIC itself, : : :" is not very clear.

We agree and have deleted this sentence as well. The modified paragraph now reads:

"Maximum TICs are decreasing, while minimum TICs stay relatively constant with increasing LWC, leading to a range of observed TICs at any given LWC. As one and the same LWC value can result from different cloud microphysical conditions…"

page 24334, lines 25-29: The authors speculate about the potential role of organic acids in DOC behavior. They probably do know the answer, but do not present it to the readers of this manuscript. This is awkward.

We have checked our data and can tell that the determined main organic acids during the event discussed here (FCE11.2) explain less than 10% of the inorganic anion deficit. We can therefore not further elucidate the role of organic acids in DOC behaviour here. To indicate this to the reader, we have added the following sentence:

"It is noted that the main organic acids mentioned above explain only less than 10% of the inorganic anion deficit for this event."

sections 3.4.1 and 3.4.2: Please provide some information about the water volumes collected on the individual stages of the collectors.

We have added 2 figures in the supplemental material, giving water volumes per stage collected with the 3-stage (Fig. S8) and 5-stage collectors (Fig. S24). In addition, we added the following to the beginning of section 3.4.1:

"Volumes of cloud water collected per stage were between 5.9 and 240 ml with typically lowest volumes on the intermediate stage (16-22 µm) and highest volumes in the smaller or larger size class, depending on the sample (see Figure S8 for details)."

To the beginning of section 3.4.2, the following was added:

"Collected cloud water volumes were from 0.55 to 15 ml, with smallest volumes typically in the 4-10 µm droplet size range and largest ones mostly for droplets >30 µm (see also Figure S24)."

Wouldn’t it make sense (despite all overlap between stages) to present LWC and CWL data as functions of droplet size?

Indeed, we had first prepared the figures this way. We found that readability is compromised, however, because of many overlapping lines and therefore switched to the bar plots.

Tab 4 heading: Replace "> 0.2“ by "< 0.2“

Thanks, done.

Referee 2

The present manuscript describes the chemical composition of clouds sampled during the HCCT-2010 campaign. It features an impressive dataset in terms of physical and chemical parameters (inorganic ions together with DOC) measured. The cloud sampling is performed using several cloud collectors allowing to analyze the cloud chemical composition as a function of the droplet size. Sampling of the interstitial phase and of the cloud water by the CVI/INT system allows estimating the scavenging of the chemicals by the cloud water. Finally, AMS measurements allow to evaluate the variability of the cloud water composition with a high time resolution. Such analysis of cloud chemistry is very rare and difficult and by themselves worthwhile additions to the scientific literature. The discussion of the results is very detailed but clear and organized in a logical way. Some results like the correlation between LWC and cloud concentrations are expected/traditional. Other approaches are rather novel. For example, the way you connect air masses collected at three campaign sites (connected air flow) is interesting and allows documenting "full cloud events". Factors controlling solute concentrations are more subject to debate since there are so many factors that can explain concentration variability... So certain of your conclusions are more or less speculative...
Overall the study is an interesting contribution to atmospheric cloud chemistry and I would support publication of the manuscript while encouraging the authors to consider the following comments.

We thank the reviewer for his or her comments and helpful remarks and will address all suggestions in the following.

1- The introduction part is too long and should be reduced. Too many details are present and need to be suppressed for more clarity. Most of the information in the introduction part is even discussed in the results part.

We agree and have removed two details in the introduction section on the control of solute concentrations, as well as omitted a large part of the discussion on solute concentration size dependencies from the Schell et al. modelling study.

2- p. 24319, line 12 : you argue that samples are stored at -20_C until analysis. For H2O2 analysis, this can lead to strong underestimation of the concentration. Indeed, the retention coefficient of H2O2 is below 1 (around 0.3 following Snider et al., 1992). This means that about 1/3 of aqueous H2O2 is retained in the crystal during freezing of the samples.

Actually, this is one of the reasons why H2O2 was stabilized/preserved immediately after sampling and prior to freezing, as described on P24322 L13-16 in the discussion paper.

3- p. 24321. Please notify the nature of the filters (teflon, nylon?).

The sentence was modified as follows:

"Cloud water from the different samplers was filtered through 0.45 µm syringe filters (IC Acrodisc 13, Polyethersulfone membrane, Pall, Dreieich, Germany) and analysed…"

4- p. 24324. Definitely, I think that CASCC2 bulk samplers are not fitted to collect successive cloud water samples... especially for trace ions. For FCE 1.1 and FCE 13.3, the concentrations of certain chemical compounds should be carefully considered. This leads also to a general comment. To my opinion, CASCC2 sampler concentrates the cloud samples due to the way they collect the water. I know that most of the cloud water collection are performed with this Teflon strands system but I prefer the system where droplets impact on flat surface. This leads to my question. Did you compare you measurements performed on cloud water collected by your various cloud water collectors (CASCC2 vs. 5 stage for example)?

Yes, we have compared both the 3-stage and the 5-stage concentration with the ones observed from the bulk CASCC2 collectors and did not find severe deviations. We didn't include it into the manuscript due to paper size reasons, but we agree that it is interesting information and have therefore now included two figures into the supplemental material showing volume-weighted mean concentrations from the multistage collectors plotted versus corresponding bulk concentrations (Fig. S3 and Fig. S4). In addition, we have changed the title of section 3.2 to "Control samples and collector intercomparison" and added the following paragraph at the end of this section:

"Comparisons of volume-weighted mean concentrations from the multistage collectors with bulk concentrations from the CASCC2 for main cloud water constituents (sulfate, nitrate, ammonium, DOC) are shown in Figure S3 and Figure S4. They reveal generally similar data between the samplers with a tendency of sometimes higher concentrations from the multistage collectors, which was, however, not consistently observed for all constituents and/or cloud events."

5- p. 24325. Your lowest value for cloud pH (3.6) seems really low. You need high concentration of strong inorganic acids (mainly nitrate and sulfate) to reach this value. If you compare for example with puy de Dôme measurements, you reach a pH value closed to 3.6 with significantly higher concentration of nitrate and sulfate. Please discuss.

A minimum pH of 3.6 does not seem extraordinary low to us, given that polluted air masses e.g. from the Rhine/Main area (Frankfurt/Main) can easily reach the rural Schmücke site. As reported by (Deguillaume et al., 2014), pH can reach even lower values in the polluted regime at Puy de Dôme. It is clear that higher concentrations of acids are needed for a pH in this range, but we would not consider this to be worth adding another discussion to the already quite long manuscript.

6- Conclusion. What is the next step? Modeling investigations to explain the results? Is it feasible with current cloud chemistry models? Please discuss.

The reviewer is right in assuming that the experimental data presented here will be used in our multiphase SPACCIM/CAPRAM model (actually has been used already, but data is not published yet). We have added the following as last bullet point to the conclusions section:

- "The comprehensive dataset obtained during HCCT-2010 will serve as a reference for the further development and evaluation of multiphase models in future studies."

Minor comments:
1- Please add in the table 1, pH values for each FCE.

Table 1 contains general information on the FCEs and is quite loaded already. We have added mean pH values to Figure 1, together with total solute concentrations, as suggested below.

2- Figure 1. Please indicate the total concentrations of measured chemicals for each FCE. This will help the readers.

Good point, done.

Reference: Snider, J. R., D. C. Montague, and G. Vali : Hydrogen peroxide retention in rime ice, J. Geophys. Res., 97 (D7), 7569–7578, 1992.

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Cloud water composition during HCCT-2010: Scavenging efficiencies, solute concentrations, and droplet size dependence of inorganic ions and dissolved organic carbon

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Abstract

Cloud water samples were taken in September/October 2010 at Mt. Schmücke in a rural, forested area in Germany during the Lagrange-type Hill Cap Cloud Thuringia 2010 (HCCT-2010) cloud experiment. Besides bulk collectors, a 3-stage and a 5-stage collector were applied and samples were analysed for inorganic ions (SO_4^{2-},NO_3^-, NH_4^+, Cl^-, Na^+, Mg^{2+}, Ca^{2+}, K^+), H_2O_2 (aq), S(IV), and dissolved organic carbon (DOC). Campaign volume-weighted mean concentrations were 191, 142, and 39 µmol L^{-1} for ammonium, nitrate, and sulfate, respectively, between 4 and 27 µmol L^{-1} for minor ions, 5.4 µmol L^{-1} for H_2O_2 (aq),

1.9 µmol L^{-1} for S(IV), and 3.9 mgC L^{-1} for DOC. The concentrations compare well to more recent European cloud water data from similar sites. On a mass basis, organic material (as DOC * 1.8) contributed 20-40% (event means) to total solute concentrations and was found to have non-negligible impact on cloud water acidity. Relative standard deviations of major ions were 60-66% for solute concentrations and 52-80% for cloud water loadings (CWLs). ~~Contrary to some earlier suggestions, t~~The similar variability of solute concentrations and CWLs together with the results of back trajectory analysis and principal component analysis, suggests that concentrations in incoming air masses (i.e. air mass history), rather than cloud liquid water content (LWC) was the main factor controlling bulk solute concentrations ~~at Mt. Schmücke~~for the cloud studied. Droplet effective radius was found to be a somewhat better predictor for cloud water total ionic content (TIC) than LWC, even though no single explanatory variable can fully describe TIC (or solute concentration) variations in a simple functional relation due to the complex processes involved. Bulk concentrations typically agreed within a factor of 2 with co-located measurements of residual particle concentrations sampled by a counterflow virtual impactor (CV) and analysed by an aerosol mass spectrometer (AMS), with the deviations being mainly caused by systematic differences and limitations of the approaches (such as outgassing of dissolved gases during residual particle sampling). Scavenging efficiencies (SEs) of aerosol constituents were 0.56-0.94, 0.79-0.99, 0.71-98, and 0.67-0.92 for SO_4^{2-}, NO_3^-, NH_4^+, and DOC, respectively, when calculated as event means with in-cloud data only. SEs estimated using data from an upwind site were substantially different in many cases, revealing the impact of gas-phase uptake (for volatile constituents) and mass losses across Mt. Schmücke likely due to physical processes such as droplet scavenging by trees and/or entrainment. Drop size-resolved cloud water concentrations of major ions SO_4^{2-}, NO_3^-, and NH_4^+ revealed two main profiles: decreasing concentrations with increasing droplet size and "U"-shapes. In contrast, profiles of typical coarse particle mode minor ions were often increasing with increasing drop size, highlighting the importance of a species' particle concentration size distribution for the development of size-resolved solute concentration patterns. Concentration differences between droplet size classes were typically <2 for major ions from the 3-stage collector and somewhat more pronounced from the 5-stage collector, while they were much larger for minor ions. Due to a better separation of droplet populations, the 5-stage collector was capable of resolving some features of solute size dependencies not seen in the 3-stage data, especially sharp concentration increases (up to a factor of 5-10) in the smallest droplets for many solutes.

1 Introduction

Clouds represent an important part of the atmospheric multiphase system. Uptake of gases, dissolution of cloud condensation nuclei (CCN) constituents, and chemical reactions lead to complex compositions of their aqueous phase, which are highly variable in time and space and droplet size. Knowledge of these compositions and their variability is crucial for understanding a number of important processes in the atmosphere, including droplet activation and growth (e.g. Taraniuk et al., 2008; Facchini et al., 1999), formation and transformation of compounds (e.g. Herrmann et al., 2015; Fahey et al., 2005), production and consumption of important oxidants (e.g. Whalley et al., 2015; Marinoni et al., 2011), or transport and deposition of pollutants (e.g. Vet et al., 2014; Fowler et al., 2009). The present contribution presents results of cloud water chemical composition and related measurements during the Hill Cap Cloud Thuringia 2010 (HCCT-2010) experiment, performed in autumn 2010 at Mt. Schmücke, Germany. It focuses on the aspects of i) main drivers of bulk cloud water solute concentrations, ii) scavenging efficiencies of aerosol constituents, and iii) size-resolved droplet composition, which will be introduced here.

Whether and to what extent *solute concentrations are controlled by LWC* has been debated in the literature. Both Möller et al. (1996) and Elbert et al. (2000) concluded from their studies that LWC was the main parameter in controlling cloud water total ionic content (TIC) and that this relationship could be described by a power law function. From a comprehensive literature survey, Elbert et al. (2000) concluded that at any given site the cloud water loading (CWL, the product of solute concentrations and LWC) would be a fairly constant value (with "fairly constant" being interpreted as max / mean ratio < 5).~~While Möller et al. (1996) acknowledged that different air pollution situations lead to strong deviations from the average power function, Elbert et al. (2000) generalized their findings to the statement that at any given site the cloud water loading (CWL, the product of solute concentrations and LWC) would be a fairly constant value (with "fairly constant" being interpreted as max/mean ratio < 5).~~ In a discussion of this proposition (Kasper-Giebl, 2002; Elbert et al., 2002), Kasper-Giebl (2002) demonstrated that a constant CWL would imply either constant scavenging efficiencies and substance concentrations in air, or opposite trends of these two parameters, neither of which can be generally regarded as true. More recently, Aleksic and Dukett (2010)

showed for a very large dataset, that the relationship of TIC ~ LWC cannot be described by a simple function, but rather by a series of exponential distributions of TIC whose means values decrease with increasing LWC. These authors as well conclude that CWL is a stochastic quantity and thus cannot be a constant. ~~In their 12 year dataset of 3300 records, CWL had a relative standard deviation (RSD, standard deviation divided by mean times 100) of 107 %. The RSD of TIC was only slightly larger (127 %), contradicting the earlier conclusions of both Möller et al. (1996) and Elbert et al. (2000) of LWC being the main driver in TIC variability.~~ In section 3.3.2 of this work the parameters controlling bulk cloud water solute concentrations are studied for the comparatively uniform conditions during HCCT-2010 (with its identical site, season and wind sector during sampling).

Scavenging efficiencies (SEs) indicate how much of a compounds' total concentration is recovered in the cloud liquid phase after cloud formation. Different approaches for its calculation exist. Cloud water concentrations and interstitial particulate and/or gaseous concentrations have been used to derive in-cloud scavenging efficiencies of non-volatile or (semi)-volatile compounds (Sellegri et al., 2003; Acker et al., 2002; Hitzenberger et al., 2000; Kasper-Giebl et al., 2000; Daum et al., 1984). Alternatively, cloud concentrations can be related to total particulate (and/or gaseous) concentrations upwind of a cloud (van Pinxteren et al., 2005; Svenningsson et al., 1997; Leaitch et al., 1986; Hegg et al., 1984) or before cloud/fog onset (Gilardoni et al., 2014; Collett et al., 2008; Noone et al., 1992). In the ideal case of a "closed system" with conserved masses, all approaches would lead to the same scavenging efficiencies. However, as real clouds and fogs are open and dynamic systems, heavily interacting with their physical and chemical environment, the different approaches might lead to different results and comparing these might allow for insights into important processes taking place in the cloud/fog system. In the present study, many (though not all) of the phases relevant for the concentrations of major cloud constituents (sulfate, nitrate, ammonium, DOC) have been measured both upwind and inside of clouds at the Schmücke and are used to calculate and compare scavenging efficiencies derived from different approaches (section 3.3.4).

In clouds, *solute concentrations typically vary across droplet size* (Bator and Collett, 1997; Rao and Collett, 1995), which has significant implications for chemical reactions in droplets (Fahey et al., 2005; Reilly et al., 2001; Hoag et al., 1999; Gurciullo and Pandis, 1997) and deposition behaviour of solutes (Moore et al., 2004b; Collett et al., 2001; Bator and Collett,

1997). A conceptual model developed by Ogren et al. (1992) qualitatively describes the variation of non-volatile solute concentrations with cloud drop size in 3 different drop size regions: Region I ranges from < 1µm to approx. 5 µm drop diameter (exact size range strongly depends on cloud properties) and contains freshly activated (or non-activated) droplets close to their equilibrium size at the prevailing supersaturation. In this so-called "equilibrium growth" region, solute concentrations sharply decrease with increasing drop size, because at their critical diameter, larger droplets are more dilute than smaller ones as a result of the interactions between the Kelvin and the Raoult effect (Pruppacher and Klett, 2010; Ogren and Charlson, 1992). Region II, ranging from approx. 5 – 50 µm, represents droplets which have freely grown by water condensation beyond their critical size. In this "condensation growth" region, solute concentrations increase with increasing drop size, because small drops grow faster than large drops (r^{-1} growth law), i.e. large drops experience less dilution as compared to smaller ones. In region III, above approx. 50 µm in diameter, coalescence of drops becomes important. As larger drops collide more efficiently with smaller (i.e. more diluted) ones, solute concentrations decrease with increasing drop size in this "coalescence growth" region.

In more detailed numerical simulations, Schell et al. (1997) studied parameters determining non-volatile solute concentrations in different droplet sizes. Their results show size dependencies which are in principle consistent with the three regions in the conceptual model of Ogren et al. (1992). However, the exact shape of the curve strongly depends on several parameters~~, one of them being~~ like the droplet growth time (cloud age), the width of the CCN number distribution (e.g. presence of coarse particles), and the soluble fraction of input aerosol particles. In some cases, the concentration increase in the Ogren et al. region II can diminish to the point of constantly decreasing solute concentrations with increasing droplet sizes nearly over the full droplet size range. ~~In a freshly formed cloud (e.g. a hill cap cloud close to its base), there is a sharp solute concentration increase for droplets with diameters between approx. 5 and 30 µm (region II in the Ogren et al. model), while for smaller and larger drops concentrations decrease with increasing size (regions I and III). With increasing cloud age (i.e. droplet growth time or height within cloud), however, the rate of concentration increase in region II diminishes, eventually leading to a profile of decreasing solute concentrations with increasing droplet sizes in this drop size range (followed by increasing concentrations for larger drops). A second important parameter was found to be the width of the CCN number distribution, which determines both the drop diameter at the transition point~~

~~from region II to III in the Ogren et al. model (i.e. increasing vs. decreasing solute concentrations with increasing drop size), as well as the maximum solute concentration at this diameter. In simulation runs with narrowed CCN distributions (either by reducing maximum particle diameters in the initial size distributions or by reducing the maximum supersaturation and thus increasing the minimum CCN size limit), the transition between regions II and III was shifted towards smaller drops (diameter < 10 µm in the extreme case) and the solute concentration at this drop size decreased drastically. Similar to the scenario of an aged cloud, this can lead to constantly decreasing solute concentrations for drop diameters larger than about 5 - 10 µm. A third parameter influencing solute drop size dependencies was reported to be the soluble fraction of input aerosol particles. While input particles with soluble fractions of 25 and 50 % yielded solute concentration profiles consistent with the Ogren et al. model (including a strong increase of concentrations in region II), solute concentrations deriving from largely insoluble particles (2 % soluble fraction) tended to decrease with increasing drop size nearly over the full drop size range.~~

Thi~~s~~ese ~~discussion of the~~ model results illustrate~~s~~ the complexity of solute concentration drop size dependencies, which is even increased in reality by many factors such as gas-phase uptake of soluble material, chemical reactions in droplets, size-dependent composition and variable mixing state of input aerosol, entrainment processes, and inhomogeneous fields of supersaturation, i.e. different histories of individual droplets (Flossmann and Wobrock, 2010; Ogren and Charlson, 1992). In addition, available instrumentation for size-resolved droplet sampling usually integrates both over extended droplet size ranges with mostly ~~–~~2 size fractions only and time periods of typically hours, yielding volume-weighted sample concentrations which can significantly blur existing concentration gradients (Moore et al., 2004a, and references therein; Ogren and Charlson, 1992). Despite such difficulties, observations of size-dependent solute concentrations are still important as available measurements especially for more than two size fractions are very sparse. In the present study, a 3-stage and a 5-stage collector were applied and the observed solute concentration size dependencies are discussed in section 3.4 in view of the above described existing knowledge.

2 Materials and methods

2.1 Cloud water sampling

Cloud water sampling took place on top of a 20 m high tower at Mt. Schmücke (Thuringia, Germany, 50°39'16.5" N, 10°46'8.5" E, 937 m asl) with several collectors. Bulk cloud water samples were collected into pre-cleaned plastic bottles using the Caltech Active Strand Cloud Water Collector Version 2 (CASCC2, Demoz et al., 1996), which has a 50% collection efficiency cut-off diameter (D_{50}) of 3.5 µm and collects droplets by inertial impaction on Teflon strands within the airflow through the instrument. To increase the collected volume of cloud water for chemical analyses, 4 individual instruments were run in parallel with a time resolution of one hour. After weighing for volume determination, the samples were pooled, aliquots for different chemical analyses were taken and aliquots as well as leftover samples were stored at -20°C until analysis. For size-resolved droplet sampling a 3-stage collector (Raja et al., 2008) with nominal D_{50} of 22, 16, and 4 µm for stages 1, 2, and 3, respectively, was used. This collector is basically a size-fractionating version of the CASCC, using Teflon strands/banks with different diameters and different spacing in the 3 stages. In addition, the CSU 5-stage collector (Moore et al., 2002) with nominal D_{50} of 30, 25, 15, 10, and 4 µm for stages 1 – 5 was operated. In contrast to the 3-stage, the 5-stage collector impacts droplets on flat surfaces downstream of jets with decreasing diameters for air acceleration (cascade impactor design). It has to be noted, that experimentally determined D_{50}s for this sampler differ somewhat from the nominal values and that, even though droplet separating characteristics have been improved over other existing multistage collectors, there is still considerable mixing of droplets of different sizes within each stage (Straub and Collett, 2002). Due to limitations of the lateral channel blower applied in this study, the 5-stage collector was operated about 10% below its nominal air flow rate of 2.0 $m^3 min^{-1}$, which likely had a modest effect on its collection characteristics and adds some uncertainty to the real cut-off diameters. Sample handling from the multistage collectors was the same as described for the bulk collectors. Before each cloud event, the samplers were cleaned by spraying deionised water into the inlet (bulk collectors) or taking apart the individual stages and rinsing all surfaces with deionised water (multistage collectors). Control samples were taken after the cleaning procedures by spraying deionised water into the samplers and handling the collected water in the same way as the real samples.

2.2 Interstitial and residual particle sampling

To complement the liquid cloud water samples, droplet residuals and interstitial particles were sampled downstream of a counter-flow virtual impactor (CVI) and an interstitial inlet (INT). The CVI/INT system was set up in a building next to the measurement tower with the inlets installed through a window at 15 m height, facing south-west direction (215°). Details of the setup can be found elsewhere (Mertes et al., 2005; Schwarzenböck et al., 2000). In brief, interstitial particles and gases are separated from cloud droplets in the CVI by a counter-flow air stream which allows only droplets larger 5 µm in diameter to enter the system. Inside the CVI the droplets are evaporated in particle-free and dry carrier air, resulting in the formation of dry residual particles consisting of non-volatile cloud water components. Volatile components can be expected to evaporate during the drying process. The INT inlet samples interstitial particles and gases by segregating droplets larger 5 µm. Downstream of INT and CVI, particles were sampled on quartz filters (MK 360, Munktell, Bärenstein, Germany, 47 mm for CVI, 24 mm for INT) with sampling durations typically varying between ca. 4 and 8 hours (some shorter and longer sampling events existed as well). Filters were stored at -20°C for later offline analysis. Online measurements of submicron particle composition were performed by two aerosol mass spectrometers (AMS, Aerodyne Research Inc., USA): a C-TOF-AMS for droplet residuals (CVI, 5 min time resolution) and a HR-TOF-AMS for non-activated particles (INT, 2.5 min time resolution). Details of the AMS measurements ~~are~~ will be given in a forthcoming companion paper of this special issue (Schneider et al., in preparation).

2.3 ~~Upwind~~ Valley site~~s~~ aerosol sampling

Next to the Schmücke in-cloud site, two more valley sites upwind and downwind of the Schmücke were installed during HCCT-2010 to characterise air masses before and after their passage through the clouds. Characterisation of incoming aerosol was performed at the upwind measurement site close to the village of Goldlauter (50°38'15"N, 10°45'14"E, 605 m asl). A full description of the instrumental setup ~~is~~will be given in a forthcoming companion paper of this special issue (Poulain et al., in preparation). In brief, a commercial monitor for aerosols and gases (MARGA 1S, Metrohm Applikon, The Netherlands) was used for continuous (1 h time resolution) determination of water-soluble inorganic trace gases and particulate ions. The MARGA operated at a sampling rate of 1 $m^3 h^{-1}$ and consisted of a PM_{10} inlet, a wet rotating denuder absorbing water-soluble gases into deionised water (10 ppm

H_2O_2 added as biocide), a steam jet aerosol collector to grow and collect aerosol particles, and 2 ion chromatography systems for online cation and anion analysis. Size-resolved particle sampling was performed using a 5-stage Berner impactor with D_{50}s of 0.05, 0.14, 0.42, 1.2, 3.5, and 10 µm and a sampling flow rate of 75 l min^{-1}. Data from the downwind site has not been used in the present contribution.

2.4 Cloud microphysical and meteorological parameters

Cloud liquid water content (LWC), droplet surface area (PSA), and effective droplet radius (R_{eff}) were measured continuously by a particle volume monitor (PVM-100, Gerber Scientific, USA), which was mounted on the roof of a building next to the measurement tower. Droplet number distributions were obtained from a forward-scattering spectrometer probe (FSSP-100, PMS Inc., Boulder, CO, USA), sitting on the top platform of the measurement tower. A Ceilometer (CHM15k, Jenoptik, Jena, Germany) was installed at the upwind site Goldlauter to derive cloud base heights (CBHs). Standard meteorological parameters (temperature, air pressure, relative humidity, wind direction, wind speed, global radiation, precipitation) were determined by automatic weather stations (Vantage Pro2, Davis Instruments Corp., Hayward, CA, USA) both at the upwind site (ca. 3 m above ground) and on the Schmücke measurement tower (ca. 22 m above ground).

2.5 Chemical analyses

Cloud water from the different samplers was filtered through 0.45 µm syringe filters (IC Acrodisc 13, Polyethersulfone membrane, Pall, Dreieich, Germany) and analysed for inorganic ions Cl^-, NO_3^-, SO_4^{2-},Na^+, NH_4^+, K^+, Mg^{2+}, and Ca^{2+} by ion chromatography (IC) with conductivity detection (ICS3000, Dionex, Dreieich, Germany). Cation separation was performed in a CS16 column (3 mm) applying a methanesulfonic acid eluent, while anions were separated using a KOH eluent in an AS18 column (2 mm). Inorganic ions from CVI and INT filters were determined by the same method after extraction in deionised water (Milli-Q, Millipore, Schwalbach, Germany) and filtration through a 0.45 µm syringe filter. Blank correction of filter data took place by substracting mean concentrations from three unloaded field blank filters.

Dissolved organic carbon (DOC) was determined from filtered cloud water samples using a TOC-V_{CPH} analyser (Shimadzu, Japan) in the NPOC (non-purgeable organic carbon) mode

(van Pinxteren et al., 2009). Hydrogen peroxide (H_2O_2) in solution was determined (in sum with organic peroxides) by fluorescence spectroscopy (Shimadzu RF-1501) following the method of Lazrus et al. (1985). To stabilize peroxides during sample storage, p-hydroxyphenylacetic acid solution (POPHA) was added to aliquots of cloud water immediately after sampling to form a stable dimer (Rao and Collett, 1995). S(IV) and its reservoir species hydroxymethanesulphonate (HMS) were determined spectrophotometrically (Lambda 900, Perkin Elmer, Waltham, MA, USA) by the pararosaniline method (Dasgupta et al., 1980). Preservation of total S(IV) and HMS took place following the procedure described by Rao and Collett (1995). Concentrations of reactive compounds at the time of sample preservation can be biased due to reactions during the collection period. The extent of such artefacts will depend on reactant concentrations and cloud water pH and cannot easily be estimated. Cloud water pH was measured immediately after sampling using an MI-410 combination micro-electrode (Microelectrodes, Inc., USA) regularly calibrated at pH 4 and 7.

2.6 Data processing and back-trajectory analysis

Cloud water data are presented either as solute concentration ($\mu mol\ L^{-1}$ or $mg\ L^{-1}$) or as CWLs (sometimes also referred to as equivalent air concentrations) in $\mu g\ m^{-3}$. CWLs are derived from the solute concentrations by multiplication with the cloud LWC (in $g\ m^{-3}$) and the molar mass of the compound (in $g\ mol^{-1}$), where necessary. For comparison of CWLs between different instruments and/or sites, concentrations were normalised to standard temperature and pressure (STP: 273 K, 1013 mbar). Ambient temperature during the time of sampling was used for normalising cloud water collector data, while room temperature was used for CVI/INT, MARGA, and AMS data (room temp. at time of calibration for the ladder one). The open-source statistical software R (R Core Team, 2015) including the ggplot2 package (Wickham, 2009) was used for data processing and plotting. Back trajectories were calculated using the PC version of the HYSPLIT model (Draxler and Rolph, 2003) with GDAS 1° resolution data from NOAA's Air Resource Laboratory (http://ready.arl.noaa.gov/archives.php). Residence times indices (RTIs) for different land cover classes (water, natural vegetation, agriculture, urban areas, bare areas) were derived as proxies for the impacts of typical emissions over these areas on the sampled air masses following the methodology described by van Pinxteren et al. (2010).

3 Results and discussion

3.1 Cloud events

Within about 1/3 of the 6 weeks HCCT-2010 campaign Mt. Schmücke was covered in clouds. Based on the project philosophy of studying aerosol cloud interactions in a Lagrange-type approach, only those clouds were sampled for which local meteorological parameters (mainly wind direction) indicated a good possibility of sampling representative air masses at all three campaign sites ("connected" air flow, see Tilgner et al., 2014) without substantial loss of material between the sites (non-precipitating clouds only). After the campaign, these events were thoroughly evaluated regarding the hypothesis of a connected air flow (Tilgner et al., 2014), leading to the so-called "Full Cloud Events" (FCEs) with conditions appropriate to compare data from the different sites in a meaningful way. In Table 1~~Table 1~~ a list of the FCEs with cloud water samples available is given together with some additional information on meteorological and cloud microphysical conditions. Note that the numbering of the events is based on all clouds occurring during HCCT-2010 and is thus non-consecutive. A total of 8 FCEs were sampled, out of which some belonged to the same cloud appearance at Mt. Schmücke, but were interrupted either by rain or wind direction out of a predefined South-West corridor (FCE11.2+3 and FCE26.1+2). Two relatively long FCEs occurred with durations of 15 h, while the other events were shorter with 2 – 7 h durations. Mean LWCs ranged between 0.15 and 0.37 g m^{-3} and were a function of the in-cloud height of the measurement site (i.e. Schmücke above cloud base, derived from upwind site cloud base height measurements). Droplet surface areas were 700 – 1400 cm^2 m^{-3} on average with effective droplet radii of about 6 – 9 µm. Mean event temperatures decreased from about 9 °C for the first FCE to 1 – 2 °C for the last events at the end of the campaign . The numbers of samples for the different instruments are given in Table 1~~Table 1~~ as well according to the time resolutions of the samplers. Overall, meteorological and cloud microphysical conditions were typical for clouds at Mt. Schmücke during this time of the year. Many more details on meteorology are given in Tilgner et al. (2014).

3.2 Control samples and collector intercomparison

To check for possible contamination, control samples were taken from the cloud water collectors in between cloud events (section 2.1) indicating a "field blank" value for the species determined. Concentration levels in these blanks showed clear differences among the

three samplers with highest values from the CASCC2 bulk sampler (Figure S1). In contrast to the two multistage collectors, the CASCC2 was not disassembled for cleaning, which indicates that the cleaning procedure applied here (spraying deionised water through the sampler) is less effective in removing leftover traces from previously sampled cloud water (or its dried residuals if cleaning was not performed directly after the end of the event). Mean concentration levels in the controls are usually <10% of cloud water concentrations for more abundant ions (ammonium, nitrate, sulfate), but can make up significant fractions (up to 100% or even more in individual samples with low concentration) for trace ions (Figure S2). Mean blank levels of H_2O_2 and DOC are 25 and 15% of cloud water concentrations on average, respectively (Figure S2). The amount of carry-over contamination in the controls depends on concentration levels in the previous sample as well as on the effectiveness of the cleaning procedure (water volume applied, dried surfaces, etc.) and will likely vary from one event to another, which hampers a correction of cloud water concentrations by the available blank data. Carry-over contamination will likely affect the first sample of a new cloud event mainly, as the inside-surfaces of the CASCC2 are continuously washed by cloud water during operation and any contamination can be expected to be removed after the first hour of sampling. In addition, a fraction of the control sample concentrations can be suspected to form by uptake of gases during control sampling for species like ammonium (from ammonia), nitrate (from nitric acid), DOC (from water-soluble volatile organic compounds, VOCs), and especially H_2O_2. Cloud water concentrations are thus reported as measured in the following.

Comparisons of volume-weighted mean concentrations from the multistage collectors with bulk concentrations from the CASCC2 for main cloud water constituents (sulfate, nitrate, ammonium, DOC) are shown in Figure S3 and Figure S4. They reveal generally similar data between the samplers with a tendency of sometimes higher concentrations from the multistage collectors, which was, however, not consistently observed for all constituents and/or cloud events.

3.3 Bulk concentrations

3.3.1 Composition overview

In Table 2 concentrations of inorganic ions, H_2O_2 (aq), S(IV), HMS, and DOC as well as cloud water pH are summarised for the events given in Table 1~~Table 1~~. The observed range of pH-values was from 3.6 to 5.3, with a mean of 4.3. Highest ion concentrations (on a molar

basis) were observed for ammonium, followed by nitrate. Sulfate, chloride, and sodium showed considerably lower concentrations, while potassium, magnesium, and calcium were lowest. Arithmetic mean concentrations of this study are compared to literature data from clouds/fogs at other European sites in Table 3. Note that some authors report arithmetic means, while others report volume-weighted mean concentrations, which are always lower for a given dataset (see Table 2). Comparability of literature pH data is even more hampered as it is either reported as arithmetic mean or derived from either arithmetic or volume-weighted mean H^+ concentrations (the first approach leading to higher values than the other ones). In general, however, concentration levels in the present study are often similar to those observed in more recent campaigns at Puy de Dôme (continental non-polluted regime, Deguillaume et al., 2014), in the Western Sudety Mountains (Blas et al., 2008), and at the Schmücke site in a previous campaign (Brüggemann et al., 2005). In contrast, data from the 1980s and 1990s often show much higher concentrations of sulfate and nitrate (Bridges et al., 2002; Herckes et al., 2002; Wrzesinsky and Klemm, 2000; Acker et al., 1998; Joos and Baltensperger, 1991; Lammel and Metzig, 1991), presumably due to the decline in European emissions of NO_x and SO_2 over the past decades (EEA, 2014). Concentrations of DOC are more sparsely available in the literature for European clouds. Mean values during HCCT-2010 compare well with data from Puy de Dôme (continental non-polluted regime, Deguillaume et al., 2014), Rax (Löflund et al., 2002) and Schmücke (Brüggemann et al., 2005). Data for H_2O_2(aq) and S(IV) are even more sparse. In the present study, H_2O_2(aq) has been found to be within the same order of magnitude as determined in similar environments (Deguillaume et al., 2014; Brüggemann et al., 2005; Löflund et al., 2002), while S(IV) is at the lower end of reported concentrations.

Average relative compositions based on volume-weighted mean concentrations (in mg L^{-1}) are shown in Figure 1 for the main cloud events. DOC was converted to DOM (dissolved organic matter) using a conversion factor of 1.8 as in previous studies (Giulianelli et al., 2014; Benedict et al., 2012; Straub et al., 2012; Collett et al., 2008). Solute concentrations are always dominated by the main ions sulfate, nitrate, and ammonium, explaining approx. 60-70 % of total determined concentrations (campaign average 62 %). Among them, nitrate represents the dominant species (approx. 30-50 % of total concentrations, average 35 %), while sulfate and ammonium comprise lower fractions of total solutes (averages of 14 and 13 %, respectively). Organic compounds contribute approx. 20-40 % (average 28 %) and are thus another main constituent of cloud water dissolved material. These fractions are similar to what has been reported for background and anthropogenic influenced conditions at Puy de

Dôme (Marinoni et al., 2004) and are – despite the different environment – strikingly similar to the 20-year mean composition of Po valley fogs with 35 %, 15 %, 18 %, and 25 % contributions of nitrate, sulfate, ammonium, and DOM, respectively (Giulianelli et al., 2014).

The ion balance of inorganic anions versus cations (including $[H^+]$) is shown in Figure 2~~Figure 2~~. An anion deficit is observed for nearly all samples, ranging up to 178 µeq L^{-1}. Inorganic anions missing from the calculation are unlikely to explain the deficit, as they will have a small impact on the ion balance only (bicarbonate < 1 µM for given pH values, bisulfite < 3.2 µM based on S(IV) and HMS data). Concentrations of a large number of organic acids were measured from the bulk cloud water samples and will be presented elsewhere (van Pinxteren et al., in preparation). Summing up the equivalent concentrations of the most abundant determined acids (formic, acetic, glycolic, oxalic, malonic, succinic, and malic acid) with consideration of their respective dissociation states depending on their pK_a values and sample pH values gives a range of 5 – 82 (average of 23) µeq L^{-1}, which explains 6 – 100% (average of 56%) of the inorganic anion deficit. In about 10% of the samples organic acid equivalent concentrations significantly exceeded the anion deficit (up to 255%), likely related to measurement uncertainties and/or non-determined cations. ~~amounting to 3 – 115 µeq L^{-1} and can thus explain a large fraction of the anion deficit.~~ Considering that the DOC fraction likely contains many more than the analytically resolved organic acids, it can be assumed that the missing anions are predominantly organic in nature and that organic acidic material had a non-negligible impact on the cloud water acidity during HCCT-2010. Similar observations have been made before in other cloud/fog systems (Straub et al., 2012; Hegg et al., 2002; Khwaja, 1995; Collett et al., 1989).

3.3.2 Factors controlling solute concentrations

In Figure 3~~Figure 3~~a the variability of observed solute concentrations for selected ions is indicated in box-plots. Variability was high both within events (max/min ratios of up to 5-8 for main ions during the longer events, and up to 5-34 for minor ions), as well as in-between events (max/min ratios of median conc. between 3 and 6 for main ions, 6-29 for minor ions). In general, cloud water solute concentration variability can be caused by i) changes in microphysical cloud conditions, e.g. supersaturation and LWC, ii) changes in CCN concentration, size distribution, and chemical composition, iii) changes in gas-phase

concentrations of soluble gases and corresponding phase equilibria, and iv) chemical reactions in the cloud water. Distinctly different concentration patterns can be observed in Figure 3~~Figure 3~~a for three ion groups from similar sources, i.e. secondary ions ammonium, nitrate, and sulfate, sea-salt ions sodium and chloride, and the biomass burning and/or soil marker potassium, indicating a dominant influence of air mass history and thus CCN concentration and composition on cloud water solute concentrations. This is most obvious for sodium and chloride, which show highest concentrations during FCEs 1.1, 22.1, and 26.1+2. During these events, back-trajectory analysis revealed a stronger influence of marine emissions (residence time indices above water surfaces were between 0.3 and 0.5, as compared to < 0.2 for the remaining events, cf. Figure S~~3~~5).

To remove any influence of LWC fluctuations, CWLs are plotted in Figure 3~~Figure 3~~b. The CWL patterns resemble those of solute concentrations to a large extent, suggesting that for our dataset CCN composition and concentrations of soluble gases (i.e. air mass history) have a stronger impact on cloud water solute concentrations than LWC variability. Relative standard deviations (RSDs) of solute concentrations (whole campaign) are 66 %, 60 %, and 60 % for sulfate, nitrate, and ammonium, respectively, and 84-125% for trace ions. RSDs of CWLs are similar, sometimes even higher, with values of 80 %, 52 %, and 66 % for sulfate, nitrate, and ammonium, respectively, and 62-96 % for trace ions. Removing LWC variability, thus, does not reduce concentration variability, at least for the LWC range in this study. This is similar to ~~the~~ observations of Aleksic and Dukett (2010) from ~~their~~ a much larger dataset and indicates that LWC is obviously an important, but not necessarily the primary control factor of solute concentrations. ~~(cf. section 1), and challenges the proposition of CWL generally being a more appropriate measure for pollution characterisation and site comparisons in cloud studies (Elbert et al., 2000, section 1)~~.

~~While LWC undoubtedly impacts solute concentrations, the additional variance it introduces to observed concentrations seems to be small, if present at all. This might be either due to effects such as enhanced gas-phase uptake or smaller activation diameter, which can be (indirectly) linked to LWC and might counteract the additional variance and/or due to much larger variance in total aerosol concentrations in the air masses the clouds are forming in. In any case, LWC does not seem to be the main driver in solute concentration differences during this study.~~

If at all, an inverse functional relationship between solute concentration and LWC (Elbert et al., 2000; Möller et al., 1996) can only be observed during single events (i.e. when CCN concentration and composition as well as gas phase concentrations might be regarded comparably constant) in our dataset. This is shown in Figure 4~~Figure 4~~a for TIC versus LWC where the color-coded single event data indicates more or less constantly decreasing TIC with increasing LWC for some events. Overall, however, the pattern approximates those observed for larger datasets (Aleksic and Dukett, 2010; Kasper-Giebl, 2002; Möller et al., 1996): Maximum TICs are decreasing, while minimum TICs stay relatively constant with increasing LWC, leading to a range of observed TICs at any given LWC.~~. Also in our dataset, LWC does therefore rather control the range of observable TICs than the actual TIC itself, with larger ranges and larger mean values at lower LWC (Aleksic and Dukett, 2010). Given that~~ As one and the same LWC value can result from different cloud microphysical conditions (e.g. few large drops vs. more small drops) and clouds with similar LWC can form in very different air masses, this is actually an expected observation. In several other cloud/fog studies relationships between TIC and/or solute concentrations with LWC were reported to be non-existent, neither (Giulianelli et al., 2014; Straub et al., 2012; Marinoni et al., 2004; Kasper-Giebl, 2002).

The reason for this ostensible contradiction to the conclusions of the studies by Möller et al. (1996) and Elbert et al. (2000) might lie in different assessments of the quality of fitted models. Möller et al. (1996) and Elbert et al. (2000) report power law fits with coefficients of determination (R^2) of 0.27 and 0.38, respectively. Even when considering these values satisfactory (on the general usefulness of R^2 especially for goodness-of-fit of nonlinear models see Spiess and Neumeyer (2010)), the presented scatter plots leave room for questioning the ability of the fitted functions to adequately represent the data.

Instead of LWC, Marinoni et al. (2004) report TIC in cloud water at Puy de Dôme to be a power function of effective droplet radius (R_{eff}), even though with similarly poor R^2 of 0.29. In Figure 4~~Figure 4~~b, TIC during HCCT-2010 is plotted against R_{eff}, which was determined by the PVM as well. In contrast to LWC, both maximum and minimum ~~LWC~~ TIC values are decreasing with increasing R_{eff} in this plot and the relationship comes indeed closer to a functional one (best fit for simple linear regression; R^2 increases from 0.14 with LWC to 0.52 with R_{eff} as explanatory variable). There is, however, still substantial unexplained TIC

variation, likely arising from different broadness and/or skewness of the droplet size spectrum and from processes like phase equilibria and/or aqueous phase reactions.

In Figure 4~~Figure 4~~c and Figure 4~~Figure 4~~d the relationships of DOC with LWC and R_{eff} are shown, which are very similar to the ones observed for TIC. Herckes et al. (2013) examine total organic carbon (TOC) concentrations against LWC for a number of different sites worldwide. A simple relationship explaining the variation across all locations could not be identified by the authors. However, their plot looks remarkably similar to the plots of TIC vs. LWC from the larger datasets referenced above (decreasing spread of concentrations with increasing LWC), indicating that the main factors controlling the organic content of fog and cloud water are the same as the ones determining inorganic ion concentrations (likely nucleation scavenging and some additional gas phase uptake).

As a further means to study the various influences on solute concentrations, principal component analysis (PCA) was performed on cloud water solute concentrations and pH, back trajectory RTIs, LWC, and R_{eff}. Factor loadings of 4 extracted principal components after Varimax rotation are shown in Table 4. The first factor is highly correlated to air mass residence times above the oceans and cloud water concentrations of sea-salt constituents sodium, magnesium, and chloride. The second factor shows high loadings for all 4 main cloud water solutes (sulfate, nitrate, ammonium, DOC), representing typical main particulate components in aged continental air masses. The third factor is highly correlated to potassium and calcium concentrations and air mass residence times above agricultural lands and likely represents a mixed soil/biomass burning influence. The fourth factor mainly includes the variability of air mass residence times above urban areas, with no strong correlation to cloud water constituents. pH shows a weak anticorrelation to this factor, which could indicate an impact of acidic pollutants in comparably fresh air masses.

LWC has a much smaller impact on the marine factor than air mass residence time above water and its loading on factor 2 is weak as well (in contrast to R_{eff}, which has a significant impact on this factor). This further supports the conclusion of LWC variability impacting solute concentrations to a lesser extent if several clouds with different air mass histories are considered.

In summary, the discussion in this section shows that no single factor is available to adequately describe the complex processes controlling solute concentrations of both inorganic and organic material in bulk cloud water. If a simple functional relationship is needed, R_{eff}

might be a somewhat better choice than LWC. The probabilistic approach of Aleksic, however, seems more appropriate: For any given LWC (and probably R_{eff} as well), solute concentrations exhibit a (non-linear) distribution, as they depend on several other variables at the same time.

3.3.3 Comparison of bulk vs. CVI concentrations

In parallel to the bulk cloud water sampling, a CVI separated droplets from the interstitial phase and enabled the chemical characterisation of residual particles from filters and online with an AMS (section 2.2). The resulting CWLs of main solutes (normalised to standard conditions) are compared to the ones obtained from bulk cloud water samples in Figure 5. As can be seen, the temporal trends are often similar from both time-resolving samplers (CASCC2 and CVI-AMS), while absolute values can differ. During FCEs 11.3, 22.1, and 26.1+2, the ratios between CASCC2 and CVI-AMS CWLs are close to 1, especially for ammonium and sulfate (see Figure S~~4~~6 for ranges of CWL ratios). During FCEs 1.1, 11.2, and 13.3, this ratio is close to 2 (median), while it can be even higher for nitrate. Time-integrated mean CWLs from CVI filters are mostly close to the values from the CVI-AMS for sulfate and nitrate (with the exception of FCE1.1), while for ammonium, they are substantially lower during 4 out of the 6 events shown. CWL deviations for DOC (for residual particle data calculated as AMS organics divided by a conversion factor of 1.8 as above) tend to be lower than for the ions and CASCC2/CVI-AMS ratios are even below 1 during FCEs 1.1, 11.2, and 26.1+2 (Figure S~~4~~6). DOC CWLs from CVI filters are not given due to unreliable data from the small masses sampled on the filters.

Possible reasons for these deviations are manifold and include i) different sampling locations in the cloud (tower versus inlet at house wall), ii) different cut-off and detection characteristics (all dissolved bulk material analysed from CASCC2, while AMS measures non-refractory submicron residual particles only), iii) different assumptions/corrections for sampling efficiency (assumption of constant sampling efficiency across droplet size spectrum for CASCC2, correction of CVI sampling efficiencies based on particle number size distributions), iv) measurement uncertainties of analytical methods, AMS, and PVM for LWC measurement, v) – for DOC – uncertainty in the OM to OC conversion factor (1.8) and inclusion of undissolved organic matter in the AMS residual organics concentration, vi) – for filter samples – potential negative artifacts from evaporation of semi-volatile particle constituents during sampling as well as uncertainty from blank correction especially for short

sampling times and low sampled masses, and vii) – very important for some species - different droplet "pretreatment", i.e. liquid collection in the bulk sampler versus evaporation of water and volatile constituents such as ammonia, nitric acid and dissolved VOCs in the CVI. Given all these uncertainties and systematic differences, a general agreement between CWLs obtained from the different samplers within a factor of 2 appears well acceptable. A notable exception with much less agreement is nitrate during FCE11.2, where bulk cloud water CWLs are about a factor of 3.5 higher than CVI concentrations. The reason for the large deviation during this event is likely an enhanced concentration of nitric acid, which is taken up as nitrate into the bulk cloud water, but can be (partly) released back to the gas phase during droplet drying in the CVI (see also the following section).

3.3.4 Scavenging efficiencies

Scavenging efficiencies (SEs) were calculated by two different approaches. "In-cloud SEs" are based on cloud water loadings and interstitial particle concentrations (both being normalised to STP) and are calculated as follows:

$$SE_{in-cloud} = \frac{CWL}{CWL+c_{int}} \quad \text{(Equation 1)}$$

with $SE_{in-cloud}$: in-cloud scavenging efficiency

CWL: cloud water loading in µg m^{-3}, either from bulk cloud water (CASCC2) or from droplet residual concentrations (CVI-AMS and CVI-Filter)

c_{int}: interstitial particle concentration in µg m^{-3} (INT-AMS or INT-Filter)

"Upwind SEs", in contrast, are based on a comparison of STP normalised CWLs and upwind concentrations, calculated as:

$$SE_{upwind} = \frac{CWL}{c_{upw}} \quad \text{(Equation 2)}$$

with SE_{upwind}: upwind scavenging efficiency

CWL: cloud water loading in µg m^{-3} from bulk cloud water (CASCC2)

c_{upw}: upwind concentration from MARGA measurements in µg m^{-3}, either particulate only or total aerosol (particulate + gaseous concentration)

The results of these calculations are shown in Figure 6~~Figure 6~~. In-cloud SEs calculated from the different samplers usually agree well except for cases where sampler intercomparison was

poor (section 3.3.3). Comparison with upwind SEs, however, reveals substantial differences, which are summarised as event means in Table 5~~Table 5~~ (for residual in-cloud SEs only the ones based on CVI/INT AMS data are given here to avoid redundancy). Mean in-cloud SEs for sulfate are usually >= 0.9 except for FCE11.2 and FCE13.3, where substantial fractions (21-44%, depending on data used) of in-cloud sulfate reside in interstitial particles. During these events particle activation curves obtained from comparing measured particle number size distributions upwind and in-cloud were comparably shallow and the critical activation diameter was larger than during other events (Figure S7~~5~~), consistent with larger fractions of submicron sulfate not being activated to cloud droplets due to cloud microphysical conditions. Consistent with our data, in-cloud SEs of sulfate between 0.52 and 0.99 have been reported for clouds at Puy de Dôme, Brocken, and Mt. Sonnblick (Sellegri, 2003; Acker et al., 2002; Hitzenberger et al., 2000; Kasper-Giebl et al., 2000), with larger values being more typical.

In contrast to in-cloud SEs, sulfate upwind SEs were mostly << 0.9, indicating incomplete mass conservation between the sites. From previous studies at the Schmücke (Brüggemann et al., 2005; Herrmann et al., 2005) and results on aerosol processing presented in a forthcoming companion paper, it is known that various physical loss processes, such as scavenging of cloud droplets by trees and/or entrainment of cleaner air masses from aloft can reduce observed concentrations of all particle constituents along the air path from upwind via Schmücke towards the downwind site. Upwind SEs being smaller than in-cloud SEs support these conclusions of physical particulate mass losses from the upwind to the in-cloud site. Only during FCE13.3 upwind SEs are found to be higher than in-cloud SEs, indicating additional sulfate mass within the cloud, which could result from chemical production, uptake of gaseous H_2SO_4 (Roth et al., 2016; Harris et al., 2014; Harris et al., 2013) and/or other processes (e.g. entrainment). Similar to sulfate, ammonium shows in-cloud SEs typically > 0.9, except for FCE13.3 (large activation diameter). Upwind SEs are similarly large if upwind particulate ammonium concentrations are considered only, but drop to mean values between 0.4 and 0.7 if gaseous upwind ammonia - which is likely to be taken up by the cloud water at least partially- is included in the balance. Consistent with the conclusions from sulfate, the lower overall upwind SEs thus likely reflect the impact of physical loss processes at the sites.

For nitrate and DOC, these comparisons look different. While in-cloud SEs are again > 0.9 in most cases, upwind SEs are > 1 in most cases, indicating additional nitrate and DOC at the in-cloud site (note that event mean DOC upwind SEs in Table 5~~Table 5~~ were calculated using

water-soluble organic carbon concentrations from impactor samples, as the MARGA analyses inorganic ions only). For DOC, this most likely results from uptake of water-soluble VOCs (e.g. acids, aldehydes, ketones) into cloud droplets. The highest value was observed for FCE11.2, where the inorganic anion deficit was highest as well (Figure 2~~Figure 2~~), indicating that a significant amount of organic material taken up from the gas-phase must have been acidic or – alternatively – neutral compounds were oxidised to organic acids upon dissolution in the cloud droplets. It is noted that the main organic acids mentioned above explain only less than 10% of the inorganic anion deficit for this event.

For nitrate, upwind SEs stay similarly high or even higher than in-cloud SEs even after considering any upwind HNO_3 measured by the MARGA. Especially when considering that nitrate likely experiences similar physical mass losses as ammonium and sulfate (which typically were on the order of 10 – 40% at the downwind site, data not shown here), this would imply a nitrate budget at the cloud site substantially larger than the sum of particulate and gaseous nitrate at the upwind site. Given that aqueous phase oxidation of NO_x to nitrate can be considered negligible (Seinfeld and Pandis, 2006) and a potential positive nitrate artefact from hydrolysis of N_2O_5 in the cloud water can be assumed to be present in similar magnitude in the wet rotating denuder samples of the MARGA system (Phillips et al., 2013), such a large budget increase of nitrate at the cloud site seems unrealistic. In addition, a comprehensive data analysis focussing on aerosol processing during FCEs (manuscript in preparation) does not yield indications for increased nitrate at a site downwind of the cloud, neither on average over all FCEs, nor specifically during FCE11.2, where nitrate enrichment was highest. Any additional nitrate in the cloud water thus needs to evaporate back to the gas phase upon cloud dissipation.

The most likely explanation for the observed discrepancy is a severe underestimation of nitric acid by the MARGA system. Accurate nitric acid determination is known to be challenging due to the "stickiness" of the molecule (Rumsey et al., 2014) and adsorption in the inlet was reported to be strongly increased when sampling air – as during FCE sampling - is near 100% RH (Neuman et al., 1999). As the inlet HDPE tubing during HCCT-2010 was approx. 3.5 m long (from PM_{10} head to denuder), significant losses of HNO_3 before denuder sampling seem likely. In a not (yet) published intercomparison of nitric acid between the MARGA unit as used during HCCT-2010 and a separate batch denuder with inlet tubing reduced to a minimum, concentration ratios between the MARGA and the reference denuder were

typically between 0.17 and 0.98 (10^{th} and 90^{th} percentile, G. Spindler and B. Stieger, personal communication). Using a value of 0.25 (lower quartile of the intercomparison) as a correction factor for nitric acid measured during HCCT-2010 (i.e. multiplying measured apparent concentrations by 4) yields upwind SEs for total nitrate between 0.7 and 1.2 (as event means), which would be more consistent with the values obtained for ammonium and sulfate.

An enrichment of cloud water nitrate has previously been observed in several studies and has usually been related to the uptake of nitric acid as the most probable explanation (Prabhakar et al., 2014; Hayden et al., 2008; Brüggemann et al., 2005; Sellegri et al., 2003; Cape et al., 1997), which is in agreement with our considerations described above.

In conclusion, the comparison of upwind and in-cloud scavenging efficiencies reveals that i) nucleation scavenging typically removed >80 %, often close to 100 % of soluble material from the particle phase upon cloud formation, ii) uptake of gaseous ammonia, nitric acid and water-soluble VOCs had an additional significant impact on observed cloud water concentrations, and iii) particulate material is clearly lost or diluted to some extent between the upwind and the in-cloud site, likely due to physical processes such as droplet scavenging by trees and/or entrainment of cleaner air masses.

3.4 Size-resolved droplet compositions

3.4.1 3-stage collector

In Figure 7 volume-weighted mean (VWM) concentrations per cloud event are shown for ions, H_2O_2, and DOC within the droplet size classes of the 3-stage collector. Even though the nominal cut-off diameters of the 3 stages are given in Figure 7, it has to be noted, that in reality significant mixing of droplets between the nominal size classes occurs due to the relatively broad collection efficiency curves (Straub and Collett, 2002). Concentrations in a given droplet size class are thus influenced by droplets from other size classes to a significant extent and the size distributions can only reflect an approximate picture of the real pattern. Volumes of cloud water collected per stage were between 5.9 and 240 ml with typically lowest volumes on the intermediate stage (16-22 µm) and highest volumes in the smaller or larger size class, depending on the sample (see Figure S8 for details).

Volume-weighted mean concentrations per event were calculated to reduce the complexity of the data set, even though information on the temporal evolution of size-resolved concentrations is lost by the averaging. Data for all individual samples taken with the 3-stage collector is given in the Supplemental Material (Figure S9~~6~~ – Figure S1~~5~~8). As can be seen there, concentrations levels of individual cloud water constituents can vary significantly within one cloud event while the general patterns of concentrations in the three droplet size classes are often quite persistent during an event (exceptions will be noted below). For the major ions sulfate, nitrate, and ammonium, two main profiles of size-resolved cloud water concentrations can be observed in the VWM data: i) decreasing concentrations with increasing drop size for FCEs 1.1, 11.2, 11.3, 13.3, and ii) profiles with minimum concentrations in medium-sized droplets on stage 2 ("U"-shaped profiles) for FCEs 22.1 and FCE26.1+2. Only for nitrate during FCE1.1 a profile of increasing concentrations with increasing drop size is observed. Concentration differences between highest and lowest values are usually within a factor of 2 with the exception of FCE11.2, where concentrations of sulfate and ammonium in large drops were a factor of 3-4 lower than in small drops (on VWM basis). The two types of profiles reflect the dominant profiles of major ions in the individual samples (Figures S~~6~~9 – Figure S~~8~~11) for most of the events. Only during FCE1.1 and mainly for sulfate and ammonium, the VWM profile does not adequately represent the individual profiles, which were rather variable during the first half of this 15h event and stabilized to a profile of increasing concentrations with increasing drop size during the second half of the event. As sampled water volumes were comparably low during the second half of the event, however, their weight to the volume-weighted mean profile is rather low. Literature data from 3-stage cloud water collectors is very sparse. Raja et al. (2008) report decreasing concentrations of main ions with increasing drop size for fog samples in the US Gulf coast region, obtained with the same collector as in the present study. Collett et al. (1995) observed U-type profiles in cloud samples obtained with a different 3-stage collector (different nominal cut-offs) from two sites in North Carolina and California, USA.

The VWM profiles of low concentration ions (chloride, sodium, magnesium, calcium, and – in part – potassium) were found to be markedly different from the major ion profiles. Concentrations were usually increasing with increasing drop size, especially for events with elevated concentrations (FCE1.1, 22.1, and 26.1+2) due to elevated impact of marine emissions on sampled air masses (cf. section 3.3.2). Also, observed concentration differences in different drop size ranges tended to be larger (up to a factor of 10) as compared to major

ion concentrations. Available literature data for minor ions in three drop size ranges reveals diverse profiles, depending on species and location (Raja et al., 2009; Collett et al., 1995).

In contrast to the ionic data, concentrations of H_2O_2 in different collector stages were comparably homogeneous, with maximum differences of 25% (or a factor of 1.3). This is likely related to the different incorporation pathway (uptake from gas-phase as compared to nucleation scavenging for the ions), which is expected to yield more similar concentrations in differently sized cloud drops, at least if equilibrium conditions are assumed (Hoag et al., 1999).

Both uptake pathways can in principle occur for DOC (VOC uptake and/or dissolution of CCN organic material). The size-resolved concentration pattern in Figure 7, however, resembles those of major ions, suggesting nucleation scavenging as the major path of DOC incorporation into cloud water during this study.

Mean pH values per event (based on VWM concentrations of H^+) are shown in Figure 8a. A similar pattern of slightly (approx. 0.1 pH units per stage) increasing values with increasing drop diameter can be observed for nearly all events and collector stages. In individual samples (Figure S1~~6~~9) differences between stages can be somewhat higher (up to approx. 0.5 pH units), but the general patterns look similar to the VWM event averages. Qualitatively, increasing pH with drop size is consistent with i) coarse (and typically less acidic) CCNs leading to larger droplets (cf. elevated concentrations of coarse particle mode constituents), and ii) reduced (diluted) concentrations of potentially acidic constituents (sulfate, nitrate, DOC) in larger drops (Collett et al., 1994).

These observations highlight the complexity of solute concentration drop size dependencies. Even for the comparably uniform conditions of the present study (same site, same season, similar air mass origins, similar heights within the cloud), different profiles can result for one and the same ion. This becomes even more obvious from individual samples (e.g. sulfate during FCE1.1, Figure S~~6~~9), where – as stated above – a number of different profiles can occur during the same cloud event. Considering that these individual samples represent volume-weighted averages over 2 hours, it is easy to imagine that with a higher time resolution of sampling the variability of observed profiles would even increase. Without detailed numerical modelling (which is beyond the scope of this study), a quantitative understanding of these profiles and their variations seems impossible. In addition, the sampler characteristics (few stages with broad collection efficiencies) together with changing droplet

size distributions in a cloud might influence the observed size dependencies. Even though drop volume size distributions were usually similar both between events (Figure S~~17~~20) and between individual samples within the events (Figure S~~18~~21), subtle changes, e.g. in the broadness of the distribution or in the abundance of large (> 30 µm) drops, can – together with the broad mixing of differently sized drops – lead to artificial modifications in the observed volume-weighted concentrations on the three stages (Moore et al., 2004a). Despite these difficulties, two broad conclusions from the 3-stage sf-CASCC ion data can be drawn: (i) main ions (sulfate, nitrate, ammonium) have similar solute concentration drop size dependencies (consistent with their presumed strong internal mixing in CCNs) and are often enriched in smaller sized droplets (even though other, especially U-type profiles do occur as well), and (ii) increasing concentrations with increasing droplet sizes, which might be expected based on the consideration of the simple Ogren et al. (1992) model (see section 1), are mainly observed if a strong coarse mode in upwind particles is present for a given constituent (e.g. for sodium, magnesium, chloride, and nitrate- during FCE1.1; cf. Figure S~~19~~22 and Figure S~~20~~23 for size distributions of inorganic ions at upwind site during FCEs). These findings are consistent with the availability of coarse CCN being an important prerequisite for such an inverse concentration – size relationship to develop (Schell et al., 1997), although other factors likely contribute to these observations as well.

3.4.2 5-stage collector

Size-resolved concentrations of ions and H_2O_2 from the 5-stage collector are given in Figure 9 in the same way as described above for the 3-stage data (event VWM and normalised data)~~.~~, Collected cloud water volumes were from 0.55 to 15 ml, with smallest volumes typically in the 4-10 µm droplet size range and largest ones mostly for droplets >30 µm (see also Figure S24). ~~c~~Concentration profiles of individual samples are shown in Figure S2~~1~~5 – Figure S~~29~~33). The number of events is smaller, as this sampler was not operated during FCE1.1 and FCE26.1+2. Due to the relatively low volume of cloud water the 5-stage collector is sampling, DOC analysis could not be performed from these samples. For major ions, the patterns are broadly consistent with the profiles of decreasing concentrations with increasing drop size observed from the 3-stage collector for FCEs 11.2, 11.3, and 13.3, with FCE22.1 showing some similarity to a U-shape (even though the concentration increase towards larger drops is observable on stage 2 only, not on stage 1 collecting the largest drops). Concentration

differences between smallest and largest droplets are somewhat more pronounced (typically a factor of about 2) as compared to the 3-stage collector (typically smaller than a factor of 2), illustrating the higher efficiency of the 5-stage collector in separating small and large drop populations. Sharpest concentration differences are usually observed between stage 4 and 5 (small droplets). This is true for basically all of the individual samples as well (Figure S251 – Figure S2933). Concentration patterns on stages 1-4, however, can vary somewhat within a single event, depending on the development of the cloud. For example, nitrate shows constantly decreasing concentrations with increasing drop sizes during the first half of FCE11.2 (Figure S226), while during the second half, concentrations in larger drops tend to increase. Similarly, ammonium concentrations develop from a maximum in medium-sized drops for the first sample to notably homogeneous concentrations across all 5 collector stages (difference of only about 30 % between smallest and larges drops) during FCE11.2 (Figure S237). The observed profiles differ from those reported from a hill cap cloud at Whiteface, NY, USA, using the same 5-stage collector (Moore et al., 2004a), where U-type profiles with highest concentrations in largest drops were observed for ammonium and nitrate, while sulfate showed increasing concentrations with increasing drop size through all 5 stages. The same study reports 5-stage concentration profiles from a fog event in Davis, CA, USA, which are more similar to those in this study, with decreasing concentrations with increasing drop size (Moore et al., 2004a).

The patterns of trace ions also show some similarity with the ones observed from the 3-stage collector, mainly in that concentrations tend to increase from medium-sized towards larger droplets for most ions and events as well. There are, however, two distinct features in the 5-stage data which are not captured by the 3-stage collector: First, similar to the main ions, the concentration increase towards larger droplets is often (though not always) observable on stage 2 only, with decreasing concentrations on stage 1 (largest drops). Second, all trace ions show a very pronounced concentration increase in smallest droplets (stage 5), with often a factor of 5-10 difference to stage 4 concentrations, which is usually not seen in the 3-stage data, where smallest droplets are mixed with much larger ones on stage 3, leading to more diluted concentrations. Literature data on size-resolved trace ion concentrations from 5-stage collectors is available only for calcium, for which a pronounced U-type profile with highest concentrations in largest drops was reported (Moore et al., 2004a), while sodium, potassium and chloride ions were mentioned to have very similar profiles.

Compared to ionic content, the concentrations of H_2O_2 are more homogeneously distributed between the collector stages (maximum deviation < 50%) - similar to what was observed from the 3-stage collector data - and a general pattern cannot be observed from the (few) data available.

Event-averaged pH values from the 5-stage collector are given in Figure 8b (for individual samples in Figure S3~~0~~4). Highest values were mostly observed in smallest droplets (stage 5) with a significant decrease towards the next droplet size range (stage 4) at least during 3 out of the 4 events. From collector stage 4 towards stage 2 (increasing drop sizes) pH values tend to increase, similar to what is observed from the 3-stage collector (Figure 8a), while in largest drops (stage 1) they decrease again (to different extents). ~~–~~Overall, pH variations between different drop size classes are not too large for the sampled clouds with maximum differences of about 0.6 pH units on event-averaged basis.

These observations are generally consistent with the findings from the 3-stage collector. However, they also highlight the higher efficiency of drop population separation of the 5-stage collector as compared to the 3-stage collector, as ratios between minimum and maximum concentrations are larger and the sharp concentration increase towards the smallest droplets (especially for trace ions) is only observed here (for volume size droplet distributions during 5-stage sampling see Figure S3~~1~~5). In addition, the observation of often decreasing concentrations from stage 2 (second-largest drops) to stage 1 (largest drops) might reflect the transition from region II (condensation growth) to region III (coalescence growth) in the Ogren et al. (1992) model (section 1), even though it must be noted that collection efficiency curves of these two stages are overlapping to a comparatively large extent (Straub and Collett, 2002). Compared to the study of Moore et al. (2004a) stressing the importance of cloud age (drop growth time) by comparing two different types of clouds/fogs, our data from more similar cloud systems highlights the impact of the size distributions of CCN constituents on the development of size-resolved concentration patterns. Both parameters were predicted to be relevant from detailed model sensitivity studies (section 1, Schell et al., 1997). In addition, despite the considerable mixing of droplets with different sizes occurring in the samplers, the data reveal the substantial differences which can exist in different droplet size classes as well as the variability of observed solute concentration profiles even under comparably similar cloud conditions. As such differences impact both chemical reactions in cloud drops and deposition efficiencies and can thus modify atmospheric sink and/or source strengths of PM

constituents (Moore et al., 2004b), further observational and modelling studies on size-resolved droplet compositions seem important.

4 Conclusions

The analysis of bulk and size-resolved cloud water samples and related measurements of 8 cloud events during HCCT-2010 has led to the following main conclusions:

- Variability of solute concentrations in bulk samples was high ~~even~~ for the ~~comparably uniform conditions~~clouds studied ~~during the campaign~~ and was caused mainly by the variability of CCN concentrations and compositions, i.e. air mass history, in contrast to earlier suggestion of LWC generally being the main driver in solute concentration variation.
- A simple functional relationship between LWC and solute concentrations was observed only within single cloud events with little variation in incoming air mass concentrations and conditions. Across several events, no single factor is available to adequately describe the complex processes determining observed solute concentrations in cloud water. If a simple function is needed, R_{eff} might be a somewhat better choice than LWC.
- Both nucleation scavenging and gas-phase uptake contributed to observed cloud water concentrations of major constituents, with the first one being especially important for sulfate and the second one for nitrate.
- Losses of particulate mass occur from the upwind to the in-cloud site, observed from different in-cloud vs. upwind scavenging efficiencies and likely related to physical loss processes such as droplet scavenging and/or entrainment.
- Solute concentration droplet size profiles can be highly variable even within single events and were only partly consistent with considerations from a simple conceptual model. The observations made highlight the importance of CCN constituents' size distributions on the development of concentration profiles, consistent with earlier numerical simulation results.
- The comprehensive dataset obtained during HCCT-2010 will serve as a reference for the further development and evaluation of multiphase models in future studies.

Supplemental material related to this article is available online at doi: …

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Tables and Figures

Table 1: Sampling times of cloud water collectors during Full Cloud Events with mean liquid water content (LWC), droplet surface area (PSA), effective droplet radius (R_{eff}), Schmücke above cloud base (SACB), temperature (T), wind speed (WS), and global radiation (GR) at Mt. Schmücke, as well as the number of samples for the different collectors.

Event	Start (CEST)	Stop (CEST)	Duration (h)	LWC ($g\ m^{-3}$)	SACB (m)	PSA ($cm^2\ m^{-3}$)	R_{eff} (µm)	T (°C)	WS ($m\ s^{-1}$)	GR ($W\ m^{-2}$)	# CASCC2	# 3-stage	# 5-stage
FCE1.1	14/09/2010 11:00	15/09/2010 02:00	15	0.24	167	1248	5.7	9.2	8.2	15	15	7	-[a]
FCE7.1	24/09/2010 23:45	25/09/2010 01:45	2	0.19	156	846	5.7	8.3	5.5	0	2	1	1
FCE11.2	01/10/2010 22:30	02/10/2010 05:30	7	0.37	237	1277	8.7	6.2	4.1	0	7	4	4
FCE11.3	02/10/2010 14:30	02/10/2010 19:30	5	0.33	225	1353	7.4	7.7	7.3	31	5	3	2
FCE13.3	06/10/2010 12:15	07/10/2010 03:15	15	0.34	185	1392	7.3	9.1	3.9	52	15	8	4
FCE22.1	19/10/2010 21:30	20/10/2010 03:30	6	0.30	222	1272	7.4	1.2	4.7	0	6	3	2
FCE26.1	24/10/2010 01:30	24/10/2010 08:30	7	0.20	174	961	7.6	2.3	8.9	0	7	3	-[a]
FCE26.2	24/10/2010 09:15	24/10/2010 11:45	2.5	0.14	141	701	7.3	1.4	9.1	43	3	1	-[a]

a) Collector not operated

Table 2: Summary of cloud water solute concentrations determined during HCCT-2010.

Compound	Unit	#	Range	median	mean	VWM
pH		60	3.6-5.3	4.56	4.29[a]	4.30[a]
SO_4^{2-}	µmol L^{-1}	60	6.2-104	33	43	39
NO_3^-	µmol L^{-1}	60	46-479	151	164	142
Cl^-	µmol L^{-1}	60	3.7-84	22	30	25
NH_4^+	µmol L^{-1}	60	64-523	182	216	191
Na^+	µmol L^{-1}	60	0.58-195	20	35	27
K^+	µmol L^{-1}	60	1.3-31	3.8	6.1	5.5
Mg^{2+}	µmol L^{-1}	60	0.63-26	3.1	5.1	4.1
Ca^{2+}	µmol L^{-1}	60	1.4-37	7	9.8	8.7
H_2O_2	µmol L^{-1}	60	0.35-17	5	5.6	5.4
S(IV)	µmol L^{-1}	34	BDL-3.6	2.1	1.9	1.9
HMS	µmol L^{-1}	34	BDL-2.7	0.76	0.87	0.91
DOC	mgC L^{-1}	60	1.3-13	4	4.4	3.9

#: Number of samples analysed
VWM: volume-weighted mean concentration
BDL: below detection limit
[a] derived from mean/VWM H^+ concentration

Table 3: Comparison of mean HCCT-2010 cloud water concentrations with literature data (arithmetic or volume-weighted means) from other European mountain sites.

Location	Date	pH	Cl^- (μM)	SO_4^{2-} (μM)	NO_3^- (μM)	NH_4^+ (μM)	Na^+ (μM)	K^+ (μM)	Mg^{2+} (μM)	Ca^{2+} (μM)	H_2O_2 (μM)	DOC (mg L^{-1})	S(IV) (μM)	Ref.
Schmücke, Germany	2010	4.3	30	43	164	216	35	6.1	5.1	9.8	5.6	4.4	1.9	This work
Puy de Dome, France[a]	2001-2011	4.3	69	60	417	233	44	18	3.8	53	4.9	12[d]		(Deguillaume et al., 2014)
Puy de Dome, France[b]	2001-2011	5.1	35	49	111	145	34	5.0	6.6	15	10	5.5[d]		(Deguillaume et al., 2014)
Sudety Mts., Poland	2003-2004	4.25	66	67	173	167	67	6	10	26				(Blas et al., 2008)
Schmücke, Germany	2001-2002	4.5	19	59	207						2.7	6.4		(Brüggemann et al., 2005)
Holme Moss, UK[c]	1994-2001		652-1711	90-208[e]	175-469	158-518	578-1563							(Beswick et al., 2003)
Rax, Austria	1999-2000	3.8	16	82	136	230	16	7	11	11		6.0[d]		(Löflund et al., 2002)
Vosges Mts., France	1998-1999	4.82	143	149	181	276	175	57	26	60				(Herckes et al., 2002)
Zinnwald, Germany	1997-1998	4.0	48	281	176	560	52	23	6	28				(Zimmermann and Zimmermann, 2002)
Waldstein, Germany	1997	4.3	54	248	481	669	65	11.5	19.5	34				(Wrzesinsky and Klemm, 2000)
Krusne Hory, Czech Rep.	1995-1996	2.96	155	625	726	203	64	20	20	68				(Bridges et al., 2002)
Brocken, Germany[c]	1992-1996	3.8-4.5	68-119	133-160	280-365	378-468	60-128	2-12	14-18	27-67				(Acker et al., 1998)
Great Dun Fell, UK	1993	4.0		91	202	321							2.7	(Laj et al., 1997)
Sonnblick, Austria	1991[f]	4.5	30	64	32	36	34	12	2.9	11				(Brantner et al., 1994)
Vosges Mts., France	1990	3.3	120	185	410	270	170	40					24	(Lammel and Metzig, 1991)
Schöllkopf, Germany	1988	4.1	90	250	400	830	70	60						(Lammel and Metzig, 1991)
Zindelen, Switzerland	1986-1987	4.8	431	447	1020	2107							85.9	(Joos and Baltensperger, 1991)

a) polluted regime
b) continental regime
c) range of annual means
d) TOC
e) nss-Sulfate
f) fall data

Table 4: Factor loadings of 4 principal components after Varimax rotation. Loadings with absolute values <= 0.2 are regarded insignificant and omitted, while those > 0.6 are regarded highly significant and printed bold.

	F1	F2	F3	F4
pH		0.53	-0.26	-0.36
LWC	-0.57	-0.32		0.47
Reff	-0.45	**-0.74**		
RTI Water	**0.84**		-0.48	
RTI NaturalVegetation	**-0.92**		0.28	
RTI Agriculture	-0.39		**0.63**	0.49
RTI Urban	0.22			**0.91**
Sulfate		**0.93**		
Nitrate		**0.73**	0.54	0.24
Ammonium		**0.97**		
Sodium	**0.95**			
Magnesium	**0.89**		0.24	0.20
Chloride	**0.95**			
Potassium			**0.87**	
Calcium	0.26	0.34	**0.72**	0.34
DOC	-0.24	**0.77**	0.50	

Table 5: Event means of upwind and in-cloud scavenging efficiencies calculated from different approaches. Numbers in brackets include both particulate and gaseous upwind concentrations, where available. See text for details.

Event	CASCC2+MARGA	CASCC2+INT-AMS	CVI/INT AMS
Ammonium			
FCE1.1	0.85 (0.39)	0.92	0.83
FCE11.2	0.95 (0.52)	0.98	0.96
FCE11.3	1.04 (0.5)	0.97	0.97
FCE13.3	0.94 (0.65)	0.80	0.71
FCE22.1	0.85 (0.69)	0.96	0.96
FCE26.1+2	1.01 (0.51)	0.95	0.90
Nitrate			
FCE1.1	0.87 (0.82)	0.95	0.86
FCE11.2	2.26 (1.86)	0.99	0.95
FCE11.3	1.16 (1.01)	0.96	0.96
FCE13.3	1.17 (1.1)	0.87	0.79
FCE22.1	1.25 (1.18)	0.98	0.96
FCE26.1+2	1.04 (0.94)	0.96	0.94
Sulfate			
FCE1.1	0.66	0.88	0.79
FCE11.2	0.55	0.79	0.69
FCE11.3	0.79	0.89	0.88
FCE13.3	0.89	0.68	0.56
FCE22.1	0.82	0.94	0.94
FCE26.1+2	0.75	0.94	0.91
DOC			
FCE1.1	1.09a	0.83	0.67
FCE11.2	3.42a	0.86	0.88
FCE11.3	1.86a	0.89	0.92
FCE13.3	1.11a	0.72	0.69
FCE22.1	1.72a	0.87	0.79
FCE26.1+2	1.45a	0.89	0.86

a) DOC from MARGA not available. PM_{10} water-soluble organic carbon from Berner impactor used instead.

Figures

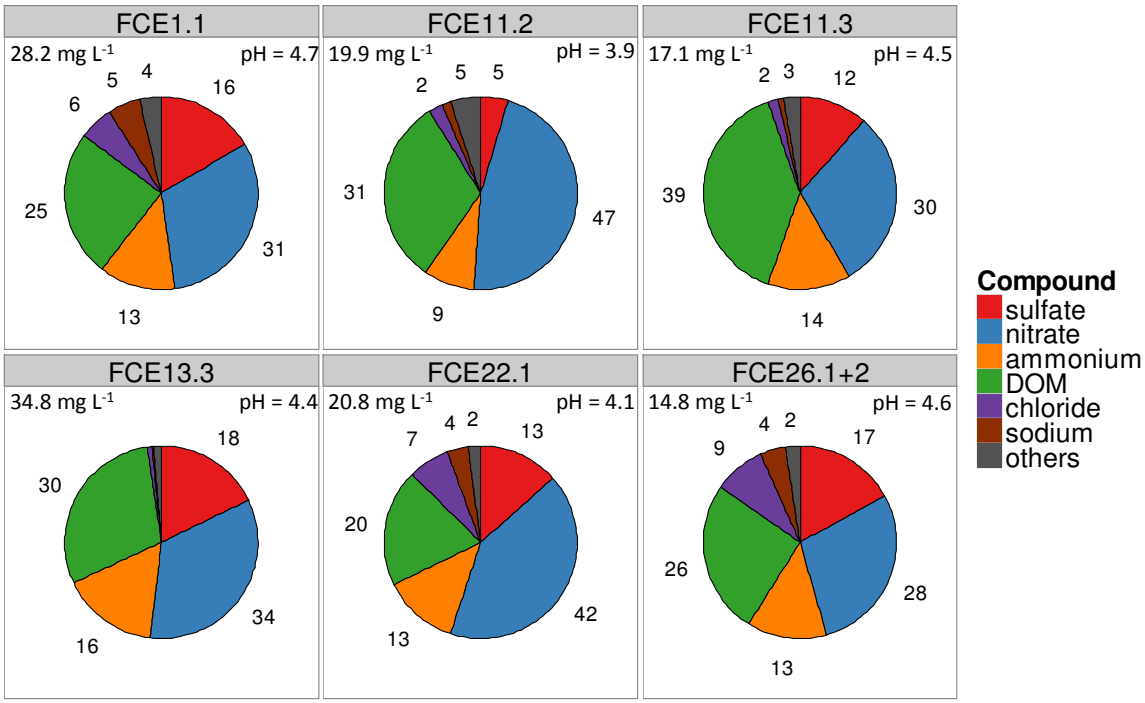

Figure 1: Volume-weighted mean composition of bulk cloud water during main events. Numbers represent percentage from total solute concentration (in mg L^{-1}). Trace solutes calcium, magnesium, potassium, H_2O_2(aq), and S(IV) are summarised as „others". DOM is calculated as DOC*1.8. Total solute concentrations and pH values derived from VWM H^+ concentrations are indicated in the upper left and reight panel corners, respectively.

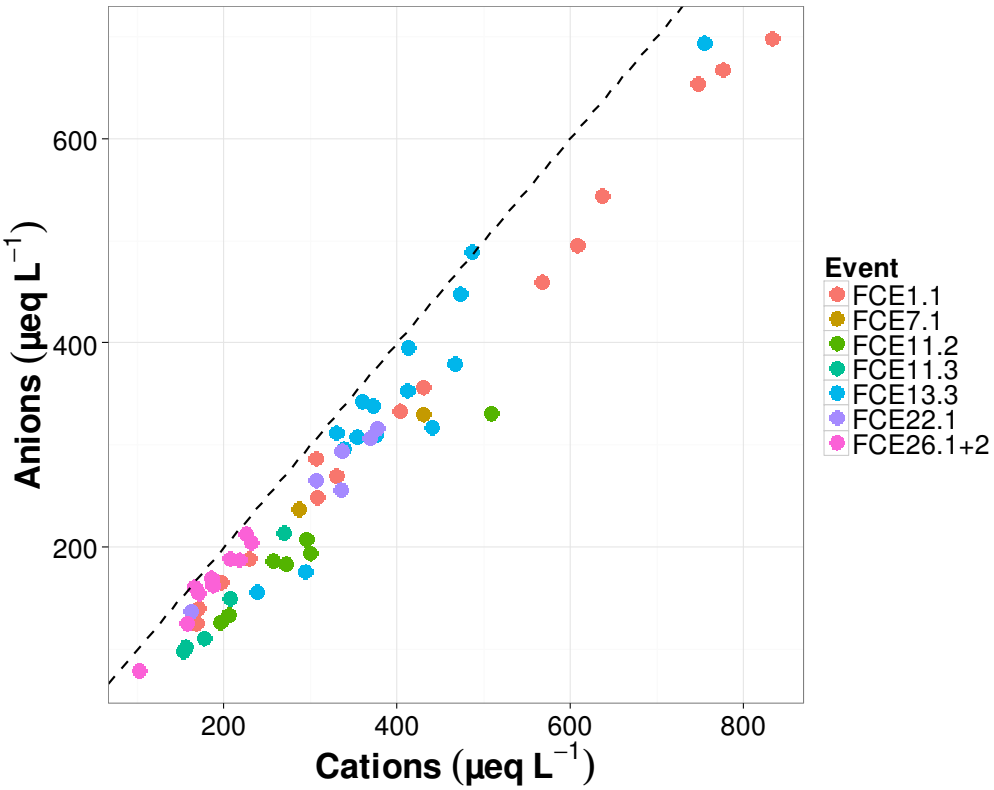

Figure 2: Ion balance on an equivalent basis for inorganic anions and cations. Dashed line is 1:1.

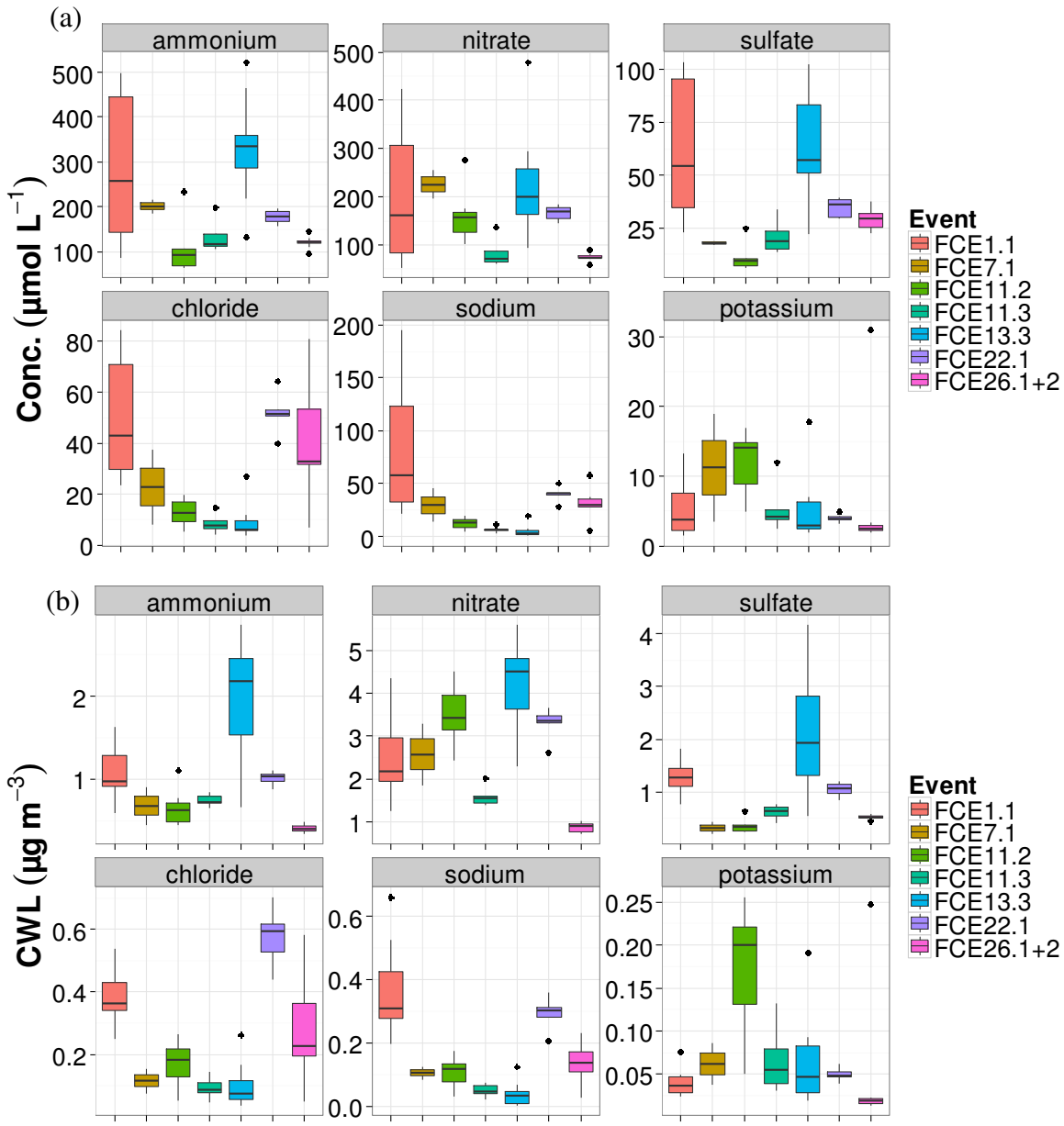

Figure 3: Variability of cloud water concentrations both within and between FCEs for selected inorganic ions. (a) Solute concentrations, (b) Cloud water loadings. Boxes indicate 25th, 50th, and 75th percentile, ~~while~~ whiskers extend to 1.5 * IQR (interquartile range)~~.~~, and dots indicate individual data points outside this range.

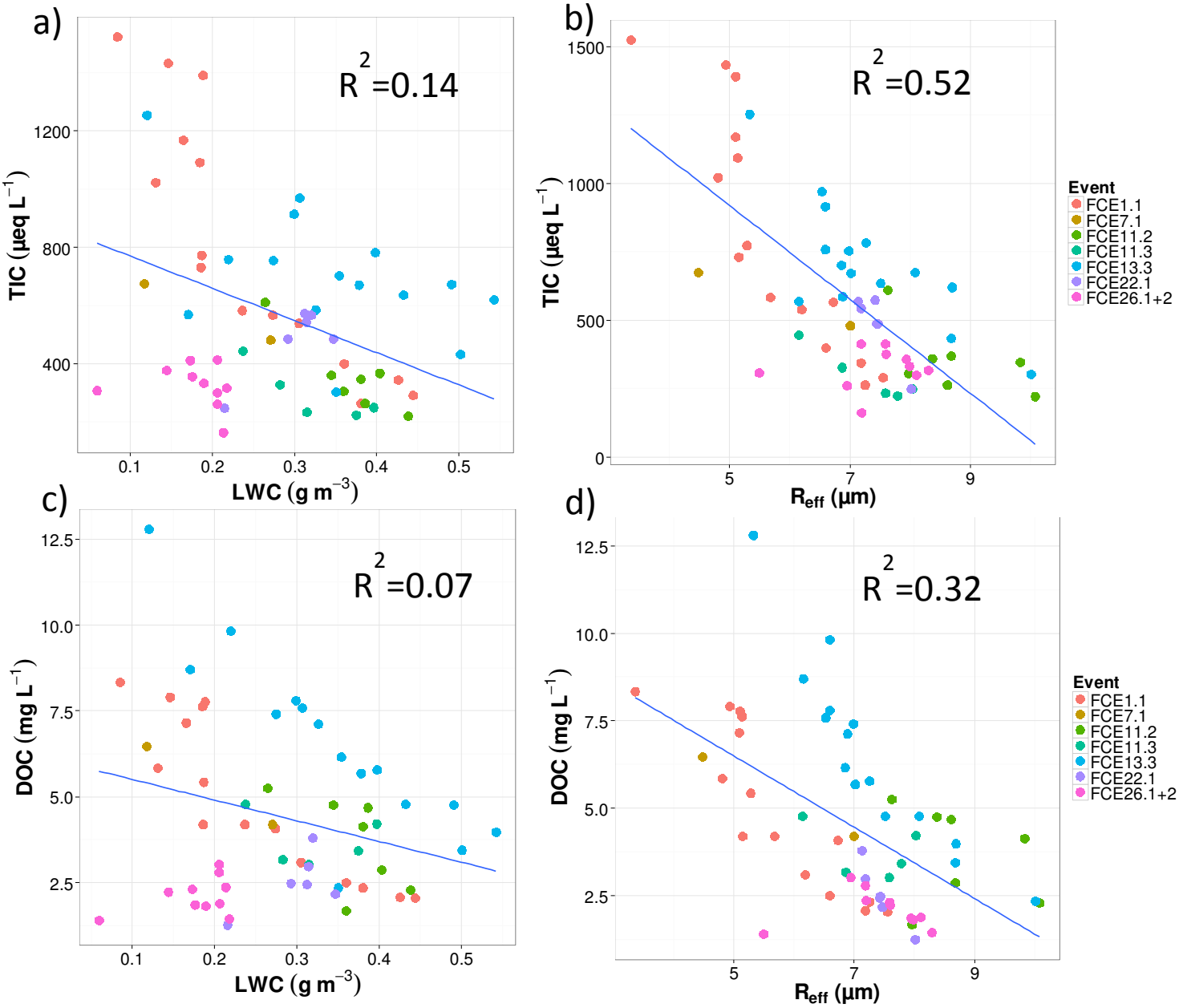

Figure 4: Relationships of total ionic content (upper panels) and dissolved organic carbon (lower panels) versus liquid water content (a and c) and effective droplet radius (b and d) for bulk cloud water samples.

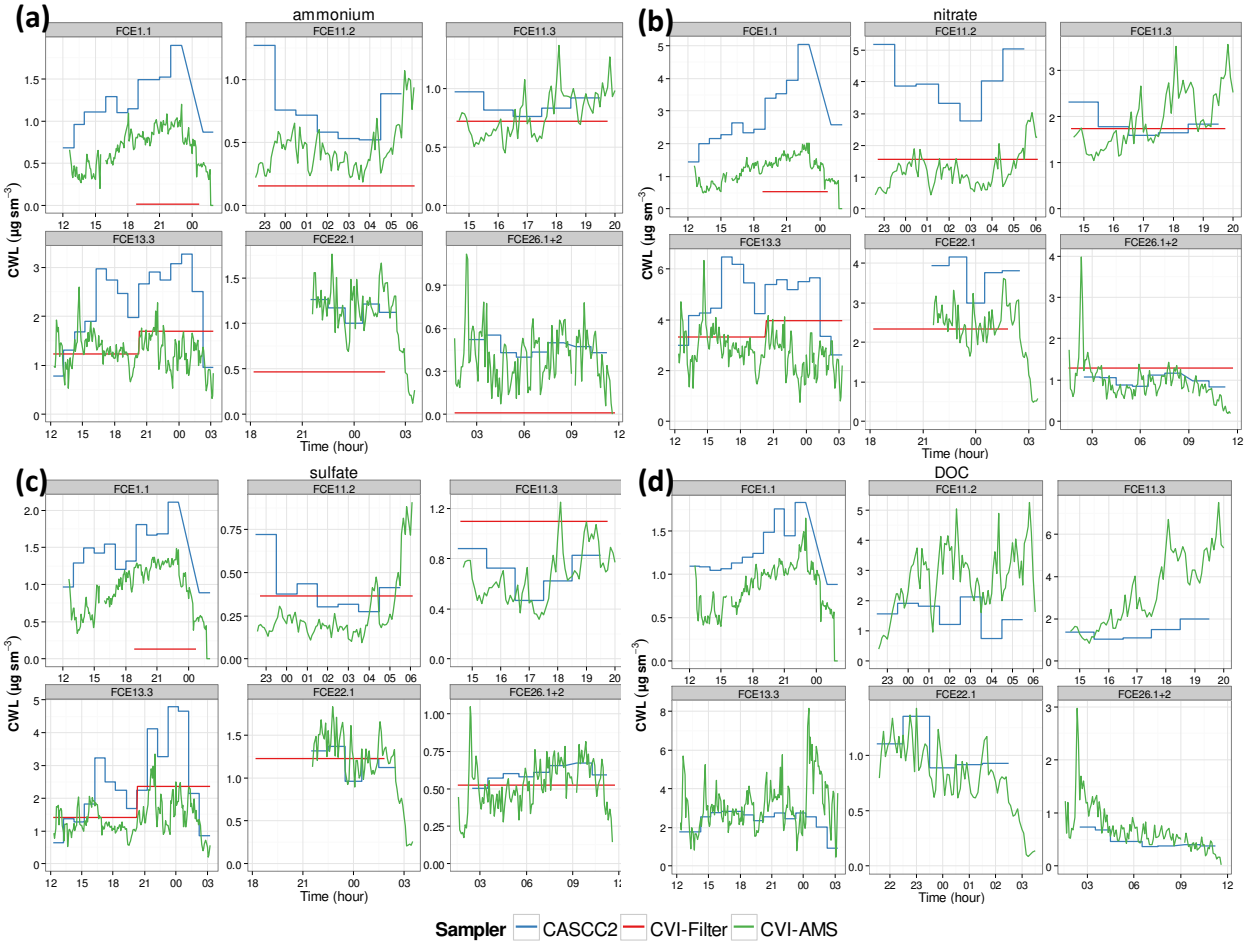

Figure 5: Comparison of cloud water loadings (normalised to standard temperature and pressure) from bulk cloud water collector (blue), quartz filter downstream CVI inlet (red), and AMS downstream CVI (green) for cloud water main constituents (a) ammonium, (b) nitrate, (c) sulfate, and (d) DOC (AMS organics/1.8).

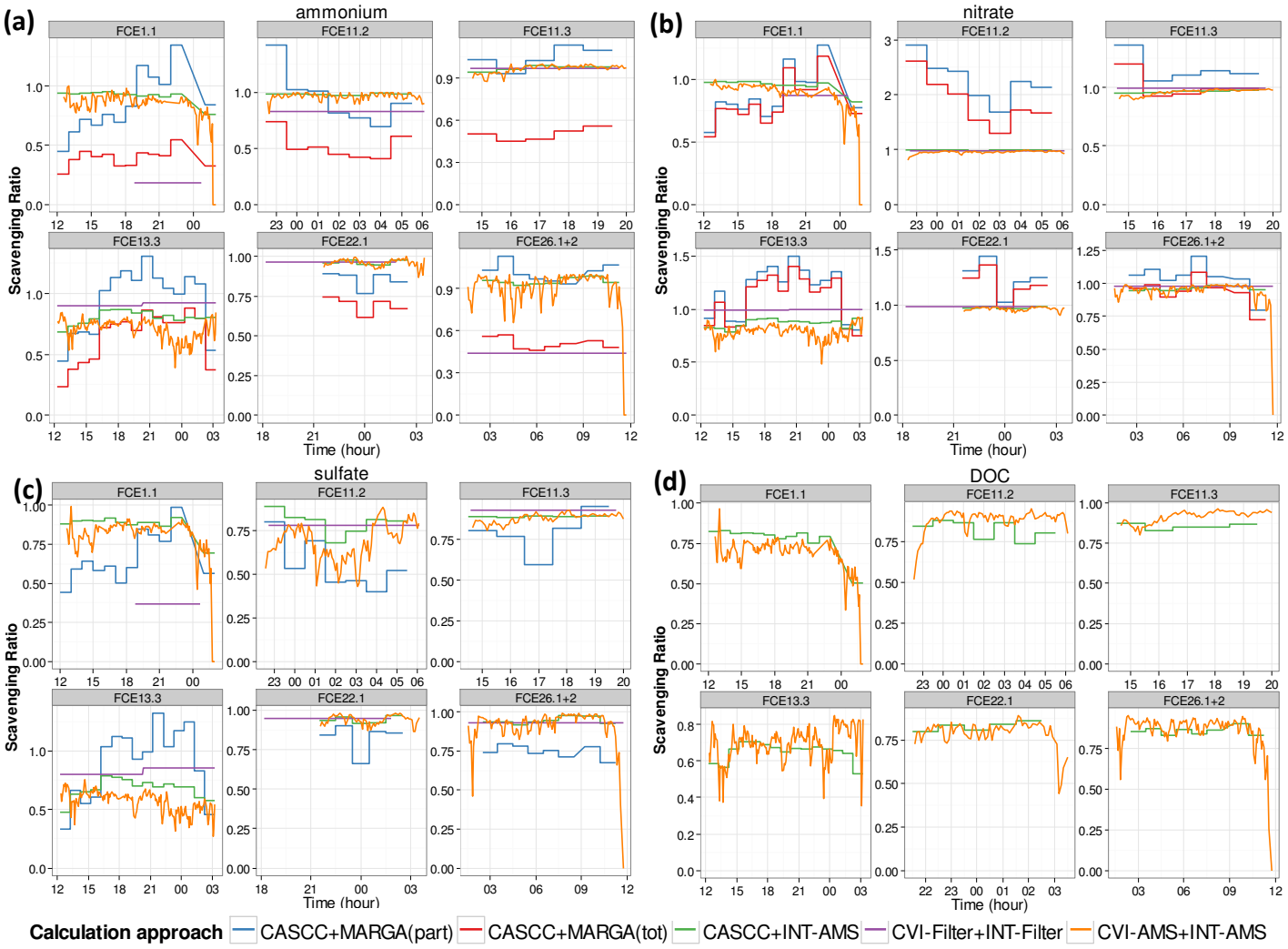

Figure 6: Cloud scavenging efficiencies for (a) ammonium, (b) nitrate, (c) sulfate, and (d) DOC, calculated as "upwind SE" from bulk cloud water loadings and upwind MARGA data (blue and red for MARGA particulate and total aerosol conentrations, respectively) and "in-cloud SEs" from bulk CWLs and interstitial AMS data (green), droplet residual and interstitial particle concentrations from filters (purple), and droplet residual and interstitial particle concentrations from AMS (orange). See text for details.

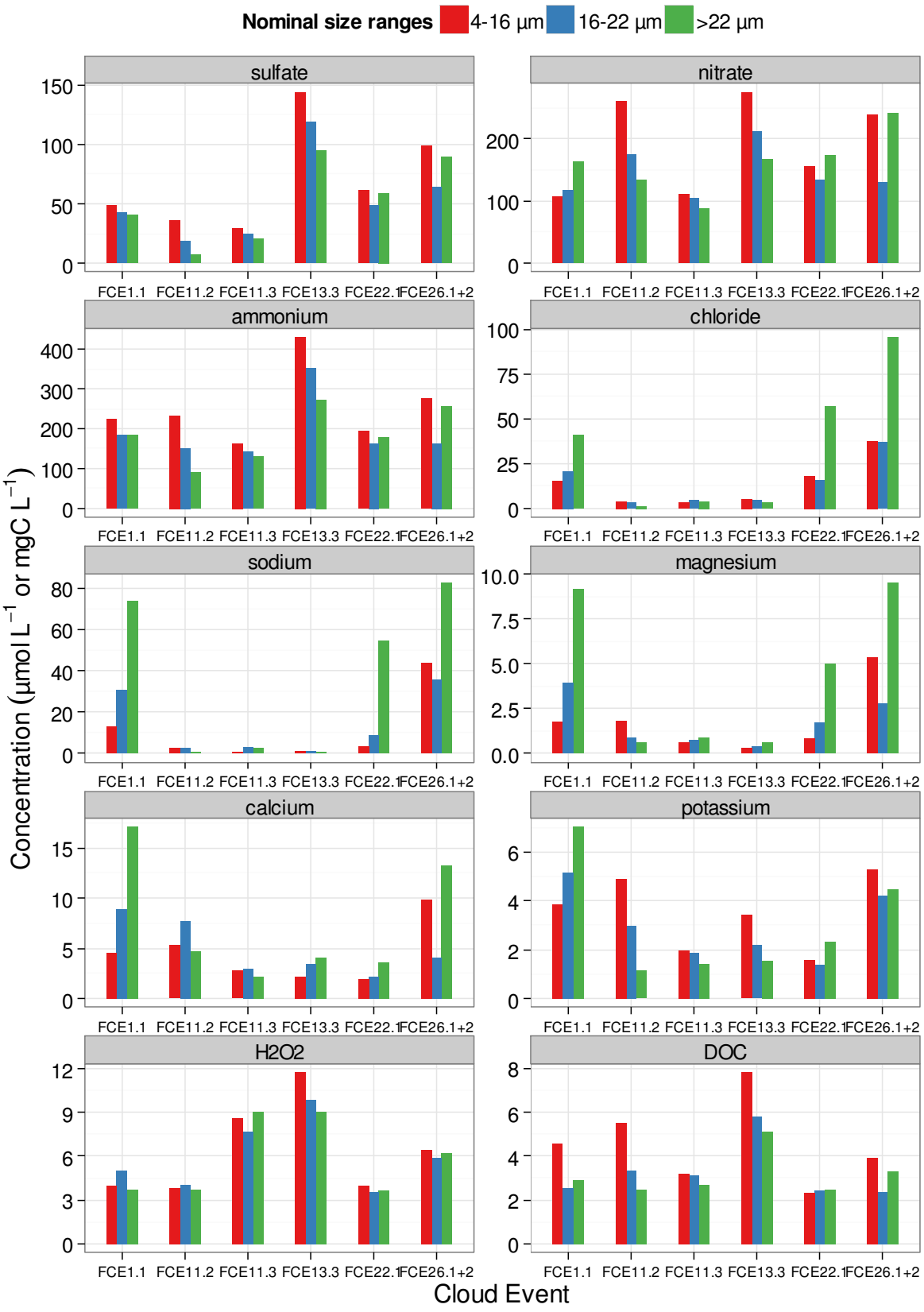

Figure 7: Size-resolved cloud water concentrations from 3-stage collector. Volume-weighted mean concentrations per event are given in µmol L^{-1} except for DOC (mgC L^{-1}).

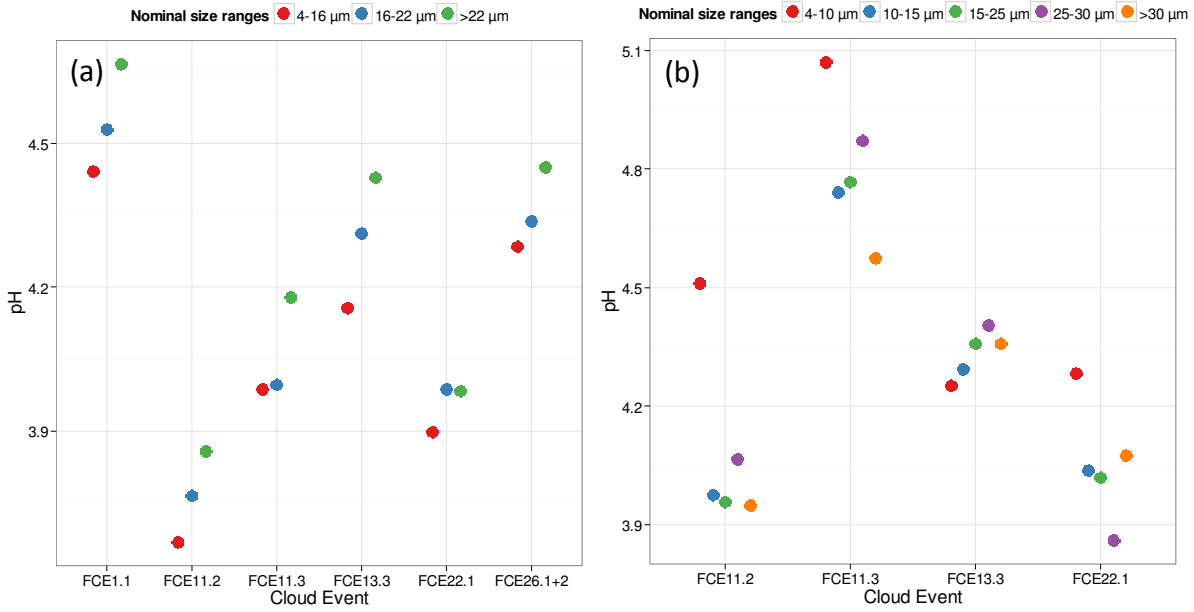

Figure 8: Mean pH values per event, calculated from volume-weighted mean concentrations of H^+ from (a) 3-stage cloud water collector, and (b) 5-stage collector.

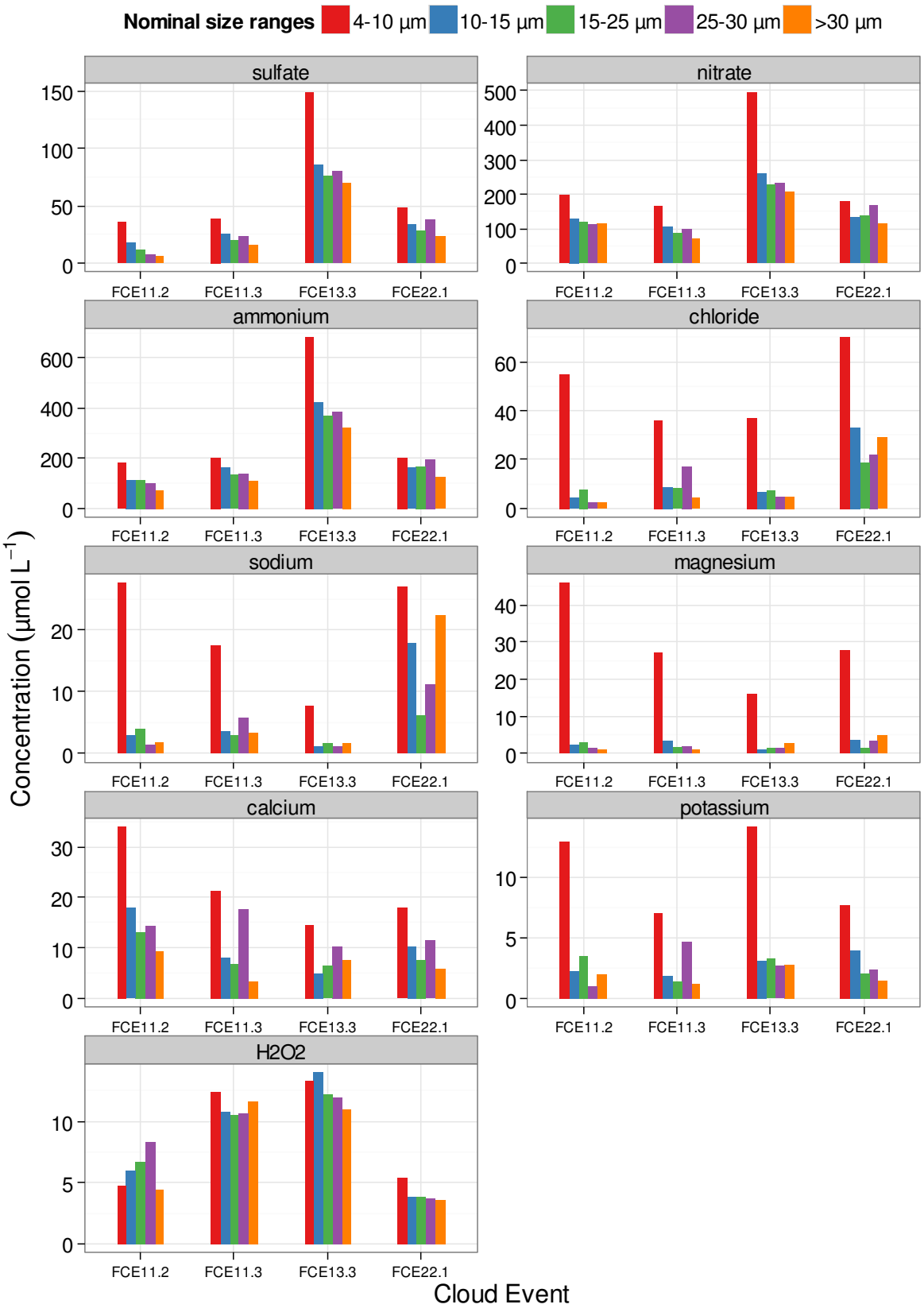

Figure 9: Size-resolved cloud water concentrations from 5-stage collector. Volume-weighted mean concentrations per event are given in µmol L^{-1}.