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Continuous measurements at the urban roadside in an Asian Megacity by Aerosol Chemical Speciation Monitor (ACSM): Particulate matter characteristics during fall and winter seasons in Hong Kong

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27 **Abstract**

28 Non-refractory submicron aerosol is characterized using an Aerosol Chemical Speciation Monitor (ACSM) in the
29 fall and winter seasons of 2013 at the roadside in an Asian megacity environment in Hong Kong. Organic aerosol
30 (OA), characterized by application of Positive Matrix Factorization (PMF), and sulfate are found dominant. Traffic-
31 related organic aerosol shows good correlation with other vehicle-related species, and cooking aerosol displays
32 clear meal-time concentration maxima and association with surface winds from restaurant areas.
33 Contributions of individual species and OA factors to high NR-PM₁ are analyzed for hourly data and daily
34 data; while cooking emissions in OA contribute to high hourly concentrations, particularly during meal
35 times, secondary organic aerosol components are responsible for episodic events and high day-to-day PM
36 concentrations. Clean periods are either associated with precipitation, which reduces secondary OA with
37 a lesser impact on primary organics, or clean oceanic air masses with reduced long-range transport and
38 better dilution of local pollution. Haze events are connected with increases in contribution of secondary
39 organic aerosol, from 30% to 50% among total non-refractory organics, and influence of continental air
40 masses.

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45 1 Introduction

46 The Special Administrative Region of Hong Kong (HKSAR) is a global logistics and finance center
47 located at the south-eastern edge of the Pearl River Delta Region (PRD), China's largest manufacturing
48 area and one of the world's most densely populated regions. Hong Kong has been plagued by deteriorating
49 air quality, attributed to local emissions from traffic, residential and commercial activity, regional
50 pollution from the PRD and long-range transport (Nie et al., 2013; Wong et al., 2013; Yuan et al., 2013).

51 High-time-resolution online instruments can characterize ambient aerosols quickly and mitigate the
52 influence of changing environmental conditions. Few real-time studies have been conducted in Hong
53 Kong aside from recent measurement campaigns conducted by high resolution aerosol mass spectrometer
54 (HR-AMS) (Lee et al., 2013; Li et al., 2013; Li et al., 2015; Huang et al., 2015). Long-term AMS studies
55 tend to be costly and time-consuming due to the complexity of the instrument. The ACSM, whose design
56 is based on the AMS but has been substantially simplified, has seen a growing trend of use due to its
57 comparative ease of operation, robustness, and sufficient time resolution (~20-60min) for studies spanning
58 months or longer (Ng et al., 2011; Sun et al., 2012, 2013a, b; Budisulistiorini et al., 2013; Canonaco et al.,
59 2013; Takahama et al., 2013; Bougiatioti et al., 2014; Petit et al., 2015; Ripoll et al., 2015; Tiitta et al.,
60 2014; Minguillón et al., 2015).

61 A high resolution aerosol mass spectrometer (HR-ToF-AMS) had been previously deployed at an urban
62 site in the Shenzhen metropolitan area and a rural site in the PRD region in the fall months of October and
63 November (He et al., 2011; Huang et al., 2011). They found that organic constituents dominate, followed
64 by sulfate which is similar to this study, but the fraction of sulfate at the rural site is larger than that of the
65 urban site. Four OA components were identified at the urban site including HOA, BBOA, LV-OOA and
66 SV-OOA, but only three OA factors without HOA were resolved at the rural site. They both reported an
67 important contribution from BBOA with about 24% of total OA.

68 We also have previously deployed an HR-ToF-AMS at the supersite of the Hong Kong University of
69 Science and Technology (HKUST) to determine typical variations in submicron species concentrations,
70 overall composition, size distributions, PMF-resolved organic factors and degree of oxygenation. The
71 supersite measurements provided valuable insights into characteristics of mainly of secondary components
72 of submicron particulate matter, with dominance of sulfate and oxygenated organic aerosol species
73 observed (Lee et al., 2013; Li et al., 2013, 2015). Subsequent work was conducted at a downtown location

74 (Mong Kok) in Hong Kong, next to the roadside, in spring 2013 to assess important primary aerosol
75 sources in the inner-city to identify contributions of long-range transport to roadside pollution, and to
76 establish characteristic concentration trends at different temporal scales. Cooking aerosol was identified
77 as the dominant component in submicron non-refractory organics, followed by traffic-related emissions
78 (Lee et al., 2015).

79 Different from previous studies in Hong Kong, this work focuses on the characterization of roadside
80 aerosol during the fall and winter seasons, when the influence of transported air mass is greatest and PM
81 pollution in Hong Kong is generally more severe. Episodic haze events were found to be mainly driven
82 by secondary aerosol rather than primary emissions, while hourly high PM concentrations were often
83 driven by cooking aerosol. Statistical methods were employed to show that the correlation of COA and
84 HOA to SV-OOA varied under different conditions and in different times of the day. While HOA showed
85 a stronger relationship to SV-OOA overall, COA can be an important contributor to SV-OOA during meal
86 times.

87

88 **2. Experimental**

89 The roadside measurement data were collected from 3 September to 31 December, 2013 in Mong Kok
90 (MK), an urban area with dense buildings and population in the Kowloon peninsula under the Hong Kong
91 Environmental Protection Department (HKEPD) project (ref.: 13-00986). The sampling site was next to
92 the road-side air quality monitoring station (AQMS) of HKEPD at the junction of the heavily trafficked
93 Nathan Road and Lai Chi Kok Road (22°19'2"N, 114°10'06"E). The distribution of businesses in the
94 vicinity varies, with restaurants mainly to the east, commercial buildings to the south and east, small shops
95 for interior decoration, furniture and electrical goods to the west and residential buildings to the north of
96 the sampling location (Lee et al., 2015). The sampling setup is described in detail in the Supporting
97 Information, Section 1.

98 Non-refractory PM₁ (NR-PM₁) species (sulfate, nitrate, ammonium, chloride, and organics) were
99 measured in-situ by an Aerodyne Aerosol Chemical Speciation Monitor (ACSM, SN: 140-154). Other
100 data including meteorological data (wind, temperature, relative humidity, solar irradiation), volatile
101 organic compounds (VOCs) measured by an online gas-chromatography system (GC955-611 and GC955-
102 811, Synspec BV), and standard criteria pollutants (NO_x, SO₂ and PM_{2.5}) were provided by the HKEPD,

103 with equipment details available from the HKEPD air quality reports (Environmental Protection
104 Department, 2013).

105 The acquired 20-minute-average data were treated according to the general ACSM data analysis protocols
106 established in previous studies (Ng et al., 2011; Sun et al., 2012), using the standard WaveMetrics Igor
107 Pro-based Data Analysis Software (Version 6.3.5.5) and incorporating calibrations for relative ionization
108 efficiency (RIE), collection efficiency (CE) and detection limit (DL). Further details on data treatment can
109 be found in the Supporting Information, Section 2.

110 Factors contributing to organic aerosol were explored using PMF (Paatero and Tapper, 1994; Zhang et al.,
111 2011) with the Igor-Pro-based PMF evaluation toolkit (PET) (Ulbrich et al., 2009). In general, PMF can
112 be used to resolve organic aerosol (OA) into factors such as hydrocarbon-like OA (HOA), cooking OA
113 (COA), semi-volatile oxygenated OA (SV-OOA), low-volatility oxygenated OA (LV-OOA), and others.
114 ME-2 analysis with the SOFI tool as applied in several studies may yield additional insights but has not
115 been applied in this study due to its ongoing development (Canonaco et al., 2013; Minguillón et al., 2015).
116 The optimal factor number was determined by inter-comparing factors' mass spectra and time series,
117 correlations between factors and related tracers, and correlations with standard mass spectra; solutions
118 with 3, 4, 5 factors at $f_{peak}=0$ and 6 factors at $f_{peak}=-0.2$ were explored, after which the optimal f_{peak}
119 value was determined by repeating the above analysis with varying f_{peak} values.

120 The 4-factor solution (HOA, COA, SV-OOA, LV-OOA) is optimal, with $Q/Q_{exp}=0.8$ and better
121 differentiation between factor time-series ($R_{pr} < 0.6$; Fig. S6). The factors also correlate well with
122 associated inorganics and external tracers (NO_3 , SO_4 , NH_4 , NO_x ; Zhang et al., 2005, 2011; Ulbrich et al.,
123 2009), e.g. HOA with NO_x , SV-OOA with NO_3 , LV-OOA with SO_4 and NH_4 (Table S4). Furthermore,
124 the resolved mass spectra of four factors exhibit good similarity (all un-centered R (R_{uc}) > 0.80) with
125 reference source mass spectra from the AMS MS database (Ulbrich, I. M., Lechner, M., and Jimenez, J.
126 L., AMS Spectral Database, url: <http://cires.colorado.edu/jimenez-group/AMSsd>; Ulbrich et al., 2009).
127 PMF diagnostic details are shown in the Supplementary Information (SI: Section 3) and Fig. S7.

128 We note that m/z 60 and 73, important makers of BBOA mass spectra (Aiken et al., 2010; Cubision et al.,
129 2011; Huang et al., 2011), were resolved not only in COA but also in SV-OOA. Their presence in SV-
130 OOA is not the result of artifacts from the PMF analysis, but were attributed to the following reasons,
131 with more details shown in the supplement (Sect.4-6). Firstly, when PMF was run using only nighttime

132 data (between 0:00 and 6:00), i.e. when there is little COA (Fig. S10), these two ions still persist with
133 similar fractional intensities in SV-OOA as at other times. Secondly, increasing the number of PMF factors
134 and adjusting the f_{peak} value did not yield a distinct satisfactory BBOA factor. Thirdly, the time series of
135 m/z 60 and 73 show weak correlation with other burning tracers (EC_residual, CO_residual), with R_{pr} of
136 about 0.2 and 0.4 respectively, but track well with SV-OOA, with R_{pr} of 0.92 and 0.93 respectively (Fig.
137 S12, Table S9).

138 In terms of the possible sources of m/z 60 and 73, we observe that these two ions showed matching peaks
139 with the COA diurnal profile and good correlations with the sum of the time series of COA and LV-OOA,
140 with R_{pr} of 0.72 and 0.78 respectively. Furthermore, the ratio of the integrated signal at m/z 60 to the total
141 signal in the organic component mass spectrum is 0.48%, which is just slightly higher than the baseline
142 level ($0.3\% \pm 0.06\%$) observed in environments without biomass burning influence and with SOA
143 dominance in ambient OA (Cubision et al., 2011). This indicates that these two ions at Mong Kok were
144 mainly imbedded in cooking emissions and background aerosol due to transport rather than in a distinct
145 source with further details shown in the supplement (Sect. 6).

146

147 **3 Results and discussion**

148 **3.1 Mass concentration and chemical composition**

149 Figure 1a and 1b display meteorological data (relative humidity, temperature, and precipitation) and mass
150 concentrations of non-refractory PM_{10} (NR- PM_{10}) species and organic aerosol (OA) components,
151 respectively, between September and December 2013. Total NR- PM_{10} concentrations vary from $2.1 \mu\text{g}/\text{m}^3$
152 to $76.4 \mu\text{g}/\text{m}^3$ with an average of $25.9 \pm 13.0 \mu\text{g}/\text{m}^3$. ACSM NR- PM_{10} concentrations co-vary with that of
153 $\text{PM}_{2.5}$ measured by TEOM ($R^2=0.64$, slope=0.59; Fig. S1); the low slope value may be caused by the
154 different size cuts of ACSM and TEOM and the presence of refractory materials such as elemental carbon
155 (and to a lesser extent mineral dust and sea salt) which the ACSM cannot detect. Overall, daily $\text{PM}_{2.5}$
156 concentrations range from $3.7 \mu\text{g}/\text{m}^3$ to $106.0 \mu\text{g}/\text{m}^3$ and are largely (90.0%) within the 24-hr air quality
157 standard of $75 \mu\text{g}/\text{m}^3$ set by the Hong Kong Air Quality Objectives (HKAQO). Days with better air quality
158 ($\text{PM}_{2.5} < 35 \mu\text{g}/\text{m}^3$) are mainly observed in the month of September and during rainy periods in the other
159 months. The prevailing winds from the ocean in September not only brings in less polluted air mass but
160 also dilutes local air pollutants more compared with other seasons (Yuan et al., 2006; Li et al., 2015).

161 Precipitation has an obvious impact on total NR-PM₁ concentrations, but as we will discuss, has a lesser
162 effect on primary organics.

163 Overall, NR-PM₁ is dominated by organics and sulfate with relative contributions of 58.2% and 23.3%
164 and average concentrations of $15.1 \pm 8.1 \mu\text{g}/\text{m}^3$ and $6.0 \pm 3.5 \mu\text{g}/\text{m}^3$, respectively (Fig. 2a). Other
165 inorganic species (ammonium, nitrate and chloride) amount to approximately 20% of NR-PM₁. The
166 dominance of organics and sulfate is consistent with previous on-line studies in urban areas (e.g., Salcedo
167 et al., 2006; Aiken et al., 2009; Sun et al., 2012, 2013b) as well as previous filter-based studies in MK
168 (e.g., Louie et al., 2005; Cheng et al., 2010 and Huang et al., 2014). The measured composition is
169 consistent with earlier HR-AMS measurements carried out at the same site in spring and summer 2013
170 (Lee et al., 2015) with very similar overall species distribution, but slightly lower measured concentrations
171 as compared to the ACSM. This is likely due to the fact that sampling took place in different time periods
172 (spring-summer 2013 for the AMS campaign, fall-winter 2013 for the ACSM campaign). In the AMS
173 study, 6 PMF aerosol factors were identified (one additional OOA factor and one additional COA factor).
174 A marked difference is observed in the distribution of primary OA (POA) and secondary OA (SOA);
175 whereas in spring and summer (AMS), POA makes up 65% of total organics, the reverse is observed for
176 fall and winter (ACSM) where POA only amounts to 42% overall. A possible reason for this discrepancy
177 is the fact that impacts of regional pollution and long-range transport are usually higher during fall and
178 winter (Yuan et al., 2013; Li et al., 2015), thus contributing more SOA.

179 Elemental carbon (EC) concentrations are significant at the Mong Kok site but not measureable by ACSM
180 due to its high refractory temperature. EC has been discussed extensively in the previously mentioned
181 filter-based studies and a brief comparison of online ECOC measurements to the results of HR-AMS
182 measurements has been presented in the HR-AMS study (Lee et al., 2015). We therefore do not discuss
183 EC in detail in this work.

184 **3.2 OA Components**

185 PMF resolved four factors, including two primary OA factors (hydrocarbon-like OA (HOA) from traffic
186 emissions and cooking OA, or COA) and two oxygenated OA factors (OOA): highly oxidized low-
187 volatility OOA (LV-OOA) and the less-oxidized semi-volatile OOA (SV-OOA; Aiken et al., 2008;
188 Jimenez et al., 2009; Tiitta et al., 2014). The mass spectra are depicted in Fig. 3. The mass concentrations
189 of primary OA factors (HOA and COA), surrogates of local emissions, constitute 42% of total organics

190 and are slightly higher than that of LV-OOA (38%; Fig. 2b). SV-OOA contributes approximately 20% to
191 total OA and is associated with both the primary organic aerosol sources and LV-OOA (see Sect. 3.2).

192 **3.2.1 Hydrocarbon-like OA (HOA)**

193 The mass spectrum of HOA is dominated by the $C_nH_{2n-1}^+$ ion series (m/z 27, 41, 55, 69, 83, 97), typical
194 of cycloalkanes or unsaturated hydrocarbon, which account for 27% of total peak intensity in the HOA
195 spectrum. The other prominent group is the $C_nH_{2n+1}^+$ ion series (m/z 29, 43, 57, 71, 85, 99), typical of
196 alkanes and accounting for 26% of the total peak. This mass spectrum is very similar to the standard HOA
197 spectrum with R_{uc} of 0.92, and its fractions of $C_nH_{2n-1}^+$ and $C_nH_{2n+1}^+$ (27%, 26%) are consistent with
198 standard ones (28%, 27%) (Ng et al., 2011). This HOA spectrum is also consistent with that resolved by
199 HR-ToF-AMS at the HKUST Supersite on the dominance of saturated C_xH_y -type ions, most notably at
200 m/z 43 and 57 (Lee et al., 2013).

201 HOA has an average concentration of $2.7 \pm 0.98 \mu\text{g}/\text{m}^3$ (Fig. 1b) and shows strong diurnal variations,
202 including a regular decrease to about $1 \mu\text{g}/\text{m}^3$ during 0:00-5:00 (Fig. 4h) which is discussed in 3.3 section
203 in detail. In addition, the temporal variation of HOA displays strong correlations with NO_x ($R_{pr}=0.69$),
204 CO ($R_{pr}=0.62$) and several VOCs (Pentane, Toluene, Benzene) as shown in Table S10.

205 The diurnal patterns of vehicle number, HOA, NO, NO_2 , NO_x and traffic-related VOCs (i-pentane, n-
206 pentane, toluene, octane, benzene, i-butane and n-butane) are depicted in Fig. 5. Vehicle counting on Lai
207 Chi Kok road next to the sampling site spanned 28 - 31 May 2013 and was provided by HKEPD (Lee et
208 al., 2015). Although these dates are different from our campaign period, they provide a useful reference
209 for the traffic conditions near the site. In general, more gasoline and diesel vehicles are observed during
210 daytime than at night. The decrease of these vehicles during 22:00–4:00 is in agreement with the diurnal
211 profile of HOA (Fig. 4h). On the other hand, liquefied petroleum gas (LPG) vehicles, which are usually
212 taxis, show slightly higher numbers during 22:00–4:00 at the site. HOA increases sharply from 1.5
213 $\mu\text{g}/\text{m}^3$ at about 6:00 to the morning peak of $3.6 \mu\text{g}/\text{m}^3$ at 9:00, and then persists at high concentrations
214 until midnight, including another peak with $3.9 \mu\text{g}/\text{m}^3$ at 17:00. The diurnal pattern of HOA is consistent
215 with that of NO_x ($\text{NO}+\text{NO}_2$), which is almost exclusively from vehicle emissions. These results are
216 consistent with the traffic conditions at MK with heavy traffic continuously after 6:00 and rush hours
217 from 7:00 to 11:00 and 16:00 to 19:00. NO_2 is the result of direct emission as well as formation from
218 NO, and it increased during daytime to reach a maximum even higher than that of NO at about 17:00.

219 Concentrations of toluene (a fuel additive) and pentane and octane (significant components in exhaust of
220 petrol vehicles; Huang et al., 2011; Wanna et al., 2008) start to increase during the morning rush hour
221 (7:00) and peak between 18:00 and 19:00. HOA and NO_x show a distinct morning peak at ~8:00 when a
222 small shoulder is also found in the VOCs. Butane, a constituent of LPG, displays a diurnal pattern
223 different from that of HOA, with higher concentrations between 22:00 and 4:00; LPG-fueled taxis are a
224 major means of transport during the nighttime and early morning, and fuel leakage during refueling may
225 contribute to the observed pattern. Furthermore, fuel leakage during refueling of LPG vehicles may
226 contribute more than diesel-fueled vehicular emissions to butane even though the number of diesel
227 fueled vehicles is slightly higher than LPG ones at that time. At last, the sampling site is near a major
228 junction serving a number of district centers (West Kowloon, Sha Tin, Tsim Sha Tsui) and is therefore
229 frequented by taxis.

230 **3.2.2 Cooking-related OA (COA)**

231 The most prominent ions of the resolved COA profile at MK were m/z 41 (mainly C₂HO⁺, C₃H₅⁺) and m/z
232 55 (mainly C₃H₃O⁺, C₄H₇⁺). Ratios of m/z 41/43 = 1.8 and m/z 55/57 = 2.2, which are distinctly larger than
233 that of HOA at 0.73 and 0.76 respectively (Fig. 4); such ratios have been widely reported for COA in
234 AMS and ACSM studies. For example, Lanz et al. (2010) reported ratios of m/z 41/43 and m/z 55/57 of
235 0.5 and 0.4 in HOA, and 1.2 and 1.2 in COA, respectively, while Sun et al. (2013a) reported 0.5 for these
236 two ratios in HOA and 2.3 for those in COA, respectively.

237 Fig. 6a shows COA concentrations sorted by wind direction in MK. The COA concentration reaches up
238 to 12 µg/m³, contributing ~60% of total organics, when easterly winds dominate, probably due to the large
239 number of restaurants located on the eastern side of the sampling site (Fig. 6a). In general, COA
240 contributes significantly to the total mass of organic aerosol with an average fraction of 24% (3.7 µg/m³),
241 in line with the 16-30% COA contributions found in several cities including London, Manchester,
242 Barcelona, Beijing, Fresno, and New York (Allen et al., 2010; Huang et al., 2010; Sun et al., 2013b; Mohr
243 et al., 2012; Ge et al., 2012). Fig. 6b and c compare the chemical composition of NR-PM₁ and OA during
244 meal times (lunch, 12:00-14:00, and dinner, 19:00-21:00) and non-meal times (0:00-6:00); the non-meal
245 period is defined by the periods of low concentration (<2 µg/m³) in the COA diurnal pattern. During dinner
246 time, the average concentration of organics increases by about 11 µg/m³ and its contributions in total NR-
247 PM₁ increase to 70%, while the concentrations of other species do not change much (Fig. 6b). As shown

248 in Fig. 6c, the increase in organic concentrations results from the increase in COA from 1.7 to 7.8 $\mu\text{g}/\text{m}^3$
249 ($\sim 360\%$ increase), and to a lesser extent increases in SV-OOA (from 1.5 to 4.5 $\mu\text{g}/\text{m}^3$, a $\sim 200\%$ increase)
250 and in HOA (from 1.4 to 3.2 $\mu\text{g}/\text{m}^3$, a $\sim 130\%$ increase). As shown in Table 1, the average concentration
251 of organics during dinner time is 5 $\mu\text{g}/\text{m}^3$ higher than that during lunch, and this increase is attributed to
252 the increase of COA and SV-OOA mass but not of HOA. This is consistent with the expectation that the
253 cooking activities at MK are higher during dinner than during lunch, while traffic during dinner is
254 comparable to or smaller than that during lunch (Fig. 4f and Fig. 4h). The increase of SV-OOA during
255 dinner time may be the result of enhanced cooking emissions and possibly less evaporation due to lower
256 ambient temperature; contributions from traffic emissions are not likely to be important since there is little
257 increase of HOA during the meal times.

258

259 3.2.3 Oxygenated OA (OOA)

260 LV-OOA is characterized by the prominent m/z 44 ion (mainly CO_2^+) and minor $\text{C}_n\text{H}_{2n-1}$ and $\text{C}_n\text{H}_{2n+1}$ ion
261 series generated by saturated alkanes, alkenes and cycloalkanes. The LV-OOA spectrum correlates well
262 with the standard LV-OOA spectrum (Fig. 3), with a R_{uc} of 0.97. The LV-OOA time series is associated
263 with that of SO_4^{2-} with a R_{pr} of 0.86 (Fig. 1), consistent with reports in the literature (DeCarlo et al., 2010;
264 He et al., 2011; Zhang et al., 2014; Tiitta et al., 2014). The LV-OOA diurnal pattern varies little, suggesting
265 that it is part of the background aerosol, possibly resulting from long range transport (Li et al., 2013; 2015).
266 SV-OOA, which is less oxidized than LV-OOA, is marked by the dominant ions of m/z 43 and m/z 44
267 mainly contributed by $\text{C}_2\text{H}_3\text{O}^+$ and CO_2^+ . The mass spectrum of SV-OOA closely resembles that of
268 'standard' SV-OOA with a R_{uc} of 0.87 (Fig. 3). Some marker fragments of COA and HOA, for example,
269 m/z 41, 43, 55, and 57, are present in the SV-OOA mass spectrum. SV-OOA concentrations are also
270 weakly associated with those of HOA and their co-emitted precursors (benzene and toluene), with R_{pr} of
271 0.58, 0.65 and 0.51 respectively. In fact, the correlation between SV-OOA and benzene is better than that
272 of HOA and benzene (0.56). The diurnal pattern of SV-OOA also shows peaks at meal times like COA.
273 Lastly, the fraction of signal at m/z 44 (f_{44} fraction) of SV-OOA at MK is twice that of the standard
274 measured by Q-AMS (Zhang et al., 2014; Tiitta et al., 2014). Together, these results suggest that SV-OOA
275 may be correlated with POA (HOA and COA), possibly due to rapid oxidation of POA and semivolatile
276 gases, which may then form SV-OOA. The variation of the average concentration of SV-OOA as a
277 function of binned LV-OOA concentration in increments and a bin width of 2 $\mu\text{g}/\text{m}^3$ is shown in Fig. S13.
278 The linear, positive relationship between SV-OOA and LV-OOA suggests that non-local formation and

279 subsequent transport may also contribute to the measured SV-OOA at MK. However, it should be
280 mentioned that ACSM resolved organic spectra have been observed to show higher f_{44} in other studies
281 (Crenn et al, 2015, Fröhlich et al., 2015) compared to HR-ToF-AMS measurements due to inherent
282 instrumental uncertainties in the determination of f_{44} . This might have caused the elevated f_{44} observed
283 in our SV-OOA spectrum.

284

285 Figure 7 displays the concentration of different OA factors (coded by color) as a function of binned O_x
286 concentration (ppb) and binned temperature ($^{\circ}C$) with a bin width of 15ppb and $5^{\circ}C$ respectively. In
287 general, the concentration of all OA factors increases as O_x increases across all temperatures. While it is
288 understood that LV-OOA and SV-OOA are correlated with O_x because they all result from similar
289 photochemical activities, the correlation between HOA and O_x is the result of the good correlation (0.78)
290 between HOA and NO_2 , which accounts for 84% of total O_x . NO_2 is partly emitted from vehicles and
291 partly formed by secondary oxidation at MK as discussed in Sect. 3.2.1. Increase in ambient temperature
292 is associated with decrease in HOA and SV-OOA, likely due to evaporation effects and partitioning, but
293 it has no obvious correlations with LV-OOA and COA.

294 To further assess the relative importance of other OA factors to the resolved SV-OOA, ordinary least
295 squares (OLS) regressions were conducted. Considering the potential influence of primary OA on the
296 regression results, the whole dataset was separated into three time periods consisting of: meal time (MT;
297 12:00 -14:00, 19:00-21:00) marked by enhanced COA; background time (BT; 0:00-6:00) marked by low
298 POA; and other time (OT; 6:00-12:00, 14:00-19:00 and 21:00-24:00). The data of each period was further
299 divided into high/low temperature (HTemp, LTemp = $T < 22.5^{\circ}C$.) and high/low O_x (Hi O_x , LO O_x = $O_x <$
300 70ppb) to reveal impacts of temperature and the degree of oxygenation on the correlations among OA
301 factors.

302 Tables 2 and 3 show the coefficients of HOA, COA and LV-OOA in the regression equation for the
303 reconstructed SV-OOA and their average concentrations during different periods under high/low
304 temperature and high/low O_x respectively. The average concentrations of HOA and SV-OOA under
305 HTemp are obviously lower than under LTemp for each period but the concentration of COA and LV-
306 OOA varies little across different temperatures (Table 2). Considering the stronger correlations between
307 HOA and SV-OOA than between COA and SV-OOA, a stronger and closer temperature dependence of
308 HOA and SV-OOA was revealed. In addition, the regression coefficients of HOA and COA during each

309 period under HTemp are much smaller than under LTemp, reflecting a weakening of their relationship
310 with SV-OOA as temperature increases.

311 Consistent with the discussion of Fig. 7, the concentrations of HOA, SV-OOA and LV-OOA under HiOx
312 are much higher than those under LOx for each period (Table 3). Besides, HOA shows a higher correlation
313 with SV-OOA under HiOx due to more intensive oxidation of HOA to SV-OOA. However, LV-OOA
314 shows a reverse trend with smaller coefficients with SV-OOA. It is probable that HiOx conditions favor
315 the conversion of SV-OOA to LV-OOA leading to smaller coefficient of LV-OOA on SV-OOA, although
316 overall most LV-OOA is considered to be from transport.

317 At last, we also can conclude that HOA overall has a stronger relationship to SV-OOA compared to COA,
318 supported by much higher coefficients of HOA than that of COA over all time periods, temperature and
319 O_x levels. Cooking emissions are not as important to SV-OOA in the BT periods but they can be important
320 during MT periods, indicated by the lowest concentration and correlation with SV-OOA during BT, but
321 highest concentration during MT periods.

322

323 **3.3 Diurnal patterns**

324 The diurnal profiles of NR-PM₁ species and OA components are depicted in Fig. 4. Total organics display
325 a diurnal pattern with two pronounced peaks during 12:00-14:00 and 19:00-21:00, corresponding to the
326 peaks of COA at lunch and dinner time respectively. In addition, organics increase at about 10 am, which
327 may be related to the increase of local emissions of HOA and COA by 2.3 µg/m³ and 1.1 µg/m³
328 respectively from 6:00 to 10:00.

329 The mass concentration of sulfate (Fig. 4b) does not show any diel variation. It is likely that sulfate, as a
330 regional pollutant, is mainly formed during long-range transport, leading to the lack of a specific diurnal
331 pattern at MK; a similar flat diurnal pattern for sulfate has also been found at the HKUST supersite in
332 Hong Kong (Lee et al., 2013; Li et al., 2015). These results differ significantly from observations in
333 Beijing and Lanzhou in China and Welgegund in South Africa (Sun et al., 2012, 2013b; Xu et al., 2014;
334 Tiitta et al., 2014) where sulfate displays an obvious increase at noon-time in summer and wet seasons
335 due to either photochemical reaction or aqueous oxidation of SO₂. The difference may result from the
336 much lower level of sulfur dioxide (SO₂) with an average of 4.6 ppb in MK compared to for example, ~32
337 ppb in Beijing, where coal combustion leads to a much higher SO₂ concentration (Lin et al., 2011). Sulfate

338 and relative humidity (RH) have almost no correlation ($R^2=0.06$) in MK, suggesting little importance of
339 local aqueous processing for the formation of sulfate.

340 Nitrate shows a slight dip around noontime, corresponding to the increase of the ambient temperature (Fig.
341 4j). Evaporative loss of particulate nitrate might outweigh the secondary production of nitrate during this
342 time. The diurnal pattern of ammonium (Fig. 4d) is very similar to that of sulfate, as expected based on
343 their commonly observed association in atmospheric particles. Chloride (Fig. 4e) has rather low
344 concentrations and shows a similar diurnal variation to that of nitrate, likely due to its volatility.

345

346 **3.4 Day-of-week patterns**

347 Fig. 8a shows the average concentration trends on individual days of the week for NR-PM₁ species and
348 Fig. 8b describes the diurnal patterns of the OA components for weekdays, Saturdays and Sundays.
349 Because of the small datasets on Saturdays and Sundays, data beyond one standard deviation from the
350 mean ($25.9 \pm 13.0 \mu\text{g}/\text{m}^3$) were removed from the whole dataset to remove the influence of episodic events
351 in this analysis. Overall, total NR-PM₁ concentrations have no obvious variation (average variation less
352 than 5%) from Monday to Saturday, but drop by 16% on Sundays compared to Saturdays. This weekend
353 difference is opposite to the result found in Beijing where higher concentrations occurred on Sundays than
354 Saturdays (Sun et al., 2013b). On the other hand, others such as Lough et al. (2006) and Rattigan et al.
355 (2010) reported that both Saturdays and Sundays had obvious traffic emissions reduction due to less
356 human activities on weekends in Los Angeles and New York.

357

358 Organics and secondary inorganics (SO₄, NH₄ and NO₃) contributed 54% and 46% respectively to the
359 concentration difference between Sundays and Saturdays in MK. The difference in organics is mainly
360 attributed to the variation of HOA, which shows very similar diurnal variations on Saturdays and
361 weekdays, but has an average decrease of 23% after 7 am on Sundays. A 37% reduction of traffic-related
362 carbonaceous aerosol on Sundays compared with weekdays in MK has been reported (Huang et al., 2014).
363 In Hong Kong many people work on Saturday, which leads to a traffic pattern similar to normal weekdays.
364 COA shows nearly the same diurnal patterns on all days, and LV-OOA and SV-OOA do not show obvious
365 variations. Overall, local emissions from traffic contribute most to the day-of-week variations in organics.

366

367 **3.5 Contributions of individual species and OA factors to high NR-PM₁**

368 Fig. 9a, 9b and 9c show the variation in hourly mass concentration of NR-PM₁ species and OA
369 components and their mass fractions as a function of hourly total NR-PM₁ mass loading. Below 50 µg/m³,
370 all aerosol species display a nearly linear increase with PM₁ mass loading, with slopes of about 0.5 for
371 organics, 0.25 for sulfate and LV-OOA, and around 0.1 for nitrate, ammonium, COA, HOA and SV-OOA
372 (Fig. 9a). While the fractions of NH₄ and organics remain relatively stable, sulfate exhibits a decrease and
373 then an increase, and NO₃ and chloride shows a gradual increase then a decrease respectively as NR-PM₁
374 increases to 50 µg/m³ (Fig. 9b, 9c). Although the mass concentrations of all organic factors increase as
375 NR-PM₁ increases, SV-OOA is the only factor that increases in mass fraction. Primary OA components
376 (HOA and COA) and transported OA (LV-OOA) show a decrease in fraction and stable contributions
377 respectively as NR-PM₁ increases to 50 µg/m³, while the contribution of SV-OOA increases sharply from
378 around 5% to 25% of total organic mass. Beyond 50 µg/m³, the mass loadings of SO₄ and organics increase,
379 while those of NH₄, NO₃ and LV-OOA remain almost constant, which differs from the observations in
380 Beijing, where NH₄ and NO₃ kept a linear increase from 50 to about 200 µg/m³ (Sun et al., 2013b; Zhang
381 et al., 2014). In terms of fractions, only COA and to a lesser extent SV-OOA, increase as NR-PM₁
382 increases further. In fact, over 80% of the high hourly NR-PM₁ concentrations (>50 µg/m³) are observed
383 during the meal-time periods with enhanced cooking activities.

384 When the hourly averages in Fig. 9 are replaced by daily averages (Fig. 10), the COA concentration varies
385 little and its fraction does not exhibit an increase but instead decreases significantly with increasing daily
386 NR-PM₁. On the other hand, the fractions of SV-OOA and LV-OOA clearly increase. This analysis
387 suggests that while cooking OA is responsible for the hourly high concentrations during meal times and
388 potential high hourly PM levels, LV-OOA/SV-OOA are responsible for episodic events and high day-to-
389 day PM levels.

390 To analyze the difference in particle composition and meteorological conditions among episodic periods
391 and clean periods, three heavy polluted episodes (19-22, 23-26 Oct and 10-13 Dec) and two clean periods
392 (17-18 Sep and 14-18 Dec), highlighted in Fig. 1, were analyzed. The average concentrations of these
393 chosen periods are larger than one standard deviation from the average concentration of the campaign
394 ($25.9 \pm 13.0 \mu\text{g}/\text{m}^3$). The composition, meteorological features (T and RH) and oxidation index (O_x and
395 f_{44}) of these five events are shown in Table 4. Clean period 1 (C1) is characterized by low NR-PM₁
396 concentrations (below 13 µg/m³), prevailing coastal wind (easterly wind), lack of rain, high ambient

397 temperature (~28 °C) and high relative humidity (~70%). Another clean period (C2) features continuous
398 precipitation with the coldest and most humid weather condition in the period studied. Haze period 1 (H1)
399 has similar temperature and humidity as C1 but is marked by mixed continental/oceanic winds. From H1
400 to the following haze period (H2), the observed wind direction shifts to reflect continental transport, with
401 a significant decrease in RH to 36%. Haze period 3 (H3), just before C2, is also dominated by continental
402 winds but with lower temperatures (~19 °C) than during other haze events.

403 Although the total NR-PM₁ of C1 (12.2 µg/m³) and C2 (11.8 µg/m³) are both only 25–30% of that during
404 haze periods, they were driven by different mechanisms. The main differences in meteorological
405 conditions between C1 and C2 are the dominance of continental wind rather than coastal wind, much
406 lower temperature and the existence of precipitation in C2. The low concentration of C1 is mainly
407 attributed to easterly wind bringing less air pollutants and diluting local air pollutants. To a lesser extent,
408 it is influenced by both particle evaporation, especially for SV-OOA, and dilution of local emissions
409 during high temperatures, which might be the reason why HOA, COA and SV-OOA in C1 are lower than
410 in C2 despite the lack of rain. The low mass loading of C2 was mainly caused by the wet deposition of
411 precipitation. It dramatically reduces the concentration of secondary species such as SO₄, NH₄, NO₃, SV-
412 OOA and LV-OOA, but not primary HOA and COA. Compared to the adjacent period H3, the total
413 organic mass reduces by 68% to an average of 8.1 µg/m³ (Table 4). Precipitation effectively removes
414 secondary particles but is less efficient for primary particles that are continuously generated locally.

415
416 With a similar continental source region as C2, the most severe pollution event H3 occurred during 10–
417 13 December with an average NR-PM₁ of 47.7 µg/m³. The persistent northerly wind continually brought
418 air masses from the PRD region into Hong Kong and led to a marked mass increase of secondary species
419 of SO₄, NH₄, NO₃, LV-OOA and SVOOA. Furthermore, H3 is characterized by the highest mass
420 concentration and relative contribution of nitrate and SV-OOA compared with other haze periods. This is
421 likely due to the average temperature of H3 being 5–6°C lower than that of other haze events.

422 In addition, although all three haze events have very similar SO₄ mass loading, there is a ~ 50% increase
423 in NH₄ concentration during the H3 episode, consistent with the increase of nitrate in that period.

424

425 The other two haze events are adjacent with influence from both continental and oceanic region in H1 and
426 continental source region in H2. The mixed pattern of source regions during H1 identified as land–sea
427 breeze (Fig. S14) can redistribute PM pollution over the whole PRD region and accumulate air pollutants

428 effectively (Lo et al., 2006; Chan and Yao, 2008; Lee et al., 2013). The pronounced high concentration of
429 LV-OOA and SV-OOA, jointly contributing 70% of total organics, reflects the oxidation of primary
430 emissions in the PRD under such cycles, which is also observed at the suburban HKUST site (Lee et al.,
431 2013). The periodic nitrate peaks in H1 with low concentration in daytime and high concentration in
432 nighttime coincide with temperature changes. During H2 period, the prevailing wind is northwesterly and
433 there is a sharp decrease in relative humidity. It is interesting to note that the dip in RH during H2 coincides
434 with the dip in sulfate, ammonium, nitrate and LV-OOA; this might be caused by decreased aqueous-
435 phase processing, and by decreased gas-particle partitioning associated with water uptake under low RH
436 for secondary aerosol particles (Sun et al., 2013a, b).

437 The fractions of f_{44} during these three haze occasions are all lower than that at HKUST (Li et al., 2013),
438 which reflects a larger abundance of the less oxygenated POA at the urban MK site. In addition, the POA
439 concentration (HOA+ COA) does not change much between clean periods and haze periods. However, its
440 relative contribution decreases from about 50% during clean periods to 30% during haze events because
441 of the pronounced variation of secondary OA as shown in Fig. 11.

442

443 **4. Conclusions**

444 The characteristics and sources of ambient submicron non-refractory particulate matter (NR-PM₁) were
445 investigated in an urban roadside environment in Hong Kong using an Aerodyne ACSM from September
446 to December, 2013; these are the first ACSM measurements in Hong Kong. Organics and sulfate dominate
447 total NR-PM₁, making up more than 50% and 20% of measured mass concentration, respectively. PMF
448 analysis of organic aerosol mass spectra yielded four characteristic organic aerosol (OA) factors:
449 hydrocarbon-like organic aerosol (HOA), cooking organic aerosol (COA), semi-volatile oxygenated
450 organic aerosol (SV-OOA) and low-volatility oxygenated organic aerosol (LV-OOA). Primary OA factors
451 (HOA and COA) from fresh emissions contribute 43% of total organics, slightly larger than that of LV-
452 OOA, which is generally a transported pollutant in this study, with about 37% of total organics. SV-OOA
453 contributes about 20% of total organics and is variably correlated with HOA, COA and LV-OOA under
454 different conditions and in different times of day. While HOA showed a stronger relationship with SV-
455 OOA overall, COA can be an important contributor to SV-OOA during meal times. In addition, the

456 transported pollutants reflected by LV-OOA displays a relatively stable correlation with SV-OOA during
457 different investigated periods (BT, MT, and OT).

458 The mass loadings of traffic related aerosol (HOA) are consistent with expected traffic count data and
459 correlate well with various vehicle-related VOCs and NO_x. Furthermore, HOA, with an average decrease
460 of 23% on Sundays, was mainly responsible for the lower organic concentrations on Sundays when
461 compared with other days. Cooking aerosol (COA) displays a well-defined diurnal variation with lunch-
462 and dinner-time peaks and contributes on average 40% of total organics during mealtimes; COA is clearly
463 associated with local easterly winds, which coincides with the location of nearby restaurants.

464 The contributions of individual species and OA factors to high NR-PM₁ were analyzed based on hourly
465 data and daily data. While cooking is responsible for the hourly high concentrations during meal times,
466 LV-OOA/SV-OOA are responsible for episodic events and high daily PM concentrations. Three heavily
467 polluted episodes and two clean periods were recorded during sampling and attributed to different
468 meteorological and circulatory conditions. The analysis of clean periods shows that precipitation has an
469 obvious deposition impact on total NR-PM₁ concentrations, but has a lesser effect on primary organics.
470 Clean ocean wind not only brings in less polluted air mass, but also dilutes the local air pollutants. During
471 this campaign, high-PM events were generally related to continental air mass influence or land-sea breeze
472 circulatory conditions, which has less influence on primary emissions but significant effects on secondary
473 particles, with a pronounced increase in the secondary OA contribution during haze events (from 30% to
474 50%).

475

476 **Tables**477 **Table 1.** Average concentrations of NR-PM₁ and OA components during lunch time, dinner time and non-meal times

Species $\mu\text{g}/\text{m}^3$	Lunch	Dinner	Non-meal
Org	18.8	23.7	10.3
SO ₄	5.8	6.1	6.3
NH ₄	2.6	2.9	3.0
NO ₃	1.4	1.8	1.6
Chl	0.1	0.2	0.2
Organic aerosol components			
HOA	3.2	3.2	1.4
COA	6.2	9.6	1.7
LV-OOA	5.8	5.4	5.6
SV-OOA	3.6	5.5	1.5

478

479

480 **Table 2.** Regression of SV-OOA vs. HOA, COA and LV-OOA and concentrations of OA factors and O_x under high and
481 low temperature (LT and HT) conditions of the three chosen periods (MT, BT and OT).

Period Temperature	Meal time (MT)		Background time (BT)		Other time (OT)	
	LTemp	HTemp	LTemp	HTemp	LTemp	HTemp
Coefficients ^a						
HOA	0.80	0.56	0.70	0.43	0.48	0.23
COA	0.29	0.15	0.22	0.00	0.31	0.11
LV-OOA	0.25	0.23	0.23	0.24	0.25	0.28
Adjusted R ²	0.90	0.81	0.83	0.57	0.85	0.73
Average Concentration ($\mu\text{g}/\text{m}^3$, ppb)						
HOA	3.71	2.85	1.60	1.18	3.51	2.88
COA	7.34	7.40	1.61	1.54	3.50	3.74
LV-OOA	5.46	5.57	5.91	5.07	5.85	5.99
SV-OOA	6.30	3.89	2.68	1.44	4.1	2.39
O _x (ppb)	83.12	85.23	58.71	53.45	75.06	76.77

482 ^a The coefficient of HOA, COA and LV-OOA in the regression equation reconstructing SV-OOA under LTemp (T <22.5 °C)
483 and HTemp (T >22.5 °C) during meal times (12:00-14:00, 19:00-21:00), background time (0:00-6:00) and other times. The
484 average temperature of the whole campaign is 22.5 °C. All entries of coefficients are significant at the 1% level (two-tailed),
485 except that of HOA/OT, which is significant at the 5% level.

486

487 **Table 3.** Regression of SV-OOA vs. HOA, COA and LV-OOA and concentrations of OA factors and Temperature under
488 high and low O_x (HiO_x and LO_x) of the four chosen periods (MT, BT and OT).

	Meal time (MT)		Background time (BT)		Other time (OT)	
	LO _x	HiO _x	LO _x	HiO _x	LO _x	HiO _x
Coefficients ^a						
HOA	0.50	1.13	0.62	0.64*	0.08*	0.52
COA	0.13	0.14	0.00	0.15	0.14	0.14
LV-OOA	0.33	0.10*	0.26	0.18	0.34	0.21
Adjusted R ²	0.73	0.86	0.73	0.80	0.67	0.78
Average Concentration (µg/m ³ , ppb)						
HOA	2.24	3.41	1.20	2.03	2.11	3.55
COA	7.31	7.57	1.59	1.73	2.77	3.71
LV-OOA	3.50	5.92	5.07	7.22	4.06	6.77
SV-OOA	3.22	5.29	1.85	2.79	1.8	3.56
Temp (°C)	23.30	23.80	21.48	20.39	22.01	22.74

489 ^a The coefficient of HOA, COA and LV-OOA in the regression equation reconstructing SV-OOA under LO_x (O_x<70 ppb)
490 and HT (O_x>70 ppb) during meal times (12:00-14:00, 19:00-21:00), background time (0:00-6:00) and other times. 70 ppb is
491 the average O_x of the whole study. All entries of coefficients are significant at 1% level (two-level), except those indicated
492 with *, which indicates significance at the 5% level.

493

494 **Table 4.** Measured and calculated parameters in four chosen periods (C1, H1, H2, H3 and C2)

495

	Clean period 1 (C1) ^a		Haze period 1 (H1)		Haze period 2 (H2)		Haze period 3 (H3)		Clean period 2 (C2)	
RH (%)	70.8		65.0		36.4		64.8		84.6	
T (°C)	27.6		25.0		23.8		18.7		13.2	
O_x (ppb)	69.6		82.0		99.5		70.4		40.9	
f₄₄	0.114		0.118		0.120		0.108		0.057	
Precip(mm)	0		0		0		0		8.9	
Wind	coastal		continental/oceanic		continental		continental		continental	
(µg/m³, %)	Conc.	Perc.	Conc.	Perc.	Conc.	Perc.	Conc.	Perc.	Conc.	Perc.
NR-PM₁	12.2		44.1		39.0		47.7		11.6	
Org	6.7	54.4	25.2	57.2	21.1	54.2	25.1	52.6	8.1	69.6
SO₄	3.8	31.2	11.8	26.8	12.1	30.9	11.4	23.8	1.5	12.8
NH₄	1.2	9.9	4.4	10.1	4.4	11.3	6.5	13.6	1.1	9.4
NO₃	0.4	3.5	2.4	5.6	1.3	3.4	4.4	9.2	0.8	7.3
Chl	0.1	1.0	0.2	0.4	0.1	0.2	0.4	0.8	0.1	0.9
HOA	1.2	18.5	3.8	15.1	3.0	14.4	4.2	16.9	2.1	26.2
COA	2.3	34.8	3.7	14.5	3.3	15.5	3.3	13.1	2.6	31.7
LV-OOA	3.0	44.8	11.5	45.4	10.2	48.4	9.9	39.6	1.8	22.0
SV-OOA	0.1	2.0	6.3	25.0	4.5	21.6	7.6	30.4	1.6	20.1

496 ^a Average of data from clean days (C1 and C2) and hazy days (H1, H2 and H3). C1: 17-18 September; H1: 19-22 October;
497 H2: 23-26 October; H3: 10-13 December; C2: 14-18 December.

498 T: temperature; RH: relative humidity; O_x: odd oxygen (O₃ + NO₂) in ppbv; f₄₄: fraction of m/z 44 in organic mass spectra.

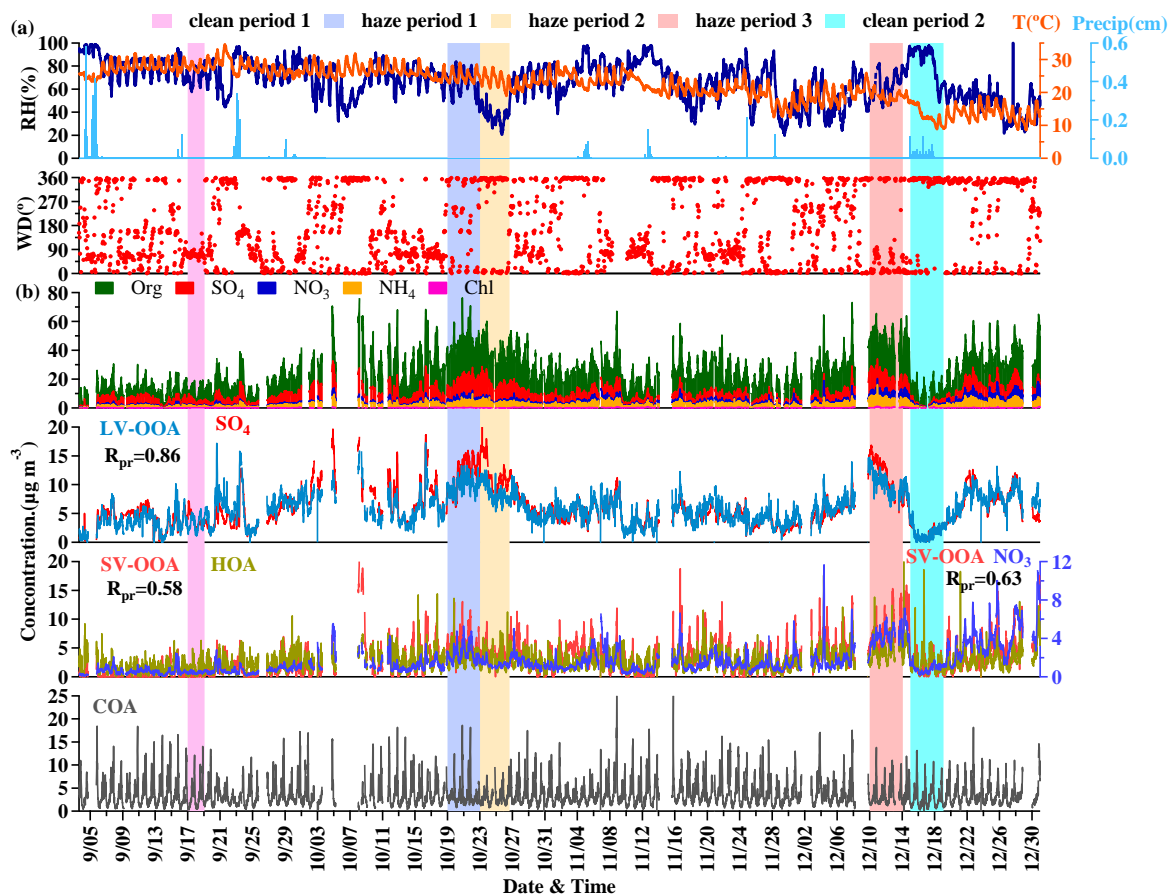
499

500 **Figures**

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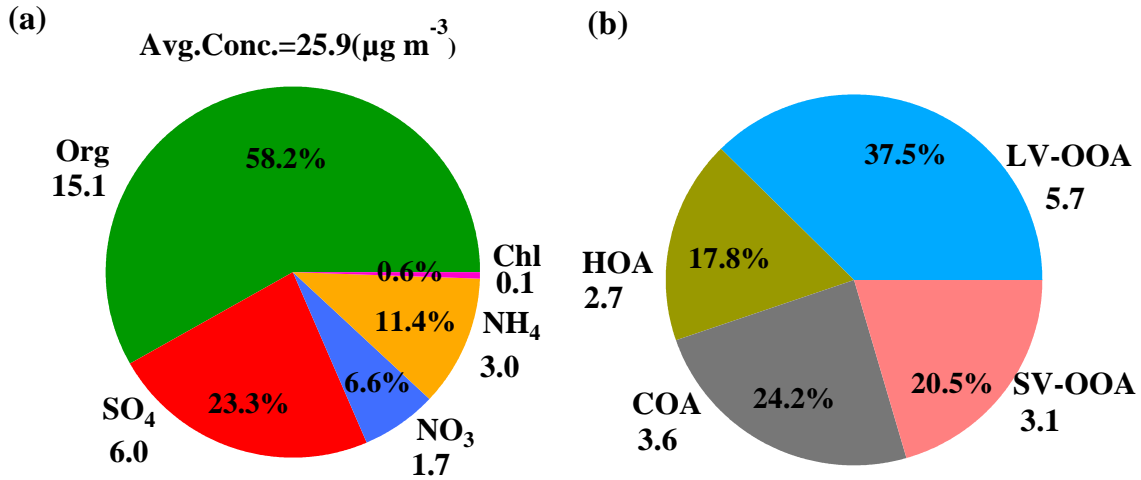
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506 **Fig. 1.** Overview of temporal variation of (a) meteorological factors (Relative Humidity, Temperature and Precipitation) and507 (b) stacked plot of non-refractory PM₁ species (Org, SO₄, NO₃, NH₄ and Chl) and non-stacked plot of organic aerosol

508 components (LV-OOA, SV-OOA, HOA and COA). Five periods: clean period 1 (C1), haze period 1 (H1), haze period 2

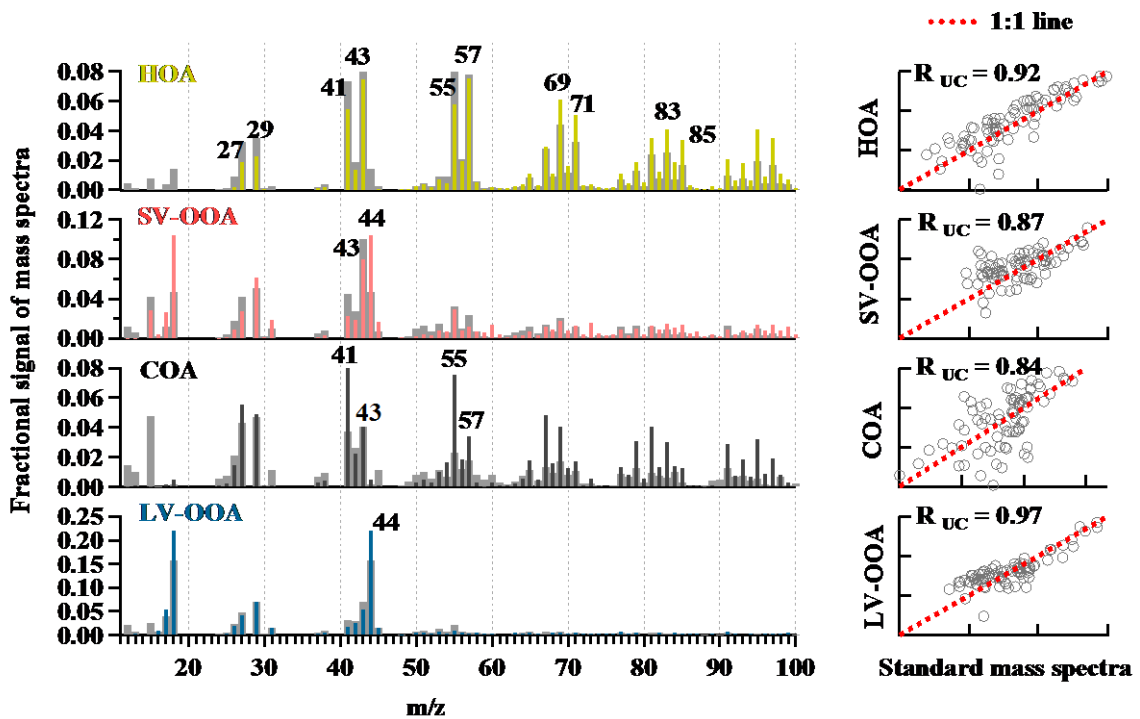
509 (H2), haze period 3 (H3) and clean period 2 (C2) are highlighted.



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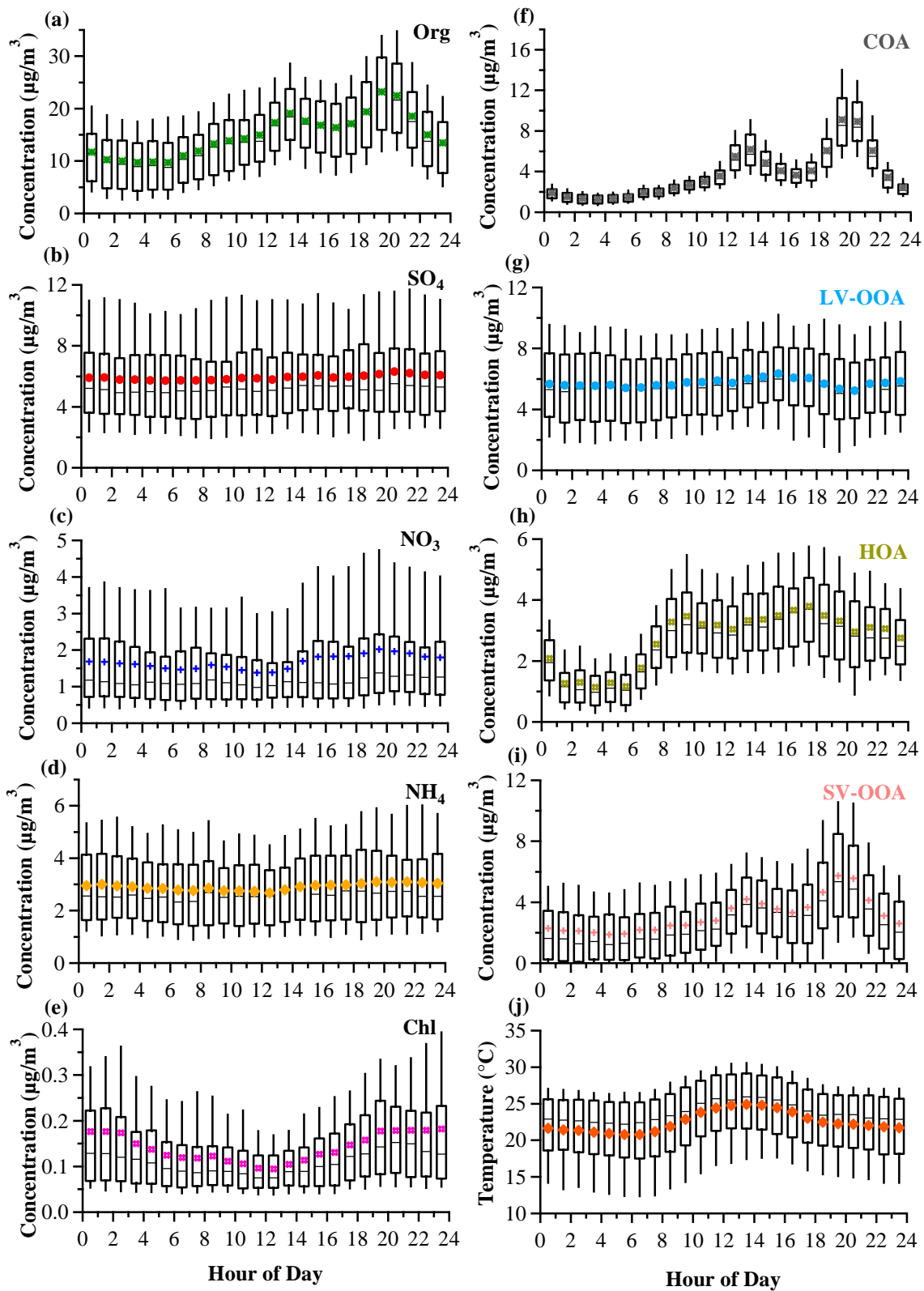
511 **Fig. 2.** Average concentration and chemical composition of (a) NR-PM₁ (Org, SO₄, NO₃, NH₄ and Chl) and (b) organic

512 aerosol (LV-OOA, SV-OOA, HOA and COA).



513

514 **Fig. 3.** Mass spectra of resolved OA components (HOA, SV-OOA, LV-OOA, COA) with the corresponding standard spectra
 515 (in gray) and the correlation with standard mass spectral profiles available on the AMS MS database (Ulbrich, I. M., Lechner,
 516 M., and Jimenez, J. L., AMS Spectral Database). The x and y axes in the right-hand graphs are mass spectra of resolved factor
 517 and the standard, respectively.

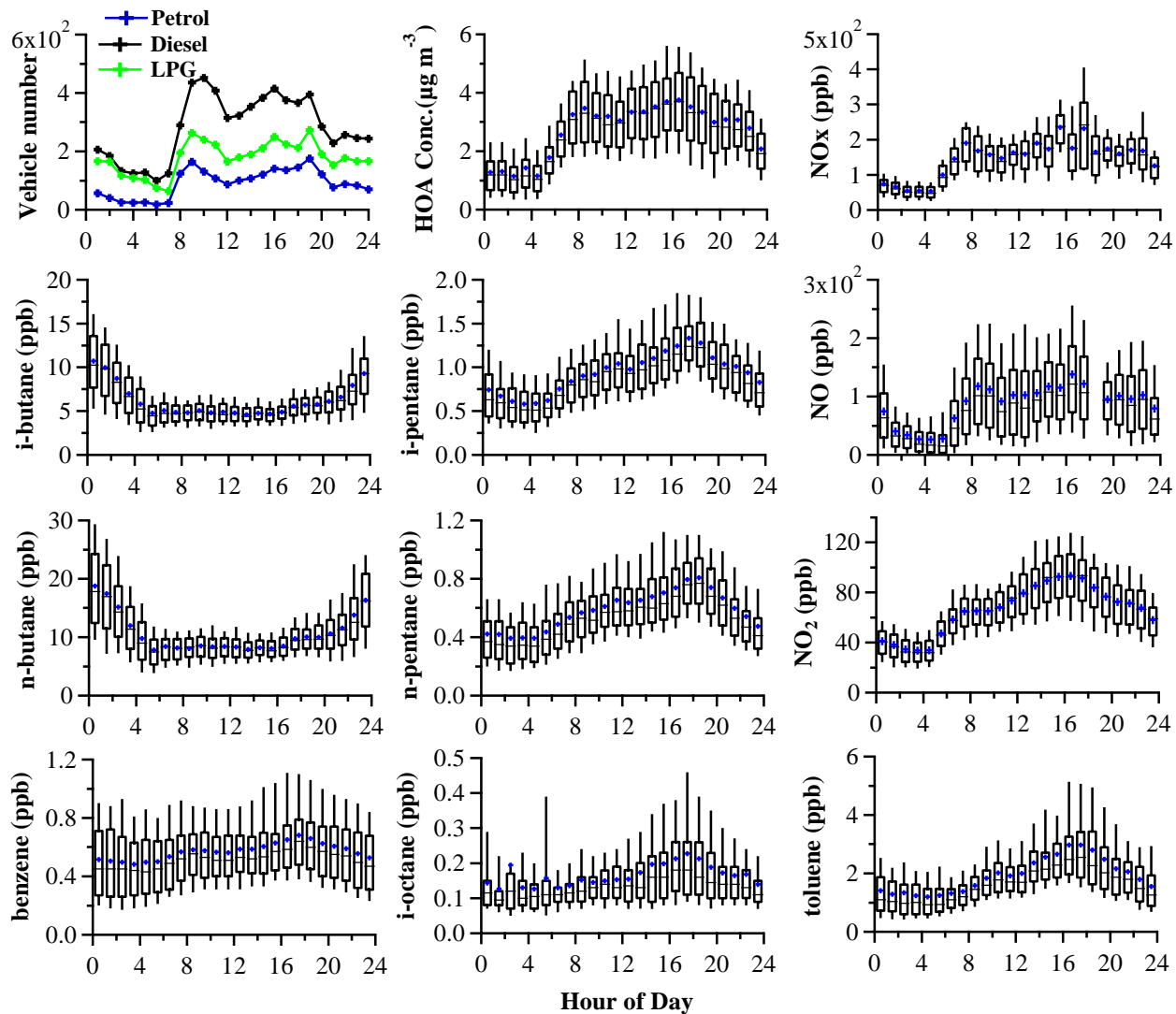


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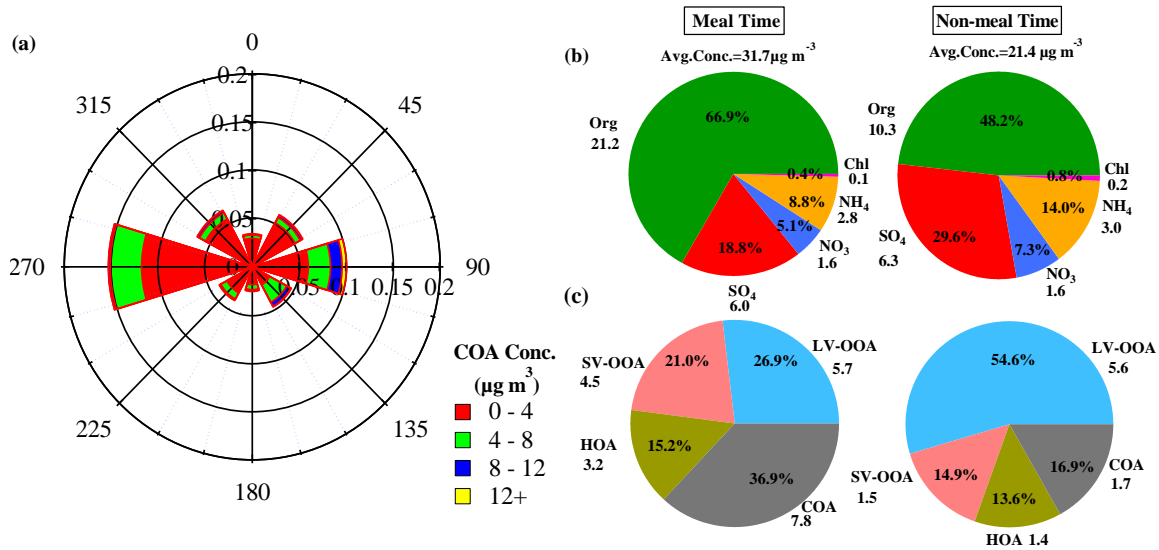
Fig. 4 Diurnal profiles of NR-PM₁ species, OA components and Temperature for the entire study with 25th and 75th percentile boxes, 10th and 90th percentile whiskers, mean as colored marker and median as black line in the whisker box.



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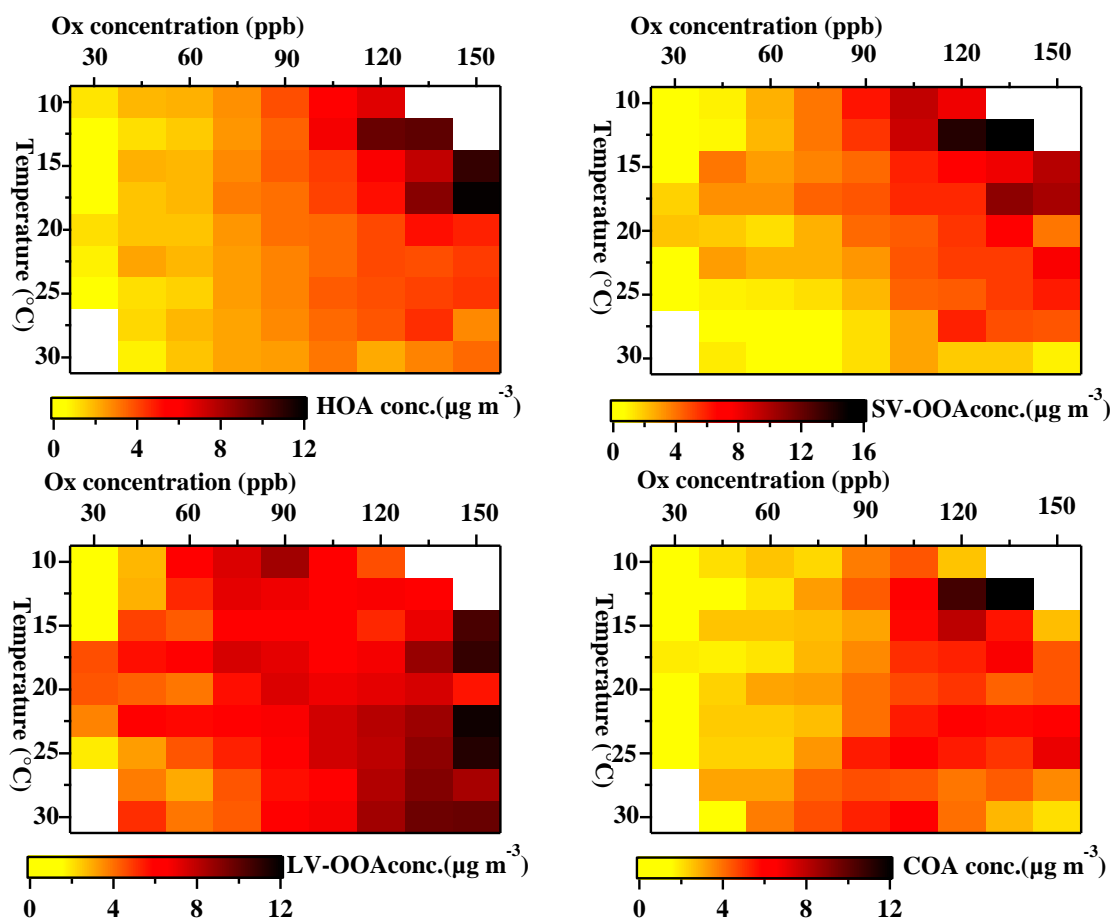
522 **Fig. 5.** Diurnal patterns of vehicle numbers at the Mong Kok site in 28 -31 May 2013 and concentrations of HOA, NO_x,
 523 NO₂, NO, i-pentane, n-pentane, i-octane, i-butane, n-butane, benzene and toluene during the whole study.

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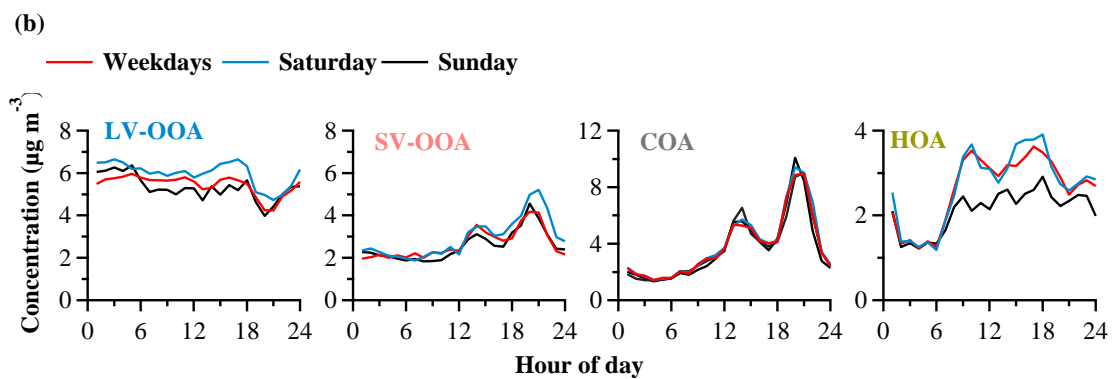
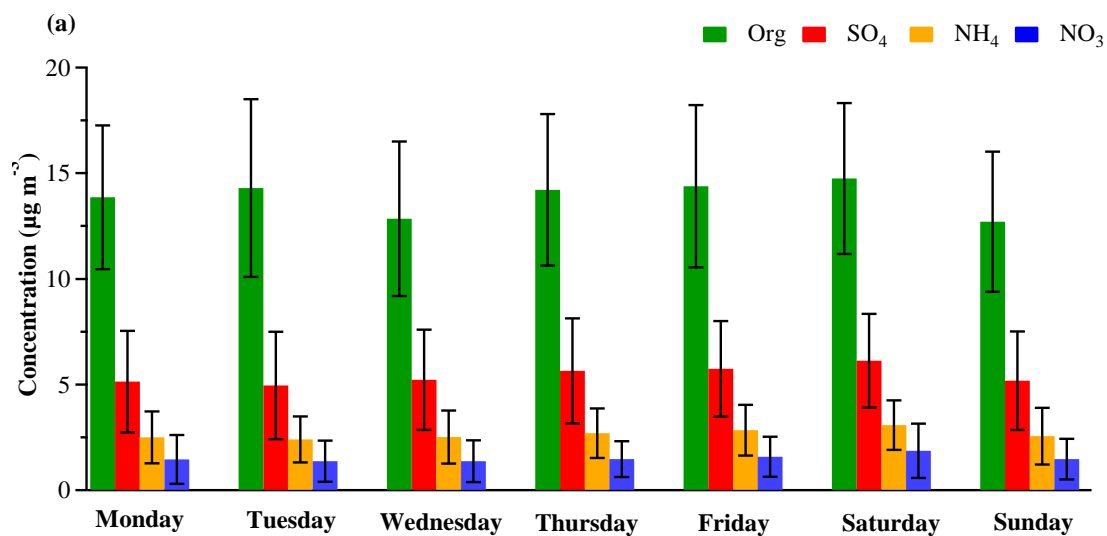
Fig. 6. (a) Wind rose plot of COA concentration. The angle and radius represent the wind direction and its probability, respectively, while color indicates COA concentration. (b) The fractional composition of NR-PM₁ species during meal times (12:00-2:00, 19:00-21:00) and non-meal time (0:00-6:00). (c) The fractional composition of OA during meal times and non-meal time, respectively.



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Fig. 7. Variation of the average concentration of OA components (HOA, SV-OOA, LV-OOA and COA) coded by color as a function of binned O_x concentration (ppb) and binned temperature (°C).

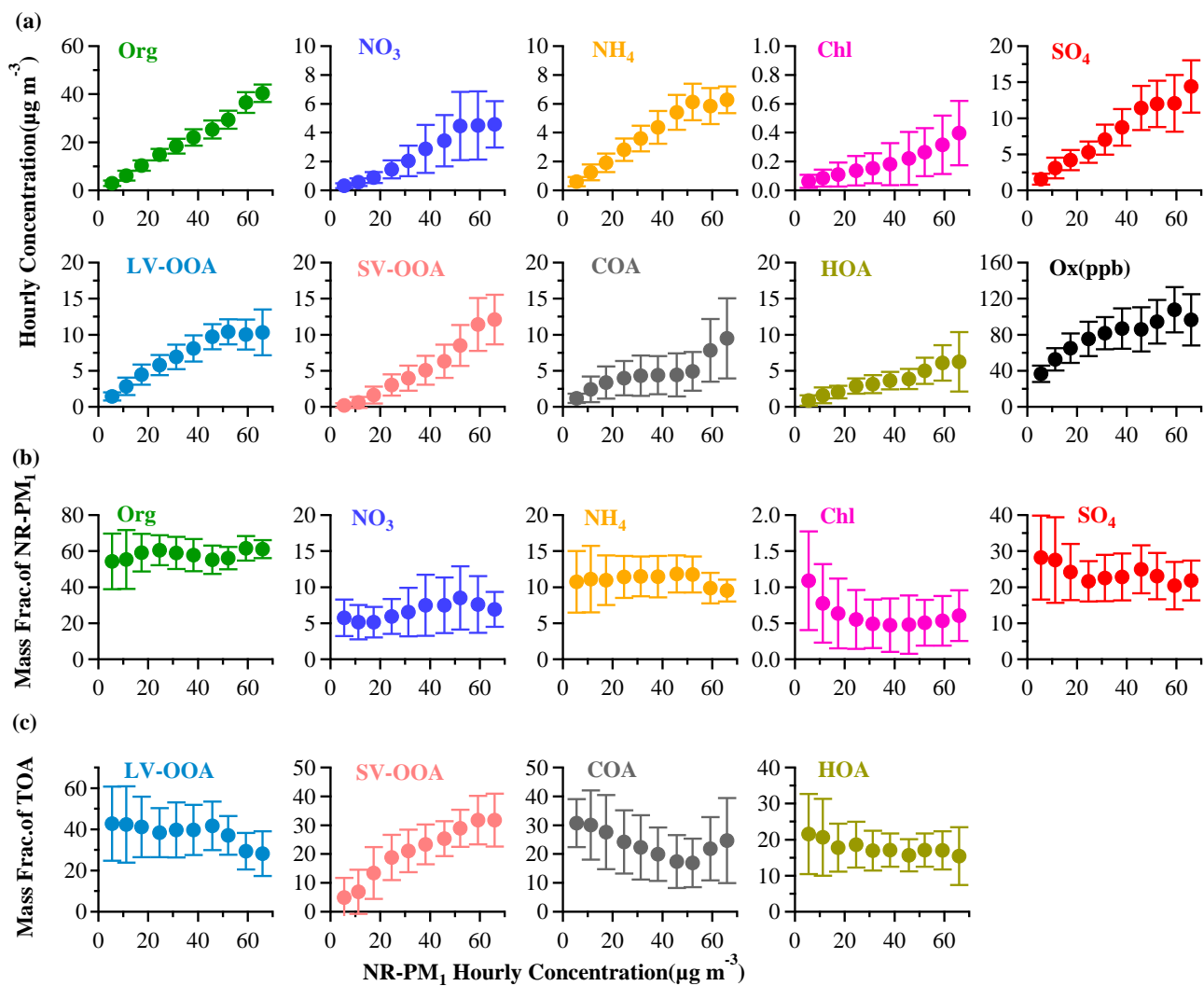


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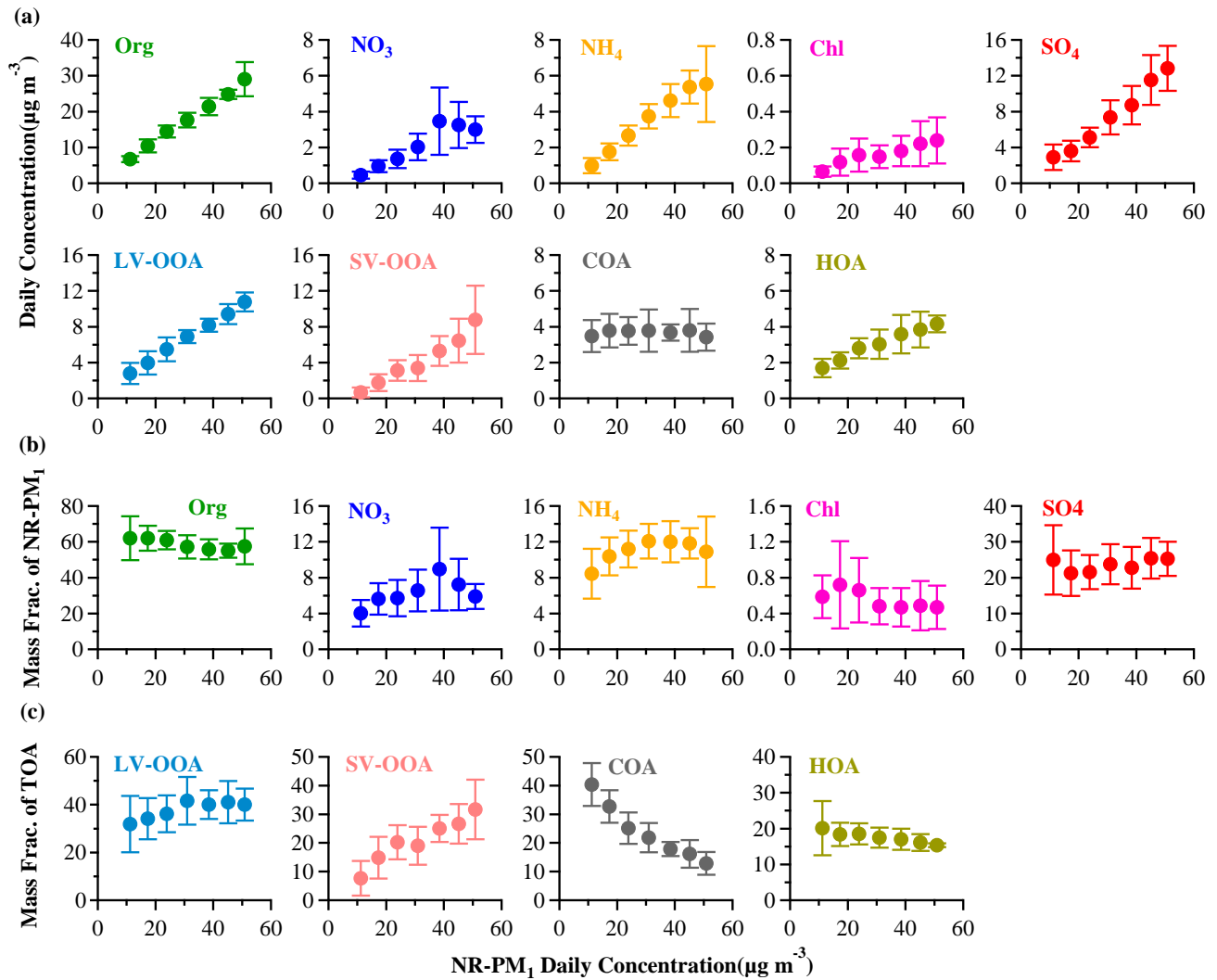
Fig. 8 (a) Day-of-week variations of NR-PM₁ species (standard deviation as vertical line) and (b) average diurnal patterns of OA components for weekdays, Saturdays and Sundays.



537

538 **Fig. 9.** (a) Variation in mass concentration of NR-PM₁ species and OA components as a function of total NR-PM₁ mass loading,
 539 and (b) mass fraction of total NR-PM₁ for NR-PM₁ species, and (c) mass fraction of total organics for OA components, as a
 540 function of total NR-PM₁ mass loading. All the mass concentrations and fractions of above species were sorted according to
 541 the hourly average NR-PM₁ mass in ascending order. The solid circles represent the average value for each concentration bin
 542 with a width of 7 μg/m³, and the vertical lines represent the standard deviations.

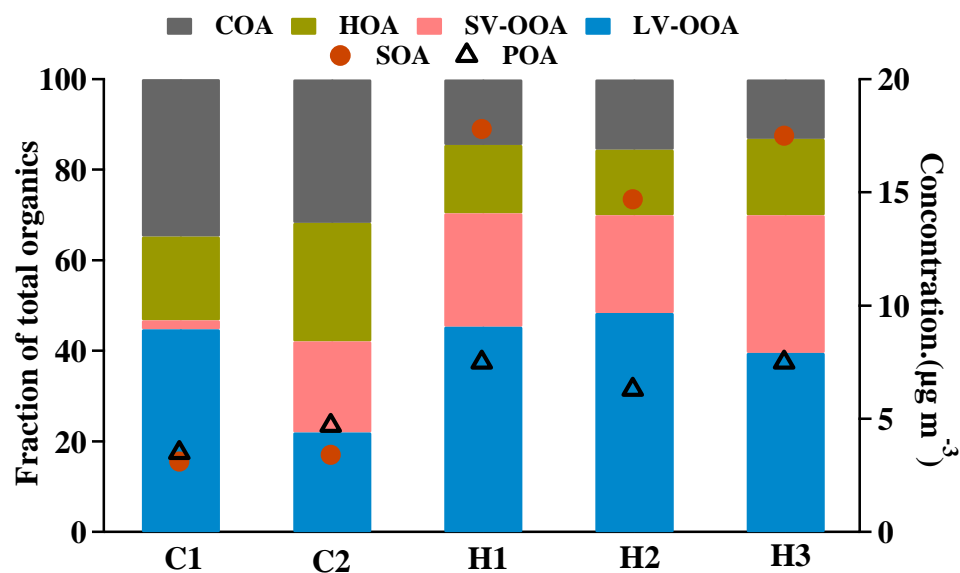
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544

545 **Fig. 10.** (a) Variation in mass concentration of NR-PM₁ species and OA components as a function of total NR-PM₁ mass
 546 loading, and (b) mass fraction of total NR-PM₁ for NR-PM₁ species, and (c) mass fraction of total organics for OA
 547 components, as a function of total NR-PM₁ mass loading. All the mass concentrations and fractions of above species were
 548 sorted according to the daily average NR-PM₁ mass in ascending order. The solid circles represent the average values for

549 each concentration bin with a width of $7 \mu\text{g}/\text{m}^3$, and the vertical lines represent the standard deviations.



550

551 **Fig.11.** Mass fraction of hydrocarbon-like organic aerosol (HOA), cooking organic aerosol (COA), semi-volatile oxygenated
552 organic aerosol (SV-OOA) and low-volatility oxygenated organic aerosol (LV-OOA) in color, and the mass concentration of
553 POA and SOA marked by triangles and circles, respectively, during five periods: clean periods (C1 and C2), and haze periods
554 (H1, H2 and H3)

555

556 Acknowledgements

557 The Aerodyne Aerosol Chemical Speciation Monitor measurements were part of the Hong Kong
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