Response to Editor's Comments
Dear Dr. Guenther,
Thank you for accepting our paper for publication in ACP. We revised the manuscript by taking all the suggested corrections:
Line 139: replace "remained" with "remaining"
Line 175: replace "oxidaiton" with "oxidation"
Line 236: insert "of a" between "consists" and "significant"
Line 528: replace "consistence" with "consistent"
(please see lines 153, 184, 239 and 445, respectively, of the revised MS). The revised MS with track changes is included in this response letter for your check.
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12 Sincerely yours,

14 Kimitaka Kawamura

- 15 Atmospheric chemistry of nitrogenous aerosols in Northeast Asia:
- 16 biological sources and secondary formation

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Abstract. To better understand the sources of nitrogenous aerosols, particularly 25 26 water-soluble organic nitrogen (WSON) and water-insoluble organic nitrogen (WION), in 27 Northeast Asia, we measured total nitrogen (TN) and water-soluble total nitrogen (WSTN) as well as nitrogen isotope ratios ($\delta^{15}N)$ of TN ($\delta^{15}N_{TN})$ and WSTN ($\delta^{15}N_{WSTN})$ in the total 28 29 suspended particles (TSP) collected from Sapporo, northern Japan for one-year period. In general, WION was more abundant ($126 \pm 117 \text{ ng m}^{-3}$) whereas WSON ($89.7 \pm 80.6 \text{ ng m}^{-3}$), 30 31 accounting for $14 \pm 11\%$ and $9.2 \pm 7.3\%$ of TN, respectively. WSON peaked in late autumn to winter (maximum 288 ng m⁻³) and WION peaked in mid spring to early summer (454 ng 32 m⁻³). $\delta^{15}N_{TN}$ (21.9 ± 4.1%) and $\delta^{15}N_{WSTN}$ (25.8 ± 8.2%) showed peaks in summer with 33 relatively high ratios in late autumn. Based on the seasonal variations of WSON and WION 34 35 together with organic tracers, fossil fuel combustion and biomass burning are found to be two 36 major sources of WSON whereas emissions of biological particles and secondary formation 37 by reactions of biogenic secondary organic species (carbonyls) with NH₃ are suggested as important source of WION. The seasonality of $\delta^{15}N_{TN}$ and $\delta^{15}N_{WSTN}$, together with the 38 39 comparisons to literature values, implies that chemical aging (including gas/particle 40 partitioning) and biomass burning are the causes of the enhanced values in summer and autumn, respectively. This study demonstrates that contributions of aerosol N from fossil fuel 41 42 combustion and biomass burning dominate in autumn/winter whereas emission of terrestrial 43 biological particles and secondary formation from biogenic hydrocarbons and subsequent 44 chemical aging in the atmosphere are important in spring/summer in Northeast Asia.

1 Introduction

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46 In East Asia, high loading of aerosol nitrogen (N) and its significant deposition onto the Earth surface are associated with enhanced agricultural usage of nitrogenous fertilizer: 40% of 47 vegetation receives nitrogen deposition in excess of 1000 mg-N m⁻² yr⁻¹ (threshold level) and 48 even more (~50%) in Japan (Dentener et al., 2006; Liu et al., 2013; Morino et al., 2011). 49 50 Emissions of NH₃ and NO_x to the air have an impact on atmospheric chemistry, visibility 51 impairment, human health and the Earth's climate system (Rollins et al., 2012; Vitousek et al., 52 1997). Further, N deposition over the Earth surface has large influence on terrestrial and 53 aquatic ecosystems via the changes in the global biogeochemical cycles (Duce et al., 2008; Galloway et al., 2004; Kim et al., 2014). Most studies of aerosol N in East Asia and other 54 regions in the world have been focused on inorganic species (NO₃⁻ and NH₄⁺) (Dentener et al., 55 56 2006; Duce et al., 2008; Liu et al., 2013). Although organic nitrogen (ON) represents a significant fraction (up to 80%) of total aerosol N and may play a critical role in 57 58 biogeochemical cycles, ON has rarely been studied (Cape et al., 2011; Matsumoto et al., 2014; 59 Zhang et al., 2008). 60 Aerosol ON is comprised of a wide range of nitrogenous compounds from semi-volatile amines to proteins and macromolecules (Cape et al., 2011; Laskin et al., 2009; Wang et al., 61 62 2010). In addition to biomass burning, the primary sources including biological particles 63 emitted from soil, vegetation, pollen, bacteria and the ocean surface are important (Cape et al., 2011; Jickells et al., 2013; Miyazaki et al., 2014; Neff et al., 2002). ON can also be produced 64 in the atmosphere via the reactions of NO₃ radical or NH₃ with biogenic/anthropogenic 65 volatile organic compounds (VOCs) followed by gas-to-particle conversion (Cornell et al., 66 67 2001; Miyazaki et al., 2014; Pratt et al., 2012). Recently, oxidation of isoprene and 68 monoterpenes by NO₃ is considered as a substantial source of organic nitrates and secondary 69 organic aerosols (SOA) (Fry et al., 2009; Pratt et al., 2012; Rollins et al., 2009; Rollins et al.,

2012). Further, chemical aging of SOA derived from both biogenic and anthropogenic 70 precursors in the presence of NH₃ in gas phase and NH₄ salts in aqueous phase can produce 71 light-absorbing nitrogenous compounds (defined as brown carbon) (Bones et al., 2010; 72 73 Updyke et al., 2012). 74 However, the relative importance of anthropogenic and biogenic emissions including 75 biomass burning is largely unknown and secondary formation of ON has been poorly 76 characterized (Cape et al., 2011; Jickells et al., 2013; Kanakidou et al., 2012) and thus land-atmosphere interactions of aerosol N are not fully understood yet because of limited 77 78 measurements of aerosol ON. Previous studies of ON have focused on water-soluble ON 79 (WSON) but very little is known about the abundances and sources of water-insoluble ON 80 (WION) particularly in continental aerosols. Interestingly, WION is much more abundant (on 81 average 34.0 nmol N m⁻³) than WSON (4.5 nmol N m⁻³) in aerosols from Lewes, Delaware 82 on the mid-Atlantic US coast (Russell et al., 2003) and marine aerosols ($55 \pm 16\%$ of total N) over the western North Pacific (Miyazaki et al., 2011). Hence, it is of interest to investigate 83 84 the seasonal variations of WSON and WION. 85 Here, we present total N (TN), water-soluble TN (WSTN), WSON and WION and nitrogen isotope ratios (δ^{15} N) of TN (δ^{15} N_{TN}) and WSTN (δ^{15} N_{WSTN}) measured in ambient 86 aerosols (TSP) collected from Sapporo, northern Japan over one-year period. The isotope 87 88 studies are helpful in understanding the origins and atmospheric processing of N (Kawamura 89 et al., 2004; Pavuluri et al., 2010). Based on WSON and WION data together with various organic tracers, δ^{15} N values and air mass trajectories as well as the comparison with previous 90 91 studies, we discuss the atmospheric chemistry of nitrogenous aerosols and their possible 92 sources with a major focus on WSON and WION in Northeast Asia.

2 Materials and methods

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2.1 Aerosol sampling

Total suspended particles (TSP) were collected from Sapporo in the western part of Hokkaido Island, northern Japan (43.07°N, 141.36°E), an ideal location for collecting the air masses delivered from Northeast Asian regions: Siberia, the Russian Far East, China, and surrounding oceans (Aggarwal and Kawamura, 2008; Yamamoto et al., 2011). The geographic details of the sampling site are described elsewhere (Pavuluri et al., 2013). TSP sampling (n = 21) was performed on the rooftop of the three-story building of the Institute of Low Temperature Science (ILTS), Hokkaido University, Sapporo during 2 September 2009 and 5 October 2010 using a pre-combusted (450°C, 4 h) quartz fiber filter and high-volume (~65 m³ h⁻¹) air sampler. Each sample was collected for ca. 2 consecutive weeks in order to obtain sufficient amount of carbon for radiocarbon (¹⁴C) analyses at organic molecular levels, another objective of this research. Filter samples were placed in a pre-combusted glass jar with a Teflon-lined screw cap and stored in a dark room at -20°C prior to analysis.

It should be noted that aerosol samples collected on quartz fiber filters might have positive (adsorption of gaseous HNO₃ and NH₃ and WSON) and negative (evaporation of particulate NH₄⁺ salts such as NH₄NO₃ and WSON) sampling artifacts (Matsumoto et al., 2014; Squizzato et al., 2013). Since the sampling time is longer in this study, the evaporative loss from the particles should be more significant than the adsorbed gases by quartz fiber filter and thus the reported concentrations may be underestimated. However, the ambient temperatures encountered in Sapporo are rather low (range of averages of each sample period: -3.30°C to 24.5°C) (Pavuluri et al., 2015) that may not cause a significant evaporative loss of N species during the sampling period, because quartz filters show a good retention for semi-volatile NH₄NO₃ sampled at an air temperature up to 21°C and ~30% at even an elevated air temperature of 35°C (Schaap et al., 2004). Therefore, we believe that our sampling technique does not have serious sampling artifacts even in summer.

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2.2 Chemical analyses

2.2.1 Determination of TN and WSTN and their N isotope ratios

TN (and WSTN) contents and their isotope ratios ($\delta^{15}N_{TN}$ and $\delta^{15}N_{WSTN}$, respectively) in TSP 123 124 samples were determined using an elemental analyzer (EA) (Carlo Erba NA 1500) and 125 EA/isotope ratio mass spectrometer (IRMS) (Finnigan MAT Delta Plus), respectively, as 126 described elsewhere (Kawamura et al., 2004; Pavuluri et al., 2010). An aliquot of filter sample (1.8 cm in diameter disc × 3) was extracted for WSTN with ~15 ml Milli Q water 127 128 (>18.3 MΩ) under ultrsonication for 20 min and filtrated with a syringe filter (GL Sciences 129 Chromatodisc Type A, 0.45 μm). The extracts were adjusted to pH 8~9 with ~0.5 ml of 0.05 130 M KOH and concentrated to dryness using a rotary evaporator under vacuum and then 131 re-dissolved in 200 µl Milli Q water. An aliquot of filter sample (1.0 cm diameter disc for TN 132 and 50 µl of water extract, a sub-sample of the 200 µl re-dissolved extract, adsorbed onto ~20 133 mg of pre-combusted Chromosorb for WSTN) was placed in a pre-cleaned tin cup and 134 introduced into EA. The resulting N2 gas was measured with thermal conductivity detector in EA and transferred to IRMS via an interface (ConFlo II) for ¹⁵N/¹⁴N measurement. The blank 135 136 levels in TN measurements are found to be 0.0 µgN. The analytical uncertainties (i.e., RSD) 137 in a single set of duplicate analyses of filter samples were within 6% for TN, 3% for WSTN, 0.28% for $\delta^{15}N_{TN}$ and 1.34% for $\delta^{15}N_{WSTN}$. The detection limit of EA determined by the 138 139 sensitivity of the detector is 0.4 µg N (Nieuwenhuize et al., 1994). 140 We also measured WSTN using a total organic carbon (TOC)/total nitrogen (TN) 141 analyzer (Shimadzu TOC-Vcsh), as reported by Miyazaki et al. (2011). Briefly, an aliquot of 142 filter (1.4 cm in diameter disc) was extracted with 10 ml Milli Q water under ultrasonication 143 for 20 min. The extracts were filtered with syringe filter (GL Sciences Chromatodisc Type A,

 $0.45~\mu m)$ and then injected into TOC/TN analyzer. The analytical uncertainty in duplicate analyses of filter samples was within 5%.

146 Concentrations of WSTN measured by EA are lower by $23 \pm 5\%$ than those measured by 147 TOC/TN analyzer. The difference was significant when the concentration of WSTN was greater than 0.5 µg m⁻³. The difference may be arisen from different combustion temperature: 148 149 1400°C for EA v.s. 680°C for TOC/TN analyzer and different procedures. We consider that 150 the WSTN measured by EA may be underestimated due to possible evaporative loss of N 151 species such as NH₃ and HNO₃ during the concentration step and/or sticking loss of N 152 species on glass walls. In fact, if the evaporative loss of N species, e.g., NH₃, is significant, $\delta^{15}N$ of the remaining WSTN should be significantly varied in proportionate to loss of 153 WSTN, because of isotopic fractionation caused by NH₄⁺-NH₃ exchange (phase partitioning) 154 reaction (Hoefs, 2009). Under equilibrium conditions, δ¹⁵N values of NH₄⁺ become higher 155 156 than those of NH₃, where the isotopic enrichment factor (ε_{NH4+-NH3}) estimated to be +33‰ (Heaton et al., 1997). However, δ¹⁵N_{WSTN} did not show any abnormal deviation in its 157 temporal trend, which is comparable to that of $\delta^{15}N_{TN}$ (see Fig. 2b), suggesting that the 158 evaporative loss of N species and thus the impact of WSTN loss on its δ^{15} N are not important. 159 160 Here, we use WSTN measured by TOC/TN analyzer to minimize the uncertainties in the 161 estimation of WSON and WION.

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2.2.2 Measurements of inorganic ions

Details in the measurements of inorganic ionic species such as NO₃, NH₄⁺ and methanesulfonate: MSA⁻ (CH₃SO₃⁻) are described elsewhere (Pavuluri et al., 2015). Briefly, an aliquot of filter sample was extracted with Milli Q water under ultrasonication and filtrated with a syringe filter (GL Sciences Chromatodisc Type A, 0.45 μm). The filtrates were then injected into an ion chromatograph (761 Compact IC, Metrohm). A calibration

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curve was prepared with a set of authentic standards to calculate the concentrations for the samples. The analytical precision was within 4%.

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2.2.3 Estimation of WSON and WION

Amounts of inorganic N (IN), i.e., sum of NO₃⁻-N and NH₄⁺-N, were calculated from the concentrations of NO₃⁻ and NH₄⁺ ions (Pavuluri et al., 2015). The amounts of ON, WSON

and WION were estimated by the following equations (Miyazaki et al., 2011);

ON = TN - IN

WSON = WSTN - IN

WION = TN - WSTN

180 Using propagating errors of each parameter, the precision of ON, WSON and WION were

estimated to be within 8%.

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2.2.4 Measurements of organic tracers

Organic tracers such as hopanes, isoprene- and monoterpene-oxidation products were
determined using a capillary gas chromatograph (Hewlett-Packard 6890) coupled to mass

186 spectrometer (Hewlett-Packard 5973) (GC/MS) as described elsewhere (Fu et al., 2010).

187 Briefly, organic tracer compounds were extracted from the filter samples with

dichloromethane/methanol (2:1; v/v) under ultrasonication. The hydroxyl and carboxyl

189 functional groups were derivatized to trimethylsilyl (TMS) ethers and esters, respectively,

190 with 50 µl of N,O-bis-(trimethylsilyl)triflouroacetamide (BSTFA) containing 1%

trimethylsilyl chloride and 10 µl of pyridine. The TMS derivatives were then measured using

a GC/MS. The analytical errors in duplicate analyses were less than 10%.

All the data reported here are corrected for the field blank collected in each season.

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Results and Discussion

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Source regions of Sapporo aerosols

198 Ten-day backward air mass trajectories arriving in Sapporo at 500 m above the ground level (AGL) were computed for every 48 h during each sample period using HYSPLIT model 200 (Draxler and Rolph, 2012) and are plotted in Fig. 1. The trajectories show that the air masses were mostly originated from Siberia passing over Northeast Asian regions and the Japan Sea during autumn, winter and spring (Fig. 1). In summer, they originated from the East China Sea and/or western North Pacific passing over the coastal region and/or the Main Island of Japan (Fig. 1). The air parcels were mostly travelled at lower than 2000 m AGL in all seasons and hence their advection from distant source regions to the receptor site should be significant.

In fact, the radiocarbon analyses showed that percent modern carbon (pMC) of total carbon and water-soluble organic carbon in our aerosols started to increase from mid to late winter toward spring (Pavuluri et al., 2013), although the growing season starts in May in Hokkaido when daily average temperatures are ≥10°C (Toma et al., 2011). Because the contribution of modern carbon from local vegetation is unlikely in winter, such increase in pMC during winter has been mainly attributed to the contributions of biomass burning aerosols via long-range transport from East Asia (Pavuluri et al., 2013). In addition, temporal variations of crustal metals, e.g., Al, Ca and Fe, in Sapporo aerosols are consistent with those of TSP from winter to mid summer with a peak in early spring, indicating a significant atmospheric transport of Asian dust from arid regions in Mongolia and China (Pavuluri et al., 2015). Therefore, we consider that the Sapporo aerosols should have largely influenced by long-range transported air masses from distant sources over Northeast Asia. However, the characteristics and seasonal variations of Sapporo aerosols could reflect the regional scenario and we do not preclude an impact from the local sources during the campaign.

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3.2 Total nitrogen (TN) and water-soluble total nitrogen (WSTN)

223 Concentrations of TN and WSTN ranged from 348 to 1,750 ng m⁻³ (average 1020 ± 466 ng m^{-3}) and 293 to 1,520 ng m^{-3} (893 ± 443 ng m^{-3}), respectively, in Sapporo aerosols during the 224 225 study period (n = 21). The mass fractions of TN in TSP ranged from 1.20 to 8.66% (average 226 $3.81 \pm 2.28\%$). As shown in Fig. 2a, although the temporal trends of TN and WSTN did 227 follow a seasonal pattern, except for few samples, their increasing and decreasing trends are 228 not quite distinct from season to season. However, on average they were higher in winter 229 followed by spring, summer and autumn (Table 1). Similarly, TN/TSP mass ratios did not 230 show any clear temporal trend during the study period but showed a drastic increase from 231 mid autumn (1.31%) to winter (8.66%). TN/TSP mass ratios (average 3.8%, range 1.2-8.7%) 232 observed in Sapporo aerosols during the campaign are lower than those (average 5%, range 233 2-15%) reported for urban aerosols from Tokyo (Kawamura et al., 1995), where 234 anthropogenic source contributions are expected to be the major source. However, wintertime 235 TN/TSP mass ratios in Sapporo aerosols (≥7.8%) are close to that reported from urban Tokyo, 236 suggesting that the contributions from anthropogenic sources such as fossil fuel combustion 237 may be significant in winter. As seen from Fig. 3, WSTN contains mostly IN (i.e., $\sum NO_3^- N + NH_4^+ - N$) (range 70.8— 238 239 100%, average $89.7 \pm 7.74\%$) whereas TN consists of a significant amount of ON (average $22.7 \pm 17.7\%$, maximum 44.7%,) that ranged from below detection limit (BDL)-569 ng m⁻³ 240 241 $(215 \pm 150 \text{ ng m}^{-3})$ during the study period. High abundances of ON in TN rather than WSTN 242 indicate a significance of WION in Sapporo aerosols. Further, WSTN showed a stronger linear relation with IN $(r^2 = 0.97)$ than that $(r^2 = 0.90)$ between TN and IN in Sapporo 243 aerosols (Fig. 3). Such deviation in the linearity suggests that the origins and/or secondary 244 formation processes of WION (i.e., TN - WSTN) may be different from that of WSTN 245

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246	(including WSON). Therefore, it is apparent that the atmospheric loading and impacts of
247	aerosol N in Northeast Asia is significantly influenced by the sources and seasonality of ON.
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249	3.3 Water-soluble organic nitrogen (WSON) and water-insoluble organic nitrogen
250	(WION)
251	3.3.1 Concentrations and mass fractions in TN
252	Concentrations of WSON and WION in Sapporo aerosols ranged from BDL to 288 ng m ⁻³
253	and BDL to 454 ng m $^{-3}$, respectively. On average, WION is more abundant (126 \pm 117 ng m $^{-3}$)
254	than WSON (89.7 \pm 80.6 ng m ⁻³). Although the temporal variations did not show any clear

256 mid spring to early summer (i.e., in growing season) (Fig. 4a). Further, seasonally averaged 257 concentrations of WSON are higher in winter followed by autumn, spring and summer 258

whereas those of WION are higher in summer followed by spring and much lower in autumn and winter, although the seasonal differences are not statistically significant (Table 1). On

seasonal pattern, WSON maximized in late autumn to mid winter whereas WION peaked in

average, mass fraction of WION in TN is 13.9 ± 11.1% (range BDL-33.1%) and WSON/TN

ratio is $9.13 \pm 7.29\%$ (BDL–30.2%) during the study period. Temporal variations of the mass

fractions of WSON and WION in TN (Fig. 4b) are similar to those of the concentrations (Fig.

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3.3.2 Comparisons with previous studies from different locale

As seen from Table 2, average concentration of WSON in Sapporo aerosols is lower than that reported in urban aerosols from Davis, California and Kofu, Japan as well as from the coastal sites: Erdemli in Turkey, Crete in Greece, and Qingdao in China. It is also lower than that reported in the forest aerosols collected from Rondônia, Brazil during intensive biomass burning period (dry season) (Mace et al., 2003a) and from Sapporo, Japan, and in the marine

271 aerosols over the Asian outflow regions: the Yellow Sea, South China Sea and western North 272 Pacific (Table 2). In contrast, average concentration of WSON in Sapporo aerosols is 273 comparable to that of forest aerosols from Fujiyoshida, Japan, but higher than that from Rondônia, Brazil during wet season. It is also higher than that of the marine aerosols from 274 275 pristine oceanic regions: Cape Grim, Australia (Mace et al., 2003b) and Oahu, Hawaii 276 (Cornell et al., 2001) and over the western North Pacific (Miyazaki et al., 2011) (Table 2). In addition, the higher end (maximum 288 ng m⁻³) of WSON is comparable to average WSON 277 278 concentration reported in urban aerosols from Davis, California and Kofu, Japan and in 279 coastal urban aerosols from Crete, Greece (Table 2). However, average mass fraction of 280 WSON in WSTN in Sapporo aerosols is found to be comparable or close to those reported 281 for urban aerosols from Kofu, Japan, coastal urban aerosols from Crete, Greece, forest 282 aerosols from Fujiyoshida, Japan and the marine aerosols over the Yellow Sea, China Sea 283 and western North Pacific (Table 2). 284 Higher concentrations of WSON reported for urban aerosols from Davis, California, 285 during late fall and winter are attributed to increased amounts of atmospheric liquid water, 286 which promote a partition of gaseous WSON to particles (Zhang et al., 2002). In Kofu and 287 Fujiyoshida (forest site), Japan, emissions from combustion sources including biomass 288 burning and plant-derived particles as well as secondary formation through the reaction of 289 NO₂ with volatile organics are considered as potential sources of WSON (Matsumoto et al., 290 2014). In the Mediterranean atmosphere, atmospheric dust from north Africa was expected as 291 dominant source of WSON at Erdemli, Turkey (Mace et al., 2003c) whereas in Crete, Greece, 292 fossil fuel combustion and biomass burning are also considered (Violaki and Mihalopoulos, 293 2010). Over the marginal seas of China (Asian outflow region): Qingdao, the Yellow Sea and 294 Sea of China, anthropogenic sources are considered as dominant origin of WSON with minor 295 contributions from marine and crustal sources (Nakamura et al., 2006; Shi et al., 2010). In the

forest aerosols from Sapporo, Japan, emissions of biological particles are considered as major source of WSON in autumn whereas secondary formation of WSON from biogenic hydrocarbons is considered as dominant source in summer (Miyazaki et al., 2014).

On the other hand, average concentration of WION ($126 \pm 117 \text{ ng m}^{-3}$) found in Sapporo aerosols is comparable to that ($112 \pm 61 \text{ ng m}^{-3}$) reported in the marine aerosols over the western North Pacific, where the sea-to-air emissions are considered as the major source as well as the secondary production via the reactions of gaseous hydrocarbons with NH₃ (Miyazaki et al., 2011). In addition, the higher end (maximum 454 ng m⁻³) of WION in Sapporo aerosols is comparable to the average WION (476 ng m⁻³) reported in the coastal aerosols from Lewes, Delaware, where biological and mineral particles are expected as major source (Russell et al., 2003).

Therefore, we consider that WSON in Sapporo aerosols may be mainly derived from anthropogenic emissions including biomass burning, although emissions from biological sources and secondary formation from gaseous hydrocarbon precursors cannot be excluded. Meanwhile, WION may be originated from emissions of biological particles and secondary formation from biogenic hydrocarbons. We further examine the possible contributions from such sources in the following section based on their comparisons with the source tracers.

3.4 Comparisons of WSON and WION with organic tracers

315 3.4.1 Possible sources of WSON

The temporal trend of WSON is found to be somewhat similar to that of hopanes (C₂₇-C₃₂ hopanoid hydrocarbons), specific biomarkers of petroleum and coal (Schauer et al., 2002), particularly in autumn and winter time Sapporo aerosols (Fig. 4c). Further, WSON showed a linear relationship with hopanes with moderate but significant correlation during the campaign (Fig. 5a). On the other hand, temporal variation of WSON from mid autumn to mid

winter is somewhat similar to that of levoglucosan (Fig. 4d), a tracer for biomass burning (Simoneit, 2002), and showed a weak linear relation throughout the campaign (Fig.5b). Contributions of WSON to TN are higher in the above seasons (Fig. 4b). In contrast, WSON was abundant in few spring (from mid April to mid May) and summer (late June and late July) samples (Fig. 4a), despite the lower levels of both hopanes and levoglucosan throughout the spring and summer (Fig. 4c,d). The lower levels of levoglucosan in spring and summer, and thus a weak correlation of WSON with levoglucosan during the campaign, should have been caused, in part, by the decomposition of levoglucosan (Hoffmann et al., 2010), because contributions of biomass burning emissions to Sapporo aerosols were considered to be significant during these periods (Pavuluri et al., 2015). Temporal trend of WSON from late spring to early summer (Fig. 4a) is similar to that

Temporal trend of WSON from late spring to early summer (Fig. 4a) is similar to that of sucrose (Fig. 4e), which is a tracer of pollens emitted from terrestrial higher plants (Fu et al., 2012). The seasonal trend of WSON (Fig. 4a) is also similar to those of biogenic secondary organic aerosols (SOA), i.e., α-pinene-SOA tracers (Kleindienst et al., 2007), and MSA⁻ in late spring to early autumn (Fig. 4f,g). MSA⁻ is produced in the atmosphere by photochemical oxidation of dimethyl sulfide (DMS) (Bates et al., 1992) that emits from terrestrial higher plants (Geng and Mu, 2006), soil in forests (Yi et al., 2010) and phytoplankton bloom in oceans (Bates et al., 1992). Therefore, MSA⁻ can be considered as a tracer to understand the importance of secondary production of WSON (and WION) from biogenic organics, although both DMS and MSA⁻ do not contain nitrogen. Interestingly, WSON showed a linear relation with sucrose (r = 0.71, $p \le 0.05$), α-pinene SOA tracers and MSA⁻ during spring and summer (Fig. 5c-e). In fact, the Spearman's rank correlations of WSON with α-pinene SOA tracers (r = 0.71) and MSA⁻ (r = 0.62) in spring and summer (r = 0.71) are statistically significant (r = 0.71). Further, it showed a positive Spearman's rank correlation with isoprene SOA tracers (r = 0.33) and β-caryophylinic acid (r = 0.37).

However, WSON shows no relation with sucrose, isoprene- and α -pinene-SOA tracers, β -caryophylinic acid and MSA⁻ when we consider the total data set (n = 21) that include all seasons during the campaign (Table 3).

The above comparisons of WSON with source tracers imply that anthropogenic emissions including biomass burning are major sources of WSON in Sapporo aerosols. Their contributions to Sapporo aerosols may have been enhanced in autumn and winter when the air masses enriched with forest fire and fossil fuel combustion products are often delivered from Siberia passing over Northeast China (Fig. 1a, b). In fact, fossil fuel consumption is significantly higher in winter than any other season in East Asia (Zhang et al., 2009). Further, emissions of biological particles and secondary production by the reaction of biogenic hydrocarbons with NO₃ radicals (Fry et al., 2009; Pratt et al., 2012) could contribute to WSON in late spring and late spring-early autumn, respectively, although we do not preclude a minor contribution from biomass burning emissions. In fact, levels of NO₃⁻ in Sapporo aerosols declined in late spring to summer followed by a gradual increase toward early autumn (Pavuluri et al., 2015). The lower levels of NO₃⁻ in summer may be in part caused by its active involvement in the reactions with hydrocarbons (Fry et al., 2009; Pratt et al., 2012; Rollins et al., 2009) to result in secondary WSON, although clean air masses from oceanic regions often arrive in Sapporo during summer (Fig. 1d).

3.4.2 Possible source and formation processes of WION

The results of Spearman's rank correlation analysis between WION and source tracers are presented in Table 3. WION shows a significant inversed correlation with hopanes and levoglucosan during the campaign (Table 3). On the contrary, temporal trend of WION is similar to those of sucrose (Fig. 4e) during late spring to early summer and of biogenic SOA tracers: isoprene- and α-pinene-SOA tracers and β-caryophyllinic acid, (Fig. 4f) and MSA-

371 (Fig. 4g) throughout the year. Further, WION showed a moderate significant positive 372 Spearman's correlation with sucrose, isoprene SOA and MSA⁻ during the study period (n = 373 21) (Table 3) and also with β-caryophyllinic acid in spring and summer (r = 0.62, $p \le 0.05$; n = 10).

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Based on the above comparisons of WION with source tracers, we suggest that the WION in Sapporo aerosols is mainly derived from emissions of biological particles such as pollens and secondary production by the reactions of biogenic hydrocarbons (containing carbonyls) with NH₃ in the atmosphere (Bones et al., 2010; Updyke et al., 2012). WSON could be abundantly produced if oxidation of biogenic hydrocarbons by NO₃ radicals is prominent. However, the SOA preferably produced by O₃-initiated oxidation of biogenic hydrocarbons (e.g., isoprene and monoterpenes) contains carbonyl groups that can react with reduced nitrogen species such as NH₃(g), NH₄⁺ and amino acids and produce high molecular weight N containing organics (Bones et al., 2010; Updyke et al., 2012), which may not be fully water-soluble. For example, glyoxal has been reported to form an imidazole (Galloway et al., 2009) and also undergo oligomerization reactions (Noziere et al., 2009) to form high molecular weight N containing organics in the presence of NH₃. In fact, Wang et al. (2010) observed high molecular weight N containing organic salts (m/z range of 250-500) at substantial intensities in the positive-ion mass spectra in urban aerosols from Shanghai, China. They interpreted the formation of such high molecular weight organic salts by Mannich reaction rather than imidazole salts alone.

Based on observations under controlled environmental conditions, Husted and Schjoerring (1996) reported that NH₃ emission from plants is enhanced with increasing leaf temperature and light intensity, which is similar to the emission of organic compounds from higher plants (Guenther, 1997). On the other hand, net emission of NH₃ from soil is enhanced in summer due to increasing soil temperatures (Wentworth et al., 2014). Therefore, enhanced

396 emissions of biogenic hydrocarbons and NH3 during growing season (mid spring to mid 397 summer) are very likely in spring and summer followed by the subsequent reactions and 398 production of WION in the atmosphere. In fact, the air masses arriving to Sapporo originate 399 from Siberia in spring and oceanic region passing over the Japanese Main Island in summer 400 (Fig. 1c, d). They should be enriched with the components derived from terrestrial and/or oceanic emissions. In addition to biogenic SOA and MSA (Fig. 4f,g), abundances of NH₄ 401 402 are relatively high in spring and summer (Pavuluri et al., 2015). Hence, secondary production of WION by the reactions of SOA (carbonyls) with NH₃ (NH₄⁺ ↔ NH₃) (Noziere et al., 2009) 403 404 should be accelerated during the warmer seasons in Northeast Asia. However, there is a need 405 to conduct the measurements of WION and source tracers at different locale in order to fully 406 understand the main source(s) of WION in atmospheric aerosols, which is a subject of future 407 research.

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3.5 Nitrogen isotope ratios of TN and WSTN: implications for source and chemical

410 aging

411 3.5.1 Seasonal variations

 $\delta^{15}N_{TN}$ ranged from 15.5 to 29.4‰ (average 21.9 ± 4.1‰) whereas $\delta^{15}N_{WSTN}$ ranged from 12.2 to 39.1‰ (25.8 ± 8.2‰) in Sapporo aerosols during the campaign period (n = 21). As shown in Fig. 2b, temporal trends of $\delta^{15}N_{TN}$ and $\delta^{15}N_{WSTN}$ were highly comparable. They both show peaks in summer with relatively high values in late autumn (Fig. 2b). Their averages, except for $\delta^{15}N_{TN}$ in autumn and winter, were significantly varied from season-to-season with higher $\delta^{15}N$ values in summer followed by spring, winter and autumn (Table 1). The seasonal trends of $\delta^{15}N_{TN}$ and $\delta^{15}N_{WSTN}$ suggest that aerosol N is significantly

influenced by season-specific source(s) and/or chemical aging of N species.

We also found that $\delta^{15}N_{WSTN}$ are higher than $\delta^{15}N_{TN}$ throughout the campaign, except 420 for few samples in autumn (Fig. 2b). An enrichment of ¹⁵N in WSTN over TN is more 421 422 significant in summer than other seasons (Table 1). It has been reported that the aerosol 423 particles collected over the controlled laboratory burns of the vegetation, which significantly contribute to WSTN, are enriched with ¹⁵N ranging from -1.3 to 13.1% (average 6.6%) 424 425 compared to the source vegetation (Turekian et al., 1998). Further, chemical aging of N 426 species, including gas-to-particle exchange (e.g., NH₄⁺ ↔ NH₃) reactions during long-range 427 transport causes the enrichment of ^{15}N in aerosol N (Pavuluri et al., 2010). In addition, $\delta^{15}N$ 428 of terrestrial plants, which significantly contribute to WION, is relatively low (e.g., -3.4 to 429 12.2‰ in plants from the San Francisco Bay area) (Cloern et al., 2002). Therefore, higher $\delta^{15}N_{WSTN}$ values than $\delta^{15}N_{TN}$ in Sapporo aerosols suggest that contributions of biomass 430 burning to WSTN and biological sources to WION (i.e. TN-WSTN) may be significant and 431 432 aerosol N has been subjected to significant chemical aging during long-range transport. The 433 enhanced aging (particularly NH₄⁺ ↔ NH₃) of nitrogenous aerosols under higher ambient temperature in summer may have caused further enrichment of ¹⁵N in WSTN (Pavuluri et al., 434 435 2010). As seen from Fig. 6, $\delta^{15}N_{TN}$ (and $\delta^{15}N_{WSTN}$) showed an inverse relation with the mass 436 437 fractions of NO₃-N and WSON in TN, but showed a linear relation with NH₄⁺-N/TN and 438 WION/TN during the study period. Their relations with NO₃-N/TN and NH₄⁺-N/TN are 439

fractions of NO_3^- -N and WSON in TN, but showed a linear relation with NH_4^+ -N/TN and WION/TN during the study period. Their relations with NO_3^- -N/TN and NH_4^+ -N/TN are statistically significant at 99% (p \leq 0.01) level but not in other two cases (Fig. 6). Such relations imply that the $\delta^{15}N_{TN}$ and $\delta^{15}N_{WSTN}$ are mainly controlled by their sources, chemical aging of nitrogenous species and abundances of NO_3^- -N and NH_4^+ -N. It is well recognized that NO_3^- originates mainly from fossil fuel combustion and biomass burning whereas NH_4^+ from biomass burning, agricultural activities and biological emissions (Delon et al., 2012; Hertel et al., 2012). In fact, NO_3^- is more abundant in mid autumn to early spring whereas

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NH₄⁺ maximize in mid spring to summer (Pavuluri et al., 2015). These results are consistent, with those of source tracers (Fig. 4), again suggesting that the contributions from biomass burning and fossil fuel combustion are important in autumn and winter whereas the biological particles in spring and biogenic emissions of gaseous species (hydrocarbons and NH₃) and

3.5.2 Comparison with literature values

subsequent secondary production in spring/summer are important.

Figure 7 compares the range (or mean) of $\delta^{15}N$ values reported for the particles emitted from point sources in mid-latitudes as well as those observed in atmospheric aerosols from different locale over the world including Sapporo aerosols. $\delta^{15}N_{TN}$ values observed in Sapporo aerosols are significantly higher than those (-19.4‰ to 15.4‰) reported for the particles emitted from point sources of fossil fuel combustion and waste incineration burning (Fig. 7). They are also higher than those reported in urban aerosols from Paris, France, where fossil fuel combustion was expected as a major source (Widory, 2007), and the marine aerosols over the western North Pacific, which are considered to be mainly derived from sea-to-air emissions (Miyazaki et al., 2011) (Fig. 7).

However, they are comparable to the higher ends of the $\delta^{15}N_{TN}$ reported in atmospheric

However, they are comparable to the higher ends of the $\delta^{15}N_{TN}$ reported in atmospheric aerosols from Jeju Island, Korea (Fig. 7), which were attributed to significant atmospheric processing of N species, emitted mainly from China, during long-range transport (Kundu et al., 2010b). Further the lower ends of $\delta^{15}N_{TN}$ in Sapporo aerosols are comparable to the higher ends of $\delta^{15}N_{TN}$ reported for the particles emitted from controlled burning of C_3 (range, 2.0% to 19.5%) and C_4 (9.8% to 22.7%) plant species in a laboratory study and those of atmospheric aerosols from Piracicaba and Amazon basin, Brazil, where biomass burning is a dominant source (Martinelli et al., 2002) (Fig. 7). In addition, they are well comparable to those reported in the forest aerosols from Rondônia, Brazil, where biomass burning was

expected as a dominant source (Kundu et al., 2010a), and urban aerosols from Chennai and Mumbai, India, which were considered to be mainly originated from biomass burning and subjected for aging (Aggarwal et al., 2013; Pavuluri et al., 2010) (Fig. 7).

On the other hand, lower values of $\delta^{15}N_{WSTN}$ observed in Sapporo aerosols are comparable to those reported in the forest aerosols from Sapporo experimental forest, Japan (Fig. 7). Although their seasonal patterns (winter minimum and summer maximum) are similar, the enrichment of ¹⁵N in Sapporo aerosols (Fig. 2b) is significantly higher than that (range of monthly averages, 9.0% to 26.0%) reported in the Sapporo forest aerosols (Miyazaki et al., 2014). Further, the range of $\delta^{15}N_{WSTN}$ values observed in Sapporo aerosols are highly comparable with $\delta^{15}N$ of aerosol NH_4^+ , which is much higher than that of gaseous NH_3 , in the atmosphere from the Maki monitoring station, Niigata, Japan (Fig. 7). The large difference in $\delta^{15}N$ between NH_3 and NH_4^+ has been attributed to gas-to-particle (equilibrium) exchange reactions in the atmosphere (Hayasaka et al., 2004) because under equilibrium conditions, particulate NH_4^+ is more enriched with NH_3^+ than gaseous NH_3^+ , where the isotopic enrichment factor ($\epsilon_{NH4+NH3}$) was estimated to be +33% (Heaton et al., 1997). Such comparisons suggest that aerosol N in Sapporo aerosols might have significantly originated from biogenic sources including biomass burning and subjected to chemical aging during long-range transport, although we do not preclude a contribution from fossil fuel combustion.

4 Summary and Conclusions

Water-soluble organic nitrogen (WSON) and water-insoluble organic nitrogen (WION) and N isotope ratios of total nitrogen ($\delta^{15}N_{TN}$) and water-soluble TN ($\delta^{15}N_{WSTN}$) were obtained for TSP aerosol samples collected from Sapporo, northern Japan during September 2009 and October 2010. WSON and WION ranged from 0.00 to 288 ng m⁻³ (average 89.7 ± 80.6 ng m⁻³) and from 0.00 to 454 ng m⁻³ (126 ± 117 ng m⁻³). Although their temporal variations did

not show clear seasonal trends, average WSON was found to be higher in winter followed by autumn, spring and summer whereas WION maximized in summer followed by spring with much lower concentration in autumn and winter. $\delta^{15}N_{TN}$ ranged from 15.5 to 29.4% (average 21.9 ± 4.1%) whereas $\delta^{15}N_{WSTN}$ ranged from 12.2 to 39.1% (25.8 ± 8.2%). Their seasonal variations showed high ratios in summer and late autumn. Comparisons of seasonal variations of WSON and WION with anthropogenic and biological source tracers together with air mass trajectories suggest that fossil fuel combustion and biomass burning are major sources of WSON whereas emissions of biological particles and secondary reactions of biogenic hydrocarbons (carbonyls) with NH₃ significantly contribute to WION in Northeast Asia. Seasonal variations of $\delta^{15}N_{TN}$ and $\delta^{15}N_{WSTN}$ and their relations to mass fractions of nitrogenous components in TN suggest that aerosol N in Sapporo aerosols is mainly originated from biogenic sources including biomass burning and subjected to chemical aging during long-range transport, although a contribution from fossil fuel combustion is important particularly in winter.

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Table 1. Seasonal and annual averages with standard deviation of concentrations of nitrogenous components and ¹⁵N isotope ratios of TN and WSTN in atmospheric aerosol (TSP) samples collected during 2 September 2009 and 5 October 2010 from Sapporo, northern Japan.

C	Autumn	Winter	Spring	Summer	Annual		
Component	(n = 8)	(n = 3)	(n = 5)	(n = 5)	(n = 19)		
Concentrations (ng m ⁻³)							
TN	790 ± 480	1456 ± 298	1116 ± 434	1023 ± 433	1074 ± 454		
WSTN	714 ± 496	1388 ± 247	986 ± 356	789 ± 351	950 ± 427		
NO_3-N	284 ± 120	362 ± 55.2	395 ± 182	73.8 ± 24.1	279 ± 172		
NH_4^+-N	338 ± 347	896 ± 97.2	511 ± 191	647 ± 310	576 ± 314		
ON	168 ± 77.1	198 ± 186	212 ± 221	302 ± 149	220 ± 157		
WSON	91.6 ± 73.6	130 ± 139	84.7 ± 99.5	67.8 ± 39.8	96.0 ± 81.9		
WION	77.5 ± 50.9	68.4 ± 51.3	130 ± 141	234 ± 144	125 ± 123		
Isotope ratios (%)							
$\delta^{15}N_{TN}$	18.91 ± 3.17	19.72 ± 3.56	23.42 ± 1.63	26.82 ± 1.95	22.47 ± 3.98		
$\delta^{15}N_{WSTN}$	19.17 ± 6.45	23.84 ± 4.13	26.82 ± 2.46	36.88 ± 2.46	26.71 ± 8.16		

Table 2. Concentrations of water-soluble organic nitrogen (WSON) and the mass fractions of WSON in water-soluble total nitrogen (WSTN) in Sapporo aerosols together with those in atmospheric aerosols from different sites in the world. ^aTotal of fine and coarse mode particles, ^bDry (biomass burning) season, ^cAsian outflow region.

Locale	Study Period	Aerosol	WSON	WSON/WSTN	Reference
		Size	(ng m ⁻³)	(%)	
Urban sites					
Sapporo, Japan	Sept 2009 to Oct 2010	TSP	89.7 ± 80.6	10.5 ± 7.5	This study
Davis, California	Aug 1997 to Jul 1998	$PM_{2.5}$	265 ± 190	23	Zhang et al. (2002)
Kofu, Japan	Aug 2009 to Jan 2013	$P{M_{10}}^a$	221	~14	Matsumoto et al. (2014)
Forest sites					
Rondônia, Brazil	Mar-May 1999	PM_{10}	49 ± 64.4	45 ± 35	Mace et al. (2003a)
Rondônia, Brazil	Sept-Oct 1999b	PM_{10}	854 ± 938	43 ± 31	Mace et al. (2003a)
Sapporo, Japan	Jun 2009 to Dec 2011	TSP	157 ± 127	20 ± 11	Miyazaki et al. (2014)
Fujiyoshida, Japan	Aug 2009 to Jan 2013	$P{M_{10}}^a$	101	~11	Matsumoto et al. (2014)
Coastal sites					
Erdemli, Turkey	Mar-May 2000	TSP	406 ± 588	${\sim}26\pm28$	Mace et al., (2003c)
Crete, Greece	2005-2006	$PM_{10}^{\ a}$	~239	13	Violaki & Mihalopoulos (2010)
Qingdao, China	Mar-Apr 2006	TSP	2520 ± 1760	~20	Shi et al. (2010)
Oceanic sites					
Yellow Sea ^c	Mar 2005	TSP	2860 ± 1780	~17	Shi et al. (2010)
Yellow Sea ^c	Apr 2006	TSP	1220 ± 812	~17	Shi et al. (2010)
South China Seac	Apr 2005	TSP	910 ± 280	~34	Shi et al. (2010)
East China Seac	Sept-Oct 2002	TSP^a	756 ± 504	~24	Nakamura et al. (2006)
East China Sea and	M 2004	TCD	224 + 266	10	N-1
western North Pacific ^c	Mar 2004	TSP ^a	224 ± 266	~10	Nakamura et al. (2006)
Cape Grim, Australia	Nov-Dec 2000	TSP	50.4 ± 79.8	~21	Mace et al. (2003b)
Cape Grim, Australia	Nov-Dec 2000	TSP	13.0 ± 13.3	~25	Mace et al. (2003b)
Oahu, Hawaii	Jul-Aug 1998	PM_{10}	46.2 ± 28	~33	Cornell et al. (2001)
Western North Pacific	Aug-Sept 2008	TSP	~17.5		Miyazaki et al. (2011)

Table 3. Spearman's rank correlation of WSON and WION with source tracers in Sapporo
 aerosols during the study period (n = 21). Significant correlation coefficients at 0.05 level are
 shown in bold.

Source tracer	Correlation coefficient (pr)		
Source tracer	WSON	WION	
Hopanes	0.48	-0. 50	
Levoglucosan	0.18	-0.57	
Sucrose	-0.02	0.38	
Isoprene-SOA tracers	-0.12	0.50	
α-Pinene-SOA tracers	0.21	0.14	
β-Caryophylinic acid	0.06	0.30	
MSA ⁻	0.09	0.59	

775 Figure captions 776 777 Fig. 1. Plots of 10-day backward air mass trajectories arriving over Sapporo at 500 AGL. 778 779 Fig. 2. Seasonal variations in (a) concentrations of TN and WSTN and (b) N isotope ratios of TN ($\delta^{15}N_{TN}$) and WSTN ($\delta^{15}N_{WSTN}$). 780 781 782 Fig. 3. Linear relations of IN (sum of NO₃-N and NH₄+-N) to (a) WSTN and (b) TN. 783 784 Fig. 4. Seasonal variations of (a) concentrations of WSON and WION, (b) mass fractions of 785 WSON and WION in TN, and (c-e) concentrations of organic molecular tracers and MSA in 786 Sapporo aerosols. Concentrations of hopanes are sum of the concentrations of C₂₇-C₃₂ 787 hopanoid hydrocarbons. Concentrations of isoprene-SOA tracers are sum of the 788 concentrations of 2-methylglyceric acid, cis-2-methyl-1,3,4-trihydroxy-1-butene, 789 3-methyl-2,3,4-trihydroxy-1-butene, trans-2-methyl-1,3,4-trihydroxy-1-butene, 790 2-methylthreitol and 2-methylerythritol whereas concentrations of α -pinene-SOA tracers are 791 sum of the concentrations of 3-hydroxyglutaric, pinonic, pinic and 792 3-methyl-1,2,3-butanetricarboxylic acids. The data of organic molecular tracers are from 793 Pavuluri et al. (2013) and MSA⁻ is from Pavuluri et al. (2015). 794 795 Fig. 5. Linear relations of WSON with (a) hopanes (C₂₇-C₃₂ hopanoid hydrocarbons) and (b) 796 levoglucosan during campaign period (n = 21) and with (c) sucrose, (d) α -pinene-SOA 797 tracers (sum of 3-hydroxyglutaric, pinonic, pinic and 3-methyl-1,2,3-butanetricarboxylic 798 acids), and (e) MSA^{-} in spring and summer (n = 10) in Sapporo aerosols.

800 Fig. 6. Scatter plots between $\delta^{15}N_{TN}$ (and $\delta^{15}N_{WSTN}$) and mass fractions of nitrogenous 801 components (NO₃-N, NH₄+N, WSON and WION) in TN in Sapporo aerosols. 802 803 Fig. 7. Range or mean N isotope ratios in the particles emitted from point sources, source substance, and atmospheric aerosols from different sites in the world. *, **, and *** show 804 gaseous NH₃, particulate NH₄⁺ and WSTN, respectively. ^aWidory (2007); ^bTurekian et al. 805 806 (1998); ^{c,g}Kundu et al. (2010a, 2010b); ^dMartinelli et al. (2002); ^ePavuluri et al. (2010); ^fAggarwal et al. (2013); ^{h,j}Miyazaki et al. (2011, 2014); ⁱHayasaka et al. (2004); ^kThis study. 807 808

Figure 1.

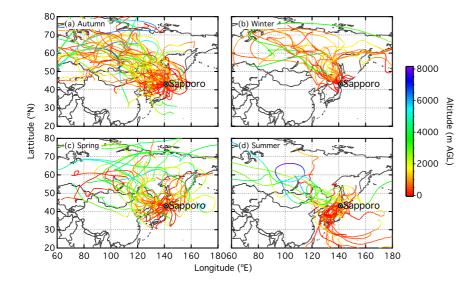


Figure 2.

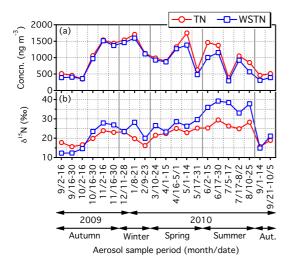


Figure 3.

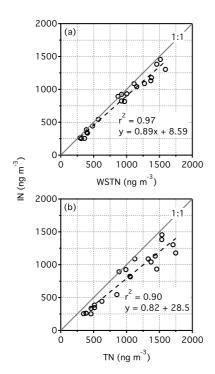
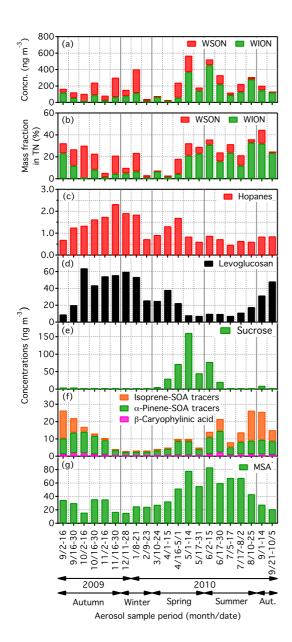
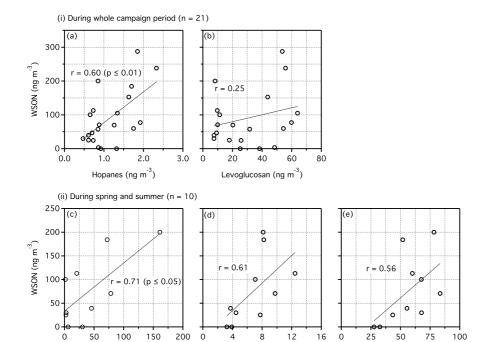


Figure 4.



821 **Figure 5.**

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 $\alpha\text{-Pinene SOA tracers (ng m}^{-3}\text{)}$

 MSA^{-} (ng m $^{-3}$)

Sucrose (ng m⁻³)

Figure 6.

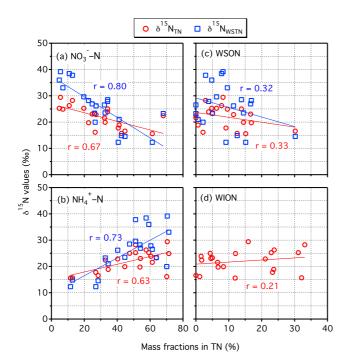


Figure 7.

