

1 **Title: Size-resolved measurements of ice nucleating particles at six locations in**
2 **North America and one in Europe**

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22

23 **Abstract**

24 Detailed information on the size of ice nucleating particles (INPs) may be useful in
25 source identification, modeling their transport in the atmosphere to improve climate predictions,
26 and determining how effectively or ineffectively instrumentation used for quantifying INPs in
27 the atmosphere captures the full INP population. In this study we report immersion-mode INP
28 number concentrations as a function of size at six ground sites in North America and one in
29 Europe using the micro-orifice uniform deposit impactor-droplet freezing technique (MOUDI-
30 DFT), which combines particle size-segregation by inertial impaction and a microscope-based
31 immersion freezing apparatus. The lowest INP number concentrations were observed at Arctic
32 and alpine locations and the highest at suburban and agricultural locations, consistent with
33 previous studies of INP concentrations in similar environments. We found that 91 ± 9 , 79 ± 17 ,
34 and 63 ± 21 % of INPs had an aerodynamic diameter $> 1 \mu\text{m}$ at ice activation temperatures of -
35 15, -20, and -25 °C, respectively, when averaging over all sampling locations. In addition, $62 \pm$
36 20 , 55 ± 18 , and 42 ± 17 % of INPs were in the coarse mode ($> 2.5 \mu\text{m}$) at ice activation
37 temperatures of -15, -20, and -25 °C, respectively, when averaging over all sampling locations.
38 These results are consistent with six out of the nine studies in the literature that have focused on
39 the size distribution of INPs in the atmosphere. Taken together, these findings strongly suggest
40 that supermicron and coarse mode aerosol particles are a significant component of the INP
41 population in many different ground-level environments. Further size-resolved studies of INPs as
42 a function of altitude are required since the size distribution of INPs may be different at high
43 altitudes due to size-dependent removal processes of atmospheric particles.

44 **1 Introduction**

45 Ice nucleating particles (INPs) are a unique class of aerosol particles that catalyze ice

46 formation under atmospheric conditions. A variety of particle types have been identified as INPs,
47 including mineral dust, black carbon, volcanic ash, glassy aerosols, and primary biological
48 particles such as bacteria, fungal spores, and pollen (see reviews by Szyrmer and Zawadzki,
49 1997; Möhler et al., 2007; Ariya et al., 2009; Després et al., 2012; Hoose and Möhler, 2012;
50 Murray et al., 2012; Yakobi-Hancock et al., 2013). Although only a small fraction of aerosol
51 particles nucleate ice (e.g. Rogers et al., 1998), INPs are important since they can lead to changes
52 in the properties and lifetimes of mixed-phase and ice clouds, ultimately affecting climate and
53 precipitation (Baker, 1997; Lohmann and Feichter, 2005; Baker and Peter, 2008; DeMott et al.,
54 2010; Creamean et al., 2013).

55 Vali et al. (2015) describes four modes of heterogeneous ice nucleation: deposition
56 nucleation, where ice forms on the INP directly from the gas phase; condensation freezing,
57 where ice nucleates during the condensing of water onto the INP; immersion freezing, where
58 crystallization is initiated by an INP within a supercooled liquid droplet; and contact freezing,
59 where the freezing of a supercooled liquid droplet is due to impaction by an INP. In this study
60 we focus on freezing via the immersion mode in dilute solution droplets, which is relevant to
61 mixed-phase cloud conditions.

62 Due to the importance of INPs for climate and precipitation, there has been a renewed
63 interest in measuring the concentrations of INPs in the atmosphere (DeMott et al., 2011). While
64 much of this work has focused on measurements of the total number concentration of INPs, there
65 has been less emphasis on determining their size distributions in the atmosphere. Information on
66 airborne INP size distributions may be particularly helpful in identifying the predominant INP
67 sources. For example, information on the size distribution of INPs may help rule out or support
68 the role of fungal spores in atmospheric ice nucleation since they are often in the supermicron

69 range (Graham et al., 2003; Elbert et al., 2007; Sesartic and Dallafior, 2011; Després et al., 2012;
70 Huffman et al., 2012). A similar approach can be used with black carbon particles, since they are
71 mainly in the submicron range (Clarke et al., 2004; Schwarz et al., 2008, 2013).

72 Previous modeling studies have shown that the transport and distribution of INPs, and
73 aerosol particles in general, are sensitive to the size of the particles assumed in the models
74 (Burrows et al., 2009; Wilkinson et al., 2011). Information on the size distributions of INPs are
75 thus needed for accurate modeling of their transport and distributions in the atmosphere (Morris
76 et al., 2004; Hoose et al., 2010a, 2010b; Sesartic et al., 2013; Haga et al., 2014; Spracklen and
77 Heald, 2014).

78 Information on the size distribution of INPs is also needed to determine if techniques
79 used to measure atmospheric INP concentrations capture the entire INP population. For example,
80 the continuous flow diffusion chamber (Rogers et al., 2001b) is often used for measuring INPs
81 (e.g. DeMott et al., 1998; Rogers et al., 2001a; Richardson et al., 2007; Pratt et al., 2009; Prenni
82 et al., 2009; Eidhammer et al., 2010; Chou et al., 2011; Friedman et al., 2011; Hoyle et al., 2011;
83 Corbin et al., 2012; Garcia et al., 2012; Tobo et al., 2013; McCluskey et al., 2014). This type of
84 instrument has the advantage of providing real-time measurements of INPs with the ability to
85 detect very large INP number concentrations, but the aerodynamic diameter of particles
86 measured with it is limited, from $d_{50} \leq 2.4 \mu\text{m}$ in some studies (e.g. Garcia et al., 2012) to $d_{50} \leq$
87 $0.75 \mu\text{m}$ in others (e.g. DeMott et al., 2003). Such techniques may miss supermicron or coarse
88 mode (i.e. larger than $2.5 \mu\text{m}$) INPs. The exact proportion of INPs missed may depend on
89 temperature. Such online instruments have typically focused on measurements below
90 approximately $-20 \text{ }^\circ\text{C}$ as sample volume considerations limit effective sampling of lower INP
91 number concentrations at warmer temperatures. The exact proportion of INPs missed may also

92 depend on altitude since the removal of atmospheric particles by wet and dry deposition in the
93 atmosphere is expected to be size dependent. As an example, supermicron particles have larger
94 dry deposition loss rates than submicron particles.

95 Previous studies of INPs as a function of size have been carried out in the field (e.g. Vali,
96 1966; Rosinski et al., 1986; Mertes et al., 2007; Santachiara et al., 2010) and in the laboratory
97 (e.g. Welti et al., 2009; O’Sullivan et al., 2015). These and additional studies are further
98 discussed in Sect. 3.2. In the current study, we add to the existing body of size-resolved INP
99 measurements by reporting ground-level INP size distributions from six locations in North
100 America and one in Europe, covering a diverse set of environments and investigating immersion
101 freezing at -15, -20, and -25 °C.

102 **2 Methods**

103 **2.1 Sampling sites**

104 The seven locations used in this study are detailed in Table 1 and shown in Fig. 1. All
105 reported sampling periods are local times. Measurements using the sampling instrumentation
106 described in the next section were made at five locations in Canada: Alert, Nunavut; the
107 Labrador Sea near Newfoundland and Labrador; Whistler Mountain, British Columbia; the
108 University of British Columbia (UBC) campus, British Columbia and Amphitrite Point, British
109 Columbia. Measurements in Canada were conducted as part of the larger NETwork on Climate
110 and Aerosols: addressing key uncertainties in Remote Canadian Environments project
111 (NETCARE; <http://netcare-project.ca/>). Measurements were also made at Saclay, France and
112 Colby, Kansas, USA.

113 **2.1.1 Alert**

114 Arctic sampling was conducted at the Dr. Neil Trivett Global Atmosphere Watch
115 Observatory in Alert, Nunavut, Canada (labeled 1 in Fig. 1; Cobbett et al., 2007) between March
116 29 and July 23, 2014. This Arctic research station is part of a global network for measuring
117 chemical and physical perturbations of the atmosphere. Aerosol particles were collected through
118 a louvered total suspended particulate (TSP) inlet (Mesa Labs Inc., Butler, NJ, USA) and 0.9 m
119 mast located on the upper level of an outdoor platform free of surrounding obstructions, and
120 were stored in the dark at -15 or 4 °C for a period of 10–112 days prior to analysis.

121 **2.1.2 Whistler Mountain**

122 The Whistler Peak High Elevation Site is located at the summit of Whistler Mountain in
123 Whistler, British Columbia, Canada (labeled 3 in Fig. 1) and operated by Environment Canada
124 (Gallagher et al., 2011; Macdonald et al., 2011). Aerosol particle collection at this alpine site
125 occurred between March 30 and April 23, 2014. The louvered TSP inlet was located
126 approximately 10 m from a chairlift operating station. Although there are no continuous
127 combustion sources at the site, sampled air may have been influenced by engine exhaust for short
128 periods of time due to nearby snowmobile operation. Samples were stored in the dark at 4 °C for
129 a period of 1–4 days prior to analysis.

130 **2.1.3 Amphitrite Point**

131 The coastal site at Amphitrite Point on Vancouver Island, British Columbia, Canada
132 (labeled 5 in Fig. 1) is operated by Environment Canada, the BC Ministry of Environment, and
133 Metro Vancouver for the continuous monitoring of aerosols and trace gases influenced by marine
134 trajectories (McKendry et al., 2014; Yakobi-Hancock et al., 2014; Mason et al., 2015b). The
135 mobile laboratory used during sampling was located approximately 100 m from the high tide line
136 of the Pacific Ocean along a rocky shoreline, separated from the ocean by a narrow row of trees

137 and shrubs approximately 2–10 m in height. Sampling took place from August 6 to August 27,
138 2013 using a louvered TSP inlet and 3 m mast. Aerosol particles were stored at room
139 temperature and analyzed within 1 day of collection.

140 **2.1.4 The Labrador Sea**

141 The Canadian Coast Guard Service vessel CCGS Amundsen serves as both an icebreaker
142 for shipping lanes and an Arctic research vessel. One set of aerosol particle samples was
143 collected from the top of the bridge of this vessel on July 11, 2014 while in the Labrador Sea off
144 the coast of Newfoundland and Labrador, Canada (labeled 2 in Fig. 1). While sampling was
145 within the marine boundary layer in the presence of sea spray aerosols, back trajectories (not
146 included) calculated using the Hybrid Single-Particle Lagrangian Integrated Trajectory
147 (HYSPPLIT4) model of the National Oceanographic and Atmospheric Administration (Draxler
148 and Rolph, 2014) indicate that the sampled air mass spent the majority of the previous 72-hour
149 period over land. Air was passed through a louvered TSP inlet and 1.5 m mast during sampling,
150 and collected aerosol particles were stored in the dark at 4 °C for a period of 45–46 days prior to
151 analysis.

152 **2.1.5 Saclay, France**

153 Aerosol particle samples were collected at the Commissariat à l’Energie Atomique
154 (CEA) Atmospheric Supersite (AS), CEA l’Orme des Merisiers. The CEA-AS Observatory is a
155 suburban area located 30 km southeast of Paris in Saclay, France (labeled 7 in Fig. 1). The CEA-
156 AS Observatory is surrounded by different sources of bioaerosols such as forest and agricultural
157 fields, and is often influenced by marine or urban air masses (Baisnée et al., 2014).
158 Measurements were made as part of the BIODTECT 2014 intensive campaign, an
159 intercomparison of bioaerosol detection methods (Sarda-Estève et al., 2014). During this study

160 period, the site was heavily influenced by urban outflow. A large set of ancillary measurements
161 was done to constrain all the particulate matter sources. Aerosol particles were sampled through
162 a TSP inlet and 10 m mast between July 15 and August 4, 2014, and were stored in the dark at 4
163 °C for a period of 55–217 days prior to analysis.

164 **2.1.6 UBC Campus**

165 Four sets of aerosol particle samples were collected from a weather station on the roof of
166 the five-story Earth Sciences Building on the UBC campus in British Columbia, Canada (labeled
167 4 in Fig. 1). The UBC campus is located on a peninsula and is surrounded by forest on three
168 sides and ocean on the fourth. The site has been classified as suburban since it is less than 10 km
169 from downtown Vancouver. Samples were collected through a TSP inlet and 0.5 m mast between
170 May 12 and May 16, 2014. The aerosol particles were stored in the dark at 4 °C for a period of
171 21–23 days prior to analysis.

172 **2.1.7 Colby, KS**

173 Aerosol particles were collected at the soybean and sorghum fields of the Kansas State
174 University Northwest Research Center in Colby, KS, USA (labeled 6 in Fig. 1). One sample was
175 collected at each location during combine harvesting from a distance approximately 3–10 m
176 downwind of the field. A third sample was also collected at the sorghum field the night
177 following harvest. Sampling took place on October 14 and 15, 2014 and samples were stored in
178 the dark at 4 °C for a period of 41–46 days prior to analysis.

179 **2.2 Size-resolved INP number concentrations**

180 INP number concentrations as a function of size and temperature were determined using
181 the micro-orifice uniform deposit impactor-droplet freezing technique (MOUDI-DFT; Huffman

182 et al., 2013; Mason et al., 2015a). This technique combines aerosol particle collection by a
183 cascade inertial impactor with sharp size-cutoff characteristics (the MOUDI; Marple et al., 1991)
184 with an established droplet freezing apparatus (the DFT) for determining immersion-mode
185 freezing properties (Koop et al., 1998; Iannone et al., 2011; Haga et al., 2013). A similar
186 approach has also been used to study deposition nucleation by particles collected from the
187 atmosphere (Wang et al., 2012; Knopf et al., 2014).

188 **2.2.1 Aerosol particle sampling**

189 Size-fractionated aerosol particle samples were collected onto hydrophobic glass cover
190 slips (HR3-215; Hampton Research, Aliso Viejo, CA, USA) using a model 110R or 120R
191 MOUDI (MSP Corp., Shoreview, MN, USA). Previous work has shown that these hydrophobic
192 glass surfaces do not cause significant heterogeneous ice nucleation (e.g. Haga et al., 2013, 2014;
193 Wheeler et al., 2015). Substrate holders were used on the impaction plates of the MOUDI to
194 reproducibly position the hydrophobic glass cover slips in regions where aerosol deposit particle
195 concentrations did not vary significantly (for details see Mason et al., 2015a). At most locations
196 MOUDI stages 2–9 were used, corresponding to particle size bins of 10–5.6, 5.6–3.2, 3.2–1.8,
197 1.8–1.0, 1.0–0.56, 0.56–0.32, 0.32–0.18, and 0.18–0.10 μm (50 % cutoff aerodynamic diameter;
198 Marple et al., 1991), respectively. The range in particle size collected at each location is given in
199 Table 1.

200 Bounce within inertial impactors such as the MOUDI can occur during aerosol sampling,
201 where particles impact the collection substrate but are not retained. This rebounding of particles
202 from the surface could possibly alter the INP number concentrations and size distributions being
203 measured. If composition is held constant, bounce is expected to increase with particle size
204 because of their greater kinetic energy (Dahneke, 1971). Hence, INP number concentrations for

205 large particle sizes may be underestimated here. Bounce is also expected to increase with
206 decreasing relative humidity (RH). Previous work has shown that having a sample RH of 70 %
207 or greater can be effective in reducing particle bounce (e.g. Winkler, 1974; Fang et al., 1991;
208 Stein et al., 1994; Vasiliou et al., 1999; Chen et al., 2011; Bateman et al., 2014), although its
209 efficacy is dependent on particle type (Winkler, 1974; Lawson, 1980; Saukko et al., 2012). For
210 six out of the seven sites investigated here, the average RH during sampling was 69% or greater.
211 Recently, results from the MOUDI-DFT and the continuous flow diffusion chamber were
212 compared during an ambient field campaign at Colorado State University (Mason et al., 2015a).
213 For particle sizes $< 2.4 \mu\text{m}$, the INP number concentrations measured by the MOUDI-DFT were
214 within experimental error of those measured by the continuous flow diffusion chamber
215 technique, suggesting that bounce was not an issue during these previous ambient field
216 measurements. Based on these previous measurements, in the current studies we do not consider
217 the issue of particle bounce when calculating INP number concentrations and size distributions.
218 Nevertheless, additional studies are warranted to better quantify the effect of bounce.

219 During sampling, the aerosols were not brought to a standardized humidity, and the RH
220 varied from site-to-site (see Table 1 for the average RH during sampling at each site). This
221 variability in RH could lead to a small variability in INP size due to differences in hygroscopic
222 growth. We further note that the duration of sample storage varied in the current study.
223 Additional studies are needed to quantify the effect of storage on INP activity.

224 **2.2.2 Freezing measurements**

225 Samples were analyzed by the DFT to determine the number concentration of particles
226 active in the immersion-freezing mode. Details of the experimental procedure can be found in
227 Mason et al. (2015a). Briefly, samples were transferred to a temperature- and humidity-

228 controlled flow cell coupled to an optical microscope equipped with a 5× magnification objective
229 (Axiolab; Zeiss, Oberkochen, Germany). Water droplets were condensed onto the sample and
230 monitored using a CCD camera recording a digital video. Since the relative humidity of the gas
231 flow during droplet condensation was held at approximately 120 %, water condensation occurred
232 uniformly on the cover slip, and growing droplets coagulated as they grew to a final size of $97 \pm$
233 $42 \mu\text{m}$ (mean diameter and 1 standard deviation (SD) uncertainty). The freezing temperature of
234 each droplet was then determined during cooling at a rate of $-10 \text{ }^\circ\text{C min}^{-1}$ using the video
235 timestamp and a resistance temperature detector located within the flow cell. Note that droplet
236 growth and coagulation occurs in the same manner for samples containing particles and clean
237 hydrophobic glass cover slips (no particles deposited). In addition, based on an analysis of
238 samples collected at Amphitrite Point, more than 99 % of particles become incorporated into the
239 droplets prior to the freezing experiments. Here we regard ice nucleation as a singular process
240 (i.e. strictly temperature-dependent) but note that the stochastic (i.e. time-dependent) component
241 to immersion freezing (Vali, 2014) may alter the median freezing temperature of a droplet by
242 $0.5\text{--}2 \text{ }^\circ\text{C}$ per decade change in cooling rate (Murray et al., 2011; Welti et al., 2012; Wright and
243 Petters, 2013; Wright et al., 2013; Wheeler et al., 2015).

244 A potential issue with the droplet freezing technique is heterogeneous ice nucleation
245 initiated by the hydrophobic glass cover slips used to collect atmospheric particles. To address
246 this issue, experiments were conducted using new hydrophobic glass cover slips that were
247 processed in the same manner as ambient samples except they were not exposed to atmospheric
248 particles drawn into the MOUDI. For the five hydrophobic glass cover slips investigated, which
249 contained 231 droplets generated during the freezing experiments, the average freezing
250 temperature was $-36.5 \pm 0.5 \text{ }^\circ\text{C}$ (1 SD). In addition, none of the droplets froze above $-33.7 \text{ }^\circ\text{C}$ in

251 these blank experiments. Since we only report INP number concentrations for temperatures from
252 -15 °C to -25 °C in this study, heterogeneous ice nucleation by the substrate is unlikely to
253 contribute to the reported INP number concentrations.

254 2.2.3 Calculating the number concentration of INPs

255 The number of INPs in the DFT, $\#INPs(T)$, was calculated using the following equation
256 which accounts for the possibility of a droplet containing multiple INPs (Vali, 1971):

$$257 \quad \#INPs(T) = -\ln\left(\frac{N_u(T)}{N_o}\right) N_o f_{0.25-0.10\text{mm}} f_{ne} \quad (1)$$

258 where $N_u(T)$ is the number of unfrozen droplets at temperature T , N_o is the total number of
259 droplets, $f_{nu,0.25-0.10\text{mm}}$ is a non-uniformity correction factor that takes into account non-uniformity
260 at the 0.25-1 mm scale, and f_{ne} is a statistical uncertainty derived for a given number of detected
261 nucleation events, with fewer nucleation events leading to greater statistical uncertainty (Koop et
262 al., 1997). For the results reported, here f_{ne} is derived for a confidence level of 0.95.

263 Equation (1) assumes that droplets in a given freezing experiment have the same volume
264 (Vali, 1971). Using droplet freezing experiments reported in Mason et al. (2015a), which are
265 similar to experiments presented here, we explored if this assumption leads to uncertainties when
266 applied to the DFT experiments. Number of INPs was first calculated from the DFT experiments
267 as described above. Second, number of INPs were calculated by first separating the droplet
268 freezing results into 2-4 bins based on droplet volume. After binning the data by droplet volume,
269 droplets in each bin are more similar in volume. Equation 1 was then used to calculate the
270 number of INPs in each bin. Finally, the total number of INPs was determined by summing the
271 numbers of INPs calculated for each bin. We found that the total number of INPs determined
272 both ways (with and without binning) agreed within the experimental uncertainty at freezing

273 temperatures of $-25\text{ }^{\circ}\text{C}$ and above for 97 % of freezing events. We also analyzed the same data
274 by first separating the droplet freezing results into 2-4 bins based on the maximum area the
275 droplets covered. Again, the number of INPs determined with and without binning was in good
276 agreement, being within the experimental uncertainty at freezing temperatures of $-25\text{ }^{\circ}\text{C}$ and
277 above for 98 % of freezing events. Based on this analysis we conclude that the application of
278 Equation (1), which assumes monodisperse droplets, to the DFT results at $-25\text{ }^{\circ}\text{C}$ and above does
279 not lead to large uncertainties.

280 In the DFT, once a droplet freezes it may grow by vapor diffusion and contact a
281 neighboring liquid droplet, causing the latter to freeze as well. To account for these non-
282 immersion freezing events, we calculated the $\#INPs(T)$ in the immersion mode using two
283 difference scenarios: (i) we calculated an upper limit by assuming that all droplets which
284 underwent the processes discussed above froze by immersion freezing; and (ii) we calculated a
285 lower limit to the INP concentration by assuming that all droplets which underwent the processes
286 discussed above remained liquid until the homogeneous freezing temperature of approximately -
287 $37\text{ }^{\circ}\text{C}$ (Wheeler et al., 2015).

288 After $\#INPs(T)$ were determined for a given freezing experiment, atmospheric INP
289 number concentrations, $[INPs(T)]$, were calculated using the following equation:

$$290 \quad [INPs(T)] = \#INPs(T) \left(\frac{A_{\text{deposit}}}{A_{\text{DFT}}V} \right) f_{\text{nu},1\text{mm}} \quad (2)$$

291 where A_{deposit} is the total area of the sample deposit on the MOUDI impaction plate, A_{DFT} is the
292 area of the sample analyzed by the DFT (1.2 mm^2 in all samples), V is the volume of air sampled
293 by the MOUDI, and $f_{\text{nu},1\text{mm}}$ is a correction factor to account for non-uniformity in particle
294 concentration of the sample deposit at the 1 mm scale. See Mason et al. (2015a) for details.

295 Values of A_{deposit} , $f_{\text{nu},0.25-0.10\text{mm}}$, $f_{\text{nu},1\text{mm}}$, and f_{ne} are given in Tables S1 and S2 and discussed in
296 Section S1 of the Supplement. Reported INP number concentrations at each location are
297 averaged over all samples and have been adjusted to standard temperature and pressure. In
298 calculating averages over all sampling locations, measurements have not been weighted by
299 sample number.

300 For the experimental conditions used in the current study the maximum number
301 concentration of INPs that could be detected was roughly 20 L^{-1} . This maximum number
302 concentration is greater than that reported in Huffman et al. (2013) because in the current studies
303 shorter sampling times were used.

304 **3 Results and discussion**

305 **3.1 INP number concentrations**

306 The total number concentration of INPs active at -15, -20, and -25 °C are shown for each
307 site in Fig. 2. Freezing events were rare at temperatures warmer than -15 °C, accounting for only
308 1.3 % of all cases, and are therefore not reported. Some of the DFT experiments proceeded such
309 that all droplets froze at temperatures slightly below -25 °C. Since this scenario prohibits
310 calculation of INP number concentrations, -25 °C is the lowest temperature reported. As
311 expected, INP number concentrations were found to increase with decreasing freezing
312 temperature with the average concentration at -25 °C ($3.8 \pm 2.9 \text{ L}^{-1}$) being more than an order of
313 magnitude larger than at -15 °C ($0.25 \pm 0.15 \text{ L}^{-1}$).

314 INP number concentrations were relatively low at the Alert and Whistler Mountain sites
315 with values of 0.05 and 0.10 L^{-1} at -15 °C, 0.22 and 0.16 L^{-1} at -20 °C, and 0.99 and 1.1 L^{-1} at -25
316 °C, respectively. These findings are consistent with previous measurements at similar locations.

317 For example, Arctic measurements of Bigg (1996) for particles $< 10 \mu\text{m}$ and Fountain and
318 Ohtake (1985) using filter samples found mean INP concentrations of 0.01 L^{-1} at $-15 \text{ }^\circ\text{C}$ and 0.13
319 L^{-1} at $-20 \text{ }^\circ\text{C}$, respectively, with both deposition and condensation modes likely possible, and
320 Prenni et al. (2007) measured an average INP number concentration for particles with and
321 aerodynamic diameter $< 1.5 \mu\text{m}$ of approximately 0.33 L^{-1} between -8 and $-28 \text{ }^\circ\text{C}$ with
322 deposition, condensation, and immersion modes likely possible. At high elevation sites, Bowers
323 et al. (2009) at Mt. Werner in Colorado and Conen et al. (2012) at the research station
324 Jungfrauoch in Switzerland measured mean immersion-mode INP number concentrations of
325 approximately 0.02 L^{-1} at -10 and $-12 \text{ }^\circ\text{C}$, respectively, using filter samples of $0.2\text{--}0.3 \mu\text{m}$ pore
326 size. High elevation sites can receive large quantities of dust, which can act as efficient INPs at
327 lower temperatures (Chou et al., 2011), but this was unlikely during our measurement period
328 based on the low INP number concentrations.

329 INP number concentrations at Amphitrite Point were 0.23 , 0.94 , and 2.15 L^{-1} at droplet
330 freezing temperatures of -15 , -20 , and $-25 \text{ }^\circ\text{C}$, respectively. Despite the predominance of marine
331 air masses being sampled, the major source of INPs at Amphitrite Point during the study period
332 was likely biological particles from local vegetation (Mason et al., 2015b). Similar values were
333 measured in the Labrador Sea, where INP number concentrations at -15 , -20 , and $-25 \text{ }^\circ\text{C}$ were
334 0.38 , 1.3 , and 2.8 L^{-1} , respectively. These concentrations are consistent with previous
335 measurements within the marine boundary layer in regions influenced by air flow off of nearby
336 coasts, for instance those of Schnell (1977) in the immersion freezing mode for sizes $> 0.45 \mu\text{m}$
337 off the coast of Nova Scotia, roughly $1100\text{--}1500 \text{ km}$ southwest of our sampling site in the
338 Labrador Sea, and Rosinski et al. (1995) over the East China Sea in the deposition and
339 condensation modes for sizes $> 0.2 \mu\text{m}$. However, INP number concentrations found during

340 marine studies can vary by several orders of magnitude with changing location as summarized by
341 Burrows et al. (2013).

342 The highest concentrations of INPs at a freezing temperature of -25 °C were found at the
343 Colby sites, where the average number concentration was 8.9 L⁻¹. Aerosol sampling was
344 conducted adjacent to soya and sorghum fields during and following periods of combine
345 operation. This high concentration of INPs is consistent with previous work of Garcia et al.
346 (2012) that showed elevated concentrations of INPs downwind of corn fields during combine
347 harvesting, and Bowers et al. (2011) who found greater INP concentrations in air above cropland
348 than above suburban or forest sites.

349 The suburban sites of Saclay and the UBC campus also showed high INP concentrations,
350 being 4.4 and 6.1 L⁻¹, respectively, at a freezing temperature of -25 °C. Both sites were likely
351 influenced by multiple sources of INPs. For example, both are in close proximity to major
352 metropolitan centers and forest vegetation, which are potential sources of anthropogenic INPs
353 (e.g. Hobbs and Locatelli, 1970; Al-Naimi and Saunders, 1985; Knopf et al., 2010, 2014; Ebert
354 et al., 2011; Corbin et al., 2012; Cziczo et al., 2013; Brooks et al., 2014) and biological INPs
355 (e.g. Vali et al., 1976; Kieft and Ruscetti, 1990; Richard et al., 1996; Hirano and Upper, 2000;
356 Diehl et al., 2002; Prenni et al., 2009; Iannone et al., 2011; Pummer et al., 2012; Huffman et al.,
357 2013; Tobo et al., 2013; Haga et al., 2014; Wright et al., 2014), respectively. The sampling site at
358 Saclay was also within 1 km of agricultural fields, an additional source of biological aerosols that
359 may act as INPs (e.g. Lindow et al., 1982; Hirano et al., 1985; Georgakopoulos and Sands, 1992;
360 Möhler et al., 2008; Bowers et al., 2011; Garcia et al., 2012; Haga et al., 2013; Morris et al.,
361 2013; Hiranuma et al., 2015).

362 **3.2 INP size distributions**

363 Figure 3a shows the relative contribution of supermicron aerosol particles to the total
364 measured INP population. Note that the same particle size range was not investigated at all
365 locations with particles in the range of 0.10–0.18 μm not being measured at Whistler Mountain
366 or Amphitrite Point, and no uncertainty is reported for the Labrador Sea measurement as only a
367 single sample was available. Averaging over all sampling locations with a 1 SD uncertainty, $91 \pm$
368 9 , 79 ± 17 , and 63 ± 21 % of INPs had an aerodynamic diameter $> 1 \mu\text{m}$ at ice activation
369 temperatures of -15, -20, and -25 $^{\circ}\text{C}$, respectively. At -15 $^{\circ}\text{C}$, the percentage of supermicron
370 INPs ranged from 78 % at Whistler Mountain up to 100 % at the Labrador Sea and Colby sites.
371 At lower temperatures, there was more variation between samples: at -20 $^{\circ}\text{C}$ the percentage of
372 supermicron INPs ranged from 52 % at Whistler Mountain to 100 % over the Labrador Sea, and
373 at -25 $^{\circ}\text{C}$ the percentage of supermicron INPs ranged from 39 % at Whistler Mountain to 95 %
374 over the Labrador Sea.

375 Figure 3b shows the fraction of INPs that are in the coarse mode, calculated by assuming
376 that half of the INPs found in the 1.8–3.2 μm MOUDI size cut were larger than 2.5 μm (i.e. INPs
377 are uniformly distributed over this size range). Measurements of the total particle size
378 distribution were not available at all locations to test this assumption. Furthermore, it is not
379 known if the INP size distribution follows the total particle size distribution *a priori*. Averaging
380 over all sampling locations, the percentage of INPs in the coarse mode was 62 ± 20 , 55 ± 18 , and
381 42 ± 17 % (1 SD) at ice activation temperatures of -15, -20, and -25 $^{\circ}\text{C}$, respectively. The
382 percentage of INPs in the coarse mode was found to range from 38 % at Saclay to 91 % in Colby
383 at -15 $^{\circ}\text{C}$, from 26 % at Whistler Mountain to 73 % in Colby at -20 $^{\circ}\text{C}$, and from 20 % at Alert to
384 64 % at the Labrador Sea at -25 $^{\circ}\text{C}$. Despite great diversity in the studied locations, each had a
385 significant contribution from coarse mode particles to the measured INP population.

386 The median sizes of INPs at ice activation temperatures of -15, -20, and -25 °C are shown
387 in Fig. 4 with the 25th and 75th percentile values. At -15 °C, the median INP size is relatively
388 large at all locations, varying from 2.1 µm at Saclay to 4.7 µm at Colby with an average of $3.2 \pm$
389 $0.9 \mu\text{m}$ (1 SD). As droplet freezing temperature decreased, the median INP size also decreased,
390 with the exception of samples from the Labrador Sea and the UBC campus. At -25 °C, the
391 median size of INPs varied from 0.83 µm at Alert to 3.1 µm at the Labrador Sea site with an
392 average of $1.9 \pm 1.0 \mu\text{m}$. The median size of the INPs was $> 1 \mu\text{m}$ in all cases with the exception
393 of the Alert and Whistler Mountain sites at a freezing temperature of -25 °C. Alert, in the polar
394 tundra at high latitude, and Whistler Mountain, at high elevation and periodically in the free
395 troposphere, are remote with fewer local sources of aerosols.

396 In Fig. 4, the difference between the 75th and 25th percentile sizes is relatively small at all
397 locations at -15 °C. A narrow INP size distribution at -15 °C is consistent with a single type or
398 class of particles dominating freezing at this temperature. With decreasing temperature, the
399 interquartile range significantly increased: the 75th and 25th percentile INP sizes decreased by an
400 average factor of 1.2 and 3.7, respectively, between -15 and -25 °C, corresponding to a 63 %
401 increase in the average interquartile range.

402 INP size distributions are further explored in Fig. 5, where the fraction of the measured
403 INP number concentration found in each MOUDI size bin is shown. Colors on the red end of the
404 scale illustrate that a large fraction of the INPs measured at a particular location belong to that
405 particle size bin. Total INP number concentrations as a function of temperature are given in Fig.
406 S1 and histograms of the INP size distributions are given in Figs. S2–S8 of the Supplement.
407 Figure 5a shows that most INPs active at -15 °C were 1–10 µm in size. In particular, when
408 averaged over all locations, 72 % of the INPs active at -15 °C were between 1.8 and 5.6 µm.

409 Furthermore, the major mode (i.e. the global maximum in a size distribution) was always larger
410 than 1.8 μm . At lower freezing temperatures the INP size distributions broadened with increased
411 contributions from smaller aerosol particles, evident by the more uniform intensity of Fig. 5b and
412 c. By $-25\text{ }^{\circ}\text{C}$, six of the seven locations had a submicron INP mode, and at Alert and Whistler
413 Mountain this was the major mode. A general broadening of the INP size distribution with
414 decreasing temperature would be expected if there were an increase in the number of particle
415 types exhibiting ice activity with decreasing activation temperature.

416 Several previous studies have conducted size-resolved INP measurements. In most of
417 these studies, the fraction of INPs larger than a given particle size was not reported, and in some
418 cases the temperatures studied were different than in the current study. To better compare our
419 data with these previous studies, we have used the literature data to calculate the fraction of INPs
420 larger than either 1, 1.2, or 2.5 μm at temperatures as close as possible to the freezing
421 temperatures we used. Details of the calculations are presented in the Supplement, and the results
422 of the calculations are summarized in Table 2.

423 Table 2 shows that in six out of the nine previous studies, a large fraction (16–100 %) of
424 the INPs was found to be supermicron in size, consistent with the current study. Here the work of
425 Rosinski et al. (1986) is considered as two separate studies given the change in the investigated
426 mode of ice nucleation. Although the condensation and immersion freezing measurements
427 reported in Rosinski et al. (1986) appear contradictory, it is important to note that that the two
428 freezing results for Rosinski et al. (1986) shown in Table 2 correspond to different temperature
429 ranges (compare $-5\text{ }^{\circ}\text{C}$ to $-6\text{ }^{\circ}\text{C}$ for condensation and $-10.8\text{ }^{\circ}\text{C}$ for immersion freezing). One
430 possibility is that small particles dominated the INP population at the warmest temperatures
431 while larger particles dominated the INP population at the colder temperatures investigated by

432 Rosinski et al. (1986).

433 The ice nucleation efficiency of particles as a function of size have also been investigated
434 in laboratory experiments (e.g. Lüönd et al. (2010), Archuleta et al. (2005), Welti et al. (2009)).
435 In general this work has shown the ice nucleation efficiency increases as particle size increases.
436 Furthermore, several studies have investigated correlations between the concentrations of INPs
437 and aerosol particles above a certain size (Richardson et al., 2007; DeMott et al., 2010; Chou et
438 al., 2011; Field et al., 2012; Huffman et al., 2013; Prenni et al., 2013; Tobo et al., 2013; Ardon-
439 Dryer and Levin, 2014; Jiang et al., 2014, 2015). For example, using data from a variety of field
440 measurements DeMott et al. (2010) observed a correlation between INP concentrations and the
441 concentration of aerosol particles $> 0.5 \mu\text{m}$, and Ardon-Dryer and Levin (2014) found that INP
442 concentrations in Israel were better correlated to the concentration of aerosol particles $2.5\text{--}10 \mu\text{m}$
443 in size than those $< 2.5 \mu\text{m}$. These results are also consistent with INPs being relatively large in
444 size.

445 **4 Summary and conclusions**

446 INP number concentrations in the immersion mode as a function of size and droplet
447 freezing temperature were determined at six locations across North America and one in Europe.
448 INP number concentrations varied by as much as an order of magnitude between locations, and
449 were generally found to be lowest at the remote sites of Alert and Whistler Mountain and highest
450 at the agricultural sites of Colby and the suburban sites of Saclay and the UBC campus,
451 consistent with previous studies. Several key findings indicate the potential importance of large
452 INPs at ground level: (1) 91 ± 9 and 62 ± 20 % of INPs measured at $-15 \text{ }^\circ\text{C}$ across all locations
453 are supermicron or in the coarse mode, respectively; (2) at the lowest temperature analyzed, -25
454 $^\circ\text{C}$, 63 ± 21 and 42 ± 17 % of INPs across all locations remained in the supermicron regime and

455 coarse mode, respectively; (3) at $-15\text{ }^{\circ}\text{C}$, the median INP size was relatively large at all locations,
456 varying from $2.1\text{ }\mu\text{m}$ at Saclay to $4.7\text{ }\mu\text{m}$ at Colby with an average of $3.2 \pm 0.9\text{ }\mu\text{m}$; and (4) at -
457 $25\text{ }^{\circ}\text{C}$, the median size INP varied from $0.83\text{ }\mu\text{m}$ at Alert to $3.1\text{ }\mu\text{m}$ above the Labrador Sea with
458 an average of $1.9 \pm 1.0\text{ }\mu\text{m}$.

459 Our measurements indicate that, when averaged over all studied locations and
460 temperatures, 78 ± 19 and $53 \pm 20\%$ of immersion-mode INPs may be missed if either
461 supermicron particles or coarse mode particles are not sampled at ground sites. As noted in Sect.
462 1, some instrumentation for measuring ambient INP number concentrations restricts the upper
463 range of sampled aerosol particles. The data presented may be useful for estimating the fraction
464 of INPs not measured with these instruments at ground sites and in different environments.

465 All measurements used in this study were conducted at ground level and, apart from
466 those at Whistler Mountain, were also close to sea level. As the large contribution of
467 supermicron and coarse mode INPs to the overall INP population noted here may not necessarily
468 hold for higher altitudes, additional size-resolved INP measurements as a function of altitude are
469 needed. In obtaining such data, careful consideration will be needed toward sampling issues with
470 aerosol inlets and transfer through sample lines on aircraft platforms. It can be anticipated that
471 supermicron particle transfer will be strongly restricted without special inlets or care, and this
472 may especially impact INP sampling.

473 A caveat to this study is that our measurements were confined to aerosol particle sizes
474 greater than either 0.10 or $0.18\text{ }\mu\text{m}$. If there were a significant contribution from INPs of smaller
475 sizes (Vali, 1966; Schnell and Vali, 1973; Pummer et al., 2012; Augustin et al., 2013; Fröhlich-
476 Nowoisky et al., 2015; O’Sullivan et al., 2015; Tong et al., 2015; Wilson et al., 2015), and these
477 smaller sizes did not coagulate or get scavenged by larger particles, the values presented here

478 would represent upper limits to the contribution of supermicron and coarse mode particles to the
479 total INP population. Additional studies exploring the relative atmospheric abundance of INPs <
480 0.10 μm are necessary (Hader et al., 2014). Future studies of the size distribution of INPs should
481 also include measurements of particle mixing state to determine if particles are internally or
482 externally mixed at the locations where the size distribution of INPs are being measured.

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502 **References**

- 503 Al-Naimi, R. and Saunders, C. P. R.: Measurements of natural deposition and condensation-
504 freezing ice nuclei with a continuous flow chamber, *Atmos. Environ.*, 19, 1871–1882,
505 doi:10.1016/0004-6981(85)90012-5, 1985.
- 506 Archuleta, C. M., DeMott, P. J. and Kreidenweis, S. M.: Ice nucleation by surrogates for
507 atmospheric mineral dust and mineral dust/sulfate particles at cirrus temperatures, *Atmos. Chem.*
508 *Phys.*, 5, 2617–2634, doi:10.5194/acp-5-2617-2005, 2005.
- 509 Ardon-Dryer, K. and Levin, Z.: Ground-based measurements of immersion freezing in the
510 eastern Mediterranean, *Atmos. Chem. Phys.*, 14, 5217–5231, doi:10.5194/acp-14-5217-2014,
511 2014.
- 512 Ariya, P. A., Sun, J., Eltouny, N. A., Hudson, E. D., Hayes, C. T. and Kos, G.: Physical and
513 chemical characterization of bioaerosols - Implications for nucleation processes, *Int. Rev. Phys.*
514 *Chem.*, 28, 1–32, doi:10.1080/01442350802597438, 2009.
- 515 Augustin, S., Wex, H., Niedermeier, D., Pummer, B., Grothe, H., Hartmann, S., Tomsche, L.,
516 Clauss, T., Voigtländer, J., Ignatius, K. and Stratmann, F.: Immersion freezing of birch pollen
517 washing water, *Atmos. Chem. Phys.*, 13, 10989–11003, doi:10.5194/acp-13-10989-2013, 2013.
- 518 Baisnée, D., Thibaudon, M., Baumier, R., McMeeking, G., Kok, G., O'Connor, D., Sodeau, J.,
519 Huffman, J. A., Lassar, W., Pierce, K., Gallagher, M., Crawford, I., Salines, G. and Sarda-
520 Estève, R.: Simultaneous Real-time Fluorescence and Microscopy Measurements of Bioaerosols
521 during the BIODTECT 2014 Campaign in Paris Area, AAAR 33rd Annual Conference,
522 Orlando, FL, October 20–24, 2014.
- 523 Baker, M. B.: Cloud Microphysics and Climate, *Science*, 276, 1072–1078,
524 doi:10.1126/science.276.5315.1072, 1997.
- 525 Baker, M. B. and Peter, T.: Small-scale cloud processes and climate, *Nature*, 451, 299–300,
526 doi:10.1038/nature06594, 2008.
- 527 Bateman, A. P., Belassein, H. and Martin, S. T.: Impactor Apparatus for the Study of Particle
528 Rebound: Relative Humidity and Capillary Forces, *Aerosol Sci. Technol.*, 48, 42–52,
529 doi:10.1080/02786826.2013.853866, 2014.
- 530 Berezinski, N. A., Stepanov, G. V. and Khorguani, V. G.: Ice-forming activity of atmospheric
531 aerosol particles of different sizes, *Lecture Notes in Physics*, edited by P. E. Wagner and G. Vali,
532 309, 709–712, Springer, Heidelberg and Berlin, Germany, 1988.
- 533 Bigg, E. K.: Ice forming nuclei in the high Arctic, *Tellus B*, 48, 223–233,
534 doi:10.3402/tellusb.v48i2.15888, 1996.
- 535 Bowers, R. M., Lauber, C. L., Wiedinmyer, C., Hamady, M., Hallar, A. G., Fall, R., Knight, R.
536 and Fierer, N.: Characterization of Airborne Microbial Communities at a High-Elevation Site
537 and Their Potential To Act as Atmospheric Ice Nuclei, *Appl. Environ. Microbiol.*, 75, 5121–
538 5130, doi:10.1128/AEM.00447-09, 2009.

539 Bowers, R. M., McLetchie, S., Knight, R. and Fierer, N.: Spatial variability in airborne bacterial
540 communities across land-use types and their relationship to the bacterial communities of
541 potential source environments, *ISME J.*, 5, 601–612, doi:10.1038/ismej.2010.167, 2011.

542 Brooks, S. D., Suter, K. and Olivarez, L.: Effects of Chemical Aging on the Ice Nucleation
543 Activity of Soot and Polycyclic Aromatic Hydrocarbon Aerosols, *J. Phys. Chem. A*, 118, 10036–
544 10047, doi:10.1021/jp508809y, 2014.

545 Burrows, S. M., Butler, T., Jöckel, P., Tost, H., Kerkweg, A., Pöschl, U. and Lawrence, M. G.:
546 Bacteria in the global atmosphere - Part 2: Modeling of emissions and transport between
547 different ecosystems, *Atmos. Chem. Phys.*, 9, 9281–9297, doi:10.5194/acp-9-9281-2009, 2009.

548 Burrows, S. M., Hoose, C., Pöschl, U. and Lawrence, M. G.: Ice nuclei in marine air: biogenic
549 particles or dust?, *Atmos. Chem. Phys.*, 13, 245–267, doi:10.5194/acp-13-245-2013, 2013.

550 Chen, S.-C., Tsai, C.-J., Chen, H.-D., Huang, C.-Y. and Roam, G.-D.: The Influence of Relative
551 Humidity on Nanoparticle Concentration and Particle Mass Distribution Measurements by the
552 MOUDI, *Aerosol Sci. Technol.*, 45, 596–603, doi:10.1080/02786826.2010.551557, 2011.

553 Chou, C., Stetzer, O., Weingartner, E., Jurányi, Z., Kanji, Z. A. and Lohmann, U.: Ice nuclei
554 properties within a Saharan dust event at the Jungfraujoch in the Swiss Alps, *Atmos. Chem.*
555 *Phys.*, 11, 4725–4738, doi:10.5194/acp-11-4725-2011, 2011.

556 Clarke, A. D., Shinozuka, Y., Kapustin, V. N., Howell, S., Huebert, B., Doherty, S., Anderson,
557 T., Covert, D., Anderson, J., Hua, X., Moore II, K. G., McNaughton, C., Carmichael, G. and
558 Weber, R.: Size distributions and mixtures of dust and black carbon aerosol in Asian outflow:
559 Physiochemistry and optical properties, *J. Geophys. Res.*, 109, D15S09,
560 doi:10.1029/2003JD004378, 2004.

561 Cobbett, F. D., Steffen, A., Lawson, G. and Van Heyst, B. J.: GEM fluxes and atmospheric
562 mercury concentrations (GEM, RGM and Hg^p) in the Canadian Arctic at Alert, Nunavut, Canada
563 (February-June 2005), *Atmos. Environ.*, 41, 6527–6543, doi:10.1016/j.atmosenv.2007.04.033,
564 2007.

565 Conen, F., Henne, S., Morris, C. E. and Alewell, C.: Atmospheric ice nucleators active ≥ -12 °C
566 can be quantified on PM₁₀ filters, *Atmos. Meas. Tech.*, 5, 321–327, doi:10.5194/amt-5-321-2012,
567 2012.

568 Corbin, J. C., Rehbein, P. J. G., Evans, G. J. and Abbatt, J. P. D.: Combustion particles as ice
569 nuclei in an urban environment: Evidence from single-particle mass spectrometry, *Atmos.*
570 *Environ.*, 51, 286–292, doi:10.1016/j.atmosenv.2012.01.007, 2012.

571 Creamean, J. M., Suski, K. J., Rosenfeld, D., Cazorla, A., DeMott, P. J., Sullivan, R. C., White,
572 A. B., Ralph, F. M., Minnis, P., Comstock, J. M., Tomlinson, J. M. and Prather, K. A.: Dust and
573 Biological Aerosols from the Sahara and Asia Influence Precipitation in the Western U.S.,
574 *Science*, 339, 1572–1578, doi:10.1126/science.1227279, 2013.

575 Cziczo, D. J., Froyd, K. D., Hoose, C., Jensen, E. J., Diao, M., Zondlo, M. A., Smith, J. B.,
576 Twohy, C. H. and Murphy, D. M.: Clarifying the Dominant Sources and Mechanisms of Cirrus
577 Cloud Formation, *Science*, 340, 1320–1324, doi:10.1126/science.1234145, 2013.

578 Dahneke, B.: The capture of aerosol particles by surfaces, *J. Colloid Interface Sci.*, 37, 342–353,
579 doi:10.1016/0021-9797(71)90302-X, 1971.

580 DeMott, P. J., Cziczo, D. J., Prenni, A. J., Murphy, D. M., Kreidenweis, S. M., Thomson, D. S.,
581 Borys, R. and Rogers, D. C.: Measurements of the concentration and composition of nuclei for
582 cirrus formation, *P. Natl. Acad. Sci. USA*, 100, 14655–14660, doi:10.1073/pnas.2532677100,
583 2003.

584 DeMott, P. J., Möhler, O., Stetzer, O., Vali, G., Levin, Z., Petters, M. D., Murakami, M., Leisner,
585 T., Bundke, U., Klein, H., Kanji, Z. A., Cotton, R., Jones, H., Benz, S., Brinkmann, M.,
586 Rzesanke, D., Saathoff, H., Nicolet, M., Saito, A., Nillius, B., Bingemer, H., Abbatt, J., Ardon,
587 K., Ganor, E., Georgakopoulos, D. G. and Saunders, C.: Resurgence in ice nuclei measurement
588 research, *B. Am. Meteorol. Soc.*, 92, 1623–1635, doi:10.1175/2011BAMS3119.1, 2011.

589 DeMott, P. J., Prenni, A. J., Liu, X., Kreidenweis, S. M., Petters, M. D., Twohy, C. H.,
590 Richardson, M. S., Eidhammer, T. and Rogers, D. C.: Predicting global atmospheric ice nuclei
591 distributions and their impacts on climate, *P. Natl. Acad. Sci. USA*, 107, 11217–11222,
592 doi:10.1073/pnas.0910818107, 2010.

593 DeMott, P. J., Rogers, D. C., Kreidenweis, S. M., Chen, Y., Twohy, C. H., Baumgardner, D.,
594 Heymsfield, A. J., and Chan, K. R.: The role of heterogeneous freezing nucleation in upper
595 tropospheric clouds: Inferences from SUCCESS, *Geophys. Res. Lett.*, 25,
596 doi:10.1029/97GL03779, 1387–1390, 1998.

597 Després, V. R., Huffman, J. A., Burrows, S. M., Hoose, C., Safatov, A. S., Buryak, G., Fröhlich-
598 Nowoisky, J., Elbert, W., Andreae, M. O., Pöschl, U. and Jaenicke, R.: Primary biological
599 aerosol particles in the atmosphere: a review, *Tellus B*, 64, 15598,
600 doi:10.3402/tellusb.v64i0.15598, 2012.

601 Diehl, K., Matthias-Maser, S., Jaenicke, R. and Mitra, S. K.: The ice nucleating ability of pollen:
602 Part II. Laboratory studies in immersion and contact freezing modes, *Atmos. Res.*, 61, 125–133,
603 doi:10.1016/S0169-8095(01)00132-6, 2002.

604 Draxler, R. R. and Rolph, G. D.: HYSPLIT (HYbrid Single-Particle Lagrangian Integrated
605 Trajectory) Model access via NOAA ARL READY Website, available at:
606 <http://ready.arl.noaa.gov/HYSPLIT.php> (last access: 27 May 2014), NOAA Air Resources
607 Laboratory, Silver Spring, MD, 2014.

608 Ebert, M., Worrigen, A., Benker, N., Mertes, S., Weingartner, E. and Weinbruch, S.: Chemical
609 composition and mixing-state of ice residuals sampled within mixed phase clouds, *Atmos.*
610 *Chem. Phys.*, 11, 2805–2816, doi:10.5194/acp-11-2805-2011, 2011.

611 Eidhammer, T., DeMott, P. J., Prenni, A. J., Petters, M. D., Twohy, C. H., Rogers, D. C., Stith,
612 J., Heymsfield, A., Wang, Z., Pratt, K. A., Prather, K. A., Murphy, S. M., Seinfeld, J. H.,
613 Subramanian, R. and Kreidenweis, S. M.: Ice Initiation by Aerosol Particles: Measured and
614 Predicted Ice Nuclei Concentrations versus Measured Ice Crystal Concentrations in an
615 Orographic Wave Cloud, *J. Atmos. Sci.*, 67, 2417–2436, doi:10.1175/2010JAS3266.1, 2010.

616 Elbert, W., Taylor, P. E., Andreae, M. O. and Pöschl, U.: Contribution of fungi to primary
617 biogenic aerosols in the atmosphere: wet and dry discharged spores, carbohydrates, and
618 inorganic ions, *Atmos. Chem. Phys.*, 7, 4569–4588, doi:10.5194/acp-7-4569-2007, 2007.

619 Fang, C. P., McMurry, P. H., Marple, V. A. and Rubow, K. L.: Effect of Flow-induced Relative
620 Humidity Changes on Size Cuts for Sulfuric Acid Droplets in the Microorifice Uniform Deposit

621 Impactor (MOUDI), *Aerosol Sci. Technol.*, 14, 266–277, doi:10.1080/02786829108959489,
622 1991.

623 Field, P. R., Heymsfield, A. J., Shipway, B. J., DeMott, P. J., Pratt, K. A., Rogers, D. C., Stith, J.
624 and Prather, K. A.: Ice in Clouds Experiment-Layer Clouds. Part II: Testing Characteristics of
625 Heterogeneous Ice Formation in Lee Wave Clouds, *J. Atmos. Sci.*, 69, 1066–1079,
626 doi:10.1175/JAS-D-11-026.1, 2012.

627 Fountain, A. G. and Ohtake, T.: Concentrations and Source Areas of Ice Nuclei in the Alaskan
628 Atmosphere, *J. Clim. Appl. Meteorol.*, 24, 377–382, 1985.

629 Friedman, B., Kulkarni, G., Beránek, J., Zelenyuk, A., Thornton, J. A. and Cziczo, D. J.: Ice
630 nucleation and droplet formation by bare and coated soot particles, *J. Geophys. Res.*, 116,
631 D17203, doi:10.1029/2011JD015999, 2011.

632 Fröhlich-Nowoisky, J., Hill, T. C. J., Pummer, B. G., Yordanova, P., Franc, G. D. and Pöschl,
633 U.: Ice nucleation activity in the widespread soil fungus *Mortierella alpina*, *Biogeosciences*, 12,
634 1057–1071, doi:10.5194/bg-12-1057-2015, 2015.

635 Gallagher, J. P., McKendry, I. G., Macdonald, A. M. and Leitch, W. R.: Seasonal and Diurnal
636 Variations in Aerosol Concentration on Whistler Mountain: Boundary Layer Influence and
637 Synoptic-Scale Controls, *J. Appl. Meteorol. Climatol.*, 50, 2210–2222, doi:10.1175/JAMC-D-
638 11-028.1, 2011.

639 Garcia, E., Hill, T. C. J., Prenni, A. J., DeMott, P. J., Franc, G. D. and Kreidenweis, S. M.:
640 Biogenic ice nuclei in boundary layer air over two U.S. High Plains agricultural regions, *J.*
641 *Geophys. Res.*, 117, D18209, doi:10.1029/2012JD018343, 2012.

642 Georgakopoulos, D. G. and Sands, D. C.: Epiphytic populations of *Pseudomonas syringae* on
643 barley, *Can. J. Microbiol.*, 38, 111–114, doi:10.1139/m92-018, 1992.

644 Graham, B., Guyon, P., Maenhaut, W., Taylor, P. E., Ebert, M., Matthias-Maser, S., Mayol-
645 Bracero, O. L., Godoi, R. H. M., Artaxo, P., Meixner, F. X., Moura, M. A. L., Rocha, C. H. E.
646 D., Van Grieken, R., Glovsky, M. M., Flagan, R. C. and Andreae, M. O.: Composition and
647 diurnal variability of the natural Amazonian aerosol, *J. Geophys. Res.*, 108, 4765,
648 doi:10.1029/2003JD004049, 2003.

649 Hader, J. D., Wright, T. P. and Petters, M. D.: Contribution of pollen to atmospheric ice nuclei
650 concentrations, *Atmos. Chem. Phys.*, 14, 5433–5449, doi:10.5194/acp-14-5433-2014, 2014.

651 Haga, D. I., Burrows, S. M., Iannone, R., Wheeler, M. J., Mason, R. H., Chen, J., Polishchuk, E.
652 A., Pöschl, U. and Bertram, A. K.: Ice nucleation by fungal spores from the classes
653 *Agaricomycetes*, *Ustilaginomycetes*, and *Eurotiomycetes*, and the effect on the atmospheric
654 transport of these spores, *Atmos. Chem. Phys.*, 14, 8611–8630, doi:10.5194/acp-14-8611-2014,
655 2014.

656 Haga, D. I., Iannone, R., Wheeler, M. J., Mason, R., Polishchuk, E. A., Fetch Jr., T., van der
657 Kamp, B. J., McKendry, I. G. and Bertram, A. K.: Ice nucleation properties of rust and bunt
658 fungal spores and their transport to high altitudes, where they can cause heterogeneous freezing,
659 *J. Geophys. Res.-Atmos.*, 118, 7260–7272, doi:10.1002/jgrd.50556, 2013.

660 Hirano, S. S., Baker, L. S. and Upper, C. D.: Ice nucleation temperature of individual leaves in
661 relation to population sizes of ice nucleation active bacteria and frost injury, *Plant Physiol.*, 77,
662 259–265, doi:10.1104/pp.77.2.259, 1985.

663 Hirano, S. S. and Upper, C. D.: Bacteria in the leaf ecosystem with emphasis on *Pseudomonas*
664 *syringae* - a pathogen, ice nucleus, and epiphyte, *Microbiol. Mol. Biol. R.*, 64, 624–653,
665 doi:10.1128/MMBR.64.3.624-653.2000, 2000.

666 Hiranuma, N., Möhler, O., Yamashita, K., Tajiri, T., Saito, A., Kiselev, A., Hoffmann, N.,
667 Hoose, C., Jantsch, E., Koop, T. and Murakami, M.: Ice nucleation by cellulose and its potential
668 contribution to ice formation in clouds, *Nat. Geosci.*, 8, 273–277, doi:10.1038/ngeo2374, 2015.

669 Hobbs, P. V and Locatelli, J. D.: Ice Nucleus Measurements at Three Sites in Western
670 Washington, *J. Atmos. Sci.*, 27, 90–100, 1970.

671 Hoose, C., Kristjánsson, J. E. and Burrows, S. M.: How important is biological ice nucleation in
672 clouds on a global scale?, *Environ. Res. Lett.*, 5, 024009, doi:10.1088/1748-9326/5/2/024009,
673 2010a.

674 Hoose, C., Kristjánsson, J. E., Chen, J.-P. and Hazra, A.: A Classical-Theory-Based
675 Parameterization of Heterogeneous Ice Nucleation by Mineral Dust, Soot, and Biological
676 Particles in a Global Climate Model, *J. Atmos. Sci.*, 67, 2483–2503,
677 doi:10.1175/2010JAS3425.1, 2010b.

678 Hoose, C. and Möhler, O.: Heterogeneous ice nucleation on atmospheric aerosols: a review of
679 results from laboratory experiments, *Atmos. Chem. Phys.*, 12, 9817–9854, doi:10.5194/acp-12-
680 9817-2012, 2012.

681 Hoyle, C. R., Pinti, V., Welti, A., Zobrist, B., Marcolli, C., Luo, B., Höskuldsson, Á., Mattsson,
682 H. B., Stetzer, O., Thorsteinsson, T., Larsen, G. and Peter, T.: Ice nucleation properties of
683 volcanic ash from Eyjafjallajökull, *Atmos. Chem. Phys.*, 11, 9911–9926, doi:10.5194/acp-11-
684 9911-2011, 2011.

685 Huffman, J. A., Prenni, A. J., DeMott, P. J., Pöhlker, C., Mason, R. H., Robinson, N. H.,
686 Fröhlich-Nowoisky, J., Tobo, Y., Després, V. R., Garcia, E., Gochis, D. J., Harris, E., Müller-
687 Germann, I., Ruzene, C., Schmer, B., Sinha, B., Day, D. A., Andreae, M. O., Jimenez, J. L.,
688 Gallagher, M., Kreidenweis, S. M., Bertram, A. K. and Pöschl, U.: High concentrations of
689 biological aerosol particles and ice nuclei during and after rain, *Atmos. Chem. Phys.*, 13, 6151–
690 6164, doi:10.5194/acp-13-6151-2013, 2013.

691 Huffman, J. A., Sinha, B., Garland, R. M., Snee-Pollmann, A., Gunthe, S. S., Artaxo, P., Martin,
692 S. T., Andreae, M. O. and Pöschl, U.: Size distributions and temporal variations of biological
693 aerosol particles in the Amazon rainforest characterized by microscopy and real-time UV-APS
694 fluorescence techniques during AMAZE-08, *Atmos. Chem. Phys.*, 12, 11997–12019,
695 doi:10.5194/acp-12-11997-2012, 2012.

696 Iannone, R., Chernoff, D. I., Pringle, A., Martin, S. T. and Bertram, A. K.: The ice nucleation
697 ability of one of the most abundant types of fungal spores found in the atmosphere, *Atmos.*
698 *Chem. Phys.*, 11, 1191–1201, doi:10.5194/acp-11-1191-2011, 2011.

699 Jiang, H., Yin, Y., Su, H., Shan, Y. and Gao, R.: The characteristics of atmospheric ice nuclei
700 measured at the top of Huangshan (the Yellow Mountains) in Southeast China using a newly

701 built static vacuum water vapor diffusion chamber, *Atmos. Res.*, 153, 200–208,
702 doi:10.1016/j.atmosres.2014.08.015, 2015.

703 Jiang, H., Yin, Y., Yang, L., Yang, S., Su, H. and Chen, K.: The characteristics of atmospheric
704 ice nuclei measured at different altitudes in the Huangshan Mountains in Southeast China, *Adv.*
705 *Atmos. Sci.*, 31, 396–406, doi:10.1007/s00376-013-3048-5, 2014.

706 Kieft, T. L. and Ruscetti, T.: Characterization of biological ice nuclei from a lichen, *J. Bacteriol.*,
707 172, 3519–3523, 1990.

708 Knopf, D. A., Alpert, P. A., Wang, B., O'Brien, R. E., Kelly, S. T., Laskin, A., Gilles, M. K. and
709 Moffet, R. C.: Microspectroscopic imaging and characterization of individually identified ice
710 nucleating particles from a case field study, *J. Geophys. Res.-Atmos.*, 119, 10365–10381,
711 doi:10.1002/2014JD021866, 2014.

712 Knopf, D. A., Wang, B., Laskin, A., Moffet, R. C. and Gilles, M. K.: Heterogeneous nucleation
713 of ice on anthropogenic organic particles collected in Mexico City, *Geophys. Res. Lett.*, 37,
714 L11803, doi:10.1029/2010GL043362, 2010.

715 Koop, T., Luo, B., Biermann, U. M., Crutzen, P. J. and Peter, T.: Freezing of HNO₃/H₂SO₄/H₂O
716 solutions at stratospheric temperatures: Nucleation statistics and experiments, *J. Phys. Chem. A*,
717 101, 1117–1133, doi:10.1021/jp9626531, 1997.

718 Koop, T., Ng, H. P., Molina, L. T. and Molina, M. J.: A New Optical Technique to Study
719 Aerosol Phase Transitions: The Nucleation of Ice from H₂SO₄ Aerosols, *J. Phys. Chem. A*, 102,
720 8924–8931, doi:10.1021/jp9828078, 1998.

721 Lawson, D. R.: Impaction surface coatings intercomparison and measurements with cascade
722 impactors, *Atmos. Environ.*, 14, 195–199, doi:10.1016/0004-6981(80)90278-4, 1980.

723 Lindow, S. E., Arny, D. C. and Upper, C. D.: Bacterial ice nucleation: a factor in frost injury to
724 plants, *Plant Physiol.*, 70, 1084–1089, doi:10.1104/pp.70.4.1084, 1982.

725 Lohmann, U. and Feichter, J.: Global indirect aerosol effects: a review, *Atmos. Chem. Phys.*, 5,
726 715–737, doi:10.5194/acp-5-715-2005, 2005.

727 Lüönd, F., Stetzer, O., Welti, A. and Lohmann, U.: Experimental study on the ice nucleation
728 ability of size-selected kaolinite particles in the immersion mode, *J. Geophys. Res.*, 115,
729 D14201, doi:10.1029/2009JD012959, 2010.

730 Macdonald, A. M., Anlauf, K. G., Leitch, W. R., Chan, E. and Tarasick, D. W.: Interannual
731 variability of ozone and carbon monoxide at the Whistler high elevation site: 2002–2006, *Atmos.*
732 *Chem. Phys.*, 11, 11431–11446, doi:10.5194/acp-11-11431-2011, 2011.

733 Marple, V. A., Rubow, K. L. and Behm, S. M.: A Microorifice Uniform Deposit Impactor
734 (MOUDI): Description, Calibration, and Use, *Aerosol Sci. Technol.*, 14, 434–446,
735 doi:10.1080/02786829108959504, 1991.

736 Mason, R. H., Chou, C., McCluskey, C. S., Levin, E. J. T., Schiller, C. L., Hill, T. C. J.,
737 Huffman, J. A., DeMott, P. J. and Bertram, A. K.: The micro-orifice uniform deposit impactor-
738 droplet freezing technique (MOUDI-DFT) for measuring concentrations of ice nucleating
739 particles as a function of size: improvements and initial validation, *Atmos. Meas. Tech.*, 8, 2449–
740 2462, doi:10.5194/amt-8-2449-2015, 2015a.

741 Mason, R. H., Si, M., Li, J., Chou, C., Dickie, R., Toom-Sauntry, D., Pöhlker, C., Yakobi-
742 Hancock, J. D., Ladino, L. A., Jones, K., Leaitch, W. R., Schiller, C. L., Abbatt, J. P. D.,
743 Huffman, J. A. and Bertram, A. K.: Ice nucleating particles at a coastal marine boundary layer
744 site: correlations with aerosol type and meteorological conditions, *Atmos. Chem. Phys. Discuss.*,
745 15, 16273–16323, doi:10.5194/acpd-8-16273-2015, 2015b.

746 McCluskey, C. S., DeMott, P. J., Prenni, A. J., Levin, E. J. T., McMeeking, G. R., Sullivan, A.
747 P., Hill, T. C. J., Nakao, S., Carrico, C. M. and Kreidenweis, S. M.: Characteristics of
748 atmospheric ice nucleating particles associated with biomass burning in the US: Prescribed burns
749 and wildfires, *J. Geophys. Res.-Atmos.*, 119, 10458–10470, doi:10.1002/2014JD021980, 2014.

750 McKendry, I. G., Christensen, E., Schiller, C. L., Vingarzan, R., Macdonald, A. M. and Li, Y.:
751 Low Ozone Episodes at Amphitrite Point Marine Boundary Layer Observatory, British
752 Columbia, Canada, *Atmos. Ocean*, 52, 271–280, doi:10.1080/07055900.2014.910164, 2014.

753 Mertes, S., Verheggen, B., Walter, S., Connolly, P., Ebert, M., Schneider, J., Bower, K. N.,
754 Cozic, J., Weinbruch, S., Baltensperger, U. and Weingartner, E.: Counterflow Virtual Impactor
755 Based Collection of Small Ice Particles in Mixed-Phase Clouds for the Physico-Chemical
756 Characterization of Tropospheric Ice Nuclei: Sampler Description and First Case Study, *Aerosol
757 Sci. Technol.*, 41, 848–864, doi:10.1080/02786820701501881, 2007.

758 Möhler, O., DeMott, P. J., Vali, G. and Levin, Z.: Microbiology and atmospheric processes: the
759 role of biological particles in cloud physics, *Biogeosciences*, 4, 1059–1071, doi:10.5194/bg-4-
760 1059-2007, 2007.

761 Möhler, O., Georgakopoulos, D. G., Morris, C. E., Benz, S., Ebert, V., Hunsmann, S., Saathoff,
762 H., Schnaiter, M. and Wagner, R.: Heterogeneous ice nucleation activity of bacteria: new
763 laboratory experiments at simulated cloud conditions, *Biogeosciences*, 5, 1425–1435,
764 doi:10.5194/bg-5-1425-2008, 2008.

765 Morris, C. E., Georgakopoulos, D. G. and Sands, D. C.: Ice nucleation active bacteria and their
766 potential role in precipitation, *J. Phys. IV*, 121, 87–103, doi:10.1051/jp4:2004121004, 2004.

767 Morris, C. E., Sands, D. C., Glaux, C., Samsatly, J., Asaad, S., Moukahel, A. R., Gonçalves, F.
768 L. T. and Bigg, E. K.: Urediospores of rust fungi are ice nucleation active at > -10 °C and harbor
769 ice nucleation active bacteria, *Atmos. Chem. Phys.*, 13, 4223–4233, doi:10.5194/acp-13-4223-
770 2013, 2013.

771 Murray, B. J., Broadley, S. L., Wilson, T. W., Atkinson, J. D. and Wills, R. H.: Heterogeneous
772 freezing of water droplets containing kaolinite particles, *Atmos. Chem. Phys.*, 11, 4191–4207,
773 doi:10.5194/acp-11-4191-2011, 2011.

774 Murray, B. J., O’Sullivan, D., Atkinson, J. D. and Webb, M. E.: Ice nucleation by particles
775 immersed in supercooled cloud droplets, *Chem. Soc. Rev.*, 41, 6519–6554,
776 doi:10.1039/c2cs35200a, 2012.

777 O’Sullivan, D., Murray, B. J., Ross, J. F., Whale, T. F., Price, H. C., Atkinson, J. D., Umo, N. S.
778 and Webb, M. E.: The relevance of nanoscale biological fragments for ice nucleation in clouds,
779 *Sci. Rep.*, 5, 8082, doi:10.1038/srep08082, 2015.

780 Pratt, K. A., DeMott, P. J., French, J. R., Wang, Z., Westphal, D. L., Heymsfield, A. J., Twohy,
781 C. H., Prenni, A. J. and Prather, K. A.: In situ detection of biological particles in cloud ice-
782 crystals, *Nat. Geosci.*, 2, 398–401, doi:10.1038/ngeo521, 2009.

783 Prenni, A. J., Harrington, J. Y., Tjernström, M., DeMott, P. J., Avramov, A., Long, C. N.,
784 Kreidenweis, S. M., Olsson, P. Q. and Verlinde, J.: Can Ice-Nucleating Aerosols Affect Arctic
785 Seasonal Climate?, *B. Am. Meteorol. Soc.*, 88, 541–550, doi:10.1175/BAMS-88-4-541, 2007.

786 Prenni, A. J., Petters, M. D., Kreidenweis, S. M., Heald, C. L., Martin, S. T., Artaxo, P., Garland,
787 R. M., Wollny, A. G. and Pöschl, U.: Relative roles of biogenic emissions and Saharan dust as
788 ice nuclei in the Amazon basin, *Nat. Geosci.*, 2, 402–405, doi:10.1038/ngeo517, 2009.

789 Prenni, A. J., Tobo, Y., Garcia, E., DeMott, P. J., Huffman, J. A., McCluskey, C. S.,
790 Kreidenweis, S. M., Prenni, J. E., Pöhlker, C. and Pöschl, U.: The impact of rain on ice nuclei
791 populations at a forested site in Colorado, *Geophys. Res. Lett.*, 40, 227–231,
792 doi:10.1029/2012GL053953, 2013.

793 Pummer, B. G., Bauer, H., Bernardi, J., Bleicher, S. and Grothe, H.: Suspendable
794 macromolecules are responsible for ice nucleation activity of birch and conifer pollen, *Atmos.*
795 *Chem. Phys.*, 12, 2541–2550, doi:10.5194/acp-12-2541-2012, 2012.

796 Richard, C., Martin, J.-G. and Pouleur, S.: Ice nucleation activity identified in some
797 phytopathogenic *Fusarium* species, *Phytoprotection*, 77, 83–92, doi:10.7202/706104ar, 1996.

798 Richardson, M. S., DeMott, P. J., Kreidenweis, S. M., Cziczo, D. J., Dunlea, E. J., Jimenez, J. L.,
799 Thomson, D. S., Ashbaugh, L. L., Borys, R. D., Westphal, D. L., Casuccio, G. S. and Lersch, T.
800 L.: Measurements of heterogeneous ice nuclei in the western United States in springtime and
801 their relation to aerosol characteristics, *J. Geophys. Res.*, 112, D02209,
802 doi:10.1029/2006JD007500, 2007.

803 Rogers, D. C., DeMott, P. J. and Kreidenweis, S. M.: Airborne measurements of tropospheric
804 ice-nucleating aerosol particles in the Arctic spring, *J. Geophys. Res.*, 106, 15053–15063,
805 doi:10.1029/2000JD900790, 2001a.

806 Rogers, D. C., DeMott, P. J., Kreidenweis, S. M. and Chen, Y.: Measurements of ice nucleating
807 aerosols during SUCCESS, *Geophys. Res. Lett.*, 25, 1383–1386, doi:10.1029/97GL03478, 1998.

808 Rogers, D. C., DeMott, P. J., Kreidenweis, S. M. and Chen, Y.: A continuous-flow diffusion
809 chamber for airborne measurements of ice nuclei, *J. Atmos. Ocean. Technol.*, 18, 725–741,
810 2001b.

811 Rosinski, J., Haagenson, P. L., Nagamoto, C. T. and Parungo, F.: Ice-forming nuclei of maritime
812 origin, *J. Aerosol Sci.*, 17, 23–46, doi:10.1016/0021-8502(86)90004-2, 1986.

813 Rosinski, J., Haagenson, P. L., Nagamoto, C. T., Quintana, B., Parungo, F. and Hoyt, S. D.: Ice-
814 forming nuclei in air masses over the Gulf of Mexico, *J. Aerosol Sci.*, 19, 539–551,
815 doi:10.1016/0021-8502(88)90206-6, 1988.

816 Rosinski, J., Nagamoto, C. T. and Zhou, M. Y.: Ice-forming nuclei over the East China Sea,
817 *Atmos. Res.*, 36, 95–105, doi:10.1016/0169-8095(94)00029-D, 1995.

818 Rucklidge, J.: The Examination by Electron Microscope of Ice Crystal Nuclei from Cloud
819 Chamber Experiments, *J. Atmos. Sci.*, 22, 301–308, 1965.

820 Santachiara, G., Di Matteo, L., Prodi, F. and Belosi, F.: Atmospheric particles acting as Ice
821 Forming Nuclei in different size ranges, *Atmos. Res.*, 96, 266–272,
822 doi:10.1016/j.atmosres.2009.08.004, 2010.

823 Sarda-Estève, R., Huffman, J. A., Gallagher, M., Thibaudon, M., Baisnee, D., Baumier, R.,
824 McKeening, G., Kok, G., Sodeau, J., O'Connor, D., Crawford, I., Flynn, M., Saari, S., Poeschl,
825 U., Gros, V., Favez, O., Amodeo, T., Sciare, J., Bonnaire, N., Lassar, W., Pierce, K., Chou, C.,
826 Bertram, A., Salines, G., Roux, J.-M., Nadal, M. H., Bossuet, C. and Olmedo, L.: BIODETEECT
827 2014 campaign in Paris area: overview of the experimental strategy and preliminary results,
828 AAAR 33rd Annual Conference, Orlando, FL, October 20–24, 2014.

829 Saukko, E., Kuuluvainen, H. and Virtanen, A.: A method to resolve the phase state of aerosol
830 particles, *Atmos. Meas. Tech.*, 5, 259–265, doi:10.5194/amt-5-259-2012, 2012.

831 Schnell, R. C.: Ice Nuclei in Seawater, Fog Water and Marine Air off the Coast of Nova Scotia:
832 Summer 1975, *J. Atmos. Sci.*, 34, 1299–1305, 1977.

833 Schnell, R. C. and Vali, G.: World-wide source of leaf-derived freezing nuclei, *Nature*, 246,
834 212–213, doi:10.1038/246212a0, 1973.

835 Schwarz, J. P., Gao, R. S., Perring, A. E., Spackman, J. R. and Fahey, D. W.: Black carbon
836 aerosol size in snow, *Sci. Rep.*, 3, 1356, doi:10.1038/srep01356, 2013.

837 Schwarz, J. P., Gao, R. S., Spackman, J. R., Watts, L. A., Thomson, D. S., Fahey, D. W.,
838 Ryerson, T. B., Peischl, J., Holloway, J. S., Trainer, M., Frost, G. J., Baynard, T., Lack, D. A., de
839 Gouw, J. A., Warneke, C. and Del Negro, L. A.: Measurement of the mixing state, mass, and
840 optical size of individual black carbon particles in urban and biomass burning emissions,
841 *Geophys. Res. Lett.*, 35, L13810, doi:10.1029/2008GL033968, 2008.

842 Sesartic, A. and Dallafior, T. N.: Global fungal spore emissions, review and synthesis of
843 literature data, *Biogeosciences*, 8, 1181–1192, doi:10.5194/bg-8-1181-2011, 2011.

844 Sesartic, A., Lohmann, U. and Storelvmo, T.: Modelling the impact of fungal spore ice nuclei on
845 clouds and precipitation, *Environ. Res. Lett.*, 8, 014029, doi:10.1088/1748-9326/8/1/014029,
846 2013.

847 Spracklen, D. V and Heald, C. L.: The contribution of fungal spores and bacteria to regional and
848 global aerosol number and ice nucleation immersion freezing rates, *Atmos. Chem. Phys.*, 14,
849 9051–9059, doi:10.5194/acp-14-9051-2014, 2014.

850 Stein, S. W., Turpin, B. J., Cai, X., Huang, P.-F. and McMurray, P. H.: Measurements of relative
851 humidity-dependent bounce and density for atmospheric particles using the DMA-impactor
852 technique, *Atmos. Environ.*, 28, 1739–1746, doi:10.1016/1352-2310(94)90136-8, 1994.

853 Szyrmer, W. and Zawadzki, I.: Biogenic and anthropogenic sources of ice-forming nuclei: A
854 review, *B. Am. Meteorol. Soc.*, 78, 209–228, 1997.

855 Tobo, Y., Prenni, A. J., DeMott, P. J., Huffman, J. A., McCluskey, C. S., Tian, G., Pöhlker, C.,
856 Pöschl, U. and Kreidenweis, S. M.: Biological aerosol particles as a key determinant of ice nuclei
857 populations in a forest ecosystem, *J. Geophys. Res.*, 118, 10100–10110, doi:10.1002/jgrd.50801,
858 2013.

859 Tong, H.-J., Ouyang, B., Nikolovski, N., Lienhard, D. M., Pope, F. D. and Kalberer, M.: A new
860 electrodynamic balance (EDB) design for low-temperature studies: application to immersion
861 freezing of pollen extract bioaerosols, *Atmos. Meas. Tech.*, 8, 1183–1195, doi:10.5194/amt-8-
862 1183-2015, 2015.

863 Vali, G.: Sizes of Atmospheric Ice Nuclei, *Nature*, 212, 384–385, doi:10.1038/212384a0, 1966.

864 Vali, G.: Quantitative Evaluation of Experimental Results on the Heterogeneous Freezing
865 Nucleation of Supercooled Liquids, *J. Atmos. Sci.*, 28, 402–409, 1971.

866 Vali, G.: Interpretation of freezing nucleation experiments: singular and stochastic; sites and
867 surfaces, *Atmos. Chem. Phys.*, 14, 5271–5294, doi:10.5194/acp-14-5271-2014, 2014.

868 Vali, G., Christensen, M., Fresh, R. W., Galyan, E. L., Maki, L. R. and Schnell, R. C.: Biogenic
869 Ice Nuclei. Part II: Bacterial Sources, *J. Atmos. Sci.*, 33, 1565–1570, 1976.

870 Vali, G., DeMott, P. J., Möhler, O. and Whale, T. F.: Technical Note: A proposal for ice
871 nucleation terminology, *Atmos. Chem. Phys.*, 15, 10263–10270, doi:10.5194/acp-15-10263-
872 2015, 2015.

873 Vasiliou, J. G., Sorensen, D. and McMurry, P. H.: Sampling at controlled relative humidity with
874 a cascade impactor, *Atmos. Environ.*, 33, 1049–1056, doi:10.1016/S1352-2310(98)00323-9,
875 1999.

876 Wang, B., Laskin, A., Roedel, T., Gilles, M. K., Moffet, R. C., Tivanski, A. V and Knopf, D. A.:
877 Heterogeneous ice nucleation and water uptake by field-collected atmospheric particles below
878 273 K, *J. Geophys. Res.*, 117, D00V19, doi:10.1029/2012JD017446, 2012.

879 Welti, A., Lüönd, F., Kanji, Z. A., Stetzer, O. and Lohmann, U.: Time dependence of immersion
880 freezing: an experimental study on size selected kaolinite particles, *Atmos. Chem. Phys.*, 12,
881 9893–9907, doi:10.5194/acp-12-9893-2012, 2012.

882 Welti, A., Lüönd, F., Stetzer, O. and Lohmann, U.: Influence of particle size on the ice
883 nucleating ability of mineral dusts, *Atmos. Chem. Phys.*, 9, 6705–6715, doi:10.5194/acp-9-6705-
884 2009, 2009.

885 Wheeler, M. J., Mason, R. H., Steunenberg, K., Wagstaff, M., Chou, C. and Bertram, A. K.:
886 Immersion Freezing of Supermicron Mineral Dust Particles: Freezing Results, Testing Different
887 Schemes for Describing Ice Nucleation, and Ice Nucleation Active Site Densities, *J. Phys. Chem.*
888 *A*, 119, 4358–4372, doi:10.1021/jp507875q, 2015.

889 Wilkinson, D. M., Koumoutsaris, S., Mitchell, E. A. D. and Bey, I.: Modelling the effect of size
890 on the aerial dispersal of microorganisms, *J. Biogeogr.*, 39, 89–97, doi:10.1111/j.1365-
891 2699.2011.02569.x, 2011.

892 Wilson, T. W., Ladino, L. A., Alpert, P. A., Breckels, M. N., Brooks, I. M., Browse, J., Burrows,
893 S. M., Carslaw, K. S., Huffman, J. A., Judd, C., Kilhau, W. P., Mason, R. H., McFiggans, G.,
894 Miller, L. A., Najera, J., Polishchuk, E., Rae, S., Schiller, C. L., Si, M., Vergara Temprado, J.,
895 Whale, T. F., Wong, J. P. S., Wurl, O., Yakobi-Hancock, J. D., Abbatt, J. P. D., Aller, J. Y.,
896 Bertram, A. K., Knopf, D. A. and Murray, B. J.: A marine biogenic source of atmospheric ice
897 nucleating particles, submitted, 2015.

898 Winkler, P.: Relative humidity and the adhesion of atmospheric particles to the plates of
899 impactors, *J. Aerosol Sci.*, 5, 235–240, doi:10.1016/0021-8502(74)90058-5, 1974.

900 Wright, T. P., Hader, J. D., McMeeking, G. R. and Petters, M. D.: High Relative Humidity as a
901 Trigger for Widespread Release of Ice Nuclei, *Aerosol Sci. Technol.*, 48, i–v,
902 doi:10.1080/02786826.2014.968244, 2014.

903 Wright, T. P. and Petters, M. D.: The role of time in heterogeneous freezing nucleation, *J.*
904 *Geophys. Res.-Atmos.*, 118, 3731–3743, doi:10.1002/jgrd.50365, 2013.

905 Wright, T. P., Petters, M. D., Hader, J. D., Morton, T. and Holder, A. L.: Minimal cooling rate
906 dependence of ice nuclei activity in the immersion mode, *J. Geophys. Res.-Atmos.*, 118, 10535–
907 10543, doi:10.1002/jgrd.50810, 2013.

908 Yakobi-Hancock, J. D., Ladino, L. A. and Abbattt, J. P. D.: Review of recent developments and
909 shortcomings in the characterization of potential atmospheric ice nuclei: focus on the tropics,
910 *Rev. Ciencias*, 17, 15–34, 2013.

911 Yakobi-Hancock, J. D., Ladino, L. A., Bertram, A. K., Huffman, J. A., Jones, K., Leaitch, W. R.,
912 Mason, R. H., Schiller, C. L., Toom-Sauntry, D., Wong, J. P. S. and Abbatt, J. P. D.: CCN
913 activity of size-selected aerosol at a Pacific coastal location, *Atmos. Chem. Phys.*, 14, 12307–
914 12317, doi:10.5194/acp-14-12307-2014, 2014.

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932 **Table 1.** The seven locations used in this study and conditions during sampling.

Location	Environment	Coordinates	Elevation (m)	Sampling period	Number of Samples	Average sampling time (h)	Average temperature (°C)	Average relative humidity (%)	Particle size range (µm) ^a
Alert, NU, Canada	Arctic	82.45° N 62.51° W	12 agl 200 asl	Mar. 29 - Jul. 23, 2014	9	17.6	-17.4	77	0.10–10
Whistler Mountain, BC, Canada	Alpine	50.06° N 122.96° W	2 agl 2182 asl	Mar. 30 - Apr. 23, 2014	4	6.7	0.8	83	0.18–10
Amphitrite Point, BC, Canada	Coastal	48.92° N 125.54° W	5.5 agl 25 asl	Aug. 6–27, 2013	34	7.8	13.8	97	0.18–10
The Labrador Sea, Canada	Marine	54.50° N 55.37° W	2 agl 15 asl	Jul. 11, 2014	1	6.2	10.9	75	0.10–10
CEA, Saclay, France	Suburban	48.70° N 2.14° E	10 agl 168 asl	Jul. 15 - Aug. 4, 2014	15	7.2	20.6	69	0.10–10
UBC campus, BC, Canada	Suburban	49.26° N 123.25° W	24 agl 120 asl	May 12–16, 2014	4	6.3	15.2	70	0.10–10
Colby, KS, USA	Agricultural	39.39° N 101.06° W and 39.39° N 101.08° W	2 agl 968 asl	Oct. 14–15, 2014	3	4.5	17.0	48	0.10–10

933 ^aAerodynamic diameter based on the 50 % cutoff of the MOUDI (Marple et al., 1991).

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943 **Table 2.** Previous size-resolved INP measurements.

Study	Location	Geographical Description	Altitude	Particle sizes investigated (μm)	Mode of ice nucleation	Ice nucleation temperature ($^{\circ}\text{C}$) ^a	Result
Rucklidge (1965)	West Plains, Missouri	Forest and pasture	4 m agl	TSP	Condensation and/or deposition	-12 to -25	$\leq 14\%$ of INPs $> 1\ \mu\text{m}$
Vali (1966)	Alberta, Canada	Western Canada	Hail melt water	TSP ^b	Immersion	-12.8	16 % of INPs $> 1.2\ \mu\text{m}^{\text{c}}$
Rosinski et al. (1986)	Central and western South Pacific Ocean	Marine	Near sea level	0.5 to > 8	Immersion	-10.8	100 % of INPs $> 1\ \mu\text{m}$
				< 0.5 to > 8	Condensation	-5 to -6	1 % of INPs $> 1\ \mu\text{m}$
Rosinski et al. (1988)	Gulf of Mexico	Marine	Near sea level	0.1 to > 4.5	Condensation	-15 to -16	45 % of INPs $> 1\ \mu\text{m}$
Berezinski et al. (1988)	European territory of the former Soviet Union	Eastern Europe	100–500 m agl	0.1 to > 100	Condensation	-15 to -20	37 % of INPs $> 1\ \mu\text{m}$
Mertes et al. (2007)	Jungfrauoch, Switzerland	Alpine	3580 m asl	0.02 to 5	Unknown (ice residual)	-17.4	$< 1\%$ of ice residuals $> 1\ \mu\text{m}^{\text{c}}$
Santachiara et al. (2010)	S. Pietro Capofiume, Italy	Rural Italy	3 m agl	TSP	Condensation	-17 to -19	47 and 30 % of INPs > 1 and $2.5\ \mu\text{m}$, respectively
		Forest during/after rainfall	4 m agl	0.32 to > 18	Immersion and deposition	-15 to -20	89 % of INPs $> 1\ \mu\text{m}$
Huffman et al. (2013)	Manitou Experimental Forest, CO, USA	Forest during dry periods	4 m agl	0.32 to > 18	Immersion and deposition	-15 to -20	46 % of INPs $> 1\ \mu\text{m}$

944 ^aWe used data at the temperatures of this study (-15, -20, and -25 $^{\circ}\text{C}$) when available, otherwise the next closest
 945 temperature was used.

946 ^bTSP = total suspended particulate.

947 ^cParticle size reported for Vali (1966) is based on filter pore size, and particle size reported in Mertes et al. (2007) is
 948 based on electrical mobility and optical measurements. In all other studies, particle size is given as the aerodynamic
 949 diameter.



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951 **Figure 1.** Sampling locations used in this study: (1) Alert, Nunavut, Canada; (2) Whistler
952 Mountain, British Columbia, Canada; (3) Amphitrite Point, British Columbia, Canada; (4) the
953 Labrador Sea, Canada; (5) CEA in Saclay, France; (6) the University of British Columbia
954 campus, British Columbia, Canada; and (7) Colby, Kansas, USA. Site coordinates are given in
955 Table 1 with details in Sect. 2.1. The image was modified from Bing Maps, 2014
956 (<http://bing.com/maps>).

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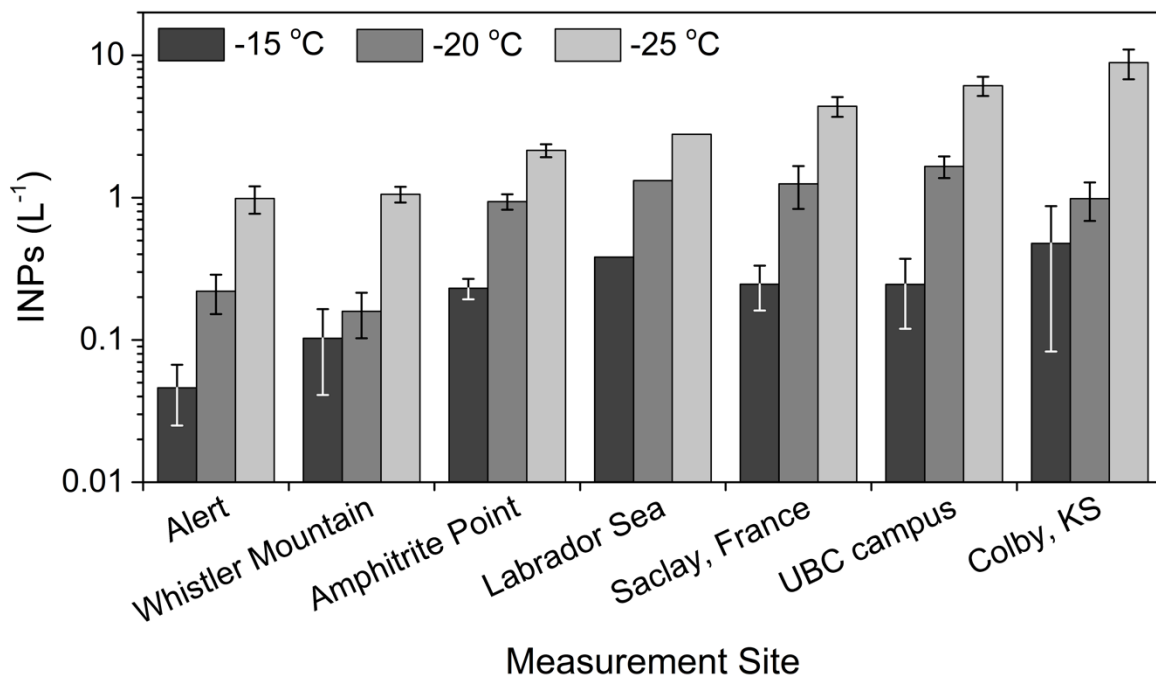
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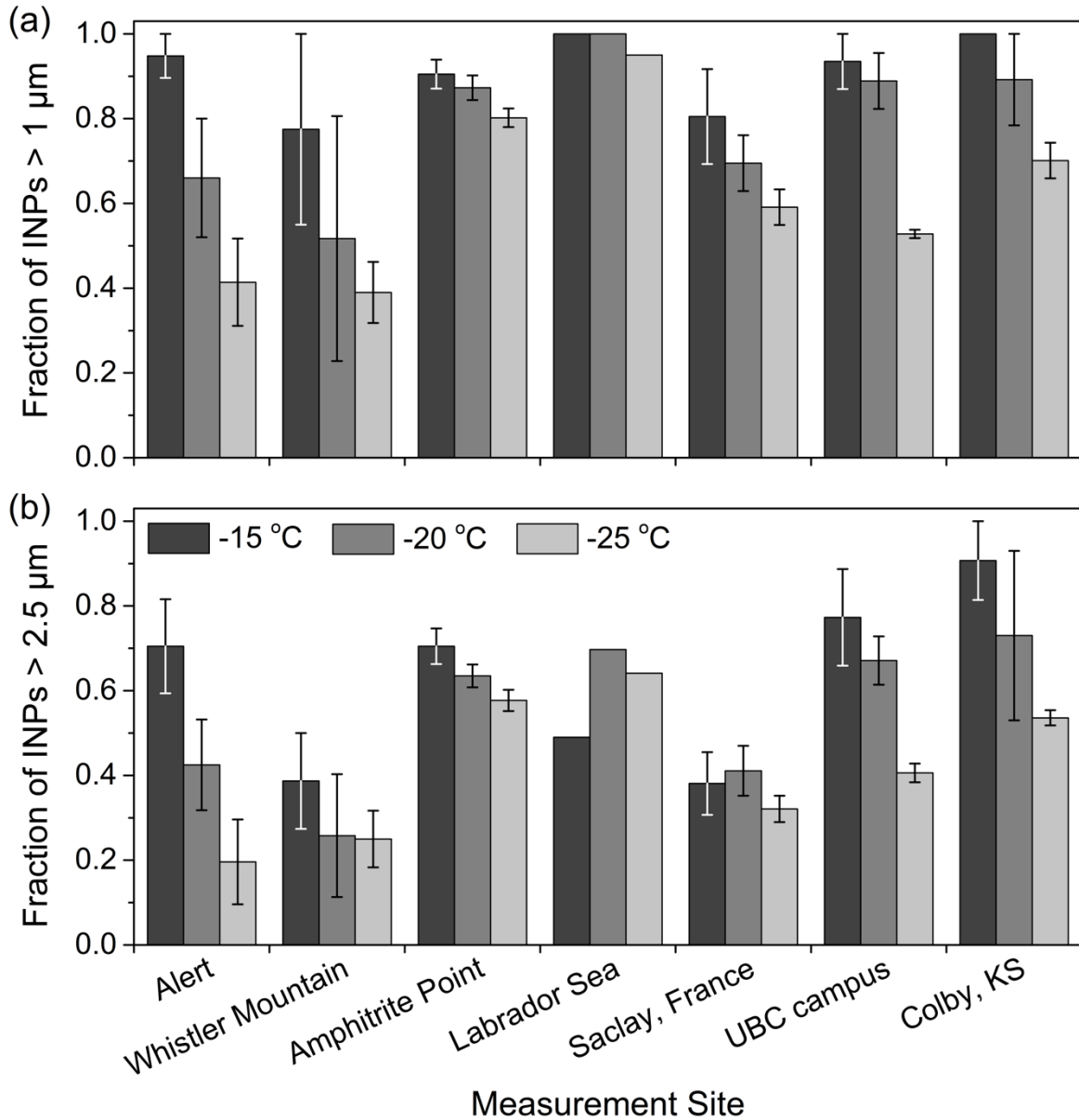
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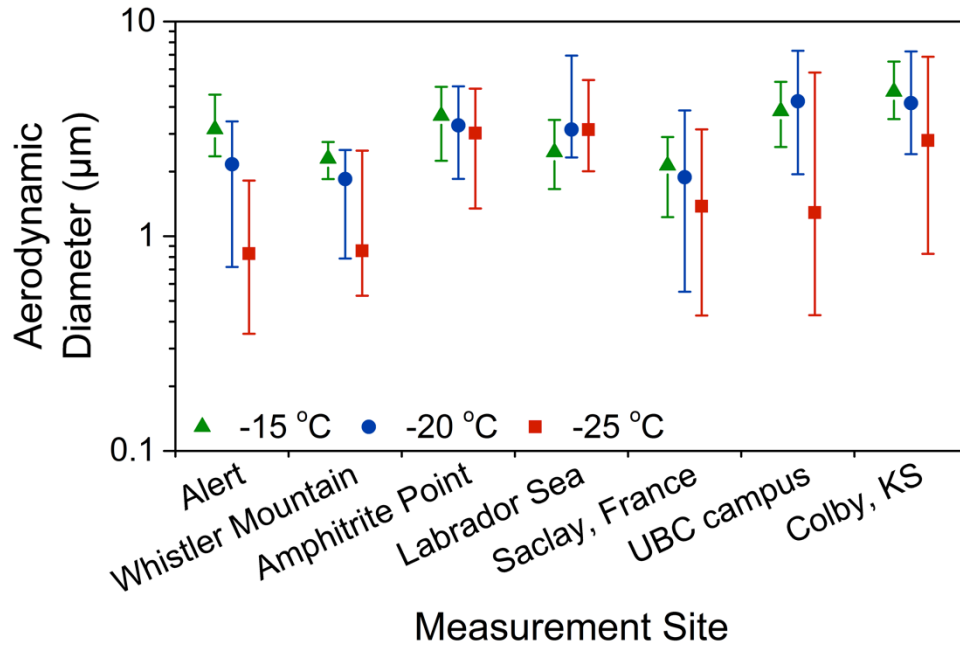


965 **Figure 2.** Mean INP number concentrations at droplet freezing temperatures of -15 °C (dark
 966 grey), -20 °C (intermediate grey), and -25 °C (light grey). Uncertainty is given as the standard
 967 error of the mean, assuming a normal distribution. As only one sample was available from the
 968 Labrador Sea, no uncertainty is reported.

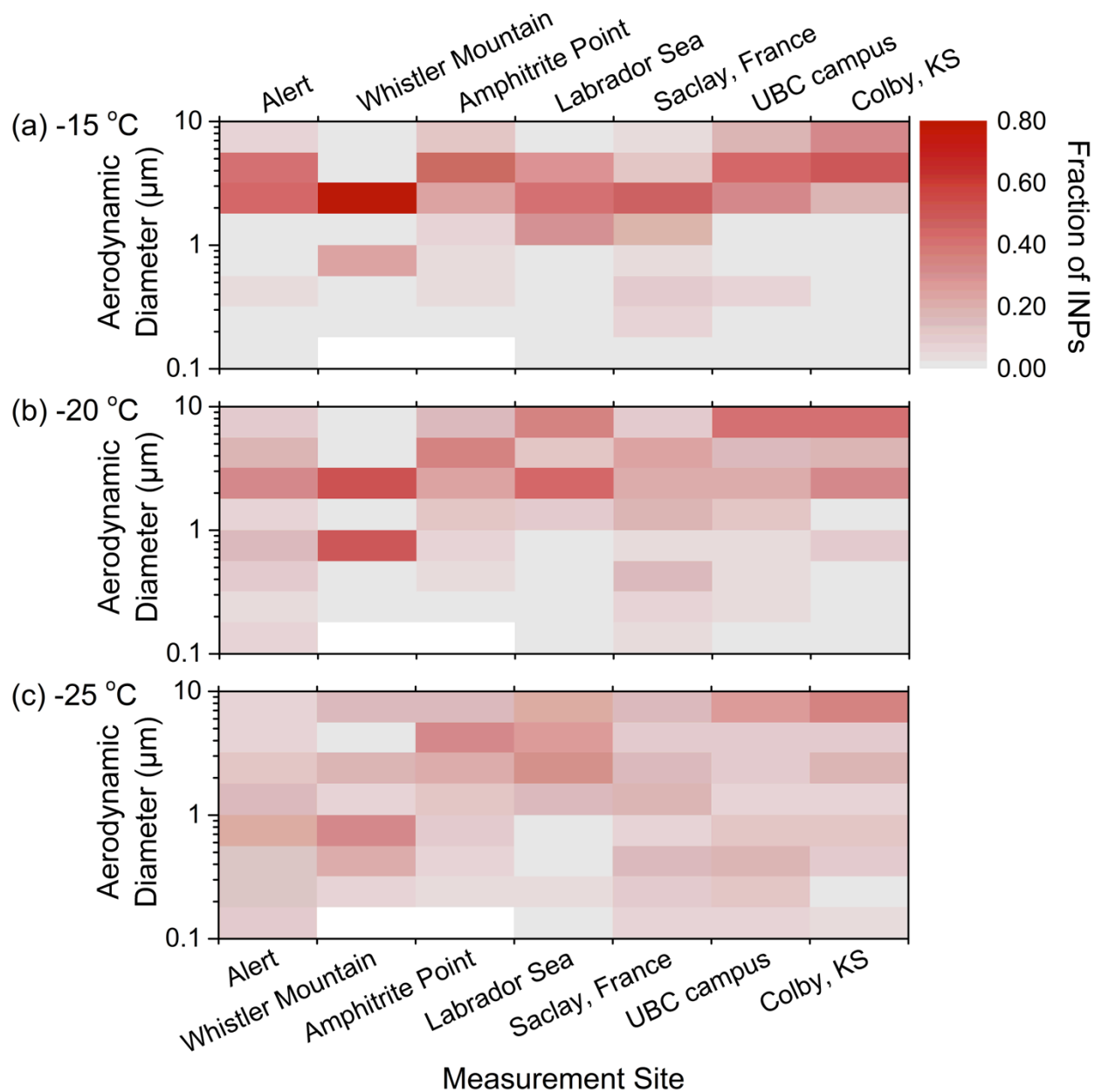


969 **Figure 3.** The mean fraction of INPs larger than (a) 1 μm and (b) 2.5 μm . Uncertainty is the
 970 standard error of the mean, assuming a normal distribution. Shading in the histogram
 971 corresponds to INP activation temperature: -15 $^{\circ}\text{C}$ is dark grey, -20 $^{\circ}\text{C}$ is an intermediate grey,
 972 and -25 $^{\circ}\text{C}$ is light grey. As 2.5 μm does not align with the size cut of a MOUDI stage, the
 973 fraction of INPs larger than 2.5 μm was found by assuming that number concentration of INPs
 974 1.8–3.2 μm in size was uniformly distributed over that size range. As only one sample was
 975 available from the Labrador Sea, no uncertainty is reported.

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980 **Figure 4.** The median size of INPs at ice-activation temperatures of -15 °C (green), -20 °C
 981 (blue), and -25 °C (red) when averaged over all analyzed samples. Upper and lower uncertainties
 982 are the 75th and 25th percentiles, respectively.



983 **Figure 5.** Fractional INP concentrations as a function of aerosol particle size, location, and
 984 activation temperature: (a) -15 °C; (b) -20 °C; and (c) -25 °C. The color bar indicates the fraction
 985 of INPs measured in each particle size bin. Aerosol particle sizes correspond to the 50 % cutoff
 986 aerodynamic diameters of the MOUDI stages (Marple et al., 1991). Missing sizes for the
 987 Whistler Mountain and Amphitrite Point sites are uncolored.