

# Concentrations and solubility of trace elements in fine particles at a mountain site, southern China: regional sources and cloud processing

T. Li<sup>1</sup>, Y. Wang<sup>1</sup>, W. J. Li<sup>2</sup>, J. M. Chen<sup>1, 2</sup>, T. Wang<sup>3</sup> and W. X. Wang<sup>2</sup>

[1]{School of Environmental Science and Engineering, Shandong University, Jinan, Shandong 250100, China}

[2]{Environment Research Institute, Shandong University, Jinan, Shandong 250100, China}

[3]{Department of Civil and Environmental Engineering, the Hong Kong Polytechnic University, Hong Kong, China}

Correspondence to: Y. Wang (wy@sdu.edu.cn)

## Abstract

The concentrations and solubility of twelve trace elements in PM<sub>2.5</sub> at Mt. Lushan, southern China, were investigated during the summer of 2011 and the spring of 2012. The average PM<sub>2.5</sub> mass was  $55.2 \pm 20.1 \mu\text{g m}^{-3}$  during the observation period. Temporal variations of all trace elements including total and water-soluble fractions with several dust storm spikes for total fraction Al and Fe were observed. The enrichment factor (EF) values were one order of magnitude higher for the water-soluble fractions versus the total fractions of trace elements. Four major emission sources were classified by principal component analysis (PCA), namely nonferrous metal mining and smelting (for Cr, As, Ba and parts of Zn), coal combustion (for Pb, Zn, Se, Cu and Mn), crustal materials (for Al and Fe) and municipal solid waste incineration (for Cd and Mo). Trajectory cluster analysis and the potential source contribution function (PSCF) consistently identified the Yangtze River Delta (YRD), the Pearl River Delta (PRD) and parts of Hunan and Jiangxi as the major source regions and pathways for anthropogenic elements, while northern China was identified for crustal elements. In

contrast, the local Jiangxi area has become the most significant contributor to the solubility of most trace elements, apart from the YRD with severe air pollution. In addition, the solubility alteration of trace elements in cloud events was investigated and transmission electron microscopy (TEM) analysis indicated that the irreversible alteration of particle morphology by cloud processing was highly responsible for the enhancement of element solubility. Our work implies an important role of regional anthropogenic pollution and cloud processing in the evolution of trace element solubility during transport.

## **1 Introduction**

Epidemiologic studies have associated long-term exposure to highly inhalable ambient fine particulate matter ( $\leq 2.5 \mu\text{m}$ ,  $\text{PM}_{2.5}$ ) with many adverse health outcomes such as cardiovascular and respiratory morbidity and mortality, whether in urban or rural areas (Hoek et al., 2013;Cao et al., 2012;Weichenthal et al., 2014). Sustained exposure to high concentrations of PM air pollution has been suggested to contribute to a decrease in life expectancy of approximately 3.0 y in China (Chen et al., 2013). The long atmospheric lifetimes of days give fine particles opportunities to be subject to long-range transboundary or intercontinental transport in the air and to deposition toward remote areas, carrying abundant anthropogenic pollutants and affecting ecosystems (Mahowald, 2011). Fine particles are also responsible for regional and global climate change and the hydrologic cycle interference through radiative forcing directly by reflecting sunlight and indirectly by changing cloud properties and precipitation acting as cloud condensation nuclei (Kaufman et al., 2002).

Although trace elements constitute only a small proportion of fine particle mass, their negative impacts on human health and ecosystems have attracted considerable attention because of their toxicity and bioaccumulation by inhalation and deposition. Toxicological and epidemiologic studies often suggest trace metals, particularly water-soluble metals which are more easily bioavailable, as the critical components harmful to the cardiopulmonary system through oxidative chemistry (Cakmak et al., 2014;Costa and Dreher, 1997). Soluble trace elements, especially the transition metals, are likely to be the primary drivers of the generation of reactive oxygen species, inducing cellular inflammation (Charrier et al., 2014;Cheung et al., 2012;Shafer et al., 2010). Besides, trace metal ions also play an important catalysis role in secondary aqueous-phase (Harris et al., 2013) and

1 particle-phase (Clements et al., 2013) sulphate formation, heterogeneous production and elimination  
2 of gas-phase hydrogen peroxide (Guo et al., 2014a) as well as multiphase cloud chemistry  
3 (Deguillaume et al., 2004). Importantly, most anthropogenic trace elements in ambient air are  
4 considered to be mainly concentrated on fine particles, strengthening their toxicity. Therefore,  
5 understanding both the concentrations and water solubility of fine particle trace elements is essential  
6 for revealing their bioavailability in ambient environments and reactivity in atmospheric chemistry.  
7 Diverse characteristics and sources of atmospheric trace elements have been investigated in various  
8 areas. For example, regional background, urban and industrial sites in Spain were found to have  
9 clearly different concentration levels and source origins of trace elements in PM (Querol et al., 2007);  
10 technogenic metalliferous fine aerosols observed in urban Montseny demonstrated the  
11 spatiotemporal variability of trace metal pollution (Moreno et al., 2011); traffic, static combustion  
12 and crustal were suggested as the main sources for Fe, Zn, Pb and Cu in PM<sub>2.5</sub> in an urban  
13 background area in Edinburgh (Heal et al., 2005); meanwhile, stationary industrial emissions from  
14 coal combustion were confirmed as the contributor to substantial atmospheric lead pollution of PM<sub>2.5</sub>  
15 in Shanghai (Chen et al., 2008). The United States has already established a spatially gridded  
16 national emissions inventory of PM<sub>2.5</sub> trace elements in speciation profiles with the dominant sources  
17 of crustal, biomass burning, coal combustion and industrial processing (Reff. et al., 2009). In Asia,  
18 atmospheric metallic element pollution by human activities has been surveyed, with the highest  
19 metal concentrations occurring in China (Fang et al., 2010). Coal combustion, nonferrous metal  
20 smelting and iron and steel manufacturing are found to be the major disproportionate sources of  
21 atmospheric hazardous trace elements in China according to the latest anthropogenic emissions  
22 inventory, which highlighted an increasing trend in air pollution by toxic heavy metals (Cheng et al.,  
23 2015).  
24 Recently, many field observations have found that even remote mountains and oceans have been  
25 polluted by anthropogenic trace elements in fine particles, significantly confirming the contributions  
26 of long-range transport (Deng et al., 2011; Schwab, 2004; Fomba et al., 2013). A few studies on dust  
27 in Asia (Takahashi et al., 2011; Hsu et al., 2013) indicate that the water solubility of trace elements  
28 could be modified during long-range transport, probably affected by emission sources, anthropogenic

pollution, acid processing and photochemical aging, etc. (Hsu et al., 2010). However, little information about fine particle trace elements has been reported with regard to the identification of source regions or pathways, and the main mechanism of trace element dissolution by atmospheric processing is still unclear.

This paper provides a dataset of the concentration and solubility (defined as the proportion of dissolved element concentration in total content) of trace elements in fine particles at Mt. Lushan, a high, rural mountain within an acid precipitation area in southern China. We focused on identifying the potential source region distributions for individual elements, aiming to evaluate the contributions of local, regional and long-range transport. Regional contributions to element solubility were also preliminarily investigated. One condensation-evaporation cloud cycle can increase aerosol solubility, and atmospheric aerosols may experience up to ten cloud cycles before removal (Desboeufs et al., 2001; Spokes et al., 1994); hence, we intensively discussed the influence of actual cloud processing on the alteration of fine particle trace element solubility, considering the rather frequent cloud events at Mt. Lushan. The size, mixing state and chemical composition of individual metal particles in cloud residues were detected using transmission electron microscopy (TEM) to investigate the effects of microphysical properties on the evolution of element solubility.

## **2 Methodology**

### **2.1 Site description**

Mt. Lushan is one of the National Geoparks of China, with an area of 302 km<sup>2</sup>, and is situated in northern Jiangxi Province, southern China, adjacent to the Yangtze River and Poyang Lake (Fig. 1). Mt. Lushan is significantly influenced by a subtropical monsoon and mountain climate, with frequent cloud events occurring from spring to autumn. The sampling site is located in Guniubei (115°59'E, 29°35'N, 1165 m a.s.l., see Fig. S1), the top of Guling town where most residents work in tourism or related services and little pollution is produced. There are intensive coal-fired power plants located at the eastern coastal area, and abundant mineral resources and many large non-ferrous industries exist in Jiangxi and in neighbouring provinces. Therefore, Mt. Lushan is a favourable platform for

observing the long-range transport of regional air pollution and for evaluating the influence of atmospheric particles on the local ecological environment.

## 2.2 Sample collection

Daily PM<sub>2.5</sub> samples were collected on quartz fibre filters (MK 360, 90 mm, Munktell, Switzerland) during the summer of 2011 (8 August–23 September) and the spring of 2012 (18 March–20 May). The filters were heated at 500 °C for 3 h prior to sampling, subsequently balanced at 25 °C and 50 ± 5% relative humidity for at least 24 h and weighed three times before and after collection. The sampling procedures were conducted during non-rainy and non-cloudy periods by employing a PM<sub>2.5</sub> impactor (TH-150A, Tianhong Co., China) with a medium flow rate of 100 L min<sup>-1</sup>. Seventy-six valid ambient samples and 6 blank filters were obtained and stored in the dark and at -20 °C for laboratory analysis.

Cloud water were collected by a improved single-stage Caltech Active Strand Cloudwater Collector (CASCC2, see the Supplement for detail). The samples were immediately filtered and stored in brown glass bottles at 4 °C with preservation of 1% v/v high pure hydrochloric acid.

The sampling system of cloud residues (Fig. S2) was employed to obtain individual cloud residues in cloud events on 11 September 2011 and 18 April 2012. Briefly, cloud residues were trapped onto copper TEM grids coated with carbon film using a single-stage cascade impactor with a 0.5 mm-diameter jet nozzle under a flow rate of 0.5 L min<sup>-1</sup>. An optical microscopy with magnification from ×500 to ×1200 was used to check the integrity of carbon film and particle distribution on the TEM grid. If suitable, the grids were sealed in a dry plastic tube and stored in desiccators at 25 °C and 20 ± 3% RH until the laboratory analysis. Otherwise, another satisfactory sample shall be collected and preserved.

## **2.3 Extraction procedure**

### **2.3.1 Total fraction (Acid)**

To obtain the total concentration of trace elements in fine particles with the least loss of semi-volatile elements, one quarter of the samples and blank filters were digested with an acid mixture (5 mL 65% HNO<sub>3</sub> and 2 mL 30% H<sub>2</sub>O<sub>2</sub>, guaranteed reagent grade, Kemiou Co., China) in closed Teflon vessels using a microwave digestion system according to a controlled gradient temperature procedure: reaching 120 °C in 5 minutes and remaining there for 5 minutes, then reaching 160 °C in 5 minutes and remaining there for 2 minutes, and finally reaching 185 °C in 5 minutes and remaining there for 15 minutes. After cooling to room temperature, the solutions were filtered and subsequently diluted to 25 mL with high-purity deionised water ( $\geq 18 \text{ M}\Omega\cdot\text{CM}$ ) in acid-cleaned brown glass bottles.

### **2.3.2 Water-soluble fraction (Water)**

Half of the samples and blank filters were extracted with 50 mL high-purity deionised water ( $\geq 18 \text{ M}\Omega\cdot\text{CM}$ ) using ultra-sonication for 1 h followed by filtration. Solutions of approximately 25 mL were preserved in acid-cleaned brown glass bottles with 1% v/v high pure hydrochloric acid to lessen adsorption of the elements by the glass walls. All solutions were stored at 4 °C until instrument analysis.

## **2.4 Chemical and physical analysis**

The concentrations of the total and water-soluble fractions of twelve trace elements (including Al, Cr, Mn, Fe, Cu, Zn, As, Se, Mo, Cd, Ba and Pb) were measured by Inductively Coupled Plasma Mass Spectrometry (ICP-MS, Agilent 7500a) based on the EPA 200.8 method using internal standard substances of Li, Sc, Ge, Y, In, Tb and Bi to eliminate matrix interference. National standard materials (soil, GSS-4, China) were digested for calculating element recoveries by the acid extraction method. All recoveries were found to be a little lower than 100% (Table S1), ranging from 93.0% for Cd to 72.7% for Se, primarily ascribed to volatilisation losses and acid mixture that is not strong enough to release all mineral constituents from the silica matrix into the solution. Reagent and filter

blanks were also prepared and analysed for background element content. The detection limits of the ICP-MS analysis for the total fraction elements determined were calculated as three times the standard deviation ( $3\delta$ ) of the six blank values (Table S1 in the Supplement). Trace elements concentrations in cloud water samples were similarly determined by ICP-MS.

Individual metal particles in cloud residues on the TEM grids were analysed with a JEM-2100 TEM operating at 200 kV accompanied by a semi-quantitative determination of elemental composition by an energy-dispersive X-ray spectrometer (EDS) that can detect elements heavier than carbon. EDS spectra were collected for only 15 s to minimise radiation exposure and potential beam damage. Copper could not be analysed because of interference from the copper TEM grid.

## **2.5 Meteorology**

Meteorology parameters during the observation periods and two cloud events were obtained from the local meteorological station (listed in Table S2). Generally, summer days exhibited higher temperature, relative humidity and wind speed than spring. In particular, the cloud events were characterized by high humidity ( $>99\%$ ), low wind speed (about  $1.0 \text{ m s}^{-1}$ ) and bad visibility (nearly 0 km).

## **2.6 Backward trajectories and potential source contribution function (PSCF)**

The Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model (PC Version 4.8) developed by the National Oceanic and Atmospheric Administration Air Resources Laboratory (NOAA-ARL) (Draxler and Rolph, 2014) was used to reconstruct the three-dimensional backward trajectories for the purpose of identifying the possible impacts of long-range transport. Three-day air mass backward trajectories initiated every 12 h arriving at Mt. Lushan at the height of 1,165 m a.s.l. were calculated, and five mean trajectory clusters were classified.

The potential source contribution function (PSCF) analysis developed by (Hopke et al., 1995) was computed using a geographical information system based software TrajStat (Wang et al., 2009) with the calculated backward trajectories and measured atmospheric pollutant concentrations, representing the conditional probability that air parcels may be responsible for concentrations higher than the

criterion level during transport to the receptor site. The PSCF value for the  $ij$ th grid cell is defined as

$$\text{PSCF}_{ij} = \frac{m_{ij}}{n_{ij}} \quad (1)$$

where  $n_{ij}$  is the total number of trajectory segment endpoints that fall in the  $ij$ th cell and  $m_{ij}$  is the number of endpoints for the same cell with arrival times at the sampling site corresponding to pollutant concentrations higher than an arbitrary criterion value. In this study, the total number of endpoints was 18,998, and the geophysical domain (90–130 °E, 15–45 °N) was divided into 4,800 grid cells with a  $0.5^\circ \times 0.5^\circ$  resolution. The average concentrations were set as criteria for PM<sub>2.5</sub> and each trace element, except for Al, for which the 75th percentile was better to distinguish between larger versus moderate regional sources. The PSCF values were multiplied by an arbitrary weight function  $W_{ij}$  to reduce the effect of small values of  $n_{ij}$  and to better reflect the uncertainty in values for these cells (Polissar et al., 2001). The weighting function reduced the PSCF values when the total number of the endpoints in a particular cell ( $n_{ij}$ ) was less than approximately three times the average value ( $n_{\text{Ave}}$ ) of the endpoints per each cell, defined as

$$W_{ij} = \begin{cases} 1.00 & n_{ij} > 3n_{\text{Ave}} \\ 0.70 & 1.5n_{\text{Ave}} < n_{ij} \leq 3n_{\text{Ave}} \\ 0.42 & n_{\text{Ave}} < n_{ij} \leq 1.5n_{\text{Ave}} \\ 0.17 & 0 < n_{ij} \leq n_{\text{Ave}} \end{cases} \quad (2)$$

### 3 Results and discussion

#### 3.1 Elemental composition

##### 3.1.1 General characterisation

The ambient atmospheric concentrations of PM<sub>2.5</sub> and trace elements including total and water-soluble fractions at Mt. Lushan are summarised in Table 1, and the comparison with typical mountains and megacities in China are listed in Table S3. The average PM<sub>2.5</sub> concentration at Mt. Lushan in spring ( $54.7 \mu\text{g m}^{-3}$ ) was approximately that of summer ( $55.9 \mu\text{g m}^{-3}$ ), more than twice the WHO air quality guideline for daily PM<sub>2.5</sub> of  $25 \mu\text{g m}^{-3}$ , but still lower than many urban sites at ground level such as Beijing ( $118.5 \mu\text{g m}^{-3}$ ) and Guangzhou ( $81.7 \mu\text{g m}^{-3}$ ) (Yang et al., 2011). It was



clearly contrary to Mt. Tai (Deng et al., 2011), where PM<sub>2.5</sub> exhibited a concentration 2–3 times higher in summer (123.1 µg m<sup>-3</sup>) than in spring (46.6 µg m<sup>-3</sup>), suggesting a less-pronounced seasonal variation of PM<sub>2.5</sub> at Mt. Lushan.

The total fraction concentrations of twelve trace elements in spring were comparable with those in summer at Mt. Lushan, with the exception of higher Al in spring. Overall, elements Al and Fe contributed to the highest concentration, and the other elements decreased from 258.3 ng m<sup>-3</sup> for Zn to 2.0 ng m<sup>-3</sup> for Mo. The dominant elements (Al, Fe and Zn) exhibited significantly lower concentrations at Mt. Lushan than Mt. Tai (1200 ng m<sup>-3</sup> for Al, 810 ng m<sup>-3</sup> for Fe and 400 ng m<sup>-3</sup> for Zn) (Deng et al., 2011) and slightly higher than Mt. Gongga (295.8 ng m<sup>-3</sup> for Al, 224.0 ng m<sup>-3</sup> for Fe and 154.6 ng m<sup>-3</sup> for Zn) (Yang et al., 2009a), probably due to the geomorphology and climatology biases. As shown in Table S3, all elements (except Ba) have lower concentration levels in comparison with Mt. Dinghu, a national nature reserve influenced by atmospheric pollution (Yang et al., 2009b) and with typical megacities in China such as Beijing, Guangzhou (Yang et al., 2011) and Shanghai (Chen et al., 2008). For water-soluble fractions, Zn transformed into the primary elements, followed by Al and Fe, with higher concentrations in summer than spring; other trace elements (from Pb to Mo) were still comparable between summer and spring.

### 3.1.2 Temporal variations

Figure 2 displays the temporal variations of trace element concentrations in PM<sub>2.5</sub> at Mt. Lushan for total and water-soluble fractions during the sampling campaigns. Most total fraction elements exhibited more intense variances in spring than in summer, with unique temporal patterns such as the highly variable pattern observed for aluminium and the relatively steady pattern of arsenium. Similar temporal patterns were observed for the water-soluble and total fractions for most elements, except for Al and Fe, which had drastic variations with many spike episodes for total fractions but lower constant concentration for water-soluble fractions. The dependence of temporal variations of water-soluble elements on their total fractions were also examined (Fig. S3). The concentrations of water-soluble Zn, Cu, Mn, As, Se and Cd were found to be rather consistent with their total fraction according to their high correlations ( $R^2 = 0.50\text{--}0.78$ ), whereas there was only a weak association

between the total concentrations and water-soluble fractions of Al, Fe and Cr.

Several spikes in element concentration were observed in specific periods. The first spike (Spike I) for all elements from 11–13 August 2011 was ascribed to the short sampling time due to the interruption by an unexpected thundershower. However, the significantly elevated total concentration of Al, Fe and Mn during 23–25 March 2012 (Spike II) and 25–26 April 2012 (Spike III) with no distinct elevation in the other elements were observed (Fig. 2), likely contributed by the dust storms originating from the Gobi desert and the Taklimakan desert in spring according to the MODIS image, the OMI Aerosol Index and air mass backward trajectory analysis (Fig. S4). These features were similar with the Asian Dust observed at Huaniao Island in the East China Sea, where concentrations of Al, Fe, Mn and Ba 3–4 times higher were observed with no significant difference for Zn, Cu, As and Cd in the air compared with non-dust days (Guo et al., 2014b). In addition, the dust storms contributed little to the water-soluble fractions of all trace elements.

## **3.2 Source identification**

### **3.2.1 Correlations between individual elements**

The Pearson correlation coefficients  $r$  between each element for both total fraction and water-soluble fraction were examined (Table S4). Moderate significant associations of Fe with elements Al, Mn and Cr are found to exist both in total and water-soluble fractions with comparable coefficients between 0.40–0.68. Correlations for other trace elements in water-soluble fraction are stronger than in total fraction, suggesting the higher possibilities of common sources for some water-soluble elements. For example, the correlation levels of Zn–Cu, Zn–Se, Pb–Zn and Pb–Se are dramatically high for water-soluble fractions ( $0.72 < r < 0.82$ ) and moderate for total fractions ( $0.25 < r < 0.52$ ), which could be attributed to their common anthropogenic sources of coal-fired power plants and industrial processes based on the suggestion of the spherical morphology and elemental composition of individual metal particles via the TEM examination (Li et al., 2014).

### **3.2.2 Anthropogenic pollution**

The enrichment factor (EF) for individual trace elements was applied to identification of their

general crustal and anthropogenic sources and evaluate the degree of pollution using the following formula:

$$EF_i = (X_i/X_R)_{\text{aerosol}} / (X'_i/X'_R)_{\text{crust}} \quad (2)$$

where  $EF_i$  is the enrichment factor of element  $i$ ;  $X_i$  and  $X_R$  are the concentrations of element  $i$  and reference element  $R$  in aerosol, respectively;  $X'_i$  and  $X'_R$  are the background contents of elements in Chinese soils (Wei et al., 1991); and Al is selected as the reference element for calculation. Element sources are classified into three groups with the following standard:  $EF < 10$  is considered to be a crustal origin without enrichment;  $10 < EF < 100$  comes from mixed origins (crustal and anthropogenic sources);  $EF > 100$  indicates air pollution from an anthropogenic origin. Figure 3 describes the EF values for both the total and water-soluble fractions of each element in a decreasing order. Elements with  $EF > 100$ , including Se, Cd, Zn, Pb, As, Mo and Cu, were found to be highly enriched, ranging from hundreds to tens of thousands in both fractions, indicating the severe anthropogenic pollution of atmospheric trace elements. Cr, Ba and Mn were likely to be from mixed origins because the majority of their EF values fell within 10–100. Fe was predominantly from a crustal origin based on its  $EF < 10$ . It is worth noting that the water-soluble fractions of most elements (except Cr and Fe) have higher EF values by approximately one order of magnitude than the total fractions, especially for elements with extremely high EF values, implying the inclination of enormous pollution toward the water-soluble fraction of anthropogenic elements.

### 3.2.3 Emission source classification

The principal component analysis (PCA) was performed in detail to classify the main sources of trace elements in  $PM_{2.5}$  at Mt. Lushan. The PCA results for trace elements in total fraction are shown in Fig. 4 with four major components identified.

The first principal component was considered primarily from nonferrous metal mining and metallurgical smelting, accounting for 34.4% of the total variance with high loadings for Cr, As, and Ba and considerable loadings for Zn and Mo. Although there were diverse EF values which suggested anthropogenic origins mixed with crustal materials ranging from dozens for Cr and Ba to thousands for As, Zn and Mo, these total fraction elements still presented considerably moderate

1 correlations. According to the provincial output statistics of nonferrous metals in 2010 (Cheng et al.,  
2 2015) and the Overall Planning of Mining Resources of 2008–2015 for Jiangxi and adjacent Hunan  
3 Province, there are abundant nonferrous minerals (such as barite, realgar and sphalerite, etc.) and  
4 many large-scale metallurgical industries within and near Jiangxi Province. In addition, high  
5 concentrations of Pb and Zn in cloud drop residues at Mt. Lushan were suggested deriving from fine  
6 particles contributed by nonferrous smelting in Jiangxi Province (Li et al., 2013).

7 The second principal component that contributed to high loadings for Pb, Cu, Se, Zn, and Mn was  
8 identified as coal combustion. Coal combustion is representatively considered to be the source of  
9 most atmospheric trace elements in China. All species in this component except Mn exhibited high  
10 EF values, indicating their anthropogenic origins. The semi-volatile Se is usually used as a tracer for  
11 determining atmospheric sources and even the pathways of coal combustion pollutants, because it is  
12 primarily produced in high temperature combustion and can undergo long-range transport with fine  
13 particles after the rapid gas-to-particle conversion (Husain et al., 2004; Wen and Carignan, 2007).  
14 Coal combustion is also becoming the major contributor to atmospheric Pb with the nationwide  
15 phase-out of leaded gasoline in China (Xu et al., 2012). The high correlations between Se and the  
16 other species in Component 2 clearly indicated the likely sources of coal combustion.

17 The third component in particular explained most of the Al and Fe which are the typical signatures of  
18 crustal materials or soils, along with relatively high correlations and low EF values.

19 Municipal solid waste incineration was regarded as the primary sources of fine particle Cd and Mo at  
20 Mt. Lushan, explaining 8.4% of the variance. Both Cd and Mo had low concentrations but high EF  
21 values, with a strong association in total fraction ( $r = 0.71$ ). Jakob et al., (1995) found that Cd in fly  
22 ash of municipal solid waste incinerators could evaporate into the atmosphere completely during heat  
23 treatment. The rapid increase in atmospheric Cd emissions from municipal solid waste incineration  
24 in recent years and the unbalanced large emissions in east-central and southeast China (Tian et al.,  
25 2012) were responsible for fine particle Cd at Mt. Lushan.

### 26 **3.3 Potential source region distributions**

27 Three-day air mass backward trajectories were calculated to estimate the potential source regions and

1 pathways of air pollutants, making full use of the geographical advantages of the high-altitude  
 2 observation site and the long-range transport capacity of fine particles. The air masses associated  
 3 with PM<sub>2.5</sub> samples were classified into five clusters (Fig. 5a). The air masses of the Southwest (SW)  
 4 cluster originated from the South China Sea and passed through the Pearl River Delta (PRD), Hunan  
 5 and Jiangxi Province with the highest occurrence frequency of 28.9%. Cluster East (E) mainly came  
 6 from the East China Sea and partly from the Circum-Bohai Sea Region, traversing the Yangtze River  
 7 Delta (YRD) and Anhui Province at a frequency of 27.6%. Air masses of cluster Local (L) were  
 8 found to be advancing toward or revolving around Mt. Lushan slowly from ground-level within areas  
 9 covering distances of approximately 200 km away from the observation site, contributing 22.4% of  
 10 the occurrence frequency. Clusters Northwest (NW) and North (N) both went through the China  
 11 Loess Plateau and accounted for 13.2 and 7.9% of the occurrence frequency, respectively. The  
 12 primary difference was that the former mainly originated from Taklimakan and the Gobi Desert,  
 13 whereas the latter was from remote Siberia and Mongolia with a much higher wind speed.

14 Figure 5b shows the averaged PM<sub>2.5</sub> concentration of different clusters. Cluster E exhibited the  
 15 highest PM<sub>2.5</sub> concentration of 67.0  $\mu\text{g m}^{-3}$ , followed by cluster L (59.6  $\mu\text{g m}^{-3}$ ) and the similar  
 16 clusters NW and N, whereas cluster SW had the lowest concentration (45.3  $\mu\text{g m}^{-3}$ ). The order of  
 17 PM<sub>2.5</sub> concentration seemed to be consistent with the PSCF result (Fig. 5c) which integrated PM<sub>2.5</sub>  
 18 concentration with air mass occurrence frequency. As shown in Fig. 5c, the YRD and southern Anhui  
 19 Province were the most likely source regions of PM<sub>2.5</sub> at Mt. Lushan. These economically developed  
 20 regions exhibit massive energy consumption and industrial production, becoming important  
 21 anthropogenic source regions in eastern China. The local area of Jiangxi, neighbouring southern  
 22 Anhui and eastern Hunan were also significant PM<sub>2.5</sub> source regions, where many nonferrous metal  
 23 smelting industries are widely distributed. However, the contributions to PM<sub>2.5</sub> at Mt. Lushan by  
 24 long-range transport of air masses from northern China (Inner Mongolia and the Loess Plateau) were  
 25 much less than from eastern China and local regions, suggesting nearly negligible impacts of mineral  
 26 dusts or soil transport on fine particle mass concentration at Mt. Lushan compared with regional  
 27 anthropogenic pollution. In addition, the PM<sub>2.5</sub> concentration in five clusters or different source  
 28 regions were highly dependent on the inorganic ions, especially the secondary ions (Li et al., 2014).

1 Interestingly, the PRD contributed much less concentrations to fine particles in spite of its high  
2 occurrence frequency, probably due to particle scavenging by frequent rainfall during the Asian  
3 summer monsoon.

4 Figure 6a–l illustrates the distinctive potential source region distributions for individual elements in  
5 fine particles at Mt. Lushan. Crustal elements Al and Fe exhibited similar source region distributions  
6 (Fig. 6a and b). The highest PSCF values in northern China including the Loess Plateau, Henan and  
7 eastern Hubei Province indicated that these areas were very important source regions and pathways  
8 for fine particle Al and Fe, with vast contributions of long-range transport of mineral dusts or soils.  
9 The PRD and eastern Hunan Province were likely to be other important source regions for Al and Fe  
10 probably due to the influence of mining activities, whereas the YRD in eastern China and the local  
11 area of Jiangxi Province made much lower contributions.

12 Fine particle As (Fig. 6c) had an almost parallel pattern of source region distribution with Fe, except  
13 the discordance that the PRD, eastern Guangxi, Hunan and Jiangxi Province were identified as the  
14 most important source regions and pathways, attributable to the emissions of the nonferrous metals  
15 mining and smelting industries. Northern China also tended to be an important potential source  
16 region, suggesting the possible contribution of natural sources for As. Similar to As, the PRD, Hunan  
17 and Jiangxi Province, as well as northern China were also potential source regions for fine particle  
18 Mn (Fig. 6d). However, the YRD area was characterised by extremely high PSCF values for Mn,  
19 indicating an important anthropogenic source region.

20 Fine particle Pb, Zn, Cu and Se (Fig. 6e–h), which were identified primarily from coal combustion  
21 by the PCA analysis, had similar source region distributions with some discrepancies, generally  
22 coinciding with the geographical distribution of coal-fired power plants in China which were  
23 aggregated within coastal areas such as the YRD, the PRD and Shandong Province (Fig. S5a). The  
24 YRD area including Jiangsu, Zhejiang and southern Anhui Province was an especially pronounced  
25 source region for Pb, Zn, Cu and Se at Mt. Lushan. The significantly high PSCF values in the YRD  
26 region were attributable to the large coal consumption by intensive power plants in Jiangsu, Zhejiang  
27 and even Shandong Province and to the differences in trace element contents in raw coal such as high  
28 Se contents in Anhui Province, both of which made great contributions to emissions of atmospheric

trace elements (Tian et al., 2014). Meanwhile, the PRD, Hunan and the local area of Jiangxi Province also played a remarkable role in source region distributions of Zn, Cu and Se at Mt. Lushan, for which coal-fired power plants and nonferrous metals smelting industries in these areas were primarily responsible. Unexpectedly, northern China rather than the PRD area was also identified as a potential source region of Pb, possibly attributed to the partial contributions of long-range transport of mineral dusts for fine particle Pb at Mt. Lushan.

As shown in Fig. 6i and j, the YRD, the PRD, eastern Hunan and western Jiangxi were identified as the common likely source regions and pathways for fine particle Cd and Mo at Mt. Lushan. The identified source regions corresponded very well with the uneven regional distribution of municipal solid waste (MSW) incineration plants in China (Fig. S5b), with 183.5, 175.1 and 146.6 kg of atmospheric Cd emitted by MSW incineration from Jiangsu, Zhejiang and Guangdong, respectively, the top three largest emitting provinces in 2010 (Tian et al., 2012). Figure 6k shows an apparently more homogeneous source region distribution for Ba, with the PRD and areas surrounding Mt. Lushan as potential source regions which have a high natural background content of barite. On the contrary, few potential source regions were identified for Cr (Fig. 6l) besides the PRD, parts of Hunan Province and a small region in central Zhejiang with low PSCF values.

It should be noted that the potential source region distributions for trace elements at Mt. Lushan were very different with  $PM_{2.5}$ . The YRD, the PRD and Hunan Province were likely to be the major source regions for most anthropogenic elements (except Cu within eastern Jiangxi), and northern China was likely for crustal elements (Al and Fe). However, the contribution probabilities of areas near Mt. Lushan for most elements were obviously lower than other source regions. This result revealed that long-range transport and regional anthropogenic sources, rather than local sources, made much more significant contributions to fine particle trace elements at Mt. Lushan, which could be ascribed to the different types and quantities of emissions as well as the protection policies for Mt. Lushan and its adjacent Poyang Lake.

In addition, the order of mass concentration of individual trace elements for five air mass clusters ending at Mt. Lushan (Fig. S6) was in accordance with the distributions of likely source regions identified by PSCF values, further suggesting greater contributions by regional sources and

long-range transport to fine particle trace elements at Mt. Lushan. For example, clusters N and NW contributed to the highest concentration for crustal Al and Fe and anthropogenic Zn and As, followed by the highest cluster SW, E and the lowest cluster L; exceptionally, cluster L made the highest contribution to the combustion element Cu.

### **3.4 Trace element solubility**

#### **3.4.1 Comparison of individual element solubility**

In this study, element solubility is calculated as the percentage of water-soluble concentration of an element divided by its total concentration. Figure S7 compares the average water solubility of individual elements in PM<sub>2.5</sub> at Mt. Lushan with other sites in decreasing order. The most soluble elements were As, Mn, Cu, Zn and Se with a solubility reaching up to 60–70%; followed by Cd, Pb, Ba and Mo with moderate solubility, approximately 40–50%; whereas the least soluble elements were Fe, Al and Cr with a solubility less than 30%. Compared with TSP over the East China Sea (Hsu et al., 2010), some elements (Mn, Cu, Ba, Mo, Fe and Al) were significantly highly soluble in PM<sub>2.5</sub> at Mt. Lushan, whereas other species (As, Zn, Se and Cd) showed the opposite with the exception of comparable Pb and Cr. Similarly, compared with the PM<sub>2.5</sub> at Edinburgh, UK (Heal et al., 2005), the elements Mn, Cu and Fe exhibited higher water solubility, whereas the elements As, Zn and Pb were comparable, and the elements Cd and Cr were lower at Mt. Lushan. A large gap in anthropogenic elements (As, Mn, Cu and Pb) between Mt. Lushan and Nanjing (Hu et al., 2012) was also observed. Experimental conditions (compared in Supplement, Table S5) such as digestion methods or analysis instruments may result in some solubility discrepancies at different sites, but they are far from the determinant.

#### **3.4.2 Effect of emission sources on element solubility**

Element solubility is associated with four major emission sources identified by EF value and PCA analysis and displayed in Fig. 7. Crustal elements with mean EF < 10 had low average solubility (< 30%) besides the highly dissolved Mn that was possibly impacted by a combustion source; the solubility of elements related to mining and metallurgy varied between < 10%–70% with a 10–1000



range in EF value; coal combustion elements characterised by high EF values over 100 exhibited a high solubility of approximately 45%–75%, contributable to the combustion process during which oxidisable elements in coal are mostly altered to easily soluble species (Quispe et al., 2012); elements produced by waste incineration were also highly soluble (40%–50%) coupled with EF > 100. The roughly increasing tendency of solubility along with EF value suggested that element solubility was closely related to the emission sources, likely indicating that crustal elements with low EF values often showed low solubility, whereas polluted elements featuring high EF values were usually highly soluble. This result coincides well with the conclusion that the dissolvable portion of aerosol elements is controlled by their dominant crustal or anthropogenic origins (Hsu et al., 2005). However, there are also some exceptions that could not be explained by emission sources. For instance, the low solubility of atmospheric Cr at Mt. Lushan could probably be ascribed to the stable crystalline structure in minerals such as chromite, with the reduced Cr (III) being the dominant species in ambient fine particles (Werner et al., 2007). Figure S8 further shows a hyperbolic or negative logarithmic relationship between element concentration and the solubility for and only for Al, Fe, Cr and Mn, similar to many inverse relationships for aerosol Fe, Al and other species reviewed by (Schulz et al., 2012), indicating the unneglected effect of total concentration on crustal trace element solubility rather than anthropogenic elements. Thus, trace element solubility may be initially determined by emission sources with various interference.

### **3.4.3 Source region contributions to element solubility**

Figure 8 illustrates the potential source regions for individual element solubility in PM<sub>2.5</sub> at Mt. Lushan with the expanding application of PSCF analysis to evaluate the local and regional contributions. It can be seen that almost all of the elements had quite different solubility distributions with their total concentrations, characterised by the highest PSCF values for element solubility in the local area of Jiangxi Province and the YRD region followed by lower PSCF values in parts of the PRD region and Hunan Province.

In this study, we used fine particle Fe for detailed explanation because there are many investigations on the solubility of aerosol Fe, which plays a critical role in the global iron cycle among terrestrial

dust, the atmosphere and the ocean, as well as in the effects on climate (Jickells et al., 2005). As shown in Fig. 8a, northern China, identified as the major source region for crustal Fe, made little contribution to its solubility. The significantly low solubility of fine particle Fe in northern China, much lower than the PSCF criteria value, was consistent with the very low soluble Fe (solubility less than 2%) from desert dusts or soils (Baker et al., 2006; Jickells et al., 2005). The experiment performed by (Mackie et al., 2006) suggests that the amount of readily released Fe does not increase after the uplift and abrasion of soils in areas with little atmospheric pollution; thus, the low solubility of the Fe from northern China could be attributed to the initial absence of dissolved Fe in dusts or soils derived from the Gobi desert or Loess Plateau. On the contrary, the relatively much higher PSCF values for Fe in the local area of Jiangxi, southern Anhui and the YRD indicated greater contributions of anthropogenic pollution in these regions to the solubility of fine particle Fe at Mt. Lushan, because the high solubility of Fe in the YRD and Anhui Province was just in accordance with the large amounts of anthropogenic emissions of SO<sub>2</sub> and NO<sub>x</sub> in these regions (Zhang et al., 2009), which could significantly promote the dissolution of aerosol Fe acting as acid precursors (Schulz et al., 2012; Solomon et al., 2009). A significant correlation (0.629) of iron solubility and sulphate concentration in our fine particle samples was further verified, similar to the correspondence of high iron solubility to the sulphur content in ambient fine particles observed by (Oakes et al., 2012). Moreover, the outstanding enhancement of fine particle Fe solubility in local regions close to Mt. Lushan was principally recognised to be an effect of cloud processing, an important atmospheric process controlling Fe dissolution in aerosols (discussed in Sect. 3.4.4). The regional distributions for the solubility of crustal Al was similar with Fe, except for the lower PSCF values near Mt. Lushan (Fig. 8b).

Anhui, Jiangsu and Jiangxi Province were identified as the most important contributors to the high solubility of fine particle As at Mt. Lushan (Fig. 8c), ascribed to the large amount of atmospheric As emissions from anthropogenic sources (approximately 200 tons in 2010) in these provinces (Cheng et al., 2015); in contrast, little contributions were made to As solubility by the PRD and Hunan Province where the highest As concentrations were provided. For the elements from coal combustion (including Mn, Pb, Zn, Cu and Se, Fig. 8d–h) and MSW incineration (including Cd and Mo, Fig. 8i

and j), the highest contributions of element solubility were uniformly focused within the local area of Jiangxi Province, the YRD and parts of Fujian Province, generally coinciding with their significantly high concentrations emitted from anthropogenic sources in eastern China. The solubility pattern for Ba (Fig. 8k) was rather similar to its concentration distribution, whereas the pattern of high solubility for Cr in Anhui Province (Fig. 8l) was very different to the concentration pattern.

It can be concluded that the solubility of fine particle trace elements from local and regional source regions were much higher than that by long-range transport from south-western China and northern China. The high contributions for particle element solubility in the YRD region should be attributed to the serious air pollution. However, on account of the relatively less air pollution released from local sources, the frequent cloud processing was considered as the significant contributor to the high solubility of trace elements in local regions near Mt. Lushan.

Additionally, the statistics on the solubility of individual fine particle elements at Mt. Lushan for five air mass categories are shown in Fig. S9. In general, the fine particle elements derived from cluster L exhibited the highest solubility whether for crustal or anthropogenic elements, closely followed by cluster E; elements from cluster SW showed moderate solubility, ascribed to the wash-out of soluble constituents by frequent rainfall in summer monsoon; elements from cluster NW and N were the least soluble, primarily impacted by crustal materials from northern China. It was demonstrated that the statistical results for most elements fairly agreed with the solubility distributions identified by PSCF analysis.

#### **3.4.4 Evolution of element solubility during cloud processing**

Cloud processing might be of great importance in affecting aerosol dissolution at Mt. Lushan where frequent cloud events occurred. To determine the evolution of element solubility during cloud processing with less disturbance of natural factors and human activities, cloud events were selected with the following conditions: (1) non-raining cloud to avoid scavenging of particles by precipitation, (2) short duration and time interval of cloud processing and particle sampling to minimise the interference of external aerosols, (3) identical air mass trajectory of the cloud and particles with low wind speed to ensure consistent origins of the air masses. Table S6 shows the alteration of individual

1 element solubility in fine particles before and after the two selected cloud events on 11 September  
2 2011 and on 18 April 2012.

3 It was evident that after cloud processing on 11 September 2011, most fine particle elements  
4 exhibited a drastic increase in solubility. For instance, post-cloud particle Al became much more  
5 soluble compared with the pre-cloud particle, with solubility changing from 1.7 to 28.7%. Other  
6 elements, such as Cu, Zn, As, Se, Cd and Ba, also showed an approximately one- to two-fold  
7 increase in solubility. The cloud processing on 18 April 2012 also brought about a significantly large  
8 number of solubility increments for the crustal elements Al and Fe, but it resulted in less increments  
9 for anthropogenic elements such as Cu, As, Se and Pb due to their already higher solubility in  
10 pre-cloud particles which might have undergone intensive cloud cycles during the prolonged  
11 residence time in the air. Generally, the evidently higher solubility for the post-cloud fine particle  
12 elements after the two cloud events, overpassing their overall average solubility, corroborated the  
13 remarkable dissolution efficiencies of aerosol elements by cloud processing at Mt. Lushan.

14 Interestingly, the increase in aerosol element solubility after both cloud events was coupled with the  
15 elevation of aerosol sulphate concentration as shown in Table S6. For example, post-cloud sulphate  
16 particles (4 h) exhibited a sharp elevation of  $10 \mu\text{g m}^{-3}$  after the cloud processing (3 h) on 11  
17 September 2011. Based on the very short intervals between cloud events and post-cloud particle  
18 sampling, the increased sulphate particles were most likely contributed by cloud water sulphate,  
19 which could be generated by the rapid atmospheric aqueous-phase oxidation of S(IV) with a rate  
20 greater than  $100\% \text{ h}^{-1}$  (Seinfeld and Pandis, 2012), much quicker than the gas-phase oxidation of  $\text{SO}_2$   
21 with a rate of approximately  $1\% \text{ h}^{-1}$  (Newman, 1981). In addition, the higher sulphate concentrations  
22 in cloud water and post-cloud particles on 11 September 2011 ( $41.70 \text{ mg L}^{-1}$  and  $24.72 \mu\text{g m}^{-3}$ ,  
23 respectively) from cluster E compared with those on 18 April 2012 ( $26.78 \text{ mg L}^{-1}$  and  $17.34 \mu\text{g m}^{-3}$ ,  
24 respectively) from cluster L were likely to be associated with the larger amount of anthropogenic  
25  $\text{SO}_2$  gases from the YRD in eastern China because acidic substances, such as sulphate formed by  
26 their precursor gases (e.g.,  $\text{SO}_2$ ), are prone to exist in fine particles and lead to acidification of  
27 aerosols (Ren et al., 2011); and the pH of cloud droplets modified by these dissolved acidic  
28 substances in particles could in turn influence the dissolution of aerosol trace elements (Deguillaume

et al., 2005). As a consequence, the enhancement of trace element solubility during cloud processing were probably dependent on the corresponding heterogeneous formation of sulphate.

To further understand the effect of cloud processing on aerosol elements dissolution, individual cloud droplet residues were collected during the two cloud events, and the micro-morphology and composition of metal particles were examined by TEM-EDS analysis. Representative TEM images and the EDS spectra of major metal particles are shown in Fig. 9. Plenty of Pb-rich, Fe-rich and fly ash particles were observed to be the dominating metal particles in the two cloud events, consistent with the major types of metal-associated particles (Pb-rich, fly ash, Fe-rich and Zn-rich) in cloud droplets at Mt. Lushan (Li et al., 2013). Almost all of metal particles were found to be internally mixed within S-rich cloud residues with the encapsulation of cloud water and presented a nearly spherical shape or aggregation, suggesting the likely sources to be high-temperature coal combustion by coal-fired power plants and industries, during which industrial gases containing abundant SO<sub>2</sub> and metals were released, followed by the generation of metal-sulphate particles (Gieré et al., 2006).

Figure 9 also demonstrates the small diameters of the metal particles in cloud residues much smaller than 1 µm, especially for the aggregation of spherical Fe nanoparticles embedded in aged S-rich residues (Fig. 9c). The unique, large specific surface of nano-sized Fe particles is able to effect a higher absorption intensity and more complicated surface chemistry during cloud processing, generating greater quantities of surface complexes with acidic anions such as sulphate, and improving the dissolution rates of iron particles (Rubasinghege et al., 2010). Shi et al. (2009) suggested that the variations in pH during cloud processing would induce the formation of amorphous Fe nanoparticles and an increase in iron solubility (reactivity) in Saharan dusts. More importantly, the solubilisation of aerosol elements caused by the change in particle morphology during cloud processing is irreversible (Deguillaume et al., 2005). Finally, along with the dissipating of cloud water in droplets, the Fe nanoparticles and sulphate residues remain in post-cloud particles, resulting in an attractive enhancement of Fe solubility. Therefore, we hypothesise that irreversible changes or alterations of particle morphology such as specific surface area, which may result from the acidification and heterogeneous reactions during cloud processing, should be the critical factor for the significant enhancement of solubility not only for Fe, but also for many other trace elements

in cloud-processed fine particles.

## 4 Conclusions

The characteristics of fine particle trace elements at the summit of a mountain in southern China were investigated. The  $\text{PM}_{2.5}$  concentration ( $55.2 \pm 20.1 \mu\text{g m}^{-3}$ ) was twice the WHO guideline but much lower than that in urban sites, with little seasonal variation. The total fractions of Al and Fe and the water-soluble fraction of Zn were the dominant elements and similar temporal variations between the total and water-soluble fractions for all species along with several spikes contributed by dust storms for Al and Fe were observed. The enrichment factor analysis implied the pollution trace elements to be more easily concentrated in water-soluble fractions with higher correlations. Nonferrous metal mining and smelting, coal combustion, crustal materials and municipal solid waste incineration were further classified as the major emission sources, which might initially determine the trace element solubility with interferences such as total concentrations. The YRD including Jiangsu, Zhejiang and parts of Anhui Province were indicated as the most important source regions of  $\text{PM}_{2.5}$  by trajectory cluster analysis and PSCF results, and this result was discrepant with the source distributions for individual trace elements. The YRD as well as the PRD, eastern Hunan and western Jiangxi were identified as the primary source regions and pathways for combustion-related elements (Mn, Pb, Zn, Cu, Se, Cd and Mo) emitted by regional coal-fired power plants and municipal solid waste incineration, whereas the long-range transport of mineral dusts from northern China was the major contributor to crustal Al and Fe. Interestingly, local sources contributed relatively less to most trace elements. In contrast, the YRD region and the local area of Jiangxi Province were very likely to make the greatest contributions to element solubility rather than long-range transport from northern or southern China. A significant enhancement in trace element solubility and corresponding increase in aerosol sulphate were observed during two selected cloud events. TEM-EDS analysis of cloud residues suggested that the irreversible alteration of particle morphology by heterogeneous reactions or acidification with sulphate during cloud processing could crucially increase element solubility.

This study highlighted the contributions of regional anthropogenic pollution and local cloud processing to trace element solubility of  $\text{PM}_{2.5}$  at Mt. Lushan. We further expect that fine particles

1 would undergo several cloud-processing cycles during long-range transport with massive clouds  
2 covering the surface of the earth, which can promote aerosol elements dissolution and increase health  
3 and ecological risks.

#### 4 **Acknowledgements**

5 This research was funded by the National Natural Science Foundation of China (21177073 and  
6 41075092). We acknowledge the NOAA Air Resources Laboratory (ARL) for provision of the  
7 HYSPLIT trajectory model. We are grateful for all the staff of Lushan Mountain Meteorological  
8 Station for offering the observation platform and their assistance during the field campaigns. Thanks  
9 also due to Taixing Yue at Environmental Monitoring Central Station of Shandong Province for the  
10 help in the ICP-MS measurements.

11

## References

- Baker, A. R., Jickells, T. D., Witt, M., and Linge, K. L.: Trends in the solubility of iron, aluminium, manganese and phosphorus in aerosol collected over the Atlantic Ocean, *March Chem.*, 98, 43-58, doi:10.1016/j.marchem.2005.06.004, 2006.
- Cakmak, S., Dales, R., Kauri, L. M., Mahmud, M., Van Ryswyk, K., Vanos, J., Liu, L., Kumarathasan, P., Thomson, E., Vincent, R., and Weichenthal, S.: Metal composition of fine particulate air pollution and acute changes in cardiorespiratory physiology, *Environ. Pollut.*, 189, 208-214, doi:10.1016/j.envpol.2014.03.004, 2014.
- Cao, J., Xu, H., Xu, Q., Chen, B., and Kan, H.: Fine particulate matter constituents and cardiopulmonary mortality in a heavily polluted Chinese city, *Environ. Health Perspect.*, 120, 373-378, doi:10.1289/ehp.1103671, 2012.
- Charrier, J. G., McFall, A. S., Richards-Henderson, N. K., and Anastasio, C.: Hydrogen peroxide formation in a surrogate lung fluid by transition metals and quinones present in particulate matter, *Environ. Sci. Technol.*, 48, 7010-7017, doi:10.1021/es501011w, 2014.
- Chen, J., Tan, M., Li, Y., Zheng, J., Zhang, Y., Shan, Z., Zhang, G., and Li, Y.: Characteristics of trace elements and lead isotope ratios in PM(2.5) from four sites in Shanghai, *J. Hazard. Mater.*, 156, 36-43, doi:10.1016/j.jhazmat.2007.11.122, 2008.
- Chen, Y., Ebenstein, A., Greenstone, M., and Li, H.: Evidence on the impact of sustained exposure to air pollution on life expectancy from China's Huai River policy, *Proc. Natl. Acad. Sci. U.S.A.*, 110, 12936-12941, 2013.
- Cheng, K., Wang, Y., Tian, H., Gao, X., Zhang, Y., Wu, X., Zhu, C., and Gao, J.: Atmospheric Emission Characteristics and Control Policies of Five Precedent-Controlled Toxic Heavy Metals from Anthropogenic Sources in China, *Environ. Sci. Technol.*, 49, 1206-1214, doi:10.1021/es5037332, 2015.
- Cheung, K., Shafer, M. M., Schauer, J. J., and Sioutas, C.: Diurnal trends in oxidative potential of coarse particulate matter in the Los Angeles Basin and their relation to sources and chemical composition, *Environ. Sci. Technol.*, 46, 3779-3787, doi:10.1021/es204211v, 2012.
- Clements, A. L., Buzcu-Guven, B., Fraser, M. P., Kulkarni, P., and Chellam, S.: Role of particulate



1 metals in heterogenous secondary sulfate formation, *Atmos. Environ.*, 75, 233-240,  
2 doi:10.1016/j.atmosenv.2013.04.038, 2013.

3 Costa, D. L., and Dreher, K. L.: Bioavailable transition metals in particulate matter mediate  
4 cardiopulmonary injury in healthy and compromised animal models, *Environ. Health Perspect.*, 105,  
5 1053-1060, 1997.

6 Deguillaume, L., Leriche, M., Monod, A., and Chaumerliac, N.: The role of transition metal ions on  
7 HO x radicals in clouds: a numerical evaluation of its impact on multiphase chemistry, *Atmos. Chem.*  
8 *Phys.*, 4, 95-110, 2004.

9 Deguillaume, L., Leriche, M., Desboeufs, K., Mailhot, G., George, C., and Chaumerliac†, N.:  
10 Transition metals in atmospheric liquid phases: Sources, reactivity, and sensitive parameters, *Chem.*  
11 *Rev.*, 105, 3388-3431, doi:10.1021/cr040649c, 2005.

12 Deng, C., Zhuang, G., Huang, K., Li, J., Zhang, R., Wang, Q., Liu, T., Sun, Y., Guo, Z., Fu, J. S., and  
13 Wang, Z.: Chemical characterization of aerosols at the summit of Mountain Tai in Central East China,  
14 *Atmos. Chem. Phys.*, 11, 7319-7332, doi:10.5194/acp-11-7319-2011, 2011.

15 Desboeufs, K. V., Losno, R., and Colin, J. L.: Factors influencing aerosol solubility during cloud  
16 processes, *Atmos. Environ.*, 35, 3529-3537, doi:10.1016/S1352-2310(00)00472-6, 2001.

17 Draxler, R. R., and Rolph, G. D.: HYSPLIT (HYbrid Single-Particle Lagrangian Integrated  
18 Trajectory) Model access via NOAA ARL READY Website, available at:  
19 <http://www.arl.noaa.gov/HYSPLIT.php> (last access: 1 February 2015), NOAA Air Resources  
20 Laboratory, College Park, MD, 2014.

21 Fang, G. C., Huang, Y. L., and Huang, J. H.: Study of atmospheric metallic elements pollution in  
22 Asia during 2000-2007, *J. Hazard. Mater.*, 180, 115-121, doi:10.1016/j.jhazmat.2010.03.120, 2010.

23 Fomba, K. W., Müller, K., van Pinxteren, D., and Herrmann, H.: Aerosol size-resolved trace metal  
24 composition in remote northern tropical Atlantic marine environment: case study Cape Verde islands,  
25 *Atmos. Chem. Phys.*, 13, 4801-4814, doi:10.5194/acp-13-4801-2013, 2013.

26 Gieré R., Blackford, M., and Smith, K.: TEM study of PM<sub>2.5</sub> emitted from coal and tire  
27 combustion in a thermal power station, *Environ. Sci. Technol.*, 40, 6235-6240, 2006.

28 Guo, J., Tilgner, A., Yeung, C., Wang, Z., Louie, P. K., Luk, C. W., Xu, Z., Yuan, C., Gao, Y., Poon,

1 S., Herrmann, H., Lee, S., Lam, K. S., and Wang, T.: Atmospheric peroxides in a polluted subtropical  
2 environment: seasonal variation, sources and sinks, and importance of heterogeneous processes,  
3 Environ. Sci. Technol., 48, 1443-1450, doi:10.1021/es403229x, 2014a.

4 Guo, L., Chen, Y., Wang, F., Meng, X., Xu, Z., and Zhuang, G.: Effects of Asian dust on the  
5 atmospheric input of trace elements to the East China Sea, March Chem., 163, 19-27,  
6 doi:10.1016/j.marchem.2014.04.003, 2014b.

7 Harris, E., Sinha, B., van Pinxteren, D., Tilgner, A., Fomba, K. W., Schneider, J., Roth, A., Gnauk, T.,  
8 Fahlbusch, B., Mertes, S., Lee, T., Collett, J., Foley, S., Borrmann, S., Hoppe, P., and Herrmann, H.:  
9 Enhanced role of transition metal ion catalysis during in-cloud oxidation of SO<sub>2</sub>, Science, 340,  
10 727-730, doi:10.1126/science.1230911, 2013.

11 Heal, M. R., Hibbs, L. R., Agius, R. M., and Beverland, I. J.: Total and water-soluble trace metal  
12 content of urban background PM<sub>10</sub>, PM<sub>2.5</sub> and black smoke in Edinburgh, UK, Atmos. Environ., 39,  
13 1417-1430, doi:10.1016/j.atmosenv.2004.11.026, 2005.

14 Hoek, G., Krishnan, R. M., Beelen, R., Peters, A., Ostro, B., Brunekreef, B., and Kaufman, J. D.:  
15 Long-term air pollution exposure and cardio-respiratory mortality: a review, Environ. Health-Glob,  
16 12, 43, doi:10.1186/1476-069X-12-43, 2013.

17 Hopke, P., Barrie, L., Li, S. M., Cheng, M. D., Li, C., and Xie, Y.: Possible sources and preferred  
18 pathways for biogenic and non-sea-salt sulfur for the high Arctic, J. Geophys.Res.-Atmos., 100,  
19 16595-16603, 1995.

20 Hsu, S.-C., Lin, F.-J., and Jeng, W.-L.: Seawater solubility of natural and anthropogenic metals  
21 within ambient aerosols collected from Taiwan coastal sites, Atmos. Environ., 39, 3989-4001,  
22 doi:10.1016/j.atmosenv.2005.03.033, 2005.

23 Hsu, S.-C., Wong, G. T. F., Gong, G.-C., Shiah, F.-K., Huang, Y.-T., Kao, S.-J., Tsai, F., Candice  
24 Lung, S.-C., Lin, F.-J., Lin, I. I., Hung, C.-C., and Tseng, C.-M.: Sources, solubility, and dry  
25 deposition of aerosol trace elements over the East China Sea, March Chem., 120, 116-127,  
26 doi:10.1016/j.marchem.2008.10.003, 2010.

27 Hsu, S.-C., Lin, F.-J., Liu, T.-H., Lin, S.-H., Kao, S.-J., Tseng, C.-M., and Huang, C.-H.: Short time  
28 dissolution kinetics of refractory elements Fe, Al, and Ti in Asian outflow-impacted marine aerosols

1 and implications, *Atmos. Environ.*, 79, 93-100, doi:10.1016/j.atmosenv.2013.06.037, 2013.

2 Hu, X., Zhang, Y., Ding, Z., Wang, T., Lian, H., Sun, Y., and Wu, J.: Bioaccessibility and health risk  
3 of arsenic and heavy metals (Cd, Co, Cr, Cu, Ni, Pb, Zn and Mn) in TSP and PM<sub>2.5</sub> in Nanjing,  
4 China, *Atmos. Environ.*, 57, 146-152, <http://dx.doi.org/10.1016/j.atmosenv.2012.04.056>, 2012.

5 Husain, L., Ghauri, B., Yang, K., Khan, A. R., and Rattigan, O. V.: Application of the SO<sub>4</sub><sup>2-</sup>/Se  
6 tracer technique to study SO<sub>2</sub> oxidation in cloud and fog on a time scale of minutes, *Chemosphere*,  
7 54, 177-183, doi:10.1016/s0045-6535(03)00531-9, 2004.

8 Jakob, A., Stucki, S., and Kuhn, P.: Evaporation of Heavy Metals during the Heat Treatment of  
9 Municipal Solid Waste Incinerator Fly Ash, *Environ. Sci. Technol.*, 29, 2429-2436,  
10 doi:10.1021/es00009a040, 1995.

11 Jickells, T. D., An, Z. S., Andersen, K. K., Baker, A. R., Bergametti, G., Brooks, N., Cao, J. J., Boyd,  
12 P. W., Duce, R. A., Hunter, K. A., Kawahata, H., Kubilay, N., laRoche, J., Liss, P. S., Mahowald, N.,  
13 Prospero, J. M., Ridgwell, A. J., Tegen, I., and Torres, R.: Global iron connections between desert  
14 dust, ocean biogeochemistry, and climate, *Science*, 308, 67-71, doi:10.1126/science.1105959, 2005.

15 Kaufman, Y. J., Tanré D., and Boucher, O.: A satellite view of aerosols in the climate system, *Nature*,  
16 419, 215-223, 2002.

17 Li, W., Wang, Y., Collett, J. L., Jr., Chen, J., Zhang, X., Wang, Z., and Wang, W.: Microscopic  
18 evaluation of trace metals in cloud droplets in an Acid precipitation region, *Environ. Sci. Technol.*,  
19 47, 4172-4180, doi:10.1021/es304779t, 2013.

20 Li, W., Chi, J., Shi, Z., Wang, X., Chen, B., Wang, Y., Li, T., Chen, J., Zhang, D., Wang, Z., Shi, C.,  
21 Liu, L., and Wang, W.: Composition and hygroscopicity of aerosol particles at Mt. Lu in South China:  
22 Implications for acid precipitation, *Atmos. Environ.*, 94, 626-636,  
23 doi:10.1016/j.atmosenv.2014.06.003, 2014.

24 Mackie, D. S., Peat, J. M., McTainsh, G. H., Boyd, P. W., and Hunter, K. A.: Soil abrasion and eolian  
25 dust production: Implications for iron partitioning and solubility, *Geochem. Geophys. Geosyst.*, 7,  
26 1-11, doi:10.1029/2006gc001404, 2006.

27 Mahowald, N.: Aerosol indirect effect on biogeochemical cycles and climate, *Science*, 334, 794-796,  
28 doi:10.1126/science.1207374, 2011.

1 Moreno, T., Querol, X., Alastuey, A., Reche, C., Cusack, M., Amato, F., Pandolfi, M., Pey, J.,  
2 Richard, A., Prévôt, A. S. H., Furger, M., and Gibbons, W.: Variations in time and space of trace  
3 metal aerosol concentrations in urban areas and their surroundings, *Atmos. Chem. Phys.*, 11,  
4 9415-9430, doi:10.5194/acp-11-9415-2011, 2011.

5 Newman, L.: Atmospheric oxidation of sulfur dioxide: A review as viewed from power plant and  
6 smelter plume studies, *Atmos. Environ.* (1967), 15, 2231-2239, doi:10.1016/0004-6981(81)90255-9,  
7 1981.

8 Oakes, M., Ingall, E. D., Lai, B., Shafer, M. M., Hays, M. D., Liu, Z. G., Russell, A. G., and Weber,  
9 R. J.: Iron solubility related to particle sulfur content in source emission and ambient fine particles,  
10 *Environ. Sci. Technol.*, 46, 6637-6644, doi:10.1021/es300701c, 2012.

11 Polissar, A. V., Hopke, P. K., and Harris, J. M.: Source regions for atmospheric aerosol measured at  
12 Barrow, Alaska, *Environ. Sci. Technol.*, 35, 4214-4226, 2001.

13 Querol, X., Viana, M., Alastuey, A., Amato, F., Moreno, T., Castillo, S., Pey, J., de la Rosa, J.,  
14 Sánchez de la Campa, A., Artíñano, B., Salvador, P., García Dos Santos, S., Fernández-Patier, R.,  
15 Moreno-Grau, S., Negral, L., Minguillón, M. C., Monfort, E., Gil, J. I., Inza, A., Ortega, L. A.,  
16 Santamaría, J. M., and Zabalza, J.: Source origin of trace elements in PM from regional background,  
17 urban and industrial sites of Spain, *Atmos. Environ.*, 41, 7219-7231,  
18 doi:10.1016/j.atmosenv.2007.05.022, 2007.

19 Quispe, D., Pérez-López, R., Silva, L. F. O., and Nieto, J. M.: Changes in mobility of hazardous  
20 elements during coal combustion in Santa Catarina power plant (Brazil), *Fuel*, 94, 495-503,  
21 <http://dx.doi.org/10.1016/j.fuel.2011.09.034>, 2012.

22 Reff, A., Bhave, P., Simon, H., Pace, T. G., Pouliot, G. A., Mobley, J. D., and Houyoux, M.:  
23 Emissions Inventory of PM<sub>2.5</sub> Trace Elements across the United States, *Environ. Sci. Technol.*, 43,  
24 5790–5796, doi:10.1021/es802930x, 2009.

25 Ren, L., Wang, W., Wang, Q., Yang, X., and Tang, D.: Comparison and trend study on acidity and  
26 acidic buffering capacity of particulate matter in China, *Atmos. Environ.*, 45, 7503-7519,  
27 doi:10.1016/j.atmosenv.2010.08.055, 2011.

28 Rubasinghege, G., Lentz, R. W., Scherer, M. M., and Grassian, V. H.: Simulated atmospheric

1 processing of iron oxyhydroxide minerals at low pH: roles of particle size and acid anion in iron  
2 dissolution, *Proc. Natl. Acad. Sci. U.S.A.*, 107, 6628-6633, doi:10.1073/pnas.0910809107, 2010.

3 Schulz, M., Prospero, J. M., Baker, A. R., Dentener, F., Ickes, L., Liss, P. S., Mahowald, N. M.,  
4 Nickovic, S., Garcia-Pando, C. P., Rodriguez, S., Sarin, M., Tegen, I., and Duce, R. A.: Atmospheric  
5 transport and deposition of mineral dust to the ocean: implications for research needs, *Environ. Sci.*  
6 *Technol.*, 46, 10390-10404, doi:10.1021/es300073u, 2012.

7 Schwab, J. J.: Aerosol chemical composition in New York state from integrated filter samples:  
8 Urban/rural and seasonal contrasts, *J. Geophys. Res.*, 109, D16S05, doi:10.1029/2003jd004078,  
9 2004.

10 Seinfeld, J. H., and Pandis, S. N.: Atmospheric chemistry and physics: from air pollution to climate  
11 change, John Wiley & Sons, New Jersey, 2012.

12 Shafer, M. M., Perkins, D. A., Antkiewicz, D. S., Stone, E. A., Quraishi, T. A., and Schauer, J. J.:  
13 Reactive oxygen species activity and chemical speciation of size-fractionated atmospheric particulate  
14 matter from Lahore, Pakistan: an important role for transition metals, *J. Environ. Monit.*, 12, 704-715,  
15 2010.

16 Shi, Z., Krom, M. D., Bonneville, S., Baker, A. R., Jickells, T. D., and Benning, L. G.: Formation of  
17 iron nanoparticles and increase in iron reactivity in mineral dust during simulated cloud processing,  
18 *Environ. Sci. Technol.*, 43, 6592-6596, 2009.

19 Solmon, F., Chuang, P. Y., Meskhidze, N., and Chen, Y.: Acidic processing of mineral dust iron by  
20 anthropogenic compounds over the north Pacific Ocean, *J. Geophys. Res.*, 114, D02305,  
21 doi:10.1029/2008jd010417, 2009.

22 Spokes, L. J., Jickells, T. D., and Lim, B.: Solubilisation of aerosol trace metals by cloud processing:  
23 A laboratory study, *Geochim. Cosmochim. Acta*, 58, 3281-3287,  
24 [http://dx.doi.org/10.1016/0016-7037\(94\)90056-6](http://dx.doi.org/10.1016/0016-7037(94)90056-6), 1994.

25 Takahashi, Y., Higashi, M., Furukawa, T., and Mitsunobu, S.: Change of iron species and iron  
26 solubility in Asian dust during the long-range transport from western China to Japan, *Atmos. Chem.*  
27 *Phys.*, 11, 11237-11252, doi:10.5194/acp-11-11237-2011, 2011.

28 Tian, H., Gao, J., Lu, L., Zhao, D., Cheng, K., and Qiu, P.: Temporal trends and spatial variation

1 characteristics of hazardous air pollutant emission inventory from municipal solid waste incineration  
2 in China, *Environ. Sci. Technol.*, 46, 10364-10371, doi:10.1021/es302343s, 2012.

3 Tian, H., Liu, K., Zhou, J., Lu, L., Hao, J., Qiu, P., Gao, J., Zhu, C., Wang, K., and Hua, S.:  
4 Atmospheric emission inventory of hazardous trace elements from China's coal-fired power  
5 plants--temporal trends and spatial variation characteristics, *Environ. Sci. Technol.*, 48, 3575-3582,  
6 doi:10.1021/es404730j, 2014.

7 Wang, Y., Zhang, X., and Draxler, R. R.: TrajStat: GIS-based software that uses various trajectory  
8 statistical analysis methods to identify potential sources from long-term air pollution measurement  
9 data, *Environ. Model. Soft.*, 24, 938-939, 2009.

10 Wei, F., Chen, J., Wu, Y., and Zheng, C.: Study of the background contents of 61 elements of soils in  
11 China, *Environmental Science (in Chinese)*, 12 (4), 12-19, 1991.

12 Weichenthal, S., Villeneuve, P. J., Burnett, R. T., van Donkelaar, A., Martin, R. V., Jones, R. R.,  
13 DellaValle, C. T., Sandler, D. P., Ward, M. H., and Hoppin, J. A.: Long-term exposure to fine  
14 particulate matter: association with nonaccidental and cardiovascular mortality in the agricultural  
15 health study cohort, *Environ. Health Perspect.*, 122, 609-615, doi:10.1289/ehp.1307277, 2014.

16 Wen, H., and Carignan, J.: Review on atmospheric selenium: Emissions, speciation and fate, *Atmos.*  
17 *Environ.*, 41, 7151-7165, doi:10.1016/j.atmosenv.2007.07.035, 2007.

18 Werner, M. L., Nico, P. S., Marcus, M. A., and Anastasio, C.: Use of Micro-XANES to Speciate  
19 Chromium in Airborne Fine Particles in the Sacramento Valley, *Environ. Sci. Technol.*, 41,  
20 4919-4924, doi:10.1021/es070430q, 2007.

21 Xu, H. M., Cao, J. J., Ho, K. F., Ding, H., Han, Y. M., Wang, G. H., Chow, J. C., Watson, J. G., Khol,  
22 S. D., Qiang, J., and Li, W. T.: Lead concentrations in fine particulate matter after the phasing out of  
23 leaded gasoline in Xi'an, China, *Atmos. Environ.*, 46, 217-224, doi:10.1016/j.atmosenv.2011.09.078,  
24 2012.

25 Yang, F., Tan, J., Zhao, Q., Du, Z., He, K., Ma, Y., Duan, F., Chen, G., and Zhao, Q.: Characteristics  
26 of PM<sub>2.5</sub> speciation in representative megacities and across China, *Atmos. Chem. Phys.*, 11,  
27 5207-5219, doi:10.5194/acp-11-5207-2011, 2011.

28 Yang, Y., Wang, Y., Wen, T., Li, W., Zhao, Y., and Li, L.: Elemental composition of PM<sub>2.5</sub> and PM<sub>10</sub>

1 at Mount Gongga in China during 2006, Atmos. Res., 93, 801-810,  
2 doi:10.1016/j.atmosres.2009.03.014, 2009a.

3 Yang, Y., Wang, Y., Wen, T., Zhao, Y., and Li, J.: Element Characterisitcs and Sources of PM2.5 at  
4 Mount Dinghu in 2006, Environmental Science (in Chinese), 30, 988-992, 2009b.

5 Zhang, Q., Streets, D. G., Carmichael, G. R., He, K., Huo, H., Kannari, A., Klimont, Z., Park, I.,  
6 Reddy, S., and Fu, J.: Asian emissions in 2006 for the NASA INTEX-B mission, Atmos. Chem.  
7 Phys., 9, 5131-5153, 2009.

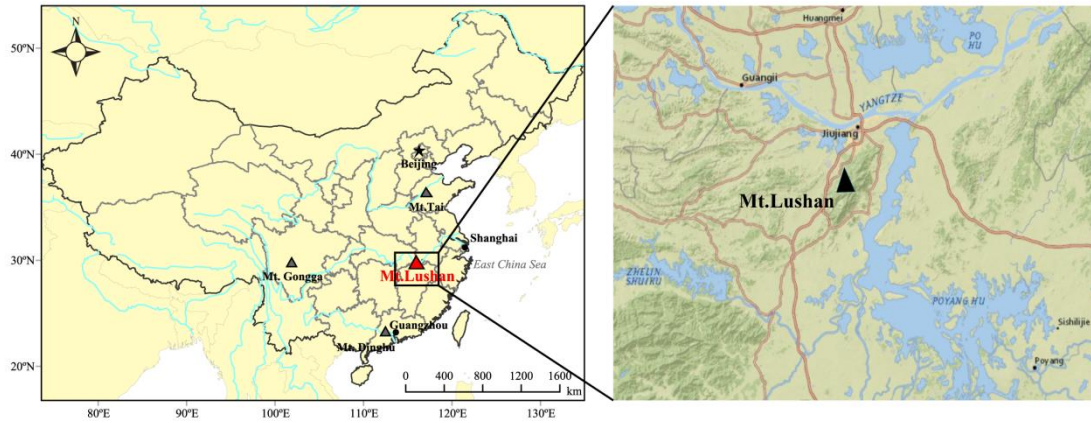
8  
9

1 **Table 1.** Concentrations (mean  $\pm$  SD) of PM<sub>2.5</sub> ( $\mu\text{g m}^{-3}$ ) and trace elements (ng m<sup>-3</sup>) including total  
2 and water soluble fraction at Mt. Lushan.

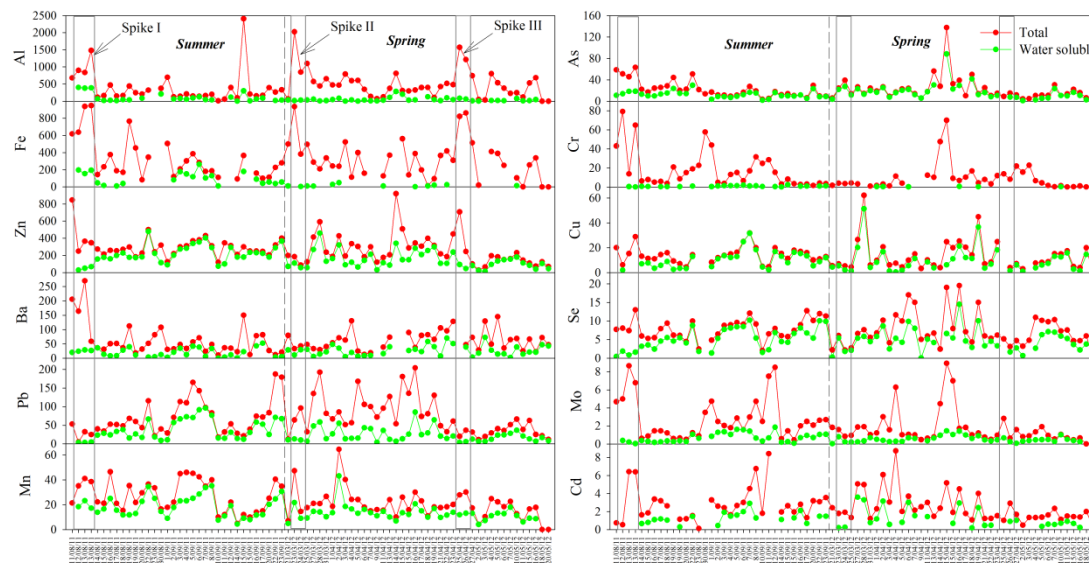
	Overall		Summer,2011		Spring,2012	
	Total	Water soluble	Total	Water soluble	Total	Water soluble
PM <sub>2.5</sub>	55.2 $\pm$ 20.1		55.9 $\pm$ 21.8		54.7 $\pm$ 18.9	
Al	449.1 $\pm$ 441.1	73.3 $\pm$ 93.5	369.1 $\pm$ 464.4	108.0 $\pm$ 121.7	515.1 $\pm$ 415.2	48.9 $\pm$ 57.3
Fe	331.1 $\pm$ 236.2	71.5 $\pm$ 70.5	330.4 $\pm$ 250.9	101.9 $\pm$ 70.5	331.6 $\pm$ 227.3	15.7 $\pm$ 12.9
Zn	258.3 $\pm$ 162.8	172.9 $\pm$ 105.8	274.3 $\pm$ 129.5	211.9 $\pm$ 104.4	245.3 $\pm$ 186.1	142.3 $\pm$ 97.5
Pb	68.2 $\pm$ 49.3	29.4 $\pm$ 23.8	65.4 $\pm$ 47.3	37.8 $\pm$ 27.5	70.5 $\pm$ 51.4	22.7 $\pm$ 18.3
Ba	63.8 $\pm$ 54.2	23.4 $\pm$ 17.0	66.0 $\pm$ 71.1	19.3 $\pm$ 15.0	61.9 $\pm$ 33.6	26.7 $\pm$ 18.0
Mn	22.2 $\pm$ 12.2	15.5 $\pm$ 8.1	24.7 $\pm$ 12.5	19.0 $\pm$ 8.8	20.2 $\pm$ 11.6	12.8 $\pm$ 6.3
As	21.5 $\pm$ 19.6	14.8 $\pm$ 11.9	22.3 $\pm$ 16.3	12.4 $0\pm 6.1$	20.9 $\pm$ 22.2	17.0 $\pm$ 15.1
Cr	13.7 $\pm$ 17.2	0.7 $\pm$ 0.5	18.2 $\pm$ 19.5	0.8 $\pm$ 0.6	9.6 $\pm$ 13.8	0.3 $\pm$ 0.1
Cu	12.4 $\pm$ 9.6	9.9 $\pm$ 8.9	13.2 $\pm$ 6.4	10.7 $\pm$ 6.9	11.7 $\pm$ 11.6	9.3 $\pm$ 10.2
Se	7.0 $\pm$ 3.3	5.0 $\pm$ 2.9	7.6 $\pm$ 2.9	5.2 $\pm$ 2.9	6.5 $\pm$ 3.5	4.8 $\pm$ 2.9
Cd	2.5 $\pm$ 1.8	1.3 $\pm$ 0.9	2.8 $\pm$ 1.9	1.3 $\pm$ 0.6	2.2 $\pm$ 1.6	1.2 $\pm$ 1.1
Mo	2.0 $\pm$ 2.0	0.6 $\pm$ 0.4	2.6 $\pm$ 2.2	0.7 $\pm$ 0.5	1.5 $\pm$ 1.7	0.5 $\pm$ 0.3

3

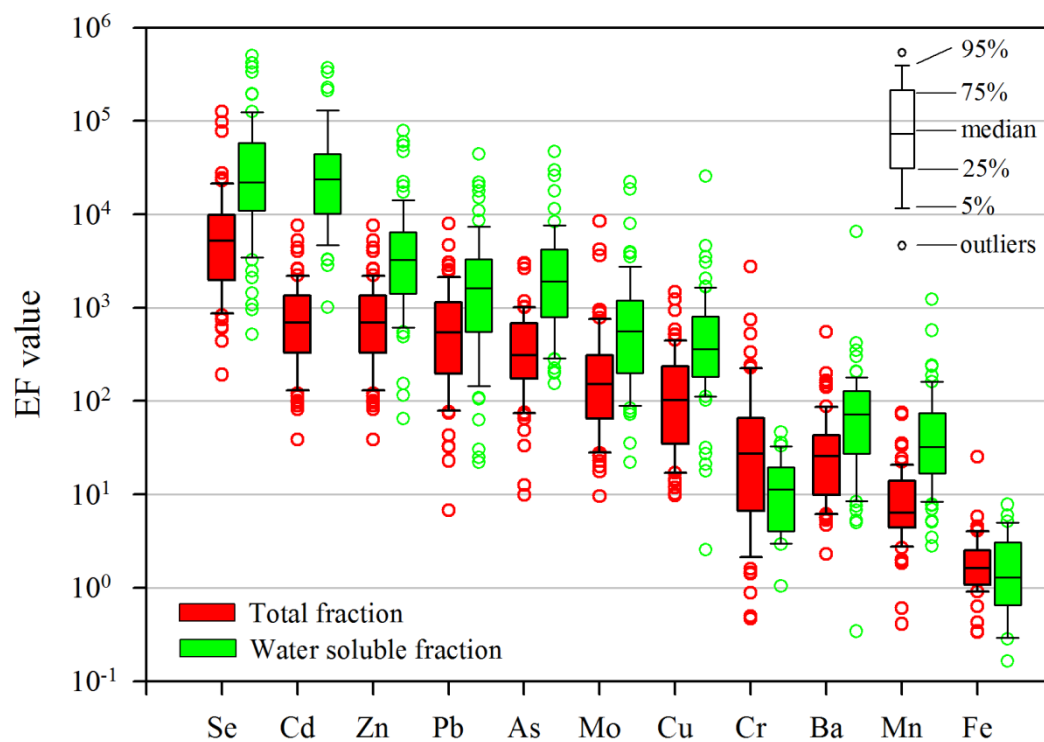




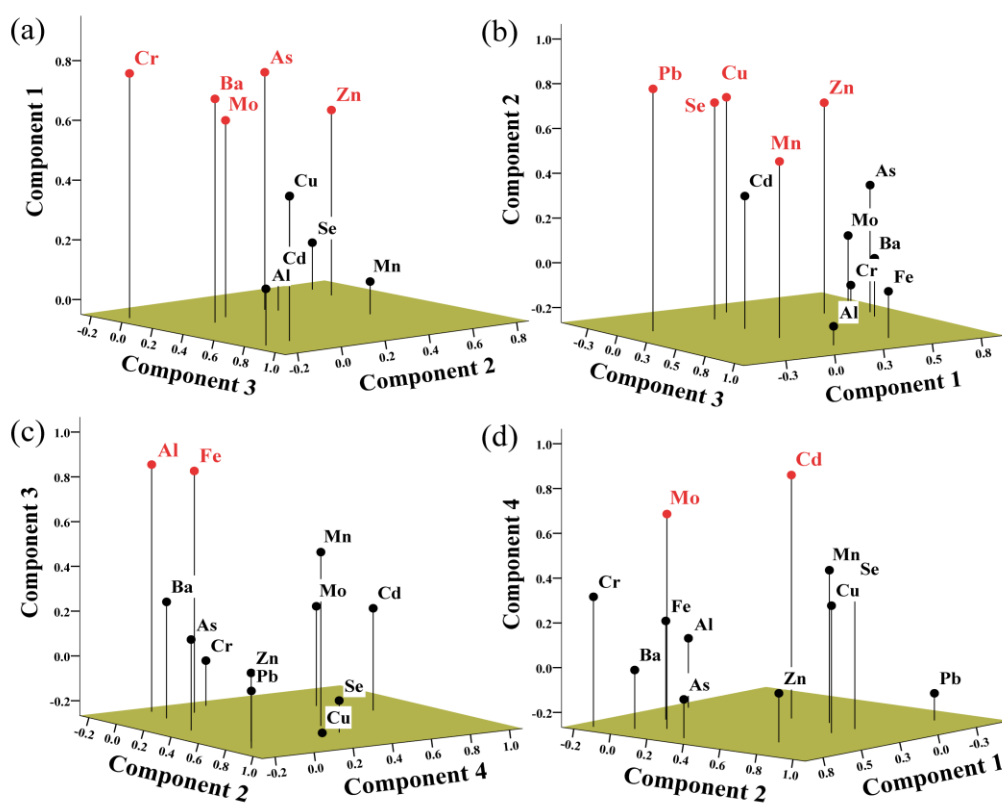
**Figure 1.** The location of Mt. Lushan and adjacent Yangtze River and Poyang Lake.



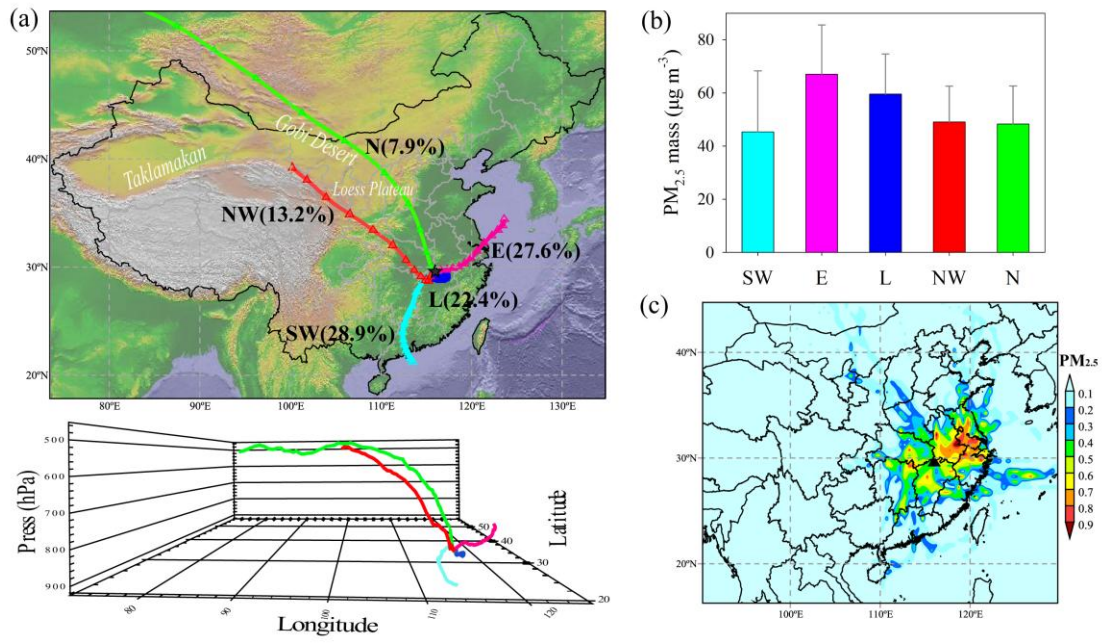
**Figure 2.** Temporal variations of individual trace element concentrations ( $\text{ng m}^{-3}$ ) in  $\text{PM}_{2.5}$  at Mt. Lushan. Boxes indicate the concentration spikes.



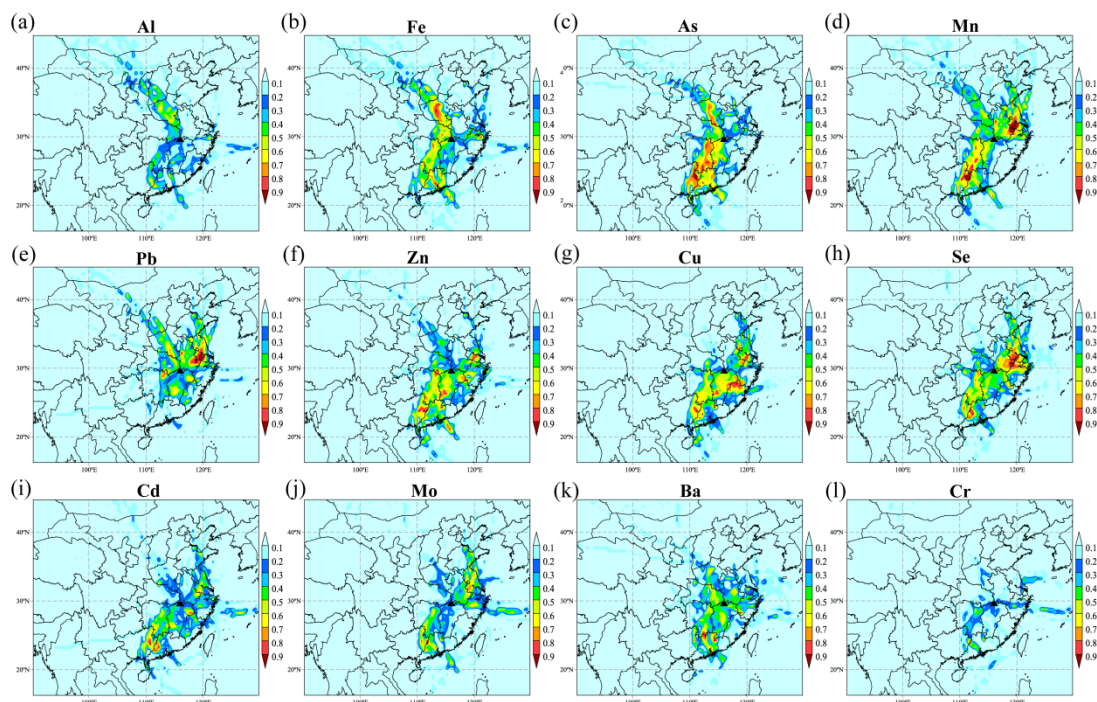
**Figure 3.** Enrichment factor (EF) values of fine particle trace elements at Mt. Lushan in total and water soluble fraction.



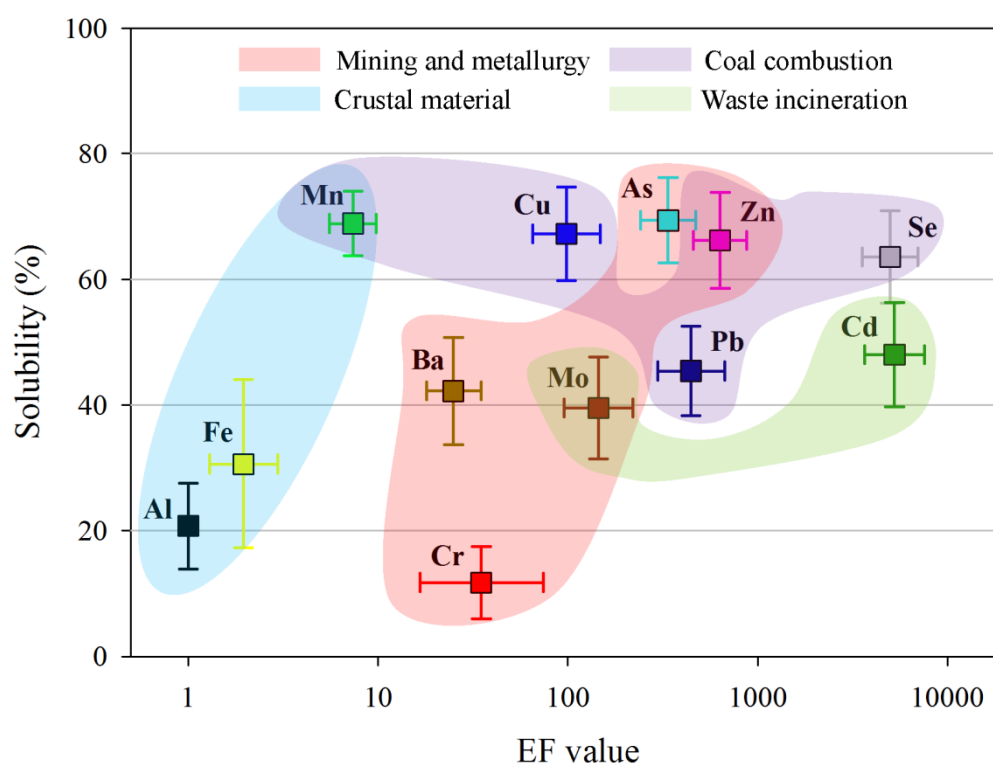
**Figure 4.** Loadings of (a) Component 1, (b) Component 2, (c) Component 3 and (d) Component 4 for trace elements in total fraction in PM<sub>2.5</sub>.



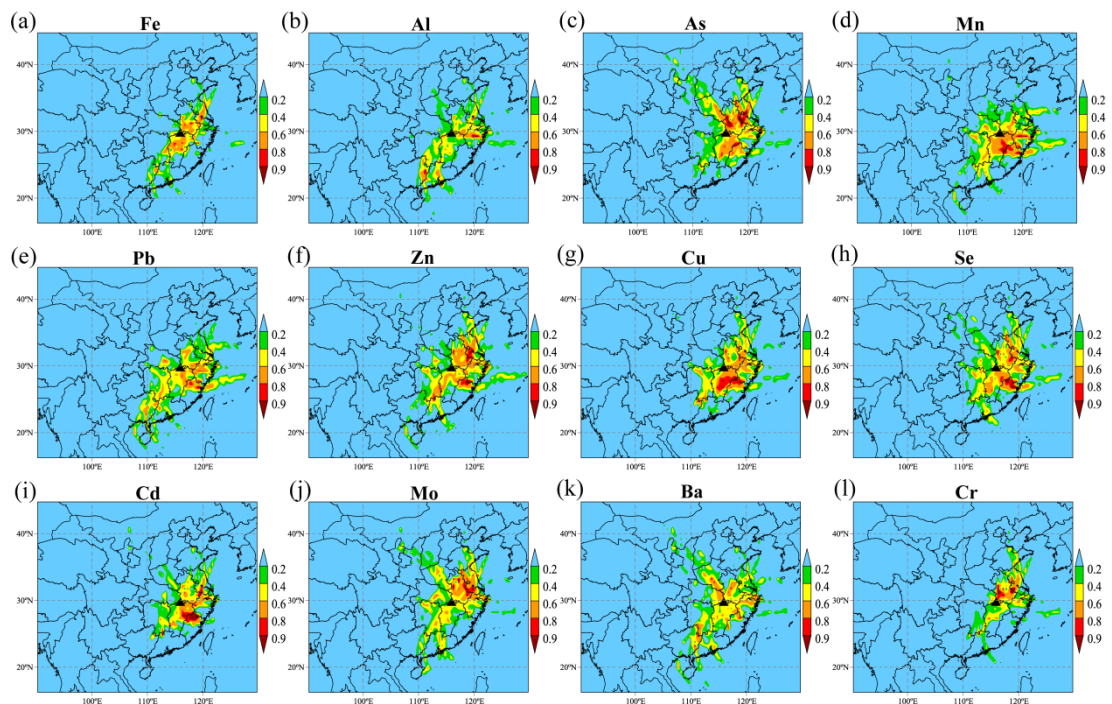
**Figure 5.** (a) Three-days mean trajectories arriving at Mt. Lushan, (b) averaged  $PM_{2.5}$  concentration of five clusters, and (c) the likely source regions of  $PM_{2.5}$  identified using PSCF plots during observation period.



**Figure 6.** Likely source regions of individual elements in  $PM_{2.5}$  at Mt. Lushan, identified using PSCF plots. Note the criteria are 75th percentile for Al and mean concentration for others.

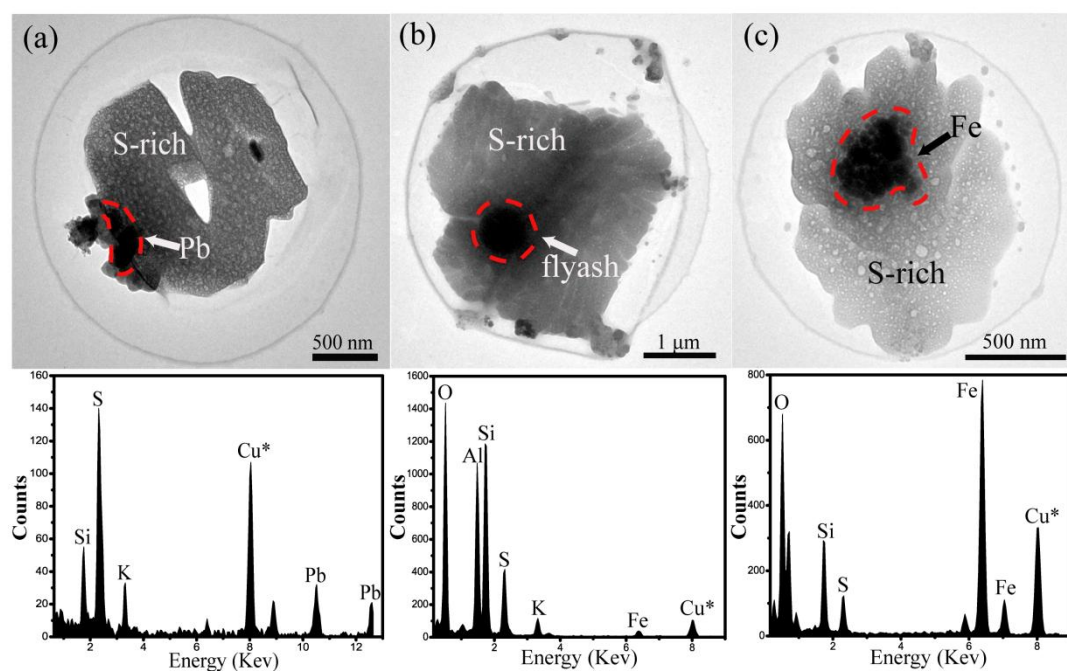


**Figure 7.** Scatter plot of aerosol element solubility vs. logarithmic EF value, computed with geometric mean in 99% confidence interval. Colored shade areas represent different sources classified by EF value and PCA analysis.



**Figure 8.** Solubility distributions in likely source regions for individual elements in PM<sub>2.5</sub> at Mt. Lushan, identified by PSCF with the mean solubility values as criteria.





**Figure 9.** Typical TEM images and corresponding EDS spectra of metal particles embedded in individual S-rich cloud residues collected during cloud events. (a) Pb-S. (b) fly ash-S. (c) Fe-S. The dotted red circles indicate the examined area of EDS. Cu\* signals result from copper TEM grid.