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Atmospheric chemistry of nitrogenous aerosols in Northeast Asia: biological sources and secondary formation

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Abstract

To better understand the sources of nitrogenous aerosols, particularly water-soluble organic nitrogen (WSON) and water-insoluble organic nitrogen (WION), in Northeast Asia, we measured total nitrogen (TN) and water-soluble total nitrogen (WSTN) as well as nitrogen isotope ratios (δ^{15} N) of TN (δ^{15} N_{TN}) and WSTN (δ^{15} N_{WSTN}) in the total suspended particles (TSP) collected from Sapporo, northern Japan for one-year period. In general, WION was more abundant $(126 \pm 117 \text{ ngm}^{-3})$ whereas WSON $(89.7\pm80.6 \text{ ng m}^{-3})$, accounting for 14 ± 11 % and 9.2 ± 7.3 % of TN, respectively. WSON peaked in late autumn to winter (maximum 288 ng m^{-3}) and WION peaked in mid spring to early summer (454 ng m⁻³). δ^{15} N_{TN} (21.9 ± 4.1%) and δ^{15} N_{WSTN} (25.8 ± 8.2%) 10 showed peaks in summer with relatively high ratios in late autumn. Based on the seasonal variations of WSON and WION together with organic tracers, fossil fuel combustion and biomass burning are found to be two major sources of WSON whereas emissions of biological particles and secondary formation by reactions of biogenic secondary organic species (carbonyls) with NH₃ are suggested as important source of 15 WION. The seasonality of $\delta^{15}N_{TN}$ and $\delta^{15}N_{WSTN}$, together with the comparisons to literature values, implies that chemical aging (including gas/particle partitioning) and biomass burning are the causes of the enhanced values in summer and autumn, respectively. This study demonstrates that contributions of aerosol N from fossil fuel com-

²⁰ bustion and biomass burning dominate in autumn/winter whereas emission of terrestrial biological particles and secondary formation from biogenic hydrocarbons and subsequent chemical aging in the atmosphere are important in spring/summer in Northeast Asia.

1 Introduction

²⁵ In East Asia, high loading of aerosol nitrogen (N) and its significant deposition onto the Earth surface are associated with enhanced agricultural usage of nitrogenous fer-



tilizer: 40 % of vegetation receives nitrogen deposition in excess of 1000 mg N m⁻² yr⁻¹ (threshold level) and even more (~ 50 %) in Japan (Dentener et al., 2006; Liu et al., 2013; Morino et al., 2011). Emissions of NH₃ and NO_x to the air have an impact on atmospheric chemistry, visibility impairment, human health and the Earth's climate system (Rollins et al., 2012; Vitousek et al., 1997). Further, N deposition over the Earth surface has large influence on terrestrial and aquatic ecosystems via the changes in the global biogeochemical cycles (Duce et al., 2008; Galloway et al., 2004; Kim et al., 2014). Most studies of aerosol N in East Asia and other regions in the world have been

focused on inorganic species (NO₃⁻ and NH₄⁺) (Dentener et al., 2006; Duce et al., 2008;
Liu et al., 2013). Although organic nitrogen (ON) represents a significant fraction (up to 80%) of total aerosol N and may play a critical role in biogeochemical cycles, ON has rarely been studied (Cape et al., 2011; Matsumoto et al., 2014; Zhang et al., 2008).

Aerosol ON is comprised of a wide range of nitrogenous compounds from semivolatile amines to proteins and macromolecules (Cape et al., 2011; Laskin et al., 2009;

- ¹⁵ Wang et al., 2010). In addition to biomass burning, the primary sources including biological particles emitted from soil, vegetation, pollen, bacteria and the ocean surface are important (Cape et al., 2011; Jickells et al., 2013; Miyazaki et al., 2014; Neff et al., 2002). ON can also be produced in the atmosphere via the reactions of NO₃ radical or NH₃ with biogenic/anthropogenic volatile organic compounds (VOCs) followed by gas-
- to-particle conversion (Cornell et al., 2001; Miyazaki et al., 2014; Pratt et al., 2012). Recently, oxidation of isoprene and monoterpenes by NO₃ is considered as a substantial source of organic nitrates and secondary organic aerosols (SOA) (Fry et al., 2009; Pratt et al., 2012; Rollins et al., 2009, 2012). Further, chemical aging of SOA derived from both biogenic and anthropogenic precursors in the presence of NH₃ in gas phase
- and NH⁺₄ salts in aqueous phase can produce light-absorbing nitrogenous compounds (defined as brown carbon) (Bones et al., 2010; Updyke et al., 2012).

However, the relative importance of anthropogenic and biogenic emissions including biomass burning is largely unknown and secondary formation of ON has been poorly characterized (Cape et al., 2011; Jickells et al., 2013; Kanakidou et al., 2012) and thus



land-atmosphere interactions of aerosol N are not fully understood yet because of limited measurements of aerosol ON. Previous studies of ON have focused on watersoluble ON (WSON) but very little is known about the abundances and sources of water-insoluble ON (WION) particularly in continental aerosols. Interestingly, WION is

- $_{5}$ much more abundant (on average 34.0 nmolNm⁻³) than WSON (4.5 nmolNm⁻³) in aerosols from Lewes, Delaware on the mid-Atlantic US coast (Russell et al., 2003) and marine aerosols (55 ± 16 % of total N) over the western North Pacific (Miyazaki et al., 2011). Hence, it is of interest to investigate the seasonal variations of WSON and WION.
- Here, we present total N (TN), water-soluble TN (WSTN), WSON and WION and 10 nitrogen isotope ratios (δ^{15} N) of TN (δ^{15} N_{TN}) and WSTN (δ^{15} N_{WSTN}) measured in ambient aerosols (TSP) collected from Sapporo, northern Japan over one-year period. The isotope studies are helpful in understanding the origins and atmospheric processing of N (Kawamura et al., 2004; Pavuluri et al., 2010). Based on WSON and WION data together with various organic tracers, δ^{15} N values and air mass trajectories as 15 well as the comparison with previous studies, we discuss the atmospheric chemistry

of nitrogenous aerosols and their possible sources with a major focus on WSON and WION in Northeast Asia.

Materials and methods 2

2.1 Aerosol sampling 20

Total suspended particles (TSP) were collected from Sapporo in the western part of Hokkaido Island, northern Japan (43.07° N, 141.36° E), an ideal location for collecting the air masses delivered from Northeast Asian regions: Siberia, the Russian Far East, China, and surrounding oceans (Aggarwal and Kawamura, 2008; Yamamoto et al.,

2011). The geographic details of the sampling site are described elsewhere (Pavuluri 25 et al., 2013). TSP sampling (n = 21) was performed on the rooftop of the three-story



building of the Institute of Low Temperature Science (ILTS), Hokkaido University, Sapporo during 2 September 2009 and 5 October 2010 using a pre-combusted (450 °C, 4 h) quartz fiber filter and high-volume ($\sim 65 \text{ m}^3 \text{ h}^{-1}$) air sampler. Each sample was collected for ca. 2 consecutive weeks in order to obtain sufficient amount of carbon for radiocarbon (¹⁴C) analyses at organic molecular levels, another objective of this research. Filter samples were placed in a pre-combusted glass jar with a Teflon-lined screw cap and stored in a dark room at -20 °C prior to analysis.

It should be noted that aerosol samples collected on quartz fiber filters might have positive (adsorption of gaseous HNO_3 and NH_3 and WSON) and negative (evaporation

- ¹⁰ of particulate NH_4^+ salts such as NH_4NO_3 and WSON) sampling artifacts (Matsumoto et al., 2014; Squizzato et al., 2013). Since the sampling time is longer in this study, the evaporative loss from the particles should be more significant than the adsorbed gases by quartz fiber filter and thus the reported concentrations may be underestimated. However, the ambient temperatures encountered in Sapporo are rather low
- ¹⁵ (range of averages of each sample period: -3.30 to 24.5 °C) (Pavuluri et al., 2015) that may not cause a significant evaporative loss of N species during the sampling period, because quartz filters show a good retention for semi-volatile NH₄NO₃ sampled at an air temperature up to 21 and ~ 30 % at even an elevated air temperature of 35 °C (Schaap et al., 2004). Therefore, we believe that our sampling technique does not have serious sampling artifacts even in summer.

2.2 Chemical analyses

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2.2.1 Determination of TN and WSTN and their N isotope ratios

TN (and WSTN) contents and their isotope ratios ($\delta^{15}N_{TN}$ and $\delta^{15}N_{WSTN}$, respectively) in TSP samples were determined using an elemental analyzer (EA) (Carlo Erba NA 1500) and EA/isotope ratio mass spectrometer (IRMS) (Finnigan MAT Delta Plus), respectively, as described elsewhere (Kawamura et al., 2004; Pavuluri et al., 2010). An



Q water (> 18.3 MΩ) under ultrsonication for 20 min and filtrated with a syringe filter (GL Sciences Chromatodisc Type A, 0.45 µm). The extracts were adjusted to pH 8 ~ 9 with 0.05 M KOH and concentrated to dryness using a rotary evaporator under vacuum and then re-dissolved in 200 µL Milli Q water. An aliquot of filter sample (1.0 cm diam-

eter disc for TN and 50 μL of water extract adsorbed onto ~ 20 mg of pre-combusted Chromosorb for WSTN) was placed in a pre-cleaned tin cup and introduced into EA. The resulting N₂ gas was measured with thermal conductivity detector in EA and transferred to IRMS via an interface (ConFlo II) for ¹⁵N/¹⁴N measurement. The analytical uncertainties in duplicate analyses of filter samples were within 6% for TN, 3% for WSTN, 0.28‰ for δ¹⁵N_{TN} and 1.34‰ for δ¹⁵N_{WSTN}.

We measured WSTN using a total organic carbon (TOC)/total nitrogen (TN) analyzer (Shimadzu TOC-Vcsh), as reported by Miyazaki et al. (2011). Briefly, an aliquot of filter (1.4 cm in diameter disc) was extracted with 10 mL Milli Q water under ultrasonication for 20 min. The extracts were filtered with syringe filter (GL Sciences Chromatodisc

¹⁵ Type A, 0.45 μm) and then injected into TOC/TN analyzer. The analytical uncertainty in duplicate analyses of filter samples was within 5 %.

Concentrations of WSTN measured by EA are lower by $23 \pm 5\%$ than those measured by TOC/TN analyzer. The difference was significant when the concentration of WSTN was beyond $0.5 \,\mu g m^{-3}$. The difference may be arisen from different combustion

- temperature: 1400 °C for EA vs. 680 °C for TOC/TN analyzer and different procedures. We consider that the WSTN measured by EA may be underestimated due to possible evaporative loss of N species such as NH₃ and HNO₃ during the concentration step and/or sticking loss of N species on glass walls. Here, we use WSTN measured by TOC/TN analyzer to minimize the uncertainties in the estimation of WSON and WION.
- ²⁵ However, $\delta^{15}N_{WSTN}$ did not show any abnormal deviation in its temporal trend, which is comparable to that of $\delta^{15}N_{TN}$ (see Fig. 2b), suggesting that the impact of WSTN loss on its $\delta^{15}N$ is negligible.



2.2.2 Measurements of inorganic ions

Details in the measurements of inorganic ionic species such as NO_3^- , NH_4^+ and methanesulfonate: MSA⁻ (CH₃SO₃⁻) are described elsewhere (Pavuluri et al., 2015). Briefly, an aliquot of filter sample was extracted with Milli Q water under ultrasonication

and filtrated with a syringe filter (GL Sciences Chromatodisc Type A, 0.45 μm). The filtrates were then injected into an ion chromatograph (761 Compact IC, Metrohm). A calibration curve was prepared with a set of authentic standards to calculate the concentrations for the samples. The analytical precision was within 4%.

2.2.3 Estimation of WSON and WION

¹⁰ Amounts of inorganic N (IN), i.e., sum of NO₃⁻-N and NH₄⁺-N, were calculated from the concentrations of NO₃⁻ and NH₄⁺ ions (Pavuluri et al., 2015). The amounts of ON, WSON and WION were estimated by the following equations (Miyazaki et al., 2011);

ON = TN – IN WSON = WSTN – IN

15 WION = TN – WSTN

2.2.4 Measurements of organic tracers

Organic tracers such as hopanes, isoprene- and monoterpene-oxidaiton products were determined using a capillary gas chromatograph (Hewlett-Packard 6890) coupled to mass spectrometer (Hewlett-Packard 5973) (GC/MS) as described elsewhere (Fu et al., 2010). Briefly, organic tracer compounds were extracted from the filter samples with dichloromethane/methanol (2:1; v/v) under ultrasonication. The hydroxyl and carboxyl functional groups were derivatized to trimethylsilyl (TMS) ethers and esters, respectively, with 50 µL of N,O-bis-(trimethylsilyl)triflouroacetamide (BSTFA) containing 1 % trimethylsilyl chloride and 10 µL of pyridine. The TMS derivatives were then



measured using a GC/MS. The analytical errors in duplicate analyses were less than 10 %.

All the data reported here are corrected for the field blank collected in each season.

3 Results and discussion

5 3.1 Source regions of Sapporo aerosols

Ten-day backward air mass trajectories arriving in Sapporo at 500 m above the ground level (a.g.l.) were computed for every 48 h during each sample period using HYSPLIT model (Draxler and Rolph, 2012) and are plotted in Fig. 1. The trajectories show that the air masses were mostly originated from Siberia passing over Northeast Asian re¹⁰ gions and the Japan Sea during autumn, winter and spring (Fig. 1). In summer, they originated from the East China Sea and/or western North Pacific passing over the coastal region and/or the Main Island of Japan (Fig. 1). The air parcels were mostly travelled at lower than 2000 m a.g.l. in all seasons and hence their advection from distant source regions to the receptor site should be significant. Further, the air masses
¹⁵ arriving in Sapporo during each sampling period (~ 2 week) were mostly originated from the similar source region (Fig. 1), suggesting that only the limited source regions may have influenced each sample despite the relatively long interval of sampling.

In fact, the radiocarbon analyses showed that percent modern carbon (pMC) of total carbon and water-soluble organic carbon in our aerosols started to increase from mid

- to late winter toward spring (Pavuluri et al., 2013), although the growing season starts in May in Hokkaido when daily average temperatures are ≥ 10°C (Toma et al., 2011). Because the contribution of modern carbon from local vegetation is unlikely in winter, such increase in pMC during winter has been mainly attributed to the contributions of biomass burning aerosols via long-range transport from East Asia (Pavuluri et al., 2012).
- ²⁵ 2013). In addition, temporal variations of crustal metals, e.g., Al, Ca and Fe, in Sapporo aerosols are consistent with those of TSP from winter to mid summer with a peak



in early spring, indicating a significant atmospheric transport of Asian dust from arid regions in Mongolia and China (Pavuluri et al., 2015). Therefore, we consider that the Sapporo aerosols should have largely influenced by long-range transported air masses from distant sources over Northeast Asia. However, the characteristics and seasonal
 variations of Sapporo aerosols could reflect the regional scenario and we do not preclude an impact from the local sources during the campaign.

3.2 Total nitrogen (TN) and water-soluble total nitrogen (WSTN)

Concentrations of TN and WSTN ranged from 348 to 1750 ngm⁻³ (average 1020 ± 466 ng m⁻³) and 293 to 15200 ng m⁻³ (893 ± 443 ng m⁻³), respectively, in Sapporo aerosols during the study period (*n* = 21). The mass fractions of TN in TSP ranged from 1.20 to 8.66 % (average 3.81±2.28 %). As shown in Fig. 2a, TN and WSTN did not show clear temporal trends. However, on average they were higher in winter followed by spring, summer and autumn (Table 1). Similarly, TN/TSP mass ratios did not show any clear temporal trend during the study period but showed a drastic increase from 1.2–8.7 %) observed in Sapporo aerosols during the campaign are lower than those (average 5 %, range 2–15 %) reported for urban aerosols from Tokyo (Kawamura et al., 1995), where anthropogenic source contributions are expected to be the major source. However, wintertime TN/TSP mass ratios in Sapporo aerosols (≥ 7.8 %) are close to that reported from urban Tokyo, suggesting that the contributions from anthropogenic

sources such as fossil fuel combustion may be significant in winter.

As seen from Fig. 3, WSTN contains mostly IN (i.e., $\sum NO_3^- N + NH_4^+ - N$) (range 70.8– 100%, average 89.7 ± 7.74%) whereas TN consists significant amount of ON (0.00– 44.7%, 22.7 ± 17.7%) that ranged from 0.00–569 ng m⁻³ (215 ± 150 ng m⁻³) during the study period. High abundances of ON in TN rather than WSTN indicate a significance of WION in Sapporo aerosols. Further, WSTN showed a stronger linear relation with IN ($r^2 = 0.97$) than that ($r^2 = 0.90$) between TN and IN in Sapporo aerosols (Fig. 3). Such deviation in the linearity suggests that the origins and/or secondary formation

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processes of WION (i.e., TN – WSTN) may be different from that of WSTN (including WSON). Therefore, it is apparent that the atmospheric loading and impacts of aerosol N in Northeast Asia is significantly influenced by the sources and seasonality of ON.

- Average mass fraction of ON in TN detected in Sapporo aerosols is lower than that (annual average 37%) reported in remote marine aerosols (TSP), but comparable to that (23%) in springtime fine aerosols ($PM_{1.0}$) from Okinawa Island, Japan, which were considered to be derived from anthropogenic emissions over China and subjected for chemical aging during long-range transport (Kunwar et al., 2015). The mass fraction of ON in TN is close to that (17.7 ± 9.68%) reported in the wintertime tropical Indian
- aerosols from Chennai that were originated from fossil fuel combustion and biomass burning emissions and subjected to significant chemical aging (Pavuluri et al., 2010). However, the loadings of WION in both Okinawa and Chennai aerosols have not been reported. These comparisons suggest that fossil fuel combustion and biomass burning followed by chemical aging may have an influence on ON loadings in Sapporo aerosols.
- ¹⁵ However, the possible contribution of ON, particularly WION, from biological sources is not clear (Miyazaki et al., 2011).

3.3 Water-soluble organic nitrogen (WSON) and water-insoluble organic nitrogen (WION)

3.3.1 Concentrations and mass fractions in TN

²⁰ Concentrations of WSON and WION in Sapporo aerosols ranged from 0.00 to 288 ng m⁻³ and 0.00 to 454 ng m⁻³, respectively. On average, WION is more abundant (126 ± 117 ng m⁻³) than WSON (89.7 ± 80.6 ng m⁻³). Although the temporal variations did not show any clear seasonal pattern, WSON maximized in late autumn to mid winter whereas WION peaked in mid spring to early summer (i.e., in growing season)
 ²⁵ (Fig. 4a). Further, seasonally averaged concentrations of WSON are higher in winter followed by autumn, spring and summer whereas those of WION are higher in summer



ences are not statistically significant (Table 1). On average, mass fraction of WION in TN is $13.9 \pm 11.1 \%$ (range 0.00-33.1 %) and WSON/TN ratio is $9.13 \pm 7.29 \%$ (0.00-30.2 %) during the study period. Temporal variations of the mass fractions of WSON and WION in TN (Fig. 4b) are similar to those of the concentrations (Fig. 4a).

5 3.3.2 Comparisons with previous studies from different locale

As seen from Table 2, average concentration of WSON in Sapporo aerosols is lower than that reported in urban aerosols from Davis, California and Kofu, Japan as well as from the coastal sites: Erdemli in Turkey, Crete in Greece, and Qingdao in China. It is also lower than that reported in the forest aerosols collected from Rondônia, Brazil during intensive biomass burning period (dry season) (Mace et al., 2003a) and from Sapporo, Japan, and in the marine aerosols over the Asian outflow regions: the Yellow Sea, South China Sea and western North Pacific (Table 2). In contrast, average concentration of WSON in Sapporo aerosols is comparable to that of forest aerosols from Fujiyoshida, Japan, but higher than that from Rondônia, Brazil during wet season.

- It is also higher than that of the marine aerosols from pristine oceanic regions: Cape Grim, Australia (Mace et al., 2003b) and Oahu, Hawaii (Cornell et al., 2001) and over the western North Pacific (Miyazaki et al., 2011) (Table 2). In addition, the higher end (maximum 288 ng m⁻³) of WSON is comparable to average WSON concentration reported in urban aerosols from Davis, California and Kofu, Japan and in coastal urban
- ²⁰ aerosols from Crete, Greece (Table 2). However, average mass fraction of WSON in WSTN in Sapporo aerosols is found to be comparable or close to those reported for urban aerosols from Kofu, Japan, coastal urban aerosols from Crete, Greece, forest aerosols from Fujiyoshida, Japan and the marine aerosols over the Yellow Sea, China Sea and western North Pacific (Table 2).
- Higher concentrations of WSON reported for urban aerosols from Davis, California, during late fall and winter are attributed to increased amounts of atmospheric liquid water, which promote a partition of gaseous WSON to particles (Zhang et al., 2002). In Kofu and Fujiyoshida (forest site), Japan, emissions from combustion sources in-



cluding biomass burning and plant-derived particles as well as secondary formation through the reaction of NO_2 with volatile organics are considered as potential sources of WSON (Matsumoto et al., 2014). In the Mediterranean atmosphere, atmospheric dust from north Africa was expected as dominant source of WSON at Erdemli, Turkey

- 6 (Mace et al., 2003c) whereas in Crete, Greece, fossil fuel combustion and biomass burning are also considered (Violaki and Mihalopoulos, 2010). Over the marginal seas of China (Asian outflow region): Qingdao, the Yellow Sea and Sea of China, anthropogenic sources are considered as dominant origin of WSON with minor contributions from marine and crustal sources (Nakamura et al., 2006; Shi et al., 2010). In the forest
- aerosols from Sapporo, Japan, emissions of biological particles are considered as major source of WSON in autumn whereas secondary formation of WSON from biogenic hydrocarbons is considered as dominant source in summer (Miyazaki et al., 2014).

On the other hand, average concentration of WION $(126 \pm 117 \text{ ng m}^{-3})$ found in Sapporo aerosols is comparable to that $(112 \pm 61 \text{ ng m}^{-3})$ reported in the marine aerosols over the western North Pacific, where the sea-to-air emissions are considered as

- ¹⁵ over the western North Pacific, where the sea-to-air emissions are considered as the major source as well as the secondary production via the reactions of gaseous hydrocarbons with NH₃ (Miyazaki et al., 2011). In addition, the higher end (maximum 454 ngm⁻³) of WION in Sapporo aerosols is comparable to the average WION (476 ngm⁻³) reported in the coastal aerosols from Lewes, Delaware, where biological
- and mineral particles are expected as major source (Russell et al., 2003). Therefore, we consider that WSON in Sapporo aerosols may be mainly derived from anthropogenic emissions including biomass burning, although emissions from biological sources and secondary formation from gaseous hydrocarbon precursors cannot be excluded. Meanwhile, WION may be originated from emissions of biological particles and secondary formation by hydrocarbons.
- ²⁵ formation from biogenic hydrocarbons.



3.4 Comparisons of WSON and WION with organic tracers

3.4.1 Possible sources of WSON

Anthropogenic sources including biomass burning are generally considered as dominant source of WSON, particularly in East Asia (Matsumoto et al., 2014; Nakamura et al., 2006; Shi et al., 2010). It has also been reported that biomass burning aerosols contain significant amounts of N containing organics (particularly N-heterocyclic alkaloids) (Laskin et al., 2009). In Sapporo aerosols, temporal trend of WSON is somewhat similar to that of hopanes (C₂₇-C₃₂ hopanoid hydrocarbons), specific biomarkers of petroleum and coal (Schauer et al., 2002), particularly in autumn and winter (Fig. 4c).

- Further, WSON showed a linear relationship with hopanes with moderate but significant correlation (r = 0.60) during the campaign (Table 3). On the other hand, temporal variation of WSON from mid autumn to mid winter is somewhat similar to that of levoglucosan (Fig. 4d), a tracer for biomass burning (Simoneit, 2002), and showed a weak linear relation throughout the campaign (Table 3). Contributions of WSON to TN are
- ¹⁵ higher in the above seasons (Fig. 4b). In contrast, WSON was abundant in few spring (from mid April to mid May) and summer (late June and late July) samples (Fig. 4a), despite the lower levels of both hopanes and levoglucosan throughout the spring and summer (Fig. 4c, d).

Temporal trend of WSON from late spring to early summer (Fig. 4a) is similar to that of sucrose (Fig. 4e), which is a tracer of pollens emitted from terrestrial higher plants (Fu et al., 2012). The seasonal trend of WSON (Fig. 4a) is also similar to those of biogenic secondary organic aerosols (SOA), i.e., α -pinene-SOA tracers (Kleindienst et al., 2007), and MSA⁻, a tracer of biogenically derived secondary sulfur species (Bates et al., 1992) in late spring to early autumn (Fig. 4f, g). WSON well correlates with sucrose (r = 0.71, $p \le 0.05$), α -pinene SOA tracers (r = 0.61) and MSA⁻ (r = 0.56) during

spring and summer (n = 10), although it does not show any relation with isoprene SOA tracers (r = 0.16) and β -caryophylinic acid (r = 0.19). However, WSON shows only a weak relation with sucrose and no relation with isoprene- and α -pinene-SOA tracers.



ers, β -caryophylinic acid and MSA⁻ when we consider the total data set (n = 21) that include all seasons during the campaign (Table 3).

The above comparisons of WSON with source tracers imply that anthropogenic emissions including biomass burning are major sources of WSON in Sapporo aerosols.

- ⁵ Their contributions to Sapporo aerosols may have been enhanced in autumn and winter when the air masses enriched with forest fire and fossil fuel combustion products are often delivered from Siberia passing over Northeast China (Fig. 1a, b). In fact, fossil fuel consumption is significantly higher in winter than any other season in East Asia (Zhang et al., 2009). Further, emissions of biological particles and secondary produc-
- tion by the reaction of biogenic hydrocarbons with NO₃ radicals (Fry et al., 2009; Pratt et al., 2012) could contribute to WSON in late spring and late spring-early autumn, respectively. In fact, levels of NO₃⁻ in Sapporo aerosols declined in late spring to summer followed by a gradual increase toward early autumn (Pavuluri et al., 2015). The lower levels of NO₃⁻ in summer may be in part caused by its active involvement in the reactions with hydrogenerate and (Fry et al., 2000). From the lower levels of NO₃⁻ in summer may be in part caused by its active involvement in the reactions with hydrogenerate and (Fry et al., 2000).
- tions with hydrocarbons (Fry et al., 2009; Pratt et al., 2012; Rollins et al., 2009) to result in secondary WSON, although clean air masses from oceanic regions often arrive in Sapporo during summer (Fig. 1d).

3.4.2 Possible source and formation processes of WION

WION shows an inversed correlation with hopanes (*r* = 0.40) and levoglucosan (*r* = 0.53, *p* ≤ 0.05) during the campaign (Table 3). On the contrary, temporal trend of WION is similar to those of sucrose (Fig. 4e) during late spring to early summer and of biogenic SOA tracers: isoprene- and *α*-pinene-SOA tracers and *β*-caryophyllinic acid, (Fig. 4f) and MSA⁻ (Fig. 4g) throughout the year. Further, WION showed a significant linear relationship with sucrose and MSA⁻ and positive correlation (although weak) with all biogenic SOA tracers (isoprene- and *α*-pinene-SOA tracers and *β*-caryophyllinic acid) (Table 3). The correlation between WION and biogenic SOA tracers is slightly improved when the outliers that existed toward higher ends in the WION data set are



omitted. It is noteworthy that concentrations of sucrose are significantly high in the samples corresponding to the outliers in the WION data set.

Based on the above comparisons of WION with source tracers, we suggest that the WION in Sapporo aerosols is mainly derived from emissions of biological particles such as pollens and secondary production by the reactions of biogenic hydrocarbons (containing carbonyls) with NH₃ in the atmosphere (Bones et al., 2010; Updyke et al., 2012). WSON could be abundantly produced if oxidation of biogenic hydrocarbons by NO₃ radicals is prominent. However, the SOA preferably produced by O₃-initiated oxidation of biogenic hydrocarbons (e.g., isoprene and monoterpenes) contains carbonyl groups that can react with reduced nitrogen species such as NH₃ (g), NH⁺₄ and amino acide and produce high molecular weight N containing organics (Bones et al., 2010).

- acids and produce high molecular weight N containing organics (Bones et al., 2010; Updyke et al., 2012), which may not be fully water-soluble. For example, glyoxal has been reported to form an imidazole (Galloway et al., 2009) and also undergo oligomerization reactions (Noziere et al., 2009) to form high molecular weight N containing organics in the presence of NH₃. In fact, Wang et al. (2010) observed high molecular
- weight N containing organic salts (m/z range of 250–500) at substantial intensities in the positive-ion mass spectra in urban aerosols from Shanghai, China. They interpreted the formation of such high molecular weight organic salts by Mannich reaction rather than imidazole salts alone.
- Based on observations under controlled environmental conditions, Husted and Schjoerring (1996) reported that NH₃ emission from plants is enhanced with increasing leaf temperature and light intensity, which is similar to the emission of organic compounds from higher plants (Guenther, 1997). On the other hand, net emission of NH₃ from soil is enhanced in summer due to increasing soil temperatures (Wentworth et al.,
- 25 2014). Therefore, enhanced emissions of biogenic hydrocarbons and NH₃ during growing season (mid spring to mid summer) are very likely in spring and summer followed by the subsequent reactions and production of WION in the atmosphere. In fact, the air masses arriving to Sapporo originate from Siberia in spring and oceanic region passing over the Japanese Main Island in summer (Fig. 1c, d). They should be enriched



with the components derived from terrestrial and/or oceanic emissions. In addition to biogenic SOA and MSA⁻ (Fig. 4f, g), abundances of NH_4^+ are relatively high in spring and summer (Pavuluri et al., 2015). Hence, secondary production of WION by the reactions of SOA (carbonyls) with NH_3 ($NH_4^+ \leftrightarrow NH_3$) (Noziere et al., 2009) should be ⁵ accelerated during the warmer seasons in Northeast Asia.

Nitrogen isotope ratios of TN and WSTN: implications for source and aging 3.5

3.5.1 Seasonal variations

 δ^{15} N_{TN} ranged from 15.5 to 29.4 ‰ (average 21.9 ± 4.1 ‰) whereas δ^{15} N_{WSTN} ranged from 12.2 to 39.1 % (25.8 ± 8.2 ‰) in Sapporo aerosols (n = 21). As shown in Fig. 2b, temporal trends of $\delta^{15}N_{TN}$ and $\delta^{15}N_{WSTN}$ were highly comparable. They both show 10 peaks in summer with relatively high values in late autumn (Fig. 2b). Their averages, except for $\delta^{15}N_{TN}$ in autumn and winter, were significantly varied from season-to-season with higher δ^{15} N values in summer followed by spring, winter and autumn (Table 1). The seasonal trends of δ^{15} N and δ^{15} N_{WSTN} suggest that aerosol N is significantly influenced by season-specific source(s) and/or chemical aging of N species. As discussed 15 earlier, contributions from biomass burning (including forest fires) and fossil fuel combustion are enhanced in autumn/winter whereas those from biological sources and secondary production from biogenic hydrocarbons are significant in spring/summer. We found that $\delta^{15}N_{WSTN}$ are higher than $\delta^{15}N_{TN}$ throughout the campaign, except for few samples in autumn (Fig. 2b). An enrichment of ¹⁵N in WSTN over TN is more sig-20 nificant in summer than other seasons (Table 1). It has been reported that the aerosol particles collected over the controlled laboratory burns of the vegetation, which significantly contribute to WSTN, are enriched with ¹⁵N ranging from -1.3 to 13.1%

(average 6.6%) compared to the source vegetation (Turekian et al., 1998). Further, chemical aging of N species, including gas-to-particle exchange (e.g., $NH_4^+ \leftrightarrow NH_3$) re-25 actions during long-range transport causes the enrichment of ¹⁵N in aerosol N (Pavuluri et al., 2010). In addition, δ^{15} N of terrestrial plants, which significantly contribute



to WION, is relatively low (e.g., -3.4 to 12.2% in plants from the San Francisco Bay area) (Cloern et al., 2002). Therefore, higher $\delta^{15}N_{WSTN}$ values than $\delta^{15}N_{TN}$ in Sapporo aerosols suggest that contributions of biomass burning to WSTN and biological sources to WION (i.e. TN-WSTN) are significant and aerosol N has been subjected to significant chemical aging during long-range transport. The enhanced aging (particularly $NH_4^+ \leftrightarrow NH_3$) of nitrogenous aerosols under higher ambient temperature in summer may have caused further enrichment of ¹⁵N in WSTN (Pavuluri et al., 2010). As seen from Fig. 5, $\delta^{15}N_{TN}$ (and $\delta^{15}N_{WSTN}$) showed an inverse relation with the mass fractions of NO₃⁻-N and WSON in TN, but showed a linear relation with NH₄⁺-N/TN and WION/TN during the study period. Their relations with NO3-N/TN and 10 NH⁺₄-N/TN are statistically significant at 99 % ($p \le 0.01$) level but not in other two cases (Fig. 5). Such relations imply that the $\delta^{15}N_{TN}$ and $\delta^{15}N_{WSTN}$ are mainly controlled by their sources, chemical aging of nitrogenous species and abundances of NO3-N and NH₄⁺-N. It is well recognized that NO₃⁻ originates mainly from fossil fuel combustion and biomass burning whereas NH⁺₄ from biomass burning, agricultural activities and biolog-15 ical emissions (Delon et al., 2012; Hertel et al., 2012). In fact, NO₃⁻ is more abundant in mid autumn to early spring whereas NH₄⁺ maximize in mid spring to summer (Pavuluri et al., 2015). These results are consistence with those of source tracers (Fig. 4), again suggesting that the contributions from biomass burning and fossil fuel combustion are important in autumn and winter whereas the biological particles in spring and biogenic emissions of gaseous species (hydrocarbons and NH₃) and subsequent secondary production in spring/summer are important. These results of δ^{15} N further support that the WSON is mainly originate from combustion sources whereas the WION from biological sources.

25 **3.5.2** Comparison with literature values

Figure 6 compares the range (or mean) of δ^{15} N values reported for the particles emitted from point sources in mid-latitudes (combustion of different fossil fuels and waste



incinerations) and controlled laboratory burning of biomass (C₃ and C₄ plant species) as well as those observed in atmospheric aerosols from different locale over the world including Sapporo aerosols. $\delta^{15}N_{TN}$ values observed in Sapporo aerosols are significantly higher than those (-19.4 to 15.4%) reported for the particles emitted from point

- ⁵ sources of fossil fuel combustion and waste incineration burning (Fig. 6). They are also higher than those reported in urban aerosols from Paris, France (Fig. 6), where fossil fuel combustion was expected as a major source (Widory, 2007), and the marine aerosols over the western North Pacific (Fig. 6), which are considered to be mainly derived from sea-to-air emissions (Miyazaki et al., 2011).
- ¹⁰ However, they are comparable to the higher ends of the $\delta^{15}N_{TN}$ reported in atmospheric aerosols from Jeju Island, Korea (Fig. 6), which were attributed to significant atmospheric processing of N species, emitted mainly from China, during long-range transport (Kundu et al., 2010b). Further the lower ends of $\delta^{15}N_{TN}$ in Sapporo aerosols are comparable to the higher ends of $\delta^{15}N_{TN}$ reported for the particles emitted from ¹⁵ controlled burning of C₃ (range, 2.0 to 19.5‰) and C₄ (9.8 to 22.7‰) plant species
- in a laboratory study and those of atmospheric aerosols from Piracicaba and Amazon basin, Brazil (Fig. 6), where biomass burning is a dominant source (Martinelli et al., 2002). In addition, they are well comparable to those reported in the forest aerosols from Rondônia, Brazil, where biomass burning was expected as a dominant source (Martinelli et al., 2002).
- ²⁰ (Kundu et al., 2010a), and urban aerosols from Chennai and Mumbai, India (Fig. 6), which were considered to be mainly originated from biomass burning and subjected for aging (Aggarwal et al., 2013; Pavuluri et al., 2010).

On the other hand, lower values of $\delta^{15}N_{WSTN}$ observed in Sapporo aerosols are comparable to those reported in the forest aerosols from Sapporo experimental for-

est, Japan (Fig. 6). Although their seasonal patterns (winter minimum and summer maximum) are similar, the enrichment of ¹⁵N in Sapporo aerosols (Fig. 2b) is significantly higher than that (range of monthly averages, 9.0 to 26.0‰) reported in the Sapporo forest aerosols (Miyazaki et al., 2014). Further, the range of $\delta^{15}N_{WSTN}$ values observed in Sapporo aerosols are highly comparable with $\delta^{15}N$ of aerosol NH⁺₄, which



is much higher than that of gaseous NH_3 , in the atmosphere from the Maki monitoring station, Niigata, Japan (Fig. 6). The large difference in δ^{15} N between NH₃ and NH₄⁺ has been attributed to gas to particle (equilibrium) exchange reactions in the atmosphere (Hayasaka et al., 2004) because under equilibrium conditions, particulate NH_4^4 is more enriched with ¹⁵N than gaseous NH₃, where the isotopic enrichment factor 5 ($\mathcal{E}_{\mathrm{NH}_4^+\mathrm{-}\mathrm{NH}_3}$ was estimated to be 33‰ (Heaton et al., 1997). Such comparisons suggest that aerosol N in Sapporo aerosols should have mainly originated from biogenic sources including biomass burning and subjected to significant aging during long-range transport, although we do not preclude a contribution from fossil fuel combustion in winter.

Summary and conclusions 4

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Water-soluble organic nitrogen (WSON) and water-insoluble organic nitrogen (WION) and N isotope ratios of total nitrogen ($\delta^{15}N_{TN}$) and water-soluble TN ($\delta^{15}N_{WSTN}$) were obtained for TSP aerosol samples collected from Sapporo, northern Japan during September 2009 and October 2010. WSON and WION ranged from 0.00 to 288 ng m⁻ 15 (average 89.7 ± 80.6 ng m⁻³) and from 0.00 to 454 ng m⁻³ (126 ± 117 ng m⁻³). Although their temporal variations did not show clear seasonal trends, average WSON was found to be higher in winter followed by autumn, spring and summer whereas WION maximized in summer followed by spring with much lower concentration in autumn and

- winter. $\delta^{15}N_{TN}$ ranged from 15.5 to 29.4 ‰ (average 21.9 ± 4.1 ‰) whereas $\delta^{15}N_{WSTN}$ 20 ranged from 12.2 to 39.1 % (25.8 ± 8.2 %). Their seasonal variations showed high ratios in summer and late autumn. Comparisons of seasonal variations of WSON and WION with anthropogenic and biological source tracers together with air mass trajectories suggest that fossil fuel combustion and biomass burning are major sources of
- WSON whereas emissions of biological particles and secondary reactions of biogenic 25 hydrocarbons (carbonyls) with NH₃ significantly contribute to WION in Northeast Asia.



Seasonal variations of $\delta^{15}N_{TN}$ and $\delta^{15}N_{WSTN}$ and their relations to mass fractions of nitrogenous components in TN suggest that aerosol N in Sapporo aerosols is mainly originated from biogenic sources including biomass burning and subjected to chemical aging during long-range transport, although a contribution from fossil fuel combustion $_{5}$ is important particularly in winter.

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Table 1. Seasonal and annual averages with SD of concentrations of nitrogenous components and ¹⁵N isotope ratios of TN and WSTN in atmospheric aerosol (TSP) samples collected during 2 September 2009 and 5 October 2010 from Sapporo, northern Japan.

Component	Autumn	Winter	Spring	Summer	Annual	
Concentrations (ng m ⁻³)						
TN	790 ± 480	1456 ± 298	1116 ± 434	1023 ± 433	1074 ± 454	
WSTN	714 ± 496	1388 ± 247	986 ± 356	789 ± 351	950 ± 427	
NO ₃ -N	284 ± 120	362 ± 55.2	395 ± 182	73.8 ± 24.1	279 ± 172	
NH₄ [∓] -N	338 ± 347	896 ± 97.2	511 ± 191	647 ± 310	576 ± 314	
ON	168 ± 77.1	198 ± 186	212 ± 221	302 ± 149	220 ± 157	
WSON	91.6 ± 73.6	130 ± 139	84.7 ± 99.5	67.8 ± 39.8	96.0 ± 81.9	
WION	77.5 ± 50.9	68.4 ± 51.3	130 ± 141	234 ± 144	125 ± 123	
Isotope ratios (‰)						
$\delta^{15} N_{TN}$	18.91 ± 3.17	19.72 ± 3.56	23.42 ± 1.63	26.82 ± 1.95	22.47 ± 3.98	
$\delta^{15} N_{WSTN}$	19.17 ± 6.45	23.84 ± 4.13	26.82 ± 2.46	36.88 ± 2.46	26.71 ± 8.16	



Table 2. Concentrations of water-soluble organic nitrogen (WSON) and the mass fractions of WSON in water-soluble total nitrogen (WSTN) in Sapporo aerosols together with those in atmospheric aerosols from different sites in the world. ^a Total of fine and coarse mode particles, ^b dry (biomass burning) season, ^c Asian outflow region.

Locale	Study Period	Aerosol	WSON	WSON/WSTN	Reference	
		Size	(ng m ⁻³)	(%)		
Urban sites						
Sapporo, Japan	Sep 2009 to Oct 2010	TSP	89.7 ± 80.6	10.5 ± 7.5	This study	
Davis, California	Aug 1997 to Jul 1998	$PM_{2.5}$	265 ± 190	23	Zhang et al. (2002)	
Kofu, Japan	Aug 2009 to Jan 2013	PM ₁₀ ^a	221	~ 14	Matsumoto et al. (2014)	
	Forest sites					
Rondônia, Brazil	Mar-May 1999	PM ₁₀	49 ± 64.4	45 ± 35	Mace et al. (2003a)	
Rondônia, Brazil	Sep–Oct 1999 ^b	PM ₁₀	854 ± 938	43 ± 31	Mace et al. (2003a)	
Sapporo, Japan	Jun 2009 to Dec 2011	TSP	157 ± 127	20 ± 11	Miyazaki et al. (2014)	
Fujiyoshida, Japan	Aug 2009 to Jan 2013	PM ₁₀ ^a	101	~ 11	Matsumoto et al. (2014)	
		Coast	al sites			
Erdemli, Turkey	Mar–May 2000	TSP	406 ± 588	$\sim 26 \pm 28$	Mace et al. (2003c)	
Crete, Greece	2005–2006	PM ₁₀ ^a	~ 239	13	Violaki and Mihalopoulos (2010)	
Qingdao, China	Mar–Apr 2006	TSP	2520 ± 1760	~ 20	Shi et al. (2010)	
Oceanic sites						
Yellow Sea ^c	Mar 2005	TSP	2860 ± 1780	~ 17	Shi et al. (2010)	
Yellow Sea ^c	Apr 2006	TSP	1220 ± 812	~ 17	Shi et al. (2010)	
South China Sea ^c	Apr 2005	TSP	910 ± 280	~ 34	Shi et al. (2010)	
East China Sea ^c	Sept–Oct 2002	TSP ^a	756 ± 504	~ 24	Nakamura et al. (2006)	
East China Sea and	Mar 2004	TSP ^a	224 ± 266	~ 10	Nakamura et al. (2006)	
Western North Pacific ^c						
Cape Grim, Australia	Nov–Dec 2000	TSP	50.4 ± 79.8	~ 21	Mace et al. (2003b)	
Cape Grim, Australia	Nov–Dec 2000	TSP	13.0 ± 13.3	~ 25	Mace et al. (2003b)	
Oahu, Hawaii	Jul–Aug 1998	PM ₁₀	46.2 ± 28	~ 33	Cornell et al. (2001)	
Western North Pacific	Aug–Sep 2008	TSP	~ 17.5		Miyazaki et al. (2011)	

ACPD 15, 12617-12652, 2015 **Atmospheric** chemistry of nitrogenous aerosols in Northeast Asia C. M. Pavuluri et al. **Title Page** Abstract Introduction Conclusions References **Tables** Figures < Back Close Full Screen / Esc **Printer-friendly Version** Interactive Discussion

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Table 3. Relations of WSON and WION with source tracers in Sapporo aerosols during the study period (n = 21).

Course trees	Correlation coefficient (r)			
Source tracer	Correlation coefficient (r)			
	WSON	WION		
Hopanes	0.60**	-0.40		
Levoglucosan	0.25	-0.53*		
Sucrose	0.28	0.63**		
Isoprene-SOA tracers	-0.31	0.26		
α -Pinene-SOA tracers	0.02	0.25		
β -Caryophylinic acid	-0.14	0.24		
MSA ⁻	0.04	0.71**		

 ** Correlation is significant at 0.01 level; * Correlation is significant at 0.05 level.





Figure 1. Plots of 10 day backward air mass trajectories arriving over Sapporo at 500 m a.g.l.











Figure 3. Linear relations of IN (sum of NO_3^- -N and NH_4^+ -N) to (a) WSTN and (b) TN.





Figure 4. Seasonal variations of **(a)** concentrations of WSON and WION, **(b)** mass fractions of WSON and WION in TN, and **(c–e)** concentrations of organic molecular tracers and MSA⁻ in Sapporo aerosols. Concentrations of isoprene-SOA tracers are sum of the concentrations of 2-methylglyceric acid, cis-2-methyl-1,3,4-trihydroxy-1-butene, 3-methyl-2,3,4-trihydroxy-1-butene, trans-2-methyl-1,3,4-trihydroxy-1-butene, 2-methylthreitol and 2-methylerythritol whereas concentrations of α -pinene-SOA tracers are sum of the concentrations of 3-hydroxyglutaric, pinonic, pinic and 3-methyl-1,2,3-butanetricarboxylic acids. The data of organic molecular tracers are from Pavuluri et al. (2013) and MSA⁻ is from Pavuluri et al. (2015).







Figure 5. Scatter plots between $\delta^{15}N_{TN}$ (and $\delta^{15}N_{WSTN}$) and mass fractions of nitrogenous components (NO₃⁻-N, NH₄⁺-N, WSON and WION) in TN in Sapporo aerosols.



Figure 6. Range or mean N isotope ratios in the particles emitted from point sources, source substance, and atmospheric aerosols from different sites in the world. *, **, and *** show gaseous NH₃, particulate NH₄⁺ and WSTN, respectively. ^aWidory (2007); ^bTurekian et al. (1998); ^{c, g}Kundu et al. (2010a, b); ^dMartinelli et al. (2002); ^ePavuluri et al. (2010); ^fAggarwal et al. (2013); ^{h, j}Miyazaki et al. (2011, 2014); ⁱHayasaka et al. (2004); ^kThis study.

