

1 Charging and Coagulation of Radioactive and Nonradioactive
2 Particles in the Atmosphere

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38

39 ABSTRACT

40 Charging and coagulation influence one another and impact the particle charge and size
41 distributions in the atmosphere. However, few investigations to date have focused on the
42 coagulation kinetics of atmospheric particles accumulating charge. This study presents three
43 approaches to include mutual effects of charging and coagulation on the microphysical evolution
44 of atmospheric particles such as radioactive particles. The first approach employs ion balance,
45 charge balance, and a bivariate population balance model (PBM) to comprehensively calculate
46 both charge accumulation and coagulation rates of particles. The second approach involves a

47 much simpler description of charging, and uses a monovariate PBM and subsequent effects of
48 charge on particle coagulation. The third approach is further simplified assuming that particles
49 instantaneously reach their steady-state charge distributions. It is found that compared to the
50 other two approaches, the first approach can accurately predict time-dependent changes in the
51 size and charge distributions of particles over a wide size range covering from the free molecule
52 to continuum regimes. The other two approaches can reliably predict both charge accumulation
53 and coagulation rates for particles larger than about 0.04 micrometers and atmospherically
54 relevant conditions. These approaches are applied to investigate coagulation kinetics of particles
55 accumulating charge in a radioactive neutralizer, the urban atmosphere, and an atmospheric
56 system containing radioactive particles. Limitations of the approaches are discussed.

57

58 **1. INTRODUCTION**

59 Atmospheric particles play an important role in airborne transport of contaminants, such as
60 radionuclides. Contaminants emitted from anthropogenic sources (e.g., nuclear plant accidents)
61 can be captured by background aerosols and then are transported together with pre-existing
62 particles. Contaminant-laden particles can be deposited onto the ground by dry and wet
63 deposition (primary contamination) and subsequently resuspended by wind or heat-driven
64 convection and moved to other areas (secondary contamination). For instance, due to these
65 atmospheric dispersion patterns, radioactive particles (e.g., ^{137}Cs) released during the Fukushima
66 accident were sampled in-situ 150 km away from the emission site (Yamauchi et al., 2012), and
67 also found in many places around the world (Hoeve and Jacobson, 2012). Similar dispersion
68 patterns of radioactive particles were observed after the Chernobyl accident (Yoshenko et al.,
69 2006a, 2006b). These examples suggest that particle deposition, which is largely affected by
70 aerosol microphysics, can determine the fate of contaminants during atmospheric transport. Thus,
71 accurate understanding of the microphysical behavior of atmospheric particles is necessary to
72 more accurately predict transport of contaminants (especially long-lived ones, such as ^{137}Cs), as
73 well as their potential environmental impacts.

74 The microphysical behavior of atmospheric particles is driven by such properties as charge and
75 size (Fuchs, 1989; Pruppacher and Klett, 1997). Atmospheric particles can acquire charge via
76 self-charging and diffusion charging. Self-charging refers to charge accumulation caused by
77 radioactive decay which typically leads to emission of electrons from particle surfaces. Diffusion
78 charging is attributed to diffusion of ions from the surrounding atmosphere onto the surface of
79 particles. Radioactive particles can be charged through these two charging mechanisms

80 (Greenfield, 1956; Yeh et al., 1976; Clement and Harrison, 1992; Clement et al., 1995;
81 Gensdarmes et al., 2001; Walker et al., 2010; Kweon et al., 2013; Kim et al., 2014; 2015) while
82 natural atmospheric particles are typically charged by diffusion charging (Hoppel, 1985; Yair
83 and Levin, 1989; Renard et al., 2013). Particle charging can modify not only the charge but also
84 the size distribution because charge on the particles modifies their coagulation rate coefficients
85 by generating electrostatic interactions (Fuchs, 1989; Tsouris et al., 1995; Chin et al., 1998).
86 Coagulation of atmospheric particles can influence their charging because the concentration of
87 atmospheric ions is affected by the particle size distribution (Yair and Levin, 1989), thereby
88 modifying the diffusion charging rates of the particles. Also, particle coagulation can result in
89 charge neutralization or accumulation on atmospheric particles (Alonso et al., 1998). These
90 effects imply that particle coagulation can influence the particle charge distribution. Thus,
91 particle charging and coagulation can mutually affect each other and simultaneously affect both
92 charge and size distributions in the atmosphere.

93 Theoretical and experimental investigations have been performed to examine the charging of
94 radioactive particles and background aerosols in the atmosphere. However, the effects of
95 coagulation of particles on the charge distribution have been frequently neglected by assuming
96 that the size distribution is constant while they are charged (Greenfield, 1956; Hoppel, 1985;
97 Yair and Levin, 1989). The assumption may be valid if the particle concentration is low or the
98 steady-state charge distribution is instantaneously attained (Hoppel, 1985; Renard et al., 2013;
99 Kim et al., 2015). If the timescale for particle charging is longer than that for particle coagulation,
100 the assumption may no longer be valid (Yair and Levin, 1989). Also, the effects of charging on
101 the particle size distribution are frequently neglected in aerosol transport models involving
102 microphysics of atmospheric particles. A possible reason for neglecting the charging effects may

103 be that the steady-state mean charge of atmospheric particles rarely may affect their coagulation
104 rates (Seinfeld and Pandis, 2006). However, neglecting electrostatic particle-particle interactions
105 may increase uncertainty of prediction results if particles can acquire multiple elementary
106 charges (e.g., radioactive particles). The simplified assumption of omitting electrostatic particle
107 interactions may create uncertainty in transport predictions of radioactive particles. Hence, it
108 may be necessary to take into account the mutual effects of particle charging and coagulation
109 processes in predicting the behavior of atmospheric particles carrying radioactive contaminants.

110 Previous attempts to consider charging effects include Oron and Seinfeld (1989a, 1989b), who
111 developed sectional approaches to simultaneously predict the behavior of charged and uncharged
112 atmospheric particles. Laakso et al. (2002) developed a general dynamic equation, including
113 charging and coagulation kinetics of atmospheric particles. Yu and Turco (2001) presented a
114 kinetic approach that involves diffusion charging and particle coagulation. However, the validity
115 of these approaches was not evaluated using analytical solutions. Alonso (1999) and Alonso et al.
116 (1998) developed analytical and numerical approaches to estimate time-dependent changes in the
117 size distributions of singly charged and neutral particles; thus, these approaches cannot be used
118 to investigate the coagulation kinetics of particles acquiring multiple elementary charges. Also,
119 none of these approaches considered self-charging; therefore, the aforementioned approaches
120 may be subject to error when they are used to simulate atmospheric dispersion of radioactive
121 plumes.

122 Our study presents three approaches to simultaneously predict time-dependent changes of the
123 charge and size distributions of radioactive and nonradioactive particles over a wide size range.
124 Development, validity, application, and limitations of these approaches are discussed.

125 2. MODEL DEVELOPMENT

126 2.1. Ion balance model

127 Many atmospheric processes can generate and remove ions in air. Typical ion sources in the
128 atmosphere involve natural and artificial radioactivity, as well as cosmic rays. Ions are generally
129 removed by ion-ion recombination and ion-particle attachment. Changes in ion concentrations by
130 these processes can be given by (Kim et al., 2015):

$$131 \frac{dn_{\text{ion}}^+}{dt} = -n_{\text{ion}}^+ \sum_k \sum_j \beta_{k,j}^+ N_{k,j} - \alpha_{rc} n_{\text{ion}}^+ n_{\text{ion}}^- + q, \quad (1)$$

$$132 \frac{dn_{\text{ion}}^-}{dt} = -n_{\text{ion}}^- \sum_k \sum_j \beta_{k,j}^- N_{k,j} - \alpha_{rc} n_{\text{ion}}^+ n_{\text{ion}}^- + q + q_e, \quad (2)$$

133 where n_{ion}^\pm refers to the number concentrations of positive or negative ions, the indices k and j
134 represent the size and number of elementary charges of particles, respectively, $\beta_{k,j}^\pm$ is the
135 attachment coefficient between a particle and an ion, $N_{k,j}$ is the number concentration of particles,
136 α_{rc} is the recombination coefficient of ions, and t is time. The first two terms of the right-hand-
137 side (RHS) of equations 1 and 2 represent the loss rate of ions due to ion-particle attachment and
138 ion-ion recombination, respectively. The third term denotes the production rate of ion pairs, q :

$$139 q = q_b + q_I, \quad (3)$$

140 where q_b is the ion production rate by cosmic rays and natural radioactivity, and q_I is the ion
141 production rate by radionuclides released by nuclear events. Electrons released by radioactive

142 decay are taken into account via the last term of eq 2. Changes in the ion concentrations may
 143 affect the electrical conductivity of the atmosphere, σ_{air} (Harrison and Carslaw, 2003):

$$144 \quad \sigma_{\text{air}} = e \left(\mu_+ n_{\text{ion}}^+ + \mu_- n_{\text{ion}}^- \right), \quad (4)$$

145 where e is the electrical charge and μ_{\pm} is the mobility of positive or negative ions. In eq 4, the
 146 terms in the parentheses of the RHS represent polar air conductivities.

147

148 2.2. Charge balance models

149 Self-charging generally accumulates positive charge on the surface of particles, while diffusion
 150 charging adds both positive and negative charges, indicating that the charging mechanisms can
 151 compete with one another. For radioactive particles involved in beta decay, time-dependent
 152 changes in their charge distributions due to competition of the charging mechanisms can be
 153 expressed by (Clement and Harrison, 1992; Kim et al., 2015):

$$154 \quad \frac{dN_{kj}}{dt} = A_{k,j-1}N_{k,j-1} - A_{k,j}N_{k,j} + \beta_{k,j-1}^+ n_{\text{ion}}^+ N_{k,j-1} - \beta_{k,j}^+ n_{\text{ion}}^+ N_{k,j} + \beta_{k,j+1}^- n_{\text{ion}}^- N_{k,j+1} - \beta_{k,j}^- n_{\text{ion}}^- N_{k,j}, \quad (5)$$

155 where A is the radioactivity of the radioactive particles. In eq 5, self-charging is represented by
 156 terms that include A , while diffusion charging is represented by terms with $\beta_{k,j}^{\pm}$. If the terms for
 157 self-charging are removed, eq 5 becomes identical to the charge balance model presented by
 158 Renard et al. (2013) who predicted electrification phenomena of aerosols in the real atmosphere.
 159 The mean value of the particle charge distributions can be given by:

160
$$\frac{dJ_k}{dt} = \frac{d}{dt} \left(\frac{\sum_j j N_{kj}}{\sum_j N_{kj}} \right). \quad (6)$$

161 The mean charge J of the radioactive particles of size k can also be approximated using a simple
 162 charge balance equation (Kim et al., 2015):

163
$$\frac{dJ_k}{dt} = A_k + \beta_{k,J}^+ n_{\text{ion}}^+ - \beta_{k,J}^- n_{\text{ion}}^-, \quad (7)$$

164 where A_k is the radioactivity of the size k particles. Similarly to eq 5, the mean charge
 165 accumulation rate of the radioactive particles (eq 7) depends on the competition between self-
 166 and diffusion charging. Eqs 5 and 7 indicate that the net charge of beta-emitting radioactive
 167 particles converges to a steady state where self-charging balances diffusion charging. The
 168 timescale, τ , needed to reach a steady state can be given by (Clement and Harrison, 1992):

169
$$\tau = \frac{1}{\beta^- \overline{n_{\text{ion}}^-}}, \quad (8)$$

170 where $\overline{\beta^-}$ is the mean attachment coefficient between negative ions and particles. At steady
 171 state, the mean charge of the radioactive particles can be approximated using (Clement et al.,
 172 1995):

173
$$J_k = \left\{ \begin{array}{ll} y - \left(\frac{y(X-1)}{\exp(2\lambda y) - 1} \right) & \lambda y > 0.22 \quad (9a) \\ y + \frac{X-1}{2\lambda} & \lambda y \leq 0.22 \quad (9b) \end{array} \right\}, \quad (9)$$

174 with $\lambda = \frac{e^2}{8\pi\epsilon_0\epsilon r_k k_B T}$, $y = \frac{\epsilon_0 A_k}{e\mu_- n_{\text{ion}}^0}$, $n_{\text{ion}}^0 = \sqrt{\frac{q}{\alpha}}$, $X = \frac{\mu_+ n_{\text{ion}}^+}{\mu_- n_{\text{ion}}^-}$,

175 where ϵ_0 is the vacuum permittivity, ϵ is the dielectric constant of the air, r is the radius of
 176 particles, k_B is the Boltzmann constant, T is the temperature, n_{ion}^0 is the mean ion concentration,
 177 and I is the ionization rate coefficient of beta-emitting radionuclides. Eq 9 suggests that the
 178 steady-state mean charge of beta-emitting radioactive particles is highly influenced by their size
 179 and decay rates, as well as the concentrations and mobilities of ions in air. The second term of
 180 the RHS of eq 9(b) represents charge accumulated only by diffusion charging; thus, it can be
 181 used to approximate the steady-state mean charge of nonradioactive particles, such as
 182 background aerosols that are externally mixed with radionuclides.

183 2.3. Population balance models

184 2.3.1. Bivariate population balance model

185 A bivariate population balance model, expressed in terms of particle volume x and charge j , can
 186 be used to predict effects of coagulation on time-dependent changes in the particle size and
 187 charge distributions. In the bivariate population balance model, the time-evolution of the number
 188 densities of charged and uncharged particles, n , due to coagulation, can be given by (Zebel, 1958;
 189 Oron and Seinfeld, 1989a&b):

190
$$\frac{\partial n(x, j)}{\partial t} = \frac{1}{2} \sum_{j'=-\infty}^{\infty} \int_0^x F_{j', j-j'}(x', x-x') n(x', j') n(x-x', j-j') dx'$$

$$- \sum_{j'=-\infty}^{\infty} \int_0^{\infty} F_{j, j'}(x, x') n(x, j) n(x', j') dx'$$
(10)

191 where F is the coagulation frequency ($\text{m}^3 \text{s}^{-1}$), which is also called the coagulation (rate)
 192 coefficient (Jacobson, 2005; Seinfeld and Pandis, 2006), and which can be obtained by
 193 multiplying the collision frequency and the collision efficiency. The two terms on the RHS of eq
 194 10 represent the production and loss rates of charged and uncharged particles by coagulation,
 195 respectively. A numerical solution of eq 10 can be obtained through the discretization of the
 196 integral terms, respectively (Oron and Seinfeld, 1989a&b). Vanni (2000) tested several sectional
 197 approaches and showed that the approach of Kumar and Ramkrishna (1996) is simpler and more
 198 accurate than other tested approaches, and also preserves mass and number of particles. Thus, the
 199 sectional approach of Kumar and Ramkrishna (1996) was used in this study to discretize the
 200 integral terms of eq 10, leading to the following discretized form:

$$201 \quad \frac{dN_{kj}}{dt} = \sum_{j'=-\infty}^{\infty} \sum_{\substack{l \geq m \\ x_{k-1} \leq x_l + x_m \leq x_{k+1}}} \left(1 - \frac{1}{2} \delta_{l,m}\right) \eta_{l,m} F_{l,m,j-j',j'} N_{l,j-j'} N_{m,j'} - \sum_{j'=-\infty}^{\infty} \sum_{l=1}^M F_{k,l,j,j'} N_{k,j} N_{l,j'}, \quad (11)$$

202 where indices l and m refer to the size bins, δ is the Kronecker delta, $\eta_{l,m}$ is a property
 203 distribution factor between two size bins given by Kumar and Ramkrishna (1996), and M is the
 204 total number of the size bins. If coagulation is induced by thermal energy (i.e., Brownian
 205 coagulation), the Brownian collision frequency β^{Br} ($\text{m}^3 \text{s}^{-1}$), which is also called the Brownian
 206 collision kernel (Jacobson, 2005), is given by (Fuchs, 1989):

$$207 \quad \beta_{kl}^{Br} = 4\pi (r_k + r_l) (D_{p,k} + D_{p,l}) \left(\frac{r_k + r_l}{r_k + r_l + \sqrt{g_k^2 + g_l^2}} + \frac{4(D_{p,k} + D_{p,l})}{(r_k + r_l) \sqrt{\bar{v}_{p,k}^2 + \bar{v}_{p,l}^2}} \right)^{-1}, \quad (12)$$

208 where D_p is the particle diffusion coefficient, g is the particle mean traveling distance, and \bar{v}_p is
 209 the particle thermal speed in air. Coagulation of charged particles is influenced by electrostatic

210 particle-particle interactions. This effect can be accounted for by multiplying the collision
 211 frequency with the collision efficiency, α^{Br} , which is also called the coalescence efficiency
 212 (Fuchs, 1989; Jacobson, 2005; Seinfeld and Pandis, 2006):

$$213 \quad \alpha_{kl}^{Br} = \frac{u}{e^u - 1} \quad (13)$$

$$214 \quad \text{with } u = \frac{j_k j_l e^2}{4\pi\epsilon_0 \epsilon (r_k + r_l) k_B T}$$

215 In eq 13, u indicates the relative importance between electrostatic potential energy and thermal
 216 energy in coagulation.

217 2.3.2. Monivariate population balance model

218 The time-evolution of the size distribution of particles can be estimated using a monivariate
 219 population balance model, with only the particle volume as the variable (Kumar and
 220 Ramkrishna, 1996):

$$221 \quad \frac{dN_k}{dt} = \sum_{\substack{l \geq m \\ x_{k-1} \leq x_l + x_m \leq x_{k+1}}} \left(1 - \frac{1}{2} \delta_{l,m}^{Kr} \right) \eta_{l,m} F_{l,m} N_l N_m - \sum_{l=1}^M F_{k,l} N_k N_l. \quad (14)$$

222 The coagulation frequency can be simply corrected using the mean charge of particles. However,
 223 the collision efficiency computed with the mean charge can be different from that with the
 224 particle charge distributions (Matsoukas, 1997). To include effects of the particle charge
 225 distributions on the coagulation frequency, eq (13) can be replaced by the average collision

226 efficiency $\bar{\alpha}$ (Clement et al., 1995), which involves interaction of all charged particles of size k
 227 with any charged particles of size l .

$$228 \quad \alpha_{kl}^{-Br} = 1 + \frac{\sum_{j_k, j_l \neq 0} N_{k, j_k} N_{l, j_l} (\alpha_{kl}^{Br} - 1)}{\sum_{j_k} N_{k, j_k} \sum_{j_l} N_{l, j_l}}. \quad (15)$$

229 The particle charge distributions needed to calculate $\bar{\alpha}^{Br}$ can be obtained by assuming a
 230 Gaussian distribution:

$$231 \quad N_{kj} = \frac{\sum_j N_{k, j}}{\sqrt{2\pi}\sigma_k} \exp\left(-\frac{(j - J_k)^2}{2\sigma_k^2}\right) \quad (16)$$

232 with

$$233 \quad \sigma^2 = y + \frac{1}{2\lambda}.$$

234 **2.4. Approaches to couple particle charging with coagulation kinetics**

235 Figure 1 shows three approaches which can be used to predict the time-evolution of the charge
 236 and size distributions of particles in the atmosphere. All the approaches can be used to simulate
 237 charging and coagulation kinetics of atmospheric particles carrying contaminants, including
 238 radioactive particles. Approach 1 is a rigorous scheme that simultaneously computes both charge
 239 accumulation and coagulation rates of particles using the ion balance model (eqs 1 and 2), the
 240 charge balance model (eq 5), and the bivariate population balance model (eq 11). Approach 2 is a
 241 simplified scheme of Approach 1, which can be used to predict the particle charge distribution
 242 using the mean charge of particles (eq 7) and the Gaussian distribution (eq 16). In order to easily

243 simulate the coagulation of charged particles, Approach 2 employs the monovariate population
244 balance model (eq 14) that corrects the collision frequency using the average collision efficiency
245 (eq 15). Approach 2 can be simplified to Approach 3 by assuming that charge accumulation rates
246 of particles instantaneously reach a steady state, with a timescale based on 5 times larger than τ
247 from eq 8. The steady-state particle charge distribution can be approximated by eqs 9 and 16. In
248 Approach 3, the collision frequency is multiplied by the average collision efficiency to include
249 the influence of electrostatic forces on coagulation. In this work, polarization of particles is not
250 taken into account.

251

252 **3. RESULTS AND DISCUSSION**

253 **3.1. Methods to simulate particle charging**

254 The three approaches attained above employ different methods to simulate charging of particles.
255 These methods were evaluated by comparing their prediction results with measurements
256 obtained using radioactive charge neutralizers (Liu and Pui, 1974; Wiedensohler and Fissan,
257 1991; Alonso et al., 1997) and radioactive particles (Gensdarmes et al., 2001). Initial conditions
258 for the simulations were determined from the measurements. The properties of ions observed
259 during the measurements are shown in Table 1. For the measurements providing the values of
260 ion mass, $\beta_{k,j}^{\pm}$ was calculated using Fuchs (1963) and Hoppel and Frick (1986). However, the
261 mass of ions was not measured during the experiments performed by Gensdarmes et al. (2001).
262 In these experiments, $\beta_{k,j}^{\pm}$ was estimated using analytical equations given by Gunn (1954) and
263 Harrison and Carslaw (2003).

264 *3.1.1. Diffusion-Charging Mechanism*

265 Figure 2 shows the steady-state charge distributions of nonradioactive particles over a wide size
266 range. Here, the particles were charged by the diffusion charging mechanism. For particles larger
267 than approximately 0.04 μm in diameter, the prediction results of all approaches were in good
268 agreement with the measurements [Figures 2 (a) and (b)]. Below 0.04- μm particle size,
269 Approach 1 accurately forecasted the particle charge distributions, but Approaches 2 and 3
270 underestimated the number concentrations of the negatively charged particles [Figure 2 (a)]
271 although the mean charge values of the particles given by all approaches were comparable.
272 Similar discrepancies were observed for the number concentrations of the positively charged
273 particles smaller than about 0.025 μm (not shown).

274 Analysis of the discrepancies suggests that they originate from the standard deviation involved in
275 the Gaussian distribution (eq 16). At a given temperature, the width of the particle charge
276 distributions can be significantly influenced by three parameters: the particle size, ion mass, and
277 ion mobility (Wiedensohler and Fissan, 1991). In Approaches 2 and 3, however, the effects of
278 the ion properties are not involved, so particle size primarily drives the standard deviation, which
279 can differ from what Approach 1 gives. When Approaches 2 and 3 used the standard deviation
280 values obtained by Approach 1, their simulation results became closer to the measurements,
281 although the discrepancies are still seen for negatively and positively charged particles smaller
282 than about 0.02 μm .

283

284

285 *3.1.2. Competition of Self-Charging and Diffusion-Charging Mechanisms*

286 In our previous work (Kim et al., 2014; 2015), it has been shown that Approaches 1 and 3 can
287 reliably simulate charging of radioactive particles. Thus, in this study, we focused on evaluating
288 the validity of Approach 2 with the experiments of Gensdarmes et al. (2001) who measured the
289 charge distributions of ^{137}Cs particles under various ionizing conditions. Ionizing rates of air
290 molecules were estimated using a linear energy transfer equation for energetic electrons emitted
291 by beta decay (Kim et al., 2015). Results of Approach 1 were included as a reference.

292 Figure 3 shows the charge accumulation on radioactive particles under two ionizing conditions:
293 $q_I = 3.7 \times 10^8 \text{ m}^{-3} \text{ s}^{-1}$ and $q_I = 7.1 \times 10^6 \text{ m}^{-3} \text{ s}^{-1}$. Approach 2 predictions are in good agreement with
294 observations and Approach 1 values. During the measurements, the self-charging rate of the
295 radioactive particles was constant because of the long half-life of ^{137}Cs (approximately 30 years),
296 suggesting that changes in their charge accumulation rates may be dominated by diffusion
297 charging rates. The ion concentrations in air can rapidly increase at the high ionizing rate
298 considered, suggesting that the diffusion charging rate of the ^{137}Cs particles quickly increased
299 and then became comparable to their self-charging rate (eq 7). The charge accumulation on the
300 radioactive particles promptly reached a steady-state value, and the particle charge distribution
301 was similar to the initial condition (Figure 4). In contrast, the time required to reach the steady-
302 state value was much longer at the low ionizing rate considered; hence, the particle charge
303 distribution shifted to the right (see Figure 4), i.e., more positive charge. The agreement observed
304 in Figure 4 between simulation results by Approach 2 and experimental data by Gensdarmes et al.
305 (2001) suggests that Approach 2 can accurately forecast the competition between self- and

306 diffusion charging on submicron particles carrying radionuclides, and precisely predict the
307 particle charge distributions.

308 *3.1.3. Timescale to Reach Steady-State Charge Accumulation Rate*

309 To evaluate the steady-state assumption of particle charging for atmospheric conditions, the
310 timescale for reaching steady-state (eq 8) is evaluated with Approaches 1 and 2. Figure 5 shows
311 time-dependent changes in the concentrations of negatively charged particles under two different
312 initial conditions of Alonso et al. (1997) who measured the charge distributions of particles of a
313 few nanometers. All particles were initially uncharged or negatively charged. Because the
314 particle size was very small, Approach 1 was used to predict the time-evolution of the particle
315 concentrations. As time elapsed, the initially uncharged particles became negatively charged by
316 capturing negative ions. The diffusion of positive ions led to the discharging of the initially
317 negatively charged particles. For the initial conditions used, the charging and discharging rates of
318 the particles reached a steady state after approximately 0.2 s, respectively. This
319 charging/discharging behavior predicted by Approach 1 is in good agreement with the
320 measurements of Alonso et al.(1997). However, the timescales obtained from eq 8 are shorter
321 than the prediction results, as well as the measurements, because Approach 1 and the
322 observations provided exact timescales, while τ in eq 8 is a scaling parameter. Similar results
323 were also obtained for different initial conditions, as well as for the ^{137}Cs particles (see Table 2).
324 Eq 8 is based on the assumptions that (i) $X \approx 1$, (ii) all particles are initially uncharged, and (iii)
325 the ion concentrations are constant (Clement and Harrison, 1992). As seen in Figure 6, when all
326 the assumptions were applied, the diffusion charging rate of the radioactive particles became -
327 $8.1 \times 10^{-3} \text{ s}^{-1}$ at the timescale provided by eq 8, which corresponds to approximately 63% of the

328 steady-state self-charging rate. If the timescale is increased by a factor of five, the diffusion
329 charging rate reaches about 99 % of the steady-state self-charging rate. Similar results were
330 obtained for other cases with radioactive particles. Thus, eq 8 is valid if the assumptions can be
331 used, and a reliable timescale to reach a steady state (e.g., 99%) can be obtained by multiplying
332 the equation by a factor of 5. However, because the assumptions cannot be used in typical
333 atmospheric conditions, such as $X \neq 1$ (Harrison and Carslaw, 2003), the steady-state assumption
334 of Approach 3 should be evaluated using Approaches 1 or 2.

335 So far, we have evaluated the validity of the methods used in the three approaches to predict
336 charge accumulation on atmospheric particles. The evaluation results suggest that the method
337 employed in Approach 1 can accurately simulate charging of particles in the free molecule ($d_p <$
338 $0.01 \mu\text{m}$), transition ($d_p = 0.01 - 0.2 \mu\text{m}$), and continuum ($d_p > 0.2 \mu\text{m}$) regimes. The methods
339 used in Approaches 2 and 3 can reliably forecast charging rate of atmospheric particles larger
340 than $0.04 \mu\text{m}$.

341

342 **3.2. Validity of the three approaches to couple particle charging with coagulation**

343 *3.2.1. Bivariate Population Balance Model for Approach 1*

344 Based on the numerical approach of Alonso et al.(1998), Alonso (1999) suggested an analytical
345 approach to simultaneously investigate charging and coagulation kinetics of nonradioactive
346 particles, smaller than $0.02 \mu\text{m}$ in diameter. Results of the analytical approach agreed well with
347 those of the numerical approach, but the applicability of both analytical approaches may be
348 limited as discussed in section Introduction. The analytical approach, however, was found to be

349 useful to validate numerical solutions of population balance equations including diffusion
350 charging and coagulation (Alonso, 1999). In this study, the analytical approach of Alonso (1999)
351 was used to evaluate Approach 1 because Approaches 2 and 3 are not applicable to particles
352 smaller than 0.02 μm , as shown in Figure 2a. In integrating eq 11, an equidistant diameter grid
353 was used for discretization. Particles were assumed to be initially uncharged, monodispersed (d_p
354 = 0.003 μm , $N_i = 10^{15} \text{ m}^{-3}$), with a constant ion concentration. The mobility of negative ions was
355 slightly greater than that of positive ions, according to the properties taken from Alonso et al.
356 (1997) (See Table 1). Figure 7 shows the time dependent concentration of particle charge classes.
357 The concentration of uncharged 0.003- μm particles decreased with time because of loss due to
358 coagulation and charging by captured ions. However, the loss of particle concentration caused by
359 diffusion charging was negligible because the ion-particle attachment coefficient was small,
360 suggesting that the time-evolution of the charged-particle concentration depends on the
361 uncharged-particle concentration. More negative than positive ions were captured by the
362 uncharged 0.003- μm particles and, thus, the concentration of the negatively charged 0.003- μm
363 particles is slightly higher than that of the positively charged 0.003- μm particles. The
364 concentrations of larger particles (e.g., charged and uncharged 0.006- μm particles) increased
365 over time because of the size growth of small particles due to coagulation, as well as the
366 diffusion-charging mechanism. These evolution patterns predicted by Approach 1 were in good
367 agreement with the prediction results given by the analytical approach. As can be seen in Figures
368 2 and 5, the ion balance and charge balance models of Approach 1 accurately predicted the
369 diffusion charging of nanoparticles, suggesting that the numerical solution of the bivariate
370 population balance model (eq 11) reliably predicts coagulation of particles acquiring charge.

371

372 3.2.2. Average Collision Efficiency of Approaches 2 and 3

373 Approaches 2 and 3 employ an average collision efficiency and are coupled to the monivariate,
374 instead of the bivariate, population balance model. These approaches provided accurate particle
375 charge distributions for various cases (e.g., Figures 2 and 4; Kim et al., 2014), suggesting that
376 their validity may be highly influenced by the accuracy of the average collision efficiency. Thus,
377 we compared simulation results of Approach 2 with those of Approach 1 to check if the average
378 collision efficiency (eq 15) can appropriately account for the influence of the charge distributions
379 of particles on their size growth via coagulation. For comparison, simulation results of Approach
380 2 using the mean charge (eq 13), as well as those for uncharged particles, were included.

381 Similarly to Oron and Seinfeld (1989 a&b), we assumed monodispersed initial size distributions
382 ($d_p = 0.1 \mu\text{m}$, $0.5 \mu\text{m}$, $1 \mu\text{m}$, $N_t = 10^{13} \text{ m}^{-3}$, and $n_{\text{ion}}^0 = 10^{16} \text{ m}^{-3}$). The geometrical grids ($x_{k+1} = 2x_k$)
383 were used to cover a wide particle-size range. Other basic assumptions were similar to those
384 considered for the validation test of Approach 1.

385 Figure 8 shows the time-evolution of the particle size distributions induced by particle charging
386 and coagulation. The simulation conditions led to the accumulation of more negative than
387 positive charges on the particles. At $t = 1 \text{ min}$, the number fraction of the negatively charged
388 particles was 0.7, while that of the positively charged and uncharged particles was 0.16 and 0.14,
389 respectively. Thus, the size growth of the particles by coagulation was suppressed due to the
390 generation of strong repulsive electrostatic forces (Approaches 1 and 2 vs Uncharged).

391 While most particles were negatively charged, some particles captured positive ions. Owing to
392 electrostatic attractive forces, the positively charged particles can more frequently coagulate with
393 the negatively charged particles and grow. Therefore, the coagulation rates predicted by

394 Approach 2 with the average collision efficiency were slightly higher than those for the case
395 assuming that all particles were negatively charged [Approach 2 (eq 13) vs Approach 2 (eq 15)].
396 These coagulation patterns predicted by Approach 2 using the average collision efficiency were
397 in good agreement with those given by Approach 1 [Approach 1 vs Approach 2 (eq 15)], as well
398 as the particle charge distributions in various size ranges (Figure S1). Similar results were
399 obtained for different initial particle size distributions ($d_p = 0.1 \mu\text{m}$; $d_p = 1 \mu\text{m}$). Thus, the
400 monovariate population balance model using the average collision efficiency can be used to
401 simultaneously investigate charging and coagulation kinetics of atmospheric particles. These
402 validation tests suggest that all the approaches developed can be used to reliably couple particle
403 charging with coagulation kinetics of atmospheric particles.

404

405 **3.3. Applications**

406 *3.3.1. Radioactive Neutralizer*

407 Radioactive neutralizers are typically used to control the charge of atmospheric particles in many
408 laboratory-scale experiments. The applicability of the three approaches to studies using
409 radioactive neutralizers was evaluated using the experiments of Alonso et al. (1998) who
410 measured the size distribution of nanometer-size particles passing by a ^{241}Am radioactive
411 neutralizer under various residence times.

412 Figure 9 shows the size distribution of negatively charged particles when the residence time, $t =$
413 0.318 s . All particles were initially uncharged, but some particles became charged by capturing
414 positive and negative ions in the neutralizer. Approach 1 accurately predicted the size

415 distribution of the negatively charged particles, while the predictions by Approaches 2 and 3
416 were different from the measurements. The particle size distributions predicted by all the
417 approaches were similar (not shown). It can be concluded that the Gaussian distribution used in
418 Approaches 2 and 3 cannot accurately predict the charge distributions of very small particles
419 (See Figure 2). Thus, Approaches 2 and 3 should not be used for particles smaller than 0.04 μm .

420 As shown in Figures 2-5, Approach 1 can accurately predict the charge accumulation rate of
421 radioactive and nonradioactive particles in the free-molecule, transition, and continuum regimes.
422 Approach 1 employs the interpolation formula of Fuchs that can be used to compute the collision
423 frequency of the particles in these regimes, revealing that this approach can also precisely predict
424 the charge distribution of larger particles undergoing coagulation. These results suggest that
425 Approach 1 can be a reasonable option to simultaneously simulate charging and coagulation of
426 particles of any size in laboratory-scale experiments.

427 *3.3.2. Charging and Coagulation of Nonradioactive Particles in Urban Atmosphere*

428 Hoppel (1985) simulated charging of 0.06- μm urban aerosols by diffusion charging; however,
429 effects of coagulation on their steady-state charge distribution were excluded from the simulation.
430 Changes in the particle charge and size distributions by charging and coagulation were
431 investigated in this work on the basis of the simulation of Hoppel (1985) for comparison. The
432 simulation time was approximately 100 min, but for completeness, we repeated and extended the
433 Hoppel (1985) simulation to 1 day. The extended results were compared with prediction results
434 of Approach 1, which involves the effects of coagulation on the particle charge distribution. It
435 was assumed that cosmic rays and natural radioactivity generate ion pairs in the atmosphere,
436 giving $q_b \approx 10^7 \text{ m}^{-3} \text{ s}^{-1}$ (Hoppel, 1985).

437 Figure 10 presents changes in the particle charge and size distributions vs time. The simulation
438 results performed by Hoppel (1985) showed that the particle charge distribution approached its
439 steady-state value after approximately 90 min. However, as time elapsed, the particles grew in
440 size due to coagulation. The size growth led to the generation of large particles capturing many
441 ions, thereby modifying the particle charge distribution.

442 The simulation results of Hoppel (1985) also indicated that the ion concentrations became
443 unchanged after reaching a steady state. However, coagulation reduced the particle number
444 concentrations which can affect the loss rate of ions by diffusion charging (see eqs 1 and 2). The
445 reduction in the particle concentrations increased the ion concentrations, thereby enhancing the
446 electrical conductivity of the postulated atmosphere (Figure 11). The ion concentrations and
447 electrical conductivity are expected to increase until ion-ion recombination becomes the major
448 ion removal mechanism. These results suggest that coagulation can affect the electrical
449 properties in the atmosphere, as well as the particle charge distribution.

450 *3.3.3. Charging and Coagulation of Radioactive Particles in the Atmosphere*

451 *3.3.3.1. Comparison with results by Greenfield (1956)*

452 Nuclear events can release particles carrying radionuclides. Greenfield (1956) simulated time-
453 evolution of the charge distribution of 0.1 μm radioactive particles emitting energetic electrons at
454 6-km altitude. Because Greenfield (1956) assumed that the particle size distribution is constant
455 for 4 hours, the influence of coagulation on the particle charge distribution was evaluated using
456 the simulation conditions postulated by Greenfield (1956).

457 Figure 12 shows changes in the particle charge distributions vs time. Both self-charging and
458 diffusion charging influenced the charge accumulation on radioactive particles. Due to many ion
459 pairs produced by beta radiation (Figure S2), positive charge accumulated on the particles by
460 self-charging was rapidly neutralized by capturing negative ions. Thus, the particle charge
461 distribution given by Approach 1 was slightly shifted to the right of the zero elementary charge
462 in Figure 12(a), although large particles with a high level of radioactivity were generated by
463 coagulation. As time elapsed, the particle charge distribution was slightly moved to the left.
464 Because the decay rates of the highly radioactive particles were reduced over time, their self-
465 charging rates also decreased, and this led to the slight movement of the charge distribution to
466 the left in Figure 12(a).

467 The discrepancies between the predictions of Approach 1 and Greenfield (1956) result mainly
468 from the ion-particle attachment coefficient used in the simulation. The values assumed by
469 Greenfield (1956) were beyond the ion-particle attachment coefficient found by other researchers
470 (e.g., Hoppel and Frick, 1986), leading to the discrepancies observed (Kim et al., 2015).

471 Beta radiation caused by radioactive decay rapidly increased the ion concentrations, thereby
472 enhancing the electrical conductivity in the atmosphere (Figure S2). In contrast to the case
473 shown in Figure 11, the ion concentrations and air conductivity significantly decreased with time
474 because the ionization rate of air molecules decreased considerably, and ion-ion recombination
475 was responsible for the change in the concentrations. Nevertheless, the air conductivity
476 enhancement by beta radiation was much higher than that by cosmic rays and natural
477 radioactivity.

478 After the Chernobyl and Fukushima accidents, short- and long-range transport of particles
479 carrying radionuclides, such as ^{137}Cs and ^{134}Cs , affected the electrical properties of the local
480 atmosphere in many places (Israelsson and Knudsen, 1986; Yamauchi et al., 2012). In particular,
481 beta radiation led to significant changes in the electrical conductivity and potential gradient in
482 the local atmosphere (Kim et al., 2015). Israelsson et al. (1987) suggested that an increase in the
483 electrical conductivity led to enhancement of lightning activities at radioactively contaminated
484 sites in Sweden. These observations reveal that the approaches developed in this study can be
485 employed to investigate the influence of radionuclides on electrification phenomena in the
486 atmosphere.

487 3.3.3.2. Steady-state assumption of radioactive particle charging

488 The steady-state assumption of particle charging can be useful to simulate coagulation of
489 radioactive particles in model studies of radioactivity transport. Charging and coagulation
490 kinetics of radioactive particles were investigated using Approaches 2 and 3 to evaluate the
491 validity of the steady-state assumption of radioactive particle charging. For comparison, the size
492 growth of particles by coagulation was simulated by assuming the Boltzmann charge distribution.

493 We used the simulation condition employed to validate the average collision efficiency, but
494 additionally presumed that radioactive decay of ^{134}Cs is responsible for the ionization of air
495 molecules. The specific radioactivity of ^{134}Cs was obtained from Clement and Harrison (1992).
496 The ionization rate of ^{134}Cs was estimated according to Kim et al. (2015). Under these conditions,
497 eq 8 revealed that $5\tau \approx 4.3$ ms. Thus, we assumed that charge accumulation rates of ^{134}Cs
498 particles instantaneously reach steady state, and evaluated this assumption for the simulation
499 conditions of $X \approx 0.7$ and ion concentration given by $q_I = I_{\text{Cs-134}} \times A_{\text{Cs-134}} \times N_I$.

500 Figure 13 shows the charge and size distributions of the ^{134}Cs particles after 2 hours of evolution.
501 The prediction results of Approach 3 were different from those of the case assuming the
502 Boltzmann charge distribution, but agreed well with those of Approach 2, suggesting that the
503 steady-state assumption of radioactive particle charging can be valid if τ is small. We also tested
504 the assumption of Approach 3 using different initial conditions (e.g., $d_p = 0.3 \mu\text{m}$), and the
505 agreement was still maintained (not shown).

506

507 **3.4. Computational costs**

508 The computational costs to predict transport of particles containing contaminants depends on the
509 number of ordinary differential equations (ODEs) solved during simulation. Thus, the number of
510 ODEs involved in the three approaches was evaluated by assuming 30 size bins, which
511 corresponds to those used in the two-moment aerosol sectional microphysics model, covering
512 particle diameters from $0.01 \mu\text{m}$ to $10 \mu\text{m}$ (Adams and Seinfeld, 2002).

513 Table 3 shows an example of the computational costs of the three approaches. For Approach 1,
514 we assumed that atmospheric particles can acquire up to fifteen elementary charges regardless of
515 their sign, thereby resulting in 932 ODEs. Because Approaches 2 and 3 employed the
516 monovariate population balance model, a fewer number of ODEs were involved in these
517 Approaches than in Approach 1, suggesting that they are computationally more efficient. For
518 instance, compared to Approach 1, Approaches 2 and 3 more quickly computed the charge
519 accumulation and coagulation rates of urban aerosols.

520 A simple way to reduce the number of ODEs included in Approach 1 is to assume that
521 atmospheric particles acquire only a few electrical charges. For example, Laakso et al. (2002)
522 assumed that submicron particles can acquire elementary charges from -5 to +5. This assumption
523 can be valid if the particle size is small [see Figure 2 (a)]. However, when the simulation
524 conditions were used, we observed loss of submicron particles because they can acquire more
525 elementary charges (Figure S3). To preserve mass and charge, one may optimize the minimum
526 number of elementary charges using the charge balance model of Approach 1, and then begin the
527 simulation of charging of particles undergoing coagulation.

528 Approach 3 includes all the physics of charging and coagulation to predict the particle
529 size/charge distribution, but, compared to Approaches 1 and 2, it is computationally more
530 suitable for use in a 3-D global transport model to predict the transport of radioactivity in the
531 environment after a radiological event such as a nuclear plant accident.

532

533 **4. CONCLUSIONS**

534 Understanding the behavior of atmospheric particles is important to accurately predict short- and
535 long-range transport of contaminants. Particle charging and coagulation processes can strongly
536 affect the behavior of atmospheric particles because these processes can change their important
537 physical and electrical properties, such as size and charge. This study has shown three
538 approaches with a wide range of complexity and applications to involve the mutual effects of
539 charging and coagulation processes in the simulation of particle charge and size distributions vs
540 time. Depending on the initial conditions, these approaches can be employed to accurately

541 predict the behavior of atmospheric particles carrying radioactive contaminants. We have shown
542 the approaches to be applicable to a wide variety of atmospheric (laboratory and field)
543 applications. The accuracy of the approaches depends on the assumptions made to reduce
544 computational cost. The developed approaches can be readily incorporated into microphysical
545 and transport models of any scale to account for charging phenomena of atmospheric particles.

546

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670

Table 1. Properties of ions used for experimental observations

References	Positive ions		Negative ions	
	Mass (AMU)	Mobility ($\text{cm}^2 \text{V}^{-1} \text{s}^{-1}$)	Mass (AMU)	Mobility ($\text{cm}^2 \text{V}^{-1} \text{s}^{-1}$)
Alonso et al. (1997)	150	1.15	80	1.65
Liu and Pui (1974) ^a	140	1.4	101	1.6
Wiedensohler and Fissan (1991)	140	1.4	101	1.6
Gensdarmes et al. (2001)	-	1.19	-	1.54

^a The properties of ions were obtained from Wiedensohler and Fissan (1991), who used a radioactive neutralizer similar to that employed by Liu and Pui (1974).

671

672

Table 2. Timescales required for particles to reach steady-state charge

Charging mechanism	Particle diameter (μm)	Steady-state mean ion concentration (m^{-3})	τ (s) (eq 8)	Numerical timescale (s) (Approach 1 or 2)	Timescale from measurements (s)	Reference
Diffusion charging	0.0071	8.0×10^{12}	0.04	0.2 ~ 0.4	0.3	Alonso et al. (1997)
	0.0070	2.0×10^{13}	0.02	0.2 ~ 0.4	0.3	
	0.0027	8.0×10^{12}	0.04	0.2 ~ 0.4	0.3	
	0.0026	2.0×10^{13}	0.02	0.2 ~ 0.4	0.3	
Diffusion charging and self-charging	0.82	2.1×10^9	171.1	> 1,000	-	Gensdarmes et al.(2001)
	0.82	1.5×10^{10}	23.6	> 200	-	
	1.05	7.6×10^8	471.6	> 3,000	-	
	1.05	7.0×10^9	51	> 500	-	

673

Table 3. Computational costs of the approaches used

Example: 30 size bins		Approach 1	Approach 2	Approach 3
	Ion balance model	2	2	-
The number of ODEs	Charge balance model	$30 \times 31 = 930$	30	-
	Population balance model		30	30
	Total	932	62	30
Computational time for urban aerosol ^{a,b} (s)		12724.2	302.6	8.5

^a Simulation conditions: $d_g = 0.116 \mu\text{m}$, $\sigma_g = 1.46$, $N_t = 6.718 \times 10^9 \text{ m}^{-3}$ and $q = 10^7 \text{ m}^{-3} \text{ s}^{-1}$ (Kim et al., 2015). The simulation time is 6 hours.

^b Computational resources: Intel(R) Core(TM)2 Duo CPU E6850 @ 3.00 GHz with 4GB RAM and Matlab ODE solver.

676 **Figure captions**

677 Figure 1. Three approaches to predict time-dependent changes in the particle size and charge
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686 Approach 2 were compared with the measurements of Gensdarmes et al.(2001).

687 Figure 5. Timescale to reach steady-state charge accumulation rates of $0.0071 \text{ }\mu\text{m}$ nanoparticles.
688 The lines are the simulation results of Approach 1. The symbols are the measurements of Alonso
689 et al.(1997). Charging timescales were estimated using eq 8 ($\tau_{\text{charging}} = 0.042 \text{ s}$ and $\tau_{\text{discharging}} =$
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694 radioactive particles.

695 Figure 7. Validation of the numerical solution for the bivariate population balance model under a
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699 Figure 8. Time-evolution of the particle size distributions predicted by the monovariate
700 population balance model with the average collision efficiency (eq 15) under a monodispersed
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704 Figure 9. Evolution of the size distribution of negatively charged particles in a ^{241}Am radioactive
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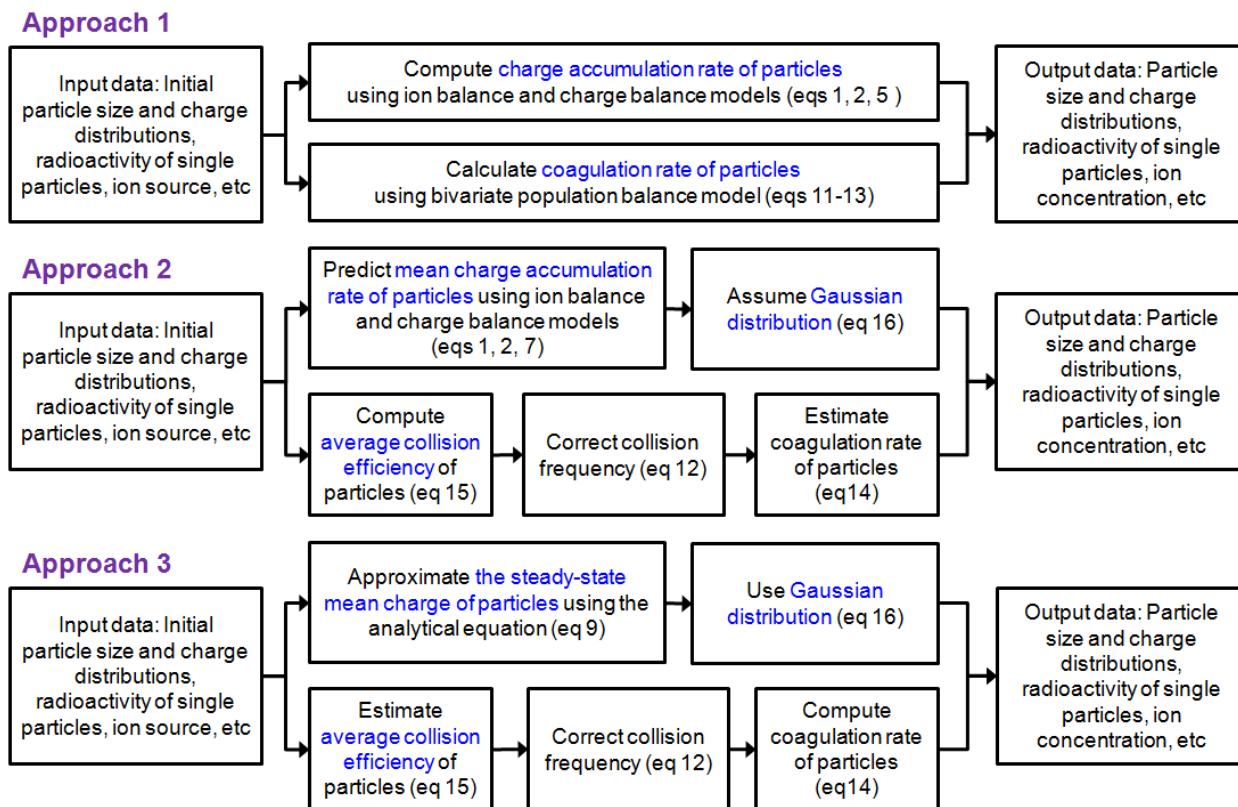
708 Figure 10. Time-evolution of the charge (a) and size (b) distributions of atmospheric particles in
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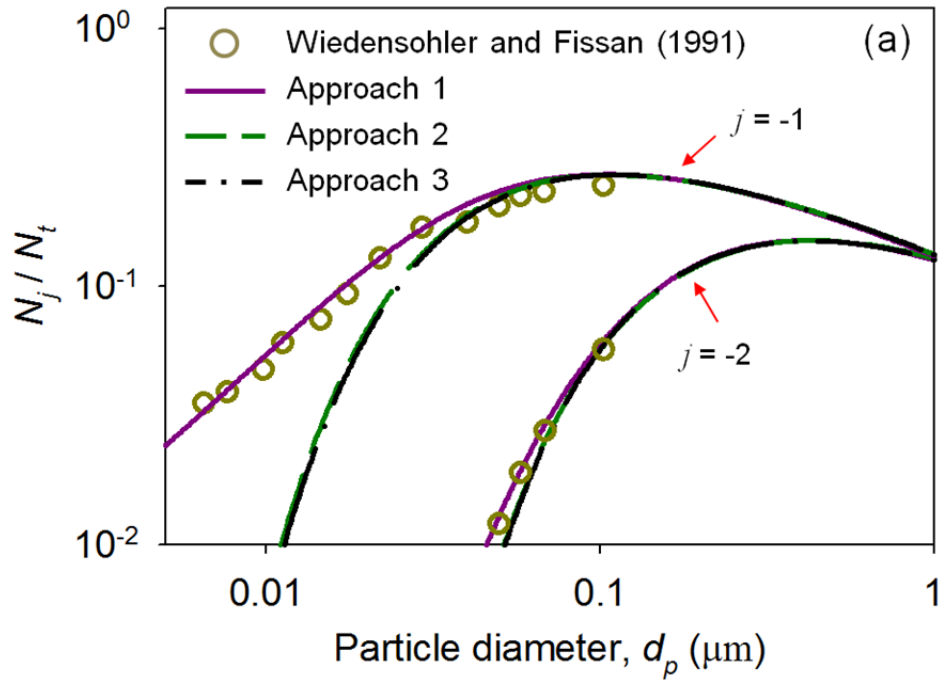
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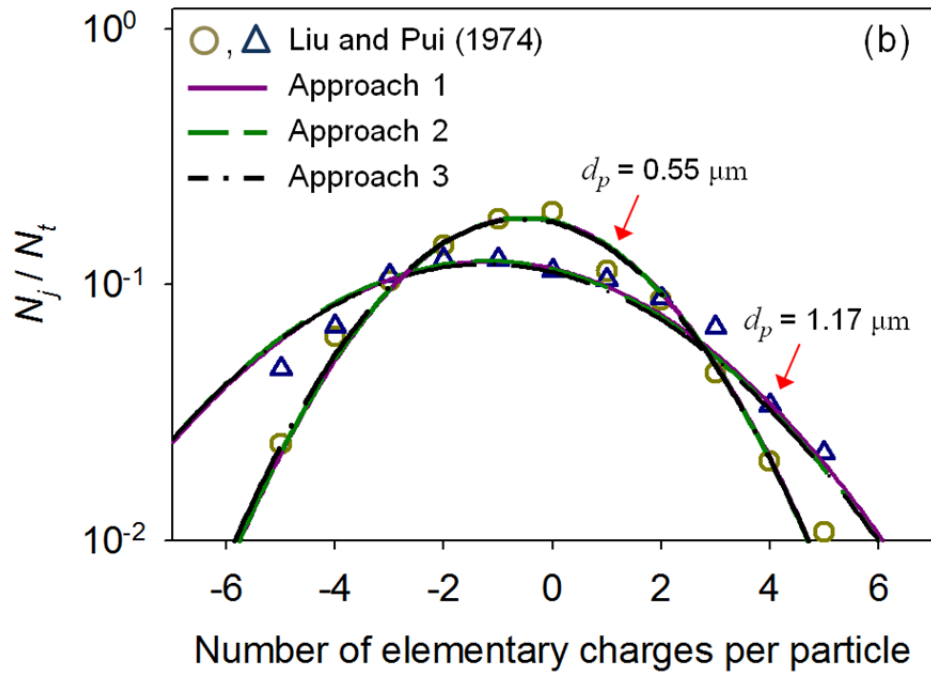
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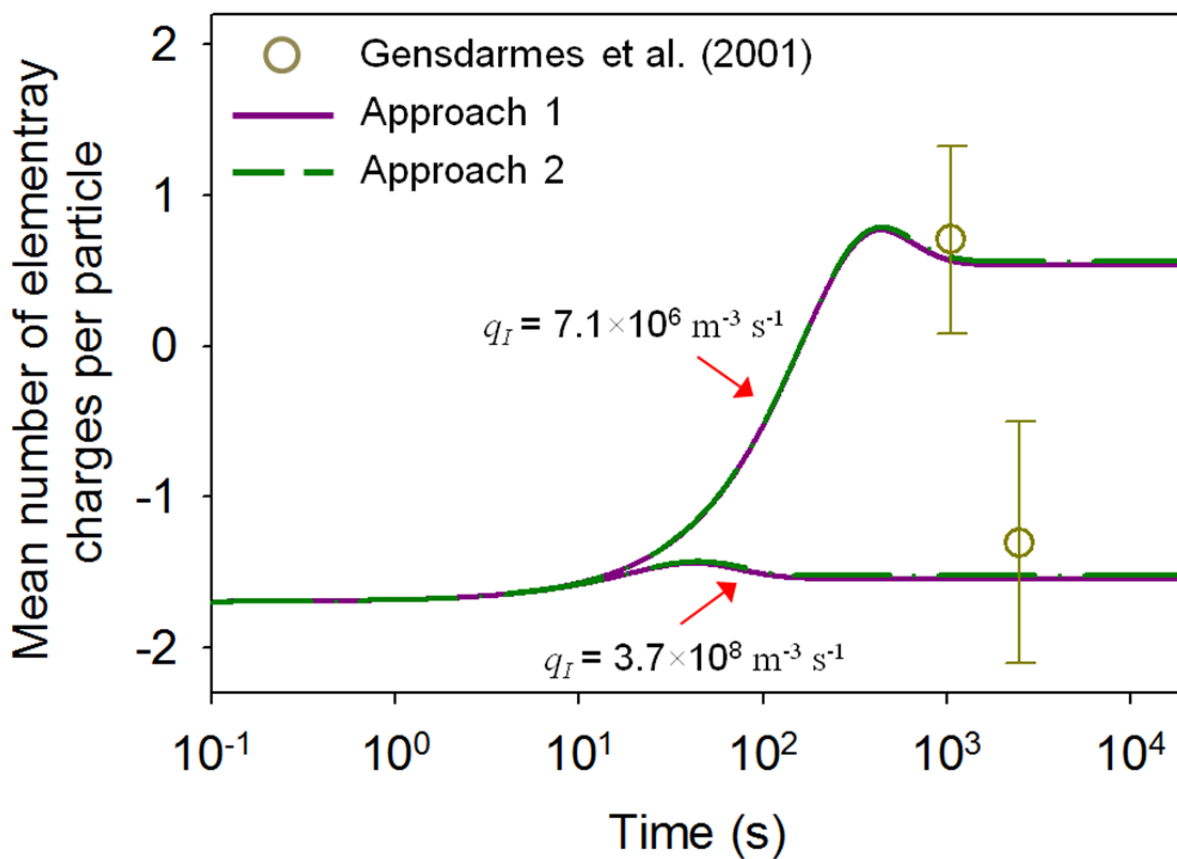


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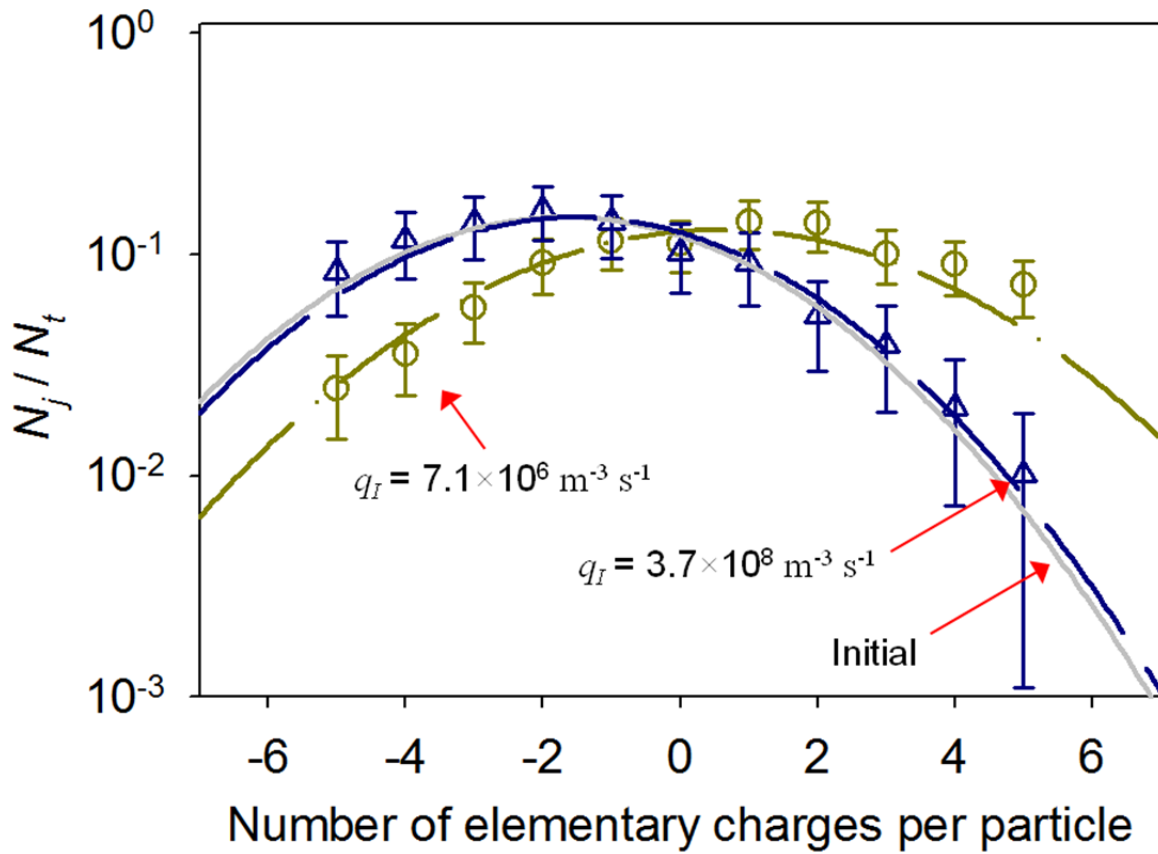
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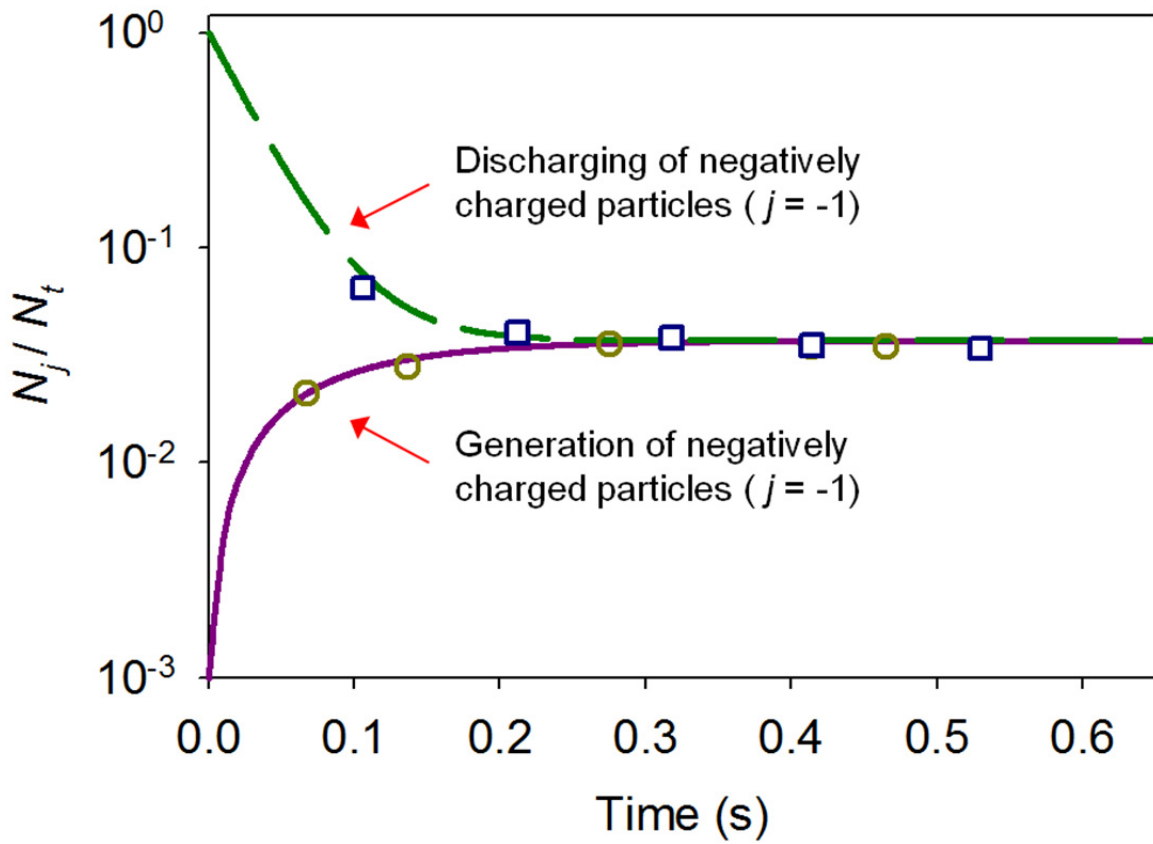
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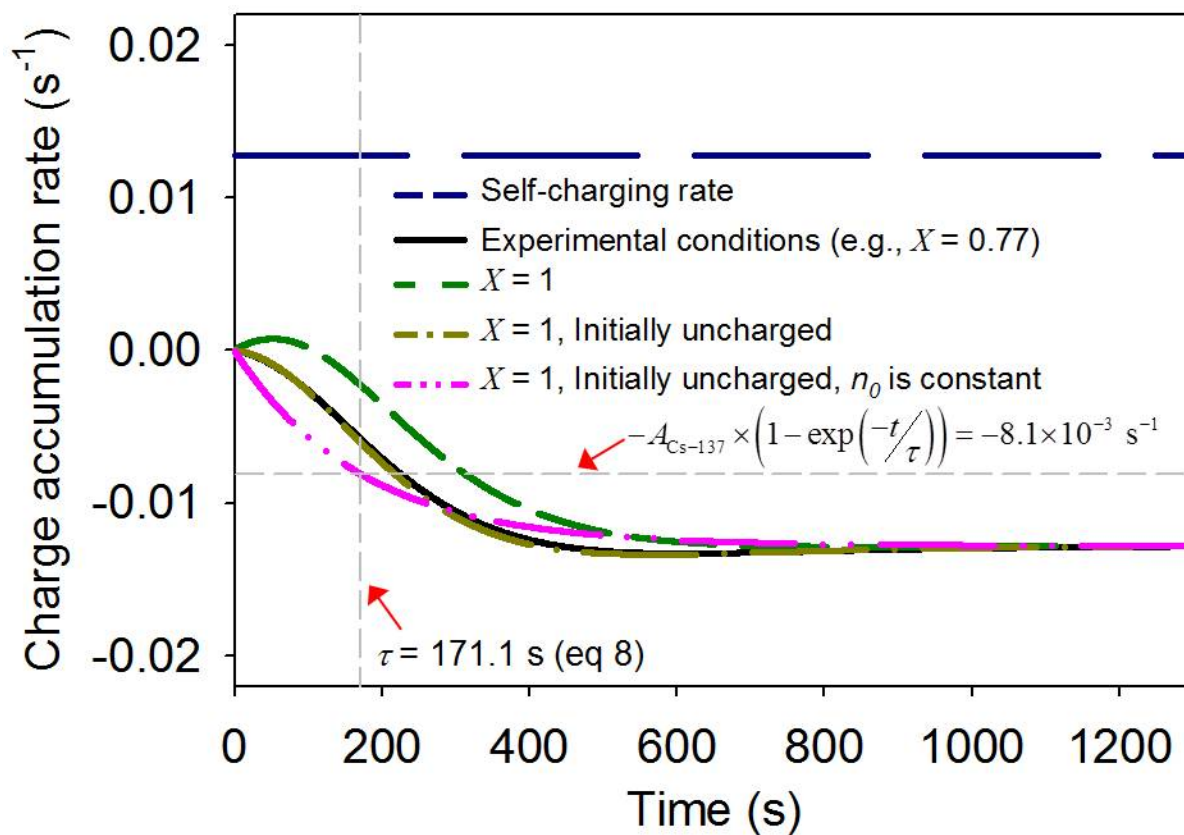
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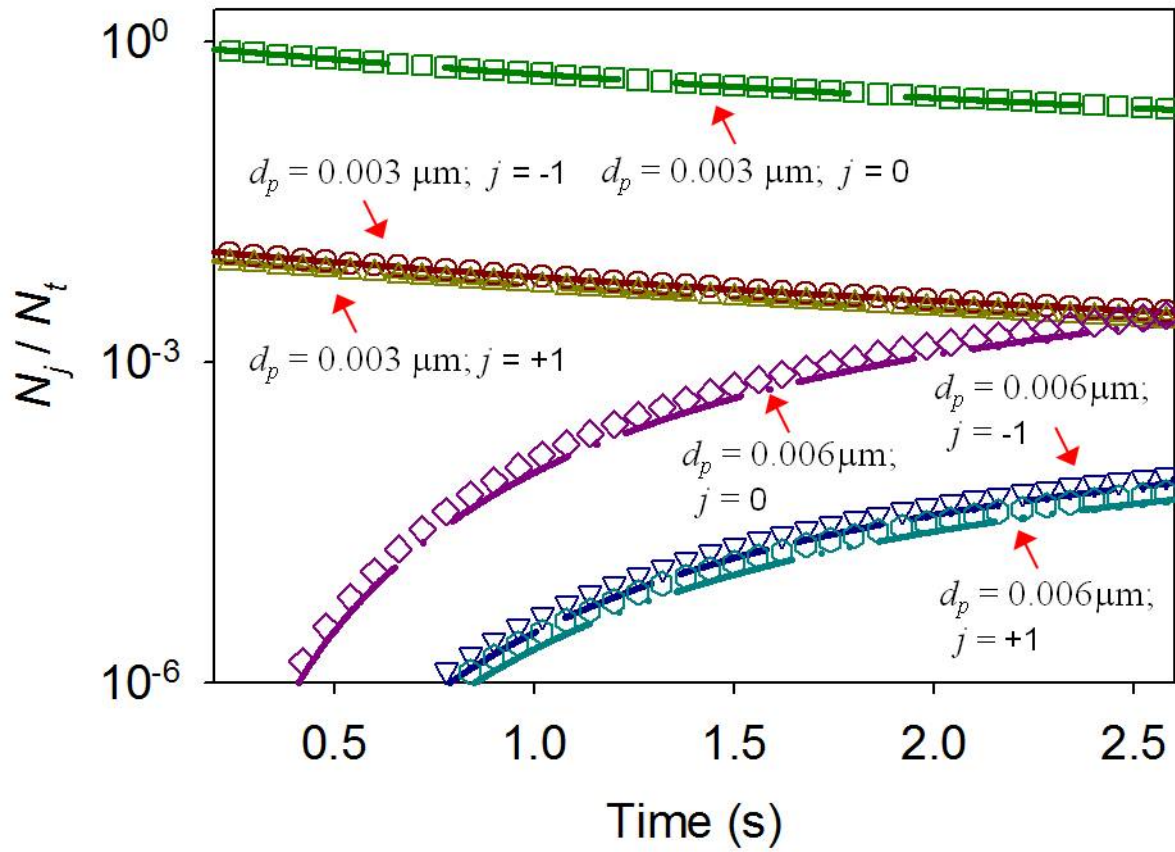
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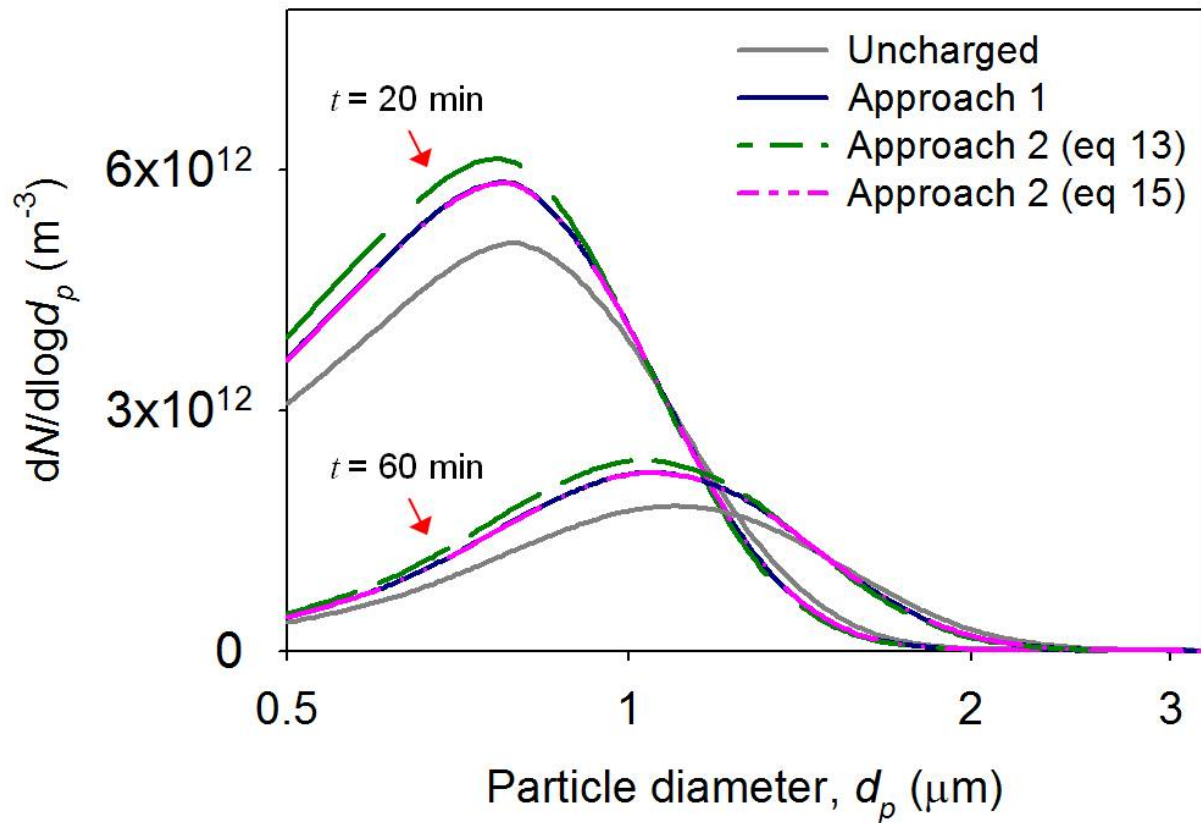
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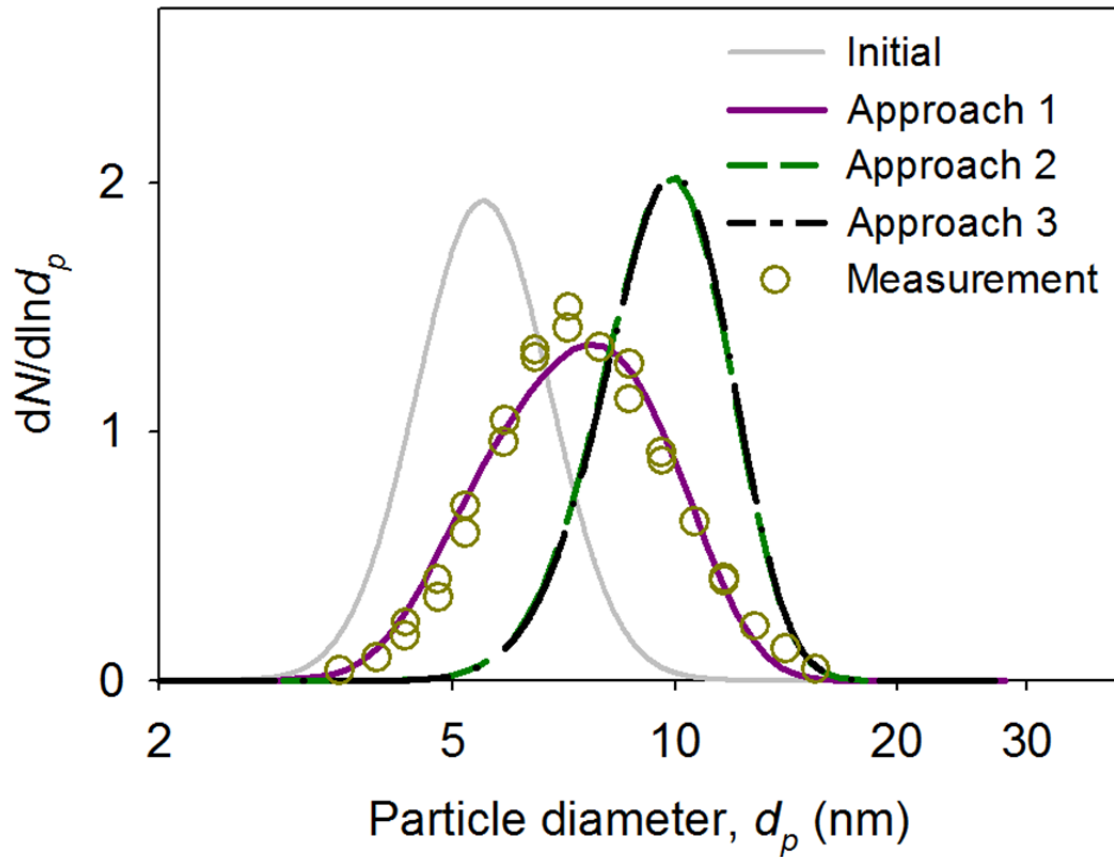
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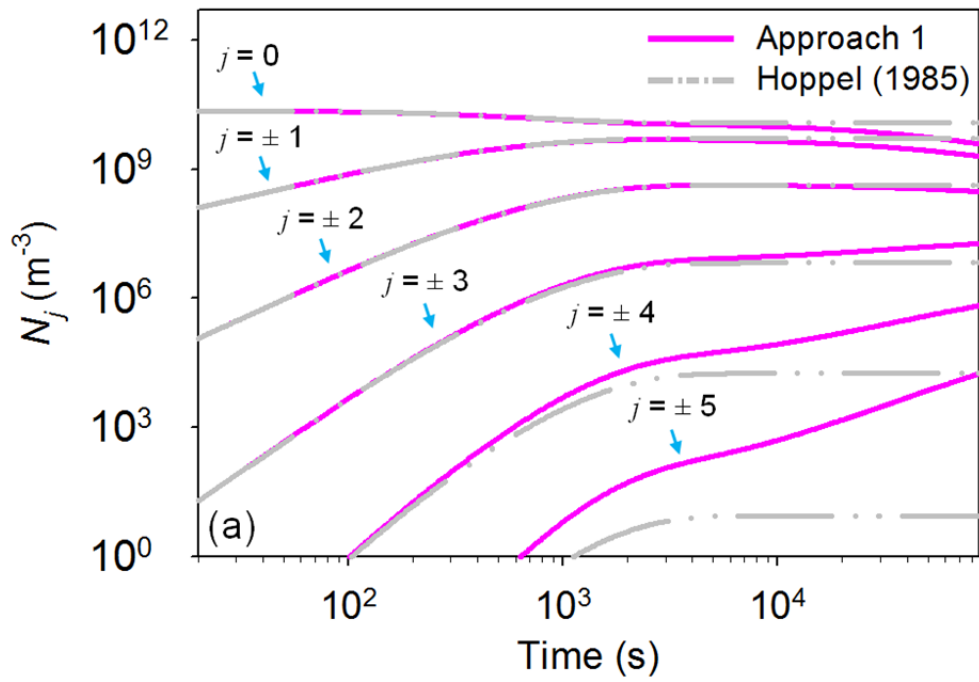
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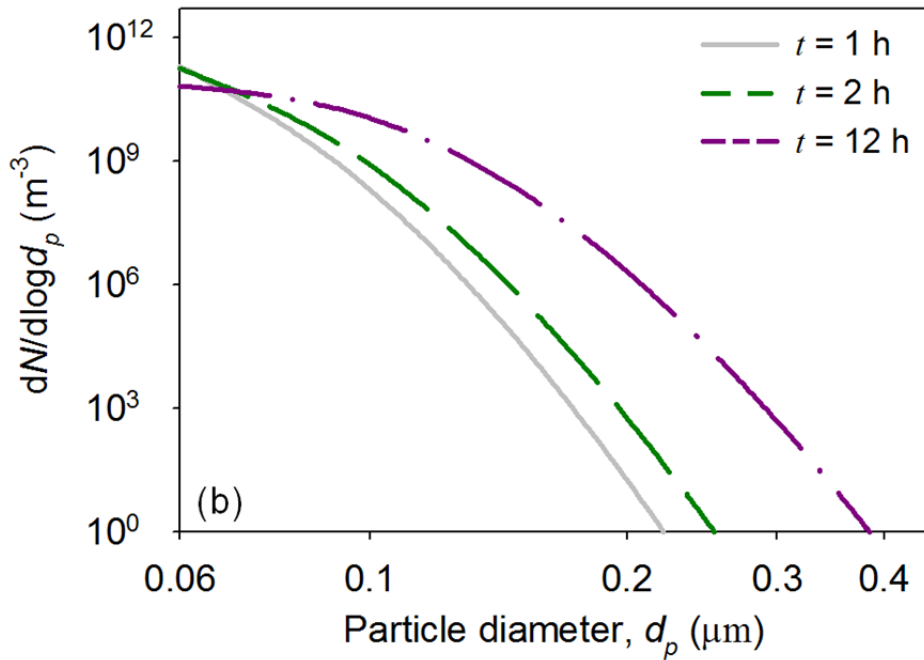
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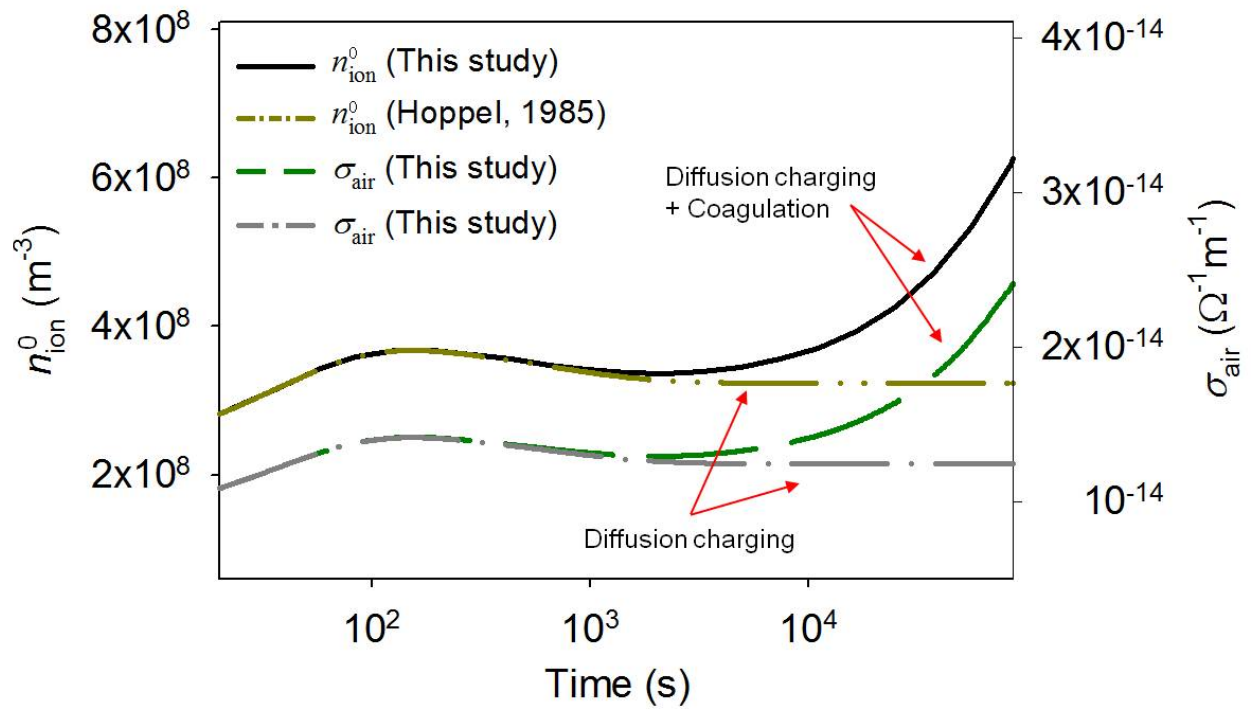


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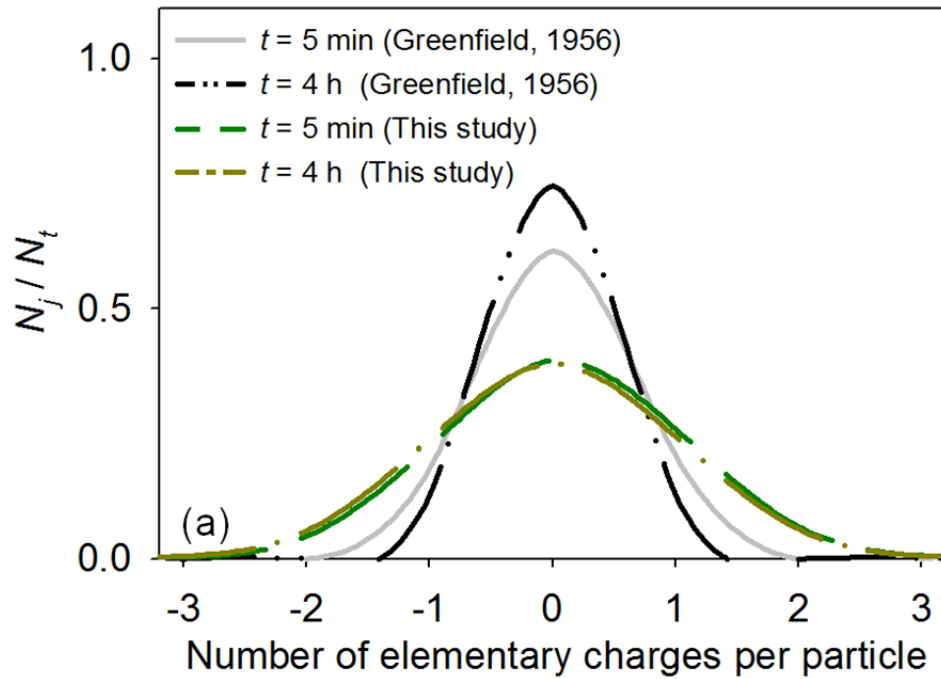
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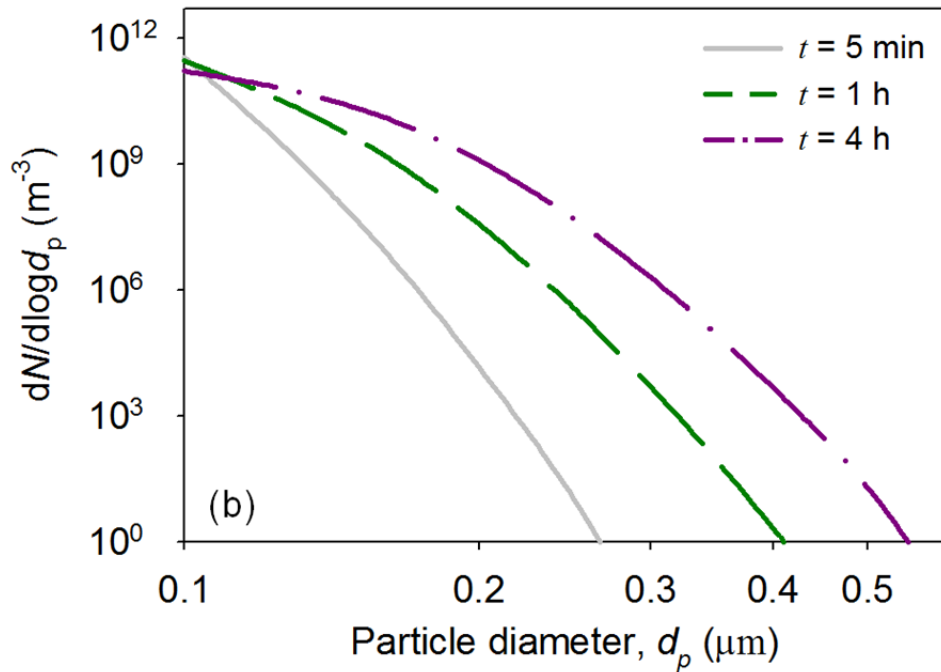
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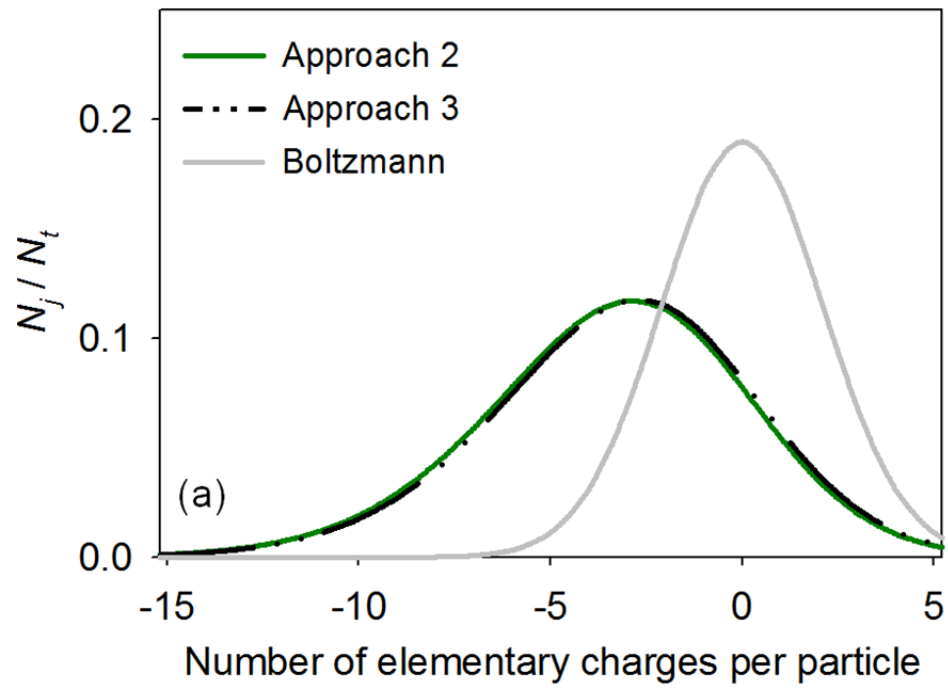
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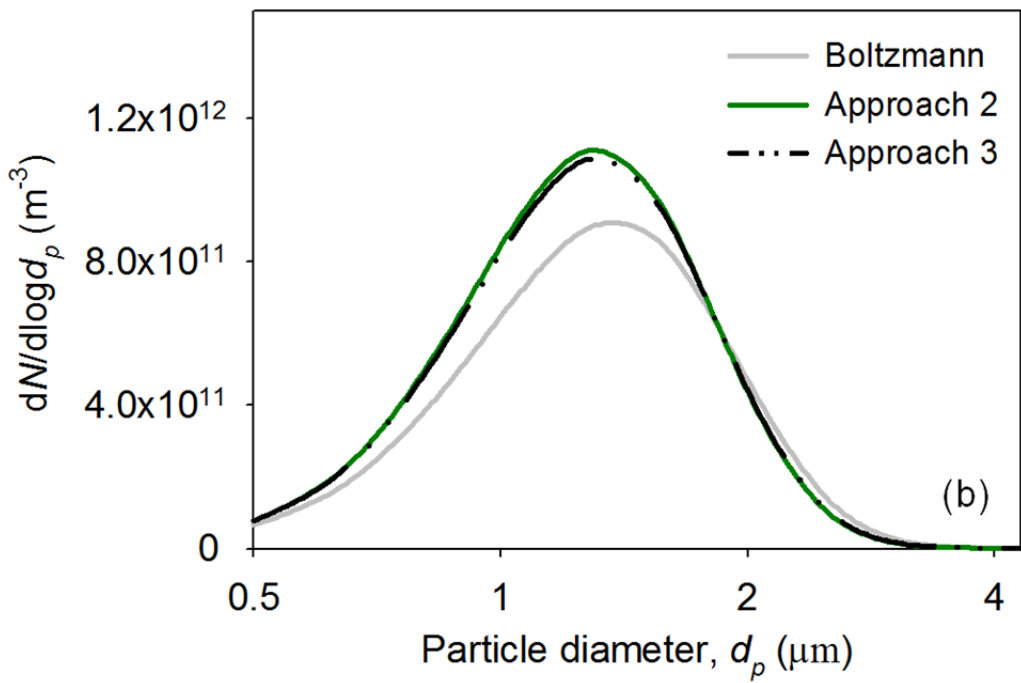
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