

1 Observations and Analysis of Organic 2 Aerosol Evolution in Some Prescribed Fire 3 Smoke Plumes

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17

18 Abstract

19 Open biomass burning is a significant source of primary air pollutants such as particulate matter
20 (PM) and non-methane organic gases (NMOG). However, the physical and chemical
21 atmospheric processing of these emissions during transport is poorly understood. Atmospheric
22 transformations of biomass burning emissions have been investigated in environmental
23 chambers, but there have been limited opportunities to investigate these transformations in the
24 atmosphere. In this study, we deployed a suite of real-time instrumentation on a Twin Otter
25 aircraft to sample smoke from prescribed fires in South Carolina, conducting measurements at
26 both the source and downwind to characterize smoke evolution with atmospheric aging. Organic
27 aerosol (OA) within the smoke plumes was quantified using an Aerosol Mass Spectrometer
28 (AMS), along with refractory black carbon (rBC) using a Single Particle Soot Photometer and
29 carbon monoxide (CO) and carbon dioxide (CO₂) using a Cavity Ring-Down Spectrometer.
30 During the two fires for which we were able to obtain aerosol aging data, normalized excess
31 mixing ratios and “export factors” of conserved species (rBC, CO, CO₂) suggested that changes
32 in emissions at the source did not account for most of the differences observed in samples of
33 increasing age. Investigation of AMS mass fragments indicated that the in-plume fractional
34 contribution ($f_{m/z}$) to OA of the primary fragment (m/z 60) decreased downwind, while the
35 fractional contribution of the secondary fragment (m/z 44) increased. Increases in f_{44} are typically
36 interpreted as indicating chemical aging of OA. Likewise, we observed an increase in the O:C
37 elemental ratio downwind, which is usually associated with aerosol aging. However, the rapid
38 mixing of these plumes into the background air suggests that these chemical transformations may
39 be attributable to the different volatilities of the compounds that fragment to these m/z in the
40 AMS. The gas-particle partitioning behavior of the bulk OA observed during the study was

41 consistent with the predictions from a parameterization developed for open biomass burning
42 emissions in the laboratory. Furthermore, we observed no statistically-significant increase in
43 total organic mass with atmospheric transport. Hence, our results suggest that dilution-driven
44 evaporation likely dominated over chemical production of SOA within our smoke plumes, likely
45 due to the fast dilution and limited aging times ($< \sim 5$ hr) that we could sample.

46 1. Introduction

47 Open biomass burning is estimated to be the largest contributor on a global scale to
48 atmospheric fine carbonaceous particulate matter (PM) (Bond et al., 2013) and the second largest
49 contributor to atmospheric non-methane organic gases (NMOG) (Akagi et al., 2011). Substantial
50 research has been focused on characterizing gas- and particle-phase primary emissions from
51 biomass burning and the development of emission inventories (Akagi et al., 2011; Burling et al.,
52 2010, 2011; Christian et al., 2003; Hosseini et al., 2013; May et al., 2014; McMeeking et al.,
53 2009; Reid et al., 2005; Urbanski, 2013; Urbanski et al., 2011; Watson et al., 2011; van der Werf
54 et al., 2010; Wiedinmyer et al., 2006, 2011; Yokelson et al., 2013a). These emissions are
55 ultimately integrated into chemical transport models used to predict regional air quality and
56 global climate impacts.

57 Organic aerosol (OA) species represent the major component by mass in the submicron
58 carbonaceous PM emitted from fires (May et al., 2014; McMeeking et al., 2009; Reid et al.,
59 2005). In general, the physical and chemical evolution of biomass-burning-derived OA in the
60 atmosphere after emission is poorly understood, in part because OA is a “metastable
61 intermediate” (Donahue et al., 2013). Since OA consists of thousands of species with a spectrum
62 of temperature-dependent saturation vapor pressures, the portion of OA that is observable as PM
63 varies with dilution and with atmospheric temperature. Further, many of the species comprising

64 OA have been shown to undergo oxidation reactions, forming secondary products with their own
65 range of volatilities. In field studies, OA in biomass burning plumes has been observed to be
66 enhanced, be depleted, or remain constant with time after emission (Akagi et al., 2012; Capes et
67 al., 2008; Cubison et al., 2011; DeCarlo et al., 2008; Jolleys et al., 2012; Vakkari et al., 2014;
68 Yokelson et al., 2009), but due to the complexities described above, attribution of these
69 transformations to specific physical and chemical processes is difficult (Heilman et al., 2014).

70 Laboratory studies have been conducted to attempt to separate these processes for
71 biomass-burning-derived OA. As part of the third Fire Lab at Missoula Experiment (FLAME-
72 III), May et al. (2013) derived a volatility distribution and related thermodynamic parameters
73 representative of the primary emissions from all of the biomass fuels studied. In that same study,
74 Hennigan et al. (2011) and Ortega et al. (2013) investigated chemical transformations of the
75 emissions using an environmental chamber and a potential aerosol mass chamber, respectively.
76 Results from both chambers demonstrated that the OA mass can be enhanced, depleted, or
77 remain roughly constant with oxidation, similar to field measurements, yet the OA always
78 became apparently more oxidized with photochemical aging, as interpreted from the organic
79 mass fragments measured via online aerosol mass spectrometry.

80 In this work, we report and interpret observations from the South Carolina fiRe
81 Emissions And Measurements (SCREAM) campaign conducted in October-November 2011
82 (Akagi et al., 2013, 2014; May et al., 2014; Sullivan et al., 2014). The objectives of SCREAM
83 were to: (1) simulate moderately intense wildfires by conducting prescribed burns at sites with
84 high fuel loadings, (2) characterize the emissions and develop estimates of emission factors and
85 emission ratios from both ground- and aircraft-based sampling, and (3) sample plumes
86 downwind as they evolved during atmospheric transport. We also sampled fires of opportunity

87 during the study. The SCREAM campaign combined simultaneous aircraft-based online
88 measurements of refractory black carbon (rBC), time-resolved non-refractory sub-micron PM
89 measurements (including OA), and time-resolved water-soluble organic carbon (WSOC) and
90 levoglucosan (LEV) measurements in addition to a suite of gas-phase compounds. Companion
91 papers have reported airborne trace gas emissions (Akagi et al., 2013), ground-based trace gas
92 emissions (Akagi et al., 2014), airborne WSOC and smoke marker emissions (Sullivan et al.,
93 2014), and airborne primary PM emissions (May et al., 2014). This paper focuses on airborne
94 observations of the OA mass concentrations and composition near the source and
95 transformations to OA mass concentration and composition during the first hours of atmospheric
96 transport.

97 2. Methods

98 Emissions from five of the seven fires sampled during SCREAM are discussed in this paper.
99 Details including fuel type, area burned, meteorology and stand history were provided by Akagi
100 et al. (2013) and are summarized briefly here. Two of the burns were conducted on the Fort
101 Jackson (FJ) Army Base (located northeast of Columbia, SC) in Blocks 9b (FJ 9b; 34°0'15" N,
102 80°52'37" W; 1 November 2011) and 22b (FJ 22b; 34°5'4" N, 80°52'16" W; 2 November 2011).
103 These burns occurred in older stands that had not been treated for a number of years, and were
104 intended to simulate wildfires. Fuel inventories indicated that the vegetation comprised primarily
105 of mature longleaf pine (*Pinus palustris*) and loblolly pine (*Pinus taeda*) with some contributions
106 of turkey oak (*Quercus laevis* Walter) and farkleberry (*Vaccinium arboretum* Marsh.).
107 Complementary ground-based measurements of emissions from the FJ burns were reported by
108 Akagi et al. (2014). The three other sampled fires were designated Georgetown (33°12'9" N,

109 79°24'6" W; 7 November 2011), Francis Marion (33°12'55" N, 79°28'34" W; 8 November
110 2011), and Bamberg (33°14'5" N, 80°56'41" W; 10 November 2011), based on the location in
111 SC where the fire occurred. Georgetown and Francis Marion were located in coastal SC, likely
112 burning coastal grasses and longleaf pine understory, respectively, based on in-flight
113 observations. The Bamberg fire, located roughly 80 km due south of the Fort Jackson site in
114 inland SC, was likely comprised of multiple fuel types, including longleaf/loblolly pine
115 understory as well as marsh grasses, based on smoke marker ratio measurements reported in
116 Sullivan et al. (2014).

117 2.1. Sample Collection

118 Smoke plumes during SCREAM were sampled via airborne measurements onboard a United
119 States Forest Service DHC-6 Twin Otter aircraft. Sampling strategies and flight tracks are
120 described in prior literature from the SCREAM study (Akagi et al., 2013; May et al., 2014;
121 Sullivan et al., 2014). Fires were initiated in mid-morning, and the aircraft initially sampled the
122 emissions near the source. Following the source characterization period, the downwind plume
123 was sampled to investigate the effect of chemical and physical aging during atmospheric
124 transport. For consistency with May et al. (2014), we defined "near-source" samples as those
125 collected within 5 km of the fire (always less than 30 minutes of aging, but most of the smoke
126 had an age of < 10 minutes based on average ambient wind speed), while downwind samples
127 were those collected at distances greater than 5 km. During flights, there were also periods of
128 out-of-plume background sampling to establish time-dependent background concentrations of
129 the species that were quantified in the plume.

130 The flight path for the FJ 9b burn is provided in Figure 1 as one example. Figure 1a
131 provides total (i.e., not background-corrected) OA mass concentration (C_{OA}), which was
132 typically between 3-7 $\mu\text{g m}^{-3}$ (average = 4.6 $\mu\text{g m}^{-3}$) outside of the plume throughout the

133 sampling domain, with the exception of higher concentrations attributable to the smoke plume.
134 Correcting the data for the background OA results in Figure 1b; here, the plume transport is more
135 distinct.

136 The first 1-2 hours of flight time was typically spent sampling near the source at ~100-
137 600 m altitude. Following this characterization period, it was possible to sample smoke
138 downwind with 1-2 hours of atmospheric aging, so we then alternated downwind cross-plume
139 samples with occasional additional source sampling. A challenge was that emissions were
140 rapidly diluted and mixed within the boundary layer, and the plumes did not penetrate into the
141 free troposphere, so visual tracking of the plumes was challenging. In fact, the flight path was
142 guided via consultation with real-time instrument output, which enabled the identification of
143 plume center and extent as well as the marking of way points. Furthermore, the plume from the
144 FJ 22b fire entered restricted air space near Columbia, SC, so it was only possible to follow this
145 plume for a short distance from the point of emission.

146 Downwind, the Twin Otter typically flew at altitudes between 500 m and 1500 m, but not
147 with sufficient detail to develop vertical profiles. All data, regardless of sampling altitude (or
148 latitude/longitude), are categorized as “within the plume” or “outside of the plume” along with
149 the additional distinction of “estimated time since emission” (please refer to Section 2.3.1).

150 2.2. Instrumentation

151 The instrumentation installed on the Twin Otter used to characterize emissions included a
152 High-Resolution Time-of-Flight Aerosol Mass Spectrometer (HR-ToF-AMS; Aerodyne
153 Research, Inc.), a Single Particle Soot Photometer (SP2; Droplet Measurement Technologies,
154 Inc.), a Cavity Ring-Down Spectrometer (CRDS; Picarro G2401; Picarro, Inc.), an airborne
155 Fourier-transform infrared spectrometer (AFTIR), a Particle-into-Liquid Sampler/Total Organic
156 Carbon and fraction collector system (Sullivan et al., 2014) and an Aircraft Integrated

157 Meteorological Measuring System (AIMMS-20) probe (Aventech Research, Inc.). The AIMMS-
158 20 provided meteorological data such as three-dimensional wind vectors, three-dimensional
159 position of the aircraft (i.e., latitude, longitude, and altitude), ambient temperature, and ambient
160 relative humidity. All sampling was conducted from a low turbulence inlet (Wilson et al., 2004)
161 followed by a non-rotating Micro Orifice Uniform Deposit Impactor (MOUDI; Marple et al.,
162 1991). The MOUDI was operated such that it served as a PM₁ selector (i.e., having 50% particle
163 transmission efficiency for particulate matter of 1 μm aerodynamic diameter with a sharpness of
164 1.08 – particles less than roughly 900 nm will be transmitted with 100% efficiency). All data
165 were adjusted to the same timestamp via alignment of peaks (thus accounting for differences in
166 both instrument clocks and instrument response times), which we referenced to the
167 HR-ToF-AMS.

168 2.2.1. Aerosol Mass Spectrometer

169 The HR-ToF-AMS (hereafter AMS) characterizes non-refractory sub-micron aerosol by
170 focusing sampled particles through an aerodynamic lens, collecting particles on a thermal
171 vaporizer, ionizing the vaporized particles via electron impaction, and detecting ions (m/z) in the
172 high-resolution time-of-flight mass spectrometer (DeCarlo et al., 2006). Using the ToF-AMS
173 data analysis toolkit SQUIRREL/PIKA (Sueper et al., 2013), aerosol mass concentrations can be
174 reconstructed from the m/z signal; for this study, we fit HR peaks for $m/z \leq 200$. These
175 concentrations are dependent on instrument parameters (e.g., ionization efficiency and vaporizer
176 collection efficiency). Ionization efficiency calibrations were performed with 350 nm ammonium
177 nitrate particles throughout the campaign, with values ranging from 1.83×10^{-7} to 2.91×10^{-7} ions
178 molecule⁻¹. Composition-dependent collection efficiencies were calculated following the
179 algorithm of Middlebrook et al. (2012), which is now built into the SQUIRREL software, for
180 each AMS sample and ranged from roughly 0.5-0.9, with a campaign-average value of 0.53. We

181 report AMS-derived emissions data of nitrate, sulfate, ammonium, and chloride elsewhere (May
182 et al., 2014). As stated in May et al. (2014), our results are potentially biased by up to a factor of
183 two due to the inherent uncertainty in our estimation of CE.

184 The AMS was mounted into National Center for Atmospheric Research GV-type aircraft
185 racks with a pressure-controlled inlet to reduce fluctuations of the pressure within the
186 aerodynamic lens (Bahreini et al., 2008). During operation, data were exclusively collected using
187 the “V-mode” of the ion time-of-flight within the mass spectrometer; since no particle time-of-
188 flight data were collected, no size-resolved information is available. AMS data were typically
189 collected with a time resolution of 6 seconds (corresponding to a distance of roughly 250-300
190 m).

191 While we obtained simultaneous measurements of gas-phase CO₂, we utilized the
192 standard correction in the fragmentation table from Allan et al. (2004), rather than explicit
193 corrections for CO₂ to account for differences within and without the plume. The AMS samples
194 particles roughly 10⁷ times more efficiently than the gas-phase. We estimate that on average, our
195 plume OA concentrations are positively biased by 0.0044 ± 0.0019% (both near the source and
196 downwind), our background OA concentrations are positively biased by 0.025 ± 0.021%, and
197 our *m/z* 44 measurements are positively biased by 0.20 ± 0.11%, all based on co-located gas-
198 phase CO₂ measurements. Consequently, we deemed this correction unnecessary as this
199 interference represents < 0.5% of our reported values.

200 2.2.2. Single Particle Soot Photometer

201 The SP2 provides operationally-defined rBC mass concentrations via laser-induced
202 incandescence (Stephens et al., 2003; Schwarz et al., 2006). Absorbing material present in
203 particles is heated to its vaporization temperature and emits radiation, which is measured by
204 optical detectors. This approach removes uncertainties due to interferences of artifacts that have

205 been observed during filter-based approaches (Kirchstetter et al., 2004) and excludes the
206 influence of “brown” carbon that can bias optical absorption methods (Andreae and Gelencsér,
207 2006; Lack et al., 2012), although it has been shown the method responds to some metals. Signal
208 is related to rBC mass via calibration procedures; during SCREAM, calibrations were performed
209 using fullerene soot. Generally, rBC mass concentrations were recorded every 6 seconds, similar
210 to the AMS. Additional details related to the SP2 operation during this campaign can be found in
211 May et al. (2014).

212 2.2.3. Cavity Ring-Down Spectrometer

213 A Picarro G2401 provided 0.5 Hz measurements of CO₂, CO, CH₄, and H₂O, which are
214 the major gas-phase emissions from combustion sources. The CRDS was calibrated in-flight
215 using mixed standards of CO₂/CO/CH₄ and procedures similar to those described by Urbanski
216 (2013). These data were applied to calculate emission ratios and emission factors of particle-
217 phase species, as described below.

218 2.3. Data Processing

219 2.3.1 Plume Identification and Age

220 As discussed previously, visual plume identification was challenging. Positive downwind
221 plume intercepts were confirmed through simultaneous spikes in measured values of OA, BC,
222 and CO. These confirmed intercepts were corroborated by simultaneous increases in smoke
223 markers (Sullivan et al., 2014) and trace gases (Akagi et al., 2013). Intercept-averaged
224 concentrations for each transect were derived by integrating the excess area (above background)
225 under the data time series curves and dividing by the elapsed time in the window of integration.

226 Sample ages (times since emission) were estimated using the distance from the source
227 and mean wind speed. Distance from the source was computed using the haversine formula and
228 the spatial coordinates measured by the AIMMS-20. Mean wind speed was also measured using

229 the AIMMS-20. Akagi et al. (2013) estimated that this approach has an uncertainty of roughly
230 30%, largely due to uncertainties in the wind speed data. Due to the plume tracking challenges,
231 plume intercepts were rarely perfectly perpendicular to the plume and were often diagonal
232 transects. Thus, a given sample can be associated with a range of estimated ages. In subsequent
233 figures, we plot the average age of a plume intercept along with error bars representing the range
234 of ages; in these figures, we do not include the estimated uncertainty of 30% on this range.

235
236 In-plume data from all research flights were corrected for local background
237 concentrations via integration under the curves in data time series between out-of-plume
238 measurements. The resulting species concentrations are “excess” concentrations and denoted by
239 the delta symbol, i.e., ΔX is the excess concentration of species X. We show background values
240 of some parameters in some of the following figures; these background concentrations represent
241 the median background concentration for the duration of the given flight. Sample background-
242 corrected data are provided in Figure 1.

243 2.3.2. Excess Emission Ratios and Emission Factors

244 Normalized excess mixing ratios (NEMR) are often used to account for transient fire
245 behavior and the dilution and mixing of plumes with background air during transport (e.g.,
246 Hobbs et al., 2003) and are defined as:

$$247 \quad NEMR_x = \frac{\Delta X}{\Delta CO} \quad (1)$$

248 where ΔX is the excess concentration of species X, and ΔCO is the background-corrected value
249 of CO. Since both numerator and denominator are excess quantities, uncertainties in their values
250 increase as the plume dilutes and in-plume concentrations approach the background
251 concentrations. Here, we report plume-integrated $NEMR_x$ for each plume interception, so our

252 values differ from the “fire-integrated” values (based on consideration of all the plume
253 intercepts) reported in May et al. (May et al., 2014). $NEMR_X$ are reported here in units of $\mu\text{g m}^{-3}$
254 ppmv-CO^{-1} ; this value can be converted to g g-CO^{-1} by multiplying by a factor of 8.7×10^{-4}
255 $\text{ppmv-CO} (\mu\text{g-CO m}^{-3})^{-1}$. Strictly speaking, $NEMR_X$ is a misnomer for aerosol mass
256 concentrations, but we utilize this terminology for consistency with the vast body of prior
257 literature.

258 Time series of instantaneous $NEMR_X$ provide information on transient smoke behavior
259 (Jolleys et al., 2014). By associating instantaneous $NEMR_X$ with time since emission,
260 physicochemical transformations can be investigated, since $NEMR_X$ accounts for dilution and
261 thus should be constant with time in the absence of sources or sinks of the species X or changes
262 in the initial emissions. The net formation of secondary organic aerosol (SOA) in smoke plumes
263 can be inferred from an increase in $NEMR_{OA}$ with distance downwind (Yokelson et al., 2009).
264 On the other hand, since OA emitted from biomass burning sources is semi-volatile, net
265 evaporation of particle-phase mass as dilution proceeds would appear as a decrease in $NEMR_{OA}$
266 (Akagi et al., 2012).

267 Emission factors (EF) are widely used descriptors of fire emissions (Ward and Radke,
268 1993; Andreae and Merlet, 2001). Their calculation relates the mass of X emitted (M_X) to the
269 mass of dry fuel consumed (M_{fuel}). In cases where the mass of fuel consumed is unknown, a
270 carbon mass balance approach can be applied, which relates the change in the concentration of X
271 relative to the background (ΔX ; $\mu\text{g m}^{-3}$) to the excess carbon concentrations (i.e., background-
272 corrected concentrations that have been converted to mg-C m^{-3}) of CO_2 , CO, total organic gases
273 (TOG = CH_4 + NMOG), and carbonaceous PM (PM_c):

$$274 \quad EF_X = \frac{M_X}{M_{\text{fuel}}} = \frac{\Delta X}{\Delta \text{CO}_2 + \Delta \text{CO} + \Delta \text{TOG} + \Delta \text{PM}_c} f_c \quad (2)$$

275 In Equation 2, the term f_C is a conversion factor representing fuel carbon content. Since we lack
276 detailed fuel information, we assume that $f_C = 0.50$, roughly the average fuel carbon content of
277 southeastern (SE) US coastal plain biomass fuels reported in laboratory studies (Burling et al.,
278 2010; May et al., 2014; McMeeking et al., 2009). Since ΔTOG and $\Delta\text{PM}_C \ll (\Delta\text{CO}_2 + \Delta\text{CO})$, for
279 convenience we approximate EF_X neglecting both ΔTOG and ΔPM_C , which results in an over-
280 estimate in EF_X of ~3-5% (Yokelson et al., 2013b). Like NEMR_X , EF_X are based on excess
281 concentrations and account for dilution, but if an “emission factor” is computed with downwind
282 data, the value obtained reflects changes in the initial emission factor plus the effect of any
283 sources or sinks of the originally-emitted species X. Hereafter, we will refer to downwind
284 “emission factors” as “export factors”, also denoted as EF_X and calculated from Equation 2; the
285 main distinction is that an export factor describes X downwind from the source, and thus may be
286 subject to both changes in the emissions as the fire burns and atmospheric transformations. We
287 report EF_X as g or mg kg-dry-fuel⁻¹.

288 3. Results and Discussion

289 During the study, only two fires provided adequate downwind aerosol data allowing us to
290 investigate in-plume aerosol physicochemical transformations: the FJ 9b fire and the Francis
291 Marion fire. As mentioned earlier, the plume from the FJ 22b fire entered restricted airspace and
292 could not be pursued. Further, the Georgetown fire was a small fire whose plume rapidly mixed
293 with the background, so downwind ΔOA was small and uncertain; the Bamberg samples
294 represented two distinct fuel types as shown elsewhere (May et al., 2014; Sullivan et al., 2014),
295 making it difficult to distinguish transformations during transport from differences in the
296 sources.

297 In Figure 2, we present composition data versus estimated time since emission for the
298 NEMR or EF for four major components present in the biomass burning smoke sampled for the
299 FJ 9b fire: OA (NEMR_{OA}; Figure 2a), rBC (NEMR_{rBC}; Figure 2b), CO (EF_{CO}; Figure 2c), and
300 CO₂ (EF_{CO2}; Figure 2d). Data near the source are presented as box-and-whisker plots (25th-75th
301 and 10th-90th percentiles); these data were collected during roughly 2.5 hours of sampling during
302 which the modified combustion efficiency (MCE) (Ward and Radke, 1993) varied between 0.900
303 and 0.930, which explains some of the variability in the data. Data up to five hours downwind
304 were obtained and are shown as open symbols. For downwind samples, vertical errors bars
305 represent estimated measurement uncertainties while horizontal error bars represent the range of
306 estimated plume ages for non-perpendicular plume transects; horizontal error bars do not account
307 for the estimated 30% measurement uncertainty in wind speed. To assess whether differences
308 near the source and downwind are statistically significant, we conducted unpaired *t*-tests. When
309 the corresponding two-tailed *p* value ≤ 0.05 , we consider the results to be significantly different;
310 conversely, if the *p* value > 0.05 , we infer that there is no significant difference.

311 We expect rBC, CO, and CO₂ to be conserved with transport since they are stable in the
312 atmosphere on the timescales considered here. Indeed, unpaired *t*-tests for the data shown in
313 Figures 2b-d indicate that there was no significant difference in the average value of these
314 species at the source and downwind (two-tailed *p* values > 0.13). Differences in mean downwind
315 EFs are attributable to measurement uncertainties, including identification of the plume edges,
316 and variability in the combustion at the source. Fitting an exponential decay with distance from
317 the source of absolute mixing ratios of CO and CO₂, we infer an average mixing rate (the inverse
318 of the dilution timescale, or the time to decay by 1/e) of 1.6 hr⁻¹ during the FJ 9b experiment.

319 Since OA is reactive and semi-volatile, it is perhaps not surprising that the downwind
320 NEMR_{OA} over 2-5 hr of atmospheric aging is significantly lower than the NEMR_{OA} at the source
321 (Figure 2a; two-tailed p value = 0.015), suggesting a net loss of emitted OA via evaporation
322 and/or reaction. As demonstrated by Akagi et al. (2013), the smoke plume was photochemically
323 active, as evident through enhancements of ozone and formaldehyde relative to the source.

324 Figure 3 is identical to Figure 2 but represents the Francis Marion burn, the only other
325 case with downwind aerosol measurements adequate to assess aging (here, up to 1.5 hr after
326 emission). Akagi et al. (2013) inferred photochemical processing was occurring in the Francis
327 Marion plume, based on observed downwind enhancements of ozone and formaldehyde relative
328 to the source. However, unlike the FJ 9b fire, none of the computed downwind NEMR and EF
329 shown in Fig. 3 were significantly different from the source (all two-tailed p values > 0.32). The
330 background OA concentrations, which we assume contribute to gas-particle partitioning of
331 emitted OA by providing additional absorptive material, were roughly 50% greater during the
332 Francis Marion fire compared to the FJ 9b fire; furthermore, the mixing rate was 20% slower for
333 the Francis Marion plume (1.3 hr^{-1}), and the plume aging was observed over a much shorter time
334 period. These factors would slow the evaporation of emitted OA, and limit the time over which
335 chemical transformations could occur and be observed. Indeed, over the first 1.5 hr after
336 emission, the data for FJ 9b shown in Figure 2 also indicated no statistically-significant change
337 in NEMR_{OA}.

338 3.1. Chemical Transformations of Organic Aerosol

339 In this section, we investigate chemical transformations of the organic aerosol, fragment
340 evolution (Figure 4), and elemental ratio analyses (Figure 5) using two approaches to for both
341 fires with adequate downwind data: the FJ 9b and Francis Marion burns. Both Ng et al. (2010)

342 and Morgan et al. (2010) demonstrated that “fresh” OA in ambient samples can be distinguished
343 by organic fragment signatures in the mass spectra (e.g., $C_3H_7^+$ at m/z 43), while “aged” OA is
344 more highly oxidized and can be distinguished by a strong contribution of CO_2^+ (m/z 44). The
345 fractional contributions of each of these fragments to the total OA concentration (e.g., $f_{44} =$
346 C_{44}/C_{OA} , where C_{44} is the mass concentration of particulate CO_2^+ , which is likely due to
347 decarboxylation on the vaporizer surface rather than CO_2 molecules being present in the aerosol
348 sample) change with atmospheric aging: f_{43} is expected to decrease and f_{44} to increase.

349 However, neither Ng et al. (2010) nor Morgan et al. (2010) directly considered the
350 influence of biomass burning. Cubison et al. (2011) and Ortega et al. (2013) thus modified the
351 approach and compared f_{60} and f_{44} to infer photochemical aging of BBOA. Levoglucosan and
352 other anhydrosugars are pyrolysis products of cellulose, and thus are used as molecular markers
353 for biomass burning emissions (Simoneit et al., 1999; Sullivan et al., 2008); these compounds
354 contribute to AMS spectra at m/z 60 ($C_2H_4O_2^+$) (Alfarra et al., 2007; Lee et al., 2010). May et al.
355 (2012) and references therein demonstrated that levoglucosan is semi-volatile at ambient
356 conditions and thus m/z 60 could decrease due to evaporation during dilution, if this finding is
357 extrapolated to all contributing species. Furthermore, Hennigan et al. (2010) demonstrated that
358 levoglucosan is reactive and chemically decays similar to the hydrocarbon-like (m/z 43)
359 fragments. Thus, f_{60} may change due to both dilution-driven evaporation and photo-oxidation
360 processes.

361 In Figure 4, we present excess f_{60} (Δf_{60}) and excess f_{44} (Δf_{44}) for the FJ 9b and Francis
362 Marion fires. These excess fragment fractional contributions were computed from background-
363 corrected m/z 60 or m/z 44 mass concentrations by dividing that excess concentration by ΔOA .
364 Thus, as the plume dilutes and becomes less distinguishable from the background, Δf_{60} and Δf_{44}

365 should remain constant if neither preferentially evaporates, reacts, or accumulates within the
366 plume. For the FJ 9b fire, the source-downwind differences for both Δf_{60} (Figure 4a) and Δf_{44}
367 (Figure 4b) are statistically significant (two-tailed p value < 0.0001). For the Francis Marion fire,
368 Δf_{60} (Figure 4c) is significantly lower downwind than at the source (two-tailed p value < 0.0001),
369 while Δf_{44} (Figure 4d) is significantly higher downwind than at the source (two-tailed p value =
370 0.029). The result for Δf_{60} is consistent with Figure 2a; that is, the decrease in Δf_{60} downwind
371 during the FJ 9b fire reflects the decrease in NEMR_{OA} . An observed decrease in Δf_{60} with no
372 decrease in OA concentration during the Francis Marion fire may be related to chemical
373 reactions of compounds that fragment to m/z 60 or to differences in the volatility of these
374 compounds compared to the bulk OA. The mechanistic driver of all transformations will be
375 explored below.

376 The increase in Δf_{44} with plume age for both fires indicates a compositional change
377 toward increasing mass fractional contributions from molecules that fragment to CO_2^+ . If only
378 dilution (and hence, evaporation) was occurring in the plumes as they moved downwind, Δf_{44}
379 should be conserved, provided its parent's volatility is similar to that of the bulk of the emitted
380 OA. The observed increase in CO_2^+ in these photochemically-active environments may indicate
381 that production of SOA occurred within the plumes, although there were no statistically-
382 significant increases in the measured downwind NEMR_{OA} , as also found in some previous field
383 studies (e.g., Capes et al., 2008; Cubison et al., 2011). On the other hand, this increase could also
384 indicate that the species fragmenting to m/z 44 are relatively less volatile than the bulk OA that
385 evaporates during transport and dilution.

386 There is experimental evidence investigating chemically-resolved volatility that is
387 consistent with the evaporation of bulk OA resulting in a relative increase in m/z 44 and a

388 relative decrease in m/z 60. Huffman et al. (2009a) demonstrated for ambient samples in two
389 different megacities that, at a given temperature in a thermodenuder, m/z 60 evaporated to a
390 greater extent than the bulk OA, while m/z 44 evaporated to a lesser extent than the bulk OA.
391 While heating OA is technically not the same as diluting OA, the response of OA to increased
392 temperature is analogous to the response of OA to increased dilution. Furthermore, Collier and
393 Zhang (2013) demonstrated that f_{44} increased with decreasing C_{OA} for vehicle test data in the
394 absence of chemistry and hypothesized that this observation was attributable to preferential
395 evaporation of less-oxidized OA species. Thus, the observed changes during SCREAM in Δf_{44}
396 and Δf_{60} may be due, at least in part, to physical changes occurring as some of the emitted OA is
397 volatilized upon dilution with ambient air.

398 Another framework for tracking the chemical evolution of OA was suggested by Heald et
399 al. (2010), who proposed the use of elemental ratios (hydrogen to carbon, H:C, and oxygen to
400 carbon, O:C) to describe photochemical aging of OA. Similar to the fragment evolution, with
401 increasing OH exposure, H:C is expected to decrease (e.g., due to hydrogen abstraction
402 reactions) and O:C is expected to increase (e.g., due to oxygen addition to alkyl radicals). In
403 Figure 5 we present the evolution of the elemental ratio of H:C and O:C during atmospheric
404 transport of the biomass burning plumes from both fires; values of the ratios for the average
405 background, out-of-plume ratio for each fire are shown as dotted lines. We assume uncertainties
406 of 31% of O:C and 10% for H:C, based on Aiken et al. (2008).

407 For both fires, the average background H:C ratio was roughly 15% greater than the H:C
408 at the source; downwind H:C values were mostly within the source variability. As the plumes
409 were transported downwind and mixed with background OA, based on measured dilution rates
410 we expected H:C to have increased toward the background values on a 2- to 3-hour timescale if

411 it were a conserved tracer. The lack of a clear increase with time since emission in both
412 experiments suggests either that loss of both H and C occurred in the plume, or increases in C
413 occurred without corresponding addition of H that would maintain the H:C observed at the
414 source. Typically, H:C decreases with increasing oxidation (Heald et al., 2010).

415 For O:C, about half the downwind values were higher than could be explained by
416 measured variability at the source, and the background OA had O:C within (but at the lower end)
417 of the range at the source. Dilution with background air was thus expected to have had little
418 impact on O:C if O:C were a conserved tracer. Like m/z 44, O:C could have increased with time
419 if photochemical production and condensation of high O:C species or photochemical aging of
420 aerosol had occurred (Kroll et al., 2011).

421 However, the observed decreases in NEMR_{OA} (whether statistically significant or not)
422 suggests that changes in H:C and O:C may potentially be induced by a solely physical process
423 (i.e., if C were lost from the aerosol phase by preferential evaporation of species that had lower
424 O:C than the average observed at the source). In fact, Huffman et al. (2009b) demonstrated that
425 O:C increased and H:C decreased with increasing evaporation of bulk OA in biomass burning
426 emissions during thermodenuder experiments. Hence, evaporative transformations may be
427 difficult to differentiate from oxidative transformations.

428 3.2. Physical Transformations of Organic Aerosol

429 A net loss of OA due solely to dilution-driven evaporation may thus be consistent with
430 the observations in Figures 2-5. However, we note that we cannot definitively state that no
431 aerosol chemistry has occurred within the plumes as they age. In the following, we assume a
432 priori knowledge that dilution-driven evaporation dominates over chemical processing and
433 explore if the volatility distribution derived by May et al. (2013) for laboratory biomass burning

434 POA can reproduce our airborne observations. If it can, no oxidative chemistry is required to
435 explain the data, although it is possible that some occurs.

436 Simulations representing the process of dilution alone are presented in Figure 6, which
437 shows EF_{OA} data (representing the emission factors near the source and export factors
438 downwind) as a function of the total mass concentration of observed organic aerosol (i.e., not
439 background corrected), C_{OA} , for six flights. Model curves were calculated using the following
440 equation (Donahue et al., 2006; Robinson et al., 2010):

$$441 \quad EF_{OA} = EF_{tot} \sum_i f_i \left(1 + \frac{C_i^*}{C_{OA}} \right)^{-1} \quad (3)$$

442 where i represents arbitrarily-chosen surrogate compounds defined by their saturation
443 concentration (C_i^* ; related to saturation vapor pressure through the ideal gas law), and f_i is the
444 mass fraction of each species i relative to the total emitted organics. The set of f_i and C_i^* is
445 referred to as a volatility distribution. Here, we utilize the volatility distribution for emissions
446 from open biomass burning that was proposed by May et al. (2013), which is comprised of
447 surrogate compounds representing seven logarithmically-spaced C_i^* bins. C_{OA} represents the
448 total OA concentration (emissions + background).

449 EF_{tot} is the emission factor of total organics (gas + particle phase) that are constrained by
450 the volatility distribution (here, all material between $C_i^* = 3 \times 10^{-3} \mu\text{g m}^{-3}$ and $3 \times 10^4 \mu\text{g m}^{-3}$, so
451 this is not equivalent to NMOG), and hence, contribute to gas-particle partitioning; EF_{tot} is likely
452 dominated by biomass-burning-derived organics but may include background semi-volatile
453 organic material that can partition into the particle phase due to the presence of the biomass
454 burning smoke. Values of EF_{tot} were inferred using Equation 3 with measured C_{OA} , calculated
455 EF_{OA} (from Equation 2), and the volatility distribution from May et al. (2013) as inputs for each
456 plume intercept. In Figure 6, the lines represent predictions based on the average EF_{tot} inferred

457 for each fire, while the shaded areas represent \pm one standard deviation in EF_{tot} . Values of EF_{tot}
458 ranged from roughly 2 g kg-fuel⁻¹ (Bamberg B) to 12 g kg-fuel⁻¹ (FJ 22b); both the FJ 9b and
459 Francis Marion fires had inferred EF_{tot} of roughly 6 g kg-fuel⁻¹. Equation 3 implies that EF_{OA}
460 (regardless of whether this represents an emission factor or export factor) decreases with
461 increasing dilution, due to the physical repartitioning of semi-volatile species.

462 There are some key assumptions to our use of Equation 3. We assume that the gas-
463 particle partitioning of the OA can be described using a parameterization derived for laboratory
464 fires, even though the OA in our samples has originated from prescribed fires in the field and
465 may be enhanced by background semi-volatile organics. We are also assuming that EF_{tot} is
466 constant in time for a given prescribed fire (i.e., it does not vary due to source variability, mixing
467 with background air, or atmospheric chemistry). Finally, we are inherently assuming that the
468 plume temperature is constant at 298 K, so dilution is the only process affecting gas-particle
469 partitioning. While these assumptions are not strictly true, they should not affect our conclusions
470 significantly on average.

471 Figures 6a and 6b provide EF_{OA} calculated near the source and downwind for the FJ 9b
472 and Francis Marion fires, respectively. Near-source data from the FJ 22b and Georgetown fires
473 are presented in Figure 6c, and from the Bamberg fires in Figure 6d. Generally, the near-source
474 data for all fires follow the expected trend, exhibiting a decrease in EF_{OA} with decreasing total
475 measured (i.e., not background-corrected) C_{OA} , as would be expected for a semi-volatile tracer
476 with the characteristics summarized by May et al. (2013); variability in near-source data arises
477 due to proximity to the source and to the center of the plume as well as the smoke production
478 rate. Downwind data (only available for Figures 6a and 6b) also generally follow the trend
479 predicted by Equation 3; indeed, downwind OA concentrations appear to be lower than

480 predicted, suggesting evaporation of emitted OA dominates over production and condensation of
481 SOA if occurring. Performing a t-test on the inferred EF_{tot} for both the FJ 9b and Francis Marion
482 fires indicates that the differences between near-source and downwind values are not statistically
483 significant (p -value > 0.1), suggesting no observable SOA production from oxidation reactions
484 (e.g., excess OA has reached equilibrium).

485 We also note that the predictions in Figure 6 are based on a composite volatility
486 distribution that best represented biomass fuels investigated in the laboratory during the
487 FLAME-III study, which has been extrapolated to the field in this study. Also, fire behavior was
488 variable during the several hours over which data were collected, as evident in the MCE
489 variability (Akagi et al., 2013); the emissions of organics has been demonstrated to vary with
490 MCE (May et al., 2014; McMeeking et al., 2009). All data are represented using a single set of
491 model inputs, which does not account for this variability with MCE. While other factors likely
492 play a role, these two are likely to be the most important. Regardless, the differences in OA
493 observed at the source and downwind for these plumes can be explained by a simple model of
494 gas-particle partitioning.

495 4. Conclusions

496 In this work, we present field observations of the physicochemical evolution of the
497 organic aerosol present in biomass burning plumes from two prescribed fires in South Carolina.
498 Downwind observations of rBC to CO ratios and emission factors of CO, and CO₂ are not
499 statistically different on average from those at the source. The downwind ratio of OA to CO was
500 significantly lower than at the source for the fire that we were able to follow downwind for up to
501 five hours of atmospheric aging. The downwind OA to CO ratio was not significantly different

502 downwind for the other fire, which may be related to the much shorter observable atmospheric
503 aging time (~2 hr).

504 We observed significant differences in downwind ratios of AMS mass fragments thought
505 to be indicative of fresh biomass burning emissions (m/z 60, which decreased) and more oxidized
506 OA species (m/z 44, which increased), consistent with prior reported laboratory photo-oxidation
507 experiments. While the observed increases in Δf_{44} (and the O:C ratio) imply the possibility of
508 SOA production within the plume, these observed changes are also consistent with differences in
509 the volatilities of the species fragmenting to m/z 60 and m/z 44 relative to the bulk OA, resulting
510 in differences in evaporation as the plume dilutes into background air.

511 Our observations and model simulations suggest that dilution-driven evaporation out of
512 the particle phase dominated over condensation of semi-volatile material into the particle phase
513 over roughly the first two hours of transport during the FJ 9b fire. After this, the OA in the
514 plume reached an apparent steady-state with the background in our observations, as there is no
515 net change to $NEMR_{OA}$ (i.e., there is no obvious dilution-driven evaporation or SOA
516 production); thus, OA transformation can be predicted with a simple gas-particle partitioning
517 model. For the Francis Marion fire, due to limited downwind data, we cannot draw a similar
518 conclusion with any certainty. The decrease in $NEMR_{OA}$ for the FJ 9b fire is consistent with
519 results from previous literature (Akagi et al., 2012; Jolleys et al., 2012, 2015); however, other
520 studies report increases in OA with increasing plume age (DeCarlo et al., 2008; Vakkari et al.,
521 2014; Yokelson et al., 2009). The exact cause of this variability in observations is unclear. These
522 remaining unexplained differences among different field studies highlight the need for additional
523 research on atmospheric physicochemical transformations of biomass burning plumes.

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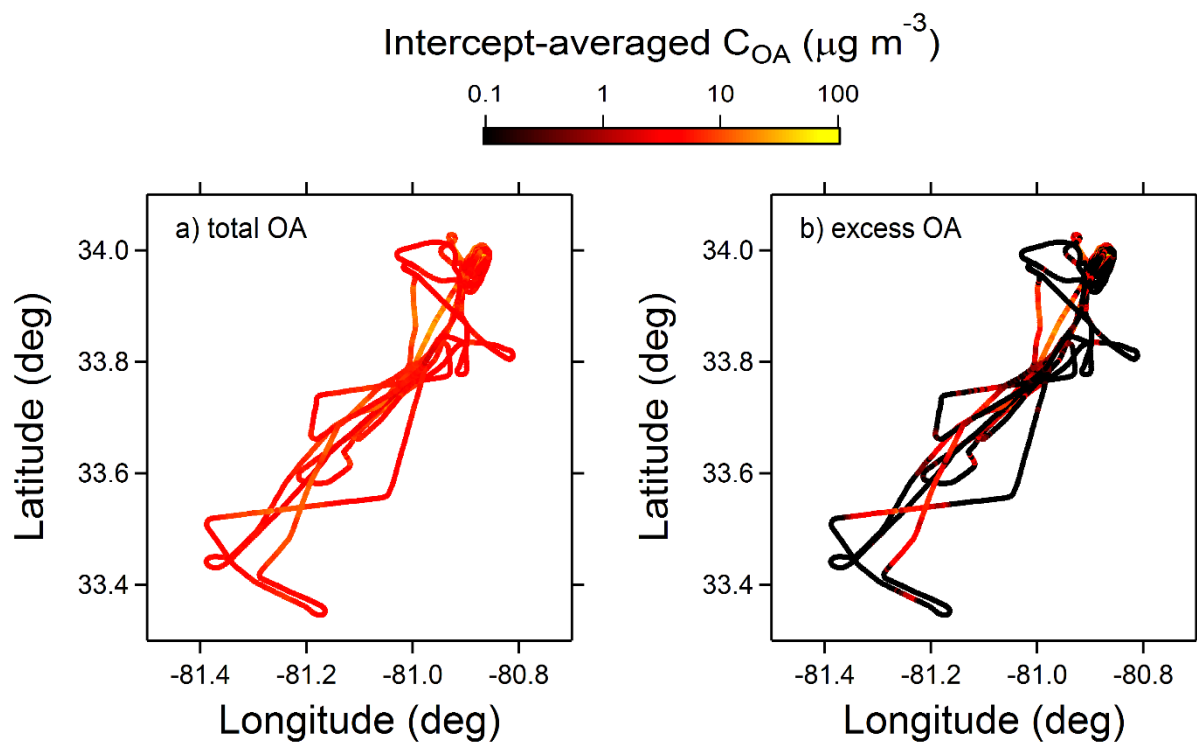
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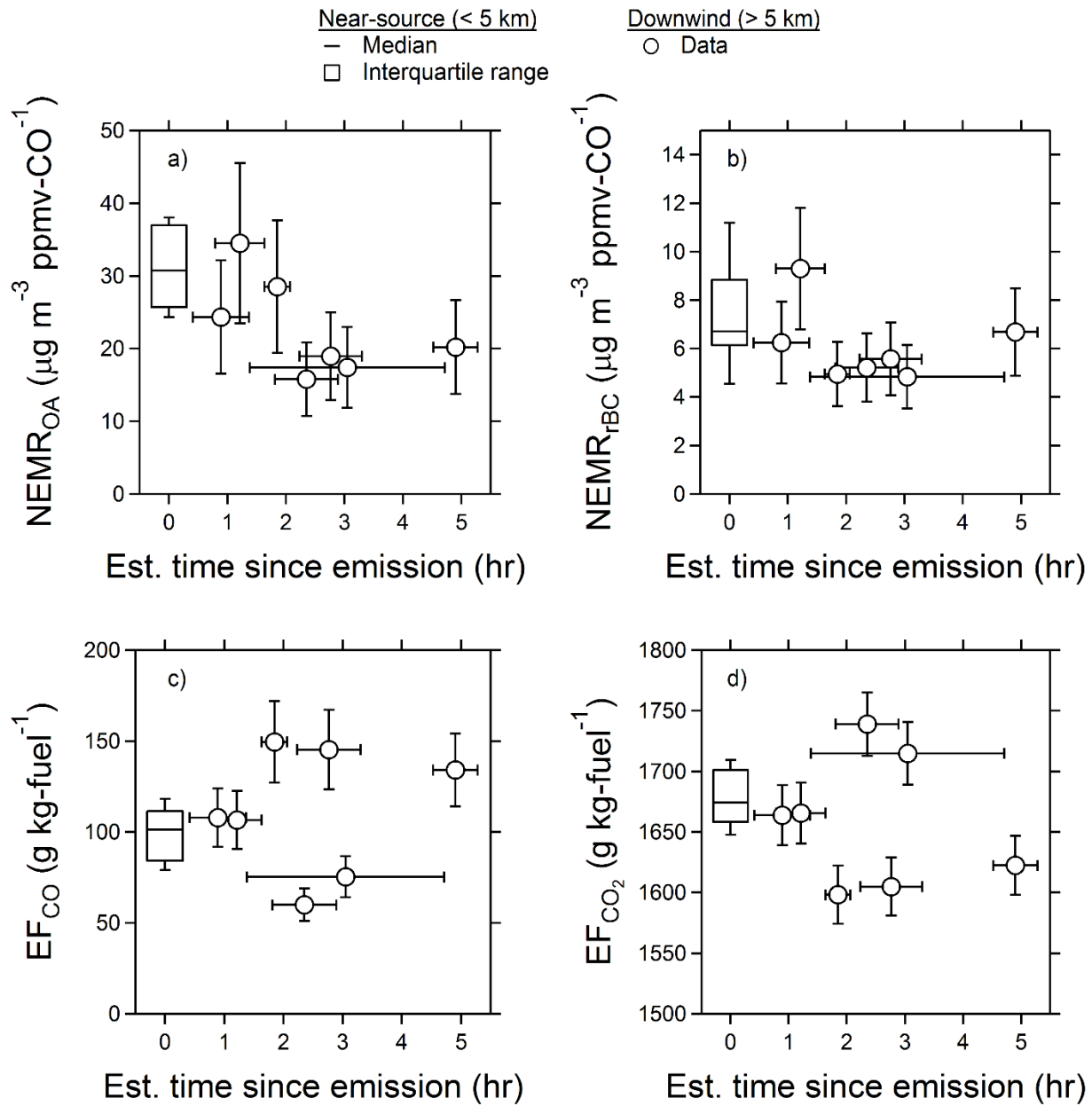
807 Figures



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809 Figure 1.

810 Flight tracks colored by **a)** total OA concentration and **b)** excess OA concentration. Due to the
811 log-scaling of intercept-averaged concentrations, the minimum value in panel b) is set to $0.1 \mu\text{g m}^{-3}$
812 m^{-3} . Removing the background OA elucidates distinct plume transport to the southwest.



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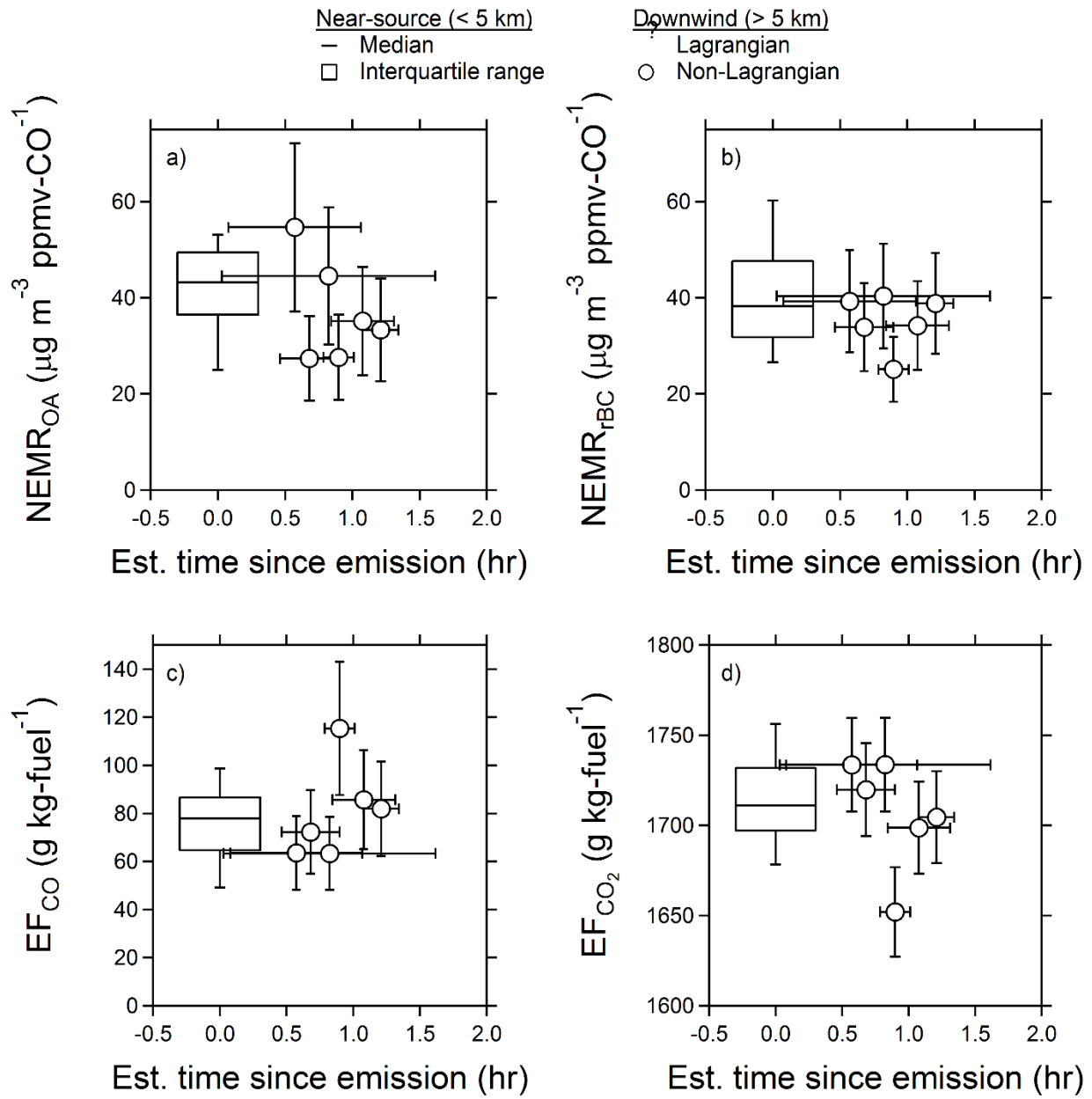
814 **Figure 2.**

815 Near-source and downwind data collected during the FJ 9b prescribed fire. **a)** the ratio of excess
 816 OA to CO; **b)** the ratio of excess rBC to CO; **c)** emission/export factor for CO; and **d)**
 817 emission/export factor of CO₂. Near-source data are represented by box-and-whisker plots
 818 (boxes: 25th and 75th percentiles; whiskers: 10th and 90th percentiles; horizontal lines: median)
 819 while downwind data are represented by markers. Error bars associated with the markers indicate
 820 range of estimated time since emission (x-direction) and measurement uncertainty (y-direction).
 821 Error bars in x direction do not account for estimated 30% accuracy of wind speed.

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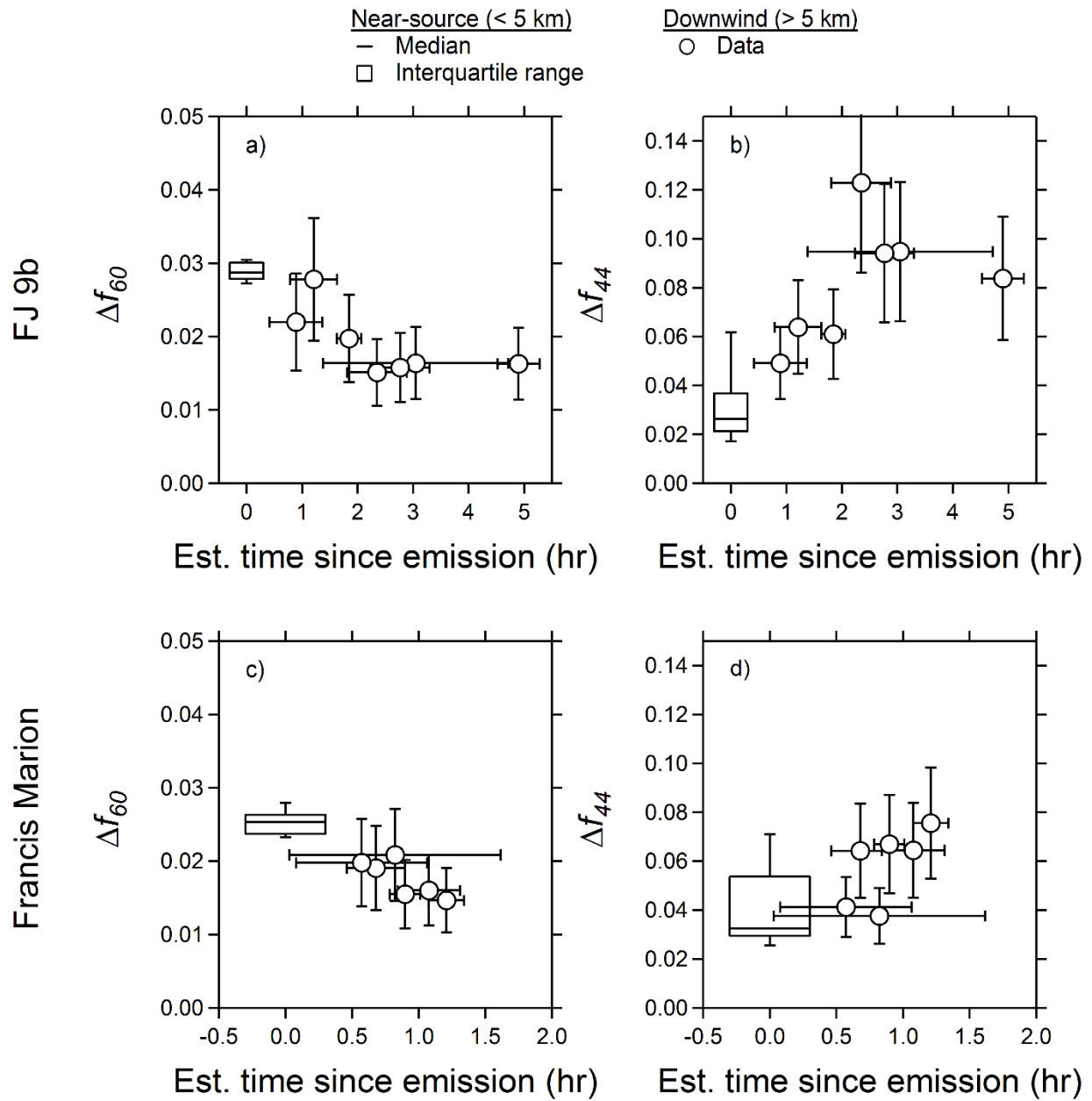
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Figure 3.

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As in Figure 2, but for the Francis Marion prescribed fire.

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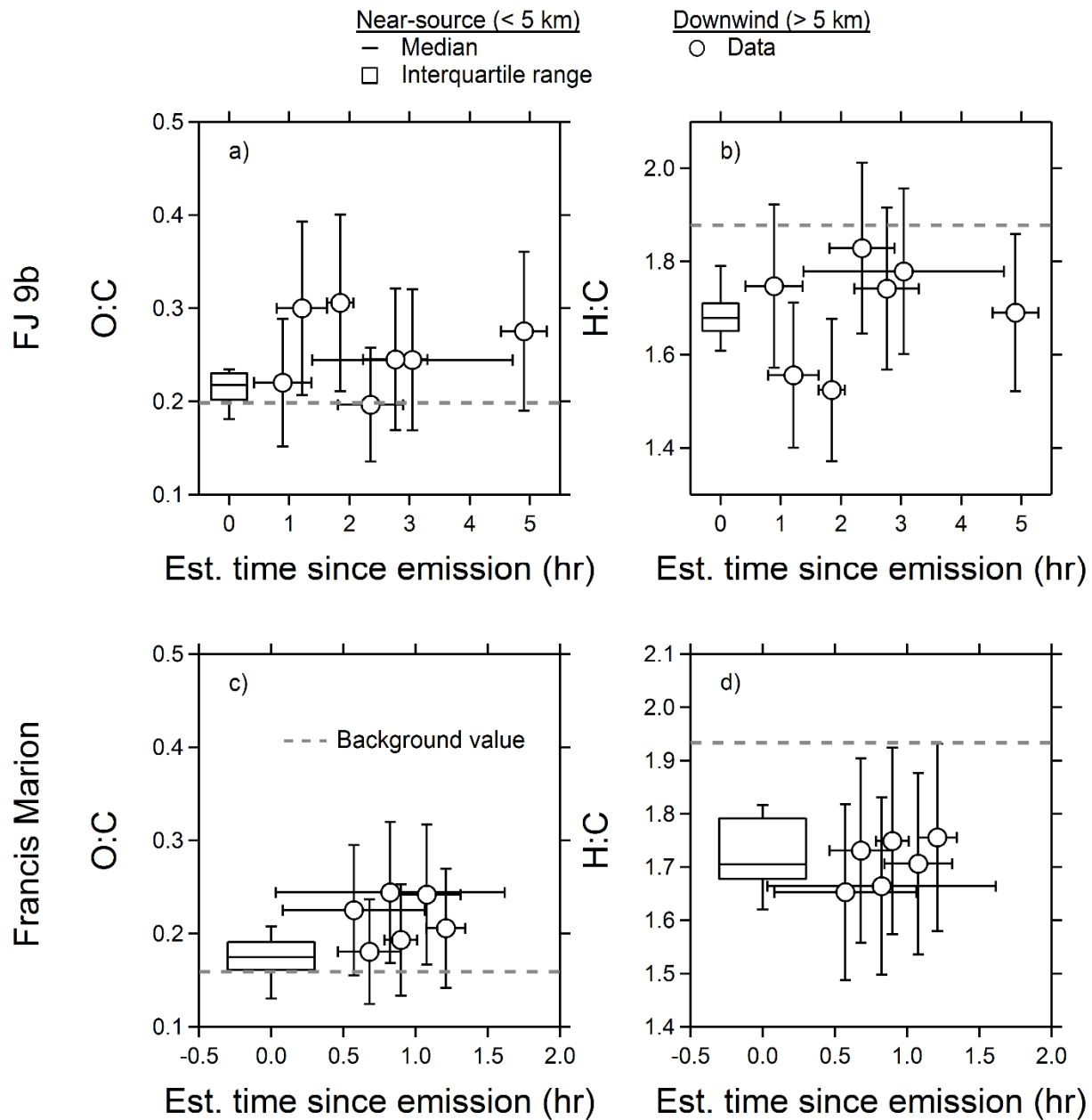
830 **Figure 4.**

831 Evolution of background-corrected AMS mass fractions. **a)** Δf_{60} for the FJ 9b fire; **b)** Δf_{44} for the
 832 FJ 9b fire; **c)** Δf_{60} for the Francis Marion fire; **d)** Δf_{44} for the Francis Marion fire. In all panels,
 833 there is a statistically-significant difference between data collected near the source and
 834 downwind. Box-and-whisker plots and markers are identical to those in Figures 2 and 3.

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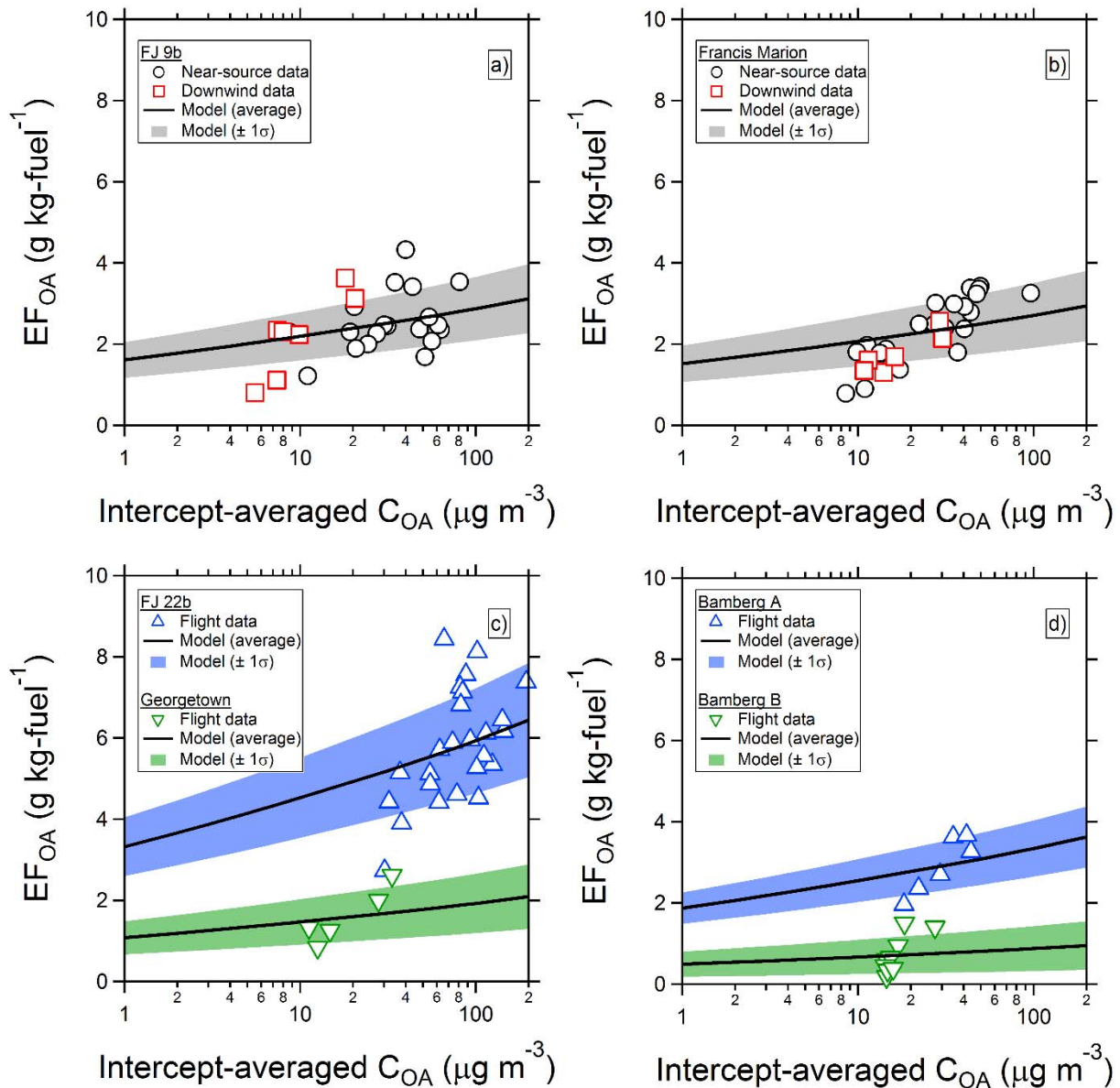
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840 **Figure 5.**

841 Evolution of elemental ratios derived from AMS data. **a)** O:C for the FJ 9b fire; **b)** H:C for the
842 FJ 9b fire; **c)** O:C for the Francis Marion fire; **d)** H:C for the Francis Marion fire. For both fires,
843 changes in O:C with increasing estimated time since emission are statistically significant.
844 Dashed line is the value of the parameter in the background measurements outside of plume
845 penetrations. Box-and-whisker plots and markers are identical to those in Figures 2 and 3.

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Figure 6.

Changes in the emission factor of excess OA due to gas-particle partitioning as a function of total observed OA. **a)** near-source (circles) and downwind (squares) data for the FJ 9b fire; **b)** near-source (circles) and downwind (squares) data for the Francis Marion fire; **c)** near-source data for the FJ 22b (upward-facing triangles) and Georgetown (downward-facing triangles) fires; and **d)** near-source data for the two fires attributed to the Bamberg site (“A”: upward-facing triangles; “B”: downward-facing triangles). Curves represent predictions using the laboratory parameterization from May et al. (2013).