

1 Chemical Characterization of Submicron Aerosol and Particle Growth
2 Events at a National Background Site (3295 m a.s.l.) on the Tibetan
3 Plateau

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19 **Abstract**

20 Atmospheric aerosols exert highly uncertain impacts on radiative forcing and also
21 have detrimental effects on human health. While aerosol particles are widely
22 characterized in megacities in China, aerosol composition, sources and particle
23 growth in rural areas in the Tibetan Plateau remain less understood. Here we present
24 the results from an autumn study that was conducted from 5 September to 15 October
25 2013 at a national background monitoring station (3295 m a.s.l.) in the Tibetan
26 Plateau. The submicron aerosol composition and particle number size distributions
27 were measured in situ with an Aerodyne Aerosol Chemical Speciation Monitor
28 (ACSM) and a Scanning Mobility Particle Sizer (SMPS). The average mass
29 concentration of submicron aerosol (PM_{10}) is $11.4 \mu\text{g m}^{-3}$ (range: $1.0 - 78.4 \mu\text{g m}^{-3}$) for
30 the entire study, which is much lower than those observed at urban and rural sites in
31 eastern China. Organics dominated PM_{10} on average accounting for 43%, followed by
32 sulfate (28%) and ammonium (11%). Positive matrix factorization analysis of ACSM
33 organic aerosol (OA) mass spectra identified an oxygenated OA (OOA) and a biomass
34 burning OA (BBOA). The OOA dominated OA composition accounting for 85% on
35 average, 17% of which was inferred from aged BBOA. The BBOA contributed a
36 considerable fraction of OA (15%) due to the burning of cow dung and straws in
37 September. New particle formation and growth events were frequently observed (80%
38 of time) throughout the study. The average particle growth rate is 2.0 nm hr^{-1} (range:
39 $0.8 - 3.2 \text{ nm hr}^{-1}$). By linking the evolution of particle number size distribution to
40 aerosol composition, we found an elevated contribution of organics during particle
41 growth periods and also a positive relationship between the growth rate and the
42 fraction of OOA in OA, which potentially indicates an important role of organics in
43 particle growth in the Tibetan Plateau.

44

45 **Keywords**

46 Tibetan Plateau; ACSM; Submicron Aerosol; OOA; BBOA; Particle Growth

47 **1 Introduction**

48 High concentration of atmosphere aerosol associated with the rapid economic
49 growth, urbanization and industrialization has become a major environmental concern
50 in China. Aerosol particles especially fine particles ($PM_{2.5}$) have large impacts on
51 human health, natural ecosystem, weather and climate, radiative balance and the
52 self-purification capacity of troposphere (Jacobson, 2001; Tie and Cao, 2009). As a
53 result, a large number of studies have been conducted to investigate the sources,
54 chemical and physical properties, and evolution processes of aerosol particles at urban
55 and rural sites in China during the last decade (Cao et al., 2007; Wu et al., 2007; He et
56 al., 2011; Gong et al., 2012; Huang et al., 2012; Huang et al., 2013; Sun et al., 2013).
57 The results showed that fine particles are mainly composed of organics, sulfate,
58 nitrate, ammonium, mineral dust, and black carbon. The sources of organic aerosol
59 (OA) were also characterized and various OA factors from distinct sources were
60 identified including primary OA (POA), e.g., hydrocarbon-like OA (HOA), cooking
61 OA (COA), biomass burning OA (BBOA) and coal combustion OA (CCOA), and
62 secondary OA (SOA), e.g., semi-volatile oxygenated OA (SV-OOA) and low
63 volatility OOA (LV-OOA) (Huang et al., 2010; Sun et al., 2010; He et al., 2011;
64 Huang et al., 2011; Xu et al., 2014a). While previous studies significantly improve our
65 understanding on the sources and chemical properties of aerosol particles, they were
66 mainly conducted in developed areas in China, including Beijing-Tianjin-Hebei, Pearl
67 River Delta and Yangtze River Delta.

68 The Tibetan Plateau (~ 2,000,000 square kilometers) is the highest plateau in the
69 world with an average altitude of over 4000 meters above sea level. The Tibetan
70 Plateau is an ideal location for charactering rural and regional background aerosol due
71 to minor influences of anthropogenic activities. However, chemical characterization
72 of aerosol particles in the Tibetan Plateau is rather limited, and therefore their sources,
73 properties, and evolution processes are poorly known. Cong et al. (2015) reported the
74 seasonal variations of various aerosol components including carbonaceous species
75 and water-soluble ionic species on the south edge of the Tibetan Plateau. Sulfate was
76 found to dominate the total ionic mass (25%) followed by nitrate. In addition, most

77 aerosol species showed pronounced season variations in the pre-monsoon period due
78 to biomass burning impacts from India and Nepal. Zhao et al. (2013) also
79 characterized the chemical composition and sources of total suspended particulate
80 (TSP) at Lulang on the southeastern TP based on one year measurement. Similar
81 seasonal variations with higher concentrations during pre-monsoon were observed.
82 The back trajectory analysis showed evident transport of air pollutants from south
83 Asia to the TP. The analysis of size-segregated aerosol samples collected at a remote
84 site in the inland Tibetan Plateau during 2012 further confirmed the high
85 concentrations of organic carbon (OC) and elemental carbon (EC) during the
86 pre-monsoon period (Wan et al., 2015), although their concentrations in PM₁ (2.38
87 and 0.08 $\mu\text{g m}^{-3}$, respectively) were much lower than those reported in eastern China.
88 Most studies above were conducted in the southeastern Tibetan Plateau.
89 Comparatively, aerosol particles showed quite different behavior in the northeastern
90 Tibetan Plateau. Li et al. (2013) investigated the sources and chemical composition of
91 fine particles collected at a remote site (Qinghai Lake) in the summer of 2010 in the
92 Tibetan Plateau. The average PM_{2.5} concentration was $22\pm 13 \mu\text{g m}^{-3}$ with sulfate and
93 carbonaceous aerosol being the two major species. Xu et al. (2014b) conducted a
94 year-long measurement of PM_{2.5} composition at the Qilian Shan Station, The annual
95 average concentration of PM_{2.5} was $9.5 \pm 5.4 \mu\text{g m}^{-3}$ with water-soluble ions
96 accounting for 39% of total mass. Water-soluble ions were dominated by sulfate (39%)
97 and showed pronounced seasonal variations. The aerosol composition, size
98 distributions, and back trajectory analysis together indicated a mixed impact of both
99 mineral dust from arid areas of northwest China and anthropogenic emissions from
100 urban areas. However, previous extensive efforts to characterize the chemical
101 properties of aerosol particles in the Tibetan Plateau heavily rely on filter
102 measurements with the duration ranging from days to weeks, real-time measurement
103 of aerosol particle composition is still very limited. A recent study by Xu et al. (2014a)
104 deployed a high-resolution time-of-flight aerosol mass spectrometer (HR-ToF-AMS)
105 at an urban site in Lanzhou in northwest China. The submicron aerosol in the city was
106 dominated by organic aerosol (47%) with a large contribution from local traffic and

107 cooking emissions (40%). To our knowledge, there is no such real-time measurement
108 of aerosol particle composition with aerosol mass spectrometer at rural sites in the
109 Tibetan Plateau yet.

110 The study of new particle formation and growth events in the Tibetan Plateau is
111 also relatively new. Since 2004, a number of studies have been conducted to
112 investigate the new particle formation (NPF) and particle growth events in various
113 environments in China (Wu et al., 2007; Wiedensohler et al., 2009; Yue et al., 2010;
114 Zhang et al., 2011b; Wang et al., 2013a; Wang et al., 2013b). The NPF events were
115 frequently observed in urban cities, rural sites, coastal regions, and mountain sites.
116 Sulfuric acid was found to play a dominant role in both NPF and subsequent particle
117 growth, while organics makes an important contributor to particle growth (Yue et al.,
118 2010). The particle growth rates varied largely depending on sites and days, yet
119 generally fell within 1 – 20 nm hr⁻¹. Kivekas et al. (2009) conducted a long-term
120 measurement of particle number size distributions at Waliguan, a regional background
121 site located approximately 140 km southwest of our sampling site. The annual
122 average particle number concentration was found to be higher than other rural sites in
123 the world. Despite this, the particle growth and its relationship to chemical species in
124 the Tibetan Plateau are rarely investigated and remain poorly understood.

125 In this study, an Aerodyne Aerosol Chemical Speciation Monitor (ACSM) was
126 first deployed at a national background monitoring site (Menyuan, Qinghai) in the
127 Tibetan Plateau for the real-time characterization of submicron aerosol composition
128 including organics, sulfate, nitrate, ammonium, and chloride from 5 September to 15
129 October, 2013. Collocated measurements including black carbon and particle number
130 size distributions were also conducted at the same site. Here we report the aerosol
131 composition and variations of submicron aerosols and investigate the sources of
132 organic aerosol with positive matrix factorization (PMF). In addition, the particle
133 growth events are also characterized and the roles of chemical species in particle
134 growth are elucidated.

135

136 **2 Experimental method**

137 2.1 Sampling site

138 The sampling site, i.e., the national atmospheric background monitoring station
139 (NBS) (37°36'30"N, 101°15'26"E, 3295 m a.s.l.) is located on the Daban Mountain in
140 Menyuan, Qinghai province (Fig. 1). The sampling site is characterized by a typical
141 Plateau continental climate with a pleasantly cool and short summer, and a long cold
142 winter. The annual average temperature is $-1\sim-2^{\circ}\text{C}$, and the precipitation is 426 - 860
143 mm. In this study, ambient temperature averaged 4.9°C ($-8.7 - 17.9^{\circ}\text{C}$) and wind
144 speed varied largely with an average value of 3 m s^{-1} . In addition, several precipitation
145 events were also observed, particularly during the first half period of this study (Fig.
146 3). The diurnal profiles of meteorological conditions including temperature, relative
147 humidity, wind speed, and wind direction are shown in Fig. S1. The sampling site is
148 relatively pristine with most areas covered by typical Tibetan Plateau plants, e.g.,
149 *potentilla fruticosa* and *kobresia* etc. There are no strong local anthropogenic source
150 emissions in this area ($\sim 741\text{ km}^2$ with a population of ~ 2000) except occasional
151 biomass burning events that were observed during this study. The capital city Xining
152 of Qinghai province with a population of 2,290,000 is approximately 160 km south of
153 the sampling site which is connected by a national road G227 with few traffic
154 vehicles.

155 2.2 Instrumentation

156 The field measurements were conducted from 5 September to 15 October 2013.
157 All the instruments were placed in an air-conditioned room with the temperature
158 maintaining at $\sim 23^{\circ}\text{C}$. The chemical compositions of non-refractory submicron
159 aerosol (NR-PM₁) species including organics (Org), sulfate (SO₄), nitrate (NO₃),
160 ammonium (NH₄) and chloride (Cl) were measured *in-situ* by an Aerodyne ACSM
161 (Ng et al., 2011b). A PM_{2.5} cyclone (Model: URG-2000-30ED) was supplied in front
162 of the sampling line to remove coarse particles larger than $2.5\text{ }\mu\text{m}$. The ambient air
163 was drawn inside the room through a 1/2 inch (outer diameter) stainless steel tube
164 using an external pump (flow rate is $\sim 3\text{ L min}^{-1}$). The sampling height is
165 approximately 2 m, and the particle residence time in the sampling tube is $\sim 5\text{ s}$. A
166 silica gel diffusion dryer was then used to dry aerosol particles before sampling into

167 the ACSM. After passing through a 100 μm critical orifice, aerosol particles between
168 30 nm – 1 μm are focused into a narrow particle beam via the aerodynamic lens in the
169 vacuum chamber, and then flash vaporized and ionized at a heated surface ($\sim 600^\circ\text{C}$).
170 The positive ions generated are finally analyzed by a commercial quadrupole mass
171 spectrometer. In this study, the mass spectrometer of ACSM was operated at a
172 scanning rate of 500 ms amu^{-1} from m/z 10 to 150. The time resolution is
173 approximately 15 min by alternating 6 cycles between the ambient air and
174 particle-free air. The detailed operation of ACSM has been given in Sun et al.(2012).

175 In addition to ACSM measurements, a Scanning Mobility Particle Sizer (TSI,
176 3936) equipped with a long Differential Mobility Analyzer (DMA) was
177 simultaneously operated to measure the particle number size distributions between
178 11.8 nm – 478.3 nm at a time resolution of 5 min. Other collocated species included
179 CO, O₃, NO_x, and SO₂ by various gas analyzers from Thermo Scientific and black
180 carbon (BC) by an Aethalometer (AE31, Magee Scientific Corp.). The meteorological
181 parameters, e.g., temperature, relative humidity, pressure, visibility, precipitation,
182 wind speed and wind direct were also recorded at the same site. All the data are
183 reported with ambient conditions at Beijing Standard Time.

184 **2.3 Data analysis**

185 The ACSM data were analyzed within Igor Pro (WaveMetrics, Inc., Oregon USA)
186 using the standard ACSM data analysis software (v.1.5.3.0). The mass concentrations
187 and chemical composition of NR-PM₁ species were obtained using the default relative
188 ionization efficiency (RIE) that is 1.4, 1.2, 1.1 and 1.3 for organics, sulfate, nitrate and
189 chloride, respectively, except ammonium (6.5) that was derived from pure ammonium
190 nitrate during ionization efficiency (IE) calibration. A collection efficiency (CE) of 0.5
191 was used to account for the incomplete detection of aerosol species (Matthew et al.,
192 2008; Middlebrook et al., 2012) because aerosol particles were dry and only slightly
193 acidic, and also the mass fraction of ammonium nitrate is not high enough to affect
194 CE significantly.

195 The sources of organic aerosol were investigated by performing Positive Matrix
196 Factorization (PMF2.exe, v 4.2) on ACSM OA mass spectra (Paatero and Tapper,

197 1994; Ulbrich et al., 2009). PMF is a standard multivariate factor analysis model
198 broadly used in the field of air pollution source apportionment. The detailed PMF
199 analysis of organic aerosol from AMS measurements, including error matrix
200 preparation, data pretreatment, selections of the optimum number of factors and
201 rotational forcing parameter (FPEAK), and the evaluation of PMF solutions was given
202 in Ulbrich et al. (2009) and Zhang et al. (2011a). In this study, the organic mass
203 spectra from m/z 12 to m/z 125 were used for the PMF analysis. Because of the
204 absence of collocated measurements, the two factor solution with $f_{\text{peak}} = 0$ and
205 Q/Q_{exp} close to 1 was chosen (see Fig. S2 for the PMF diagnostic plots). The two
206 factors including a biomass burning OA (BBOA) and an oxygenated OA (OOA) were
207 identified. The two OA factors showed largely different factor profiles and time series
208 indicating their distinct sources.

209 **3 Results and discussion**

210 **3.1 Mass concentration and chemical composition of submicron aerosol**

211 Figure 2 shows a comparison of the total PM_{10} mass ($\text{NR-PM}_{10} + \text{BC}$) with that
212 determined from the SMPS measurements. Assuming spherical particles, the SMPS
213 number concentrations were converted to the mass concentrations using
214 chemically-resolved particle density that was estimated from the chemical
215 composition of PM_{10} (Salcedo et al., 2006). As shown in Fig. 2, the time series of PM_{10}
216 tracks well with that of SMPS measurements ($r^2 = 0.87$). The slope of 0.52 is likely
217 due to the limited size range of SMPS measurements (12 – 478 nm) by missing a
218 considerable fraction of large particles that ACSM can measure. The PM_{10} mass varied
219 dramatically throughout the study with hourly average concentration ranging from 1.0
220 to $78.4 \mu\text{g m}^{-3}$. The average mass concentration of PM_{10} ($\pm 1\sigma$) for the entire study is
221 $11.9 (\pm 8.5) \mu\text{g m}^{-3}$, which is $\sim 3 - 4$ times lower than those observed at rural sites in
222 China ($29.9 - 44.1 \mu\text{g m}^{-3}$) (Huang et al., 2011; Hu et al., 2013; Huang et al., 2013;
223 Zhang et al., 2014). It is also approximately twice lower than that ($24.5 \mu\text{g m}^{-3}$)
224 measured at an urban site in Lanzhou in the Tibetan Plateau (Xu et al., 2014a). While
225 the average PM_{10} mass concentration in this study is close to those observed at the
226 remote sites in Asia, e.g., Okinawa ($14.5 \mu\text{g m}^{-3}$) (Zhang et al., 2007a) and Fukue

227 (12.0 $\mu\text{g m}^{-3}$) (Takami et al., 2005) in Japan, and Jeju (8.6 $\mu\text{g m}^{-3}$) in Korea (Topping
228 et al., 2004), it is much higher than those reported at rural/remote sites in north
229 America and Europe, e.g., Chebogue (2.9 $\mu\text{g m}^{-3}$), Storm Peak (2.1 $\mu\text{g m}^{-3}$) and
230 Hyytiälä (2.0 $\mu\text{g m}^{-3}$), and even comparable to the loadings at urban sites, e.g., New
231 York City (12 $\mu\text{g m}^{-3}$), Pittsburgh (15 $\mu\text{g m}^{-3}$), Manchester (14.0 $\mu\text{g m}^{-3}$) (Zhang et al.,
232 2007a). These results suggest that the NBS is a typical rural site in Asia, yet with
233 higher background concentrations compared to those in other continents.

234 Figure 3 shows the time series of mass concentrations and mass fractions of
235 aerosol species in PM_{10} . The average PM_{10} composition is dominated by organics and
236 sulfate on average accounting for 43% and 28%, respectively. Black carbon and
237 chloride represent small fractions contributing 4.5% and 1.2%, respectively to PM_{10} .
238 As shown in Fig. 1, the aerosol composition at the NBS is largely different from that
239 observed at the urban site in the Tibetan Plateau (Xu et al., 2014a). In particular,
240 sulfate shows ~60% higher contribution, yet BC is more than twice lower than that
241 observed at the urban site (Fig. 1). Xu et al. (2014a) found that 47% of BC was from
242 local traffic emissions which well explained the higher contribution of BC at the
243 urban site. Compared to this study, the average composition of PM_{10} measured by the
244 AMS at other rural sites in China showed similar dominance of sulfate (25 – 34%)
245 except Changdao Island (19%), yet overall higher contributions of nitrate because
246 most these rural sites are close to urban areas with high NO_x emissions. The sulfate
247 contributions become more dominant (36 – 64%) at remote sites in East Asia which
248 are far away from urban areas. The increase of sulfate contribution is associated with
249 a large reduction of nitrate contribution (< 5%). Such a change in aerosol bulk
250 composition at rural/remote sites in East Asia is shown Fig. 1. Overall, organics
251 comprises the major fraction of PM_{10} , contributing approximately one third of the total
252 mass at most sites. While sulfate plays a dominant role in PM_{10} at remote sites, nitrate
253 shows the highest contribution at the rural sites in eastern China. Such compositional
254 differences illustrate the different sources of sulfate and nitrate. While sulfate is
255 dominantly from regional sources and transport, nitrate is more likely influenced by
256 anthropogenic NO_x emissions over a smaller regional areas.

257 Aerosol species also varied dramatically throughout the study. For example, the
258 organics increased rapidly from $2.9 \mu\text{g m}^{-3}$ to $77.8 \mu\text{g m}^{-3}$ in one hour on 21
259 September. While sulfate remained small variations, nitrate, chloride, and BC showed
260 similar steep increases as organics indicating strong impacts of local biomass burning
261 (Zhang et al., 2015). Rapid decreases of aerosol species due to the precipitation of
262 scavenging or wind direction change were also frequently observed. For a better
263 understanding aerosol composition under variable meteorological conditions and
264 sources, five episodes with two of them from clean periods are shown in Fig. 3d. The
265 aerosol composition varied largely among different episodes. While the average PM_{10}
266 mass concentrations during the two clean episodes are similar (3.6 and $3.8 \mu\text{g m}^{-3}$),
267 the episode of Clean2 shows much higher contribution of organics (48% vs. 34%)
268 with slightly lower sulfate (33% vs. 36%) than Clean 1, consistent with their different
269 air mass trajectories (Fig. S3). The other three episodes show $\sim 5 - 8$ times higher
270 mass concentration of PM_{10} ($17.6 - 27.2 \mu\text{g m}^{-3}$) than the two clean episodes. The Ep1
271 is dominated by organics (70%), almost twice of those during the other two episodes
272 suggesting a largely different source. The relative contributions of sulfate and
273 organics during Ep2 and Ep3 are different although the nitrate contribution is similar.
274 These results suggest that the national background site is subject to the influences of
275 air masses from different sources, some of which are enriched with OA while others
276 are dominated with aerosols mainly composed of ammonium sulfates. We also
277 noticed that the two clean periods showed overall higher contribution of sulfate and
278 lower contribution of nitrate compared to the three pollution episodes. The possible
279 reasons were likely due to that the air masses during clean periods were either from a
280 longer transport when ammonium nitrate was deposited or evaporated due to dilution
281 processes, or from less anthropogenic influenced regions with low NO_x emissions.

282 The aerosol particle acidity was evaluated using the ratio of measured NH_4^+
283 ($\text{NH}_4^+_{\text{meas}}$) to the predicted NH_4^+ ($\text{NH}_4^+_{\text{pred}} = 18 \times (2 \times \text{SO}_4 / 96 + \text{NO}_3 / 62 + \text{Cl} / 35.5)$) that
284 needs to fully neutralize sulfate, nitrate, and chloride (Zhang et al., 2007b). The
285 $\text{NH}_4^+_{\text{meas}}$ correlates tightly with $\text{NH}_4^+_{\text{pred}}$ ($r^2 = 0.95$), yielding a regression slope of
286 0.80. The results suggest that aerosol particles at the NBS are overall acidic. Similar

287 acidic particles were also observed at other rural sites in China, e.g., Jiaxing in
288 Yangtze River Delta (Huang et al., 2013), Kaiping in Pearl River Delta (Huang et al.,
289 2011), Yufa in Beijing (Takegawa et al., 2009), and Qilianshan Mountain in the
290 northeast of the Qinghai–Xizang Plateau (Xu et al., 2015). As a comparison, the
291 aerosol particles in the urban city Lanzhou in the Tibetan Plateau were overall
292 neutralized (Xu et al., 2014a). One of the explanations is that more SO₂ is oxidized to
293 sulfate during the transport while gaseous ammonia is not enough to neutralize the
294 newly formed sulfate. This is supported by the overall higher contribution of sulfate at
295 rural/remote sites than that at urban sites. Also note that the newly formed sulfate
296 particles during the frequent NPF events might also have played a role.

297 **3.2 Diurnal variations**

298 The diurnal cycles of aerosol species and PM₁ are shown in Fig. 4a. The PM₁
299 shows a pronounced diurnal cycle with the concentration ranging from 7.9 to 13.4 μg
300 m⁻³. The PM₁ shows a visible peak at noon time and then has a gradual decrease
301 reaching the minimum approximately at 16:00. After that, the PM₁ starts to build up
302 and reaches the highest level at midnight. Such a diurnal cycle is similar to those of
303 SO₂ and CO (Fig. 4d), which likely indicates that the major source of PM₁ at the NBS
304 is from regional transport. All aerosol species present similarly pronounced diurnal
305 cycles to PM₁ with the lowest concentrations occurring approximately at 16:00,
306 indicating that the diurnal cycles of aerosol species were mainly driven by the
307 dynamics of planetary boundary layer. Organics dominated PM₁ composition
308 throughout the day varying from 38% - 51%. The concentration of organics at 16:00
309 is approximately twice lower than that at midnight. Sulfate shows the largest noon
310 peak among all aerosol species, consistent with those of SO₂ and CO. The sulfate
311 contributes more than 25% to PM₁ with the highest contribution as much as 33%
312 between 12:00 – 14:00. Nitrate and chloride shows relatively stable concentrations
313 before 11:00 and then gradually decreased to low ambient levels during daytime.
314 Such diurnal variations still exist after considering the dilution effects of boundary
315 layer height using the conserved tracer CO as a reference (Fig. 4c). This indicates that
316 gas-particle partitioning affected by temperature and humidity has played an

317 important role in driving the diurnal variations of nitrate and chloride. Consistently,
318 the nitrate contribution to PM₁ during late afternoon is ~7-8% which is much lower
319 than that (> 12%) in the early morning. The diurnal variation of BC is different from
320 that observed at the urban site in the Tibetan Plateau where the pronounced morning
321 peak due traffic influences was observed (Xu et al., 2014a). In fact, BC has a good
322 correlation with secondary nitrate ($r^2 = 0.59$) indicating that BC is likely dominantly
323 from regional transport. This is also supported by the low ambient levels of NO_x (2.5
324 – 5.1 µg m⁻³). The contribution of BC to PM₁ is relatively constant, which is ~ 4 – 5%
325 throughout the day.

326 **3.3 OA composition and sources**

327 PMF analysis of ACSM OA mass spectra identified two factors, i.e., a biomass
328 burning OA (BBOA) and an oxygenated OA (OOA). The mass spectra and time
329 series of the two OA factors are shown in Fig. 5.

330 **3.3.1 BBOA**

331 The mass spectrum of BBOA resembles to that of standard BBOA ($r^2 = 0.82$)
332 which is characterized by a prominent peak of m/z 60 (1.1% of total signal), a tracer
333 m/z for biomass burning aerosols (Aiken et al., 2009; Cubison et al., 2011; Hennigan
334 et al., 2011). The fraction of m/z 60 in BBOA (1.1%) is also much higher than ~0.3%
335 in the absence of biomass burning impacts (Cubison et al., 2011). BBOA correlates
336 tightly with m/z 60 ($r^2 = 0.82$) and also chloride ($r^2 = 0.52$). The ratio of BBOA to m/z
337 60 is 55.6, which is higher than that of fresh BBOA (34.5) measured during the
338 second Fire Lab at Missoula Experiment (FLAME II) (Lee et al., 2010). One of the
339 explanations is that BBOA in the ambient is more aged because the m/z 60 related
340 levoglucosan can be rapidly oxidized in the atmosphere (Hennigan et al., 2010).
341 Indeed, Zhang et al. (2015) reported a much higher ratio of aged BBOA to m/z 60
342 (74.8) than fresh BBOA (16.8) during two harvest seasons in Nanjing, China. The
343 time series of BBOA shows periodically large peaks, particularly on the days of 21
344 and 22 September, which were mainly from the burning of a large amount of straws in
345 the south-west region. Relatively high concentration of BBOA was also observed at
346 the end of the campaign due to the burning of cow dung for heating purpose because

347 of the low temperature. The average concentration of BBOA is $0.8 (\pm 1.5) \mu\text{g m}^{-3}$ for
348 the entire study on average accounting for 15% of total OA. Although the average
349 BBOA contribution is much lower than those measured in PRD, e.g., Jiaying ($\sim 3.9 \mu\text{g}$
350 m^{-3} , 30.1%) (Huang et al., 2013), Kaiping ($\sim 1.36 \mu\text{g m}^{-3}$, 24.5%) (Huang et al., 2011),
351 and Shenzhen ($\sim 5.2 \mu\text{g m}^{-3}$, 29.5%) (He et al., 2011), the contribution of BBOA
352 during some strong BB plumes can reach up to 40%, e.g., 21-22 September,
353 indicating a large impact of biomass burning on OA at the national background site.
354 BBOA showed a pronounced diurnal cycle which is similar to that of chloride (Fig.
355 5c). The BBOA concentration increased rapidly from 18:00 and reached a maximum
356 in 2 hours, likely indicating that the burning of straws and cow dung mainly occurred
357 during this period of time. As a result, the contribution of BBOA to total OA increased
358 from $\sim 10\%$ to more than 20%.

359 **3.3.2 Oxygenated organic aerosols (OOA)**

360 Similar to previously reported OOA (Zhang et al., 2005), the mass spectrum of
361 OOA in this study is characterized by a prominent m/z 44 peak (mainly CO_2^+). The
362 mass spectrum of OOA also resembles to that of low-volatility OOA ($r^2 = 0.88$) (Ng
363 et al., 2011a), yet with higher fraction of m/z 44 (f_{44}). Much higher fraction of m/z 44
364 in ACSM OOA spectrum than that from HR-ToF-AMS was reported recently by a
365 comprehensive evaluation of the ACSM (Fröhlich et al., 2015). The results also
366 showed that f_{44} has minor impacts on the mass concentrations of OOA factors,
367 although it varies largely by a factor of 0.6 – 1.3. The average mass concentration of
368 OOA is $4.1 \mu\text{g m}^{-3}$, on average accounting for 85% of total OA. The OOA
369 contribution is much higher than those reported at urban sites in summer ($\sim 60\%$)
370 (Huang et al., 2010; Sun et al., 2012; Xu et al., 2014a), and also higher than those
371 ($\sim 70\%$) observed at rural sites in China (Hu et al., 2013; Huang et al., 2013). These
372 results suggest that organic aerosol was highly aged and well processed at the NBS. In
373 addition, aqueous-processing of OA at nighttime associated with high RH might also
374 played a role in forming the highly oxidized OA. The diurnal cycle of OOA was
375 similar to that of PM_{10} , which showed a small peak before noon time followed by a
376 subsequent decrease until 16:00. The OOA dominated OA throughout the day varying

377 from 80 – 90%, indicating that OA at the NBS was mainly composed of secondary
378 organic aerosol.

379 Previous studies have shown the ubiquitously tight correlations between sulfate
380 and highly oxidized OA because of their similar secondary nature over regional scales
381 (Zhang et al., 2005; DeCarlo et al., 2010). While the OOA correlates well with
382 secondary sulfate for most of the time in this study, several periods with largely
383 different correlations were also observed (Fig. 6). As shown in Fig. 5b and 6, the
384 weak correlation events mainly occurred during periods with strong biomass burning
385 impacts were observed. However, it cannot be resolved by extending PMF solution to
386 more than 2 factors because of the limitation of PMF technique in source
387 apportionment analysis. Similar different correlations between sulfate and LV-OOA
388 were also observed during two research flights in Mexico City and the Central
389 Mexican Plateau (DeCarlo et al., 2010). Following the approach suggested by
390 DeCarlo et al. (2010), we performed a post-processing technique with external tracers
391 on the further apportionment of OOA. We first assume that OOA and sulfate have
392 similar sources during periods in the absence of biomass burning impacts, which is
393 supported by their tight correlations ($r^2 = 0.74$). An average OOA/SO₄ ratio of 1.04,
394 i.e., (OOA/SO₄)_{NBB}, was obtained by performing a linear regression analysis on OOA
395 versus SO₄. We then assume that SO₄ is completely from non-biomass burning (NBB)
396 sources during BB-impact periods. This assumption is rational because previous
397 studies have found that fresh biomass burning emits a very small or negligible
398 fraction of sulfate (Levin et al., 2010). The sulfate-related OOA can be calculated as
399 OOA × [OOA/SO₄]_{NBB}, and the excess OOA that is from different sources is then
400 determined as:

$$401 \quad \text{OOA}_{\text{post-processed}} = \text{OOA} - \text{SO}_4 \times [\text{OOA}/\text{SO}_4]_{\text{NBB}} \quad (1)$$

402 Because the post-processed OOA shows high concentrations during BB periods, we
403 conclude that it's very likely an aged BBOA that was mixed with OOA. In fact, the
404 mass spectrum of OOA_{post-processed} is similar to that of OOA. The fraction of m/z 60 (f_{60})
405 is 0.29%, which is very close to ~0.3% for non-biomass burning organic aerosol
406 (Aiken et al., 2008). Smog chamber experiments have shown that fresh BBOA can be

407 rapidly oxidized within 3 – 4.5 hours (Hennigan et al., 2011). While f_{44} increases
408 significantly, f_{60} quickly decreases to a value close to ~0.3%. Similarly, a recent study
409 in Nanjing resolved an aged BBOA factor with its spectrum resembling to that of
410 OOA yet with much lower f_{60} (Zhang et al., 2015). The average concentration of aged
411 BBOA is $0.82 (\pm 2.65) \mu\text{g m}^{-3}$, accounting for 17% of OA for the entire study. The
412 contribution of aged BBOA is close to that of fresh BBOA, which might indicate that
413 half of BBOA has been aged. Still, the sum of fresh and aged BBOA highly correlates
414 with m/z 60 ($r^2 = 0.81$, slope = 136.1). The fresh and aged BBOA together accounted
415 for 33% of the total OA suggesting that BBOA was a large local source of OA during
416 the observational period. With the post-processing technique, the sulfate-related OOA
417 contributed 67% on average of total OA, which is close to those observed at other
418 rural sites in e.g., Kaiping (Huang et al., 2011) and Changdao (Hu et al., 2013).

419 **3.4 Chemistry of particle growth**

420 Figure 7a shows the evolution of size distributions of particle number
421 concentrations for the entire study. New particle formation and growth events (NPE)
422 were observed almost every day (27 days in 34 days). Most NPE started at ~11:00
423 (The time of sunrise is 2 hours behind of Beijing standard time) and persisted more
424 than half day except some NPE were interrupted by either precipitation events or
425 strong winds. The average particle number size distributions during NPE and non
426 event days (non-NPE) are shown in Fig. 7b. Both NPE and non-NPE show broad size
427 distributions with higher number concentrations occurring during NPE. Three modes
428 with geometric mean diameter (GMD) peaking at 28 nm, 43 nm, and 104 nm,
429 respectively were resolved using a log-normal distribution fitting (Seinfeld and Pandis,
430 2006). The largest mode (104 nm) dominated the total number of particles accounting
431 for ~70%. In contrast, the average size distribution during non-NPE was characterized
432 by a bi-modal distribution with the GMD peaking at 59 nm and 146 nm, respectively.
433 The peak diameters were shifted to the larger sizes compared to those during NPE.
434 Such a size shift from clean days to polluted days was also observed previously in
435 Beijing (Yue et al., 2010). Also, the two modes showed almost equivalent
436 contributions to the total number of particles. The average particle number

437 concentration for the entire study is $2.4 \times 10^3 \text{ cm}^{-3}$, which is nearly an order of
438 magnitude lower than those reported at rural sites in eastern China (Wu et al., 2007),
439 but close to that ($2.03 \times 10^3 \text{ cm}^{-3}$) observed at Mount Waliguan which is a remote site
440 located nearby (Kivekas et al., 2009). The particle size was further segregated into
441 small Aitken mode (20 – 40 nm, N_{20-40}), large Aitken mode (40 – 100 nm, N_{40-100}),
442 and Accumulation mode (100 – 470 nm, N_{Accu}) particles. The time series and diurnal
443 cycles of particle numbers for three different sizes are shown in Fig. 7c, d. The N_{20-40}
444 presented sharp peaks almost in everyday corresponding to new particle formation
445 events. The diurnal cycle of N_{20-40} showed that the number concentration started to
446 increase at approximately 11:00 (150 cm^{-3}) and reached a maximum at 14:00 (770
447 cm^{-3}). In contrast, the N_{40-100} and N_{Accu} showed largely different diurnal cycles from
448 that of N_{20-40} , indicating their different sources. In fact, the diurnal cycles of N_{40-100}
449 and N_{Accu} are remarkably similar to those of aerosol species, suggesting that the large
450 particles are more likely from regional transport.

451 Figure 8 shows the diurnal evolution of particle number size distributions, aerosol
452 composition, and gaseous species during NPE and non-NPE days. The particle
453 number size distributions during NPE were characterized by distinct bimodal
454 distributions showing a persistent larger mode with the GMD peaking at ~ 100 nm,
455 and a smaller mode below 50 nm. The particle growth started at approximately 11:00
456 from ~ 20 nm, and continued to grow slowly until ~ 45 nm at mid-night. The maximum
457 size particles can grow in this study is generally smaller than those ($\sim 60 - 70$ nm)
458 observed at urban and rural sites in Beijing (Wang et al., 2013a), which is likely due
459 to the much lower concentrations of aerosol species and precursors. All aerosol
460 species however showed decreases during the particle growth period between 12:00 –
461 17:00, and the gaseous CO and SO₂ showed similar variations as aerosol species. By
462 excluding the dilution effect of PBL using CO as a tracer, we found that organics was
463 the only species showing a gradual increase during the particle growth period (Fig. 8a)
464 while other species remained minor changes or even slightly decreased. The
465 contribution of organics to PM₁ also showed a corresponding increase from 40% to
466 47%. These results suggest that organics might have played a dominant role in

467 particle growth at the national background site. Our conclusion is consistent with the
468 recent findings that organics, particularly oxidized organic aerosol species, play a
469 more important role than ammonium sulfate in particle growth (Dusek et al., 2010;
470 Ehn et al., 2014; Setyan et al., 2014). Also note that the contribution of organics to
471 PM₁ during NPE (~40 – 50%) is overall higher than that during non-NPE (~30 –
472 40%), while the sulfate contribution is correspondingly lower (~20 – 30% vs. 30 –
473 40%), which further supports the important role of organics during NPE. The particle
474 growth was mixed with anthropogenic sources from 17:00 which are indicated by
475 synchronous enhancements of both aerosol species and gaseous precursors. One of
476 possible reasons is due to the air mass transport from downwind urban areas.

477 The diurnal evolution of particle size distributions and aerosol composition
478 during non-NPE is largely different from that during NPE. The particle number size
479 distributions and mass concentrations of aerosol species showed a dramatic variation
480 at noon time (12:00), indicating a very different chemical and/or physical process
481 between the first and the second half day. The aerosol particles showed an evident
482 growth from ~50 nm to 60 nm during the first 6 hours, which is likely a continuation
483 of previous NPE. Compared to the early stage of particle growth during NPE, the
484 particle growth during non-NPE is associated with synchronous increases of both
485 organics and sulfate. The results indicate that both organics and sulfate contribute to
486 the particle growth after mixed with anthropogenic sources from ~18:00 in the
487 previous day.

488 We further calculated the particle growth rates (GR) for NPE events without
489 interruptions due to meteorological changes using Eq. (2).

$$490 \quad \text{GR} = \frac{\Delta D_m}{\Delta t} \quad (2)$$

491 where D_m is the geometric mean diameter from the log-normal fitting, ΔD_m is the
492 difference of diameter during the growth period and Δt is the duration of growth time.
493 The calculated GR and the corresponding average chemical composition and fraction
494 of OOA during the growth period are shown in Fig. 9a. The GR ranges from 0.8 nm
495 h⁻¹ to 3.2 nm h⁻¹ with an average of 2.0 nm h⁻¹. The GR in this study is overall

496 consistent with those observed at remote and/or forest sites (Eisele and McMurry,
497 1997; Weber et al., 1997), yet generally smaller than those measured at urban and
498 polluted rural sites (Yue et al., 2010; Shen et al., 2011; Zhang et al., 2011b) where
499 abundant condensable vapor and high concentrations of particulate matter facilitate
500 the growth of particles (Wang et al., 2013a). By linking GR to aerosol composition,
501 we found that GR at the background site is positively related to the fraction of
502 oxidized OA, which likely indicate the important role of oxidized secondary organic
503 aerosol in particle growth (Ehn et al., 2014). Zhang et al. (2011b) also observed a
504 tight correlation between OOA and GR in urban Beijing supporting the important role
505 of OOA in particle growth. Further investigation is needed for a better understanding
506 of the role of organic aerosol, particularly oxidized OA, in the new particle formation
507 and particle growth at the regional background site.

508 **4 Conclusions**

509 The aerosol particle composition and particle number size distributions were
510 measured at a national background monitoring station in the Tibetan Plateau (3295 m,
511 a.s.l.) from 5 September to 15 October 2013. The average mass concentration of PM₁
512 is 11.4 (\pm 8.5) $\mu\text{g m}^{-3}$ for the entire study, which is lower than those observed at urban
513 and rural sites in eastern China. Organics constituted the major fraction of PM₁, on
514 average accounting for 43% followed by sulfate (28%) and ammonium (11%).
515 Several periods with the contribution of organics as much as 70% due to biomass
516 burning impacts were also observed. All aerosol species presented similar diurnal
517 cycles that were mainly driven by the dynamics of planetary boundary layer and
518 regional transport. PMF source apportionment analysis resolved a secondary OOA
519 and a primary BBOA. OOA dominated OA composition accounting for 85% on
520 average with the rest being BBOA. A post-processing technique based on the
521 correlation of OOA and sulfate separated an aged BBOA which on average accounted
522 for 17% of OA. New particle formation and particle growth events were frequently
523 observed during this study. The particle growth rates varied from 0.8 to 3.2 nm hr⁻¹
524 with an average growth rate of 2.0 nm hr⁻¹. Organics was found to be the only species
525 with gradually increased contribution to PM₁ during NPE. Also, higher contribution

526 of organics during NPE than non-NPE days was observed. These results potentially
527 illustrate the important role of organics in particle growth. Further analysis showed a
528 positive correlation of particle growth rate with the fraction of OOA suggesting that
529 oxidized OA plays a critical role contributing to the particle growth.

530

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783 Tibetan Plateau, China: Implications for pollution transport from South Asia, *J. Geophys.*
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785 **Tables**

786 Table 1. A summary of average mass concentrations ($\mu\text{g m}^{-3}$) of PM_{10} species during
 787 five episodes and the entire study. The 30 min detection limit (DLs) of the ACSM is
 788 also shown (Sun et al., 2012).

789

	Org	SO ₄	NO ₃	NH ₄	Cl	BC	PM ₁₀
Entire Study	4.9	3.2	1.2	1.4	0.14	0.51	11.9
Clean1	1.2	1.3	0.25	0.58	0.02	0.22	3.6
Clean2	1.8	1.3	0.24	0.45	0.03	-	3.8
Ep1	19.1	2.1	2.7	1.4	0.53	1.4	27.2
Ep2	6.3	5.9	2.0	2.6	0.18	0.57	17.6
Ep3	10.8	6.0	2.9	2.6	0.39	1.0	23.7
DLs	0.54	0.07	0.06	0.25	0.03		

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792 **Figure Captions:**

793 Figure 1. Map of the sampling site (Menyuan, Qinghai). Also shown is the chemical
794 composition of submicron aerosols measured at selected rural/remote sites in East
795 Asia except Lanzhou, an urban site in northwest China. The detailed information of
796 the sampling sites is presented in Table S1.

797 Figure 2. Comparison of the mass concentrations of PM_{10} (NR- PM_{10} +BC) measured
798 by the ACSM and Aethalometer with that by the SMPS ($D_m = 12 - 478$ nm): (a) time
799 series and (b) scatter plot.

800 Figure 3. Time series of (a-c) meteorological variables including T (temperature), RH
801 (relative humidity), Precip. (precipitation), WS (wind speed), WD (wind direction),
802 and Vis (visibility), (d) mass concentrations and (e) mass fractions of PM_{10} species .
803 The pie charts show the average chemical composition of PM_{10} for five episodes.

804 Figure 4. Average diurnal cycles of (a) mass concentration; (b) mass fraction of PM_{10}
805 species; (c) ratios of aerosol species to CO, and (d) gaseous species. The local sunrise
806 and sunset was around 7:00 and 19:00, respectively.

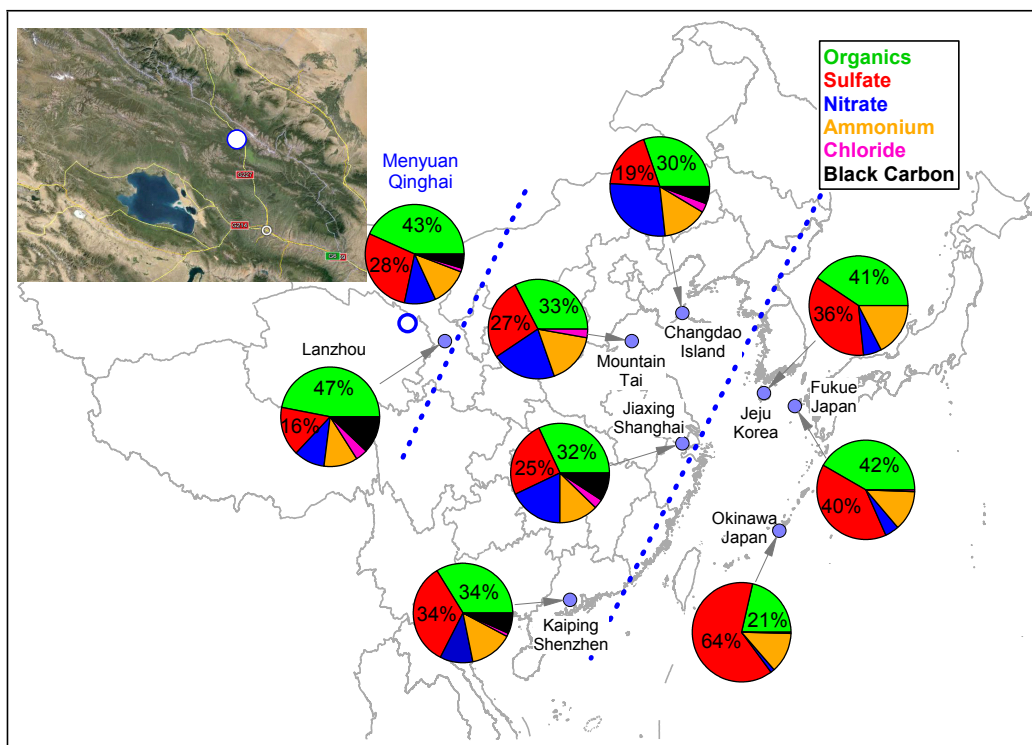
807 Figure 5. (a) Mass spectra and (b) time series of mass concentrations of BBOA and
808 OOA. The standard average mass spectra of BBOA and OOA in Ng et al. (2011) are
809 also shown for the comparison. The pie chart in (b) shows the average composition of
810 OA for the entire study.

811 Figure 6. Scatter plot of OOA versus SO_4 during BB and NBB periods. The data
812 points are color coded by BBOA concentrations. The pie chart shows the average
813 composition of OA with post-processed OOA ($= OOA - SO_4 \times [OOA/SO_4]_{NBB}$).

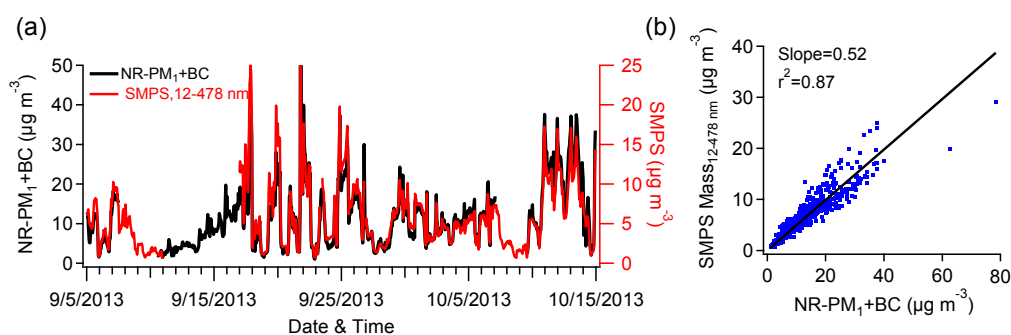
814 Figure 7. (a) The evolution of particle number size distributions; (b) average particle
815 number size distributions during NPE and non-NPE; (c,d) time series and diurnal
816 cycles of particle number concentrations for three different sizes. The log-normal
817 distribution fitting of each mode is shown in (b) as dash lines. The sunrise time was
818 around 7:00.

819 Figure 8. Diurnal evolution of particle size distributions, aerosol composition, gaseous
820 precursors, and the ratios of aerosol species to CO during (a) NPE and (b) non-NPE.
821 The sunrise time was approximately 7:00.

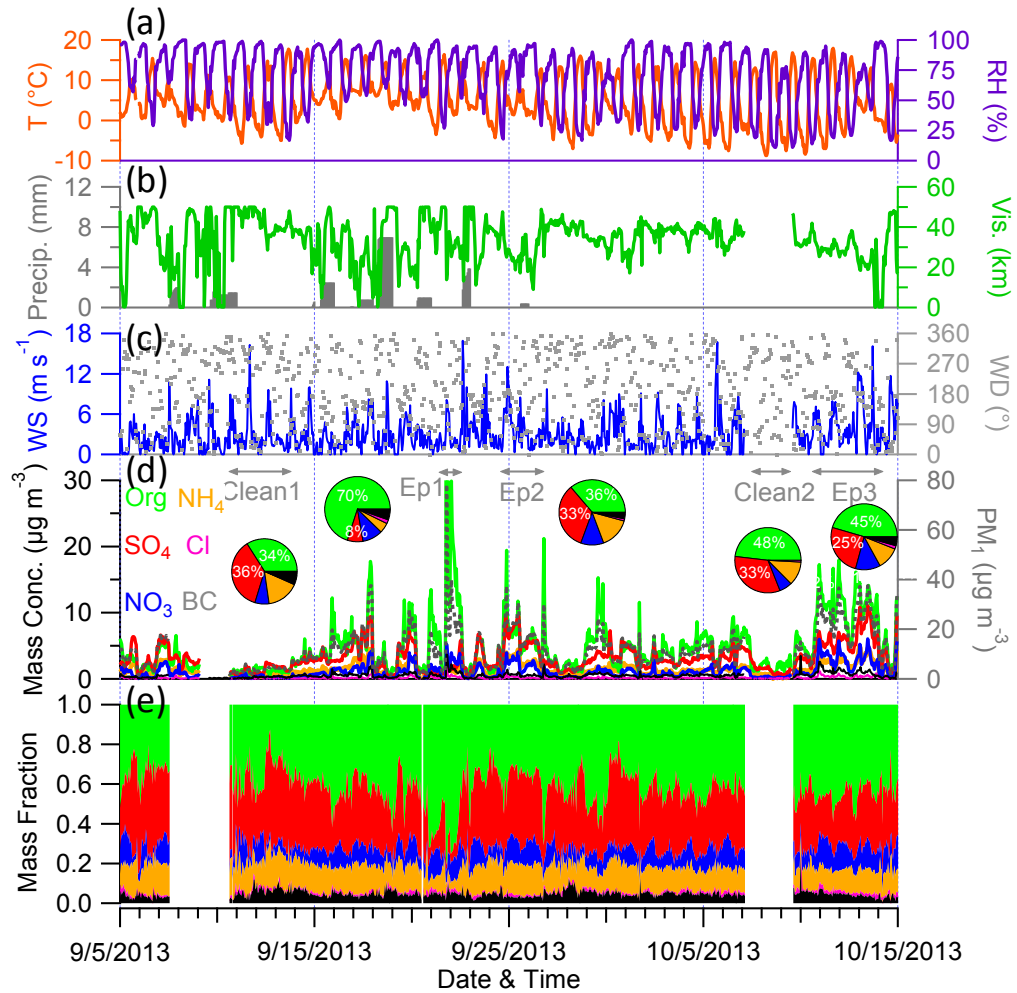
822 Figure 9. (a) Time series of OOA/ PM_{10} , particle growth rates, and average chemical
823 composition during particle growth periods; (b) correlation of growth rate with
824 OOA/ PM_{10} . The data points are color coded by the PM_{10} mass concentration.



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 826 Figure 1. Map of the sampling site (Menyuan, Qinghai). Also shown is the chemical
 827 composition of submicron aerosols measured at selected rural/remote sites in East
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 829 the sampling sites is presented in Table S1.
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 833 Figure 2. Comparison of the mass concentrations of PM₁ (NR-PM₁ +BC) measured
 834 by the ACSM and Aethalometer with that by the SMPS ($D_m = 12 - 478$ nm): (a) time
 835 series and (b) scatter plot.



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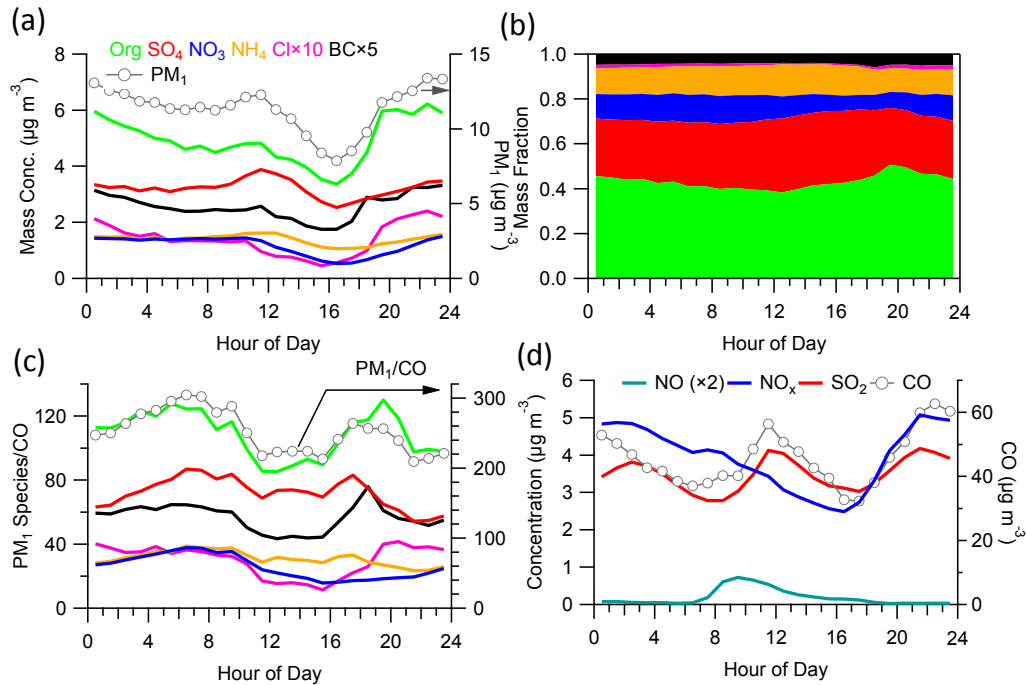
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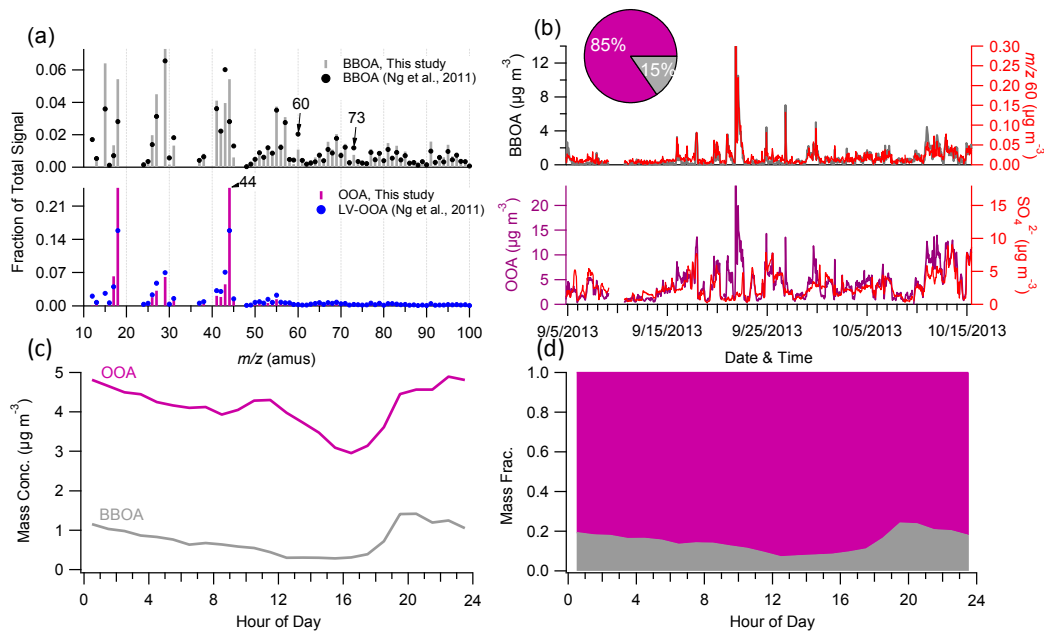
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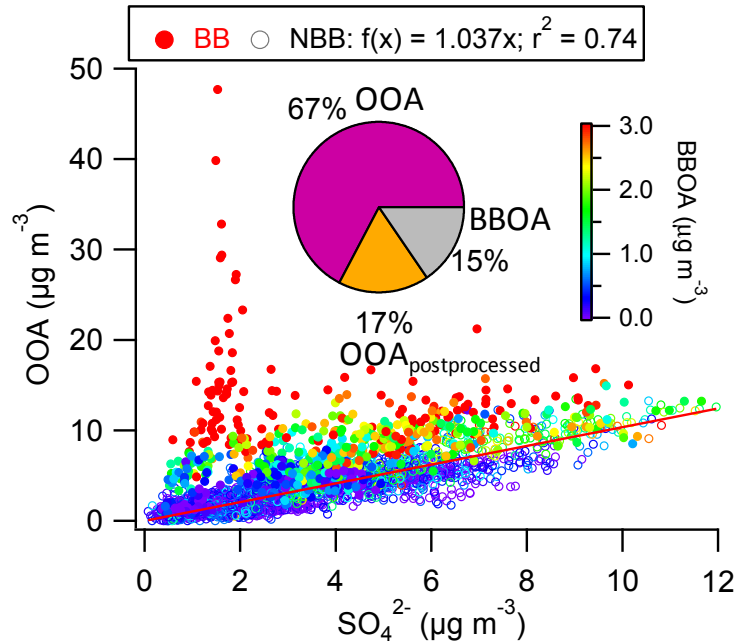
Figure 3. Time series of (a-c) meteorological variables including T (temperature), RH (relative humidity), Precip. (precipitation), WS (wind speed), WD (wind direction), and Vis (visibility), (d) mass concentrations and (e) mass fractions of PM₁ species. The pie charts show the average chemical composition of PM₁ for five episodes.



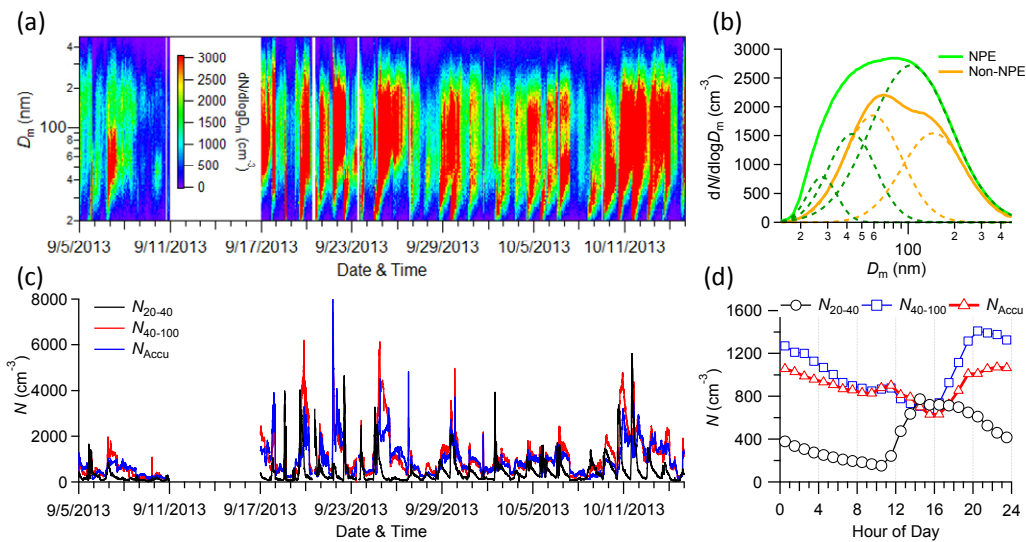
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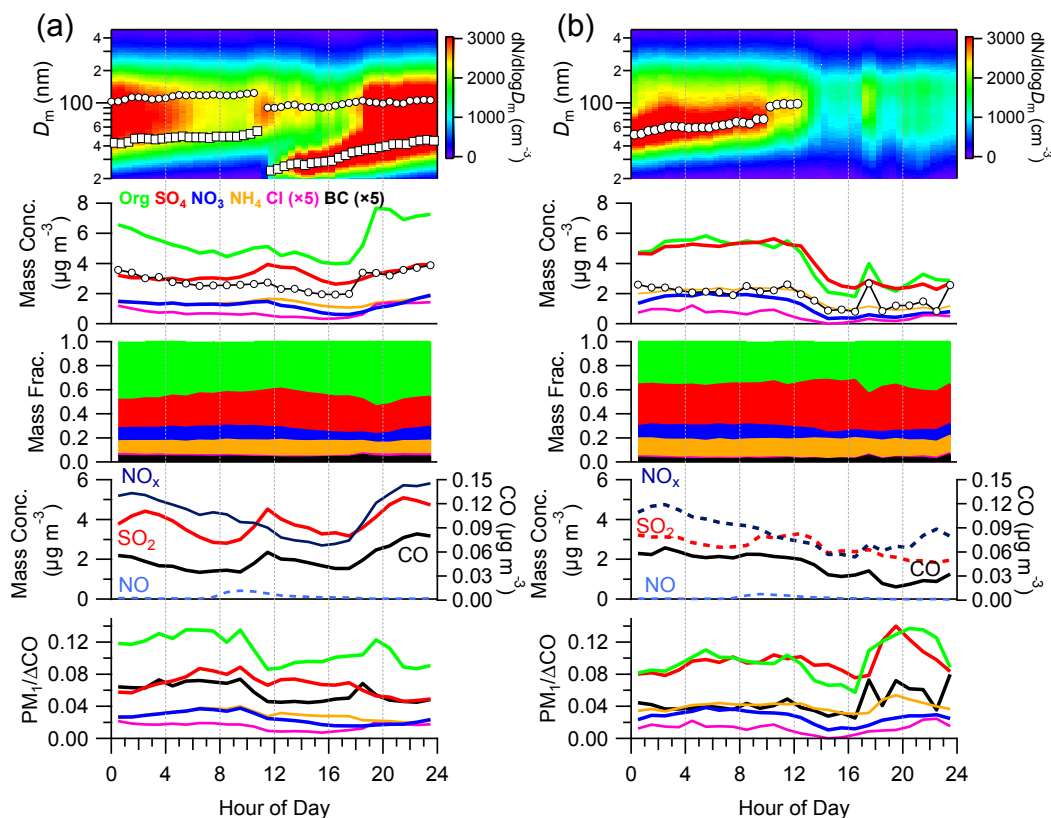
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 846 Figure 5. (a) Mass spectra and (b) time series of mass concentrations of BBOA and
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 852 Figure 6. Scatter plot of OOA versus SO_4 during BB and NBB periods. The data
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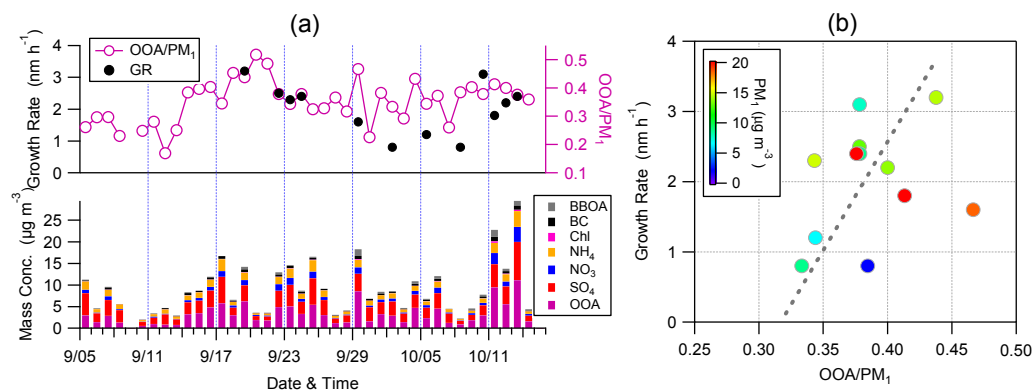
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868 Figure 9. (a) Time series of OOA/PM₁, particle growth rates, and average chemical
 869 composition during particle growth periods; (b) correlation of growth rate with
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